

Figure S1: Correlation plot of $C_{10}H_{16}O_2$ and $C_{20}H_{32}N_2O_8$. (inset) R^2 values for the comparison between $C_{10}H_{16}O_2$ and the series of $C_{20}H_{32}N_2O_x$ ($x = 8 - 12$).

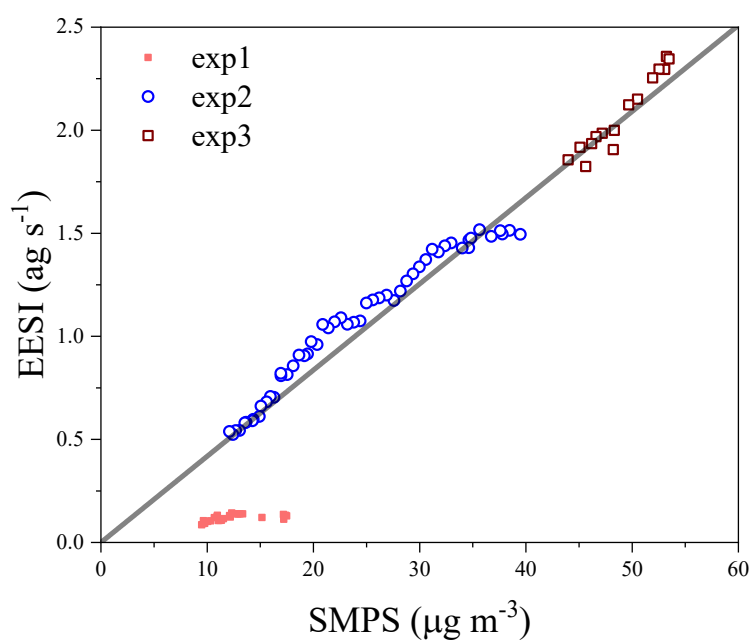


Figure S2: EESI mass flux compared to the measured SMPS mass for experiments 1 – 3, when the EESI was present. The total mass flux measured by the EESI is the ion signal (s^{-1}) converted to measured mass per second by scaling each molecular formula according to its mass.

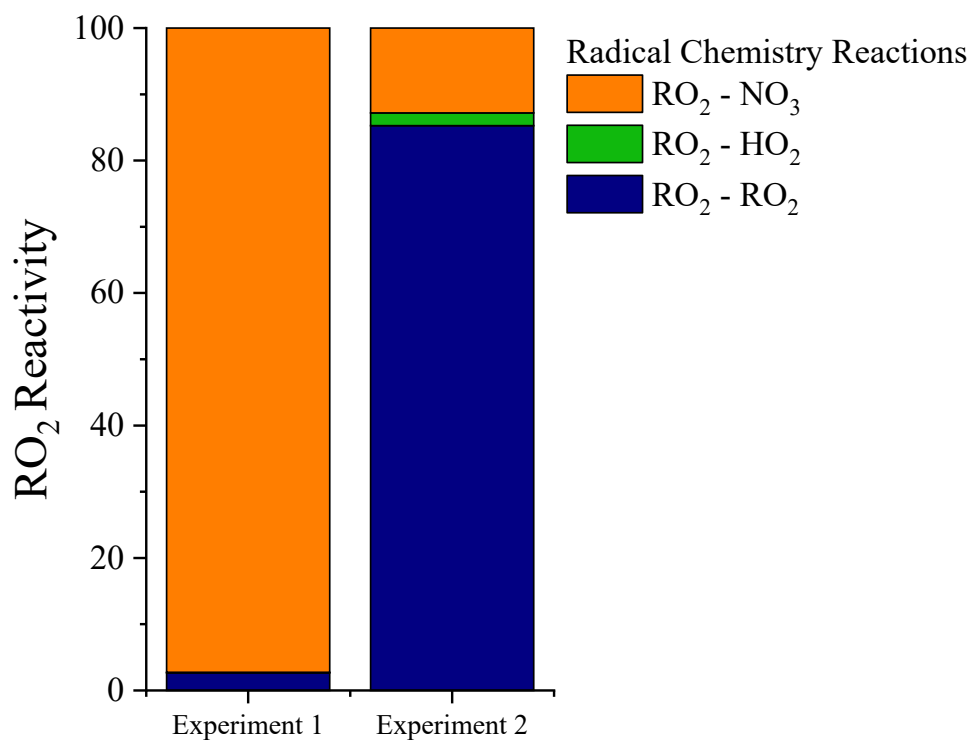


Figure S3: Integrated radical reactivity of RO₂ radicals in experiment 1 and experiment 2 modeled with the Master Chemical Mechanism (MCM) (Jenkin et al., 2003; Saunders et al., 2003). The absolute values will differ depending on the reaction rates of RO₂ – RO₂, but this provides insight into the reactions controlling the initial composition of the particle phase.

