

# Impact of stratospheric air and surface emissions on tropospheric nitrous oxide during ATom

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**Abstract.** We measured the global distribution of tropospheric N<sub>2</sub>O mixing ratios during the NASA airborne Atmospheric Tomography (ATom) mission. ATom measured concentrations of ~300 gas species and aerosol properties in 647 vertical profiles spanning the Pacific, Atlantic, Arctic, and much of the Southern Ocean basins, from nearly Pole to Pole, over four seasons (2016–2018). We measured N<sub>2</sub>O concentrations at 1 Hz using a Quantum Cascade Laser Spectrometer. We introduced a new spectral retrieval method to account for the pressure and temperature sensitivity of

the instrument when deployed on aircraft. This retrieval strategy improved the precision of our ATom QCLS N<sub>2</sub>O measurements by a factor of 3 (based on the stdev. of calibration measurements). Our measurements show that most of the variance of N<sub>2</sub>O mixing ratios in the troposphere is driven by the influence of N<sub>2</sub>O-depleted stratospheric air, especially at mid and high latitudes. We observe the downward propagation of lower N<sub>2</sub>O mixing ratios (compared to surface stations) that tracks the influence of stratosphere-troposphere exchange through the tropospheric column down to the surface. The highest N<sub>2</sub>O mixing ratios occur close to the equator, extending through the boundary layer and free troposphere. We observed influences from a complex and diverse mixture of N<sub>2</sub>O sources, with emission source types identified using the rich suite of chemical species measured on ATom and with the geographical origin calculated using an atmospheric transport model. Although ATom flights were mostly over the oceans, the most prominent N<sub>2</sub>O enhancements were associated with anthropogenic emissions, including industry, oil and gas, urban and biomass burning, especially in the tropical Atlantic outflow from Africa. Enhanced N<sub>2</sub>O mixing ratios are mostly associated with pollution-related tracers arriving from the coastal area of Nigeria. Peaks of N<sub>2</sub>O are often associated with indicators of photochemical processing, suggesting possible unexpected source processes. In most of the cases, the results show the difficulty of separating the mixture of different sources in the atmosphere that may contribute to uncertainties in the N<sub>2</sub>O global budget. The extensive data set from ATom will help improve the understanding of N<sub>2</sub>O emission processes and their representation in global models.

## 1. Introduction

Nitrous oxide (N<sub>2</sub>O) is a powerful greenhouse gas and, due to its oxidation to NO<sub>x</sub>, a major contributor to both stratospheric ozone loss and to passivation of stratospheric oxy-halogen radicals (Forster et al., 2007; Ravishankara et al., 2009). The increasing rate of 0.93 ppb yr<sup>-1</sup> of atmospheric N<sub>2</sub>O since the industrial revolution implies significant (~30%) imbalance between emission rates and destruction in the stratosphere. Seasonal cycles in tropospheric N<sub>2</sub>O are driven by both stratosphere-to-troposphere exchange and by surface emissions (Nevison et al., 2011; Assonov et al., 2013; Thompson et al., 2014a). Most N<sub>2</sub>O emissions are attributed to microbial nitrification and denitrification in natural and cultivated soils, freshwaters and oceans, plus emissions related to human activities such as biomass burning and industrial emissions (Butterbach-Bahl et al., 2013; Saikawa et al., 2014; Thompson et al., 2014a; Upstill-Goddard et al., 2017; WMO, 2018).

Much effort has been made to reduce the uncertainties in the individual components of the N<sub>2</sub>O global budget (e.g., Tian et al., 2012, 2020; Xiang et al., 2013; Thompson et al., 2014a, b; Ganesan et al., 2020; Yang et al., 2020). Recent estimates of global total N<sub>2</sub>O emission to the atmosphere from bottom-up and top-down methods average 17 Tg N yr<sup>-1</sup> (12.2–23.5 from bottom-up analysis, and 15.9–17.7 Tg N yr<sup>-1</sup> from top-down approaches, Tian et al., 2020). The most recent estimates of the global ocean emission of N<sub>2</sub>O range between 2.5 and 4.3 Tg N yr<sup>-1</sup> (~ 20% of total emissions), with the tropics, upwelling coastal areas and subpolar regions as the major contributors to these fluxes (Yang et al., 2020; Tian et al., 2020). However, the magnitude of marine N<sub>2</sub>O emissions is subject to large uncertainty, due to spatial

and temporal heterogeneity (Nevison et al., 1995, 2005; Ganesan et al., 2020; Yang et al., 2020). According to Tian et al. (2020), anthropogenic sources account for ~ 43% of the global N<sub>2</sub>O emissions (7.3 Tg N yr<sup>-1</sup>), with industry and biomass burning emissions estimated to be 1.6 to 1.9 Tg N yr<sup>-1</sup> respectively (Syakila and Kroeze, 2011; Tian et al., 2020) and the rest from agriculture. N<sub>2</sub>O emissions from biogenic sources and fires in Africa are estimated at 3.3 ± 1.3 Tg N<sub>2</sub>O yr<sup>-1</sup> (Valentini et al., 2014). Agricultural N<sub>2</sub>O emission estimates (up to ~ 37%) range between 2.5 and 5.8 Tg N yr<sup>-1</sup>, and between 4.9 and 6.5 Tg N yr<sup>-1</sup> in the case of natural soils (Kort et al., 2008; 2010; Syakila and Kroeze, 2011; Tian et al., 2020). Recent estimates of N<sub>2</sub>O emissions from fertilized tropical and subtropical agricultural systems are 3 ± 5 kg N ha<sup>-1</sup> y<sup>-1</sup> (Albanito et al., 2017). Most of these estimates are derived from short-term local-scale in-situ measurements and are difficult to extrapolate with confidence to large regions or to the globe.

In the atmosphere, N<sub>2</sub>O is destroyed by oxidation (10%, O(<sup>1</sup>D) reaction) and photolysis (90%, 190-230 nm photolysis) in the upper stratosphere (> 20 km altitude; SPARC, 2013), which makes it a good candidate for tracing the air exchange between the stratosphere and the troposphere (Hintsa et al., 1998; Nevison et al., 2011; Assonov et al., 2013; Krause et al., 2018). Atmospheric models tend to underestimate the inter-hemispheric N<sub>2</sub>O gradient, which Thompson et al. (2014a) attribute to an overestimation of N<sub>2</sub>O emissions in the Southern Ocean, an underestimate of the Northern Hemisphere emissions, and/or an overestimate of stratosphere-to-troposphere exchange in the Northern Hemisphere. Overall, the largest uncertainties of modelled N<sub>2</sub>O emissions are found in tropical South America and South Asia (Thompson et al., 2014b).

We present atmospheric N<sub>2</sub>O altitude profiles at high temporal resolution collected during the NASA Atmospheric Tomography (ATom) mission. ATom was a global scale airborne deployment conducted over a 3-year period (2016–2018) using the NASA DC-8 aircraft. In ATom, the DC-8 flew vertical profiles (0.2–13 km) nearly continuously while measuring mixing ratios of ~300 trace gases and aerosol physical and chemical properties over the Pacific and Atlantic basins from nearly Pole to Pole and during each of the four seasons. Each deployment started and ended in Palmdale (California, USA). Each deployment (1 – 4) generally consisted of a loop southward from the Arctic through the central Pacific, across the Southern Ocean to South America, northward through the Atlantic, and across Greenland and the Arctic Ocean. During ATom-3 and -4, two additional flights from Punta Arenas (Chile) sampled the Antarctic troposphere and Upper Troposphere/Lower Stratosphere (UT/LS) to 80° S.

In this work we focus on the measurements taken during January–February 2017 (ATom-2), September–October 2017 (ATom-3), and April - May 2018 (ATom-4) (no Quantum Cascade Laser spectrometer (QCLS) N<sub>2</sub>O data are available for ATom-1 in Aug. 2016). The motivation for this paper is twofold. Firstly, we present a new retrieval strategy to account for the pressure and temperature dependence of laser-based instruments, specifically for the use of quantum cascade laser spectrometers on aircraft. Secondly, we report on the global distribution of N<sub>2</sub>O from the surface to 13 km, and examine the processes contributing to the variability of tropospheric N<sub>2</sub>O based on the vertical profiles of N<sub>2</sub>O and a broad variety of co-variate chemical species and aerosol properties.

## 110 2 Instrument specifications, spectral analysis and calibration

### 2.1 Specifications of the QCLS

We measured N<sub>2</sub>O mixing ratios with the Harvard/NCAR/Aerodyne Research Inc. Quantum Cascade Laser Spectrometer (QCLS). This instrument was previously deployed on the NCAR/NSF Gulfstream V for the HIAPER Pole-to-Pole Observations mission (HIPPO, Wofsy et al., 2011; [https://www.eol.ucar.edu/field\\_projects/hippo](https://www.eol.ucar.edu/field_projects/hippo)) and the  
115 O<sub>2</sub>/N<sub>2</sub> Ratio and CO<sub>2</sub> Southern Ocean Study (ORCAS, Stephens et al., 2018; [https://www.eol.ucar.edu/field\\_projects/orcas](https://www.eol.ucar.edu/field_projects/orcas)). Detailed information about the spectrometer configuration can be found in Jiménez et al. (2005, 2006) and Santoni et al. (2014). A brief description follows.

QCLS provides continuous (1 Hz) measurements of N<sub>2</sub>O, methane (CH<sub>4</sub>) and carbon monoxide (CO) using two thermoelectrically-cooled pulsed quantum cascade lasers, a 76 m pathlength multiple pass absorption cell (~ 0.5 L  
120 volume), and two liquid-nitrogen cooled solid-state HgCdTe detectors. All these components are mounted on a temperature-stabilized, vibrationally isolated optical bench. The temperature in QCLS is controlled by Peltier elements coupled with a closed-circuit recirculating fluid kept at  $288.0 \pm 0.1$  K. QCLS measures CH<sub>4</sub> and N<sub>2</sub>O by scanning the spectral interval of  $1275.45 \pm 0.15$  cm<sup>-1</sup>. A second laser is used to scan CO at  $2169.15 \pm 0.15$  cm<sup>-1</sup>. The supply currents to QCLS are ramped at a rate of 3.8 kHz to scan the laser frequency for 200 channels (steps in frequency) in laser 1 and  
125 50 channels in laser 2, with an extra 10 channels to measure the laser shut off (zero-light level). The spectra and fit residual for CH<sub>4</sub>, N<sub>2</sub>O and CO are shown in Fig. S1 of the Supplement. Mixing ratios are derived at a rate of 1 Hz by a least-squares spectral fit assuming a Voigt line profile at the pressure and temperature measured inside the sample cell and using molecular line parameters from the HIGH-resolution TRANsmission molecular absorption database (HITRAN, Rothman et al., 2005). The temperature and pressure inside the cell are monitored with a 30 k $\Omega$  thermistor  
130 and a capacitance manometer (133 hPa full scale), respectively.

During sampling, the air passes through a 50-tube Nafion drier to remove the bulk water vapor. A Teflon diaphragm pump downstream of the cell reduces the air pressure to ~60 hPa. Both ambient air and calibration gases pass through a Teflon dry-ice trap to reduce the dew point to -70 °C. After ATom-1, we added a bypass between the inlet and the instrument to increase the flushing rate of the inlet and inlet tubing. The calibration sequence includes 2 minutes of  
135 Ultra-High Purity zero air, followed by 1 minute each of low- and high-mixing ratio gases every 30 minutes (see Fig. S2). During ATom-1 and -2, we measured zero air every 15 minutes, and every 30 minutes during ATom-3 and -4. A data logger (CR10X, Campbell Scientific) was used to automate the sampling sequence. The CR10X controlled the pressure controller on the cell and managed the data transfer.

We use gas cylinders traceable to the National Oceanic and Atmospheric Administration World Meteorological  
140 Organization scales for calibration (NOAA-WMO-X2004A scale for CH<sub>4</sub>, WMO-X2014A for CO and NOAA-2006A for N<sub>2</sub>O). These gas standards were recalibrated before, during and after the deployments to maintain traceability. The

low-mixing ratio gas cylinder contained  $298.5 \pm 0.3$  ppb of  $\text{N}_2\text{O}$ ,  $1692.4 \pm 0.2$  ppb of  $\text{CH}_4$  and  $119.1 \pm 0.3$  ppb for  $\text{CO}$ . The high-mixing ratio gas cylinder contained  $399.1 \pm 0.3$  ppb of  $\text{N}_2\text{O}$ ,  $2182.5 \pm 0.3$  ppb of  $\text{CH}_4$  and  $192.8 \pm 0.5$  ppb of  $\text{CO}$ , respectively. Detailed information on calibrations of the gas cylinders used during ATom are in Table S1 of the Supplement.

QCLS also measures carbon dioxide ( $\text{CO}_2$ ) in a separate unit. Detailed information about QCLS  $\text{CO}_2$  measurements can be found in Santoni et al. (2014).

## 2.2 Spectral analysis and calibration

The QCLS was damaged during shipping to the deployment site before the start of ATom-1, and the resulting alteration in the optical alignment modified the sensitivity of the instrument to temperature and pressure changes during aircraft maneuvers. This increased sensitivity was observed in all ATom deployments. At a constant altitude, instrumental precision was similar to the precision measured during HIPPO (see Allan-Werle variance analysis in Fig. 2 in Santoni et al. 2014 for HIPPO and Fig. S3 for ATom), but drifts were observed during altitude changes due to the effects of changes in cabin pressure and temperature on the spectral location of interference fringes that arise in the optical path outside the sample cell. In addition, flight altitude changes could mechanically stress the optical elements surrounding the cell, further modulating fringes or changing the shape of the detected laser intensity profile. These spectral artifacts ultimately reduced the accuracy of mixing ratios retrieved from spectral fitting. The spectral artifacts most strongly affected the measurements of  $\text{CH}_4$  and  $\text{N}_2\text{O}$ . Several post-processing methods using the TDL-Wintel software were explored to improve the precision and accuracy of ATom QCLS  $\text{N}_2\text{O}$  data, most with little success. Since the measured spectra were all saved, it is possible to re-fit the data with different fit parameters. A limited number of interference fringes may be included in the set of fitting functions. However, none of the previously used full re-fitting strategies significantly improved the data accuracy.

We have achieved significant improvement in the precision and accuracy of the ATom QCLS  $\text{N}_2\text{O}$  data using a new method dubbed the “Neptune algorithm”, developed by Aerodyne Research, Inc., and that has been further developed and applied to the data sets described here. Using this algorithm, the precision of the retrieved  $\text{N}_2\text{O}$  data measured with the damaged QCLS, was similar to that reported in HIPPO. The Neptune algorithm generates corrections to the mixing ratios retrieved from the original fits by associating specific spectral features with anomalies in retrieved mixing ratios observed during calibrations, i.e., during intervals when the mixing ratios are held constant. The spectral baseline is defined as the spectral channels outside the boundaries of the spectral lines of the target gas. Fluctuations in the spectral baselines are quantified for the entire data set by means of principal component analysis (PCA). PCA provides an efficient description of the spectral fluctuations, naturally producing an ordered set from strongest to weakest of orthogonal vectors (spectral forms), each with an amplitude history spanning the data set. The PCAs are defined by an optimization procedure during calibrations, when mixing ratio fluctuations are designed to be  $\sim 0$ . The finite fluctuations in retrieved mixing ratios during calibrations are fit in the spectral space of the baseline as linear

combinations of the leading PCA vector amplitudes, creating a linear combination of amplitudes of spectral fluctuations that predict errors in the mixing ratios for each gas for an entire flight. The error-producing linear combination of amplitudes of PCA spectral fluctuations produces a full set of anomaly estimates, which is subtracted from the retrieved mixing ratios during the flight. The computational time for a 10-hour long dataset is only seconds, so variations in the algorithm's parameters (i.e. how many PCAs are retained) can be optimized rapidly.

