

Response to Anonymous Referee #2 on

“Large hemispheric difference in nucleation mode aerosol concentrations in the lowermost stratosphere at mid and high latitudes”

5 Christina J Williamson et al.
2021-03-16

10 We thank the referee for their thorough and pertinent review. We have addressed the comments as detailed here below and are grateful from the improvement this has made to the manuscript.

15 *The description of how box modelling is done needs to be detailed better. In its current form it is really difficult to understand how to simulations are done. The reader is pointed to Kupc et al., 2020 for the description of the box model setup. However, in that paper both MAIA and TOMAS models are used, so it would be easier for the reader to understand the modelling part if it was briefly summarized in this paper.*

We have expanded the description of MAIA at line 166 to read as follows:

20 “To more quantitatively assess the effects of thermodynamics on NPF in the LMS, box modeling is performed using the Model of Aerosols and Ions in the Atmosphere (MAIA), (Lovejoy et al., 2004;Kazil and Lovejoy, 2007;Kazil et al., 2007). MAIA describes the oxidation of SO₂ to gaseous H₂SO₄, the nucleation of neutral and negative H₂SO₄-H₂O clusters, aerosol growth by sulfuric acid condensation/evaporation, and particle coagulation. The production rate of H₂SO₄ is
25 calculated assuming that the reaction of SO₂+OH is the rate limiting step of the oxidation of SO₂ to form H₂SO₄ (Lovejoy et al., 1996). Nucleation is described with laboratory thermochemical data for H₂SO₄ and H₂O uptake and loss by small neutral and negative clusters (Curtius et al., 2001;Lovejoy and Curtius, 2001;Froyd and Lovejoy, 2003;Hanson and Lovejoy, 2006). The thermochemical data for uptake and loss of H₂SO₄ and H₂O by large sulfuric acid aerosol (>> 5
30 sulfuric acid molecules) are based on the liquid drop model and H₂SO₄ and H₂O vapor pressures over bulk solutions. These were calculated with a computer code (provided by S. L. Clegg, personal communication, 2007) which adopts experimental data from Giauque et al. (1960) and (Clegg et al., 1994). The thermochemical data for intermediate sized particles are a smooth interpolation of the data for small and large aerosol particles. The model uses 20 linear bins in
35 which H₂SO₄ content increases by 1 molecule per bin, and 50 geometric bins in which H₂SO₄ content increases by a factor of 1.45 per bin, covering a dry (312.15 K, 10% RH) particle diameter range of ~0.5–800 nm.

MAIA operates along trajectories with changing pressure, temperature and relative humidity
40 (Kazil and Lovejoy, 2007) in the temperature range 180-320 K and the relative humidity range 1-101 %, which includes upper troposphere conditions. MAIA parametrizes the OH diurnal cycle as a half-sine centered around noon with a prescribed noon OH concentration, while setting the nighttime OH concentration to 0. The length of the daytime period is calculated from the day of year and location. Atmospheric ionization rates due to galactic cosmic rays are calculated as a
45 function of latitude, altitude, and solar cycle phase by a model of energetic particle transport in the Earth’s atmosphere (O'Brien, 2005). The transformation between geographic and

geomagnetic coordinates is calculated with GEOPACK (<http://geo.phys.spbu.ru/~tsyganenko/modeling.html>) and the International Geomagnetic Reference Field 12 coefficients (<https://www.ngdc.noaa.gov/AGA/vmod/igrf.html>).

Aerosol nucleation rates calculated from the experimental thermochemical data of neutral and charged H₂SO₄/H₂O cluster formation that are used in MAIA (Kazil and Lovejoy, 2007) compare well with neutral and charged H₂SO₄/H₂O nucleation rates measured in the European Organization for Nuclear Research (CERN) Cosmics Leaving Outdoor Droplets (CLOUD) chamber (Kirkby et al., 2011). Global model simulations, either using a parameterization of neutral and charged H₂SO₄/H₂O nucleation based on the CERN CLOUD chamber measurements, or nucleation rates calculated from the experimental thermochemical data used in MAIA (Kazil et al., 2010) show a good agreement in the global mean profile of total (> 3 nm) aerosol concentration (Määttä et al., 2018).

