



I In-depth study of the formation processes of single atmospheric particles in

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the southeastern margin of Tibetan Plateau

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14 Abstract

The unique geographical location of the Tibetan Plateau (TP) plays an important role in 15 regulating global climate change, but the impacts of the chemical components and 16 17 atmospheric processing on the size distribution and mixing state of individual particles are rarely explored in the southeastern margin of the TP, which is a transport channel for 18 19 pollutants from Southeast Asia during the pre-monsoon season. Thus a single-particle aerosol 20 mass spectrometer (SPAMS) was deployed to investigate how the local emissions of chemical composition interact with the transporting particles and assess the mixing state of 21 different particle types and secondary formation in this study. The TP particles were classified 22 into six main types: the rich-potassium (rich-K) type was the largest fraction of the total 23 particles (30.9%), followed by the biomass burning (BB) type (18.7%). Most particle types 24 25 were mainly transported from the surroundings and cross-border of northern Myanmar; but the air masses from northeastern India and Myanmar show a greater impact on the number 26 fraction of BB (31.7%) and Dust (18.2%) types, respectively. Besides, the two episodes 27 events with high particle concentrations showed that the differences in the meteorological 28 conditions in the same air clusters could cause significant changes in chemical components, 29 30 especially the Dust and EC-aged types changed by a sum of 93.6% and 72.0% respectively. 31 Ammonium and Dust particles distribute at a relatively larger size (~ 600 nm), but the size 32 peak of other types is present at ~ 440 nm. The easily volatilized nitrate ($^{62}NO_3^{-}$) during the 33 transport process leads the more abundant sulfate (97HSO₄⁻) to mix internally with the TP particles. $C_2H_3O^+$, $HC_2O_4^-$, NH_4^+ , NO_3^- , and HSO_4^- , severed as the indicators of secondary 34 formation, are present in the atmospheric aging process of photo-oxidation and aqueous 35 36 reaction by a linear correlation with O_x (O₃+NO₂) and relative humidity (RH). This study provides insights that can improve the knowledge of particle composition and size, mixing 37 state, and aging mechanism at high time resolution over the TP region. 38

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40 Keywords

41 Southeastern Tibetan Plateau, Pre-monsoon, Individual particles, Chemical characteristics,

42 Atmospheric aging,





43 1 Introduction

Atmospheric aerosols have complex components and sources and can be coated with 44 inorganic or organic materials during transport and atmospheric processing (Crippa et al., 45 46 2013). When further coated through coagulation, condensation, and photochemical oxidation, their physical and chemical properties and optical properties will change greatly, making their 47 impact on the air more uncertain (Jacobson, 2002; Zaveri et al., 2010; Matsui, 2016; 48 Budisulistiorini et al., 2017). The ability of aerosol particles to affect atmospheric conditions 49 50 is dependent on their sizes, chemical compositions, and mixing states (Mc Figgans et al., 51 2006; Dusek et al., 2006; Ma et al., 2012). For example, dust aerosol is an important factor 52 affecting climate change through the interactions of various physical processes such as the direct radiative effect (Mahowald et al., 2014; Shao et al., 2011; Wang et al., 2022); the 53 54 formation of secondary chemical components often aggravates regional pollution. However, the ability of secondary formation has regional differences due to the variations in precursors, 55 source strengths, and meteorological conditions (Pratt et al., 2011; Volkamer et al., 2006). 56 The influence of these complex chemical components on aerosol size and mixing state varies 57 58 with the pollution sources and/or atmospheric formation mechanism, which have been widely studied in urban of low altitude (Pratt et al., 2011; Liu et al., 2020a; Xu et al., 2017). 59 Furthermore, a previous study found that the migration or formation of low-volatile 60 component (such as nitrate and organic matter) could effectively be reduced due to 61 evaporation during the upward transportation process (Liu et al., 2020b), which further alter 62 63 the chemical compositions and the particle sizes. In addition, the transportation of the aerosols to a relatively cleaner environment prevails the formation of secondary chemicals at 64 a high altitude (Liu et al., 2020b). Therefore, a comprehensive investigation of the detailed 65 66 characteristic of aerosol formation and mixing states is required to understand their 67 environmental effects in low-, and high-altitude.

As a typical high-altitude region, the Tibetan Plateau (TP) has the highest and largest mountain area in the world, which is the most sensitive and obvious indicator of climate change in the entire Asian continent (Yao, et al., 2012; Chen and Bordoni, 2014; Immerzeel et al., 2010). In recent decades, many studies have shown that the melting and retreat of glaciers





72 in the TP regions is accelerating, and the main reason is attributable to anthropogenic emissions, such as greenhouse gases and aerosols (William et al., 2010; Luo et al., 2020; Hua 73 et al., 2019). Atmospheric aerosols also can influence the properties and life span of clouds as 74 75 cloud condensation nuclei, and affect local hydrological cycles and monsoon patterns by changing the microphysical properties of clouds (Oian et al., 2011; Seinfeld and Pandis, 2012, 76 77 Xia et al., 2007; Gettelman et al., 2013; Kumar et al., 2017). The southern part of the TP is always affected by the transport of more polluted air from South Asia along the mountain 78 valleys, especially during the pre-monsoon (i.e., March-May) with the southwest prevailing 79 80 wind (Cao et al., 2010; Chan et al., 2017; Zhu et al., 2017; Zhao et al., 2017; Han et al., 2020). Most studies have focused on the influence of optical properties; however, a few studies have 81 been conducted on aerosol components within the plateau. 82

