

Simulations of winter ozone in the Upper Green River Basin, Wyoming, using WRF-Chem

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Abstract.

In both the Upper Green River Basin (UGRB) of Wyoming and the Uintah Basin of Utah, strong wintertime ozone (O₃) formation episodes leading to O₃ ~~concentrations~~ mixing ratios occasionally exceeding 70 ~~ppb~~ parts per billion (ppb) have been observed over the last two decades. Wintertime O₃ events in the UGRB were first observed in 2005 and since then have continued to be observed intermittently when meteorological conditions are favorable, despite significant efforts to reduce emissions. While O₃ formation has been successfully simulated using observed volatile organic compound (VOC) and nitrogen oxide (NO_x) ~~concentrations~~ mixing ratios, successful simulation of these wintertime episodes using emission inventories in a 3-D photochemical model has remained elusive. An accurate 3-D photochemical model driven by an emission inventory is critical to understand the spatial extent of high O₃ events and which emission sources have the most impact on O₃ formation.

10 In the winter of 2016-2017 (December 2016 - March 2017) several high O₃ events were ~~recorded~~ observed with 1-hour ~~concentrations~~ mixing ratios exceeding 70 ppb. This study uses the Weather Research Forecasting model with chemistry (WRF-Chem) to simulate one of the high O₃ events observed in the UGRB during March of 2017. The WRF-Chem simulations were carried out using the 2014 edition of the Environmental Protection Agency National Emissions Inventory (EPA-NEI 2014v2), which, unlike previous versions, includes estimates of emissions from non-point oil and gas production sources. Simulations

15 were carried out with two different chemical mechanisms: the Model for Ozone and Related Chemical Tracers (MOZART) and the Regional Atmospheric Chemistry Mechanism (RACM), and the results were compared with data from 7 weather and air quality monitoring stations in the UGRB operated by Wyoming Department of Environmental Quality (WYDEQ). The simulated meteorology compared favorably to observations in terms of predicting with regard to temperature inversions, surface temperature, and wind speeds. Notably, because of snow cover present in the basin, the photolysis surface albedo had to be modified to form predict O₃ exceeding in excess of 70 ppb, though although the models were relatively insensitive to the exact photolysis albedo if it was over 0.65. O₃ precursors NO_x and NMHC, i.e., NO_x and volatile organic compounds (VOCs), are predicted similarly in simulations with both chemistry mechanisms, but simulated NMHC-VOC mixing ratios are a factor of six or more lower than the observations, while NO_x ~~mixing ratios are also underpredicted~~. Sensitivity studies show that increasing VOC and NO_x is also underpredicted but to a lesser degree. Sensitivity simulations revealed that increasing NO_x

25 and VOC emissions to match ~~observed mixing ratios had a relatively small effect on the ozone produced~~. However, further

~~increases in NO_x created much more ozone, suggesting ozone formation~~ observations produced slightly more O₃ compared to baseline runs, but an additional sensitivity simulation with doubled NO_x emissions resulted in a considerable increase in O₃ formation. These results suggest that O₃ formation in the basin is ~~very~~ most sensitive to NO_x emissions.

1 Introduction

30 Tropospheric ozone (O_3) is a secondary pollutant that is harmful to human health, plants, and other animals when at elevated levels (Fuhrer et al., 1997; Ebi and McGregor, 2008). The current 2015 US National Ambient Air Quality Standard (NAAQS) for the 8-h-8-hour average O_3 mixing ratio is 70 parts per ~~billions~~ billion (ppb). As of August 14, 2020, the 2015 NAAQS standard for the 8-h-8-hour average O_3 mixing ratio has been proposed to be retained (EPA, 2020). ~~Any hourly occurrence~~ While below the NAAQS threshold, all hourly occurrences of O_3 ~~concentration greater mixing ratios greater than~~ or equal to the NAAQS standard is 70 ppb are referred to as ~~an~~ O_3 ~~event~~ events throughout this paper. In the past decades, there has been a significant increase in wintertime as well as summertime O_3 events in the western US (Cooper et al., 2012).

According to the US Energy Information Administration (EIA), in 2018, Wyoming was the 8th largest producer of oil and natural gas in the United States, with a majority of the natural gas production coming from the Upper Green River Basin (UGRB). Specifically, the UGRB accounts for 60% of the state's natural gas production and 16% of its oil produc-
40 tion (Wyoming State Geological Survey; WSGS, 2020). As of 2017, there were 5506 total wells (5436 producing wells) in the Jonah and Pinedale fields ~~that constitute within~~ the UGRB, a 5.7% increase in the total and 5.9% increase in the producing wells in the UGRB compared with those in 2016 (<http://pipeline.wyo.gov/FieldReportYear.cfm>). By September 2020, there ~~had been was an~~ 8.8% ~~of~~ increase in the total wells since 2017 and a 14.6% increase in oil and gas producing wells in the UGRB.

45 The formation of O_3 has traditionally been an urban summertime phenomena because of the need for strong solar intensity and sufficient volatile organic compounds (VOCs). Elevated ~~concentrations mixing ratios~~ of wintertime O_3 in a few rural US basins have been associated with the rapid development of natural gas and oil production fields (Mansfield and Hall, 2013; Edwards et al., 2014; Ahmadov et al., 2015; Field et al., 2015a, b). Such elevated O_3 events can occur in winter under specific meteorological conditions: a snow-covered ground that provides high albedo that increases solar intensity while also preventing
50 solar heating of the ground (Carter and Seinfeld, 2012) and weak/calm winds. Combined, these conditions result in a persistent temperature inversion and little horizontal/vertical transport, which provides the conditions needed for the photochemical production and build up of O_3 (Mansfield and Hall, 2018).

Several studies have been carried out to understand the meteorological and chemical processes leading to high wintertime O_3 events in western US oil and gas basins. These studies have focused on observational measurements (Schnell et al., 2009; Olt-
55 mans et al., 2014b; Rappenglück et al., 2014; Field et al., 2015b; Lyman and Tran, 2015), aircraft measurements (Oltmans et al., 2014a), statistical models (Mansfield and Hall, 2013), box models (Carter and Seinfeld, 2012; Edwards et al., 2013, 2014), and 3-D photochemical models (Rodriguez et al., 2009; Ahmadov et al., 2015; Matichuk et al., 2017). Most of these studies have ~~been carried out in focused on~~ the UGRB and Utah's Uintah Basin (UB), and both basins have been identified as regions exceeding the NAAQS (Lyman and Tran, 2015). These studies have shown the principal role played by emissions from oil
60 and natural gas production fields in the formation of wintertime O_3 . However, the assessment of wintertime O_3 formation in these regions poses serious challenges because each basin has complex topography and meteorological conditions along with poorly constrained precursor (VOC and nitrogen oxide (NO_x)) ~~emission~~ emissions. One shortfall of all previous studies is

that most of them have not utilized an existing emission inventory to model O₃ formation. Rather, these studies have utilized observed atmospheric concentrations-levels of precursors to model O₃ formation, thus making it difficult to assess how future expansion of production or various emission reductions will affect O₃ formation.

Schnell et al. (2009) summarized the confluence of three major factors for wintertime O₃ formation: (i) the extensive production of oil and natural gas that releases NO_x and VOCs or hydrocarbons (HCs) into the atmosphere, (ii) calm wind conditions, and (iii) high albedo caused by snow accumulation at the surface that leads to a strong temperature inversion. A strong inversion traps O₃ and its precursors near the ground; if the inversion persists for several days, the concentrations-mixing ratios of O₃ and its precursors increase. The high surface albedo also provides additional shortwave radiation for photochemistry compared to a dry-snow-free landscape.

Some studies have specifically pointed out the importance of deep snow cover or high surface albedo in the formation of wintertime O₃. Oltmans et al. (2014b) and Rappenglück et al. (2014) noted that in March 2011, the UGRB experienced high hourly O₃ concentrations-mixing ratios exceeding 150 ppb, which was associated with the deepest snow cover of the season. In addition, Oltmans et al. (2014b) also pointed out that for the period with snow coverage on the ground, the sum of incoming and reflected ultraviolet levels was almost 80% higher than the period with no snow cover, addressing-highlighting the impact of fresh snow accumulation during high O₃ events. Rappenglück et al. (2014) noted a significant increase in the background O₃ concentration-mixing ratio from around 40 ppb in January to 60 ppb in March 2011, owing to the changes in the meteorological conditions and chemical processes each-month-that-change-the-pollutant-concentration-that-ultimately-affect-pollutant-levels.

Numerous measurement studies have pointed out the important roles played by topography and both meteorological and chemical processes in the UGRB, leading to different O₃ and precursor concentrations-mixing ratios within each basin and from year to year. Field et al. (2015b) carried out air quality measurements in the UGRB for two consecutive winters (2011 and 2012) at a site located 5 km southeast of a Wyoming Department of Environmental Quality (WYDEQ) air quality and weather monitoring station (Boulder). They measured O₃, reactive nitrogen compounds, methane (CH₄), total non-methane hydrocarbon (NMHC), carbon monoxide (CO), and other standard meteorological parameters. The lower-concentration-of-observed-O₃-ambient-NMHC-mixing-ratios in 2012 were associated-with-lower-NMHC-concentrations,-which-was-lower-compared-to-2011,-lower-than-in-2011,-which-resulted-in-lower-observed-mixing-ratios-of-O₃-in-2012. Furthermore, Lyman and Tran (2015) measured O₃ and meteorological parameters at different location in the UB and observed a negative correlation between the O₃ concentration-mixing ratio and station elevation. The stations at higher elevations showed very few O₃ exceedance events compared to those at lower elevation. As mentioned by Schnell et al. (2009), the-a prolonged inversion period traps O₃ near the basin floor due to low wind speeds and limited vertical transport, hence reducing O₃ concentrations-mixing ratios at the higher elevations. Additionally, Oltmans et al. (2014a) conducted 7 aircraft flights in the UB and found that the-high-O₃-concentrations-were-confined-in-mixing-ratios-were-confined-to the shallow inversion layer, namely 300-400 m above the ground.

Mansfield and Hall (2013) used a statistical model to accurately predict O₃ formation, but they noted challenges in extending the findings from one basin to another, as factors such as the thermal inversion and snow cover that play an important role in wintertime O₃ formation vary among basins. They used quadratic regression models to predict the daily O₃ concentrations-mixing ratios in the UB and UGRB and found that the high O₃ events in the UB and UGRB occurred primarily in February

and March, respectively. However, the most intense inversion periods in both basins occurred in January. For both the UB and UGRB, they concluded that these high O₃ events were highly sensitive to the solar radiation, which intensifies as the year progresses.

Carter and Seinfeld (2012) used a box model to study NO_x-limited and VOC-limited regimes in the UGRB. They found that the concentrations-mixing ratios of NO, NO₂ and NMHC, and VOC/NO_x ratios varied both spatially and temporally within the basin. Hence, they suggested that equal attention needs to be given to the geographical distribution of O₃ precursors and the local meteorology. Edwards et al. (2013) utilized the Dynamically Simple Model of Atmospheric Chemical Complexity (DSMACC), a photochemical box model with a very thorough-detailed chemical mechanism, to assess the sensitivity of NO_x and VOC along with radical precursors¹ for O₃ production in the UB. Using this model, with input of observed O₃ precursors, they were able to accurately simulate relatively small amounts of O₃ formation in the absence of snow cover in 2013. Furthermore, Edwards et al. (2014) demonstrated that the same model could simulate large amounts of O₃ production in the UB when snow cover was present, and they emphasized the importance of carbonyl photolysis in the radical chemistry.

There have been a few studies that have utilized 3-D photochemical models to simulate high O₃ events in western US oil and gas basins, though to date there has not been a successful 3-D photochemical modeling study that has simulated high wintertime O₃ in the UGRB. Rodriguez et al. (2009) applied the Comprehensive Air Quality Model with Extensions (CAMx) to assess the impacts of the development of oil and gas fields in the western US on the air quality of various parks and national wilderness areas in the inter-mountain west of the US for 2002. They concluded that the model captured the general trend in O₃ on a monthly scale; however, the model did not capture wintertime O₃ formation events occurring during strong inversions. Ahmadov et al. (2015) used the Weather Research Forecasting model coupled with Chemistry (WRF-Chem, version 3.5.1) to study wintertime O₃ pollution in the UB. To account for the emissions from the oil and gas sector, they employed two different emission scenarios. The first emission dataset was the US EPA National Emission Inventory 2011 version 1 (NEI2011; bottom-up) and the second emission dataset was derived from in situ aircraft and ground-based measurements (top-down). They reported an underestimation of hydrocarbons (CH₄ and other VOCs) and an overestimation of NO_x emissions in the NEI2011 inventory compared to the top-down emission scenario. Ahmadov et al. (2015) found that the model simulation using the bottom-up NEI2011 inventory underestimated the high O₃ concentrations-mixing ratios observed in the UB and that it was necessary to utilize observed concentrations-mixing ratios of VOCs and NO_x to successfully simulate observed O₃ mixing ratios. Additionally, Matichuk et al. (2017) used WRF and Community Multiscale Air Quality (CMAQ) model to study a 10-day high-ozone-episode-high O₃ event in 2013 in the UB. Similar to Ahmadov et al. (2015), they also used the NEI2011 emission dataset, but they found that the CMAQ model did not reproduce the observed O₃, NO_x, and VOC levels in the UB. Furthermore, Matichuk et al. (2017) identified a positive temperature bias and overestimation of the daytime planetary boundary layer height in the WRF simulations, which was hypothesized to be the reason for the underestimation of O₃, NO_x, and VOCs from-in the CMAQ model.

As outlined above, wintertime O₃ production requires a thermal inversion as well as sufficiently deep snow (i.e., deep enough to cover most of the vegetation) over a larger area; hence, not all winters experience high O₃ concentrations-mixing

¹Formaldehyde, nitrous acid, and nitryl chloride.

ratios. Additionally, reported emissions from oil and gas have been significantly reduced over the last decade WYDEQ (2018). In the winter of 2005 and 2006, the newly installed WYDEQ monitoring stations at Boulder, Daniel South, and Jonah observed multiple occurrences of high O₃ ~~concentrations~~ mixing ratios that exceeded the existing 1997 8-hour O₃ standard (84 ppb, 135 WYDEQ, 2018). Since 2005, WYDEQ has operated regular annual O₃ monitoring in the UGRB, and several air quality and weather monitoring stations have been added in the basin. In recent years (most notably 2008, 2011, 2017, 2019, and 2020), elevated wintertime O₃ events have been observed in the UGRB, with hourly O₃ ~~concentrations~~ mixing ratios exceeding 70 ppb for several days in each year. The formation and occurrence of elevated wintertime O₃ ~~concentrations~~ mixing ratios is an unusual event compared to its urban summertime formation. In July 2012, the UGRB was declared as a marginal non-
140 attainment area for O₃ by the US EPA (Rappenglück et al., 2014). In the winter of 2012, there were only 3 days in which the 8-hour averaged O₃ mixing ratios exceeded 75 ppb (NAAQS 2008), while in the winter of 2011, there were 7 days of exceedance (Field et al., 2015b) at a site located near the Boulder station. ~~In~~ Moreover, in March 2017, the Boulder station observed several hours of an hourly averaged O₃ ~~concentration~~ mixing ratio exceeding 70 ppb (NAAQS 2015).

Given the continued occurrence of high O₃ events in the UGRB, the lack of modeling studies aimed at understanding the
145 formation of O₃ in the basin, and plans to continue development of the basin, it is important to develop a photochemical model capable of reproducing high O₃ events of the recent past in order to understand how these events can be prevented in the future. The main goal in this study is to assess if a photochemical model (particularly WRF-Chem) operating with NEI emissions can simulate wintertime O₃ formation in the UGRB. Successful simulation of O₃ events would mean the model could then be utilized to assess effective emission control in preventing future O₃ events as well as the impact of future development on O₃
150 formation. This study primarily focuses on one of the elevated wintertime O₃ events in the winter of 2017 ~~;(a 4-day period from Mar 3 to Mar 7, 2017,)~~ because 2017 was an active year for elevated O₃ in the UGRB (WYDEQ, 2018). The observed hourly O₃ mixing ratios during the period exceeded 70 ppb (NAAQS 2015) for several hours at ~~several~~ multiple air quality monitoring stations in the UGRB. ~~For our O₃ simulations, we have chosen to simulate the 2017 season because this was the most recent year with sustained periods of high O₃ when this project began in 2019. It is most useful to simulate O₃ events from recent years (versus modeling events in 2011) because basin-wide emission estimates from the State DEQ have decreased significantly over the last decade with potential impacts on both ozone precursor concentrations and VOC:NOx ratios. Also, we do not have emissions for oil and gas from 2011.~~ In this paper, the results from WRF-Chem simulations for the given period are analyzed, aimed at understanding the production of O₃ in the UGRB.

2 Methods

160 This section describes the study area, model setup, datasets, methods, and preprocessing tools utilized in the WRF-Chem simulations and to validate the model results.

2.1 Study Region

The focus area of this study is the UGRB. The UGRB is a valley located in Sublette County in western Wyoming, with the Wyoming Range to its west, the Gros Ventre Range to its north, and the Wind River Range to its east. There are 7 weather and air quality monitoring stations operated by the WYDEQ in or near the UGRB: ~~BP~~–Big Piney, ~~B~~–~~Boulder~~, ~~DS~~–~~Boulder~~, Daniel South, ~~JS~~–Juel Spring, ~~M~~–Moxa Arch, ~~P~~–~~Pinedale~~ and ~~SP~~–~~Pinedale~~ and South Pass, whose exact locations are shown in the upper panel of ~~Figure Fig.~~ 1. In addition, the geographical information related to these stations is provided in Table 1. Five of the stations (~~BP~~, ~~B~~, ~~DS~~, ~~JS~~, and ~~P~~Big Piney, Boulder, Daniel South, Juel Spring, and Pinedale) are in close proximity to each other and lie in the basin where wind, and pollutant transport can be affected by the mountains to the east, west, and north. Stations ~~B~~ and ~~P~~Boulder and Pinedale lie in close proximity to the Pinedale Anticline and Jonah Field Developments (PAJF). The natural gas and oil development fields are located southwest of stations ~~B~~ and ~~P~~Boulder and Pinedale, as shown in the bottom panel of ~~Figure Fig.~~ 1 (Toner et al., 2019). The other two stations (~~M~~ and ~~SP~~Moxa Arch and South Pass) lie further away from the basin. ~~Station SP~~ and the PAJF. Station South Pass is located in the foothills of the Wind River Range and has the highest elevation, and station ~~M~~Moxa Arch is the southernmost and lowest in elevation and is located in close proximity to an interstate highway (I-80).

