1	Regional heterogeneities in the emission of airborne primary sugar
2	compounds and biogenic secondary organic aerosols in the East Asian
3	outflow: Evidence for coal combustion as a source of levoglucosan
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5	M. Mozammel Haque <sup>1,2,3</sup> , Yanlin Zhang <sup>1,2*</sup> Srinivas Bikkina <sup>4</sup> , Meehye Lee <sup>5</sup> , and Kimitaka
6	Kawamura <sup>3,4*</sup>
7	
8	<sup>1</sup> Yale-NUIST Center on Atmospheric Environment, International Joint Laboratory on Climate and
9 10	Environment Change (ILCEC), Nanjing University of Information Science & Technology, Nanjing, 210044, China
11	<sup>2</sup> School of Applied Meteorology, Nanjing University of Information Science & Technology, Nanjing
12 13	210044, China <sup>3</sup> Institute of Low Temperature Science, Hokkaido University, Sapporo 060-0819, Japan
13 14	<sup>4</sup> Chubu Institute for Advanced Studies, Chubu University, Kasugai 487-8501, Japan
15	<sup>5</sup> Department of Earth and Environmental Sciences, Korea University, Anam-dong, Sungbuk-gu, Seoul
16	136-701, South Korea
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26	
27	*Corresponding author
28	E-mail: dryanlinzhang@outlook.com (Yan-Lin Zhang)
29	E-mail: kkawamura@isc.chubu.ac.jp (Kimitaka Kawamura)
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### 31 ABSTRACT

32 Biomass burning (BB) significantly influences the chemical composition of organic aerosols 33 (OA) in the East Asian outflow. Source apportionment of BB-derived OA is an influential 34 factor for understanding their regional emissions, which is crucial for reducing uncertainties 35 in their projected climate and health-effects. We analyzed here three different classes of 36 atmospheric sugar compounds (anhydrosugars, primary sugars, and sugar alcohols) and two 37 types of biogenic secondary organic aerosol (BSOA) tracers (isoprene- and monoterpene 38 derived SOA products) in a year-long collected total suspended particulate matter (TSP) from 39 an island-based receptor site in South Korea, Gosan. We investigate seasonal variations in the 40 source emissions of BB-derived OA using mass concentrations of anhydrosugars and radiocarbon (<sup>14</sup>C-) isotopic composition of organic carbon (OC) and elemental carbon (EC) in 41 42 ambient aerosols. Levoglucosan (Lev) is the most abundant anhydrosugar, followed by 43 galactosan (Gal) and mannosan (Man). Strong correlations of Lev with Gal and Man, along 44 with their ratios (*Lev/Gal:* 6.65  $\pm$  2.26; *Lev/Man:* 15.1  $\pm$  6.76) indicate the contribution from 45 hardwood burning emissions. The seasonal trends revealed that the BB impact is more 46 pronounced in winter and fall, as evidenced by the high concentrations of anhydrosugars. 47 Likewise, significant correlations among three primary sugars (i.e., glucose, fructose, and 48 sucrose) emphasized the contribution of airborne pollen. The primary sugars showed higher 49 concentrations in spring/summer than winter/fall. The fungal spore tracer compounds (i.e., 50 arabitol, mannitol, and erythritol) correlated well with trehalose (i.e., a proxy for soil organic 51 carbon), suggesting the origin from airborne fungal spores and soil microbes in the East 52 Asian outflow. These sugar alcohols peaked in summer, followed by spring/fall and winter. 53 Monoterpene-derived SOA tracers were most abundant compared to isoprene-SOA tracers. 54 Both BSOA tracers were dominant in summer, followed by fall, spring, and winter. The 55 source apportionment based on multiple linear regressions and diagnostic mass ratios 56 together revealed that BB emission mostly contributed from hardwood and crop-residue burning. We also found significant positive linear relationships of <sup>14</sup>C-based nonfossil- and 57 58 fossil-derived organic carbon fractions with Lev-C along with the comparable regression 59 slopes, suggesting the importance of BB and coal combustion sources in the East Asian 60 outflow.

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Keywords: Biomass burning tracers, primary biological aerosol particles, biogenic SOA
 tracers, radiocarbon-based source apportionment, organic aerosols, East Asian outflow

### 64 **1. Introduction**

Organic aerosols (OA), which account for a major fraction of up to 50% of airborne total 65 66 suspended particulate matter, have considerable effects on regional and global climate by 67 absorbing or scattering sunlight (Kanakidou et al., 2005). However, the climate effects of OA 68 are involved with large uncertainties due to our limited understanding of the contributing 69 sources. OA can be derived from both primary emissions and secondarily formed species. 70 Sugars are an important group of water-soluble, primary organic compounds whose 71 concentrations are significant in atmospheric aerosols over the continent (Jia and Fraser, 72 2011; Fu et al., 2008; Yttri et al., 2007; Graham et al., 2003). Anhydrosugars such as 73 levoglucosan, galactosan, and mannosan are the key tracers of biomass burning (BB) 74 emissions (Simoneit, 2002). Sugar alcohols, along with glucose, trehalose and sucrose are 75 mostly originated from primary biological particles such as fungal spores, pollen, bacteria, 76 viruses, and vegetative debris (Graham et al., 2003; Simoneit et al., 2004a; Bauer et al., 2008; 77 Deguillaume et al., 2008). Primary sugars and sugar alcohols are predominantly present in the 78 coarse mode aerosols, accounting for 0.5-10% of atmospheric aerosol carbon matter (Yttri et 79 al., 2007; Pio et al., 2008).

80 Secondary organic aerosol (SOA) is a large fraction of OA, while there were only 81 limited studies about the key factors controlling SOA formation. The SOA formation 82 significantly increases with the enhancement of the ambient aerosol mass (Liu et al., 2018). 83 SOA is formed by both homogenous and heterogeneous reactions of volatile organic 84 compounds (VOCs) in the atmosphere (Surratt et al., 2010; Robinson et al., 2007; Claeys et 85 al., 2004). On a global estimation, biogenic VOCs (BVOCs) such as isoprene, monoterpenes 86 (e.g.,  $\alpha/\beta$ -pinene), and sesquiterpenes (e.g.,  $\beta$ -caryophyllene) are one order of magnitude 87 higher than those of anthropogenic VOCs (e.g., toluene) (Guenther et al., 2006). The global emissions of annual BVOCs were estimated to be 1150 TgC yr<sup>-1</sup>, accounting for 44% 88 89 isoprene and 11% monoterpenes (Guenther et al., 1995). Isoprene is highly reactive and 90 promptly reacts with oxidants such as O<sub>3</sub>, OH, and NO<sub>x</sub> in the atmosphere to form SOA 91 (Kroll et al., 2005, 2006; Ng et al., 2008; Surratt et al., 2010; Bikkina et al., 2021), estimated to be 19.2 TgC yr<sup>-1</sup>, consisting of ~70% of the total SOA budget (Heald et al., 2008). 92 93 Monoterpenes are important sources of biogenic secondary organic aerosol (BSOA), 94 considering  $\alpha$ -pinene as major species, accounting for ~35% of the global monoterpenes 95 emissions (Griffin et al., 1999).

Anthropogenic activities such as coal and biofuel combustion over East Asia,
including China, are responsible for the vast emission of OA (Huebert et al., 2003; Zhang et

98 al., 2016). Understanding the ambient levels OA in the East Asian outflow is crucial for 99 assessing their regional climatic effects. As part of this effort, the Korean Climate 100 Observatory at Gosan (KCOG), a supersite located in South Korea, is an ideal location for 101 investigating the atmospheric outflow characteristics from East Asia (Fu et al., 2010a; Kundu 102 et al., 2010; Ramanathan et al., 2007; Kawamura et al., 2004; Arimoto et al., 1996). For 103 instance, primary OA associated with soil/desert dust in East Asia, along with forest fires in 104 Siberia/northeastern China, are transported over Gosan in spring (Wang et al., 2009a). BSOA 105 during long-range transport from the continent and open ocean, as well as local vegetation, 106 can significantly contribute to Gosan aerosols. Although these investigations were carried out 107 almost a decade ago, no such observations are available in contemporary times from Gosan. 108 Here, we attempt to understand the current states of East Asian OA using both the molecular 109 marker approach and radiocarbon data of carbonaceous components.

110 The KCOG, located on the western side of Jeju Island adjacent to the Yellow Sea and the East China Sea, is facing the Asian continent but is isolated from public areas of the 111 112 island (Kawamura et al., 2004). Simoneit et al. (2004b) have documented during the ACE-113 Asia campaign that OA from the BB and fossil fuel combustion sources are transported along 114 with desert dust to the KCOG during continental outflow. An intensive campaign was 115 organized at the KCOG during spring 2005 to observe the physical properties of East Asian 116 aerosols while two dust events were detected (Nakajima et al., 2007). Here, we focus on the 117 characterization of airborne anhydrosugars, primary sugars, sugar alcohols, and BSOA 118 tracers from the KCOG. Gosan is influenced by the continental outflow from East Asia 119 during winter, spring and fall, whereas the site is influenced by the maritime air masses from 120 the Pacific Ocean and other marginal seas during summer. This makes the KCOG ideal for 121 characterizing the regional heterogeneities in the emissions of organic compounds in the East 122 Asian outflow based on the total suspended particulate (TSP) samples collected during April 123 2013-April 2014.

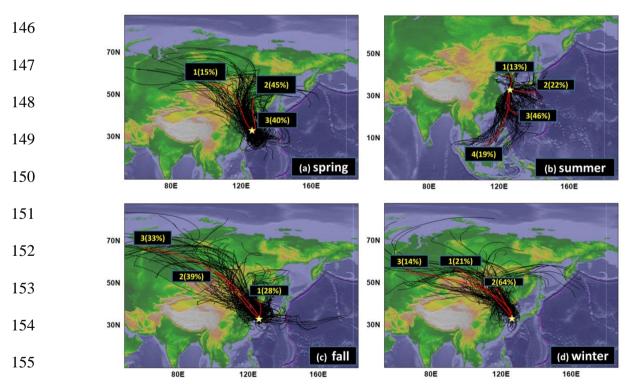
### 124 **2. Methods**

### 125 **2.1. Aerosol sampling and prevailing meteorology**

TSP samples were collected on pre-combusted (450°C for 6 h) quartz fiber filters (20 cm × 25 cm, Pallflex) at the KCOG (33.17 °N, 126.10 °E, see Figure 1), South Korea. To get enough signal for the radiocarbon measurements, each TSP sample was collected for 10–14 days from April 2013 to April 2014. Twenty-one samples were collected using a high-volume air sampler (Kimoto AS-810, ~65 m<sup>3</sup> h<sup>-1</sup>) installed on the rooftop of a trailer house (~3 m above the ground). After the collection, the aerosol filters were transferred to a pre132 combusted (450°C for 6 h) glass jar (150 mL) equipped with a Teflon-lined screw cap and 133 transported to the laboratory in Sapporo. These TSP samples were stored in a dark freezer 134 room at -20°C until the analysis. Three field blank filters were also collected during the 135 campaign.

