

1 **Measurement report: characterization and sources of the ~~ambient~~ secondary organic carbon in a**
2 **Chinese megacity over five years from 2016 to 2020**

3 Meng Wang¹, Yusen Duan², Wei Xu³, Qiyuan Wang⁴, Zhuozhi Zhang¹, Qi Yuan¹, Xinwei Li¹, Shuwen Han¹, Haijie
4 Tong¹, Juntao Huo², Jia Chen², Shan Gao⁵, Zhongbiao Wu⁶, Long Cui⁴, Yu Huang⁴, Guangli Xiu⁷, Junji Cao^{4, 8},
5 Qingyan Fu^{2, *}, Shun-cheng Lee^{1, *}

6 ¹Department of Civil and Environmental Engineering, The Hong Kong Polytechnic University, Hung Hom, Hong
7 Kong

8 ²Shanghai Environmental Monitoring Center, Shanghai, China

9 ³School of Physics, Ryan Institute's Centre for Climate & Air Pollution Studies, and Marine Renewable Energy Ireland,
10 National University of Ireland Galway, University Road, Galway, H91 CF50, Ireland

11 ⁴State Key Laboratory of Loess and Quaternary Geology, Institute of Earth Environment, Chinese Academy of
12 Sciences, Xi'an 710061, China

13 ⁵Zhejiang Tianlan Environmental Protection Technology Co., Ltd., Hangzhou 311202, China

14 ⁶Department of Environmental Engineering, Zhejiang University, 866 Yuhangtang Road, Hangzhou, 310058, China

15 ⁷State Environmental Protection Key Laboratory of Environmental Risk Assessment and Control on Chemical Process,
16 School of Resources and Environmental Engineering, East China University of Science and Technology, Shanghai
17 200237, China

18 ⁸Key Laboratory of Middle Atmosphere and Global Environment Observation, Institute of Atmospheric Physics,
19 Chinese Academy of Sciences, Beijing 100029, China

20
21 *Correspondence to:* shun-cheng.lee@polyu.edu.hk (S.C. Lee) and qingyanf@sheemc.cn (Q.Y. Fu).

22 **Abstract**

23 To investigate impact factors and source area of secondary organic aerosols in the Yangtze River Delta (YRD) region,
24 a world-class urban agglomeration in China, long-term measurements of organic carbon (OC) and elementary carbon
25 (EC) in particulate matter of less than 2.5 μm (PM_{2.5}) with hourly time resolution were conducted at a regional site in
26 Shanghai from 2016 to 2020. Based on the five-year measurements, the interannual, monthly, seasonal, and diurnal
27 variations in OC and EC, as well as OC subtypes, i.e., secondary OC (SOC) and primary OC (POC), apportioned by
28 the novel statistical model of the minimum R² method, and the formation pathways of SOC, are presented. By
29 examining the relationship between SOC and temperature, as well as relative humidity (RH), we show that SOC
30 formation is greatly enhanced at high temperatures (>30 °C), while it is inversely correlated with RH. In particular,
31 we show that the photochemical formation of SOC is the major formation pathway even in winter when solar radiation
32 was supposedly less intense than in summer, which is different from that in north China plain where aqueous phase
33 chemistry is found to be an important SOC formation pathway. Moreover, increased SOC concentrations are also
34 found to be associated with high wind speed (>5 m s⁻¹) in winter, which is increased by 29.1% (2.62 $\mu\text{g m}^{-3}$) when
35 compared to that during lower winds, suggesting regional sources of SOC in winter. By analyzing the potential source
36 regions using the concentration weighted trajectory (CWT), the geographic regions of SOC are found to be mainly
37 associated with transport from outside Shanghai (SOC > 3.5 $\mu\text{g m}^{-3}$) including central and southern Anhui, Zhejiang,
38 and Fujian. The results from this study provide critical information about the long-term trend of carbonaceous aerosol,

39 in particular SOC, in one of the largest megacities in the world and are helpful to develop pollution control measures
40 from a long-term planning perspective.

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42 **Keywords:** PM_{2.5}; Carbonaceous aerosols; Secondary organic carbon (SOC); Long-term observation; Concentration-
43 weighted trajectory (CWT)

44 1 Introduction

45 Carbonaceous aerosols account for 20-90% of the submicron aerosol mass (Jimenez et al., 2009; Kroll et al., 2011).
46 It affects the physical and chemical properties of the atmosphere, including radiative forcing, hygroscopicity, and
47 toxicity (Hopke, 1991; Pope and Dockery, 2006; Bond et al., 2013). Carbonaceous components are classified
48 experimentally into three fractions: elemental carbon (EC), carbonate carbon (~~CC~~) and organic carbon (OC) (Turpin
49 et al., 2000). EC is a primary pollutant that can be directly emitted from fossil fuel combustion and biomass burning
50 (Cao et al., 2003; Galindo et al., 2019). Carbonate carbon is mainly in natural mineral dust and building/demolition
51 dust and exists in the coarse fraction (Chow and Watson, 2002; Chang et al., 2017), while, OC is composed of hundreds
52 of organic compounds, forming a complex mixture with different chemical and physical properties and accounting for
53 a major fraction of the carbonaceous aerosol (Chatterjee et al., 2021). OC can either be emitted directly from e.g.,
54 combustion processes, vehicular exhaust and cooking and is termed as primary OC (POC). It can also be formed in
55 the atmosphere by gas-to-particle oxidation reactions, ~~terms-termed~~ as secondary organic carbon (SOC) (Salvador et
56 al., 2021; Hallquist et al., 2009). Carbonaceous aerosols are among the major constituents of atmospheric aerosols and
57 their quantification is necessary for understanding the role of aerosols ~~in-issues-varying~~ from the regional visibility
58 degradation to health effects and global climate change (Wu et al., 2012; Mauderly and Chow, 2008).

59 Since carbonaceous aerosols are indispensable for probing atmospheric aging processes of organic aerosols and
60 formulating effective emission control policies, there have been a number of studies in China. In the 1980s, Dod et al.
61 (1986) first published a study on carbonaceous aerosol in Beijing in three seasons (i.e., spring, summer and winter),
62 demonstrating that ambient carbonaceous aerosols (soot in the study) were derived principally from coal combustion,
63 especially in winter. With the development of analytical techniques and in-depth research on carbonaceous aerosols,
64 a number of studies on the carbonaceous aerosols have emerged in many Chinese cities, such as Beijing (He et al.,
65 2001; Zhang et al., 2007; Ji et al., 2019), Xi'an (Cao et al., 2005; Han et al., 2009; Shen et al., 2014), Chengdu (Wang
66 et al., 2013; Tao et al., 2013), Shanghai (Cao et al., 2013; Zhu et al., 2015; Li et al., 2019), Guangzhou (Cao et al.,
67 2004; Ho et al., 2014; Wang et al., 2016), and Hong Kong (Lee et al., 2006; Ho et al., 2002; Ho et al., 2019). ~~Cao et
68 al. (2007) conducted the first nationwide simultaneous measurements of carbonaceous aerosols in 14 cities in China
69 in the winter and summer of 2003, revealing the seasonal and regional sources of carbonaceous aerosols across China.~~
70 However, long-term (e.g., 5 years) analysis of carbonaceous aerosols in the megacities is currently lacking, limiting
71 our understanding of the trend evolution of carbonaceous aerosols and the ability to evaluate the effectiveness of air
72 quality policies such as “Action Plan on Prevention and Control of Air Pollution” (Zhang et al., 2019).

73 Shanghai is one of the megacities with the most rapid economic and social development in the Yangtze River Delta
74 (YRD), China (Lin et al., 2014). Along with rapid economic growth and urbanization, the consequent degradation of
75 air quality has been recognized (Fu et al., 2008; Wang et al., 2015). Hence, the Chinese government unveiled its 5-
76 year “Action Plan on Prevention and Control of Air Pollution” in 2013, a comprehensive guideline that calls for
77 nationwide improvements in air quality by 2017, aiming to cut PM_{2.5} levels by 20% in the regions of YRD (Mep,

78 2013). Over the past decade, extensive studies have been launched to investigate the impact of carbonaceous aerosol
79 on air pollution in Shanghai. The PM_{2.5} reduction targets in Shanghai have been met to date (Zhang et al., 2019).
80 However, it is not well understood how the PM_{2.5} components, particularly carbonaceous aerosol, were evolving over
81 recent years, with different components likely demonstrating distinct temporal evolution. For example, while Shanghai
82 has witnessed a decrease in EC concentration, from an annual average value of 2.81 $\mu\text{g m}^{-3}$ in 2010 to 2.11 $\mu\text{g m}^{-3}$ in
83 2014, it also saw a small increase in OC concentration, from an annual average value of 7.09 $\mu\text{g m}^{-3}$ in 2010 to 7.83
84 $\mu\text{g m}^{-3}$ in 2014 (Chang et al., 2017). To grasp more complete information on the variation, evolution and sources of
85 the carbonaceous aerosol, especially in the post- “Action Plan on Prevention and Control of Air Pollution” era,
86 continuous and highly time-resolved measurements of carbonaceous aerosol over multiple years are necessary but are
87 currently lacking.

88 In this study, we conducted a long-term field campaign at a regional site in the YRD region from 2016 to 2020.
89 Hourly time-resolved OC and EC in PM_{2.5} were measured in a supersite in Shanghai. The secondary organic carbon
90 (SOC) was estimated by the minimum R² method (MRS) (Wu and Yu, 2016). The characteristics of carbonous aerosol
91 pollution and their seasonal and diurnal variations are discussed. Furthermore, we explored the meteorological factor
92 effects on carbonaceous aerosol concentrations in different levels of PM_{2.5}. To attain a better understanding of the
93 temporal variations of SOC in different PM_{2.5} levels and source areas, we identified the main source areas of SOC by
94 employing backward trajectory clusters and the Concentration weighted trajectory (CWT) model based on the Hybrid
95 Single Particle Lagrangian Integrated Trajectory (HYSPPLIT) analysis. The purpose of this study is to improve the
96 understanding of the variation and sources of SOC in ~~the~~ PM_{2.5} fraction. The long-term data presented in this study
97 provides critical information that can evaluate the effectiveness of the current air pollution control policies and are
98 informing to develop future pollution control measures.

99 2 Experiment and method

100 2.1 Observation site

101 The sampling site is in Qingpu District of western Shanghai, named Dianshan Lake (DSL) supersite
102 (31.09°N, 120.98°E, ~ 15 m above ground) (Fig. S1). It is ~7 km east of Dianshan Lake, ~ 50 km from downtown
103 Shanghai, situated at the intersection area of Jiangsu, Shanghai and Zhejiang. It is in a suburban area in the Yangtze
104 River Delta, surrounded by farmland and vegetated lands. There are two highways (G318 and G50, ~1 km to the site)
105 but no large industries near the sampling site. DSL station is a supersite maintained by the Shanghai Environmental
106 Monitoring Center and the monitoring data is incorporated into the national regional air automatic monitoring network
107 of China. The site is considered suitable to investigate the regional air quality and transport of air pollutants in YRD
108 region (Jia et al., 2020). In this study, a five-year intensive campaign was conducted at DSL site from January 2016
109 to December 2020, of which the observations were suspended from July to September 2019 (5% of the data) due to
110 site maintenance.

111 2.2 Instruments and measurements

112 PM_{2.5} mass concentrations were determined automatically by a tapered-element oscillating microbalance monitor
113 (TEOM, Thermo FH62C-14, USA). The sampling flow rate of the TEOM was 16.7 L min⁻¹. The uncertainty of the
114 hourly measurement is $\pm 1.50 \text{ mg m}^{-3}$, and the detection limit is 0.1 $\mu\text{g m}^{-3}$. In the study, the PM_{2.5} concentration was

115 converted to hourly means. O₃ and NO₂ were measured by an online analyzer (Model O342M, Environmental S.A,
116 FRA; model 42i, Thermo Environmental Instruments, USA).

