Levels of persistent organic pollutants (POPs) in the Antarctic atmosphere over time (1980 to 2021) and estimation of their atmospheric half-lives.

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Abstract

35 Persistent organic pollutants (POPs) are synthetic compounds that were intentionally produced in large quantities and have been distributed in the global environment, originating a threat due to their persistence, bioaccumulative potential and toxicity. POPs reach the Antarctic continent through long-range atmospheric transport. In these areas low temperatures play a significant role in the environmental fate of POPs, retaining them for a long-time due to cold trapping by diffusion and wet deposition, acting as net sink for many POPs. However, in the current context of climate change, remobilization of POPs

40 trapped for decades in water, ice, and soil, is happening. Therefore, continuous monitoring of POPs in polar air is necessary to assess whether there is a recent re-release of historical pollutants back to the environment. We reviewed the scientific literature on atmospheric levels of several POPs families (polychlorinated biphenyls PCBs, hexachlorobenzene HCB, hexachlorocyclohexanes HCHs, and DDT) from 1988 to 2021. We estimated the atmospheric half-life using characteristic decreasing times (TD). We observed that HCB levels in the Antarctic atmosphere were higher than the other target OCs, but

45 HCB also displayed higher fluctuations and did not show a significant decrease over time. Conversely, the atmospheric levels of HCHs, and some,DDTs, and PCBs have decreased significantly. The estimated atmospheric half-lives for POPs decreased in the following order: 4,4' DDE (13.5 years) > 4,4' DDD (12.8 years) > 4,4' DDT (7.4 years) > 2,4′ DDE (6.4 years) > 2,4′ DDT (6.3 years) > α-HCH (6 years) > HCB (6 years) > γ-HCH (4.2 years), while for PCB congeners they decreased in the following order: PCB 153 (7.6 years) > PCB 138 (6.5 years) > PCB 101 (4.7 years) > PCB 180 (4.6 years) > PCB 28 (4 years)

50 > PCB 52 (3.7 years) > PCB 118 (3.6 years). For HCH isomers and PCBs, the Stockholm Convention ban on POPs did have an impact on decreasing their levels during the last decades. Nevertheless, their ubiquity in the Antarctic atmosphere shows the problematic issues related to highly persistent synthetic chemicals.

1 Introduction

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      Persistent Organic Pollutants (POPs) are a group of toxic chemicals primarily produced and used by the agricultural, industrial, and household applications during the third industrial revolution (Safe, 1994; Qiu et al., 2004; Jayaraj et al., 2016). In the last three decades, studies have reported that POPs levels have soared in the environment worldwide, as these chemicals are highly stable and resistant to degradation (Pennington, 2001). This persistence and their hydrophobicity result in POPs

60   bioaccumulation within organisms and biomagnification along food webs (Hop et al., 2002; Fisk et al., 2001a; 2001b; Borga and Di Guardo, 2005), where they may elicit toxic effects, such as endocrine disruption, threatening the health of both wildlife and humans (Brown et al., 2014; Bourgeon et al., 2012). Given their detrimental effects, 31 substances/substance groups of POPs are currently regulated internationally by the Stockholm Convention (SC), which seeks to reduce and eliminate POPs production and use (UNECE, 1998; UNEP, 2006). However, despite regulatory action among SC signatory nations,

65   considerable levels of POPs are still detected in water, atmosphere, biota, and sediments worldwide due to their persistence, potential for long range transport, as well as their current emission sources (e.g., Vergara et al., 2019; Vasseghian et al., 2021; Avila et al., 2021; Die et al., 2021; García-Cegarra et al., 2021). Of utmost concern, these toxic pollutants are present in the environmental compartments of regions far from emission sources that have previously been considered pristine areas, including polar regions (Galbán-Malagón et al., 2013a; 2013b; 2013c; Pozo et al., 2017; Wu et al., 2020; Azcune et al., 2022;

70   Xie et al., 2022).

      The Antarctic continent is the most remote region from primary sources of POPs (Von Waldow et al., 2010). POPs reach Antarctica mainly through long-range atmospheric transport (LRAT), which generally occurs by the process known as "grasshopping", consisting of successive atmospheric volatilizations and depositions (Blais et al., 2007; Brown and Wania, 2008; Bengtson-Nash, 2011; Jurado and Dachs, 2008). Ocean currents also contribute to their transport processes, albeit at

75   longer timescales since the Antarctic Circumpolar Current acts as a barrier limiting oceanic transport of POPs to the Antarctic continent (Bengtson-Nash et al., 2010). The "barrier theory" has been questioned by Lozoya et al. (2022) for the South Shetland Islands, where the current experiences topographical forcing through the Drake Passage. Finally, another minor transport process is biological, mediated by migratory biota (Braune et al., 2005; Wild et al., 2022). In addition, there may be local sources of POPs, such as research stations and tourist hotspots, that can contribute to detectable and sometimes elevated

80   concentrations of POPs. For example, PCBs have been reported in the vicinity of such local sources (Larsson et al., 1992; Risebrough et al. 1990; Hale et al., 2008). The low temperatures of Antarctica play an important role in the environmental fate of POPs, repressing re-volatilization processes and favoring cold trapping (Wania and Mackay, 1996, Casal et al. 2019), limiting any potential degradation, and enhancing bioaccumulation. In this context, several studies show that polar regions act as a net sink for many POPs; Antarctica is a vast continent covered in ice surrounded by the southern ocean, hence chemicals

85   deposited through LRAT will first deposit in these compartments (Mackay and Wania, 1995; Kallenborn et al., 1998; Dickhut

et al., 2005; Giogia et al., 2008; Cincinelli et al., 2009; Baek et al., 2011; Cabrerizo et al., 2017; Galbán-Malagón et al., 2012, 2013a, 2013c; Montone et al., 2013). For example, there is evidence supporting oceanic sequestration by the biological pump during blooms burying these compounds on the seafloor (Galbán-Malagón et al., 2013a, 2013c) or by biodegradation due to the microbial loop (Galbán-Malagón et al., 2013d). In the context of rapid climate change experienced in polar regions, the
90 remobilization of POPs previously trapped for decades in water, ice, and soil is expected (Nizzetto et al., 2010; Ma et al., 2011; Cabrerizo et al., 2013). The re-emission of POPs to the environment will affect global efforts to moderate human and environmental exposure to these toxic compounds (Bigot et al., 2016), therefore, continuous monitoring of POPs levels in polar abiotic matrices is necessary to assess to what extent such re-emissions to the atmosphere occur.

