

1 *Supplement of*

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3 Characteristics and degradation of organic aerosols from cooking
4 sources based on hourly observation of organic molecular markers in
5 urban environment

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22 individual factors to PM_{2.5} (b) and to OC (c)

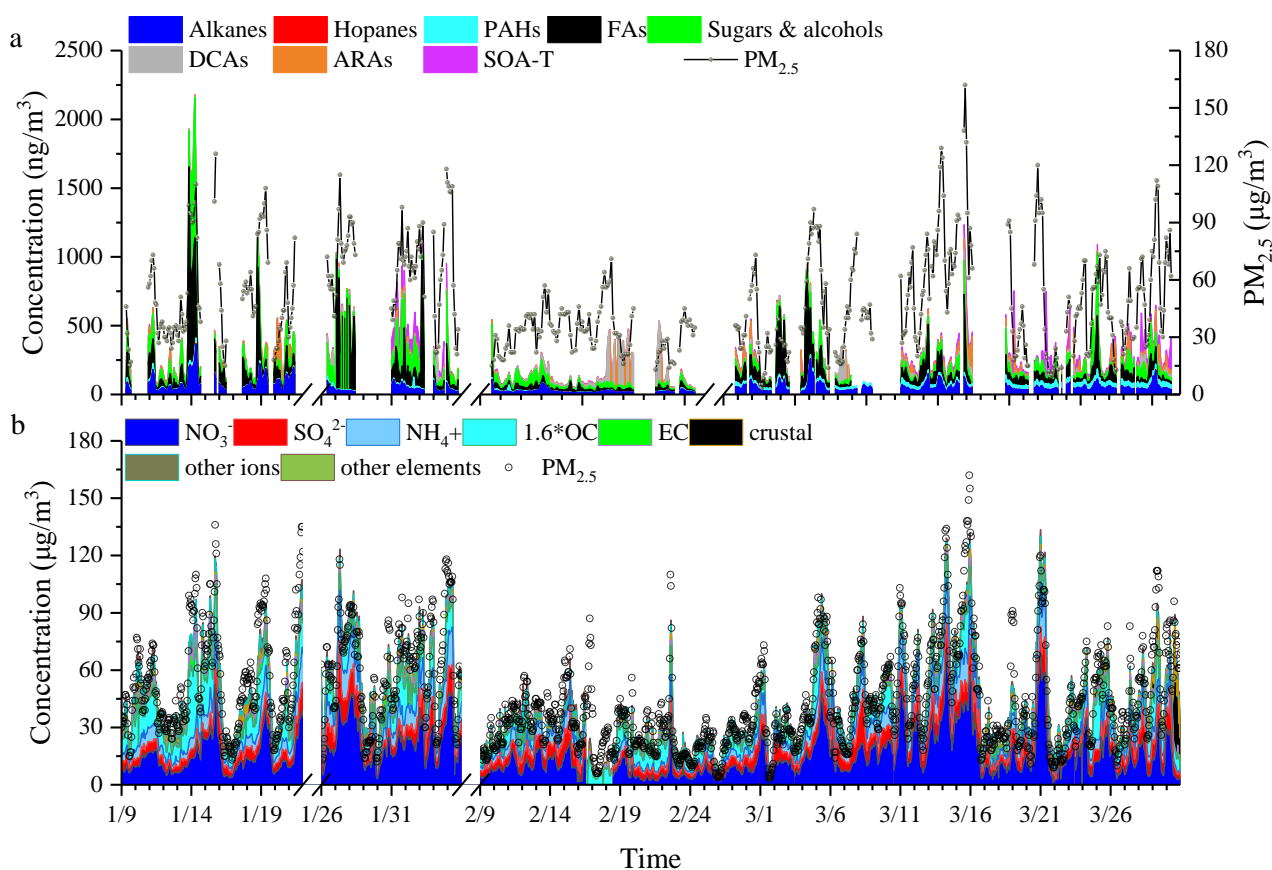
23 Table S1. PM_{2.5} and its chemical components included in the PMF analysis

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25 **Section S1. Observation instruments and factors**

