

1 **Variations and sources of volatile organic compounds (VOCs)**
2 **in urban region: insights from measurements on a tall tower**

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27 **Abstract**

28 Volatile organic compounds (VOCs) are key precursors of ozone and particulate
29 matter, which are the two dominant air pollutants in urban environments. However,
30 compositions and sources of VOCs in urban air aloft were rarely reported so far. To
31 address this matter, highly time-resolved measurements of VOCs were made by proton-
32 transfer-reaction time-of-flight mass spectrometer (PTR-ToF-MS) at a 450-m platform
33 on the Canton Tower in Guangzhou, China. A combination of *in-situ* measurements and
34 modeling techniques was used to characterize variations and sources of VOCs. Five
35 sources were identified from positive matrix factorization (PMF) analysis, namely
36 daytime-mixed (e.g., biogenic emissions and secondary formation), visitor-related (e.g.,
37 human breath, cooking, and volatilization of ethanol-containing products),
38 vehicular+industrial, regional transport, and volatile chemical product (VCP)-
39 dominated (i.e., volatilization of personal care products), contributing on average to
40 21%, 30%, 28%, 10%, and 11% of total VOCs (TVOC) mixing ratios, respectively. We
41 observe that contributions of the visitor-related source, mainly composed of ethanol,
42 followed well with the variation in visitor numbers on the tower. The VCP-dominated
43 source only had an average contribution of ~ 5.7 ppb during the campaign, accounting
44 for a small fraction (11%) of TVOC mixing ratios but a large fraction (22%) of the total
45 OH reactivity. However, large fractions of reactive VOC species, e.g., monoterpenes
46 (49%), were attributed to the VCP-dominated source, indicating important
47 contributions of VCPs to ambient concentrations of these species in urban environments.
48 Vertical profiles of air pollutants (namely NO_x, ozone, O₃, and PM_{2.5}), measured at 5
49 m, 118 m, 168 m, and 488 m, exhibited more evident gradients at night than in the
50 daytime owing to the stronger stability of the nocturnal boundary layer. Mixing ratios
51 of VOC species during the nighttime generally decreased with time when the 450-m
52 platform was located in the nocturnal residual layer and markedly increased when
53 impacted by emissions at ground. The results in this study demonstrated composition

- 54 characteristics and sources of VOCs in urban air aloft, which could provide valuable
- 55 implications in making control strategies of VOCs and secondary air pollutants.

56 **1 Introduction**

57 Volatile organic compounds (VOCs) are important trace gases in the atmosphere
58 and are composed of myriad chemical species (Pallavi et al., 2019; Wang et al., 2020a;
59 Gkatzelis et al., 2021). Except for their direct adverse impacts on human health (Zhang
60 et al., 2013), VOCs are also important precursors of secondary pollutants such as ozone
61 and secondary aerosol (Vo et al., 2018; Zhou et al., 2019; Qin et al., 2021). Reduction
62 in ambient VOCs concentrations is the key for synergistic control of both ozone and
63 particle pollution. However, it is highly challenging for this target due to complex
64 sources and chemical transformations of VOCs in urban environments (Yuan et al.,
65 2012; Mo et al., 2016; Zhu et al., 2019).

66 In addition to compiling accurate emission inventories (bottom-up method)
67 (Zheng et al., 2013; An et al., 2021), the combination of *in-situ* measurements and
68 receptor models (top-down method) was widely adopted to quantitatively apportion
69 sources of ambient VOCs (Baudic et al., 2016; Liu et al., 2016; Fan et al., 2021; Pernov
70 et al., 2021). Concentrations of various VOC species can be measured by offline and
71 online techniques. Gas chromatography-flame ionization detector/mass spectrometry
72 (GC-FID/MS) combined with stainless steel canisters are the most popular offline
73 technique (Guo et al., 2011; Yuan et al., 2013; Zhang et al., 2013; Qin et al., 2021).
74 Automated online GC-FID system and high time resolution mass spectrometer, such as
75 proton-transfer-reaction mass spectrometer (PTR-MS) and chemical ionization mass
76 spectrometer (CIMS), are popular online techniques (de Gouw and Warneke, 2007;
77 Wang et al., 2020a; Wang et al., 2020c; Fan et al., 2021; Ye et al., 2021). However,
78 VOCs measurements made by both online and offline instruments are markedly
79 affected by very local emission sources, particularly in urban environments, when they
80 are usually deployed at ground level. This is highly important for studies aiming to
81 characterize variations and sources of ambient VOCs at large spatial scales (such as a
82 city or city clusters) based on measurements of only one site. To address this concern,

83 VOCs measurements made in the upper part of the planetary boundary layer (PBL) may
84 be a better choice due to the well mixing of surface emissions when being transported
85 upward from sources to observation sites (Hu et al., 2015a; Hu et al., 2015b; Squires et
86 al., 2020).

87 As reported in the literature, *in-situ* measurements of VOCs at high altitudes (e.g.,
88 hundreds of meters or several kilometers above ground level) were predominantly made
89 using the combination of offline techniques and samples collected by various platforms
90 such as aircrafts (Geng et al., 2009; Xue et al., 2011; Benish et al., 2020), tethered
91 balloons (Zhang et al., 2018; Wu et al., 2020b; Wang et al., 2021; Wu et al., 2021), high
92 buildings and towers (Ting et al., 2008; Mo et al., 2020), and unmanned aerial vehicles
93 (UAVs) (Vo et al., 2018; Liu et al., 2021). These offline measurements were
94 predominantly used to reveal vertical variations of VOCs concentrations, impacts of
95 VOCs degradation chemistry on the formation of secondary pollutants, and source
96 characteristics of the species of interest. Offline measurements made at high altitudes
97 were generally not capable of fully characterizing temporal variations of concentrations
98 and source characteristics of VOCs due to strict limitations in their time resolution and
99 sample sizes. In this condition, online VOCs measurements with fast response at high
100 altitudes are required. Lack of available platforms has been a key limited factor for
101 conducting online VOCs measurements at high altitudes in China. For instance, the
102 combined utilization of aircraft and online spectrometer (such as PTR-MS) has been
103 widely used in North America to measure VOCs concentrations in the lower
104 troposphere (Hornbrook et al., 2011; Müller et al., 2016; Yuan et al., 2016; Koss et al.,
105 2017; Fry et al., 2018; Chen et al., 2019), while it is quite difficult in China due to the
106 lack of professional research aircraft and the strict control of airspace. Tethered balloons
107 and UAVs are generally not suitable for online VOCs measurements due to their limited
108 payloads (Dieu Hien et al., 2019). Tower-based platforms provide another path for
109 online VOCs measurements at high altitudes in urban environments. However, tower-
110 based online measurements of VOCs were only reported in Beijing, China so far

111 ([Squires et al., 2020](#); [Zhang et al., 2020](#)).

112 In this study, continuous online VOCs measurements, including more than 200
113 species with a time resolution of 10 s, were made at a 450-m platform on the Canton
114 Tower in the Pearl River Delta (PRD) region, China during August–November 2020. A
115 combination of the VOCs measurements and the positive matrix factorization (PMF)
116 receptor model was used to provide new insights into the concentrations, temporal
117 variations, and source contributions of VOCs in urban region.

118 **2 Methods and materials**

119 **2.1 Site description and field campaign**

120 The PRD region is one of the most developed city clusters in China with more
121 than 70 million residents by 2020 and is suffering from air pollution problems (e.g.,
122 ozone and secondary aerosol) ([Wang et al., 2017](#); [Wang et al., 2020b](#); [Yan et al., 2020](#);
123 [Li et al., 2022](#)). In this study, VOCs measurements were made at the Canton Tower
124 (CTT, 23.11°N, 113.33°E) in Guangzhou, a large city in PRD (Figure S1), from August
125 18 to November 5 in 2020. The CTT has a total height of 610 m including the shaft on
126 the top (Figure S1(c)). The observation was conducted in a room (Figure S1) at the 450-
127 m Look Out platform ([Jin et al., 2022](#)), which is a ramp with stairs and is located on the
128 top of the main body of the CTT. The observation room is located below the ramp and
129 a sampling port is reserved on the wall outside the tower. A louver is located ~3 m below
130 the sampling port. The 450-m Look Out platform is a famous tourist attraction with an
131 opening time of local time (LT, UTC+8) 10:00–22:30, and visitors could walk around
132 for a panorama of downtown Guangzhou. On each day, there are two busiest tourist
133 hours, roughly at LT 11:00–14:00 and 18:00–21:00, on the 450-m platform. In addition,
134 there are three restaurants between 376 and 423 m. The VOCs measurements were
135 interrupted during October 8–12 due to instrument malfunction.

136 2.2 VOCs measurements

137 VOCs measurements were made using a high-resolution proton-transfer-reaction
138 quadrupole interface time-of-flight mass spectrometer (PTR-QiToF-MS, Ionicon
139 Analytik, Innsbruck, Austria) with both hydronium ion (H_3O^+) (Yuan et al., 2017; Wu
140 et al., 2020a) and nitric oxide ion (NO^+) chemistry (Wang et al., 2020a). The H_3O^+ and
141 NO^+ modes were automatically switched with 22 min for the H_3O^+ mode and 12 min
142 for the NO^+ mode during the campaign. In this study, only VOCs measurements made
143 in the H_3O^+ mode were used for analysis. In H_3O^+ mode, the PTR-QiToF-MS was
144 operated with a drift tube pressure of 3.8 mbar, a drift tube temperature of 120 °C, and
145 a drift tube voltage of 760 V, resulting in an E/N (E refers to electric field and N refers
146 to number density of buffer gas in the drift tube) value of ~ 120 Td (Townsend). Raw
147 data of PTR-ToF-MS were processed and analyzed using Tofware software (Tofwerk
148 AG, v3.0.3) and please refer to our previous works (Wang et al., 2020a; Wu et al., 2020a)
149 for more details. Signals of 3035 ions with m/z up to 510 were obtained at time
150 resolutions of 10 s. To measure VOCs concentrations outside the tower, a ~ 5 m long
151 Perfluoroalkoxy (PFA) Teflon tubing (OD: 1/4") was used to connect the inlet of the
152 instrument and the sampling port (Figure S1). The PFA Teflon tubing has been proven
153 to be effective in measuring ambient concentrations of VOCs (Deming et al., 2019; Liu
154 et al., 2019) and has been widely used in field studies (de Gouw et al., 2003a; Hu et al.,
155 2011; Wu et al., 2020a). Air sample in the tubing was drawn by a pump at a flow rate
156 of ~ 5 L min^{-1} . Blank measurements were performed automatically at the last 2 min of
157 the H_3O^+ mode by passing ambient air through a platinum catalyst heated to 365 °C.

158 A gas standard with 35 VOC species (Table S1) was used for calibrations of the
159 PTR-ToF-MS once per day. Ten organic acids and nitrogen-containing VOC species
160 were calibrated using a liquid calibration unit in the laboratory. Sensitivities of the
161 remaining VOC species were determined using the quantification method based on
162 reaction kinetics of the PTR-ToF-MS (Wu et al., 2020a; He et al., 2022). Impacts of the
163 change in ambient humidity on measured signals of the PTR-ToF-MS were removed

164 using humidity-dependence curves of VOC species determined in the laboratory (Wang
165 et al., 2020a; Wu et al., 2020a). The limit of detection (LOD) for a VOC species was
166 defined as the concentration when the signal-to-noise ratio (SNR) equals to 3 (Yuan et
167 al., 2017). Average mixing ratios, LODs, sensitivities, chemical formula, and suggested
168 compounds of 225 VOC species used in this study are summarized in Table S1.

