# Light Absorption by Brown Carbon over the South-East Atlantic Ocean

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**Abstract.** Biomass burning emissions often contain brown carbon (BrC), which represents a large family of lightabsorbing organics that are chemically complex, <u>thus making it and therefore</u> difficult to estimate their absorption of incoming solar radiation, resulting in large uncertainties in the estimation of the global direct radiative effect

- of aerosols. Here we investigate the contribution of BrC to the total light absorption of biomass burning aerosols over the South-East Atlantic Ocean with different optical models utilizing a suite of airborne measurements from the ORACLES 2018 campaign. by introducing a<u>A</u>n effective refractive index of black carbon (BC),  $m_{eBC}=m_{eBC}=1.95$ +i $k_{eBC}$ , that characterizes accounts for the absorptivity of all possible absorbing components at 660 nm wavelength was introduced to facilitate the attribution of absorption at shorter wavelengths, i.e. 470 nm. Most
- 30 values of the imaginary part of the <u>effective</u> refractive index,  $k_{eBC}$ , were larger than those commonly used for BC from biomass burning emissions, suggesting contributions from absorbers <u>beyond-besides</u> BC at 660 nm. The TEM-EDX single particle analysis further suggests that these long-wavelength absorbers might include iron oxides, as iron is found to be present only when large values of  $k_{eBC}$  are derived. Using this effective BC refractive

5 Absorption attribution using the Bruggeman mixing Mie model suggests a minor BrC contribution of 4 % at 530 nm, while its removal would triple the BrC contribution to the total absorption at 470 nm obtained using the AAE (absorption Ångström exponent) attribution method. Another key finding was that estimates of the BrC contribution at 470 nm from the commonly used AAE (absorption Ångström exponent) attribution method (<5%) are much lower than the BrC contribution estimates (R<sub>BrC,470</sub>) using our new methodology that accounts for contributions from both BrC and non carbonaceous, long wavelength absorbers, such as magnetite. Thus, it is recommended that the application of any optical properties-based attribution method use absorption coefficients at the longest possible wavelength to minimize the influence of BrC-at the long wavelength, and to account for potential contributions from other absorbing materials.

#### **1** Introduction

- 15 Black carbon (BC) and brown carbon (BrC) are the two main light-absorbing carbonaceous aerosols that play a significant role in Earth's radiative forcing and climate (Bond et al., 2013; Laskin et al., 2015; Brown et al., 2018). BC is the principal atmospheric particulate absorber, and that absorbs strongly over the entire solar spectrum (Bond and Bergstrom, 2006). Biomass burning (BB) contributes approximately 2/3 of the global primary organic aerosol (OA) budget, which is currently treated in most climate models treat-as "white carbon" that only 20 scatters and does not absorb light (Bond et al., 2013). However, studies show that BB OA contains a substantial amount of BrC, which predominantly absorbs at short visible and near-UV wavelengths (Chen and Bond, 2010; Lack et al., 2012; Saleh et al., 2014; Taylor et al., 2020). Feng et al. (2013) found that the absorption of BrC can shift BB direct radiative forcing to positive values. The contribution of BrC to the total absorption by carbonaceous aerosols is estimated to be ~20-50 % with a global radiative effect of ~0.03-0.6 W m<sup>-2</sup> (Kirchstetter and Thatcher, 25 2012; Wang et al., 2013; Feng et al., 2013; Saleh et al., 2015; Jo et al., 2016). However, values reported by these studies are subject to substantial uncertainties, and investigations of BrC properties are still very much in a developmental stage (Szopa et al., 2021). Thus, in order to realize a substantive reduction in the uncertainty, a better attribution of BrC light absorption is required-must be first accomplished.
- To date, sSeveral methods have been employed to investigate the light absorption of BrC. The AAE
   (absorption Ångström exponent) attribution method utilizes the different spectral dependences of the absorption by BC and BrC to determine the contribution from BrC at short wavelengths (Lack and Langridge, 2013; Wang et al., 2018a; Taylor et al., 2020). Most earlier studies assumed the AAE of BC (AAE<sub>BC</sub>) to be unity, although

this assumption may introduce large uncertainties (Lack and Langridge, 2013) because AAE<sub>BC</sub> varies with size, mixing state, and wavelength (Lack and Cappa, 2010; Fig. 2 in Liu et al., 2015; Liu et al., 2018). Revised methods with more realistic values of  $AAE_{BC}$  have been proposed, such as using the AAE from two long-visible wavelengths (Taylor et al., 2020) or that from the Mie model (Wang et al., 2018a). A major drawback of these methods is that they are suitable only for mixtures of BC and BrC, and if other absorbing materials are present,

- such as dusts, more detailed information is needed to account for their contribution to light absorption at the longer wavelengths. Another methodology involves measuring the absorption of organics that have been extracted with either an organic solvent or water and represents, thus far, the only way to directly measure BrC absorption (Wong et al., 2019). The drawback to this approach is that not all organics can be extracted with one or two solvents, as
- 10 highlighted by the work of Chen and Bond (2010), who reported extraction efficiencies of  $\sim 70$  % in water and ~90 % in methanol. This inability to extract all organics means that insoluble organic substances remain unknown since they are not measured. This, in turn, could lead to cases where the absorption properties of the extracted organics might be different from those derived from in situ measurements. In addition, this method is carried out offline and requires elaborate laboratory analysis. Still another approach to estimate the absorption of BrC is
- 15 through optical closure, determine the BrC absorption as the difference of the total measured absorption and that of BC calculated using Mie theory (Saleh et al., 2014; Liu et al., 2015). In this approach, the accuracy of BrC absorption relies heavily on the accuracy of BC absorption calculation. Values commonly used for the refractive index of BC (*m*<sub>BC</sub>) from BB emissions in these calculations have ranged from 1.5+0.3i to 1.95+0.79i (Liu et al., 2015; Chylek et al., 2019; Taylor et al., 2020; Kahnert and Kanngießer, 2020), which will lead to large differences
- 20 in BC absorption simulation results (Taylor et al., 2020). To date there is no consensus on the best value of  $m_{\rm BC}$ . Another factor influencing the BC absorption calculation, and hence the estimated BrC absorption, is the mixing state of BC and non-BC components within particles. Liu et al. (2015) used the core-shell (CS) Mie model and Rayleigh-Debye-Gans approximation to investigate the effect of BC microphysics on the estimation of BrC absorption and found it to be highly sensitive to the model treatment. Saleh et al. (2014) compared internal and 25 external mixtures of BC and BrC and found internally mixed cases yielded smaller BrC absorption than externally mixed ones. Similar to the AAE attribution method, the presence of absorbing materials other than BC and BrC can lead to errors in the attribution of BrC absorption. For example, light absorbing iron oxides (FeOx) was found to be common in field studies, especially for BB emissions (Ito et al., 2018), yet few studies perform measurements of size distributions and chemical composition by of thisese particle types, making it difficult to separate their contribution to the total absorption from that of BrC.
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The savannah regions in Africa experience widespread annual BB events from July to October, which are estimated to account to approximately 1/3 of global BB emissions (van der Werf et al., 2010). These aerosols are transported westward over the South-East Atlantic, making the region of the west coast of southern Africa and the South-Eastern Atlantic as that with the largest BrC absorption aerosol optical depth (Brown et al., 2018), this regionand hence an ideal natural laboratory for investigating the absorption of BrC from BB emissions. In this study, we estimate the absorption of BrC using the optical closure method utilizing in-situ aircraft measurements and offline single particle analysis from the ORACLES (ObseRvations of Aerosols above CLouds and their

