

1 **Measurement report: Long emission-wavelength chromophores dominate the light absorption**
2 **of brown carbon in Aerosols over Bangkok: impact from biomass burning**

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21 **Abstract:** Chromophores represent an important portion of light-absorbing species, i.e. brown carbon.
22 Yet knowledge on what and how chromophores contribute to aerosol light absorption is still sparse. To
23 address this problem, we examined soluble independent chromophores in a set of year-round aerosol
24 samples from Bangkok. The water-soluble fluorescent chromophores identified via excitation-
25 emission matrix (EEM) spectroscopy and follow-up parallel factor analysis could be mainly assigned
26 as humic-like substances and protein-like substances, which differed in their EEM pattern from that of
27 the methanol-soluble fraction. The emission wavelength of fluorescent chromophores in
28 environmental samples tended to increase compared with that of the primary combustion emission,
29 which could be attributed to secondary formation or the aging process. Fluorescent indices inferred
30 that these light-absorbing chromophores were not significantly humified and comprised a mixture of
31 organic matter of terrestrial and microbial origin, which exhibited a different characteristic from
32 primary biomass burning and coal combustion results. A multiple linear regression analysis revealed
33 that larger fluorescent chromophores that were oxygen-rich and highly aromatic with high molecular
34 weights, were the key contributors of light absorption, preferably at longer emission wavelength ($\lambda_{\text{max}} >$
35 500 nm). Positive matrix factorization analysis further suggested that up to 50% of these responsible
36 chromophores originated from biomass burning emissions.

37

38 1. Introduction

39 Atmospheric aerosols play a substantial role in climate change through radiative forcing
40 (Alexander et al., 2008). Carbonaceous aerosols mainly include organic carbon (OC) and elemental
41 carbon (EC). Brown carbon (BrC) is a specific type of OC that absorbs radiation efficiently in the near-
42 ultraviolet and visible (UV-vis) range (Laskin et al., 2015; Kirchstetter et al., 2004) and may contribute
43 15% or more of total light absorption over the UV-vis spectrum (Kirchstetter and Thatcher, 2012; Liu
44 et al., 2013). This fraction can significantly affect atmospheric chemistry, air quality, and climate
45 change (Marrero-Ortiz et al., 2018; Laskin et al., 2015). Forest fires, residential heating by wood and
46 coal, biogenic release, and secondary formation contribute to BrC in the atmosphere (Laskin et al.,
47 2015). Many studies have indicated that the optical properties of BrC may significantly evolve as a
48 result of atmospheric processes such as oxidation (Fan et al., 2020), solar irradiation (Wong et al.,
49 2017), and relative humidity (Kasthuriarachchi et al., 2020). These factors cause variability in the
50 chemical compositions and levels of BrC across source regions and receptors, resulting in a high degree
51 of uncertainty regarding the effects of BrC (Dasari et al., 2019; Xie et al., 2019).

52 Light absorption of BrC is associated with its molecular composition and chemical structure
53 (Song et al., 2019; Lin et al., 2018; Mo et al., 2018; Jiang et al., 2020). Detailed structural
54 characterization of BrC compounds is essential to understand their sources and chemical processes in
55 the atmosphere. High-resolution mass spectrometry (HRMS) is a powerful tool for molecular-level
56 chemical analysis of organic aerosols (Laskin et al., 2018). Combinations of offline high-performance
57 liquid chromatography (HPLC), a photodiode array detector, and HRMS allow the chemical
58 characterization of aerosols specific to BrC (Lin et al., 2018; Lin et al., 2016; Lin et al., 2015; Lin et al.,
59 2017). With these combination approaches, nitroaromatics, aromatic acids, phenols, polycyclic
60 aromatic hydrocarbons and their derivatives are basically identified as BrC chromophores (Wang et
61 al., 2020b; Yan et al., 2020). However, it should be noted that it is difficult to ionize some organic
62 compounds for detection using HRMS, and even for those that can be detected, HRMS can only
63 provide possible molecular structures based on empirical deduction (Song et al., 2018; Lin et al., 2015).
64 The isomeric complexity of natural organic matter may have exceeded achievable one-dimensional
65 chromatographic resolution (Hawkes et al., 2018), and therefore, the majority of components in the
66 BrC mixture remain undetermined.

67 Excitation-emission matrix (EEM) fluorescence spectroscopy detects bulk chromophores in a
68 solution (Chen et al., 2016b). Chromophores can be revealed by EEM with information on their
69 chemical structures associated with molecular weight, aromatic rings, conjugated systems, etc (Wu et
70 al., 2003). For example, a redshift in emission spectral maxima can be caused by an increase in the

71 number of aromatic rings condensed in a straight chain, conjugated double bonds, or formational
72 changes that permit vibrational energy losses of the promoted electrons (Wu et al., 2003). A significant
73 Stokes shift with emission wavelength can be observed in aged secondary organic aerosols (SOA)
74 using EEM spectroscopy (Lee et al., 2013). Parallel factor (PARAFAC) analysis has been widely used
75 to decompose the EEM spectral signature into independent underlying components (Han et al.,
76 2020; Yue et al., 2019; Wu et al., 2019; Chen et al., 2019b), adding valuable information to absorbance-
77 based measurements (Yan and Kim, 2017). This technique helps to categorize groups of similar
78 fluorophores or chromophores or similar optical properties, thereby allowing a better understanding of
79 the chemical properties of BrC, while it should be noted that not all chromophores in BrC compounds
80 are fluorescence (Chen et al., 2019a). There is evidence that BrC absorption is closely correlated with
81 fluorescent chromophores (Huo et al., 2018). However, the intrinsic relationship between fluorescent
82 chromophores and BrC absorption has not been explored.

83 Southeast Asia is subject to intensive regional biomass burning, the emissions from which may
84 contribute to atmospheric brown clouds (Ramanathan et al., 2007; Laskin et al., 2015). The contribution
85 of biomass burning to aerosol optical depth was evaluated to be more than 56% over this region (Huang
86 et al., 2013). Despite many studies focused on the characterization of atmospheric black carbon (BC)
87 (See et al., 2006; Fujii et al., 2014; Permadi et al., 2018), studies on BrC in the region are still limited.
88 A recent study in Singapore indicated that water-soluble OC (WSOC) exhibited strong wavelength
89 dependence and even higher values of BrC absorption than those from Korea, India, China, and Nepal
90 (Adam et al., 2020), indicating abundant water-soluble BrC in the air over Southeast Asia.

91 This study was performed to explore the relationships between EEM chromophores and BrC light
92 absorption in soluble aerosol organic matter. A set of year-round aerosol samples from Bangkok,
93 Thailand, was analyzed. Water-soluble and methanol-soluble BrC in the aerosol samples were
94 characterized by EEM followed by statistical analyses to retrieve information on the contributions of
95 fluorescent chromophores to BrC light absorption, as well as their emission sources. This study
96 provides a comprehensive dataset on seasonal variability in the light absorption properties, sources,
97 and chemical components of BrC, which may be useful for improving further modeling and field
98 observation.

99 **2. Experiment**

100 **2.1. Sample Collection and Extraction.**

101 Eighty-five total suspended particulate (TSP) samples were collected on the roof (57 m above
102 ground level) of the Faculty of Environment, Kasetsart University (100°57' E and 13°85' N) in
103 Bangkok, Thailand (Fig. S1). Detailed information about the sampling site is presented elsewhere

104 (Wang et al., 2020a). Sampling was performed from January 18, 2016 to January 28, 2017, and the
105 sampling period was divided into four seasons: the pre-hot season (January 18–February 28, 2016),
106 hot season (March 2–May 30, 2016), monsoon (June 2–October 30, 2016), and cool season (November
107 1, 2016–January 28, 2017). Table S1 lists the average meteorological data in the four seasons.
108 Generally, during the sampling period, the hot season was characterized by high temperatures and wind
109 speeds, and the monsoon season by high humidity. TSP samples were collected over 24 h using a high-
110 volume ($0.3 \text{ m}^3 \text{ min}^{-1}$) sampler with quartz-fiber filters (QFFs, prebaked for 6 h at $450 \text{ }^\circ\text{C}$). All samples
111 and field blanks were stored under dark conditions at $-20 \text{ }^\circ\text{C}$ until analysis.

