

1 **In-depth study of the formation processes of single atmospheric particles in**  
2 **the southeastern margin of Tibetan Plateau**

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13

14 **Abstract**

15 The unique geographical location of the Tibetan Plateau (TP) plays an important role in  
16 regulating global climate change, but the impacts of the chemical components and  
17 atmospheric processing on the size distribution and mixing state of individual particles are  
18 rarely explored in the southeastern margin of the TP, which is a transport channel for  
19 pollutants from Southeast Asia to the TP during the pre-monsoon season. Thus a  
20 single-particle aerosol mass spectrometer (SPAMS) was deployed to investigate how the local  
21 emissions of chemical composition interact with the transporting particles and assess the  
22 mixing state of different particle types and secondary formation in this study. The TP particles  
23 were classified into six distinct types, mainly including the largest fraction of the  
24 rich-potassium (rich-K) type in the total particles (30.9%), followed by the biomass burning  
25 (BB) type (18.7%). Most particle types were mainly transported from surroundings the  
26 sampling site and along the Sino-Myanmar border; but the air mass trajectories from  
27 northeastern India and Myanmar show a greater impact on the number fraction of BB (31.7%)  
28 and Dust (18.2%) types, respectively. Then, the two episodes with high particle  
29 concentrations showed that the differences in the meteorological conditions in the same  
30 trajectory clusters could cause significant changes in chemical components, especially the  
31 Dust and EC-aged types, which changed by a sum of 93.6% and 72.0%, respectively.  
32 Ammonium and Dust particles distribute at a relatively larger size ( $\sim 600$  nm), but the size  
33 peak of other types is present at  $\sim 440$  nm. Compared with the abundant sulfate ( $^{97}\text{HSO}_4^-$ ),  
34 the low nitrate ( $^{62}\text{NO}_3^-$ ) internally mixed in TP particles is mainly due to the fact that nitrate  
35 is more volatilized during the transport process. The formation mechanism of secondary  
36 speciation demonstrate that the formation capacity of atmospheric oxidation is presumably  
37 affected by the convective transmission and the regional transport in TP. However, the  
38 relative humidity (RH) could significantly promote the formation of secondary species,  
39 especially  $^{97}\text{HSO}_4^-$  and  $^{18}\text{NH}_4^+$ . This study provides new insights into the particle  
40 composition and size, mixing state, and aging mechanism in high time resolution over the TP  
41 region.

42 **Keywords**

43 Southeastern Tibetan Plateau, Individual particles, Chemical characteristics, mixing state,

44 Secondary formation

## 45 **1 Introduction**

46 Atmospheric aerosols have complex components and sources and can be coated with  
47 inorganic or organic materials during transport and atmospheric processing (Crippa et al.,  
48 2013), and then its sizes, chemical compositions, mixing states and optical properties would  
49 change greatly, leading to its influence in the atmosphere more uncertain (Jacobson, 2002;  
50 Zaveri et al., 2010; Matsui, 2016; Budisulistiorini et al., 2017; Ma et al., 2012). Currently, the  
51 influences of the complex chemical components on aerosol size and mixing state show large  
52 regional differences due to the variations in the pollution sources, atmospheric formation  
53 mechanism and meteorological conditions, which have been widely studied in an urban area  
54 at a low altitude (Pratt et al., 2011; Liu et al., 2020a; Xu et al., 2017; Wang et al., 2022).  
55 However, Liu et al (2020) have found that the migration or formation of low-volatile  
56 component (such as nitrate and organic matter) could effectively be reduced due to  
57 evaporation during the upward transportation process, which further alter the chemical  
58 compositions and the particle sizes. The transportation of the aerosols to a relatively cleaner  
59 environment prevails the formation of secondary chemicals at a high altitude (Liu et al.,  
60 2020b). Therefore, a comprehensive investigation of the detailed characteristic of aerosol  
61 formation and mixing states is required to understand their environmental effects in low-, and  
62 high-altitude.

63 As a typical high-altitude region, the Tibetan Plateau (TP) has the highest and largest  
64 mountain area in the world, which is the most sensitive and obvious indicator of climate  
65 change in the entire Asian continent (Liu et al., 2017; Chen and Bordoni, 2014; Immerzeel et  
66 al., 2010). Numerous studies have shown that the melting and retreat of glaciers in the TP  
67 regions is accelerating in recent decades, largely attributed to anthropogenic emissions, such  
68 as greenhouse gases and aerosols (Luo et al., 2020; Hua et al., 2019). Atmospheric aerosols  
69 also can act as cloud condensation nuclei to impact the local hydrological cycles and  
70 monsoon patterns by changing the microphysical properties and life span of clouds (Qian et  
71 al., 2011; Gettelman et al., 2013; Kumar et al., 2017). The southern part of the TP is always  
72 affected by the transport of more polluted air from South Asia along the mountain valleys,

73 especially during the pre-monsoon (i.e., March-May) with the southwest prevailing wind  
74 (Chan et al., 2017; Zhao et al., 2017; Han et al., 2020). Most studies have focused on the  
75 optical characteristics within the TP; however, only a few research has been conducted on  
76 aerosol components.

77 Present researches on aerosol components over the TP mostly focus on exploring the  
78 influence of light-absorbing carbon aerosols and dust particles on climate change by optical  
79 or offline sampling methods (e.g., Wang et al., 2019a; Liu et al., 2021). There is a lack of  
80 studies on the chemical composition, mixing states, and formation mechanism of aerosols in  
81 the southeast margin and even the entire TP, especially using high-time resolved  
82 measurements. Although time-integrated sampling with filter collection followed by  
83 laboratory analyses has been widely adopted for the chemical characterization of aerosols (Li  
84 et al., 2022a; Shen et al., 2015; Zhang et al., 2013), the drawbacks of the traditional approach  
85 need to get attention, including the low time resolution, high detection limit, and time- and  
86 labor-intensive procedures. Therefore, more advanced aerosol measurement equipment with  
87 high-time resolution are developed, for example the aerosol chemical speciation monitor  
88 (ACSM) and aerosol mass spectrometer (AMS) (Ng et al., 2011; Canagaratna et al., 2007) are  
89 mainly achieved the online observation datasets of non-refractory submicron aerosol  
90 (including the mass concentration of sulfate, nitrate, ammonium, chloride and organic; and  
91 mass spectral of organic). This is beneficial to recognize the dynamic processes of source  
92 emission of organic matter in the atmosphere (Du et al., 2015; Zhang et al., 2019a).  
93 Meanwhile, aerosol time-of-flight mass spectrometry (ATOFMS) (Dall'Osto et al., 2014) and  
94 single particle aerosol mass spectrometer (SPAMS) (Zhang et al., 2020) are popular for  
95 characterizing atmospheric individual particles. These devices can determine the chemical  
96 composition and size distribution of the particles in detail, such as the dynamic processes of  
97 chemical aging, mixing state and transporting (Liang et al., 2022; Li et al., 2022b; Zhang et  
98 al., 2019b). To the best knowledge, the advanced measurement device has not yet been  
99 applied for the studies conducted in TP, leading to a lack of in-depth research on the PM<sub>2.5</sub>  
100 pollution in TP, especially in the southeastern margin, which hinders our understanding of the  
101 distribution characteristics and formation mechanism of aerosol components in high-altitude  
102 regions.

103 The southeastern margin of the TP is an important transitional zone between the  
104 high-altitude TP and the low-altitude Yungui Plateau (Wang et al., 2019a; Zhao et al., 2017),  
105 is an ideal place for investigating the impacts of pollutants transport and formation in the  
106 high-altitude zone. In this study, continuous field observation of individual particles (SPAMS)  
107 was deployed on the southeastern margin of the TP during the pre-monsoon period, to (i)  
108 investigate the changes of chemical characteristics between transport and local fine particles  
109 during pre-monsoon, (ii) determine the size distributions and mixing states of different  
110 particle types, and (iii) assess the contributions of photooxidation and aqueous reaction to the  
111 formation of the secondary species. These results would expand our understanding of the  
112 chemical components, size distribution, mixing state and aging pathways of aerosols in the  
113 high-altitude areas over the TP and surrounding areas.

## 114 **2 Methodology**

### 115 **2.1 Sampling site**

116 Intensive one-month field observation was deployed at the rooftop (~ 10 m above ground  
117 level) of the Lijiang Astronomical Station, Chinese Academy of Sciences (3260 m above sea  
118 level; 26°41'24"N, 100°10'48"E), Gaomeigu County, Yunnan Province, China, during the  
119 pre-monsoon period (from April 14<sup>th</sup> to May 13<sup>th</sup>, 2018). The nearest residential area is the  
120 Gaomeigu village (3–5 km away) with a small population size of 113 residents in 27  
121 households. Villagers earn a living by farming (e.g., potato and autumn rape), and biomass is  
122 the major domestic fuel (Li et al., 2016). The sampling site is surrounded by rural and  
123 mountainous areas and has no obvious industry or traffic emissions. During the total  
124 observation period, the average temperature (T) and relative humidity (RH) are  $8.4 \pm 3.1$  °C  
125 and  $69\% \pm 21\%$ , respectively. The wind speed (WS) is  $2.2 \pm 1.2$  m·s<sup>-1</sup> with the prevailing  
126 wind in the north and northeastern (Fig. S1).

### 127 **2.2 On-line instrument**

128 A detailed operational principle and the calibrations of the single-particle aerosol mass  
129 spectrometer (SPAMS, Hexin Analytical Instrument Co., Ltd., Guangzhou, China) has been  
130 described elsewhere (Li et al., 2011). Briefly, individual particles are drawn into SPAMS

131 through a critical orifice. The particles are focused and accelerated, then aerodynamically  
132 sized by two continuous diode Nd: YAG laser beams (532 nm), subsequently desorbed and  
133 ionized by a pulsed laser (266 nm) triggered exactly based on the velocity of the specific  
134 particle. The generated of positive and negative molecular fragments are recorded with the  
135 corresponding size of individual particles. In summary, a velocity, a detection time, and an  
136 ion mass spectrum are recorded for each ionized particle, while there is no mass spectrum for  
137 not ionized particles. The velocity could be converted to  $d_{va}$  based on a calibration using  
138 polystyrene latex spheres (PSL, Thermo Scientific Corp., Palo Alto, USA) with predefined  
139 sizes. The average ambient pressure is 690 hPa (in a range of 685–694 hPa) during the  
140 measurements and calibration. Particles measured by SPAMS mostly are within the size  
141 range of vacuum aerodynamic diameter ( $d_{va}$ ) 0.2–2.0  $\mu\text{m}$ . A hollow silicone dryer was  
142 installed in front of the inlet. This reduces the uncertainty of particle collection efficiency due  
143 to the changes of humidity in sampled airs.

144 Meteorological parameters, including the temperature ( $^{\circ}\text{C}$ ), RH (%), WS ( $\text{m}\cdot\text{s}^{-1}$ ), and  
145 wind direction (WD) were continuously achieved using an automatic weather station (Model  
146 MAWS201, Vaisala HydroMet, Helsinki, Finland) in a 5-min resolution, and the planetary  
147 boundary layer (PBL) was acquired from the website (<https://rda.ucar.edu/datasets/ds083.2>,  
148 last access: 17 April 2020) in a 1-hour resolution. Gaseous concentrations (ppbv) were  
149 obtained using a multiple gas analyzer (Thermo Scientific Corp.), including ozone ( $\text{O}_3$ , model  
150 49i) and nitrogen oxides ( $\text{NO}_x$ , model 42i) in a 5-min resolution. The SPAMS and gas  
151 analyzers are co-located in the same position, and the weather station was uncovered outside  
152 ~5 m from the sampling house. Time series of SPAMS particles, gaseous concentrations (NO,  
153  $\text{NO}_x$ ,  $\text{O}_3$ , and CO) and meteorological parameters (PBL, temperature, RH, WD, and WS)  
154 were shown in Fig. S2.

### 155 **2.3 Individual particle classification**

156 During the observation period, a total of 461,876 ambient particles with the size ( $d_{va}$ ) of  
157 0.2–2.0  $\mu\text{m}$  were collected, including 55,583 in Episode 1 (E1; from April 18<sup>th</sup> 08:00 local  
158 time (LT) to April 19<sup>th</sup> 08:00 LT) and 62,110 in Episode 2 (E2; from April 26<sup>th</sup> 17:00 LT to  
159 April 28<sup>th</sup> 02:00 LT). The analyzed particles are classified into 1,557 groups using an adaptive

160 resonance theory neural network (ART-2a) with a vigilance factor of 0.8, a learning rate of  
161 0.05, and 20 iterations (Song et al., 1999). Finally, eight major particle clusters [i.e.,  
162 potassium-rich (rich-K), biomass burning (BB), organic carbon (OC), Ammonium, aged  
163 element carbon (EC-aged), Dust, sodium (Na)-potassium (K)-containing (NaK-SN), and iron  
164 (Fe)-lead (Pb)-containing (Metal)] with distinct chemical patterns were manually combined,  
165 which represent ~99.7 % of the population of the detected particles. The remaining particles  
166 are grouped as “Other”. The characteristics of the positive and negative mass spectra (MS) of  
167 each particle type are shown in Fig. S3. A detailed description of classification criteria for  
168 individual particles and the characteristic ion fragments for each particle type can be found in  
169 Text S1. The criteria used for searching the some secondary species in the SPAMS datasets  
170 are summarized in Table S2.

## 171 **2.4 Trajectory-related analysis**

172 To determine the influence of regional transport on different particles at the southeastern  
173 margin of the TP, the trajectory clusters analysis was carried out using the 72-h backward air  
174 mass trajectories at arrival heights of 500 m above ground level. The trajectories were  
175 calculated with the Hybrid Single-Particle Lagrangian Integrated Trajectory model (Draxler  
176 and Hess, 1998), and the meteorological data were obtained from the Global Data  
177 Assimilation System (GDAS; <ftp://arlftp.arlhq.noaa.gov/pub/archives/gdas1>, last access: 6  
178 April, 2022). The cluster analysis employs a Euclidean-oriented distance definition to  
179 differentiate and cluster the major spatial features of the inputting trajectories. Details of the  
180 trajectory clustering method can be found in Sirois and Bottenheim (1995). To investigate the  
181 effects of transport on the chemical characteristic of the individual particles, trajectories with  
182 particle number concentrations high than the 75<sup>th</sup> percentile are considered as pollution (Liu  
183 et al., 2021).

