1	Insights into the significant increase of ozone during COVID-
2	19 in a typical urban city of China
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18	Abstract
19	The outbreak of COVID-19 promoted strict restrictions to human activities in China, which led to
20	dramatic decrease in most air pollutant concentrations (e.g., PM <sub>2.5</sub> , PM <sub>10</sub> , NO <sub>x</sub> , SO <sub>2</sub> , and CO).
21	However, obvious increase of ozone $(O_3)$ concentrations was found during the lockdown period in
22	most urban areas of China. In this study, we conducted a field measurement targeting ozone and its
23	key precursors by utilizing a novel proton transfer reaction time-of-flight mass spectrometer (PTR-
24	TOF-MS) in Changzhou, which is representative for the Yangtze River Delta (YRD) city cluster of
25	China. We further applied the integrated methodology including machine learning, observation-based
26	model (OBM), and sensitivity analysis to get insights into the reasons causing the obvious increase of

27 ozone. Major findings include: (1) By deweathered calculation, we found changes in precursor 28 emissions contributed 1.46 ppby to the increase in the observed O<sub>3</sub> during the Full-lockdown period in 29 2020, while meteorology constrained 3.0 ppbv of  $O_3$  in the Full-lockdown period of 2019. (2) By 30 using an OBM model, we found that although significant reduction of O<sub>3</sub> precursors was observed 31 during Full-lockdown period, the photochemical formation of O<sub>3</sub> was stronger than that during the 32 Pre-lockdown period. (3) The  $NO_x/VOCs$  ratio dropped dramatically from 1.84 during Pre-lockdown 33 to 0.79 in Full-lockdown period, which switched O<sub>3</sub> formation from VOCs-limited regime to the 34 boundary of  $NO_{x}$ - and VOC-limited regime. Additionally, box model results suggested that the 35 decrease in  $NO_x/VOCs$  ratio during Full-lockdown period could increase the MeanO<sub>3</sub> by 2.4 ppbv. 36 Results of this study give insights into the relationship between  $O_3$  and its precursors in urban area, and demonstrate reasons causing the obvious increase of  $O_3$  in most urban areas of China during the 37 38 COVID-19 lock-down period. This study also underlines the necessity of controlling anthropogenic 39 OVOCs, alkenes, and aromatics in the sustained campaign of reducing  $O_3$  pollution in China.

40 Keywords: Ozone; VOCs; PTR-TOF-MS; COVID-19

## 41 **1. Introduction**

42 At the end of 2019, a tragic coronavirus (COVID-19) occurred, which has caused over 271 million global infection and over 4.51 million deaths as of this writing (12<sup>th</sup> Feb 2022). To protect 43 44 people's health, China adopted strict measures to control the spread of this pandemic. Thirty provinces, 45 autonomous regions and municipalities have launched Full-lockdown response (also known as Level I response, roughly from 24<sup>th</sup> Jan to 25<sup>th</sup> Feb 2020) as early as 24<sup>th</sup> Jan 2020 (Shen et al., 2021; Li et al., 46 2020; Huang et al., 2020). With the effective control of COVID-19 in China, the emergency response 47 48 level in most provinces (except Hubei province, the hardest-hit region) gradually downgraded to Partial-lockdown (Level II and Level III response, roughly after 25th Feb 2020) (Li et al., 2020), and 49 50 work resumption started. During Full-lockdown period, all the social events that may cause crowds 51 (excluding transportation and industries that maintained the basic operation of society) were severely 52 restricted. Affected by the pandemic, many factories were shut down, and the on-road traffic volume

53 and construction activities have been reduced significantly (Zheng et al., 2020). During Full-lockdown 54 period, dramatic decrease of air pollutants (e.g., PM<sub>2.5</sub>, NO<sub>2</sub>, BC) were found in China, especially in 55 urban areas (Fan et al., 2021; Gao et al., 2021; Li et al., 2020; Xu et al., 2020). Surprisingly, marginal 56 increases of  $O_3$  were observed during the lockdown period in YRD region, and this seems to be 57 contradictory to the decrease of most air pollutants (Li et al., 2020). However, as suggested by 58 previous studies, the formation of O<sub>3</sub> is significantly influenced by NO<sub>4</sub>/VOCs ratio and 59 meteorological conditions (temperature, relative humidity and actinic flux) (Zhang et al., 2020a; 60 Zhang et al., 2020b). Therefore, it is essential to investigate the changes of meteorological and 61 emissions conditions to figure out reasons causing the increase of  $O_3$  during this pandemic.

62 Previous studies on the O<sub>3</sub> pollution in the YRD region have often focused on the more populated 63 metropolitan areas, such as Shanghai and Nanjing, which are considerably far away from the industrial zones that are essentially responsible for the sources of O3 precursors (Li et al., 2019; Zhang 64 65 et al., 2020b). Changzhou, located in the center of the Yangtze River Delta (YRD) region, is a typical city with fast urbanization, heavy industrial structure, huge energy consumption, increasing vehicle 66 67 stocks and frequent air pollution. Therefore, it provides a more representative environment to fully 68 elucidate the mechanism underlying the  $O_3$  pollution in the YRD region (Shi et al., 2020). In a 69 companion paper (Jensen et al., 2021), we also demonstrated that Changzhou is representative for the 70 region by analyzing both surface observations and satellite data. According to previous studies, the 71 anthropogenic VOCs emission in Changzhou was around 9~12.6×10<sup>4</sup> tons/year, among which 72 industries was the dominant source, accounting for 27~47% of the total VOC emissions (Cheng et al., 73 2016; Fu et al., 2013). It is notable that industrial sources together contributed over 80% of 74 anthropogenic VOC emissions (Sun et al., 2019). Apart from industrial sources, vehicle exhaust 75 accounted for 9%~14% of total VOC emissions (Sun et al., 2019). However, rare observation 76 regarding VOCs characteristics during COVID-19 in Changzhou has been conducted.

Highly time-resolved measurements of VOCs are generally much sparse and could not be easily expanded during the lockdowns. This limits our understanding of how VOCs changed and how the formation of ozone was affected. Here, we used a novel proton transfer reaction time-of-flight mass spectrometer (PTR-TOF-MS, Tofwerk, Model Vocus Elf, CHE) to conduct online observation of VOCs in Changzhou. The characteristics of VOCs and the variations of general air pollutants in each emergency response period were analyzed. Additionally, ozone formation during each period was investigated by an OBM model. Although terrifying impact has been caused by the COVID-19, it provided a rare experiment to analyze the variations of VOCs and NO<sub>x</sub> due to changes of anthropogenic activities in a typical city of China. Furthermore, results of this study offer theoretical support for formulating refined ozone management policy in China.

