



1 Exploring the inorganic composition of the Asian Tropopause  
2 Aerosol Layer using medium-duration balloon flights

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7 *Hazel Vernier<sup>1</sup>, Neeraj Rastogi<sup>2</sup>, Hongyu Liu<sup>3,4</sup>, Amit Kumar Pandit<sup>3</sup>, Kris Bedka<sup>4</sup>, Anil Patel<sup>2</sup>,*  
8 *Madineni Venkat Ratnam<sup>5</sup>, Buduru Suneel Kumar<sup>6</sup>, Bo Zhang<sup>3</sup>, Harish Gadhavi<sup>2</sup>, Frank*  
9 *Wienhold<sup>7</sup>, Gwenael Berthet<sup>1</sup>, Jean-Paul Vernier<sup>3,4</sup>*

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11 1. Laboratoire de Physique et Chimie de l'Environnement et de l'Espace (LPC2E), France  
12 2. Physical Research Laboratory, Ahmedabad, India  
13 3. National Institute of Aerospace, Hampton, VA, USA  
14 4. NASA Langley Research Center, Hampton, VA, USA  
15  
16 5. National Atmospheric Research Laboratory, Gadanki, India  
17 6. TIFR Balloon Facility, Hyderabad, India  
18 7. ETH, Zürich, Switzerland

19  
20 Correspondence to: hazel.vernier ([hazel@cnrs-orleans.fr](mailto:hazel@cnrs-orleans.fr))

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33 **Abstract.** Satellite observations have revealed an enhanced aerosol layer near the tropopause  
34 over Asia during the summer monsoon, called the Asian Tropopause Aerosol Layer (ATAL). In  
35 this work, aerosol particles in the ATAL were collected with a balloon-borne impactor near the  
36 tropopause region over India, using extended duration balloon flights, in summer 2017 and  
37 winter 2018. Their chemical composition was further investigated by quantitative analysis using  
38 offline ion chromatography. Nitrate ( $\text{NO}_3^-$ ) and nitrite ( $\text{NO}_2^-$ ) were found to be the dominant ions  
39 in the collected aerosols with values ranging between 87-343  $\text{ng/m}^3$  STP during the summer  
40 campaign. In contrast, sulfate ( $\text{SO}_4^{2-}$ ) levels were found above the detection limit ( $>10 \text{ ng/m}^3$   
41 STP) only in winter. In addition, we determined the origin of the air masses sampled during the



1 flights through analysis of back trajectories along with convective influence. The results obtained  
2 therein were put into a context of large-scale transport and aerosol distribution with GEOS-Chem  
3 chemical transport model simulations. The first flight of summer 2017 which sampled air mass  
4 within the Asian monsoon anticyclone (AMA), influenced by convection over Western China,  
5 was associated with particle size radius (0.05-2 $\mu$ m). In contrast, the second flight sampled air  
6 mass at the edge of the AMA associated with larger particle size radius (>2 $\mu$ m) with higher  
7 nitrite concentration. The sampled air masses in winter 2018 were likely affected by smoke from  
8 the Pacific Northwest fire event in Canada, which occurred 7 months prior to our campaign,  
9 leading to concentration enhancements of SO<sub>4</sub><sup>2-</sup> and Ca<sup>2+</sup>. Overall, our results suggest that  
10 nitrogen-containing particles represent a large fraction of aerosols populating the ATAL, in  
11 agreement with the results from aircraft measurements during the StratoClim campaign.  
12 Furthermore, GEOS-Chem model simulations suggest that lightning NO<sub>x</sub> emissions had a  
13 significant impact on the production of nitrate aerosols sampled during the summer 2017.

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## 1. Introduction

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### 1.1 Asian Summer Monsoon and the transport of pollution

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Rapid economic growth in Asia over the past two decades have led to serious environmental threats on water and air qualities. Every winter, pollutants can be observed through satellites in the form of a grayish veil of particulate matter referred to as the Asian Brown Cloud (Ramanathan and



1 Crutzen, 2003). Temperature inversion blocks dangerous life-threatening pollutants near the  
2 ground such as particles with diameter less than 2.5 microns ( $PM_{2.5}$ ) with levels up to 10 times  
3 higher than the World Health Organization recommendations. In summer, the Southwest Asian  
4 Monsoon (SAM) discharges polluted air over very long distances. About 20% of air masses in the  
5 tropical lower stratosphere have been in contact with air in the boundary layer in Asia according  
6 to trajectory calculations (Orbe et al., 2015). Polluted air masses transported from the boundary  
7 layer to higher altitudes are confined within the Asian Monsoon Anticyclone (AMA) (Ploeger et  
8 al., 2017). In the AMA, pollution is accumulated and is further dispersed over a large area of the  
9 Northern Hemisphere reaching longitudes from  $10^{\circ}$ - $140^{\circ}$ E, and latitudes from  $10^{\circ}$ -  $40^{\circ}$ N (Park et  
10 al., 2007; Randel et al., 2010; Ungermann et al., 2016). The air exported from the AMA influences  
11 the composition of the entire lowermost stratosphere of the Northern Hemisphere (Ploeger et al.,  
12 2017; Santee et al., 2017; Yu et al., 2017). Deep convective clouds represent conduits for air  
13 pollution to reach the Upper Troposphere and Lower Stratosphere (UTLS) region. Aerosols in the  
14 UTLS have longer residence times than those in the lower troposphere, influencing the chemistry  
15 of the atmosphere and the Earth's climate (Rasch et al., 2008). In addition, they also affect the  
16 concentration of chemical species through changes in photolysis rates and heterogeneous reactions  
17 (Pitari et al., 2014). It has been further reported that aerosols in the UTLS can impact climate by  
18 altering properties of cirrus clouds via homogeneous or heterogeneous ice nucleation (Li et al.,  
19 2005; Liu et al., 2009; Yin et al., 2012; Fadnavis et al., 2013).

20 A layer of aerosol enhancements observed by the Cloud-Aerosol Lidar and Pathfinder Satellite  
21 Observations (CALIPSO) and the Stratospheric Aerosol and Gas Experiment (SAGE) II (Vernier  
22 et al., 2011; Thomason and Vernier, 2013), also known as the Asian Tropopause Aerosol Layer  
23 (ATAL), coincide with the presence of enhanced trace gas pollutants (carbon monoxide (CO),  
24 hydrogen cyanide (HCN), etc.) in the UTLS region. Satellite and balloon-borne measurements  
25 (Vernier et al., 2011, 2015) confirmed the presence of ATAL at altitudes of 14-18 km, connected  
26 to the AMA. The positive trend in UTLS aerosols inferred from satellites observations since the  
27 late 90's may reflect the increasing influence of anthropogenic emissions on stratospheric aerosol  
28 levels. Indeed, global chemical transport model simulations suggest that sulfate, nitrate and organic  
29 aerosols produced from gas-phase precursors populates the UTLS region over Asia in various  
30 relative fractions during the summer monsoon (Brabec et al., 2012; Gu et al., 2016; Fairlie et al.,  
31 2020).



1           1.2 What is the significance of ATAL's composition?

2   The ATAL constitutes one of the most important sources of UTLS aerosols in the absence of  
3   volcanic eruptions (Vernier et al., 2011). It has the potential to affect the Earth's radiative balance  
4   (Vernier et al., 2015), stratospheric ozone chemistry, and the properties of cirrus clouds. For  
5   example, an increase of solid particle concentration relative to the liquid background aerosol levels  
6   could trigger heterogeneous freezing and the formation of cirrus clouds at a lower relative humidity  
7   with respect to ice (Cziczo et al., 2015; Wang et al., 2020). Model simulations suggest that the  
8   ATAL represents 20% of the total column surface area density in the stratosphere of the Northern  
9   Hemisphere (Yu et al., 2018) with potential halogen heterogeneous chemistry on aerosols that can  
10   affect ozone trends (Solomon et al., 2016). The types of aerosols populating the ATAL could affect  
11   those chemical processes. Finally, the presence of absorbing aerosols (e.g., soot) in the UTLS  
12   could shift the level of zero net radiative heating upward and enhance troposphere-to-stratosphere  
13   transport (Yu et al., 2015).