The detection of chemicals in remote regions serves as direct empirical evidence of a compound's persistence and
95 potential for long-term environmental transport (Bengston-Nash et al., 2017). POPs were first reported in Antarctic biota in the 1960s (Sladen et al., 1966; Tatton and Ruzicka, 1967), sparking interest in studying the transport, fate, and levels present in different environmental compartments. Through the collation of decades of coordinated monitoring data of POPs in the Arctic atmosphere, studies have explored the fate, sources, and long-range transport of POPs in the Northern Hemisphere (Hung et al., 2010, 2016; Wu et al., 2010, 2011). A general downward trend of many airborne POPs has been demonstrated in
100 the Arctic (Hung et al., 2010, 2016; Kong et al., 2014). However, continuous and consistent atmospheric measurements on POPs in Antarctica are limited, due to the remote geographical location and complex climatic conditions of this continent, which put logistical constraints on any monitoring programme. These knowledge gaps make it difficult to understand the fundamental patterns of POPs in this area (Bengston-Nash, 2011), as well as to facilitate systematic comparison with studies conducted in the Arctic.

105 This paper presents the first systematic review of the most reported POPs in the Antarctic atmosphere, allowing to summarize the data collected by the different studies and compare the concentrations recorded over the years and at the different sampling sites. Such compilation allows to identify temporal trends and calculate the atmospheric half-lives of the predominant POPs being monitored to provide insights into expected impacts of environmental remobilization under changing Antarctic conditions.

110 **2 Methods**

2.1 Compilation of bibliographical data

We reviewed all published studies on atmospheric levels of the most reported POP families in the Antarctic
atmosphere (polychlorinated biphenyls (PCBs), hexachlorobenzene (HCB), hexachlorocyclohexane (HCH) and
dichlorodiphenyltrichloroethane (DDT) and its degradation products) from 1988 to 2021. An exhaustive search was performed
in the Web of Science and Scopus databases using the words "Persistent Organic Pollutants", "atmospheric" and "Antarctica",
including only articles written in English; excluding from the analysis references that do not refer to a good quality assurance
and quality control during the chemical analysis, or if the levels of field blanks were not reported. A total of 34 publications
were found, from which we retrieved data on the levels reported, the year in which the samples were collected, and the sampling
sites (Table 1, 2 and 3). We worked exclusively with the levels of the target compounds in the gas phase e, obtained from
active and passive sampling. Furthermore, compounds scarcely reported in the Antarctic atmosphere, such as polybrominated
diphenyl ethers (PBDEs), Polycyclic aromatic hydrocarbons (PAHs), per-and poly-fluoroalkyl substances (PFASs) , were
excluded.

2.2 Statistical analysis

To evaluate the differences between the levels, present in West Antarctica and East Antarctica, a non-parametric U-
Mann Whitney variance analysis was conducted. All the analyses were performed using the R statistical software. (R Core
Team, 2022). To estimate the trend in the change of concentrations, a linear regression was performed between the natural
logarithm of the concentrations for each year studied.

2.3 Estimation of characteristic decreasing times (TD)

Atmospheric half-lives were estimated by deriving the e-folding or characteristic decreasing times (T_D), following the
methodology of Galbán-Malagón et al. (2013a). The half-life is defined as the time needed to decrease the atmospheric
concentration by 35% (e^{-1}) of its initial concentration, which is given by 0.69 T_D. First, only the studies that reported all the
values recorded for each sample were used (Table S1 and S2). These studies were ordered by year of sampling, and their
respective T_D was calculated by least squares adjusting the concentrations to Eq. (1):

$$\ln C_{Atm} = -k_d t + b \qquad\qquad (1)$$

Where k_d is the inverse of the e-folding time T_D (in years), t is the time in years and b is the independent term. T_D was not calculated for β-HCH, due to the limited data available.

145   **3 Results and Discussion**

3.1 Organochlorine Pesticides (OCPs)

OCPs represent most of the POPs listed in the Stockholm Convention. These organic compounds have been widely
150   produced and commercialized since the 1950s for agricultural use and vector control (UNECE, 1998; UNEP, 2006). The application of technical HCH in agriculture has been banned since the early 1980s, while DDT, Lindane (γ-HCH), and HCB were banned in the 1990s (UNECE, 1998; UNEP, 2001). OCPs were first reported in Antarctic marine biota in the late 60's by Sladen et al. (1966) and Taton & Ruzicka (1967). To date, their levels in different environmental compartments continue to be reported (e.g., Vergara et al., 2019; Wu et al., 2020; Krasnobaev et al., 2020; Xie et al., 2022).
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3.1.1 Atmospheric levels of organochlorine pesticides (OCPs)

In the Arctic atmosphere, HCB concentrations are the highest of any OCPs (De March et al., 1998). Similarly, atmospheric concentrations of HCB reported from the Antarctic have been observed to be higher than the other target OCPs
160   (Table 1 and 2), being the most frequently detected and abundant POP in the Antarctic atmosphere (Kallenborn et al., 2013; Wang et al., 2018; Hao et al., 2019; Wu et al., 2020). Temporal patterns of atmospheric HCB concentrations in the Antarctic show significant inter-annual fluctuations with low but significant decreasing trende ($p<0.001$, See table 4)), with a higher variability   over time specially in the last decade (Fig. 1A). A clear decrease in concentrations is shown until about 2010, thereafter a large variability of data is shown where the trend seems to be changing, however there is a lack of sufficient data
165   to be able to confirm this trend. The maximum values were reported by Hao et al. (2019), during the 2012-2018 sampling period on King George Island (Table 1). Such increases in HCB gaseous levels could be mainly associated with re-emission from environmental surfaces (water, soil, and snow) shifting from a reservoir to a secondary source of this compound on the Antarctic continent. HCB is the most persistent OCPs chemical assessed here, as suggested before (Galbán-Malagón et al. 2013). In addition, there may still be an important influence of transport from current primary sources (i.e., combustion and
170   thermal processes) on a global scale and unintentional formation during thermal processing or combustion of chlorine-

containing materials (Barber et al., 2005). The trend shown in Fig 1A points that concentrations of HCB in the Antarctic atmosphere may be regionally dependent, and maybe highly in pace to the climate/environmental change processes occurring in different Antarctic regions.