112 WSOC was prepared by ultrasonication extraction of filter punches with ultra-pure deionized
113 water (resistivity of $> 18.2 \text{ M}\Omega$). The methanol-soluble OC (MSOC) fraction was then obtained by
114 extracting the freeze-dried residue on the same QFFs after water extraction with HPLC-grade methanol,
115 which is used for water-insoluble fractions (Chen and Bond, 2010). It is worth noting that the MSOC
116 in this study is not necessarily like that of the same name in other studies. The extract solutions were
117 passed through $0.22\text{-}\mu\text{m}$ PTFE filters and subjected to follow-up UV-vis absorption and fluorescence
118 spectral analysis. The mass concentrations of WSOC and MSOC were measured, and the method are
119 shown in the Supplement.

120 **2.2. Absorption Spectra and Fluorescence Spectra.**

121 The extract solutions were placed in quartz cells with a path-length of 1 cm and subjected to
122 analysis using an fluorometer (Aqualog; Horiba Scientific, USA). Absorption spectra and EEM spectra
123 were obtained simultaneously using this instrument. The contribution of solvents was subtracted from
124 the extract spectra. UV-vis absorption spectra were scanned in the range of 239 to 800 nm with a step
125 size of 3 nm. The Fluorescence spectra were recorded with emission wavelength (E_m) ranging from
126 247.01 to 825.03 nm and excitation wavelength (E_x) ranging from 239 to 800 nm. The wavelength
127 increments of the scans for E_m and E_x were 4.66 and 3 nm, respectively. The calculation of optical
128 parameters and the relative contributions of BrC to total aerosol light absorption are presented in the
129 Supplement.

130 **2.3. Factor analysis**

131 In this study, we built a PARAFAC model, based on 85 TSP sample fluorescence (samples \times E_x
132 \times E_m : $85 \times 188 \times 125$, 85-model). Original EEM spectra were corrected and decomposed via
133 PARAFAC analysis with reference to earlier methods using drEEM toolbox version 2.0 with MATLAB
134 software (<http://models.life.ku.dk/drEEM>, last access: June 2014) (Murphy et al., 2013; Andersson and
135 Bro, 2000). The absorbance, all below 1 at 239 nm, was deemed suitable for correcting the EEM

136 spectra for inner filter effects (IFE) (Luciani et al., 2009;Gu and Kenny, 2009;Fu et al., 2015), and the
137 sample EEM spectra, and blanks were normalized relative to the Raman peak area of ultrapure
138 deionized water collected on the same day to correct fluorescence in Raman Units (RU) (Murphy et
139 al., 2013;Murphy et al., 2010). Spectra with $E_m > 580$ nm and $E_x < 250$ nm were removed to eliminate
140 noisy data. The non-negativity constraint is necessary to obtain reasonable spectra, and signals of first-
141 order Rayleigh, Raman, and second-order Rayleigh scattering in the EEM spectra were removed using
142 the interpolation method (Bahram et al., 2006). The two- to nine-component PARAFAC model was
143 explored, within the context of spectral loading, core consistency, and residual analysis (Figs. S2–S5).
144 Finally, seven and six components were identified in the WSOC and MSOC fractions, which explained
145 99.89% and 99.76% of the variance, respectively. Both the seven- and six-component PARAFAC
146 solutions passed the split-half analysis with the split style of “S₄C₆T₃”, and residuals were examined
147 to ensure that there was no systematic variation. The parameters obtained from the PARAFAC model
148 were used to calculate the approximate abundance of each component, expressed as F_{max} (in RU),
149 corresponding to the maximum fluorescence intensity for a particular sample.

150 Fluorescence indices based on intensity ratios that provide insight into the origins of dissolved
151 BrC, such as the humification index (HIX) (the ratio of average emission intensity in the 435–480-nm
152 range to that in the 300–345-nm range following excitation at 254 nm, which was used to reflect the
153 degree of humification) (Zsolnay et al., 1999), the biological index (BIX) (the ratio of emission
154 intensities at 380 and 430 nm following excitation at 310 nm, reflecting autochthonous biological
155 activity in water samples) (Huguet et al., 2009), and fluorescence index (FI) (the ratio of emission
156 intensities at 470 and 520 nm following excitation at 370 nm, reflecting the possibility of microbial
157 origin and for examining differences in precursor organic materials) (Lee et al., 2013;Murphy et al.,
158 2018).

159 **2.4. Statistical analysis**

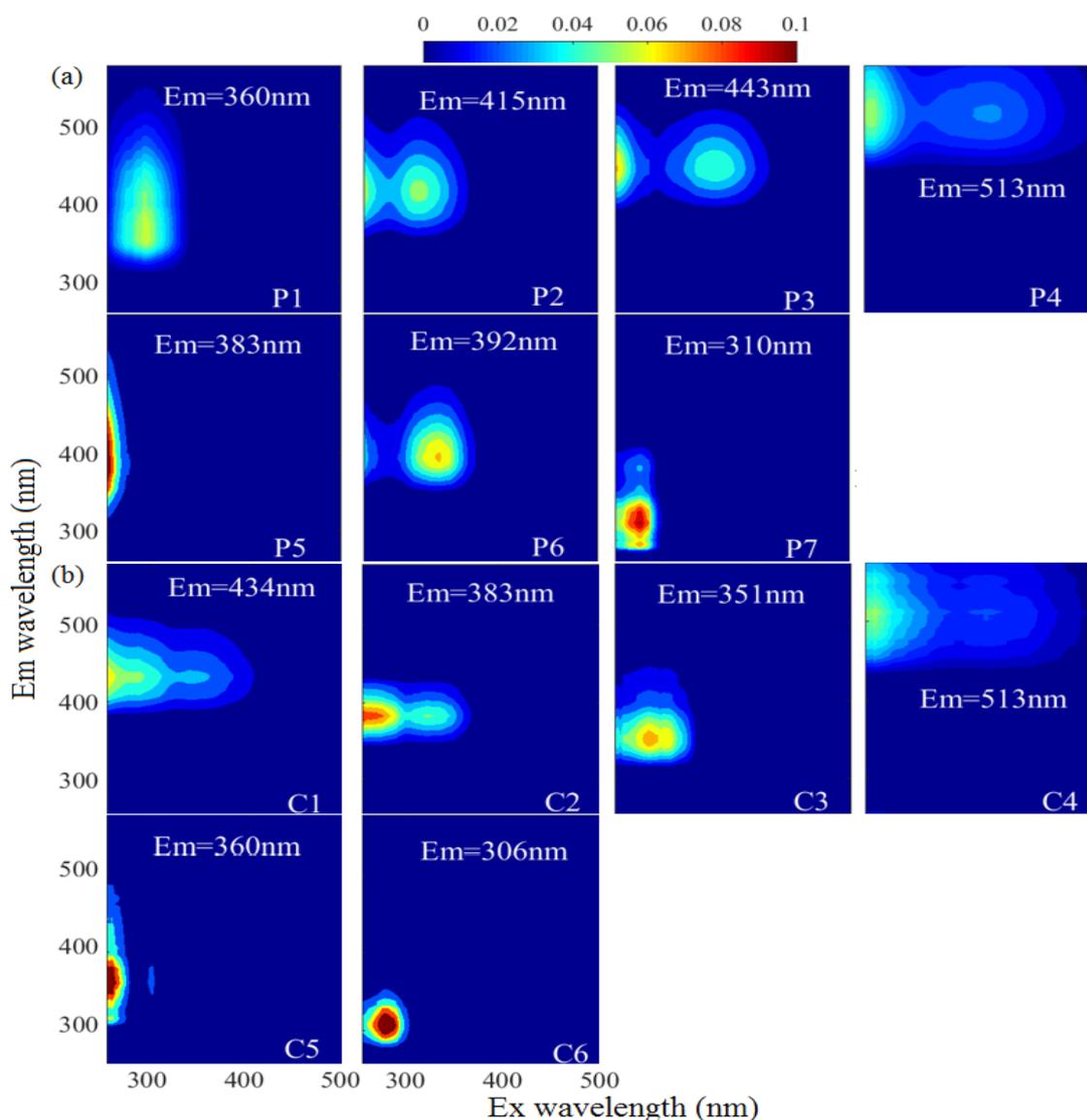
160 A hierarchical cluster method was used to classify aerosol samples based on the relative
161 contributions of PARAFAC components to the respective samples. The Squared Euclidean distance
162 method was used to evaluate the distances between samples, and the Between-group linkage method
163 was chosen for hierarchical cluster analysis. The multiple linear regression (MLR) model was applied
164 to elucidate the relationship between fluorescent chromophores and light absorption of BrC using a
165 stepwise screening process. Analyses were performed using SPSS software (SPSS Inc., Chicago, IL,
166 USA).

