

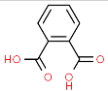
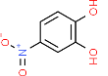
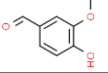
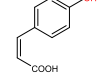
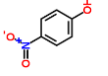
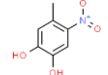
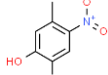
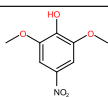
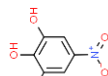
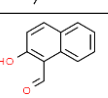
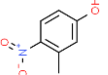




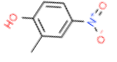
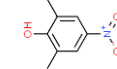
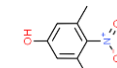
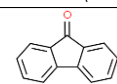
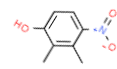
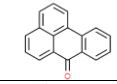
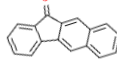
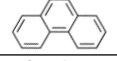
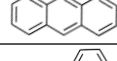
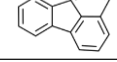
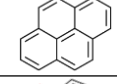
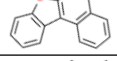
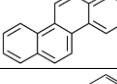
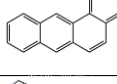
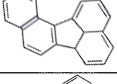
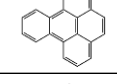
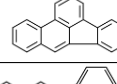
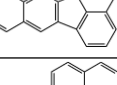
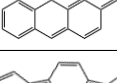
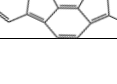
Supplemental Information

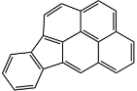

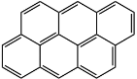
Brown Carbon Aerosol in Polluted Urban Air of North China Plain: Day-night Differences in the Chromophores and Optical Properties

Yuquan Gong et al.

Table S1. Retention Times (RTs), Abbreviation, Elemental Formulas, and Assigned Structures of Identified BrC Chromophores. (*) represent tentative structure components.

Peak #	RT (min)	Candidate compound	Abbreviation	m/z & ion. Mech.	Formula	Reference	Unambiguous/Tentative structure
1#	1.69	Isoquinoline	ISO	130.065 [M & H] ⁺	C ₉ H ₇ N	Banerjee & Zare, 2015	
2#	2.03	Leucoline	LEU	130.065 [M & H] ⁺	C ₉ H ₇ N	Banerjee & Zare, 2015	
3#	6.06	Phthalic acid	PA	149.023 [M & H] ⁺	C ₈ H ₄ O ₃	He et al., 2018	
4#	7.21	4-nitrocatechol	4NC	154.014 [M & H] ⁻	C ₆ H ₅ NO ₃	Huang et al. 2020	
5#	7.87	vanillin	VAN	153.054 [M & H] ⁺	C ₈ H ₈ O ₃	Huang et al., 2021	
6#	9.15	p-cis-coumaric acid	PCCA	163.040 [M & H] ⁻	C ₉ H ₈ O ₃	Smith, Kinney, & Anastasio, 2016	
7#	11.48	4-nitrophenol	4NP	139.019 [M & H] ⁻	C ₆ H ₅ NO ₃	Wang et al., 2017	
8#	13.53	4-methyl-5-nitrocatechol	4M5NC	168.030 [M & H] ⁻	C ₇ H ₇ NO ₄	Kitanovski et al., 2012	
9#	13.82	3-methyl-6-nitrocatechol	3M6NC	168.030 [M & H] ⁻	C ₇ H ₇ NO ₄	Lin et al., 2015	
10#	16.06	4-nitrosyringol	4NS	198.040 [M & H] ⁻	C ₈ H ₉ NO ₅	Lin et al., 2017	
11#	18.23	3-methyl-5-nitrocatechol	3M5NC	168.030 [M & H] ⁻	C ₇ H ₇ NO ₄	Wang et al., 2017	
12#	18.72	1-Formyl-2-naphthol	1F2N	171.045 [M & H] ⁻	C ₁₁ H ₈ O ₂	Rao et al., 2003 (*)	
13#	19.43	3-methyl-4-nitrophenol	3M4NP	152.035 [M & H] ⁻	C ₇ H ₇ NO ₃	Wang et al., 2017	
14#	20.06	1,2-acenaphthylenedione	1,2ACE	183.044 [M & H] ⁺	C ₁₂ H ₆ O ₂	Bandowe et al., 2014	

