1	Regional heterogeneities in the emission of airborne primary sugar
2	compounds and biogenic secondary organic aerosols in the East Asian
3	outflow: Evidence for coal combustion as a source of levoglucosan
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### 31 ABSTRACT

32 Biomass burning (BB) significantly influences the chemical composition of organic aerosols 33 (OA) in the East Asian outflow. Source apportionment of BB-derived OA is an influential 34 factor for understanding their regional emissions, which is crucial for reducing uncertainties 35 in their projected climate and health-effects. We analyzed here three different classes of 36 atmospheric sugar compounds (anhydrosugars, primary sugars, and sugar alcohols) and two 37 types of biogenic secondary organic aerosol (BSOA) tracers (isoprene- and monoterpene 38 derived SOA products) in a year-long collected total suspended particulate matter (TSP) from 39 an island-based receptor site in South Korea, Gosan. We investigate seasonal variations in the 40 source-emissions of BB-derived OA using mass concentrations of anhydrosugars and radiocarbon (<sup>14</sup>C-) isotopic composition of organic carbon (OC) and elemental carbon (EC) in 41 42 ambient aerosols. Levoglucosan (Lev) is the most abundant anhydrosugar, followed by 43 galactosan (Gal) and mannosan (Man). Strong correlations of Lev with Gal and Man, along 44 with their ratios (*Lev/Gal:* 6.65  $\pm$  2.26; *Lev/Man:* 15.1  $\pm$  6.76) indicate the contribution from 45 hardwood burning emissions. The seasonal trends revealed that the BB impact is more 46 pronounced in winter and fall, as evidenced by the high concentrations of anhydrosugars. 47 Likewise, significant correlations among three primary sugars (i.e., glucose, fructose, and 48 sucrose) emphasized the contribution of airborne pollen. The primary sugars showed higher 49 concentrations in spring/summer than winter/fall. The fungal spore tracer compounds (i.e., 50 arabitol, mannitol, and erythritol) correlated well with trehalose (i.e., a proxy for soil organic 51 carbon), suggesting the origin from airborne fungal spores and soil microbes in the East 52 Asian outflow. These sugar alcohols peaked in summer, followed by spring/fall and winter. 53 Monoterpene-derived SOA tracers were most abundant compared to isoprene-SOA tracers. 54 Both BSOA tracers were dominant in summer, followed by fall, spring, and winter. The 55 source apportionment based on multiple linear regressions and diagnostic mass ratios 56 together revealed that BB emission mostly contributed from hardwood and crop-residue burning. We also found significant positive linear relationships of <sup>14</sup>C-based nonfossil- and 57 58 fossil-derived organic carbon fractions with Lev-C along with the comparable regression 59 slopes, suggesting the importance of BB and coal combustion sources in the East Asian 60 outflow.

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Keywords: Biomass burning tracers, primary biological aerosol particles, biogenic SOA
 tracers, radiocarbon-based source apportionment, organic aerosols, East Asian outflow

## 64 **1. Introduction**

Organic aerosols (OA), which account for a major fraction of up to 50% of airborne total 65 66 suspended particulate matter, have considerable effects on regional and global climate by 67 absorbing or scattering sunlight (Kanakidou et al., 2005). However, the climate effects of OA 68 are involved with large uncertainties due to our limited understanding of the contributing 69 sources. OA can be derived from both primary emissions and secondarily formed species. 70 Sugars are an important group of water-soluble, primary organic compounds whose 71 concentrations are significant in atmospheric aerosols over the continent (Jia and Fraser, 72 2011; Fu et al., 2008; Yttri et al., 2007; Graham et al., 2003). Anhydrosugars such as 73 levoglucosan, galactosan, and mannosan are the key tracers of biomass burning (BB) 74 emissions (Simoneit, 2002). Sugar alcohols, along with glucose, trehalose and sucrose are 75 mostly originated from primary biological particles such as fungal spores, pollen, bacteria, 76 and viruses, and vegetative debris (Graham et al., 2003; Simoneit et al., 2004a; Bauer et al., 77 2008; Deguillaume et al., 2008). Primary sugars and sugar alcohols are predominantly 78 present in the coarse mode aerosols, accounting for 0.5-10% of atmospheric aerosol carbon 79 matter (Yttri et al., 2007; Pio et al., 2008).

80 Secondary organic aerosol (SOA) is a large fraction of OA, while there were only 81 limited studies about the key factors controlling SOA formation. The SOA formation 82 significantly increased with the enhancement of the ambient aerosol mass (Liu et al., 2018). 83 SOA is formed by both homogenous and heterogeneous reactions of volatile organic 84 compounds (VOCs) in the atmosphere (Surratt et al., 2010; Robinson et al., 2007; Claeys et 85 al., 2004). On a global estimation, biogenic VOCs (BVOCs) such as isoprene, monoterpenes 86 (e.g.,  $\alpha/\beta$ -pinene), and sesquiterpenes (e.g.,  $\beta$ -caryophyllene) are one order of magnitude 87 higher than those of anthropogenic VOCs (e.g., toluene) (Guenther et al., 2006). The global emissions of annual BVOCs were estimated to be 1150 TgC yr<sup>-1</sup>, accounting for 44% 88 89 isoprene and 11% monoterpenes (Guenther et al., 1995). Isoprene is highly reactive and 90 promptly reacts with oxidants such as O<sub>3</sub>, OH, and NO<sub>x</sub> in the atmosphere to form SOA 91 (Kroll et al., 2005, 2006; Ng et al., 2008; Surratt et al., 2010; Bikkina et al., 2021), estimated to be 19.2 TgC yr<sup>-1</sup>, consisting of ~70% of the total SOA budget (Heald et al., 2008). 92 93 Monoterpenes are important sources of biogenic secondary organic aerosol (BSOA), 94 considering  $\alpha$ -pinene as major species, accounting for ~35% of the global monoterpenes 95 emissions (Griffin et al., 1999).

Anthropogenic activities such as coal and biofuel combustion over East Asia,
including China, are responsible for the vast emission of OA (Huebert et al., 2003; Zhang et

98 al., 2016). Understanding the ambient levels OA in the East Asian outflow is crucial for 99 assessing their regional climatic effects. As part of this effort, the Korean Climate 100 Observatory at Gosan (KCOG), a super site located in South Korea, is an ideal location for 101 investigating the atmospheric outflow characteristics from East Asia (Fu et al., 2010a; Kundu 102 et al., 2010; Ramanathan et al., 2007; Kawamura et al., 2004; Arimoto et al., 1996). For 103 instance, primary OA associated with soil/desert dust in East Asia, along with forest fires in 104 Siberia/northeastern China, are transported over Gosan in spring (Wang et al., 2009a). BSOA 105 during long-range transport from the continent and Open Ocean, as well as local vegetation, 106 can significantly contribute to Gosan aerosols. Although these investigations were carried out 107 almost a decade ago, no such observations are available in contemporary times from Gosan. 108 Here, we attempt to understand the current states of East Asian OA using both the molecular 109 marker approach and radiocarbon data of carbonaceous components.

110 The KCOG, located on the western side of Jeju Island adjacent to the Yellow Sea and the East China Sea, is facing the Asian continent but is isolated from public areas of the 111 112 island (Kawamura et al., 2004). Simoneit et al. (2004b) have documented during the ACE-113 Asia campaign that OA from the BB and fossil fuel combustion sources are transported along 114 with desert dust to the KCOG during continental outflow. An intensive campaign was 115 organized at the KCOG during spring 2005 to observe the physical properties of East Asian 116 aerosols while two dust events were detected (Nakajima et al., 2007). Here, we focus on the 117 characterization of airborne anhydrosugars, primary sugars, sugar alcohols, and BSOA 118 tracers from the KCOG. Gosan is influenced by the continental outflow from East Asia 119 during winter, spring and fall, whereas the site is influenced by the maritime air masses from 120 the Pacific Ocean and other marginal seas. This makes the KCOG ideal for characterizing the 121 regional heterogeneities in the emissions of organic compounds in the East Asian outflow 122 based on the TSP samples collected during April 2013-April 2014.

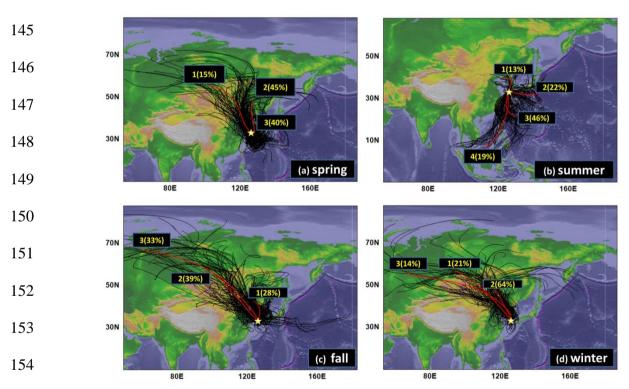
## 123 **2. Methods**

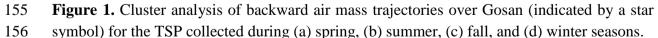
# 124 **2.1. Aerosol sampling and prevailing meteorology**

Total suspended particles (TSP) were collected on pre-combusted (450°C for 6 h) quartz fiber filters (20 cm  $\times$  25 cm, Pallflex) at the KCOG (33.17 °N, 126.10 °E, see Figure 1), South Korea. To get enough signal for the radiocarbon measurements, each TSP sample was collected for 10–14 days from April 2013 to April 2014. Twenty-one samples were collected using a high-volume air sampler (Kimoto AS-810, ~65 m<sup>3</sup> h<sup>-1</sup>) installed on the rooftop of a trailer house (~3 m above the ground). After the collection, aerosol filters were transferred to a pre-combusted (450°C for 6 h) glass jar (150 mL) equipped with a Teflon-lined screw cap and transported to the laboratory in Sapporo. These TSP samples were stored in a dark
freezer room at -20°C until the analysis. Three field blank filters were also collected during
the campaign.

135 The ambient temperatures at the Gosan site were on average 6.9°C in winter, 14.1°C 136 in spring, 27.0°C in summer, and 17.1°C in fall. Likewise, the average relative humidity was 137 found to be highest in summer (71.3%), followed by spring (64.9%), fall (63.5%), and winter 138 (54.7%). Gosan is influenced by the pollution sources in East Asia during winter as well as 139 other transition periods (spring and fall) due to the prevailing westerlies. In contrast, winds in 140 summer blew mostly from the western North Pacific (WNP) by the easterly winds. The 141 spring season is, in particular, important for the transport of mineral dust mixed with polluted 142 OA to Gosan (Kundu et al., 2010).