5 intEractionS) 2018 campaign (Redemann et al., 2020). An effective refractive index of BC ( $m_{eBC}$ ), which attempts to capture the absorption of all possible absorbing components at 660 nm, is introduced to facilitate the absorption attribution at shorter wavelengths. The core-shell model and homogeneous models are applied and compared in this study. The range of values of the organic aerosolOA mass absorption coefficient (MAC<sub>OA,470</sub>) and contribution of BrC to the total absorption at 470 nm ( $R_{BrC,470}$ ) using the optical closure method are <u>also</u> obtained.

# 10 2 Methods



# 2.1 Site and Instrumentation

Figure 1.÷ (upper left panel) overview of the flight locations during ORACLES 2018 investigated in this study, and (all other panels) 7-day back trajectories and fire radiative power (FRP) from MODIS 6 Collection Active Fire Detections (MCD14ML) at the end day of the trajectory (only data with confidence >50 % are used here).

ORACLES was a three-year NASA-funded <u>airborne</u> field campaign to investigate the influence of BB emissions from southern Africa on regional and global climate (Redemann et al., 2020). We investigated aerosol optical properties from straight and level runs during seven research flights (RF) from the ORACLES 2018 campaign (Fig. 1): RF05\_1, RF05\_2, RF05\_3, RF06\_1, RF06\_2, RF10, and RF11. The specifications for each flight can be found in Table 1 in Dang et al. (2021). These flights were chosen as they provided investigations of aerosol properties from-in the marine boundary layer (MBL) and in the-free troposphere (FT), from relatively

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aged aerosols to highly aged aerosols, and are comprehensive in online and offline measurements with few missing data.  $RF05_1$ -occurred in mixed cloud and aerosol layer at an altitude of ~1 km,and -RF05\_2 wereas within the MBL at an altitude of ~940 and 380 m, respectively; the others flights were in the aerosol layer in the above the MBLFT at an altitude equal to or greater than 2 km. The aging time is taken as the first interception of

5 back trajectories with fires; note this age is strictly minimum age as air can get repeated injections of smoke. The aerosol/plume age, modelled with a two-week forecast using the Weather Research and Aerosol Aware Microphysics (WRF-AAM) model (Thompson and Eidhammer, 2014), The age-was estimated to be ~155 h11 days for RF05\_1 and RF05\_2, and ~40 h7 days for RF05\_3, RF06\_1, RF06\_2, RF10, and RF11 (Fig. S1). The age of RF05\_2 was uncertain as its 7 day back trajectories were within the MBL and did not reach the fires, but based on the result from the Weather Research and Aerosol Aware Microphysics (WRF-AAM) model, it was ~9 h older than RF05\_1, and thus the most aged among all samples.

Particle properties measured were non-refractory submicron aerosol composition from an Aerosol Mass Spectrometer (AMS), mass concentrations of refractory BC and mixing state of BC containing particles from a Single Particle Soot Photometer (SP2), particle number size distribution (PNSD) from an Ultra-High-Sensitivity 15 Aerosol Spectrometer (UHSAS) and Aerodynamic Particle Sizer (APS), light absorption at 470, 530, and 660 nm from a Particle Soot Absorption Photometer (PSAP), light scattering at 450, 550, and 700 nm from a nephelometer (Neph), and particle morphology and elemental composition from filter samples that were analysed by a transmission electron microscopy (TEM) coupled with energy dispersive X ray (EDX). The mass and number concentrations of the refractory BC particles and the single-particle mass and mixing state were determined by 20 the Single-Particle Soot Photometer (SP2, Droplet Measurement Technologies). Coating thicknesses of rBCcontaining particles were derived from scattering measurements using the leading-edge-only (LEO) method of Gao et al. (2007). The SP2 detects BC particles with core diameters from 80-650 nm; a lognormal function was fit to BC number size distribution to account for BC particles with core diameters outside this range (Schwarz et al., 2006). The mass equivalent diameter of BC was calculated as  $(6m/\rho_{BC}\pi)^{1/3}$ , where m and  $\rho_{BC}$  represent the 25 mass and density of BC, respectively, with  $\rho_{BC}$  assumed to be 1.8 g cm<sup>-3</sup> (Bond and Bergstrom, 2006). The uncertainty of the measured BC mass concentration and coating thicknesses are estimated to be 20 % and 22 %, respectively (Laborde et al., 2012). The mass ratio (MR) -of non-BC substance to BC for BC-containing particles was determined by integrating the coating mass for all particles in a given time window MR =  $\frac{\sum_{i} (D_{p,i}^{3} - D_{c,i}^{3})^{*} \rho_{BC-free}}{\sum_{i} D_{c,i}^{3} + \rho_{BC}}$ where  $D_p$  and  $D_c$  represent the diameter of coated BC particle and BC core, respectively; the *i* denotes the *i*<sup>th</sup>

30 particle in the that investigated time window. The determination of  $\rho_{BC-free}$  is calculated with the density of various salts (determined will be discussed in Section 2.2) and OA using the volume mixing rule.

<u>A Particle Soot Absorption Photometer (PSAP, Radiance Research) was used to determine 1 Hz aerosol</u> absorption coefficients at wavelengths 470, 530, and 660 nm, which were corrected with the wavelength-averaged <u>Virkkula correction (Virkkula, 2010)</u> and smoothed to 10 s to reduce noise. The uncertainty of absorption <u>coefficients is estimated to be 20 % (Fischer et al., 2010).</u>

5 <u>A High-Resolution Time-of-Flight Aerosol Mass Spectrometer (HR-ToF-AMS, Aerodyne Research)</u> measured mass concentrations of sulphate (SO<sub>4</sub><sup>2-</sup>), nitrate (NO<sub>3</sub><sup>-</sup>), ammonium (NH<sub>4</sub><sup>+</sup>), chloride (Cl<sup>-</sup>), and OA from particles with vacuum aerodynamic diameters between 50 to 500 nm.