Peak #	RT (min)	Candidate compound	Abbreviation	m/z & ion. Mech.	Formula	Reference	Unambiguous/Tentative structure
15#	20.53	1,8-naphthalic anhydride	1,8NA	199.038[M & H] ⁺	C ₁₂ H ₆ O ₃	Bandowe et al., 2014	
16#	21.95	2-methyl-4-nitrophenol	2M4NP	152.035 [M & H] ⁻	C ₇ H ₇ NO ₃	Yuan et al., 2020	
17#	26.08	2,6-Dimethyl-4-nitrophenol	2,6D4NP	166.051 [M & H] ⁻	C ₈ H ₉ NO ₃	Yuan et al., 2020	
18#	26.39	3,5-Dimethyl-4-nitrophenol	3,5D4NP	166.051 [M & H] ⁻	C ₈ H ₉ NO ₃	Fischer & Mathivanan, 1988 ^(*)	
19#	27.12	9-fluorenone	9FLU	181.064 [M & H] ⁺	C ₁₃ H ₈ O	Huang et al. 2020	
20#	28.21	2,3-Dimethyl-4-nitrophenol	2,3D4NP	166.051 [M & H] ⁻	C ₈ H ₉ NO ₃	Fischer & Mathivanan, 1988 ^(*)	
21#	36.93	Benzanthrone	BEN	231.079 [M & H] ⁺	C ₁₇ H ₁₀ O	Huang et al. 2020	
22#	38.41	Benzo[b]fluoren-11-one	BbF11O	231.080 [M & H] ⁺	C ₁₇ H ₁₀ O	Yuan et al., 2020	
23#	39.27	Phenanthrene	PHE	178.23 [M & H] ⁺	C ₁₄ H ₁₀	Ho et al., 2009	
24#	39.88	Anthracene	ANT	178.22[M & H] ⁺	C ₁₄ H ₁₀	Alcanzare, 2006	
25#	41.23	Fluoranthene	FLU	202.25[M & H] ⁺	C ₁₆ H ₁₀	Lee & Lane, 2010	
26#	41.7	Pyrene	PYR	202.25[M & H] ⁺	C ₁₆ H ₁₀	Ho et al., 2009	
27#	42.48	Benzo[b]naphtho[1,2-d]furan	BbN[1,2d]F	218.25[M & H] ⁺	C ₁₆ H ₁₀ O	Standards	
28#	43.46	Chrysene	CHR	228.28[M & H] ⁺	C ₁₈ H ₁₂	Bandowe et al., 2014	
29#	43.63	Benzo(a)anthracene	BaA	240.36[M & H] ⁺	C ₁₈ H ₁₂	Huang et al. 2020	
30#	44.94	Benzo(j)fluoranthene	BjF	252.31[M & H] ⁺	C ₂₀ H ₁₂	Standards	
31#	45.23	Benzo(e)pyrene	BeP	252.31[M & H] ⁺	C ₂₀ H ₁₂	Ho et al., 2009 ^(*)	
32#	45.29	Benzo(b)fluoranthene	BbF	252.31[M & H] ⁺	C ₂₀ H ₁₂	Yuan et al., 2020	
33#	45.46	Benzo(k)fluoranthene	BkF	252.31[M & H] ⁺	C ₂₀ H ₁₂	Standards	
34#	45.79	Benzo(a)pyrene	BaP	252.31[M & H] ⁺	C ₂₀ H ₁₂	Ho et al., 2009	
35#	46.73	Indeno[1,2,3-cd]fluoranthene	I[1,2,3cd]F	276.33[M & H] ⁺	C ₂₂ H ₁₂	Alcanzare, 2006 ^(*)	

