

Response to Reviewer2

General comments:

*This paper investigates the secondary inorganic aerosol formation in cities in China focusing on nitrate aerosol, using observed data from a Chinese network of station and a WRF-Chem mesoscale model to analyze the response of PM_{2.5} to NO_x emission changes on a seasonal basis. The main findings are the almost linear response of secondary inorganic nitrogen to NO_x emissions changes during summer while in winter the system is highly non-linear. Sensitivities to ammonia and sulfur dioxide emissions, which have been shown to be important, are discussed only based on existing literature. **The paper is within the scope of ACP and could be suitable for publication in the journal after a number of improvements suggested below.***

Major comments:

My major comment is that the authors miss to show a clear view of the interplay between O₃, SO₂, NO_x, NH₃ and PM_{2.5}. The authors provide some information in this respect in the conclusion but the manuscript can be largely improved in this respect. Therefore, I would suggest that the authors check on relevant references and integrate some of them in the introduction or discussion session. Some are listed below but more might be found by a literature search.

Response: Accepted. Several improvements are performed in the revised manuscript as below:

- (1) To show the responses of nitrate and PM_{2.5} pollution to the cooperative controls of NO_x, SO₂, NH₃ and VOC_s, three series of additional simulations (see *Line 160-165* and *Table 1*; C_{S-N}, C_{N-N} and C_{V-N}) are conducted with the anthropogenic emissions of SO₂, NH₃ and VOC_s in China cut by 20%, 40%...and 80%. Our results emphasize that atmospheric NH₃ and VOC_s are effective in controlling the particulate nitrate pollution, whereas decreasing the SO₂ and NO_x emissions may have counter-intuitive effects on nitrate aerosols. For details, see *Sec. 3.3.4*.
- (2) More relevant references are supplemented in the introduction section to explain the interplay between O₃, NO_x and PM_{2.5}. For details, see *Line 77-86*.
- (3) We extend the discussions about the formation pathways for nitrate calculated in the WRF-Chem model and their comparisons with recent assessments in China and worldwide. The calculated contributions (60.19–91.71% for NO₂+OH oxidation and 8.29–39.81% for heterogeneous pathways) are in line

with previous assessments in China and globally (Alexander et al., 2009; Alexander et al., 2020; Wang et al., 2019; Wang et al., 2017; Luo et al., 2021; Liu et al., 2020; He et al., 2020). For details, see *Line 148-153* and *Line 283-297*.

Specific Comments:

Line 189: capture

Response: Accepted. For details, see *Line 205*.

Line 194: MB

Response: Accepted. This sentence has been rephrased as “The model statistical evaluations show a mean bias (MB)...”. For details, see *Line 210-211*.

Line 211-212: that the consideration of the optimized... reduces...

Response: Accepted. For details, see *Line 228*.

Lines 228: was more than ...

Response: Accepted. For details, see *Line 246*.

Line 232: (because OM has more than C, O and H) I suggest adding ‘mainly’ for the hydrogen and oxygen

Response: Accepted. For details, see *Line 250*.

Line 235: could you provide standard deviation for percent fractions ?

Response: Accepted. We give the ranges for the percent fractions. This sentence is rephrased as “The three SIA components together account for 32.3–57.4% (48.6% on average) and 27.7–70.9% (56.9% on average) of the total PM_{2.5} mass...”. For details, see *Line 253-254*.

Lines 236: SIA are the dominant components of PM_{2.5} – 48.6% is not the majority – it is below 50%

Response: Accepted. This sentence is rephrased as “...become the primary components of PM_{2.5} in the two periods”. For details, see *Line 255*.

Lines 237-242: please explain the drivers, are temperature, liquid water content of the aerosol impacting these ratios

Response: Accepted. The seasonal distinctions for sulfur and nitrogen oxidation ratios have also been reported by Shu et al. (2019). They attributed the weakened conversion from NO₂ to nitrate in summer to the volatility and evaporative loss of nitrate (Sun et al., 2012). The sharp increase of particles and moderate ambient humidity in winter also benefit the heterogeneous formation of SIA, leading to high NOR and SOR (Wang et al., 2012). For details, see *Line 267-274*.

Lines 258-259: discussion could profit from additional literature – also heterogeneous reactions on dust can produce nitrate aerosol (see references above)

Response: Accepted. We extend the discussions about the formation pathways for nitrate calculated in the WRF-Chem model and their comparisons with recent assessments in China and worldwide. The calculated contributions (60.19–91.71% for NO₂+OH oxidation and 8.29–39.81% for heterogeneous pathways) are in line with previous assessments in China and globally (Alexander et al., 2009; Alexander et al., 2020; Wang et al., 2019; Wang et al., 2017; Luo et al., 2021; Liu et al., 2020; He et al., 2020). The nitrate formation from heterogeneous reactions is moderately underestimated in this study, possibly due to the uncertainties of heterogeneous uptake coefficients and unclear reaction mechanisms applied in the model (Li et al., 2019; Xue et al., 2016; He et al., 2014). For details, see *Line 283-297*.