We determine single particle mixing state and elemental composition from offline analysis of particles sampled on Paella TEM grids. Around 50 particles per filter were analyzed. A JEOL<sup>™</sup> JEM-2010F FEG-TEM with a ThermoNoran<sup>™</sup> EDX detector was used. TEM was performed at 200 KeV accelerating voltage, with a take-off angle of 15.9 degrees for X-ray emission from the sample. The elemental weight percentage per particle was determined using the NSS software (Thermo Electron Corporation) with Cliff-Lorimer Absorbance

The light absorption from the PSAP was corrected using the wavelength-average correction scheme following Virkkula (2010) and Pistone et al. (2019). All\_in-situ instruments were mounted in the NASA P-3 aircraft and operated under dry conditions. except for the Neph, whose humidity is not controlled. Data for BC mass concentration less than 0.1 μg/m<sup>-3</sup> or absorption coefficient at 660 nm less than 1.5 Mm<sup>-1</sup> were not included. Measurements were averaged to 10 s and adjusted to STP values at 273.15 K and 1013 hPa.

correction method. Detailed descriptions can be found in Dang et al. (2021).

## 2.2 Optical Models

- From the TEM images, mMost of the particles were found to be nearly spherical from the images obtained by TEM, with >70% of the particles having aspect ratios (the largest and smallest characteristic sizes of arbitrarily shaped particle) smaller than <1.5, Ttherefore, we applied Mie theory in our to determine aerosol optical simulationsproperties. Models such as T-matrix or discrete dipole approximation model were not used as they would require a large number of free parameters (Scarnato et al., 2013; He et al., 2016). We investigated the sensitivity of our closure simulations to four different models for the BC-containing particles the ideal coreshell (CS) model and three homogeneous mixing models: 1) the volume mixing (VM) model, 2) the Maxwell-Garnett (MG) model, and 3) the Bruggeman (BG) model. We did not investigate the externally mixing model, as our TEM analysis showed that most of the particles were internally mixed (Dang et al., 2021). The number size distribution of BC-containing particles was obtained from the BC core size distribution and the BC 2-D size and</li>
- 30 mixing state (i.e. coating thickness) distribution from SP2. In our optical closure, particles were separated into BC containing particles and BC free particles, whose size distributions were calculated with the PNSD of all particles from the UHSAS and APS, and BC 2 D size and mixing state (i.e. coating thickness) distribution of BCcontaining particles from the SP2. We applied the aforementioned four models to BC containing particles.

Detailed descriptions and inputs of <u>all-the</u> four models can be found in Section S12 in the supplement. We assumed that the non-BC components in both populations were homogeneously well-mixed and calculated the refractive index of the mixture,  $m_{BC-free}$ , with the VM rule, which assumes that the index of refraction of the mixture is the sum of the volume-weighted indices of refraction of the individual components. The mass concentrations of SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup>, and Cl<sup>-</sup> measured by AMS were converted to those of inorganic salts to facilitate calculation, using

the simplified ion-pairing scheme proposed by Gysel et al. (2007), modified as follows:

$$n_{NH_4Cl} = n_{Cl^-} =$$

$$n_{NH_4NO_3} = \min(n_{NO_3^-}, n_{NH_4^+} - n_{Cl^-}) \min(.-)$$
(1)
(2)

$$n_{NH_4HSO_4} = \max\left(0, \min\left(2n_{SO_4^{2-}} - n_{NH_4^+} + n_{NO_3^-} + n_{Cl^-}, n_{NH_4^+} - n_{NO_3^-} - n_{Cl^-}\right)\right) = \max$$
(3)

$$n_{(NH_4)_2SO_4} = \max\left(0, \min\left(n_{SO_4^{2-}}, n_{NH_4^+} - n_{NO_3^-} - n_{Cl^-} - n_{SO_4^{2-}}\right)\right) = \max\left(4\right)$$

$$n_{KNO_3} = \max\left(0, \min\left(n_{NO_3^-}, n_{NH_4^+} - n_{Cl^-} - n_{NO_3^-} - 2n_{SO_4^{--}}\right)\right) = \max$$
(5)

$$n_{K2SO_4} = \max\left(0, \min\left(n_{SO^{2-}}, n_{SO^{2-}} - n_{NH^+} + n_{NO^-_2} + n_{Cl^-}\right)\right) - \max\left(6\right)$$

where  $n^{\text{mol}}$  represents the number of moles. As potassium salts are the most frequently detected salts from TEM-EDX analysis (Dang et al., 2021), the residual anions were assumed to be combined with potassium, i.e. KNO<sub>3</sub> and K<sub>2</sub>SO<sub>4</sub>-; <u>however</u>, <u>Note that</u> a small fraction of NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup> detected by <u>the</u> AMS might be organics and thus lack an accompanying cation. Densities of the various salts and OA used to convert the mass concentrations to volume concentrations were taken from Kuang et al. (2020; Table 1) and Liu et al. (2015). Th<u>eus</u>, *m*<sub>BC-free</sub> is

as:

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$$m_{BC-free} = \left(\sum_{i=salts} m_i * V_i + m_{OA} * V_{OA}\right) / \left(\sum_{i=salts} V_i + V_{OA}\right)$$
(7)

where m<sub>i</sub> and V<sub>i</sub> represent the refractive indices and volumes of aforementioned salts; moA and VoA are those for OA. The refractive indices of salts are taken from (Fierz-Schmidhauser et al.; (2010); and Cotterell et al.; (2017)
Table 1 in Kuang et al. (2020). The real part of the refractive index of OA varies from 1.35 to 1.7 (Lack and Cappa, 2010; Liu et al., 2013; Saleh et al., 2014; Moise et al., 2015); 1.55 is used in this study. OA is assumed to be non-absorbing in the calculation. The average and standard deviation of calculated m<sub>BC-free</sub> is 1.52±0.015, consistent with those used in BrC studies in literature (Saleh et al., 2013; Liu et al., 2021, 2015). The quantity m<sub>BC-free</sub> is assumed to be the same value for both BC-free particles and the non-BC component in BC-containing particles.
The particle offective refractive indices for the VM\_MC\_ and PG medals can then he obtained by using the

The <u>particle</u> effective refractive indices for the VM, MG, and BG models can then be obtained <u>by-using</u> the corresponding mixing rules (Section S12 in the supplement) with  $m_{BC-free}$  and  $m_{BC}$ . The <u>determination of the</u> refractive ind<u>exices</u> of BC and OA-will be <u>presented discussed</u> in the following section.

### 2.3 Optical calculation procedure



Figure 2<sub>2</sub>: Schematic diagram of the iterative derivation of BrC absorption attribution performed in this study. Cells in white, yellow, and blue represent instrument, measurements, and calculations, respectively. The green box is the optimization step used to determine the values of the <u>imaginary part of the</u> effective refractive index of BC, <u>*mk*eBC</u>-and the refractive index of OA, *m*OA, once the deviation of measured and calculated seattering and absorption coefficients

at 660 nm ( $\sigma_{abs,660}^{cal}$  and  $\sigma_{abs,660}^{mea}$ ) are minimized. <u> $k_{eBC}$  and  $m_{OA}$  are is</u> held constant at investigated wavelengths, i.e. 470, 530, and 660 nm. The imaginary part of the refractive index of OA,  $k_{OA}$ , was restricted to 0 in this calculation.