175   The reported atmospheric concentrations of ΣHCHs in Antarctica from 1980-2019 show a decreasing trend over time (Table 1; Fig. 1 B and C), with significant differences in inter-annual levels (P<0.05). The maximum concentration of HCHs was 170 pg/m^3, reported in 1980-1982 (Tanabe et al., 1982; 1983), and progressively lower concentrations reaching values under detection levels and below 1 pg/m^3 are reported from 2003 to 2019 (Gambaro et al., 2005; Cincinelli et al., 2009; Baek et al., 2011; Galbán-Malagón et al., 2013b; Kallenborn et al., 2013; Pozo et al., 2017; Cabrerizo et al., 2017; Wu et al., 2020; Bigot et al., 2016; Hao et al., 2019). The γ-HCH isomer was found at high concentrations in Antarctica between 1989 and 1990, with a maximum atmospheric concentration of 118 pg/m^3 in 1988 at Ross Island, by Larson et al. (1992) (Fig. 1C, Table S1). Decreasing concentrations are then reported for γ-HCH in 2000, which is unsurprising if fresh sources have been removed, given the lower volatility and higher water solubility of this isomer. On the other hand, the α-HCH isomer, is found to increase since 2006 (Baek et al., 2011; Galbán-Malagón et al., 2013b; Hao et al., 2019), compared to the concentrations recorded during 2001-2004 by Dickhut et al. (2005) and Cincinelli et al. (2009).

185   Published studies reporting gaseous levels for DDT and their isomers from 1988-2018 were lower than the rest of the target OCPs, and like HCHs, the DDTs showed a decreasing trend over the years (Table 2, Fig. 2), with significant inter-annual differences (p<0.05) for compounds 4,4'-DDT, 4,4'-DDE, 2,4'-DDT and 2,2'-DDE, and non-significant annual differences (p>0.05) for compounds 4,4'-DDD and 2,4'-DDD.

    To date, atmospheric concentrations of HCB, α-HCH, β-HCH, and γ-HCH isomers have been studied over much of 190 the Antarctic continent, both in West Antarctica (Kallenborn et al., 1998; Montone et al., 2005; Dickhut et al., 2005; Baek et al., 2011; Galbán-Malagón et al., 2013c; Khairy et al., 2016; Hao et al., 2019), and in East Antarctica (Tanabe et al., 1982, 1983; Lakaschus et al., 2002; Larsson et al., 1992; Jantunen et al., 2004; Bidleman et al., 1993; Gambaro et al., 2005; Cincinelli et al., 2009; Kallenborn et al., 2013; Pozo et al., 2017; Cabrerizo et al., 2017; Wu et al., 2020; Bigot et al., 2016). The detected concentrations of HCB and α-HCHs indicate significant spatial differences (P<0.05), with higher atmospheric concentrations 195 in West Antarctica than in East Antarctica (Table S.4). The γ-HCH isomer did not show spatial differences between the two zones (P>0.05) (Table S.4). The U-Mann Whitney variance analysis was not performed for the β-HCH isomer, because all levels reported in East Antarctica were below the detection limit.

3.1.2. Atmospheric half-lives of organochlorine pesticides (OCPs)

200   The estimated Atmospheric half-lives for OCPs were estimated for the compounds that showed a significant trend in relation to the sampling time (p<0.05 or lower). The estimated half lives decreased in the following order trend decreased in

the following order: 4,4'-DDT (17.2 years) > 2,4'-DDT (14.4 years) > α-HCH (14.3years) > HCB (14.0 years) > γ-HCH (10.1 years) (more details are given in Table 4). The higher atmospheric half-life values estimated in this study for DDTs isomers, compared to the values estimated for HCHs and HCB might be related to the years in which these compounds were banned,

205 since DDTs were banned approximately 10 years after HCHs isomers. It may also be due to continuous production and use of DDTs in some parts of the world due to exemptions to the Stockholm Convention. The estimated values are higher than the atmospheric half-lives reported by other authors, such as Atkinson (1986); Howard (1991); Mortimer & Connel (1995); and Kelly et al. (1994), whose estimated and published values do not exceed one year. However, the methodologies employed differ from the one used in the present study, where Atkinson (1986); Howard (1991); and Mortimer & Connel (1995), were

210 based on rate constant of gas-phase reaction with OH radical for trichlorobiphenyls, while Kelly et al. (1994), were based on atmospheric transformation lifetime. On the other hand, if we compare studies with similar methodology, the study by Venier & Hites (2010) in Great Lakes shows that the half-life estimates for α-HCH and γ-HCH are in a similar range to our estimates, while the one obtained for 4,4'-DDDT is slightly lower. (Fig 4a) Likewise, according to the half-life estimates by Wong et al. (2021), HCB shows higher values than those reported by us, but they report similar values than ours for α-HCH, γ-HCH, 2,4'-

215 DDDT and 4,4'-DDDT (Fig. 4a).

Polar areas are often considered to be a net sink for POPs. Studies have documented that α-HCH and γ-HCH exchanges preferentially from air to water, with this diffusion being the predominant atmospheric deposition mechanism (Galbán-Malagón et al., 2013a,c; Dickhut et al., 2005; Cincinelli et al., 2009; Jantunen et al., 2004; Lohmann et al., 2009; Xie et al., 2011; Zhang et al., 2012; Huang et al., 2013). Once deposited onto surface waters, they are susceptible to sequestration

220 by the biological pump (Galbán-Malagón et al., 2013a,c), as well as to degradation driven by hydrolysis and biodegradation to a minor extent (Harner et al., 2000; Helm et al., 2002; Galbán-Malagón et al., 2013c). These processes minimize the opportunity for re-entry to the atmosphere through volatilization. The lower half-lives values for HCHs may be related to their lower Henry's law constant when compared to other POPs. On the contrary, to our knowledge, no degradation processes have been documented for HCB in surface water and furthermore, conditions close to air-water equilibrium have been reported for

225 this compound (Cincinelli et al., 2009; Galbán-Malagón et al., 2013c). Similarly, DDTs are more hydrophobic with much higher K_{OW} values than HCHs (Table S3), so they are rapidly removed from seawater as particles sink (Lohmann et al., 2007). Thus, it is possible that the high half-lives estimated for DDTs and their metabolites DDD and DDE may be due to unknown current primary and secondary sources (Voldner and Li, 1995; Channa et al., 2012, Li et al., 2020).

