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1 Secondary aerosol formation in marine Arctic environments: A model measurement

2 comparison at Ny-Ålesund

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23 Abstract

- 24 In this study, we modeled the aerosol particle formation along air mass trajectories arriving at
- 25 the remote Arctic research stations Gruvebadet (67 m a.s.l) and Zeppelin (474 m a.s.l), Ny-
- 26 Ålesund during May 2018. The aim of this study was to improve our understanding of
- 27 processes governing secondary aerosol formation in remote Arctic marine environments. We
- 28 run the Lagrangian chemistry transport model ADCHEM, along air mass trajectories
- 29 generated with FLEXPART v10.4. The air masses arriving at Ny-Ålesund spend most of their
- 30 time over the open ice-free ocean. In order to capture the secondary aerosol formation from
- the DMS emitted by phytoplankton on the ocean surface, we implemented a recently
- 32 developed comprehensive DMS and halogen multi-phase oxidation chemistry scheme,
- 33 coupled with the widely used Master Chemical Mechanism (MCM).
- 34 The modeled median particle number size distributions are in close agreement with the
- 35 observations in the marine influenced boundary layer at near sea surface Gruvebadet site.
- 36 However, while the model reproduces the accumulation mode particle number concentrations
- at Zeppelin, it overestimates the Aitken mode particle number concentrations by a factor of





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- \sim 5.5. We attribute this to the deficiency of the model to capture the complex orographic
- 39 effects on the boundary layer dynamics at Ny-Ålesund. The model also reproduces the
- 40 average vertical particle number concentration profiles within the boundary layer (0-600 m
- 41 a.s.l.) above Gruvebadet, as measured with Condensation Particle Counters (CPCs) on board
- 42 an Unmanned Aircraft Systems (UAS).
- 43 The model successfully reproduces the observed Hoppel minima, often seen in particle
- 44 number size distributions at Ny-Ålesund. The model also supports the previous experimental
- 45 findings that ion mediated H₂SO₄-NH₃ nucleation can explain the observed new particle
- 46 formation in the marine Arctic boundary layer in the vicinity of Ny-Ålesund. Precursors
- 47 resulting from gas and aqueous phase DMS chemistry contribute to the subsequent growth of
- 48 the secondary aerosols. The growth of particles is primarily driven via H₂SO₄ condensation
- 49 and formation of methane sulfonic acid (MSA) through the aqueous-phase ozonolysis of
- 50 methane sulfinic acid (MSIA) in cloud and deliquescent droplets.

51 1. Introduction

52 The Earth's radiation budget is influenced both directly and indirectly by aerosols, which scatter and absorb the incoming short-wave radiation (direct effect) and serve as cloud 53 condensation nuclei (CCN, indirect effect), affecting both short and long-wave radiation 54 (Gantt et al., 2014; Oshima et al., 2020; Park et al., 2017; Scott et al., 2014). The Arctic 55 environments are susceptible to perturbations in the radiation balance, with some estimates 56 suggesting that, compared to the global average, the Arctic is warming at twice the rate, a 57 phenomenon termed as Arctic amplification (AMAP, 2011, 2017; Tunved et al., 2013). The 58 warming of the Arctic polar environment has accelerated sea ice loss, leading to a rapid 59 60 decline in the extent and duration of snow cover and increase in permafrost thaw (AMAP, 2011, 2017; Bengtsson et al., 2013). 61

62 The Arctic aerosol number concentration shows a pronounced seasonal variation, where the

- 63 late winter and early spring period is characterized by elevated accumulation mode aerosol
- 64 concentrations, accompanied by trace gases (mostly anthropogenic with long-range
- 65 transported trace elements such as sulfates, soot, and Peroxy Acyl Nitrates (PANs)). This
- 66 annually recurring phenomenon in late winter and spring is termed the Arctic Haze (Barrie,
- 67 1986; Lupi et al., 2016; Tunved et al., 2013). This contrasts with the summer period, when the
- atmospheric new particle formation is observed at Arctic sites, most likely due to low
- 69 background aerosol concentrations, increased photo-chemistry and biological activity
- 70 (Engvall et al., 2008; Heintzenberg et al., 2017; Tunved et al., 2013).





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The climate change driven Arctic sea ice loss has a profound impact on natural aerosol 71 production. Arrigo and van Dijken, 2015 found that decreasing and thinning of sea ice 72 73 increased the rates of phytoplankton net primary production by $\sim 20\%$ between the years 1998 74 and 2009. This can lead to an increase in the emissions of primary biogenic precursors such as dimethyl sulfide (DMS), nitrogen volatiles (e.g. alkyl-amines) (Dall'Osto et al., 2017a; 75 Dallósto et al., 2017b) and biological iodine species (Cuevas et al., 2018). DMS is emitted 76 77 into the atmosphere via air-sea gas exchanges (Park et al., 2017; Uhlig et al., 2019), and accounts for ~80% of global natural sulfur emissions (Kettle and Andreae, 2000; Uhlig et al., 78 2019). Methane sulfonic acid (MSA) and sulphuric acid (H₂SO₄) is formed via DMS gas-79 phase oxidation by OH and halogen species (Cl, Br) (Hoffmann et al., 2016; Kim et al., 2021; 80 Wollesen de Jonge et al., 2021). MSA and H₂SO₄, together with ammonia (NH₃) or amines, 81 act as precursors contributing to new particle formation (NPF) and subsequently to CCN 82 production, influencing cloud formation and radiative balance (Berndt et al., 2020; Dallósto et 83 al., 2017; Hoffmann et al., 2016; Kim et al., 2021). NH₃ plays a major role in particle 84 formation through stabilization of sulfuric acid clusters (Beck et al., 2021; Jokinen et al., 85 2018; Olenius et al., 2013). Depending on local parameters such as ocean pH, salinity and 86 temperature, global oceans can act either as a source or sink of NH_3 (Paulot et al., 2015). 87 88 Apart from participating in cluster formation, NH₃ influences the pH of marine aerosols by neutralizing the acid (H_2SO_4 and MSA) in the particles (Paulot et al., 2015). Though a few 89 90 potential sources of NH₃ are known, for example coastal sea bird colonies, pockets of open water and melting sea ice in summertime Arctic, the magnitude of the emissions remain 91 92 uncertain (Dall'Osto et al., 2019; Riddick et al., 2012; Wentworth et al., 2016). 93 DMS oxidation chemistry has been under focus, but uncertainties in climate predictions persist since the chemical transport models (CTMs) and global climate models (GCMs) 94 95 employ fixed MSA and SO₂ yields from gas-phase oxidation of DMS to calculate aerosol formation (Hertel et al., 1994; Hoffmann et al., 2016; Kloster et al., 2006; Wollesen de Jonge 96 97 et al., 2021). Including a detailed multi-phase (aqueous-phase chemistry coupled with gasphase chemistry) DMS chemistry in numerical models can overcome these uncertainties 98 (Barnes et al., 2006; Campolongo et al., 1999). Reaction intermediates such as dimethyl 99 sulfoxide (DMSO), dimethyl sulfone (DMSO2), methane sulfinic acid (MSIA) are water-100 soluble, and experiments have shown that neglecting aqueous phase chemistry leads to either 101 an under-estimation of modeled MSA (Campolongo et al., 1999), or an over-estimation of 102 gaseous SO₂ compared to measured values (Hoffmann et al., 2016). For example, the 103

104 temperature dependent ratio of MSA/non-sea-salt SO_4^{2-} (nss- SO_4^{2-}) is often used to estimate