MAIA is run along back-trajectories, initiated at the aircraft location, which were calculated using the Traj3D trajectory model (Bowman, 1993) and the National Center for Environmental Prediction (NCEP) global forecast system (GFS) meteorology (2015). NCEP provides temperature, relative humidity, and pressure along the trajectories for the MAIA runs. The initial SO₂ concentration and the H₂SO₄ condensation sink of the initial aerosol size distribution were estimated from ATom observations at similar latitudes and altitudes. The geometric mean diameter (46 nm) and geometric standard deviation (2.8) of the initial aerosol size distribution were obtained by fitting a lognormal mode to the size distribution observed at the ATom measurement locations. The noon concentration of OH in the simulations was set to 3×10⁶ molec. cm⁻³. This estimate agrees well with aircraft-measured OH concentrations during ATom (Kupc et al., 2020)."

The comparison between CEDS emission rates and observed aerosol concentrations in Figure 11 is problematic since it in now way takes into account the transport of SO₂. As mentioned in the text, observations were made mostly outside of the flight corridors and there the number concentrations of ultrafine particles did not correlate with SO₂ concentration. Can it be that new particle formation occurred near flight corridors and these particles were transported to the regions of aircraft observations?

Regarding Fig 11 and the comparison of measured SO₂ flux with CEDS emissions, we have indeed omitted to explicitly address transport. We have noted in the manuscript the lifetime of SO₂ as around 1 month (line 394, and SM section 2) and zonal transport from flight corridors to the most distant observations from those corridors is on the order of half a day (line 386). Using the median observed windspeed of 10-30 m/s (line 385) we can also see that the complete zonal mixing timescale is on the order for 1 month, approximately equal to the lifetime of SO₂. Since the jet core with peak wind speeds of 40-60 m/s is typically located on the tropopause in the 30-50N region (Manney et al., 2014), and the peak in SO₂ is also located in this region, the zonal mixing time could be as little as one to two weeks. We believe this justifies the presentation of zonally averaged SO₂ in Fig 11 parts b and c. We agree with the referee that this was not adequately addressed in the original manuscript and so propose to include the following explanation at line 386:

"Based on these measured windspeeds, zonal mixing is expected on timescales of around 1 month, which is approximately equal to the lifetime of SO₂. However, since the jet core with peak wind speeds of 40-60 m/s is typically located on the tropopause in the 30-50N region

(Manney et al., 2014), and the peak in SO₂ is also located in this region, the zonal mixing time could be as little as one to two weeks. For this reason, we present zonally averaged SO₂ concentrations in figure 11b and c.”

The referee asks an interesting question about whether NPF could be occurring within flight corridors and then transported to the ATom flight paths where we observed them. It is important to note that SO₂ lifetimes are estimated to be much larger than nucleation mode aerosol lifetimes (1 month (line 394) for SO₂, compared with a few days (line 147)) for nucleation mode aerosols). We therefore do not see a mechanism for transporting newly formed particles from regions of higher SO₂ concentrations to regions of lower SO₂ concentrations, given the assumption that SO₂ is driving nucleation. Where there is a lack of correlation between SO₂ and nucleation mode aerosol, we believe this is more likely to be due a different chemical source for aerosol nucleation, lack of measurement precision or the different sinks and lifetimes of SO₂ and aerosols.

Page 2, Lines 36-37: Solomon et al., 2011 does not discuss aerosol size distributions. Wouldn't Williamson et al., 2019 reference be more suitable reference here?

We thank the referee for catching our mistake with the Solomon et al 2011 on lines 36-37. We think the Williamson et al 2019 reference is not ideal here, because that paper concerns ATom observations in the tropics, which generally did not penetrate into the stratosphere. Instead, we have corrected this to:

“Aerosols in the lowermost stratosphere (LMS) are highly variable, even in the absence of major volcanic eruptions (Solomon et al., 2011), and models currently struggle to reproduce observed aerosol size distributions in this region of the atmosphere (Murphy et al., 2020).”

Page 2, Lines 126-130: Where is this information about condensation and coagulation rates used?

The condensation rates explained in the methods section line 126-130 are used to calculate the condensation sink for comparison with the MAIA box modeling presented in Figure 7 and discussed at lines 170 and 203-230. The coagulation rates are used in the supplementary material section S2 to estimate the lifetime of particles in the LMS (Fig S4), which is then referenced in the main manuscript at line 147 and subsequently. To clarify this in the text we have added the following sentence at line 132:

“Condensation and coagulation rates will be used in this analysis to relate our observations to theory and models, and to estimate particle lifetimes.”

Page 5, Line 146: oder → order

We have corrected the spelling mistake on line 146 and thank the referee for spotting that.

