1 Secondary aerosol formation in marine Arctic environments: A model measurement

2 comparison at Ny-Ålesund

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27 Abstract

28 In this study, we modeled the aerosol particle formation along air mass trajectories arriving at the remote Arctic research stations Gruvebadet (67 m a.s.l) and Zeppelin (474 m a.s.l), Ny-29 Ålesund during May 2018. The aim of this study was to improve our understanding of 30 processes governing secondary aerosol formation in remote Arctic marine environments. We 31 run the Lagrangian chemistry transport model ADCHEM, along air mass trajectories 32 generated with FLEXPART v10.4. The air masses arriving at Ny-Ålesund spent most of their 33 time over the open ice-free ocean. In order to capture the secondary aerosol formation from 34 the DMS emitted by phytoplankton from the ocean surface, we implemented a recently 35 developed comprehensive DMS and halogen multi-phase oxidation chemistry scheme, 36 coupled with the widely used Master Chemical Mechanism (MCM). 37

38 The modeled median particle number size distributions are in close agreement with the 39 observations in the marine influenced boundary layer at near sea surface Gruvebadet site.

57 observations in the marme innucleed boundary layer at near sea surface or avolated site.

40 However, while the model reproduces the accumulation mode particle number concentrations

at Zeppelin, it overestimates the Aitken mode particle number concentrations by a factor of
~5.5. We attribute this to the deficiency of the model to capture the complex orographic
effects on the boundary layer dynamics at Ny-Ålesund. However, the model reproduces the
average vertical particle number concentration profiles within the boundary layer (0-600 m
a.s.l.) above Gruvebadet, as measured with Condensation Particle Counters (CPCs) on board
an Unmanned Aircraft Systems (UAS).

47 The model successfully reproduces the observed Hoppel minima, often seen in particle number size distributions at Ny-Ålesund. The model also supports the previous experimental 48 findings that ion mediated H₂SO₄-NH₃ nucleation can explain the observed new particle 49 formation in the marine Arctic boundary layer in the vicinity of Ny-Ålesund. Precursors 50 resulting from gas and aqueous phase DMS chemistry contribute to the subsequent growth of 51 the secondary aerosols. The growth of particles is primarily driven via H₂SO₄ condensation 52 and formation of methane sulfonic acid (MSA) through the aqueous-phase ozonolysis of 53 methane sulfinic acid (MSIA) in cloud and deliquescent droplets. 54

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56 **1. Introduction**

The Earth's radiation budget is influenced both directly and indirectly by aerosols, which 57 scatter and absorb the incoming short-wave radiation (direct effect) and serve as cloud 58 condensation nuclei (CCN, indirect effect), affecting both short and long-wave radiation 59 (Gantt et al., 2014; Oshima et al., 2020; Park et al., 2017; Scott et al., 2014). The Arctic 60 environments are susceptible to perturbations in the radiation balance, with some estimates 61 62 suggesting that, compared to the global average, the Arctic is warming at three times the rate, a phenomenon termed as Arctic amplification (AMAP, 2011, 2017, 2021; Lenssen et al., 63 64 2019; Tunved et al., 2013). The warming of the Arctic polar environment has accelerated sea ice loss, leading to a rapid decline in the extent and duration of snow cover and increase in 65 permafrost thaw (AMAP, 2011, 2017; Bengtsson et al., 2013). 66

The Arctic aerosol number concentration shows a pronounced seasonal variation, where the late winter and early spring period is characterized by elevated accumulation mode aerosol concentrations, accompanied by trace gases (mostly anthropogenic with long-range transported trace elements such as sulfates, soot, and Peroxy Acyl Nitrates (PANs)). This annually recurring phenomenon in late winter and spring is termed the Arctic Haze (Barrie, 1986; Lupi et al., 2016; Tunved et al., 2013). This contrasts with the summer period, when the atmospheric new particle formation is observed at Arctic sites, most likely due to low
background aerosol concentrations, increased photo-chemistry and biological activity
(Engvall et al., 2008; Heintzenberg et al., 2017; Tunved et al., 2013).

76 The climate change driven Arctic sea ice loss has a profound impact on natural aerosol production. Arrigo and van Dijken, (2015) found that decreasing and thinning of sea ice 77 increased the rates of phytoplankton net primary production by ~20% between the years 1998 78 79 and 2009. This can lead to an increase in the emissions of primary biogenic precursors such as dimethyl sulfide (DMS), nitrogen volatiles (e.g. alkyl-amines) (Dall'Osto et al., 2017a; 80 Dallósto et al., 2017b) and biological iodine species (Cuevas et al., 2018). DMS is emitted 81 into the atmosphere via air-sea gas exchanges (Park et al., 2017; Uhlig et al., 2019), and 82 accounts for ~80% of global natural sulfur emissions (Kettle and Andreae, 2000; Uhlig et al., 83 2019). Methane sulfonic acid (MSA) and sulphuric acid (H₂SO₄) is formed via DMS gas-84 85 phase oxidation by OH and halogen species (Cl, Br) (Hoffmann et al., 2016; Kim et al., 2021; Wollesen de Jonge et al., 2021). MSA and H_2SO_4 , together with ammonia (NH₃) or amines, 86 87 act as precursors contributing to new particle formation (NPF) and subsequently to CCN production, influencing cloud formation and radiative balance (Berndt et al., 2020; Dall'Osto 88 89 et al., 2017b; Hoffmann et al., 2016; Kim et al., 2021, Jang et al., 2021; Park et al., 2021). NH₃ plays a major role in particle formation through stabilization of sulfuric acid clusters 90 91 (Beck et al., 2021; Jokinen et al., 2018; Olenius et al., 2013). Depending on local parameters 92 such as ocean pH, salinity and temperature, global oceans can act either as a source or sink of 93 NH₃ (Paulot et al., 2015). Apart from participating in cluster formation, NH₃ influences the pH of marine aerosols by neutralizing the acid (H₂SO₄ and MSA) in the particles (Paulot et 94 al., 2015). Though a few potential sources of NH₃ are known, for example coastal sea bird 95 colonies, pockets of open water and melting sea ice in summertime Arctic, the magnitude of 96 the emissions remain uncertain (Dall'osto et al., 2019a; Riddick et al., 2012; Wentworth et al., 97 2016). 98

99 DMS oxidation chemistry has been under focus, but uncertainties in climate predictions 100 persist since the chemical transport models (CTMs) and global climate models (GCMs) employ fixed MSA and SO₂ yields from gas-phase oxidation of DMS to calculate aerosol 101 102 formation (Hertel et al., 1994; Hoffmann et al., 2016; Kloster et al., 2006; Wollesen de Jonge et al., 2021). Including a detailed multi-phase (aqueous-phase chemistry coupled with gas-103 104 phase chemistry) DMS chemistry in numerical models can overcome these uncertainties (Barnes et al., 2006; Campolongo et al., 1999). Reaction intermediates such as dimethyl 105 106 sulfoxide (DMSO), dimethyl sulfone (DMSO₂), methane sulfinic acid (MSIA) are water107 soluble, and experiments have shown that neglecting aqueous phase chemistry leads to either 108 an under-estimation of modeled MSA (Campolongo et al., 1999), or an over-estimation of gaseous SO₂ compared to measured values (Hoffmann et al., 2016). For example, the 109 temperature dependent ratio of MSA/non-sea-salt SO₄²⁻ (nss-SO₄²⁻) is often used to estimate 110 the contribution of DMS to sulfate budget (Ayers et al., 1999; Barnes et al., 2006). 111 Campolongo et al., 1999 showed that modeling studies which included a multi-phase DMS 112 chemistry can bridge the gap between temperature-dependent observations and modeled 113 MSA/nss-SO₄²⁻. Incorporating reactive halogens species over marine environments is crucial 114 in determining the DMS oxidation pathways to either SO₂ or MSA, the aging of marine 115 aerosols and the radiative properties of marine clouds (Hoffmann et al., 2016). Modeling 116 studies have shown that Cl⁻ and BrO⁻ radicals in the gas phase act as important DMS sinks 117 (Chen et al., 2018; Wollesen de Jonge et al., 2021), further underlining the role of halogen-118 DMS chemistry in the marine boundary layer. 119