## 184 **3 Results and Discussion**

### 185 **3.1 Characteristics of particle composition**

186 Table 1 summarizes the numbers of concentrations, relative percentages and  
187 characteristic ions of each particle type. The most dominant particle type in Gaomeigu during

188 pre-monsoon is rich-K, accounting for an average of 30.9% of the total resolved particles,  
189 followed by BB (18.7%), OC (12.8%), Ammonium (11.9%), EC-aged (10.9%) and Dust  
190 (10.7%). Their characteristics of mass spectrum and possible sources are described in  
191 supplemental information of text S1 in detail. Similar to the results of some studies in urban  
192 areas, rich-K or carbonaceous-containing type is the dominant particle type (15-50%) (Xu et  
193 al., 2018; Wang et al., 2019b; Li et al., 2022). Combined with the previous studies and the  
194 characteristics of the mass spectrum (Fig. S3a) in this study, the rich-K particles are  
195 contributed by biomass burning and traffic emission, because that extensive works usually  
196 identify abundant  $^{39}\text{K}^+$  signals for biomass burning (Pratt et al., 2011; Chen et al., 2017),  
197 while the presence of phosphate ( $m/z$   $^{79}\text{PO}_3^-$ ) indicates the vehicle exhaust (Yang et al., 2017).  
198 The results of the correlation between seven variables (Fig. S4) show that rich-K type was  
199 strongly correlated with Ammonium ( $r=0.84$ ) and EC-aged ( $r=0.90$ ) types, follow well  
200 correlated with OC ( $r=0.70$ ) and BB ( $r=0.68$ ) types, further demonstrate that rich-K particles  
201 type is from traffic emission and biomass burning, and is affected by secondary formation  
202 during the atmospheric aging in southeastern TP. It is worth noting that few research have  
203 captured the high proportion of Ammonium particles as shown in this study (Shen et al., 2017;  
204 Xu et al., 2018), which is ascribed to the conversion of ammonia ( $\text{NH}_3$ ) precursor emitted  
205 from large-scale agricultural activities and mountain forest (Engling et al., 2011; Li et al.,  
206 2013). It is necessary to point out that 60% of Ammonium particles contain signals of amine  
207 fragment ( $m/z$  58,  $\text{C}_2\text{H}_5\text{NH}=\text{CH}_2^+$ ), implying their similar formation pathway (Zhang et al.,  
208 2012). Moreover, the amine-containing particle represented 12.5% of the total ambient  
209 particles, which is significantly higher than that in some urban areas at low altitudes (around  
210 2%) (Cahi et al., 2012; Zhang et al., 2015; Li et al., 2017) but is comparable to that at  
211 observed sites with high RH, or during fog and cloud events at a high altitude ( $> 9\%$ ) (Roth et  
212 al., 2016; Lin et al., 2019). This suggests that the formation of amines under high RH and fog  
213 conditions might exist in the Gaomeigu area (with an altitude of 3260 m), for example, the  
214 high relative fraction of amine-containing particle corresponds to a high RH (Fig. S5), and  
215 the existence of amine sources govern the ammonium formations (Bi et al., 2016; Rehbein et  
216 al., 2011). The relatively larger fraction of Dust particles is related to the short-time  
217 occurrences of dust events in spring (Fig. S6), leading to a wide contribution ranging between



218 10% and 70% in the period of 19:00 LT on April 16<sup>th</sup> to 10:00 LT on April 17<sup>th</sup>.

219 Fig. 1 shows the diurnal variations of each particle type. The rich-K, BB and OC  
220 particles decrease after midnight until 06:00 LT, possibly explained by the curtailment of  
221 local traffic and biomass-burning activities at nighttime even though both the planetary  
222 boundary layer (PBL) height and WS decrease (Fig. S7). Then, their concentrations rapidly  
223 increase in the morning (around 07:00 LT) due to more pollutants from biomass burning and  
224 traffic emissions at the upwind region. The increases of PBL height and WS also lead to the  
225 transported of air pollutants from the surrounding regions to the sampling site (Liu et al.,  
226 2021). At 11:00 LT, the particle concentrations sharply decrease till 16:00–17:00 LT, caused  
227 by the pollutant dispersion with the continuing increases of the PBL height and WS.  
228 Increasing trends are observed after 17:00 LT due to the pollutant accumulation with the  
229 reduction of PBL height and WS. In contrast, the Ammonium, EC-aged and Dust particles  
230 show a unimodal pattern of the daily diurnal variation (Fig. 1d–f). From 00:00 to 06:00 LT,  
231 minor fluctuation of particle concentrations of Ammonium, EC-aged and Dust is observed for  
232 these particle types. After that, their levels continuously elevate until ~11:00 LT due to the  
233 regional transport, traffic emission and fugitive dust (Text S2). While the PBL height and WS  
234 increase continuously, the Ammonium, EC-aged and Dust types begin to decline from 12:00  
235 to 17:00 LT. The subsequent increases of these three types after 17:00 LT are attributed to the  
236 reduction of PBL height, as a result of the accumulation of pollutants in the near-surface  
237 atmosphere.

238 Based on the transport pathways, four air mass clusters are identified to investigate the  
239 effect of regional transport on the major particle types (i.e., rich-K, BB, OC, Ammonium,  
240 EC-aged and Dust) (Fig. 2). Cluster 1, 3 and 4 are originated from northeastern Myanmar,  
241 accounting for 59.8%, 33.2% and 4.6% of the total trajectories, respectively. Cluster 1 had an  
242 average percentage of 32.7%, 18.5%, 12.0%, 12.5%, 11.1% and 8.9%, respectively, on the  
243 rich-K, BB, OC, Ammonium, EC-aged and Dust particles (Table S1). Cluster 3 and 4 have  
244 the comparable contributions of OC (15.5% and 12.5%, *respectively*), increased of BB  
245 (19.3% and 26.8%, *respectively*) and decreased of rich-K (26.8% and 25.2%, *respectively*),  
246 Ammonium (10.4% and 7.7%, *respectively*) and EC-aged (7.7% and 6.3%, *respectively*), to

247 those of Cluster 1, but with a high contribution of Dust (16.6%), which refer Cluster 3 and 4  
248 to as dust and biomass burning pollution. However, Cluster 1 is more influenced by  
249 compound pollution, mainly including secondary formation, biomass burning, and traffic  
250 emissions. The diurnal variations of the BB and OC fractions are similar which rapidly  
251 elevate at 07:00 LT (Fig. S8) due to the increased contribution of biomass burning and traffic  
252 emissions from Cluster 1, Ammonium and EC-aged particles (peak at 07:00 LT) caused by  
253 the effect of Cluster 1 and 3 together. A stable diurnal variation of rich-K fraction is mainly  
254 due to its large proportion and diverse sources. The similar diurnal trend of Clusters 3 and 4  
255 are both associated with dust contributions, which decrease at 04:00 LT and increase at noon.  
256 The increased nighttime particles could be attributed to the pollutant accumulation with the  
257 decreased PBL height. Cluster 2 originate from the northeastern India and passes over  
258 Bangladesh. This cluster accounts for only 2.4 % of the total trajectories, in which ~30.8%  
259 and ~35.9% are mainly associated with the rich-K and BB particles, respectively. Although  
260 Cluster 2 and 4 are composed of a small fraction of total trajectories (2.4% and 4.6 %,  
261 respectively), BB and dust particles are identified as the major pollutants, suggesting  
262 significant influences from India and northeastern Myanmar during the campaign.

263 A more in-depth investigation of the characteristics of the main particle types in the  
264 southeastern Tibet Plateau was conducted during two episode periods when the number  
265 concentration of particles was high (i.e., E1: from 08:00 LT April 18<sup>th</sup> to 08:00 LT April 19<sup>th</sup>,  
266 2018; E2: 17:00 LT April 26<sup>th</sup> to 02:00 LT April 28<sup>th</sup>, 2018) (Fig. S6). Even though the two  
267 episodes are contributed by Cluster 1, the chemical components show significant differences  
268 (Table 1). During E1, the average fractions of the rich-K, BB, OC, Ammonium, EC-aged, and  
269 Dust particle are 29.0%, 11.5%, 8.1%, 17.5%, 10.0% and 20.3%, respectively, different from  
270 39.3%, 14.2%, 10.0%, 13.5%, 17.2%, and 1.3% respectively, during E2. It can be seen that  
271 the major changed factor of the Dust particle is 93.6% lower during E2 than E1, whereas the  
272 EC-aged particle shows a reversible of 72.0% higher during E2. Meanwhile, rich-K, BB and  
273 OC particles also increase by 35.5%, 23.5% and 23.4% respectively during E1 compared to  
274 E2. For the air mass clusters (Fig. S9), E1 and E2 exhibit minor differences, mostly  
275 originating from northern Myanmar and the Sino-Burmese border, but not identical regions.

276 The Dust particles that are much lower during E2 than E1 could be explained by higher WS  
277 (on average of  $2.7 \pm 1.0$  m/s versus  $0.4 \pm 0.5$  m/s) (Fig. S9) and PBL height ( $771 \pm 717$  m  
278 versus  $560 \pm 549$  m) (Fig. S10). The Dust particles are mainly formed by re-suspension in the  
279 local areas. In addition, the quick thrown-up dust belongs to more coarse size particles, which  
280 are out of the detection range of the SPAMS. However, due to the larger dust particles  
281 deposited more easily under the low WS and the stagnant air conditions during E1, more  
282 suspended dust particles of small size fall in the detection range of SPAMS. Moreover, the  
283 increased PBL height and WS could speed up the transportation of pollutants from multiple  
284 sources (e.g., traffic and biomass burning emissions) to the observation site, leading to  
285 elevate the fraction of EC-aged, rich-K, BB, and OC particles during E2. The decreased  
286 Ammonium fraction during E2 is potentially explained by the reductions in the secondary  
287 pollutant formation with declines of RH (from  $73.9\% \pm 23.9\%$  to  $53.1\% \pm 14.9\%$ ), in  
288 comparison to those during E1.

### 289 **3.2 Characteristics of size distribution and mixing state**

290 The aerodynamic size distributions of all particle types are shown in Fig. 3. According to  
291 the characteristics of the average MS (Text S1 and Fig. S3), rich-K, BB, OC and EC-aged  
292 particles originated from the similar sources of solid-fuel combustion or vehicle emission  
293 Their size distribution thus presents within a small-scale ( $\sim 440$  nm) (Fig. S11a). However, the  
294 relative percentage of each particle type is distinct with different size ranges, possibly due to  
295 the unique atmospheric processing. For example, as shown in Fig. 3a, the proportions of  
296 rich-K and BB types increases along with the increase in particle size from 200 to 420 nm,  
297 then decrease. OC and EC-aged types are mainly distributed in relatively small sizes, and  
298 their proportions gradually decrease when the size ranges become larger. Ammonium and  
299 Dust types are mainly distributed in large sizes of  $\sim 600$  nm (Fig. S11a). The proportion of  
300 Ammonium particles gradually increases with the increase of particle size and peaks at 740  
301 nm, the relatively large size distribution is ascribed to the intense atmospheric aging during  
302 regional transport (Text S1). The proportion of Dust particles gradually increases with a  
303 size  $> 560$  nm and peaks at  $1.48 \mu\text{m}$ . This is consistent with the fact that dust is a coarse  
304 particle, generally formed at the roadside and fly ash.

305 Compared with the total particle size distribution, the peak values of the six main  
306 particle types show minor differences ( $< 80$  nm) during the two different episode periods (Fig.  
307 11b,c). However, the percentage of the six particle types distribute in wider size ranges during  
308 E2 than during E1 possibly due to the more intensive atmospheric aging. Similarly, during the  
309 two episodes (Fig. 3b,c), the relatively high fraction of the rich-K and BB particles are more  
310 affected by the primary emissions when their peak value concentrate at  $< 300$  nm, and  $> 300$   
311 nm are more related to the aging process (Li et al., 2022b; Bi et al., 2011). Relatively greater  
312 fluctuation for the large-size fractions ( $> 1.1$   $\mu\text{m}$ ) could be explained by the low particle  
313 concentration (a number less than 20). It should be pointed out that further application of this  
314 method would require a co-located particle-sizing instrument to scale the size-resolved  
315 particle detection efficiency. Both particle composition and size-dependent are the  
316 predominant impacting factors on the particle detection efficiency of the SPAMS (Wenzel et  
317 al., 2003; Yang et al., 2017; Healy et al., 2013).

318 To investigate the mixing state of the secondary species in the six main particle types,  
319 the number fractions of six secondary markers ( $^{97}\text{HSO}_4^-$ ,  $^{195}\text{H}(\text{HSO}_4)_2^-$ ,  $^{62}\text{NO}_3^-$ ,  $^{18}\text{NH}_4^+$ ,  
320  $^{58}\text{C}_2\text{H}_5\text{NHCH}_2^+$  and  $^{89}\text{HC}_2\text{O}_4^-$ ) are calculated (Fig. 4). The presences of amine ( $m/z$   
321  $^{58}\text{C}_2\text{H}_5\text{NHCH}_2^+$ ) and sulfuric acid ( $m/z$   $^{195}\text{H}(\text{HSO}_4)_2^-$ ) signals are possibly indicative of the  
322 water uptake (Chen et al., 2019) and acidic property of the particles (Rehbein et al., 2011),  
323 respectively. The mixing states are obtained by the ratio of the number concentration of the  
324 selected ions to each particle type.

