



# Nonlinear responses of particulate nitrate to NO<sub>x</sub> emission controls in

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#### 9 Abstract

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Nitrate is an increasingly important component of fine particulate matter (PM<sub>2.5</sub>) in Chinese cities. The production of nitrate is not only related to the abundance of its precursor, but also supported by atmospheric photochemical oxidants. The control of nitrogen oxides (NO<sub>x</sub>) emissions may thereby lead to nonlinear changes of nitrate concentrations, raising a new challenge to the current emission control actions in China. This paper uses comprehensive measurements and a regional meteorologychemistry model with optimized mechanisms to establish the nonlinear responses between particulate nitrate and NO<sub>x</sub> emission controls in the megalopolises of China. Nitrate is an essential component of PM<sub>2.5</sub> in eastern China, accounting for 9.4–15.5% and 11.5-32.1% of the PM<sub>2.5</sub> mass for the warm and cold seasons. The hypothetical NO<sub>x</sub> emission reduction scenarios (-10%~-80%) during summer-autumn result in almost linearly lower PM<sub>2.5</sub> by -2.65% in Beijing-Tianjin-Hebei (BTH) and -2.79% in Yangtze River Delta (YRD) per 10% cut of NO<sub>x</sub> emissions, whereas they increase the oxidant levels and lead to a rather complicated response of PM components in winter. Wintertime nitrate is found to increase by 4.28% in BTH and 4.60% in YRD, with higher dinitrogen pentoxide (N2O5) intermediate products produced from





26 increased ozone introduced by lower NO<sub>x</sub> emissions. An inflexion point appears at 27 40-50% NO<sub>x</sub> emission reduction, and a further cut in NO<sub>x</sub> emission is predicted to cause -8.74% reduction of nitrate for BTH and -10.59% for YRD per 10% cut of 28 29 NO<sub>x</sub> emissions. In addition, the 2012–2016 NO<sub>x</sub> control strategy actually leads to no 30 change or even increase of nitrate in some areas (8.82% in BTH and 14.41% in YRD) during winter. This paper helps understand the nonlinear aerosol and photochemistry 31 32 feedbacks, and defines the effectiveness of proposed mitigations for the increasingly 33 serious nitrate pollution in China.

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#### 1 Introduction

mass during haze events in China (Huang et al., 2014a; Zhao et al., 2013). Since the enactment of the Air Pollution Action Plan in 2013, the Chinese government has taken drastic measures to reduce the emissions of sulfur dioxide (SO<sub>2</sub>), nitrogen oxides (NO<sub>x</sub>) and primary PM<sub>2.5</sub>, leading to significant decreases in sulfate and overall PM<sub>2.5</sub> concentrations in cities (Silver et al., 2018; Li et al., 2021a; Wang et al., 2017). Meanwhile, the nitrogen/sulfur (N/S) ratio in PM<sub>2.5</sub> increased significantly and nitrate had been the main component of PM<sub>2.5</sub> (16-45%) during haze episodes, despite a more than 20% reduction in the concentrations of its precursor NO<sub>x</sub> (Shao et al., 2018; Wen et al., 2018; Zhai et al., 2019). The increasingly serious nitrate pollution has emerged to be the new emphasis of air pollution controls in China. Nitrate formation involves complex multiphase chemical reactions. In the daytime, nitrogen dioxide (NO2) reacts with hydroxyl radical (OH) to produce nitric acid (HNO<sub>3</sub>). With excess ammonium (NH<sub>3</sub>), low temperature and insufficient sulphuric acid, this reaction can proceed quickly and produce high ammonium nitrate (Seinfeld and Pandis, 2006). In the nighttime, however, high-concentration NO<sub>2</sub> reacts with ozone (O<sub>3</sub>) to produce the nitrate radical (NO<sub>3</sub>) and dinitrogen pentoxide (N<sub>2</sub>O<sub>5</sub>). The heterogeneous hydrolysis of N<sub>2</sub>O<sub>5</sub> on wet particles is the main pathway for nocturnal nitrate formation (56–97%) (He et al., 2018; Pathak et al., 2011; Xue et al., 2014). Nitrate chemistry is not only related to the abundance of its precursor NO<sub>x</sub>, but also supported by atmospheric oxidants (OH and O<sub>3</sub>) produced from the photochemical reactions of NO<sub>x</sub> and volatile organic compounds (VOC<sub>s</sub>) (Meng et al.,

Secondary inorganic aerosols (SIA), including sulfate (SO<sub>4</sub><sup>2-</sup>), nitrate (NO<sub>3</sub><sup>-</sup>) and

ammonium (NH<sub>4</sub><sup>+</sup>) account for 30–60% of the total fine particulate matter (PM<sub>2.5</sub>)





61 between particulate nitrate and NO<sub>x</sub> emissions is nonlinear depending on the ozone 62 chemical sensitivity regime (Pun and Seigneur, 2001; Nguyen and Dabdub, 2002). 63 Pun and Seigneur (2001) showed that the daytime HNO<sub>3</sub> production was more 64 sensitive to the concentration of atmospheric oxidants, and that in the VOC-limited 65 regime the decrease of HNO<sub>3</sub> production due to the NO<sub>x</sub> emission control might be 66 offset by the increase of OH. Nguyen and Dabdub (2002) calculated a detailed isopleth between nitrate and NO<sub>x</sub> emission; they found that the reduction of NO<sub>x</sub> 67 68 emission resulted in a decrease of nitrate in the NO<sub>x</sub>-limited regime, and an increase of nitrate under extreme conditions in the VOC-limited regime. Despite that, the 69 70 single-site box model results could not distinguish the regional differences among 71 chemical regimes; the basic hypotheses in box models to predict nitrate production, 72 e.g., stagnant atmosphere or fixed NO<sub>2</sub>/NO<sub>y</sub> ratio, are also unreasonable in the real 73 atmosphere. 74 As an important precursor for both fine particles and ozone, the strict control of NO<sub>x</sub> emission has started in China since the 12<sup>th</sup> Five-Year Plan (Zheng et al., 2018). 75 76 A confounding factor is that, for most cities in China, the production of O<sub>3</sub> is usually 77 limited by VOC<sub>s</sub> (Liu et al., 2010). The control of NO<sub>x</sub> emission has therefore 78 resulted in an increase of O<sub>3</sub> concentrations in recent years (Li et al., 2021a; Li et al., 79 2019a; Kalsoom et al., 2021), implying complex impacts on nitrate formation. Some 80 simulations thought that the NO<sub>x</sub> emission increase in 2005–2012 resulted in an increase of nitrate by 3.4% yr<sup>-1</sup> in eastern China (Geng et al., 2017; Wang et al., 81 82 2013), and the following NO<sub>x</sub> emission control resulted in a decrease of nitrate by 3– 83 14% (Wang et al., 2014). Recent evidence from field observations (Fu et al., 2020) 84 and numerical simulations (Dong et al., 2014) suggested that the NO<sub>x</sub> emission