5 Figure 3. Modelled (blue and orange markers) and measured (green markers) absorption coefficients ( $\sigma_{abs}$ ) at PSAP wavelengths for RF06\_1. Variables are modelled with two  $m_{BC}$  values (shown in the legend) using the CS, MG, BG, and VM models (specified on the top of each plot). OA is assumed to be non-absorbing with the refractive index  $m_{OA}$ of 1.55+0i. The horizontal lines in the boxes represent the median value, the boxes represent 25<sup>th</sup> to 75<sup>th</sup> percentile, the whiskers represent 1.5 inter-quartile range, and the diamonds represent outliers.

wavelength (nm)

wavelength (nm)

Figure 3: Modelled (blue and orange markers) and measured (green markers) absorption coefficients ( $\sigma_{abs}$ ) and scattering coefficients ( $\sigma_{sea}$ ) at PSAP wavelengths for RF06\_1. Variables are modelled with two commonly used  $m_{BC}$ values (shown on the top) and two values of  $m_{OA}$  using the CS, MG, BG, and VM models (specified on the right). OA is assumed to be non-absorbing with the refractive index  $m_{OA}$  equal to 1.5+0i (blue markers) and 1.65+0i (orange markers), respectively. Values of  $\sigma_{sea}$  measured by the nephelometer at 450, 550, and 700 nm are interpolated to PSAP wavelengths, 470, 530, and 660 nm using the scattering Ångström exponent (Section S3 in the supplement).

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The numerical value selected for  $m_{BC}$  has a large impact on the absorption attributed to BC (Fig. 3), and hence on the estimated absorption by BrC, but as noted above, there is no consensus on the best value of  $m_{BC}$ . Additionally, the lack of measurements of possible absorbers other than BC and BrC such as magnetite makes it difficult to correctly estimate the BrC absorption, so approaches that take these components into consideration are required. Here, wWe introduce an effective refractive index of BC,  $m_{eBC} = 1.95 + ik_{eBC}$ , to represent the refractive index of BC and any other absorbing components at 660 nm in this study. The  $m_{eBC}$  would be the same as  $m_{BC}$  if the absorption at 660 nm is solely contributed by BC; however, if other absorbing components are present, the

imaginary part  $(k_{BC})$  of  $m_{eBC}$  would be greater than that  $(k_{BC})$  of  $m_{BC}$ , the amount of which depends on the

- 15 absorptivity and relative amount of these absorbers. The *m*<sub>eBC</sub> is defined as *m*<sub>eBC</sub> = neBC+ik<sub>eBC</sub>, where *n*<sub>eBC</sub> and *k*<sub>eBC</sub> are the real and imaginary parts of *m*<sub>eBC</sub>, respectively. The *m*<sub>eBC</sub> is proposed based on three considerations. Firstly, the numerical value selected for *m*<sub>BC</sub> has a large impact on the absorption attributed to BC (Fig. 3 and Fig. S3), and hence on the estimated absorption by BrC, but as note above, there is no consensus on the best value of *m*<sub>BC</sub> yet. Secondly, the lack of measurements of possible absorbers other than BC and BrC such as magnetite
- 20 makes it difficult to correctly estimate the BrC absorption, so approaches that take these components into consideration are required. The real part of *m*<sub>eBC</sub> is held constant at 1.95, the upper bound of the values commonly used for BC (Bond and Bergstrom, 2006; Saleh et al., 2013; Liu et al., 2015; Kahnert and Kanngießer, 2020). It has only a minor influence on absorption calculations (Liu et al., 2021): sensitivity test of the calculated absorption to the real part of *m*<sub>eBC</sub> when the latter is varied from 1.75 to 2.26 shows that, at the extreme case of RF10 with the thickest coating in this study, the absorption increased less than 5 % at 660 nm. Thirdly, aged BrC is usually considered to be non absorbing at long visible wavelengths (Chen and Bond, 2010; Lack et al., 2012), thus we assume *m*<sub>eBC</sub> does not include the absorptivity of BrC and therefore the BrC absorption can be differentiated from absorption from other substances. This assumption has not considered tar balls, which are ubiquitous in fresh plumes and absorbing at infrared wavelengths; considering the aging time of our samples, the absorption of tar
- 30 balls is expected to be small or negligible (Sedlacek III et al., 2018; Adler et al., 2019; Li et al., 2019). The m<sub>eBC</sub> is named the effective refractive index of BC as it is supposed to be contributed mainly or entirely by BC. The m<sub>eBC</sub> would be the same as m<sub>BC</sub> if the absorption at 660 nm is solely contributed by BC; however, if other absorbing components are present, the imaginary part (k<sub>eBC</sub>) of m<sub>eBC</sub> would be greater than that

 $(k_{BC})$  of  $m_{BC}$ , the amount of which depends on the absorptivity and relative amount of these absorbers. The retrieval of  $m\underline{k}_{eBC}$  is illustrated in Fig. 2; the real refractive index of OA  $(n_{OA})$  is retrieved as well because it has an impact, albeit small, on the BC absorption calculation (Fig. 3 and Fig. S3). Note that OA is assumed to be non-absorbing; thus, with the imaginary refractive index  $k_{OA} = \underline{is}$  taken as zero $\theta$  in this whole calculation. We determined the  $\underline{v}\underline{V}$  alues of  $\underline{m}_{eBC} - \underline{k}_{eBC}$  and  $\underline{m}_{OA}$  by were determined by minimizing the sums of the squares,  $\chi^2$ , of the relative

differences between the measured and calculated <del>seattering and </del>absorption coefficients at 660 nm for each <del>of the</del> four models:

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$$\chi^{2}(m_{eBC}, m_{OA}) = \left(\frac{\sigma_{sca,660}^{cal} - \sigma_{sca,660}^{mea}}{\sigma_{sca,660}^{mea}}\right)^{2} + \left(\frac{\sigma_{abs,660}^{cal} - \sigma_{abs,660}^{mea}}{\sigma_{abs,660}^{mea}}\right)^{2} \chi^{2}(k_{eBC}) = \left(\frac{\sigma_{abs,660}^{cal} - \sigma_{abs,660}^{mea}}{\sigma_{abs,660}^{mea}}\right)^{2},$$
(8)

where σ<sup>cal</sup><sub>ab5,660</sub> and σ<sup>mea</sup><sub>ab5,660</sub> are the calculated and measured absorption coefficients at 660 nm, respectively. The <u>v</u>Values of *n*<sub>eBC</sub>, *k*<sub>eBC</sub> and *n*<sub>OA</sub> areis varied searched from 1.5 to 4, 0.3 to 2.6,5 over a greater range than the values of *k*<sub>BC</sub> recorded in literature (Chylek et al., 2019; Kahnert and Kanngießer, 2020 Taylor et al., 2020), -and with a span of 0.01-1.35 to 1.7 (Laek and Cappa, 2010; Liu et al., 2013; Saleh et al., 2014; Moise et al., 2015), respectively over greater range than those recorded in literature. The *k*<sub>eBC</sub> vValues of with χ<sup>2</sup>/W greater than 10%0.1 are excluded, where N stands for the total data number. The quantities and are the calculated and measured absorption coefficients at 660 nm, respectively, and and are the calculated and measured seattering coefficients
at 660 nm, respectively, where is determined by interpolated using the scattering Ångström exponent from seattering coefficients measured by the nephelometer (Section S3 in the supplement). Only absorption is used in the calculation of for RF05\_1 and RF05\_2, as measured and simulated scattering coefficients are not comparable (discussed in detail in Section S1.2 in the supplement):

$$\chi^{2}(m_{eBC}, m_{OA}) = \sum_{\lambda = 660} \left( \frac{\sigma_{abs,\lambda}^{cal} - \sigma_{abs,\lambda}^{mea}}{\sigma_{abs,\lambda}^{mea}} \right)^{2} \tag{9}$$

The absorption of BrC at the long visible, i.e. 660 nm, is expected to be very small or negligible
considering the aged nature of aerosols in our study (Sedlacek III et al., 2018; Adler et al., 2019; Li et al., 2019). Therefore, we assume that the contribution of BrC to k<sub>eBC</sub> can be neglected, and use it to differentiate the BrC absorption from that of other substances, i.e. BC and any other absorbers. This assumption of nil contribution of BrC to k<sub>eBC</sub> yields a lower bound of BrC absorption at 530 and 470 nm. An evaluation of the influence of this assumption to the BrC absorption at shorter wavelengths is discussed in Section S2 in the supplement. As
illustrated in Fig. 2, the absorption at shorter wavelengths is assumed to be determined by that from BrC, and that from all other absorbing components, the latter of which is characterized entirely by meters. We assume the meter absorption coefficients at 470 and 530 nm (σ<sup>eal</sup> absorption and σ<sup>eal</sup> absorption for each flight using the different models. This is supported by e values of m<sub>BC</sub>- that it isand mo<sub>A</sub>-are usually considered to be very weakly dependent on, or

independent of, wavelength in the visible spectrum (e.g., Chang and Charalampopoulos, 1990; Moteki et al., 2010; Saleh et al., 2014). As\_<u>will be</u> discussed later in Section 3.1, <u>other</u> absorbers that contributinge to  $m_{eBC}$ , if any, besides BC agre mainly regarded hypothesized to be as magnetite, whose refractive index is generally also mostly invariant\_\_between wavelengths 470 and 660 nm\_(Amaury et al., unpublished data, http://www.astro.uni-

- 5 jena.de/Laboratory/OCDB/mgfeoxides.html; Ackerman and Toon, 1981; Zhang et al., 2015). Thus, the assumption of constant *m*<sub>eBC</sub> and *m*<sub>OA</sub>-between 470 and 660 nm is reasonable in our study, although it may lead to underover estimations of BrC absorption for highly aged particles, as discussed further in Section 3.2. In addition, the absorption by BrC may be slightly underestimated without incorporating tar balls into the calculation. As illustrated in Fig. 2, the absorption at shorter wavelengths is assumed to be determined by that from BrC, and that from all other absorbing components, the latter of which is characterized entirely by *m*<sub>eBC</sub>.
  - Therefore, the absorption of BrC at wavelength  $\lambda$  ( $\sigma_{abs,BrC,\lambda}$ ) is can be calculated as the difference between measured absorption coefficients ( $\sigma_{abs,\lambda}^{mea}$ ) and the onesthat calculated with  $m_{eBC}$  ( $\sigma_{abs,\lambda}^{cal}$ ):

$$\sigma_{abs,BrC,\lambda} = \sigma_{abs,\lambda}^{mea} - \sigma_{abs,\lambda}^{cal}$$
(9)

The fractional contribution of BrC to the total absorption at wavelength  $\lambda$ , quantity  $R_{BrC,\lambda}$ , is defined as

$$R_{Brc,\lambda} = \sigma_{abs,Brc,\lambda} / \sigma_{abs,\lambda}^{mea} \tag{10}$$

is the fractional contribution of BrC to the total absorption at wavelength  $\lambda$ , which and is assumed to be zero at 15 660 nm in our study. The mass absorption coefficient cross section (MAC) of OA (MACOA,) at wavelength  $\lambda_{a}$ MACOA,  $\lambda_{a}$ , can be determined from  $\sigma_{abs,BrC,\lambda}$  and the mass concentration of OA,  $M_{OA}$ , as:

$$MAC_{OA,\lambda} = \sigma_{abs,BrC,\lambda}/M_{OA}$$
(11)

Similarly, the mass absorption coefficient of BC (MAC<sub>BC</sub>) is defined as:  $MAC_{BC,\lambda} = \sigma_{abs,\lambda}^{mea}/M_{BC}$ , where  $\sigma_{abs,\lambda}^{mea}$  and  $M_{BC}$  represent the measured absorption coefficient <u>at wavelength  $\lambda$ </u> and the mass concentration of BC<sub>2</sub> respectively.

# 20 **3** Results and discussion

### 3.1 Effective refractive index of BC (*m*<sub>eBC</sub>)

In this section, we present and discuss the results of *mk*<sub>eBC</sub> derived from the aforementioned four models using the AMS, SP2, APS, and UHSAS data for each sampleflight. Uncertainties of *k*<sub>eBC</sub> retrieved from various models have been determined by Monte-Carlo uncertainty analysis in Section S3 in the supplement. The values of *k*<sub>eBC</sub> for each sample flight determined by the various models are shown in Figure 4. No *m*<sub>eBC</sub> result for RF05\_3 was obtained with the CS model because no *m*<sub>eBC</sub> value has been achieved because the criteria of *χ*<sup>2</sup> smaller than 0.1 in the retrieval was not met, as the Aabsorption coefficients calculated with the largest possible *m*<sub>BC</sub> with the CS model using the largest possible *m*<sub>BC</sub> (i.e. 2.26+1.26i) underestimate was only half of the measurements for <u>RF05\_3</u> by twice (Fig. S23).