167 **3. Results and Discussion**

168 3.1. EEM of dissolved organic substances.

169 Fluorescence spectra coupled with PARAFAC results can provide more information about the
170 chemical structures of chromophores. Figure 1 and Table S2 show the seven-component (P1–7)
171 PARAFAC solutions of WSOC in the samples of aerosol over Bangkok, the peaks of which fell mainly
172 into the humic-like and protein-like chromophore regions in the plots. Components P2, P3, P4, and P6
173 were identified as humic-like substances (HULIS) (Chen et al., 2017a; Stedmon and Markager,
174 2005; Wu et al., 2019; Chen et al., 2003). A second peak was observed at a high excitation wavelength
175 for these components, indicating the existence of a large number of condensed aromatic moieties,
176 conjugated bonds, and nonlinear ring systems (Matos et al., 2015). Among them, P2, P3, and P4 had a
177 longer emission wavelength (> 400 nm) than P6, likely due to the low probability of fluorescence
178 emission from quinonoid $n-\pi^*$ transitions (Cory and McKnight, 2005). P3 produced similar spectra to
179 those of aqueous reaction products of hydroxyacetone with glycine (Gao and Zhang, 2018), and
180 dissolved organic matter (DOM) in the surface water of Xiangxi Bay and Three Gorges Reservoir
181 (Wang et al., 2019). P6 had a peak similar to those in the fluorescence spectra of N-containing SOA
182 species formed by α -pinene under ozonolysis and photooxidation with NH_3 in a flow reactor (Babar et
183 al., 2017) as well as pyridoxine (Pohlker et al., 2012), indicating a possible biological source. P5 was
184 similar to a previously identified fluorophore in $\text{PM}_{2.5}$ from Xi'an (Chen et al., 2019b). P1 and P7
185 could be assigned as protein-like organic matter (PLOM) due to their short emission wavelengths (Wu
186 et al., 2003). Specifically, P7 resembled a tyrosine-like fluorophore (Zhou et al., 2019; Chen et al., 2003)
187 and may be related to non-N-containing species (Chen et al., 2016b).

188 The MSOC fraction extracted from the filter residue after water extraction produced fluorescence
189 signals with fluorescence patterns different from those of the WSOC fraction, indicating a different
190 chemical composition from that of WSOC. Thus, WSOC with the addition of MSOC may provide a
191 more comprehensive description of the optical and chemical characteristics of BrC compared to
192 WSOC alone. Six components (C1–C6) were resolved for the MSOC. Among them, C1 and C2 were
193 associated with shorter excitation wavelengths (< 250 nm) but longer emission wavelengths (> 380
194 nm), indicating the presence of fulvic-like substances (Chen et al., 2003; Mounier et al., 1999). C6
195 produced a pattern similar to that of tyrosine-like fluorescence (Stedmon and Markager, 2005).
196 Although C4 had a similar EEM spectrum as P4 of WSOC, the two components were chemically
197 different in polarity, suggesting different behaviors in the environment (Ishii and Boyer, 2012). Note
198 that there were no special chemical structures for the different types of chromophores, and therefore,
199 the origins and chemical structures of HULIS and PLOM studied here are not necessarily like those
200 with the same names in other types of organic matter.



201

202 **Figure 1.** The fluorescent components identified by the PARAFAC (parallel factor) analysis for EEM of water-
 203 soluble organic carbon (P1–P7, WSOC, a) and methanol-soluble organic carbon (C1–C6, MSOC, b) in the aerosol
 204 samples over Bangkok in Thailand (n=85). The color represents that the intensity was normalized to set the maximum
 205 as 0.1.

206 To further explore the potential sources of the EEM-PARAFAC components, we added 60 source
 207 samples to the matrices. The source sample EEM data were described in our previous study (Tang et
 208 al., 2020b), including those of 33 biomass-burning samples (IDs: 1–33), 17 coal-combustion samples
 209 (IDs: 34–50) samples, eight tunnel samples (IDs: 51–58) and two vehicle-exhaust samples from trucks
 210 (IDs: 59–60), which are important sources of BrC in the atmosphere. This, in combination with our
 211 Bangkok field samples, yielded a new matrix ($145 \times 188 \times 125$, 145-model) for modeling. PARAFAC
 212 analysis successfully decomposed the dataset, and the output was the same as for the 85-model. The
 213 component solutions are presented in Fig. S6. To validate the stability of the model after loading by
 214 the new matrix, the Tucker congruence coefficient (TCC) was calculated to determine the similarity

215 of two fluorescence spectra between the two models (refer to Text S3 of Supplement). Note that a
216 higher TCC value would indicate a higher degree of similarity of the spectra. As shown in Table S2
217 and Fig. S7, high TCC values were found as expected between the 85-model components and the 145-
218 model components, indicating that the two models identified similar fluorescent chromophores. It
219 should be noted that one additional fluorescent component was identified each for the WSOC and
220 MSOC fractions in the new 145-model, respectively, but these components were only highly
221 characterized by source emission samples, as reported in our previous study (Tang et al., 2020b).

222 Using the distribution proportions of the EEM-PARAFAC fitted components (145-model), we
223 conducted hierarchical cluster analysis of the mixed ambient and source samples. The results are
224 shown in Figs. S9 and S10. For the WSOC fraction, all aerosol samples from Bangkok and tunnel
225 samples were assigned to cluster A, whereas biomass-burning and coal-combustion aerosols were
226 assigned to clusters C and D, respectively. This implied that the fluorescent chromophore types could
227 be somewhat related to the emission precursors of the aerosol components. However, the distribution
228 of fluorescent chromophores varied clearly between the ambient aerosols and source samples. The
229 ambient aerosol samples contained higher levels of fluorescent chromophores with longer emission
230 wavelengths that were related to humic-like or fulvic-like chromophores (components 145M-P1 (P1
231 component in 145-model), 145M-P5, and 145M-P6), whereas the primary biomass-burning and coal-
232 combustion samples contained high-intensity fluorescent chromophores with shorter emission
233 wavelengths that were related to protein-like fluorescence (145M-P2 and 145M-P4). These
234 phenomena was similarly reported previously, i.e., protein-like substances produce compounds with
235 similar fluorescence properties as humic substances under irradiation conditions (Bianco et al., 2014).
236 Similar differences between field samples and source samples were found for the MSOC fraction.
237 Therefore, our results confirmed that chemical reactions or “aging” in the atmosphere greatly modifies
238 the chromophore patterns of emission sources by both bleaching the source chromophores or
239 producing new chromophores and, at least in this case, shifts the chromophore emission wavelength
240 toward longer wavelengths, i.e., from protein-like to fulvic-like (Bianco et al., 2014; Bianco et al.,
241 2016; Lee et al., 2013).

242 **3.2. Fluorescence-derived indices**

243 The ratios of fluorescence intensity from specific spectral regions of an EEM were used as
244 indicators for the relative contributions of organic matter derived from terrestrial or microbial sources
245 in natural waters (Shimabuku et al., 2017; Birdwell and Engel, 2010; Mcknight et al., 2001). HIX was
246 initially introduced to estimate the degree of maturation of DOM in soil (Zsolnay et al., 1999),
247 representing the degree of humification of organic matter, for which higher HIX values also indicate