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## 158 **2.2. Extraction and analysis of organic compounds**

159 Approximately  $3.14 \text{ cm}^2$  filter cuts were extracted with dichloromethane/methanol (2:1; v/v). 160 The extracts were concentrated using a rotary evaporator under vacuum and then blown down 161 to near dryness with pure nitrogen gas. The dried residues were subsequently reacted with N, 162 O-bis(trimethylsilyl)trifluoroacetamide containing 1% trimethylchlorosilane (BSTFA+1% TMCS, SUPELCO<sup>©</sup>, Sigma-Aldrich<sup>®</sup>) and pyridine at 70 °C for 3 h to derive OH and COOH 163 164 groups of polar organic compounds to trimethylsilyl ethers and esters, respectively. After the 165 derivatization followed by the addition of a known amount of internal standard solution (Tridecane; 1.43 ng  $L^{-1}$  in n-hexane), the derivatized extracts were injected onto a gas 166 chromatograph (Hewlett-Packard model 6890 GC) coupled to a mass spectrometer (Hewlett-167 168 Packard model 5973, MSD) (GC-MS). More details on the quantification of polar organic 169 compounds using GC-MS are described in Haque et al. (2019).

170 The target compounds (anhydrosugars, primary sugars, sugar alcohols, and BSOA tracers) were separated on a DB-5MS fused silica capillary column (30 m x 0.25 mm i.d., 0.5 171 172 µm film thickness) using helium as a carrier gas at a flow rate of 1.0 ml min<sup>-1</sup>. The GC oven temperature was programmed from 50°C for 2 min and then increased from 50 to 120°C at 173 30°C min<sup>-1</sup> and to 300°C at 6°C min<sup>-1</sup> with a final isotherm hold at 300°C for 16 min. The 174 175 sample was injected in a splitless mode with the injector temperature at 280°C. The MS was operated at 70 eV and scanned from 50 to 650 Da on an electron impact (EI) mode. Mass 176 177 spectral data were acquired and processed using the Chemstation software. The organic 178 compounds were identified individually by comparison with retention times and mass spectra 179 of authentic standards and NIST library and literature data of mass fragmentation patterns 180 (Medeiros and Simoneit, 2007). For assessing the recoveries, ~100-200 ng of the standard 181 solution was spiked on the blank filter and analyzed as a real sample. Overall, the average 182 recoveries were found to be 80-104% for target compounds. The field and laboratory blank 183 filters (n = 3) were also analyzed by the same procedures as a real sample. Target compounds 184 were not found in the field blanks. The analytical errors based on concentrations by replicate 185 sample analyses (n = 3) were less than 15%.

# 186 **2.3. Carbon fractions analysis**

187 Organic carbon (OC) and elemental carbon (EC) were analyzed using a thermal-optical 188 transmittance method with a Sunset Laboratory carbon analyzer following the NIOSH protocol (Birch and Cary, 1996), and detailed procedures were given elsewhere (Zhang et al., 189 2016). Furthermore, a portion of 2.54 cm<sup>2</sup> of each sample filter was extracted with 15 mL 190 191 ultrapure water (resistivity > 18.2 M $\Omega$ cm, Sartorius arium 611 UV) with ultrasonication for 192 30 min. The water extracts were then filtered through a membrane disc filter for water-193 soluble organic carbon (WSOC) analysis by a total organic carbon (TOC) analyzer 194 (Shimadzu, TOC-Vcsh) (Boreddy et al., 2018). The concentrations of WSOC were corrected 195 by field blanks. The analytical errors in the triplicate analyses were less than 5% for WSOC.

### 196 2.4. Radiocarbon isotopic composition of total carbon (TC) and EC

The concentrations of TC in the TSP samples were determined using an elemental analyzer. 197 For the radiocarbon isotopic composition ( $\Delta^{14}$ C), the aerosol filter punches were exposed for 198 ~12 h to HCl fumes in a vacuum desiccator. Subsequently, these filters were analyzed for 199  $\Delta^{14}$ C on a modified elemental analyzer coupled via a gas interface to Accelerator Mass 200 201 Spectrometer Mini Carbon Dating System (MICADAS) at the University of Bern, 202 Switzerland (Salazar et al., 2015). The evolved CO<sub>2</sub> of TC from the elemental analyzer was 203 passed through a moisture trap (Sicapent, Merck) and isolated from other residual gasses 204 using a temperature-controlled zeolite trap. The purified CO<sub>2</sub> was introduced through a gas interface system to MICADAS, where  ${}^{14}C/{}^{12}C$  ratios are measured according to the analytical 205 procedures detailed in Zhang et al. (2016). Likewise, the evolved CO<sub>2</sub> of elemental carbon 206 from the Sunset Lab OC/EC analyzer using the Swiss 4S protocol (Zhang et al., 2012), was 207 directed to the MICADAS and measured for the  ${}^{14}C/{}^{12}C$  ratio relative to standard calibration 208 gas. These results were expressed as fractions of modern carbon  $(f_M)$  by normalizing with a 209  $\delta^{13}$ C value of the reference standard in the year 1950 (-25‰) according to Stuiver and Polach 210 (1997) for the fractionation effects. The  $f_{\rm M}(\rm OC)$  can be estimated by using the  $f_{\rm M}(\rm TC)$  and 211  $f_{\rm M}({\rm EC})$  in an isotope mass balance equation (Zhang et al., 2015). Additionally, we estimated 212 the relative contributions of OC and EC from the nonfossil and fossil sources (fnonfossil and 213  $f_{\text{fossil}}$ , respectively) using the following equations. 214

215 
$$f_{\text{nonfossil-OC}} = f_{\text{M}}(\text{OC-sample})/f_{\text{M}}(\text{OC-ref}); f_{\text{M}}(\text{OC-ref}) \approx 1.07 \pm 0.04$$
 (1)  
216  $f_{\text{nonfossil-EC}} = f_{\text{M}}(\text{EC-sample})/f_{\text{M}}(\text{EC-ref}); f_{\text{M}}(\text{EC-ref}) \approx 1.10 \pm 0.05$  (2)  
217  $f_{\text{fossil-OC}} = 1 - f_{\text{nonfossil-OC}}$  (3)  
218  $f_{\text{fossil-EC}} = 1 - f_{\text{nonfossil-EC}}$  (4)

The reference values of OC and EC were obtained from Mohn et al. (2008). Using the fractions of  $f_{\text{fossil-OC}}$  and  $f_{\text{nonfossil-OC}}$ , we can, therefore, estimate the mass concentration of ambient organic carbon (OC-ambient) from fossil and nonfossil sources (OC<sub>fossil</sub> and OC<sub>nonfossil</sub>, respectively).

223 
$$OC_{nonfossil} = f_{nonfossil-OC} \times [OC]_{ambient}$$
 (5)

$$OC_{fossil} = f_{fossil-OC} \times [OC]_{ambient}$$
(6)

225 More details on the radiocarbon isotopic composition data over Gosan were reported 226 elsewhere (Zhang et al., 2016).

227 **3. Results and discussion** 

### 228 **3.1. Trajectory and cluster analysis**

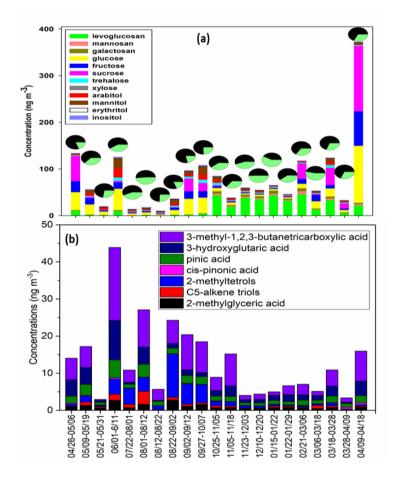
229 Backward air mass trajectories are useful for assessing the impact of local versus regional 230 source emissions over Gosan. Seven-day isentropic backward air mass trajectories were 231 computed using the hybrid single-particle Lagrangian integrated trajectory model (HYSPLIT, 232 version 4: Stein et al., 2015) over the KCOG for the sampling period using the 233 meteorological datasets of the Global Data Assimilation System (GDAS) network. The 234 trajectory endpoint files from the HYSPLIT model were further used for the cluster analysis 235 using the Trajstat package (Wang et al., 2009b) for all four seasons (Figure 1). Although 236 cluster analysis revealed the predominance of continental transport in the spring, fall, and 237 winter seasons, the air masses over the KCOG in summer mostly originated from the WNP. 238 Since spring is a transition of winds switching from westerlies to easterlies, Gosan is likely 239 influenced by the long-range transport of dust, pollution, and sea-salt aerosols.

240 The vertical mixing of pollutants within the boundary layer height also plays an important role in controlling the strength of continental outflow alongside regional 241 242 meteorology. For instance, the mixing height of air parcels from the HYSPLIT model is 243 mostly confined to 1000 m in winter but somewhat increased towards the spring and fall 244 seasons (Figure S1). This vertical enhancement in the boundary layer height facilitates the 245 transport of mineral dust particles from the arid and semiarid regions in East Asia along with 246 urban pollutants to Gosan in spring and fall compared to winter. However, the strength of the 247 continental outflow somewhat depends on several factors, including source emissions, 248 meteorology, and mixing height of air parcels.

249 Gosan is influenced by three types of air masses in spring (Figure 1a), from the 250 Mongolian Desert (cluster 1: 15%), North China (cluster 2: 45%), and from the Yellow Sea 251 (cluster 3: 40%). In contrast, the easterlies from the WNP in summer mostly influenced the 252 composition of TSP over the KCOG. This inference is based on the cluster analysis for 253 summer samples (Figure 1b), which showed four regimes, including transport from the Sea of 254 Japan (cluster 1: 13%), WNP (cluster 2: 22%), South China Sea (cluster 3: 46%), and East 255 China Sea (cluster 4: 19%). In contrast, cluster analysis revealed three major transport 256 regimes from East Asia in fall and winter (Figure 1c-d). However, there are subtle differences 257 that exist between winter and fall in terms of influence from nearby versus distant pollution 258 sources. For instance, long-range transport of air masses from west Mongolia (cluster 1: 259 21%) and the Russian Far East (cluster 3: 14%) exerted a weak influence on the TSP sampled 260 over Gosan in winter. Besides, we observed a somewhat larger impact of air masses from the 261 North China Plain over Gosan (cluster 2: 64%) in winter. In contrast, Gosan is less influenced

by air masses originating from the North China Plain, contributing ~28% (cluster 1) than
those from Mongolia (cluster 2: 39%) and the Russian Far East (cluster 3: 33%) in fall.
Therefore, the impact of East Asian outflow is stronger in winter than in spring and fall.

