



# 1 Wet deposition in the remote western and central Mediterranean

# 2 as a source of trace metals to surface seawater

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31 Abstract. This study reports the only recent characterisation of two contrasted wet deposition events 32 collected during the PEACETIME cruise in the Mediterranean open seawater, and their impact on 33 trace metals (TMs) marine stocks. Rain samples were analysed for Al, 12 trace metals (TMs 34 hereafter, including Co, Cd, Cr, Cu, Fe, Mn, Mo, Ni, Pb, Ti, V and Zn) and nutrients (N, P, DOC) 35 concentrations. The first rain sample collected in the Ionian Sea (rain ION) was a wet typical 36 regional background deposition event whereas the second rain collected in the Algerian Basin (rain 37 FAST) was a Saharan dust wet deposition. The concentrations of TMs in the two rain samples were significantly lower compared to concentrations in rains collected at coastal sites reported in the 38 literature, suggesting either less anthropogenic influence in the remote Mediterranean environment, 39 40 or decreased emissions during the last decades in the Mediterranean Sea. The TMs inventories in 41 the surface microlayer and mixed layer (0-20m) at ION and FAST stations before and after the 42 events, compared to atmospheric fluxes, showed that the atmospheric inputs were a significant 43 source of particulate TMs for both layers. At the scale of the western and central Mediterranean, the 44 atmospheric inputs were of the same order of magnitude as marine stocks within the ML for dissolved Fe, Co and Zn, underlining the role of the atmosphere in their biogeochemical cycle in 45 46 the stratified Mediterranean Sea. In case of intense wet dust deposition event, the contribution of atmospheric inputs could be critical for dissolved stocks of the majority of TMs. 47





## 48 1. Introduction

49 Atmospheric deposition of continental aerosol has long been recognized to influence trace element 50 concentrations in remote oceanic surface waters (Buat-Ménard and Chesselet, 1979; Hardy, 1982; 51 Buat-Ménard, 1983). In particular, the Mediterranean Sea (Med Sea) is an oligotrophic environment 52 where marine biosphere growth is nutrient-limited during the long Mediterranean summer season 53 characterized by a strong thermal stratification of surface waters (The Mermex Group, 2011). The 54 Mediterranean atmosphere is characterized by the permanent presence of anthropogenic aerosols 55 from industrial and domestic activities around the basin (e.g., Sciare et al., 2003; Kanakidou et al., 2011). In addition to this anthropogenic background, the Mediterranean basin is also subject to 56 seasonal contributions of particles from biomass fires in summer (Guieu et al., 2005) and to intense 57 58 sporadic Saharan dust inputs (e.g., Loÿe-Pilot and Martin, 1996; Vincent et al., 2016). Several 59 studies emphasized that the atmospheric deposition of aerosols, notably through wet deposition, 60 plays a significant role in marine cycles of both nutrient, such as nitrogen (N) and phosphorus (P) 61 (e.g. Pulido-Villena et al., 2010; Richon et al., 2018 a and b; Violaki et al., 2018) and micronutrients, such as iron (Fe) (Bonnet and Guieu, 2006). Recently, atmospheric dust inputs were identified to 62 have a fertilizing effect on the plankton stocks and fluxes, even in the presence of relatively high 63 64 nutrients and Fe marine concentrations (Ridame et al., 2011; reviewed in Guieu and Ridame, 2021). 65 Mackey et al. (2012) showed that TMs provided by dust deposition could explain this fertilizing 66 effect. Indeed, some TMs, including Mn, Co, Ni (Mackey et al., 2012), Cu (Annett et al., 2008) and 67 Zn (Morel et al., 1991), play physiological roles for phytoplanktonic organisms. These TMs are 68 present in very low concentrations in oligotrophic systems such as the Med Sea, possibly limiting the phytoplankton growth (Pinedo-Gonzàles et al. 2015) and implying the role of dust deposition as 69 70 source of TMs for planktonic communities. On the other hand, atmospheric deposition of European 71 aerosol particles was identified to have a negative effect on chlorophyll concentrations (Gallisai et 72 al. 2014), by providing trace metals, as Cu, at toxic levels (Jordi et al. 2012).

73 The atmospheric deposition of TMs in the Mediterranean is related to both dust and anthropogenic aerosols deposition (Desboeufs et al., 2018). The role of dust deposition as a source of TMs was 74 75 observed from the correlation between the atmospheric deposition of mineral dust, and the 76 enrichment of dissolved TMs (Cd, Co, Cu, Fe) in the Mediterranean Sea surface microlayer (Tovar-77 Sánchez et al., 2014). For the water column, the adding of dissolved Fe and Mn was emphasized in 78 mesocosm experiments after dust addition mimicking intense wet dust deposition (Wuttig et al., 79 2013). Yet, the direct impact of wet deposition events on TMs concentrations in surface seawater 80 has never been studied *in situ*, whether in the Mediterranean or in other oceanic regions. Moreover,





81 the two key criteria used to assess the potential impact of TMs and nutrients wet deposition are their 82 respective concentration (or flux) and solubility, i.e., the partitioning between dissolved and total 83 concentrations in rainwaters. Indeed, it is considered that the dissolved fraction of nutrients and 84 TMs can be directly assimilated by the phytoplankton. Few studies focussed on concentrations of 85 TMs in rainwater samples was collected around the Mediterranean basin: Al-Momani et al., (1998), 86 Kanellopoulou, (2001) and Özsoy and Örnektekin, (2009) in the eastern basin, and Losno (1989), 87 Guieu et al. (1997), Frau et al., (1996), Chester et al., (1997) and Guerzoni et al. (1999b) in the western basin. These studies led to highly variable TMs concentration and solubility, illustrating a 88 more general large variability of TMs inputs during wet depositions in the Mediterranean Sea 89 90 (reviewed in Desboeufs, 2021). All these studies were performed at coastal sites. Offshore samples of rainwater have rarely been studied in the literature so far. In the Mediterranean, to our knowledge, 91 92 trace element concentrations from only three rain samples collected at sea in April 1981 have been reported in a PhD thesis (Dulac, 1986). However, due to the continental and local source of pollution 93 94 and the variety of anthropogenic aerosol sources (Amato et al., 2016), the TMs rain composition of 95 the coastal zone could not be representative of atmospheric deposition in the remote Mediterranean.

96 The PEACETIME cruise (ProcEss studies at the Air-sEa Interface after dust deposition in the 97 MEditerranean Sea) performed in spring 2017 aimed at studying the impacts of atmospheric 98 deposition, in particular Saharan dust events, on the physical, chemical and biological processes in 99 this marine oligotrophic environment (Guieu et al., 2020). We investigated here the concentration 100 and solubility of TMs and nutrients of two rainwaters sampled in the Mediterranean open sea during 101 the cruise. We compare our results on TMs concentrations and solubility in rainwater with previous 102 studies based on rainwater samples collected at coastal sites to investigate potential differences with the open sea. Additionally, to assess the impact of wet deposition on the surface TMs 103 104 concentrations, surface seawater including the surface micro-layer was concurrently collected for the first time with rain samples. 105

106 2. Sampling and methods

## 107 2.1 Sampling and chemical analysis of rainwater

The PEACETIME oceanographic campaign (<u>https://doi.org/10.17600/17000300</u>) took place in the
western and central Mediterranean Sea on-board the French research vessel (R/V) *Pourquoi Pas ?*between 11 May and 10 June 2017, i.e. at the beginning of the Mediterranean stratification season
(Guieu et al., 2020). The rain collector was installed on the upper deck (22 m above sea level)





112 where no on-board activities were taking place to avoid contamination. A rain collector was equipped with an on-line filtration system to directly separate the dissolved and particulate fractions 113 114 (details of the filtration system are available in Heimburger et al., 2012) allowing for the calculation of solubility of TMs in the rainwater at the time of collection. The filtration device was equipped 115 116 with a Nuclepore® polycarbonate membrane (PC) filter (porosity: 0.2 µm, diameter: 47 mm), and 117 the diameter of the funnel of the collector was 24 cm. The rain collectors were installed only when 118 rain was expected and kept closed within an acid-washed sealed plastic film until the rainfall began. 119 All the sampling materials were thoroughly washed in the laboratory prior to the cruise departure (washing protocol described in Heimburger et al., 2012). No stabilizing setup was used to keep the 120 121 funnel level during the pitch and roll of the ship, preventing a precise assessment of the height of rainfall from the collected water volume. During the rain sampling, the ship was always facing the 122 123 wind to avoid contamination by the smoke of the ship itself, as the chimney was situated on the 124 lower deck and behind the collector.

125 Immediately after sampling, the collector was disassembled under the laminar flow hood inside a clean-room on-board laboratory. The dissolved fraction was separated in 4 aliquots dedicated to i) 126 127 dissolved organic carbon (DOC) by high-temperature catalytic oxidation (HTCO) on a Shimadzu 128 total organic carbon analyzer (as described in Van Wambeke et al 2021), ii) major ions by ion 129 chromatography (IC), iii) metals analysis by inductively coupled plasma methods (ICP), and iv) pH 130 measurement. For ICP measurements, the sample was acidified immediately to 1% by volume of ultra-pure nitric acid: 67-69%, Ultrapur, Normatom®, VWR. For IC analysis, the sample was 131 132 immediately frozen. The filter (particulate fraction) was dried under the laminar flow hood, and then put in a storage box and packed with a plastic bag to avoid contamination. After returning to 133 the laboratory, filters were acid digested by using the adapted protocol from Heimburger et al. 134 (2012) as follows: filters were placed in tightly capped Savillex<sup>™</sup> PFA digestion vessels with 4 mL 135 of a mixture of HNO<sub>3</sub> (67-69%, Ultrapur, Normatom<sup>®</sup>, VWR), H<sub>2</sub>O and HF acids (40%, Ultrapur, 136 137 Normatom<sup>®</sup>, VWR) in a proportion (3: 1: 0.5), then heated in an oven at 130°C for 14 hours. After 138 cooling, acid was completely evaporated on a heater plate (ANALAB, 250, A4) at 140°C for about 2h, then 0.5 mL of H<sub>2</sub>O<sub>2</sub> (30-32%, Romil-UpA<sup>™</sup>) and 1 mL of the acidified water (2% HNO<sub>3</sub>) was 139 140 added to the vessels and heated during 30 min. to dissolve the dry residue in the bottom of the 141 vessels; finally, 12 mL of acidified water (1% HNO<sub>3</sub>) was added to obtain 13.5 mL of solution in a 142 tube for ICP-MS analyses.