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16           1.3 What is known about ATAL's composition?

17   The composition of the ATAL is a very active research topic. Energy-dispersive X-ray analysis  
18   (EDX) of aerosols sampled near 10-12 km onboard commercial aircraft as part of the Civil Aircraft  
19   for the Regular Investigation of the atmosphere Based on an Instrument Container (CARIBIC)  
20   program, at the bottom part of the ATAL, suggests a ratio between carbon and sulfur in the range  
21   2-10 (Vernier et al., 2015). Aircraft Limb InfraRed measurements carried out during the  
22   StratoClim campaign in Nepal and India show the presence of ammonium nitrate in aerosol  
23   particles, validating satellite observations from the Cryogenic Infrared Spectrometers and  
24   Telescopes for the Atmosphere (CRISTA), and Michelson Interferometer for Passive Atmospheric  
25   Sounding MIPAS (Höpfner et al., 2019).

26   The aerosol particles in the ATAL are looked upon as an insignia of the presence of pollution in  
27   the monsoon circulation from large SO<sub>2</sub> and NO<sub>x</sub> emissions in South and SE Asia. Human-induced  
28   biomass burning (Van der A et al., 2008), fossil fuel combustion (Ghude et al., 2009; Bouman et  
29   al., 2002), wildfires (Goode et al., 2000; Andrae and Merlet, 2001), and lightning (Martin et al.,  
30   2007) are the significant anthropogenic, and natural sources of NO<sub>x</sub>. Soil biogenic emission of



1 NO<sub>x</sub> represents a large fraction of total NO<sub>x</sub> (Jalié et al., 2004). Reactive nitrogen is emitted from  
2 the tropical soils by microbial processes as NO (Yienger and Levy, 1995; Conrad et al., 1996).  
3 Investigations of the composition of the aerosol particles in the ATAL are exiguous, although  
4 preliminary data from balloon borne measurements indicate the presence of nitrate aerosol  
5 particles (Vernier et al., 2015, 2018). Recent in situ aerosol mass spectrometric measurements also  
6 reveal the presence of nitrate, ammonium and sulfate within the ATAL (Höpfner et al., 2016).

7 Here, we investigate the inorganic composition of the ATAL over India during the summer  
8 monsoon and winter using a balloon-borne aerosol impactor system with offline Ion  
9 Chromatography (IC) analysis. Section 2 describes the concept of the balloon experiment and the  
10 impactor system. The IC analysis of the samples collected during two balloon flights in 2017 and  
11 on the ground, as well as that of winter 2018 is described in Section 3. Section 4 compares those  
12 results obtained from balloon-borne measurements and satellite observations. Finally, the origin  
13 of the air masses sampled during those flights is assessed in Section 5 through back-trajectory  
14 analysis combined with convective proxies. The GEOS-Chem model simulations are presented in  
15 Section 6 to put the measurements in a context of regional aerosol transport and distribution,  
16 followed by summary and conclusions in Section 7.

## 17 **2. Balloon flights, instrumentation, and chemical analysis approach**

### 18 *2.1 Rationale for the experiment*

20 Contingent on measurements during the 2015 Balloon borne measurement campaigns of the Asian  
21 Tropopause Aerosol Layer (BATL) campaign, a concentration of 20 particle/cm<sup>3</sup> was found near  
22 the tropopause for aerosol radius greater than 75 nm (Vernier et al., 2018). It translates into a mass  
23 concentration of 40 ng/m<sup>3</sup> assuming that the aerosols were liquid sulfate droplets. During that time,  
24 the lower detection limit for the IC instrument at NASA Langley Research Center was around 20  
25 ng/m<sup>3</sup>. In order to reach the detection limit of sulfate aerosols, one would need to sample at least  
26 0.5 m<sup>3</sup> assuming the sulfate concentration above. Based upon those results and weight limitations,  
27 we decided to use an impactor with a flow rate of 7 lpm which would need to float in the UTLS  
28 region for several hours to sample sufficient volume of air (2 hours of sampling = 0.84 m<sup>3</sup>).

### 29 *2.2 Balloon experiment*



1 We used zero-pressure plastic balloons to achieve a float near the tropopause and sample enough  
2 aerosols to reach the detection limit of the IC. The Tata Institute of Fundamental Research Balloon  
3 Facility (TIFR-BF) in Hyderabad, India provided the infrastructure to conduct the experiment. 300  
4 to 500 m<sup>3</sup> polyethylene balloons manufactured by TIFR were used for the Zero-Pressure flights  
5 (ZF) to carry a communication/control package developed by TIFR, a science module including a  
6 meteorological radiosonde, a Compact Optical Backscatter and Aerosol Detector (COBALD)  
7 (Vernier et al., 2015; Yu et al., 2017), an aerosol impactor, and a ballast module at the end of the  
8 flight train. A schematic diagram shown in Fig.1 (top panel) describes a typical balloon flight.  
9 During the ascent, atmospheric pressure decreases allowing gas inside the balloon to occupy a  
10 large space (stage 2). The equilibrium point is reached when the hydrogen escapes from the side  
11 escape tubes attached at the bottom of the balloon, until the inside pressure equals the outside  
12 pressure (stage 3) leading to the pressure differential to 0 (zero-pressure balloon). The float altitude  
13 depends upon the volume of the balloon, the density of gas, as well as the total weight of the system  
14 following simple Archimedes principle. Extreme cold temperatures near the tropopause affect the  
15 float due to radiative cooling, leading to a reduction of the buoyancy force, which entrain the  
16 descent of the system (stage 4). To counterbalance this effect, ballast shots are released from a  
17 container to reduce the total weight (stage 5) leading to the ascent of the balloon.

### 18 2.3 Balloon-borne Aerosol Impactor

19 We developed the Balloon-borne Aerosol Impactor (BAI) for the ZF flights. This aerosol sampler  
20 is comprised of a 4-stage impactor, a vacuum pump, a volumetric flow controller, and a Raspberry-  
21 PI based controller connected to a meteorological sonde. The mechanical part of the impactor was  
22 designed by *California Measurements, Inc.* and is based upon the principle of inertia, where the  
23 flow and the instrument dimension determine the size cutoff at different stages. The size cutoff in  
24 radius for the 4 stages (S1, S2, S3 and S4) of the impactor is 2, 0.5, 0.15 and 0.05 μm at 7 lpm.  
25 The pump is controlled electronically based on pressure measurements from the meteorological  
26 sonde. Our objective is to sample aerosols within the ATAL region, to achieve this the pump was  
27 switched on below 150 hPa (~14 km) and switched off above 70 hPa (~18 km).

28 In 2017, we conducted a series of balloon flights using the BAI together with a COBALD sonde  
29 for aerosol backscatter measurements of cloud and aerosol layers encountered by the BAI. The  
30 time-height evolution of the 3 ZFs is shown in Fig.1 (bottom). The maximum flight duration was



1 obtained through ZF3 with a float time of nearly 2h 50min above 150 hPa and below 70 hPa. The  
2 oscillation of the balloon trajectories is due to the cooling of the gas inside the balloon and  
3 subsequent release of ballast to regain higher altitudes. The BAI was preserved in a foam box  
4 containing dry ice, during transportation to TIFR where the filters were immediately unloaded and  
5 stored in 47 mm Petri dishes which were frozen at -24 °C until further analysis at Physical Research  
6 Laboratory, Ahmedabad, India. Fig. 2 represents the time evolution of altitude, temperature, and  
7 relative humidity inside the box containing the impactor where the different phases of the  
8 experiment are mentioned.