Page 8, Lines 229-237: Here you discuss that the SO2 concentrations (40 pptv) were higher than the median observed values. Why was this value chosen? Was this level required to initiate nucleation? It is said that SO2 concentrations decreased to the observed concentrations. Was NPF still ongoing at these levels?

MAIA was run only at a few discrete levels SO₂ because of constraints on computing time. The pertinent levels were 20 and 40 pptv, results of which are shown in Fig 7. As the referee points out, this is above the median observed values. NPF did occur at 20 pptv, but slower growth meant that, for higher condensation sinks, observable increases in number concentrations of particles larger than 2.65 nm were only seen on trajectories with higher RH_w at these SO₂ concentrations (Fig. 7f). At lower condensation sinks, increases in number concentrations of particles larger than 2.65 nm were clearly observable with SO₂ at 20 pptv (Fig. 7h).

Section 6: The first three paragraphs explain how aerosol emissions are calculated. The motivation for this procedure is unclear to me. Is this done in order to obtain higher temporal resolution emission rates from CEDS monthly fields?

In section 6 the first paragraph explains why we are considering aircraft as sources for small particles and SO₂. The second paragraph introduces the CEDS database and explains the relevant sources used therein. The third paragraph details how we calculate expected emissions of nucleation mode aerosol using the CEDS database, using the data on aircraft emissions therein, and literature values to convert from SO₂ emissions to nucleation mode particle emissions, since the nucleation mode particle emissions were not, to our knowledge, included in the CEDS database. We would like to understand better the aspects of this that are unclear, or potentially duplicating CEDS fields in order to make any necessary corrections.

Page 13, Line 400: Can higher observed SO₂ concentrations be due to transport from regions with higher emissions?

In the SH LMS, the higher observed SO₂ concentrations relative to CEDS emissions referred to on line 400 may well be due to transport, or perhaps small, more local sources, but we do not have the data to make any firm conclusions regarding this.

Page 14, Line 430: "influence from fires may suppress NPF in the LMS, or that the additional surface area from biomass burning particles shortens the lifetime of newly formed particles". Aren't these two the same thing?

On line 430, we were attempting to reference the influence of biomass burning on two distinct processes – the nucleation of aerosols, and the subsequent growth of those particles. The referee has shown that this lacked clarity, so we have modified the sentence from the original "suggesting that the influence from fires may suppress NPF in the LMS, or that the additional surface area from biomass burning particles shortens the lifetime of newly formed particles" to

"suggesting that the additional surface area from biomass burning particles may reduce nucleation mode number concentrations in the LMS through two mechanisms, suppressing the formation of particles by increasing the condensation sink, and shortening the lifetime of particles that do form by increasing the coagulation sink."

Unsolicited corrections

We would like to alert the referee to some unsolicited corrections we would like to make to the manuscript that came up in the course of addressing referee comments. These are detailed here below. We would also like to alert the referee to proposed corrections in our response to referee #1.

Extraneous comma removed line 57

For clearer reading, Line 62 change from
"Ammonia and amines have been shown to contribute to NPF"

190 to
 “Ammonia and amines have been shown to be involved in NPF”

Line 82: “lowermost stratosphere” changed to “LMS” for consistency.

195 Line 123 “CH3Cl” corrected to “CH₃Cl”

Line 127
 “using Fuchs expression for the coagulation rate coefficient(Seinfeld and Pandis, 2006)”
 corrected to
 200 “using the Fuchs expression for the coagulation rate coefficient (Seinfeld and Pandis, 2006)”

Line 141 changed from “in’ to “by” in “This stratospheric definition is consistent with that used
 by Murphy et al. (2020)”.

205 We noted a missing parenthesis on line 195 and so added this.

Line 227, alteration made in order to read better from
 “20 of the 55 trajectories experience more humid air, indicative of UT or tropopause conditions.”
 to
 210 “The other 20 trajectories experienced more humid air, indicative of UT or tropopause
 conditions.”

The full description of Asian Summer Monsoon (ASM) has been placed on the first usage at line
 253 instead of where it was mistakenly put on line 424.