The dissolved fraction was analysed by IC (IC 850 Metrohm) for the inorganic and organic anions
(NO<sub>2</sub><sup>-</sup>, NO<sub>3</sub><sup>-</sup>, PO<sub>4</sub><sup>3-</sup>, SO<sub>4</sub><sup>2-</sup>, F<sup>-</sup>, Cl<sup>-</sup>, Br<sup>-</sup>, HCOO<sup>-</sup>, CH<sub>3</sub>COO<sup>-</sup>, C<sub>2</sub>H<sub>5</sub>COO<sup>-</sup>, MSA, C<sub>2</sub>O<sub>4</sub><sup>2-</sup>) and for the





- 145 cation  $NH_4^+$  (Mallet et al., 2017). On the other hand, the dissolved fraction and solution from digestion of particulate fractions were analysed by ICP-AES (Inductively Coupled Plasma Atomic 146 147 Emission Spectrometry, Spectro ARCOS Ametek®) for major elements (Al, Ca, K, Mg, Na, S) (Desboeufs et al., 2014) and by HR-ICP-MS (High Resolution Inductively Coupled Plasma Mass 148 Spectrometry, Neptune Plus TM at Thermo Scientific TM) for TMs: Cd, Co, Cr, Cu, Fe, Mn, Mo, Ni, 149 150 P, Pb, Ti, V and Zn. The speciation of dissolved P was estimated by determining the dissolved 151 inorganic phosphorus (DIP) from phosphate concentrations expressed as P and the dissolved 152 organic phosphorus (DOP) from the difference between total dissolved phosphorus (TDP), obtained 153 by ICP-MS, and DIP, obtained by IC.
- In order to estimate the contamination of sampling and analytical protocols, 3 blanks of rain samples 154 (collected on-board during the cruise with the same protocol without rain events) were used and 155 processed. The procedural limit of detections (LoD) were defined as 3 x standard deviation of blank 156 157 samples both for dissolved and particulate fractions estimated after acid digestion. All samples 158 dissolved and particulate concentrations were higher than LoD, except for NO<sub>2</sub><sup>-</sup> in the two rain samples. The blank concentrations represented 10.2% in average for TMs and were typically lower 159 160 than 20% of the sample concentrations, except for Cd (52%) and Mo (43%) in the dissolved fraction. 161 For the sample concentration computations, we subtracted these blanks values to elemental 162 concentrations obtained in rain samples.
- 163 2.2 Atmospheric ancillary measurements
- 164
- The PEGASUS (Portable Gas and Aerosol Sampling UnitS, www.pegasus.cnrs.fr) mobile platform 165 166 of LISA is a self-contained facility based on two standard 20-feet containers, adapted with air-167 conditioning, rectified power, air intake and exhausts for sampling and online measurements of 168 atmospheric aerosols and gaseous compounds, and their analysis (Formenti et al., 2019). During the PEACETIME cruise, only the sampling module of the facility was deployed on the starboard side 169 170 of deck 7 of the R/V. The PEGASUS instrumental payload of relevance to this paper included 171 measurements of the major gases such as NOx, SO<sub>2</sub>, O<sub>3</sub> and CO by online analysers (Horiba APNA, 172 APSA, APOA and PICARRO respectively; 2-min resolution, detection limit 0.5 ppb and 1 ppb for CO), that were used to estimate origins of collected air masses. 173 From the first of June (not operational before), additional measurements by an ALS450<sup>®</sup> Rayleigh-174
- 175 Mie lidar (Leosphere <sup>TM</sup>; Royer et al., 2011) was used to monitor the vertical distribution of aerosols
- 176 over time and the associated integrated columns. The vertical lidar profiles were analysed to yield
- the apparent backscatter coefficient (ABC) corrected from the molecular transmission, as well as





178 the volume depolarisation ratio (VDR). The inversion procedure (Chazette et al., 2016; 2019) to retrieve the aerosol extinction coefficient (unit km<sup>-1</sup>) uses a vertical-dependent lidar ratio that takes 179 180 into account two aerosol layers. The first layer corresponds to marine aerosols in the marine boundary layer (MBL), the second to a desert aerosol layer that can extend between ~1 and 6 km 181 182 above mean sea level (amsl). In accordance with Chazette et al. (2016), for the same wavelength 183 and region, the lidar ratios were set to 25 and 55 sr, respectively. The vertical profile of the aerosol 184 extinction coefficient was retrieved from 0.2 km amsl on with a vertical resolution of 15 m. Based on these profiles, the integrated column content of dust aerosols was estimated using a specific 185 extinction cross-section of 1.1 m<sup>2</sup> g<sup>-1</sup> as proposed by Raut and Chazette (2009). 186

187 In addition, detailed meteorological data such as air and sea temperature, atmospheric pressure,

relative humidity, atmospheric pressure, heat flux and wind speed and direction were provided on

189 a 30 seconds timestep basis by the ship's permanent instrumentation.

#### 190 2.3 Sampling and analysis of dissolved TMs in seawater

191 Before and after each rain, seawater samples were collected in the surface microlayer (SML: <1 192 mm) and subsurface seawater (SSW: <1-m depth) (Tovar-Sánchez et al., 2020, this special issue). 193 SML and SSW samples were collected from a pneumatic boat 0.5-1 nautical mile away from the 194 R/V in order to avoid any potential contamination. SML samples were collected using a glass plate sampler (Stortini et al., 2012; Tovar-Sánchez et al., 2019) which had previously been cleaned with 195 acid overnight and rinsed thoroughly with ultrapure water (MQ-water). The 39 x 25 cm silicate 196 glass plate had an effective sampling surface area of 1950 cm<sup>2</sup> considering both sides. In order to 197 198 check for procedural contamination, SML blanks were collected at some stations on board of the 199 pneumatic boat by rinsing the glass plate with ultra-pure water and collecting 0.5 L of ultra-pure 200 water using the glass plate system. The surface microlayer thickness was calculated following the formula of Wurl (2009). SSW was collected by using an acid-washed Teflon tubing connected to a 201 peristaltic pump. The total fraction (i.e. T-SML) was directly collected from the glass plate system 202 203 without filtration in a 0.5 L acid cleaned LDPE bottles, while the dissolved fraction in the SML (i.e. 204 D-SML) and SSW (i.e. D-SSW) was filtered in situ through an acid-cleaned polypropylene cartridge 205 filter (0.22 µm; MSI, Calyx®).

TMs samples were also collected in the water column using the trace metals clean (TMC) rosette before and after the rains (Bressac et al., 2021). Although rosette deployments were performed over the whole water column, we focus here on the 0-20 m marine mixed-layer (ML). The water column was sampled using the TMC titanium rosette mounted with 24 teflon-coated Go-Flo bottles.





210 Immediately after recovery, the Go-Flo bottles were transferred inside a class-100 clean laboratory container. Seawater samples were directly filtered from the bottles through acid-cleaned 0.2-µm 211 212 capsule filters (Sartorius Sartobran-P-capsule 0.45/0.2-µm). All samples were acidified on board to pH <2 with Ultrapure-grade HCl under a class-100 HEPA laminar flow hood. Metals (namely Cd, 213 214 Co, Cu, Ni, Mo, V, Zn and Pb) were pre-concentrated using an organic extraction method (Bruland 215 et al., 1979) and quantified by ICP-MS (Perkin Elmer ELAN DRC-e). In order to breakdown metal-216 organic complexes and remove organic matter (Achterberg et al., 2001; Milne et al., 2010), total fraction samples (i.e. T-SML) were digested prior to the pre-concentration using a UV system 217 218 consisting of one UV (80 W) mercury lamp that irradiated the samples (contained in quartz bottles) 219 during 30 min. The accuracy of the pre-concentration method and analysis for TMs was established using Seawater Reference Material (CASS 6, NRC-CNRC) with recoveries ranging from 89% for 220 221 Mo to 108% for Pb. Due to the complexity of the analytical method, all the TMC samplings were not analysed for these metals. Overall, 1 or 2 depths were obtained in the mixed layer (0-20 m). 222 223 Dissolved Fe and Al concentrations were also measured on board. Dissolved Fe concentrations were 224 measured using an automated Flow Injection Analysis (FIA) with online preconcentration and 225 chemiluminescence detection (Bonnet and Guieu, 2006), and dissolved Al concentrations using the fluorometric method described by Hydes and Liss (1976). Sampling and analysis for dissolved Fe 226 227 and Al concentrations are fully described in Bressac et al. (2021), and covered at least 4 depths in 228 the 0-20 m mixed layer.

### 229 2.4 Enrichment factor and solubility

In order to better constrain the origin of TMs in the rain samples, their enrichment factor (EF; Rahn
1976) relative to the Earth's crust was calculated based on their total concentrations (dissolved +
particulate fractions) as:

$$EF = \frac{\left([X]/[Al]\right)sample}{\left([X]/[Al]\right)crust}$$
(1)

where [X]/[Al] is the ratio between an element X and Al concentrations in rainwater samples (at the numerator), and in the Earth's crust (denominator) from Rudnick and Gao (2003). Aluminium is currently used as a reference element as it only has a crustal origin. For a given TM, EF >1 indicates an enrichment with respect to the average composition of the Earth's crust. To account for the soil composition variability of mineral dust atmospheric sources, TMs with an EF value >10 are considered derived from non-crustal sources (Rahn, 1976).