Recent DMS+OH oxidation experiments performed in the AURA chamber at Aarhus 120 121 University show that MSA dominates the secondary aerosol mass formation (Rosati et al., 2021). Aerosol dynamics model simulations which intended to replicate the observations 122 123 during these AURA experiments, using the DMS gas-phase chemistry scheme from the Master Chemical Mechanism, MCMv3.3.1, (Jenkin et al., 1997, 2015; Saunders et al., 2003), 124 125 substantially underestimates the particle mass and number concentrations and the MSA:SO4²⁻ ratio (Rosati et al., 2021, Wollesen de Jonge, 2021). Based on these findings, Wollesen de 126 Jonge et al. (2021) developed a new DMS multi-phase chemistry scheme based on MCM 127 v3.3.1, CAPRAM DMS module 1.0 (DM1.0) (Hoffmann et al., 2016), a subset of the multi-128 phase halogen chemistry mechanism CAPRAM Halogen Module 2.0 (HM2.0) (Bräuer et al., 129 2013) and new reactions leading to the formation of hydroperoxymethyl thioformate 130 (HPMTF). With the new DMS multi-phase chemistry mechanism, the aerosol dynamics 131 model could capture the observed particle number concentrations and secondary PM MSA 132 and SO₄²⁻ during DMS oxidation experiments performed at both dry and humid conditions at 133 0 °C and 20 °C in the AURA chamber. For more details on the DMS, halogen and multi-phase 134 chemistry scheme used in ADCHEM, the reader is referred to the article and supplement of 135 Wollesen de Jonge et al., (2021). 136

The aim of this work is to understand the processes and DMS oxidation products governing the formation and growth of the secondary aerosol in pristine remote marine Arctic region. To facilitate this, we have implemented the above mentioned DMS multi-phase chemistry mechanism into ADCHEM (see Methods section) and modeled the aerosol formation along

air mass trajectories arriving at Ny-Ålesund. We compared the model results with 141 observations from Zeppelin (78°56' N, 11°53' E, 474 m a.s.l) and Gruvebadet (78°92' N, 142 11°90' E, 67 m a.s.l). These two sites represent remote marine Arctic conditions. Gruvebadet 143 144 represents ground-level concentrations as it is well within the boundary layer (BL). Zeppelin on the other hand, is most often above the BL in winter months and sometimes below the BL 145 during spring and summer months (Traversi et al., 2020). This implies that Zeppelin is more 146 influenced by long range transport, and Gruvebadet by more local effects (Traversi et al., 147 2020). This, demonstrates the complexity involved in capturing the atmospheric mixing and 148 secondary aerosol concentrations at Ny-Ålesund. The reason is that Svalbard has an 149 orographically complex terrain comprising of mountains, glaciers, fjords and flat lands that 150 introduce various micro-meteorological phenomena (Rader et al., 2021; Schemann and Ebell, 151 152 2020).

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154 **2. Methods**

Using the combined multi-phase DMS chemistry mechanism by Wollesen de Jonge et al., 155 (2021), MCMv3.3.1 and the monoterpene peroxy radical autoxidation mechanism (PRAM, 156 Roldin et al., 2019; Xavier et al., 2019) we simulated aerosol particle formation within the 157 marine boundary layer (MBL) upwind and at Ny-Ålesund between 1st - 25h May 2018, using 158 the Aerosol Dynamics, gas and particle-phase CHEMistry and radiative transfer model 159 ADCHEM (Öström et al., 2017; Roldin et al., 2011, 2019). We ran ADCHEM as a 160 161 Lagrangian model along the air mass trajectories arriving at Zeppelin every 3 hours during the selected period (in total 200 trajectory simulations). FLEXPART v10.4 was used to calculate 162 the air mass trajectories and potential emission sensitivity fields (Pisso et al., 2019; Stohl et 163 al., 2005). The simulation results for the vertical distribution of newly formed aerosol 164 (particle diameters < 12 nm) were validated against concurrent measurement data available 165 166 from the ALADINA (Application of Light-Weight Aircraft for Detecting in situ Aerosol) campaign, wherein a UAS was used to investigate horizontal and vertical distribution of 167 168 aerosol profiles in the marine boundary layer (ABL) (Lampert et al., 2020). Additionally, 169 modeled particle number size distributions and PM₁₀ chemical compositions were compared 170 to the available measured particle number size distributions and PM₁₀ filter samples at both Gruvebadet and Zeppelin measurement stations. 171

172 2.1 Air mass trajectories and potential emission sensitivity fields

173 We employed the Lagrangian particle dispersion model FLEXible PARTicle 174 (FLEXPARTv10.4) to assess the emission sensitivities or "footprints" of air-masses origin arriving at Zeppelin during the simulation period. FLEXPART is a stochastic model used to 175 176 compute dispersion of hypothetical particles, based on mean, turbulent and diffusive flows which can be run backwards in time to estimate air mass history at a site (Pisso et al., 2019). 177 European Center for Medium-Range Weather Forecasts (ECMWF) ERA5 reanalysis 178 meteorology with 137 height levels, 1-hour temporal and 0.5° x 0.5° spatial resolution, was 179 used as an input to FLEXPART (ERA5 hourly data on single levels from 1979 to present. 180 Copernicus Climate Change Service (C3S) Climate Data Store (CDS). last access 30th April 181 2021, 10.24381/cds.adbb2d47, ERA5 hourly data on pressure levels from 1979 to present. 182 Copernicus Climate Change Service (C3S) Climate Data Store (CDS). last access 30th April 183 184 2021, 10.24381/cds.bd0915c6). The air-mass history was simulated 7-day backwards in time and arriving at Zeppelin (474 m a.s.l) every 3 hours (at 00:00, 03:00, 06:00, 09:00, 12:00, 185 186 15:00, 15:00 and 21:00 UTC) for the entire simulation period (1st - 25th May 2018).

187 FLEXPART calculated normalized emission sensitivity fields were combined with oceanic emissions (DMS, dibromomethane, tribromomethane, iodomethane), NH₃ from sea-bird 188 189 colonies and anthropogenic emissions (NH₃, SO₂, CO, NO_x) derived from global inventories (see section 2.2). This was done to obtain representative emissions that consider the complete 190 emsission source regions along the trajectories, upwind of the measurement station. 191 192 Additional meteorological parameters such as temperature, pressure, sea surface temperature, specific humidity and cloud liquid water content from ERA5 reanalysis dataset were extracted 193 along the trajectories and provided as inputs to ADCHEM. 194

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196 **2.2** Gas and primary particle emissions

197 Emissions of gas-phase biogenic volatile organic compounds (VOCs) α -pinene, β -pinene Δ 3carene, limonene, isoprene and β -caryophyllene were modeled with a 1 - dimensional version 198 of MEGAN v2.04 (Model of Emissions of Gases and Aerosols from Nature 2.04) (Guenther 199 et al., 2006). Gas-phase emissions of marine halogens such as tribromomethane (CHBr₃), 200 dibromomethane (CH₂Br₂), iodomethane (CH₃I) were retrieved from CAMS-OCE Global 201 202 oceanic emissions (CAMS-GLOB-OCE) which are available as daily means with a spatial resolution of 0.5°x0.5° (Granier et al., 2019; Ziska et al., 2013). CAMS-GLOB-OCE also 203 204 provides gas-phase DMS emissions with the same temporal and spatial resolution (Granier et al., 2019) calculated with the air-sea flux parameterization and emission fluxes described in 205

