Secondary ice production during the break-up of freezing water drops on impact with ice particles

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Abstract. We experimentally investigated collisions of supercooled water drops (~5 mm in diameter) with ice particles of a similar size ($\sim 6 \text{ mm in diameter}$) placed on a glass slide at temperatures $T \geq > -12$ °C. Our results showed that secondary drops were generated during both the spreading and retraction phase of the supercooled water drop impact. The secondary drops generated during the spreading phase were emitted too fast to quantify. However, quantification of the secondary drops

5 generated during the retraction phase with diameters > 0.1 mm showed that 5–10 secondary drops formed per collision, with approximately 30% of the secondary drops freezing over a temperature range of between -4 °C $-\leq$ T \leq to -12 °C. Our investigation provides the first dedicated laboratory study of collisions of supercooled water drops with ice particles as a secondary ice production mechanism. Our results suggest that this secondary ice production mechanism may be significant for ice formation in atmospheric clouds containing large supercooled drops and ice particles.

10 1 Introduction

Most surface rainfall events that occur across the globe are associated with the ice-phase within clouds in the Earth's atmosphere (Field and Heymsfield, 2015), as are severe weather events such as freezing rain, hail and thunderstorms (Changnon, 2003; Púčik et al., 2019; Elsom, 2001). Therefore, understanding the processes which govern ice formation in clouds is crucial for determining their effects on both climate and weather.

- Where <u>subzero</u> temperatures are warmer than the homogeneous freezing point, T > of -35 °C, supercooled water drops can heterogeneously freeze via <u>solid a subset of aerosol</u> particles present in the atmosphere. These solid This <u>subset of aerosol</u> particles, called ice nucleating particles (INPs), are relatively rare, and while number concentrations of INPs vary in time and space, they <u>are typically less than typically fall between 1 × 10⁻⁵ L⁻¹ to 1 L⁻¹ at temperatures $T \sim \sim -10$ °C (Kanji et al., 2017). Yet, observed ice particle concentrations in mixed–phase clouds can be several orders of magnitude higher than</u>
- 20 concentrations predicted from ice particles forming due to INPs (e.g. Crawford et al., 2012; Lloyd et al., 2015; Lasher-Trapp et al., 2016; Ladino et al., 2017). Ice can also form at <u>T-temperatures</u> > -35 °C via secondary ice production (SIP) where new ice particles are formed from pre-existing ice particles. However, our understanding of ice formation from SIP mechanisms is incomplete (e.g. see reviews by Field et al., 2017; Korolev and Leisner, 2020), resulting in poor representation of SIP

mechanisms in numerical weather prediction (NWP) models. As such, NWP models underestimate the concentrations of ice

25 particles in mixed-phase clouds, severely for cumuliform clouds (Crawford et al., 2012).

Several SIP mechanisms have been identified and studied both in the laboratory and theoretically, but only the rime-splintering SIP mechanism is widely implemented in NWP models. Active between -3 °C $\leq T \leq t_0$ -8 °C, rime-splintering occurs when supercooled water drop diameters are < 13 µm or and > 24 µm (Hallett and Mossop, 1974; Mossop and Hallett, 1974; Mossop, 1978). Another SIP mechanism, the fragmentation of freezing drops, has received a significant proportion of

- 30 laboratory based SIP investigations. Fragmentation due to freezing drizzle drops or raindrops can occur over a wider temperature range between 0 °C <T < to -32 °C than rime-splintering, but quantification between laboratory studies varies significantly (see Table 1 of Korolev and Leisner, 2020, for a summary). A range in diameters of freezing supercooled water drops has also been investigated between laboratory studies from 4 μ m to 1000 μ m (see Table 1 of Korolev and Leisner, 2020, for a summary). While other SIP mechanisms exist (e.g. ice-ice collisions, sublimation fragmentation), the attention , especially of laboratory
- 35 studies , has overwhelmingly focussed on the <u>SIP mechanisms of rime</u>-splintering and fragmentation due to freezing dropsSIP mechanisms. Furthermore, unidentified SIP mechanisms may also exist.

In this paper, we present a SIP mechanism involving the formation of secondary drops from the collision of a supercooled water drop with <u>an a larger</u> ice particle. This SIP mechanism has been investigated via a theoretical study by Phillips et al. (2018) , but not directly via laboratory studies , although indirect laboratory studies of who referred to it as 'Mode 2' as it

- 40 involves collisions of supercooled water drops with more massive ice particles resulting in fragmentation of the supercooled water drop. Ice contained in some of the secondary drops was assumed to initiate freezing, yielding secondary ice fragments. By contrast, 'Mode 1' involved either collisions of supercooled water drops with less massive ice particles resulting in spherical freezing of the supercooled water drop or activation of immersed INPs, with a quasi-spherical outer ice shell that fragments. While there are no dedicated laboratory studies of this SIP mechanism involving collisions of supercooled water drops
- 45 with more massive ice particles or activation of INPs immersed in them, there are laboratory studies that have indirectly studied aspects of this processdo exist. For example, a similar mechanism was alluded to by Latham and Warwicker (1980) in their experimental investigation of charge transfer during interactions between hailstones and supercooled water drops. They observed that frost could occasionally be broken during impact, thus forming new ice particles. Although this was an unwanted outcome of their experiments it provided some hints of a potential SIP mechanism during the interactions between ice particles
- 50 and supercooled raindrops. Later, Schremb et al. (2018) studied the fluid flow and solidification of supercooled water drops on elevated ice targets, briefly observing the formation of secondary drops from the rim of the supercooled water drop during impact. However, for both of these studies no quantification of the secondary drops was made.