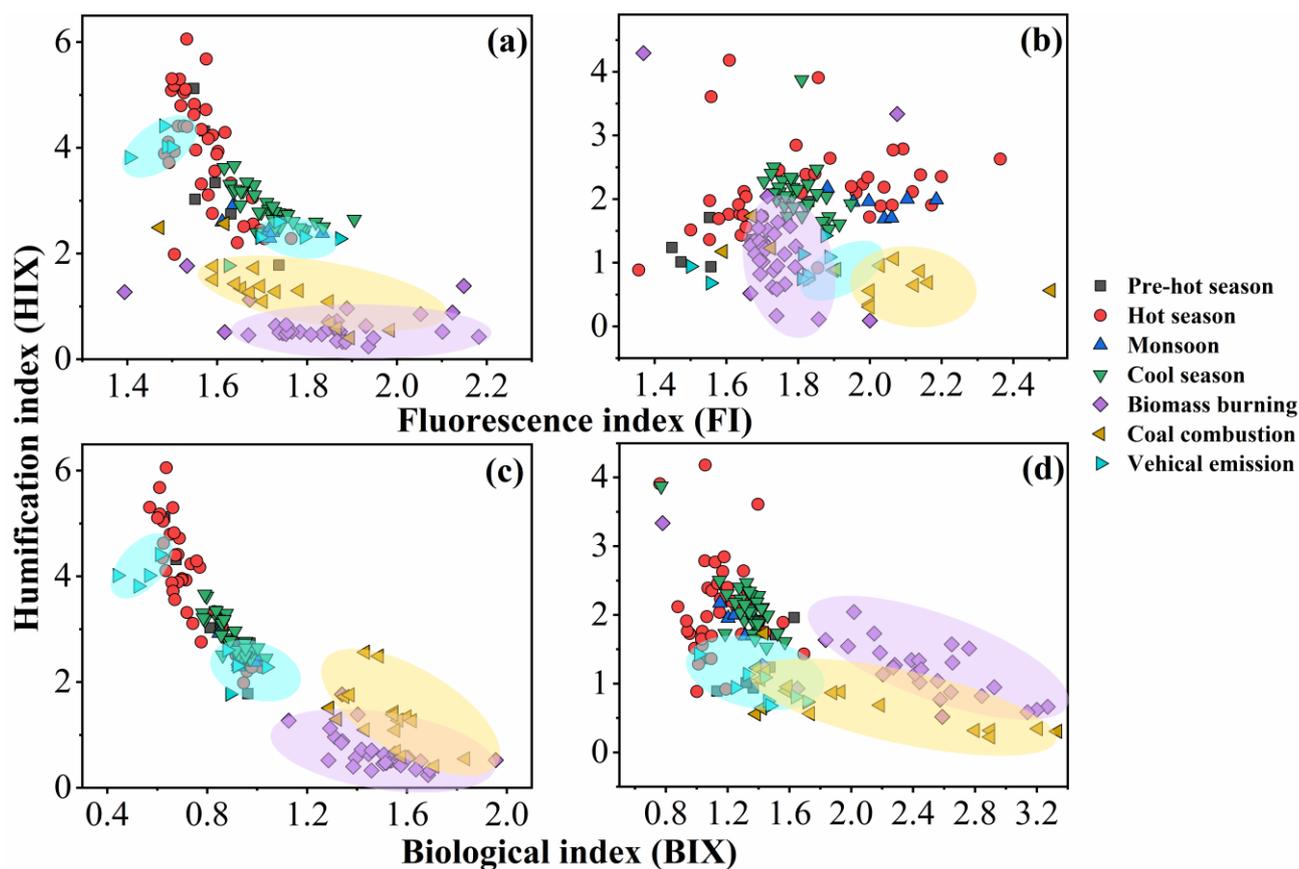
248 higher degree of polycondensation (low H/C ratio) and aromaticity (Qin et al., 2018). Generally, high
 249 HIX values (> 10) correspond to strongly humified or aromatic organics, principally of terrestrial
 250 origin, whereas low values (< 4) are indicative of autochthonous or microbial origin. As shown in
 251 Table 1 and Fig. 2, the HIX values were 3.4 ± 0.99 and 2.0 ± 0.59 for WSOC and MSOC, respectively,
 252 in aerosol samples from Bangkok. All HIX values were less than 10, which could be viewed as a
 253 nominal cutoff below which DOM is not significantly humified (Birdwell and Valsaraj, 2010; Zsolnay
 254 et al., 1999; Huguet et al., 2009). Figure 2 shows the HIX values in primary biomass-burning and coal-
 255 combustion samples, which were much lower than those in the ambient samples, indicating that the
 256 lower values of HIX in the atmosphere likely correspond to freshly introduced material. Lee et al.
 257 (2013) reported that fresh SOA had low HIX values, but these values increased significantly upon
 258 aging with ammonia. The much higher HIX values in the WSOC compared to the MSOC suggest that
 259 WSOC may have a higher degree of aromaticity or a more condensed chemical structure. Our previous
 260 study revealed that MSOC has a higher molecular weight but lower aromaticity index than the
 261 corresponding WSOC in combustion experiment aerosol samples, indicating a more aliphatic structure
 262 in the MSOC (Tang et al., 2020b). The HIX values of WSOC were highest in the hot season (3.9 ± 1.1),
 263 followed by the pre-hot season (3.3 ± 1.1), cool season (2.9 ± 0.36), and monsoon (2.5 ± 0.22), whereas
 264 those of the MSOC tended to be higher in the hot and cool seasons than in the monsoon and pre-hot
 265 seasons. The HIX values in the WSOC fraction were comparable to those of water-soluble organic
 266 aerosols in the high Arctic atmosphere (mean: 2.9) (Fu et al., 2015) and higher than those of water-
 267 soluble aerosols (1.2 ± 0.1 in winter and 2.0 ± 0.3 in summer) over northwest China (Qin et al., 2018),
 268 likely indicating a higher degree of chromophore humification.

269 **Table 1** Seasonal averages of the concentration of organic carbon (OC), elemental carbon (EC), water-soluble organic
 270 carbon (WSOC), and methanol-soluble organic carbon (MSOC), BrC absorption, fluorescence indices and
 271 levoglucosan level for aerosol samples collected from Bangkok in Thailand. Pre-hot season is from January 18 to
 272 February 29, 2016; hot season is from March 2 to May 31, 2016; monsoon is from June 2 to October 30, 2016; cool
 273 season is from November 1, 2016 to January 28, 2017.

	Annual (n=85) Ave \pm sd	Pre-Hot season (n=7) Ave \pm sd	Hot season (n=41) Ave \pm sd	Monsoon (n=7) Ave \pm sd	Cool season (n=30) Ave \pm sd
^a OC ($\mu\text{g C m}^{-3}$)	12 \pm 7.3	19 \pm 9.3	9.6 \pm 6.7	6.5 \pm 0.97	16 \pm 5.6
^a EC ($\mu\text{g C m}^{-3}$)	1.4 \pm 0.48	2.0 \pm 0.45	1.2 \pm 0.47	1.2 \pm 0.15	1.5 \pm 0.40
^a OC/EC	8.9 \pm 5.2	9.6 \pm 3.4	8.4 \pm 6.8	5.4 \pm 0.51	10 \pm 2.5
			WSOC		
$\mu\text{g C m}^{-3}$	6.2 \pm 4.2	9.9 \pm 5.7	5.3 \pm 4.1	2.6 \pm 0.31	7.4 \pm 3.4

AAE (330–400 nm)	5.1±0.68	5.0±0.52	5.4±0.56	6.2±0.11	4.5±0.34
Abs ₃₆₅ (Mm ⁻¹)	5.6±4.9	10±7.4	4.5±4.5	1.2±0.21	7.2±4.1
MAE ₃₆₅ (m ² g ⁻¹ C)	0.83±0.25	0.96±0.19	0.78±0.23	0.45±0.06	0.95±0.21
FI	1.6±0.10	1.6±0.09	1.6±0.08	1.7±0.07	1.7±0.07
BIX	0.82±0.13	0.83±0.14	0.74±0.13	0.92±0.05	0.89±0.07
HIX	3.4±0.99	3.3±1.1	3.9±1.1	2.5±0.22	2.9±0.36
			MSOC		
µg C m ⁻³	6.0±3.4	9.2±4.0	4.3±2.9	3.9±0.86	8.1±2.6
AAE (330–400 nm)	5.2±0.94	4.9±0.69	5.5±1.1	5.1±0.15	4.7±0.55
Abs ₃₆₅ (Mm ⁻¹)	1.7±1.4	1.9±1.6	1.0±0.99	0.72±0.23	2.7±1.4
MAE ₃₆₅ (m ² g ⁻¹ C)	0.26±0.12	0.19±0.08	0.23±0.11	0.19±0.06	0.33±0.11
FI	1.8±0.20	1.5±0.20	1.8±0.23	2.0±0.10	1.8±0.06
BIX	1.2±0.18	1.4±0.20	1.2±0.19	1.3±0.09	1.3±0.14
HIX	2.0±0.59	1.3±0.41	2.1±0.68	1.9±0.17	2.1±0.42
^a Levogluconan (ng C m ⁻³)	222±485	362±438	185±654	42±16	280±185
^a Levogluconan/TSP (×10 ⁻³)	2.9±2.9	3.4±3.1	2.3±3.6	1.9±0.98	3.9±1.8

a: described elsewhere (Wang et al., 2020a).