117 Organic carbon (OC) and elemental carbon (EC) were measured online by a Sunset Semi-Continuous Carbon
118 Analyzer (Sunset Laboratory, Forest Grove, Oregon, USA) using the thermal-optical transmittance method at a flow
119 rate of 8 L min⁻¹. This instrument can provide hourly time-resolved OC and EC analyses. The detection limit of OC
120 and EC are 0.2 and 0.04 μg m⁻³, respectively.

121 The meteorological parameters including ambient temperature (T), relative humidity (RH), wind speed (WS),
122 and wind direction (WD) were obtained at the sampling site using the Visala (WXT520, Vaisala Ltd., Finland)
123 automatic weather station at hourly time resolution. The uncertainty of ambient temperature, RH, WS and WD are ±
124 0.1 °C, ± 3%, ± 0.3 m s⁻¹ and 3°, respectively. The data is collected every minute and converted to hourly means.

125 **2.3 Estimation of secondary organic carbon (SOC) by minimum R² (MRS) method**

126 Since EC is a tracer for primary POC from combustion sources, EC-tracer method has been widely used for
127 separating POC and SOC (Cao et al., 2007). In this study, an innovative EC-tracer method was used to estimate SOC
128 named the minimum correlation coefficient (MRS) method (Wu and Yu, 2016). The concentrations of SOC were
129 estimated as follows

$$130 \text{POC} = \text{EC} \times (\text{OC}/\text{EC})_{\text{primary}} \quad (1)$$

$$131 \text{SOC} = \text{OC} - \text{POC} \quad (2)$$

132 where OC and EC are the concentrations measured in the sample, (OC/EC)_{primary} is an estimate of the primary OC/EC
133 ratio through calculating a hypothetical set of (OC/EC)_{primary}. The hypothetical (OC/EC)_{primary} that generates the
134 minimum correlation coefficient (R²) SOC values was determined by seeking the minimum R² between SOC and EC.
135 This method may result in negative SOC concentrations for those periods when the estimated (OC/EC)_{primary} value
136 was higher than the measured OC/EC ratio. Although these data increase the uncertainty of the method, we assumed
137 these points were free of SOC formation. Since the relative contributions of different primary emission sources would
138 vary from month to month (Table S1), we calculated (OC/EC)_{primary} according to this method for each month from
139 2016 to 2020 in Shanghai (Fig. S2-S6).

140 **2.4 Back trajectory and concentration-weighted trajectory (CWT) model**

141 To determine the influences of regional transport on SOC at Shanghai, we calculated 72 h air mass back trajectory
142 of the central location at 500 m above the ground level. The trajectories were calculated with the NOAA Hybrid Single
143 Particle Lagrangian Integrated Trajectory (HYSPLIT4.0) model (Draxler and Rolph, 2003). The meteorological data
144 were from the Global Data Assimilation System (GDAS). The model was run eight times per day at starting times of
145 00:00, 03:00, 6:00, 9:00, 12:00, 15:00, 18:00, and 21:00 local time (LT), respectively. The relative parameter settings
146 in the model had also been used in the literature (Wang et al., 2018; Lin et al., 2019).

147 The concentration-weighted trajectory (CWT) approach was used to investigate the potential transport of pollution
148 (Fleming et al., 2012) on the interface of ZeFir (Petit et al., 2017). For the CWT calculations, the entire geographic
149 region covered by the 3-day backward trajectories was separated into 7920 grid cells of 0.5° latitude × 0.5° longitude.
150 Each grid cell was assigned a residence-time-weighted concentration obtained by the hourly averaged SOC
151 concentration associated with the trajectories that crossed that grid cell (Hsu et al., 2003). The CWT is defined as:

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$$C_{ij} = \frac{\sum_{l=1}^M C_l \tau_{ijl}}{\sum_{l=1}^M \tau_{ijl}} \quad (3)$$

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where C_{ij} is the average weighted concentration in the grid cell (ij th); C_l is the measured SOC concentration on the arrival of trajectory l ; τ_{ijl} is the number of trajectory endpoints in the ij th grid cell by trajectory l ; and M is the total number of trajectories.

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3 Results and discussion

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3.1 Temporal variations of carbonaceous aerosol

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3.1.1 Interannual variations

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Summary statistics for carbonaceous aerosol concentrations (EC, OC, POC, and SOC), as well as total $PM_{2.5}$, from 1 January 2016 to 31 December 2020 are presented in Table 1. During the entire observation period, the EC concentration ranged from 0.01-11.6 $\mu\text{g m}^{-3}$, and the five-year average concentration was $1.28 \pm 0.95 \mu\text{g m}^{-3}$. Annually, the EC concentration measured at Dianshan Lake has essentially decreased year by year in parallel over the five years. The average concentration of EC was highest in 2016, with an annual average of $1.50 \pm 1.17 \mu\text{g m}^{-3}$, while the average concentration of EC in 2020 was the lowest ($1.00 \pm 0.64 \mu\text{g m}^{-3}$). Therefore, compared to EC in 2016, the annual EC concentration reduced by $\sim 50\%$ in 2020.

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Different from EC, the average concentration of OC was the highest in 2017 (average $6.32 \pm 3.52 \mu\text{g m}^{-3}$). Since 2018, the average concentration of OC has decreased year by year with the lowest annual level of $4.99 \pm 2.93 \mu\text{g m}^{-3}$ found in 2020. It is worth noting that although the average concentration of OC in 2016 was lower than that in 2017, the maximum concentration of OC in 2016 was 1.41-1.61 times that of other years.

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OC subtypes of POC and SOC were apportioned using the novel MRS statistical model (See the method section). Different from the trend observed for EC, the apportioned POC increased year to year from 2016 to 2019, reaching the maximum value in 2019 ($3.76 \pm 2.55 \mu\text{g m}^{-3}$). However, it dropped sharply in 2020 [likely due to lockdown caused by the Covid-19 \(Jia et al., 2020\)](#). Therefore, the POC/EC ratio was changing for different years, and using a fixed POC/EC value over multiple years might bias the POC as well as SOC. In contrast, the changing trend of SOC was consistent with OC, which was the maximum in 2017 (average SOC concentration $2.98 \pm 2.25 \mu\text{g m}^{-3}$). In the next three years, the annual average concentration of SOC decreased on a year-to-year basis, reaching the lowest in 2020 ($1.53 \pm 1.35 \mu\text{g m}^{-3}$).

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3.1.2 Seasonal and monthly variations

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The seasonal variations of carbonaceous aerosol concentrations are illustrated in Fig. 1. The season-wise average concentrations of EC ranged from 0.92 (summer of 2019) to 1.90 $\mu\text{g m}^{-3}$ (winter of 2016), while OC ranged from 4.35 (summer of 2012) to 7.83 $\mu\text{g m}^{-3}$ (winter of 2016). For EC, OC, as well as OC subtypes (POC and SOC), similar seasonal variations are observed with generally higher average carbonaceous aerosol concentrations in autumn and winter and lower levels in spring and summer, except for a slightly higher concentration of EC in the summer of 2017 due to the boost in intensive pollution episodes (indicated by a significantly higher value in 95th percentile than that in other years). Higher concentrations of EC were observed in winter for the other four years, which could be caused by the stagnation of the atmosphere and the stronger influence of regional transport during wintertime (Chen et al.,

187 2017).

188 There is a consistent pattern for the seasonal variations of POC concentrations, the concentration levels of POC in
189 spring, summer, and fall were generally lower than that in the winter, reflecting generally locally-dominated POC
190 emissions in Dianshan Lake. In particular, POC concentrations in winter were 5.40, 3.88, 4.10, 4.13, and 3.97 $\mu\text{g m}^{-3}$,
191 and 2.04, 1.31, 1.40, 1.38, and 1.41 times higher than those in the summer for 2016, 2017, 2018, 2019, and 2020,
192 respectively.

193 SOC concentrations were estimated to be ranging from 0.13 ~ 10.70 $\mu\text{g m}^{-3}$ (spring), 1.04 ~ 19.41 $\mu\text{g m}^{-3}$ (summer),
194 0.02 ~ 25.37 $\mu\text{g m}^{-3}$ (autumn), and 0.03 ~ 37.14 $\mu\text{g m}^{-3}$ (winter) over the five years. Comparatively, there was no clear
195 trend in the seasonal changes of SOC over the five years, which can be explained by their complexity in terms of the
196 sources and formation processes. Indeed, in contrast to POC, SOC is the mixed product of the aging of the primary
197 emissions and secondary formation from precursor gases, which could vary significantly in different seasons. For
198 instance, strong solar radiation tends to facilitate photochemical reactions and thus enhance the formation of volatile
199 organic compounds (VOCs) to organic aerosols in summer (Tuet et al., 2017), while the increased anthropogenic
200 emissions (e.g., biomass burning and coal burning emissions) will also lead to a significant increase in SOC during
201 the harvest period and heating season (Zhang et al., 2013; Wang et al., 2020).

202 Monthly, the average mass concentrations of carbonaceous aerosols show relatively large variations in this study
203 (Table S1), with the average value ranging from 0.56 (October 2017) to 2.22 (December 2017) $\mu\text{g m}^{-3}$ for EC, while
204 it ranged from 3.39 (October 2017) to 9.00 $\mu\text{g m}^{-3}$ (December 2017) for OC. The month of December presented the
205 highest EC and OC average concentration (EC: $1.81 \pm 1.36 \mu\text{g m}^{-3}$; OC: $7.27 \pm 5.03 \mu\text{g m}^{-3}$) throughout the study
206 period. The lowest month for carbonaceous aerosols concentration was in August (EC: $0.94 \pm 0.52 \mu\text{g m}^{-3}$; OC: 4.47
207 $\pm 2.56 \mu\text{g m}^{-3}$). These are consistent with the previous study in Shanghai from 2010 to 2014 (Chang et al., 2017).
208 Table S1 also shows the monthly mean POC and SOC concentrations at our study site for the whole 5-year period.
209 POC shows similar variations to OC, with higher average concentrations in the cold season (from November to
210 February next year) and lower ones in the warm season (from April to October). The highest average POC
211 concentration was $4.97 \pm 3.97 \mu\text{g m}^{-3}$ (December), and the lowest POC average concentration was $2.23 \pm 1.34 \mu\text{g m}^{-3}$
212 (August). In contrast, the SOC average concentration was the highest in July ($3.43 \pm 3.12 \mu\text{g m}^{-3}$), which accounted
213 for 58.1% on average of OC in the same month.

214 3.1.3 Weekend–weekday pattern and diurnal variations

215 Fig. 2 shows diurnal patterns of carbonaceous aerosols during weekdays and weekends in four seasons, as well as
216 over the entire study period. Consistently, EC shows a distinctive diurnal pattern for different seasons or the whole
217 period (Fig. 2), which is characterized by two peaks occurring in the morning (around 08:00 local time) and during
218 the evening (around 20:00 local time), corresponding well with the morning and evening rush hours, coupled by
219 shallow mixing layer heights. It is worth noting that the peak of EC during the morning peak is higher than the evening
220 peak, and the difference between the two peaks is the largest in winter and the smallest in spring.

221 Different from EC, the daily variation of OC does not show a consistent pattern (Fig. 2). OC shows a peak at around
222 noon in spring and summer, while the peak time in autumn and winter is advanced to about 10:00 in the morning. The
223 peak appearing near noon can reflect the contribution of photochemical reaction to OC. In particular, the apportioned
224 SOC shows increased concentrations at a similar time. This phenomenon is especially obvious in spring and summer,
225 while no clear change in the concentration of SOC is found in winter. Additionally, in autumn and winter, OC also
226 shows a peak at 22:00, which is partly due to the primary emission as evident by the simultaneous increase in POC,

227 while such increase is absent in SOC (Fig. 2).

