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2 **High Enrichment of Heavy Metals in Fine**

3 **Particulate Matter through Dust Aerosol Generation**

4

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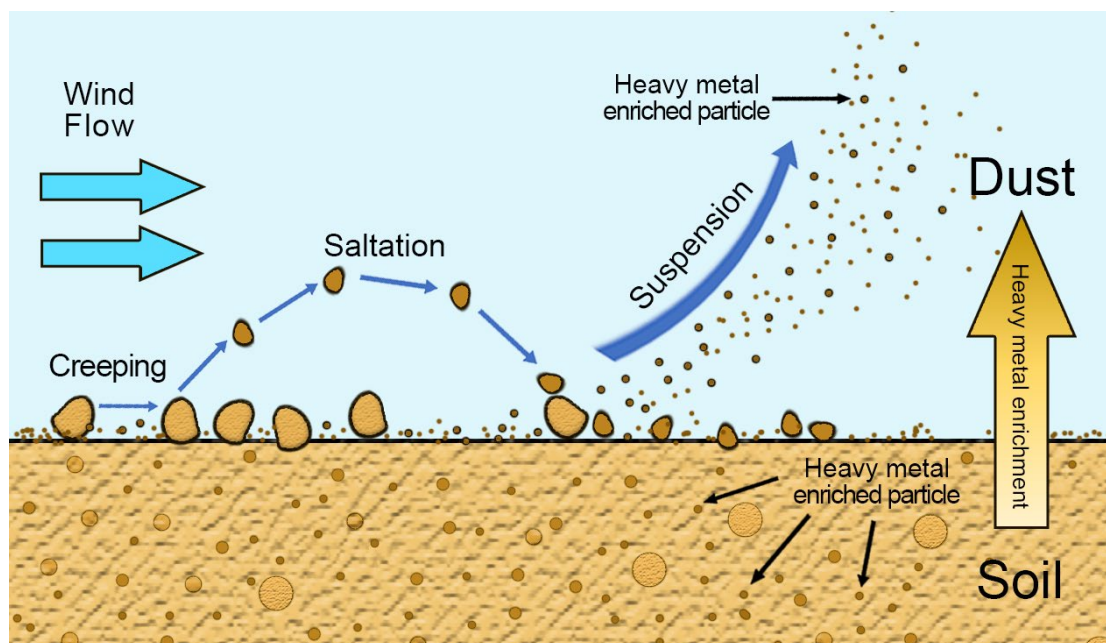
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28

29 **Abstract**

30 Dust is a major source of atmospheric aerosols. Its chemical composition is often assumed to be  
31 similar to the parent soil. However, this assumption has not been rigorously verified. Here, we  
32 generated dust aerosols from soils to determine if there is particle size-dependent selectivity of  
33 heavy metals in the dust generation. Mn, Cd, Pb and other heavy metals were found to be highly  
34 enriched in fine ( $PM_{2.5}$ ) dust aerosols, which can be up to ~6.5-fold. To calculate the contributions  
35 of dust to atmospheric heavy metals, regional air quality models usually use the dust chemical  
36 profiles from the US EPA's SPECIATE database, which does not capture the correct size-dependent  
37 selectivity of heavy metals in dust aerosols. Our air quality modeling for China demonstrates that  
38 the calculated contribution of fine dust aerosols to atmospheric heavy metals, as well as their cancer  
39 risks, could have significant errors without using proper dust profiles.

40 **Graphical Abstract**



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42

43 **Short Summary**

44 Dust is a major source of atmospheric aerosols. Its chemical composition is often assumed to be  
45 similar to the parent soil. However, this assumption has not been rigorously verified. Dust aerosols  
46 are mainly generated by wind erosion, which may have some chemical selectivity. Mn, Cd and Pb  
47 were found to be highly enriched in fine (PM<sub>2.5</sub>) dust aerosols. In addition, estimation of heavy  
48 metal emission from dust generation by air quality models may have errors without using proper  
49 dust profiles.

50

## 51 **1 Introduction**

52 The major sources of natural aerosols include mineral dust aerosols produced by wind erosion  
53 (Prospero et al., 2002). Dust aerosols are influenced by regional atmospheric circulation, soil  
54 characteristics and local weather conditions (Bryant, 2013; Ding et al., 2005; Huebert et al., 2003;  
55 Liu et al., 2004; Yang et al., 2008), mainly generated and aerosolized when strong wind passes over  
56 soil or sandy areas (Gillette and Goodwin, 1974). Recent studies show mineral dust aerosol accounts  
57 for approximately 40% of the mass fraction of natural atmospheric aerosol, with an estimated annual  
58 flux of  $\sim 2,000 \text{ Tg}\cdot\text{yr}^{-1}$  (Alfaro, 2008; Griggs and Noguera, 2002; Huneus et al., 2011; Textor et al.,  
59 2006). As the second-largest natural source of atmospheric aerosols in terms of mass flux, dust  
60 aerosol has a profound impact on the ecosystem (Middleton et al., 2019), especially the climate  
61 (Evan et al., 2014; Kok et al., 2018; Shao et al., 2013). Interactions between dust aerosols and water  
62 vapor play a critical role in cloud condensation and ice nucleation processes (Kaufman et al., 2002;  
63 Tang et al., 2016). Dust particles can be transported on large scales (Shao and Dong, 2006), and  
64 could act as a medium to transport toxic compounds, including heavy metals, which significantly  
65 harm human health, particularly the human respiratory system and even cause premature death  
66 (Urrutia-Pereira et al., 2021).

67 Atmospheric studies often assume that the chemical composition of aerosolized dust is similar to  
68 the parent soil (Gunawardana et al., 2012; Zhuang et al., 2001). The chemical composition of dust  
69 aerosol consists of a key part in source apportionment modeling (Balakrishna and Pervez, 2009;  
70 Samiksha et al., 2017; Santos et al., 2017; Ying et al., 2018). A critical approach in source  
71 apportionment modeling is the chemical transport model, which predicts the dust aerosol on global

72 and regional scales based on the prior knowledge of source emission, atmospheric transport, and  
73 chemical reaction process. SPECIATE is the EPA's speciation profiles repository of air pollution  
74 sources of volatile organic compounds (VOCs) and particulate matter (PM). Therefore, the US  
75 EPA's SPECIATE database is an important product to convert total emissions from specific sources  
76 into the speciated emissions needed for the chemical transport model. The previous study has  
77 combined the US EPA's SPECIATE database and air quality model to predict dust aerosols (Ying et  
78 al., 2018), based on the assumption of the chemical composition of dust aerosols is similar to the  
79 resuspended soil profiles.

80 Yet, dust generation and aerosolization are complex processes, which may have some chemical  
81 selectivity. Most small dust particles ( $< 20 \mu\text{m}$ ) are produced either by wind erosion, which leads to  
82 soil movements such as creeping, saltation, and suspension (Burezq, 2020) or sandblasting process,  
83 which leads soil particles ( $\sim 75 \mu\text{m}$ ) to be lifted by the wind, move in ballistic trajectories due to the  
84 combined effect of aerodynamic force and gravity force (Grini and Zender, 2004; Shao and Raupach,  
85 1993; Shao et al., 1996). The sandblasting efficiency of a soil particle is highly sensitive to its size  
86 (Grini and Zender, 2004; Grini et al., 2002). In addition, the chemical composition of soil particles  
87 can also vary with particle size. As smaller soil particles are more easily ejected, dust aerosol  
88 particles are unlikely to have exactly the same composition as their parent soils (Perlwitz et al., 2015;  
89 Wu et al., 2022). Dust deposited samples were the dust samples collected on the road or other  
90 surfaces using a brush and plastic tray (Shangguan et al., 2022), while dust aerosol samples were  
91 collected by filtering the air. Dust aerosols were produced by the ballistic impacts of wind-driven  
92 sand grains (Kok et al., 2023). Indeed, some previous studies do find that in the deposited dust  
93 samples (not dust aerosol samples), smaller particles tend to contain higher amounts of heavy metals

94 (Naderizadeh et al., 2016; Parajuli et al., 2016; Becagli et al., 2020). However, the heavy metal  
95 profiles for dust aerosols from the US EPA's SPECIATE database seem to have no such enrichment  
96 between each particle size, as Table S1 reports profile 41350 as an example. Although these profiles  
97 have been widely used in air quality modeling works (Lowenthal et al., 2010; Simon et al., 2010;  
98 Ashrafi et al., 2018), they were actually measured in the 1970s and 1980s with the resuspension of  
99 soil samples, which placed soil in a glass tube and drew air flow to blow and suspend the soil  
100 particles to the air (Miller et al., 1972). This method is not likely to produce realistic dust aerosols,  
101 as it does not simulate sandblasting process properly. It is not known whether using such a  
102 problematic dust profile could significantly impact air quality model calculations.