2.2 Model Setup

Simulations of O₃ formation in the UGRB were conducted using WRF-Chem (Skamarock et al., 2008) version 3.9.1. WRF-Chem is a fully coupled model, in which its atmospheric chemistry component is directly coupled to the meteorological component of the model (Grell et al., 2005). The meteorological and air quality components of the model use the same transport and physics schemes as well as the same vertical and horizontal grid structure. This is beneficial over models such as CAMx and CMAQ where the meteorological and the atmospheric chemistry components are run separately. Ahmadov et al. (2015) also pointed out the benefit of WRF-Chem, which helped in the proper simulation of pollutant accumulation in shallow inversion layers. ~~The model configuration with,~~ including the physical and chemical parameterizations used ~~for the study~~in this study, is shown in Table 2. Figure 1 shows the model domain and terrain height, which is centered on the UGRB. The model domain is represented by a grid of 200 x 200 x 60 points with a horizontal grid spacing of 4 km; vertical grids extend up to 100 hPa, with 60-m grid spacing near the surface and 250-m grid spacing at the top of the model.

2.3 Datasets

The National Centers for Environmental Prediction (NCEP) North American Regional Reanalysis (NARR) (Mesinger et al., 2006) was used for the initial and boundary meteorological conditions for the simulations in this study. The data are available on a Lambert conformal conical grid with a grid spacing of approximately 0.3 degrees (32 km). The 3-hourly fields with 29 vertical pressure levels from 1000 to 100 hPa were used in this study to initialize and provide the lateral boundary conditions for the WRF-Chem simulations to study O₃ formation.

The NEI data were used for emissions in the ~~the~~-WRF-Chem simulations. ~~The~~ Specifically, the data for natural gas and oil sources were obtained from the US EPA NEI-2014 dataset (version 2, hereafter; NEI2014v2) released in February 2018 (US-EPA, 2018). The NEI2014v2 data were the latest emission inventory available at the time of the initiation of this study and is available at a 12-km horizontal resolution. This particular version of the emission dataset incorporates the ~~processes~~ emissions associated with the exploration, drilling, and production of oil, gas, and coal-bed CH₄ wells in the UGRB. The EPA emission estimates are the most widely used and easily available ~~estimate data~~ that include most potential emission sources ~~that could impact air quality. However, previous comparisons by Alvarez et al. (2018); Robertson et al. (2020) impacting air~~ quality, although some previous studies (e.g., Alvarez et al., 2018; Robertson et al., 2020) have pointed out underestimations of CH₄ emissions ~~for the from~~ oil and gas extraction basins in EPA estimates compared to ~~their~~ observations. To account for the transport of chemical species into the model domain, 6-hourly data from the Community Atmosphere Model with Chemistry (CAM-CHEM; Emmons et al. (2020)) were used in the simulations.

The observed meteorological and air quality data from the aforementioned 7 weather and air quality monitoring stations were obtained from the WYDEQ website. The data are available in 5-minute and hourly formats. The hourly data were used for this study for a direct comparison of meteorological parameters, such as temperature and wind speed, and chemical species, such as O₃, NO_x, CH₄, and NMHC, with the simulated results. The NHMC data were only available at the Boulder ~~station~~ site as this was the only station equipped to report these results.

2.4 WRF-Chem simulations

The O₃ formation simulations focus on a 4-day period from Mar 3 to Mar 7, 2017. A spin-up period was not explicitly considered in this study owing to the computational expense of each simulation and that O₃ generally does not start increasing until nearly 24 hours into the simulation; however, the results from the first day should still be viewed with caution. For all simulations, the model physics and photolysis surface albedos were modified to account for the effect of snow on photolysis in the model. The default photolysis albedo in the model is 0.15 because the model was primarily developed for summertime photochemistry. The default photolysis albedo is much lower than what is commonly observed during winter when the surface is covered with snow. Under the default albedo of 0.15, the simulations drastically underestimated O₃ formation (as shown in the results below). This study is intended to study *wintertime* photochemistry of O₃. We thus require a higher albedo to represent a snow-covered surface. Hence, in an effort to simulate a range of potential surface conditions, multiple albedo sensitivity simulations were carried out. A similar study using WRF-Chem with RACM chemistry was carried out by Ahmadov et al. (2015) in the UB, Utah, where they set the surface albedo to 0.85 in their simulations of wintertime O₃ production. ~~As~~ However, as noted by Mansfield and Hall (2013), for wintertime O₃ formation, factors such as the thermal inversion and snow cover play an important role and they vary among ~~the basins. Hence basins, and thus~~ the findings and characteristics of wintertime O₃ formation cannot be extended from one basin to another. ~~Speciallly, Therefore, in this study,~~ surface albedos of 0.55, 0.65, 0.75, 0.85 and 0.95 were used for the sensitivity study and fixed to 0.85 in the model for further analysis based previous estimates of snow albedo in the region (Ahmadov et al., 2015) and sufficient O₃ formation in the UGRB ~~using 0.85 surface albedo.~~

In this study, two different chemistry mechanisms ~~are were~~ used: (i) the Model for Ozone and Related Tracers (MOZART) and (ii) the Regional Atmospheric Chemistry Mechanism (RACM). The MOZART chemistry mechanism has been widely used ~~model~~ to study O₃ formation and transport around the world (Hauglustaine et al., 1998; Murazaki and Hess, 2006; Beig and Singh, 2007; Yarragunta et al., 2019). In the UB, RACM has been successfully used to simulate O₃ production due to oil and natural gas production in winter when observed levels of VOCs and NO_x ~~were~~ x were used as inputs (Ahmadov et al., 2015). Based on the findings from Ahmadov et al. (2015), the important point noted by Mansfield and Hall (2013), and the MOZART and RACM mechanisms being widely used chemical mechanisms to study O₃ both globally and regionally, the simulations were carried out with these two chemical mechanisms to understand which chemical mechanism provided the best ~~comparison with observed~~ prediction of O₃ compared to observations and its precursors in the UGRB. The WRF-Chem namelist options used for MOZART and RACM are provided in ~~the supplemental section A2 of this paper in Figures Appendix A, Figs. A1 and A2, respectively.~~ Despite all Where possible, the same namelist options ~~used in these models, the were used for both models. However, regarding the photolysis option, the~~ simulations with MOZART ~~use used~~ photolysis option 4, which is the updated TUV photolysis option ~~that was setup based on recent advanced in the understanding of photolysis rates that was configured~~ to work with only a few chemistry mechanism schemes in WRF-Chem v3.9.1. ~~While, while~~ the RACM simulations ~~use used~~ photolysis option 1, which is the Madronich photolysis scheme. ~~With the current setup for photolysis option 4 in the WRF-Chem v3.9.1 it does not work with RACM chemistry mechanism. This study uses photolysis option 4 for MOZART simulation as it produces higher O₃ compared to when photolysis option 1 was used (Figure not shown).~~

~~Additionally, some key points that were considered to achieve the goals of this study and needed to reproduce the results are as follows: (i) sufficient surface albedo to represent the effect of snow cover and depth on the meteorological conditions, (ii) correct photolysis albedo to represent the wintertime conditions for the chemical mechanisms to reproduce sufficient O₃, and (iii) NEI data as well as CAM-CHEM global emissions data processed separately for each chemical mechanisms, as different mechanisms lump chemical species differently and are also driven by different chemical reactions.~~

2.5 Preprocessing

The EPA anthro emiss tool provided by the Atmospheric Chemistry Observations & Modeling² (ACOM) division at the National Center for Atmospheric Research (NCAR) was used for preprocessing the emissions in this study. This tool creates anthropogenic emission files from the NEI datasets ~~for lat/lon grids~~ that can be ingested into the ~~WRF model. The WRF-Chem model. Because the~~ MOZART and RACM chemistry mechanisms use different species ~~grouping; hence groupings~~, the emission inventory files were processed separately for each mechanism. ~~Mozbe; Additionally, mozbc,~~ which is also provided by ACOM, was ~~also~~ used in this study. ~~The mozbe tool maps the to map the chemical~~ species from the ~~Community Atmosphere Model with Chemistry (CAM-CHEM)~~ global dataset to ~~WRF WRF-Chem~~ fields that can easily be ~~ingested into WRF-Chem used~~ as initial and boundary conditions.

For simulations using the MOZART chemical mechanism, two other WRF-Chem utilities were also used: exo_coldens and wesely. The exo_coldens utility ~~helps read O₃ and O₂ reads O₃ and O₂~~ climatological atmospheric column ~~values~~

²<https://www2.acom.ucar.edu/wrf-chem/wrf-chem-tools-community>

260 ~~data~~ rather than using fixed values, and this is coupled to ~~an updated the aforementioned updated TUV~~ photolysis option
(~~photophot_opt=4~~). For dry deposition in MOZART, ~~an additional file is required that allows the wesely utility is used to~~
~~account~~ for seasonal changes in dry deposition. ~~The additional information is provided using the wesely utility.~~ Both the
exo_coldens and wesely utilities read the ~~WRF input files as well as emission files for the MOZART chemistry mechanism~~
~~WRF-Chem input files and emission files~~ to produce additional data files ~~that can be read by the for~~ WRF-Chem model.
265 ~~simulations conducted with the MOZART chemistry mechanism.~~

~~The NEI2014v2 dataset provides emissions covering the model domain, but the advection of chemical species into the
domain through the lateral boundaries must also be considered. The WRF-Chem simulations in this study used the NEI2014v2
emission data re-gridded to the WRF-Chem domain. The initial and boundary conditions of the simulations were updated every
24 hours for each simulations using the CAM-CHEM data.~~

270 2.6 ~~Temperature Inversion Analysis and Surface Meteorology~~ Model Validation

To study the ability of the model to replicate observed meteorological conditions in the UGRB, we study the temperature
inversion, weak winds, and surface temperature. ~~The~~ Owing to differences in data availability (e.g., observations and emission
inventories), two different periods are selected for meteorology validation. Specifically, the temperature inversion was studied
using the WRF model (without chemistry) for 2011 ~~with the same model configuration~~, while the surface meteorology ~~in~~
275 ~~the~~ was validated using the focus period in March 2017, and the details of these analyses are provided below. Additionally,
wintertime O₃ was measured in the UGRB from February to March 2017 by the WYDEQ, providing an additional dataset to
validate the WRF-Chem ~~simulations for model~~, albeit with a strict focus on O₃ precursors.

2.6.1 Temperature Inversion Validation

Owing to the aforementioned data availability limitations, in addition to the WRF-Chem simulations focused on the high
280 O₃ event in Mar 2017 were utilized. The reason for the discrepancy is due to differences in data availability between the
different periods. For model validation, the simulation results were compared with, additional simulations using WRF but
without chemistry were conducted for the entire winter of 2011 (Dec 1 2010 to Mar 31 2011; hereafter referred to as IOP11),
encompassing the period during which vertical profiles of temperature and O₃ from ~~ozonesonde data collected during two~~
~~intensive operational period (IOPs) in 2011. The temperature inversion was studied to validate the ability of WRF model~~
285 ~~meteorology to simulate inversions in the basin. The data from year 2011 was utilized because the ozonesondes were collected~~
~~by the WYDEQ Air Quality Department (AQD) conducted two IOPs in winter 2011 (MSI, 2011, Feb 28 to Mar 2 and Mar~~
~~9 to Mar 12). This is the only year for which vertically resolved meteorological data were available from radiosondes. The~~
~~observed vertical data for the temperature inversion was also obtained from the WYDEQ website. The~~ The IOP events were
identified based on the conditions (deep snow and large spatial coverage in the study area, development of an inversion, and
290 calm surface winds) that support elevated O₃ ~~concentrations~~ mixing ratios. During each IOP period, 3-4 ozonesondes were
launched adjacent to the Boulder station (see Fig. 1) each day, providing vertical profiles of O₃ mixing ratio, temperature, and
wind speed. ~~The WRF simulation was carried out for the entire winter of~~ We note that the data from year 2011 (Dec 1 2010

to Mar 31 2011), which includes both IOP periods and the high O₃ events of the winter of 2011. We were utilized because this is the only year for which vertically resolved meteorological data were available from radiosondes. Further, we understand that the ability of the model to simulate one event (i.e., the vertical structure for a few days in 2011) does not indicate that it will perform accurately again. However, with given that basin-wide emission estimates from the state DEQ have decreased significantly over the last decade with potential impacts on both O₃ precursor mixing ratios and VOC:NO_x ratios, as well as the unavailability of emissions for oil and gas from 2011, the lack of data, we are forced to either not examine the vertical structure at all or instead find an analog that can provide some level of confidence in the IOP2011 simulation is only used herein to validate simulated temperature inversions, and the focus of the remainder of this work is on the high O₃ events that occurred in Mar 2017.

2.6.2 Surface Meteorology Validation

The surface meteorology in the WRF-Chem simulations detailed in Section 2.4 were validated against observations collected at 7 monitoring stations (Big Piney, Boulder, Daniel South, Juel Spring, Moxa Arch, Pinedale and South Pass) by WYDEQ, with a focus on temperature and wind speed, two factors that greatly impact the accumulation of O₃, as described in Section 1. While wind direction is another important variable for pollutant transport, given the low wind speeds that occur during high O₃ events, the model's ability to replicate the vertical structure of the lower troposphere during high-O₃ events. We chose the later and proceeded with the no chemistry simulations for the IOPs in 2011. The simulation will hereafter referred to as IOP11 precisely simulate the observed wind directions is of lesser important to this study. Moreover, such a comparison could be greatly affected by sampling issues owing to the high variability in observed and simulated wind directions under nearly calm conditions.

2.7 VOC and NO_x Sensitivity Study

WYDEQ carried out a winter wintertime O₃ study from February to March 2017, coinciding with the high O₃ event that is the focus of this study. On several days during this study speciated VOC canisters period, speciated VOC canister measurements were collected between 04:00 to 07:00 MST at Boulder, Big Piney, Juel Spring, and Moxa Arch (MSI, 2017). This study uses the The speciated VOC data from Boulder all stations on Mar 3, 2017 to compare with the, were compared with the respective station model data. The observed values from the canister and models are Table B1 in Appendix B shows ratios of canister-observed speciated VOC mixing ratios to the simulated values. The Moxa Arch site is relatively far from the emission sources and is not representative of the main O₃ formation region. The other sites show variability, although all sites show significant underestimates of both VOCs and NO_x. The Boulder site has the largest underestimates of reactive BTEX species, while some species have larger underestimates at other sites. Given this comparison and the site-to-site variability in the model-observation comparison, VOCs measured at the Boulder site appear to be a reasonable, though aggressive, basis for adjusting the emissions in the model. For NO_x, because the data are available for the entire study period, factors for NO and NO₂ were calculated taking into account the entire study period (Mar 3 to Mar 7, 2017; see Fig. B1 in Appendix B for details on the time series), as shown in Table 5. Three different simulations for

Using the above model–observation comparison, four additional simulations using each chemistry mechanism were carried out to study the effect of adjusted NO_x sensitivity to NO_x and VOC emissions. The precursors' emissions were adjusted in each simulation based on the factors shown in Table 5. The factors for each VOC species are kept the same in each simulation, while the factors for NO and NO_2 are adjusted for each case. The emissions factors for VOC in Table 5 is emission adjustment factors for VOCs in Table 5 were calculated by dividing the observed value values (canister) by the model-simulated value values for the same time period. However, because NO_x data is available for the entire study period, factors for NO and NO_2 are calculated taking into account the entire study period (Mar 3 to Mar 7 2017), with The goal of this analysis and additional sensitivity simulations performed was to 1) constrain adjust NO and NO_2 to better match the observations and 2) test the sensitivity of O_3 production to NO_x levels. The time series of NO and NO_2 at Boulder is shown in the supplemental section Figure B1. In the second simulation the NO_x levels. In all, 4 additional sensitivity simulations were conducted using the aforementioned emission adjustment factors for VOCs are kept as it is while the factors for NO and NO_2 are reduced to half and in the third simulation the factors for NO and NO_2 are doubled keeping the factors for VOC the same. Dry deposition of gaseous species is turned off in all simulations for the sensitivity study. NO_x : 1) increased VOCs only, 2) increased NO_x only, 3) increased VOCs and NO_x , and 4) increased VOCs and doubled NO_x (here, “doubled” indicates that the emission adjustment factor was doubled).

The adjustments for certain VOC species that are lumped in the model chemistry requires slightly more explanation. For alkanes, the concentration mixing ratios of all observed alkane species larger than propane (butane up to undecane) were summed, and then this sum was used to adjust the lumped model species (bigalk for MOZART and HC5 for RACM). In RACM, the species TOL is a combination of toluene and a fraction (0.293) of benzene. Accordingly, a fraction of the observed benzene was added to the observed toluene to adjust this variable. Also, in RACM, the species HC3 is a combination of methanol, ethanol, and a fraction (0.519) propane. The DEQ observations do not include methanol or ethanol, meaning only the observed propane mixing ratio was used to modify this variable. Finally, in both models the lumped xylene parameter includes trimethylbenzene. Accordingly, the observed concentrations mixing ratios of xylene and trimethylbenzenes were added to create the emission grouped to calculate the emission adjustment factor for xylene.

350 3 Results and Discussion

To simulate O_3 formation in the UGRB, we first validated We first validate the WRF model's performance in simulating the observed vertical temperature profile and surface meteorology during strong inversions (see Section 2.6.1 for details). After determining that WRF was is able to reasonably reproduce the meteorological conditions necessary for O_3 formation, we studied analyze O_3 formation with the WRF-Chem model using two different chemical mechanisms and multiple sensitivity simulations.

3.1 Validation of WRF Model Meteorology

3.1.1 Temperature Inversion

Owing to the importance of thermal inversions for the build up of θ_{O_3} in wintertime events, we first ~~explored~~explore the ability of the model to simulate temperature inversions within the selected modeling framework. Vertical profiles of the observed temperature and O_3 mixing ratio during the most recent IOP (Feb 28 to Mar 2 and Mar 9 to Mar 12, 2011) ~~were~~are compared with the simulated vertical temperature profiles from ~~simulations with WRF~~WRF simulations during the same time period (IOP11, ~~Figure~~Fig. 2). Although 7 days ~~were~~are identified as the IOP period, the results from only 4 days are discussed due to ozonesonde data availability. Because these simulations ~~were completed~~are performed to compare meteorology and not chemistry, the WRF model without chemistry ~~was~~is used, and simulated O_3 is not available. We ~~did~~do not aim to simulate O_3 events from 2011 because emissions have changed dramatically since 2011 and there is ~~not a~~no good inventory that includes oil and gas sources for that period. Observed O_3 is presented only to demonstrate how O_3 formation follows the inversion events.