206 (Lana et al., 2011; Nightingale et al., 2000). NH₃ emissions from seabird colonies were 207 acquired from a global emission inventory (Riddick et al., 2012). To account for additional NH_3 fluxes from the open ocean, we used an estimated sea surface equilibrium $NH_{3(g)}$ 208 209 saturation concentration of 0.5 nmol/m³ (12.2 ppt at standard temperature and pressure (STP)) which approximately correspond to a surface ocean ammonium concentration of 0.125 210 mmol/m³ (or ~3ppb, calculated based on equation 3 & 4 from Wentworth et al., 2016) at a sea 211 surface temperature of +2 °C. The sea surface temperature for the study period varied between 212 -2 °C-23°C along the trajectories. The estimated surface ocean ammonium concentrations is in 213 214 close agreement with the concentration estimated by the global ocean biogeochemical model COBALT (Stock et al., 2014) in the North Atlantic ocean, but up to a factor of ~5 higher than 215 the concentrations simulated with other ocean biogeochemical models and/or model setups 216 (Paulot et al., 2015). Therefore, we performed model sensitivity runs with a sea surface 217 equilibrium NH_{3(g)} concentration of 0.1 and 1 nmol/m³. The NH_{3(g)} equilibrium saturation 218 219 concentrations represent the ambient surface gas-phase concentration at which the air-sea flux 220 changes direction, with a net downward flux from air to sea if the ambient NH_{3(g)} exceeds the equilibrium gas concentrations and vice versa (Wentworth et al., 2016). For the anthropogenic 221 trace gas and primary particle emissions, we used the CAMS-GLOB-ANT v2.1 inventory, 222 223 with a spatial resolution of $0.1^{\circ} \times 0.1^{\circ}$ (Granier et al., 2019).

224 In this work, we used the sea surface temperature (SST) and wind speed dependent sea-spray 225 aerosol (SSA) emission parameterization by Sofiev et al. (2011), (further referred to as Sofiev11). Sofiev11 used a modified source function based on the parameterization of 226 Monahan et al. (1986) and experiments by Mårtensson et al. (2003) and SEAS campaign by 227 Clarke et al. (2006). The modified source function in Sofiev11 provides extrapolated SSA 228 emissions between size ranges of 10 nm-10 µm, with appropriate correction functions 229 employed for SST deviating from 298.15 K (Sofiev et al., 2011). Sofiev11 SSA 230 parameterization shows that with increasing temperatures, emission flux for larger particles 231 increases while the emission fluxes for smaller particles decreases (Barthel et al., 2019; 232 Sofiev et al., 2011). We performed sensitivity tests using the temperature and wind speed 233 dependent SSA parameterization by Salter et al., (2015), (further referred to as Salter15). 234 Both the Salter15 and Sofiev11 are valid between 10 nm-10 µm. Model simulation 235 comparisons between Sofiev11 and Salter15 have shown that the SSA parameterization from 236 Sofiev11 has a stronger temperature dependence and higher particle number concentration 237 emissions in the Aitken mode but result in lower PM₁₀ emissions at temperatures below 25 °C 238 (Barthel et al., 2019). 239

241 **2.2 ADCHEM**

For this study, ADCHEM was employed as a 1 - dimensional column model with 40 242 logarithmically vertical layers, extending up to ~2600m. The model time step used for 243 simulations was 30 seconds. The vertical atmospheric turbulent diffusion was solved using a 244 modified Grisogono turbulent diffusivity scheme (Jeričević et al., 2010; Öström et al., 2017; 245 Roldin et al., 2019). The ADCHEM aerosol module includes new particle formation, 246 247 Brownian coagulation, condensation and evaporation of particles, and finally the dry and wet 248 deposition of both particles and gases. The particle number size distributions were represented 249 using 100 size bins ranging from 1.07 nm to 10 µm dry diameter. Clouds were assumed to be 250 present in the model grid cells when the bulk liquid water content (LWC, extracted along the trajectory from ERA5 datasets) was greater than 0.01 g m⁻³. As a default, we used a constant 251 252 cloud supersaturation (S) of 0.25% and the particles were activated into cloud droplets, if the calculated water vapor supersaturation above the particle surface (S_c , calculated using Köhler 253 254 theory) was smaller than S. For sulfate dominated aerosol particles this corresponds to a minimum dry particle activation diameter of ~80 nm (see Figure S10). During the cloud 255 256 processing, each activated cloud droplet was assumed to take up an equal amount of liquid 257 water corresponding to the total bulk LWC divided by the calculated number concentration of activated cloud droplets. The gas-liquid droplet mass transfer and dissolution of 50 species in 258 total, including HCl, HNO₃, H₂SO₄, NH₃, HIO₃, H₂O₂, O₃, OH, BrO, NO₃, DMSO, MSIA, 259 MSA and HPMTF and their irreversible reactions in the interstitial and activated cloud 260 droplets are treated by the multi-phase chemistry mechanism (see Wollesen de Jonge et al. 261 (2021) for details). The kinetic pre-processor (KPP) (Damian et al., 2002) was used to 262 generate the multi-phase chemistry mechanism used in this study. 263

Recent observations of NPF at Ny-Ålesund have confirmed the importance of ion-mediated 264 265 H₂SO₄-NH₃ nucleation in spring with MSA and H₂SO₄ condensation contributing to the subsequent growth of particles (Beck et al., 2021; Lee et al., 2020). In this work, the 266 267 Atmosphere Cluster Dynamics Code (ACDC) (McGrath et al., 2012; Olenius et al., 2013) was coupled with ADCHEM (Roldin et al., 2019). ACDC was used to model NPF, which 268 269 involved H₂SO₄ clustering with NH₃ via both neutral and ion-induced pathways with an ionization rate of 1.7 cm⁻³s⁻¹. ACDC was used to solve the evolution of molecular H₂SO₄-NH₃ 270 271 clusters by considering the loss of clusters by collisions, evaporation or coagulation scavenging onto larger aerosol particles. At each time step, the flux of clusters (up to ~ 5 272

H₂SO₄ and 5 NH₃ each) growing out of the ACDC molecule-cluster domain represents the
NPF rate. These newly formed clusters are assigned to the corresponding smallest particle
size bin at 1.07 nm in diameter in ADCHEM, which then simulates the condensational growth
of particles and losses due to evaporation, coagulation, and wet and dry deposition.

277 For all simulations, we used model output from the closest height levels which can represent

278 Gruvebadet (model height of 73.5 m a.s.l) and Zeppelin (model height of 486.0 m a.s.l).

279 Sensitivity Tests

Alongside the main ADCHEM simulations, *BaseCase*, we performed nine complementary 280 281 scenario runs to assess the impact of different processes on the modeled aerosol concentrations. We performed simulations without aerosol in-cloud processing (Cloudoff), to 282 283 check the impact of in-cloud processing on the growth of aerosols. We investigated the effect 284 of higher PM_{10} particle emissions on the chemical composition of secondary aerosols, using 285 the sea-spray emission parameterization based on Salter et al., 2015 (SalterSSA). Simulations were conducted to assess the impact of lower and higher ammonia sources over the open 286 ocean (LowNH₃, HighNH₃). A sensitivity test without precipitation (NoPrecip) was performed 287 to test the influence of precipitation on number concentration and particle composition. Since 288 289 cloud supersaturation is critical to the activation of particles and is highly uncertain, we performed two simulations with low and high cloud supersaturation (S=0.1%, SSat=0.1 and 290 S=0.4%, Ssat=0.4) to test its impact on the modelled particle distributions. This corresponds 291 to minimum dry particle activation diameters of ~150 nm and ~60 nm respectively, for sulfate 292 rich aerosol particles (see Figure S10). We performed a simulation without new particle 293 formation (NPFoff), and finally one simulation without the dissolution and irreversible 294 aqueous chemistry of the intermediate DMS oxidation products, SO₂ and halogens 295 (woDissolution), implying that MSA, H₂SO₄ and HIO₃ is only formed in the gas-phase. Table 296 297 1. summarizes the setup for different model sensitivity test.