# 87 **2. Methodology**

#### 88 2.1 Field measurement

89 The field campaign was conducted from 8<sup>th</sup> Jan to 31<sup>st</sup> Mar 2020 at a sampling site located on the 90 rooftop of a building at Changzhou Environmental Monitoring Center (CEMC, 31.76° N, 119.96° E), 91 which was approximately 15 m above ground level. As a typical urban monitoring station, this site is 92 in the center of Changzhou city, surrounded by residential and commercial area, which is also adjacent 93 to the main transportation junction in Changzhou (Figure 1). According to local epidemic prevention policies, we roughly classified the measurement periods into three stages: Pre-lockdown (8th January 94 to 23rd January 2020), Full-lockdown (25th January to 24th February 2020), Partial-lockdown (25th 95 96 February to 28<sup>th</sup> March 2020) as defined in a study of the Yangtze River Delta (Li Li et al., 2020).

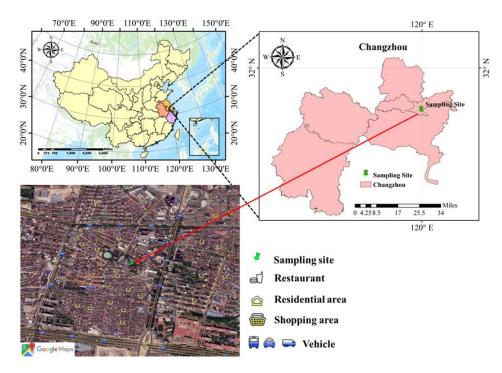




Figure 1. Location of the sampling site in Changzhou.

From Jan 8th to Mar 27th, 2020, the concentrations of traditional air pollutants (PM2.5, PM10, NOx, 99 100 SO<sub>2</sub>, CO, O<sub>3</sub>) as well as meteorological parameters were monitored by a series of analyzers (Table 101 1Error! Reference source not found.). In particular, 87 VOCs species were quantified, 59 of which 102 were identified, by a PTR-TOF-MS with time resolution of 1 min. Detailed measurement techniques 103 and quality assurance and control has been documented in detail in our companion paper (Jensen et al., 104 2021). Here, we just briefly introduce the measurement. The air samples were directly drawn into a 3 105 m-long tube connected to the instrument. A priming pump, with flow rate of 4 L/min, was used to 106 reduce the retention time of the gas sample in the tube. To avoid blocking of inlet tube caused by 107 particles, a particulate filter was assembled at the front of the inlet tube. The pressure of the ion source 108 was set as 2 mbar and the temperature of the reaction chamber was set to 90 °C during the observation. 109 VOCs are ionized by reactions with  $H_3O^+$  ions from a discharge, and the product ions are detected by 110 a time-of-light mass analyzer (m/ $\Delta$ m FMHW of 950 at m/O 107). The PTR-TOF-MS can detect most 111 unsaturated hydrocarbons and VOCs with functional groups but cannot detect species with proton 112 affinities lower than that of water, namely alkanes and small alkenes. Eighteen standard gases 113 (including acetonitrile, acetaldehyde, acrolein, acetone, isoprene, butanone, 2-butanone, benzene, 2pentanone, ethyl acetate, toluene, methyl isobutyl ketone, styrene, xylene, trimethylbenzene, naphthalene, a-pinene, and 1,3-dichlorobenzene) with concentrations of 1 ppmv were used for the calibration of the PTR-TOF-MS. In addition, a built-in calibration system was used to control the zero and standard gases.

Species/Parameter	Experimental Technique				
T, RH, WS, WD and P	2000WX, Airmax, USA				
O3	400E, API, USA				
NO <sub>x</sub> (NO and NO <sub>2</sub> )	T200, API, USA				
SO <sub>2</sub>	T100, API, USA				
СО	T300, API, USA				
PM <sub>2.5</sub>	5030, Thermo Fisher, USA				
PM10	5030, Thermo Fisher, USA				
VOCs	Vocus Elf, Tofwerk, CHE				

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## 119

## 120 2.2 Observation-based model

121 An OBM model coupled with MCM v3.3.1 was utilized to investigate the atmospheric oxidation 122 capability and the radical chemistry. Detailed information about the chemistry mechanism is available 123 on MCM website (http://mcm.leeds.ac.uk/MCM/, last access 8 Jul 2021). More than 5800 chemical 124 species and 17000 reactions are included in this mechanism. The photolysis frequencies (J values) 125 were calculated as a function of solar zenith angle, altitude using lookup tables, calculated using the 126 Tropospheric Ultraviolet and Visible (TUV) model (Wolfe et al., 2016). Dilution mixing within the 127 boundary layer is considered. However, as a 0-D model, vertical or horizonal transport of airmasses are not involved. The observed meteorological parameters (T, RH, P), trace gases (NO, NO<sub>2</sub>, CO, SO<sub>2</sub>, 128 129 and VOCs) were used to constrain the model. Before each simulation, the model was run 3 days as 130 spin-up to reach a stable state. According to the definition of atmospheric oxidation capability (AOC), 131 AOC is quantified by Eq (1) (Gever et al., 2001).

$$AOC = \sum_{i=1}^{N} k_{Y_i - X}[Y_i][X]$$
 (1)

where  $Y_i$  are the primary pollutants (e.g., VOCs, CH<sub>4</sub>, and CO); *X* are atmospheric oxidants (OH, O<sub>3</sub>, NO<sub>3</sub>);  $k_{Yi}$  are the bimolecular rate constants for the reactions of  $Y_i$  and *X*. A high value of AOC indicates fast scavenge of primary air pollutants. Additionally, OH reactivity ( $k_{OH}$ ), defined as the reaction rate coefficients multiplied by the concentrations of the reactants with OH, is also widely used as an indicator of AOC. The value of  $k_{OH}$  depends on both the abundances and compositions of primary pollutants and can be calculated by Eq (2).

$$k_{OH} = \sum_{i} k_{OH+X_i} \times [X_i] \tag{2}$$

where  $k_{OH+Xi}$  are the reaction rate coefficients of reaction OH+ $X_i$ ; Xi are the concentrations of pollutants (VOC, NO<sub>2</sub>, CO, OVOC etc.) (Zhu et al., 2020).

140 **2.3 Trend Analysis** 

141 Mann-Kendall (MK) trend test is a widely used non-parametric test method (Pathakoti et al., 142 2021; Zhang et al., 2013). It is applicable to all distributions (that is, the data does not need to meet the 143 assumption of normal distribution), but the data should have no serial correlation. If the data has serial 144 correlation, it will have an impact on the significance level (p value). In this study, the MK trend 145 analysis was performed for individual VOC concentrations during Pre-lockdown and Full-lockdown 146 period. By using the "feast" R package, no obvious serial correlation of individual VOC is found. 147 Therefore, the observed VOC data is suitable for MK test. Detailed description and the calculation 148 formula of MK trend test could be found in the study of Pathakoti et al. (2021) and Alhathloul et al. 149 (2021). A positive z value from the MK test indicates increasing trend of the target compound. On the 150 contrary, a negative z value suggests the target compound was decreasing.

