

Wet deposition in the remote western and central Mediterranean as a source of trace metals to surface seawater

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Abstract. This study reports the only recent characterisation of two contrasted wet deposition events collected during the PEACETIME cruise in the open Mediterranean Sea (Med Sea), and their impact on trace metal (TM) marine stocks. Rain samples were analysed for Al, 12 TMs (Co, Cd, Cr, Cu, Fe, Mn, Mo, Ni, Pb, Ti, V and Zn) and nutrients (N, P, dissolved organic carbon) concentrations. The first rain sample collected in the Ionian Sea (Rain ION) was a typical regional background wet deposition event whereas the second rain sample collected in the Algerian Basin (Rain FAST) was a Saharan dust wet deposition event. Even in remote Med Sea, the background TM inputs presented an anthropogenic signature, except for Fe, Mn and Ti. The concentrations of TMs in the two rain samples were significantly lower compared to concentrations in rains collected at coastal sites reported in the literature, due to the decrease of anthropogenic emissions during the preceding decades. The atmospheric TM inputs were mainly as dissolved forms even in dusty Rain FAST. The TM stocks in the mixed layer (ML, 0-20 m) at the FAST station before and after the event showed that the atmospheric inputs were a significant supply of particulate TMs and of dissolved Fe and Co for surface seawater. Even if the wet deposition delivers TMs mainly as soluble forms, the post-deposition aerosol dissolution could to be a key additional pathway in the supply of dissolved TMs. At the scale of the western and central Mediterranean, the atmospheric inputs were of the same order of magnitude as ML stocks for dissolved Fe, Co and Zn, highlighting the role of the atmosphere in their biogeochemical cycles in the stratified Mediterranean Sea. In case of intense dust-rich wet deposition events, the role of atmospheric inputs as [an](#) external source was extended to dissolved Co, Fe, Mn, Pb and Zn. Our results suggest that the wet deposition constitutes only a source of some of dissolved TMs for Med surface waters. The contribution of dry deposition on the atmospheric TM inputs ~~need~~[needs](#) to be investigated.

1. Introduction

Atmospheric deposition of continental aerosol has long been recognized to influence trace element concentrations in remote oceanic surface waters (Buat-Ménard and Chesselet, 1979; Hardy, 1982; Buat-Ménard, 1983). In particular, the Mediterranean Sea (Med Sea) is an oligotrophic environment where marine microbial growth is nutrient-limited during the long Mediterranean summer season, which is characterized by a strong thermal stratification of surface waters (The Mermex Group, 2011). The Mediterranean atmosphere is characterized by the permanent presence of anthropogenic aerosols from industrial and domestic activities around the basin (e.g., Sciare et al., 2003; Kanakidou et al., 2011). In addition to this anthropogenic background, the Mediterranean basin is also subject to seasonal contributions of particles from biomass fires in summer (Guieu et al., 2005) and to intense sporadic Saharan dust inputs (e.g., Lojze-Pilot and Martin, 1996; Vincent et al., 2016). Several studies have emphasized that the atmospheric deposition of aerosols, notably through wet deposition, plays a significant role in the marine cycles of nutrient, such as nitrogen (N) and phosphorus (P) (e.g. Pulido-Villena et al., 2010; Richon et al., 2018 a and b; Violaki et al., 2018) and micronutrients, such as iron (Fe) (Bonnet and Guieu, 2006). Recently, atmospheric dust inputs were identified to have a fertilizing effect on planktonic stocks and fluxes, even in the presence of relatively high dissolved N, P and Fe concentrations (Ridame et al., 2011; reviewed in Guieu and Ridame, 2021). Mackey et al. (2012) showed that TMs provided by dust deposition could explain this fertilizing effect. Indeed, many TMs, including Mn, Co, Ni (Mackey et al., 2012), Cu (Annett et al., 2008) and Zn (Morel et al., 1991), play physiological roles for phytoplanktonic organisms. These TMs are present in very low concentrations in oligotrophic systems such as the Med Sea, possibly limiting (or co-limiting) phytoplankton growth (Pinedo-Gonzàles et al. 2015) and pointing to the importance of dust deposition as [a](#) source of TMs for planktonic communities. On the other hand, atmospheric deposition of European aerosols was identified to have a negative effect on chlorophyll concentrations (Gallissai et al. 2014), by providing Cu, at toxic levels (Jordi et al. 2012).

The atmospheric deposition of TMs in the Mediterranean is related to both dust and anthropogenic aerosol deposition (Desboeufs et al., 2018). The role of dust deposition as a source of TMs was observed from the correlation between the atmospheric deposition of mineral dust, and the enrichment of dissolved TMs (Cd, Co, Cu, Fe) in the Med Sea surface microlayer (SML)(Tovar-Sánchez et al., 2014). For the water column, the adding of dissolved Fe and Mn was emphasized in mesocosm experiments after dust addition mimicking intense wet dust deposition (Wuttig et al., 2013). Bacconnais et al. (2019) showed that the dissolved Cu isotope signature in surface waters could be related to Saharan dust deposition in the Southern Med Sea. By comparison based on

94 annual or monthly deposition measurements and marine concentrations, the potential role of
95 atmospheric deposition as source of dissolved Co and Fe for Mediterranean surface waters was also
96 pointed ~~out~~ (Bonnet and Guieu, 2006; Dulaquais et al., 2017). Due to ~~Mediterranean~~ sporadic and
97 intense ~~Mediterranean~~ storms, ~~the rain events by~~ scavenging ~~loaded~~ air masses ~~loaded~~ with
98 anthropogenic aerosols or Saharan dust could lead to higher ~~TM~~ deposition ~~TM~~ fluxes ~~by rain~~ than
99 dry deposition (Desboeufs et al., 2021). Moreover, even if annual wet and dry deposition are
100 equivalent in Mediterranean (Theodosi et al., 2010), wet deposition is known to provide
101 ~~soluble~~ ~~dissolved~~, and potentially bioavailable forms of TMs (Jickells et al., 2016). Yet, to the best
102 of our knowledge, the direct impact of wet deposition on TM concentrations in surface seawater
103 has not been studied ~~in aat~~ same location by concurrently collecting both rainwater and seawater
104 samples before this work, whether in the Med Sea or in other oceanic regions.

105 Two key criteria used to assess the potential impact of TMs and nutrients are their respective
106 concentrations (or fluxes) and fractional solubility (solubility hereafter), i.e., the partitioning
107 between dissolved and total concentrations in rainwaters. Indeed, it is considered that the dissolved
108 fraction of nutrients and TMs is a surrogate for their bioavailability (Jickells et al., 2016). Few
109 studies have reported concentrations of TMs in rainwater samples collected around the
110 Mediterranean basin: Al-Momani et al., (1998), Kanellopoulou, (2001) and Özsoy and Örnektekin,
111 (2009) in the eastern basin, and Losno (1989), Guieu et al. (1997), Frau et al., (1996), Chester et al.,
112 (1997) and Guerzoni et al. (1999b) in the western basin. These studies led to highly variable TM
113 concentrations and solubilities, illustrating the large variability of TM inputs during wet deposition
114 events in the Med Sea (reviewed in Desboeufs, 2021). All these studies were performed at coastal
115 sites. Offshore samples of rainwater have rarely been reported in the literature. In the Mediterranean,
116 to our knowledge, trace element concentrations from only three rain samples collected at sea in
117 April 1981 have been reported in a PhD thesis (Dulac, 1986). However, due to the continental and
118 local sources of pollution, the variety of anthropogenic aerosol sources (Amato et al., 2016) and the
119 physico-chemical processes during atmospheric aerosol transport (chemical ageing, dispersion,
120 gravitational settling, in-cloud reactivity..)(Weinzierl et al., 2017), the TM rain composition of the
121 coastal zone may not be representative of atmospheric deposition to the remote Mediterranean.

122 The PEACETIME cruise (ProcEss studies at the Air-sEa Interface after dust deposition in the
123 MEiterranean Sea) performed in spring 2017 aimed to study the impacts of atmospheric
124 deposition, in particular Saharan dust events, on the physical, chemical and biological processes in
125 this marine oligotrophic environment (Guieu et al., 2020). We investigated the concentration and
126 solubility of TMs and nutrients from two rain events sampled in the central and western Med Sea

127 during the cruise. We compared our results with the existing literature on TM concentrations in
128 offshore and coastal rain samples dating of ~~1990's~~1990s to investigate the time evolution of TM
129 concentrations and the potential differences with the open sea. Additionally, to assess the impact of
130 wet deposition on the surface TM concentrations, surface seawater and rain water were concurrently
131 collected. This is the first time TM data for these atmospheric and marine compartments in Med
132 Sea have all been discussed at the same time and in a same place.

133 2. Sampling and methods

134 2.1 Sampling and chemical analysis of rainwater

135 The PEACETIME oceanographic campaign (<https://doi.org/10.17600/17000300>) took place in the
136 western and central Med Sea on-board the French R/V *Pourquoi Pas?* between 11 May and 10 June
137 2017, i.e. at the beginning of the Mediterranean stratification season (Guieu et al., 2020). The rain
138 collector was installed on the upper deck (22 m above sea level) where no on-board activities were
139 taking place to avoid contamination. The rain collector was equipped with an on-line filtration
140 system to separate the dissolved and particulate fractions at the time of collection (details of the
141 filtration system are available in Heimbürger et al., 2012) allowing for the calculation of solubility
142 of TMs in the rainwater. The filtration device was equipped with a Nuclepore® polycarbonate
143 membrane filter (porosity: 0.2 µm, diameter: 47 mm). The diameter of the funnel of the collector
144 was 24 cm. The rain collector was installed only when rain was expected and covered by an acid-
145 washed sealed plastic film when not in use. All the sampling materials were thoroughly acid-washed
146 in the laboratory prior to the cruise departure (washing protocol described in Heimbürger et al.,
147 2012). No stabilizing device was used to keep the funnel level during the pitch and roll of the ship,
148 preventing a precise assessment of the height of rainfall from the collected water volume. During
149 the rain sampling, the R/V was always facing the wind to avoid contamination by the ship's exhaust,
150 as the chimney was situated on the lower deck and behind the collector.

151 Immediately after sampling, the collector was disassembled under a laminar flow hood inside an
152 on-board clean-room laboratory. The dissolved fraction was separated into four aliquots dedicated
153 to i) TM determination by inductively coupled plasma coupled methods (ICP), ii) major ion analysis
154 by ion chromatography (IC), iii) dissolved organic carbon (DOC) determination by high-
155 temperature catalytic oxidation (HTCO) on a Shimadzu total organic carbon analyzer (as described
156 in van Wambeke et al., 2021a), and iv) pH measurement. For ICP measurements, the sample was
157 acidified immediately to 1% by volume of ultra-pure nitric acid (67-69%, Ultrapur, Normatom®,

158 VWR). For IC analysis, the filtrate was immediately frozen. For DOC analysis, the filtrate was
159 collected into pre-combusted glass ampoule and acidified to pH 2 with phosphoric acid. Ampoule
160 was immediately sealed and stored in the dark at room temperature.

161 The filter (particulate fraction) was dried under the laminar flow hood, and then put in a storage box
162 and packed with a plastic bag to avoid contamination. After returning to the laboratory, filters were
163 acid digested by using the protocol adapted from Heimbürger et al. (2012) as follows: filters were
164 placed in tightly capped Savillex™ PFA digestion vessels with 4 mL of a mixture of HNO₃ (67-
165 69%, Ultrapur, Normatom®, VWR), H₂O and HF acids (40%, Ultrapur, Normatom®, VWR) in a
166 proportion of 3: 1: 0.5, then heated in an oven at 130°C for 14 hours. After cooling, the acid solution
167 was completely evaporated on a heater plate (ANALAB, 250, A4) at 140°C for about 2h, then 0.5
168 mL of H₂O₂ (30-32%, Romil-UpA™) and 1 mL of the acidified water (2% HNO₃) was added to the
169 vessels and heated for 30 min. to dissolve the dry residue in the bottom of the vessels; finally, 12
170 mL of acidified water (1% HNO₃) was added to obtain 13.5 mL of solution in a tube for ICP
171 analyses.

