



1	Oxygenated VOCs as significant but varied contributors
2	to VOC emissions from vehicles
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23 Abstract:

Vehicular emission is an important source for volatile organic compounds (VOCs) in 24 urban and downwind regions. In this study, we conducted a chassis dynamometer study 25 to investigate VOC emissions from vehicles using gasoline, diesel, and liquefied 26 petroleum gas (LPG) as fuel. Time-resolved VOC emissions from vehicles are 27 chemically characterized by a proton-transfer-reaction time-of-flight mass 28 spectrometry (PTR-ToF-MS) with high frequency. Our results show that emission 29 factors of VOCs generally decrease with the improvement of emission standard for 30 gasoline vehicles, whereas variations of emission factors for diesel vehicles with 31 emission standards are more diverse. Mass spectra analysis of PTR-ToF-MS suggest 32 that cold start significantly influence VOCs emission of gasoline vehicles, while the 33 34 influences are less important for diesel vehicles. Large differences of VOC emissions 35 between gasoline and diesel vehicles are observed with emission factors of most VOC 36 species from diesel vehicles were higher than gasoline vehicles, especially for most oxygenated volatile organic compounds (OVOCs) and heavier aromatics. These results 37 indicate quantification of heavier species by PTR-ToF-MS may be important in 38 characterization of vehicular exhausts. Our results suggest that VOC pairs (e.g. C₁₄ 39 aromatics/toluene ratio) could potentially provide good indicators for distinguishing 40 emissions from gasoline and diesel vehicles. The fractions of OVOCs in total VOC 41 emissions are determined by combining measurements of hydrocarbons from canisters 42 and online observations of PTR-ToF-MS. We show that OVOCs contribute $7.7\% \pm 6.2\%$ 43 of gasoline vehicles of the total VOC emissions, while the fractions are significantly 44 higher for diesel vehicles (40-77%), highlighting the importance to detect these OVOC 45 species in diesel emissions. Our study demonstrated that the large number of OVOC 46 species measured by PTR-ToF-MS are important in characterization of VOC emissions 47 from vehicles. 48





50 1. Introduction

Volatile organic compounds (VOCs) are important trace components in the 51 troposphere, as important precursors of ground-level ozone (Shao et al., 2009) and 52 53 secondary organic aerosol (SOA) (Seinfeld and Pandis, 2006;Kansal, 2009;Ziemann and Atkinson, 2012). As the result, it is particularly important to identify emission 54 sources of VOCs in the atmosphere. Vehicular emission is an important source of VOCs 55 in cities around the world (Liu et al., 2008; Parrish et al., 2009), contributing 56 57 approximately 25% to total VOC emissions in China (Ou et al., 2015; Wu et al., 2016;Sun et al., 2018). In order to control atmospheric pollution in urban and 58 surrounding regions, it is necessary to understand source profiles and emission 59 characteristics of VOCs from vehicles. 60

Emissions of VOCs from vehicles have been investigated extensively from 61 tunnel studies (Cui et al., 2018;Zhang et al., 2018;Song et al., 2020), on-road mobile 62 measurements (Li et al., 2017), and chassis dynamometer tests (Guo et al., 2011; Wang 63 et al., 2013; Yang et al., 2018). Previous studies demonstrated that fuel types of vehicles 64 strongly impact VOC emissions. Aromatics along with other hydrocarbons are known 65 as compounds with high emissions in exhausts of gasoline vehicles (Wang et al., 66 2013;Ly et al., 2020). Some carbonyl compounds contribute significantly to emissions 67 of diesel vehicles, at fractions much higher than gasoline vehicles (Tsai et al., 2012;Qiao 68 et al., 2012; Yao et al., 2015; Mo et al., 2016). Moreover, there are still a large number 69 70 of unidentifiable compounds in diesel vehicles (May et al., 2014). Furthermore, VOC emissions significantly decreased for stricter emission standards (Cao et al., 2016). In 71 order to reduce emissions of most primary pollutants, more stringent emission standards 72 73 and after-treatment devices have been implemented. The emission standard of China 74 VI has already been implemented in July of 2019 in a few key cities in China and in 75 July of 2021 nationwide. The emission limits for various air pollutants emitted by 76 vehicles are significantly lower under the China VI emission standard (Wu et al., 2017). With the continuous development of engine and exhaust after-treatment technologies, 77 emission characteristics of VOCs from vehicles may change and need to be frequently 78





79 updated.

Oxygenated volatile organic compounds (OVOCs) were found to be an important 80 group in vehicle exhausts, accounting for more than 50% of the total VOC emissions 81 for diesel vehicles (Schauer et al., 1999; Yao et al., 2015; Mo et al., 2016). Traditionally, 82 VOCs are collected in the canister or Tedlar bags, and then analyzed by gas 83 chromatography-mass spectrometer/flame ionization detector (GC-MS/FID), mainly 84 reporting emissions of hydrocarbons (Wang et al., 2017;Qi et al., 2019). Previous work 85 usually collected 2,4-dinitrophenyhydrazine (DNPH) cartridges and analyzed using 86 high-performance liquid chromatography (HPLC) for carbonyls (aldehydes and 87 ketones), which are both time-consuming and prone to contaminations (Mo et al., 88 2016;Han et al., 2019). 89

The large variability of VOC emissions under different engine activities or 90 driving conditions require characterization of vehicular emissions at higher time 91 92 resolution. Proton-transfer-reaction mass spectrometry (PTR-MS) has been used in a number of studies for measurements of vehicle emissions. VOCs from vehicle exhausts 93 under various driving and operational modes were measured by PTR-MS onboard a 94 mobile laboratory (Zavala et al., 2006;Zavala et al., 2009). Drozd et al. (2016) used a 95 PTR-MS to emphasize the importance of cold start for vehicles, concluding that VOC 96 97 emissions during cold start were equal to a 200 miles distance of driving during hot stabilized condition. Proton-transfer-reaction time-of-flight mass spectrometry (PTR-98 ToF-MS) can provide more powerful detection of various VOCs, thanks to the 99 measurements of whole mass spectra and high mass resolution (Cappellin et al., 100 101 2012; Yuan et al., 2017). More OVOC species could be quantified from the measured mass spectra based on parameterization methods for sensitivity of instrument 102 (Sekimoto et al., 2017; Wu et al., 2020). 103

In this study, we applied a PTR-ToF-MS along with a suite of other instruments to measure VOCs emitted from gasoline, diesel, and liquefied petroleum gas (LPG) vehicles. We investigated emission factors from different fuel types and emission standards for representative VOC species exhausted from these vehicles. We used the dataset to analyze contributions of various VOC groups to total VOC emissions in





109 different types of vehicles.