26 Hourly data for meteorological parameters, gaseous pollutants (NO₂ and O₃), PM_{2.5} and its major chemical
27 composition during the campaign were measured online. The meteorological parameters were obtained from a
28 meteorological monitor (WXT520, VAISALA Inc., FL), which uses ultrasound to measure wind speed and direction, and
29 a PTU module to measure atmospheric pressure, temperature and humidity using capacitive measurements; O₃ and NO₂
30 were measured by a ozone analyzer (49i-PS, Thermo Fisher Scientific, US) and NO_x (MODEL450i, Thermo Fisher
31 Scientific, US) analyzer respectively; PM_{2.5} mass concentration was measured by an online particulate matter monitor
32 (BAM1020, Met One Inc., US) using the β-ray method; the concentration of the carbonaceous component of PM_{2.5} was
33 measured using a semi-continuous OC/EC analyzer (RT-4, Sunset Laboratory Inc, US) (Nicolosi et al., 2018; Zhang et al.,
34 2017); ions and elements were measured by a MARGA ionic online analyzer (ADI2080, Metrohm, CHN) (Makkonen et
35 al., 2012) and a atmospheric elements online monitor (EHM-X200, Tianrui, CHN).

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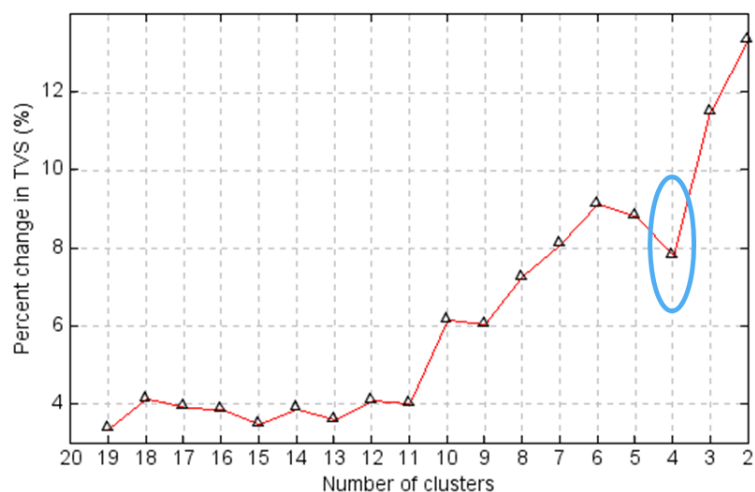


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38 **Figure S1. Time series of PM_{2.5}, chemical components and organic molecular tracers (PAHs, polycyclic aromatic**
39 **hydrocarbons; DCAs, dicarboxylic acids; FAs, fatty acids; ARAs, aromatic acids; SOA-T, secondary organic aerosol tracers;**
40 **crustal=2.20×[Al]+2.49×[Si]+1.63×[Ca]+2.42×[Fe]+1.94×[Ti] (Huang et al., 2014).)**

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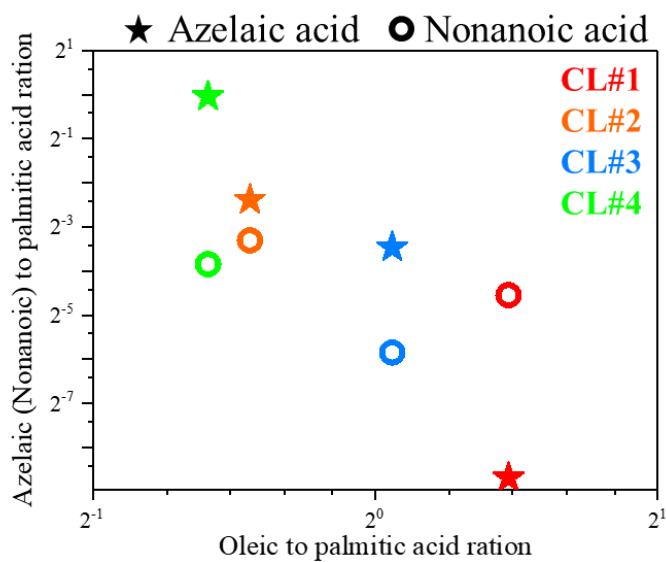
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Figure S2. The variation of the total spatial variance (Choose the most appropriate clustering case based on the change in total spatial variance (TSV). The point raised down by TSV were selected as the optimal number of clusters, and the optimal solution of four clusters was finally extracted in this study (He et al., 2020).)