#### 9 2.4 Analysis of major ions in aerosol samples

10 Aerosol samples were extracted in deionized water (Milli-Q, specific resistance  $\geq 18.2$  M $\Omega$ .  
11 cm) in sterile polypropylene vials for 30 minutes (3 intervals of 10 minutes each) using  
12 ultrasonication. The extract was further analyzed for water-soluble inorganic species (WSIS, such  
13 as Na<sup>+</sup>, K<sup>+</sup>, Mg<sup>2+</sup>, Ca<sup>2+</sup>, NH<sub>4</sub><sup>+</sup>, Cl<sup>-</sup>, NO<sub>2</sub><sup>-</sup>, NO<sub>3</sub><sup>-</sup>, and SO<sub>4</sub><sup>2-</sup>) using an ion chromatograph (IC model-  
14 Dionex ICS-5000 DC-5). For calibration, 1000 mg/L stock solution of each cation (using Merck  
15 high purity analytical grade NaNO<sub>2</sub>, (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, KNO<sub>3</sub>, CaCl<sub>2</sub>·2H<sub>2</sub>O, and Mg metal) were  
16 prepared. In addition, mixed standards were prepared by diluting stock solutions in polypropylene  
17 vials, thus satisfying the primary requirement of instrument calibration for cations. Similarly, anion  
18 multi-element standard-II (1000 mg/L in H<sub>2</sub>O, HC 409399, Merck) was diluted subsequently as  
19 instrument calibration for anions. Post extraction, the extract of each sample was then separated  
20 and eluted in the cation column (DIONEX IonPac<sup>TM</sup> SC16, 5 × 250mm), and anion column  
21 (DIONEX Ion Pac<sup>TM</sup> AS23, 4 × 250mm) via the interaction with the mobile phases, i.e., 30 mM  
22 methyl sulphonic acid (MSA) for cation and a mixture of 4.5 mM carbonate + 0.8 mM bi-carbonate  
23 solutions for anions. The quantification of each ion was then performed using the conductivity  
24 detector. Several blanks were also analyzed in the same way as the sample, and blank corrected  
25 from ionic concentrations are reported. As the concentrations of different species were too low in  
26 UTLS aerosol samples, only those values which were at least two times higher than their respective  
27 blanks are reported. More than 50% of samples were repeated for the reproducibility and found to  
28 vary between 2 to 20% for all the analyzed ions. To validate the analysis, Dionex six cation-I  
29 standard (product number 040187) and Dionex seven anion standard-II (Part #57590) were diluted  
30 and checked in the respective cation and anion calibration curves which were found within  $\pm 10\%$   
31 relative standard deviation (RSD).



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### 3. Results of IC analysis

3 Figure 3 shows the concentration of ions from ground (GND), and two ZF2, and ZF3 flight samples  
4 collected during the summer 2017 campaign, in comparison with the only flight results of the  
5 winter 2018 campaign (ZF Winter). In GND samples,  $\text{Na}^+$  and  $\text{Ca}^{2+}$  cations are seen on S1 and S2  
6 with corresponding anions ( $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$ , and  $\text{NO}_2^-$ ) co-existing at the same stage. High  $\text{NH}_4^+$  is  
7 observed only on S3 with a concentration of  $212 \text{ ng/m}^3 \text{ STP}$ .  $\text{K}^+$  was also seen on S3 with a  
8 concentration of  $26 \text{ ng/m}^3 \text{ STP}$  (fine mode) that could have originated from biomass burning. City  
9 pollution from Hyderabad is likely the source of those aerosols observed on the GND filters. Flight  
10 ZF2 and ZF3 show significant amounts of  $\text{NO}_3^-$  and  $\text{NO}_2^-$  ( $87\text{-}343 \text{ ng/m}^3 \text{ STP}$ ) with traceable  
11 amounts of proxies for mineral dust ( $\text{Ca}^{2+}$ ) and biomass burning ( $\text{K}^+$ ). Other species were below 5  
12 (for cations) to 10 (for anions)  $\text{ng/m}^3 \text{ STP}$ , the detection limit of the IC instrument for our analytical  
13 setup. Charge balance was not achieved due to higher negative charge mainly from  $\text{NO}_3^-$  and  $\text{NO}_2^-$   
14 than the positive charge mainly from  $\text{NH}_4^+$ ,  $\text{Ca}^{2+}$ , and  $\text{K}^+$  (Fig. 3), implying the existence of  $\text{NO}_3^-$   
15 and  $\text{NO}_2^-$  in forms rather than salt. For instance, nitric acid trihydrate (NAT,  $\text{HNO}_3 \cdot 3\text{H}_2\text{O}$ ) could  
16 be another aerosol cluster in which  $\text{NO}_3^-$  may be present in the tropical UTLS (Voigt et al., 2000).  
17 However, recent measurements from StratoClim campaign show that solid ammonium nitrate  
18 dominates the aerosol population of the ATAL with concentration up to  $1.5 \mu\text{g/m}^3$  during summer  
19 (Hopfner et al., 2019). In contrast, we did not find a significant amount of ammonium in our ZF  
20 flight samples during the summer, but did find the same in the flight samples of winter (Fig. 3).  
21 Overall, the concentration of nitrate ( $80\text{-}100 \text{ ng/m}^3 \text{ STP}$ ) found on both flights seems to be lower  
22 than the levels observed during StratoClim (Hopfner et al., 2019). In the only successful flight  
23 during the winter 2018,  $\text{Na}^+$  and  $\text{K}^+$  were almost inexistent. In comparison, the proxy of mineral  
24 dust ( $\text{Ca}^{2+}$ ) was present on all four stages with traceable amounts, and could be associated with  
25  $\text{SO}_4^{2-}$  which was also found on all 4 stages (Fig. 3 Bottom).

26 Balloon-borne and aircraft sampling techniques have been used since the early 70's to study the  
27 composition of aerosols in the UTLS region (Lazarus et al., 1970). While sulfate tends to be stable  
28 enough to be collected and further analyzed without major chemical transformation, other nitrate-  
29 containing particles can be more unstable.  $\text{NO}_3^-$  salts apart from  $\text{NH}_4\text{NO}_3$  are not significantly  
30 volatile after sampling (Newman et al., 1993). However, the dissociation of  $\text{NH}_4\text{NO}_3$  into gas-  
31 phase  $\text{HNO}_3$  and  $\text{NH}_3$  increases sharply with increasing temperature and relative humidity





1 (Seinfeld et al., 1982; Lightstone et al., 2000), leading to a significant loss of particulate nitrate  
2 (PN). Between payload descent, sample recovery, and transportation, the observed temperature  
3 stayed below 35°C at 65% RH (Fig. 2), indicating that there was likelihood of the dissociation of  
4  $\text{NH}_4\text{NO}_3$  between recovery and transportation. The slight retention of  $\text{HNO}_3$  (gas) on PTFE filter  
5 could represent a significant source of particulate nitrate on filters at low concentration and was  
6 used in the past to estimate stratospheric  $\text{HNO}_3$  (Lazarus et al., 1970).

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#### 8 **4. COBALD and CALIOP backscatter measurements during ZF2**

9 ZF2 was launched to be collocated in space and time (within 20 km and 1h) with satellite  
10 observations from the CALIOP lidar onboard the CALIPSO satellite. Fig. 4 (Top) shows  
11 Scattering Ratio (SR) and Color Index (CI) profiles from COBALD (470 nm and 940 nm) together  
12 with CALIOP SR and Volume Depolarization profiles at 532 nm. The CI and both balloon and  
13 satellite observations show a layer between 13.5 and 16 km with high depolarization (CALIOP)  
14 and high color ratio (COBALD), likely made of aspherical particles. ZF2 flight floated near 16-17  
15 km for more than 2h (Fig.1, bottom). The time series (Fig.4, bottom) indicate that the  
16 measurements took place above 150 hPa (43 min after takeoff), pressure threshold below which  
17 the pump connected to the impactor was switched on. In the thin ice cloud, the pump was on for  
18 nearly 16 min. while most of the sampling was done for more than 1 h in a cloud-free region  
19 enhanced with aerosols above.