169 **2.3 Other measurements**

170 During the CTT campaign, a CO₂ and H₂O Gas Analyzer (Model: Li-840A, Licor
171 Inc., USA) was deployed to measure carbon dioxide (CO₂, ppm in dry air) and humidity
172 (mmol mol⁻¹). In addition, four air quality automatic monitoring stations are located at
173 ground level (~5 m), 118 m, 168 m, and 488 m of the CTT, which report hourly
174 concentrations of ozone, NO, NO₂, NO_x, and PM_{2.5} along with meteorological
175 parameters, namely temperature (T), relatively humidity (RH), and pressure (Mo et al.,
176 2020). Mass concentrations of gaseous pollutants were reported at 25 °C and 1013.25
177 hPa and were converted to mixing ratios (ppb) accordingly. Contour plots of vertical
178 profiles of NO_x, ozone, Ox (O₃+NO₂), and PM_{2.5} concentrations were made using the
179 bilinear method in Igor software (v8.04). Linear interpolations for concentrations of
180 these pollutants were performed on both spatial (altitude) and temporal scales. A
181 ceilometer (CL31, Vaisala, Finland) deployed on the Panyu Campus of Jinan University
182 (23.02°N, 113.41°E, Figure S1), approximately 13.5 km to the southeast of the CTT,
183 was used to measure planetary boundary layer height (PBLH) during the campaign. In
184 addition, measurements of VOCs and CO₂ made on the campus of Guangzhou Institute
185 of Geochemistry (GIG), Chinese Academy of Sciences (23.15°N, 113.36°E, ~25 m
186 above ground level) during September–November 2018 (Wang et al., 2020a; Wang et
187 al., 2020c; Wu et al., 2020a) were used for comparison with those measured on the CTT.
188 *p* values were obtained using the Student's *t*-test to determine statistical significance
189 levels of differences. The GIG site is located approximately 5.7 km to the northeast of
190 the CTT. Measurements of VOCs and CO₂ at the GIG site were made using the same
191 instruments as those at the CTT site.

192 **2.4 PMF receptor model**

193 The PMF receptor model was used to quantitatively analyze sources of the VOCs
194 measurements made at the 450-m platform. The PMF model has been widely used to
195 determine source contributions of measured VOCs concentrations in previous studies
196 (Yuan et al., 2012; Pallavi et al., 2019; Pernov et al., 2021). A simple description of the
197 PMF model was provided in the Supplementary Information (SI).

198 The PMF model was performed on 225 VOC species (Table S1) in this study. In
199 preparation of PMF input data, measured concentrations of a VOC species below the
200 LOD were replaced with half of the LOD and corresponding uncertainties were
201 assigned to 5/6 of the LOD. Missing samples of a VOC species were replaced with its
202 median value during the campaign and corresponding uncertainties were set as values
203 equal to three times the median value (Zhang et al., 2013; Pernov et al., 2021; Qin et
204 al., 2021). During the CTT campaign, the measured ethanol concentrations were
205 impacted by the change in the number of visitors (a detailed discussion in Section 3.3)
206 and exhibited strong variations (Figure 1). Thus, measurement uncertainties of ethanol
207 calculated by Eq. (S3) were reduced by a factor of 5 to increase its weight in PMF
208 analysis, which successfully resolved factors representing visitor influences and reduce
209 residuals of PMF solution from over 20% to ~14%. The PMF analysis was performed
210 using the PMF Evaluation Tool (v3.05) with Igor Pro (Ulbrich et al., 2009).

211 **3 Results and discussion**

212 **3.1 Overview of field measurements during the campaign**

213 As shown in Figure 1, concentrations of various species and meteorological
214 parameters all exhibited strong variations during the campaign. Daily mean ozone
215 mixing ratios varied in the range of 17.8–105.0 ppb with an average (\pm standard
216 deviation) of 55.1 ± 18.3 ppb. Daily mean total VOCs (TVOC) mixing ratios, including
217 a total of 225 species, varied between 23.9–124.2 ppb with an average of 62.1 ± 21.8
218 ppb. Daily mean NO_x mixing ratios varied in the range of 7.9–31.6 ppb with an average

219 of 13.6 ± 3.8 ppb. Measured CO_2 mixing ratios exhibited strong variability with daily
220 mean values ranging from 403.5 to 471.4 ppm. Ethanol was the most abundant VOC
221 species, accounting on average for 23.5% of measured TVOC mixing ratios. Daily
222 mean ethanol mixing ratios varied between 4.3–53.4 ppb with an average of 15.3 ± 9.1
223 ppb. Toluene was the most abundant aromatic species and had an average mixing ratio
224 of 1.4 ± 0.9 ppb. Daily mean temperatures varied in the range of 17.7–29.0 °C with an
225 average of 23.2 ± 3.0 °C. Daily mean RH varied between 39.3%–85.0% with an average
226 of $71.6\% \pm 10.3\%$. In general, the observation site was predominantly influenced by
227 hot and moist air masses from August 18 to October 4, but cooler and dryer air masses
228 from October 5 to November 5.

229 The most abundant 10 VOC species measured by PTR-ToF-MS during the CTT
230 campaign were ethanol, methanol, acetic acid, formaldehyde, acetone, ethyl acetate,
231 acetaldehyde, hydroxyacetone+propionic acid, toluene, and C8 aromatics, contributing
232 to over 70% of TVOC mixing ratios. As shown in Figure 2, the 225 VOC species were
233 classified into six categories, namely C_xH_y (i.e., hydrocarbons), $\text{C}_x\text{H}_y\text{O}_1$ (i.e., VOC
234 species containing one oxygen atom), $\text{C}_x\text{H}_y\text{O}_2$ (i.e., VOC species containing two
235 oxygen atoms), $\text{C}_x\text{H}_y\text{O}_{\geq 3}$ (i.e., VOC species containing three or more oxygen atoms),
236 N/S containing species (i.e., VOC species containing nitrogen or sulfur atoms), and
237 siloxanes (Wu et al., 2020a; He et al., 2022). The most abundant category was $\text{C}_x\text{H}_y\text{O}_1$,
238 which had an average contribution of 67% to TVOC mixing ratios, but only contributed
239 to 40% of total OH reactivity. The $\text{C}_x\text{H}_y\text{O}_2$ and $\text{C}_x\text{H}_y\text{O}_{\geq 3}$ categories contributed to 22%
240 and 1% of TVOC mixing ratios, respectively. C_xH_y only accounted for 9% of TVOC
241 mixing ratios but contributed to 37% of the total OH reactivity, indicating more reactive
242 VOC species in this category. Concentrations of N/S containing species and siloxanes
243 were generally lower than 0.5 ppb and totally contributed to ~1% of TVOC mixing
244 ratios.

245 At ground level, each VOCs category accounted for comparable fractions in
246 TVOC mixing ratios and the total OH reactivity to those measured at 450 m. However,

247 the majority of the C_xH_y , $C_xH_yO_{\geq 3}$, and N/S containing species measured at 450 m had
248 lower mixing ratios than those measured at ground level (Figures 2(b) and S2),
249 implying their predominant contributions from surface emission sources. Most of the
250 $C_xH_yO_1$ and $C_xH_yO_2$ species measured at 450 m had comparable mixing ratios to those
251 measured at the ground level. However, mixing ratios of some $C_xH_yO_2$, $C_xH_yO_{\geq 3}$, and
252 N/S containing species measured at 450 m were higher than those measured at ground
253 level, which can be attributable to either enhancement of their emissions on the 450-m
254 platform or more secondary formation from oxidation of VOCs (e.g., C_xH_y and $C_xH_yO_1$
255 species). The differences in contributions of VOCs categories to the total concentrations
256 and OH reactivity imply that sources of the VOCs measurements made at 450 m and
257 the ground level are different.

258 **3.2 Diurnal variations in selected VOC species**

259 Average diurnal profiles of nine selected VOC species measured by PTR-ToF-MS
260 during the CTT campaign are demonstrated in Figure 3. Measurement results at GIG in
261 2018 are also shown for comparison to investigate differences in their diurnal variation
262 patterns and likely sources. In addition, average diurnal profiles of the selected VOC
263 species on working and non-working days (including weekends and public holidays
264 when the 450-m platform had more visitors) during the CTT campaign are compared
265 to explore potential emissions from visitors. Average diurnal variations in ratios of
266 concentrations of selected VOC species measured on non-working days to those
267 measured on working days were also calculated, as shown in Figure S3. Meteorological
268 factors, namely temperature and RH, exhibited insignificant differences between
269 working and non-working days (Figure S4). Thus, the differences in VOCs
270 concentrations between working and non-working days were not notably impacted by
271 the change in meteorological conditions.

272 Diurnal profiles of aromatic species, including benzene, toluene, and C8 aromatics
273 measured at 450 m exhibited similar variability with minima occurring between LT
274 12:00–16:00. Aromatics with higher chemical reactivity could be removed more rapidly

275 by reactions with hydroxyl radicals (OH) in the daytime (Yuan et al., 2012; Wu et al.,
276 2020a). In addition, rapid elevation of the daytime PBL could enhance the dilution of
277 chemical species, leading to rapid decreases in their concentrations (Sangiorgi et al.,
278 2011; Zhang et al., 2018). The two effects are the two most important factors for
279 controlling diurnal profiles of aromatics measured at 450 m. By contrast, diurnal
280 profiles of aromatics measured at ground displayed a different pattern with two peaks
281 occurring in the morning (LT 07:00–08:00) and evening (LT 19:00–22:00), respectively.
282 Diurnal patterns of aromatics are consistent with that of NO_x (a typical tracer of traffic
283 emissions in urban region) at ground level but were different from that of NO_x at 450
284 m (Figure 4). Therefore, measured concentrations of aromatics, particularly for benzene,
285 were markedly affected by traffic emissions at ground level, but contributed by more
286 complex sources at 450 m. The differences in diurnal profiles of aromatics between
287 working and non-working days were insignificant ($p>0.05$), implying minor
288 contributions from visitor-related emissions. On working days, toluene concentrations
289 measured at 450 m were more affected by traffic emissions as manifested by the two
290 remarkable peaks in the morning and late afternoon.