1997). Using a box model, some studies have determined that the relationship

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85 reduction in China could result in an increase of nitrate in winter through increased 86 photochemical oxidants and nocturnal N<sub>2</sub>O<sub>5</sub> chemistry, but a decrease in other seasons. 87 In the next 5–10 years, SO<sub>2</sub> emissions might level off in China, while NO<sub>x</sub> emissions 88 will become stringently controlled to ensure further air quality improvements (Zheng 89 et al., 2018). Accurately understanding the nonlinear aerosol and photochemistry 90 feedbacks is crucial to resolve the emerging nitrate pollution and to establish 91 reasonable air pollution control strategies in China. To address this issue, we use comprehensive measurements and a regional 92 93 meteorology-chemistry model combined with hypothetical NO<sub>x</sub> emission scenarios to 94 establish the nonlinear response relationships between particulate nitrate and NO<sub>x</sub> 95 emission controls in the megalopolises of China. The model configurations, numerical 96 designs and observational data are presented in Sect. 2. Sect. 3 discusses the results. 97 Finally, a summary is presented in Sect. 4. 98 2 Materials and Methods 99 2.1 Model setup and experimental designs 100 This study uses the Weather Research and Forecasting-Chemistry (WRF-Chem) 101 model version 4.1 developed by Grell et al. (2002) to simulate the meteorology and 102 atmospheric chemistry. The mesoscale meteorology and air quality simulations of 103 WRF-Chem have been improved in terms of incorporating the satellite-derived land

The modeling domain covers two main megalopolises of China and its adjacent areas—the Beijing-Tianjin-Hebei (BTH) region and the Yangtze River Delta (YRD) region (Fig. 1). The modeling framework is configured with 81×86 grid cells at 25 km horizontal resolution. The model is run with an 84-hour model cycle, with the first 12

surface parameters (Li et al., 2014; Li et al., 2017), and optimizing the SIA formation

pathways enhanced by mineral aerosols (Li et al., 2019b; Huang et al., 2014b).





110 hours discarded as spin-up time and model outputs of each model cycle to provide 111 chemical initial conditions for the subsequent overlapping 84-hour simulation. The 6-112 hour, 1°×1° National Centers for Environmental Prediction Final (NCEP/FNL) 113 analysis fields are regularly input for the model initial and lateral boundary 114 meteorological conditions. 115 The model physical configurations include the YSU boundary layer scheme 116 (Noh et al., 2003), the RRTMG radiation scheme (Iacono et al., 2008), the Noah land surface scheme (Ek et al., 2003) and the Lin microphysics scheme (Lin et al., 1983). 117 118 We have further updated the land cover type and vegetation data in WRF mesoscale 119 model with the latest land surface parameters derived from Moderate Resolution 120 Imaging Spectroradiometer (Li et al., 2014; Li et al., 2017). 121 The atmospheric chemistry is simulated using the Carbon Bond Mechanism 122 version Z (CBMZ) (Zaveri and Peters, 1999) gas-phase chemistry module coupled 123 with a four-bin sectional Model for Simulating Aerosol Interactions and Chemistry 124 (MOSAIC) (Zaveri et al., 2008). The aqueous-phase chemistry is based on the 125 Carnegie Mellon University (CMU) scheme including 50 species and more than 100 126 reactions (Fahey and Pandis, 2001). Formation of SIA in the model accounts for the 127 gas-phases oxidation of SO2 and NO2, and aqueous-phase oxidation of SO2 by 128 hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) and O<sub>3</sub> in cloud. We have also optimized the SIA 129 formation pathways by including the aqueous SO<sub>2</sub> oxidation catalyzed by mineral 130 ions and heterogeneous uptake of SO2, NO2, NO3, N2O5 and HNO3 on mineral 131 aerosols in the MOSAIC aerosol module (Li et al., 2019b; Huang et al., 2014b). 132 Anthropogenic emissions are adopted from the 2016 Multi-resolution Emission 133 Inventory for China (MEIC) and the 2010 MIX-Asia emission inventory for regions 134 outside of mainland China developed by Tsinghua University (http://meicmodel.org).





135 Biogenic emissions are calculated online using the Model of Emissions of Gases and 136 Aerosols from Nature (Guenther et al., 2006). 137 A series of WRF-Chem simulations are designed as summarized in Table 1. In 138 the baseline simulation (denoted as the B0 scenario), the anthropogenic emissions in 139 China remain unchanged at the usual levels. Simulation N0 is the same as the base 140 case B0, but it only considers the NO<sub>2</sub>+OH gas-phase oxidation pathway for the 141 production of nitrate aerosol. The B0 and N0 simulations are designed to distinguish 142 the formation mechanisms for the high concentrations of nitrate aerosols during the 143 warm and cold seasons. A group of sensitivity scenarios (C1~C8) are designed with 144 the perturbed anthropogenic NO<sub>x</sub> emissions in China cut by 10%, 20%...and 80%, 145 respectively. The differences between B0 and C1~C8 simulations are calculated to 146 illustrate the responses of particulate pollution in China's megacities to the NO<sub>x</sub> 147 emission reduction scenarios. Another simulation (E1) is designed with the 148 anthropogenic emission of NO<sub>x</sub> in China set to the 2012 level to show the impacts of 149 2012–2016 NO<sub>x</sub> control strategy on particulate pollution. 150 For all simulation scenarios, two month-long periods during the Campaign on Air Pollution and Urban Meteorology in Yangtze River Delta (CAPUM-YRD)— 151 152 August 15 to September 16 (Period I) and November 24 to December 26 (Period II) in 153 2016, are simulated to represent the warm and cold seasons, respectively (Shu et al., 154 2019). The complete simulation consists of thirteen 84-hour model cycles with the 155 first 6 days as spin-up for chemistry and the remaining model outputs for analysis. 156 2.2 Weather and air pollutants data 157 Surface meteorological observations at 186 land-based automatic stations in 158 China (Fig. 1) are collected for model meteorological validation, including hourly





