

1 **Mist Cannon Trucks Can Exacerbate the**
2 **Formation of Water-Soluble Organic Aerosol and**
3 **PM_{2.5} Pollution in the Road Environment**

4

5 Yu Xu¹, Xin-Ni Dong², Chen He³, Dai-She Wu⁴, Hong-Wei Xiao¹, Hua-Yun Xiao^{1,*}

6

7 ¹School of Environmental Science and Engineering, Shanghai Jiao Tong University,
8 Shanghai 200240, China

9 ²Jiangxi Province Science and Technology Information Institute, Nanchang 330000,
10 China

11 ³State Key Laboratory of Heavy Oil Processing, China University of Petroleum, Beijing
12 102249, China

13 ⁴School of Resource, Environmental and Chemical Engineering, Nanchang University,
14 Nanchang 330031, China

15

16

17

18 *Corresponding author: Hua-Yun Xiao

19 E-mail: xiaohuayun@sjtu.edu.cn

20 Phone: +86-173-0183-7060

21

22

23

24

25 **Abstract:** Mist cannon trucks have been widely applied in megacities in China to
26 reduce the road dust, since they are considered to be more water-saving and efficient
27 than the traditional sprinkling truck. However, their effect on the formation of water-
28 soluble organic compounds and the pollution control of fine particle (PM_{2.5}) remains
29 unknown. We characterized the chemical composition variations in PM_{2.5} collected on
30 the road sides during the simulated operations of mist cannon truck and traditional
31 sprinkling truck via Fourier transform ion cyclotron resonance mass spectrometry and
32 ion chromatography. The mass concentrations of water-soluble organic carbon in PM_{2.5}
33 showed a significant increase (62–70%) after air spraying. Furthermore, we found that
34 water-soluble organic compounds, particularly organic nitrates, increased significantly
35 via the interactions of reactive gas-phase organics, atmospheric oxidants, and aerosol
36 liquid water after air spraying, although the air spraying had a better effect on
37 suppressing road dust than the ground aspersion. Moreover, the formation of PM_{2.5} on
38 the road segment where the mist cannon truck passed by was promoted, with an increase
39 of up to 13% in mass concentration after 25–35 minutes, on average. Thus, the
40 application of mist cannon trucks potentially worsens the road atmospheric
41 environment through the increase in PM_{2.5} level and the production of a large number
42 of water-soluble organic compounds in PM_{2.5}. The overall results provide not only
43 valuable insights to the formation processes of water-soluble organic compounds
44 associated with aerosol liquid water in the road environment but also management
45 strategies to regulate the mist cannon truck operation in China.

46

47 **Keywords:** Mist cannon truck; Water mist; **Water-soluble organic compounds**; PM_{2.5};

48 Process and mechanism

49

50 **1 Introduction**

51 Over the past decade, the demand for effective road dust control has grown
52 dramatically due to the upgraded environmental protection policies. Traditionally, the
53 sprinkling trucks with the ground aspersion work well for vehicle-generated road dust.
54 The newly developed mist cannon trucks are able to spray water mist up to 120 m away
55 and 100 m high, with a droplet diameter of as small as 5 μm. They are considered to be
56 more water-saving and efficient than the traditional sprinkling truck (the ground
57 aspersion), **although the relevant argumentation work is rarely systematically studied**
58 **or reported. In recent years**, the mist cannon trucks have been widely utilized by the
59 local environmental bureau to achieve the target of strict emission control in megacities
60 in China (Wang et al., 2022). Traffic-related emissions contribute a huge amount of
61 volatile organic compounds (VOCs), nitrogen oxides (NO_x), ammonia (NH₃), and fine
62 particles (PM_{2.5}) to the urban atmosphere, which exerts adverse impacts on human
63 health and climate change (Yang et al., 2022; Deng et al., 2020). However, **no study has**
64 **investigated whether and how the water mist sprayed by mist cannon truck affects the**
65 **road atmospheric environment.**

66 The mist cannon trucks can spray a large amount of fine water mist in a short time,
67 **hence** the air humidity of local road environment where the mist cannon truck passed
68 by will increase sharply. Aerosol liquid water (ALW) exists in the condensed phase as

69 a function of gas and particle chemical composition, particle concentration, temperature
70 (T), and relative humidity (RH) (Xu et al., 2020b; Nguyen et al., 2015). An increase in
71 RH can promote the rise in ALW concentration (Guo et al., 2015). **It is well documented**
72 **that** ALW plays important role in the formation of secondary organic aerosol (SOA) via
73 the partitioning of gas-phase water-soluble organics to the particle phase and
74 subsequent reactions in the particle phase (Sareen et al., 2017; Carlton and Turpin,
75 2013). The severe haze episodes in Beijing can even be partly attributed to the
76 interactions between ALW (or high RH) and aerosol organic components (Li et al., 2019;
77 Wang et al., 2021). In particular, 40–80% of fossil-fuel-derived primary organic
78 aerosols were found to be water-soluble (Qiu et al., 2019). However, there are large
79 knowledge gaps in our current understanding on ALW-related organic compound
80 formation in the road environment with mist cannon truck operation.

81 Although few studies have systematically evaluated the ability of mist cannon
82 truck to remove road dust, it is easy to understand that the tiny water droplets generated
83 by mist cannon system can indeed capture coarse particles (i.e., dropping dust to the
84 ground) more effectively than the water column sprayed by the traditional sprinkling
85 truck. However, fine particles (PM_{2.5}) are commonly regarded as a major threat to urban
86 atmospheric environment and human health (Yue et al., 2020). Thus, assessing the
87 impact of the mist cannon truck operation on road PM_{2.5} pollution is of great
88 significance for guiding future environmental protection initiatives.

89 In this study, we simulated the operation scenes of mist cannon truck and
90 traditional sprinkling truck on the sides of the urban road (Nanchang, eastern China).

91 We collected ambient PM_{2.5} samples in these scenes. The molecular compositions of
92 water-soluble organic matter (WSOM) in the collected samples were resolved using
93 ultrahigh-resolution Fourier transform ion cyclotron resonance mass spectrometry (FT-
94 ICR MS). Moreover, we also presented the measurement of other chemical components
95 (e. g., water-soluble ions) in PM_{2.5} samples and the predicted ALW concentration. The
96 concentrations of PM_{2.5} were also monitored on the road segment where the mist
97 cannon truck passed by. The overall results will shed light on the impact of spraying
98 water mist by mist cannon truck on the formation of water-soluble organic compound
99 and its implications for PM_{2.5} pollution control in the urban road environment for the
100 first time.

101

102 **2 Materials and methods**

103 **2.1 Study site and sample collection**

104 A branch road (provincial capital north 2nd road) located in the centre of Nanchang
105 (Eastern China) was selected as the study area (Fig. 1a). This area is characterized by
106 heavy traffic and high population density. There are no typical pollution sources, such
107 as factories and garbage treatment plants, within 30 km of the study area. The trees on
108 both sides of the road are very tall and luxuriant (Fig. 1a), which makes the atmosphere
109 in this road environment relatively stable and rarely disturbed by strong winds. The
110 dominant tree species in this area are camphor trees (*Cinnamomum Camphora*). Thus,
111 the region is expected to be influenced by both anthropogenic (vehicle exhausts) and
112 biogenic VOCs.

113 Two sampling points on the road side with a distance of approximately 70 m were

114 selected, which were affected by air spray and ground aspersion, respectively (**Fig. 1b–**
115 **e**). The air spraying at a height of 8 m above the ground was to simulate the water mist
116 sprayed by the mist cannon truck. In contrast, the ground aspersion with a height of 0.4
117 m above the ground was designed to simulate the operation of traditional sprinkling
118 truck. The residence time of the fine water mist sprayed by the mist cannon truck in the
119 air is assumed to be several minutes to tens of minutes, which mainly depends on the
120 ambient temperature, RH, and wind speed. As mentioned above, the luxuriant trees on
121 both sides of the road cause the atmosphere in the road environment to be relatively
122 stable. Moreover, we found that each simulated air spray can maintain a high level of
123 RH for several minutes (about 4–8 minutes). Thus, the frequency of spraying water
124 (Milli-Q water, 18.2 MΩ cm) was set to 1 minute spraying every 8 minutes in this study.
125 This spraying operation can prevent the resuspension of road dust as much as possible.
126 The spraying was controlled by an intelligent timing irrigation equipment (Nadster,
127 China). The diameter of the pores in the nozzle for air spraying is less than 0.8 mm,
128 while that in the nozzle for ground aspersion is approximately 6 mm.

129 PM_{2.5} samples were collected onto prebaked (500°C for ~10 hours) quartz fiber
130 filters (Pallflex, Pall Corporation, USA) using a high-volume air sampler (Series 2031,
131 Laoying, China). Sampling at the above-mentioned two sites was simultaneously
132 performed from 23 March to 26 March, 2021. The duration of each aerosol sampling
133 was approximately 4 h (9:00–13:00 LT) every day. We observed that the traffic flow on
134 25 March was higher than that on other days. The weather of sampling periods was
135 cloudy to sunny, sunny, sunny, and shower (only one short precipitation event in the

136 sampling period). The average ambient T was approximately 21 °C. The average
137 ambient RH in those periods (23–25 March) ranged between 50% and 60%. The
138 average RH can reach 84–87% after air spraying. On 26 March, the average ambient
139 RH was as high as 80% during the period of 9:00–13:00. Thus, the water spraying
140 operation was stopped when the samples were collected on 26 March. All samples were
141 stored at –28 °C prior to the analysis. In addition, the mass concentrations of PM_{2.5} were
142 measured online (Thermo Scientific 5030i, USA) from July to August 2021 near the
143 trunk road where the mist cannon truck was frequently operated (**Fig. 1a**). **Typically,**
144 **the misting cannon trucks were operated back and forth on specific road sections to**
145 **prevent the resuspension of dust. Thus, the PM_{2.5} online monitoring was performed after**
146 **the misting cannon truck passed through the monitoring point several times.**
147 Specifically, PM_{2.5} concentrations near road (81 road, Nanchang) were recorded at 5-
148 minute intervals from 10 minutes before to 50 minutes after the mist cannon truck
149 passed by.

150

151 **2.2 Chemical analysis**

152 A portion of each filter sample was ultrasonically extracted with Milli-Q water
153 (MQW). WSOM in the extracting solution was further extracted using the typical solid
154 phase extraction (SPE) method (Dittmar et al., 2008; Qiao et al., 2020; Xie et al., 2020).
155 Briefly, the cartridge (PPL, 0.5 g, Agilent) was rinsed with 18 mL of methanol (LC-MS
156 grade, Thermo Fisher) and 18 mL of HCl solution (pH = 2) in turn. Subsequently, the
157 extracts with acidity adjusting (pH = 2) were injected into cartridge. Acidified MQW

158 (18 mL) and normal MQW (6 mL) were added in turn to remove the salt and chloride
159 ion trapped in the cartridge. After drying under a stream of N₂, trapped OM was eluted
160 using 15 mL of methanol. The eluted solution was concentrated to 4 mL and then
161 preserved at -28 °C until analysis. The molecular compositions of WSOM in PM_{2.5}
162 samples were determined using a Bruker Apex Ultra Fourier transform ion cyclotron
163 resonance mass spectrometry (FT-ICR MS) (Bruker, Germany) coupled to an Apollo II
164 Electrospray ionization (ESI) (He et al., 2020). The samples were injected into the
165 ionization source at 250 μL h⁻¹ through a syringe pump. The instrument was operated
166 in the negative-ion mode, with a spray shield voltage of 4.0 kV. One hundred and
167 twenty-eight continuous scans were acquired on each analysis to increase the signal-to-
168 noise ratio of the mass spectrum. Blank was analyzed with the same procedure. Detailed
169 methodologies and data quality have been described elsewhere (He et al., 2020; He et
170 al., 2019).