Observations within mixed-phase clouds often show ice crystal number concentrations higher than the numbers of ice nucleating particles present in the atmosphere. For instance, ice particle number concentrations exceeding $100 L^{-1}$, in shallow

55 convection with cloud-top temperature no lower than -12 °C, have been observed over the UK (Crawford et al., 2012). Furthermore, thin mixed-phase layer clouds have been observed to continually generate snow (Westbrook and Illingworth, 2013). Conventional thinking would suggest that the ice in mixed-phase layer clouds should fall out, leaving the layer 'depleted' of INPs; however, the observations clearly show that ice continues to form in these clouds over time.

The rime-splintering SIP mechanism has been successful in predicting the glaciation of mixed-phase clouds in many cases,

- 60 especially those involving a warm cloud base creating sufficiently large cloud-drops in the rime-splintering temperature region between -3 °C to -8 °C (e.g. Harris-Hobbs and Cooper, 1987; Blyth and Latham, 1993, 1997; Phillips et al., 2001, 2005; Crosier et al., 201 . However, there are also numerous cases where significant concentrations of ice observed in clouds cannot be explained by the rime-splintering SIP mechanism. Hobbs and Rangno (1985) compiled tables of aircraft observations from a wide range of cloud environments. They found that the maximum ice particle concentrations were independent of the cloud-top temperature
- 65 but were strongly dependent on the broadness of the supercooled drop spectrum near the cloud-top, with approximately half of the clouds exhibiting ice enhancement.

In this paper, we describe a set of experiments performed at the University of Manchester to determine the freezing fraction of secondary drops (Φ, formed) formed in the splash during the collision of a mm sized 5 mm diameter supercooled raindrop on an a 6 mm diameter ice particle, providing the first laboratory quantification of this SIP mechanism. The theory of collisions

70 of a supercooled water drops with an ice particle from Phillips et al. (2018) is given in Section ?? This freezing fraction (Φ) is the ratio of secondary drops the freeze to all such drops emitted. The experimental setup is described in Section 2. The results are presented in Section 3, and the discussion in Section 4. Finally, the conclusions are given in Section 5.

2 Theory

The theoretical study by Phillips et al. (2018) considered the fragmentation of freezing water drops during collisions of supercooled

75 water drops with ice particles more massive, referred to as 'Mode 2'. When collided with more massive ice particles, the spherical symmetry of a supercooled water drop is not maintained, with the potential to fragment at impact, emitting secondary drops. Due to a lack of laboratory investigations on this SIP mechanism, Phillips et al. (2018) used dimensionless energy (DE) as described by Testik et al. (2011) to determine if fragmentation occurred based on the ratio between the kinetic energy (K_0) at impact and surface energy:

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$$K_0 = \frac{1}{2} \left(\frac{m \times m_i}{m + m_i} \right) (v - v_i)^2$$

$$DE = \frac{K_0}{\gamma_{liq}\pi D^2}$$

where *m* is the mass of the supercooled drop, m_i is the mass of the ice particle, *v* is the velocity of the supercooled water drop, v_i is the velocity of the ice particle, γ_{lia} is the surface tension of liquid water and D is the diameter of the drop.

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A power law was fitted to the experimental data of ? for a 2.5 mm water drop impacting a rough copper hemisphere at room temperature to estimate the number of secondary drops formed, $N_s = 3 \times max(DE - DE_{crit}, 0)$ where $D_{crit} \sim 0.2$. After accounting for the fractional depletion of liquid water at the end of stage 1 of freezing, the number of secondary drops (N_s) formed in the collision of a supercooled water drop with an ice particle more massive was determined. This yielded the number of secondary ice particles per collision (N_t) :

where f is the initial fraction by mass of a drop frozen at stage 1 of freezing $f(T) = -c_w T/L_f$ given that c_w is the specific heat capacity of water, T is the initial freezing temperature (°C) and L_f is the specific latent heat of freezing; and DE_{crit} is equal to ~0.2. Finally, Phillips et al. (2018) hypothesised that $\Phi(T) = min[4f(T), 1]$ such that $\Phi = 0.5$ at -10 °C.

Despite these theoretical hypotheses and approximations, implementation of 'Mode 2' in a parcel model with bin microphysics 95 gave adequate agreement with Ice Cloud Experiment–Tropical (ICE–T) observational data (Lawson et al., 2015).

2 Experimental Setup

A schematic of the experimental setup is shown in Fig. 1. The setup was purpose–built to study the impact of a supercooled water drop on an ice particle. For this study, we used two configurations of the experimental setup. The first configuration was used to study the drop impact with a high–speed camera (Chronos 1.4, Kron Technologies Inc.) equipped with a microscopic lens (Kron Technologies Inc.) and a 0.5× barlow lens (Kron Technologies Inc.) in a side–on view. The second configuration was used to quantify the fraction of secondary drops that froze after impact with the ice particle using two Raspberry Picameras (Raspberry PiCamera Module V2) referred to as RPicams, with a polarising filter (Standard 55 mm Circular Polariser) attached to one camera. At present, the two configurations are not compatible to work concurrently. Recordings using the high–speed camera were recorded at 1069 frames per second (fps) and recordings using the RPicameras-RPicams were recorded at 24

105 frames per secondfps.