- the contribution of DMS to sulfate budget (Avers et al., 1999; Barnes et al., 2006). 105 Campolongo et al., 1999 showed that modeling studies which included a multi-phase DMS 106 107 chemistry can bridge the gap between temperature-dependent observations and modeled MSA/nss-SO₄²⁻. Incorporating reactive halogens species over marine environments is crucial 108 in determining the DMS oxidation pathways to either SO₂ or MSA, the aging of marine 109 aerosols and the radiative properties of marine clouds (Hoffmann et al., 2016). Modeling 110 111 studies have shown that Cl and BrO radicals in the gas phase act as important DMS sinks (Chen et al., 2018; Wollesen de Jonge et al., 2021), further underlining the role of halogen-112 DMS chemistry in the marine boundary layer. 113 114 Recent DMS+OH oxidation experiments performed in the AURA chamber at Aarhus University show that MSA dominates the secondary aerosol mass formation (Rosati et al., 115 116 2021). Aerosol dynamics model simulations which intended to replicate the observations 117 during these AURA experiments, using the DMS gas-phase chemistry scheme from the 118 Master Chemical Mechanism, MCMv3.3.1, (Jenkin et al., 1997, 2015; Saunders et al., 2003), substantially underestimates the particle mass and number concentrations and the MSA:SO₄ 119 120 (Rosati et al., 2021, Wollesen de Jonge, 2021). Based on these findings, Wollesen de Jonge et 121 al. (2021) developed a new DMS multi-phase chemistry scheme based on MCM v3.3.1, CAPRAM DMS module 1.0 (DM1.0) (Hoffmann et al., 2016), a subset of the multi-phase 122 halogen chemistry mechanism CAPRAM Halogen Module 2.0 (HM2.0) (Bräuer et al., 2013) 123 and new reactions leading to the formation of hydroperoxymethyl thioformate (HPMTF). 124 125 With the new DMS multi-phase chemistry mechanism, the aerosol dynamics model could capture the observed particle number concentrations and secondary PM MSA and SO42-126 127 during DMS oxidation experiments performed at both dry and humid conditions at 0 °C and 128 20 °C in AURA. In this work, we have implemented the above mentioned DMS multi-phase chemistry 129 mechanism into ADCHEM (see Methods section) and modeled the aerosol formation along 130 air mass trajectories arriving at Ny-Ålesund. We compared the model results with 131 observations from Zeppelin (78°56' N, 11°53' E, 474 m a.s.l) and Gruvebadet (78°92' N, 11°90' 132 E, 67 m a.s.l). These two sites represent remote marine Arctic conditions. Gruvebadet 133 represents ground-level concentrations as it is well within the boundary layer (BL). Zeppelin 134 on the other hand, is most often above the BL in winter months and sometimes below the BL 135 136 during spring and summer months (Traversi et al., 2020). This implies that Zeppelin is often
- 137 influenced by long range transport, and Gruvebadet by local short-range effects (Traversi et
- al., 2020). This, demonstrates the complexity involved in capturing the atmospheric mixing





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- and secondary aerosol concentrations at Ny-Ålesund. The reason is that Svalbard has an
- 140 orographically complex terrain comprising of mountains, glaciers, fjords and flat lands that
- 141 introduce various micro-meteorological phenomena (Rader et al., 2021; Schemann and Ebell,
- 142 2020).

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144 **2. Methods**

145 Using the combined multi-phase DMS chemistry mechanism by Wollesen de Jonge et al.,

146 (2021), MCMv3.3.1 and the monoterpene peroxy radical autoxidation mechanism (PRAM,

147 Roldin et al., 2019; Xavier et al., 2019) we simulated aerosol particle formation within the

148 marine boundary layer (MBL) upwind and at Ny-Ålesund between 1st - 25^h May 2018, using

149 the Aerosol Dynamics, gas and particle-phase CHEMistry and radiative transfer model

150 ADCHEM (Öström et al., 2017; Roldin et al., 2011, 2019). We ran ADCHEM as a

151 Lagrangian model along the air mass trajectories arriving at Zeppelin every 3 hours during the

selected period (in total 200 trajectory simulations). FLEXPART v10.4 was used to calculate

the air mass trajectories and potential emission sensitivity fields (Pisso et al., 2019; Stohl et

al., 2005). The simulation results for the vertical distribution of newly formed aerosol (size <

155 12 nm) were validated against concurrent measurement data available from the ALADINA

- 156 (Application of Light-Weight Aircraft for Detecting in situ Aerosol) campaign, wherein a
- 157 UAS was used to investigate horizontal and vertical distribution of aerosol profiles in the
- 158 marine boundary layer (ABL) (Lampert et al., 2020). Additionally, modeled particle number
- size distributions and PM₁₀ chemical compositions were compared to the available measured
- 160 particle number size distributions and PM_{10} filter samples at both Gruvebadet and Zeppelin
- 161 measurement stations.

162 2.1 Air mass trajectories and potential emission sensitivity fields

163 We employed the Lagrangian particle dispersion model FLEXible PARTicle

164 (FLEXPARTv10.4) to assess the emission sensitivities or "footprints" of air-masses origin

- arriving at Zeppelin during the simulation period. FLEXPART is a stochastic model used to
- 166 compute dispersion of hypothetical particles, based on mean, turbulent and diffusive flows
- 167 which can be run backwards in time to estimate air mass history at a site (Pisso et al., 2019).
- 168 European Center for Medium-Range Weather Forecasts (ECMWF) ERA5 reanalysis
- 169 meteorology with 137 height levels, 1-hour temporal and 0.5° x 0.5° spatial resolution, was
- used as an input to FLEXPART (ERA5 hourly data on single levels from 1979 to present.





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- 171 Copernicus Climate Change Service (C3S) Climate Data Store (CDS). last access 30th April
- 172 2021, 10.24381/cds.adbb2d47, ERA5 hourly data on pressure levels from 1979 to present.
- 173 Copernicus Climate Change Service (C3S) Climate Data Store (CDS). last access 30th April
- 174 2021, 10.24381/cds.bd0915c6). The air-mass history was simulated 7-day backwards in time
- and arriving at Zeppelin (474 m a.s.l) every 3 hours (at 00:00, 03:00, 06:00, 09:00, 12:00,
- 176 15:00, 15:00 and 21:00 UTC) for the entire simulation period (1st 25th May 2018).
- 177 FLEXPART calculated normalized emission sensitivity fields were combined with oceanic
- 178 emissions (DMS, dibromomethane, tribromomethane, iodomethane), NH₃ from sea-bird
- 179 colonies and anthropogenic emissions (NH₃, SO₂, CO, NO_x) derived from global inventories
- 180 (see section 2.2). This was done to obtain representative emissions that consider the complete
- 181 emsission source regions along the trajectories, upwind of the measurement station.
- 182 Additional meteorological parameters such as temperature, pressure, sea surface temperature,
- 183 specific humidity and cloud liquid water content from ERA5 reanalysis dataset were extracted
- along the trajectories and provided as inputs to ADCHEM.
- 185

186 2.2 Gas and primary particle emissions

- 187 Emissions of gas-phase biogenic volatile organic compounds (VOCs) α -pinene, β -pinene Δ 3-
- 188 carene, limonene, isoprene and β-caryophyllene were modeled with a 1 dimensional version
- of MEGAN v2.04 (Model of Emissions of Gases and Aerosols from Nature 2.04) (Guenther
- et al., 2006). Gas-phase emissions of marine halogens such as tribromomethane (CHBr₃),
- dibromomethane (CH₂Br₂), iodomethane (CH₃I) were retrieved from CAMS-OCE Global
- 192 oceanic emissions (CAMS-GLOB-OCE) which are available as daily means with a spatial
- resolution of 0.5°x0.5° (Granier et al., 2019; Ziska et al., 2013). CAMS-GLOB-OCE also
- provides gas-phase DMS emissions with the same temporal and spatial resolution (Granier et
- al., 2019) calculated with the air-sea flux parameterization and emission fluxes described in
- 196 (Lana et al., 2011; Nightingale et al., 2000). NH₃ emissions from seabird colonies were
- acquired from a global emission inventory (Riddick et al., 2012). To account for additional
- 198 NH₃ fluxes from the open ocean, we used an estimated sea surface equilibrium $NH_{3(g)}$
- saturation concentration of 0.5 nmol/m³ (12.2 ppt at standard temperature and pressure (STP))
- which approximately correspond to a surface ocean ammonium concentration of 0.125
- 201 mmol/m³ (or ~3ppb, calculated based on equation 3 & 4 from Wentworth et al., 2016) at a sea
- 202 surface temperature of +2 °C. The sea surface temperature for the study period varied between
- 203 0 °C-14 °C along the trajectories. The estimated surface ocean ammonium concentrations is in





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close agreement with the concentration estimated by the global ocean biogeochemical model 204 COBALT (Stock et al., 2014) in the North Atlantic ocean, but up to a factor of ~5 higher than 205 206 the concentrations simulated with other ocean biogeochemical models and/or model setups 207 (Paulot et al., 2015). Therefore, we performed model sensitivity runs with a sea surface equilibrium NH_{3(g)} concentration of 0.1 and 1 nmol/m³. The NH_{3(g)} equilibrium saturation 208 concentrations represent the ambient surface gas-phase concentration at which the air-sea flux 209 changes direction, with a net downward flux from air to sea if the ambient $\mathrm{NH}_{3(g)}$ exceeds the 210 equilibrium gas concentrations and vice versa (Wentworth et al., 2016). For the anthropogenic 211 trace gas and primary particle emissions, we used the CAMS-GLOB-ANT v2.1 inventory, 212 with a spatial resolution of 0.1°x0.1° (Granier et al., 2019). 213 In this work, we used the sea surface temperature (SST) and wind speed dependent sea-spray 214 215 aerosol (SSA) emission parameterization by Sofiev et al. (2011), (further referred to as Sofiev11). Sofiev11 used a modified source function based on the parameterization of 216 217 Monahan et al. (1986) and experiments by Mårtensson et al. (2003) and SEAS campaign by Clarke et al. (2006). The modified source function in Sofiev11 provides extrapolated SSA 218 emissions between size ranges of 10 nm-10 µm, with appropriate correction functions 219 220 employed for SST deviating from 298.15 K (Sofiev et al., 2011). Sofiev11 SSA parameterization shows that with increasing temperatures, emission flux for larger particles 221 increases while the emission fluxes for smaller particles decreases (Barthel et al., 2019; 222 Sofiev et al., 2011). We performed sensitivity tests using the temperature and wind speed 223 224 dependent SSA parameterization by Salter et al., (2015), (further referred to as Salter15). Both the Salter15 and Sofiev11 are valid between 10 nm-10 µm. Model simulation 225 226 comparisons between Sofiev11 and Salter15 have shown that the SSA parameterization from 227 Sofiev11 has a stronger temperature dependence and higher particle number concentration 228 emissions in the Aitken mode but result in lower PM₁₀ emissions at temperatures below 25 °C (Barthel et al., 2019). 229