230 **3.2 Polychlorinated Biphenyls (PCBs)**

Like OCPs, polychlorinated biphenyls (PCBs) were among the first groups of POPs to be listed under the Stockholm Convention and are characterized by their high chemical stability. Prior to their regulatory control in the 1970s, commercial mixtures of PCBs were widely used in many industrial applications, such as fluids in transformers and capacitors, hydraulic fluids, lubricating oils, and as additives in pesticides, inks and paints, flame retardants, plasticizers, sealants for wood and cement surfaces, among others (Kennish, 1997; FAO/UNEP 1992).

PCBs were first reported in Antarctica in the 1960s and 1970s (Risebrough et al., 1968, 1976), and since then, numerous studies have reported their levels in air, water, sediments, snow, and biota on the Antarctic continent (e.g., Kallenborn et al., 1998; Fuoco et al., 1995; Gupta et al., 1996; Weber et al., 2003; Kim et al., 2015). Here, we selected 7 indicator PCB congeners (28, 52, 101, 118, 138, 153 and 180) considering that they are the most reported PCB congeners worldwide, including Antarctica.

3.2.1 Atmospheric levels of polychlorinated biphenyls (PCBs)

The atmospheric concentrations of Σ_7PCBs reported by the reviewed studies were below those of the target OCPs (Table 3). Overall, the levels of Σ_7PCBs reported from 1988 to 2018 showed a decreasing trend over time (Table 3 and 4, Fig. 3), with significant differences in their levels ($p<0.05$). Congeners 28 and 52 recorded the highest concentrations on King George Island, with values of 69.9 pg/m^3 in 1995, and 33.2 pg/m^3 in 1996, reported by Montone et al. (2005; 2003) (Fig. 3 A and B, Table S2). In contrast, the lowest concentrations of all target PCBs were reported for congener 180, ranging from not detected (n.d) to 3.4 pg/m^3 (Fig. 3G, Table S2).

Like OCPs, atmospheric concentrations of the seven PCB congeners have been reported over most of the Antarctic zone, covering the West Antarctic zone (Montone et al., 2003; 2005; Kallenborn et al., 1998; Baek et al., 2011; Galbán-Malagón et al., 2013c; Li et al., 2012; Khairy et al., 2016; Wang et al., 2017; Hao et al., 2019; Wu et al., 2020) and eastern Antarctica (Larson et al., 1992; Gambaro et al., 2005; Kallenborn et al., 2013; Pozo et al., 2017; Cabrerizo et al., 2017). Significant spatial differences ($p<0.05$) were observed in the atmospheric concentrations of congeners 28, 52, 101 and 138, with higher concentrations in West Antarctica than East Antarctica, while there was no significant difference among sites for congeners 101, 118 and 153 ($p>0.05$). These differences are consistent with the different atmospheric patterns over the Antarctic peninsula regions, with entrance of air-masses from the north, and more permanent wet deposition events by snow and rain, increasing the regional concentrations of POPs (Casal et al. 2019, Casas et al. 2021).

3.2.2. Atmospheric half-lives of polychlorinated biphenyls (PCBs)

The estimated atmospheric half-lives for target PCBs decreased in the following order: PCB 153 (7.6 years) > PCB 138 (6.5 years) > PCB 101 (4.7 years) > PCB 180 (4.6 years) > PCB 28 (3.9 years) > PCB 52 (3.7 years) > PCB 118 (3.6 years) (Table 4). The estimated half-lives were directly proportional to the congener's Henry's law constant (HLC) values. (Table S3). Studies by Atkinson (1986) and Sinkkonen & Paasivirta (2000) reported half-lives lower than those estimated in the present work, where none of the estimated half-lives for these compounds exceeded 1 year. However, the methodology of both studies differs from that of the present study, calculating the half-lives of the compounds by means of the rate constant of gas-phase reaction with the OH. Regarding studies using a similar methodology, the atmospheric half-lives estimated by Venier & Hites (2010) in the Great Lakes (United States and Canada) and by Wong et al. (2021) in the Arctic were higher relative to our results for PCBs 28, 52, 101 and 118. They were in a similar range for PCBs 138, 153 and 180 (Fig. 4b).

Studies have documented that the biological pump is highly efficient for PCBs with high hydrophobicity, i.e., high K_{OW} values (Table S3) (Dachs et al., 2002, Galbán-Malagón et al., 2012; Galbán-Malagón et al., 2013a), thus reducing their revolatilization. The estimated atmospheric half-lives, however, do not reflect lower values for the compounds with higher K_{OW} (e.g. PCB 138, 153 and 180), so other factors may be influencing the high estimated half-lives of the more hydrophobic PCBs. One of these factors could be the presence of local sources of certain PCB congeners, since it has been reported that near the research stations higher PCB concentrations are monitored, compared to sites farther away from these stations, specifically PCB congeners 28, 52, 56 and 101 (Li et al., 2012b; Montone et al., 2003). Furthermore, remobilization of PCBs stored in soils and ice (Cabrerizo et al. 2013, Casal et al. 2019) could be another factor modulating the surface, and thus atmospheric, concentrations of POPs.

3.3 Influence of global climate change on the dynamics of persistent organic pollutants (POPs) in the Antarctic continent.