136 The ambient temperatures at the Gosan site were on average 6.9°C in winter, 14.1°C in spring, 27.0°C in summer, and 17.1°C in fall. Likewise, the average relative humidity was 137 138 found to be highest in summer (71.3%), followed by spring (64.9%), fall (63.5%), and winter 139 (54.7%). Gosan is influenced by the pollution sources in East Asia during winter as well as 140 other transition periods (spring and fall) due to the prevailing westerlies. In contrast, winds in 141 summer blew mostly from the western North Pacific (WNP) by the easterly winds. The 142 spring season is, in particular, important for the transport of mineral dust mixed with polluted 143 OA to Gosan (Kundu et al., 2010).

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- 145



**Figure 1.** Cluster analysis of backward air mass trajectories over Gosan (indicated by a star symbol) for the TSP collected during (a) spring, (b) summer, (c) fall, and (d) winter seasons.

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## 159 2.2. Extraction and analysis of organic compounds

160 Approximately  $3.14 \text{ cm}^2$  filter cuts were extracted with dichloromethane/methanol (2:1; v/v).

161 The extracts were concentrated using a rotary evaporator under vacuum and then blown down

162 to near dryness with pure nitrogen gas. The dried residues were subsequently reacted with N, O-bis(trimethylsilyl)trifluoroacetamide containing 1% trimethylchlorosilane (BSTFA+1% 163 TMCS, SUPELCO<sup>©</sup>, Sigma-Aldrich<sup>®</sup>) and pyridine at 70 °C for 3 h to derive OH and COOH 164 groups of polar organic compounds to trimethylsilyl ethers and esters, respectively. After the 165 166 derivatization followed by the addition of a known amount of internal standard solution (Tridecane: 1.43 ng L<sup>-1</sup> in n-hexane), the derivatized extracts were injected onto a gas 167 chromatograph (Hewlett-Packard model 6890 GC) coupled to a mass spectrometer (Hewlett-168 169 Packard model 5973, MSD) (GC-MS). More details on the quantification of polar organic 170 compounds using GC-MS are described in Haque et al. (2019).

171 The target compounds (anhydrosugars, primary sugars, sugar alcohols, and BSOA 172 tracers) were separated on a DB-5MS fused silica capillary column (30 m x 0.25 mm i.d., 0.5 µm film thickness) using helium as a carrier gas at a flow rate of 1.0 ml min<sup>-1</sup>. The GC oven 173 174 temperature was programmed from 50°C for 2 min and then increased from 50 to 120°C at 30°C min<sup>-1</sup> and to 300°C at 6°C min<sup>-1</sup> with a final isotherm hold at 300°C for 16 min. The 175 sample was injected in a splitless mode with the injector temperature at 280°C. The MS was 176 177 operated at 70 eV and scanned from 50 to 650 Da on an electron impact (EI) mode. Mass 178 spectral data were acquired and processed using the Chemstation software. The organic 179 compounds were identified individually by comparison with retention times and mass spectra 180 of authentic standards and NIST library and literature data of mass fragmentation patterns 181 (Medeiros and Simoneit, 2007). For assessing the recoveries, ~100-200 ng of the standard 182 solution was spiked on the blank filter and analyzed as a real sample. Overall, the average 183 recoveries were found to be 80-104% for target compounds. The field and laboratory blank 184 filters (n = 3) were also analyzed by the same procedures as a real sample. Target compounds 185 were not found in the field blanks. The analytical errors based on concentrations by replicate 186 sample analyses (n = 3) were less than 15%.

### 187 **2.3. Carbon fractions analysis**

188 Organic carbon (OC) and elemental carbon (EC) were analyzed using a thermal-optical 189 transmittance method with a Sunset Laboratory carbon analyzer following the NIOSH 190 protocol (Birch and Cary, 1996), and detailed procedures were given elsewhere (Zhang et al., 191 2016). Furthermore, a portion of 2.54 cm<sup>2</sup> of each sample filter was extracted with 15 mL 192 ultrapure water (resistivity > 18.2 M $\Omega$ cm, Sartorius arium 611 UV) with ultrasonication for 193 30 min. The water extracts were then filtered through a membrane disc filter for water-194 soluble organic carbon (WSOC) analysis by a total organic carbon (TOC) analyzer 195 (Shimadzu, TOC-Vcsh) (Boreddy et al., 2018). The concentrations of WSOC were corrected

by field blanks. The analytical errors in the triplicate analyses were less than 5% for WSOC.

# 197 **2.4. Radiocarbon isotopic composition of total carbon (TC) and EC**

198 The concentrations of TC in the TSP samples were determined using an elemental analyzer. For the radiocarbon isotopic composition ( $\Delta^{14}$ C), the aerosol filter punches (1.5 cm<sup>2</sup>) were 199 exposed for ~12 h to HCl fumes in a vacuum desiccator. Subsequently, the punches were 200 analyzed for  $\Delta^{14}$ C on a modified elemental analyzer coupled via a gas interface to Accelerator 201 Mass Spectrometer Mini Carbon Dating System (MICADAS) at the University of Bern. 202 203 Switzerland (Salazar et al., 2015). The evolved CO<sub>2</sub> of TC from the elemental analyzer was 204 passed through a moisture trap (Sicapent, Merck) and isolated from other residual gasses 205 using a temperature-controlled zeolite trap. The purified CO<sub>2</sub> was introduced through a gas interface system to MICADAS, where  ${}^{14}C/{}^{12}C$  ratios are measured according to the analytical 206 procedures detailed in Zhang et al. (2016). Likewise, the evolved CO<sub>2</sub> of elemental carbon 207 from the Sunset Lab OC/EC analyzer using the Swiss 4S protocol (Zhang et al., 2012), was 208 directed to the MICADAS and measured for the <sup>14</sup>C/<sup>12</sup>C ratio relative to standard calibration 209 gas. These results were expressed as fractions of modern carbon  $(f_M)$  by normalizing with a 210 211  $\delta^{13}$ C value of the reference standard in the year 1950 (-25‰) according to Stuiver and Polach 212 (1997) for the fractionation effects. The  $f_{\rm M}(\rm OC)$  can be estimated by using the  $f_{\rm M}(\rm TC)$  and  $f_{\rm M}({\rm EC})$  in an isotope mass balance equation (Zhang et al., 2015). Additionally, we estimated 213 214 the relative contributions of OC and EC from the nonfossil and fossil sources ( $f_{nonfossil}$  and 215  $f_{\text{fossil}}$ , respectively) using the following equations.

216	$f_{\text{nonfossil-OC}} = f_{\text{M}}(\text{OC-sample})/f_{\text{M}}(\text{OC-ref}); f_{\text{M}}(\text{OC-ref}) \approx 1.07 \pm 0.04$	(1)
217	$f_{\text{nonfossil-EC}} = f_{\text{M}}(\text{EC-sample})/f_{\text{M}}(\text{EC-ref}); f_{\text{M}}(\text{EC-ref}) \approx 1.10 \pm 0.05$	(2)
218	$f_{\text{fossil-OC}} = 1 - f_{\text{nonfossil-OC}}$	(3)

 $219 f_{\text{fossil-EC}} = 1 - f_{\text{nonfossil-EC}} (4)$ 

The reference values of OC and EC were obtained from Mohn et al. (2008). Using the fractions of  $f_{\text{fossil-OC}}$  and  $f_{\text{nonfossil-OC}}$ , we can, therefore, estimate the mass concentration of ambient organic carbon (OC-ambient) from fossil and nonfossil sources (OC<sub>fossil</sub> and OC<sub>nonfossil</sub>, respectively).

224 
$$OC_{nonfossil} = f_{nonfossil-OC} \times [OC]_{ambient}$$
 (5)

225 
$$OC_{fossil} = f_{fossil-OC} \times [OC]_{ambient}$$
 (6)

226 More details on the radiocarbon isotopic composition data over Gosan were reported 227 elsewhere (Zhang et al., 2016).

#### 228 **3. Results and discussion**

### 229 **3.1. Trajectory and cluster analysis**

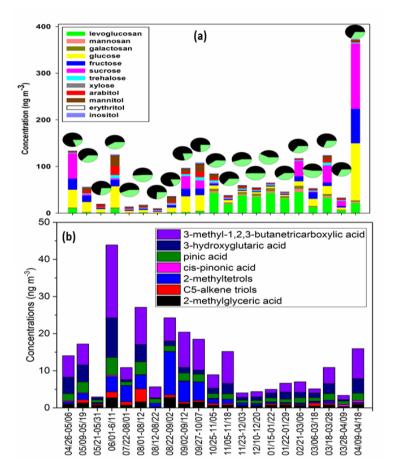
230 Backward air mass trajectories are useful for assessing the impact of local versus regional 231 source emissions over Gosan. Seven-day isentropic backward air mass trajectories were 232 computed using the hybrid single-particle Lagrangian integrated trajectory model (HYSPLIT, 233 version 4: Stein et al., 2015) over the KCOG for the sampling period using the 234 meteorological datasets of the Global Data Assimilation System (GDAS) network. The 235 trajectory endpoint files from the HYSPLIT model were further used for the cluster analysis 236 using the Trajstat package (Wang et al., 2009b) for all four seasons (Figure 1). Although 237 cluster analysis revealed the predominance of continental transport in the spring, fall, and 238 winter seasons, the air masses over the KCOG in summer mostly originated from the WNP. 239 Since spring is a transition of winds switching from westerlies to easterlies, Gosan is likely 240 influenced by the long-range transport of dust, pollution, and sea-salt aerosols.

241 The vertical mixing of pollutants within the boundary layer height also plays an 242 important role in controlling the strength of continental outflow alongside regional 243 meteorology. For instance, the mixing height of air parcels from the HYSPLIT model is 244 mostly confined to 1000 m in winter but somewhat increased towards the spring and fall 245 seasons (Figure S1). This vertical enhancement in the boundary layer height facilitates the 246 transport of mineral dust particles from the arid and semiarid regions in East Asia along with 247 urban pollutants to Gosan in spring and fall compared to winter. However, the strength of the 248 continental outflow somewhat depends on several factors, including source emissions, 249 meteorology, and mixing height of air parcels.

