



1 **The impact of chlorine chemistry combined with**
2 **heterogeneous N₂O₅ reactions on air quality in China**

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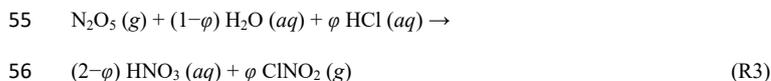
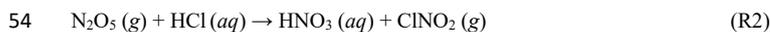
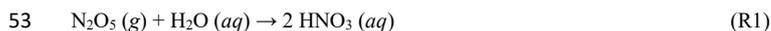
14 **Abstract:** The heterogeneous reaction of N₂O₅ on Cl-containing aerosols (N₂O₅ – ClNO₂ chemistry)
15 plays a key role in chlorine activation, NO_x recycling and consequently O₃ formation. In this study, we
16 use the GEOS-Chem model with additional anthropogenic and biomass burning chlorine emissions
17 combined with updated parameterizations for N₂O₅ – ClNO₂ chemistry (i.e. the uptake coefficient of
18 N₂O₅ ($\gamma_{\text{N}_2\text{O}_5}$) and the ClNO₂ yield (ϕ_{ClNO_2})) to investigate the impacts of chlorine chemistry on air quality
19 in China, the role of N₂O₅ – ClNO₂ chemistry, as well as their sensitivities to chlorine emissions and
20 parameterizations for $\gamma_{\text{N}_2\text{O}_5}$ and ϕ_{ClNO_2} . The model evaluation with multiple data sets observed across
21 China demonstrated significant improvement especially regarding the simulation of Cl⁻, N₂O₅ and ClNO₂
22 with the updates in chlorine emissions and N₂O₅ – ClNO₂ chemistry. Total tropospheric chlorine
23 chemistry could increase annual mean MDA8 O₃ by up to 4.5 ppbv but decrease PM_{2.5} by up to 7.9 μg
24 m⁻³ in China, 83% and 90% of which could be attributed to the effect of N₂O₅ – ClNO₂ chemistry. The
25 heterogeneous uptake of N₂O₅ on chloride-containing aerosol surfaces is an important loss pathway of
26 N₂O₅ as well as a important source of O₃, and hence is particularly useful in elucidating the commonly
27 seen ozone underestimations. The importance of chlorine chemistry largely depends on both chlorine
28 emissions and the parameterizations for N₂O₅ – ClNO₂ chemistry. With the additional chlorine emissions
29 annual mean maximum daily 8-hour average (MDA8) O₃ in China could be increased by up to 3.5 ppbv.
30 The corresponding effect on PM_{2.5} concentrations varies largely with regions, with an increase of up to
31 4.5 μg m⁻³ in the North China Plain but a decrease of up to 3.7 μg m⁻³ in the Sichuan Basin. On the other
32 hand, even with the same chlorine emissions, the effects on MDA8 O₃ and PM_{2.5} in China could differ
33 by 48% and 27%, respectively between different parameterizations.



34 1 Introduction

35 Chlorine (Cl) plays an important role in atmospheric chemistry in both the stratosphere and the
36 troposphere, primarily via the reactions of Cl atom with various atmospheric trace gases including
37 dimethyl sulfide, methane, and other volatile organic compounds (VOCs). The chemistry of Cl is quite
38 similar with that of hydroxyl radicals (OH) while Cl atom reacts up to 2 orders of magnitude faster with
39 some VOCs than OH (Atkinson et al., 2006). Studies have shown that Cl accounts for around 2.5% –
40 2.7% of the global CH₄ oxidation in the troposphere, and the contribution varies across regions, reaching
41 up to 10% – 15% in Cape Verde and ~20% in east China (Lawler et al., 2011; Hossaini et al., 2016). Cl
42 atom, therefore, is regarded as a potentially important tropospheric oxidant.

43 In general, Cl atom can be produced from the photo-dissociation and the oxidation of chlorinated
44 organic species (e.g. CH₃Cl, CH₂Cl₂ and CHCl₃) and inorganic chlorine species (i.e. HCl and Cl₂).
45 Recently, nitryl chloride (ClNO₂), formed through the heterogeneous reaction between dinitrogen
46 pentoxide (N₂O₅) and chloride-containing aerosols, is found to be another important source of
47 tropospheric Cl atoms in polluted regions (Liu et al., 2018; Haskins et al., 2019; Choi et al., 2020). The
48 heterogeneous formation of ClNO₂ and the consequent photolysis can be described by reactions R1 –
49 R4 shown below (Finlayson-Pitts et al., 1989; Osthoff et al., 2008). The net reactions of R1 (N₂O₅
50 hydrolysis on none-chloride-containing aerosols) and R2 (uptake of N₂O₅ on chloride-containing
51 aerosols) could be expressed as R3, in which the ClNO₂ yield (i.e. φ_{ClNO_2} , defined as the molar ratio of
52 produced ClNO₂ to total reacted N₂O₅) represents the fraction of N₂O₅ reacting via R2.



58 Estimates based on model simulations have suggested that ClNO₂ provides a source of Cl atoms totaling
59 0.66 Tg Cl a⁻¹, with the vast majority (95%) being in the Northern Hemisphere. The relative contribution



60 of ClNO₂ to global tropospheric Cl atoms is 14% on average and exhibits clear regional variations
61 (Sherwen et al., 2016). For example, the study by Riedel et al. (2012) reported that the relative
62 contribution is approximately 45% in Los Angeles based on a simple box model combined with local
63 observations.

64 The heterogeneous formation of ClNO₂ also serves as a reservoir for reactive nitrogen at night. The rapid
65 photolysis of ClNO₂ at daytime (R4) not only releases highly reactive Cl atom but also recycles NO₂ back
66 to the atmosphere, which as well significantly affect the daytime photochemistry (Thornton et al., 2010;
67 Riedel et al., 2014). It was suggested that the heterogeneous reaction between N₂O₅ and chloride-
68 containing aerosol could increase monthly mean values of the maximum daily 8h average (MDA8) O₃
69 concentrations by 1.0 – 8.0 ppbv in most Northern Hemisphere regions (Sarwar et al., 2014; Wang et al.,
70 2019). The reaction also impacts secondary aerosol formation, mainly through the recycling of NO_x
71 (Staudt et al., 2019; Mitroo et al., 2019). For example, Sarwar et al. (2014) estimated that ClNO₂
72 production decreases nitrate by 3.3% in winter and 0.3% in summer averaged over the entire Northern
73 Hemisphere. The influence of the heterogeneous formation of ClNO₂ in China is even larger due to the
74 polluted environment, leading to an increase in ozone concentrations by up to 7 ppbv, and a decrease in
75 total nitrate by up to 2.35 μg m⁻³ on monthly mean basis (Li et al., 2016; Sarwar et al., 2014)

76 There are two key parameters determining the heterogeneous uptake of N₂O₅, including the uptake
77 coefficient of N₂O₅ ($\gamma_{\text{N}_2\text{O}_5}$) and the ClNO₂ yield (ϕ_{ClNO_2}). The most widely used parameterization for $\gamma_{\text{N}_2\text{O}_5}$
78 and ϕ_{ClNO_2} was proposed by Bertram and Thornton (2009), which is based on the laboratory studies with
79 considerations of temperature, relative humidity (RH), aerosol water content, concentrations of nitrate
80 and chloride, and specific surface area. However, recent field and model studies have shown that this
81 parameterization would overestimate both $\gamma_{\text{N}_2\text{O}_5}$ and ϕ_{ClNO_2} , especially in regions with high Cl levels
82 (McDuffie et al., 2018b; McDuffie et al., 2018a; Xia et al., 2019; Chang et al., 2016; Hong et al., 2020;
83 Yu et al., 2020). The discrepancies could be partly attributed to the complexity of atmospheric aerosols
84 (e.g. mixing state and complex coating materials) in contrast to the simple proxies used in laboratory
85 studies (Yu et al., 2020). Several parameterizations updated from the one by Bertram and Thornton (2009)
86 have been proposed by more recent studies based on field measurements and box model studies (Yu et
87 al., 2020; McDuffie et al., 2018a; McDuffie et al., 2018b; Xia et al., 2019). However, a full evaluation of



88 the representativeness of different parameterizations for the $\text{N}_2\text{O}_5 - \text{ClNO}_2$ chemistry and the associated
89 impacts on ambient air quality is not available yet.

90 In addition to the parameterization, the influence of the $\text{N}_2\text{O}_5 - \text{ClNO}_2$ chemistry is also sensitive to
91 chlorine emissions. In early modelling studies, global tropospheric chlorine is mainly from sea salt
92 aerosols (SSA), and most of the chlorine over continental regions in North America and Europe is
93 dominated by the long-range transport of SSA (Wang et al., 2019; Sherwen et al., 2017). The importance
94 of anthropogenic chlorine emissions, which were ignored in most studies, has been raised recently based
95 on both field measurements and model simulations (Le Breton et al., 2018; Yang et al., 2018; Wang et
96 al., 2020; Hong et al., 2020). The study by Wang et al. (2020b) suggested that anthropogenic emissions
97 could dominate reactive chlorine in China, resulting in an increase in $\text{PM}_{2.5}$ and Ozone by up to $3.2 \mu\text{g}$
98 m^{-3} and 1.9 ppbv on annual mean basis, respectively. The comprehensive effects of anthropogenic
99 chlorine on air quality as well as their sensitivities to different parameterizations for $\text{N}_2\text{O}_5 - \text{ClNO}_2$
100 chemistry, however, has not been investigated in previous studies.

101 In this work, we use the GEOS-Chem model to investigate the impacts of chlorine chemistry including
102 the heterogeneous $\text{N}_2\text{O}_5 - \text{ClNO}_2$ chemistry on air quality in China. Multiple observational data sets,
103 including N_2O_5 , ClNO_2 , O_3 , $\text{PM}_{2.5}$ and its chemical species from different representative sites across
104 China, are used to assess the model performance. With comprehensive chlorine emissions as well as
105 appropriate parameterizations for $\text{N}_2\text{O}_5 - \text{ClNO}_2$ chemistry, our objectives are: 1) to improve the model's
106 performance regarding the simulation of particulate chloride, ClNO_2 , N_2O_5 , $\text{PM}_{2.5}$ and O_3 concentrations;
107 and 2) to extend the investigation on the effects of chlorine chemistry on both $\text{PM}_{2.5}$ and ozone pollution
108 in China as well as their sensitivities to anthropogenic chlorine emissions and the parameterizations for
109 $\text{N}_2\text{O}_5 - \text{ClNO}_2$ chemistry.

110 **2 Methodology**

111 **2.1 GEOS-Chem Model**

112 The GEOS-Chem model (version 12.9.3, <http://www.geos-chem.org>) is driven by assimilated



113 meteorological fields GEOS-FP from the NASA Global Modeling and Assimilation Office (GMAO) at
114 NASA Goddard Space Flight Center. The simulation in this study was conducted in a nested-grid model
115 with a native horizontal resolution of $0.25^\circ \times 0.3125^\circ$ (latitude \times longitude) and 47 vertical levels over
116 East Asia ($70^\circ - 140^\circ$ E, 15° S – 55° N). The dynamical boundary conditions were from a global simulation
117 with a horizontal resolution of $2^\circ \times 2.5^\circ$. We initialized the model with a 1-month spin up followed by a
118 1-year simulation for 2018. The simulation included a detailed representation of coupled NO_x – ozone –
119 VOCs – aerosol – halogen chemistry (Sherwen et al., 2016). Previous studies have demonstrated the
120 ability of GEOS-Chem to reasonably reproduce the magnitude and seasonal variation of surface ozone
121 and particulate matter over East Asia and China (Wang et al., 2013; Geng et al., 2015; Li et al., 2019).

122 2.1.1 Chlorine Chemistry

123 The GEOS-Chem model includes a comprehensive chlorine chemistry mechanism coupled with bromine
124 and iodine chemistry. Full details could be found in the study of Wang et al. (2019b). Briefly, the model
125 includes 12 gas-phase inorganic chlorine species (Cl , Cl_2 , Cl_2O_2 , ClNO_2 , ClNO_3 , ClO , ClOO , OCIO ,
126 BrCl , ICl , HOCl , HCl), 3 gas-phase organic chlorine (CH_3Cl , CH_2Cl_2 , CHCl_3), and aerosol Cl^- in two
127 size bins (fine mode with radius $\leq 0.5 \mu\text{m}$ and coarse mode with radius $> 0.5 \mu\text{m}$). The gas-aerosol
128 equilibrium between HCl and Cl^- is calculated with ISORROPIA II (Fountoukis and Nenes, 2007) as
129 part of the $\text{H}_2\text{SO}_4 - \text{HCl} - \text{HNO}_3 - \text{NH}_3$ – non-volatile cations (NVCs) thermodynamic system, where
130 Na^+ is used as a proxy for NVCs.

