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Regional heterogeneities in the emission of airborne primary sugar compounds and biogenic secondary organic aerosols in the East Asian outflow: Evidence for coal combustion as a source of levoglucosan M. Mozammel Haque^{1,2,3}, Yanlin Zhang^{1,2*} Srinivas Bikkina⁴, Meehye Lee⁵, and Kimitaka Kawamura^{3,4} ¹Yale-NUIST Center on Atmospheric Environment, International Joint Laboratory on Climate and Environment Change (ILCEC), Nanjing University of Information Science & Technology, Nanjing, 210044, China ²School of Applied Meteorology, Nanjing University of Information Science & Technology, Nanjing 210044, China ³Institute of Low Temperature Science, Hokkaido University, Sapporo 060-0819, Japan ⁴Chubu Institute for Advanced Studies, Chubu University, Kasugai 487-8501, Japan ⁵Department of Earth and Environmental Sciences, Korea University, Anam-dong, Sungbuk-gu, Seoul 136-701, South Korea

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29 ABSTRACT

30 Biomass burning (BB) significantly influences the chemical composition of organic aerosols 31 (OA) in the East Asian outflow. Source apportionment of BB-derived OA is an influential 32 factor for understanding their regional emissions, which is crucial for reducing uncertainties in their projected climate and health-effects. We analyzed here three different classes of 33 34 atmospheric sugar compounds (anhydrosugars, primary sugars, and sugar alcohols) and two 35 types of biogenic secondary organic aerosol (BSOA) tracers (isoprene- and monoterpene 36 derived SOA products) in a year-long collected total suspended particulate matter (TSP) from 37 an island-based receptor site in South Korea, the Gosan. We investigate seasonal variations in 38 the source-emissions of BB-derived OA using mass concentrations of anhydrosugars and 39 radiocarbon (¹⁴C-) isotopic composition of organic carbon (OC) and elemental carbon (EC) in 40 ambient aerosols. Levoglucosan (Lev) is the most abundant anhydrosugar, followed by 41 galactosan (Gal) and/or mannosan (Man). Strong correlations of Lev with Gal and Man as 42 well as their high mass ratios such as Lev/Gal (6.7±2.2) and Lev/Man (15.0±6.7), indicate the 43 contribution from hardwood burning emissions. The seasonal trends revealed that the 44 biomass-burning impact is more pronounced in winter and fall, as evidenced from the high 45 concentrations of anhydrosugars. Likewise, significant correlations were observed among 46 three primary sugars (i.e., glucose, fructose, and sucrose), emphasizing the contribution from 47 airborne pollen. The primary sugars showed higher concentrations in spring/summer than 48 winter/fall. The fungal spore tracer compounds (i.e., arabitol, mannitol, and erythritol) 49 correlated well with trehalose (i.e., a proxy for soil organic carbon), suggesting the origin 50 from airborne fungal spores and soil microbes in the East Asian outflow. These sugar 51 alcohols peaked in summer, followed by spring/fall and winter. Monoterpene-derived SOA 52 tracers were most abundant compared to isoprene-SOA tracers. Both BSOA tracers were 53 dominant in summer, followed by fall, spring, and winter. The source apportionment based 54 on multiple linear regressions, diagnostic mass ratios, and positive matrix factorization 55 analysis altogether revealed that biomass burning (41.9%) and biogenic SOA (21.1%) mostly dictates the OA loading in the ambient aerosols from East Asian outflow. We also found 56 significant positive linear relationships of ¹⁴C-based nonfossil- and fossil-derived organic 57 carbon fractions with Lev-C along with the comparable regression slopes, suggesting the 58 59 importance of BB and coal combustion sources in the East Asian outflow. 60

Keywords: Biomass burning tracers, primary biological aerosol particles, biogenic SOA
 tracers, radiocarbon-based source apportionment, organic aerosols, East Asian outflow





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63 **1. Introduction**

64 Organic aerosols (OA), which account for a major fraction up to 50% of airborne total 65 suspended particulate matter, have considerable effects on regional and global climate by 66 absorbing or scattering sunlight (Kanakidou et al., 2005). However, the climate effects of OA 67 are involved with large uncertainties due to our limited understanding of the contributing 68 sources. OA can be derived from both primary emissions and secondarily formed species. 69 Sugars are an important group of water-soluble, primarily organic compounds whose 70 concentrations are significant in atmospheric aerosols over the continent (Jia and Fraser, 71 2011; Fu et al., 2008; Yttri et al., 2007; Graham et al., 2003). Anhydrosugars such as 72 levoglucosan, galactosan, and mannosan are the key tracers of BB emissions (Simoneit, 73 2002). Sugar alcohols, along with glucose, trehalose and sucrose are mostly originated from 74 primary biological particles such as fungal spores, pollen, bacteria, and viruses, and 75 vegetative debris (Graham et al., 2003; Simoneit et al., 2004a; Bauer et al., 2008; 76 Deguillaume et al., 2008). Primary sugars and sugar alcohols are predominantly present in the 77 coarse mode aerosols, accounting for 0.5-10% of atmospheric aerosol carbon matter (Yttri et 78 al., 2007; Pio et al., 2008).

79 Secondary organic aerosol (SOA) is a large fraction of OA, while the key factors 80 controlling SOA formation are still poorly understood. The SOA formation significantly 81 increased with the enhancement of the ambient aerosol mass (Liu et al., 2018). SOA is 82 formed by both homogenous and heterogeneous reactions of volatile organic compounds 83 (VOCs) in the atmosphere (Surratt et al., 2010; Robinson et al., 2007; Claeys et al., 2004). On 84 a global estimation, biogenic VOCs (BVOCs) such as isoprene, monoterpenes (e.g., α/β -85 pinene), and sesquiterpenes (e.g., β -caryophyllene) are one order of magnitude higher than those of anthropogenic VOCs (e.g., toluene) (Guenther et al., 2006). The global emissions of 86 87 annual BVOCs were estimated to be 1150 TgC/yr, accounting for 44% isoprene and 11% 88 monoterpenes (Guenther et al., 1995). Isoprene is highly reactive and promptly reacts with 89 oxidants such as O₃, OH, NO_x in the atmosphere to form SOA (Kroll et al., 2005, 2006; Ng et 90 al., 2008; Surratt et al., 2010; Bikkina et al., 2021), estimated to be 19.2 TgC yr⁻¹, consisting 91 of ~70% of the total SOA budget (Heald et al., 2008). Monoterpenes are also considered to be 92 important sources of biogenic secondary organic aerosol (BSOA), accounting for ~35% of 93 the global BVOCs emissions (Griffin et al., 1999).

Atmospheric aerosols in East Asia, particularly in China, are characterized by coal and biofuel combustion as a major energy source, causing vast-emission of OA into the atmosphere (Huebert et al., 2003; Zhang et al., 2016). Understanding the ambient levels OA





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97 in the East Asian outflow is crucial for assessing their regional climatic effects. As part of 98 this effort, Korean Climate Observatory at Gosan (KCOG), a super site located in South 99 Korea has been chosen for investigating the atmospheric outflow characteristics from East 100 Asia (Fu et al., 2010a; Kundu et al., 2010; Ramanathan et al., 2007; Kawamura et al., 2004; 101 Arimoto et al., 1996). For instance, primary OA associated with soil/desert dust in East Asia 102 along with forest fires in Siberia/northeastern China are transported over Gosan in the spring 103 season (Wang et al., 2009a). BSOA during long-range transport from the continent and open 104 ocean, as well as local vegetation, can significantly contribute to Gosan aerosols. Although 105 these investigations were carried out almost a decade ago, no such observations are available 106 in contemporary times from Gosan. Here, we attempt to understand the current states of East 107 Asian OA using both molecular marker approach and radiocarbon data of carbonaceous 108 components.

109 The site of KCOG, located on the western side of Jeju Island adjacent to the Yellow 110 Sea and the East China Sea, is facing the Asian continent but is isolated from public areas of 111 the island (Kawamura et al., 2004). Simoneit et al., (2004b) have documented during the 112 ACE-Asia campaign that OA from the BB, and fossil fuel combustion sources are transported 113 along with desert dust to KCOG during continental outflow. An intensive campaign was organized at the KCOG during spring 2005 to observe the physical properties of East Asian 114 115 aerosols while two dust events were detected (Nakajima et al., 2007). Here, we focus on the 116 characterization of airborne anhydrosugars, primary sugars, sugar alcohols, and BSOA 117 tracers from the KCOG. Gosan is influenced by the continental outflow from East Asia 118 during winter, spring and fall, whereas the site is influenced by the maritime air masses from 119 the Pacific Ocean and other marginal seas. This makes KCOG ideal for characterizing the 120 regional heterogeneities in the emissions of organic compounds in the East Asian outflow 121 based on the TSP samples collected during April 2013-April 2014.

122 **2. Methods**

123 2.1. Aerosol Sampling and Prevailing Meteorology

Total suspended particles (TSP) were collected on pre-combusted (450°C for 6 h) quartz fiber filters (20 cm × 25 cm, Pallflex) at the KCOG (33.17 °N, 126.10 °E, see Figure 1), South Korea. To get enough signal for the radiocarbon measurements, each TSP sample was collected for 10–14 days from April 2013 to April 2014. A total of twenty-one samples were collected using a high-volume air sampler (Kimoto AS-810, ~65 m³ h⁻¹) installed on the rooftop of a trailer house (~3 m above the ground). After the collection, aerosol filters were transferred to a pre-combusted (450°C for 6 h) glass jar (150 mL) equipped with a Teflon-





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lined screw cap and transported to the laboratory in Sapporo. These TSP samples were stored
in a dark freezer room at -20°C until the analysis. Three field blank filters were also collected
during the campaign.

