

1 **Secondary aerosol formation in marine Arctic environments: A model measurement** 2 **comparison at Ny-Ålesund**

3 Carlton Xavier¹, Metin Baykara^{1,2}, Robin Wollesen de Jonge³, Barbara Altstädter⁴, Petri
4 Clusius¹, Ville Vakkari^{5,6}, Roseline Thakur¹, Lisa Beck¹, Silvia Becagli⁷, Mirko Severi⁷, Rita
5 Traversi⁷, Radovan Krejci^{8,9}, Peter Tunved^{8,9}, Mauro Mazzola¹⁰, Birgit Wehner¹¹, Mikko
6 Sipilä¹, Markku Kulmala¹, Michael Boy^{1,12}, Pontus Roldin³

7 ¹Institute for Atmospheric and Earth Systems Research, University of Helsinki, P.O. Box 64, 00014 Helsinki,
8 Finland

9 ²Climate and Marine Sciences Department, Eurasia Institute of Earth Sciences, Istanbul Technical University,
10 Maslak 34469, Istanbul, Turkey

11 ³Division of Nuclear Physics, Department of Physics, Lund University, P. O. Box 118 SE-221 00 Lund, Sweden

12 ⁴Institute of Flight Guidance, Technische Universität Braunschweig, 38108 Braunschweig, Germany

13 ⁵Atmospheric Chemistry Research Group, Chemical Resource Beneficiation, North-West University,
14 Potchefstroom, South Africa

15 ⁶Finnish Meteorological Institute, POBox 503, FI-00101 Helsinki, Finland

16 ⁷Department of Chemistry, University of Florence, Sesto Fiorentino, 50019 Florence, Italy

17 ⁸Department of Environmental Science, Stockholm University, Stockholm, Sweden

18 ⁹Bolin Centre for Climate Research, Stockholm University, Stockholm, Sweden

19 ¹⁰National Research Council of Italy, Institute of Polar Sciences (CNR-ISP)

20 ¹¹Institute of Tropospheric Research, 04318 Leipzig, Germany

21 ¹²LUT School of Engineering Science, Lappeenranta-Lahti University of Technology, P.O.Box 20, 53851
22 Lappeenranta, Finland

23

24 **Correspondence:** Carlton Xavier (carlton.xavier@helsinki.fi) and Pontus Roldin
25 (pontus.roldin@nuclear.lu.se)

26

27 **Abstract**

28 In this study, we modeled the aerosol particle formation along air mass trajectories arriving at
29 the remote Arctic research stations Gruebadet (67 m a.s.l) and Zeppelin (474 m a.s.l), Ny-
30 Ålesund during May 2018. The aim of this study was to improve our understanding of
31 processes governing secondary aerosol formation in remote Arctic marine environments. We
32 run the Lagrangian chemistry transport model ADCHEM, along air mass trajectories
33 generated with FLEXPART v10.4. The air masses arriving at Ny-Ålesund spent most of their
34 time over the open ice-free ocean. In order to capture the secondary aerosol formation from
35 the DMS emitted by phytoplankton from the ocean surface, we implemented a recently
36 developed comprehensive DMS and halogen multi-phase oxidation chemistry scheme,
37 coupled with the widely used Master Chemical Mechanism (MCM).

38 The modeled median particle number size distributions are in close agreement with the
39 observations in the marine influenced boundary layer at near sea surface Gruebadet site.
40 However, while the model reproduces the accumulation mode particle number concentrations

41 at Zeppelin, it overestimates the Aitken mode particle number concentrations by a factor of
42 ~ 5.5 . We attribute this to the deficiency of the model to capture the complex orographic
43 effects on the boundary layer dynamics at Ny-Ålesund. However, the model reproduces the
44 average vertical particle number concentration profiles within the boundary layer (0-600 m
45 a.s.l.) above Gruvebadet, as measured with Condensation Particle Counters (CPCs) on board
46 an Unmanned Aircraft Systems (UAS).

47 The model successfully reproduces the observed Hoppel minima, often seen in particle
48 number size distributions at Ny-Ålesund. The model also supports the previous experimental
49 findings that ion mediated $\text{H}_2\text{SO}_4\text{-NH}_3$ nucleation can explain the observed new particle
50 formation in the marine Arctic boundary layer in the vicinity of Ny-Ålesund. Precursors
51 resulting from gas and aqueous phase DMS chemistry contribute to the subsequent growth of
52 the secondary aerosols. The growth of particles is primarily driven via H_2SO_4 condensation
53 and formation of methane sulfonic acid (MSA) through the aqueous-phase ozonolysis of
54 methane sulfinic acid (MSIA) in cloud and deliquescent droplets.

55

56 **1. Introduction**

57 The Earth's radiation budget is influenced both directly and indirectly by aerosols, which
58 scatter and absorb the incoming short-wave radiation (direct effect) and serve as cloud
59 condensation nuclei (CCN, indirect effect), affecting both short and long-wave radiation
60 (Gantt et al., 2014; Oshima et al., 2020; Park et al., 2017; Scott et al., 2014). The Arctic
61 environments are susceptible to perturbations in the radiation balance, with some estimates
62 suggesting that, compared to the global average, the Arctic is warming at three times the rate,
63 a phenomenon termed as Arctic amplification (AMAP, 2011, 2017, 2021; Lenssen et al.,
64 2019; Tunved et al., 2013). The warming of the Arctic polar environment has accelerated sea
65 ice loss, leading to a rapid decline in the extent and duration of snow cover and increase in
66 permafrost thaw (AMAP, 2011, 2017; Bengtsson et al., 2013).

67 The Arctic aerosol number concentration shows a pronounced seasonal variation, where the
68 late winter and early spring period is characterized by elevated accumulation mode aerosol
69 concentrations, accompanied by trace gases (mostly anthropogenic with long-range
70 transported trace elements such as sulfates, soot, and Peroxy Acyl Nitrates (PANs)). This
71 annually recurring phenomenon in late winter and spring is termed the Arctic Haze (Barrie,
72 1986; Lupi et al., 2016; Tunved et al., 2013). This contrasts with the summer period, when the

73 atmospheric new particle formation is observed at Arctic sites, most likely due to low
74 background aerosol concentrations, increased photo-chemistry and biological activity
75 (Engvall et al., 2008; Heintzenberg et al., 2017; Tunved et al., 2013).

76 The climate change driven Arctic sea ice loss has a profound impact on natural aerosol
77 production. Arrigo and van Dijken, (2015) found that decreasing and thinning of sea ice
78 increased the rates of phytoplankton net primary production by ~20% between the years 1998
79 and 2009. This can lead to an increase in the emissions of primary biogenic precursors such as
80 dimethyl sulfide (DMS), nitrogen volatiles (e.g. alkyl-amines) (Dall'Osto et al., 2017a;
81 Dall'osto et al., 2017b) and biological iodine species (Cuevas et al., 2018). DMS is emitted
82 into the atmosphere via air-sea gas exchanges (Park et al., 2017; Uhlig et al., 2019), and
83 accounts for ~80% of global natural sulfur emissions (Kettle and Andreae, 2000; Uhlig et al.,
84 2019). Methane sulfonic acid (MSA) and sulphuric acid (H_2SO_4) is formed via DMS gas-
85 phase oxidation by OH and halogen species (Cl, Br) (Hoffmann et al., 2016; Kim et al., 2021;
86 Wollesen de Jonge et al., 2021). MSA and H_2SO_4 , together with ammonia (NH_3) or amines,
87 act as precursors contributing to new particle formation (NPF) and subsequently to CCN
88 production, influencing cloud formation and radiative balance (Berndt et al., 2020; Dall'Osto
89 et al., 2017b; Hoffmann et al., 2016; Kim et al., 2021, Jang et al., 2021; Park et al., 2021).
90 NH_3 plays a major role in particle formation through stabilization of sulfuric acid clusters
91 (Beck et al., 2021; Jokinen et al., 2018; Olenius et al., 2013). Depending on local parameters
92 such as ocean pH, salinity and temperature, global oceans can act either as a source or sink of
93 NH_3 (Paulot et al., 2015). Apart from participating in cluster formation, NH_3 influences the
94 pH of marine aerosols by neutralizing the acid (H_2SO_4 and MSA) in the particles (Paulot et
95 al., 2015). Though a few potential sources of NH_3 are known, for example coastal sea bird
96 colonies, pockets of open water and melting sea ice in summertime Arctic, the magnitude of
97 the emissions remain uncertain (Dall'osto et al., 2019a; Riddick et al., 2012; Wentworth et al.,
98 2016).

99 DMS oxidation chemistry has been under focus, but uncertainties in climate predictions
100 persist since the chemical transport models (CTMs) and global climate models (GCMs)
101 employ fixed MSA and SO_2 yields from gas-phase oxidation of DMS to calculate aerosol
102 formation (Hertel et al., 1994; Hoffmann et al., 2016; Kloster et al., 2006; Wollesen de Jonge
103 et al., 2021). Including a detailed multi-phase (aqueous-phase chemistry coupled with gas-
104 phase chemistry) DMS chemistry in numerical models can overcome these uncertainties
105 (Barnes et al., 2006; Campolongo et al., 1999). Reaction intermediates such as dimethyl
106 sulfoxide (DMSO), dimethyl sulfone ($DMSO_2$), methane sulfinic acid (MSIA) are water-

107 soluble, and experiments have shown that neglecting aqueous phase chemistry leads to either
108 an under-estimation of modeled MSA (Campolongo et al., 1999), or an over-estimation of
109 gaseous SO₂ compared to measured values (Hoffmann et al., 2016). For example, the
110 temperature dependent ratio of MSA/non-sea-salt SO₄²⁻ (nss-SO₄²⁻) is often used to estimate
111 the contribution of DMS to sulfate budget (Ayers et al., 1999; Barnes et al., 2006).
112 Campolongo et al., 1999 showed that modeling studies which included a multi-phase DMS
113 chemistry can bridge the gap between temperature-dependent observations and modeled
114 MSA/nss-SO₄²⁻. Incorporating reactive halogens species over marine environments is crucial
115 in determining the DMS oxidation pathways to either SO₂ or MSA, the aging of marine
116 aerosols and the radiative properties of marine clouds (Hoffmann et al., 2016). Modeling
117 studies have shown that Cl⁻ and BrO⁻ radicals in the gas phase act as important DMS sinks
118 (Chen et al., 2018; Wollesen de Jonge et al., 2021), further underlining the role of halogen-
119 DMS chemistry in the marine boundary layer.

120 Recent DMS+OH oxidation experiments performed in the AURA chamber at Aarhus
121 University show that MSA dominates the secondary aerosol mass formation (Rosati et al.,
122 2021). Aerosol dynamics model simulations which intended to replicate the observations
123 during these AURA experiments, using the DMS gas-phase chemistry scheme from the
124 Master Chemical Mechanism, MCMv3.3.1, (Jenkin et al., 1997, 2015; Saunders et al., 2003),
125 substantially underestimates the particle mass and number concentrations and the MSA:SO₄²⁻
126 ratio (Rosati et al., 2021, Wollesen de Jonge, 2021). Based on these findings, Wollesen de
127 Jonge et al. (2021) developed a new DMS multi-phase chemistry scheme based on MCM
128 v3.3.1, CAPRAM DMS module 1.0 (DM1.0) (Hoffmann et al., 2016), a subset of the multi-
129 phase halogen chemistry mechanism CAPRAM Halogen Module 2.0 (HM2.0) (Bräuer et al.,
130 2013) and new reactions leading to the formation of hydroperoxymethyl thioformate
131 (HPMTF). With the new DMS multi-phase chemistry mechanism, the aerosol dynamics
132 model could capture the observed particle number concentrations and secondary PM MSA
133 and SO₄²⁻ during DMS oxidation experiments performed at both dry and humid conditions at
134 0 °C and 20 °C in the AURA chamber. For more details on the DMS, halogen and multi-phase
135 chemistry scheme used in ADCHEM, the reader is referred to the article and supplement of
136 Wollesen de Jonge et al., (2021).

137 The aim of this work is to understand the processes and DMS oxidation products governing
138 the formation and growth of the secondary aerosol in pristine remote marine Arctic region. To
139 facilitate this, we have implemented the above mentioned DMS multi-phase chemistry
140 mechanism into ADCHEM (see Methods section) and modeled the aerosol formation along

141 air mass trajectories arriving at Ny-Ålesund. We compared the model results with
142 observations from Zeppelin (78°56' N, 11°53' E, 474 m a.s.l) and Gruvebadet (78°92' N,
143 11°90' E, 67 m a.s.l). These two sites represent remote marine Arctic conditions. Gruvebadet
144 represents ground-level concentrations as it is well within the boundary layer (BL). Zeppelin
145 on the other hand, is most often above the BL in winter months and sometimes below the BL
146 during spring and summer months (Traversi et al., 2020). This implies that Zeppelin is more
147 influenced by long range transport, and Gruvebadet by more local effects (Traversi et al.,
148 2020). This, demonstrates the complexity involved in capturing the atmospheric mixing and
149 secondary aerosol concentrations at Ny-Ålesund. The reason is that Svalbard has an
150 orographically complex terrain comprising of mountains, glaciers, fjords and flat lands that
151 introduce various micro-meteorological phenomena (Rader et al., 2021; Schemann and Ebell,
152 2020).

153

154 **2. Methods**

155 Using the combined multi-phase DMS chemistry mechanism by Wollesen de Jonge et al.,
156 (2021), MCMv3.3.1 and the monoterpene peroxy radical autoxidation mechanism (PRAM,
157 Roldin et al., 2019; Xavier et al., 2019) we simulated aerosol particle formation within the
158 marine boundary layer (MBL) upwind and at Ny-Ålesund between 1st - 25^h May 2018, using
159 the Aerosol Dynamics, gas and particle-phase CHEMistry and radiative transfer model
160 ADCHEM (Öström et al., 2017; Roldin et al., 2011, 2019). We ran ADCHEM as a
161 Lagrangian model along the air mass trajectories arriving at Zeppelin every 3 hours during the
162 selected period (in total 200 trajectory simulations). FLEXPART v10.4 was used to calculate
163 the air mass trajectories and potential emission sensitivity fields (Pisso et al., 2019; Stohl et
164 al., 2005). The simulation results for the vertical distribution of newly formed aerosol
165 (particle diameters < 12 nm) were validated against concurrent measurement data available
166 from the ALADINA (Application of Light-Weight Aircraft for Detecting in situ Aerosol)
167 campaign, wherein a UAS was used to investigate horizontal and vertical distribution of
168 aerosol profiles in the marine boundary layer (ABL) (Lampert et al., 2020). Additionally,
169 modeled particle number size distributions and PM₁₀ chemical compositions were compared
170 to the available measured particle number size distributions and PM₁₀ filter samples at both
171 Gruvebadet and Zeppelin measurement stations.

172 **2.1 Air mass trajectories and potential emission sensitivity fields**

173 We employed the Lagrangian particle dispersion model FLEXible PARTicle
174 (FLEXPARTv10.4) to assess the emission sensitivities or “footprints” of air-masses origin
175 arriving at Zeppelin during the simulation period. FLEXPART is a stochastic model used to
176 compute dispersion of hypothetical particles, based on mean, turbulent and diffusive flows
177 which can be run backwards in time to estimate air mass history at a site (Pisso et al., 2019).
178 European Center for Medium-Range Weather Forecasts (ECMWF) ERA5 reanalysis
179 meteorology with 137 height levels, 1-hour temporal and 0.5° x 0.5° spatial resolution, was
180 used as an input to FLEXPART (*ERA5 hourly data on single levels from 1979 to present*.
181 *Copernicus Climate Change Service (C3S) Climate Data Store (CDS)*. last access 30th April
182 2021, 10.24381/cds.adbb2d47, *ERA5 hourly data on pressure levels from 1979 to present*.
183 *Copernicus Climate Change Service (C3S) Climate Data Store (CDS)*. last access 30th April
184 2021, 10.24381/cds.bd0915c6). The air-mass history was simulated 7-day backwards in time
185 and arriving at Zeppelin (474 m a.s.l) every 3 hours (at 00:00, 03:00, 06:00, 09:00, 12:00,
186 15:00, 15:00 and 21:00 UTC) for the entire simulation period (1st - 25th May 2018).

187 FLEXPART calculated normalized emission sensitivity fields were combined with oceanic
188 emissions (DMS, dibromomethane, tribromomethane, iodomethane), NH₃ from sea-bird
189 colonies and anthropogenic emissions (NH₃, SO₂, CO, NO_x) derived from global inventories
190 (see section 2.2). This was done to obtain representative emissions that consider the complete
191 emission source regions along the trajectories, upwind of the measurement station.
192 Additional meteorological parameters such as temperature, pressure, sea surface temperature,
193 specific humidity and cloud liquid water content from ERA5 reanalysis dataset were extracted
194 along the trajectories and provided as inputs to ADCHEM.