The experimental setup is operated in a cold room which can achieve a base temperature as low as -50 °C and provided the means of achieving a supercooled environment. The experimental setup was housed in a Bosch strut/Perspex panel frame to prevent the accidental introduction of frost particles during the experiments. A glass slide was supported on 3D printed plastic stilts approximately 10 cm in height which had a fan attached to dissipate the heat emitted from the polarised light source

- 110 (LCD monitor). The temperature of the glass slide was monitored using a K-type thermocouple attached to the glass slide with aluminium tape. The relative humidity was not measured , but will be below ice saturation, and possibly very small ice fragments were not observed due to sublimation preventing growth to visible sizes. The ice particles were prefabricated by freezing ultrapure water drops (Endotoxin–Free UP H2O, Merck) of approximately 6 mm in diameter on a glass slide coated in a water repellent (Rain–X) using a Peltier cooling system. The typical freezing shape of the ice particle is shown in Fig.
- 115 1. A pipette was modified to allow an ultrapure water drop (Endotoxin–Free UP H2O, Merck) at room temperature with a diameter of approximately 5 mm to be placed on the pipette using a disposable needle (22 gauge, sterile) and syringe. The modified pipette was held in a 3D printed tipper mechanism parallel to the glass slide, and the water drop was allowed to reach thermal equilibrium with the cold room for 90 s. The supercooled drop was released from the modified pipette perpendicular to the glass slide and was controlled by an Arduino and servo motor. The tipper was attached to an x-y translator (modified 3D)
- 120 printer) to allow multiple drop impact experiments using the same glass slide.

As the drop height and initial supercooled water drop diameter before impact (*D*) were kept constant at 1.36 m and 5 mm $_{\overline{7}}$ respectively, the normal impact velocity ($\forall_0 V_0$) for all experiments was 5.2 m s⁻¹. The terminal velocity of a 5 mm diameter drop is approximately 9 m s⁻¹ (Gunn and Kinzer, 1949). Initially, the impact velocity may seem unrealistic. However, the ice particle in these experiments was held stationary on a glass slide, but in the atmosphere the ice particle would also be falling.

- 125 The terminal velocity will depend on the ice particle shape, but for aggregates of similar size it is typically around 1 m s⁻¹ Locatelli and Hobbs (1974). The differential velocity between the supercooled water drop and the ice particle ice particle will be less than 9 m s⁻¹ dependent on the nature of the ice particle. While such large droplets are rare in the atmosphere the purpose here is to demonstrate that the process is a potential secondary ice mechanism. The supercooled water drop and the ice particle/glass slide were in thermal equilibrium for all experiments.
- 130 The temperature range investigated was between -4 °C to -12 °C. As the temperature of water decreases, the surface tension (σ) and viscosity $(\lambda \mu)$ of water increases (Hrubý et al., 2014; Dehaoui et al., 2015). In fluid dynamics, the Weber number, We = $\rho DV_0^2/\sigma$, and Reynolds number, Re = $\rho DV_0/\mu$, are used to relate inertial forces to the inertial forces of the fluid to its interfacial and viscous forces , respectively. respectively. In this case, the fluid is the supercooled water drop. The inertial forces are is from the initial impact velocity of the supercooled water drop, and the interfacial (surface tension) and viscous forces are
- 135 properties of the supercooled water drop. Taking into account the temperature dependent values of surface tension and viscosity of the supercooled water between -4 °C \leq T \leq to -12 °C, the We and Re number ranges obtained were 1747 \leq We \leq 1772 and 8781 \leq Re \leq 12240, respectively.

We conducted 32 experiments using the RPicams configuration during quantification of the freezing fraction of secondary drops and the data is given in Table A1.

140 3 Results

From our high-speed and RPicams recordings we present a schematic diagram of the formation of secondary drops from a supercooled water drop impact on an ice particle on a glass slide in Fig. 2. The We and Re numbers used were sufficiently large, i.e. We >> 2.5 and Re >> 25, such that inertia dominated the spreading of the thin film (Roisman, 2009). Surface tension and viscosity forces were therefore considered negligible during the spreading phase of the drop (Roisman, 2009), as was the wettability of the surface (Antonini et al., 2012). Figure 2(a) depicts the filament-like structures which were ejected during the spreading phase of the drop impact. We were unable to track the positions of these secondary drops or quantify them with our current high-speed camera or RPicams configurations. As the kinetic energy is transferred from that of a vertical to horizontal motion at impact, the water drop spread out radially, and instabilities at the rim were also observed. Figure 2(b) depicts the retraction of the drop, which caused the instabilities to 'pinch off' or rupture, followed by partial rebound. On

150 superhydrophobic surfaces, rupturing of the instabilities has been attributed to surface tension (Zhang et al., 2020). Our glass slide, coated in a water-repellent, is probably superhydrophobic, and surface tension is likely the cause of the rupture of the rim instabilities.



Figure 1. Schematic diagram of the experimental setup. Components labeled (i) were used in the high–speed configuration and (ii) were used in the RPicams configuration. The setup was operated in a cold room to achieve a supercooled environment.



Figure 2. A schematic diagram of a supercooled water drop impact on an ice particle on a glass slide and subsequent secondary drop formation during (a) the spreading phase and (b) the retraction phase.

3.1 Drop impact: high-speed recordings

We performed control experiments at room temperature (23 °C) and several supercooled temperatures using the high–speed camera configuration to characterise the water drop (diameter of 5 mm) impact of impacting the glass slide. Figure 3 shows the