291 Isoprene and monoterpenes exhibited distinct diurnal variation patterns during the
292 two campaigns. As reported in (Gómez et al., 2020; Tan et al., 2021), diurnal profiles
293 of isoprene and monoterpenes concentrations in non-urban regions usually displayed
294 unimodal patterns with a peak occurring at noon due to the strong light/temperature-
295 dependence of biogenic emissions. In this study, isoprene concentrations at 450 m
296 plateaued during the daytime and were slightly higher on non-working days than those
297 on working days, implying large contributions from visitor-related emissions. The
298 diurnal profile of monoterpenes measured at 450 m exhibited a bimodal pattern with
299 two peaks at LT 14:00 and 20:00, which was roughly in accordance with diurnal peaks
300 of visitor numbers on the 450-m platform. In addition, monoterpenes concentrations at
301 450 m were significantly ($p<0.01$) higher on non-working days (particularly during the
302 busiest tourist hours) than on working days, confirming significant contributions from

303 visitor-related or cooking emissions (Klein et al., 2016). The diurnal profiles of methyl
304 vinyl ketone (MVK) + methacrolein (MACR) demonstrated similar shapes to ozone at
305 both 450 m and ground level with maxima occurring between LT 13:00–15:00 (Figure
306 4), consistent with MVK+MACR as photooxidation products of isoprene (Greenberg
307 et al., 1999; Zhao et al., 2021). The concentrations of MVK+MACR during the daytime
308 on non-working days were also significantly ($p<0.01$) higher than those on working
309 days, which are consistent with isoprene observations.

310 Acetone, methanol, and ethanol are abundant OVOC species in urban atmosphere.
311 Diurnal profiles of acetone measured at both 450 m and the ground level were
312 characterized by higher concentrations in the daytime, suggesting predominant
313 contributions from daytime sources, such as vegetation emissions and photooxidation
314 of hydrocarbons (Hu et al., 2013; Gkatzelis et al., 2021). In addition, acetone
315 concentrations at 450 m were higher on non-working days than on working days,
316 implying prominent contributions from visitor-related emissions. Diurnal profiles of
317 methanol and ethanol measured at ground level were characterized by a bimodal pattern
318 with two peaks occurring in the morning (LT 08:00) and evening (LT 20:00),
319 respectively, confirming strong contributions from traffic emissions. However,
320 methanol concentrations measured at 450 m exhibited weak diurnal variability and
321 lower concentrations on non-working days, indicating that they were less affected by
322 visitor-related emissions. The diurnal profile of ethanol at 450 m displayed two peaks
323 at LT 13:00 and 19:00, respectively, which was in accordance with the two busiest
324 tourist hours of the 450-m platform. In addition, ethanol concentrations at 450 m on
325 non-working days were significantly ($p<0.01$) higher than those on working days,
326 particularly in the opening hours of the 450-m platform. These results suggest that the
327 ethanol concentrations measured at 450 m were largely contributed by visitor-related
328 emissions.

329 To further explore spatial scales of emission source regions for different VOC
330 species, autocorrelation profiles of their time series were calculated by offsetting time

331 from -120 to 120 min. As indicated in previous studies (Hayes et al., 2013; Hu et al.,
332 2016), concentrations of a species that is more affected by local sources would have a
333 narrower autocorrelation profile. As shown in Figure 4, peak widths of autocorrelation
334 profiles for different species at 450 m strongly varied. Autocorrelation profiles of
335 monoterpenes, toluene, ethanol, methanol, and isoprene were relatively narrower (even
336 narrower than the autocorrelation profile of NO_x), and thus sources of these species
337 had more local characteristics. Autocorrelation profiles of benzene, C₈ aromatics,
338 acetone, and MVK+MACR were much flatter (but narrower than the autocorrelation
339 profile of ozone and O_x), indicating that concentrations of these species were more
340 contributed by sources at larger spatial scales. By contrast, peak widths of the
341 autocorrelation profiles for different species (except for ethanol) were comparable to
342 that of NO_x. Therefore, concentrations of the selected VOC species were notably
343 contributed by local traffic emissions at ground level but contributed by more complex
344 sources on larger spatial scales at 450 m.

345 **3.3 Impacts of visitor-related emissions on VOCs measurements**

346 As introduced in section 2.1, the CTT campaign was conducted in August-
347 November of 2020, during which visitors were required to wear masks when visiting
348 the CTT and ethanol-containing products were widely used to prevent the spread of the
349 COVID-19 pandemic. For example, medicinal alcohol (75%) spray was widely used to
350 wipe public utilities and 75%-ethanol bacteriostatic gel was extensively used as
351 sanitizer for hands. The total usage of ethanol-containing products was closely
352 associated with the number of visitors. This can be manifested by the diurnal profiles
353 of some VOC species (e.g., ethanol) that exhibited similar variation patterns to that of
354 the number of visitors at the 450-m platform, as shown in Figure 3. In addition, the
355 restaurants are located ~30 m below the observation site and emission intensities of
356 VOCs (e.g., monoterpenes) from cooking-related sources were also closely associated
357 with the number of visitors. Therefore, the VOCs measurements were inevitably
358 affected by visitor-related emissions, such as human breath, cooking, and volatilization

359 of ethanol-containing and personal care products (Veres et al., 2013).

360 As shown in Figure 5(a), the diurnal profile of CO₂ measured at 450 m increased
361 between LT 09:00–20:00, which was different from those measured at ground level.
362 The higher CO₂ mixing ratios at 450 m were predominantly contributed by human
363 breath due to the absence of combustion sources. Measured ethanol mixing ratios were
364 well correlated with those of CO₂ ($R^2=0.36$, $p<0.01$) during the CTT campaign (Figure
365 5(b)), indicating that ethanol concentrations, as well as its variations, were
366 predominantly determined by the change in the number of visitors on the tower. In
367 addition, the CO₂ mixing ratios on non-working days, especially during the busiest
368 tourist hours, were significantly ($p<0.01$) higher than those on working days. The 450-
369 m platform was closed during October 13-15 as the result of the influence of Typhoon
370 Kompasu. On these days, mixing ratios of ethanol, CO₂, and monoterpenes exhibited
371 similar variation patterns to benzene (a typical tracer of traffic emissions), as shown in
372 Figure 5(c). However, mixing ratios of ethanol, CO₂, and monoterpenes exhibited quite
373 different variation patterns from benzene when the 450-m platform was re-open
374 (October 16–21). For instance, mixing ratios of ethanol, CO₂, and monoterpenes
375 generally decreased from LT 12:00 to 18:00 between October 13–15, but markedly
376 increased during the same period between October 16–21. Therefore, it can be
377 concluded that the VOCs measurements made at 450 m were significantly affected by
378 visitor-related emissions, which will be quantitatively assessed using the PMF analysis
379 in following sections.

380 **3.4 Source analysis of VOCs measurements**

381 In this study, a five-factor solution for the PMF analysis was chosen as the optimal
382 result. Figure 6 displays source profiles ($m/z \leq 150$, the full range of the mass spectra
383 is shown in Figure S7) of the five PMF factors along with average diurnal profiles of
384 their contributions. The five factors were assigned to likely sources of daytime-mixed,
385 visitor-related, vehicular+industrial, regional transport, and volatile chemical product
386 (VCP)-dominated according to characteristics of their source profiles and temporal

387 variations, which are detailedly discussed in the *SI*.

388 The visitor-related source predominantly includes contributions from human
389 breath and volatilization of ethanol-containing and personal care products.
390 Contributions of the visitor-related source had the narrowest autocorrelation profile
391 among the five factors (Figure 6(g)), confirming its most local characteristics. As shown
392 in Figure 7, the visitor-related source had the largest contributions (15.9 ± 19.6 ppb),
393 accounting for 30% of the average TVOC mixing ratio. In addition, contributions of the
394 visitor-related source accounted for a larger fraction of TVOC mixing ratios on non-
395 working days (33%) than those on working days (28%), as shown in Figures 7 and S8.
396 It should be noted that visitor-related emissions belonged to highly local sources on the
397 450 m platform and were not typical of the VOCs measurements in the upper boundary
398 layer. The vehicular+industrial source mainly includes contributions from vehicular
399 exhausts and emissions of various industrial processes. Contributions of the
400 vehicular+industrial source (15.1 ± 18.3 ppb) were comparable to those of the visitor-
401 related source, accounting for 28% of the average TVOC mixing ratio. As anticipated,
402 the vehicular+industrial source contributed to a smaller fraction of TVOC mixing ratios
403 on non-working days (26%) than those on working days (30%). The VCP-dominated
404 source predominantly includes contributions from the volatilization of VCPs in urban
405 environments. The VCP-dominated source had an average contribution of 5.7 ± 5.4 ppb,
406 accounting for 11% of the average TVOC mixing ratio. The average contribution of the
407 VCP-dominated source in this study was comparable to those (~ 6.0 ppb) measured in
408 New York City (Gkatzelis et al., 2021). However, VCPs contributed to over 50% of
409 anthropogenic VOCs emissions in New York City, which is much greater than the
410 fraction in this study (11%, and it will increase to 16% when contributions of the visitor-
411 related source were removed). In comparison to large cities in U.S., traffic and
412 industrial emissions were still dominant sources of ambient VOCs in Chinese cities.
413 However, VCPs emissions should also be given more attention as the VCP-dominated
414 (22%) and vehicular+industrial (23%) sources had comparable contributions to the total

415 OH reactivities, as shown in Figure 7(f).

416 The daytime-mixed source predominantly includes contributions from biogenic
417 emissions and photooxidation products of various VOCs. The daytime-mixed source
418 had an average contribution of 11.6 ± 12.6 ppb, accounting for 21% of the average
419 TVOC mixing ratio. It exhibited consistent diurnal variation patterns on both working
420 and non-working days but had larger contributions in the daytime on non-working days
421 (Figure 6). This may be attributed to the enhanced formation of secondary OVOC
422 species as manifested by the higher ozone concentrations on non-working days (Figure
423 S9). The regional transport source mainly includes contributions from advection
424 transport of aged air masses. Contributions of the regional transport source had the
425 flattest autocorrelation profile, implying its most regional characteristics. Only a small
426 fraction (<5%) of reactive chemical species such as aromatics were attributed to this
427 factor, leading to the lowest contribution to the total OH reactivity. Contributions of the
428 regional transport source accounted for 13% of the TVOC mixing ratio when affected
429 by continental airflows, but only accounted for 3% when affected by marine airflows
430 (Figure S10). By contrast, contributions of the other factors displayed weak
431 dependences on wind direction.

432 As shown in Figure 8, source apportionment of the selected VOC species (Figure
433 3) discussed in section 3.2 were further investigated. The vehicular+industrial source
434 had the largest contribution (36%) to benzene. The daytime-mixed source also
435 contributed to 18% of measured benzene mixing ratios. In addition, more than 20% of
436 benzene was attributed to the VCP-dominated source. In contrast to benzene, toluene
437 was predominantly attributed to the vehicular+industrial (93%) and visitor-related (7%)
438 sources. The average ratio of toluene to benzene was 5.7 ppb/ppb during the CTT
439 campaign (Figure S11), further confirming primary contributions of toluene from
440 vehicular and industrial emissions (Wu et al., 2016; Zhou et al., 2019; Xia et al., 2021).
441 The vehicular+industrial source also accounted for the largest fractions of C8 and C9
442 aromatics. In addition, 26% of C8 aromatics and 38% of C9 aromatics were attributed

443 to the VCP-dominated source. The other three sources in total contributed to less than
444 10% of concentrations of C8 and C9 aromatics. These results indicate that VCPs are
445 important sources of aromatics in urban environments but they were rarely identified
446 in previous studies.