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Table 1. Model sensitivity tests performed alongside the main *BaseCase* simulations to test the effect of different parameters on secondary aerosol formation. These sensitivity tests focus on the role of in-cloud processing and aqueous phase chemistry, the NH₃ emissions from open ocean, SSA parameterization and cloud supersaturation. The sea surface equilibrium $NH_{3(g)}$ concentrations in ppt are provided in the brackets.

Simulation	In-cloud	NH _{3(eq)} (nmol/m ³ ,	SSA	Precipitation
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	Processing	ppt)	parameterization	
BaseCase	On	0.5 (12.2)	Sofiev11	On
SalterSSA	On	0.5 (12.2)	Salter15	On
Cloudoff	On	0.5 (12.2)	Sofiev11	On
LowNH3, HighNH3	On	0.1 (2.4) 1.0 (24)	Sofiev11	On
NoPrecip	On	0.5 (12.2)	Sofiev11	Off
SSat0.4, SSat0.1	On	0.5 (12.2)	Sofiev11	On
NPFoff	On	0.5 (12.2)	Sofiev11	On
WoDissolution	On, but no dissolution and irreversible chemistry of intermediate DMS oxidation products	0.5 (12.2)	Sofiev11	On

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305 2.3 Measurements

We utilized comprehensive measurements from the Ny-Ålesund Research station sites, 306 Zeppelin observatory (Platt et al., 2022) and Gruvebadet during the period of 1st - 25th May 307 2018. Since 2017, the atmospheric observatory at Gruvebadet, which is located about 700 m 308 southwest of Ny-Ålesund village at almost sea level (67 m.s.l), hosted Neutral cluster and Air 309 Ion Spectrometer (NAIS, Manninen et al., 2010; Mirme and Mirme, 2013) for semi-310 permanent measurements. Here we use NAIS measured number size distribution of naturally 311 charged (ions) in diameter size ranges between 0.8 nm-40 nm and neutral particles in the size 312 range of 2.5 nm-42 nm, with a temporal resolution of two seconds. 313

314 During the measurement period, a scanning mobility particle sizer (SMPS), was operated to measure particle number size distribution in the diameter size range of 10 - 470 nm at 315 Zeppelin. Concurrent SMPS data (TSI 3034, 54 channels) with diameter size ranging from 10 316 to 470 nm from Gruvebadet were also available (Dall'osto et al., 2019b; Moroni et al., 2020), 317 thus, enabling us to compare the modeled particle number size distribution with the measured 318 size distributions at both measurement stations. Daily resolution continuous aerosol samples 319 with PM₁₀ cutoff were collected at Gruvebadet using a Tecore Skypost low-volume sampler 320 (Amore et al., 2022). The detection limit for Na⁺ was 0.0001 µg m⁻³ and 0.0002 µg m⁻³ for Cl⁻, 321

NH4⁺ and SO4²⁻. Since the field blank medians at Gruvebadet were less than 1 percentile of
sampled values, the field blanks were not subtracted from the sampled values (Amore et al.,
2022).

Vertical particle number concentration profiles were obtained using UAS ALADINA (Bärfuss 325 et al., 2018; Lampert et al., 2020), which was operated during the simulation period. 326 ALADINA were operated up to a height of 850 m a.s.l., thus can be used for a potential 327 328 closure between the two different research sites of Gruvebadet and Zeppelin. ALADINA is equipped with two condensation particle counters (CPCs Model 3007, TSI Inc., St. Paul, 329 MN, USA), measuring in the size ranges of 3 nm - 2 μ m (CPC1) and ~12 nm - 2 μ m (CPC2) 330 (Lampert et al., 2020; Petäjä et al., 2020). The difference between CPC1 and CPC2 provides 331 an estimate of particle number concentrations in the size of 3 - 12 nm (N_{3-12}), which was used 332 as an indicator of NPF. Alongside the CPCs, a host of other instruments measuring 333 334 meteorological parameters were operated in unison, the description of which can be found in Bärfuss et al., (2018) and Lampert et al., (2020). 335

336 Evaluating temporal aspects of model performance

The modeled PM_{10} inorganic chemical composition was evaluated against the measured PM_{10} inorganic chemical composition using statistical estimates such as, normalized mean bias (NMB), Pearson correlation coefficient (r), root mean squared error (RMSE) and fraction of predictions within a factor of 2 of the observed values (FAC2). These tests were used to evaluate modeled values (M_i) against observation values (O_i) at both the measurement sites.

342 Pearson correlation coefficient was calculated using the formula:

343

$$r = \frac{1}{n} \sum_{i=1}^{n} \frac{\left(O_i - \bar{O}\right)}{\sigma_O} \frac{\left(M_i - \bar{M}\right)}{\sigma_M}$$
(Eq 1)

344

345 Where σ_0 and σ_M are standard deviations of the observed and modeled values, respectively.

Normalized mean bias (NMB) indicates if the predictions are over or underestimating the observed values, with the factor representing the under or over estimation. NMB was calculated using Eq. 2:

$$NMB = \frac{\sum_{i=1}^{n} (M_i - O_i)}{\sum_{i=1}^{n} O_i}$$
(Eq 2)

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351 Root mean squared error (RMSE) was calculated using Eq. 3:

$$RMSE = \sqrt{\sum_{i=1}^{n} \frac{(M_i - O_i)^2}{n}}$$
(Eq 3)

353

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FAC2 is a robust metric defined as the percentage of predictions which are within a factor of 2 of the observed values (Eq. 4):

356

$$Fac2 = 0.5 \le \frac{M}{O} \le 2.0$$
 (Eq 4)

357 3. Results and Discussion

In the following sections, we analyze and evaluate the model results against comprehensive measurements in Ny-Ålesund. In sub-section 3.1, we focus on the particle number size distributions at both sites, followed by gas-phase concentrations and PM_{10} inorganic chemical composition (sub-section 3.2) and the vertical nano-particle concentration profiles (subsection 3.3). Finally, in sub-section 3.4, we analyze the results from the model sensitivity tests.

364

365 **3.1 Particle number size distributions**

Figure 1(a) and (b) show the observed and predicted particle number size distributions at Gruvebadet for the *BaseCase* simulation. Figure 1(a) includes SMPS observations starting from 10 to 470 nm and NAIS observations for neutral particles in the range 2.5 nm-10 nm (boundary marked by the black line) since NAIS data below 2.5 nm cannot be relied upon, owing to the presence of corona generated ions (Jayaratne et al., 2017; Manninen et al., 2011, 2016).

In the *BaseCase* simulations model captures particle formation on 2nd May followed by an 372 increasing number of Aitken and accumulation mode particles during the days of 3rd - 4th 373 May, which is the result of more polluted air masses arriving at Ny-Ålesund from the 374 375 European continent (Figure S1). Similarly, the model reproduces the particle formation on the 20th of May, specifically in the size ranges 2-8 nm, but overestimates the Aitken mode and 376 accumulation mode particle concentration on the 21st of May. However, the model tends to 377 underestimate the nucleation mode particle number concentrations between 10-25 nm (N_{10-25} 378 nm) around noon, and overestimate the concentrations during the morning and evening (Figure 379

380 S2a). The model and measurements show an apparent time delay in the formation of new 381 particles larger than 10 nm. While the measurements show a peak at 11 am the simulated N_{10} - $_{25 \text{ nm}}$ shows a maximum at 3 am and 6 pm. The modeled $N_{10-25 \text{ nm}}$ maximum around 6 pm is 382 383 likely a result of the formation of new particles around noon, which grow to >10 nm in diameter during the afternoon and evening by condensation of H₂SO₄. The predicted Aitken 384 $(N_{25-100 \text{ nm}})$ and accumulation mode particle concentrations $(N_{>100 \text{ nm}})$ which form few days 385 upwind of the station are overall, in good agreement with the measurements, which show a 386 minor diurnal trend (Figure 2b-c). The measurements indicate that at Gruvebadet, $N_{10-25 \text{ nm}}$ 387 388 contributes the most significant fraction of measured total number concentrations with 45.3%, while N_{25-100 nm} and N_{100-470 nm} contribute 30.5% and 23.94% respectively. However, the 389 simulations predict greater contribution of Aitken mode (~53.85%) to total number 390 concentration, with the N_{10} - 25 nm and $N_{100-470}$ nm accounting for ~ 36.58% and 9.57% 391 392 respectively (Figure S2).

