

# Transport-driven aerosol differences above and below the canopy of a mixed deciduous forest

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**Abstract.** Exchanges of energy and mass between the surrounding air and plant surfaces occur below, within, and above a forest's vegetative canopy. The canopy also can lead to vertical gradients in light, trace gases, oxidant availability, turbulent mixing, and properties and concentrations of organic aerosols (OA). In this study, a high-resolution time-of-flight aerosol mass spectrometer is used to measure non-refractory submicron aerosol composition and concentration above (30m) and below (6m) a forest canopy in a mixed deciduous forest at the Program for Research on Oxidants: Photochemistry, Emissions, and Transport tower in northern Michigan during the summer of 2016. Three OA factors are resolved using positive matrix factorization: more-oxidized oxygenated organic aerosol (MO-OOA), isoprene-epoxydiol-derived organic aerosol (IEPOX-OA), and 91Fac (a factor characterized with a distinct fragment ion at  $m/z$  91) from both the above- and below-canopy inlets. MO-OOA was most strongly associated with long-range transport from more polluted regions to the south, while IEPOX-OA and 91Fac were associated with shorter-range transport and local oxidation chemistry. Overall vertical similarity in aerosol composition, degrees of oxidation, and diurnal profiles between the two inlets was observed throughout the campaign, which implies that rapid in-canopy transport of aerosols is efficient enough to cause relatively consistent vertical distributions of aerosols at this scale. However, four distinct vertical gradient episodes are identified for OA, with vertical concentration differences (above-canopy minus below-canopy concentrations) in total OA of up to 0.8  $\mu\text{g}/\text{m}^3$ , a value that is 42% of the campaign average OA concentration of 1.9  $\mu\text{g}/\text{m}^3$ . The magnitude of these differences correlated with concurrent vertical differences in either sulfate aerosol or ozone. These differences are likely driven by a combination of long-range transport mechanisms, canopy-scale mixing and local chemistry. These results emphasize the importance of including vertical and horizontal transport mechanisms when interpreting trace gas and aerosol data in forested environments.

## 34 1 Introduction

35 Aerosols play a key role in the energy balance of the Earth's climate system by scattering and absorbing incoming solar  
36 radiation and by impacting cloud lifetime and reflectivity (IPCC, 2007). These climatic effects depend strongly on the  
37 chemical speciation of the aerosol particles. Approximately 20-90% of submicron aerosol mass worldwide on average has  
38 been predicted to be organic material (Kanakidou et al., 2005), and this is supported by field studies in a variety of urban,  
39 urban downwind, and rural locations across the Northern Hemisphere (Jimenez et al., 2009; Zhang et al., 2007). The bulk of  
40 this organic material is thought to be secondary organic aerosol (SOA), which is formed in the atmosphere by the reaction of  
41 volatile organic compounds (VOCs) with oxidants such as hydroxyl radical (OH), ozone (O<sub>3</sub>), nitrate radical, and chlorine  
42 atom, with the resulting products then partitioning to the particle phase.

43 Precursor VOCs that contribute to SOA formation are emitted from both anthropogenic and biogenic sources. Biogenic  
44 VOCs (BVOCs) primarily are emitted into the atmosphere from terrestrial vegetation, and on a global scale, emissions of  
45 BVOCs exceed those of anthropogenic VOCs (Fehsenfeld et al., 1992; Guenther et al., 2000, 1995). Major SOA precursor  
46 BVOCs include isoprene (C<sub>5</sub>H<sub>8</sub>) and terpenes. To date, more than 5000 terpene compounds have been identified, including  
47 monoterpenes (C<sub>10</sub>), sesquiterpenes (C<sub>15</sub>), and diterpenes (C<sub>20</sub>) (Geron et al., 2000). However, factors such as the addition  
48 of functional groups (aldehydes, alcohols, carboxylic acids, alkyl nitrate, etc.) and the wide variety of possible reaction  
49 pathways rapidly increase the number of relevant atmospheric VOCs beyond what is initially emitted (Goldstein and  
50 Galbally, 2007).

51 If other loss processes, including deposition, were not considered, the final atmospheric fate of the carbon associated  
52 with BVOCs would be oxidation to carbon dioxide. However, partitioning of oxidation products to the aerosol phase as  
53 SOA interrupts this oxidation sequence. Partitioning of VOC oxidation products between the gas and aerosol phases  
54 depends on multiple factors such as the phase and concentration of pre-existing primary OA (POA) or SOA, particulate-  
55 phase SOA reactions, and the presence of aerosol liquid water (ALW) (Seinfeld and Pandis, 2006). As a result of the large  
56 number of precursor VOCs, the highly non-linear oxidation chemistry, and the presence of multiple aerosol phases in the  
57 atmosphere, SOA formation is complex and relatively poorly understood (Goldstein and Galbally, 2007).

58 The physical environment strongly impacts these chemical SOA formation processes. For example, in forested areas in  
59 which BVOC emissions are prevalent, the exchanges of energy and mass between the forest and the atmosphere are  
60 influenced by the forest's vegetation canopy. Absorption of light by a canopy can diminish the amount of radiation that is  
61 received below the canopy, influencing photolysis rates of photolabile species (Baldocchi et al., 1995; Brown et al., 2005;  
62 Fuentes et al., 2007; Makar et al., 2017; Schulze et al., 2017) and oxidant availability (Fuentes et al., 2007). Loss of BVOC  
63 oxidation products to deposition within the canopy also has been found to be an important factor in determining the  
64 oxidative capacity of a forested environment (Pugh et al., 2009).

65 Vertical transport, also influenced by the canopy, likewise impacts the concentrations of BVOC and SOA in forested  
66 environments. Roughness elements created by the leaves, branches, and stems in a dense vegetative canopy combined with  
67 above-canopy wind shear leads to coherent structures (Finnigan, 2000). These turbulent flow structures contribute to the  
68 fluxes of heat, energy, and matter in forest canopies (Thomas and Foken, 2007). The physical motion of coherent structures  
69 occurs via two main mechanisms: upward "bursts," in which air is ejected upward from the canopy into the atmosphere, and  
70 downward "sweeps," in which air is directed downward from the atmosphere into the canopy. Vertically resolved sonic  
71 anemometer measurements, which provide temperature and three-dimensional wind velocity components at each vertical  
72 measurement location, can be used to derive in-canopy mixing metrics.

73 The relative magnitudes of timescales for turbulent transport and chemical processing govern how trace compounds are  
74 distributed within the canopy (Baldocchi et al., 1995; Fuentes et al., 2007; Steiner et al., 2011). A modeling study by Gao et  
75 al. (1993) found that in-canopy chemical processing of isoprene occurs on a much longer timescale than turbulent transport,  
76 making in-canopy reactions less important compared to the turbulent transport and emission of isoprene in determining its

77 in-canopy concentrations. On the other hand, for compounds with an estimated chemical loss timescale that is roughly  
78 equivalent to the timescales of turbulent transport (e.g., O<sub>3</sub>-initiated oxidation of the sesquiterpene  $\beta$ -caryophyllene), rapid  
79 in-canopy chemical loss could dominate (Stroud et al., 2005). Furthermore, partitioning to the aerosol-phase was inferred as  
80 a potential reason for observations of decreased mixing ratios of oxidation products of very reactive BVOCs in a ponderosa  
81 pine forest canopy (Holzinger et al., 2005).

82 Several studies have attempted to model vertical profiles of trace gases and aerosols in a forest canopy considering both  
83 chemistry and turbulent transport. Bryan et al. (2012) found that forest canopy-atmosphere interactions were highly  
84 sensitive to turbulent mixing parameterizations during a field campaign in northern Michigan. Differences in highly reactive  
85 BVOCs and BVOC oxidation products have been estimated above and below the canopy in modeling and measurement  
86 efforts (Alwe et al., 2019; Ashworth et al., 2015; Holzinger et al., 2005; Schulze et al., 2017; Stroud et al., 2005; Wolfe and  
87 Thornton, 2011). Schulze et al. (2017) found that rapid through-canopy transport (minimum in-canopy residence time of 10  
88 min) leads to relatively consistent simulated above- and below-canopy SOA composition and concentration.

89 Because in-canopy mixing plays a role in the vertical distribution of trace gases and aerosols in a forest canopy, vertical  
90 differences in OA components and other inorganic aerosol species such as sulfate (SO<sub>4</sub>) could be caused by the degree of  
91 mixing between the above- and below-canopy environments. During the PROPHET-CABINEX 2009 campaign, the degree  
92 of atmosphere-canopy coupling between the above-canopy atmosphere and the forest was analyzed by Steiner et al. (2011).  
93 In this study, the degree of coupling was calculated using the ratio of the kinematic heat flux above the canopy to the  
94 kinematic heat flux in the upper canopy. Opposing kinematic heat flux directions (negative ratios) imply that the below-  
95 canopy environment is uncoupled from the atmosphere. In 2009, coupling conditions ranged between strong coupling  
96 (greater than zero but less than the threshold value defined by the slope of a regression between the two heat fluxes), weak  
97 coupling (greater than the defined threshold value), and uncoupled (negative). Uncoupled conditions occurred most  
98 commonly in the early morning hours between 4AM and 8AM local time. This set of hours represents approximately 30%  
99 of every day over the whole 2009 study period. This suggests that early morning hours may contribute to more instances of  
100 uncoupled canopy-atmosphere conditions but that coupling between the forest canopy and the atmosphere occurs a majority  
101 of the time (Steiner et al., 2011).

102 Past field studies, such as those above a tropical forest in Brazil and a temperate forest in California, found that in-  
103 canopy SOA formation, deposition, and thermal gradient effects on gas-particle partitioning all influence net OA fluxes  
104 (Farmer et al., 2013). In the same study, the authors found that oxygenated OA tended to deposit in the canopy, whereas the  
105 forest canopy released less oxygenated OA. The source of OA fluxes between forests and the atmosphere has been  
106 associated with vertical turbulent transport between the forest atmosphere and the surface layer directly above the canopy;  
107 however, there have been few studies that have provided high temporal resolution measurements of both OA composition  
108 throughout a forest canopy and canopy mixing strength. In a mixed forest in Ontario, Canada, Gordon et al. (2011) found  
109 that the frequent occurrence of net upward aerosol flux was associated with decoupled canopy conditions where entrainment  
110 of particle-free air from above the canopy created a positive flux above the forest. On the other hand, Whitehead et al.  
111 (2010) found that the particle number concentration and submicron particle composition in the trunk space (i.e., below-  
112 canopy) and above-canopy environments showed minimal variation with height during daytime due to stronger turbulence  
113 and mixing conditions. In light of these limited studies, there is a need for additional measurements and data to inform the  
114 exchange of aerosols between the forest and the atmosphere.

115 Recent work from the Program for Research on Oxidants and PHotochemistry, Emissions and Transport (PROPHET)  
116 during the Atmospheric Measurements of Oxidants in Summer (AMOS) campaign in 2016 has indicated that flux of  
117 isoprene and monoterpenes at the canopy-atmosphere boundary represents over half of the net carbon flux, while oxygenated  
118 VOCs (OVOCs) constitute a majority of the species with detectable VOC fluxes (192 of the 236 species with identified  
119 molecular formulas). The authors report that the observed and modelled net carbon flux during the campaign was upward  
120 (canopy emission) during the campaign in 2016 (Millet et al., 2018). Vertical gradients of BVOCs in this mixed forest  
121 environment vary greatly depending on canopy vegetation height, primary emission versus secondary production, and  
122 diurnal variability (Alwe et al., 2019). Correlation analysis of VOC vertical gradients suggests that formic acid (HCOOH)

originates from a secondary photochemical production. During the same campaign, distinct sample-to-sample variability in the molecular-level aerosol composition (73 +/- 8%) was observed despite less variability in elemental composition of the bulk OA, indicating the chemical complexity of functionalized OA at the site (Ditto et al., 2018).

Here, measurements using an Aerodyne high-resolution time-of-flight aerosol mass spectrometer (HR-ToF-AMS) are used to characterize ambient aerosol above and below the canopy in a mixed, deciduous forest during the PROPHET-AMOS campaign in 2016, with the aim of evaluating quantitatively potential chemical and physical phenomena leading to observed differences between particulate matter above and below the forest canopy. These measurements provide data for validation of forest canopy-atmosphere exchange models and allow an assessment of turbulent transport in the forest canopy and the resulting impacts on above- and below-canopy SOA composition and concentration. These data also could assist in the determination of whether or not SOA is forming within the forest canopy.

## 2 Methods

### 2.1 Site description

Measurements were made during the PROPHET-AMOS 2016 campaign from 1 July – 31 July 2016. The PROPHET site (45.55° N, 84.78° W) is situated in a temperate, mixed deciduous forest in the northern portion of Michigan's Lower Peninsula at the University of Michigan Biological Station (UMBS). The surrounding forest consists of aspens (60.9%), northern hardwoods (16.6%; maple, beech, birch, ash, and hemlocks), upland conifers (13.3%; white and red pines), non-forest cover types (7.6%; bracken ferns/grass/developed/road), and northern red oak (1.6%) (Cooper et al., 2001; Bergen and Dronova, 2007). The forests in this region are currently undergoing succession, where the dominant aspens have matured and are now being replaced by northern hardwoods and pines (Bergen and Dronova, 2007). It is expected that this new forest composition will shift BVOC emissions from an isoprene-dominated environment to one that is more influenced by monoterpenes (Toma and Bertman, 2012). The height of the forest canopy varies, but the mean canopy height is approximately 22.5 meters (VanReken et al., 2015). The site's physical layout and site meteorology have been described elsewhere (Carroll et al., 2001; Cooper et al., 2001). The PROPHET site features a 31-m scaffolding tower, allowing measurements to be made at variable heights within and above the vegetation canopy.

Due to the sparse surrounding population, the PROPHET site at UMBS has minimal local anthropogenic influences. The closest major urban centers include Detroit, Michigan (350 km to the southeast, population: 672,795), Milwaukee, Wisconsin (350 km to the southwest, population: 595,047), and Chicago, Illinois (450 km to the southwest, population: 2,704,958) (Carroll et al., 2001). The closest nearby towns include Pellston, Michigan (5 km to the west, population: 828), Petoskey, Michigan (30 km to the southwest, population: 5,749), and Cheboygan, Michigan (30 km to the northeast, population: 4,726). Population totals are based on 2016 population estimates for cities and towns in the United States (US) from the US Census Bureau (US Census Bureau, 2018).

The mean temperature during the campaign was 20.6 +/- 4.6 °C (mean +/- one standard deviation), which is 3-4°C warmer than the mean temperature during the 2009 PROPHET-CABINEX campaign (16.9 °C) (VanReken et al., 2015). Temperature conditions during this study are consistent with mean summertime temperatures from studies in 2008 and 2010 (Toma and Bertman, 2012) and with historical mean temperature data for the month of July from the Pellston Regional Airport (Figure S1 in the Supplemental Information (SI)). The mean relative humidity (RH) during the campaign was 73.8 +/- 17.5% , which is similar to conditions during the CABINEX campaign (74.5 +/- 17.5%) (VanReken et al., 2015). Winds recorded at the top of the PROPHET tower originated mostly from the west, southwest, and northwest (as shown in Figure S2 in the SI). The historical average precipitation according to the National Atmospheric Deposition Program (NADP) National Trends Network for the month of July from 1979-2015 for UMBS is 69.0 +/- 37.8 mm. During the PROPHET-AMOS 2016 campaign, the total accumulated precipitation was 79.0 mm, indicating that precipitation at the site during the campaign was within one standard deviation of the historical July average (NADP, 2016).

## 165 2.2 Instrumentation and sampling

166 A summary of the trace-gas measurements, instrumentation, and meteorological parameters on board the University of  
167 Houston/Rice University Mobile Air Quality Laboratory (MAQL) are summarized in Table S1 of the SI. Details and  
168 operation of the MAQL have been described previously in the literature (Leong et al., 2017; Wallace et al., 2018). The  
169 MAQL was situated approximately 10 meters to the east of the PROPHET tower. A photograph of the MAQL in stationary  
170 sampling mode is shown in Figure S3 in the SI. Table S2 in the SI lists the measurements, measurement methods, and  
171 sampling heights of other participating institutions at PROPHET-AMOS 2016 during the campaign. Measurements are  
172 reported in local time (LT, Eastern Daylight Time (EDT)) during the campaign.