134 The ambient temperatures at the Gosan site were on average 6.9°C in winter, 14.1°C 135 in spring, 27.0°C in summer, and 17.1°C in fall. Likewise, the average relative humidity was 136 found to be highest in summer (71.3%), followed by spring (64.9%), fall (63.5%), and winter 137 (54.7%). Gosan is influenced by the pollution sources in East Asia during winter as well as other transition periods (spring and fall) due to the prevailing westerlies. In contrast, winds in 138 139 summer blew mostly from the western North Pacific (WNP) by the easterly winds. The 140 spring season is, in particular, important for the transport of mineral dust mixed with polluted 141 OA to Gosan (Kundu et al., 2010).



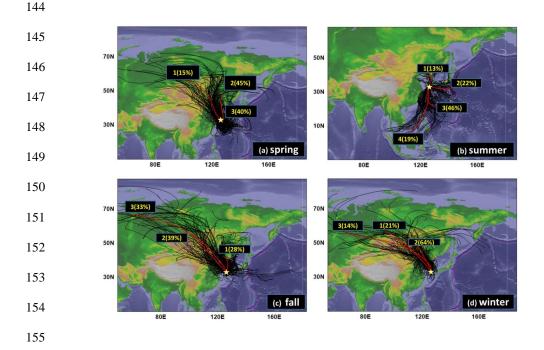


Figure 1. Cluster analysis of backward air mass trajectories over Gosan (indicated by a star symbol) for the TSP collected during (a) spring, (b) summer, (c) fall, and (d) winter seasons.

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160 2.2. Extraction and Analysis of Organic Compounds

Approximately 3.14 cm² filter cuts were extracted with dichloromethane/methanol (2:1; v/v). 161 162 The extracts were concentrated using a rotary evaporator under vacuum and then blown down to near dryness with pure nitrogen gas. The dried residues were subsequently reacted with N, 163 164 O-bis(trimethylsilyl)trifluoroacetamide containing 1% trimethylchlorosilane (BSTFA+1% TMCS, SUPELCO[©], Sigmaaldrich[®]) and pyridine at 70 °C for 3 h to derive OH and COOH 165 166 groups of polar organic compounds to trimethylsilyl ethers and esters, respectively. After the derivatization followed by addition of a known amount of internal standard solution 167 (Tridecane; 1.43 ng L⁻¹ in n-hexane), the derivatized extracts were injected on to a gas 168 169 chromatograph (Hewlett-Packard model 6890 GC) coupled to a mass spectrometer (Hewlett-170 Packard model 5973, MSD) (GC-MS). More details on the quantification of polar organic 171 compounds using GC-MS are described in Haque et al. (2019).

172 The target compounds (anhydrosugars, primary sugars, sugar alcohols, and BSOA 173 tracers) were separated on a DB-5MS fused silica capillary column (30 m x 0.25 mm i.d., 0.5 um film thickness) using helium as a carrier gas at a flow rate of 1.0 ml min⁻¹. The GC oven 174 temperature was programmed from 50°C for 2 min and then increased from 50 to 120°C at 175 30°C min⁻¹ and to 300°C at 6°C min⁻¹ with a final isotherm hold at 300°C for 16 min. The 176 177 sample was injected in a splitless mode with the injector temperature at 280°C. The MS was 178 operated at 70 eV and scanned from 50 to 650 Da on an electron impact (EI) mode. Mass 179 spectral data were acquired and processed using the Chemstation software. The organic 180 compounds were identified individually by comparison with retention times and mass spectra 181 of authentic standards and NIST library and literature data of mass fragmentation patterns 182 (Medeiros and Simoneit, 2007). For assessing the recoveries, ~100-200 ng of the standard 183 solution was spiked on the blank filter and analyzed as a real sample. Overall, the average 184 recoveries were found to be 80-104% for target compounds. The field and laboratory blank filters (n = 3) were also analyzed by the same procedures as a real sample. Target compounds 185 186 were not found in the field blanks. The analytical errors based on concentrations by replicate 187 sample analyses (n = 3) were less than 15%.

188 2.3. Radiocarbon isotopic composition of TC and EC

189 The concentrations of total carbon (TC) in TSP samples were analyzed using an elemental 190 analyzer. For the radiocarbon isotopic composition (Δ^{14} C), the aerosol filter punches were 191 exposed for ~12 h to HCl fumes in a vacuum desiccator. Subsequently, these filters were

analyzed for Δ^{14} C on a modified elemental analyzer coupled via a gas interface to Accelerator





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193 Mass Spectrometer Mini Carbon Dating System (MICADAS) at the University of Bern, 194 Switzerland (Salazar et al., 2015). The evolved CO₂ of TC from the elemental analyzer was 195 passed through a moisture trap (Sicapent, Merck) and isolated it from other residual gasses 196 using a temperature-controlled zeolite trap. The purified CO₂ was introduced through a gas interface system to MICADAS, where ${}^{14}C/{}^{12}C$ ratios are measured according to the analytical 197 procedures detailed in Zhang et al. (2016). Likewise, the evolved CO₂ of elemental carbon 198 from the Sunset Lab OC/EC analyzer using Swiss 4S protocol (Zhang et al., 2012), was 199 directed to the MICDAS and measured for the ¹⁴C/¹²C ratio relative to standard calibration 200 gas. These results were expressed as fractions of modern carbon (f_M) by normalizing with a 201 δ^{13} C value of the reference standard in the year 1950 (-25‰) according to Stuiver and Polach 202 203 (1997) for the fractionation effects. The $f_{\rm M}(\rm OC)$ can be estimated by using the $f_{\rm M}(\rm TC)$ and 204 $f_{\rm M}({\rm EC})$ in an isotope mass balance equation (Zhang et al., 2015). Additionally, we estimated 205 the relative contributions of OC and EC from the nonfossil and fossil sources ($f_{nonfossil}$ and 206 f_{fossil} , respectively) using the following equations. 207 $f_{\text{nonfossil-OC}} = f_{\text{M}}(\text{OC-sample})/f_{\text{M}}(\text{OC-ref}); f_{\text{M}}(\text{OC-ref}) \approx 1.07 \pm 0.04$ (1) $f_{\text{nonfossil-EC}} = f_{\text{M}}(\text{EC-sample})/f_{\text{M}}(\text{EC-ref}); f_{\text{M}}(\text{EC-ref}) \approx 1.10 \pm 0.05$ 208 (2)209 $f_{\text{fossil-OC}} = 1 - f_{\text{nonfossil-OC}}$ (3) 210 (4) $f_{\text{fossil-EC}} = 1 - f_{\text{nonfossil-EC}}$ 211 The reference values of OC and EC were obtained from Mohn et al. (2008). Using 212 the fractions of $f_{\text{fossil-OC}}$ and $f_{nonfossil-OC}$, we can, therefore, estimate the mass concentration of 213 ambient organic carbon (OC-ambient) from fossil and nonfossil sources (OC_{fossil} and 214 OC_{nonfossil}, respectively). 215 $OC_{nonfossil} = f_{nonfossil-OC} \times [OC]_{ambient}$ (5)

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$$OC_{fossil} = f_{fossil-OC} \times [OC]_{ambient}$$
(6)

217 More details on the radiocarbon isotopic composition data over Gosan were reported218 elsewhere (Zhang et al., 2016).

219 3. Results and discussion

220 **3.1. Trajectory/Cluster Analysis**

Backward air mass trajectories are useful for assessing the impact of local versus regional source-emissions over Gosan. Seven-day isentropic backward air mass trajectories were computed using the hybrid single-particle lagrangian integrated trajectory model (HYSPLIT, version 4: Stein et al., 2015) over KCOG for the sampling period using the meteorological





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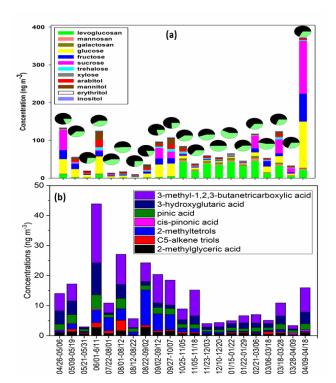
datasets of Global Data Assimilation System (GDAS) network. The trajectory endpoint files from the HYSPLIT model were used for the cluster analysis using the Trajstat package (Wang et al., 2009b) for all four seasons (Figure 1). Although cluster analysis revealed the predominance of continental transport in spring, fall, and winter seasons, the air masses over KCOG in summer mostly originated from the WNP. Since spring is a transition of winds switching from westerlies to easterlies, Gosan is likely influenced by the long-range transport of dust, pollution, and sea-salt aerosols.

232 The vertical mixing of pollutants in the boundary layer height also plays an 233 important role in controlling the strength of continental outflow along with regional 234 meteorology. For instance, the mixing height of air parcels computed by the HYSPLIT model is mostly confined within 1000 m in winter but somewhat increases towards spring and fall 235 236 seasons (Figure S1). This vertical enhancement in the boundary layer height facilitates the 237 transport of mineral dust particles from the arid and semiarid regions in East Asia along with 238 urban pollutants to Gosan in spring and fall compared to winter. However, the strength of 239 continental outflow is somewhat dependent on several factors, including source-emissions, 240 meteorology, and mixing height of air parcels.