159 data of 2 m air temperature, 2 m relative humidity and 10 m wind speed. These data 160 are archived at the U. S. National Climatic Data Center (NCDC) (Smith et al., 2011). 161 Air pollutants data at the national air quality monitoring network and regional 162 supersites of China (Fig. 1) are used for model chemical validation. Six routine air 163 pollutants including particulate matter with aerodynamic diameter less than 10 µm 164 (PM<sub>10</sub>), PM<sub>2.5</sub>, SO<sub>2</sub>, NO<sub>2</sub>, carbon monoxide (CO) and O<sub>3</sub> are monitored and reported 165 hourly by Chinese National Environmental Monitoring Center (CNEMC) network 166 (available at http://websearch.mep.gov.cn/). This nationwide monitoring network 167 includes 1597 sites covering 454 cities in mainland China, as shown in Fig. 1. 168 Additionally, four comprehensive atmospheric environment supersites in YRD 169 including Dianshanhu (DSH; 31.1°N, 121.0°E), Pudong (PD; 31.2° N, 121.5°E), 170 Nanjing (NJ; 32.1°N, 118.8°E) and Hangzhou (HZ; 30.3°N, 120.2°E) measured the 171 mass concentrations of PM2.5, water-soluble ions (sulfate, nitrate, ammonium, sodium, 172 chloride, potassium, calcium and magnesium), carbonaceous aerosols (elemental 173 carbon (EC) and organic carbon (OC)) and gaseous pollutants (SO<sub>2</sub>, NO<sub>2</sub>, CO and O<sub>3</sub>) 174 during the CAPUM-YRD campaign. Details for the methods and data at the four 175 supersites are described in Shu et al. (2019).

### 3 Results and discussions

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#### 3.1 Model weather and chemical validation

Model evaluations indicate that the WRF-Chem model is able to simulate the weather and atmospheric pollution characteristics in China. The simulated magnitudes of surface temperature by WRF-Chem in general agree with actual observations, with a correlation efficient (*R*) of 0.89 and 0.94, and a normalized mean bias (NMB) of -0.55% and -0.80% respectively in Period I and Period II (Table 2). Underestimation of relative humidity (-5.65% in Period I and -6.56% in Period II) is common in the





184 WRF simulation and is attributed to the influence of the boundary layer 185 parameterization on the weather forecast (Bhati and Mohan, 2018; Gomez-Navarro et 186 al., 2015). Clear overestimation of wind speed (23.72% in Period I and 40.64% in 187 Period II) might be because of the unresolved topography in WRF (Jimenez et al., 188 2013; Li et al., 2014). 189 The predicted concentrations of routine air pollutants also faithfully captures the 190 spatial and seasonal patterns of observed surface PM2.5, SO2, NO2 and O3 levels in 191 both seasons (Fig. 2). Both simulations and observations display high air pollutant 192 concentrations in the vicinity of North China Plain (NCP) and eastern China, but with 193 higher O<sub>3</sub> levels in the warm seasons and oppositely higher PM<sub>2.5</sub> and other gaseous 194 pollutants in winter. The model statistical evaluations show a MB of -3.66, -1.14, 4.7 and 18.32  $\mu$ g m<sup>-3</sup>, and NMB of -9.92, -6.46, 16.47 and 7.72% for PM<sub>2.5</sub>, SO<sub>2</sub>, NO<sub>2</sub> 195 196 and O<sub>3</sub> in Period I, and a relatively larger MB of -27.31, -11.65, 1.27 and -39.01 μg 197 m<sup>-3</sup>, and NMB of -29.82, -28.11, 2.40 and -31.05% in Period II, respectively (Table 198 3). The uncertainty in emissions data, the absence of secondary organic aerosol in 199 MOSAIC aerosol chemistry or the simulated wind errors may be responsible for the 200 larger atmospheric chemical biases in winter, which has been extensively discussed in 201 some studies (Zhao et al., 2016; Li et al., 2021a). 202 As the most important component of PM<sub>2.5</sub>, reasonable representation of SIA is 203 imperative to PM<sub>2.5</sub> simulation. Evaluations with measurements of PM<sub>2.5</sub> components 204 at the four supersites of eastern China show that the model performs reasonably in 205 simulating the seasonal variations and proportions of aerosol species in PM<sub>2.5</sub>, but it is 206 biased low by 10-40% in simulating the magnitudes of SIA concentrations (Fig. 3). 207 The model underestimation is -1.8, -2.2 and -2.2  $\mu$ g m<sup>-3</sup> for sulfate, nitrate and ammonium, respectively, in Period I, and -2.6, -4.3 and -3.4  $\mu$ g m<sup>-3</sup> in Period II. The 208

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210 vary from 2.4 in Period I to 0.6 in Period II. Our previous work (Li et al., 2019) has 211 confirmed that the optimized aqueous and heterogeneous SIA formation pathways in 212 WRF-Chem significantly reduce the model biases by 41.38% for sulfate and 44.55% 213 for nitrate during the CAPUM-YRD campaign of 2016. Recent studies highlighted 214 that the remaining SIA simulation biases may be attributed to the missing aqueous 215 oxidation of SO<sub>2</sub> by NO<sub>2</sub> on alkaline aerosols under humid conditions (Wang et al., 216 2016; Cheng et al., 2016). 217 3.2 Air pollution and aerosol composition characteristics 218 Chemical composition analyses of major gaseous and particulate air pollutants 219 suggest large seasonal variations of air pollution characteristics in China (Fig. 2). 220 Mainly emitted from combustion sources, atmospheric pollutants accumulate in the 221 densely industrialized and populated megalopolises of China, with a hotspot along 222 Beijing, Hebei, Shandong and their adjacent cities frequently exceeding China's 223 National Ambient Air Quality Standards. The average concentrations of PM<sub>2.5</sub>, SO<sub>2</sub>, NO2 and daily-maximum O3 in China's routine air quality monitoring network are 224 225 36.88, 17.65, 28.53 and 237.45  $\mu$ g m<sup>-3</sup> for Period I, and 91.59, 41.45, 53.01 and 226 125.62 μg m<sup>-3</sup> for Period II. The PM<sub>2.5</sub>, SO<sub>2</sub> and NO<sub>2</sub> concentrations show obvious 227 increases by 148.35%, 134.84% and 85.80% during winter. The maximum PM<sub>2.5</sub> concentration recorded in the winter period could be more than 600 µg m<sup>-3</sup>, which is 228 229 the highest value ever recorded in 2016 and leads to the "orange" air quality alert. 230 The further analyses of PM<sub>2.5</sub> mass concentrations, major PM<sub>2.5</sub> components and 231 gases at the four supersites in YRD are presented in Fig. 4-5. Organic matter (OM) is

model also captures the large change of S/N ratios from the warm to cold seasons, that

obtained by multiplying the OC concentrations by a factor of 1.6, accounting for the