#### 20 **5. Convective influence**

21 16-year climatological assessment of cirrus clouds, their microphysical, and optical properties  
22 were observed using a ground-based lidar over Gadanki, India (Pandit et al., 2015). When  
23 compared with those obtained by CALIOP, the observations showed the presence of cirrus clouds  
24 of geometrical thickness less than 2 km. Moreover, the increasing fraction of sub-visible cirrus  
25 clouds between 1998-2003 probably modified on the temperature and the water vapor budget in  
26 the Tropical Tropopause Layer (Pandit et al., 2015). To study the impact of convection on our  
27 measurements, we calculate back-trajectories from ZF2 and ZF3 using the Langley Trajectory  
28 Model (LaTM) driven by GEOS-5 winds (Fairlie et al., 2014) and locate the intersection with  
29 anvils and deep convective clouds observed through Cloud Top Brightness Temperature from the  
30 HIMAWARI-8 satellite (Vernier et al., 2018). Figure 5 shows the position of those 5-day back-



1 trajectories (black lines) and deep convection influence (red dots). Air sampled during ZF2 on 15<sup>th</sup>  
2 August 2017 travelled along two branches influenced by convection over southern/eastern China  
3 and western China, respectively. Air masses sampled by ZF3 originated from convection over  
4 Laos, Myanmar, the Bay of Bengal, and possibly local convection over the Indian Eastern Shore  
5 close to the measurement location.

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## 7 **6. Comparison with GEOS-Chem simulations**

8 We conduct GEOS-Chem model simulations to put our observations in a context of large-scale  
9 transport and distribution of atmospheric composition. GEOS-Chem is a state-of-the-art global 3-  
10 D chemical transport model that includes fully coupled ozone-NO<sub>x</sub>-VOC-aerosol chemistry for  
11 both troposphere and stratosphere (Bey et al., 2001; Park et al., 2004; Eastham et al., 2014). We  
12 use here the model version 11-01 ([http://wiki.seas.harvard.edu/geos-chem/index.php/GEOS-](http://wiki.seas.harvard.edu/geos-chem/index.php/GEOS-Chem_v11-01)  
13 [Chem\\_v11-01](http://wiki.seas.harvard.edu/geos-chem/index.php/GEOS-Chem_v11-01)). A previous version of the model was used to study the origins of aerosols in the  
14 ATAL by Gu et al. (2016) and Fairlie et al. (2020). The model simulates black carbon (Park et  
15 al., 2003), primary and secondary organic aerosols (Pyle et al., 2010), sulfate-nitrate-ammonium  
16 aerosols (Park et al., 2004), mineral dust (Fairlie et al., 2007; Ridley et al., 2014), and sea salt  
17 (Jaegle et al., 2011), treated as an external mixture. Anthropogenic emissions use the EDGAR  
18 database (Olivier & Berdowski, 2001), with regional options, including the MIX inventory over  
19 East Asia (Li et al., 2014) and the EPA/NEI 2011 inventory over North America (Travis et al.,  
20 2016). Biofuel emissions are from Yevich and Logan (2003). Carbonaceous aerosol emissions  
21 are provided by Bond et al. (2007). Biogenic emissions are calculated by the MEGAN model  
22 (Guenther et al., 2012). Biomass burning emissions use the Quick-Fire Emissions Dataset  
23 (QFED; Darmenov & da Silva, 2015). Lightning NO<sub>x</sub> emissions (LNO<sub>x</sub>) are as described by  
24 Murray et al. (2012) and match the Lightning Imaging Sensor and the Optical Transient Detector  
25 (LIS/OTD) climatological observations of lightning flashes. Volcanic SO<sub>2</sub> emissions are  
26 provided by the AeroCom project (data available from [www.geos-chem.org](http://www.geos-chem.org)). The model  
27 simulations are driven by the Modern-Era Retrospective analysis for Research and Applications  
28 (MERRA-2) reanalysis from the NASA Global Modeling and Assimilation Office (Gelaro et al.,  
29 2017). For computational efficiency, MERRA-2 fields have been mapped from the native grid to  
30 2.5° by 2° horizontal resolution for input to GEOS-Chem. Further, we used the simulations with



1 and without lightning  $\text{NO}_x$  emissions to understand the contribution of lightning to the formation  
2 of nitrate aerosol.

3 In situ chemical analysis are compared with GEOS-Chem simulations. Fig. 6 shows the maps of  
4 CO, nitrate, sulfate, and black carbon (BC) aerosol concentrations averaged over 100-150 hPa at  
5 22 UTC for Aug. 15<sup>th</sup> and 21<sup>st</sup>, respectively, during ZF2 and ZF3 flights (white dots on the map).  
6 On Aug. 15<sup>th</sup> CO, BC and nitrate aerosol concentrations are enhanced over West China, Nepal and  
7 northeastern India with the center of the anticyclone positioned over West China. Measurements  
8 made during ZF2 were obtained from air masses with enhanced CO, BC and nitrate as suggested  
9 by the model. On the contrary on Aug. 21<sup>st</sup> during ZF3, the position of the anticyclone was shifted  
10 to the east and the flight apparently sampled air at the edge of the anticyclone where CO, BC and  
11 nitrate are significantly lower. The simulated nitrate concentrations near the location of ZF2 and  
12 ZF3 is spatially inhomogeneous with variations between 30 and 2700 ng/m<sup>3</sup> across South India.  
13 Figure 7 shows the time series of model 3-hourly CO, sulfate and nitrate concentrations averaged  
14 over 100-150hPa within the model grid-point where Hyderabad is located during August 2017.  
15 The measured nitrate concentration during ZF2 and ZF3 near ~100ng/m<sup>3</sup> are within the ranges of  
16 values simulated within 24h of the observations. The results of the GEOS-chem model simulation  
17 signifies that lightning  $\text{NO}_x$  could significantly (up to ~75% on August 10th) contribute to the  
18 formation of nitrates during certain time periods. The lifetime of  $\text{NO}_x$  is approximately 3h in the  
19 region of the outflow of thunderstorms due to the production of methyl proxy nitrate and alkyl,  
20 and multifunctional nitrates. The  $\text{NO}_x$  lifetime is believed to increase downwind from the outflow.  
21 Also shown in Fig.7 are nitrate concentrations attributed to lightning as determined by the  
22 difference between simulations with and without lightning  $\text{NO}_x$  emissions. ZF2 and ZF3 occurred  
23 during a period where the levels of nitrate were relatively small (<100 ng/m<sup>3</sup> STP), with minimal  
24 influence of lightning  $\text{NO}_x$  emissions. Nevertheless, CO levels are slightly higher during ZF2 (80  
25 ppbv) than ZF3 (60 ppbv), indicating that measurements made during ZF2 may have been more  
26 influenced by pollution. The latter is also reflected by the higher BC levels during ZF2 in the  
27 model.