A shallow mixing height can be seen in each profile. The residual layer above the ground appears to be well mixed early in the simulation; hence, we can see fairly uniform O_3 ~~concentrations in the vertical~~high concentrations mixing ratios in the ~~observed profiles~~observed profiles. ~~High mixing ratios~~High mixing ratios of O_3 ~~were~~are observed on Mar 1-2, 2011. On these days, a strong inversion is observed with a shallow mixing height of around 500 m agl, which prevents vertical mixing, thus leading to a build up of O_3 precursors ~~that then lead to high concentrations and high mixing ratios~~of O_3 that increase in the afternoon ~~MSI (2011)~~(MSI, 2011). On Mar 2, 2011 (third row), higher morning O_3 ~~was~~is observed compared to the previous day, presumably due to the persistent inversion, which is validated by the observation of high hydrocarbon ~~concentrations~~mixing ratios in the afternoon of Mar 2 (MSI, 2011).

For the days discussed here, the simulated temperature is 2 to 4 °C warmer than the observed temperature, except for Mar 9, 2011 (~~Figure~~Fig. 2, last row), where it is 2 to 5 °C colder than the observed temperature near the surface. During the morning hours, the simulated temperatures follow the observed temperatures fairly well; however, the simulated inversion height is slightly elevated. In both the observations and the model, the inversion height increases through the day and the inversion strength (difference in maximum vs. surface temperature) decreases. However, the model seems to increase the inversion height slightly too much while also decreasing the strength of the inversion. Overall, the model simulation of the inversion events ~~was~~is deemed adequate to proceed.

3.1.2 Surface Meteorology

Given the model's ability to ~~relatively accurately~~reasonably represent temperature inversions, at least based on our comparison with available data from 2011, we further assess the model's ability to predict surface meteorology focusing on the target period of high O_3 in March 2017. It is important to highlight again that vertical data are not available for the selected time period. We utilize observations from the high O_3 events of 2017 because the seven ground stations measure basic meteorological

parameters. It is crucial for the photochemical model to simulate low temperatures and calm winds to be able to replicate high O₃ ~~concentrations~~ mixing ratios (Schnell et al., 2009).

390 The observed 2-m temperature data for Pinedale ~~were are~~ unavailable, hence the temperature ~~correlation~~ correlations for only six stations are shown in Figure Fig. 3. Both simulations show good correlation with the observed temperatures, and the correlation coefficients do not show any sensitivity to the different chemistry mechanisms at the Juel Spring and Moxa Arch stations. However, RACM shows higher correlation coefficients compared to MOZART at other stations ~~Although the~~ (except Boulder). The difference in the correlation coefficients for the different chemistry ~~mechanism is small, it is~~ mechanisms is
395 small and likely due to radiation feedbacks between the chemistry and meteorology ~~in these mechanisms and as well as~~ internal model variability (Bassett et al., 2020). Furthermore, the temperature bias between the observed and simulated datasets is below 3 °C at all stations (Table 3), and all of the data points lie in close proximity to the one-to-one lines. Overall, the simulations show good correlation with the observed 2-m temperatures.

As mentioned earlier, calm ~~wind speeds~~ winds are an essential meteorological condition for the photochemical production
400 of wintertime O₃ because they are necessary for the accumulation of O₃ precursors. The correlation between observed and simulated wind speeds is shown in Figure Fig. 4. The correlation coefficients are calculated for each data point (hourly) for the entire study period, although only wind speeds from 0 to 10 m s⁻¹ are shown given the focus of the study ~~is calm~~ periods on calm periods with high O₃ mixing ratios. For all stations except South Pass, a majority of the data points are clustered below or around 4 m s⁻¹ ~~, which means that for the majority of the time, both the observed and simulated wind~~ speeds are less than or equal to 4 m s⁻¹ in both the observations and model simulations. The differences in the correlation
405 coefficient coefficients between different simulations are due to internal model variability ~~of the model (Bassett et al., 2020)~~ Therefore (Bassett et al., 2020). Furthermore, the relatively low correlation coefficients may be the result of small variations ~~of~~ in low wind speeds. To test this ~~idea notion~~ and to verify that calm periods were successfully simulated when they occurred are successfully simulated, Table 4 shows the percentage of ~~the times the time that the~~ simulated and observed wind speeds are
410 less than or equal to different thresholds (3, 4 and 5 m s⁻¹). For example, at Boulder, both the simulated wind ~~speed from~~ speeds using MOZART and the observed wind ~~speed speeds~~ are less than or equal to 3 m s⁻¹ for 98.33% of the hourly periods analyzed analysis period, while for RACM this figure is 90.77%. Again, the chosen thresholds are based on the interest in studying calm wind ~~speed speeds~~ in the basin, which limit pollutant transport/dispersion and enable pollutant accumulation near the surface. Therefore, even though the correlation coefficients between the modeled and observed winds are relatively
415 low, we conclude from the results in Table 4 that WRF with either chemistry mechanism is able to successfully predict low winds the large majority of the time they occur wind speeds for most of the study period. An analysis of the diurnal variability of winds ~~showed good also shows reasonable~~ qualitative agreement between the model and observations in terms of the timing of increasing and decreasing wind speeds each day (Figure, especially on days with elevated O₃ mixing ratios (figure not shown)).

3.2 Baseline Simulation and O₃ Production

420 Given the aforementioned ability of the model to adequately simulate the key meteorological conditions needed for O₃ production and accumulation, we now turn to an analysis of the chemical mechanisms and their ability to ~~produce~~ reproduce the

observed hourly periods ~~with~~ of high O₃. At first, O₃ formation was simulated in the UGRB using the ~~MOZART chemistry mechanism~~ RACM chemistry mechanism, and it was noted that the modeled ~~concentrations~~ mixing ratios were dramatically below ~~the~~ observed O₃ levels. However, the default WRF-Chem model has a low photolysis albedo (0.15) as it was intended
425 to simulate summertime O₃, ~~which does not typically occur over high-albedo surfaces~~. We modified the photolysis albedo in the model based on more typically wintertime conditions following Ahmadov et al. (2015), who noted that in the UB, it was necessary to increase the photolysis albedo to accurately simulate O₃ production. ~~In an effort to understand~~ Further, in this study, additional simulations were conducted to analyze the sensitivity of O₃ formation to the photolysis albedo in the WRF-Chem model, ~~we performed a sensitivity test. As described in the methods section, we carried out several albedo sensitivity~~
430 ~~simulations with various albedo settings ranging from 0.55 to 0.95~~ (spanning albedos representative of partially snow-covered vegetation to fresh, deep snow) and compared the results ~~to the results with~~ with those obtained using the default albedo of 0.15. All of the albedo sensitivity tests used the RACM chemical mechanism. Figure 5 compares the default albedo (0.15) with different photolysis albedo settings (0.65 and 0.85). It is evident that the default photolysis albedo produces much lower O₃ ~~concentrations~~ mixing ratios at all stations. However, when the model is altered to use an albedo of 0.85, the diurnal variation
435 and high O₃ peaks are captured relatively well, although there is some variability from station to station. For the remainder of the simulations in this paper, a photolysis albedo of 0.85 is used, which is the same albedo used by Ahmadov et al. (2015) in the UB.

Setting a fixed photolysis albedo of 0.85, we next compared simulations using two different chemistry mechanisms available in WRF-Chem: MOZART and RACM. Figure 6 compares the time series of simulated hourly O₃ ~~concentrations~~ mixing
440 ratios from four different simulations with dry deposition of gas species included and not included in both MOZART and RACM simulations at several UGRB monitoring stations. The hourly averaged observed background daily O₃ mixing ratio is approximately 55 ppb at all stations. During the afternoon hours, most of the stations have hourly O₃ mixing ratios greater than 70 ppb. The observed O₃ ~~concentrations~~ mixing ratios are highest at the Boulder site, which is likely because it lies in close proximity to the PAJF production facilities and is thus closer to the main sources of VOC precursors than the other sites. For
445 Moxa Arch and South Pass, the observed O₃ ~~concentrations~~ mixing ratios are lower because they do not lie in close proximity to the wells and also lie further from the basin.

~~It is important to note that one difference between the MOZART and RACM simulations used in this study is the photolysis option (phot_opt = 4 for MOZART and phot_opt = 1 for RACM), which could affect O₃ production. As RACM is not coupled to phot_opt = 4, an addition sensitivity simulation was performed using option 1 with MOZART, which led to less O₃ compared with using option 4, albeit with better agreement with the observations. As such, we elected to use phot_opt = 4 for subsequent simulations but note that some of the difference between RACM and MOZART may be attributed to the photolysis scheme used with the former leading to less O₃ production.~~

450

To better understand the chemistry mechanisms' sensitivity to dry deposition, we also compare the diurnal variation of O₃ ~~concentrations~~ mixing ratios from MOZART and RACM with dry deposition turned ~~on and off in both simulations~~ off at the 7
455 monitoring stations. The ~~RACM simulation with~~ justification for these additional simulations is that RACM's dry deposition of gas-phase species (~~RACM_ddOn~~) does not produce sufficient O₃ to replicate the observed O₃ ~~concentrations~~ (Figure mixing

ratios (Fig. 6; orange lines). However, when dry deposition of gas species is included in MOZART (MOZ_ddOn; Figure 6; purple lines), the simulation performs better compared to RACM_ddOn. In MOZART, when dry deposition is turned on, it adjusts the deposition rate over snow surfaces (owing to the use of the weesly pre-processing tool that adjust the seasonal change in dry deposition), where the loss is expected to be greatly reduced. On the contrary, RACM does not adjust the dry deposition rate over such snow-covered surfaces, hence the simulation with dry deposition turned off dry deposition is likely too high in RACM and could explain the underestimate of O₃. Thus, we turned off dry deposition to mimic the very slow deposition of gas-phase species over a snow-covered surface surfaces (i.e., RACM_ddOff). Despite this markable large differences in the results from the simulations when the dry deposition of gas species is included, when dry deposition is turned off, both MOZART (MOZ_ddOff; Figure Fig. 6; blue lines) and RACM (RACM_ddOff; Figure Fig. 6; red lines) produces similar concentrations-mixing ratios of O₃, suggesting that a large reason for the disparity in results under the different chemistry mechanisms is related to the formulation of dry deposition of gas species.

Focusing on Shifting focus to the individual sites, at Big Piney and Daniel South, which are located on the eastern side of the Wyoming range, all four simulations overestimate the first O₃ event (Mar 03-3 2017 at 15:00 local time). However, as noted earlier, the model results from the first day should be viewed with caution. Moreover, the MOZ_ddOff and RACM_ddOff simulations capture the diurnal cycle of O₃ reasonably well at Boulder, while they overestimate the high O₃ event at Pinedale on Mar 03-3, 2017, 17:00 local time, which is well captured by MOZ_ddOn. However, the simulations miss the higher O₃ concentrations-mixing ratios at Juel Spring. Overall, both the MOZ_ddOff and RACM_ddOn-ddOff simulations do reasonably well at simulating the O₃ mixing ratios in the UGRB for the selected study period and capturing the diurnal variation of O₃, a first for a photochemical model using an existing emissions inventory, although it is important to remember that this was only possible after adjusting the photolysis albedo in the model and, in the case of RACM, turning of-off dry deposition of gas-phase species. Due to their better performance in estimating observed O₃ the results from MOZ_ddOff and RACM_ddOff simulations will be discussed in the following analyses and the simulations will be referred to as MOZ17 and RACM17 respectively.

To better understand the differences in the simulated and observed O₃ concentrations-mixing ratios, we next looked at the precursor (NO_x) concentrations-mixing ratios. Figure 7 shows the time series of hourly NO_{x-x} at the 7 monitoring stations, along with results from MOZ17 and RACM17. The observed hourly mixing ratios of NO_{x-x} at Big Piney, Boulder and Pinedale are higher than the other stations. These three stations are all near small towns in the region with Pinedale being the largest of the towns and Pinedale having notably higher NO_{x-x} than the others. The NO_x-mixing ratio is primarily affected by its emission rate in the region. At Pinedale, the higher observed concentrations are most likely due to the fact that the station is near the city of Pinedale where Moreover, in Pinedale, there are sources of NO_{x-x} that are not related to oil and gas, most notably residential wood burning. However, residential wood burning is not well represented in the emission inventory; thus, the model is expected to underestimate NO_x from this source_x in such areas. The elevated observed NO_x concentrations_x mixing ratios compare well with the observed PM2.5 concentrations at Pinedale (Figure C1 Fig. C1 in Appendix C), which supports the conclusion that wood burning is a strong NO_x-source in these areas_x source in this area. The simulated concentrations of NO_x seems mixing ratios of NO_x are less sensitive to the different chemical mechanisms, emphasizing that the emissions dominate concentrations, the mixing ratios and not chemical loss mechanisms. The NO_{x-x} mixing ratios are underestimated by

both simulations even during the high O₃ events. Although the simulated NO_x concentrations x mixing ratios at Daniel South is are higher compared to the other stations, the observed data are missing. The observed and simulated NO_x concentrations x mixing ratios at South Pass are low and show little variability, emphasizing as expected given that this station is further from the oil and gas production region. Overall, the simulations underestimate the observed NO_x concentrations x mixing ratios to varying degrees depending on the location and do not capture the diurnal cycle well, which poses a juxtaposition given the reasonably good agreement between simulated and observed O₃ mixing ratios.

The top panel in Figure 8 compares In the top panel of Fig. 8, we next compare the simulated NMHC concentrations mixing ratios (plotted on the left; primary y-axis) and observed NMHC concentrations mixing ratios at the Boulder station (plotted on the secondary y-axis). The Boulder station is the only monitoring site in the basin that measures either NMHC or CH₄. In addition, the MOZART³ and RACM⁴ chemical mechanisms lump the VOC species differently. The bottom panel of Figure Fig. 8 shows the observed O₃ concentrations mixing ratios at the Boulder station during the same time period showing that the accumulation of NMHC leads to the production of O₃. The Although the temporal evolution of NMHC is well captured by the simulations, the magnitudes of the simulated NMHC concentrations mixing ratios are lower by a factor of approximately 6 compared with the observation. Both RACM17 and MOZ17 give very similar NMHC mixing ratios. In fact, because the chemical production of O₃ does not remove a large amount of the NMHC present. When it was discovered that the model simulated model simulated VOC mixing ratios were dramatically different from substantially lower than the observations at the Boulder site, we employed University of Wyoming mobile laboratory data to confirm that the Boulder site does not record anomalously high mixing ratios relative to the surrounding area that would all be within the same grid-cell in the model (as the station sits in a small valley). The mobile lab does did not measure NMHC, but both the mobile lab and the Boulder station measure measured CH₄ enhancements, which are a reasonable proxy for VOC enhancements, thus enabling us to see if CH₄ measurements made by the lab in the region surrounding the Boulder site were significantly different than those reported by the site. Hence, we analyzed We compared the CH₄ concentrations (a proxy for VOC concentrations) mixing ratios collected by the mobile lab during an O₃ event in 2020, the closest year to our study period for which data are available. The WYDEQ Boulder site data were within 25% of the data collected by the mobile lab near the monitoring site (Figure D1 Fig. D1 in Appendix D). This observation indicates that the difference between the simulated and observed NMHC mixing ratios is not the result of anomalously high mixing ratios at the Boulder site, but concluded that and thus the NMHC mixing ratio measured

³methylperoxy radical, methyl hydroperoxide, formaldehyde, methanol, ethene, ethane, acetaldehyde, ethanol and its oxides, acetic acid, glyoxal, glycolaldehyde, ethylperoxy radical, ethyl hydroperoxide, acetylperoxy radical, peracetic acid, peroxy acetyl nitrate, propene, propane and its oxides, acetone, hydroxyacetone, methylglyoxal, organic nitrate, lumped alkenes (C>3), methyl ethyl ketone and its oxides, methyl vinyl ketone, methacrolein, methacryloyl peroxyacetyl nitrate, peroxy radicals, lumped alkanes (C>3) and their oxides, isoprene, unsaturated hydroxyhydroperoxide, lumped unsaturated hydroxycarbonyl, unsaturated dicarbonyl, lumped isoprene nitrate, lumped aromatics and their oxides, and lumped monoterpenes and their oxides

⁴ethane, alkanes, alcohols, esters, alkynes, ethene, terminal alkenes, internal alkenes, butadiene and other anthropogenic dienes, isoprene, alpha-pinene and other cyclic terpenes, delta-limonene and other cyclic diene-terpenes, toluene, xylene, cresol, formaldehyde, acetaldehyde, ketones, glyoxal, methylglyoxal and other alpha-carbonyl aldehydes, unsaturated dicarbonyls, methacrolein and unsaturated monoaldehydes, unsaturated dihydroxyl dicarbonyl, hydroxy ketone, organic nitrate, peroxyacetyl nitrate and higher saturated PANs, unsaturated PANs, methyl hydrogen peroxide, higher organic peroxides, peroxyacetic acid, formic acid, acetic acid and higher acids, methyl peroxy radicals, aromatic peroxy radicals, acetyl peroxy and its saturated and unsaturated radicals

at the Boulder site is an accurate representation in the region. ~~Although the overall temporal trend in the NMHC mixing ratio is well captured by the simulations, both MOZ17 and RACM17 dramatically underpredict the NMHC mixing ratios.~~

520 It is very intriguing that both chemical mechanisms are able to reasonably replicate the O₃ ~~concentrations-mixing ratios~~ at the monitoring sites despite the fact that NMHC ~~concentrations-mixing ratios~~ in the model are approximately 6 times lower than those observed at the Boulder monitoring site ~~and NO_x is also generally underestimated~~. The mobile lab results strongly suggest that this discrepancy is not due to non-representative measurements at the Boulder monitoring site. This leaves the possibilities that the simulated NMHC ~~compounds~~ are much more reactive than the actual ~~NMHC/NMHCs~~, that some other
525 feature of the chemistry is too active in the model, and/or that the UGRB will continue to experience high O₃ events even at much lower NMHC levels because ~~ozone-O₃~~ production is predominantly determined by NO_{x-x} availability. In terms of the possibility that the chemistry in the model is too active, it is important to note that the RACM17 chemistry successfully simulated O₃ events in the UB when observed NO_{x-x} and speciated VOCs were input (Ahmadov et al., 2015). The sensitivity to adjustments in speciated VOC and NO_{x-x} emission is discussed later in ~~the manuscript in~~ Section 4.