### 173 2.2.1 Trace gases

174 Below-canopy trace gases including nitric oxide (NO), nitrogen dioxide (NO<sub>2</sub>), total reactive nitrogen (NO<sub>y</sub>), O<sub>3</sub>, carbon  
175 monoxide (CO), and sulfur dioxide (SO<sub>2</sub>) were measured from a common inlet on the sampling arm of the MAQL. Reported  
176 values of nitrogen oxides (NO<sub>x</sub>) represent the sum of NO and NO<sub>2</sub>, while reported values of NO<sub>y</sub> include NO<sub>x</sub> and its  
177 reservoir species. These measurements were taken at a height of 6 m above ground level.

### 178 2.2.2 HR-ToF-AMS

179 The HR-ToF-AMS (Aerodyne Research Inc., USA) was used to measure non-refractory submicron particulate matter  
180 (NR-PM<sub>1</sub>) from the MAQL; measured composition includes OA, SO<sub>4</sub>, nitrate (NO<sub>3</sub>), ammonium (NH<sub>4</sub>), and chloride (Cl).  
181 Detailed descriptions of the operation and principles of the HR-ToF-AMS have been provided elsewhere (DeCarlo et al.,  
182 2007). In brief, particles are sampled through a 100-micron critical orifice and are focused into a particle beam using an  
183 aerodynamic lens. After traversing a vacuum chamber, the particles in the beam impact onto a tungsten vaporizer heated to  
184 600°C. The vapors formed are ionized by electron impact ionization at 70 eV. The resulting ions are detected using a high-  
185 resolution time-of-flight mass spectrometer. The HR-ToF-AMS was operated in a high mass sensitivity mode, referred to as  
186 V-mode. Ionization efficiency (IE) calibrations were performed at the beginning and end of the campaign using  
187 monodisperse 300 nanometer NH<sub>4</sub>NO<sub>3</sub> particles. Gas-phase interferences were subtracted from the data based on the  
188 observed signal when ambient air was sampled through a filter. Filter zeros were run each day at varying times of day.

189 Sampling was performed from the raised 6-m inlet on the common sampling arm of the MAQL (below canopy) and  
190 from a 30-m inlet on the PROPHET tower (above canopy). Copper tubing was used for the sampling inlets, and each inlet  
191 was fitted with cyclones to remove particles larger than 2.5 microns in diameter. Prior to the HR-ToF-AMS critical orifice,  
192 air was sampled through a nafion dryer to dry the sample flow. A three-way valve was used to alternate HR-ToF-AMS  
193 sampling between the above- and below-canopy inlets at 10-minute intervals.

### 194 2.2.3 PTR-QiToF

195 Above-canopy mixing ratios and fluxes, along with in-canopy vertical gradients, were measured for a wide array of  
196 VOCs during PROPHET-AMOS 2016 by Proton Transfer Reaction-Quadrupole interface Time-Of-Flight mass spectrometry  
197 (PTR-QiToF). A detailed description of the sampling configuration, calibration and zeroing procedures, humidity  
198 corrections, and instrumental performance during the campaign is provided by Alwe et al. (2019) and Millet et al. (2018).  
199 Briefly, six identical 45-m inlet lines (0.5" OD/0.375" ID PFA, each heated to 50°C) were installed on the PROPHET tower  
200 to sample from 34 m, 21 m, 17 m, 13 m, 9 m, and 5 m above ground level. Sample flow was maintained at ~40 standard  
201 L/min (SLM) for the 34 m inlet line and at >5-10 SLM for the others. Each hour, 30 min was spent sampling from the 34 m  
202 inlet to quantify above-canopy VOC mixing ratios and fluxes. The remainder of each hour was spent characterizing in-  
203 canopy vertical gradients by sequentially sampling from the other inlets (for 5 min apiece) followed by a 5-min instrumental  
204 blank.

## 205 2.3 HR-ToF-AMS data and positive matrix factorization analysis

206 Data analysis for HR-ToF-AMS data was performed in Igor Pro v6.37 using the SQUIRREL v1.57 (SeQUential Igor  
207 data RetRiEval) and PIKA v1.16 (Peak Integration by Key Analysis) analysis toolkits. High-resolution mass spectral fitting  
208 was performed on HR-ToF-AMS V-mode data. Ratios such as the oxygen-to-carbon elemental ratio (O:C) and hydrogen-to-  
209 carbon elemental ratio (H:C) were determined according to the “improved-ambient” (IA) method (Canagaratna et al., 2015).  
210 The IA method uses specific ion fragments to correct for compositional biases, and this method has been shown to calculate  
211 accurately the elemental ratios of organic laboratory standards that are more representative of oxidized, ambient OA species  
212 (Canagaratna et al., 2015). The default values for relative IE were used for each of the following species: OA (1.4), SO<sub>4</sub>  
213 (1.2), NH<sub>4</sub> (4), NO<sub>3</sub> (1.1), and Chl (1.3), where values in parentheses refer to the ratio of the IE of the given species with  
214 respect to the value of IE of NO<sub>3</sub> obtained during routine IE calibrations using NH<sub>4</sub>NO<sub>3</sub>.

215 To account for the effects of aerosol composition on the transmission efficiency of aerosols to the detection region of the  
216 HR-ToF-AMS, a chemical composition-dependent (and therefore time-dependent) collection efficiency (CDCE) was applied  
217 to the HR-ToF-AMS data (Middlebrook et al., 2012), which led to a campaign average CDCE of 0.77 +/- 0.18. During the  
218 campaign, the HR-ToF-AMS time resolution was 40 seconds from 2 July 02:00 to 19 July 10:00 LT, after which the HR-  
219 ToF-AMS time resolution was changed to 30 seconds between 22 July 11:00 and 31 July 15:00 LT (due to instrumental  
220 changes after an HR-ToF-AMS power supply failure). Calculated detection limits of NR-PM<sub>1</sub> species are included in Table  
221 S3.

222 Positive matrix factorization (PMF) is a mathematical model in which measured data are decomposed into a  
223 combination of factors that have varying contributions throughout a time series. Here, the PMF model has been applied to  
224 HR-ToF-AMS data to retrieve OA factors that contain information regarding OA sources, chemical properties, and/or  
225 atmospheric processing (Jimenez et al., 2009; Ulbrich et al., 2009; Zhang et al., 2011). Subtypes of OA extracted from OA  
226 mass spectra using PMF often include (but are not limited to): more-oxidized oxygenated OA (MO-OOA), less-oxidized  
227 oxygenated OA (LO-OOA), hydrocarbon-like OA (HOA), biomass burning organic aerosol (BBOA), and cooking organic  
228 aerosol (COA) (Cubison et al., 2011; Jimenez et al., 2009; Mohr et al., 2012; Zhang et al., 2007, 2005). The OOA-related  
229 factors are generally considered to be associated with SOA, while subtypes such as HOA, BBOA, and COA are presumed to  
230 correspond to POA. Factors associated with isoprene-derived epoxydiol OA (IEPOX-OA) have also been identified using  
231 PMF using enhanced signals at m/z 82 in their OA mass spectra (Hu et al., 2015). A detailed description of the PMF model  
232 (Paatero and Tapper, 1994; Paatero, 1997; Ulbrich et al., 2009; Xu et al., 2015) is included in the SI.

233 The results of the separate PMF analyses on above- and below-canopy OA are included in the SI. For the above-canopy  
234 OA data, a summary of the PMF factor selection (Table S4), factor time series correlations with external data (Table S5),  
235 factor mass spectra correlations with reference mass spectra (Table S6), time series of PMF model residuals (Figure S4), and  
236 mass spectra and time series for possible two- to five-factor PMF solutions (Figure S5 - Figure S8) are shown in the SI.  
237 VOCs measured above-canopy via PTR-QiToF at the top of the PROPHET tower are defined in Table S7, and factor time  
238 series correlations with the time series of these VOCs are shown in Table S8. PMF diagnostics, such as the mass spectra and  
239 time series correlation amongst factors (Figure S9), FPEAK and SEED diagnostic plots (Figure S10), results of the FPEAK  
240 analysis (Table S9), model residual diagnostic plots (Figure S11), and results from bootstrapping analysis (Figure S12) are  
241 shown in the SI. Finally, Figure S13 displays the time series and high-resolution mass spectra of the optimal solution for the  
242 above-canopy OA dataset.

243 For the below-canopy OA data, a summary of the PMF factor selection (Table S10), factor time series correlations with  
244 external data (Table S11), factor mass spectra correlations with reference mass spectra (Table S12), mass spectra and time  
245 series of possible two- to five-factor PMF solutions (Figure S14 - Figure S17), factor time series correlations with VOCs  
246 measured via PTR-QiToF at the 34-m inlet on the PROPHET tower (Table S13), mass spectra and time series correlations  
247 amongst factors (Figure S18), FPEAK and SEED diagnostic plots (Figure S19), results from FPEAK analysis (Table S14  
248 and Table S15), model residual diagnostic plots (Figure S20), and results from bootstrapping analysis (Figure S21) are

shown in the SI. Finally, the time series and high-resolution mass spectra of the optimal three-factor solution for below-canopy OA are shown in Figure S22.

## 2.4 HYSPLIT backward trajectory analysis

### 2.4.1 Trajectory cluster analysis

Backward-trajectories are used in this study to determine the origin of air masses arriving at the field site using the Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT) model (Draxler and Hess, 1998; Stein et al., 2015). Meteorological data from the US Eta data assimilation system archive at 40-km spatial resolution (EDAS40) are used for HYSPLIT trajectory calculations. The EDAS40 data output is constructed using forecasted data from the Eta model, which utilizes observations from surface, aircraft, and satellite data to predict meteorological parameters such as pressure, wind speed, and wind direction (Cooper et al., 2001).

In order to assess the influence of air mass histories on aerosols at each site, a cluster analysis was performed on backward trajectories using MeteInfo v1.4.9R2 and the TrajStat v1.4.4R8 package (Wang, 2014; Wang et al., 2009). The angle distance clustering type is used in this study and calculates the angular distance between two backward trajectories as seen from the site, using methods outlined in Sirois and Bottenheim (1995). The number of suitable clusters is chosen based on the slope of the percentage change in total spatial variation versus number of clusters and a visual inspection of the mean trajectories of the cluster numbers.

### 2.4.2 Weighted potential source contribution function (WPSCF) analysis

In addition to a backward-trajectory cluster analysis performed for bulk aerosol properties and gas-phase species, two-day HYSPLIT backward-trajectories initiated from the PROPHET site at 500 m above ground level are used in a weighted potential source contribution function (WPSCF) analysis for OA factors. WPSCF analysis is performed in MeteInfo v1.4.9R2 using the TrajStat v1.4.4R8 package and results are plotted using ESRI's ArcMap v10.1 (Wang, 2014; Wang et al., 2009). Similar PSCF analyses using backward-trajectories have been performed previously using aerosol properties (Bondy et al., 2017; Chang et al., 2017; Polissar, 1999; Schulze et al., 2018).

The number of backward-trajectory endpoints falling within a given grid cell with coordinates  $(i, j)$  is defined as  $n_{ij}$ . The number of instances in which backward-trajectories ending at a given grid cell have a value (i.e., OA factor mass concentration) higher than an arbitrarily set criterion value is defined as  $m_{ij}$ . The PSCF value for a cell at location  $(i, j)$  is then defined as follows:

$$PSCF_{ij} = \frac{m_{ij}}{n_{ij}} \quad (1)$$

When calculating values of PSCF, some grid cells will contain only a small number of backward-trajectory endpoints. In order to reduce the high uncertainties related to a limited number of endpoints falling within a grid cell in PSCF analysis, a weighting function is applied to the trajectory numbers following the methods of Polissar et al. (2001):

$$W_{ij} = \begin{cases} 1.00 & 80 < n_{ij} \\ 0.70 & 20 < n_{ij} \leq 80 \\ 0.42 & 10 < n_{ij} \leq 20 \\ 0.05 & n_{ij} \leq 10 \end{cases} \quad (2)$$

In this study, the domain of analysis is set to the geographical extents of the two-day HYSPLIT backward trajectories initiated from the PROPHET site. A grid cell size of  $0.5^\circ$  by  $0.5^\circ$  is used. Median values of the OA factors are used as the arbitrary criterion values. Overall, the WPSCF analysis allows for an identification of potential source areas, where higher WPSCF values within a region indicate a higher likelihood that this region results in observed values higher than the criterion values.

## 286 2.5 Sonic anemometer data-processing

287 Turbulence measurements during the campaign were obtained from five sonic anemometers installed on the PROPHET  
288 tower at the following measurement heights: 34m (CSAT 3B, Campbell Scientific Inc.), 29m (81000, RM Young), 21m  
289 (CSAT 3, Campbell Scientific Inc.), 13m (CSAT 3, Campbell Scientific Inc.), and 5m (CSAT 3, Campbell Scientific Inc.).  
290 The sonic anemometer at 34 m was operated continuously during the campaign while data are only available from the lower  
291 heights from 9 July – 29 July 2016. High-frequency data are de-spiked (data points outside of 3.5-standard deviations are  
292 removed) and then separated into 30-minute windows to apply a tilt correction such that the x-axis is rotated into the  
293 direction of the mean wind velocity (Foken, 2008). Reynolds decomposition is then applied to the three-dimensional wind  
294 components ( $u, v, w$ ), so each variable (e.g.,  $u$ ) is separated into its mean ( $\bar{u}$ ) and fluctuating component ( $u'$ ). The friction  
295 velocity,  $u^*$ , is defined then as

$$296 \quad u^*{}^2 = -\overline{u'w'} \quad (3)$$

297 Any 30-minute periods that experienced rain (as measured by the rain-gauge at the UMBS AmeriFlux tower), weak winds  
298 (winds less than  $0.5 \text{ m s}^{-1}$  at the top sonic anemometer), or wind directed through the tower were excluded due to potential  
299 interference.