195

196 **2.2 Gas and primary particle emissions**

197 Emissions of gas-phase biogenic volatile organic compounds (VOCs) α -pinene, β -pinene Δ 3-
198 carene, limonene, isoprene and β -caryophyllene were modeled with a 1 - dimensional version
199 of MEGAN v2.04 (Model of Emissions of Gases and Aerosols from Nature 2.04) (Guenther
200 et al., 2006). Gas-phase emissions of marine halogens such as tribromomethane (CHBr₃),
201 dibromomethane (CH₂Br₂), iodomethane (CH₃I) were retrieved from CAMS-OCE Global
202 oceanic emissions (CAMS-GLOB-OCE) which are available as daily means with a spatial
203 resolution of 0.5°x0.5° (Granier et al., 2019; Ziska et al., 2013). CAMS-GLOB-OCE also
204 provides gas-phase DMS emissions with the same temporal and spatial resolution (Granier et
205 al., 2019) calculated with the air-sea flux parameterization and emission fluxes described in

206 (Lana et al., 2011; Nightingale et al., 2000). NH_3 emissions from seabird colonies were
207 acquired from a global emission inventory (Riddick et al., 2012). To account for additional
208 NH_3 fluxes from the open ocean, we used an estimated sea surface equilibrium $\text{NH}_{3(g)}$
209 saturation concentration of 0.5 nmol/m^3 (12.2 ppt at standard temperature and pressure (STP))
210 which approximately correspond to a surface ocean ammonium concentration of 0.125
211 mmol/m^3 (or $\sim 3\text{ppb}$, calculated based on equation 3 & 4 from Wentworth et al., 2016) at a sea
212 surface temperature of $+2 \text{ }^\circ\text{C}$. The sea surface temperature for the study period varied between
213 $-2 \text{ }^\circ\text{C}$ - $23 \text{ }^\circ\text{C}$ along the trajectories. The estimated surface ocean ammonium concentrations is in
214 close agreement with the concentration estimated by the global ocean biogeochemical model
215 COBALT (Stock et al., 2014) in the North Atlantic ocean, but up to a factor of ~ 5 higher than
216 the concentrations simulated with other ocean biogeochemical models and/or model setups
217 (Paulot et al., 2015). Therefore, we performed model sensitivity runs with a sea surface
218 equilibrium $\text{NH}_{3(g)}$ concentration of 0.1 and 1 nmol/m^3 . The $\text{NH}_{3(g)}$ equilibrium saturation
219 concentrations represent the ambient surface gas-phase concentration at which the air-sea flux
220 changes direction, with a net downward flux from air to sea if the ambient $\text{NH}_{3(g)}$ exceeds the
221 equilibrium gas concentrations and vice versa (Wentworth et al., 2016). For the anthropogenic
222 trace gas and primary particle emissions, we used the CAMS-GLOB-ANT v2.1 inventory,
223 with a spatial resolution of $0.1^\circ \times 0.1^\circ$ (Granier et al., 2019).

224 In this work, we used the sea surface temperature (SST) and wind speed dependent sea-spray
225 aerosol (SSA) emission parameterization by Sofiev et al. (2011), (further referred to as
226 Sofiev11). Sofiev11 used a modified source function based on the parameterization of
227 Monahan et al. (1986) and experiments by Mårtensson et al. (2003) and SEAS campaign by
228 Clarke et al. (2006). The modified source function in Sofiev11 provides extrapolated SSA
229 emissions between size ranges of 10 nm - $10 \text{ }\mu\text{m}$, with appropriate correction functions
230 employed for SST deviating from 298.15 K (Sofiev et al., 2011). Sofiev11 SSA
231 parameterization shows that with increasing temperatures, emission flux for larger particles
232 increases while the emission fluxes for smaller particles decreases (Barthel et al., 2019;
233 Sofiev et al., 2011). We performed sensitivity tests using the temperature and wind speed
234 dependent SSA parameterization by Salter et al., (2015), (further referred to as Salter15).
235 Both the Salter15 and Sofiev11 are valid between 10 nm - $10 \text{ }\mu\text{m}$. Model simulation
236 comparisons between Sofiev11 and Salter15 have shown that the SSA parameterization from
237 Sofiev11 has a stronger temperature dependence and higher particle number concentration
238 emissions in the Aitken mode but result in lower PM_{10} emissions at temperatures below $25 \text{ }^\circ\text{C}$
239 (Barthel et al., 2019).

240

241 **2.2 ADCHEM**

242 For this study, ADCHEM was employed as a 1 - dimensional column model with 40
243 logarithmically vertical layers, extending up to ~2600m. The model time step used for
244 simulations was 30 seconds. The vertical atmospheric turbulent diffusion was solved using a
245 modified Grisogono turbulent diffusivity scheme (Jeričević et al., 2010; Öström et al., 2017;
246 Roldin et al., 2019). The ADCHEM aerosol module includes new particle formation,
247 Brownian coagulation, condensation and evaporation of particles, and finally the dry and wet
248 deposition of both particles and gases. The particle number size distributions were represented
249 using 100 size bins ranging from 1.07 nm to 10 μm dry diameter. Clouds were assumed to be
250 present in the model grid cells when the bulk liquid water content (LWC, extracted along the
251 trajectory from ERA5 datasets) was greater than 0.01 g m^{-3} . As a default, we used a constant
252 cloud supersaturation (S) of 0.25% and the particles were activated into cloud droplets, if the
253 calculated water vapor supersaturation above the particle surface (S_c , calculated using Köhler
254 theory) was smaller than S . For sulfate dominated aerosol particles this corresponds to a
255 minimum dry particle activation diameter of ~80 nm (see Figure S10). During the cloud
256 processing, each activated cloud droplet was assumed to take up an equal amount of liquid
257 water corresponding to the total bulk LWC divided by the calculated number concentration of
258 activated cloud droplets. The gas-liquid droplet mass transfer and dissolution of 50 species in
259 total, including HCl, HNO₃, H₂SO₄, NH₃, HIO₃, H₂O₂, O₃, OH, BrO, NO₃, DMSO, MSIA,
260 MSA and HPMTF and their irreversible reactions in the interstitial and activated cloud
261 droplets are treated by the multi-phase chemistry mechanism (see Wollesen de Jonge et al.
262 (2021) for details). The kinetic pre-processor (KPP) (Damian et al., 2002) was used to
263 generate the multi-phase chemistry mechanism used in this study.

264 Recent observations of NPF at Ny-Ålesund have confirmed the importance of ion-mediated
265 H₂SO₄-NH₃ nucleation in spring with MSA and H₂SO₄ condensation contributing to the
266 subsequent growth of particles (Beck et al., 2021; Lee et al., 2020). In this work, the
267 Atmosphere Cluster Dynamics Code (ACDC) (McGrath et al., 2012; Olenius et al., 2013) was
268 coupled with ADCHEM (Roldin et al., 2019). ACDC was used to model NPF, which
269 involved H₂SO₄ clustering with NH₃ via both neutral and ion-induced pathways with an
270 ionization rate of 1.7 $\text{cm}^{-3}\text{s}^{-1}$. ACDC was used to solve the evolution of molecular H₂SO₄-NH₃
271 clusters by considering the loss of clusters by collisions, evaporation or coagulation
272 scavenging onto larger aerosol particles. At each time step, the flux of clusters (up to ~ 5

273 H₂SO₄ and 5 NH₃ each) growing out of the ACDC molecule-cluster domain represents the
 274 NPF rate. These newly formed clusters are assigned to the corresponding smallest particle
 275 size bin at 1.07 nm in diameter in ADCHEM, which then simulates the condensational growth
 276 of particles and losses due to evaporation, coagulation, and wet and dry deposition.

277 For all simulations, we used model output from the closest height levels which can represent
 278 Gruvebadet (model height of 73.5 m a.s.l) and Zeppelin (model height of 486.0 m a.s.l).

279 Sensitivity Tests

280 Alongside the main ADCHEM simulations, *BaseCase*, we performed nine complementary
 281 scenario runs to assess the impact of different processes on the modeled aerosol
 282 concentrations. We performed simulations without aerosol in-cloud processing (*Cloudoff*), to
 283 check the impact of in-cloud processing on the growth of aerosols. We investigated the effect
 284 of higher PM₁₀ particle emissions on the chemical composition of secondary aerosols, using
 285 the sea-spray emission parameterization based on Salter et al., 2015 (*SalterSSA*). Simulations
 286 were conducted to assess the impact of lower and higher ammonia sources over the open
 287 ocean (*LowNH₃*, *HighNH₃*). A sensitivity test without precipitation (*NoPrecip*) was performed
 288 to test the influence of precipitation on number concentration and particle composition. Since
 289 cloud supersaturation is critical to the activation of particles and is highly uncertain, we
 290 performed two simulations with low and high cloud supersaturation ($S=0.1\%$, $SSat=0.1$ and
 291 $S=0.4\%$, $Ssat=0.4$) to test its impact on the modelled particle distributions. This corresponds
 292 to minimum dry particle activation diameters of ~150 nm and ~60 nm respectively, for sulfate
 293 rich aerosol particles (see Figure S10). We performed a simulation without new particle
 294 formation (*NPFoff*), and finally one simulation without the dissolution and irreversible
 295 aqueous chemistry of the intermediate DMS oxidation products, SO₂ and halogens
 296 (*woDissolution*), implying that MSA, H₂SO₄ and HIO₃ is only formed in the gas-phase. Table
 297 1. summarizes the setup for different model sensitivity test.

298

299 **Table 1.** Model sensitivity tests performed alongside the main *BaseCase* simulations to test
 300 the effect of different parameters on secondary aerosol formation. These sensitivity tests focus
 301 on the role of in-cloud processing and aqueous phase chemistry, the NH₃ emissions from open
 302 ocean, SSA parameterization and cloud supersaturation. The sea surface equilibrium NH_{3(g)}
 303 concentrations in ppt are provided in the brackets.

Simulation	In-cloud	NH _{3(eq)} (nmol/m ³ ,	SSA	Precipitation
------------	----------	--	-----	---------------

	Processing	ppt)	parameterization	
<i>BaseCase</i>	On	0.5 (12.2)	Sofiev11	On
<i>SalterSSA</i>	On	0.5 (12.2)	Salter15	On
<i>Cloudoff</i>	On	0.5 (12.2)	Sofiev11	On
<i>LowNH₃, HighNH₃</i>	On	0.1 (2.4) 1.0 (24)	Sofiev11	On
<i>NoPrecip</i>	On	0.5 (12.2)	Sofiev11	Off
<i>SSat0.4, SSat0.1</i>	On	0.5 (12.2)	Sofiev11	On
<i>NPFoff</i>	On	0.5 (12.2)	Sofiev11	On
<i>WoDissolution</i>	On, but no dissolution and irreversible chemistry of intermediate DMS oxidation products	0.5 (12.2)	Sofiev11	On

304

305 **2.3 Measurements**

306 We utilized comprehensive measurements from the Ny-Ålesund Research station sites,
307 Zeppelin observatory (Platt et al., 2022) and Gruvebadet during the period of 1st - 25th May
308 2018. Since 2017, the atmospheric observatory at Gruvebadet, which is located about 700 m
309 southwest of Ny-Ålesund village at almost sea level (67 m.s.l), hosted Neutral cluster and Air
310 Ion Spectrometer (NAIS, Manninen et al., 2010; Mirme and Mirme, 2013) for semi-
311 permanent measurements. Here we use NAIS measured number size distribution of naturally
312 charged (ions) in diameter size ranges between 0.8 nm-40 nm and neutral particles in the size
313 range of 2.5 nm-42 nm, with a temporal resolution of two seconds.

314 During the measurement period, a scanning mobility particle sizer (SMPS), was operated to
315 measure particle number size distribution in the diameter size range of 10 - 470 nm at
316 Zeppelin. Concurrent SMPS data (TSI 3034, 54 channels) with diameter size ranging from 10
317 to 470 nm from Gruvebadet were also available (Dall'osto et al., 2019b; Moroni et al., 2020),
318 thus, enabling us to compare the modeled particle number size distribution with the measured
319 size distributions at both measurement stations. Daily resolution continuous aerosol samples
320 with PM₁₀ cutoff were collected at Gruvebadet using a Tecore Skypost low-volume sampler
321 (Amore et al., 2022). The detection limit for Na⁺ was 0.0001 µg m⁻³ and 0.0002 µg m⁻³ for Cl⁻,

322 NH_4^+ and SO_4^{2-} . Since the field blank medians at Gruvebadet were less than 1 percentile of
 323 sampled values, the field blanks were not subtracted from the sampled values (Amore et al.,
 324 2022).

325 Vertical particle number concentration profiles were obtained using UAS ALADINA (Bärfuss
 326 et al., 2018; Lampert et al., 2020), which was operated during the simulation period.
 327 ALADINA were operated up to a height of 850 m a.s.l., thus can be used for a potential
 328 closure between the two different research sites of Gruvebadet and Zeppelin. ALADINA is
 329 equipped with two condensation particle counters (CPCs Model 3007, TSI Inc., St. Paul,
 330 MN, USA), measuring in the size ranges of 3 nm - 2 μm (CPC1) and ~12 nm - 2 μm (CPC2)
 331 (Lampert et al., 2020; Petäjä et al., 2020). The difference between CPC1 and CPC2 provides
 332 an estimate of particle number concentrations in the size of 3 - 12 nm (N_{3-12}), which was used
 333 as an indicator of NPF. Alongside the CPCs, a host of other instruments measuring
 334 meteorological parameters were operated in unison, the description of which can be found in
 335 Bärfuss et al., (2018) and Lampert et al.,(2020).

336 **Evaluating temporal aspects of model performance**

337 The modeled PM_{10} inorganic chemical composition was evaluated against the measured PM_{10}
 338 inorganic chemical composition using statistical estimates such as, normalized mean bias
 339 (NMB), Pearson correlation coefficient (r), root mean squared error (RMSE) and fraction of
 340 predictions within a factor of 2 of the observed values (FAC2). These tests were used to
 341 evaluate modeled values (M_i) against observation values (O_i) at both the measurement sites.

342 Pearson correlation coefficient was calculated using the formula:

$$343 \quad r = \frac{1}{n} \sum_{i=1}^n \frac{(O_i - \bar{O})}{\sigma_O} \frac{(M_i - \bar{M})}{\sigma_M} \quad (\text{Eq 1})$$

345 Where σ_o and σ_M are standard deviations of the observed and modeled values, respectively.

346 Normalized mean bias (NMB) indicates if the predictions are over or underestimating the
 347 observed values, with the factor representing the under or over estimation. NMB was
 348 calculated using Eq. 2:

$$349 \quad NMB = \frac{\sum_{i=1}^n (M_i - O_i)}{\sum_{i=1}^n O_i} \quad (\text{Eq 2})$$

350

351 Root mean squared error (RMSE) was calculated using Eq. 3:

$$352 \quad RMSE = \sqrt{\sum_{i=1}^n \frac{(M_i - O_i)^2}{n}} \quad (Eq\ 3)$$

353

354 FAC2 is a robust metric defined as the percentage of predictions which are within a factor of
355 2 of the observed values (Eq. 4):

$$356 \quad Fac2 = 0.5 \leq \frac{M}{O} \leq 2.0 \quad (Eq\ 4)$$

357 **3. Results and Discussion**

358 In the following sections, we analyze and evaluate the model results against comprehensive
359 measurements in Ny-Ålesund. In sub-section 3.1, we focus on the particle number size
360 distributions at both sites, followed by gas-phase concentrations and PM₁₀ inorganic chemical
361 composition (sub-section 3.2) and the vertical nano-particle concentration profiles (sub-
362 section 3.3). Finally, in sub-section 3.4, we analyze the results from the model sensitivity
363 tests.

364

365 **3.1 Particle number size distributions**

366 Figure 1(a) and (b) show the observed and predicted particle number size distributions at
367 Gruvebadet for the *BaseCase* simulation. Figure 1(a) includes SMPS observations starting
368 from 10 to 470 nm and NAIS observations for neutral particles in the range 2.5 nm-10 nm
369 (boundary marked by the black line) since NAIS data below 2.5 nm cannot be relied upon,
370 owing to the presence of corona generated ions (Jayaratne et al., 2017; Manninen et al., 2011,
371 2016).