131 The heterogeneous uptake of N_2O_5 on aerosol surfaces leading to the production of ClNO_2 and HNO_3
132 has also been included in GEOS-Chem with parameterizations for $\gamma_{\text{N}_2\text{O}_5}$ and ϕ_{ClNO_2} proposed by McDuffie
133 et al. (2018b; 2018a) by default (hereinafter referred to as McDuffie Parameterization). The $\gamma_{\text{N}_2\text{O}_5}$ can be
134 described by Eq. 1:

$$135 \quad \frac{1}{\gamma_{\text{N}_2\text{O}_5}} = \frac{1}{\gamma_{\text{core}}} + \frac{1}{\gamma_{\text{coat}}} \quad \text{Eq. 1}$$

136 Where γ_{core} represents the reactive uptake of inorganic aerosol core, and γ_{coat} represents the retardation of
137 the organic coating. More details of the parameterization can be referred to McDuffie et al. (2018b).

138 ϕ_{ClNO_2} is calculated following the study of Bertram and Thornton (2009), but is reduced by 75% based
139 on the observations conducted in eastern U.S. and offshore in spring 2015 (Lee et al., 2018). It could be



140 described by Eq. 2:

$$141 \quad \varphi_{\text{ClNO}_2} = 0.25 \times \left(\frac{k_2[\text{H}_2\text{O}]}{k_3[\text{Cl}^-]} + 1 \right)^{-1} \quad \text{Eq. 2}$$

142 Where $[\text{H}_2\text{O}]$ and $[\text{Cl}^-]$ are the concentrations of aerosol liquid water content and aerosol chloride,
143 respectively, and $k_3/k_2 = 450$. The McDuffie parameterization can reproduce observed mean values of
144 $\gamma_{\text{N}_2\text{O}_5}$ and φ_{ClNO_2} in eastern U.S. (McDuffie et al., 2018b; McDuffie et al., 2018a), but is still of great
145 uncertainty.

146 Recently, Yu et al. (2020) proposed new parameterizations of $\gamma_{\text{N}_2\text{O}_5}$ and φ_{ClNO_2} based on the direct
147 measurements at two sites in northern and southern China representing different atmospheric conditions.

148 The parameterizations (hereinafter referred to as Yu Parameterization) are described by Eq. 3 – 4:

$$149 \quad \gamma_{\text{N}_2\text{O}_5} = 6.12 \times 10^6 \times \frac{[\text{H}_2\text{O}]V}{c \cdot S_a} \left(1 - \frac{1}{0.033 \times \frac{[\text{H}_2\text{O}]}{[\text{NO}_3^-]} + 3.4 \times \frac{[\text{Cl}^-]}{[\text{NO}_3^-]} + 1} \right) \quad \text{Eq. 3}$$

$$150 \quad \varphi_{\text{ClNO}_2} = \left(1 + \frac{[\text{H}_2\text{O}]}{105[\text{Cl}^-]} \right)^{-1} \quad \text{Eq. 4}$$

151 Where c is an average gas-phase thermal velocity of N_2O_5 ; V and S_a are particle volume and surface area
152 densities, respectively; $[\text{H}_2\text{O}]$, $[\text{Cl}^-]$ and $[\text{NO}_3^-]$ are the concentrations of aerosol liquid water content,
153 aerosol chloride and aerosol nitrate, respectively.

154 In this study, we updated the $\text{N}_2\text{O}_5 - \text{ClNO}_2$ chemistry in GEOS-Chem with the Yu parameterization.
155 Additional simulation cases were also performed to evaluate the representativeness of both the Yu and
156 McDuffie Parameterizations regarding the simulation of N_2O_5 , ClNO_2 , O_3 , $\text{PM}_{2.5}$ and its chemical
157 speciation in China. Detailed description of the model setup for related cases is provided below in Section
158 2.1.3.

159 2.1.2 Emissions

160 The study uses the Hemispheric Transport of Air Pollution (HTAPv2, <http://www.htap.org/>) based on the
161 emission of 2010 as a global anthropogenic inventory. This inventory is overwritten by a regional
162 emission inventory MIX (with a horizontal resolution of $0.25^\circ \times 0.25^\circ$) over East Asia based on the
163 emission in 2017, which is developed for the Model Inter-Comparison Study for Asia (MICS-Asia) and
164 covers all major anthropogenic sources in 30 Asian countries and regions (Li et al., 2017). In addition,



165 anthropogenic emissions of black carbon (BC) and organic carbon (OC) in Guangdong province, China
166 ($109^{\circ} - 117^{\circ}$ E, $20^{\circ} - 26^{\circ}$ N) are overwritten by a more recent high-resolution inventory ($9 \text{ km} \times 9 \text{ km}$)
167 described by Huang et al. (2021). Biomass burning emissions are from the Global Fire Emissions
168 Database (GFED4) (Van et al., 2010) with a 3-hour time resolution. And the biogenic emissions of VOCs
169 are calculated based on the Model of Emissions of Gases and Aerosols from Nature (MEGAN2.1)
170 (Guenther et al., 2006).

171 Table 1 lists Cl emissions from all sources in the model. The global tropospheric chlorine by default in
172 the model is mainly from the mobilization of Cl^- from SSA distributed over two size bins (fine and coarse
173 modes) (Wang et al., 2019), which is computed online as the integrals of the size-dependent source
174 function depending on wind speeds and sea surface temperatures (Jaeglé et al., 2011). During the
175 simulation year of 2018, SSA contributes $6.5 \times 10^4 \text{ Gg Cl}^-$, most of which however are distributed over
176 the ocean due to its relatively short lifetime (~ 1.5 days) (Choi et al., 2020). The release of atomic Cl
177 from organic chlorine (CH_3Cl , CH_2Cl_2 and CHCl_3) via the oxidation by OH and Cl is also included in
178 the model by default. These organic chlorine gases are mainly of biogenic marine origin (Simmonds et
179 al., 2006), with a mean tropospheric lifetime longer than 250 days (Wang et al., 2020), and are simulated
180 in the model by imposing fixed surface concentrations as described by Schmidt et al. (2016). Total
181 emissions of Cl atom from CH_3Cl , CH_2Cl_2 , and CHCl_3 are estimated to be 3.8, 2.4, and $0.70 \text{ Gg Cl a}^{-1}$,
182 respectively.

183 Considering the importance of anthropogenic chlorine in China, we have further updated chlorine
184 inventories in the model to account for anthropogenic HCl, Cl_2 and fine particulate Cl^- , as well as biomass
185 burning HCl and Cl^- emissions (also shown in Table 1). For fine particulate Cl^- from both anthropogenic
186 and biomass burning, the emissions are estimated based on $\text{PM}_{2.5}$ emissions from MIX and GFED4
187 inventories combined with the emission ratios of fine particulate Cl^- versus $\text{PM}_{2.5}$ for different emission
188 sectors adopted from the study of Fu et al. (2018). Estimated Cl^- emissions from anthropogenic and
189 biomass burning are 379 and 120 Gg, respectively, comparable to the results of 486 Gg in total for the
190 year of 2014 by Fu et al. (2018). The anthropogenic emissions of HCl and Cl_2 are from ACEIC
191 (Anthropogenic Chlorine Emissions Inventory for China) (Liu et al., 2018) and are estimated to be 218
192 and 8.9 Gg Cl in China, respectively. For HCl from biomass burning, the emission factors from Lobert



193 et al. (1999) are used for different types of biomass provided in GFED4, and a total emission of 30 Gg
194 Cl is obtained in China in 2018. Total implemented chlorine emission for the simulation year of 2018 is
195 756 Gg Cl.

196 Figure 1 shows the distribution of Cl emissions from the sources mentioned above. Anthropogenic and
197 biomass burning emissions of HCl are concentrated in the Northeast Plain (NP), North China Plain (NCP),
198 Yangtze River Delta (YRD), and Sichuan Basin (SCB), and could be up to 320 kg Cl km⁻² a⁻¹ in the SCB.
199 Emissions of HCl are low in South China, mainly due to the low chlorine content of coal in these regions
200 (Hong et al., 2020). The relative contribution of biomass burning to total HCl emissions in China is 14%
201 on average, but could become dominant in the NP due to the discrepancies in the spatial distributions of
202 anthropogenic and biomass burning emissions. The anthropogenic Cl₂ emissions have a similar spatial
203 distribution to that of HCl, but are one order of magnitude lower than HCl emissions. The distribution of
204 non-sea salt Cl⁻ emissions is also similar to that of HCl and Cl₂, except that non-sea salt Cl⁻ emissions
205 are also high in Central China. In contrast, emissions of sea salt Cl⁻ (Fig. S1) are mainly distributed over
206 the ocean, implying limited influences over inland due to rapid deposition during transport. The spatial
207 distributions of different organic chlorine sources are similar, with maximums (~ 0.5 kg Cl km⁻² a⁻¹) in
208 coastal regions (Fig. S1).

209 2.1.3 Model setup for different simulation cases

210 In this study, we conducted a series of simulation cases to investigate the effects of chlorine chemistry
211 on air quality in China, the role of N₂O₅ – ClNO₂ chemistry, and the associated sensitivities to chlorine
212 emissions as well as the parameterizations for N₂O₅ – ClNO₂ chemistry. Detailed model setup for those
213 cases is listed in Table 2. The Base case is the one with all updates in this study, including additional
214 chlorine sources from anthropogenic and biomass burning emissions as well as improved N₂O₅ – ClNO₂
215 chemistry represented by the Yu parameterization. The NoEm case is conducted with a similar setup as
216 the Base case but only includes chlorine emissions from SSA and organic chlorine sources so as to
217 evaluate the model improvement originated from the updated chlorine emissions through the comparison
218 with the Base case. The McDuffie case is also performed using the McDuffie instead of Yu
219 parameterization for $\gamma_{\text{N}_2\text{O}_5}$ and ϕ_{ClNO_2} while keeping others the same as the Base case so as to evaluate the
220 discrepancies originated from different parameterizations for the N₂O₅ – ClNO₂ chemistry.



221 In addition, the NoHet case sets ϕ_{ClNO_2} to zero (i.e. $[\text{Cl}^-] = 0$ in Eq. 3 and 4) but keeps others the same as
222 the Base case so as to evaluate the importance of the of $\text{N}_2\text{O}_5 - \text{ClNO}_2$ chemistry through the comparison
223 with the Base case. Similarly, combined with three more sensitivity cases (NoChem, NoEmHet and
224 NoAll, see details in Table 2), the study provides an overall evaluation of the importance of tropospheric
225 chlorine chemistry as well as its sensitivities to chlorine emissions and the parameterizations for $\text{N}_2\text{O}_5 -$
226 ClNO_2 chemistry in the model.

227 2.2 Observations

228 Multiple observed data sets were applied in this study to evaluate the performance of GEOS-Chem
229 simulation, including the concentrations of chemical species of $\text{PM}_{2.5}$ from three representative sites,
230 located in south (Guangzhou, 23.14° N, 113.36° E), east (Dongying, 37.82° N, 119.05° E) and north
231 (Gucheng, 37.36° N, 115.96° E) China, respectively (Fig. S2). Concentrations of SO_4^{2-} , NO_3^- , NH_4^+ , Cl
232 and organic matter (OM) in $\text{PM}_{2.5}$ were measured by High Resolution Time-of-Flight Aerosol Mass
233 Spectrometer (HR-ToF-AMS; Aerodyne Research Inc., USA, Decarlo et al. (2006)) from October 2 to
234 November 18, 2018 (with a time resolution of 1 minute) at Guangzhou site (Chen et al., 2021b), and
235 from March 18 to April 21, 2018 (with a 1-minute time resolution) at Dongying site. Concentrations of
236 these species were measured by an Aerodyne Quadruple Aerosol Chemical Speciation Monitor (ACSM;
237 Aerodyne Research Inc., USA, Ng et al. (2011)) from November 11 to December 18 in 2018, with a time
238 resolution of 2 minutes at Gucheng site (Li et al., 2021).

239 Concentrations of N_2O_5 and ClNO_2 (with a time resolution of 1 minute) were also measured at
240 Guangzhou site by a Chemical Ionization Mass Spectrometer (CIMS, THS Instruments Inc., Atlanta,
241 (Kercher et al., 2009)) from September 25 to November 12 in 2018 (Ye et al., 2021). To have a thorough
242 evaluation of the representativeness of different parameterizations for $\gamma_{\text{N}_2\text{O}_5}$ and ϕ_{ClNO_2} , observations of
243 ClNO_2 and N_2O_5 at six more sites across China from previous studies (see Table S1 and Fig. S2) are also
244 used in this study. It should be noted that model results sampled at those sites for comparison were
245 simulated in the same months but different years while ignoring the uncertainties associated with the
246 interannual variability.