241 Gosan is influenced by three types of air masses in spring (Figure 1a), from 242 Mongolian Desert (cluster 1: 15%), North China (cluster 2: 45%), and the Yellow Sea 243 (cluster 3: 40%). In contrast, prevailing easterlies from the WNP mostly influence the 244 composition of TSP in summer. The cluster analysis for summer samples (Figure 1b) showed 245 four regimes, including transport from the Sea of Japan (cluster 1: 13%), WNP (cluster 2: 246 22%), South China Sea (cluster 3: 46%), and East China Sea (cluster 4: 19%). In contrast, 247 cluster analysis revealed three major transport regimes from East Asia in fall and winter 248 (Figure 1c-d). However, there exist subtle differences between winter and fall in terms of 249 influence from nearby versus distant pollution sources. For instance, long-range transport of 250 air masses from west Mongolia (cluster 1: 21%) and Russian Far East (cluster 3: 14%) 251 weakly influence Gosan in winter. Besides, we observed somewhat a larger impact of air 252 masses from the North China Plain over Gosan (cluster 2: 64%) in winter. In contrast, Gosan 253 is less influenced by air masses originating from the North China Plain, contributing $\sim 28\%$ 254 (cluster 1) than those from Mongolia (cluster 2: 39%) and Russian Far East (cluster 3: 33%) 255 in fall. Therefore, the impact of East Asian outflow is stronger in winter than in spring and 256 fall.







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Figure 2. (a) Cumulative concentration levels of anhydrosugars, primary sugars, and sugar alcohols (i.e., represented by bars), and depicting the contributions of nonfossil (green color) and fossil (black color) organic carbon (i.e., pie charts), (b) Cumulative concentration levels of isoprene- and monoterpene SOA tracers in each TSP sample collected over Gosan.

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265 **3.2. Temporal and Seasonal Variability of sugars**

266 The temporal/seasonal trends of sugar compounds over KCOG provide useful information on 267 the emission strengths of various sources in the East Asian outflow. All three anhydrosugars 268 showed similar temporal and seasonal trends with higher concentrations in winter and fall than spring and summer (Figure 2a). As levoglucosan and two other anhydrosugars 269 270 (mannosan and galactosan) mostly originate from the pyrolysis of cellulose/hemicellulose, 271 their higher loading along with more nonfossil fraction of OC (Figure 2a; pie charts) in 272 winter and fall indicate the impact of BB emissions. The MODIS satellite-based fire counts 273 (Figure S1) together with cluster analysis in winter and fall (Figure 1) have revealed an 274 influence of active BB emissions in the North China Plain, Mongolia and Russian Far East. 275 The temporal trends of glucose, fructose, and sucrose exhibited less variability throughout the 276 sampling period; however, a slight increase in their concentration was observed in





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spring/summertime TSP samples (Figure 2a). Glucose and fructose have origins from leaf fragments and pollen species (Fu et al., 2012a). Sucrose is a potential tracer for airborne pollen (Fu et al., 2012a) and late spring/early summer is often regarded as a season for "pollen-allergies". Therefore, similar temporal trends of glucose and fructose with sucrose indicate their common source, pollens (Figure 2a).

282 BB also contributes to xylose and, hence, the temporal variability of xylose is 283 mimicking that of anhydrosugars. Trehalose is a primary sugar and can be used as a tracer for organic carbon associated with soil dust particles (Fu et al., 2012a). Besides, the temporal 284 285 variability of trehalose closely resembled that of fungal spore tracers (arabitol, mannitol, and 286 erythritol), showing high concentrations in spring, summer, and fall seasons (Figure 2a) (Zhu et al., 2015a; Fu et al., 2012a). KCOG is influenced by the large scale of advection of mineral 287 288 dust from East Asia to the WNP during these three seasons (Tyagi et al., 2017; Huebert et al., 2003). The impact of mineral dust in the East Asian outflow can be traced by the high 289 concentrations of non-sea-salt Ca²⁺ in the TSP samples from Gosan (Arimoto et al., 1996). 290 Similar temporal trends of trehalose and nss-Ca²⁺, particularly in spring samples (Figure S2), 291 292 suggest that the origin of fungal spores over Gosan is associated with the impact of Kosa 293 (Asian dust) events. The temporal variability of inositol is different from other sugar alcohols 294 (Figure 2a).

295 The seasonally averaged concentrations of all the anhydrosugars and xylose are 296 somewhat higher in winter/fall than spring/summer (Figure 3a-d), possibly due to a greater 297 influence of long-range transport from East Asia. In contrast, glucose, fructose, and sucrose 298 peaked in spring but decreased in other seasons (Figure 3e-g), mainly due to the contribution 299 of airborne pollen. Trehalose showed higher concentrations in fall and summer, followed by 300 spring and winter (Figure 3h). Arabitol, mannitol, and erythritol showed higher 301 concentrations in summer/fall than winter and spring (Figure 3i-k). This seasonal trend is 302 consistent with those of soil-derived fungal spores. This feature was also observed in aerosols 303 collected from the oceanic island (Okinawa) in the WNP during the impact of East Asian 304 outflow (Zhu et al., 2015a). The seasonally averaged mass concentrations of inositol are 305 highest in spring, followed by summer, fall and winter (Figure 31). Overall, the molecular compositions of anhydrosugars are characterized by the predominance of levoglucosan 306 307 followed by galactosan and mannosan (Figure S3). Although the temporal variability of 308 primary sugars in the TSP samples from Gosan showed a characteristic peak of glucose 309 and/or sucrose (Figure 2a), the seasonally averaged distributions are somewhat different





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- 310 (Figure S3). The molecular distributions of sugar alcohols are characterized by high loadings
- 311 of arabitol and mannitol, followed by erythritol and inositol (Figure S3).

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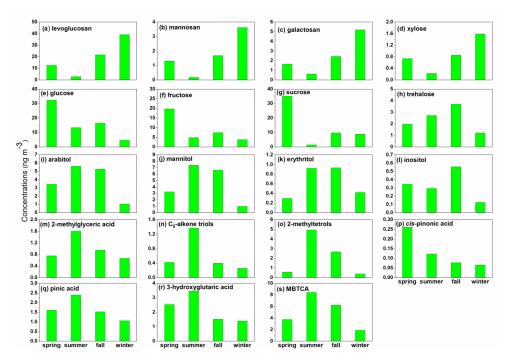


Figure 3. Seasonal variability of atmospheric levels of sugar compounds and BSOA tracers
 in TSP samples from Gosan during April 2013-April 2014.

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317 **3.3. Temporal and Seasonal Variability of BSOA tracers**

Six isoprene-SOA tracers such as 2-methylglyceric acid (2-MGA), three C5-alkene triols, and 318 319 two 2-methyltetrols (2-methylthreitol and 2-methylerythritol) (2-MTs) were identified in Gosan aerosol samples (Table 1). The sum of isoprene-SOA tracers ranged from 0.35 to 15.1 320 ng m⁻³ (avg. 3.69 ng m⁻³) with the predominance of 2-MTs (avg. 2.09 ng m⁻³). 2-MGA is the 321 second most abundant isoprene-SOA tracer (avg. 0.99 ng m⁻³), a high-generation product 322 323 probably formed by further photooxidation of methacrolein and methacrylic acid. A similar 324 molecular composition was observed over the North Pacific and California Coast (Fu et al., 325 2011). All the isoprene SOA tracers exhibited similar temporal variations with higher 326 concentrations in summer/spring months compared to autumn and winter (Figure 2b). Conversely, four monoterpene-SOA tracers, i.e., cis-pinonic acid, pinic acid, 3-327





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328 hydroxyglutaric acid (3-HGA), and 3-methyl-1,2,3-butanetricarboxylic acid (MBTCA), were 329 detected in Gosan samples (Table 1). Total concentrations of monoterpene-SOA tracers were found to be 1.65 to 35.5 ng m⁻³ (avg. 9.24 ng m⁻³) with a high concentration of MBTCA 330 (avg. 5.11 ng m^{-3}). All the monoterpene-SOA tracers showed similar temporal trends with 331 332 high values in summer/spring periods than autumn/winter (Figure 2b). Nevertheless, cis-333 pinonic acid was ascribed somewhat different temporal variability with other monoterpene-334 SOA tracers. It can be assumed that *cis*-pinonic acid might be further photo-oxidized to form 335 MBTCA (Szmigielski et al., 2007).

336 The seasonally averaged isoprene-SOA tracers were characterized with high 337 concentrations in summer, followed by spring/fall and winter (Figure 3m-o). It is of interest to note that two fall samples (KOS984; 2-12 Sep and KOS986; 27 Sep to 07 Oct) exhibited a 338 339 high concentration of 2-MTs over Gosan (Figure 2b). We presumed that local vegetation might contribute significantly to the formation of 2-MTs as it is a first-generation product. 340 341 Moreover, 2-MTs can be derived from the open ocean under low NO_x conditions (Hu et al., 342 2013). 3-HGA and pinic acid showed somewhat higher concentrations in summer/spring than 343 fall/winter due to the growing vegetation (Figure 3q-r). Cis-pinonic acid was more abundant 344 in spring compared to summer (Figure 3p) because of its photo-degradation, as discussed 345 above. In contrast, MBTCA was found to be dominant in summer/fall than spring/winter 346 (Figure 3s). The formation of MBTCA could be enhanced in fall season during atmospheric 347 transport from the East Asian continent. The molecular distributions of isoprene-SOA tracers 348 were characterized by a high loading of 2-MTs, followed by 2-MGA and C₅-alkene triols in 349 all seasons (Figure S3). Molecular composition of monoterpene-SOA tracers was dominated 350 by MBTCA, followed by 3-HGA, pinic acid, and *cis*-pinonic acid in all seasons (Figure S3). 351 Overall, BSOA tracers were found to be most abundant in summer, followed by fall, spring, 352 and winter (Table 1).