530 The spatial variation in the formation and dissipation of O₃ and its precursors for the high O₃ event on Mar 4, 2017, is shown in ~~Figures-Figs.~~ 9, 10, and 11 for O₃, NO_x, and VOCs, respectively, from the MOZ17 simulation, and similarly, ~~Figures-Figs.~~ 12, 13, and 14 show the results from RACM17. In both simulations, the formation and build up of O₃ is seen around noon local time (~~Figure-Figs.~~ 9c and ~~Figure-12c~~). In the late afternoon (at 16:00 local time), the O₃ ~~concentration-mixing ratios~~ reaches its maximum of 124 ppb in MOZ17 (~~Figure-Fig.~~ 9d) and 138 ppb in RACM17 (~~Figure-Fig.~~ 12d). Although higher O₃
535 ~~concentrations-mixing ratios~~ are found locally in RACM17, these dissipate rather quickly compared to MOZ17, demonstrating that there are subtle differences in the chemical mechanisms. For both simulations, the highest O₃ ~~concentration-is-mixing ratios are~~ seen closer to the Big Piney, Boulder, Daniel South and Pinedale stations, though none of the stations are simulated to have the highest ~~concentrationsmixing ratios~~. If compared closely with the well locations in ~~Figure-Fig.~~ 1, the highest O₃ ~~concentrations-overlap the location-mixing ratios overlap the locations~~ of the wells. The simulations ~~with different chemistry~~
540 ~~mechanisms~~ show a similar temporal trend in O₃ formation, which can also be seen in ~~Figure-Fig.~~ 6, although the highest ~~concentrations-mixing ratios~~ differ by approximately 10 ppb. The O₃ mixing ratios at Juel Spring, Moxa Arch, and South Pass are comparatively lower. The wind speeds are also stronger ($> 5 \text{ m s}^{-1}$) around these stations (~~Figure-Figs.~~ 9d and 12d). Particularly, ~~around-near~~ South Pass, the wind speeds are around 15 m s^{-1} . With the lack of mountains surrounding these stations and comparatively higher wind speeds, pollutant ~~concentrations-mixing ratios~~ can be easily diluted and dissipated.

545 To better understand the formation, accumulation, and dissipation of O₃ precursors, i.e., NO_{x-x} and VOCs, the diurnal and spatial variations are shown for both simulations (~~Figures-Figs.~~ 10, 11, 13 and 14). The simulations suggest that, as expected, most NO_{x-x} sources are in the production region for oil and gas, though the Pinedale results show that the inventory is missing some anthropogenic sources of NO_{x-x}, especially residential wood burning. The high ~~concentrations-of-NO_x-mixing ratios of NO_x~~ along the bottom of the figures are due to I-80 and not oil and gas infrastructure. Both chemical mechanisms show a
550 similar trend in NO_{x-x} with the build up of NO_{x-concentrations-x mixing ratios} in the morning at 08:00 local time (~~Figures~~ ~~Figs.~~ 10b and 13b) ~~the higher concentrations-, higher mixing ratios~~ at noon local time (~~Figures-Figs.~~ 10c and 13c), ~~which is~~ a few hours before the higher ~~concentrations-mixing ratios~~ of O₃ are simulated, and ~~the lower pollutant concentration-lower~~

pollutant mixing ratios at 16:00 local time (Figures Figs. 10d and 13d) when the O₃ concentrations-mixing ratios are the highest. It is important to note-remember that the simulations underestimates the NO_x-concentrations underestimate the NO_x mixing ratios and the simulated NO_x-concentrations-x mixing ratios do not vary largely among the simulations using different chemical mechanisms. Similar to the diurnal NO_x-profile-variability in NO_x, the diurnal profile-of-variability in VOCs from both simulations (Figures Figs. 11 and 14) also shows a similar trend in the distribution-of-VOCs-in-the-basin, with higher VOC concentrations-mixing ratios occurring a few hours before the higher O₃ concentrations-mixing ratios are simulated. Overall, the simulations capture the diurnal variation of the-O₃ and its precursors reasonably well, however, the simulated concentrations-. However, the simulated mixing ratios of the precursors are lower-low compared to the respective observations, thus warranted further analysis into the sensitivity of simulated O₃ production to the precursor emissions.

4 VOC and NO_x-x Sensitivity Analysis

The results from the baseline simulations using both the MOZART and RACM chemistry mechanisms show that despite the low NO_x-and-VOC-concentrations-found-in-the-model-x and VOC mixing ratios using the NEI2014v2 emissions, both models produce O₃ enhancements during the observed ozone-high O₃ episodes. To understand the potential sensitivity of O₃ formation in the UGRB to VOC and NO_x-x levels, additional simulations were performed by comparing the baseline simulation results to observed VOC and NO_x-levels-x mixing ratios, then adjusting the emissions by the ratio between modeled-and-observed values, and rerunning the model, the observed and modeled values, as outlined in Section 2.7. These adjusted emissions were then used for additional sensitivity simulations. The factors used to adjust the emissions are displayed-shown in Table 5. Of note-is-that The additional simulations for the sensitivity analysis all have dry deposition turned off, similar to the baseline runs. Moreover, the data from the Boulder station during the entire model-run-period-was-entire study period are used to adjust NO_x-concentrations-the NO_x mixing ratios because continuous NO_x-data-is-available-while-VOC data-is-only-available-during the-collection-period-x data are available, as described in Section 2.7. It is clear from Table 5 that significant-enhancements-are required-for-the-model-significantly-underestimates all VOC species, but-especially-large-adjustments-had-to-be-made-to-the-thus requiring large emission enhancement factors to be applied to ensure that the model simulations produce similar VOC mixing ratios to the observations. More specifically, especially large adjustments are required for aromatic species. This result indicates that not only are the modeled VOC emissions too small, -but-that-but also the resulting mixture has less-aromatic-species-lower aromatic species mixing ratios than what is observed, making the modeled VOC mixture significantly less reactive than the observed VOC mixture.

Figure 15 compares the time series of The O₃ concentration at seven monitoring stations among different NO_x-sensitivity simulations -. All simulations were done with the same amount of VOC, which was increased to match observations according to Table 5. The model run with NO_x-and-VOC-increased-mixing ratios from the aforementioned VOC and NO_x sensitivity simulations are shown in Figs. 15 and 16 and compared with data from the seven monitoring stations. Specifically, Fig. 15 shows the results for increased NO_x and VOCs to observed levels-does-increase-ozone-formation-, which does increase O₃ formation but perhaps by less than would have been anticipated given the dramatic increases in VOC emissions and reactivity.

In fact, these runs do ~~an excellent a good~~ job of replicating the ~~ozone-O₃~~ observed at the Boulder site in the RACM model while ~~moderately overpredicting ozone only moderately over-predicting O₃~~ formation with the MOZART ~~chemistry. Doubling the NO_x mechanism. Further, an additional doubling of NO_x~~ causes a large jump in the ~~ozone-O₃~~ predicted at all sites by both the RACM and MOZART chemistry schemes, ~~while cutting the NO_x in half reduces ozone formation. To explore the~~
590 ~~sensitivity to NO_x vs. VOC further, additional simulations were conducted where only NO_x and only VOCs were adjusted from the baseline run (Fig. 16). The simulation where only NO_x was adjusted results in moderate increases in the simulated O₃ mixing ratios at all sites. These model runs and across the entire study period. When NO and NO₂ emissions were kept at their baseline levels and VOC emissions were adjusted, the simulations produce slight increases in O₃ at some sites, especially with the MOZART chemistry scheme, but at other sites the O₃ mixing ratios are not always elevated. Rather, in these cases,~~
595 ~~the timing of O₃ formation changes, with increases seen earlier in the day and sometimes even lower peaks in the modeled O₃ mixing ratio compared with the baseline run. These results are interesting and somewhat unusual, but further analysis was not pursued because by increasing VOCs while not adjusting NO_x, the VOC:NO_x ratio for this model run is far outside what is actually observed in the basin and is thus considered highly unrealistic. All together, these sensitivity simulations strongly suggest that ozone-O₃ formation in the basin is predominantly limited by the NO_x available availability of NO_x rather than~~
600 ~~being controlled mainly by VOC concentrations. The runs the VOC mixing ratios. Further, the results suggest that if more NO_x was available x is available,~~ the basin might see ~~even~~ higher levels of ~~ozone-O₃~~ than currently observed. ~~To further investigate this, the formaldehyde:NO₂ (HCHO:NO₂) ratio for all VOC and NO_x sensitivity simulations is presented in Fig. 17. This ratio has been used in previous studies as a proxy for VOC-limited and NO_x-limited conditions (Liu et al., 2021). Here, the ratio is well above 1 during the high O₃ events for all simulations, with the only decrease being observed for the simulations with~~
605 ~~only increased NO_x emissions. These results further suggest that O₃ formation in the basin is strongly controlled by NO_x availability (Liu et al., 2021).~~

5 Conclusions

Over the past decade, there have been a number of elevated wintertime O₃ events in the UGRB, WY, with ~~concentrations mixing ratios~~ often exceeding 70 ppb. ~~Ozone-These~~ events, though much less severe than observed a decade ago, have continued
610 despite significant efforts to reduce emissions from oil and gas production. This fact drives the need for a photochemical model to better understand what is happening and to determine what emission reductions might effectively reduce ~~ozone-O₃~~. This study, to the ~~the~~-best of the authors' knowledge, is the first to utilize the EPA-NEI2014v2 emissions inventory with a fully coupled meteorology and chemistry model (WRF-Chem) to simulate O₃ events in the UGRB. The utilization of NEI2014v2 is key because this is the first NEI version to integrate non-point oil and gas sources, which are a dominant driver of the
615 ~~ozone formation in teh-O₃ formation in the~~ UGRB. Additionally, this study compared the results of two different chemistry mechanisms (MOZART and RACM), focusing on their ability to ~~replicate the concentrations-reproduce the mixing ratios~~ of O₃. Neither chemistry mechanism can reproduce these high O₃ events without modifying the default surface ~~albedo-of the base-and photolysis albedo in the~~ model. Furthermore, ~~the dry deposition of gas species in RACM needed to be modified to~~

620 ~~better represent slower losses to snow surfaces for this chemistry to create similar amounts of ozone to the MOZART model,~~
~~which differences in dry deposition also affected the simulated accumulation of O₃, where the MOZART scheme accounts for~~
dry deposition changes with snow cover. ~~When dry deposition is turned off in both chemistry schemes, they produce while~~
~~RACM does not. Thus, dry deposition was turned off to reduce inter-model differences, and these simulations produced~~ similar
amounts of ~~ozone~~O₃.

~~The performance of the model~~

625 ~~The model performance with regard to~~ meteorology in the vertical was validated using vertical profiles of observed temper-
ature during two ~~IOP periods from a earlier study~~ ~~IOPs from an earlier period~~ (Feb 28 to Mar 2 and Mar 9 to Mar 12, 2011).
Vertical data ~~is were~~ only available from this time period. Although the simulated temperature ~~is was~~ 2 to 4 °C warmer than the
observed temperature, the simulation captured the inversion layer near the surface. To validate the ~~the~~ model's ability to predict
the surface meteorology, 2-m temperature and wind speed from two WRF-Chem simulations (MOZ17 and RACM17) were
630 compared with the observations at 7 weather stations during the 2017 ~~study~~ period. The simulated 2-m temperature showed a
good correlation with observations at all stations. The simulated periods of low wind speeds also showed good agreement with
the observed calm winds, though variability in the exact magnitude of the low winds results in relatively ~~poor low~~ correlation
coefficients.

To study the model's ability to ~~replicate simulate~~ high O₃ events, we analyzed ~~concentrations mixing ratios~~ of O₃ and its
635 precursors (NO_{x-x} and VOCs). The baseline simulations captured the high O₃ ~~concentrations mixing ratios~~ reasonably well
at most of the stations, even though simulated levels of NO_{x-x} and VOCs were dramatically lower than observations. The
low modeled ~~concentrations of NO_x mixing ratios of NO_x~~ and VOCs suggest that emissions in NEI2014v2 are too low in the
UGRB. Spatial plots of O₃ and its precursors show the predicted spatial extent of O₃ formation and that the models suggest the
monitoring sites are close to, but not at, the location of maximum O₃. Sensitivity studies where the levels of NO_{x-x} and VOCs
640 were ~~adjusted increased~~ to better match observations ~~demonstrate demonstrated~~ that dramatically increasing VOC emissions
and increasing the reactivity of the VOC mixture does not dramatically increase simulated ~~ozone concentrations (though they~~
~~do go up)~~O₃ ~~mixing ratios~~. Rather, the ~~ozone~~O₃ levels appear to be predominantly controlled by NO_{x-x} availability. Because
the RACM chemistry has previously been shown to perform reasonably well at simulating O₃ events in the UB (Ahmadov
et al., 2015) and again performs well in the current study when VOC and NO_{x-x} levels are adjusted to match observations,
645 this study presents the possibility that O₃ might be able to be formed in the UGRB at significantly lower ~~NMHC VOC~~ levels
than are currently observed.

Code and data availability. The WRF and WRF-Chem models are freely available online (<https://github.com/wrf-model/WRF>). The emis-
sion preprocessing tools and NEI emission data can be found at <https://www2.acom.ucar.edu/wrf-chem/wrf-chem-tools-community>. The
WYDEQ data can be obtained from <https://www.wyvisnet.com>.

650 *Author contributions.* SG, ZL, and SM designed the study and conducted the model simulations, analysis, and comparison with observations. TT and SR assisted with the model configuration and setup.

Competing interests. The authors declare that they have no conflict of interest.

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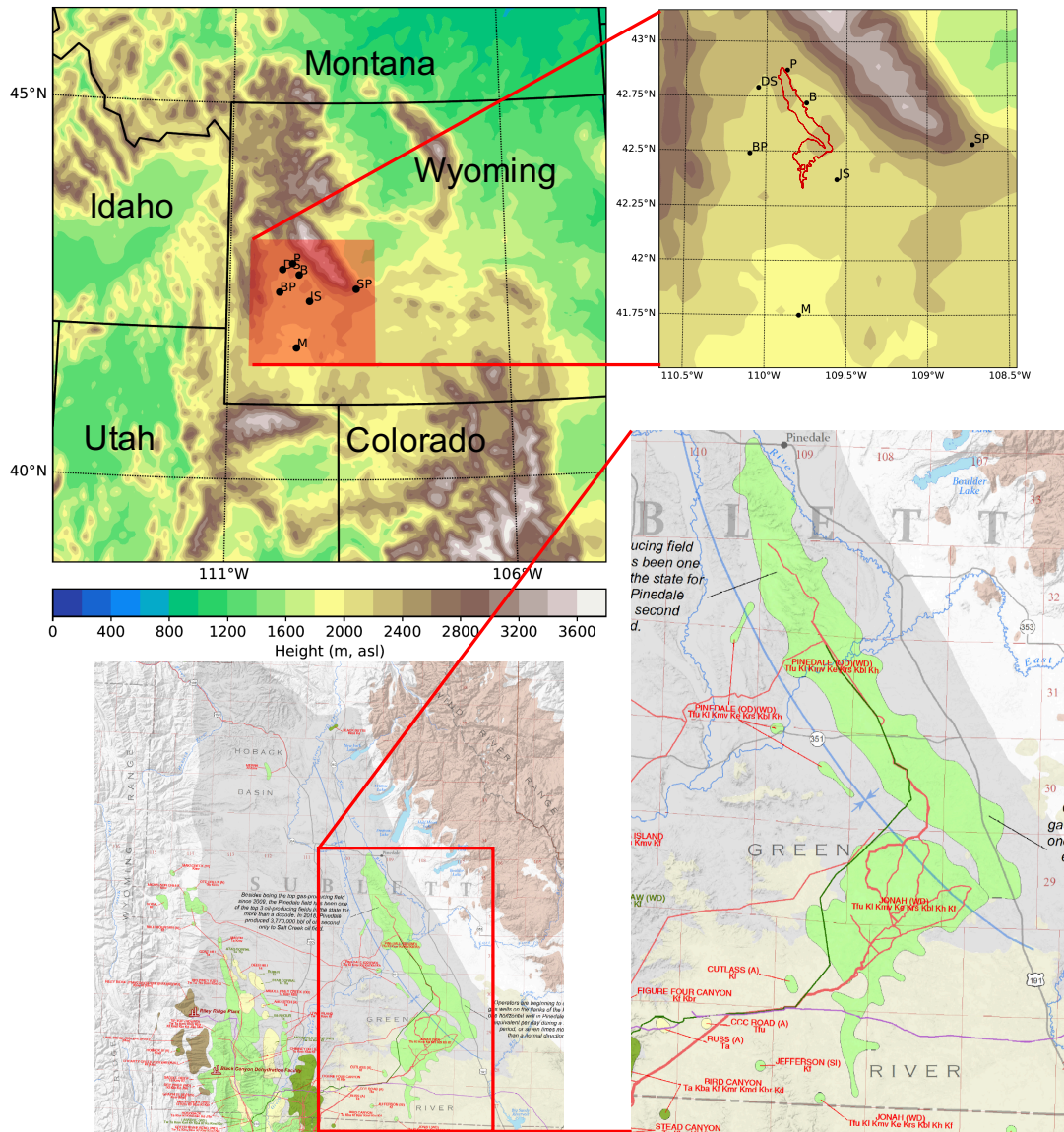


Figure 1. WRF domain (4 km x 4 km grid spacing) with WRF-derived terrain height (upper panels), along with 7 weather and air quality monitoring stations in Upper Green River Basin ([BP - Big Piney](#), [B - Boulder](#), [DS - Daniel South](#), [JS - Juel Spring](#), [M - Moxa Arch](#), [P - Pinedale](#), [SP - South Pass](#); shown by the red box). The red outline on the top-right plot is the approximate location of the Pinedale and Jonah Anticline Fields derived from the WSGS data depicted in the lower panels. The exact locations of the oil and natural gas wells in UGRB are also shown for reference in the bottom panels. The oil and gas facility data depicted in the lower panels are from Toner et al. (2019), ©WSGS.