Sen's slope, a non-parametric test proposed by Sen (1968), is used in this study to assess the rate of change in individual VOC concentrations. The Sen's slope is selected since it is insensitive to outliers, and does not require a normal distribution of residuals. Sen's slope (Q) is mathematically represented by the following equations.

$$Q = median(SS_{ij}) \tag{3}$$

$$SS_{ij} = \frac{x_j - x_i}{j - i}, 1 \le i \le j \le n$$

$$\tag{4}$$

where  $x_j$  and  $x_i$  are concentrations of VOC specie x at time *j* and *i* ( $1 \le i \le j \le n$ ), respectively. SS<sub>ij</sub> is the linear slope between time *i* and *j*, and Q is the median of SS<sub>ij</sub>. Positive and negative Q values indicate increasing or decreasing trend of VOC specie x, respectively.

#### 158 **2.4 Deweathered model**

159 The observed concentrations of  $O_3$  could be influenced by meteorological conditions, emissions 160 and/or chemistry. The emissions and chemistry are being treated together and separated from 161 meteorology by the deweathered approach based on the random forest (RF). Hourly data of Unix date (number of seconds since 1970-01-01), Julian day, weekday, hour of day, wind speed (WS), wind 162 163 direction (WD), temperature (T), relative humidity (RH), and pressure (P), which are available during 164 the whole observation, were used for the deweathered calculation of  $O_3$ . The missing data was 165 replaced by linear interpolation. Training of the models was conducted on 80% of the input data and 166 the other 20% was withheld from training. To avoid the disadvantage of overfitting during the training of RF, a process called bagging (or bootstrap aggregation) was adopted. Bagging results in new, 167 168 sampled set called out-of-bag (OOB) data. A decision tree is then grown on the OOB data. Therefore, 169 all the decision trees are grown on different observations and avoid the overfitting (Grange and David 170 (2019)). To determine the value of number of trees (ntree), number of samples (nsample), and the 171 minimal node size, a series of random forests were performed under different choices of ntree, 172 nsample, and minimal node size. The results suggest that the highest coefficient of determination ( $\mathbb{R}^2$ , 173 0.84) was obtained when ntree, nsample and minimal node size was set as 300, 300, and 5, 174 respectively (Table S1 and S2). More details of this model could be found in the study of Grange and 175 David (2019). The uncertainty of the deweather model is obtained by growing 50 random forest 176 models with the hyperparameters described above, which is the same method as Grange and Carslaw 177 (2019). The mean and standard error of the predicted  $O_3$  concentrations is shown in Figure S1, and 178 results of the model are stable during the 50 runs. The differences in observed O<sub>3</sub> concentrations 179  $(O_{3,Obs})$  and deweathered  $O_3$  concentrations  $(O_{3,Normal})$  were regarded as the concentrations contributed by meteorology  $(O_{3,Met})$ , which is consistent with the definition in Li et al. (2021). Correspondingly, the differences in  $O_{3,Normal}$  concentrations in different periods represent the influence of emissions, since the  $O_{3,Normal}$  has already removed the influence of meteorological conditions.

# 183 **3. Results and discussion**

# 184 **3.1 Overview of the field campaign**

185 Figure 2 shows the meteorological conditions during the observation. During the whole experiment, the prevailing WD was southeast. The average T and RH was  $9.9 \pm 5.1^{\circ}$ C and  $58.9 \pm$ 186 187 17.1%, respectively. Compared to Pre-lockdown period, the concentrations of PM<sub>2.5</sub>, PM<sub>10</sub>, SO<sub>2</sub>, NO, 188 NO<sub>2</sub>, TVOC and CO during Full-lockdown period decreased by 48%, 42%, 11%, 65%, 58%, 33% and 189 39%, respectively. It should be noted that the decreasing ratio of VOC/NO<sub>x</sub> is around 1.75, suggesting 190 that the lockdown policy has stronger influence on NO<sub>x</sub> emissions than VOC emissions. The  $O_3$ 191 concentrations during the same period in 2020 and 2019 are summarized in Table 2. Considering the 192 influence of Chinese New Year, the corresponding period in 2019 was decided according to lunar 193 calendar. It should be noted that, compared to Full-lockdown period in 2019, the mean  $O_3$ 194 concentration in 2020 is obviously higher (5.5 ppbv, Figure 2). Meanwhile, the average  $O_3$ 195 concentrations in Full-lockdown period in 2020 was 67% higher than that during Pre-lockdown period 196 in 2020. To roughly analyze the cause of the obvious increase of  $O_3$  during Full-lockdown period in 197 2020, we summarized the temperature (T) and relative humidity (RH) in Table 2. The T and RH in 198 Full-lockdown period in 2020 was ~1.6 °C higher and 6.1% lower than that in the same period in 2019, 199 while the P and WS were comparable during the same period in 2020 and 2019 (Table 2). The 200 relatively higher T was in favor of O<sub>3</sub> formation during the Full-lockdown period in 2020. As for RH, 201 the influence on  $O_3$  is nonlinear (Zhang et al., 2020), and based on our sensitivity test, lower RH could 202 lead to decrease or increase of  $O_3$  concentration (Figure S2). Overall, changes in  $O_3$  concentrations 203 could be a result of the joint effect of meteorological conditions and emissions/chemistry, the 204 following sections would discuss these influences respectively.

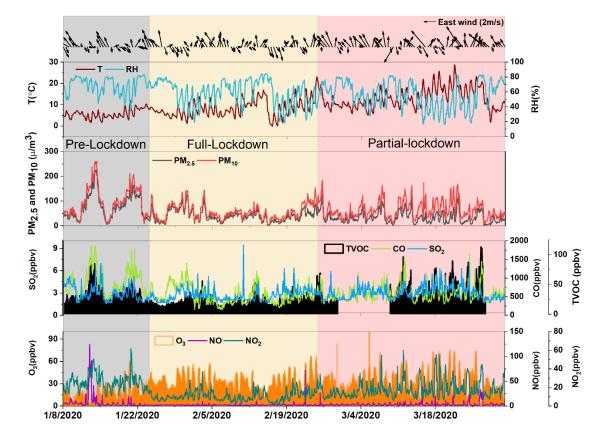


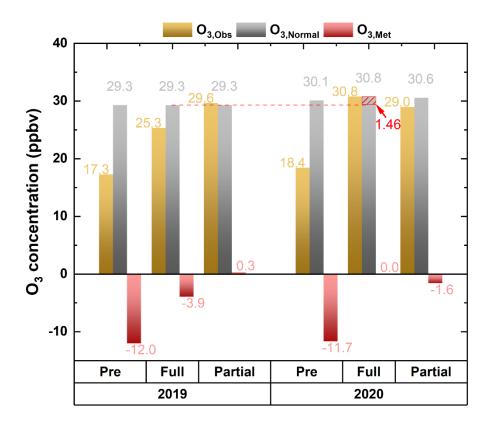
Figure 2 Time series of meteorological parameters and air pollutants during the whole observation.
 Table 2 Comparison of average meteorological conditions during Pre-lockdown, Full-lockdown, and Partial lockdown in 2020 and the same period in 2019.

