

1 **Composited analyses of the chemical and physical characteristics of co-**
2 **polluted days by ozone and PM_{2.5} over 2013-2020 in the Beijing–Tianjin–Hebei**
3 **region**

4
5 Huibin Dai¹, Hong Liao^{1*}, Ke Li¹, Xu Yue¹, Yang Yang¹, Jia Zhu¹, Jianbing Jin¹,
6 Baojie Li¹, Xingwen Jiang²

7
8 ¹Jiangsu Key Laboratory of Atmospheric Environment Monitoring and Pollution
9 Control, Jiangsu Collaborative Innovation Center of Atmospheric Environment and
10 Equipment Technology, School of Environmental Science and Engineering, Nanjing
11 University of Information Science & Technology, Nanjing 210044, China

12 ²Institute of Plateau Meteorology, China Meteorological Administration, Chengdu,
13 Sichuan 610072, China

14
15 *Correspondence to: Hong Liao (hongliao@nuist.edu.cn)

17 **Abstract.**

18 The co-polluted days by ozone (O₃) and PM_{2.5} (particulate matter with an
19 aerodynamic equivalent diameter of 2.5 μm or less) (O₃&PM_{2.5}PD) were frequently
20 observed in the Beijing–Tianjin–Hebei (BTH) region in warm seasons (April–October)
21 of 2013–2020. We applied the 3-D global chemical transport model (GEOS-Chem) to
22 investigate the chemical and physical characteristics of O₃&PM_{2.5}PD by composited
23 analyses of such days that were captured by both the observations and the model. Model
24 results showed that, when O₃&PM_{2.5}PD occurred, the concentrations of hydroxyl
25 radical and total oxidant, sulfur oxidation ratio, and nitrogen oxidation ratio were all
26 high, and the concentrations of sulfate at the surface were the highest among all
27 pollution types. We also found unique features in vertical distributions of aerosols
28 during O₃&PM_{2.5}PD; concentrations of PM_{2.5} decreased with altitude near the surface
29 but remained stable at 975–819 hPa. Process analyses showed that secondary aerosols
30 (nitrate, ammonium and sulfate) had strong chemical productions at 913–819 hPa,
31 which were then transported downward, resulting in the quite uniform vertical profiles
32 at 975–819 hPa in O₃&PM_{2.5}PD. **The weather patterns for O₃&PM_{2.5}PD were**
33 **characterized by anomalous high-pressure system at 500 hPa as well as strong**
34 **southerlies and high RH at 850 hPa. The latter resulted in the strong chemical**
35 **productions around 850 hPa in O₃&PM_{2.5}PD.** The physical and chemical characteristics
36 of O₃&PM_{2.5}PD are quite different from those of polluted days by either O₃ alone or
37 PM_{2.5} alone, which have important implications for air quality management.

38

39 **Keywords:** Co-occurrence, Ozone and PM_{2.5}, Pollution, Meteorological parameters.

40

41 **1. Introduction**

42 Surface ozone (O_3) and $PM_{2.5}$ (particulate matter with an aerodynamic equivalent
43 diameter of 2.5 micrometers or less) are important air pollutants in the atmosphere that
44 have harmful effects on public health (Gao and Ji, 2018; Jiang et al., 2019), ecosystems
45 (Ren et al., 2011; Yue et al., 2017), and crops (Wang et al., 2005; Wang et al., 2007).
46 Surface O_3 is a secondary pollutant produced by photochemical oxidation of volatile
47 organic compounds (VOCs) and nitrogen oxides ($NO_x \equiv NO+NO_2$) in the presence of
48 intense ultraviolet light, and the major $PM_{2.5}$ components (nitrate (NO_3^-), ammonium
49 (NH_4^+), sulfate (SO_4^{2-}), black carbon (BC), organic carbon (OC)) are **mainly** caused by
50 anthropogenic emissions of aerosols and aerosol precursors. Although surface O_3 and
51 $PM_{2.5}$ have different formation mechanisms, they are coupled through the common
52 precursors (NO_x and VOCs) and photochemical reactions (Chu et al., 2020). Since 2013,
53 stringent clean air actions have been implemented to improve air quality in China (State
54 Council of the People's Republic of China, 2013, 2018). However, O_3 concentrations
55 increased unexpectedly, while $PM_{2.5}$ concentrations decreased drastically in the past
56 years (Li et al., 2019). The co-polluted days by O_3 and $PM_{2.5}$ (concentrations of both
57 O_3 and $PM_{2.5}$ exceed the national air quality standards on the same day, hereafter
58 referred to as O_3 & $PM_{2.5}$ PD) were also reported (Dai et al., 2019). Therefore, it is
59 fundamental to examine the chemical and physical characteristics of O_3 & $PM_{2.5}$ PD.

60 The Beijing–Tianjin–Hebei (BTH) region is the most populated region in northern
61 China. In the past few years, concentrations of O_3 and $PM_{2.5}$ in the BTH were among
62 the highest in China. The **observations** from China National Environmental Monitoring
63 Center (CNEMC) showed that the mean and maximum MDA8 (daily maximum 8-h
64 average) O_3 in North China in summer of 2019 were 83 ppb and 129 ppb, respectively,
65 and the summer mean MDA8 O_3 increased with a trend of 3.3 ppb a^{-1} over 2013–2019
66 (Li et al., 2020). Gong et al. (2020) reported that O_3 polluted days (i.e., MDA8 O_3

67 concentration exceeds 80 ppb) in May-July in the BTH increased from 35 days in the
68 year of 2014 to 56 days in 2018. As for observed PM_{2.5}, the concentration averaged
69 over BTH had a decreasing trend of 10 $\mu\text{g m}^{-3} \text{ yr}^{-1}$ over 2013-2019, and the mean value
70 was $79 \pm 17 \mu\text{g m}^{-3}$ over these years (Li et al., 2020). BTH also had the highest
71 frequency and intensity of severe haze pollution days (i.e., days with daily mean PM_{2.5}
72 concentration exceeding 150 $\mu\text{g m}^{-3}$) in China from 2013 to 2017, with an observed
73 mean frequency of 21.2 d yr^{-1} and an observed mean intensity of 231.6 $\mu\text{g m}^{-3}$ (Dang
74 and Liao, 2019).

75 The interactions between O₃ and PM_{2.5} have been reported in previous studies.
76 Zhu et al. (2019) examined the spatial-temporal characteristics of the correlations
77 between observed O₃ and PM_{2.5} at 1497 sites in China for 2016 and found that O₃-
78 PM_{2.5} had the highest positive correlations (correlation coefficients > +0.7) in July in
79 southern China and the largest negative correlations (r values < -0.5) during January in
80 northern China. Li et al. (2019) used the GEOS-Chem model to analyze the O₃-PM_{2.5}
81 relationship in northern China and found that O₃ production was suppressed under high
82 PM_{2.5} conditions (PM_{2.5} concentrations > 60 $\mu\text{g m}^{-3}$) because of the reactive uptake of
83 hydrogen oxide radicals (HO_x) by aerosol particles. Chu et al. (2020) analyzed the
84 observed daily PM_{2.5} and O₃ concentrations in 114 cities in China during years of 2013-
85 2018 and found that the correlations between O₃ and PM_{2.5} tended to change from
86 negative in 2013 to positive in 2018 in China as air quality improved.

87 Few previous studies have examined the co-occurrence of O₃ and PM_{2.5} pollution
88 (MDA8 O₃ > 80 ppb and PM_{2.5} > 75 $\mu\text{g m}^{-3}$). Zong et al. (2021) used the obliquely
89 rotated principal component analysis in the T-mode (T-PCA) method to identify the
90 synoptic weather pattern associated with O₃&PM_{2.5}PD in eastern China during
91 summer of 2015–2018, and found that O₃&PM_{2.5}PD were associated with a stable

92 western Pacific subtropical high ridge, which brought warm and moist air flow from
93 the East China Sea to the eastern China to promote hygroscopic growth of fine
94 particulate matter in BTH and northern YRD. Dai et al. (2021) analyzed O₃&PM_{2.5}PD
95 in the YRD for April-October of 2013-2019 by using observations and reported that
96 the co-polluted days occurred mainly in April (29.6% of co-polluted days occurred in
97 April), May (23.0%), June (19.5%), and October (10.8%) under meteorological
98 conditions of higher relative humidity, higher surface air temperature, and lower wind
99 speed relative to the days with O₃ pollution alone. Qin et al. (2021) investigated
100 O₃&PM_{2.5}PD by using the hourly observed concentrations of water-soluble ions, OC,
101 and elemental carbon (EC) in 2019 in cities of Nanjing and Changzhou. They found
102 that inorganic aerosols mainly existed as NH₄NO₃ and the correlation coefficients
103 between the secondary components NO₃⁻, NH₄⁺, and SO₄²⁻ were relatively high during
104 O₃&PM_{2.5}PD in 2019, indicating a significant formation of secondary inorganic
105 aerosols. Although these studies have discussed the meteorological conditions and
106 some chemical characteristics of O₃&PM_{2.5}PD, the understanding of O₃&PM_{2.5}PD
107 was quite limited because of the limited observations of chemical species involved.

108 In this work, we take advantage of the comprehensive chemical mechanism of
109 the global chemical transport model to have better understanding of O₃&PM_{2.5}PD. We
110 apply the 3-D global chemical transport model (GEOS-Chem) to simulate
111 O₃&PM_{2.5}PD in BTH in years of 2013-2020, and investigate the chemical and
112 physical characteristics of O₃&PM_{2.5}PD by composited analyses of such days that are
113 captured by both the observations and the model. The objectives of this study are: 1)
114 to examine the underlying chemical mechanisms for O₃&PM_{2.5}PD in BTH for warm
115 seasons (April-October) of 2013-2020 by comparing O₃&PM_{2.5}PD with polluted days
116 by O₃ alone or by PM_{2.5} alone, and 2) to identify the weather patterns that are

117 associated with O₃&PM_{2.5}PD in BTH. The observations, the reanalyzed
118 meteorological data, the GEOS-Chem model, and the process analysis are described
119 in Section 2. The observed O₃&PM_{2.5}PD are presented in Section 3.1. The evaluation
120 of simulated concentrations of O₃ and PM_{2.5} as well as the simulated pollution days by
121 O₃ and/or PM_{2.5} are shown in Section 3.2. The underlying mechanisms of
122 O₃&PM_{2.5}PD are shown in Section 3.3. In Section 3.4, the meteorological conditions
123 for the co-occurrence of O₃ and PM_{2.5} pollution are investigated. The conclusions are
124 presented in Section 4.

125

126 **2. Methods**

127 **2.1 Observed O₃ and PM_{2.5} concentrations**

128 Hourly concentrations of PM_{2.5} and O₃ in China over the years of 2013-2020
129 were taken from the public website of CNEMC (<https://air.cnemc.cn:18007/>,
130 CNEMC, 2022). To ensure data quality, the daily mean PM_{2.5} concentration was
131 calculated when there were valid data for more than 20 h during that day and the
132 MDA8 O₃ concentration was calculated when there were valid data for at least 6 h for
133 each 8 h. For the calculation of monthly and annual mean concentrations, the number
134 of days with valid concentrations had to be more than 15 in each month. The spatial
135 distribution of the 79 valid sites within BTH (37-41°N, 114-118°E, the black
136 rectangle) is shown in Fig. 1. For model evaluation, the observed concentrations were
137 averaged over sites within each of the 0.5° latitude × 0.625° longitude MERRA-2
138 grid cell. There are 18 model grids in BTH. Note that the observed O₃ concentrations
139 from this network have a unit of μg m⁻³. For the consistency of observed and
140 simulated O₃ concentrations, 1 μg m⁻³ of O₃ is approximately 0.5 ppb under the
141 conditions of 298 K and 1013 hPa. The observed O₃ concentrations reported by the

142 CNEMC were under standard conditions of 273 K and 1013 hPa before 31 August
143 2018 and were under standard conditions of 298 K and 1013 hPa afterwards
144 (http://www.mee.gov.cn/xxgk2018/xxgk/xxgk01/201808/t20180815_629602.html),
145 which were accounted for as O₃ concentrations were converted to ppb.

146 According to the National Ambient Air Quality Standard of China (GB3095-
147 2012), O₃ (PM_{2.5}) concentration exceeds the national air quality standard if the MDA8
148 O₃ (daily mean PM_{2.5}) concentration is higher than 160 μg m⁻³ (75 μg m⁻³). In this
149 study, we define O₃ polluted days (hereafter called ‘O₃PD’) for days with MDA8 O₃
150 concentration > 160 μg m⁻³, PM_{2.5} polluted days (hereafter called ‘PM_{2.5}PD’) with
151 daily mean PM_{2.5} concentration > 75 μg m⁻³, and the co-pollution days by O₃ and
152 PM_{2.5} (O₃&PM_{2.5}PD) with daily MDA8 O₃ concentration > 160 μg m⁻³ as well as the
153 daily mean PM_{2.5} concentration > 75 μg m⁻³.