447 Isoprene and monoterpenes are widely known tracers of biogenic emissions
448 (Millet et al., 2016; Zhao et al., 2021). However, the daytime-mixed source only
449 contributed to 16% of measured isoprene mixing ratios. By contrast, more than 70% of
450 isoprene were attributed to the visitor-related (38%) and VCP-dominated (35%) sources.
451 As for monoterpenes, more than 95% of the measured mixing ratios were attributed to
452 the visitor-related (47%) and VCP-dominated (49%) sources. The average ratio of
453 monoterpene to isoprene mixing ratios at 450 m was 0.84 in the daytime (LT 08:00–
454 18:00), which was significantly ($p < 0.01$) greater than that at the ground level (0.05)
455 (Figure S11). It further confirms strong contributions of monoterpenes from visitor-
456 related emissions at the 450-m platform. The daytime-mixed source did not exhibit
457 discernible contributions to monoterpenes. This agrees well with the results in New
458 York City where monoterpene mixing ratios were primarily attributed to anthropogenic
459 sources such as VCPs, cooking, and building materials (Coggon et al., 2021; Gkatzelis
460 et al., 2021). These results suggest that emission intensities of isoprene and
461 monoterpenes may be highly underestimated in urban regions if their anthropogenic
462 emissions are overlooked or less considered. This is exceedingly important for air
463 quality models when estimating formation of ozone and secondary organic aerosol
464 driven by the oxidation of isoprene and monoterpene. As the key photooxidation
465 products of isoprene, nearly 60% of MVK+MACR were attributed to the daytime-
466 mixed source. The visitor-related, regional transport, and VCP-dominated sources
467 contributed to comparable fractions (11%–15%) of MVK+MACR. Therefore,
468 anthropogenic emissions are also important sources of MVK+MACR in urban
469 environments.

470 As shown in Figure 8, 39% of acetone was attributed to the daytime-mixed source.

471 The vehicular+industrial (19%) and VCP-dominated (21%) sources accounted for
472 comparable fractions of measured acetone mixing ratios. The visitor-related source had
473 the lowest contribution (7%) to acetone. As for methanol, the vehicular+industrial
474 source accounted for the largest fraction (38%), followed by the daytime-mixed (22%),
475 regional transport (17%), VCP-dominated (14%), and visitor-related (9%) sources.
476 These results reveal that VCPs also contributed significantly to ambient concentrations
477 of acetone and methanol and should be carefully considered when estimating their total
478 emission intensities from anthropogenic sources. Ethanol was predominantly attributed
479 to the visitor-related source. Therefore, the enhanced ethanol mixing ratios were not
480 capable of representing its characteristic concentrations in urban environments.
481 Although the absence of synchronous ground-level measurements, we can speculate
482 that ethanol concentrations at ground level were also increased during the outbreak of
483 the COVID-19 pandemic due to the extensive usage of ethanol-containing products.
484 The enhancement of ethanol concentrations can contribute significantly to the increase
485 in atmospheric OH reactivity (Millet et al., 2012; de Gouw et al., 2017; de Gouw et al.,
486 2018) and then regulate the formation of secondary pollutants. Therefore, impacts of
487 the ethanol enhancement on ambient air quality should be explicitly investigated in
488 future studies due to the wide report of ozone enhancement during the outbreak of the
489 COVID-19 pandemic (Huang et al., 2020; Qi et al., 2021).

490 Acetonitrile is widely used as a typical tracer of biomass burning sources in
491 previous studies (de Gouw et al., 2003b; Zhang et al., 2020; Tan et al., 2021). However,
492 biomass burning source was not identified in this study because acetonitrile was not
493 predominantly attributed to a single factor. In addition to the visitor-related source, the
494 other four sources also had large contributions to acetonitrile. As indicated by (Huangfu
495 et al., 2021), it is not always suitable, particularly in urban environments, to use absolute
496 concentrations of acetonitrile as the indication of biomass burning sources. The ratio of
497 acetonitrile to CO is a better indicator to identify whether VOCs measurements are
498 predominantly contributed by biomass burning emissions. The average ratio of

499 acetonitrile to CO was only 0.09 (ppb ppm⁻¹) during the CTT campaign (Figure S11),
500 indicating negligible contributions from biomass burning sources. In addition to the
501 daytime-mixed (22%) and vehicular+industrial (26%) sources, the VCP-dominated
502 source (31%) also had large contributions to acetonitrile in urban environments.

503 **3.5 Vertical distributions of air pollutants concentrations**

504 As introduced in section 2.1, hourly concentrations of some air pollutants were
505 routinely measured at four automatic sites on the CTT. Figure 9 shows contour plots of
506 vertical profiles of NO_x, ozone, Ox (O₃+NO₂), and PM_{2.5} concentrations in September
507 2020. Concentrations of the four pollutants all exhibited stratified structures between
508 488 m and the ground level. Higher mixing ratios of ozone and Ox predominantly
509 occurred at higher altitudes, while higher NO_x mixing ratios mainly occurred at ground
510 level. By contrast, higher PM_{2.5} concentrations were observed at both middle altitudes
511 and ground level.

512 To further clarify vertical distribution patterns of air pollutants concentrations,
513 their composite profiles for daytime (LT 08:00–18:00), nighttime (LT 19:00–05:00),
514 and the whole day in the campaign were determined, respectively, as shown in Figure
515 10. Vertical profiles of air pollutant concentrations exhibited similar shapes both in
516 daytime and nighttime. NO_x mixing ratios decreased from the ground level to 488 m,
517 suggesting intensive surface emissions around the CTT. Ozone mixing ratios rapidly
518 increased from the ground level to 488 m, which was consistent with the results reported
519 in previous studies (Velasco et al., 2008; Li et al., 2018; Zhang et al., 2019; Li et al.,
520 2021b). The positive gradients of ozone profiles are mainly caused by enhanced NO
521 titration (NO+O₃=O₂+NO₂) and dry deposition near ground. Ox mixing ratios also
522 increased from the ground level to 488 m but exhibited weaker gradients in comparison
523 to ozone. Vertical profiles of PM_{2.5} concentrations exhibited similar shapes to NO_x.
524 Daily mean concentrations of PM_{2.5} and Ox were well correlated at the four altitudes
525 with *r* values varying in the range of 0.61–0.82, suggesting prominent contributions of
526 secondary formation to ambient PM concentrations. Moreover, the correlation

527 coefficients between Ox and PM_{2.5} concentrations at 488 m (0.82) were greater than
528 those at ground level (0.78), as they were less affected by nearby vehicular emissions.
529 This is consistent with the work by (Yan et al., 2020), who reported that secondary
530 components contributed to ~80% of PM_{2.5} concentrations in PRD over the 2008–2019
531 period.

532 As shown in Figures 9 and 10, vertical profiles of air pollutant concentrations
533 exhibited weaker gradients in the daytime than in the nighttime. Therefore, the daytime
534 VOCs chemistry may have minor differences between the ground level and the 450-m
535 site due to strong vertical mixing of chemical species in the planetary boundary layer
536 (PBLH>450 m, as shown in Figure S12). In the nighttime, the oxidative products (such
537 as organic nitrates and OVOCs) of unsaturated hydrocarbons, predominantly initiated
538 by nitrate radicals (NO₃) and ozone, are also important precursors of secondary aerosol
539 (Warneke et al., 2004; Brown et al., 2011; Ng et al., 2017; Liebmann et al., 2019).
540 However, it is highly challenging to investigate the nighttime VOCs chemistry with
541 only ground-level measurements due to the rapid removal of NO₃ radicals and ozone
542 by enhanced NO titration (Geyer and Stutz, 2004; Stutz et al., 2004; Brown et al., 2007).
543 In this condition, the nocturnal residual layer, separated from nocturnal boundary layer
544 and remained, to a large extent, the chemical composition of the daytime atmosphere,
545 could provide an ideal place for investigating nighttime VOCs chemistry. Oxidative
546 products of VOCs in the residual layer could be mixed downward with the expansion
547 of the PBL during the daytime (Geyer and Stutz, 2004; Stutz et al., 2004; Li et al.,
548 2021a), contributing to the formation of ozone and secondary aerosol at ground level.
549 Investigation of the nighttime VOCs chemistry was one of the initial purposes of this
550 study. Unfortunately, the 450-m site was rarely located in the nocturnal residual layer
551 during the CTT campaign due to frequent occurrences of cloudy and rainy weather. The
552 average nighttime PBLH in Guangzhou was approximately stabilized at 500 m during
553 the campaign (Figure S12), implying notable impacts from surface emissions on the
554 measurements made at 450 m.

555 In addition to the measured PBLH data, formation of the residual layer at 450 m
556 could be also identified by comparing differences of ozone mixing ratios between 488
557 m and the ground level. Without fresh NO emissions, ozone mixing ratios in the
558 nocturnal residual layer were markedly higher than at ground level and exhibited weak
559 variability throughout the nighttime (Caputi et al., 2019; Udina et al., 2020). By contrast,
560 surface ozone mixing ratios are generally very low (close to zero) due to enhanced
561 titration by freshly emitted NO and strong inhibition of atmospheric vertical mixing
562 (Ma et al., 2011; Chen et al., 2020). In this study, the data collected between September
563 27–30 was one of the cases discussed above and was used to briefly describe behaviors
564 of some representative VOC species (namely ethanol, monoterpene, styrene, phenol,
565 and toluene) at 450 m.

566 As shown in Figure 11, ozone mixing ratios measured at ground level (10.2 ± 10.4
567 ppb) were significantly ($p < 0.01$) lower than those at 488 m (44.2 ± 19.6 ppb) on the
568 nighttime of September 27–30, indicating formation of the nocturnal residual layer
569 lower than 450 m. On the nighttime of September 27–28, ozone mixing ratios at 488 m
570 slightly fluctuated around 46.8 ppb between LT 19:00–00:00 and suddenly decreased
571 to 28.4 ppb at LT 01:00 on September 28. The sudden decrease in ozone at LT 01:00
572 was accompanied by slight increases in both NO_x and VOCs but notable decreases in
573 NO_x and NO at ground level, indicating a transitory intrusion of surface fresh emissions
574 into the residual layer. On September 28, ozone mixing ratios at 488 m slightly
575 decreased from 33.0 to 31.5 ppb from LT 02:00 to 05:00, during which mixing ratios
576 of NO_x and VOCs all decreased in different degrees. The continuous decreases in both
577 toluene and ethanol between LT 02:00–05:00 confirm that the VOCs measurements at
578 450 m were free of interferences by fresh emissions due to their large contributions
579 from vehicular exhausts (Figure 8). Toluene mixing ratios decreased by 43% from LT
580 02:00 to 05:00, which was larger than those (12–27%) of the other VOC species shown
581 in Figure 11. However, the NO₃ reactivity (characterized by reaction rate constants of
582 VOC species to NO₃ radical, k_{NO_3}) of toluene ($k_{NO_3} = 7 \times 10^{-17}$ cm³ molecule⁻¹ s⁻¹) is

583 exceedingly lower than those of the other unsaturated VOC species (k_{NO_3} varies in the
584 magnitudes of 10^{-12} cm³ molecule⁻¹ s⁻¹) (Atkinson and Arey, 2003; Atkinson et al.,
585 2006). Therefore, the decline of unsaturated VOC species in the nocturnal residual layer
586 may not be all attributed to the degradation chemistry initiated by NO₃ radicals or ozone.