Over the past decades, global climate changes and the effects of increasing temperatures have been observed in the northern and southern hemispheres (Hung et al., 2022). Increases in ambient temperature can influence physical and chemical processes and ecosystem changes. For example, it has been reported that increasing ambient temperature will affect the dynamics and exchange of POPs between different environmental matrices. Some studies have exposed the relationship between climate change and POPs concentrations (Vorkamp et al., 2022; Potapowicz et al., 2018), describing that POPs are temporarily stored in sediments/soils and can be released into the environment with thawing permafrost (Potapowicz et al., 2018) and that an increase in POPs availability following iceberg calving (NIC, 2014) or increased soil remobilization by up to 45% (Cabrerizo

et al., 2013). In addition, several studies show that seawater, snow, and presumably Antarctic soil are becoming critical secondary sources for POP remobilization (Cabrerizo et al., 2012, 2013; Klanová et al., 2008; Casal et al., 2019).

On the other hand, the mean annual air temperature along the western Antarctic Peninsula has been reported to have increased by as much as 3.4 °C. In addition, the mid-winter temperature increased by 6.0 °C over the past 50 years, making the region

295 one of the most critically affected by climate warming (Vaughan et al., 2003; Turner et al., 2005). However, evidence from field sampling indicated that, to date, there is no relationship between atmospheric ambient temperature and atmospheric concentrations of HCH and HCB in East Antarctica (Bengtson-Nash, 2017). On the other hand, increasing ambient temperature leads to decreased snow cover, nutrient runoff from land to sea, and increased bioavailability of nutrients on land, causing an increase in primary producers on land and sea (Wasley et al., 2006). In this context, it has been demonstrated in aquatic systems

300 that an increase in primary productivity is vital in the sedimentation processes of POPs from the surface to the aquatic bottom through the biological pump (Larson et al., 2000; Galbán-Malagón et al., 2018), this process contributes mainly to the removal of POPs from the environment, despite the adverse effects of an increase in primary productivity in ecosystems (e.g., increase in organic matter). Thus, global warming in Antarctica not only implies a temperature change but also leads to multiple processes that can affect the biogeochemical dynamics of POPs, which plays an essential role in the environmental fate of

305 POPs. Because depending on the future effects of climate change, Antarctica can act as a secondary source of POPs through the re-volatilization of these compounds or as a sink, contributing to the decrease of their environmental levels. Therefore, future exploration of the impact of climate change is necessary and establishes the importance of establishing long-term monitoring networks.

310 3.4 Potential sources of bias

As presented here, several factors can be considered as sources of bias from historical data analysis. First, in the time frame of this study (1980-2021), analytical instrumentation and laboratory techniques exhibited dramatic change, particularly with the advent of advanced mass spectrometry (MS) over electron capture detection (ECD) or novel calibration techniques based on

315 isotopically labeled standards (Azcune et al., 2022). Therefore, recent data are generated by more sophisticated techniques and modern laboratory QA/QC criteria, which need to be included in data reported before 2000. On the other hand, we also included studies using active and passive sampling, but no major differences in the values obtained were observed (See Tables 1 and 2).

The published information from Antarctica is reduced to a group of individual experiences in different geographical locations

320 of international teams working in the field under different conditions and levels of competence that are difficult to obtain and analyze. One might expect that studies that show a strong track record in Antarctic research, reporting POP levels over a time

series, might have greater validity due to constancy and consistency in both sampling and the types of analyses used (e.g., Larson et al., 1992; Baek et al., 2011; Hao et al., 2019).

In summary, the source of bias, related to the technological advancement of the analyses of the collected samples, could have
325   relevance in the observed variability of the historical trends of HCB and HCH (see Fig. 1 A to C). In these cases, it is suggested to continue with dedicated monitoring of these POPs in the coming years to obtain robust observations and conclusions on the degradation of POPs in the Antarctic atmosphere.

4 Conclusion

330

In the present review, a clear trend of decreasing concentrations of PCBs and most targeted OCPs in the Antarctic atmosphere from 1980 to 2018 is documented. In response to the hypothesis raised historically about the decrease in atmospheric levels of historical POPs (Vecchiato et al., 2015). However, it opens the door to study new families of pollutants for which there is already analytical capacity not available in previous decades. In the case of HCH isomers, DDT and PCB congeners, high
335   atmospheric concentrations were reported during 1990-1999 decade but these compounds were highly restricted since 1970s. After that date, a strong decrease was observed in the Antarctic atmosphere, which shows that the Stockholm Convention ban on POPs did have the intended impact on the (atmospheric) concentrations over time. However, these compounds are still ubiquitous in the Antarctic atmosphere with atmospheric half-lives of more than 3 years. On the other hand, the revised atmospheric levels of HCB show a decrease in the decade of its prohibition (1990), however, from the year 2000 onwards,
340   they show strong fluctuations in literature, with values even higher than those reported in 1990. It is noteworthy to take into account, that a decrease of the atmospheric concentrations does not imply neither a decrease of the total POPs in Antactica, an issue that will require future work. In fact, a re-emission of HCB and other POPs from environmental surfaces, such as water, soil, and snow, product of its high stability in the environment, potential sources of bias. Studies to date in Antarctica do not allow conclusions to be drawn about the influence of temperature on the environmental fate of POPs on the Antarctic continent.
345   This is due to the lack of consistent time series data as historically conducted in the Arctic. Moreover, our results point to the importance of periodic monitoring and the need to establish monitoring networks with continuous sampling campaigns, not only with aim to monitor the legacy POPs, as well as to identify new pollutants that have the potential to reach Antarctica (e.g., new flame retardants, per- and polyfluoroalkyl substances (PFAS), and polycyclic aromatic hydrocarbons (PAHs), among others). There is increasing evidence of the presence of emerging compounds in different environmental matrices in Antarctica,
350   however, the current surveillance of atmospheric pollutants is related to specific research groups, instead of coordinated efforts

between countries with Antarctic presence, where continuous monitoring networks could be generated with the inclusion of various persistent toxic chemicals, as analogous to the efforts done by the Arctic Monitoring and Assessment Program (AMAP), or the Integrated Atmospheric Deposition Network (IADN) in the Great Lakes.