240 The relative solubility of TMs in the two rainwater events was calculated as:





241 
$$S_X \% = \frac{[X] dissolved}{[X] total} \ge 100$$
(2)

242 where  $S_X$ % is the relative solubility (in %) of an element X in the rainwater, [X]<sub>dissolved</sub> and [X]<sub>total</sub>

243 are its soluble and total concentration, respectively.

### 244 2.5 Atmospheric deposition fluxes

245 Impacts on biogeochemical cycles and ecosystem functioning after a rain event occur on time scales of a few days (2-3), and space scales of tens of km (about 20-50 km within the radius of the ship's 246 247 position). In the specific context of oceanographic cruising, the documentation of these impacts is 248 restricted to the vertical dimension at the prescribed temporal scale. In this vertical dimension, the exchange of TMs across the ML was controlled both by atmospheric inputs over the R/V position 249 250 and by advection from surrounding water masses that may have been impacted by surrounding rainfall. Therefore, we had to consider this process in our estimation of the atmospheric fluxes 251 contributions. For this purpose, the atmospheric fluxes have to be integrated to the extent of the rain 252 253 area that can impact the marine surface layers. We derived wet deposition fluxes by considering the 254 total precipitation accumulated during the duration of the rain over the area around the R/V location. Thus, the wet deposition fluxes in our rain samples were calculated by multiplying the volume 255 256 weighted mean rainfall concentration by the total precipitation. The total precipitation of the rain 257 events was issued from the hourly total precipitation accumulated during the rain events over the 258 region from ERA5 ECMWF reanalysis (Herbasch et al., 2018) and from the rain rate composite 259 radar products from the European OPERA database (Saltikoff et al., 2019), when it is possible. Although subject to uncertainties (Morin et al., 2003), a surface-based weather radar is probably the 260 best tool to estimate rainfall in the surroundings of the R/V. However, the OPERA database does 261 262 not include Italian radars, which anyway did not cover the central area of the Ionian Sea during the 263 cruise. ERA5 data are available on regular latitude-longitude grids at 0.25° x 0.25° resolution. The 264 accumulated precipitation was taken from the grid-points spanning the ship's location, more or less  $0.25^{\circ}$  around the central grid-point for integrating the regional variability. Surface rain rate radar 265 composite images were available every 15 minutes with a spatial resolution of 2 km x 2 km. The 266 accumulated precipitation was the sum of integral rain rates during the rain duration averaged over 267 268 the radar pixels spanning the ship location within a radius of about 25 km around the ship location.

### 269 2.6 Stocks in the surface seawater

270 For the surface microlayer (SML), stocks of TMs were estimated from the integration of TMs

271 concentration over the thickness of the layer. The thickness ranged from 32 to 43 µm and from 26

to 43 μm at ION and FAST, respectively (Tovar-Sanchez et al., 2020).





273 The trace metals stocks within the ML were calculated by trapezoidal integrations of marine concentrations from SSW and TMC rosette samplings. The upper water column was stratified along 274 275 the cruise transect (Taillandier et al., 2020), with a ML ranging from 7 to 21 m (11 to 21 m at ION station and 11 to 19 m at FAST station (Van Wambeke et al., 2020). The ML depth (MLD) 276 277 fluctuations, for example due to wind peaks associated with rain events, could create rapidly 278 changing conditions of vertical advection from deeper waters. However, with no significant increase 279 in TMs concentrations being observed below the ML down to about 50 m (not shown), the enrichment observed in the ML after rainfalls could not be attributed to any mixing with deeper 280 281 water due to high wind. In consequence, stocks in the ML have been integrated over a constant depth range of 0-20 m for comparison, as Bressac et al. (2021). For Cu, Fe, Ni, Zn, stocks were 282 283 estimated both for the dissolved and particulate fractions in the SML and ML, for Co, Cd, Mo, Pb 284 and V for the dissolved fraction only in the ML and for both fractions in the SML and for Mn and 285 Ti only for the particulate fraction in the ML.

The partitioning coefficient between the particulate and dissolved phases (Kd
=[particulate]/[dissolved]) was used to investigate exchanges between dissolved and particulate
pools of TMs.

## 289 **3. Results**

## 290 3.1 General conditions

291 The general meteorological conditions during the cruise indicated that the ION and FAST stations 292 were highly affected by cloudy weather conditions. During these periods, 2 significant rains 293 occurred on the R/V position and have been collected: The first rain (hereafter Rain ION) was collected during the 4-day ION station in the Ionian Sea in the early morning of 29/05 at 03:08 294 295 UTC, and the second rain (hereafter Rain FAST) occurred during the 5-day "Fast action" station, (hereafter 'FAST') in the Algerian Basin during the night of 05/06 at 00:36 UTC (Table 1). The two 296 297 rain samples coincide with peaks in relative humidity and wind speed, and minima in air 298 temperature (not shown).

#### 299 Table 1: Information regarding the two rains collected during the PEACETIME cruise.

Sample	Sampling time	Station name (dates) and rain location	Estimated total precipitation
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Rain ION	29 May 2017, 03:08-04:00 (UTC) 05:08–06:00 (local time)	ION (25–29 May) 35.36°N, 19.92°E	3.5 ±1.2 mm	
Rain FAST	05 June 2017, 00:36-01:04 (UTC) 02:36–03:04 (local time)	FAST (2–7 June) 37.94°N, 2.91°E	6.0 ±1.5 mm	

300

## 301 *3.1.1. Rain ION*

302 The ERA 5 data reanalysis shows 2 periods of precipitation in the surrounding of the ship's position, i.e. in the morning and evening of June 26 (not shown) and in the night between June 28 and 29, in 303 agreement with on-board visual observations. The rain event collected at ION was the product of a 304 large cloud system, covering an area of about 90 000 km<sup>2</sup> around the R/V position, spreading over 305 the Ionian and Aegean Seas (Fig. 1). No radar measurements being available for this area, the 306 307 accumulated rate was estimated from ERA 5 data reanalysis on the grid-point corresponding to the ION station and was 3.5  $\pm$ 1.2 mm around the R/V position ( $\pm$ 0.25°). The wash-out of the 308 atmospheric particles was revealed by the decrease in aerosol number concentrations monitored on-309 310 board from about 1900 to 300 part.cm<sup>-3</sup> (supplementary material Fig. S1). Air mass back-trajectories showed that the scavenged air masses came from Greece both in the marine boundary layer and in 311 312 the free troposphere (Fig. S1). The satellite observations showed low aerosol optical thickness during this period (not shown), meaning low amounts of aerosols in the atmospheric column. No 313 314 significant European pollution influence was monitored by on-board measurements during this 315 event, with major gas mixing ratio and aerosol concentrations in the average values of the cruise 316 (Fig. S1) and typical of clean atmospheric concentrations, i.e. under detection limit for NOx, 1.2 ppb for SO<sub>2</sub>, 51 ppb for O<sub>3</sub>, 80 ppb for CO and 3000 part.cm<sup>-3</sup>. On this basis, this wet event was 317 318 representative of a Mediterranean background marine rain event.







319

Figure 1: Total precipitation (mm) between May 28 at 20:00 UTC and 29 at 10:00 UTC from ERA5
ECMWF reanalysis. The red circle indicates the R/V position.

## 322 3.1.2. Rain FAST

323 As detailed in Guieu et al. (2020), the FAST station position was decided on the basis of regional model forecast runs and satellite observations, in the purpose to catch a wet dust deposition event. 324 Significant dust emissions were observed from NASCube (http://nascube.univ-lille1.fr/, Gonzales 325 326 and Briottet, 2017) over North Africa from the night of 30-31 May, then new dust emissions in the 327 night from 3 to 4 June in Algeria and southern Morocco associated with a northward atmospheric 328 flux. On 30 May, the SEVIRI AOD satellite product (https://www.icare.univ-lille.fr/data-329 access/browse-images/geostationary-satellites/, Thieuleux et al., 2005) confirmed the presence of 330 atmospheric dust in a cloudy air mass over the western part of the Mediterranean, and from 2 June the export of a dust plume from North Africa south of the Balearic Islands with high AOD (>0.8) 331 332 on the Alboran Sea was observed (Fig. S2). The dust plume was transported to the NE up to Sardinia 333 on June 4, with AOD<0.5 in all the area and clear sky with low AOD was left west of 4°E on June 334 5.

On-board lidar measurements (Fig. 2 a,b,c) showed that the aerosol plume was present over the ship position from 2 June at 21:00 UTC until the rain event, and corresponded to a dust aerosol layer well highlighted by the high depolarization. The dust plume was concentrated between 3 and 4 km





338 at the beginning of the station, then expanded to the end of the day on June 3 down to the marine 339 boundary layer (about 500 m amsl). The mass integrated contents of dust aerosols derived from the 340 profiles of aerosol extinction ranged from a minimum of  $0.18 \pm 0.005$  g m<sup>-2</sup> just before the rain to a

341 maximum of 0.24  $\pm$ 0.009 g m<sup>-2</sup>, where standard deviations indicate the temporal variability (1

342 sigma).



343





344 Figure 2: On-board lidar-derived a) Apparent backscatter coefficient (ABC), (b) Temporal evolution 345 (in Local Time) of the lidar-derived volume depolarization ratio (VDR) where the dust plume is 346 highlighted for values higher than ~1.7 (yellow to red colors) and the rain by values higher than 3 347 (indicated by the arrow), and c) vertical profiles of the aerosol extinction coefficient (AEC) in cloud 348 free condition, integrated over 3 periods along the dust plume event, noted A, B and C in figure a. The 349 grey shade represents the root mean square (rms) variability along the time of the measurement. The 350 dust layer is highlighted on the profiles. The mean aerosol optical thickness is given in the boxed legend 351 with its temporal variability (1 sigma). The location of the marine boundary layer (MBL) is also 352 pointed.