234 exhibit high levels, with average values of 18.8 µg m<sup>-3</sup> for Period I and 37.1 µg m<sup>-3</sup> 235 for Period II. The three SIA components together account for 48.6% and 56.9% of the 236 total PM<sub>2.5</sub> mass concentrations, and become the dominant components of PM<sub>2.5</sub> in 237 the two periods. The ratios of sulfate, nitrate and ammonium in total PM<sub>2.5</sub> range from 238 13.5-28.9%, 9.4-15.5% and 9.4-14.9% at the four supersites for Period I, and 9.2-239 20.3%, 11.5–32.1% and 7.0–19.8% for Period II, respectively. The higher proportion 240 of nitrate in PM<sub>2.5</sub> than that of sulfate during winter, with a sulfur/nitrogen (S/N) ratio 241 of 0.64, is in accordance with recent observations during other winter haze periods in 242 China (Shao et al., 2018; Zhang et al., 2018; Zhang et al., 2019). They emphasized 243 that since the enactment of Clean Air Action Plan in 2013, the PM<sub>2.5</sub> components had 244 changed clearly with decreasing contributions from coal combustion. 245 The high proportions of sulfate and nitrate in PM<sub>2.5</sub> could be related to the high 246 oxidation rates of SO<sub>2</sub> and NO<sub>2</sub>. The observed average values of sulfur oxidation ratio  $(SOR=[SO_4^{2-}]/([SO_4^{2-}]+[SO_2]))$  and nitrogen oxidation ratio  $(NOR=[NO_3^{-}]/([NO_3^{-}]))$ 247 248 [+[NO<sub>2</sub>])) are 0.41 and 0.13 in Period I, and 0.33 and 0.21 in Period II, indicating 249 enhanced secondary oxidation formation. Figure 6 illustrates the contributions of each 250 formation pathway for the high concentrations of nitrate calculated from B0 and E0 251 simulations. It is shown that on a daily basis the gas-phase oxidation production of 252 HNO<sub>3</sub> and its subsequent partitioning to the aerosol phase is the principal formation 253 route for particulate nitrate, with the average contributions of 60.19% for BTH and 254 91.71% for YRD in Period I and 75.14% for BTH and 85.94% for YRD in Period II, 255 heterogeneous hydrolysis of N<sub>2</sub>O<sub>5</sub> and other reactive nitrogen gases contributed to the 256 remaining nitrate particularly in BTH with high aerosol loading. These results are in 257 line with previous assessments (Alexander et al., 2009; Wen et al., 2018; Sun et al.,

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258 2018), which reported that the global tropospheric nitrate burden is dominated by nitrate formation via NO<sub>2</sub>+OH (76%), followed by N<sub>2</sub>O<sub>5</sub> hydrolysis (18%).

#### 3.3 Nonlinear responses of PM<sub>2.5</sub> to NO<sub>x</sub> emissions and their policy implication

#### 3.3.1 PM<sub>2.5</sub>-NO<sub>x</sub> and O<sub>3</sub>-NO<sub>x</sub> responses in the warm and cold seasons

NO<sub>x</sub> is key in atmospheric chemistry and serves as an important precursor for both ozone and secondary aerosols. We conduct a series of simulations with perturbed  $NO_x$  emissions (-10%~-80%) to assess the responses of PM<sub>2.5</sub> mass concentrations to NO<sub>x</sub> emissions in two megalopolises of China (Fig. 7). The WRF-Chem simulation results show that the responses of surface PM<sub>2.5</sub> concentrations to NO<sub>x</sub> emissions vary in different seasons and display strong nonlinear in winter. To better quantify their effectiveness, we define the NO<sub>x</sub> emission control efficiency (β), denoting the percentage change of PM<sub>2.5</sub> or its components concentrations in response to successive 10% cut of NO<sub>x</sub> emission. In Period I (Aug-Sep), the PM<sub>2.5</sub>-NO<sub>x</sub> responses are closer to a linear function, reflecting a stronger sensitivity to NO<sub>x</sub> emission changes in the warm season. The surface PM<sub>2.5</sub> concentrations decrease almost linearly as we gradually reduce  $NO_x$  emissions in China, with the  $\beta$  values of -2.65%in BTH and -2.79% in YRD. However, the PM<sub>2.5</sub>-NO<sub>x</sub> emission responses in Period II (Nov-Dec) display strong nonlinearity and are analogous to a bell-shaped distribution for both regions. The NO<sub>x</sub> emission reduction within the first 50% would even increase surface mean PM<sub>2.5</sub> concentrations by +1.33% in BTH, and this  $\beta$  value increases to +1.42% in YRD with the first 40% NO<sub>x</sub> emission reduction. Subsequently, the PM<sub>2.5</sub> responses shift towards a similar linear pattern, with a β value of -3.47% in BTH and -3.89% in YRD.