## 300 3 Results and discussion

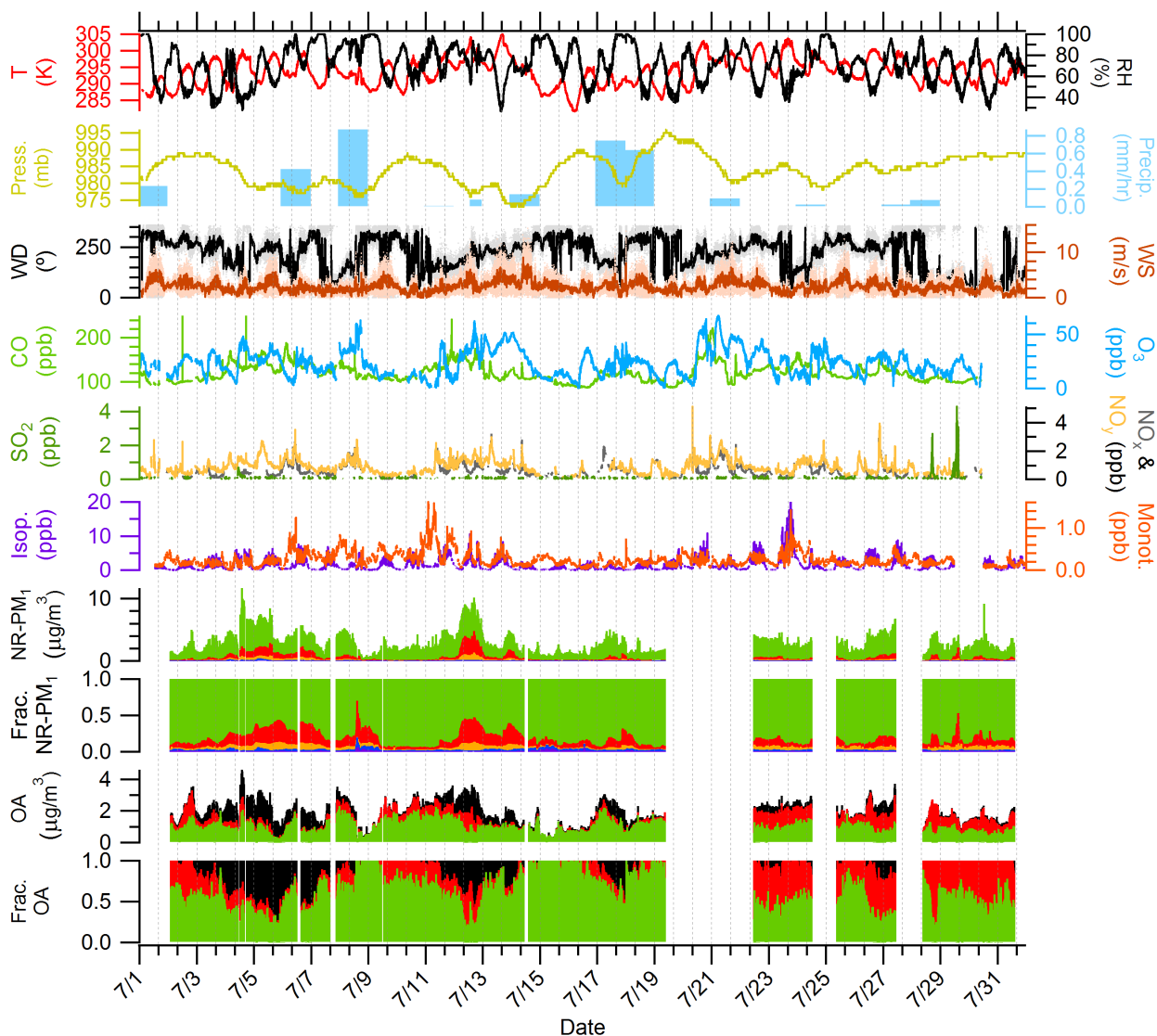
### 301 3.1 Backward-trajectory cluster analysis

302 Two-day backward trajectories are initiated from the PROPHET site and calculated at one-hour intervals from the  
303 beginning to the end of the campaign (1 July 00:00 – 31 July 21:00 LT) at 500 meters above ground level, an elevation  
304 selected to be within the boundary layer during the day and to avoid trajectory interaction with the surface. A total of 742  
305 two-day backward-trajectories are calculated over the course of the campaign. Cluster analysis resulted in three directional  
306 clusters: southerly (299 of 742), northeasterly (192 of 742), and northwesterly (251 of 742) (as shown in Figure S23 in the SI  
307 as Cluster 1, 2, and 3 respectively). Further description of cluster number selection is discussed in Figure S24 in the SI.  
308 Similar to the air mass history analyses at the PROPHET site in Cooper et al. (2001) and VanReken et al. (2015), eight-hour  
309 transitional periods between backward-trajectory classifications were removed from analysis because it is likely that the  
310 chemical species from these different air masses are mixed. In total, transitional periods composed 22% of the total number  
311 of backward-trajectories (165 of 742 total), and the remaining 577 are used for further analysis. Mean values of  
312 anthropogenically-influenced species such as  $\text{SO}_4$  and benzene were not statistically significantly different ( $p < 0.01$ )  
313 between trajectories from northeasterly and northwesterly clusters, so trajectories from these two clusters are grouped to  
314 represent a “northerly” air mass type. During the PROPHET-AMOS study, northerly transport occurred during 60% of the  
315 study period (341 of 577), while southerly transport occurred 40% of the time (236 of 577). This type of air-mass  
316 classification is consistent with previous summertime studies at the PROPHET site where northerly transport occurred 44%  
317 (1998), 60% (2009), and 57% (2014) of the time and southerly transport occurred 24% (1998), 29% (2009), and 43% (2014)  
318 of the time (Cooper et al., 2001; Gunsch et al., 2017; VanReken et al., 2015).

### 319 3.2 Non-refractory submicron time series and bulk chemical composition

320 Figure 1 shows the time series of the mass concentrations of  $\text{OA}$ ,  $\text{SO}_4$ ,  $\text{NH}_4$ ,  $\text{NO}_3$ , and  $\text{Chl}$  as measured by the HR-ToF-  
321 AMS. Mass concentrations plotted in Figure 1 represent time series data from both the above- and below-canopy sampling  
322 inlets. The average total  $\text{NR-PM}_{10}$  (sum of mass concentrations of  $\text{OA}$ ,  $\text{SO}_4$ ,  $\text{NH}_4$ ,  $\text{NO}_3$ , and  $\text{Chl}$ ) is  $2.3 \pm 1.5 \mu\text{g m}^{-3}$ . High  
323  $\text{NR-PM}_{10}$  concentration episodes (7/3-7/7 and 7/11-7/14) are strongly influenced by southerly air masses advecting





**Figure 1: Overview of time series of the following from top to bottom: temperature (T) and RH; pressure (Press.) and precipitation (Precip.); wind direction (WD) and wind speed (WS); CO and O<sub>3</sub>; SO<sub>2</sub>, NO<sub>x</sub>, and NO<sub>y</sub>; isoprene (Isop.) and total monoterpenes (Monot.); particulate OA (green), NO<sub>3</sub> (blue), SO<sub>4</sub> (red), NH<sub>4</sub> (orange), and Chl (purple); fraction of species to total NR-PM<sub>1</sub>, OA factors derived from PMF, and fractional contribution of OA factors to total OA. OA factors are colored as follows: MO-OOA (black), IEPOX-OA (red), and 91Fac (green). Precipitation data are provided from the NADP site in Cheboygan County, MI. Trace gas data (CO, O<sub>3</sub>, SO<sub>2</sub>, NO<sub>x</sub>, and NO<sub>y</sub>) are measured from the 6-m inlet on the MAQL. Meteorological data (T, RH, Press., WD, and WS) and VOC data (Isop. and Monot.) are measured from the 34-m inlet on the PROPHET tower. The OA time series include data from both heights (6 m and 30 m), with the sampling switched at 10-minute intervals.**

to the site, as confirmed by the HYSPLIT backward-trajectory clusters shown in Figure S23 in the SI. Northerly backward-trajectories originated over clean, remote areas in Canada, while southerly backward-trajectories originated over more anthropogenically influenced areas. OA is the dominant NR-PM<sub>1</sub> component over the entire campaign, representing

approximately 84.2% of the NR-PM<sub>1</sub> mass. SO<sub>4</sub> contributes the second highest average mass fraction to NR-PM<sub>1</sub> (10.7%) followed by NH<sub>4</sub> (3.1%), NO<sub>3</sub> (1.6%), and Chl (0.4%). During periods of northerly flow, OA represents 89.5% of the average NR-PM<sub>1</sub> mass, while SO<sub>4</sub>, NH<sub>4</sub>, NO<sub>3</sub> and Chl represent 6.9%, 1.9%, 1.3%, and 0.4%, respectively. Periods of southerly flow decrease the relative contribution of OA to 75.5% and Chl to 0.3% while increasing the relative contribution of SO<sub>4</sub> to 16.8%, NH<sub>4</sub> to 5.6%, and NO<sub>3</sub> to 1.3%. The increased fractional contribution of OA during periods of northerly-originating air is consistent with results from VanReken et al. (2015), who found that water-soluble organics dominated aerosol mass during periods of “clean” northerly flow at the PROPHET site. Sheesley et al. (2004) also found the aerosol organic carbon composition at a remote site in the upper peninsula of Michigan (approximately 100 miles from the PROPHET site) was greatly influenced by the source region of the air parcel. This study found that both stagnant and northerly air parcels contained higher concentrations of pinonic acid and limited amounts of primary emission tracer compounds, while anthropogenically influenced air parcels from the south and northwest contained higher concentrations of aromatic and aliphatic dicarboxylic acids.

Diurnal plots for OA, SO<sub>4</sub>, NH<sub>4</sub>, NO<sub>3</sub>, O:C, and H:C are shown in Figure S25 in the SI. The diurnal profiles of OA, SO<sub>4</sub>, and NH<sub>4</sub> are all relatively flat and do not have a clear diurnal trend. The lack of clear diurnal variations for SO<sub>4</sub> is consistent with regional transport as the source of SO<sub>4</sub> during this campaign. The diurnal variations of NO<sub>3</sub> show increases in the morning, with a maximum around approximately 10:00 local time, and lower concentrations in the afternoon.

Using the backward-trajectory clustering results described in Section 3.1, mean values of NR-PM<sub>1</sub>, NR-PM<sub>1</sub> species, OA elemental ratios, meteorological parameters, and trace gases for the entire campaign, northerly, and southerly backward-trajectory clusters are shown in Table 1. Overall, relative to northerly air masses, southerly air masses were found to be warmer (~2 °C), more humid (~5%), and associated with higher concentrations of NO<sub>x</sub>, NO<sub>y</sub>, O<sub>3</sub>, and benzene. Furthermore, on average, southerly air masses had higher concentrations of NR-PM<sub>1</sub>, OA, SO<sub>4</sub>, NH<sub>4</sub>, and NO<sub>3</sub>. Higher levels of OA oxidation (8% difference) based on O:C are also observed during periods of southerly flow (O:C = 0.69 versus 0.75 for northerly versus southerly flow, respectively). An additional metric of oxidation, the oxidation state of carbon (OSc), indicates that the degree of oxidation is higher during periods of southerly flow (OSc = -0.12 versus 0.06 for northerly versus southerly flow, respectively, where  $OSc = 2 * O:C - H:C$ ) (Kroll et al., 2011). The factor of 5 differences between northerly and southerly SO<sub>4</sub> is consistent with the increased influence of SO<sub>2</sub> point sources from electric-generating units south of the site in Ohio, Indiana, and Illinois. Overall, these observations demonstrate that anthropogenic influence from southerly flow directly affects NR-PM<sub>1</sub> mass concentrations, NR-PM<sub>1</sub> composition, the degree of OA oxidation, and trace gas mixing ratios at this site. Results are in agreement with VanReken et al. (2015), where southerly air masses or those “anthropogenically impacted” were found to contain higher aerosol loadings (in terms of aerosol volume, particle number, and median particle diameter) and hygroscopicity, along with increased trace gases abundances, as compared to northerly air masses or “clean” regimes.

Oxygen-containing ion families (C<sub>x</sub>H<sub>y</sub>O<sub>z+1</sub><sup>+</sup>) represent over 50% of the campaign-averaged OA high-resolution mass spectrum, and this high degree of oxygenation is reflected in an average O:C ratio of 0.71 +/- 0.08 and an average H:C ratio of 1.49 +/- 0.06. A distinct peak at m/z 44 in the average mass spectrum accounts for 14.1% of the total OA. The peak at m/z 44 is mainly composed of the CO<sub>2</sub><sup>+</sup> ion (96.0% of m/z 44). The ratio of m/z 44 to the total signal in the organic mass spectrum (f<sub>44</sub>), a surrogate for O:C and an indicator for photochemical aging, in this study is 0.14. Together with the average O:C ratio (0.71) observed in this study, these values are consistent with OOA observed across AMS datasets (Jimenez et al., 2009; Ng et al., 2010). Other prominent ions in the campaign-averaged high-resolution mass spectrum include m/z 55, 82, and 91. Fragments at m/z 55 represent 2.4% of the total OA, and are representative of both oxygenated and hydrocarbon fragments, such as C<sub>3</sub>H<sub>3</sub>O<sup>+</sup> (50.2% of m/z 55) and C<sub>4</sub>H<sub>7</sub><sup>+</sup> (42.9% of m/z 55). The possible OA sources leading to increased signal at m/z 82 (0.46% of total OA) and 91 (0.62% of total OA) will be discussed in the following section. Figure S25 in the SI shows the diurnal variations of O:C and H:C, indicating relatively stable diurnal cycles. Increases in H:C are observed starting at 9:00 before reaching a maximum at midday (13:00) followed by a slow decrease between the hours of 13:00 and 20:00, while an opposite pattern is observed for mean values of O:C.

384 **Table 1: Campaign-averaged values (+/- one standard deviation from the mean) of mass concentrations of NR-PM<sub>1</sub>, VOC and**  
385 **trace gas mixing ratios, and meteorological parameters, as well as hourly-averaged values associated with northerly and southerly**  
386 **backward-trajectory clusters. Northerly and southerly air masses are defined in Section 3.1 using a cluster analysis of HYSPLIT**  
387 **two-day backward trajectories.**

Parameter	Campaign	Northerly	Southerly
NR-PM <sub>1</sub> (μg m <sup>-3</sup> )	2.3 ± 1.5	1.8 ± 0.7	3.6 ± 2.0
OA (μg m <sup>-3</sup> )	1.9 ± 1.0	1.6 ± 0.6	2.6 ± 1.3
SO <sub>4</sub> (μg m <sup>-3</sup> )	0.3 ± 0.4	0.1 ± 0.1	0.7 ± 0.6
NH <sub>4</sub> (μg m <sup>-3</sup> )	0.1 ± 0.1	0.04 ± 0.04	0.2 ± 0.2
NO <sub>3</sub> (μg m <sup>-3</sup> )	0.04 ± 0.03	0.02 ± 0.01	0.06 ± 0.05
Chl (μg m <sup>-3</sup> )	0.01 ± 0.00	0.01 ± 0.00	0.01 ± 0.00
O:C	0.7 ± 0.1	0.7 ± 0.1	0.8 ± 0.1
H:C	1.5 ± 0.1	1.5 ± 0.0	1.4 ± 0.00
OSc	-0.1 ± 0.2	-0.1 ± 0.1	0.1 ± 0.2
Isoprene (ppb) <sup>a, b</sup>	1.6 ± 1.9	1.5 ± 1.5	1.8 ± 1.5
Monoterpenes (ppb) <sup>a, b</sup>	0.3 ± 0.2	0.2 ± 0.1	0.2 ± 0.2
Benzene (ppb) <sup>a</sup>	0.04 ± 0.02	0.03 ± 0.01	0.05 ± 0.02
NO (ppt) <sup>c</sup>	24.1 ± 36.4	21.5 ± 16.2	29.7 ± 29.5
NO <sub>2</sub> (ppt) <sup>c</sup>	593.0 ± 445.8	369.4 ± 257.7	872.0 ± 478.6
NO <sub>y</sub> (ppt) <sup>c</sup>	934.5 ± 490.0	610.5 ± 271.8	1283.9 ± 412.6
O <sub>3</sub> (ppb) <sup>c</sup>	24.1 ± 11.9	20.1 ± 8.3	30.5 ± 13.2
CO (ppt) <sup>c</sup>	120.6 ± 21.5	109.6 ± 12.6	134.7 ± 21.7
SO <sub>2</sub> (ppb) <sup>b, c</sup>	31.3 ± 174.2	45.0 ± 212.0	23.1 ± 64.2
Temperature (°C) <sup>c</sup>	20.6 ± 4.6	20.3 ± 4.5	22.4 ± 4.5
Relative Humidity (%) <sup>c</sup>	73.8 ± 17.3	68.6 ± 17.0	73.5 ± 17.6

388 <sup>a</sup> VOC measurements using University of Minnesota's PTR-QiToF from the 34-m inlet on the PROPHET tower.

389 <sup>b</sup> Table entries where the differences between the northerly and southerly backward-trajectories mean values are not  
390 statistically significant ( $p < 0.01$ ) using a two-sample t-test.

391 <sup>c</sup> Trace gas and meteorological parameters measured from onboard the MAQL.