103

104 Here we examined the enrichment of heavy metals in the laboratory-generated dust aerosols. A dust  
105 aerosol generator that mimics realistic sandblasting and saltation was used to generate dust aerosol  
106 from a collection of soil samples (Lafon et al., 2014). The concentrations of heavy metals in soil  
107 and dust aerosols were measured by an inductively coupled plasma mass spectrometer (ICP-MS).  
108 In this study, some heavy metals, such as Mn, Cd, Zn and Pb, were found to be highly enriched in  
109 dust aerosols. Especially, the enrichment factors would be much higher for smaller dust aerosols. In  
110 addition, we also utilized a single particle aerosol mass spectrometer (SPAMS) to study heavy metal-  
111 containing dust aerosols before, during, and after a dust storm. Regional air quality models usually  
112 use problematic dust composition profiles from the US EPA's SPECIATE database. Herein we  
113 modeled the contribution of dust aerosol to atmospheric heavy metal loadings, utilizing a range of  
114 dust aerosol profiles determined in this laboratory study as well as the SPECIATE profile, to  
115 investigate whether using a proper dust profile is critical to air quality modeling and cancer risk

116 calculations.

## 117 **2 Materials and methods**

### 118 **2.1 Soil sample collection**

119 Fourteen samples were collected from the top 10 cm of the natural soil profile from various locations  
120 in dust source regions and Shanghai, China (Table S2, Fig. S1). S1-S4 were collected from dust  
121 sources on the northern slope of Yinshan Mountain in central inner Mongolia and the adjacent areas  
122 of the Hunshandake Sandy Land, S5-S12 were collected from dust sources of Hexi Corridor and  
123 Alxa Plateau, S13 was collected in Xinjiang Province, in the dust sources of the Taklimakan Desert,  
124 and S14 was sampled from Shanghai Yangpu District. Although the soil (S14) collected in Shanghai  
125 does not originate from a dust source region, it can still produce dust aerosols in some cases. For  
126 example, under dry weather conditions, the soil surface in the Shanghai area could serve as a  
127 significant local contributor to the generation of dust aerosols (Liu et al., 2016; Liu et al., 2020).  
128 During the prevailing dust storm periods from March to May, Shanghai is primarily influenced by  
129 dust originating from the western Inner Mongolia Gobi, deserts in the Tibetan Plateau, and arid  
130 deserts in northwest China (Fu et al., 2010; Fu et al., 2014; Sun et al., 2017). Soil texture  
131 determination was conducted according to the method outlined in a previous study (Kettler et al.,  
132 2001). Soil texture characterization was conducted based on the method outlined in a previous study  
133 (Kettler et al., 2001). Soil particle dispersion was achieved by adding hexametaphosphate (HMP)  
134 and sodium hydroxide (NaOH) to a soil sample (particle size < 2 mm) and shaking it for 16 hours.  
135 The percentage of sand and silt was obtained using a Laser Scattering Particle Size Distribution  
136 Analyzer (LA-960). Further details can be found in the SI. As shown in Table S2, they represent

137 several soil types: S1 was silty loam; S2, S4, S7, S10, S11 and S12 were sand; S3 was sandy loam;  
138 S5 and S6 were loam; S8 and S13 were loam sand; S9 and S14 were silty clay loam. Before dust  
139 aerosol generation, soil samples were placed in a fume hood and left to dry, without stirring or other  
140 treatment, before aerosolization. Fine and coarse dust aerosols (PM<sub>2.5</sub> and PM<sub>10</sub>) were produced  
141 with a GAMEL dust aerosol generator, which can realistically simulate the sandblasting process.  
142 Then, the pH of the soil was measured. Detailed information can be found in [Fig. S1](#) and [Table S2](#).

## 143 **2.2 Laboratory dust aerosol generation and collection**

144 A laboratory dust generator (GAMEL: “Générateur d’Aérosol Minéral En Laboratoire”) ([Lafon et](#)  
145 [al., 2014](#)) was used to produce dust aerosols from the soil samples. The GAMEL dust generator can  
146 realistically simulate the sandblasting process. Wind tunnels have the advantage of realistically  
147 simulating the generation of dust aerosols. However, conducting this study has certain drawbacks.  
148 These include the requirement for a substantial quantity of parent soils and the significant cost  
149 associated with eliminating ambient aerosol interference ([Alfaro et al., 1997](#); [Lafon et al., 2006](#);  
150 [Alfaro, 2008](#)). In GAMEL's dust production system, 10 g of each soil sample was added to a PTFE  
151 flask, which was agitated by a shaker simulating the sandblasting process to produce dust aerosols.  
152 A constant flow of particle-free air was passed through the dust-generating flask. The optimal  
153 generation parameter of the shaker was set at a frequency of 500 cycles/min according to [Lafon et](#)  
154 [al., 2014](#) with an airflow rate of 8 liter/min controlled by a Mass Flow Controller (MFC, Sevenstar,  
155 Beijing Sevenstar Flow Co., LTD). The sample stream was filtered through a cyclone and particles  
156 were collected on a 47 mm PVC film held in a metal frame filter holder (Pall Gelman, Port  
157 Washington, NY, USA). Dust-PM<sub>2.5</sub> and dust-PM<sub>10</sub> were obtained with or without an 8LPM cyclone,



158 respectively. The running time was 1 min. To obtain more dust aerosols in different size ranges, size-  
159 fractionated particle sampling of dust aerosols was carried out with a 10-stage Micro-Orifice  
160 Uniform Deposit Impactor (MOUDI 110R; MSP) with size cut points of 10  $\mu\text{m}$ , 5.6  $\mu\text{m}$ , 3.2  $\mu\text{m}$ ,  
161 1.8  $\mu\text{m}$ , 1.0  $\mu\text{m}$ , and 0.56  $\mu\text{m}$ . Analysis of the size distribution and chemical composition of dust  
162 generated by GAMEL and dust generated under natural conditions has shown that the GAMEL  
163 generator can produce realistic dust aerosol (Lafon et al., 2014). All the dust aerosol mass collected  
164 is shown in Table S3 and S4. The instrument setup is illustrated in Fig. S2.

165

### 166 **2.3 Analysis of laboratory-generated dust aerosols**

167 The dust aerosol samples collected were weighed with an analytical balance and then put into 25 ml  
168 digestion tubes with 6 ml 69%  $\text{HNO}_3$  symmetrically. The temperature program of Microwave  
169 Digestion (Anton Paar) was as follows: initial temperature of 100  $^\circ\text{C}$  held for 5 min, then ramped  
170 to 140  $^\circ\text{C}$  for 5 min, and finally at 180  $^\circ\text{C}$  for 60 min. The whole process was holding 120 min.  
171 According to this study (Chang et al., 1984), almost all the heavy metal elements in the natural soil  
172 and dust aerosol in concentrated nitric acid were extracted using this experimental procedure. After  
173 digestion, the solution was acid-fed at 120  $^\circ\text{C}$  for 1.5 h, then deionized water (conductivity 18.25  
174  $\text{M}\Omega$ ) was added, the volume was constant with a 25 mL volumetric flask and then passed through a  
175 0.45  $\mu\text{m}$  membrane. The samples were diluted with 2%  $\text{HNO}_3$  by 4 times for further analysis. Three  
176 blank PVC film samples were digested using the same method for background control.