372 In the *BaseCase* simulation the model captures particle formation on 2nd May followed by an
373 increasing number of Aitken and accumulation mode particles during the days of 3rd - 4th
374 May, which is the result of more polluted air masses arriving at Ny-Ålesund from the
375 European continent (Figure S1). Similarly, the model reproduces the particle formation on the
376 20th of May, specifically in the size ranges 2-8 nm, but overestimates the Aitken mode and
377 accumulation mode particle concentration on the 21st of May. However, the model tends to
378 underestimate the nucleation mode particle number concentrations between 10-25 nm (N_{10-25}
379 nm) around noon, and overestimate the concentrations during the morning and evening (Figure

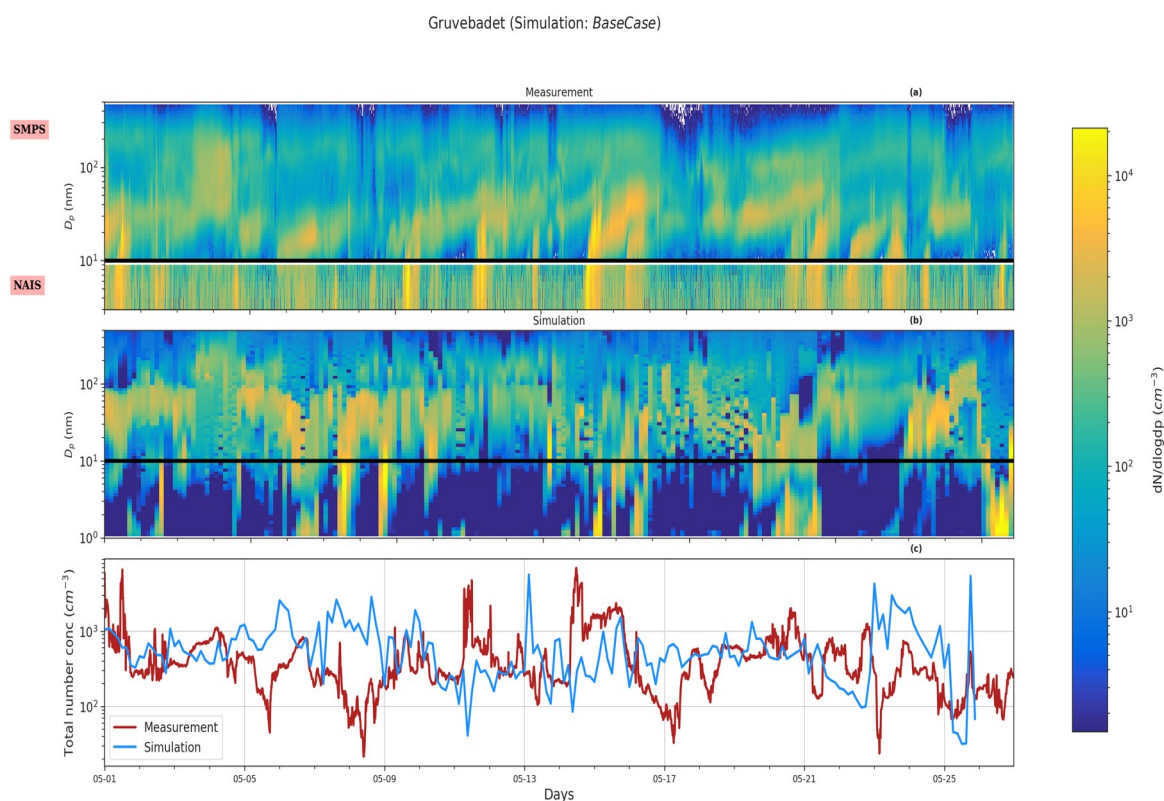
380 S2a). The model and measurements show an apparent time delay in the formation of new
381 particles larger than 10 nm. While the measurements show a peak at 11 am the simulated N_{10-}
382 $_{25 \text{ nm}}$ shows a maximum at 3 am and 6 pm. The modeled $N_{10-25 \text{ nm}}$ maximum around 6 pm is
383 likely a result of the formation of new particles around noon, which grow to >10 nm in
384 diameter during the afternoon and evening by condensation of H_2SO_4 . The predicted Aitken
385 ($N_{25-100 \text{ nm}}$) and accumulation mode particle concentrations ($N_{>100 \text{ nm}}$) which form few days
386 upwind of the station are overall, in good agreement with the measurements, which show a
387 minor diurnal trend (Figure 2b-c). The measurements indicate that at Gruebadet, $N_{10-25 \text{ nm}}$
388 contributes the most significant fraction of measured total number concentrations with 45.3%,
389 while $N_{25-100 \text{ nm}}$ and $N_{100-470 \text{ nm}}$ contribute 30.5% and 23.94% respectively. However, the
390 simulations predict greater contribution of Aitken mode (~53.85%) to total number
391 concentration, with the $N_{10-25 \text{ nm}}$ and $N_{100-470 \text{ nm}}$ accounting for ~ 36.58% and 9.57%
392 respectively (Figure S2).

393 Figure 2 shows the measured size distribution in panel (a) and simulated size distribution in
394 panel (b) for Zeppelin. At Zeppelin, the model overestimates the number concentration in
395 nucleation and Aitken modes (also cf. Figure S3, supplementary). The particle number size
396 distribution measurements at Zeppelin indicate that the relative contribution of the three
397 modes (nucleation, Aitken and accumulation) varies to some extent when compared to
398 Gruebadet. Measurements show that at Zeppelin, $N_{10-25 \text{ nm}}$ contributes ~33.46%, $N_{25-100 \text{ nm}}$
399 46.43% and $N_{>100 \text{ nm}}$ 20.11% to the total particle number concentrations. The model predicts
400 lower relative contribution of $N_{10-25 \text{ nm}}$ (26.94%), and a greater contribution of $N_{25-100 \text{ nm}}$
401 (63.44%) to the total simulated particle number concentrations. The diurnal trends at Zeppelin
402 agree well with earlier measurements conducted at Zeppelin in spring by Ström et al. (2009).
403 Additionally, the measured diurnal pattern at Zeppelin varies in comparison to Gruebadet. At
404 Zeppelin, the $N_{10-25 \text{ nm}}$ concentrations peak in the afternoon and evening. The modeled

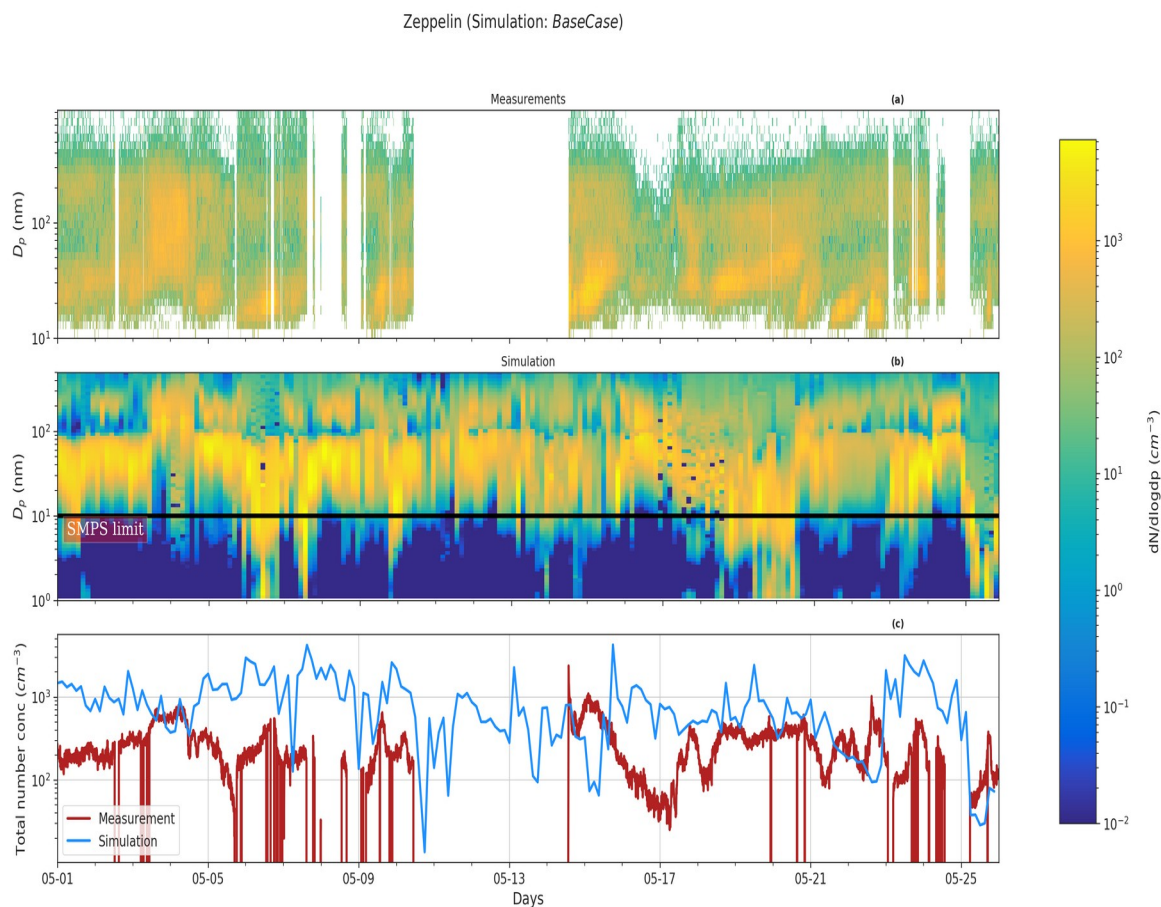
405 $N_{10-25 \text{ nm}}$ shows only a weak diurnal trend. It should be noted that the measurements show a
406 time delay of around 3 hours in the peak $N_{10-25 \text{ nm}}$ concentrations at the two sites (Figure S2
407 and S3). This is possibly be a result of vertical mixing and dilution effects modulating the
408 observed particle number concentrations at sites situated at different altitudes, similar to
409 observation made at Zeppelin and Corbel by Ström et al. (2009).

410 ADCHEM considers the formation of new particles via both the ion-mediated and neutral
411 H_2SO_4 - NH_3 clustering pathways. Beck et al (2021) observed dominant contribution of
412 negatively charged H_2SO_4 - NH_3 clusters to secondary particle formation in May 2017 at Ny-

413 Ålesund, with HIO_3 playing a small role in the initial particle formation. However, the
 414 discrepancy in the modeled and observed diurnal trends of $N_{10-25\text{ nm}}$ could indicate that there
 415 are other sources or vapors that might potentially contribute to the particle formation. Other
 416 possible NPF mechanism may involve amines (Olenius et al., 2013) and pure biogenic highly
 417 oxidized molecule (HOM) (neutral and ion induced) nucleation (Kirkby et al., 2016). We
 418 speculate that the exclusion of these other mechanisms (HIO_3 , H_2SO_4 -amines and HOM
 419 driven particle formation) might result in the discrepancies in the modeled and observed
 420 particle number concentration diurnal trends. HIO_3 induced particle formation could, e.g. play
 421 an important role if the air masses upwind of Ny-Ålesund traverse over the sea-ice covered
 422 regions (Baccarini et al., 2020; Beck et al., 2021).



423 **Figure 1.** Particle number size distribution at Gruvebadet for *BaseCase*. The panel (a) shows
 424 the measurement data for the period 1-25th May from SMPS (10 nm-470 nm) and NAIS (2.5 -
 425 10 nm), the panel (b) provides the modeled particle size distribution and panel (c) shows the
 426 total measured and simulated number concentrations. The black line at 10 nm denotes the
 427 boundary above which SMPS data starts and NAIS data ends. The abscissa indicates the time
 428 for the entire simulated duration. The ordinate in Figure 1 for both panels (a) and (b) indicates
 429 the particle diameter (D_p , nm).

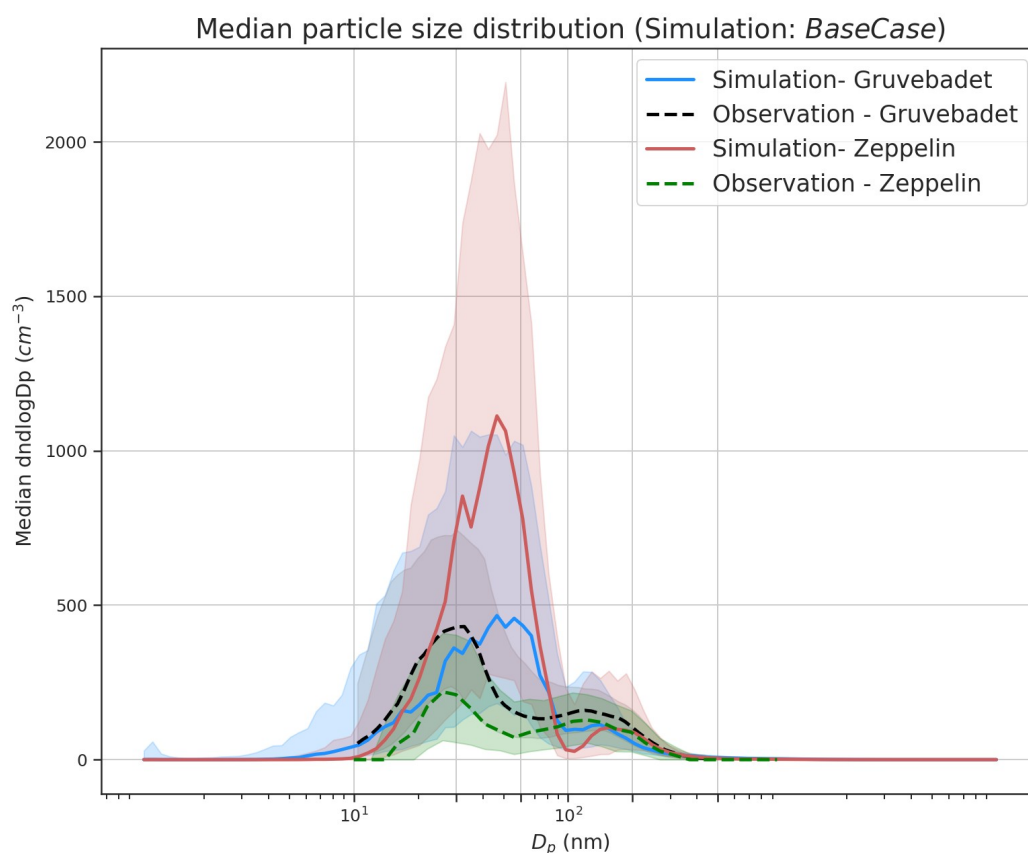


431 **Figure 2.** Particle size distribution at Zeppelin. The panel (a) shows the measurement data for
 432 the period 1-25th May from SMPS, the panel (b) provides the simulated particle size
 433 distribution and panel (c) shows the total measured and simulated number concentrations for
 434 the *BaseCase* simulations. The abscissa and ordinates are similar to Figure 1.

435

436 Figure 3 presents the median particle number size distribution for the *BaseCase* simulation at
 437 both Zeppelin and Gruvebadet, with respective 25th and 75th percentiles, for the entire selected
 438 period. At Gruvebadet, the modeled and measured median particle number size distributions
 439 are in reasonable agreement for both Aitken and accumulation mode. However, the model
 440 over predicts the median Aitken mode concentrations at Zeppelin by a factor ~ 5.5 . The
 441 modeled Aitken mode peak at both measurement sites is ~ 50 nm, while the measured Aitken
 442 mode peak is ~ 30 nm. Though the modeled accumulation mode peak is at a larger size (~ 150
 443 nm), compared to the measured accumulation mode peak (~ 110 nm), the predicted value is
 444 slightly lower than the monthly averaged accumulation mode peak location measured at
 445 Zeppelin in earlier studies (~ 160 - 170 nm, Dall'Osto et al., 2019).

446 The discrepancy between the modeled and measured particle concentrations at Zeppelin can
 447 be caused by the underlying complexity of modeling the boundary layer dynamics at an
 448 elevated site, such as Zeppelin. The vertical mixing of aerosols along the up-slope or down-
 449 slope of a mountain site is difficult, if not impossible for a 1- dimensional column model,
 450 since it is unable to capture the topographical influence on locally varying wind speeds or
 451 latent and sensible heat fluxes (Mikkola, 2020; Wainwright et al., 2012).



452 **Figure 3.** Median particle number size distribution at Gruvebadet and Zeppelin for both
 453 modeled (*BaseCase* simulations) and measured values. The shaded areas indicate the 25th and
 454 75th percentiles for both model and measured median particle number size distribution. At
 455 Zeppelin, the simulated median size distribution is calculated for periods only when SMPS
 456 data were available.