154

155 **2.2 Reanalyzed meteorological fields**

156 Meteorological fields were obtained from the Version 2 of Modern Era
157 Retrospective-analysis for Research and Application (MERRA2), which were
158 generated by the NASA Global Modeling and Assimilation Office (GMAO). The
159 MERRA2 data have a horizontal resolution of 0.5° latitude × 0.625° longitude and
160 72 vertical layers (Molod et al., 2015). To analyze the meteorological conditions for
161 O₃&PM_{2.5}PD, vertical pressure velocity (OMEGA), planetary boundary layer height
162 (PBLH), temperature (T), relative humidity (RH), surface incoming shortwave flux
163 (SWGDN) are used. Note that the temporal resolution for PBLH, T, and SWGDN is
164 1h, and that for OMEGA and RH is 3h. Daily mean geopotential heights at 850 and
165 500 hPa from the National Center for Environmental Prediction (NCEP) and National
166 Center for Atmospheric Research (NCAR) global reanalysis with a resolution of 2.5°

167 latitude by 2.5° longitude are also utilized in this study.

168

169 **2.3 Observed aerosol optical depth**

170 We obtained the version 3 datasets of observed daily aerosol optical depth
171 (AOD) of level 2 (improved cloud screened and quality-assured) from Aerosol
172 Robotic Network (AERONET, https://aeronet.gsfc.nasa.gov/new_web/index.html)
173 established by NASA and LOA-PHOTONS (Giles et al., 2019). Three sites in the
174 BTH region have observations available over 2013-2020, including Beijing (39.97°N ,
175 116.38°E), Beijing-CAMS (39.93°N , 116.31°E), and Xianghe (39.75°N , 116.96°E).
176 The AOD values at 440 nm and 675 nm at these three sites are analyzed in this study.

177

178 **2.4 GEOS-Chem model**

179 We simulated O_3 and $\text{PM}_{2.5}$ using the nested version of the 3-D global chemical
180 transport model (GEOS-Chem, version 11-01) driven by the MERRA2
181 meteorological data. The nested domain was set over Asia ($60^\circ\text{-}150^\circ\text{E}$, $11^\circ\text{S-}55^\circ\text{N}$)
182 with a horizontal resolution of 0.5° latitude \times 0.625° longitude, and the chemical
183 boundary conditions were provided by the global GEOS-Chem simulation with 2.5°
184 latitude \times 2.5° longitude horizontal resolution.

185 The GEOS-Chem model includes fully coupled O_3 - NO_x -hydrocarbon and
186 aerosol chemistry mechanism (Bey et al., 2001; Pye et al., 2009) to simulate aerosols
187 including SO_4^{2-} (Park et al., 2004), NO_3^- (Pye et al., 2009), NH_4^+ , BC and OC (Park
188 et al., 2003), mineral dust (Fairlie et al., 2007), and sea salt (Alexander et al., 2005) as
189 well as the gas-phase pollutants such as NO_x and O_3 . Over the Asian domain, the
190 anthropogenic emissions of OC, BC, carbon monoxide (CO), sulfur dioxide (SO_2),

191 NO_x, ammonia (NH₃), and VOCs were obtained from the Multi-resolution Emission
192 Inventory for China (MEIC), which includes emissions from industry, power,
193 residential and transportation sectors for years of 2014-2017 (Li et al., 2017; Zheng et
194 al., 2018), 2019 and 2020 (Zheng et al., 2021). Emissions in 2018 were obtained by
195 the interpolation of those in 2017 and 2019 for each grid due to the lack of publicly
196 accessible emission inventories for that year. The biogenic emissions in GEOS-Chem
197 are simulated using MEGAN v2.1 (Guenther et al., 2012).

198 The hourly O₃ and PM_{2.5} concentrations for the years of 2013-2020 were
199 simulated by the GEOS-Chem model which were driven by MERRA-2
200 meteorological fields. The model was spined up for 6 months before the integration
201 over the studied time period.

202

203 **2.5 Process analysis**

204 Process analysis (PA) was applied to identify the relative importance of
205 atmospheric processes in O₃&PM_{2.5}PD. PA has been widely used in previous studies
206 to examine the key processes contributing to air pollution episodes (Gonçalves et al.,
207 2009; Dang and Liao, 2019; Gong and Liao, 2019) as well as the interannual and
208 decadal variations of air pollutants (Mu and Liao, 2014; Lou et al., 2015). Five major
209 processes that influence O₃ and PM_{2.5} concentrations were diagnosed at every time
210 step, including net chemical production, dry deposition, horizontal advection, vertical
211 advection, and diffusion, for the regional pollution days (days with more than half of
212 the sites in BTH experiencing pollutions). We carried out PA for O₃SPD (exclude
213 O₃&PM_{2.5}PD from O₃PD), PM_{2.5}SPD (exclude O₃&PM_{2.5}PD from PM_{2.5}PD), and
214 O₃&PM_{2.5}PD over BTH.

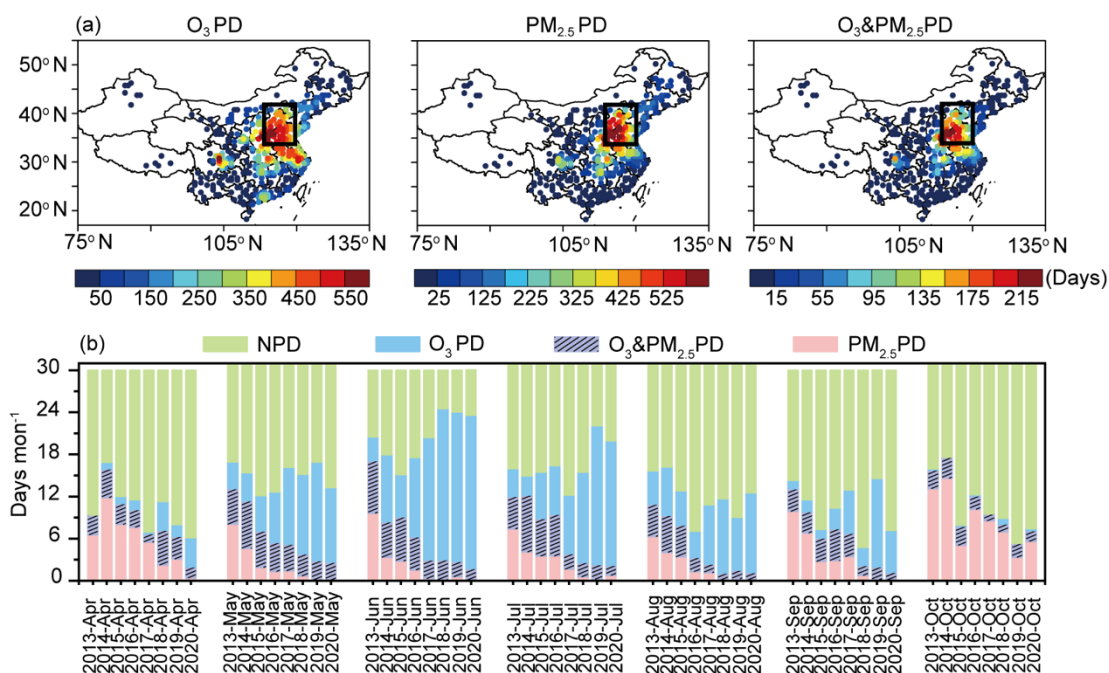
215

216 **3. Results**

217 **3.1 Observed polluted days by O₃ and PM_{2.5}**

218 Figure 1a shows the spatial distributions of observed numbers of O₃PD,
219 PM_{2.5}PD, and O₃&PM_{2.5}PD summed over the warm seasons (April-October) of 2013-
220 2020. The spatial distributions of polluted days in each year are shown in Fig. S1. The
221 numbers of O₃PD, PM_{2.5}PD, and O₃&PM_{2.5}PD were high in BTH, which were,
222 respectively, 524.3 344.6, and 128.1 days from observations, as the values were
223 averaged over all sites in BTH. The high numbers of O₃PD, PM_{2.5}PD, and
224 O₃&PM_{2.5}PD in BTH were associated with the highest anthropogenic emissions (NO_x
225 and NMVOCs) in this region (Dang et al., 2021).

226 Figure 1b shows the numbers of days averaged over all sites in BTH for non-
227 polluted days (NPD, MDA8 O₃ < 80 ppb and PM_{2.5} < 75 µg m⁻³), O₃PD,
228 O₃&PM_{2.5}PD, and PM_{2.5}PD in each month of warm seasons from 2013 to 2020. O₃PD
229 and O₃&PM_{2.5}PD mainly occurred in May, June, and July, while PM_{2.5}PD mainly
230 appeared in April and October. The monthly numbers of O₃&PM_{2.5}PD (PM_{2.5}PD)
231 declined from 2013 to 2020, with the fastest drop in June, from 7.5 (17.1) days in
232 June 2013 to 1.8 (1.8) days in June 2020. On the contrary, the numbers of O₃PD kept
233 increasing, especially in June, from 10.9 days in June 2013 to 23.6 days in June 2020.
234 The reductions in O₃&PM_{2.5}PD were associated with the large reductions in PM_{2.5}
235 since the implementation of the Clean Air Action in 2013.

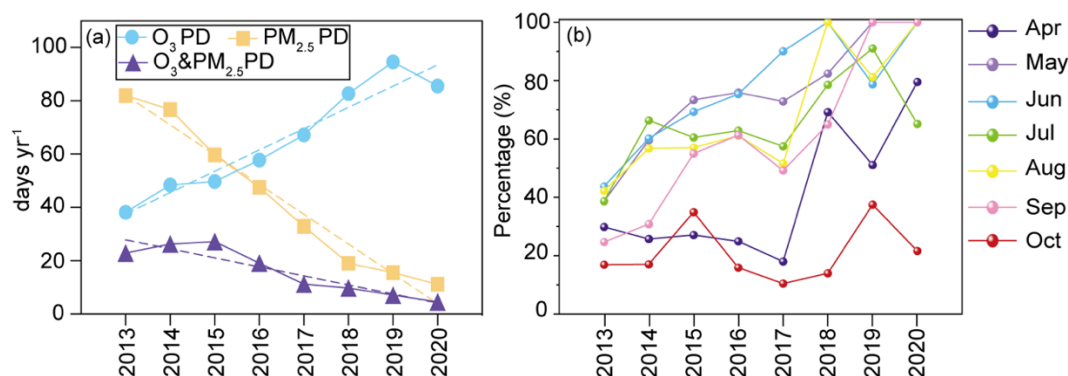


236

237 **Figure 1.** (a) Spatial distributions of observed numbers of O₃PD, PM_{2.5}PD, and
 238 O₃&PM_{2.5}PD summed over April-October of 2013-2020. The solid black rectangle
 239 indicates the BTH region. (b) The observed numbers of NPD (non-polluted days,
 240 green), O₃PD (blue + purple with slashes), O₃&PM_{2.5}PD (purple with slashes), and
 241 PM_{2.5}PD (pink + purple with slashes) averaged over all sites in BTH from April to
 242 October in 2013 to 2020.

243 Figure 2a shows the linear trends of observed O₃PD, PM_{2.5}PD, and O₃&PM_{2.5}PD
 244 in warm seasons of 2013-2020 averaged over the BTH. O₃PD showed an upward
 245 trend of 7.9 days yr⁻¹ from 2013 to 2020. However, the numbers of PM_{2.5}PD and
 246 O₃&PM_{2.5}PD decreased over 2013-2020, with linear trends of -11.2 and -3.4 days yr⁻¹,
 247 respectively. Figure 2b shows the changes in percentage of O₃&PM_{2.5}PD in PM_{2.5}PD
 248 from 2013 to 2020 for each month. It should be noted that, when PM_{2.5}PD occurred,
 249 the proportions of O₃&PM_{2.5}PD had an upward trend from 2013 to 2020. In May,
 250 June, August, and September of 2020, the proportions of O₃&PM_{2.5}PD in PM_{2.5}PD
 251 reached 100%, indicating that PM_{2.5} pollution was accompanied by O₃ pollution in

252 recent years.



253

254 **Figure 2.** (a) The trends of observed O₃PD, PM_{2.5}PD, and O₃&PM_{2.5}PD in warm
255 seasons from 2013 to 2020 averaged over all sites in BTH. The blue, yellow and purple
256 solid lines (dashed lines) represent the numbers (liner trend) of O₃PD, PM_{2.5}PD, and
257 O₃&PM_{2.5}PD, respectively. (b) The percentage of O₃&PM_{2.5}PD in PM_{2.5}PD for April

258 to October in 2013 to 2020. The polluted days were averaged over all sites in BTH.