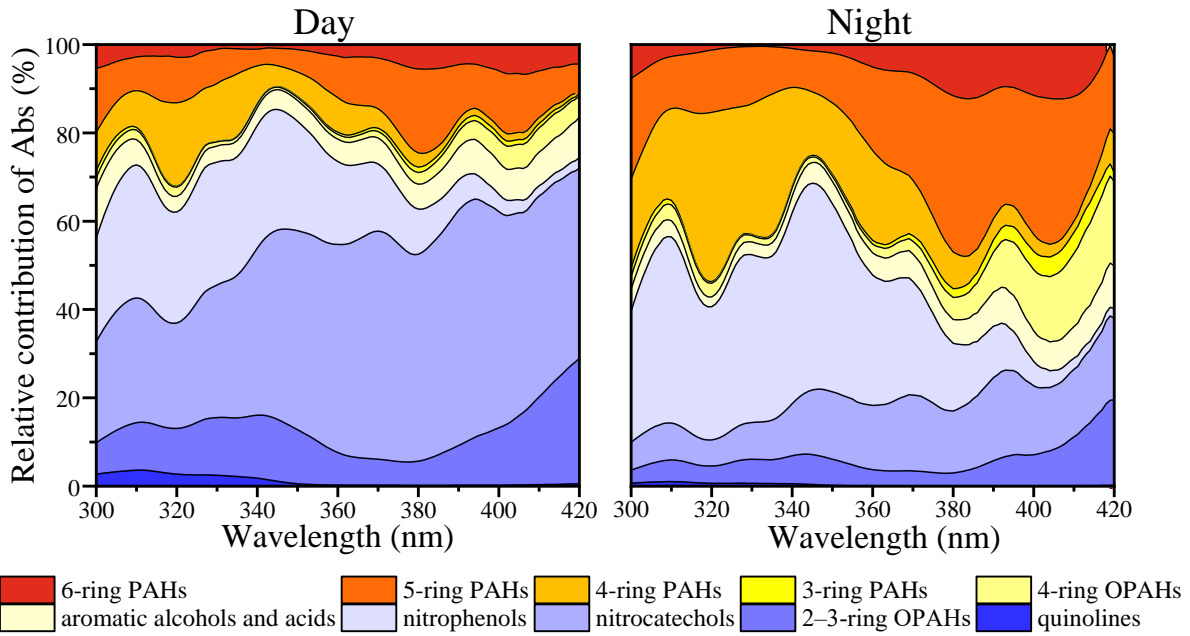
Peak #	RT (min)	Candidate compound	Abbreviation	m/z & ion. Mech.	Formula	Reference	Unambiguous/Tentative structure
36#	47.43	Indeno(1,2,3-cd)pyrene	I[1,2,3cd]P	276.33[M & H] ⁺	C ₂₂ H ₁₂	Standards	
37#	47.53	Benzo(g,h,i)perylene	B(g,h,i)P	276.33[M & H] ⁺	C ₂₂ H ₁₂	Alcanzare, 2006	
38#	48.25	Anthanthrene	ANTHA	276.33[M & H] ⁺	C ₂₂ H ₁₂	Standards ^(*)	

10 **Table S2.** Average (\pm standard deviation) values $\text{Abs}_{365\text{nm}}$, $\text{MAE}_{365\text{nm}}$, and AAE of WS-BrC and WIS-BrC, as well as concentrations of WSOC and WISOC,
 11 measured organic species in the PM 2.5 aerosols from the urban. ^a represents the determination of the HULIS extraction solution. Here $\text{Abs}_{365, \text{MS-BrC}}$ is the
 12 light absorption coefficient of methanol-soluble BrC at 365 nm.