172 The dissolved fractions and the solutions issued of digestion of the particulate fractions were
173 analysed by ICP-AES (Inductively Coupled Plasma Atomic Emission Spectrometry, Spectro
174 ARCOS Ametek®, Desboeufs et al., 2014) for major elements (Al, Ca, K, Mg, Na, S) and by HR-
175 ICP-MS (High Resolution Inductively Coupled Plasma Mass Spectrometry, Neptune Plus™ at
176 Thermo Scientific™) for P and TMs: Cd, Co, Cr, Cu, Fe, Mn, Mo, Ni, Pb, Ti, V and Zn. The
177 dissolved fraction was analysed by IC (IC 850 Metrohm) for the inorganic and organic anions (NO₂⁻
178 , NO₃⁻, PO₄³⁻, SO₄²⁻, F⁻, Cl⁻, Br⁻, HCOO⁻, CH₃COO⁻, C₂H₅COO⁻, MSA, C₂O₄²⁻) and for the cation
179 NH₄⁺ (Mallet et al., 2017). Only TMs, major nutrients, i.e. N and P species, and Al are discussed in
180 this manuscript. The speciation of dissolved P was estimated by determining dissolved inorganic
181 phosphorus (DIP) from phosphate concentrations expressed as P and the dissolved organic
182 phosphorus (DOP) from the difference between total dissolved phosphorus (TDP), obtained by HR-
183 ICP-MS, and DIP, obtained by IC. The dissolved inorganic nitrogen (DIN) were defined as sum of
184 NO₂⁻, NO₃⁻ and NH₄⁺, expressed as N.

185 In order to estimate the contamination of sampling and analytical protocols, three blanks of rain
186 samples, collected on-board during the cruise, were used and processed. Blank samples were
187 prepared by rinsing the funnel with 150 mL of ultrapure water (18.2 mΩ·cm) with the same protocol
188 of rain collection. The procedural limit of detections (LoD) were defined as 3 x standard deviation
189 of blank samples both for dissolved and particulate fractions estimated after acid digestion. All

190 dissolved and particulate sample concentrations were higher than LoD, except for NO₂ in the two
191 rain samples. The blank concentrations represented 10.2% on average for TMs and were typically
192 lower than 20% of the sample concentrations, except for Cd (52%) and Mo (43%) in the dissolved
193 fraction. Blank concentrations were subtracted from all sample concentrations.

194 **2.2 Atmospheric ancillary measurements**

195
196 The PEGASUS (Portable Gas and Aerosol Sampling UnitS, www.pegasus.cnrs.fr) mobile platform
197 of LISA is a self-contained facility based on two standard 20-feet containers, adapted with air-
198 conditioning, rectified power, air intake and exhausts for sampling and online measurements of
199 atmospheric aerosols and gaseous compounds, and their analysis (Formenti et al., 2019). During the
200 PEACETIME cruise, only the sampling module of the facility was deployed on the starboard side
201 of deck 7 of the R/V. The PEGASUS instrumental payload of relevance to this paper included
202 measurements of the major gases such as NO_x, SO₂, O₃ and CO by online analysers (Horiba APNA,
203 APSA, APOA and PICARRO respectively; 2-min resolution, for all gases was detection limit 0.5
204 ppb and 1 ppb for CO). These gases were used to estimate the origins of sampled air.

205 From the first of June 2017 (not operational before), additional measurements by an ALS450®
206 Rayleigh-Mie lidar (Leosphere™; Royer et al., 2011) were used to monitor the vertical distribution
207 of aerosols over time and the associated integrated columns. The vertical lidar profiles were
208 analysed to yield the apparent backscatter coefficient (ABC) corrected from the molecular
209 transmission, as well as the volume depolarisation ratio (VDR). The inversion procedure (Chazette
210 et al., 2016 and 2019) to retrieve the aerosol extinction coefficient (unit km⁻¹) used a vertical-
211 dependent lidar ratio that took into account two aerosol layers. The first layer corresponded to
212 marine aerosols in the marine boundary layer (MBL), the second to a desert aerosol layer that could
213 extend between ~1 and 6 km above mean sea level (amsl). In accordance with Chazette et al. (2016),
214 for the same wavelength and region, the lidar ratios were set to 25 and 55 sr, respectively. The
215 vertical profile of the aerosol extinction coefficient was retrieved from 0.2 km amsl upwards with a
216 vertical resolution of 15 m. Based on these profiles, the integrated column content of dust aerosols
217 was estimated using a specific extinction cross-section of 1.1 m² g⁻¹ as proposed by Raut and
218 Chazette (2009).

219 In addition, detailed meteorological data such as air and sea temperature, atmospheric pressure,
220 relative humidity, atmospheric pressure, heat flux and wind speed and direction were provided on
221 a 30 second time step basis by the ship's permanent instrumentation.

222 **2.3 Sampling and analysis of dissolved TMs in seawater**

223 TM samples were also collected in the water column using a titanium trace metals clean (TMC)
224 rosette (mounted with 24 teflon-coated Go-Flo bottles) before and after the rain event (Bressac et
225 al., 2021). Although rosette deployments were performed over the whole water column, we focus
226 here on the 0-20 m marine mixed-layer (ML). Immediately after recovery, the Go-Flo bottles were
227 transferred inside a class-100 clean laboratory container. Seawater samples were directly filtered
228 from the bottles through acid-cleaned 0.2 µm capsule filters (Sartorius Sartobran-P-capsule
229 0.45/0.2-µm). All samples were acidified on-board to pH<2 with Ultrapure-grade HCl under a class-
230 100 HEPA laminar flow hood. Metals (namely Cd, Co, Cu, Ni, Mo, V, Zn and Pb) were pre-
231 concentrated using an organic extraction method (Bruland et al., 1979) and quantified by ICP-MS
232 (Perkin Elmer ELAN DRC-e) in the home laboratory. The accuracy of the pre-concentration method
233 and analysis for TMs was established using Seawater Reference Material (CASS 6, NRC-CNRC)
234 with recoveries ranging from 89% for Mo to 108% for Pb. Due to the complexity of the analytical
235 method, all the TMC samplings were not analysed for Cd, Co, Cu, Ni, Mo, V, Zn and Pb. Overall,
236 1 or 2 depths were obtained in the mixed layer (0-20 m). Dissolved Fe and Al concentrations were
237 also measured on-board. Dissolved Fe concentrations were measured using an automated Flow
238 Injection Analysis (FIA) with online preconcentration and chemiluminescence detection (Bonnet
239 and Guieu, 2006), and dissolved Al concentrations using the fluorometric method described by
240 Hydes and Liss (1976). Sampling and analysis for dissolved Fe and Al concentrations are fully
241 described in Bressac et al. (2021), and covered at least four depths in the 0-20 m mixed layer.

242 **2.4 Concurrent sampling strategy**

243 Our sampling strategy of collecting seawater before and after rains were to trace the fate of
244 atmospheric TMs and nutrients in the water column after wet deposition events in an open ocean.
245 The sampling time-resolution was optimized to obtain results before and after rains, however this
246 strategy was very dependent on meteorological conditions. The time chart of the sampling of rain
247 and surface water is presented in Fig. 1.

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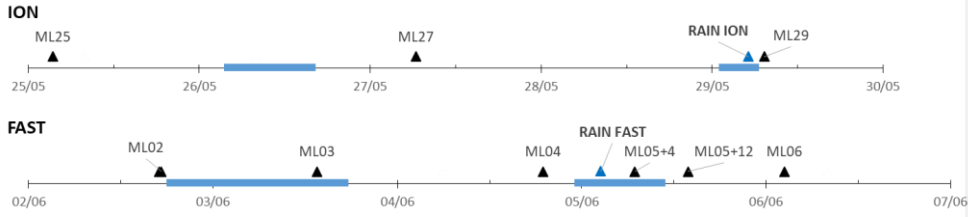


Figure 1: Sampling chronology during the ION and FAST stations for the ML and rains. The blue periods correspond to rainfall in the vicinity of the R/V position (from ERA 5 reanalysis and radar imagery, see section 3.1.), and blue triangles to the rainfall on the R/V position. Samplings were performed 4 days (ML25) and 2 days (ML27) before and 2 h (ML29) after Rain ION, and at a higher frequency at FAST: 57 (ML02), 37 (ML03) and 7.5 (ML04) hours before and 4.5 (ML05+4), 12 (ML05+12), 24 hours (ML06) after Rain FAST.

The potential enrichment of the ML ~~from the Rain FAST~~ was estimated by comparing atmospheric wet deposition fluxes and marine stocks in particulate and dissolved fraction and by calculating the difference (delta) in TM stocks before and after rain. Due to the long time (48h) between ML27 and ML29 and hence the possible change of water masses between the two samplings excluding to have reliable background concentration levels, the enrichment of the ML was not calculated for ION station. Dust rain deposition over the FAST station area started on 3 June (see section 3.1). Bressac et al. (2021) showed that the dust signature, traced by changes in Al and Fe particulate stocks in the ML, was already visible from the ML03 sampling, but not for dissolved stocks. However, they presumed also that the water mass sampled before deposition (ML02) was different from the one sampled during the rest of the time-series (ML03 to ML06) but was representative of particulate background level. In consequence, we defined the enrichment of ~~seawater layers~~ the ML as the difference between the stocks after rains (ML05+4) and the initial seawater stocks (ML02 for particulate stocks (before dust signature) and ML04 for dissolved stocks (the closest sampling before the rain)).

2.5 Enrichment factor and solubility

In order to better constrain the origin of TMs in the rain samples, their enrichment factors (EF; Rahn 1976) relative to the Earth's crust were calculated based on their total concentrations (dissolved + particulate fractions) as:

$$EF = \frac{([X]/[Al])_{sample}}{([X]/[Al])_{crust}} \quad (1)$$

where $[X]/[Al]$ is the ratio between an element X and Al concentrations in rainwater samples (at the numerator), and in the upper continental crust (denominator) from Rudnick and Gao (2003). Aluminium is currently used as a reference element as it predominantly has a crustal origin. For a given TM, $EF > 1$ indicates an enrichment with respect to the average composition of the Earth's crust. To account for the soil composition variability of mineral dust sources, TMs with an EF value > 10 are considered significantly enriched, which points to a non-crustal source (Rahn, 1976). For most metals, enrichment shows important input from anthropogenic sources, due to their low content in other non-crustal sources such as sea spray or biogenic aerosols (Jickells et al., 2016).

The fractional solubility of TMs in the two rainwater events was calculated as:

$$S_X\% = \frac{[X]_{dissolved}}{[X]_{total}} \times 100 \quad (2)$$

where $S_X\%$ is the solubility (in %) of an element X in the rainwater, $[X]_{dissolved}$ and $[X]_{total}$ are its soluble and total concentration, respectively.

2.6 Atmospheric wet deposition fluxes

Impacts on biogeochemical cycles and ecosystem functioning following a rain event occur on time scales of a few days (2-3), and space scales of tens of km (about 20-50 km within the radius of the R/V position). In the specific context of oceanographic cruising, the documentation of these impacts is restricted to the vertical dimension at the prescribed temporal scale. In this vertical dimension, the exchange of TMs into the ML was controlled both by atmospheric inputs over the R/V position and by advection from surrounding water masses that may have been impacted by surrounding rainfall. Therefore, we had to consider this process in our estimation of the atmospheric fluxes contributions. For this purpose, the atmospheric fluxes have to be integrated to the extent of the rain area that can impact the marine surface layers. We derived wet deposition fluxes by considering the total precipitation accumulated during the duration of the rain over the area around the R/V position (in a radius of ~25 km). Thus, the wet deposition fluxes were calculated by multiplying the TM concentrations ($\mu\text{g L}^{-1}$ or $\mu\text{mol L}^{-1}$) in our rain samples by the total precipitation (mm) in this area. The total precipitation of the rain events was issued from the hourly total precipitation accumulated during the rain events over the region from ERA5 ECMWF reanalysis (Herbasch et al., 2018) and from the rain rate composite radar products from the European OPERA database (Saltikoff et al., 2019), when it was possible. Although subject to uncertainties (Morin et al., 2003), a surface-based weather radar is probably the best tool to estimate rainfall in the surroundings of the R/V since this method is a direct measurement of precipitation with both a best time and spatial resolution in comparison of model estimations. However, the OPERA database does not include Italian radars,

which did not cover the central area of the Ionian Sea during the cruise anyway. ERA5 data were available on regular latitude-longitude grids at $0.25^\circ \times 0.25^\circ$ resolution. The accumulated precipitation was taken from the grid-points spanning the R/V location, $\pm 0.25^\circ$ around the central grid-point for integrating the regional variability. Surface rain rate radar composite images were available every 15 minutes with a spatial resolution of 2 km x 2 km. The accumulated precipitation was the sum of integrated rain rates during the rain duration averaged over the radar pixels spanning the R/V location within a radius of about 25 km around.