215 Line 309 “Because the lifetime of these particles is ~ days” changed to “Because the lifetime of
 these particles is on the order of days”

Line 341 “from” added to “through NPF resulting from SO₂ oxidation”

220 Line 402 “higher observed concentrations ... are also likely” corrected from “is also likely”

Line 405 “we get a total flux” change to “we determine a total flux”

225 Line 435 “ASM-sources particles” corrected to “ASM-sourced particles”
 Line 465 “volcano emissions” changed to “volcanic emissions”

Line 451 “the only eruption to occur between the tropopause height: corrected to “the only
 eruption to reach between the tropopause height”

230 For clarity, line 495 has been rewritten from
 “How this would be achieved, and the potential consequences, both intended effects and side-
 effects are highly uncertain.”
 to

235 “How this would be achieved, and the potential consequences of such actions (both the intended effects and any unintended side-effects) are highly uncertain.”

We noted in the SM line 44 a forward slash had accidentally been used instead of a period, and have corrected this.

240

Additional references have been added at line 53 and the order of references changed to “New particle formation (NPF) has been well documented in a variety of locations in the planetary boundary layer and free troposphere (Clarke et al., 1998;Clarke et al., 2013;Kulmala et al., 2013;Williamson et al., 2019).”

245

Similarly, at line 284 which now reads
New particles have previously been shown to ... form in the tropical UT (Clarke et al., 1998;Clarke and Kapustin, 2002;Clarke et al., 2013;Williamson et al., 2019)

250 Figure 3: “level” removed from a,b y-axis and corrections made to caption text: “except for” changed to “except”, “grew” changed to “grey”

Figure 4: y-axes for NH and SH changed to be the same range to make comparison easier

255 Figure 8: greater than and less than symbols in legend corrected

Figure S3. We have added a more through explanation of the figure and how this was calculated from the data in the caption, and included an example mass spectrum to illustrate this. The new proposed figure and caption are as follows:

260

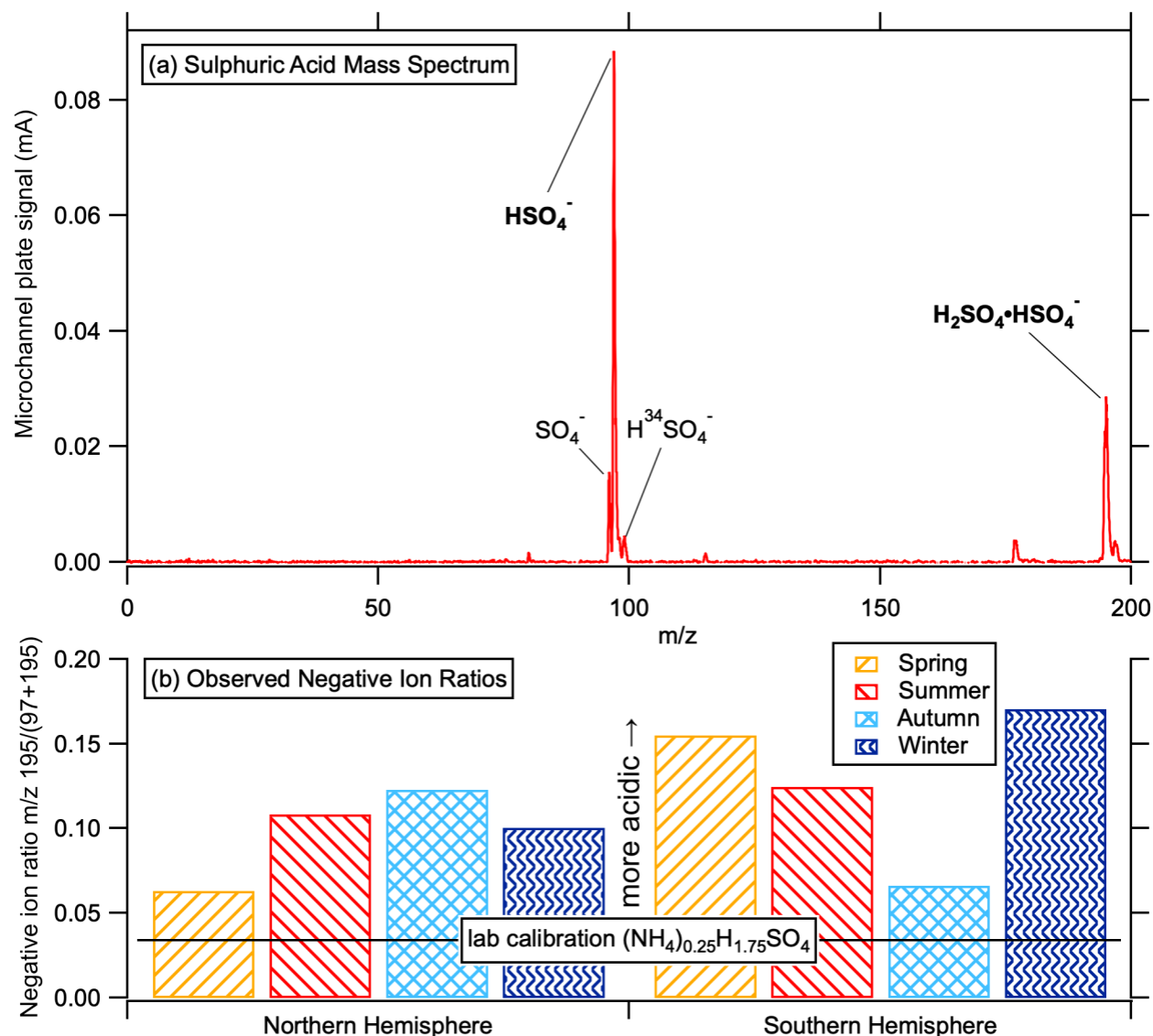


Figure S1: Aerosol acidity in the LMS. a) An example negative ion spectrum of an acidic sulphate particle. This spectrum is from a $0.39\ \mu\text{m}$ diameter particle in the stratosphere at 12.2 km and 310 ppbv of O_3 on 20171009. Laboratory calibrations show that the $\text{H}_2\text{SO}_4 \cdot \text{HSO}_4^-$ peak is very small or non-existent for particles composed of ammonium sulphate and the cluster ion peak increases with acidity until it is a large peak for nearly pure sulfuric acid. b) The bars for the NH (left) and SH (right), separated by season, show the average ratio of the size of the cluster peak at $m/z\ 195$ to the sum of the peaks at $m/z\ 195$ and 97. The averages are for particles when O_3 concentrations were 250 to 350 ppbv in the stratosphere. The averages are also for particles between 0.35 and $0.6\ \mu\text{m}$ diameter because in the stratosphere most particles of that size originated in the stratosphere. Lab calibrations of particles composed of $(\text{NH}_4)_{0.25}\text{H}_{1.75}\text{SO}_4$ had a negative ion ratio $m/z\ 195/(97+195)$ of 0.034, therefore we consider ratios higher than this (more acidic) to contain less than 0.25 mole fraction ammonium. Rather than analyzing possible differences in the acidity with season, here we emphasize that stratospheric particles in all seasons and both hemispheres are highly acidic. This sets limits on the possible concentration of gas phase ammonia. Calculations of uptake from the gas phase show that a continuous 1 pptv of gas phase ammonia could add 0.25 mol fraction ammonium to sulfuric acid particles in less than a week.