The Neptune-PCA analysis improved the overall precision by a factor of 4 for CH<sub>4</sub> and a factor of 3 in the case of N<sub>2</sub>O with respect to the precision of the original retrievals, as measured by the standard deviation of retrieved mixing ratios during calibrations. The repeatability of the retrieved calibrations was 0.2 ppb for N<sub>2</sub>O and 1 ppb for CH<sub>4</sub> (Fig. S4). The laser path of the CH<sub>4</sub>/N<sub>2</sub>O laser was realigned between ATom-1 and -2 and the Neptune retrieval was applied to CH<sub>4</sub> and N<sub>2</sub>O measurements corresponding to the ATom-2, -3 and -4 deployments. Mixing ratios of CH<sub>4</sub> and N<sub>2</sub>O could not be retrieved during ATom-1 because light levels were too low for the CH<sub>4</sub>/N<sub>2</sub>O laser due to the damage-induced misalignment.

The steps involved in the Neptune correction process were as follows.

1) We paired the mixing ratio records with the corresponding spectra (1-s resolution) for each species (CH<sub>4</sub> and N<sub>2</sub>O).

2) We grouped the mixing ratios and spectra by type as calibrations (zeros, low span and high span) and air samples, and in time. The spectral data were thus arranged in an array, with point number in the spectrum as x, and spectrum number as y. We calculated an average spectrum for each group type and subtracted these from each individual spectrum within a group.

3) We zeroed-out the spectral arrays at the positions of the absorption lines to concentrate on the fluctuations observed in the baseline and to prevent the PCA from finding line-depth fluctuations as relevant vectors during the calibrations. Some degree of smoothing (in x) was applied to the subtracted spectra, so that high-frequency fluctuations, which have little influence on the mixing ratio determination, are not represented. An example of such a processed spectral array is shown in Fig. 1a.

4) We applied PCA to the whole line-zeroed spectral array to evaluate the fluctuations. PCA is applied in two steps: multiply the spectral array by its transpose, to generate an autocovariance array; and then perform singular value decomposition on the autocovariance array. The PCA generated an efficient description of how the baseline of the spectrum changed with cabin pressure and temperature. The description of spectral fluctuations is made in terms of a set of products of vectors and amplitudes.

5) We fit the spectra to the PCAs to express mixing ratio fluctuations during the set of calibrations and zeros as a linear combination of PCA vector histories. The number of vector histories that we included in the fit typically is limited to less than 30, because the weaker PCA amplitudes tend to just describe random noise.

The linear combination of amplitudes that links spectral fluctuations in the baseline to mixing ratio fluctuations during calibrations, was then applied to the full data set. That generated the retrieval errors for uncalibrated mixing

210 ratios, for the whole time series. We subtracted the errors from the initial retrievals from TDLWintel-QCLS software  
and computed calibrated mixing ratios using the corrected retrievals for both calibrations and samples. An example of  
the result of the Neptune algorithm on the N<sub>2</sub>O samples and calibrations for the ATom-4 flight on 12 May 2018 is  
shown in Fig. 1b. The approach used here to minimize the effect of changes in pressure and temperature in optical  
215 instruments is based on the observation of fluctuations of the baseline during calibrations. Hence, in those cases where  
altitude changes occurred during sampling but not during any of the calibrations for an individual flight, this  
methodology will not provide any improvement. Due to frequent calibrations, we did not observe this rare scenario in  
the whole mission. To evaluate the ultimate accuracy of our measurements we compare the QCLS N<sub>2</sub>O measurements  
with other on-board N<sub>2</sub>O measurements as well as with the surface N<sub>2</sub>O measurements of stations located along the  
flight tracks.

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### 3 Accuracy of N<sub>2</sub>O measurements from QCLS

We evaluated N<sub>2</sub>O mixing ratios measured by QCLS against three other instruments measuring N<sub>2</sub>O on the NASA DC-  
8 aircraft during ATom. In addition, we compared the set of 4 airborne measurements to data from the flask sampling  
network at ground stations from the NOAA Global Monitoring Laboratory (GML, <https://www.esrl.noaa.gov/gmd/>) to  
225 evaluate the differences between the airborne data and the ground-based measurements in the NOAA reference network.

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#### 3.1 Comparison between airborne N<sub>2</sub>O measurements

Measurements of N<sub>2</sub>O on the DC-8 during ATom were obtained by four instruments (i) the Unmanned Aircraft Systems  
Chromatograph for Atmospheric Trace Species (UCATS, Hintsa et al., 2021), (ii) the PAN and other Trace  
230 Hydrohalocarbon ExpeRiment (PANTHER; Moore et al., 2006; Wofsy et al., 2011), (iii) the Programmable Flask  
Package Whole Air Sampler (PFP; Montzka et al., 2019), and (iv) our 1-Hz QCLS.

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We compared QCLS, PANTHER and UCATS in 10s intervals as provided in the ATom merged file,  
MER10\_DC8\_ATOM-1.nc, available at the Oak Ridge National Laboratory Distributed Active Archive Center (ORNL-  
DAAC, Wofsy et al., 2018, <https://doi.org/10.3334/ORNLDAAC/1581>). The ATom file MER-PFP merged with the  
235 PFP sampling interval, also available in this repository, was used to compare QCLS and PFP data. The one-to-one  
comparison between these instruments showed an approximately 1 ppb positive bias in N<sub>2</sub>O mixing ratios from the  
QCLS (see Fig. 2A1–2A3). The 95% confidence interval of the mean difference of the pairs (95% C.I.) was  $0.75 \pm 0.04$   
ppb between QCLS and PANTHER,  $1.13 \pm 0.03$  ppb between QCLS and UCATS, and  $1.18 \pm 0.09$  ppb between QCLS  
and PFP, respectively, for the full data set (ATom-2, -3 and -4). Information about the coefficients of the linear fit for  
240 each instrument comparison and the 95% C.I. of the difference for each pair are shown in Table of S2. The offset that  
QCLS N<sub>2</sub>O shows against PFP N<sub>2</sub>O coincides with the offset already reported by Santoni et al. (2014) during HIPPO  
in 2009-2011, which may be attributed to our calibration procedure. PFP is the reference instrument on board, analyzed  
with excellent precision and accuracy.

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### 245 3.2 Comparison between airborne and surface measurements of N<sub>2</sub>O

We evaluate the traceability of lower-troposphere N<sub>2</sub>O mixing ratios by ATom by comparing the four airborne instruments with the surface measurements of N<sub>2</sub>O from the NOAA flask sampling network. If during a flight, a surface station was encountered within a latitude range of 5 degrees north and south with respect the flight track, that surface station was used in the study.

250 A mean value of N<sub>2</sub>O within that latitude grid of +/- 5 degrees and between 1 to 4 km altitude of instrument was compared with the mean N<sub>2</sub>O at the surface station observed between +/- 5 days relative to the flight (due to the non-daily frequency of flask samples). We chose the altitude range between 1 to 4 km to agree with the low free troposphere conditions that characterized most of the selected ground stations. Information about the surface stations used here is shown in Table of S.3 of the Supplement.

255 The comparison of the whole data set (ATom-2, -3, -4) shows an overall overestimation of N<sub>2</sub>O mixing ratios by QCLS and PANTHER with respect to the surface data of  $1.37 \pm 0.35$  and  $0.44 \pm 0.51$  ppb (95% C. I.), respectively. In contrast, UCATS and PFP showed  $0.27 \pm 0.37$  and  $0.008 \pm 0.34$  ppb (95% C. I.), low bias with respect to the surface data, respectively (Fig.2B1-2B4). Due to the excellent agreement between PFP and the surface stations and the consistent offset that QCLS shows against PFP and the stations, in the following sections, QCLS N<sub>2</sub>O data presented  
260 in this publication is corrected by subtracting the offset with respect to the PFP data on-board in each deployment:  $1.03 \pm 0.13$  ppb in AT-2,  $1.49 \pm 0.19$  ppb in AT-3, and  $1.18 \pm 0.17$  ppb in AT-4. The final official archive data file includes a new column where these corrections have been applied (N<sub>2</sub>O\_QCLS\_ad).

These results show the very close comparability of ATom airborne N<sub>2</sub>O instruments (differences < 0.5 ppb for UCATS and PANTHER instruments, 0 ppb in the case of PFP) relative to the surface stations and demonstrate the  
265 feasibility of using ATom N<sub>2</sub>O measurements to evaluate the impact of stratospheric air and meridional transport of N<sub>2</sub>O emissions in N<sub>2</sub>O tropospheric column measurements over the ocean basins. In the following section, we define the boundary conditions based on the NOAA Greenhouse Gas Marine Boundary Layer Reference from the NOAA GML Carbon Cycle Group (NOAA/ESRL GML CCGG, <https://www.esrl.noaa.gov/gmd/ccgg/mbl/>) for evaluating the impact of stratospheric air and meridional transport of N<sub>2</sub>O emissions in N<sub>2</sub>O tropospheric column over the ocean  
270 basins. The NOAA-MBL N<sub>2</sub>O product is synthetic latitude profile generated at 0.05 sine latitude and weekly resolution (<https://www.esrl.noaa.gov/gmd/ccgg/mbl/>) from individual flask measurements of marine boundary-layer surface stations distributed along the two ocean basins, and provides the scenario to evaluate the traceability of aircraft measurements relative to ground measurements in remote sites (<https://doi.org/10.15138/53g1-x417>).

## 275 4. Results and Discussion

The vertical profiles of N<sub>2</sub>O from ATom provide a global overview of the N<sub>2</sub>O distribution in the troposphere with observations over the Pacific and Atlantic basins. For this study we do not include data collected over and close to land.



In ATom, N<sub>2</sub>O ranged between 280 and 335 ppb over the oceans. In each season, the lowest N<sub>2</sub>O mixing ratios are observed at high latitudes (HL, > 60°) in the UT/LS (8–12.5 km) in air transported downward from the stratosphere. The highest N<sub>2</sub>O mixing ratios are found close to the equator (30° S–30° N, 326 to 335 ppb), extending along the tropospheric column up to 6 km and influenced by convective activity over the tropical regions (Kort et al., 2011; Santoni et al., 2014). At mid-latitudes (ML, 30°–60° N), tropospheric N<sub>2</sub>O values ranged between 322 and 333 ppb. Tropospheric N<sub>2</sub>O tends to increase towards northern latitudes as a result of higher anthropogenic emissions in the Northern Hemisphere relative to the Southern Hemisphere. More details on the variability of N<sub>2</sub>O mixing ratios along the tropospheric column are described in Section S1.

We study the impact of N<sub>2</sub>O sources and stratospheric air on the N<sub>2</sub>O column based on the anomalies (enhancements and depletions) we observed in the airborne N<sub>2</sub>O mixing ratios relative to the N<sub>2</sub>O “background” defined here as the NOAA-MBL product. We use the NOAA-MBL product to constrain a latitudinal gradient of N<sub>2</sub>O mixing ratios at the surface for each deployment. These data have been widely used to estimate the N<sub>2</sub>O background (Assonov et al., 2013; Nevison et al., 2011). More information about the NOAA-MBL product and the latitudinal gradient of their measurements is discussed in Section S2. This approach highlights the extra information that aircraft profiles can provide. Cross-sections of N<sub>2</sub>O anomalies are shown in Fig. 3. The data describe the overall homogeneity of N<sub>2</sub>O in the troposphere (30% of the anomalies ranged between ± 0.5 ppb). We consider the ± 0.5 ppb interval to account for the day-to-day and seasonal variability of N<sub>2</sub>O. Episodes of N<sub>2</sub>O depletion (< -0.5 ppb) related to the influence of stratospheric air are observed in 53.5% of the aircraft samples during ATom-2 to -4, whereas episodes of N<sub>2</sub>O enhancement (> 0.5 ppb) related to the contribution of N<sub>2</sub>O sources account for 16.5% of the calculated anomalies.

Trajectories and associated surface influence functions were computed using the Traj3D model (Bowman, 1993) and wind fields from the National Center for Environmental Prediction Global Forecast System (NCEP GFS). Model trajectories were initialized at receptors spaced one minute apart along the ATom flight tracks, and followed backwards for 30 days, and reported at 3-hour resolution. From these trajectories we calculated the surface influence for each receptor point (footprints in units of ppt/(nmol m<sup>-2</sup> s<sup>-1</sup>)). The footprint can be convolved with a known flux inventory of a non-reactive gas to calculate the expected enhancement/depletion of that gas for each receptor point.

#### 4.1 Impact of stratospheric air on tropospheric N<sub>2</sub>O mixing ratios during ATom

We observed the strongest depletions (> 5 ppb) in N<sub>2</sub>O mixing ratios at high latitudes and altitudes, consistent with stratospherically influenced air (Fig. 3). Stratosphere-troposphere exchange processes allow stratospheric-depleted N<sub>2</sub>O to be distributed throughout the troposphere. The NOAA surface network shows a seasonal minimum of N<sub>2</sub>O 2–4 months later than the stratospheric polar vortex break-up season. This seasonal minimum is observed at the surface around May in the southern hemisphere and around July in the northern hemisphere (see Fig. S8 and S9) (*cf.* Nevison et al., 2011 and references therein). The enhanced downwelling of the Brewer–Dobson circulation (BDC) in late winter–spring, reinforces the downward transport of stratospheric air depleted in N<sub>2</sub>O throughout the free troposphere (1–8

km), as observed in October in the southern hemisphere (ATom-3, Fig. 3c and 3f) and in May in the North Atlantic (ATom-4, Fig 3e). The N<sub>2</sub>O depletion is likely the result of stratospheric air being moved downwards by the BDC and trapped by the polar vortex, with a more pronounced effect in the southern hemisphere where the polar vortex is stronger. These results support previous work suggesting that downward transport of stratospheric air with low N<sub>2</sub>O exerts a strong influence on the variance of tropospheric N<sub>2</sub>O mixing ratios (Nevison et al., 2011; Assonov et al, 2013).

The impact of stratosphere-to-troposphere transport can be studied by combining information on tracers of stratospheric air such as ozone (O<sub>3</sub> from the NOAA - NO<sub>y</sub>O<sub>3</sub>; Bourgeois et al., 2020), sulfur hexafluoride (SF<sub>6</sub> from PANTHER), CFC<sub>12</sub> (from PANTHER) and carbon monoxide (CO from QCLS). These tracers are usually used either because they are strongly produced in the stratosphere (e.g. O<sub>3</sub>) or because they are tracers of anthropogenic emissions in the troposphere with a strong stratospheric sink (e.g., CO, SF<sub>6</sub> and CFC<sub>12</sub>). In addition, meteorological parameters such as potential vorticity (PV), the product of absolute vorticity and thermodynamic stability (PV was generated by GEOS5-FP for ATom) can be used to trace the stratosphere-to-troposphere transport.