259

260 3.2 Simulated polluted days and model evaluation

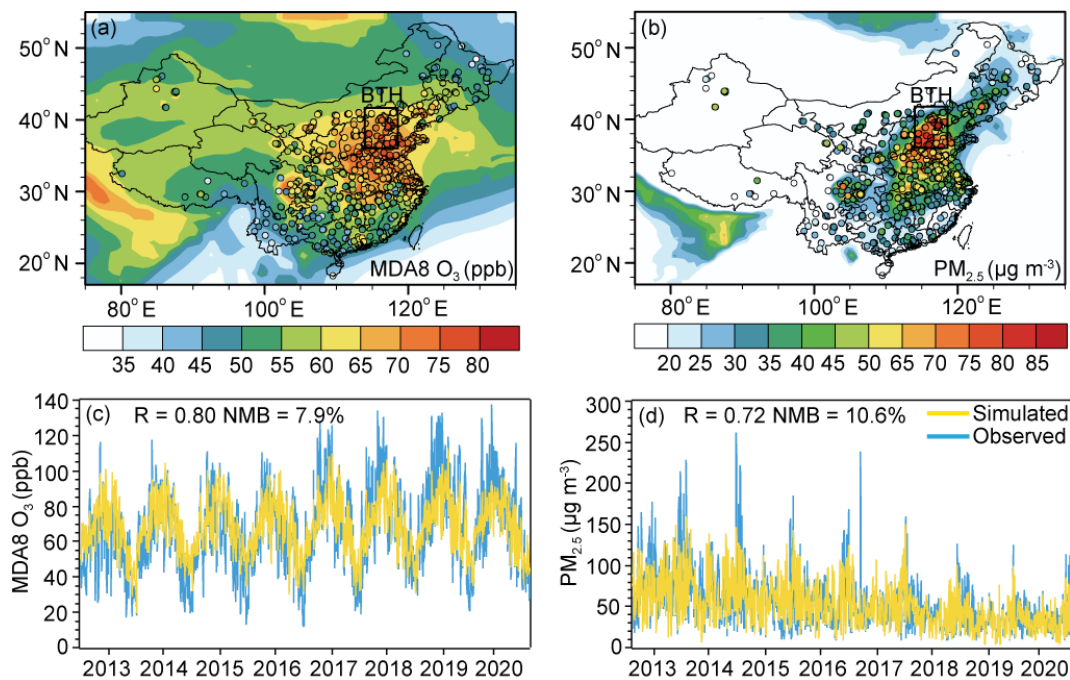
261 3.2.1 Simulated surface-layer MDA8 O₃ and PM_{2.5} concentrations

262 Figures 3a and 3b show, respectively, the spatial distributions of simulated and
263 observed surface-layer concentrations of MDA8 O₃ and PM_{2.5} in China, as the
264 concentrations are averaged over the warm seasons (April-October) of 2013-2020.

265 The concentrations of MDA8 O₃ and PM_{2.5} were both high in BTH. Averaged over
266 BTH and the studied time period, the observed concentrations of MDA8 O₃ and PM_{2.5}
267 were 58.1 ppb and 60.3 $\mu\text{g m}^{-3}$, respectively, while the simulated values were 68.0 ppb
268 and 61.1 $\mu\text{g m}^{-3}$, respectively. Figures 3c and 3d compare the time series of observed
269 and simulated daily MDA8 O₃ and PM_{2.5} concentrations averaged over the BTH. The
270 simulated daily concentrations of MDA8 O₃ (PM_{2.5}) in eight warm seasons have a
271 normalized mean bias (NMB) of 7.9% (10.6%). The model generally captures the
272 daily variations (peaks and troughs) in the observed MDA8 O₃ and PM_{2.5}

273 concentrations, with R values of 0.80 and 0.72, respectively. It should be noted that
 274 mineral dust and sea-salt aerosols were not considered in this study, because they are
 275 not the major aerosol components in China and the concentrations are generally low
 276 based on previous measurements (Xuan et al., 2000; Ye et al., 2003; Duan et al., 2006;
 277 Zhao et al., 2013). However, excluding dust and sea salt may lead to low biases in
 278 simulated PM_{2.5} concentrations.

279 Due to the lack of the publicly accessible long-term observations of PM_{2.5}
 280 components in China, we compared the simulated SO₂ and NO₂ (precursors for SO₄²⁻
 281 and NO₃) with observations from CNEMC in Fig. S2. The simulated daily mean
 282 concentrations of NO₂ (SO₂) agree well with the observations from CNEMC with R
 283 of 0.82 (0.78) and MB of -14.9% (9.3%).



284
 285 **Figure 3.** Spatial distributions of simulated (shades) and observed (CNEMC, dots)
 286 surface-layer concentrations of (a) MDA8 O₃ (ppb) and (b) PM_{2.5} (µg m⁻³) averaged
 287 over the eight warm seasons (April to October, 2013–2020). The solid black rectangle
 288 in (a) and (b) indicates the BTH region. Simulated and observed daily concentrations

289 of surface-layer (c) MDA8 O₃ and (d) PM_{2.5} averaged over BTH. The correlation
290 coefficient (R) and normalized mean bias (NMB) are also shown for (c) and (d).

291 $NMB = (\sum_{i=1}^N (M_i - O_i) / \sum_{i=1}^N (O_i)) \times 100\%$, where O_i and M_i are the observed and
292 simulated concentrations, respectively, i refers to the i^{th} day, and N is the total number
293 of days.

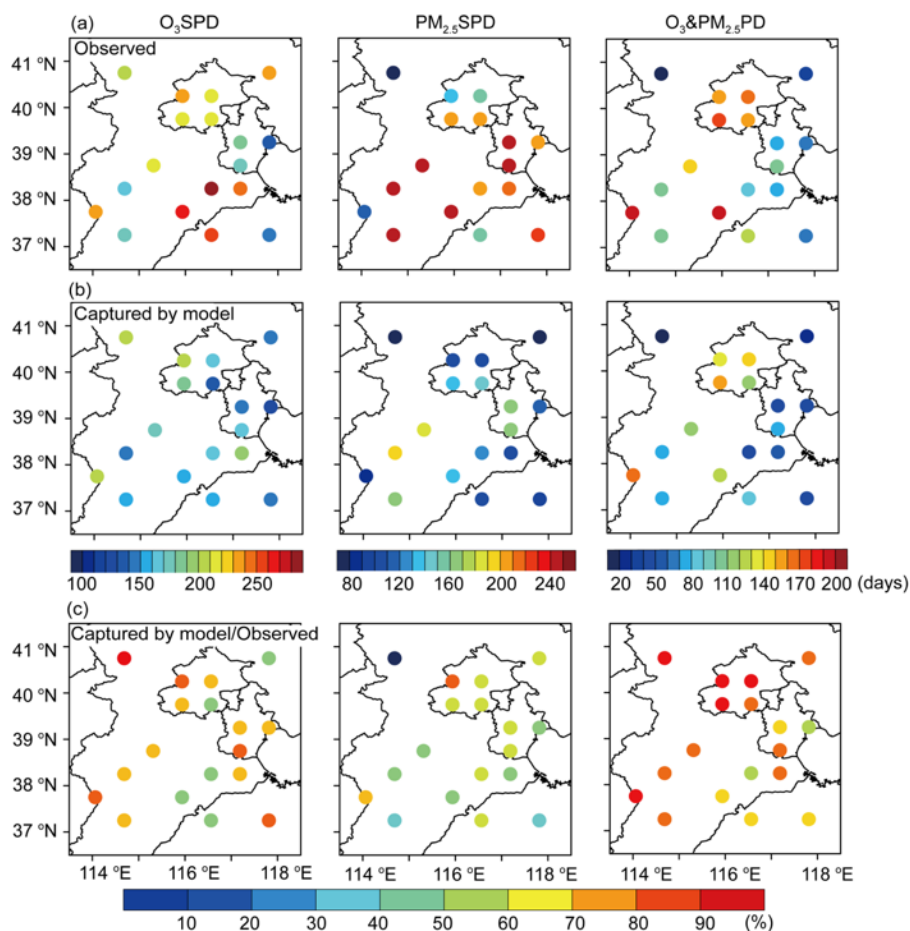
294 **3.2.2 Simulated O₃PD, PM_{2.5}PD, and O₃&PM_{2.5}PD**

295 Figure S3 shows the capability of the model in capturing the polluted days.
296 Although the GEOS-Chem model well reproduces the spatial distributions of observed
297 MDA8 O₃ and PM_{2.5} concentrations, it underestimates the numbers of O₃PD, PM_{2.5}PD,
298 and O₃&PM_{2.5}PD because of the model's deficiency in capturing the peak
299 concentrations of air pollutants. Such deficiency was also reported in previous studies
300 that used the GEOS-Chem model or the weather Research and Forecasting with
301 Chemistry (WFR-chem) model (Zhang et al., 2016; Ni et al., 2018; Gong and Liao,
302 2019; Dang and Liao, 2019). Therefore, to identify O₃PD, PM_{2.5}PD, and O₃&PM_{2.5}PD
303 using model results, we utilized lower thresholds by considering the NMBs of simulated
304 MDA8 O₃ and PM_{2.5} concentrations in each of 18 grids of BTH. Taking the grid of
305 Beijing as an example, simulated MDA8 O₃ and PM_{2.5} had NMBs of -22.0% and -
306 26.9%, respectively, as the simulated concentrations were compared with observations
307 for days with observed concentrations higher than the national air quality standards over
308 the warm seasons of 2013-2020. We then adjusted the threshold of O₃PD in this grid to
309 be 62.4 ppb (80 ppb×78%) and that of PM_{2.5}PD to be 54.8 μg m⁻³ (75 μg m⁻³×73.1%).
310 These adjusted thresholds were also used to identify O₃&PM_{2.5}PD. Such approach was
311 also used in previous studies to better capture the pollution events based on the
312 simulations (Dang and Liao, 2019; Gong and Liao, 2019). With the adjusted thresholds,
313 56-93% of the observed O₃PD, PM_{2.5}PD, and O₃&PM_{2.5}PD can be captured by the

314 model (Fig. S3e).

315 **3.2.3 Simulated O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD**

316 Since O₃PD or PM_{2.5}PD encompasses O₃&PM_{2.5}PD, we further define O₃ single
317 pollution days (hereafter called “O₃SPD”, which is to exclude O₃&PM_{2.5}PD from O₃PD)
318 and PM_{2.5} single pollution days (hereafter called “PM_{2.5}SPD”, which is to exclude
319 O₃&PM_{2.5}PD from PM_{2.5}PD) for the purpose of obtaining the characteristics of
320 different polluted days. Figures 4a and 4b show, respectively, the spatial distributions
321 of numbers of O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD from observations and from the
322 GEOS-Chem model using the adjusted thresholds. Considering the total of polluted
323 days in 18 grids in BTH, observed O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD were,
324 respectively, 3937, 3698, and 2024 days, in which 75.0% (2954/3937), 58.1%
325 (2148/3698), and 79.7% (1614/2024) were captured by observation and simulation
326 simultaneously (Fig. 4c). In addition, the numbers of observed and captured O₃SPD,
327 PM_{2.5}SPD, and O₃&PM_{2.5}PD in each month are shown in Fig. S4. The model has a
328 fairly good capability of capturing the observed polluted days in each month.



329

330 **Figure 4.** Spatial distributions of (a) observed numbers of O₃SPD, PM_{2.5}SPD, and
 331 O₃&PM_{2.5}PD, (b) numbers of polluted days that were observed and also captured by
 332 the GEOS-Chem model with adjusted thresholds, and (c) percentages of observed
 333 polluted days that were captured by the model with adjusted thresholds. The values
 334 were calculated for the warm months (April to October) of 2013-2020.

335

336 3.3 Chemical characteristics of polluted days by O₃ and PM_{2.5}

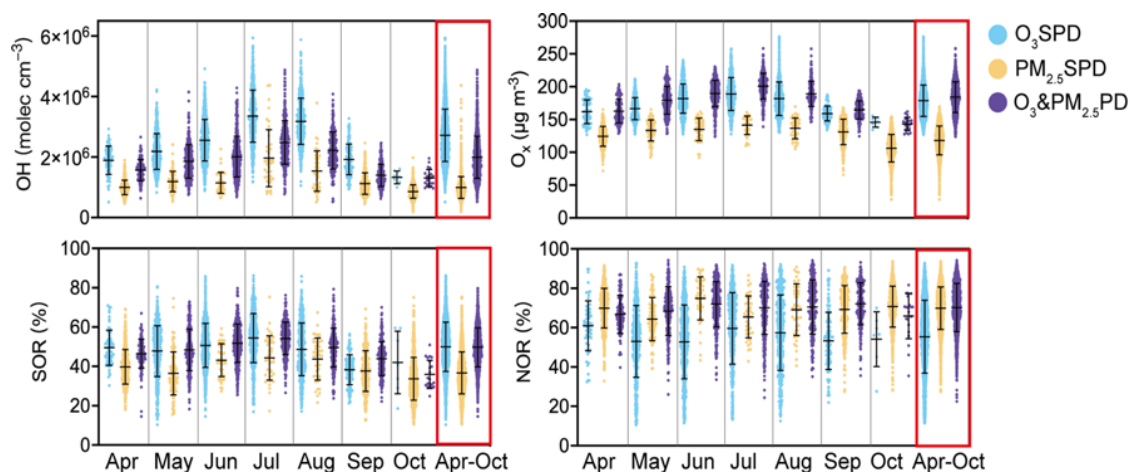
337 In this section, to investigate the chemical characteristics of O₃SPD, PM_{2.5}SPD
 338 and O₃&PM_{2.5}PD, we present first the simulated atmospheric oxidants in 3.3.1, and
 339 then show the simulated surface concentrations and vertical profiles of PM_{2.5} and
 340 MDA8 O₃ in 3.3.2 and 3.3.3, respectively, followed by the process analysis in 3.3.4.
 341 The observed AOD values to verify the model results are presented in 3.3.5.

