

High contribution of anthropogenic combustion sources to atmospheric inorganic reactive nitrogen in South China evidenced by isotopes

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Abstract: Due to the intense release of reactive nitrogen (Nr) from anthropogenic activity, the source layout of atmospheric nitrogen aerosol has changed. The inorganic nitrogen (NH_4^+ and NO_3^-) was essential part of atmospheric nitrogen aerosol and accounted for 69%. To comprehensively clarify the level, sources, and environmental fate of NH_4^+ and NO_3^- , their concentrations and stable isotopes ($\delta^{15}\text{N}$) in fine particulate matters ($\text{PM}_{2.5}$) were measured in a subtropical megacity of South China. N- NH_4^+ and N- NO_3^- contributed 45.8% and 23.2% to total nitrogen (TN), respectively. The source contributions of NH_4^+ and NO_3^- were estimated by $\delta^{15}\text{N}$, which suggested that anthropogenic combustion activities including coal combustion, biomass burning, and vehicles were dominant sources. Especially, biomass burning was the predominant source of NH_4^+ (27.9%). Whereas, coal combustion was the dominant source of NO_3^- (40.4%). This study emphasized the substantial impacts of human activities on inorganic Nr. With the rapid development of industry and transportation, nitrogen emissions will be even higher. The promotion of clean energy and efficient use of biomass would help reduce nitrogen emissions and alleviate air pollution.

30 1. Introduction

31 Nitrogenous aerosols are ubiquitous in environment and play an important role as
32 nutrients in ecosystems(Bhattarai et al., 2019). With the massive combustion of fossil fuels and
33 the development of livestock, the proportion of TN in particulate matter (PM) ranges from 1.2%
34 to 17.0% and has shown a rapid increase in the last few decades(Bhattarai et al., 2019;
35 Galloway et al., 2004; Holland et al., 1999). Mostly nitrogenous aerosols formed from
36 atmospheric Nr will be deposited into terrestrial and aquatic ecosystems(Huang et al., 2015).
37 Excessive external nitrogen deposition accelerates nitrogen loss in soil, decreases species
38 diversity, disturbs terrestrial ecosystems, and leads to eutrophication in aquatic
39 ecosystems(Breemen, 2002; Wedin and Tilman, 1996; Yang et al., 2015). Furthermore,
40 nitrogenous aerosols have adverse impacts on the climate, air quality, and human
41 health(Bhattarai et al., 2019; Song et al., 2021).

42 N-NO_3^- and N-NH_4^+ as inorganic Nr are dominant species in the deposition of
43 nitrogen(Zhu et al., 2015). N-NH_4^+ was the highest in nitrogen deposition, and NH_4^+ was
44 gradually considered to be an important component of secondary inorganic aerosols (SIA)(Sun
45 et al., 2021). NH_3 , the precursor of NH_4^+ , is a vital atmospheric alkaline gas, which can
46 participate in nucleation to promote new particles generation, and can react with acid gas to
47 produce ammonium sulfate and ammonium nitrate(Dunne et al., 2016; Fu et al., 2017). The
48 excessive NH_3 emission from anthropogenic sources will partially offset the benefits of
49 reducing SO_2 and NO_x and trigger urban haze in China(Sun et al., 2021; Meng et al., 2018;
50 Pan et al., 2018a). In many urban environments, NO_3^- has replaced sulfate as the component
51 with the highest proportion in SIA. NO_x , precursors of NO_3^- , are also closely related to the
52 formation of atmospheric oxidants and exert important effects on atmospheric oxidation. In
53 addition, NH_4NO_3 in PM plays an increasingly important role in promoting the formation of
54 sulfate and organic matter, and has profound effect on the physical and chemical properties of
55 PM(Liu et al., 2021; Liu et al., 2020; Hodas et al., 2014). Therefore, to mitigate nitrogen
56 deposition and air pollution, the control of NH_4^+ (NH_3) and NO_3^- (NO_x) should not be neglected.

57 Considerable efforts have been made to comprehensively understand the budget of
58 atmospheric NH_4^+ and NO_3^- . $\delta^{15}\text{N}$ is effective to quantify sources contribution of nitrogenous

59 species(Elliott et al., 2007). The anthropogenic combustion sources (combustion of coal,
60 biomass, and gasoline) play a key role in the emission of NO_3^- (NO_x) in many regions of China
61 suggested by $\delta^{15}\text{N}$ (Zong et al., 2020), which also have large effects on NH_3 (Chen et al., 2022b).
62 NH_3 is released by agricultural sources (agricultural activity and livestock) and non-agricultural
63 sources (fossil fuel combustion and vehicle)(Bhattarai et al., 2019). A previous study showed
64 that agricultural source was the dominant source (80%-90%) of NH_3 in China(Kang et al.,
65 2016). However, NH_3 emissions from agricultural source have been reduced due to intensive
66 farming and efficient fertilization(Wang et al., 2022). The incomplete burning of biomass leads
67 to massive NH_3 emissions and is gradually to be the second largest non-agricultural source of
68 NH_3 (Yu et al., 2020), which may be responsible for the lag of the decline in air pollutants
69 deposition behind the reduction in emission of precursors(Zhao et al., 2022b). Biomass burning
70 in the suburbs also has a potential impact on urban NH_3 (Xiao et al., 2020). As for urban NH_3 ,
71 combustion sources (including coal combustion, vehicles emission, and biomass burning) were
72 gradually becoming dominant sources in recent years verified by $\delta^{15}\text{N-NH}_x$ ($\text{NH}_3+\text{NH}_4^+$)(Xiao
73 et al., 2020; Pan et al., 2018b). In addition, the super clean emission of coal-fired power plant
74 and strict emission standards of vehicles will change the source layout of NH_4^+ and NO_3^- .
75 Selective catalytic reduction technology equipped with vehicles and industrial boiler reduces
76 NO_x but increases NH_3 emissions(Meng et al., 2017; Pan et al., 2016). The occurrence of haze
77 in North China was closely related to NH_3 emissions from combustion sources(Pan et al., 2018a;
78 Pan et al., 2018b). NH_4^+ and NO_3^- are the main components of SIA and play a vital role in the
79 formation of secondary aerosol(Meng et al., 2017), so it is necessary to revisit their sources.

80 Nr emissions from densely populated subtropical areas increased rapidly with the high
81 development of industry and transportation(Wang et al., 2013). Guangzhou is the core megacity
82 in the South subtropical region of China, where the atmospheric environment is complex and
83 the atmospheric oxidation level is high(Tan et al., 2019). The high emissions of inorganic
84 nitrogen from anthropogenic combustion sources have serious and profound impacts on the
85 environment. In this study, we aimed to comprehensively clarify the level of inorganic Nr and
86 revisit the source layout of atmospheric inorganic Nr .

87 2. Experimental and theoretical methods

88 2.1. Sampling and Chemical concentration analysis

89 PM_{2.5} samples (n=66) were collected from May 2017 to June 2018 in Guangzhou
90 (23.13°N, 113.27°E). Details of sample collection can be found in our previous study(Jiang et
91 al., 2021a). The chemical components including water-soluble ions (i.e., NH₄⁺, K⁺, Na⁺, Ca²⁺,
92 Mg²⁺, Cl⁻, NO₃⁻, and SO₄²⁻), organic carbon (OC), element carbon (EC), and organic molecular
93 markers (e.g., levoglucosan) were analyzed in our previous studies (SI Text S1)(Jiang et al.,
94 2021a; Jiang et al., 2021b). Moreover, meteorological parameters (temperature, relative
95 humidity (RH), atmosphere pressure, and wind speed) and the concentration of trace gases (CO,
96 SO₂, NO, NO₂, and O₃) were acquired by online instruments (details shown in SI Text S1). A
97 circular punch (r=1cm) of the sample filter was wrapped in a tin boat and then measured in an
98 elemental analyzer to determine the concentrations of TN.