355 **Conflict of Interest.**

The authors declare no conflict of interest.

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Table 1. HCB and HCHs levels (pg/m^{-3}) in Antarctic atmosphere since 1980 to present. $\sum$n indicates the number of isomers included in the study.

Sampling area	Type of sampling	Year	HCB	a-HCH	y-HCH	$\sum$HCHs	$\sum$ n	Reference
Southern ocean	Active	1980-1981				90-170		Tanabe et al., 1982
Southern ocean	Active	1981-1982				44-170		Tanae et al., 1983
Cape town and Newmayer Station	Active	1999		0.36	0.15			Lakaschus et al., 2002
Ross Island	Passive	1988 - 1999			25.8 (0.5 - 118)			Larson et al., 1992
East Antarctica	Passive	1990	62.6 (40-78)	3.2 (2.8-3.6)	2.4 (1.1-5.6)	5.7	2	Bidleman et al., 1993
Signy Island	Active	1994 - 1995		2.8	21.8	26.97	3	Kallenborn et al., 1998
East Antarctica	Passive	1997 - 1998		1.06 (0.81 - 1.4)				Jantunen et al., 2004
Terranova bay	Passive	1993	21 (n.d - 28)			13 (5 - 20.0)	3	Kallenborn et al., 1998
Ross Island	Active	1995	(<0.6 - 25.3)			3.9-32.5	2	Montone et al., 2005
west of the Antarctic Peninsula and southwest of Adelaide Island	Active	2001 - 2002	19.4 (<5 - 32.1)	0.3 (<0.05- 0.52)	0.755 (<0.02-2.98)			Duckhut et al. 2005
Terra Nova Bay	Active	2003-2004	11.4 (6.0 - 20)			0.8 (0.3 - 1.2)	2	Gambaro et al., 2005
Terra Nova Bay	Active	2003-2004	11.4 (5.93 - 20.4)			0.22 (0.1 - 0.35)	2	Cincinelli et al., 2009
Ny-Ålesund, King George Island, and Chuuk	Passive	2005-2009						Baek et al., 2011
South Scotia	Active	2008	8.1 (2.18 - 15.82)	1.7(0.06-5.84)	4.6 (1.5-7.1)			Galbán-Malagón et a., 2013b
Wedell	Active	2009	19.5 (2.4 - 30.1)	0.16 (0.05- 2.09)	0.84 (0.1- 1.87)			Galbán-Malagón et a., 2013b
Bransfield sea	Active	2009	16.7 (3.3 - 34.24)	0.14 (0.04- 0.46)	1.15 (0.2-3)			Galbán-Malagón et a., 2013b
Bellingshausen	Active	2009	42.9 (27.31 - 49.71)	0.26 (0.22- 0.16)	0.14 (0.07- 0.19)			Galbán-Malagón et a., 2013b
Palmer station	Active	2010	34 (26.2 – 37.7)	0.81 – 1.68	0.87 – 2.31			Khairy et al., 2016
Station troll / Queen Maud Land	Active	2010	22.9					Kallenborn et al. 2013

Ross Sea	Passive	2010-2011	22.8 (0.8 - 50)	0.5 (n.d-0.5)	n.d	0.5 (n.d-0.5)	2	Pozo et al., 2017
Antarctic Plateau	Active	2011	(0.67-2.7)		BD-2.7			Cabrerizo et al. 2016
Antarctic marginal seas	Active	2013-2014	2.6 (0.081 - 10)			(n.d - 6.8)	3	Wu et al., 2020
Southern Ocean between Australia and Antarctica	Active	2014	(<22 - 35)	<0.13-1.1	<0.70-4.3	n.d - 3.65	3	Bigot et al., 2016
King George Island	Passive	2012-2018	163 (99.2 - 252)	1.4 (0.5-13.6)	0.1-7.9	0.7 - 22.3	4	Hao et al., 2019

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Table 2. PCBs levels (pg/m^{-3}) in Antarctic atmosphere since 1980 to present. ∑n indicates the number of congeners included in the study.

Sampling area	Type of sampling	Year	∑PCBs	∑n	Reference
Ross Island	Passive	1988 - 1990	15.2	6	Larson et al. 1992
King George Island	Active	1993 - 1994	20.8 (12.09 - 42.8)	10	Montone et al., 2001
Signy Island	Active	1994 - 1995	(0.01-17.2)	22	Kallenborn et al., 1998
Ross Island	Active	1995	62.4	11	Montone et al., 2005
King George Island	Active	1996-1996	37.4 (12.1 - 92.6)	10	Montone et al., 2003
Terra Nova Bay	Active	2003-2004	1.06 (0.61-1.78)	61	Gambaro et al., 2005
Ny-Ålesund, King George Island, and Chuuk	Passive	2005-2009	60.3(22.8 - 87.1)	11	Baek et al., 2011
Ny-Ålesund, King George Island, and Chuuk	Passive	2005-2009	19.8 (11.1-31.9)	205	Baek et al., 2011
ICEPOS	Active	2005	16.84 (7.12- 25.65)	25	Galbán-Malagón et al. 2013c
South Scotia sea	Active	2008	45.13 (6.2 - 78.9)	25	Galbán-Malagón et al. 2013c
Antarctic peninsula	Active	2009	12.13 (1.8- 38.1)	25	Galbán-Malagón et al. 2013c
Polish beach	Active	2009	(2.1 - 3.1)	25	Galbán-Malagón et al. 2013c
Livingston Island	Active	2009	7.23 (3.5 12.9)	25	Galbán-Malagón et al. 2013c
King George Island	Passive	2009 - 2010	1.142	7	Li et al 2012
King George Island	Passive	2009 - 2010	36.837	19	Li et al 2012
King George Island, Antarctica.	Passive	2009 - 2010	4.34	7	Li et al. 2012/2
Station troll / Queen Maud Land	Active	2010	0.5	32	Kallenborn et al., 2013
Palmer station	Active	2010	12	29	Khairy et al., 2016
Ross Sea	Passive	2010-2011	0.46 (0.14-1.13)	7	Pozo et al. 2017
Antarctic Plateau	Active	2011	(0.8-27)	26	Cabrerizo et al. 2016
King George Island	Active	2011-2014	5.39 (0.91-35.9)	7	Wang et al. 2017
King George Island	Active	2011-2014	5.87- 72.7 (26.1)	20	Wang et al. 2017
King George Island	Pasive	2010-2018	10.4 (1.5 - 29.7)	19	Hao et al., 2019
Antarctic marginal seas	Active	2013-2014	1.1 (nd-6.7)	14	Wu et al. 2020
King George Island and Ardley Island	Passive		0.078 (nd - 0.137)	6	Wang et al. 2015

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Table 3. Estimated atmospheric decrementing times of examined compounds POPs.