2.7 Stocks in the surface seawater

The trace metals stocks within the ML were calculated by trapezoidal integrations of marine concentrations from TMC rosette samplings. The upper water column was stratified along the cruise transect (Taillandier et al., 2020), with a ML depth (MLD) ranging from 7 to 21 m (11 to 21 m at ION and 11 to 19 m at FAST (Van Wambeke et al., 2020). The MLD fluctuations, for example due to wind peaks associated with rain events, could create rapidly changing conditions of vertical advection from deeper waters. However, with no significant increase in TM concentrations being observed below the ML down to about 50 m (not shown), the enrichment observed in the ML after rain could not be attributed to any mixing with deeper water due to high wind. In consequence, stocks in the ML have been integrated over a constant depth range of 0-20 m for comparison, in accordance with Bressac et al. (2021). For Cu, Fe, Ni and Zn, stocks were estimated both for the dissolved and particulate fractions in the ML, for Co, Cd, Mo, Pb and V only for the dissolved fraction (particulate fraction was not analysed) and for Mn and Ti only for the particulate fraction. The particulate and dissolved fractions of TM stock or fluxes will be mentioned as pTM and dTM respectively.

The partitioning coefficient between the particulate and dissolved phases ($K_d = [\text{particulate}]/[\text{dissolved}]$) was used to investigate exchange between the dissolved and particulate pools of TMs.

3. Results

3.1 Atmospheric conditions during wet deposition events

The general meteorological conditions during the cruise indicated that the ION and FAST stations were highly affected by cloudy weather conditions. During these periods, two significant rain events

occurred over the R/V position and were sampled: The first sample (Rain ION) was collected during the 4-day ION station occupation in the Ionian Sea in the early morning of 29/05/17 at 03:08 UTC. The second rain event (Rain FAST) occurred during the 5-day "Fast action" station occupation, ('FAST') in the Algerian Basin during the night of 05/06/17 at 00:36 UTC (Table 1). The two rain sample collections coincided with peaks in relative humidity and wind speed, and minima in air temperature (not shown).

Table 1: Information regarding the two rains collected during the PEACETIME cruise.

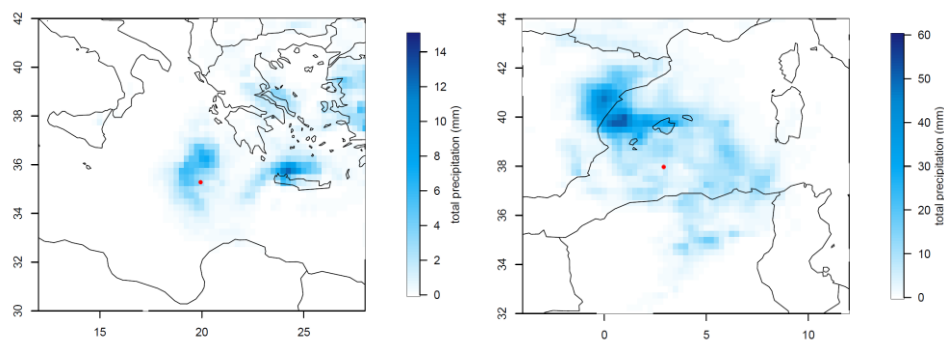
Sample	Sampling time	Station name (dates) and rain location	Estimated total precipitation
Rain ION	29 May 2017, 03:08-04:00 (UTC) 05:08–06:00 (local time)	ION (25–29 May) 35.36°N, 19.92°E	3.5 ±1.2 mm
Rain FAST	05 June 2017, 00:36-01:04 (UTC) 02:36–03:04 (local time)	FAST (2–7 June) 37.94°N, 2.91°E	6.0 ±1.5 mm

Tableau mis en forme

3.1.1. Rain ION

The ERA 5 data reanalysis shows two periods of precipitation in the vicinity of the R/V position, i.e. in the morning and evening of 26 June (not shown) and in the night between 28 and 29 June 2017, in agreement with on-board visual observations. The rain event collected at ION was the product of a large cloud system, covering an area of about 90 000 km² around the R/V position, spreading over the Ionian and Aegean Seas (Fig. 2). As no radar measurements were available for this area, the accumulated precipitation (3.5 ±1.2 mm) was estimated from ERA 5 data reanalysis on the grid-point corresponding to the ION station (±0.25°) around the R/V position. The wash-out of the atmospheric particles was revealed by the decrease in aerosol number concentrations monitored on-board from about 1900 to 300 part.cm⁻³ (supplementary material Fig. S1). Air mass back-trajectories showed that the scavenged air masses came from Greece both in the marine boundary layer and in the free troposphere (Fig. S1). The satellite observations also showed low aerosol optical thickness during this period (not shown), meaning low amounts of aerosols in the atmospheric column. No significant European pollution influence was monitored by on-board measurements during this event, with major gas mixing ratios and aerosol concentrations close to the average values of the cruise (Fig. S1), i.e. below the limit of detection for NO_x, 1.2 ppb for SO₂,

361 51 ppb for O₃, 80 ppb for CO and 3000 part.cm⁻³. On this basis, this wet event was representative
 362 of a Mediterranean background marine rain event.



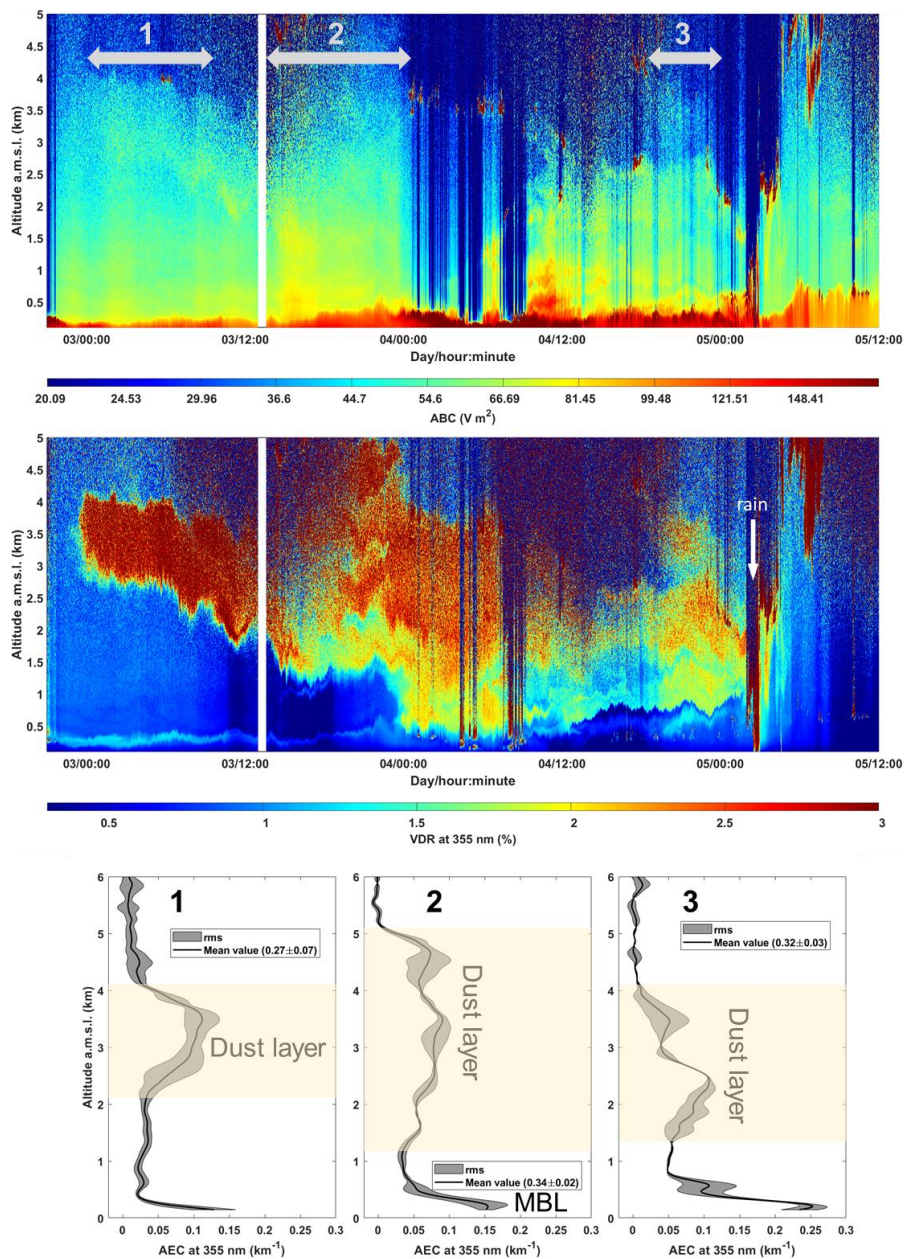
363
 364 **Figure 2: Total precipitation (mm) between 28 May at 20:00 UTC and 29 May 2017 at 10:00 UTC**
 365 **(Rain ION, left figure) and between the 4 June at 20:00 and 5 June 2017 at 09:00 UTC (Rain FAST**
 366 **right figure) from ERA5 ECMWF reanalysis. The red circle indicates the R/V position. Note different**
 367 **scales for total precipitation.**

368 3.1.2. Rain FAST

369 As detailed in Guieu et al. (2020), the FAST position was decided on the basis of regional model
 370 forecast runs and satellite observations, for the purpose of catching a wet dust deposition event.
 371 Significant dust emissions were observed from NASCube (<http://nascube.univ-lille1.fr/>, Gonzales
 372 and Briottet, 2017) over North Africa from the night of 30-31 May 2017, then new dust emissions
 373 in the night from 3 to 4 June 2017 in Algeria and southern Morocco associated with a northward
 374 atmospheric flux. On 30 May, the SEVIRI AOD satellite product ([https://www.icare.univ-](https://www.icare.univ-lille.fr/data-access/browse-images/geostationary-satellites/)
 375 [lille.fr/data-access/browse-images/geostationary-satellites/](https://www.icare.univ-lille.fr/data-access/browse-images/geostationary-satellites/), Thieuleux et al., 2005) confirmed the
 376 presence of atmospheric dust in a cloudy air mass over the western part of the Mediterranean, and
 377 from 2 June the export of a dust plume from North Africa south of the Balearic Islands with high
 378 AOD (>0.8) on the Alboran Sea was observed (Fig. S2). The dust plume was transported to the NE
 379 up to Sardinia on 4 June, with AOD<0.5. Clear sky with low AOD was left west of 4°E on 5 June.

380 On-board lidar measurements (Fig. 3 a,b,c) showed that the aerosol plume was present over the R/V
 381 position from 2 June 2017 at 21:00 UTC until the rain event, and corresponded to a dust aerosol
 382 layer highlighted by the high depolarization. The dust plume was concentrated between 3 and 4 km
 383 at the beginning of the station occupation, then expanded down to the marine boundary layer (about
 384 500 m amsl) by the end of the day on 3 June 2017. The mass integrated concentrations of dust

385 aerosols derived from the profiles of aerosol extinction ranged from a minimum of $0.18 \pm 0.005 \text{ g}$
386 m^{-2} just before the rain to a maximum of $0.24 \pm 0.009 \text{ g m}^{-2}$, where standard deviations indicate the
387 temporal variability (1 sigma).



388

389 **Figure 3: On-board lidar-derived: a. apparent backscatter coefficient (ABC) (top panel), b. temporal**
 390 **evolution of the lidar-derived volume depolarization ratio (VDR) (middle panel in local time) where**

the dust plume is highlighted for values higher than ~ 1.7 (yellow to red colours) and the rain by values higher than 3 (indicated by the white arrow), and c. vertical profiles of the aerosol extinction coefficient in cloud free condition, integrated over 3 periods, noted 1, 2 and 3 on the top panel, along the dust plume event (bottom panels). The grey shade represents the root mean square (rms) variability along the time of the measurement. The dust layer is highlighted on the profiles. The mean aerosol optical thickness is given in the boxed legend with its temporal variability (1 sigma). The location of the marine boundary layer (MBL) is also pointed.