Periods	Date	P (hPa)	RH (%)	Т (°С)	Precipitation (mm)	WS (m/s)
Pre-lockdown	(2020.1.8-1.24)	1025.4	84.9	4.8	0.13	1.8
Same period in 2019	(2019.1.19-2.4)	1025.6	72.7	5.2	0.05	1.9
Full-lockdown	(2020.1.25-2.24)	1025.6	73.0	7.3	0.09	2.1
Same period in 2019	(2019.2.5-3.7)	1024.1	79.1	5.7	0.15	2.1
Partial-lockdown	(2020.2.25-3.31)	1018.9	69.5	12.1	0.11	2.4
Same period in 2019	(2019.3.8-4.12)	1017.6	64.0	13.8	0.02	2.0

# **3.2 Mechanism affecting the obvious O<sub>3</sub> increase**

#### **3.2.1 Meteorological perspective**

212 Deweathered  $O_3$  concentrations were calculated based on the model described in Section 2.4. The 213 difference between  $O_{3,Obs}$  and  $O_{3,Normal}$  can be regarded as the meteorological influence ( $O_{3,Met}$ ). In 214 addition, the difference between  $O_{3,Normal}$  concentrations in different years could be considered as the 215 influence of emissions ( $O_{3,Emi}$ ). Figure 3 exhibited the average  $O_{3,Obs}$ ,  $O_{3,Normal}$ ,  $O_{3,Met}$  during the same 216 periods in 2019 and 2020, respectively. It is obvious that the O<sub>3,Obs</sub> during Pre-lockdown period is 217 much lower than that during Full-lockdown period in both years, which was partly attributed to 218 negative influence of meteorological condition during Pre-lockdown period (Figure 3). This is 219 consistent with the increasing temperature and solar radiation, which could significantly contribute to 220 the increase in ozone concentration, from Pre-lockdown to Full-lockdown period. It should be noted 221 that meteorology constrained O<sub>3</sub> concentrations by 3.9 ppbv during the Full-lock down period in 2019. 222 Apart from the influence of meteorological condition, the O<sub>3.Normal</sub> in Full-lockdown period in 2020 is 223 still 1.46 ppbv and 0.64 ppb higher than that during Full-lockdown period in 2019 and that during Pre-224 lockdown period in 2020, indicating that improper decline of precursor emissions was possibly the 225 key reason for the obvious increase of O<sub>3</sub> during Full-lockdown period in 2020.



226 227 Figure 3. Comparison of observed (Obs), weather-normalized (Normal), and meteorological-factors-infected

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# 229 **3.2.2 Ambient VOCs**

230 As mentioned above, the changes in  $O_3$  precursor emissions strongly affected the  $O_{3,Obs}$ , and the 231 changes in VOCs and NO<sub>x</sub> emissions would eventually be reflected by the observed concentrations of 232 individual VOCs and NO<sub>x</sub>. Therefore, the concentrations of each VOC group in different periods were 233 summarized (Figure 4). OVOCs dominated the total VOCs (TVOC) concentrations during the whole 234 observation, with a daily average concentration of  $21.44 \pm 10.27$  ppbv. During Full-lockdown period, 235 the TVOC dropped to 22.19  $\pm$ 7.9 ppbv from 32.78  $\pm$  13.81 ppbv, which was mainly affected by the 236 decrease in industrial activities and traffic volume. This is proved by the trend of traffic volume, 237 VOCs emission and traffic/industrial-derived VOCs (Text S1 and Figure S3). In addition, Jensen et al. 238 (2021) found the VOC emissions from most industries in Changzhou share the same "U-shape" trend 239 as our study. The most obvious drop was found in aromatics ( $\sim$ 54%), followed by OVOCs ( $\sim$ 27%), alkenes (~26%), nitrogen hydrocarbon (~25%), and other VOCs (~21%). Additionally, the 240 241 discrepancy of daytime and nighttime VOCs concentrations during different periods were compared 242 (Figure 4 (A)). The concentration of each VOCs group exhibited higher values during nighttime, 243 which was caused by the low atmospheric oxidation condition and the low atmospheric boundary layer height (Maji et al., 2020; Valach et al., 2015). 244

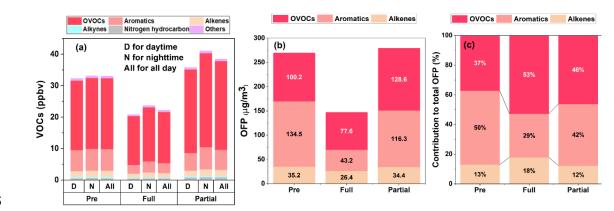
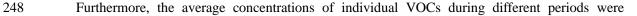




Figure 4. Comparison of daytime and nighttime VOCs concentrations (A), average OFP (B), and contribution
 to total OFP (C) during different periods.



summarized in Figure 5. Total 42 VOC species exhibited an 'U' shape trend during the whole observation, while formaldehyde (HCHO) and methanol showed an obvious increasing pattern. It should be noted that the measurement of HCHO could be strongly influenced by humidity. Since within the drift tube, the back reaction, which converse the protonated HCHO back into HCHO, is highly humidity dependent (Inomata et al., 2008; Warneke et al., 2011).

254 To quantitatively evaluate the changes of individual VOC concentrations from Pre-lockdown to 255 Full-lockdown period, when the variations of each VOCs are obvious, we applied MK trend test and 256 Sen's slope analysis based on the hourly average VOCs concentration data (Table S3). Table 3 lists the 257 top 10 VOCs species with decreasing pattern (with negative Q value) from Pre-lockdown to Full-258 lockdown period. Toluene, benzene and xylene exhibited the most significant decreasing pattern, with a slope of  $7.73 \times 10^{-4}$ ,  $7.36 \times 10^{-4}$ , and  $7.20 \times 10^{-4}$  ppbv h<sup>-1</sup>, respectively. As for NO<sub>x</sub> and TVOC, the slope 259 was  $-1.62 \times 10^{-2}$  and 5.48  $10^{-3}$  ppb h<sup>-1</sup> (Table S3). This result is consistent with the drastic drop of 260 261 industrial activities and traffic volumes, which are key sources of aromatics and NO<sub>x</sub>, from Pre-262 lockdown to Full-lockdown period. Other VOCs, such as ethyl-acetate, acetic acid, acetaldehyde, 263 diethyl sulfide, ethanol, butanol and acrolein are also tightly associated with industrial processes, 264 thereby showed decreasing trend from Pre-lockdown to Full-lockdown period. Additionally, the 265 average diurnal variations of acetonitrile, dimethyl formamide (DMF), and styrene, which are tracers 266 of biomass burning and industrial emission, respectively, exhibited significant reduction during Full-267 lockdown period (Figure S4), also indicating strong decrease in these emissions. However, formaldehyde and methanol exhibited increasing trend, with a slope of  $12.78 \times 10^{-4}$  and  $6.35 \times 10^{-4}$  ppbv 268 269 h<sup>-1</sup>, respectively. This could be explained by the secondary formation of HCHO and methanol, which 270 was promoted under better oxidation condition in Full-lockdown period.