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Figure 2. (a) Cumulative concentration levels of anhydrosugars, primary sugars, and sugar alcohols (i.e., represented by bars), and depicting the contributions of nonfossil (green color) and fossil (black color) organic carbon (i.e., pie charts), (b) Cumulative concentration levels of isoprene- and monoterpene-SOA tracers in each TSP sample collected over Gosan.

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# 273 **3.2. Temporal and seasonal variability of sugars**

274 The temporal/seasonal trends of sugar compounds over the KCOG provide useful 275 information on the emission strengths of various sources in the East Asian outflow. All three 276 anhydrosugars showed similar temporal and seasonal trends with higher concentrations in 277 winter and fall than spring and summer (Figures 2a and S2). As levoglucosan and other two 278 anhydrosugars (mannosan and galactosan) the pyrolysis products of are 279 cellulose/hemicellulose, their higher concentrations along with an increase in nonfossil

280 fraction of OC (Figure 2a; pie charts) in TSP from winter and fall revealed the impact of BB 281 emissions. The MODIS satellite-based fire counts (Figure S1) together with cluster analysis 282 in winter and fall (Figure 1) have revealed an influence of active BB emissions in the North 283 China Plain, Mongolia, and the Russian Far East. The temporal trends of glucose, fructose, 284 and sucrose exhibited less variability throughout the sampling period; however, we observed 285 a slight increase in their concentration towards spring/summer (Figure 2a). Glucose and 286 fructose have origins from leaf fragments and pollen species (Fu et al., 2012a). Sucrose is a 287 potential tracer for airborne pollen (Fu et al., 2012a) and late spring/early summer is often 288 regarded as a season of "pollen-allergies". Therefore, the similar temporal trends of glucose and fructose with sucrose indicate their common source, pollens (Figure 2a). Since glucose, 289 fructose, and sucrose showed moderately significant correlations ( $R^2 = 0.44-0.48$ , p<0.01) 290 291 with levoglucosan in winter, it is somewhat possible that BB source emission could also 292 influence the concentrations of these saccharides in this season (Hague et al., 2019; Fu et al., 293 2008).

294 BB also contributes to xylose and, hence, the temporal variability of xylose is 295 mimicking that of the anhydrosugars. Trehalose is a primary sugar and a useful tracer for 296 organic carbon associated with soil dust particles (Fu et al., 2012a). The temporal variability 297 of trehalose closely resembled that of the fungal spore tracers (arabitol, mannitol, and 298 erythritol), showing high concentrations in the spring, summer, and fall seasons (Figures 2a 299 and S2) (Zhu et al., 2015a; Fu et al., 2012a). The KCOG is under the influence of a large-300 scale advection of mineral dust from East Asia to the WNP during these three seasons (Tyagi et al., 2017; Huebert et al., 2003). The mineral dust transport from East Asia to the WNP can 301 be traced by the high concentrations of non-sea-salt Ca<sup>2+</sup> in the TSP samples from Gosan 302 (Arimoto et al., 1996). Similar temporal trends of trehalose and nss-Ca<sup>2+</sup>, particularly in 303 304 spring samples (Figure S3), suggest that the abundance of OA specific to fungal spores over 305 Gosan is likely associated with the Kosa (Asian dust) events.

The major sources of arabitol and mannitol are airborne fungal spores (Bauer et al., 306 307 2008), accompanying detritus from mature leaves (Pashynska et al., 2002). Heald and 308 Spracklen (2009) reported that mannitol and arabitol are considerably associated with 309 terrestrial biosphere activity. Inositol is largely derived from the developing leaves in summer 310 (Pashynska et al., 2002) and BB in winter (Fu et al., 2010b). Zhu et al. (2015b) found a 311 similar seasonal behavior of inositol with those of other sugar alcohols with the 312 predominance in summer, associated with microbial activities in local forests from Okinawa. Inositol showed a moderately significant correlation with levoglucosan ( $R^2 = 0.33$ , p< 0.01) 313

in winter; however, there were no positive linear relationships between levoglucosan and
other sugar alcohols, implying a partial emission of inositol from the BB during winter in
Gosan aerosols. Therefore, the temporal variability of inositol differs from that of the other
sugar alcohols (Figure S2). The sources of sugar compounds are further discussed in section
3.4.

319 The seasonally averaged concentrations of all the anhydrosugars and xylose are 320 higher in winter/fall than spring/summer (Figure 3a-d), possibly due to a greater influence of 321 long-range transport from East Asia. In contrast, glucose, fructose, and sucrose peaked in 322 spring but decreased in the other seasons (Figure 3e-g), mainly because of the contribution of 323 airborne pollen. Trehalose showed higher concentrations in fall and summer, followed by 324 spring and winter (Figure 3h). Arabitol, mannitol, and erythritol showed higher 325 concentrations in summer/fall than in winter and spring (Figure 3i-k). This seasonal trend is 326 consistent with those of soil-derived fungal spores. This feature is consistent with earlier 327 observations from a remote oceanic island in the WNP (Okinawa) during the impact of East Asian outflow (Zhu et al., 2015a). The seasonally averaged mass concentrations of inositol 328 329 are highest in spring, followed by summer, fall, and winter (Figure 31). Overall, the molecular 330 compositions of anhydrosugars showed the predominance of levoglucosan followed by 331 galactosan and mannosan (Figure S4). Galactosan is more abundant in crop-residue burning 332 emissions than mannosan (Engling et al., 2009; Sheesley et al., 2003). It is very much likely 333 that the impact of crop-residue burning emissions in East Asia over Gosan is more prominent 334 in winter/spring. Such high abundances of galactosan over mannosan were found in the North 335 China Plain (Fu et al., 2008) and in the Indo-Gangetic Plain outflow sampled over the Bay of 336 Bengal (Bikkina et al., 2019). Although the temporal variability of primary sugars in the TSP 337 samples from Gosan showed a characteristic peak of glucose and sucrose (Figure 2a), the 338 seasonally averaged distributions are different (Figure S4). The molecular distributions of 339 sugar alcohols are characterized by high loadings of arabitol and mannitol, followed by 340 erythritol and inositol (Figure S4).

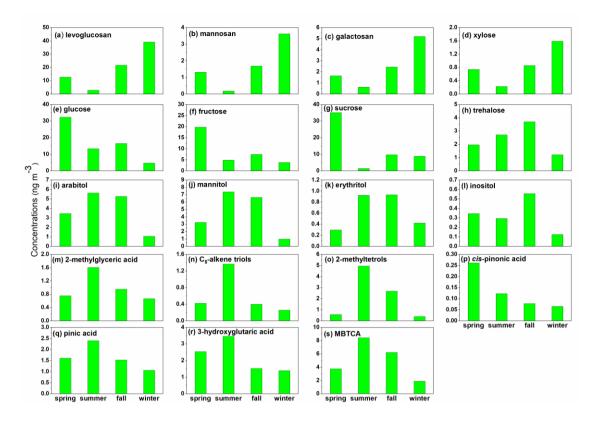


Figure 3. Seasonal variability of the atmospheric levels of sugar compounds and BSOA
 tracers in TSP samples from Gosan during April 2013-April 2014.

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# 346 **3.3. Temporal and seasonal variability of BSOA tracers**

We identified six isoprene-SOA tracers such as 2-methylglyceric acid (2-MGA), three  $C_5$ -347 348 alkene triols, and two 2-methyltetrols (2-methylthreitol and 2-methylerythritol) (2-MTs) in Gosan aerosol samples (Table 1). The sum of the isoprene-SOA tracers ranged from 0.35 to 349 15.1 ng m<sup>-3</sup> (avg. 3.69 ng m<sup>-3</sup>) with the predominance of 2-MTs (avg. 2.09 ng m<sup>-3</sup>). 2-MGA is 350 the second most abundant isoprene-SOA tracer (avg. 0.99 ng m<sup>-3</sup>), a high-generation product 351 352 probably formed by further photooxidation of methacrolein and methacrylic acid. A similar 353 molecular composition was observed over the North Pacific and California Coast (Fu et al., 354 2011). All the isoprene-SOA tracers exhibited similar temporal variations with higher 355 concentrations in the summer/spring months compared to autumn and winter (Figure 2b). Conversely, four monoterpene-SOA tracers, i.e., cis-pinonic acid, pinic acid, 3-356 hydroxyglutaric acid (3-HGA), and 3-methyl-1,2,3-butanetricarboxylic acid (MBTCA), were 357 358 detected in the Gosan samples (Table 1). The total concentrations of monoterpene-SOA tracers were found to be 1.65 to 35.5 ng m<sup>-3</sup> (avg. 9.24 ng m<sup>-3</sup>) with a high concentration of 359 MBTCA (avg. 5.11 ng  $m^{-3}$ ). All the monoterpene-SOA tracers showed similar temporal 360 trends with high values in summer/spring periods than autumn/winter (Figure 2b). 361

Nevertheless, *cis*-pinonic acid was ascribed a somewhat different temporal variability with other monoterpene-SOA tracers. It is likely that *cis*-pinonic acid might be further photooxidized to form MBTCA (Szmigielski et al., 2007).

365 The seasonal average of isoprene-SOA tracers showed high concentrations in summer, followed by spring/fall and winter (Figure 3m-o). One key feature of the data 366 367 presented here is that two fall samples (KOS984; 2-12 September and KOS986; 27 368 September to 07 October) exhibited high concentrations for 2-MTs over Gosan (Figure 2b). 369 We presumed that local vegetation might contribute significantly to the formation of 2-MTs 370 as it is a first-generation product. Moreover, 2-MTs can be derived from the open ocean 371 under low NO<sub>x</sub> conditions (Hu et al., 2013). 3-HGA and pinic acid showed somewhat higher 372 concentrations in summer/spring than fall/winter due to the growing vegetation (Figure 3q-r). 373 Cis-pinonic acid was more abundant in spring compared to summer (Figure 3p) because of its 374 photo-degradation, as discussed above. In contrast, MBTCA was dominant in summer/fall 375 than spring/winter (Figure 3s). Here, the formation of MBTCA could be enhanced in fall 376 during atmospheric transport from East Asia. The molecular distributions of isoprene-SOA 377 tracers were characterized by a high loading of 2-MTs, followed by 2-MGA and C<sub>5</sub>-alkene 378 triols in all seasons (Figure S4). The molecular composition of monoterpene-SOA tracers was 379 dominated by MBTCA, followed by 3-HGA, pinic acid, and cis-pinonic acid in all seasons 380 (Figure S4). Overall, BSOA tracers were found to be most abundant in summer, followed by 381 fall, spring, and winter (Table 1). Interestingly, it is likely that secondary OA undergoes 382 much faster cycling than the primary sugar compounds, considering the feasibility of 383 photooxidation. This would mean a slight underestimation of BSOA over the KCOG in the 384 East Asian outflow and, hence, their atmospheric abundances over Gosan reflect a lower 385 limit.