99 2.2. Isotope analysis

100 The δ¹⁵N-NO₃⁻ and δ¹⁸O-NO₃⁻ values in PM_{2.5} were analyzed by methods of nitrous oxide
101 (N₂O), which was described in previous study in detail(Zong et al., 2017). Briefly, NO₃⁻ was
102 reduced to NO₂⁻ using cadmium powder and imidazole solution, and N₂O was made by adding
103 NaN₃ to NO₂⁻ solution. The production of 75nmol N₂O gas was needed to measure. The N₂O
104 gas produced by above processes was measured by MAT253 stable isotope mass spectrometer.
105 The values of δ¹⁸O and δ¹⁵N were expressed in per mil (‰) shown in Eq. (1) and (2), relative
106 to the international oxygen and nitrogen isotope standard, respectively.

$$107 \delta^{15}\text{N} = \left[\frac{(^{15}\text{N}/^{14}\text{N})_{\text{sample}}}{(^{15}\text{N}/^{14}\text{N})_{\text{standard}}} - 1 \right] * 1000 \quad (1)$$

$$108 \delta^{18}\text{O} = \left[\frac{(^{18}\text{O}/^{16}\text{O})_{\text{sample}}}{(^{18}\text{O}/^{16}\text{O})_{\text{standard}}} - 1 \right] * 1000 \quad (2)$$

109 The δ¹⁵N-NH₄⁺ was measured by methods of hypobromite oxidation coupled with
110 reduction of hydroxylamine hydrochloride(Sun et al., 2021). Briefly, NH₄⁺ was oxidated to
111 NO₂⁻ using alkaline hypobromite (BrO⁻), and N₂O was made by adding sodium arsenite and
112 hydrochloric acid to NO₂⁻ solution. The production of 120 nmol N₂O gas was needed to
113 measure. The N₂O gas produced by above processes was measured by MAT253 stable isotope
114 mass spectrometer. The values of δ¹⁵N were expressed in per mil (‰), Eq. (1). To ensure the

115 stability of the instrument, standard samples were tested for every ten samples. The standard
116 deviation of replicates was generally less than 0.4‰, 0.8‰, and 0.5‰ for $\delta^{15}\text{N-NO}_3^-$, $\delta^{18}\text{O-}$
117 NO_3^- , and $\delta^{15}\text{N-NH}_4^+$, respectively. The instrumental values of $\delta^{15}\text{N-NO}_3^-$ and $\delta^{18}\text{O-NO}_3^-$ were
118 corrected by multi-point correction ($\delta^{18}\text{O } r^2=0.99$, $\delta^{15}\text{N } r^2=0.999$) based on international
119 standards (IAEA-NO-3, USGS32, USGS34, and USGS35). The measured values of $\delta^{15}\text{N-NH}_4^+$
120 were also corrected by multi-point correction ($r^2=0.999$) based on international standards
121 (IAEA-N1, USGS25, and USGS26). In addition, ^7Be and ^{210}Pb were acquired and details were
122 shown in [SI Text S1](#).

123 **2.3. IsoSource and Bayesian mixing model**

124 **IsoSource model.** IsoSource model was released by Environmental Protection Agency
125 (EPA), could calculate ranges of source contributions to a mixture based on conservation of
126 isotopic mass when number of sources is too large to permit a unique solution and provide the
127 distribution of source proportions (Phillips et al., 2005). IsoSource model coupled with $\delta^{15}\text{N-}$
128 NH_3 of atmospheric initial and potential sources (shown in [Table 1](#)) were applied to quantify
129 the contribution of various sources to NH_3 . Nitrogen fertilizers application, livestock, human
130 waste, biomass burning, coal combustion, and vehicles were considered as sources of NH_3 in
131 this study, details shown in [SI Text S2](#). Atmospheric initial $\delta^{15}\text{N-NH}_3$ was calculated by
132 following Eq. (3).

$$133 \quad \delta^{15}\text{N-NH}_{3\text{-initial}} = \delta^{15}\text{N-NH}_4^+ - \varepsilon(\text{NH}_4^+ - \text{NH}_3) \times (1 - f) \quad (3)$$

134 Where, $\delta^{15}\text{N-NH}_4^+$ and $\delta^{15}\text{N-NH}_{3\text{-initial}}$ represent the $\delta^{15}\text{N}$ of particulate NH_4^+ and
135 atmospheric initial NH_3 , respectively. $\varepsilon(\text{NH}_4^+ - \text{NH}_3)$ represents the isotope fractionation factor
136 in the gaseous NH_3 conversion to particulate NH_4^+ in the atmosphere. The f value represents
137 the proportion of the initial NH_3 converted to NH_4^+ , referring to NH_3 and NH_4^+ observed in
138 Guangzhou (Liao et al., 2014).

139 The $\varepsilon(\text{NH}_4^+ - \text{NH}_3)$ value is temperature dependent (Huang et al., 2019), which can be
140 deduced from (Urey, 1947), as shown in Eq. (4). The atmospheric average temperature was
141 24.5°C in our sampling period, and the corresponding $\varepsilon(\text{NH}_4^+ - \text{NH}_3)$ value was 34.2‰
142 calculated by Eq. (4). In addition, the $\varepsilon(\text{NH}_4^+ - \text{NH}_3)$ in Guangzhou was estimated to be 32.4‰
143 according to Eq. (8). Eq. (8) was deduced by Eq. (5-7). According to Eq. (8), a linear fitting

144 equation was observed between $f\text{NH}_4^+$ and $\delta^{15}\text{N-NH}_4^+$ (**Fig. S1**), and the absolute value of the
 145 slope (32.4‰) was equal to $\varepsilon(\text{NH}_4^+-\text{NH}_3)$. The $\varepsilon(\text{NH}_4^+-\text{NH}_3)$ average of the two methods (34.2‰
 146 and 32.4‰) was 33.3‰ and approximated to the experimental isotope enrichment factor
 147 (33‰)(Heaton et al., 1997). Therefore, 33‰ was used for deducing the $\delta^{15}\text{N}$ of the initial NH_3 .

$$148 \quad \varepsilon_{(\text{NH}_4^+-\text{NH}_3)} = 12.4678 * \frac{1000}{T+273.15} - 7.6694 \quad (4)$$

$$149 \quad \delta^{15}\text{N-NH}_4^+ - \delta^{15}\text{N-NH}_3 = \varepsilon_{(\text{NH}_4^+-\text{NH}_3)} \quad (5)$$

$$150 \quad f\text{NH}_4^+ + f\text{NH}_3 = 1 \quad (6)$$

$$151 \quad \delta^{15}\text{N-NH}_4^+ * f\text{NH}_4^+ + (\delta^{15}\text{N-NH}_4^+ - \varepsilon_{(\text{NH}_4^+-\text{NH}_3)}) * (1 - f\text{NH}_4^+) = \delta^{15}\text{N} \quad (7)$$

$$152 \quad \delta^{15}\text{N-NH}_4^+ = -\varepsilon_{(\text{NH}_4^+-\text{NH}_3)} * f\text{NH}_4^+ + (\delta^{15}\text{N} + \varepsilon_{(\text{NH}_4^+-\text{NH}_3)}) \quad (8)$$

153 Where, T represents the atmospheric temperature (°C). $\delta^{15}\text{N-NH}_4^+$ and $\delta^{15}\text{N-NH}_3$
 154 represent the $\delta^{15}\text{N}$ of particulate NH_4^+ and atmospheric NH_3 , respectively. $\delta^{15}\text{N}$ represents the
 155 sum of $\delta^{15}\text{N-NH}_4^+$ and $\delta^{15}\text{N-NH}_3$. $f\text{NH}_3$ and $f\text{NH}_4^+$ represent the proportion of atmospheric
 156 NH_3 and particulate NH_4^+ , respectively.