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231 **2.2 ADCHEM**

For this study, ADCHEM was employed as a 1 - dimensional column model with 40

logarithmically vertical layers, extending up to ~2600m. The model time step used for

simulations was 30 seconds. The vertical atmospheric turbulent diffusion was solved using a

modified Grisogono turbulent diffusivity scheme (Jeričević et al., 2010; Öström et al., 2017;

236 Roldin et al., 2019). The ADCHEM aerosol module includes new particle formation,





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- Brownian coagulation, condensation and evaporation of particles, and finally the dry and wet 237 deposition of both particles and gases. The particle number size distributions were represented 238 239 using 100 size bins ranging from 1.07 nm to 10 μ m dry diameter. Clouds were assumed to be present in the model grid cells when the bulk liquid water content (LWC, extracted along the 240 trajectory from ERA5 datasets) was greater than 0.01 g m⁻³. As a default, we used a constant 241 cloud supersaturation (S) of 0.5% and the particles were activated into cloud droplets, if the 242 243 calculated water vapor supersaturation above the particle surface (S_c , calculated using Köhler theory) was smaller than S. During the cloud processing, each activated cloud droplet was 244 assumed to take up an equal amount of liquid water corresponding to the total bulk LWC 245 divided by the calculated number concentration of activated cloud droplets. The gas-liquid 246 droplet mass transfer and dissolution of 50 species in total, including HCl, HNO₃, H₂SO₄, 247 NH₃, HIO₃, H₂O₂, O₃, OH, BrO, NO₃, DMSO, MSIA, MSA and HPMTF and their irreversible 248 reactions in the interstitial and activated cloud droplets are treated by the multi-phase 249 chemistry mechanism (see Wollesen de Jonge et al. (2021) for details). The kinetic pre-250 processor (KPP) (Damian et al., 2002) was used to generate the multi-phase chemistry 251 mechanism used in this study. 252 Recent observations of NPF at Ny-Ålesund have confirmed the importance of ion-mediated 253 H₂SO₄-NH₃ nucleation in spring with MSA and H₂SO₄ condensation contributing to the 254 subsequent growth of particles (Beck et al., 2021; Lee et al., 2020). In this work, the 255 Atmosphere Cluster Dynamics Code (ACDC) (McGrath et al., 2012; Olenius et al., 2013) was 256 257 coupled with ADCHEM (Roldin et al., 2019). ACDC was used to model NPF, which involved H₂SO₄ clustering with NH₃ via both neutral and ion-induced pathways with an 258 259 ionization rate of 1.7 cm 3 s⁻¹. ACDC was used to solve the evolution of molecular H₂SO₄-NH₃ clusters by considering the loss of clusters by collisions, evaporation or coagulation 260 261 scavenging onto larger aerosol particles. At each time step, the flux of clusters (up to ~ 5 H₂SO₄ and 5 NH₃ each) growing out of the ACDC molecule-cluster domain represents the 262 263 NPF rate. These newly formed clusters are assigned to the corresponding smallest particle size bin at 1.07 nm in diameter in ADCHEM, which then simulates the condensational growth 264 of particles and losses due to evaporation, coagulation, and wet and dry deposition. 265 For all simulations, we used model output from the closest height levels which can represent 266 Gruvebadet (model height of 73.5 m a.s.l) and Zeppelin (model height of 486.0 m a.s.l). 267
- 268 Sensitivity Tests

269 Alongside the main ADCHEM simulations, *BaseCase*, we performed nine complementary





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270 scenario runs to assess the impact of different processes on the modeled aerosol concentrations. We performed simulations without aerosol in-cloud processing (Cloudoff), to 271 272 check the impact of in-cloud processing on the growth of aerosols. We investigated the effect of higher PM₁₀ particle emissions on the chemical composition of secondary aerosols, using 273 the sea-spray emission parameterization based on Salter et al., 2015 (SalterSSA). Simulations 274 275 were conducted to assess the impact of lower and higher ammonia sources over the open 276 ocean (LowNH₃, HighNH₃). A sensitivity test without precipitation (NoPrecip) was performed to test the influence of precipitation on number concentration and particle composition. Since 277 cloud supersaturation is critical to the activation of particles and is highly uncertain, we 278 performed two simulations with low and high cloud supersaturation (S=0.2%, SSat=0.2 and 279 280 S=0.8%, Ssat=0.8) to test its impact on the modelled particle distributions. We performed a simulation without new particle formation (NPFoff), and finally one simulation without the 281 dissolution and irreversible aqueous chemistry of the intermediate DMS oxidation products, 282 SO₂ and halogens (woDissolution), implying that MSA, H₂SO₄ and HIO₃ is only formed in the 283 gas-phase. Table 1. summarizes the setup for different model sensitivity test. 284

285

286 Table 1. Model sensitivity tests performed alongside the main *BaseCase* simulations to test

the effect of different parameters on secondary aerosol formation. These sensitivity tests focus

288 on the role of in-cloud processing and aqueous phase chemistry, the NH₃ emissions from open

289 ocean, SSA parameterization and cloud supersaturation. The sea surface equilibrium $NH_{3(g)}$

290 concentrations in ppt are provided in the brackets.

Simulation	In-cloud Processing	NH _{3(equilibrium)} (nmol/m ³ , ppt)	SSA parameterization	Precipitation
BaseCase	On	0.5 (12.2)	Sofiev11	On
SalterSSA	On	0.5 (12.2)	Salter 15	On





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Cloudoff	Off	0.5 (12.2)	Sofiev11	On
LowNH _{3,}	On	0.1 (2.4)	Sofiev11	On
HighNH ₃		1 (24)		
NoPrecip	On	0.5 (12.2)	Sofiev11	Off
SSat0.8, SSat0.2	On	0.5 (12.2)	Sofiev11	On
NPFoff	On	0.5 (12.2)	Sofiev11	On
WoDissolution	On, but no dissolution and irreversible chemistry of intermediate DMS oxidation products	0.5 (12.2)	Sofiev11	On

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292 **2.3 Measurements**

293 We utilized comprehensive measurements from the Ny-Ålesund sites, Zeppelin and

294 Gruvebadet during the period of 1st - 25th May 2018. Since 2017, the atmospheric observatory

at Gruvebadet, which is located about 700 m southwest of Ny-Ålesund village at almost sea

296 level (67 m.s.l), hosted Neutral cluster and Air Ion Spectrometer (NAIS, Manninen et al.,

2010; Mirme and Mirme, 2013) for semi-permanent measurements. Here we use NAIS

298 measured number size distribution of naturally charged (ions) in diameter size ranges between

299 0.8 nm-40 nm and neutral particles in the size range of 2.5 nm-42 nm, with a temporal

300 resolution of two seconds.





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- During the measurement period, a scanning mobility particle sizer (SMPS), was operated to 301 measure particle number size distribution in the diameter size range of 10 - 470 nm at 302 303 Zeppelin. Concurrent SMPS data (TSI 3034, 54 channels) with diameter size ranging from 10 to 470 nm from Gruvebadet were also available (Dall'osto et al., 2019; Moroni et al., 2020), 304 thus, enabling us to compare the modeled particle number size distribution with the measured 305 size distributions at both measurement stations. Daily resolution continuous aerosol samples 306 307 with PM₁₀ cutoff were collected at Gruvebadet using a Tecore Skypost low-volume sampler (Amore et al., 2022). The detection limit for Na⁺ was 0.0001 μ g m⁻³ and 0.0002 μ g m⁻³ for Cl⁻, 308 NH_4^+ and SO_4^{2-} . Since the field blank medians at Gruvebadet were less than 1 percentile of 309 sampled values, the field blanks were not subtracted from the sampled values (Amore et al., 310 2022). 311 312 Vertical particle number concentration profiles were obtained using UAS ALADINA (Bärfuss et al., 2018; Lampert et al., 2020), which was operated during the simulation period. 313 314 ALADINA were operated up to a height of 850 m a.s.l., thus can be used for a potential closure between the two different research sites of Gruvebadet and Zeppelin. ALADINA is 315 equipped with two condensation particle counters (CPCs Model 3007, TSI Inc., St. Paul, 316
- 317 MN, USA), measuring in the size ranges of 3 nm 2 μ m (CPC1) and ~12 nm 2 μ m (CPC2)
- 318 (Lampert et al., 2020; Petäjä et al., 2020). The difference between CPC1 and CPC2 provides
- an estimate of particle number concentrations in the size of 3 12 nm (PN₃₋₁₂), which was
- used as an indicator of NPF. Alongside the CPCs, a host of other instruments measuring
- 321 meteorological parameters were operated in unison, the description of which can be found in
- 322 Bärfuss et al., (2018) and Lampert et al., (2020).