Rain was observed by weather radar images in the area of the R/V neighbouring from 3 June at 7:00 UTC. The rain recorded around FAST was associated with two periods of rain: from 7:00 to 14:00 UTC on 03/06, and from 16:00 UTC on 04/06 to 06:00 UTC on 05/06. For this latter case, a rain front ($100\,000\text{ km}^2$), moving eastward from Spain and North Africa regions, reached the FAST the night between the 4 and 5 June (Fig. 4). Wet deposition between the 4 and early 5 June in the FAST station area were confirmed by radar imagery, showing several other instances of rain around the R/V position before and after the rain sampling (Fig. 4). Continuous on-board lidar measurements confirmed the below-cloud deposition during the rain event of early 5 June (Fig. 3b). Rain FAST was a wet deposition event occurring at the end of an episode of transport of Saharan dust, whereas precipitation on the 3 June occurred during the maximum of the dust plume (Fig. 3b and S2). The surface concentrations of gas and particles, measured on-board, suggest no clear dust or anthropogenic influence in the atmospheric boundary layer during this period of wet deposition, in agreement with back trajectories of low altitude air masses (Fig S2.), presuming no local mixing between dust and anthropogenic particles into rain samples. The total precipitation estimated from radar rainfall yield an accumulated precipitation of $6.0 \pm 1.5\text{ mm}$ ($\pm 25\text{ km}$ around the R/V position), in agreement with ECMWF reanalysis ERA5 (Fig. 2) for the wet deposition on the night of 4-5 June ($5.7 \pm 1.4\text{ mm}$ in the grid-point spanning the R/V position, i.e. $\pm 0.25^\circ$).

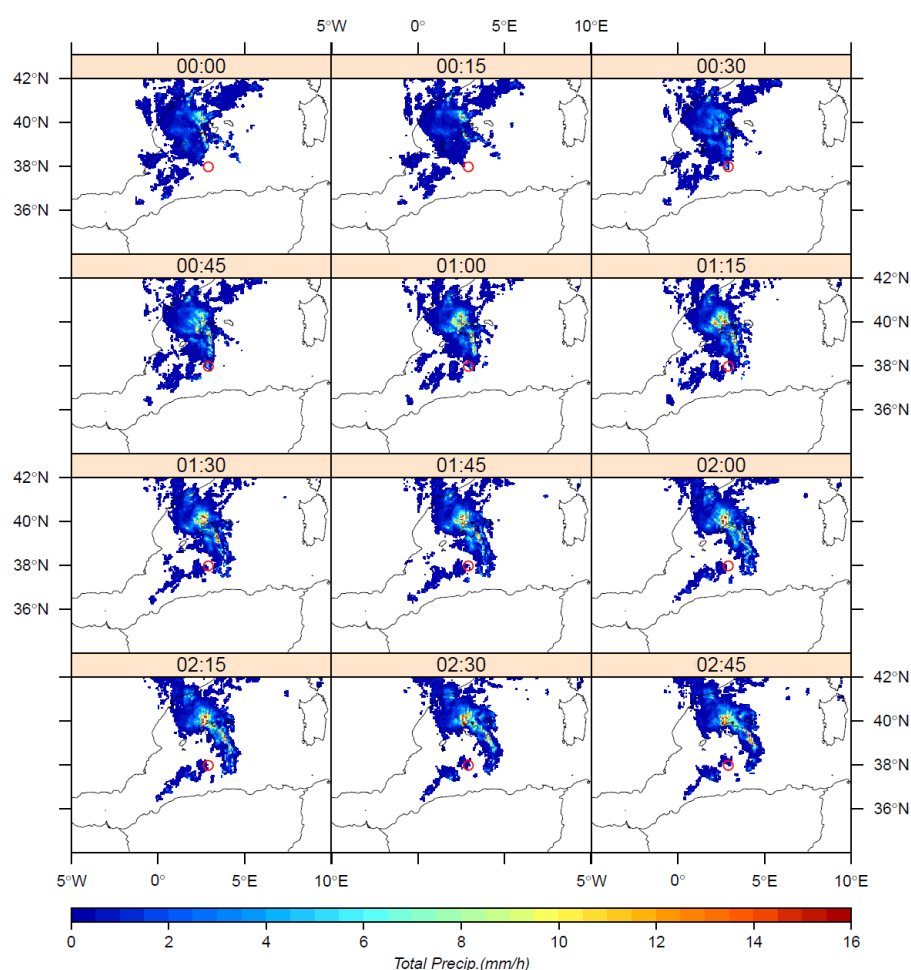


Figure 4: Rain rates (mm/h) during the night between the 4 and 5 June, when Rain FAST was collected on-board, issued from European rain radar composites (OPERA programme) of 5 June between 00:00 and 02:45 UTC.

3.2. Chemical composition of rains

Dissolved and total concentrations of nutrients and TMs in the rain samples are presented in Table 2. Among all measured dissolved concentrations, nitrate was the most abundant nutrient, followed by ammonium (Table 2). The nitrite concentration was below the limit of detection for the two rain samples. Regarding TMs in rain, Fe and Zn presented the highest concentrations both in the dissolved fraction and in total deposition with the same order of magnitude (10 to 25 $\mu\text{g L}^{-1}$). Co,

Cd and Mo had the lowest concentrations ($<0.1 \mu\text{g L}^{-1}$ in both events, associated to the greatest uncertainties due to LoD), whereas the other TM concentrations ranged between 0.1 and $10 \mu\text{g L}^{-1}$ (Table 2). Concentrations of nutrients and the majority of TMs were higher in the dust-rich rain, except dissolved Pb (similar concentrations in both rain samples) and Cr (3 times higher concentration in Rain ION relative to Rain FAST).

Table 2: Dissolved and total concentrations of nutrients and TMs in the two rain samples collected during the PEACETIME cruise in $\mu\text{g.L}^{-1}$ or ng.L^{-1} and $\mu\text{mol.L}^{-1}$ or nmol.L^{-1} in the parentheses (sd = standard deviation from three replicates).

		Rain ION				Rain FAST				
		Dissolved		Total		Dissolved		Total		
		concentrations	$\pm sd$	concentrations	$\pm sd$	concentrations	$\pm sd$	concentrations	$\pm sd$	
Nutrients	NO ₃ ⁻	$\mu\text{g L}^{-1} (\mu\text{mol L}^{-1})$	1185 (19.1)	71 (1.1)			3694 (59.6)	222 (3.6)		
	NH ₄ ⁺	$\mu\text{g L}^{-1} (\mu\text{mol L}^{-1})$	366 (20.3)	11 (0.6)			654 (36.3)	19 (1.1)		
	DIN	$\mu\text{g L}^{-1} (\mu\text{mol L}^{-1})$	552 (39.4)	82 (1.7)			1343 (96)	241 (17)		
	PO ₄ ³⁻	$\mu\text{g L}^{-1} (\text{nmol L}^{-1})$	18.1 (189)	0.5 (6)			19.0 (200)	0.6 (6)		
	DIP	$\mu\text{g L}^{-1} (\text{nmol L}^{-1})$	5.87 (189)	0.18 (6)			6.20 (200)	0.19 (6)		
	DOP	$\mu\text{g L}^{-1} (\text{nmol L}^{-1})$	8.63 (278)	1.94 (75)			4.91 (158)	1.56 (57)		
	TP	$\mu\text{g L}^{-1} (\text{nmol L}^{-1})$	14.51 (468)	2.52 (81)	16.6 (536)	1.0 (33)	11.11 (358)	1.95 (63)	58.7 (1894)	3.5
	DIN/DIP		(208)				(480)			
	DOC	$(\mu\text{mol L}^{-1})$	(105.7)	(2.2)			(95.5)	(1.2)		
Metals	Al	$\mu\text{g L}^{-1} (\text{nmol L}^{-1})$	13.0 (480)	0.8 (30)	14.6 (540)	0.9 (32)	23.4 (867)	0.7 (24)	440 (16308)	7
	Cu	$\mu\text{g L}^{-1} (\text{nmol L}^{-1})$	0.71 (11.1)	0.02 (0.3)	0.73 (11.5)	0.02 (0.3)	1.15 (18.0)	0.04 (0.6)	1.63 (25.7)	0.06
	Fe	$\mu\text{g L}^{-1} (\text{nmol L}^{-1})$	15.1 (270)	0.4 (6)	17.9 (321)	0.6 (11)	19.2 (344)	0.1 (2)	231 (4140)	7
	Mn	$\mu\text{g L}^{-1} (\text{nmol L}^{-1})$	0.55 (10.0)	0.02 (0.3)	0.60 (10.9)	0.02 (0.4)	3.17 (57.8)	0.07 (1.2)	5.26 (95.7)	0.12
	Ni	$\mu\text{g L}^{-1} (\text{nmol L}^{-1})$	0.52 (8.8)	0.02 (0.3)	0.67 (11.4)	0.02 (0.4)	0.59 (10.1)	0.02 (0.4)	0.84 (14.3)	0.03
	Ti	$\mu\text{g L}^{-1} (\text{nmol L}^{-1})$	0.48 (10.0)	0.04 (0.8)	0.65 (13.6)	0.48 (3.2)	0.22 (4.7)	0.01 (0.1)	33.36 (697)	0.51
	V	$\mu\text{g L}^{-1} (\text{nmol L}^{-1})$	0.37 (7.4)	0.01 (0.2)	0.38 (7.42)	0.01 (0.25)	1.37 (26.9)	0.03 (0.5)	2.02 (39.7)	0.04
	Zn	$\mu\text{g L}^{-1} (\text{nmol L}^{-1})$	24.8 (379)	0.8 (12)	25.3 (387)	0.8 (12)	22.7 (347)	0.6 (8)	26.3 (402)	0.7
	Cd	$\text{ng L}^{-1} (\text{pmol L}^{-1})$	12.9 (115)	6.4 (57)	13.1 (117)	6.3 (56)	20.2 (180)	10.3 (92)	23.7 (210)	6.8
	Co	$\text{ng L}^{-1} (\text{pmol L}^{-1})$	44 (749)	13 (229)	47.4 (804)	14.5 (246)	82 (1386)	20 (347)	157 (2661)	28
	Cr	$\text{ng L}^{-1} (\text{pmol L}^{-1})$	241 (4636)	16 (300)	628 (12079)	5 (95)	79 (1522)	14 (260)	443 (8514)	43
	Mo	$\text{ng L}^{-1} (\text{pmol L}^{-1})$	28 (288)	10 (106)	4.1 (43)	1.4 (14)	82 (855)	11 (113)	92.1 (960)	16.2
	Pb	$\text{ng L}^{-1} (\text{pmol L}^{-1})$	170 (822)	11 (54)	175.1 (845)	1.4 (7)	166 (801)	9 (41)	604 (2917)	19

3.3. Marine concentrations

All the TMs had significantly higher concentrations in the ML compared to water below the ML and deeper (e.g. for Fe: Bressac et al., 2021), in agreement with a stratified profile associated with atmospheric inputs. The pTM and dTM concentrations within the ML (0-20 m) are displayed in Fig. 5. Concentrations were of the same order of magnitude at ION and FAST. The TMs were mainly in dissolved forms in the ML (K_d from 0.006 to 0.5), except for Fe (K_d around 2) and one outlier value for Zn (1.3) (Fig. 5).

At both stations, the highest TM concentrations in the surface seawater were found for dMo ($\sim 120 \text{ nM}$), these values are the first measurements published in Med Sea and are in agreement with the

443 high abundance of dMo in open seawater in other oceanic regions (~107 nM, Smedley and
 444 Kinniburgh, 2017). Fe is the most abundant in the particulate fraction in the ML (~4 nM). All the
 445 pTM and dTM concentrations measured during the cruise were within the range previously
 446 published for the Med Sea (Sherrell and Boyle, 1988; Saager et al., 1993; Morley et al., 1997; Yoon
 447 et al., 1999; Wuttig et al., 2013; Baconnais et al., 2019; Migon et al., 2020, GEOTRACES-
 448 IDP2021). However, dCo concentrations (from 10 to 20 pM) were among the lowest ones measured
 449 during stratification period in Western Med Sea (~120 pM, Dulaquais et al., 2017). Zn presented
 450 the largest range of concentrations within the ML both in the dissolved and particulate fractions (0.6
 451 to 19 nM and 61 to 3300 pM respectively), due to some high concentrations. However, the
 452 concentrations stayed in the typical range of values reported for the Med Sea (Bethoux et al., 1990,
 453 Yoon et al., 1999), even if we cannot exclude a possible contamination for these outlier
 454 concentrations.