The distinct forms of PM<sub>2.5</sub>-NO<sub>x</sub> emission responses in the warm and cold





 $NO_x$  controls (Fig. 8–9) under the seasonal ozone chemical sensitivity regimes. The photochemical indicator of  $\Delta[O_3]_{NO_x}/\Delta[O_3]_{VOC_s}$  with a critical value of 1.0 is used to investigate the season-varying ozone sensitivity in China, which is calculated as the ratio of ozone concentration changes under 20%  $NO_x$  emission reduction to that under 20%  $VOC_s$  emission reduction (Fig. S1). The results indicate a strong VOC-limited ozone chemistry across China during winter, while either VOC-limited regime over a large portion of NCP and eastern China or  $NO_x$ -limited regime in northern and western China during summer-autumn, as also indicated from previous studies (Xie et al., 2014; Dong et al., 2014; Liu et al., 2010). We find larger  $O_3$  and OH productions under lower  $NO_x$  emission conditions in both seasons, particularly in Period II (Nov–Dec) with an increase rate of 28.51% and 36.92% in BTH and 34.36% and 33.6% in YRD per 10% cut of  $NO_x$  emission than the base case (Fig. 8–9(b, d)). The SIA formation chemistry is highly limited by atmospheric oxidants produced from the  $NO_x$ - $VOC_s$ - $O_3$  photochemical cycles. The nonlinear  $O_3$ - $NO_x$  responses indicate a rather complicated aerosol and photochemistry feedback in megacities.

# 3.3.2 Nonlinear responses of particulate nitrate to NO<sub>x</sub> emissions

The SIA formation chemistry is basically driven by atmospheric oxidants levels, and a reduction of  $NO_x$  emission may have counter-intuitive effects on SIA components by controlling atmospheric oxidants levels. The calculated SIA components for each emission scenario in both months show that sulfate has minor changes with reducing  $NO_x$  emissions but nitrate and ammonium aerosols can be substantially decreased/increased (Fig. 8–9).

Response of sulfate to the  $NO_x$  emissions is more predictable and determined by the changes of atmospheric oxidant levels since that the conversion of  $SO_2$  to sulfate is mainly driven by OH in the gas-phase and by dissolved  $H_2O_2$  or  $O_3$  in the presence





309 shaped distribution as that of O<sub>3</sub>-NO<sub>x</sub> response curve, with a fitted curve in Eq. 1. The 310 surface sulfate concentrations has minor changes by -6.69%~+2.33% in BTH and by 311 -9.55%  $\sim +3.47\%$  in YRD under the  $-10\sim -80\%$  NO<sub>x</sub> emission reduction scenarios.  $[SO_4^2] = -2.45\Delta E_{NOx}^2 - 2.15\Delta E_{NOx} + 5.90$  in BTH 312 (Eq. 1)  $[SO_4^{2-}] = -2.26\Delta E_{NO_X}^2 - 1.31\Delta E_{NO_X} + 6.65$  in YRD 313 314 where  $[SO_4^{2-}]$  is the surface mean concentration of sulfate (µg m<sup>-3</sup>);  $\Delta E_{NOx}$  is the 315 percentage change of NO<sub>x</sub> emission (%). 316 As expected, the production of nitrate reflects a strong sensitivity to NO<sub>x</sub> and it 317 decreases linearly with the  $NO_x$  emission control, with a  $\beta$  value of -10.89% in BTH 318 and -11.39% in YRD, which further leads to a decrease of ammonium concentrations 319 by -4.10% in BTH and -4.73% in YRD. The formation of nitrate involves the 320 NO<sub>2</sub>+OH→HNO<sub>3</sub> gas-phase oxidation and the heterogeneous hydrolysis of N<sub>2</sub>O<sub>5</sub>. 321 The strong sensibility of particulate nitrate in response to NO<sub>x</sub> decrease can be 322 explained by its synchronously suppressive production of intermediate products 323 HNO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub>. For example, when the NO<sub>x</sub> emission is cut by 20%, the surface 324 NO<sub>2</sub> concentration in BTH drops by 21.96% but O<sub>3</sub> increases slightly by 2.56% due to 325 greater VOC availability in warm season, leading to substantial reductions in the 326 surface HNO<sub>3</sub> (-16.72%) and N<sub>2</sub>O<sub>5</sub> (-8.94%) concentrations. 327 In Period II (Nov-Dec), we find opposite results with bell-shaped distributions 328 for nitrate-NO<sub>x</sub> response (Eq. 2) and ammonium-NO<sub>x</sub> response (Eq. 3), and linearly 329 increasing sulfate concentrations ( $\beta$ =+1.34% in BTH and +2.29% in YRD) in the 330 VOC-poor environment, leading to small PM<sub>2.5</sub> changes in winter. Such nonlinear 331 nitrate-NO<sub>x</sub> emission responses can be explained by the derived excess oxidants as we 332 gradually reduce NO<sub>x</sub> emissions in each scenario. It is noted that in winter the nitrate-

of fog or cloud. In Period I (Aug-Sep), the sulfate-NOx response follows a bell-

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NO<sub>x</sub> response is determined by the production of N<sub>2</sub>O<sub>5</sub> intermediate product, which is more sensitive to the concentration of atmospheric oxidants under low NO<sub>x</sub> emission reduction conditions. The significant increase of surface O<sub>3</sub> in each NO<sub>x</sub> emission scenario leads to an enhancement of N<sub>2</sub>O<sub>5</sub> level from 10% to more than 100%, which is produced from the  $NO_2 \xrightarrow{O_3} NO_3 \xrightarrow{NO_2} N_2O_5$  chemical reaction and is a crucial intermediate product for nitrate formation. In spite of the HNO3 concentration remaining nearly unchanged or decreasing slightly by less than 5% in response to NO<sub>x</sub> control, nitrate is found to increase (β=+4.28% in BTH and +4.60% in YRD) with higher N<sub>2</sub>O<sub>5</sub> produced from the increased ozone introduced by attenuated titration. An inflexion point appears at the 40–50% NO<sub>x</sub> emission reduction scenario, and a further reduction in NO<sub>x</sub> emission is predicted to cause -8.74% and -6.18% reductions of particulate nitrate and ammonium for BTH, and -10.59% and -8.17% for YRD.  $[NO_3^-]$ =-34.54 $\Delta E_{NOx}^2$ -30.66 $\Delta E_{NOx}$ +10.52 in BTH (Eq. 2) $[NO_3^-]=-36.53\Delta E_{NOx}^2-26.94\Delta E_{NOx}+9.70$  in YRD  $[NH_4^+] = -9.12\Delta E_{NOx}^2 - 8.73\Delta E_{NOx} + 5.40$  in BTH (Eq. 3) $[NH_4^+] = -10.55\Delta E_{NOx}^2 - 8.36\Delta E_{NOx} + 4.58 \text{ in YRD}$ where [NO<sub>3</sub><sup>-</sup>] and [NH<sub>4</sub><sup>+</sup>] are the surface mean concentrations (µg m<sup>-3</sup>) of nitrate and ammonium, respectively. These results reveal that the increase in atmospheric oxidants in response to NO<sub>x</sub> control can offset the decreasing precursor concentrations and further enhance the formation of secondary nitrate, as recently found during the COVID-19 pandemic (Huang et al., 2020; Li et al., 2021b). Our results provide insights for developing

mitigation strategies for the ubiquitous secondary aerosols in winter haze of China.