28 We extracted CO, nitrate and sulfate concentration from the GEOS-Chem simulation along the  
29 calculated trajectories initialized from ZF2 and ZF3 measurement locations in Fig. 8. The lines  
30 are colored according to the balloon GPS altitudes which are used to initialize the trajectory  
31 model. Fig.5 shows that GEOS-chem could simulate convective activities reaching levels



1 between 14-15 km as confirmed by cloud top heights (black circle) derived from HIMAWARI-8  
2 crossed by trajectories originating from the troposphere for both ZF2 and ZF3. ZF2 was  
3 influenced by convective activities over Western China while ZF3 sampled air masses originated  
4 from convection in SE Asia (Myanmar, Laos). CO levels with initial altitudes near 14-15 km  
5 (green color) for ZF2 are shown to decrease from 120 ng/m<sup>3</sup> to 80 ng/m<sup>3</sup> along the back-  
6 trajectories confirming the influence of Chinese pollution and its progressive dilution. At the  
7 same initial altitudes, The CO levels along ZF3 back-trajectories are significantly lower near 50-  
8 80 ng/m<sup>3</sup> possibly indicating minimal impacts of polluted sources. Sulfate concentrations are  
9 much higher (100-200 ng/m<sup>3</sup>) for air parcels initialized near 16-17 km for ZF2 and ZF3 likely  
10 indicating stratospheric sources while air parcels near 14-15 km show levels below 100 ng/m<sup>3</sup>.  
11 We note that sulfate along the trajectories influenced by Chinese pollution during ZF2 increase  
12 significantly 50h before our measurements which could indicate the formation of sulfate aerosol  
13 from SO<sub>2</sub>. The levels of NO<sub>3</sub> show significant variability along the trajectories for both cases  
14 but more pronounced in ZF3 with levels above 400 ng/m<sup>3</sup> emphasizing again the likely  
15 importance of LNO<sub>x</sub> in the production of nitrate aerosols.

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## 18 7. Summary and Conclusions

19 The chemical composition of the ATAL has been investigated using offline IC analysis of aerosol  
20 impacted samples collected onboard the zero-pressure balloon flights as part of the BATAL  
21 campaigns. The measurements of the 2017 summer campaign indicate the dominating presence of  
22 nitrate and nitrite aerosols with concentrations between 88 and 374 ng/m<sup>3</sup> STP. Our first flight  
23 (ZF2) on 15<sup>th</sup> August 2017, occurred within the AMA and thus sampled air masses therein. In situ  
24 measurements revealed the presence of NO<sub>3</sub><sup>-</sup>, and NO<sub>2</sub><sup>-</sup> aerosols (60-200 ng/m<sup>3</sup> STP) of size  
25 ranging between 0.05-2μm. The second flight (ZF3) on 21<sup>st</sup> August 2017, however, occurred at  
26 the edge of the anticyclone and subsequent in situ measurements revealed the presence of larger  
27 particle size NO<sub>3</sub><sup>-</sup> and NO<sub>2</sub><sup>-</sup> aerosols at higher concentrations (87.3-343 ng/m<sup>3</sup> STP). Throughout  
28 the flights during the 2017 summer campaign, sulfate aerosol remained below the detection limit  
29 of the system (10 ng/m<sup>3</sup> STP) much lower than the results from the GEOS-Chem model simulation  
30 (80-120 ng/m<sup>3</sup> STP) which may suggest that sulfate removal is insufficient in the model. Unlike



1 the summer,  $\text{Ca}^{2+}$  and  $\text{SO}_4^{2-}$  were found on all four stages in sizes ranging between 0.2-0.05 $\mu\text{m}$   
2 together with traces of  $\text{NH}_4^+$  which couldn't be quantified in the winter campaign. The winter  
3 flight sampled residual from the 2017 Canadian wildfires which affected stratospheric aerosol  
4 loadings for several months.

5 We study the influence of convection on those measurements using back trajectory calculations  
6 collocated with geostationary satellite observations. We show that ZF2 and ZF3 were influenced  
7 by convection over Western China, the Bay of Bengal as well as Myanmar, Thailand, and Laos.  
8 While the model seems to represent convection in the upper troposphere (14-15 km) with the rapid  
9 ascension of air parcels, the model ability to simulate convective influence at higher altitudes seem  
10 to be limited.

11 We used the GEOS-chem model simulations with and without lightning  $\text{NO}_x$  emissions to  
12 understand the contribution of lightning to nitrate aerosol. The flights, ZF2 (Aug. 15<sup>th</sup>) and ZF3  
13 (Aug. 21<sup>st</sup>) occurred during a period where the levels of nitrate were relatively small (<100  $\text{ng}/\text{m}^3$   
14 STP), with minimal influence of lightning  $\text{NO}_x$  in contrast with other periods largely affected by  
15 nitrate produced by LNO $_x$ . The chemical information extracted from GEOS-Chem along the  
16 trajectories indicates that ZF2 was influenced by Chinese pollutions associated with high levels of  
17 CO where smaller nitrate particles were found which could also indicate the influence of new  
18 particle formation.

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22 *Data availability.* We plan to keep those data on the Langley Archive data base together with the  
23 model results

24

25 *Author contributions.* HV led the preparation of the paper. AP and NR contributed to the chemical  
26 analysis of the balloon samples. MVR, HG, AP, JPV, DF, SK, GB organized the balloon flights.  
27 FW contributed to the analysis of the COBALD data. HL and BZ performed GEOS-Chem model  
28 simulations and assisted with model output analysis. KB analyzed HIMAWARI-8

29

30 *Competing interests.* The authors declare that they have no conflict of interest.



1 *Acknowledgement.* HL, TDF, KMB, BZ, and JPV acknowledge funding support from the NASA  
2 Atmospheric Composition Modeling and Analysis Program (ACMAP) and the Upper  
3 Atmospheric Research Program (UARP). NASA Center for Computational Sciences (NCCS)  
4 provided supercomputing resources. The GEOS-Chem model is managed by the Atmospheric  
5 Chemistry Modeling Group at Harvard University with support from NASA ACPMAP and MAP  
6 programs.

7 *Financial support.* This research has been supported by XXX and the NASA Atmospheric  
8 Composition Modeling and Analysis Program (ACMAP).  
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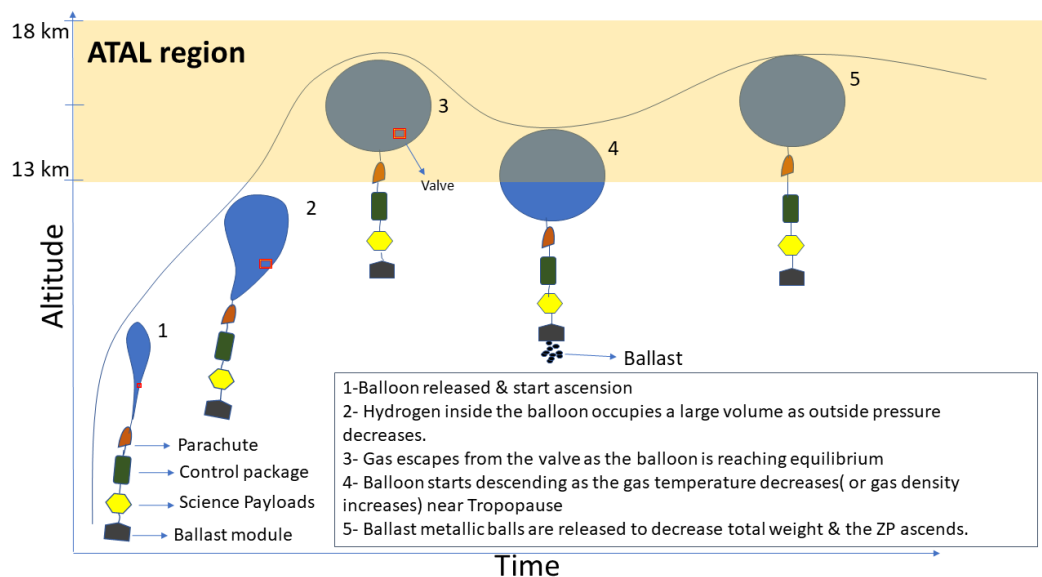
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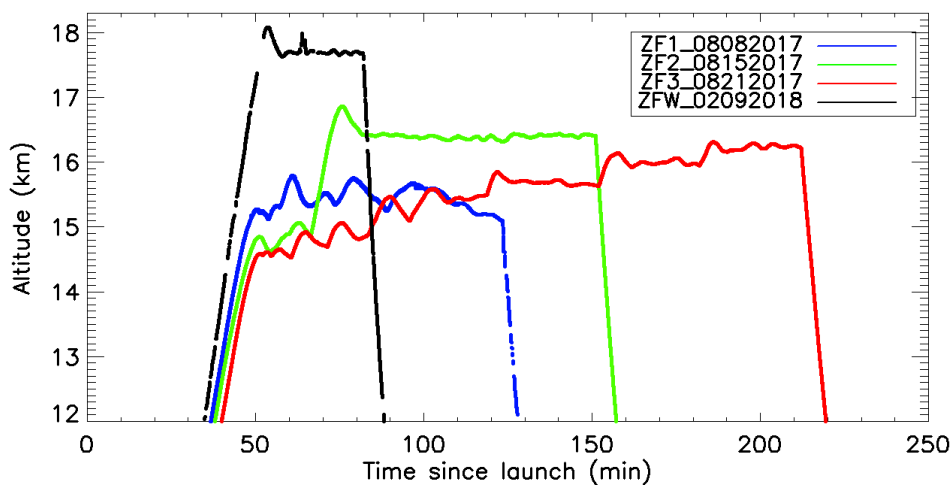
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1 **Figures.**