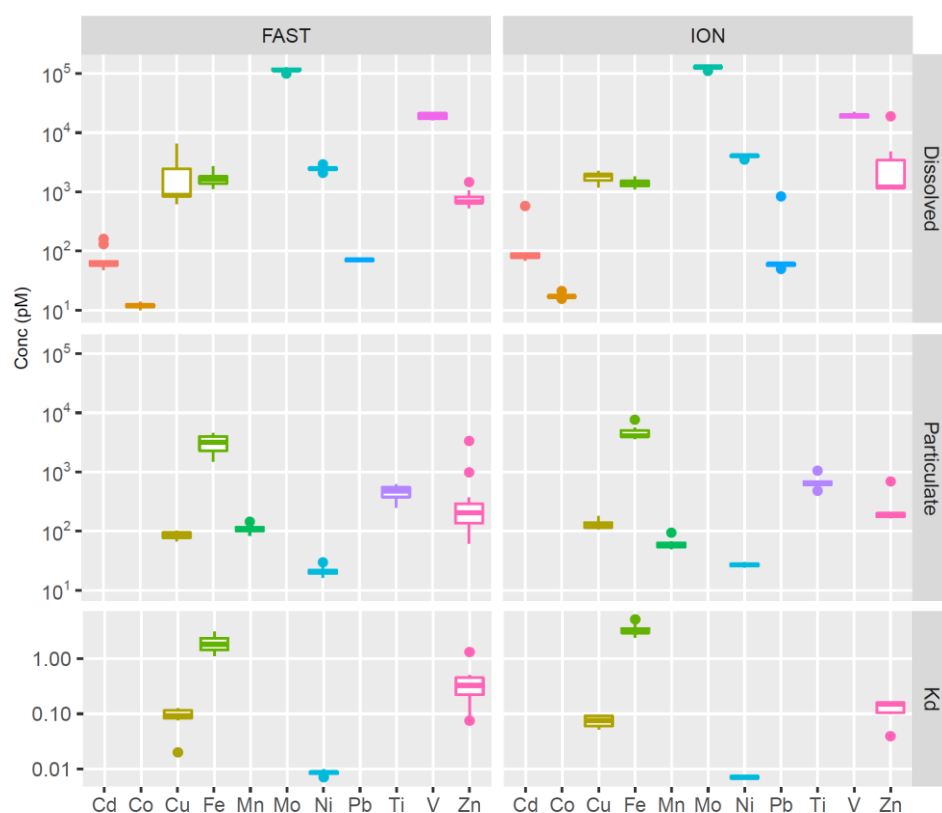


Figure 5: Box plots of dissolved (upper panels) and particulate (middle panels) marine concentrations (pM) and Kd values (lower panels) for the different TMs within the ML at ION (right panels) and FAST (left panels). In the box plots, the box indicates the interquartile range, i.e. the 25th and the 75th percentile, and the line within the box marks the median. The whiskers indicate the quartiles ± 1.5 times the interquartile range. Points above and below the whiskers indicate outliers outside the 10th and 90th percentile.

4. Discussion

4.1. Composition of rain collected over the remote Mediterranean Sea

4.1.1. Concentrations

Regarding nutrients, nitrogen species concentrations in rain samples were in good agreement with those reported in Mediterranean rain samples, ranging from 1130 to 5100 $\mu\text{g L}^{-1}$ for NO_3^- and between 207 and 1200 $\mu\text{g L}^{-1}$ for NH_4^+ (Loye-Pilot et al., 1990; Avila et al., 1997; Al Momani et al., 1998; Herut et al., 1999; Violaki et al., 2010; Izquieta-Rojano et al., 2016; Nehir and Koçak, 2018). The FAST rain concentrations were within the published range, whereas the ION rain was in the low range, confirming a background signature at this station. The rainwater samples presented a large dominance of N in comparison to P, as observed from the N/P ratio derived from DIN/DIP (Table 2) ranging from 208 at ION to 480 at FAST. Previous observations showed a predominance of N relative to P in the atmospheric bulk deposition over the Mediterranean coast, with average ratio about 100, the highest reaching 170 for DIN/DIP and 1200 for DIN/TDP (Markaki et al., 2010, Desboeufs et al., 2018). The highest ratio could be linked to a washout effect of gaseous N species (NO_x and NH_3) by rain (Ochoa-Hueso et al., 2011). At the two stations, observed NO_x concentrations were below the limit of detection in the boundary layer before wet deposition. The presence of nitrate and ammonium in the background aerosols has been observed during recent campaigns in the remote Mediterranean atmosphere (e.g. Mallet et al., 2019). To our knowledge, no data are available on both P and N concentrations in Mediterranean aerosols. The lowest concentrations of P relative to N in aerosol particles in the Mediterranean were observed during the cruise (DIN/TDP ranged from 13 to 790, Fu et al., in prep.). The TDP concentrations were consistent with the average value of 8.4 $\mu\text{g L}^{-1}$ measured in African dust rain samples collected in Spain over the 1996-2008 period (Izquierdo et al., 2012). Inorganic phosphorus predominated in the dust-rich rain, whereas organic P was dominant in the background rain as the contribution of DOP to the TDP was 60% and 44% in Rain ION and Rain FAST, respectively. The DOP/TDP ratio presents a very large range in Mediterranean rains, spanning from 6% in Spanish dusty rain samples (Izquierdo et

al., 2012) to 75-92% in rains from Crete (Violaki et al., 2018). A reason for this wide range could be that Mediterranean European aerosols, as opposed to Saharan dust particles, are dominated by organic phosphorus compounds associated with bacteria (Longo et al., 2014).

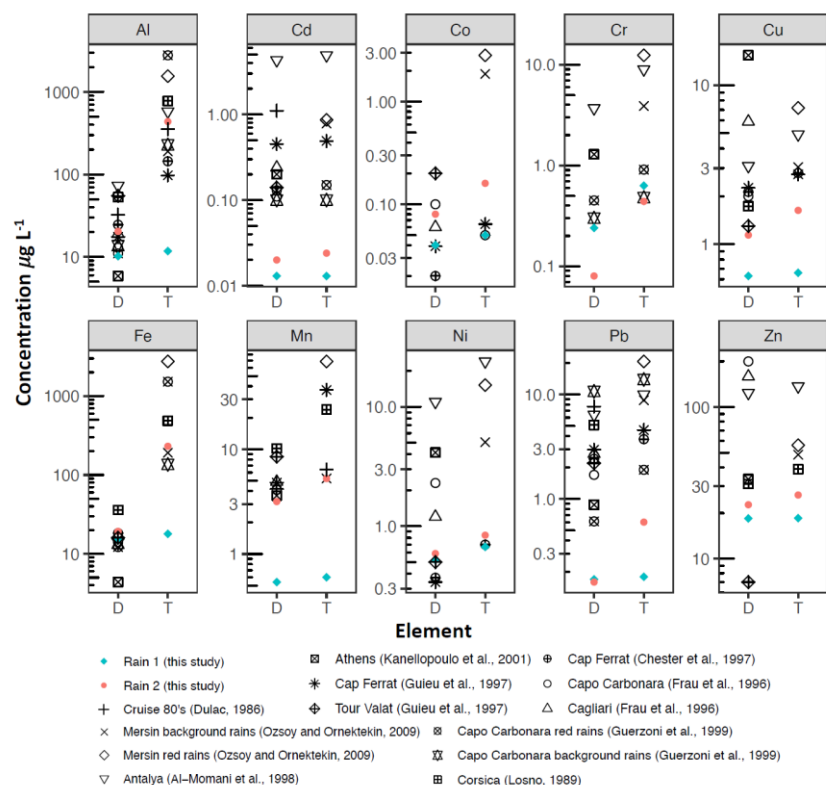


Figure 6: Comparison of dissolved (D) and total (T) TM concentrations to previous studies in the eastern and western Mediterranean Sea.

The dissolved and total TM concentrations in the PEACETIME rains were lower than those reported in coastal areas (eastern Basin: Özsoy and Örnektekin, 2009; Al-Momani et al., 1998; Kanellopoulou et al, 2001 and western Basin: Guieu et al., 1997; Guerzoni et al., 1999b; Chester et al., 1997; Losno, 1989; Frau et al., 1996), especially for the background Rain ION (Fig. 6). Most of the referenced works on coastal rainwaters date from the late 1990s. There is a continuous decline of TM emissions since ~~90's~~ 1990s due to regulatory efforts (Pacyna et al., 2007). The subsequent decrease of the anthropogenic Cd and Pb imprint on atmospheric inputs from European coasts to open sea is well documented (OSPAR, 2008; Travníkov et al., 2012; Geotraces IDP21). Since the

phasing-out of leaded automobile gasoline, the decrease of atmospheric Pb concentrations is also observed in Mediterranean atmosphere (Migon et al., 2008). The low TM concentrations of ION and FAST rain samples, in particular Cd and Pb, suggest a pronounced decrease of TM inputs in open Mediterranean due to environmental mitigation on TM emissions. Moreover, the coastal deposition is generally not representative of open sea inputs, e.g. due to proximity of anthropogenic sources in coastal areas. Thus, the ~~90s~~1990s data should not be used as a current reference for open Mediterranean rain composition.

4.1.2. Enrichment factor

EF and solubility values of TMs and P observed during the two rain events were very contrasted (Fig. 7). In Rain ION, almost all elements were significantly enriched relative to the upper continental crust ($EF > 10$, and up to $\sim 10^3$ for Cd and Zn), whereas in Rain FAST, only Zn (73), Cd (48) and Mo (15) were enriched. Only Ti, Fe, and Mn did not present a significant enrichment ($EF < 10$) in Rain ION, in agreement with previous studies in the Mediterranean environment showing that these metals are mainly associated with mineral dust in atmospheric deposition (e.g. Guieu et al., 2010; Desboeufs et al., 2018). Yet, for Fe and Mn, an influence of non-crustal sources in Rain ION could be suspected through a clear increase in the EF values compared to FAST. Mo is the most abundant TM in seawater (Smedley and Kinniburgh, 2017) and in particular in Med Sea (see section 3.3). Thus, at the difference of other TMs, the non-crustal part of Mo could be associated to seasalt aerosols rather than anthropogenic signal. The EF ratio relative to seawater ($(Mo/Na)_{seawater} = 8.9 \cdot 10^{-7}$ in mass ratio, Millero, 2013) were 7.4 for Rain ION and 4.6 for Rain FAST confirming the marine origin of this element in both rain samples. The anthropogenic origin of TMs and P have been reported by several studies on atmospheric deposition monitoring in the western Mediterranean (e.g., Guieu et al., 2010; Sandroni and Migon, 2002; Desboeufs et al., 2018). For example, Desboeufs et al. (2018) showed that there is a large contribution of anthropogenic combustion sources to the P, Cr, V and Zn background deposition fluxes. Aerosol composition monitoring over the Mediterranean coastal area showed the role of land-based sources and ship traffic sources on TM contents (Bove et al., 2016; Becagli et al., 2017). As all the deposition measurements sites were located in coastal areas, it was difficult to exclude the influence of these local sources for explaining the observed anthropogenic contribution. Here, EF values showed a clear anthropogenic signature for P and all TMs except Ti, Fe, Mn and Mo in the offshore Rain ION sample. In particular, the EF of Zn in Rain ION was on average five times higher than the EF found in the rain samples previously studied from coastal sites in the Mediterranean region (Özsoy and Örnektekin, 2009; Al-Momani et al., 1998; Losno, 1989). Nevertheless, extremely high enrichments

of Zn in rainwater have been reported from island sites in the Med Sea, for example Frau et al. (1996) reported geometric mean EF of ~6500 in both dust-rich and dust-poor rains from two sites in southern Sardinia, and Fu et al. (2017) reported EF >1000 for Zn in atmospheric insoluble bulk (wet+dry) deposition on Lampedusa Island. As previously discussed (section 3.1.1.), Rain ION was representative of a Mediterranean background marine rain event. The Zn EF at ION was the same order of magnitude as at these island sites which suggests a high anthropogenic background signal of Zn even in open Med. More generally, the high EFs in Rain ION mean that even over the remote Med Sea, the chemical composition of background aerosol particles is likely continuously impacted by anthropogenic sources.