342 **3.3.1 Atmospheric oxidants of O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD**

343 Figure 5 shows the boxplots of daily concentrations of hydroxyl radical (OH) and
344 total oxidant ($O_x = O_3 + NO_2$) from the model for days of O₃SPD, PM_{2.5}SPD, and
345 O₃&PM_{2.5}PD that were observed and also captured by the model (samples in Fig. 4b)
346 in the warm seasons of 2013-2020 in 18 grids of BTH. The levels of OH and O_x
347 characterize the atmospheric oxidation capacity, following Hu et al. (2020) and Chan
348 et al. (2017). The concentrations of OH were the highest in O₃SPD, with an averaged
349 value of 2.8×10^6 molec cm⁻³, followed by that in O₃&PM_{2.5}PD (2.0×10^6 molec cm⁻³)
350 and in PM_{2.5}SPD (1.0×10^6 molec cm⁻³). Due to the lack of publicly accessible
351 observations of OH in BTH, we compare the simulated OH concentrations with
352 observations reported in the literature (Table S1). The simulated OH concentrations
353 agree closely with the observed values. In Wangdu of BTH, while the observed daily
354 maximum OH concentrations in summer of 2014 were in the range of $5\text{-}15 \times 10^6$ molec
355 cm⁻³ (Tan et al., 2016), the simulated OH concentrations in the same time period in this
356 work were $3.7\text{-}9.5 \times 10^6$ molec cm⁻³. In Beijing in summer of 2017, the observed daily
357 mean OH concentration was 5.8×10^6 molec cm⁻³ (Woodward et al., 2020) and our
358 simulated value was 2.4×10^6 molec cm⁻³.

359 The mean values of O_x were, respectively, 178.7, 118.1, and 184.1 $\mu\text{g m}^{-3}$ in
360 O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD, indicating that the atmospheric oxidation
361 capacity was strong in O₃&PM_{2.5}PD, which favored the production of secondary
362 components of PM_{2.5}. Figure 5 also shows sulfur oxidation ratio (SOR, $n\text{-SO}_4^{2-} / (n\text{-SO}_4^{2-}$
363 $+ n\text{-SO}_2)$, where $n\text{-SO}_4^{2-}$ and $n\text{-SO}_2$ are the concentrations of SO_4^{2-} and SO_2 ,
364 respectively) and nitrogen oxidation ratio (NOR, $n\text{-NO}_3^- / (n\text{-NO}_3^- + n\text{-NO}_2)$, where $n\text{-NO}_3^-$
365 NO_3^- and $n\text{-NO}_2$ are the concentrations of NO_3^- and NO_2 , respectively). SOR and NOR
366 are measures of the conversion degrees of sulfur and nitrogen, respectively (Zhu et al.,

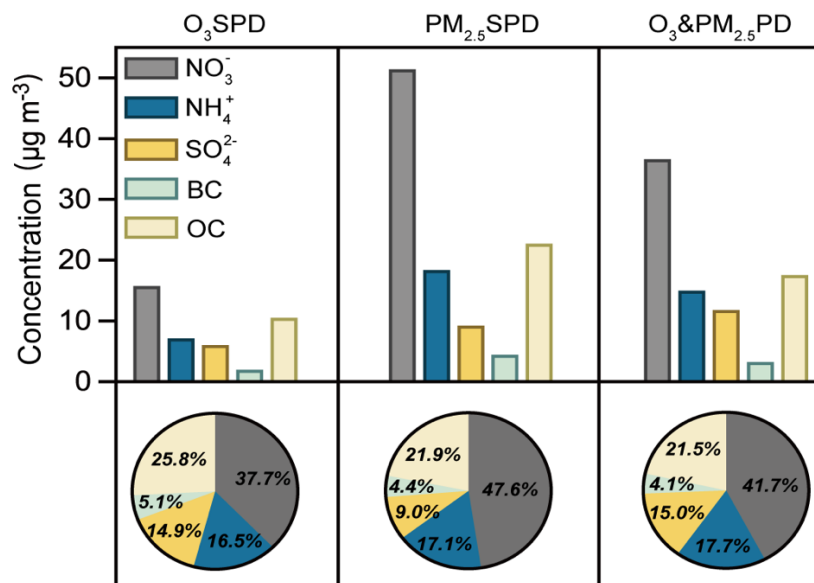
367 2019). In O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD, the averaged values of SOR were
 368 50.0%, 36.7%, and 49.7%, and those of NOR were 55.4%, 70.0%, and 70.2%,
 369 respectively. The high SOR and NOR in O₃&PM_{2.5}PD indicated the strong formation
 370 of SO₄²⁻ and NO₃⁻ that were promoted by high atmospheric oxidation capacity. The
 371 monthly variations of OH, O_x, and SOR were similar (Fig. 5), with the highest values
 372 in summer, owing to the high temperature that promoted high concentrations of
 373 oxidants and SOR. It is interesting that SOR and O_x values were higher in O₃&PM_{2.5}PD
 374 than in O₃SPD or in PM_{2.5}SPD during May-August. Similarly, NOR values were higher
 375 in O₃&PM_{2.5}PD than in O₃SPD or in PM_{2.5}SPD in May and July-September. Overall,
 376 the O₃&PM_{2.5}PD occurred with high levels of atmospheric oxidants, SOR, and NOR,
 377 leading to combined increases in O₃ and PM_{2.5} concentrations.



378 **Figure 5.** The boxplots of surface-layer hydroxyl radical (OH, molec cm⁻³), total
 379 oxidant (O_x, μg m⁻³), sulfur oxidation ratio (SOR, %), nitrogen oxidation ratio
 380 (NOR, %) for model-captured O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD in 18 grids of
 381 BTH in the months of April to October from 2013 to 2020. The whiskers represent the
 382 standard deviation, the black line represents the mean value of the samples.

384 **3.3.2 Surface-layer concentrations of PM_{2.5} components in O₃SPD, PM_{2.5}SPD, and**
 385 **O₃&PM_{2.5}PD**

386 The simulated concentrations of PM_{2.5} components (NO₃⁻, NH₄⁺, SO₄²⁻, BC, and
 387 OC, averaged over 18 grids of BTH are shown in Fig. 6 for days of O₃SPD, PM_{2.5}SPD,
 388 and O₃&PM_{2.5}PD in the warm seasons of 2013-2020 that were observed and also
 389 captured by the model. While the mean concentrations of NO₃⁻, NH₄⁺, BC, and OC,
 390 were all the highest in PM_{2.5}SPD, SO₄²⁻ concentration was the highest in O₃&PM_{2.5}PD.
 391 In O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD, the mean concentrations of SO₄²⁻ were 6.2,
 392 9.4, and 11.97 μg m⁻³, respectively, and the percentages of SO₄²⁻ in PM_{2.5} were 14.9%,
 393 9.0%, and 15.0%, respectively. In July and August, the concentrations of SO₄²⁻ and
 394 MDA8 O₃ in O₃&PM_{2.5}PD were the highest compared with those in O₃SPD and
 395 PM_{2.5}SPD (Fig. S5).

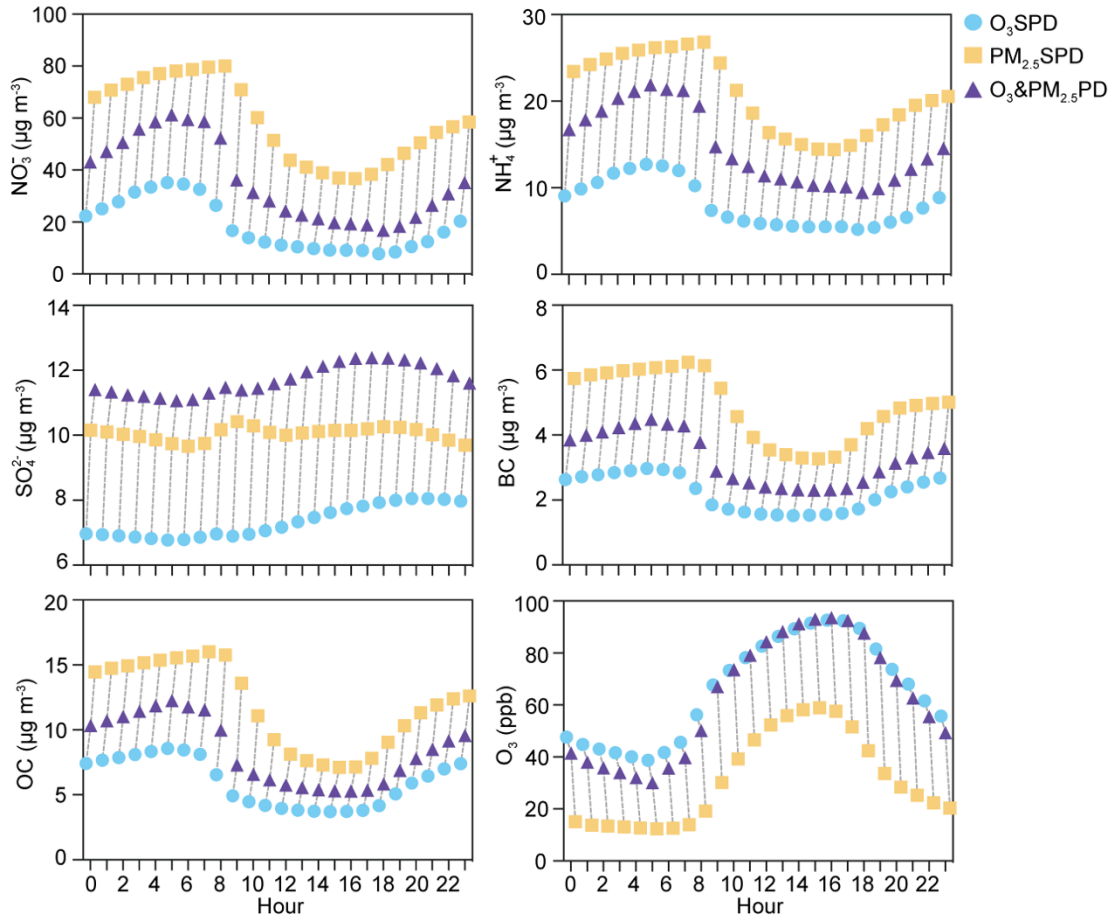


396
 397 **Figure 6.** The concentrations of PM_{2.5} components (μg m⁻³) and percentages of PM_{2.5}
 398 components (%) at the surface for NO₃⁻, NH₄⁺, SO₄²⁻, BC, and OC. The values were
 399 averaged over the model-captured O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD in the months
 400 of April to October of 2013-2020 in BTH.

401 Figure 7 presents the hourly concentrations of NO₃⁻, NH₄⁺, SO₄²⁻, BC, OC, and O₃
 402 for model-captured O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD over all 18 grids of BTH in

403 the warm seasons from 2013-2020. Concentrations of NO_3^- and NH_4^+ had similarities
404 in diurnal variations, all of which reached the highest values in the early morning (5:00
405 local time (LT) in O_3SPD and $\text{O}_3\&\text{PM}_{2.5}\text{PD}$, 7:00-8:00 LT in $\text{PM}_{2.5}\text{SPD}$) and had the
406 lowest values in the late afternoon (18:00 LT in O_3SPD and $\text{O}_3\&\text{PM}_{2.5}\text{PD}$, 16:00 LT in
407 $\text{PM}_{2.5}\text{SPD}$). Concentrations of BC and OC peaked at the same time as those of NO_3^-
408 and NH_4^+ and had the lowest values at 15:00 LT in O_3SPD , $\text{PM}_{2.5}\text{SPD}$, and
409 $\text{O}_3\&\text{PM}_{2.5}\text{PD}$. The diurnal variations in NO_3^- , NH_4^+ , BC, OC reflected the diurnal
410 variations in PBLH (shown in Fig. S6), which generally reached their highest
411 concentrations before the sudden uplift of PBLH in the early morning (times for uplift
412 of PBLH: 6:00 LT in O_3SPD and $\text{O}_3\&\text{PM}_{2.5}\text{PD}$, 7:00 LT in $\text{PM}_{2.5}\text{SPD}$). Compared to
413 O_3SPD and $\text{O}_3\&\text{PM}_{2.5}\text{PD}$, the PBLH of $\text{PM}_{2.5}\text{SPD}$ was lower and uplifted one hour
414 later, which was more favorable for the accumulation of aerosols. During the daytime,
415 PBLH in $\text{O}_3\&\text{PM}_{2.5}\text{PD}$ was between O_3SPD and $\text{PM}_{2.5}\text{SPD}$.

416 It is worth noting that the diurnal variations of SO_4^{2-} were different from those of
417 other aerosol species, with the highest values at 20:00, 9:00, and 16:00 LT in O_3SPD ,
418 $\text{PM}_{2.5}\text{SPD}$, and $\text{O}_3\&\text{PM}_{2.5}\text{PD}$, respectively, and the lowest values in early morning and
419 night (5:00 LT in O_3SPD and $\text{O}_3\&\text{PM}_{2.5}\text{PD}$, 23:00 LT in $\text{PM}_{2.5}\text{SPD}$). For the diurnal
420 variation of O_3 , the highest values occurred during the daytime (16:00 LT in O_3SPD
421 and $\text{O}_3\&\text{PM}_{2.5}\text{PD}$, 15:00 LT in $\text{PM}_{2.5}\text{SPD}$) and the lowest values appeared at 5:00 LT
422 in all the cases. Therefore, in $\text{O}_3\&\text{PM}_{2.5}\text{PD}$, the time of the highest value of SO_4^{2-} was
423 the same as that of O_3 , indicating that SO_4^{2-} and O_3 were produced synergistically
424 during the daytime with strong atmospheric oxidation.