Line 267: nonlinear behaviour in winter

Response: Accepted. For details, see *Line 305*.

Section 3.3.1 & 3.3.3, I miss here the discussion on the involvement of SO₂ emission changes to PM_{2.5} aerosols and thereby to O₃ and nitrate aerosol

Response: Accepted. A series of simulations (C_{S-N}; see *Line 160-165* and *Table 1*) are supplemented to show the responses of nitrate and PM_{2.5} pollution to the SO₂ emission controls. The SO₂ emission reduction, although effective in reducing sulfate and PM_{2.5} (*Fig. 12* in the revised manuscript), is not successful in regulating the nitrate pollution due to the chemical competition in nitrate and sulfate formations (Geng et al., 2017; Wang et al., 2013). Changes in nitrate concentration are linearly associated with the SO₂ emission reductions, with the β values of 2.90% during summer-autumn and 1.35% during winter. For details, see *Sec. 3.3.4* and *Fig. 12*.

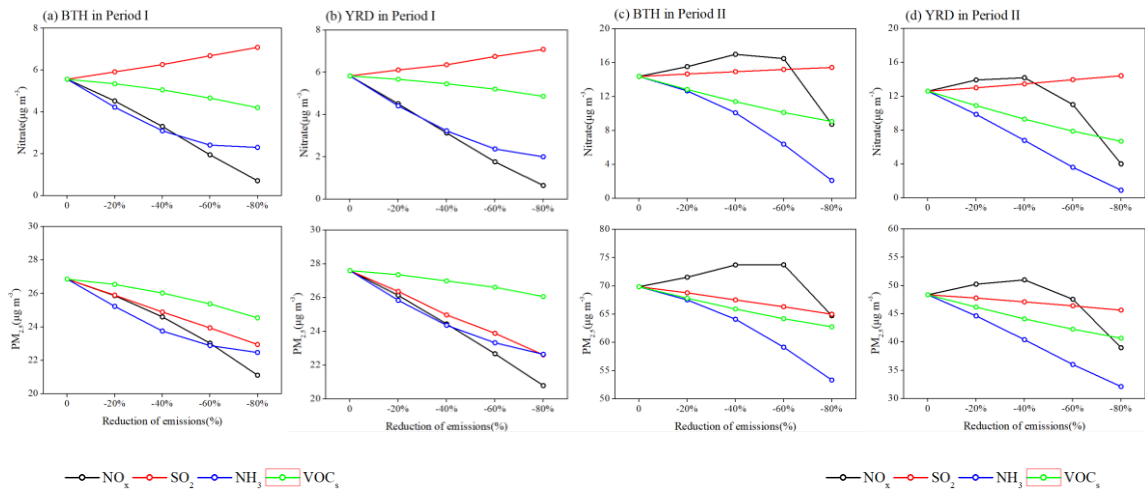


Fig. 12. Responses of the surface nitrate (upper panels) and PM_{2.5} (bottom panels) concentrations to the emission reduction scenarios of NO_x, SO₂, NH₃ and VOC_s during Period I (a, b) and Period II (c, d).

Line 310: have minor changes

Response: Accepted. This sentence is rephrased as “The β values for surface sulfate change by $-0.74\% \sim +1.16\%$ in BTH and $-1.54\% \sim +0.17\%$ in YRD under the $-10 \sim -80\%$ NO_x emission reduction scenarios”. For details, see *Line 348-350*.

Lines 312, 313, 345-348: How significant are these correlations?

Response: Accepted. The R^2 values are supplemented in Eq. 1-3. For details, see *Line 351-352* and *Line 368-371*.

Line 320: do not forget acid replacement reactions (e.g. reactions on dust or sea-salt aerosol)

Response: Accepted. This sentence is rephrased as “The formation of nitrate mainly involves the $\text{NO}_2 + \text{OH} \rightarrow \text{HNO}_3$ gas-phase oxidation and the heterogeneous hydrolysis of N_2O_5 and other nitrogenous gases”. For details, see *Line 358-360*.

Lines 324-325: do not forget here the reduction of $\text{NO} + \text{O}_3$ titration reaction

Response: Accepted. This sentence is rephrased as “...due to the reduction of $\text{NO} + \text{O}_3$ titration reaction and the greater VOC availability in the warm season”. For details, see *Line 365-366*.

Line 337: reactions

Response: Accepted. For details, see *Line 381*.

Line 391 move 'during summer-autumn' to the end of line 392

Response: Accepted. For details, see *Line 484*.

Lines 408-415 and 419-427 would fit better in introduction or discussion. Please move appropriately.

Response: Accepted. Three series of additional simulations (see *Line 160-165* and *Table 1*; C_{S-N} , C_{N-N} and C_{V-N}) are conducted to show the responses of nitrate and $PM_{2.5}$ pollution to the cooperative controls of NO_x , SO_2 , NH_3 and VOCs, respectively. These texts have been moved to *Sec. 3.3.4*. For details, see *Sec. 3.3.4*.

References:

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