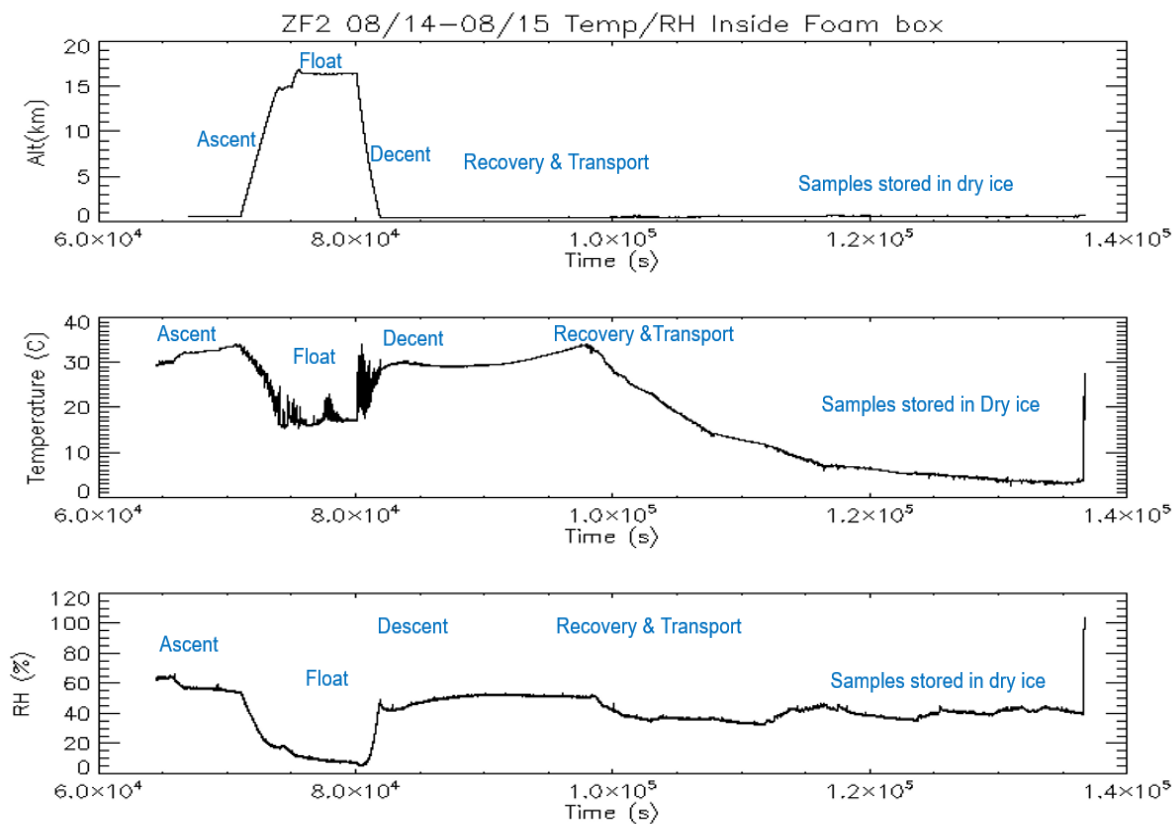


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4 Figure 1. (Top). Schematic diagram of the zero-pressure flight concept. (Bottom) Time-height  
 5 curves of the GPS altitudes of the 3 zero-pressure flights during summer 2017, in comparison  
 6 with that of winter 2018, launched from TIFR-BF, Hyderabad, India.



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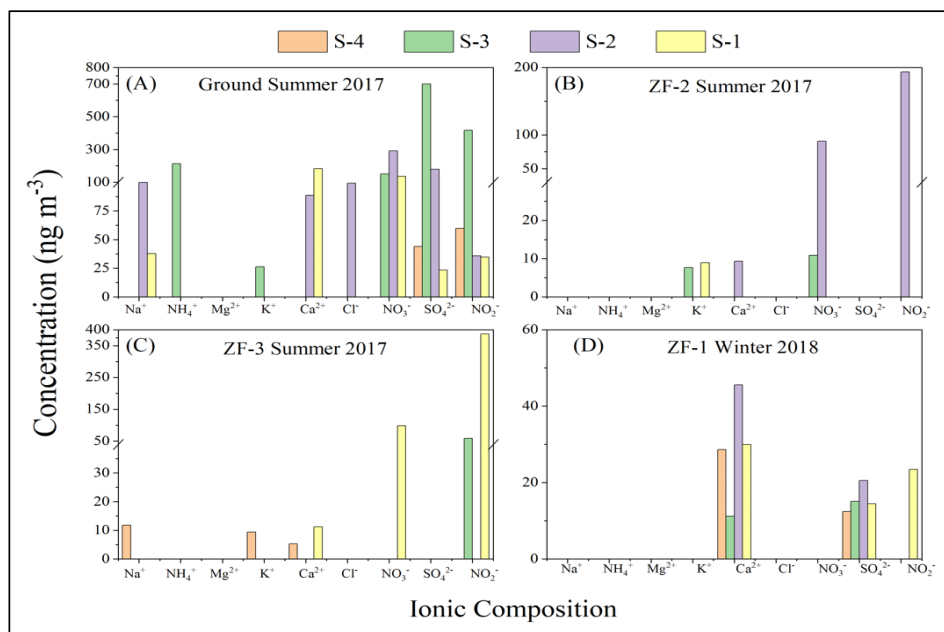
2 Figure 2. Time series of the Altitude, Temperature, and Relative humidity profiles of the  
3 samples inside the foam box during ZF2 flight.