References

- 280 NCEP GFS 0.25-degree global forecast grids historical archive. Systems, N. U. R. D. A. C. a. I., and Lab (Eds.), 2015.
- Bowman, K. P.: Large-Scale Isentropic Mixing Properties of the Antarctic Polar Vortex from Analyzed Winds, *J Geophys Res-Atmos*, 98, 23013-23027, Doi 10.1029/93jd02599, 1993.
- 285 Clarke, A. D., Varner, J. L., Eisele, F., Mauldin, R. L., Tanner, D., and Litchy, M.: Particle production in the remote marine atmosphere: Cloud outflow and subsidence during ACE 1, *J Geophys Res-Atmos*, 103, 16397-16409, Doi 10.1029/97jd02987, 1998.
- Clarke, A. D., and Kapustin, V. N.: A pacific aerosol survey. Part I: A decade of data on particle production, transport, evolution, and mixing in the troposphere, *Journal of the Atmospheric Sciences*, 59, 363-382, Doi 10.1175/1520-0469(2002)059<0363:Apaspi>2.0.Co;2, 2002.
- 290 Clarke, A. D., Freitag, S., Simpson, R. M. C., Hudson, J. G., Howell, S. G., Brekhovskikh, V. L., Campos, T., Kapustin, V. N., and Zhou, J.: Free troposphere as a major source of CCN for the equatorial pacific boundary layer: long-range transport and teleconnections, *Atmos Chem Phys*, 13, 7511-7529, 10.5194/acp-13-7511-2013, 2013.
- 295 Clegg, S. L., Rard, J. A., and Pitzer, K. S.: Thermodynamic properties of 0–6 mol kg^{−1} aqueous sulfuric acid from 273.15 to 328.15 K, *J. Chem. Soc., Faraday Trans.*, 90, 1875-1894, 10.1039/FT9949001875, 1994.
- Curtius, J., Froyd, K. D., and Lovejoy, E. R.: Cluster Ion Thermal Decomposition (I): Experimental Kinetics Study and ab Initio Calculations for HSO₄-(H₂SO₄)_x(HNO₃)_y, *The Journal of Physical Chemistry A*, 105, 10867-10873, 10.1021/jp0124950, 2001.
- 300 Froyd, K. D., and Lovejoy, E. R.: Experimental Thermodynamics of Cluster Ions Composed of H₂SO₄ and H₂O. 1. Positive Ions, *The Journal of Physical Chemistry A*, 107, 9800-9811, 10.1021/jp027803o, 2003.
- Giauque, W. F., Hornung, E. W., Kunzler, J. E., and Rubin, T. R.: The Thermodynamic Properties of Aqueous Sulfuric Acid Solutions and Hydrates from 15 to 300°K.1, *Journal of the American Chemical Society*, 82, 62-70, 10.1021/ja01486a014, 1960.
- 305 Hanson, D. R., and Lovejoy, E. R.: Measurement of the Thermodynamics of the Hydrated Dimer and Trimer of Sulfuric Acid, *The Journal of Physical Chemistry A*, 110, 9525-9528, 10.1021/jp062844w, 2006.
- 310 Kazil, J., and Lovejoy, E. R.: A semi-analytical method for calculating rates of new sulfate aerosol formation from the gas phase, *Atmos. Chem. Phys.*, 7, 3447-3459, 10.5194/acp-7-3447-2007, 2007.

- Kazil, J., Lovejoy, E. R., Jensen, E. J., and Hanson, D. R.: Is aerosol formation in cirrus clouds possible?, *Atmos. Chem. Phys.*, 7, 1407-1413, 10.5194/acp-7-1407-2007, 2007.
- Kazil, J., Stier, P., Zhang, K., Quaas, J., Kinne, S., O'Donnell, D., Rast, S., Esch, M., Ferrachat, S., Lohmann, U., and Feichter, J.: Aerosol nucleation and its role for clouds and Earth's radiative forcing in the aerosol-climate model ECHAM5-HAM, *Atmos Chem Phys*, 10, 10733-10752, 10.5194/acp-10-10733-2010, 2010.
- Kirkby, J., Curtius, J., Almeida, J., Dunne, E., Duplissy, J., Ehrhart, S., Franchin, A., Gagné, S., Ickes, L., Kürten, A., Kupc, A., Metzger, A., Riccobono, F., Rondo, L., Schobesberger, S., Tsagkogeorgas, G., Wimmer, D., Amorim, A., Bianchi, F., Breitenlechner, M., David, A., Dommen, J., Downard, A., Ehn, M., Flagan, R. C., Haider, S., Hansel, A., Hauser, D., Jud, W., Junninen, H., Kreissl, F., Kvashin, A., Laaksonen, A., Lehtipalo, K., Lima, J., Lovejoy, E. R., Makhmutov, V., Mathot, S., Mikkilä, J., Minginette, P., Mogo, S., Nieminen, T., Onnela, A., Pereira, P., Petäjä, T., Schnitzhofer, R., Seinfeld, J. H., Sipilä, M., Stozhkov, Y., Stratmann, F., Tomé, A., Vanhanen, J., Viisanen, Y., Vrtala, A., Wagner, P. E., Walther, H., Weingartner, E., Wex, H., Winkler, P. M., Carslaw, K. S., Worsnop, D. R., Baltensperger, U., and Kulmala, M.: Role of sulphuric acid, ammonia and galactic cosmic rays in atmospheric aerosol nucleation, *Nature*, 476, 429-433, 10.1038/nature10343, 2011.
- Kulmala, M., Kontkanen, J., Junninen, H., Lehtipalo, K., Manninen, H. E., Nieminen, T., Petaja, T., Sipilä, M., Schobesberger, S., Rantala, P., Franchin, A., Jokinen, T., Jarvinen, E., Aijala, M., Kangasluoma, J., Hakala, J., Aalto, P. P., Paasonen, P., Mikkilä, J., Vanhanen, J., Aalto, J., Hakola, H., Makkonen, U., Ruuskanen, T., Mauldin, R. L., Duplissy, J., Vehkamäki, H., Back, J., Kortelainen, A., Riipinen, I., Kurten, T., Johnston, M. V., Smith, J. N., Ehn, M., Mentel, T. F., Lehtinen, K. E. J., Laaksonen, A., Kerminen, V. M., and Worsnop, D. R.: Direct Observations of Atmospheric Aerosol Nucleation, *Science*, 339, 943-946, 10.1126/science.1227385, 2013.
- Kupc, A., Williamson, C. J., Hodshire, A. L., Kazil, J., Ray, E., Bui, T. P., Dollner, M., Froyd, K. D., McKain, K., Rollins, A., Schill, G. P., Thames, A., Weinzierl, B. B., Pierce, J. R., and Brock, C. A.: The potential role of organics in new particle formation and initial growth in the remote tropical upper troposphere, *Atmos. Chem. Phys.*, 20, 15037-15060, 10.5194/acp-20-15037-2020, 2020.
- Lovejoy, E. R., Hanson, D. R., and Huey, L. G.: Kinetics and Products of the Gas-Phase Reaction of SO₃ with Water, *The Journal of Physical Chemistry*, 100, 19911-19916, 10.1021/jp962414d, 1996.
- Lovejoy, E. R., and Curtius, J.: Cluster Ion Thermal Decomposition (II): Master Equation Modeling in the Low-Pressure Limit and Fall-Off Regions. Bond Energies for HSO₄-(H₂SO₄)_x(HNO₃)_y, *The Journal of Physical Chemistry A*, 105, 10874-10883, 10.1021/jp012496s, 2001.
- Lovejoy, E. R., Curtius, J., and Froyd, K. D.: Atmospheric ion-induced nucleation of sulfuric acid and water, *J Geophys Res-Atmos*, 109, Artn D08204 10.1029/2003jd004460, 2004.