392

### 393 3.3 Organic aerosol source apportionment

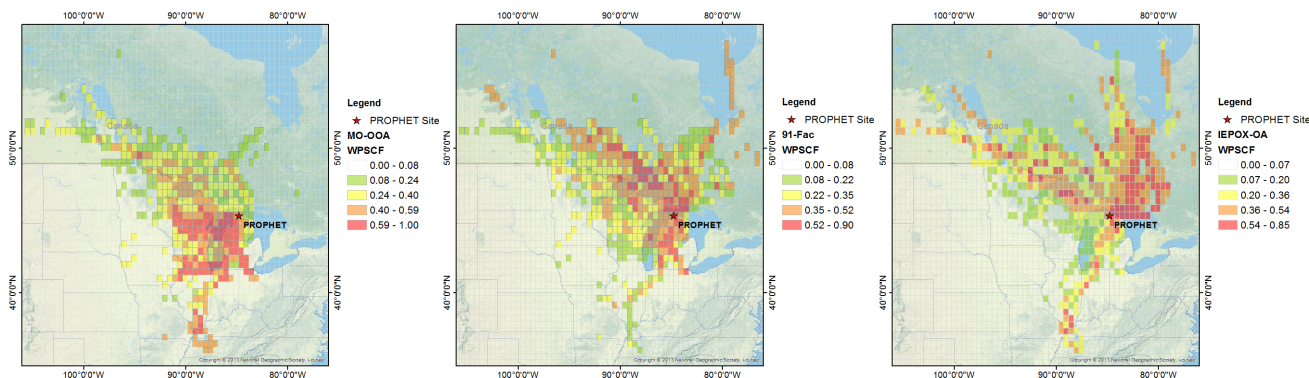
394 A three-factor solution is obtained for both the above-canopy inlet at 30 meters above the UMBS forest floor (A-MO-  
395 OOA, A-IEPOX-OA, A-91Fac) and the below-canopy inlet at 6 meters within the UMBS canopy (B-MO-OOA, B-IEPOX-  
396 OA, and B-91Fac). The 91Fac factor represents OA with enhanced signal at  $m/z$  91 in the mass spectrum. The addition of  
397 more factors to the PMF solution beyond three factors resulted in less physically meaningful and interpretable factors. Thus,  
398 the three-factor solution is considered the optimal solution for the above- and below-canopy OA datasets. In this study, seed  
399 = 0 was chosen for both PMF solutions as there was minimal variation in  $Q/Q_{\text{exp}}$  across the 50 seeds. Solutions at FPEAK =  
400 0 were chosen because solutions at other FPEAK values did not present improved correlations between factors and reference  
401 mass spectra.

402 Time series of mass concentrations and time series of fractional contributions to total OA are shown in Figure 1. The  
403 O:C ratio of each of the OA factors are as follows: 0.65 (IEPOX-OA), 0.71 (91Fac), and 0.89-0.90 (MO-OOA), all of which  
404 indicate the high degree of oxygenation in each of the factors. For the above-canopy PMF solution, 91Fac makes the largest

405 contribution to the total OA (43.8%), followed by IEPOX-OA (32.8%) and MO-OOA (23.4%). For the below-canopy PMF  
 406 solution, 91Fac also makes the largest contribution to the total OA (42.5%), followed by IEPOX-OA (34.0%) and MO-OOA  
 407 (23.5%). Details on the chemical composition, mass spectral characteristics and diurnal profiles of each factor are discussed  
 408 further in the SI.

409 Hourly averages of the above-canopy OA factors are paired with hourly two-day backward-trajectories for WPSCF  
 410 analysis. Results from WPSCF analysis using above-canopy OA data (Figure 2) indicate that A-MO-OOA predominantly  
 411 originates from southerly air masses, as supported by external measurements data such as benzene, OVOCs, carbonyls, and  
 412 SO<sub>4</sub>. Air masses that originate from the south pass over the large urban centers of Chicago, Milwaukee, and Detroit. This  
 413 further supports the aged, transported nature and anthropogenic influences of the observed A-MO-OOA at this site. In  
 414 contrast, no strong indication of a distinct source region is observed for 91Fac. This could suggest a more localized source  
 415 of 91Fac in relation to the site. Combining these WPSCF results with correlations of 91Fac with VOC masses corresponding  
 416 to monoterpene oxidation suggests that 91Fac is sourced from local biogenic, monoterpene-oxidation related chemistry (Xu  
 417 et al., 2018). Finally, WPSCF results for A-IEPOX-OA indicate that this OA factor coincides with more northerly airflow  
 418 with some contributions from northwesterly flow. These northerly source regions driving A-IEPOX-OA correspond to more  
 419 rural, less anthropogenically influenced locations in Canada. Interestingly, an instance of high WPSCF values for A-  
 420 IEPOX-OA and A-MO-OOA can be traced back to areas near the southeastern tip of Missouri and western Tennessee,  
 421 implying long-range transport of these OA factors. Backward-trajectories associated with this instance of high WPSCF  
 422 values for IEPOX-OA pass over areas in the high isoprene-emitting region in the Ozarks of southern Missouri, which is  
 423 commonly referred to as the “isoprene volcano” (Carlton and Baker, 2011; Wiedinmyer et al., 2005). The median mass  
 424 concentrations of above-canopy OA factors used as the criterion values for the WPSCF method were A-MO-OOA: 0.31  $\mu\text{g m}^{-3}$ ,  
 425 A-91Fac: 0.68  $\mu\text{g m}^{-3}$ , and A-IEPOX-OA: 0.58  $\mu\text{g m}^{-3}$ . Overall, WPSCF analysis indicates that transport from southerly  
 426 flow (A-MO-OOA), local sources (A-91Fac), and transport from northerly flow (A-IEPOX-OA) are related to the OA  
 427 factors observed at this site.

428



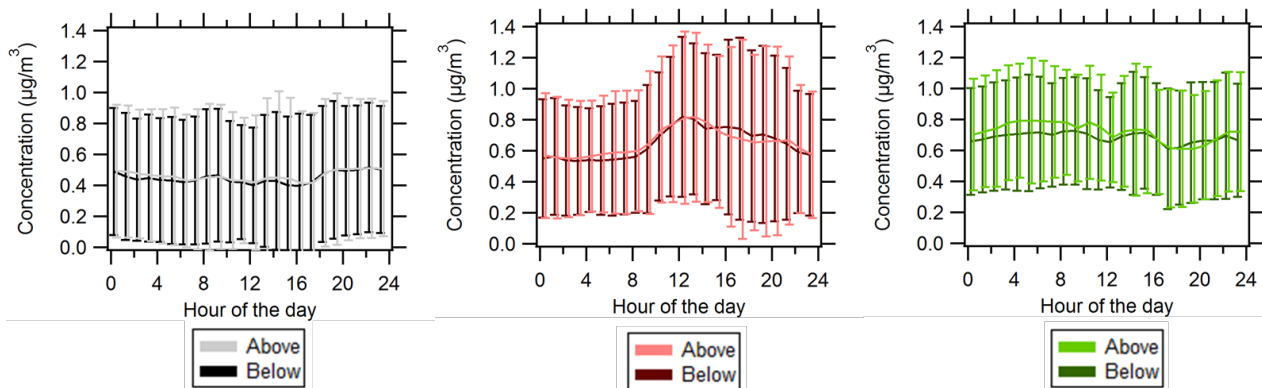
429

430 **Figure 2: WPSCF backward-trajectory analysis maps for the following hourly-averaged OA factors from left to right: above-**  
 431 **canopy MO-OOA, above-canopy 91Fac and above-canopy IEPOX-OA. The geographical location of the PROPHEt site is**  
 432 **represented by a red star. WPSCF analysis is performed using HYSPLIT two-day backward trajectories, a grid cell size of 0.5° by**  
 433 **0.5°, and the geographical domain of the extents of the two-day backward trajectories. Median mass concentrations of above-**  
 434 **canopy OA factors (A-MO-OOA: 0.31  $\mu\text{g m}^{-3}$ , A-91Fac: 0.68  $\mu\text{g m}^{-3}$ , and A-IEPOX-OA: 0.58  $\mu\text{g m}^{-3}$ ) are used as the criterion**  
 435 **WPSCF values. Color scales correspond to probability of source regions for each respective OA factor, and it should be noted that**  
 436 **the range and gradation of the color scale changes between each plot.**

437

438 Overall, PMF analysis for both sampling inlets at the site indicates that the OA is a combination of MO-OOA, 91Fac,  
 439 and IEPOX-OA. The dominant OA factor is 91Fac, and during periods of southerly flow MO-OOA contributes relatively  
 440 more (time series of the fractional contributions to total OA shown from both inlets is shown in Figure 1 and is also shown

separately for each inlet in Figure S13 and Figure S22 in the SI). On average, source apportionment results indicate that the OA at both inlets generally had similar average fractional OA contributions and degrees of oxidation. Diurnal profiles for above- and below-canopy OA are shown in Figure 3 and indicate that diurnal profiles and variations are similar between the two inlets for each OA factor. It is worth noting the diurnal pattern of the IEPOX-OA factor, likely indicating a relatively small influence of local isoprene emissions.



**Figure 3: Diurnal profiles of (left) MO-OOA, (middle) IEPOX-OA, and (right) 91Fac where solid lines represent average values and whiskers represent one standard deviation from the mean. Darker colors represent below-canopy OA (B-MO-OOA, B-IEPOX-OA, B-91Fac) while lighter colors represent above-canopy OA (A-MO-OOA, A-IEPOX-OA, A-91Fac). Data for below-canopy OA factors have been offset by 15 minutes simply to aid plot interpretation.**

### 3.4 Vertical characterization of above- and below-canopy NR-PM<sub>1</sub> and OA factors

#### 3.4.1 Similarity between above- and below-canopy environments

Mean values of NR-PM<sub>1</sub>, NR-PM<sub>1</sub> species, OA elemental ratios, meteorological parameters, VOCs, and trace gases from the above- and below-canopy inlets are summarized in Table 2. Each of the mean values for the parameters listed in Table 2 are within one standard deviation of each other for the below- and above-canopy sampling heights. However, results from Wilcoxon rank-sum tests indicate that the medians for a majority of the above- and below-canopy parameters are significantly different ( $p = 0.05$ ). To provide a comparison of the two inlets on the same time scale, scatter plots of 30-minute averaged values are shown in Figure 4 for OA factors, total OA, SO<sub>4</sub>, and O:C. For reference, a 1:1 line is shown on each scatter plot. Values falling within one standard deviation are similar between above and below canopy, while values deviating farther from the 1:1 line indicate values in which the above- or below-canopy environment had differing concentrations. Figure 4 illustrates that mass concentrations of OA factors, total OA, and SO<sub>4</sub> were similar over a majority of the campaign and suggests that the above- and below-canopy environments were generally coupled from a PM perspective. It also appears that total OA, MO-OOA, IEPOX-OA, and 91Fac show increases above canopy relative to below canopy at the highest end of the range of the measurements.

Results from a recent study from the PROPHET-AMOS 2016 campaign by Millet et al. (2018) indicate that flux of isoprene and monoterpenes at the canopy-atmosphere boundary represents over half of the net carbon flux, while OVOCs constitute a majority of the species with detectable VOC fluxes. Overall, the authors report that the observed and modeled net carbon flux during the campaign was upward (canopy emission) during the campaign in 2016. Additionally, in-canopy gradients of directly emitted BVOCs, such as isoprene and monoterpenes, indicate patterns that are consistent with their respective temperature, light, and physical emission dependencies (higher concentrations in-canopy for isoprene and higher

in the lower canopy for monoterpenes). On the other hand, in-canopy gradients of secondary products from BVOC oxidation, such as acetic acid and glycoaldehyde, indicate patterns consistent with net nighttime uptake and a weak peak

**Table 2: Campaign-averaged values (+/- one standard deviation) of mass concentrations of NR-PM<sub>1</sub>, VOC and trace gas mixing ratios, and meteorological parameters above and below the canopy.**

Parameter <sup>a, b</sup>	Above	Below
NR-PM <sub>1</sub> (µg m <sup>-3</sup> )	2.4 ± 1.6	2.3 ± 1.5
OA (µg m <sup>-3</sup> )	1.9 ± 1.0	1.8 ± 1.0
SO <sub>4</sub> (µg m <sup>-3</sup> )	0.3 ± 0.5	0.3 ± 0.4
NH <sub>4</sub> (µg m <sup>-3</sup> )	0.1 ± 0.2	0.1 ± 0.2
NO <sub>3</sub> (µg m <sup>-3</sup> )	0.04 ± 0.04	0.04 ± 0.03
Chl (µg m <sup>-3</sup> )	0.01 ± 0.00	0.01 ± 0.00
O:C	0.7 ± 0.1	0.7 ± 0.1
H:C <sup>c</sup>	1.5 ± 0.1	1.5 ± 0.1
OSc	0.0 ± 0.2	-0.1 ± 0.2
Isoprene (ppb)	1.6 ± 1.9	1.8 ± 2.2
Monoterpenes (ppb)	0.3 ± 0.2	0.3 ± 0.3
Benzene (ppb)	0.04 ± 0.02	0.04 ± 0.02
NO (ppt)	36.0 ± 56.9	24.1 ± 36.4
NO <sub>2</sub> (ppt)	564.8 ± 393.2	593.0 ± 448.8
O <sub>3</sub> (ppb)	32.2 ± 12.2	24.1 ± 11.9
Temperature (°C) <sup>c</sup>	20.7 ± 4.1	20.6 ± 4.6
Relative Humidity (%)	71.5 ± 17.3	73.8 ± 17.3

<sup>a</sup> Summary statistics for NR-PM<sub>1</sub> measurements were calculated using 5-minute averaged data, VOC measurements using 1-minute averaged data (University of Minnesota’s PTR-QiToF 5-m inlet and 34-m inlet), NO<sub>x</sub> above-canopy measurements using 5-minute averaged data (University of Toronto), NO<sub>x</sub> and O<sub>3</sub> below-canopy measurements using 5-minute averaged data (MAQL), and O<sub>3</sub> 1-minute averaged data (CU-Boulder’s 27-m inlet on the AmeriFlux Tower).

<sup>b</sup> Percent of data below detection limit for NR-PM<sub>1</sub>: NH<sub>4</sub> (4% / 4%) and Chl (59% / 63%), VOCs: isoprene (30% / 30%), monoterpenes (56% / 36%), benzene (100% / 100%), and NO (37% / 22%), where percentages are shown for above- and below-canopy data, respectively. Unless otherwise stated, the percentage below detection limit for all other parameters is 0%.

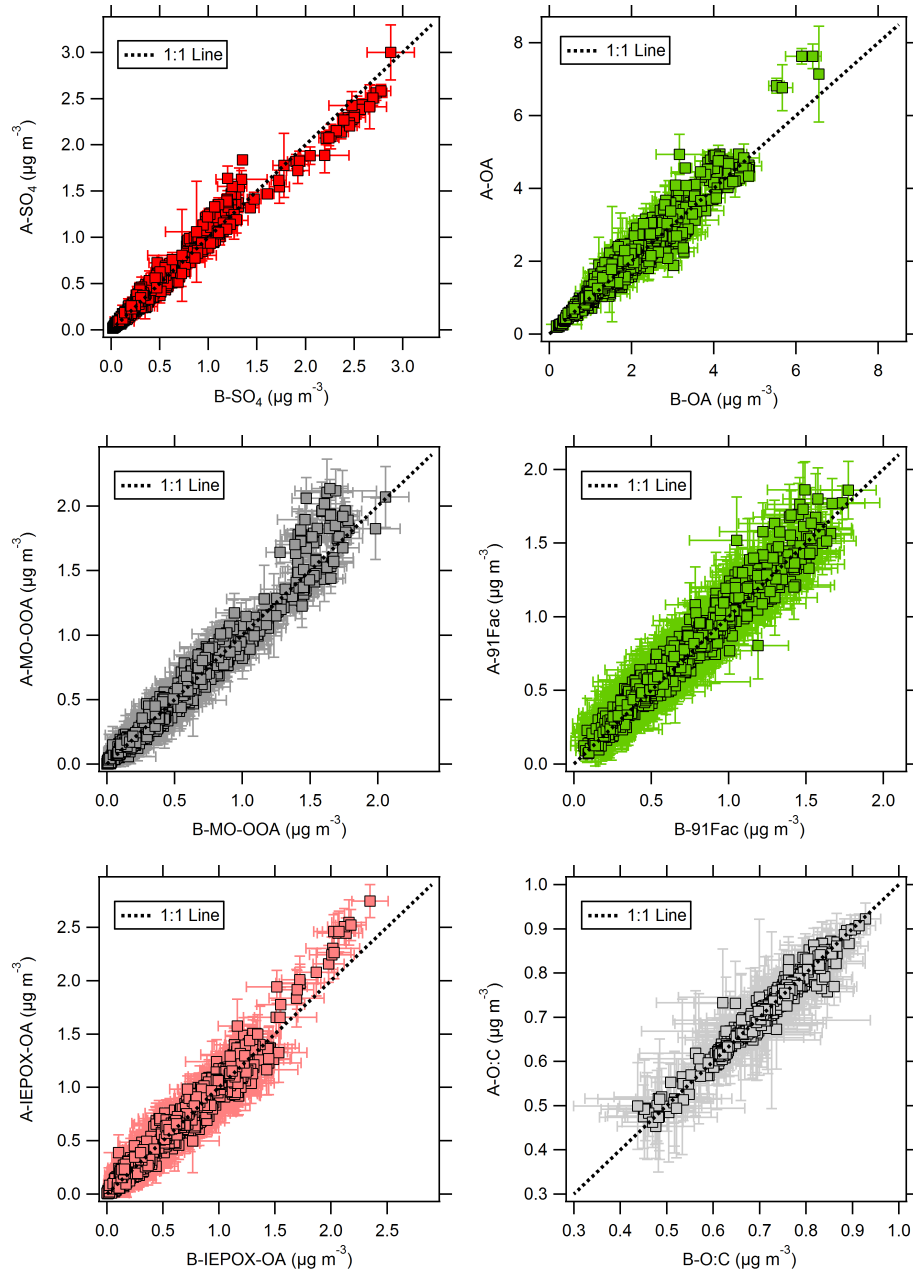
<sup>c</sup> Table entries where the null hypothesis (“equal medians”) between the above- and below-canopy data cannot be rejected and was not statistically significant (p > 0.05) using a two-sided, non-parametric Wilcoxon rank-sum test. All other table entries indicate that the null hypothesis (“equal medians”) can be rejected at the statistical significance level (p < 0.05).

concentration near the mid-canopy. Correlation analysis of these secondary oxidation products and HCOOH indicates that HCOOH likely originates from a secondary source in this environment (Alwe et al., 2019). In the present study, the overall homogeneity in OA factors implies that, despite vertical gradients in trace gases and BVOCs from primary emission and secondary production, turbulent mixing of aerosols between the forest canopy and the atmosphere is efficient.