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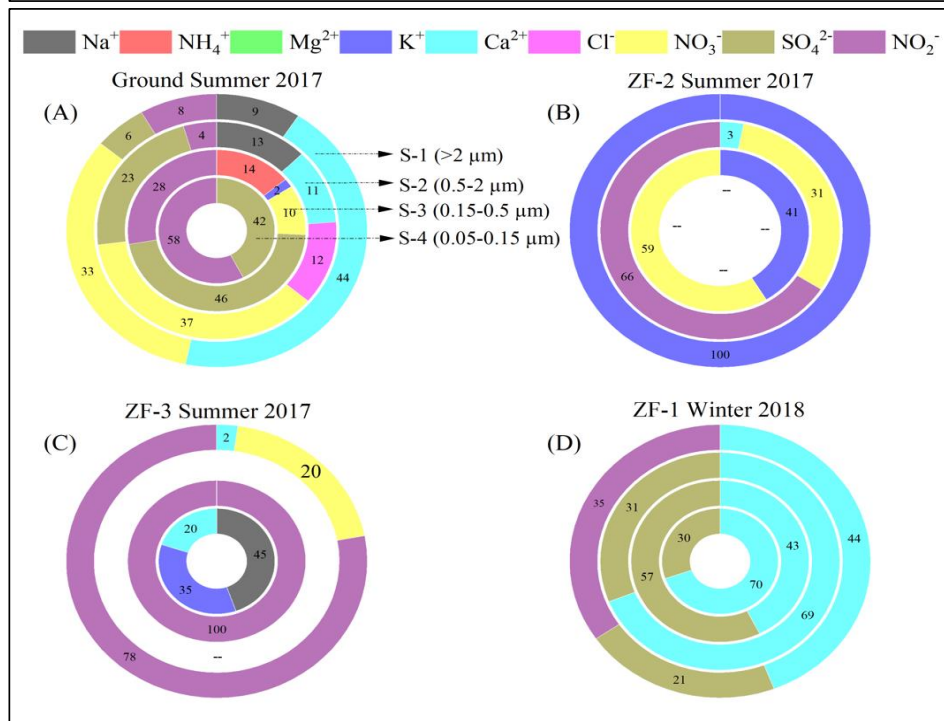
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3 Figure 3. Results from the analysis of inorganic aerosol. (Top) Aerosol ionic composition of the  
 4 filters collected on (A) the ground, during (B) ZF2 and (C) ZF3 of Summer 2017, and (D) ZF1 in



1 Winter 2018. (Bottom) Percentage distribution of individual ions. S1 to S4 indicates the four  
2 stages of the impactor. The size cut off is 2, 0.5, 0.15 and 0.05 micron for S1, S2, S3 and S4,  
3 respectively.

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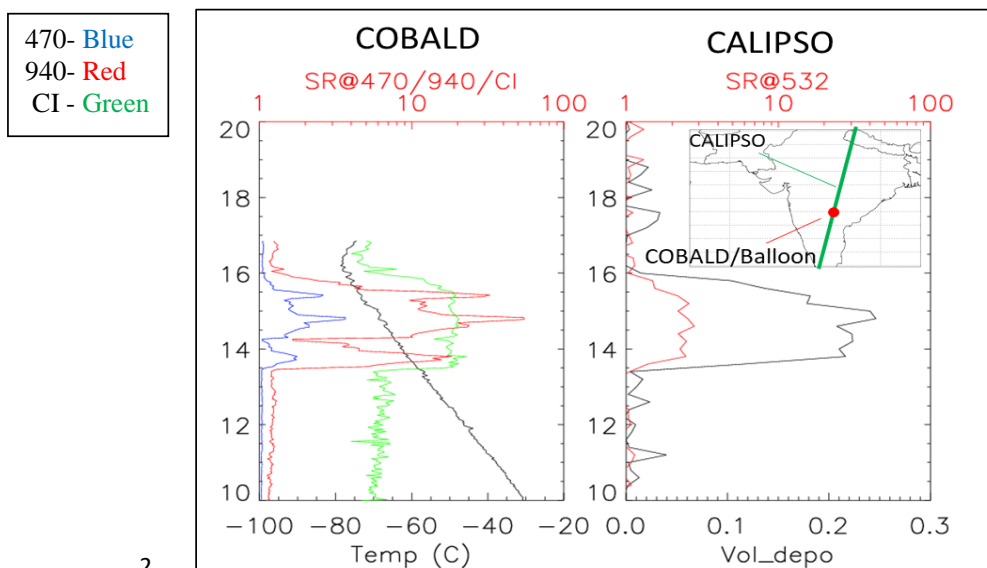
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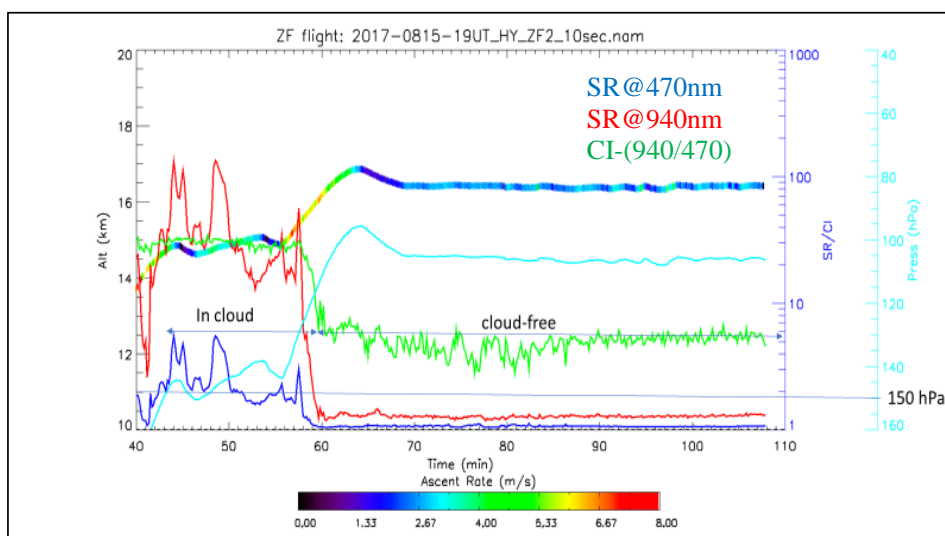




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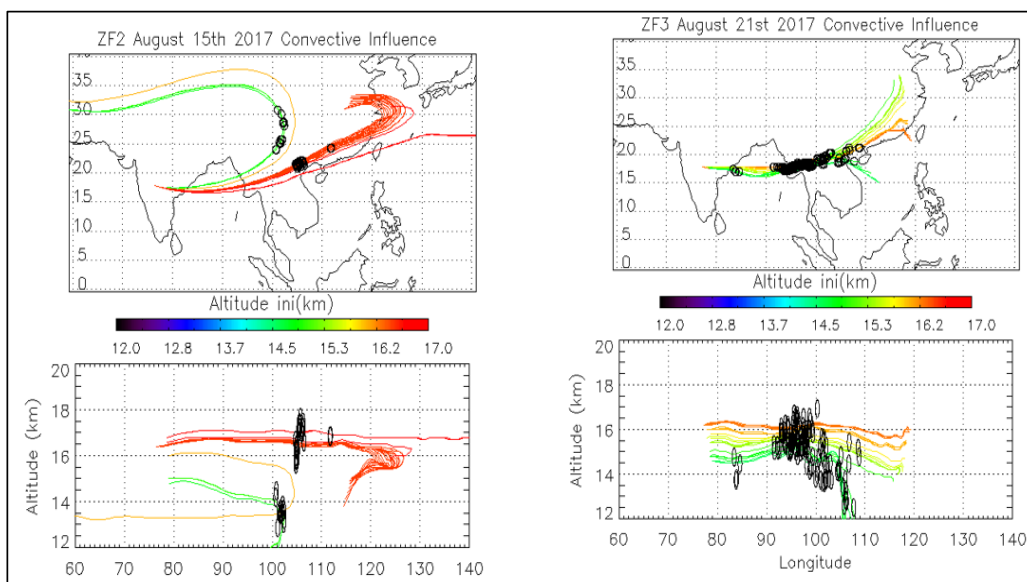
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5 Figure 4. (Top) COBALD balloon *In Situ* and CALIOP Satellite Scattering Ratios (SR), volume  
 6 depolarization and Color Index (CI) profiles collocated in time and space (within 20 km and 1 h)  
 7 on August, 15<sup>th</sup> at 19 UT. (Bottom) Time series along ZF2 of Scattering Ratios (SR) at 940 nm  
 8 and 470 nm from COBALD and GPS altitude (colored with ascent rate) and measured pressure  
 9 from the Imet radiosonde.