250 Gosan is influenced by three types of air masses in spring (Figure 1a), from the 251 Mongolian Desert (cluster 1: 15%), North China (cluster 2: 45%), and from the Yellow Sea 252 (cluster 3: 40%). In contrast, the easterlies from the WNP in summer mostly influenced the 253 composition of TSP over the KCOG. This inference is based on the cluster analysis for 254 summer samples (Figure 1b), which showed four regimes, including transport from the Sea of Japan (cluster 1: 13%), WNP (cluster 2: 22%), South China Sea (cluster 3: 46%), and East 255 256 China Sea (cluster 4: 19%). In contrast, cluster analysis revealed three major transport 257 regimes from East Asia in fall and winter (Figure 1c-d). However, there are subtle differences 258 that exist between winter and fall in terms of influence from nearby versus distant pollution 259 sources. For instance, long-range transport of air masses from west Mongolia (cluster 1: 260 21%) and the Russian Far East (cluster 3: 14%) exerted a weak influence on the TSP sampled 261 over Gosan in winter. Besides, we observed a somewhat larger impact of air masses from the

North China Plain over Gosan (cluster 2: 64%) in winter. In contrast, Gosan is less influenced
by air masses originating from the North China Plain, contributing ~28% (cluster 1) than
those from Mongolia (cluster 2: 39%) and the Russian Far East (cluster 3: 33%) in fall.
Therefore, the impact of East Asian outflow is stronger in winter than in spring and fall.

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Figure 2. (a) Cumulative concentration levels of anhydrosugars, primary sugars, and sugar
 alcohols (i.e., represented by bars), and depicting the contributions of nonfossil (green color)
 and fossil (black color) organic carbon (i.e., pie charts), (b) Cumulative concentration levels
 of isoprene- and monoterpene-SOA tracers in each TSP sample collected over Gosan.

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# 274 **3.2. Temporal and seasonal variability of sugars**

275 The temporal/seasonal trends of sugar compounds over the KCOG provide useful 276 information on the emission strengths of various sources in the East Asian outflow. All three 277 anhydrosugars showed similar temporal and seasonal trends with higher concentrations in 278 winter and fall than spring and summer (Figures 2a and S2). As levoglucosan and two other 279 anhydrosugars (mannosan and galactosan) the pyrolysis products of are

280 cellulose/hemicellulose, their higher concentrations along with an increase in nonfossil fraction of OC (Figure 2a; pie charts) in TSP from winter and fall revealed the impact of BB 281 282 emissions. The MODIS satellite-based fire counts (Figure S1) together with cluster analysis 283 in winter and fall (Figure 1) have revealed an influence of active BB emissions in the North 284 China Plain, Mongolia, and the Russian Far East. The temporal trends of glucose, fructose, 285 and sucrose exhibited less variability throughout the sampling period; however, we observed 286 a slight increase in their concentration towards spring/summer (Figure 2a). Glucose and 287 fructose have origins from leaf fragments and pollen species (Fu et al., 2012a). Sucrose is a 288 potential tracer for airborne pollen (Fu et al., 2012a) and late spring/early summer is often 289 regarded as a season of "pollen-allergies". Therefore, the similar temporal trends of glucose 290 and fructose with sucrose indicate their common source, pollens (Figure 2a). Since glucose, fructose, and sucrose showed moderately significant correlations ( $R^2 = 0.44-0.48$ , p<0.01) 291 292 with levoglucosan in winter, it is somewhat possible that BB source emission could also 293 influence the concentrations of these saccharides in this season (Haque et al., 2019; Fu et al., 294 2008).

295 BB also contributes to xylose and, hence, the temporal variability of xylose is 296 mimicking that of the anhydrosugars. Trehalose is a primary sugar and a useful tracer for 297 organic carbon associated with soil dust particles (Fu et al., 2012a). The temporal variability 298 of trehalose closely resembles that of the fungal spore tracers (arabitol, mannitol, and 299 erythritol), showing high concentrations in the spring, summer, and fall seasons (Figures 2a 300 and S2) (Zhu et al., 2015a; Fu et al., 2012a). The KCOG is under the influence of a large-301 scale advection of mineral dust from East Asia to the WNP during these three seasons (Tyagi 302 et al., 2017; Huebert et al., 2003). The mineral dust transport from East Asia to the WNP can be traced by the high concentrations of non-sea-salt Ca<sup>2+</sup> in the TSP samples from Gosan 303 (Arimoto et al., 1996). Similar temporal trends of trehalose and nss-Ca<sup>2+</sup>, particularly in 304 spring samples (Figure S3), suggest that the abundance of OA specific to fungal spores over 305 306 Gosan is likely associated with the Kosa (Asian dust) events.

The major sources of arabitol and mannitol are airborne fungal spores (Bauer et al., 2008), accompanying detritus from mature leaves (Pashynska et al., 2002). Heald and Spracklen (2009) reported that mannitol and arabitol are considerably associated with terrestrial biosphere activity. Inositol is largely derived from the developing leaves in summer (Pashynska et al., 2002) and BB in winter (Fu et al., 2010b). Zhu et al. (2015b) found a similar seasonal behavior of inositol with those of other sugar alcohols with the predominance in summer, associated with microbial activities in local forests from Okinawa. Inositol showed a moderately significant correlation with levoglucosan ( $R^2 = 0.33$ , p< 0.01) in winter; however, there were no positive linear relationships between levoglucosan and other sugar alcohols, implying a partial emission of inositol from the BB during winter in Gosan aerosols. Therefore, the temporal variability of inositol differs from that of the other sugar alcohols (Figure S2). The sources of sugar compounds are further discussed in section 3.4.

320 The seasonally averaged concentrations of all the anhydrosugars and xylose are 321 higher in winter/fall than spring/summer (Figure 3a-d), possibly due to a greater influence of 322 long-range transport from East Asia. In contrast, glucose, fructose, and sucrose peaked in 323 spring but decreased in the other seasons (Figure 3e-g), mainly because of the contribution of 324 airborne pollen. Trehalose showed higher concentrations in fall and summer, followed by 325 spring and winter (Figure 3h). Arabitol, mannitol, and erythritol showed higher 326 concentrations in summer/fall than in winter and spring (Figure 3i-k). This seasonal trend is 327 consistent with those of soil-derived fungal spores. This feature is consistent with earlier 328 observations from a remote oceanic island in the WNP (Okinawa) during the impact of East 329 Asian outflow (Zhu et al., 2015a). The seasonally averaged mass concentrations of inositol 330 are highest in spring, followed by summer, fall, and winter (Figure 31). Overall, the molecular 331 compositions of anhydrosugars showed the predominance of levoglucosan followed by 332 galactosan and mannosan (Figure S4). Galactosan is more abundant in crop-residue burning 333 emissions than mannosan (Engling et al., 2009; Sheesley et al., 2003). It is very much likely 334 that the impact of crop-residue burning emissions in East Asia over Gosan is more prominent 335 in winter/spring. Such high abundances of galactosan over mannosan were found in the North China Plain (Fu et al., 2008) and in the Indo-Gangetic Plain outflow sampled over the Bay of 336 337 Bengal (Bikkina et al., 2019). Although the temporal variability of primary sugars in the TSP 338 samples from Gosan showed a characteristic peak of glucose and sucrose (Figure 2a), the 339 seasonally averaged distributions are different (Figure S4). The molecular distributions of 340 sugar alcohols are characterized by high loadings of arabitol and mannitol, followed by 341 erythritol and inositol (Figure S4).

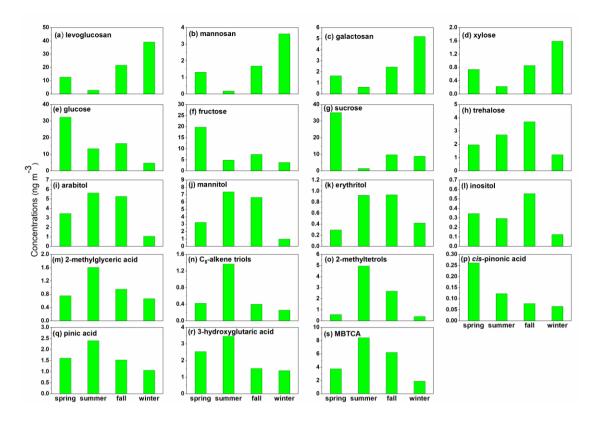


Figure 3. Seasonal variability of the atmospheric levels of sugar compounds and BSOA
 tracers in TSP samples from Gosan during April 2013-April 2014.

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## 347 **3.3. Temporal and seasonal variability of BSOA tracers**

We identified six isoprene-SOA tracers such as 2-methylglyceric acid (2-MGA), three  $C_5$ -348 349 alkene triols, and two 2-methyltetrols (2-methylthreitol and 2-methylerythritol) (2-MTs) in Gosan aerosol samples (Table 1). The sum of the isoprene-SOA tracers ranged from 0.35 to 350 15.1 ng m<sup>-3</sup> (avg. 3.69 ng m<sup>-3</sup>) with the predominance of 2-MTs (avg. 2.09 ng m<sup>-3</sup>). 2-MGA is 351 the second most abundant isoprene-SOA tracer (avg. 0.99 ng m<sup>-3</sup>), a high-generation product 352 353 probably formed by further photooxidation of methacrolein and methacrylic acid. A similar 354 molecular composition was observed over the North Pacific and California Coast (Fu et al., 355 2011). All the isoprene-SOA tracers exhibited similar temporal variations with higher 356 concentrations in the summer/spring months compared to autumn and winter (Figure 2b). Conversely, four monoterpene-SOA tracers, i.e., cis-pinonic acid, pinic acid, 3-357 hydroxyglutaric acid (3-HGA), and 3-methyl-1,2,3-butanetricarboxylic acid (MBTCA), were 358 359 detected in the Gosan samples (Table 1). The total concentrations of monoterpene-SOA tracers were found to be 1.65 to 35.5 ng m<sup>-3</sup> (avg. 9.24 ng m<sup>-3</sup>) with a high concentration of 360 MBTCA (avg. 5.11 ng  $m^{-3}$ ). All the monoterpene-SOA tracers showed similar temporal 361 trends with high values in summer/spring periods than autumn/winter (Figure 2b). 362

Nevertheless, *cis*-pinonic acid exhibited a somewhat different temporal variability than the other monoterpene-SOA tracers. It is likely that *cis*-pinonic acid might be further photooxidized to form MBTCA (Szmigielski et al., 2007).