323 Evaluating temporal aspects of model performance

- 324 The modeled PM_{10} inorganic chemical composition was evaluated against the measured PM_{10}
- inorganic chemical composition using statistical estimates such as, normalized mean bias
- 326 (NMB), Pearson correlation coefficient (r), root mean squared error (RMSE) and fraction of
- 327 predictions within a factor of 2 of the observed values (FAC2). These tests were used to
- 328 evaluate modeled values (M_i) against observation values (O_i) at both the measurement sites.

329 Pearson correlation coefficient was calculated using the formula:

$$r = \frac{1}{n} \sum_{i=1}^{n} \frac{(O_i - \bar{O})}{\sigma_O} \frac{(M_i - \bar{M})}{\sigma_M}$$
(Eq 1)
331

332 Where σ_0 and σ_M are standard deviations of the observed and modeled values, respectively.





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- 333 Normalized mean bias (NMB) indicates if the predictions are over or underestimating the
- 334 observed values, with the factor representing the under or over estimation. NMB was
- 335 calculated using Eq. 2:

$$NMB = \frac{\sum_{i=1}^{n} (M_i - O_i)}{\sum_{i=1}^{n} O_i}$$
(Eq 2)

336 337

338 Root mean squared error (RMSE) was calculated using Eq. 3:

339
$$RMSE = \sqrt{\sum_{i=1}^{n} \frac{(M_i - O_i)^2}{n}}$$
(Eq 3)

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FAC2 is a robust metric defined as the percentage of predictions which are within a factor of2 of the observed values (Eq. 4):

$$Fac2 = 0.5 \le \frac{M}{O} \le 2.0$$
 (Eq 4)

344 3. Results and Discussion

In the following sections, we analyze and evaluate the model results against comprehensive measurements in Ny-Ålesund. In sub-section 3.1, we focus on the particle number size distributions at both sites, followed by gas-phase concentrations and PM₁₀ inorganic chemical composition (sub-section 3.2) and the vertical nano-particle concentration profiles (subsection 3.3). Finally, in sub-section 3.4, we analyze the results from the model sensitivity tests.

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352 **3.1 Particle number size distributions**

353 Figure 1(a) and (b) show the observed and predicted particle number size distributions at

354 Gruvebadet for the BaseCase simulation. Figure 1(a) includes SMPS observations starting

from 10 to 470 nm and NAIS observations for neutral particles in the range 2.5 nm-10 nm

356 (boundary marked by the black line) since NAIS data below 2.5 nm cannot be relied upon,

- owing to the presence of corona generated ions (Jayaratne et al., 2017; Manninen et al., 2011,
- 358 2016).

359 In the *BaseCase* simulations, ADCHEM reproduces the general trends of the observed

 360 particle number concentrations. For example, the model captures particle formation on 2^{nd}





- May followed by an increasing number of Aitken and accumulation mode particles during the 361 days of 3rd - 4th May, which is the result of more polluted air masses arriving at Ny-Ålesund 362 363 from the European continent (Figure S1). Similarly, the model reproduces the particle formation on the 20th of May, specifically in the size ranges 2-8 nm, but overestimates the 364 Aitken mode and accumulation mode particle concentration on the 21^{st} of May. In general, the 365 model predicts the formation of new particles with reasonable accuracy during the selected 366 367 period. However, the model tends to underestimate the nucleation mode particle number concentrations between 10-25 nm (PN10-25 nm) around noon, and overestimate the 368 concentrations during the morning and evening (Figure 2a). The model and measurements 369 show an apparent time delay in the formation of new particles larger than 10 nm. While the 370 measurements show a peak at 11 am the simulated $PN_{10-25 \text{ nm}}$ shows a maximum at 3 am and 6 371 pm. The modeled $PN_{10-25 nm}$ maximum around 6 pm is likely a result of the formation of new 372 particles around noon, which grow to >10 nm in diameter during the afternoon and evening 373 by condensation of H_2SO_4 . The predicted Aitken ($PN_{25-100 \text{ nm}}$) and accumulation mode particle 374 concentrations (PN>100 nm) which form few days upwind of the station are overall, in good 375 agreement with the measurements, which show a minor diurnal trend (Figure 2b-c). The 376 measurements indicate that at Gruvebadet, PN10-25 nm contributes the most significant fraction 377 378 of measured total number concentrations with 45.3%, while PN_{25-100 nm} and PN_{100-470 nm} contribute 30.5% and 23.94% respectively. However, the simulations predict greater 379 380 contribution of Aitken mode (\sim 53.85%) to total number concentration, with the PN_{10-25 nm} and $PN_{100-470 \text{ nm}}$ accounting for ~ 36.58% and 9.57% respectively (Figure S2). 381 382 Figure 2 shows the measured size distribution in panel (a) and simulated size distribution in 383 panel (b) for Zeppelin. At Zeppelin, the model overestimates the number concentration in nucleation and Aitken modes (also cf. Figure S3, supplementary). The particle number size 384 385 distribution measurements at Zeppelin indicate that the relative contribution of the three modes (nucleation, Aitken and accumulation) varies to some extent when compared to 386 387 Gruvebadet. Measurements show that at Zeppelin, PN_{10-25 nm} contributes ~33.46%, PN_{25-100 nm} 46.43% and $PN_{>100 \text{ nm}}$ 20.11% to the total particle number concentrations. The model predicts 388 lower relative contribution of PN10-25 nm (26.94%), and a greater contribution of PN25-100 nm 389 (63.44%) to the total simulated particle number concentrations. The diurnal trends at Zeppelin 390 agree well with earlier measurements conducted at Zeppelin in spring by Ström et al. (2009). 391 Additionally, the measured diurnal pattern at Zeppelin varies in comparison to Gruvebadet. At 392 Zeppelin, the PN_{10-25 nm} concentrations peak in the afternoon and evening. The modeled 393
- $PN_{10-25 \text{ nm}}$ shows only a weak diurnal trend. It should be noted that the measurements show a





- 395 time delay of around 3 hours in the peak $PN_{10-25 nm}$ concentrations at the two sites (Figure S2
- and S3). This is possibly be a result of vertical mixing and dilution effects modulating the
- 397 observed particle number concentrations at sites situated at different altitudes, similar to
- observation made at Zeppelin and Corbel by Ström et al. (2009).
- 399 ADCHEM considers the formation of new particles via both the ion-mediated and neutral
- 400 H₂SO₄-NH₃ clustering pathways. Beck et al (2021) observed dominant contribution of
- 401 negative H₂SO₄ -NH₃ clusters to secondary particle formation in May 2017 at Ny-Ålesund,
- 402 with HIO₃ playing a small role in the initial particle formation. However, the discrepancy in
- 403 the modeled and observed diurnal trends of $PN_{10-25 nm}$ could indicate that there are other
- 404 sources or vapors that might potentially contribute to the particle formation. Other possible
- 405 NPF mechanism may involve amines (Olenius et al., 2013) and pure biogenic highly oxidized
- 406 molecule (HOM) (neutral and ion induced) nucleation (Kirkby et al., 2016). We speculate that
- 407 the exclusion of these other mechanisms (HIO₃, H₂SO₄-amines and HOM driven particle
- 408 formation) might result in the discrepancies in the modeled and observed particle number
- 409 concentration diurnal trends. HIO3 induced particle formation could, e.g. play an important
- 410 role if the air masses upwind of Ny-Ålesund traverse over the sea-ice covered regions
- 411 (Baccarini et al., 2020; Beck et al., 2021).







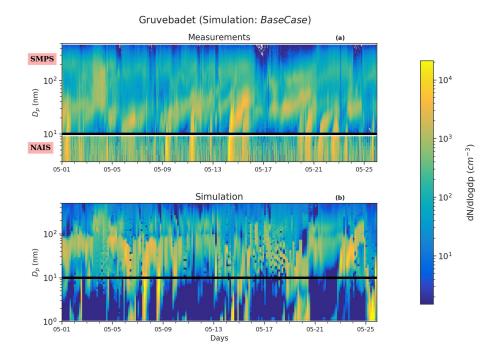


Figure 1. Particle number size distribution at Gruvebadet for *BaseCase*. The panel (a) shows the measurement data for the period $1-25^{\text{th}}$ May from SMPS (10 nm-470 nm) and NAIS (2.5 -10 nm) and the panel (b) provides the modeled particle size distribution. The black line at 10 nm denotes the boundary above which SMPS data starts and NAIS data ends. The abscissa indicates the time for the entire simulated duration. The ordinate in Figure 1 for both panels (a) and (b) indicates the particle diameter (D_p, nm).