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## 3.3.3 Impacts of 2012–2016 NO<sub>x</sub> control strategy on particulate pollution

During the 12th Five-Year Plan period (2011-2015), a series of end-of-pipe pollutant controls (e.g., Selective Catalytic Reduction techniques) were carried out for power, industry and transportation sectors. These measures effectively controlled the national NO<sub>x</sub> emissions by 22.8% from 2012 to 2016 (MEIC v1.3) in China. To quantify the effects of recent NO<sub>x</sub> control measures on the levels of photochemical oxidants and particulate nitrate, we conduct an additional simulation with NO<sub>x</sub> emissions set to the levels of 2012 in E1. The model simulations (Fig. 10) suggest that reducing China's NO<sub>x</sub> emissions alone from 2012 to 2016 leads to an average -24.93%~-8.62% decrease of NO<sub>x</sub> concentrations in the surface layer. As previously pointed out, the 2012-2016 NO<sub>x</sub> emission control measures lead to increased O<sub>3</sub> and OH levels in winter, which offset the effectiveness of NO<sub>x</sub> emission reduction in alleviating winter nitrate. No obvious declines in the winter nitrate levels are observed and even increase in some areas (+8.82% in BTH and 14.41% in YRD; Fig. S2-S3). As shown, the largest PM<sub>2.5</sub> responses shift towards the southern Hebei and central China provinces, where the wintertime PM<sub>2.5</sub> concentrations are particularly high in this region. The substantial emission changes from 2012 to 2016 lower the PM<sub>2.5</sub> air pollution by up to -1.84% in BTH and -3.52% in YRD for Period I and oppositely increase PM<sub>2.5</sub> by 2.36% in BTH and 4.67% in YRD for Period II. The past NO<sub>x</sub> emission control strategy leads to increased atmospheric oxidant levels and deteriorate particulate pollution in winter due to the nonlinear photochemistry and aerosol chemical feedbacks, without regard to the other control measures. This conclusion is also supported by evidence from the

recent field observations (Fu et al., 2020).

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#### 4 Conclusions

via controlling emissions of SO<sub>2</sub> and NO<sub>x</sub>, but raised a new question of how effective NO<sub>x</sub> emission controls can be on the mitigation of emerging nitrate and ozone air pollution. We use comprehensive measurements and a regional meteorologychemistry model with optimized mechanisms to establish the nonlinear responses between particulate nitrate and NO<sub>x</sub> emission controls in the megalopolises of China. Nitrate is an essential component of PM<sub>2.5</sub> in eastern China, accounting for 9.4– 15.5% and 11.5–32.1% of the total PM<sub>2.5</sub> mass for the warm and cold seasons, respectively. We find that the efficiency of PM<sub>2.5</sub> reduction is highly sensitive to NO<sub>x</sub> emission and it varies in different seasons depending on the ozone chemical regimes. The reduction of NO<sub>x</sub> emissions during summer-autumn results in almost linearly lower PM<sub>2.5</sub> by -2.65% in BTH and -2.79% in YRD per 10% cut of NO<sub>x</sub> emissions, whereas it increases the atmospheric oxidant levels (28.51% and 36.92% for O<sub>3</sub> and OH in BTH and 34.36% and 33.6% in YRD) and leads to a rather complicated response of the PM components in winter. Nitrate is found to increase ( $\beta$ =+4.28% in BTH and +4.60% in YRD) in winter with higher N<sub>2</sub>O<sub>5</sub> intermediate produced from the increased ozone introduced by attenuated titration, despite the nearly unchanged or slightly decreased HNO<sub>3</sub> concentrations in response to NO<sub>x</sub> control. An inflexion point appears at 40–50% NO<sub>x</sub> emission reduction, and a further reduction of NO<sub>x</sub> emission is predicted to cause -8.74% reductions of particulate nitrate for BTH and -10.59% for YRD. In addition, the 2012–2016 NO<sub>x</sub> emission control strategy leads to -24.93%~-8.62% decrease of surface NO<sub>x</sub>, and no change or even increase of wintertime nitrate in BTH (+8.82%) and YRD (14.41%).

Recent air pollution actions have significantly lowered the PM<sub>2.5</sub> levels in China





404 Our results emphasize that future PM<sub>2.5</sub> pollution mitigation strategies should 405 also focus on reducing the key oxidants involved in secondary aerosol production. 406 VOC<sub>s</sub>, which is not a direct precursor for SIA, is effective in SIA controls due to their 407 influence on atmospheric oxidation cycles (Tsimpidi et al., 2008; Womack et al., 2019; 408 Nguyen and Dabdub, 2002). Womack et al. (2019) evaluated the impacts of 409 traditional ozone mitigation strategies on ammonium nitrate production in the Salt 410 Lake Valley, USA. They found that the ammonium nitrate aerosol pollution is 411 responsive to VOC<sub>s</sub> control and not initially responsive to NO<sub>x</sub> control. Tsimpidi et al. 412 (2008) also showed that the reduction of VOC<sub>s</sub> emissions caused a marginal increase 413 of PM<sub>2.5</sub> during summer in eastern United States, whereas it resulted in a decrease of 414 oxidant levels and 5-20% reduction of both inorganic and organic PM<sub>2.5</sub> components during winter. Larger and synchronized NO<sub>x</sub> and VOC<sub>s</sub> emissions reductions are 415 416 required to overcome the adverse effects of nonlinear photochemistry and aerosol 417 chemical feedbacks. 418 Atmospheric NH<sub>3</sub> also acts as a critical neutralizing species for SIA production 419 and efficient haze mitigation (Liu et al., 2019). Atmospheric chemistry modeling 420 indicated that controlling NH<sub>3</sub> emission would significantly reduce the population-421 weighted PM<sub>2.5</sub> concentration by 6.2–21% with 60–100% NH<sub>3</sub> reductions in January, 422 implying the need to consider NH<sub>3</sub> emission controls when designing the PM<sub>2.5</sub> 423 pollution mitigation strategies (Wen et al., 2021). China's anthropogenic emissions are 424 estimated to decrease by 62% for SO<sub>2</sub> and 17% for NO<sub>x</sub> over 2010-2017, while 425 NH<sub>3</sub> emissions slightly increased by 1% (Zheng et al., 2018). The recent "Three-year 426 Action Plan Fighting for a Blue Sky" calls for agricultural NH3 emission controls but 427 without a specific reduction target. Such emission changes would emphasize the need 428 to jointly consider multi-pollutants emissions controls for mitigating SIA air pollution.