Furthermore, this similarity between the above- and below-canopy environments suggests that the chemical timescales of SOA formation processes likely are long relative to residence times due to turbulent mixing (Foken et al., 2012), assuming relatively constant background levels. Ultimately, the observed similarity agrees with previous modeling work that predicts similar SOA mass loadings at these two heights (Ashworth et al., 2015; Schulze et al., 2017). The results shown in the present work also are in agreement with measurements at the site in 2009, showing that aerosol gradients on the PROPHET

499 tower “existed at times between the above-canopy (31.4m) and understory environments (5 m),” but that the understory  
 500 conditions were generally similar to that of the above-canopy conditions (VanReken et al., 2015). The vertical similarity in  
 501

502



503

504

505 **Figure 4: Scatter plots of above- (A) and below- (B) canopy hourly-averaged values for: (top from left to right) SO<sub>4</sub>**  
 506 **and total OA, (middle from left to right) MO-OOA and 91Fac, and (bottom) IEPOX-OA and campaign O:C. The**  
 507 **three-way valve switched between the two inlets at 10-minute intervals, so 30-minute averaged mass concentrations**  
 508 **allows for a comparison of both inlets on the same time basis. Above-canopy values are plotted on the y-axis of each**  
 509 **plot and below-canopy values are plotted on the x-axis of each plot. Averages are shown with squares; whiskers**



510 represent one standard deviation from the mean. For reference, a 1:1 line is shown with each plot. Outliers on the  
511 O:C plot corresponding to periods of precipitation were removed.

512  
513 NR-PM<sub>1</sub> is also in agreement with findings in the Amazon forest, where a balance between upward and downward fine  
514 particle fluxes was found (Rizzo et al., 2010) and in a southeast Asian rainforest, where PM<sub>1</sub> did not show significant  
515 variations with height during the daytime (Whitehead et al., 2010).

### 516 3.4.2 Episodes of vertical differences in NR-PM<sub>1</sub>

517 Episodes of vertical differences in NR-PM<sub>1</sub> between the two inlets were observed, and four such episodes are described  
518 here: Episode #1 (Period: 7/3/2016 19:30 to 7/5/2016 15:00 LT), Episode #2 (11 July 15:00 to 12 July 23:00 LT), Episode  
519 #3 (7/16/2016 21:30 to 7/19/2016 08:30 LT), and Episode #4 (7/26/2016 08:30 to 7/31/2016 13:30 LT). Episodes were  
520 defined as sustained periods in which a vertical difference in OA or SO<sub>4</sub> was greater than or equal to concentrations  
521 representing  $\geq 25\%$  of the campaign averaged OA or SO<sub>4</sub>. The “vertical difference,” symbolized as  $\Delta$  in Figure 5, is defined  
522 as the difference between above- and below-canopy values, where the positive convention indicates larger concentrations  
523 above the canopy. Episodes #1, #3, and #4 indicate increased above-canopy OA concentrations, while Episode #2 indicates  
524 increased below-canopy SO<sub>4</sub> concentrations. Episodes with higher above-canopy NR-PM<sub>1</sub> ranged up to  $\sim 1.0 \mu\text{g m}^{-3}$  higher  
525 in OA and  $\sim 0.5 \mu\text{g m}^{-3}$  higher in SO<sub>4</sub> relative to equivalent above/below mass ( $\Delta = 0$ ). Figure 5 shows time series of vertical  
526 differences in OA factors and SO<sub>4</sub> and estimates of friction velocity ( $u^*$ ) at five different heights (not equal to those for the  
527 VOC measurements) on the PROPHET tower over the campaign. Calculated from three-dimensional wind velocity data,  $u^*$   
528 is a function of the shear stress at the surface and is used in this study as a metric of in-canopy mixing (Equation 5).

529 The sum of the delta values for each OA factor ( $\sum\Delta\text{OA Factors}$ ) for the episodes are shown in Figure 6 with  
530 corresponding observations of  $u^*$  and vertical differences in SO<sub>4</sub> and O<sub>3</sub>. Figure 6 shows that vertical differences in OA  
531 correlate with vertical differences in SO<sub>4</sub> for Episodes #1 and #4, while  $\sum\Delta\text{OA Factors}$  during Episode #3 coincides with  
532 higher above-canopy O<sub>3</sub> concentrations. Correlations between chemical species for Episode #2 are weaker. Correlation  
533 coefficients ( $r$ ) for each episode are 0.69 (Episode #1  $\sum\Delta\text{OA Factors}$ -Sulfate), -0.44 (Episode #2 Sulfate- $u^*$ ), 0.34 (Episode  
534 #3  $\sum\Delta\text{OA Factors}$ -Ozone), and 0.59 (Episode #4  $\sum\Delta\text{OA Factors}$ -Sulfate). These episodes have different OA factor  
535 composition: MO-OOA contributing a larger percentage during Episode #1 and #2, 91Fac contributing roughly half in  
536 Episode #3, and IEPOX-OA contributing more than half of the total OA vertical difference in Episode #4.

### 537 3.5 The role of mixing and particulate SO<sub>4</sub> in vertical differences in OA

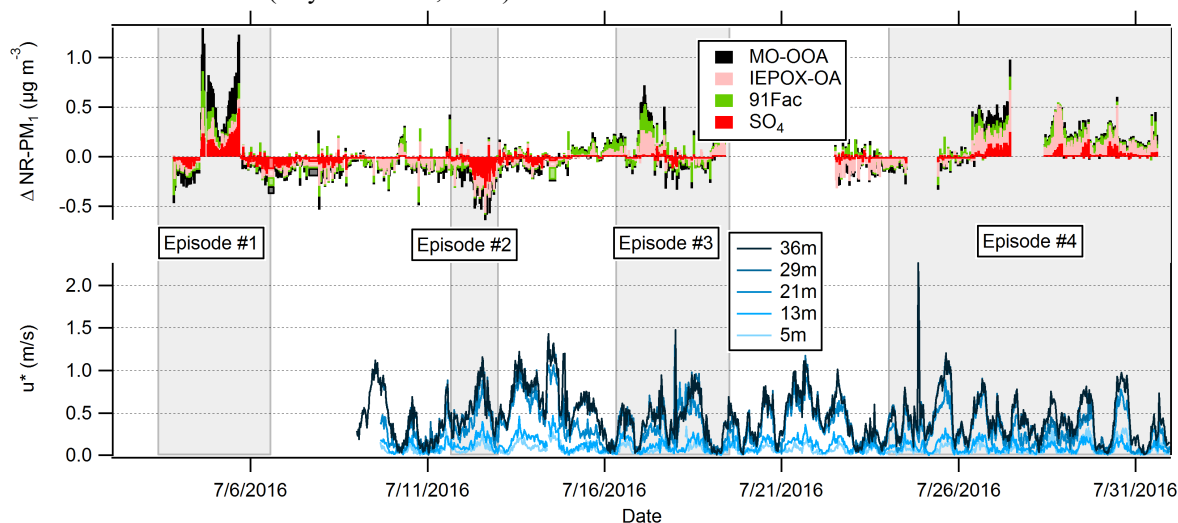
538 The diurnal profile of vertical difference episodes for total OA, SO<sub>4</sub>, and OA factors in 2016 indicate there is no clear  
539 diurnal pattern observed for any of these species (Figure S26 in the SI). Maximum vertical differences for SO<sub>4</sub>, total OA,  
540 and MO-OOA are observed around 3PM. For total OA, a gradual increase occurs at 9AM before reaching a maximum  
541 vertical difference at 3PM, suggesting that the episodes of higher above-canopy NR-PM<sub>1</sub> begin shortly after the most  
542 common observed hours of canopy uncoupling from the site in 2009. The initiation of these events may also be attributed to  
543 venting of the nocturnal boundary layer as it breaks up in the morning hours, as observed for events of upward particle  
544 number fluxes by Whitehead et al. (2010).

545 To assess the agreement between the occurrence of episodes and micrometeorological measurements of in-canopy  
546 mixing, episode-specific data for Episodes #2-4 are shown in Figure 6. During Episode #1,  $u^*$  data are available only at  
547 36m, so no friction velocity data are shown, preventing a full analysis similar to those performed for the other three episodes.  
548 However, based on similarity in sulfate enhancements, it can be assumed that Episode #1 is somewhat similar to Episode #4.

549 During Episode #2, based on the backward trajectory cluster analysis shown in Table 1 and the WPSCF results, regional  
550 transport is the likely source of the PM, including the enhancements below canopy. Prior to this episode, SO<sub>4</sub> and MO-OOA  
551 are uniform from below-to-above canopy. The below-canopy enhancement then reaches up to  $0.3 \mu\text{g m}^{-3}$  of SO<sub>4</sub>.



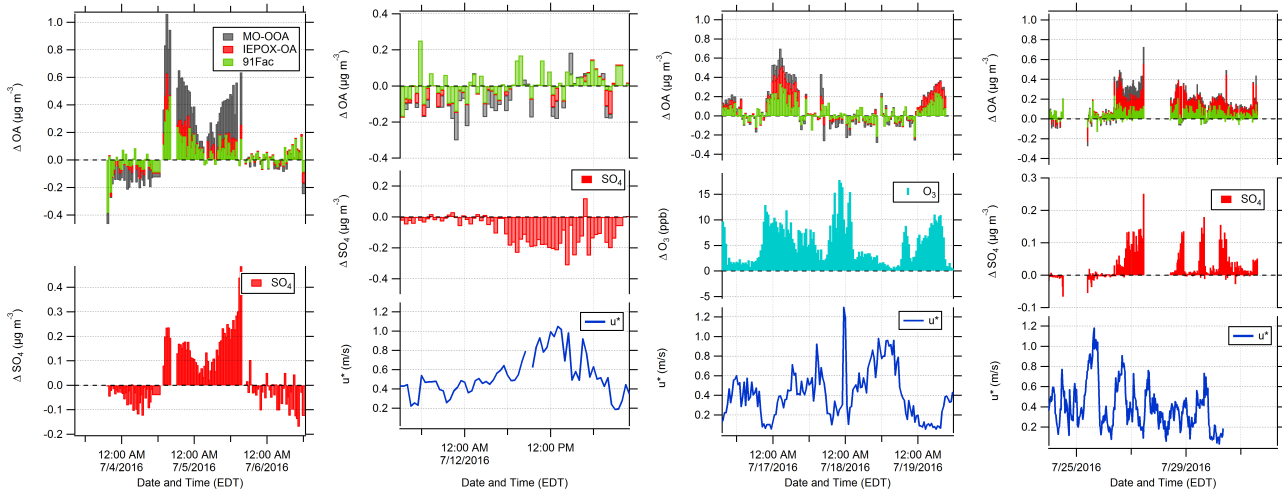
Downmixing of clean air from aloft could lead to lower concentrations above canopy, meaning that the observed difference would be caused by decreasing above-canopy concentrations, not increasing below-canopy concentrations. However, below-canopy concentrations are observed to increase while above-canopy concentrations decrease. Without a local, below-canopy source, this indicates that pollutants from above canopy are mixed into the below-canopy region at a rate faster than they are lost to deposition. Regional pollutants, advected to the site above the boundary layer, can be mixed down to the surface with daytime boundary layer growth, as has been found in previous aircraft campaigns (Berkowitz et al., 1998; Thornberry et al., 2001). Downward transport of air masses from the surrounding region has also been hypothesized to contribute to higher ratios of organic nitrogen to organic carbon ratios in water-soluble aerosols within a forest canopy relative to its forest floor (Miyazaki et al., 2014).



**Figure 5: Time series of (top) observed NR-PM<sub>1</sub> vertical differences between above- and below-canopy inlets and (bottom) friction velocity ( $u^*$ ) in m/s at five different heights on the PROPHET tower. Vertical differences are defined as  $\Delta$  Species, where  $\Delta$  refers to the Above-Canopy minus the Below-Canopy mass concentration. Note that the elevations for  $u^*$  are slightly different than those for trace gas sampling.**

During Episodes #3 and #4, periods of above-canopy enhancement in OA factors coincide with relatively lower  $u^*$  ( $u^* < 0.2$  m/s during Episode #3 and  $u^* < 0.6$  m/s during Episode #4, with the average  $u^*$  considering all heights over the duration of the episode), with lower friction velocities and greater above-canopy enhancement in Episode 3. In the case of Episode #3, lower in-canopy mixing, mostly occurring during nighttime, agrees well with periods of higher above-canopy OA factor and  $O_3$  concentrations. In this case, both IEPOX-OA and 91Fac contribute to the OA enhancement, likely due to strong photochemistry both locally and during transport (as indicated by increased  $O_3$ ). Episode #4 is the longest duration episode; however, the agreement between low-mixing and above-canopy PM enhancements is more variable during this case. Similar to Episode #3, the lowest values of friction velocity occur during the nighttime or early morning. Despite the temporal misalignment of  $\Delta$ OA and low  $u^*$ , it should be noted that friction velocities lower than the campaign average ( $\sim 0.4$  m/s) are observed during the latter periods of Episode #4 (7/28-7/30), which is consistent with the low mixing hypothesis presented for Episode #3. Therefore, it appears that stagnant conditions above the canopy created an environment where canopy exchange became limited and air masses did not fully penetrate into the canopy; aerosol deposition below canopy potentially enhanced the positive delta values. This scenario could also promote in-canopy OA accumulation, but this does not appear to have occurred, implying that other factors contributed to these vertical differences. Instead, it appears that increased sulfate loading was associated with this transport and that this increased sulfate above the canopy led to enhancements above

582 canopy, particularly of IEPOX-OA relative to 91Fac, as shown in Figure 6. This also could be related to associated changes  
 583 in aerosol liquid water.  
 584



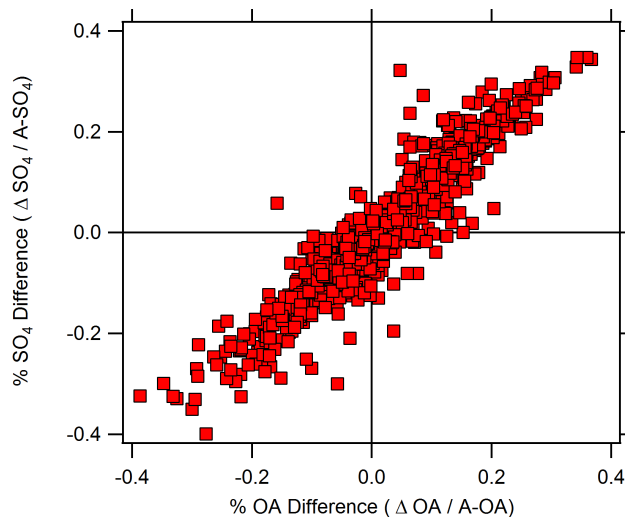
585  
 586 **Figure 6: Time series of observed NR-PM<sub>1</sub> vertical difference episodes: (from left to right) Episode #1, Episode #2,**  
 587 **Episode #3, and Episode #4. Data shown for Episode #1, Episode #2, and Episode #4 include vertical differences in**  
 588 **30-minute averaged above- and below-canopy SO<sub>4</sub>, while data shown for Episode #3 show vertical differences in 30-**  
 589 **minute averaged above- and below-canopy O<sub>3</sub>. The O<sub>3</sub> data were measured from the AmeriFlux Tower at 6 meters**  
 590 **and 27 meters and were provided courtesy of CU-Boulder. Friction velocity measurements at 29m on the PROPHET**  
 591 **tower are also provided for Episodes #2-4.**

592  
 593 The probability distributions of  $\Delta\text{SO}_4$  and  $\Delta\text{O}_3$  between episodes and the average for the entire PROPHET campaign are  
 594 compared in Figure S27 of the SI. The cumulative distribution function (CDF) for  $\Delta\text{SO}_4$  indicates that Episodes #1 and #4  
 595 had higher probabilities of positive  $\Delta\text{SO}_4$  (denoting higher above-canopy concentrations), while Episode #2 had higher  
 596 probabilities of negative  $\Delta\text{SO}_4$  (denoting higher below-canopy concentrations). The episode-specific CDFs for #1 and #2 are  
 597 also different from the total campaign CDF, implying that the vertical differences observed were distinct events. The  
 598 average O<sub>3</sub> difference between the 6- and 27-m inlets on the AmeriFlux tower throughout the campaign indicated that O<sub>3</sub>  
 599 was on average 5 ppb greater at the 27-m inlet, with the difference reaching as high as 34 ppb on 6 July and 20 July. The  
 600 probability distribution of vertical O<sub>3</sub> differences observed during Episode #3 is markedly different than those observed  
 601 during Episode #1, Episode #4, and the remainder of the campaign. This indicates that O<sub>3</sub> had a larger probability of being  
 602 between 0-10 ppb higher above the canopy during this episode compared to the rest of the campaign.