110 2. Materials and methods

111 2.1 Tested vehicles and the chassis dynamometer study methods

In this study, we conducted chassis dynamometer measurements to investigate 112 VOC emissions from vehicles using gasoline, diesel, LPG as fuel. All gasoline vehicles 113 114 are light-duty-gasoline-vehicle (LDGV) with the emission standards from China I to 115 China VI, whereas diesel vehicles can be classified into light-duty-diesel-truck (LDDT), middle-duty-diesel-truck (MDDT), heavy-duty-diesel-truck (HDDT), and bus 116 117 associated with emission standards of China III to China V. In addition, the test vehicles using LPG are all taxis, which are under mandatory scrappage after 8 years of driving 118 119 in China; as a result only China IV and China V for LPG vehicles were tested. Among the 38 vehicles we tested, a fraction of vehicles was measured several times, with a total 120 of 62 experiments measured. The detailed information for test vehicles is summarized 121 in Table S1 and Table S2. 122

123 The short transient driving cycle (GB 18285-2018, Figure S1a), as one of the 124 widely used test methods for vehicle emissions in China (Li et al., 2012; Wang et al., 2013), was used for measurements of gasoline vehicles and LDDT, each running for 125 three to five times. The short transient driving cycle methods were initially adapted 126 based on emission regulations of the Economic Commission for Europe (ECE) cycle 127 (Yao et al., 2003), which is developed and used in European countries (Laurikko, 1995). 128 The short transient driving cycle consist of four conditions, namely idling, acceleration, 129 deceleration and uniform speed, as shown in Fig. S1. For the MDDT and HDDT, we 130 customized a step-by-step test method, in which the vehicle accelerates to $20 \text{ km} \cdot h^{-1}$, 131 40 km h^{-1} and 60 km h^{-1} in sequence after the engine activates, keeping at 20 km h^{-1} 132 and 40 km h⁻¹ for 2 minutes, and 60 km h⁻¹ for 1 minute, respectively (Fig. S1) (Li et 133 al., 2021;Liu et al., 2021;Liao et al., 2021). In addition, the cold start was tested for a 134 number of vehicles after a cold soak for more than 12 hours at ambient temperature 135 136 (20-25 °C) before engine started. The measurements of cold start are compared to measurements of hot start after a ~ 10 minutes break for the vehicles after previous 137





138 measurement. More details about cold start and hot start in this campaign can be found

in Li et al. (2021).

A custom-built sampling and dilution system for vehicles combining online and 140 offline sampling techniques was used in this study. As shown in Fig. S2, a portable 141 emission measurement system (PEMS, SEMTECH-DS, Sensors. USA) was employed 142 to measure emissions of CO, CO₂, NO_x, and total hydrocarbon (THC) directly from the 143 tailpipe of vehicles. A custom-built dilution system (Li et al., 2021;Liao et al., 2021) 144 was used for dilution of vehicular emissions, achieving dilution ratios of 10-100 for 145 different vehicles. After dilution, CO₂ and CO were measured using a Li-840A 146 CO₂/H₂O Gas Analyzer (Licor, Inc. USA) and a Thermo 48i-TLE analyzer (Thermo 147 Fisher Scientific Inc. USA), respectively. Measurements of CO₂ before and after the 148 149 dilution system was used to determine the dilution ratio for each test (see details in Fig. 150 S3).

151 2.2 VOC measurements using PTR-ToF-MS

In this study, a Proton Transfer Reaction Quadrupole interface Time-of-Flight 152 Mass Spectrometer (PTR-QiToF-MS) (Ionicon Analytik, Innsbruck, Austria) with 153 H₃O⁺ chemistry was used to measure VOCs (Sulzer et al., 2014). The mass spectra of 154 PTR-ToF-MS was recorded every 1 s as to capture characteristics of VOC species from 155 vehicle exhausts in real-time. Background measurements of the instrument were 156 performed using sampled air through a custom-built platinum catalytical converter 157 158 heated to 365 °C for 30 s before vehicle starts in each test. The more detailed setting parameters for the instrument can be found elsewhere (Wu et al., 2020;Wang et al., 159 2020a;He et al., 2022). Data analysis of PTR-ToF-MS was performed using the Tofware 160 161 software package (version 3.0.3, Tofwerk AG, Switzerland) (Stark et al., 2015).

A 23-component gas standard (Linde Spectra) was used for daily calibration of PTR-ToF-MS during the campaign. VOC sensitivities from automatical calibrations indicated quite stable instrumental performance for most of the VOC species (Fig. S4). Another gas standard with 35-component VOCs (Apel Riemer Environmental Inc.) was used for calibrations during the later period of this campaign to include more VOC





167 species in the calibration. The Liquid Calibration Unit (LCU, Ionicon Analytik, Innsbruck, Austria) was used to calibrate a total of 11 organic acids and nitrogen-168 containing species (Table S3). The limits of detection for calibrated VOC species are 169 170 below 100 ppt for the 1-s measurement, except for ethanol (423 ppt) and formic acid (166 ppt). Additionally, the humidity dependence for a few VOC species in PTR-ToF-171 MS (Yuan et al., 2017;Koss et al., 2018) were corrected using humidity-dependence 172 curves determined in the laboratory, as previously shown in Wu et al. (2020). To 173 quantify the ion signals without calibration, we determine the sensitivities based on the 174 kinetics of proton-transfer reactions of H₃O⁺ with VOCs (Cappellin et al., 175 2012;Sekimoto et al., 2017). The relationship between VOCs sensitivity and kinetic 176 rate constants for the same instrument has been reported in Wu et al. (2020) and He et 177 al. (2022). The corrected sensitivities as a function of kinetic rate constants for proton-178 transfer reactions of H₃O⁺ with VOCs during this campaign is shown in Fig. S5. The 179 180 fitted line is used to determine sensitivities of uncalibrated species, and the uncertainty 181 of the concentrations for uncalibrated species are determined to be around 50%.