**Author contribution** 





430 Mengmeng Li developed the model code, designed the numerical experiments, 431 and wrote the original draft. Zihan Zhang carried out the numerical experiments. Shu 432 Li and Bingliang Zhuang validated and analyzed the model results. Tijian Wang and 433 Min Xie reviewed and revised the manuscript. 434 **Competing interests** 435 The authors declare that they have no conflict of interest. Acknowledgement 436 This study is funded by the National Natural Science Foundation of China 437 (41975153, 42077192 and 41775056), the National Key Basic Research Development 438 Program of China (2019YFC0214603, 2020YFA0607802), and the Emory 439 440 University-Nanjing University Collaborative Research Grant. 441 Data availability statement 442 The WRF-Chem model 4.1 available version is at 443 http://www2.mmm.ucar.edu/wrf/users/downloads.html. The NCEP FNL data are 444 accessible at the National Center for Atmospheric Research (NCAR) Research Data Archive (RDA; http://rda.ucar.edu/datasets/ds083.2/). The MEIC anthropogenic 445 446 emission inventories are available at www.meicmodel.org, and for more information, 447 please contact Q. Zhang (qiangzhang@tsinghua.edu.cn). The surface weather data are accessible at the Integrated Surface Database (https://www.ncdc.noaa.gov/isd/data-448 449 access). The surface air pollutants and aerosol species data are provided by Chinese 450 National Environmental Monitoring Center (http://www.cnemc.cn/en/) and archived 451 at https://doi.org/10.6084/m9.figshare.12818807.v1.





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Table 1. The emission scenarios in WRF-Chem numerical experiments

Simulation scenarios	Descriptions
В0	Base simulation under the 2016 emission conditions.
N0	Same as B0, but only consider the gas-phase oxidation pathway for the production of nitrate aerosol.
$C_N$ (N=1/2//8)	Same as B0, but anthropogenic $NO_x$ emissions are reduced by 10%, 20%80%, respectively, relative to the usual levels in 2016.
E1	Same as B0, but anthropogenic $NO_x$ emissions are replaced using the MEIC inventory in 2012.

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**Table 2**. Statistical evaluations of the model meteorological performance

Variable	Obs	Sim	R a	MB <sup>a</sup>	NMB <sup>a</sup>	ME <sup>a</sup>	RMSE <sup>a</sup>
Period I (15 August to 16 September)							
Temperature (°C)	24.04	23.91	0.89	-0.13	-0.55%	1.98	2.63
Humidity (%)	70.89	66.88	0.78	-4.01	-5.65%	11.07	14.67
Wind speed (m s <sup>-1</sup> )	2.46	3.04	0.50	0.58	23.72%	1.38	1.83
Period II (24 November to 26 December)							
Temperature (°C)	3.43	3.40	0.94	-0.03	-0.80%	2.18	2.83
Humidity (%)	69.85	65.27	0.63	-4.58	-6.56%	13.51	17.88
Wind speed (m s <sup>-1</sup> )	2.61	3.66	0.55	1.06	40.64%	1.70	2.23

<sup>a</sup> R: correlation efficient; MB: mean bias; NMB: normalized mean bias; ME: mean

error; RMSE: root mean square error.

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Table 3. Statistical evaluations of the model chemical performance

Variable	Obs	Sim	MB	NMB	Obs	Sim	MB	NMB
	Period I			Period II				
PM <sub>2.5</sub>	36.88	33.22	-3.66	-9.92%	91.59	64.28	-27.31	-29.82%
$\mathrm{SO}_2$	17.65	16.51	-1.14	-6.46%	41.45	29.80	-11.65	-28.11%
$NO_2$	28.53	33.23	4.70	16.47%	53.01	54.28	1.27	2.40%
Daily- maximum O <sub>3</sub>	237.45	255.77	18.32	7.72%	125.62	86.61	-39.01	-31.05%



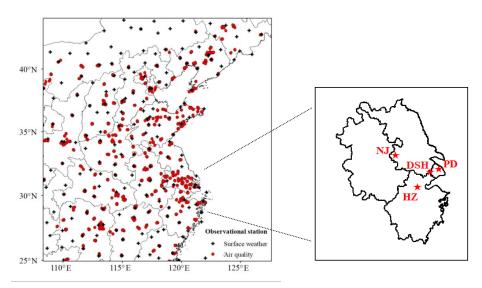
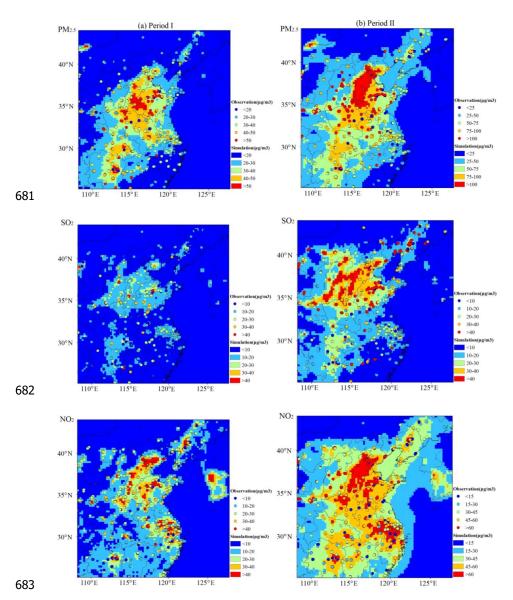


Fig. 1. WRF-Chem domain configuration and observational stations. Black crosses:
surface weather stations; Red dots: CNEMC routine air quality monitoring stations;

Red stars: surface supersites in YRD.