(a) Temperature: 23	°C			
t = 0 ms (I)	t = 17 ms (S)	t = 33 ms (S)	t = 50 ms (S)	t = 67 ms (S)
4				
t = 83 ms (R)	t = 100 ms (R)	t = 117 ms (R)	t = 133 ms (R + B)	t = 150 ms (R)
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t = 167 ms (R)	t = 183 ms (R + B)	t = 200 ms (R)	t = 217 ms (R)	t = 233 ms (R)
t = 250 ms (R)	t = 267 ms (R)	t = 283 ms (PR)	t = 300 ms (B + PR)	t = 317 ms (PR)
	The second se			
(b) Temperature: -5	°C			
(b) Temperature: -5 t = 0 ms (I + E)	°C t = 17 ms (S + E)	t = 33 ms (S + E)	t = 50 ms (S + E)	t = 67 ms (S + E)
(b) Temperature: -5 t = 0 ms (I + E)	°C t = 17 ms (S + E)	t = 33 ms (S + E)	t = 50 ms (S + E)	t = 67 ms (S + E)
(b) Temperature: -5 t = 0 ms (I + E) t = 83 ms (R)	°C t = 17 ms (S + E) t = 100 ms (R)	t = 33 ms (S + E)	t = 50 ms (S + E) t = 133 ms (R)	t = 67 ms (S + E)
(b) Temperature: -5 t = 0 ms (I + E) t = 83 ms (R)	C t = 17 ms (S + E) t = 100 ms (R)	t = 33 ms (S + E) t = 117 ms (R)	t = 50 ms (S + E) t = 133 ms (R)	t = 67 ms (S + E)
(b) Temperature: -5 t = 0 ms (1 + E) t = 83 ms (R) t = 83 ms (R) t = 167 ms (R)	°C t = 17 ms (S + E) t = 100 ms (R) t = 183 ms (R)	t = 33 ms (S + E) t = 117 ms (R) t = 200 ms (R)	t = 50 ms (S + E) t = 133 ms (R) t = 217 ms (R)	t = 67 ms (S + E) t = 150 ms (R) t = 233 ms (R)
(b) Temperature: -5 t = 0 ms (1 + E) t = 83 ms (R) t = 167 ms (R)	°C t = 17 ms (S + E) t = 100 ms (R) t = 183 ms (R)	t = 33 ms (S + E) t = 117 ms (R) t = 200 ms (R)	t = 50 ms (S + E) t = 133 ms (R) t = 217 ms (R)	t = 67 ms (S + E) t = 150 ms (R) t = 233 ms (R)
(b) Temperature: -5 t = 0 ms (I + E) t = 83 ms (R) t = 167 ms (R) t = 250 ms (R)	<pre>c t = 17 ms (S + E) t = 100 ms (R) t = 183 ms (R) t = 267 ms (R) </pre>	t = 33 ms (S + E) t = 117 ms (R) t = 200 ms (R) t = 283 ms (PR)	t = 50 ms (S + E) t = 133 ms (R) t = 217 ms (R) t = 300 ms (PR)	t = 67 ms (S + E) t = 150 ms (R) t = 233 ms (R) t = 317 ms (PR)

Figure 3. Frames from the high–speed camera configuration of a water drop impact on a glass slide when both water drop and glass slide are at (a) room temperature (23 °C) and (b) -5 °C. The impact phase (I), spreading phase (S), secondary drop formation/ejection during the spreading phase (E), retraction phase (R), secondary drop formation due to receding break-up (B) and partial rebound (PR) of the water drop are indicated in the frames. Arrows indicate secondary drop formation during the retraction phase of the water drop.

t = 0 ms (I + E)	t = 17 ms (S + E)	t = 33 ms (S + E)	t = 50 ms (S + E)	t = 67 ms (S + E)
t = 83 ms (R)	t = 100 ms (R)	t = 117 ms (R)	t = 133 ms (R)	t = 150 ms (R + B)
	President of	Philip Del	a think you	
t = 167 ms (R + B)	t = 183 ms (R)	t = 200 ms (R)	t = 217 ms (R)	t = 233 ms (R)
t = 167 ms (R + B)	t = 183 ms (R)	t = 200 ms (R)	t = 217 ms (R)	t = 233 ms (R)
t = 167 ms (R + B) ↓ ↓ t = 250 ms (PR)	t = 183 ms (R) t = 267 ms (PR)	t = 200 ms (R)	t = 217 ms (R)	t = 233 ms (R)

Figure 4. Frames from the high–speed camera setup configuration of a supercooled water drop impact on an ice particle when both drop and ice particle are at -5 °C. The impact phase (I), spreading phase (S), secondary drop formation/ejection during the spreading phase (E), retraction phase (R), secondary drop formation due to receding break-up (B) and partial rebound (PR) of the water drop are indicated in the frames. Arrows indicate secondary drop formation during the retraction phase of the supercooled water drop.

frames from a high–speed recording of (a) a water drop impact on the glass slide at room temperature and (b) a supercooled water drop impact at -5 $^{\circ}$ C.

On impact with the glass slide, the water drop deformed and spread radially outwards as a thin film bordered by a thicker rim. Instabilities at the rim were observed for both the room temperature drop and the supercooled drop at -5 °C. The supercooled

- 160 drop shown in Fig. 3(b) ejected straight filament–like structures at an angle to the glass surface close to the impact and these filament–like structures disintegrated into secondary drops. This was in contrast to the impact of the water drop at room temperature drop-where no ejection of filament–like structures was observed. During the retraction phase, some of the rim instabilities pinched off from the thin film in the experiments with the water drop at room temperature forming secondary drops, in a process called 'receding break–up'. In contrast, no receding break–up was observed for the supercooled drop.
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Figure 4 shows the frames of a supercooled water drop impacting the side of an ice particle at -5 °C. Similar to the supercooled water drop on a glass slide, filament–like structures, which dissipated into secondary drops, formed at or close to impact with the glass slide/ice particle. Unlike the impact of a supercooled water drop on a bare glass slide, secondary drops formed via receding break–up. These secondary drops were observed around the parts of the rim of the thin film which contacted the ice particle.