587 On the nighttime of September 28–29, the PBLH was higher than 500 m between
588 LT 19:00–00:00, resulting in notable decreases in ozone and increases in NO_x and
589 VOCs. As shown in Figure 11, the 450-m site may locate in the residual layer after LT
590 01:00. However, the rapid decrease in mixing ratios of NO_x and VOCs between LT
591 01:00–05:00 were not likely caused by chemical removal due to the rapid increase in
592 ozone. Regional transport of aged air masses (characterized by high ozone and low NO_x
593 mixing ratios) may be responsible for the rapid decline in various VOC species in the
594 early morning of September 29. On the nighttime of September 29–30, the 450-m site
595 may be impacted by surface fresh emissions as mixing ratios of ozone, NO_x, and VOCs
596 all decreased between LT 19:00–01:00 and simultaneously increased between LT
597 02:00–05:00. NO_x and toluene mixing ratios generally increased between LT 12:00–
598 18:00 during September 27–29, which were quite different from their average diurnal
599 variation patterns during the whole campaign (Figures 3 and 4). As discussed above,
600 the 450-m site was located in the nocturnal residual layer during September 27–29.
601 Therefore, emissions of pollutants from surface sources could be mixed upward to the
602 measurement site only when the PBLH was higher than 450 m. Furthermore, the PBL
603 was relatively lower and rapidly shrank in the afternoon, leading to the accumulation
604 of chemical species at 450 m.

605 In summary, the VOCs measurements made by PTR-ToF-MS at the 450-m site
606 can be used to characterize variations in VOC species from their primary emissions
607 during the nighttime. Nevertheless, the oxidative degradation processes of VOCs in the
608 nighttime were not well captured. It is highly difficult to provide more information on
609 the nighttime chemistry of VOC species solely depending on their temporal variations.
610 We believe that the oxidative degradation of reactive VOC species did occur in the

611 nocturnal residual layer due to the coexistence of high concentrations of NO_x and ozone.
612 Measurement techniques that targeting oxidation products (e.g., ToF-CIMS) and
613 numerical models should be jointly used to deeply analyze the nighttime chemistry of
614 VOCs in the nocturnal residual layer and quantitatively evaluate their impacts on
615 ambient air quality during the daytime.

616 **4 Conclusions**

617 Continuous measurements of VOCs mixing ratios were made by PTR-ToF-MS at
618 450 m on the CTT in PRD, China from August 18–November 5, 2020. In addition to
619 some specific VOC species (such as ethanol and monoterpenes) that were intensively
620 emitted by visitor-related sources, mixing ratios of most VOC species at 450 m were
621 generally lower than those at ground level. Due to intensive emissions from visitor-
622 related sources, mixing ratios of some VOC species were significantly higher on non-
623 working days than those on working days. The VOCs mixing ratios measured at 450 m
624 also exhibited different diurnal variations from those at ground level, indicating that
625 they were contributed by more mixed sources at larger spatial scales. Five sources,
626 namely daytime-mixed, visitor-related, vehicular+industrial, regional transport, and
627 VCP-dominated, were determined by the PMF model, contributing to 21%, 30%, 28%,
628 10%, and 11% of the average TVOC mixing ratio, respectively. In addition to the
629 daytime-mixed and visitor-related sources, the other three sources all had relatively
630 lower contributions on non-working days than on working days. The VCP-dominated
631 source contributed an average of 5.7 ppb to TVOC mixing ratios, which was
632 comparable to those reported in American cities ([Gkatzelis et al., 2021](#)). However, the
633 VCP-dominated source accounted for a much smaller fraction (11%) of measured
634 TVOC mixing ratios in this study than in U.S. cities (>50%). Therefore, the reduction
635 in anthropogenic VOC emissions from traffic and industrial sources are still priorities
636 of current air pollution control for Chinese cities. Though smaller fraction of VOCs
637 contributed by VCPs was observed in this study compared to cities in U.S. ([McDonald](#)

638 [et al., 2018](#); [Gkatzelis et al., 2021](#)), large fractions of key VOC species (such as
639 monoterpenes and some aromatic species) were attributed to the VCP-dominated
640 source. In addition, the VCP-dominated (22%) and vehicular+industrial sources (23%)
641 had comparable contributions to the total OH reactivity. Therefore, VCPs emissions
642 should be given more attentions when making control strategies for VOCs in urban
643 region..

644 The vertical distribution patterns of NO_x, ozone, O_x, and PM_{2.5} concentrations
645 were investigated using measurements made at four different heights on the CTT.
646 Vertical profiles of NO_x and PM_{2.5} generally exhibited negative gradients, while
647 vertical profiles of ozone demonstrated positive gradients. In addition, the vertical
648 gradients of air pollutant concentrations were larger in the nighttime than in the daytime,
649 predominantly owing to stronger stability of the nocturnal boundary layer. The 450-m
650 site was rarely located in the nocturnal residual layer as cloudy and rainy weather
651 dominated during the campaign. The selected case revealed that the NO₃⁻ or O₃-
652 initiated degradation chemistry may be not the sole path for the removal of unsaturated
653 VOC species in nighttime. The degradation chemistry of reactive VOC species in the
654 nocturnal residual layer and their impacts on ground-level air quality could be further
655 investigated in combination with model simulations in future studies.

656 **Data availability**

657 The observational data used in this study are available from corresponding authors
658 upon request.

659 **Author contributions**

660 XBL and BY designed the research. XBL, BY, SHW, CLW, JL, ZJL, XJH, YBHF,
661 CLP, CP, JPQ, CHW, YCY, MC, HDZ, WDY, XMW, and MS contributed to the data
662 collection and data analysis. XBL and BY performed the PMF analysis with
663 contributions from YXS, SXY, and SH. XBL and BY wrote the paper. All the coauthors

664 discussed the results and reviewed the paper.

665 **Competing interests**

666 The authors declare that they have no conflict of interest.

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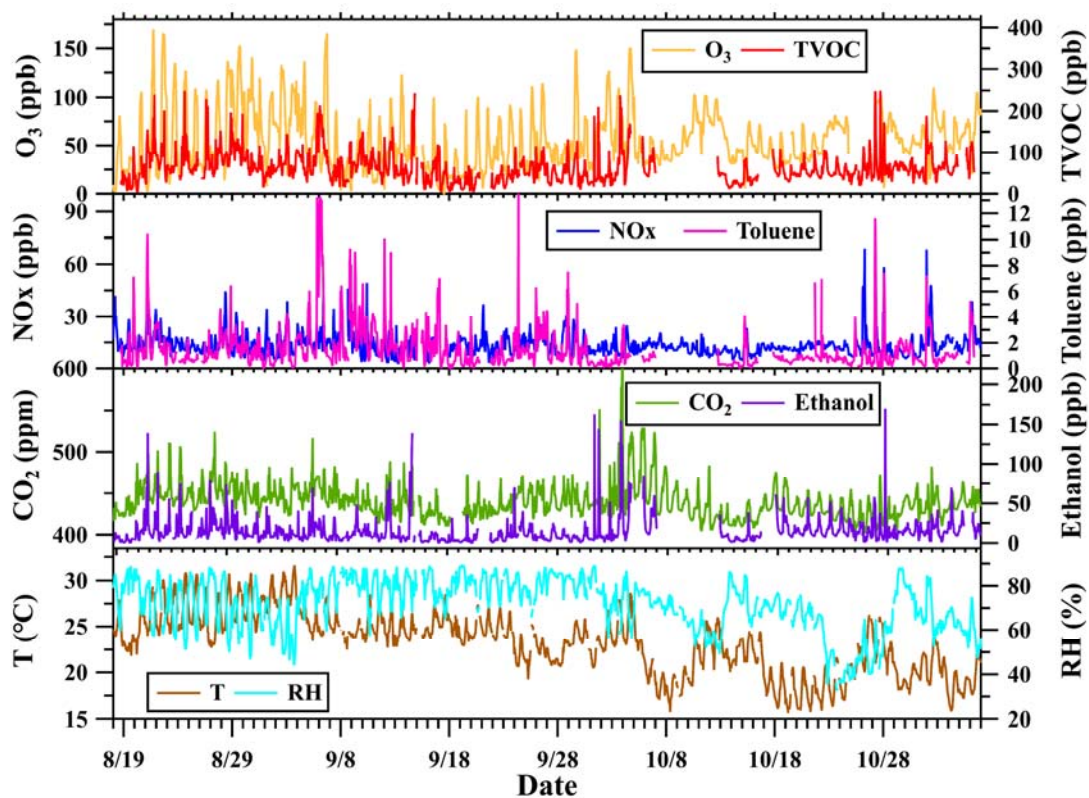
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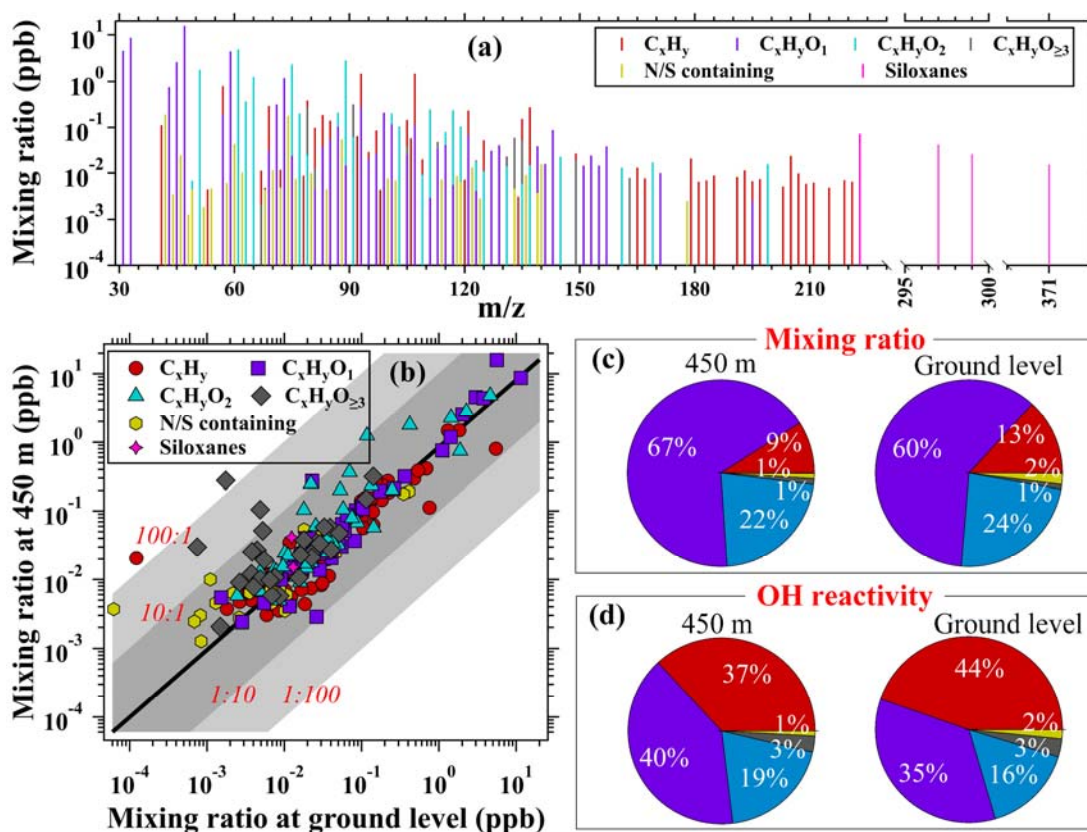
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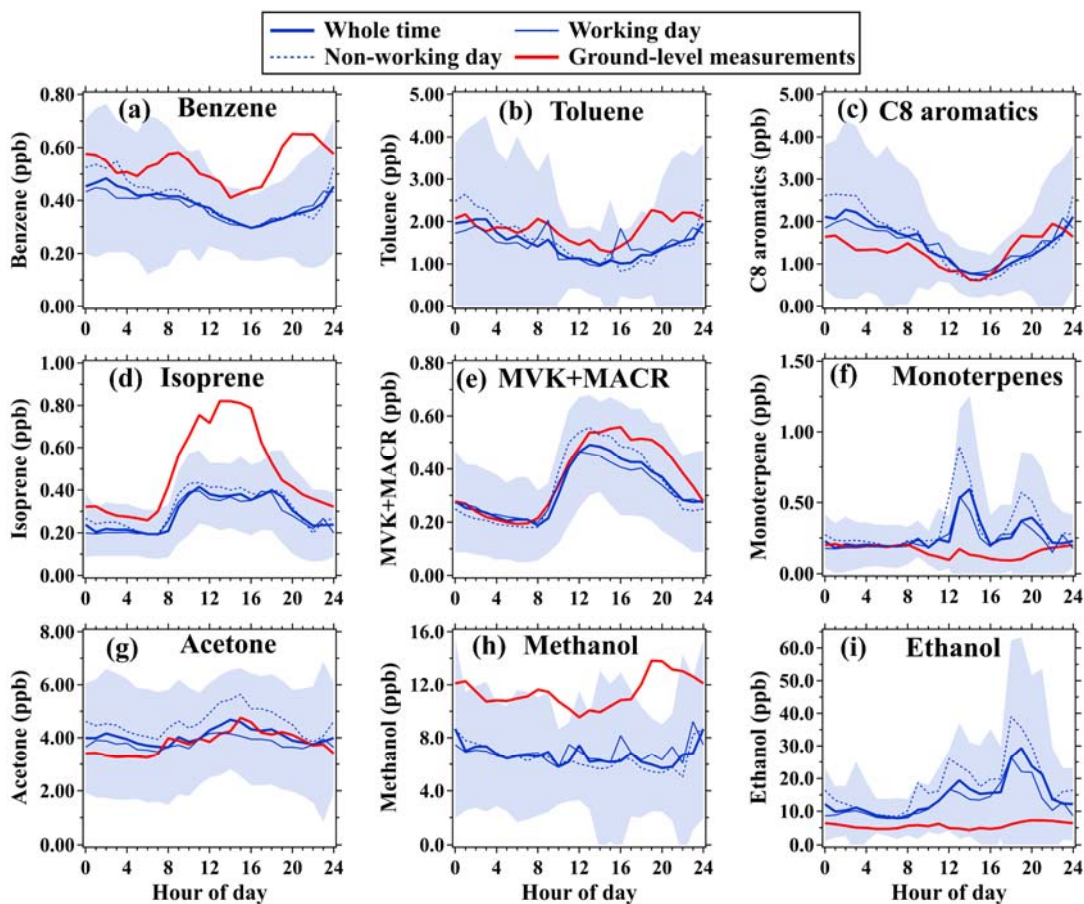
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1135 **Figure 1.** Time series of concentrations of some typical chemical species along with
 1136 meteorological parameters (hourly averages) during the CTT campaign. Temperature
 1137 (T), relative humidity (RH), concentrations of ozone and NOx were measured at 488
 1138 m. Concentrations of VOCs, ethanol, and CO₂ were measured at 450 m.