425

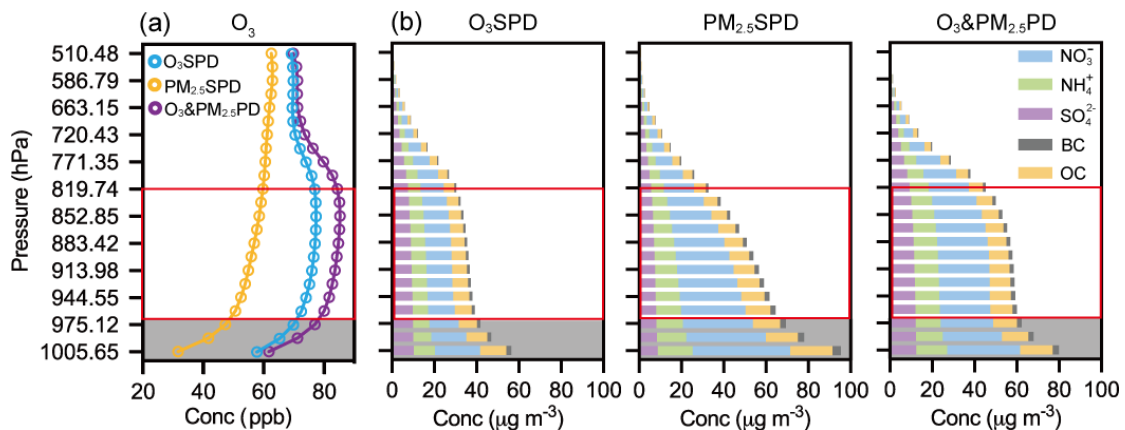
426 **Figure 7.** The hourly concentrations of NO_3^- , NH_4^+ , SO_4^{2-} , BC, OC, and O_3 averaged
 427 over the model-captured O_3SPD , $\text{PM}_{2.5}\text{SPD}$, and $\text{O}_3 \& \text{PM}_{2.5}\text{PD}$ in BTH in the months
 428 of April to October of 2013-2020.

429 **3.3.3 Vertical distributions of O_3 and $\text{PM}_{2.5}$ in O_3SPD , $\text{PM}_{2.5}\text{SPD}$, and**
 430 **$\text{O}_3\&\text{PM}_{2.5}\text{PD}$**

431 The simulated vertical distributions of O_3 and $\text{PM}_{2.5}$ averaged over the 18 grids of
 432 BTH and the O_3SPD , $\text{PM}_{2.5}\text{SPD}$, and $\text{O}_3\&\text{PM}_{2.5}\text{PD}$ in warm seasons of 2013-2020 are
 433 shown in Fig. 8. The vertical distribution of O_3 in O_3SPD was similar to that in
 434 $\text{O}_3\&\text{PM}_{2.5}\text{PD}$ (Fig. 8a). In these two cases, concentrations of O_3 increased from the
 435 surface to about 975 hPa, remained high between 975 and 819 hPa, and decreased with
 436 altitude between 819 and 663 hPa. Although the magnitudes of O_3 were close at the
 437 surface (61.9 ppb in $\text{O}_3\&\text{PM}_{2.5}\text{PD}$ and 58.1 ppb in O_3SPD), the concentration of O_3

438 averaged over 975 and 819 hPa was 10.4% higher in O₃&PM_{2.5}PD than in O₃SPD,
 439 which was a very unique feature of O₃&PM_{2.5}PD. For the case of PM_{2.5}SPD, the
 440 concentrations of O₃ were the lowest among the three cases and increased gently with
 441 altitude above 975 hPa.

442 Figure 8b shows the vertical distributions of PM_{2.5} components. In all the cases,
 443 PM_{2.5} concentrations were the highest at the surface, and decreased with altitude from
 444 the surface to 975 hPa. However, concentrations of PM_{2.5} were quite stable between
 445 975 and 819 hPa for O₃SPD (about 36.4 μg m⁻³) and O₃&PM_{2.5}PD (about 58.1 μg m⁻³),
 446 corresponding to the stable O₃ levels at these altitudes in these two cases (Fig. 8a). For
 447 PM_{2.5}SPD, while PM_{2.5} concentration at the surface was the highest among the three
 448 cases, it decreased rapidly between 975 and 819 hPa. The averaged PM_{2.5} concentration
 449 between 975 and 819 hPa was 52.4 μg m⁻³ in PM_{2.5}SPD, which was lower than that in
 450 O₃&PM_{2.5}PD.



451

452 **Figure 8.** The vertical distributions of (a) concentrations of O₃ (ppb) and (b) PM_{2.5}

453 components (μg m⁻³) of NO₃⁻, NH₄⁺, SO₄²⁻, BC, OC averaged over the model-

454 captured O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD in BTH in the months of April to

455 October of 2013-2020.

456 To further investigate the differences in vertical profiles of NO₃⁻, NH₄⁺, SO₄²⁻, BC,

457 OC, and PM_{2.5} in O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD, the ratios of concentration at
 458 975 hPa to that at the surface as well as the concentration at 819 hPa to that at 975 hPa
 459 are shown in Table 1. The concentration of PM_{2.5} decreased largely, with the ratio of
 460 PM_{2.5}(975 hPa) / PM_{2.5}(1005 hPa) of 0.78 in O₃&PM_{2.5}PD and of 0.74 in PM_{2.5}SPD. For each
 461 of the PM_{2.5} components, the ratios near the surface (from 1005 to 975 hPa, gray shaded
 462 area in Fig. 8) were close in the three types of pollution. While the ratios of NO₃⁻, NH₄⁺,
 463 BC, OC were in the range of 0.65-0.80, the ratios of SO₄²⁻ were about 0.93-0.98,
 464 indicating that SO₄²⁻ concentrations were quite uniform from the surface to 975 hPa in
 465 all three types of pollution.

466 **Table 1.** The ratios at 975 and 1005 hPa (gray shaded area in Fig. 8) and at 819 and
 467 975 hPa (red frame in Fig. 8) of NO₃⁻, NH₄⁺, SO₄²⁻, BC, OC, and PM_{2.5} in O₃SPD,
 468 PM_{2.5}SPD, and O₃&PM_{2.5}PD in BTH region.

		NO ₃ ⁻	NH ₄ ⁺	SO ₄ ²⁻	BC	OC	PM _{2.5}
Conc _{819 hPa} / Conc _{975 hPa}	O ₃ SPD	0.95	0.90	0.85	0.73	0.73	0.86
	PM _{2.5} SPD	0.64	0.68	0.81	0.64	0.63	0.67
	O ₃ &PM _{2.5} PD	0.94	0.91	0.87	0.79	0.77	0.89
Conc _{975hPa} / Conc _{1005 hPa}	O ₃ SPD	0.65	0.77	0.96	0.69	0.70	0.74
	PM _{2.5} SPD	0.72	0.76	0.93	0.67	0.65	0.73
	O ₃ &PM _{2.5} PD	0.72	0.80	0.98	0.76	0.73	0.78

469
 470 In the upper layers (975-819 hPa, red rectangle in Fig. 8), the changes in
 471 concentrations of pollutants with altitude in PM_{2.5}SPD were quite different from those
 472 in O₃&PM_{2.5}PD and O₃SPD. The decline of PM_{2.5} from 975 to 819 hPa was slow in
 473 O₃&PM_{2.5}PD (PM_{2.5}(819 hPa) /PM_{2.5}(975 hPa) = 0.89) and O₃SPD (0.86) and fast in

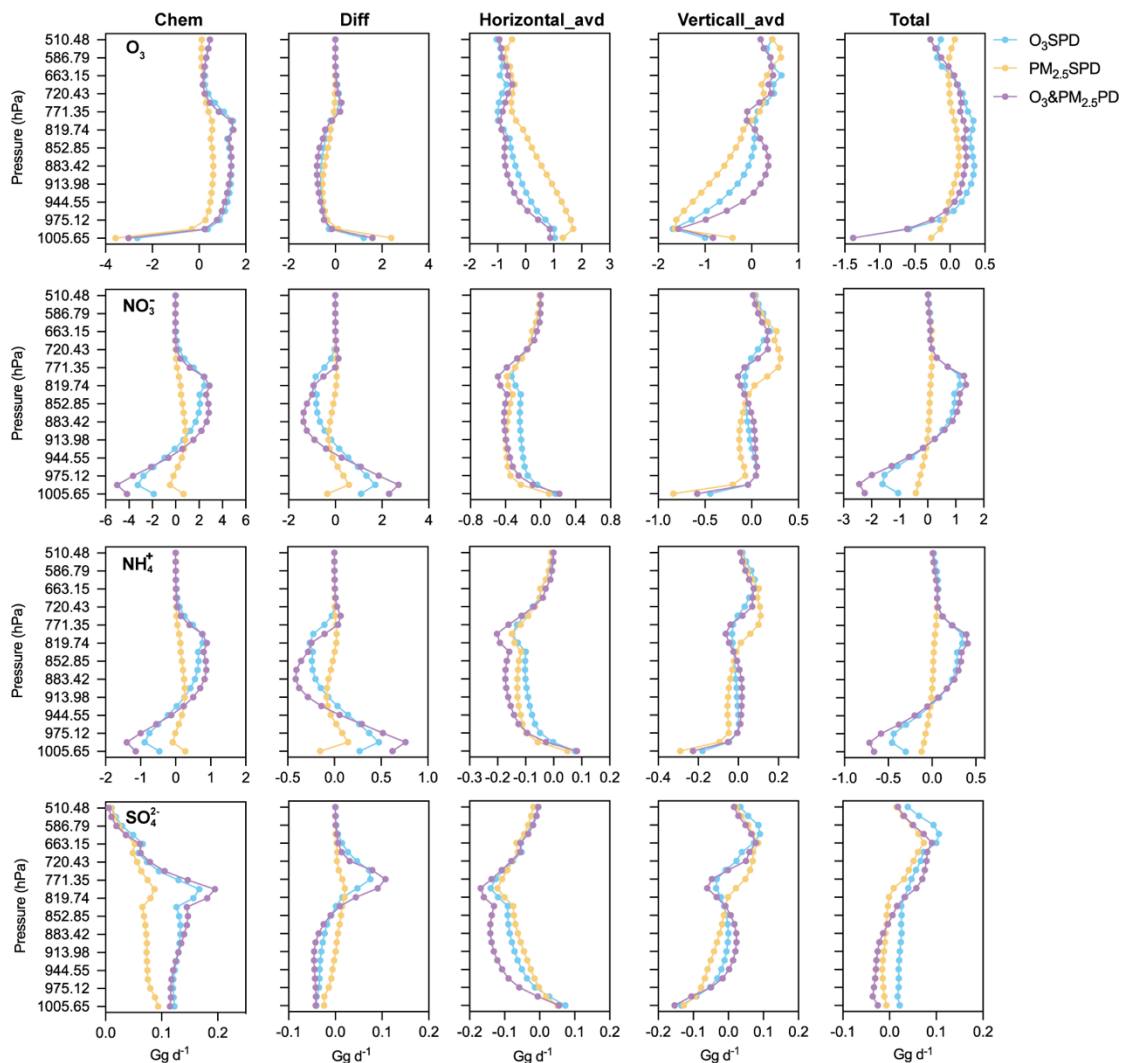
474 PM_{2.5}SPD (0.67). Considering that the variation of BC with altitude was mainly driven
475 by meteorology (Sun et al., 2020), the vertical variations of other components that
476 differed significantly from BC indicated the influences of chemical processes. In
477 PM_{2.5}SPD, NO₃⁻, NH₄⁺, OC had about the same ratio as BC (0.64) (with large decreases
478 with height), except for SO₄²⁻ concentration that had a ratio of 0.81. In O₃&PM_{2.5}PD,
479 the ratios of NO₃⁻, NH₄⁺, SO₄²⁻ were, 0.94, 0.91, 0.87, respectively, which were much
480 higher than the value of BC (0.79), indicating NO₃⁻, NH₄⁺, SO₄²⁻ were quite uniform
481 in the layers of 975-819 hPa with the influence of chemical processes, which will be
482 discussed further in Sect. 3.3.4 below.

483 **3.3.4 Process analyses for O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD**

484 The process analysis (PA) is applied to identify the relative importance of
485 atmospheric processes in the three types of pollution. Figure 9 shows the net changes
486 in O₃, NO₃⁻, NH₄⁺, SO₄²⁻ by the processes of chemical production (Chem), horizontal
487 advection (Horizontal_adv), vertical advection (Vertical_avd), and diffusion (Diff,
488 vertical PBL mixing process) in the GEOS-Chem model, as well as the total of all these
489 processes (i.e., Chem + Diff + Horizontal_adv + Vertical_avd) in O₃SPD, PM_{2.5}SPD,
490 and O₃&PM_{2.5}PD.