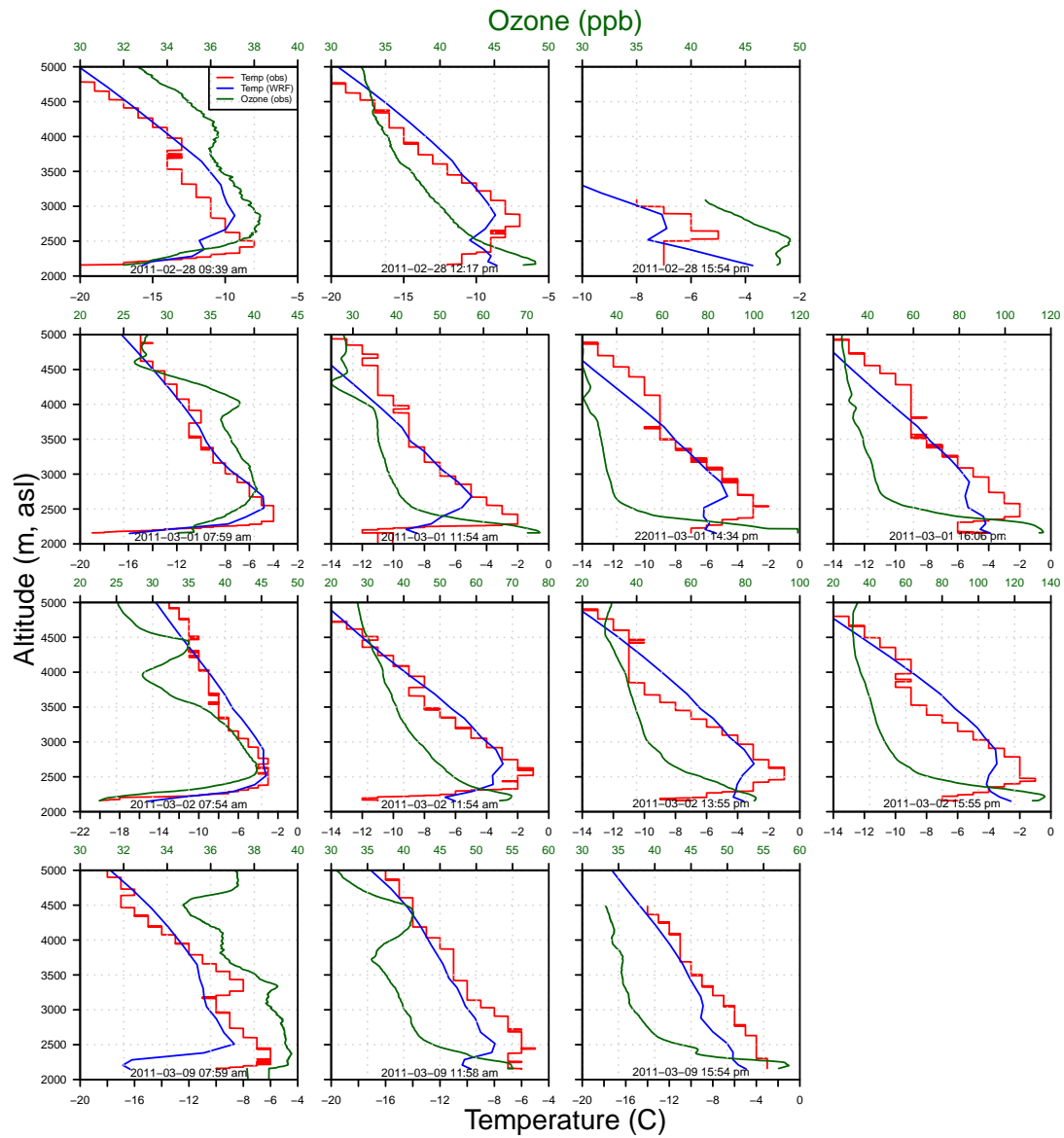


Figure 2. The vertical profile Vertical profiles of O₃ (ppb, green) and temperature (°C, red) from ozonesondes launched in 2011 by WYDEQ compared to WRF-simulated temperature (°C, blue) for 4 days. Each row represents 3-4 ozonesondes launched in one day.

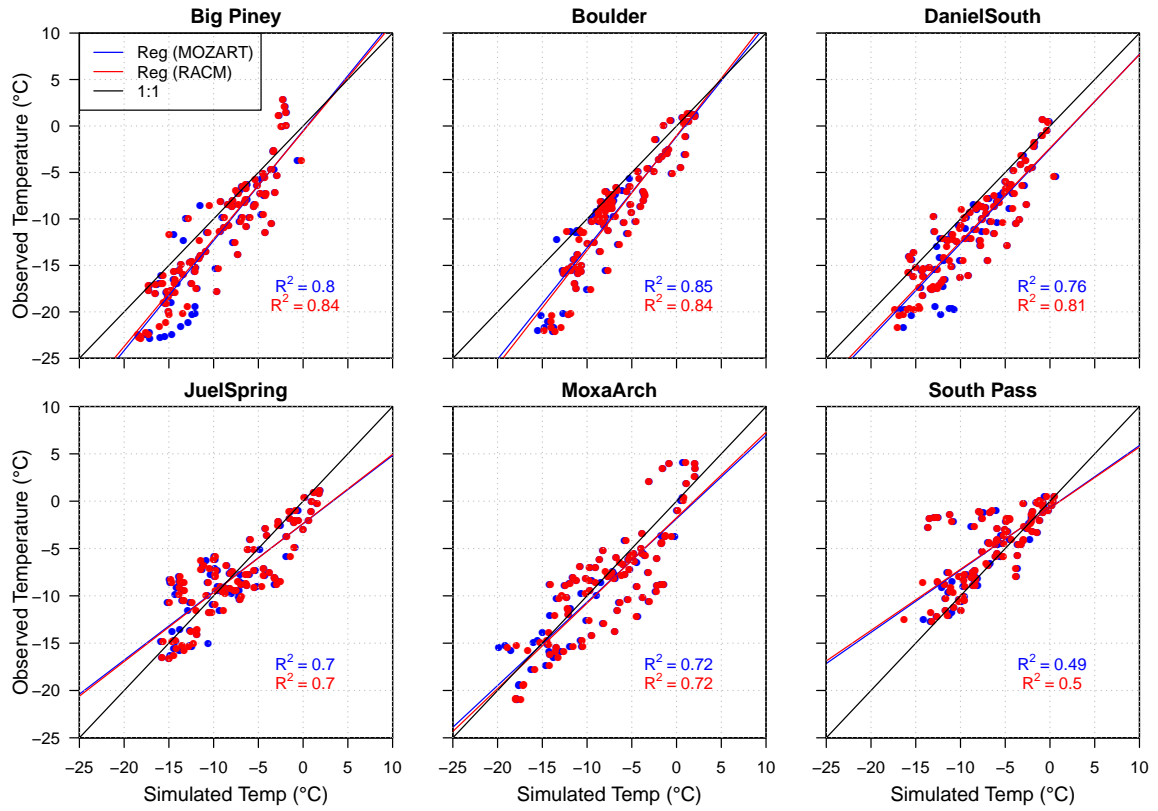


Figure 3. Correlation between recoreded-observed and simulated 2-m temperature at six monitoring stations. The data points and regressing regression line for MOZ17-MOZART are shown in blue and same for RACM17-RACM are shown in red. The one-to-one lines are represented by black lines in each plot.

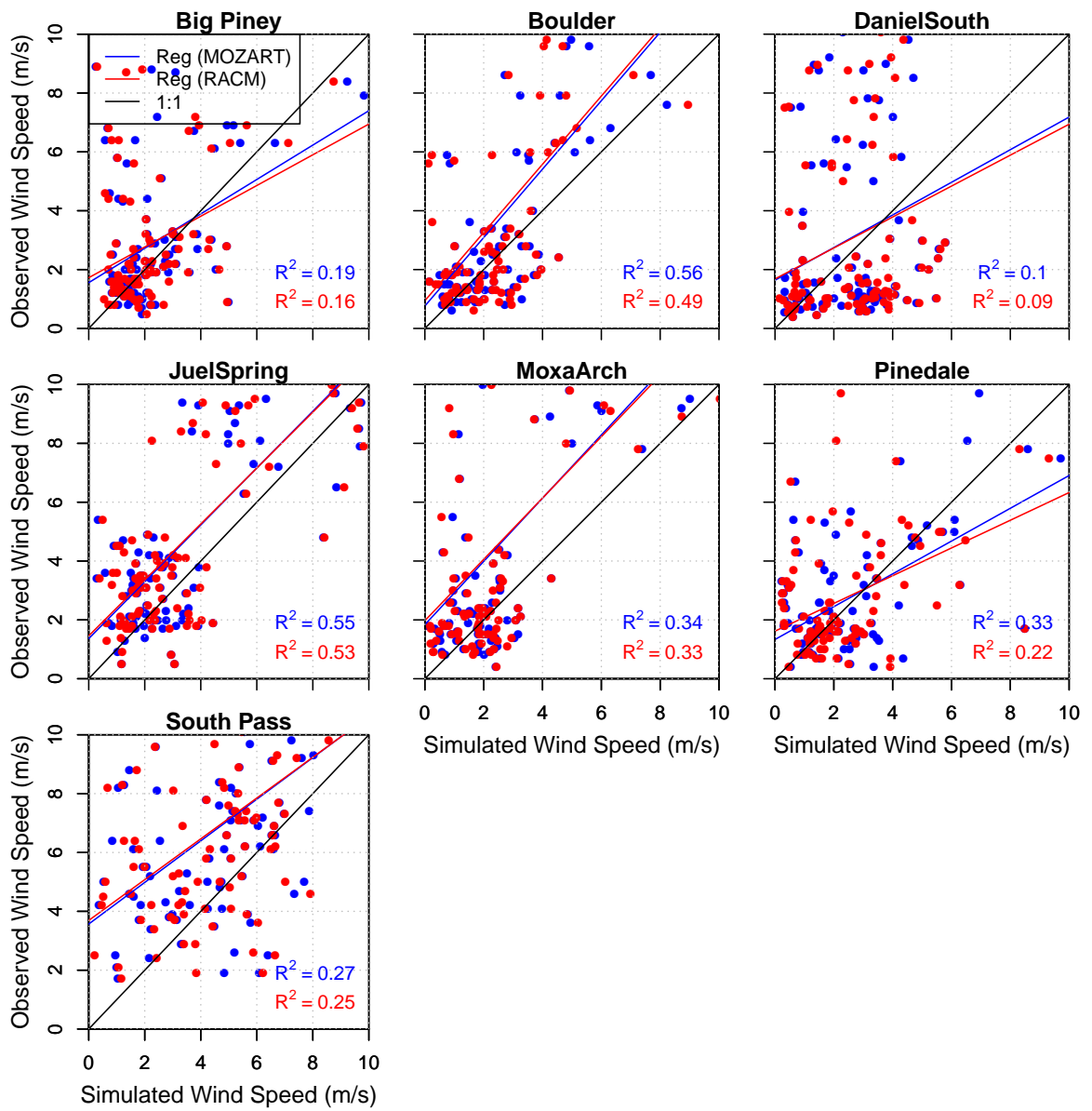


Figure 4. Similar to [Figure Fig. 3](#) but for wind speed.

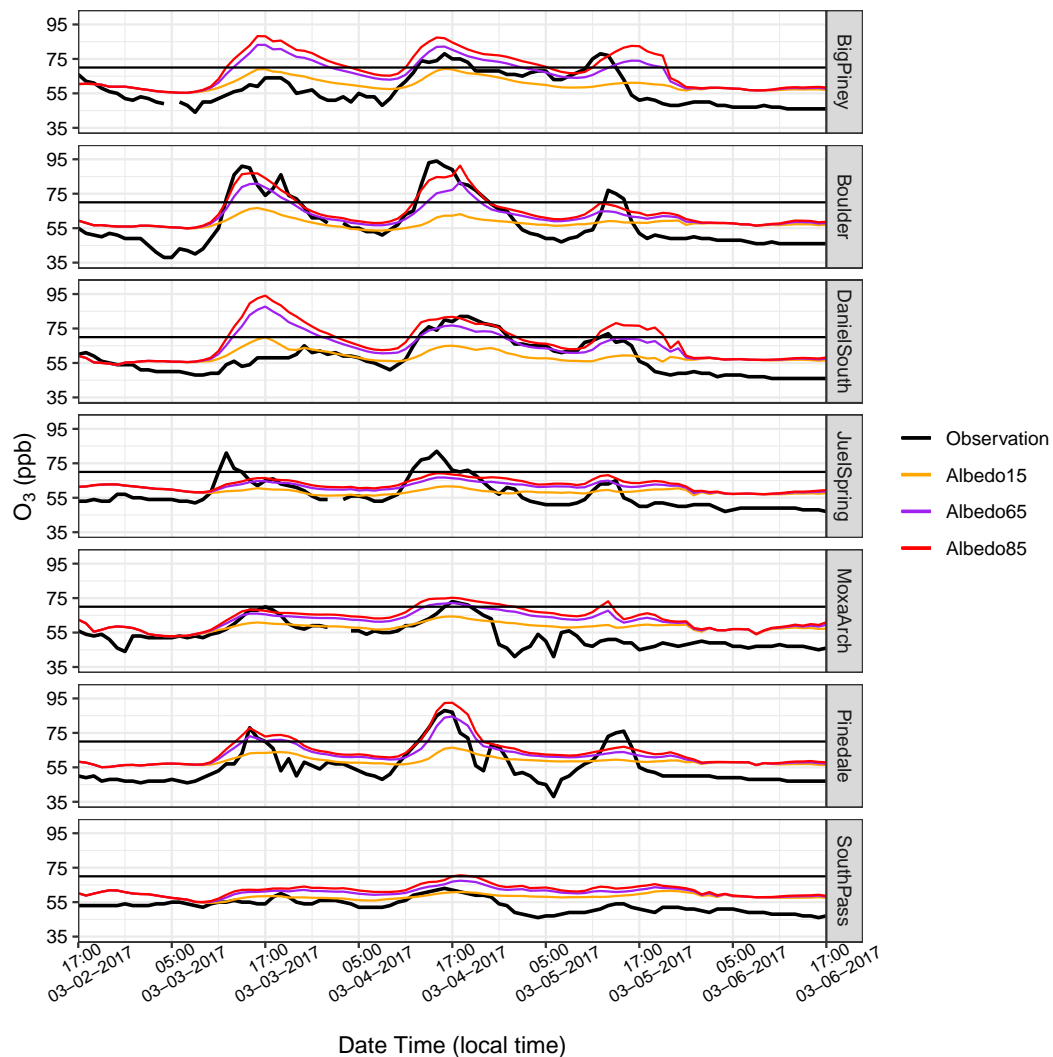


Figure 5. Albedo sensitivity for the WRF-Chem simulation at seven monitoring stations. The observed O_3 concentrations-mixing ratios at each station are shown in black lines, the orange lines represent the results from the default photolysis albedo of 0.15, and the purple and red lines are the modified photolysis albedos of 0.65 and 0.85, respectively. The NAAQS 2015 standard is shown by the black dotted lines on each plot.

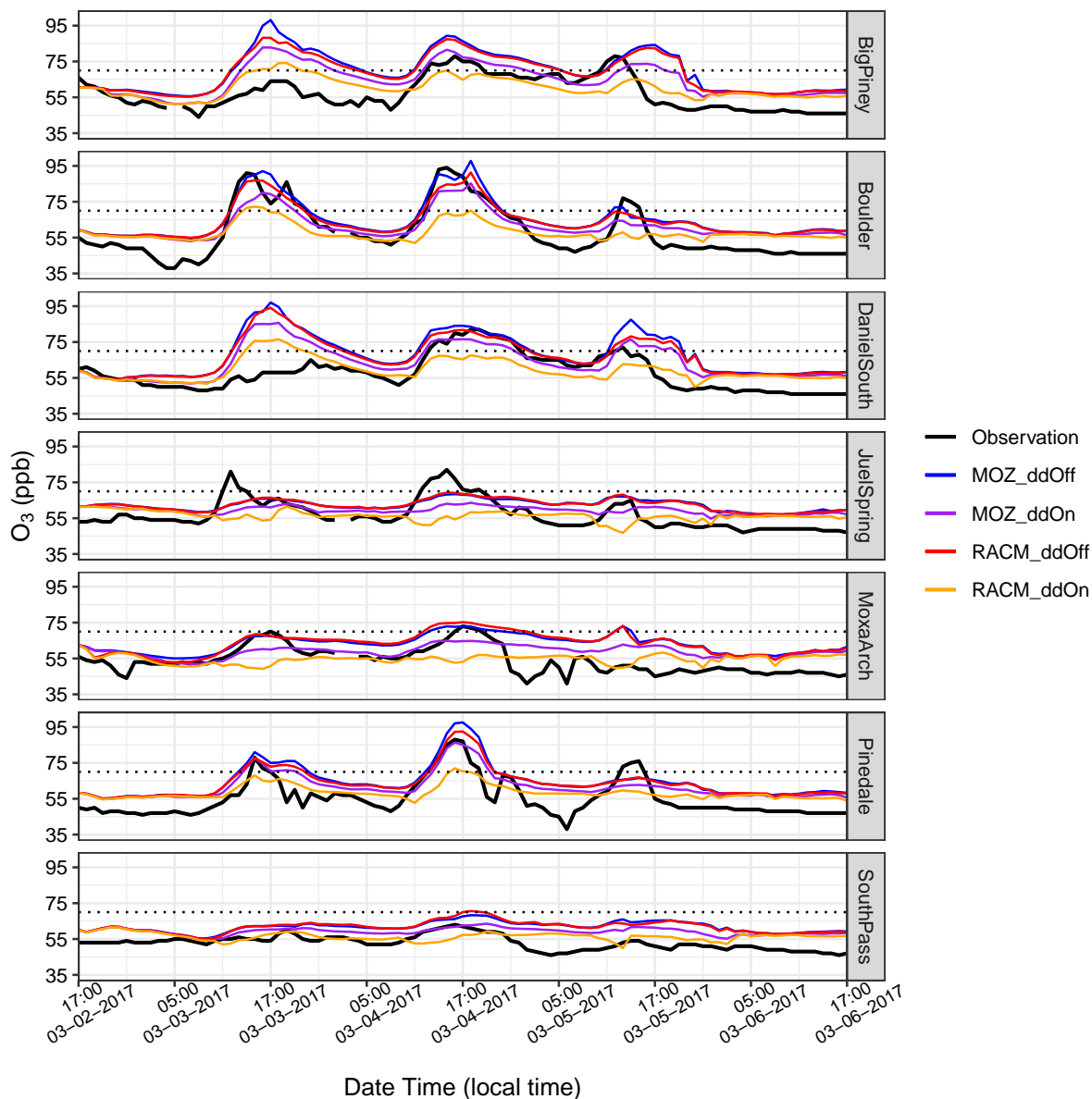


Figure 6. Time series of O_3 concentrations- O_3 mixing ratios at 7 monitoring stations for the time period of Mar 3 to Mar 7, 2017, along with the 8-hour National Ambient Air Quality Standard, 2015 (dotted black lines). **MOZart**-**MOZART** simulation with dry deposition of gas species not included is represented by blue lines and for **Raem**-**RACM** simulation without the dry deposition is represented by red lines. The **MOZart**-**MOZART** and **Raem**-**RACM** simulations with the inclusion of dry deposition of gaseous species are represented by purple and orange lines respectively.