457

458 Another detectable feature in the median particle number size distribution is the diameter of
 459 the Hoppel minimum (Hoppel et al., 1985, 1986), and the role of in-cloud processing in
 460 forming this minimum. A Hoppel minimum is often observed in marine air masses (Fossum

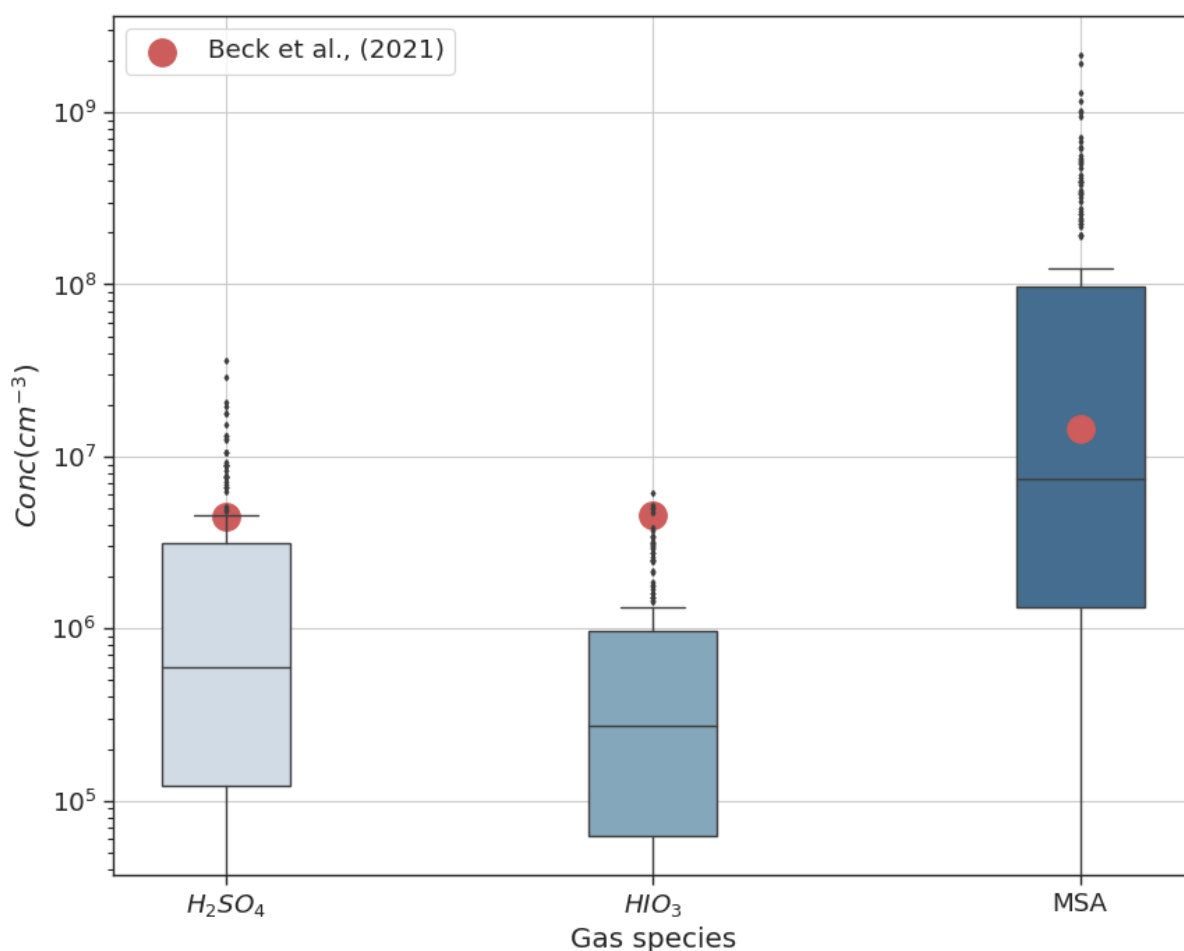
461 et al., 2018; Tunved et al., 2013; Zheng et al., 2018) and is attributed to in-cloud processing of
462 aerosols, with chemical processing (e.g., sulfate production via oxidation of dissolved SO₂)
463 (Feingold and Kreidenweis, 2000; Hoppel et al., 1986), and coalescence of droplets playing a
464 key role (Flossmann and Wobrock, 2019; Hoppel et al., 1986; Hoppel and Frick, 1990; Noble
465 and Hudson, 2013). It has been estimated that, on average, aerosols take part in about 10 non-
466 precipitating cloud cycles before it is removed from the atmosphere by wet scavenging
467 (Hoose et al., 2008; Hoppel et al., 1986; Rosenfeld et al., 2014). These non-precipitating
468 cloud cycles facilitate the formation of hygroscopic accumulation mode particles, with low
469 critical supersaturation (S_c) that readily activates to cloud droplets during subsequent cloud
470 cycles, thus growing to larger sizes. This is because the activated particles undergo chemical
471 processing, gas-to-particle conversions, coalescence and coagulation with other interstitial
472 particles. Upon evaporation of water, the emerging dry particles have a larger size and lower
473 S_c , leading to a minimum being formed between the un-activated and activated cloud droplets
474 (Herenz et al., 2018; Hudson et al., 2015; Noble and Hudson, 2013). The diameter at which
475 the Hoppel minimum is observed varies depending on the cloud supersaturation and particle
476 composition (Hoppel et al., 1986; Hudson et al., 2015), with Hoppel minima sizes observed in
477 ranges from 60 nm at Zeppelin Ny-Ålesund to around 90 nm at Tuktoyatuk, Canada (Herenz
478 et al., 2018; Tunved et al., 2013).

479 The median particle number size distribution in Figure 3 shows that at both stations, the
480 measured Hoppel minima is around ~ 60 nm, while the simulated Hoppel minima are around
481 the size of ~ 100 nm at both sites. This difference in location of Hoppel minima can be
482 attributed to the assumed value of $S=0.25\%$ in the model. The value of S used in the model
483 lies in the range of typical marine stratocumulus clouds, which can vary between 0.1 - 1%
484 (Fossum et al., 2018; Quinn et al., 2017). With $S=0.25\%$ the sulfate dominating particles in
485 the Arctic marine boundary layer will be activated into cloud droplets if their diameter is
486 greater than ~85 nm in diameter (Fig. S10).

487 **3.2 Gas and particle-phase chemical composition of important precursors**

488 Figure 4 shows the range of simulated gas-phase concentrations of DMS oxidation products
489 H₂SO₄, MSA and HIO₃ for the entire period at height levels representing Gruvebadet. The
490 mean measurement values (red dots) represent gas-phase concentrations for the same species
491 from an earlier 2017 May campaign performed at Gruvebadet by Beck et al., (2021).
492 Measurements of H₂SO₄ at Gruvebadet from May 2017 indicate monthly mean concentrations
493 around ~10⁶ # cm⁻³ (Beck et al., 2021). The modeled H₂SO₄ concentrations at Gruvebadet are

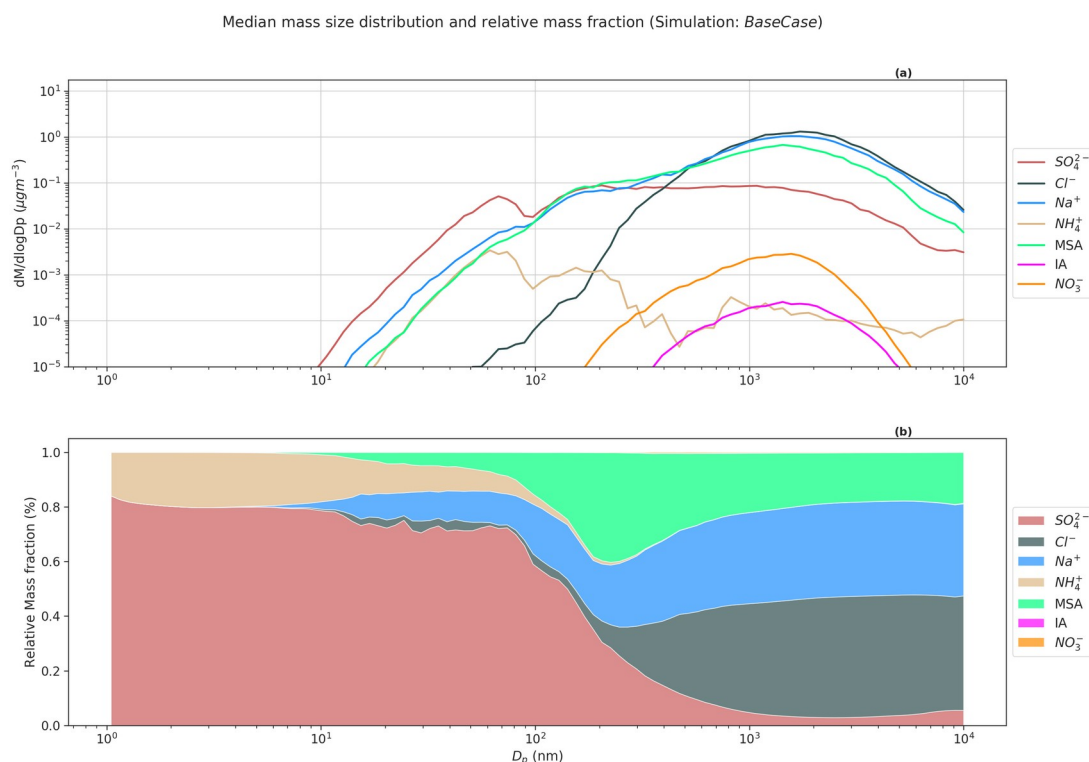
494 $3 \times 10^6 \text{ # cm}^{-3}$, implying a reasonably good model performance in predicting gaseous
 495 precursor concentrations. The simulated gas concentrations of MSA (10^5 - 10^8 # cm^{-3}) also
 496 agrees well with the measurements made at Gruebadet in May 2017 by (Beck et al., 2021),
 497 wherein they measured daily averages of MSA gas concentrations in the order of 10^7 # cm^{-3} .
 498 The low modeled values of MSA and DMSO gas phase concentrations at the height
 499 representing Zeppelin (c.f. Figure S4 supplementary e.g. between 15/05 - 17/05) coincide
 500 with the period where the planetary boundary layer height (PBLH) is below the altitude of
 501 Zeppelin station (cf. Figure S5 supplementary). Overall, we can conclude that the modeled
 502 precursor gas concentrations at the two measurement sites are, in general, good agreement
 503 with earlier measurements at the two sites.



504 **Figure 4.** Gas-phase concentrations at Gruebadet for the *BaseCase* simulations. The red dots
 505 indicate the mean measured values from an earlier 2017 May campaign conducted at
 506 Gruebadet by Beck et al., (2021).

507 Figure 5 (a) shows the simulated median mass size distribution of compounds Cl^- , Na^+ , MSA,
 508 SO_4^{2-} , NH_4^+ , and NO_3^- for the *BaseCase* runs in the lowest model layer. Figure 5(a) indicates
 509 that the nucleation mode particles are composed mainly of SO_4^{2-} and NH_4^+ , while MSA, Cl^-

510 and Na^+ dominate PM for larger particles. The observed and modeled high $\text{MSA}_{(\text{g})}$
511 concentrations in comparison to $\text{H}_2\text{SO}_{4(\text{g})}$ at Ny-Ålesund is not reflected in the respective
512 vapor contribution to the nano-particle growth. This is because, in contrast to H_2SO_4 , MSA is
513 not a non-volatile condensable compound. The gas-to-particle partitioning of MSA requires
514 co-condensation and dissolution of (NH_3) (Hodshire et al., 2019) or the existence of cations
515 such as Na^+ which decreases the particle acidity ($[\text{H}^+]$). Figure 5(b) shows the relative mass
516 fraction of the above-mentioned compounds to PM at different sizes. SO_4^{2-} and NH_4^+
517 dominate the mass for particles in the nucleation and Aitken mode. SO_4^{2-} contributes ~74%
518 and ~71% to nucleation and Aitken mode PM, with its contribution decreasing for
519 accumulation (100 nm-1 μm) and coarse ($>1 \mu\text{m}$) mode PM (~6% and 3.36% respectively)
520 (Table S1). NH_4^+ contribution follows a similar trend, as SO_4^{2-} , with 12.34% and 6.95%
521 contribution to nucleation and Aitken mode PM, but insignificant for accumulation and coarse
522 mode PM (Table S1). The loss of primary sea spray aerosols due to wet scavenging promoted
523 the growth of secondary aerosol particles in the nucleation and Aitken mode by NH_4^+ and
524 SO_4^{2-} as seen in Figure 5 (b). Na^+ (~32.9%), Cl^- (~39.5%) and MSA (20.45%) are the
525 dominant contributors to accumulation and coarse mode PM. In the *BaseCase* simulations,
526 gas-phase SO_2 dissolves in the cloud droplets, and is oxidized by H_2O_2 into SO_4^{2-} (Wollesen
527 de Jonge et al., 2021). Previous modeling studies have shown that a very small fraction of
528 MSA is formed in the gas phase. Instead, most MSA is formed via ozonolysis of MSIA in the
529 aqueous phase (Hoffmann et al., 2016; Wollesen de Jonge et al., 2021). It should be noted that
530 HIO_3 and NO_3^- have an insignificant contribution to total PM_{10} , amounting to ~0.05% and
531 0.17% respectively.

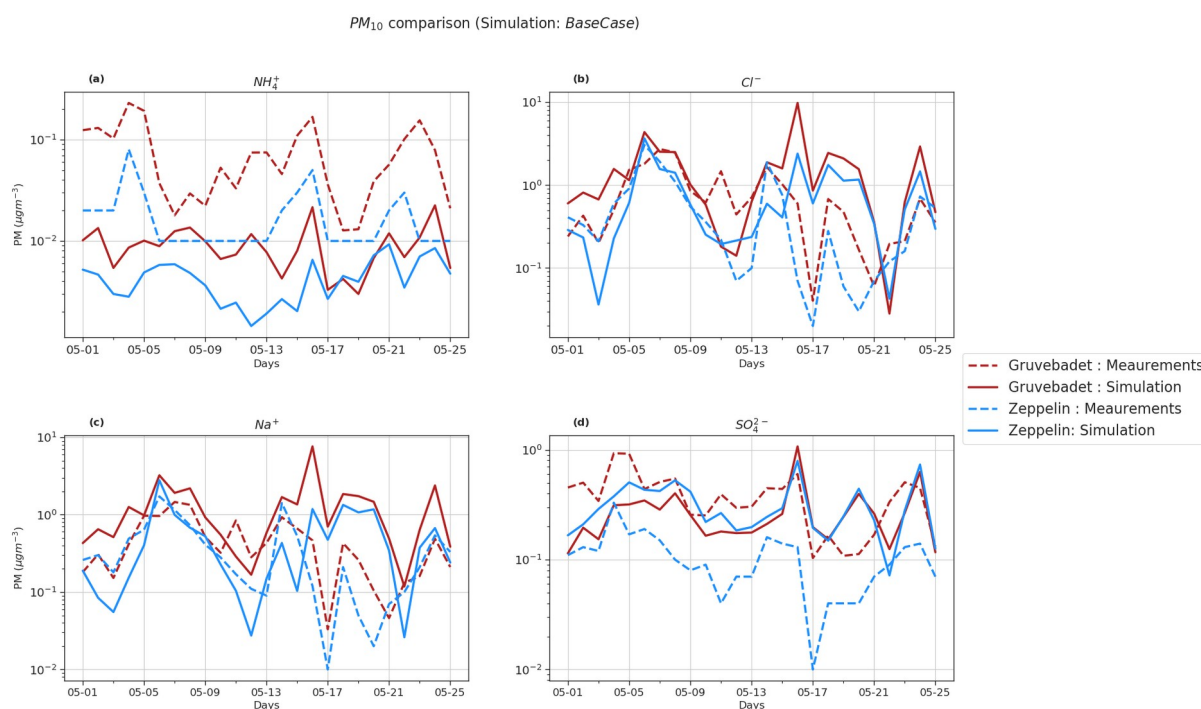


532 **Figure 5.** Simulated median mass size distribution for *BaseCase* simulations. The upper panel
 533 (a) shows the median mass size distribution for compounds Cl^- , Na^+ , MSA, SO_4^{2-} , NH_4^+ , HIO_3
 534 and NO_3^- for the entire size distribution ranging from 1.07 nm-10 μm . The lower panel (b)
 535 shows the relative mass fractions or contribution of compounds Cl^- , Na^+ , MSA, SO_4^{2-} , NH_4^+ ,
 536 HIO_3 and NO_3^- to total non-refractory PM at different sizes.

537 Figure 6 compares the daily PM_{10} filter measurements to the modeled values at both
 538 measurement stations. The model prediction of PM_{10} Cl^- , Na^+ , SO_4^{2-} and NH_4^+ , was
 539 evaluated using statistical metrics such as NMB, FAC2, correlation coefficient (r) and RMSE
 540 (Table 3). Though the model does well in simulating the trends of PM_{10} SO_4^{2-} , Na^+ and Cl^- at
 541 Zeppelin (r values of 0.35, 0.51 and 0.6 respectively), it is unable to predict the NH_4^+ trends
 542 accurately ($r = -0.08$).

543 Pearson correlation (r -values) at Gruvebadet are in the range of 0.29-0.34 for PM_{10} NH_4^+ ,
 544 SO_4^{2-} , Na^+ and Cl^- implying that the model trends are reasonably consistent with the measured
 545 trends. However, at Gruvebadet the NMB values for PM_{10} NH_4^+ and SO_4^{2-} are underpredicted
 546 (NMB = -0.88 and -0.28 respectively), while PM_{10} Na^+ and Cl^- show a large overprediction
 547 (1.81 and 1.05) in the modelled values. In contrast, at Zeppelin, the modeled PM SO_4^{2-} is
 548 overestimated (NMB=1.96). Likewise, large RMSE and negligible FAC2 values, for PM_{10}
 549 Na^+ , and Cl^- imply discrepancies between the predicted and measured values, indicating that
 550 the model is overestimating PM_{10} SO_4^{2-} , Na^+ and Cl^- at Gruvebadet and PM_{10} SO_4^{2-} at

551 Zeppelin. In summary, the model tends to overpredict PM_{10} Na^+ , Cl^- and SO_4^{2-} concentrations,
 552 but on the other hand, does reasonably well in predicting the daily measured trends.
 553 Additionally, the modeled PM_{10} Cl^-/Na^+ molar ratio at Gruvebadet and Zeppelin is ~ 0.79 and
 554 ~ 0.95 , respectively. This is much higher than the observed PM_{10} Cl^-/Na^+ molar ratio at both
 555 sites (~ 0.39). One likely reason for this is the overestimated sea spray aerosol emissions. The
 556 PM_{10} Cl^-/Na^+ molar ratios give a measure of the acidic nature of aerosol, since increased
 557 condensation of strong acid MSA and H_2SO_4 increases acidity of aerosols thereby causing
 558 loss of Cl^- (dechlorination) as HCl (Ayers et al., 1999; Frey et al., 2020). Thus, increased
 559 availability of H_2SO_4 and MSA in particle phase in Aitken mode particles results in acid-
 560 induced Cl^- loss from sea-spray particles.
 561



562 **Figure 6.** PM_{10} comparison of *BaseCase* simulations with daily filter samples from
 563 Gruvebadet and Zeppelin for the entire modeled period. Panel (a) shows PM_{10} NH_4^+ , (b)
 564 shows PM_{10} Cl^- , (c) shows PM_{10} Na^+ and (d) shows PM_{10} SO_4^{2-} filter samples. The dotted
 565 lines in each panel indicate measurement values, and the solid line denotes simulated values.
 566 The ordinate is plotted in log scale to better visualize the low values.