182 2.3 Other VOC measurements

Whole air samples were collected using canisters after the dilution system for 183 determination of hydrocarbons emitted from various vehicles. All the canisters were 184 sent to the laboratory for analysis by an offline GC-MS/FID system, with a total 95 185 hydrocarbons calibrated by Photochemical Assessment Monitoring Stations (PAMS) 186 187 and TO-15 standard mixtures (Table S4). Due to the difference of sampling (e.g., times and dilution ratios), we compared emission factors from PTR-ToF-MS and the offline 188 canister-GC-MS/FID, obtaining consistent results, except for gasoline vehicles with 189 190 China I (Fig. S6c).

An instrument based on Hantzsch reaction-absorption method was used to measure formaldehyde (Zhu et al., 2020). Good agreement for formaldehyde between PTR-ToF-MS and the Hantzsch instrument was obtained (Fig. S6a). An iodide-adduct time-of-flight chemical ionization mass spectrometer (I⁻ ToF-CIMS, Aerodyne Research, Inc.) (Wang et al., 2020c;Ye et al., 2021) was used to measure organic acids,





196 hydrogen cyanide (HCN), and isocyanic acid (HNCO) from vehicles (Li et al., 2021).

197 As shown in Fig. S6b, formic acid measured by PTR-ToF-MS and I ToF-CIMS showed

198 reasonable agreement.

199 2.4 Emission factors and emission ratios calculation

In this study, we determine emission factors of VOC species in two different 200 approaches: the mileage-based emission factors (mg·km⁻¹) as the mass of these VOCs 201 exhausted per kilometer driving of vehicles, and the fuel-based emission factors 202 203 $(mg \cdot kg_{fuel})$ as the mass of VOCs per kilogram of fuel burned by the vehicles. In addition, emission ratios of VOCs to combustion tracers (usually CO) are widely 204 applied in vehicle emissions in urban regions, as the result we determine emission 205 ratios to CO in ppb ppm⁻¹ as well. More details about the determination of emission 206 factors and emission ratios can be found in Sect. 1 in the Supplement. 207

The average emission factors for various types of vehicles are determined from 208 arithmetic means for different emission standards of vehicles. As for diesel vehicles, 209 the average emission factors are obtained from the arithmetic means of LDDT, MDDT, 210 211 HDDT, and bus. Besides, we also calculate emission factors and emission ratios from weighted means based on the fractions of gasoline and diesel vehicles with different 212 emission standards in China (MEEPRC, 2019;Li et al., 2021) (See Sect. 1 in the 213 214 Supplement for details). In order to evaluate the uncertainties of obtained emission factors, the average limit of detection for VOC species are used to estimate the limit 215 216 of detection for the determined emission factors (more details can be found in Sect. 2 in the Supplement). 217

218 **3. Results and discussions**

219 **3.1** Characteristics of the VOC emissions in the vehicles

Time series of several aromatics and OVOC species measured by PTR-ToF-MS for a selected gasoline vehicle associated with emission standard of China I and a LDDT associated with China IV emission standard are shown in Fig. 1. Both tests started with cold engines for the two vehicles. Benzene and toluene are typical aromatic species exhausted by vehicles. As shown in Fig. 1a, high concentrations of benzene and toluene





225 exhausted by the gasoline vehicle were observed as the engine started. The concentrations of the two species continued to increase until ~ 2 min after the engine 226 started, and then dropped rapidly before a minor increase during the acceleration 227 condition. These observations are similar to the previous results from PTR-MS 228 measurements in Drozd et al. (2016). Acetaldehyde and acetone are important OVOC 229 species emitted from vehicles. They show similar temporal variations as benzene and 230 toluene. However, concentrations of acetaldehyde and acetone were much lower than 231 the two aromatics after engine started. Compared to the concentrations at engine start-232 up for the gasoline vehicle (the first cycle), concentrations of the VOCs are 3.0 to 40 233 times lower during the gasoline vehicle running at hot stabilized condition (the third 234 cycle). As shown in Fig. 1 for the diesel vehicle, enhanced emissions from cold start 235 236 are minor, which is different from the gasoline vehicle. The concentration of these VOCs at engine start-up for the diesel vehicle are only 1.3 to 2.5 times higher than the 237 238 periods as the diesel vehicle running at hot stabilized condition. It indicates that the impact of the engine start-up in diesel vehicles on emissions is much lower than 239 gasoline vehicles. It might be a combined effect of cold engine and operation 240 241 temperature of the after-treatment device. In contrast to the gasoline vehicle, we observe higher concentrations of the two OVOC species than the two aromatics species from 242 the diesel vehicle. These higher OVOC concentrations in diesel vehicle exhausts are in 243 line with the observations of organic acids using the I- ToF-CIMS from the same 244 campaign (Li et al., 2021). 245

Based on the high time-resolution measurements of PTR-ToF-MS, we 246 247 determined emission factors of various VOC species from different vehicles. Fig. 2 shows the determined average mileage-based emission factors of benzene, toluene, 248 acetaldehyde, and acetone for various types of vehicles (also tabulated in the 249 Supplement table). In general, we observe a downward trend for emissions factors of 250 gasoline vehicles from China I to China VI emission standards for the four 251 representative VOC species. Emission factors of the four species for China VI vehicles 252 are 12 to 25 times lower than emissions for China I vehicles, indicating that newer 253 emission standards successfully reduced VOC emissions of gasoline vehicles. The 254