171 Another filter cut was ultrasonically extracted with MQW for the determination of
172 water-soluble organic carbon (WSOC), water soluble total nitrogen (WSTN), and
173 inorganic ions. The mass concentrations of WSOC and WSTN in samples were
174 measured with a total organic carbon/total nitrogen analyser (Elementar vario, Germany)
175 (Xu et al., 2019). The mass concentration of WSOC was converted to that of WSOM
176 using a conversion factor of 1.8 (Müller et al., 2017; Finessi et al., 2012; Yttri et al.,
177 2007; Simon et al., 2011). The mass concentrations of water-soluble inorganic ions,
178 such as SO₄²⁻, NO₃⁻, NH₄⁺, and K⁺, were measured using an ion chromatography
179 system (Dionex, Thermo, USA) (Xu et al., 2020a; Xu et al., 2022). The mass

180 concentration of water-soluble organic nitrogen (WSON) was calculated as the
181 difference in the concentrations between WSTN and inorganic nitrogen (i.e., $\text{NO}_3^- \text{N}$
182 + $\text{NO}_2^- \text{N}$ + $\text{NH}_4^+ \text{N}$) (Xu et al., 2020b). Ambient T and RH were measured using a
183 temperature and humidity monitor (CEM 9880M, China).

184

185 **2.3 Compound categorization and ALW prediction**

186 The molecular formulas assigned from FT-ICR MS were classified into four main
187 compound groups in this study. These identified groups include CHO (containing only
188 C, H, and O), CHON (containing C, H, O, and N), CHOS (containing C, H, O, and S),
189 and CHONS (containing C, H, N, O, and S) (Song et al., 2018). The double-bond
190 equivalent (DBE) was calculated to describe the unsaturation degree of **the** organic
191 compounds (see details in **Sect. S1** in the Supplement) (Schmidt et al., 2017; Qiao et
192 al., 2020). The modified aromaticity index (AI_{mod}) was also calculated to reflect the
193 aromaticity of organic molecules (see details in **Sect. S1** in the Supplement) (Schmidt
194 et al., 2017; Koch and Dittmar, 2006). The carbon oxidation state (OSc) was used to
195 indicate the evolving composition of aerosol organics that underwent oxidation
196 processes (Kroll et al., 2011). The details of OSc calculation is shown in **Sect. S1** in the
197 Supplement. According to the oxygen-to-carbon (O/C) and hydrogen-to-carbon (H/C)
198 elemental ratios, the identified molecular formulas were further classified into five
199 compound categories, including unsaturated aliphatic-like (**UA**), highly unsaturated-
200 like (**HU**), highly aromatic-like (**HA**), polycyclic aromatic-like (**PA**), and saturated-like
201 (**Sa**) molecules (Su et al., 2021). These classified compound categories were visualized

202 in the van Krevelen diagram (see details in **Sect. S1** in the Supplement).

203 The thermodynamic model ISORROPIA-II was applied to calculate the mass
204 concentrations of ALW driven by inorganic components (Guo et al., 2015; Tan et al.,
205 2017; Xu et al., 2022). The model predicts the inorganic ALW based on particle mass
206 concentrations of inorganic species, RH, and T (see details in **Sect. S2** in the
207 Supplement). Particle hygroscopicity is also influenced by organics in aerosol particles
208 (Sareen et al., 2013; Cruz and Pandis, 2000). The impact of organic fraction on aerosol
209 water is complex and depends on the composition of organic matter (Nguyen et al.,
210 2016). In this study, the mass concentration of water associated with aerosol organic
211 fraction was predicted according to the previously reported model with the Zdanovskii–
212 Stokes–Robinson mixing rule (see details in **Sect. S2** in the Supplement) (Nguyen et
213 al., 2016; Nguyen et al., 2015).

214

215 **3 Results and discussion**

216 **3.1 Chemical characteristics of PM_{2.5} in different road segments**

217 **Figure 2** compares the differences in the chemical composition of PM_{2.5} collected
218 in the air spray road segment and ground aspersion road segment. **The mass fraction of**
219 **WSOM was the highest** irrespective of the weather and road section where the samples
220 were collected, which accounted for 30–40% of the total water-soluble components
221 (**Fig. 2a–d**). From 23 to 25 March, the mass concentrations and fractions of WSOM
222 (/WSOC) were higher in the air spray road segment than in the ground aspersion road
223 segment. For the case without water spray treatment (as reference group), the chemical

224 compositions of PM_{2.5} samples collected from those two adjacent road segments only
225 showed small differences in the concentrations (**Fig. 2d, h, i**). It suggested that the
226 impact of background PM_{2.5} or meteorological factor on PM_{2.5} composition or level was
227 similar between these two study sites. Obviously, the variations in the mass
228 concentrations and fractions of WSOM from the air spray road segment to the ground
229 aspersion road segment can be attributed to the differences in water-soluble SOA yield
230 or formation pathway caused by different water spray treatments.

231 The variation pattern of ALW was similar to that of WSOC during the study
232 periods (**Fig. 2e–h**). Moreover, the mass concentrations of WSON also tended to
233 decrease from the air spray road segment to the ground aspersion road segment. Linear
234 regression analysis for all data showed that the mass concentrations of ALW ($n = 8$)
235 were significantly positively ($P < 0.01$) correlated with those of WSOC and WSON,
236 with the R² values of 0.84 and 0.75, respectively. The results were consistent with those
237 obtained by the previous studies conducted in an agriculture area in Italy (Hodas et al.,
238 2014) and a suburban forest site in Tokyo (Xu et al., 2020b). Moreover, these studies
239 by Hodas et al. (2014) and Xu et al. (2020b) suggested that the ALW dependence of
240 reactive gas uptake and subsequent aqueous reactions significantly contributed the
241 production of WSOC and WSON. Thus, the increase in ALW concentration after air
242 spraying can promote the formation of water-soluble organic compounds in PM_{2.5} in
243 the road environment.

244 Nitrate and sulfate were the most abundant inorganic components (**Fig. 2a–d**),
245 which have been identified as typical factors controlling ALW (Hodas et al., 2014).

246 From the air spray road segment to the ground aspersion road segment, the decrease in
247 nitrate concentration was more significant than that in sulfate concentration (**Fig. 2i–**
248 **k**). Moreover, the concentration of nitrate significantly correlated with that of ALW (P
249 < 0.01 , $R^2 = 0.7$). In contrast, the sulfate did not show a strong correlation with ALW
250 ($R^2 = 0.3$). As we know, the gas-phase oxidation of NO_2 by hydroxyl radical ($\bullet\text{OH}$) to
251 form nitric acid (HNO_3) is an important pathway for the formation of daytime nitrate
252 aerosol (Fu et al., 2020; Chen et al., 2020). Hydroxyl radical can be rapidly produced
253 by O_3 photolysis under conditions with abundant water vapour and sunlight (as in this
254 study) (Li et al., 2022), which is undoubtedly beneficial for the production of HNO_3 .
255 Thus, in the region with large NO_x and ammonia emissions (originated from vehicle
256 exhausts (Yang et al., 2022)), the formation of daytime nitrate aerosol could be
257 promoted by enhanced RH (24–43% of increase) caused by air spraying. This is also
258 partly supported by the thermodynamics of ammonium nitrate formation (Mozurkewich,
259 1993; Hodas et al., 2014). Additionally, it has been suggested that the formation of
260 nitrate and ALW is mutually reinforcing (Chen et al., 2022). Thus, the increase in ALW
261 concentration after air spraying was mainly driven by RH and locally (traffic emissions)
262 formed nitrate aerosol.

263 Interestingly, Ca^{2+} and Mg^{2+} showed a significant increase in the concentration
264 from the air spray road segment to the ground aspersion road segment (**Fig. 2i–k**),
265 which was contrary to the case of other components (e.g., WSOC, ALW, and nitrate).
266 In addition, during 26 March without water spray treatment, the differences in both Ca^{2+}
267 and Mg^{2+} concentrations between those two adjacent road segments were almost

268 negligible (**Fig. 2I**). It is well known that Ca^{2+} and Mg^{2+} are typical crustal materials
269 and are mainly enriched in atmospheric coarse particles (Chen and Chen, 2008). Thus,
270 a decrease in Ca^{2+} and Mg^{2+} concentrations after air spraying implied that the water mist
271 sprayed by mist cannon truck had a better effect on suppressing road dust than the
272 ground aspersion by traditional sprinkling truck.

273

274 **3.2 Molecular characteristics of water-soluble organic compounds**

275 Thousands of molecular formulas (5966–8102) were observed in WSOM in $\text{PM}_{2.5}$
276 collected from road environment (**Table 1**). The CHO molecular formulas (1089–2037)
277 accounted for 20–25% in all molecular formulas. Further, CHO compounds were
278 classified by the number of oxygen atoms in their molecules, according to which the
279 subgroups ranged from O_2 to O_{15} (**Fig. 3**). The number and intensity of dominated O_5 –
280 O_{10} subgroups accounted for 72–85% and 71–86% of the total compounds, respectively;
281 moreover, these percentages were higher than the results reported for aerosols in
282 Beijing (Xie et al., 2020). The average H/C and O/C ratios of CHO compounds varied
283 from 1.08 to 1.24 and from 0.42 to 0.49, respectively (**Table S1**). The average O/C
284 ratios were higher than the value (0.33 ± 0.11) obtained from typical urban aerosols
285 (Beijing, China), while the H/C ratios showed relatively small differences between our
286 results and observation results in Beijing (1.14 ± 0.37) (Xie et al., 2020). **In addition,**
287 **another study performed in Beijing showed that the average O/C and H/C ratios of**
288 **organic aerosols were in the range of 0.47–0.53 and 1.52–1.63, respectively (Hu et al.,**
289 **2017).** These dissimilarities might be attributed to the more complex sources of urban

290 aerosols compared to aerosols collected from road environment.