603 Based on Episode #1 and Episode #4, one of several potential driving factors of vertical differences for OA is a vertical  
 604 difference in particulate SO<sub>4</sub>. Figure 7 indicates the vertical differences in OA and SO<sub>4</sub> relative to their respective above-  
 605 canopy concentrations. This metric is defined as “% PM Difference,” where PM is the given PM constituent and is equal to  
 606 the  $\Delta\text{PM}$  constituent divided by the above-canopy PM concentration. Data that fall in the upper right and bottom left  
 607 quadrants correspond to higher above- and higher below-canopy concentrations, respectively. The vertical difference above  
 608 or below the canopy can be up to approximately 35% of the total available SO<sub>4</sub> or OA at the site. Figure 7 indicates that  
 609 there is a strong linear relationship between the %SO<sub>4</sub> Difference and %OA Difference during the campaign, even outside of  
 610 the Episodes defined here.

611 The concurrent features of vertical differences in OA factors and SO<sub>4</sub> observed in Episodes #1, #2, and #4 and OA  
 612 factors and O<sub>3</sub> in Episode #3 suggest that long-range regional transport (enhanced MO-OOA, SO<sub>4</sub>, and/or O<sub>3</sub>) is a key first  
 613 step in causing these differences. Local through-canopy mixing then determines whether the enhancement is above

614 (Episodes #1, #3, and #4) or below (Episode #2) the canopy. The transported material can then impact local chemistry.  
 615 Increased  $\text{SO}_4$  likely increases aerosol hygroscopicity resulting in increased ALW, which could result in increased  
 616 partitioning of semi-volatile organics to the condensed phase (i.e., IEPOX-OA) (Budisulistiorini et al., 2017; El-Sayed et al.,  
 617 2018; Marais et al., 2016). Pre-existing OA (i.e., MO-OOA) transported with  $\text{SO}_4$  aerosol could also play a role in the  
 618 partitioning of OA above the canopy during these episodes. Vertical gradients in  $\text{O}_3$  coinciding with Episode #3 (higher  $\text{O}_3$   
 619 above) could initiate relatively local formation of extremely low volatility compounds, which are first-generation oxidation  
 620 products of  $\alpha$ -pinene and  $\text{O}_3$  (Jokinen et al., 2015). Based on the WPSCF and backward trajectory clustering results, cooler,  
 621 northerly air masses associated with IEPOX-OA also could promote partitioning of semi-volatile organics to the particle  
 622 phase. It is important to note that it is difficult to conclude the exact chemical mechanisms that are influencing these events  
 623 due to their low mass loadings and episodic nature and the consistently higher above-canopy  $\text{O}_3$  concentrations during the  
 624 campaign.  
 625



626  
 627 **Figure 7: Scatter plot of % difference in  $\text{SO}_4$  (y-axis) and OA (x-axis) for the entire campaign. The % difference is**  
 628 **calculated as  $\Delta \text{Species} / \text{Above-Canopy Species concentration}$ , and is representative of the normalized  $\Delta \text{Species}$**   
 629 **concentration.**  
 630

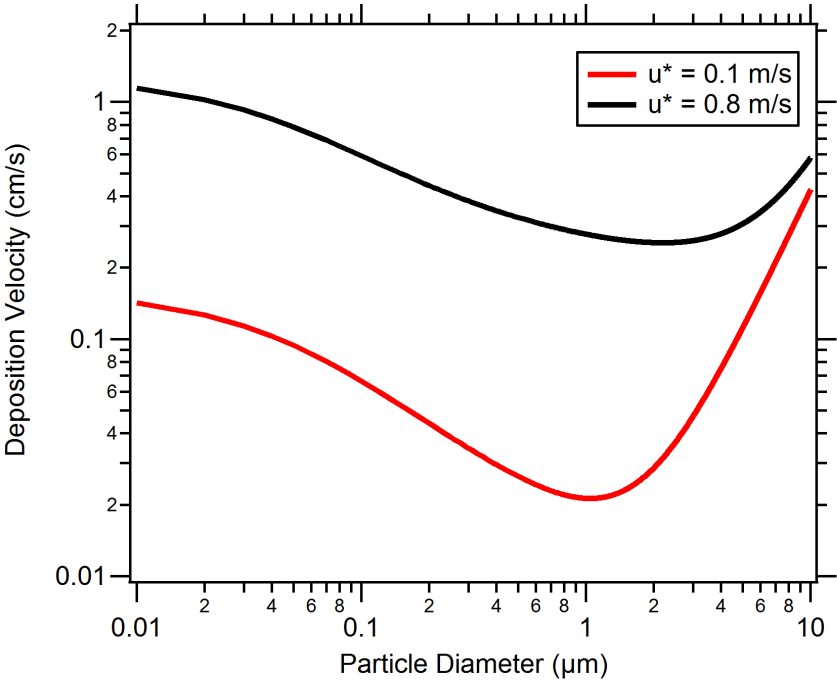
631 Understanding these events is important because uncoupled forest-canopy conditions have been observed in a number of  
 632 locations (Foken et al., 2012; Whitehead et al., 2010) and could indicate differences between above-canopy and surface level  
 633 PM. One-dimensional modeling has revealed that the timescales of turbulent transport inside a forest canopy can be much  
 634 shorter (minutes) than the timescale of aerosol dynamics and deposition (hours) (Rannik et al., 2016), but this evaluation  
 635 suggests that advective episodes can cause vertical gradients that are not locally driven. Above-canopy OA and particle  
 636 fluxes have been observed in other studies (Farmer et al., 2013; Pryor et al., 2007), suggesting the need for careful evaluation  
 637 of whether differences are driven by local chemistry, long-range transport, or a combination thereof.  
 638

### 639 3.6 Particle dry deposition model

640 To investigate broadly the effects of canopy mixing on particle deposition in a forest canopy, a particle dry deposition  
 641 model is used. Note that we are not applying this through the canopy but to illustrate the relationship between deposition  
 642 and  $u^*$ . The resistance model for particle dry deposition assumes that the deposition process is controlled by three  
 643 resistances in series: aerodynamic, quasi-laminar, and canopy resistance (Seinfeld and Pandis, 2006). Details on the particle

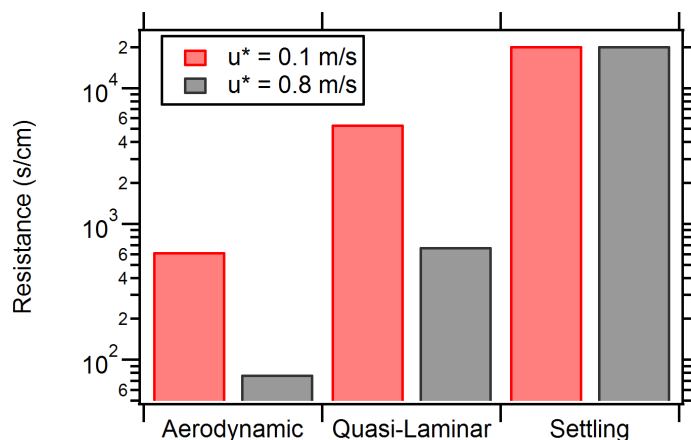
dry deposition model are included in the SI. Conditions and parameters representative of the land use category, season, and forest canopy present at the PROPHET site are used as model inputs.

Figure 8 displays model results for deposition velocity as a function of particle diameter under stable atmospheric conditions. For illustration, test cases are shown for  $u^*$  values for low in-canopy mixing ( $u^* = 0.1$  m/s) and high in-canopy mixing ( $u^* = 0.8$  m/s). Values of  $u^*$  were chosen based on the range of  $u^*$  values observed during the campaign at 29m on the PROPHET tower. For the submicron particle diameter range (0.1 to 1 micron), a characteristic minimum in deposition velocity is observed for both test cases at 1 micron diameter. Deposition modeling indicates that higher deposition velocities are achieved in the high in-canopy mixing test case, implying that there is less canopy transport resistance to the surface and more deposition in the canopy. This corresponds to Episode #2.



**Figure 8: Particle dry deposition velocity resistance model results plotted by particle diameter and friction velocity. Two test cases of low and high friction velocity, representative of data collected at the PROPHET site, are plotted in red and black solid lines, respectively. Land-use conditions and seasonal parameters similar to the PROPHET campaign were used as inputs to the particle dry deposition model. Additional model parameters are provided in the SI.**

Deposition modeling also indicates relatively lower deposition velocities for the low mixing condition test case, which suggests that particles aloft are transported less efficiently into the forest (similar to the conditions present for Episodes #3 and #4). A comparison of the deposition resistances (shown in Figure 9) indicates that aerodynamic and quasi-laminar resistances in the low-mixing test case are an order of magnitude greater than the high-mixing case. The settling removal resistance is the same between both test cases because it is merely a function of diameter. A greater aerodynamic resistance limits turbulent transport from the above-canopy layer to the surface layer and greater quasi-laminar layer resistance limits transport to just above the surface by lowering particle impactation. In total, this result emphasizes the importance of in-canopy transport and mixing in governing particle concentrations in forests.



**Figure 8: Deposition and particle settling (accounts for the effect of sedimentation) resistances comparison for low and high friction velocities. Resistance comparison assumes a 1 micron particle diameter.**

#### 4 Conclusions

In this study, source apportionment of OA using separate PMF analyses on inlets situated above and below a forest canopy resulted in three OA factors at a site in northern Michigan. Similarity in OA composition, concentration, and diurnal profiles was observed between the two inlets, suggesting that turbulent transport efficiently mixes OA across the canopy. However, OA factor vertical differences between the two inlets were observed during four separate episodes. During these episodes, vertical differences were both positive (greater concentrations above the canopy) and negative (greater concentrations below canopy). NR-PM<sub>1</sub> composition amongst these episodes was unique (Episode #1: MO-OOA dominant, Episode #2: SO<sub>4</sub> dominant, Episode #3: 91Fac dominant, Episode #4: IEPOX-OA dominant). Furthermore, the vertical difference concentrations can be 35% of the total available SO<sub>4</sub> or OA at the site. It has also been shown that relative vertical differences in SO<sub>4</sub> and OA are linearly associated, suggesting the role of regional SO<sub>4</sub> in vertical profiles of NR-PM<sub>1</sub> in forested environments.

Using micrometeorological measurements, it has been shown in this work that periods of low in-canopy mixing conditions can be associated with the timing of vertical difference episodes where above-canopy concentrations are larger than those below. The opposite scenario is hypothesized to be associated with transport of air from aloft or with accumulation of material below canopy more rapidly than it is deposited. Either way, these results suggest that canopy mixing impacts particle levels. However, under low-mixing scenarios, it appears that enhancements as defined here depend on both these conditions and the presence of transported O<sub>3</sub> or SO<sub>4</sub> to enhance IEPOX-OA and 91Fac levels.

To the knowledge of the author, this is one of only a few studies that have assessed vertical profiles in OA factors above and below a forest canopy, and the first study to observe episodes of vertical differences in OA factors above and below a forest canopy. This work adds to the existing literature on aerosol chemistry in a forest canopy environment presented by Rizzo et al. (2010) and Whitehead et al. (2010) and to literature on vertical profiles of OA in an urban environment by Öztürk et al. (2013).

To investigate the effects of forest canopies on SOA formation, small-scale models, such as those described in Schulze et al. (2017) and Ashworth et al. (2015), have been developed. The OA data from this work can be used to validate such models and are particularly relevant to these modeling efforts, as the models were developed using campaign data from the PROPHET site in 2009. Vertical transport and horizontal advection are both sources of uncertainty in current forest canopy-atmosphere exchange models, which are designed to focus on local processes. Ultimately, the vertical similarity in

700 NR-PM<sub>1</sub> OA composition observed in this study implies that it may be valid to assume that below-canopy OA composition  
701 is generally representative of the OA composition in the atmospheric layer directly above the canopy and vice versa.  
702 However, this work highlights that advection of regional pollution into forested regions can lead to in-canopy gradients that  
703 are not present under purely local conditions.  
704

705 **Data Availability**

706 Data are available through contacting the corresponding author.

707 **Author Contributions**

708 A.A.T.B. prepared the manuscript with input from all authors. A.A.T.B., H.W.W. and R.J.G. conceived of the study and  
709 operated and analyzed the data from the HR-ToF-AMS. S.K. and A.L.S. investigated friction velocity and in-canopy  
710 mixing. H.D.A. and D.B.M. collected and analyzed VOC data. J.H.F., M.H.E. and S.A. collected and analyzed trace gas  
711 data and operated the MAQL.