255 decline of emission factors for the four species with newer emission standards for diesel vehicles are in the range of 1.1 to 7.4 times from China III to China V, compared to 4.5 256 to 5.4 times reduction from China III to China V for gasoline vehicles. Emission factors 257 of benzene and toluene from diesel vehicles are in the range of 0.8 to 7.4 mg km^{-1} and 258 0.3 to 5.8 mg·km⁻¹, which are comparable to emission factors from gasoline vehicles 259 with China IV to China VI emission standards. This is different from observations of 260 the two OVOC species (acetaldehyde and acetone), with much higher emission factors 261 from diesel vehicles (8.0 to 27.9 mg km⁻¹ for acetaldehyde and 0.8 to 10.0 mg km⁻¹ for 262 acetone) than almost all gasoline vehicles (a maximum of 3.9 mg·km⁻¹ for acetaldehyde 263 and a maximum of 3.2 mg·km⁻¹ for acetone). Higher emission factors from diesel 264 vehicles are also observed for many other common OVOC species, as shown in Fig. 3. 265 As the largest OVOCs emitted from gasoline vehicles $(4.6 \pm 5.1 \text{ mg} \cdot \text{km}^{-1})$, methanol is 266 found to be the only common OVOC species, with lower emission factors from diesel 267 268 vehicles than gasoline vehicles. The high emissions of OVOCs from diesel vehicles may be related to combustion processes in diesel vehicles, with more excess air into 269 combustion cylinder resulting in higher oxygen contents and more oxidation processes 270 271 during fuel combustion (Pang et al., 2008; Qiao et al., 2012). Finally, the determined emission factors of the four VOC species from LPG vehicles are much lower than both 272 273 gasoline and diesel vehicles.

274 3.2 Analysis VOCs of PTR-ToF-MS mass spectra

275 In addition to typical VOC species shown above, PTR-ToF-MS detected abundant signals for a large number of ions. The determined average mileage-based 276 emission factors for all detected VOC species are shown as mass spectra in Fig. 4. VOC 277 278 species measured by PTR-ToF-MS were divided into groups according to chemical formula, namely hydrocarbon species only containing C and H atoms (C_xH_y), OVOCs 279 280 (C_xH_yO_z), species containing nitrogen and/or sulfur atoms (N/S-containing), and some other ions (others). We observe similar mass spectra of emission factors for gasoline 281 vehicles with different emission standards (Fig. S7). Highest emission factors from 282 gasoline vehicles (Fig. 4a) are detected as hydrocarbons, including C6 to C10 aromatics. 283





284 A few OVOC species, namely methanol, ethanol, formaldehyde, acetaldehyde and acetone, are also observed as the largest emissions. In contrast to gasoline vehicles, the 285 largest emissions from diesel vehicles were attributed to a few low-molecular-weight 286 OVOC species, including formaldehyde, acetaldehyde, formic acid, and acetic acid, 287 followed by a large number of hydrocarbon species. Comparison between the mass 288 spectra of gasoline and diesel vehicle emissions suggest that emissions from diesel 289 vehicles are more evenly distributed among different VOC species, as reflected by 50 290 and 140 species contributing more than 1% of the total emissions for gasoline and diesel 291 vehicles, respectively. As shown in Fig. 3b, many hydrocarbon ions in the range of m/z292 150-200 still account for significant fractions of emissions from diesel vehicles, 293 294 whereas only one species in this m/z range contribute more than 1‰ of emissions from 295 gasoline vehicles. These results demonstrate that diesel vehicles emit more heavier hydrocarbons than those from gasoline vehicles, which is consistent with observations 296 297 in previous studies (Gentner et al., 2012; Erickson et al., 2014). It should be noted that the signals of C16H22O4H (m/z=279) were higher during the tests based on determined 298 emission factors. However, we suspect that it may be emitted artifacts from the 299 300 sampling or dilution system as it mainly showed higher signals in the latter period of each test when sampling materials absorb more heat from vehicle exhausts (Fig. S8), 301 and thus it is not included in Fig. 3 (details in the Sect. 2 in the Supplement). 302

The scatterplot of carbon oxidation states $(\overline{OS_c})$ as a function of carbon number 303 (n_c) provides a framework for describing bulk chemical properties of organics (Kroll 304 et al., 2011). The details of $\overline{OS_C}$ calculation is included in Sect. 3 in the Supplement. 305 306 The results from gasoline and diesel vehicles are compared in Fig. 5 (LPG vehicles are shown in Fig. S9). It is apparent that ions with carbon oxidation states between -2.0 to 307 0 comprise main emissions for each carbon number for both gasoline and diesel 308 vehicles. It is interesting to observe that averaged $\overline{OS_c}$ for $n_c > 6$ increase as the carbon 309 number decrease for both gasoline and diesel vehicles, whereas the opposite trends are 310 observed for $n_c < 5$. The averaged $\overline{OS_c}$ in diesel vehicles for n_c between 1 and 5 are 311 significantly higher than those in gasoline vehicles, as the result of high emissions of 312 C_2 to C_5 low-molecular-weight OVOCs. Fig. 5c further shows that emission factors of 313





314 most VOC species from diesel vehicles were higher than gasoline vehicles, except a number of species occupying in the right-bottom corner of the two-dimensional space. 315 The determined mass spectra of PTR-ToF-MS in terms of emission factor for 316 317 different types of vehicles can be used to explore the dependence of various VOC emissions to different factors. Fig. 6a-b shows scatterplots of the average mileage-318 based emission factors of VOCs between cold start and hot start for gasoline and diesel 319 vehicles, respectively. We observe strong correlation between emission factors from 320 cold start and hot start tests (R=0.99 and 0.92) and generally consistent ratios between 321 cold start and hot start for different types of VOC species for both gasoline and diesel 322 vehicles, indicating that variation behaviors are similar for different species and thus 323 chemical compositions of VOC emissions are comparable between different start 324 conditions. It is obvious that emission factors of VOCs during cold start are 325 significantly higher than those during hot start for gasoline vehicles (slope=0.40), 326 327 whereas similar emissions factors between cold start and hot start are derived for diesel vehicles (slope=0.84). These results suggest that gasoline vehicles are more 328 329 significantly influenced by cold start, as the result of compositions in gasoline fuel are 330 more volatile than diesel fuel (US NRC, 1996). We further explore the effects of emission standards to VOCs emission factors by comparing determined emission 331 332 factors between China I and China V for gasoline vehicle (Fig. 6c, also see China III versus China V and China V versus China VI in Fig. S10) and between China III and 333 China V for LDDT (Fig. 6d, also see China III versus China V for MDDT and HDDT 334 in Fig. S10). Comparison of both gasoline and diesel vehicles demonstrate newer 335 336 emission standards successfully decreased VOC emissions. Based on the derived slopes, we obtain VOCs emission factors reduced by a factor of 10 for gasoline 337 vehicles from China I to China V (a factor of 5 reduction from China III to China V 338 and a factor of 2.5 reduction for China V to China VI), and a factor of 2 reduction for 339 LDDT from China III to China V (a factor of 1.5 and 8 reduction for MDDT and 340 HDDT from China III to China V). The reduction ratio for gasoline vehicles from 341 China I to China V are generally similar for most VOC species, except that some 342 OVOC species with smaller reduction ratios. The reduction ratios for LDDT vehicles 343





from China III to China V show large variability for different species. The lowest reduction ratios (a factor of ~2) are observed for the low-molecular weight OVOC species associated with largest emissions, while the reduction ratios for hydrocarbons and higher-molecular weight OVOCs are in the range of a factor of 10-100. These results indicate the after-treatment device for diesel vehicles may effectively reduce emissions of some heavier VOC species, though the after-treatment devices do not aim for VOCs control.