Figure 2 shows the measured size distribution in panel (a) and simulated size distribution in 393 394 panel (b) for Zeppelin. At Zeppelin, the model overestimates the number concentration in nucleation and Aitken modes (also cf. Figure S3, supplementary). The particle number size 395 396 distribution measurements at Zeppelin indicate that the relative contribution of the three modes (nucleation, Aitken and accumulation) varies to some extent when compared to 397 398 Gruvebadet. Measurements show that at Zeppelin, $N_{10-25 \text{ nm}}$ contributes ~33.46%, $N_{25-100 \text{ nm}}$ 399 46.43% and $N_{>100 \text{ nm}}$ 20.11% to the total particle number concentrations. The model predicts lower relative contribution of $N_{10-25 \text{ nm}}$ (26.94%), and a greater contribution of $N_{25-100 \text{ nm}}$ 400 (63.44%) to the total simulated particle number concentrations. The diurnal trends at Zeppelin 401 agree well with earlier measurements conducted at Zeppelin in spring by Ström et al. (2009). 402 403 Additionally, the measured diurnal pattern at Zeppelin varies in comparison to Gruvebadet. At Zeppelin, the $N_{10-25 \text{ nm}}$ concentrations peak in the afternoon and evening. The modeled 404

 $N_{10-25 \text{ nm}}$ shows only a weak diurnal trend. It should be noted that the measurements show a time delay of around 3 hours in the peak $N_{10-25 \text{ nm}}$ concentrations at the two sites (Figure S2 and S3). This is possibly be a result of vertical mixing and dilution effects modulating the observed particle number concentrations at sites situated at different altitudes, similar to observation made at Zeppelin and Corbel by Ström et al. (2009).

ADCHEM considers the formation of new particles via both the ion-mediated and neutral
H₂SO₄ -NH₃ clustering pathways. Beck et al (2021) observed dominant contribution of

413 Ålesund, with HIO₃ playing a small role in the initial particle formation. However, the 414 discrepancy in the modeled and observed diurnal trends of $N_{10-25 \text{ nm}}$ could indicate that there are other sources or vapors that might potentially contribute to the particle formation. Other 415 416 possible NPF mechanism may involve amines (Olenius et al., 2013) and pure biogenic highly oxidized molecule (HOM) (neutral and ion induced) nucleation (Kirkby et al., 2016). We 417 speculate that the exclusion of these other mechanisms (HIO₃, H₂SO₄-amines and HOM 418 driven particle formation) might result in the discrepancies in the modeled and observed 419 particle number concentration diurnal trends. HIO₃ induced particle formation could, e.g. play 420 an important role if the air masses upwind of Ny-Ålesund traverse over the sea-ice covered 421 regions (Baccarini et al., 2020; Beck et al., 2021). 422





Zeppelin (Simulation: BaseCase)



Figure 2. Particle size distribution at Zeppelin. The panel (a) shows the measurement data for the period 1-25th May from SMPS, the panel (b) provides the simulated particle size distribution and panel (c) shows the total measured and simulated number concentrations for the *BaseCase* simulations. The abscissa and ordinates are similar to Figure 1.

435

Figure 3 presents the median particle number size distribution for the BaseCase simulation at 436 both Zeppelin and Gruvebadet, with respective 25th and 75th percentiles, for the entire selected 437 438 period. At Gruvebadet, the modeled and measured median particle number size distributions 439 are in reasonable agreement for both Aitken and accumulation mode. However, the model over predicts the median Aitken mode concentrations at Zeppelin by a factor ~ 5.5 . The 440 modeled Aitken mode peak at both measurement sites is ~50 nm, while the measured Aitken 441 mode peak is ~30 nm. Though the modeled accumulation mode peak is at a larger size (~150 442 nm), compared to the measured accumulation mode peak (~110 nm), the predicted value is 443 slightly lower than the monthly averaged accumulation mode peak location measured at 444 Zeppelin in earlier studies (~160-170 nm, Dall'Osto et al., 2019). 445

The discrepancy between the modeled and measured particle concentrations at Zeppelin can be caused by the underlying complexity of modeling the boundary layer dynamics at an elevated site, such as Zeppelin. The vertical mixing of aerosols along the up-slope or downslope of a mountain site is difficult, if not impossible for a 1- dimensional column model, since it is unable to capture the topographical influence on locally varying wind speeds or latent and sensible heat fluxes (Mikkola, 2020; Wainwright et al., 2012).



Figure 3. Median particle number size distribution at Gruvebadet and Zeppelin for both modeled (*BaseCase* simulations) and measured values. The shaded areas indicate the 25th and 75th percentiles for both model and measured median particle number size distribution. At Zeppelin, the simulated median size distribution is calculated for periods only when SMPS data were available.

457

458 Another detectable feature in the median particle number size distribution is the diameter of 459 the Hoppel minimum (Hoppel et al., 1985, 1986), and the role of in-cloud processing in 460 forming this minimum. A Hoppel minimum is often observed in marine air masses (Fossum

461 et al., 2018; Tunved et al., 2013; Zheng et al., 2018) and is attributed to in-cloud processing of 462 aerosols, with chemical processing (e.g., sulfate production via oxidation of dissolved SO₂) (Feingold and Kreidenweis, 2000; Hoppel et al., 1986), and coalescence of droplets playing a 463 464 key role (Flossmann and Wobrock, 2019; Hoppel et al., 1986; Hoppel and Frick, 1990; Noble and Hudson, 2013). It has been estimated that, on average, aerosols take part in about 10 non-465 precipitating cloud cycles before it is removed from the atmosphere by wet scavenging 466 (Hoose et al., 2008; Hoppel et al., 1986; Rosenfeld et al., 2014). These non-precipitating 467 cloud cycles facilitate the formation of hygroscopic accumulation mode particles, with low 468 469 critical supersaturation (S_c) that readily activates to cloud droplets during subsequent cloud cycles, thus growing to larger sizes. This is because the activated particles undergo chemical 470 processing, gas-to-particle conversions, coalescence and coagulation with other interstitial 471 particles. Upon evaporation of water, the emerging dry particles have a larger size and lower 472 S_c , leading to a minimum being formed between the un-activated and activated cloud droplets 473 474 (Herenz et al., 2018; Hudson et al., 2015; Noble and Hudson, 2013). The diameter at which 475 the Hoppel minimum is observed varies depending on the cloud supersaturation and particle composition (Hoppel et al., 1986; Hudson et al., 2015), with Hoppel minima sizes observed in 476 ranges from 60 nm at Zeppelin Ny-Ålesund to around 90 nm at Tuktoyatuk, Canada (Herenz 477 et al., 2018; Tunved et al., 2013). 478

479 The median particle number size distribution in Figure 3 shows that at both stations, the 480 measured Hoppel minima is around ~ 60 nm, while the simulated Hoppel minima are around 481 the size of ~ 100 nm at both sites. This difference in location of Hoppel minima can be attributed to the assumed value of S=0.25% in the model. The value of S used in the model 482 483 lies in the range of typical marine stratocumulus clouds, which can vary between 0.1 - 1%(Fossum et al., 2018; Quinn et al., 2017). With S=0.25% the sulfate dominating particles in 484 the Arctic marine boundary layer will be activated into cloud droplets if their diameter is 485 greater than ~ 85 nm in diameter (Fig. S10). 486

487 3.2 Gas and particle-phase chemical composition of important precursors

488 Figure 4 shows the range of simulated gas-phase concentrations of DMS oxidation products

489 H₂SO₄, MSA and HIO₃ for the entire period at height levels representing Gruvebadet. The

490 mean measurement values (red dots) represent gas-phase concentrations for the same species

491 from an earlier 2017 May campaign performed at Gruvebadet by Beck et al., (2021).