325 The most abundant of  $^{97}\text{HSO}_4^-$  and  $^{18}\text{NH}_4^+$  fraction are seen in Ammonium (99% and  
326 94%, *respectively*) and EC-aged (92% and 31%, *respectively*) particles, whereas much low  
327 fraction of  $^{62}\text{NO}_3^-$  is found (2% and 7%, *respectively*). These suggest that ammonium sulfate  
328 is not a predominant form instead of ammonium nitrate (Zhang et al., 2013). The high  
329 contribution of  $^{97}\text{HSO}_4^-$  in EC-containing particles also suggests a significant influence of  
330 anthropogenically emitted sulfate precursors (e.g.,  $\text{SO}_2$ ) on the aging of EC-containing  
331 particles at the high altitude (Peng et al., 2016; Zhang et al., 2017a). Meanwhile, relatively  
332 high number fractions of  $^{195}\text{H}(\text{HSO}_4)_2^-$  and  $^{58}\text{C}_2\text{H}_5\text{NHCH}_2^+$  are also observed in Ammonium  
333 (63% and 60%) and EC-aged (4% and 19%) particles. These abundant mixtures potentially  
334 represent the high hygroscopicity of Ammonium and EC-aged particles, and their ability to

335 neutralize the acidic particles of Ammonium particle (Sorooshian et al., 2007). Then, a  
336 moderate fraction of  $^{97}\text{HSO}_4^-$  and  $^{18}\text{NH}_4^+$  are seen on the rich-K (65%, 7%) and OC (56%,  
337 4%) particles. In contrast, more  $^{62}\text{NO}_3^-$  fraction contribute to the rich-K (38%) and OC (68%)  
338 particles, mainly affected by vehicle emissions and biomass burning (Text S1). Following BB  
339 (18%) and Dust (6%) particles are found in a relatively low fraction of  $^{97}\text{HSO}_4^-$ , while the  
340 moderate  $^{62}\text{NO}_3^-$  accounts for 45% of the BB particle but only 3% of the Dust particle.  
341 Combined with the results of the minor  $^{18}\text{NH}_4^+$  fraction (<1%) in BB and Dust particles  
342 suggests a relatively low degree of aging. In addition, oxalate ( $^{89}\text{HC}_2\text{O}_4^-$ ), a representative  
343 component of secondary organic formation is mainly mixed with BB (13%) and rich-K (12%)  
344 particles. This is because the substantial precursors of oxalic acid, including acetate  
345 ( $^{59}\text{C}_2\text{H}_3\text{O}_2^-$ ), methylglyoxal ( $^{71}\text{C}_3\text{H}_3\text{O}_2^-$ ), glyoxylate ( $^{73}\text{C}_2\text{HO}_3^-$ ), are emitted from biomass  
346 burning, and then oxalate heterogeneously formed in BB-related particles (Zhang et al.,  
347 2019b; Zauscher et al., 2013). A relatively low fraction (<5%) of oxalate-containing particles  
348 in OC, Ammonium, EC-aged and Dust particles is potentially limited by the contributions of  
349 precursor oxalic acid.

350 Compared to the mixing state of individual particle in urban or suburban areas that are  
351 located close to emission sources (Chen et al., 2016; Dall'Osto and Harrison, 2012; Zhang et  
352 al., 2017a; Li et al., 2022b), the high fractions of sulfate and ammonium at the high altitude  
353 area demonstrate a high degree of aging of the individual particles, whereas the low fraction  
354 of nitrate with high volatility indicates its loss during transportation processing.

355 The number fractions of six markers in the four trajectories were used to further  
356 investigate the impacts of regional transport. As shown in Fig. 5(a,c), the dominant mixing  
357 ion types in each particles (except for Dust) are similar among the four clusters. For Cluster 1,  
358 the number fractions of  $^{97}\text{HSO}_4^-$  and  $^{89}\text{HC}_2\text{O}_4^-$  have larger values in five particle types  
359 (except for Dust type) than those in other trajectories. Similar to Cluster 1, Cluster 3 and 4 are  
360 impacted by regional transport from northeastern Myanmar, the fractions of the six markers  
361 also similar in OC, Ammonium, and EC-aged types. However,  $^{97}\text{HSO}_4^-$  in Cluster 3 and 4 is  
362 reduced in rich-K, BB and Dust types, while  $^{62}\text{NO}_3^-$  is increased in rich-K and decreased in  
363 Dust types, compared with Cluster 1. As discussed in Section 3.1, these results demonstrate  
364 that the aging degree of Cluster 3 and 4 might be lower than that of Cluster 1. For Cluster 2,

365 the fraction of  $^{97}\text{HSO}_4^-$  is obviously decreased in rich-K, BB and EC-aged types but slightly  
366 increased in Dust type (Fig. 5f). Such pattern inverse the observations in rich-K, OC and Dust  
367 types for  $^{62}\text{NO}_3^-$  ions. These variations in Cluster 2 are more likely due to the influences of  
368 biomass-burning activities from surrounding the sampling site, rather than regional transport.  
369 Furthermore, Cluster 2 is associated with regional transport from northeastern India along the  
370 afternoon to nighttime (from 15:00 LT on 11 May to 07:00 LT on 12 May), which is  
371 favorable to the nitrate formation  $\text{N}_2\text{O}_5$  by heterogeneous hydrolysis (Wang et al., 2017; Ding  
372 et al., 2021). However, these cases are infrequent, as only 2% of trajectories are associated  
373 with Cluster 2.

374 During E1, more than 50% of  $^{97}\text{HSO}_4^-$  fractions are mixed in the rich-K (81%), OC  
375 (62%), Ammonium (100%), EC-aged (98%) particles (Fig. S12), low in BB (37%) and Dust  
376 (4%) particles. Dissimilar with E1, the number fraction of  $^{97}\text{HSO}_4^-$  in Dust increases to 34%  
377 during E2, potentially associated with the enhancement by secondary formation during  
378 regional transport. However, the mixing state of  $^{195}\text{H}(\text{HSO}_4)_2^-$ ,  $^{62}\text{NO}_3^-$ ,  $\text{NH}_4^+$  and oxalate  
379 fractions are similar between the two episode events. The  $^{58}\text{C}_2\text{H}_5\text{NHCH}_2^+$  fractions are  
380 significantly higher in E2 than E1 for Ammonium (67% versus 31%) and EC-aged particles  
381 (48% versus 17%), due to the relatively higher hygroscopic behavior (i.e., RHs) (Sorooshian  
382 et al., 2007).

### 383 **3.3 Formation process of the high number concentration particle episodes**

384 Photochemical oxidation and aqueous-phase reaction are the key formation pathways of  
385 secondary species (Link et al., 2017; Xue et al., 2014; Jiang et al., 2019). The oxidant  $\text{O}_x$  ( $\text{O}_3$   
386 +  $\text{NO}_2$ ) concentration and RH usually serve as indicators of the degree of photochemical  
387 oxidation (Wood et al., 2010) and aqueous-phase reaction (Ervens et al., 2011), *receptively*,  
388 though the current  $\text{O}_x$  and RH conditions obtained using the in-situ measurement are not  
389 indicative of the past conditions experienced by the particle. Thus, the relative number  
390 fractions of  $^{43}\text{C}_2\text{H}_3\text{O}^+$ ,  $^{89}\text{HC}_2\text{O}_4^-$ ,  $^{62}\text{NO}_3^-$ ,  $^{97}\text{HSO}_4^-$  and  $^{18}\text{NH}_4^+$ -containing particles to the total  
391 detected particles were selected to provide a rough speculative of the secondary formation  
392 mechanism in TP ambient conditions (Liang et al., 2022). The correlations of the number  
393 fraction of each secondary species with the  $\text{O}_x$  concentrations ( $\text{O}_x$ ) during daytime (from  
394 06:00 to 20:00 LT) and RH during nighttime (from 20:00 to 06:00 LT next day) are used to

395 reveal the formation pathways during the two episodes (Li et al., 2022).

396 As illustrated in Fig. 6, for E1,  $^{43}\text{C}_2\text{H}_3\text{O}^+$ ,  $^{89}\text{HC}_2\text{O}_4^-$ ,  $^{97}\text{HSO}_4^-$ , and  $^{18}\text{NH}_4^+$  show  
397 significant negative linear correlations with  $\text{O}_x$  ( $p < 0.01$ ), and the correlation strengths range  
398 from moderate to strong ( $r = -0.51$  to  $-0.81$ ). However, the  $^{62}\text{NO}_3^-$  fraction shows an upward  
399 trend with an insignificant correlation ( $r = 0.33$ ,  $p > 0.05$ ) with the increase in  $\text{O}_x$   
400 concentration. For E2,  $^{43}\text{C}_2\text{H}_3\text{O}^+$  shows weak correlation with  $\text{O}_x$  ( $r = 0.37$ ,  $p > 0.05$ ), but  
401 strong correlations with  $^{89}\text{HC}_2\text{O}_4^-$ ,  $^{97}\text{HSO}_4^-$ , and  $^{18}\text{NH}_4^+$  ( $r = 0.81\sim 0.92$ ,  $p < 0.01$ ). It should  
402 be noted that  $^{62}\text{NO}_3^-$  has a strong negative correlation ( $r = -0.85$ ,  $p < 0.01$ ) with  $\text{O}_x$ . In general,  
403 the opposite linear relationship between secondary aerosol and  $\text{O}_x$  during E1 and E2 might be  
404 influenced by reasons of i) the relatively low secondary formations because of the small  
405 amount of precursors emitting from anthropogenic activities around the sampling site (Li et  
406 al., 2016); ii) higher dilution rate of the particles formed in the atmosphere with the rapid rise  
407 of PBL height during E1 than E2 (Fig. S13a); iii) the degrees of contributions of regional  
408 transport due to the low WS ( $0.5 \pm 0.6 \text{ m s}^{-1}$ ) during E1 and the high WS ( $3.1 \pm 1.0 \text{ m s}^{-1}$ )  
409 during E2, respectively (Fig. S9). Therefore, for E1, the increases of  $\text{NO}_3^-$  fraction could be  
410 influenced by the local nitrate formation, while the declines of other secondary components  
411 should be ascribed to the reduced contribution of regional transport. For E2, the decreased of  
412  $\text{NO}_3^-$  fraction could be caused by the relatively higher volatilization loss of nitrate than other  
413 components through the regional transport. Additionally, previous study proves that the  
414 formations of organic nitrate species (such as  $^{27}\text{CHN}^+$ ,  $^{30}\text{NO}^+$ ,  $^{43}\text{CHO}_1\text{N}^+$  and  $\text{CHO}_x\text{N}^+$ )  
415 through the  $\text{NO}+\text{RO}_2$  pathway dominate 80% of the total nitrate production in tropical  
416 forested regions during summertime (Alexander et al., 2009). Aruffo et al (2022) also found  
417 that low  $\text{NO}_x$  (i.e.  $< 6 \text{ ppbv}$ ), compared to  $2.3 \pm 0.8 \text{ ppbv}$  in this study, could even promote  
418 the particle-phase partitioning of the lower volatility of organonitrates. These results suggest  
419 that the secondary organic species have different formation capacities through  
420 photo-oxidation reactions, among which the rate of  $\text{HSO}_4^-$  formation (slop=0.017) is the  
421 highest. Increased with  $\text{O}_x$  concentration during E2, the concentration levels of secondary  
422 organic species of  $\text{C}_2\text{H}_3\text{O}^+$  (18%-28%) imperceptibly rise, while the oxalate fraction  
423 significantly increase by 7%-20%.

424 Considering that the oxalate is abundant mixed in rich-K (14%), BB (15%), EC-aged

425 (5%), and Dust (6%) particles in Cluster 1 (Fig. 5), and the increased contributions of rich-K  
426 (39.3%), BB (14.2%) and EC-aged (17.2%) types during E2 (Table 1), the apparent formation  
427 of oxalate might be due to the enhancement of regional transport. Particularly, this presents  
428 the nearby biomass burning and combustion activities produce more precursor species of  
429 oxalate (Sullivan et al., 2007; Kundu et al., 2010; Zhang et al., 2017b).

430 Fig. 7 illustrates that the number fractions of  $^{43}\text{C}_2\text{H}_3\text{O}^+$ ,  $^{89}\text{HC}_2\text{O}_4^-$ ,  $^{97}\text{HSO}_4^-$ , and  $^{18}\text{NH}_4^+$   
431 have moderate to strong positive correlations with RH ( $r = 0.70\sim 0.81$ ,  $p < 0.01$  or  $0.05$ ) in the  
432 nighttime during the two episodes, except that  $^{43}\text{C}_2\text{H}_3\text{O}^+$  during E2 ( $p = 0.48$ ) and  $^{89}\text{HC}_2\text{O}_4^-$   
433 during E1 ( $p = 0.12$ ). Furthermore,  $^{62}\text{NO}_3^-$  fraction has no obvious changes with insignificant  
434 correlation with RH during E1 ( $p = 0.43$ ) and presents a moderate negative correlation with  
435 RH ( $r = 0.69$ ,  $p < 0.01$ ) during E2. As shown in Fig. 7e, the highest aqueous formation rate of  
436  $\text{HSO}_4^-$  is mainly due to the properties of low volatile and high hygroscopic of sulfate (Wang  
437 et al., 2016; Zhang et al., 2019c; Sun et al., 2013). Compared with that during E2  
438 (slop=0.014), the decreased formation rate of  $\text{HSO}_4^-$  during the E1 (slop=0.009) may be  
439 because the decreases of aerosol acidity in higher RH > 80% (Huang et al., 2019; Meng et al.,  
440 2014; Tian et al., 2021). And the increased contributions of regional transport due to the high  
441 WS ( $2.4 \pm 0.8 \text{ m s}^{-1}$ ) during E2 are compared with the low WS ( $0.08 \pm 0.08 \text{ m s}^{-1}$ ) during E1  
442 (Fig. S9). The fair production rate of  $\text{NH}_4^+$  during the E1 (slop= 0.005) and E2 (slop=0.006)  
443 demonstrate that an aqueous-phase reaction could effectively promote ammonium formation.  
444 Meanwhile, a slightly larger slop of  $\text{NH}_4^+$  during E2 could be also affected by the increased  
445 contributions of regional transport. Compared with those during E1, the inverse generation  
446 rates of two secondary organic species (i.e.,  $\text{C}_2\text{H}_3\text{O}^+$  and  $\text{HC}_2\text{O}_4^-$ ) during E2 are possibly  
447 caused by the different formation pathways with a variety of RH levels or distinct regional  
448 transports. For example,  $\text{C}_2\text{H}_3\text{O}^+$  shows a strong correlation with RH ( $r = 0.70$ ,  $p < 0.05$ )  
449 during E1 (slop=0.003) but has insignificant correlation during E2. This could be explained  
450 by high RHs that could effectively promote secondary organic formation during E1. In  
451 addition, the  $\text{HC}_2\text{O}_4^-$  fraction increases slightly (9.7-13.1%) during E1 is potentially ascribed  
452 to more abundant Dust-type particles (20.3%) which compose of high calcium (Ca) (Fig. S14)  
453 that favor the formation of metal oxalate complexes (i.e., Ca oxalate). At high RHs ( $93.4 \pm$   
454  $7.6\%$ ), if oxalate ions are dissolved in the aqueous phase with the presence of Ca ions, the Ca



455 oxalate complexes can precipitate because of their low hygroscopic and insoluble natures  
456 (Furukawa and Takahashi, 2011). This could offset the oxalate formation in the  
457 aqueous-phase reaction. However, significant linear increases (slop=0.003) with RH ( $r = 0.81$ ,  
458  $p < 0.01$ ) during E2 demonstrate that the aqueous-phase reaction effectively promotes the  
459 oxalate formation (Cheng et al., 2017; Meng et al., 2020). No obvious change and  
460 insignificant correlation between  $^{62}\text{NO}_3^-$  and RH are found during E1, potentially attributed  
461 to the decreases of  $\text{NO}_2$  concentration ( $3.7 \pm 0.4$  ppbv) in the local atmosphere. Meanwhile,  
462 high RHs could promote organonitrates formation (Fang et al., 2021; Fry et al., 2014). The  
463 linearity between  $^{62}\text{NO}_3^-$  and RH ( $r = 0.69$ ,  $p < 0.01$ ) significantly decreases during E2,  
464 mostly due to the losses of the volatile compound through the regional transport (Fig. S15).