247 In addition, we also use observed hourly data of O_3 and $\text{PM}_{2.5}$ published by the China National



248 Environmental Monitoring Center (CNEMC, <http://www.cnemc.cn/sss/>, last access on June 20, 2021)
249 to evaluate the model's overall performance in China. The network was launched in 2013 as part of the
250 Clean Air Action Plan, and includes ~1500 stations located in 370 cities by 2018 (Fig. S2).

251 **3 Results and discussion**

252 **3.1 Improved model performance with updated chlorine emissions and N₂O₅ – ClNO₂ chemistry**

253 Figure 2 shows time series of observed and simulated Cl⁻ concentrations at the Guangzhou, Dongying,
254 and Gucheng sites. The observations show the lowest Cl⁻ concentrations at the Guangzhou site ($0.55 \pm$
255 $0.52 \mu\text{g m}^{-3}$), although the site is the closest to the ocean among all three sites, while the highest
256 concentrations ($4.7 \pm 3.3 \mu\text{g m}^{-3}$) are observed at the Gucheng site, away from the sea. Moderate
257 concentrations of Cl⁻ is observed at the Dongying site, around $1.1 \pm 0.82 \mu\text{g m}^{-3}$. The relatively higher
258 concentrations observed inland again suggest the dominance of non-sea salt Cl⁻ in China, as mentioned
259 before in the Introduction.

260 The comparison between observations and simulated results from the NoEm case shows a serious
261 underestimate of Cl⁻ concentrations, with normalized mean bias (NMB) ranging from -96% to -79%,
262 suggesting the missing of significant chlorine sources in addition to sea salt chlorine. In contrast, the
263 Base case with updated chlorine emissions exhibits much higher Cl⁻ concentrations and can successfully
264 reproduce observations, with average of 0.77 ± 0.54 , 0.71 ± 0.52 , and $4.5 \pm 2.4 \mu\text{g m}^{-3}$ at the Guangzhou,
265 Dongying, and Gucheng sites, respectively. The increase in Cl⁻ concentrations in the Base case compared
266 with the NoEm case is the most significant at the Gucheng site, by a factor of 24 (from 0.19 to $4.5 \mu\text{g m}^{-3}$
267 on average). The NMB in the Base case therefore has decreased to -36% – 39%. The slight
268 underestimates at the Dongying site in the Base case could be to some extent explained by the bias in
269 GFED4, which underestimates emissions from agricultural fires due to their small size and short duration
270 as suggested by the study of Zhang et al. (2020). In spite of that, the model with the Base case well
271 reproduces the overall distribution of the observed particulate chloride concentrations in China. The
272 correlation coefficients (r) between observed and model results at the three sites also increase from -0.05
273 – 0.61 in the NoEm case to 0.40 – 0.71 in the Base case. The significant improvement in the model



274 performance again suggests sources other than SSA play a key role in Cl⁻ concentrations in China.

275 The comparison between observed and simulated N₂O₅ (Fig. 3a) shows that NMB for the NoEm case are
276 -58%, 150%, 108% and 25% at the Guangzhou, Wangdu, Taizhou and Mount Tai sites, respectively. In
277 contrast, the corresponding NMB for the Base case are much smaller, -57%, 48%, 91% and 18%,
278 respectively. The improvement in the Base case is apparent at most sites, implying that additional
279 chlorine emissions combined with the updated N₂O₅ – ClNO₂ chemistry with Yu parameterization could
280 effectively increase the uptake coefficient of N₂O₅. Little improvement is found at the Guangzhou site (-
281 58% in the NoEm case vs. -57% in the Base case). Previous studies also found an underestimation of
282 N₂O₅ in the Pearl River Delta (PRD), which could be partly explained by the underestimation of the
283 sources (e.g. NO₂) and/or the overestimation of the sink of N₂O₅ there (Dai et al., 2020; Li et al., 2016).

284 The N₂O₅ results from the McDuffie case, which uses the McDuffie parameterization (a default setting
285 in GEOS-Chem, see Section 2.1.1 and 2.1.3) instead of Yu parameterization are also shown in Fig. 3a.
286 The NMB for the McDuffie case are -53%, 154%, 143% and 37% at the Guangzhou, Wangdu, Taizhou
287 and Mount Tai sites, respectively. The comparison indicates that Yu parameterization can reproduce
288 observed N₂O₅ better in China in general, while McDuffie parameterization tends to overestimate N₂O₅
289 concentrations, which could be attributed to the suppression from the organic coatings leading to a much
290 lower $\gamma_{\text{N}_2\text{O}_5}$ (see Eq. 1 for McDuffie parameterization). Similarly, Morgan et al. (2015) found significant
291 underestimates in $\gamma_{\text{N}_2\text{O}_5}$ when the effect of organic coatings is considered. The effect of organic coatings
292 strongly depends on organic composition, particle phase state, and the environmental factors (e.g. RH
293 and temperature), and thus remains poorly quantified (McDuffie et al., 2018b; Morgan et al., 2015). As
294 suggested by the study of Yu et al. (2020), the bias in $\gamma_{\text{N}_2\text{O}_5}$ in McDuffie parameterization could be to
295 some extent explained by the fact that the parameterization does not differentiate between water-soluble
296 and water-insoluble organic fractions and simplifies the morphology and phase state, resulting in an
297 underestimation of the solubility and the diffusivity of N₂O₅ in organic matter.

298 For the comparison of ClNO₂ (Fig. 3b), we use the mean nighttime (excluding the data at local time of
299 10:00 – 16:00) maximum mixing ratio, as suggested by Wang et al. (2019b and 2020b). Observed ClNO₂
300 is high in Guangzhou (1121 pptv) and Wangdu (~ 990 pptv), followed by Changping (~ 500 pptv) and
301 Beijing (~ 430 pptv). The lowest concentrations are obtained at Mount Tai and Mount TaiMoShan (~ 150



302 and 120 pptv, respectively) due to relatively clean condition at high altitude. The comparison between
303 observed and simulated ClNO₂ at different sites also suggests a better model performance for the Base
304 case with NMB in the range of -28% – 22%, compared with the NMB of -77% – -31% and -59% – -36%
305 for the NoEm and McDuffie cases, respectively. The large difference between the NoEm and Base cases
306 again emphasizes the important role of non-sea salt chlorine in the formation of ClNO₂. The
307 underestimates in McDuffie parameterization on the other hand may suggest that the scaling factor of
308 0.25 applied to ϕ_{ClNO_2} in Eq. 2 is not appropriate for the atmospheric condition in China. A scaling factor
309 of 0.5 instead of 0.25 may fit better with observations used in the study. More field measurements and
310 model evaluations are required to come up with a more precise scaling factor better representing ϕ_{ClNO_2}
311 in China. Overall, with updated chlorine emissions and Yu parameterization for $\gamma_{\text{N}_2\text{O}_5}$ and ϕ_{ClNO_2} , the Base
312 case agrees better with both the magnitude and the spatial variation of observed N₂O₅ and ClNO₂ in
313 China.

314 To further elucidate how the model behaves in reproducing the spatial distribution of ozone and PM_{2.5}
315 through the incorporation of the additional chlorine emissions and updated N₂O₅ – ClNO₂ chemistry with
316 Yu parameterization, simulated MDA8 O₃ and PM_{2.5} from different cases were compared with
317 observations across China. Figure 4 shows observed annual mean MDA8 O₃ and PM_{2.5} in 2018 in China
318 from CNEMC compared with the model results from the Base case. The observed annual mean MDA8
319 O₃ and PM_{2.5} are 49 ppbv and 39 $\mu\text{g m}^{-3}$ respectively in 2018 in China. Model results from the Base case
320 could generally reproduce observed spatial and seasonal variations of annual mean MDA8 O₃ and PM_{2.5}
321 concentrations, with NMB of -26% and 3.6% and r of 0.83 and 0.81 respectively.

322 Table 3 also summarized the model performance on both annual and seasonal scales regarding the
323 simulation of O₃ and PM_{2.5} from different cases. For the comparison with observed MDA8 O₃, although
324 different simulation cases show a similar range of r , the Base case tends to have a slightly smaller bias
325 in general, with NMB of -26% on annual average (-49% – -5.5% on seasonal mean) vs. -28% (-54% – -
326 5.9%) in the NoEm and -27% (-53% – -5.2%) in the McDuffie case. For the comparison with observed
327 PM_{2.5}, the NMB bias from the Base case is 3.6% on annual average (-6.3% – 28% on seasonal mean).
328 Compared with the NoEm case, there is some improvement in summer (5.0% vs. 3.9%) and winter (-
329 7.9% vs. -4.3%) but slightly larger bias in autumn (26% vs. 28%). The McDuffie case on the other hand



330 produces slightly higher $PM_{2.5}$ concentrations, with NMB of 5.6%. Regarding the simulation of
331 particulate chemical species (Table S2), although the model performance varies with sites and species,
332 the Base case demonstrates better agreement with observations compared with the NoEm and McDuffie
333 cases in general.

334 On the whole, the model performance is better with the additional anthropogenic and biomass burning
335 chlorine emissions combined with updated $N_2O_5 - ClNO_2$ chemistry represented by Yu parameterization.
336 Therefore, the following investigation of the impacts of chlorine chemistry on air quality in China as
337 well as their sensitivities to chlorine emissions and parameterizations for $N_2O_5 - ClNO_2$ chemistry is
338 mainly based on the Base case.

339 **3.2 Impacts of tropospheric chlorine chemistry on air quality and the role of $N_2O_5 - ClNO_2$** 340 **chemistry**

341 To comprehensively quantify the importance of chlorine chemistry, we conducted a sensitivity case in
342 which all related tropospheric chlorine chemistry was turned off (NoChem, also listed in Table 2). The
343 differences between the Base and NoChem cases (Fig. 5 and Fig. S3) could thus represent the impact of
344 the chlorine chemistry. The comparison shows that chlorine chemistry could increase annual mean
345 nighttime max $ClNO_2$ surface concentrations by 243 pptv averaged in China (up to 1548 pptv in the
346 SCB). The increase in annual mean Cl atoms is 1.7×10^3 molec cm^{-3} averaged in Chian (up to 7×10^3
347 molec cm^{-3} in coastal regions), which further lead to an increase in surface annual mean MDA8 O_3 and
348 OH by 1.1 ppbv and 3.8×10^4 molec cm^{-3} averaged in China (up to 4.5 ppbv and 1.2×10^5 molec cm^{-3} in
349 the SCB) respectively through VOCs oxidation. In contrast, annual mean $PM_{2.5}$ surface concentrations
350 are decreased by $0.91 \mu g m^{-3}$ averaged in China (up to $7.9 \mu g m^{-3}$ in the SCB), mainly due to the decrease
351 of NO_3^- and NH_4^+ (up to $6.4 \mu g m^{-3}$ and $1.9 \mu g m^{-3}$ respectively), although SO_4^{2-} concentrations are
352 increased slightly by up to $1.2 \mu g m^{-3}$ in the SCB (Fig. S3).

353 Previous studies suggested that tropospheric chlorine drives a global decrease of O_3 by catalytic
354 production of bromine and iodine radicals from hypohalous acids (HOX, where X = Br and I) (Schmidt
355 et al., 2016; Wang et al., 2019). On the contrary, $N_2O_5 - ClNO_2$ chemistry can enhance O_3 through the
356 production of Cl atoms and the recycling of NO_x . Therefore, we further investigate the role that $N_2O_5 -$



357 ClNO₂ chemistry plays in tropospheric chlorine chemistry through the comparison between the Base and
358 NoHet (Fig. 6 and Fig. S4). The comparison illustrates that the N₂O₅ – ClNO₂ chemistry could result in
359 a significant production of ClNO₂, reaching 600 – 1400 pptv for annual mean nighttime max surface
360 concentrations in the NCP, and up to 1546 pptv in the SCB. The change in the surface concentrations of
361 Cl atoms (an annual mean increase of $1 - 4 \times 10^3$ molec cm⁻³ in central and eastern China) is mainly due
362 to the photolysis of ClNO₂ and accounts for 74% of total change in annual mean Cl atoms due to all
363 tropospheric chlorine chemistry in China, which are consistent with the results from the previous study
364 by Liu et al. (2017).

365 Consequently, the annual mean MDA8 O₃ surface concentrations are increased by 1.5 – 3 ppbv in central
366 and eastern China and by up to 3.8 ppbv in the SCB, accounting for 83% of total change in annual mean
367 MDA8 O₃ due to all tropospheric chlorine chemistry in China. It is interesting to note that while MDA8
368 O₃ surface concentrations show maxima in summer and minima in winter in general, the influence of
369 N₂O₅ – ClNO₂ chemistry on O₃ concentrations exhibits a different seasonality. The increase in seasonal
370 mean MDA8 O₃ concentrations is the largest in winter (by up to 6.5 ppbv in the SCB), but is less than
371 1.5 ppbv in summer. This is because of more accumulation of N₂O₅ and ClNO₂ in dark conditions in long
372 winter nights (Sarwar et al., 2014).