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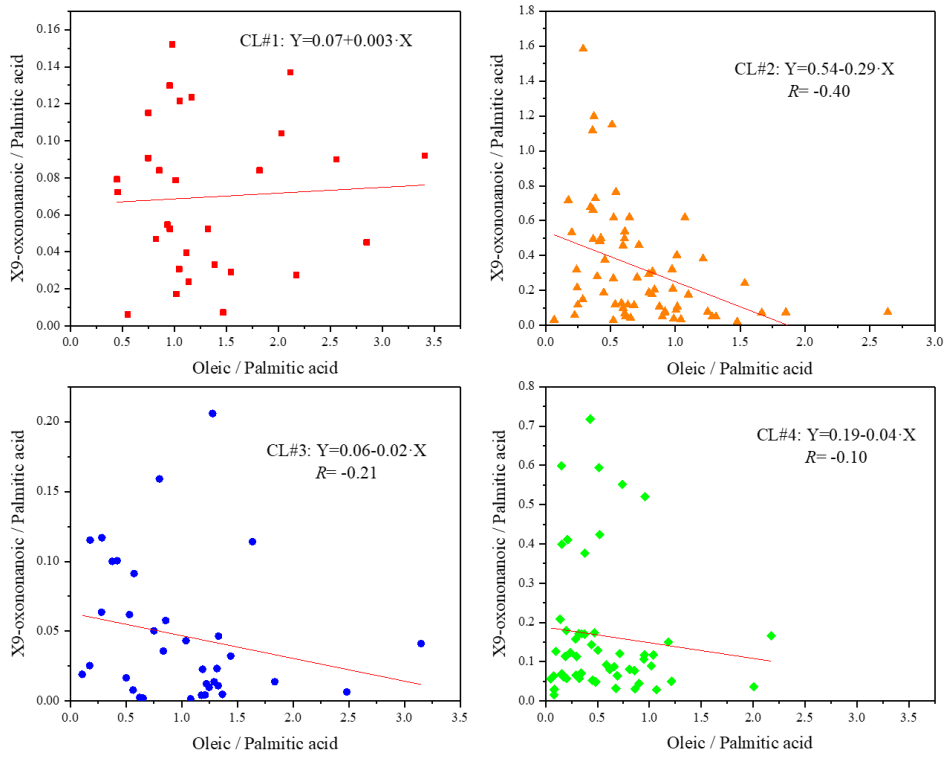


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Figure S3. Correlation of Azelaic (Nonanoic) acid/palmitic acid ratio and oleic acid/palmitic acid ratio for the ambient samples

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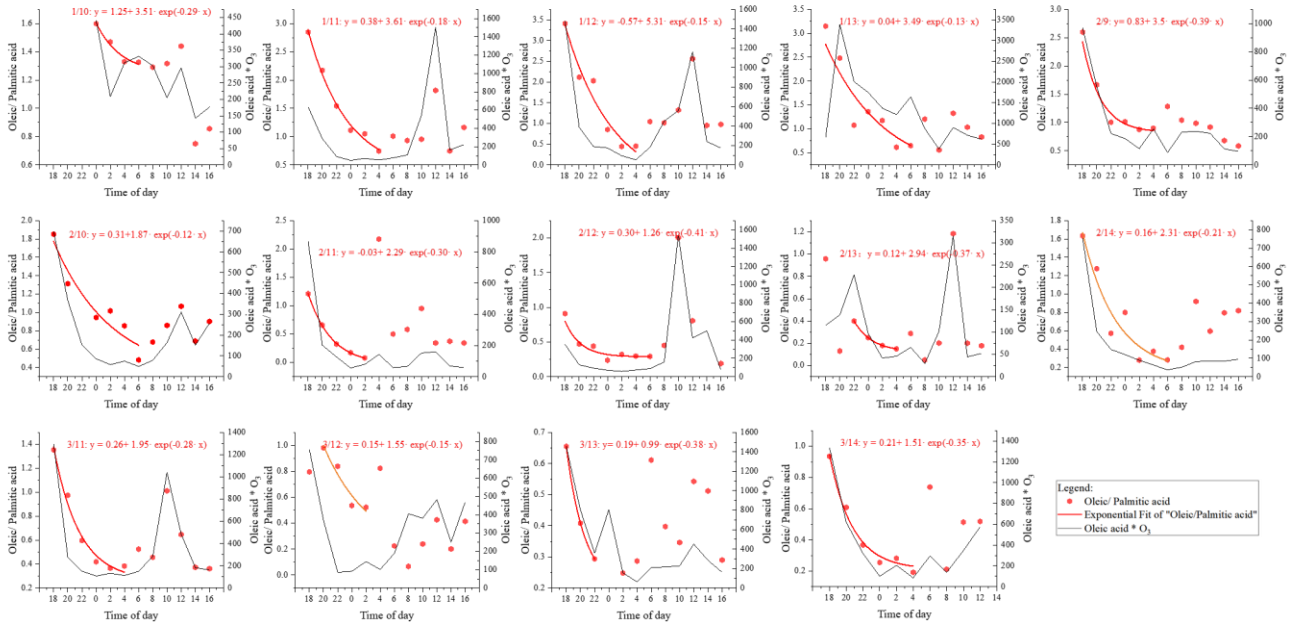
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Figure S4. Correlation of X9-oxononanoic acid with oleic acid normalized by palmitic acid for the ambient samples under different sources of air masses