228 In terms of weekdays and weekends variation, the average concentration of EC during weekends in spring and
229 autumn is higher than that of weekdays (spring $EC_{\text{weekdays}}=1.25 \mu\text{g m}^{-3}$; $EC_{\text{weekends}}=1.30 \mu\text{g m}^{-3}$, autumn
230 $EC_{\text{weekdays}}=1.19 \mu\text{g m}^{-3}$; $EC_{\text{weekends}}=1.24 \mu\text{g m}^{-3}$), while the difference between weekday and weekends is small in
231 summer (both at $\sim 1.10 \mu\text{g m}^{-3}$). Only the winter EC_{weekdays} ($1.56 \mu\text{g m}^{-3}$) is higher than EC_{weekends} ($1.46 \mu\text{g m}^{-3}$). The
232 weekday and weekend variation observed at this site is different when compared to previous studies. Specifically,
233 according to a previous literature report (Chang et al., 2017), the observational data from 2010 to 2014 showed that
234 the concentration of EC on working days was greater than that on weekends because the traffic volume was
235 significantly higher on weekdays than on weekends, consistent with the location of the sampling site that is near
236 national highways (2 km away). However, Shanghai has officially implemented a traffic restriction system in 2016.
237 In this study, the sampling site is located near tourist attractions and is not in the traffic restricted area of Shanghai,
238 which is near the national expressway entering and leaving Shanghai (the straight-line distance is no more than two
239 kilometers). It is speculated that the heavy traffic flow due to the attraction of the nearby tourist sites during spring
240 and autumn weekends may lead to high EC emissions.

241 POC and SOC show different weekly patterns. Specifically, the concentration on working days in winter is higher
242 than that on weekends, while the SOC weekend in spring is slightly higher than that of weekdays, and the weekdays
243 of other seasons are higher than weekends (Fig. 2). This indicates that there is no significant decline in anthropogenic
244 activity on the weekends compared to weekdays. Enhanced anthropogenic emissions could be caused by no limit on
245 driving vehicles based on license plates on weekends. Human activities increase near the sampling site, leading to
246 increased VOC emissions and more SOC generation. Below we discuss more the sources of SOC and the impact of
247 meteorological parameters on its formation.

248 3.2 Insights into the formation pathways of SOC

249 3.2.1 Relationship between SOC and temperature

250 Examining the relationship between SOC vs. meteorological parameters (e.g., temperature, RH, and wind speed)
251 could provide more information on the formation and transformation of ambient SOC. Fig. S7 shows the statistics on
252 the concentration distribution of SOC in different temperature bins. Specifically, the mean value of SOC concentration
253 was $2.42 \mu\text{g m}^{-3}$ ($T < 0 \text{ }^\circ\text{C}$), $2.32 \mu\text{g m}^{-3}$ ($0 \text{ }^\circ\text{C} < T < 10 \text{ }^\circ\text{C}$), $2.06 \mu\text{g m}^{-3}$ ($10 \text{ }^\circ\text{C} < T < 20 \text{ }^\circ\text{C}$), $1.98 \mu\text{g m}^{-3}$ ($20 \text{ }^\circ\text{C} <$
254 $T < 30 \text{ }^\circ\text{C}$), and $3.82 \mu\text{g m}^{-3}$ ($T > 30 \text{ }^\circ\text{C}$) during the study period. Therefore, while the concentration of SOC does not
255 show a linear increase with the increase in temperature, at $T > 30 \text{ }^\circ\text{C}$, the SOC is significantly higher than in other
256 groups. We further use the T-test in different temperature groups and find that their difference is statistical significantly
257 (Fig. S8).

258 To investigate the temperature impacts on the formation of secondary organic aerosols, we divided the dataset
259 into four groups based on $PM_{2.5}$ concentrations for all seasons (Fig. 3). The clean periods were defined for $PM_{2.5}$
260 concentration $< 15 \mu\text{g m}^{-3}$, the transition periods were defined for $15 \mu\text{g m}^{-3} < PM_{2.5} < 35 \mu\text{g m}^{-3}$, the less polluted
261 days were defined for $35 \mu\text{g m}^{-3} < PM_{2.5} < 100 \mu\text{g m}^{-3}$, and the severe haze periods were defined for conditions with
262 $PM_{2.5} > 100 \mu\text{g m}^{-3}$. The definition of clean and haze periods is based on the national primary ambient air quality
263 standards for annual and daily mean $PM_{2.5}$ concentrations (i.e., 15 and $35 \mu\text{g m}^{-3}$, respectively). Below we show that
264 the promotion of SOC at high temperatures ($>30 \text{ }^\circ\text{C}$) is held true for pollution levels in all seasons.

265 Specifically, during the clean period in spring, the SOC concentration in Dianshan Lake in spring showed a trend

266 of first decreasing and then increasing with temperature. When $10\text{ }^{\circ}\text{C} < T < 20\text{ }^{\circ}\text{C}$, the average SOC concentration
267 was the lowest ($1.13\text{ }\mu\text{g m}^{-3}$). However, when $T > 30\text{ }^{\circ}\text{C}$, the highest SOC concentration ($3.41\text{ }\mu\text{g m}^{-3}$) was more than
268 doubled. Under the transition and mild pollution conditions, the change of SOC also showed a minimum value at
269 $10\text{ }^{\circ}\text{C} < T < 20\text{ }^{\circ}\text{C}$, but the concentration at this low point increased with the intensification of pollution. On heavy
270 pollution days, when the temperature is less than $30\text{ }^{\circ}\text{C}$, the temperature has no obvious promoting effect on the
271 generation of SOC. Similarly, during the clean period in summer, the effect of temperature increase on SOC was not
272 significant. However, under transition and pollution conditions (including light pollution and severe pollution), the
273 average concentration of SOC will increase significantly with the increase in temperature. Especially during periods
274 of severe pollution, the SOC concentration increased from $1.93\text{ }\mu\text{g m}^{-3}$ ($10\text{ }^{\circ}\text{C} < T < 20\text{ }^{\circ}\text{C}$) to $9.30\text{ }\mu\text{g m}^{-3}$ ($T > 30\text{ }^{\circ}\text{C}$).
275 In autumn, except for the clean days when the mean SOC concentration was the highest ($1.46\text{ }\mu\text{g m}^{-3}$) at $10\text{ }^{\circ}\text{C} < T <$
276 $20\text{ }^{\circ}\text{C}$, the average SOC concentration in the pollution period increases with the increase of temperature for other
277 periods. In comparison, winter SOC (The bottom panel in Fig. 3) is most significantly affected by temperature during
278 periods of severe pollution. During the severe pollution period, when $20\text{ }^{\circ}\text{C} < T < 30\text{ }^{\circ}\text{C}$, the average concentration of
279 SOC was higher than the average concentration of SOC under all conditions in other seasons, on average, reaching
280 $10.0\text{ }\mu\text{g m}^{-3}$.

281 In order to further verify the effect of temperature on the SOC concentration under various pollution conditions,
282 we conducted the Pearson correlation test between different temperature intervals and SOC concentration in each
283 period (Fig. S9). The results show that during the clean period, the Pearson correlation coefficient between temperature
284 and SOC concentration is only 0.31 ($T < 0\text{ }^{\circ}\text{C}$), indicating that the effect of temperature on the average concentration
285 of SOC is not significant during the clean period. The highest values of Pearson's correlation coefficients appeared at
286 $T > 30\text{ }^{\circ}\text{C}$ under transitional and lightly polluted conditions, but none of them exceeded 0.5. However, during the
287 period of heavy pollution at $T > 30\text{ }^{\circ}\text{C}$, the Pearson correlation coefficient between SOC and the temperature increased
288 to 0.62 (Fig. S9), demonstrating a more significant role of temperature in driving SOC formation during the heavy
289 pollution periods. High temperatures can boost the emission of SOC precursors, e.g., biogenic VOCs as well as
290 anthropogenic VOCs from e.g., solvent use (Zheng et al., 2018). Moreover, high temperatures are usually associated
291 with a strong solar radiation (Shrestha et al., 2019), which can promote the photochemical oxidation of both biogenic
292 and anthropogenic VOCs, increasing SOC concentrations.

293

294 3.2.2 Relationship between SOC and RH

295 Fig. 4 shows the diurnal variations of SOC concentrations and RH in four different $\text{PM}_{2.5}$ groups. In general, RH is at
296 its highest in the early morning and lowest between 13:00 and 15:00 noon. Most of the peaks of SOC during the
297 pollution period of each season appear at the lowest RH value at noon, while such a pattern is not observed during
298 clean periods. Specifically, during clean periods in spring, the daily average of RH is 70.4%, and the daily average
299 concentration of SOC is $1.26\text{ }\mu\text{g m}^{-3}$. The peak of SOC appeared at 1:00 am when the relative humidity reached 77.4%,
300 and then the relative humidity continues to increase gradually, reaching the highest value of the day (RH: 87.0%) at
301 7:00. However, the concentration of SOC does not change significantly, all around $1.30\text{ }\mu\text{g m}^{-3}$. This indicates that on
302 clean days, SOC is not significantly affected by photochemistry. In contrast, during more polluted periods in spring,
303 SOC shows an increased concentration ($> 2.10\text{ }\mu\text{g m}^{-3}$) at 15:00, which is due to photochemical oxidation which
304 overcomes the dilution effects caused by the increased planetary boundary layer in the afternoon. In summer, the
305 change of RH in different $\text{PM}_{2.5}$ ranges is not obvious with a mean RH of 78%, but the difference in SOC concentration

306 is significant. The daily average concentration of SOC in severe pollution is roughly 5 times that of clean days. In the
307 clean periods of summer, the nighttime peak of SOC is $1.61 \mu\text{g m}^{-3}$, which is larger than the daytime peak of $1.52 \mu\text{g}$
308 m^{-3} . With the intensification of the pollution degree, the difference between the peak daytime SOC and the peak
309 nighttime SOC gradually increased in summer with low RH associated with high SOC in the afternoon.

310 During heavy pollution in winter, RH does not change significantly between 0:00 and 6:00, and the nighttime
311 peak of SOC appears at 1:00 (SOC: $3.75 \mu\text{g m}^{-3}$). During the day, the RH gradually decreased to the lowest value of
312 50.4% at 14:00. At the same time, the concentration of SOC increases significantly and remains at a high concentration
313 level from 9:00 to 16:00, suggesting the photochemical formation of SOC is still very efficient and important even in
314 winter when solar radiation was supposedly less intense than in summer.

315 3.2.3 Photochemical formation of SOC

316 The concentration of oxidant O_x ($\text{O}_x = \text{O}_3 + \text{NO}_2$) is usually used as a proxy to indicate the atmospheric oxidizing
317 capacity associated with photochemical reactions (Wang et al., 2017). The daily O_x minimum occurred in the morning
318 followed by a sharp increase to a peak in the afternoon in all seasons (Fig. S10). Similarly, SOC also a large increase
319 in the afternoon in all seasons, with peak concentrations in the range of $2.40\text{-}3.00 \mu\text{g m}^{-3}$ (Fig. S10). The concurrent
320 increase in SOC and O_x in the afternoon suggests photochemical formation was a dominant formation pathway for
321 SOC even in winter. This is different from the formation pathways of SOC in north China, where aqueous phase
322 chemistry is often reported to be the major formation pathway of SOC in winter (Lin et al., 2020; An et al., 2019; Sun
323 et al., 2015; Chen et al., 2019).