373 There also exhibits an obvious decrease in the annual mean surface concentrations of PM_{2.5} attributed to
374 the N₂O₅ – ClNO₂ chemistry, ranging from 1.5 to 4.5 μg m⁻³ in central and eastern China (accounting for
375 90% of total change in annual mean PM_{2.5} due to all tropospheric chlorine chemistry in China). The
376 decrease is more significant in autumn and winter in China, with a range of 3.5 – 5.5 μg m⁻³ in central
377 and eastern China, and being up to 11 μg m⁻³ in the SCB. In contrast, the decrease in PM_{2.5} is less than 2
378 μg m⁻³ in summer in China. The change in PM_{2.5} is mainly due to the decrease in NO₃⁻ (up to 6.2 μg m⁻³
379 in the SCB on annual average). In addition, NH₄⁺ is also decreased by up to 1.8 μg m⁻³ in the SCB on
380 annual average, following the pattern of ΔNO₃⁻. This is because NH₃ is in excess in most regions in China
381 (Xu et al., 2019) and the formation of ClNO₂ via R2 could hinder the formation of HNO₃ and shift the
382 partitioning between NH₃ and NH₄⁺ towards NH₃. Unlike the change in NO₃⁻ and NH₄⁺, the N₂O₅ – ClNO₂
383 chemistry increases surface SO₄²⁻ concentration slightly, which could be explained by the enhancements
384 of atmospheric oxidation associated with the increase in Cl atoms, OH and O₃, facilitating the formation



385 of secondary aerosols (Sarwar et al., 2014).

386 On the other hand, the effect of tropospheric chlorine chemistry without the $\text{N}_2\text{O}_5 - \text{ClNO}_2$ chemistry is
387 much smaller (Fig. S5), leading to an increase of up to 0.7 ppbv in inland China and a decrease of 0.3 –
388 0.5 ppbv in coastal regions for annual mean MDA8 O_3 concentrations. The increase is probably
389 associated with Cl atoms from photolysis of gas-phase chlorine, especially non-sea salt Cl_2 in inland
390 China, while the decrease at coastal regions is mainly due to catalytic production of bromine and iodine
391 radicals originated from sea-salt aerosols. The comparison demonstrates the dominance of $\text{N}_2\text{O}_5 - \text{ClNO}_2$
392 chemistry in total tropospheric chlorine chemistry in China.

393 3.3 The effect of $\text{N}_2\text{O}_5 - \text{ClNO}_2$ chemistry in response to chlorine emissions

394 Since both $\gamma_{\text{N}_2\text{O}_5}$ and ϕ_{ClNO_2} in the Yu parameterization are highly dependent on $[\text{Cl}^-]$, the effect of N_2O_5
395 – ClNO_2 chemistry on air quality is thus sensitive to chlorine emissions. Figure 7 shows the effects of
396 additional chlorine emissions from anthropogenic and biomass burning sources on annual mean surface
397 concentrations of different species (Cl^- , Cl atom, OH, MDA8 O_3 , $\text{PM}_{2.5}$ and NO_3^-) in China, calculated
398 as the differences between the Base and the NoEm case. With the implementation of the additional
399 chlorine emissions, the particulate Cl^- concentration increased significantly in inland China, with the
400 largest increase in the SCB ($4.5 \mu\text{g m}^{-3}$) and little change in west China. The increase is in the range of
401 $1.5 - 3.5 \mu\text{g m}^{-3}$ in the NCP and $< 0.5 \mu\text{g m}^{-3}$ in South China. The spatial distribution of ΔCl atoms is
402 also consistent with that of the additional chlorine emissions and ΔCl^- , showing the largest increment in
403 the SCB (about $4.5 - 5 \times 10^3 \text{ molec cm}^{-3}$). There is also a moderate increase in Cl atoms in the NP and
404 NCP, with a range of $1.5 - 4 \times 10^3 \text{ molec cm}^{-3}$. Only a minor increase of Cl atoms is found in South China
405 ($< 1 \times 10^3 \text{ molec cm}^{-3}$).

406 The reactions of VOCs with Cl atoms lead to a further increase in OH (an annual mean increase of $2 - 9$
407 $\times 10^4 \text{ molec cm}^{-3}$ in central and eastern China) and O_3 (Wang et al., 2016; Sarwar et al., 2014). The
408 increase in O_3 could also be due to the recycling of NO_x back to the atmosphere associated with the
409 photolysis of ClNO_2 (R4). The increase in MDA8 O_3 surface concentrations ranges from 0.5 to 3 ppbv
410 in central and eastern China, and reaches up to 3.5 ppbv in the SCB on annual average. The impacts of
411 chlorine sources on O_3 formation also vary with seasons. Although O_3 pollution is generally severe in



412 summer, the change in MDA8 O₃ due to chlorine sources is relatively minor, with maxima of 0.7 ppbv
413 in the SCB and < 0.5 ppbv in most other regions averaged in summer. In contrast, the increase is most
414 obvious in winter, with maxima of 5.2 ppbv in the SCB on seasonal average.

415 The effects of the additional chlorine emissions on surface PM_{2.5} concentrations is complicated. The NCP
416 shows the largest increase (3 – 4.5 μg m⁻³ on annual average), mainly due to the increase in Cl⁻, which
417 could also promote N₂O₅ – ClNO₂ chemistry and lead to more NO₃⁻ production (Chen et al., 2021a). In
418 contrast, the SCB exhibits both an increase (by up to 4.2 μg m⁻³) and a decrease (by up to 3.7 μg m⁻³). The
419 decrease of PM_{2.5} in the SCB is mainly due to the large decrease of NO₃⁻ there. In the SCB, nitrate
420 formation is dominated by the heterogeneous hydrolysis of N₂O₅ (Tian et al., 2019) while the additional
421 Cl⁻ could hinder the path of N₂O₅ hydrolysis due to competition with the path of ClNO₂ formation.
422 Consequently, the additional chlorine emissions result in a decrease of NO₃⁻ up to 5.6 μg m⁻³ in SCB on
423 annual average.

424 In addition, NH₄⁺ concentrations could also be affected through the reaction of R5:



426 In the NP and NCP where anthropogenic and biomass burning emissions of HCl are high, the annual
427 mean NH₄⁺ surface concentrations are increased by 0.5 – 1.5 μg m⁻³ (Fig. S6). NH₄⁺ concentrations are
428 also affected by the gas-particle partitioning equilibrium, and decrease as the pH value gets higher (or
429 increase with H⁺ concentrations). Therefore, the competition between the N₂O₅ – ClNO₂ chemistry and
430 N₂O₅ hydrolysis could also affect the formation of NH₄⁺. In other words, increased Cl⁻ concentrations
431 could result in less H⁺ and thus less NH₄⁺. Consequently, there also exists some decrease in NH₄⁺
432 concentrations in the SCB associated with the large decrease in NO₃⁻ concentrations. In contrast, little
433 change is found for surface SO₄²⁻ concentrations, less than 0.5 μg m⁻³ in most regions of China (Fig. S6).

434 It is worth mentioning that the effects of the additional chlorine emissions work mainly through the N₂O₅
435 – ClNO₂ chemistry. Without this heterogeneous chemistry, the increase of chlorine emissions shows only
436 a minor change in Cl atoms (< 10³ molec cm⁻³ in China, estimated as the difference between the NoHet
437 and NoEmHet cases in Fig. S7). The impact of chlorine emissions on O₃ concentrations also weakens
438 when the N₂O₅ – ClNO₂ chemistry is turned off, with an increase of 0.5 – 1 ppbv in MDA8 O₃ on annual
439 average (vs. 0.5 – 3 ppbv mentioned above).



440 On the other hand, the impacts of $\text{N}_2\text{O}_5 - \text{ClNO}_2$ chemistry on air quality in inland China would be
441 seriously underestimated if the additional anthropogenic and biomass burning chlorine sources are
442 ignored. If only seas salt chlorine emission is included in the simulation, the increase of ClNO_2 surface
443 concentrations resulted from $\text{N}_2\text{O}_5 - \text{ClNO}_2$ chemistry only occurs in coastal regions due to the
444 heterogeneous uptake of N_2O_5 on sea salt chloride aerosols (by up to 260 pptv on annual average,
445 indicated by the difference between the NoEm and NoEmHet cases, Fig. S8). Consequently, the increase
446 in Cl atoms and MDA8 O_3 surface concentrations is found mainly in coastal regions. For instance, annual
447 mean MDA8 O_3 concentrations are increased by up to 2 ppbv in coastal regions, but by less than 0.5
448 ppbv in inland China. In other words, the dominance of $\text{N}_2\text{O}_5 - \text{ClNO}_2$ chemistry in the impact of chlorine
449 chemistry on air quality in China is to large extent driven by the additional chlorine emissions.

450 **3.4 The effect of $\text{N}_2\text{O}_5 - \text{ClNO}_2$ chemistry in response to parameterizations for $\gamma_{\text{N}_2\text{O}_5}$ and ϕ_{ClNO_2}**

451 It should be noted that the impact of $\text{N}_2\text{O}_5 - \text{ClNO}_2$ chemistry on air quality not only depends on the
452 amount of chlorine emissions, but is also sensitive to the parameterizations for $\gamma_{\text{N}_2\text{O}_5}$ and ϕ_{ClNO_2} . As
453 discussed earlier (Fig. 3a), there exists a large difference in simulated N_2O_5 between the Base and NoEm
454 cases at the Wangdu site, implying the sensitivity of $\gamma_{\text{N}_2\text{O}_5}$ to chlorine emissions in the Yu
455 parameterization and thus the importance of non-sea salt chlorine emissions in inland China. The
456 comparison between the Base and NoHet cases (not shown here) also suggests that the heterogeneous
457 uptake of N_2O_5 on chloride-containing aerosol surfaces in the Yu Parameterization is an important loss
458 pathway of N_2O_5 , and should not be ignored.

459 In contrast, N_2O_5 concentrations have little dependence on chlorine emissions in the McDuffie
460 parameterization (Fig. 3a). This insensitivity to chlorine emissions could be expected from Eq. 1 where
461 $\gamma_{\text{N}_2\text{O}_5}$ relies mainly on total concentrations of inorganic species, of which chlorine is only a minor
462 component. The smaller dependence of $\gamma_{\text{N}_2\text{O}_5}$ on concentrations of Cl together with the lower value of
463 ϕ_{ClNO_2} make the results from the McDuffie case less sensitive to chlorine emissions, producing less
464 ClNO_2 and Cl atoms compared with the Base case (Yu parameterization) although with the same emission.
465 Consequently the McDuffie case produces less O_3 , with annual mean surface concentrations of MDA8
466 O_3 lower by 0.47 ppbv averaged in China (by up to 2 ppbv in the SCB), but results in more $\text{PM}_{2.5}$ (0.63
467 $\mu\text{g m}^{-3}$ averaged in China and up to 4.7 $\mu\text{g m}^{-3}$ in the SCB on annual mean basis) mainly due to NO_3^- (Fig.



468 8). In other words, compared to the Base case with the Yu parameterization, the impacts of chlorine
469 emissions on annual MDA8 O₃ and PM_{2.5} in the McDuffie case has been decreased by 48% and 27%
470 averaged in China, respectively. Therefore, even with the same amounts of emissions, the impacts of
471 N₂O₅ – ClNO₂ chemistry on air quality varies significantly with different parameterizations.

472 4 Conclusions

473 Considering the importance of chlorine chemistry in modulating the O₃ and PM_{2.5} as well as the
474 previously ignored chlorine emission from anthropogenic and biomass burning, we updated the GOES-
475 Chem model in this study with comprehensive chlorine emissions and a new parameterization based on
476 the study of Yu et al. (2020) for N₂O₅ – ClNO₂ chemistry, followed by the extensive evaluation of model
477 performance. Through the utilization of a large number of observational datasets, we found a substantial
478 improvement has been achieved by the additional chlorine emissions, with NMB decreasing from -96%
479 – -79% to -36% – 39% for Cl⁻ simulation. The comparison with observed N₂O₅ and ClNO₂ also indicates
480 better model performance with Yu parameterization while $\gamma_{\text{N}_2\text{O}_5}$ and ϕ_{ClNO_2} are underestimated in
481 McDuffie parameterization (a default setting in GEOS-Chem), resulting in larger model bias. The
482 simulation of O₃ and PM_{2.5} also agrees better with observations in general in the Base case (with the
483 additional chlorine emissions and Yu parameterization) than the others.