Present aerosol components studies conducted in TP mostly focus on exploring the 83 influence of light-absorbing carbon aerosols and dust particles on climate change by optical 84 85 or offline sampling methods (e.g., Wang et al., 2019a; Liu et al., 2021). There is a lack of studies on the full composition, mixing states, and formation mechanism of aerosols in the 86 87 southeast margin and even the entire TP, especially using high-time resolved measurements. 88 Although time-integrated sampling with filter collection followed by laboratory analyses has been widely adopted for the chemical characterization of aerosols (Chen et al., 2015; Li et al., 89 90 2022a; Shen et al., 2015; Zhang et al., 2013). Drawbacks of the traditional approach include 91 low time resolution, high detection limit, and time- and labor-intensive procedures. More advanced aerosol online measurement devices with a high temporal resolution, such as the 92 aerosol chemical speciation monitor (ACSM) and aerosol mass spectrometer (AMS) (Ng et 93 94 al., 2011; Canagaratna et al., 2007) show the inability to measure the mixing state of particles and limit to obtain the information of non-refractory submicron aerosol. AMS/ACSM mainly 95 used to provide online observation datasets with high temporal resolution (including the mass 96 concentration of sulfate, nitrate, ammonium, chloride, and organic; and corresponding mass 97 98 spectral), which is beneficial to obtain the dynamic processes of source emission in the atmosphere (Du et al., 2015; Zhang et al., 2019a). At the same time, aerosol time-of-flight 99 mass spectrometry (ATOFMS) (Prather et al., 1994), and single particle aerosol mass 100 101 spectrometer (SPAMS) (Li et al., 2011), are popular for characterizing atmospheric individual





102 particles. These devices could determine the full chemical composition and their size distribution, to achieve more detailed information, such as the dynamic processes of chemical 103 aging, mixing state, and transport of the aerosols (Wang et al., 2016; Zhang et al., 2015, 104 105 2016). To the best knowledge, the advanced measurement device has not yet been applied for the studies conducted in TP, leading to a lack of in-depth research on the PM_{2.5} pollution in 106 TP, especially in the southeastern margin. The shortage of information hinders our 107 108 understanding of the distribution characteristics and formation mechanism of full aerosol components in high-altitude regions. 109

The southeastern margin of the TP is an important transitional zone between the 110 high-altitude TP and the low-altitude Yungui Plateau (Wang et al., 2019a; Zhao et al., 2017), 111 an ideal place for investigating the impacts of pollutants transport and formation in the 112 high-altitude zone. In this study, a high-time resolution field observation of individual 113 particles (SPAMS) was deployed on the southeastern margin of the TP during the 114 115 pre-monsoon, to continuously (i) investigate the changes of chemical characteristics between transport and local fine particles during pre-monsoon, (ii) determine the particle size 116 distributions, and the mixing states of different particle types, and (iii) assess the 117 118 contributions of photooxidation and aqueous reaction to the formation of the secondary 119 species. These results would expand our understanding of the chemical components, size 120 distribution, mixing state, and aging pathways of aerosols in the high-altitude areas in the TP 121 and surrounding areas.

122 2 Methodology

123 **2.1 Observation site**

Intensive 1-month field observation was deployed at the rooftop (~ 10 m above the ground) of the Lijiang Astronomical Station, Chinese Academy of Sciences (3260 m above sea level; 26°41′24″N, 100°10′48″E), Gaomeigu County, Yunnan Province, China, during the pre-monsoon period (from April 14th to May 13th, 2018). The nearest residential area is the Gaomeigu village (3–5 km away) with a small population size of 113 residents in 27 households. The villagers make a living by farming (e.g., potato and autumn rape), and biomass is the main residential fuel (Li et al., 2016). The site is surrounded by rural and





mountainous areas and has no obvious industry or traffic emissions. During the observation period, the average temperature (T) and relative humidity (RH) are 8.4 ± 3.1 °C and $69\% \pm 21\%$, respectively. The wind speed (WS) is 2.2 ± 1.2 m·s⁻¹ with the prevailing wind in the north and northeastern (Fig. S1).

135 **2.2 On-line instrument**

136 A detailed operational principle and the calibrations of the single-particle aerosol mass 137 spectrometer (SPAMS, Hexin Analytical Instrument Co., Ltd., Guangzhou, China) has been 138 described elsewhere (Li et al., 2011). Briefly, individual particles are drawn into SPAMS 139 through a critical orifice. The particles are focused and accelerated, then aerodynamically sized by two continuous diode Nd: YAG laser beams (532 nm), subsequently desorbed and 140 141 ionized by a pulsed laser (266 nm) triggered exactly based on the velocity of the specific particle. The generated of positive and negative molecular fragments are recorded with the 142 corresponding size of individual particles. In summary, a velocity, a detection moment, and 143 144 an ion mass spectrum are recorded for each ionized particle, while there is no mass spectrum 145 for not ionized particles. The velocity could be converted to $d_{\rm va}$ based on a calibration using polystyrene latex spheres (PSL, Thermo Scientific Corp., Palo Alto, USA) with predefined 146 sizes. The average ambient pressure is 690 hPa (in a range of 685-694 hPa) during the 147 measurements and calibration. Particles measured by SPAMS mostly are within the size 148 range of vacuum aerodynamic diameter (d_{va}) 0.2–2.0 µm. 149

Meteorological parameters, including the planetary boundary layer (PBL), temperature 150 (°C), RH (%), WS (m·s⁻¹), and wind direction (WD) were continuously achieved using an 151 automatic weather station (Model MAWS201, Vaisala HydroMet, Helsinki, Finland) at a time 152 resolution of 5 min. Gaseous concentrations (ppbv) were obtained using a multiple gas 153 analyzer (Thermo Scientific Corp.), including ozone (O3, model 49i) and nitrogen oxides 154 (NO_x, model 42i) in a 5-min resolution. The SPAMS and gas analyzers are co-located in the 155 same position, while the weather station was uncovered outside ~ 5 m from the sampling 156 house. Time series of SPAMS particles, gaseous concentrations (NO, NO_x, O₃, and CO) and 157 158 meteorological parameters (PBL, temperature, RH, WD, and WS) were shown in Fig. S2.