Overall, the interhemispheric gradient of N<sub>2</sub>O is much smaller than that of CO and SF<sub>6</sub> (Fig. 4), but the difference for each species is driven by larger anthropogenic emissions in the northern hemisphere. The tracer-tracer correlations shown in Fig. 4 show different patterns. The linear trend between N<sub>2</sub>O and O<sub>3</sub> or CFC-12 highlights the role of depletion (N<sub>2</sub>O and CFC-12) and production (O<sub>3</sub>) in the stratosphere (Fig. 4a1, 4a4). When N<sub>2</sub>O is plotted against the anthropogenic tracers, CO and SF<sub>6</sub>, two distinct trends are observed. Tropospheric N<sub>2</sub>O can be identified as the horizontal band containing high N<sub>2</sub>O (> 328 ppb) and variable CO and SF<sub>6</sub>, whereas the vertical band with variable N<sub>2</sub>O and small changes in CO and SF<sub>6</sub>, shows the mixing between tropospheric air and stratospheric air depleted in N<sub>2</sub>O (Fig. 4a1–4a3). The N<sub>2</sub>O versus CO plot shows an L-shaped (bimodal) curve similar to those typically observed on O<sub>3</sub>-CO correlations during events of stratosphere-to-troposphere air mass mixing (Fig. 4a2, Krause et al., 2018). A quasi-vertical line in the N<sub>2</sub>O–CO plot (e.g. constant CO) is indicative of a strong impact of stratospheric air, where CO shows the stratospheric equilibrium mixing ratio (Krause et al., 2018). The lower the CO background, the greater the influence of the stratospheric air during the air mass mixing (North Atlantic high latitudes in Fig. 4a2) and vice versa. A strong correlation is also indicative of rapid mixing between the two air masses. During ATom, the strongest impact of stratospheric air was observed in the Pacific mid and high latitudes in February (ATom-2) and in the Atlantic in May (ATom-4, Fig. S11). At the Pacific northern mid and high latitudes (NMHL > 30° N), we found a consistent linear relationship between N<sub>2</sub>O and O<sub>3</sub>, with a relatively constant N<sub>2</sub>O/O<sub>3</sub> slope (-0.05 to -0.04) during all seasons. Linear correlations between N<sub>2</sub>O and CFC-12 highlight the dominant influence of stratospheric air depleted in these two substances in the range of mixing ratios observed at mid and high latitudes (Fig. S11).

During spring, the mid-latitudes are strongly impacted by stratospheric air due to the occurrence of tropopause folds and cut-off lows to the south of the westerly subtropical jets (Hu et al., 2010 and references therein). The stronger depletion of N<sub>2</sub>O mixing ratios observed over the Atlantic relative to the Pacific during spring is due to a greater number of deep stratosphere-to-troposphere transport events at middle latitudes in the region between May and July (Fig. 3e;

Cuevas et al., 2013 and references therein). Anomalies in PV relative to its mean latitudinal distribution in the free troposphere (2–8 km) highlight events of strong downward transport of stratospheric air. Negative PV, N<sub>2</sub>O, CO, and CFC-12 anomalies (positive for O<sub>3</sub>) describe the transport of stratospheric air into the troposphere in the SH, whereas positive PV and negative N<sub>2</sub>O, CO, and SF<sub>6</sub> anomalies (positive for O<sub>3</sub>) describe the downward transport of stratospheric air in the NH (Fig. 4b1-4b4). The correlations between N<sub>2</sub>O and PV, and the similarities with CFC-12, indicate that stratosphere-to-troposphere exchange leads to variations of tropospheric N<sub>2</sub>O up to 10 ppb in the higher latitudes for the altitudes covered during the flights. This influence is notably larger than the 2–4 ppb enhancements associated with regional emissions (see below).

#### 355 4.2 Impact of emissions on tropospheric N<sub>2</sub>O mixing ratios during ATom

During ATom, episodes of N<sub>2</sub>O positive anomalies relative to the surface station MBL reference occur close to the equator (Fig. 3a-c), and in a few locations at mid latitudes, in both ocean basins across all seasons. We use the information from the vertical profiles including back-trajectories and correlated chemical tracers to trace the origins of these enhancements including: CO, CH<sub>4</sub>, and CO<sub>2</sub>, measured by the QCLS and NOAA Picarro 2401-m, hydrogen cyanide (HCN), sulfur dioxide (SO<sub>2</sub>), hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) and peroxyacetic acid (PAA) measured with the California Institute of Technology Chemical Ionization Mass Spectrometers (CIT-CIMS, Crouse et al., 2006; St. Clair et al., 2010), ammonium (NH<sub>4</sub><sup>+</sup>), sulfate (SO<sub>4</sub><sup>2-</sup>) and nitrate (NO<sub>3</sub><sup>-</sup>) and organic aerosols (OA) from the Colorado University Aircraft High-Resolution Time-of-Flight Aerosol Mass Spectrometer (HR-AMS, DeCarlo et al., 2006; Canagaratna et al., 2007; Jimenez, et al., 2019; Guo et al., 2020; Hodzic, et al., 2020), NO<sub>y</sub> from the NOAA NO<sub>y</sub>O<sub>3</sub> 4-channel chemiluminescence (CL, Ryerson et al., 2019), CH<sub>2</sub>Br<sub>2</sub>, CH<sub>3</sub>CN, benzene and propane from the NCAR Trace Organic Gas Analyzer (TOGA, Apel et al., 2019) instrument, and atmospheric potential oxygen (APO  $\approx$  O<sub>2</sub> + 1.1 x CO<sub>2</sub>) from the NCAR Airborne Oxygen Instrument (AO2, Stephens et al., 2020).

We calculate the correlations between N<sub>2</sub>O and the mentioned species in three layers (0-2000 m, 2000-4000 m, and 4000-6000 m). Correlation coefficients in each layer for a given profile were calculated using a minimum threshold of 15 data points per layer. These profiles show that many of the most prominent enhancements of N<sub>2</sub>O are closely associated with pollutants such as HCN, CH<sub>3</sub>CN, H<sub>2</sub>O<sub>2</sub>, and other pollutants associated with combustion and photochemical air pollution. Some profiles show peaks closely correlated with SO<sub>2</sub> and enhanced PM1 particles, and vertical gradients were sometimes correlated with gradients of APO and HCN.

Several N<sub>2</sub>O peaks were observed together with enhancements of H<sub>2</sub>O<sub>2</sub> and PAA, which are primarily formed from chemistry occurring in the atmosphere. For the altitude range 2–4 km, regressions produced  $r^2 > 0.7$  for 16 profiles of N<sub>2</sub>O vs. H<sub>2</sub>O<sub>2</sub> and 15 profiles of N<sub>2</sub>O vs. HCN (a tracer for combustion of biomass), but only three such profiles produced these strong associations for both H<sub>2</sub>O<sub>2</sub> and HCN in common. Some of these profiles, shows also correlated enhancements of SO<sub>2</sub> and NO (9 profiles  $r^2 > 0.6$ ). This result raises the question as to whether globally significant production of N<sub>2</sub>O may be occurring in heterogeneous reactions involving SO<sub>2</sub>, NO redox chemistry and HONO nearby

380 to strong sources of reactive pollutants, as has been observed in heavily polluted atmospheres (Wang et al., 2020), or  
as theorized to occur in the plumes of refineries or power plants (e.g. Pires and Rossi, 1997).

In most cases, because we were sampling in the middle of the oceans and not over the source regions, the distinction  
between the different sources contributing to the observed N<sub>2</sub>O enhancements is not possible. We also observe that the  
impact of the different sources to N<sub>2</sub>O mixing ratios is regionally dependent. Here we describe, with some examples,  
385 the sources contributing to the major enhancements of N<sub>2</sub>O observed during ATom by oceanic regions, although we  
cannot pinpoint precisely the source processes.

#### 4.2.1 N<sub>2</sub>O enhancements over the Pacific

Episodes of N<sub>2</sub>O enhancements were frequently observed at the southern Pacific Ocean mid latitudes, linked by the  
associated footprints to emissions over the continents. In this region, N<sub>2</sub>O enhancements are predominantly associated  
390 with airmasses with enhanced H<sub>2</sub>O<sub>2</sub>, PAA and CO. For example, consider Fig. 5 showing data from profile 12 on 3  
May 2018, at 49.5–50° S near the Dateline. A distinct peak in N<sub>2</sub>O of 1 ppb amplitude, at 1700 m altitude, is significantly  
correlated with enhancements in CH<sub>3</sub>CN. These associations and the footprints suggest a regional contribution from  
fuel types from the industrial zone of Australia (Fig. 5c) which is also confirmed by the aerosol characterization from  
PALMS (not shown for brevity). In this profile, close to the surface, the lowest QCLS N<sub>2</sub>O mixing ratios agree with  
395 the NOAA MBL N<sub>2</sub>O (dashed line in Fig. 5b). At higher altitudes (2.5–6 km), strong correlations between N<sub>2</sub>O, H<sub>2</sub>O<sub>2</sub>,  
PAA, CO and HCN, but not SO<sub>2</sub>, suggest the influence of biomass burning from central Australia (3–5 km) and from  
South America (6 km) (Fig. 5b and 5c middle and right panels, and Fig. S11f). The relatively low mixing ratios of  
short-lived trace gases (PAA, H<sub>2</sub>O<sub>2</sub> and PM<sub>1</sub> aerosols with lifetimes ranging from hours to few days), and the surface  
400 influence based on the back trajectories (Fig. S13a), indicate that most of these profiles sampled significantly aged air  
masses transported for extended periods over the South Pacific.

At the equatorial Pacific, episodes of N<sub>2</sub>O enhancements were frequently associated with a mixture of potential  
marine, industrial and biomass burning emissions. Atmospheric potential oxygen (APO) is a tracer primarily of oxygen  
exchange with the oceans, defined as deviations in the oxygen-to-nitrogen ratio ( $\delta(\text{O}_2/\text{N}_2)$ ) corrected for changes in O<sub>2</sub>  
405 due to terrestrial photosynthesis and respiration, and also mostly for influences from combustion (Stephens et al., 1998),

$$\delta\text{APO} = \delta(\text{O}_2/\text{N}_2) + 1.1 / X_{\text{O}_2} (X_{\text{CO}_2} - 350) \quad (1)$$

Here  $\delta(\text{O}_2/\text{N}_2)$  is the deviation in O<sub>2</sub>:N<sub>2</sub> ratio (per meg), 1.1 is an approximation to the O<sub>2</sub>:CO<sub>2</sub> ratio for  
410 photosynthesis and respiration,  $X_{\text{O}_2}$  is the mole fraction of O<sub>2</sub> in dry air, and  $X_{\text{CO}_2}$  is the mole fraction of CO<sub>2</sub> in the air  
sample (dry,  $\mu\text{mol mol}^{-1}$ ). Since APO primarily tracks oxygen exchange between the ocean and the atmosphere,  
depletions in APO can indicate marine N<sub>2</sub>O emissions from areas with strong upwelling (Leucker et al., 2003; Ganesan  
et al., 2020). However, APO is also sensitive to pollution such as biomass burning and fossil fuel combustion (Lueker

et al., 2001) and because both N<sub>2</sub>O and APO have meridional gradients resulting from many influences, correlations  
415 can result simply from sampling air transported from different latitudes. In ATom, 9 profiles showed significant  
correlations ( $r^2 > 0.7$ ) between N<sub>2</sub>O and APO (or  $\delta(\text{O}_2:\text{N}_2)$ , which has lower measurement noise), for altitude bins 0–2  
km (8) and 2–4 km (1) where backtrajectories trace their origin close to the North American west coast and the  
Mauritanian coast, and in the equatorial Pacific. The median slope of regressions of APO vs. N<sub>2</sub>O for these profiles in  
ATom was -0.04, and the mean -0.05 ( $\pm 0.04$ ,  $1\sigma$ ) ppb per meg, very similar to the range found by Ganesan et al. (2020)  
420 and Leuker et al. (2003) in coastal areas.

An example is shown in Fig. 6 for 1 May 2018. We observed high correlation between N<sub>2</sub>O and APO ( $r^2 = 0.66$ )  
between 0 and 4 km altitude. At these altitudes we also see enhancements in dibromomethane (CH<sub>2</sub>Br<sub>2</sub>), a tracer of  
phytoplankton biomass (Liu et al., 2013 and references therein), consistent with a marine biological flux of halogenated  
VOCs (Asher et al., 2019), dimethyl sulfide (DMS) and methanesulfonic acid (MSA), main particulate product of DMS  
425 oxidation in the MBL. However, on this flight, the footprints and the surface ocean influence (Fig. S12b) indicate that  
this N<sub>2</sub>O gradient represents a difference in sampling a marine air mass from the south near the surface and a more  
continentally air mass from the east at 4 km (Fig. 6A–C). Close to the surface, the lowest QCLS N<sub>2</sub>O mixing ratios  
agree with the NOAA MBL N<sub>2</sub>O at the origin of the airmasses suggested by the footprints (25° S, dashed red line in  
Fig. 6b), whereas at 4 km, the lowest QCLS N<sub>2</sub>O mixing ratios agree with the NOAA MBL N<sub>2</sub>O (dotted blue line in  
430 Fig. 6b). Thus, the N<sub>2</sub>O:APO correlation most likely represents the latitudinal and ocean-land gradients established for  
a combination of reasons, with higher APO and lower N<sub>2</sub>O from higher southern latitudes away from continents. In this  
particular flight, N<sub>2</sub>O variations were more noticeable between 4 and 6 km height, in relation to biomass burning plumes  
from fires occurring in Venezuela and the Caribbean, in agreement with simultaneous enhancements on CO, HCN  
mixing ratios (Fig. 6A, C and Fig. S12); and oil and gas pollution sources near coastlines in the first 2 km with increasing  
435 SO<sub>2</sub>. The nature of these emissions was also confirmed by the aerosol characterization (from the PALMS instrument,  
figure not shown).

#### 4.2.2 N<sub>2</sub>O enhancements over the Atlantic

The Atlantic basin ATom flights saw much more continental influence than the Pacific. Around 30° N, in the  
440 North Atlantic during winter, we observe small enhancements of N<sub>2</sub>O that contrast with the overall influence of  
stratospheric air on the tropospheric column (AT-2, Fig. 3d). The contribution is much higher during the fall season  
(AT-3, Fig. 3f). Several episodes of N<sub>2</sub>O enhancements are associated with enhancements of CH<sub>4</sub>, CO, and HCN. We  
also observe some episodes where N<sub>2</sub>O increases while CO<sub>2</sub> decreases (figure not shown), which could reflect the  
accumulation of agricultural emissions over the summer or just greater sampling of Northern Hemisphere summer air  
445 masses, whereas increases of N<sub>2</sub>O with CO are indicators of urban pollution and together with HCN are associated with  
a few episodes of biomass burning.