291 The average H/C and O/C ratios of CHON compounds ranged from 1.05 to 1.21
292 and 0.42 to 0.51, respectively (**Table S1**). The H/C ratio ranges of CHON compounds
293 in this study overlapped with those measured in previous studies (Su et al., 2021; Xie
294 et al., 2020). However, the O/C ratios of CHON compounds were relatively higher in
295 road-derived aerosols than in aerosols (0.36 ± 0.12) or snow ($0.32\text{--}0.37$) collected in
296 urban areas (building roof) (Su et al., 2021; Xie et al., 2020). This difference might be
297 associated with the influence of source strength (e.g., vehicle exhausts) and atmospheric
298 oxidation capacity. The number of CHON formulas (1501–2685) was much higher than
299 that of CHO formulas (**Table 1**). The assigned CHON formulas were further divided
300 into CHON₁ (N₁O₂–N₁O₁₆), CHON₂ (N₂O₂–N₂O₁₃), and CHON₃ (N₃O₂–N₃O₁₃) groups
301 (**Fig. 3**). CHON₁ was found to be the dominant nitrogen-containing species in all
302 samples, which was consistent with previous reports on urban aerosols and snow (Su et
303 al., 2021; Xie et al., 2020). Moreover, the CHON₁ compounds with O/N > 2 contributed
304 99.2–100.0% to total CHON₁ species in all samples. The CHON₂ compounds with O/N >
305 2 accounted for 90.2–100.0% of total CHON₂ species. For CHON₃ group, the
306 proportion of nitrogen-containing compounds with O/N > 2 was 53.0–61.8%. The
307 CHON species with O/N > 2, which allows an assignment of oxidized-form nitrogen,
308 are preferentially ionized in negative electrospray ionization mode (Lin et al., 2015; Su
309 et al., 2021). Studies on the compositions of organic matter in urban rainwater and
310 aerosols have suggested that numerous CHON compounds contained oxidized nitrogen
311 function groups (e.g., –ONO₂) and that NO_x-related oxidation processes can be

312 responsible for the formation of these CHON compounds (Altieri et al., 2009; Lee et
313 al., 2016). Thus, the CHON compounds with $O/N > 2$ in our $PM_{2.5}$ samples can be
314 assumed to be **mostly** in an oxidized form (e.g., organic nitrates).

315 **Figure 3** also shows the differences in the number of CHO and CHON species
316 between the air spray road segment and ground aspersion road segment. The abundance
317 of each O_n subgroup in CHO compounds considerably enhanced after air spraying,
318 especially the subgroups of O_5 – O_{11} . In contrast, the number of CHO species for these
319 two cases without water spray treatment showed a considerably **smaller** difference (**Fig.**
320 **3d**). In general, the total number of CHO compounds increased significantly after air
321 spraying (**Fig. 3** and **Table 1**). However, there was no significant change for the total
322 number of CHO species between the two cases without water spray treatment. These
323 findings implied that **the** increased ALW after air spraying can **substantially contribute**
324 to the formation of CHO compounds with a more oxygenated state.

325 The number of CHON compounds decreased significantly from the air spray road
326 segment to the ground aspersion road segment, a variation pattern of which was similar
327 to that of CHO compounds (**Fig. 3** and **Table 1**). Furthermore, the decrease **in the**
328 **number of molecules** from the air spray road segment to the ground aspersion road
329 segment was more remarkable for $CHON_1$ compounds than for $CHON_2$ compounds. In
330 contrast, the variation in the number of $CHON_3$ molecules after air spraying was less
331 significant than that of $CHON_{1-2}$ compounds. In addition, insignificant change in the
332 number of CHON compounds was found in $PM_{2.5}$ collected in the two road segments
333 without water spray (**Fig. 3d**). The potentially high abundance of organic nitrates has

334 been suggested in these road-derived aerosols, as mentioned above. Thus, the increase
335 in the number of nitrogen-containing compounds after air spraying indicated that the
336 interactions among ALW, traffic-derived reactive nitrogen, and ambient VOCs play an
337 important role in organic nitrogen compound formation in PM_{2.5}. **This consideration**
338 **can also be partly supported by the result** obtained by Xu et al. (2020b) in a suburban
339 forest in Tokyo, Japan. The authors suggested that ALW is a key driver for the formation
340 of aerosol WSON through secondary processes associated with atmospheric reactive
341 nitrogen and biogenic VOCs (Xu et al., 2020b). For the sulfur-containing compounds,
342 their molecular **numbers** just showed a relatively small change after air spraying (**Fig.**
343 **S1**). This suggested that the impact of ALW on sulfur-containing compound formation
344 was weaker than that of ALW on the formation of CHO and CHON compounds in this
345 road environment.

346

347 **3.3 CHO and CHON species formed under the influence of increased ALW**

348 The molecular compositions of CHO compounds in PM_{2.5} in the van Krevelen
349 diagram were scattered across wider ranges in the air spray road segment than in the
350 ground aspersion road segment, particularly in the sunny days (24 and 25 March) (**Fig.**
351 **4**). **Moreover, common CHO molecules accounted for 39% (sunny day) – 63% (cloudy**
352 **to sunny day) and 90–95% of CHO molecules in PM_{2.5} collected from the air spray and**
353 **ground aspersion road segments, respectively (Table 1). In contrast, common CHO**
354 **molecules contributed 81–85% of CHO molecules in PM_{2.5} collected from two road**
355 **segments without water spray (I vs II).** These results can be explained by the increased

356 molecular diversity caused by ALW-related atmospheric processes. It also implied the
357 importance of photochemical reactions in CHO compound formation. Further, the
358 unique CHO compounds were identified between PM_{2.5} samples in the air spray road
359 segment (/no water spray road segment (I)) and the ground aspersion road segment (/no
360 water spray road segment (II)) (**Fig. S2**). On 23 March and 24 March, the newly
361 emerging CHO compounds after air spraying were dominated by unsaturated aliphatic-
362 like and highly unsaturated-like compounds. However, both unsaturated-like species
363 (unsaturated aliphatic-like and highly unsaturated-like) and aromatic-like species
364 (highly aromatic-like and polycyclic aromatic-like) contributed significantly to the
365 newly emerging CHO compounds after air spraying on 25 March when the ALW and
366 traffic flow were higher than other days (**Fig. S2**). Obviously, the formation of those
367 unique CHO compounds was closely associated with increased ALW.

368 **Figure 5** shows the OSc values of the identified unique CHO molecules. The OSc
369 values of these CHO molecules were higher than those of primary vehicle exhausts (–
370 2.0 to –1.9) (Aiken et al., 2008). The OSc values of the secondary organic aerosol
371 formed via the reactions of anthropogenic and biogenic VOCs (e.g., isoprene,
372 monoterpene, toluene, alkane, and alkene) and oxidants (e.g., O₃ and/or •OH) varied
373 from –1.1 to +1.0 (Kroll et al., 2011; Li et al., 2021b), which was within the OSc value
374 ranges of CHO molecules measured in this study. In addition, hydrocarbon-like organic
375 aerosol (HOA) likely linked with primary vehicle exhausts (Sun et al., 2016) only
376 accounted for less than 6% of total unique CHO compounds (**Fig. 5**). **Although it is**
377 **difficult to classify all CHO molecules in Fig. 5, these identified unique CHO molecules**

378 can at least suggest that the water mist from air spraying can promote the formation of
379 CHO compounds and increase their molecular diversity. As mentioned previously, there
380 are dense trees on both sides of the road. Thus, these newly emerging CHO compounds
381 can be largely attributed to secondary processes associated with oxidation of vehicle
382 exhausts and biogenic VOCs by O₃ and/or •OH. In addition, we also observed increased
383 oligomerization (e.g., methylglyoxal (C₃H₄O₂) to form oligomers (C₄₋₇H₆₋₁₀O₅) in
384 particle phase) of CHO compounds after air spraying (Ma et al., 2022), particularly on
385 25 March with a high ALW and a large traffic flow (Fig. S3c). The overall results
386 implied that the water mist sprayed by mist cannon trucks can indeed enhance the
387 abundance and diversity of CHO compounds in PM_{2.5} via promoting gas-to-particle
388 partitioning of gas-phase oxidation products of VOCs and subsequent aqueous-phase
389 reactions.

390 For the CHON compounds, their molecular compositions were scattered across an
391 increased range in the van Krevelen diagram after air spraying compared to the
392 observations in the ground aspersion case (Fig. S4), which was also indicated by the
393 low proportion of common CHON molecules (45–55%) in total CHON compounds for
394 the case with air spraying treatment (Table 1). Moreover, these newly emerging CHON
395 molecules after air spraying showed a high diversity, as shown in Fig. S5. The group of
396 CHON₁ was the dominant nitrogen-containing compound in the identified unique
397 CHON compounds. On 23 March and 24 March, the main unique CHON compounds
398 emerged during air spraying were unsaturated aliphatic-like and highly unsaturated-like
399 nitrogen-containing species. The number of highly aromatic-like and polycyclic

400 aromatic-like compounds that newly emerged also increased significantly following
401 increased traffic flow and ALW (25 March). The results suggested that the increase in
402 ALW concentration after air spraying can facilitate the formation of particle-phase
403 nitrogen-containing compounds (Hallquist et al., 2009; Xu et al., 2020b).

404 Organic nitrates have been supposed to be abundant in our PM_{2.5} samples collected
405 from road environment. It is well documented that atmospheric organic nitrates are
406 primary, secondary, and tertiary byproducts of **reactions among** anthropogenic and
407 biogenic VOCs, atmospheric oxidants (e.g., O₃, •OH, and nitrate radicals), and NO_x
408 (Lee et al., 2016; Yeh and Ziemann, 2014; Su et al., 2021). In addition, numerous
409 organic nitrates are known as semivolatile compounds, which **are able to partition**
410 between the gas and particle phases when they are oxidized or photolyzed (Bean and
411 Hildebrandt Ruiz, 2016). Recently, an oxidation and hydrolysis mechanism associated
412 with **the formation of atmospheric organic nitrates** has been proposed to interpret the
413 potential origins or precursors of CHON compounds (Su et al., 2021). **In this study**, we
414 found that 68–82% of newly emerging CHON₁ compounds after air spraying can be
415 explained by oxidation (e.g., R₁OH)-product (e.g., R₁ONO₂) pair proposed by Su et al.
416 (2021) (**Fig. 6a–c**). It indicated that these newly emerging CHON₁ compounds **after air**
417 **spraying** were largely derived from the oxidation of CHO species under existence of
418 NO_x. A similar pattern was also observed in the newly emerging CHON₂ compounds
419 (**Fig. S6**). It should be pointed out that vehicle exhausts and roadside vegetation are
420 important sources for VOCs and NO_x in this road environment. However, **the number**
421 **of unique CHON₁ and CHON₂ compounds identified in the comparative cases without**

422 water spray treatment was much less than that identified in the comparative cases with
423 air spraying and ground aspersion treatments (Fig. 6 and Fig. S6). In particular, we did
424 not observe significant oligomerization of CHON compounds after air spraying (Fig.
425 S7). Thus, these results suggested that the increase in ALW caused by air spraying can
426 facilitate the formation of organic nitrates via CHO compounds as potential precursors
427 under the presence of NO_x.

428

429 3.4 Environmental implication

430 Misting cannon sprayers are commonly applied in agriculture for the distribution
431 of fertilizers, pesticides, and herbicides. In recent years, misting cannon trucks are
432 widely developed and regraded as an excellent option for road dust control due to the
433 production of tiny water droplets that can drop dust to the ground. In particular, it is a
434 high-performance system of spraying disinfection in the road environment during the
435 COVID-19 epidemic period. For the first time, we provide a detailed characterization
436 of chemical compositions in road-derived PM_{2.5} under the influence of air spraying.
437 Recent study conducted in a rural site (Shanghai, China) has suggested that gaseous
438 water-soluble organic compounds mainly partitioned to the organic phase (organic shell)
439 under the condition of RH less than 80% (relatively low ALW) but to ALW under the
440 humid condition (RH > 80%, as air spraying operation), highlighting the importance of
441 high ALW in SOA formation processes (Lv et al., 2022). This is because aerosols can
442 exist in a phase-separated form with an inorganic core and an organic shell (Yu et al.,
443 2019; Li et al., 2021a; Ushijima et al., 2021). Our results verified the formation of

444 numerous new CHO and CHON compounds by ALW-related promoting effect (Fig. 7).
445 In particular, the mass concentrations of WSOC in PM_{2.5} increased by 62–70% after air
446 spraying. Clearly, although the air spraying by mist cannon system could exert a better
447 effect on suppressing road dust than the ground aspersion, as discussed previously, air
448 pollution induced by increased water-soluble organic compounds in PM_{2.5} will be
449 exacerbated in the road environment.