159 2.3 Individual particle classification





160 During the observation period, a total of 461,876 ambient particles with the size (d_{va}) of 0.2–2.0 µm were collected, including 55,583 in Episode I (E1; from April 18th 08:00 to April 161 19th 08:00) and 62,110 in Episode II (E2; from April 26th 17:00 to April 28th 02:00). The 162 analyzed particles are classified into 1,557 clusters using an adaptive resonance theory neural 163 network (ART-2a) with a vigilance factor of 0.8, a learning rate of 0.05, and 20 iterations 164 (Song et al., 1999). Finally, eight major particle clusters [i.e., Potassium-rich (rich-K), 165 Biomass Burning (BB), Organic Carbon (OC), Ammonium, Aging Element Carbon 166 (EC-aged), Dust, Sodium, Potassium-containing (NaK-SN), and Iron (Fe)-Lead 167 (Pb)-containing (Metal)] with distinct chemical patterns were manually combined, which 168 represent ~99.7 % of the population of the detected particles. The remaining particles are 169 grouped as "Other". The characteristics of the positive and negative mass spectra (MS) of 170 each particle type are shown in Fig. S3. A detailed description of classification criteria for 171 individual particles and the characteristic ion fragments for each particle type can be found in 172 173 Text S1.

174 2.4 Trajectory-related analysis

To determine the influence of regional transport on different particles at Gaomeigu, the 175 trajectory clusters analysis was carried out using the 72-h backward air mass trajectories at 176 500 m above the ground level. The trajectories were calculated with the Hybrid 177 Single-Particle Lagrangian Integrated Trajectory model (Draxler and Hess, 1998), and the 178 meteorological data were obtained from the Global Data Assimilation System (GDAS; 179 ftp://arlftp.arlhq.noaa.gov/pub/archives/gdas1, last access: 6 April, 2022). The cluster analysis 180 employs a Euclidean-oriented distance definition to differentiate and cluster the major spatial 181 features of the inputting trajectories. Details of the trajectory clustering method can be found 182 in Sirois and Bottenheim (1995). To investigate the effects of transport on the chemical 183 characteristic of the individual particles, trajectories with particle number concentrations high 184 than the 75th percentile are considered as pollution (Liu et al., 2021). 185

186 **3 Results and Discussion**

187 **3.1 Characteristics of particle composition**





188 Table 1 summarizes the numbers of concentrations, relative percentages, and characteristic ions of each particle type. The most dominant particle type in Gaomeigu during 189 pre-monsoon is rich-K, accounting for an average of 30.9% of the total resolved particles, 190 191 followed by BB (18.7%), OC (12.8%), Ammonium (11.9%), EC-aged (10.9%), and Dust 192 (10.7%). Similar to the results of some studies in urban areas, rich-K or 193 carbonaceous-containing type is the dominant particle type (15-50%) (Li et al., 2014; Zhang et al., 2015; Shen et al., 2017; Zhang et al., 2017; Xu et al., 2018). Differently, few 194 195 researchers can capture the high proportion of Ammonium particles as shown in this study 196 (Shen et al., 2017; Xu et al., 2018), which is ascribed to the conversion of ammonia (NH₃) precursor emitted from large-scale agricultural activities and mountain forest (Engling et al., 197 2011; Li et al., 2013). It is necessary to point out that 60% of Ammonium particles contain 198 signals of diethylamine (DEA, ${}^{58}C_2H_5NHCH_2^+$), implying their similar formation pathway 199 (Zhang et al., 2012). Moreover, the DEA-containing particle represented 12.5% of the total 200 201 ambient particles, which is significantly higher than that in some urban areas at low altitudes (around 2%) (Cahi et al., 2012; Pratt and Prather, 2010; Zhang et al., 2015; Li et al., 2017) 202 203 but is comparable to the observed in processing of high RH, fog and cloud events at a high 204 altitude (> 9%) (Roth et al., 2016; Lin et al., 2019). This suggests that the formation of amines under high RH and fog condition might exist in the Gaomeigu area (with an altitude 205 206 of 3260 m), for example the high relative fraction of DEA-containing particle corresponds to 207 a high RH (Fig. S4), and the existence of amine sources govern the ammonium formations (Bi et al., 2016; Rehbein et al., 2011; Zhang et al., 2012). The relatively larger fraction of 208 Dust particles is related to the short-time occurrences of dust events in spring (Fig. S5), 209 210 leading to a wide contribution ranging between 10% and 70% in the period of 19:00 on April 16th to 10:00 on April 17th. 211

Fig. 1 shows the diurnal variations of each particle type. The rich-K, BB, and OC particles decrease after midnight until 06:00, possibly explained by the curtailment of local traffic and biomass-burning activities even though both the planetary boundary layer (PBL) height and WS decrease (Fig. S6). Then, they rapidly increase around 07:00 when the pollutants from biomass burning are transported from the upwind region as the PBL rises (Liu et al., 2021). At 11:00, the particle counts sharply decrease till 16:00–17:00, caused by





218 the pollutant dispersion with the increases of the PBL height and WS. Increasing trends are observed after 17:00 due to the reduction of PBL height and WS. In contrast, the Ammonium, 219 EC-aged, and Dust particles show a unimodal pattern of the daily diurnal variation (Fig. 1d-f). 220 221 From 00:00 to 06:00, minor fluctuation of particle concentrations of Ammonium, EC-aged, 222 and Dust is observed for these particle types. After that, they continuously elevate until 12:00 due to the regional transport, traffic emission, and road dust from upwind areas (Text S2). 223 224 While the PBL height and WS increase continuously, the Ammonium, EC-aged, and Dust particles begin to decline from 12:00 to 17:00. The subsequent increases of these three 225 particles after 17:00 are attributed to the reduction of PBL height, as a result of the 226 accumulation of pollutants in the near-surface atmosphere. 227