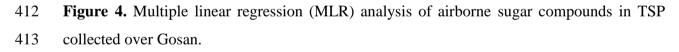
366 The seasonal average of isoprene-SOA tracers showed high concentrations in summer, followed by spring/fall and winter (Figure 3m-o). One key feature of the data 367 368 presented here is that two fall samples (KOS984; 2-12 September and KOS986; 27 369 September to 07 October) exhibited high concentrations for 2-MTs over Gosan (Figure 2b). 370 We presumed that local vegetation might contribute significantly to the formation of 2-MTs 371 as they are first-generation products. Moreover, 2-MTs can be derived from the open ocean 372 under low NO<sub>x</sub> conditions (Hu et al., 2013). 3-HGA and pinic acid showed somewhat higher 373 concentrations in summer/spring than fall/winter due to the growing vegetation (Figure 3q-r). 374 Cis-pinonic acid was more abundant in spring compared to summer (Figure 3p) because of its photo-degradation, as discussed above. In contrast, MBTCA was more dominant in 375 376 summer/fall than in spring/winter (Figure 3s). Here, the formation of MBTCA could be 377 enhanced in fall during atmospheric transport from East Asia. The molecular distributions of 378 isoprene-SOA tracers were characterized by a high loading of 2-MTs, followed by 2-MGA 379 and C<sub>5</sub>-alkene triols in all seasons (Figure S4). The molecular composition of monoterpene-380 SOA tracers was dominated by MBTCA, followed by 3-HGA, pinic acid, and cis-pinonic 381 acid in all seasons (Figure S4). Overall, BSOA tracers were found to be most abundant in 382 summer, followed by fall, spring, and winter (Table 1). Interestingly, it is likely that 383 secondary OA undergoes much faster cycling than the primary sugar compounds, considering 384 the feasibility of photooxidation. This would mean a slight underestimation of BSOA over 385 the KCOG in the East Asian outflow and, hence, their atmospheric abundances over Gosan 386 reflect a lower limit.

387 Kang et al. (2018b) reported that monoterpene-SOA tracers were more abundant than 388 isoprene-SOA tracers in spring-summer over the East China Sea, which is consistent with this 389 study. Although the mass budget calculations showed that isoprene and monoterpenes are 390 largely emitted by terrestrial plants, the open ocean can also contribute to isoprene and 391 monoterpenes significantly (Conte et al., 2020; Shaw et al., 2010; Broadgate et al., 1997). Air 392 mass back trajectory and cluster analysis (Figure 1) implied that the air masses mostly 393 originated from the ocean during summer. This means the open ocean significantly 394 contributed to isoprene-SOA production. However, terrestrial sources from the continent also 395 substantially enhanced the formation of BSOA. For example, one sample (KOS979; 1-11

June 2013) during summer showed the highest loading of BSOA tracers (Figure 2b) when air
masses were transported from the continent (Zhang et al., 2016).

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399	Lev	6	0 60 120		0 60 140	, ,	1 3 5 7		0 10 20	
400	φ = 0.9 Ma	an								
401	0.98 0.9	91 Gal								3 8
402	0.1	02 -0.09	Gluc	<u>}</u>		21.2				
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404	8 0.09 0.1 6 0.09	0.09	0.89	0.94	Suc	<u>.</u>				<u></u>
405 406	0.96 0.3	88 0.95	-0.01	0.04	0.19	Xyl				00 10 20
407	φ -0.05 -0.	03 -0.05	0.25	0.08	0.02	-0.07	Treha			
408	-0.22 -0	.2 -0.19	0.26	0.06	-0.08	-0.21	0.86	Arab	•	
409	<sup>8</sup> = −0.25 −0.	22 -0.22	0.3	0.11	-0.03	-0.26	0.91	0.95	Manit	
410	0 -0.	11 -0.01	-0.07	-0.21	-0.3	-0.09	0.59	0.57	0.59	Erythr
411	<sup>♀</sup> -0.24 -0. <sup>♀</sup> -0.24 -0.	18 -0.22 0 3 6	0.59	0.45	0.33	-0.23	0.76	0.72	0.82	0.34 Inos



## 415 **3.4. Source apportionment- regression analysis and diagnostic ratios**

416 The anhydrosugars strongly correlated with xylose (Figure 4), suggesting their common 417 source as BB emission in East Asia. Fu et al. (2012a) analyzed pollen from different tree species (e.g., White birch, Chinese willow, Peking willow, Sugi, Hinoki), which are enriched 418 in sucrose (182-37,300  $\mu$ g g<sup>-1</sup>), glucose (378-3601  $\mu$ g g<sup>-1</sup>) and fructose (162-1813  $\mu$ g g<sup>-1</sup>). In 419 420 our samples, sucrose strongly correlated with glucose and fructose (Figure 4), suggesting 421 their origin from plant-derived airborne pollen. Likewise, a strong correlation was found 422 between arabitol and mannitol, indicating a mutual origin from a similar type of fungal spores 423 (Zhu et al., 2015a; Fu et al., 2012a; Bauer et al., 2008; Yttri et al., 2007). Bauer et al. (2008) ascribed weak correlations between arabitol and mannitol to the diverse nature of ambient fungal spores. Furthermore, both sugar alcohols correlated well with trehalose, a tracer for soil organic carbon (Fu et al., 2012a). This observation suggests their common origin from soil organic matter associated with fungal spores. Erythritol also originates from fungal spores; however, its abundance is affected by the atmospheric aging process. Kessler et al.

spores; however, its abundance is affected by the atmospheric aging process. Kessler et al.
(2010) reported that erythritol is degraded during long-range transport in 12.7 days.
Consequently, arabitol and mannitol were moderately correlated with erythritol in the Gosan
samples due to the degradation of the latter sugar alcohol in the East Asian outflow.

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432 The linear relationship of levoglucosan (Lev) with mannosan (Man), galactosan (Gal), 433 and  $nss-K^+$  provides useful information on the type of BB emissions (hardwood, softwood, or 434 crop-residue). Ratios of Lev/Man and Lev/(Man + Gal) can be useful to distinguish BB and 435 coal combustion contributions. The average ratios of Lev/Man (15.1  $\pm$  6.76) and Lev/(Man + 436 Gal) (4.27  $\pm$  1.23) in Gosan aerosols are much closer to those from wood burning and coal 437 combustion sources emissions, respectively (Yan et al., 2018). It reveals that Lev could 438 originate from both biomass and coal burning source emissions, which is consistent with the 439 linear relationship between Lev-C and the fossil-/nonfossil carbon fraction (section 3.6). 440 Furthermore, different types of biomass are characterized by distinct Lev/Man ratios. For 441 instance, Lev/Man ratios from the softwood burning emissions (3-10) differ from those of 442 hardwood (15-25) and crop-residues (>40) (Singh et al., 2017; Schmidl et al., 2008a, b; Fu et 443 al., 2008; Engling et al., 2006, 2009; Fine et al., 2001, 2004). We found that the Lev/Man 444 ratios (Table 2) over the KCOG overlap between seasons and are somewhat close to that of 445 hardwood burning emissions in northern China, Mongolia, and the Russian Far East, as 446 corroborated by the backward air mass trajectories and MODIS fire counts (Figures 1 and 447 S1). Besides,  $Lev/K^+$  and *Man/Gal* ratios in summer differ from those of other seasons (Table 448 2). Cheng et al. (2013) have apportioned qualitatively the source contributions of 449 anhydrosugars over a receptor site based on the comparison of Lev/K<sup>+</sup> and Lev/Man ratios in 450 aerosols to those from various source profiles compiled from the literature. This approach of 451 using the mass ratios of Lev/Man and  $Lev/K^+$  has been proven useful for deciphering the 452 difference in BB-derived OA (Bikkina et al., 2019).

Here, we adopted this methodology to ascertain the likely contributing sources of anhydrosugars, which are BB tracers from different seasons (Figure 5). This source apportionment relies on the fact that the  $Lev/K^+$  ratio from the softwood burning (10-1000) is higher than from hardwood (1-100) (Fine et al., 2004). In contrast, Lev/Man ratios for softwood are lower than those of hardwood burning (10-100) (Fine et al., 2004). Likewise,

the Lev/Man and Lev/K<sup>+</sup> ratios from grasses and crop-residues are 10-100 and 0.01-1.0, 458 respectively (Bikkina et al., 2019). On a similar note, the  $Lev/K^+$  ratios from the burning of 459 460 pine needles (0.1-1.0) somewhat overlap with those from hardwood burning emissions but 461 are characterized by distinct Lev/Man ratios (Bikkina et al., 2019). The burning of dead 462 leaves (duff) showed higher  $Lev/K^+$  ratios than those of pine needles and grasses, but their 463 Lev/Man ratio is on the lower side than for the former biomass type and the softwood burning 464 emissions. However, Lev is more susceptible to degradation by photooxidation with OH radicals during atmospheric transport (half-life: <2.2 days) (Hennigan et al., 2010) and this 465 466 would cause lower abundances of this anhydrosugar. Hence, the hardwood  $Lev/K^+$  ratios could slightly shift downwards. Therefore, caution is required while interpreting the ambient 467 data from a receptor site (Bikkina et al., 2019). Overlapping the seasonal data on this scatter 468 469 plot of *Lev*/K<sup>+</sup> versus *Lev/Man* (Figure 5) clearly reveals a mixed contribution of burning of 470 hardwood and crop-residue in the East Asian outflow. It should be noted that the 471 photooxidation process during atmospheric transport is also applicable for low concentrations and poor correlations of other primary saccharides. 472



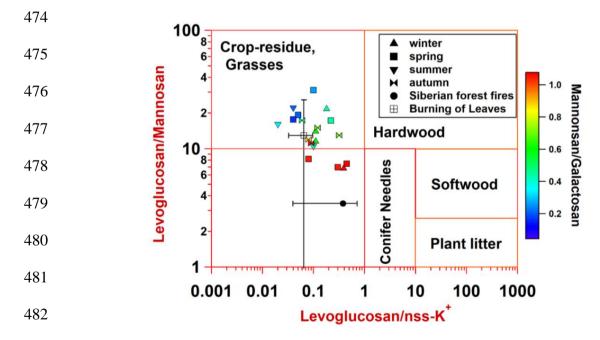
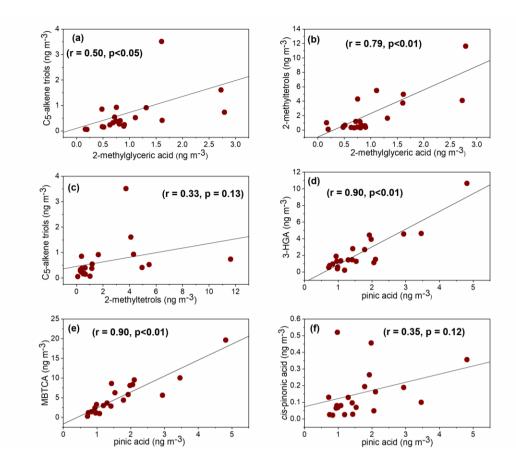


Figure 5. Scatter plot of levoglucosan/K<sup>+</sup> versus levoglucosan/mannosan ratios in TSP
collected over Gosan during April 2013-April 2014. The data for Siberian forest fires and
burning leaves were adopted from Sullivan et al. (2008).