177

178 The heavy metal content was determined by an inductively coupled plasma mass spectrometer (ICP-

179 MS; Agilent, 8900). Before analysis, tuning procedures including plasma parameter, ion  
180 transmission path, quadrupole mass spectrometer, and detector had been done. During analysis,  
181 standard solutions were prepared at concentrations of 0, 1, 2, 5, 10, 20, 50, and 100 µg/L. "In, Bi,  
182 and Rn" were used as internal standard elements, and were introduced into the nebulizer by mixing  
183 with the sample to be tested and the standard solution in the sampling pipeline by online addition,  
184 and the instrument drift and matrix effect were compensated. After each analysis of a sample, 2%  
185 dilute nitric acid was used to clean the injection line for 1 min, and then continue to collect the  
186 second sample to eliminate the memory effect of the previous sample.

187 A scanning electron microscope (SEM; Phenom Pro) equipped with an energy-dispersive X-ray  
188 detector was used for morphologies of particle examination at the voltage of 10 kV. All the samples  
189 (soil, PM<sub>2.5</sub> and PM<sub>10</sub>) were on the carbon conductive adhesive, then spray platinum to improve the  
190 conductivity. Here, the parent soil of S10 and generated PM<sub>2.5</sub> and PM<sub>10</sub> were examined.

191 Statistical analysis was performed using SPSS Statistics. The correlation analysis was conducted  
192 through Spearman's correlation and the significant difference was used with an independent sample  
193 T-test.

#### 194 **2.4 Ambient dust aerosol measurements**

195 On May 23<sup>rd</sup>, 2018 (LT), on-site field measurements were conducted in Shanghai to assess the  
196 ambient dust particles. The measurements indicated an average wind speed of 2.2 m/s, which  
197 corresponds to a level of a floating dust storm with a visibility of up to 10 km. The sampling was  
198 located on the sixth floor of the Environmental Science Building in Jiangwan Campus, Fudan  
199 University, a typical residential area in a heavily polluted urban area. The chemical composition of

200 individual ambient particles was measured by single-particle aerosol mass spectrometry (SPAMS,  
201 Hexin Co., Ltd). Detailed information on SPAMS is available elsewhere (Li et al., 2011). An  
202 adaptive resonance theory-based clustering method (ART-2a) was used to classify the mass spectra  
203 generated and identify dust/heavy-metal-containing particles (Sullivan et al., 2007). The Hybrid  
204 Single-Particle Lagrangian Integrated Trajectory HYSPLIT-4 model developed by the ARL (Air  
205 Resources Laboratory) of the NOAA (National Oceanic and Atmospheric Administration), USA,  
206 was employed to compute hourly resolved 48 h air mass backward trajectories at 500 m arrival  
207 height (Lv et al., 2021; Pongkiatkul and Kim Oanh, 2007).

208

## 209 **2.5 Air quality model configuration and application**

210 The source-oriented Community Multiscale Air Quality (CMAQ) model v5.0.1 with an expanded  
211 Stratospheric and Air Pollution Research-99 (SAPRC-99) photochemical mechanism was applied  
212 to simulate PM<sub>2.5</sub> levels and track the sources of primary PM<sub>2.5</sub> (PPM<sub>2.5</sub>) in China during the entire  
213 year of 2013 (Guenther et al., 2012; Ying et al., 2018). The simulation domain covered China and  
214 its surrounding countries, with a horizontal resolution of 36 × 36 km<sup>2</sup> (127 × 197 grids).  
215 Anthropogenic emissions were based on the Multi-resolution Emission Inventory for China (MEIC,  
216 v1.3, 0.25° × 0.25°, <http://www.meicmodel.org>). Biogenic emissions were generated by the Model  
217 of Emissions of Gases and Aerosols from Nature (MEGAN) v2.1 (Guenther et al., 2012). The  
218 meteorological inputs for the CMAQ model were calculated by the Weather Research and  
219 Forecasting (WRF) model (<https://www2.mmm.ucar.edu/wrf/users>).

220

221 Five major source contributions (windblown dust, residential, transportation, power generation and  
222 industrial sources) to PM<sub>2.5</sub> were investigated based on the inventory-observation-constrained  
223 emission factors (Ying et al., 2018). Three control trials were conducted for each heavy metal  
224 according to measured soil, dust-PM<sub>2.5</sub> and the SPECIATE datasets from the four regions (three dust  
225 sources and Shanghai). It is worth noting that the emission factors for areas outside these four  
226 regions were estimated using Inverse Distance Weight (IDW) spatial interpolation methods. These  
227 methods were based on the dataset of emission factors within these four regions, which represent  
228 the amount of heavy metal emitted per kilogram of dust (Zhang and Tripathi, 2018). Each heavy  
229 metal source concentration from dust aerosol and all four sources were used to quantify the  
230 contribution on heavy metal concentrations in the atmospheric dust aerosols, which can be  
231 represented in Equation 1:

$$232 \quad R = \frac{E_i \times s_i \times a}{\sum_{i=1}^5 E_i \times s_i} \quad \text{Equation 1}$$

233 Where  $E_i$  is the PPM<sub>2.5</sub> emission from  $i^{th}$  source,  $s_i$  is the emission factor of the specific heavy metal  
234 from  $i^{th}$  source,  $a$  is the concentration of heavy metal in measured soil, dust-PM<sub>2.5</sub>, and the  
235 SPECIATE datasets.  $E_i$ ,  $s_i$ , and  $a$  are the values for dust.

236 In addition, the human health risk of heavy metals was assessed. Three main routes of chemical  
237 daily intake (CDI, mg kg<sup>-1</sup> day<sup>-1</sup>) of air heavy metals were: (1) direct ingestion of particles or gases  
238 existed in the air (CDI<sub>ing</sub>); (2) inhalation of suspended particles through mouth and nose (CDI<sub>inh</sub>);  
239 and (3) daily absorption of heavy metals through the skin (CDI<sub>dermal</sub>) (Luo et al., 2012). To assess  
240 the carcinogenic and non-carcinogenic effects of heavy metals, we evaluated these effects in 13 age  
241 groups ranging from birth to ≤80 years old. These age groups are as follows: <1, 1 to <2, 2 to <3, 3  
242 to <6, 6 to <11, 11 to <16, 16 to <20, 21 to <31, 31 to <51, 51 to <61, 61 to <71, 71 to <81, and ≥81

243 years (Gholizadeh et al., 2019b). The variables and values used for estimating human exposure to  
 244 heavy metals were obtained from the U.S. Environmental Protection Agency (USEPA) and the U.S.  
 245 Department of Energy (USDoE) (Moya et al., 2011; Doe, 2011).  $CDI_{ing}$ ,  $CDI_{inh}$ , and  $CDI_{dermal}$   
 246 were calculated as:

$$248 \quad CDI_{ing} = C \times \frac{IR_{ing} \times EF \times ED}{BW \times AT} \times 10^{-6} \quad \text{Equation 2}$$

$$249 \quad CDI_{dermal} = C \times \frac{SA \times AF \times ABS_d \times EF \times ED}{BW \times AT} \times 10^{-6} \quad \text{Equation 3}$$

$$250 \quad CDI_{inh} = C \times \frac{IR_{inh} \times ET \times EF \times ED}{BW \times AT} \times 10^{-6} \quad \text{Equation 4}$$

251 Moreover, the total carcinogenic risk  
 252 (TCR) for each heavy metal were calculated by:

$$253 \quad \text{carcinogenic risk} = CDI_{ing,dermal,inh} \times CSF \quad \text{Equation 5}$$

$$254 \quad TCR = \sum risk = CDI_{ing} \times CSF_{ing} + CDI_{inh} \times IUR +$$

$$255 \quad CDI_{dermal} \times CSF_{ing}/ABS_{GI} \quad \text{Equation 6}$$

256  
 257 Here the  $IR_{ing}$  was Ingestion rate ( $\text{mg day}^{-1}$ ),  $EF$  was exposure frequency ( $\text{day year}^{-1}$ ),  $ED$  was  
 258 exposure duration (year),  $BW$  was body weight (kg),  $AT$  was Averaging time (day),  $SA$  was total  
 259 body skin surface area ( $\text{m}^2$ ),  $AF$  was skin adherence factor ( $\text{mg cm}^{-2}$ ),  $ET$  was exposure time (hour  
 260  $\text{day}^{-1}$ ),  $ABS_d$  was dermal absorption factor,  $IR_{inh}$  inhalation rate ( $\text{m}^3 \text{day}^{-1}$ ),  $ABS_{GI}$  was  
 261 gastrointestinal absorption factor,  $CSF$  was cancer slope factor. The values of these parameters could  
 262 be found in the previous study (Gholizadeh et al., 2019a).