567

568 **Table 3:** Evaluation of modeled PM_{10} values at both sites of Gruvebadet (G) and Zeppelin (Z)
 569 for the four particle-phase species Cl^- , Na^+ , SO_4^{2-} and NH_4^+ .

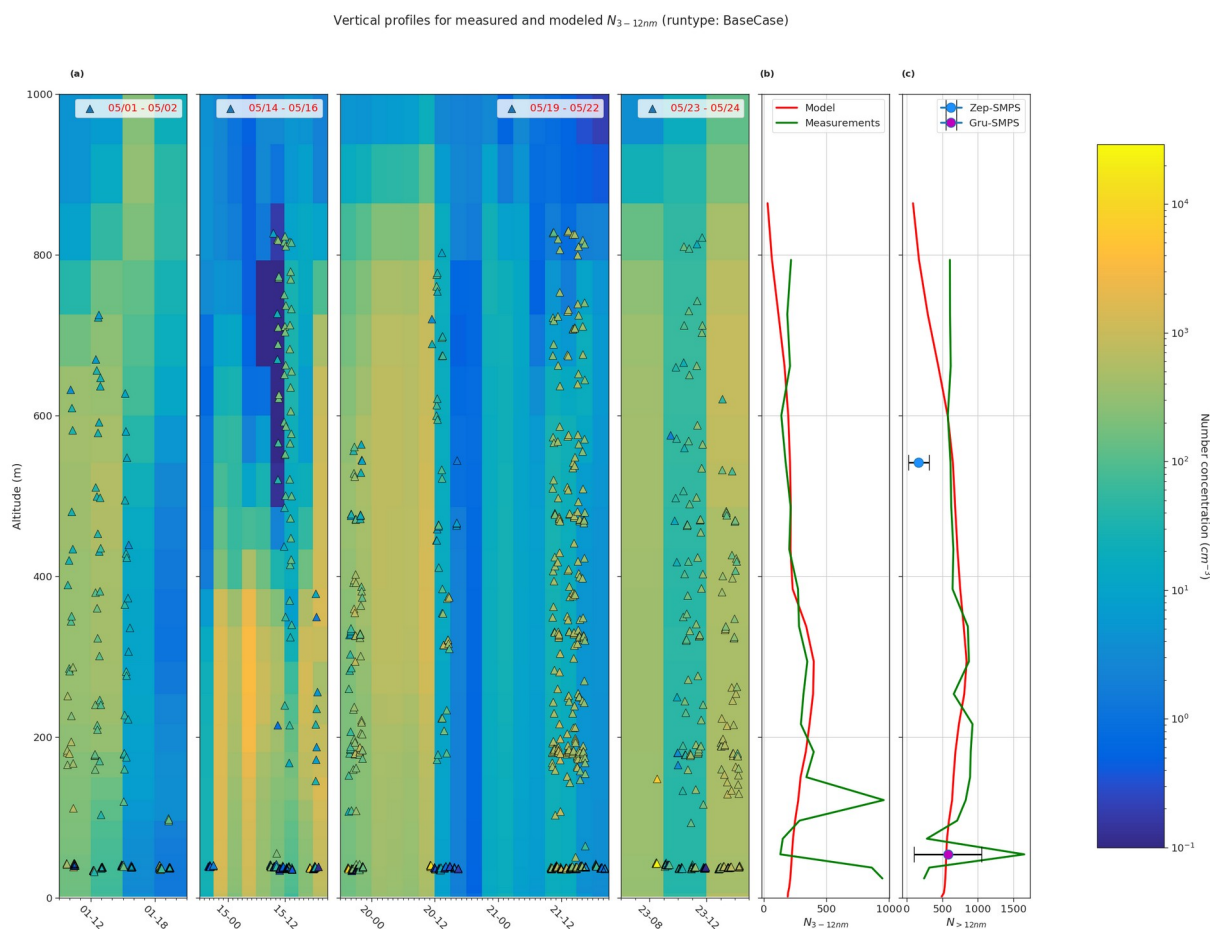
Species	Normalized mean	Correlation	RMSE ($\mu\text{g m}^{-3}$)	FAC2
---------	-----------------	-------------	-------------------------------	------

	bias factor (NMB)	coefficient (r)		
NH ₄ ⁺	-0.88 ^G , -0.76 ^Z	0.34 ^G , -0.08 ^Z	0.09 ^G , 0.02 ^Z	0.04 ^G , 0.2 ^Z
SO ₄ ²⁻	-0.28 ^G , 1.96 ^Z	0.33 ^G , 0.35 ^Z	0.27 ^G , 0.26 ^Z	0.6 ^G , 0.24 ^Z
Na ⁺	1.81 ^G , 0.36 ^Z	0.29 ^G , 0.51 ^Z	1.67 ^G , 0.55 ^Z	0.4 ^G , 0.48 ^Z
Cl ⁻	1.05 ^G , 0.39 ^Z	0.24 ^G , 0.60 ^Z	2.08 ^G , 0.74 ^Z	0.24 ^G , 0.44 ^Z

570

571 **3.3 Vertical profiles of ultra-fine particle**

572 Figure 7 (a) shows the measured vertical $N_{3-12 \text{ nm}}$ concentrations from CPC onboard the UAS
573 for four measurement periods overlayed onto simulated vertical profiles. Figure 7 (b) and (c)
574 show the mean vertical profiles for $N_{3-12 \text{ nm}}$ and $N_{>12 \text{ nm}}$ for both the *BaseCase* simulation and
575 UAS measurements for the entire selected period. The model underestimates the measured N_{3-12}
576 $N_{3-12 \text{ nm}}$ and $N_{>12 \text{ nm}}$ vertical particle number concentrations below 200 m a.s.l. The NMB for N_{3-12}
577 $N_{3-12 \text{ nm}}$ and $N_{>12 \text{ nm}}$ is -0.28 and -0.14, respectively, implying that the model underestimates the
578 particle number concentrations. Both the modeled and measured mean particle number
579 concentrations for $N_{3-12 \text{ nm}}$ and $N_{>12 \text{ nm}}$ are in good agreement between the heights of 200-600
580 m a.s.l. The lower calculated concentrations of modeled mean particle number concentrations
581 above 600 m a.s.l is most likely affected by higher turbulence in the transition zone from the
582 boundary layer to the free troposphere, which might cause a large mixing of aerosol particles.
583 It should be noted that, at Gruvebadet, the mean SMPS particle number concentrations are in
584 good agreement with the modeled particle number concentrations. However, at the altitude of
585 the Zeppelin station, both the model and UAS measurements of $N_{>12 \text{ nm}}$ are substantially
586 higher (factor of 4) than the mean particle number concentrations measured with the SMPS at
587 Zeppelin. This finding further strengthens the conclusion that the complex orography at Ny-
588 Ålesund highly affects the variability in the vertical scale, which may cause this discrepancy
589 in the observed and modeled particle number concentrations at Zeppelin (see section 3.1). The
590 UAS measurements were carried out at the airport on Ny-Ålesund (and the UAS was flown
591 around Ny-Ålesund) where the boundary layer measurements, like the model, most likely
592 resemble the general Arctic marine boundary layer conditions. Figure S9 shows the influence
593 of different sensitivity simulations on the modeled vertical particle number concentrations.
594 The large spread in the modeled vertical particle number concentrations in Figure S9,
595 highlights the importance of constraining uncertain parameters such as cloud supersaturation
596 and NH₃ gas emissions, to better simulate secondary aerosol formation in marine polar
597 regions.



599 **Figure 7.** Comparison of vertical profiles of measured particle number concentration and
 600 *BaseCase* simulation. Panel (a) shows measured particle number concentration between 3 -12
 601 nm ($N_{3-12\text{ nm}}$, triangles), from CPC onboard the UAS during 4 periods 05/01- 05/02, 05/14 -
 602 05/16, 05/19- 05/22, and 05/23- 05/24 (in legend) overlaid onto the simulated N_{3-12nm} for
 603 the same periods. Panel (b) shows the simulated and measured mean N_{3-12nm} and panel (c)
 604 show the $N_{>12nm}$ for the selected period. Additionally, Panel (c) also shows the mean SMPS
 605 particle concentrations at both Gruvebadet and Zeppelin. The horizontal bars for the mean
 606 SMPS values represent the standard deviation.

607

608 3.4 Sensitivity Tests

609 In this section, we will discuss the results from the sensitivity tests that we performed to
 610 complement the main *BaseCase* simulations. The settings of different sensitivity tests are
 611 described in Table 1.

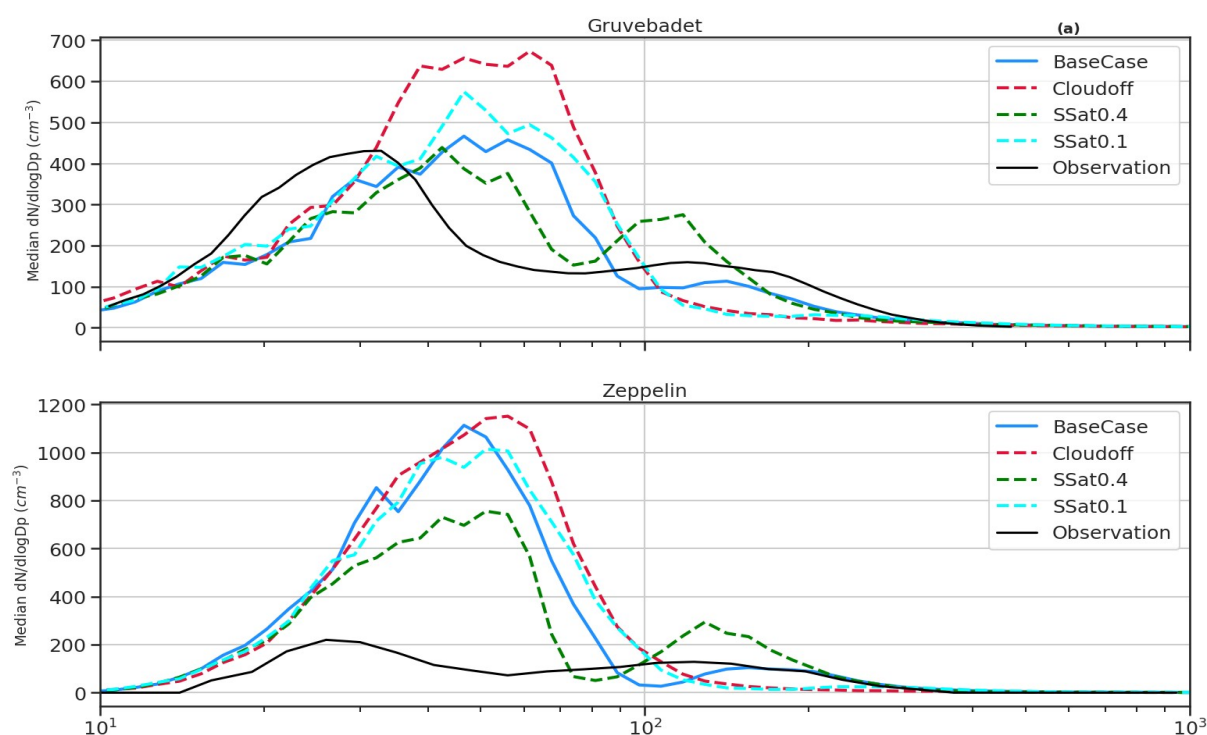
612 Median particle size distribution for sensitivity tests

613 The sensitivity study *Cloudoff* was performed to test how in-cloud processing affects the
614 formation of larger particles, especially the accumulation mode (Figure 8). In the *Cloudoff*
615 test, in-cloud processing was switched off in the model and the RH was set to just below
616 supersaturation (99.9999%) in the model grid cell where clouds (RH=100.5%) exists in the
617 *BaseCase* runs. The aim of the *Cloudoff* simulation was to investigate if the model can
618 capture the observed accumulation mode without aerosol cloud processing. It is clear from
619 Figure 8 (a) and (b) that in *Cloudoff* simulations, the median size distribution lacks the
620 accumulation mode and Hoppel minima and has a higher Aitken mode particle concentration
621 compared to either *BaseCase* or the measured median size distribution. This further
622 emphasizes the importance of in-cloud processing in activation of particles to CCN sizes and
623 their growth to larger sizes. However, it should be noted, that other processes such as
624 Brownian scavenging by larger cloud droplets could result in the shift in particles from the
625 Aitken mode to accumulation mode (as seen in median measured size distribution, Figure 8,
626 Noble and Hudson, (2019)). Another noteworthy point in *Cloudoff* simulations is the larger
627 number concentration of particles <10 nm compared to other cases. One plausible reason is
628 the lack of activated cloud droplets, since the large surface areas of activated droplets are
629 efficient at Brownian scavenging of smaller particles (Hudson et al., 2015). Likewise, the
630 median particle number size distribution from the sensitivity tests with lower cloud
631 supersaturation (S) of 0.1% $SSat=0.1$, reduces the accumulation mode particles, since there
632 are fewer particles with $S_c < S$ available for activation. Increasing S to 0.4% increases
633 accumulation mode particles, since more particles with $S_c < S$ are activated to cloud droplets
634 (Aitken mode concentration decreases with respect to *BaseCase* simulations, since more
635 smaller particles are activated into cloud droplets). Therefore, simulated results show that
636 increasing the cloud supersaturation results in a higher number of smaller particles being
637 activated into cloud droplets and shifts the simulated Hoppel minima close to the measured
638 sizes. Figure S6, in supplementary shows median particle size distribution for all sensitivity
639 tests.

640 The *SalterSSA* sensitivity test underestimates both the Aitken and accumulation mode
641 concentrations at Gruvebadet (Figure S5, supplementary). The Salter sea-spray
642 parameterization produces ~ 2 magnitudes fewer Aitken mode particles compared to Sofiev
643 et. al, 2011, while the coarse mode particle emissions using *SalterSSA* parameterization are
644 higher than Sofiev et. al 2011. This can cause MSA, H₂SO₄ and NH₃ to partition onto coarse
645 mode particles rather than contributing to NPF and growth of the nucleation and Aitken mode
646 particles, which substantially lowers the Aitken and accumulation mode number

647 concentrations. The *NPFoff* simulation from Figure S6 shows lower Aitken mode
 648 concentrations, implying that the main contributor to Aitken mode particle number
 649 concentrations are the secondary aerosols rather than the primary sea-salt particles.

650



651 **Figure 8:** Median size distribution at Gruvebadet (panel (a)) and Zeppelin (panel (b)) for all
 652 the sensitivity tests *Cloudoff*, *SSat=0.4*, and *SSat=0.1* (colored dashed lines) including
 653 *BaseCase* (blue solid line) and observations (black solid line).

654 Another parameter of uncertainty is the concentration of NH₃ in the marine atmosphere. The
 655 *LowNH₃* simulations, as expected, result in lower Aitken mode particles, whereas *HighNH₃*
 656 simulations show an overprediction of Aitken mode concentrations (Figure S6,
 657 supplementary). This underlines the necessity of constraining ocean and marine emissions of
 658 NH₃ to better predict the aerosol particle formation in marine polar environments.