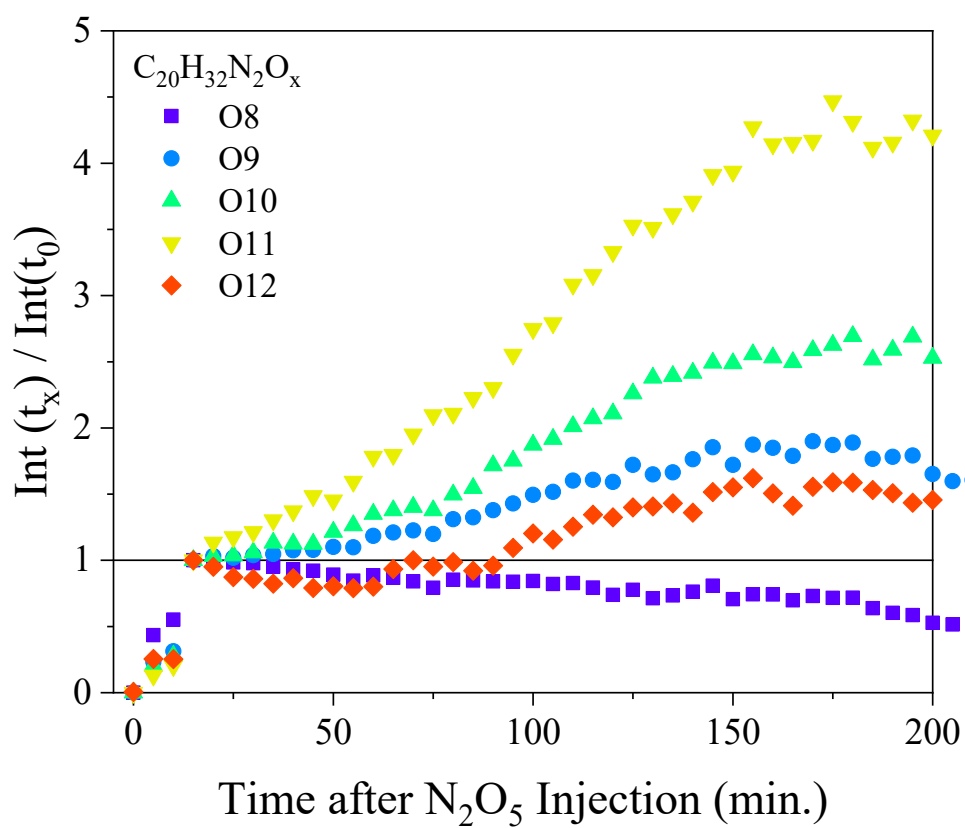


Figure S4: Normalized $C_{20}H_{32}N_2O_x$ dinitrate signals relative to $t = 15$ min mass from experiment 3. $t = 15$ min represents the maximum in particle mass, which is why this time was chosen.