353 Rainfalls were observed by weather radar images in the neighbouring area of the R/V from 3 June 354 at 7:00 UTC. The rainfalls recorded around FAST station were associated with 2 periods of rain: the 03/06 from 7:00 to 14:00 UTC, and from the 04/06 at 16:00 UTC to 05/06 at 06:00 UTC. For 355 this latter, a rain front (100 000 km<sup>2</sup>), moving eastward from Spain and North Africa regions, 356 reached the FAST station the night between the 4 and 5 June (Fig. 3). Wet deposition between the 357 358 4 and early 5 June in the FAST station area were confirmed by radar imagery, showing several other 359 rain spots around the R/V position before and after the rain sampling (Fig. 3). Continuous on-board lidar measurements confirmed the below-cloud deposition during the rain event of early 5 June (Fig. 360 2b). Rain FAST was a wet deposition event occurring at the end of an episode of transport of 361 Saharan dust, whereas precipitation of the 3 June occurred during the maximum of the dust plume 362 363 (Fig. 2b and S2). The surface concentrations of gas and particles, measured on-board, suggest no 364 clear dust or anthropogenic influence in the atmospheric boundary layer during the period of wet deposition, in agreement with back trajectories of low altitude air masses (Fig S2.), presuming no 365 366 local mixing between dust and anthropogenic particles into rain samples. The total precipitation estimated from radar rainfall estimates yield an accumulated precipitation of  $6.0 \pm 1.5$  mm ( $\pm 25$  km 367 around), in agreement with ECMWF reanalysis ERA5 (Fig. S2) for the wet deposition on the night 368 of 4-5 June (5.7  $\pm$ 1.4 mm in the grid-point spanning the R/V position, i.e.  $\pm$ 0.25° around). 369







370

371 Figure 3: Rain rates (mm/h) during the night between the 4 and 5 June, when Rain Fast has been

collected on board, issued from European rain radar composites (OPERA programme) of June 5
between 00:00 and 02:45 UTC.

## 374 **3.2.** Chemical composition of rains

375 Dissolved and total concentrations of nutrients and TMs in the rain samples are presented in Table 376 2. Among all measured dissolved concentrations,  $NO_3^-$  was the most abundant nutrient, followed 377 by ammonium (Table 2). The nitrite concentration was under the limit of detection for the two rain 378 samples. Regarding TMs in rain, Fe and Zn presented the highest concentrations in rain samples 379 with the same order of magnitude (10 to 25 µg L<sup>-1</sup>). Co, Cd and Mo had the lowest concentrations





- 380 (<0.1  $\mu$ g L<sup>-1</sup> in both events), whereas the other TMs concentrations ranged between 0.1 and 10  $\mu$ g
- 381  $L^{-1}$  in both rains (Table 2). Concentrations of nutrients and the majority of TMs were higher in the
- 382 dust-rich rain, except Pb (similar concentrations in both rains) and Cr (3 times higher concentration
- 383 in Rain ION relative to Rain FAST).

384 Table 2: Dissolved and total concentrations of nutrients and TMs in the two rains collected during the

385 PEACETIME cruise in  $\mu g_{L}$ <sup>-1</sup> or  $ng_{L}$ <sup>-1</sup> and  $\mu mol_{L}$ <sup>-1</sup> or  $nmol_{L}$ <sup>-1</sup> in the parentheses (sd = standard deviation

## **386** from three replicates).

		Rain ION			Rain FAST					
		Dissolved		Total		Dissolved		Total		
		concentrations	±	sd	concentrations	±sd	concentrations	±sd	concentrations	±s
NO3 <sup>-</sup>	μg L <sup>-1</sup> (μm ol L <sup>-1</sup> )	1185 (19.1)	71	(1.1)			3694 (59.6	222 (3.6		
$\mathrm{NH_{4}^{+}}$	μg L <sup>-1</sup> (μm ol L <sup>-1</sup> )	366 (20.3)	11	(0.6)			654 (36.3	19 (1.1		
DIN	μg L <sup>-1</sup> (μm ol L <sup>-1</sup> )	552 (39.4)	82	(1.7)			1343 (96)	241 (17		
PO43-	μg L <sup>-1</sup> (nm ol L <sup>-1</sup> )	18.1 (189)	0.5	(6)			19.0 (200)	0.6 (6)		
DIP	μg L <sup>-1</sup> (nm ol L <sup>-1</sup> )	5.87 (189)	0.18	(6)			6.20 (200)	0.19 (6)		
DOP	$\mu gL^{\text{-}1}(\text{nm ol}L^{\text{-}1})$	8.63 (278)	1.94	(75)			4.91 (158)	1.56 (57)		
TP	μg L <sup>-1</sup> (nm ol L <sup>-1</sup> )	14.51 (468)	2.52	(81)	16.6 (536)	1.0 (33)	11.11 (358)	1.95 (63)	58.7 (1894)	3.5
DIN/DIP	,	(208)					(480)			
DOC	(µmol L <sup>-1</sup> )	(105.7)		(2.2)			(95.5)	(1.2)		
Al	$\mu gL^{\text{-l}}(\text{nm ol}L^{\text{-l}})$	13.0 (480)	0.8	(30)	14.6 (540)	0.9 (32)	23.4 (867)	0.7 (24)	440 (16308)	7
Cu	$\mu gL^{\text{-}1}(\text{nm ol}~L^{\text{-}1})$	0.71 (11.1)	0.02	(0.3)	0.73 (11.5)	0.02 (0.3)	1.15 (18.0)	0.04 (0.6)	1.63 (25.7)	0.06
Fe	$\mu g L^{-1}$ (nm of $L^{-1}$ )	15.1 (270)	0.4	(6)	17.9 (321)	0.6 (11)	19.2 (344)	0.1 (2)	231 (4140)	7
Mn	$\mu g  L^{\text{-}1}  (\text{nm ol}   L^{\text{-}1})$	0.55 (10.0)	0.02	(0.3)	0.60 (10.9)	0.02 (0.4)	3.17 (57.8)	0.07 (1.2)	5.26 (95.7)	0.12
Ni	$\mu g  L^{\text{-}1}  (\text{nm ol}   L^{\text{-}1})$	0.52 (8.8)	0.02	(0.3)	0.67 (11.4)	0.02 (0.4)	0.59 (10.1)	0.02 (0.4)	0.84 (14.3)	0.03
Ti	$\mu g  L^{\text{-}1}  (\text{nm ol}   L^{\text{-}1})$	0.48 (10.0)	0.04	(0.8)	0.65 (13.6)	0.48 (3.2)	0.22 (4.7)	0.01 (0.1)	33.36 (697)	0.51
V	$\mu g  L^{1}(\text{nm ol}   L^{1})$	0.37 (7.4)	0.01	(0.2)	0.38 (7.42)	0.01 (0.25)	1.37 (26.9)	0.03 (0.5)	2.02 (39.7)	0.04
Zn	$\mu g  L^{1}(\text{nm ol}   L^{1})$	24.8 (379)	0.8	(12)	25.3 (387)	0.8 (12)	22.7 (347)	0.6 (8)	26.3 (402)	0.7
Cđ	$ng L^{-1}(pm ol L^{-1})$	12.9 (115)	6.4	(57)	13.1 (117)	6.3 (56)	20.2 (180)	10.3 (92)	23.7 (210)	6.8
Co	$ng L^{-1} (pm ol L^{-1})$	44 (749)	13	(229)	47.4 (804)	14.5 (246)	82 (1386)	20 (347)	157 (2661)	28
Cr	$ng L^{-1}(pm ol L^{-1})$	241 (4636)	16	(300)	628 (12079)	5 (95)	79 (1522)	14 (260)	443 (8514)	43
Mo	$ng L^{-1}(pm ol L^{-1})$	28 (288)	10	(106)	4.1 (43)	1.4 (14)	82 (855)	11 (113)	92.1 (960)	16.2
Pb	$ng L^{-1}(pm ol L^{-1})$	170 (822)	11	(54)	175.1 (845)	1.4 (7)	166 (801)	9 (41)	604 (2917)	19
	NO3 NH4 <sup>+</sup> DIN PO4 <sup>3-</sup> DIP DOP TP DIN/DIP DOC Al Cu Fe Mn Ni Ti V Zn Cd Co Cr Mo Pb	NO <sub>3</sub> μgL <sup>-1</sup> (µmd L <sup>-1</sup> )           NH4*         μgL <sup>-1</sup> (µmd L <sup>-1</sup> )           DIN         μgL <sup>-1</sup> (nmd L <sup>-1</sup> )           PQ4 <sup>3-</sup> µgL <sup>-1</sup> (nmd L <sup>-1</sup> )           DIP         µgL <sup>-1</sup> (nmd L <sup>-1</sup> )           DOP         µgL <sup>-1</sup> (nmd L <sup>-1</sup> )           DP         µgL <sup>-1</sup> (nmd L <sup>-1</sup> )           DIP         µgL <sup>-1</sup> (nmd L <sup>-1</sup> )           DIV/DIP         DOC         (µmol L <sup>-1</sup> )           Al         µgL <sup>-1</sup> (nmd L <sup>-1</sup> )           Cu         µgL <sup>-1</sup> (nmd L <sup>-1</sup> )           Mn         µgL <sup>-1</sup> (nmd L <sup>-1</sup> )           Ni         µgL <sup>-1</sup> (nmd L <sup>-1</sup> )           V         µgL <sup>-1</sup> (nmd L <sup>-1</sup> )           V         µgL <sup>-1</sup> (nmd L <sup>-1</sup> )           Cd         ngL <sup>-1</sup> (nmd L <sup>-1</sup> )           Cd         ngL <sup>-1</sup> (pmd L <sup>-1</sup> )           Cr         ngL <sup>-1</sup> (pmd L <sup>-1</sup> )           Cr         ngL <sup>-1</sup> (pmd L <sup>-1</sup> )           Pb         ngL <sup>-1</sup> (pmd L <sup>-1</sup> )	$\begin{tabular}{ c c c c c c c } \hline \hline & $	$\begin{tabular}{ c c c c c c c } \hline \hline Dissolved \\ \hline \hline concentrations & \pm \\ \hline \hline concentrations & \hline $	$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	$\begin{tabular}{ c c c c c c c c c c c c c c c c c c c$	$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$