263

## 264 **3 Results and discussion**

### 265 **3.1 Enrichment of heavy metals in fine dust aerosols**

266

267 [Fig. S3-S4](#) show the absolute concentrations of heavy metals in dust aerosols and their parent soils.

268 The concentrations of heavy metals in dust-PM<sub>10</sub> were similar to soil concentrations, which showed  
269 a significant correlation between soils and PM<sub>10</sub> (p<0.01) ([Fig. S5](#)). While the concentrations of  
270 heavy metals in dust-PM<sub>2.5</sub> were higher than those in soils, especially Mn, Ni, Cu and Zn, showed  
271 significant differences (p<0.001) ([Fig. S6](#)). This trend was consistent across all soil samples. The  
272 enrichment factor (EF) of heavy metals in dust aerosols relative to the parent soils was calculated  
273 with Equation 7.

$$274 \quad EF = \frac{C_1/m_1}{C_0/m_0} \quad \text{Equation 7}$$

275 Where C<sub>1</sub> is the heavy metal concentration in dust-PM; m<sub>1</sub> is the mass of dust-PM collected on the  
276 filter; m<sub>0</sub> is the mass of soil in the ICP-MS sample, and C<sub>0</sub> is the heavy metal concentration of the  
277 soil.

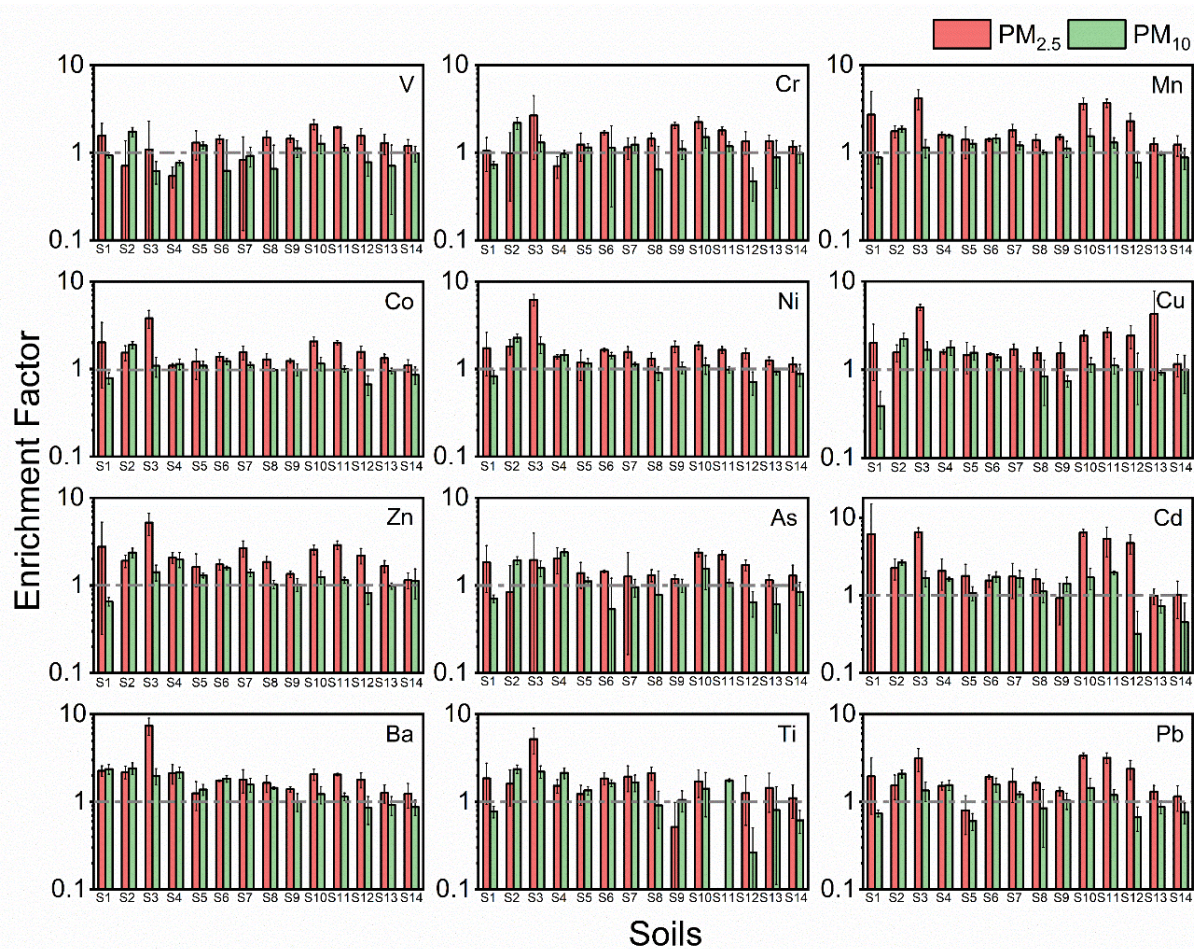
278

279 [Figures 1](#) and [S7](#) show that many heavy metals were highly enriched in fine dust aerosols (PM<sub>2.5</sub>),  
280 i.e., their absolute concentrations were significantly higher in fine dust particles than in the parent  
281 soil ([Fig. S6](#)). V, Cr, Mn, Co, Ni, Cu, Zn, As, Cd, Ba, Ti, and Pb were all enriched in dust-PM<sub>2.5</sub>  
282 during the process of dust formation. The following trend of heavy metal enrichment was  
283 established for dust-PM<sub>2.5</sub>: Cd > Zn > Ba > Cu > Mn > Pb > Ni > Ti > Co > As > Cr > V. Notably,  
284 the EFs of Cd were greater than 5 for soil S1, S10 and S11. No other literature has reported the  
285 enrichment of Cd or other heavy metals in dust aerosols. However, there is one study showing the

286 enrichment of water-soluble ions during dust aerosol production from soil (Wu et al., 2022). It  
287 reports that the EFs of  $\text{Ca}^{2+}$  ranged from approximately 5.6 to 223.1, and the EF values of  $\text{Mg}^{2+}$   
288 were between approximately 2.1 and 90.3 for dust- $\text{PM}_{2.5}$  from Sandy soils in the Taklamakan Desert.  
289 In this study, it is found that the EF of Cd and other metals falls within the range of EF for these  
290 water-soluble ions, consistent with the value reported by Wu et al., (2022). Fig. 1 also illustrates  
291 that all heavy metals were more highly enriched in smaller  $\text{PM}_{2.5}$  dust particles compared to larger  
292  $\text{PM}_{10}$  dust particles. For example, the Cd's EF reached ~6.4 and ~1.7 for dust- $\text{PM}_{2.5}$  and dust  $\text{PM}_{10}$ ,  
293 respectively, from soil S1. Most dust- $\text{PM}_{2.5}$  should originate from the small colloids in soil, which  
294 are defined as soil particles with less than 2  $\mu\text{m}$  in diameter. These soil colloids usually carry large  
295 amounts of negative charges, which can help adsorb many cations in soil, including various heavy  
296 metal ions (Brady and Weil, 2008). Thus, heavy metals are enriched in small soil aggregates. During  
297 the sandblasting process, the smaller soil grains, with higher heavy metal concentrations, are more  
298 likely to be ejected and form dust aerosols. The particle size dependence of heavy metal enrichment  
299 could have significant ramifications for the health impacts of dust aerosols. The dust aerosol size  
300 distribution of dust (Fig. S8) was also measured by an Aerodynamic Particle Sizer (APS,  
301 APS Model 3321; TSI Inc.; USA). It is found that the peak of the particle size distribution of dust  
302 aerosol was approximately at 2~3  $\mu\text{m}$ . Similarly, the scanning electron microscope (SEM) images  
303 of these dust aerosols (generated by S10) also show the presence of a large number of particles with  
304 sizes of 2~3  $\mu\text{m}$ . As particle size decreased, the shape of particles changed from flakes to rods,  
305 which means a larger surface area (Fig. S9). When examining the impact of soil texture on dust  
306 aerosol enrichment, first, notable variations were observed in the EF values from one soil texture,  
307 such as sandy soils, specifically S2, S4, S7, S10, S11, and S12. To assess the significance of these