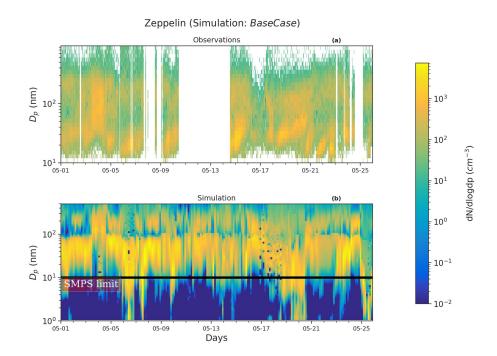


Figure 2. Particle size distribution at Zeppelin. The panel (a) shows the measurement data for
the period 1-25th May from SMPS and the panel (b) provides the simulated particle size
distribution for the *BaseCase* simulations. The abscissa and ordinates are similar to Figure 1.

422

Figure 3 presents the median particle number size distribution for the BaseCase simulation at 423 both Zeppelin and Gruvebadet, with respective 25th and 75th percentiles, for the entire selected 424 period. At Gruvebadet, the modeled and measured median particle number size distributions 425 426 are in reasonable agreement for both Aitken and accumulation mode. However, the model over predicts the median Aitken mode concentrations at Zeppelin by a factor ~ 5.5. The 427 modeled Aitken mode peak at both measurement sites is ~50 nm, while the measured Aitken 428 mode peak is \sim 30 nm. Though the modeled accumulation mode peak is at a larger size (\sim 150 429 nm), compared to the measured accumulation mode peak (~110 nm), the predicted values are 430 in good agreement with the monthly averaged accumulation mode peak location measured at 431 Zeppelin in earlier studies (~160-170 nm, Dall'Osto et al., 2019). 432

433 The discrepancy between the modeled and measured particle concentrations at Zeppelin can





17

- 434 be caused by the underlying complexity of modeling the boundary layer dynamics at an
- 435 elevated site, such as Zeppelin. The vertical mixing of aerosols along the up-slope or down-
- 436 slope of a mountain site is difficult, if not impossible for a 1- dimensional column model,
- 437 since it is unable to capture the topographical influence on locally varying wind speeds or
- 438 latent and sensible heat fluxes (Mikkola, 2020; Wainwright et al., 2012).

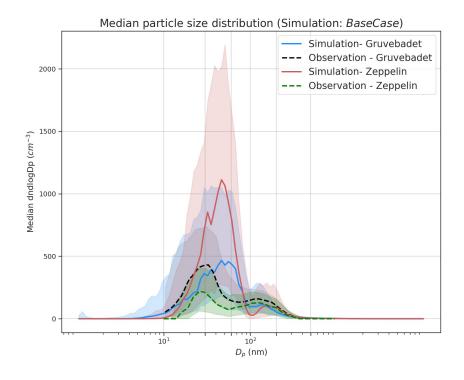


Figure 3. Median particle number size distribution at Gruvebadet and Zeppelin for both
modeled (*BaseCase* simulations) and measured values. The shaded areas indicate the 25th and
75th percentiles for both model and measured median particle number size distribution. At
Zeppelin, the simulated median size distribution is calculated for periods only when SMPS
data were available.

444

Another detectable feature in the median particle number size distribution is the diameter of
the Hoppel minimum (Hoppel et al., 1985, 1986), and the role of in-cloud processing in
forming this minimum. A Hoppel minimum is often observed in marine air masses (Fossum
et al., 2018; Tunved et al., 2013; Zheng et al., 2018) and is attributed to in-cloud processing of





18

- aerosols, with chemical processing (e.g., sulfate production via oxidation of dissolved SO₂) 449 (Feingold and Kreidenweis, 2000; Hoppel et al., 1986), and coalescence of droplets playing a 450 key role (Flossmann and Wobrock, 2019; Hoppel et al., 1986; Hoppel and Frick, 1990; Noble 451 and Hudson, 2013). It has been estimated that, on average, aerosols take part in about 10 non-452 precipitating cloud cycles before it is removed from the atmosphere by wet scavenging 453 (Hoose et al., 2008; Hoppel et al., 1986; Rosenfeld et al., 2014). These non-precipitating 454 455 cloud cycles facilitate the formation of hygroscopic accumulation mode particles, with low critical supersaturation (S_c) that readily activates to cloud droplets during subsequent cloud 456 cycles, thus growing to larger sizes. This is because the activated particles undergo chemical 457 processing, gas-to-particle conversions, coalescence and coagulation with other interstitial 458 particles. Upon evaporation of water, the emerging dry particles have a larger size and lower 459 S_{c_2} leading to a minimum being formed between the un-activated and activated cloud droplets 460 (Herenz et al., 2018; Hudson et al., 2015; Noble and Hudson, 2013). The diameter at which 461 the Hoppel minimum is observed varies depending on the cloud supersaturation and particle 462 composition (Hoppel et al., 1986; Hudson et al., 2015), with Hoppel minima sizes observed in 463 ranges from 60 nm at Zeppelin Ny-Ålesund to around 90 nm at Tuktoyatuk, Canada (Herenz 464 et al., 2018; Tunved et al., 2013). 465
- 466 The median particle number size distribution in Figure 3 shows that at both stations, the
- 467 measured Hoppel minima is around ~ 60 nm, while the simulated Hoppel minima are around
- the size of ~ 100 nm at both sites. This difference in location of Hoppel minima can be
- 469 attributed to the assumed value of S=0.5% in the model. The value of S used in the model lies
- in the range of typical marine stratocumulus clouds, which can vary between 0.1 1%
- 471 (Fossum et al., 2018; Quinn et al., 2017).

472 3.2 Gas and particle-phase chemical composition of important precursors

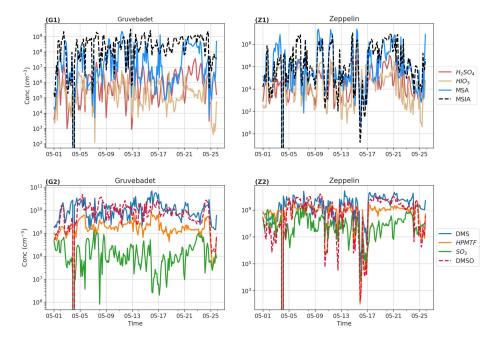
- 473 Figure 4 shows the simulated gas-phase precursor and main DMS oxidation product
- 474 concentrations including H₂SO₄, MSA, MSIA, HPMTF, SO₂ and DMSO, for the entire period
- 475 at the height levels representing Gruvebadet (G1 and G2), and Zeppelin (Z1 and Z2). The SO₂
- 476 gas-phase concentrations are in the order of 10^{6} - $10^{9} \, \text{\# cm}^{-3}$ (with monthly mean values 1.7 x
- 477 $10^8 \,\#\,\mathrm{cm}^{-3}$), which is a factor of 2.3 higher than the average concentrations measured for
- 478 spring 7.6 $\times 10^7 \, \text{\# cm}^{-3}$ by (Lee et al., 2020) at Zeppelin. The monthly mean simulated H₂SO₄
- 479 gas phase concentrations (6.8 x $10^5 \,\# \, \text{cm}^{-3}$) also agree well with the estimated H_2SO_4 proxy
- 480 (Eq. S1, supplementary) spring average values of $7.5 \times 10^5 \text{ } \text{ cm}^{-3}$ (Lee et al., 2020) at
- 481 Zeppelin. Measurements of H₂SO₄ at Gruvebadet from May 2017 indicate monthly mean





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- 482 concentrations around $\sim 10^6 \, \text{\# cm}^{-3}$ (Beck et al., 2021). The modeled H₂SO₄ concentrations at
- 483 Gruvebadet are $3 \times 10^6 \# \text{ cm}^{-3}$, implying a reasonably good model performance in predicting
- 484 gaseous precursor concentrations. Simulated gas concentrations of MSA $(10^5-10^8 \,\#\, \text{cm}^{-3} \text{ also})$
- agrees well with the measurements made at Gruvebadet in May 2017 by (Beck et al., 2021),
- 486 wherein they measured daily averages of MSA gas concentrations in the order of $10^7 \, \text{\# cm}^{-3}$.
- 487 The low modeled values of MSA and DMSO gas phase concentrations at the height
- 488 representing Zeppelin (e.g. between 15/05 17/05) coincide with the period where the
- 489 planetary boundary layer height (PBLH) is below the altitude of Zeppelin station (cf. Figure
- 490 S4 supplementary). Overall, we can conclude that the modeled precursor gas concentrations at
- 491 the two measurement sites are, in general, good agreement with earlier measurements at the
- 492 two sites.



Gas concentrations (Simulation: BaseCase)

- Figure 4. Gas-phase concentrations for the *BaseCase* simulations. The upper two panels (G1) and (Z1) show the gas-phase concentrations at Gruvebadet and Zeppelin respectively, for compounds H_2SO_4 (red), HIO₃ (gold), MSA (cyan), MSIA (dotted black) and the lower panels
- 497 (G2) and (Z2) show the gas-phase concentrations for DMS (blue), HPMTF (orange), SO₂





20

498 (green) and DMSO (dotted red). Note the different ordinate scales.