157 **Bayesian mixing model.** $\delta^{15}\text{N}$ were used for tracing source based on conservation of
 158 isotopic mass. Bayesian mixing model improved upon linear mixing models by explicitly
 159 considering uncertainty in prior information and isotopic equilibrium fractionation. Recently,
 160 Bayesian mixing model was applied to trace the sources of atmospheric pollutants(Zong et al.,
 161 2017; Zong et al., 2020). The model coupled with $\delta^{15}\text{N-NO}_3^-$ and $\delta^{18}\text{O-NO}_3^-$ were used to
 162 identify the formation process and quantify the sources contribution of NO_3^- .

163 In Central Pearl River Delta (PRD), NO_3^- formed through $\cdot\text{OH}$ and N_2O_5 pathways
 164 contributed to 94% simulated by CAMQ model (Qu et al., 2021). In this study, only $\cdot\text{OH}$ and
 165 N_2O_5 formation pathways were considered. Details of NO_3^- formation pathway were also
 166 shown in **SI Text S2**. The atmospheric $\delta^{18}\text{O-NO}_3^-$ can be expressed by Eq. (9). The $[\delta^{18}\text{O-}$
 167 $\text{HNO}_3]_{\text{OH}}$ can be further expressed by Eq. (10) assuming no kinetic isotope fractionation
 168 (Walters and Michalski, 2016). And $[\delta^{18}\text{O-HNO}_3]_{\text{H}_2\text{O}}$ can be estimated by Eq. (11) (Walters and
 169 Michalski, 2016). The $\delta^{18}\text{O}$ values in tropospheric H_2O , NO_x , O_3 , and OH were within a certain
 170 range. The tropospheric $\delta^{18}\text{O-H}_2\text{O}$, $\delta^{18}\text{O-NO}_x$, $\delta^{18}\text{O-O}_3$, and $\delta^{18}\text{O-OH}$ ranged from -25‰ to
 171 0‰(Baskaran et al., 2011; Walters and Michalski, 2016), 112‰ to 122‰ (Michalski et al.,

172 2014; Walters and Michalski, 2016), 90‰ to 122‰, and -15‰ to 0‰, respectively (Fang et al.,
173 2011; Johnston and Thiemens, 1997). Therefore, the γ (the contribution of $\cdot\text{OH}$ formation
174 pathway) can be estimated by $f\text{NO}_2$ and oxygen isotope fractionation i.e., $\alpha_{\text{NO}_2/\text{NO}}$, $\alpha_{\text{OH}/\text{H}_2\text{O}}$,
175 and $\alpha_{\text{N}_2\text{O}_5/\text{NO}_2}$. The oxygen isotope fractionations are temperature dependent and can be
176 estimated by Eq. (13) and **Table S1**. The $f\text{NO}_2$ varied from 0.20 to 0.95 (Zong et al., 2017;
177 Walters et al., 2016). Based on $\delta^{18}\text{O}-\text{NO}_3^-$, $\delta^{18}\text{O}-\text{H}_2\text{O}$, $\delta^{18}\text{O}-\text{NO}_x$, $\delta^{18}\text{O}-\text{O}_3$, and temperature
178 (Eq. (9-13)), γ (maximum γ and minimum γ) was estimated by Monte Carlo simulation nested
179 in Bayesian mixing model (Zong et al., 2017). Assuming no kinetic isotope fractionation, the
180 nitrogen isotope fractionation value in the formation process of NO_3^- (εN) was calculated by
181 Eq. (13-16) combined with γ and temperature (Zong et al., 2017; Walters and Michalski, 2016;
182 Walters et al., 2016). The εN value in our sampling period was $5.1 \pm 2.5\%$, which was
183 comparable to that in Beijing (average 6.5‰) (Fan et al., 2020). The contributions of different
184 sources to atmospheric NO_x were quantified by Bayesian mixing model coupled with εN , $\delta^{15}\text{N}-$
185 atmospheric- NO_3^- , and $\delta^{15}\text{N}-\text{NO}_x$ endmembers shown in **Table 1**. We considered coal
186 combustion, mobile traffic sources, biomass burning, and soil microbial process as dominant
187 atmospheric NO_x sources in Guangzhou, details shown in **SI Text S2**. The specific details of
188 Bayesian mixing model were reported by our previous studies (Zong et al., 2017; Zong et al.,
189 2020).

$$190 \delta^{18}\text{O}-\text{NO}_3^- = \gamma \times [\delta^{18}\text{O}-\text{NO}_3^-]_{\text{OH}} + (1 - \gamma) \times [\delta^{18}\text{O}-\text{NO}_3^-]_{\text{H}_2\text{O}} = \gamma \times [\delta^{18}\text{O}-\text{HNO}_3]_{\text{OH}} +$$

$$191 (1 - \gamma) \times [\delta^{18}\text{O}-\text{HNO}_3]_{\text{H}_2\text{O}} \quad (9)$$

$$192 [\delta^{18}\text{O}-\text{HNO}_3]_{\text{OH}} = \frac{2}{3} [(\delta^{18}\text{O}-\text{NO}_2)]_{\text{OH}} + \frac{1}{3} [\delta^{18}\text{O}-\text{OH}]_{\text{OH}} = \frac{2}{3} \left[\frac{1000 \times ({}^{18}\alpha_{\text{NO}_2/\text{NO}} - 1)(1 - f_{\text{NO}_2})}{(1 - f_{\text{NO}_2}) + ({}^{18}\alpha_{\text{NO}_2/\text{NO}} \times f_{\text{NO}_2})} + \right.$$

$$193 \left. [\delta^{18}\text{O}-\text{NO}_x] \right] + \frac{1}{3} [(\delta^{18}\text{O}-\text{H}_2\text{O}) + 1000 \times ({}^{18}\alpha_{\text{OH}/\text{H}_2\text{O}} - 1)] \quad (10)$$

$$194 [\delta^{18}\text{O}-\text{HNO}_3]_{\text{H}_2\text{O}} = \frac{5}{6} (\delta^{18}\text{O}-\text{N}_2\text{O}_5) + \frac{1}{6} (\delta^{18}\text{O}-\text{H}_2\text{O}) \quad (11)$$

$$195 \delta^{18}\text{O}-\text{N}_2\text{O}_5 = \delta^{18}\text{O}-\text{NO}_2 + 1000 \times ({}^{18}\alpha_{\text{N}_2\text{O}_5/\text{NO}_2} - 1) \quad (12)$$

$$196 1000({}^m\alpha_{\text{X}/\text{Y}} - 1) = \frac{\text{A}}{\text{T}^4} \times 10^{10} + \frac{\text{B}}{\text{T}^3} \times 10^8 + \frac{\text{C}}{\text{T}^2} \times 10^6 + \frac{\text{D}}{\text{T}} \times 10^4 \quad (13)$$