- 350 Määttänen, A., Merikanto, J., Henschel, H., Duplissy, J., Makkonen, R., Ortega, I. K., and
Vehkamäki, H.: New Parameterizations for Neutral and Ion-Induced Sulfuric Acid-Water Particle
Formation in Nucleation and Kinetic Regimes, *Journal of Geophysical Research: Atmospheres*,
123, 1269-1296, <https://doi.org/10.1002/2017JD027429>, 2018.
- 355 Manney, G. L., Hegglin, M. I., Daffer, W. H., Schwartz, M. J., Santee, M. L., and Pawson, S.:
Climatology of Upper Tropospheric–Lower Stratospheric (UTLS) Jets and Tropopauses in
MERRA, *J Climate*, 27, 3248-3271, 10.1175/jcli-d-13-00243.1, 2014.
- Murphy, D. M., Froyd, K. D., Bourgeois, I., Brock, C. A., Kupc, A., Peischl, J., Schill, G. P.,
Thompson, C. R., Williamson, C. J., and Yu, P.: Radiative and chemical implications of the size
and composition of aerosol particles in the existing or modified global stratosphere, *Atmos.*
Chem. Phys. Discuss., 2020, 1-32, 10.5194/acp-2020-909, 2020.
- 360 O'Brien, K.: The theory of cosmic-ray and high-energy solar-particle transport in the
atmosphere, in, 29-44, 2005.
- Seinfeld, J. H., and Pandis, S. N.: *Atmospheric chemistry and physics from air pollution to
climate change*, 2nd ed., Wiley, Hoboken, N.J., 1203 S. pp., 2006.
- 365 Solomon, S., Daniel, J. S., Neely, R. R., Vernier, J.-P., Dutton, E. G., and Thomason, L. W.: The
Persistently Variable “Background” Stratospheric Aerosol Layer and Global Climate Change,
Science, 333, 866-870, 10.1126/science.1206027, 2011.
- 370 Williamson, C. J., Kupc, A., Axisa, D., Bilsback, K. R., Bui, T., Campuzano-Jost, P., Dollner, M.,
Froyd, K. D., Hodshire, A. L., Jimenez, J. L., Kodros, J. K., Luo, G., Murphy, D. M., Nault, B. A., Ray,
E. A., Weinzierl, B., Wilson, J. C., Yu, F. Q., Yu, P. F., Pierce, J. R., and Brock, C. A.: A large source
of cloud condensation nuclei from new particle formation in the tropics, *Nature*, 574, 399-+,
10.1038/s41586-019-1638-9, 2019.