Figure 5. Selected frames from the impact of a supercooled water drop on an ice particle at -4°C using the RPicams configuration. Frames (a)–(c) before, at and \sim 10 s after impact using the camera with no polarising filter. Red arrows in (c) indicate the number of secondary drops formed. Frame (d) shows the difference between (a) and (c). Frames (e)–(g) before, at and \sim 10 s after impact using the camera with a polarising filter. The white arrow in (h) indicates the frozen secondary drop. Frame (h) shows the difference between (e) and (g).

170 3.2 Determining the freezing fraction of the secondary drops: RPicams

We performed supercooled water drop impacts on ice particles over a temperature range from between -4 °C \leq T \leq to -12 °C. To unambiguously identify if a secondary drop had frozen, we used a polarising filter with a polarised light source, exploiting the birefringent properties of ice. Figure 5 shows selected frames of a supercooled water drop impact at -4 °C using the RPicams configuration. The top row of Fig. 5 shows frames from the camera with no polarising filter (a) before, (b) at and (c) ~10 s

- 175 after impact. The number of secondary drops observed are indicated by red arrows in Fig. 5 (c). The difference between the Fig. 5 (a) and (c) is presented in Fig. 5 (d) clearly indicating the secondary drops formed. The bottom row shows frames from the camera with a polarising filter (e) before, (f) at and (g) ~ 10 s after impact. The frozen secondary drop is indicated by a white arrow in Fig. 5 (g). The difference between the Fig. 5 (e) and (g) is presented in Fig. 5 (h) clearly indicating the frozen secondary drop formed.
- For this particular experiment, five secondary drops formed, of which one froze, giving a freezing fraction, $\Phi = 0.2$. During these experiments, two types of supercooled water impacts occurred: direct impact on the ice particle and partial impact on the ice particle. These different impacts arose due to practical difficulties with consistently impacting the ice particle with supercooled water drop due to changes in viscosity of water at different temperatures. For the experiment shown in Fig. 5, the impact was a side impact towards the top left of the ice particle as indicated in Fig. 5 (b). The RPicams configuration only
- 185 observed the larger > 0.1 mm diameter drops formed during retraction of the thin film. The smaller secondary drops (< 0.1 mm diameter) observed at impact from the high–speed configuration were not observed using this configuration as the minimum drop diameter the RPicams could detect was 0.1 mm.</p>



Figure 6. (a) The average freezing fraction of the secondary drops($\overline{\Phi}$) (left y-axis, black triangles) and (b) the average number of secondary drops(right y-axis, \overline{N}_s (blue circles) as a function of temperature. Average data included both direct and partial collisions. The error bars represent the standard error in the temperature intervals which are listed in Table A2 & A3.

Figure 6(a) shows the average freezing fraction of secondary drops formed when a supercooled water drop with a diameter of 5 mm collided with an ice particle, Φ, plotted on the left y-axis as a function of temperature. The raw data can be found
in Table A1 and the averaged data of the freezing fraction of secondary drops in Table A2. The average number of secondary drops, N_s, liquid or solid, is plotted on the right y-axis of shown in Fig. 6(b) as a function of temperature, which reached a maximum at approximately -7.5 °C. The averaged data of the number of secondary drops can be found in Table A3.

4 Discussion

From our high-speed and RPicams recordings we present a schematic diagram of the formation of secondary drops from a 195 supercooled water drop impact on an ice particle in Fig. 2. The We and Re numbers used were sufficiently large, i.e. We » 2.5 and Re » 25, such that inertia dominated the spreading of the thin film (Roisman, 2009). Surface tension and viscosity forces were considered negligible during the spreading phase of the drop (Roisman, 2009), as was the wettability of the surface (Antonini et al., 2012). Figure 2(a) depicts the filament–like structures which were ejected during the spreading phase of the drop impact. These filament–like ejections were classified as forming from both prompt– (Pan et al., 2010) and corona–type

- 200 splashes (Ashida et al., 2020), which may be due to the loose definition for prompt and corona splash mechanisms within the literature (Josserand and Thoroddsen, 2016). We were unable to track the positions of these secondary drops or quantify them with our current high-speed camera or RPicams configurations. As the kinetic energy is transferred from that of a vertical to horizontal motion at impact, the water drop spread out radially, and instabilities at the rim were also observed. Figure 2(b) depicts the retraction of the drop, which caused the instabilities to 'pinch off' or rupture, followed by partial rebound. On
- 205 superhydrophobic surfaces, rupturing of the instabilities has been attributed to surface tension (Zhang et al., 2020). Our glass slide, coated in a water-repellent, is probably superhydrophobic, and surface tension is likely the cause of the rupture of the rim instabilities.

As the As the ice particles were placed on a flat glass slide, during impact, the supercooled water drop spread across the ice particle and on to the glass slide where the larger > 0.1 mm diameter sized secondary drops formed. We acknowledge that the glass slide presents an artificially flat surface compared to atmospheric conditions. However, a study by Schremb et al. (2018) showed that, on an elevated ice surface, the thin film of a supercooled water drop with a diameter of ~4 mm and similar We and Re numbers at -14 °C was ejected and subsequently ruptured, forming secondary drops. While quantification was not the focus of their study, it was observed that the rim of the supercooled water drop was largely frozen, but and only some of the secondary drops were observed as ice. The size of the secondary drops formed in the study by Schremb et al. (2018) is

- 215 comparable to our secondary drops despite the different generation mechanism, and the supplementary videos indicate that 10s of secondary drops were formed. Furthermore, water drops with diameters between ~3–4 mm colliding with a steel disk of ~4 mm in diameter Rozhkov et al. (2002) (Rozhkov et al., 2002) and water drops with diameters of 6 mm colliding with an iron cylinder of the same diameter Villermaux and Bossa (2011) produced numerous secondary drops, although again no quantification was made(Villermaux and Bossa, 2011) produced 100s of secondary drops. Clearly, when a flat surface like the
- 220 glass slide used in this study is not present, secondary drops are still formed, emitted from the rim of the thin film during impact. Whether this impacts on However, there is much uncertainty about the number of secondary drops formed is

In addition, the ice particle in our experiments is in a fixed position on the glass slide, whereas, in the atmosphere, the ice particle is in free fall. When the faster-moving supercooled water drop collides with the ice particle, the ice particle will move in response to the collision, likely affecting the formation of the secondary drops and their subsequent freezing. However, currently, it is difficult to ascertain without further studies.