499

500	Figure 5 (a) shows the simulated median mass size distribution of compounds Cl ⁻ , Na ⁺ , MSA,
501	SO_4^{2-} , NH_4^+ , and NO_3^- for the <i>BaseCase</i> runs in the lowest model layer. Figure 5(a) indicates
502	that the nucleation mode particles are composed mainly of SO_4^{2-} and NH_4^+ , while MSA, Cl ⁻
503	and Na^+ dominate PM for larger particles. The observed and modeled high $MSA_{(g)}$
504	concentrations in comparison to $H_2SO_{4(g)}$ at Ny-Ålesund is not reflected in the respective
505	vapor contribution to the nano-particle growth. This is because, in contrast to H ₂ SO ₄ , MSA is
506	not a non-volatile condensable compound. The gas-to-particle partitioning of MSA requires
507	co-condensation and dissolution of (NH ₃) (Hodshire et al., 2019) or the existence of cations
508	such as Na ⁺ which decreases the particle acidity ([H ⁺]). Figure 5(b) shows the relative mass
509	fraction of the above-mentioned compounds to PM at different sizes. SO_4^{2-} and NH_4^+
510	dominate the mass for particles in the nucleation and Aitken mode. SO_4^{2-} contributes ~74%
511	and ~71% to nucleation and Aitken mode PM, with its contribution decreasing for
512	accumulation (100 nm-1 μ m) and coarse (>1 μ m) mode PM (~6% and 3.36% respectively)
513	(Table 2). NH_4^+ contribution follows a similar trend, as SO_4^{2-} , with 12.34% and 6.95%
514	contribution to nucleation and Aitken mode PM, but insignificant for accumulation and coarse
515	mode PM (Table 2). The loss of primary sea spray aerosols due to wet scavenging promoted
516	the growth of secondary aerosol particles in the nucleation and Aitken mode by NH_4^+ and
517	SO_4^{2-} as seen in Figure 5 (b). Na ⁺ (~32.9%), Cl ⁻ (~39.5%) and MSA (20.45%) are the
518	dominant contributors to accumulation and coarse mode PM. In the BaseCase simulations,
519	gas-phase SO ₂ dissolves in the cloud droplets, and is oxidized by H ₂ O ₂ into SO ₄ ²⁻ (Wollesen
520	de Jonge et al., 2021). Previous modeling studies have shown that a very small fraction of
521	MSA is formed in the gas phase. Instead, most MSA is formed via ozonolysis of MSIA in the
522	aqueous phase (Hoffmann et al., 2016; Wollesen de Jonge et al., 2021). It should be noted that
523	HIO_3 and NO_3^- have an insignificant contribution to total PM_{10} , amounting to ~0.05% and
524	0.17% respectively.

525

Table 2: The table shows simulated fractional contribution of different compounds to total

527 PM in different size regimes of nucleation (total $PM_{<25nm})$, Aitken (total $PM_{25\,\text{--}\,100\,\text{nm}})$ and

528 accumulation - coarse (total PM_{>100nm}) mode.

Species	Total	Total Aitken	Total	Total coarse	Total PM ₁₀
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	nucleation mode PM fraction (PM _{<25nm}) (%)	mode PM fraction (PM _{25 - 100 nm}) (%)	accumulation mode PM fraction (PM ₁₀₀ nm-1 µm) (%)	mode PM fraction (PM>1 µm) (%)	fractional contribution (%)
SO4 ²⁻	73.99	71.00	5.96	3.36	6.67
$\mathrm{NH_4^+}$	12.34	6.95	0.13	0.06	0.21
Cl	2.36	1.98	39.96	43.36	39.54
Na ⁺	8.02	11.92	32.90	34.17	32.91
MSA	3.26	8.12	20.58	18.85	20.45
HIO ₃	0.004	0.004	0.05	0.05	0.05
NO ₃ -	0.006	0.01	0.17	0.15	0.17

Median mass size distribution and relative mass fraction (Simulation: BaseCase)

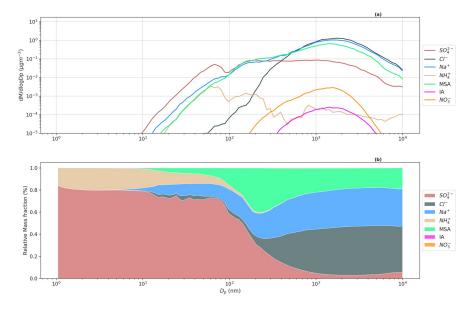


Figure 5. Simulated median mass size distribution for BaseCase simulations. The upper panel 530 (a) shows the median mass size distribution for compounds Cl⁻, Na⁺, MSA, SO₄²⁻, NH₄⁺, HIO₃ 531 and NO₃⁻ for the entire size distribution ranging from 1.07 nm-10 μ m. The lower panel (b) 532 shows the relative mass fractions or contribution of compounds Cl⁻, Na⁺, MSA, SO₄²⁻, NH₄⁺, 533 HIO₃ and NO₃⁻ to total non-refractory PM at different sizes.





- measurement stations. The model prediction of PM₁₀ Cl, Na⁺, SO₄² and NH₄⁺, was evaluated 536 using statistical metrics such as NMB, FAC2, correlation coefficient (r) and RMSE (Table 3). 537 538 Though the model does well in simulating the trends of $PM_{10} SO_4^{2-}$, Na⁺ and Cl⁻ at Zeppelin (r values of 0.35, 0.51 and 0.6 respectively), it is unable to predict the NH_4^+ trends accurately (r 539 540 = -0.08). Pearson correlation (r -values) at Gruvebadet are in the range of 0.29-0.34 for $PM_{10} NH_4^+$, 541 SO_4^{2-} , Na⁺ and Cl⁻ implying that the model trends are reasonably consistent with the measured 542 trends. However, at both Gruvebadet the NMB values for Na⁺ and Cl⁻ are quite large (1.81 543 544 and 1.05), indicating a large overprediction of the predicted values, while $PM_{10} NH_4^+$ and
- SO_4^{2-} at Gruvebadet is underpredicted (NMB = -0.88 and -0.28 respectively). In contrast, at
- Zeppelin, the modeled PM SO₄²⁻ is overestimated (NMB=1.96). Likewise, large RMSE and
- negligible FAC2 values, for $PM_{10} Na^+$, and Cl^- imply discrepancies between the predicted and
- measured values, indicating that the model is overestimating PM_{10} Na⁺ and Cl⁻ at Gruvebadet
- and PM_{10} SO₄²⁻ at Zeppelin. In summary, the model tends to overpredict PM_{10} Na⁺, Cl⁻ and
- SO_4^{2-} concentrations, but on the other hand, does reasonably well in predicting the daily
- measured trends. Additionally, the modeled PM_{10} Cl⁻/Na⁺ molar ratio at Gruvebadet and
- 552 Zeppelin is ~0.79 and ~0.95, respectively. This is much higher than the observed PM_{10}
- 553 Cl⁻/Na⁺ molar ratio at both sites (\sim 0.39). One likely reason for this is the overestimated sea
- spray aerosol emissions. The PM_{10} Cl/Na⁺ molar ratios give a measure of the acidic nature of
- aerosol, since increased condensation of strong acid MSA and H₂SO₄ increases acidity of
- aerosols thereby causing loss of Cl⁻ (dechlorination) as HCl (Ayers et al., 1999; Frey et al.,
- 557 2020). Thus, increased availability of H₂SO₄ and MSA in particle phase in Aitken mode
- 558 particles results in acid-induced Cl⁻ loss from sea-spray particles.







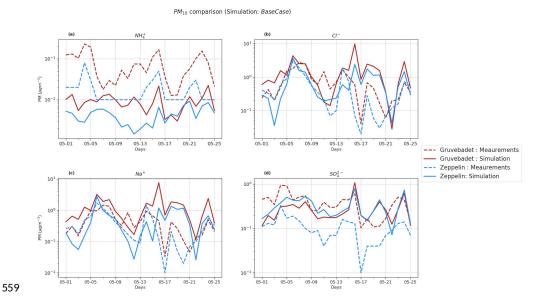


Figure 6. PM₁₀ comparison of *BaseCase* simulations with daily filter samples from
Gruvebadet and Zeppelin for the entire modeled period. Panel (a) shows PM₁₀ NH₄⁺, (b)
shows PM₁₀ Cl⁻, (c) shows PM₁₀ Na⁺ and (d) shows PM₁₀ SO₄²⁻ filter samples. The dotted lines
in each panel indicate measurement values, and the solid line denotes simulated values. The
ordinate is plotted in log scale to better visualize the low values.