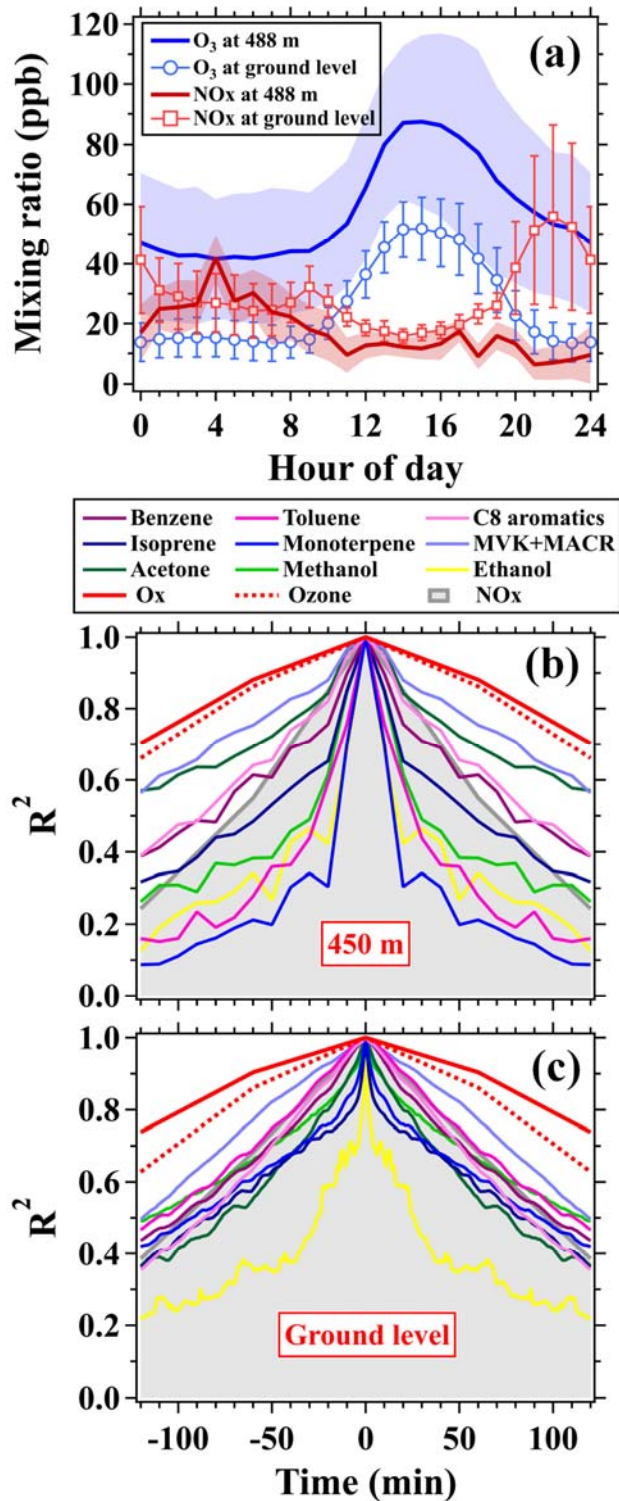
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1140 **Figure 2.** (a) Average mass spectra of VOCs (including 225 species) obtained by PTR-
 1141 ToF-MS measured at 450 m during the CTT campaign. (b) Scatter plots of the average
 1142 VOC mixing ratios measured at 450 m during the CTT campaign versus those measured
 1143 at ground level during the GIG campaign; The black solid line indicates the ratio of 1:1;
 1144 The dark grey shaded areas indicate the ratios of 10:1 and 1:10; The light grey shaded
 1145 areas indicate the ratios of 100:1 and 1:100. (c-d) Average contribution percentages of
 1146 the six VOCs categories to their total concentrations and OH reactivities at 450 m and
 1147 the ground level, respectively; Only the VOCs species that have known reaction rate
 1148 constants with OH radical (Table S1) were used for calculation.



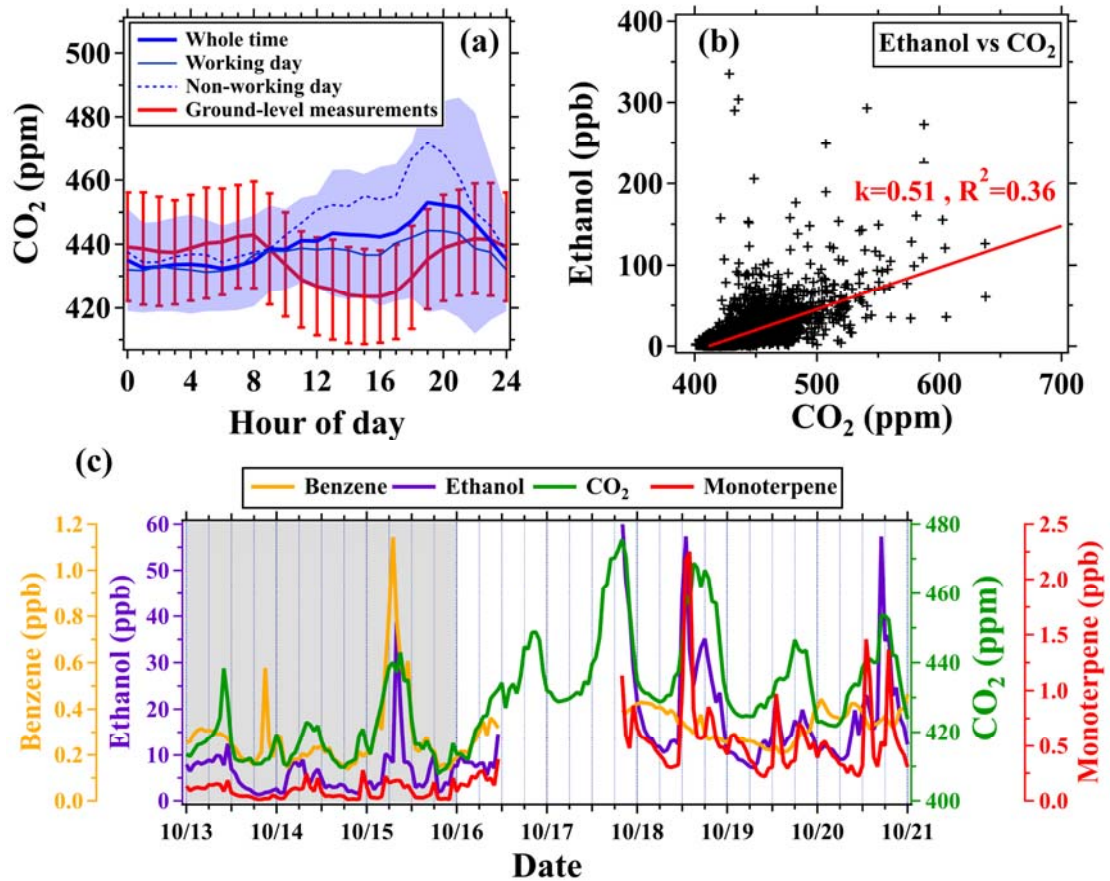
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1150 **Figure 3.** Diurnal variations in mixing ratios of selected VOC species measured by
 1151 PTR-ToF-MS. Thick blue solid lines and shaded areas represent averages and standard
 1152 deviations, respectively, during the CTT campaign (August 18–November 05, 2020).
 1153 Red solid lines represent averages during the GIG campaign (September 11–November
 1154 19, 2018). Thin blue solid and dashed lines represent averages in working days and
 1155 non-working (including weekends and public holidays) days, respectively, during the
 1156 CTT campaign.