Compounds	T$_D$ (Years)	95% Confident Interval	R^2	*p-value*	Equation
HCB	14.0	10.6-20.7	12.03	<0,0001	LnCg = -0.04931*year + 100.2
a-HCH	14.3	12.4-17.0	32.55	<0,0001	LnCg = -0.04817*year + 96.17
y-HCH	10.1	8.6-12.3	44.63	<0,0001	LnCg = -0.06837*year + 136.9
4,4′ DDT	17.2	11.8-31.7	23.54	<0,0001	LnCg = -0.04015*year + 79.50
2,4 DDT	14.4	9.8-27.3	37.55	<0.001	LnCg = -0.04794*year + 94.99
2,4 DDE	17.6	9.2-232	15.44	<0.05	LnCg = -0.03916*year + 77.93
PCB 28	3.9	3.2-5.2	43.08	<0,0001	LnCg = -0.1748*year + 351.2
PCB 52	3.7	3.2-4.3	63.53	<0,0001	LnCg = -0.1887*year + 378.7
PCB 101	4.7	4.0-5.6	67.42	<0,0001	LnCg = -0.1480*year + 295.8
PCB 118	3.6	3.0-4.3	55.91	<0,0001	LnCg = -0.1930*year + 385.8
PCB 138	6.5	5.3-8.3	40.7	<0,0001	LnCg = -0.1066*year + 212.7
PCB 153	7.6	6.0-10.4	31.59	<0,0001	LnCg = -0.09071*year + 181.2
PCB 180	4.6	3.3-8.0	24.64	<0,0001	LnCg = -0.1486*year + 296.2

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Figure 1. Atmospheric levels (pg/m³) of HCB (A), α-HCH (B), β-HCH (C) and γ-HCH (D), over time.

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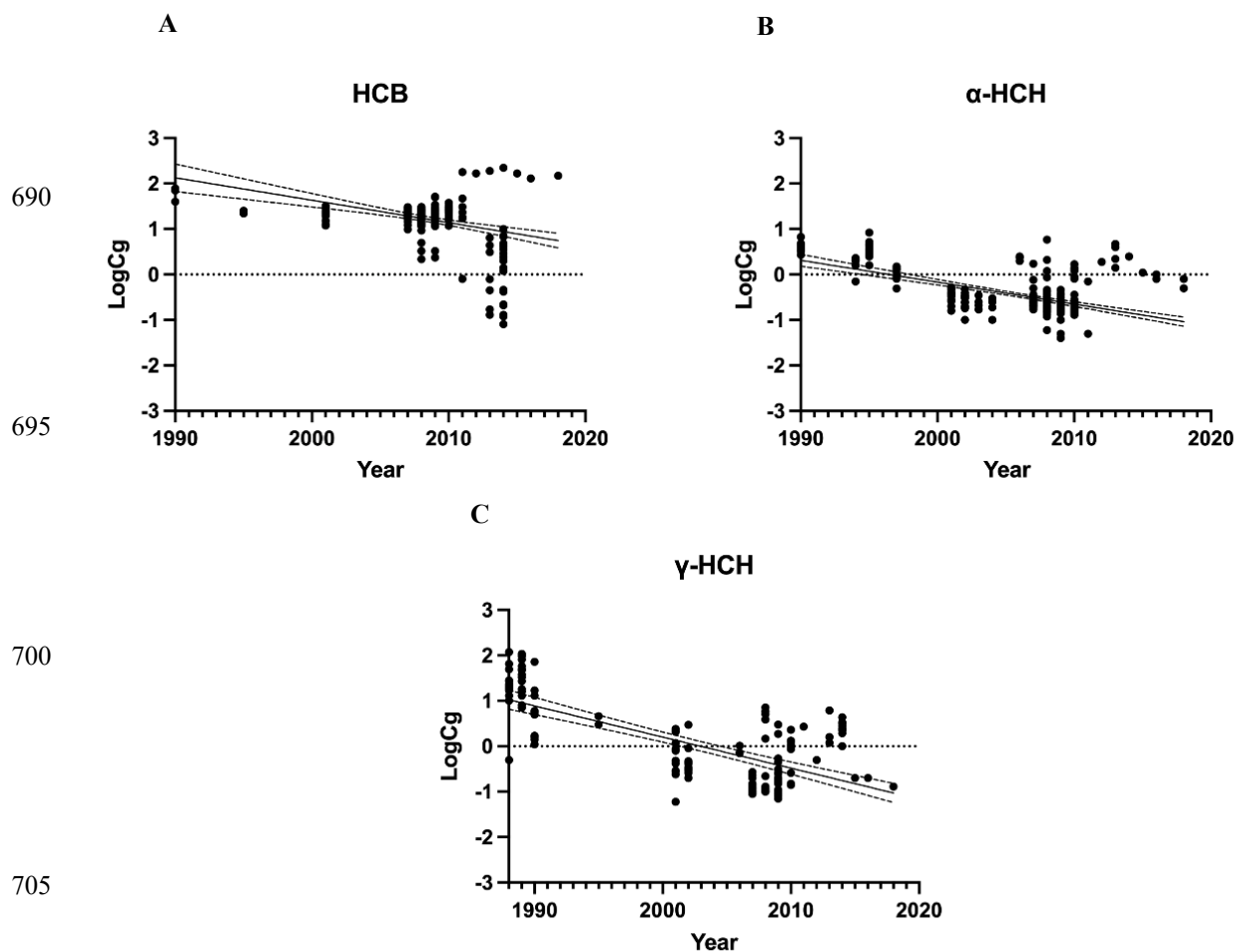
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Figure 2. Atmospheric levels (pg/m³) of 2,4'-DDT (A), 4,4′-DDT (B), 2,4'- DDE (C), over time.