Based on the transport pathways, four air masses clusters are identified to investigate the 228 effect of regional transport on the major particle types (i.e., rich-K, BB, OC, Ammonium, 229 EC-aged, and Dust) (Fig. 2). The most dominant air masses are Cluster 1, 3 and 4 from 230 231 northeastern Myanmar, accounting for 59.8%, 33.2% and 4.6% of the total trajectories, respectively. Cluster 1 had an average percentage of 31.2%, 20.2%, 13.7%, 11.8%, 10.9%, 232 and 6.5%, respectively, on the rich-K, BB, OC, Ammonium, EC-aged, and Dust particles. 233 Clusters 3 and 4 had comparable contributions of BB, OC, Ammonium, and EC-aged to those 234 of Cluster 1, but with a high contribution of Dust, which approximately 12.2% and 18.2% of 235 Clusters 3 and 4 are referred to as dust pollution. The diurnal variations of the BB and OC 236 fractions are similar which rapidly elevate at 07:00 (Fig. S7) due to the increased contribution 237 238 of biomass burning and vehicle emissions from Cluster 1, Ammonium and EC-aged particles (peak at 07:00) caused by the effect of Cluster 1 and 3 together. The similar diurnal trend of 239 Clusters 3 and 4 are both associated with dust contributions, which decrease at 04:00 and 240 241 increase at noon. The increased nighttime particles could be attributed to the pollutant 242 accumulation with the decreased PBL height. Cluster 2 originated from northeastern India 243 and passes over Bangladesh. This cluster accounts for only 2.4 % of the total trajectories, in 244 which ~30.1% and ~31.7% are mainly associated with the rich-K and BB particles, respectively. Even though Clusters 2 and 4 are composed of a small fraction of total 245 trajectories (2.4% and 4.6%, respectively), BB and dust particles are identified as the major 246





247 pollutants, suggesting significant influences from India and northeastern Myanmar during the

248 campaign.

249 **3.2** Characteristics of size distribution and mixing state

250 Fig. 3 shows the size distributions of each particle type. Corresponding the average MS (Text S1 and Fig. S2), rich-K, BB, and OC, EC-aged particles have similar sources from 251 252 vehicle emission or solid-fuel combustion, their size distribution presents at small-scale 253 $(\sim 440 \text{ nm})$ (Fig. 3a). However, the relative proportion of each particle type is distinct under 254 different sizes range, maybe due to the different atmospheric processing in ambient. For 255 example, as shown in Fig. 3b, the percentage of rich-K increases from 17% to 44% along 256 with the increase of particle size from 200 to 420 nm, and then decreases to <10% at 900 nm; 257 a similar thing, the percentage of BB increases from 9% to 27% with the increasing sizes of 200 to 420 nm, and then decreases to <10% at 660 nm. However, the OC and EC-aged types 258 are mainly distributed in relatively small sizes, the percentage of them gradually decreases 259 from 31% and 36% to 9% when size range from \sim 200 to 500 and \sim 400 nm, respectively. 260 261 Notably, the Ammonium and Dust are mainly distributed in large sizes of ~600 nm (Fig. 3a). The Ammonium particles gradually increase from 1.6% to 29% from 440 to 740 nm, then 262 decline to <10% at 1.2 µm. The relatively large size distribution of the Ammonium type is 263 ascribed to the intense atmospheric aging during regional transport (Text S1). The percentage 264 of Dust particles gradually increases from 10% at 560 nm to 60% at 1.48 µm. This is 265 consistent with consist to the fact that the dust is a coarse particle, generally formed at the 266 roadside and fly ash. 267

To investigate the mixing state of the secondary species in the six main particle types, several number fractions of secondary markers (i.e., ${}^{97}\text{HSO}_4^-$, ${}^{195}\text{H}(\text{HSO}_4)_2^-$, ${}^{62}\text{NO}_3^-$, ${}^{18}\text{NH}_4^+$, ${}^{58}\text{C}_2\text{H}_5\text{NHCH}_2^+$ and ${}^{89}\text{HC}_2\text{O}_4^-$) are selected (Fig. 4). Amine particles are characterized by ion signals of amine at m/z ${}^{58}\text{C}_2\text{H}_5\text{NHCH}_2^+$ (diethylamine, DEA) (Angelino et al., 2001; Moffet et al., 2008) and sulfuric acid at m/z ${}^{195}\text{H}(\text{HSO}_4)_2^-$, which is indicative of acidic particles (Rehbein et al., 2011).

The most abundant number fraction of ${}^{97}\text{HSO}_4^-$ and ${}^{18}\text{NH}_4^+$ in Ammonium (99% and 94%) and EC-aged (92% and 31%) particles, but the proportion of ${}^{62}\text{NO}_3^-$ is lower (2% and





276 7%), suggest that ammonium sulfate is not a predominant form instead of ammonium nitrate. Meanwhile, a high number fraction of ¹⁹⁵H(HSO₄)₂⁻ and DEA is also observed in Ammonium 277 (63% and 60%) and EC-aged (4% and 19%) particles. These abundant mixtures may 278 279 represent the high hygroscopicity of Ammonium and EC-aged particles, and their ability to neutralize the acidic particles of Ammonium particle (Sorooshian et al., 2007). Then, a 280 moderate proportion of 97 HSO₄⁻ and 18 NH₄⁺ are seen on the rich-K (65%, 7%) and OC (56%, 281 4%) particles. In contrast, more ⁶²NO₃⁻ contributes to the rich-K (38%) and OC (68%) 282 particles, mainly affected by vehicle emissions (Text S1). Followed by BB (18%) and Dust 283 (6%) particles are found a low number fraction of 97 HSO₄⁻, the moderate 62 NO₃⁻ accounts for 284 45% of the BB particle and only 3% of the Dust particle, and ${}^{18}NH_4^+$ is minor (<1%), which 285 is suggested the degree of BB and Dust particles aging are low. In addition, representing the 286 component of secondary organic formation, oxalate (89HC2O4) is mainly mixed with BB 287 (13%) and rich-K (12%) particles, because they have similar sources from biomass burning. 288 289 A relatively low fraction (<5%) of the oxalate-containing particles in the OC, Ammonium, 290 EC-aged, and Dust particles are a result of their source origins.