#### 465 **4 Conclusions**

466 This study presents the chemical composition, size distribution, mixing state and  
467 secondary formation of individual particles in the southeastern margin of TP, China during  
468 the pre-monsoon season using a high-resolution SPAMS. The finding shows that the rich-K  
469 (30.9%) and BB types (18.7%) are the two dominant aerosol particles in the remote area;  
470 followed by the OC (12.8%), Ammonium (11.9%), EC-aged (10.9%), and Dust (10.7%) types;  
471 the NaK-SN, Metal and Others particle types contributed 0.3–2.8% to the total ambient  
472 particles. By interpreting the mass spectra and diurnal trends, the major particle types are  
473 mainly from traffic emission, biomass burning, secondary formation and fly ash, while the  
474 dynamics of the PBL height could also affect their contributions. The observed change in the  
475 number fraction of the particle types was mainly influenced by air mass (97.61% of the total  
476 trajectories) from northeastern Myanmar, and significantly contributed to rich-K and BB  
477 types. The particle types show distinct size distributions. The two critical particle types of  
478 rich-K and BB appear in a unimodal pattern, the fractions of OC and EC-aged gradually  
479 decrease with the increase of the particle sizes, but Ammonium and Dust types show the  
480 opposite. Sulfate is the major secondary species and is highly mixed with rich-K, Ammonium  
481 and EC-aged types. Nitrate has a relatively low mixing ratio due to its higher volatility than  
482 sulfate during regional transportation, except for BB and OC types. During the entire study  
483 campaign, two episodes with the high number concentration of particles occur but with  
484 significant differences in each particle fraction due to the different meteorological conditions

485 (RH, WS, etc.). Meanwhile, the different meteorological conditions also lead to an inverse  
486 linear correlation between the indicators of secondary formation, including  $C_2H_3O^+$ ,  $HC_2O_4^-$ ,  
487  $NH_4^+$ ,  $NO_3^-$  and  $HSO_4^-$ . and  $O_x$  ( $O_3+NO_2$ ) during episode 1 and 2 periods; however, they  
488 present a positive linear correlation with relative humidity (RH), except for  $NO_3^-$  shown the  
489 negative linear correlation with RH due to the low precursors concentration and potential  
490 organonitrates formation. These results demonstrated that the capacity of atmospheric aging  
491 of photo-oxidation and aqueous reaction have complex influencing factors. Although the  
492 detailed formation pathways and their percentage contributions to secondary species are not  
493 quantitatively estimated in this study, our results have important implications for the various  
494 possibilities affecting the characteristic of chemical components, size distribution, mixing  
495 states, and formation mechanism of aerosols in the southeast TP. More depth investigations  
496 concerning the evolution mechanisms of secondary aerosols are encouraged since TP is a  
497 significant regulator to global climate change.

498 *Data availability.* The data presented in this study are available at the Zenodo data archive  
499 <https://doi.org/10.5281/zenodo.7336857>.

500

501 *Competing interests.* The authors declare that they have no conflict of interest.

502

503 *Author contributions.* QW and JC designed the campaign. WR conducted field measurements.  
504 LL, QW, JT, and YZ made data analysis and interpretation. LL and QW wrote the paper. All  
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506

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## 516 **Reference**

517 Alexander, B., Hastings, M. G., Allman, D. J., Dachs, J., Thornton, J. A., and Kunasek, S. A.: Quantifying  
518 atmospheric nitrate formation pathways based on a global model of the oxygen isotopic composition  
519 ( $\Delta^{17}\text{O}$ ) of atmospheric nitrate, *Atmos. Chem. Phys.*, 9, 5043–5056,  
520 <https://doi.org/10.5194/acp-9-5043-2009>, 2009.

521 Aruffo, E., Wang, J., Ye, J., Ohno, P., Qin, Y., Stewart, M., McKinney, K., Di Carlo, P., and Martin, S. T.:  
522 Partitioning of Organonitrates in the Production of Secondary Organic Aerosols from  $\alpha$ -Pinene  
523 Photo-Oxidation, *Environ. Sci. Technol.*, 56, 5421–5429, <https://doi.org/10.1021/acs.est.1c08380>,  
524 2022.

525 Bi, X. H., Zhang, G. H., Li, L., Wang, X. M., Li, M., Sheng, G. Y., Fu, J. M., and Zhou, Z.: Mixing state of  
526 biomass burning particles by single particle aerosol mass spectrometer in the urban area of PRD,  
527 China, *Atmos. Environ.*, 45, 3447–3453, <https://doi.org/10.1016/j.atmosenv.2011.03.034>, 2011.

528 Bi, X. H., Lin, Q. H., Peng, L., Zhang, G. H., Wang, X. M., Brechtel, F. J., Chen, D. H., Li, M., Peng, P. A.,  
529 Sheng, G. Y., and Zhou, Z.: In situ detection of the chemistry of individual fog droplet residues in the  
530 Pearl River Delta region, China, *J. Geophys. Res. Atmos.*, 121, 9105–9116,  
531 <https://doi.org/10.1002/2016jd024886>, 2016.

532 Budisulistiorini, S. H., Riva, M., Williams, M., Chen, J., Itoh, M., Surratt, J. D., and Kuwata, M.:

533 Light-absorbing brown carbon aerosol constituents from combustion of Indonesian peat and biomass,  
534 *Environ. Sci. Technol.*, 51, 4415–4423, <https://doi.org/10.1021/acs.est.7b00397>, 2017.

535 Canagaratna, M. R., Jayne, J. T., Jimenez, J. L., Allan, J. D., Alfarra, M. R., Zhang, Q., Onasch, T. B.,  
536 Drewnick, F., Coe, H., Middlebrook, A., Delia, A., Williams, L. R., Trimborn, A. M., Northway, M. J.,  
537 DeCarlo, P. F., Kolb, C. E., Davidovits, P., and Worsnop, D. R.: Chemical and microphysical  
538 characterization of ambient aerosols with the aerodyne aerosol mass spectrometer, *Mass Spectrom.*  
539 *Rev.*, 26, 185–222, <https://dx.doi.org/10.1002/mas.20115>, 2007.

540 Chan, C. Y., Wong, K. H., Li, Y. S., Chan, Y., and Zhang, X.D.: The effects of Southeast Asia fire activities  
541 on tropospheric ozone, trace gases and aerosols at a remote site over the Tibetan Plateau of Southwest  
542 China, *Tellus B*, 58B, 310–318, <https://doi.org/10.1111/j.1600-0889.2006.00187.x>, 2017.

543 Chen, Y., Cao, J. J., Huang, R. J., Yang, F. M., Wang, Q. Y., and Wang, Y. C.: Characterization, mixing  
544 state, and evolution of urban single particles in Xi'an (China) during wintertime haze days, *Sci. Total*  
545 *Environ.*, 573, 937–945, <https://doi.org/10.1016/j.scitotenv.2016.08.151>, 2016.

546 Chen, Y., Wenger, J. C., Yang, F. M., Cao, J. J., Huang, R. J., Shi, G. M., Zhang, S. M., Tian, M., and Wang,  
547 H. B.: Source characterization of urban particles from meat smoking activities in Chongqing, China  
548 using single particle aerosol mass spectrometry, *Environ. Pollut.*, 228, 92–101,  
549 <https://doi.org/10.1016/j.envpol.2017.05.022>, 2017.

550 Chen, Y., Tian, M., Huang, R. J., Shi, G. M., Wang, H. B., Peng, C., Cao, J. J., Wang, Q. Y., Zhang, S. M.,  
551 Guo, D. M., Zhang, L. M., and Yang, F. M.: Characterization of urban amine-containing particles in  
552 southwestern China: seasonal variation, source, and processing, *Atmos. Chem. Phys.*, 19, 3245–3255,  
553 <https://doi.org/10.5194/acp-19-3245-2019>, 2019.

554 Chen, J. Q., and Bordoni, S.: Orographic effects of the Tibetan Plateau on the East Asian Summer  
555 Monsoon: An energetic perspective, *J. Climate.*, 27, 3052–3072,  
556 <https://doi.org/10.1175/JCLI-D-13-00479.1>, 2014.

557 Cheng, C. L., Li, M., Chan, C. K., Tong, H. J., Chen, C. H., Chen, D. H., Wu, D., Li, L., Wu, C., Cheng, P.,  
558 Gao, W., Huang, Z. X., Li, X., Zhang, Z. J., Fu, Z., Bi, Y. R., Zhou, Z.: Mixing state of acid  
559 containing particles in the rural area of Pearl River Delta, China: implication the formation mechanism  
560 of oxalic acid, *Atmos. Chem. Phys.*, 17, 9519–9533, <https://doi.org/10.5194/acp-17-9519-2017>, 2017.

561 Crippa, M., DeCarlo, P. F., Slowik, J. G., Mohr, C., Heringa, M. F., Chirico, R., Poulain, L., Freutel, F.,  
562 Sciare, J., Cozic, J., Di Marco, C. F., Elsasser, M., Nicolas, J. B., Marchand, N., Abidi, E.,  
563 Wiedensohler, A., Drewnick, F., Schneider, J., Borrmann, S., Nemitz, E., Zimmermann, R., Jaffrezo, J.  
564 L., Prévôt, A. S. H., and Baltensperger, U.: Wintertime aerosol chemical composition and source  
565 apportionment of the organic fraction in the metropolitan area of Paris, *Atmos. Chem. Phys.*, 13,  
566 961–981, <https://doi.org/10.5194/acp-13-961-2013>, 2013.

567 Dall'Osto, M., Beddows, D.C.S., Gietl, J. K., Olatunbosun, O. A., Yang, X. G., and Harrison, R. M.:  
568 Characteristics of tyre dust in polluted air: studies by single particle mass spectrometry (ATOFMS),  
569 *Atmos. Environ.*, 94, 224–230, <https://doi.org/10.1016/j.atmosenv.2014.05.026>, 2014.

570 Dall'Osto, M., and Harrison, R. M.: Urban organic aerosols measured by single particle mass spectrometry  
571 in the megacity of London, *Atmos. Chem. Phys.*, 12, 4127–4142,  
572 <http://dx.doi.org/10.5194/acp-12-4127-2012>, 2012.

573 Ding, J., Dai, Q. L., Zhang, Y. F., Xu, J., Huangfu, Y. Q., Feng, Y. C.: Air humidity affects secondary  
574 aerosol formation in different pathways, *Sci. Total Environ.*, 759, 143540–143549,  
575 <https://doi.org/10.1016/j.scitotenv.2020.143540>, 2021.

576 Draxler, R. and Hess, G.: An overview of the HYSPLIT\_4 modelling system for trajectories, *Aust.*

577 Meteorol. Mag., 47, 295–308, 1998.

578 Du, W., Sun, Y. L., Xu, Y. S., Jiang, Q., Wang, Q. Q., Wang, W., Wang, F., Bai, Z. P., Zhao, X. D., and  
579 Yang, Y. C.: Chemical characterization of submicron aerosol and particle growth events at a national  
580 background site (3295 m a.s.l.) on the Tibetan Plateau, *Atmos. Chem. Phys.*, 15, 10811–10824,  
581 <https://doi.org/10.5194/acp-15-10811-2015>, 2015.

582 Engling, G., Zhang, Y. N., Chan, C. Y., Sang, X. F., Lin, M., Ho, K. F., Li, Y. S., Lin, C. Y., and Lee, J. J.:  
583 Characterization and sources of aerosol particles over the southeastern Tibetan Plateau during the  
584 Southeast Asia biomass-burning season, *Tellus B*, 63, 117–128,  
585 <https://doi.org/10.1111/j.1600-0889.2010.00512.x>, 2011.

586 Ervens, B., Turpin, B. J., and Weber, R. J.: Secondary organic aerosol formation in cloud droplets and  
587 aqueous particles (aqSOA): a review of laboratory, field and model studies, *Atmos. Chem. Phys.*, 11,  
588 11069–11102, <https://doi.org/10.5194/acp-11-11069-2011>, 2011.

589 Fang, X. Z., Liu, Y. Y., Li, K. J., Wang, T., Deng, Y., Feng, Y. Q., Yang, Y., Cheng, H. Y., Chen, J. M., and  
590 Zhang, L. W.: Atmospheric Nitrate Formation through Oxidation by Carbonate Radical, *ACS Earth  
591 Space Chem.*, 5, 1801–1811, <https://doi.org/10.1021/acsearthspacechem.1c00169>, 2021.

592 Fry, J. L., Draper, D. C., Barsanti, K. C., Smith, J. N., Ortega, J., Winkler, P. M., Lawler, M. J., Brown, S.  
593 S., Edwards, P. M., Cohen, R. C., and Lee, L.: Secondary Organic Aerosol Formation and Organic  
594 Nitrate Yield from NO<sub>3</sub> Oxidation of Biogenic Hydrocarbons, *Environ. Sci. Technol.*, 48,  
595 11944–11953, <https://doi.org/10.1021/es502204x>, 2014.

596 Furukawa, T., and Takahashi, Y.: Oxalate metal complexes in aerosol particles: implications for the  
597 hygroscopicity of oxalate-containing particles, *Atmos. Chem. Phys.*, 11, 4289–4301,  
598 <https://doi.org/10.5194/acp-11-4289-2011>, 2011.