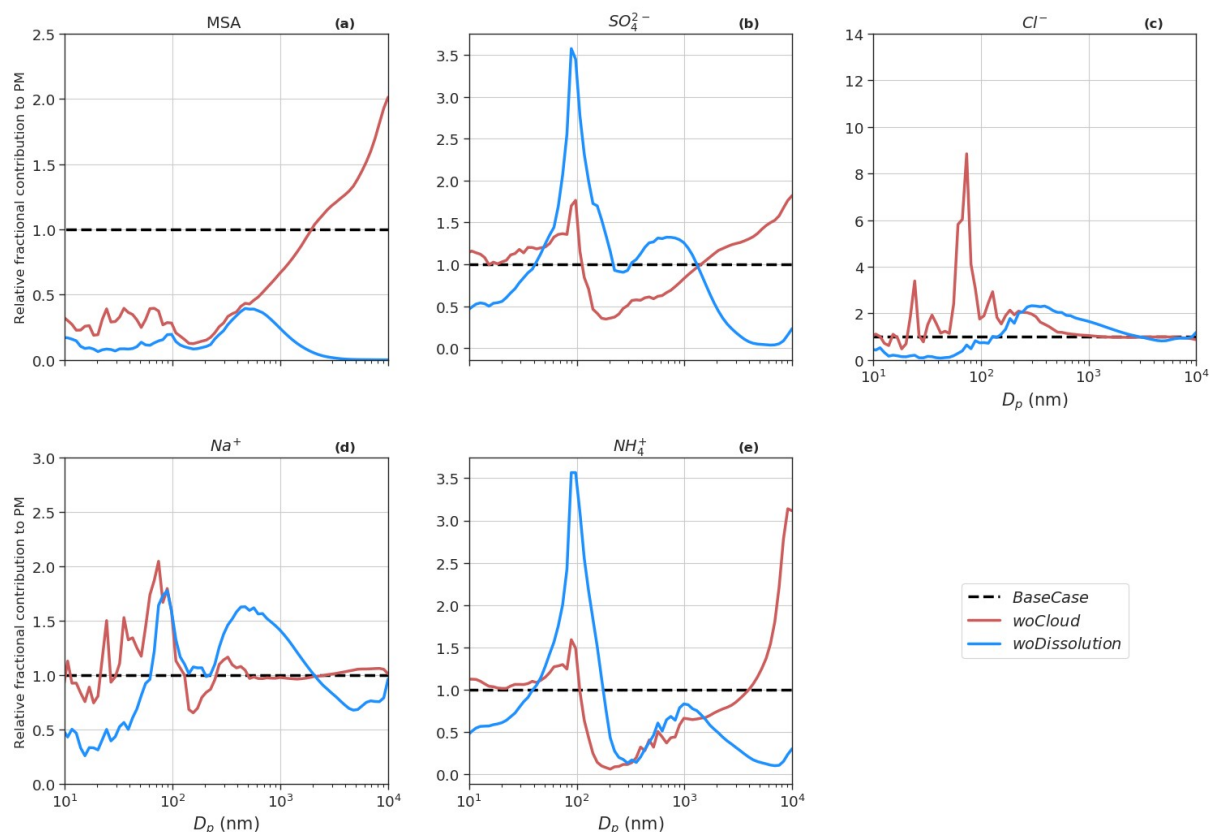
659 Particle phase comparison for sensitivity tests

660 Figure 9 shows the contribution of constituent compounds to PM at different particle sizes
 661 with respect to the *BaseCase* simulation. The overall mean contribution of SO₄²⁻ and MSA to
 662 total PM₁₀ decreased by ~8% and 11% respectively, in *Cloudoff* runs compared to the
 663 *BaseCase* simulations. It is expected that in non-cloud conditions there is a reduction in SO₄²⁻
 664 and MSA PM contribution because of the reduced partitioning of gaseous SO₂ to the cloud
 665 droplets (for PM SO₄²⁻ formation) and inhibition of MSIA ozonolysis in the cloud droplets

666 (leading to PM MSA formation) (Chen et al., 2018; Hoffmann et al., 2016; Wollesen de Jonge
667 et al., 2021). This is observed for accumulation mode particles between size ranges of 100 nm
668 to 1 μm which is characterized by lower SO_4^{2-} and MSA PM. On the other hand, PM SO_4^{2-} and
669 MSA increase for coarse mode particles ($> 1 \mu\text{m}$). Without cloud droplet activation the
670 deliquescent sea spray coarse mode particles become a major liquid water reservoir where
671 MSIA and to a lesser extent SO_2 are dissolved and oxidized into MSA and SO_4^{2-} , which partly
672 explains the increase in PM MSA and SO_4^{2-} for sizes $> 1 \mu\text{m}$. The results from *Cloudoff*
673 simulation agrees with the findings from Wollesen de Jonge et al., 2021, who found that MSA
674 was almost exclusively formed in the aqueous phase via MSIA ozonolysis in cloud droplets
675 and deliquescent particles during and in between in-cloud periods. PM SO_4^{2-} in *Cloudoff* runs
676 is mainly driven via condensation of H_2SO_4 , since an increase in SO_2 gas-phase
677 concentrations ($\sim 42\%$ with respect to *BaseCase*) promoted gas-phase H_2SO_4 production
678 (increase of $\sim 44\%$ with respect to *BaseCase*), and therefore H_2SO_4 derived PM SO_4^{2-} .

679 In the *woDissolution* simulation, all the PM MSA and SO_4^{2-} are a result of the condensation of
680 $\text{MSA}_{(g)}$ and $\text{H}_2\text{SO}_{4(g)}$, since irreversible aqueous-phase chemistry is switched off. The overall
681 contribution of PM SO_4^{2-} to the total PM_{10} increases by $\sim 12\%$ relative to the *BaseCase* run,
682 while on the other hand, the contribution of PM MSA decreases by $\sim 87\%$ (relative to
683 *BaseCase*). The lower PM_{10} MSA in *woDissolution* simulation emphasizes the importance of
684 aqueous-phase formation of MSA to the growth of particles. The effect of precipitation on
685 modeled PM (*NoPrecip*) indicates an increase in PM Na^+ and Cl^- of $\sim 112\%$ and 119%
686 respectively, as compared to *BaseCase* (Figure S8). This is because of the decrease in the wet
687 deposition of aerosol and sea-spray particles by rain events and below cloud scavenging. The
688 consequence of neglecting precipitation results in increased condensation sink for H_2SO_4 and
689 NH_3 (increase of 62% and 22% in PM SO_4^{2-} , NH_4^+ respectively), but since sea-spray aerosols
690 are not scavenged by the wet removal process, the overall fractional contribution to PM by
691 SO_4^{2-} , NH_4^+ and MSA is lower relative to *BaseCase* runs.

692 *SalterSSA* simulation results in higher PM Cl^- and Na^+ (470% and 371% increase
693 respectively) compared to *BaseCase* runs. This is because Salter15 SSA parameterization
694 produces larger mass emission fluxes in size ranges $> 1 \mu\text{m}$ compared to Sofiev11 SSA
695 parameterization (Barthel et al., 2019). Additionally, there is an increase of $\sim 19\%$ in PM
696 MSA, largely due to formation of MSA in larger deliquescent coarse mode particles.

Contribution to PM relative to *BaseCase*

698 **Figure 9:** Contribution of constituent compounds, namely, MSA (panel (a)), SO_4^{2-} (panel
 699 (b)), Cl^- (panel(c)), Na^+ (panel (d)) and NH_4^+ (panel (e)) to PM with respect to *BaseCase* (the
 700 black dotted line).

701 4. Summary and conclusions

702 In this work, we attempt to simulate secondary aerosol formation at remote Arctic sites of
 703 Gruvebadat and Zeppelin, Ny-Ålesund, during the period of 1st - 25th of May 2018. We used
 704 the 1-dimensional column model ADCHEM which was run along FLEXPART generated
 705 Lagrangian trajectories. Since the air mass spend most of their time over the open ocean
 706 upwind of Ny-Ålesund, we use a comprehensive multi-phase DMS chemistry scheme coupled
 707 with MCMv3.3.1 and PRAM.

708 In the model, new particles are formed via ion-mediated $\text{H}_2\text{SO}_4\text{-NH}_3$ nucleation, with the
 709 initial particle growth mainly driven by condensation of H_2SO_4 , while the secondary PM_{10}
 710 MSA and SO_4^{2-} contribution was mainly formed by oxidation of MSIA and SO_2 in the
 711 aqueous phase. At Gruvebadet, the modeled median particle number size distribution agrees
 712 reasonably well with the measurements, however, at Zeppelin, the simulated Aitken mode

713 median concentration is overestimated by a factor of 5.5. This relatively large discrepancy in
714 modeled and measured particle size distributions at Zeppelin, and likewise the large
715 difference between the measured particle number size distributions at Gruvebadet and
716 Zeppelin, can to a large extent be explained by the orographic effects at Zeppelin which
717 distorts the atmospheric boundary layer dynamics. Thus, while the model generally is able to
718 capture the particle number size distribution dynamics in the marine boundary layer, as
719 measured at the near sea level Gruvebadet site, it generally cannot capture the observations at
720 the mountain station of Zeppelin, which often lies above the boundary layer and may
721 experience free tropospheric conditions. This is also supported by the fact that $N_{>12\text{ nm}}$
722 concentrations measured with the UAS above Ny-Ålesund airport agrees well with the
723 modeled particle number concentrations, at the same altitude as Zeppelin. However, both the
724 model and UAS $N_{>12\text{ nm}}$ concentrations is a factor of 4 higher than the $N_{>12\text{ nm}}$ observation at
725 Zeppelin.

726 Both the measured and modeled particle size distribution, at both stations, show a distinct
727 Hoppel minima, which can be explained by in-cloud processing. Model sensitivity runs with
728 varying cloud supersaturation indicate that a cloud supersaturation of 0.4% or higher is
729 required for the model to capture the observed Hoppel minima. Furthermore, model
730 sensitivity runs show that the Aitken mode particle number concentrations are dominated by
731 contribution of secondary aerosols rather than primary emissions. The modeled PM_{10} Cl^- and
732 Na^+ is positively correlated when compared to PM_{10} filter samples. The main driver for
733 secondary aerosol particle growth is the formation of MSA via aqueous phase ozonolysis of
734 the DMS oxidation product MSIA. This demonstrates the importance of multi-phase DMS
735 chemistry in capturing the size resolved secondary aerosol growth in marine polar regions.

736 The sensitivity studies indicate that it is important to limit the uncertainties in parameters such
737 as cloud supersaturation and NH_3 emissions over open oceans to get a better constraint on
738 secondary aerosol formation and its subsequent climatic effects. This work was a first attempt
739 to simulate new particle and secondary aerosol formation in marine polar regions using a
740 process based chemistry transport model that includes a comprehensive multi-phase DMS and
741 halogen chemistry mechanism, detailed gas-molecular cluster and aerosol dynamics. In future
742 studies, we aim to implement ADCHEM for extended studies in polar marine and remote
743 continental regions where different atmospheric constituents such as HIO_3 , terpenes and
744 amines drive secondary aerosol formation.

745

746 Acknowledgments

747 The ALADINA study was funded by the German Research Foundation under grants LA
748 2907/5-3 and WI 1449/22-3. M. Sipilä acknowledges Academy of Finland (296628) and the
749 European Research Council (ERC) under the European Union's Horizon 2020 research and
750 innovation programme (GASPARCON, grant agreement no. 714621). This project has
751 received funding from the Swedish Research Council Formas project no. 2018-01745-
752 COBACCA, Swedish Research Council VR project no. 2019-05006, the Crafoord foundation
753 through project number 20210969. The presented research has been also been funded by the
754 Academy of Finland (Center of Excellence in Atmospheric Sciences) grant no. 4100200.

755 The authors would like to thank Tinja Olenius from the Swedish Meteorological and
756 Hydrological (SMHI) for help with the implementation of ACDC in ADCHEM. We would
757 also like to acknowledge the invaluable contribution of computational resources from CSC-IT
758 Center for Science, Finland. The authors would like to thank Noora Hyytinen from the
759 University of Oulu and University of Eastern Finland for providing the Henry's law
760 coefficient and dissolution constants which were used in the multi-phase chemistry.

761 Observations at Zeppelin station were supported by Swedish Environmental Protection
762 Agency (Naturvårdsverket) and by "Arctic Climate Across Scales (ACAS)" project funded by
763 Knut and Alice Wallenberg Foundation and by project IWCAA funded by agency FORMAS.
764 Authors would like to also thank for large support to Norwegian Polar Institute (NPI).

765

766 Author Contributions

767 CX, PR, MBoy, BA and BW planned and designed the study. PR, RWdJ and CX developed
768 and setup the ADCHEM model. CX, MB and VV performed the FLEXPART model
769 simulations. BA, BW, RoTh, and RT provided the measurement data. Resources were
770 provided by PR and MBoy. CX, PR and MBoy wrote the original draft, which included
771 visualizations made by CX and PR. All other authors discussed the results and contributed to
772 the final manuscript.

773

774 Competing interests

775 The authors declare that they do have no conflict of interest.

776 References

777 AMAP: Snow, Water, Ice and Permafrost in the Arctic (SWIPA): Climate Change and the

- 778 Cryosphere. [online] Available from:
779 <http://amap.no/swipa/%5Cnhttp://amap.no/swipa/CombinedReport.pdf%5Cnhttp://amap.no/swipa/SWIPA2011ExecutiveSummaryV2.pdf%5Cnhttp://amap.no/swipa/SwipaSimpleSummary-SEC.pdf>, 2011.
- 782 AMAP: AMAP, 2017. Snow, Water, Ice and Permafrost in the Arctic (SWIPA) 2017. Arctic
783 Monitoring and Assessment Programme (AMAP), Oslo, Norway. xiv + 269 pp. [online]
784 Available from: [https://www.amap.no/documents/doc/snow-water-ice-and-permafrost-in-the-](https://www.amap.no/documents/doc/snow-water-ice-and-permafrost-in-the-arctic-swipa-2017/1610)
785 [arctic-swipa-2017/1610](https://www.amap.no/documents/doc/snow-water-ice-and-permafrost-in-the-arctic-swipa-2017/1610), 2017.
- 786 AMAP: AMAP Arctic Climate Change Update 2021: Key Trends and Impacts, , 16 pp
787 [online] Available from: <https://www.amap.no/documents/download/6759/inline>, 2021.
- 788 Amore, A., Giardi, F., Becagli, S., Caiazza, L., Mazzola, M., Severi, M. and Traversi, R.:
789 Source apportionment of sulphate in the High Arctic by a 10 yr-long record from Gruvebadet
790 Observatory (Ny-Ålesund, Svalbard Islands), *Atmos. Environ.*, 270(July 2021),
791 doi:10.1016/j.atmosenv.2021.118890, 2022.
- 792 Arrigo, K. R. and van Dijken, G. L.: Continued increases in Arctic Ocean primary production,
793 *Prog. Oceanogr.*, 136, 60–70, doi:10.1016/j.pcean.2015.05.002, 2015.
- 794 Ayers, G. P., Gillett, R. W., Caine, J. M. and Dick, A. L.: Chloride and bromide loss from
795 sea-salt particles in Southern Ocean air, *J. Atmos. Chem.*, 33(3), 299–319,
796 doi:10.1023/A:1006120205159, 1999.
- 797 Baccarini, A., Karlsson, L., Dommen, J., Duplessis, P., Vüllers, J., Brooks, I. M., Saiz-Lopez,
798 A., Salter, M., Tjernström, M., Baltensperger, U., Zieger, P. and Schmale, J.: Frequent new
799 particle formation over the high Arctic pack ice by enhanced iodine emissions, *Nat.*
800 *Commun.*, 11(1), 1–11, doi:10.1038/s41467-020-18551-0, 2020.
- 801 Bärffuss, K., Pätzold, F., Altstädter, B., Kathe, E., Nowak, S., Bretschneider, L., Bestmann, U.
802 and Lampert, A.: New setup of the UAS ALADINA for measuring boundary layer properties,
803 atmospheric particles and solar radiation, *Atmosphere (Basel)*, 9(1),
804 doi:10.3390/atmos9010028, 2018.
- 805 Barnes, I., Hjorth, J. and Mihalopoulos, N.: Dimethyl sulfide and dimethyl sulfoxide and their
806 oxidation in the atmosphere, *Chem. Rev.*, 106(3), 940–975, doi:10.1021/cr020529+, 2006.
- 807 Barrie, L. A.: Arctic air pollution: An overview of current knowledge, *Atmos. Environ.*,
808 20(4), 643–663, doi:10.1016/0004-6981(86)90180-0, 1986.
- 809 Barthel, S., Tegen, I. and Wolke, R.: Do new sea spray aerosol source functions improve the
810 results of a regional aerosol model?, *Atmos. Environ.*, 198(May 2018), 265–278,
811 doi:10.1016/j.atmosenv.2018.10.016, 2019.
- 812 Beck, L. J., Sarnela, N., Junninen, H., Hoppe, C. J. M., Garmash, O., Bianchi, F., Riva, M.,
813 Rose, C., Peräkylä, O., Wimmer, D., Kausiala, O., Jokinen, T., Ahonen, L., Mikkilä, J.,
814 Hakala, J., He, X. C., Kontkanen, J., Wolf, K. K. E., Cappelletti, D., Mazzola, M., Traversi,
815 R., Petroselli, C., Viola, A. P., Vitale, V., Lange, R., Massling, A., Nøjgaard, J. K., Krejci, R.,