The <u>V</u>variations of  $k_{eBC}$  shows a similar patterns among different models (Fig. 4). <u>The Those vv</u>alues derived from the CS and VM models are the highest and lowest, respectively, <u>consistent with modelling results</u> for absorption in previous studies (e.g. Taylor et al. 2020), and, and those the values derived from the MG and BG models are between the other two <u>models</u> and <del>are</del> very close to each other, <u>consistent in agreement with modelling</u>

- 5 <u>the results for absorption results calculated with various models in previous studies-Taylor et al. 2020</u> (Lesins et al., 2002; Fierce et al., 2017a; Taylor et al., 2020). The values of  $k_{eBC}$  for RF05\_1, RF05\_2, RF06\_1, and RF06\_2 from the CS model have greatly exceeded the largest value of  $k_{BC}$  of 1.26 commonly used values of  $m_{BC}$ -(grey shaded region in Fig. 4), determined by Moteki et al. (2010) for soot particles from an urban source; (Tokyo). Except for RF10, Llittle confidence has been placed in the value of  $k_{eBC}$  from the CS model except for RF10 for
- 10 two main reasons. First, the bulk-MR for all samples is less than 2, and application of the CS model may can be inappropriate for particles with such low MR values (Liu et al., 2017). BesidesFurthermore, the TEM analysis found few core-shell structures (e.g., example of RF05\_1 in Fig. 5) among all-investigated samples, with most BC-containing particles composed of condensed BC nodules internally mixed with salts/organics (e.g., example of RF05\_3 in Fig. 5) or aggregates attached to non-BC components (e.g., example of RF05\_2 in Fig. 5). Figure 5
- 15 shows <u>R</u>representative TEM images of different types of BC-containing particles for each flight are shown in Fig. 5. The use of CS model for RF10 was supported by its high MR value of 7.4, which has exceeded the lower bound of MR for the use of CS model proposed by Liu et al. (2017). AltTalthough a fair number of BC particles homogeneously mixed with salts and OA were detected as well from TEM for RF10 (Fig. 5)., which seems to resemble homogeneous models, this sample is thickly coated, with MR equal to 7.4 and MR100 equal to 22.5,
- 20 largely exceeded the threshold value of MR for the CS model specified in Liu et al. (2017). In addition, results of the BrC contribution to the total absorption also suggest that homogeneous models might be inappropriate for RF10, as discussed in detail in Section 3.3.





Figure 4<sub>2</sub>: <u>Values Boxplot</u> of  $k_{eBC}$  (elosed dots<u>left axis</u>) derived from different models and <u>the MR values</u> (orange open dots, <u>top right</u> \*axis) for each flight. Error bars of MR represent 20<sub>2</sub>% uncertainty. The <u>top right</u> \*y-axis is-us<u>es</u> the log scale. <u>Light and dark gGrey shaded region shows</u> for  $k_{eBC}$  smaller than 1.26, the largest marks the range of values of  $k_{BC}$  value commonly used fodetermined by Moteki et al. (2010), and 0.79, the largest value of commonly used  $k_{BC_{7a}}$ 



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Figure 5<sub>2</sub>÷ Elemental composition and representative microphysical images of different types of BC-containing particles from TEM analysis for each flight. Int BC and Ext BC represent internally and externally mixed BC particles, respectively. For example, Int BC+K stands for internally mixed BC with potassium salts. The y-axis uses a x<sup>1/2</sup> scale.

As discussed in <u>Section 2.3</u>,  $\underline{mk}_{eBC}$  should be the same as  $\underline{mk}_{BC}$  if the absorption at 660 nm arises only from BC; otherwise, higher  $\underline{k}_{eBC}$  values indicate larger contributions of absorbers other than BC at 660 nm higher  $\underline{k}_{eBC}$  value indicates larger absorptivity of absorbers at 660 nm. The values of  $\underline{k}_{eBC}$  for RF05\_1, RF05\_2, RF05\_3,

RF06\_1, and RF06\_2, for which iron (Fe) was detected from the TEM-EDX (as shown in the bar chart in Fig. 5), are mostly larger than unity and greater than those for RF10 and RF11, for which no iron was detected. Therefore, we suspect higher  $k_{eBC}$  values may be related to iron oxides (FeOx) that absorbs at 660 nm. We have examined all possible compounds of detected elements from EDX and found that magnetite is the only compound that is

- 5 not rare in the atmosphere and also absorbs strongly at 660 nm. Magnetite shows a strong and uniform absorptivity over the visible spectrum with reported imaginary refractive index ranging from 0.58 to 1.0 (Amaury et al., unpublished data, http://www.astro.uni-jena.de/Laboratory/OCDB/mgfeoxides.html; Ackerman and Toon, 1981; Zhang et al., 2015), and therefore can contribute to the high values of *k*<sub>eBC</sub> at 660 nm. Studies show that magnetite can be transformed from Fe(III) at high temperatures, such as goethite and hematite (Till et al., 2015; Ito et al.,
- 10 2018), which happen to be the two most abundant forms of FeOx in African dust (Formenti et al., 2014). Therefore, we hypothesize that a part of the magnetite might be converted from Fe(III) during biomass burning. Additionally, magnetite can be emitted from anthropogenic activities, such as steel manufacturing, oil combustion, and engines and brakes of motor vehicles (Machemer, 2004; Liati et al., 2015; Moteki et al., 2017; Kurisu et al., 2019). The significance of anthropogenic magnetite in radiative forcing has been investigated and highlighted in recent
- 15 studies (Moteki et al., 2017; Ito et al., 2018; Lamb et al., 2021). Although there is no investigation on magnetite in sub-Saharan Africa yet, the industrial and motor vehicle emissions in Africa are likely to contribute magnetite. Furthermore, the pyrometallurgical process is a widely used extraction method in copper mining, a major industry in the central African Copperbelt (https://www.pyrometallurgy.co.za/PyroSA/, Vítková et al., 2010; Sikamo et al., 2016; Shengo et al., 2019), of which iron is a common unwanted slag element (Meter et al., 1999). Therefore, we
- suspect it might also contribute magnetite considering the high temperature of the pyrometallurgical process. Therefore, we suspect higher *k*<sub>eBC</sub> values may be related to iron oxides (FeOx) that absorbs at 660 nm, which we suspect to be magnetite based on the following three points. Firstly, magnetite absorbs strongly and uniformly over the visible spectrum with the imaginary refractive index ranging from 0.58 to 1.0 according to literature (Amaury et al., unpublished data, http://www.astro.uni jena.de/Laboratory/OCDB/mgfeoxides.html; Ackerman and Toon, 1981; Zhang et al., 2015), and can thereby contribute to the high values of *k*<sub>eBC</sub>-at 660 nm. Secondly, magnetite can be transformed from Fe(III) at high temperatures, such as goethite and hematite (Till et al., 2015; Ito et al., 2018), which happen to be the two most abundant forms of FeOx in African dust (Formenti et al., 2014). Therefore, it is speculated that a part of the magnetite is converted from Fe(III) during biomass burning. Thirdly,
- 30 emissions (Liati et al., 2015; Ito et al., 2018; Kurisu et al., 2019). Its significance in radiative forcing and carbon cycle has been investigated and highlighted in recent studies (Moteki et al., 2017; Ito et al., 2018; Lamb et al., 2021). Although there is no study on pyrogenic iron in sub Saharan Africa yet, we speculate that anthropogenic combustion emissions in Africa, especially in coastal areas, may also contribute magnetite.