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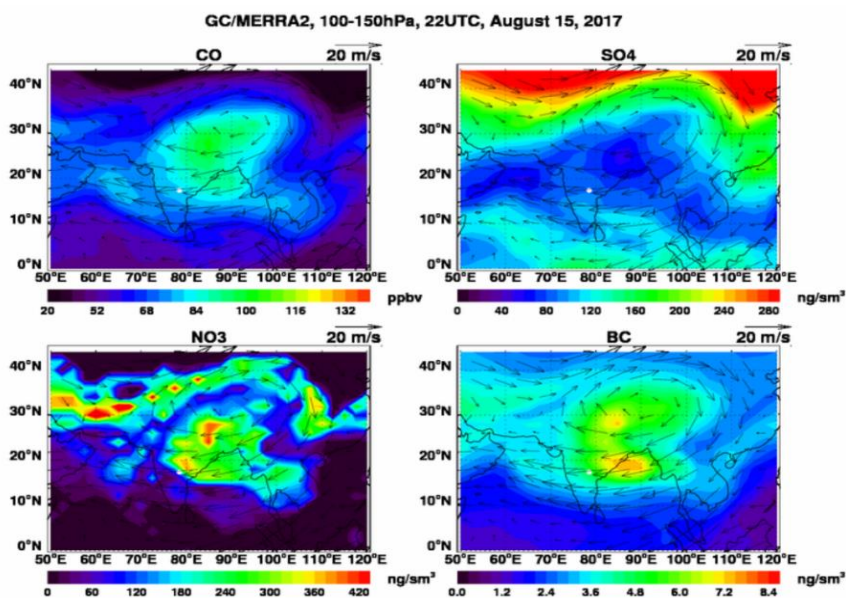


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3 Figure 5. Back-trajectories initialized from ZF2 (08/15) and ZF3 (08/21) measurements between  
4 150 hPa and 70 hPa. Black dots along the trajectories are the position of convective systems  
5 intersecting air masses sampled during the balloon flight.



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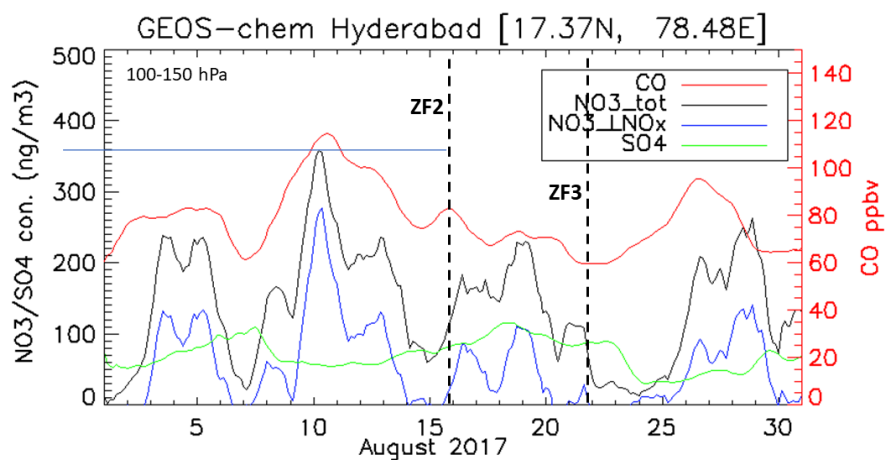
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Figure 6. GEOS-Chem model-simulated carbon monoxide (CO, ppbv), nitrate ( $\text{NO}_3^-$ ,  $\text{ng}/\text{m}^3$  STP), sulfate ( $\text{SO}_4^{2-}$ ,  $\text{ng}/\text{m}^3$  STP), and black carbon (BC,  $\text{ng}/\text{m}^3$  STP) concentrations averaged over 100-150hPa at 22UTC, August 15<sup>th</sup> (top panels), and August 21<sup>st</sup> (bottom panels), 2017.



1 Standard temperature and pressure (STP) are 298K and 1013.25hPa, respectively. Arrows denote  
2 wind direction while the white dot indicates sampling location, Hyderabad, India.

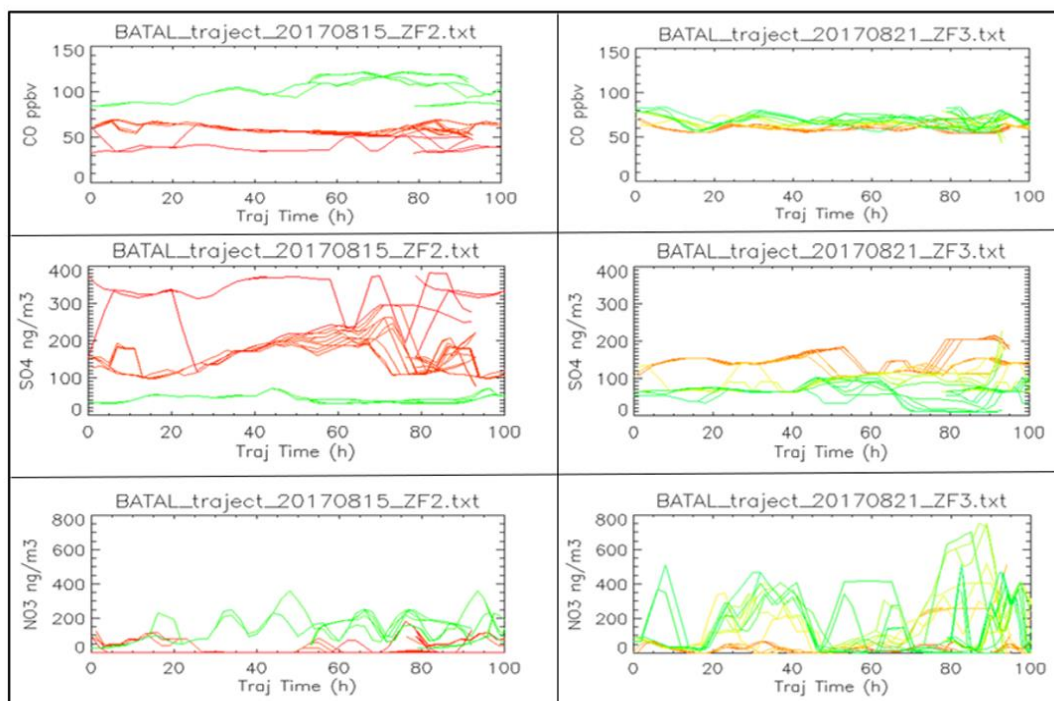
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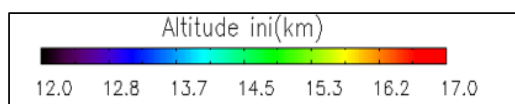
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2 Figure 7. Time series of simulated 3-hourly CO, SO<sub>4</sub><sup>2-</sup>, and NO<sub>3</sub><sup>-</sup> concentrations averaged over  
3 100-150hPa at Hyderabad during the ZF2 and ZF3 flights on 15<sup>th</sup> Aug. & 21<sup>st</sup> Aug., 2017. Also  
4 shown are concentrations of NO<sub>3</sub><sup>-</sup> due to lightning NO<sub>x</sub> emissions (NO<sub>3</sub>\_LNO<sub>x</sub>). See text for  
5 details.

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3 Figure 8. GEOS-Chem model-simulated CO, SO<sub>4</sub><sup>2-</sup>, and NO<sub>3</sub><sup>-</sup> concentrations extracted along the  
4 trajectory lines during flights ZF2 and ZF3 (Fig. 5).

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