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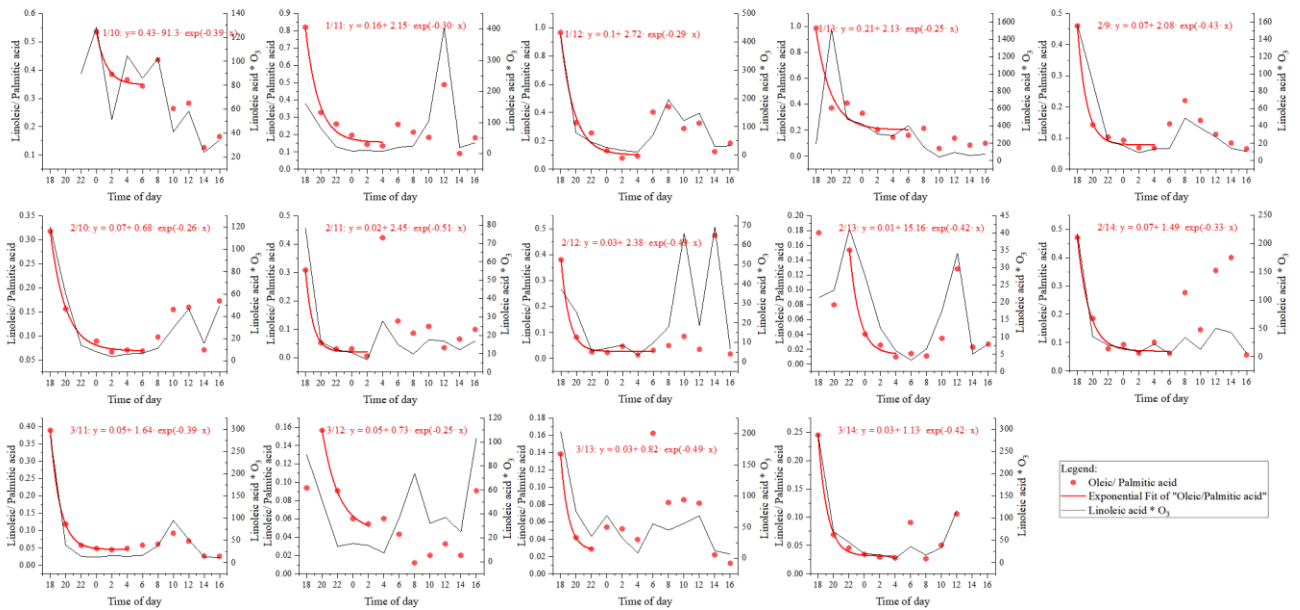
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Figure S5. Day-to-day fitting of oleic acid normalized by palmitic acid

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60 **Figure S6. Day-to-day fitting of linoleic acid normalized by palmitic acid**

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62 **Section S2. Source apportionment using PMF**