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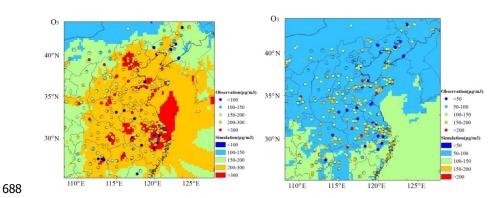




**Fig. 2**. Spatial patterns of the surface average PM<sub>2.5</sub>, NO<sub>2</sub>, SO<sub>2</sub> and daily-maximum O<sub>3</sub> concentrations in Period I (left panels) and Period II (right panels) from the WRF-Chem modeling (shaded contours) and routine air quality observations (dots).

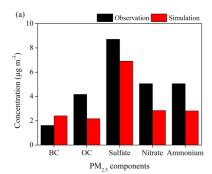


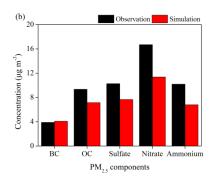




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Fig. 2. Continued.





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**Fig. 3**. Comparisons of surface  $PM_{2.5}$  components from WRF-Chem simulations and observations in Period I (a) and Period II (b) at the four supersites in YRD.



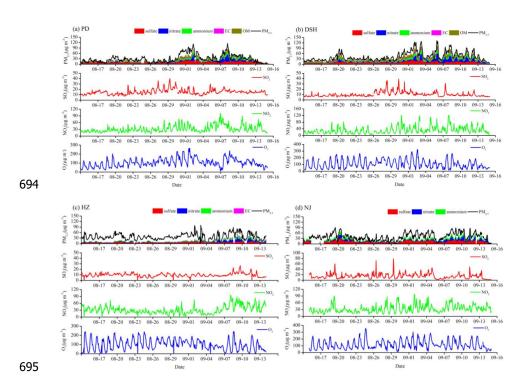


Fig. 4. Observed aerosol composition and gaseous pollutants concentrations at thefour supersites during Period I.





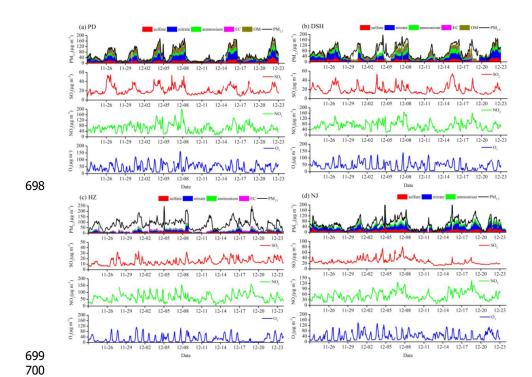
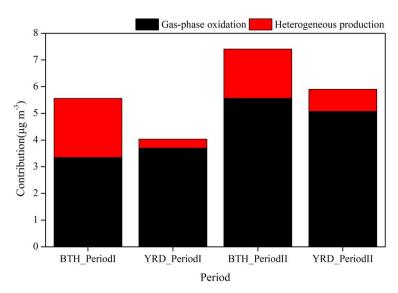


Fig. 5. Same as Fig. 3, but for Period II.

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**Fig. 6**. Contributions of gas-phase oxidation and heterogeneous production to the surface nitrate concentrations for the BTH and YRD regions in two seasons.

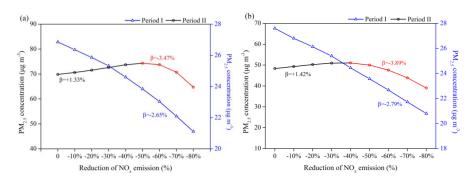


Fig. 7. Responses of surface  $PM_{2.5}$  concentrations to the  $NO_x$  emission reduction scenarios in (a) BTH and (b) YRD. The calculated  $NO_x$  emission control efficiency ( $\beta$ ) is also marked in the figure.

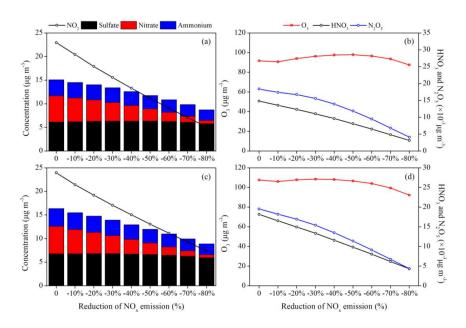
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**Fig. 8**. Responses of the concentrations of surface SIA and key atmospheric trace gases ( $NO_2$ ,  $O_3$ ,  $HNO_3$  and  $NO_3$ ) to the  $NO_x$  emission reduction scenarios in (a, b) BTH and (c, d) YRD during Period I.

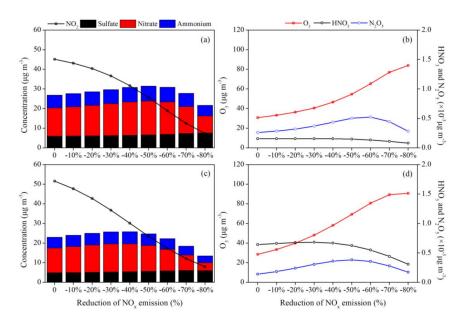


Fig. 9. Same as Fig. 7, but for Period II.

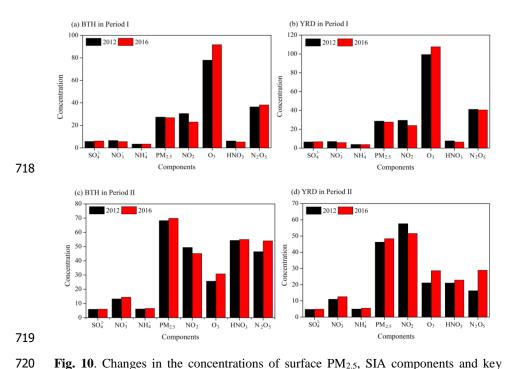
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**Fig. 10**. Changes in the concentrations of surface  $PM_{2.5}$ , SIA components and key atmospheric trace (NO<sub>2</sub>, O<sub>3</sub>, HNO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub>) due to the 2012–2016 NO<sub>x</sub> emission reductions in China estimated as the differences between the base simulation and E1 scenario. The units are ppt for HNO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub>, and  $\mu$ g m<sup>-3</sup> for other chemical species.