450 To further reveal the influence of air spraying on PM_{2.5} pollution on the roadside,
451 we investigated the time series of percentage variation in PM_{2.5} mass concentration after
452 mist cannon truck operation at a low-speed (< 30 km h⁻¹) (Fig. 8). It should be pointed
453 out that misting cannon trucks usually operate back and forth on specific road sections
454 to prevent the resuspension of dust. After the misting cannon truck passed through the
455 monitoring site several times, repeated online PM_{2.5} monitoring (*n* = 34, within a month)
456 was performed to exclude the impact of dispersion and traffic flow on analysis results.
457 Accordingly, the resuspension of road dust was expected to exert a relatively minor
458 impact on the PM_{2.5} level near road. The concentration of PM_{2.5} showed an increasing
459 trend after the mist cannon truck passed the monitoring point for 15 minutes. Thus, the
460 water droplets sprayed by the mist cannon truck cannot directly cause an increase in
461 PM_{2.5} concentration, suggesting that the increased PM_{2.5} should be secondarily formed
462 after water mist spraying (~15 minutes). This consideration was also supported by a
463 significant increase in the concentration and number of water-soluble organic
464 compounds after air spraying (Fig. 2 and Fig. 3). After the mist cannon truck has passed
465 for 25–35 minutes, the increase proportion of PM_{2.5} concentration on the roadside

466 gradually reached the maximum (~13%, on average). Subsequently, the proportion of
467 increase in PM_{2.5} concentration gradually decreased, reaching ~6% at 50 min after the
468 mist cannon truck was operated. In addition, the width of the road segment (81 road,
469 Nanchang, China) where PM_{2.5} was monitored is very large (~43 m), implying that the
470 water mist sprayed by mist cannon truck should exert a greater promoting effect on the
471 formation of PM_{2.5} on conventional urban roads. The overall results suggested that mist
472 cannon truck cannot effectively reduce the PM_{2.5} level in the road environment, but lead
473 to aggravation of PM_{2.5} pollution.

474 The chemical composition of fine aerosol particles in the urban road atmosphere
475 is highly complex, including a lot of harmful organic compounds (polycyclic aromatic
476 hydrocarbons and nitro-aromatics), as indicated by our measurements and previous
477 study (Tong et al., 2016). Emissions from vehicles and roadside vegetation are
478 important anthropogenic and biogenic sources of reactive gas-phase OC and key
479 precursors to form SOA in urban areas (Gentner et al., 2012; Xu et al., 2020b; Tong et
480 al., 2016). In particular, organic aerosol composition in the road environment can be
481 strongly impacted by vehicle emissions (e.g., VOCs and NO_x) (Tong et al., 2016).
482 Inhalation of the particles containing harmful organics can be responsible for a number
483 of adverse health effects (Künzli et al., 2000). However, the wide application of mist
484 cannon truck by local environmental protection department undoubtedly accelerates the
485 formation processes of **water-soluble organic compounds** and PM_{2.5} associated with
486 ALW, which will further worsen the urban road environment and cause health hazards
487 to walking residents. Thus, the present study provides the crucial information for the

488 decision makers to regulate the mist cannon truck operation in many cities in China.

489 **4 Conclusions**

490 We investigated the changes in the chemical composition of PM_{2.5} collected from
491 the road sides of the urban road during the simulated operations of mist cannon truck
492 and traditional sprinkling truck. Moreover, we also conducted online measurement of
493 PM_{2.5} concentration in the urban road segment where the mist cannon truck passed by.
494 The mass concentrations of WSOC in PM_{2.5} increased significantly (62–70%) from the
495 ground aspersion road segment to the air spray road segment. Similarly, ALW also
496 showed a significant increase after air spraying. We found that the mass concentration
497 of ALW ($n = 8$) in PM_{2.5} was significantly **positively** correlated with that of WSOC and
498 WSON. Thus, the increase of ALW after air spraying can promote the formation of
499 particle-phase water-soluble organic compounds in the road environment. In addition,
500 the decrease in Ca²⁺ and Mg²⁺ concentrations after air spraying suggested that the water
501 mist sprayed by mist cannon truck has a better effect on suppressing road dust than the
502 ground aspersion by traditional sprinkling truck.

503 A comparison in the number of CHO and CHON species between the air spraying
504 and ground aspersion road segments suggested that the increase in ALW after air
505 spraying can enhance the abundance and diversity of CHO and CHON compounds in
506 PM_{2.5}. The **unique** CHO compounds **formed** after air spraying can be largely attributed
507 to secondary processes associated with oxidation of vehicle exhausts and biogenic
508 VOCs by oxidants and oligomerization. Organic nitrates were considered to be the
509 abundant nitrogen-containing compounds in PM_{2.5}. Furthermore, we found that **the**

510 **increased** organic nitrates were largely derived from the oxidation of CHO species
511 under the existence of NO_x.

512 After the mist cannon truck has passed for 25–35 minutes, PM_{2.5} concentration on
513 the road side increased by up to 13%, on average. The proportion of increase in PM_{2.5}
514 concentration gradually decreased to ~6% at 50 min after the mist cannon truck was
515 operated. The overall results suggested that although mist cannon truck **could suppress**
516 **road dust better than** the traditional sprinkling truck, air pollution induced by increased
517 aerosol water-soluble organic compounds and PM_{2.5} levels will be exacerbated in the
518 urban road environment. Our findings provide new insights into the formation
519 processes of aerosol water-soluble organic compounds associated with the water mist
520 sprayed by mist cannon truck in the road atmospheric environment.

521

522 **Data Availability.** The data presented in this work are available upon request from the
523 corresponding authors.

524

525 **Supplement.** The supplement related to this article is available online at:
526 <https://doi.org/10.5194/acp-2022-735>.

527

528 **Author contributions.** YX, H-Y.X, and DW designed the study; YX, XD and H-W.X
529 performed field measurements; YX and CH performed chemical analysis and data
530 analysis; YX wrote the original manuscript; and YX reviewed and edited the manuscript.

531 **Competing interests.** The authors declare no competing financial interest.

532

533 **Disclaimer.** Publisher's note: Copernicus Publications remains neutral with regard to
534 jurisdictional claims in published maps and institutional affiliations.

535

536 **Acknowledgements.** The authors are very grateful to Chen-Xi Li for his kind and
537 valuable comments to improve the manuscript. Yu Xu acknowledges the Shanghai
538 Sailing Program (grant no. 22YF1418700) and the Natural (Youth) Science Foundation
539 of Jiangxi, China (grant no. 20212BAB213039).

540

541 **Financial support.** This research has been supported by the Shanghai Sailing Program
542 (grant no. 22YF1418700) and the Natural (Youth) Science Foundation of Jiangxi, China
543 (grant no. 20212BAB213039).

544

545 **Review statement.** This paper was edited by Annele Virtanen and reviewed by two
546 anonymous referees.

547

548 **References**

549 Aiken, A. C., DeCarlo, P. F., Kroll, J. H., Worsnop, D. R., Huffman, J. A., Docherty, K.
550 S., Ulbrich, I. M., Mohr, C., Kimmel, J. R., Sueper, D., Sun, Y., Zhang, Q.,
551 Trimborn, A., Northway, M., Ziemann, P. J., Canagaratna, M. R., Onasch, T. B.,
552 Alfarra, M. R., Prevot, A. S. H., Dommen, J., Duplissy, J., Metzger, A.,
553 Baltensperger, U., and Jimenez, J. L.: O/C and OM/OC Ratios of Primary,

554 Secondary, and Ambient Organic Aerosols with High-Resolution Time-of-
555 Flight Aerosol Mass Spectrometry, *Environ. Sci. Technol.*, 42, 4478-4485,
556 10.1021/es703009q, 2008.

557 Altieri, K. E., Turpin, B. J., and Seitzinger, S. P.: Oligomers, organosulfates, and
558 nitrooxy organosulfates in rainwater identified by ultra-high resolution
559 electrospray ionization FT-ICR mass spectrometry, *Atmos. Chem. Phys.*, 9,
560 2533-2542, 10.5194/acp-9-2533-2009, 2009.

561 Bean, J. K. and Hildebrandt Ruiz, L.: Gas-particle partitioning and hydrolysis of
562 organic nitrates formed from the oxidation of α -pinene in environmental
563 chamber experiments, *Atmos. Chem. Phys.*, 16, 2175-2184, 10.5194/acp-16-
564 2175-2016, 2016.

565 Carlton, A. and Turpin, B.: Particle partitioning potential of organic compounds is
566 highest in the Eastern US and driven by anthropogenic water, *Atmos. Chem.*
567 *Phys.*, 13, 10203-10214, 2013.

568 Chen, H. Y. and Chen, L. D.: Importance of anthropogenic inputs and continental-
569 derived dust for the distribution and flux of water-soluble nitrogen and
570 phosphorus species in aerosol within the atmosphere over the East China Sea, *J.*
571 *Geophys. Res.: Atmos.*, 113, D11. DOI: 10.1029/2007JD009491., 2008.

572 Chen, X., Wang, H., Lu, K., Li, C., Zhai, T., Tan, Z., Ma, X., Yang, X., Liu, Y., Chen,
573 S., Dong, H., Li, X., Wu, Z., Hu, M., Zeng, L., and Zhang, Y.: Field
574 Determination of Nitrate Formation Pathway in Winter Beijing, *Environmental*
575 *Science & Technology*, 54, 9243-9253, 10.1021/acs.est.0c00972, 2020.

576 Chen, Y., Wang, Y., Nenes, A., Wild, O., Song, S., Hu, D., Liu, D., He, J., Hildebrandt
577 Ruiz, L., Apte, J. S., Gunthe, S. S., and Liu, P.: Ammonium Chloride Associated
578 Aerosol Liquid Water Enhances Haze in Delhi, India, *Environ. Sci. Technol.*,
579 10.1021/acs.est.2c00650, 2022.

580 Cruz, C. N. and Pandis, S. N.: Deliquescence and hygroscopic growth of mixed
581 inorganic– organic atmospheric aerosol, *Environ. Sci. Technol.*, 34, 4313-4319.
582 <https://doi.org/4310.1021/es9907109>, 2000.

583 Deng, F., Lv, Z., Qi, L., Wang, X., Shi, M., and Liu, H.: A big data approach to
584 improving the vehicle emission inventory in China, *Nat. Commun.*, 11, 2801,
585 10.1038/s41467-020-16579-w, 2020.

586 Dittmar, T., Koch, B., Hertkorn, N., and Kattner, G.: A simple and efficient method for
587 the solid-phase extraction of dissolved organic matter (SPE-DOM) from
588 seawater, *Limnol. Oceanogr.: Meth.*, 6, 230–235, 2008.

589 Finessi, E., Decesari, S., Paglione, M., Giulianelli, L., Carbone, C., Gilardoni, S., Fuzzi,
590 S., Saarikoski, S., Raatikainen, T., and Hillamo, R.: Determination of the
591 biogenic secondary organic aerosol fraction in the boreal forest by NMR
592 spectroscopy, *Atmos. Chem. Phys.*, 12, 941-959. [https://doi.org/910.5194/acp-](https://doi.org/910.5194/acp-5112-5941-2012)
593 5112-5941-2012, 2012.