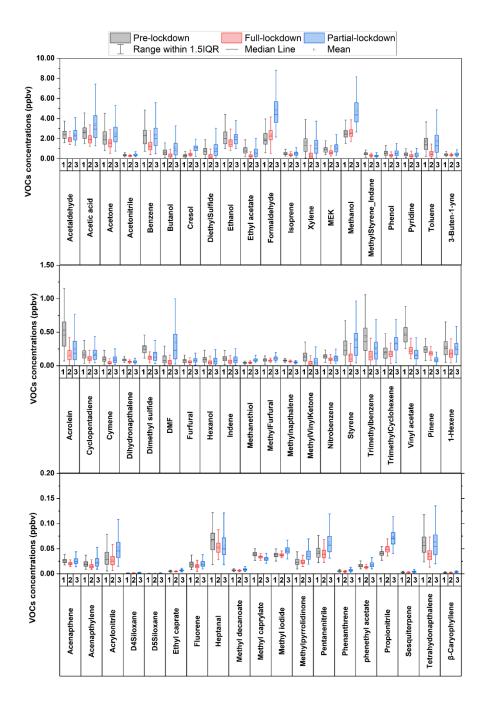


 Figure 5. Concentrations of individual VOC species during different period.

\*MEK, DMF, are abbreviation of Methyl ethyl ketone and dimethylformamide, respectively.

Table 3. Top 10 VOCs with decreasing trend from Pre-lockdown to Full-lockdown

Z value	Q *10000 (ppbv h <sup>-1</sup> )	VOC	Z value	Q *10000 (ppbv h <sup>-1</sup> )
-14.02	-7.73	Acetaldehyde	-10.31	-3.95
-9.65	-7.36	Diethyl sulfide	-9.15	-3.16
-12.38	-7.20	Ethanol	-5.48	-3.09
-18.53	-5.20	Butanol	-10.42	-2.83
-6.79	-4.12	Acrolein	-15.48	-2.76
	-14.02 -9.65 -12.38 -18.53	-14.02     -7.73       -9.65     -7.36       -12.38     -7.20       -18.53     -5.20	-14.02     -7.73     Acetaldehyde       -9.65     -7.36     Diethyl sulfide       -12.38     -7.20     Ethanol       -18.53     -5.20     Butanol	-14.02     -7.73     Acetaldehyde     -10.31       -9.65     -7.36     Diethyl sulfide     -9.15       -12.38     -7.20     Ethanol     -5.48       -18.53     -5.20     Butanol     -10.42

#### 280 **3.2.3 Chemistry perspective**

The reactivities of different VOCs vary significantly, hence, ozone formation potential (OFP) is used in this study to assess the potential contribution of active VOCs (including alkenes, aromatics and OVOCs) to  $O_3$  formation on the same basis, and it can be calculated by formula (5):

$$OFP_i = MIR_i \times [VOC_i] \tag{5}$$

284 where  $MIR_i$  is the ozone formation potential coefficient for a given VOC species i in the maximum 285 increment reaction of O<sub>3</sub>, acquired from Carter (2009);  $[VOC_i]$  is the concentration of VOC species i 286 (in  $\mu g/m^3$ ). It should be noted that OFP does not indicate O<sub>3</sub> concentration but only serves as a 287 reference for the potential O<sub>3</sub> produced via the degradation of VOCs. The time series of total OFP is 288 shown in Figure 6. The average OFP in Pre-lockdown, Full-lockdown, and Partial-lockdown period 289 was  $269.4 \pm 146.0$ ,  $147.2 \pm 72.4$ ,  $279.3 \pm 168.6 \,\mu g/m^3$ , respectively. The trend of the total OFP 290 indicates the drastic decrease of VOCs reactivities from Pre-lockdown to Full-lockdown period. 291 During Pre-lockdown period, aromatics were the dominant OFP contributor (49%), followed by 292 OVOCs (38%) and alkenes (13%) (Figure 4). Among VOCs, xylene exhibited the maximum OFP 293 value (68.6 ± 59.3  $\mu$ g/m<sup>3</sup>), followed by acetaldehyde (28.8 ± 6.4  $\mu$ g/m<sup>3</sup>), toluene (25.7 ± 20.1  $\mu$ g/m<sup>3</sup>) trimethylbenzene ( $25.4 \pm 15.8 \ \mu g/m^3$ ), and formaldehyde ( $22.7 \pm 9.1 \ \mu g/m^3$ ) (Figure S5). Compared to 294 295 Pre-lockdown period, the OFP of aromatics decreased dramatically (-91.2  $\mu$ g/m<sup>3</sup>) during Full-296 lockdown period (Figure 4 (B)), which was mainly attributed to the rapid decline of human activities 297 (e.g., transportation and industry). However, the OFP of alkenes and OVOCs only decreased by 8.9 and 22.5 µg/m<sup>3</sup>, respectively. During the PTR-TOF-MS observation, the most abundant alkenes and 298 aromatics are 1-hexene and isoprene, with the  $k_{OH}$  of 100 and 57  $\times 10^{-12}$  cm<sup>3</sup> molecule<sup>-1</sup> s<sup>-1</sup>, 299

300 respectively (Atkinson and Arey, 2003). The fast degradation of these alkenes could attribute to the 301 relatively smaller change of OFP from alkenes. As for OVOCs, the secondary formation could 302 compensate the decrease in primary emissions. The OFP values of aromatics and alkenes during Pre-303 lockdown and Partial-lockdown period are comparable, but OVOCs exhibited higher OFP 304 contribution (~46%) in Partial-lockdown period, which could be attributed to the higher AOC, 305 enhanced solar radiation and temperature during Partial-lockdown period. To compare the average 306 reactivity of VOCs during different periods, we calculated the mean MIR, derived by dividing the 307 total OFP by total VOC concentration, in each period. A higher MIR means stronger capability of 308 VOCs to produce ozone. As shown in Figure 7, the average MIR during Pre-lockdown, Full-lockdown, and Partial-lockdown period was 3.85, 3.53 and 3.68 (g O<sub>3</sub>/g VOC), respectively. This result suggests 309 310 that VOC species composition in Full-lockdown is more conductive to ozone formation than that in 311 Pre-lockdown, and Partial-lockdown period. However, the formation of O<sub>3</sub> was sensitive to the ratio 312 of  $NO_x/VOCs$  and meteorological conditions, which can be significantly different in each period. As 313 shown in Figure 7, the average  $NO_x/VOCs$  ratio in the three periods (shown in) was 1.84, 0.79, and 314 0.84, respectively, suggesting more NO<sub>x</sub> was reduced than VOCs during Full-lockdown period, which 315 could further influence the sensitivity of O<sub>3</sub> formation.