492 Measurements of H₂SO₄ at Gruvebadet from May 2017 indicate monthly mean concentrations

493 around $\sim 10^6 \, \text{\# cm}^{-3}$ (Beck et al., 2021). The modeled H₂SO₄ concentrations at Gruvebadet are

3 x 10^6 # cm⁻³, implying a reasonably good model performance in predicting gaseous 494 precursor concentrations. The simulated gas concentrations of MSA $(10^5-10^8 \text{ # cm}^{-3})$ also 495 agrees well with the measurements made at Gruvebadet in May 2017 by (Beck et al., 2021), 496 497 wherein they measured daily averages of MSA gas concentrations in the order of $10^7 \, \text{\# cm}^{-3}$. The low modeled values of MSA and DMSO gas phase concentrations at the height 498 representing Zeppelin (c.f. Figure S4 supplementary e.g. between 15/05 - 17/05) coincide 499 with the period where the planetary boundary layer height (PBLH) is below the altitude of 500 Zeppelin station (cf. Figure S5 supplementary). Overall, we can conclude that the modeled 501 502 precursor gas concentrations at the two measurement sites are, in general, good agreement with earlier measurements at the two sites. 503



Figure 4. Gas-phase concentrations at Gruvebadet for the *BaseCase* simulations. The red dots indicate the mean measured values from an earlier 2017 May campaign conducted at Gruvebadet by Beck et al., (2021).

Figure 5 (a) shows the simulated median mass size distribution of compounds Cl⁻, Na⁺, MSA, SO₄²⁻, NH₄⁺, and NO₃⁻ for the *BaseCase* runs in the lowest model layer. Figure 5(a) indicates that the nucleation mode particles are composed mainly of SO₄²⁻ and NH₄⁺, while MSA, Cl⁻

and Na⁺ dominate PM for larger particles. The observed and modeled high MSA_(g) 510 concentrations in comparison to H₂SO_{4(g)} at Ny-Ålesund is not reflected in the respective 511 vapor contribution to the nano-particle growth. This is because, in contrast to H₂SO₄, MSA is 512 513 not a non-volatile condensable compound. The gas-to-particle partitioning of MSA requires co-condensation and dissolution of (NH₃) (Hodshire et al., 2019) or the existence of cations 514 such as Na⁺ which decreases the particle acidity ([H⁺]). Figure 5(b) shows the relative mass 515 fraction of the above-mentioned compounds to PM at different sizes. SO42- and NH4+ 516 dominate the mass for particles in the nucleation and Aitken mode. SO_4^{2-} contributes ~74% 517 and ~71% to nucleation and Aitken mode PM, with its contribution decreasing for 518 accumulation (100 nm-1 µm) and coarse (>1 µm) mode PM (~6% and 3.36% respectively) 519 (Table S1). NH₄⁺ contribution follows a similar trend, as SO_4^{2-} , with 12.34% and 6.95% 520 contribution to nucleation and Aitken mode PM, but insignificant for accumulation and coarse 521 mode PM (Table S1). The loss of primary sea spray aerosols due to wet scavenging promoted 522 523 the growth of secondary aerosol particles in the nucleation and Aitken mode by $NH_{4^{+}}$ and SO₄²⁻ as seen in Figure 5 (b). Na⁺ (~32.9%), Cl⁻ (~39.5%) and MSA (20.45%) are the 524 dominant contributors to accumulation and coarse mode PM. In the BaseCase simulations, 525 gas-phase SO₂ dissolves in the cloud droplets, and is oxidized by H_2O_2 into SO₄²⁻ (Wollesen 526 de Jonge et al., 2021). Previous modeling studies have shown that a very small fraction of 527 528 MSA is formed in the gas phase. Instead, most MSA is formed via ozonolysis of MSIA in the aqueous phase (Hoffmann et al., 2016; Wollesen de Jonge et al., 2021). It should be noted that 529 HIO₃ and NO₃⁻ have an insignificant contribution to total PM₁₀, amounting to ~0.05% and 530 0.17% respectively. 531





Figure 5. Simulated median mass size distribution for *BaseCase* simulations. The upper panel (a) shows the median mass size distribution for compounds Cl⁻, Na⁺, MSA, SO₄²⁻, NH₄⁺, HIO₃ and NO₃⁻ for the entire size distribution ranging from 1.07 nm-10 μ m. The lower panel (b) shows the relative mass fractions or contribution of compounds Cl⁻, Na⁺, MSA, SO₄²⁻, NH₄⁺, HIO₃ and NO₃⁻ to total non-refractory PM at different sizes.

Figure 6 compares the daily PM_{10} filter measurements to the modeled values at both measurement stations. The model prediction of PM_{10} Cl⁻, Na⁺, SO4²⁻ and NH4⁺, was evaluated using statistical metrics such as NMB, FAC2, correlation coefficient (*r*) and RMSE (Table 3). Though the model does well in simulating the trends of PM_{10} SO4²⁻, Na⁺ and Cl⁻ at Zeppelin (*r* values of 0.35, 0.51 and 0.6 respectively), it is unable to predict the NH4⁺ trends accurately (*r* = -0.08).



20

Zeppelin. In summary, the model tends to overpredict $PM_{10} Na^+$, Cl⁻ and SO_4^{2-} concentrations, 551 552 but on the other hand, does reasonably well in predicting the daily measured trends. Additionally, the modeled PM_{10} Cl⁻/Na⁺ molar ratio at Gruvebadet and Zeppelin is ~0.79 and 553 554 ~0.95, respectively. This is much higher than the observed PM_{10} Cl⁻/Na⁺ molar ratio at both sites (~ 0.39). One likely reason for this is the overestimated sea spray aerosol emissions. The 555 PM_{10} Cl⁻/Na⁺ molar ratios give a measure of the acidic nature of aerosol, since increased 556 condensation of strong acid MSA and H₂SO₄ increases acidity of aerosols thereby causing 557 loss of Cl⁻ (dechlorination) as HCl (Ayers et al., 1999; Frey et al., 2020). Thus, increased 558 availability of H₂SO₄ and MSA in particle phase in Aitken mode particles results in acid-559 induced Cl⁻ loss from sea-spray particles. 560



Figure 6. PM_{10} comparison of *BaseCase* simulations with daily filter samples from Gruvebadet and Zeppelin for the entire modeled period. Panel **(a)** shows PM_{10} NH_{4^+} , **(b)** shows PM_{10} Cl^- , **(c)** shows PM_{10} Na^+ and **(d)** shows PM_{10} $SO_{4^{2^-}}$ filter samples. The dotted lines in each panel indicate measurement values, and the solid line denotes simulated values. The ordinate is plotted in log scale to better visualize the low values.