308 variations, a one-way Analysis of Variance (ANOVA) was conducted using SPSS. In ANOVA, the  
309 *p-value* represents the probability of obtaining the observed differences in means (or more extreme  
310 differences) by random chance alone, assuming no true difference between the groups. A *p-value*  
311 below a predetermined significance level (commonly 0.05) indicates significant differences between  
312 the means of the compared groups. Specifically, for sandy soil, analysis results reveal significant  
313 variations between these six soils in terms of the EF values for both dust-PM<sub>2.5</sub> (*p-value*=0.004<0.05)  
314 and dust-PM<sub>10</sub> (*p-value*=0<0.05) (Table S5 and S6). These results indicate that there are significant  
315 differences in the EFs of heavy metals within the sandy soil group. Then, the variation between soil  
316 types was analyzed. For the six different types of soil samples, the results of ANOVA showed  
317 significant differences in the EFs of dust-PM<sub>2.5</sub> (*p-value* =0<0.05) and dust-PM<sub>10</sub> (*p-value* =0<0.05)  
318 among these soil types (Table S7 and S8). The differences among the six soils from different soil  
319 types were greater than those observed among the different soils in the same soil type, indicating a  
320 potential role of soil type in affecting EFs, which would require further study to elucidate. Detailed  
321 information was found in SI of [Texture S3](#) and [Table S5-S10](#).





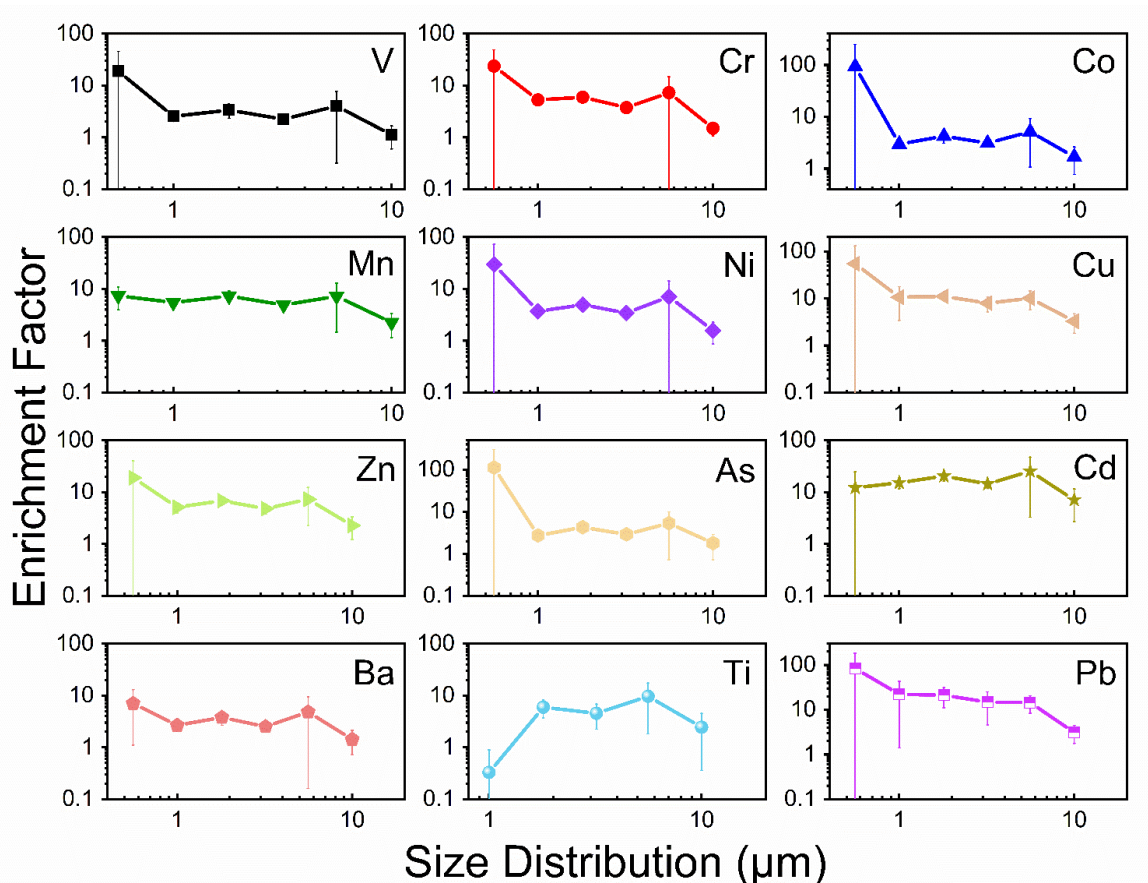
322

323 **Figure 1.** Enrichment Factors of PM<sub>2.5</sub> and PM<sub>10</sub>. Enrichment factors of heavy metals in dust  
 324 aerosols from soil S1-S14; red represents PM<sub>2.5</sub> and green represents PM<sub>10</sub>. The grey dotted line  
 325 represents the EF as 1. The whiskers on the bars represent the standard deviations of triplicates.

326

327 To investigate the link between dust particle size and heavy metal EFs in greater detail, a MOUDI  
 328 impactor was used to collect dust-PM from 0.56 to 10  $\mu\text{m}$  (absolute concentration obtained in [Fig.](#)  
 329 [S10](#)). Consistent with the results discussed above, the EFs for some heavy metals, such as Pb,  
 330 significantly increased with decreasing particle diameter ( $r = -1$ ,  $p < 0.01$ ) ([Fig. 2](#)). For the smallest  
 331 dust particles (0.56~1.0  $\mu\text{m}$ ), the EFs for Pb was approximately 83, an order of magnitude greater  
 332 than the EFs (~3) for the largest dust particles (>10  $\mu\text{m}$ ). V, Cr, Co, Mn, Ni, Cu, Zn, As, and Ba  
 333 show consistent trends, with EFs increasing as the particle size decreases. In detail, V (ranging from

334 ~1.1 to ~18.9), Cr (ranging from ~1.5 to ~23.7), Co (ranging from ~1.7 to ~93.7), Mn (ranging from  
 335 ~2.3 to ~7.4), Ni (ranging from ~1.6 to ~29.7), Cu (ranging from ~3.3 to ~54.3), Zn (ranging from  
 336 ~2.3 to ~19.0), As (ranging from ~1.8 to ~112.3), and Ba (ranging from ~1.4 to ~7.0), as the particle  
 337 size decreases from 10  $\mu\text{m}$  to 0.56  $\mu\text{m}$ . These results demonstrate that some heavy metals are indeed  
 338 enriched in smaller soil particles, which could be aerosolized during the sandblasting process. The  
 339 particle size dependence of heavy metal enrichment could have significant ramifications for the  
 340 health impacts of dust aerosols. In contrast, Cd's EFs remain relatively unchanged with varying  
 341 particle sizes. On the other hand, Ti exhibits an opposite trend, with EF values decreasing as the  
 342 particle size decreasing, and the reason for this difference requires further study.