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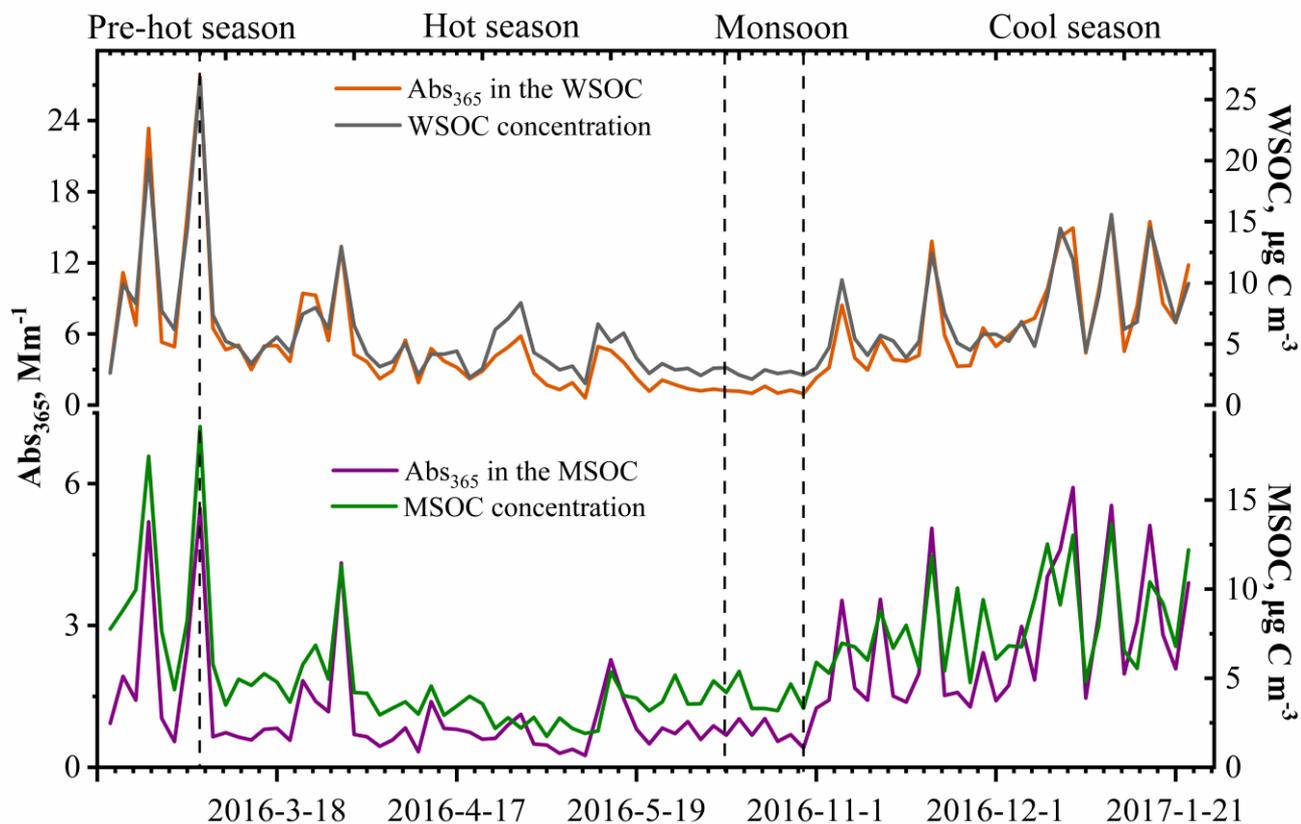
276 **Figure 2.** Fluorescence index (FI), biological index (BIX), and humification index (HIX) of water-soluble organic
 277 carbon (WSOC, a, c) and methanol-soluble organic carbon (MSOC, b, d) in aerosol samples from Bangkok, Thailand,
 278 as well as source emission samples including biomass burning, coal combustion and vehicle emission which were
 279 encircled by a violet, yellow, and blue region, respectively. Note that the fluorescence characteristic of source samples
 280 was described elsewhere (Tang et al., 2020b), but the fluorescence indices was first reported in this study. Pre-hot
 281 season is from January 18 to February 29, 2016; hot season is from March 2 to May 31, 2016; monsoon is from June
 282 2 to October 30, 2016; cool season is from November 1, 2016 to January 28, 2017.

283 The BIX and FI were previously proposed as proxies for the contribution of biogenic organic
 284 matter and autochthonous biological activity in natural water, respectively (Fu et al., 2015; Qin et al.,
 285 2018). For example, the FI decreased by up to 20% indicating that the samples appeared increasingly
 286 like “terrestrial” DOM, whereas the BIX increased by up to 37% indicating that the samples became
 287 more “autochthonous” in character (Murphy et al., 2018; Gabor et al., 2014). FI values ≤ 1.4 correspond
 288 to terrestrially derived organics and higher aromaticity, whereas values ≥ 1.9 correspond to microbial
 289 sources and a lower aromatic carbon content (Mcknight et al., 2001). An increase in BIX is related to
 290 an increase in the contribution of microbially derived organics, with high values (> 1) shown to
 291 correspond to a predominantly biological or microbial origin of DOM and the presence of organic
 292 matter freshly released into water, whereas values ≤ 0.6 indicate the presence of little biological
 293 material (Huguet et al., 2009).

294 The FI and BIX values of the Bangkok aerosol samples are summarized in Table 1 and Fig. 2.
295 The FI values of the WSOC and MSOC were 1.6 ± 0.10 and 1.8 ± 0.20 , respectively, suggesting that
296 these chromophores are representative of both terrestrially and microbially derived organic matter. The
297 BIX values of the WSOC and MSOC were 0.82 ± 0.13 and 1.2 ± 0.18 , respectively. Almost all BIX
298 values were greater than 0.6 in the two fractions, suggesting biological or microbial contribution. Lee
299 et al. (2013) reported that the BIX values of SOA samples averaged 0.6 and increased upon aging. In
300 addition, the results of our source samples showed that primary biomass-burning and coal-combustion
301 samples had high FI and BIX values (Fig. 2). These results indicate that these chromophores in
302 Bangkok were likely freshly introduced or derived from biomass burning and coal combustion. Further,
303 an increase in BIX in the MSOC in comparison with the WSOC was observed in primary biomass-
304 burning and coal-combustion samples, consistent with the Bangkok samples. The BIX values were
305 similar to those in the WSOC in Arctic aerosols (0.6–0.96, mean: 0.72), which were within the extreme
306 values for the predominance of humic- or protein-like fluorophores (Fu et al., 2015). BIX values
307 exhibited the opposite trend from HIX values, with low BIX values in the hot season. This may be
308 explained by a previous study showing that a high BIX appears to indicate little humification (Birdwell
309 and Engel, 2010). It should be noted that the fluorescence indices (FI, BIX, and HIX) were first applied
310 for aquatic and soil organic compounds and further extended to the atmosphere due to the similarities
311 in the properties of organic matter (Graber and Rudich, 2006). However, the values observed for
312 primary biomass burning and coal combustion in this study differ from with the previously established
313 fluorescence standards for aquatic environments and soil. Therefore, caution is required when using
314 these indices to appoint source of atmospheric chromophores (Wu et al., 2021).

315 **3.3. Optical properties of dissolved BrC.**

316 Figure 3 shows the variations in soluble OC concentrations and the corresponding light absorption
317 coefficient at 365 nm (Abs_{365}). In general, the Abs_{365} closely tracked the variations in the mass
318 concentrations of WSOC and MSOC ($p < 0.000$, $R^2 = 0.95$ and $p < 0.000$, $R^2 = 0.75$, respectively) (Fig.
319 S11), indicating that the portions of BrC in both fractions were considerably stable. Furthermore, light
320 absorption at 365nm were higher in the pre-hot season, hot season, and cool season than that in the
321 monsoon season. According to the levoglucosan level that generally regarded as biomass burning
322 tracers and the ratios of levoglucosan/TSP (Table 1), we infer that the non-monsoon season were more
323 affected by biomass burning and also showed high absorption.



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325 **Figure 3.** Time series plots of water-soluble organic carbon (WSOC) and methanol-soluble organic carbon (MSOC)
 326 concentration ($\mu\text{g C m}^{-3}$) and water- and methanol-extract light absorption coefficient at 365 nm (Abs_{365}) (Mm^{-1}) in
 327 the aerosol samples from Bangkok, Thailand during 2016–2017.