Our results are inconsistent with the observations in other field studies (Zhang et al., 2015; Dall'Osto and Harrison, 2012; Li et al., 2022b), in which the carbonaceous-containing particles are more mixed with sulfate than nitrate, additionally, the rich-K and Dust particles are inclined to mix with nitrate. The dissimilarity could be potentially ascribed to the high emission of sulfate from coal combustion, biomass burning, and vehicles to the rich-K, Ammonium, and EC-aged particles (Yang et al., 2017; Li et al., 2022b), and the higher loss of more volatile nitrate than sulfate during the airmass transportation.

298 **3.3** Formation process of the high number concentration particle episodes

A more in-depth investigation of the characteristics of the main particle types in the southeastern Tibet Plateau was conducted during two episode periods when the number concentration of particles was high (i.e., E1: from 08:00 April 18th to 08:00 April 19th, 2018; E2: 17:00 April 26th to 02:00 April 28th, 2018). Even though the two episode events are mostly contributed by Cluster 1, the chemical components show significant differences (Table 1). During E1, the average fractions of the rich-K, BB, OC, Ammonium, EC-aged, and Dust particle are 29.0%, 11.5%, 8.1%, 17.5%, 10.0% and 20.3%, respectively, different from





306 39.3%, 14.2%, 10.0%, 13.5%, 17.2%, and 1.3% respectively, during E2. It can be seen that the Dust particle is the major changed factor, which is 93.6% lower during E2 than E1, 307 whereas the EC-aged particle shows a reversible of 72.0% higher during E2. The rich-K, BB, 308 309 and, OC particles show 22.9%-35.5% differences between the two episode periods. For the air mass clusters (Fig. S8), E1 and E2 exhibit minor differences, mostly originating from 310 311 northern Myanmar and the Sino-Burmese border, but not identical regions. The Dust particles that are much lower during E2 than E1 could be explained by higher WS (on average of $2.7 \pm$ 312 1.0 m/s versus 0.4 ± 0.5 m/s) (Fig. S10) and PBL height (771 ± 717 m versus 560 ± 549 m). 313 314 The Dust particles are mainly formed by re-suspension in the local areas. In addition, the quick thrown-up dust belongs to more coarse size particles, which are out of the detection 315 316 range of the SPAMS. However, due to the larger dust particles deposited more easily under the low WS and the stagnant air conditions during E1, more suspended dust particles of small 317 size fall in the detection range of SPAMS. Moreover, the increased PBL height and WS could 318 319 speed up the transportation of pollutants from multiple sources (e.g., traffic and biomass 320 burning emissions) to the observation site, leading to evaluate the EC-aged, rich-K, BB, and 321 OC particles. The decreased Ammonium particle during E2 is potentially explained by the reductions in the secondary pollutant formation with declines of RH (from $73.9\% \pm 23.9\%$ to 322 $53.1\% \pm 14.9\%$), the oxidation capacities [O₃: from 82.3 ± 5.5 ppbv to 76.8 ± 8.4 ppbv; and 323 324 NO_x: from, 3.9 ± 0.8 to 2.7 ± 0.8 ppb), in comparison to those during E1.

325 In terms of particle size distribution, the peak value of the rich-K, BB, and EC-aged show minor differences (< 80 nm) between two episode periods (Fig. 5a). Expressed in 326 relative proportion (Fig. 5b), the rich-K and BB particles exhibit bimodal distributions, while 327 328 a peak at < 300 nm affected by the primary emissions and > 300 nm associated with the aging process (Li et al., 2022b; Bi et al., 2011). In addition, compared with that during E1, rich-K 329 particles distribute in a wider size range and remind in a high percentage (> 20%) at $\sim 1 \, \mu m$ 330 during E2 due to the atmospheric aging of the airmass. A similar aging process leads to the 331 332 particle size growth of BB, OC, Ammonium, and EC-aged particles, as well as a wider size 333 distribution. Due to relatively low concentration, the size distribution of Dust particles greatly fluctuates. 334

335 During E1, more than 50% of 97 HSO₄⁻ fractions are mixed in the rich-K (81%), OC





(62%), Ammonium (100%), EC-aged (98%) particles (Fig. 6), lower than in BB (37%) and Dust (4%) particles. Dissimilar with E1, the number fraction of 97 HSO₄⁻ increases to 34% during E2, potentially associated with the enhancement by secondary formation. However, the mixing state of 195 H(HSO₄)₂⁻), 62 NO₃⁻, NH₄⁺and oxalate fractions are similar between the two episodes events. The fractions of DEA are significantly higher in E2 than E1 for Ammonium (67% versus 31%) and EC-aged particles (48% versus 17%), mainly due to the high hygroscopic behavior (i.e., higher RH) (Sorooshian et al., 2007).

Photochemical oxidation and aqueous-phase reaction are the key formation pathways of 343 secondary species (Robinson et al., 2007; Link et al., 2017; Xue et al., 2014; Jiang et al., 344 2019). Generally, the O_x ($O_3 + NO_2$) concentration and RH serve as indicators of the degree of 345 photochemical oxidation (Wood et al., 2010) and aqueous-phase reaction (Ervens et al., 2011). 346 In this study, we chose the relative number fractions of ⁴³C₂H₃O⁺, ⁸⁹HC₂O₄⁻, ⁶²NO₃⁻, ⁹⁷HSO₄⁻, 347 and ¹⁸NH₄⁺-containing particles to the total detected particles to indicate the secondary 348 349 formation (Liang et al., 2022), respectively. The correlations between the number fraction of each secondary species with the oxidant concentrations ($O_x = O_3 + NO_2$) and RH are used to 350 351 reflect the formation pathways during the two events (Chen et al., 2016).