484 Total tropospheric chlorine chemistry could increase Cl atoms by up to 7×10^3 molec cm⁻³, and leads to
485 an increase of up to 4.5 ppbv in MDA8 O₃ but a decrease of up to 7.9 μg m⁻³ in PM_{2.5} concentrations on
486 an annual mean basis in China. The decrease in PM_{2.5} is mainly associated with the decrease in NO₃⁻ and
487 NH₄⁺, by up to 6.4 and 1.9 μg m⁻³, respectively. The results also indicate that the N₂O₅ – ClNO₂ chemistry
488 dominate the impact of chlorine chemistry, accounting for 83% and 90% of total change in O₃ and PM_{2.5}
489 concentrations. In other words, the chlorine chemistry without the N₂O₅ – ClNO₂ chemistry has a minor
490 effect on annual mean MDA8 O₃ (less than 0.7 ppbv) and PM_{2.5} (less than 1.5 μg m⁻³) concentrations in
491 China. This mechanism is particularly useful in elucidating the commonly seen O₃ underestimations (e.g.
492 Ma et al. (2019)).

493 The effect of N₂O₅ – ClNO₂ chemistry is sensitive to chlorine emissions. With the additional



494 anthropogenic and biomass burning sources, simulated $PM_{2.5}$ concentrations are increased by up to 4.5
495 $\mu\text{g m}^{-3}$ in the NCP but decreased by up to $3.7 \mu\text{g m}^{-3}$ in the SCB on an annual basis. The latter is mainly
496 driven by the decrease of NO_3^- due to the competition between the formation of ClNO_2 and HNO_3 upon
497 the uptake of N_2O_5 on aerosol surfaces. The additional emissions also increase Cl atoms and OH in China
498 associated with the photolysis of ClNO_2 , consequently leading to an increase of annual mean MDA8 O_3
499 concentrations by up to 3.5 ppbv. In contrast, the significance of the $\text{N}_2\text{O}_5 - \text{ClNO}_2$ chemistry especially
500 over inland China would be severely underestimated if only sea salt chlorine is considered, with only a
501 slight increase in MDA8 O_3 (< 0.5 ppbv) and a minor decrease in $PM_{2.5}$ ($< 1.5 \mu\text{g m}^{-3}$) in inland China.
502 Moreover, we found the importance of chlorine chemistry not only depends on the amount of emissions,
503 but is also sensitive to the parameterizations for $\text{N}_2\text{O}_5 - \text{ClNO}_2$ chemistry. Although with the same
504 emission, the effects on MDA8 O_3 and $PM_{2.5}$ in China from the McDuffie case are lower compared to
505 the results with the Yu parameterization: differing by 48% and 27% in the annual average, respectively.

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514 **Competing interests.**

515 The authors declare that they have no conflict of interest.



516 **Data availability**

517 The data used in this study are available upon request from Qiaoqiao Wang (qwang@jnu.edu.cn).

518 **References**

519 Atkinson, R., Baulch, D. L., Cox, R. A., Crowley, J. N., Hampson, R. F., Hynes, R. G., Jenkin, M. E.,
520 Rossi, M. J., Troe, J., and Subcommittee, I.: Evaluated kinetic and photochemical data for
521 atmospheric chemistry: Volume II - gas phase reactions of organic species, *Atmos. Chem. Phys.*, 6,
522 3625-4055, 10.5194/acp-6-3625-2006, 2006.

523 Bertram, T. H. and Thornton, J. A.: Toward a general parameterization of N₂O₅ reactivity on aqueous
524 particles: the competing effects of particle liquid water, nitrate and chloride, *Atmospheric Chemistry
525 & Physics Discussions*, 9, 191-198, 2009.

526 Chang, W. L., Brown, S. S., Stutz, J., Middlebrook, A. M., Bahreini, R., Wagner, N. L., Dubé, W. P.,
527 Pollack, I. B., Ryerson, T. B., and Riemer, N.: Evaluating N₂O₅ heterogeneous hydrolysis
528 parameterizations for CalNex 2010, *J. Geophys. Res.: Atmos.*, 121, 5051-5070,
529 <https://doi.org/10.1002/2015JD024737>, 2016.

530 Chen, C., Zhang, H., Yan, W., Wu, N., Zhang, Q., and He, K.: Aerosol water content enhancement leads
531 to changes in the major formation mechanisms of nitrate and secondary organic aerosols in winter
532 over the North China Plain, *Environ. Pollut.*, 287, 117625,
533 <https://doi.org/10.1016/j.envpol.2021.117625>, 2021a.

534 Chen, W., Ye, Y., Hu, W., Zhou, H., Pan, T., Wang, Y., Song, W., Song, Q., Ye, C., Wang, C., Wang, B.,
535 Huang, S., Yuan, B., Zhu, M., Lian, X., Zhang, G., Bi, X., Jiang, F., Liu, J., Canonaco, F., Prevot,
536 A. S. H., Shao, M., and Wang, X.: Real-Time Characterization of Aerosol Compositions, Sources,
537 and Aging Processes in Guangzhou During PRIDE-GBA 2018 Campaign, *J. Geophys. Res.: Atmos.*,
538 126, e2021JD035114, <https://doi.org/10.1029/2021JD035114>, 2021b.

539 Choi, M. S., Qiu, X., Zhang, J., Wang, S., Li, X., Sun, Y., Chen, J., and Ying, Q.: Study of Secondary
540 Organic Aerosol Formation from Chlorine Radical-Initiated Oxidation of Volatile Organic
541 Compounds in a Polluted Atmosphere Using a 3D Chemical Transport Model, *Environ. Sci.
542 Technol.*, 54, 13409-13418, 10.1021/acs.est.0c02958, 2020.

543 Dai, J., Liu, Y., Wang, P., Fu, X., Xia, M., and Wang, T.: The impact of sea-salt chloride on ozone through
544 heterogeneous reaction with N₂O₅ in a coastal region of south China, *Atmos. Environ.*, 236, 117604,
545 <https://doi.org/10.1016/j.atmosenv.2020.117604>, 2020.

546 DeCarlo, P. F., Kimmel, J. R., Trimborn, A., Northway, M. J., Jayne, J. T., Aiken, A. C., Gonin, M., Fuhrer,



- 547 K., Horvath, T., Docherty, K. S., Worsnop, D. R., and Jimenez, J. L.: Field-Deployable, High-
548 Resolution, Time-of-Flight Aerosol Mass Spectrometer, *Anal. Chem.*, 78, 8281-8289,
549 10.1021/ac061249n, 2006.
- 550 Finlayson-Pitts, B. J., Ezell, M. J., and Pitts, J. N. J.: Formation of chemically active chlorine compounds
551 by reactions of atmospheric NaCl particles with gaseous N₂O₅ and ClONO₂, *Cheminform*, 20, 241-
552 244, 1989.
- 553 Fountoukis, C. and Nenes, A.: ISORROPIA II: a computationally efficient thermodynamic equilibrium
554 model for K⁺-Ca²⁺-Mg²⁺-NH₄⁺-Na⁺-SO₄²⁻-NO₃⁻-Cl⁻-H₂O aerosols, *Atmos. Chem. Phys.*, 7, 4639-
555 4659, 10.5194/acp-7-4639-2007, 2007.
- 556 Fu, X., Wang, T., Wang, S., Zhang, L., Cai, S., Xing, J., and Hao, J.: Anthropogenic Emissions of
557 Hydrogen Chloride and Fine Particulate Chloride in China, *Environ. Sci. Technol.*, 52, 1644, 2018.
- 558 Geng, G., Zhang, Q., Martin, R. V., Van Donkelaar, A., Huo, H., Che, H., Lin, J., and He, K.: Estimating
559 long-term PM_{2.5} concentrations in China using satellite-based aerosol optical depth and a chemical
560 transport model, *Remote Sens. Environ.*, 166, 262-270, 2015.
- 561 Guenther, A., Karl, T., Harley, P., Wiedinmyer, C., Palmer, P. I., and Geron, C.: Estimates of global
562 terrestrial isoprene emissions using MEGAN (Model of Emissions of Gases and Aerosols from
563 Nature), *Atmos. Chem. Phys.*, 6, 3181-3210, 10.5194/acp-6-3181-2006, 2006.
- 564 Haskins, J. D., Lee, B. H., Lopez-Hilifiker, F. D., Peng, Q., Jaeglé, L., Reeves, J. M., Schroder, J. C.,
565 Campuzano-Jost, P., Fibiger, D., McDuffie, E. E., Jiménez, J. L., Brown, S. S., and Thornton, J. A.:
566 Observational Constraints on the Formation of Cl₂ From the Reactive Uptake of ClONO₂ on Aerosols
567 in the Polluted Marine Boundary Layer, *J. Geophys. Res.: Atmos.*, 124, 8851-8869,
568 <https://doi.org/10.1029/2019JD030627>, 2019.
- 569 Hong, Y., Liu, Y., Chen, X., Fan, Q., Chen, C., Chen, X., and Wang, M.: The role of anthropogenic
570 chlorine emission in surface ozone formation during different seasons over eastern China, *Sci. Total
571 Environ.*, 723, 137697, <https://doi.org/10.1016/j.scitotenv.2020.137697>, 2020.
- 572 Hossaini, R., Chipperfield, M. P., Saiz-Lopez, A., Fernandez, R., Monks, S., Feng, W., Brauer, P., and
573 von Glasow, R.: A global model of tropospheric chlorine chemistry: Organic versus inorganic
574 sources and impact on methane oxidation, *J. Geophys. Res.: Atmos.*, 121, 14,271-214,297,
575 <https://doi.org/10.1002/2016JD025756>, 2016.
- 576 Huang, Z., Zhong, Z., Sha, Q., Xu, Y., Zhang, Z., Wu, L., Wang, Y., Zhang, L., Cui, X., Tang, M., Shi,
577 B., Zheng, C., Li, Z., Hu, M., Bi, L., Zheng, J., and Yan, M.: An updated model-ready emission
578 inventory for Guangdong Province by incorporating big data and mapping onto multiple chemical
579 mechanisms, *Sci. Total Environ.*, 769, 144535, <https://doi.org/10.1016/j.scitotenv.2020.144535>,
580 2021.



- 581 Jaeglé, L., Quinn, P. K., Bates, T. S., Alexander, B., and Lin, J. T.: Global distribution of sea salt aerosols:
582 new constraints from in situ and remote sensing observations, *Atmos. Chem. Phys.*, 11, 3137-3157,
583 10.5194/acp-11-3137-2011, 2011.
- 584 Kercher, J. P., Riedel, T. P., and Thornton, J. A.: Chlorine activation by N₂O₅: simultaneous, in situ
585 detection of ClNO₂ and N₂O₅ by chemical ionization mass spectrometry, *Atmospheric
586 Measurement Techniques Discussions*, 2009.
- 587 Lawler, M. J., Sander, R., Carpenter, L. J., Lee, J. D., and Saltzman, E. S.: HOCl and Cl₂ observations
588 in marine air, *Atmos. Chem. Phys.*, 11, 8115-8144, 10.5194/acp-11-7617-2011, 2011.
- 589 Le Breton, M., Hallquist, Å. M., Pathak, R. K., Simpson, D., Wang, Y., Johansson, J., Zheng, J., Yang,
590 Y., Shang, D., Wang, H., Liu, Q., Chan, C., Wang, T., Bannan, T. J., Priestley, M., Percival, C. J.,
591 Shallcross, D. E., Lu, K., Guo, S., Hu, M., and Hallquist, M.: Chlorine oxidation of VOCs at a semi-
592 rural site in Beijing: Significant chlorine liberation from ClNO₂ and subsequent gas and particle
593 phase Cl-VOC production, *Atmos. Chem. Phys.*, 18, 13013-13030, 10.5194/acp-18-13013-2018,
594 2018.
- 595 Lee, B. H., Lopez-Hilfiker, F. D., Schroder, J. C., Campuzano-Jost, P., Jimenez, J. L., McDuffie, E. E.,
596 Fibiger, D. L., Veres, P. R., Brown, S. S., Campos, T. L., Weinheimer, A. J., Flocke, F. F., Norris, G.,
597 O'Mara, K., Green, J. R., Fiddler, M. N., Bililign, S., Shah, V., Jaeglé, L., and Thornton, J. A.:
598 Airborne Observations of Reactive Inorganic Chlorine and Bromine Species in the Exhaust of Coal-
599 Fired Power Plants, *J. Geophys. Res.: Atmos.*, 123, 11,225-211,237,
600 <https://doi.org/10.1029/2018JD029284>, 2018.
- 601 Li, G., Su, H., Ma, N., Tao, J., Kuang, Y., Wang, Q., Hong, J., Zhang, Y., Kuhn, U., Zhang, S., Pan, X.,
602 Lu, N., Tang, M., Zheng, G., Wang, Z., Gao, Y., Cheng, P., Xu, W., Zhou, G., Zhao, C., Yuan, B.,
603 Shao, M., Ding, A., Zhang, Q., Fu, P., Sun, Y., Pöschl, U., and Cheng, Y.: Multiphase chemistry
604 experiment in Fogs and Aerosols in the North China Plain (McFAN): integrated analysis and
605 intensive winter campaign 2018, *Faraday Discuss.*, 226, 207-222, 10.1039/D0FD00099J, 2021.
- 606 Li, K., Jacob, D. J., Liao, H., Shen, L., Zhang, Q., and Bates, K. H.: Anthropogenic drivers of 2013–2017
607 trends in summer surface ozone in China, *P. Natl. Acad. Sci.*, 116, 422-427,
608 10.1073/pnas.1812168116, 2019.
- 609 Li, M., Zhang, Q., Kurokawa, J. I., Woo, J. H., He, K., Lu, Z., Ohara, T., Song, Y., Streets, D. G.,
610 Carmichael, G. R., Cheng, Y., Hong, C., Huo, H., Jiang, X., Kang, S., Liu, F., Su, H., and Zheng,
611 B.: MIX: a mosaic Asian anthropogenic emission inventory under the international collaboration
612 framework of the MICS-Asia and HTAP, *Atmos. Chem. Phys.*, 17, 935-963, 10.5194/acp-17-935-
613 2017, 2017.
- 614 Li, Q., Zhang, L., Wang, T., Tham, Y. J., Ahmadov, R., Xue, L., Zhang, Q., and Zheng, J.: Impacts of
615 heterogeneous uptake of dinitrogen pentoxide and chlorine activation on ozone and reactive
616 nitrogen partitioning: improvement and application of the WRF-Chem model in southern China,