$$197 \varepsilon\text{N} = \gamma \times \varepsilon(\delta^{15}\text{N}-\text{NO}_3^-)_{\text{OH}} + (1 - \gamma) \times \varepsilon(\delta^{15}\text{N}-\text{NO}_3^-)_{\text{H}_2\text{O}}$$

$$= \gamma \times \varepsilon(\delta^{15}\text{N}-\text{HNO}_3)_{\text{OH}} + (1 - \gamma) \times \varepsilon(\delta^{15}\text{N}-\text{HNO}_3)_{\text{H}_2\text{O}} \quad (14)$$

$$\varepsilon(\delta^{15}\text{N}-\text{HNO}_3)_{\text{OH}} = \varepsilon(\delta^{15}\text{N}-\text{NO}_2)_{\text{OH}} = 1000 \times \left[\frac{(^{15}\alpha_{\text{NO}_2/\text{NO}}-1)(1-f_{\text{NO}_2})}{(1-f_{\text{NO}_2})+(^{15}\alpha_{\text{NO}_2/\text{NO}} \times f_{\text{NO}_2})} \right] \quad (15)$$

$$\varepsilon(\delta^{15}\text{N}-\text{HNO}_3)_{\text{H}_2\text{O}} = \varepsilon(\delta^{15}\text{N}-\text{N}_2\text{O}_5)_{\text{H}_2\text{O}} = 1000 \times (^{15}\alpha_{\text{N}_2\text{O}_5/\text{NO}_2} - 1) \quad (16)$$

Where, γ is the contribution of $\cdot\text{OH}$ formation pathway to NO_3^- , εN is the nitrogen isotope fractionation value. f_{NO_2} is the fraction of NO_2 in the total NO_x . $^{18}\alpha_{\text{NO}_2/\text{NO}}$, $^{18}\alpha_{\text{OH}/\text{H}_2\text{O}}$, $^{18}\alpha_{\text{N}_2\text{O}_5/\text{NO}_2}$ are the oxygen isotope equilibrium fractionation factors between NO_2 and NO , $\cdot\text{OH}$ and H_2O , N_2O_5 and NO_2 , respectively. $^{15}\alpha_{\text{NO}_2/\text{NO}}$ and $^{15}\alpha_{\text{N}_2\text{O}_5/\text{NO}_2}$ are the nitrogen isotope equilibrium fractionation factor between NO_2 and NO , N_2O_5 and NO_2 , respectively.

Table 1. The estimation of $\delta^{15}\text{N}-\text{NH}_3$ and $\delta^{15}\text{N}-\text{NO}_x$ from various sources.

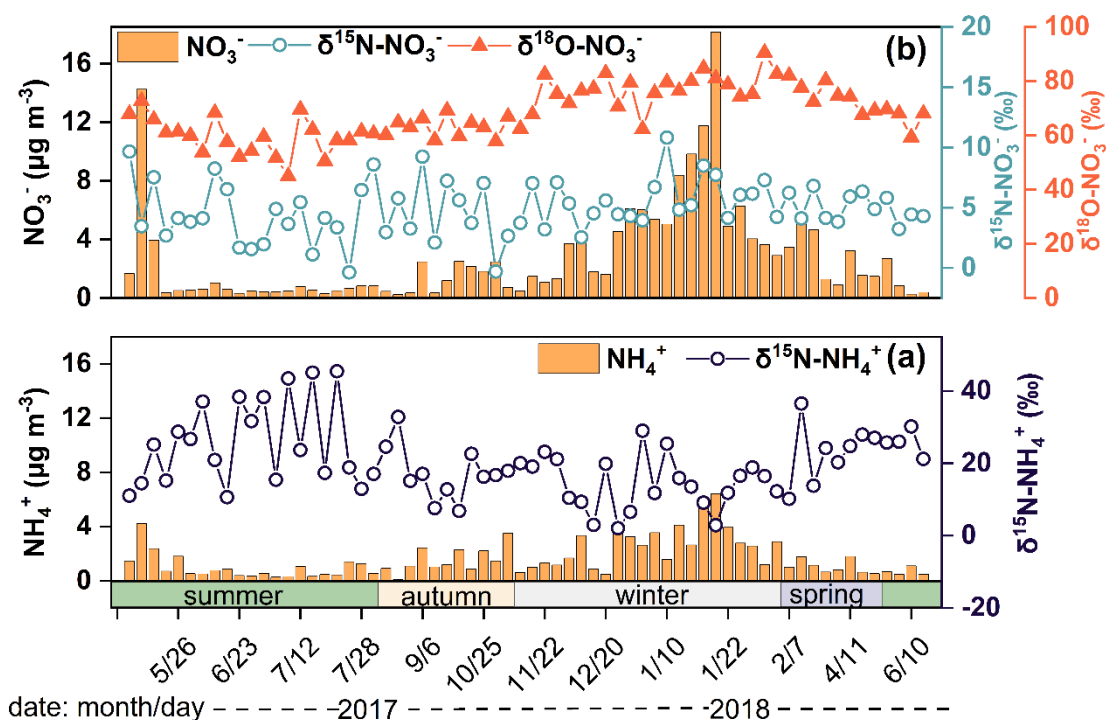
Source	$\delta^{15}\text{N}-\text{NH}_3(\text{‰})$	References
Biomass burning	17.5±7.8	(Kawashima and Kurahashi, 2011; Xiao et al., 2020)
Coal combustion	-2.5±6.4	(Felix et al., 2013; Pan et al., 2016)
Urban traffic	6.6±2.1	(Walters et al., 2020)
Fertilizer	-28.3±5.8	(Bhattarai et al., 2021; Chang et al., 2016; Felix et al., 2013; Bhattarai et al., 2020)
Livestock	-18.3±7.7	(Bhattarai et al., 2021; Chang et al., 2016; Felix et al., 2013; Bhattarai et al., 2020)
Urban waste	-22.8±3.6	(Bhattarai et al., 2021; Chang et al., 2016)
Source	$\delta^{15}\text{N}-\text{NO}_x(\text{‰})$	References
Biomass burning	1.04±4.13	(Zong et al., 2017; Fibiger and Hastings, 2016; Zong et al., 2022)
Coal combustion	13.72±4.57	(Zong et al., 2017; Felix et al., 2015; Felix et al., 2012)
Mobile source	-7.25±7.80	(Zong et al., 2017; Walters et al., 2015)
Soil microbial process	-33.77±12.16	(Zong et al., 2017; Felix and Elliott, 2013)

3. Results and discussion

3.1. Concentration and seasonal variation of NH_4^+ and NO_3^-

The concentration of NH_4^+ and NO_3^- in $\text{PM}_{2.5}$ was $1.6 \pm 1.3 \mu\text{g m}^{-3}$ and $2.8 \pm 3.4 \mu\text{g m}^{-3}$, contributed 18.7% and 32.6% to SIA. The concentration of $\text{N}-\text{NH}_4^+$ and $\text{N}-\text{NO}_3^-$ was $1.2 \pm 1.0 \mu\text{g m}^{-3}$ and $0.6 \pm 0.8 \mu\text{g m}^{-3}$, contributed 45.8% and 23.2% to TN, respectively; thus, NH_4^+ and NO_3^- were essential part of nitrogen aerosols. NH_4^+ and NO_3^- showed similar seasonal variations with higher concentrations in winter than in summer (**Fig. 1**). During winter the air mass was often dry and cold with low wind speed, which meant the decrease of the atmospheric self-purification capability. In addition, primary combustion sources related to fossil fuel and biomass burning always showed significant increase in North China in winter, which greatly