567

561

Table 3: Evaluation of modeled PM₁₀ values at both sites of Gruvebadet (G) and Zeppelin (Z)
for the four particle-phase species Cl⁻, Na⁺, SO4²⁻ and NH4⁺.

	bias factor (NMB)	<mark>coefficient (r)</mark>		
NH4 ⁺	-0.88 ^G , -0.76 ^Z	$0.34^{\rm G}, -0.08^{\rm Z}$	$0.09^{\rm G}, 0.02^{\rm Z}$	$0.04^{\rm G}, 0.2^{\rm Z}$
SO ₄ ²⁻	-0.28 ^G , 1.96 ^Z	0.33 ^G , 0.35 ^Z	$0.27^{\rm G}, 0.26^{\rm Z}$	$0.6^{\rm G}, 0.24^{\rm Z}$
Na ⁺	1.81 ^G , 0.36 ^Z	0.29 ^G , 0.51 ^Z	1.67 ^G , 0.55 ^Z	$0.4^{\rm G}, 0.48^{\rm Z}$
Cl ⁻	$1.05^{\rm G}, 0.39^{\rm Z}$	$0.24^{\rm G}, 0.60^{\rm Z}$	2.08 ^G ,0.74 ^Z	$0.24^{\rm G}, 0.44^{\rm Z}$

570

571 **3.3 Vertical profiles of ultra-fine particle**

572 Figure 7 (a) shows the measured vertical $N_{3-12 \text{ nm}}$ concentrations from CPC onboard the UAS for four measurement periods overlayed onto simulated vertical profiles. Figure 7 (b) and (c) 573 show the mean vertical profiles for $N_{3-12 \text{ nm}}$ and $N_{>12 \text{ nm}}$ for both the *BaseCase* simulation and 574 UAS measurements for the entire selected period. The model underestimates the measured N_{3-} 575 $_{12 \text{ nm}}$ and $N_{>12 \text{ nm}}$ vertical particle number concentrations below 200 m a.s.l. The NMB for N_{3-12} 576 $_{nm}$ and $N_{>12 nm}$ is -0.28 and -0.14, respectively, implying that the model underestimates the 577 578 particle number concentrations. Both the modeled and measured mean particle number concentrations for N_{3-12nm} and $N_{>12nm}$ are in good agreement between the heights of 200-600 579 580 m a.s.l. The lower calculated concentrations of modeled mean particle number concentrations above 600 m a.s.l is most likely affected by higher turbulence in the transition zone from the 581 582 boundary layer to the free troposphere, which might cause a large mixing of aerosol particles. 583 It should be noted that, at Gruvebadet, the mean SMPS particle number concentrations are in good agreement with the modeled particle number concentrations. However, at the altitude of 584 585 the Zeppelin station, both the model and UAS measurements of $N_{>12nm}$ are substantially higher (factor of 4) than the mean particle number concentrations measured with the SMPS at 586 587 Zeppelin. This finding further strengthens the conclusion that the complex orography at Ny-588 Ålesund highly affects the variability in the vertical scale, which may cause this discrepancy 589 in the observed and modeled particle number concentrations at Zeppelin (see section 3.1). The UAS measurements were carried out at the airport on Ny-Ålesund (and the UAS was flown 590 591 around Ny-Ålesund) where the boundary layer measurements, like the model, most likely resemble the general Arctic marine boundary layer conditions. Figure S9 shows the influence 592 of different sensitivity simulations on the modeled vertical particle number concentrations. 593 The large spread in the modeled vertical particle number concentrations in Figure S9, 594 highlights the importance of constraining uncertain parameters such as cloud supersaturation 595 and NH₃ gas emissions, to better simulate secondary aerosol formation in marine polar 596 regions. 597





Figure 7. Comparison of vertical profiles of measured particle number concentration and 599 BaseCase simulation. Panel (a) shows measured particle number concentration between 3-12 600 nm (N_{3-12 nm}, triangles), from CPC onboard the UAS during 4 periods 05/01- 05/02, 05/14 -601 05/16, 05/19- 05/22, and 05/23- 05/24 (in legend) overlayed onto the simulated N_{3-12nm} for 602 603 the same periods. Panel (b) shows the simulated and measured mean N_{3-12nm} and panel (c) show the $N_{>12nm}$ for the selected period. Additionally, Panel (c) also shows the mean SMPS 604 605 particle concentrations at both Gruvebadet and Zeppelin. The horizontal bars for the mean 606 SMPS values represent the standard deviation.

607

608 3.4 Sensitivity Tests

In this section, we will discuss the results from the sensitivity tests that we performed to
complement the main *BaseCase* simulations. The settings of different sensitivity tests are
described in Table 1.

612 Median particle size distribution for sensitivity tests

613 The sensitivity study *Cloudoff* was performed to test how in-cloud processing affects the 614 formation of larger particles, especially the accumulation mode (Figure 8). In the Cloudoff 615 test, in-cloud processing was switched off in the model and the RH was set to just below 616 supersaturation (99.9999%) in the model grid cell where clouds (RH=100.5%) exists in the BaseCase runs. The aim of the Cloudoff simulation was to investigate if the model can 617 capture the observed accumulation mode without aerosol cloud processing. It is clear from 618 Figure 8 (a) and (b) that in *Cloudoff* simulations, the median size distribution lacks the 619 accumulation mode and Hoppel minima and has a higher Aitken mode particle concentration 620 621 compared to either BaseCase or the measured median size distribution. This further emphasizes the importance of in-cloud processing in activation of particles to CCN sizes and 622 their growth to larger sizes. However, it should be noted, that other processes such as 623 624 Brownian scavenging by larger cloud droplets could result in the shift in particles from the Aitken mode to accumulation mode (as seen in median measured size distribution, Figure 8, 625 626 Noble and Hudson, (2019)). Another noteworthy point in *Cloudoff* simulations is the larger 627 number concentration of particles <10 nm compared to other cases. One plausible reason is the lack of activated cloud droplets, since the large surface areas of activated droplets are 628 629 efficient at Brownian scavenging of smaller particles (Hudson et al., 2015). Likewise, the 630 median particle number size distribution from the sensitivity tests with lower cloud 631 supersaturation (S) of 0.1% SSat=0.1, reduces the accumulation mode particles, since there are fewer particles with $S_c < S$ available for activation. Increasing S to 0.4% increases 632 accumulation mode particles, since more particles with $S_c < S$ are activated to cloud droplets 633 (Aitken mode concentration decreases with respect to BaseCase simulations, since more 634 635 smaller particles are activated into cloud droplets). Therefore, simulated results show that increasing the cloud supersaturation results in a higher number of smaller particles being 636 637 activated into cloud droplets and shifts the simulated Hoppel minima close to the measured 638 sizes. Figure S6, in supplementary shows median particle size distribution for all sensitivity

639 tests.

The *SalterSSA* sensitivity test underestimates both the Aitken and accumulation mode concentrations at Gruvebadet (Figure S5, supplementary). The Salter sea-spray parameterization produces ~ 2 magnitudes fewer Aitken mode particles compared to Sofiev et. al, 2011, while the coarse mode particle emissions using *SalterSSA* parameterization are higher than Sofiev et. al 2011. This can cause MSA, H₂SO₄ and NH₃ to partition onto coarse mode particles rather than contributing to NPF and growth of the nucleation and Aitken mode particles, which substantially lowers the Aitken and accumulation mode number concentrations. The *NPFoff* simulation from Figure S6 shows lower Aitken mode
concentrations, implying that the main contributor to Aitken mode particle number
concentrations are the secondary aerosols rather than the primary sea-salt particles.

650



Figure 8: Median size distribution at Gruvebadet (panel (a)) and Zeppelin (panel (b)) for all
the sensitivity tests *Cloudoff*, *SSat=0.4*, and *SSat=0.1* (colored dashed lines) including *BaseCase* (blue solid line) and observations (black solid line).

Another parameter of uncertainty is the concentration of NH₃ in the marine atmosphere. The *LowNH*₃ simulations, as expected, result in lower Aitken mode particles, whereas *HighNH*₃ simulations show an overprediction of Aitken mode concentrations (Figure S6, supplementary). This underlines the necessity of constraining ocean and marine emissions of NH₃ to better predict the aerosol particle formation in marine polar environments.