- 617 Atmos. Chem. Phys., 16, 14875-14890, 10.5194/acp-16-14875-2016, 2016.
- 618 Liu, X., Qu, H., Huey, L. G., Wang, Y., Sjostedt, S., Zeng, L., Lu, K., Wu, Y., Hu, M., Shao, M., Zhu, T.,
619 and Zhang, Y.: High Levels of Daytime Molecular Chlorine and Nitryl Chloride at a Rural Site on
620 the North China Plain, Environ. Sci. Technol., 51, 9588-9595, 10.1021/acs.est.7b03039, 2017.
- 621 Liu, Y., Fan, Q., Chen, X., Zhao, J., Ling, Z., Hong, Y., Li, W., Chen, X., Wang, M., and Wei, X.:
622 Modeling the impact of chlorine emissions from coal combustion and prescribed waste incineration
623 on tropospheric ozone formation in China, Atmos. Chem. Phys., 18, 2709-2724, 10.5194/acp-18-
624 2709-2018, 2018.
- 625 Lobert, J. M., Keene, W. C., Logan, J. A., and Yevich, R.: Global chlorine emissions from biomass
626 burning: Reactive Chlorine Emissions Inventory, J. Geophys. Res.: Atmos., 104, 8373-8389,
627 <https://doi.org/10.1029/1998JD100077>, 1999.
- 628 Ma, M., Gao, Y., Wang, Y., Zhang, S., Leung, L. R., Liu, C., Wang, S., Zhao, B., Chang, X., Su, H.,
629 Zhang, T., Sheng, L., Yao, X., and Gao, H.: Substantial ozone enhancement over the North China
630 Plain from increased biogenic emissions due to heat waves and land cover in summer 2017, Atmos.
631 Chem. Phys., 19, 12195-12207, 10.5194/acp-19-12195-2019, 2019.
- 632 McDuffie, E. E., Fibiger, D. L., Dubé, W. P., Lopez-Hilfiker, F., Lee, B. H., Jaeglé, L., Guo, H., Weber,
633 R. J., Reeves, J. M., Weinheimer, A. J., Schroder, J. C., Campuzano-Jost, P., Jimenez, J. L., Dibb, J.
634 E., Veres, P., Ebben, C., Sparks, T. L., Wooldridge, P. J., Cohen, R. C., Campos, T., Hall, S. R.,
635 Ullmann, K., Roberts, J. M., Thornton, J. A., and Brown, S. S.: ClNO₂ Yields From Aircraft
636 Measurements During the 2015 WINTER Campaign and Critical Evaluation of the Current
637 Parameterization, J. Geophys. Res.: Atmos., 123, 12,994-913,015,
638 <https://doi.org/10.1029/2018JD029358>, 2018a.
- 639 McDuffie, E. E., Fibiger, D. L., Dubé, W. P., Lopez-Hilfiker, F., Lee, B. H., Thornton, J. A., Shah, V.,
640 Jaeglé, L., Guo, H., Weber, R. J., Michael Reeves, J., Weinheimer, A. J., Schroder, J. C.,
641 Campuzano-Jost, P., Jimenez, J. L., Dibb, J. E., Veres, P., Ebben, C., Sparks, T. L., Wooldridge, P.
642 J., Cohen, R. C., Hornbrook, R. S., Apel, E. C., Campos, T., Hall, S. R., Ullmann, K., and Brown,
643 S. S.: Heterogeneous N₂O₅ Uptake During Winter: Aircraft Measurements During the 2015
644 WINTER Campaign and Critical Evaluation of Current Parameterizations, J. Geophys. Res.: Atmos.,
645 123, 4345-4372, <https://doi.org/10.1002/2018JD028336>, 2018b.
- 646 Mitroo, D., Gill, T. E., Haas, S., Pratt, K. A., and Gaston, C. J.: ClNO₂ Production from N₂O₅ Uptake
647 on Saline Playa Dusts: New Insights into Potential Inland Sources of ClNO₂, Environ. Sci. Technol.,
648 53, 7442-7452, 2019.
- 649 Morgan, W. T., Ouyang, B., Allan, J. D., Aruffo, E., Di Carlo, P., Kennedy, O. J., Lowe, D., Flynn, M. J.,
650 Rosenberg, P. D., Williams, P. I., Jones, R., McFiggans, G. B., and Coe, H.: Influence of aerosol
651 chemical composition on N₂O₅ uptake: airborne regional measurements in northwestern Europe,
652 Atmos. Chem. Phys., 15, 973-990, 10.5194/acp-15-973-2015, 2015.



- 653 Ng, N. L., Herndon, S. C., Trimborn, A., Canagaratna, M. R., Croteau, P. L., Onasch, T. B., Sueper, D.,
654 Worsnop, D. R., Zhang, Q., Sun, Y. L., and Jayne, J. T.: An Aerosol Chemical Speciation Monitor
655 (ACSM) for Routine Monitoring of the Composition and Mass Concentrations of Ambient Aerosol,
656 *Aerosol Sci. Technol.*, 45, 780-794, 10.1080/02786826.2011.560211, 2011.
- 657 Osthoff, H. D., Roberts, J. M., Ravishankara, A. R., Williams, E. J., Lerner, B. M., Sommariva, R., Bates,
658 T. S., Coffman, D., Quinn, P. K., Dibb, J. E., Stark, H., Burkholder, J. B., Talukdar, R. K., Meagher,
659 J., Fehsenfeld, F. C., and Brown, S. S.: High levels of nitryl chloride in the polluted subtropical
660 marine boundary layer, *Nat. Geosci.*, 1, 324-328, 10.1038/ngeo177, 2008.
- 661 Riedel, T. P., Bertram, T. H., Crisp, T. A., Williams, E. J., Lerner, B. M., Vlasenko, A., Li, S. M., Gilman,
662 J., De Gouw, J., and Bon, D. M.: Nitryl Chloride and Molecular Chlorine in the Coastal Marine
663 Boundary Layer, *Environ. Sci. Technol.*, 46, 10463-10470, 2012.
- 664 Sarwar, G., Simon, H., Xing, J., and Mathur, R.: Importance of tropospheric ClNO₂ chemistry across the
665 Northern Hemisphere, *Geophys. Res. Lett.*, 2014.
- 666 Schmidt, J. A., Jacob, D. J., Horowitz, H. M., Hu, L., Sherwen, T., Evans, M. J., Liang, Q., Suleiman, R.
667 M., Oram, D. E., Le Breton, M., Percival, C. J., Wang, S., Dix, B., and Volkamer, R.: Modeling the
668 observed tropospheric BrO background: Importance of multiphase chemistry and implications for
669 ozone, OH, and mercury, *J. Geophys. Res.: Atmos.*, 121, 11,819-811,835,
670 <https://doi.org/10.1002/2015JD024229>, 2016.
- 671 Sherwen, T., Evans, M. J. J., Sommariva, R., Hollis, L. D. J., Ball, S., Monks, P., Reed, C., Carpenter, L.,
672 Lee, J. D., and Forster, G.: Effects of halogens on European air-quality, *Faraday Discuss.*, 200, 2017.
- 673 Sherwen, T., Schmidt, J. A., Evans, M. J., Carpenter, L. J., Großmann, K., Eastham, S. D., Jacob, D. J.,
674 Dix, B., Koenig, T. K., Sinreich, R., Ortega, I., Volkamer, R., Saiz-Lopez, A., Prados-Roman, C.,
675 Mahajan, A. S., and Ordóñez, C.: Global impacts of tropospheric halogens (Cl, Br, I) on oxidants
676 and composition in GEOS-Chem, *Atmos. Chem. Phys.*, 16, 12239-12271, 10.5194/acp-16-12239-
677 2016, 2016.
- 678 Simmonds, P. G., Manning, A. J., Cunnold, D. M., McCulloch, A., O'Doherty, S., Derwent, R. G.,
679 Krummel, P. B., Fraser, P. J., Dunse, B., Porter, L. W., Wang, R. H. J., Grealley, B. R., Miller, B. R.,
680 Salameh, P., Weiss, R. F., and Prinn, R. G.: Global trends, seasonal cycles, and European emissions
681 of dichloromethane, trichloroethene, and tetrachloroethene from the AGAGE observations at Mace
682 Head, Ireland, and Cape Grim, Tasmania, *J. Geophys. Res.: Atmos.*, 111,
683 <https://doi.org/10.1029/2006JD007082>, 2006.
- 684 Staudt, S., Gord, J. R., Karimova, N. V., McDuffie, E. E., Brown, S. S., Gerber, R. B., Nathanson, G. M.,
685 and Bertram, T. H.: Sulfate and Carboxylate Suppress the Formation of ClNO₂ at Atmospheric
686 Interfaces, *ACS Earth and Space Chemistry*, 3, 1987-1997, 10.1021/acsearthspacechem.9b00177,
687 2019.