352 As shown in Fig. 7, compared with E1 and E2, the linear relationships between the secondary aerosols and O_x are the opposite. During E1, ⁴³C₂H₃O⁺, ⁸⁹HC₂O₄⁻, ⁹⁷HSO₄⁻, ¹⁸NH₄⁺ 353 354 show significant negative linear correlation with O_x (p < 0.01), and the correlation strengths 355 range from moderate to strong (r = $-0.51 \sim -0.81$), except that $^{62}NO_3^{-1}$ fraction shows a certain higher but has no significant correlation with O_x (r = 0.33, p > 0.05). This might be 356 357 influenced by the pollutant dispersion with the increased PBL height when O_x was evaluated 358 (Fig. S9), offsetting the relatively low secondary formations (i.e., both precursors and local anthropogenic emissions are low) near the study location (Li et al., 2016). However, the 359 number fraction of NO_3^- exhibits an upward trend with the increase of O_x concentration, due 360 to the rise of NO₂ concentration (Fig. S9). During E1, the increase of NO₃⁻ could be generally 361 ascribed to the local NO₂ emission, while the declines of other secondary components might 362 be due to the reduced contribution of regional transportation to the other precursors. During 363 E2, ${}^{43}C_2H_3O^+$ has less significant correlation with O_x (r = 0.37, p > 0.05), but with strong 364 correlations with ${}^{89}\text{HC}_2\text{O}_4^-$, ${}^{97}\text{HSO}_4^-$, and ${}^{18}\text{NH}_4^+$ (r = 0.81~0.92, p < 0.01). It should be noted 365





366 that ${}^{62}NO_3^-$ has a strong negative correlation (r = 0.85, p < 0.01) with O_x. These results suggest that photo-oxidation reactions have promoted secondary formation, among which the 367 rate of HSO₄⁻ formation (slop = 0.017) is the highest. Increased with O_x concentration, the 368 369 secondary organic species of $C_2H_3O^+$ (18%-28%) imperceptibly raise, and the oxalate also increases by 7%-20%. This result shows that the secondary organic species were different in 370 371 the capacity of atmospheric oxidation formation, the obviously formed and accumulated secondary acids might be due to the formation of ammonium oxalate by adsorption of NH₃ 372 gas (Sullivan et al., 2007; Nie et al., 2012; Kawamura and Bikkina, 2016; Lin et al., 2019). 373 374 Increasing with O_x concentration, the relative fraction of NO₃⁻ decreases, attributed to its relatively volatile and difficult remote transport during the aging process, and the formation 375 of organic nitrate. The previous study proves that the formations of organic nitrate species 376 (such as ${}^{27}CHN^+$, ${}^{30}NO^+$, ${}^{43}CHO_1N^+$, and CHO_xN^+) through the NO+RO₂ pathway dominate 377 80% of the total nitrate in tropical forested regions during summertime (Alexander et al., 378 379 2009). Aruffo et al (2022) also found that low NO_x (e.g., ≤ 6 ppb) (2.3 ± 0.8 ppbv in this study) could even promote the particle-phase partitioning of the lower volatility of organonitrates. 380

Fig. 8 illustrates that ⁴³C₂H₃O⁺, ⁸⁹HC₂O₄⁻, ⁹⁷HSO₄⁻, and ¹⁸NH₄⁺ have moderate to strong 381 positive correlations with RH (r = $0.70 \sim 0.81$, p < 0.01 or 0.05) during the two episode events, 382 except that ${}^{43}C_2H_3O^+$ during E2 (p = 0.48) and ${}^{89}HC_2O_4^-$ during E1 (p = 0.12). Furthermore, 383 384 ⁶²NO₃⁻ fraction also has no obvious changes and insignificant correlation with RH during E1 385 (p = 0.43) and presents a moderate negative correlation with RH (r = 0.69, p < 0.01) during E2. As shown in Fig. 8e, the aqueous formation rate of HSO_4^- is higher during the E1 (slop is 386 387 0.014) than in E2 (slop is 0.009) due to the low volatile and high hygroscopicity of sulfate 388 (Wang et al., 2016; Zhang et al., 2019b; Sun et al., 2013). More favorable meteorological conditions during E1, including lower WS (0.08 ± 0.08 m s⁻¹) and temperature (3.9 ± 0.8 °C), 389 and higher RH (93.4 \pm 7.6%), lead to the higher formation rate of HSO₄⁻ than that during E2 390 (with meteorological parameters of 2.4 ± 0.8 m s⁻¹, 6.9 ± 1.2 °C, 60.7 ± 8.7 %, respectively). 391 392 The NH₄⁺ species also have a greater production rate during the E1 (slop is 0.005) and E2 (slop is 0.006) due to the high RH. More abundant NH₃ precursors transported from the 393 surrounding under the high WS and acidic anion fraction (i.e., sulfate, and oxalate) advance 394 395 the NH_4^+ formation. Compared with those during E1, the secondary organic species (e.g.,





396 $C_2H_3O^+$ and $HC_2O_4^-$) show inversed generation rates during E2. In addition, $C_2H_3O^+$ shows a strong correlation with RH (r = 0.70, p < 0.05) during E1 (slop is 0.003) but has insignificant 397 correlation during E2. However, the HC₂O₄⁻ fraction has a slight higher (9.7%-13.1%) during 398 399 E1 and increases of correlation with RH (r = 0.81, p < 0.01) during E2 (slop is 0.003). These could be explained by the elevated ammonium oxalate and its precursor ((i.e.,59C2H3O2-, 400 ⁷¹C₃H₃O₂⁻, ⁷³C₂HO₃⁻) concentrations from biomass burning (Ervens et al., 2011; Li et al., 401 2022b). The linearity between $^{62}NO_{3}^{-}$ and RH (r = 0.69, p < 0.01) decreases during E2, 402 mostly due to the low NO₂ concentration (2.6 \pm 0.7 ppb) which further decreases with 403 elevating O₃ (Fig. S10). Meanwhile, high RH could promote organonitrates formation (Fang 404 et al., 2021; Fry et al., 2014). No obvious change and insignificant correlation between NO₃⁻ 405 and RH during E1, potentially attributed to the decreases of NO₂ concentration (3.7 ± 0.4 ppb) 406 in the local atmosphere. 407