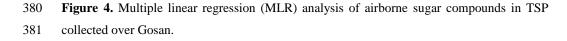
353 Kang et al. (2018b) reported that monoterpene-SOA tracers were more abundant than 354 isoprene-SOA tracers in spring-summer over the East China Sea, which is consistent with this 355 study. The major fraction of isoprene and monoterpenes is emitted from terrestrial plants; 356 however, the open ocean can emit isoprene and monoterpenes significantly (Conte et al., 2020; Shaw et al., 2010; Broadgate et al., 1997). Air mass back trajectory and cluster analysis 357 358 (Figure 1) implied that air masses mostly originated from the ocean during summer. It 359 indicates that the open ocean significantly contributed to isoprene-SOA production. However, 360 terrestrial sources from the continent also substantially enhanced the formation of BSOA. For 361 Example, one sample (KOS: 979; 1-11 June 2013) during summer showed the highest





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362	loading of BSOA tracers (Figure 2b), when air masses are transported from the continent
363	(Zhang et al., 2016).
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371	⁹ / ₈ 0.09 0.27 0.09 0.89 0.94 Suc
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377	0 -0.11 -0.01 -0.07 -0.21 -0.3 -0.09 0.59 0.57 0.59 Erythr
378	$\begin{bmatrix} 9 \\ -0.24 \\ 8 \end{bmatrix} = -0.24 \begin{bmatrix} -0.22 \\ -0.18 \end{bmatrix} \begin{bmatrix} -0.22 \\ 0.59 \end{bmatrix} \begin{bmatrix} 0.45 \\ -0.45 \end{bmatrix} \begin{bmatrix} 0.33 \\ -0.23 \end{bmatrix} \begin{bmatrix} -0.23 \\ 0.76 \end{bmatrix} \begin{bmatrix} 0.72 \\ 0.72 \end{bmatrix} \begin{bmatrix} 0.82 \\ 0.34 \end{bmatrix} \begin{bmatrix} 0.34 \\ 0.034 \end{bmatrix} $ Inos
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383 3.4. Source Apportionment- Regression Analysis and Diagnostic Ratios

Anhydrosugars strongly correlated with xylose (Figure 4), suggesting their common source as BB emission in East Asia. Fu et al. (2012a) analyzed pollen from different tree species (*e.g.*, White birch, Chinese willow, Peking willow, Sugi, Hinoki), which are enriched in sucrose (182-37,300 μ g g⁻¹), glucose (378-3601 μ g g⁻¹) and fructose (162-1813 μ g g⁻¹). In our samples, sucrose strongly correlated with glucose and fructose (Figure 4), suggesting their





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389 origin from plant-derived airborne pollen. Likewise, a strong correlation was found between 390 arabitol and mannitol, indicating a mutual origin from a similar type of fungal spores (Zhu et 391 al., 2015a; Fu et al., 2012a; Bauer et al., 2008; Yttri et al., 2007). Bauer et al. (2008) ascribed 392 weak correlations between arabitol and mannitol to the diverse nature of ambient fungal 393 spores. Furthermore, both sugar alcohols correlated well with trehalose, a tracer for soil 394 organic carbon (Fu et al., 2012a). This observation suggests their common origin from soil 395 organic matter associated with fungal spores. Erythritol also originates from fungal spores; 396 however, its abundance is affected by the atmospheric aging process. Kessler et al. (2010) 397 reported that erythritol is degraded during long-range transport in 12.7 days. Consequently, 398 arabitol and mannitol were moderately correlated with erythritol in the Gosan samples due to 399 the degradation of latter sugar alcohol in the East Asian outflow.

400 The linear relationship of levoglucosan (Lev) with mannosan (Man), galactosan (Gal), 401 and nss-K⁺ provide useful information on the type of BB emissions (hardwood, softwood, or 402 crop-residue). Different types of biomass are characterized by the distinct Lev/Man ratios. For 403 instance, Lev/Man ratios from the softwood burning emissions (3-10) are different from those 404 of hardwood (15-25) and crop-residues (>40) (Singh et al., 2017; Schmidl et al., 2008a, b; Fu 405 et al., 2008; Engling et al., 2006, 2009; Fine et al., 2001, 2004). We found the Lev/Man ratios 406 (Table 2) over KCOG overlap between seasons and somewhat close to that of hardwood 407 burning emissions in northern China, Mongolia, and Russian Far East, as corroborated by the 408 backward air mass trajectories and MODIS fire counts (Figures 1 and S1). Besides, Lev/K^+ 409 and Man/Gal ratios in summer are different from those of other seasons (Table 2). Cheng et 410 al. (2013) have apportioned qualitatively the possible source contributions of anhydrosugars 411 over a receptor site based on the comparison of Lev/K^+ and Lev/Man ratios in aerosols to 412 those from various sources profiles compiled from the literature. This approach of using the mass ratios of Lev/Man and Lev/K⁺ has been proven useful for deciphering the difference in 413 414 BB-derived OA (Bikkina et al., 2019).

415 Here, we adopted this methodology to ascertain the likely contributing sources of 416 anhydrosugars, which are BB tracers from different seasons (Figure 5). This source 417 apportionment relies on the fact that Lev/K^+ from the softwood burning (10-1000) is 418 somewhat higher than that of hardwood (1-100) (Fine et al., 2004). In contrast, Lev/Man 419 ratios for softwood are lower than those of hardwood burning (10-100) (Fine et al., 2004). 420 Likewise, Lev/K⁺ and Lev/Man ratios from grasses and crop-residues are 10-100 and 0.01-421 1.0, respectively (Bikkina et al., 2019). On a similar note, the Lev/K^+ ratios from the burning 422 of pine needles (0.1-1.0) somewhat overlap with hardwood burning emissions but are

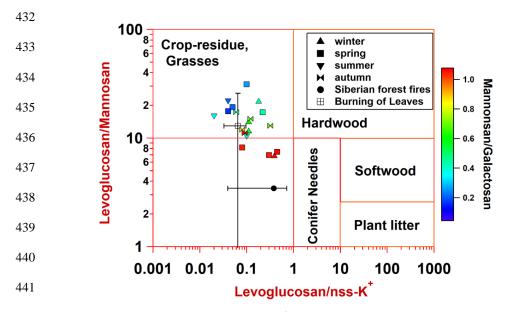


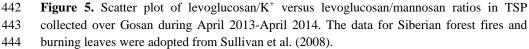


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423 characterized by distinct Lev/Man ratios (Bikkina et al., 2019). The burning of dead leaves (duff) showed higher Lev/K^+ than that of pine needles and grasses, but their Lev/Man is on 424 425 the lower side than the former biomass type and the softwood burning emissions. However, 426 Lev is more susceptible to degradation during transport and, hence, the hardwood Lev/K^+ 427 ratios could slightly shift downwards. Therefore, caution is required while interpreting the 428 ambient data from a receptor site (Bikkina et al., 2019). Overlapping the seasonal data on this 429 scatter plot of Lev/K⁺ versus Lev/Man (Figure 5) clearly revealed a mixed contribution of 430 burning of hardwood and crop-residue in the East Asian outflow.

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The correlation coefficients and diagnostic ratios of BSOA tracers specify their source origin or formation pathway. 2-MGA showed significant correlation with 2-MTs (r = 0.79, p<0.01) and C₅-alkene triols (r = 0.50, p<0.05) (Figure 6a-b), suggesting their similar formation pathway or common sources of isoprene SOA tracers. However, a poor correlation coefficient between 2-MTs and C₅-alkene triols (r = 0.33, p = 0.13) (Figure 6c) indicates their different formation process over the Gosan atmosphere. Wang et al. (2005) documented that polyols are formed from isoprene through diepoxy derivatives, which further convert into 2-

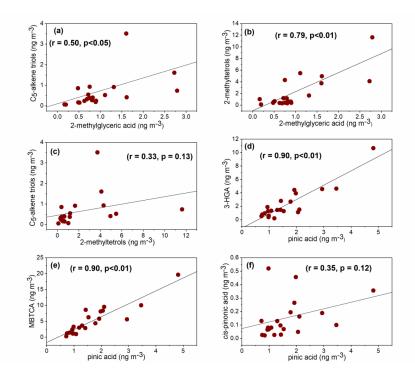




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453 MTs by acid-catalyzed hydrolysis. On the other hand, C5-alkene triols are produced from the 454 precursor of hydroxyperoxy radicals that are initially derived from isoprene through 455 rearrangement reactions (Surratt et al., 2006). It can be noted that the formation mechanisms of 2-MGA and 2-MTs are different while depending on the NO_x concentrations (Surratt et al., 456 457 2010). Thus, the ratio of 2-MGA/2-MTs attributes to the influence of NO_x on isoprene SOA 458 formation. We found a low ratio of 2-MGA/2-MTs (0.67) (Figure S4a, Table 2) in summer, 459 implying enhancement of 2-MTs formation over the open ocean due to the low-NO_x environment in the ocean-atmosphere. On the contrary, 2-MGA/2-MTs ratios for other 460 461 seasons were >1.0 (Figure S4a, Table 2), indicating an elevated formation of 2-MGA through 462 continental high NO_x condition, which is consistent with air masses back trajectory.

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Figure 6. Pearson linear correlation coefficient analysis of BSOA tracers in Gosan TSP
 aerosols during April 2013-April 2014.

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468 *Cis*-pinonic acid showed poor correlation with pinic acid (r = 0.35, p = 0.12) (Figure 469 6f), suggesting that they formed from different monoterpenes such as α -pinene, β -pinene, or





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 δ -limonene. In contrast, pinic acid exhibited a strong linear correlation with 3-HGA (r = 0.90, p<0.01) and MBTCA (r 0.90, p<0.01) (Figure 6d-e), implying their similar sources. It should be noted that the formation processes of pinic acid, 3-HGA, and MBTCA are different because pinic acid is a first-generation product, and 3-HGA and MBTCA are high-generation products (Claeys et al., 2007; Müller et al., 2012; Szmigielski et al., 2007). The ratio of cispinonic acid + pinic acid to MBTCA (P/M) is used to evaluate the aging of monoterpene SOA.