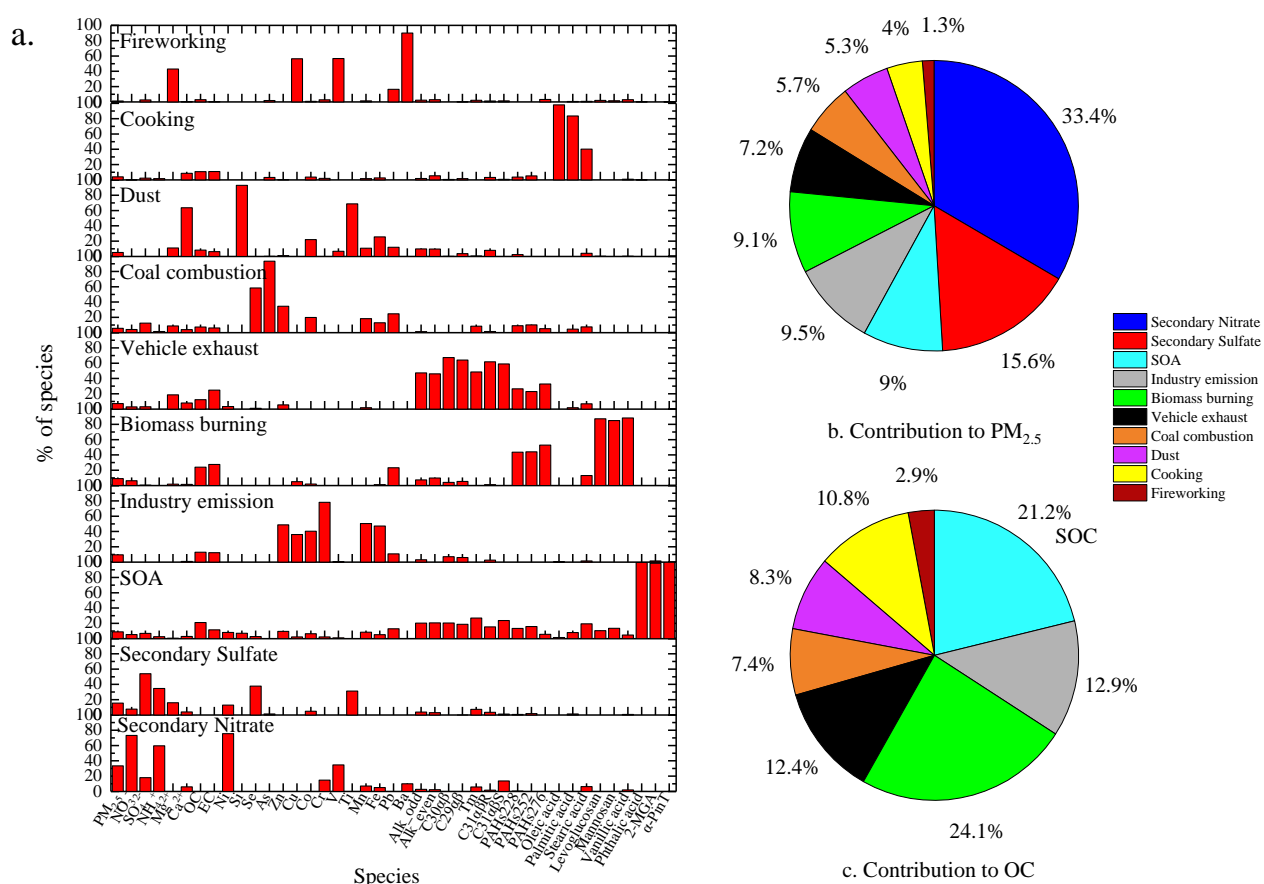
63 Table S1 lists the input PM_{2.5} and its components in the PMF modeling. The preferential input species for PMF are
 64 those with high abundance and source specific (Norris et al., 2014). Generally, organic markers with lower volatility and
 65 lower reactivity were selected. Figure S7(a) shows the PMF-resolved source profiles and time series of source
 66 contributions to OC for each source factor. Figure S7(b, c) shows the campaign-average percentage source contributions
 67 to OC and PM_{2.5} from the PMF result (The time period is the whole observation period using TAG from January to
 68 March 2021, that is, the period of Figure S1). Here, we only briefly present the identification of each source factor.

69 A total of 10 factors are identified. Among them, seven are primary sources, they are industrial emission, biomass
 70 burning, vehicle exhaust, coal combustion, dust, cooking and fireworking. Three secondary sources, namely, secondary
 71 nitrate, secondary sulfate and SOA factor (Li et al., 2020; Wang et al., 2017).