The influence of different regions on the N<sub>2</sub>O mixing ratios over the Atlantic is shown on 14 May 2018 (Fig. 7). This profile shows the contribution to tropospheric N<sub>2</sub>O from western Europe pollution transported down over the Mauritanian coast, combined with biomass burning emissions, urban and industrial emissions from southern Africa and the Middle East (between 1.5 and 3 km), and polluted air masses from South America and the African west coast mixed with the oceanic contribution to N<sub>2</sub>O (~10 km, Fig. 7A–C and Fig. S13). The aerosol characterization (PALMS, not shown) indicates mineral dust and biomass burning emissions influencing the atmospheric layer between 1 to 6 km and oil combustion influence below 4 km. At high altitudes, N<sub>2</sub>O enhancements are caused by the interception of polluted airmasses from South America and the African west coast mixed with the oceanic contribution to N<sub>2</sub>O (~10 km). The N<sub>2</sub>O:APO correlations for the feature between 1.5 and 3 km most likely represent depletion of APO by industrial combustion, stoichiometrically consistent with the observed increases in CO<sub>2</sub> and CH<sub>4</sub> for this feature.

During ATom, we observed large contributions to tropospheric N<sub>2</sub>O over the Atlantic Ocean from Africa, with some influence from Europe and South America. In the subtropical and tropical regions over the Atlantic during AT-2, we found strong correlations between N<sub>2</sub>O and H<sub>2</sub>O<sub>2</sub> and PAA, HCN, CO, CO<sub>2</sub>, SO<sub>2</sub>, OA, NH<sub>4</sub><sup>+</sup> and SO<sub>4</sub><sup>2-</sup> between 0 and 2.5 km, representing the combined influence of photochemistry ( $r_{\text{N}_2\text{OvsPAA}}^2 = 0.94$ ), biomass burning events from the Congo region ( $r_{\text{N}_2\text{OvsHCN}}^2 = 0.95$ ) and industrial production of N<sub>2</sub>O from oil and gas emissions from the Niger River Delta in Africa ( $r_{\text{N}_2\text{OvsSO}_2}^2 = 0.84$ ). An example is shown in Fig. 8 for 15 February 2017 (Fig. S12 and land contribution in Fig. S13).

To understand the origin of the enhancements in N<sub>2</sub>O, we calculated the enhancement expected in the atmosphere based on monthly mean estimates of anthropogenic emission from the Emissions Database for Global Atmospheric Research (EDGAR, <http://edgar.jrc.ec.europa.eu/>). We convolved the calculated surface influence (footprint) with the inventory to calculate the N<sub>2</sub>O enhancement expected for each receptor. We also calculated the contribution of each region and source sector to the overall enhancement. This allowed us to quantify the dominant sources for various layers within each profile. Each of the calculated enhancements were then compared to the enhancement in N<sub>2</sub>O observed for the profiles. The observed N<sub>2</sub>O enhancements were calculated relative to the NOAA MBL reference (Fig 8a, dashed red line) for each 10s observation, with background concentrations chosen for locations close to the origin of the airmass indicated by the surface influence (shown as dashed and dotted lined on the N<sub>2</sub>O altitude profiles in Figs 5 – 8). We also included a 0.4 ppb uncertainty for the observed enhancements based on our measurement precision.

In the Atlantic during ATom 2 (Feb 2017; Fig 9), the largest N<sub>2</sub>O enhancement is attributed to African agriculture (peaking at 2 ppb at 2 km), with smaller but significant influence from Asia and Europe (0.5 ppb each at 2-4 km, Fig. S14). The observed and modeled N<sub>2</sub>O enhancements agree within an order of magnitude for the profile, but the model underestimates the high altitude (4-7 km) N<sub>2</sub>O enhancement by <1 ppb and overestimates the lower altitude enhancement (2-4 km) by ~1 ppb. The difference in N<sub>2</sub>O enhancement could be due to a strong latitudinal gradient in N<sub>2</sub>O across this profile or the timing of emission of N<sub>2</sub>O sampled along this single profile compared to a monthly mean estimate from the inventory. Strong correlations between N<sub>2</sub>O and HCN ( $r^2 = 0.95$ ), CO and CH<sub>3</sub>CN suggest a source

of N<sub>2</sub>O from burning emissions also contribute to the N<sub>2</sub>O enhancement (Fig. 8 and Fig. S12). However, when we convolved the monthly mean fire contributions from the Global Fire Emissions Database (GFED, <https://www.globalfire.org>) with the surface influence footprints (as described above), we find that the wildfire produced N<sub>2</sub>O is minimal for this profile (~0.2 ppb), suggesting fires of anthropogenic or urban origin might be the source of that contribution (Figs. 8A–C, 9, S12 and S13).

## 5. Conclusions

N<sub>2</sub>O mixing ratios at 1Hz were obtained during the NASA ATom airborne program by applying a new spectral retrieval method to account for the pressure and temperature sensitivity of quantum cascade laser spectrometers when deployed on aircraft. This method improved the precision of our QCLS N<sub>2</sub>O measurements (based on the stdev. of calibration measurements) by a factor of 3, allowing us to provide N<sub>2</sub>O measurements to the level of precision shown in previous aircraft missions.

The N<sub>2</sub>O altitude profiles observed during ATom show that tropospheric N<sub>2</sub>O variability is strongly driven by the influence of stratospheric air depleted in N<sub>2</sub>O, especially at middle and high latitudes. At high latitudes, our profiles showed a strong depleted N<sub>2</sub>O signal around the time of the vortex break-up season, persisting for several months. Combining the information from N<sub>2</sub>O profiles and other chemical tracers such as CO, SF<sub>6</sub>, O<sub>3</sub> and CFC-12 we traced the propagation of stratospheric air along the tropospheric column down to the surface. This transport dominates the N<sub>2</sub>O seasonal cycle and creates the seasonal surface minima 2–3 months after the peak stratosphere-troposphere exchange in spring.

The high resolution of this data set (10 s) allowed us to study the factors influencing the enhancements in the N<sub>2</sub>O tropospheric mixing ratios associated with biomass burning, human activities such as urban and industrial emissions. The highest N<sub>2</sub>O mixing ratios are close to the Equator, extending through the tropospheric column. Episodes of the strongest N<sub>2</sub>O enhancements were observed close to the Equator and also in a number of locations at mid-latitudes. We use the information given by the vertical profiles of N<sub>2</sub>O and a variety of chemical tracers together with footprints computed every 60 s along the flight track to identify and trace the sources of these N<sub>2</sub>O enhancements. N<sub>2</sub>O enhancement events were more frequent in the Atlantic than in the Pacific.

Over the Atlantic, excess N<sub>2</sub>O together with the co-occurrence of other pollutants suggested that industrial and urban N<sub>2</sub>O emissions originated in distant locations such as western and southern Africa, the Middle East, Europe and South America may be significantly greater than the emissions from biomass burning in Africa. This view is supported by our observations of a strong contribution to N<sub>2</sub>O from oil and gas emissions from the Niger River Delta in Africa. Possibly the correlations observed between N<sub>2</sub>O and SO<sub>2</sub> ( $r^2 = 0.90$ ) could be used to estimate N<sub>2</sub>O emissions from oil and gas.

Over the southern Pacific Ocean and the tropical Atlantic Ocean, we observed a significant number (>12) of profiles where enhancements in N<sub>2</sub>O were associated with increased H<sub>2</sub>O<sub>2</sub> and PAA, and notably less well correlated with HCN or CO. Since H<sub>2</sub>O<sub>2</sub> and PAA are products of photochemical pollution, this observation raised the question as to whether significant N<sub>2</sub>O may be produced by heterogenous processes involving HONO or NO<sub>x</sub> reactions in acidic aerosols close to sources, or in very heavily polluted areas. It is hard to make a definite conclusion based on measurements so far from the most active regions. Studies directed at understanding this question would have to be carried out directly in the polluted areas. Because agricultural activities do not have unique tracer signatures, we were not able to distinguish contributions from cultivated and natural soils to N<sub>2</sub>O emissions from the ATom data. Previous airborne studies have observed these inputs, using flights in agricultural areas (Kort et al, 2008), and at towers in these regions (e.g., Nevison et al, 2017; Miller et al., 2008).

Our study shows that airborne campaigns such as ATom can help trace the origins of biomass burning and industrial emissions and investigate their impact on the variability of tropospheric N<sub>2</sub>O, providing unique signatures in vertical profiles and with covariate tracers. We hope that the information provided by the global tropospheric N<sub>2</sub>O profiles from the ATom mission will help better diagnose and reduce uncertainties of atmospheric chemical transport models for constraining the N<sub>2</sub>O global budget.

530

#### **Appendix A: List of frequently used symbols and acronyms**

	<i>Description</i>	<i>Acronym</i>
	Atmospheric Potential Oxygen	APO
535	Atmospheric Tomography	ATom
	California Institute of Technology - Chemical Ionization Mass Spectrometer	CIT-CIMS
	CU Aircraft High-Resolution Time-of-Flight Aerosol Mass Spectrometer	HR-AMS
	Global Monitoring Laboratory	GML
	HIAPER Pole-to-Pole Observations	HIPPO
540	High Latitudes	HL
	High-resolution TRANsmision molecular absorption database	HITRAN
	Marine Boundary Layer	MBL
	Middle Latitudes	ML
	Modern-Era Retrospective analysis for Research and Applications 2 model	MERRA2
545	National Center for Environmental Prediction Global Forecast System model	NCEP GFS
	National Oceanic and Atmospheric Administration	NOAA
	NCAR Airborne Oxygen Instrument	AO2
	NOAA Halocarbons and other Atmospheric Trace Species Flask Sampling Program	NOAA-HATS
	NOAA NOyO <sub>3</sub> 4-channel chemiluminescence	CL
550	Northern Hemisphere	NH
	PAN and other Trace Hydrohalo-carbon ExpeRiment	PANTHER
	Particle Analysis by Laser Mass Spectrometry instrument	PALMS
	Potential Vorticity	PV
	Principal Component Analysis	PCA



555	Programmable Flask Package Whole Air Sampler	PFP
	Quantum Cascade Laser Spectrometer	QCLS
	Southern Hemisphere	SH
	Stochastic Time-Inverted Lagrangian Transport Model	STILT
	Trace Organic Gas Analyzer	TOGA
560	Unmanned Aircraft Systems Chromatograph for Atmospheric Trace Species	UCATS
	Upper Troposphere/Lower Stratosphere	UT/LS
	World Meteorological Organization	WMO

565 *Data availability.* Data from the ATom mission can be found on the NASA ESPO archive (<https://espoarchive.nasa.gov>) at <https://espoarchive.nasa.gov/archive/browse/atom>, and in the ATom data repository at the NASA/ORNL DAAC: <https://doi.org/10.3334/ORNLDAAC/1581> (Wofsy et al., 2018).

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*Author Contribution.* Y. Gonzalez did the data analysis and wrote and revised the paper. S.C. Wofsy and R. Commane actively contributed to the design of the study and data analysis. J.B. MacManus designed the Neptune software for spectral re-analysis and contributed to the writing. R. Commane and B.C. Daube performed and analyzed QCLS measurements of CH<sub>4</sub>, N<sub>2</sub>O and CO and contributed to the discussions. E. Manninen and L. D. Schiferl contributed to the data analysis. K. McKain performed and analyzed NOAA Picarro measurements of CH<sub>4</sub>, CO and CO<sub>2</sub>. J.W. Elkins, E.J. Hints and F. Moore performed and analyzed N<sub>2</sub>O, SF<sub>6</sub> and CFC-12 measurements from PANTHER and UCATS instruments. F. Moore, S. Montzka and C. Sweeney performed and analyzed N<sub>2</sub>O measurements with the Programmable Flask Package Whole Air Sampler (PFP). P. O. Wennberg, J. Crouse, M. Kim and H. M. Allen performed and analyzed CIT-CIMS measurements of HCN and SO<sub>2</sub> shown here. K. Froyd performed and analyzed PALMS measurements. J.L. Jimenez, P. Campuzano-Jost and B. A. Nault performed and analyzed HR-AMS spectrometer of a variety of aerosols. E. Ray provided backtrajectories for each minute during the flight tracks and P. Newman provided the GEOS5 FP meteorological products. T.B. Ryerson, I. Bourgeois, J. Peischl and C. R. Thompson performed and analyzed NO<sub>y</sub>O<sub>3</sub> measurements of NO<sub>y</sub> and O<sub>3</sub>. B. B. Stephens and E. J. Morgan performed and analyzed AO<sub>2</sub> and the Medusa Whole Air Sampler measurements of O<sub>2</sub>/N<sub>2</sub> and CO<sub>2</sub> and assisted with the interpretation. E. C. Apel and R. S. Hornbrook performed and analyzed TOGA measurements of volatile organic compounds. All coauthors provided comments on the paper.