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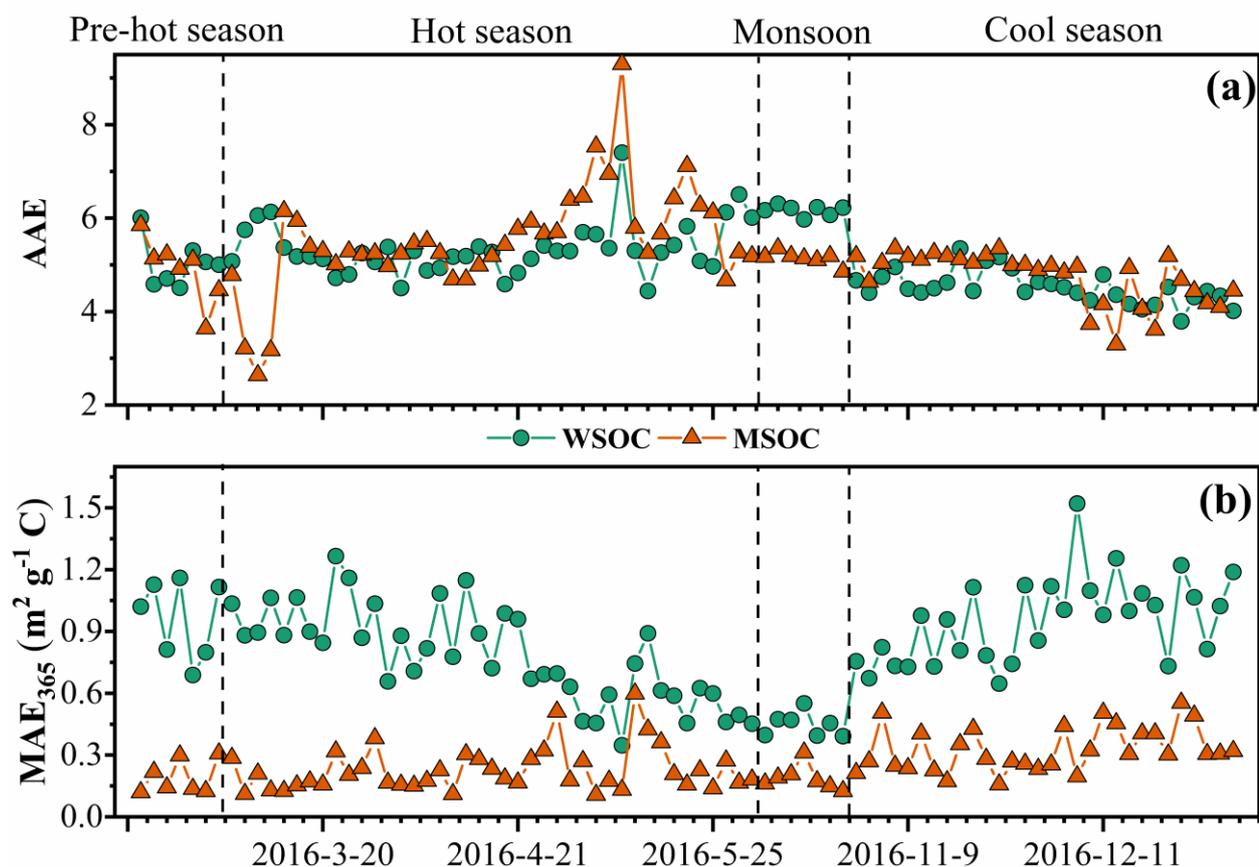
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The absorption Ångström exponent (AAE) and mass absorption efficiency (MAE) are important optical parameters reflecting the spectral dependence and light absorption ability of BrC, respectively. The magnitude of the AAE reflects the differences in BrC source and atmospheric processes (Lack et al., 2013). Typically, the AAE value is close to 1 when light absorption is dominated by soot (Kirchstetter et al., 2004), roughly 1–3 for simulated biomass-burning aerosols (Hopkins et al., 2007), and up to 6–7 for water-soluble HULIS in biomass burning-impacted aerosols (Hoffer et al., 2006). The AAE values of the WSOC and MSOC between 330 and 400 nm in this study were up to 5.1 ± 0.68 and 5.2 ± 0.94 (Fig. 4), respectively, indicating strong wavelength dependence in the light absorption capability. These high values show that BrC tends to absorb more solar irradiation over ultraviolet wavelengths, which is comparable to BC absorption as shown in Fig. S12. These observations indicate that BrC has important impacts on photochemical reactions in the atmosphere (Barnard et al., 2008). The AAE values in this study are similar to those of water-soluble BrC over biomass burning-impacted regions, such as Beijing (Mo et al., 2018; Yan et al., 2015) and Guangzhou (Liu et al., 2018), but lower than those of aerosols from simulated biomass-burning and coal-combustion experiments (Fan et al., 2018; Tang et al., 2020a; Li et al., 2018). However, it should be noted that the BrC AAE varies in the

343 atmosphere. Dasari et al. (2019) reported that AAE values of water-soluble BrC increase continuously
 344 due to photolysis of chromophores and atmospheric oxidation during long-range transport over the
 345 Indo-Gangetic Plain (IGP). In addition, pH changes can cause the absorption spectra of some BrC
 346 species to shift to longer wavelengths upon deprotonation, decreasing AAE values (Mo et al., 2017).
 347 The pH values of WSOC fraction for all the samples were within the range of 5–7, generally thinking
 348 it didn't affect the absorbance according to a prior study (Chen et al., 2016a).



349
 350 **Figure 4.** Time series plots of Absorption Ångström exponent (AAE, a), the mass absorption efficiency at 365 nm
 351 (MAE_{365} , b) in the water-soluble organic carbon (WSOC) and methanol-soluble organic carbon (MSOC) in aerosols
 352 samples from Bangkok in Thailand during 2016–2017.

353 The MAE at 365 nm (MAE_{365}) of the WSOC was $0.83 \pm 0.25 \text{ m}^2 \text{ g}^{-1} \text{ C}$, which was higher than that
 354 of the MSOC ($0.26 \pm 0.12 \text{ m}^2 \text{ g}^{-1} \text{ C}$), indicating that more water-soluble BrC with stronger light
 355 absorption capability could be extracted with ultrapure deionized water, whereas water-insoluble BrC
 356 is characterized by lower light absorption capability over Bangkok. These results were consistent with
 357 those from vehicular exhaust samples in our previous study, where MAE_{365} values of the WSOC
 358 ($0.71 \pm 0.30 \text{ m}^2 \text{ g}^{-1} \text{ C}$) were higher than those of the MSOC ($0.26 \pm 0.09 \text{ m}^2 \text{ g}^{-1} \text{ C}$) (Tang et al., 2020b).
 359 Opposite results have been shown for primary biomass burning and coal combustion (Tang et al.,
 360 2020b). Similarly, Bikkina et al. (2020) observed that the marine-impacted aerosols of the Bay of