351 **3.3** Non-target analysis for comparison between gasoline and diesel

352 vehicles

As shown in the previous section, the analysis of PTR-ToF-MS mass spectra 353 354 provide rich information on understanding the influences of VOC emissions from vehicles. This detailed information provided by the PTR-ToF-MS also offer an 355 opportunity to systematically compare emissions between gasoline and diesel vehicles. 356 The scatterplot of the determined average emission factors of various VOC species 357 358 between gasoline and diesel vehicles is shown in Fig. 7. Large difference of VOC compositions emitted from gasoline and diesel vehicles are observed, as indicated by 359 the low correlation of the data points (R=0.24). A limited number of VOC species, 360 including C6-C10 aromatics and some N/S-containing species (e.g. C7H5N) are 361 associated with higher emission factors from gasoline vehicles, whereas the obtained 362 emission factors of most VOC species emitted from diesel vehicles are higher, 363 especially most OVOC species. For example, formic acid is found to be one of the 364 most significant emission species in diesel vehicles, with emission factors three orders 365 of magnitude higher than that of gasoline vehicles. In addition, emission factors of 366 HCN from gasoline vehicles are similar to those from diesel vehicles. These results 367 are consistent with the measurements using the I ToF-CIMS from the same campaign, 368 369 as shown in Li et al. (2021).

The scatterplot shown in Fig. 7 can also be expressed in terms of the determined fuel-based emission factors between gasoline and diesel vehicles (Fig. S11). Generally, similar variability is obtained except the determined slope of the data points, with





higher slopes determined from the scatterplot based on fuel-based emission factor (0.19 versus 0.15). The difference between the slopes reflects the different average mileage for the same weight of fuel between gasoline (9.7 km·kg_{fuel}⁻¹) and diesel vehicles (7.1 km·kg_{fuel}⁻¹), as demonstrated for emission factors of CO₂ in Table S5.

From the comparison gasoline and diesel vehicles, we can also observe profound 377 differences in relative changes of emission factors for analogous compounds series. The 378 emission factors of C6-C10 aromatics are apparently higher for gasoline vehicles than 379 diesel vehicles, whereas emission factors for larger aromatics $(n_c>11)$ from diesel 380 vehicles start to exceed gasoline vehicles. This interesting behavior is the result of 381 different variations of emission factors for gasoline and diesel vehicles as carbon 382 383 number increases. As shown in Fig. 8, emission factors of aromatics from gasoline vehicles start to rapidly decrease at $n_c=10$ (a factor of 5 for each additional carbon for 384 C10-C15), while the emission factors of aromatic for diesel vehicles demonstrate a 385 386 relatively flat pattern between C₆ and C₁₅, only with significantly decrease for $n_c>15$. 387 Based on Fig. 8, we determine that emissions of aromatics with $n_c \ge 10$ in gasoline and diesel vehicles are account for 14% and 63% of total aromatic emissions, again suggest 388 389 the importance of heavier aromatics in emissions from diesel vehicles. It also highlights that quantification of these heavier species by PTR-ToF-MS may be important in 390 characterization of vehicular exhausts, especially diesel vehicles. 391

In addition to aromatics, the relative changes of emission factors for carbonyls 392 with carbon number are apparently different between gasoline and diesel vehicles (Fig. 393 7 and Fig. 8b). Emission factors of carbonyls tend to decrease as carbon number 394 395 increase for both gasoline and diesel vehicles. The decrease magnitudes are observed to be comparable from C_1 - C_6 carbonyls for gasoline (97.6%) and diesel vehicles 396 (97.4%). However, as $n_c > 6$, the decrease of carbonyl emissions factors for diesel 397 vehicles become smaller, result in larger emissions factors than gasoline vehicles for 398 this range of carbon number. 399

The above discussions demonstrate that emission characteristics of aromatics and
 OVOCs are significantly different between gasoline and diesel vehicles. As the result,
 the ratios of VOC pairs can be identified to distinguish emissions of gasoline and diesel





403 vehicles. Fig. 9 shows the scatterplots of four representative VOCs (benzene, C14 aromatics, formaldehyde, and acetaldehyde) versus toluene based on the determined 404 emission factors. The data points for each VOCs pair clearly show distinct separation 405 406 between gasoline vehicles and diesel vehicles, with apparently higher slopes for diesel vehicles than gasoline vehicles, as the result of much larger emission factors of toluene 407 from gasoline vehicles and lower emission factors of the four representative VOCs 408 from diesel vehicles. The benzene/toluene ratio in gasoline and diesel vehicle are 409 determined as 0.48 and 1.24 mg·mg⁻¹ (corresponding to 0.57 and 1.46 ppb·ppb⁻¹ that 410 are more widely used in ambient studies). The difference of benzene/toluene ratio 411 between gasoline and diesel vehicles has been reported in previous studies, and our 412 results are generally consistent with these previous results (Chan et al., 2002;Barletta 413 et al., 2005; Qiao et al., 2012; Kumar et al., 2020). Compared to benzene/toluene ratio, 414 the difference of C14 aromatics/toluene ratio between gasoline and diesel vehicles are 415 416 more substantial (a factor of 3800). The remarkable larger emission factors of C_{14} aromatics from diesel vehicles suggest that diesel vehicles can be a significant or even 417 418 predominated source for higher molecular aromatics. The enormous difference of C14 419 aromatics/toluene ratio (and also other higher aromatics/toluene) between gasoline and diesel vehicles indicate these ratios could potentially provide good indicators for 420 421 separation of gasoline and diesel vehicles in ambient or tunnel studies. Similar 422 discrepancies are observed for formaldehyde/toluene and acetaldehyde/toluene ratios between gasoline and diesel vehicles. These ratios may not be able to be used as 423 indicators for distinguish gasoline and diesel vehicles in ambient studies, since 424 425 secondary sources may complicate the observed ratios in ambient air. However, these results strongly suggest that diesel vehicles can be important in emissions of these 426 OVOC species, though the number of diesel vehicles are smaller than gasoline 427 vehicles in many countries, e.g. China and U.S (Wallington et al., 2013; Yao et al., 428 429 2015;Huang et al., 2021).