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## References

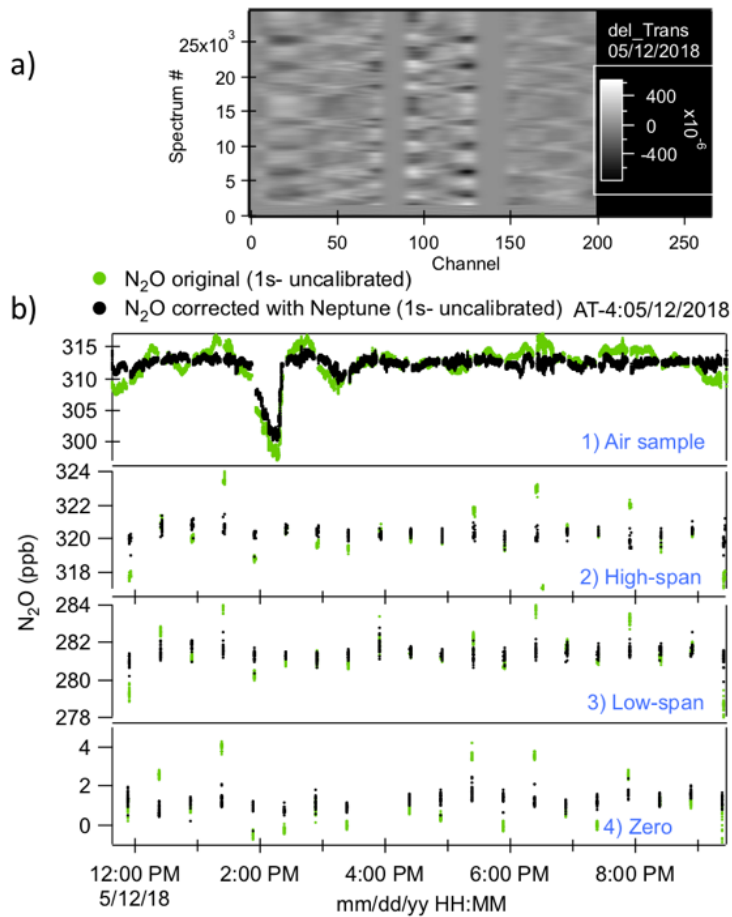
- Albanito, F., Lebender, U., Cornulier, T., Sapkota, T. B., Brentrup, F., Stirling, C., and Hillier, J.: Direct Nitrous Oxide  
Emissions from Tropical and Sub-Tropical Agricultural Systems - A Review and Modelling Of Emission Factors,  
610 *Sci. Rep.*, 7, 44235, doi: 10.1038/srep44235, 2017.
- Apel, E., Asher, E. C., Hills, A. J. and Hornbrook, R. S.: ATom: L2 Volatile Organic Compounds (VOCs) from the  
Trace Organic Gas Analyzer (TOGA), ORNL DAAC, <https://doi.org/10.3334/ORNLDAAC/1749>, 2019.
- Asher, E., Hornbrook, R. S., Stephens, B. B., Kinnison, D., Morgan, E. J., Keeling, R. F., Atlas, E. L., Schauffler, S.  
M., Tilmes, S., Kort, E. A., Hoecker-Martínez, M. S., Long, M. C., Lamarque, J.-F., Saiz-Lopez, A., McKain, K.,  
615 Sweeney, C., Hills, A. J., and Apel, E. C.: Novel approaches to improve estimates of short-lived halocarbon  
emissions during summer from the Southern Ocean using airborne observations, *Atmos. Chem. Phys.*, 19, 14071–  
14090, <https://doi.org/10.5194/acp-19-14071-2019>, 2019.
- Bourgeois I., Peischl, J., Thompson, C. R., Aikin, K. C., Campos, T., Clark, H., Commane, R., Daube, B., Diskin, G.  
W., Elkins, J. W., Gao, R.-S., Gaudell, A., Hints, E. J., Johnson, B. J., Kivi, R., McKain, K., Moore, F. L., Parrish,  
620 D. D., Querel, R., Ray, E., Sánchez, R., Sweeney, C., Tarasick, D. W., Thompson, A. M., Thouret, V., Witte, J. C.,  
Wofsy, S. C. and Ryerson, T. B.: Global-scale distribution of ozone in the remote troposphere from ATom and  
HIPPO airborne field missions, *Atmos. Chem. Phys.*, 20, 10611–10635, [https://doi.org/10.5194/acp-20-10611-  
2020](https://doi.org/10.5194/acp-20-10611-2020), 2020.
- Bowman, K. P.: Large-scale isentropic mixing properties of the Antarctic polar vortex from analyzed winds, *J. Geophys.*  
625 *Res. Atmos.*, 98(D12), 23, 013, <https://doi.org/10.1029/93JD02599>, 1993.
- Brock, C. A., Williamson, C., Kupc, A., Froyd, K. D., Erdesz, F., Wagner, N., Richardson, M., Schwarz, J. P., Gao, R.-  
S., Katic, J. M., Campuzano-Jost, P., Nault, B. A., Schroder, J. C., Jimenez, J. L., Weinzierl, B., Dollner, M., Bui,  
T., and Murphy, D. M.: Aerosol size distributions during the Atmospheric Tomography Mission (ATom): methods,  
uncertainties, and data products, *Atmos. Meas. Tech.*, 12, 3081–3099, <https://doi.org/10.5194/amt-12-3081-2019>,  
630 2019.
- Butterbach-Bahl, K., Baggs, E. M., Dannenmann, M., Kiese, R., and Zechmeister-Boltenstern, S.: Nitrous oxide  
emissions from soils: how well do we understand the processes and their controls? *Phil. Trans. R. Soc. B*, 368,  
20130122, <https://doi.org/10.1098/rstb.2013.0122>, 2013.

- 635 Canagaratna, M. R., Jayne, J. T., Jimenez, J. L., Allan, J. D., Alfarra, M. R., Zhang, Q., Onasch, T. B., Drewnick, F., Coe, H., Middlebrook, A., Delia, A., Williams, L. R., Trimborn, A. M., Northway, M. J., Decarlo, P. F., Kolb, C. E., Davidovits, P. and Worsnop, D. R.: Chemical and microphysical characterization of ambient aerosols with the Aerodyne Aerosol Mass Spectrometer, *Mass Spectrom. Rev.*, 26(2), 185–222, 2007.
- 640 Castaldi, S., de Grandcourt, A., Rasile, A., Skiba, U., and Valentini, R.: CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O fluxes from soil of a burned grassland in Central Africa, *Biogeosciences*, 7, 3459–3471, <https://doi.org/10.5194/bg-7-3459-2010>, 2010.
- Castaldi, S., Ermice, A., Strumia, S.: Fluxes of N<sub>2</sub>O and CH<sub>4</sub> from soils of savannas and seasonally-dry ecosystems, *J. Biogeogr.*, 33, 401–415, <https://doi.org/10.1111/j.1365-2699.2005.01447.x>, 2006.
- Crouse, J. D., McKinney, K. A., Kwan, A. J. and Wennberg, P. O.: Measurement of Gas-Phase Hydroperoxides by Chemical Ionization Mass Spectrometry, *Anal. Chem.*, 78, 19, 6726–6732, <https://doi.org/10.1021/ac0604235>, 2006.
- 645 Cuevas, E., González, Y., Rodríguez, S., Guerra, J. C., Gomez-Peláez, A. J., Alonso-Pérez, S., Bustos, J., and Milford, C.: Assessment of atmospheric processes driving ozone variations in the subtropical North Atlantic free troposphere, *Atmos. Chem. Phys.*, 13, 1973–1998, <https://doi.org/10.5194/acp-13-1973-2013>, 2013.
- DeCarlo, P. F., Kimmel, J. R., Trimborn, A., Northway, M. J., Jayne, J. T., Aiken, A. C., Gonin, M., Fuhrer, K., Horvath, T., Docherty, K. S., Worsnop, D. R. and Jimenez, J. L.: Field-deployable, high-resolution, time-of-flight aerosol mass spectrometer, *Anal. Chem.*, 78(24), 8281–8289, 2006.
- 650 Forster, P., Ramaswamy, V., Artaxo, P., Berntsen, T., Betts, R., Fahey, D. W., Haywood, J., Lean, J., Lowe, D. C., Myhre, G., Nganga, J., Prinn, R., Raga, G., Schulz, M., and Van Dorland, R.: Changes in Atmospheric Constituents and in Radiative Forcing, in: *Climate Change 2007: The Physical Science Basis. Contribution of Working Group I to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change*, Cambridge University Press Cambridge, UK and New York, NY, USA, 2007.
- 655 Froyd, K., Murphy, D., Sanford, T. J., Thomson, D. S., Wilson, J., Pfister, L. and Lait, L. R.: Aerosol composition of the tropical upper troposphere, *Atmos. Chem. Phys.*, 9, 4363–4385, <https://doi.org/10.5194/acp-9-4363-2009>, 2009.
- Ganesan, A. L., Manizza, M., Morgan, E. J., Harth, C. M., Kozlova, E., Lueker, T., Manning, A. J., Lunt, M.F., Mühle, J., Lavric, J. V., Heimann, M., Weiss, R. F. and Rigby, M.: Marine Nitrous Oxide Emissions From Three Eastern Boundary Upwelling Systems Inferred From Atmospheric Observations, *Geophysical Research Letters*, 47, e2020GL087822, <https://doi.org/10.1029/2020GL087822>, 2020.
- 660 Guo, H., Campuzano-Jost, P., Nault, B. A., Day, D. A., Schroder, J. C., Dibb, J. E., Dollner, M., Weinzierl, B. and Jimenez, J. L.: The Importance of Size Ranges in Aerosol Instrument Intercomparisons: A Case Study for the ATom Mission, *Atmos. Meas. Tech. Discuss.*, doi:10.5194/amt-2020-224, 2020.
- 665 Hintsä, E., Boering, K. A., Weinstock, E. M., Anderson, J. G., Gary, B. L., Pfister, L., Daube, B. C., Wofsy, S. C., Loewenstein, M., Podolske, J. R., Margitan, J. J., and Bu, T. T.: Troposphere-to-stratosphere transport in the lowermost stratosphere from measurements of H<sub>2</sub>O, CO, N<sub>2</sub>O and O<sub>3</sub>, *Geophys. Res. Lett.*, 25, 14, 2655–2658, <https://doi.org/10.1029/98GL01797>, 1998.
- 670 Hintsä, E. J., Moore, F. L., Hurst, D. F., Dutton, G. S., Hall, B. D., Nance, J. D., Miller, B. R., Montzka, S. A., Wolton, L. P., McClure-Begley, A., Elkins, J. W., Hall, E. G., Jordan, A. F., Rollins, A. W., Thornberry, T. D., Watts, L. A., Thompson, C. R., Peischl, J., Bourgeois, I., Ryerson, T. B., Daube, B. C., Pittman, J. V., Wofsy, S. C., Kort, E., Diskin, G. S., and Bui, T. P.: UAS Chromatograph for Atmospheric Trace Species (UCATS) – a versatile instrument for trace gas measurements on airborne platforms, *Atmos. Meas. Tech. Discuss.*, <https://doi.org/10.5194/amt-2020-496>, in review, 2021.
- 675 Hodzic, A., Campuzano-Jost, P., Bian, H., Chin, M., Colarco, P. R., Day, D. A., Froyd, K. D., Heinold, B., Jo, D. S., Katich, J. M., Kodros, J. K., Nault, B. A., Pierce, J. R., Ray, E., Schacht, J., Schill, G. P., Schroder, J. C., Schwarz, J. P., Sueper, D. T., Tegen, I., Tilmes, S., Tsigaridis, K., Yu, P. and Jimenez, J. L.: Characterization of organic aerosol across the global remote troposphere: a comparison of ATom measurements and global chemistry models, *Atmos. Chem. Phys.*, 20(8), 4607–4635, 2020.
- 680 Hu, K., Lu, R., Wang, D.: Seasonal climatology of cut-off lows and associated precipitation patterns over Northeast China, *Meteorol. Atmos. Phys.*, 106:37–48, <https://doi.org/10.1007/s00703-009-0049-0>, 2010.

- Jimenez, J. L., Campuzano-Jost, P., Day, D. A., Nault, B. A., Price, D. J. and Schroder, J. C.: ATom: L2 Measurements from CU High-Resolution Aerosol Mass Spectrometer (HR-AMS), , doi:10.3334/ORN LDAAC/1716, 2019.
- 685 Jiménez, R., Herndon, S., Shorter, J. H., Nelson, D. D., McManus, J. B., and Zahniser, M. S.: Atmospheric trace gas measurements using a dual quantum-cascade laser mid-infrared absorption spectrometer, *Proc. SPIE*, 5738, 318–331, <https://doi.org/10.1117/12.597130>, 2005.
- 690 Jiménez, R., Park, S., Daube, B. C., McManus, J. B., Nelson, D. D., Zahniser, M. S., and Wofsy, S. C.: A new quantum-cascade laser-based spectrometer for high-precision airborne CO<sub>2</sub> measurements, 13<sup>th</sup> WMO/IAEA Meeting of Experts on Carbon Dioxide Concentration and Related Tracers Measurement Techniques, WMO/TD-No. 1359; GAW Report- No. 168, 100–105, 2006.
- Kort, E. A., Eluszkiewicz, J., Stephens, B. B., Miller, J. B., Gerbig, C., Nehr Korn, T., Daube, B. C., Kaplan, J. O., Houweling, S., and Wofsy, S. C.: Emissions of CH<sub>4</sub> and N<sub>2</sub>O over the United States and Canada based on a receptor-oriented modeling framework and COBRA-NA atmospheric observations, *Geophys. Res. Letters*, vol. 35, L18808, doi:10.1029/2008GL034031, 2008.
- 695 Kort, E. A., Andrews, A. E., Dlugokencky, E., Sweeney, C., Hirsch, A., Eluszkiewicz, J., Nehr Korn, T., Michalak, A., Stephens, B., Gerbig, C., Miller, J. B., Kaplan, J., Houweling, S., Daube, B. C., Tans, P. and Wofsy, S. C.: Atmospheric constraints on 2004 emissions of methane and nitrous oxide in North America from atmospheric measurements and a receptor-oriented modeling framework, *J. Integr. Environ. Sci.*, 7:S1, 125-133, <https://doi.org/10.1080/19438151003767483>, 2010.
- 700 Kort, E. A., Patra, P. K., Ishijima, K., Daube, B. C., Jiménez, R., Elkins, J., Hurst, D., Moore, F. L., Sweeney, C. and Wofsy, S. C.: Tropospheric distribution and variability of N<sub>2</sub>O: Evidence for strong tropical emissions, *Geophys. Res. Lett.*, 38, L15806, <https://doi.org/10.1029/2011GL047612>, 2011.
- 705 Krause, J., Hoor, P., Engel, A., Plöger, F., Groöß, J.-U., Bönisch, H., Keber, T., Sinnhuber, B.-M., Woiwode, W. and Oelhaf, H.: Mixing and ageing in the polar lower stratosphere in winter 2015–2016, *Atmos. Chem. Phys.*, 18, 6057–6073, <https://doi.org/10.5194/acp-18-6057-2018>, 2018.
- Lin, J. C., Gerbig, C., Wofsy, S. C., Andrews, A. E., Daube, B. C., Davis, K. J. and Grainger, C. A.: A near-field tool for simulating the upstream influence of atmospheric observations: The Stochastic Time- Inverted Lagrangian Transport (STILT) model, *J. Geophys. Res. Atmos.*, 108(D16), 4493, <https://doi.org/10.1029/2002JD003161>, 2003.
- 710 Liu, Y., Yvon-Lewis, S. A., Thornton, D. C. O., Butler, J. H., Bianchi, T. S., Campbell, L., Hu, L. and Smith, R. W.: Spatial and temporal distributions of bromoform and dibromomethane in the Atlantic Ocean and their relationship with photosynthetic biomass, *J. Geophys. Res. Oceans*, vol. 118, 3950–3965, <https://doi.org/10.1002/jgrc.20299>, 2013.
- 715 Lueker, T. J., Keeling, R. F. and Dubey, M. K.: The oxygen to Carbon Dioxide Ratios observed in Emissions from a Wildfire in the Northern California, *Geophysical Research Letters*, vol. 28, no. 12, pages 2413-2416, doi: 10.1029/2000GL011860, 2001.
- Lueker, T. J., Walker, S. J., Vollmer, M. K., Keeling, R. F., Nevison, C. D., Weiss, R. F. and Garcia, H. E.: Coastal upwelling air-sea fluxes revealed in atmospheric observations of O<sub>2</sub>/N<sub>2</sub>, CO<sub>2</sub> and N<sub>2</sub>O, *Geophys. Res. Lett.*, vol. 30, no. 6, 1292, <https://doi.org/10.1029/2002GL016615>, 2003.
- 720 Lueker, T. J.: Coastal upwelling fluxes of O<sub>2</sub>, N<sub>2</sub>O, and CO<sub>2</sub> assessed from continuous atmospheric observations at Trinidad, California, *Biogeosciences*, 1, 101–111, <https://doi.org/10.5194/bg-1-101-2004>, 2004.
- Montzka, S., Moore, F. and Sweeney, C.: ATom: L2 Measurements from the Programmable Flask Package (PFP) Whole Air Sampler, ORNL DAAC, doi:10.3334/ORN LDAAC/1746, 2019.
- 725 Moore, F., Dutton, G., Elkins, J. W., Hall, B., Hurst, D., Nance, J. D., and Thompson, T: PANTHER Data from SOLVE-II Through CR-AVE: A Contrast Between Long- and Short-Lived Compounds, American Geophysical Union, Fall Meeting 2006, abstract #A41A-0025, 2006.
- Nevison, C. D., Dlugokencky, E., Dutton, G., Elkins, J. W., Fraser, P., Hall, B., Krummel, P. B., Langenfelds, R. L., O'Doherty, S., Prinn, R. G., Steele, L. P., and Weiss, R. F.: Exploring causes of interannual variability in the seasonal cycles of tropospheric nitrous oxide, *Atmos. Chem. Phys.*, 11, 3713–3730, <https://doi.org/10.5194/acp-11-3713-2011>, 2011.
- 730