599 Gettelman, A., Morrison, H., Terai, C. R., and Wood, R.: Microphysical process rates and global  
600 aerosolecloud interactions, *Atmos. Chem. Phys.*, 13, 9855–9867,  
601 <https://doi.org/10.5194/acp-14-9099-2014>, 2013.

602 Han, H., Wu, Y., Liu, J., Zhao, T. L., Zhuang, B. L., Wang, H. L., Li, Y. C., Chen, H. M., Zhu, Y., Liu, H.  
603 N., Wang, Q. G., Li, S., Wang, T. J., Xie, M., and Li, M. M.: Impacts of atmospheric transport and  
604 biomass burning on the inter-annual variation in black carbon aerosols over the Tibetan Plateau,  
605 *Atmos. Chem. Phys.*, 20, 13591–13610, <https://doi.org/10.5194/acp-20-13591-2020>, 2020.

606 Healy, R. M., Sciare, J., Poulain, L., Crippa, M., Wiedensohler, A., Prévôt, A.S.H., Baltensperger, U.,  
607 Sarda-Estève, R., McGuire, M. L., Jeong, C. H., McGillicuddy, E., O’Connor, I. P., Sodeau, J. R.,  
608 Evans, G. J., and Wenger, J. C.: Quantitative determination of carbonaceous particle mixing state in  
609 Paris using single-particle mass spectrometer and aerosol mass spectrometer measurements, *Atmos.  
610 Chem. Phys.*, 13, 9479–9496, <http://dx.doi.org/10.5194/acp-13-9479-2013>, 2013.

611 Hua, S., Liu, Y. Z., Luo, R., Shao, T. B., Zhu, Q. Z.: Inconsistent aerosol indirect effects on water clouds  
612 and ice clouds over the Tibetan Plateau, *Int. J. Climatol.*, 40, 3832–3848,  
613 <https://doi.org/10.1002/joc.6430>, 2019.

614 Huang, X. J., Zhang, J. K., Luo, B., Luo, J. Q., Zhang, W., and Rao, Z. H.: Characterization of oxalic  
615 acid-containing particles in summer and winter seasons in Chengdu China, *Atmos. Environ.*, 198,  
616 133–141, <https://doi.org/10.1016/j.atmosenv.2018.10.050>, 2019.

617 Immerzeel, W. W., van Beek, L. P. H., and Bierkens, M. F. P.: Climate change will affect the Asian water  
618 towers, *Science*, 328, 1382–1385, <https://doi.org/10.1126/science.1183188>, 2010.

619 Jacobson, M. Z.: Analysis of aerosol interactions with numerical techniques for solving coagulation,  
620 nucleation, condensation, dissolution, and reversible chemistry among multiple size distributions, *J.*

621 Geophys. Res., 107(D19), 4366, <https://doi.org/10.1029/2001JD002044>, 2002.

622 Jiang, H. H., Frie, A. L., Lavi, A., Chen, J. Y., Zhang, H., Bahreini, R., and Lin, Y. H.: Brown carbon  
623 formation from nighttime chemistry of unsaturated heterocyclic volatile organic compounds, *Environ.*  
624 *Sci. Technol. Lett.*, 6, 184190, <https://doi.org/10.1021/acs.estlett.9b00017>, 2019.

625 Kumar, M., Raju, M. P., Singh, R. K., Singh, A. K., Singh, R. S., and Banerjee, T.: Wintertime  
626 characteristics of aerosols over middle Indo-Gangetic Plain: vertical profile, transport and radiative  
627 forcing, *Atmos. Res.*, 183, 268–282, <https://doi.org/10.1016/j.atmosres.2016.09.012>, 2017.

628 Kundu, S., Kawamura, K., Andreae, T. W., Hoffer, A., and Andreae, M. O.: Molecular distributions of  
629 dicarboxylic acids, ketocarboxylic acids and alpha-dicarbonyls in biomass burning aerosols:  
630 implications for photochemical production and degradation in smoke layers, *Atmos. Chem. Phys.*, 10  
631 (5), 2209–2225, <https://doi.org/10.5194/acp-10-2209-2010>, 2010.

632 Lian, X. F., Zhang, G. H., Yang, Y. X., Lin, Q. H., Fu, Y. Z., Jiang, F., Peng, L., Hu, X. D., Chen, D. H.,  
633 Wang, X. M., Peng, P. A., Sheng, G. Y., and Bi, X. H.: Evidence for the Formation of Imidazole from  
634 Carbonyls and Reduced Nitrogen Species at the Individual Particle Level in the Ambient Atmosphere,  
635 *Environ. Sci. Technol. Lett.*, 8, 9–15, <https://dx.doi.org/10.1021/acs.estlett.0c00722>, 2021.

636 Liang, Z. C., Zhou, L. Y., Cuevas, R. A., Li, X. Y., Cheng, C. L., Li, M., Tang, R. Z., Zhang, R. F., Lee  
637 Patrick K. H., Lai, Alvin C. K., and Chan, C.K.: Sulfate Formation in Incense Burning Particles: A  
638 Single-Particle Mass Spectrometric Study, *Environ. Sci. Technol. Lett.*, 9, 718–725,  
639 <https://doi.org/10.1021/acs.estlett.2c00492>, 2022.

640 Li, C. L., Bosch, C., Kang, S. C., Andersson, A., Chen, P. F., Zhang, Q. G., Cong, Z. Y., Chen, B., Qin, D.  
641 H., and Gustafsson, Ö.: Sources of black carbon to the Himalayan–Tibetan Plateau glaciers, *Nat.*  
642 *Commun.*, 7, 12574, <https://doi.org/10.1038/ncomms12574>, 2016b.

643 Li, C. L., Bosch, C., Kang, S. C., Andersson, A., Chen, P. F., Zhang, Q. G., Cong, Z. Y., Tripathee, L., and  
644 Örtjanb, G.: <sup>14</sup>C characteristics of organic carbon in the atmosphere and at glacier region of the Tibetan  
645 Plateau, *Sci. Total Environ.*, 832, 155020, <https://doi.org/10.1016/j.scitotenv.2022.155020>, 2022a.

646 Li, L., Huang, Z. X., Dong, J. G., Li, M., Gao, W., Nian, H. Q., Fu, Z., Zhang, G. H., Bi, X. H., Cheng, P.,  
647 and Zhou, Z.: Real time bipolar time-of-flight mass spectrometer for analyzing single aerosol particles,  
648 *Int. J. Mass Spectrom.*, 303, 118–124, <https://doi.org/10.1016/j.ijms.2011.01.017>, 2011.

649 Li L., Wang, Q. Y., Zhang, Y., Liu, S. X., Zhang, T., Wang, S., Tian, J., Chen, Y., Hang Ho, S. S., Han, Y.,  
650 and Cao, J.J.: Impact of reduced anthropogenic emissions on chemical characteristics of urban aerosol  
651 by individual particle analysis, *Chemosphere*, 303, 135013,  
652 <https://doi.org/10.1016/j.chemosphere.2022.135013>, 2022b.

653 Li, J. J., Wang, G. H., Wang, X. M., Cao, J. J., Sun, T., Cheng, C. L., Meng, J. J., Hu, T. F., and Liu, S. X.:  
654 Abundance, composition and source of atmospheric PM<sub>2.5</sub> at a remote site in the Tibetan Plateau,  
655 China, *Tellus B*, 65, <http://dx.doi.org/10.3402/tellusb.v65i0.20281>, 2013.

656 Li, Y. J., Sun, Y. L., Zhang, Q., Li, X., Li, M., Zhou, Z., and Chan, C. K.: Real-time chemical  
657 characterization of atmospheric particulate matter in China: A review, *Atmos. Environ.*, 158, 270–304,  
658 <http://dx.doi.org/10.1016/j.atmosenv.2017.02.027>, 2017.

659 Lin, Q. H., Bi, X. H., Zhang, G. H., Yang, Y. X., Peng, L., Lian, X. F., Fu, Y. Z., Li, M., Chen, D. H., Miller,  
660 M., Ou, J., Tang, M. J., Wang, X. M., Peng, P. A., Sheng, G. Y., and Zhou, Z.: In-cloud formation of  
661 secondary species in iron-containing particles, *Atmos. Chem. Phys.*, 19, 1195–1206,  
662 <https://doi.org/10.5194/acp-19-1195-2019>, 2019.

663 Link, M. F., Kim, J., Park, G., Lee, T., Park, T., Babar, Z. B., Sung, K., Kim, P., Kang, S., Kim, J. S., Choi,  
664 Y., Son, J., Lim, H. J., and Farmer, D.K.: Elevated production of NH<sub>4</sub>NO<sub>3</sub> from the photochemical

665 processing of vehicle exhaust: Implications for air quality in the Seoul Metropolitan Region, *Atmos.*  
666 *Environ.*, 156, 95–101, <https://doi.org/10.1016/j.atmosenv.2017.02.031>, 2017.

667 Liu, Y. Z., Zhu, Q. Z., Huang, J. P., Hua, S., and Jia, R.: Impact of dust-polluted convective clouds over the  
668 Tibetan Plateau on downstream precipitation, *Atmos. Environ.*, 209, 67–77,  
669 <https://doi.org/10.1016/j.atmosenv.2019.04.001>, 2019.

670 Liu, Q., Liu, D. T., Gao, Q., Tian, P., Wang, F., Zhao, D. L., Bi, K., Wu, Y. Z., Ding, S., Hu, K., Zhang, J.  
671 L., Ding, D. P., and Zhao, C. S.: Vertical characteristics of aerosol hygroscopicity and impacts on  
672 optical properties over the North China Plain during winter, *Atmos. Chem. Phys.*, 20, 3931–3944,  
673 <https://doi.org/10.5194/acp-20-3931-2020>, 2020a.

674 Liu, D. T., Hu, K., Zhao, D. L., Ding, S., Wu, Y. F., Zhou, C., Yu, C. J., Tian, P., Liu, Q., Bi, K., Wu, Y. Z.,  
675 Hu, B., Ji, D. S., Kong, S. F., Ouyang, B., He, H., Huang, M. Y., and Ding, D.P.: Efficient Vertical  
676 Transport of Black Carbon in the Planetary Boundary Layer, *Geo. Res. Lett.*, 47, 1–10, <https://doi.org/10.1029/2020GL088858>, 2020b

678 Liu, H. K., Wang, Q. Y., Xing, L., Zhang, Y., Zhang, T., Ran, W. K., and Cao, J. J.: Measurement report:  
679 quantifying source contribution of fossil fuels and biomass-burning black carbon aerosol in the  
680 southeastern margin of the Tibetan Plateau, *Atmos. Chem. Phys.*, 21, 973–987,  
681 <https://doi.org/10.5194/acp-21-973-2021>, 2021.

682 Liu, X. D., Dong, B. W., Yin, Z. Y., Smith, R. S., Guo, Q. C.: Continental drift and plateau uplift control  
683 origination and evolution of Asian and Australian monsoons, *Sci. Rep.*, 7, 40344, <https://doi.org/10.1038/srep40344>, 2017.

685 Luo, M., Liu, Y. Z., Zhu, Q. Z., Tang, Y. H., and Alam, K.: Role and mechanisms of black carbon affecting  
686 water vapor transport to Tibet, *Remote Sens.*, 12, 231, <https://doi.org/10.3390/rs12020231>, 2020.

687 Ma, L., Li, M., Huang, Z. X., Li, L., Gao, W., Nian, H. Q., Zou, L. L., Fu, Z., Gao, J., Chai, F. H., and  
688 Zhou, Z.: Real time analysis of lead-containing atmospheric particles in Beijing during springtime by  
689 single particle aerosol mass spectrometry, *Chemosphere*, 154, 454–462,  
690 <https://doi.org/10.1016/j.chemosphere.2016.04.001>, 2016.

691 Ma, X., Yu, F., and Luo, G.: Aerosol direct radiative forcing based on GEOS-Chem-APM and uncertainties,  
692 *Atmos. Chem. Phys.*, 12, 5563–5581, <https://doi.org/10.5194/acp-12-5563-2012>, 2012.

693 Matsui, H.: Black carbon simulations using a size- and mixingstate-resolved three-dimensional model: 2.  
694 Aging timescale and its impact over East Asia, *J. Geophys. Res. Atmos.*, 121, 1808–1821,  
695 <https://doi.org/10.1002/2015jd023999>, 2016.

696 Meng, J. J., Wang, G. H., Li, J. J., Cheng, C. L., Ren, Y. Q., Huang, Y., Cheng, Y. T., Cao, J. J., and Zhang, T.:  
697 Seasonal characteristics of oxalic acid and related SOA in the free troposphere of Mt. Hua, central  
698 China: implications for sources and formation mechanisms, *Sci. Total Environ.*, 493, 1088–1097,  
699 <https://doi.org/10.1016/j.scitotenv.2014.04.086>, 2014.

700 Meng, J. J., Liu, X. D., Hou, Z. F., Yi, Y. N., Yan, L., Li, Z., Cao, J.J., Li, J. J., Wang, G. H.: Molecular  
701 characteristics and stable carbon isotope compositions of dicarboxylic acids and related compounds in  
702 the urban atmosphere of the North China Plain: implications for aqueous phase formation of SOA  
703 during the haze periods, *Sci. Total Environ.*, 705, 135256,  
704 <https://doi.org/10.1016/j.scitotenv.2019.135256>, 2020.

705 Ng, N. L., Canagaratna, M. R., Jimenez, J. L., Chhabra, P. S., Seinfeld, J. H., and Worsnop, D. R.: Changes  
706 in organic aerosol composition with aging inferred from aerosol mass spectra, *Atmos. Chem. Phys.*,  
707 11(13), 6465–6474, <https://doi.org/10.5194/acp-11-6465-2011>, 2011.

708 Peng, J. F., Hu, M., Guo, S., Du, Z. F., Zheng, J., Shang, D. J., Zamora, M. L., Zeng, L. M., Shao, M., Wu,

709 Y. S., Zheng, J., Wang, Y., Glen, C. R., Collins, D. R., Molina, M. J., and Zhang, R. Y.: Markedly  
710 enhanced absorption and direct radiative forcing of black carbon under polluted urban environments,  
711 *P. Natl. Acad. Sci. USA*, 113, 4266–4271, <https://doi.org/10.1073/pnas.1602310113>, 2016.

712 Pratt, K. A., Hatch, L. E., and Prather, K. A.: Seasonal volatility dependence of ambient particle phase  
713 amines, *Environ. Sci. Technol.*, 43, 5276–5281, <https://doi.org/10.1021/es803189n>, 2009.