The EF values of TMs for Rain FAST were significantly lower than for Rain ION (Fig. 7) but similar to Saharan rains (Guerzoni et al., 1999b; Özsoy and Örnektekin, 2009), confirming the dust signature for this rain. The comparison between dust-rich and background rains generally reveals a net difference of concentrations (at least higher by a factor 3 in dust-rich), notably for Al, Fe, Mn and Cr (Guerzoni et al., 1999b; Özsoy and Örnektekin, 2009). Here, an increase in concentrations between rains ION and FAST was observed for the majority of TMs: Al (x28), Ti (x50), Mo (x23), Fe (x13), Mn (x9), V (x5), Pb (x3.5), Co(x3) and Cu (x2) but also for P(x4) (Table 2). The combination of higher concentrations and EF values <10 found in Rain FAST show that the dust contribution was important on deposition fluxes of many TMs and P during this event. However, the high Al concentrations in rain FAST drives mathematically down EF values, masking potentially other source signature.

4.1.3. Solubility

The solubility values were also higher in Rain ION than in the dusty Rain FAST, except for Mo for which the difference between both rain samples was not significant (Fig. 7). For Rain ION, TMs and P presented solubility higher than 78%, except for Cr (38%). In Rain FAST, solubility values <10% were observed for Al and Fe, more than 10 times lower than in Rain ION. For the other TMs, the highest difference in solubility was observed for Pb whose solubility decreased from 97% in Rain ION to 27% in Rain FAST. In a review on TM solubility in Mediterranean rainwaters collected in coastal areas, Desboeufs (2021) emphasize the large range of solubility for all the TMs: Fe (0.8-41%), Cr (6-80%), Pb (5-90%), Ni (22-93%), Mn (16-95%), Cu (22-96%), Zn (14-99%), V (35-99%) and Cd (72-99%). The solubility ranges found in this study were generally consistent with those reviewed by Desboeufs (2021). Moreover, the Mn solubility values in FAST (60%) and ION (92%) rains are close to those reported by Dulac (1986) from a dust-rich (57%) and an

anthropogenic (83%) rain collected at sea in the Ligurian Sea and west of Sardinia in April 1981. Only Fe solubility (84%) found in Rain ION was higher than the average values previously reported. In the Rain FAST, Fe solubility was 8%, this is 10 times lower than the average Fe solubility in 10 dust-rich rains collected on the southeastern coast of Sardinia by Guerzoni et al. (1999b), but consistent with Saharan dust deposition collected in the Atlantic Ocean (Sedwick et al., 2007; Baker et al., 2013). It is known that anthropogenic Fe is more soluble than Fe-bearing dust (Desboeufs et al., 2005, Jickells et al., 2016). Regarding the evolution of TM emissions (see section 4.1.1), we suspect that this difference could be due to a higher contribution of anthropogenic signal for Fe in dust-rich rains in 90's-1990s in Sardinia that in the recent rain samples.



Figure 7: Enrichment Factors (EF, upper panel) and solubility (% , bottom panel) of phosphorus (P) and TMs ordered by increasing EF in the two rainwater samples.

Few studies have compared TM solubility between dust-rich and anthropogenic rains in Mediterranean. An increase in solubility values from dust-rich to background rains was observed for Mn in offshore rains as mentioned above (Dulac, 1986) and for Al, Cr, Fe and Pb (but only a slight increase for Cd) in Sardinia (Guerzoni et al., 1999b). The decrease in solubility from background to dust-rich rains was observed for P in Spain by Izquierdo et al. (2012), with values of

solubility decreasing from 25% to 7%. Here, our results on the two rain samples confirm the lowest solubility of TMs in dust-rich rains (except for Mo). Aerosol leaching experiments showed that metal dissolution from aerosol particles can be also influenced by a number of parameters, such as pH, presence of dissolved organic complexing ligands, in-cloud processing, particle origin and load (Desboeufs et al., 1999; Bonnet and Guieu, 2004; Desboeufs et al., 2005; Paris and Desboeufs, 2013; Heimbürger et al., 2013; Jickells et al., 2016). However, it is known that metals that are mainly associated with crustal aluminosilicate mineral lattices such as Fe and Ti have very low solubility values, due to the difficulties to breaking bonds in the lattices (Journet et al., 2008; Jickell et al., 2016). Regarding metal partitioning in rainwater, the role of pH and of the particulate mineral dust loading, reflecting the dust versus anthropogenic signature, were identified as the main controls of TM solubility in the Mediterranean rainwater (Özsoy and Örnektekin, 2009; Theodosi et al., 2010). From our two rain samples, it is difficult to propose a control explaining the difference in solubility values. However, the pH values were very close in the two samples (Table 2), excluding a pH effect on solubility values. A much lower solubility of TMs in Rain FAST is consistent with the EFs indicating a crustal origin of TMs in Rain FAST (Fig. 7). As discussed before, background atmosphere in Med Sea seems to be continuously influenced by anthropogenic particles. Even in dusty-rain FAST, it is highly probable that a part of metals presented an anthropogenic imprint, not visible on EF values but with solubility similar than in Rain ION. Thus, the decrease of solubility between the two rain samples could be either due to the lowest solubility of TMs in mineral dust (as suggested by aerosol leaching experiments), either results from the presence of mineral dust which by increasing the TM total concentrations overwhelms the anthropogenic background signal, or both. The case of Mo is unique, since its solubility was comparable in Rains ION and FAST. As discussed from EF values Mo was associated to seasalt aerosols in both rain samples, explaining the similarity of solubility.

4.2. Atmospheric wet deposition as a source of TMs to the surface seawater

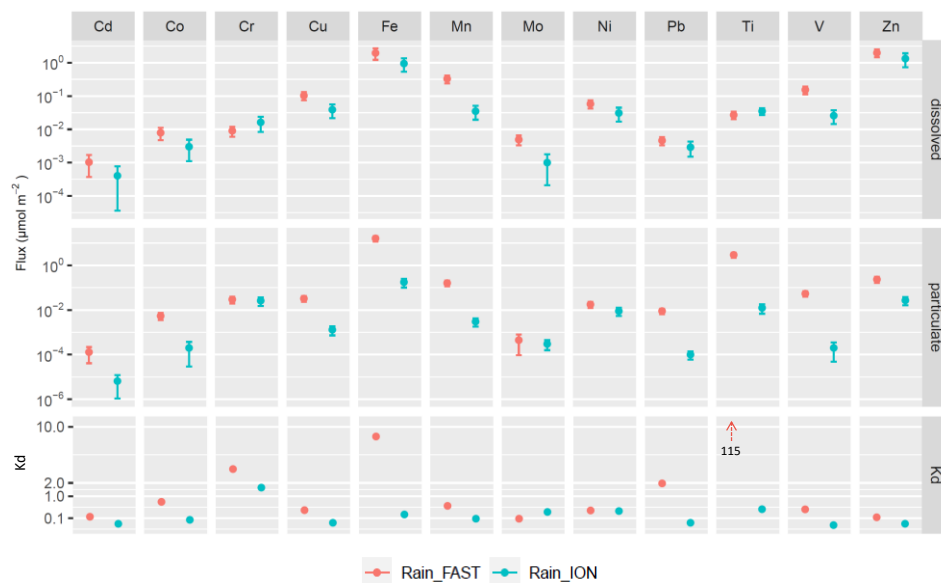
4.2.1. Atmospheric fluxes

As mentioned before, the two collected rains were part of large rain systems, associated with patchy rainfalls that lasted several hours or days (see section 3.1). This spatio-temporal variability led to heterogeneity in both rainwater concentrations and accumulated precipitation across the studied region. Such spatial variability has been observed by Chance et al. (2015) in the Atlantic Ocean. Moreover, even weak lateral advection can transfer surface water impacted by intense precipitation in the vicinity of the vessel. On this basis, the spatial extrapolation of wet deposition fluxes seems

616 subject to large uncertainties (almost 100% RSD) when the rain samples are not collected across
617 the rain area (Chance et al., 2015). To best counteract this effect, spatial variability was taken into
618 account to quantify the total precipitation i.e. 3.5 ± 1.2 mm for rain ION and 6.0 ± 1.5 mm for rain
619 FAST (see section 3.1) in order to quantify the wet deposition fluxes (here ~25% RSD for total
620 precipitation and 1 to 74% RSD for concentrations).

621 From the total (dissolved + particulate) Al concentration measured in the Rain FAST sample, we
622 estimated the wet mineral dust deposition flux at 65 ± 18 mg m⁻², assuming 7.1% Al in dust (Guieu
623 et al., 2002). The vertical distribution of dust particles (Fig. 3b) and the absence of high Al
624 concentrations close to the sea surface (Fu personal comm.) indicate that dust dry deposition can be
625 neglected. Based on the increase in total Al in the upper 20 m of the column water following the
626 deposition events and using 7.1% for Al in dust, Bressac et al. (2021) derived an average dust
627 deposition flux of ~55 mg m⁻² at FAST, which is comparable to our estimate. Although low
628 compared to deposition fluxes reported in the western Mediterranean (Bergametti et al., 1989; Lojze-
629 Pilot and Martin, 1996; TERNON et al., 2010), our flux estimates are in the same order of magnitude
630 of the most intense weekly dust deposition fluxes calculated more recently in Corsica between 2011
631 and 2013 (14% of fluxes >50 mg m⁻²) and is comparable to the mean weekly flux (93 mg m⁻²)
632 reported for Majorca during the same period (Vincent et al., 2016). The columnar aerosol
633 concentration during the dust event at FAST being estimated to be between 0.18 and 0.24 g m⁻² (see
634 section 3.1.), the expected maximum values of atmospheric dust flux could be in this range. The
635 comparison with the estimated flux indicates that the atmospheric column was probably not totally
636 washed-out by the short rain event. Indeed, Fig. 3b shows that a significant depolarization was
637 observed immediately after the rain ended on the R/V, before atmospheric advection could have
638 brought dusty air possibly not affected by rain. Satellite products (Fig. S2) confirm that on 5 June,
639 the dusty air mass was transported ~~farther~~[further](#) to the north-east from the station where it was
640 replaced by clear air.

641



642

643 **Figure 8: Dissolved (upper panels) and particulate (middle panels) wet deposition fluxes ($\mu\text{mol m}^{-2}$),**
 644 **and Kd (lower panels) for the different TMs estimated from the two rains sampled on-board,**
 645 **considering the standard deviation on the TM concentrations and the spatial variability of total**
 646 **precipitation over the area of sampling (Rain ION in blue and Rain FAST in red). Note different scales**
 647 **on the y axes and that Kd value for Ti (115) in Rain FAST is out of scale.**

648 The atmospheric dissolved and particulate wet deposition fluxes of TMs, derived from the chemical
 649 composition and total precipitation of rain samples, are presented in Fig. 8. Co, Mo and Cd
 650 presented the lowest fluxes in the two rainfalls. Zn and Fe fluxes were on the same order of
 651 magnitude and were the highest dissolved fluxes compared to the other TMs in the two rains. The
 652 Kd values ($0.1 < \text{Kd} < 1$), in consistence with solubility values higher than 50% (Fig. 7), show that
 653 whatever the rain sample, the atmospheric TM inputs were, mainly dissolved, except Fe, Pb and Ti
 654 in Rain FAST and Cr in the two rains (Fig. 8). The comparison shows that almost all the dTM fluxes
 655 were higher in the dusty rain, except Cr and Ti due to their low solubility in Rain FAST. Our results
 656 suggest that dust deposition resulted in higher atmospheric inputs of TMs than from low perturbed-
 657 anthropogenic background rain, even here in the case of the moderate deposition input flux reported.
 658 With more than an order of magnitude difference in deposition fluxes [between](#) the two rain events,
 659 this most notably be the case for particulate Fe, Mn, Pb and Ti, typically poorly soluble from desert
 660 dust source. Dissolved Cd, Co, Cu, Mn and V fluxes present the highest increases between

background Rain ION and dusty Rain FAST. Yet, these former elements (except Mn) are usually considered issued from anthropogenic sources. The orders of magnitude found in this study could be used as a benchmark to estimate atmospheric inputs of TMs from wet deposition to the western Med Sea. However, we must keep in mind that annual and long-term deposition fluxes of desert dust-related elements, as Fe, Mn and Ti (e.g. TERNON et al., 2010), but also nitrogen species (e.g., Richon et al., 2018b), are dominated by a few atypical intense deposition events in the Med Sea, as is the case in many other oceanic regions (Duce et al., 1991).