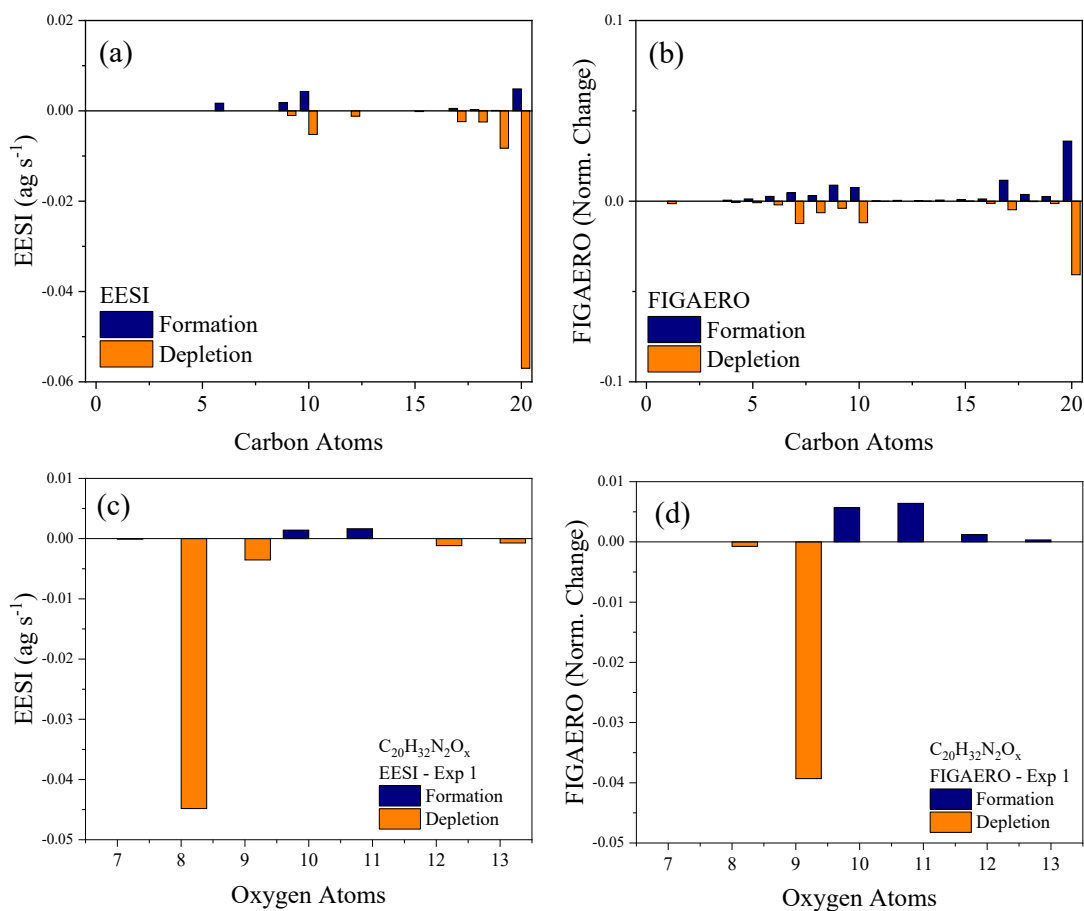


Figure S5: (a) and (b) Carbon distribution of the molecules formed (blue) and depleted (orange) in the particle phase during aging for both the EESI (a) and FIGAERO-CIMS (b), respectively for exp 1. (c & d) The change in the oxygen distribution for the $C_{20}H_{32}N_2O_{8-13}$ molecules during dark aging for the EESI-ToF (c) and FIGAERO-CIMS (d), respectively.

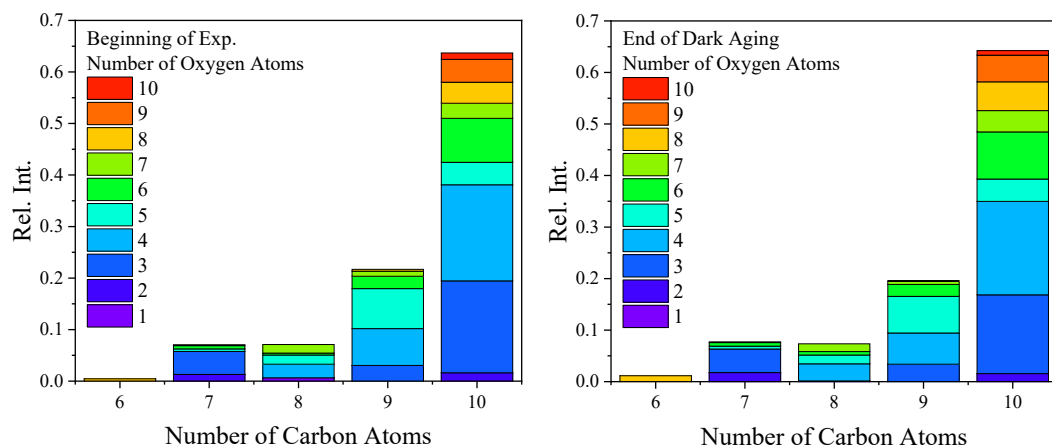


Figure S6: Monomer carbon and oxygen atom distribution for the beginning of the experiment $t_{early} = 10$ min and end of dark aging $t = 240$ min for exp 2.