Components	This study		Li et al. (2020)		Huang et al. (2021)		Li et al. (2021)	
	Day	Night	Day	Night	Xi'an	Beijing	Day	Night
WSOC ($\mu\text{g m}^{-3}$)	17.29 \pm 14.49	12.90 \pm 13.36	22.1 \pm 8.0	21.7 \pm 10.4	12.4 \pm 6.50 ^a	6.4 \pm 3.80 ^a	/	/
WISOC ($\mu\text{g m}^{-3}$)	29.78 \pm 22.39	31.07 \pm 12.47	21.9 \pm 10.1	26.2 \pm 17.3	20.80 \pm 7.90	16.30 \pm 8.90	/	/
Abs_{365, WS-BrC} (Mm^{-1})	46.04 \pm 38.91	35.68 \pm 35.50	19.2 \pm 6.8	19.9 \pm 9.5	31.50 \pm 16.40 ^a	15.00 \pm 9.50 ^a	/	/
Abs_{365, MS-BrC} (Mm^{-1})	79.86 \pm 66.50	82.69 \pm 55.84	/	/	/	/	50.0 \pm 5.00	75.0 \pm 7.50
Abs_{365, WIS-BrC} (Mm^{-1})	27.90 \pm 24.80	40.88 \pm 23.42	17.2 \pm 8.2	26.7 \pm 15.8	33.90 \pm 16.40	26.10 \pm 18.40		
MAE_{365, WS-BrC} ($\text{m}^2\text{g}^{-1}\text{C}^{-1}$)	2.58 \pm 0.14	2.88 \pm 0.24	0.92 \pm 0.21	0.94 \pm 0.28	1.80 \pm 0.30 ^a	1.80 \pm 0.40 ^a	/	/
MAE_{365, WIS-BrC} ($\text{m}^2\text{g}^{-1}\text{C}^{-1}$)	1.02 \pm 0.49	1.43 \pm 0.83	0.85 \pm 0.34	1.05 \pm 0.28	1.50 \pm 0.50	1.50 \pm 0.40	1.73 \pm 0.64	2.13 \pm 0.65
AAE_{WS-BrC}	5.10 \pm 0.28	5.51 \pm 0.40	5.14 \pm 0.2	5.07 \pm 0.72	8.20 \pm 1.00	9.40 \pm 2.60	/	/
AAE_{WIS-BrC}	6.36 \pm 0.45	6.97 \pm 0.80	5.94 \pm 0.12	6.15 \pm 0.24	5.4 \pm 0.20	5.7 \pm 0.20	5.16 \pm 1.15	4.07 \pm 0.87

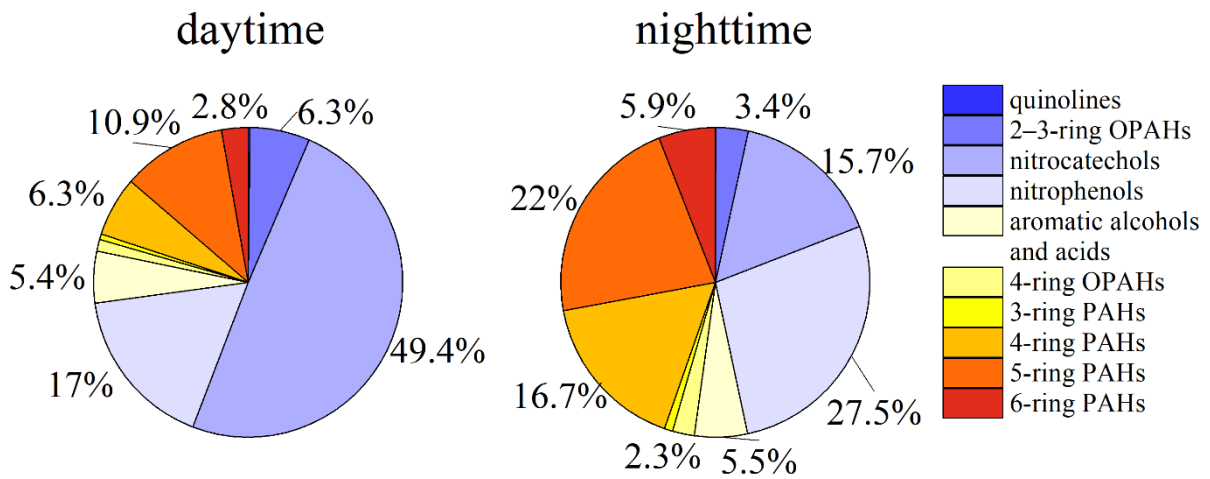
14 **Table S3.** The identified 38 BrC chromophores are divided into ten subgroups.