487 The correlation coefficients and diagnostic ratios of BSOA tracers specify their source origin or formation pathway. Nevertheless, atmospheric stability or reactivity of BSOA 488 489 tracers through the photooxidation during long-range transport may bias the correlation 490 coefficients in Gosan aerosols. 2-MGA showed a significant correlation with 2-MTs (r =491 0.79, p<0.01) and C<sub>5</sub>-alkene triols (r = 0.50, p<0.05) (Figure 6a-b), suggesting their similar 492 formation pathway or common sources of isoprene-SOA tracers. However, a poor correlation 493 coefficient between 2-MTs and C<sub>5</sub>-alkene triols (r = 0.33, p = 0.13) (Figure 6c) indicates their 494 different formation process over the Gosan atmosphere. Wang et al. (2005) documented that 495 polyols are formed from isoprene through diepoxy derivatives, which further convert into 2-496 MTs by acid-catalyzed hydrolysis. On the other hand, C<sub>5</sub>-alkene triols are produced from the 497 precursor of hydroxyperoxy radicals that are initially derived from isoprene through 498 rearrangement reactions (Surratt et al., 2006). It can be noted that the formation mechanisms 499 of 2-MGA and 2-MTs are different while depending on the NO<sub>x</sub> concentrations (Surratt et al., 500 2010). Thus, the ratio of 2-MGA/2-MTs attributes to the influence of NO<sub>x</sub> on isoprene-SOA 501 formation. We found a low ratio of 2-MGA/2-MTs (0.67) (Figure S5a, Table 2) in summer, 502 implying enhancement of 2-MTs formation over the open ocean due to the low-NO<sub>x</sub> 503 environment in the ocean atmosphere. On the contrary, the 2-MGA/2-MTs ratios for other 504 seasons were >1.0 (Figure S5a, Table 2), indicating an elevated formation of 2-MGA through 505 continental high NO<sub>x</sub> condition, which is consistent with the air masses back trajectory.



508 **Figure 6.** Pearson linear correlation coefficient analysis of BSOA tracers in Gosan TSP 509 aerosols during April 2013-April 2014.

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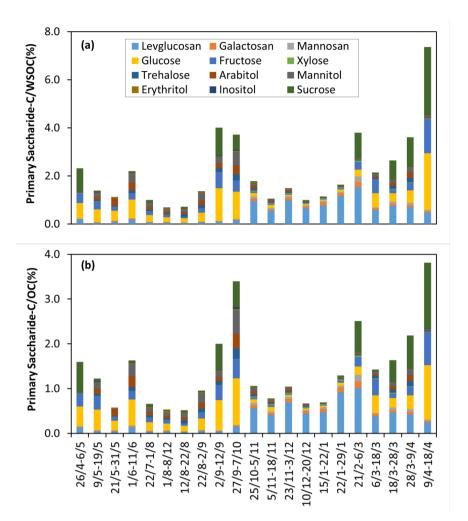
511 *Cis*-pinonic acid showed a weak correlation with pinic acid (r = 0.35, p = 0.12) 512 (Figure 6f), suggesting that different atmospheric reactivity of cis-pinonic/pinic acids during 513 transport would cause such poor correlation. In contrast, pinic acid exhibited a strong positive 514 linear correlation with 3-HGA (r = 0.90, p<0.01) and MBTCA (r 0.90, p<0.01) (Figure 6d-e), 515 implying their similar sources. It should be noted that the formation processes of pinic acid, 516 3-HGA, and MBTCA are different because pinic acid is a first-generation product, and 3-517 HGA and MBTCA are high-generation products (Claeys et al., 2007; Müller et al., 2012; 518 Szmigielski et al., 2007). The ratio of *cis*-pinonic acid + pinic acid to MBTCA (P/M) is used 519 to evaluate the aging of monoterpene-SOA.

520 A low P/M ratio suggests the transformation of *cis*-pinonic and pinic acids to 521 MBTCA and thus relatively aged monoterpene-SOA, whereas a high ratio reflects relatively 522 fresh monoterpene-SOA (Gómez-González et al., 2012; Ding et al., 2014). Gómez-González 523 et al. (2012) reported aged monoterpene-SOA (P/M = 0.84) from a Belgian forest site while 524 fresh chamber-produced  $\alpha$ -pinene-SOA tracers showed P/M ratios of 1.51 to 3.21 (Offenberg et al., 2007). The average ratio of P/M in this study was 0.62 with the low value in summer (Figure S5b, Table 2), which is lower than those of Guangzhou (fresh monoterpene-SOA; 28.9) while the air masses originated from southern China (Ding et al., 2014). This observation indicates that monoterpenes SOA have undergone substantial aging during transport to Gosan, particularly in summer when extensive photochemical oxidation occurred due to the high temperature and intense solar radiation.

### 531 **3.5. Relative abundances in WSOC and OC**

532 Levoglucosan is the most abundant anhydrosugar, contributing 0.05-1.54% of WSOC and 533 0.03-1.02% of OC. Likewise, sucrose, glucose, and fructose were more abundant primary 534 sugars, contributing 0.01-2.83%, 0.03-1.22%, and 0.02-0.74% of WSOC, respectively. 535 Contributions of the three primary sugars varied from 0.05% to 3.41% of OC. Arabitol and 536 mannitol are the most abundant sugar alcohols, whose contribution to WSOC and OC ranged 537 from 0.02% to 0.93% and 0.01% to 0.85%, respectively. Figure 7 depicts the contribution of 538 sugar compounds to WSOC and OC in TSP collected over Gosan during the study period. 539 We also compared the atmospheric abundances of sugar compounds from Gosan with the 540 literature data (Table 3). This comparison revealed the less influence of BB tracer compounds 541 (i.e., anhydrosugar levels) over Gosan, a factor of 5-10 times lower than those reported for 542 the BB-influenced source regions in China and East Asia (Fu et al., 2012b; Kang et al., 543 2018a; Wang and Kawamura, 2005; Wang et al., 2012). However, the levels of anhydrosugar 544 over Gosan are higher than those observed over the remote Canadian High Arctic site (Fu et 545 al., 2009a). In contrast, Gosan is characterized by high concentrations of primary sugars 546 compared to other remote sampling sites in Table 3. This is because of the overwhelming 547 contribution of primary sugars associated with soil dust particles over Gosan during the East 548 Asian outflow. Such high loadings of primary sugars were observed from other remote island 549 receptor sites in the WNP (Okinawa) during the spring season (Zhu et al., 2015a). Likewise, 550 the concentrations of sugar alcohols from Gosan are similar to those from other receptor sites 551 influenced by the East Asian outflow (Verma et al., 2018; Zhu et al., 2015a).

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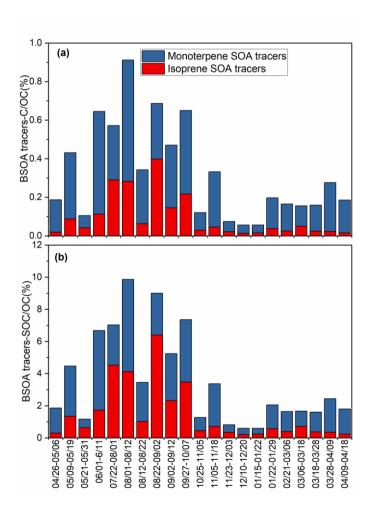


**Figure 7.** Contribution of primary saccharide-C in (a) WSOC(%) and (b) OC(%) in TSP collected over Gosan during April 2013-April 2014.

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560 The contributions of isoprene-SOA tracers to ambient OC (0.01-0.40%, avg. 0.09%) 561 and WSOC (0.02-0.57%, 0.13%) were lower than those of monoterpene-derived SOA (0.04-562 0.63%, 0.23% for OC and 0.06-0.82%, 0.33% for WSOC). The contributions of isoprene 563 oxidation products to OC and WSOC were found to be highest in summer (0.23% and 0.32%, 564 respectively), followed by fall (0.09% and 0.13%), spring (0.04% and 0.06%), and winter 565 (0.02% and 0.03%). Likewise, the contributions of monoterpene-SOA products to aerosol OC 566 and WSOC exhibited the highest value in summer (0.40% and 0.55%, respectively), followed 567 by fall (0.24% and 0.35%), spring (0.18% and 0.27%), and winter (0.10% and 0.14%). We 568 found that the contribution of BSOA products to the carbonaceous components was twice in 569 summer (Figure 8, Table 2). This means BSOA formation occurred in summer to a greater 570 extent due to the intensive BVOCs emission with key factors of meteorological parameters 571 (higher temperature and radiation), i.e., higher concentrations of ozone, and other oxidizing 572 agents (NO<sub>x</sub>, OH, etc.). It should be pointed out that the fraction of WSOC in OC (WSOC/OC) is often prone to photochemical aging and, hence, contributes to SOA. More 573 574 specifically, the WSOC/OC > 0.4 over a receptor site indicates aged aerosols with a 575 significant SOA contribution (Haque et al., 2019; Boreddy et al., 2018). In our study, the 576 average WSOC/OC ratio observed was 0.68 with the highest ratio in summer (0.72), although 577 a subtle difference existed in winter/autumn (> 0.68), implying the presence of aged OA over 578 the KCOG. Such higher abundances of SOA over Gosan during transport from East Asia is 579 mainly due to photochemical aging of anthropogenic (fossil fuel/biomass combustion) 580 emissions. Huang et al. (2014) reported significant SOA formation from the fossil 581 fuel/biomass combustion precursor VOCs in winter over China.

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Figure 8. Contribution of (a) isoprene- and monoterpene-SOA tracers-C in ambient OC(%)
and (b) isoprene- and monoterpene-SOA tracers-SOC in ambient OC(%) in Gosan aerosol
samples during April 2013-April 2014.