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## 388 3.3. Marine concentrations and stocks

All the TMs had significantly higher concentrations in the ML compared to deep water masses, in 389 agreement with a stratified profile associated with atmospheric input. The particulate and dissolved 390 trace metal concentrations within the ML (0-20 m) and the SML are displayed in Fig. 4. 391 392 Concentrations were of the same order of magnitude in the two studied stations, except for the 393 particulate phase in the SML where the concentrations of Cu and Co were significantly lower at 394 ION station. The TMs were mainly in dissolved forms in the ML, except for Fe, whose dissolved and particulate concentrations were in the same order of magnitude. On the contrary, the particulate 395 396 phase contribution dominated for TMs in the SML, in particular at the ION station. At both stations, 397 the highest TMs concentrations in the surface seawater were found for Mo in the dissolved fraction,





398 in agreement with the abundance of dissolved Mo in seawater (Smedley and Kinniburgh, 2017), and Fe in the particulate fraction. All the particulate and dissolved TMs concentrations measured 399 400 during the cruise were representative for the Mediterranean Sea (Sherrell and Boyle, 1988; Saager et al., 1993; Morley et al., 1997; Yoon et al., 1999; Wuttig et al., 2013; Baconnais et al., 2019; 401 402 Migon et al., 2020). Zn presented the largest range of concentrations within the ML both in the 403 particulate and dissolved phases, due to some high concentrations. However, the concentrations 404 stayed in the typical range of values found in the Mediterranean Sea (Bethoux et al., 1990, Yoon et al., 1999). In the SML, the concentrations were lower than in the ML and Pb dominated both in 405 406 dissolved and particulate phases. Tovar-Sanchez et al. (2020) showed that the TMs concentrations 407 in the SML during the PEACETIME campaign were generally lower than those previously 408 measured in the Mediterranean Sea, except in the particulate phase during the FAST station after 409 dust deposition.



410

Figure 4: Boxplots of dissolved (left panels) and particulate (right panels) marine concentrations (pM) for the different TMs within the ML (upper panels) and the SML (lower panels) at ION and FAST stations. In the box plots, the box indicates the interquartile range, i.e. the 25th and the 75 th percentile, and the line within the box marks the median. The whiskers indicate the quartiles ±1.5 times the interquartile range. Points above and below the whiskers indicate outliers outside the 10 th and 90th percentile.





## 417 4. Discussion

## 418 4.1. Composition of rain collected over the remote Med Sea

#### 419 *4.1.1. Concentrations*

420 Regarding nutrients, nitrogen species concentrations in rain samples were in good agreement with those reported in Mediterranean rain samples, ranging from 1130 to 5100 µg L<sup>-1</sup> for NO3<sup>-</sup> and 421 between 207 and 1200 µg L<sup>-1</sup> for NH<sub>4</sub><sup>+</sup> (Loye-Pilot et al., 1990; Avila et al., 1997; Al Momani et 422 al., 1998; Herut et al., 1999; Violaki et al., 2010; Izquieta-Rojano et al., 2016; Nehir and Kocak, 423 424 2018). The FAST rain concentrations were in the average values, whereas the ION rain was in the 425 low range, confirming a background signature. The rainwater samples presented a large dominance of N in comparison to P, as observed from the N/P ratio derived from DIN/DIP (Table 2) ranging 426 427 from 208 at ION to 480 at FAST. Previous observations showed a predominance of N relative to P in the atmospheric bulk deposition over the Mediterranean coast, with ratio higher than Redfield 428 429 ratio (Markaki et al., 2010, Desboeufs et al., 2018). The highest ratio observed reached 1200 (in 430 case of DIN/TDP), but were on average around 100 in bulk deposition. The highest ratio could be 431 linked to a washout effect of the gaseous N species (as NOx and NH<sub>3</sub>) by rain (Ochoa-Hueso et al., 432 2011). At the two stations, no high NOx concentrations were observed in the boundary layer before 433 wet deposition. The presence of nitrate and ammonium in the background aerosols has been emphasized during recent campaigns in the remote Mediterranean atmosphere (e.g. Mallet et al., 434 435 2019). To our knowledge, no data are available on both P and N concentrations in Mediterranean 436 aerosols. The lowest concentrations of P relative to N in aerosol particles in Mediterranean have 437 been observed during the cruise (Fu et al., in prep.). The TDP concentrations were consistent with the average value of 8.4 µg L<sup>-1</sup> measured in African dust rain samples collected in Spain over the 438 1996-2008 period (Izquierdo et al., 2012). Inorganic phosphorus predominated in the dust-rich rain, 439 440 whereas organic P was dominant in the background rain as the contribution of DOP to the TDP was 441 60% and 44% in Rain ION and Rain FAST, respectively. The DOP/TDP ratio presents a very large 442 range in Mediterranean rains, spanning from 6% in Spanish dusty rain samples (Izquierdo et al., 443 2012) to 75-92% in rains from Crete Islands (Violaki et al., 2018). A reason for this wide range 444 could be that Mediterranean European aerosols, as opposed to Saharan dust particles, are dominated 445 by organic phosphorus compounds associated with bacteria (Longo et al., 2014).







446

# Figure 5: Comparison of dissolved (D) and total (T) TMs concentrations along with data from 14 former studies carried out in the eastern and western Mediterranean Sea.

The dissolved and total TMs concentrations in the PEACETIME rains were lower than those 449 450 reported in coastal areas (eastern Basin: Özsoy and Örnektekin, 2009; Al-Momani et al., 1998; 451 Kanellopoulou et al, 2001 and western Basin: Guieu et al., 1997; Guerzoni et al., 1999b; Chester et al., 1997; Losno, 1989; Frau et al., 1996) (Fig. 5), notably for the background Rain ION. This 452 suggests the probable effect of both local anthropogenic influence at coastal sites due to higher 453 454 aerosol concentrations in comparison to remote Mediterranean (Fu et al., in prep.) and due to the 455 reduction of anthropogenic emission for some elements since most of the referenced works on 456 coastal rainwaters date from the late 1990s. This is particularly true for Cd and Pb whose emissions have strongly decreased over the last decades, notably due to removal of lead in gasoline and 457 reduction of coal combustion (Pacyna et al., 2007). This has resulted in a clear decrease in the 458 particulate concentrations of these metals in the Mediterranean atmosphere (Migon et al., 2008), 459





460 consistent with the fact that concentrations in the PEACETIME rains are one to two orders of 461 magnitude lower than reported in the literature. For these metals, the discrepancy is also observed 462 between the concentrations in our open-sea rain samples and the concentrations measured in three 463 rains collected at sea in April 1981 (Dulac, 1986), confirming the large decrease of concentrations 464 could be related with the decrease of emission. Thus, our results show the former literature cannot 465 be used as current reference about coastal rain composition due to recent environmental mitigation 466 on metals emissions.

#### 467 4.1.2. Enrichment factor

468 EF and solubility values of TMs and P observed during the two rain events were very contrasted (Fig. 6). In Rain ION, almost all elements were significantly enriched relative to earth crust (EF 469 >10, and up to  $\sim 10^3$  for Cd and Zn), whereas in Rain FAST, only Zn (73), Cd (48) and Mo (15) 470 471 were slightly enriched. Only Ti, Fe, and Mn did not present a significant enrichment (EF < 10) in Rain ION, in agreement with previous studies in the Mediterranean environment showing that these 472 473 metals are mainly associated with mineral dust in atmospheric deposition (e.g., Guieu et al., 2010; Desboeufs et al., 2018). Both rains, EF of Zn was in average five times higher than EF found in the 474 475 rains previously studied in the Mediterranean region (Özsoy and Örnektekin, 2009; Al-Momani et 476 al., 1998; Losno, 1989). However, extremely high enrichments of Zn in rainwater have also been 477 reported by Frau et al. (1996) with a geometric mean EF of about 6500 in both crust-rich and crustpoor rains from two sites in southern Sardinia. Fu et al. (2017) also reported EF higher than 1000 478 for Zn in atmospheric bulk deposition in Lampedusa Island. The anthropogenic origin of particulate 479 TMs and P have been reported by several studies on atmospheric deposition in the western 480 Mediterranean (e.g., Guieu et al., 2010; Sandroni and Migon, 2002; Desboeufs et al., 2018). For 481 482 example, Desboeufs et al. (2018) showed that there is a large contribution of anthropogenic 483 combustion sources in P, Cr, V and Zn background deposition fluxes. Aerosol composition 484 monitoring over Mediterranean coastal area showed the role of land-based sources and ship traffic sources on TMs contents (Bove et al., 2016; Becagli et al., 2017). However, all these sampling sites 485 486 were located in coastal areas, where it was difficult to discriminate the potential local influences. 487 Here, even if the on-board atmospheric gas and particle measurements did not show a specific 488 anthropogenic influence during the period of Rain ION, the particles scavenged by this rain 489 presented a clear anthropogenic signature for all TMs except Ti, Fe and Mn; however for Fe and 490 Mn, an influence of non-crustal sources in Rain ION is visible through a clear increase in the EF 491 values compared to FAST (Fig. 6). This means that even over remote Med Sea, the chemical





492 composition of background aerosol particles is likely continuously impacted by anthropogenic493 sources.

494 Moreover, the EF values of TMs for Rain FAST were significantly lower than for Rain ION (Fig. 495 6) but similar to Saharan rains (Guerzoni et al., 1999b; Özsoy and Örnektekin, 2009). The 496 comparison between dust-rich and background rains generally reveals a net difference of concentrations (at least higher by a factor 3 in dust-rich), notably for Al, Fe, Mn and Cr (Guerzoni 497 et al., 1999b; Özsoy and Örnektekin, 2009). Such contrast was indeed observed for Al, Fe and Mn 498 499 in the PEACETIME rains (Fig. 5), but also for Cu and Pb. The combination of higher concentrations 500 and EF values <10 found in Rain FAST confirm that the dust contribution was important on 501 deposition fluxes of many TMs.