217 increased the concentration of atmospheric pollutants in Guangzhou by long-range
 218 transportation. However, during summer, the air mass from sea was relatively clean with high
 219 wind speed facilitating the diffusion of pollutants. Moreover, high temperature in summer was
 220 conducive to the decomposition of NH_4NO_3 (Song et al., 2008). Thus, the levels of NH_4^+ and
 221 NO_3^- were lower in summer. In addition, concentrations of NH_4^+ and NO_3^- in our study, were
 222 lower than North China [Beijing(Wu et al., 2019; Fan et al., 2022), Tianjin(Xiang et al., 2022),
 223 Shijiazhuang(Xiang et al., 2022), and Harbin(Sun et al., 2021)], East China [Nanchang(Xiao
 224 et al., 2020)], and Central China [Wuhan and Changsha(Xiao et al., 2020; Zong et al., 2020)],
 225 suggested the level of air pollution in Guangzhou has been alleviated to a certain extent.
 226 Therefore, it is necessary to conduct comprehensive study on the emission sources of NH_4^+ and
 227 NO_3^- to take more effective measures to mitigate air pollution.
 228



229
 230 **Figure 1.** The concentration and $\delta^{15}\text{N}$ of NH_4^+ (a) and concentration, $\delta^{15}\text{N}$, and $\delta^{18}\text{O}$ of NO_3^-
 231 (b).

232 3.2. Characteristic and seasonal variation in $\delta^{15}\text{N-NH}_4^+$ and source apportionment of 233 NH_4^+

234 The $\delta^{15}\text{N-NH}_4^+$ values over Guangzhou ranged from 2.1‰ to 45.5‰, with an annual mean
235 of 20.2 ± 10.1 ‰. In our study, the $\delta^{15}\text{N-NH}_4^+$ values were comparable to those at suburban sites
236 (Fig. S2) such as sites in Japan (22.1 ± 8.3 ‰, 16.1 ± 6.6 ‰)(Kawashima and Kurahashi, 2011)
237 and Korea (Jeju Island, 17.4 ± 4.9 ‰)(Kundu et al., 2010) but heavier than those in polluted
238 regions, such as Guangzhou during summer haze(average 7.17‰)(Liu et al., 2018) and Beijing
239 (-37.1 ‰ to 5.8‰)(Pan et al., 2016). $\delta^{15}\text{N-NH}_4^+$ values were lower in autumn (17.3‰) and
240 winter (14.4‰) than in spring (22.5‰) and summer (25.7‰), which was similar to the trends
241 in Japan(Kawashima and Kurahashi, 2011).

242 The seasonal differences in $\delta^{15}\text{N-NH}_4^+$ values were significant between warm
243 (summer/spring) and cold seasons (winter/ autumn) ($p < 0.05$). The $\delta^{15}\text{N-NH}_4^+$ was affected by
244 the ratio of $\text{NH}_4^+ / (\text{NH}_3 + \text{NH}_4^+)$ (Eq. (8) and Fig. S1). A linear fitting equation was observed
245 between $\text{NH}_4^+ / (\text{NH}_3 + \text{NH}_4^+)$ and $\delta^{15}\text{N-NH}_4^+$, and the absolute value of the slope (32.4)
246 approximated the isotope equilibrium fractionation value (33%) between atmospheric NH_3 and
247 NH_4^+ (Fig. S1). The linear fitting suggested that the lower the NH_4^+ proportion was, the heavier
248 the $\delta^{15}\text{N-NH}_4^+$ value. The lower NH_4^+ level was accordance with higher $\delta^{15}\text{N-NH}_4^+$ in summer,
249 which was the opposite of winter. In addition, previous study suggested that the marked
250 variation in $\delta^{15}\text{N-NH}_4^+$ values was largely controlled by the emission sources of NH_3 , the
251 precursor gas of NH_4^+ (Liu et al., 2018). According to the $\delta^{15}\text{N-NH}_4^+$ results, the source of NH_4^+
252 was assigned as biomass burning (27.9 ± 16.4 %), coal combustion (16.0 ± 3.9 %), vehicles
253 (19.8 ± 5.3 %), fertilizer (10.9 ± 6.1 %), livestock (12.7 ± 5.8 %), and urban waste (11.9 ± 6.1 %),
254 shown in Fig. 2a.

255 In our study, non-agriculture sources were the dominators of NH_4^+ (75.6%). Unexpectedly,
256 the contribution of biomass burning was the highest. Especially, from late June to July, the
257 contribution of biomass burning enhanced, which possibly resulted from sugarcane leaf
258 burning. The $\delta^{15}\text{N}$ in sugarcane leaf was as high as 38‰(Martinellia et al., 2002). The $\delta^{15}\text{N}$ of
259 NH_4^+ formed from NH_3 released by sugarcane leaves burning was 44.1‰ (SI Text S3), which
260 was consistent with the highest $\delta^{15}\text{N-NH}_4^+$ values (45.5‰ and 45.1‰) in July. In PRD, south
261 winds prevail in July and the sampling site is located downwind of sugarcane planting area.

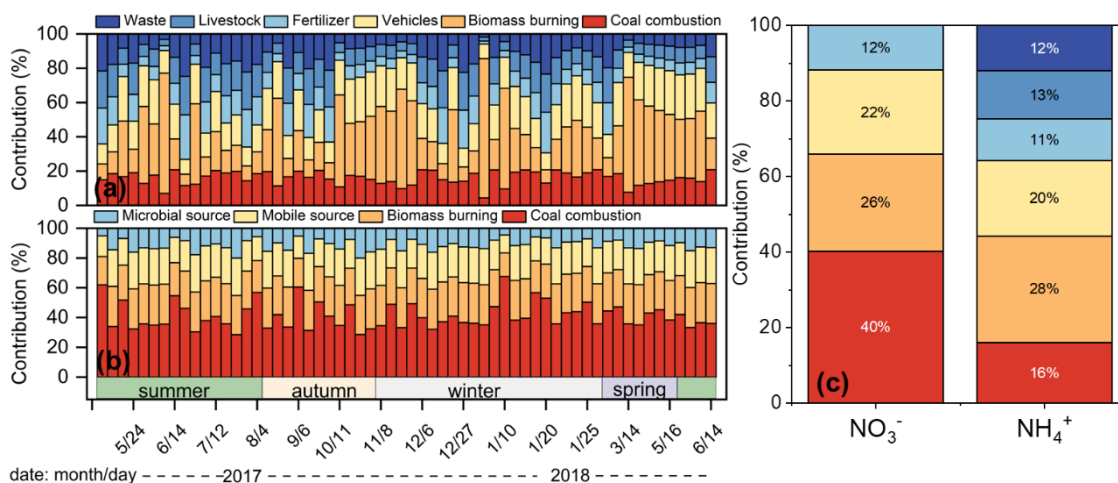
262 Therefore, the air mass to the sampling site might carry the pollutants related to sugarcane leaf
263 burning. K^+ is a typical biomass burning tracer(Cui et al., 2018). Considering the impact of
264 primary emission intensity, $[NH_4^+/EC]$ and $[K^+/EC]$ were used to calculate the correlation
265 coefficient ($r=0.435$, $p < 0.01$), which verified NH_4^+ was influenced by biomass burning. In
266 recent years, biomass burning has been gradually identified as an important source of
267 NH_4^+ (Meng et al., 2017; Xiao et al., 2020). The results based on emission inventories showed
268 that the contribution of residential biomass combustion to NH_3 ranged from 33% to 53% in
269 China(Meng et al., 2017). According to $\delta^{15}N$, biomass burning contributed 18% [Harbin, East
270 North China](Sun et al., 2021), 46%[Wuhan, South Central China], 40% [Changsha, South
271 Central China](Xiao et al., 2020), 35% [Nanchang, East China](Xiao et al., 2020), and 23%
272 [Guangzhou, South China](Chen et al., 2022a) to NH_4^+ . Particularly, in Guangzhou the
273 contribution of biomass burning in the ground was higher than that in Guangzhou tower with
274 a height of 488 meters, suggested that the influence of regional biomass burning(Chen et al.,
275 2022a). Furthermore, 7Be mainly originates from upper atmosphere, whereas ^{210}Pb is derived
276 from terrestrial surface(Jiang et al., 2021b). High level of 7Be observed in ground suggested
277 the sink influence of upper atmosphere. 7Be and ^{210}Pb are chemically stable and with unique
278 sources, which can effectively reflect the transport of continental air mass and the air exchange
279 between stratosphere and troposphere. In our study, the correlation coefficient between NH_4^+
280 and ^{210}Pb ($r=0.701$, $p < 0.01$) was higher than that between NH_4^+ and 7Be ($r=0.432$, $p < 0.01$),
281 suggested that NH_4^+ was mainly affected by regional emission. Therefore, biomass burning
282 exerted essential influence on NH_4^+ level, which should no longer be ignored.