343  
 344 **Figure 2.** Enrichment factors of heavy metals in dust aerosols with different particle size ranges.  
 345 The EF data were produced from the Soil S10, with diameters at above 10  $\mu\text{m}$ , 5.6-10  $\mu\text{m}$ , 3.2-5.6  
 346  $\mu\text{m}$ , 1.8-3.2  $\mu\text{m}$ , 1.0-1.8  $\mu\text{m}$  and 0.56-1.0  $\mu\text{m}$ . The whiskers on the bars represent the standard

347 deviations of triplicates.

348

### 349 **3.2 Modeling of the contributions of dust aerosols to atmospheric heavy metals** 350 **using the dust profiles from this study and the SPECIATE datasets**

351 It is necessary to know the sources of atmospheric heavy metals to effectively control their emission.

352 Air quality models with emission inventories can estimate the contributions of various sources to

353 atmospheric heavy metals. However, when estimating heavy metal emissions from dust production,

354 some widely used air quality models, such as the CMAQ model, typically use dust profiles from the

355 US EPA's SPECIATE datasets. As discussed in the introduction, this dust profile may be outdated

356 and cannot reflect realistic dust compositions. We used the CMAQ model to assess the potential

357 impact of dust aerosol profile in atmospheric dust aerosol using our measured profile and the profile

358 (No. 41350) from the SPECIATE datasets. The model tracked heavy metals in PM<sub>2.5</sub> in China for

359 the year 2013 (see Methods) from five major sources: windblown dust, residential, transportation,

360 power generation, and industry.

361

362 [Figure 3](#) shows the modeled contributions of the dust source to the Cr and Pb concentrations in

363 PM<sub>2.5</sub> for China, using the measured soil, dust-PM<sub>2.5</sub> profiles from this study, as well as the

364 SPECIATE composition profiles (see Methods). In addition, the modeled results for other metals,

365 such as As, Cu, Mn, Ti, and Zn were presented in [Fig. S11-15](#).

366

367 For atmospheric Cr, it is clear that the scenario of applying SPECIATE database significantly

368 underestimates the contribution of dust aerosol, with the highest value of  $\sim 0.08 \mu\text{g}/\text{m}^3$ , when

369 compared to the scenario of applying the measured dust-PM<sub>2.5</sub> profiles, which had the highest value  
370 of  $\sim 0.14 \mu\text{g}/\text{m}^3$ . For Pb, as shown in the right column of Fig. 3, the scenario of applying  
371 SPECIATE profile overestimates the contribution of dust aerosol, with the value up to  $\sim 0.4 \mu\text{g}/\text{m}^3$ ,  
372 when compared to the scenario of applying the measured dust-PM<sub>2.5</sub> profiles, which had the highest  
373 value of  $\sim 0.14$ . Uncertainties associated with the use of SPECIATE have also been identified in  
374 previous studies (Ho et al., 2003; Xia et al., 2017). Specifically, the dust PM<sub>2.5</sub> source profiles  
375 obtained from local studies indicated that SPECIATE overestimated the contributions of  
376 atmospheric K and Al by approximately 23%, while underestimating the contributions of Ca and  
377 Na by 50%. Additionally, the model represents the annual average data for the year 2013. Although  
378 there are some field studies conducted in the same year (Wang et al., 2021; Shi et al., 2018), there  
379 is no available annual average data for a direct comparison with the model results. These results  
380 demonstrate that the modeled heavy metal distribution in the atmosphere is quite sensitive to the  
381 input of dust composition profile, strongly suggesting that using a proper dust composition profile  
382 is a key in such air quality modeling.

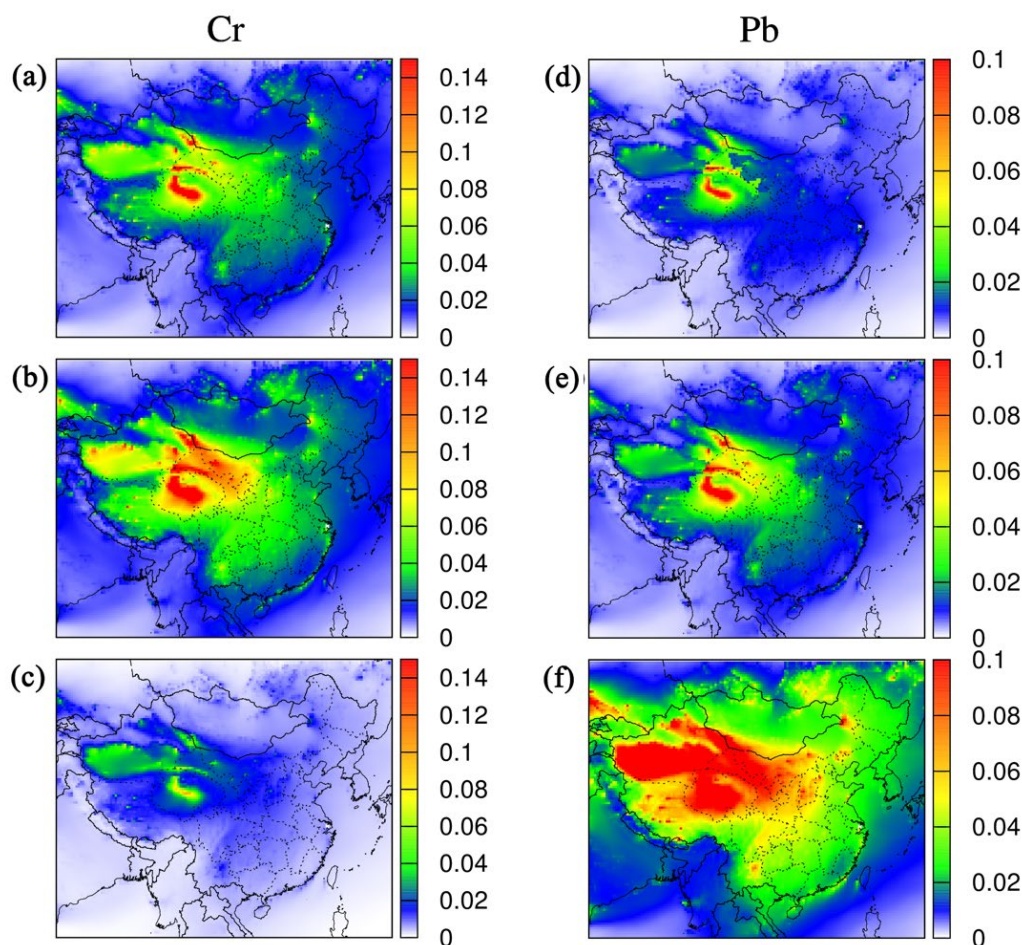
383

384 As discussed in the Introduction, many atmospheric studies assume that dust aerosol composition  
385 is similar to the composition of its parent soil. Here we also use the soil composition as an input  
386 dust profile in the model calculation to see how the modeled results are compared to that using the  
387 dust-PM<sub>2.5</sub> profile. For Cr, an obvious elevation of contribution was found by comparing the map  
388 using soil (a) and dust-PM<sub>2.5</sub> (b) profiles, with the hotspots of contribution ( $\sim 0.14 \mu\text{g}/\text{m}^3$ )  
389 distributed in northwest China. The region with dust aerosol contribution ranged from 0.02 to 0.08  
390  $\mu\text{g}/\text{m}^3$  covers most areas in China by using the dust-PM<sub>2.5</sub> profile. In contrast, the application of

391 the soil profile to the model reveals a significantly reduced area where the modeled Cr concentration  
392 from dust aerosols falls within the range of 0.02 to 0.08  $\mu\text{g}/\text{m}^3$ . For Pb, a significant difference is  
393 also found. The high contribution areas are also mainly distributed in northwest China for scenarios  
394 of applying soil and dust profiles, with a value up to 0.1  $\mu\text{g}/\text{m}^3$ . While the area with low dust  
395 aerosol contribution ( $<0.02 \mu\text{g}/\text{m}^3$ ) shrinks considerably in the scenario of applying soil profile.