magnetite can be emitted by anthropogenic combustion, such as steel manufacturing, oil combustion, and vehicle

Among all flights, RF05\_3, collected at the uppermost aerosol layer (Fig. 4), is an exception among all flights, since the values of the absorption coefficient at 660 nm calculated with all four models are considerably smaller than the measured ones (Fig. S23). The MG, BG, and VM models yield median  $k_{eBC}$  values greater than 1.6;5, and no values was attained met the criteria of  $\chi^2 < 0.1$  with for the CS model. The MAC<sub>BC</sub> of RF05\_3 is the

- 5 highest among all those investigatedflights with values (mean±standard deviation) of 20.0±0.8, 17.8±0.8, and 14.3±0.7 m<sup>2</sup> g<sup>-1</sup> at 470, 530, and 660 nm, respectively. The absorption enhancement, *E*<sub>Abs</sub>, defined as the ratio of MAC<sub>BC</sub> to the value for uncoated BC reported by Bond and Bergstrom (2006), is 2.3±0.1 for all three wavelengths. To the best of our knowledge, except for modelling or laboratory studies of thickly coated particles (Bond et al., 2006; Jacobson, 2012; Peng et al., 2016), such high values of *E*<sub>Abs</sub>, particularly at long visible wavelengths, are rarely reported in field measurements (Cui et al., 2016). Taylor et al. (2020) presented relatively high MAC<sub>BC</sub> values of 20±4, 15±3, and 12±2 m<sup>2</sup> g<sup>-1</sup> at 405, 514, and 655 nm, respectively and an *E*<sub>Abs</sub> of 1.85±0.45 for the
- CLARIFY\_2017 (<u>Cloud-Aerosol-Radiation Interaction and Forcing 2017 measurement campaign;</u> Haywood et al., (2020); however, these values, which are still-generally comparable or smaller than those for RF05\_3. However, pParticles in CLARIFY 2017 are universally thickly coated, with median MR values of 8-12, and are
- 15 therefore would be expected to have large values of *E*<sub>abs</sub>believed to have a largely enhanced lensing effect, which is not the case for as opposed to RF05\_3, which for which had a median MR (Fig. 4) was equal to of 1.6 and MR<sub>100</sub> was equal to 5.3. The We calculated absorption coefficients for RF05\_3 at 660 nm using the VM model, which usually yields a higher estimation of the measurements (Taylor et al., 2020), using and a *m*<sub>BC</sub> of 1.95+0.79i and *m*<sub>OA</sub> of 1.65+0i, an upper bound of the commonly used BC refractive index (Liu et al., 2021). Uncertainties
- 20 related to calculated absorption coefficients can be found in Section S3 in the supplement. The calculated absorption, which is very likely an overestimation of the measurement, is <u>was</u>-only able to explain ~57 %9% of the measured absorption coefficient (Fig. S2), implying that the remaining 41%amount of absorption which cannot be attributed to BC was might result fromeontributed by other absorbers. This huge-unexplained portion of absorption from other absorbers in this flightRF05\_3 is at least twice as high as that in other flights (e.g. Fig. 3) and Fig. S2). Back-trajectories show that air parcels in RF05\_3 passed by the Copperbelt in Zambia (Fig. 1), where the pyrometallurgical process in copper mining might contribute large amounts of magnetite-may indicate a much larger amount of magnetite compared to other flights. In addition, we noticed titanium (Ti) on the particles analyzed from RF05\_3. Formenti et al. (2014) found the mineral dust in western Africa consists mainly of clays, quartz, iron, and titanium oxides, which together represent at least 92% of the dust mass. Although titanium oxides are not absorbing, several forms of Ti have been reported to strongly absorb at visible wavelengths (Pflüger and Fink, 1997; Palm et al., 2018). Therefore, we suspect this large-unexplained absorption in RF05\_3 may also be

# 3.2 Mass absorption coefficient of OA (MACOA)

related to absorbing titanium compounds.



Figure 6<sub>2</sub>÷ Values of MAC<sub>0A,470</sub> (closed dotsboxes) derived using different models and the mass concentrations of OA, *M*<sub>0A</sub> (orange open-dots, top xleft y-axis), for each flight. Error bars represent the 10<sup>th</sup>-and 90<sup>th</sup>-percentile1σ standard deviation. The scale of both x-axes is symmetric log. The horizontal lines of boxplot represent the median value, the boxes represent 25<sup>th</sup> to 75<sup>th</sup> percentile, the whiskers represent 1.5 inter-quartile range. RF05\_1 and RF05\_2 were measured in the MBL with plume agd larger than 11 days, other flights were measured in the FT with aging time around 7 days. The aerosol/plume age provided by a two-week forecast using the Weather Research and Aerosol Aware Microphysics (WRF-AAM) model.

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Values of MAC<sub>OA,470</sub> calculated from Eq. 1<u>1</u><sup>2</sup> using the different models are shown in Fig. 6. <u>The</u> <u>uncertainty of MAC<sub>OA,470</sub> has been determined using Monte-Carlo uncertainty analysis in Section S3 in the</u> <u>supplement.</u> The values of MAC<sub>OA,470</sub> from the three homogeneous models were fairly close to each other, <u>as</u> <u>expected from previous studies</u> (Lesins et al., 2002); while MAC<sub>OA,470</sub> from the CS model was much greater, highlighting the importance of <del>a proper</del> model treatment.\_From Section 3.1, results of RF10 from the CS model and the results of other flights calculated with homogeneous models, and other flights from homogeneous models,

are more plausible, with <u>the values (mean±standard deviation)</u> of MAC<sub>0A,470</sub> for RF05\_3, RF06\_1, RF06\_2, RF10, and RF11 ranging from  $0.3025\pm0.2734 \text{ m}^2 \text{ g}^{-1}\text{m}^2/\text{g}}$  to  $0.6843\pm0.0812 \text{ m}^2 \text{ g}^{-1}\text{m}^2/\text{g}}$ . Our This result at 470 nm is slightly lower or generally comparable to the value of  $0.31\pm0.09 \text{ m}^2 \text{ g}^{-1}\text{m}^2/\text{g}}$  at 405 nm for highly aged aerosols sampled downwind of ORACLES in CLARIFY 2017 campaign (Taylor et al., 2020), suggesting BrC bleaching during transport considering our result is at a much longer wavelength (Che et al., 2021).