594 Fu, X., Wang, T., Gao, J., Wang, P., Liu, Y., Wang, S., Zhao, B., and Xue, L.: Persistent
595 Heavy Winter Nitrate Pollution Driven by Increased Photochemical Oxidants in
596 Northern China, *Environmental Science & Technology*, 54, 3881-3889,
597 10.1021/acs.est.9b07248, 2020.

598 Gentner, D. R., Isaacman, G., Worton, D. R., Chan, A. W. H., Dallmann, T. R., Davis,
599 L., Liu, S., Day, D. A., Russell, L. M., Wilson, K. R., Weber, R., Guha, A.,
600 Harley, R. A., and Goldstein, A. H.: Elucidating secondary organic aerosol from
601 diesel and gasoline vehicles through detailed characterization of organic carbon
602 emissions, *P. Natl. Acad. Sci. U. S. A.*, 109, 18318-18323,
603 doi:10.1073/pnas.1212272109, 2012.

604 Guo, H. Y., Xu, L., Bougiatioti, A., Cerully, K. M., Capps, S. L., Hite Jr, J., Carlton, A.,
605 Lee, S. H., Bergin, M., and Ng, N.: Fine-particle water and pH in the
606 southeastern United States, *Atmos. Chem. Phys.*, 15, 5211-5228.
607 <https://doi.org/5210.5194/acp-5215-5211-2015>, 2015.

608 Hallquist, M., Wenger, J. C., Baltensperger, U., Rudich, Y., Simpson, D., Claeys, M.,
609 Dommen, J., Donahue, N., George, C., and Goldstein, A.: The formation,
610 properties and impact of secondary organic aerosol: current and emerging issues,
611 *Atmos. Chem. Phys.*, 9, 5155-5236, 2009.

612 He, C., Pan, Q., Li, P., Xie, W., He, D., Zhang, C., and Shi, Q.: Molecular composition
613 and spatial distribution of dissolved organic matter (DOM) in the Pearl River
614 Estuary, China, *Environ. Chem.*, 17, 10.1071/EN19051, 2019.

615 He, C., Zhang, Y., Li, Y., Zhuo, X., Li, Y., Zhang, C., and Shi, Q.: In-House Standard
616 Method for Molecular Characterization of Dissolved Organic Matter by FT-ICR
617 Mass Spectrometry, *ACS Omega*, 5, 11730-11736, 10.1021/acsomega.0c01055,
618 2020.

619 Hodas, N., Sullivan, A. P., Skog, K., Keutsch, F. N., Collett Jr, J. L., Decesari, S.,

620 Facchini, M. C., Carlton, A. G., Laaksonen, A., and Turpin, B. J.: Aerosol liquid
621 water driven by anthropogenic nitrate: Implications for lifetimes of water-
622 soluble organic gases and potential for secondary organic aerosol formation,
623 *Environ. Sci. Technol.*, 48, 11127-11136.
624 <https://doi.org/11110.11021/es5025096>, 2014.

625 Hu, W., Hu, M., Hu, W. W., Zheng, J., Chen, C., Wu, Y., and Guo, S.: Seasonal
626 variations in high time-resolved chemical compositions, sources, and evolution
627 of atmospheric submicron aerosols in the megacity Beijing, *Atmos. Chem.*
628 *Phys.*, 17, 9979-10000, 10.5194/acp-17-9979-2017, 2017.

629 Koch, B. P. and Dittmar, T.: From mass to structure: an aromaticity index for high-
630 resolution mass data of natural organic matter, *Rapid Commun. Mass Spectrom.*,
631 20, 926-932, <https://doi.org/10.1002/rcm.2386>, 2006.

632 Kroll, J. H., Donahue, N. M., Jimenez, J. L., Kessler, S. H., Canagaratna, M. R., Wilson,
633 K. R., Altieri, K. E., Mazzoleni, L. R., Wozniak, A. S., Bluhm, H., Mysak, E. R.,
634 Smith, J. D., Kolb, C. E., and Worsnop, D. R.: Carbon oxidation state as a metric
635 for describing the chemistry of atmospheric organic aerosol, *Nat. Chem.*, 3, 133-
636 139, 10.1038/nchem.948, 2011.

637 Künzli, N., Kaiser, R., Medina, S., Studnicka, M., Chanel, O., Filliger, P., Herry, M.,
638 Horak, F., Jr., Puybonnieux-Textier, V., Quénel, P., Schneider, J., Seethaler, R.,
639 Vergnaud, J. C., and Sommer, H.: Public-health impact of outdoor and traffic-
640 related air pollution: a European assessment, *Lancet (London, England)*, 356,
641 795-801, 10.1016/s0140-6736(00)02653-2, 2000.

642 Lee, B. H., Mohr, C., Lopez-Hilfiker, F. D., Lutz, A., Hallquist, M., Lee, L., Romer, P.,
643 Cohen, R. C., Iyer, S., Kurtén, T., Hu, W., Day, D. A., Campuzano-Jost, P.,
644 Jimenez, J. L., Xu, L., Ng, N. L., Guo, H., Weber, R. J., Wild, R. J., Brown, S.
645 S., Koss, A., Gouw, J. d., Olson, K., Goldstein, A. H., Seco, R., Kim, S., McAvey,
646 K., Shepson, P. B., Starn, T., Baumann, K., Edgerton, E. S., Liu, J., Shilling, J.
647 E., Miller, D. O., Brune, W., Schobesberger, S., D'Ambro, E. L., and Thornton,
648 J. A.: Highly functionalized organic nitrates in the southeast United States:
649 Contribution to secondary organic aerosol and reactive nitrogen budgets, *P. Natl.*
650 *Acad. Sci. U. S. A.*, 113, 1516-1521, doi:10.1073/pnas.1508108113, 2016.

651 Li, W., Teng, X., Chen, X., Liu, L., Xu, L., Zhang, J., Wang, Y., Zhang, Y., and Shi, Z.:
652 Organic Coating Reduces Hygroscopic Growth of Phase-Separated Aerosol
653 Particles, *Environmental Science & Technology*, 55, 16339-16346,
654 10.1021/acs.est.1c05901, 2021a.

655 Li, X., Song, S., Zhou, W., Hao, J., Worsnop, D., and Jiang, J.: Interactions between
656 aerosol organic components and liquid water content during haze episodes in
657 Beijing, *Atmos. Chem. Phys.*, 19, 12163-12174, 10.5194/acp-19-12163-2019,
658 2019.

659 Li, X., Zhang, Y., Shi, L., Kawamura, K., Kunwar, B., Takami, A., Arakaki, T., and Lai,
660 S.: Aerosol Proteinaceous Matter in Coastal Okinawa, Japan: Influence of Long-
661 Range Transport and Photochemical Degradation, *Environmental Science &*
662 *Technology*, 56, 5256-5265, 10.1021/acs.est.1c08658, 2022.

663 Li, Y., Zhao, J., Wang, Y., Seinfeld, J. H., and Zhang, R.: Multigeneration Production

664 of Secondary Organic Aerosol from Toluene Photooxidation, *Environ. Sci.*
665 *Technol.*, 55, 8592-8603, 10.1021/acs.est.1c02026, 2021b.

666 Lin, P., Liu, J., Shilling, J. E., Kathmann, S. M., Laskin, J., and Laskin, A.: Molecular
667 characterization of brown carbon (BrC) chromophores in secondary organic
668 aerosol generated from photo-oxidation of toluene, *Phys. Chem. Chem. Phys.*,
669 17, 23312-23325, 10.1039/C5CP02563J, 2015.

670 Lv, S., Wang, F., Wu, C., Chen, Y., Liu, S., Zhang, S., Li, D., Du, W., Zhang, F., Wang,
671 H., Huang, C., Fu, Q., Duan, Y., and Wang, G.: Gas-to-Aerosol Phase
672 Partitioning of Atmospheric Water-Soluble Organic Compounds at a Rural Site
673 in China: An Enhancing Effect of NH₃ on SOA Formation, *Environ. Sci.*
674 *Technol.*, 56, 3915-3924, 10.1021/acs.est.1c06855, 2022.

675 Ma, W., Zheng, F., Zhang, Y., Chen, X., Zhan, J., Hua, C., Song, B., Wang, Z., Xie, J.,
676 Yan, C., Kulmala, M., and Liu, Y.: Weakened Gas-to-Particle Partitioning of
677 Oxygenated Organic Molecules in Liquified Aerosol Particles, *Environmental*
678 *Science & Technology Letters*, 10.1021/acs.estlett.2c00556, 2022.

679 Mozurkewich, M.: The dissociation constant of ammonium nitrate and its dependence
680 on temperature, relative humidity and particle size, *Atmos. Environ.*, 27, 261-
681 270, [https://doi.org/10.1016/0960-1686\(93\)90356-4](https://doi.org/10.1016/0960-1686(93)90356-4), 1993.

682 Müller, A., Miyazaki, Y., Tachibana, E., Kawamura, K., and Hiura, T.: Evidence of a
683 reduction in cloud condensation nuclei activity of water-soluble aerosols caused
684 by biogenic emissions in a cool-temperate forest, *Sci. Rep.*, 7, 8452. DOI:
685 8410.1038/s41598-41017-08112-41599., 2017.

686 Nguyen, T. K. V., Capps, S. L., and Carlton, A. G.: Decreasing Aerosol Water Is
687 Consistent with OC Trends in the Southeast U.S, *Environ. Sci. Technol.*, 49,
688 7843-7850. <https://doi.org/7810.1021/acs.est.7845b00828>, 2015.

689 Nguyen, T. K. V., Zhang, Q., Jimenez, J. L., Pike, M., and Carlton, A. G.: Liquid water:
690 ubiquitous contributor to aerosol mass, *Environ. Sci. Tech. Lett.*, 3, 257-263.
691 <https://doi.org/210.1021/acs.estlett.1026b00167>, 2016.

692 Qiao, W., Guo, H., He, C., Shi, Q., Xiu, W., and Zhao, B.: Molecular Evidence of
693 Arsenic Mobility Linked to Biodegradable Organic Matter, *Environ. Sci.*
694 *Technol.*, 54, 7280-7290. <https://doi.org/7210.1021/acs.est.7280c00737>, 2020.

695 Qiu, Y., Xie, Q., Wang, J., Xu, W., Li, L., Wang, Q., Zhao, J., Chen, Y., Chen, Y., Wu,
696 Y., Du, W., Zhou, W., Lee, J., Zhao, C., Ge, X., Fu, P., Wang, Z., Worsnop, D.
697 R., and Sun, Y.: Vertical Characterization and Source Apportionment of Water-
698 Soluble Organic Aerosol with High-resolution Aerosol Mass Spectrometry in
699 Beijing, China, *ACS Earth Space Chem.*, 3, 273-284,
700 [10.1021/acsearthspacechem.8b00155](https://doi.org/10.1021/acsearthspacechem.8b00155), 2019.

701 Sareen, N., Schwier, A., Lathem, T., Nenes, A., and McNeill, V. F.: Surfactants from the
702 gas phase may promote cloud droplet formation, *P. Natl. Acad. Sci. U. S. A.*,
703 110, 2723-2728. <https://doi.org/2710.1073/pnas.1204838110>, 2013.