712 **Competing Interests**

713 The authors declare they have no conflict of interest.

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721 **References**

722 Alwe, H.D., Millet, D.B., Chen, X., Raff, J.D., Payne, Z.C., Fledderman, K., 2019. Oxidation of Volatile Organic  
723 Compounds as the Major Source of Formic Acid in a Mixed Forest Canopy. *Geophysical Research Letters* 0.  
724 <https://doi.org/10.1029/2018GL081526>  
725 Ashworth, K., Chung, S.H., Griffin, R.J., Chen, J., Forkel, R., Bryan, A.M., Steiner, A.L., 2015. FORest Canopy  
726 Atmosphere Transfer (FORCAsT) 1.0: a 1-D model of biosphere–atmosphere chemical exchange. *Geoscientific*  
727 *Model Development* 8, 3765–3784. <https://doi.org/10.5194/gmd-8-3765-2015>  
728 Baldocchi, D., Guenther, A.B., Harley, P., Klinger, L., Zimmerman, P., Lamb, B., Westberg, H., 1995. The fluxes and air  
729 chemistry of isoprene above a deciduous hardwood forest. *Philosophical Transactions: Physical Sciences and*  
730 *Engineering* 351, 279–296.  
731 Bergen, K.M., Dronova, I., 2007. Observing succession on aspen-dominated landscapes using a remote sensing-ecosystem  
732 approach. *Landscape Ecology* 22, 1395–1410. <https://doi.org/10.1007/s10980-007-9119-1>  
733 Berkowitz, C.M., Fast, J.D., Springston, S.R., Larsen, R.J., Spicer, C.W., Doskey, P.V., Hubbe, J.M., Plastringe, R., 1998.  
734 Formation mechanisms and chemical characteristics of elevated photochemical layers over the northeast United  
735 States. *J. Geophys. Res.* 103, 10631–10647. <https://doi.org/10.1029/97JD03751>

736 Bondy, A.L., Wang, B., Laskin, A., Craig, R.L., Nhliziyo, M.V., Bertman, S., Pratt, K.A., Shepson, P.B., Ault, A.P., 2017.  
737 Inland Sea Spray Aerosol Transport and Incomplete Chloride Depletion: Varying Degrees of Reactive Processing  
738 Observed during SOAS. *Environ. Sci. Technol.* <https://doi.org/10.1021/acs.est.7b02085>

739 Brown, S.S., Osthoff, H.D., Stark, H., Dubé, W.P., Ryerson, T.B., Warneke, C., de Gouw, J.A., Wollny, A.G., Parrish, D.D.,  
740 Fehsenfeld, F.C., Ravishankara, A.R., 2005. Aircraft observations of daytime NO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub> and their implications  
741 for tropospheric chemistry. *Journal of Photochemistry and Photobiology A: Chemistry* 176, 270–278.  
742 <https://doi.org/10.1016/j.jphotochem.2005.10.004>

743 Bryan, A.M., Bertman, S.B., Carroll, M.A., Dusanter, S., Edwards, G.D., Forkel, R., Griffith, S., Guenther, A.B., Hansen,  
744 R.F., Helmig, D., Jobson, B.T., Keutsch, F.N., Lefer, B.L., Pressley, S.N., Shepson, P.B., Stevens, P.S., Steiner,  
745 A.L., 2012. In-canopy gas-phase chemistry during CABINEX 2009: sensitivity of a 1-D canopy model to vertical  
746 mixing and isoprene chemistry. *Atmospheric Chemistry and Physics* 12, 8829–8849. [https://doi.org/10.5194/acp-](https://doi.org/10.5194/acp-12-8829-2012)  
747 12-8829-2012

748 Budisulistiorini, S.H., Nenes, A., Carlton, A.G., Surratt, J.D., McNeill, V.F., Pye, H.O.T., 2017. Simulating Aqueous-Phase  
749 Isoprene-Epoxydiol (IEPOX) Secondary Organic Aerosol Production During the 2013 Southern Oxidant and  
750 Aerosol Study (SOAS). *Environmental Science & Technology* 51, 5026–5034.  
751 <https://doi.org/10.1021/acs.est.6b05750>

752 Canagaratna, M.R., Jimenez, J.L., Kroll, J.H., Chen, Q., Kessler, S.H., Massoli, P., Hildebrandt Ruiz, L., Fortner, E.,  
753 Williams, L.R., Wilson, K.R., Surratt, J.D., Donahue, N.M., Jayne, J.T., Worsnop, D.R., 2015. Elemental ratio  
754 measurements of organic compounds using aerosol mass spectrometry: characterization, improved calibration, and  
755 implications. *Atmospheric Chemistry and Physics* 15, 253–272. <https://doi.org/10.5194/acp-15-253-2015>

756 Carlton, A.G., Baker, K.R., 2011. Photochemical Modeling of the Ozark Isoprene Volcano: MEGAN, BEIS, and Their  
757 Impacts on Air Quality Predictions. *Environmental Science & Technology* 45, 4438–4445.  
758 <https://doi.org/10.1021/es200050x>

759 Carroll, M.A., Bertman, S.B., Shepson, P.B., 2001. Overview of the Program for Research on Oxidants: PHotochemistry,  
760 Emissions, and Transport (PROPHET) summer 1998 measurements intensive. *Journal of Geophysical Research*  
761 106.

762 Chang, Y., Deng, C., Cao, F., Cao, C., Zou, Z., Liu, S., Lee, X., Li, J., Zhang, G., Zhang, Y., 2017. Assessment of  
763 carbonaceous aerosols in Shanghai, China – Part 1: long-term evolution, seasonal variations, and meteorological  
764 effects. *Atmos. Chem. Phys.* 17, 9945–9964. <https://doi.org/10.5194/acp-17-9945-2017>

765 Cooper, O.R., Moody, J.L., Thornberry, T.D., Town, M.S., Carroll, M.A., 2001. PROPHET 1998 meteorological overview  
766 and air-mass classification. *Journal of Geophysical Research* 106.

767 Cubison, M.J., Ortega, A.M., Hayes, P.L., Farmer, D.K., Day, D., Lechner, M.J., Brune, W.H., Apel, E., Diskin, G.S.,  
768 Fisher, J.A., Fuelberg, H.E., Hecobian, A., Knapp, D.J., Mikoviny, T., Riemer, D., Sachse, G.W., Sessions, W.,  
769 Weber, R.J., Weinheimer, A.J., Wisthaler, A., Jimenez, J.L., 2011. Effects of aging on organic aerosol from open  
770 biomass burning smoke in aircraft and laboratory studies. *Atmospheric Chemistry and Physics* 11, 12049–12064.  
771 <https://doi.org/10.5194/acp-11-12049-2011>

772 DeCarlo, P.F., Kimmel, J.R., Trimborn, A., Northway, M.J., Jayne, J.T., Aiken, A.C., Gonin, M., Fuhrer, K., Horvath, T.,  
773 Docherty, K.S., others, 2006. Field-deployable, high-resolution, time-of-flight aerosol mass spectrometer.  
774 *Analytical Chemistry* 78, 8281–8289.

775 Ditto, J.C., Barnes, E.B., Khare, P., Takeuchi, M., Joo, T., Bui, A.A.T., Lee-Taylor, J., Eris, G., Chen, Y., Aumont, B.,  
776 Jimenez, J.L., Ng, N.L., Griffin, R.J., Gentner, D.R., 2018. An omnipresent diversity and variability in the chemical  
777 composition of atmospheric functionalized organic aerosol. *Communications Chemistry* 1, 75.  
778 <https://doi.org/10.1038/s42004-018-0074-3>

779 Draxler, R.R., Hess, G.D., 1998. An overview of the HYSPLIT\_4 modeling system for trajectories, dispersion, and  
780 deposition. *Australian meteorological magazine* 47.

781 El-Sayed, M.M.H., Ortiz-Montalvo, D.L., Hennigan, C.J., 2018. The effects of isoprene and NO<sub>x</sub> on secondary organic  
782 aerosols formed through reversible and irreversible uptake to aerosol water. *Atmospheric Chemistry and Physics*  
783 18, 1171–1184. <https://doi.org/10.5194/acp-18-1171-2018>

Farmer, D.K., Chen, Q., Kimmel, J.R., Docherty, K.S., Nemitz, E., Artaxo, P.A., Cappa, C.D., Martin, S.T., Jimenez, J.L., 2013. Chemically Resolved Particle Fluxes Over Tropical and Temperate Forests. *Aerosol Science and Technology* 47, 818–830. <https://doi.org/10.1080/02786826.2013.791022>

Fehsenfeld, F.C., Calvert, J.G., Fall, R., Goldan, P., Guenther, A., Hewitt, C.N., Lamb, B., Shaw, L., Trainer, M., Westberg, H., Zimmerman, P., 1992. Emissions of volatile organic compounds from vegetation and the implications for atmospheric chemistry. *Global Biogeochemical Cycles* 6.

Finnigan, J., 2000. Turbulence in plant canopies. *Annual Review of Fluid Mechanics* 32, 519–571.

Foken, T., Meixner, F.X., Falge, E., Zetzsch, C., Serafimovich, A., Bargsten, A., Behrendt, T., Biermann, T., Breuninger, C., Dix, S., Gerken, T., Hunner, M., Lehmann-Pape, L., Hens, K., Jocher, G., Kesselmeier, J., Lüers, J., Mayer, J.-C., Moravek, A., Plake, D., Riederer, M., Rütz, F., Scheibe, M., Siebicke, L., Sörgel, M., Staudt, K., Trebs, I., Tsokankunku, A., Welling, M., Wolff, V., Zhu, Z., 2012. Coupling processes and exchange of energy and reactive and non-reactive trace gases at a forest site – results of the EGER experiment. *Atmospheric Chemistry and Physics* 12, 1923–1950. <https://doi.org/10.5194/acp-12-1923-2012>

Fuentes, J.D., Wang, D., Bowling, D.R., Potosnak, M., Monson, R.K., Goliff, W.S., Stockwell, W.R., 2007. Biogenic Hydrocarbon Chemistry within and Above a Mixed Deciduous Forest. *Journal of Atmospheric Chemistry* 56, 165–185. <https://doi.org/10.1007/s10874-006-9048-4>

Gao, W., Wesely, M.L., Doskey, P.V., 1993. Numerical modeling of the turbulent diffusion and chemistry of NO<sub>x</sub>, O<sub>3</sub>, isoprene, and other reactive trace gases in and above a forest canopy. *Journal of Geophysical Research: Atmospheres* 98, 18339–18353. <https://doi.org/10.1029/93JD01862>

Geron, C., Rasmussen, R., R. Arnsts, R., Guenther, A., 2000. A review and synthesis of monoterpene speciation from forests in the United States. *Atmospheric Environment* 34, 1761–1781. [https://doi.org/10.1016/S1352-2310\(99\)00364-7](https://doi.org/10.1016/S1352-2310(99)00364-7)

Goldstein, A.H., Galbally, I.E., 2007. Known and unexplored organic constituents in the earth’s atmosphere. *Environmental Science & Technology*.

Gordon, M., Staebler, R.M., Liggio, J., Vlasenko, A., Li, S.-M., Hayden, K., 2011. Aerosol flux measurements above a mixed forest at Borden, Ontario. *Atmospheric Chemistry and Physics* 11, 6773–6786. <https://doi.org/10.5194/acp-11-6773-2011>

Guenther, A., Geron, C., Pierce, T., Lamb, B., Harley, P., Fall, R., 2000. Natural emissions of non-methane volatile organic compounds, carbon monoxide, and oxides of nitrogen from North America. *Atmospheric Environment* 34, 2205–2230.

Guenther, A., Hewitt, C.N., Erickson, D., Fall, R., Geron, C., Graedel, T., Harley, P., Klinger, L., Lerdau, M., McKay, W.A., Pierce, T., Scholes, B., Steinbrecher, R., Tallamraju, R., Taylor, J., Zimmerman, P., 1995. A global model of natural volatile organic compound emissions. *Journal of Geophysical Research* 100.

Gunsch, M.J., May, N.W., Wen, M., Bottenus, C.L.H., Gardner, D.J., VanReken, T.M., Bertman, S.B., Hopke, P.K., Ault, A.P., Pratt, K.A., 2018. Ubiquitous influence of wildfire emissions and secondary organic aerosol on summertime atmospheric aerosol in the forested Great Lakes region. *Atmospheric Chemistry and Physics* 18, 3701–3715. <https://doi.org/10.5194/acp-18-3701-2018>

Holzinger, R., Lee, A., Paw, K.T., Goldstein, U.A.H., 2005. Observations of oxidation products above a forest imply biogenic emissions of very reactive compounds. *Atmospheric Chemistry and Physics* 5, 67–75.

Hu, W.W., Campuzano-Jost, P., Palm, B.B., Day, D.A., Ortega, A.M., Hayes, P.L., Krechmer, J.E., Chen, Q., Kuwata, M., Liu, Y.J., de Sá, S.S., McKinney, K., Martin, S.T., Hu, M., Budisulistiorini, S.H., Riva, M., Surratt, J.D., St. Clair, J.M., Isaacman-Van Wertz, G., Yee, L.D., Goldstein, A.H., Carbone, S., Brito, J., Artaxo, P., de Gouw, J.A., Koss, A., Wisthaler, A., Mikoviny, T., Karl, T., Kaser, L., Jud, W., Hansel, A., Docherty, K.S., Alexander, M.L., Robinson, N.H., Coe, H., Allan, J.D., Canagaratna, M.R., Paulot, F., Jimenez, J.L., 2015. Characterization of a real-time tracer for isoprene epoxydiols-derived secondary organic aerosol (IEPOX-SOA) from aerosol mass spectrometer measurements. *Atmospheric Chemistry and Physics* 15, 11807–11833. <https://doi.org/10.5194/acp-15-11807-2015>

IPCC (Ed.), 2007. Climate change 2007: the physical science basis: contribution of Working Group I to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change. Cambridge University Press, Cambridge ; New York.



Jimenez, J.L., Canagaratna, M.R., Donahue, N.M., Prevot, A.S.H., Zhang, Q., Kroll, J.H., DeCarlo, P.F., Allan, J.D., Coe, H., Ng, N.L., Aiken, A.C., Docherty, K.S., Ulbrich, I.M., Grieshop, A.P., Robinson, A.L., Duplissy, J., Smith, J.D., Wilson, K.R., Lanz, V.A., Hueglin, C., Sun, Y.L., Tian, J., Laaksonen, A., Raatikainen, T., Rautiainen, J., Vaattovaara, P., Ehn, M., Kulmala, M., Tomlinson, J.M., Collins, D.R., Cubison, M.J., Dunlea, E., Huffman, J.A., Onasch, T.B., Alfarra, M.R., Williams, P.I., Bower, K., Kondo, Y., Schneider, J., Drewnick, F., Borrmann, S., Weimer, S., Demerjian, K., Salcedo, D., Cottrell, L., Griffin, R., Takami, A., Miyoshi, T., Hatakeyama, S., Shimojo, A., Sun, J.Y., Zhang, Y.M., Dzepina, K., Kimmel, J.R., Sueper, D., Jayne, J.T., Herndon, S.C., Trimborn, A.M., Williams, L.R., Wood, E.C., Middlebrook, A.M., Kolb, C.E., Baltensperger, U., Worsnop, D.R., 2009. Evolution of Organic Aerosols in the Atmosphere. *Science* 326, 1525–1529. <https://doi.org/10.1126/science.1180353>

Jokinen, T., Berndt, T., Makkonen, R., Kerminen, V.-M., Junninen, H., Paasonen, P., Stratmann, F., Herrmann, H., Guenther, A.B., Worsnop, D.R., Kulmala, M., Ehn, M., Sipilä, M., 2015. Production of extremely low volatile organic compounds from biogenic emissions: Measured yields and atmospheric implications. *Proceedings of the National Academy of Sciences* 112, 7123–7128. <https://doi.org/10.1073/pnas.1423977112>

Kanakidou, M., Seinfeld, J.H., Pandis, S.N., Barnes, I., Dentener, F.J., Facchini, M.C., Dingenen, R.V., Ervens, B., Nenes, A., Nielsen, C.J., others, 2005. Organic aerosol and global climate modelling: a review. *Atmospheric Chemistry and Physics* 5, 1053–1123.