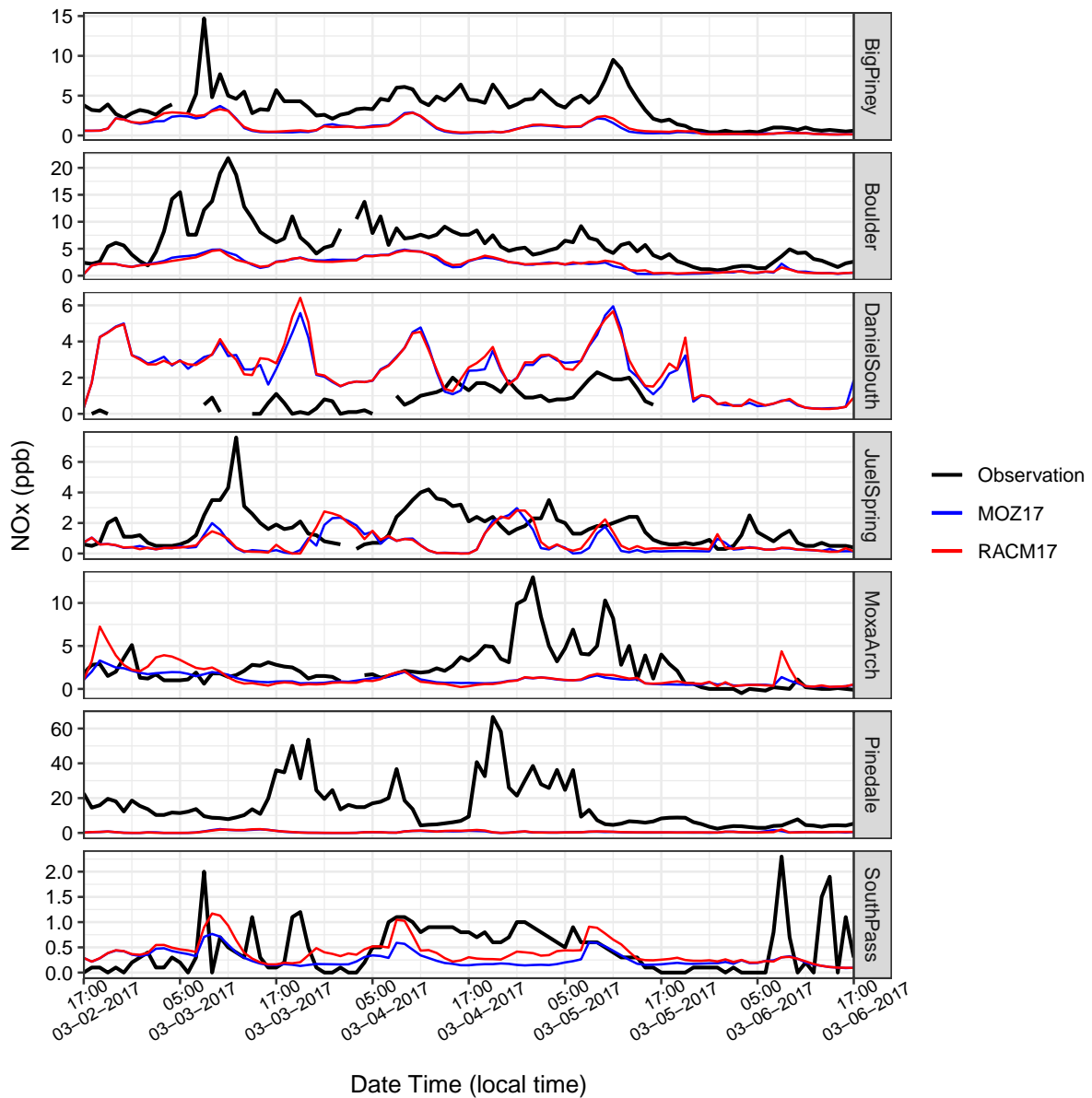


Figure 7. Similar to [Figure Fig. 6](#) but for NO_x concentrations- NO_x mixing ratios (note the different y-scale for each station).

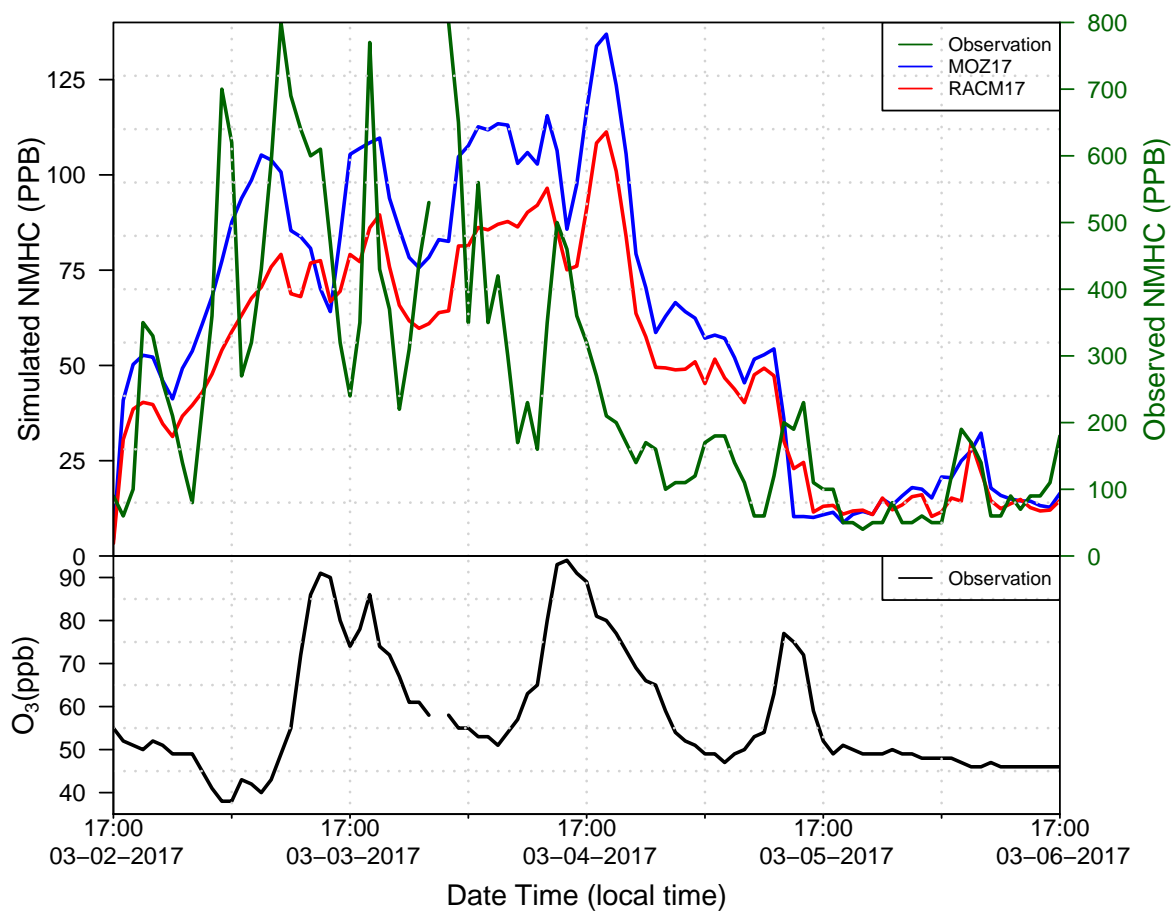


Figure 8. Time series of NMHC (top) and O₃ (bottom) at the Boulder site.

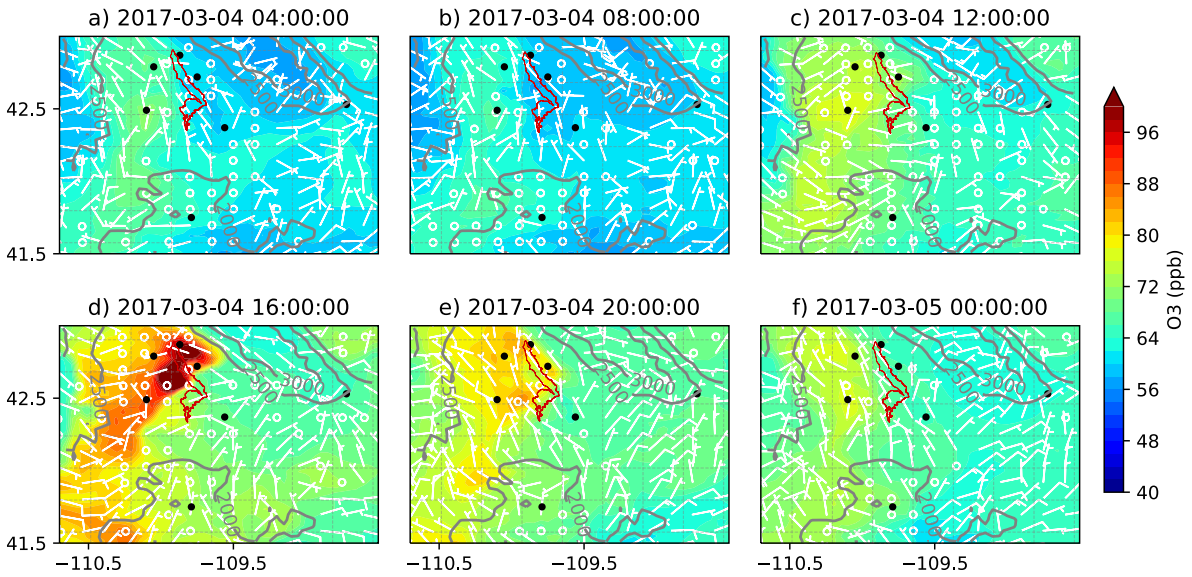


Figure 9. The formation and dissipation of O₃ concentrations over the basin using MOZART chemistry with dry-deposition of gas species turned off (MOZ_ddOff) for the Θ_3 -O₃ event on Mar 04, 2017, starting at 04:00 and ending at 24:00, with an interval of four hours in two consecutive figures. All times in the figure are in local time (UTC - 7 hours). The black dots are the location of the seven WYDEQ stations, and the red outline is an approximate location of the PAJF development.

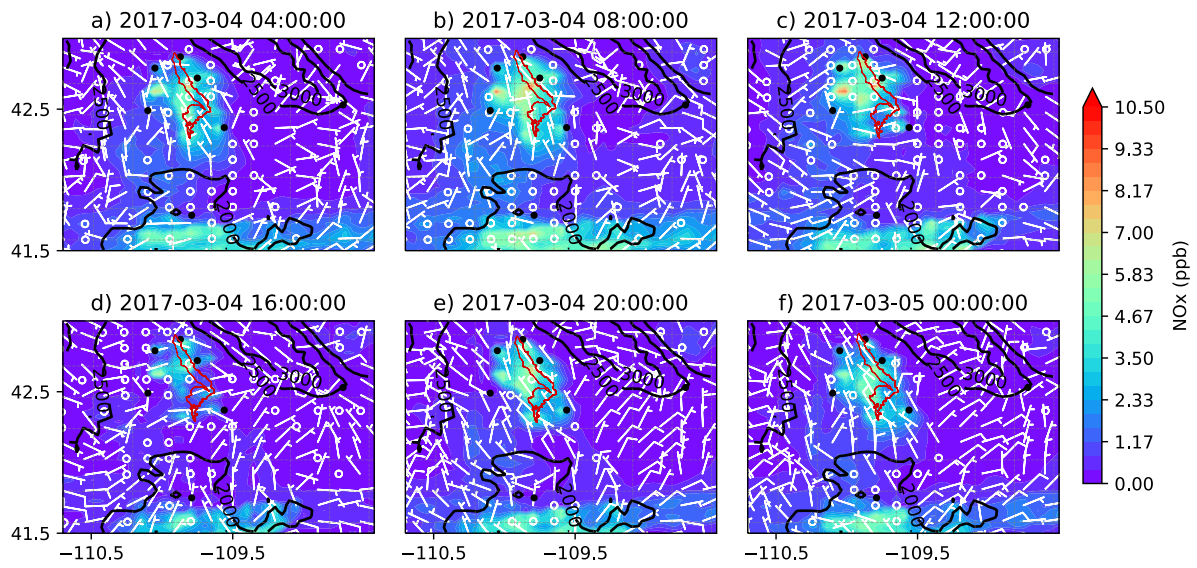


Figure 10. Similar to Figure Fig. 9 but for NO_x concentrations x mixing ratios.

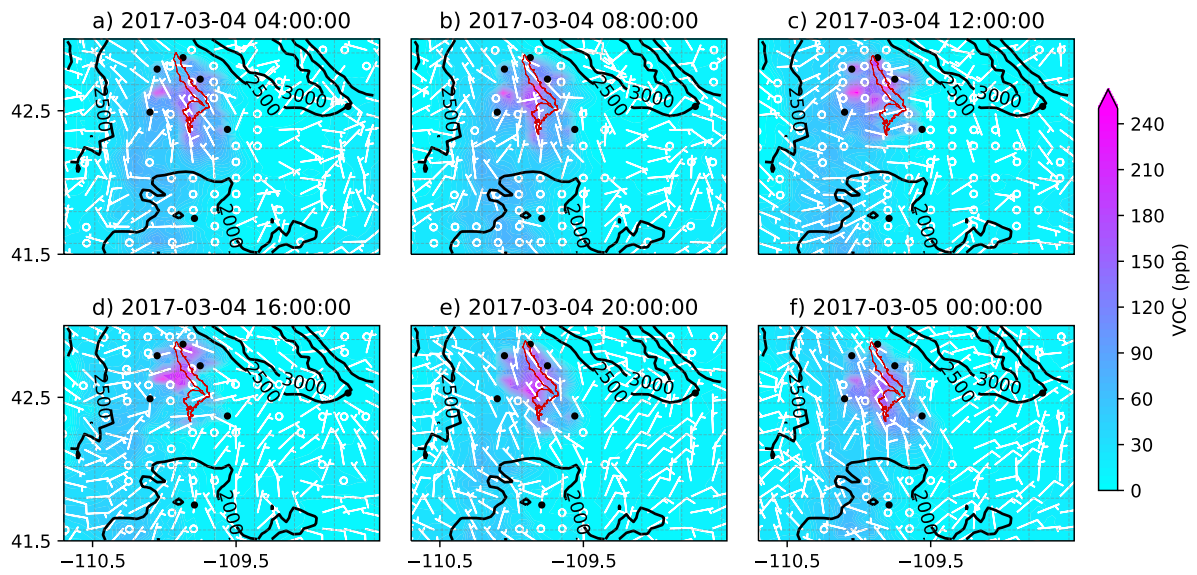


Figure 11. Similar to [Figure Fig. 9](#) but for the [concentrations mixing ratios](#) of VOCs.

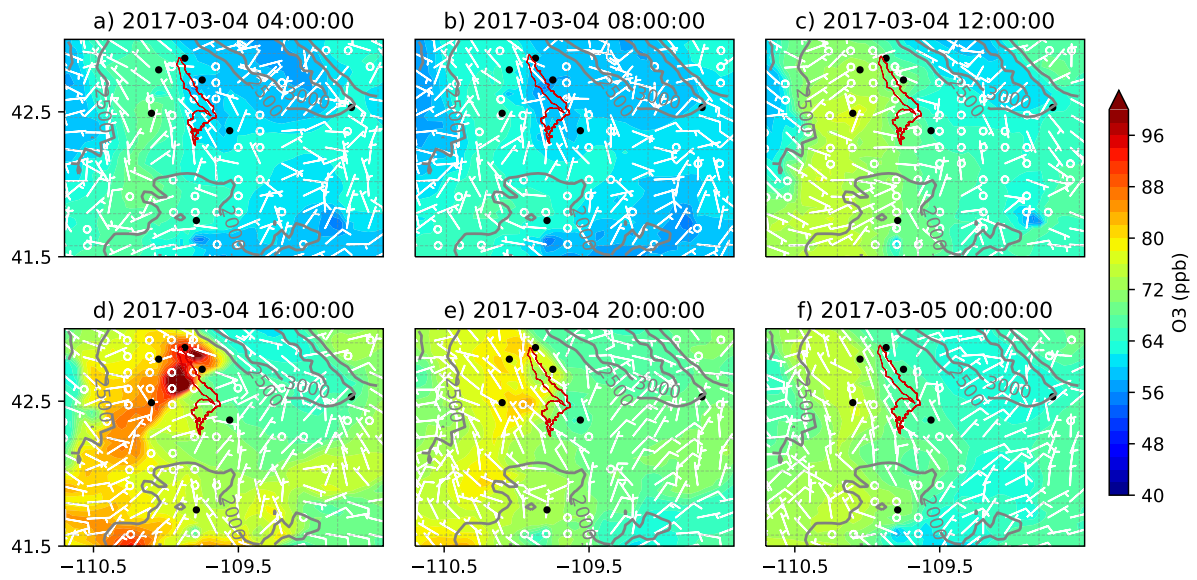


Figure 12. ~~The simulated~~ Simulated O₃ ~~concentrations-mixing ratios~~ over the UGRB using RACM chemistry with dry deposition of gas species turned off (RACM_ddOff) for the O₃ event on Mar 04, 2017.

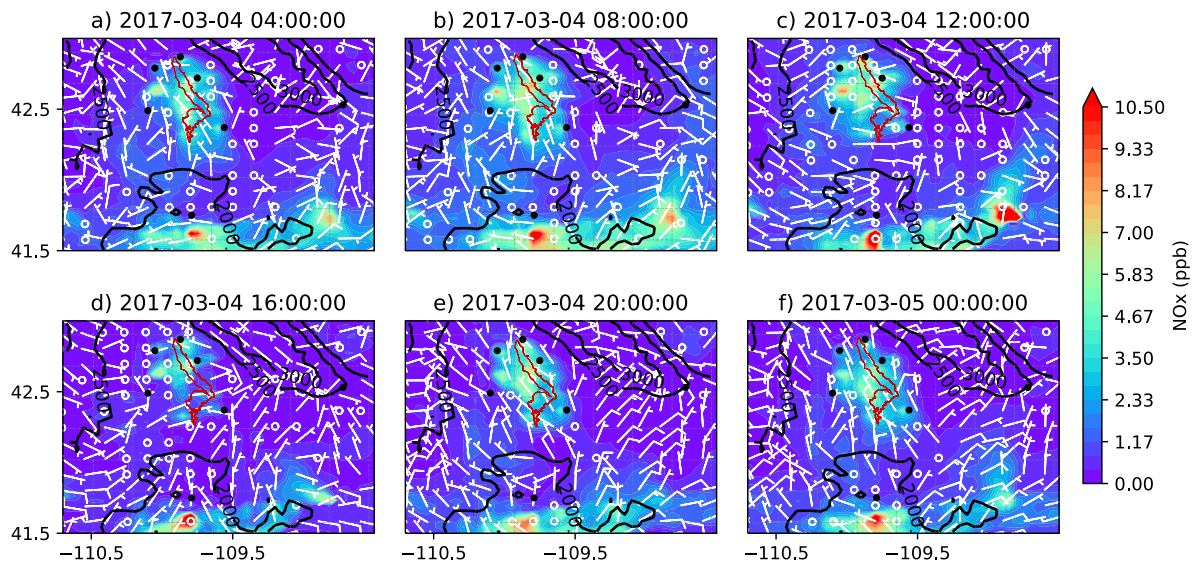


Figure 13. Similar to Figure Fig. 12 but for NO_x concentrations mixing ratios.