Categories	Subgroups	Candidate compound
I	quinolines	Isoquinoline
		Leucoline
II	2-3-ring OPAHs	1-Formyl-2-naphthol
		1,2-acenaphthylenedione
		1,8-naphthalic anhydride
		9-fluorenone
III	nitrocatechols	4-nitrocatechol
		4-methyl-5-nitrocatechol
		3-methyl-6-nitrocatechol
		3-methyl-5-nitrocatechol
IV	nitrophenols	4-nitrophenol
		3-methyl-4-nitrophenol
		2-methyl-4-nitrophenol
		2,6-Dimethyl-4-nitrophenol
		3,5-Dimethyl-4-nitrophenol
		2,3-Dimethyl-4-nitrophenol
V	aromatic alcohols and acids	Phthalic acid
		vanillin
		p-cis-coumaric acid
		4-nitrosyringol
VI	4-ring OPAHs	Benzanthrone
		Benzo[b]fluoren-11-one
		Benzo[b]naphtho[1,2-d]furan
VII	3-ring PAHs	Phenanthrene
		Anthracene
VIII	4-ring PAHs	Fluoranthene
		Pyrene
		Chrysene
		Benzo(a)anthracene
IX	5-ring PAHs	Benzo(j)fluoranthene
		Benzo(e)pyrene
		Benzo(b)fluoranthene
		Benzo(k)fluoranthene
		Benzo(a)pyrene
X	6-ring PAHs	Indeno[1,2,3-cd]fluoranthene
		Indeno(1,2,3-cd)pyrene
		Benzo(g,h,i)perylene
		Anthanthrene



15

16 **Figure S1.** Relative contributions of light absorption of ten BrC subgroups during the day and night.



17

18 **Figure S2.** Light-absorbing contributions of ten BrC subgroups at 365nm in the daytime and nighttime.