324 The positive relationship between SOC and O_x was well presented in four different $\text{PM}_{2.5}$ bins in different seasons
325 (Fig. 5). In spring, SOC was positively correlated with O_x with the concentrations of SOC during the haze periods \sim
326 $1.8\text{-}3.2$ times higher than those during the clean periods. The SOC in summer and fall showed a similar trend with
327 higher levels of O_x significantly associated with the increased SOC concentrations. The average concentration of SOC
328 reached its highest during the severe haze period in summer and autumn with an average SOC of $>6.00 \mu\text{g m}^{-3}$ when
329 O_x was $>200 \mu\text{g m}^{-3}$. For the $\text{PM}_{2.5}$ bin of $>100 \mu\text{g m}^{-3}$ in winter, the concentration of SOC showed a significant
330 increase ($>6.00 \mu\text{g m}^{-3}$; mean value) from $<4.00 \mu\text{g m}^{-3}$ when O_x increased to $>200 \mu\text{g m}^{-3}$ from $<50 \mu\text{g m}^{-3}$. In
331 contrast, the increase in SOC for other $\text{PM}_{2.5}$ bins was less significant in winter, due to the generally low O_x for $\text{PM}_{2.5}$
332 of $<100 \mu\text{g m}^{-3}$ (Fig. 5).

333 3.2.4 Relationship between SOC and wind speed/direction

334 Wind speed is an important factor controlling the concentrations of carbonaceous aerosols. In this study, EC and
335 POC concentrations show evident WS dependence, with higher concentrations in association with lower wind speeds
336 (Fig. S11). This is consistent with the general pattern that pollution episodes are likely to occur under lower wind
337 speeds ($\text{WS} < 1 \text{ m s}^{-1}$) (Ren, 2018). At the same time, the relationship between the concentration of carbonaceous
338 aerosols and wind speed can also reflect that its main contribution comes from local emissions or regional transmission.
339 In particular, in spring, summer, and autumn, the concentration of carbonaceous aerosols decreased with the increase
340 in wind speed, indicating that in these seasons, local emissions at low wind speeds are the main contribution of
341 carbonaceous aerosols. It was worth noting that in winter, on the one hand, the concentration of carbonaceous aerosol
342 under each wind speed gradient is higher than that of other seasons. On the other hand, when the wind speed is higher
343 than 4.5 m s^{-1} , the concentration of carbonaceous aerosol is also increased. Specifically, the concentration of EC

344 increased by 12.4%, while POC increased by 11.7%, indicating the contribution of the transport in winter to
345 carbonaceous aerosols.

346 In contrast, SOC is affected differently by wind speed. The dependence of SOC concentrations on wind speed is
347 shown in Fig. 6a. In spring, the concentration of SOC is about $2 \mu\text{g m}^{-3}$, and the concentration gradient of SOC
348 increases slightly with the increase in wind speed. When the wind speed is greater than 1.5 m s^{-1} and less than 2 m s^{-1} ,
349 the concentration of SOC reaches the highest value of $2.15 \mu\text{g m}^{-3}$. When the wind speed is less than 0.5 m s^{-1} , the
350 SOC concentration is $1.37 \mu\text{g m}^{-3}$, and when the wind speed is greater than 5 m s^{-1} , the SOC concentration is $1.78 \mu\text{g}$
351 m^{-3} , with an increase of 29.8%. In summer, the concentration of SOC decreases with the wind speed gradient. When
352 the wind speed is 2~2.5 m/s, the SOC concentration is the highest ($2.79 \mu\text{g m}^{-3}$). When the wind speed is greater than
353 5 m/s, the SOC concentration is at its lowest ($1.40 \mu\text{g m}^{-3}$). In autumn, the SOC does not appear to change significantly
354 ($\sim 2 \mu\text{g m}^{-3}$) when the wind speed gradient gradually increased. In winter, when the wind speed is less than 4.5 m s^{-1} ,
355 the SOC concentration is about $2 \mu\text{g m}^{-3}$ (mean value is $2.03 \mu\text{g m}^{-3}$). When the wind speed is greater than 4.5 m s^{-1} ,
356 the SOC concentration increases to $2.45 \mu\text{g m}^{-3}$. It is worth noting that when the wind speed is greater than 5 m s^{-1} ,
357 the concentration of SOC increases by 29.1% ($2.62 \mu\text{g m}^{-3}$), reaching the highest average concentration of SOC under
358 different wind speed gradients in winter. This shows that the main contribution of Shanghai SOC in winter comes
359 from regional transmission.

360 The seasonal bivariate polar plots of SOC concentrations for 2016 – 2020 were shown in Fig. 6b. The high
361 concentration load of SOC near the sampling site in all seasons mainly occurs in the southwest direction and under
362 the condition of low wind speed (WS less than 4 m/s). The concentration distributions of SOC (Fig. 6b) and OC were
363 similar in spring (Fig. S12c), and the highest concentration area appeared in the southwest region. The distribution
364 and loading of SOC with a high concentration in summer (SOC $> 4 \mu\text{g m}^{-3}$) is closer to the sampling point (dense
365 distribution in WS around 2 m s^{-1}), further proving the previous conclusion that the main contribution of SOC in
366 summer from a local build. The relationship between SOC and wind direction remains unchanged in autumn. However,
367 the high SOC loading area is still located southwest of the sampling point, but the concentration is significantly lower
368 than that in summer and autumn which is closer to the sampling point. In addition, in the southeast direction, the area
369 with wind speed greater than 6 m s^{-1} has a high loading area of SOC, and it is speculated that this part of SOC may
370 be transmitted from the area near the east coast.

371 3.3 Analysis of potential source regions of SOC

372 The CWT results demonstrate the spatial distributions of SOC in the form of the SOC weighted 72-h backward
373 trajectories (Fig. 7). The CWT results are generally consistent with the corresponding polar plots as shown in Section
374 3.2.4. Specifically, the potential source areas with high CWT values for SOC were located in the surroundings of
375 Shanghai. In spring, SOC mainly comes from North China and the middle and lower reaches of the Yangtze River,
376 specifically from central Anhui, southern Jiangsu, central and northern Zhejiang, and northern Fujian; in summer, the
377 high SOC values in southern Shanghai (SOC $> 3.5 \mu\text{g m}^{-3}$) mainly come from central and southern Anhui, Zhejiang,
378 Fujian; there are also great contributions from offshore (the northern South China Sea and East China Sea). The SOC
379 in Shanghai in autumn mainly comes from the northern and central regions of Zhejiang. Southern Jiangsu is the main
380 source of SOC in Shanghai in winter, followed by northern Zhejiang; on the other hand, the northern long-distance
381 transmission from the North China Plain further extends to Inner Mongolia, Mongolia and the Russian border.

382 We further analyzed the potential sources of Shanghai SOC under different PM concentrations (Fig. 8). The
383 concentrations of SOC during the clean period were all lower than $2 \mu\text{g m}^{-3}$, and there were three main source

384 pathways, namely, the northern of the North China Plain, Inner Mongolia and eastern Mongolia; the Yellow Sea and
385 the Korean Peninsula; Zhejiang Province and northern Fujian Province. The source area of SOC during the transition
386 period was further expanded, and the concentration of SOC in the main area was between 2 and 3 $\mu\text{g m}^{-3}$, which was
387 basically consistent with the source area coverage during the cleaning period. It is worth noting that a high SOC
388 loading appeared in the coastal area of Fujian, presumably related to secondary aerosols transported by oceanic air
389 masses. During the transition periods ($35 < \text{PM} < 100 \mu\text{g m}^{-3}$), the source area of SOC expanded, and the area with
390 SOC concentration greater than 3.5 $\mu\text{g m}^{-3}$ became clearer, mainly concentrated in Zhejiang, Fujian and offshore areas.
391 The main transmission areas are eastern Inner Mongolia, Hebei Province, and the North China Plain, all the way
392 southward to Anhui, connecting with the source areas of Zhejiang Province and Fujian Province. During high pollution
393 periods, the areas with SOC concentrations higher than 4 $\mu\text{g m}^{-3}$ were mainly concentrated in two areas, one was from
394 southeastern Mongolia in the north, through eastern Inner Mongolia and Hebei Province, through Shandong and
395 Jiangsu and finally to Shanghai; the other source area was in the south of Shanghai Zhejiang Province, Fujian Province,
396 and eastern Jiangxi Province. Overall, the main potential source areas of Shanghai SOC are the Yangtze River Delta,
397 North China Plain, northern China, Inner Mongolia, and eastern China provinces and offshore areas, such as Zhejiang
398 Province, Fujian Province, and the South China Sea.

399 **4 Conclusions**

400 In this study, the hourly mass concentration of OC and EC in $\text{PM}_{2.5}$ were continuously measured from 1 January
401 2016 to 31 December 2020 at a supersite in Shanghai. OC subtypes of POC and SOC were estimated by the novel
402 MRS method. Based on the five-year measurements, the interannual, monthly, seasonal, and diurnal variations in OC
403 and EC, as well as OC subtypes are presented. By examining the relationship between SOC and meteorological
404 parameters (e.g., temperature, RH, and wind speed), as well as O_x , the sources, formation and transformation
405 mechanisms of ambient SOC are revealed. We show that SOC formation is greatly enhanced at high temperatures
406 ($>30^\circ\text{C}$), while it is inversely correlated with RH. In particular, we show that the photochemical formation of SOC is
407 still very efficient and is the major formation pathway even in winter when solar radiation was supposedly less intense
408 than in summer. High EC and POC concentrations are found to be associated with low wind speeds, which is consistent
409 with their primary nature from local emission. Moreover, increased SOC concentrations are also found to be associated
410 with high wind speed ($>5 \text{ m s}^{-1}$) in winter, which is increased by 29.1% (2.62 $\mu\text{g m}^{-3}$) when compared to that during
411 lower winds, suggesting regional sources of SOC in winter. By analyzing the potential source regions using the CWT
412 algorithm, the geographic regions of SOC are found to be mainly associated with transport from outside Shanghai
413 ($\text{SOC} > 3.5 \mu\text{g m}^{-3}$) including central and southern Anhui, Zhejiang, and Fujian. Our results elucidate the trends in
414 SOC and POC over recent years in Shanghai, which are important for evaluating the effectiveness of the air pollution
415 control measures and holding important implications for policy making. Given that SOC was associated with high
416 temperature and regional transport, global warming is likely increasing the importance of SOC. Since SOC is regional,
417 combined efforts in reducing regional sources of SOC precursors are needed to further reduce the air pollution events
418 in Shanghai.

419 **Data availability**

420 The data presented in this study are available at the Zenodo data archive <https://doi.org/10.5281/zenodo.6473085>

421 (Wang et al., 2022).

422 **Supplement.**

423 The supplement related to this article is available online at:

424

425 **Declaration of competing interest**

426 The authors declare that they have no known competing financial interests or personal relationships that could have
427 appeared to influence the work reported in this paper.

428 **Credit authorship contribution statement**

429 Meng Wang: Conceptualization, Methodology, Validation, Formal analysis, Writing- original draft.

430 Yusen Duan: Methodology, Formal analysis.

431 Wei Xu: Validation, Formal analysis.

432 Qiyuan Wang: Conceptualization, Writing, Review and Editing.

433 Zhuozhi Zhang: Formal analysis, Writing, Review and Editing.

434 Qi Yuan: Formal analysis, Methodology.

435 Xinwei Li: Investigation, Methodology.

436 Shuwen Han: Validation, Investigation.

437 Haijie Tong: Writing, Review and Editing.

438 Juntao Huo: Investigation, Methodology, and Validation.

439 Jia Chen: Investigation, Methodology, and Validation.

440 Shan Gao: Methodology, Validation, Formal analysis.

441 Zhongbiao Wu: Review and editing, Funding acquisition.

442 Long Cui: Formal analysis, Investigation.

443 Yu Huang: Writing, Review and Editing.

444 Junji Cao: Writing, Review and Editing.

445 Qingyan Fu: Writing, Review editing, Funding acquisition, and Resources.

446 Shun-cheng Lee: Writing, Review editing, Funding acquisition, and Supervision.

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457 **References**

458 An, Z., Huang, R.-J., Zhang, R., Tie, X., Li, G., Cao, J., Zhou, W., Shi, Z., Han, Y., Gu, Z., and Ji, Y.: Severe haze in
459 northern China: A synergy of anthropogenic emissions and atmospheric processes, *Proceedings of the National
460 Academy of Sciences*, 116, 8657, 10.1073/pnas.1900125116, 2019.