491 For O₃, the net changes of O₃ by all processes were positive at altitudes of 975-
492 819 hPa in O₃&PM_{2.5}PD and O₃SPD, in which Chem had the largest positive
493 contribution (about 1.5 Gg d⁻¹), indicating O₃ is chemically produced at these layers.
494 For NO₃⁻ and NH₄⁺, the nets of all processes increased mass concentrations at 913-819
495 hPa in O₃&PM_{2.5}PD and O₃SPD, in which Chem and Vertical_avd were positive and
496 Chem had the largest positive contribution. The vertical profiles of Chem were similar
497 for NO₃⁻ and NH₄⁺, both of which had the largest positive values at 913-819 hPa (2.83
498 Gg d⁻¹ for NO₃⁻ and 0.88 Gg d⁻¹ for NH₄⁺), leading to higher concentrations of NO₃⁻

499 and NH_4^+ in $\text{O}_3\&\text{PM}_{2.5}\text{PD}$ than in O_3SPD and $\text{PM}_{2.5}\text{SPD}$. Chem and Diff of SO_4^{2-}
 500 were different from those of NO_3^- and NH_4^+ . For SO_4^{2-} , Chem was positive from the
 501 surface to 510 hPa with a peak around 819 hPa, and Diff was positive at 819-771 hPa
 502 but negative from 819 hPa to the surface, which resulted in the uniform SO_4^{2-} profile
 503 as shown in Fig. 8. Chem for SO_4^{2-} was the highest around 819 hPa in $\text{O}_3\&\text{PM}_{2.5}\text{PD}$,
 504 which was related to the strong liquid-phase chemical formation of SO_4^{2-} (Fig. S7). In
 505 addition to Chem, Vertical_avd also had positive contributions to the net changes in O_3 ,
 506 NO_3^- , NH_4^+ , and SO_4^{2-} at 944-819 hPa. Vertical_avd was negative at 819 hPa and
 507 positive between 944 to 819 hPa, implying that the pollutants were transported from
 508 819 hPa to 944 hPa in $\text{O}_3\&\text{PM}_{2.5}\text{PD}$.



509

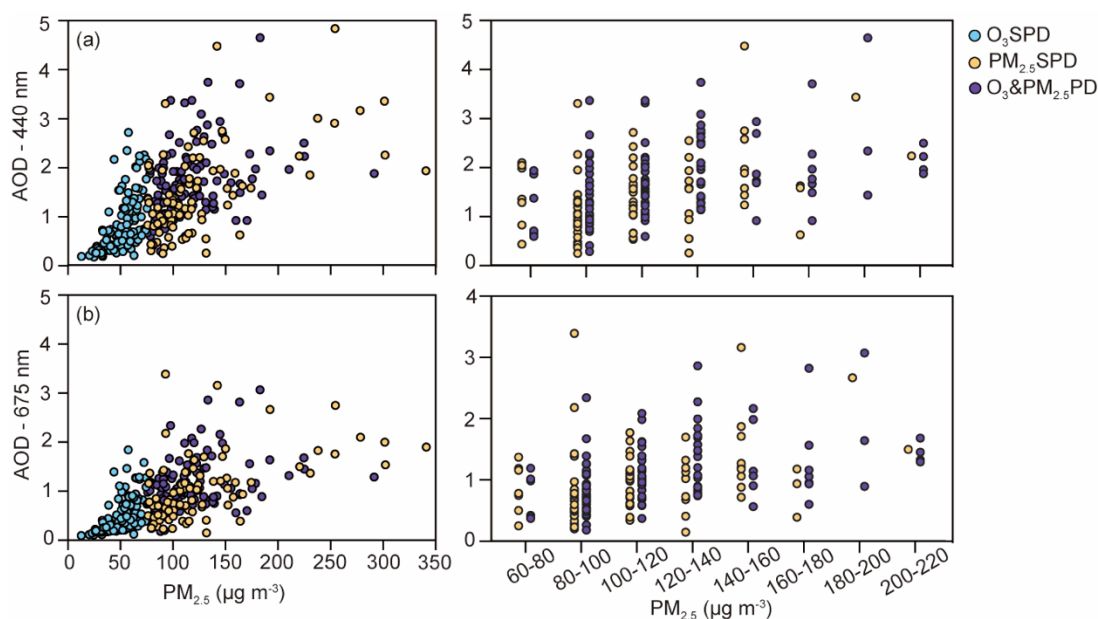
510 **Figure 9.** The vertical profiles of net changes in O_3 , NO_3^- , NH_4^+ , and SO_4^{2-} ($Gg\ d^{-1}$)
511 over BTH by each and total of processes. The values were averaged over the model-
512 captured regional O_3 SPD, $PM_{2.5}$ SPD, and O_3 & $PM_{2.5}$ PD in April-October of 2013-
513 2020.

514 Overall, NO_3^- , NH_4^+ , and SO_4^{2-} all had larger chemical productions at 913-819
515 hPa in O_3 & $PM_{2.5}$ PD compared to those in O_3 SPD and $PM_{2.5}$ SPD, accompanied by
516 strong vertical transport from 819 hPa to near the surface, resulting in the quite uniform
517 vertical profiles at 975-819 hPa in O_3 & $PM_{2.5}$ PD. In addition, the vertical profiles of net
518 changes in $PM_{2.5}$ over BTH are shown in Fig. S8 for these three cases. Since NO_3^- ,
519 NH_4^+ , and SO_4^{2-} were the major components of $PM_{2.5}$, the PA of $PM_{2.5}$ is similar to that
520 of each component.

521 **3.3.5 Observed AOD in O_3 SPD, $PM_{2.5}$ SPD, and O_3 & $PM_{2.5}$ PD**

522 To try to support the model result that O_3 & $PM_{2.5}$ PD had more uniform vertical
523 profile than $PM_{2.5}$ SPD from the surface to 819 hPa altitude, we present the scatter plots
524 of observed AOD (at 440 nm and 675 nm) versus observed $PM_{2.5}$ concentrations in
525 O_3 SPD, $PM_{2.5}$ SPD, and O_3 & $PM_{2.5}$ PD in Fig. 10. AERONET observations of AOD
526 from 2013 to 2020 are available at three sites in BTH (that is, Beijing (39.97°N, 116.38°
527 E), Beijing-CAMS (39.93°N, 116.31°E), Xianghe (39.75°N, 116.96°E)). At Beijing
528 (39.97°N, 116.38°E), AOD (440nm and 675nm) increased with $PM_{2.5}$ concentration in
529 all three types of pollution. However, under the same levels of surface $PM_{2.5}$
530 concentration, AOD values in O_3 & $PM_{2.5}$ PD were higher than in $PM_{2.5}$ SPD, implying
531 that the column burdens of aerosols were generally higher in O_3 & $PM_{2.5}$ PD than in
532 $PM_{2.5}$ SPD, which may support the unique vertical distribution of $PM_{2.5}$ in O_3 & $PM_{2.5}$ PD
533 shown in Fig. 8b. The scatter plots at Beijing-CAMS and Xianghe sites are similar and

534 are shown in Fig. S9.



535

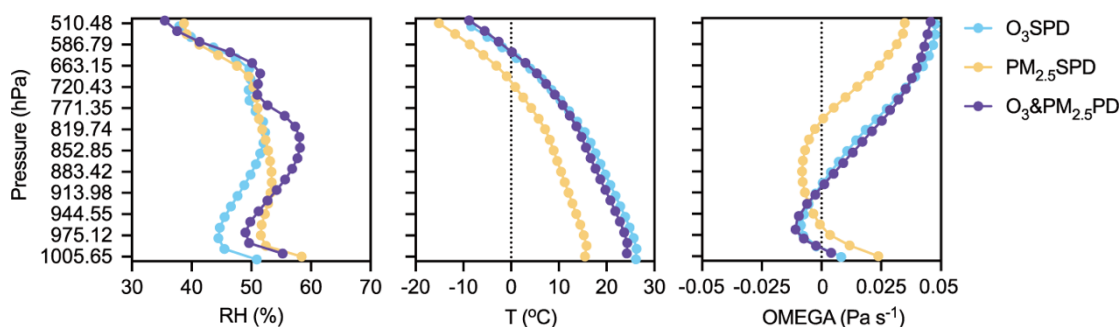
536 **Figure 10.** The scatterplots of (a) AOD (440 nm) and (b) AOD (675 nm) versus
537 observed PM_{2.5} concentrations in O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD in Beijing
538 (39.97°N, 116.38°E) in April-October of 2013-2020.

539

540 3.4 Meteorological conditions for O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD over BTH

541 Figure 11 shows the vertical profiles of RH, T, and OMEGA for O₃SPD, PM_{2.5}SPD,
542 and O₃&PM_{2.5}PD captured by the model over BTH in the months of April to October
543 from 2013-2020. It should be noted that O₃&PM_{2.5}PD had an unique vertical
544 distribution of RH. Near the surface, the values of RH in O₃&PM_{2.5}PD were between
545 those in O₃SPD and PM_{2.5}SPD. However, in the upper layers (883-771 hPa),
546 O₃&PM_{2.5}PD had the highest RH among the three cases with a peak value of 58.2%.
547 As a result, the strongest aqueous chemical production of SO₄²⁻ (aqueous oxidation of
548 SO₂ by H₂O₂) occurred in O₃&PM_{2.5}PD around 819 to 771 hPa (Fig. S7). The vertical
549 profiles of temperature were similar in the three types of pollution, with the lowest
550 temperature in PM_{2.5}SPD. The vertical profiles of OMEGA were different in the three

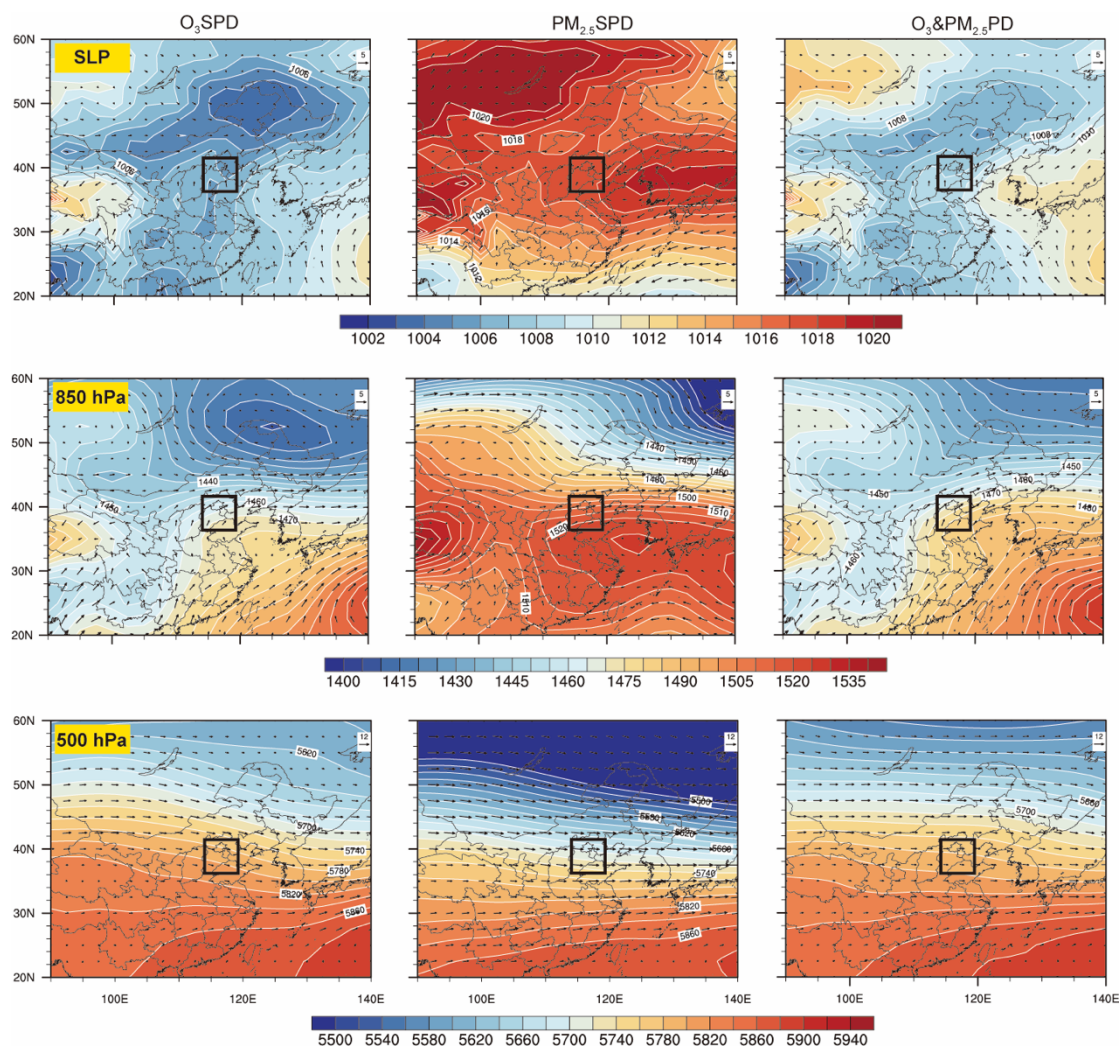
551 cases. In O₃SPD and O₃&PM_{2.5}PD, OMEGA had positive values around 819 hPa,
 552 indicating a strong sinking airflow, leading to a downward transport of pollutants.
 553 Under O₃&PM_{2.5}PD, the average values of PBLH and SWGDN were 946.1 m and
 554 257.2 W m⁻², respectively, which were higher (lower) than those in PM_{2.5}SPD (O₃SPD)
 555 (Fig. S10).



556
 557 **Figure 11.** The vertical profiles of RH (%), T (°C), and OMEGA (Pa s⁻¹) averaged over
 558 BTH and over the model-captured regional O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD in in
 559 April-October of 2013-2020.