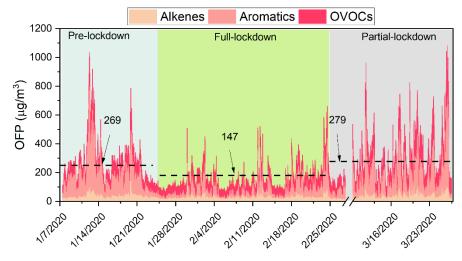


Figure 6. Time series of OFP during the whole observation period (dash lines represent the average OFP value)

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during each period)

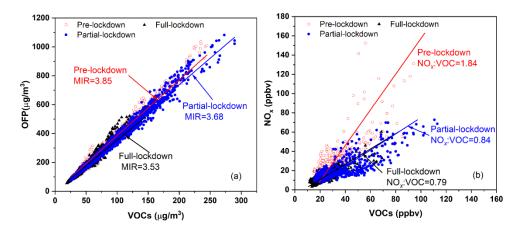
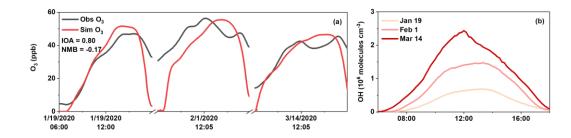




Figure 7. Plot of 1-hour averaged MIR and NOx vs VOCs during three periods.

321 To investigate the detailed formation mechanism of  $O_3$  in each period, three cases (January 19<sup>th</sup>), 322 February 1<sup>st</sup>, March 14<sup>th</sup>) with stagnant meteorological conditions were chosen. The index of 323 agreement (IOA) of  $O_3$  is 0.80, indicating that the model can capture the daytime variation of  $O_3$ . The simulated daytime OH concentrations exhibited an increasing trend from January 19th to March 14th, 324 325 with an average value of  $0.36 \pm 0.27 \times 10^6$ ,  $0.75 \pm 0.54 \times 10^6$  and  $1.18 \pm 0.78 \times 10^6$  molecules cm<sup>-3</sup>, 326 respectively. This could be attributed to the increasing solar radiation and temperature from January to 327 March. To analyze the atmospheric oxidation, we calculated the AOC according to Eq(1). The average daytime AOC on Jan 19<sup>th</sup>, Feb 1<sup>st</sup>, and Mar 14<sup>th</sup> was  $0.26 \pm 0.35$ ,  $0.23 \pm 0.33$ , and  $0.31 \pm 0.38$ 328 molecules cm<sup>-3</sup> s<sup>-1</sup>, respectively (Figure 9). Comparatively, these values are much lower than those 329 330 simulated for Shanghai and Beijing (Liu et al., 2012; Zhu et al., 2020; Zhang et al., 2021) in summer, 331 mainly due to the meteorological conditions in winter season. It is notable that the simulated OH on Jan 19th was significantly lower than that on Feb 1st, but the AOC on Jan 19th was comparable to that 332 333 on Feb 1st. This should be ascribed to the abundant primary pollutants, which efficiently react with OH, 334 during Pre-lockdown period.

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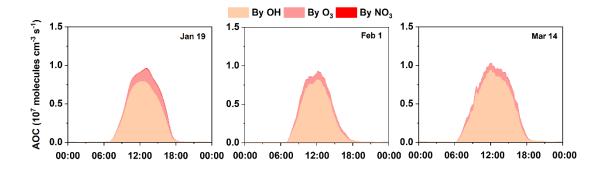


cases.

337 Figure 8. Comparison of simulated and observed O<sub>3</sub> (a) and simulated daytime OH concentrations (b) in three

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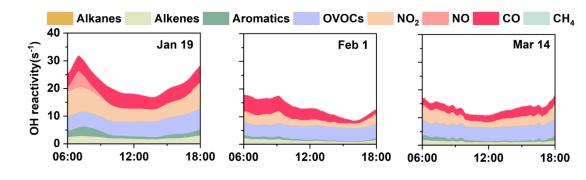




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#### Figure 9. Diurnal variation of AOC in three cases

341 The daytime variations of OH reactivity calculated by OBM model are exhibited in Figure 10, 342 including the contribution from measured pollutants (e.g., VOCs, NO<sub>x</sub>, and CO) and model-simulated 343 species (OVOCs). Generally, the  $k_{OH}$  assessed at Changzhou was in the range of 9~32 s<sup>-1</sup>, which was comparable to that calculated for other cities in China (e.g., Shanghai 4.6~25 s<sup>-1</sup>, Zhu et al., 2020, 344 Chongqing 15~25 s<sup>-1</sup>, Tan et al., 2019 and Beijing 15~25 s<sup>-1</sup>, Tan et al., 2019). It is obvious that OH 345 reactivity peaked in the morning, with maximum values of 31.76, 17.98, and  $17.30 \text{ s}^{-1}$ , respectively. 346 347 The OH reactivity from  $NO_2$  exhibited obvious daytime variation, especially during the morning rush hour, which lead to the peak  $k_{OH}$  value during morning. The OH reactivity ( $k_{OH}$ ) on Feb 1<sup>st</sup> was much 348 349 lower than that in the other two cases, which was mainly due to the abundance of emissions during 350 Pre-lockdown and Partial-lockdown period. Compared to Jan 19th, the koH from NO<sub>2</sub> on Feb 1st and Mar 14<sup>th</sup> showed lower levels, with an average value of 2.62 and 3.35 s<sup>-1</sup>, respectively. This 351 352 corresponds with the dramatic drop of traffic volume during lockdown periods. Similarly, compared to Jan 19<sup>th</sup>, the k<sub>OH</sub> from alkenes and aromatics were lower on Feb 1<sup>st</sup> and Mar 14<sup>th</sup>. As k<sub>OH</sub> from OVOC, 353