283 In addition, with the acceleration of urbanization, combustion sources related to fossil
284 fuels have become the main sources of NH_3 . In previous studies, the source of NH_x ($NH_3+NH_4^+$)
285 was mainly from agricultural activity due to rough way of farming(Chang et al., 2016; Pan et
286 al., 2020). However, with the improvement of efficient fertilization practices, agricultural NH_3
287 decreased significantly(Wang et al., 2022). Fossil fuels, such as coal and gasoline, are major
288 energies for production and domestic using, and their contribution to NH_3 has become
289 increasingly important. In North China, fossil fuel combustion contributed 92% to NH_3 during
290 hazes(Zhang et al., 2020; Pan et al., 2016). In previous study of Guangzhou, the contribution
291 of NH_3 from fossil source in ground observations (43%) was higher than the observed in

292 Guangzhou tower (18%), indicated the importance of locally related fossil fuel combustion
293 source(Chen et al., 2022a). In our study, vehicle emission and coal combustion contributed
294 $19.8\pm 5.3\%$ and $16.0\pm 3.9\%$ of NH_4^+ respectively, which was lower than North China but higher
295 than agricultural sources. The share of NH_3 from vehicle exhaust was estimated to be 18.8%
296 based on the emission factor of NH_3 from on road vehicles in Guangzhou, which was similar
297 to our results(Liu et al., 2014). The selective catalytic reduction process for vehicle can reduce
298 NO_x , but increased emission of NH_3 , which has confirmed as an important source of NH_3 (Heeb
299 et al., 2006; Meng et al., 2017). Despite the efforts of government to promote electric vehicles
300 in recent years, their share is still relatively low (about 5%). As increasing car ownership, this
301 has an important impact on atmospheric NH_3 . Coal combustion was the second most important
302 source of fossil combustion after vehicle emissions in our study, although the contribution was
303 lower than in North China(Wu et al., 2019; Zhang et al., 2020; Pan et al., 2016). The absence
304 of heating in Guangzhou may explain the lower contribution of coal combustion compared to
305 the North. On an annual basis, the contribution of fossil fuel-related combustion sources in our
306 study (35.8%) was comparable to that in North China (37%-52%)(Pan et al., 2018a).

307 The source contributions of NH_4^+ in our study were compared to other regions, shown in
308 **Fig. S3**. The combustion related sources (biomass burning, coal combustion, and vehicle) have
309 gradually become the dominant source of urban atmospheric NH_3 . Biomass burning and
310 vehicle could emit massive carbon monoxide (CO)(Li and Wang, 2007; Wang et al., 2005). In
311 Guangzhou, NH_4^+ was positively related to CO ($r=0.637$, $p < 0.01$), which confirmed
312 combustion sources played a key role in NH_4^+ . From a historical perspective, NH_3 emissions
313 from anthropogenic combustion and industry have been steadily increasing since 1960(Meng
314 et al., 2017). The optimization of energy structure and encouragement of the development of
315 new energy vehicle would be hopeful to reduce NH_3 . The results of this study would be
316 conducive to reducing NH_3 scientifically and effectively and would relieve the pressure on the
317 reduction from agricultural source.

318



319

320 **Figure 2.** The sources apportionment results of atmospheric NH₄⁺ (a) and NO₃⁻ (b) in
 321 Guangzhou, and the comparison of sources results between NH₄⁺ and NO₃⁻ (c).

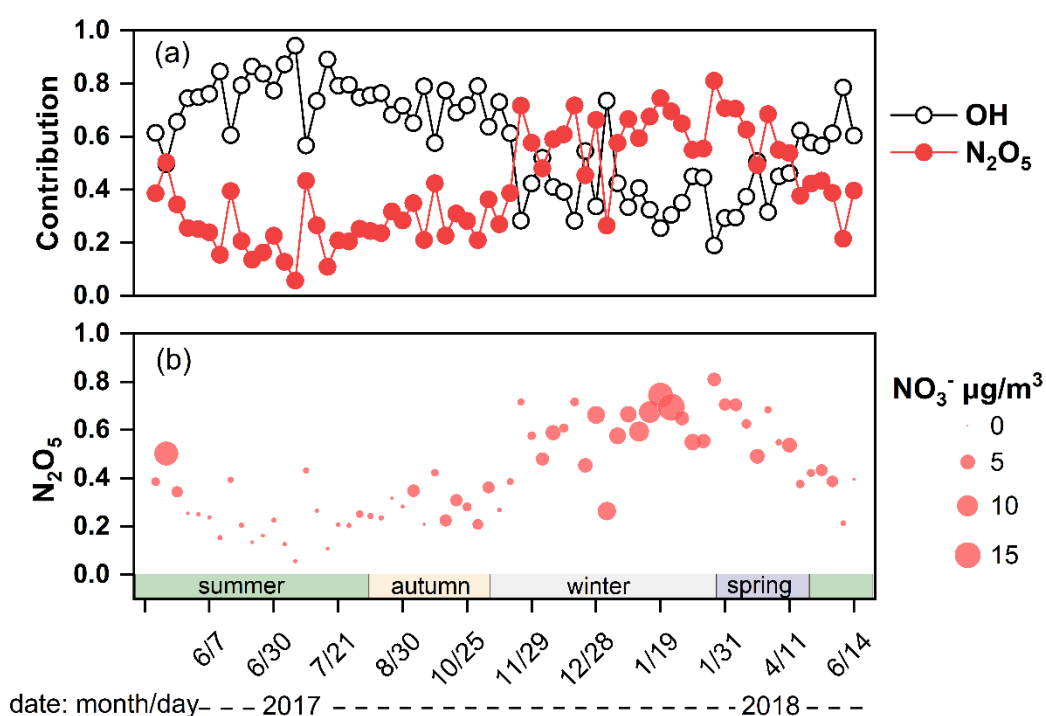
322 **3.3. Characteristic and seasonal variation in $\delta^{18}\text{O}-\text{NO}_3^-$ and $\delta^{15}\text{N}-\text{NO}_3^-$ and source**
 323 **apportionment of NO_3^-**