386 Kang et al. (2018b) reported that monoterpene-SOA tracers were more abundant than 387 isoprene-SOA tracers in spring-summer over the East China Sea, which is consistent with this 388 study. Although the mass budget calculations showed that isoprene and monoterpenes are 389 largely emitted by terrestrial plants; however, the open ocean can also contribute to isoprene 390 and monoterpenes significantly (Conte et al., 2020; Shaw et al., 2010; Broadgate et al., 1997). 391 Air mass back trajectory and cluster analysis (Figure 1) implied that the air masses mostly 392 originated from the ocean during summer. This means the open ocean significantly 393 contributed to isoprene-SOA production. However, terrestrial sources from the continent also 394 substantially enhanced the formation of BSOA. For example, one sample (KOS979; 1-11

June 2013) during summer showed the highest loading of BSOA tracers (Figure 2b) when air
masses were transported from the continent (Zhang et al., 2016).

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407	° −0.25 −0.22 −(	0.22 0.3 0.11	-0.03 -0.26 0.91	0.95 Manit
409	0 -0.11 -0	0.01 -0.07 -0.21	-0.3 -0.09 0.59	0.57 0.59 Erythr
410	;	0.22 0.59 0.45	0.33 -0.23 0.76	0.72 0.82 0.34 Inos

411 Figure 4. Multiple linear regression (MLR) analysis of airborne sugar compounds in TSP412 collected over Gosan.

413

# 414 **3.4. Source apportionment- regression analysis and diagnostic ratios**

415 Anhydrosugars strongly correlated with xylose (Figure 4), suggesting their common source as 416 BB emission in East Asia. Fu et al. (2012a) analyzed pollen from different tree species (e.g., White birch, Chinese willow, Peking willow, Sugi, Hinoki), which are enriched in sucrose 417  $(182-37,300 \ \mu g \ g^{-1})$ , glucose  $(378-3601 \ \mu g \ g^{-1})$  and fructose  $(162-1813 \ \mu g \ g^{-1})$ . In our 418 419 samples, sucrose strongly correlated with glucose and fructose (Figure 4), suggesting their 420 origin from plant-derived airborne pollen. Likewise, a strong correlation was found between 421 arabitol and mannitol, indicating a mutual origin from a similar type of fungal spores (Zhu et 422 al., 2015a; Fu et al., 2012a; Bauer et al., 2008; Yttri et al., 2007). Bauer et al. (2008) ascribed

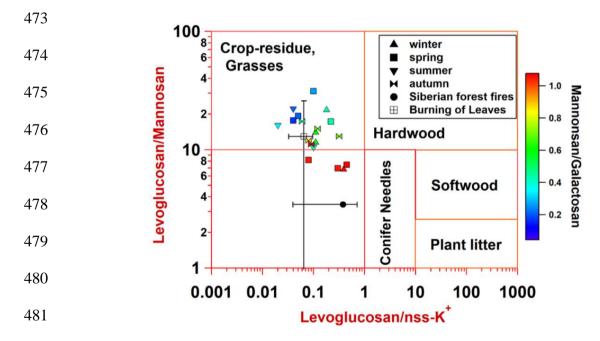
423 weak correlations between arabitol and mannitol to the diverse nature of ambient fungal 424 spores. Furthermore, both sugar alcohols correlated well with trehalose, a tracer for soil 425 organic carbon (Fu et al., 2012a). This observation suggests their common origin from soil 426 organic matter associated with fungal spores. Erythritol also originates from fungal spores; 427 however, its abundance is affected by the atmospheric aging process. Kessler et al. (2010) 428 reported that erythritol is degraded during long-range transport in 12.7 days. Consequently, 429 arabitol and mannitol were moderately correlated with erythritol in the Gosan samples due to 430 the degradation of the latter sugar alcohol in the East Asian outflow.

431 The linear relationship of levoglucosan (Lev) with mannosan (Man), galactosan (Gal), 432 and  $nss-K^+$  provide useful information on the type of BB emissions (hardwood, softwood, or 433 crop-residue). Ratios of Lev/Man and Lev/(Man + Gal) can be useful to distinguish BB and 434 coal combustion contributions. The average ratios of Lev/Man (15.1  $\pm$  6.76) and Lev/(Man + 435 Gal) (4.27  $\pm$  1.23) in Gosan aerosols are much closer to those from wood burning and coal 436 combustion sources emissions, respectively (Yan et al., 2018). It reveals that Lev could 437 originate from both biomass and coal burning source emissions, which is consistent with the 438 linear relationship between Lev-C and the fossil-/nonfossil carbon fraction (section 3.6). 439 Furthermore, different types of biomass are characterized by distinct Lev/Man ratios. For 440 instance, Lev/Man ratios from the softwood burning emissions (3-10) differ from those of 441 hardwood (15-25) and crop-residues (>40) (Singh et al., 2017; Schmidl et al., 2008a, b; Fu et 442 al., 2008; Engling et al., 2006, 2009; Fine et al., 2001, 2004). We found that the Lev/Man 443 ratios (Table 2) over the KCOG overlap between seasons and are somewhat close to that of 444 hardwood burning emissions in northern China, Mongolia, and the Russian Far East, as 445 corroborated by the backward air mass trajectories and MODIS fire counts (Figures 1 and 446 S1). Besides,  $Lev/K^+$  and *Man/Gal* ratios in summer differ from those of other seasons (Table 447 2). Cheng et al. (2013) have apportioned qualitatively the source contributions of 448 anhydrosugars over a receptor site based on the comparison of Lev/K<sup>+</sup> and Lev/Man ratios in 449 aerosols to those from various sources profiles compiled from the literature. This approach of 450 using the mass ratios of Lev/Man and  $Lev/K^+$  has been proven useful for deciphering the 451 difference in BB-derived OA (Bikkina et al., 2019).

Here, we adopted this methodology to ascertain the likely contributing sources of anhydrosugars, which are BB tracers from different seasons (Figure 5). This source apportionment relies on the fact that the  $Lev/K^+$  ratio from the softwood burning (10-1000) is higher than from hardwood (1-100) (Fine et al., 2004). In contrast, Lev/Man ratios for softwood are lower than those of hardwood burning (10-100) (Fine et al., 2004). Likewise,

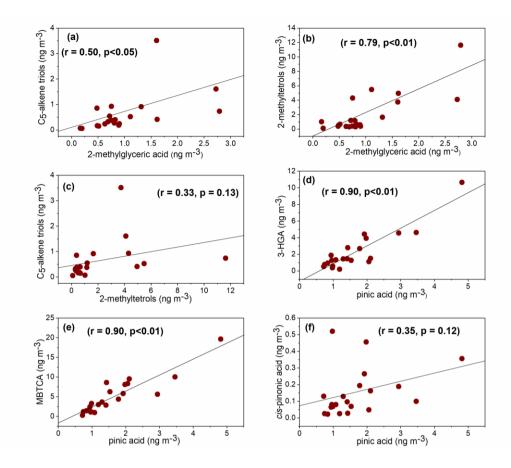
the Lev/Man and Lev/K<sup>+</sup> ratios from grasses and crop-residues are 10-100 and 0.01-1.0, 457 respectively (Bikkina et al., 2019). On a similar note, the  $Lev/K^+$  ratios from the burning of 458 459 pine needles (0.1-1.0) somewhat overlap with those from hardwood burning emissions but 460 are characterized by distinct Lev/Man ratios (Bikkina et al., 2019). The burning of dead 461 leaves (duff) showed higher  $Lev/K^+$  ratios than those of pine needles and grasses, but their Lev/Man ratio is on the lower side than for the former biomass type and the softwood burning 462 463 emissions. However, Lev is more susceptible to degradation by photooxidation with OH radicals during atmospheric transport (half-life: <2.2 days) (Hennigan et al., 2010) and this 464 465 would cause lower abundances of this anhydrosugar. Hence, the hardwood  $Lev/K^+$  ratios could slightly shift downwards. Therefore, caution is required while interpreting the ambient 466 data from a receptor site (Bikkina et al., 2019). Overlapping the seasonal data on this scatter 467 468 plot of *Lev*/K<sup>+</sup> versus *Lev/Man* (Figure 5) clearly revealed a mixed contribution of burning of 469 hardwood and crop-residue in the East Asian outflow. It should be noted that the 470 photooxidation process during atmospheric transport is also applicable for low concentrations and poor correlations of other primary saccharides. 471





482 Figure 5. Scatter plot of levoglucosan/K<sup>+</sup> versus levoglucosan/mannosan ratios in TSP
483 collected over Gosan during April 2013-April 2014. The data for Siberian forest fires and
484 burning leaves were adopted from Sullivan et al. (2008).

486 The correlation coefficients and diagnostic ratios of BSOA tracers specify their source origin or formation pathway. Nevertheless, atmospheric stability or reactivity of BSOA 487 488 tracers through the photooxidation during long-range transport may bias the correlation 489 coefficients in Gosan aerosols. 2-MGA showed a significant correlation with 2-MTs (r =490 0.79, p<0.01) and C<sub>5</sub>-alkene triols (r = 0.50, p<0.05) (Figure 6a-b), suggesting their similar 491 formation pathway or common sources of isoprene-SOA tracers. However, a poor correlation 492 coefficient between 2-MTs and C<sub>5</sub>-alkene triols (r = 0.33, p = 0.13) (Figure 6c) indicates their 493 different formation process over the Gosan atmosphere. Wang et al. (2005) documented that 494 polyols are formed from isoprene through diepoxy derivatives, which further convert into 2-495 MTs by acid-catalyzed hydrolysis. On the other hand, C<sub>5</sub>-alkene triols are produced from the 496 precursor of hydroxyperoxy radicals that are initially derived from isoprene through 497 rearrangement reactions (Surratt et al., 2006). It can be noted that the formation mechanisms 498 of 2-MGA and 2-MTs are different while depending on the NO<sub>x</sub> concentrations (Surratt et al., 499 2010). Thus, the ratio of 2-MGA/2-MTs attributes to the influence of NO<sub>x</sub> on isoprene-SOA 500 formation. We found a low ratio of 2-MGA/2-MTs (0.67) (Figure S5a, Table 2) in summer, 501 implying enhancement of 2-MTs formation over the open ocean due to the low-NO<sub>x</sub> 502 environment in the ocean atmosphere. On the contrary, the 2-MGA/2-MTs ratios for other 503 seasons were >1.0 (Figure S5a, Table 2), indicating an elevated formation of 2-MGA through 504 continental high NO<sub>x</sub> condition, which is consistent with the air masses back trajectory.