461 Bond, T. C., Doherty, S. J., Fahey, D. W., Forster, P. M., Berntsen, T., DeAngelo, B. J., Flanner, M. G., Ghan, S.,
462 Kärcher, B., Koch, D., Kinne, S., Kondo, Y., Quinn, P. K., Sarofim, M. C., Schultz, M. G., Schulz, M., Venkataraman,
463 C., Zhang, H., Zhang, S., Bellouin, N., Guttikunda, S. K., Hopke, P. K., Jacobson, M. Z., Kaiser, J. W., Klimont, Z.,
464 Lohmann, U., Schwarz, J. P., Shindell, D., Storelvmo, T., Warren, S. G., and Zender, C. S.: Bounding the role of black
465 carbon in the climate system: A scientific assessment, *Journal of Geophysical Research: Atmospheres*, 118, 5380-5552,
466 <https://doi.org/10.1002/jgrd.50171>, 2013.

467 Cao, J. J., Lee, S. C., Ho, K. F., Zhang, X. Y., Zou, S. C., Fung, K., Chow, J. C., and Watson, J. G.: Characteristics of
468 carbonaceous aerosol in Pearl River Delta Region, China during 2001 winter period, *Atmos. Environ.*, 37, 1451-1460,
469 10.1016/S1352-2310(02)01002-6, 2003.

470 Cao, J. J., Lee, S. C., Ho, K. F., Zou, S. C., Fung, K., Li, Y., Watson, J. G., and Chow, J. C.: Spatial and seasonal
471 variations of atmospheric organic carbon and elemental carbon in Pearl River Delta Region, China, *ATMOSPHERIC
472 ENVIRONMENT*, 38, 4447-4456, 10.1016/j.atmosenv.2004.05.016, 2004.

473 Cao, J. J., Zhu, C. S., Tie, X. X., Geng, F. H., Xu, H. M., Ho, S. S. H., Wang, G. H., Han, Y. M., and Ho, K. F.:
474 Characteristics and sources of carbonaceous aerosols from Shanghai, China, *ATMOSPHERIC CHEMISTRY AND
475 PHYSICS*, 13, 803-817, 10.5194/acp-13-803-2013, 2013.

476 Cao, J. J., Wu, F., Chow, J. C., Lee, S. C., Li, Y., Chen, S. W., An, Z. S., Fung, K. K., Watson, J. G., Zhu, C. S., and
477 Liu, S. X.: Characterization and source apportionment of atmospheric organic and elemental carbon during fall and
478 winter of 2003 in Xi'an, China, *ATMOSPHERIC CHEMISTRY AND PHYSICS*, 5, 3127-3137, 10.5194/acp-5-3127-
479 2005, 2005.

480 Cao, J. J., Lee, S. C., Chow, J. C., Watson, J. G., Ho, K. F., Zhang, R. J., Jin, Z. D., Shen, Z. X., Chen, G. C., Kang, Y.
481 M., Zou, S. C., Zhang, L. Z., Qi, S. H., Dai, M. H., Cheng, Y., and Hu, K.: Spatial and seasonal distributions of
482 carbonaceous aerosols over China, *Journal of Geophysical Research: Atmospheres*, 112,
483 <https://doi.org/10.1029/2006JD008205>, 2007.

484 Chang, Y., Deng, C., Cao, F., Cao, C., Zou, Z., Liu, S., Lee, X., Li, J., Zhang, G., and Zhang, Y.: Assessment of
485 carbonaceous aerosols in Shanghai, China – Part 1: long-term evolution, seasonal variations, and meteorological
486 effects, *Atmos. Chem. Phys.*, 17, 9945-9964, 10.5194/acp-17-9945-2017, 2017.

487 Chatterjee, A., Mukherjee, S., Dutta, M., Ghosh, A., Ghosh, S. K., and Roy, A.: High rise in carbonaceous aerosols
488 under very low anthropogenic emissions over eastern Himalaya, India: Impact of lockdown for COVID-19 outbreak,
489 *Atmospheric Environment*, 244, 117947, <https://doi.org/10.1016/j.atmosenv.2020.117947>, 2021.

490 Chen, D., Cui, H., Zhao, Y., Yin, L., Lu, Y., and Wang, Q. g.: A two-year study of carbonaceous aerosols in ambient
491 PM_{2.5} at a regional background site for western Yangtze River Delta, China, *Atmospheric Research*, 183, 351-361,

492 2017.

493 Chen, L., Zhu, J., Liao, H., Gao, Y., Qiu, Y., Zhang, M., Liu, Z., Li, N., and Wang, Y.: Assessing the formation and
494 evolution mechanisms of severe haze pollution in the Beijing–Tianjin–Hebei region using process analysis, *Atmos.*
495 *Chem. Phys.*, 19, 10845-10864, 10.5194/acp-19-10845-2019, 2019.

496 Chow, J. C. and Watson, J. G.: Review of PM_{2.5} and PM₁₀ Apportionment for Fossil Fuel Combustion and Other
497 Sources by the Chemical Mass Balance Receptor Model, *Energy & Fuels*, 16, 222-260, 10.1021/ef0101715, 2002.

498 Dod, R. L., Giaque, R. D., Novakov, T., Weihan, S., Quipeng, Z., and Wenzhi, S.: Sulfate and carbonaceous aerosols
499 in Beijing China, *Atmospheric Environment (1967)*, 20, 2271-2275, [https://doi.org/10.1016/0004-6981\(86\)90317-3](https://doi.org/10.1016/0004-6981(86)90317-3),
500 1986.

501 Draxler, R. R. and Rolph, G. D.: HYSPLIT (HYbrid Single-Particle Lagrangian Integrated Trajectory) model access
502 via NOAA ARL READY website (<http://www.arl.noaa.gov/ready/hysplit4.html>). NOAA Air Resources Laboratory,
503 Silver Spring, 2003.

504 Fleming, Z. L., Monks, P. S., and Manning, A. J.: Review: Untangling the influence of air-mass history in interpreting
505 observed atmospheric composition, *Atmospheric Research*, 104-105, 1-39,
506 <https://doi.org/10.1016/j.atmosres.2011.09.009>, 2012.

507 Fu, Q., Zhuang, G., Wang, J., Xu, C., Huang, K., Li, J., Hou, B., Lu, T., and Streets, D. G.: Mechanism of formation
508 of the heaviest pollution episode ever recorded in the Yangtze River Delta, China, *Atmospheric Environment*, 42,
509 2023-2036, <https://doi.org/10.1016/j.atmosenv.2007.12.002>, 2008.

510 Galindo, N., Yubero, E., Clemente, A., Nicolás, J., Navarro-Selma, B., and Crespo, J.: Insights into the origin and
511 evolution of carbonaceous aerosols in a mediterranean urban environment, *Chemosphere*, 235, 636-642, 2019.

512 Hallquist, M., Wenger, J. C., Baltensperger, U., Rudich, Y., Simpson, D., Claeys, M., Dommen, J., Donahue, N.,
513 George, C., and Goldstein, A.: The formation, properties and impact of secondary organic aerosol: current and
514 emerging issues, *Atmospheric chemistry and physics*, 9, 5155-5236, 2009.

515 Han, Y. M., Cao, J. J., Chow, J. C., Watson, J. G., An, Z. S., and Liu, S. X.: Elemental carbon in urban soils and road
516 dusts in Xi'an, China and its implication for air pollution, *ATMOSPHERIC ENVIRONMENT*, 43, 2464-2470,
517 10.1016/j.atmosenv.2009.01.040, 2009.

518 He, K., Yang, F., Ma, Y., Zhang, Q., Yao, X., Chan, C. K., Cadle, S., Chan, T., and Mulawa, P.: The characteristics of
519 PM_{2.5} in Beijing, China, *Atmospheric Environment*, 35, 4959-4970, 2001.

520 Ho, K. F., Lee, S. C., Yu, J. C., Zou, S. C., and Fung, K.: Carbonaceous characteristics of atmospheric particulate
521 matter in Hong Kong, *SCIENCE OF THE TOTAL ENVIRONMENT*, 300, 59-67, 10.1016/S0048-9697(02)00281-4,
522 2002.

523 Ho, K. F., Engling, G., Ho, S. S. H., Huang, R. J., Lai, S. C., Cao, J. J., and Lee, S. C.: Seasonal variations of
524 anhydrosugars in PM_{2.5} in the Pearl River Delta Region, China, *TELLUS SERIES B-CHEMICAL AND PHYSICAL*
525 *METEOROLOGY*, 66, 10.3402/tellusb.v66.22577, 2014.

526 Ho, S. S. H., Li, L., Qu, L., Cao, J., Lui, K. H., Niu, X., Lee, S.-C., and Ho, K. F.: Seasonal behavior of water-soluble
527 organic nitrogen in fine particulate matter (PM_{2.5}) at urban coastal environments in Hong Kong, *Air Quality*,
528 *Atmosphere & Health*, 12, 389-399, 10.1007/s11869-018-0654-5, 2019.

529 Hopke, P.: Receptor Modeling for Air Quality Management, *Issues in Environmental Science & Technology*, 8, 289–
530 300, 1991.

531 Ji, D., Gao, M., Maenhaut, W., He, J., Wu, C., Cheng, L., Gao, W., Sun, Y., Sun, J., Xin, J., Wang, L., and Wang, Y.:
532 The carbonaceous aerosol levels still remain a challenge in the Beijing-Tianjin-Hebei region of China: Insights from

533 continuous high temporal resolution measurements in multiple cities, *Environment International*, 126, 171-183,
534 <https://doi.org/10.1016/j.envint.2019.02.034>, 2019.

535 Jia, H., Huo, J., Fu, Q., Duan, Y., Lin, Y., Jin, X., Hu, X., and Cheng, J.: Insights into chemical composition, abatement
536 mechanisms and regional transport of atmospheric pollutants in the Yangtze River Delta region, China during the
537 COVID-19 outbreak control period, *Environmental Pollution*, 267, 115612,
538 <https://doi.org/10.1016/j.envpol.2020.115612>, 2020.

539 Jimenez, J. L., Canagaratna, M. R., Donahue, N. M., Prevot, A. S. H., Zhang, Q., Kroll, J. H., DeCarlo, P. F., Allan, J.
540 D., Coe, H., Ng, N. L., Aiken, A. C., Docherty, K. S., Ulbrich, I. M., Grieshop, A. P., Robinson, A. L., Duplissy, J.,
541 Smith, J. D., Wilson, K. R., Lanz, V. A., Hueglin, C., Sun, Y. L., Tian, J., Laaksonen, A., Raatikainen, T., Rautiainen,
542 J., Vaattovaara, P., Ehn, M., Kulmala, M., Tomlinson, J. M., Collins, D. R., Cubison, M. J., Dunlea, J., Huffman, J. A.,
543 Onasch, T. B., Alfarra, M. R., Williams, P. I., Bower, K., Kondo, Y., Schneider, J., Drewnick, F., Borrmann, S., Weimer,
544 S., Demerjian, K., Salcedo, D., Cottrell, L., Griffin, R., Takami, A., Miyoshi, T., Hatakeyama, S., Shimono, A., Sun,
545 J. Y., Zhang, Y. M., Dzepina, K., Kimmel, J. R., Sueper, D., Jayne, J. T., Herndon, S. C., Trimborn, A. M., Williams,
546 L. R., Wood, E. C., Middlebrook, A. M., Kolb, C. E., Baltensperger, U., and Worsnop, D. R.: Evolution of Organic
547 Aerosols in the Atmosphere, *Science*, 326, 1525-1529, <https://doi.org/10.1126/science.1180353>, 2009.