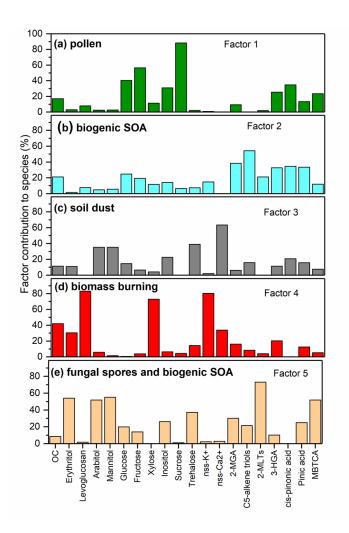
477 A low P/M ratio suggests possible transformation of cis-pinonic and pinic acids to 478 MBTCA and thus relatively aged monoterpene-SOA, whereas a high ratio reflects relatively 479 fresh monoterpene-SOA (Gómez-González et al., 2012; Ding et al., 2014). Gómez-González et al. (2012) reported aged monoterpene-SOA (P/M = 0.84) from a Belgian forest site while 480 481 fresh chamber-produced α -pinene-SOA tracers showed P/M ratios of 1.51 to 3.21 (Offenberg 482 et al., 2007). The average ratio of P/M in this study showed 0.62 with the low value in 483 summer (Figure S4b, Table 2), which is lower than those of Guangzhou (fresh monoterpene-484 SOA; 28.9) while the air masses originated from southern China (Ding et al., 2014). It 485 indicates that monoterpene SOA in Gosan aerosol was relatively aged, particularly in summer 486 when extensive photochemical oxidation occurred due to the high temperature and intense 487 solar radiation.

488 3.5. Source Apportionment - Positive Matrix Factorization (PMF) Analysis

489 We used the positive matrix factorization (PMF) analysis of airborne sugar compounds and 490 BSOA tracers to assess their sources over Gosan. Five-factor profiles were chosen that could 491 best explain the data (Figure 7). In factor 1, we found higher loadings of sucrose (88%), 492 fructose (57%), and glucose (40%). Sucrose is a well-known tracer for airborne pollen as 493 these types of biogenic sources also contain significant levels of glucose and fructose; we attribute this factor to contribution from "pollen". Likewise, factor 2 has higher loadings of 494 495 C₅-alkene triols (54%), 2-MGA (38%), *cis*-pinonic acid (34%), pinic acid (33%), and 3-HGA 496 (33%). These SOA tracers are formed by the photochemical oxidation of BVOCs (e.g., 497 isoprene and monoterpene), which are emitted from terrestrial vegetation and oceanic 498 phytoplankton. Therefore, we attribute factor 2 to biogenic secondary oxidation products. Factor 3 was abundantly loaded by non-sea-salt fraction of calcium (nss- Ca^{2+}), implying that 499 the component is associated with soil dust as $nss-Ca^{2+}$ is a specific tracer of Earth surface soil 500 501 (Athanasopoulou et al., 2010; Brahney et al., 2013). It should be noted that significant 502 loadings were also observed for trehalose (39%), glucose (35%), and fructose (35%) in factor 503 3, indicating their association with soil microbes.







504

505 **Figure 7.** Positive Matrix Factorization (PMF)- based Source profiles resolved for the 506 measured primary saccharides in TSP collected over Gosan during April 2013-April 2014.

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We found higher loadings of levoglucosan (83%), nss-K⁺ (80%), and xylose (73%) in factor 4. Large contributions were observed for the levoglucosan and K⁺ (hemicellulose/cellulose pyrolysis tracers) and xylose (wood sugar). We, therefore, ascribe this factor to BB emissions. Factor 5 is dominated by 2-MTs (73%), MBTCA (53%), mannitol (55%), erythritol (54%), and arabitol (52%). 2-MTs and MBTCA are BSOA tracers, which are produced by the photo-oxidation of isoprene and monoterpene, respectively. In contrast, arabitol, mannitol and erythritol are typical tracers of airborne fungal spores (Lewis





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and Smith, 1967; Bauer et al., 2008). Thus, factor 5 is associated with mixed sources of 'fungal spores' and 'biogenic secondary oxidation products'. PMF results were further utilized to calculate the relative contributions from various sources to OC over Gosan (Figure S5). BB was the dominant source of OC over Gosan (41.9%), followed by BSOA (21.1%), pollen (17.1%), and soil dust (11.4%).

521 3.6. Relative Abundances in WSOC and OC

522 Levoglucosan is the most abundant anhydrosugar, contributing 0.05-1.54% of WSOC and 523 0.03-1.02% of OC. Likewise, sucrose, glucose, and fructose were more abundant primary 524 sugars, contributing 0.01-2.83%, 0.03-1.22%, and 0.02-0.74% of WSOC, respectively. 525 Contributions of the three primary sugars largely varied from 0.05% to 3.41% of OC. 526 Arabitol and mannitol are the most abundant sugar alcohols, whose contribution to WSOC 527 and OC ranged from 0.02% to 0.93% and 0.01% to 0.85%, respectively. Figure 8 depicts the 528 contribution of sugar compounds to WSOC and OC in TSP collected over Gosan during the 529 study period. We also compared the atmospheric abundances of sugar compounds from 530 Gosan with the literature data (Table 3). This comparison revealed the less influence of BB 531 tracer compounds (i.e., anhydrosugar levels) over Gosan, that is, a factor of 5-10 times lower 532 than those reported for the BB-influenced source regions in China and East Asia (Fu et al., 533 2012b; Kang et al., 2018a; Wang and Kawamura, 2005; Wang et al., 2012). However, the 534 levels of anhydrosugar over Gosan are higher than those observed over the remote Canadian 535 High Arctic site (Fu et al., 2009a). In contrast, Gosan is characterized by high concentrations 536 of primary sugars compared to other remote sampling sites in Table 3. This could be 537 explained by the overwhelming contribution of primary sugars associated with soil dust 538 particles over Gosan during the East Asian outflow. Such high loadings of primary sugars 539 were observed from other remote island receptor sites in the WNP (Okinawa) during the spring season (Zhu et al., 2015a). Likewise, the concentrations of sugar alcohols from Gosan 540 541 are somewhat similar to those from other receptor sites influenced by the East Asian outflow 542 (Verma et al., 2018; Zhu et al., 2015a).

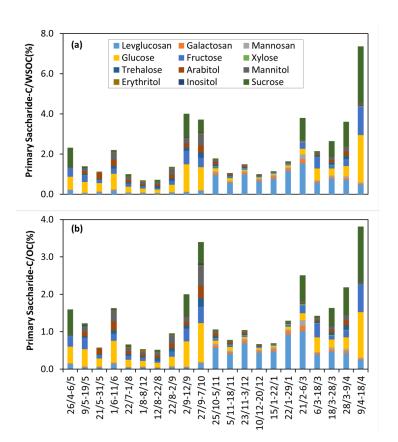
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Figure 8. Contribution of primary saccharide-C in (a) WSOC(%) and (b) OC(%) in TSP
collected over Gosan during April 2013-April 2014.

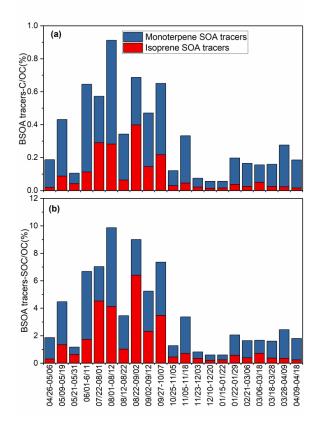
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551 Contributions of isoprene-SOA tracers to ambient OC (0.01-0.40%, ave. 0.09%) and 552 WSOC (0.02-0.57%, 0.13%) were lower than monoterpene-derived SOA (0.04-0.63%, 553 0.23% for OC and 0.06-0.82%, 0.33% for WSOC). The contributions of isoprene oxidation 554 products to OC and WSOC were found to be highest in summer (0.23%) and 0.32%, 555 respectively), followed by fall (0.09% and 0.13%), spring (0.04% and 0.06%), and winter 556 (0.02% and 0.03%). Likewise, the contributions of monoterpene-SOA products to aerosol OC 557 and WSOC exhibited highest value in summer (0.40% and 0.55%, respectively), followed by 558 fall (0.24% and 0.35%), spring (0.18% and 0.27%), and winter (0.10% and 0.14%). We 559 found that the contribution of BSOA products to carbonaceous components was twice in 560 summer than those of other seasons (Figure 9, Table 2). This result indicates that SOA 561 formation was occurred in summer to a greater extent due to the intensive BVOCs emission 562 with key factors of meteorological parameters (higher temperature and radiation).





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Figure 9. Contribution of (a) isoprene- and monoterpene SOA tracers-C in ambient OC(%)
and (b) isoprene- and monoterpene SOA tracers-SOC in ambient OC(%) in Gosan aerosol
samples during April 2013-April 2014.

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569 We estimated secondary organic carbon (SOC) derived from isoprene and 570 monoterpenes, using the measured values of BSOA tracers and following the SOA tracer-571 based method first proposed by Kleindienst et al. (2007). A summary of the estimated SOC is 572 provided in Table 2. The contribution of isoprene to SOC was calculated 2.26 to 97.4 ngC m⁻ ³ (avg. 23.7 ngC m⁻³), accounting for 1.45% of OC and 35.5% of total BSOC with the 573 574 predominance in summer (3.56% and 46.6% for OC and SOC, respectively). The estimation of monoterpene SOA to SOC (avg. 40.1 ngC m⁻³) was observed around two times higher than 575 isoprene (Table 2). Interestingly, the contribution of monoterpene-derived SOC to ambient 576 OC (3.65%) was dominant in summer, but monoterpene SOC to total SOC (72.1%) was most 577





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abundant in spring (Figure 9b). The seasonal distributions of biogenic SOC (Table 2) implied
that a substantial amount of SOC was formed from monoterpenes during the spring season.
The estimated biogenic SOC in this study is almost one order of magnitude lower than those
from the continental site in south China urban area (PRD: 446 ngC m⁻³) (Ding et al., 2012);
however, much higher than a remote site in the Canadian High Arctic (Alert: 9.4 ngC m⁻³)
(Fu et al., 2009b) and comparable with over the East China Sea (Kang et al., 2018b).