72 Secondary nitrate factor is identified by high contributions of nitrate and ammonium. The secondary sulfate factor is
 73 characterized by high loadings of sulfate and ammonium. The SOA factor is characterized by high loadings of an
 74 anthropogenic SOA tracer (phthalic acid), isoprene SOA tracer (2-methylglyceric acid) and α -pinene SOA tracers (3-
 75 hydroxyglutaric acid, pinic acid and cis-pinonic acid) (Wang et al., 2017). The profile of industrial emission contains
 76 high loadings of Cr, Zn, Fe and Mn (Men et al., 2019; Pant and Harrison, 2013). Industry activities related to steel
 77 production, plating, and metallurgy often emit a large amount of these metallic elements. Biomass burning is identified
 78 by high loadings of levoglucosan and mannosan (Feng et al., 2013; Wang et al., 2019). The sixth factor contains a high

79 abundance of n-alkanes and hopanes, and is identified to be vehicle exhaust ([Pant and Harrison, 2013](#); [Wang et al., 2017](#)).
 80 Coal combustion is identified by high loadings of Se, As and Pb ([Chen et al., 2013](#); [Wang et al., 2017](#)), and the dust
 81 factor is distinguished by crustal elements (ions) Ca, Si, and Ti. The cooking factor is distinguished by fatty acids (oleic
 82 acid, palmitic acid and stearic acid) ([Li et al., 2020](#)). The fireworking factor is identified by high loadings of flammable
 83 metals such as Mg^{2+} , Cu and Ba, etc.

84 During the whole observation period with TAG, although the cooking factor contributes only a small fraction of
 85 $PM_{2.5}$ (4%), it accounts for 10.8% of the total OC, indicating the importance of cooking emissions to OM in the urban
 86 metropolis.



87 **Figure S7. PMF resolved factor profiles (percentage of each species in each factor) (a); percentage contributions of individual**
 88 **factors to $PM_{2.5}$ (b) and to OC (c)**
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Table S1. PM_{2.5} and its chemical components included in the PMF analysis

naming	grouping	unit	concentration
Alk_odd	<i>n</i> -C ₂₅ , <i>n</i> -C ₂₇ , <i>n</i> -C ₂₉ , <i>n</i> -C ₃₁ and <i>n</i> -C ₃₃	ng/m ³	14.83
Alk_even	<i>n</i> -C ₂₄ , <i>n</i> -C ₂₆ , <i>n</i> -C ₂₈ , <i>n</i> -C ₃₀ and <i>n</i> -C ₃₂	ng/m ³	17.65
C30αβ	17α(H)21β(H)-hopane	ng/m ³	0.27
C29αβ	17α(H)21β(H)-30-norhopane	ng/m ³	0.21
Tm	17α(H)-22,29,30-trisnorhopane	ng/m ³	0.1
C31αβR	17α(H)21β(H)-(22R)-homohopane	ng/m ³	0.1
C31αβS	17α(H)21β(H)-(22S)-homohopane	ng/m ³	0.17
PAHs228	Benzo[a]anthracene, Chrysene	ng/m ³	0.96
PAHs252	Benzo[b+k]fluoranthene, Benzo[a]pyrene	ng/m ³	1.44
PAHs276	Benzo[g,h,i]perylene, Indeno[1,2,3-cd]pyrene	ng/m ³	1.67
Oleic acid	Oleic acid	ng/m ³	28.09
Palmitic acid	Palmitic acid	ng/m ³	48.86
Stearic acid	Stearic acid	ng/m ³	19.94
Levogluconan	Levogluconan	ng/m ³	54.12
Mannosan	Mannosan	ng/m ³	4.23
Vanillic acid	Vanillic acid	ng/m ³	1.27
Phthalic acid	Phthalic acid	ng/m ³	16.71
2-MGA	2-Methylglyceric acid	ng/m ³	1.78
α-PinT	3-Hydroxyglutaric acid, Pinic acid, Cis-pinonic acid	ng/m ³	14.95
NO ₃ ⁻		μg/m ³	17.46
SO ₄ ²⁻		μg/m ³	7.50
NH ₄ ⁺		μg/m ³	7.59
Other ions	Mg ²⁺ , Ca ²⁺	μg/m ³	0.46
OC		μg/m ³	5.98
EC		μg/m ³	1.87
Crustal elements	Si, Ti, Fe	μg/m ³	0.22
Other elements	Ni, Se, As, Zn, Cu, Co, Cr, V, Mn, Pb, Ba	μg/m ³	0.61
PM _{2.5}		μg/m ³	49.93
O ₃		μg/m ³	49.09
NO ₂		μg/m ³	45.27

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