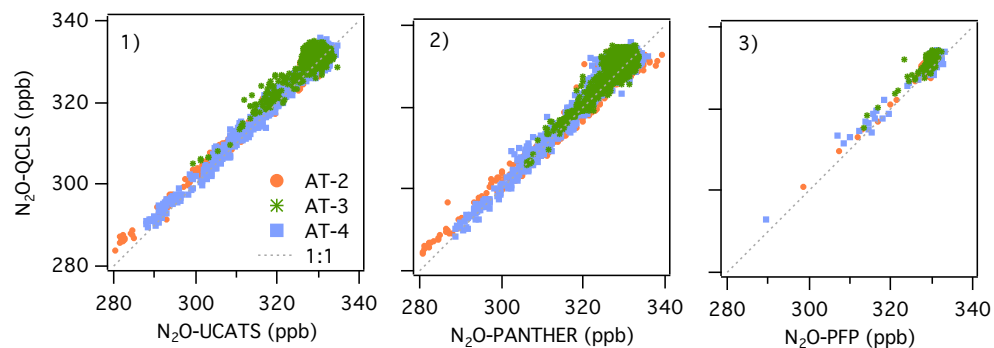
- Nevison, C.D., Keeling, R. F., Weiss, R. F., Popp, B. N., Jin, X., Fraser, P. J., Porter, L. W., and Hess, P. G.: Southern Ocean ventilation inferred from seasonal cycles of atmospheric N<sub>2</sub>O and O<sub>2</sub>/N<sub>2</sub> at Cape Grim, Tasmania, *Tellus*, 57B, 218–229, 2005.
- 735 Nevison, C.D., Weiss, R. F. and Erickson III, D. J.: Global oceanic emissions of nitrous oxide, *J. Geophys. Res. Oceans*, 100, C8, 5809-15820, <https://doi.org/10.1029/95JC00684>, 1995.
- Pires M. and Rossi, M. J.: The Heterogeneous Formation of N<sub>2</sub>O in the Presence of Acidic Solutions: Experiments and Modeling, *International Journal of Chemical Kinetics*, Vol. 29, 869–891, 1997.
- 740 Randerson, J. T., van der Werf, G. R., Giglio, L., Collatz, G. J. and Kasibhatla, P. S.: Global Fire Emissions Database, Version 4.1 (GFEDv4). ORNL DAAC, Oak Ridge, Tennessee, USA, <https://doi.org/10.3334/ORNLDAAAC/1293>, 2018
- Ravishankara, A. R., Daniel, J. S. and Portmann, R. W.: Nitrous oxide (N<sub>2</sub>O): the dominant ozone-depleting substance emitted in the 21st century., *Science*, 326(5949), 123–5, <https://doi.org/10.1126/science.1176985>, 2009.
- 745 Rothman, L.S., Jacquemart, D., Barbe, A., Benner, D. C., Birk, M., Brown, L.R., Carleer, M.R., Chackerian Jr., C., Chance, K., Couders, L.H., Dana, V., Devi, V.M., Flaud, J.-M., Gamache, R.R., Goldman, A., Hartmann, J.-M., Jucks, K.W., Maki, A.G., Mandin, J.-Y., Massie, S.T., Orphal, J., Perrin, A., Rinsland, C.P., Smith, M.A.H., Tennyson, J., Tolchenov, R.N., Toth, R.A., Vander Auwera, J., Varanasi, P., Wagner, G.: The HITRAN 2004 molecular spectroscopic database, *Journal of Quantitative Spectroscopy & Radiative Transfer* 96, 139–204, 2005.
- 750 Ryerson, T. B., Thompson, C., Peischl, J. and Bourgeois, I.: ATom: L2 In Situ Measurements from NOAA Nitrogen Oxides and Ozone (NOyO3) Instrument, ORNL DAAC, 2019, <https://doi.org/10.3334/ORNLDAAAC/1734>, 2019.
- Saikawa, E., Prinn, R. G., Dlugokencky, E., Ishijima, K., Dutton, G. S., Hall, B. D., Langenfelds, R., Tohjima, Y., Machida, T., Manizza, M., Rigby, M., O'Doherty, S., Patra, P. K., Harth, C. M., Weiss, R. F., Krummel, P. B., van der Schoot, M., Fraser, P. J., Steele, L. P., Aoki, S., Nakazawa, T., and Elkins, J. W. : Global and regional emissions estimates for N<sub>2</sub>O, *Atmos. Chem. Phys.*, 14, 4617–4641, <https://doi.org/10.5194/acp-14-4617-2014>, 2014.
- 755 Santoni, G. W., Daube, B. C., Kort, E. A., Jiménez, R., Park, S., Pittman, J. V., Gottlieb, E., Xiang, B., Zahniser, M. S., Nelson, D. D., McManus, J. B., Peischl, J., Ryerson, T. B., Holloway, J. S., Andrews, A. E., Sweeney, C., Hall, B., Hintsa, E. J., Moore, F. L., Elkins, J. W., Hurst, D. F., Stephens, B. B., Bent, J., and Wofsy, S. C.: Evaluation of the airborne quantum cascade laser spectrometer (QCLS) measurements of the carbon and greenhouse gas suite – CO<sub>2</sub>, CH<sub>4</sub>, N<sub>2</sub>O, and CO – during the CalNex and HIPPO campaigns, *Atmos. Meas. Tech.*, 7, 1509-1526, <https://doi.org/10.5194/amt-7-1509-2014>, 2014.
- 760 St. Clair, J.M., McCabe, D. C., Crouse, J. D., Steiner, U. and Wennberg, P. O.: Chemical ionization tandem mass spectrometer for the *in situ* measurement of methyl hydrogen peroxide, *Rev. Sci. Instrum.*, 81, 094102, <https://doi.org/10.1063/1.3480552>, 2010.
- 765 SPARC Report on the Lifetimes of Stratospheric Ozone-Depleting Substances, Their Replacements, and Related Species, M. Ko, P. Newman, S. Reimann, S. Strahan (Eds.), SPARC Report No. 6, WCRP-15/2013.
- Stephens, B. B., Keeling, R. F., Heimann, M., Six, K. D., Murnane, R. and Caldeira, K.: Testing global ocean carbon cycle models using measurements of atmospheric O<sub>2</sub> and CO<sub>2</sub> concentration, *Global Biogeochemical cycles*, vol. 12, no. 12, pages 213-230, 1998.
- 770 Syakila, A. and Kroeze, C.: The global nitrous oxide budget revisited, *Greenhouse Gas Measurement and Management*, 1:1, 17-26, <https://doi.org/10.3763/ghgmm.2010.0007>, 2011.
- 775 Thompson, R. L., Ishijima, K., Saikawa, E., Corazza, M., Karstens, U., Patra, P. K., Bergamaschi, P., Chevallier, F., Dlugokencky, E., Prinn, R. G., Weiss, R. F., O'Doherty, S., Fraser, P. J., Steele, L. P., Krummel, Vermeulen, A., Tohjima, Y., Jordan, A., Haszpra, L., Steinbacher, M., Van der Laan, S., Aalto, T., Meinhardt, F., Popa, M. E., Moncrieff, J., and Bousquet, P.: TransCom N<sub>2</sub>O model inter-comparison – Part 2: Atmospheric inversion estimates of N<sub>2</sub>O emissions, *Atmos. Chem. Phys.*, 14, 6177–6194, <https://doi.org/10.5194/acp-14-6177-2014>, 2014b.
- 780 Thompson, R. L., Patra, P. K., Ishijima, K., Saikawa, E., Corazza, M., Karstens, U., Wilson, C., Bergamaschi, P., Dlugokencky, E., Sweeney, C., Prinn, R. G., Weiss, R. F., O'Doherty, S., Fraser, P. J., Steele, L. P., Krummel, P. B., Saunio, M., Chipperfield, M., and Bousquet, P.: TransCom N<sub>2</sub>O model inter-comparison – Part 1: Assessing the influence of transport and surface fluxes on tropospheric N<sub>2</sub>O variability, *Atmos. Chem. Phys.*, 14, 4349–4368, <https://doi.org/10.5194/acp-14-4349-2014>, 2014a.

- Tian, H., Lu, C., Chen G., Tao, B., Pan, S., Del Grosso, S. J., Xu, X., Bruhwiler, L., Wofsy, S.C., Kort, E. A. and Prior, S. A.: Contemporary and projected biogenic fluxes of methane and nitrous oxide in North American terrestrial ecosystems, *Front Ecol Environ* 2012; 10(10): 528–536, <https://doi.org/10.1890/120057>, 2012.
- 785 Tian, H., Xu, R., Canadell, J.G. et al.: A comprehensive quantification of global nitrous oxide sources and sinks, *Nature*, 586, 248–256, <https://doi.org/10.1038/s41586-020-2780-0>, 2020.
- Upstill-Goddard, R. C., Salter, M. E., Mann, P. J., Barnes, J., Poulsen, J., Dinga, B., Fiske, G. J., and Holmes, R. M.: The riverine source of tropospheric CH<sub>4</sub> and N<sub>2</sub>O from the Republic of Congo, Western Congo Basin, *Biogeosciences*, 14, 2267–2281, <https://doi.org/10.5194/bg-14-2267-2017>, 2017.
- 790 Valentini, R., Arneth, A., Bombelli, A., Castaldi, S., Cazzolla Gatti, R., Chevallier, F., Ciais, P., Grieco, E., Hartmann, J., Henry, M., Houghton, R. A., Jung, M., Kutsch, W. L., Malhi, Y., Mayorga, E., Merbold, L., Murray-Tortarolo, G., Papale, D., Peylin, P., Poulter, B., Raymond, P. A., Santini, M., Sitch, S., Vaglio Laurin, G., van der Werf, G. R., Williams, C. A., and Scholes, R. J.: A full greenhouse gases budget of Africa: synthesis, uncertainties, and vulnerabilities, *Biogeosciences*, 11, 381–407, doi:10.5194/bg-11-381-2014, 2014.
- 795 Wang, J., Li, J., Ye, J. et al: Fast sulfate formation from oxidation of SO<sub>2</sub> by NO<sub>2</sub> and HONO observed in Beijing haze. *Nat. Commun.*, 11, 2844, <https://doi.org/10.1038/s41467-020-16683-x>, 2020.
- WMO: WMO Greenhouse Gas Bulletin., 2018.
- Wofsy, S. C., Afshar, S., Allen, H. M., Apel, E., Asher, E. C., Barletta, B., Bent, J., Bian, H., Biggs, B. C., Blake, D. R., Blake, N., Bourgeois, I., Brock, C. A., Brune, W. H., Budney, J. W., Bui, T. P., Butler, A., Campuzano-Jost, P., Chang, C. S., Chin, M., Commane, R., Correa, G., Crounse, J. D., Cullis, P. D., Daube, B. C., Day, D. A., Dean-Day, J. M., Dibb, J. E., DiGangi, J. P., Diskin, G. S., Dollner, M., Elkins, J. W., Erdesz, F., Fiore, A. M., Flynn, C. M., Froyd, K., Gesler, D. W., Hall, S. R., Hanisco, T. F., Hannun, R. A., Hills, A. J., Hints, E. J., Hoffman, A., Hornbrook, R. S., Huey, L. G., Hughes, S., Jimenez, J. L., Johnson, B. J., Katich, J. M., Keeling, R. F., Kim, M. J., Kupc, A., Lait, L. R., Lamarque, J.-F., Liu, J., McKain, K., McLaughlin, R. J., Meinardi, S., Miller, D. O., 800 Montzka, S. A., Moore, F. L., Morgan, E. J., Murphy, D. M., Murray, L. T., Nault, B. A., Neuman, J. A., Newman, P. A., Nicely, J. M., Pan, X., Paplawsky, W., Peischl, J., Prather, M. J., Price, D. J., Ray, E., Reeves, J. M., Richardson, M., Rollins, A. W., Rosenlof, K. H., Ryerson, T. B., Scheuer, E., Schill, G. P., Schroder, J. C., Schwarz, J. P., St. Clair, J. M., Steenrod, S. D., Stephens, B. B., Strode, S. A., Sweeney, C., Tanner, D., Teng, A. P., Thames, A. B., Thompson, C. R., Ullmann, K., Veres, P. R., Vieznor, N., Wagner, N. L., Watt, A., Weber, R., 805 Weinzierl, B., Wennberg, P., Williamson, C. J., Wilson, J. C., Wolfe, G. M., Woods, C. T., and Zeng, L. H.: ATom: Merged Atmospheric Chemistry, Trace Gases, and Aerosols, ORNL DAAC, Oak Ridge, Tennessee, USA, <https://doi.org/10.3334/ORNLDAAAC/1581>, 2018.
- Wofsy, S. C., the HIPPO Science Team, and Cooperating Modelers and Satellite Teams: HIAPER Pole-to-Pole Observations (HIPPO): Fine grained, global scale measurements for determining rates for transport, surface 815 emissions, and removal of climatically important atmospheric gases and aerosols, *Philos. T. Roy. Soc. A*, 369, 2073–2086, <https://doi.org/10.1098/rsta.2010.0313>, 2011.
- Xiang, B., Miller, S. M., Kort, E. A., Santoni, G. W., Daube, B. C., Commane, R., Angevine, W. M., Ryerson, T. B., Trainer, M. K., Andrews, A. E., Nehr Korn, T., Tian, H., and Wofsy, S. C.: Nitrous oxide (N<sub>2</sub>O) emissions from California based on 2010 CalNex airborne measurements, *J. Geophys. Res. Atmos.*, 118, 2809–2820, 820 <https://doi.org/10.1002/jgrd.50189>, 2013.
- Yang, Simon, Chang, B. X., Warner, M. J., Weber, T. S., Bourbonnais, A. M., Santoro, A. E., Kock, A., Sonnerup, R. E., Bullister, J. L. Wilson, S. T. and Bianchi, D.: Global reconstruction reduces the uncertainty of oceanic nitrous oxide emissions and reveals a vigorous seasonal cycle, *P. Natl. Acad. Sci. USA*, 117 (22), 11954–11960; 825 <https://doi.org/10.1073/pnas.1921914117>, 2020.