507 **Figure 6.** Pearson linear correlation coefficient analysis of BSOA tracers in Gosan TSP 508 aerosols during April 2013-April 2014.

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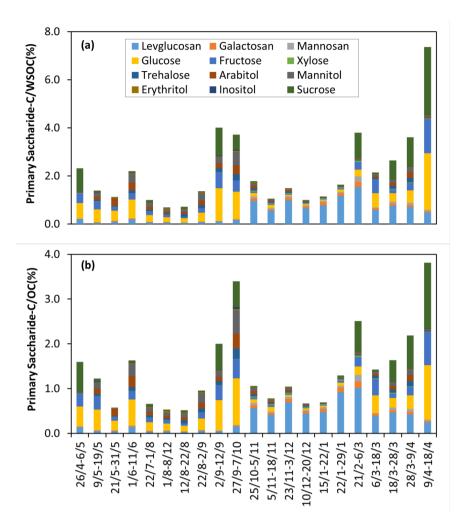
510 *Cis*-pinonic acid showed a weak correlation with pinic acid (r = 0.35, p = 0.12) 511 (Figure 6f), suggesting that different atmospheric reactivity of cis-pinonic/pinic acids during 512 transport would cause such poor correlation. In contrast, pinic acid exhibited a strong positive 513 linear correlation with 3-HGA (r = 0.90, p<0.01) and MBTCA (r 0.90, p<0.01) (Figure 6d-e), 514 implying their similar sources. It should be noted that the formation processes of pinic acid, 515 3-HGA, and MBTCA are different because pinic acid is a first-generation product, and 3-516 HGA and MBTCA are high-generation products (Claeys et al., 2007; Müller et al., 2012; 517 Szmigielski et al., 2007). The ratio of *cis*-pinonic acid + pinic acid to MBTCA (P/M) is used 518 to evaluate the aging of monoterpene-SOA.

519 A low P/M ratio suggests the transformation of *cis*-pinonic and pinic acids to 520 MBTCA and thus relatively aged monoterpene-SOA, whereas a high ratio reflects relatively 521 fresh monoterpene-SOA (Gómez-González et al., 2012; Ding et al., 2014). Gómez-González 522 et al. (2012) reported aged monoterpene-SOA (P/M = 0.84) from a Belgian forest site while 523 fresh chamber-produced  $\alpha$ -pinene-SOA tracers showed P/M ratios of 1.51 to 3.21 (Offenberg et al., 2007). The average ratio of P/M in this study showed 0.62 with the low value in summer (Figure S5b, Table 2), which is lower than those of Guangzhou (fresh monoterpene-SOA; 28.9) while the air masses originated from southern China (Ding et al., 2014). This observation indicates that monoterpenes SOA have undergone substantial aging during transport to Gosan, particularly in summer when extensive photochemical oxidation occurred due to the high temperature and intense solar radiation.

## 530 **3.5. Relative abundances in WSOC and OC**

531 Levoglucosan is the most abundant anhydrosugar, contributing 0.05-1.54% of WSOC and 532 0.03-1.02% of OC. Likewise, sucrose, glucose, and fructose were more abundant primary 533 sugars, contributing 0.01-2.83%, 0.03-1.22%, and 0.02-0.74% of WSOC, respectively. 534 Contributions of the three primary sugars varied from 0.05% to 3.41% of OC. Arabitol and 535 mannitol are the most abundant sugar alcohols, whose contribution to WSOC and OC ranged 536 from 0.02% to 0.93% and 0.01% to 0.85%, respectively. Figure 7 depicts the contribution of 537 sugar compounds to WSOC and OC in TSP collected over Gosan during the study period. 538 We also compared the atmospheric abundances of sugar compounds from Gosan with the 539 literature data (Table 3). This comparison revealed the less influence of BB tracer compounds 540 (i.e., anhydrosugar levels) over Gosan, a factor of 5-10 times lower than those reported for 541 the BB-influenced source regions in China and East Asia (Fu et al., 2012b; Kang et al., 542 2018a; Wang and Kawamura, 2005; Wang et al., 2012). However, the levels of anhydrosugar 543 over Gosan are higher than those observed over the remote Canadian High Arctic site (Fu et 544 al., 2009a). In contrast, Gosan is characterized by high concentrations of primary sugars 545 compared to other remote sampling sites in Table 3. This is because of the overwhelming 546 contribution of primary sugars associated with soil dust particles over Gosan during the East 547 Asian outflow. Such high loadings of primary sugars were observed from other remote island 548 receptor sites in the WNP (Okinawa) during the spring season (Zhu et al., 2015a). Likewise, 549 the concentrations of sugar alcohols from Gosan are similar to those from other receptor sites 550 influenced by the East Asian outflow (Verma et al., 2018; Zhu et al., 2015a).

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**Figure 7.** Contribution of primary saccharide-C in (a) WSOC(%) and (b) OC(%) in TSP collected over Gosan during April 2013-April 2014.

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559 The Contributions of isoprene-SOA tracers to ambient OC (0.01-0.40%, avg. 0.09%) 560 and WSOC (0.02-0.57%, 0.13%) were lower than those of monoterpene-derived SOA (0.04-561 0.63%, 0.23% for OC and 0.06-0.82%, 0.33% for WSOC). The contributions of isoprene 562 oxidation products to OC and WSOC were found to be highest in summer (0.23% and 0.32%, 563 respectively), followed by fall (0.09% and 0.13%), spring (0.04% and 0.06%), and winter 564 (0.02% and 0.03%). Likewise, the contributions of monoterpene-SOA products to aerosol OC 565 and WSOC exhibited the highest value in summer (0.40% and 0.55%, respectively), followed 566 by fall (0.24% and 0.35%), spring (0.18% and 0.27%), and winter (0.10% and 0.14%). We 567 found that the contribution of BSOA products to the carbonaceous components was twice in summer (Figure 8, Table 2). This means BSOA formation occurred in summer to a greater 568 569 extent due to the intensive BVOCs emission with key factors of meteorological parameters 570 (higher temperature and radiation), i.e., higher concentrations of ozone, and other oxidizing 571 agents (NO<sub>x</sub>, OH, etc.). It should be pointed out that the fraction of WSOC in OC (WSOC/OC) is often prone to photochemical aging and, hence, contributes to SOA. More 572 573 specifically, the WSOC/OC > 0.4 over a receptor site indicates aged aerosols with a 574 significant SOA contribution (Haque et al., 2019; Boreddy et al., 2018). In our study, the 575 average WSOC/OC ratio was observed 0.68 with the highest in summer (0.72), although 576 subtle difference exhibited in winter/autumn (> 0.68), implying the presence of aged OA over 577 the KCOG. Such higher abundances of SOA over Gosan during transport from East Asia is 578 mainly due to photochemical aging of anthropogenic (fossil fuel/biomass combustion) 579 emissions. Huang et al. (2014) reported significant SOA formation from the fossil 580 fuel/biomass combustion precursor VOCs in winter over China.

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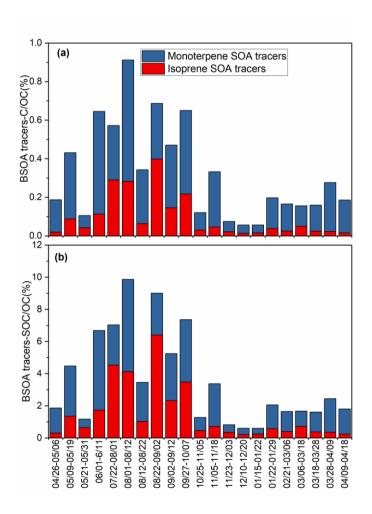


Figure 8. Contribution of (a) isoprene- and monoterpene-SOA tracers-C in ambient OC(%)
and (b) isoprene- and monoterpene-SOA tracers-SOC in ambient OC(%) in Gosan aerosol
samples during April 2013-April 2014.

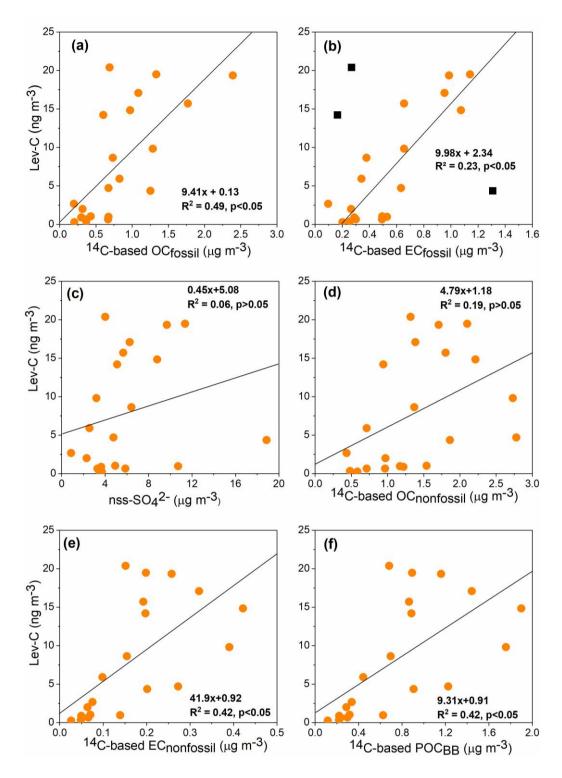
587 We estimated secondary organic carbon (SOC) derived from isoprene and monoterpenes, using the measured values of BSOA tracers and following the SOA tracer-588 589 based method first proposed by Kleindienst et al. (2007). A summary of the estimated SOC is 590 provided in Table 2. The contribution of isoprene to SOC was calculated 2.26 to 97.4 ngC m<sup>-</sup> <sup>3</sup> (avg. 23.7 ngC m<sup>-3</sup>), accounting for 1.45% of OC and 35.5% of total BSOC with the 591 predominance in summer (3.56% and 46.6% for OC and SOC, respectively). The estimation 592 of monoterpene-SOA to SOC (avg. 40.1 ngC  $m^{-3}$ ) was observed around two times higher than 593 594 that of isoprene-SOA (Table 2). Interestingly, the contribution of monoterpene-derived SOC 595 to ambient OC (3.65%) was dominant in summer, but monoterpene SOC to total SOC 596 (72.1%) was most abundant in spring (Figure 8b). The seasonal distributions of biogenic 597 SOC (Table 2) implied that a substantial amount of SOC was formed from monoterpenes in 598 spring. The estimated biogenic SOC at the KCOG is almost one order of magnitude lower 599 than those from other continental sites in China urban areas (e.g., Pearl River Delta: 446 ngC m<sup>-3</sup>) (Ding et al., 2012). However, the estimated biogenic SOC load from the KCOG is much 600 601 higher than that reported from a remote site in the Canadian High Arctic (Alert: 9.4 ngC m<sup>-3</sup>) 602 (Fu et al., 2009b) and comparable with over the East China Sea (Kang et al., 2018b).