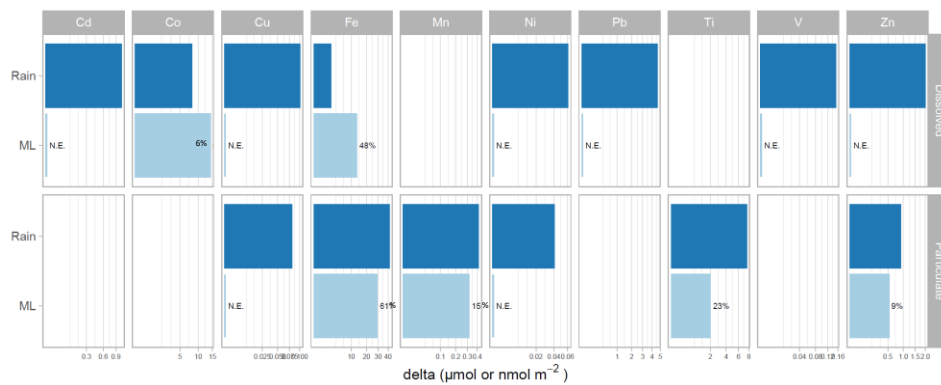
4.2.2. Comparison between TM wet deposition inputs and marine stocks at FAST

~~As previously discussed, the atmospheric concentrations measured in this study were the most recent measurements in remote Med Sea, and were inferior to 90's measurements in coastal zone.~~

As described in Guieu et al. (2020), marine dynamic conditions at FAST were favourable to observe any change in the water masses strictly attributed to external inputs coming from the atmosphere on a short time scale. The impact of the dust wet ~~depositions~~deposition on nutrients stocks in the Mediterranean surface waters is discussed in details in van Wanbeke et al. (2020) and Pulido-Villena et al. (2021). To briefly summarise, both nitrate and DIP increased in the ML following the rain. Although the closure of the N and P budgets had to necessarily take into account post-deposition processes such as new nutrient transfer through the microbial food web (uptake, remineralisation, and adsorption/desorption processes on sinking particles), it was shown that wet deposition was a significant source of nutrients for the ML during the cruise. For example, atmospheric supply of phosphate could contribute to 90% of new production at FAST (Pulido-Villena et al., 2021). Bressac et al. (2021) studied the response to Al and Fe cycles to dust deposition during the cruise. They showed that total Fe and Al stocks were increased by dust wet deposition, and that the dissolved Fe atmospheric inputs were transient in the ML and were accumulated in the subsurface waters (100-1000m). The low depth-resolution of marine TM concentration samplings prevent the possibility to make TM inventories on the water column. We focus here on the role of dust wet deposition events as a source of TMs to the surface mixed-layer, except Mo due its marine origin in the rain samples. The deltas of TM stocks before and after Rain FAST compared to the atmospheric fluxes are presented in Fig. 9.

After Rain FAST, ML stocks increased for pFe (+61%), pMn (+15%), pTi (+23%) and pZn (+9%) and for dCo (+6%) and dFe (+46%) (Fig. 9). The behaviour of pFe, pMn, pTi and pZn is consistent with observations on pAl stock which increased by 78% after rain (Bressac et al., 2021). The atmospheric particulate inputs were higher to the marine particulate deltas for all the TMs (Fig. 9),

693 supporting that the marine stock increases resulted from atmospheric inputs. The pTM inputs by
 694 rain was also observed from Kd increase in the ML for Fe (1.7 to 1.9) and for Zn (0.33 to 0.57), in
 695 agreement with the highest Kd values in rain fluxes (Fig. 7) relative to marine ratio (Fig. 4). The
 696 increase Kd was even observed for Cu (0.044 to 0.091) and Ni (0.0081 to 0.0086) for which no
 697 change in stock could be evidenced. It is probable that the rain inputs for these TMs be masked in
 698 the uncertainties of stock estimations. As dissolved concentrations of Mn and Ti were not measured
 699 in the ML, it is not possible to estimate the Kd values. However, the particulate Mn/Al ratio fell
 700 from 0.27 before the rain to 0.008 after rain, in accordance with the rain ratio (0.002), confirming
 701 the impact of rain inputs on marine particulate stocks of Mn and Al. For Ti, the particulate Ti/Al
 702 ratios were very close in the ML (0.048) and in the rain (0.045), making it impossible to observe a
 703 rain effect. Here, the observed deltas were ~~inferior to~~ less than the rain inputs. However, the marine
 704 sampling were 4h after rain. Bressac et al. (2021) showed a decrease of 40% of pAl signal in 24h
 705 which could be explained either by a rapid settling of dust in the ML, or by lateral advection. The
 706 dynamic of pFe, pMn and pTi stocks, with a decrease with time after rain (not shown), is consistent
 707 with dust removal in the ML. The pCu and pNi inputs represented 4% and 10% of particulate marine
 708 stocks. The dust removal could overwhelm the signal of enrichment of these metals. However, we
 709 cannot exclude that the pCu and pNi inputs were masked by the uncertainties of stock calculations.



710
 711 **Figure 9: Comparison between TM wet deposition fluxes (dark blue) and TM marine stock deltas**
 712 **(before and after the rain) in the ML (light blue) at FAST. Dissolved = upper panels and particulate =**
 713 **lower panels. Marine stocks increase are expressed in absolute values (Cd, Co and Pb stocks and fluxes**
 714 **in nmol m⁻², and the other TMs in μmol m⁻²) and in relative values (%). N.E.: not enhanced (increase**
 715 **<5%).**

716 For dissolved stocks, no enrichment was observed for Cd, Cu, Ni, Pb and Zn. This is not surprising
 717 for Cd, Cu, Ni, Pb and V since atmospheric dissolved inputs represent less than 1% of marine stocks.
 718 As expected from the comparison between inputs and stocks before rain, the marine dCo and dFe
 719 stocks were increased after dust deposition. However, the marine deltas within the ML for Co and
 720 Fe was 2 and 7 times higher than what could be provided from the atmospheric dissolved inputs. A
 721 lateral transport and cumulative effect of previous and surrounding wet deposition events (from 3
 722 June) could explain this positive mismatch. Several removal processes occurring in the ocean, such
 723 as biological uptake (Morel et al., 2003; Noble et al., 2008), passive scavenging by adsorption onto
 724 particles (Wuttig et al., 2013, Bressac and Guieu., 2013, Mackey et al., 2015; Migon et al., 2020)
 725 or chemical processes as precipitation (Wagener et al., 2010, Ye et al., 2011) are known to decrease
 726 dissolved TM concentrations in the first 24 ~~or to~~ 48 hours after atmospheric deposition. During the
 727 cruise, Bressac et al. (2021) show that dust deposition represented a significant input of dissolved
 728 Fe in the ML only on a timescale of hours, due to scavenging or precipitation processes. It is highly
 729 probable that these removal processes limited the cumulative effect of dTM inputs. A post-
 730 deposition dissolution of Fe and Mn in seawater was reported during dust seeding experiments
 731 simulating wet dust deposition in mesocosms over several hours or days (Wagener et al., 2008;
 732 Wuttig et al., 2013; Desboeufs et al., 2014). Mackey et al. (2015) show that in case of dry deposition,
 733 aerosol Co and Fe dissolution in seawater can be gradual and continue up to 7 days after contacting
 734 seawater. Considering the excess of TMs in marine stocks relative to rain inputs, our observations
 735 are consistent with the dissolution of Co and Fe in surface seawater following the first hours after
 736 wet dust deposition. Tovar-Sanchez et al., (2020) proposed that UV irradiation and the enrichment
 737 in organic matter in the SML could enhanced the dissolution processes and the diffusive transfer
 738 between SML and surface ~~sea-water~~seawater. They estimated that the residence times ranged from
 739 23 min. for pFe to 3.5 h for pCo in the SML after Rain FAST. From these timescales, the
 740 contribution of dissolution then diffusive process of dCo and dFe from SML to ML in the post-
 741 depositional dissolution processes could be considered. The increase of dissolved stocks suggested
 742 also that in the first hours after rain the dissolution processes were predominant on removal
 743 processes in the behaviour of dCo and dFe. Co and Fe being known to limit phytoplankton biomass,
 744 a biological uptake could be predominant if marine biota was limited for these elements. This was
 745 probably not the case here. The behaviour of Zn is more complicated to interpret since no
 746 enrichment was observed whereas the atmospheric dZn inputs represented around 12% of marine
 747 stocks. This unexpected behaviour could be explained by the predominant role of removal processes
 748 relative to post-depositional dissolution. The high solubility of Zn in Rain FAST (~86%), relative

749 to solubility values of Fe (8%) or Co (52%), suggests that all the soluble Zn was already dissolved
750 in the rain inputs before contacting with seawater.

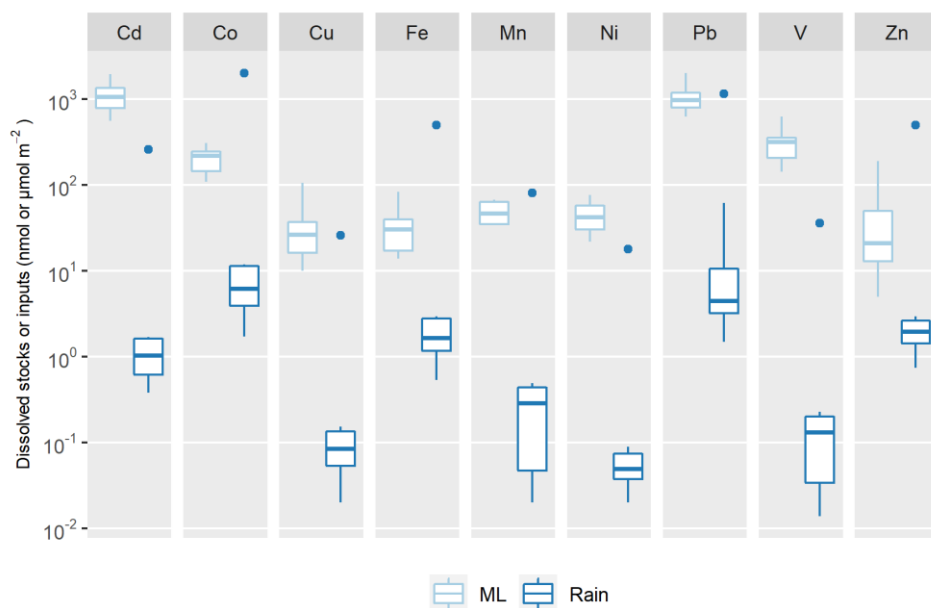
751 Finally, our results show that the studied atmospheric dust event was a net source of particulate
752 TMs and dissolved Fe and Co for the ML at FAST. Due to various marine post-deposition processes,
753 it is more complicated to observe the effect of wet deposition on dissolved stocks. Even if the wet
754 deposition delivered TMs already as soluble forms (Fig. 8), our results showed that the wet
755 deposition constitutes only a source of some of dissolved TMs for surface waters ~~in Med Sea~~. The
756 post-deposition dissolution of particulate rain inputs could represent an additional pathway of
757 dissolved TM supply for the surface ocean, notably for low soluble TMs in wet deposition. Thus,
758 ~~using only~~ the dissolved atmospheric ~~inputs fluxes~~ could ~~be~~ underestimate ~~from the only~~
759 ~~measurements dissolved atmospheric inputs of atmospheric fluxes TMs to ML inventories~~. On a
760 timescale of hours, the Fe inventory was the most impacted by the dusty rain input, both in dissolved
761 and particulate phases, confirming that the dust-rich rains are an external source of Fe to the surface
762 Med Sea (Bonnet and Guieu, 2006, Bressac et al., 2021).