714 Pratt, K. A., Murphy, S. M., Subramanian, R., DeMott, P. J., Kok, G. L., Campos, T., Rogers, D. C., Prenni,  
715 A. J., Heymsfield, A. J., Seinfeld, J. H., and Prather, K. A.: Flight-based chemical characterization of  
716 biomass burning aerosols within two prescribed burn smoke plumes, *Atmos. Chem. Phys.*, 11,  
717 12549–12565, <https://doi.org/10.5194/acp-11-12549-2011>, 2011.

718 Qian, Y., Flanner, M. G., Leung, L. R., and Wang, W.: Sensitivity studies on the impacts of Tibetan Plateau  
719 snowpack pollution on the Asian hydrological cycle and monsoon climate, *Atmos. Chem. Phys.*, 11,  
720 1929–1948, <https://doi.org/10.5194/acp-11-1929-2011>, 2011.

721 Rehbein, P. J., Jeong, C. H., McGuire, M. L., Yao, X., Corbin, J. C., and Evans, G. J.: Cloud and fog  
722 processing enhanced gas-to-particle partitioning of trimethylamine, *Environ. Sci. Technol.*, 45,  
723 4346–4352, <https://doi.org/10.1021/es1042113>, 2011.

724 Roth, A., Schneider, J., Klimach, T., Mertes, S., van Pinxteren, D., Herrmann, H., and Borrmann, S.:  
725 Aerosol properties, source identification, and cloud processing in orographic clouds measured by  
726 single particle mass spectrometry on a central European mountain site during HCCT-2010, *Atmos.*  
727 *Chem. Phys.*, 16, 505–524, <https://doi.org/10.5194/acp-16-505-2016>, 2016.

728 Shen, L. J., Wang, H. L., Yin, Y., Chen, J. H., and Chen, K.: Observation of atmospheric new particle  
729 growth events at the summit of mountain Tai (1534 m) in Central East China, *Atmos. Environ.*, 201,  
730 148–157, <https://doi.org/10.1016/j.atmosenv.2018.12.051>, 2019.

731 Shen, L. J., Wang, H. L., Lü, S., Zhang, X. H., Yuan, J., Tao, S. K., Zhang, G. J., Wang, F., and Li, L.:  
732 Influence of pollution control on air pollutants and the mixing state of aerosol particles during the 2nd  
733 World Internet Conference in Jiaying, China, *J. Clean. Prod.*, 149, 436–447,  
734 <https://doi.org/10.1016/j.jclepro.2017.02.114>, 2017.

735 Shen, R. Q., Ding, X., He, Q. F., Cong, Z. Y., Yu, Q. Q., and Wang, X. M.: Seasonal variation of secondary  
736 organic aerosol tracers in Central Tibetan Plateau, *Atmos. Chem. Phys.*, 15, 8781–8793,  
737 <https://doi.org/10.5194/acp-15-8781-2015>, 2015.

738 Sirois, A. and Bottenheim, J. W.: Use of backward trajectories to interpret the 5-year record of PAN and O<sub>3</sub>  
739 ambient air concentrations at Kejimikujik National Park, Nova Scotia, *J. Geophys. Res.*, 100,  
740 2867–2881, <https://doi.org/10.1029/94JD02951>, 1995.

741 Song, X. H., and Hopke, P. K.: Classification of single particles analyzed by ATOFMS using an artificial  
742 neural network, ART-2A, *Anal. Chem.*, 71, 860–865, <https://doi.org/10.1021/ac9809682>, 1999.

743 Sorooshian, A., Lu, M. L., Brechtel, F. J., Jonsson, H., Feingold, G., Flagan, R. C., and Seinfeld, J. H.: On  
744 the source of organic acid aerosol layers above clouds, *Environ. Sci. Technol.*, 41, 4647–4654,  
745 <https://doi.org/10.1021/es0630442>, 2007.

746 Sullivan, R. C., Guazzotti, S. A., Sodeman, D. A., and Prather, K. A.: Direct observations of the  
747 atmospheric processing of Asian mineral dust, *Atmos. Chem. Phys.*, 7, 1213–1236,  
748 <https://doi.org/10.5194/acp-7-1213-2007>, 2007.

749 Sun, Y. L., Wang, Z. F., Fu, P. Q., Jiang, Q. J., Yang, T., Li, J., and Ge, X. L.: The impact of relative  
750 humidity on aerosol composition and evolution processes during wintertime in Beijing, China, *Atmos.*  
751 *Environ.*, 77, 927–934, <https://doi.org/10.1016/j.atmosenv.2013.06.019>, 2013.

752 Tian, J., Wang, Q. Y., Zhang, Y., Yan, M. Y., Liu, H. K., Zhang, N. N., Ran, W. K., and Cao, J. J.: Impacts



753 of primary emissions and secondary aerosol formation on air pollution in an urban area of China  
754 during the COVID-19 lockdown, *Environ. Int.*, 150, 106426–14,  
755 <https://doi.org/10.1016/j.envint.2021.106426>, 2021.

756 Wang, A. Q., Xie, X. N., Liu, X. D., and Yin, Z. Y.: Direct Radiative Effect (DRE) of Dust Aerosols on  
757 West African and East Asian Monsoon: The Role of Ocean-Atmosphere Interactions, *J. Geophys. Res.*  
758 *Atmos.*, 127, 1–20, <https://doi.org/10.1029/2021JD035917>, 2022.

759 Wang, G. H., Zhang, R. Y., Gomez, M. E., Yang, L. X., Zamora, M. L., Hu, M., Lin, Y., Peng, J. F., Guo, S.,  
760 Meng, J. J., Li, J. J., Cheng, C. L., Hu, T. F., Ren, Y. Q., Wang, Y. S., Gao, J., Cao, J. J., An, Z. S.,  
761 Zhou, W. J., Li, G. H., Wang, J. Y., Tian, P. F., Marrero-Ortiz, W., Secret, J., Du, Z. F., Zheng, J.,  
762 Shang, D. J., Zeng, L. M., Shao, M., Wang, W. G., Huang, Y., Wang, Y., Zhu, Y. J., Li, Y. X., Hu, J. X.,  
763 Pan, B. W., Cai, L., Cheng, Y. T., Ji, Y. M., Zhang, Y., Rosenfeld, D., Liss, P. S., Duce, R. A., Kolb, C.  
764 E., and Molina, M. J.: Persistent sulfate formation from London Fog to Chinese haze, *P. Natl. Acad.*  
765 *Sci. USA*, 113(48), 13630–13635, <https://doi.org/10.1073/pnas.1616540113>, 2016.

766 Wang, H. C., Lu, K. D., Chen, X. R., Zhu, Q. D., Chen, Q., Guo, S., Jiang, M. Q., Li, X., Shang, D. J., Tan,  
767 Z. F., Wu, Y. S., Wu, Z. J., Zou, Q., Zheng, Y., Zeng, L. M., Zhu, T., Hu, M., Zhang, Y. H.: High  
768 N<sub>2</sub>O<sub>5</sub> concentrations observed in urban Beijing: implications of a large nitrate formation pathway,  
769 *Environ. Sci. Technol. Lett.*, 4, 416–420, <https://doi.org/10.1021/acs.estlett.7b00341>, 2017.

770 Wang, H. L., An, J. L., Shen, L. J., Zhu, B., Xia, L., Duan, Q., and Zou, J. N.: Mixing state of ambient  
771 aerosols in Nanjing city by single particle mass spectrometry, *Atmos. Environ.*, 132, 123–132,  
772 <https://dx.doi.org/10.1016/j.atmosenv.2016.02.032>, 2016.

773 Wang, Q. Y., Han, Y. M., Ye, J. H., Liu, S. X., Pongpiachan, S., Zhang, N. N., Han, Y. M., Tian, J., Wu, C.,  
774 Long, X., Zhang, Q., Zhang, W. Y., Zhao, Z. Z., and Cao, J. J.: High contribution of secondary brown  
775 carbon to aerosol light absorption in the southeastern margin of Tibetan Plateau, *Geophys. Res. Lett.*,  
776 46, 4962–4970, <https://doi.org/10.1029/2019GL082731>, 2019a.

777 Wang, H. L., Shen, L. J., Yin, Y., Chen, K., Chen, J. H., and Wang, Y. S.: Characteristics and mixing state  
778 of aerosol at the summit of Mount Tai (1534 m) in Central East China: First measurements with  
779 SPAMS, *Atmos. Environ.*, 213, 273–284, <https://doi.org/10.1016/j.atmosenv.2019.06.021>, 2019b.

780 Wang, Q. Y., Cao, J. J., Han, Y. M., Tian, J., Zhu, C. S., Zhang, Y., Zhang, N. N., Shen, Z. X., Ni, H. Y.,  
781 Zhao, S. Y., and Wu, J. R.: Sources and physicochemical characteristics of black carbon aerosol from  
782 the southeastern Tibetan Plateau: internal mixing enhances light absorption, *Atmos. Chem. Phys.*, 18,  
783 4639–4656, <https://doi.org/10.5194/acp-18-4639-2018>, 2018.

784 Wenzel, R. J., Liu, D.-Y., Edgerton, E. S., and Prather, K. A.: Aerosol time-of-flight mass spectrometry  
785 during the Atlanta Supersite Experiment: 2. Scaling procedures, *J. Geophys. Res.*, 108, 8427,  
786 <http://dx.doi.org/10.1029/2001jd001563>, 2003.

787 Wood, E. C., Canagaratna, M. R., Herndon, S. C., Onasch, T. B., Kolb, C. E., Worsnop, D. R., Kroll, J. H.,  
788 Knighton, W. B., Seila, R., Zavala, M., Molina, L. T., DeCarlo, P. F., Jimenez, J. L., Weinheimer, A. J.,  
789 Knapp, D. J., Jobson, B. T., Stutz, J., Kuster, W. C., and Williams, E. J.: Investigation of the  
790 correlation between odd oxygen and secondary organic aerosol in Mexico City and Houston, *Atmos.*  
791 *Chem. Phys.*, 10, 8947–8968, <https://doi.org/10.5194/acp-10-8947-2010>, 2010.

792 Xu, L. L., Wu, X., Hong, Z. Y., Zhang, Y. R., Deng, J. J., Hong, Y. W., and Chen, J. S.: Composition,  
793 mixing state, and size distribution of single submicron particles during pollution episodes in a coastal  
794 city in southeast China, *Environ. Sci. Pollut. Res.*, 26, 1464–1473,  
795 <https://doi.org/10.1007/s11356-018-3469-x>, 2018.

796 Xu, W. Q., Han, T. T., Du, W., Wang, Q. Q., Chen, C., Zhao, J., Zhang, Y. J., Li, J., Fu, P. Q., Wang, Z. F.,

797 Worsnop, D. R., and Sun, Y. L.: Effects of aqueous-phase and photochemical processing on secondary  
798 organic aerosol formation and evolution in Beijing, China, *Environ. Sci. Technol.*, 51, 762–770,  
799 <https://doi.org/10.1021/acs.est.6b04498>, 2017.

800 Xue, J., Griffith, S. M., Yu, X., Lau, A.K.H., and Yu, J. Z.: Effect of nitrate and sulfate relative abundance  
801 in PM<sub>2.5</sub> on liquid water content explored through half-hourly observations of inorganic soluble  
802 aerosols at a polluted receptor site, *Atmos. Environ.*, 99, 24–31,  
803 <https://doi.org/10.1016/j.atmosenv.2014.09.049>, 2014.

804 Yang, J., Ma, S. X., Gao, B., Li, X. Y., Zhang, Y. J., Cai, J., Li, M., Yao, L. A., Huang, B., and Zheng, M.:  
805 Single particle mass spectral signatures from vehicle exhaust particles and the source apportionment  
806 of on-line PM<sub>2.5</sub> by single particle aerosol mass spectrometry, *Sci. Total Environ.*, 593, 310–318,  
807 <https://doi.org/10.1016/j.scitotenv.2017.03.099>, 2017.

808 Zauscher, M. D., Wang, Y., Moore, M. J. K., Gaston, C. J., and Prather, K. A.: Air Quality Impact and  
809 Physicochemical Aging of Biomass Burning Aerosols during the 2007 San Diego Wildfires, *Environ.*  
810 *Sci. Technol.*, 47, 7633–7643, <https://doi.org/10.1021/es4004137>, 2013.

811 Zaveri, R. A., Barnard, J., Easter, R., Riemer, N., and West, M.: Particle-resolved simulation of aerosol size,  
812 composition, mixing state, and the associated optical and cloud condensation nuclei activation  
813 properties in an evolving urban plume, *J. Geophys. Res.*, 115, D17210,  
814 <https://doi.org/10.1029/2009JD013616>, 2010.

815 Zhang, G. H., Bi, X. H., Chan, L. Y., Li, L., Wang, X. M., Feng, J. L., Sheng, G. Y., Fu, J. M., Li, M., and  
816 Zhou, Z.: Enhanced trimethylamine-containing particles during fog events detected by single particle  
817 aerosol mass spectrometry in urban Guangzhou, China, *Atmos. Environ.*, 55, 121–126,  
818 <https://doi.org/10.1016/j.atmosenv.2012.03.038>, 2012.

819 Zhang, G. H., Han, B. X., Bi, X. H., Dai, S. X., Huang, W., Chen, D. H., Wang, X. M., Sheng, G. Y., Fu, J.  
820 M., and Zhou, Z.: Characteristics of individual particles in the atmosphere of Guangzhou by single  
821 particle mass spectrometry, *Atmos. Res.*, 153, 286–295,  
822 <https://doi.org/10.1016/j.atmosres.2014.08.016>, 2015.

823 Zhang, J. K., Luo, B., Zhang, J. Q., Ouyang, F., Song, H. Y., Liu, P. C., Cao, P., Schäfer, K., Wang, S. G.,  
824 Huang, X. J., and Lin, Y. F.: Analysis of the characteristics of single atmospheric particles in Chengdu  
825 using single particle mass spectrometry, *Atmos. Environ.*, 157, 91–100,  
826 <https://doi.org/10.1016/j.atmosenv.2017.03.012>, 2017a.