659 Particle phase comparison for sensitivity tests

Figure 9 shows the contribution of constituent compounds to PM at different particle sizes with respect to the *BaseCase* simulation. The overall mean contribution of SO_4^{2-} and MSA to total PM₁₀ decreased by ~8% and 11% respectively, in *Cloudoff* runs compared to the *BaseCase* simulations. It is expected that in non-cloud conditions there is a reduction in SO_4^{2-} and MSA PM contribution because of the reduced partitioning of gaseous SO_2 to the cloud droplets (for PM SO_4^{2-} formation) and inhibition of MSIA ozonolysis in the cloud droplets

(leading to PM MSA formation) (Chen et al., 2018; Hoffmann et al., 2016; Wollesen de Jonge 666 et al., 2021). This is observed for accumulation mode particles between size ranges of 100 nm 667 to1 µm which is characterized by lower SO4²⁻ and MSA PM. On the other hand, PM SO4²⁻ and 668 669 MSA increase for coarse mode particles (> 1 µm). Without cloud droplet activation the deliquescent sea spray coarse mode particles become a major liquid water reservoir where 670 MSIA and to a lesser extent SO₂ are dissolved and oxidized into MSA and SO₄²⁻, which partly 671 explains the increase in PM MSA and SO_4^{2-} for sizes > 1 µm. The results from *Cloudoff* 672 simulation agrees with the findings from Wollesen de Jonge et al., 2021, who found that MSA 673 was almost exclusively formed in the aqueous phase via MSIA ozonolysis in cloud droplets 674 and deliquescent particles during and in between in-cloud periods. PM SO42- in Cloudoff runs 675 is mainly driven via condensation of H2SO4, since an increase in SO2 gas-phase 676 677 concentrations (~42% with respect to BaseCase) promoted gas-phase H₂SO₄ production (increase of ~44% with respect to *BaseCase*), and therefore H_2SO_4 derived PM SO_4^{2-} . 678

In the *woDissolution* simulation, all the PM MSA and SO₄²⁻ are a result of the condensation of 679 MSA_(g) and H₂SO_{4(g)}, since irreversible aqueous-phase chemistry is switched off. The overall 680 contribution of PM SO₄²⁻ to the total PM₁₀ increases by ~12% relative to the *BaseCase* run, 681 while on the other hand, the contribution of PM MSA decreases by ~87% (relative to 682 BaseCase). The lower PM₁₀ MSA in *woDissolution* simulation emphasizes the importance of 683 684 aqueous-phase formation of MSA to the growth of particles. The effect of precipitation on modeled PM (NoPrecip) indicates an increase in PM Na⁺ and Cl⁻ of ~112% and 119% 685 686 respectively, as compared to BaseCase (Figure S8). This is because of the decrease in the wet deposition of aerosol and sea-spray particles by rain events and below cloud scavenging. The 687 688 consequence of neglecting precipitation results in increased condensation sink for H_2SO_4 and NH₃ (increase of 62% and 22% in PM SO₄²⁻, NH₄⁺ respectively), but since sea-spray aerosols 689 are not scavenged by the wet removal process, the overall fractional contribution to PM by 690 SO₄²⁻, NH₄⁺ and MSA is lower relative to *BaseCase* runs. 691

692 *SalterSSA* simulation results in higher PM Cl⁻ and Na⁺ (470% and 371% increase 693 respectively) compared to *BaseCase* runs. This is because Salter15 SSA parameterization 694 produces larger mass emission fluxes in size ranges > 1 μ m compared to Sofiev11 SSA 695 parameterization (Barthel et al., 2019). Additionally, there is an increase of ~19% in PM 696 MSA, largely due to formation of MSA in larger deliquescent coarse mode particles.

Contribution to PM relative to BaseCase



Figure 9: Contribution of constituent compounds, namely, MSA (panel (a)), SO_4^{2-} (panel (b)), Cl⁻ (panel(c)), Na⁺ (panel (d)) and NH₄⁺ (panel (e)) to PM with respect to *BaseCase* (the black dotted line).

701 4. Summary and conclusions

In this work, we attempt to simulate secondary aerosol formation at remote Arctic sites of Gruvebadat and Zeppelin, Ny-Ålesund, during the period of 1st - 25th of May 2018. We used the 1-dimensional column model ADCHEM which was run along FLEXPART generated Lagrangian trajectories. Since the air mass spend most of their time over the open ocean upwind of Ny-Ålesund, we use a comprehensive multi-phase DMS chemistry scheme coupled with MCMv3.3.1 and PRAM.

In the model, new particles are formed via ion-mediated H_2SO_4 -NH₃ nucleation, with the initial particle growth mainly driven by condensation of H_2SO_4 , while the secondary PM₁₀ MSA and SO_4^{2-} contribution was mainly formed by oxidation of MSIA and SO_2 in the aqueous phase. At Gruvebadet, the modeled median particle number size distribution agrees reasonably well with the measurements, however, at Zeppelin, the simulated Aitken mode 713 median concentration is overestimated by a factor of 5.5. This relatively large discrepancy in 714 modeled and measured particle size distributions at Zeppelin, and likewise the large difference between the measured particle number size distributions at Gruvebadet and 715 716 Zeppelin, can to a large extent be explained by the orographic effects at Zeppelin which distorts the atmospheric boundary layer dynamics. Thus, while the model generally is able to 717 capture the particle number size distribution dynamics in the marine boundary layer, as 718 measured at the near sea level Gruvebadet site, it generally cannot capture the observations at 719 720 the mountain station of Zeppelin, which often lies above the boundary layer and may experience free tropospheric conditions. This is also supported by the fact that $N_{>12}$ nm 721 concentrations measured with the UAS above Ny-Ålesund airport agrees well with the 722 modeled particle number concentrations, at the same altitude as Zeppelin. However, both the 723 model and UAS $N_{>12nm}$ concentrations is a factor of 4 higher than the $N_{>12nm}$ observation at 724 725 Zeppelin.

Both the measured and modeled particle size distribution, at both stations, show a distinct 726 727 Hoppel minima, which can be explained by in-cloud processing. Model sensitivity runs with varying cloud supersaturation indicate that a cloud supersaturation of 0.4% or higher is 728 729 required for the model to capture the observed Hoppel minima. Furthermore, model sensitivity runs show that the Aitken mode particle number concentrations are dominated by 730 731 contribution of secondary aerosols rather than primary emissions. The modeled PM₁₀ Cl⁻ and Na⁺ is positively correlated when compared to PM₁₀ filter samples. The main driver for 732 733 secondary aerosol particle growth is the formation of MSA via aqueous phase ozonolysis of the DMS oxidation product MSIA. This demonstrates the importance of multi-phase DMS 734 chemistry in capturing the size resolved secondary aerosol growth in marine polar regions. 735

736 The sensitivity studies indicate that it is important to limit the uncertainties in parameters such 737 as cloud supersaturation and NH₃ emissions over open oceans to get a better constraint on 738 secondary aerosol formation and its subsequent climatic effects. This work was a first attempt 739 to simulate new particle and secondary aerosol formation in marine polar regions using a 740 process based chemistry transport model that includes a comprehensive multi-phase DMS and halogen chemistry mechanism, detailed gas-molecular cluster and aerosol dynamics. In future 741 742 studies, we aim to implement ADCHEM for extended studies in polar marine and remote 743 continental regions where different atmospheric constituents such as HIO₃, terpenes and 744 amines drive secondary aerosol formation.

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766 Author Contributions

767 CX, PR, MBoy, BA and BW planned and designed the study. PR, RWdJ and CX developed 768 and setup the ADCHEM model. CX, MB and VV performed the FLEXPART model 769 simulations. BA, BW, RoTh, and RT provided the measurement data. Resources were 770 provided by PR and MBoy. CX, PR and MBoy wrote the original draft, which included 771 visualizations made by CX and PR. All other authors discussed the results and contributed to 772 the final manuscript.

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774 Competing interests

775 The authors declare that they do have no conflict of interest.

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