548 Kroll, J. H., Donahue, N. M., Jimenez, J. L., Kessler, S. H., Canagaratna, M. R., Wilson, K. R., Altieri, K. E., Mazzoleni,
549 L. R., Wozniak, A. S., and Bluhm, H.: Carbon oxidation state as a metric for describing the chemistry of atmospheric
550 organic aerosol, *Nature chemistry*, 3, 133-139, 2011.

551 Lee, S. C., Cheng, Y., Ho, K. F., Cao, J. J., Louie, P. K. K., Chow, J. C., and Watson, J. G.: PM1.0 and PM2.5
552 Characteristics in the Roadside Environment of Hong Kong, *Aerosol Science and Technology*, 40, 157-165,
553 10.1080/02786820500494544, 2006.

554 Li, H., Wang, D., Cui, L., Gao, Y., Huo, J., Wang, X., Zhang, Z., Tan, Y., Huang, Y., Cao, J., Chow, J. C., Lee, S.-c.,
555 and Fu, Q.: Characteristics of atmospheric PM2.5 composition during the implementation of stringent pollution
556 control measures in shanghai for the 2016 G20 summit, *Science of The Total Environment*, 648, 1121-1129,
557 <https://doi.org/10.1016/j.scitotenv.2018.08.219>, 2019.

558 Lin, C., Ceburnis, D., Huang, R.-J., Canonaco, F., Prévôt, A. S. H., Dowd, C., and Ovadnevaite, J.: Summertime
559 Aerosol over the West of Ireland Dominated by Secondary Aerosol during Long-Range Transport, *Atmosphere*, 10,
560 59, 2019.

561 Lin, C., Huang, R.-J., Xu, W., Duan, J., Zheng, Y., Chen, Q., Hu, W., Li, Y., Ni, H., Wu, Y., Zhang, R., Cao, J., and
562 O'Dowd, C.: Comprehensive Source Apportionment of Submicron Aerosol in Shijiazhuang, China: Secondary
563 Aerosol Formation and Holiday Effects, *ACS Earth and Space Chemistry*, 4, 947-957,
564 10.1021/acsearthspacechem.0c00109, 2020.

565 Lin, Y., Huang, K., Zhuang, G., Fu, J. S., Wang, Q., Liu, T., Deng, C., and Fu, Q.: A multi-year evolution of aerosol
566 chemistry impacting visibility and haze formation over an Eastern Asia megacity, Shanghai, *Atmospheric*
567 *Environment*, 92, 76-86, <https://doi.org/10.1016/j.atmosenv.2014.04.007>, 2014.

568 Mauderly, J. L. and Chow, J. C.: Health Effects of Organic Aerosols, *Inhalation Toxicology*, 20, 257-288,
569 10.1080/08958370701866008, 2008.

570 MEP, P. R. C. M. o. E. P.: The state council issues action plan on prevention and control of air pollution introducing
571 ten measures to improve air quality, available at:
572 http://english.mep.gov.cn/News_service/infocus/201309/t20130924_260707.htm, 2013.

573 Petit, J. E., Favez, O., Albinet, A., and Canonaco, F.: A user-friendly tool for comprehensive evaluation of the

574 geographical origins of atmospheric pollution: Wind and trajectory analyses, *Environmental Modelling & Software*,
575 88, 183-187, <https://doi.org/10.1016/j.envsoft.2016.11.022>, 2017.

576 Pope, C. A. and Dockery, D. W.: Health Effects of Fine Particulate Air Pollution: Lines that Connect, *Journal of the*
577 *Air & Waste Management Association*, 56, 709-742, 10.1080/10473289.2006.10464485, 2006.

578 Ren, H.: The organic molecular composition, diurnal variation, and stable carbon isotope ratios of PM_{2.5} in Beijing
579 during the 2014 APEC summit, *Environmental pollution*, v. 243, pp. 919-928-2018 v.2243,
580 10.1016/j.envpol.2018.08.094, 2018.

581 Salvador, C. M. G., Tang, R., Priestley, M., Li, L., Tsiligiannis, E., Le Breton, M., Zhu, W., Zeng, L., Wang, H., Yu,
582 Y., Hu, M., Guo, S., and Hallquist, M.: Ambient nitro-aromatic compounds – biomass burning versus secondary
583 formation in rural China, *Atmos. Chem. Phys.*, 21, 1389-1406, 10.5194/acp-21-1389-2021, 2021.

584 Shen, Z. X., Cao, J. J., Zhang, L. M., Liu, L., Zhang, Q., Li, J. J., Han, Y. M., Zhu, C. S., Zhao, Z. Z., and Liu, S. X.:
585 Day-night differences and seasonal variations of chemical species in PM₁₀ over Xi'an, northwest China,
586 *ENVIRONMENTAL SCIENCE AND POLLUTION RESEARCH*, 21, 3697-3705, 10.1007/s11356-013-2352-z,
587 2014.

588 Shrestha, A. K., Thapa, A., and Gautam, H.: Solar Radiation, Air Temperature, Relative Humidity, and Dew Point
589 Study: Damak, Jhapa, Nepal, *International Journal of Photoenergy*, 2019, 8369231, 10.1155/2019/8369231, 2019.

590 Sun, Y. L., Wang, Z. F., Du, W., Zhang, Q., Wang, Q. Q., Fu, P. Q., Pan, X. L., Li, J., Jayne, J., and Worsnop, D. R.:
591 Long-term real-time measurements of aerosol particle composition in Beijing, China: seasonal variations,
592 meteorological effects, and source analysis, *Atmos. Chem. Phys.*, 15, 10149-10165, 10.5194/acp-15-10149-2015,
593 2015.

594 Tao, J., Zhang, L. M., Engling, G., Zhang, R. J., Yang, Y. H., Cao, J. J., Zhu, C. S., Wang, Q. Y., and Luo, L.: Chemical
595 composition of PM_{2.5} in an urban environment in Chengdu, China: Importance of springtime dust storms and biomass
596 burning, *ATMOSPHERIC RESEARCH*, 122, 270-283, 10.1016/j.atmosres.2012.11.004, 2013.

597 Tuet, W. Y., Chen, Y., Xu, L., Fok, S., Gao, D., Weber, R. J., and Ng, N. L.: Chemical oxidative potential of secondary
598 organic aerosol (SOA) generated from the photooxidation of biogenic and anthropogenic volatile organic compounds,
599 *Atmos. Chem. Phys.*, 17, 839-853, 10.5194/acp-17-839-2017, 2017.

600 Turpin, B. J., Saxena, P., and Andrews, E.: Measuring and simulating particulate organics in the atmosphere: problems
601 and prospects, *Atmospheric Environment*, 34, 2983-3013, [https://doi.org/10.1016/S1352-2310\(99\)00501-4](https://doi.org/10.1016/S1352-2310(99)00501-4), 2000.

602 Wang, J. Z., Ho, S. S. H., Ma, S. X., Cao, J. J., Dai, W. T., Liu, S. X., Shen, Z. X., Huang, R. J., Wang, G. H., and Han,
603 Y. M.: Characterization of PM_{2.5} in Guangzhou, China: uses of organic markers for supporting source apportionment,
604 *SCIENCE OF THE TOTAL ENVIRONMENT*, 550, 961-971, 10.1016/j.scitotenv.2016.01.138, 2016.

605 Wang, M., Duan, Y., Xu, W., Wang, Q., Zhang, Z., Yuan, Q., Li, X., Han, S., Tong, H., Huo, J., Chen, J., Gao, S., Wu,
606 Z., Cui, L., Huang, Y., Xiu, G., Cao, J., Fu, Q., and Lee, S.-c.: Measurement report: characterization and sources of
607 the ambient secondary organic carbon in a Chinese megacity over five years from 2016 to 2020,
608 <https://doi.org/10.5281/zenodo.6473085>, 2022.

609 Wang, Q., Cao, J., Shen, Z., Tao, J., Xiao, S., Luo, L., He, Q., and Tang, X.: Chemical characteristics of PM_{2.5} during
610 dust storms and air pollution events in Chengdu, China, *Particuology*, 11, 70-77, 2013.

611 Wang, Q., Cao, J., Han, Y., Tian, J., Zhu, C., Zhang, Y., Zhang, N., Shen, Z., Ni, H., and Zhao, S.: Sources and
612 physicochemical characteristics of black carbon aerosol from the southeastern Tibetan Plateau: internal mixing
613 enhances light absorption, *Atmospheric Chemistry and Physics*, 18, 4639, 2018.

614 Wang, Q., Huang, R., Zhao, Z., Cao, J., Ni, H., Tie, X., Zhu, C., Shen, Z., Wang, M., Dai, W., Han, Y., Zhang, N., and

615 Prévôt, A. S. H.: Effects of photochemical oxidation on the mixing state and light absorption of black carbon in the
616 urban atmosphere of China, *Environmental Research Letters*, 12, 044012, <https://doi.org/10.1088/1748-9326/aa64ea>,
617 2017.

618 Wang, S., Nan, J., Shi, C., Fu, Q., Gao, S., Wang, D., Cui, H., Saiz-Lopez, A., and Zhou, B.: Atmospheric ammonia
619 and its impacts on regional air quality over the megacity of Shanghai, China, *Scientific Reports*, 5, 15842,
620 10.1038/srep15842, 2015.

621 Wang, Y., Wang, Q., Ye, J., Li, L., Zhou, J., Ran, W., Zhang, R., Wu, Y., and Cao, J.: Chemical composition and sources
622 of submicron aerosols in winter at a regional site in Beijing-Tianjin-Hebei region: Implications for the Joint Action
623 Plan, *Science of The Total Environment*, 719, 137547, <https://doi.org/10.1016/j.scitotenv.2020.137547>, 2020.

624 Wu, C. and Yu, J. Z.: Determination of primary combustion source organic carbon-to-elemental carbon (OC/EC) ratio
625 using ambient OC and EC measurements: secondary OC-EC correlation minimization method, *Atmos. Chem. Phys.*,
626 16, 5453-5465, 10.5194/acp-16-5453-2016, 2016.

627 Wu, C., Ng, W. M., Huang, J., Wu, D., and Yu, J. Z.: Determination of elemental and organic carbon in PM_{2.5} in the
628 Pearl River Delta region: inter-instrument (Sunset vs. DRI model 2001 thermal/optical carbon analyzer) and inter-
629 protocol comparisons (IMPROVE vs. ACE-Asia protocol), *Aerosol Science and Technology*, 46, 610-621, 2012.

630 Zhang, Q., Zheng, Y., Tong, D., Shao, M., Wang, S., Zhang, Y., Xu, X., Wang, J., He, H., Liu, W., Ding, Y., Lei, Y., Li,
631 J., Wang, Z., Zhang, X., Wang, Y., Cheng, J., Liu, Y., Shi, Q., Yan, L., Geng, G., Hong, C., Li, M., Liu, F., Zheng, B.,
632 Cao, J., Ding, A., Gao, J., Fu, Q., Huo, J., Liu, B., Liu, Z., Yang, F., He, K., and Hao, J.: Drivers of improved
633 PM_{2.5} air quality in China from 2013 to 2017, *Proceedings of the National Academy of Sciences*, 116,
634 24463-24469, 10.1073/pnas.1907956116, 2019.

635 Zhang, R., Jing, J., Tao, J., Hsu, S. C., Wang, G., Cao, J., Lee, C. S. L., Zhu, L., Chen, Z., Zhao, Y., and Shen, Z.:
636 Chemical characterization and source apportionment of PM_{2.5} in Beijing: seasonal perspective, *Atmos.*
637 *Chem. Phys.*, 13, 7053-7074, 10.5194/acp-13-7053-2013, 2013.

638 Zhang, R. J., Cao, J. J., Lee, S. C., Shen, Z. X., and Ho, K. F.: Carbonaceous aerosols in PM₁₀ and pollution gases in
639 winter in Beijing, *JOURNAL OF ENVIRONMENTAL SCIENCES*, 19, 564-571, 10.1016/S1001-0742(07)60094-1,
640 2007.

