Supplement of

The Positive Effect of Formaldehyde on the Photocatalytic Renoxification of Nitrate on TiO₂ Particles

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The supporting information has 8 pages, 1 table, and 11 figure.

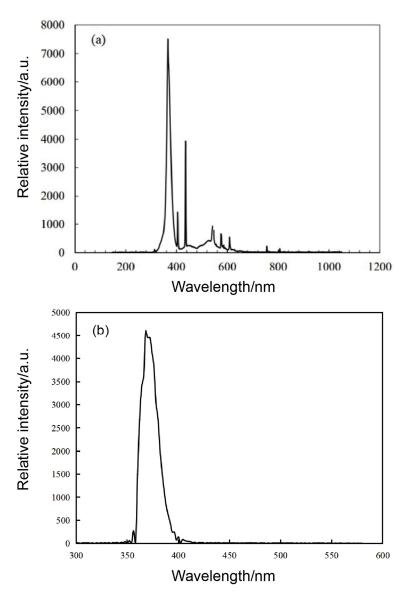


Figure S1. Spectral energy distribution of (a) 365 nm tube lamps and (b) 365 nm LED lamps.

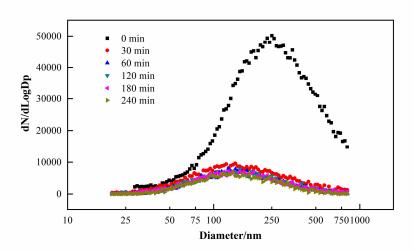


Figure S2. Changes of particle size distribution of TiO₂ particles in environmental chamber with

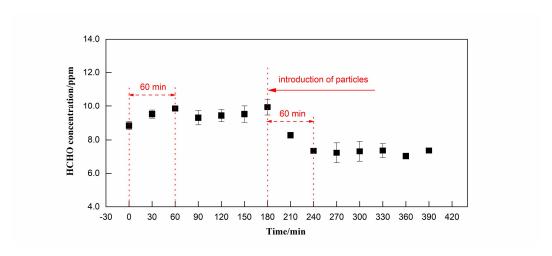


Figure S3. Changes of HCHO concentration in the environmental chamber before and after the introduction of particles over time.

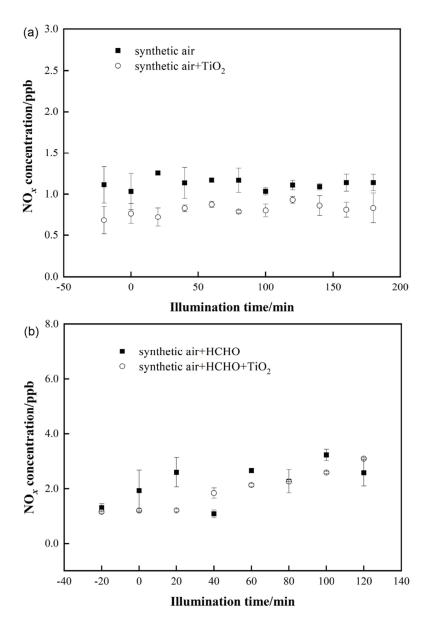


Figure S4. Changes of NO_x concentration in environmental chamber in (a) "synthetic air" and "synthetic air + TiO_2 " system, (b) "synthetic air + HCHO" and "synthetic air + HCHO + TiO_2 " system. 365 nm tube lamps were used during the blank experiment.

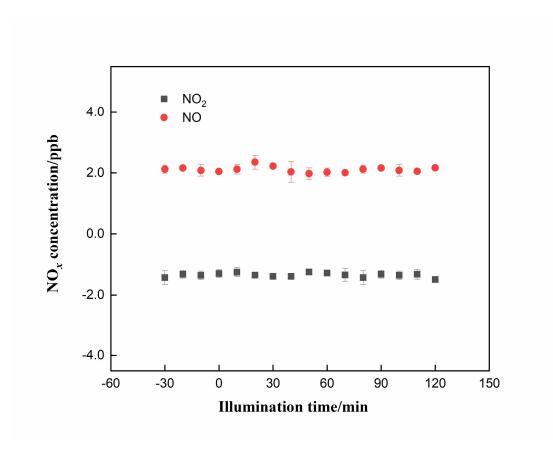


Figure S5. Effect of illumination on the release of NO and NO₂ over 4 wt.% KNO₃-TiO₂ at 293 K and 0.8% of relative humidity. 365 nm LED lamps were used during the illumination experiment.