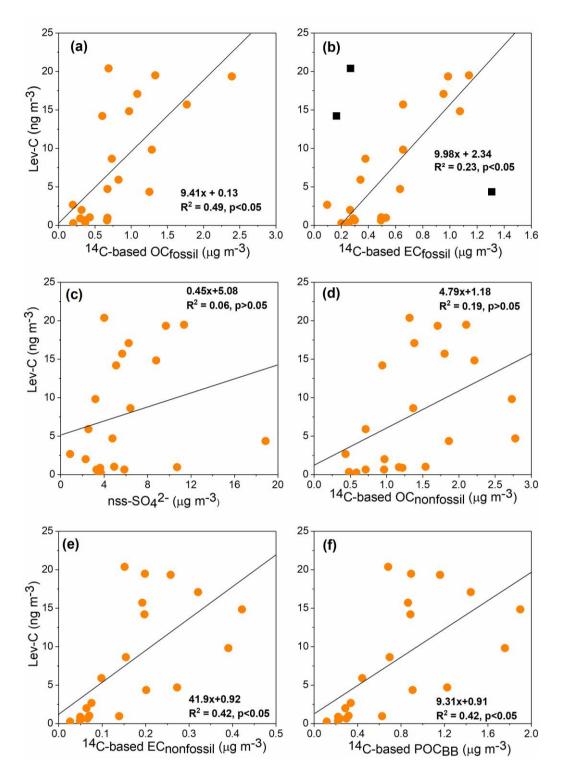
588 We estimated secondary organic carbon (SOC) derived from isoprene and monoterpenes, using the measured values of BSOA tracers and following the SOA tracer-589 590 based method first proposed by Kleindienst et al. (2007). A summary of the estimated SOC is 591 provided in Table 2. The contribution of isoprene to SOC was calculated 2.26 to 97.4 ngC m<sup>-</sup> <sup>3</sup> (avg. 23.7 ngC m<sup>-3</sup>), accounting for 1.45% of OC and 35.5% of total BSOC with the 592 predominance in summer (3.56% and 46.6% for OC and SOC, respectively). The estimation 593 of monoterpene-SOA to SOC (avg. 40.1 ngC  $m^{-3}$ ) was observed around two times higher than 594 595 that of isoprene-SOA (Table 2). Interestingly, the contribution of monoterpene-derived SOC 596 to ambient OC (3.65%) was dominant in summer, but monoterpene SOC to total SOC 597 (72.1%) was most abundant in spring (Figure 8b). The seasonal distributions of biogenic 598 SOC (Table 2) imply that a substantial amount of SOC was formed from monoterpenes in 599 spring. The estimated biogenic SOC at the KCOG is almost one order of magnitude lower 600 than that from other continental sites in Chinese urban areas (e.g., Pearl River Delta: 446 ngC m<sup>-3</sup>) (Ding et al., 2012). However, the estimated biogenic SOC load from the KCOG is much 601 602 higher than that reported from a remote site in the Canadian High Arctic (Alert: 9.4 ngC m<sup>-3</sup>) 603 (Fu et al., 2009b) and comparable with that over the East China Sea (Kang et al., 2018b).

604 It should be noted that concentrations of SOA tracers cannot always provide the 605 actual contribution of the biogenic source to ambient organic aerosol mass. For example, loadings of monoterpene-SOA tracers were lower in sample KOS999 (28 March - 9 April 606 2014) (3.02 ng m<sup>-3</sup>) compared to sample KOS1000 (9 - 18 April 2014) (14.4 ng m<sup>-3</sup>), whereas 607 the estimated contribution of SOC to ambient OC showed an opposite trend (KOS999: 608 609 2.08%; KOS1000: 1.56%). This result demonstrates that the estimation of SOC is an important factor in evaluating the contribution of BSOA to organic aerosol mass. We 610 calculated biogenic OC using radiocarbon (<sup>14</sup>C) data following the method proposed by 611 Szidat et al. (2006). Biogenic OC showed a poor correlation with biogenic SOC (r = 0.36, p =612 613 0.09) but a significant linear relationship with primary sugars (i.e., glucose, fructose and 614 sucrose) (r = 0.54, p<0.5), suggesting that primary bioaerosols from plant-derived airborne 615 pollen dictate biogenic OC over Gosan.

## 616 **3.6. Significance of fossil fuel as a source for levoglucosan**

The ambient *Lev* levels showed a significant linear relationship with the  $OC_{fossil}$ , suggesting the fossil source contribution of this molecular marker (Figure 9a). However, such a significant correlation was not evident between  $OC_{fossil}$  and other major sugar compounds. Until recently, *Lev* has been thought to originate primarily from the hemicellulose/cellulose pyrolysis of vegetation and, hence, can be employed as a powerful tracer for biomass smoke

- particles (Fraser and Lakshmanan, 2000; Simoneit et al., 1999). Nevertheless, residential coals (e.g., lignite and bituminous coal) have been shown to contain high concentrations of '*Lev*' but also emit traces of *Man and Gal* (Kourtchev et al., 2011; Fabbri et al., 2008). Recently, Yan et al. (2018) found a significant linear relationship between the <sup>14</sup>C-based fossil fraction of WSOC and *Lev-C* in the aerosols generated from coal combustion and the ambient aerosol samples. Therefore, the prevailing linear relationship between  $OC_{fossil}$  and
- 628 *Lev-*C in the Gosan samples (Figure 9a) is likely due to a common source contribution from
- 629 coal combustion in East Asia.



**Figure 9.** Linear regression analysis between levoglucosan in terms of its carbon content (*Lev-C*) and <sup>14</sup>C-based mass concentrations of (a) organic carbon and (b) elemental carbon of fossil origin (OC<sub>fossil</sub> and EC<sub>fossil</sub>, respectively), (c) nss-SO<sub>4</sub><sup>2-</sup>, (d) nonfossil derived organic carbon (OC<sub>nonfossil</sub>), (e) nonfossil derived elemental carbon (EC<sub>nonfossil</sub>), and (f) biomass burning derived primary OC (POC<sub>BB</sub>) in TSP collected over Gosan during April 2013-April 2014. In panel (b), the squares represent three outliers (i.e., samples with rather high and low *Lev/EC* ratios; please see text for more details).

638 The slope of the linear regression between Lev-C and  $OC_{fossil}$  (0.0094; Figure 9a) is 639 higher than those documented for the coal combustion source in China (~0.004±0.007) (Yan 640 et al., 2018). Moreover, Lev-C moderately correlated with the EC<sub>fossil</sub> with the regression 641 slope (~0.01) in the Gosan samples (Figure 9b), being comparable to that observed for the 642 coal combustion in China (0.044±0.076) (Yan et al., 2018). It should be noted that excluding 643 the three outliers as shown in Figure 9b (black square), Lev-C showed a stronger correlation with  $EC_{fossil}$  ( $R^2 = 0.74$ , p < 0.05). Of these, two outliers in winter (KOS995: 22-29 January 644 2014; KOS996: 21 February - 3 March 2014) have higher Lev-C levels over that of EC<sub>fossil</sub>, 645 646 when air mass trajectories showed the impact of BB emissions in the North China Plain. In contrast, the third outlier in summer (KOS979: 1-11 June 2013) has a lower Lev-C/EC<sub>fossil</sub>, 647 while air parcels transported from nearby cities in China, Korea, and Japan, thus, have more 648 649 contribution from vehicular emissions. Overall, both regression slopes are, thus, the 650 representative nature of Lev-C/OC<sub>fossil</sub> and Lev-C/EC<sub>fossil</sub> in the East Asian outflow. Lev-C and  $nss-SO_4^{2-}$  exhibited a poor correlation (Figure 9c), although both were transported from East 651 Asia. 652

Lev-C exhibited a rather weaker correlation with  $OC_{nonfossil}$  (R<sup>2</sup> = 0.19) than with 653  $EC_{nonfossil}$  ( $R^2 = 0.42$ ) over Gosan during the study period (Figure 9d-e). This could be likely 654 655 because OC<sub>nonfossil</sub> has contributions from the BB and the secondary formation process or the primary biogenic sources. The contribution of primary OC generated from BB (POC<sub>BB</sub>) to 656 OC<sub>nonfossil</sub> was taken from Zhang et al. (2016). In their study, the <sup>14</sup>C-based EC<sub>nonfossil</sub> levels 657 were scaled by a factor to constrain the  $POC_{BB}$  (Zhang et al., 2016). Here the conversion 658 659 factor is '4.5' (range: 3-10), which is a median value representing the primary OC/EC ratio 660 from BB emissions ((POC/EC)<sub>BB</sub>).

661

 $POC_{BB} = EC_{nonfossil} \times (POC/EC)_{BB}$ 

(7)

662 Lev-C showed a somewhat improved linear correlation with POC<sub>BB</sub> than with OC<sub>nonfossil</sub> 663 (Figure 9f). It is apparent from Figure 9 that the regression slopes are comparable, indicating the contribution to Lev from both coal combustion and BB emissions over Gosan. The 664 prevailing weak linear relationship (moderate correlation) of Lev with nonfossil and fossil 665 carbon fractions is likely the result of photo-degradation of Lev during atmospheric transport. 666 667 This result would mean that the higher atmospheric abundance of Lev and its pronounced linear relationships with the nonfossil and fossil carbon fractions implies a much stronger 668 669 impact of both source emissions in East Asia during the continental outflow in winter and 670 spring.

671 Overall, we present a new finding on the contribution of coal combustion sources in East 672 Asia in controlling the atmospheric levels of Lev apart from the traditional biomass/biofuel 673 burning emissions. This is based on the prevailing linear relationship between the 674 radiocarbon based nonfossil-EC and Lev in the year-round TSP samples collected from the 675 KCOG site in Jeju Island. The Gosan supersite is the best location to understand how the 676 chemical composition of source emissions from East Asia affects the outflow regions in 677 winter and spring. Recent studies have highlighted the potential contribution of Lev from residential coal combustion in China (Yan et al., 2018), with an estimated annual emission of 678 679 ~2.2 Gg of Lev from domestic coal combustion (Wu et al., 2021). Given this background 680 information, the prevailing significant linear relationship between Lev and nonfossil-EC (p-681 value < 0.05) over the KCOG clearly emphasizes the need for reconsideration of the previous 682 assessments on the impact of BB in East Asian outflow to the WNP. Additionally, this 683 dataset is further compared with the molecular distributions and relative abundances of 684 organic tracers in the TSP samples collected over Gosan during 2001, a decade ago (Fu et al., 685 2012a). This comparison allows us to better understand the regional changes in the emission 686 sources (e.g., fugitive dust, BB, and fossil-fuel combustion) on a decadal basis.