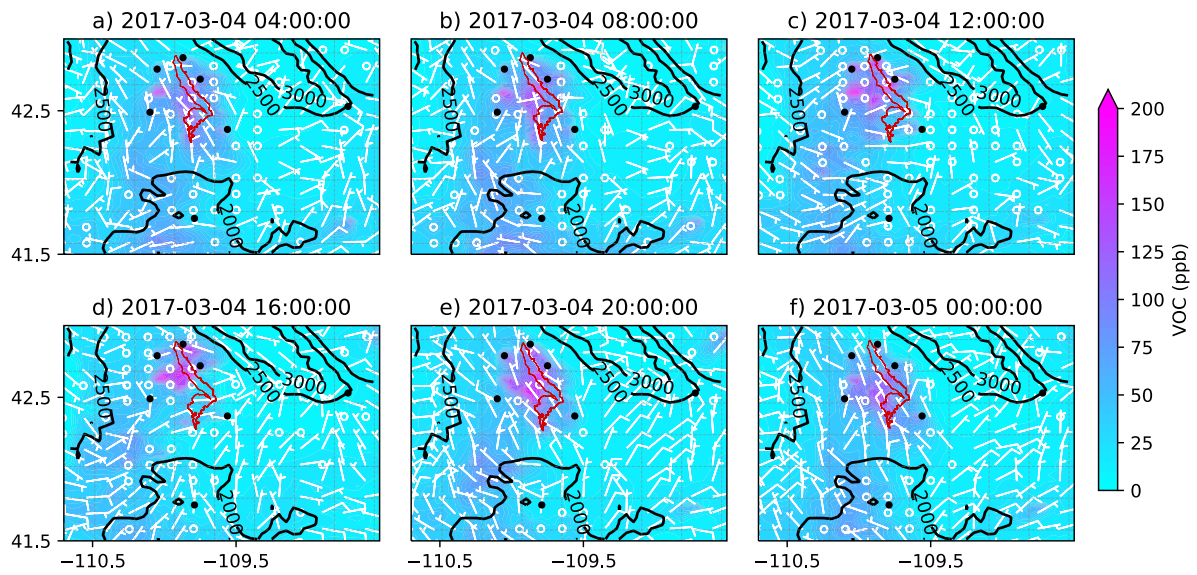


Figure 14. Similar to the [Figure Fig_12](#) but for VOC concentrations [mixing ratios](#).

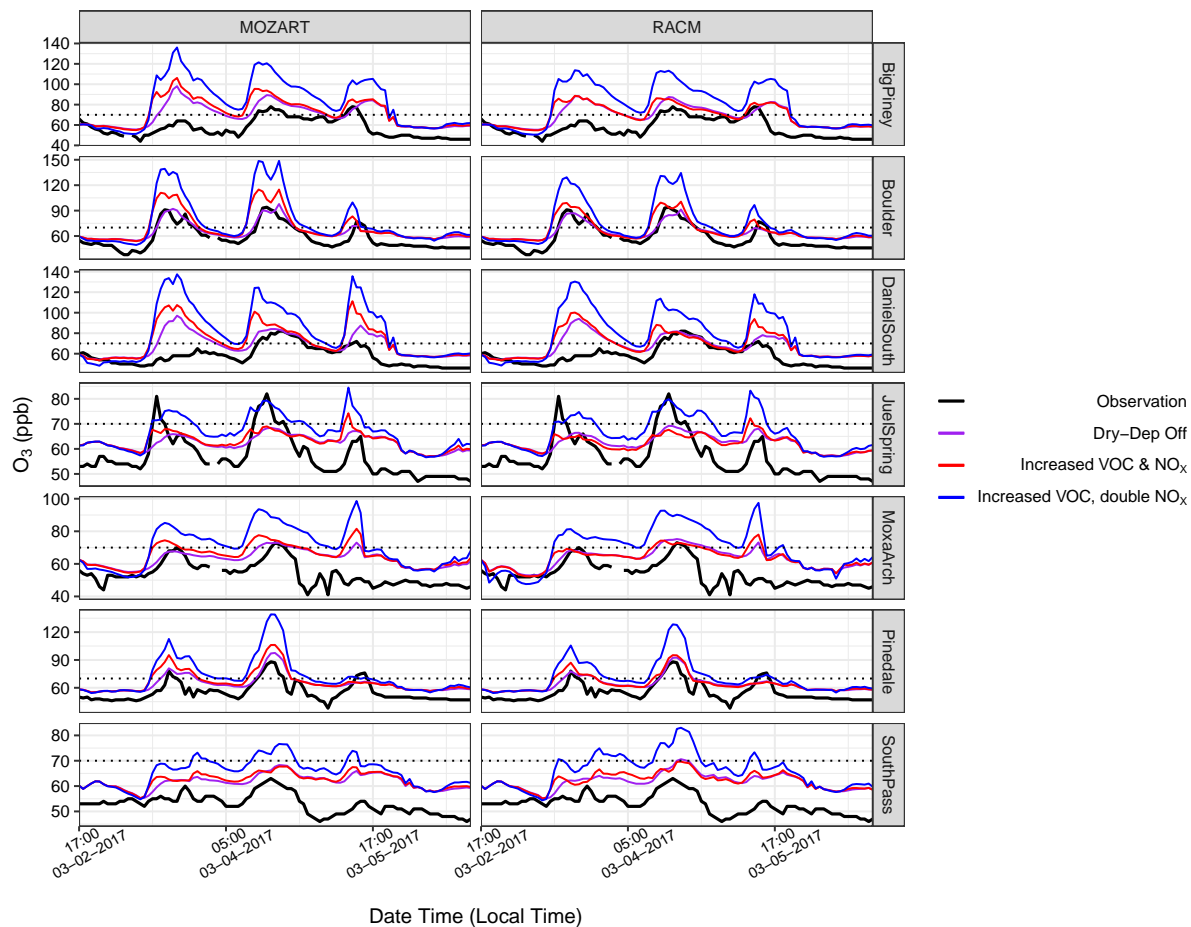


Figure 15. Time series of O_3 concentration-mixing ratios for the VOC and NO_x sensitivity simulation simulations (red, increased VOC and NO_x ; blue, increased VOC, double NO_x) at seven monitoring stations, along with the baseline simulation with dry deposition turned on/off. Note the different Y-scale y-scales for each station.

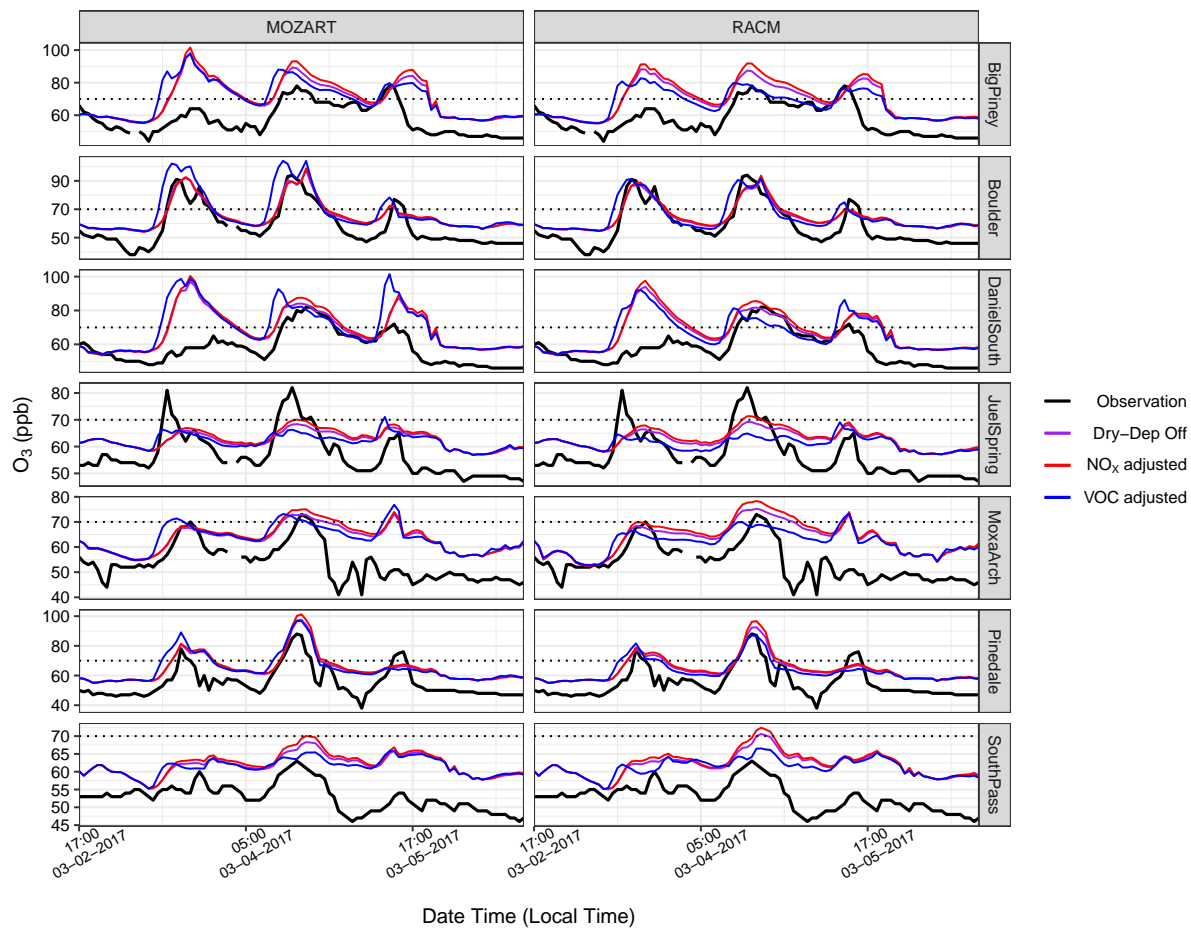


Figure 16. Similar to Fig. 15 but with only NO_x adjusted (red) and only VOCs adjusted (blue), along with the baseline simulation with dry deposition turned off. Note the different y-scales for each station.

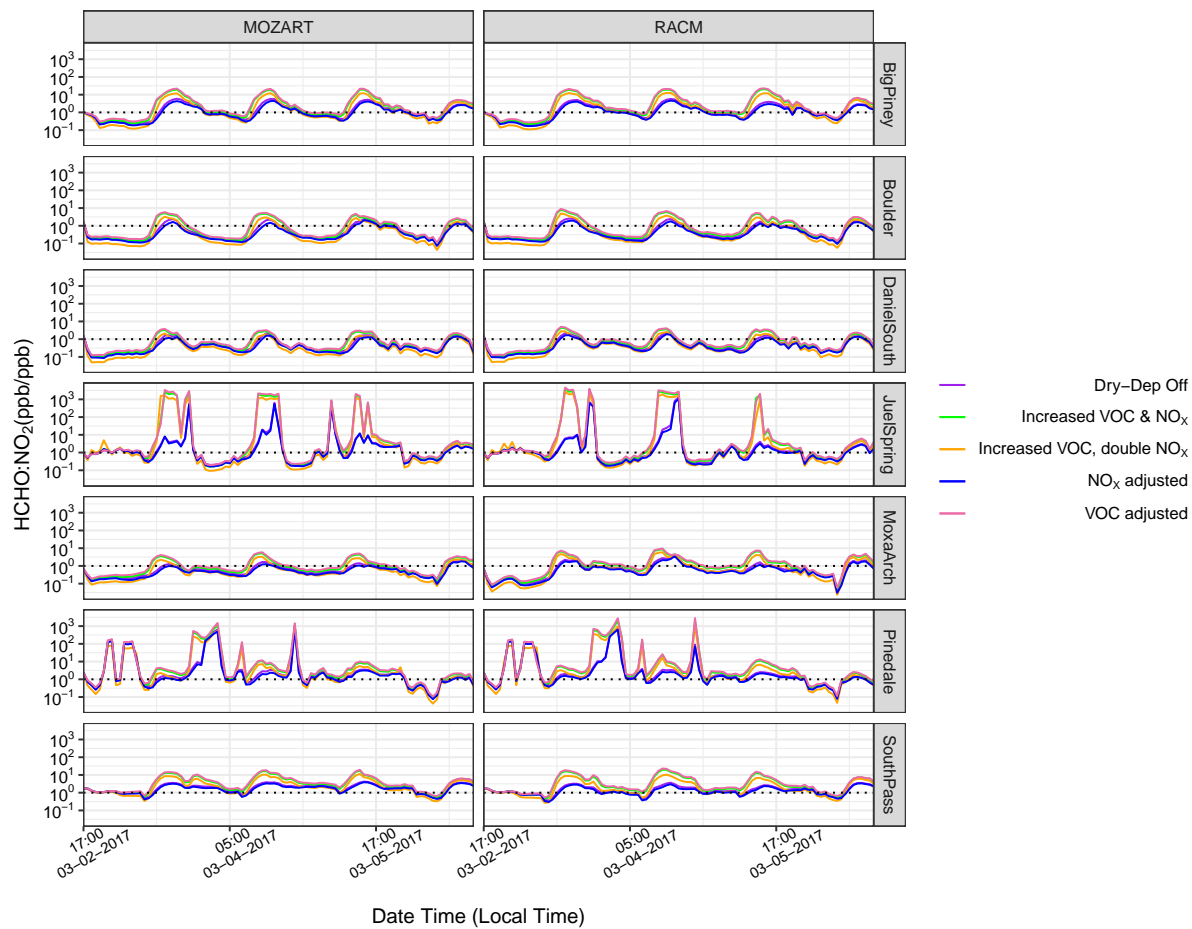


Figure 17. Time series of the HCHO:NO₂ ratio for the baseline simulations with dry deposition turned off (MOZ17 and RACM17; purple lines), as well as all of the sensitivity simulations at the seven monitoring stations. The dotted lines show the HCHO:NO₂ ratio of 1.

List of Tables

Table 1. ~~The coordinates~~ Coordinates and elevations of each weather and monitoring station in the UGRB. (Source: www.wyvisnet.com)

Station	Latitude (°N)	Longitude (°W)	Elevation (ft)
Big Piney	42.49	110.10	6,850
Boulder	42.72	109.75	7,110
Daniel South	42.79	110.05	7,129
Juel Spring	42.37	109.56	7,037
Moxa Arch	41.75	109.79	6,450
Pinedale	42.87	109.87	7,188
South Pass	42.53	108.72	8,287

Table 2. Model configuration for the base WRF and WRF-Chem Simulation

	Details
Boundary Conditions	NARR
Domain Size	800 km x 800 km x 24 km
Time step	12
Horizontal Grid Spacing	4 km (200 points x 200 points)
Vertical Levels	60 (stretched)
Microphysics Scheme	Morrison double-moment scheme (Morrison et al., 2005)
Boundary Layer Scheme	MYJ (Janjić, 1994)
Radiation Scheme (LW and SW)	RRTMG (Iacono et al., 2008)
Land Surface Scheme	Noah-MP (Yang et al., 2011)
Emission Inventory	US EPA NEI-2014 version 2 (US-EPA, 2018)
Chemical Boundary	CAM-CHEM updated every 24h updated every 6h with CAM-CHEM data (Emmons et al., 2020)
Dry deposition of gas species	turned off
have_bcs_chem	gets lateral boundary data from wrfbdy
Photolysis	Madronich TUV photolysis (phot_opt =1 in RACM and phot_opt=4 in MOZART)

Table 3. Temperature Bias (in °C) for the MOZ17 and RACM17 simulations.

	MOZ17 (°C)	RACM17 (°C)
Big Piney	2.29	2.14
Boulder	2.55	2.68
Daniel South	2.62	2.43
Juel Spring	0.18	0.25
Moxa Arch	0.9	0.98
South Pass	-1.53	-1.6

Table 4. ~~The percentage~~ Percentage of the data points that are less than or equal to the given threshold (in m s^{-1}) when the observed wind speed is also less than or equal to the same threshold.

Stations	MOZART			RACM		
	≤ 3.0	≤ 4.0	≤ 5.0	≤ 3.0	≤ 4.0	≤ 5.0
Big Piney	89.47	86.04	84.78	89.47	86.04	84.78
Boulder	98.33	86.84	76.75	90.77	83.54	74.16
Daniel South	79.41	92.31	82.02	77.94	91.14	82.95
Juel Spring	63.16	81.7	89.61	64.28	82.85	87.34
Moxa Arch	81.16	79.75	77.90	80	78.75	77.01
Pinedale	94.12	93.75	98.83	90.14	91.46	96.59
South Pass	38.46	52.94	68.75	45.45	50.00	66.00

Table 5. Data from the canister at Boulder station on Mar 3 2017 collected between 04:00 to 07:00 ~~MST~~local time. Data from the baseline simulation (MOZ17 and RACM17) for the same time period at Boulder.