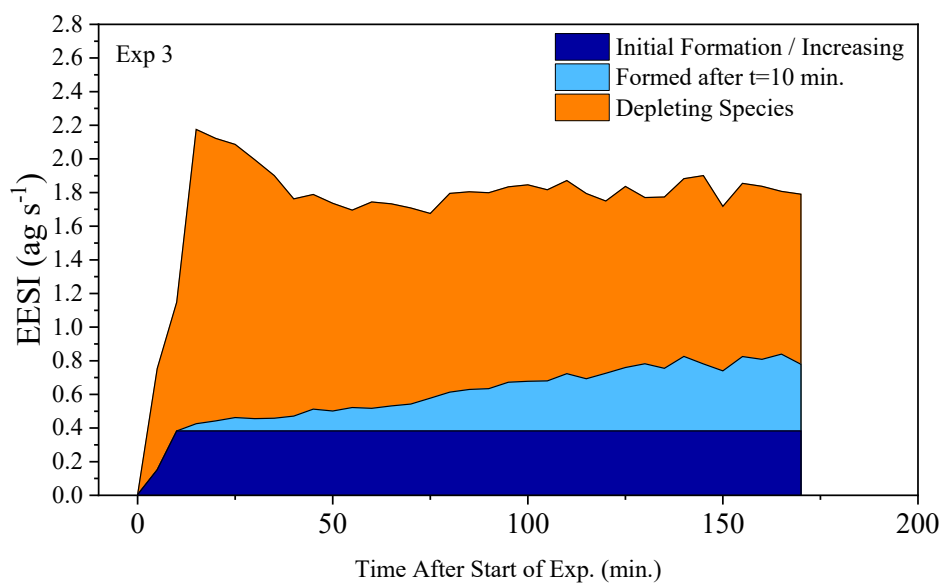


Figure S7: Total EESI-ToF intensity plotted as a function of the contribution from different sources, particle phase formation products determined from those molecular formula that increase during dark aging after $t = 10$ min, while those decreasing are shown in orange. The data is shown for experiment 3. The species forming in the dark are split into the dark blue (intensity at $t = 15$ min) and the fraction increasing after that time in light blue.

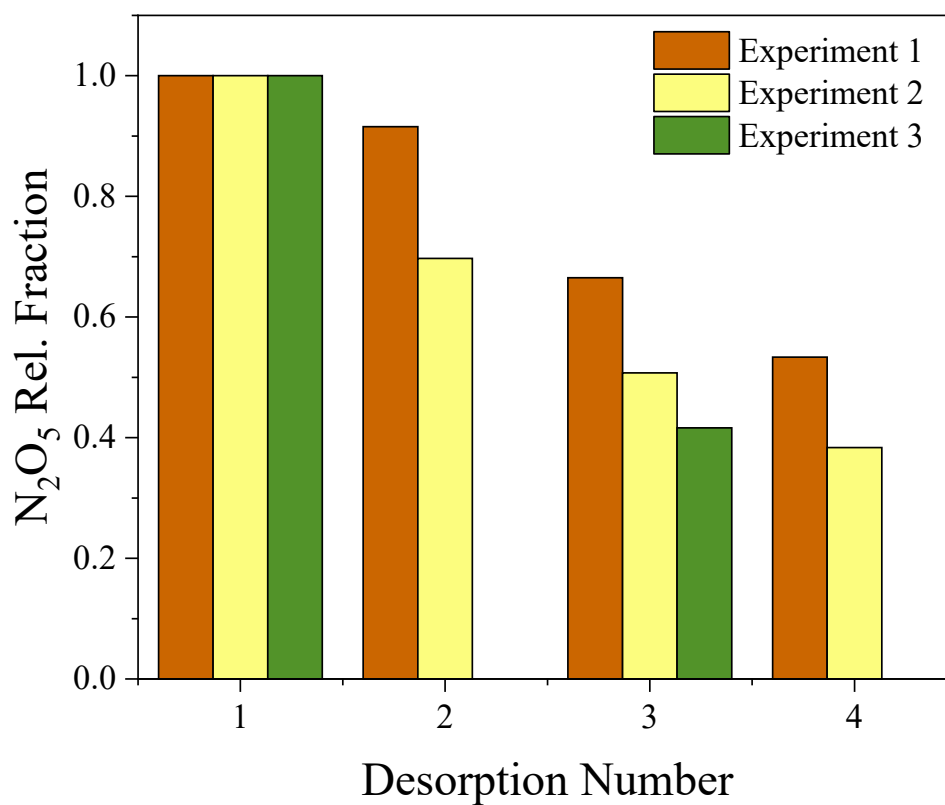


Figure S8: N₂O₅ particle phase relative fraction normalized to the initial desorption measured by the FIGAERO-CIMS. Estimated mass concentrations in the particle phase are 0.03 – 0.06 $\mu\text{g m}^{-3}$. The time between each desorption was about 1-1.5h.

References:

Jenkin, M. E., Saunders, S. M., Wagner, V., and Pilling, M. J.: Protocol for the development of the Master Chemical Mechanism, MCM v3 (Part B): tropospheric degradation of aromatic volatile organic compounds, *Atmos. Chem. Phys.*, 3, 181-193, 2003, 10.5194/acp-3-181-2003.

Saunders, S. M., Jenkin, M. E., Derwent, R. G., and Pilling, M. J.: Protocol for the development of the Master Chemical Mechanism, MCM v3 (Part A): tropospheric degradation of non-aromatic volatile organic compounds, *Atmos. Chem. Phys.*, 3, 161-180, 2003, 10.5194/acp-3-161-2003.