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A schematic diagram of a supercooled water drop impact on an ice particle and subsequent secondary drop formation during (a) the spreading phase and (b) the retraction phase.

We observed a decrease in how this will influence secondary drop formation and freezing without further investigations into the mechanisms of secondary drop formation at T < -8 °C. Quantification of our secondary drops was made during the retraction phase of the supercooled water drop impact, and on an elevated ice particle.

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Another factor that will influence the generation of secondary drops is the ice particle shape. Our ice particles have a pointed tip, as shown in Fig. 2, which is a typical shape formed when a liquid water drop is frozen on a cold substrate (Snoeijer and Brunet, 2012), but not representative of atmospheric ice particles. According to Phillips et al. (2018), who refer to this SIP mechanism as 'Mode 2', for it to occur, the supercooled water drops spent a prolonged time in contact with must

- 235 have a diameter larger than 150 µm and the ice particle . For example, the first drops formed during retraction were observed at t ~133 ms in Fig. 4 at -5 °C, whereas, the first secondary drops were observed at t ~5 ms for the supercooled more massive still. In the atmosphere, ice particles which are larger than 150 µm are typically irregular in shape (Korolev and Sussman, 2000) . A study by Zhang et al. (2020) shows that at room temperature, water drop impact on an elevated ice target studied by Schremb et al. (2018) at -14curved surfaces induce additional fragmentation mechanisms compared to flat surfaces. Therefore,
- 240 we expect the irregular shape of an ice particle to affect the fragmentation mechanisms of the supercooled water drop and thus secondary drop formation.

We observed a decrease in the number of secondary drops formed during receding break-up as temperature decreased below -8 °C. If we consider our results at T \leq 11 °C, the average number of secondary drops formed was 1.6 per collision, of which an average number of 0.4 secondary drops froze per collision. Moreover, Fig. Figure 7 shows the frames after a supercooled water

- drop impact with an ice particle for the experiments at T < between $-11^{\circ}C$. Here, it is clear to see how the lower temperature impacted on the retraction phase of °C to $-12^{\circ}C$ which was the range where the smallest number of secondary drops formed. At these temperatures, the supercooled water drop , and hence, secondary drop formation. The froze either during the spreading phase or in the early stages of the retraction phase. As the growth velocity of ice in supercooled water increases with decreasing temperature. At T = , e.g. at -2 °C it is around 0.2 cm s⁻¹, whereas at T=-10 °C it is around 5 cm s⁻¹ (see Pruppacher and
- 250 Klett, 1997, chapter 16). This, along with the transfer of heat away from the freezing drop, results in a finite amount of time that a given size of liquid water can remain in contact with ice without completely freezing onto the ice, which may explain why a decrease in secondary drops was observed. We believe the decrease in secondary drop formation at T < temperatures below -8 °C may be due to the artificially flat geometry presented by the glass slide and to the large size of the incident drop, both factors which prolonged the interaction time between the supercooled water drop and ice. For example, the supplementary</p>
- 255 videos from Schremb et al. (2018) showed several-10s of secondary drops forming even-at -14 °C after impact on an elevated ice target, more than we observed at our lowest temperatures temperature of -12 °C.

The averaged freezing fraction of secondary drops $\overline{\Phi}$ in Fig. 6 shows that the freezing fraction at T>-12°Cis fairly consistent with $\overline{\Phi} \sim 0.3$. We believe that the freezing fraction of the secondary drops is independent of the number of drops formed. For example, at T \leq -11 °C the number of secondary drops formed reduces significantly, but the averaged freezing fraction of

- 260 Whilst the freezing mechanism of the secondary drops was not specifically studied in this work, we consider the following mechanisms. The freezing of supercooled water drops occurs in two stages. The first stage is characterised by the formation of ice dendrites throughout the supercooled water drop. The latent heat from the formation of the ice dendrites is released during this stage, warming the temperature of the supercooled water drop to ~ 0 °C. The second stage is characterised by the freezing of the secondary drops remains consistent with warmer temperature collisions. Also, Schremb et al. (2018) suggested
- 265 that seeding ice crystals transported during the initial spreading phase when ice shear rates were highest could explain why



Figure 7. Frames from the RPicams configuration approximately 10 s after a supercooled water drop impact for experiments at T \leq -11 °C. The top panel shows frames from the RPicam with no polarising filter and the bottom panel shows frames from the RPicam with a polarising filter.

freezing occurred predominately around the remaining supercooled water drop and is controlled by the loss of latent heat due to the supercooled water drop surroundings. Stage 1 of freezing is fast and the time taken for this stage to complete (t_i) can be estimated from the following equation (Macklin and Payne, 1967):

$$t_i \approx \frac{\delta_R}{G} \tag{1}$$

270 where δ_R is the thickness of the layer of supercooled water on the ice particle and *G* is the growth velocity of ice which is temperature dependent.