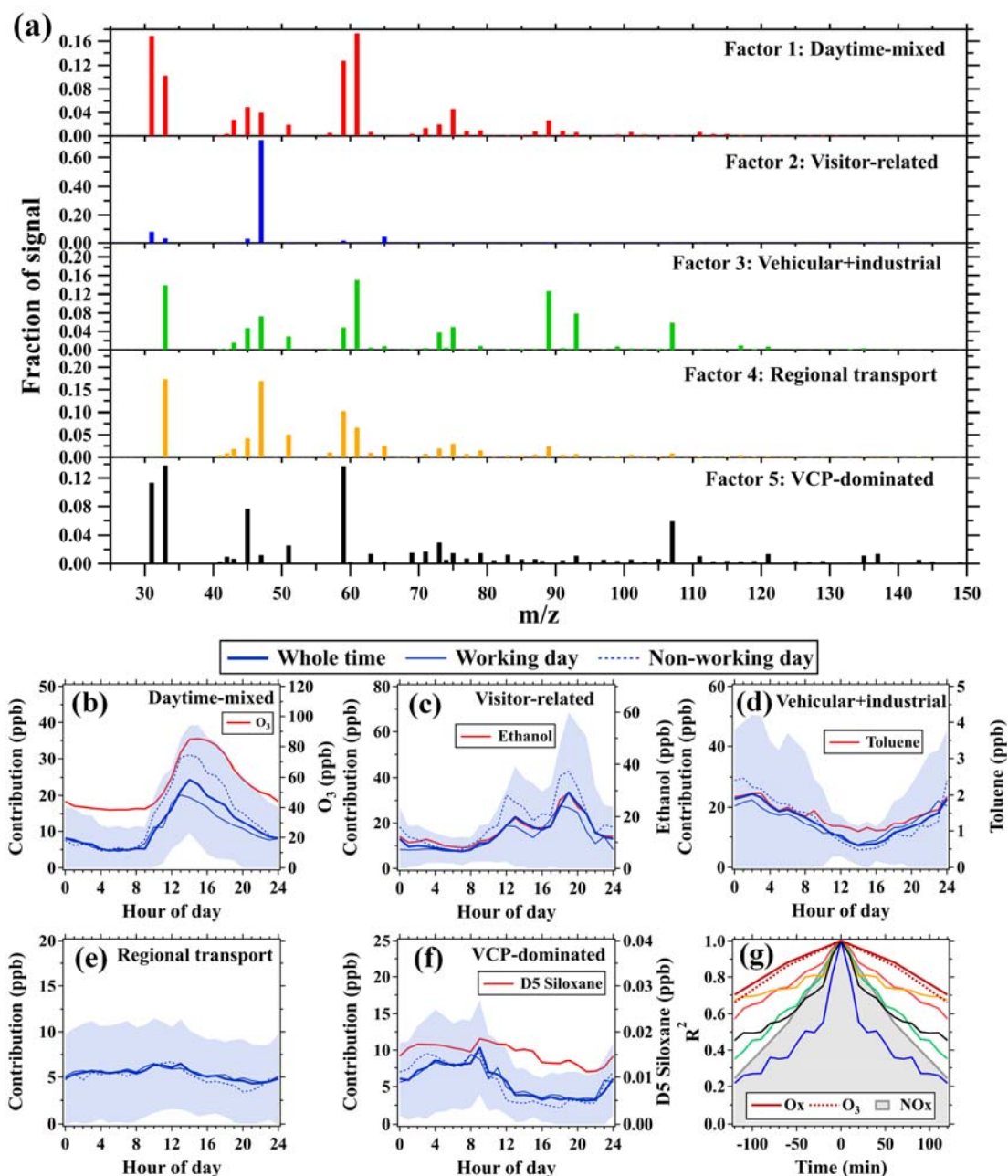


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 1158 **Figure 4.** (a) Diurnal profiles of ozone and NOx mixing ratios measured at the 488-m
 1159 site (mean \pm standard deviation) and the surface site (mean \pm 0.5 standard deviation)
 1160 on the CTT. (b) Autocorrelation of the time series of ozone (488 m), NOx (488 m), Ox
 1161 (488 m), and selected VOC species (450 m) during the CTT campaign. (c)
 1162 Autocorrelation of the time series of the selected VOC species at ground level during

1163 the GIG campaign; Autocorrelation of the time series of ozone, NO_x, and O_x in panel
1164 (c) are calculated using the measurements made at the surface site of Canton Tower
1165 during the CTT campaign.



1166
 1167 **Figure 5.** (a) Diurnal variations in CO₂ mixing ratios at 450 m and the ground level,
 1168 respectively. (b) Scatterplots of 10-min mean mixing ratios of ethanol versus CO₂
 1169 measured at 450 m during the CTT campaign.(c) Time series of benzene, ethanol, CO₂,
 1170 and monoterpene mixing ratios measured at 450 m from October 13 to 21; The grey
 1171 shaded area indicates the period (October 13–21) when the 450-m platform was closed
 1172 due to the influence of Typhoon Kompasu.



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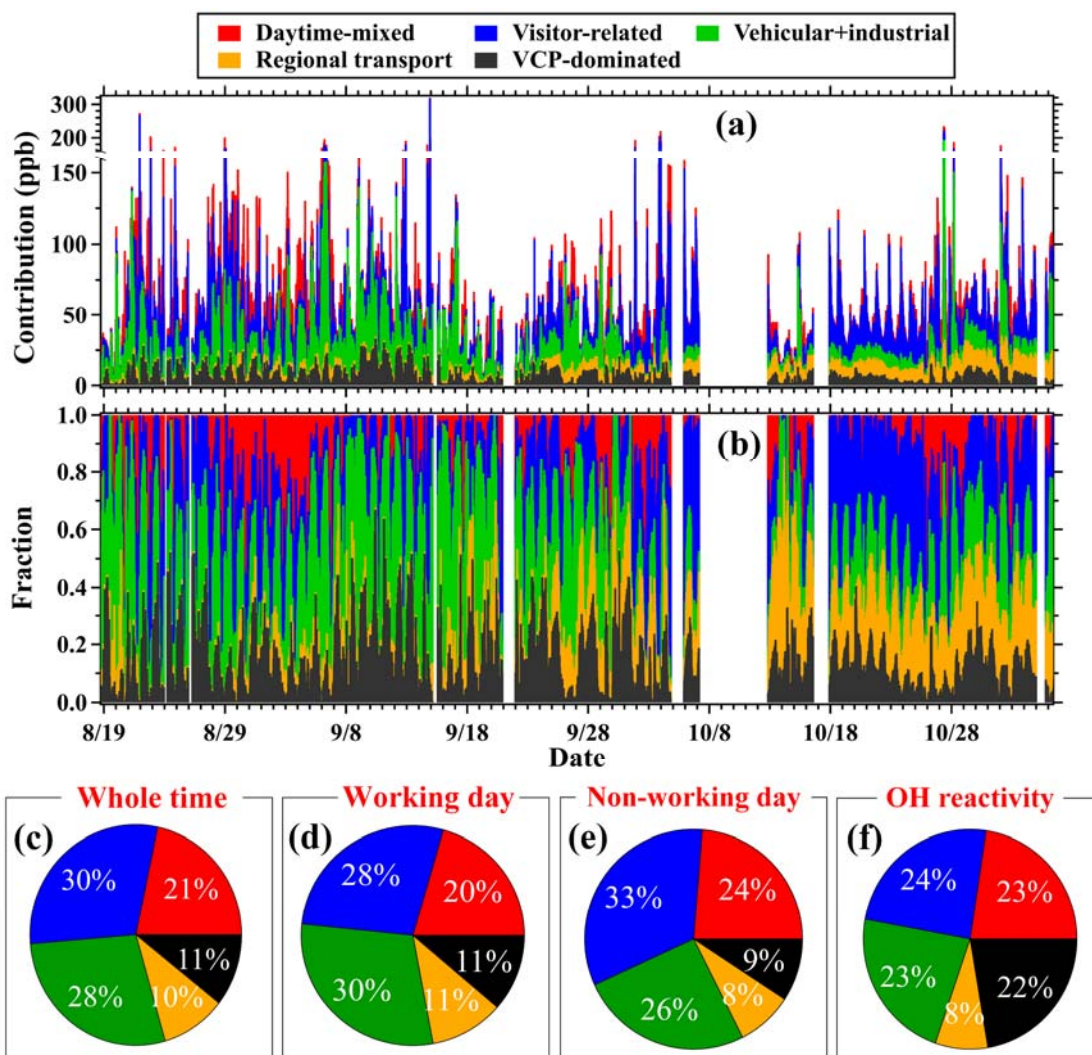
1174 **Figure 6.** (a) Factor profiles ($m/z \leq 150$) of the five PMF factors; Factor profiles with

1175 a full range of the mass spectra are provided in Figure S7. (b-f) Average diurnal profiles

1176 of the five PMF factors and source tracers. (g) Autocorrelation of the time series of the

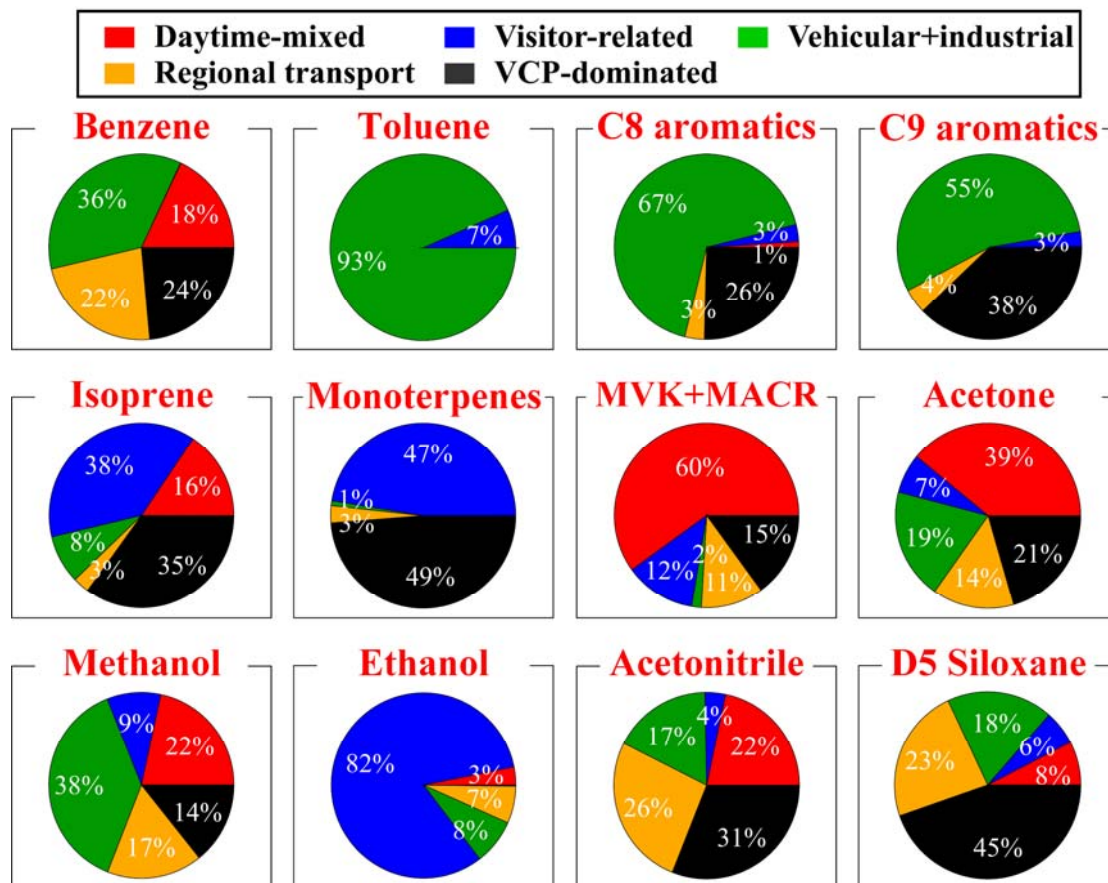
1177 five PMF factors along with Ox, ozone, and NO_x mixing ratios at 488 m; Colors of

1178 lines are consistent with the five factors in panel (a).