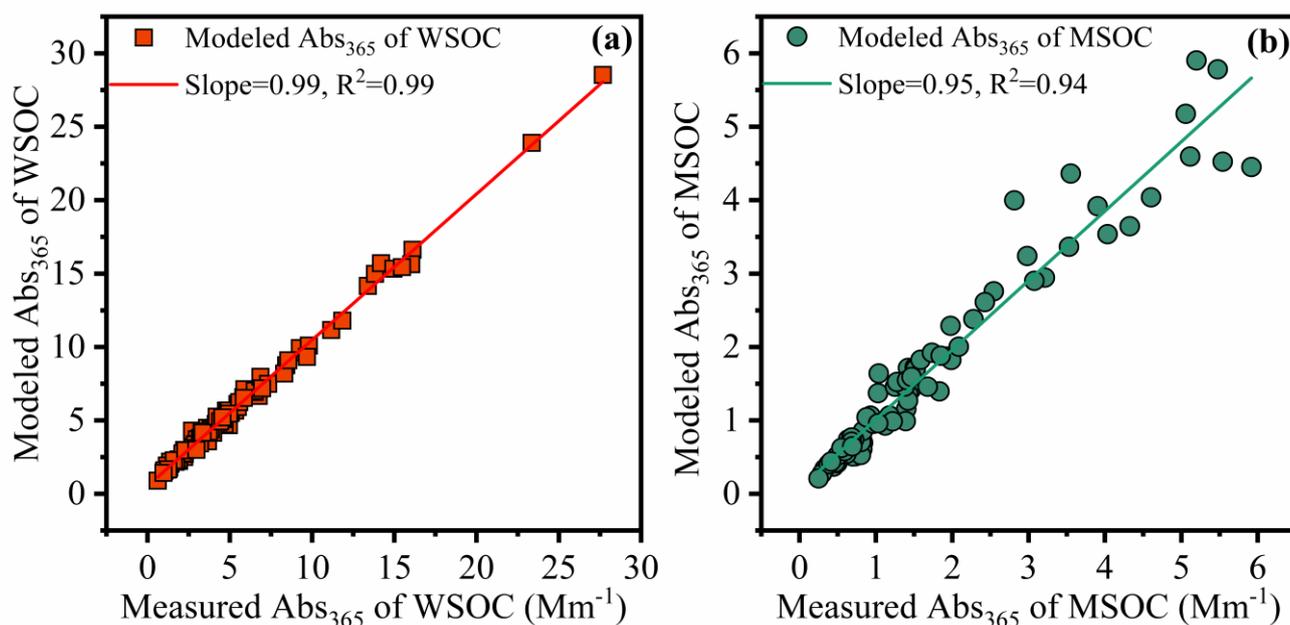
361 Bengal showed higher MAE₃₆₅ values in the WSOC fraction than MSOC fraction (only extract using
362 methanol), and they explained it due to two plausible reasons. First, the BrC aerosols over Bay of
363 Bengal have a contribution from a different source (i.e., maritime influence) and contain BrC-
364 chromophores that are more soluble in water than methanol. Secondary, there could be significant
365 photobleaching effects of different chromophores. However, Wu et al. (2020b) reported that the
366 MAE₃₆₅ values of methanol-extracts are higher than those of WSOC in winter, whereas the situation
367 is reversed in summer. Therefore, we infer that the different sources and atmospheric processes would
368 impact the distribution of water-soluble and methanol-soluble chromophores. Considering the high
369 temperature and humidity (Table S1), and tropical monsoon climate in Thailand, it seems to promote
370 more water-soluble chromophores over Thailand. As not all water-insoluble components can be
371 extracted with methanol, the observed light absorption by MSOC would therefore likely reflect the
372 lower limit. Table S3 shows a comparison of the MAE values of Bangkok aerosols with those of other
373 regions, indicating a medium light absorption capacity. The MAE₃₆₅ values of the water-soluble
374 fraction in this study were comparable to those of Nanjing (Chen et al., 2018), Guangzhou (Liu et al.,
375 2018), and Beijing in summer (Yan et al., 2015), but lower than those of PM_{2.5} from Singapore (Adam
376 et al., 2020), PM₁₀ from Godavari, Nepal, in the pre-monsoon season (Wu et al., 2019), and smoke
377 particles from biomass burning and coal combustion (Park and Yu, 2016; Fan et al., 2018; Tang et al.,
378 2020b). Lower MAE₃₆₅ values of both fractions were observed in the monsoon season than in the non-
379 monsoon seasons, likely due to the heavy monsoon rains that effectively remove soluble gases and
380 aerosols (Lawrence and Lelieveld, 2010) and/or reduce biomass-burning activity (levoglucosan level
381 in Table 1). A previous study reported similar findings in the USA in that the MAE₃₆₅ was
382 approximately three-fold higher in biomass burning-impacted samples than in non-biomass burning-
383 impacted samples (Hecobian et al., 2010). Another study in the central Tibetan Plateau highlighted that
384 BrC emitted by biomass burning has stronger light absorption capability than does secondary BrC
385 formed in the atmosphere (Wu et al., 2018). On the Indo-China peninsula, Bangkok receives 99% of
386 the fire-derived aerosols from December to April (Lee et al., 2017), which may explain the high
387 absorption levels in the non-monsoon seasons.

388 **3.4. Chromophores responsible for BrC light absorption.**

389 EEM analysis enables the probing of the chemical structure of DOM because of its ability to
390 distinguish among different classes of organic matter (Wu et al., 2003). Generally, BrC absorption is
391 related to the chromophores within it and is susceptible to change with variations in chemical
392 properties, e.g., oxidation level (Mo et al., 2018), degree of unsaturation (Jiang et al., 2020), molecular
393 weight (Tang et al., 2020b; Di Lorenzo et al., 2017), functional groups (Chen et al., 2017b), molecular

394 composition, etc (Song et al., 2019; Lin et al., 2018). The fluorescence intensity of each EEM
395 component was shown to be associated with light absorption indices, such as MAE₃₆₅ and AAE, of
396 HULIS in controlled crop straw-combustion experiments (Huo et al., 2018). As a linear relationship
397 between organic matter concentration and fluorescence intensity can be assumed for very dilute
398 samples due to the IFE (Murphy et al., 2013), we have corrected our fluorescence data for IFE using
399 absorbance to enable “clean ” correlation analysis (as shown in Fig. S13 a, b). The linear regression
400 slopes in the scatter plots of Abs₃₆₅ versus WSOC or MSOC could mathematically represent the
401 average MAE values of WSOC or MSOC at 365 nm, respectively (Fig. S11 a, b). The phenomenon
402 indicates that both fluorescence and Abs₃₆₅ data point to similar relationships between sources or
403 chemical processes with organic matter concentrations, and therefore, we attempted to link the
404 fluorescence results to BrC absorption. It should be noted that light-absorbing substances in
405 atmospheric particulate matter are not necessarily all fluorescent, such as nitrophenol compounds,
406 which are a type of BrC commonly found in the atmospheric particulate matter; however, there is no
407 strong fluorescence signal with which to scan the nitrophenol standards (Chen et al., 2019a).

408 In order to evaluate the light absorption from different fluorescent chromophores, we used MLR
409 to explore the relationship between the fluorescence intensities of chromophores and Abs₃₆₅. In this
410 study, light absorption properties were treated as the dependent variables, and the fluorescence were
411 independent variables. During MLR, insignificant fluorescent components were excluded from the
412 regression using a stepwise screening process to avoid overfitting ($F_{inclusion}: p < 0.05$; $F_{elimination}: p >$
413 0.10). The MLR statistical metrics are listed in Tables S4 and S5. For the independent variables with
414 significant correlations with the dependent variable ($p < 0.05$), or with positive contributions to the
415 independence, Abs₃₆₅, they will be retained in the statistical model as the efficiency factors to Abs₃₆₅.
416 Thus, for the WSOC fraction, a revised model (regression 3) equation was used with an adjusted R²
417 of 0.995. The final optimized equations were $Abs_{365} = 0.765 \times P4 + 0.051 \times P2 + 0.091 \times P7$, for the
418 WSOC fraction, and $Abs_{365} = 0.238 \times C4$ for the MSOC fraction (Table S5). The model errors for
419 water-soluble and methanol-soluble Abs₃₆₅ were -5.5% – 64% and -34% – 58% , respectively. The
420 predicted Abs₃₆₅ values fit the measured values well (Fig. 5, slope = 0.99 and 0.95, and R² = 0.99 and
421 0.94 for WSOC and MSOC, respectively).



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Figure 5. Linear correlation analysis between modeling Abs₃₆₅ using multiple linear regression (MLR) analysis and measured Abs₃₆₅ in the water-soluble organic carbon (WSOC, a) and methanol-soluble organic carbon (MSOC, b) in aerosols samples from Bangkok in Thailand during 2016–2017, respectively. Note that the fluorescent intensities of parallel factor (PARAFAC) model results (fluorescent components) were used as variables in MLR analysis.

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For water-soluble BrC, the P4 component had the largest coefficient with Abs₃₆₅, which was much higher than those for P2 and P7. The C4 component had the largest coefficient with Abs₃₆₅ for methanol-soluble BrC. These results indicate that the light absorption by BrC is more dependent on chromophores with longer emission wavelengths (P4 and C4). These characteristics also indicate that the strongly absorbing substances in BrC probably originate from large conjugated electron functional groups or include donor and acceptor molecules for charge-transfer interactions (Del Vecchio and Blough, 2004; Cory and McKnight, 2005). Kellerman et al. (2015) reported that these components are highly aromatic and oxygen-rich with high apparent molecular weight. These important findings highlight that larger chromophores may be the most persistent BrC species in the atmosphere and hence exert the greatest influence for perturbing the global radiative balance.