430 **3.4 OVOC fractions in VOC emissions**

431 Emission factors of various VOC species measured by PTR-ToF-MS from





432 different vehicles are summarized in Fig. 10. As shown in Fig. 10a, the determined average mileage-based emission factors of total VOC ions from diesel vehicles were 433 much higher than gasoline and LPG vehicles. Fig. 10b-d quantified the proportions of 434 different categories of ions measured by PTR-ToF-MS. The determined average 435 mileage-based emission factors of C_xH_y accounted for the largest fraction in gasoline 436 vehicles ($84\% \pm 5.9\%$), and lower fractions in diesel ($47\% \pm 16\%$) and LPG vehicles 437 $(32\% \pm 0.7\%)$. OVOCs account for larger fractions in diesel $(49\% \pm 16\%)$ and LPG 438 vehicles (58% \pm 3.7%), while they only account for 13% \pm 6.1% of emissions from 439 gasoline vehicles. The fractions of different OVOC groups generally demonstrate a 440 downward trend from $C_xH_yO_1$ to $C_xH_yO_{\geq 3}$, and OVOCs with more than two oxygen 441 atoms only occupy small percentages (0-7%) in vehicle exhausts, indicating low 442 443 emissions of these species.

Combined with measurements of other VOCs from canisters measured by GC-444 445 MS/FID, the fractions of OVOCs in total VOC emissions can determined for different vehicles (details in Sect. 4 in the Supplement) (Fig. 11). OVOCs account for $7.7\% \pm$ 446 6.2% of total VOC emissions for gasoline vehicles. The OVOC fractions for gasoline 447 448 vehicles are generally comparable for different emission standards and cold/hot start, except somewhat higher fractions for China VI from hot start (Fig. S12). The OVOC 449 fractions obtained in this study for gasoline vehicles are generally consistent with 450 previous results (Cao et al., 2016; Wang et al., 2020b) (Fig. 11). Among these studies, 451 the OVOC fractions determined for gasoline with 10% ethanol (E10) (Roy et al., 2016) 452 $(22\% \pm 11\%)$ are apparently higher. The fractions of OVOCs in total VOC emissions 453 454 for diesel vehicles are $77\% \pm 15\%$, $68\% \pm 15\%$, $73\% \pm 14\%$, and $40\% \pm 10\%$ for LDDT, MDDT, HDDT, and bus, respectively. The variations of OVOC fractions with emission 455 standards are observed to be mixed among different types of diesel vehicles (Fig. S12). 456 The OVOC fractions from diesel vehicles are obviously higher than those in gasoline 457 vehicles, indicating the importance of OVOCs in VOC emissions for diesel vehicles. 458 Compared to previous studies (Tsai et al., 2012;Qiao et al., 2012;Cao et al., 2016;Mo 459 et al., 2016), determined OVOC fractions for diesel vehicles in this study are higher. If 460 only considering carbonyls among various types of OVOCs measured by PTR-ToF-MS, 461





462 the OVOC fractions determined in this study are more comparable with previous studies (Fig. 11), since most previous studies only detected carbonyls among various 463 types of OVOCs. Finally, we determine that OVOCs account for $41\% \pm 8.6\%$ of total 464 VOC emissions for LPG vehicles, which is also higher than in one previous study 465 (Wang et al., 2020b) with only carbonyls and a few esters/alcohols included. These 466 results stress that the large number of OVOCs measured by PTR-ToF-MS are important 467 in characterization of VOC emissions from vehicles. It should be noted that the OVOC 468 fractions obtained here only reflect exhaust emissions. Evaporative emissions may be 469 associated with different fractions of various VOC groups, which may be more related 470 to fuel compositions (Rubin et al., 2006;Huang et al., 2021). 471

472 **4. Conclusions**

In this work, we conducted a chassis dynamometer study to measure VOC 473 474 emissions from gasoline, diesel, and LPG vehicles using PTR-ToF-MS along with other offline and online measurement techniques. Using this dataset, we provide emission 475 factors of many VOCs from these three different types of vehicles associated with 476 various emission standards in China. Our results show that emission factors of VOCs 477 generally decrease with the increased stringency of emission standards for gasoline 478 vehicles, whereas variations of emission factors for diesel vehicles with emission 479 standards are more diverse. Mass spectra analysis of PTR-ToF-MS suggest that cold 480 start significantly influence VOCs emission of gasoline vehicles, while the influences 481 482 are smaller for diesel vehicles.

We observe large differences of VOC emissions between gasoline and diesel 483 vehicles based on PTR-ToF-MS measurements. Emission factors of most VOC species 484 485 from diesel vehicles were higher than gasoline vehicles, especially for most OVOCs and heavier aromatics. The substantial larger emission factors of some OVOCs 486 487 emission factors for diesel vehicles indicate potentially dominant emissions of these species from diesel vehicles among vehicular emissions. Our results suggest that VOC 488 pairs (e.g. C₁₄ aromatics/toluene ratio) could potentially provide good indicators for 489 490 distinguishing emissions between gasoline and diesel vehicles.