584 It should be noted that concentrations of SOA tracers cannot always provide the actual contribution of the biogenic source to ambient organic aerosol mass. For example, 585 loadings of monoterpene-SOA tracers were lower in sample KOS999 (28 March - 9 April 586 2014) (3.02 ng m⁻³) compared to sample KOS1000 (9 - 18 April 2014) (14.4 ng m⁻³), whereas 587 estimated contribution of SOC to ambient OC showed an opposite trend (KOS999: 2.08%; 588 589 KOS1000: 1.56%). This result demonstrates that the estimation of SOC is an important factor to evaluate the contribution of BSOA to organic aerosol mass. We calculated biogenic OC 590 591 using radiocarbon (^{14}C) data following the method proposed by Szidat et al. (2006). Biogenic OC showed poor correlation with biogenic SOC (r = 0.36, p = 0.09) but significant linear 592 593 relationship with primary sugars (i.e., glucose, fructose and sucrose) (r = 0.54, p<0.5), 594 suggesting that primary bioaerosols from plant-derived airborne pollen dictate on biogenic 595 OC in Gosan aerosols.

596 **3.7. Significance of Fossil Fuel as a source for Levoglucosan**

The ambient levoglucosan levels showed a significant linear relationship with the OC_{fossil}, 597 598 suggesting the fossil source contribution of this molecular marker (Figure 10a). However, 599 such a significant correlation was not evident between OC_{fossil} and other major sugar 600 compounds. Until recently, Lev has been thought to originate primarily from the 601 hemicellulose/cellulose pyrolysis of vegetation and, hence, can be employed as a powerful tracer for biomass smoke particles (Fraser and Lakshmanan, 2000; Simoneit et al., 1999). 602 Residential coals (e.g., lignite and bituminous coal) have been shown to contain high 603 concentrations of 'Lev' but also emit traces of Man and Gal (Kourtchev et al., 2011; Fabbri et 604 al., 2008). Recently, Yan et al. (2018) found a significant linear relationship between ¹⁴C-605 606 based fossil fraction of WSOC and Lev-C in the aerosols generated from coal combustion and the ambient aerosol samples. Therefore, the prevailing linear relationship between OC_{fossil} 607 608 and Lev-C in Gosan samples (Figure 10a) is likely due to a common source contribution from 609 coal combustion in East Asia.





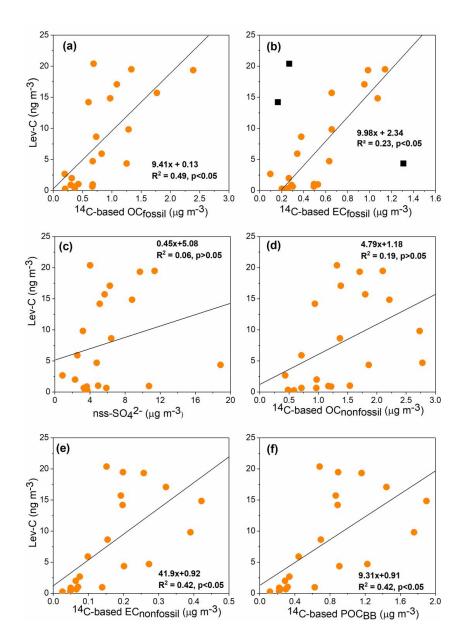




Figure 10. Linear regression analysis between levoglucosan in terms of their carbon content (*Lev-C*) and ¹⁴C-based mass concentrations of (a) organic carbon and (b) elemental carbon from fossil origin (OC_{fossil} and EC_{fossil} , respectively), (c) nss- SO_4^{2-} , (d) nonfossil derived organic carbon ($OC_{nonfossil}$), (e) nonfossil derived elemental carbon ($EC_{nonfossil}$), and (f) biomass burning derived primary OC (POC_{BB}) in TSP collected over Gosan during April 2013-April 2014. In panel (b), the squares represent three outliers (i.e., samples with rather high and low *Lev*/EC ratios; please see text for more discussion).





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618 The slope of the linear regression between Lev-C and OC_{fossil} (0.0094; Figure 10a) is 619 higher than those documented for the coal combustion source in China ($\sim 0.004 \pm 0.007$) (Yan 620 et al., 2018). Moreover, Lev-C moderately correlated with the EC_{fossil} with the regression 621 slope (~0.01) in Gosan samples (Figure 10b), being comparable to that observed for the coal 622 combustion in China (0.044±0.076) (Yan et al., 2018). It should be noted that excluding the 623 three outlier as shown in Figure 10b (black square), Lev-C showed stronger correlation with EC_{fossil} (R² = 0.74, p < 0.05). Of these, two outliers in winter (KOS995: 22-29 January 2014; 624 KOS996: 21 February - 3 March 2014) have higher Lev-C levels over that of EC_{fossil}, when 625 626 air mass trajectories showed the impact of BB emissions in the North China Plain. In 627 contrast, the third outlier in summer (KOS979: 1-11 Jun'2013) has a lower Lev-C/EC_{fossil}, 628 while air parcel transported from nearby cities in China, Korea and Japan, thus, have more 629 contribution from vehicular emissions. Overall, both the regression slopes are, thus, the 630 representative nature of Lev-C/OCfossil and Lev-C/ECfossil in the East Asian outflow. Lev-C and nss-SO₄²⁻ exhibited a poor correlation (Figure 10c) although both were transported from East 631 632 Asia.

Lev-C exhibited a rather weak correlation with $OC_{nonfossil}$ (R² = 0.19) than that with 633 $EC_{nonfossil}$ ($R^2 = 0.42$) over Gosan during the study period (Figure 10d-e). This could be likely 634 because OC_{nonfossil} has contributions from the BB and the secondary formation process or the 635 primary biogenic sources. The contribution of primary OC generated from BB (POC_{BB}) to 636 OC_{nonfossil} was taken from Zhang et al. (2016). In their study, the ¹⁴C-based EC_{nonfossil} levels 637 were scaled by a factor to constrain the POC_{BB} (Zhang et al., 2016). Here the conversion 638 639 factor is '4.5' (range: 3-10), which is a median value representing the primary OC/EC ratio 640 from BB emissions ((POC/EC)_{BB}).

641

$$POC_{BB} = EC_{nonfossil} \times (POC/EC)_{BB}$$

(7)

642 *Lev-*C showed a somewhat improved linear correlation with POC_{BB} than with the $OC_{nonfossil}$ 643 (Figure 10f). It is apparent from Figure 10 that the regression slopes are somewhat 644 comparable, indicating their contribution to *Lev* from both coal combustion and BB 645 emissions over Gosan.

646 4. Conclusions

We investigated seasonal variations of primary organic components such as anhydrosugars, primary sugars, and sugar alcohols and biogenic secondary organic aerosol (BSOA) tracers (isoprene- and monoterpene-derived SOA products) in ambient aerosols from Gosan, Jeju Island. Among the detected sugar compounds, levoglucosan was dominant in winter/fall, whereas glucose and sucrose were more abundant in spring/summer. The seasonal trends





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652 documented that BB impact is more significant in winter/fall and the primary bioaerosol 653 particles are important in spring/summer. Strong linear relationships of levoglucosan (Lev) 654 with galactosan (Gal) and mannosan (Man) together with their mass ratios of Lev/Gal 655 (6.7 ± 2.2) and Lev/Man (15.0±6.7), reflect their contribution from hardwood burning 656 emissions. On a similar note, sucrose showed significant correlations with glucose and 657 fructose, suggesting their origin from airborne pollen. The significant positive linear 658 relationship was found between the fungal spore tracer compounds (i.e., arabitol, mannitol 659 and erythritol) and trehalose (e.g., a proxy for soil organic carbon), implying their 660 contribution from airborne fungal spores and soil microbes in the East Asian outflow. These 661 sugar alcohol tracers are abundantly emitted in summer, followed by fall, spring, and winter.

662 Distributions of biogenic SOA tracers were characterized by a predominance of 663 monoterpene- than isoprene-derived oxidation products in Gosan aerosols. All the BSOA 664 tracers were dominant in summer, followed by fall/spring, and then winter. The low ratio of cis-pinonic acid + pinic acid to MBTCA (P/M) demonstrated that monoterpene SOA was 665 666 relatively aged over Gosan aerosols. The estimated secondary organic carbon (SOC) with the 667 predominance in summer indicates that substantial SOA formation occurred in summer due to the favorable meteorological conditions. The backward air mass trajectories and source 668 669 apportionment studies (i.e., linear regressions, diagnostic mass ratios, and positive matrix factorization analysis) entirely demonstrated that BB and biogenic SOA contribution mostly 670 671 dominate the ambient OA loadings in the East Asian outflow. Interestingly, Lev-C exhibited significant positive linear relationships with ¹⁴C-based nonfossil and fossil organic carbon 672 673 fractions along with the comparable regression slopes. This result reveals that BB and coal 674 (lignite) combustion both are important source for levoglucosan in the East Asian outflow.

Although there is some evidence that *Lev* could originate from the combustion of brown coals (*e.g.*, lignite) in China; however, our observations are from the KCOG (receptor site) also hinted for the fossil source contribution of this molecular marker in the regional influx of the East Asian outflow. Therefore, attribution of ambient *Lev* levels over the WNP to the impact of BB emission may cause large uncertainty.