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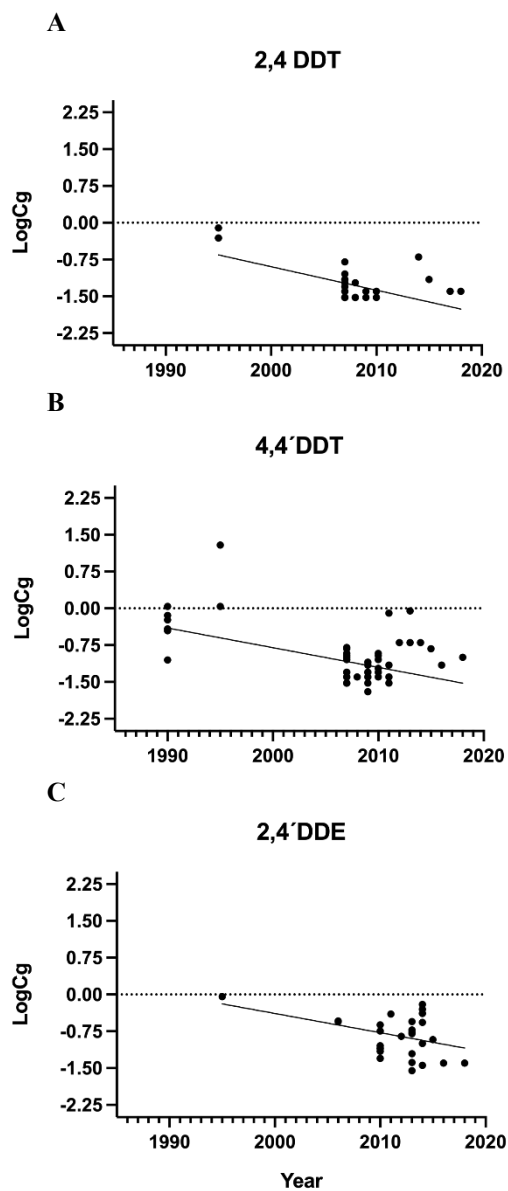
Figure 3. Atmospheric levels (pg/m³) of PCB-28 (A), PCB-52 (B), PCB-101 (C), PCB-118 (D), PCB-138 (E), PCB-153 (F) and PCB-180 (G), over time.

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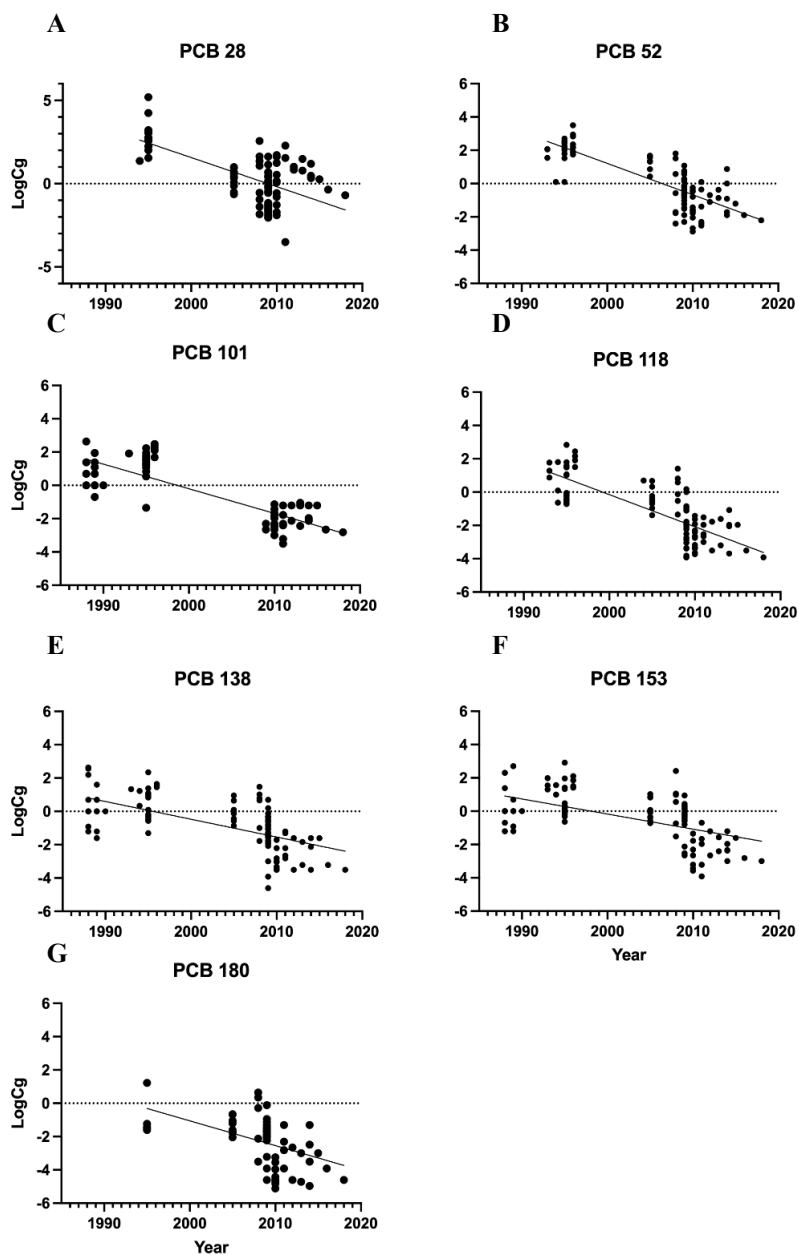
Figure 4. Comparison among the estimated atmospheric residence time obtained in the present work compared with similar
estimations from the Great Lakes and the Arctic for a) HCB, A-HCH, DDX, and b) PCBs