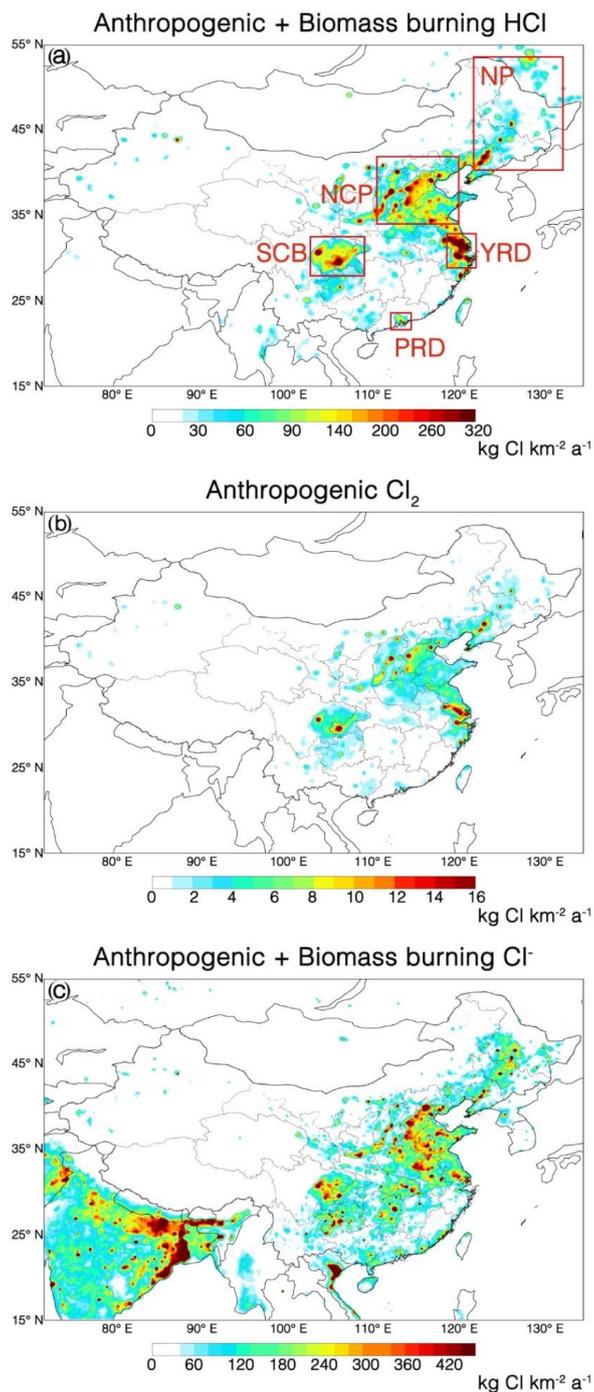
- 688 Tian, M., Liu, Y., Yang, F., Zhang, L., Peng, C., Chen, Y., Shi, G., Wang, H., Luo, B., Jiang, C., Li, B.,
689 Takeda, N., and Koizumi, K.: Increasing importance of nitrate formation for heavy aerosol pollution
690 in two megacities in Sichuan Basin, southwest China, *Environ. Pollut.*, 250, 898-905,
691 <https://doi.org/10.1016/j.envpol.2019.04.098>, 2019.
- 692 Van, d. W., G. R., Randerson, J. T., Giglio, L., Collatz, G. J., Mu, M., Kasibhatla, P. S., Morton, D. C.,
693 Defries, R. S., Jin, Y., and Van Leeuwen, T. T.: Global fire emissions and the contribution of
694 deforestation, savanna, forest, agricultural, and peat fires (1997–2009), *Atmos. Chem. Phys.*, 10,
695 16153-16230, 10.5194/acp-10-11707-2010, 2010.
- 696 Wang, T., Tham, Y. J., Xue, L., Li, Q., Zha, Q., Wang, Z., Poon, S. C. N., Dubé, W. P., Blake, D. R.,
697 Louie, P. K. K., Luk, C. W. Y., Tsui, W., and Brown, S. S.: Observations of nitryl chloride and
698 modeling its source and effect on ozone in the planetary boundary layer of southern China, *J.*
699 *Geophys. Res.: Atmos.*, 121, 2476-2489, <https://doi.org/10.1002/2015JD024556>, 2016.
- 700 Wang, X., Jacob, D. J., Fu, X., Wang, T., and Liao, H.: Effects of Anthropogenic Chlorine on PM 2.5 and
701 Ozone Air Quality in China, *Environ. Sci. Technol.*, 54, 9908-9916, 2020.
- 702 Wang, X., Jacob, D. J., Eastham, S. D., Sulprizio, M. P., Zhu, L., Chen, Q., Alexander, B., Sherwen, T.,
703 Evans, M. J., Lee, B. H., Haskins, J. D., Lopez-Hilfiker, F. D., Thornton, J. A., Huey, G. L., and
704 Liao, H.: The role of chlorine in global tropospheric chemistry, *Atmos. Chem. Phys.*, 19, 3981-4003,
705 10.5194/acp-19-3981-2019, 2019.
- 706 Wang, Y., Shen, L., Wu, S., Mickley, L., He, J., and Hao, J.: Sensitivity of surface ozone over China to
707 2000–2050 global changes of climate and emissions, *Atmos. Environ.*, 75, 374-382, 2013.
- 708 Xia, M., Wang, W., Wang, Z., Gao, J., Li, H., Liang, Y., Yu, C., Zhang, Y., Wang, P., Zhang, Y., Bi, F.,
709 Cheng, X., and Wang, T.: Heterogeneous Uptake of N₂O₅ in Sand Dust and Urban Aerosols
710 Observed during the Dry Season in Beijing, *Atmosphere*, 10, 204, 2019.
- 711 Xu, Z., Liu, M., Zhang, M., Song, Y., Wang, S., Zhang, L., Xu, T., Wang, T., Yan, C., Zhou, T., Sun, Y.,
712 Pan, Y., Hu, M., Zheng, M., and Zhu, T.: High efficiency of livestock ammonia emission controls
713 in alleviating particulate nitrate during a severe winter haze episode in northern China, *Atmos. Chem.*
714 *Phys.*, 19, 5605-5613, 10.5194/acp-19-5605-2019, 2019.
- 715 Yang, X., Wang, T., Xia, M., Gao, X., Li, Q., Zhang, N., Gao, Y., Lee, S., Wang, X., Xue, L., Yang, L.,
716 and Wang, W.: Abundance and origin of fine particulate chloride in continental China, *Sci. Total*
717 *Environ.*, 624, 1041-1051, <https://doi.org/10.1016/j.scitotenv.2017.12.205>, 2018.
- 718 Ye, C., Yuan, B., Lin, Y., Wang, Z., Hu, W., Li, T., Chen, W., Wu, C., Wang, C., Huang, S., Qi, J., Wang,
719 B., Wang, C., Song, W., Wang, X., Zheng, E., Krechmer, J. E., Ye, P., Zhang, Z., Wang, X., Worsnop,
720 D. R., and Shao, M.: Chemical characterization of oxygenated organic compounds in the gas phase
721 and particle phase using iodide CIMS with FIGAERO in urban air, *Atmos. Chem. Phys.*, 21, 8455-
722 8478, 10.5194/acp-21-8455-2021, 2021.



723 Yu, C., Wang, Z., Xia, M., Fu, X., Wang, W., Tham, Y. J., Chen, T., Zheng, P., Li, H., Shan, Y., Wang, X.,
724 Xue, L., Zhou, Y., Yue, D., Ou, Y., Gao, J., Lu, K., Brown, S. S., Zhang, Y., and Wang, T.:
725 Heterogeneous N₂O₅ reactions on atmospheric aerosols at four Chinese sites: improving model
726 representation of uptake parameters, *Atmos. Chem. Phys.*, 20, 4367-4378, 10.5194/acp-20-4367-
727 2020, 2020.

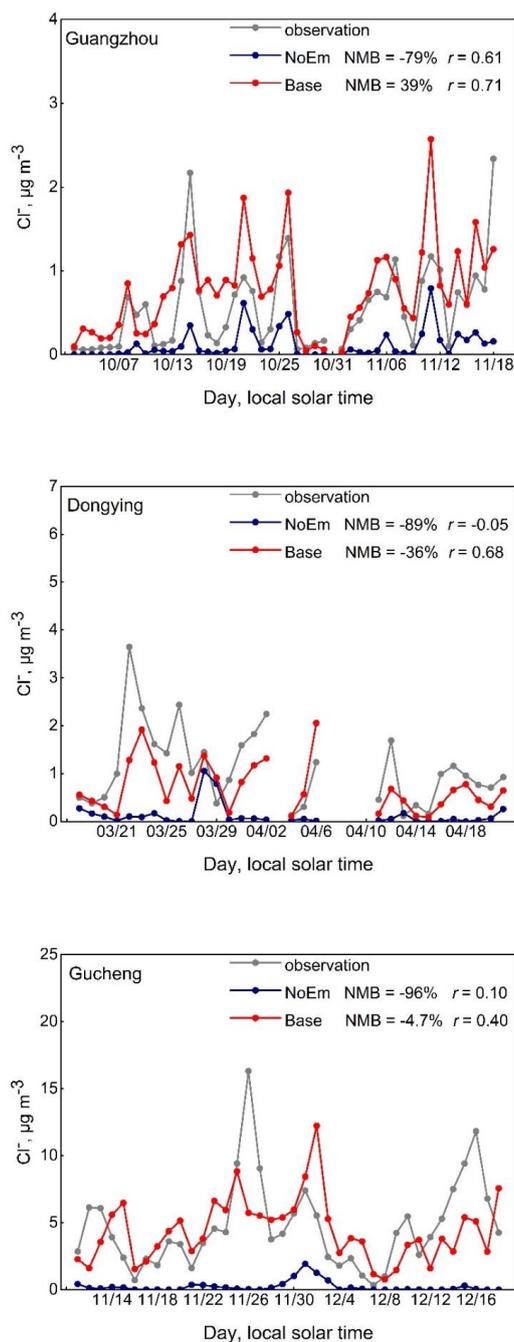
728 Zhang, T., de Jong, M. C., Wooster, M. J., Xu, W., and Wang, L.: Trends in eastern China agricultural
729 fire emissions derived from a combination of geostationary (Himawari) and polar (VIIRS) orbiter
730 fire radiative power products, *Atmos. Chem. Phys.*, 20, 10687-10705, 10.5194/acp-20-10687-2020,
731 2020.

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Figure 1. Annual emissions of (a) HCl, (b) Cl₂, (c) non-sea salt Cl⁻.



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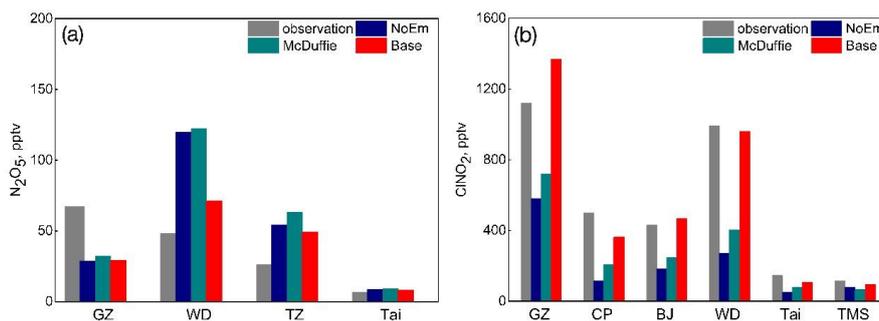
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Figure 2. Time series of simulated and observed particulate Cl^- concentrations at the Guangzhou, Dongying and Gucheng sites.



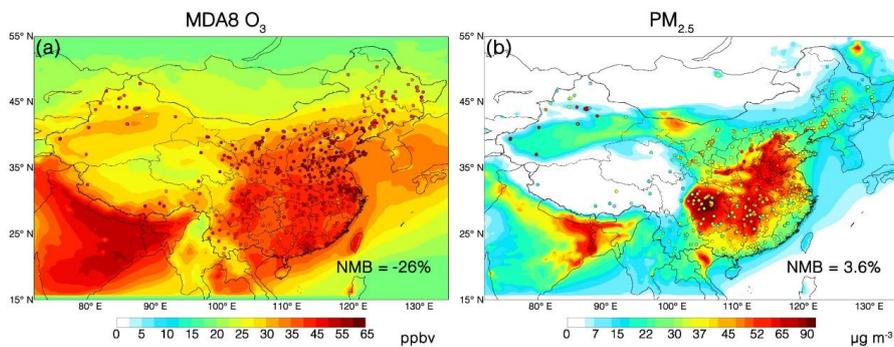
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741 **Figure 3. Comparison of observed and simulated (a) averaged N_2O_5 concentrations and (b) mean nighttime**
 742 **maximum mixing ratio of $ClNO_2$ concentrations at different sites. GZ: Guangzhou; WD: Wangdu; TZ:**
 743 **Taizhou; Tai: Mount Tai; CP: Changping; BJ: Beijing; TMS: Mount TaiMoShan**

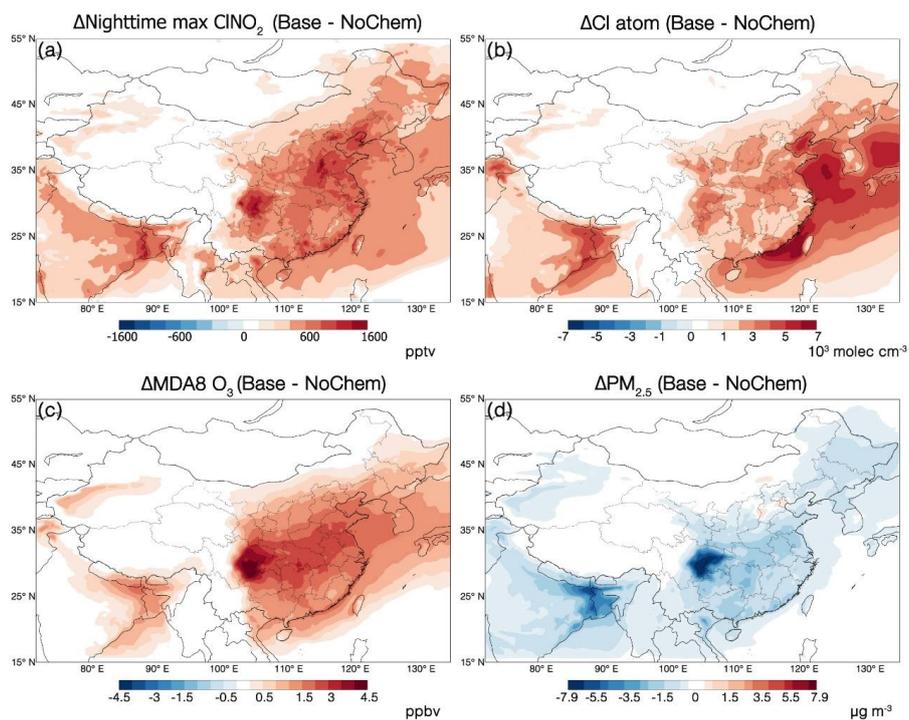
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746 **Figure 4. Annual mean surface concentrations of (a) MDA8 O_3 and (b) $PM_{2.5}$ over China in 2018. GEOS-**
 747 **Chem model values from the Base case are shown as contours. Observations from China National**
 748 **Environmental Monitoring Center (CNEMC) are shown as circles.**