704 Sareen, N., Waxman, E. M., Turpin, B. J., Volkamer, R., and Carlton, A. G.: Potential
705 of aerosol liquid water to facilitate organic aerosol formation: assessing
706 knowledge gaps about precursors and partitioning, *Environ. Sci. Technol.*, 51,
707 3327-3335, 2017.

708 Schmidt, F., Koch, B. P., Goldhammer, T., Elvert, M., Witt, M., Lin, Y.-S., Wendt, J.,
709 Zabel, M., Heuer, V. B., and Hinrichs, K.-U.: Unraveling signatures of
710 biogeochemical processes and the depositional setting in the molecular
711 composition of pore water DOM across different marine environments,
712 *Geochim. Cosmochim. Ac.*, 207, 57-80.
713 <https://doi.org/10.1016/j.gca.2017.1003.1005>, 2017.

714 Simon, H., Bhave, P., Swall, J., Frank, N., and Malm, W.: Determining the spatial and
715 seasonal variability in OM/OC ratios across the US using multiple regression,
716 *Atmos. Chem. Phys.*, 11, 2933-2949. [https://doi.org/2910.5194/acp-2911-2933-](https://doi.org/2910.5194/acp-2911-2933-2011)
717 2011, 2011.

718 Song, J., Li, M., Jiang, B., Wei, S., Fan, X., and Peng, P. a.: Molecular Characterization
719 of Water-Soluble Humic like Substances in Smoke Particles Emitted from
720 Combustion of Biomass Materials and Coal Using Ultrahigh-Resolution
721 Electrospray Ionization Fourier Transform Ion Cyclotron Resonance Mass
722 Spectrometry, *Environ. Sci. Technol.*, 52, 2575-2585, [10.1021/acs.est.7b06126](https://doi.org/10.1021/acs.est.7b06126),
723 2018.

724 Su, S., Xie, Q., Lang, Y., Cao, D., Xu, Y., Chen, J., Chen, S., Hu, W., Qi, Y., Pan, X.,
725 Sun, Y., Wang, Z., Liu, C.-Q., Jiang, G., and Fu, P.: High Molecular Diversity
726 of Organic Nitrogen in Urban Snow in North China, *Environ. Sci. Technol.*,
727 [10.1021/acs.est.0c06851](https://doi.org/10.1021/acs.est.0c06851), 2021.

728 Sun, Y., Du, W., Fu, P., Wang, Q., Li, J., ge, X., Zhang, q., Zhu, C., Ren, L., Xu, W.,
729 Zhao, J., Han, T., Worsnop, D., and Wang, Z.: Primary and secondary aerosols

730 in Beijing in winter: Sources, variations and processes, *Atmos. Chem. Phys.*, 16,
731 8309-8329, 10.5194/acp-16-8309-2016, 2016.

732 Tan, H., Cai, M., Fan, Q., Liu, L., Li, F., Chan, P. W., Deng, X., and Wu, D.: An analysis
733 of aerosol liquid water content and related impact factors in Pearl River Delta,
734 *Sci. Total Environ.*, 579, 1822-1830.
735 <https://doi.org/10.1016/j.scitotenv.2016.1811.1167>, 2017.

736 Tong, H., Kourtchev, I., Pant, P., Keyte, I., O'Connor, I. P., Wenger, J., Pope, F., Harrison,
737 R., and Kalberer, M.: Molecular composition of organic aerosols at urban
738 background and road tunnel sites using ultra-high resolution mass spectrometry,
739 *Faraday Discuss.*, 189, 51-68, 10.17863/CAM.5910, 2016.

740 Ushijima, S. B., Huynh, E., Davis, R. D., and Tolbert, M. A.: Seeded Crystal Growth of
741 Internally Mixed Organic–Inorganic Aerosols: Impact of Organic Phase State,
742 *The Journal of Physical Chemistry A*, 125, 8668-8679,
743 10.1021/acs.jpca.1c04471, 2021.

744 Wang, J., Gui, H., Yang, Z., Yu, T., Zhang, X., and Liu, J.: Real-world gaseous emission
745 characteristics of natural gas heavy-duty sanitation trucks, *J. Environ. Sci.*, 115,
746 319-329, <https://doi.org/10.1016/j.jes.2021.06.023>, 2022.

747 Wang, J., Ye, J., Zhang, Q., Zhao, J., Wu, Y., Li, J., Liu, D., Li, W., Zhang, Y., Wu, C.,
748 Xie, C., Qin, Y., Lei, Y., Huang, X., Guo, J., Liu, P., Fu, P., Li, Y., Lee, H. C.,
749 Choi, H., Zhang, J., Liao, H., Chen, M., Sun, Y., Ge, X., Martin, S. T., and Jacob,
750 D. J.: Aqueous production of secondary organic aerosol from fossil-fuel
751 emissions in winter Beijing haze, *Proceedings of the National Academy of*

752 Sciences, 118, e2022179118, 10.1073/pnas.2022179118, 2021.

753 Xie, Q., Sihui, S., Chen, S., Xu, Y., Cao, D., Chen, J., Ren, L., Yue, S., Zhao, W., Sun,
754 Y., Wang, Z., Tong, H., Su, H., Cheng, Y., Kawamura, K., Jiang, G., Liu, C.-Q.,
755 and Fu, P.: Molecular characterization of firework-related urban aerosols using
756 Fourier transform ion cyclotron resonance mass spectrometry, *Atmos. Chem.*
757 *Phys.*, 20, 6803-6820, 10.5194/acp-20-6803-2020, 2020.

758 Xu, Y., Wu, D. S., Xiao, H. Y., and Zhou, J. X.: Dissolved hydrolyzed amino acids in
759 precipitation in suburban Guiyang, southwestern China: Seasonal variations and
760 potential atmospheric processes, *Atmos. Environ.*, 211, 247-255.
761 <https://doi.org/210.1016/j.atmosenv.2019.1005.1011>, 2019.

762 Xu, Y., Xiao, H., Wu, D., and Long, C.: Abiotic and Biological Degradation of
763 Atmospheric Proteinaceous Matter Can Contribute Significantly to Dissolved
764 Amino Acids in Wet Deposition, *Environ. Sci. Technol.*, 54, 6551-6561.
765 <https://doi.org/6510.1021/acs.est.6550c00421>, 2020a.

766 Xu, Y., Dong, X.-N., Xiao, H.-Y., Zhou, J.-X., and Wu, D.-S.: Proteinaceous Matter and
767 Liquid Water in Fine Aerosols in Nanchang, Eastern China: Seasonal Variations,
768 Sources, and Potential Connections, *J. Geophys. Res.: Atmos.*, 127,
769 e2022JD036589. <https://doi.org/036510.031029/032022JD036589>, 2022.

770 Xu, Y., Miyazaki, Y., Tachibana, E., Sato, K., Ramasamy, S., Mochizuki, T., Sadanaga,
771 Y., Nakashima, Y., Sakamoto, Y., Matsuda, K., and Kajii, Y.: Aerosol Liquid
772 Water Promotes the Formation of Water-Soluble Organic Nitrogen in
773 Submicrometer Aerosols in a Suburban Forest, *Environ. Sci. Technol.*, 54,

774 1406-1414. <https://doi.org/10.1021/acs.est.1409b05849>, 2020b.

775 Yang, D., Zhu, S., Ma, Y., Zhou, L., Zheng, F., Wang, L., Jiang, J., and Zheng, J.:
776 Emissions of Ammonia and Other Nitrogen-Containing Volatile Organic
777 Compounds from Motor Vehicles under Low-Speed Driving Conditions,
778 *Environ. Sci. Technol.*, 56, 5440-5447, [10.1021/acs.est.2c00555](https://doi.org/10.1021/acs.est.2c00555), 2022.

779 Yeh, G. K. and Ziemann, P. J.: Alkyl Nitrate Formation from the Reactions of C8–C14
780 n-Alkanes with OH Radicals in the Presence of NO_x: Measured Yields with
781 Essential Corrections for Gas–Wall Partitioning, *J. Phys. Chem. A*, 118, 8147-
782 8157, [10.1021/jp500631v](https://doi.org/10.1021/jp500631v), 2014.

783 Yttri, K. E., Aas, W., Bjerke, A., Cape, J., Cavalli, F., Ceburnis, D., Dye, C., Emblico,
784 L., Facchini, M., and Forster, C.: Elemental and organic carbon in PM₁₀: a one
785 year measurement campaign within the European Monitoring and Evaluation
786 Programme EMEP, *Atmos. Chem. Phys.*, 7, 5711-5725.
787 <https://doi.org/10.5194/acp-5717-5711-2007>, 2007.

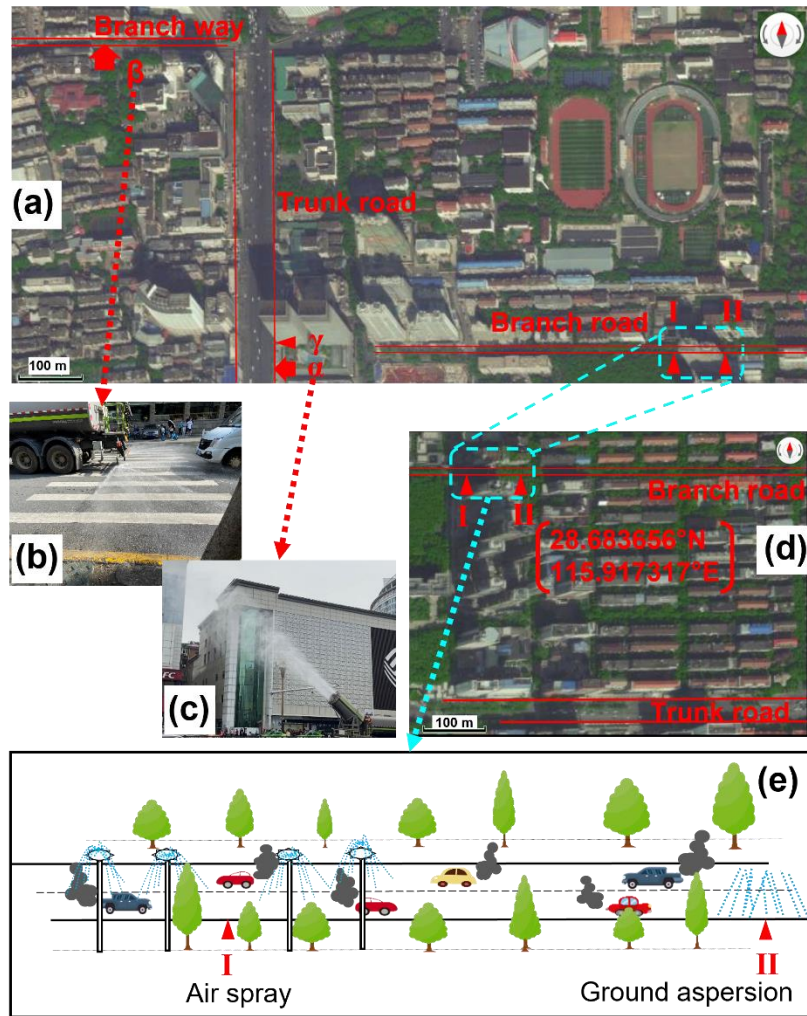
788 Yu, H., Li, W., Zhang, Y., Tunved, P., Dall'Osto, M., Shen, X., Sun, J., Zhang, X., Zhang,
789 J., and Shi, Z.: Organic coating on sulfate and soot particles during late summer
790 in the Svalbard Archipelago, *Atmos. Chem. Phys.*, 19, 10433-10446,
791 [10.5194/acp-19-10433-2019](https://doi.org/10.5194/acp-19-10433-2019), 2019.