491 Based on measurements of PTR-ToF-MS, CxHy ions account for the largest fraction in gasoline vehicles ($84\% \pm 5.9\%$), whereas OVOC ions are the largest 492 contributor in the mass spectra of emissions from diesel ($49\% \pm 16\%$) and LPG vehicles 493 $(58\% \pm 3.7\%)$. In the end, the fractions of OVOCs in total VOC emissions are 494 determined by combining hydrocarbons measurements from canister results and online 495 measurements of PTR-ToF-MS. We show that OVOCs contribute 7.7% \pm 6.2% of 496 gasoline vehicles of the total VOC emissions, while the fractions are significantly 497 higher for diesel vehicles (40-77%), highlighting the importance to detect these OVOC 498 species in diesel emissions. 499

This study shows significant contributions of OVOCs in VOC emissions from 500 various vehicles, especially diesel vehicles. As a consequence, vehicular emissions may 501 account for considerable proportions for primary emissions of these OVOCs in urban 502 regions. Emissions of many OVOC species are currently not fully represented in 503 504 emission inventories of VOCs, which may in turn affect the prediction ability of air quality models in urban regions. In this study, OVOC species are mainly quantified 505 from PTR-ToF-MS measurements by taking into account all signals in the mass spectra, 506 507 which stress that the large number of OVOC species measured by PTR-ToF-MS are important in characterization of VOC emissions from vehicles. 508

509 Data availability

510 Data are available from the authors upon request.

511 Author contribution

512 BY designed the research. ZBY, JYZ, BY, QES organized vehicle test 513 measurements. SHW, CHW, CMW, TGL, JPQ, QES, and MMZ contributed to data 514 collection. SHW performed the data analysis, with contributions from TGL, XJH, YBH, 515 XBL, and QES. SHW and BY prepared the manuscript with contributions from other 516 authors. All the authors reviewed the manuscript.

517 **Competing interests**

518 The authors declare that they have no known competing financial interests or 519 personal relationships that could have appeared to influence the work reported in this





520 paper.

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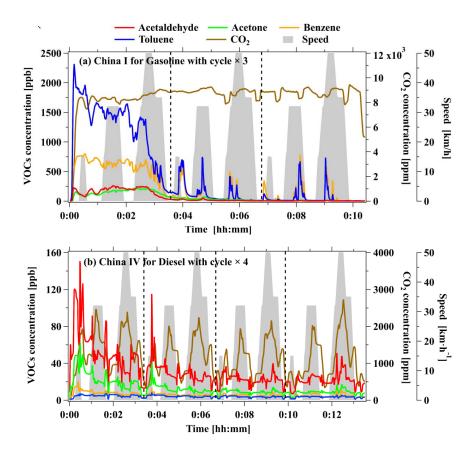


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Figure 1. Real-time concentrations of acetaldehyde, acetone, benzene, toluene, and CO₂ for (a) a gasoline vehicle with emission standard of China I and (b) a light-duty diesel vehicle with emission standard of China IV. The two vehicles were both cold started. The gray shadows represent the speed of the vehicles on the chassis dynamometer.





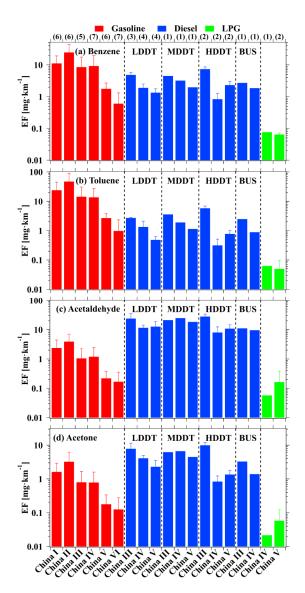
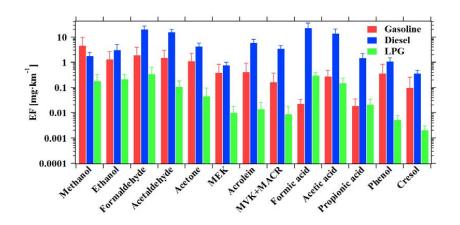




Figure 2. The determined average mileage-based emission factors (mg·km⁻¹) for (a) benzene, (b) toluene, (c) acetaldehyde, and (d) acetone for vehicles with different emission standards. The numbers above the top axis represent the number of all experiments (including multiple measurements for individual test vehicle) for each emission standard. Error bars represent standard deviations of emission factors for the specific emission standard.







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Figure 3. The determined emission factors of representative OVOC species from
different types of vehicles. Error bars represent standard deviations of the emission
factors for the VOCs.





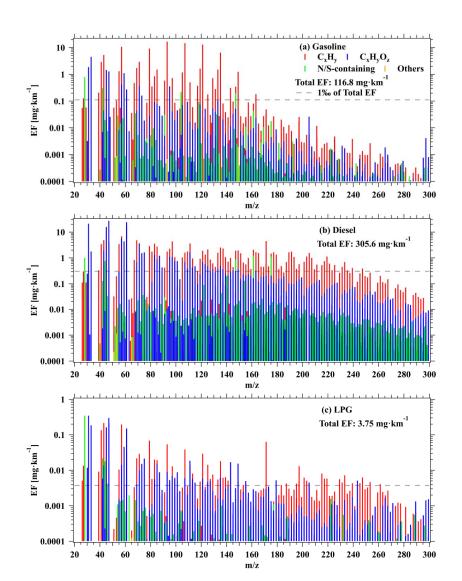


Figure 4. The determined average mileage-based emission factors of VOC species
measured by PTR-ToF-MS from (a) gasoline, (b) diesel, and (c) LPG vehicles. The
gray dashed lines represent 1‰ of total VOCs emission factors.





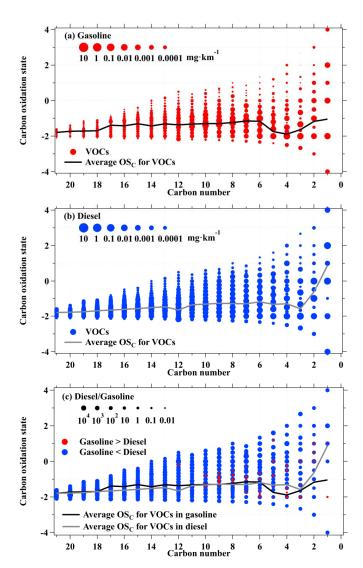


Figure 5. The two-dimensional space of $\overline{OS_c} - n_c$ with data points sized coded using emission factors of VOC species from (a) gasoline and (b) diesel vehicles, and (c) the ratio of emission factors of diesel vehicle relative to gasoline vehicle. The black and gray lines are the average $\overline{OS_c}$ of each carbon number for VOC species in gasoline and diesel vehicles, respectively.





