

Impact of HO₂ aerosol uptake on radical levels and O₃ production during summertime in Beijing

Dyson et al.

Reply to Anonymous referee #2

We thank the referee for their comments, addressed individually below. The referee's comments are given in normal type with our response in italics. Changes to the manuscript are then given in red type.

1. Since the measured and modeled OH, HO₂, and RO₂ concentrations have been discussed in detail in Whalley et al., 2021, the authors should focus this paper on the impact of HO₂ uptake on the modeled concentrations. In that light, I would recommend removing Section 2.2 and referencing the Whalley et al., 2021 ACP paper (and updating the reference to the discussion paper).

We have revised the manuscript to move Section 2.2 to the SI.

2. I would also suggest moving the description of the LN/Q and absolute O₃ sensitivity calculation (lines 570-603) to section 2 after the model description, and instead focusing on the results in Section 3.

We have revised the manuscript, moving the description of LN/Q and O₃ sensitivity to section 2.

3. The authors provide a brief description regarding potential reasons for the discrepancy between the modeled radical concentrations with the measurements, which are discussed in detail in Whalley et al. (2021). However, at first read the description here does not appear to be consistent with the description in Whalley et al. For example, line 462 states that the overprediction of HO₂ by the model may be due to “an under-prediction in the rate of reaction of RO₂ with NO to produce a different RO₂ species...” while the conclusion in Whalley et al. 2021 is that the “propagation rate of RO₂ to HO₂ may be substantially slower than assumed.” While I believe the reasoning is consistent between the two papers, the wording here could be clarified to remove any potential confusion.

We agree with the referee and for clarity have revised line 460 onwards in the manuscript to read:

“the over-prediction of HO₂ could be due, in part, to the propagation rate of RO₂ to HO₂ being significantly slower than currently included in the model. This could be due to a lack of understanding of the rate of reaction of RO₂ with NO to produce different RO₂ species, i.e. RO₂ + NO → RO₂’, which would lead to propagation of RO₂ to different, more oxidised RO₂ species, competing with the recycling of RO₂ via RO₂ to give HO₂. It is also possible, that the overestimation in the propagation rate of RO₂ to HO₂ could be due to a lack of RO₂ autoxidation pathways included within the model which could lead to the formation of highly oxygenated molecules as opposed to HO₂. The higher, measured RO₂ concentrations could, therefore, suggest that the lifetime of total RO₂ is longer than currently considered within the model.”

4. Similar to that described in Sakamoto et al. (2019), the authors include uptake of RO₂ radicals into account when analyzing the impact of aerosol uptake on ozone production sensitivity. Given the authors suggestion that the overprediction of HO₂ and underprediction of RO₂ is due to isomerization of complex RO₂ or RO radicals that effectively increases the lifetime of RO₂ radicals and slows the propagation of RO₂ to HO₂, can the authors comment on whether uptake of RO₂ radicals in this scenario could impact the concentration of RO₂ radicals and the rate of ozone production? What effective lifetime of RO₂ radicals would heterogeneous uptake be competitive and impact RO₂ concentrations? Perhaps include a plot similar to Figure 8 for RO₂ loss to address this?

We would expect that if the uptake of RO₂ to aerosol is significant, then the concentration of RO₂ radicals within the model would decrease, in turn decreasing the HO₂ concentration due to an increase in RO₂ radical sinks. This would slow the propagation of RO₂ to HO₂, and therefore decrease the HO₂ radical overprediction within the model if this process is included. Looking at Table 4 in this work, at maximum 7.3 % of total loss of HO₂ in MCM_gamma is via uptake. Li et al., (2020) measured ambient uptake of isoprene-RO₂ and gave the mean loss of HO₂ onto aerosols as 0.0014 s⁻¹, just under double the loss rate of RO₂ onto aerosols, at 0.0008 s⁻¹. Looking at Figure 3 in Whalley et al., (2021), a full comparison of median production and destruction rates for OH, HO₂, total RO₂ and RO_x for the Beijing Summer campaign is given, showing the gas phase loss of RO₂ to be much higher than HO₂. This, coupled with the lower loss rate of RO₂ measured by Li et al., (2020), suggests the RO₂ uptake would not be significant to impact RO₂ concentrations for this campaign.

Li, J., Kohno, N., Sakamoto, Y., Pham, H.G., Murano, K., Sato, K., Nakayama, T. and Kajii, Y., 2022. Potential factors contributing to ozone production in AQUAS–Kyoto campaign in summer 2020: Natural source-related missing OH reactivity and heterogeneous HO₂/RO₂ loss. Environmental Science & Technology, 56(18), pp.12926-12936.

Whalley, L.K., Slater, E.J., Woodward-Massey, R., Ye, C., Lee, J.D., Squires, F., Hopkins, J.R., Dunmore, R.E., Shaw, M., Hamilton, J.F. and Lewis, A.C., 2021. Evaluating the sensitivity of radical chemistry and ozone formation to ambient VOCs and NO_x in Beijing. Atmospheric Chemistry and Physics, 21(3), pp.2125-2147.