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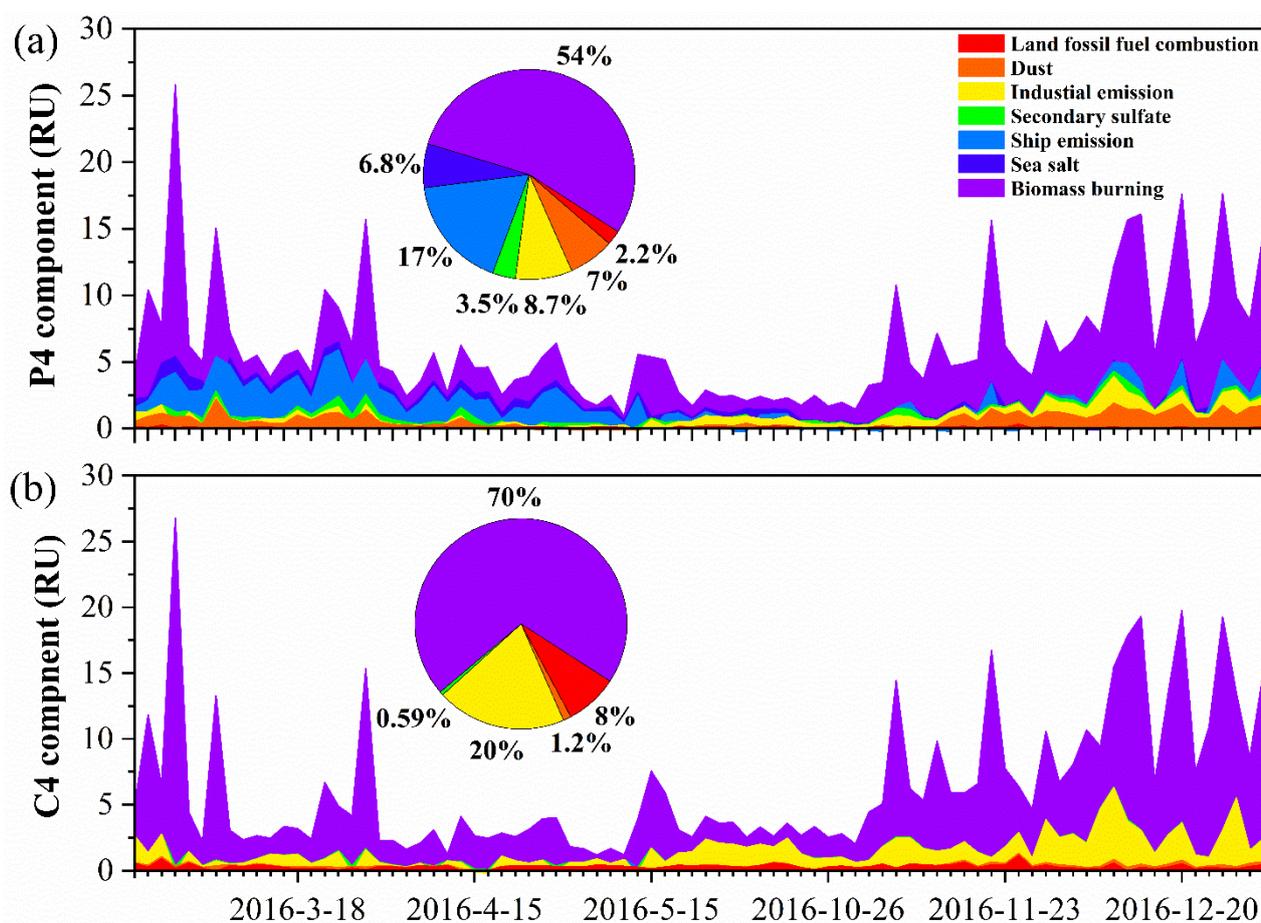
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To further interpret the BrC source profiles as real-world TSP sources, we examined 84 (minus one missing value) TSP samples from Bangkok using the US EPA PMF5.0 model. All samples were merged together to form an 84 × 30 dataset (84 samples with 30 species). The initial data of positive matrix factorization input were from our previous study (Wang et al., 2020a). We further added Abs₃₆₅ values of WSOC and MSOC, and the fluorescence intensities (in RU) of P2, P4, P7, and C4 components to the model. A seven-factor solution was achieved that provided the most physically reasonable source profiles (Fig. S14), including ship emission, secondary sulfate, dust, land fossil-fuel combustion, sea salt, biomass burning, and industrial emission, consistent with our previous study

445 (Wang et al., 2020a). Figure S15 shows the contributions of the above sources to light absorption at
 446 $\lambda = 365$ nm, which represent the fraction of BrC for each factor. Biomass burning was found to be the
 447 main source of BrC over Bangkok; 58% and 74% for water-soluble and methanol-soluble BrC,
 448 respectively. These were comparable to previous observations using a similar approach in Xi'an (55%)
 449 (Wu et al., 2020a). The time-series of Abs₃₆₅ of WSOC and MSOC contributed by factors shows the
 450 high biomass burning contribution is related to the higher local fire spots (i.e., pre-hot season, hot
 451 season, and cool season) and/or air mass from the continent (Fig. S16–S17). Jiang et al. (2021)
 452 observed increases in biomass burning contributions to BrC absorption during the winter period that
 453 was dominant in continental-origin air masses. Furthermore, the P4 and C4 components, which were
 454 more closely associated with Abs₃₆₅, could be mostly attributed to biomass burning (54% and 70%,
 455 respectively) as shown in Fig. 6. Our previous study showed that biomass burning accounted for a
 456 considerably large portion (mean: 26%) of the TSP mass concentration in the same samples (Wang et
 457 al., 2020a). This result suggests that biomass burning makes a significant contribution to not only
 458 particulate matter but also BrC light absorption.



459

460 **Figure 6.** The time-series of P4 component of the WSOC (a) and C4 of the MSOC (b) in TSP samples over Bangkok
461 in Thailand contributed by each factor resolved by positive matrix factorization.

462 **4. Conclusions**

463 This study presents a comprehensive analysis of water- and methanol-soluble chromophores in
464 aerosol samples over Bangkok in Thailand during 2016–2017. EEM combining with PARAFAC
465 analysis showed that the identified fluorescent components were humic-like and protein-like
466 substances but different patterns in the WSOC and MSOC, indicating different chemical compositions.
467 By adding three-source fluorescence into the original PARAFAC model, we found that chromophores
468 with longer emission wavelengths in the atmosphere may be due to atmospheric chemical reactions or
469 “aging” by both bleaching the source chromophores or producing new chromophores. We also suggest
470 that caution is required when using fluorescence indices to appoint source of atmospheric
471 chromophores. In addition, more water-soluble BrC with stronger light absorption capability could be
472 extracted with ultrapure deionized water over Bangkok (0.83 ± 0.25 vs. 0.26 ± 0.12 $\text{m}^2 \text{g}^{-1} \text{C}$), and both
473 water-soluble and methanol-soluble BrC exhibited a high light-absorption in non-monsoon seasons
474 due to the influence of biomass burning. The MLR analysis showed that both the light absorption of
475 BrC at 365 nm in the two fractions was significantly dependent on the special fluorescent
476 chromophores with longer emission wavelength that are generally highly aromatic and oxygen-rich
477 with high apparent molecular weight. Positive matrix factorization model results further showed that
478 biomass burning was main contributor of these fluorescent chromophores (up to 50%). In summary,
479 this study provides a new insight into BrC absorption and sources, which may promote the application
480 of EEM spectroscopy to predict and model the light absorption of BrC in the atmosphere.

481 *Data availability.* The data used in this study are available upon request. Please contact Guangcai
482 Zhong (gczhong@gig.ac.cn).

483 *Supplement.* The supplement related to this article is available.

484 *Author contributions.* JT, GZ, JL, and GZ (Guangcai Zhong) designed the experiment. JT and JW
485 carried out the measurements and analyzed the data. JW and SB organized and performed the
486 samplings. JT (Jianhui Tang) supported the fluorescence instruments and laboratory. CT and HJ
487 supported the models. JT wrote the paper. JL, GZ (Guangcai Zhong), YC, YM, BZ, XG, and GZ
488 reviewed and commented on the paper.

489 *Competing interests.* The authors declare that they have no conflict of interest.

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