324 **3.3.1. Seasonal variation of $\delta^{18}\text{O}-\text{NO}_3^-$**

325 The $\delta^{18}\text{O}-\text{NO}_3^-$ in Guangzhou was $68.1 \pm 9.7\text{‰}$ (44.9‰ to 90.5‰) comparable to that in
 326 precipitation (66.3‰, ranging from 33.4‰ to 86.2‰)(Fang et al., 2011), but lower than those
 327 regions with weak light intensity, such as Beihuangcheng Island (ranging from 49.4‰ to
 328 103.9‰)(Zong et al., 2017) and Bermuda Islands (cold season $76.9 \pm 6.3\text{‰}$) (Hastings et al.,
 329 2003). In this study, $\delta^{18}\text{O}-\text{NO}_3^-$ was higher in winter and spring than in summer and autumn,
 330 which was similar to the seasonal variation in $\delta^{18}\text{O}-\text{NO}_3^-$ in previous studies (Fang et al., 2011;
 331 Gobel et al., 2013). On the one hand, $\delta^{18}\text{O}-\text{NO}_3^-$ value was associated with the formation
 332 pathways of NO₃⁻. The results simulated by Bayesian mixing model suggested that the
 333 contributions of N₂O₅ channel to NO₃⁻ were 56.8%, 58.9%, 29.2%, and 27.0% in winter, spring,
 334 autumn, and summer, respectively. The $\delta^{18}\text{O}$ value of NO₃⁻ formed by N₂O₅ channel is higher
 335 than that by ·OH pathway (SI Text S2). The night in cold season was longer than that in warm
 336 season, which favored NO₃⁻ formation through N₂O₅ channel. In addition, the illumination
 337 intensity was weakened in cold season compared with that in warm season, which constrained
 338 the production of ·OH(Zong et al., 2020; Tan et al., 2019; Wang et al., 2017). Thus, the
 339 contribution of the N₂O₅ channel in cold season was higher than that in warm season.

340 Furthermore, concentration of NO_3^- was high when contribution of N_2O_5 channel enhanced
 341 (**Fig. 3**), suggested NO_3^- pollution was related to N_2O_5 hydrolysis pathway. The air mass to
 342 Guangzhou was derived from the South China Sea in summer and the North continental region
 343 in winter. The higher $\delta^{18}\text{O}-\text{NO}_3^-$ and NO_3^- concentration might be affected by long-range and
 344 high-altitude transport from North China, which might carry abundant precursors. Massive
 345 NO_3^- could be formed by N_2O_5 hydrolysis at high altitude and transported to the ground. The
 346 index of $f(^7\text{Be}, ^{210}\text{Pb})$ was expressed in **SI Text S1** and could reflect the influence of atmospheric
 347 dynamic transport on aerosol pollutants(Jiang et al., 2021b). Generally, air masses with low
 348 values of $f(^7\text{Be}, ^{210}\text{Pb})$ suggested that pollutants were associated with continental surface
 349 emission, whereas high $f(^7\text{Be}, ^{210}\text{Pb})$ were influenced by long-range transport from upper air
 350 masses. The contribution of N_2O_5 channel was positively correlated with $f(^7\text{Be}, ^{210}\text{Pb})$ ($r=0.319$,
 351 $p < 0.05$), indicated the long-range transport influence of upper air mass on N_2O_5 channel. For
 352 example, on 25 January 2018, the contribution of N_2O_5 channel (nitrate) was 81.1% ($3.6 \mu\text{g m}^{-3}$)
 353 ³), when the upper air mass was from North China. However, on 7 July 2017, the N_2O_5 channel
 354 (nitrate) contributed only 5.7% ($0.5 \mu\text{g m}^{-3}$) corresponding to the air mass mainly from the
 355 South China Sea transported at low-altitude (**Fig. S4**).

356



357

358 **Figure 3.** The contribution of the OH radical oxidation and N₂O₅ hydrolysis pathway to NO₃⁻
359 (a). The vertical position of dots corresponded to the contribution of N₂O₅ pathway and the size
360 of the dots corresponded to the concentration of NO₃⁻ (b).

361 $\delta^{18}\text{O-NO}_3^-$ decreased from 76.7‰ in 2014 to 68.1‰ in 2017-2018(Zong et al., 2020),
362 which indicated that $\cdot\text{OH}$ channel became more important in Guangzhou. The enhanced
363 contribution of $\cdot\text{OH}$ pathway indicated the increasing atmospheric oxidation capacity. In recent
364 years, although the concentration of PM_{2.5} in Guangzhou has significantly decreased, the
365 photochemical pollution caused by high O₃ concentrations was not optimistic(Tan et al., 2019).
366 The O₃ concentration in the PRD showed a fluctuating upward trend from 2013 to 2020;
367 especially in 2017-2018, O₃ concentrations were at high levels (Environmental Status Bulletin
368 of Guangdong Province Fig. S5). In our study, the NO₃⁻ formation pathway inferred from $\delta^{18}\text{O-}$
369 NO₃⁻ proved the enhancement of atmospheric oxidation capacity.

370 3.3.2. Seasonal variation of $\delta^{15}\text{N-NO}_3^-$ and source apportionment of NO₃⁻

371 **Seasonal variation of $\delta^{15}\text{N-NO}_3^-$.** The $\delta^{15}\text{N-NO}_3^-$ in Guangzhou was $4.9\pm 2.2\%$ (-0.4‰
372 to 10.8‰), which was similar to the wet deposition(Fang et al., 2011). The $\delta^{15}\text{N-NO}_3^-$ was
373 comparable to that from the Northeast United States (6.8‰)(Elliott et al., 2009), and lower
374 than regions in China, where NO₃⁻ was predominantly derived from anthropogenic sources,
375 such as Heshan in Guangdong ($7.50\pm 3.30\%$)(Su et al., 2020), Beihuangcheng Island
376 ($8.2\pm 6.2\%$)(Zong et al., 2017), and Beijing ($12.1\pm 3.3\%$)(Fan et al., 2022). Nevertheless, the
377 $\delta^{15}\text{N-NO}_3^-$ in this study was significantly higher than those from clean background regions,
378 where NO₃⁻ was mainly from natural sources, such as the coast of Antarctica (-
379 $12.0\pm 15.6\%$)(Savarino et al., 2007) and Bermuda ($-2.1\pm 1.5\%$ warm season, $-5.9\pm 3.3\%$ cold
380 season)(Hastings et al., 2003). The values of $\delta^{15}\text{N-NO}_3^-$ in winter, spring, summer, and autumn
381 were 5.6‰, 5.3‰, 4.4‰, and 4.5‰, respectively. The $\delta^{15}\text{N-NO}_3^-$ in winter and summer
382 showed significant difference ($p < 0.05$). The values of $\delta^{15}\text{N-NO}_3^-$ were influenced by
383 atmospheric processes and emission sources(Elliott et al., 2009). For N₂O₅ channel, NO₃⁻ is
384 characterized by higher $\delta^{15}\text{N}$ values(Freyer et al., 1993; Elliott et al., 2009). The N₂O₅ channel
385 was the predominant formation pathway of NO₃⁻ in winter, which was in accordance with the
386 seasonal variation in $\delta^{15}\text{N-NO}_3^-$. In addition, the difference in $\delta^{15}\text{N-NO}_3^-$ reflected the variation

387 in the emission source of NO_3^- . $\delta^{15}\text{N}-\text{NO}_x$ from coal combustion was relatively high. In winter,
388 the higher $\delta^{15}\text{N}-\text{NO}_3^-$ was probably related to long-range transport from North, where coal
389 combustion enhanced in winter.