603 It should be noted that concentrations of SOA tracers cannot always provide the 604 actual contribution of the biogenic source to ambient organic aerosol mass. For example, loadings of monoterpene-SOA tracers were lower in sample KOS999 (28 March - 9 April 605 2014) (3.02 ng m<sup>-3</sup>) compared to sample KOS1000 (9 - 18 April 2014) (14.4 ng m<sup>-3</sup>), whereas 606 607 the estimated contribution of SOC to ambient OC showed an opposite trend (KOS999: 608 2.08%; KOS1000: 1.56%). This result demonstrates that the estimation of SOC is an 609 important factor in evaluating the contribution of BSOA to organic aerosol mass. We calculated biogenic OC using radiocarbon (<sup>14</sup>C) data following the method proposed by 610 Szidat et al. (2006). Biogenic OC showed a poor correlation with biogenic SOC (r = 0.36, p =611 612 0.09) but a significant linear relationship with primary sugars (i.e., glucose, fructose and 613 sucrose) (r = 0.54, p<0.5), suggesting that primary bioaerosols from plant-derived airborne 614 pollen dictate on biogenic OC over Gosan.

# 615 **3.6. Significance of fossil fuel as a source for levoglucosan**

The ambient *Lev* levels showed a significant linear relationship with the  $OC_{fossil}$ , suggesting the fossil source contribution of this molecular marker (Figure 9a). However, such a significant correlation was not evident between  $OC_{fossil}$  and other major sugar compounds. Until recently, *Lev* has been thought to originate primarily from the hemicellulose/cellulose pyrolysis of vegetation and, hence, can be employed as a powerful tracer for biomass smoke

- particles (Fraser and Lakshmanan, 2000; Simoneit et al., 1999). Nevertheless, residential coals (*e.g.*, lignite and bituminous coal) have been shown to contain high concentrations of '*Lev*' but also emit traces of *Man and Gal* (Kourtchev et al., 2011; Fabbri et al., 2008). Recently, Yan et al. (2018) found a significant linear relationship between the <sup>14</sup>C-based fossil fraction of WSOC and *Lev-C* in the aerosols generated from coal combustion and the ambient aerosol samples. Therefore, the prevailing linear relationship between  $OC_{fossil}$  and *Lev-C* in the Gosan samples (Figure 9a) is likely due to a common source contribution from
- 628 coal combustion in East Asia.



629

**Figure 9.** Linear regression analysis between levoglucosan in terms of their carbon content (*Lev-C*) and <sup>14</sup>C-based mass concentrations of (a) organic carbon and (b) elemental carbon of fossil origin (OC<sub>fossil</sub> and EC<sub>fossil</sub>, respectively), (c) nss-SO<sub>4</sub><sup>2-</sup>, (d) nonfossil derived organic carbon (OC<sub>nonfossil</sub>), (e) nonfossil derived elemental carbon (EC<sub>nonfossil</sub>), and (f) biomass burning derived primary OC (POC<sub>BB</sub>) in TSP collected over Gosan during April 2013-April 2014. In panel (b), the squares represent three outliers (i.e., samples with rather high and low *Lev*/EC ratios; please see text for more details).

The slope of the linear regression between Lev-C and  $OC_{fossil}$  (0.0094; Figure 9a) is 637 638 higher than those documented for the coal combustion source in China (~0.004±0.007) (Yan 639 et al., 2018). Moreover, Lev-C moderately correlated with the EC<sub>fossil</sub> with the regression 640 slope (~0.01) in the Gosan samples (Figure 9b), being comparable to that observed for the 641 coal combustion in China (0.044±0.076) (Yan et al., 2018). It should be noted that excluding 642 the three outliers as shown in Figure 9b (black square), Lev-C showed a stronger correlation with  $EC_{fossil}$  ( $R^2 = 0.74$ , p < 0.05). Of these, two outliers in winter (KOS995: 22-29 January 643 2014; KOS996: 21 February - 3 March 2014) have higher Lev-C levels over that of EC<sub>fossil</sub>, 644 645 when air mass trajectories showed the impact of BB emissions in the North China Plain. In contrast, the third outlier in summer (KOS979: 1-11 June 2013) has a lower Lev-C/EC<sub>fossil</sub>, 646 647 while air parcels transported from nearby cities in China, Korea, and Japan, thus, have more 648 contribution from vehicular emissions. Overall, both the regression slopes are, thus, the 649 representative nature of Lev-C/OC<sub>fossil</sub> and Lev-C/EC<sub>fossil</sub> in the East Asian outflow. Lev-C and  $nss-SO_4^{2-}$  exhibited a poor correlation (Figure 9c), although both were transported from East 650 651 Asia.

Lev-C exhibited a rather weak correlation with  $OC_{nonfossil}$  (R<sup>2</sup> = 0.19) than that with 652  $EC_{nonfossil}$  ( $R^2 = 0.42$ ) over Gosan during the study period (Figure 9d-e). This could be likely 653 654 because OC<sub>nonfossil</sub> has contributions from the BB and the secondary formation process or the primary biogenic sources. The contribution of primary OC generated from BB (POC<sub>BB</sub>) to 655 OC<sub>nonfossil</sub> was taken from Zhang et al. (2016). In their study, the <sup>14</sup>C-based EC<sub>nonfossil</sub> levels 656 were scaled by a factor to constrain the  $POC_{BB}$  (Zhang et al., 2016). Here the conversion 657 658 factor is '4.5' (range: 3-10), which is a median value representing the primary OC/EC ratio 659 from BB emissions ((POC/EC)<sub>BB</sub>).

660

 $POC_{BB} = EC_{nonfossil} \times (POC/EC)_{BB}$ 

(7)

661 Lev-C showed a somewhat improved linear correlation with POC<sub>BB</sub> than with the OC<sub>nonfossil</sub> (Figure 9f). It is apparent from Figure 9 that the regression slopes are comparable, indicating 662 663 their contribution to Lev from both coal combustion and BB emissions over Gosan. The prevailing weak linear relationship (moderate correlation) of Lev with nonfossil and fossil 664 665 carbon fractions is likely the result of photo-degradation of Lev during atmospheric transport. 666 This result would mean that the higher atmospheric abundance of Lev and its pronounced linear relationships with the nonfossil and fossil carbon fractions implies a much stronger 667 668 impact of both source emissions in East Asia during the continental outflow in winter and 669 spring.

670 Overall, we present a new finding on the contribution of coal combustion sources in East 671 Asia in controlling the atmospheric levels of Lev apart from the traditional biomass/biofuel 672 burning emissions. This is based on the prevailing linear relationship between the 673 radiocarbon based nonfossil-EC and Lev in the year-round TSP samples collected from the 674 KCOG site in Jeju Island. The Gosan supersite is the best location to understand how the 675 chemical composition of source-emissions from East Asia affects the outflow regions in 676 winter and spring. Recent studies have highlighted the potential contribution of Lev from 677 residential coal combustion in China (Yan et al., 2018), with estimated annual emission of 678 ~2.2 Gg of Lev from domestic coal combustion (Wu et al., 2021). Given this background 679 information, the prevailing significant linear relationship between Lev and nonfossil-EC (p-680 value < 0.05) over the KCOG clearly emphasizes the need for reconsideration of the previous 681 assessments on the impact of BB in East Asian outflow to the WNP. Additionally, this 682 dataset is further compared with the molecular distributions and relative abundances of 683 organic tracers in the TSP samples collected over Gosan during 2001, a decade ago (Fu et al., 684 2012a). This comparison allows us to better understand the regional changes in the emission 685 sources (e.g., fugitive dust, BB, and fossil-fuel combustion) on a decadal basis.

## 686 **4. Conclusions**

687 We investigated seasonal variations of primary organic components such as anhydrosugars, 688 primary sugars, and sugar alcohols and biogenic secondary organic aerosol (BSOA) tracers 689 (isoprene- and monoterpene-derived SOA products) in ambient aerosols from Gosan, Jeju 690 Island. Among the detected sugar compounds, levoglucosan was dominant in winter/fall, 691 whereas glucose and sucrose were more abundant in spring/summer. The seasonal trends 692 documented that BB impact is more significant in winter/fall and the primary bioaerosol 693 particles are important in spring/summer. Diagnostic ratios of levoglucosan, galactosan, and 694 mannosan reflect that emissions from BB are mostly dominated by hardwood. The significant 695 linear relationship of sucrose with glucose and fructose suggests their origin from airborne 696 pollen. On a similar note, trehalose showed a significant positive correlation with arabitol, 697 mannitol, and erythritol, implying their contribution from airborne fungal spores and soil 698 microbes over the KCOG.

Distributions of biogenic SOA tracers were characterized by a predominance of monoterpene- than isoprene-derived oxidation products in Gosan aerosols. The BSOA tracers were formed in summer to a greater extent, followed by fall/spring and then winter. The low ratio of *cis*-pinonic acid + pinic acid to MBTCA (P/M) demonstrated that monoterpene-SOA 703 was relatively aged over Gosan aerosols. The estimated secondary organic carbon (SOC) 704 with the predominance in summer shows that substantial BSOA formation occurred in 705 summer due to favorable meteorological conditions. The backward air mass trajectories and 706 source apportionment studies entirely demonstrated that emission from East Asia 707 significantly dominates the ambient OA mass over KCOG. Interestingly, evoglucosan-C708 exhibited a significant positive correlation with nonfossil and fossil organic carbon fractions, 709 along with the comparable regression slopes. This result reveals that BB and coal (lignite) 710 combustion both are prominent sources for levoglucosan in the East Asian outflow.