565

566 **Table 3**: Evaluation of modeled PM₁₀ values at both sites of Gruvebadet (G) and Zeppelin (Z)

Species	Normalized mean bias factor (NMB)	Correlation coefficient (r)	RMSE (µg m ⁻³)	FAC2
$\mathrm{NH_4}^+$	$-0.88^{\rm G}, -0.76^{\rm Z}$	0.34^{G} , -0.08^{Z}	$0.09^{\rm G}, 0.02^{\rm Z}$	$0.04^{\rm G}, 0.2^{\rm Z}$
Na ⁺	1.81 ^G , 0.36 ^Z	$0.29^{\rm G}, 0.51^{\rm Z}$	$1.67^{\rm G}, 0.55^{\rm Z}$	0.4 ^G , 0.48 ^Z
Cl	1.05 ^G , 0.39 ^z	$0.24^{\rm G}, 0.60^{\rm Z}$	2.08 ^G ,0.74 ^Z	0.24 ^G , 0.44 ^Z
SO4 ²⁻	-0.28 ^G , 1.96 ^Z	0.33 ^G , 0.35 ^Z	$0.27^{\rm G}, 0.26^{\rm Z}$	0.6 ^G , 0.24 ^Z

567 for the four particle-phase species Cl^{-} , Na^{+} , SO_4^{2-} and NH_4^{+} .

568

569 **3.3 Vertical profiles of ultra-fine particle**

570 Figure 7 (a) shows the measured vertical $PN_{3-12 nm}$ concentrations from CPC onboard the UAS

571 for four measurement periods overlayed onto simulated vertical profiles. Figure 7 (b) and (c)

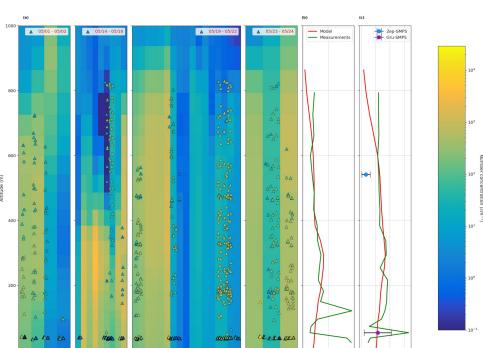




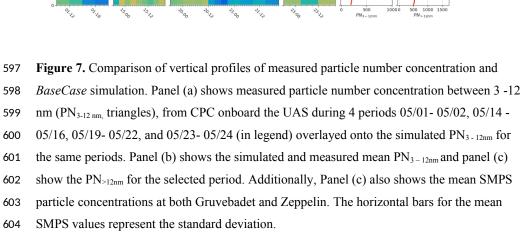
572	show the mean vertical profiles for $PN_{3-12 nm}$ and $PN_{>12 nm}$ for both the <i>BaseCase</i> simulation and
573	UAS measurements for the entire selected period. The model underestimates the measured
574	$PN_{^{3\text{-}12}\text{nm}}$ and $PN_{^{>}12\text{nm}}$ vertical particle number concentrations are underestimated by the model
575	below 200 m a.s.l. The NMB for $PN_{3-12 \text{ nm}}$ and $PN_{>12 \text{ nm}}$ is -0.28 and -0.14, respectively,
576	implying that the model underestimates the particle number concentrations. Both the modeled
577	and measured mean particle number concentrations for $\text{PN}_{\text{3-12nm}}$ and $\text{PN}_{\text{>12nm}}$ are in good
578	agreement between the heights of 200-600 m a.s.l. The lower calculated concentrations of
579	modeled mean particle number concentrations above 600 m a.s.l is most likely affected by
580	higher turbulence in the transition zone from the boundary layer to the free troposphere,
581	which might cause a large mixing of aerosol particles. It should be noted that, at Gruvebadet,
582	the mean SMPS particle number concentrations are in good agreement with the modeled
583	particle number concentrations. However, at the altitude of the Zeppelin station, both the
584	model and UAS measurements of $PN_{>12nm}$ are substantially higher (factor of 4) than the mean
585	particle number concentrations measured with the SMPS at Zeppelin. This finding further
586	strengthens the conclusion that the complex orography at Ny-Ålesund highly affects the
587	variability in the vertical scale, which may cause this discrepancy in the observed and
588	modeled particle number concentrations at Zeppelin (see section 3.1). The UAS
589	measurements were carried out at the airport on Ny-Ålesund (and the UAS was flown around
590	Ny-Ålesund) where the boundary layer measurements, like the model, most likely resemble
591	the general Arctic marine boundary layer conditions. Figure S8 shows the influence of
592	different sensitivity simulations on the modeled vertical particle number concentrations. The
593	large spread in the modeled vertical particle number concentrations in Figure S8, highlights
594	the importance of constraining uncertain parameters such as cloud supersaturation and NH ₃
595	gas emissions, to better simulate secondary aerosol formation in marine polar regions.







Vertical profiles for measured and modeled PN3-12nm (runtype: BaseCase)



605

606 3.4 Sensitivity Tests

607 In this section, we will discuss the results from the sensitivity tests that we performed to

608 complement the main *BaseCase* simulations.

609 Median particle size distribution for sensitivity tests

610 The sensitivity study *Cloudoff* was performed to test how in-cloud processing affects the



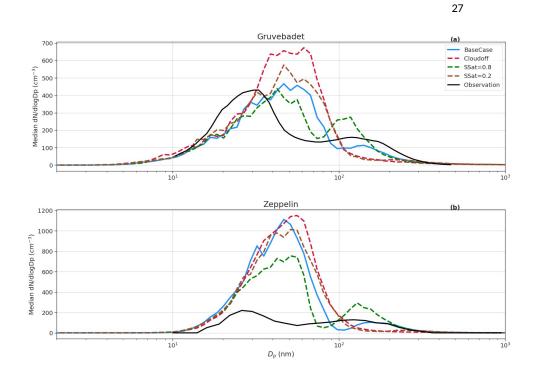


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formation of larger particles, especially the accumulation mode (Figure 8). In the *Cloudoff* 611 test, in-cloud processing was switched off in the model and the RH was set to just below 612 613 supersaturation (99.9999%) in the model grid cell where clouds (RH=100.5%) exists in the BaseCase runs. The aim of the Cloudoff simulation was to investigate if the model can 614 capture the observed accumulation mode without aerosol cloud processing. It is clear from 615 Figure 8 (a) and (b) that in *Cloudoff* simulations, the median size distribution lacks the 616 617 accumulation mode and Hoppel minima and has a higher Aitken mode particle concentration compared to either BaseCase or the measured median size distribution. This further 618 emphasizes the importance of in-cloud processing in activation of particles to CCN sizes and 619 their growth to larger sizes. Another noteworthy point in *Cloudoff* simulations is the larger 620 number concentration of particles <10 nm compared to other cases. One plausible reason is 621 the lack of activated cloud droplets, since the large surface areas of activated droplets are 622 efficient at Brownian scavenging of smaller particles (Hudson et al., 2015). Likewise, the 623 median particle number size distribution from the sensitivity tests with lower cloud 624 supersaturation (S) of 0.2% SSat=0.2, reduces the accumulation mode particles, since there 625 are fewer particles with $S_c < S$ available for activation. Increasing S to 0.8% increases 626 accumulation mode particles, since more particles with $S_c < S$ are activated to cloud droplets 627 628 (Aitken mode concentration decreases with respect to *BaseCase* simulations, since more smaller particles are activated into cloud droplets). Therefore, simulated results show that 629 630 increasing the cloud supersaturation results in a higher number of smaller particles being activated into cloud droplets and shifts the simulated Hoppel minima close to the measured 631 632 sizes. Figure S5, in supplementary shows median particle size distribution for all sensitivity 633 tests. The SalterSSA sensitivity test underestimates both the Aitken and accumulation mode 634 635 concentrations at Gruvebadet (Figure S5, supplementary). The Salter sea-spray parameterization produces ~ 2 magnitudes fewer Aitken mode particles compared to Sofiev 636 637 et. al, 2011, while the coarse mode particle emissions using *SalterSSA* parameterization are higher than Sofiev et. al 2011. This can cause MSA, H₂SO₄ and NH₃ to partition onto coarse 638 mode particles rather than contributing to NPF and growth of the nucleation and Aitken mode 639 640 particles, which substantially lowers the Aitken and accumulation mode number concentrations. The NPFoff simulation from Figure S5 shows lower Aitken mode 641 concentrations, implying that the main contributor to Aitken mode particle number 642 concentrations are the secondary aerosols rather than the primary sea-salt particles. 643







- 645 Figure 8: Median size distribution at Gruvebadet (panel (a)) and Zeppelin (panel (b)) for all
- 646 the sensitivity tests *Cloudoff*, *SSat*=0.8, and *SSat*=0.2 (colored dashed lines) including

647 BaseCase (blue solid line) and observations (black solid line).

- 648 Another parameter of uncertainty is the concentration of NH₃ in the marine atmosphere. The
- $LowNH_3$ simulations, as expected, result in lower Aitken mode particles, whereas $HighNH_3$
- simulations show an overprediction of Aitken mode concentrations (Figure S5,