827 Zhang, G. H., Lin, Q. H., Peng, L., Yang, Y. X., Fu, Y. Z., Bi, X. H., Li, M., Chen, D. H., Chen, J. X., Cai,  
828 Z., Wang, X. M., Peng, P. A., Sheng, G. Y., Zhou, Z.: Insight into the in-cloud formation of oxalate  
829 based on in situ measurement by single particle mass spectrometry, *Atmos. Chem. Phys.*, 17 (22),  
830 13891–13901, <https://doi.org/10.5194/acp-17-13891-2017>, 2017b.

831 Zhang, G. H., Han, B. X., Bi, X. H., Dai, S. X., Huang, W., Chen, D. H., Wang, X. M., Sheng, G. Y., Fu, J.  
832 M., and Zhou, Z.: Characteristics of individual particles in the atmosphere of Guangzhou by single  
833 particle mass spectrometry, *Atmos. Res.*, 153, 286–295,  
834 <https://dx.doi.org/10.1016/j.atmosres.2014.08.016>, 2015.

835 Zhang, G. H., Lian, X. F., Fu, Y. Z., Lin, Q. H., Li, L., Song, W., Wang, Z. Y., Tang, M. J., Chen, D. H., Bi,  
836 X. H., Wang, X. M., and Sheng, G. Y.: High secondary formation of nitrogen-containing organics  
837 (NOCs) and its possible link to oxidized organics and ammonium, *Atmos. Chem. Phys.*, 20,  
838 1469–1481, <https://doi.org/10.5194/acp-20-1469-2020>, 2020.

839 Zhang, X. H., Xu, J. Z., Kang, S. C., Zhang, Q., and Sun, J. Y.: Chemical characterization and sources of  
840 submicron aerosols in the northeastern Qinghai–Tibet Plateau: insights from high-resolution mass

841 spectrometry, *Atmos. Chem. Phys.*, 19, 7897–7911, <https://doi.org/10.5194/acp-19-7897-2019>, 2019a.

842 Zhang, G. H., Lin, Q. H., Peng, L., Yang, Y. X., Jiang, F., Liu, F. X., Song, W., Chen, D. H., Cai, Z., Bi, X.

843 H., Miller, M., Tang, M. J., Huang, W. L., Wang, X. M., Peng, P. A., Shen, G. Y.: Oxalate Formation

844 Enhanced by Fe-Containing Particles and Environmental Implications, *Environ. Sci. Technol.*, 53,

845 1269–1277, <https://doi.org/10.1021/acs.est.8b05280>, 2019b.

846 Zhang, N. N., Cao, J. J., Xu, H. M., and Zhu, C. S.: Elemental compositions of PM<sub>2.5</sub> and TSP in Lijiang,

847 southeastern edge of Tibetan Plateau during pre-monsoon period, *Particuology*, 11(1), 63–69,

848 <https://doi.org/10.1016/j.partic.2012.08.002>, 2013.

849 Zhang, S. P., Xing, J., Sarwar, G., Ge, Y. L., He, H., Duan, F. K., Zhao, Y., He, K. B., Zhu, L. D., Chu, B.

850 W.: Parameterization of heterogeneous reaction of SO<sub>2</sub> to sulfate on dust with coexistence of NH<sub>3</sub> and

851 NO<sub>2</sub> under different humidity conditions, *Atmos. Environ.*, 208, 133–140,

852 <https://doi.org/10.1016/j.atmosenv.2019.04.004>, 2019c.

853 Zhao, S. Y., Tie, X. X., Long, X., and Cao, J. J.: Impacts of Himalayas on black carbon over the Tibetan

854 Plateau during summer monsoon, *Sci. Total Environ.*, 598, 307–318,

855 <https://doi.org/10.1016/j.scitotenv.2017.04.101>, 2017.

856

857 Table 1. The number concentrations, average percentages and characteristic ions of nine types  
 858 of particles during the entire campaign, and the average percentages of the major six particle  
 859 types during two episodes.

Type	Number count	Fraction in total (%)	Episode 1 (%)	Episode 2 (%)	Tracer ions
rich-K	151040	30.9	29.0	39.3	$^{39}\text{K}^+$ , $^{26}\text{CN}^-$ , $^{42}\text{CNO}^-$ , $^{46}\text{NO}_2^-$ , $^{62}\text{NO}_3^-$ , $^{97}\text{HSO}_4^-$
BB	91322	18.7	11.5	14.2	$^{39}\text{K}^+$ , levoglucosan ( $^{45}\text{CHO}_2^-$ , $^{59}\text{C}_2\text{H}_3\text{O}_2^-$ , $^{71}\text{C}_3\text{H}_3\text{O}_2^-$ , $^{73}\text{C}_3\text{HO}_3^-$ ), $^{26}\text{CN}^-$ , $^{35,37}\text{Cl}^-$ , $^{42}\text{CNO}^-$ , $^{46}\text{NO}_2^-$ , $^{62}\text{NO}_3^-$ , $^{97}\text{HSO}_4^-$
OC	62446	12.8	8.1	10.0	$^{27}\text{C}_2\text{H}_3^+$ , $^{37}\text{C}_3\text{H}^+$ , $^{38}\text{C}_3\text{H}_2^+$ , $^{39}\text{K}^+/\text{C}_3\text{H}_3^+$ , $^{43}\text{C}_2\text{H}_3\text{O}^+$ , $^{51}\text{C}_4\text{H}_3^+$ , $^{26}\text{CN}^-$ , $^{42}\text{CNO}^-$ , $^{46}\text{NO}_2^-$ , $^{62}\text{NO}_3^-$ , $^{97}\text{HSO}_4^-$
Ammonium	58317	11.9	17.5	13.5	$^{12}\text{C}^+$ , $^{18}\text{NH}_4^+$ , $^{39}\text{K}^+$ , $^{58}\text{C}_2\text{H}_5\text{NHCH}_2^+$ , $^{97}\text{HSO}_4^-$ , $^{195}\text{H}(\text{HSO}_4)_2^-$
EC-aged	53337	10.9	10.0	17.2	$\text{C}_n^\pm$ (n=1~5), $^{39}\text{K}^+$ , $^{97}\text{HSO}_4^-$
Dust	52533	10.7	20.3	1.3	$^{40}\text{Ca}^+$ , $^{56}\text{CaO}^+$ , $^{16}\text{O}^-$ , $^{17}\text{OH}^-$ , $^{76}\text{SiO}_3^-$ , $^{79}\text{PO}_3^-$
NaK-SN	13726	2.8	na	na	$^{23}\text{Na}^+$ , $^{39}\text{K}^+$ , $^{62}\text{NO}_3^-$ , $^{97}\text{HSO}_4^-$
Metal	4672	1.0	na	na	$^{51}\text{V}^+$ , $^{56}\text{Fe}^+$ , $^{64,66,68}\text{Zn}^+$ , $^{206,207,208}\text{Pb}^+$
Others	1580	0.3	na	na	No obvious characteristic peaks

860

861 **Figure captions:**

862 **Figure 1.** Box and whisker diurnal plots of the number concentration of the main particle  
863 types (a) rich-Potassium (K), (b) Biomass burning (BB), (c) Organic carbon (OC), (d)  
864 Ammonium, (e) Element carbon (EC)-aged, (f) Dust in hourly resolution. The lower, middle,  
865 and upper lines of the boxes denote the 25<sup>th</sup>, 50<sup>th</sup>, and 75<sup>th</sup> percentiles. The lower and upper  
866 whiskers represent the 10<sup>th</sup> and 90<sup>th</sup> percentiles, respectively. Average values are shown in  
867 white dots.

868 **Figure 2.** Maps of the mean HYSPLIT back trajectory clusters (72 h) at the height of 500 m  
869 during the whole field observation. Embedded pie chart represents the relative fraction of  
870 each particle type in the four clusters.

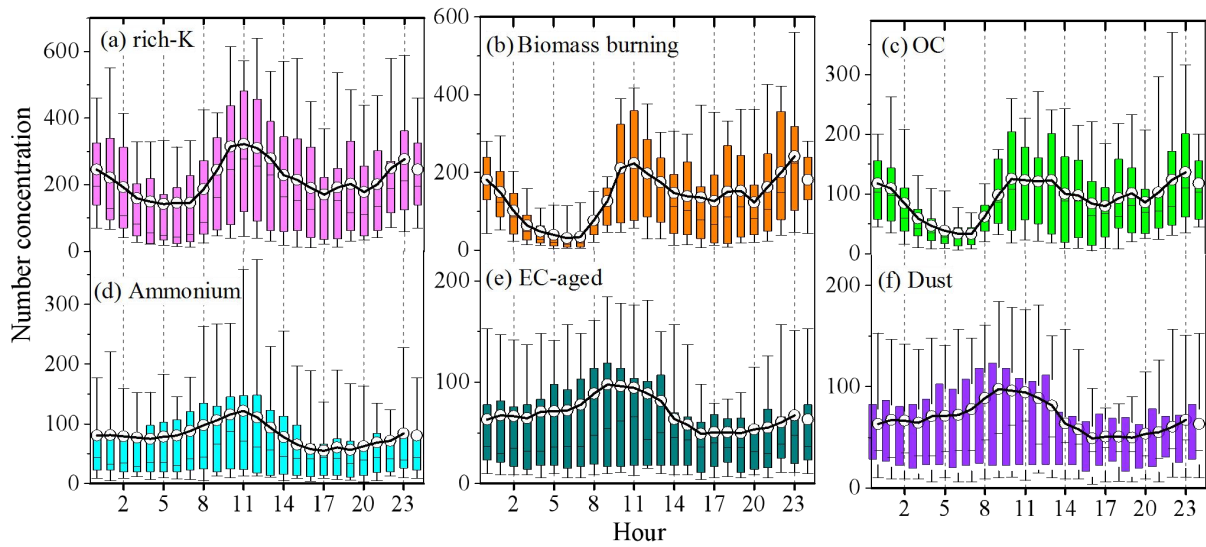
871 **Figure 3.** Size distributions of the relative number fraction (%) of the total particles for nine  
872 groups during (a) the total sampling campaign and two episodes of (b) E1 and (c) E2.

873 **Figure 4.** Number fractions of secondary markers associated with the six particle types  
874 (rich-K, BB, OC, Ammonium, EC-aged, Dust) during the whole observation. Secondary  
875 species include sulfate ( $^{97}\text{HSO}_4^-$ ), sulfuric acid ( $^{195}\text{H}(\text{HSO}_4)_2^-$ ), nitrate ( $^{62}\text{NO}_3^-$ ), ammonium  
876 ( $^{18}\text{NH}_4^+$ ), amine ( $^{58}\text{C}_2\text{H}_5\text{NHCH}_2^+$ ), and oxalate ( $^{89}\text{HC}_2\text{O}_4^-$ ) ions.

877 **Figure 5.** Number fractions of secondary markers associated with the six particle types (i.e.,  
878 rich-K, BB, OC, Ammonium, EC-aged, and Dust) in four clusters. Secondary species  
879 abbreviations as in Figure 4.

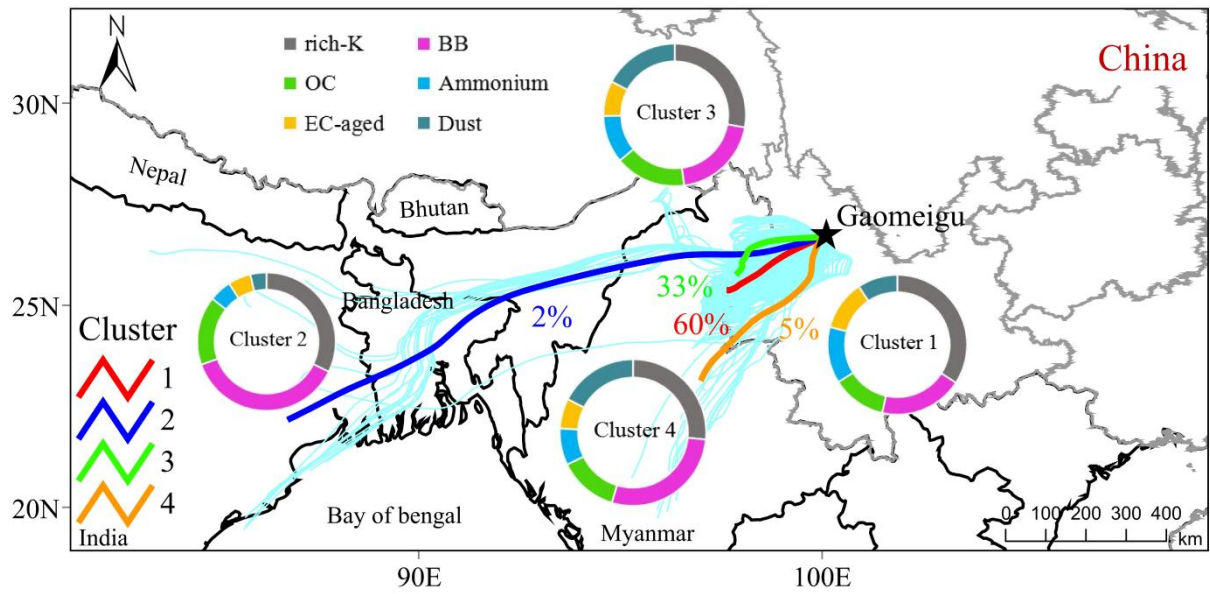
880 **Figure 6.** Correlations between the relative number fractions of the secondary species (a)  
881  $^{43}\text{C}_2\text{H}_3\text{O}^+$ , (b)  $^{89}\text{HC}_2\text{O}_4^-$ , (c)  $^{18}\text{NH}_4^+$ , (d)  $^{62}\text{NO}_3^-$ , (e)  $^{97}\text{HSO}_4^-$  and  $\text{O}_x$  concentration during E1  
882 (blue square) and E2 (red dot).

883 **Figure 7.** Correlations between the relative number fractions of the secondary species (a)  
884  $^{43}\text{C}_2\text{H}_3\text{O}^+$ , (b)  $^{89}\text{HC}_2\text{O}_4^-$ , (c)  $^{18}\text{NH}_4^+$ , (d)  $^{62}\text{NO}_3^-$ , (e)  $^{97}\text{HSO}_4^-$  and relative humidity (RH)  
885 during E1 (cyan dot) and E2 (orange square).