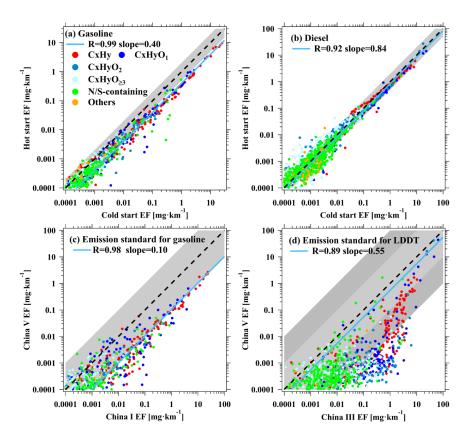


Figure 6. Scatterplots of VOCs emission factors between cold start and hot start for gasoline (a) and diesel vehicles (b). Scatterplots of VOCs emission factors between China I and China V emission standard for gasoline vehicles (c) and between China III and China V emission standard for diesel vehicles (d). Each data point indicates a VOC species measured by PTR-ToF-MS. The blue lines are the fitted results for all data points. The black dashed lines represent 1:1 ratio, and the shaded areas represent ratios of a factor of 2 in (a) and (b), and a factor of 10 and 100 in (c) and (d).





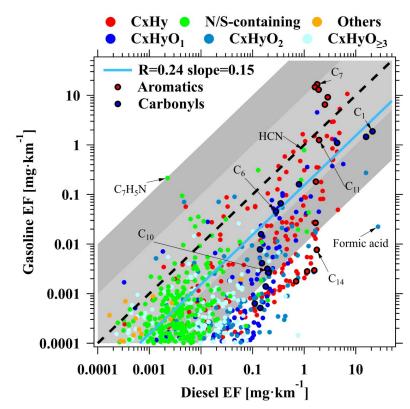
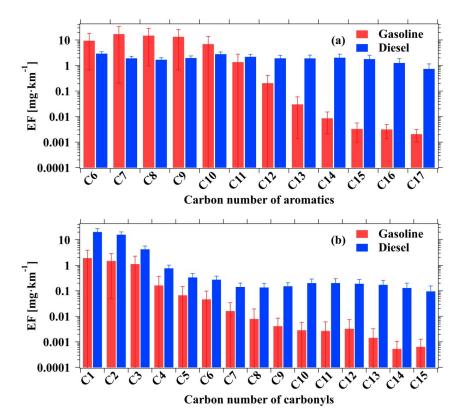


Figure 7. Scatterplot of VOCs emission factors between gasoline and diesel vehicles.
Each data point indicates a VOC species measured by PTR-ToF-MS. The blue line is
the fitted result for all data points. The black line represents 1:1 ratio, and the shaded
areas represent ratios of a factor of 10 and 100.

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Figure 8. The determined emission factors of (a) aromatics and (b) carbonyls for each
carbon number from gasoline and diesel vehicles. Error bars represent standard
deviations of the emission factors for the VOCs of different carbon number.





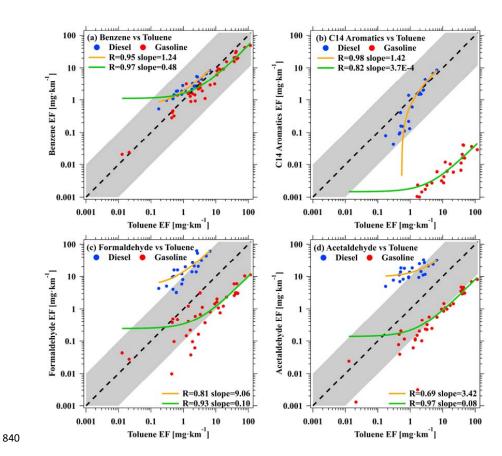


Figure 9. Scatterplots of the determined mileage-based emission factors of (a) benzene versus toluene, (b) C₁₄ aromatics versus toluene, (c) formaldehyde versus toluene, and (d) acetaldehyde versus toluene for gasoline and diesel vehicles. Each data point represents each test vehicle in this study. The green and orange lines are the fitted results for gasoline and diesel vehicle. The black line represents 1:1 ratio, and the shaded areas represent ratio of a factor of 10.





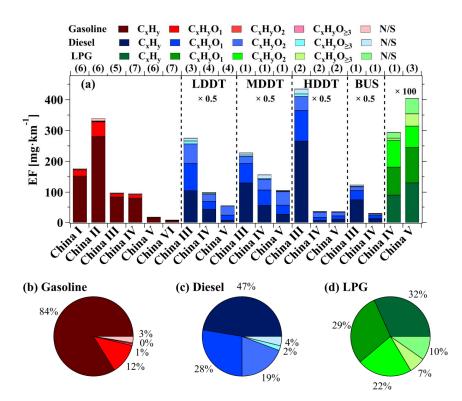


Figure 10. (a) The determined average emission factors for different emission standard from gasoline, diesel (×0.5), and LPG (×100) vehicles measured by PTR-ToF-MS. The different ion categories are discussed in the manuscript. Fractions of the determined average emission factors of VOCs ions in different ion categories from (b) gasoline, (c) diesel, and (d) LPG vehicles. The numbers above the top axis represent the number of all experiments (including multiple measurements for individual test vehicle) for each emission standard.





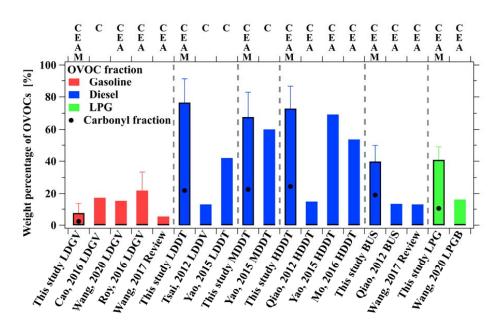


Figure 11. Comparison of OVOCs fractions determined in this study and those in previous studies. Error bars represent the standard deviations of the weight percentage of OVOCs. The C, E, A, M above the top axis represent the four groups of OVOCs measured in this study or previous studies, including Carbonyl: C, Ester/Ether: E, Alcohol: A, Multiple-functional: M.