#### 355 Full-lockdown period.



#### Figure 10. Daytime variation of OH reactivity in three cases

358 To investigate the variation of O<sub>3</sub> during different periods, the formation and loss pathways of O<sub>3</sub> 359 were calculated (Figure 11). The formation of  $O_3$  ( $P(O_3)$ ) was dominated by HO<sub>2</sub>+NO and RO<sub>2</sub>+NO 360 pathways. Although the average MIR during Full-lockdown period was the minimum among the three periods, the  $P(O_3)$  on Feb 1<sup>st</sup> was higher than that on Jan 19<sup>th</sup>. This could be attributed to the higher 361 362 AOC and better photochemical conditions during Full-lockdown period. Similarly, much higher  $P(O_3)$ was found on March 14th. To avoid the influence of meteorological conditions and test the potential 363 364 mean  $O_3$  (Mean $O_3$ ) concentrations under different  $NO_x/VOCs$  ratios, a series of scenario analyses 365 were performed based on the average condition during the whole observation, and the isopleths of 366 MeanO<sub>3</sub> concentrations are exhibited in Figure. 12. Note that the value of temperature and photolysis 367 frequencies (J values) in the scenario analyses could be higher than the actual value during Pre-368 lockdown period and could further lead to overestimation of simulated MeanO<sub>3</sub> during Per-lockdown 369 period. Additionally, the VOCs concentrations mentioned in this section only represent the VOC 370 species in the MCM mechanism. By connecting the inflection points in each  $O_3$  isopleth, we get the 371 ridge line, which divides the whole regime into NO<sub>x</sub>-sensitive and VOCs-sensitive regimes (Figure. 372 12). During Pre-lockdown period, the  $O_3$  formation was in VOC-limited regime (triangles in Figure. 373 12), with an average  $NO_x/VOC$  ratio of 1.84. As for Full-lockdown period, significant decrease of 374 NOx and VOC emissions was observed, and the  $NO_x/VOCs$  ratio dropped to 0.79, which gradually switched the  $O_3$  formation to the junction of VOCs-limited and  $NO_x$ -limited regimes, especially on 375 376 Feb 16<sup>th</sup> and Feb 17<sup>th</sup> (circles in the red rectangle in Figure. 12), when the O<sub>3</sub> formation went into 377  $NO_x$ -limited regime. During Partial-lockdown period, increasing of VOCs and  $NO_x$  emission again 378 dragged the formation of  $O_3$  back into VOCs-limited regime (triangles in Figure. 12). Interestingly, 379 although a great deal of NO<sub>x</sub> and VOCs emissions were diminished during Full-lockdown period, the 380 average Mean $O_3$  in Full-lockdown was supposed to be 2.4 ppbv higher than that in Pre-lockdown 381 period. This result is consistent with the trend of the observed MDA8  $O_3$  and the results of the deweathered calculation. Therefore, expect for the influence of meteorology, the improper NO<sub>x</sub>/VOCs 382 383 reduction ratio and further influence on chemistry was the key reason for the obvious increase of  $O_3$ 384 during Full-lockdown period in Changzhou in 2020.

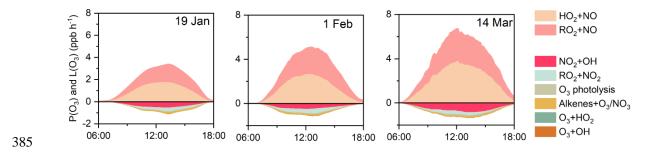
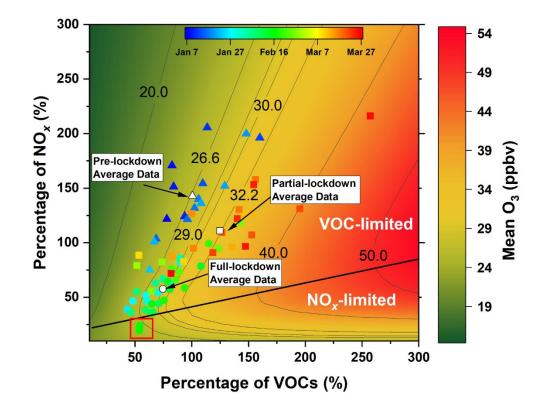




Figure 11. Daytime variation of P(O<sub>3</sub>) and L(O<sub>3</sub>) in three cases



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Figure. 12 MeanO<sub>3</sub> isopleth. The colored circles, triangles, and rectangles represent the daily average

89 concentrations of NO<sub>x</sub> and VOCs during Pre lockdown, Full-lockdown, and Partial-lockdown period,

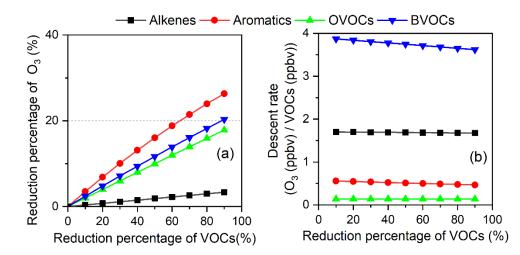
390 391 respectively. The white circle, triangle, and rectangle indicates the average  $NO_x$  and VOCs concentrations

during Pre lockdown, Full-lockdown, and Partial-lockdown period, respectively.

392 The scenario analyses raise a question: how much  $O_3$  would change as a function of reduction of 393 NO<sub>x</sub> and VOCs? Therefore, the reduction percentage of O<sub>3</sub> ( $\Delta$ O<sub>3</sub>/O<sub>3</sub>) during Pre-lockdown period as a 394 function of reduction of VOCs and NOx were calculated, and the result could be regarded as a 395 potential to control  $O_3$  pollution. Based on the VOCs species in MCM v3.3.1, we classified the 396 measured VOCs into four groups: alkenes (n-butene); aromatics (including benzene, toluene, phenol, 397 xylene, styrene, cresol, and trimethylbenzene); OVOCs (including methanol, ethanol, formaldehyde, 398 aldehyde, acrolein, methyl vinyl ketone, methyl ethyl ketone, ethyl acetate, methyl isobutyl ketone, 399 hexanol, and heptanal); and BVOCs (isoprene, pinene, and caryophyllene). The results in Figure 13(a) 400 indicate that more reduction potential of  $O_3$  could be achieved by diminishing aromatics, followed by 401 BVOCs, OVOCs, and alkenes. It should be noted that many light alkanes and active alkenes, such as 402 ethene and propene, could not be measured by the PTR-TOF-MS and might further lead to the 403 underestimation of ozone production from alkanes and alkenes. Additionally, this comparison has a 404 drawback of being influenced by the concentrations of VOCs. To normalize the influence of 405 concentrations of VOCs, the descent rate of  $O_3 (\Delta O_3 (ppbv) / \Delta VOCs (ppbv))$  as a function of 406 reduction percentage of VOCs were calculated (Figure 13 (b)).  $O_3$  exhibited the highest dependence 407 on BVOCs, with an average descent rate of  $3.74 \pm 0.09$  ppbv/ppbv. Differing from the result in Figure 408 13 (a), diminishing alkenes could lead to decrease of  $O_3$  by an average declining rate of  $1.69 \pm 0.01$ 409 ppbv/ppbv. On the contrary, reduction of NO<sub>x</sub> would lead to increase of  $O_3$ , with an average rate of  $1.29 \pm 0.21$  ppbv/ppbv (Figure S6). Although the descent rate of O<sub>3</sub> turned to decrease and the 410 411 sensitivity of  $O_3$  formation get into NOx-limited regime when over 70% of NOx were eliminated, it 412 still causes net increase of O<sub>3</sub>.