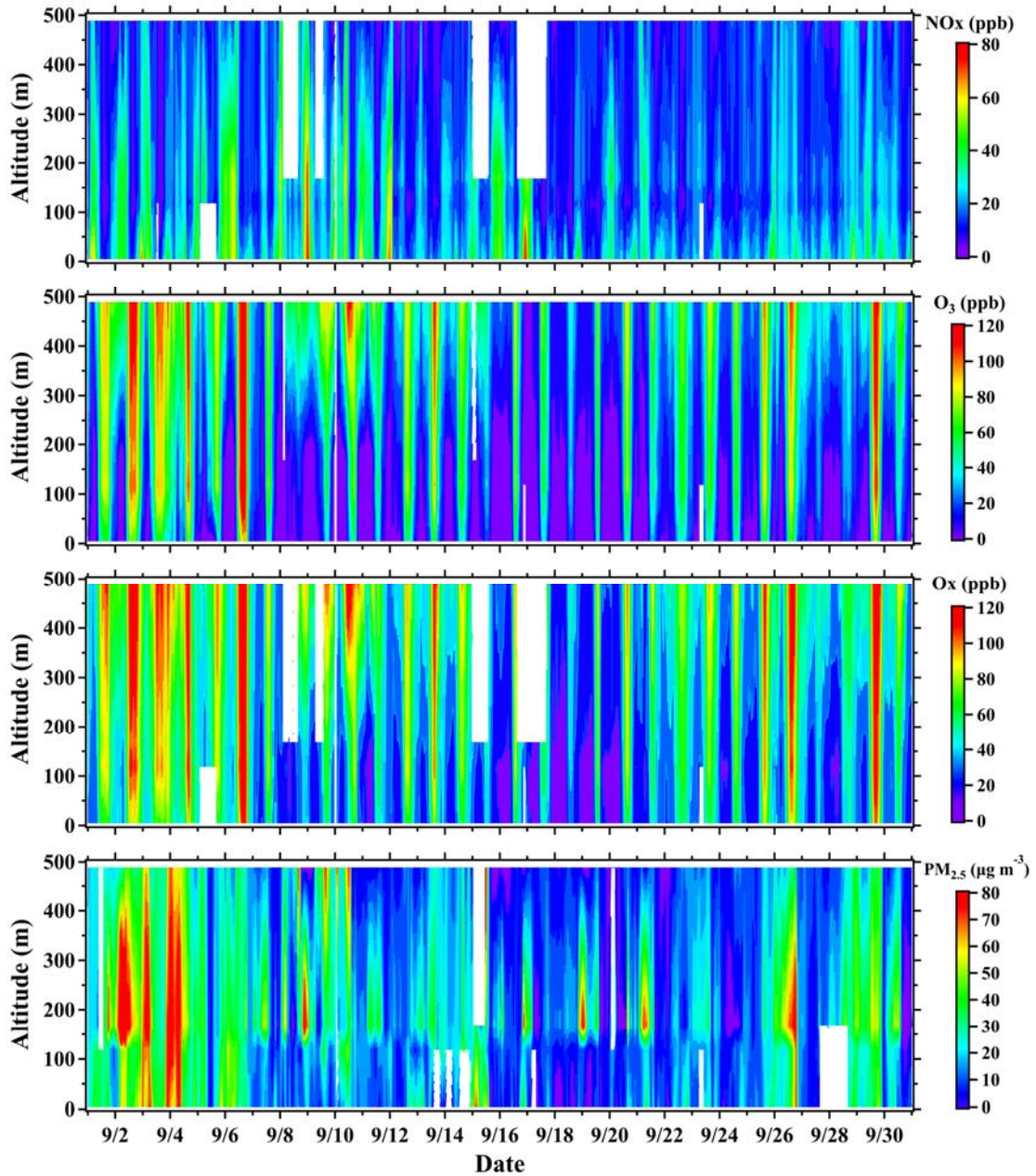
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1180 **Figure 7.** (a-b) Stacked time series of factor fractions and factor contributions for the
 1181 PMF analysis. (c-e) Average contribution percentages of the five PMF factors to (c-e)
 1182 the total VOCs concentrations in the whole time, working days, and non-working days
 1183 and (f) the total OH reactivities during the CTT campaign. In panel (d), only the VOCs
 1184 species that have known reaction rate constants with OH radical (Table S1) were used.



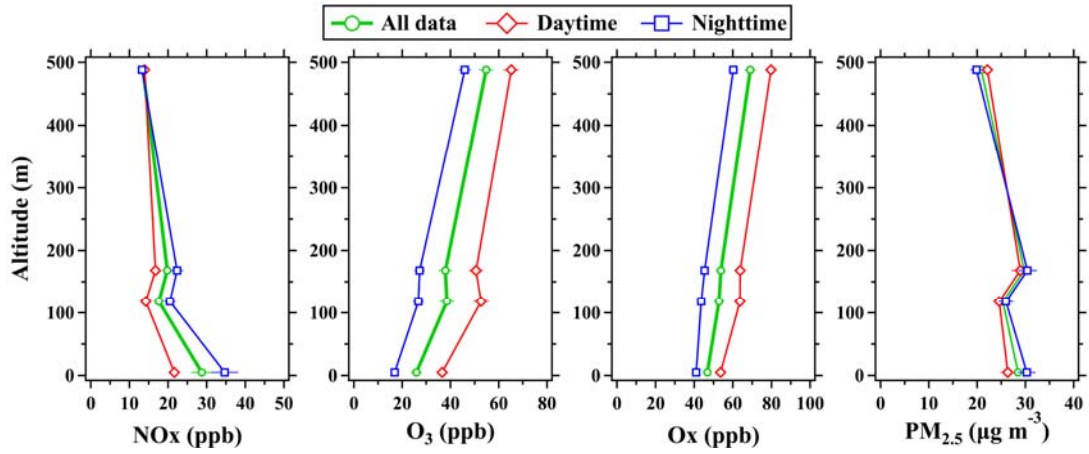
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1186 **Figure 8.** (a) Average contribution percentages of the five PMF factors to
 1187 concentrations of the 9 selected VOC species during the CTT campaign.



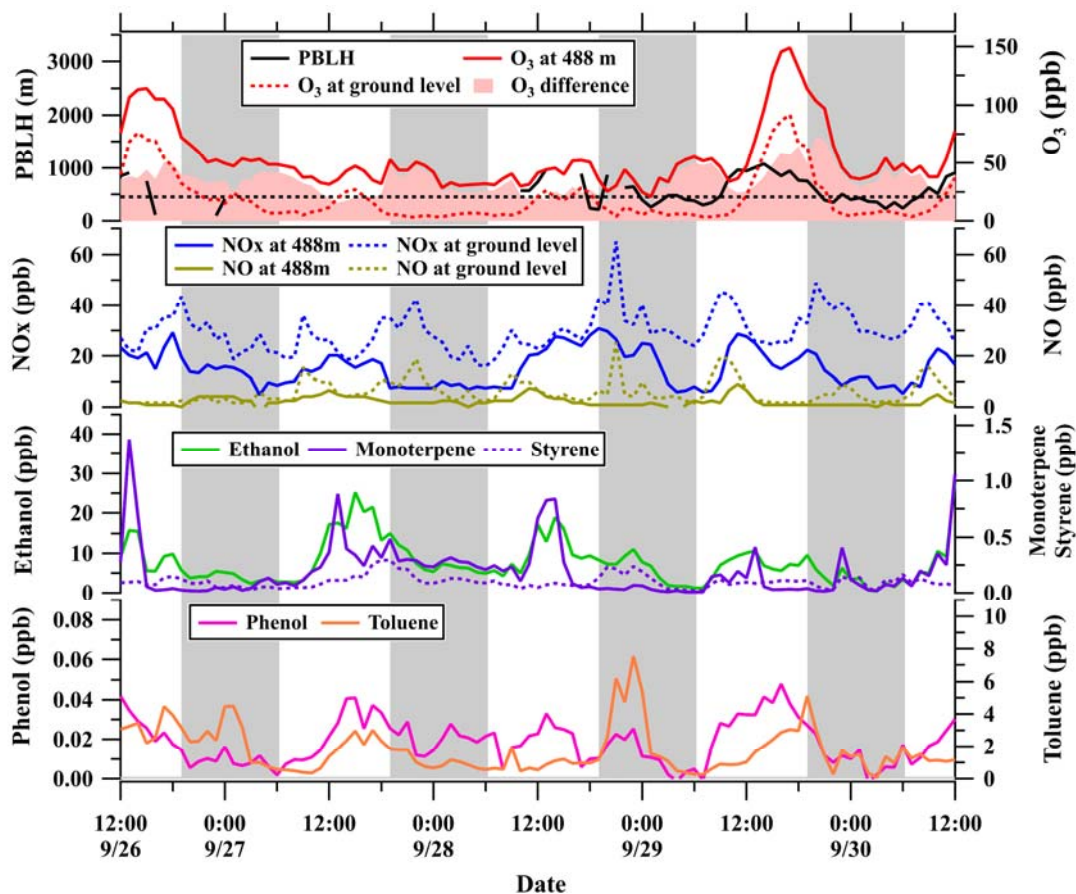
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1189 **Figure 9.** Time series of vertical profiles for O₃, NO_x, Ox (O₃+NO₂), and PM_{2.5}
 1190 concentrations in September during the CTT campaign. The contour plots are made
 1191 using the measurements from the four CTT sites (5 m, 118 m, 168 m, and 488 m).



1192

1193 **Figure 10.** Average vertical profiles of O₃, NO_x, Ox (O₃+NO₂), and PM_{2.5}
 1194 concentrations (mean ± 0.1 standard deviations) measured at the four CTT sites (5 m,
 1195 118 m, 168 m, 488 m) during the campaign. Daytime refers to the time between LT
 1196 08:00–18:00; nighttime refers to the time between LT 19:00–05:00.



1197

1198 **Figure 11.** Time series of O₃, NO_x, NO, ethanol, monoterpene, styrene, phenol, and
 1199 toluene mixing ratios along with planetary boundary layer height (PBLH) during
 1200 September 26–30. O₃ difference refers to the differences in ozone mixing ratios
 1201 between 488 m and 5 m. Grey shaded areas indicate nighttime periods (LT 19:00–
 1202 05:00).