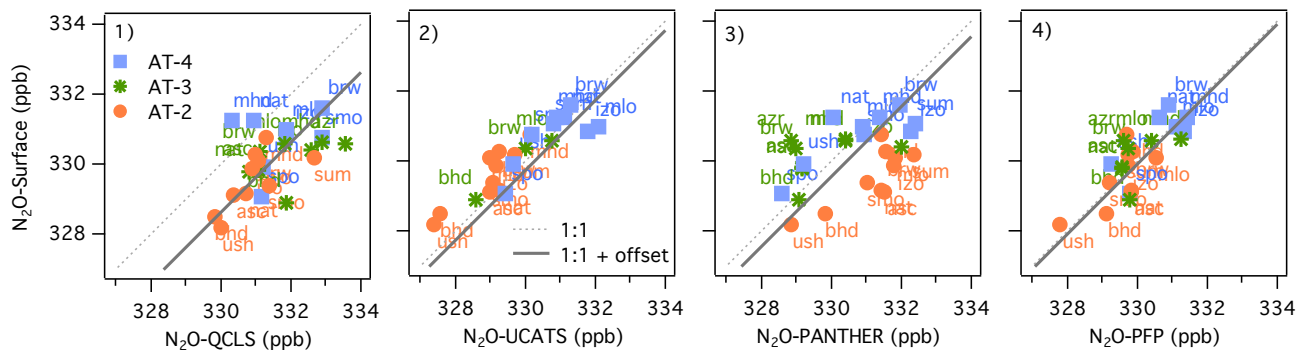


830 Figure 1: (a) A processed spectral array from the ATom-4 flight on 12 May 2018. “Channel” represents a point number in the spectra. Spectra have been grouped by type (i.e., calibration, ambient), with averages subtracted, absorption lines zeroed out (near channels 75, 140, and 225), and smoothed to close to the linewidth. Shifts in fringe phases during altitude changes are apparent. (b) Time series of ambient air samples, high-span, low-span, and zeros for the same flight as (a). Green dots are the original N<sub>2</sub>O data record. Black dots are the N<sub>2</sub>O data corrected with Neptune (no calibration applied at this point).

(a) Aircraft Instrument Comparison



(b) Aircraft Instrument vs Surface Comparison



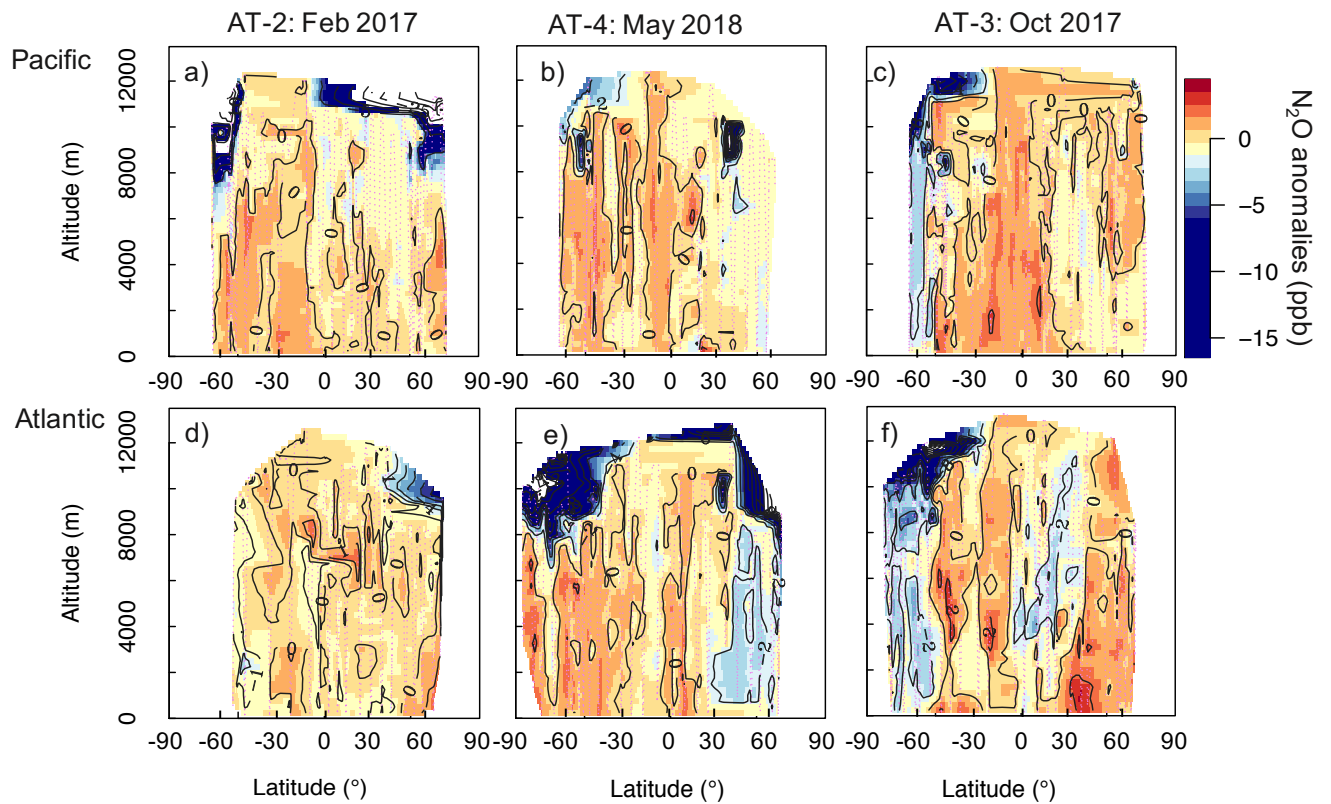
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**Figure 2: (a) Comparisons between Neptune-corrected QCLS  $N_2O$  and (1) UCATS  $N_2O$ , (2) PANTHER  $N_2O$ , and (3) PFP  $N_2O$  for ATom-2 (orange circles), ATom-3 (green stars), and -4 (blue squares). We used the 10s averaged merged file to compare QCLS, UCATS and PANTHER data. The PFP flask samples have a longer sampling time (30s to few minutes). The 1:1 line is shown as a dashed line. (b) Comparisons between NOAA  $N_2O$  surface flask measurements and Neptune-corrected and airborne data from (1) QCLS  $N_2O$ , (2) UCATS  $N_2O$ , (3) PANTHER  $N_2O$ , (4) and PFP  $N_2O$  for ATom-2, -3, and -4, similar to A1–A3. The solid line shows the 1:1 relationship + offset. For B1–B4 plots, the airborne data are the mean  $N_2O$  values within  $\pm 5^\circ$  latitude of each surface station and between 1 and 4 km.**

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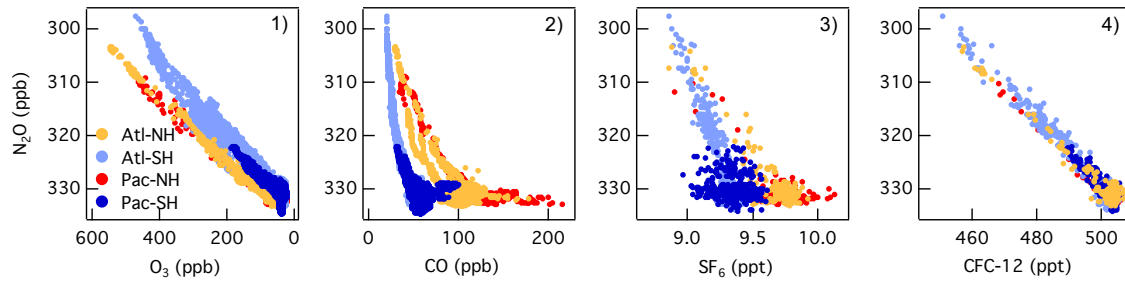




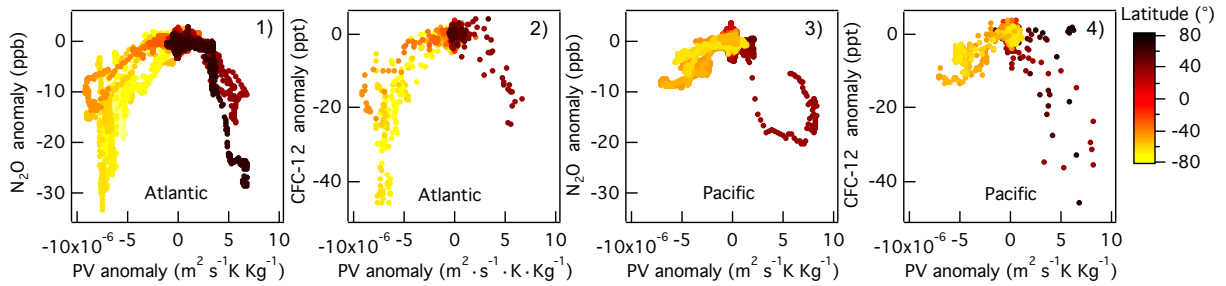
850 **Figure 3: Cross-sections of N<sub>2</sub>O anomalies (ppb), representing the differences between the airborne N<sub>2</sub>O (10-s resolution) and the surface N<sub>2</sub>O mixing ratios interpolated to 0.25° latitude and 250 m altitude for each deployment. Shown are the N<sub>2</sub>O anomalies over (a)–(c) the Pacific and (d)–(f) the Atlantic, and each column represents a deployment (ordered by season, ATom-2, -4 and -3). The color-scale ranges from -15 to 5 ppb. Values between -50 and -15 ppb, observed at the highest altitudes (>10 km) are shown in white for a better visualization of small changes in positive anomalies. Lilac dashed lines represent the flight tracks. Black contours are areas of the N<sub>2</sub>O anomalies.**

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(a) Tracer-Tracer correlations in the Northern Hemisphere Middle and High latitudes during Spring (AT-4)



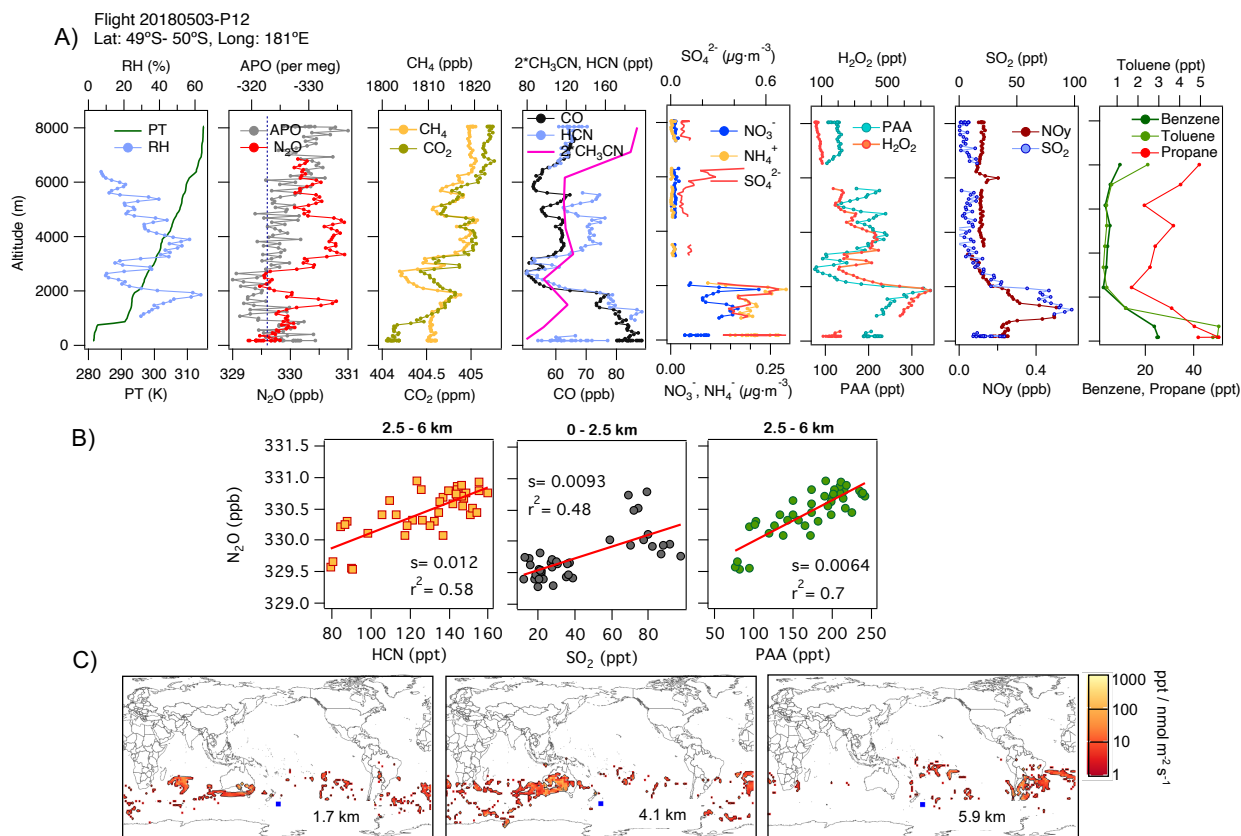
(b) Pot. Vorticity vs  $N_2O$  and CFC-12 anomalies (AT-4)



860 **Figure 4: (a) Correlations between  $N_2O$  and  $O_3$  (a1),  $CO$  (a2),  $SF_6$  (a3), and  $CFC-12$  (a4) at mid and high latitudes ( $30^\circ-85^\circ N$ ) during Northern Hemisphere spring (ATom-4). The data are colored as a function of the ocean basin and hemisphere: Pacific North Mid-High Latitudes (Pac-NH,  $>30^\circ N$ ) in red, Pacific South Mid-High Latitudes (Pac-SH,  $<30^\circ S$ ) in dark blue, Atlantic South Mid-High Latitudes (Atl-SH,  $<30^\circ S$ ) in light blue and Atlantic North Mid-High Latitudes (Atl-NH,  $>30^\circ N$ ) in orange. Note that the  $N_2O$  and  $O_3$  axes are reversed. (b) Correlations between anomalies in potential vorticity relative to its mean latitudinal distribution in the free troposphere (2–8 km) and anomalies in  $N_2O$  (b1, b3) and  $CFC-12$  (b2, b4) as a function of latitude during spring (ATom-4) over the Pacific and Atlantic basins. Mid-latitudes are shown in orange in the SH and clear brown in the NH.**