Although there is some evidence that levoglucosan could originate from the combustion of brown coals (*e.g.*, lignite) in China, our observations from the KCOG (receptor site) also hinted at the fossil source contribution of this molecular marker in the regional influx of the East Asian outflow. Therefore, attribution of ambient levoglucosan levels over the WNP to the impact of BB emission may cause large uncertainty.

## 716 Data availability

717 The data used in this paper are available upon request from the corresponding author.

## 718 Author contributions

719 KK and YLZ designed the research. ML collected the aerosol samples. MMH and SB

performed the analysis of aerosol samples. MMH wrote the paper under the guidance of YLZ

and KK. All authors were actively involved in the discussion of the paper.

# 722 **Competing interests**

The authors declare that they have no conflict of interest.

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aerosol samples from Gosan.

Species	Annual	Summer	Fall	Winter	Spring
	$Avg.^{a} \pm S.D.^{b}$ Min. <sup>c</sup> , Max. <sup>d</sup>	Avg. ± S.D. Min., Max.			
Anhydrosugars					
Levoglucosan (Lev)	$17.6 \pm 16.8 \\ 0.60, 45.9$	$\begin{array}{c} 2.92 \pm 3.89 \\ 0.60, \ \ 9.81 \end{array}$	$21.7 \pm 19.0$ 2.30, 43.9	$39.2 \pm 6.60$ 32.0, 45.9	$12.7 \pm 11.6$ 1.45, 33.4
Mannosan (Man)	$1.57 \pm 1.82$ 0.05, 6.74	$0.18 \pm 0.24$ 0.05, 0.61	$1.69 \pm 1.49$ 0.13, 3.66	$3.63 \pm 2.28$ 1.47, 6.74	$1.31 \pm 1.57$ 0.08, 4.08
Galactosan (Gal)	$2.28 \pm 2.10$ 0.14, 6.78	$0.64 \pm 0.68$ 0.14, 1.82	$2.45 \pm 2.13$ 0.35, 4.92	$5.21 \pm 1.64$ 3.40, 6.78	$1.65 \pm 1.26$ 0.50, 3.88
Primary Sugars	,	,			,
Glucose	$18.8 \pm 27.1$ 2.45, 122	$13.4 \pm 18.2$ 2.45, 45.6	$16.5 \pm 15.7$ 2.68, 33.6	$4.74 \pm 3.14$ 2.88, 9.44	$32.4 \pm 41.1$ 4.87, 122
Fructose	$10.3 \pm 15.9$ 0.97, 74.0	$\begin{array}{l} 4.90 \pm 5.15 \\ 0.97, \ 13.7 \end{array}$	$7.48 \pm 6.99$ 1.71, 16.2	$3.82 \pm 4.45$ 1.56, 10.5	$19.8 \pm 25.0$ 2.69, 74.0
Sucrose	$16.1 \pm 32.2$ 0.26, 140	$1.46 \pm 1.05$ 0.68, 3.28	$9.74 \pm 12.1$ 0.76, 27.2	$8.87 \pm 16.3$ 0.42, 33.3	$35.1 \pm 50.5$ 0.26, 140
Trehalose	$\begin{array}{c} 2.42 \pm 1.97 \\ 0.65, \ \ 7.03 \end{array}$	$\begin{array}{c} 2.72 \pm 2.53 \\ 0.97, \ \ 6.98 \end{array}$	$3.71 \pm 2.29$ 1.18, 7.03	$\begin{array}{c} 1.21 \pm 0.42 \\ 0.72, \ 1.63 \end{array}$	$\begin{array}{c} 1.98 \pm 1.55 \\ 0.65, \ \ 5.33 \end{array}$
Xylose	$\begin{array}{c} 0.81 \pm 0.65 \\ 0.04, \ \ 2.03 \end{array}$	$\begin{array}{c} 0.23 \pm 0.23 \\ 0.04, \ 0.63 \end{array}$	$0.86 \pm 0.68 \\ 0.14, 1.70$	$\begin{array}{c} 1.59 \pm 0.37 \\ 1.21, \ \ 2.03 \end{array}$	$\begin{array}{c} 0.74 \pm 0.55 \\ 0.16, \ 1.68 \end{array}$
Sugar alcohols					
Arabitol	$3.96 \pm 4.24$ 0.47, 18.7	$5.64 \pm 7.46$ 1.20, 18.7	$5.27 \pm 3.51$ 2.27, 10.9	$\begin{array}{c} 1.06 \pm 0.60 \\ 0.47, \ \ 1.91 \end{array}$	$3.47 \pm 2.19$ 1.18, 6.30
Mannitol	$4.61 \pm 5.54$ 0.25, 22.0	$7.39 \pm 8.65$ 1.71, 22.0	$6.64 \pm 6.03$ 1.69, 16.7	$\begin{array}{c} 0.99 \pm 0.45 \\ 0.55, \ 1.60 \end{array}$	$3.24 \pm 2.67$ 0.25, 7.03
Erythritol	$\begin{array}{c} 0.62 \pm 0.43 \\ 0.12, \ 1.52 \end{array}$	$\begin{array}{c} 0.92 \pm 0.53 \\ 0.42, \ 1.52 \end{array}$	$\begin{array}{c} 0.93 \pm 0.33 \\ 0.55, \ 1.27 \end{array}$	$\begin{array}{c} 0.42 \pm 0.27 \\ 0.16, \ 0.80 \end{array}$	$\begin{array}{c} 0.30 \pm 0.12 \\ 0.12, \ \ 0.48 \end{array}$
Inositol	$\begin{array}{c} 0.34 \pm 0.39 \\ 0.04, \ 1.53 \end{array}$	$\begin{array}{c} 0.29 \pm 0.41 \\ 0.08, \ 1.03 \end{array}$	$\begin{array}{c} 0.56 \pm 0.60 \\ 0.10, \ \ 1.53 \end{array}$	$\begin{array}{c} 0.13 \pm 0.07 \\ 0.08, \ \ 0.23 \end{array}$	$\begin{array}{c} 0.35 \pm 0.28 \\ 0.04, \ \ 0.73 \end{array}$
Isoprene-SOA tracers					
2-MGA	$\begin{array}{c} 0.99 \pm 0.70 \\ 0.17, \ \ 2.79 \end{array}$	$1.61 \pm 1.17 \\ 0.17, \ 2.79$	$\begin{array}{c} 0.95 \pm 0.43 \\ 0.51, \ 1.61 \end{array}$	$\begin{array}{c} 0.67 \pm 0.14 \\ 0.49, \ \ 0.81 \end{array}$	$\begin{array}{c} 0.76 \pm 0.35 \\ 0.20, \ \ 1.32 \end{array}$
Σ2-MLTs	$1.04 \pm 1.40$ 0.05, 5.81	$\begin{array}{c} 2.48 \pm 1.98 \\ 0.51, \ \ 5.81 \end{array}$	$1.34 \pm 1.16$ 0.33, 2.74	$\begin{array}{c} 0.20 \pm 0.05 \\ 0.15, \ 0.26 \end{array}$	$\begin{array}{c} 0.29 \pm 0.25 \\ 0.05, \ 0.82 \end{array}$
$\Sigma$ C5-alkene triols	$\begin{array}{c} 0.20 \pm 0.25 \\ 0.02, \ \ 1.17 \end{array}$	$\begin{array}{c} 0.46 \pm 0.44 \\ 0.02,  1.17 \end{array}$	$\begin{array}{c} 0.13 \pm 0.05 \\ 0.05, \ \ 0.18 \end{array}$	$\begin{array}{c} 0.09 \pm 0.03 \\ 0.05, \ \ 0.13 \end{array}$	$\begin{array}{c} 0.14 \pm 0.11 \\ 0.02, \ \ 0.30 \end{array}$
Monoterpene-SOA trace		,			,
<i>cis</i> -pinonic acid	$\begin{array}{c} 0.15 \pm 0.14 \\ 0.02, \ \ 0.52 \end{array}$	$\begin{array}{c} 0.12 \pm 0.13 \\ 0.03, \ 0.36 \end{array}$	$\begin{array}{c} 0.08 \pm 0.06 \\ 0.02, \ \ 0.16 \end{array}$	$\begin{array}{c} 0.07 \pm 0.03 \\ 0.02, \ \ 0.10 \end{array}$	$\begin{array}{c} 0.26 \pm 0.17 \\ 0.08, \ \ 0.52 \end{array}$
pinic acid	$\begin{array}{c} 1.67 \pm 1.01 \\ 0.72, \ \ 4.81 \end{array}$	$\begin{array}{c} 2.40 \pm 1.67 \\ 0.99, \ \ 4.81 \end{array}$	$\begin{array}{c} 1.53 \pm 0.57 \\ 0.75, \ \ 2.11 \end{array}$	$\begin{array}{c} 1.07 \pm 0.25 \\ 0.84, \ 1.42 \end{array}$	$1.61 \pm 0.78$ 0.72, 2.94
3-HGA	$2.30 \pm 2.38$ 0.19, 10.6	$3.46 \pm 4.39$ 0.19, 10.6	$\begin{array}{c} 1.52 \pm 0.77 \\ 0.76, \ \ 2.79 \end{array}$	$\begin{array}{c} 1.40 \pm 0.39 \\ 0.94, \ 1.88 \end{array}$	$2.54 \pm 1.83$ 0.38, 4.57
MBTCA	$5.11 \pm 4.54$ 0.29, 19.6	$8.44 \pm 6.88$ 3.00, 19.6	$6.24 \pm 3.64$ 1.17, 9.49	$1.90 \pm 0.85$ 0.97, 2.83	$3.77 \pm 2.96$ 0.29, 8.08

<sup>a</sup>Average, <sup>b</sup>Standard deviation, <sup>c</sup>Minimum, <sup>d</sup>Maximum. 2-MGA: 2-methylglyceric acid, 2-MLTs: 2-methyltetrols, 3-HGA: 3-hydroxyglutaric acid, MBTCA: 3-methyl-1,2,3-butanetricarboxylic acid.

1126	Table 2. Statistical summary of diagnostic ratios and carbonaceous components contribution in Gosan
$1127 \\ 1128$	aerosols.