From Fig. 4, we can estimate that the rim of the supercooled water drop. For our secondary drops formed at T >, which is also the thickest part of the supercooled water drop, is approximately 0.78 mm. Taking this value for δ_B and given that the growth velocity of ice at -5 °C is approximately 1 cm s⁻¹ (Pruppacher and Klett, 1997, chapter 16), then $t_i \approx -12 \text{ °C}$, we

- 275 believe a similar mechanism occurred and that the majority of the frozen secondary drops nucleated from seeding ice crystals rather than dendritic growth from the ice particle. This may also translate to the smaller secondary drops observed with the high-speed camera configuration as 0.078 s. Figure 4 shows the time-scale for the retraction phase is of the order 0.1 s. It is plausible that the initial ice dendrites can propagate through the supercooled water drop and that water containing these ice dendrites may then break off during the retraction phase and initiate freezing. The second phase of freezing will take longer, but
- 280 as long as the drop contains ice dendrites it will eventually freeze. This explanation is also proposed by Schremb et al. (2018) and Phillips et al. (2018) who suggest that seeding ice crystals are transported during the initial spreading phase. Alternative freezing mechanisms include the formation of a thin, unobserved film of liquid water present on the glass slide after the retraction phase. The contact between the thin film of water and the ice particle could induce freezing in the drops formed during the very early stages of spreading . thin film, which could then trigger freezing in the seemingly detached secondary drop.
- 285 Mechanical agitation or shock may also play a role in the freezing of the secondary drops (Alkezweeny, 1969; Czys, 1989).

Regardless of the freezing mechanism, the glass slide will likely have some influence, and it will be pertinent to remove this in future investigations.

4.1 Atmospheric Implications

Observations within mixed phase clouds often show ice crystal number concentrations higher than the numbers of ice nucleating

- 290 particles present in the atmosphere. For instance, ice particle number concentrations exceeding $100 L^{-1}$, in shallow convection with cloud-top temperature no lower than -12As a proof-of-concept investigation, we studied supercooled water drops with diameters of 5 mm and ice particles with diameters of 6 mm as larger sizes of supercooled water drops were easier to work with experimentally. While these sizes are not necessarily representative of cloud conditions, theoretically, this new SIP mechanism should occur where supercooled water drop diameters are > 150 °C, have been observed over the UK (Crawford et al., 2012).
- 295 Furthermore, thin mixed-phase layer clouds have been observed to continually generate snow (Westbrook and Illingworth, 2013) . Conventional thinking would suggest that the ice in mixed-phase layer clouds should fall out, leaving the layer 'depleted' of INPs; however, the observations clearly show that ice continues to form in these clouds over time.

The rime-splintering SIP mechanism has been successful in predicting the glaciation of mixed-phase clouds in many cases (e.g. Harris-Hobbs and Cooper, 1987; Blyth and Latham, 1993, 1997; Phillips et al., 2001, 2005; Crosier et al., 2011; Huang et al., 2017)

- 300 . However, there are also numerous cases where significant concentrations of ice observed in clouds cannot be explained by the rime-splintering SIP mechanism. Hobbs and Rangno (1985) compiled tables of aircraft observations from a wide range of cloud environments where enhancements in ice particle concentrations over the background INP concentrations occurred. They found that ice enhancement was independent of the cloud-top temperature, but was strongly dependent on the broadness of the supercooled drop spectrum near the cloud-top. μm and the ice particles more massive still. Supercooled water drops
- and ice particles are present within a variety of different clouds. For example, Hobbs and Rangno (1990) presented aircraft observations in small polar-maritime cumuli that displayed ice enhancement with cloud bases too cold for large cloud-droplets (> 24 µm) between -3 and -8 °C as required for rime splintering. Their discussion highlighted that ice enhancement proceeded in two stages. The first stage consisted of the formation of frozen drops, < 400 µm diameter, and small graupel particles, < 1 mm diameter. The second stage was characterised by the appearance of high concentrations of vapour–grown ice crystals
 in the upper regions of the cloud. A key finding of this series of papers was that high concentrations of small ice particles
- appeared simultaneously with frozen drizzle drops. Furthermore, Rangno and Hobbs (2001) showed that large supercooled drops were often a requirement for ice enhancement in moderately cooled Arctic stratiform clouds, and ice enhancement was often coincident with observations of large supercooled raindrops.

Supercooled drizzle drops and raindrops are common in convective clouds (e.g. Crawford et al., 2012; Taylor et al., 2016), as are large ice particles. Hence, because there is a broad continuum of drizzle and raindrop sizes, where the larger drops freeze

as are large ice particles. Hence, because there is a broad continuum of drizzle and raindrop sizes, where the larger drops freeze first, followed by accretion of the smaller unfrozen drops that 'Mode 2', the collision_indicates that collisions of supercooled water drops with ice particles more massive , may be of importance in a wide range of clouds.

5 Conclusions and Future Work

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In this study, we confirmed that during collisions of supercooled water drops with ice particles, frozen secondary drops formed over the temperature range of between -4 $^{\circ}$ C to -12 $^{\circ}$ C. Our main findings are:

- 1. Approximately 5 to 10 secondary drops are formed during the receding break-up of the retraction phase of a supercooled water drop (D = 5 mm) after collision with an ice particle (D = 6 mm) placed on a glass slide.
 - 2. An average of 30% of these secondary drops formed froze over the temperature range of between -4 °C to -12 °C.
 - 3. Experiments with a high-speed camera highlighted that secondary drops formed as a jet of smaller droplets produced
 - separately from the receding break–up of the drop. No quantification of the freezing fraction of the secondary ice drops can currently be made.
- Further work is needed to investigate the secondary drop formation without the influence of the glass slide, and this-

One of the main experimental challenges of this work was dropping the supercooled water drop consistently onto the ice particle which limited the amount of experiments we could perform. As shown in Table A1, the majority of the successful impacts were classified as partial hits despite the intention for them to be direct hits. While partial hits are expected in clouds,