792 Yue, H., He, C., Huang, Q., Yin, D., and Bryan, B. A.: Stronger policy required to
793 substantially reduce deaths from PM_{2.5} pollution in China, *Nat. Commun.*, 11,
794 1462, [10.1038/s41467-020-15319-4](https://doi.org/10.1038/s41467-020-15319-4), 2020.

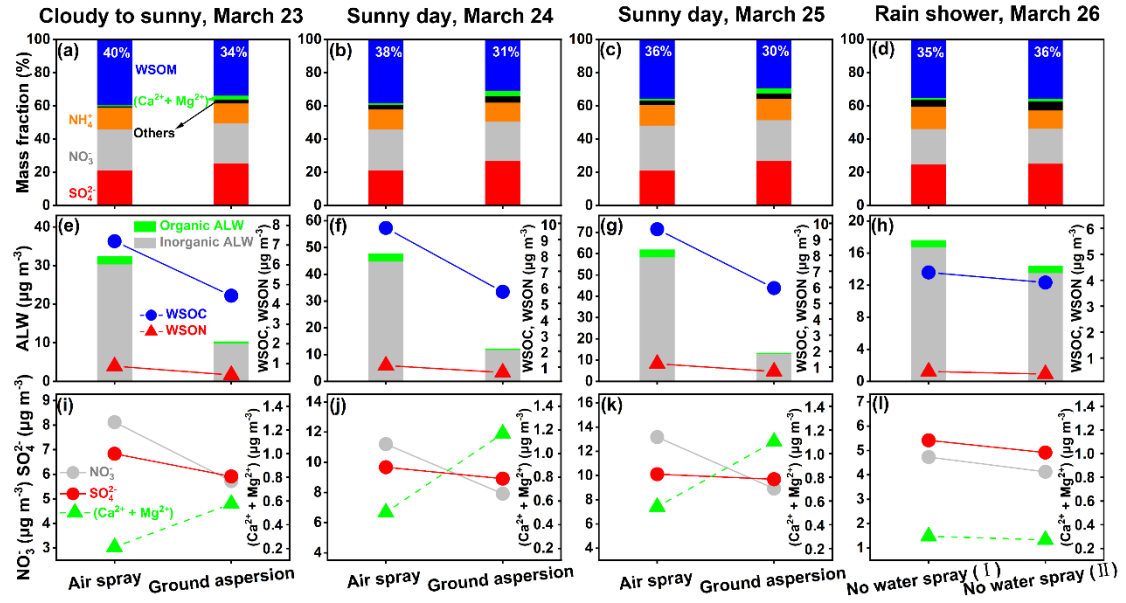
795

Table 1. The number of compounds in different subgroups in different samples and the number fractions of common molecules in the same subgroup in different samples.

Sample type (Date: Mar. 23–26)	Total	CHO	Common (CHO)	Common fraction (CHO)	CHON	Common (CHON)	Common fraction (CHON)
Air spray (23)	7069	1766		63%	2375		55%
Ground aspersion (23)	6166	1233	1104	90%	1803	1308	73%
Air spray (24)	7317	1861		53%	2447		50%
Ground aspersion (24)	6067	1098	990	90%	1501	1225	82%
Air spray (25)	8102	2037		39%	2685		45%
Ground aspersion (25)	5966	845	804	95%	1464	1209	83%
No water spray (I) (26)	6998	1621		81%	2120		74%
No water spray (II) (26)	6990	1539	1315	85%	1963	1579	80%



1
2 **Figure 1.** Map (Baidu, China) showing (a) the study area. The symbol of “ γ ” indicates
3 the location where $PM_{2.5}$ mass concentration was monitored before and after the mist
4 cannon truck passed by. The symbols of “ α ” and “ β ” refer to the locations where (b) the
5 mist cannon truck photograph and (c) the traditional sprinkling truck photograph were
6 taken, respectively. The symbol of “I” indicates that (d) the sampling was conducted on
7 the air spray road segment or no water spray road segment (I). The symbol of “II”
8 indicates that (d) the sampling was conducted on the ground aspersion road segment or
9 no water spray road segment (II). Figure (e) shows the conceptual diagram of the
10 sampling campaign.
11

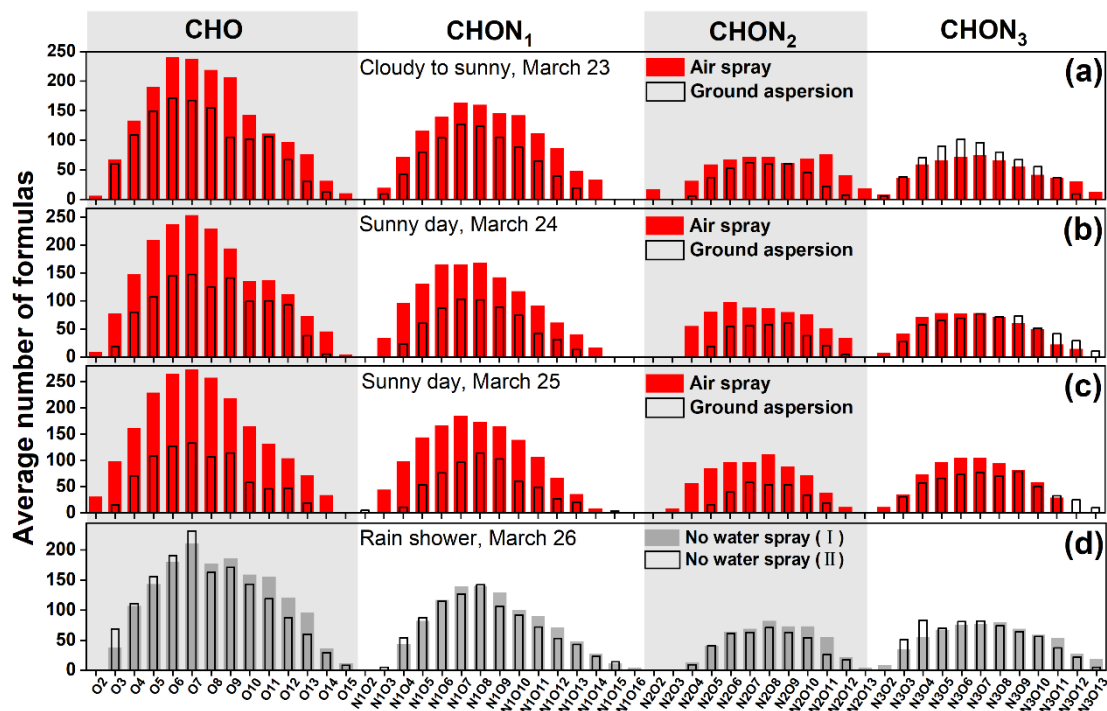


12

13 **Figure 2.** The mass fractions of the chemical components in PM_{2.5}: (a, b, c) air spray
 14 vs ground aspersion and (d) no water spray (I) vs no water spray (II). The mass
 15 concentrations of nitrate and ammonium, as well as the sum of calcium and magnesium
 16 concentrations in PM_{2.5}: (e, f, g) air spray vs ground aspersion and (h) no water spray
 17 (I) vs no water spray (II). The mass concentrations of ALW, WSOC, and WSOM in
 18 PM_{2.5}: (i, j, k) air spray vs ground aspersion and (l) no water spray (I) vs no water spray
 19 (II).

20

21



22

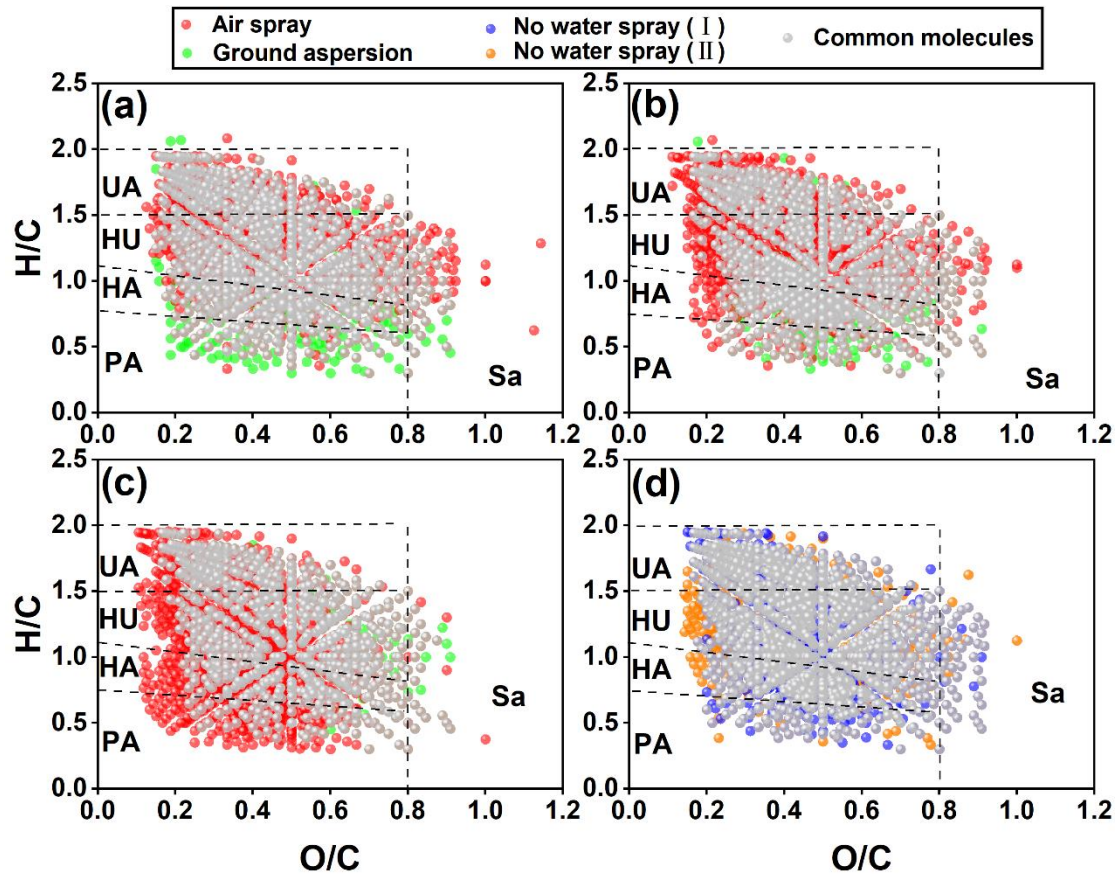
23 **Figure 3.** Classification of CHO and CHON species into subgroups according to the

24 number of O atoms in their molecules in WSOM in PM_{2.5} collected from different cases:

25 (a, b, c) air spray vs ground aspersion and (d) no water spray (I) vs no water spray (II).