560 Figure 12 shows the composited weather patterns for regional O₃SPD, PM_{2.5}SPD,
 561 and O₃&PM_{2.5}PD (over 50% cities in the BTH experienced the pollution) that were
 562 captured by the model in April-October of 2013-2020. The weather patterns of
 563 O₃&PM_{2.5}PD were similar to some extent to those of O₃SPD but were quite different
 564 from those of PM_{2.5}SPD. In O₃&PM_{2.5}PD, the BTH region was controlled by westerlies
 565 and an anomalous high-pressure system at 500 hPa (Figure S11). At 850 hPa, BTH was
 566 at the west boundary of an anomalous anticyclone, and the associated strong anomalous
 567 southerlies at 850 hPa brought moist air to BTH (Fig. S12 and S13), resulting in a high
 568 RH that was beneficial to the aqueous chemical production of SO₄²⁻ in O₃&PM_{2.5}PD.
 569 In O₃SPD, BTH was under the influence of the high pressure ridge of the Western
 570 Pacific Subtropical High (WPSH) at 850 hPa. Besides, the Northeast Cold Vortex was
 571 located to the southwest of BTH at 850 hPa in O₃SPD, leading to dry and warm
 572 conditions, which was favorable for the formation of O₃. In PM_{2.5}SPD, the BTH region

573 was under the influence of both the continental high and the WPSH at 850 hPa. At the
 574 surface, BTH was under the influence of a uniform high pressure with very weak winds
 575 and hence stagnate atmosphere, which was conducive to the accumulation of PM_{2.5}.



576
 577 **Figure 12.** Composites of wind field (m s^{-1}) with SLP (sea level pressure) and with
 578 geopotential height at 850 hPa and 500 hPa for regional O₃SPD, PM_{2.5}SPD, and
 579 O₃&PM_{2.5}PD that were captured by the model in April-October of 2013-2020. The
 580 solid black rectangle indicates BTH region.

581

582 4. Conclusions

583 We used the observed hourly concentrations of O₃ and PM_{2.5} from CNEMC and
 584 the model results from the nested-grid version of the GEOS-Chem model to examine

585 the chemical and physical characteristics of the co-polluted days by O₃ and PM_{2.5}
586 (O₃&PM_{2.5}PD) over the BTH region for eight warm seasons (April-October) from
587 2013 to 2020. The characteristic of O₃&PM_{2.5}PD were compared with those of the
588 polluted days by O₃ alone (O₃SPD) and by PM_{2.5} alone (PM_{2.5}SPD). In April-October
589 of 2013-2020, the observed O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD were 2954, 2148,
590 and 1614 days, respectively, in which 75.0% (2954/3937), 58.1% (2148/3698), and
591 79.7% (1614/2024) were captured by the GEOS-Chem model, respectively. We
592 carried out composited analyses of the chemical and physical characteristics for
593 O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD by using the samples (days) captured by both
594 the observations and the model.

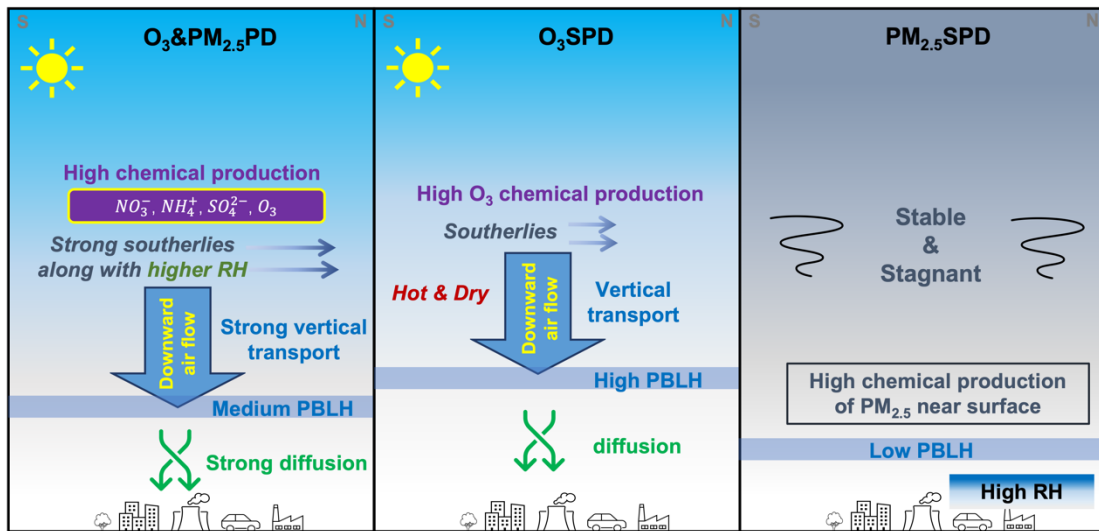
595 The chemical characteristics of O₃&PM_{2.5}PD were found to be different from
596 those of O₃SPD, PM_{2.5}SPD at the surface. O₃&PM_{2.5}PD occurred with high levels of
597 atmospheric oxidants (high OH and O_x), with higher SOR and NOR compared to those
598 in O₃SPD and PM_{2.5}SPD, leading to high concentrations of both O₃ and PM_{2.5}. At the
599 surface, the composited concentrations of NO₃⁻, NH₄⁺, BC, and OC were the highest in
600 PM_{2.5}SPD, while the composited concentration of SO₄²⁻ was the highest in
601 O₃&PM_{2.5}PD. There was a strong formation of SO₄²⁻ during the daytime in
602 O₃&PM_{2.5}PD in the oxidative atmosphere.

603 We also found unique features of the vertical distributions of O₃ and PM_{2.5} in
604 O₃&PM_{2.5}PD. Concentrations of PM_{2.5} were stable and high between 975 and 819 hPa
605 in O₃&PM_{2.5}PD, unlike those in PM_{2.5}SPD that decreased rapidly with the altitude. In
606 O₃&PM_{2.5}PD, the vertical profiles of NO₃⁻, NH₄⁺, and SO₄²⁻ were quite uniform at
607 975-819 hPa, corresponding to the stable O₃ concentrations at these altitudes. The
608 process analysis (PA) showed that NO₃⁻, NH₄⁺, and SO₄²⁻ all had larger chemical
609 productions at altitudes of 913-819 hPa in O₃&PM_{2.5}PD compared to those in O₃SPD

610 and PM_{2.5}SPD. The chemical production of SO₄²⁻ had large positive values from the
611 surface to about 500 hPa. The Vertical_avd also had positive contributions to the net
612 changes in O₃, NO₃⁻, NH₄⁺, and SO₄²⁻ at 944-819 hPa in O₃&PM_{2.5}PD. Therefore, the
613 strong chemical productions at 913-819 hPa accompanied by the downward transport
614 resulted in the quite uniform vertical profiles at 975-819 hPa in O₃&PM_{2.5}PD.

615 Figure 13 summarizes the chemical and physical characteristics in O₃&PM_{2.5}PD,
616 O₃SPD, and PM_{2.5}SPD in the BTH region. In O₃&PM_{2.5}PD, the strong chemical
617 productions of O₃, NO₃⁻, NH₄⁺, and SO₄²⁻ occurred at high altitudes of 913-819 hPa
618 where RH was high, and the accompanied downward airflow caused the stable
619 concentrations at 944-819 hPa. The composited PBLH in O₃&PM_{2.5}PD was about 946.1
620 m, and the strong mixed diffusion underneath the PBLH led to high concentrations of
621 pollutants at the ground level. In contrast, O₃SPD occurred in hot and dry atmosphere
622 with composited PBLH of 1073.5 m. Strong O₃ chemical production occurred around
623 819 hPa, and O₃ was then transported to the surface by downward air flow. The
624 atmosphere was stable and stagnate when PM_{2.5}SPD occurred, with the lowest PBLH
625 of 681.8 m. High RH (high chemical formation of PM_{2.5}) and the accumulation of
626 aerosols led to the highest surface-layer PM_{2.5} in PM_{2.5}SPD.

627 To summarize, O₃&PM_{2.5}PD were characterized by high O_x, SOR, and NOR,
628 uniform vertical profiles at 975-819 hPa, which were caused by an anomalous high-
629 pressure system at 500 hPa, strong southerlies and high RH at 850 hPa. Meteorological
630 parameters around 850 hPa promoted strong chemical production of secondary aerosols
631 and downward transport, resulting in the unique vertical profiles and high surface
632 concentrations in O₃&PM_{2.5}PD.



633

634 **Figure 13.** A schematic diagram of chemical and physical and characteristics in

635 O₃SPD, PM_{2.5}SPD, and O₃&PM_{2.5}PD in BTH region.

636 ***Data availability.***

637 The GEOS-Chem model is available at <https://geos-chem.seas.harvard.edu> (last
638 access: 5 August 2022). The observed hourly surface concentrations of air pollutants
639 are derived from the China National Environmental Monitoring Center
640 (<https://air.cnemc.cn:18007/>, CNEMC, 2022). The simulation results are available
641 upon request from the corresponding author (hongliao@nuist.edu.cn).

642

643 ***Author contributions.***

644 HD and HL conceived the study and designed the experiments. HD performed model
645 simulations and analysed the data. KL, XY, YY, JZ, JJ, and BL provided useful
646 comments on the paper. HD and HL prepared the paper, with contributions from all co-
647 authors.

648

649 ***Competing interests.***

650 The authors declare that they have no conflict of interest.

651

652 ***Acknowledgements.***

653 We acknowledge the CNEMC for making their data publicly available. We
654 acknowledge the efforts of GEOS-Chem working groups for developing and
655 managing the model.

656

657 ***Financial support.***

658 This work was supported by the National Natural Science Foundation of China (Grant
659 No. 42021004), the National Key Research and Development Program of China
660 (Grant No. 2019YFA0606800), and the Carbon Peak Carbon Neutral Science and
661 Technology Innovation Foundation of Jiangsu Province (BK20220031).

662

663 **References**

- 664 Alexander, B., Park, R. J., Jacob, D. J., Li, Q. B., Yantosca, R. M., Savarino, J., Lee,
665 C. C. W., and Thiemens, M. H.: Sulfate formation in sea-salt aerosols: Constraints
666 from oxygen isotopes, *J. Geophys. Res.-Atmos.*, 110, D10307,
667 <https://doi.org/10.1029/2004jd005659>, 2005.
- 668 Bey, I., Jacob, D. J., Yantosca, R. M., Logan, J. A., Field, B. D., Fiore, A. M., Li, Q. B.,
669 Liu, H. G. Y., Mickley, L. J., and Schultz, M. G.: Global modeling of tropospheric
670 chemistry with assimilated meteorology: Model description and evaluation, *J.*
671 *Geophys. Res.-Atmos.*, 106, 23073–23095, <https://doi.org/10.1029/2001jd000807>,
672 2001.
- 673 Chu, B. W., Ma, Q. X., Liu, J., Ma, J. Z., Zhang, P., Chen, T. A., Feng, Q. C., Wang. C.
674 Y., Yang, N., Ma, H. N., Ma, J. J., Russell, A. G., He, H.: Air Pollutant Correlations
675 in China: Secondary Air Pollutant Responses to NO_x and SO₂ Control, *Environ.*
676 *Sci. Tech. Let.*, 7(10), 695-700, <http://dx.doi.org/10.1021/acs.estlett.0c00403>,
677 2020.
- 678 Dai, H. B., Zhu, J., Liao, H., Li, J. D., Liang, M. X., Yang, Y., Yue, X.: Co-occurrence
679 of ozone and PM_{2.5} pollution in the Yangtze River Delta over 2013-2019:
680 spatiotemporal distribution and meteorological conditions, *Atmos. Res.*, 249,
681 105363, <https://doi.org/10.1016/j.atmosres.2020.105363>, 2021.
- 682 Dang, R. J. and Liao, H.: Severe winter haze days in the Beijing-Tianjin-Hebei region
683 from 1985-2017 and the roles of anthropogenic emissions and meteorology, *Atmos.*
684 *Chem. Phys.*, 19, 10801-10816, <https://doi.org/10.5194/acp-19-10801-2019>, 2019.

685 Dang, R. J., Liao, H., and Fu, Y.: Quantifying the anthropogenic and meteorological
686 influences on summertime surface ozone in China over 2012-2017, *Sci. Total.*
687 *Environ.*, 754, 142394, <https://doi:10.1016/j.scitotenv.2020.142394>, 2021.

688 Duan, F., He, K., Ma, Y., Yang, F., Yu, X., Cadle, S. H., Chan, T., and Mulawa, P. A.:
689 Concentration and chemical characteristics of PM_{2.5} in Beijing, China: 2001–
690 2002, *Sci. Total Environ.*, 355(1–3), 264–275,
691 <https://doi:10.1016/j.scitotenv.2005.03.001>, 2006.

692 Fairlie, T. D., Jacob, D. J., and Park, R. J.: The impact of transpacific transport of
693 mineral dust in the United States, *Atmos. Environ.*, 41, 1251–1266,
694 <https://doi.org/10.1016/j.atmosenv.2006.09.048>, 2007.

695 Gao, Y. and Ji, H. B.: Microscopic morphology and seasonal variation of health effect
696 arising from heavy metals in PM_{2.5} and PM₁₀: One-year measurement in a densely
697 populated area of urban Beijing, *Atmos. Res.*, 212, 213–226, [https://doi.org/](https://doi.org/10.1016/j.atmosres.2018.04.027)
698 [10.1016/j.atmosres.2018.04.027](https://doi.org/10.1016/j.atmosres.2018.04.027), 2018.