- 816 Karlsson, L., Zieger, P., Jang, S., Lee, K., Vakkari, V., Lampilahti, J., Thakur, R. C., Leino,
817 K., Kangasluoma, J., Duplissy, E. M., Siivola, E., Marbouti, M., Tham, Y. J., Saiz-Lopez, A.,
818 Petäjä, T., Ehn, M., Worsnop, D. R., Skov, H., Kulmala, M., Kerminen, V. M. and Sipilä, M.:
819 Differing Mechanisms of New Particle Formation at Two Arctic Sites, *Geophys. Res. Lett.*,
820 48(4), 1–11, doi:10.1029/2020GL091334, 2021.
- 821 Bengtsson, L., Hodges, K. I., Koumoutsaris, S., Zahn, M. and Berrisford, P.: The changing
822 energy balance of the polar regions in a warmer climate, *J. Clim.*, 26(10), 3112–3129,
823 doi:10.1175/JCLI-D-12-00233.1, 2013.
- 824 Berndt, T., Chen, J., Møller, K. H., Hyttinen, N., Prisle, N. L., Tilgner, A., Hoffmann, E. H.,
825 Herrmann, H. and Kjaergaard, H. G.: SO₂ formation and peroxy radical isomerization in the
826 atmospheric reaction of OH radicals with dimethyl disulfide, *Chem. Commun.*, 56(88),
827 13634–13637, doi:10.1039/d0cc05783e, 2020.
- 828 Bräuer, P., Tilgner, A., Wolke, R. and Herrmann, H.: Mechanism development and modelling
829 of tropospheric multiphase halogen chemistry: The CAPRAM Halogen Module 2.0 (HM2), *J.*
830 *Atmos. Chem.*, 70(1), 19–52, doi:10.1007/s10874-013-9249-6, 2013.
- 831 Campolongo, F., Saltelli, A., Jensen, N. R., Wilson, J. and Hjorth, J.: The role of multiphase
832 chemistry in the oxidation of dimethylsulphide (DMS). A latitude dependent analysis, *J.*
833 *Atmos. Chem.*, 32(3), 327–356, doi:10.1023/A:1006154618511, 1999.
- 834 Chen, Q., Sherwen, T., Evans, M. and Alexander, B.: DMS oxidation and sulfur aerosol
835 formation in the marine troposphere: A focus on reactive halogen and multiphase chemistry,
836 *Atmos. Chem. Phys.*, 18(18), 13617–13637, doi:10.5194/acp-18-13617-2018, 2018.
- 837 Clarke, A. D., Owens, S. R. and Zhou, J.: An ultrafine sea-salt flux from breaking waves:
838 Implications for cloud condensation nuclei in the remote marine atmosphere, *J. Geophys. Res.*
839 *Atmos.*, 111(6), doi:10.1029/2005JD006565, 2006.
- 840 Cuevas, C. A., Maffezzoli, N., Corella, J. P., Spolaor, A., Vallelonga, P., Kjær, H. A.,
841 Simonsen, M., Winstrup, M., Vinther, B., Horvat, C., Fernandez, R. P., Kinnison, D.,
842 Lamarque, J. F., Barbante, C. and Saiz-Lopez, A.: Rapid increase in atmospheric iodine levels
843 in the North Atlantic since the mid-20th century, *Nat. Commun.*, 9(1), 1–6,
844 doi:10.1038/s41467-018-03756-1, 2018.
- 845 Dall’osto, M., Beddows, D. C. S., Tunved, P., Harrison, R. M., Lupi, A., Vitale, V., Becagli,
846 S., Traversi, R., Park, K. T., Jun Yoon, Y., Massling, A., Skov, H., Lange, R., Strom, J. and
847 Krejci, R.: Simultaneous measurements of aerosol size distributions at three sites in the
848 European high Arctic, *Atmos. Chem. Phys.*, 19(11), 7377–7395, doi:10.5194/acp-19-7377-
849 2019, 2019a.
- 850 Dall’osto, M., Beddows, D. C. S., Tunved, P., Harrison, R. M., Lupi, A., Vitale, V., Becagli,
851 S., Traversi, R., Park, K. T., Jun Yoon, Y., Massling, A., Skov, H., Lange, R., Strom, J. and
852 Krejci, R.: Simultaneous measurements of aerosol size distributions at three sites in the
853 European high Arctic, *Atmos. Chem. Phys.*, 19(11), 7377–7395, doi:10.5194/acp-19-7377-
854 2019, 2019b.

- 855 Dall'Osto, M., Ovadnevaite, J., Paglione, M., Beddows, D. C. S., Ceburnis, D., Cree, C.,
856 Cortés, P., Zamanillo, M., Nunes, S. O., Pérez, G. L., Ortega-Retuerta, E., Emelianov, M.,
857 Vaqué, D., Marrasé, C., Estrada, M., Sala, M. M., Vidal, M., Fitzsimons, M. F., Beale, R.,
858 Airs, R., Rinaldi, M., Decesari, S., Facchini, M. C., Harrison, R. M., O'Dowd, C. and Simó,
859 R.: Antarctic sea ice region as a source of biogenic organic nitrogen in aerosols, *Sci. Rep.*,
860 7(1), 1–10, doi:10.1038/s41598-017-06188-x, 2017a.
- 861 Dall'Osto, M., Beddows, D. C. S., Tunved, P., Krejci, R., Ström, J., Hansson, H. C., Yoon, Y.
862 J., Park, K. T., Becagli, S., Udusti, R., Onasch, T., Ódowd, C. D., Simó, R. and Harrison, R.
863 M.: Arctic sea ice melt leads to atmospheric new particle formation, *Sci. Rep.*, 7(1), 1–10,
864 doi:10.1038/s41598-017-03328-1, 2017b.
- 865 Damian, V., Sandu, A., Damian, M., Potra, F. and Carmichael, G. R.: The kinetic
866 preprocessor KPP - A software environment for solving chemical kinetics, *Comput. Chem.*
867 *Eng.*, 26(11), 1567–1579, doi:10.1016/S0098-1354(02)00128-X, 2002.
- 868 Feingold, G. and Kreidenweis, S.: Does cloud processing of aerosol enhance droplet
869 concentrations?, *J. Geophys. Res. Atmos.*, 105(D19), 24351–24361,
870 doi:10.1029/2000JD900369, 2000.
- 871 Flossmann, A. I. and Wobrock, W.: Cloud processing of aerosol particles in marine
872 stratocumulus clouds, *Atmosphere (Basel)*, 10(9), doi:10.3390/atmos10090520, 2019.
- 873 Fossum, K. N., Ovadnevaite, J., Ceburnis, D., Dall'Osto, M., Marullo, S., Bellacicco, M.,
874 Simó, R., Liu, D., Flynn, M., Zuend, A. and O'Dowd, C.: Summertime Primary and
875 Secondary Contributions to Southern Ocean Cloud Condensation Nuclei, *Sci. Rep.*, 8(1), 1–
876 14, doi:10.1038/s41598-018-32047-4, 2018.
- 877 Frey, M. M., Norris, S. J., Brooks, I. M., Anderson, P. S., Nishimura, K., Yang, X., Jones, A.
878 E., Nerentorp Mastromonaco, M. G., Jones, D. H. and Wolff, E. W.: First direct observation
879 of sea salt aerosol production from blowing snow above sea ice, *Atmos. Chem. Phys.*, 20(4),
880 2549–2578, doi:10.5194/acp-20-2549-2020, 2020.
- 881 Gantt, B., He, J., Zhang, X., Zhang, Y. and Nenes, A.: Incorporation of advanced aerosol
882 activation treatments into CESM/CAM5: Model evaluation and impacts on aerosol indirect
883 effects, *Atmos. Chem. Phys.*, 14(14), 7485–7497, doi:10.5194/acp-14-7485-2014, 2014.
- 884 Granier, C., Darras, S., Gon, H. D. van der, Jana, D., Elguindi, N., Bo, G., Michael, G., Marc,
885 G., Jalkanen, J.-P., Kuenen, J., Lioussé, C., Quack, B., Simpson, D. and Sindelarova, K.: The
886 Copernicus Atmosphere Monitoring Service global and regional emissions (April 2019
887 version), (April) [online] Available from: [https://hal.archives-ouvertes.fr/hal-](https://hal.archives-ouvertes.fr/hal-02322431)
888 [02322431%0Ahttps://hal.archives-ouvertes.fr/hal-02322431/document](https://hal.archives-ouvertes.fr/hal-02322431/document), 2019.
- 889 Guenther, A., Karl, T., Harley, P., Weidinmyer, C., Palmer, P. I. and Geron, C.: Edinburgh
890 Research Explorer Estimates of global terrestrial isoprene emissions using MEGAN (Model
891 of Emissions of Gases and Aerosols from Nature) and Physics Estimates of global terrestrial
892 isoprene emissions using MEGAN (Model of Emissions of Gases an, *Atmos. Chem. Phys.*,
893 (6), 3181–3210, 2006.

- 894 Herenz, P., Wex, H., Henning, S., Kristensen, T. B., Rubach, F., Roth, A., Borrmann, S.,
895 Bozem, H., Schulz, H. and Stratmann, F.: Measurements of aerosol and CCN properties in the
896 Mackenzie River delta (Canadian Arctic) during spring--summer transition in May 2014,
897 *Atmos. Chem. Phys.*, 18(7), 4477–4496, doi:10.5194/acp-18-4477-2018, 2018.
- 898 Hertel, O., Christensen, J. and Hov, Ø.: Modelling of the end products of the chemical
899 decomposition of DMS in the marine boundary layer, *Atmos. Environ.*, 28(15), 2431–2449,
900 doi:10.1016/1352-2310(94)90395-6, 1994.
- 901 Hodshire, A. L., Campuzano-Jost, P., Kodros, J. K., Croft, B., Nault, B. A., Schroder, J. C.,
902 Jimenez, J. L. and Pierce, J. R.: The potential role of methanesulfonic acid (MSA) in aerosol
903 formation and growth and the associated radiative forcings, *Atmos. Chem. Phys.*, 19(5),
904 3137–3160, doi:10.5194/acp-19-3137-2019, 2019.
- 905 Hoffmann, E. H., Tilgner, A., Schrödner, R., Bräuer, P., Wolke, R. and Herrmann, H.: An
906 advanced modeling study on the impacts and atmospheric implications of multiphase
907 dimethyl sulfide chemistry, *Proc. Natl. Acad. Sci. U. S. A.*, 113(42), 11776–11781,
908 doi:10.1073/pnas.1606320113, 2016.
- 909 Hoose, C., Lohmann, U., Bennartz, R., Croft, B. and Lesins, G.: Global simulations of aerosol
910 processing in clouds, *Atmos. Chem. Phys.*, 8(23), 6939–6963, doi:10.5194/acp-8-6939-2008,
911 2008.
- 912 Hoppel, W. A. and Frick, G. M.: Submicron aerosol size distributions measured over the
913 tropical and South Pacific, *Atmos. Environ. Part A, Gen. Top.*, 24(3), 645–659,
914 doi:10.1016/0960-1686(90)90020-N, 1990.
- 915 Hoppel, W. A., Fitzgerald, J. W. and Larson, R. E.: Aerosol size distributions in air masses
916 advecting off the east coast of the United States., *J. Geophys. Res.*, 90(D1), 2365–2379,
917 doi:10.1029/JD090iD01p02365, 1985.
- 918 Hoppel, W. A., Frick, G. M. and Larson, R. E.: Effect of nonprecipitating clouds on the
919 aerosol size distribution in the marine boundary layer, *Geophys. Res. Lett.*, 13, 125–128,
920 1986.
- 921 Hudson, J. G., Noble, S. and Tabor, S.: Cloud supersaturations from CCN spectra Hoppel
922 minima, *J. Geophys. Res.*, 120(8), 3436–3452, doi:10.1002/2014JD022669, 2015.
- 923 Jang, S., Park, K. T., Lee, K., Yoon, Y. J., Kim, K., Chung, H. Y., Jang, E., Becagli, S., Lee,
924 B. Y., Traversi, R., Eleftheriadis, K., Krejci, R. and Hermansen, O.: Large seasonal and
925 interannual variations of biogenic sulfur compounds in the Arctic atmosphere (Svalbard;
926 78.9g N, 11.9g E), *Atmos. Chem. Phys.*, 21(12), 9761–9777, doi:10.5194/acp-21-9761-2021,
927 2021.
- 928 Jenkin, M. E., Saunders, S. M. and Pilling, M. J.: The tropospheric degradation of volatile
929 organic compounds: A protocol for mechanism development, *Atmos. Environ.*, 31(1), 81–
930 104, doi:10.1016/S1352-2310(96)00105-7, 1997.
- 931 Jenkin, M. E., Young, J. C. and Rickard, A. R.: The MCM v3.3.1 degradation scheme for

- 932 isoprene, *Atmos. Chem. Phys.*, 15(20), 11433–11459, doi:10.5194/acp-15-11433-2015, 2015.
- 933 Jeričević, A., Kraljević, L., Grisogono, B., Fagerli, H. and Večenaj, Ž.: Parameterization of
934 vertical diffusion and the atmospheric boundary layer height determination in the EMEP
935 model, *Atmos. Chem. Phys.*, 10(2), 341–364, doi:10.5194/acp-10-341-2010, 2010.
- 936 Jokinen, T., Sipilä, M., Kontkanen, J., Vakkari, V., Tisler, P., Duplissy, E. M., Junninen, H.,
937 Kangasluoma, J., Manninen, H. E., Petäjä, T., Kulmala, M., Worsnop, D. R., Kirkby, J.,
938 Virkkula, A. and Kerminen, V. M.: Ion-induced sulfuric acid–ammonia nucleation drives
939 particle formation in coastal Antarctica, *Sci. Adv.*, 4(11), 1–7, doi:10.1126/sciadv.aat9744,
940 2018.
- 941 Kettle, A. J. and Andreae, M. O.: Flux of dimethylsulfide from the oceans: A comparison of
942 updated data sets and flux models, *J. Geophys. Res. Atmos.*, 105(D22), 26793–26808,
943 doi:10.1029/2000JD900252, 2000.
- 944 Kim, I., Zhang, M., Kim, K. and Park, K.: First high-frequency underway observation of dms
945 distribution in the southern ocean during austral autumn, *Atmosphere (Basel)*, 12(1), 1–10,
946 doi:10.3390/atmos12010122, 2021.
- 947 Kirkby, J., Duplissy, J., Sengupta, K., Frege, C., Gordon, H., Williamson, C., Heinritzi, M.,
948 Simon, M., Yan, C., Almeida, J., Trostl, J., Nieminen, T., Ortega, I. K., Wagner, R., Adamov,
949 A., Amorim, A., Bernhammer, A. K., Bianchi, F., Breitenlechner, M., Brilke, S., Chen, X.,
950 Craven, J., Dias, A., Ehrhart, S., Flagan, R. C., Franchin, A., Fuchs, C., Guida, R., Hakala, J.,
951 Hoyle, C. R., Jokinen, T., Junninen, H., Kangasluoma, J., Kim, J., Krapf, M., Kurten, A.,
952 Laaksonen, A., Lehtipalo, K., Makhmutov, V., Mathot, S., Molteni, U., Onnela, A., Perakyla,
953 O., Piel, F., Petaja, T., Praplan, A. P., Pringle, K., Rap, A., Richards, N. A. D., Riipinen, I.,
954 Rissanen, M. P., Rondo, L., Sarnela, N., Schobesberger, S., Scott, C. E., Seinfeld, J. H.,
955 Sipilä, M., Steiner, G., Stozhkov, Y., Stratmann, F., Tomé, A., Virtanen, A., Vogel, A. L.,
956 Wagner, A. C., Wagner, P. E., Weingartner, E., Wimmer, D., Winkler, P. M., Ye, P., Zhang,
957 X., Hansel, A., Dommen, J., Donahue, N. M., Worsnop, D. R., Baltensperger, U., Kulmala,
958 M., Carslaw, K. S. and Curtius, J.: Ion-induced nucleation of pure biogenic particles, *Nature*,
959 533(7604), 521–526, doi:10.1038/nature17953, 2016.
- 960 Kloster, S., Feichter, J., Maier-Reimer, E., Six, K. D., Stier, P. and Wetzal, P.: DMS cycle in
961 the marine ocean-atmosphere system - A global model study, *Biogeosciences*, 3(1), 29–51,
962 doi:10.5194/bg-3-29-2006, 2006.
- 963 Lampert, A., Altstädter, B., Bärfuss, K., Bretschneider, L., Sandgaard, J., Michaelis, J.,
964 Lobitz, L., Asmussen, M., Damm, E., Käthner, R., Krüger, T., Lüpkes, C., Nowak, S., Peucker,
965 A., Rausch, T., Reiser, F., Scholtz, A., Sotomayor-Zakharov, D., Gaus, D., Bansmer, S.,
966 Wehner, B. and Pätzold, F.: Unmanned aerial systems for investigating the polar atmospheric
967 boundary layer-technical challenges and examples of applications, *Atmosphere (Basel)*,
968 11(4), doi:10.3390/ATMOS11040416, 2020.
- 969 Lana, A., Bell, T. G., Simó, R., Vallina, S. M., Ballabrera-Poy, J., Kettle, A. J., Dachs, J.,
970 Bopp, L., Saltzman, E. S., Stefels, J., Johnson, J. E. and Liss, P. S.: An updated climatology