641 Zheng, B., Tong, D., Li, M., Liu, F., Hong, C., Geng, G., Li, H., Li, X., Peng, L., Qi, J., Yan, L., Zhang, Y., Zhao, H.,
642 Zheng, Y., He, K., and Zhang, Q.: Trends in China's anthropogenic emissions since 2010 as the consequence of clean
643 air actions, *Atmos. Chem. Phys.*, 18, 14095-14111, 10.5194/acp-18-14095-2018, 2018.

644 Zhu, C. S., Cao, J. J., Zhou, J. M., Liu, S. X., Dai, W. T., Zhang, T., Zhao, Z. Z., Shen, Z. X., Li, H., and Wang, P.: A
645 Case Study of Chemical Characteristics of Daytime and Nighttime Ambient Particles in Shanghai, China,
646 *ATMOSPHERE*, 6, 1141-1153, 10.3390/atmos6081141, 2015.

649 **Table 1** Averages (\pm one standard deviation), medians, and ranges (minimum to maximum) associated standard
650 deviations for the OC, EC, POC, SOC and PM_{2.5} concentrations (in $\mu\text{g m}^{-3}$) from Jan. 2016 to Dec. 2020.

		2016	2017	2018	2019	2020	Whole study
	Average	1.50±1.17	1.23±0.88	1.31±0.88	1.31±0.89	1.00±0.64	1.28±0.95
EC	Median	1.18	1.01	1.04	1.07	0.82	1.01

	Range	0.07~11.57	0.01~6.27	0.01~9.07	0.08~6.85	0.14~5.46	0.01~11.57
OC	Average	6.03±4.01	6.32±3.52	5.79±3.58	5.40±3.16	4.99±2.93	5.75±3.53
	Median	4.93	5.61	4.87	4.53	4.15	4.83
	Range	0.77~41.85	0.41~29.49	0.78~29.77	0.78~25.96	0.57~26.40	0.41~41.85
POC	Average	3.48±3.23	3.34±2.40	3.61±2.67	3.76±2.55	3.45±2.27	3.52±2.67
	Median	2.48	2.72	2.81	3.06	2.83	2.77
	Range	0.13~37.14	0.02~19.41	0.03~22.55	0.19~20.71	0.42~17.05	0.02~37.14
SOC	Average	2.56±1.94	2.98±2.25	2.17±1.75	1.64±1.20	1.53±1.35	2.24±1.87
	Median	2.10	2.38	1.71	1.41	1.20	1.76
	Range	0.01~18.13	0.01~25.79	0.01~19.87	0.01~18.84	0.01~14.87	0.01~25.79
TCA	Average	7.53±5.06	7.55±4.29	7.10±4.38	6.72±3.98	5.98±3.50	7.03±4.36
	Median	6.10	6.66	5.98	5.64	4.99	5.88
	Range	0.94~53.42	0.44~31.91	1.07~34.65	0.96~31.74	0.83~30.20	0.44~53.42
PM_{2.5}	Average	53.0±36.16	44.9±31.48	45.16±34.22	48.18±32.82	40.14±28.96	46.50±33.25
	Median	43.0	37.0	35.0	38.0	31.0	37.0
	Range	1.0~219.0	1.0~299.0	1.0~258.0	1.0~220.0	1.0~236.0	1.0~299.0

651 *TCA (total carbon aerosol) = EC+OC

652 *2016: Jan. 2016-Dec. 2016; 2017: Jan. 2017-Dec. 2017; 2018: Jan. 2018-Dec. 2018; 2019: Jan. 2019-Dec. 2019;

653 2020: Jan. 2016-Dec. 2020;

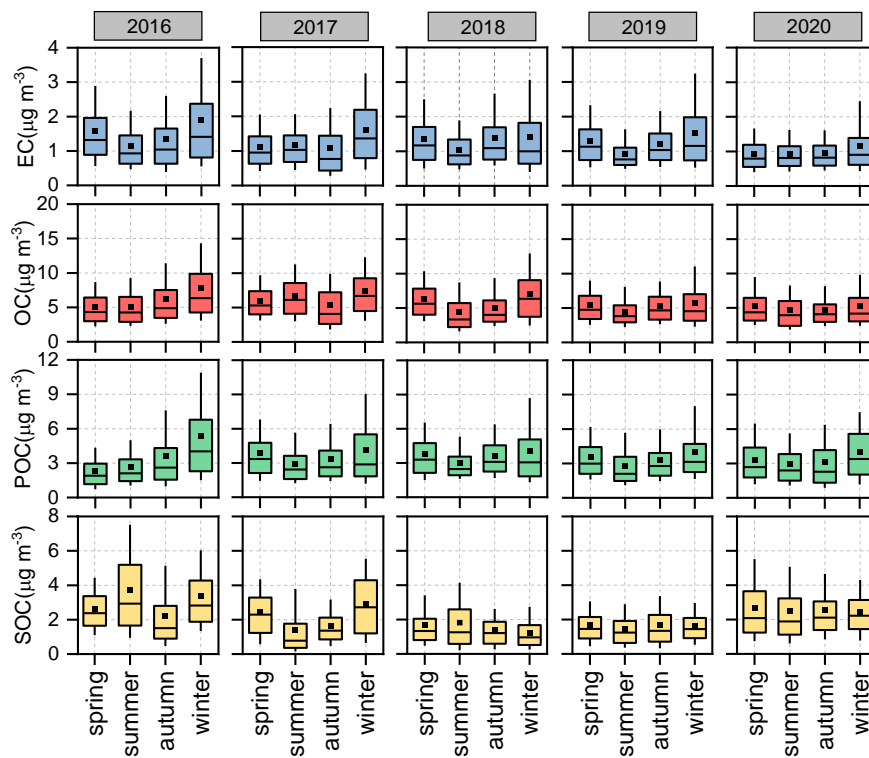


Figure 1 Seasonal variations of carbonaceous aerosol concentrations during weekdays and weekends over different years in Diashan Lake. (Spring: March, April, and May; summer: June, July, and August; Autumn: September, October, and November; Winter: January, February, and December). The box represents the 25th to 75th percentiles, the horizon line represents median, and the 10th and the 90th percentiles are the bottom and top whiskers, respectively.

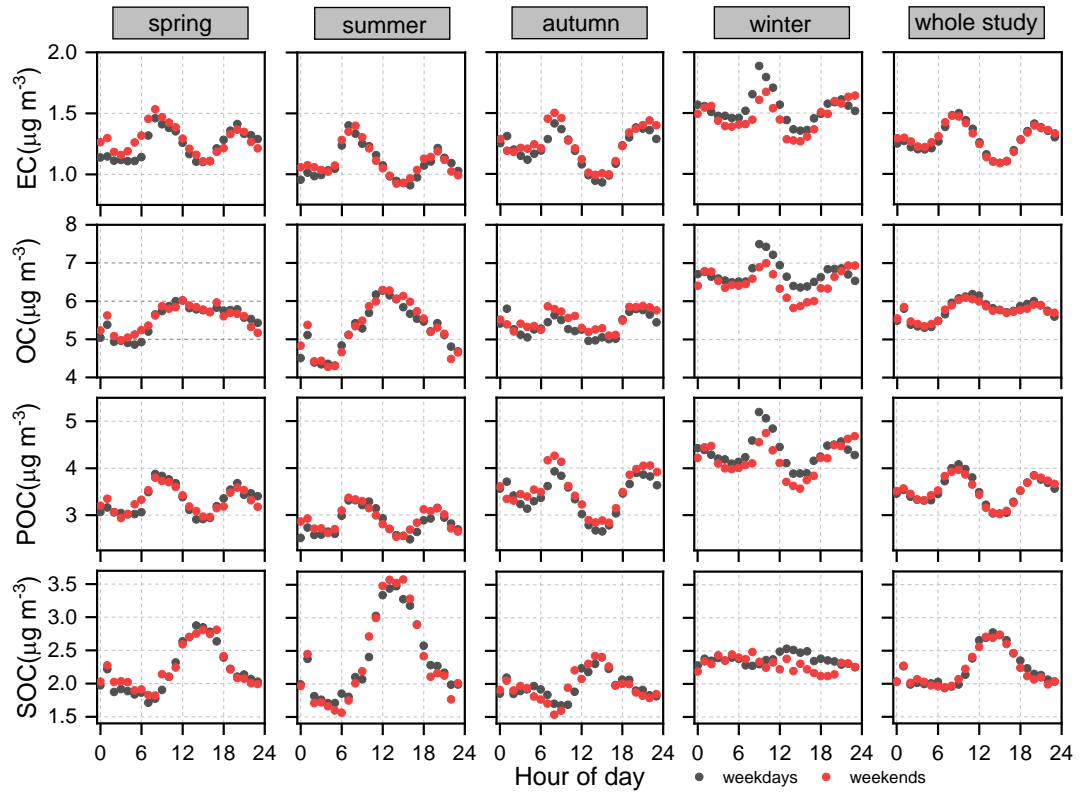


Figure 2 Diurnal variations of carbonaceous aerosol concentrations during weekdays and weekends in four seasons and the whole study period in Dianshan Lake.

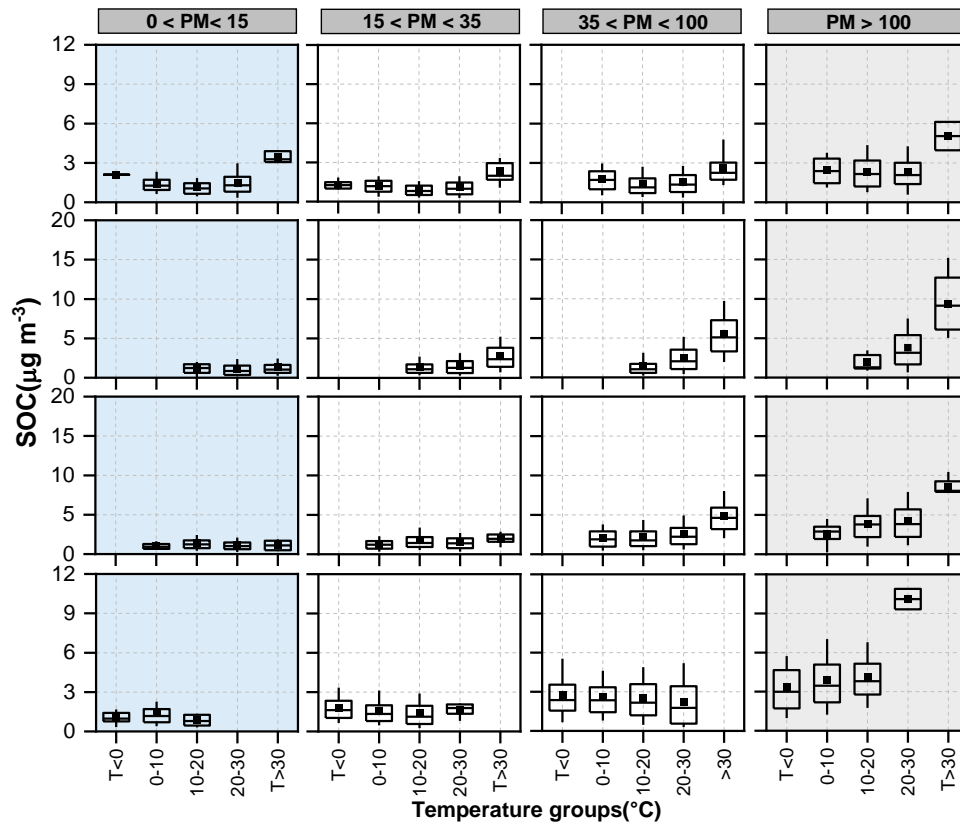


Figure 3 The SOC dependence of temperature in four different PM_{2.5} groups for each season during 2016-2020. The box represents the 25th to 75th percentiles, the horizon line represents the median, and the 10th and the 90th percentiles are the bottom and top whiskers, respectively.