26

27

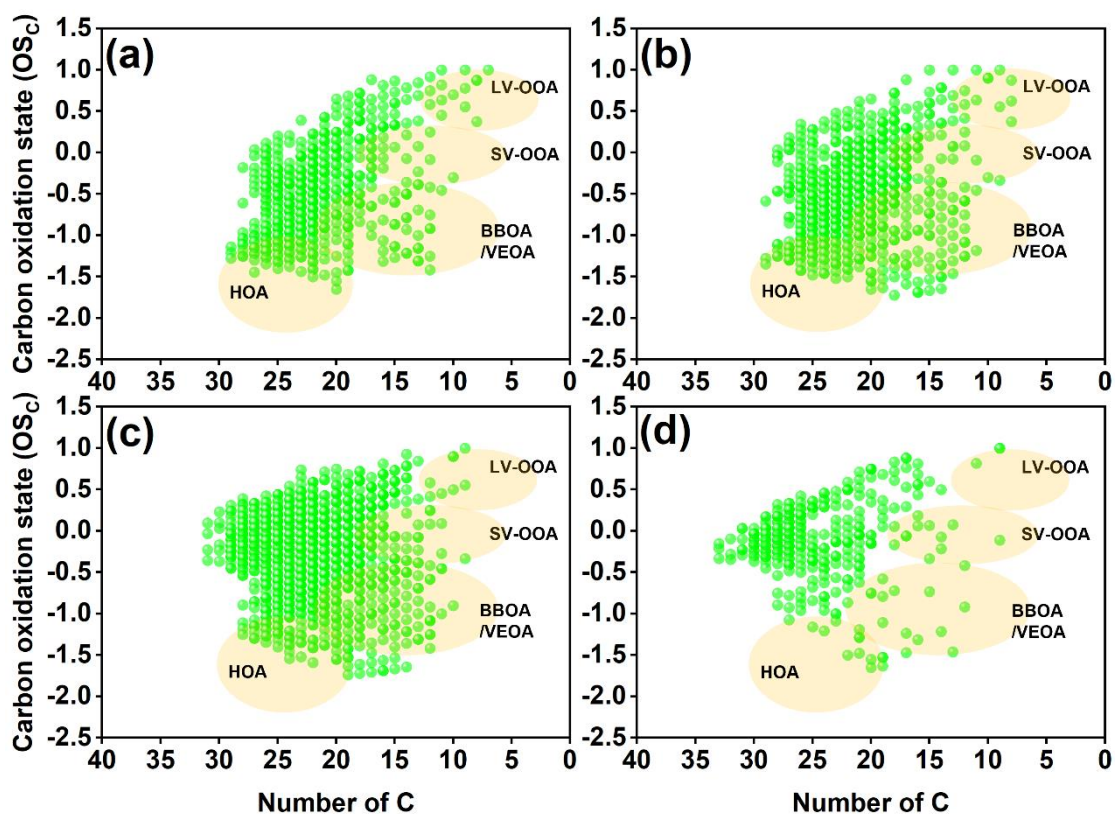


29

30 **Figure 4.** Van Krevelen diagrams of CHO compounds in WSOM in PM_{2.5} collected
 31 from different cases: air spray vs ground aspersion on (a) March 23, (b) March 24, and
 32 (c) March 25 and two road segments without water spray (I vs II) on (d) March 26. The
 33 classifications of compounds include **unsaturated aliphatic-like (UA)**, **highly**
 34 **unsaturated-like (HU)**, **highly aromatic-like (HA)**, **polycyclic aromatic-like (PA)**, and
 35 **saturated-like (Sa)** molecules.

36

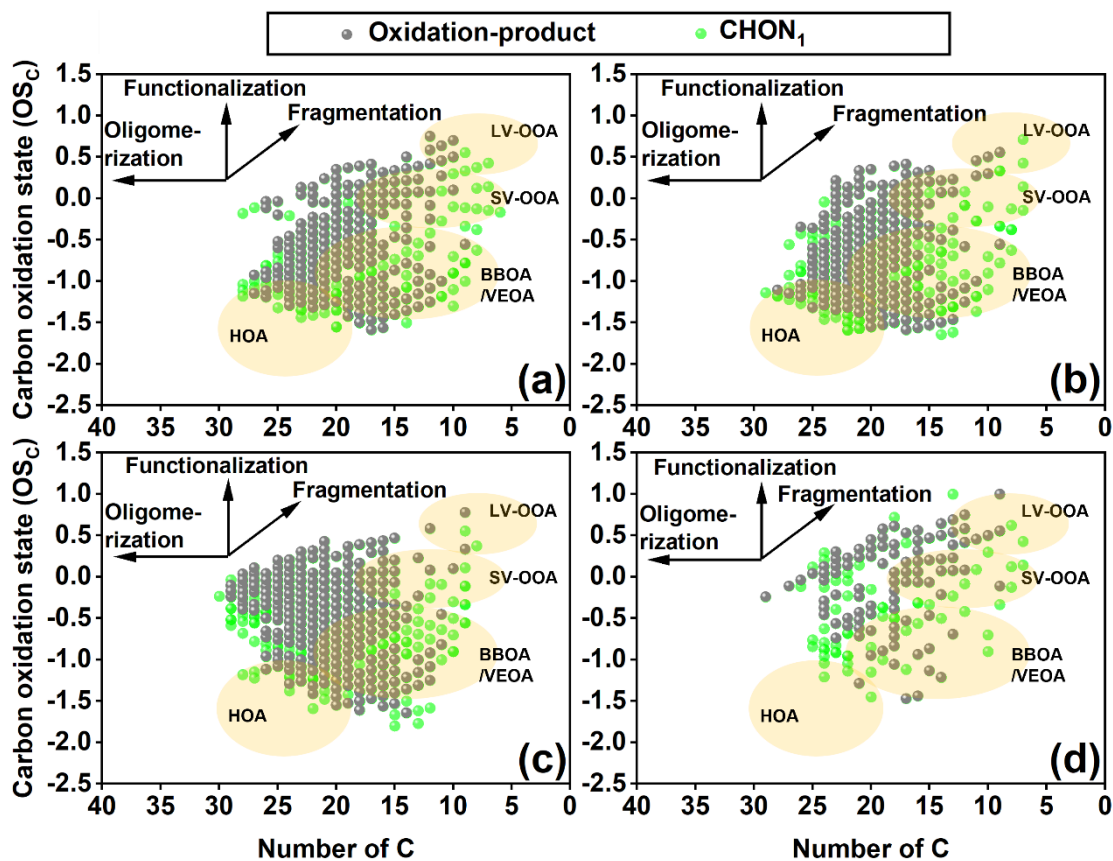
37



38

39 **Figure 5.** OS_C of unique CHO molecules in WSOM in $PM_{2.5}$ collected from different
 40 cases: air spray vs ground aspersion on (a) March 23, (b) March 24, and (c) March 25
 41 and two road segments without water spray (I vs II) on (d) March 26. For the above
 42 comparative cases, the unique CHO compounds indicate the CHO molecules identified
 43 in $PM_{2.5}$ collected from the air spray (/no water spray-I) road segments. The light orange
 44 background represents areas of HOA (hydrocarbon-like organic aerosol), BBOA and
 45 VEOA (biomass burning and vehicle emission organic aerosols) (Kroll et al., 2011;
 46 Tong et al., 2016), SV-OOA (semivolatile oxidized organic aerosol), and LV-OOA (low-
 47 volatility oxidized organic aerosol) (Kroll et al., 2011).

48

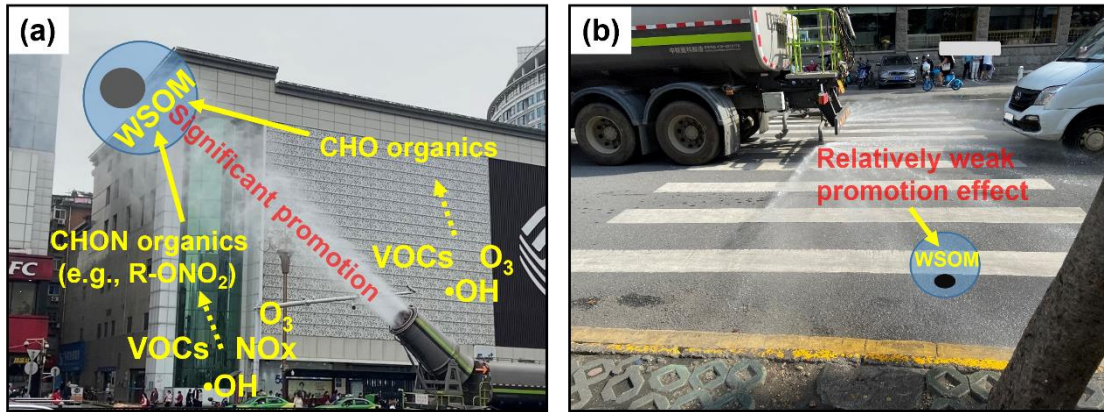


49

50 **Figure 6.** OS_C of unique CHON₁ molecules in WSOM in PM_{2.5} collected from different
 51 cases: air spray vs ground aspersion on (a) March 23, (b) March 24, and (c) March 25
 52 and two road segments without water spray (I vs II) on (d) March 26. For the above
 53 comparative cases, the unique CHON₁ compounds indicate the CHON₁ molecules
 54 identified in PM_{2.5} collected from the air spray (/no water spray-I) road segments. The
 55 light orange background represents areas of HOA, BBOA and VEOA, SV-OOA, and
 56 LV-OOA. The grey circles refer to the identified oxidation-product pairs.

57

58



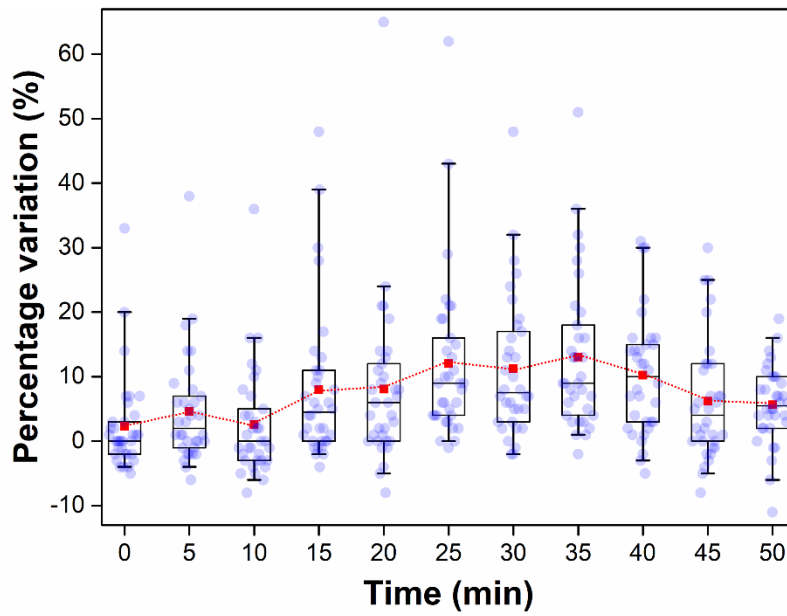
59

60 **Figure 7.** Conceptual picture showing the influence of (a) mist cannon truck and (b)
 61 traditional sprinkling truck on water-soluble organic matter (WSOM) formation in the
 62 urban road environment.

63

64

65



66

67 **Figure 8.** The time series of percentage variations in $PM_{2.5}$ mass concentrations after
68 mist cannon truck operation ($n = 34$). Each box encompasses the 25th–75th percentiles.

69 Whiskers are the 5th and 95th percentiles. The solid lines and squares inside boxes
70 indicate the median and mean. All individual data are also presented as circles.

71