680 Data availability

681 The data used in this paper are available upon the request from the corresponding author.

682 Author contributions

YLZ and KK designed the research. ML collected the aerosol samples. MMH and SBperformed the analysis of aerosol samples. MMH wrote the paper under the guidance of YLZ

and KK. All authors were actively involved in the discussion of the paper.





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686	Competing	interests
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687 The authors declare that they have no conflict of interest.

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- 692

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1062 **Table 1.** Concentrations of identified sugar compounds and BSOA tracers $(ng m^{-3})$ in the atmospheric 1063 aerosol samples from Gosan.

Species	Annual	Summer	Fall	Winter	Spring
	Avg. ^a \pm S.D. ^b Min. ^c , Max. ^d	Avg. ± S.D. Min., Max.			
Anhydrosugars					
Levoglucosan (Lev)	$\begin{array}{l} 17.6 \pm 16.8 \\ 0.60, 45.9 \end{array}$	$\begin{array}{c} 2.92 \pm 3.89 \\ 0.60, \ \ 9.81 \end{array}$	21.7 ± 19.0 2.30, 43.9	39.2 ± 6.60 32.0, 45.9	12.7 ± 11.6 1.45, 33.4
Mannosan (Man)	$\begin{array}{c} 1.57 \pm 1.82 \\ 0.05, 6.74 \end{array}$	$\begin{array}{c} 0.18 \pm 0.24 \\ 0.05, \ 0.61 \end{array}$	$\begin{array}{c} 1.69 \pm 1.49 \\ 0.13, \ \ 3.66 \end{array}$	3.63 ± 2.28 1.47, 6.74	$\begin{array}{c} 1.31 \pm 1.57 \\ 0.08, \ \ 4.08 \end{array}$
Galactosan (Gal)	$\begin{array}{rr} 2.28 \pm 2.10 \\ 0.14, & 6.78 \end{array}$	$\begin{array}{l} 0.64 \pm 0.68 \\ 0.14, \ 1.82 \end{array}$	$\begin{array}{c} 2.45 \pm 2.13 \\ 0.35, \ \ 4.92 \end{array}$	$\begin{array}{l} 5.21 \pm 1.64 \\ 3.40, \ \ 6.78 \end{array}$	$\begin{array}{c} 1.65 \pm 1.26 \\ 0.50, 3.88 \end{array}$
Primary Sugars					
Glucose	$\begin{array}{c} 18.8 \pm 27.1 \\ 2.45, \ 122 \end{array}$	$\begin{array}{c} 13.4 \pm 18.2 \\ 2.45, \ \ 45.6 \end{array}$	16.5 ± 15.7 2.68, 33.6	4.74 ± 3.14 2.88, 9.44	32.4 ± 41.1 4.87, 122
Fructose	$\begin{array}{c} 10.3 \pm 15.9 \\ 0.97, \ \ 74.0 \end{array}$	$\begin{array}{l} 4.90 \pm 5.15 \\ 0.97, 13.7 \end{array}$	7.48 ± 6.99 1.71, 16.2	3.82 ± 4.45 1.56, 10.5	19.8 ± 25.0 2.69, 74.0
Sucrose	16.1 ± 32.2 0.26, 140	1.46 ± 1.05 0.68, 3.28	9.74 ± 12.1 0.76, 27.2	$\begin{array}{c} 8.87 \pm 16.3 \\ 0.42, \ \ 33.3 \end{array}$	35.1 ± 50.5 0.26, 140
Trehalose	$\begin{array}{c} 2.42 \pm 1.97 \\ 0.65, \ \ 7.03 \end{array}$	$\begin{array}{c} 2.72 \pm 2.53 \\ 0.97, 6.98 \end{array}$	3.71 ± 2.29 1.18, 7.03	$\begin{array}{c} 1.21 \pm 0.42 \\ 0.72, \ 1.63 \end{array}$	$\begin{array}{c} 1.98 \pm 1.55 \\ 0.65, \ \ 5.33 \end{array}$
Xylose	$\begin{array}{c} 0.81 \pm 0.65 \\ 0.04, \ \ 2.03 \end{array}$	$\begin{array}{c} 0.23 \pm 0.23 \\ 0.04, \ \ 0.63 \end{array}$	$\begin{array}{l} 0.86 \pm 0.68 \\ 0.14, \ 1.70 \end{array}$	$\begin{array}{c} 1.59 \pm 0.37 \\ 1.21, \ \ 2.03 \end{array}$	$\begin{array}{c} 0.74 \pm 0.55 \\ 0.16, 1.68 \end{array}$
Sugar alcohols					
Arabitol	3.96 ± 4.24 0.47, 18.7	5.64 ± 7.46 1.20, 18.7	5.27 ± 3.51 2.27, 10.9	$\begin{array}{c} 1.06 \pm 0.60 \\ 0.47, \ \ 1.91 \end{array}$	3.47 ± 2.19 1.18, 6.30
Mannitol	4.61 ± 5.54 0.25, 22.0	$\begin{array}{c} 7.39 \pm 8.65 \\ 1.71, \ \ 22.0 \end{array}$	6.64 ± 6.03 1.69, 16.7	$\begin{array}{c} 0.99 \pm 0.45 \\ 0.55, \ 1.60 \end{array}$	3.24 ± 2.67 0.25, 7.03
Erythritol	$\begin{array}{c} 0.62 \pm 0.43 \\ 0.12, \ 1.52 \end{array}$	$\begin{array}{c} 0.92 \pm 0.53 \\ 0.42, \ 1.52 \end{array}$	$\begin{array}{c} 0.93 \pm 0.33 \\ 0.55, \ 1.27 \end{array}$	$\begin{array}{c} 0.42 \pm 0.27 \\ 0.16, \ 0.80 \end{array}$	$\begin{array}{c} 0.30 \pm 0.12 \\ 0.12, \ \ 0.48 \end{array}$
Inositol	$\begin{array}{c} 0.34 \pm 0.39 \\ 0.04, \ 1.53 \end{array}$	$\begin{array}{c} 0.29 \pm 0.41 \\ 0.08, \ 1.03 \end{array}$	$\begin{array}{l} 0.56 \pm 0.60 \\ 0.10, \ 1.53 \end{array}$	$\begin{array}{c} 0.13 \pm 0.07 \\ 0.08, \ \ 0.23 \end{array}$	$\begin{array}{c} 0.35 \pm 0.28 \\ 0.04, \ \ 0.73 \end{array}$
Isoprene SOA tracers					
2-MGA	$\begin{array}{c} 0.99 \pm 0.70 \\ 0.17, \ \ 2.79 \end{array}$	1.61 ± 1.17 0.17, 2.79	$\begin{array}{c} 0.95 \pm 0.43 \\ 0.51, \ 1.61 \end{array}$	$\begin{array}{c} 0.67 \pm 0.14 \\ 0.49, \ \ 0.81 \end{array}$	$\begin{array}{c} 0.76 \pm 0.35 \\ 0.20, \ \ 1.32 \end{array}$
Σ2-MLTs	$\begin{array}{c} 1.04 \pm 1.40 \\ 0.05, \ \ 5.81 \end{array}$	$\begin{array}{c} 2.48 \pm 1.98 \\ 0.51, \ \ 5.81 \end{array}$	1.34 ± 1.16 0.33, 2.74	$\begin{array}{c} 0.20 \pm 0.05 \\ 0.15, \ 0.26 \end{array}$	$\begin{array}{c} 0.29 \pm 0.25 \\ 0.05, \ 0.82 \end{array}$
Σ C5-alkene triols	$\begin{array}{c} 0.20 \pm 0.25 \\ 0.02, \ \ 1.17 \end{array}$	0.46 ± 0.44 0.02, 1.17	$\begin{array}{c} 0.13 \pm 0.05 \\ 0.05, \ \ 0.18 \end{array}$	$\begin{array}{c} 0.09 \pm 0.03 \\ 0.05, \ \ 0.13 \end{array}$	$\begin{array}{c} 0.14 \pm 0.11 \\ 0.02, \ \ 0.30 \end{array}$
Monoterpene SOA trac	ers				
cis-pinonic acid	$\begin{array}{c} 0.15 \pm 0.14 \\ 0.02, \ \ 0.52 \end{array}$	$\begin{array}{c} 0.12 \pm 0.13 \\ 0.03, \ 0.36 \end{array}$	$0.08 \pm 0.06 \\ 0.02, 0.16$	$\begin{array}{c} 0.07 \pm 0.03 \\ 0.02, \ \ 0.10 \end{array}$	$\begin{array}{c} 0.26 \pm 0.17 \\ 0.08, \ 0.52 \end{array}$
pinic acid	$\begin{array}{c} 1.67 \pm 1.01 \\ 0.72, \ \ 4.81 \end{array}$	2.40 ± 1.67 0.99, 4.81	$\begin{array}{c} 1.53 \pm 0.57 \\ 0.75, \ \ 2.11 \end{array}$	$\begin{array}{c} 1.07 \pm 0.25 \\ 0.84, \ 1.42 \end{array}$	$\begin{array}{c} 1.61 \pm 0.78 \\ 0.72, \ \ 2.94 \end{array}$
3-HGA	2.30 ± 2.38 0.19, 10.6	3.46 ± 4.39 0.19, 10.6	1.52 ± 0.77 0.76, 2.79	$\begin{array}{c} 1.40 \pm 0.39 \\ 0.94, \ 1.88 \end{array}$	$\begin{array}{c} 2.54 \pm 1.83 \\ 0.38, \ \ 4.57 \end{array}$
MBTCA	5.11 ± 4.54 0.29, 19.6	8.44 ± 6.88 3.00, 19.6	6.24 ± 3.64 1.17, 9.49	$\begin{array}{c} 1.90 \pm 0.85 \\ 0.97, \ \ 2.83 \end{array}$	3.77 ± 2.96 0.29, 8.08

^aAverage, ^bStandard deviation, ^cMinimum, ^dMaximum. 2-MGA: 2-methylglyceric acid, 2-MLTs: 2-methyltetrols, 3-HGA: 3-hydroxyglutaric acid, MBTCA: 3-methyl-1,2,3-butanetricarboxylic acid.