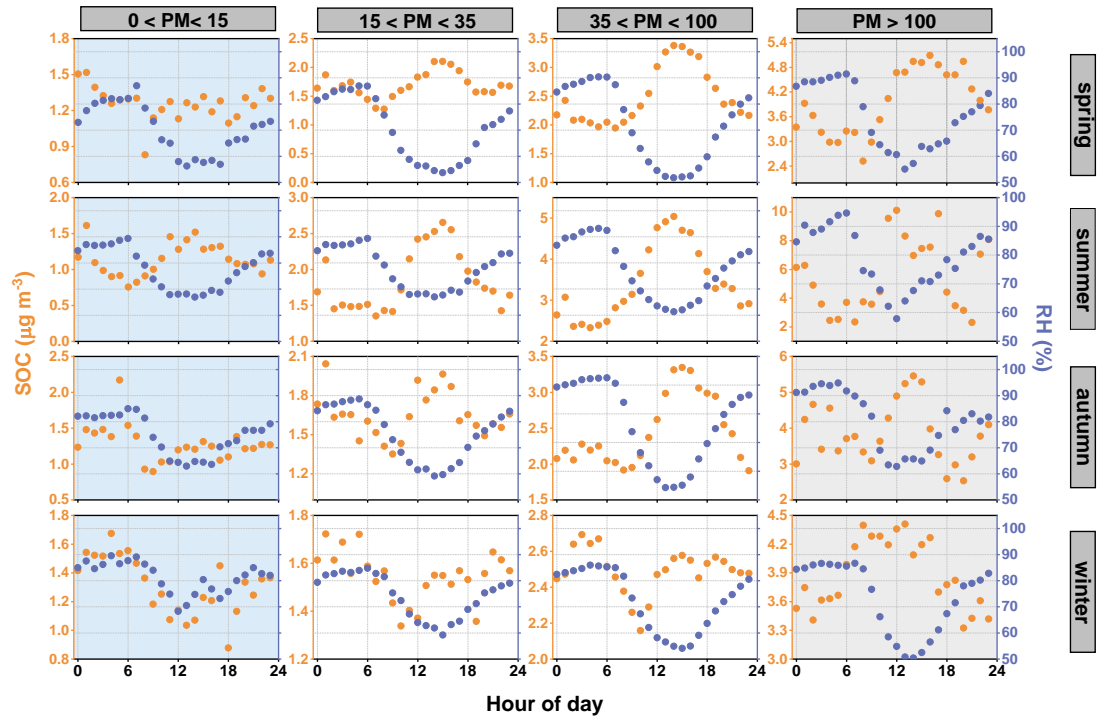


Figure 4 Diurnal variations of SOC concentrations and RH in four different PM_{2.5} groups for each season during 2016-2020.

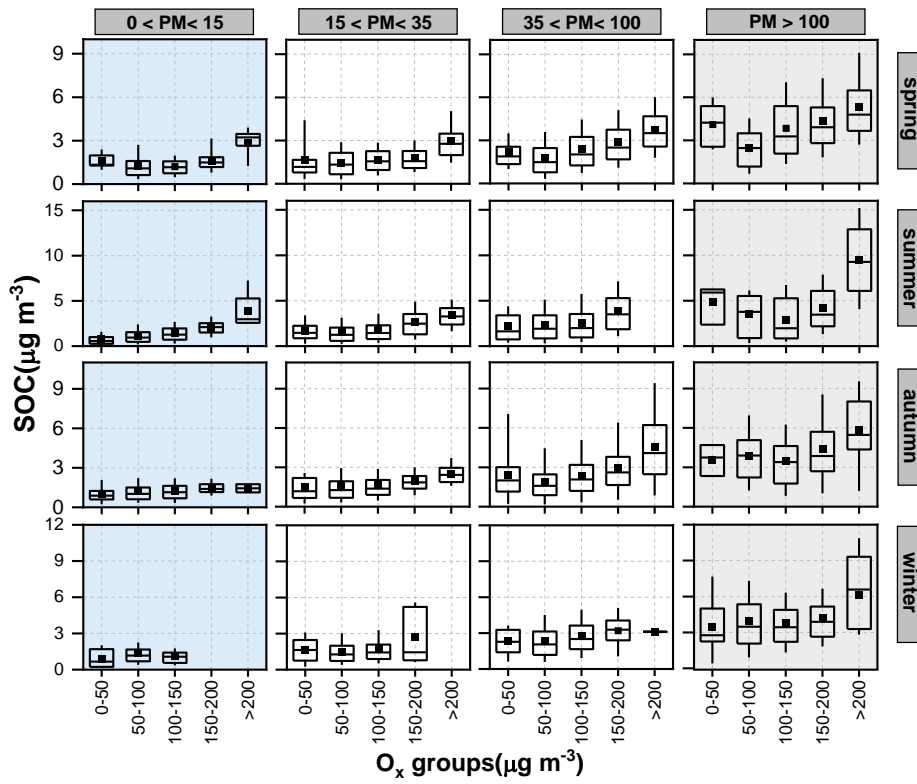


Figure 5 The SOC dependence of O_x in four different PM_{2.5} groups for each season during 2016-2020. The box represents the 25th to 75th percentiles, the horizon line represents the median, and the 10th and the 90th percentiles are the bottom and top whiskers, respectively.

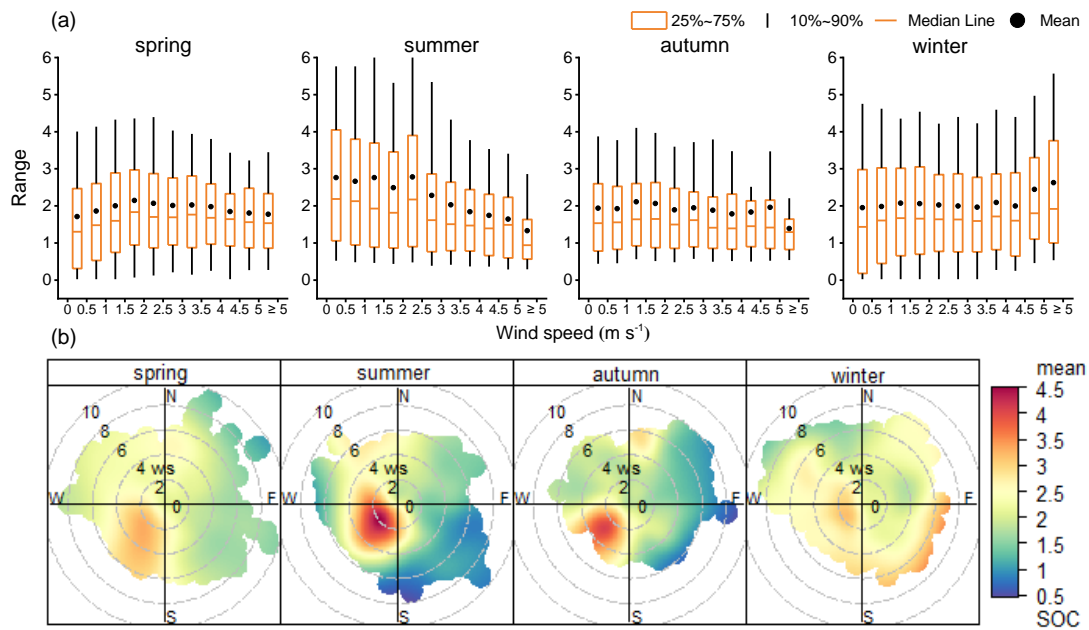


Figure 6 (a) Box plots of SOC mass concentrations as a function of wind speed sectors over the entire sampling period; (b) Bivariate polar plots of seasonal SOC concentrations ($\mu\text{g m}^{-3}$) over the entire sampling period in Dianshan Lake.

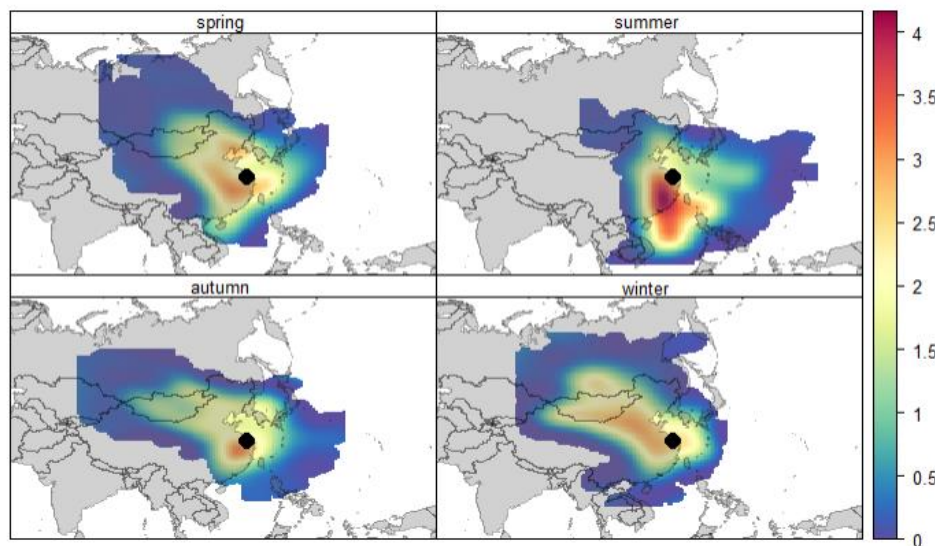


Figure 7 Back trajectory concentrations showing mean SOC concentrations ($\mu\text{g m}^{-3}$) based on the CWT approach in four seasons.

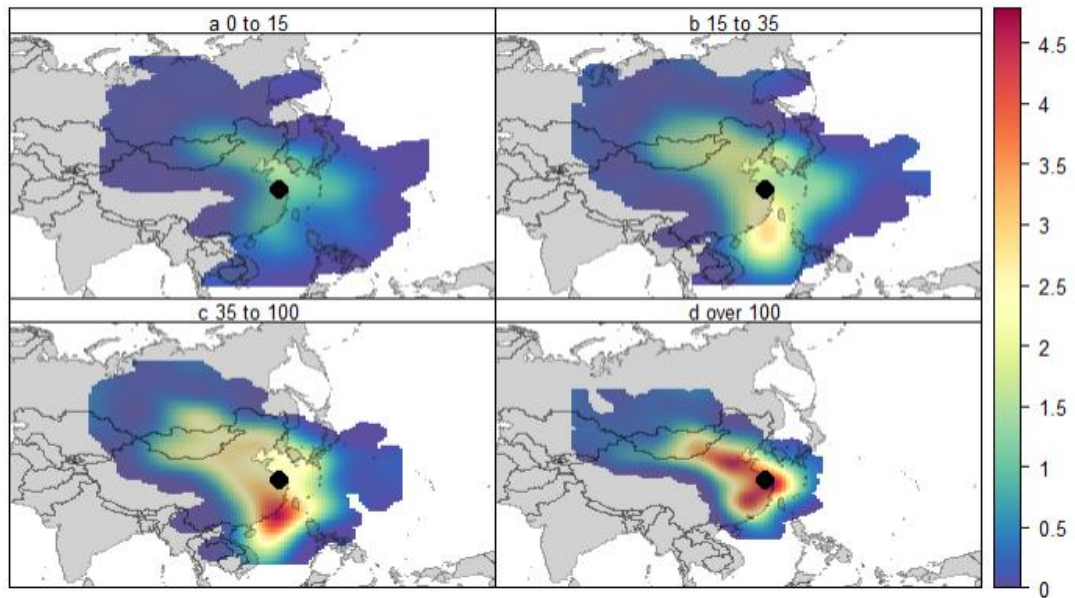


Figure 8 Back trajectory concentrations showing mean SOC concentrations ($\mu\text{g m}^{-3}$) based on the CWT approach in 4 different PM groups.