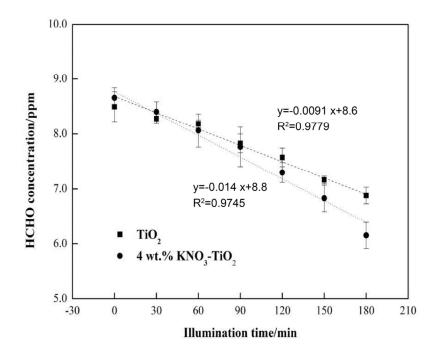


Figure S6. Photodegradation curve of HCHO on TiO₂ and 4 wt.% KNO₃-TiO₂ particles under 365 nm LED lamps at 293 K and 0.8% of relative humidity.

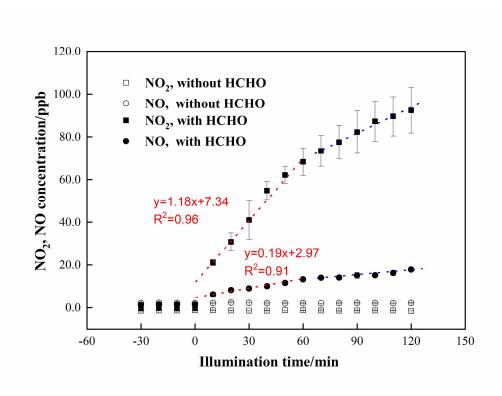


Figure S7. Effect of HCHO on the production of NO and NO₂ over 4 wt.% KNO₃-TiO₂ particles at 293 K and 0.8% of relative humidity. 365 nm LED lamps were used during the illumination experiment. The initial concentration of HCHO was about 9 ppm.

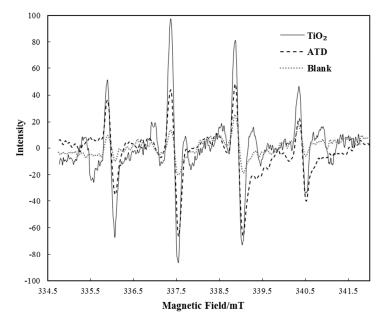


Figure S8. ESR spectra of irradiated TiO₂ and ATD particles.

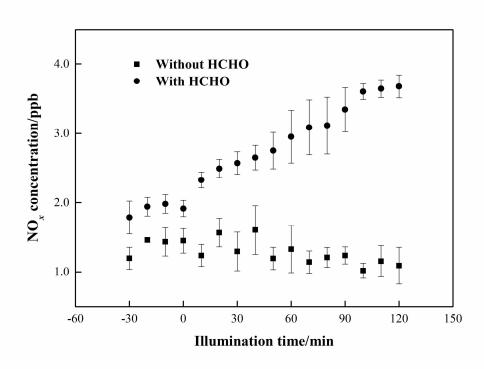


Figure S9. Effect of HCHO on the renoxification processes of Arizona Text Dust (ATD) at 293 K and 0.8% of relative humidity. 365 nm LED lamps were used during the illumination experiment. The initial concentration of HCHO was about 9 ppm.

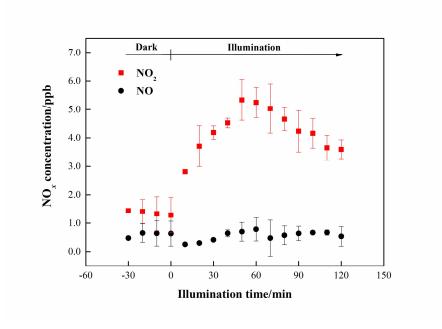


Figure S10. The release of NO₂ and NO with 365 nm LED lamps illumination over 4 wt.% KNO₃-TiO₂ particles at 293 K and 0.8% RH. The initial concentration of HCHO was about 1.0 ppm.

Table S1. Arizona Test Dust (ATD) chemical composition.

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Chemical composition	Weight percentage (%)
SiO ₂	68-76
Al_2O_3	10-15
Fe_2O_3	2.0-5.0
Na_2O	2.0-4.0
CaO	2.0-5.0
MgO	1.0-2.0
TiO_2	0.5-1.0
K_2O	2.0-5.0

^{*}Loss on Ignition 2-5 %