408 4 Conclusions

409 This study presents the chemical composition, size distribution, mixing state, and 410 secondary formation of individual particles in the southeastern margin of TP, China during 411 the pre-monsoon season using a high-resolution SPAMS. The finding shows that the rich-K 412 (30.9%) and BB types (18.7%) are the two dominant aerosol particles in the remote area; followed by the OC (12.8%), Ammonium (11.9%), EC-aged (10.9%), and Dust (10.7%) types; 413 414 the NaK-SN, Metal and Others types contributed 0.3–2.8% to the total amibent particles. By 415 interpreting the mass spectra and diurnal trends, the major particle types are mainly from traffic emission, biomass burning, secondary formation, and fly ash, while the dynamics of 416 the PBL height could also affect the contributions of these particles. The observed change in 417 418 the number fraction of the particle types was mainly influenced by air masses (97.61% of the 419 total trajectories) from northeastern Myanmar, and significantly contributed to rich-K and BB types. The particle types show distinct size distributions. The two most critical particle types 420 of rich-K and BB appear in a unimodal pattern, the fractions of OC and EC-aged gradually 421 422 decrease with the increase of the particle sizes, but Ammonium and Dust types show the opposite. Sulfate is the major secondary species and is highly mixed with rich-K, Ammonium, 423 and EC-aged types. Nitrate has a relatively low mixing ratio due to its higher volatility than 424 sulfate during regional transportation, except for BB and OC types. During the entire study 425





campaign, two air episodes with the high number concentration particle occurred but with 426 significant differences in each particle fraction due to the different meteorological conditions 427 (RH, WS, etc.). The results of the formation mechanism of secondary species demonstrate 428 that the formation capacity of atmospheric oxidation is affected by the PBL height, but the 429 relative humidity (RH) could promote the formation of secondary species, especially 97HSO₄-430 and ¹⁸NH₄⁺. The results of this study provide useful information concerning the detailed 431 characteristic of aerosol components, size distribution, and mixing states in the southeast TP, 432 and highlight the importance of the cross-border transport and formation mechanism of 433 aerosols in high-altitude regions. 434





- 435 Data availability. The data presented in this study are available at the Zenodo data archive
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- 437
- 438 *Competing interests.* The authors declare that they have no conflict of interest.
- 439
- 440 *Author contributions.* QW and JC designed the campaign. WR conducted field measurements.
- 441 LL, QW, JT, and YZ made data analysis and interpretation. LL and QW wrote the paper. All
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794 Table 1. The number concentrations, average percentages and characteristic ions of nine types

of particles during the entire study period, and the average percentages of the major six

796 particle types during two episodes.

Туре	Number count	Fraction in total (%)	Episode 1 (%)	Episode 2 (%)	Tracer ions
rich-K	151040	30.9	29.0	39.3	³⁹ K ⁺ , ²⁶ CN ⁻ , ⁴² CNO ⁻ , ⁴⁶ NO ₂ ⁻ , ⁶² NO ₃ ⁻ , ⁹⁷ HSO ₄ ⁻
BB	91322	18.7	11.5	14.2	${}^{39}\mathrm{K}^+, \text{levoglucosan} ({}^{45}\mathrm{CHO_2}^-, {}^{59}\mathrm{C_2H_3O_2}^-, {}^{71}\mathrm{C_3H_3O_2}^-, {}^{73}\mathrm{C_3HO_3}^-, {}^{26}\mathrm{CN}^-, {}^{35,37}\mathrm{Cl}^-, {}^{42}\mathrm{CNO}^-, {}^{46}\mathrm{NO_2}^-, {}^{62}\mathrm{NO_3}^-, {}^{97}\mathrm{HSO_4}^-$
OC	62446	12.8	8.1	10.0	${}^{27}C_2H_3^+, {}^{37}C_3H^+, {}^{38}C_3H_2^+, {}^{39}K^+/C_3H_3^+, {}^{43}C_2H_3O^+, {}^{51}C_4H_3^+, \\ {}^{26}CN^-, {}^{42}CNO^-, {}^{46}NO_2^-, {}^{62}NO_3^-, {}^{97}HSO_4^-$
Ammonium	58317	11.9	17.5	13.5	${}^{12}\text{C}^+, {}^{18}\text{NH}_4^+, {}^{39}\text{K}^+, {}^{58}\text{C}_2\text{H}_5\text{NHCH}_2^+, {}^{97}\text{HSO}_4^-, {}^{195}\text{H}(\text{HSO}_4)_2^-$
EC-aged	53337	10.9	10.0	17.2	C_n^{\pm} (n =1 ~ 5), 39 K ⁺ , 97 HSO ₄ ⁻
Dust	52533	10.7	20.3	1.3	${}^{40}\text{Ca}^+$, ${}^{56}\text{CaO}^+$, ${}^{16}\text{O}^-$, ${}^{17}\text{OH}^-$, ${}^{76}\text{SiO}_3^-$, ${}^{79}\text{PO}_3^-$
NaK-SN	13726	2.8	na	na	23 Na ⁺ , 39 K ⁺ , 62 NO ₃ ⁻ , 97 HSO ₄ ⁻
Metal	4672	1.0	na	na	${}^{51}V^+$, ${}^{56}Fe^+$, ${}^{64,66,68}Zn^+$, ${}^{206,207,208}Pb^+$
Others	1580	0.3	na	na	No obvious characteristic peaks

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798 Figure captions:

- **Figure 1.** Box and whisker diurnal plots of the number concentration of the main particle types (a) rich-Potassium (K), (b) Biomass burning (BB), (c) Organic carbon (OC), (d) Ammonium, (e) Element carbon (EC)-aged, (f) Dust in hourly resolution. The lower, middle, and upper lines of the boxes denote the 25th, 50th, and 75th percentiles. The lower and upper whiskers show the 10th and 90th percentiles, respectively. Average values are shown in white dots.
- **Figure 2.** Maps of the mean HYSPLIT back trajectory clusters (72 h) at the height of 500 m
- during the whole field observation; the table embedded in the figure is the number
- 807 concentration and relative fraction of the main six particle types in each cluster.
- 808 Figure 3. Size distributions of (a) the total number particle counts, (b) the relative
- 809 percentages (%) of the total particles for nine groups during the sampling campaign.
- 810 Figure 4. Number fractions of secondary markers associated with the six particle types
- 811 (rich-K, BB, OC, Ammonium, EC-aged, Dust). Secondary species include sulfate (⁹⁷HSO₄⁻),
- sulfuric acid (195H(HSO4)2⁻), nitrate (62NO3⁻), ammonium (18NH4⁺), DEA (diethylamine,
- 813 ${}^{58}C_2H_5NHCH_2^+$), and oxalate (${}^{89}HC_2O_4^-$) ions.
- 814 **Figure 5.** Size distributions of the (a, c) number concentrations and (b, d) fractions of the
- 815 major six particle types (rich-K, BB, OC, Ammonium, EC-aged, Dust) during two episodes
- 816 of (a,b) E1 and (c, d) E2.
- 817 **Figure 6.** Number fractions of secondary markers associated with the six particle types
- 818 (rich-K, BB, OC, Ammonium, EC-aged, Dust) in two episodes events of E1 and E2: sulfate
- 819 (97 HSO₄⁻), sulfuric acid (195 H(HSO₄)₂⁻), nitrate (62 NO₃⁻), ammonium (18 NH₄⁺), DEA
- 820 (diethylamine, ${}^{58}C_2H_5NHCH_2^+$), and oxalate (${}^{89}HC_2O_4^-$).
- 821 Figure 7. Correlations between the relative number fractions of the secondary species (a)
- $4^{3}C_{2}H_{3}O^{+}$, (b) $8^{9}HC_{2}O_{4}^{-}$, (c) $1^{8}NH_{4}^{+}$, (d) $6^{2}NO_{3}^{-}$, (e) $9^{7}HSO_{4}^{-}$ and O_{x} concentration during E1
- 823 (blue square) and E2 (red dot).
- 824 Figure 8. Correlations between the relative number fractions of the secondary species (a)





- $825 \quad {}^{43}C_2H_3O^+, \text{ (b) } {}^{89}HC_2O_4^-, \text{ (c) } {}^{18}NH_4^+, \text{ (d) } {}^{62}NO_3^-, \text{ (e) } {}^{97}HSO_4^- \text{ and relative humidity (RH)}$
- 826 during E1 (cyan dot) and E2 (orange square).







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828 Figure 1. Box and whisker diurnal plots of the number concentration of the main particle types (a)

829 rich-Potassium (K), (b) Biomass burning (BB), (c) Organic carbon (OC), (d) Ammonium, (e) Element

830 carbon (EC)-aged, (f) Dust in hourly resolution. The lower, middle, and upper lines of the boxes denote the

831 25th, 50th, and 75th percentiles. The lower and upper whiskers show the 10th and 90th percentiles,

832 respectively. Average values are shown in white dots.





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Figure 2. Maps of the mean HYSPLIT back trajectory clusters (72 h) at the height of 500 m during the

835 whole field observation; the table embedded in the figure is the number concentration and relative fraction

836 of the main six particle types in each cluster.







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total particles for nine groups during the sampling campaign.





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841 842 Figure 4. Number fractions of secondary markers associated with the six particle types (rich-K, BB, OC,

Ammonium, EC-aged, Dust). Secondary species include sulfate (97HSO4-), sulfuric acid (195H(HSO4)2-), 843

844 nitrate (⁶²NO₃⁻), ammonium (¹⁸NH₄⁺), DEA (diethylamine, ⁵⁸C₂H₅NHCH₂⁺), and oxalate (⁸⁹HC₂O₄⁻) ions.







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846 Figure 5. Size distributions of the (a, c) number concentrations and (b, d) fractions of the major six particle

847 types (rich-K, BB, OC, Ammonium, EC-aged, Dust) during two episodes of (a,b) E1 and (c, d) E2.







849 Figure 6. Number fractions of secondary markers associated with the six particle types (rich-K, BB, OC,

850 Ammonium, EC-aged, Dust) in two episodes events of E1 and E2: sulfate (⁹⁷HSO₄⁻), sulfuric acid

851 $({}^{195}\text{H}(\text{HSO}_4)_2^{-})$, nitrate $({}^{62}\text{NO}_3^{-})$, ammonium $({}^{18}\text{NH}_4^{+})$, DEA (diethylamine, ${}^{58}\text{C}_2\text{H}_5\text{NHCH}_2^{+})$, and oxalate 852 $({}^{89}\text{HC}_2\text{O}_4^{-})$.







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- Figure 7. Correlations between the relative number fractions of the secondary species (a) ${}^{43}C_2H_3O^+$, (b)
- 855 ${}^{89}\text{HC}_2\text{O}_4^-$, (c) ${}^{18}\text{NH}_4^+$, (d) ${}^{62}\text{NO}_3^-$, (e) ${}^{97}\text{HSO}_4^-$ and O_x concentration during E1 (blue square) and E2 (red 856 dot).







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Figure 8. Correlations between the relative number fractions of the secondary species (a) ${}^{43}C_2H_3O^+$, (b) 89HC₂O₄⁻, (c) ${}^{18}NH_4^+$, (d) ${}^{62}NO_3^-$, (e) ${}^{97}HSO_4^-$ and relative humidity (RH) during E1 (cyan dot) and E2 (orange square).

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