502 *4.1.3. Solubility* 

503 The solubility values were also larger in Rain ION than in the dusty Rain FAST, except for Mo for which the difference between both rain samples is not significant (Fig. 6). For Rain ION, TMs and 504 505 P presented solubility higher than 78%, except for Cr (38%). In Rain FAST, solubility values <10% 506 were observed for Al and Fe, more than 10 times lower than in Rain ION. For the other TMs, the highest difference in solubility was observed for Pb whose solubility decreased from 97% in Rain 507 508 ION to 27% in Rain FAST. In a review on TMs solubility in Mediterranean rainwaters collected in coastal areas, Desboeufs (2021) emphasize the large range of solubility for all the TMs: Fe (0.8-509 510 41%), Cr (6-80%), Pb (5-90%), Ni (22-93%), Mn (16-95%), Cu (22-96%), Zn (14-99%), V (35-511 99%) and Cd (72-99%). The solubility ranges found in this study were generally consistent with those reviewed by Desboeufs (2021). In particular, the Mn solubility values in FAST (60%) and 512 513 ION (92%) rains are close to those reported by Dulac (1986) from a dust-rich (57%) and an 514 anthropogenic (83%) rain collected at sea in the Ligurian Sea and west of Sardinia in April 1981, respectively. Only Fe solubility (84%) found in Rain ION was higher than the average values 515 516 previously reported. In the Rain FAST, Fe solubility was 8%, this is 10 times lower than average Fe solubility in 10 dust-rich rains collected on the southeastern coast of Sardinia by Guerzoni et al. 517 518 (1999b), but consistent with Saharan dust wet deposition collected in the Atlantic Ocean (Chance et al. 2015; Powell et al., 2015; Baker et al., 2017). 519







FAST ION

Figure 6: Enrichment Factors (EF, upper panel) and solubility (%, bottom panel) of phosphorus (P) and TMs
ordered by increasing EF in the two rainwater samples.

523 Few studies compared TMs solubility between dust-rich and background rains in the Mediterranean. 524 In Sardinia, Guerzoni et al. (1999b) observed an increase in solubility values from dust-rich to 525 background rains for Al, Cr, Fe and Pb (and hardly for Cd), and reported an inverse relationship between the particle concentration and solubility of Al, Fe, Pb and Cd. Similarly, Theodosi et al. 526 (2010) showed a decrease in TMs solubility with the increase in dust load in rains collected on Crete 527 528 Island. In those two studies, this decrease was dependent on the considered metal, with Pb 529 presenting the highest decrease in solubility. The decrease in solubility from background to dust-530 rich rains was also observed for P in Spain by Izquierdo et al. (2012), with values of solubility 531 decreasing from 25% to 7%, and for Mn in offshore rains as mentioned above (Dulac, 1986). Metal partitioning in rainwater can be influenced by a number of parameters, such as pH, presence of 532 533 dissolved organic complexing ligands, in-cloud processes, particle origin and load (Desboeufs et 534 al., 1999; Bonnet and Guieu, 2004; Paris and Desboeufs, 2013; Heimburger et al., 2013). However, 535 the particulate desert dust load, reflecting dust vs anthropogenic signature, is the main control of TMs solubility in the Mediterranean rainwater (Özsoy and Örnektekin, 2009; Theodosi et al., 2010). 536 537 A much lower solubility of TMs in Rain FAST than in Rain ION (except for Mo) is consistent with



539



the EFs in Rain FAST (Fig. 6), although no correlation between solubility and EF values could be

540 despite a >20 times higher EF in Rain ION. As Mo solubility is seldom studied in the literature, we

observed. The case of Mo is unique, since its solubility was comparable in Rains ION and FAST

541 could not conclude on the reason for this particular outcome. It is interesting to note that despite the

desert signature, the majority of metals have solubility greater than 50% in rain FAST.

#### 543 4.2. Atmospheric wet deposition as a source of TMs to the surface seawater

#### 544 4.2.1. Atmospheric fluxes

As mentioned before, the two collected rains were part of large rain systems, associated with patchy 545 546 rainfalls that lasted several hours or days (section 3.1). This spatio-temporal variability led to 547 heterogeneity in both rainwater concentrations and accumulated precipitation across the studied 548 region. Such spatial variability has been observed by Chance et al. (2015) in the Atlantic Ocean. 549 Moreover, even weak lateral advection can transfer surface water impacted by intense precipitation in the vicinity of the vessel. On this basis, the spatial extrapolation of wet deposition fluxes seems 550 subject to a large uncertainty when the rain samples are not collected across the rain area (Chance 551 et al., 2015). To best counteract this effect, spatial variability was taken into account to quantify the 552 553 total precipitation i.e.  $3.5 \pm 1.2$  mm for rain ION and  $6.0 \pm 1.5$  mm for rain FAST (see 3.1) in order 554 to quantify the wet deposition fluxes.

From the total (dissolved + particulate) Al concentration measured in the rain FAST sample, we 555 556 estimated the wet mineral dust deposition flux at  $65 \pm 18$  mg m<sup>-2</sup>, assuming 7.1% Al in dust (Guieu et al., 2002). The vertical distribution of dust particles (Fig. 2b) and the absence of high Al 557 558 concentrations close to the sea surface (Fu et al., in prep.) indicate that dust dry deposition can be neglected. Based on the increase in total Al in the upper 20 m of the column water following the 559 deposition events, Bressac et al. (2021) derived an average dust deposition flux of  $\sim$ 55 mg m<sup>-2</sup> at 560 FAST station, comparable to our estimate. Although low compared to deposition fluxes reported in 561 562 the western Mediterranean (Bergametti et al., 1989; Loÿe-Pilot and Martin, 1996; Ternon et al., 563 2010), such flux values are among the most intense weekly dust deposition fluxes recorded more 564 recently in Corsica between 2011 and 2013 and is equivalent to the mean weekly flux observed in Majorca Island during the same period (Vincent et al., 2016). The aerosol columnar during the dust 565 event being estimated between 0.18 and 0.24 g  $m^{-2}$  (see 3.1.), the expected maximum values of 566 atmospheric dust flux could be in this range. The comparison with the estimated flux indicates that 567 568 the atmospheric column was probably not totally washed-out by the short rain event. Indeed, Fig. 569 2b shows that a significant depolarization was observed immediately after the rain ended on the





- 570 ship, before atmospheric advection could have brought dusty air possibly not affected by rain.
- 571 Satellite products (Fig. S2) confirm that on 5 June, the dusty air mass was transported farther to the
- 572 north-east from the station where it was replaced by clear air.



573

Figure 7: Dissolved (upper panels) and particulate (lower panels) wet deposition fluxes (μmol m<sup>-2</sup>) for
the different TMs estimated from the two rains sampled on board, considering the standard deviation
on the TMs concentrations and the spatial variability of total precipitation over the area of sampling
(Rain ION in blue and Rain FAST in red).

578 The atmospheric dissolved and particulate wet deposition fluxes of TMs, derived from the chemical 579 composition of rain samples and total precipitation, are presented in Fig. 7. Co, Mo and Cd presented the lowest fluxes in the two rainfalls. Zn and Fe fluxes were in the same order of 580 581 magnitude and were the highest dissolved fluxes compared to the other TMs in the two rains. The 582 comparison showed that almost all the dissolved TMs fluxes were higher in the dusty rain, except 583 Cr and Ti. For the particulate phase, the fluxes were mainly increased by dust deposition for Co, 584 Fe, Mn, Pb, Ti and V. Our results emphasized that dust deposition, even here in case of a moderate deposition flux, enabled higher atmospheric inputs of TMs than a low perturbed-anthropogenic 585 586 background rain. This is notably the case for dissolved Cd, Cu, Mn, V and Zn and for particulate 587 Fe and Ti with more than one order of magnitude fluxes difference between the two rains. The orders of magnitude found in this study could be used as "typical atmospheric fluxes" to estimate 588 589 atmospheric inputs of TMs by a rain event to the western Med Sea. We must keep in mind, however, 590 that annual and long-term deposition fluxes of dust-related elements (e.g. Ternon et al., 2010), but





- also nitrogen (e.g., Richon et al., 2018b), in the Med Sea, are dominated by a few atypical intense
- 592 deposition events when they occurred.
- 593 4.2.2. Comparison between TMs wet deposition inputs and marine stocks at ION and FAST stations

Marine sampling sequences carried out before and after rains were, to our knowledge, the first direct 594 595 observations to trace the fate of atmospheric metals and nutrients in the water column after wet deposition event. The time chart of the sampling of rain and column water (surface microlayer, 596 subsurface seawater and mixed layer) is presented in Fig. 8. The impact of the two wet depositions 597 on nutrients stocks in the Mediterranean surface waters is discussed in van Wanbeke et al. (2020) 598 599 and Pulido-Villena et al. (2021). Both nitrate and DIP increased in the ML following the rains. Although the closure of the N and P budgets had to take into account post-deposition processes as 600 new nutrient transfer through the microbial food web (uptake, remineralisation, and 601 602 adsorption/desorption processes on sinking particles), it was shown that wet deposition was a significant source of nutrients for ML during the cruise. We focus here on the role of TMs deposition 603 604 as a source of metals to the column water. To do so, we estimated the potential enrichment of SML 605 and ML from the rain by calculating the difference (delta) in TMs stocks before and after rains.