Species	Observation	Simulated Mixing Ratios		Emission Adjustment Factors	
		MOZART	RACM	MOZART	RACM
Ethene (ppbv)	6.96	6.75	6.02	1	1.15
Ethane (ppbv)	124.4	34.06	30.7	3.6	4
Propane (ppbv)	46.28	17.31	10.86 <u>8.211</u>	2.7	2.2 <u>10.86</u>
Alkane (ppbv)	38.61 ^a	13.76 ^b	12.41	2.8	3.1
Benzene (ppbv)	4.95	0.12	-	40	-
Toluene (ppbv)	6.5	0.12	0.11	52	72
Propene (ppbv)	1.77	0.04	0.04	44	44
Xylene (ppbv)	5.65 ^c	0.029	0.031	194	182.25
NO (ppb)	0.42	0.38	0.39	1.10	1.07
NO ₂ (ppb)	5.71	1.90	1.89	3	3.01

^a~~Sum of i-Butane, n-Butane, i-Pentane, n-Pentane, 2-Methylpentane, 3-Methylpentane, n-Hexane, 2,4-Dimethylpentane, 3-Methylhexane, 2,2,4-Trimethylpentane, n-Heptane, 2-Methylheptane, n-Octane, n-Nonane, n-Decane and Undecane~~

^b~~Sum of i-Butane, n-Butane, i-Pentane, n-Pentane, 2-Methylpentane, 3-Methylpentane, n-Hexane, 2,4-Dimethylpentane, 3-Methylhexane, 2,2,4-Trimethylpentane, n-Heptane, 2-Methylheptane, n-Octane, n-Nonane, n-Decane and Undecane~~

^cSum of m,p-Xylene, o-Xylene, 1,3,5-Trimethylbenzene and 1,2,4-Trimethylbenzene

Appendix A: ~~Data and methods~~

1 ~~Comparison with Mobile Laboratory Data~~

Appendix A: Chemistry Namelist Options Used for MOZART and RACM Chemistry Mechanism

760 Figures A1 and A2 present additional details regarding the WRF-Chem namelist options used in the MOZART and RACM simulations.

```
&chem
  kemit                = 1,
  chem_opt             = 202,
  bioemdt             = 30,
  photdt              = 30,
  chemdt              = 0,
  io_style_emissions  = 2,
  emiss_inpt_opt      = 102,
  emiss_opt           = 10,
  emiss_opt_vol       = 0,
  chem_in_opt         = 1,
  phot_opt            = 4,
  gas_drydep_opt      = 0,
  aer_drydep_opt      = 0,
  bio_emiss_opt       = 0,
  dust_opt            = 0,
  dmsemis_opt         = 0,
  seas_opt            = 0,
  gas_bc_opt          = 1,
  gas_ic_opt          = 1,
  aer_bc_opt          = 1,
  aer_ic_opt          = 1,
  gaschem_onoff       = 1,
  aerchem_onoff       = 0,
  wetscav_onoff       = 0,
  cldchem_onoff       = 0,
  vertmix_onoff       = 1,
  chem_conv_tr        = 0,
  conv_tr_wetscav     = 0,
  conv_tr_aqchem      = 0,
  biomass_burn_opt    = 0,
  have_bcs_chem       = .true.,
  aer_ra_feedback     = 0,
  aer_op_opt          = 0,
  opt_pars_out        = 0,
  diagnostic_chem     = 2,
  has_o3_exo_coldens = .true.
/
```

Figure A1. Namelist for chemistry options used for the simulation using MOZART chemistry mechanism.

```
&chem
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  chem_in_opt   = 1,
  gaschem_onoff = 1,
  aerchem_onoff = 0,
  vertmix_onoff = 1,
  chem_conv_tr  = 0,
  gas_drydep_opt = 0,
  aer_drydep_opt = 0,
  diagnostic_chem = 2,
  chemdt        = 0,
  bioemdt       = 30,
  emiss_inpt_opt = 1,
  emiss_opt     = 3,
  kemit         = 10,
  io_style_emissions = 2,
  aircraft_emiss_opt = 0,
  bio_emiss_opt = 0,
  phot_opt     = 1,
  photdt       = 30,
  wetscav_onoff = 0,
  cldchem_onoff = 0,
  conv_tr_wetscav = 0,
  conv_tr_aqchem = 0,
  seas_opt     = 0,
  dust_opt     = 0,
  dmsemis_opt  = 0,
  biomass_burn_opt = 0,
  have_bcs_chem = .true.,
  gas_bc_opt   = 1,
  gas_ic_opt   = 1,
  aer_bc_opt   = 1,
  aer_ic_opt   = 1,
  aer_ra_feedback = 0,
  opt_pars_out = 0,
/
```

Figure A2. Namelist for chemistry options used for the simulation using RACM chemistry mechanism.

Appendix B: Sensitivity Analysis

Table B1 presents ratios of observed VOCs and NO_x to simulated values at the four stations with speciated VOC observations. Note that the NO_x ratios are computed for the duration of the model simulation as shown in Fig. B1, whereas the ratios for the VOCs are only based on simulated means during the 4-hour observation time window.

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Table B1. Emission adjustment factor for four stations calculated based on the speciated canister data on Mar 3 2017 collected between 04:00 to 07:00 local time and baseline simulation (MOZ17 and RACM17).

	Big Piney		Boulder		Juel Spring		Moxa Arch	
	MOZART	RACM	MOZART	RACM	MOZART	RACM	MOZART	RACM
Ethene (ppbv/ppb)	0.47	0.42	1	1.15	21.7	37.5	19.1	22.84
Ethane (ppbv/ppb)	2.12	1.9	3.6	4	12.77	19.1	10.2	15.85
Propane (ppbv/ppb)	2.06	6.7	2.7	10.86	12.10	47.9	8	39.51
Alkane (ppbv/ppb)	1.73	1.54	2.8	3.1	8.2	10	4.26	5.26
Benzene (ppbv/ppb)	16.07	-	40	-	10.5	-	20.5	-
Toluene (ppbv/ppb)	18.05	21.82	52	72	15.32	25.9	13.5	20
Propene (ppbv/ppb)	22.7	18.16	44	44	61.67	51.5	-	-
Xylene (ppbv/ppb)	74.77	55.32	194	182.25	158	80.9	79.5	66.31
NO (ppb/ppb)	2.24	2.32	1.10	1.07	3.4	3.43	1.54	1.66
NO2 (ppb/ppb)	3.7	3.37	3	3.01	2.7	2.4	2.7	2.1

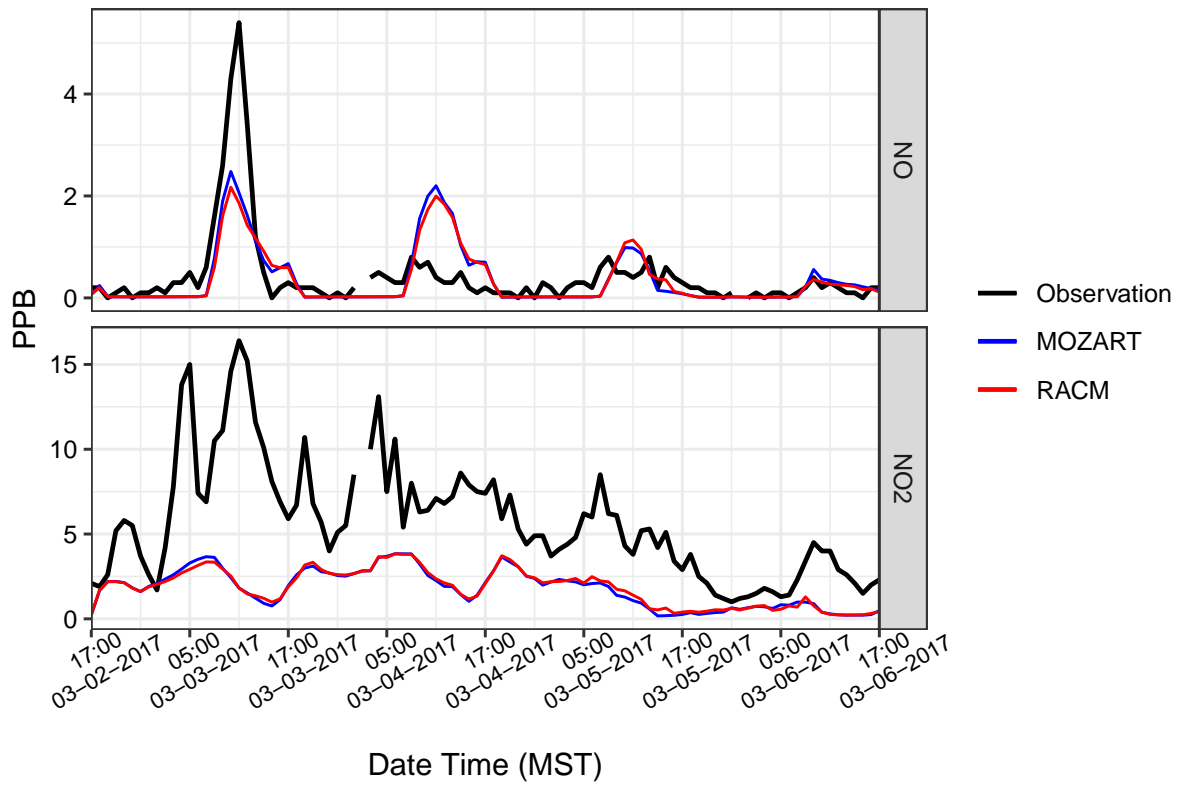


Figure B1. Time series of observed and modeled NO and NO₂ at Boulder.

Appendix C: Additional Data for Pinedale

To further example the model–observation disparity at the Pinedale location, which we attribute largely to wood burning emissions that are not included in the NEI dataset, Fig. compared observed PM_{2.5} and NO_x at Pinedale. The strong correlation is a good indicator that the high NO_x levels are related to the burning of wood that also causes enhanced PM_{2.5}.

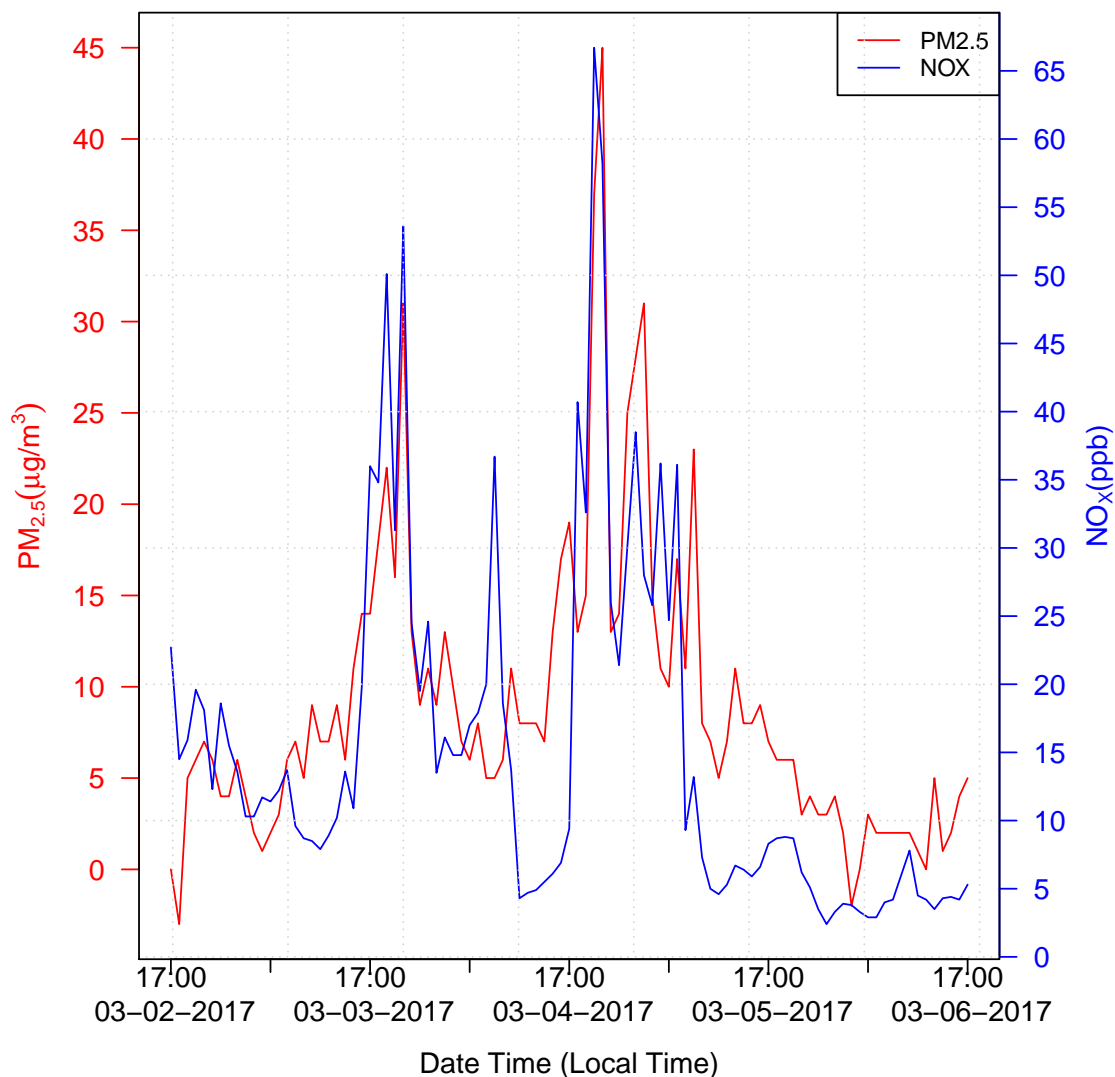


Figure C1. Time series of observation of PM_{2.5} and NO_x at Pinedale.

770 **Appendix D: Comparison with Mobile Laboratory Data**

Methane (CH₄) data from a Picarro Cavity Ringdown Spectrometer (CRDS; model G2204) on-board University of Wyoming mobile laboratory Robertson et al. (2020) were used to validate the CH₄ ~~concentrations from the Wyoming Department of Environmental Quality (WYDEQ)~~ mixing ratios from the WYDEQ Boulder station. The CRDS was modified by Picarro Inc. to sample at ~~2-Hz~~ 2 Hz. The National Institute of Standards and Technology (NIST) traceable ($\pm 1\%$) CH₄ in an ultrapure air mixture with a CH₄ ~~concentration~~ mixing ratio of 2.576 ppm was used to calibrate the Picarro instrument Robertson et al. (2020).

Due to data availability, we compared the hourly CH₄ data from WYDEQ with the 1-s data from the UW mobile laboratory. ~~The data were from~~, as shown in Fig. D1. Only data from the time period over which the UW mobile lab was in the UGRB are shown, corresponding to 11:00 am to 8:00 pm local time; ~~the time period when UW mobile laboratory was driving in and~~ around the UGRB.

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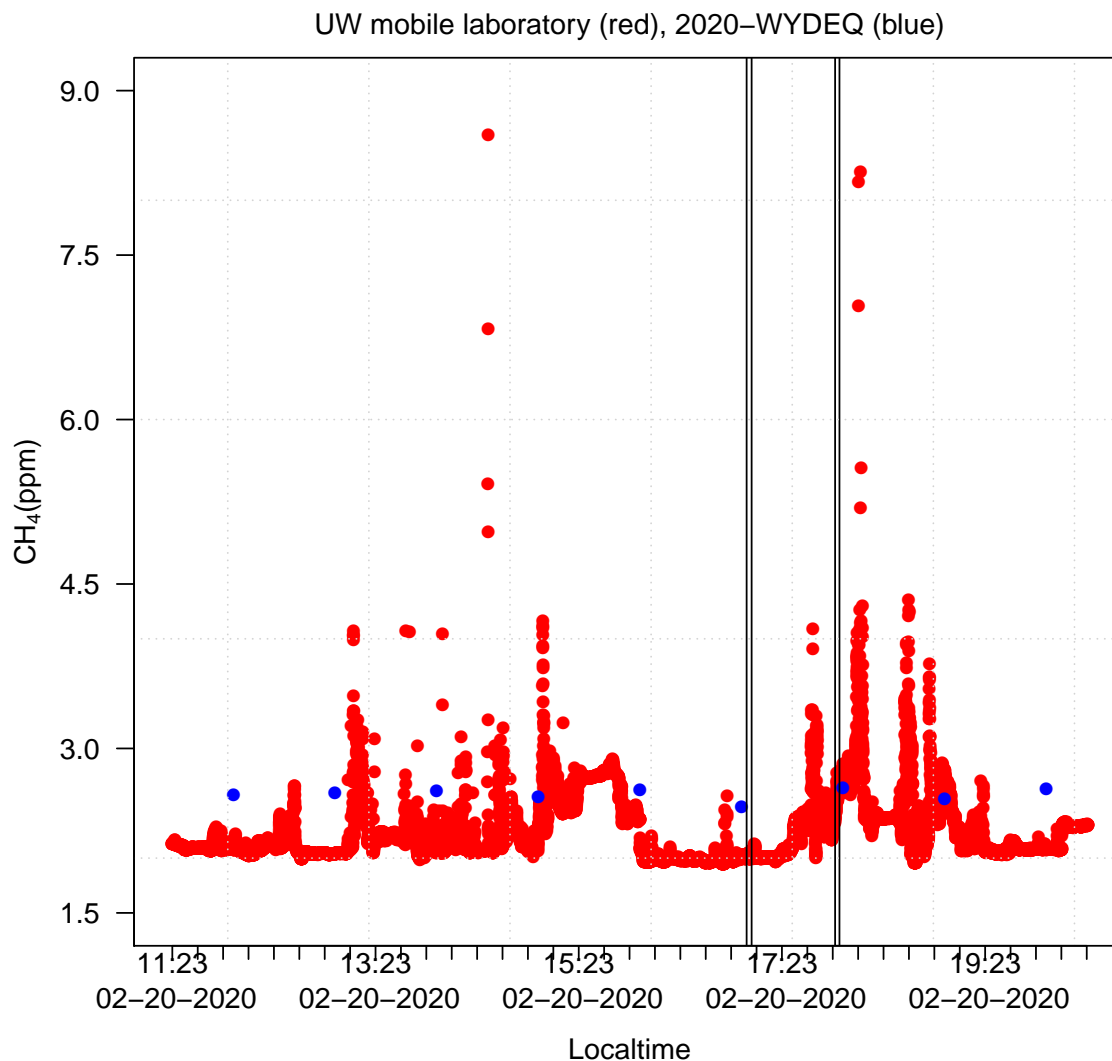


Figure D1. Time series comparison of CH₄ from the UW mobile laboratory (red) and WYDEQ Boulder site (blue) for Feb 20, 2020. The black vertical lines mark the times when the mobile laboratory was passing through the WRF grid box where the WYDEQ Boulder site is located.

D1 ~~Chemistry namelist options used for MOZART and RACM chemistry mechanism~~

~~Namelist for chemistry options used for the simulation using MOZART chemistry mechanism.~~

~~Namelist for chemistry options used for the simulation using RACM chemistry mechanism.~~

Appendix E: Supplemental Figures

- 785 ~~Time series of observed and modeled NO and NO₂ at Boulder.~~
~~Time series of observation of PM_{2.5} and NO_x at Pinedale.~~
~~Time series of observed and modeled NO and NO₂ at Boulder.~~