651 supplementary). This underlines the necessity of constraining ocean and marine emissions of

652 NH₃ to better predict the aerosol particle formation in marine polar environments.

653 Particle phase comparison for sensitivity tests

- 654 Figure 9 shows the contribution of constituent compounds to PM at different particle sizes
- with respect to the *BaseCase* simulation. The overall mean contribution of SO_4^{2-} and MSA to
- total PM_{10} decreased by ~8% and 11% respectively, in *Cloudoff* runs compared to the
- 657 BaseCase simulations. It is expected that in non-cloud conditions there is a reduction in SO_4^{2-}
- and MSA PM contribution because of the reduced partitioning of gaseous SO₂ to the cloud
- 659 droplets (for PM SO₄²⁻ formation) and inhibition of MSIA ozonolysis in the cloud droplets
- 660 (leading to PM MSA formation) (Chen et al., 2018; Hoffmann et al., 2016; Wollesen de Jonge
- 661 et al., 2021). This is observed for accumulation mode particles between size ranges of 100 nm





- to 1 μ m which is characterized by lower SO₄²⁻ and MSA PM. On the other hand, PM SO₄²⁻ and 662 MSA increase for coarse mode particles (> 1 μ m). Without cloud droplet activation the 663 664 deliquescent sea spray coarse mode particles become a major liquid water reservoir where MSIA and to a lesser extent SO₂ are dissolved and oxidized into MSA and SO₄²⁻, which partly 665 explains the increase in PM MSA and SO_4^{2-} for sizes > 1 μ m. The results from *Cloudoff* 666 667 simulation agrees with the findings from Wollesen de Jonge et al., 2021, who found that MSA 668 was almost exclusively formed in the aqueous phase via MSIA ozonolysis in cloud droplets and deliquescent particles during and in between in-cloud periods. PM SO4²⁻ in Cloudoff runs 669 is mainly driven via condensation of H₂SO₄, since an increase in SO₂ gas-phase 670 concentrations ($\sim 42\%$ with respect to *BaseCase*) promoted gas-phase H₂SO₄ production 671 (increase of ~44% with respect to *BaseCase*), and therefore H_2SO_4 derived PM SO_4^{2-} . 672 In the *woDissolution* simulation, all the PM MSA and SO_4^{2-} are a result of the condensation of 673 MSA(g) and H2SO4(g), since irreversible aqueous-phase chemistry is switched off. The overall 674 contribution of PM SO₄²⁻ to the total PM₁₀ increases by ~12% relative to the BaseCase run, 675 while on the other hand, the contribution of PM MSA decreases by $\sim 87\%$ (relative to 676 *BaseCase*). The lower PM_{10} MSA in *woDissolution* simulation emphasizes the importance of 677 678 aqueous-phase formation of MSA to the growth of particles. The effect of precipitation on modeled PM (NoPrecip) indicates an increase in PM Na⁺ and Cl⁻ of ~112% and 119% 679 respectively, as compared to *BaseCase* (Figure S7). This is because of the decrease in the wet 680 681 deposition of aerosol and sea-spray particles by rain events and below cloud scavenging. The 682 consequence of neglecting precipitation results in increased condensation sink for H₂SO₄ and NH₃ (increase of 62% and 22% in PM SO₄²⁻, NH₄⁺ respectively), but since sea-spray aerosols 683 684 are not scavenged by the wet removal process, the overall fractional contribution to PM by SO₄²⁻, NH₄⁺ and MSA is lower relative to *BaseCase* runs. 685 SalterSSA simulation results in higher PM Cl⁻ and Na⁺ (470% and 371% increase respectively) 686 compared to BaseCase runs. This is because Salter15 SSA parameterization produces larger 687 mass emission fluxes in size ranges > 1 µm compared to Sofiev11 SSA parameterization 688 689 (Barthel et al., 2019). Additionally, there is an increase of ~19% in PM MSA, largely due to
- 690 formation of MSA in larger deliquescent coarse mode particles.
- 691







Contribution to PM relative to BaseCase

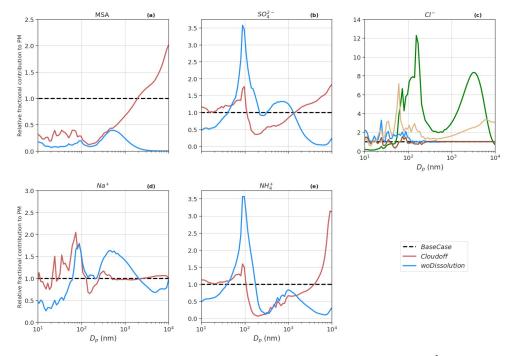


Figure 9: Contribution of constituent compounds, namely, MSA (panel (a)), SO_4^{2-} (panel (b)), Cl⁻ (panel(c)), Na⁺ (panel (d)) and NH₄⁺ (panel (e)) to PM with respect to *BaseCase* (the black dotted line).

695 4. Summary and conclusions

696 In this work, we attempt to simulate secondary aerosol formation at remote Arctic sites of

697 Gruvebadat and Zeppelin, Ny-Ålesund, during the period of 1st - 25th of May 2018. We used

698 the 1-dimensional column model ADCHEM which was run along FLEXPART generated

699 Lagrangian trajectories. Since the air mass spend most of their time over the open ocean

vupwind of Ny-Ålesund, we use a comprehensive multi-phase DMS chemistry scheme coupled

701 with MCMv3.3.1 and PRAM.

To 2 In the model, new particles are formed via ion-mediated H_2SO_4 -NH₃ nucleation, with the

initial particle growth mainly driven by condensation of H_2SO_4 , while the secondary PM_{10}

MSA and SO_4^{2-} contribution was mainly formed by oxidation of MSIA and SO_2 in the

aqueous phase. At Gruvebadet, the modeled median particle number size distribution agrees

- reasonably well with the measurements, however, at Zeppelin, the simulated Aitken mode
- 707 median concentration is overestimated by a factor of 5.5. This relatively large discrepancy in





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modeled and measured particle size distributions at Zeppelin, and likewise the large 708 difference between the measured particle number size distributions at Gruvebadet and 709 710 Zeppelin, can to a large extent be explained by the orographic effects at Zeppelin which distorts the atmospheric boundary layer dynamics. Thus, while the model generally is able to 711 capture the particle number size distribution dynamics in the marine boundary layer, as 712 measured at the near sea level Gruvebadet site, it generally cannot capture the observations at 713 714 the mountain station of Zeppelin, which often lies above the boundary layer and may experience free tropospheric conditions. This is also supported by the fact that PN>12 nm 715 concentrations measured with the UAS above Ny-Ålesund airport agrees well with the 716 modeled particle number concentrations, at the same altitude as Zeppelin. However, both the 717 model and UAS PN>12nm concentrations is a factor of 4 higher than the PN>12nm observation at 718 719 Zeppelin. Both the measured and modeled particle size distribution, at both stations, show a distinct 720 721 Hoppel minima, which can be explained by tin-cloud processing. Model sensitivity runs with 722 varying cloud supersaturation indicate that a cloud supersaturation of 0.5% or higher is required for the model to capture the observed Hoppel minima. Furthermore, model 723 724 sensitivity runs show that the Aitken mode particle number concentrations are dominated by contribution of secondary aerosols rather than primary emissions. The modeled PM₁₀ Cl⁻ and 725 Na⁺ is positively correlated when compared to PM₁₀ filter samples. The main driver for 726 secondary aerosol particle growth is the formation of MSA via aqueous phase ozonolysis of 727 728 the DMS oxidation product MSIA. This demonstrates the importance of multi-phase DMS chemistry in capturing the size resolved secondary aerosol growth in marine polar regions. 729 The sensitivity studies indicate that it is important to limit the uncertainties in parameters such 730 731 as cloud supersaturation and NH₃ emissions over open oceans to get a better constraint on secondary aerosol formation and its subsequent climatic effects. This work was a first attempt 732 to simulate new particle and secondary aerosol formation in marine polar regions using a 733 process based chemistry transport model that includes a comprehensive multi-phase DMS and 734 735 halogen chemistry mechanism, detailed gas-molecular cluster and aerosol dynamics. In future 736 studies, we aim to implement ADCHEM for extended studies in polar marine and remote 737 continental regions where different atmospheric constituents such as HIO₃, terpenes and amines drive secondary aerosol formation. 738

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- 754

755 Author Contributions

- 756 CX, PR, MBoy, BA and BW planned and designed the study. PR, RWdJ and CX developed
- 757 and setup the ADCHEM model. CX, MB and VV performed the FLEXPART model
- simulations. BA, BW, RoTh, and RT provided the measurement data. Resources were
- 759 provided by PR and MBoy. CX, PR and MBoy wrote the original draft, which included
- visualizations made by CX and PR. All other authors discussed the results and contributed to
- 761 the final manuscript.

762

763 Competing interests

The authors declare that they do have no conflict of interest.

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