Figure 8: Sampling chronology during the ION and FAST stations for SML, SSW and ML. The blue periods correspond to rainfall in the station area (after ERA 5 reanalysis and radar imagery, see section 3.1). Samplings were performed 4 days and 2 days before and 2 h after Rain ION, and at a higher frequency at the FAST station: 57, 37 and 7.5 hours before and 4.5, 12, 24 hours after Rain FAST. SML and SSW samples could not be collected immediately before and after the rains because of bad weather conditions, and were collected 3 and 4 days before Rain ION, and 57 and 20 h before and 30 h after Rain FAST.

At ION, no SML sampling was done after rain, preventing the study of the rain effect. For ML, the large variability in total and dissolved stocks between the two casts ML25 and ML27 before the rain makes the establishment of a background concentration levels before rain difficult. ML27 was used as initial conditions since it is the closest sampling from post-rain measurements (ML29). As





618 mentioned previously, dust rain deposition over the FAST station area started on 3 June. Bressac et 619 al. (2021) showed that the dust signature, traced by changes in Al and Fe stocks in the ML, was 620 already visible from the ML03 sampling. We defined the enrichment of seawater layers as the 621 difference between the maximum stocks after rains (from SML 04 or 06 and ML05+4) and the 622 initial seawater stocks (SML02 and ML02).

At ION, only particulate Cu (+27%) and Zn (+44%) stocks increased in the ML after the rain. Even if the dissolved forms of Cu and Zn predominated in the ML, this increase was accompanied by increasing Kd values, i.e. in the particulate/dissolved partitioning (0.07 vs 0.12 for Cu and 0.14 vs 0.2 for Zn). This was also the case for Fe (Kd increased from 2.6 to 4.3), although no significant

difference (<5%) was observed on the particulate Fe stock. The Kd values in the ION rain sample</li>being higher than in the marine stock before the rain, that suggests the wet deposition at ION is

629 mainly an additional source of particulate TMs.

At FAST, the ML stocks increased in the particulate phase for Fe (+61%), Mn (+15%) and Zn 630 631 (+9%) and in the dissolved phase for Cu (+9%), Fe (+46%), Pb (+8%) and Zn (+15%) (Fig. 9). In 632 addition of marine inventories, the particulate TMs input by rain was also observed on Kd values 633 and total X/Al in the ML. For example, Kd(Fe) increased from 0.14 to 0.17 in ML and its Kd was 634 0.25 in the rain. Even for Ni for which no change in stock could be evidenced, Kd(Ni) decreased 635 from 0.1 to 0.07 and its Kd in the rain was 0.006. For Mn/Al which fell from 0.27 before the rain to 0.008 after the rain (ML05+4), in accordance with the rain ratio (0.004). In the SML, the dissolved 636 and particulate stocks increased following rains for all TMs, from a factor 1.5 (Mo) to 10 (Fe) for 637 638 the dissolved phase and from a factor 1.6 (Ni) to 67 (Fe) for the particulate phase (Fig. 9).







Figure 9: Comparison between TMs wet deposition fluxes (in green) and TMs marine stock delta
(before and after the rain) in the SML (in blue) and in the ML (in red) at FAST. Dissolved = upper
panels and particulate = lower panels. Marine stocks increase are expressed in absolute values (Cd,
Co and Pb stocks in nmol m<sup>-2</sup>, and the other TMs in µmol m<sup>-2</sup>) and in relative values (%). N.E.: not
enhanced (increase <5%).</li>

645 The comparison between the observed enhancements in the SML stocks and the rain inputs at FAST 646 (Fig. 9) indicates that the atmospheric fluxes support all observed deltas. Indeed, the atmospheric 647 particulate and dissolved fluxes of TMs were 2 to 4 orders of magnitude higher than the mean stocks 648 present in the SML, except Mo which was in the same order of magnitude. In the ML, the magnitude of atmospheric particulate inputs was higher or similar to the particulate marine delta of Fe, Mn and 649 650 Zn. For Cu, Fe, Pb and Zn, the increase in dissolved stocks within the ML was 2 to 10 times higher 651 than what could provide the atmospheric inputs. As described in Guieu et al. (2020), marine 652 dynamical conditions at FAST were favorable to observe any change in the water masses strictly 653 attributed to external inputs coming from the atmosphere on a short time scale. However, a 654 cumulative effect of previous and surrounding wet deposition events could explain this difference 655 between atmospheric inputs from rain FAST and increase of marine dissolved stocks. We cannot 656 exclude the possibility of lateral transport of metals from surrounding waters being enriched by the 657 rain events of June 3 for example, as revealed by the increase in the 0-20 m Fe and Al inventories 658 (Bressac et al., 2021). Another hypothesis to explain that higher stock increases of metals in the ML 659 than the one derived from atmospheric deposition is related to post-deposition processes. Indeed, 660 once deposited, the atmospheric particulate fraction could still be partly solubilized in seawater, as 661 the solubilisation of TMs (e.g. Fe) could occur over several hours or days (Wagener et al., 2008; 662 Wuttig et al., 2013; Desboeufs et al., 2014). This could lead to an underestimate of dissolved TMs 663 atmospheric inputs. Moreover, the time lag between the rain and the first SML sampling (1 day) does not allow us to conclude on the role played by SML as a "trap" of the added dissolved metals 664 by rain. However, results showed clearly the increase of dissolved TMs in the SML even 24 h after 665 666 the rain (Fig. 9). Even if this increase could be due to dissolution processes (Tovar-Sanchez et al., 2020), we cannot exclude that the residence time of dissolved atmospheric TMs in the SML was 667 668 sufficient to mask the atmospheric inputs in the ML05+4 sample. It is also known that dissolved 669 concentrations in the ML are subject to various biological processes such as phytoplankton uptake 670 (Morel et al., 2003). The comparison between ML05+4 and ML05+12 samples performed after the rain shows that the dissolved and particulate stocks decreased quickly for all the TMs (not shown), 671 672 in agreement with the predominance of removal processes (sedimentation, biological transfer, 673 adsorption) on these stocks. However, the rate of decrease depended on the TMs, showing that some





674 removal processes predominate over others depending on the metal. For example, the dissolved 675 metals decrease could correspond to scavenging onto particles, which is a common physical-676 chemical process occurring in the ocean for Fe (Wagener et al., 2010; Bressac and Guieu., 2013) or Co (Migon et al., 2020). Finally, our results show that dust wet deposition was a net source of all 677 678 the studied trace metals for SML both in the dissolved and particulate fraction. For ML, atmospheric 679 dust inputs were also a net source of particulate Fe, Mn, and Zn, and dissolved Cu, Fe, Pb and Zn. 680 Due to various marine post-deposition processes, it was more complicated to observe the effect of wet deposition on dissolved stocks, explaining why the SML and ML particulate stocks were more 681 682 impacted by rain than the dissolved stocks. On a timescale of hours, Fe inventory was the most 683 impacted by the dusty rain input, both in dissolved and particulate phase, confirming that the dust-684 rich rains are a net source of Fe to the surface Mediterranean Sea (Bonnet and Guieu, 2006, Bressac 685 et al., 2021).

686 4.2.3. Comparison between TMs wet atmospheric inputs and marine stocks at the scale Western
687 and Central Mediterranean Sea

As observed from dissolved TMs stocks measured before and after the rains, a large part of 688 689 uncertainties in the data analysis results from various removal processes of TMs after wet 690 deposition, which could have time resolution shorter than the sampling step. In order to limit the 691 effect on these potential processes in data analysis, here we further study the role of wet deposition 692 by comparing atmospheric dissolved fluxes to marine dissolved stocks by using TMs profiles in the 693 ML at all 13 marine stations, i.e. 22 ML samplings, throughout the whole cruise (Fig. 10). Indeed, 694 considering that the collected rains were originating from large rain systems covering more than 50 000 km<sup>2</sup> around the sampling zone and were typical of Mediterranean wet deposition, we 695 hypothesized that they could have occurred in any of the explored areas during the cruise. 696 Exceptional intense dust deposition events have been recorded in the Mediterranean, reaching 20 g 697  $m^{-2}$  (Bonnet et al., 2006). Sporadic and intense wet dust deposition higher than 1 g m<sup>-2</sup> are regularly 698 699 observed in the spring in the western Mediterranean basin (e.g., Vincent et al., 2016). At the 700 beginning of the cruise, an intense wet dust deposition event (not collected) occurred over the South of Sardinia and over the Tyrrhenian Sea with fluxes reaching about 9 g m<sup>-2</sup> (Bressac et al., 2021). 701 In order to take into account the effect of such an event, we also estimated the atmospheric fluxes 702 703 of dissolved metals based on a 9 g  $m^{-2}$  wet dust deposition event considering solubility values found 704 in the rain FAST (Fig. 10). The metal solubility decreasing with increasing dust load (Theodosi et 705 al., 2010), this estimation constitutes probably a maximum value of the dissolved inputs of trace 706 metals by such a dust deposition event. In addition to removal processes, the impact of rain inputs





- 707 on TMs marine stocks is also controlled by MLD fluctuations that we ignored in the work described
- above by using a fixed depth for FAST an ION stations. However, the variability of this MLD (7-
- 709 21 m during the cruise, typical of Med thermal stratified period) could change the marine budgets
- 710 by a factor of 3. So we considered the measured MLD (Van Wambeke et al., 2020) for calculating
- 711 marine budgets of TMs at each station.



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Figure 10: Comparison of marine stocks in the ML at all the stations occupied during the PEACETIME cruise with atmospheric inputs estimated (1) from ION and FAST rains (Boxes) and (2) from an intense wet dust deposition event of 9 g m<sup>-2</sup> (blue dots). Cd, Co and Pb stocks are in nmol m<sup>-2</sup>, and the other TMs in µmol m<sup>-2</sup>. For Mn, marine stocks are derived from surface concentrations close to Corsica coasts (Wuttig et al., 2013: samples OUT at 0, 5 and 10 m) and in the Ionian Sea (Saager et al., 1993: Bannock basin at 0, 10, and 15 m), as no measurement is available from the PEACETIME cruise. Boxes and whiskers as in Fig. 4.