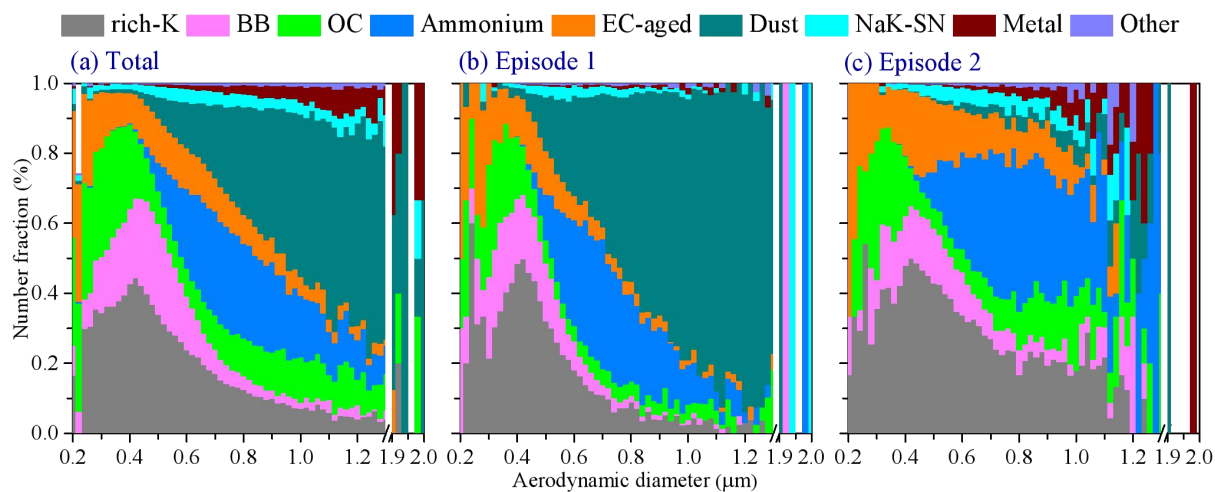
886

887 Figure 1. Box and whisker diurnal plots of the number concentration of the main particle types (a)  
 888 rich-Potassium (K), (b) Biomass burning (BB), (c) Organic carbon (OC), (d) Ammonium, (e) Element  
 889 carbon (EC)-aged, (f) Dust in hourly resolution. The lower, middle, and upper lines of the boxes denote the  
 890 25<sup>th</sup>, 50<sup>th</sup>, and 75<sup>th</sup> percentiles. The lower and upper whiskers represent the 10<sup>th</sup> and 90<sup>th</sup> percentiles,  
 891 respectively. Average values are shown in white dots.



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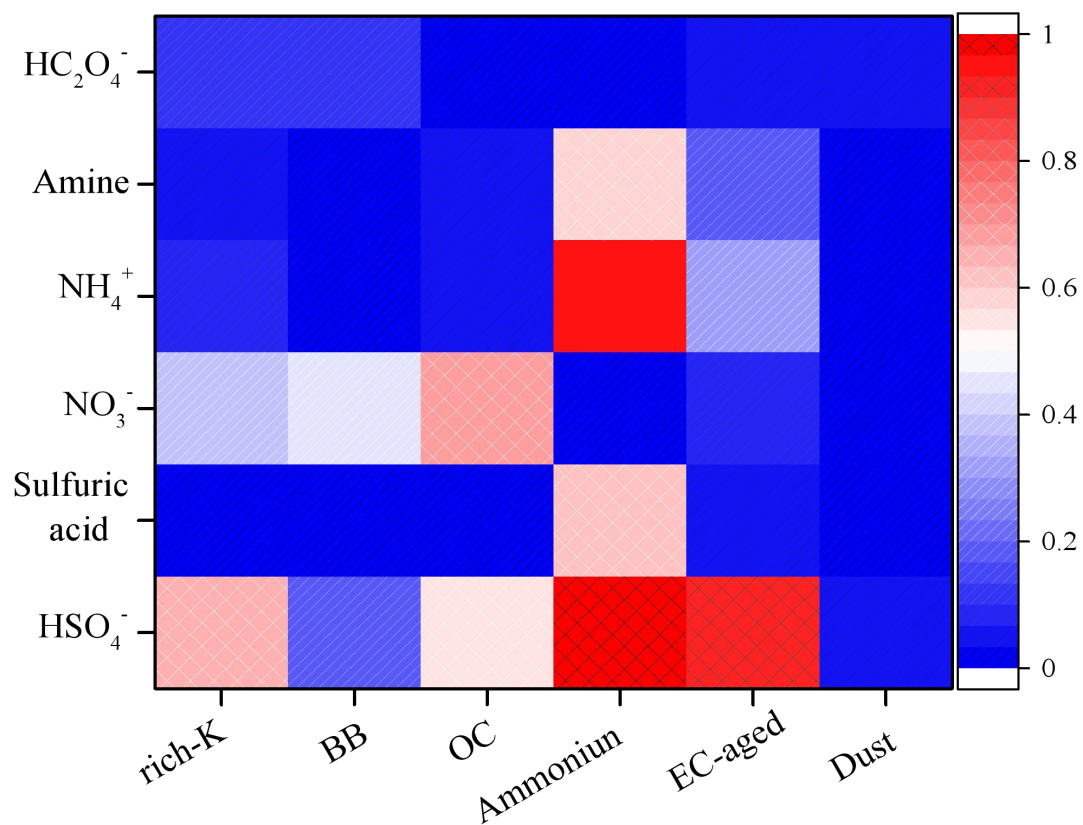
893 Figure 2. Maps of the mean HYSPLIT back trajectory clusters (72 h) at the height of 500 m during the  
 894 whole field observation. Embedded pie chart represents the relative fraction of each particle type in the  
 895 four clusters.



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897 Figure 3. Size distributions of the relative number fraction (%) of the total particles for nine groups during  
 898 (a) the total sampling campaign and two episodes of (b) E1 and (c) E2.

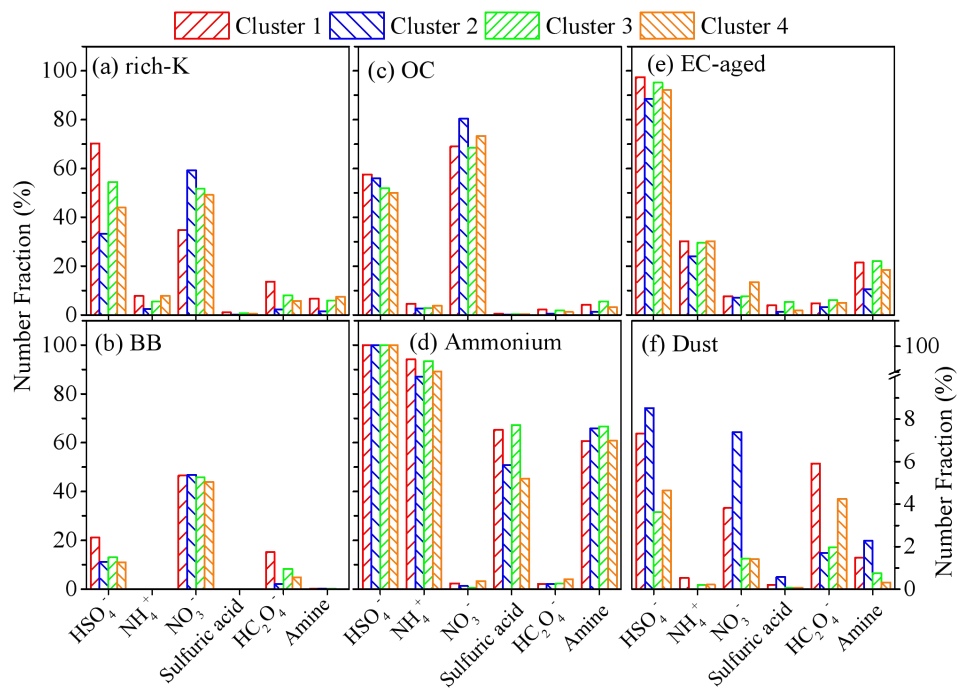




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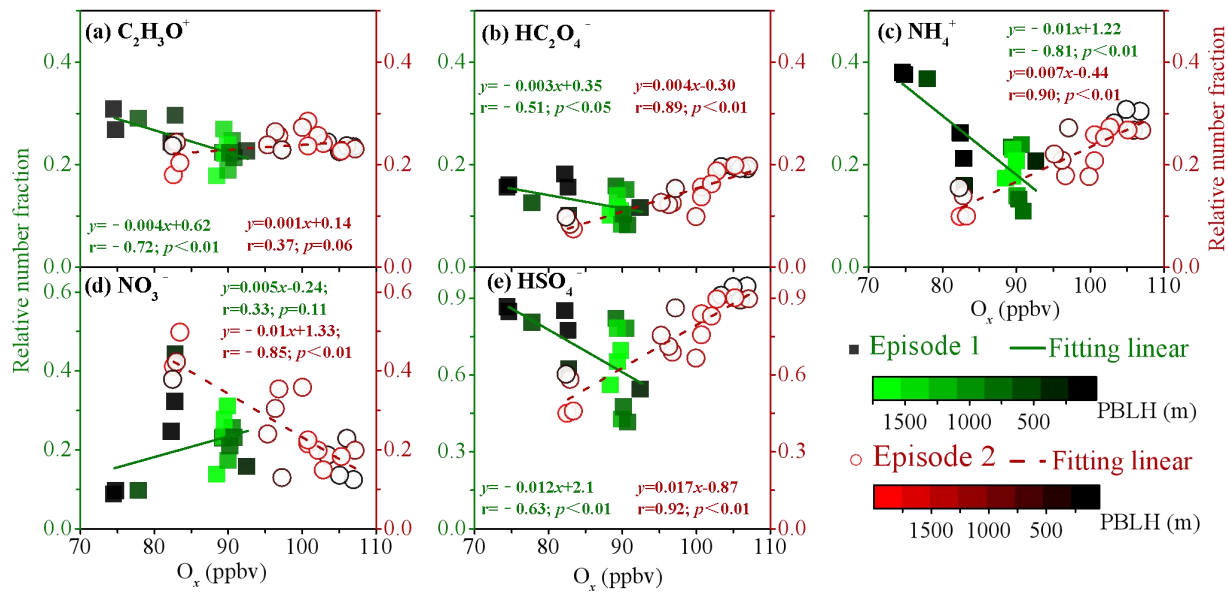
900 Figure 4. Number fractions of secondary markers associated with the six particle types (rich-K, BB, OC,  
 901 Ammonium, EC-aged, Dust) during the whole observation. Secondary species include sulfate (<sup>97</sup>HSO<sub>4</sub><sup>-</sup>),  
 902 sulfuric acid (<sup>195</sup>H(HSO<sub>4</sub>)<sub>2</sub><sup>-</sup>), nitrate (<sup>62</sup>NO<sub>3</sub><sup>-</sup>), ammonium (<sup>18</sup>NH<sub>4</sub><sup>+</sup>), amine (<sup>58</sup>C<sub>2</sub>H<sub>5</sub>NHCH<sub>2</sub><sup>+</sup>), and oxalate  
 903 (<sup>89</sup>HC<sub>2</sub>O<sub>4</sub><sup>-</sup>) ions.

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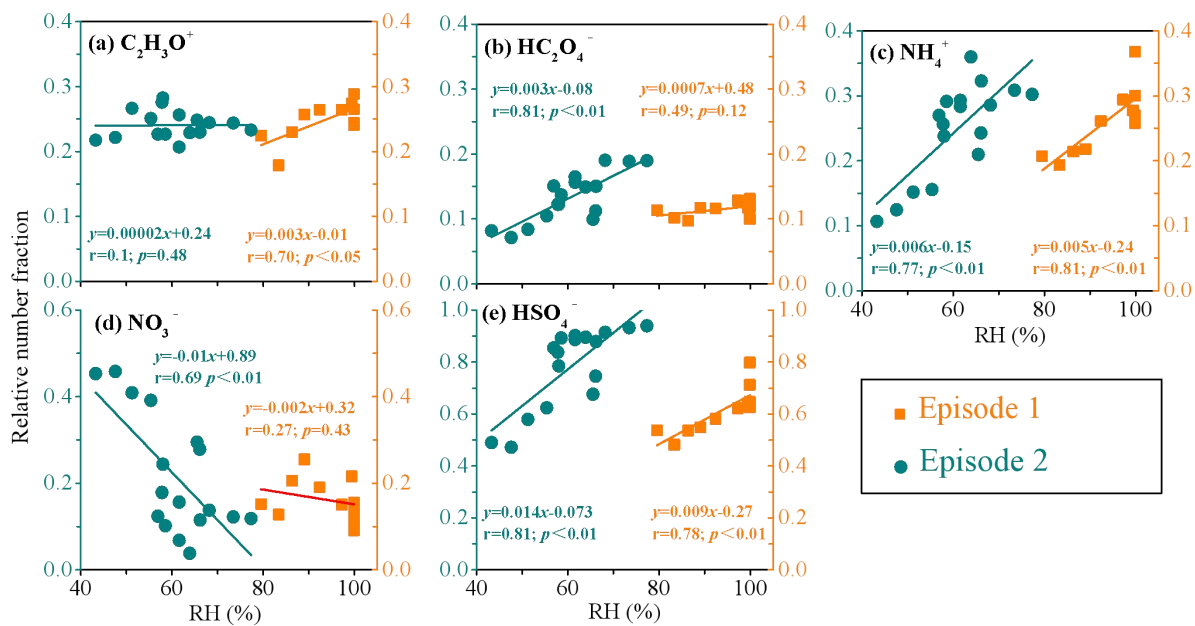
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Figure 5. Number fractions of secondary markers associated with the six particle types (i.e., rich-K, BB, OC, Ammonium, EC-aged, and Dust) in four clusters. Secondary species abbreviations as in Figure 4.



909

910 Figure 6. Correlations between the relative number fractions of the secondary species (a)  $^{43}C_2H_3O^+$ , (b)  
 911  $^{89}HC_2O_4^-$ , (c)  $^{18}NH_4^+$ , (d)  $^{62}NO_3^-$ , (e)  $^{97}HSO_4^-$  and  $O_x$  concentration during E1 (blue square) and E2 (red  
 912 dot).



913

914 Figure 7. Correlations between the relative number fractions of the secondary species (a)  $^{43}C_2H_3O^+$ , (b)  
 915  $^{89}HC_2O_4^-$ , (c)  $^{18}NH_4^+$ , (d)  $^{62}NO_3^-$ , (e)  $^{97}HSO_4^-$  and relative humidity (RH) during E1 (cyan dot) and E2  
 916 (orange square).

917