699 Giles, D. M., Sinyuk, A., Sorokin, M. S., Schafer, J. S., Lyapustin, A.: Advancements
700 in the aerosol robotic network (aeronet) version 3 database – automated near real-
701 time quality control algorithm with improved cloud screening for sun photometer
702 aerosol optical depth (aod) measurements, *Atmos. Meas. Tech.*, 12, 169–209,
703 <https://doi.org/10.5194/amt-12-169-2019>. 2019.

704 Gonçalves, M., Jiménez-Guerrero, P., and Baldasano, J. M.: Contribution of
705 atmospheric processes affecting the dynamics of air pollution in South-Western
706 Europe during a typical summer-time photochemical episode, *Atmos. Chem. Phys.*,

707 9, 849–864, <https://doi.org/10.5194/acp-9-849-2009>, 2009.

708 Gong, C. and Liao, H.: A typical weather pattern for the ozone pollution events in North
709 China, *Atmos. Chem. Phys.*, 19, 13725–13740, [https://doi.org/10.5194/acp-19-](https://doi.org/10.5194/acp-19-13725-2019)
710 13725-2019, 2019.

711 Gong, C., Liao, H., Zhang, L., Yue, X., Dang, R. J., Yang, Y.: Persistent ozone pollution
712 episodes in North China exacerbated by regional transport, *Environ. Pollut.*, 265,
713 115056, <https://doi:10.1016/j.envpol.2020.115056>, 2020.

714 Guenther, A. B., Jiang, X., Heald, C. L., Sakulyanontvittaya, T., Duhl, T., Emmons, L.
715 K., and Wang, X.: The Model of Emissions of Gases and Aerosols from Nature
716 version 2.1 (MEGAN2.1): an extended and updated framework for modeling
717 biogenic emissions, *Geosci. Model Dev.*, 5, 1471–1492,
718 <https://doi.org/10.5194/gmd-5-1471-2012>, 2012.

719 Jiang, N., Li, L., Wang, S., Li, Q., Dong, Z., Duan, S., Zhang, R., Li, S.: Variation
720 tendency of pollution characterization, sources, and health risks of PM_{2.5}-bound
721 polycyclic aromatic hydrocarbons in an emerging megacity in China: based on
722 three-year data, *Atmos. Res.*, 217, 81–92, [https://doi.org/10.1016/j.](https://doi.org/10.1016/j.atmosres.2018.10.023)
723 [atmosres.2018.10.023](https://doi.org/10.1016/j.atmosres.2018.10.023), 2019.

724 Li, K., Jacob, D. J., Liao, H., Zhu, J., Shah, V., Shen, L., Bates, K., Zhang, Q., Zhai, S.
725 X.: A two-pollutant strategy for improving ozone and particulate air quality in
726 China, *Nat. Geosci.*, 12, 906–910, <https://doi.org/10.1038/s41561-019-0464-x>,
727 2019.

728 Li, K., Jacob, D. J., Shen, L., Lu, X., Smedt, D. I., Liao, H.: Increases in surface ozone

729 pollution in China from 2013 to 2019: anthropogenic and meteorological
730 influences, *Atmos. Chem. Phys.*, 20, 11423-11433, [https://doi.org/10.5194/acp-](https://doi.org/10.5194/acp-20-11423-2020)
731 20-11423-2020, 2020.

732 Li, M., Wang, L., Liu, J.: Exploring the regional pollution characteristics and
733 meteorological formation mechanism of PM_{2.5} in North China during 2013-2017,
734 *Environ. Int.*, 134, 105283, <https://doi.org/10.1016/j.envint.2019.105283>, 2019.

735 Li, M., Zhang, Q., Kurokawa, J.-I., Woo, J.-H., He, K., Lu, Z., Ohara, T., Song, Y.,
736 Streets, D. G., Carmichael, G. R., Cheng, Y., Hong, C., Huo, H., Jiang, X., Kang,
737 S., Liu, F., Su, H., and Zheng, B.: MIX: a mosaic Asian anthropogenic emission
738 inventory under the international collaboration framework of the MICS-Asia and
739 HTAP, *Atmos. Chem. Phys.*, 17, 935–963, [https://doi.org/10.5194/acp-17-935-](https://doi.org/10.5194/acp-17-935-2017)
740 2017, 2017.

741 Liu, Y. X., Zhao, Q. B., Hao, X., Zhao, J. R., Zhang, Y., Yang, X., Fu, Q. Y., Xu, X. Y.,
742 Wang, X. F., Huo, J. T., Chen, J. M.: Increasing surface ozone and enhanced
743 secondary organic carbon formation at a city junction site: An epitome of the
744 Yangtze River Delta, China (2014–2017), *Environ. Pollut.*, 265, 0269-7491,
745 <https://doi.org/10.1016/j.envpol.2020.114847>, 2020.

746 Lou, S. J., Liao, H., Yang, Y., and Mu, Q.: Simulation of the interannual variations of
747 tropospheric ozone over China: Roles of variations in meteorological parameters
748 and anthropogenic emissions, *Atmos. Environ.*, 122, 839–851,
749 <https://doi.org/10.1016/j.atmosenv.2015.08.081>, 2015.

750 Molod, A., Takacs, L., Suarez, M., and Bacmeister, J.: Development of the GEOS-5

751 atmospheric general circulation model: evolution from MERRA to MERRA2,
752 Geosci. Model Dev., 8, 1339–1356, <https://doi.org/10.5194/gmd-8-1339-2015>,
753 2015.

754 Mu, Q. and Liao, H.: Simulation of the interannual varia- tions of aerosols in China:
755 role of variations in meteo- rological parameters, Atmos. Chem. Phys., 14, 9597–
756 9612, <https://doi.org/10.5194/acp-14-9597-2014>, 2014.

757 Nan, J. L., Wang, S. S., Guo, Y. L., Xiang, Y. J., Zhou. B.: Study on the daytime OH
758 radical and implication for its relationship with fine particles over megacity of
759 Shanghai, China, Atmos. Environ., 154, 167-178,
760 <https://doi.org/10.1016/j.atmosenv.2017.01.046>, 2017.

761 Ni, R., Lin, J., Yan, Y., and Lin, W.: Foreign and domestic contributions to springtime
762 ozone over China, Atmos. Chem. Phys., 18, 11447–11469,
763 <https://doi.org/10.5194/acp-18-11447-2018>, 2018.

764 Park, R. J., Jacob, D. J., Chin, M., and Martin, R. V.: Sources of carbonaceous aerosols
765 over the United States and implica- tions for natural visibility, J. Geophys. Res.-
766 Atmos., 108, 4355, <https://doi.org/10.1029/2002jd003190>, 2003.

767 Park, R. J., Jacob, D. J., Field, B. D., Yantosca, R. M., and Chin, M.: Natural and
768 transboundary pollution influences on sulfate-nitrate-ammonium aerosols in the
769 United States: Im- plications for policy, J. Geophys. Res.-Atmos., 109, D15204,
770 <https://doi.org/10.1029/2003jd004473>, 2004.

771 Pye, H. O. T., Liao, H., Wu, S., Mickley, L. J., Jacob, D. J., Henze, D. K., and Seinfeld,
772 J. H.: Effect of changes in climate and emissions on future sulfate-nitrate-

773 ammonium aerosol lev- els in the United States, *J. Geophys. Res.-Atmos.*, 114,
774 D01205, <https://doi.org/10.1029/2008jd010701>, 2009.

775 Qin, Y., Li, J. Y., Gong, K. J., Wu, Z., Chen, M. D., Qin, M. M., Huang, L., Hu, J. L.:
776 Double high pollution events in the Yangtze River Delta from 2015 to 2019:
777 Characteristics, trends, and meteorological situations, *Sci. Total. Environ.*, 792,
778 148349, <https://doi.org/10.1016/j.scitotenv.2021.148349>, 2021.

779 Ren, W., Tian, H., Tao, B., Chappelka, A., Sun, G., Lu, C., Liu, M., Chen, G., Xu, X.:
780 Impacts of tropospheric ozone and climate change on net primary productivity and
781 net carbon exchange of China's forest ecosystems, *Glob. Ecol. Biogeogr.*, 20, 391–
782 406, <https://doi.org/10.1111/j.1466-8238.2010.00606.x>, 2011.

783 Sun, T., Wu, C. and Wu, D.: Time-resolved black carbon aerosol vertical distribution
784 measurements using a 356-m meteorological tower in Shenzhen, *Theor. Appl.*
785 *Climatol.*, 140, 1263–1276, <https://doi.org/10.1007/s00704-020-03168-6>, 2020.

786 Tan, Z. F., Fuchs, H., and Lu, K. D.: Radical chemistry at a rural site (Wangdu) in the
787 North China Plain: Observation and model calculations of OH, HO₂ and RO₂
788 radicals, *Atmos. Chem. Phys.*, 17(1): 663–690, [https://doi.org/10.5194/acp-17-](https://doi.org/10.5194/acp-17-663-2017)
789 [663-2017](https://doi.org/10.5194/acp-17-663-2017), 2017.

790 Wang, H., Kiang, C., Tang, X., Zhou, X., Chameides, W. L.: Surface ozone: a likely
791 threat to crops in Yangtze delta of China, *Atmos. Environ.*, 39, 3843–3850,
792 <https://doi.org/10.1016/j.atmosenv.2005.02.057>, 2005.

793 Wang, X., Manning, W., Feng, Z., Zhu, Y.: Ground-level ozone in China: distribution
794 and effects on crop yields, *Environ. Pollut.*, 147 (2), 394–400, [https://](https://doi.org/10.1016/j.envpol.2013.08.011)

795 doi.org/10.1016/j.envpol.2006.05.006, 2007.

796 Woodward-Massey, R., Slater, E. J., Alen, J.: Implementation of a chemical background
797 method for atmospheric OH measurements by laser-induced fluorescence:
798 characterisation and observations from the UK and China, *Atmos. Meas. Tech.*,
799 13(6): 3119–3146, <https://doi.org/10.5194/amt-13-3119-2020>, 2020.

800 Xuan, J., Liu, G., and Du, K.: Dust emission inventory in northern China, *Atmos.*
801 *Environ.*, 34(26), 4565–4570, [https://doi:10.1016/S1352-2310\(00\)00203-X](https://doi:10.1016/S1352-2310(00)00203-X), 2000.

802 Ye, B., Ji, X., Yang, H., Yao, X., Chan, C. K., Cadle, S. H., Chan, T., and Mulawa, P.
803 A.: Concentration and chemical composition of PM_{2.5} in Shanghai for a 1-year
804 period, *Atmos. Environ.*, 37(4), 499–510, [https://doi:10.1016/S1352-](https://doi:10.1016/S1352-2310(02)00918-4)
805 [2310\(02\)00918-4](https://doi:10.1016/S1352-2310(02)00918-4), 2003.

806 Yue, X., Unger, N., Harper, K., Xia, X., Liao, H., Zhu, T., Xiao, J., Feng, Z., Li, J.:
807 Ozone and haze pollution weakens net primary productivity in China, *Atmos.*
808 *Chem. Phys.*, 17, 6073–6089, <https://doi.org/10.5194/acp-2016-1025>, 2017.

809 Zhang, Y. and Wang, Y.: Climate-driven ground-level ozone extreme in the fall over the
810 Southeast United States, *P. Natl. Acad. Sci. USA*, 113, 10025–10030,
811 <https://doi.org/10.1073/pnas.1602563113>, 2016.

812 Zhao, X. J., Zhao, P. S., Xu, J., Meng, W., Pu, W. W., Dong, F., He, D., and Shi, Q. F.:
813 Analysis of a winter regional haze event and its formation mechanism in the North
814 China Plain, *Atmos. Chem. Phys.*, 13, 5685–5696, [https://doi:10.5194/acp-13-](https://doi:10.5194/acp-13-5685-2013)
815 [5685-2013](https://doi:10.5194/acp-13-5685-2013), 2013.

816 Zheng, B., Tong, D., Li, M., Liu, F., Hong, C., Geng, G., Li, H., Li, X., Peng, L., Qi, J.,

817 Yan, L., Zhang, Y., Zhao, H., Zheng, Y., He, K., and Zhang, Q.: Trends in China's
818 anthropogenic emissions since 2010 as the consequence of clean air actions,
819 Atmos. Chem. Phys., 18, 14095–14111, [https://doi.org/10.5194/acp-18-14095-](https://doi.org/10.5194/acp-18-14095-2018)
820 2018, 2018.

821 Zheng, B., Zhang, Q., Geng, G., Chen, C., Shi, Q., Cui, M., Lei, Y., He, K.: Changes in
822 China's anthropogenic emissions and air quality during the COVID-19 pandemic
823 in 2020, Earth Syst. Sci. Data, 13, 2895–2907, [https://doi.org/10.5194/essd-13-](https://doi.org/10.5194/essd-13-2895-2021)
824 2895-2021, 2021.

825 Zhu, J., Chen, L., Liao, H., Dang, R. J.: Correlations between PM_{2.5} and ozone over
826 China and associated underlying reasons, Atmosphere, 10(7), 352,
827 <https://doi.org/10.3390/atmos10070352>, 2019.

828 Zong, L., Yang, Y., and Gao, M.: Large-scale synoptic drivers of co-occurring
829 summertime ozone and PM_{2.5} pollution in eastern China, Atmos. Chem. Phys., 11,
830 9105-9124, <https://doi.org/10.5194/acp-21-9105-2021>, 2021.