390 **Source apportionment of NO_3^- .** Based on the Bayesian mixing model coupled with $\delta^{15}\text{N}-$
391 NO_3^- , NO_3^- sources were assigned as coal combustion $40.4\pm 8.7\%$, biomass burning $25.6\pm 2.1\%$,
392 mobile sources (vehicles) $22.3\pm 3.1\%$, and microbial process $11.7\pm 3.8\%$. **Figure 2b** and **Fig.**
393 **S6** showed the source contribution of NO_3^- in Guangzhou and other regions in China,
394 respectively. Compared to earlier periods (2013-2014), the concentration of NO_3^- from vehicle
395 and coal combustion decreased significantly (Zong et al., 2020), which resulted from the stricter
396 vehicle emission standard, promotion of new energy electric vehicles, and ultraclean
397 transformation of coal combustion (Guangdong province, 2014; Tang et al., 2019). However,
398 almost all production and domestic segments rely on energy generated from coal combustion,
399 which was still dominant source of NO_3^- in 2017-2018. Coal combustion was affected not only
400 by local emissions but also by external air mass transmission. The contribution of coal
401 combustion was higher in winter than in summer, which probably related to the long-range
402 transportation from the North. Taking 10 January 2018 as an example, the contribution of coal
403 combustion sources to NO_3^- was 67.5%, and the corresponding air mass was from the North
404 and transmitted to Guangzhou through high altitude. However, the air mass on 26 July 2017
405 was mainly from the South China Sea, which was transmitted through low-altitude to
406 Guangzhou. The contribution of coal burning to NO_3^- on 26 July 2017 was 28.5% lower than
407 that on 10 January 2018.

408 As non-fossil combustion source, biomass burning was also an important source of NO_3^-
409 and accounted for 25.6%. The contribution of biomass burning and vehicle was stable
410 throughout a year. Generally, high intensity biomass burning occurred in winter in Guangdong
411 province (dry season, i.e., from November to March) (Xu et al., 2019). K^+ is a typical tracer of
412 biomass burning. The concentration of K^+ enhanced in winter ($0.4\mu\text{g}/\text{m}^3$) was higher than that
413 in summer ($0.2\mu\text{g}/\text{m}^3$) and autumn ($0.2\mu\text{g}/\text{m}^3$), respectively, indicating enhancement of
414 biomass burning intensity. Also, NO_3^- concentration of biomass burning remarkably enhanced
415 in winter ($1.2\mu\text{g}/\text{m}^3$), and was higher than that in summer ($0.4\mu\text{g}/\text{m}^3$) and autumn ($0.3\mu\text{g}/\text{m}^3$),
416 respectively. However, coal combustion also enhanced in winter due to the demand for heating

417 in North China. Our sampling site was influenced by the air mass with high coal combustion
418 contribution from the North by long-range transportation, which may reduce the contribution
419 of biomass burning relatively. Thus, the contribution of biomass burning showed stable
420 compared with coal combustion. Another non-fossil source is related to soil microbial activity
421 and only contributed 11.7% to NO_3^- , which was unexpectedly lower than the results in earlier
422 periods (2013-2014). Generally, the microorganisms in soil emit NO through nitrification or
423 denitrification, which was affected by the amount of carbon and nitrogen nutrients in soil(Hall
424 and Matson, 1996). In earlier periods, due to the higher level of aerosols, the amount of
425 nutrients settling in soil was also higher, which was exemplified by the observation of dry and
426 wet deposition in Guangzhou(He et al., 2022; Zheng et al., 2020). In addition, the reduction of
427 cultivated land from 2013 to 2018 might also reduce the contribution of microbial source
428 emissions. Therefore, emissions from natural sources were also influenced by human activities
429 to some extent. The contribution of microbial process was higher in summer than in winter. In
430 summer, higher RH and temperature were favorable for the intense activity of soil
431 microorganisms(Zong et al., 2017). The contributions of microbial processes to NO_3^- also
432 decreased in winter compared with summer at regional background sites and five Chinese
433 megacities, including Guangzhou(Zong et al., 2017; Zong et al., 2020).

434 The sources comparison between NO_3^- and NH_4^+ was shown in **Fig. 2c**. Coal combustion,
435 biomass burning, and vehicles were three significant sources of NO_3^- and NH_4^+ . Coal
436 combustion and biomass burning were the dominant sources of NO_3^- and NH_4^+ , respectively.
437 The vehicles were also an important source of atmospheric inorganic Nr contributed to 22.3%
438 and 19.8% of NO_3^- and NH_4^+ , respectively. Recently, the government has actively taken many
439 measures to reduce the pollution from vehicles, such as stricter automobile emission standards
440 and the promotion of new energy vehicles. However, due to the large vehicle ownership base,
441 the pollutants emitted from vehicles are not optimistic. In addition, vehicles emissions could
442 contribute half of the fresh secondary organic aerosol in urban environment(Zhang et al., 2022;
443 Zhao et al., 2022a).

444 **4. Conclusions**

445 A year-long field observation was conducted in Guangzhou to clarify the atmospheric fate
446 of inorganic nitrogen aerosol. Inorganic nitrogen species were the most essential component of
447 TN including NH_4^+ (45.8%) and NO_3^- (23.2%), which are also dominant components of SIA
448 and play a key role in China haze. The $\delta^{15}\text{N}$ is a powerful tool to quantify the source
449 contribution of NH_4^+ and NO_3^- , which suggested that anthropogenic combustion sources (coal
450 combustion, biomass burning, and vehicles) were the dominant sources.

451 Anthropogenic combustion sources contributed 63.2% to NH_4^+ higher than agricultural
452 sources (23.6%). NH_3 largely facilitates the formation of sulfate and nitrate. Meanwhile, sulfate
453 and nitrate promote each other with positive feedback effect, which could trigger haze. In
454 megacities of China, the focus of NH_3 reduction should be on anthropogenic combustion
455 sources, especially on biomass burning, which might be responsible for the lag of the decline
456 in the deposition of air pollutions behind the reduction in emission (Zhao et al., 2022b). In
457 addition, anthropogenic combustion sources accounted for 88.3% of NO_3^- . Coal combustion
458 and vehicles contributed 40.4% and 22.3% to NO_3^- , respectively. Despite a series of measures
459 to reduce emissions of NO_x , fossil fuels, as the main energy for production and living, will still
460 inevitably emit a large amount of NO_x . Our results emphasized that the emission of
461 atmospheric inorganic nitrogen is largely related to anthropogenic combustion sources. The
462 development and promotion of clean energy and efficient use of biomass are conducive to the
463 deep reduction of atmospheric nitrogen.

464 **Data availability**

465 The original data of this research (stable nitrogen isotopes and inorganic nitrogen
466 concentrations) are available at Mendeley data (Li and Li, 2023). The Iso Source model was
467 downloaded from Environmental Protection Agency, via their website:
468 https://www.epa.gov/sites/default/files/2015-11/isosourcev1_3_1.zip.

469 **Author contributions**

470 Funding acquisition: Jun Li

471 Investigation: Tingting Li, Zeyu Sun, and Hongxing Jiang
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473 Project Administration: Jun Li
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475 Software: Tingting Li, Zeyu Sun, and Chongguo Tian
476 Validation: Tingting Li and Jun Li
477 Writing – original draft: Tingting Li
478 Writing – review & editing: Jun Li

479 **Competing interests**

480 The authors declare that they have no conflict of interest.

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