763

764 4.2.3. Comparison between TM wet atmospheric inputs and marine stocks in the western and central 765 Mediterranean Sea

766 As discussed from dissolved TM stocks at FAST, high surface marine TM stocks masked any
767 additional input for several TMs. However, the collected rains (FAST and ION) were originating
768 from large rain systems covering more than 50 000 km² around the sampling zone and were typical
769 of Mediterranean wet deposition. The wet deposition could have ~~so~~ occurred in any of the explored
770 areas during the cruise. Here we further study the role of wet deposition by comparing atmospheric
771 dissolved fluxes to marine dissolved stocks from TM profiles in the ML at all 13 marine stations,
772 i.e. 22 ML samplings, throughout the whole cruise (Fig. 10). Exceptional intense dust deposition
773 events have been recorded in the Mediterranean, reaching 20 g m⁻² (Bonnet and Guieu, 2006).
774 Sporadic and intense wet dust deposition higher than 1 g m⁻² are observed in the spring in the
775 western Mediterranean basin (e.g., Vincent et al., 2016). At the beginning of the cruise, an intense
776 wet dust deposition event (not collected) occurred over the South of Sardinia and over the
777 Tyrrhenian Sea with fluxes reaching about 9 g m⁻² (Bressac et al., 2021). In order to take into
778 account the effect of such intense event, we also estimated the atmospheric fluxes of dissolved TMs
779 based on a 9 g m⁻² wet dust deposition event using solubility values estimated from Rain FAST
780 (Fig. 10). Fe, Mn and Pb solubility decreases with increasing dust load in Mediterranean rain

781 samples (Theodosi et al., 2010), suggesting that this estimation ~~is probably~~ could be a maximum
782 flux for such deposition event. However, recent studies from aerosol collected in Atlantic Ocean
783 showed that Co and Mn solubility was little affected by dust load ~~at the difference of~~ in contrast to
784 Fe (Baker et al., 2020). The impact of rain inputs on TM marine stocks is also controlled by MLD
785 fluctuations which we ignored in this work by using a fixed ML depth at FAST and ION. As the
786 variability of this MLD (7-21 m during the cruise, typical of Med thermal stratified period) could
787 change the marine budgets by a factor of 3, we used the measured MLD (Van Wambeke et al.,
788 2020) at each station for calculating the TM marine stocks.



789
790 **Figure 10: Comparison of marine stocks in the ML at all the stations occupied during the**
791 **PEACETIME cruise (light blue boxes) with atmospheric deposition fluxes estimated (1) from ION and**
792 **FAST rains (dark blue boxes) and (2) from an intense wet dust deposition event of 9 g m⁻² (blue dots).**
793 **Cd, Co and Pb stocks and fluxes are in nmol m⁻², and the other TMs in μmol m⁻². For Mn, marine**
794 **stocks are derived from surface concentrations close to Corsica coasts (Wuttig et al., 2013: samples**
795 **OUT at 0, 5 and 10 m) and in the Ionian Sea (Saager et al., 1993: Bannock basin at 0, 10, and 15 m),**
796 **as no measurement is available from the PEACETIME cruise. Boxes and whiskers as in Fig. 4.**

797 Applying to the whole transect, the atmospheric inputs, obtained from our rain composition, were
798 at least 100-fold smaller than the dissolved stocks in the mixed layer, except for Co, Fe and Zn. The

799 atmospheric inputs represented more than 30% of the dissolved Zn stocks and 10 to 18% for Fe.
800 For Co, the maximum atmospheric fluxes estimated during the cruise represented 5% of stocks.
801 This significant input of dFe and dCo is in agreement with our field observations in the FAST ML.
802 Here the comparison is based only on dissolved TMs in rain water, yet as discussed previously, the
803 post-deposition [depositionprocessing](#) of atmospheric particles in the water column could further
804 enrich the marine dissolved stocks for low soluble TMs, as Co, Fe, Mn, Ni, Pb or Ti. The surface
805 seawater could be significantly affected by the deposition of these dissolved elements in the case of
806 wet dust deposition. In the case of the intense dust deposition event, the dissolved inputs are of the
807 same order of magnitude as marine stocks for Co, Fe, Mn, Pb and Zn. The enrichment in dFe and
808 dMn was previously observed by Wuttig et al. (2013) after artificial dust seeding in large
809 mesocosms (simulating a wet deposition event of 10 g m^{-2}). The marine TM concentrations
810 measured during the cruise being typical of Mediterranean surface seawater concentrations, we can
811 conclude that wet deposition events were an external supply of dissolved Fe, Co and Zn for the Med
812 Sea, and more generally for all TMs in case of intense wet dust deposition, during the period of
813 thermal stratification.

814 5. Conclusions

815 This study provides both the dynamic properties and chemical characterization of two rain events
816 collected in the open Mediterranean Sea, concurrently with TM marine stocks in surface seawater.
817 Our results are the only recent report of TM concentrations, EFs and fractional solubility values in
818 rain samples collected in the remote Med Sea. By highlighting the discrepancy between TM
819 concentrations with the previous offshore and coastal rain studies, this work demonstrates the need
820 to provide a new and recent database on TM composition in Mediterranean rains in order to estimate
821 the role of atmospheric TM deposition. We have shown the representativeness of Rain FAST as
822 typical of Saharan dust wet deposition in its chemical composition as well in its magnitude and
823 extent, whereas Rain ION is more-typical of an anthropogenic background rain for the remote Med
824 Sea. On this basis, we suggest using the chemical composition of PEACETIME rains as a new
825 reference for studies of TMs on wet deposition in the Med Sea. This study was focused on wet
826 deposition, yet dry deposition is also an important source of TMs to surface waters in Med Sea
827 (Theodosi et al., 2010). As the wet deposition fluxes decreased since the [90's1990s](#) due to
828 mitigation, it is highly probable that the dry deposition fluxes were also changed. Further
829 measurements of dry deposition in open Med Sea are needed in order to estimate its contribution in
830 TM atmospheric inputs.

831 Since atmospheric TMs have been identified as critical oligo-nutrients for the marine biosphere, it
832 is important to study the response of the receiving waters to atmospheric inputs. This study is the
833 first to provide *in situ* evidence that atmospheric wet deposition constitutes a significant external
834 source for some of these elements to surface stratified Mediterranean seawater. We recommend that
835 the original approach developed here is used in other parts of the world where atmospheric wet
836 deposition is thought to impact the marine biosphere, such as in HNLC (High Nutrient, Low
837 Chlorophyll) regions.

838

839

840 **Data availability.** Guieu et al., Biogeochemical dataset collected during the PEACETIME cruise. SEANOE.
841 <https://doi.org/10.17882/75747> (2020). Atmospheric Data are accessible on [http://www.obs-](http://www.obs-vlfr.fr/proof/php/PEACETIME/peacetime.php)
842 [vlfr.fr/proof/php/PEACETIME/peacetime.php](http://www.obs-vlfr.fr/proof/php/PEACETIME/peacetime.php).

843 **Author contributions.** KD and FF designed the study and wrote the manuscript; FF, ST, JFD, Ch.G made
844 the on-board atmospheric measurements and sampling during the cruise; FF, ST and JD analysed the rain
845 samples; MB, ATS, and ARR made the marine TMs sampling and analyses; PF was the reference scientist
846 of PEGASUS, AF and FM managed all the technical preparation of atmospheric samplings, PC analysed the
847 lidar data; KD, FD and Ce.G designed the cruise strategy; KD and Ce.G coordinated the PEACETIME
848 project, FD coordinated the ChArMEx funding request, and near-real time and forecast survey of
849 atmospheric conditions during the cruise; all the authors commented on the manuscript and contributed to
850 its improvement.

851 **Competing interests.** The authors declare that they have no conflict of interest.

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853 low-chlorophyll (LNLC) ocean: effects on marine life today and in the future (ACP/BG inter- journal SI)”.
854 It is not associated with a conference.

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873

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1207 **Figure Captions:**
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1209 **Figure 1: Sampling chronology during the ION and FAST stations for the ML and rains.** The blue
1210 periods correspond to rainfall in the vicinity of the R/V position (from ERA 5 reanalysis and radar
1211 imagery, see section 3.1.), and blue triangles to the rainfall on the R/V position. Samplings were
1212 performed 4 days (ML25) and 2 days (ML27) before and 2 h (ML29) after Rain ION, and at a higher
1213 frequency at FAST station: 57 (ML02), 37 (ML03) and 7.5 (ML04) hours before and 4.5 (ML05+4), 12
1214 (ML05+12), 24 hours (ML06) after Rain FAST.

1215 **Figure 2: Total precipitation (mm) between 28 May at 20:00 UTC and 29 May 2017 at 10:00 UTC**
1216 **(Rain ION, left figure) and between the 4 June at 20:00 and 5 June 2017 at 09:00 UTC (Rain FAST**
1217 **right figure) from ERA5 ECMWF reanalysis. The red circle indicates the R/V position.**

1218 **Figure 3: On-board lidar-derived: apparent backscatter coefficient (ABC) (top panel), temporal**
1219 **evolution of the lidar-derived volume depolarization ratio (VDR) (middle panel in local time) where**
1220 **the dust plume is highlighted for values higher than ~1.7 (yellow to red colours) and the rain by values**
1221 **higher than 3 (indicated by the white arrow), and vertical profiles of the aerosol extinction coefficient**
1222 **in cloud free condition, integrated over 3 periods, noted 1, 2 and 3 on the top panel, along the dust**
1223 **plume event (bottom panels). The grey shade represents the root mean square (rms) variability along**
1224 **the time of the measurement. The dust layer is highlighted on the profiles. The mean aerosol optical**
1225 **thickness is given in the boxed legend with its temporal variability (1 sigma). The location of the marine**
1226 **boundary layer (MBL) is also pointed.**

1227 **Figure 4: Rain rates (mm/h) during the night between the 4 and 5 June, when Rain FAST was collected**
1228 **on-board, issued from European rain radar composites (OPERA programme) of 5 June between 00:00**
1229 **and 02:45 UTC.**

1230 **Figure 5: Box plots of dissolved (upper panels) and particulate (middle panels) marine concentrations**
1231 **(pM) and Kd values (lower panels) for the different TMs within the ML at ION (right panels) and**
1232 **FAST (left panels). In the box plots, the box indicates the interquartile range, i.e. the 25th and the 75**
1233 **th percentile, and the line within the box marks the median. The whiskers indicate the quartiles ± 1.5**
1234 **times the interquartile range. Points above and below the whiskers indicate outliers outside the 10th**
1235 **and 90th percentile.**

1236 **Figure 6: Comparison of dissolved (D) and total (T) TM concentrations to previous studies in the eastern and**
1237 **western Mediterranean Sea.**

1238 **Figure 7: Enrichment Factors (EF, upper panel) and solubility (% , bottom panel) of phosphorus (P) and TMs**
1239 **ordered by increasing EF in the two rainwater samples.**

1240 **Figure 8:** Dissolved (upper panels) and particulate (middle panels) wet deposition fluxes ($\mu\text{mol m}^{-2}$),
1241 and Kd (lower panels) for the different TMs estimated from the two rains sampled on-board,
1242 considering the standard deviation on the TM concentrations and the spatial variability of total
1243 precipitation over the area of sampling (Rain ION in blue and Rain FAST in red). Note different scales
1244 on the y axes and that Kd value for Ti (115) in Rain FAST is out of scale.

1245 **Figure 9:** Comparison between TM wet deposition fluxes (dark blue) and TM marine stock deltas
1246 (before and after the rain) in the ML (light blue) at FAST. Dissolved = upper panels and particulate =
1247 lower panels. Marine stocks increase are expressed in absolute values (Cd, Co and Pb stocks and fluxes
1248 in nmol m^{-2} , and the other TMs in $\mu\text{mol m}^{-2}$) and in relative values (%). N.E.: not enhanced (increase
1249 $<5\%$).

1250 **Figure 10:** Comparison of marine stocks in the ML at all the stations occupied during the
1251 PEACETIME cruise (light blue boxes) with atmospheric deposition fluxes estimated (1) from ION and
1252 FAST rains (dark blue boxes) and (2) from an intense wet dust deposition event of 9 g m^{-2} (blue dots).
1253 Cd, Co and Pb stocks and fluxes are in nmol m^{-2} , and the other TMs in $\mu\text{mol m}^{-2}$. For Mn, marine
1254 stocks are derived from surface concentrations close to Corsica coasts (Wuttig et al., 2013: samples
1255 OUT at 0, 5 and 10 m) and in the Ionian Sea (Saager et al., 1993: Bannock basin at 0, 10, and 15 m),
1256 as no measurement is available from the PEACETIME cruise. Boxes and whiskers as in Fig. 4.

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