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1065 Table 2. Statistical summary of diagnostic ratios and carbonaceous components contribution in Gosan aerosols.

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Species	Annual	Summer	Fall	Winter	Spring
	Avg. ^a \pm S.D. ^b	Avg. \pm S.D.	Avg. \pm S.D.	Avg. \pm S.D.	Avg. \pm S.D.
	Min. ^c , Max. ^d	Min., Max.	Min., Max.	Min., Max.	Min., Max.
Diagnostic ratios					
Lev/Man	$\begin{array}{l} 15.1 \pm 6.76 \\ 6.81, 31.3 \end{array}$	$\begin{array}{l} 17.1 \pm 8.21 \\ 8.58, \ \ 28.4 \end{array}$	$\begin{array}{c} 13.7 \pm 2.51 \\ 11.1, \ 17.3 \end{array}$	$\begin{array}{c} 13.5 \pm 6.24 \\ 6.81, \ \ 21.7 \end{array}$	$\begin{array}{c} 15.4 \pm 8.76 \\ 6.98, \ \ 31.3 \end{array}$
Man/Gal	$\begin{array}{c} 0.55 \pm 0.32 \\ 0.10, \ \ 1.07 \end{array}$	$\begin{array}{c} 0.28 \pm 0.13 \\ 0.10, \ \ 0.42 \end{array}$	$\begin{array}{c} 0.64 \pm 0.20 \\ 0.37, \ 0.90 \end{array}$	$\begin{array}{c} 0.66 \pm 0.27 \\ 0.43, \ 1.05 \end{array}$	$\begin{array}{c} 0.61 \pm 0.43 \\ 0.16, \ 1.07 \end{array}$
$Lev/K^{*}\times 10^{-2}$	$\begin{array}{c} 5.73 \pm 5.65 \\ 0.65, \ \ 23.2 \end{array}$	$\begin{array}{c} 1.00 \pm 0.52 \\ 0.65, \ 1.91 \end{array}$	3.94 ± 2.51 1.26, 6.30	$\begin{array}{c} 12.3 \pm 7.87 \\ 6.42, \ \ 23.2 \end{array}$	6.65 ± 4.48 1.77, 15.3
$Lev/OC \times 10^{-2}$	$\begin{array}{c} 0.73 \pm 0.66 \\ 0.08, \ \ 2.29 \end{array}$	$\begin{array}{c} 0.14 \pm 0.10 \\ 0.08, \ \ 0.31 \end{array}$	$\begin{array}{l} 0.84 \pm 0.61 \\ 0.12, \ 1.56 \end{array}$	$\begin{array}{c} 1.60 \pm 0.67 \\ 0.99, \ \ 2.29 \end{array}$	$\begin{array}{l} 0.56 \pm 0.40 \\ 0.11, \ 1.05 \end{array}$
Lev/WSOC $\times 10^{-2}$	$\begin{array}{c} 1.09 \pm 0.97 \\ 0.10, 3.46 \end{array}$	$\begin{array}{c} 0.20 \pm 0.13 \\ 0.10, \ \ 0.42 \end{array}$	$\begin{array}{c} 1.25 \pm 0.94 \\ 0.23, \ \ 2.21 \end{array}$	$\begin{array}{c} 2.33 \pm 0.90 \\ 1.48, \ \ 3.46 \end{array}$	$\begin{array}{c} 0.92 \pm 0.65 \\ 0.13, \ 1.69 \end{array}$
MGA/MLTs	$\begin{array}{c} 2.18 \pm 1.59 \\ 0.33, \ \ 5.40 \end{array}$	$\begin{array}{c} 0.67 \pm 0.43 \\ 0.33, \ 1.33 \end{array}$	$\begin{array}{c} 1.04 \pm 0.49 \\ 0.41, \ 1.55 \end{array}$	$\begin{array}{l} 3.60 \pm 1.43 \\ 1.90, \ \ 5.40 \end{array}$	$\begin{array}{c} 3.27 \pm 1.17 \\ 1.60, \ \ 4.75 \end{array}$
^e P/MBTCA	$\begin{array}{c} 0.62 \pm 0.61 \\ 0.17, \ \ 5.90 \end{array}$	$\begin{array}{c} 3.22 \pm 0.62 \\ 2.47, \ \ 3.89 \end{array}$	$\begin{array}{l} 3.62 \pm 1.67 \\ 1.52, \ \ 5.90 \end{array}$	$\begin{array}{c} 1.68 \pm 0.64 \\ 0.83, \ \ 2.38 \end{array}$	$\begin{array}{c} 1.73 \pm 1.07 \\ 0.34, \ \ 3.32 \end{array}$
Carbonaceous components					
Isoprene derived SOC (µgC m ⁻³)	$\begin{array}{c} 23.7 \pm 23.9 \\ 2.26, \ \ 97.4 \end{array}$	$\begin{array}{l} 51.0 \pm 32.4 \\ 8.08, \ 97.4 \end{array}$	$\begin{array}{l} 25.9 \pm 17.9 \\ 8.54, \ \ 45.7 \end{array}$	$\begin{array}{l} 8.52 \pm 1.07 \\ 7.46, \ \ 9.87 \end{array}$	$\begin{array}{c} 11.3 \pm 6.80 \\ 2.26, \ \ 24.9 \end{array}$
Isoprene SOC to OC (%)	$\begin{array}{c} 1.45 \pm 1.74 \\ 0.21, \ \ 6.40 \end{array}$	$\begin{array}{l} 3.56 \pm 2.18 \\ 1.03, \ \ 6.40 \end{array}$	$\begin{array}{c} 1.46 \pm 1.38 \\ 0.35, \ \ 3.48 \end{array}$	$\begin{array}{c} 0.35 \pm 0.17 \\ 0.21, \ \ 0.57 \end{array}$	$\begin{array}{c} 0.57 \pm 0.38 \\ 0.24, \ 1.35 \end{array}$
Isoprene SOC to total SOC (%)	$\begin{array}{l} 35.5 \pm 15.5 \\ 13.5, \ \ 71.1 \end{array}$	$\begin{array}{l} 46.6 \pm 20.3 \\ 26.0, \ \ 71.1 \end{array}$	$\begin{array}{l} 38.1 \pm 10.4 \\ 21.1, \ \ 47.3 \end{array}$	31.6 ± 7.05 24.0, 39.7	$\begin{array}{l} 27.9 \pm 15.5 \\ 13.5, \ 54.0 \end{array}$
Monoterpene SOC (µgC m ⁻³)	40.1 ± 33.3 7.18, 154	$\begin{array}{l} 62.7 \pm 56.5 \\ 19.2, 154 \end{array}$	$\begin{array}{l} 40.7 \pm 20.0 \\ 11.8, \ 57.7 \end{array}$	$\begin{array}{c} 19.3 \pm 5.61 \\ 14.0, \ \ 25.2 \end{array}$	35.5 ± 23.5 7.18, 62.7
Monoterpene SOC to OC (%)	$\begin{array}{l} 2.0 \pm 1.47 \\ 0.37, \ \ 5.74 \end{array}$	$\begin{array}{l} 3.65 \pm 1.58 \\ 2.44, \ \ 5.74 \end{array}$	$\begin{array}{c} 2.15 \pm 1.45 \\ 0.48, \ \ 3.88 \end{array}$	$\begin{array}{c} 0.87 \pm 0.58 \\ 0.37, \ 1.48 \end{array}$	$\begin{array}{c} 1.58 \pm 0.84 \\ 0.54, \ \ 3.13 \end{array}$
Monoterpene SOC to total SOC (%)	64.5 ± 15.5 28.9, 86.5	$53.4 \pm 20.3 \\ 28.9, \ 74.0$	61.9 ± 10.4 52.7, 78.9	$\begin{array}{c} 68.4 \pm 7.05 \\ 60.3, \ \ 76.1 \end{array}$	$\begin{array}{l} 72.1 \pm 15.5 \\ 46.0, \ 86.5 \end{array}$

^eP: *cis*-pinonic acid + pinic acid





 Table 3. Comparisons of the concentration (ng m⁻³) of anhydrosugars, sugar and sugar alcohols in Gosan aerosols with those from different sites around the world.

Sampling sites	Anhydrosugars	Primary sugars	Sugar alcohols	References
Gosan, South Korea	21.4	48.4	9.53	This study
Chennai, India	131	12.5	6.13	Fu et al., 2010b
Mt. Tai, China (June)	224	61.1	125	Fu et al., 2012b
Alert, Canada	0.17	0.49	0.30	Fu et al., 2009a
Okinawa, western North Pacific	3.53	62.0	29.5	Zhu et al., 2015a, b
Chichijima, western North Pacific	1.13	23.3	23.4	Verma et al., 2015, 2018
Mt. Hua, China (Non-dust strom)	57.8	92.5	22.4	Wang et al., 2012
Mt. Hua, China (Dust strom)	44.5	162	25.7	Wang et al., 2012
Nanjing, China	209 (Lev)	12.7	50.8	Wang and Kawamura, 2005
Beijing, China	117	24.7	11.8	Kang et al., 2018a
Belgrade, Serbia	425 (Lev)	116	98.4	Zangrando et al., 2016
Maine, USA	13.9	28	8.31	Medeiros et al., 2006
Crete, Greece	14.4	32.3	6.53	Theodosi et al., 2018