Species	Annual	Summer	Fall	Winter	Spring
	$Avg.^{a} \pm S.D.^{b}$ Min. <sup>c</sup> , Max. <sup>d</sup>		Avg. ± S.D. Min., Max.	Avg. ± S.D. Min., Max.	Avg. ± S.D. Min., Max.
Diagnostic ratios					
Lev/Man	$\begin{array}{c} 15.1 \pm 6.76 \\ 6.81,  31.3 \end{array}$	$\begin{array}{l} 17.1 \pm 8.21 \\ 8.58, \ \ 28.4 \end{array}$	$\begin{array}{c} 13.7 \pm 2.51 \\ 11.1, \ 17.3 \end{array}$	$\begin{array}{c} 13.5 \pm 6.24 \\ 6.81, \ \ 21.7 \end{array}$	$15.4 \pm 8.76$ 6.98, 31.3
Man/Gal	$\begin{array}{c} 0.55 \pm 0.32 \\ 0.10, \ 1.07 \end{array}$	$\begin{array}{c} 0.28 \pm 0.13 \\ 0.10, \ 0.42 \end{array}$	$\begin{array}{c} 0.64 \pm 0.20 \\ 0.37, \ 0.90 \end{array}$	$\begin{array}{c} 0.66 \pm 0.27 \\ 0.43, \ 1.05 \end{array}$	$\begin{array}{c} 0.61 \pm 0.43 \\ 0.16, \ 1.07 \end{array}$
Lev/(Man + Gal)	$\begin{array}{l} 4.27 \pm 1.23 \\ 2.18, \ \ 6.56 \end{array}$	$\begin{array}{c} 3.12 \pm 0.78 \\ 2.18, \ \ 4.03 \end{array}$	$\begin{array}{l} 5.08 \pm 0.32 \\ 4.71, \ \ 5.51 \end{array}$	$\begin{array}{l} 4.85 \pm 1.35 \\ 3.49, \ \ 6.56 \end{array}$	$\begin{array}{l} 4.17 \pm 1.33 \\ 2.50, \ \ 6.45 \end{array}$
$Lev/K^{\scriptscriptstyle +}\times 10^{\rm -2}$	$\begin{array}{c} 5.73 \pm 5.65 \\ 0.65, \ \ 23.2 \end{array}$	$\begin{array}{c} 1.00 \pm 0.52 \\ 0.65, \ 1.91 \end{array}$	$3.94 \pm 2.51$ 1.26, 6.30	$\begin{array}{c} 12.3 \pm 7.87 \\ 6.42, \ \ 23.2 \end{array}$	$6.65 \pm 4.48$ 1.77, 15.3
$Lev/OC \times 10^{-2}$	$\begin{array}{c} 0.73 \pm 0.66 \\ 0.08, \ \ 2.29 \end{array}$	$\begin{array}{c} 0.14 \pm 0.10 \\ 0.08, \ \ 0.31 \end{array}$	$\begin{array}{l} 0.84 \pm 0.61 \\ 0.12, \ 1.56 \end{array}$	$\begin{array}{c} 1.60 \pm 0.67 \\ 0.99, \ \ 2.29 \end{array}$	$\begin{array}{l} 0.56 \pm 0.40 \\ 0.11, \ 1.05 \end{array}$
Lev/WSOC $\times 10^{-2}$	$\begin{array}{c} 1.09 \pm 0.97 \\ 0.10,  3.46 \end{array}$	$\begin{array}{c} 0.20 \pm 0.13 \\ 0.10, \ 0.42 \end{array}$	$\begin{array}{c} 1.25 \pm 0.94 \\ 0.23, \ \ 2.21 \end{array}$	$\begin{array}{c} 2.33 \pm 0.90 \\ 1.48, \ \ 3.46 \end{array}$	$\begin{array}{c} 0.92 \pm 0.65 \\ 0.13, \ 1.69 \end{array}$
2-MGA/2-MLTs	$\begin{array}{c} 2.18 \pm 1.59 \\ 0.33, \ 5.40 \end{array}$	$\begin{array}{c} 0.67 \pm 0.43 \\ 0.33, \ 1.33 \end{array}$	$\begin{array}{c} 1.04 \pm 0.49 \\ 0.41, \ 1.55 \end{array}$	$3.60 \pm 1.43$ 1.90, 5.40	$3.27 \pm 1.17$ 1.60, 4.75
<sup>e</sup> P/MBTCA	$\begin{array}{c} 0.62 \pm 0.61 \\ 0.17, \ \ 5.90 \end{array}$	$\begin{array}{c} 3.22 \pm 0.62 \\ 2.47, \ \ 3.89 \end{array}$	$\begin{array}{l} 3.62 \pm 1.67 \\ 1.52, \ \ 5.90 \end{array}$	$\begin{array}{c} 1.68 \pm 0.64 \\ 0.83, \ \ 2.38 \end{array}$	$1.73 \pm 1.07$ 0.34, 3.32
Carbonaceous components					
Isoprene derived SOC (µgC m <sup>-3</sup> )	$\begin{array}{c} 23.7 \pm 23.9 \\ 2.26, \ \ 97.4 \end{array}$	$\begin{array}{l} 51.0 \pm 32.4 \\ 8.08, \ 97.4 \end{array}$	$\begin{array}{l} 25.9 \pm 17.9 \\ 8.54, \ \ 45.7 \end{array}$	$\begin{array}{l} 8.52 \pm 1.07 \\ 7.46, \ \ 9.87 \end{array}$	$\begin{array}{c} 11.3 \pm 6.80 \\ 2.26, \ \ 24.9 \end{array}$
Isoprene SOC to OC (%)	$\begin{array}{c} 1.45 \pm 1.74 \\ 0.21, \ \ 6.40 \end{array}$	$\begin{array}{l} 3.56 \pm 2.18 \\ 1.03, \ \ 6.40 \end{array}$	$\begin{array}{c} 1.46 \pm 1.38 \\ 0.35, \ \ 3.48 \end{array}$	$\begin{array}{c} 0.35 \pm 0.17 \\ 0.21, \ \ 0.57 \end{array}$	$\begin{array}{l} 0.57 \pm 0.38 \\ 0.24, \ 1.35 \end{array}$
Isoprene SOC to total SOC (%)	$35.5 \pm 15.5$ 13.5, 71.1	$\begin{array}{l} 46.6 \pm 20.3 \\ 26.0, \ \ 71.1 \end{array}$	$38.1 \pm 10.4$ 21.1, 47.3	$31.6 \pm 7.05$ 24.0, 39.7	$\begin{array}{l} 27.9 \pm 15.5 \\ 13.5, \ 54.0 \end{array}$
Monoterpene SOC (µgC m <sup>-3</sup> )	$40.1 \pm 33.3$ 7.18, 154	$\begin{array}{l} 62.7 \pm 56.5 \\ 19.2,  154 \end{array}$	$\begin{array}{l} 40.7\pm 20.0\\ 11.8, \ 57.7\end{array}$	$\begin{array}{c} 19.3 \pm 5.61 \\ 14.0, \ \ 25.2 \end{array}$	$35.5 \pm 23.5$ 7.18, 62.7
Monoterpene SOC to OC (%)	$\begin{array}{c} 2.0 \pm 1.47 \\ 0.37, \ \ 5.74 \end{array}$	$3.65 \pm 1.58$ 2.44, 5.74	$\begin{array}{c} 2.15 \pm 1.45 \\ 0.48, \ \ 3.88 \end{array}$	$\begin{array}{c} 0.87 \pm 0.58 \\ 0.37, \ 1.48 \end{array}$	$\begin{array}{c} 1.58 \pm 0.84 \\ 0.54, \ \ 3.13 \end{array}$
Monoterpene SOC to total SOC (%)	$\begin{array}{l} 64.5 \pm 15.5 \\ 28.9, \ 86.5 \end{array}$	$53.4 \pm 20.3 \\ 28.9, \ 74.0$	$\begin{array}{l} 61.9 \pm 10.4 \\ 52.7, \ \ 78.9 \end{array}$	$\begin{array}{c} 68.4 \pm 7.05 \\ 60.3, \ \ 76.1 \end{array}$	$\begin{array}{c} 72.1 \pm 15.5 \\ 46.0, \ 86.5 \end{array}$

<sup>e</sup>P: *cis*-pinonic acid + pinic acid

Sampling sites	Sampling type	Sampling time	Anhydro- sugars	Primary sugars	Sugar alcohols	References
Gosan, South Korea	TSP	Summer Fall Winter Spring	3.74 25.9 48.0 15.7	22.8 38.3 20.2 90.0	14.3 13.4 2.60 7.36	This study
Chennai, India	PM <sub>10</sub>	Summer Winter	127 134	15.5 11.4	7.44 4.81	Fu et al., 2010b
Mt. Tai, China	TSP	Summer (June)	224	61.1	125	Fu et al., 2012b
Alert, Canada	TSP	Winter Spring	0.32 0.02	1.14 0.18	0.25 0.36	Fu et al., 2009a
Okinawa, western North Pacific	TSP	Summer Autumn Winter Spring	0.93 2.58 6.04 3.44	73.5 56.0 34.4 101	62.9 30.7 6.03 31.2	Zhu et al., 2015a, b
Chichijima, western North Pacific	TSP	Summer Autumn Winter Spring	0.32 0.85 2.40 0.94	32.8 22.0 14.2 24.2	38.6 35.1 3.93 15.5	Verma et al., 2015, 2018
Mt. Hua, China (Non-dust storm)	$PM_{10}$	April	57.8	92.5	22.4	Wang et al., 2012
Mt. Hua, China (Dust storm)	PM <sub>10</sub>	April	44.5	162	25.7	Wang et al., 2012
Nanjing, China	PM <sub>2.5</sub>	Summer Winter	151 (Lev) 268 (Lev)	59.3 42.3	11.8 13.4	Wang and Kawamura, 2005
Beijing, China	PM <sub>2.5</sub>	Summer Autumn Winter Spring	14.5 129 254 81.4	6.63 17.2 41.5 33.9	3.31 13.7 17.8 12.3	Kang et al., 2018a
Belgrade, Serbia	TSP	Autumn	425 (Lev)	116	98.4	Zangrando et al., 2016
Maine, USA	$PM_1$	May- October	13.9	28	8.31	Medeiros et al., 2006
Crete, Greece	$PM_{10}$	Year-round	14.4	32.3	6.53	Theodosi et al., 2018

**Table 3.** Comparisons of the mean concentration (ng m<sup>-3</sup>) of anhydrosugars, sugar, and sugar alcohols in Gosan aerosols with those from different sites around the world.