Kroll, J.H., Donahue, N.M., Jimenez, J.L., Kessler, S.H., Canagaratna, M.R., Wilson, K.R., Altieri, K.E., Mazzoleni, L.R., Wozniak, A.S., Bluhm, H., Mysak, E.R., Smith, J.D., Kolb, C.E., Worsnop, D.R., 2011. Carbon oxidation state as a metric for describing the chemistry of atmospheric organic aerosol. *Nature Chemistry* 3, 133–139. <https://doi.org/10.1038/nchem.948>

Kruijt, B., Malhi, Y., Lloyd, J., Norbre, A.D., Miranda, A.C., Pereira, M.G.P., Culf, A., Grace, J., 2000. Turbulence Statistics Above And Within Two Amazon Rain Forest Canopies. *Boundary-Layer Meteorology* 94, 297–331. <https://doi.org/10.1023/A:1002401829007>

Leong, Y.J., Sanchez, N.P., Wallace, H.W., Cevik, B.K., Hernandez, C.S., Han, Y., Flynn, J.H., Massoli, P., Floerchinger, C., Fortner, E.C., Herndon, S., Bean, J.K., Ruiz, L.H., Jeon, W., Choi, Y., Lefer, B., Griffin, R.J., 2017. Overview of surface measurements and spatial characterization of submicrometer particulate matter during the DISCOVER-AQ 2013 campaign in Houston, TX. *Journal of the Air & Waste Management Association* 67, 854–872. <https://doi.org/10.1080/10962247.2017.1296502>

Makar, P.A., Staebler, R.M., Akingunola, A., Zhang, J., McLinden, C., Kharol, S.K., Pabla, B., Cheung, P., Zheng, Q., 2017. The effects of forest canopy shading and turbulence on boundary layer ozone. *Nature Communications* 8, 15243. <https://doi.org/10.1038/ncomms15243>

Marais, E.A., Jacob, D.J., Jimenez, J.L., Campuzano-Jost, P., Day, D.A., Hu, W., Krechmer, J., Zhu, L., Kim, P.S., Miller, C.C., Fisher, J.A., Travis, K., Yu, K., Hanisco, T.F., Wolfe, G.M., Arkinson, H.L., Pye, H.O.T., Froyd, K.D., Liao, J., McNeill, V.F., 2016. Aqueous-phase mechanism for secondary organic aerosol formation from isoprene: application to the southeast United States and co-benefit of SO<sub>2</sub> emission controls. *Atmospheric Chemistry and Physics* 16, 1603–1618. <https://doi.org/10.5194/acp-16-1603-2016>

Middlebrook, A.M., Bahreini, R., Jimenez, J.L., Canagaratna, M.R., 2012. Evaluation of Composition-Dependent Collection Efficiencies for the Aerodyne Aerosol Mass Spectrometer using Field Data. *Aerosol Science and Technology* 46, 258–271. <https://doi.org/10.1080/02786826.2011.620041>

Millet, D.B., Alwe, H.D., Chen, X., Deventer, M.J., Griffis, T.J., Holzinger, R., Bertman, S.B., Rickly, P.S., Stevens, P.S., Léonardis, T., Locoge, N., Dusanter, S., Tyndall, G.S., Alvarez, S.L., Erickson, M.H., Flynn, J.H., 2018. Bidirectional Ecosystem–Atmosphere Fluxes of Volatile Organic Compounds Across the Mass Spectrum: How Many Matter? *ACS Earth and Space Chemistry* 2, 764–777. <https://doi.org/10.1021/acsearthspacechem.8b00061>

Miyazaki, Y., Fu, P., Ono, K., Tachibana, E., Kawamura, K., 2014. Seasonal cycles of water-soluble organic nitrogen aerosols in a deciduous broadleaf forest in northern Japan. *J. Geophys. Res. Atmos.* 119, 1440–1454. <https://doi.org/10.1002/2013JD020713>

Mohr, C., DeCarlo, P.F., Heringa, M.F., Chirico, R., Slowik, J.G., Richter, R., Reche, C., Alastuey, A., Querol, X., Seco, R., Peñuelas, J., Jiménez, J.L., Crippa, M., Zimmermann, R., Baltensperger, U., Prévôt, A.S.H., 2012. Identification

and quantification of organic aerosol from cooking and other sources in Barcelona using aerosol mass spectrometer data. *Atmospheric Chemistry and Physics* 12, 1649–1665. <https://doi.org/10.5194/acp-12-1649-2012>

NADP, 2016. National Atmospheric Deposition Program, National Trends Network (NTN) Site Details [WWW Document]. URL <http://nadp.slh.wisc.edu/data/sites/siteDetails.aspx?net=NTN&id=MI09> (accessed 6.21.18).

Ng, N.L., Canagaratna, M.R., Zhang, Q., Jimenez, J.L., Tian, J., Ulbrich, I.M., Kroll, J.H., Docherty, K.S., Chhabra, P.S., Bahreini, R., Murphy, S.M., Seinfeld, J.H., Hildebrandt, L., Donahue, N.M., DeCarlo, P.F., Lanz, V.A., Prévôt, A.S.H., Dinar, E., Rudich, Y., Worsnop, D.R., 2010. Organic aerosol components observed in Northern Hemispheric datasets from Aerosol Mass Spectrometry. *Atmospheric Chemistry and Physics* 10, 4625–4641. <https://doi.org/10.5194/acp-10-4625-2010>

Öztürk, F., Bahreini, R., Wagner, N.L., Dubé, W.P., Young, C.J., Brown, S.S., Brock, C.A., Ulbrich, I.M., Jimenez, J.L., Cooper, O.R., Middlebrook, A.M., 2013. Vertically resolved chemical characteristics and sources of submicron aerosols measured on a Tall Tower in a suburban area near Denver, Colorado in winter. *J. Geophys. Res. Atmos.* 118, 2013JD019923. <https://doi.org/10.1002/2013JD019923>

Paatero, P., 1997. Least squares formulation of robust non-negative factor analysis. *Chemometrics and Intelligent Laboratory Systems* 37, 23–35.

Paatero, P., Tapper, U., 1994. Positive matrix factorization: A non-negative factor model with optimal utilization of error estimates of data values. *Environmetrics* 5, 111–126.

Polissar, A., 1999. The aerosol at Barrow, Alaska: long-term trends and source locations. *Atmospheric Environment* 33, 2441–2458. [https://doi.org/10.1016/S1352-2310\(98\)00423-3](https://doi.org/10.1016/S1352-2310(98)00423-3)

Polissar, A.V., Hopke, P.K., Harris, J.M., 2001. Source Regions for Atmospheric Aerosol Measured at Barrow, Alaska. *Environ. Sci. Technol.* 35, 4214–4226. <https://doi.org/10.1021/es0107529>

Pryor, S.C., Larsen, S.E., Sørensen, L.L., Barthelmie, R.J., Grönholm, T., Kulmala, M., Launiainen, S., Rannik, Ü., Vesala, T., 2007. Particle fluxes over forests: Analyses of flux methods and functional dependencies. *Journal of Geophysical Research* 112. <https://doi.org/10.1029/2006JD008066>

Pugh, T. a. M., MacKenzie, A.R., Hewitt, C.N., Langford, B., Edwards, P.M., Furneaux, K.L., Heard, D.E., Hopkins, J.R., Jones, C.E., Karunaharan, A., Lee, J., Mills, G., Misztal, P., Moller, S., Monks, P.S., Whalley, L.K., 2010. Simulating atmospheric composition over a South-East Asian tropical rainforest: performance of a chemistry box model. *Atmospheric Chemistry and Physics* 10, 279–298. <https://doi.org/10.5194/acp-10-279-2010>

Rannik, Ü., Zhou, L., Zhou, P., Gierens, R., Mammarella, I., Sogachev, A., Boy, M., 2016. Aerosol dynamics within and above forest in relation to turbulent transport and dry deposition. *Atmospheric Chemistry and Physics* 16, 3145–3160. <https://doi.org/10.5194/acp-16-3145-2016>

Rizzo, L.V., Artaxo, P., Karl, T., Guenther, A.B., Greenberg, J., 2010. Aerosol properties, in-canopy gradients, turbulent fluxes and VOC concentrations at a pristine forest site in Amazonia. *Atmospheric Environment* 44, 503–511. <https://doi.org/10.1016/j.atmosenv.2009.11.002>

Schulze, B.C., Wallace, H.W., Bui, A.T., Flynn, J.H., Erickson, M.H., Alvarez, S., Dai, Q., Usenko, S., Sheesley, R.J., Griffin, R.J., 2018. The impacts of regional shipping emissions on the chemical characteristics of coastal submicron aerosols near Houston, TX. *Atmospheric Chemistry and Physics* 18, 14217–14241. <https://doi.org/10.5194/acp-18-14217-2018>

Schulze, B.C., Wallace, H.W., Flynn, J.H., Lefer, B.L., Erickson, M.H., Jobson, B.T., Dusanter, S., Griffith, S.M., Hansen, R.F., Stevens, P.S., VanReken, T., Griffin, R.J., 2017. Differences in BVOC oxidation and SOA formation above and below the forest canopy. *Atmospheric Chemistry and Physics* 17, 1805–1828. <https://doi.org/10.5194/acp-17-1805-2017>

Seinfeld, J.H., Pandis, S.N., 2006. *Atmospheric chemistry and physics : from air pollution to climate change*. Hoboken, N.J. : John Wiley, ©2006.

Sheesley, R.J., Schauer, J.J., Bean, E., Kenski, D., 2004. Trends in Secondary Organic Aerosol at a Remote Site in Michigan's Upper Peninsula. *Environmental Science & Technology* 38, 6491–6500. <https://doi.org/10.1021/es049104q>

Sirois, A., Bottenheim, J.W., 1995. Use of backward trajectories to interpret the 5-year record of PAN and O<sub>3</sub> ambient air concentrations at Kejimikujik National Park, Nova Scotia. *Journal of Geophysical Research* 100, 2867. <https://doi.org/10.1029/94JD02951>

Stein, A.F., Draxler, R.R., Rolph, G.D., Stunder, B.J.B., Cohen, M.D., Ngan, F., 2015. NOAA's HYSPLIT Atmospheric Transport and Dispersion Modeling System. *Bulletin of the American Meteorological Society* 96, 2059–2077. <https://doi.org/10.1175/BAMS-D-14-00110.1>

Steiner, A.L., Pressley, S.N., Botros, A., Jones, E., Chung, S.H., Edburg, S.L., 2011. Analysis of coherent structures and atmosphere-canopy coupling strength during the CABINEX field campaign. *Atmospheric Chemistry and Physics* 11, 11921–11936. <https://doi.org/10.5194/acp-11-11921-2011>

Stroud, C., Makar, P., Karl, T., Guenther, A., Geron, C., Turnipseed, A., Nemitz, E., Baker, B., Potosnak, M., Fuentes, J.D., 2005. Role of canopy-scale photochemistry in modifying biogenic-atmosphere exchange of reactive terpene species: Results from the CELTIC field study. *Journal of Geophysical Research: Atmospheres* 110. <https://doi.org/10.1029/2005JD005775>

Thomas, C., Foken, T., 2007. Flux contribution of coherent structures and its implications for the exchange of energy and matter in a tall spruce canopy. *Boundary-Layer Meteorology* 123, 317–337. <https://doi.org/10.1007/s10546-006-9144-7>

Thornberry, T., Carroll, M.A., Keeler, G.A., Sanford, S., Bertman, S.B., Pippin, M.R., Ostling, K., Grossenbacher, J.W., Shepson, P.B., Cooper, O.R., Moody, J.L., Stockwell, W.R., 2001. Observations of reactive oxidized nitrogen and speciation of NO<sub>y</sub> during the PROPHET summer 1998 intensive. *Journal of Geophysical Research* 106, 359–386.

Toma, S., Bertman, S., 2012. The atmospheric potential of biogenic volatile organic compounds from needles of white pine (*Pinus strobus*) in Northern Michigan. *Atmospheric Chemistry and Physics* 12, 2245–2252. <https://doi.org/10.5194/acp-12-2245-2012>

Ulbrich, I.M., Canagaratna, M.R., Zhang, Q., Worsnop, D.R., Jimenez, J.L., 2009. Interpretation of organic components from Positive Matrix Factorization of aerosol mass spectrometric data. *Atmospheric Chemistry and Physics* 9, 2891–2918.

US Census Bureau, 2018. City and Town Population Totals: 2010-2016 [WWW Document]. URL <https://www.census.gov/data/tables/2016/demo/popest/total-cities-and-towns.html> (accessed 1.24.18).

VanReken, T.M., Mwaniki, G.R., Wallace, H.W., Pressley, S.N., Erickson, M.H., Jobson, B.T., Lamb, B.K., 2015. Influence of air mass origin on aerosol properties at a remote Michigan forest site. *Atmospheric Environment* 107, 35–43. <https://doi.org/10.1016/j.atmosenv.2015.02.027>

Wallace, H.W., Sanchez, N.P., Flynn, J.H., Erickson, M.H., Lefer, B.L., Griffin, R.J., 2018. Source apportionment of particulate matter and trace gases near a major refinery near the Houston Ship Channel. *Atmospheric Environment* 173, 16–29. <https://doi.org/10.1016/j.atmosenv.2017.10.049>

Wang, Y.Q., 2014. MeteInfo: GIS software for meteorological data visualization and analysis: Meteorological GIS software. *Meteorological Applications* 21, 360–368. <https://doi.org/10.1002/met.1345>

Wang, Y.Q., Zhang, X.Y., Draxler, R.R., 2009. TrajStat: GIS-based software that uses various trajectory statistical analysis methods to identify potential sources from long-term air pollution measurement data. *Environmental Modelling & Software* 24, 938–939. <https://doi.org/10.1016/j.envsoft.2009.01.004>

Whitehead, J.D., Gallagher, M.W., Dorsey, J.R., Robinson, N., Gabey, A.M., Coe, H., McFiggans, G., Flynn, M.J., Ryder, J., Nemitz, E., Davies, F., 2010. Aerosol fluxes and dynamics within and above a tropical rainforest in South-East Asia. *Atmospheric Chemistry and Physics* 10, 9369–9382. <https://doi.org/10.5194/acp-10-9369-2010>

Wiedinmyer, C., Greenberg, J., Guenther, A., Hopkins, B., Baker, K., Geron, C., Palmer, P.I., Long, B.P., Turner, J.R., Pétron, G., Harley, P., Pierce, T.E., Lamb, B., Westberg, H., Baugh, W., Koerber, M., Janssen, M., 2005. Ozarks Isoprene Experiment (OZIE): Measurements and modeling of the “isoprene volcano.” *Journal of Geophysical Research: Atmospheres* 110. <https://doi.org/10.1029/2005JD005800>

Wolfe, G.M., Thornton, J.A., 2011. The Chemistry of Atmosphere-Forest Exchange (CAFE) Model – Part 1: Model description and characterization. *Atmospheric Chemistry and Physics* 11, 77–101. <https://doi.org/10.5194/acp-11-77-2011>

Xu, L., Pye, H.O.T., He, J., Chen, Y.L., Murphy, B.N., Ng, N.L., 2018. Experimental and model estimates of the contributions from biogenic monoterpenes and sesquiterpenes to secondary organic aerosol in the southeastern United States. *Atmospheric Chemistry and Physics* 18, 12613–12637. <https://doi.org/10.5194/acp-18-12613-2018>

Xu, L., Suresh, S., Guo, H., Weber, R.J., Ng, N.L., 2015. Aerosol characterization over the southeastern United States using high-resolution aerosol mass spectrometry: spatial and seasonal variation of aerosol composition and sources with a

982 focus on organic nitrates. *Atmospheric Chemistry and Physics* 15, 7307–7336. [https://doi.org/10.5194/acp-15-7307-](https://doi.org/10.5194/acp-15-7307-2015)  
 983 2015  
 984 Zhang, Q., Jimenez, J.L., Canagaratna, M.R., Allan, J.D., Coe, H., Ulbrich, I., Alfarra, M.R., Takami, A., Middlebrook,  
 985 A.M., Sun, Y.L., Dzepina, K., Dunlea, E., Docherty, K., DeCarlo, P.F., Salcedo, D., Onasch, T., Jayne, J.T.,  
 986 Miyoshi, T., Shimonono, A., Hatakeyama, S., Takegawa, N., Kondo, Y., Schneider, J., Drewnick, F., Borrmann, S.,  
 987 Weimer, S., Demerjian, K., Williams, P., Bower, K., Bahreini, R., Cottrell, L., Griffin, R.J., Rautiainen, J., Sun,  
 988 J.Y., Zhang, Y.M., Worsnop, D.R., 2007. Ubiquity and dominance of oxygenated species in organic aerosols in  
 989 anthropogenically-influenced Northern Hemisphere midlatitudes: Ubiquity and dominance of oxygenated OA.  
 990 *Geophysical Research Letters* 34, 6. <https://doi.org/10.1029/2007GL029979>  
 991 Zhang, Q., Jimenez, J.L., Canagaratna, M.R., Ulbrich, I.M., Ng, N.L., Worsnop, D.R., Sun, Y., 2011. Understanding  
 992 atmospheric organic aerosols via factor analysis of aerosol mass spectrometry: a review. *Analytical and*  
 993 *Bioanalytical Chemistry* 401, 3045–3067. <https://doi.org/10.1007/s00216-011-5355-y>  
 994 Zhang, Q., Worsnop, D.R., Canagaratna, M.R., Jimenez, J.L., 2005. Hydrocarbon-like and oxygenated organic aerosols in  
 995 Pittsburgh: insights into sources and processes of organic aerosols. *Atmospheric Chemistry and Physics* 5, 3289–  
 996 3311.