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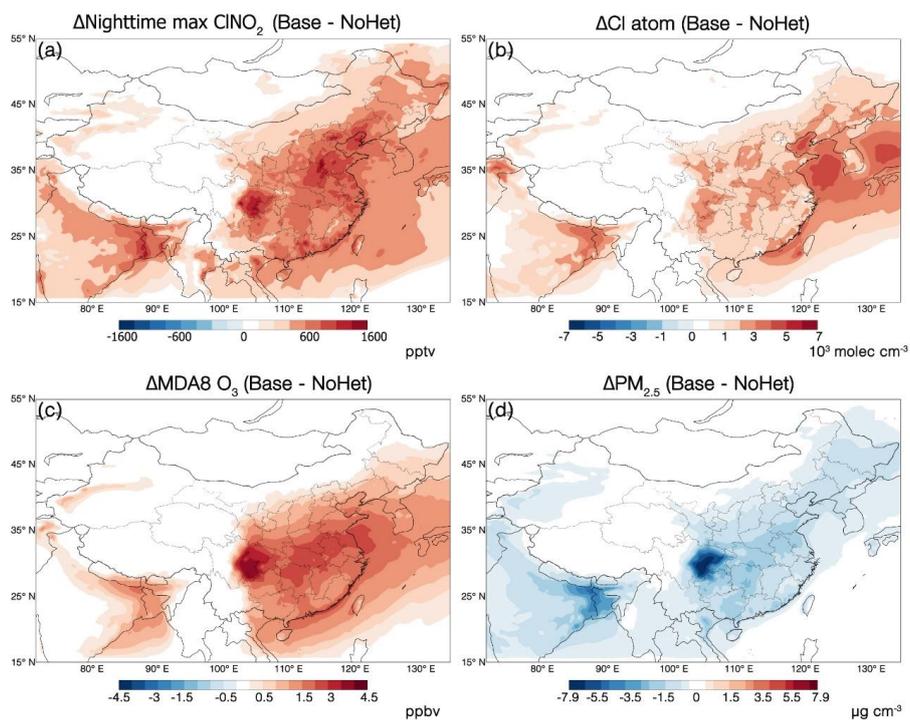
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Figure 5. Effects of chlorine chemistry on annual mean surface concentrations of (a) nighttime max ClNO_2 , (b) Cl atom, (c) MDA8 O_3 and (d) $\text{PM}_{2.5}$ in China, estimated as the differences between the Base and NoChem cases.



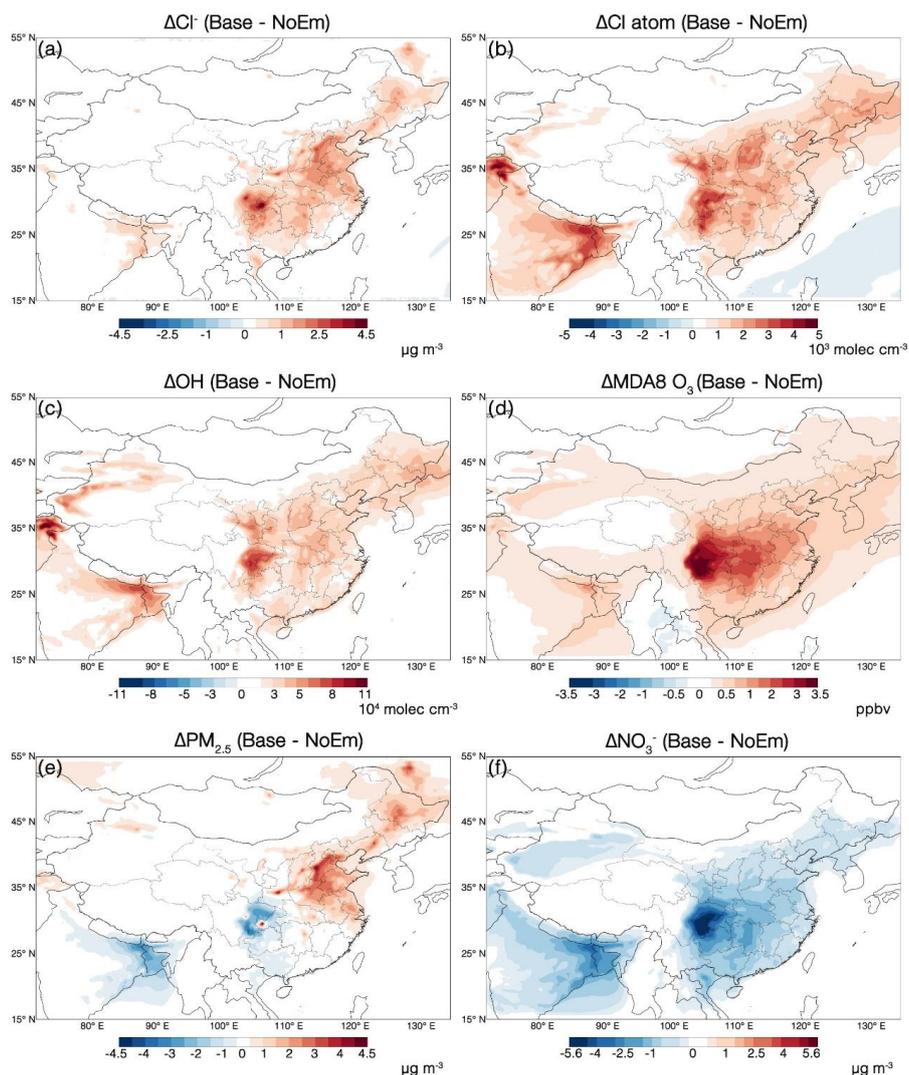
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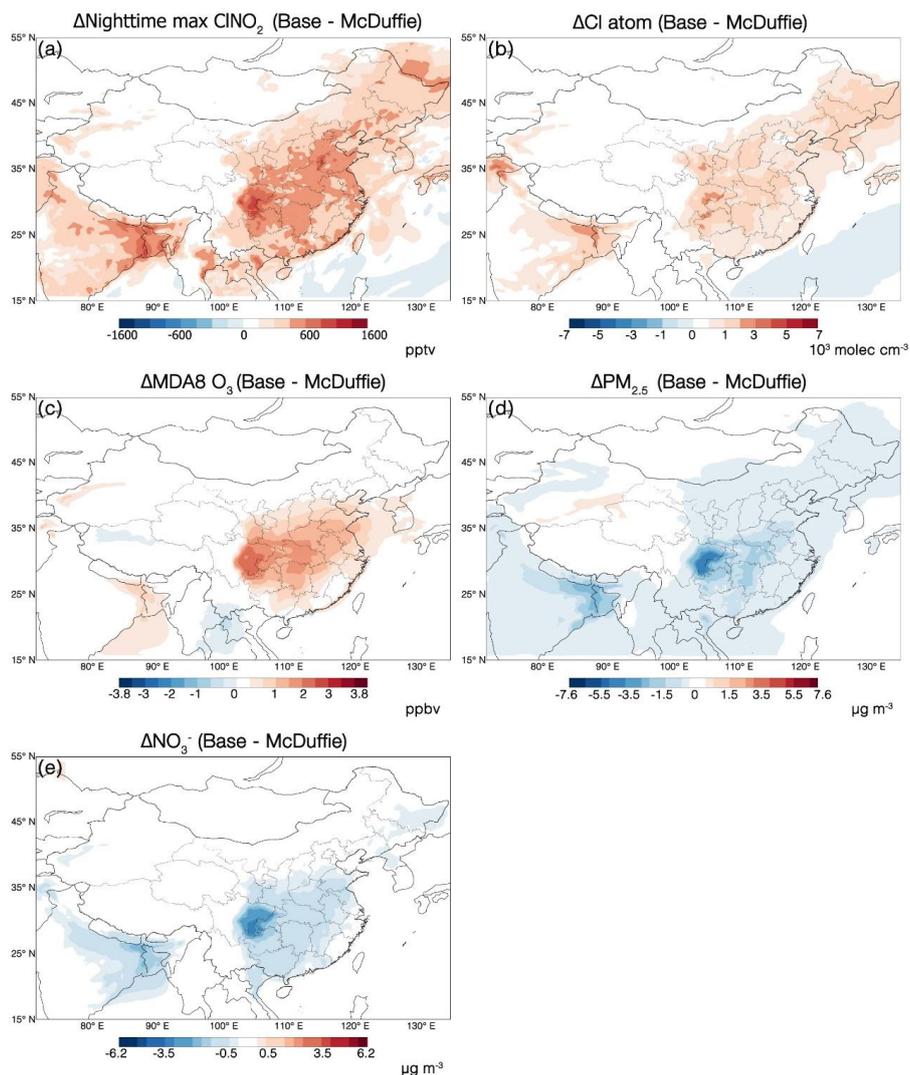
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Figure 6. Effects of N_2O_5 - ClNO_2 chemistry on annual mean surface concentrations of (a) nighttime max ClNO_2 , (b) Cl atom, (c) MDA8 O_3 and (d) $\text{PM}_{2.5}$ in China, estimated as the differences between the Base and NoHet cases.



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Figure 7. Effects of anthropogenic and biomass burning chlorine emissions on annual mean surface concentrations of (a) Cl⁻, (b) Cl atom, (c) OH, (d) MDA8 O₃, (e) PM_{2.5} and (f) NO₃⁻ in China, estimated as the differences between the Base and NoEm cases.



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765 **Figure 8.** Effects of different parameterizations on annual mean surface concentrations of (a) nighttime max
766 CINO_2 , (b) Cl atom, (c) MDA8 O_3 , (d) $\text{PM}_{2.5}$, and (e) NO_3^- in China, estimated as the differences between the
767 Base and McDuffie cases.

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Table 1. Chlorine emissions in China in the model.

| Sources | By default (Gg Cl a ⁻¹) | Updated in this study (Gg Cl a ⁻¹) |
|--|--|---|
| Sea salt Cl ⁻ | 6.5×10 ⁴ | 6.5×10 ⁴ |
| Anthropogenic HCl | 0 | 218 |
| Biomass burning HCl | 0 | 30 |
| Anthropogenic Cl ₂ | 0 | 8.9 |
| Anthropogenic Cl ⁻ | 0 | 379 |
| Biomass burning Cl ⁻ | 0 | 120 |
| CH ₃ Cl ^a | 3.8 | 3.8 |
| CH ₂ Cl ₂ ^a | 2.4 | 2.4 |
| CHCl ₃ ^a | 0.70 | 0.70 |

777 ^a: Sources are shown in terms of the chemical release (e.g. +Cl, +OH, +hv)

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Table 2. Model setup of all simulation cases

| Cases | N ₂ O ₅ uptake (γ _{N2O5}) | ClNO ₂ production (φ _{ClNO2}) | Other tropospheric chlorine chemistry | Anthropogenic and biomass burning inorganic chlorine emissions |
|----------|--|--|--|--|
| Base | Yu et al. (2020) | Yu et al. (2020) | Full | Yes |
| McDuffie | McDuffie et al. (2018a, 2018b) | McDuffie et al. (2018a, 2018b) | Full | Yes |
| NoEm | Yu et al. (2020) | Yu et al. (2020) | Full | None |
| NoHet | Yu et al. (2020) but with [Cl ⁻] = 0 | None | Full | Yes |
| NoChem | Yu et al. (2020) but with [Cl ⁻] = 0 | None | None | Yes |
| NoEmHet | Yu et al. (2020) but with [Cl ⁻] = 0 | None | Full | None |
| NoAll | Yu et al. (2020) but with [Cl ⁻] = 0 | None | None | None |

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Table 3. Normalized mean bias (NMB) and correlation coefficients (*r*) between observed and simulated

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MDA8 O₃ and PM_{2.5} concentrations during 2018 in China

| Species | Time | Base | | McDuffie | | NoEm | |
|---------------------|------------------|-------|----------|----------|----------|-------|----------|
| | | NMB | <i>r</i> | NMB | <i>r</i> | NMB | <i>r</i> |
| MDA8 O ₃ | Annual | -26% | 0.83 | -27% | 0.83 | -28% | 0.82 |
| | MAM ^a | -35% | 0.87 | -36% | 0.87 | -36% | 0.87 |
| | JJA ^b | -5.5% | 0.50 | -5.2% | 0.48 | -5.9% | 0.48 |
| | SON ^c | -24% | 0.79 | -26% | 0.78 | -28% | 0.76 |
| | DJF ^d | -49% | 0.81 | -53% | 0.8 | -54% | 0.80 |
| PM _{2.5} | Annual | 3.6% | 0.81 | 5.6% | 0.81 | 2.3% | 0.80 |
| | MAM | -6.3% | 0.52 | -4.9% | 0.53 | -6.2% | 0.52 |
| | JJA | 3.9% | 0.70 | 4.6% | 0.70 | 5.0% | 0.70 |
| | SON | 28% | 0.79 | 32% | 0.80 | 26% | 0.79 |
| | DJF | -4.3% | 0.82 | -2.6% | 0.82 | -7.9% | 0.82 |

783 ^a: March, April, and May (Spring)

784 ^b: June, July, and August (Summer)

785 ^c: September, October, and November (Autumn)

786 ^d: December, January, and February (Winter)

787