### 687 **4. Conclusions**

688 We investigated seasonal variations of primary organic components such as anhydrosugars, 689 primary sugars, sugar alcohols and BSOA tracers (isoprene- and monoterpene-derived SOA 690 products) in ambient aerosols from Gosan, Jeju Island. Among the detected sugar 691 compounds, levoglucosan was dominant in winter/fall, whereas glucose and sucrose were 692 more abundant in spring/summer. The seasonal trends documented that BB impact is more 693 significant in winter/fall and the primary bioaerosol particles are important in spring/summer. 694 Diagnostic ratios of levoglucosan, galactosan, and mannosan reflect that emissions from BB 695 are mostly dominated by hardwood. The significant linear relationship of sucrose with 696 glucose and fructose suggests their origin from airborne pollen. On a similar note, trehalose 697 showed a significant positive correlation with arabitol, mannitol, and erythritol, implying 698 their contribution from airborne fungal spores and soil microbes over the KCOG.

Distributions of biogenic SOA tracers were characterized by a predominance of monoterpene- than isoprene-derived oxidation products in Gosan aerosols. The BSOA tracers were formed in summer to a greater extent, followed by fall/spring and then winter. The low ratio of *cis*-pinonic acid + pinic acid to MBTCA demonstrated that monoterpene-SOA was relatively aged over Gosan aerosols. The estimated SOC with the predominance in summer shows that substantial BSOA formation occurred in summer due to favorable meteorological conditions. The backward air mass trajectories and source apportionment studies entirely demonstrated that emission from East Asia significantly dominates the ambient OA mass over KCOG. Interestingly, levoglucosan-C exhibited a significant positive correlation with nonfossil and fossil organic carbon fractions, along with the comparable regression slopes. This result reveals that BB and coal (lignite) combustion both are prominent sources for levoglucosan in the East Asian outflow.

Although there is some evidence that levoglucosan could originate from the combustion of brown coals (e.g., lignite) in China, our observations from the KCOG (receptor site) also hint at the fossil source contribution of this molecular marker in the regional influx of the East Asian outflow. Therefore, attribution of ambient levoglucosan levels over the WNP to the impact of BB emission may cause large uncertainty.

### 716 Data availability

717 The data used in this paper are available upon request from the corresponding author.

## 718 Author contributions

- 719 KK and YLZ designed the research. ML collected the aerosol samples. MMH and SB
- performed the analysis of aerosol samples. MMH wrote the paper under the guidance of YLZ
- and KK. All authors were actively involved in the discussion of the paper.

### 722 Competing interests

The authors declare that they have no conflict of interest.

## 724 Acknowledgments

- 725 We acknowledge the financial supports of the Japan Society for the Promotion of Science
- 726 (JSPS) through Grant-in-Aid No. 24221001 and the National Natural Science Foundation of
- 727 China (Grant No 41977305).
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aerosol samples from Gosan.

Species	Annual	Summer	Fall	Winter	Spring
	$Avg.^{a} \pm S.D.^{b}$ Min. <sup>c</sup> , Max. <sup>d</sup>	Avg. ± S.D. Min., Max.	Avg. ± S.D. Min., Max.	Avg. ± S.D. Min., Max.	Avg. ± S.D. Min., Max.
Anhydrosugars					
Levoglucosan (Lev)	$17.6 \pm 16.8 \\ 0.60, 45.9$	$\begin{array}{c} 2.92 \pm 3.89 \\ 0.60, \ \ 9.81 \end{array}$	$21.7 \pm 19.0$ 2.30, 43.9	$39.2 \pm 6.60$ 32.0, 45.9	$12.7 \pm 11.6$ 1.45, 33.4
Mannosan (Man)	$1.57 \pm 1.82$ 0.05, 6.74	$0.18 \pm 0.24$ 0.05, 0.61	$1.69 \pm 1.49$ 0.13, 3.66	$3.63 \pm 2.28$ 1.47, 6.74	$1.31 \pm 1.57$ 0.08, 4.08
Galactosan (Gal)	$2.28 \pm 2.10$ 0.14, 6.78	$0.64 \pm 0.68$ 0.14, 1.82	$2.45 \pm 2.13$ 0.35, 4.92	$5.21 \pm 1.64$ 3.40, 6.78	$1.65 \pm 1.26$ 0.50, 3.88
Primary Sugars	,	011., 1102	0.000, 0.02	0.1.0, 0.1.0	0.000, 0.000
Glucose	$18.8 \pm 27.1$ 2.45, 122	$13.4 \pm 18.2$ 2.45, 45.6	$16.5 \pm 15.7$ 2.68, 33.6	$4.74 \pm 3.14$ 2.88, 9.44	$32.4 \pm 41.1$ 4.87, 122
Fructose	$10.3 \pm 15.9$ 0.97, 74.0	$4.90 \pm 5.15$ 0.97, 13.7	$7.48 \pm 6.99$ 1.71, 16.2	$3.82 \pm 4.45$ 1.56, 10.5	$19.8 \pm 25.0 \\ 2.69, 74.0$
Sucrose	$16.1 \pm 32.2$ 0.26, 140	$1.46 \pm 1.05$ 0.68, 3.28	$9.74 \pm 12.1$ 0.76, 27.2	$8.87 \pm 16.3$ 0.42, 33.3	$35.1 \pm 50.5$ 0.26, 140
Trehalose	$\begin{array}{c} 2.42 \pm 1.97 \\ 0.65, \ \ 7.03 \end{array}$	$\begin{array}{c} 2.72 \pm 2.53 \\ 0.97, \ \ 6.98 \end{array}$	$3.71 \pm 2.29$ 1.18, 7.03	$1.21 \pm 0.42$ 0.72, 1.63	$1.98 \pm 1.55$ 0.65, 5.33
Xylose	$\begin{array}{c} 0.81 \pm 0.65 \\ 0.04, \ \ 2.03 \end{array}$	$\begin{array}{c} 0.23 \pm 0.23 \\ 0.04, \ 0.63 \end{array}$	$\begin{array}{c} 0.86 \pm 0.68 \\ 0.14, \ 1.70 \end{array}$	$\begin{array}{c} 1.59 \pm 0.37 \\ 1.21, \ \ 2.03 \end{array}$	$\begin{array}{c} 0.74 \pm 0.55 \\ 0.16, \ 1.68 \end{array}$
Sugar alcohols					
Arabitol	$\begin{array}{c} 3.96 \pm 4.24 \\ 0.47,  18.7 \end{array}$	$5.64 \pm 7.46$ 1.20, 18.7	$5.27 \pm 3.51$ 2.27, 10.9	$\begin{array}{c} 1.06 \pm 0.60 \\ 0.47, \ 1.91 \end{array}$	$3.47 \pm 2.19$ 1.18, 6.30
Mannitol	$4.61 \pm 5.54$ 0.25, 22.0	$7.39 \pm 8.65$ 1.71, 22.0	$6.64 \pm 6.03$ 1.69, 16.7	$0.99 \pm 0.45$ 0.55, 1.60	$3.24 \pm 2.67$ 0.25, 7.03
Erythritol	$0.62 \pm 0.43$ 0.12, 1.52	$\begin{array}{c} 0.92 \pm 0.53 \\ 0.42, \ 1.52 \end{array}$	$\begin{array}{c} 0.93 \pm 0.33 \\ 0.55, \ 1.27 \end{array}$	$0.42 \pm 0.27$ 0.16, 0.80	$\begin{array}{c} 0.30 \pm 0.12 \\ 0.12, \ \ 0.48 \end{array}$
Inositol	$\begin{array}{c} 0.34 \pm 0.39 \\ 0.04, \ 1.53 \end{array}$	$\begin{array}{c} 0.29 \pm 0.41 \\ 0.08, \ 1.03 \end{array}$	$\begin{array}{c} 0.56 \pm 0.60 \\ 0.10, \ 1.53 \end{array}$	$\begin{array}{c} 0.13 \pm 0.07 \\ 0.08, \ \ 0.23 \end{array}$	$\begin{array}{c} 0.35 \pm 0.28 \\ 0.04, \ 0.73 \end{array}$
Isoprene-SOA tracers					
2-MGA	$0.99 \pm 0.70 \\ 0.17, 2.79$	$1.61 \pm 1.17$ 0.17, 2.79	$0.95 \pm 0.43$ 0.51, 1.61	$0.67 \pm 0.14 \\ 0.49, 0.81$	$\begin{array}{c} 0.76 \pm 0.35 \\ 0.20, \ \ 1.32 \end{array}$
Σ2-MLTs	$1.04 \pm 1.40$ 0.05, 5.81	$2.48 \pm 1.98$ 0.51, 5.81	$1.34 \pm 1.16$ 0.33, 2.74	$0.20 \pm 0.05$ 0.15, 0.26	$0.29 \pm 0.25$ 0.05, 0.82
$\Sigma$ C5-alkene triols	$0.20 \pm 0.25$ 0.02, 1.17	$0.46 \pm 0.44$ 0.02, 1.17	$0.13 \pm 0.05$ 0.05, 0.18	$0.09 \pm 0.03$ 0.05, 0.13	$0.14 \pm 0.11$ 0.02, 0.30
Monoterpene-SOA trace					
<i>cis</i> -pinonic acid	$\begin{array}{c} 0.15 \pm 0.14 \\ 0.02,  0.52 \end{array}$	$\begin{array}{c} 0.12 \pm 0.13 \\ 0.03, \ 0.36 \end{array}$	$0.08 \pm 0.06$ 0.02, 0.16	$\begin{array}{c} 0.07 \pm 0.03 \\ 0.02, \ \ 0.10 \end{array}$	$\begin{array}{c} 0.26 \pm 0.17 \\ 0.08, \ \ 0.52 \end{array}$
pinic acid	$1.67 \pm 1.01$ 0.72, 4.81	$2.40 \pm 1.67$ 0.99, 4.81	$1.53 \pm 0.57$ 0.75, 2.11	$1.07 \pm 0.25$ 0.84, 1.42	$1.61 \pm 0.78$ 0.72, 2.94
3-HGA	$2.30 \pm 2.38$ 0.19, 10.6	$3.46 \pm 4.39$ 0.19, 10.6	$1.52 \pm 0.77$ 0.76, 2.79	$1.40 \pm 0.39$ 0.94, 1.88	$2.54 \pm 1.83$ 0.38, 4.57
MBTCA	$5.11 \pm 4.54$ 0.29, 19.6	$8.44 \pm 6.88$ 3.00, 19.6	$6.24 \pm 3.64$ 1.17, 9.49	$1.90 \pm 0.85$ 0.97, 2.83	$3.77 \pm 2.96$ 0.29, 8.08

<sup>a</sup>Average, <sup>b</sup>Standard deviation, <sup>c</sup>Minimum, <sup>d</sup>Maximum. 2-MGA: 2-methylglyceric acid, 2-MLTs: 2-methyltetrols, 3-HGA: 3-hydroxyglutaric acid, MBTCA: 3-methyl-1,2,3-butanetricarboxylic acid.

