## Supplementary material: Evaluating the contribution of the unexplored photochemistry of aldehydes on the tropospheric levels of hydrogen.

Maria Paula Perez-Peña, Jenny A. Fisher, Dylan B. Millet, Hisashi Yashiro, Ray L. Langenfelds, Paul B. Krummel, Scott H. Kable

For a vertical profile comparison of the molecular hydrogen  $(H_2)$  we used the records from the Aircraft (AIA) flask s ampling data from K rummel et al., 2021r measured over T asmania. The seasonal average ratios of the H<sub>2</sub> measured at varying heights with respect to the ground values (from 1991 to 2000) were plotted against the average model estimates from 2015 and 2016 (see Figure S7). SON and DJF were the seasons for which GEOS-Chem represented best the evolution of H<sub>2</sub> with altitude. On the other hand, for MAM and JJA, the model was unable to capture the vertical gradient of the observations attributing more H<sub>2</sub> in height than the reported in the average nine year trend. This might be caused by enhanced modelled vertical transport of H<sub>2</sub>.

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Figure S1: Time series for selected modelled species in molecules  $cm^{-3}$  from the baseline box model simulation in London starting at noon on the  $22^{nd}$  of July 2012.



Figure S2: Time series for selected modelled species in molecules  $cm^{-3}$  from the baseline box model simulation in Cape Verde starting at midnight on the  $2^{nd}$  of January 2015.



Figure S3: Time series for selected modelled species in molecules  $cm^{-3}$  from the baseline box model simulation in Borneo starting at 18:00 on the  $11^{th}$  of July 2008.



Figure S4: Percentage change in the total rate of production of  $H_2$  from aldehydes for varying quantum yields calculated relative to the baseline simulation at the three tested sites for 1 modelled day (local time)



Figure S5: Percentage change in the total rate of production of  $H_2$  from aldehydes for varying quantum yields at the three tested sites for 5 modelled days (local time)



Figure S6: Average modelled mixing rations of  $H_2$  per season estimated with GEOS-Chem at 500 hPa for 2015 and 2016.



Figure S7: Seasonal comparison between average modelled (triangles) and average observed (circles)  $H_2$  estimated ratios at different heights with respect to the surface layer. Modelled averaged values are from 2015 and 2016. Observations averaged values are those from the Aircraft (AIA) flask sampling data from Krummel et al., 2021r



Figure S8: Comparison between baseline-selected in-situ data, baseline-sampled data measured with a PDD detector (blue and orange line) and flask baseline sampled by CSIRO at Cape Grim from April 2015.



Figure S9: Seasonal cycle comparisons of CO at five sites from the CSIRO dataset (Krummel et al., 2021a, 2021b, 2021c, 2021d, 2021e, 2021f, 2021g, 2021h, 2021i) reported Flask Data for 2015 and 2016, the dash line with circle markers corresponds to observed values and the continuous line with triangle markers corresponds to modelled values in GEOS-Chem.



Figure S10: Percentage difference on the chemical production of  $H_2$  between the 1% production of  $H_2$  scenario and the baseline for a year a) At the surface level, b) At 500 hPa, c) Longitude section at 2 N and d) Latitude section at -70 W



Figure S11: Percentage difference on the chemical production of  $H_2$  between the 1% production of  $H_2$  scenario and the baseline for a year at the surface layer from the photolysis of a) methylglyoxal b) acetaldehyde, HPALD, glycolaldehyde, methacrolein, lumped aldehydes with more than 3 carbon atoms RCHO (integrated)



Figure S12: Simulated mixing ratios of methylglyoxal at the surface layer for January and July 2015. The modelled magnitudes of methylglyoxal in our GEOS-Chem simulation are comparable to those shown by Fu et al., 2008 in their Figure 2b.



Figure S13: Percentage difference on the modelled mixing ratios of  $H_2$  between the 1% production of  $H_2$  scenario and the baseline for a year a) At the surface level, b) At 500 hPa, c) Longitude section at 2deg N and d) Latitude section at -70deg W