396

397 The applied dust enrichment factors to modeled Cr in  $\text{PM}_{2.5}$  had an even stronger impact on modeled  
398 source apportionment (Fig. 3a-3b). The average dust source contribution to the total  $\text{PM}_{2.5}$  Cr  
399 concentration over China was calculated to be 0.03, and 0.05  $\mu\text{g}/\text{m}^3$  in the scenarios of applying  
400 soil and dust profiles, respectively. The model results for As, Cu, Mn, Ti and Zn (Fig. S11-S15) also  
401 show similar trends, indicating applying realistic enrichment factors to heavy metal concentrations  
402 in fine dust aerosols is critical to accurately model the sources of atmospheric heavy metals. These  
403 results demonstrate that it is not appropriate to assume dust aerosol composition is equal to soil  
404 composition, at least in air quality modeling.



405

406 **Figure 3.** Modeling of the contributions of dust aerosols to atmospheric Cr and Pb concentrations.

407 These results use the dust profiles of measured soil (a, d), dust-PM<sub>2.5</sub> (b, e), and the SPECIATE  
 408 datasets (c, f). The unit is  $\mu\text{g}/\text{m}^3$ .

409 [Figure 4](#) shows the Total Carcinogenic Risk (TCR) of the modeled atmospheric heavy metals (Cu,

410 Pb and Zn) for each province in Mainland, China. The modeled results using the dust-PM<sub>2.5</sub> and the

411 SPECIATE profiles are compared here. The carcinogenic risks lower than  $10^{-6}$  are considered

412 negligible, and risks above  $10^{-4}$  are not accepted by most international regulatory agencies ([Cheng](#)

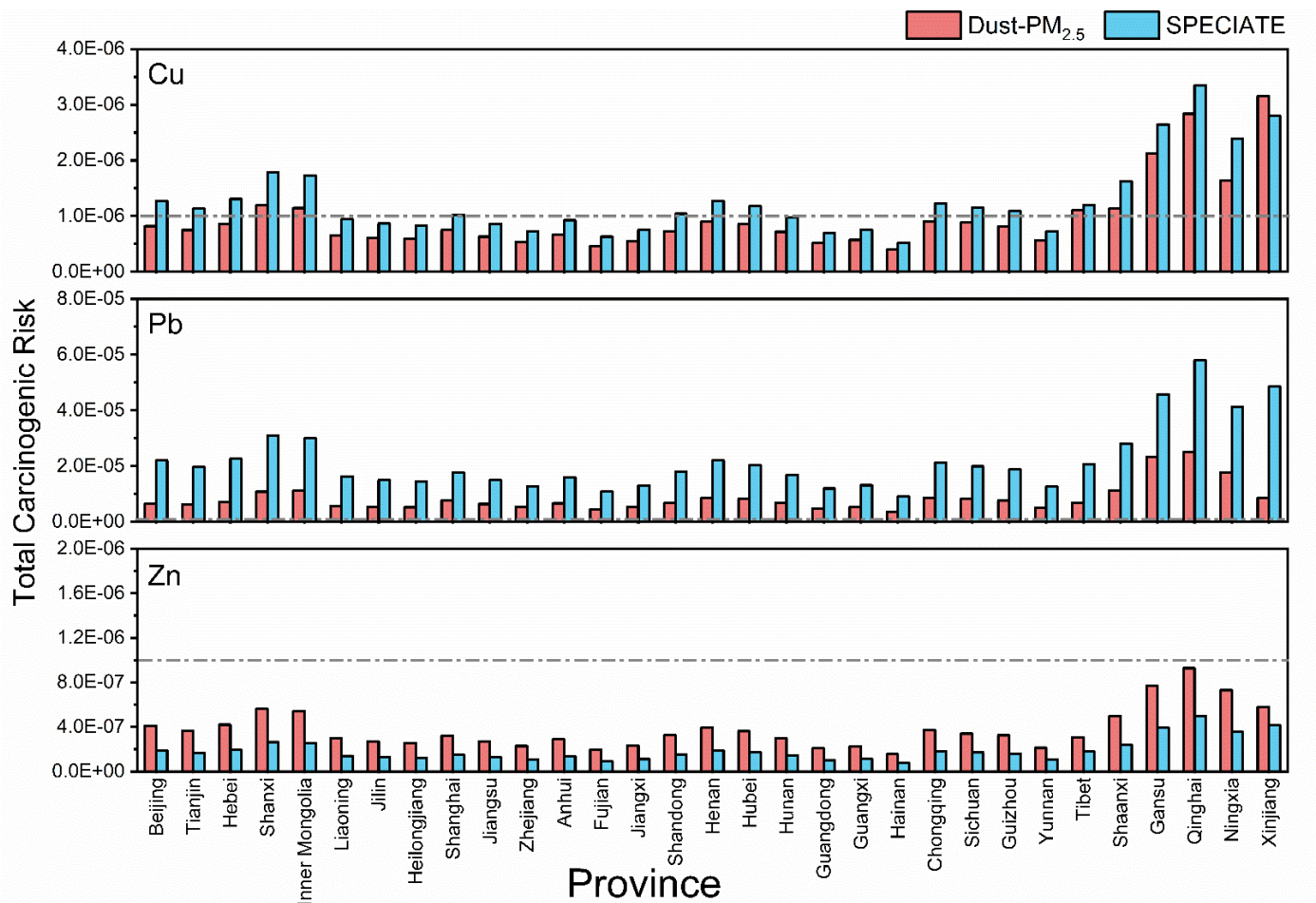
413 [et al., 2015](#); [Epa, 1989](#); [Luo et al., 2012](#)). For Cu, it is evident that using the SPECIATE profile

414 overestimated (the difference range up to  $\sim 7.5 \times 10^{-7}$ ) the TCR in China compared to using the dust-

415 PM<sub>2.5</sub> profile, as some regions exceed  $10^{-6}$ , the threshold value. For Pb, although all regions were

416 above  $10^{-6}$ , the TCR using the SPECIATE profile was greatly overestimated (the difference range is

417  $\sim 5.5 \times 10^{-6}$  -  $4.0 \times 10^{-5}$ ). The model results for Zn showed that all regions were not above  $10^{-6}$  but  
 418 significantly underestimated risks using the SPECIATE profile. This indicates that the health risk  
 419 assessment is also sensitive to dust composition profiles. Using the SPECIATE profile might be  
 420 problematic for assessing these risks.



421  
 422 **Figure 4.** Comparison of the total carcinogenic risk (TCR) of the modeled atmospheric heavy metals  
 423 for each province in Mainland, China between using the dust-PM<sub>2.5</sub> and SPECIATE profiles. Here,  
 424 the TCR of Cu, Pb and Zn were calculated. The grey dotted line is  $10^{-6}$ , the threshold value for  
 425 health concerns.  
 426