19 **References**

- 20 Alcanzare, R. J. C.: Polycyclic aromatic compounds in wood soot extracts from Henan, China, 2006.
- 21 Bandowe, B. A. M., Meusel, H., Huang, R.-j., Ho, K., Cao, J., Hoffmann, T., & Wilcke, W. PM_{2.5}-bound
22 oxygenated PAHs, nitro-PAHs and parent-PAHs from the atmosphere of a Chinese megacity: Seasonal
23 variation, sources and cancer risk assessment. *Sci Total Environ.*, 473-474, 77-87.
24 <https://doi.org/https://doi.org/10.1016/j.scitotenv.2013.11.108>, 2014.
- 25 Banerjee, S. and Zare, R. N.: Syntheses of isoquinoline and substituted quinolines in charged microdroplets,
26 *Angew. Chem.*, 127, 15008-15012, 2015.
- 27 Fischer, A., & Mathivanan, N. Formation of dienones on the reaction of cresols, xylenols, and 2-naphthol
28 with nitrogen dioxide: Observation of keto tautomers of nitrophenols. *Tetrahedron lett.*, 29(16), 1869-1872,
29 1988.
- 30 He, X., Huang, X. H. H., Chow, K. S., Wang, Q., Zhang, T., Wu, D., & Yu, J. Z. Abundance and Sources of
31 Phthalic Acids, Benzene-Tricarboxylic Acids, and Phenolic Acids in PM_{2.5} at Urban and Suburban Sites
32 in Southern China. *ACS Earth and Space Chem.*, 2(2), 147-158.
33 <https://doi.org/10.1021/acsearthspacechem.7b00131>, 2018.
- 34 Ho, K., Ho, S. S. H., Lee, S., Cheng, Y., Chow, J. C., Watson, J. G., Louie, P. K. K., & Tian, L. Emissions of
35 gas-and particle-phase polycyclic aromatic hydrocarbons (PAHs) in the Shing Mun Tunnel, Hong Kong.
36 *Atmos. Environ.*, 43(40), 6343-6351, 2009.
- 37 Huang, R.-J., Yang, L., Shen, J., Yuan, W., Gong, Y., Guo, J., Cao, W., Duan, J., Ni, H., & Zhu, C. Water-
38 insoluble organics dominate brown carbon in wintertime urban aerosol of China: chemical characteristics
39 and optical properties. *Environ. Sci. Technol.*, 54(13), 7836-7847, 2020.
- 40 Huang, R.-J., Yang, L., Shen, J., Yuan, W., Gong, Y., Ni, H., Duan, J., Yan, J., Huang, H., & You, Q.
41 Chromophoric Fingerprinting of Brown Carbon from Residential Biomass Burning. *Environ. Sci. Technol.*
42 *Lett.*, 2021.
- 43 Kitanovski, Z., Grgić, I., Yasmeen, F., Claeys, M., & Čusak, A. Development of a liquid chromatographic
44 method based on ultraviolet-visible and electrospray ionization mass spectrometric detection for the
45 identification of nitrocatechols and related tracers in biomass burning atmospheric organic aerosol. *Rapid*
46 *Commun. Mass Spectrom.*, 26(7), 793-804, 2012.
- 47 Lee, J., & Lane, D. A. Formation of oxidized products from the reaction of gaseous phenanthrene with the
48 OH radical in a reaction chamber. *Atmos. Environ.*, 44(20), 2469-2477, 2010.
- 49 Li, J., Zhang, Q., Wang, G., Li, J., Wu, C., Liu, L., Wang, J., Jiang, W., Li, L., & Ho, K. F. Optical properties
50 and molecular compositions of water-soluble and water-insoluble brown carbon (BrC) aerosols in
51 northwest China. *Atmos. Chem. Phys.*, 20(8), 4889-4904, 2020.
- 52 Li, X., Zhao, Q., Yang, Y., Zhao, Z., Liu, Z., Wen, T., Hu, B., Wang, Y., Wang, L., & Wang, G. Composition
53 and sources of brown carbon aerosols in megacity Beijing during the winter of 2016. *Atmos. Res.*, 262,
54 105773, 2021.
- 55 Lin, P., Bluvshstein, N., Rudich, Y., Nizkorodov, S. A., Laskin, J., & Laskin, A. Molecular Chemistry of
56 Atmospheric Brown Carbon Inferred from a Nationwide Biomass Burning Event. *Environ. Sci. Technol.*,
57 51(20), 11561-11570. <https://doi.org/10.1021/acs.est.7b02276>, 2017.
- 58 Lin, P., Liu, J., Shilling, J. E., Kathmann, S. M., Laskin, J., & Laskin, A. Molecular characterization of brown
59 carbon (BrC) chromophores in secondary organic aerosol generated from photo-oxidation of toluene. *Phys.*
60 *Chem. Chem. Phys.*, 17(36), 23312-23325, 2015.
- 61 Rao, P. V., Rao, C. P., Wegelius, E. K., & Rissanen, K. 2-hydroxy-1-naphthaldehyde-derived Schiff bases:
62 synthesis, characterization, and structure. *J. Chem. Crystallogr.*, 33(2), 139-147.

63 <https://doi.org/10.1023/A:1023226925997>, 2003.

64 Smith, J. D., Kinney, H., & Anastasio, C. Phenolic carbonyls undergo rapid aqueous photodegradation to
65 form low-volatility, light-absorbing products. *Atmos. Environ.*, 126, 36-44.
66 <https://doi.org/https://doi.org/10.1016/j.atmosenv.2015.11.035>, 2016.

67 Wang, X., Gu, R., Wang, L., Xu, W., Zhang, Y., Chen, B., Li, W., Xue, L., Chen, J., & Wang, W. Emissions
68 of fine particulate nitrated phenols from the burning of five common types of biomass. *Environ. Pollut.*,
69 230, 405-412, 2017.

70 Yuan, W., Huang, R.-J., Yang, L., Guo, J., Chen, Z., Duan, J., Wang, T., Ni, H., Han, Y., & Li, Y.
71 Characterization of the light-absorbing properties, chromophore composition and sources of brown carbon
72 aerosol in Xi'an, northwestern China. *Atmos. Chem. Phys.*, 20(8), 5129-5144, 2020.