- 971 of surface dimethylsulfide concentrations and emission fluxes in the global ocean, *Global*
972 *Biogeochem. Cycles*, 25(1), 1–17, doi:10.1029/2010GB003850, 2011.
- 973 Lee, H., Lee, K., Lunder, C. R., Krejci, R., Aas, W., Park, J., Park, K.-T., Lee, B. Y., Yoon,
974 Y. J. and Park, K.: Atmospheric new particle formation characteristics in the Arctic as
975 measured at Mount Zeppelin, Svalbard, from 2016 to 2018, *Atmos. Chem. Phys.*, 20(21),
976 13425–13441, doi:10.5194/acp-20-13425-2020, 2020..
- 977 Lenssen, N. J. L., Schmidt, G. A., Hansen, J. E., Menne, M. J., Persin, A., Ruedy, R. and
978 Zyss, D.: Improvements in the GISTEMP Uncertainty Model, *J. Geophys. Res. Atmos.*,
979 124(12), 6307–6326, doi:10.1029/2018JD029522, 2019.
- 980 Lupi, A., Busetto, M., Becagli, S., Giardi, F., Lanconelli, C., Mazzola, M., Udisti, R.,
981 Hansson, H. C., Henning, T., Petkov, B., Ström, J., Krejci, R., Tunved, P., Viola, A. Pietro
982 and Vitale, V.: Multi-seasonal ultrafine aerosol particle number concentration measurements
983 at the Gruvebadet observatory, Ny-Ålesund, Svalbard Islands, *Rend. Lincei*, 27, 59–71,
984 doi:10.1007/s12210-016-0532-8, 2016.
- 985 Mårtensson, E. M., Nilsson, E. D., de Leeuw, G., Cohen, L. H. and Hansson, H.-C.:
986 Laboratory simulations and parameterization of the primary marine aerosol production, *J.*
987 *Geophys. Res. Atmos.*, 108(D9), doi:https://doi.org/10.1029/2002JD002263, 2003.
- 988 Mikkola, J.: Slope and valley winds in the Himalayas as simulated by the Weather Research
989 and Forecasting model, , 42, 2020.
- 990 Monahan, E. C., Spiel, D. E. and Davidson, K. L.: A Model of Marine Aerosol Generation
991 Via Whitecaps and Wave Disruption, in *Oceanic Whitecaps: And Their Role in Air-Sea*
992 *Exchange Processes*, edited by E. C. Monahan and G. Mac Niocaill, pp. 167–174, Springer
993 Netherlands, Dordrecht., 1986.
- 994 Moroni, B., Ritter, C., Crocchianti, S., Markowicz, K., Mazzola, M., Becagli, S., Traversi, R.,
995 Krejci, R., Tunved, P. and Cappelletti, D.: Individual Particle Characteristics, Optical
996 Properties and Evolution of an Extreme Long-Range Transported Biomass Burning Event in
997 the European Arctic (Ny-Ålesund, Svalbard Islands), *J. Geophys. Res. Atmos.*, 125(5), 1–17,
998 doi:10.1029/2019JD031535, 2020.
- 999 Nightingale, P. D., Liss, P. S. and Schlosser, P.: Measurements of air-sea gas transfer during
1000 an open ocean algal bloom, *Geophys. Res. Lett.*, 27(14), 2117–2120,
1001 doi:10.1029/2000GL011541, 2000.
- 1002 Noble, S. and Hudson, J. G.: Cloud supersaturations and Hoppel minima, *AIP Conf. Proc.*,
1003 1527(September), 706–709, doi:10.1063/1.4803368, 2013.
- 1004 Noble, S. R. and Hudson, J. G.: Effects of Continental Clouds on Surface Aitken and
1005 Accumulation Modes, *J. Geophys. Res. Atmos.*, 124(10), 5479–5502,
1006 doi:10.1029/2019JD030297, 2019.
- 1007 Olenius, T., Kupiainen-Määttä, O., Ortega, I. K., Kurtén, T. and Vehkamäki, H.: Free energy
1008 barrier in the growth of sulfuric acid-ammonia and sulfuric acid-dimethylamine clusters, *J.*

- 1009 Chem. Phys., 139(8), doi:10.1063/1.4819024, 2013.
- 1010 Oshima, N., Yukimoto, S., Deushi, M., Koshiro, T., Kawai, H., Tanaka, T. Y. and Yoshida,
1011 K.: Global and Arctic effective radiative forcing of anthropogenic gases and aerosols in MRI-
1012 ESM2.0, Prog. Earth Planet. Sci., 7(1), doi:10.1186/s40645-020-00348-w, 2020.
- 1013 Öström, E., Putian, Z., Schurgers, G., Mishurov, M., Kivekäs, N., Lihavainen, H., Ehn, M.,
1014 Rissanen, M. P., Kurtén, T., Boy, M., Swietlicki, E. and Roldin, P.: Modeling the role of
1015 highly oxidized multifunctional organic molecules for the growth of new particles over the
1016 boreal forest region, Atmos. Chem. Phys., 17(14), 8887–8901, doi:10.5194/acp-17-8887-
1017 2017, 2017.
- 1018 Park, K. T., Jang, S., Lee, K., Yoon, Y. J., Kim, M. S., Park, K., Cho, H. J., Kang, J. H.,
1019 Udusti, R., Lee, B. Y. and Shin, K. H.: Observational evidence for the formation of DMS-
1020 derived aerosols during Arctic phytoplankton blooms, Atmos. Chem. Phys., 17(15), 9665–
1021 9675, doi:10.5194/acp-17-9665-2017, 2017.
- 1022 Park, K. T., Yoon, Y. J., Lee, K., Tunved, P., Krejci, R., Ström, J., Jang, E., Kang, H. J., Jang,
1023 S., Park, J., Lee, B. Y., Traversi, R., Becagli, S. and Hermansen, O.: Dimethyl Sulfide-
1024 Induced Increase in Cloud Condensation Nuclei in the Arctic Atmosphere, Global
1025 Biogeochem. Cycles, 35(7), 1–15, doi:10.1029/2021GB006969, 2021.
- 1026 Paulot, F., Jacob, D. J., Johnson, M. T., Bell, T. G., Baker, A. R., Keene, W. C., Lima, I. D.,
1027 Doney, S. C. and Stock, C. A.: Global oceanic emission of ammonia: Constraints from
1028 seawater and atmospheric observations, Global Biogeochem. Cycles, 29(8), 1165–1178,
1029 doi:10.1002/2015GB005106, 2015.
- 1030 Petäjä, T., Duplissy, E., Tabakova, K., Schmale, J., Altstädter, B. and Ancellet, G.: Overview :
1031 Integrative and Comprehensive Understanding on Polar Environments (iCUPE) – concept
1032 and initial results, , 8551–8592, 2020.
- 1033 Pisso, I., Sollum, E., Grythe, H., Kristiansen, N., Cassiani, M., Eckhardt, S., Arnold, D.,
1034 Morton, D., Thompson, R., Groot Zwaafink, C., Evangeliou, N., Sodemann, H., Haimberger,
1035 L., Henne, S., Brunner, D., Burkhardt, J., Fouilloux, A., Brioude, J., Philipp, A., Seibert, P. and
1036 Stohl, A.: The Lagrangian particle dispersion model FLEXPART version 10.3, Geosci. Model
1037 Dev. Discuss., (January), 1–67, doi:10.5194/gmd-2018-333, 2019.
- 1038 Platt, S. M., Hov, Ø., Berg, T., Breivik, K., Eckhardt, S., Hansson, H., Heintzenberg, J.,
1039 Hermansen, O., Heslin-rees, D. and Holmén, K.: Atmospheric composition in the European
1040 Arctic and 30 years of the Zeppelin Observatory , Ny-Ålesund, , 3321–3369, 2022.
- 1041 Quinn, P. K., Coffman, D. J., Johnson, J. E., Upchurch, L. M. and Bates, T. S.: Small fraction
1042 of marine cloud condensation nuclei made up of sea spray aerosol, Nat. Geosci., 10(9), 674–
1043 679, doi:10.1038/ngeo3003, 2017.
- 1044 Rader, F., Traversi, R., Severi, M., Becagli, S., Müller, K.-J., Nakoudi, K. and Ritter, C.:
1045 Overview of Aerosol Properties in the European Arctic in Spring 2019 Based on In Situ
1046 Measurements and Lidar Data, Atmosphere (Basel), 12(2), doi:10.3390/atmos12020271,
1047 2021.

- 1048 Riddick, S. N., Dragosits, U., Blackall, T. D., Daunt, F., Wanless, S. and Sutton, M. A.: The
1049 global distribution of ammonia emissions from seabird colonies, *Atmos. Environ.*, 55, 319–
1050 327, doi:10.1016/j.atmosenv.2012.02.052, 2012.
- 1051 Roldin, P., Swietlicki, E., Schurgers, G., Arneth, A., Lehtinen, K. E. J., Boy, M. and Kulmala,
1052 M.: Development and evaluation of the aerosol dynamics and gas phase chemistry model
1053 ADCHEM, *Atmos. Chem. Phys.*, 11(12), 5867–5896, doi:10.5194/acp-11-5867-2011, 2011.
- 1054 Roldin, P., Ehn, M., Kurtén, T., Olenius, T., Rissanen, M. P., Sarnela, N., Elm, J., Rantala, P.,
1055 Hao, L., Hyttinen, N., Heikkinen, L., Worsnop, D. R., Pichelstorfer, L., Xavier, C., Clusius,
1056 P., Öström, E., Petäjä, T., Kulmala, M., Vehkamäki, H., Virtanen, A., Riipinen, I. and Boy,
1057 M.: The role of highly oxygenated organic molecules in the Boreal aerosol-cloud-climate
1058 system, *Nat. Commun.*, 10(2019), 4370, doi:10.1038/s41467-019-12338-8, 2019.
- 1059 Rosati, B., Christiansen, S., Wollesen de Jonge, R., Roldin, P., Jensen, M. M., Wang, K.,
1060 Moosakutty, S. P., Thomsen, D., Salomonsen, C., Hyttinen, N., Elm, J., Feilberg, A., Glasius,
1061 M. and Bilde, M.: New Particle Formation and Growth from Dimethyl Sulfide Oxidation by
1062 Hydroxyl Radicals, *ACS Earth Sp. Chem.*, acsearthspacechem.0c00333,
1063 doi:10.1021/acsearthspacechem.0c00333, 2021.
- 1064 Rosenfeld, D., Andreae, M. O., Asmi, A., Chin, M., de Leeuw, G., Donovan, D. P., Kahn, R.,
1065 Kinne, S., Kivekäs, N., Kulmala, M., Lau, W., Schmidt, K. S., Suni, T., Wagner, T., Wild, M.
1066 and Quaas, J.: Global observations of aerosol-cloud-precipitation-climate interactions, *Rev.*
1067 *Geophys.*, 52(4), 750–808, doi:10.1002/2013RG000441, 2014.
- 1068 Salter, M. E., Zieger, P., Acosta Navarro, J. C., Grythe, H., Kirkevåg, A., Rosati, B., Riipinen,
1069 I. and Nilsson, E. D.: An empirically derived inorganic sea spray source function
1070 incorporating sea surface temperature, *Atmos. Chem. Phys.*, 15(19), 11047–11066,
1071 doi:10.5194/acp-15-11047-2015, 2015.
- 1072 Saunders, S. M., Jenkin, M. E., Derwent, R. G. and Pilling, M. J.: Protocol for the
1073 development of the Master Chemical Mechanism, MCM v3 (Part A): Tropospheric
1074 degradation of non-aromatic volatile organic compounds, *Atmos. Chem. Phys.*, 3(1), 161–
1075 180, doi:10.5194/acp-3-161-2003, 2003.
- 1076 Schemann, V. and Ebell, K.: Simulation of mixed-phase clouds with the ICON large-eddy
1077 model in the complex Arctic environment around Ny-Ålesund, *Atmos. Chem. Phys.*, 20(1),
1078 475–485, doi:10.5194/acp-20-475-2020, 2020.
- 1079 Scott, C. E., Rap, A., Spracklen, D. V., Forster, P. M., Carslaw, K. S., Mann, G. W., Pringle,
1080 K. J., Kivekäs, N., Kulmala, M., Lihavainen, H. and Tunved, P.: The direct and indirect
1081 radiative effects of biogenic secondary organic aerosol, *Atmos. Chem. Phys.*, 14(1), 447–470,
1082 doi:10.5194/acp-14-447-2014, 2014.
- 1083 Sofiev, M., Soares, J., Prank, M., De Leeuw, G. and Kukkonen, J.: A regional-to-global
1084 model of emission and transport of sea salt particles in the atmosphere, *J. Geophys. Res.*
1085 *Atmos.*, 116(21), doi:10.1029/2010JD014713, 2011.
- 1086 Stock, C. A., Dunne, J. P. and John, J. G.: Global-scale carbon and energy flows through the

- 1087 marine planktonic food web: An analysis with a coupled physical–biological model, *Prog.*
1088 *Oceanogr.*, 120, 1–28, doi:<https://doi.org/10.1016/j.pocean.2013.07.001>, 2014.
- 1089 Stohl, A., Sodemann, H., Eckhardt, S., Frank, A., Seibert, P. and Wotawa, G.: The Lagrangian
1090 particle dispersion model FLEXPART version 8 .2, 2005.
- 1091 Ström, J., Engvall, A. C., Delbart, F., Krejci, R. and Treffeisen, R.: On small particles in the
1092 Arctic summer boundary layer: Observations at two different heights near Ny-Ålesund,
1093 Svalbard, *Tellus, Ser. B Chem. Phys. Meteorol.*, 61 B(2), 473–482, doi:[10.1111/j.1600-](https://doi.org/10.1111/j.1600-0889.2008.00412.x)
1094 [0889.2008.00412.x](https://doi.org/10.1111/j.1600-0889.2008.00412.x), 2009.
- 1095 Traversi, R., Becagli, S., Severi, M., Caiazzo, L., Mazzola, M., Lupi, A., Fiebig, M.,
1096 Hermansen, O., Krejci, R. and Unit, A. S.: Arctic haze in a climate changing world: the 2010-
1097 2020 trend (HAZECLIC) 4, , (0000), 104–117, 2020.
- 1098 Tunved, P., Ström, J. and Krejci, R.: Arctic aerosol life cycle: Linking aerosol size
1099 distributions observed between 2000 and 2010 with air mass transport and precipitation at
1100 Zeppelin station, Ny-Ålesund, Svalbard, *Atmos. Chem. Phys.*, 13(7), 3643–3660,
1101 doi:[10.5194/acp-13-3643-2013](https://doi.org/10.5194/acp-13-3643-2013), 2013.
- 1102 Uhlig, C., Damm, E., Peeken, I., Krumpfen, T., Rabe, B., Korhonen, M. and Ludwichowski,
1103 K. U.: Sea ice and water mass influence dimethylsulfide concentrations in the central arctic
1104 ocean, *Front. Earth Sci.*, 7(July), 1–15, doi:[10.3389/feart.2019.00179](https://doi.org/10.3389/feart.2019.00179), 2019.
- 1105 Wainwright, C. D., Pierce, J. R., Liggio, J., Strawbridge, K. B., MacDonald, A. M. and
1106 Leaitch, R. W.: The effect of model spatial resolution on Secondary Organic Aerosol
1107 predictions: A case study at Whistler, BC, Canada, *Atmos. Chem. Phys.*, 12(22), 10911–
1108 10923, doi:[10.5194/acp-12-10911-2012](https://doi.org/10.5194/acp-12-10911-2012), 2012.
- 1109 Wentworth, G. R., Murphy, J. G., Croft, B., Martin, R. V., Pierce, J. R., Côté, J. S.,
1110 Courchesne, I., Tremblay, J. É., Gagnon, J., Thomas, J. L., Sharma, S., Toom-Sauntry, D.,
1111 Chivulescu, A., Lefebvre, M. and Abbatt, J. P. D.: Ammonia in the summertime Arctic
1112 marine boundary layer: Sources, sinks, and implications, *Atmos. Chem. Phys.*, 16(4), 1937–
1113 1953, doi:[10.5194/acp-16-1937-2016](https://doi.org/10.5194/acp-16-1937-2016), 2016.
- 1114 Wollesen de Jonge, R.: Supplement of Secondary aerosol formation from dimethyl sulfide –
1115 improved mechanistic understanding based on smog chamber experiments and modelling, ,
1116 9955–9976, 2021.
- 1117 Wollesen de Jonge, R., Elm, J., Rosati, B., Christiansen, S., Hyttinen, N., Lüdemann, D.,
1118 Bilde, M. and Roldin, P.: Secondary aerosol formation from dimethyl sulfide – improved
1119 mechanistic understanding based on smog chamber experiments and modelling, *Atmos.*
1120 *Chem. Phys.*, 1–33, doi:[10.5194/acp-2020-1324](https://doi.org/10.5194/acp-2020-1324), 2021.
- 1121 Xavier, C., Rusanen, A., Zhou, P., Dean, C., Pichelstorfer, L., Roldin, P. and Boy, M.:
1122 Aerosol mass yields of selected biogenic volatile organic compounds – a theoretical study
1123 with nearly explicit gas-phase chemistry, *Atmos. Chem. Phys.*, 19(22), 13741–13758,
1124 doi:[10.5194/acp-19-13741-2019](https://doi.org/10.5194/acp-19-13741-2019), 2019.

- 1125 Zheng, G., Wang, Y., Aiken, A. C., Gallo, F., Jensen, M. P., Kollias, P., Kuang, C., Luke, E.,
1126 Springston, S., Uin, J., Wood, R. and Wang, J.: Marine boundary layer aerosol in the eastern
1127 North Atlantic: Seasonal variations and key controlling processes, *Atmos. Chem. Phys.*,
1128 18(23), 17615–17635, doi:10.5194/acp-18-17615-2018, 2018.
- 1129 Ziska, F., Quack, B., Abrahamsson, K., Archer, S. D., Atlas, E., Bell, T., Butler, J. H.,
1130 Carpenter, L. J., Jones, C. E., Harris, N. R. P., Hepach, H., Heumann, K. G., Hughes, C.,
1131 Kuss, J., Krüger, K., Liss, P., Moore, R. M., Orlikowska, A., Raimund, S., Reeves, C. E.,
1132 Reifenhäuser, W., Robinson, A. D., Schall, C., Tanhua, T., Tegtmeier, S., Turner, S., Wang,
1133 L., Wallace, D., Williams, J., Yamamoto, H., Yvon-Lewis, S. and Yokouchi, Y.: Global sea-
1134 to-air flux climatology for bromoform, dibromomethane and methyl iodide, *Atmos. Chem.*
1135 *Phys.*, 13(17), 8915–8934, doi:10.5194/acp-13-8915-2013, 2013.