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Aldehyde	MCM ID	$\overrightarrow{\textbf{Photolysis}}{\textbf{products}^a}$	$\mathbf{JID} \; J \; \mathbf{1\%} \; \mathbf{qy}$	J 2% qy	J 5% qy	J 10% qy
Formaldehyde	$\mathrm{HCHO}^{b}$	H2 + CO	J12 -	-	-	-
Glyoxal	$GLYOX^b$	$\mathrm{CO}+\mathrm{CO}+\mathrm{H2}$	J31 -	-	-	-
Acetaldehyde	CH3CHO	$ ext{CH2CO}^{c} +  ext{H2}$	J62 4.37E-07	8.74E-07	2.18E-06	4.37E-06
Propanal	C2H5CHO	C2H4 + H2 + CO	J63 5.68E-07	1.14E-06	2.84E-06	5.68E-06
Butanal	C3H7CHO	C3H6 + H2 + CO	J64 6.76E-07	1.35E-06	3.38E-06	6.76E-06
Isobutylaldehyde	IPRCHO	C3H6 + H2 + CO	J65 5.49E-07	1.10E-06	2.74E-06	5.49E-06
Acrolein	ACR	C2H2 + H2 + CO	J66 2.44E-06	4.89E-06	1.22E-05	2.44E-05
Methacrolein	MACR	C3H4 + H2 + CO	J67 3.05E-07	6.09E-07	1.52E-06	3.05E-06
Methylglyoxal	MGLYOX	$CH2CO^{c} + H2 + CO$	J68 4.42E-07	8.83E-07	2.21E-06	4.42E-06
HPALD	HCOCH2OOH	H2 + Other products	J69 4.65E-08	9.30E-08	2.32E-07	4.65E-07
Glycolaldehyde	HOCH2CHO	HCHO + H2 + CO	J70 1.07E-07	2.15E-07	5.37E-07	1.07E-06
3-Hydroxy 2- Methylpropanal	HOIPRCHO	H2 + Other products	$J71^{d} 6.76E-07$	1.10E-06	3.38E-06	6.76E-06
Lactaldehyde	СНЗСНОНСНО	H2 + Other	$J71^{d} 6.76E-07$	1.10E-06	3.38E-06	6.76E-06
3-Oxopentanal	CO3C4CHO	H2 + Other	$J71^{d} 6.76E-07$	1.10E-06	3.38E-06	6.76E-06
3-Hydroxy-	HO3C4CHO	H2 + Other	$J71^{d} 6.76E-07$	1.10E-06	3.38E-06	6.76E-06
3- Hudrouwproponal	HOC2H4CHO	H2 + Other	$J71^{d} 6.76E-07$	1.10E-06	3.38E-06	6.76E-06
2- Usedward (	HO3C3CHO	H2 + Other	$J71^{d} 6.76E-07$	1.10E-06	3.38E-06	6.76E-06
Hydroxybutanal 4- Hydroxybutanal	НОСЗН6СНО	products H2 + Other products	$J71^{d} 6.76E-07$	1.10E-06	3.38E-06	6.76E-06

Table S1: List of aldehydes in the MCM, along with photolysis products and rates  $(J [s^{-1}])$  for the H<sub>2</sub> channel estimated with 1%, 2%, 5% and 10% quantum yields applied across wavelengths.

 $^{a}$  Expressed with the identification used in the MCM.

b Species in MCM that already include the photolytic generation of H<sub>2</sub>. The photolysis rates were kept the same for all tests.

cThe ketene  $CH_2CO$  generated in the photolysis channel was replaced with glycolaldehyde as described in the text. <sup>d</sup> In absence of aldehyde-specific cross section measurements, the photolysis rate of butanal was used as a

surrogate in each case.

Site ID	Site name	Lat	Lon	Hemisn	here Lat.	Gas	$Start^a$	$End^{b}$	N. obs <sup>c</sup>	Mean	Mean	-	BMSE	MB	NMB	MFB	NME	FE	FB
		5			band	3				Mod.	Obs.							1	1
										[ddd]	[ddd]								
ALT	Alert	82.4991	-62.3415	HN	HNH	$H_2$	2015-01	2016-12	24	499.91	489.93	0.92	12.97	9.98	2.04	0.02	2.20	2.19	2.02
MLO	Mauna Loa	19.5362	-155.5762	HN	HNH	$H_2$	2015-01	2016-12	24	533.35	536.51	0.56	9.59	-3.16	-0.59	-0.01	1.25	1.27	-0.60
CFA	Cape Ferguson	-19.2773	147.0584	HS	LSH	$H_2$	2015-01	2016-12	20	486.22	546.30	0.70	60.64	-60.08	-11.00	-0.12	11.00	11.66	-11.66
CGO	Cape Grim	-40.6822	144.6883	HS	LSH	$H_2$	2015-01	2016-12	24	509.00	547.37	0.77	38.79	-38.37	-7.01	-0.07	7.01	7.27	-7.27
MQA	Macquarie Island	-54.4985	158.9385	HS	HSH	$H_2$	2015-01	2016-12	24	514.49	547.78	0.88	33.43	-33.29	-6.08	-0.06	6.08	6.27	-6.27
CYA	Casey	-66.2833	110.5167	HS	HSH	$H_2$	2015-01	2016-12	24	516.65	546.93	0.92	30.39	-30.28	-5.54	-0.06	5.54	5.69	-5.69
MAA	Mawson	-67.6047	62.8706	HS	HSH	$H_2$	2015-01	2016-12	14	518.69	551.72	0.81	33.16	-33.03	-5.99	-0.06	5.99	6.17	-6.17
SPO	South Pole	-89.9969	-24.8	HS	HSH	$H_2$	2015-01	2016-12	24	516.67	547.13	0.92	30.57	-30.47	-5.57	-0.06	5.57	5.73	-5.73
<sup>2</sup> Corroen	onde to the fret de	oto oroilo	hlo in tho	11120000	comonte fro	m Kr		+ al (903	1 i i) 1160d	in the o	منامليمان	l of							
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metrics																			

Table S2: Calculated metrics at each measuring site from the Krummel et al., 2021j, 2021k, 2021l, 2021m, 2021n, 2021o, 2021p, 2021q dataset

<sup>b</sup>Corresponds to the last date available in the measurements from Krummel et al., (2021a-i) used in the calculation of

metrics.  $^c\mathrm{Number}$  of measurements used to compare during the period covered by the start  $^a$  and end  $^b$  dates.