Applying to the whole transect, the atmospheric inputs, obtained from our rain composition, were
at least 100-fold smaller than the dissolved stocks in the mixed layer, except for Co, Fe and Zn. The
atmospheric inputs represented more than 30% of the dissolved Zn stocks and 10 to 18% for Fe.
This significant input of dissolved Fe is in agreement with our field observations in the ML. For
Co, the maximum atmospheric fluxes estimated during the cruise represented >10% of stocks. Here





725 the comparison is based only on dissolved TMs in rain water, yet as discussed previously, the solubilisation post-deposition of atmospheric particles in the water column could further enrich the 726 727 marine dissolved stocks. In the case of the intense dust deposition event, the dissolved inputs are of the same order of magnitude as marine stocks for Co, Fe, Mn, Pb and Zn. The enrichment in 728 729 dissolved Fe and Mn was previously observed by Wuttig et al. (2013) after artificial dust seeding in large mesocosms (simulating a wet deposition event of 10 g m<sup>-2</sup>). The surface seawater could be 730 731 significantly affected by the deposition of these dissolved elements in the case of wet dust deposition. The marine TMs concentrations measured during the cruise being typical of 732 733 Mediterranean surface seawater concentrations, we conclude from these comparisons that wet 734 deposition events, notably wet dust deposition events, prove to be an external source of dissolved TMs for the Mediterranean Sea during the period of thermal stratification. 735

# 736 **5.** Conclusions

737 This study provides both the dynamical properties and chemical characterization of two rainwaters collected in the open Mediterranean Sea, concurrently with TMs marine stocks in surface seawater. 738 739 Our results are the only recent report of concentrations, EF and solubility values for TMs in rain 740 samples collected in remote Med Sea. By highlighting the discrepancy between TMs concentrations 741 with the previous and coastal rain studies, this work demonstrates the need to provide a new and recent database on metal composition in Mediterranean rains in order to estimate the role of 742 743 atmospheric TMs deposition. We showed the representativeness of rain FAST as typical Saharan 744 dust wet deposition as well in its chemical composition as in its magnitude and extent, whereas Rain 745 ION is a typical low perturbed anthropogenic background rain of remote Med Sea. On this basis, 746 we suggest to use the chemical composition of PEACETIME rains as a new reference for the studies 747 of TMs on wet deposition in Med Sea.

Since atmospheric TMs have been identified as critical oligo-nutrients for marine biosphere, our study is the first *in situ* evidence that atmospheric wet deposition constitutes a significant external source for some of these elements to surface stratified Mediterranean seawater. Our results show that the original approach developed here is very relevant in this purpose and could be used in other part of the world where atmospheric deposition is suspected to impact marine biosphere, as HNLC areas.

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756 Data availability. Guieu et al., Biogeochemical dataset collected during the PEACETIME cruise. SEANOE.
757 https://doi.org/10.17882/75747 (2020). Atmospheric Data are accessible on http://www.obs-vlfr.fr/proof/php/PEACETIME/peacetime.php.

759 Author contributions. KD and FF designed the study and wrote the manuscript; FF, ST, JFD, Ch.G made 760 the on-board atmospheric measurements and sampling during the cruise; FF, ST and JD analysed the rain 761 samples; MB, ATS, and ARR made the marine TMs sampling and analyses; PF was the reference scientist 762 of PEGASUS, AF and FM managed all the technical preparation of atmospheric samplings, PC analysed the 763 lidar data; KD, FD and Ce.G designed the cruise strategy; KD and Ce.G coordinated the PEACETIME 764 project, FD coordinated the ChArMEx funding request, and near-real time and forecast survey of 765 atmospheric conditions during the cruise; all the authors commented on the manuscript and contributed to 766 its improvement.

767 **Competing interests.** The authors declare that they have no conflict of interest.

768 Special issue statement. This article is part of the special issue "Atmospheric deposition in the low-nutrient-

- 769 low-chlorophyll (LNLC) ocean: effects on marine life today and in the future (ACP/BG inter- journal SI)".
- 770 It is not associated with a conference.

771 Acknowledgements. The authors wish to thank Thierry Alix the captain of the R/V Pourquoi Pas ? as well 772 as the whole crew and technical staff for their involvement in the scientific operation. We gratefully thank 773 Thibaut Wagener for his involvement in the trace-metals clean marine sampling and Mickaël Tharaud for 774 the HR-ICP-MS analysis. We thank the Leosphere Technical support team and especially Alexandre Menard 775 for their remote assistance with LIDAR repair under difficult off-shore conditions. Hélène Ferré and the 776 AERIS/SEDOO service are acknowledged for real-time collection during the cruise of maps from 777 operational satellites and forecast models used in this study, with appreciated contributions of EUMETSAT 778 and AERIS/ICARE for MSG/SEVIRI products. EUMETNET is acknowledged for providing the pan-779 European weather radar composite images through its OPERA programme. We acknowledge the US 780 National Oceanic and Atmospheric Administration (NOAA) Air Resources Laboratory (ARL) for the 781 provision of the HYSPLIT (HYbrid Single-Particle Lagrangian Integrated Trajectory) model via NOAA 782 ARL READY website (http://ready.arl.noaa.gov) used in this publication. This study is a contribution to 783 the PEACETIME project (http://peacetime-project.org; last accessed 05/04/2021), a joint initiative of the 784 MERMEX and ChArMEx programmes supported by CNRS-INSU, IFREMER, CEA and Météo-France as 785 part of the decadal meta-programme MISTRALS coordinated by CNRS-INSU. PEACETIME was endorsed 786 as a process study by GEOTRACES and is also a contribution to IMBER and SOLAS international programs.

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**Figure Captions:** 



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1090	reanalysis. The red circle indicates the R/V position.
1091	Figure 2: On-board lidar-derived a) Apparent backscatter coefficient (ABC), (b) Temporal evolution (in
1092	Local Time) of the lidar-derived volume depolarization ratio (VDR) where the dust plume is highlighted for
1093	values higher than ~1.7 (yellow to red colors) and the rain by values higher than 3 (indicated by the arrow),
1094	and c) vertical profiles of the aerosol extinction coefficient (AEC) in cloud free condition, integrated over 3
1095	periods along the dust plume event, noted A, B and C in figure a. The grey shade represents the root mean
1096	square (rms) variability along the time of the measurement. The dust layer is highlighted on the profiles. The
1097	mean aerosol optical thickness is given in the boxed legend with its temporal variability (1 sigma). The
1098	location of the marine boundary layer (MBL) is also pointed.

Figure 1: Total precipitation (mm) between May 28 at 20:00 UTC and 29 at 10:00 UTC from ERA5 ECMWF

- Figure 3: Rain rates (mm/h) during the night between the 4 and 5 June, when Rain Fast was collected onboard, issued from European rain radar composites (OPERA programme) of June 5 between 00:00 and 02:45
  UTC.
- Figure 4: Boxplots of dissolved (left panels) and particulate (right panels) marine concentrations (pM) for the different TMs within the ML (upper panels) and the SML (lower panels) at ION and FAST stations. In the box plots, the box indicates the interquartile range, i.e. the 25<sup>th</sup> and the 75<sup>th</sup> percentile, and the line within the box marks the median. The whiskers indicate the quartiles ±1.5 times the interquartile range. Points above
- and below the whiskers indicate outliers outside the 10<sup>th</sup> and 90<sup>th</sup> percentile.
- Figure 5: Comparison of dissolved (D) and total (T) TMs concentrations along with data from 14 formerstudies carried out in the eastern and western Mediterranean Sea.
- Figure 6: Enrichment Factors (EF, upper panel) and solubility (%, bottom panel) of phosphorus (P) and TMsordered by increasing EF in the two rainwater samples.
- 1111 Figure 7: Dissolved (upper panels) and particulate (lower panels) wet deposition fluxes (µmol m<sup>-2</sup>) for the

1112 different TMs estimated from the two rains sampled on-board, considering the standard deviation on the TMs

- 1113 concentrations and the spatial variability of total precipitation over the area of sampling (Rain ION in blue
- and Rain FAST in red).
- Figure 8: Sampling chronology during the ION and FAST stations for SML, SSW and ML. The blue periods correspond to rainfall in the station area (after ERA 5 reanalysis and radar imagery, see section 3.1). Samplings were performed 4 days and 2 days before and 2 h after Rain ION, and at a higher frequency at the FAST station: 57, 37 and 7.5 hours before and 4.5, 12, 24 hours after Rain FAST. SML and SSW samples could not be collected immediately before and after the rains because of bad weather conditions, and were
- collected 3 and 4 days before Rain ION, and 57 and 20 h before and 30 h after Rain FAST.





- 1121 Figure 9: Comparison between TMs wet deposition fluxes (in green) and TMs marine stock delta (before
- and after the rain) in the SML (in blue) and in the ML (in red) at FAST. Dissolved = upper panels and
- 1123 particulate = lower panels. Marine stocks increase are expressed in absolute values (Cd, Co and Pb stocks in
- 1124 nmol m<sup>-2</sup>, and the other TMs in  $\mu$ mol m<sup>-2</sup>) and in relative values (%). N.E.: not enhanced (increase <5%).
- 1125 Figure 10: Comparison of marine stocks in the ML at all the stations occupied during the PEACETIME
- 1126 cruise with atmospheric inputs estimated (1) from ION and FAST rains (Boxes) and (2) from an intense wet
- 1127 dust deposition event of 9 g m<sup>2</sup> (blue dots). Cd, Co and Pb stocks are in nmol m<sup>2</sup>, and the other TMs in
- 1128 µmol m<sup>-2</sup>. For Mn, marine stocks are derived from surface concentrations close to Corsica coasts (Wuttig et
- al., 2013: samples OUT at 0, 5 and 10 m) and in the Ionian Sea (Saager et al., 1993: Bannock basin at 0, 10,
- and 15 m), as no measurement is available from the PEACETIME cruise. Boxes and whiskers as in Fig. 4.

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