1124	Table 2. Statistical summary of diagnostic ratios and carbonaceous components contribution in Gosan
1125 1126	aerosols.

Species	Annual	Summer	Fall	Winter	Spring	
	$Avg.^{a} \pm S.D.^{b}$ Min. <sup>c</sup> , Max. <sup>d</sup>	Avg. ± S.D. Min., Max.	Avg. ± S.D. Min., Max.	Avg. ± S.D. Min., Max.	Avg. $\pm$ S.D. Min., Max.	
Diagnostic ratios						
Lev/Man	$\begin{array}{l} 15.1 \pm 6.76 \\ 6.81,  31.3 \end{array}$	$\begin{array}{l} 17.1 \pm 8.21 \\ 8.58, \ \ 28.4 \end{array}$	$\begin{array}{c} 13.7 \pm 2.51 \\ 11.1, \ 17.3 \end{array}$	$\begin{array}{c} 13.5 \pm 6.24 \\ 6.81, \ \ 21.7 \end{array}$	$\begin{array}{c} 15.4 \pm 8.76 \\ 6.98, \ \ 31.3 \end{array}$	
Man/Gal	$\begin{array}{c} 0.55 \pm 0.32 \\ 0.10, \ \ 1.07 \end{array}$	$\begin{array}{c} 0.28 \pm 0.13 \\ 0.10, \ \ 0.42 \end{array}$	$\begin{array}{c} 0.64 \pm 0.20 \\ 0.37, \ 0.90 \end{array}$	$\begin{array}{c} 0.66 \pm 0.27 \\ 0.43, \ 1.05 \end{array}$	$\begin{array}{c} 0.61 \pm 0.43 \\ 0.16, \ 1.07 \end{array}$	
Lev/(Man + Gal)	$\begin{array}{l} 4.27 \pm 1.23 \\ 2.18, \ \ 6.56 \end{array}$	$3.12 \pm 0.78$ 2.18, 4.03	$\begin{array}{l} 5.08 \pm 0.32 \\ 4.71, \ \ 5.51 \end{array}$	$\begin{array}{l} 4.85 \pm 1.35 \\ 3.49, \ \ 6.56 \end{array}$	$\begin{array}{l} 4.17 \pm 1.33 \\ 2.50, \ \ 6.45 \end{array}$	
$\text{Lev/K}^+ \times 10^{-2}$	$\begin{array}{c} 5.73 \pm 5.65 \\ 0.65, \ \ 23.2 \end{array}$	$\begin{array}{c} 1.00 \pm 0.52 \\ 0.65, \ 1.91 \end{array}$	$3.94 \pm 2.51$ 1.26, 6.30	$\begin{array}{c} 12.3 \pm 7.87 \\ 6.42, \ \ 23.2 \end{array}$	$6.65 \pm 4.48$ 1.77, 15.3	
Lev/OC $\times 10^{-2}$	$\begin{array}{c} 0.73 \pm 0.66 \\ 0.08, \ \ 2.29 \end{array}$	$\begin{array}{c} 0.14 \pm 0.10 \\ 0.08, \ \ 0.31 \end{array}$	$0.84 \pm 0.61$ 0.12, 1.56	$1.60 \pm 0.67$ 0.99, 2.29	$0.56 \pm 0.40 \\ 0.11, 1.05$	
Lev/WSOC $\times 10^{-2}$	$\begin{array}{c} 1.09 \pm 0.97 \\ 0.10,  3.46 \end{array}$	$\begin{array}{c} 0.20 \pm 0.13 \\ 0.10, \ \ 0.42 \end{array}$	$\begin{array}{c} 1.25 \pm 0.94 \\ 0.23, \ \ 2.21 \end{array}$	$\begin{array}{c} 2.33 \pm 0.90 \\ 1.48, \ \ 3.46 \end{array}$	$\begin{array}{c} 0.92 \pm 0.65 \\ 0.13, \ 1.69 \end{array}$	
2-MGA/2-MLTs	$\begin{array}{c} 2.18 \pm 1.59 \\ 0.33, \ \ 5.40 \end{array}$	$\begin{array}{c} 0.67 \pm 0.43 \\ 0.33, \ 1.33 \end{array}$	$\begin{array}{c} 1.04 \pm 0.49 \\ 0.41, \ \ 1.55 \end{array}$	$3.60 \pm 1.43$ 1.90, 5.40	$3.27 \pm 1.17$ 1.60, 4.75	
°P/MBTCA	$\begin{array}{l} 0.62 \pm 0.61 \\ 0.17, \ \ 5.90 \end{array}$	$3.22 \pm 0.62$ 2.47, 3.89	$\begin{array}{l} 3.62 \pm 1.67 \\ 1.52, \ \ 5.90 \end{array}$	$1.68 \pm 0.64$ 0.83, 2.38	$\begin{array}{c} 1.73 \pm 1.07 \\ 0.34, \ \ 3.32 \end{array}$	
Carbonaceous components						
Isoprene derived SOC (μgC m <sup>-3</sup> )	$\begin{array}{c} 23.7 \pm 23.9 \\ 2.26, \ 97.4 \end{array}$	$\begin{array}{l} 51.0 \pm 32.4 \\ 8.08, \ 97.4 \end{array}$	$\begin{array}{l} 25.9 \pm 17.9 \\ 8.54, \ \ 45.7 \end{array}$	$\begin{array}{l} 8.52 \pm 1.07 \\ 7.46, \ \ 9.87 \end{array}$	$\begin{array}{c} 11.3 \pm 6.80 \\ 2.26, \ \ 24.9 \end{array}$	
Isoprene SOC to OC (%)	$\begin{array}{l} 1.45 \pm 1.74 \\ 0.21, \ \ 6.40 \end{array}$	$\begin{array}{l} 3.56 \pm 2.18 \\ 1.03, \ \ 6.40 \end{array}$	$\begin{array}{c} 1.46 \pm 1.38 \\ 0.35, \ \ 3.48 \end{array}$	$\begin{array}{c} 0.35 \pm 0.17 \\ 0.21, \ \ 0.57 \end{array}$	$\begin{array}{c} 0.57 \pm 0.38 \\ 0.24, \ 1.35 \end{array}$	
Isoprene SOC to total SOC (%)	$\begin{array}{c} 35.5 \pm 15.5 \\ 13.5, \ \ 71.1 \end{array}$	$\begin{array}{l} 46.6 \pm 20.3 \\ 26.0, \ \ 71.1 \end{array}$	$38.1 \pm 10.4$ 21.1, 47.3	$\begin{array}{c} 31.6 \pm 7.05 \\ 24.0,  39.7 \end{array}$	$\begin{array}{c} 27.9 \pm 15.5 \\ 13.5, \ 54.0 \end{array}$	
Monoterpene SOC (µgC m <sup>-3</sup> )	$40.1 \pm 33.3$ 7.18, 154	$\begin{array}{l} 62.7 \pm 56.5 \\ 19.2,  154 \end{array}$	$\begin{array}{l} 40.7 \pm 20.0 \\ 11.8, \ 57.7 \end{array}$	$\begin{array}{c} 19.3 \pm 5.61 \\ 14.0, \ \ 25.2 \end{array}$	$\begin{array}{l} 35.5 \pm 23.5 \\ 7.18, \ \ 62.7 \end{array}$	
Monoterpene SOC to OC (%)	$\begin{array}{l} 2.0 \pm 1.47 \\ 0.37, \ \ 5.74 \end{array}$	$3.65 \pm 1.58$ 2.44, 5.74	$\begin{array}{c} 2.15 \pm 1.45 \\ 0.48, \ \ 3.88 \end{array}$	$\begin{array}{c} 0.87 \pm 0.58 \\ 0.37, \ 1.48 \end{array}$	$\begin{array}{c} 1.58 \pm 0.84 \\ 0.54, \ \ 3.13 \end{array}$	
Monoterpene SOC to total SOC (%)	$\begin{array}{l} 64.5 \pm 15.5 \\ 28.9, \ 86.5 \end{array}$	$53.4 \pm 20.3 \\ 28.9, \ 74.0$	$61.9 \pm 10.4$ 52.7, 78.9	$\begin{array}{c} 68.4 \pm 7.05 \\ 60.3, \ \ 76.1 \end{array}$	$\begin{array}{c} 72.1 \pm 15.5 \\ 46.0, \ 86.5 \end{array}$	

<sup>e</sup>P: *cis*-pinonic acid + pinic acid

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Sampling sites	Sampling type	Sampling time	Anhydro- sugars	Primary sugars	Sugar alcohols	References
Gosan, South Korea	TSP	Summer Fall Winter Spring	3.74 25.9 48.0 15.7	22.8 38.3 20.2 90.0	14.3 13.4 2.60 7.36	This study
Chennai, India	$PM_{10}$	Summer Winter	127 134	15.5 11.4	7.44 4.81	Fu et al., 2010b
Mt. Tai, China	TSP	Summer (June)	224	61.1	125	Fu et al., 2012b
Alert, Canada	TSP	Winter Spring	0.32 0.02	1.14 0.18	0.25 0.36	Fu et al., 2009a
Okinawa, western North Pacific	TSP	Summer Autumn Winter Spring	0.93 2.58 6.04 3.44	73.5 56.0 34.4 101	62.9 30.7 6.03 31.2	Zhu et al., 2015a, b
Chichijima, western North Pacific	TSP	Summer Autumn Winter Spring	0.32 0.85 2.40 0.94	32.8 22.0 14.2 24.2	38.6 35.1 3.93 15.5	Verma et al., 2015, 2018
Mt. Hua, China (Non-dust storm)	PM <sub>10</sub>	April	57.8	92.5	22.4	Wang et al., 2012
Mt. Hua, China (Dust storm)	PM <sub>10</sub>	April	44.5	162	25.7	Wang et al., 2012
Nanjing, China	PM <sub>2.5</sub>	Summer Winter	151 (Lev) 268 (Lev)	59.3 42.3	11.8 13.4	Wang and Kawamura, 2005
Beijing, China	PM <sub>2.5</sub>	Summer Autumn Winter Spring	14.5 129 254 81.4	6.63 17.2 41.5 33.9	3.31 13.7 17.8 12.3	Kang et al., 2018a
Belgrade, Serbia	TSP	Autumn	425 (Lev)	116	98.4	Zangrando et al., 2016
Maine, USA	$PM_1$	May- October	13.9	28	8.31	Medeiros et al., 2006
Crete, Greece	$PM_{10}$	Year-round	14.4	32.3	6.53	Theodosi et al., 2018

**Table 3.** Comparisons of the mean concentration (ng m<sup>-3</sup>) of anhydrosugars, sugar, and sugar alcohols in Gosan aerosols with those from different sites around the world.