- as well as direct hits, we also conducted many experiments where the supercooled water drop missed the ice particle. One method of achieving better control of the supercooled water drop impact could be via growth and supercooling of a water drop at the end of a needle similar to the system shown in Schremb et al. (2018). Compared to our current mechanism, which involved tilting a pipette to allow the supercooled water drop to roll off, the supercooled water drop would remain fixed to a certain point before detaching under gravity, making it easier to drop consistently in the same position.
 - Another experimental challenge we would like to address is quantifying the secondary drops formed during the spreading phase of the supercooled water drop during impact. Thoroddsen et al. (2012) quantified secondary drops ejected with velocities of up to 100 m s⁻¹ using an ultra-high-speed camera capable of recording at 1000000 fps, and we could use a similar setup. We could then exploit the birefringent properties of ice to determine whether these ejected secondary drops froze.
- 340 The number of secondary drops per collision is sensitive to geometry and material of collision, even for drops of the same size. We quantify about 10 per collision, Schremb et al. (2018) observed 10s of collision for impacts on elevated ice surface. Rozhkov et al. (2002) observe 100s for drop impacts on steel disks at room temperature, as do Villermaux and Bossa (2011) for drop impacts on iron cylinders at room temperatures. Consequently, after addressing the above challenges and elevating the ice particle off the glass surface, which may be achieved simply by fixing the ice particle on a wire, further work is needed
- 345 to investigate, more systematically, this new SIP mechanism over a larger range of range of experimental parameters, not limited to: supercooled drop sizes, temperatures, and interaction parameters supercooled water drop-to-ice particle size ratios, ice particle shapes, temperatures, drop height (and hence impact velocity), airflow, relative humidity conditions and chemical compositions of the supercooled water drop.

Appendix A

	Temperature (°C)	Total number of secondary drops	Frozen secondary drops
Direct	-4.2	14	1
	-4.2	0	0
	-5.3	7	6
	-5.5	10	2
	-7.8	12	2
	-9.9	7	0
Partial	-3.8	16	5
	-4.0	5	1
	-4.0	8	5
	-4.3	9	6
	-5.6	5	0
	-5.6	5	1
	-5.8	9	5
	-6.0	4	1
	-6.0	8	2
	-6.1	2	1
	-6.1	12	3
	-7.7	17	7
	-8.0	5	0
	-8.0	11	7
	-8.1	8	1
	-8.5	16	0
	-9.4	0	0
	-9.4	21	6
	-9.8	11	4
	-10.0	2	2
	-10.1	10	6
	-11.3	0	0
	-11.5	4	1
	-11.8	4	1
	-11.9	0	0
	-11.9	0	0

Table A1. Total number of satellite secondary drops and the number of frozen satellite secondary drops for each experiment

Table A2. The mean $(\bar{\Phi})$ and standard deviation (σ) of the fraction of frozen secondary drops within a specified temperature interval (T interval) along with the number of experiments (n) within the T interval, the average degree of supercooling within the temperature interval (\bar{T}) and the error in the sample mean $(\sigma_{\bar{\Phi}})$.

T inter	val (°C)	\bar{T} (°C)	$\bar{\Phi}$	σ	n	$\sigma_{ar{\Phi}}$
-3.8	-4.3	-4.1	0.4 0.38	0.3 0.26	5	0.1- 0.12
-5.3	-5.8	-5.6	0.4-0.36	0.3 0.34	5	0.2- 0.15
-6.0	-6.1	-6.1	0.3-0.44	0.1-0.38	4	0.1- 0.19
-7.7	-7.8	-7.8	0.3 -0.29	0.2 <u>0.17</u>	2	0.1-0.12
-8.0	-8.5	-8.2	0.2 -0.19	0.3 0.30	4	0.2- 0.15
-9.4	-9.9	-9.7	0.2 0.22	0.2 0.19	3	0.1-0.11
-10.0	-10.1	-10.1	0.8 -0.80	0.3 0.28	2	0.2 -0.20
-11.3	-11.9	-11.7	0.3- 0.25	0 -0 <u>.00</u>	2	0 0.00€

Table A3. The mean (\bar{N}_s) and standard deviation (σ) of the number of secondary drops within a specified temperature interval (T interval) along with the number of experiments (n) within the T interval, the average degree of supercooling within the temperature interval (\bar{T}) and the error in the sample mean $(\sigma_{\bar{N}_s})$.

T inter	val (°C)	\bar{T} (°C)	\bar{N}_s	σ	n	$\sigma_{\bar{N_s}}$
-3.8	-4.3	-4.1	8.7	5.9	6	2.4
-5.3	-5.8	-5.6	7.2	2.3	5	1.0
-6.0	-6.1	-6.1	6.5	4.4	4	2.2
-7.7	-7.8	-7.8	14.5	3.5	2	2.5
-8.0	-8.5	-8.2	10.0	4.7	4	2.3
-9.4	-9.9	-9.6	9.8	8.8	4	4.4
-10.0	-10.1	-10.1	6.0	5.7	2	4.0
-11.3	-11.9	-11.7	1.6	2.2	5	1.0

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Data availability. All datasets are provided in Appendix A.

Video supplement. All video recordings from the high–speed configuration and RPicams configuration are deposited in Figshare, a FAIR-aligned data repository and can be accessed at https://doi.org/10.48420/c.5476557.

355 *Author contributions.* VTJP and PJC conceived the original study. RLJ and PJC designed the new experimental setup with advice from VTJP. RLJ and PJC performed the experiments. RLJ analysed the data and wrote the paper. VTJP and PJC provided comments on the paper.

Competing interests. The authors declare that they have no conflict of interest

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