### 427 3.3 Field observation before, during and after a dust storm

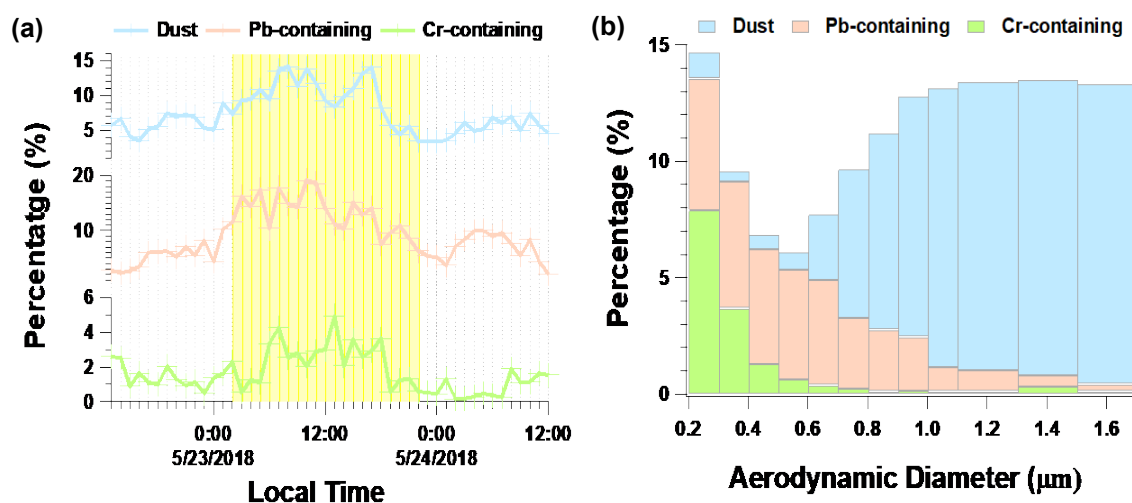
428 Our modeling results suggest that dust aerosol could be a major source of multiple heavy metals in  
429 PM<sub>2.5</sub> in China. Therefore, dust storms should significantly increase the concentrations of heavy  
430 metals in PM<sub>2.5</sub>. To test this idea, we studied a dust-storm plume, which originated from Mongolia  
431 and arrived in Shanghai (Huang et al., 2010) on 23 May 2018 (Fig. S16). Real-time single-particle  
432 mass spectra were generated by a single-particle mass spectrometer. Single-particle mass  
433 spectrometry can offer detailed information on the chemically-resolved mixing state at the single-  
434 particle level. According to the similarities of the mass-to-charge ratio and peak intensity of  
435 characterized signals, “Dust” particles were classified via an adaptive resonance theory-based  
436 clustering method (ART-2a, see Method). The number fraction of *Dust* particles was ~4.94% before  
437 and after the dust storm and it increased to ~9.73% during the dust storm episode (Fig. 5a).

438  
439 *Dust* particle mass spectra also contained ion markers indicative of an array of heavy metals ( $m/z$   
440 55[Mn<sup>+</sup>], 51[V<sup>+</sup>], 207[Pb<sup>+</sup>], 63[Cu<sup>+</sup>], 75[As<sup>+</sup>], 91[AsO<sup>+</sup>], 52[Cr<sup>+</sup>], -84[CrO<sub>2</sub><sup>-</sup>], -100[CrO<sub>3</sub><sup>-</sup>]) (red  
441 sticks in Fig. S17), indicating the existence of heavy metals in the ambient dust aerosols. The time  
442 series of Pb-containing and Cr-containing particle number fractions showed similar trends to the  
443 *Dust* particles. When the dust storm arrived, both Pb-containing and Cr-containing particle fractions  
444 increased as the dust cluster fraction increased. Before and after the dust storm, the percentages of  
445 Pb-containing and Cr-containing particles that overlapped with the *Dust* cluster were 41% and 32%,  
446 respectively. However, this overlapped ratio increased to 86% and 71% during the dust storm  
447 episode. The increase of heavy metal particles in step with the dust particles indicated that the dust  
448 particles could be the dominant source of these heavy metal species during this dust storm episode.



449

450 We further analyzed the size-resolved number fraction of dust aerosol, Pb-containing, and Cr-  
451 containing particles during the dust storm episode (Fig. 5b). The number fraction of *Dust* particles  
452 increased with increasing aerodynamic diameter. For particles above 1.0  $\mu\text{m}$ , *Dust* accounted  
453 for >12% of the total particles during the storm. However, the Pb-containing and Cr-containing  
454 particles made up a larger number fraction of analyzed particles with decreasing particle diameter  
455 size (< 1  $\mu\text{m}$ ). The number fractions of Pb-containing and Cr-containing particles were 5.7% and  
456 7.9% of all mass spectra for particles from 0.2-0.3  $\mu\text{m}$ . This result was consistent with our laboratory  
457 results that there is high heavy metal enrichment in smaller dust particles and our modeling results  
458 that dust aerosol is likely a major source of atmospheric Pb and Cr over China.



459

460 **Figure 5.** Ambient dust aerosol measurements. (a) Temporal variation of the percentages of dust  
461 aerosol, Pb-containing, and Cr-containing particle clusters. The yellow shadow represents the dust  
462 storm episode. (b) Size-resolved number fraction of dust aerosol, Pb-containing, and Cr-containing  
463 particle clusters.

464

## 465 **4 Environmental implications**

466 In this study, many heavy metals were found to be highly enriched in fine (PM<sub>2.5</sub>) dust aerosols  
467 compared to their concentrations in the parent soils. We propose that heavy metals tend to be  
468 enriched in smaller soil aggregates (Ikegami et al., 2014). During the sandblasting process, the  
469 heavy metal enriched smaller soil aggregates are more likely to be ejected and form dust aerosols.  
470 This work finds that dust aerosols from different soils may have a range of heavy metal enrichment  
471 factors. To study the transfer of heavy metals from soils to the air, it is critical to have a complete  
472 set of enrichment factors for each major soil type. There exists a difference among the heavy metal  
473 enrichment factors from different soil samples. The variability in the EFs is likely due to differences  
474 in soil properties (soil texture and size distribution etc.) which may affect the sandblasting/saltation  
475 process. For example, the enrichment factors of heaviest metals for Soil S1, S10 and S11 were  
476 higher than other soils. The detailed reason is still unknown and needs further exploration. Moreover,  
477 air quality models, including CMAQ models and various CMB models, often use the dust chemical  
478 profiles from the US EPA's SPECIATE to calculate the contribution of fine dust aerosols to  
479 atmospheric heavy metals, which are outdated and could lead to significant errors in estimating the  
480 emission of heavy metals through dust generation. Without using proper dust profiles in estimating  
481 heavy metal emissions from dust generation, the contribution of fine dust aerosols to atmospheric  
482 heavy metals, and its associated health risks are likely significantly mistaken.

## 483 **5. Conclusions**

484 Dust generation and aerosolization are complex processes that may have certain chemical selectivity.

485 Here, we deployed a laboratory generator to produce dust aerosol with a realistic sandblasting  
486 process. The concentrations of heavy metals (including V, Cr, Mn, Co, Ni, Cu, Zn, As, Cd, Ba, Ti,  
487 and Pb) in soils and fine (PM<sub>2.5</sub>) and coarse (PM<sub>10</sub>) dust aerosols were measured. With research  
488 efforts to elucidate the enrichment process of heavy metal in dust aerosols compared to their parent  
489 soils, our results fill the knowledge gaps of the compositional variation of heavy metal between the  
490 parent soils and the generated dust aerosols. Mn, Cd, Pb and other heavy metals were found to be  
491 highly enriched in fine (PM<sub>2.5</sub>) dust aerosols, which can be up to ~6.5-fold. These findings were  
492 also consistent with our field observation results. In addition, air quality models often use an  
493 outdated heavy metal profile for dust aerosols from the US EPA's SPECIATE database, which seems  
494 to lack enrichment between each particle size. We modeled the impact of the contribution of heavy  
495 metals in dust aerosol and their health risks in CMAQ, a widely used air quality model, and  
496 determined that atmospheric heavy metal concentrations over China, which drastically changed  
497 when we applied different dust profiles, such as the measured soil, dust-PM<sub>2.5</sub> profiles from this  
498 study, as well as the SPECIATE composition profiles. Our air quality modeling for China  
499 demonstrates that the calculated contribution of fine dust aerosols to atmospheric heavy metals, as  
500 well as their cancer risks, could have significant errors without using proper dust profiles.

## 501 **Supplement**

502 The supplement related to this article is available online at: <http://dx.doi.org/0.17632/byg6xk2fg9.1>.

## 503 **Data availability**

504 All data supporting this study and its findings will be available in an online data repository at:

505 <http://dx.doi.org/10.17632/wpphf8rd33.1>.

## 506 **Author contributions**

507 X.W. and J.C. conceptualized the work and designed the experiments. H.Z. and S.Z. led the air

508 quality modeling work. Q.G. lead the experimental work of heavy metal enrichment measurements.

509 J.Z. led the field observation. K.Z., Q.W., S.C., S.W., J.H., X.L. and H.C. helped in experimental

510 works. L.Z., L.W., Z.W., X.Y. and H.Z. helped in the experimental design and data analysis. Q.Y.

511 provided the data required for the air quality modeling. All authors contributed to the paper's writing.

512

## 513 **Competing interests**

514 The authors declare no competing interests.

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