Although diminishing BVOCs seems to the most efficient way to restrain O<sub>3</sub> pollution, most of
BVOCs were emitted directly from plants and could not be easily controlled. Besides, huge number of

415 OVOCs (such as formaldehyde, aldehyde, methanol, ethanol, methyl vinyl ketone, methyl ethyl 416 ketone, etc.) could be directly emitted from anthropogenic processes or secondary generated from the 417 oxidation of precursors (such as alkenes and aromatics), which complicates the control of OVOCs. 418 Therefore, considering the reduction potential and descent rate of O<sub>3</sub>, more efforts are needed on the 419 control of alkenes and aromatics.



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Figure 13. Reduction percentage of O<sub>3</sub> as a function of reduction percentage of VOCs (a); descent rate of O<sub>3</sub> as a
function of reduction percentage of VOCs (b).

**3.3 Uncertainty analysis** 

#### function of reduction percentage of voc

424 Due to limitations in the observations, several issues should be noted in the application of the 425 OBM model to evaluate the local chemistry in the present study. Firstly, deficiency of the observation of C2~C5 alkenes and alkanes could lead to underestimation of the simulated O3. We can only obtain 426 427 the C2~C5 alkenes and alkanes concentrations from the observation during the autumn of 2018 at the 428 same site. To analyze the uncertainties from this disadvantage, we have done simulation by including 429 assumed diurnal variation of ethene, propene, butene, ethane, propane and butane which are key 430  $C2\sim C5$  alkenes and alkanes at this site, in the model. On average, adding 0.5~2 times alkenes or 431 alkanes could lead to 1.65%~9.49% or 1.37~5.36% increase of simulated O<sub>3</sub>, respectively (Figure S7 432 and S8). In addition, the deficiency of  $C2\sim C5$  has potential to cause uncertainty in  $O_3$  formation 433 potential. To quantify this impact, the EKMA analysis with the hypothetical diurnal variation of 434 C2~C5 was also performed. Generally, adding C2~C5 alkenes and alkanes in the model would lead to

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435 slight increase of the simulated  $O_3$ , and could not obviously change the shape of  $O_3$  isopleth (Figure 436 S9). Therefore, the influence of the deficiency of  $C2\sim C5$  alkenes and alkanes on the  $O_3$  formation 437 sensitivity is negligible. It should be noted that, this sensitivity analysis is based on the "hypothetical" 438 diurnal variation of C2~C5 alkenes and alkanes, which would bring in uncertainty. We hope a wider 439 range of VOCs would be monitored simultaneously in future field campaign and avoid this deficiency. 440 Secondly, the photolysis frequencies (J values) were calculated as a function of solar zenith angle, 441 altitude using lookup tables, calculated using the Tropospheric Ultraviolet and Visible (TUV) model, 442 which could lead to uncertainty in the simulation of  $O_3$ . Hence, we analysis the influence of J values 443 by increasing or decreasing the photolysis rates by 10% and 20%. Results showed that the simulated 444  $O_3$  could decrease or increase by 25.14% or 21.73%, respectively, when photolysis rates were 445 decreased or increased by 20% (Figure S10). In addition, the J values, which directly or indirectly 446 influence the recycling of  $RO_x$ , could lead to uncertainty in the calculation of AOC and  $k_{OH}$ . Based on 447 above sensitivity analysis, we found the relative changes in AOC and  $k_{OH}$  by 1% changes in J values 448 was 1.07% and 0.14%, respectively. Therefore, the J values is recommended to be measured during 449 future observations.

# 450 **4. Conclusions**

451 After the outbreak of COVID-19, strict epidemic prevention measures have been adopted 452 throughout China, leading to dramatic decrease in traffic volume and industrial activities. Affected by 453 the decrease of number of vehicles on the road, non-essential industrial productivity, and associated 454 pollutant emissions, most of the air pollutants (e.g., PM<sub>2.5</sub>, PM<sub>10</sub>, NO, NO<sub>2</sub>, SO<sub>2</sub>, and VOCs) dropped 455 to a lower level during lockdown period (especially during Full-lockdown period). However,  $O_3$ 456 increased compared to that during the same period in 2019 in many urban areas of China. To figure 457 out the reasons for this obvious increase of  $O_3$ , the characteristics of  $O_3$  precursors (NO<sub>x</sub>, VOCs) 458 during Pre-lockdown, Full-lockdown, and Partial-lockdown periods in Changzhou were analyzed. 459 Although this study was conducted in single city of China, the representativeness of Changzhou 460 guaranteed the applicability of the results the YRD region. Results suggested that the decrease of 461 human activities during Full-lockdown period significantly suppressed the emissions of  $NO_x$  and 462 VOCs, which further lead to dramatic drop in the concentrations of most VOCs, especially aromatics. As a result, the NOx/VOCs ratios dropped from 1.84 at Pre-lockdown period to 0.79 during Full-463 464 lockdown period. By deweathered calculation, we found that meteorology constrained  $O_3$ 465 concentration by 3.9 ppbv during Full-lockdown period in 2019, but exhibited negligible influence on 466 that during the same period in 2020. However, compared to Full-lockdown period in 2019, changes in 467 precursor emissions led to 1.46 ppby increase in O<sub>3</sub> concentrations during the same period in 2020. To 468 verify this result, a box model was used to simulate the formation of  $O_3$ . Results show that the AOC 469 level during Full-lockdown was comparable to that during Pre-lockdown period, but the formation 470 rate of  $O_3$  was much higher during Full-lockdown period. By scenario analysis, we found the decrease of NO<sub>x</sub> and VOCs in Full-lockdown period dragged the formation of  $O_3$  from VOC-sensitive regime 471 472 to the junction of VOCs- and NO<sub>x</sub>-limited regime, and the average simulated MeanO<sub>3</sub> in Full 473 lockdown period could be 2.4 ppbv higher than that in Pre-lockdown period. Although the 474 deweathered model and OBM model shows differences in the emission-derived change of  $O_3$ , the 475 results together point out that the improper reduction of  $NO_x$  and VOCs was the key reason for the 476 obvious increase of O<sub>3</sub> during Full-lockdown period in 2020. Overall, the outbreak of COVID-19 has 477 caused devastation over the world. However, it provided an extreme experiment to investigate the  $O_3$ 478 formation under strict emission control policies and provided insights into the policy formulation for 479 diminishing O<sub>3</sub> pollution in the YRD region. The data indicate that the concentrations of VOCs and 480  $NO_x$  have changed dramatically during the pandemic, a common situation also found in other Chinese 481 cities, and led to the switch of  $O_3$  formation sensitivity. These results have a clear indication that, in 482 the future, more efforts should be paid on the reduction ratio of anthropogenic VOCs and NOx.

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