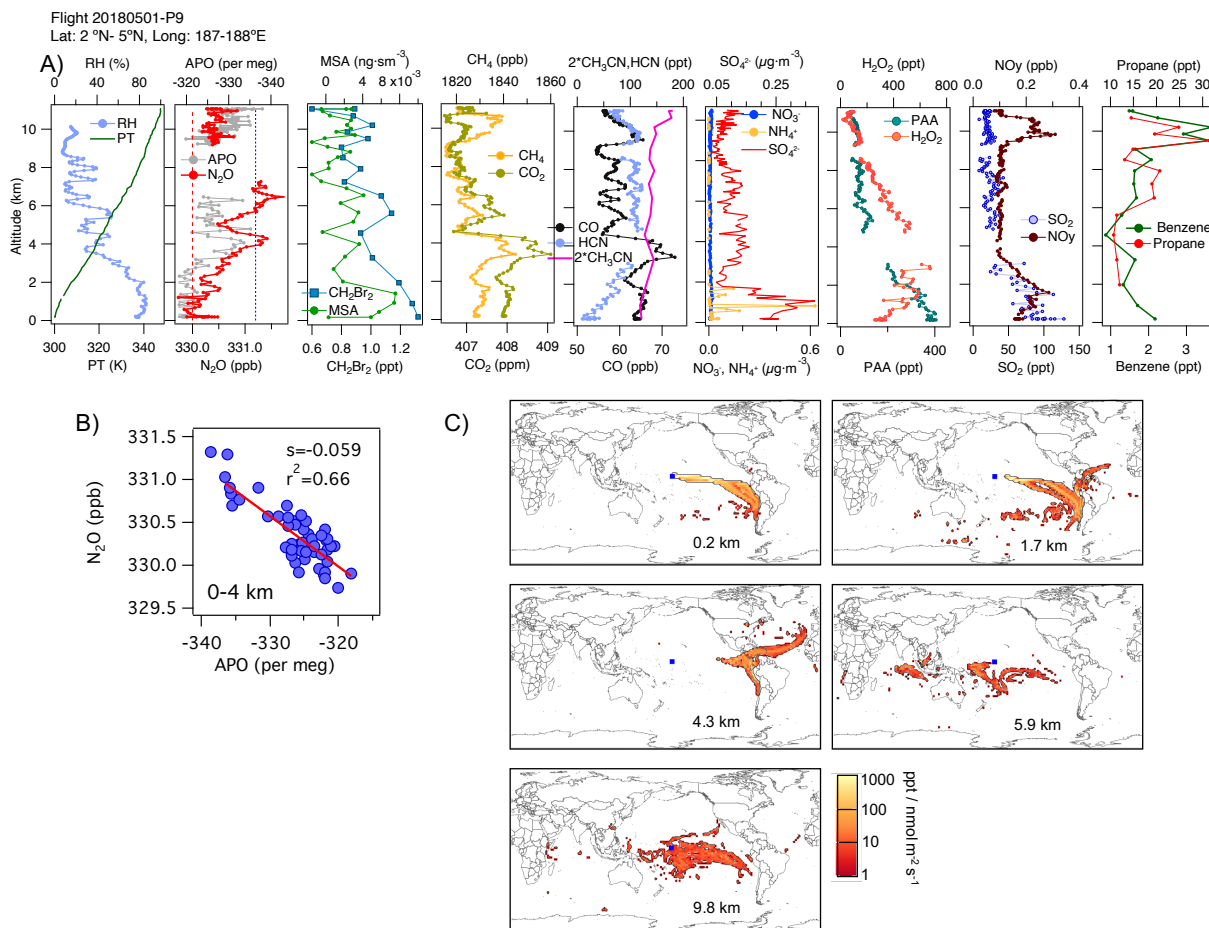
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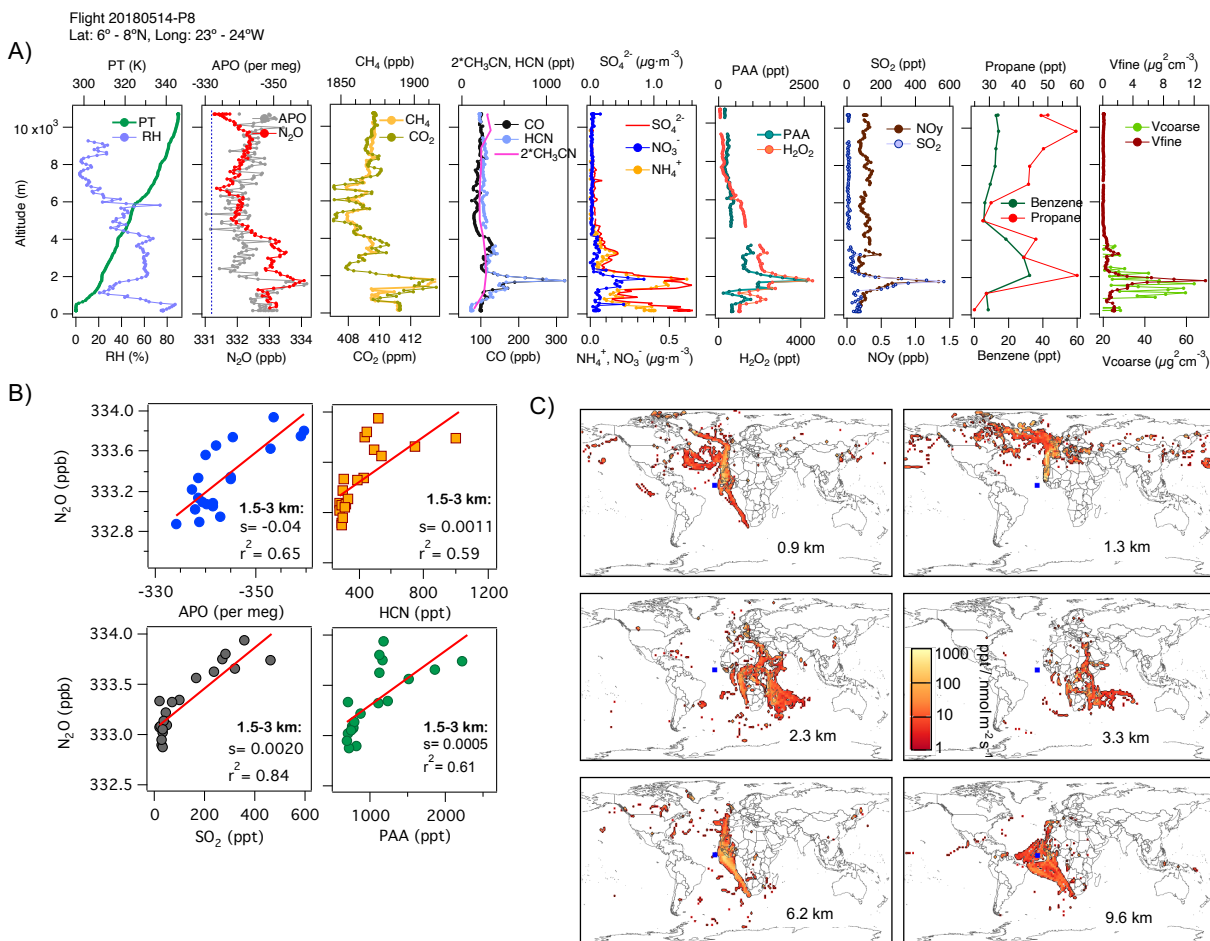
875 **Figure 5: (A)** Vertical profiles of Potential Temperature (PT), Relative Humidity (RH), N<sub>2</sub>O, APO, CH<sub>4</sub>, CO<sub>2</sub>, CO, HCN, CH<sub>3</sub>CN, NO<sub>3</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup>, SO<sub>4</sub><sup>2-</sup>, H<sub>2</sub>O<sub>2</sub>, PAA (CH<sub>3</sub>C(O)OOH), SO<sub>2</sub>, NO<sub>y</sub>, benzene, toluene and propane from profile 12 on 3 May 2018. Dotted blue line in plot A2 represents the NOAA-MBL reference (N<sub>2</sub>O-MBL) at the latitude of the flight. **(B)** Correlations between N<sub>2</sub>O and HCN and PAA for altitudes between 2.5 and 6 km and between N<sub>2</sub>O and SO<sub>2</sub> for altitudes between 0 and 2.5 km indicate an admixture of marine, biomass burning, urban sources, and oil and gas industry contributions to N<sub>2</sub>O mixing ratios (s represents the slope of the linear fit). **(C)** Footprint maps tracing surface regions influencing mixing ratios measured at the altitude ranges of 1–2, 2.5–5 and 5–7 km, respectively. Blue squares show the sample locations. Values below 3 ppt / (nmol m<sup>-2</sup> s<sup>-1</sup>) are not included. Note that the APO axes are reversed.

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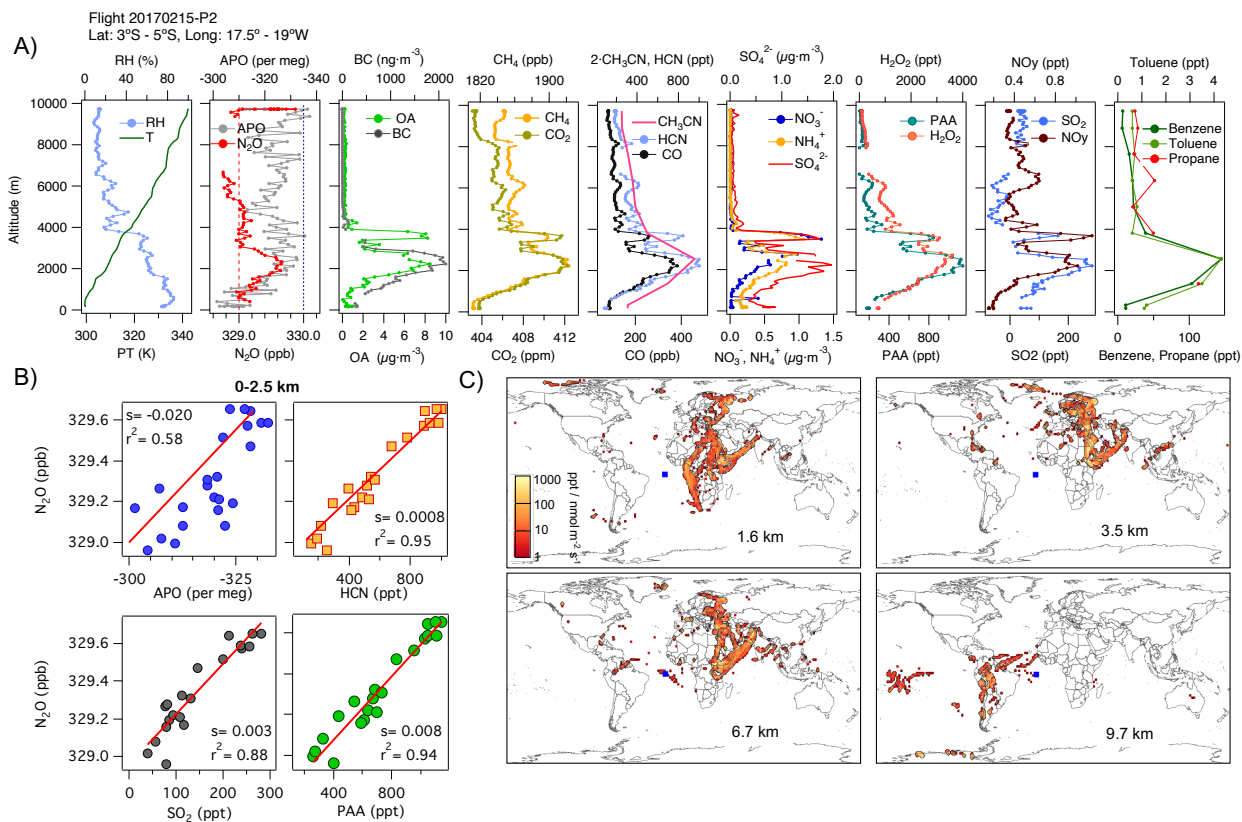
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**Figure 7: (A)** Vertical profiles of PT, RH, and tracers,  $\text{N}_2\text{O}$ , APO,  $\text{CH}_4$ ,  $\text{CO}_2$ , CO, HCN,  $\text{CH}_3\text{CN}$ ,  $\text{NO}_3^-$ ,  $\text{NH}_4^+$ ,  $\text{SO}_4^{2+}$ ,  $\text{H}_2\text{O}_2$ , PAA,  $\text{SO}_2$ ,  $\text{NO}_y$ , benzene, propane as well as the volume of coarse and fine particles, corresponding to profile 8 on 14 May 2018. Dotted blue line in plot A2 represents the NOAA-MBL reference ( $\text{N}_2\text{O}$ -MBL) at the latitude of the flight. **(B)** Correlations between  $\text{N}_2\text{O}$  and APO, HCN,  $\text{SO}_2$ , and propane between 1 and 3 km show possible contributions from marine upwelling, biomass burning and oil and gas industry, supported by the footprints ( $s$  represents the slope of the linear fit). **(C)** Footprint maps tracing surface regions influencing mixing ratios measured at the altitude ranges of 0–1, 2–4, 4–5, 5–7 and 7–10 km, respectively. Blue square shows the sample point. Values below 3 ppt /  $\text{nmol m}^{-2} \text{s}^{-1}$  are not included. Note that the APO axes are reversed.

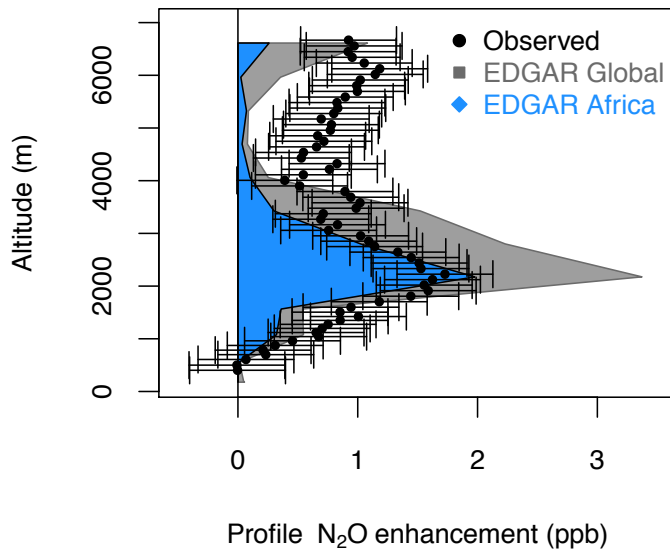
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905 **Figure 8: (A)** Vertical profiles of PT, RH, and tracers, N<sub>2</sub>O, APO, organic aerosols (OA), black carbon (BC), CH<sub>4</sub>, CO<sub>2</sub>, CO, HCN, CH<sub>3</sub>CN, NO<sub>3</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup>, SO<sub>4</sub><sup>2+</sup>, H<sub>2</sub>O<sub>2</sub>, PAA, SO<sub>2</sub>, NO<sub>y</sub>, benzene, toluene and propane, corresponding to profile 2 on 15 February 2017. Dotted blue line in plot A2 represents the NOAA-MBL reference (N<sub>2</sub>O-MBL) at the latitude of the flight, and red dashed line shows the NOAA-MBL at the origin of the southern airmasses shown by the footprints below 2 km (20°S). **(B)** Correlations between N<sub>2</sub>O and APO, HCN and SO<sub>2</sub>, for data observed below 2.5 km, indicate an admixture of marine, biomass burning, urban sources, and oil and gas industry contributions to N<sub>2</sub>O mixing ratios (s represents the slope of the linear fit). **(C)** Footprint maps tracing surface regions influencing mixing ratios measured at the altitude ranges of 0–2, 2–3, 3–4 and 4–7 km, respectively. Blue squares show the sample point. Values below 3 ppt / nmol m<sup>-2</sup> s<sup>-1</sup> are not included in the footprint plot. Note that APO axes are reversed.

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915 **Figure 9.** N<sub>2</sub>O enhancements estimated by EDGAR for the globe (grey polygons) and for the African region (blue polygons) and the observed QCLS-N<sub>2</sub>O enhancement relative to the NOAA-MBL N<sub>2</sub>O reference at the origin of the southern airmasses shown by the footprints below 2 km for the profile 2017/02/15-P2 (20°S, 329 ppb shown in Fig. 8). QCLS-N<sub>2</sub>O are expressed at 10s resolution and receptors were calculated every 60s.

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