- RF05\_1 and RF05\_2, whose particles are equally or even moreas least as aged than as those in CLARIFY,
  showed unexpectedly high MAC<sub>0A,470</sub> values of 1.184±0.654 m<sup>2</sup> g<sup>-1</sup>m<sup>2</sup>/g and 1.072.38±1.8960 m<sup>2</sup> g<sup>-1</sup>m<sup>2</sup>/g,
  respectively. Although the values of M<sub>0A</sub> for these samples are low (Fig. 6), measurements from all instruments are within the detection limit, which would argue that these values are valid. Wang et al. (2018) used airborne measurements to constrain their global model and found the best MAC<sub>0A</sub> to represent the measurements i i was 1.33 m<sup>2</sup> g<sup>-1</sup>m<sup>2</sup>/g for freshly emitted BB OA at 365 nm. Lin et al. (2017) investigated relatively fresh BB aerosols subject
- 10 to atmospheric processinges during a night-long BB event in an urban environment and got-reported a MAC of 0.9 m<sup>2</sup> g<sup>-1</sup>m<sup>2</sup>/g for water extractable BrC at 470 nm under the peak BB episode. Our values of MAC<sub>0A,470</sub> for RF05\_1 and RF05\_2 are much higher than those for much fresher BB aerosols, which seems unrealistic, as BrC bleaching is expected to occur during transport (Hems et al., 2021; Che et al., 2021). One possible explanation is that secondary BrC formation occurred, perhaps through aqueous-phase chemistry during transport (Hems et al., 2021).
- 15 2021), for-as\_RF05\_1 and RF05\_2, which were sampled in or close to the MBL with higher RHs. Saleh et al. (2013) reported that secondary BrC can be more absorbing than primary BrC at short visible wavelengths; however, to the best of our knowledge, such high MAC<u>like those in RF05\_1 and RF05\_2</u> for secondary BrC have not been documented (Kasthuriarachchi et al., 2020).
- As wWe assume a constant m<sub>eBC</sub> over the investigated spectrum, i.e. 470-660 nm<sub>5</sub>; while the m<sub>eBC</sub> would
  be underestimated at shorter wavelengths if there are some components with strong absorption at short visible but not at long visible wavelengths, the m<sub>eBC</sub> would be underestimated at shorter wavelengths and therefore lead to an overestimation of MAC<sub>0A,470</sub> would be overestimated. Hematite, whose imaginary refractive index ranges from 0 to 1.0 at 470 nm and 0 to ~0.24 at 670 nm (Zhang et al., 2015b; Go et al., 2021), is the second most abundant FeOx in western African dust (Formenti et al., 2014). Particles in RF05\_1 and RF05\_2 experienced approximately over 116 days of transport and thus may have had more opportunities to mix with hematite and therefore lead to an overestimation of MAC<sub>0A</sub>. While lacking measurements on these absorbers to determine the compound, it is difficult to verify our speculation. A modified SP2 has been reported by (Yoshida et al., 2016) that can discriminate black-coloured magnetite and red-coloured hematite; thus, we recommend adding such measurement capabilities in future BB investigations over Africa and the South-East Atlantic region.
- 30 **3.3** Contribution of BrC to total absorption (*R*<sub>BrC</sub>)



Figure 7<sub>2</sub>÷ Contribution of BrC to the total absorption obtained from optical closure approach in this study (boxes,  $R_{BrC,470}$ ) and from the AAE attribution method (orange squares,  $R_{BrC,470,AAE}$ ) at 470 nm. The grey zone indicates the  $R_{BrC,470}$  range of 58-22 15% in this study. Orange shades of  $R_{BrC,470,AAE}$  represent the 10<sup>th</sup> and 90<sup>th</sup> percentile. The orange diamond is the contribution of WSOC (water soluble organic carbon) to the total absorption,  $R_{WSOC,365}$  (Nenes A., personal communication).





Figure 8, Hlustration of the attribution of absorption for RF06\_1. Blue area represents the absorption of BC, calculated with  $m_{BC}$ =1.95+0.79i and  $m_{OA}$ =1.65+0i-using BG model. The upper border of the yellow region represents absorption coefficients calculated with corresponding  $m_{eBC}$  and  $m_{OA}$ -for RF06\_1 using BG model assuming non-absorbing OA; the yellow area illustrates the absorption by absorbing components besides BC and BrC, calculated as the difference between the two absorption coefficients calculated by the BG model using  $m_{eBC}$  and  $m_{BC}$ , respectively. The upper border of the red region is drawn with measured absorption coefficients at 470, 530, and 660 nm. The red region stands for the BrC absorption coefficient calculated from the AAE attribution method with Eq. 134. The 2\_% and 11\_% are the proportions of BrC at 470 nm estimated from the AAE attribution at 470, 530, and 660 nm from the optical closure method. Uncertainties can be found in Table S2 in the supplement.

The contribution of BrC to the total absorption at 470 nm,  $R_{BrC,470}$ , which was calculated using the optical closure method from Eq. 104, is shown in Fig. 7. The uncertainties using various models can be found in Section S3 in the supplement. Similar to the results for MAC<sub>OA,470</sub>, the MG, BG, and VM models yield fairly close results for  $R_{BrC,470}$ , while those from the CS model are higher, consistent with previous studies (Lesins et al., 2002; Taylor et al., 2020). It is noted that the values of  $R_{BrC,470}$  from MG, BG, and VM models at 470 nm for RF10 are the lowest among all flights, with an average of  $3.6\pm 2.6$  %. near zero, which is unrealistic Little confidence is placed on this value for because the following three reasons. Firstly, RF10 was sampled after approximately 40 hours of transport; although there are considerable photobleaching and oxidative whitening of BrC when removed from its source for a day or more, it is unlikely that BrC will be completely bleached with a 40 h aging time (Hems et al., 2021). Secondly, samples-particles in this study with similar aging times, i.e.age in RF05\_3, RF06\_1, RF06\_2, and RF11<sub>3</sub> all showed a much greater higher value of  $R_{BrC,470}$  entribution of BrC to absorption. FinallyIn addition,

Nenes et al. (personal communication) reported the contribution of water-soluble BrC organic carbon (WSOC) to the total absorption at 365 nm of 20<u>% (Fig. 7) for particles measured</u>. As our RF10 was measured 10 min earlier at a similar location in the plume 10 min later than RF10, and with no distinct changes were observed in either the meteorological conditions or aerosol properties.<sub>5</sub> Therefore, we expect the AAE of BrC at 365/470 wavelength

- 5 pair, AAE<sub>365/470,BrC</sub>, derived from  $\sigma_{abs,365,BrC}^{mea}$  and  $\sigma_{abs,470,BrC}^{cal}$  these two results (absorption coefficients of BrC at 365 and 470 nm) to be within the range of ~2-11 those from in literature (Laskin et al., 2015). Since  $\sigma_{abs,365,BrC}^{mea}$  the absorption coefficient of BrC at 365 nm is is unknown, we calculated  $\sigma_{abs,365,extrapolate}$  by extrapolating the measured  $\sigma_{abs,470}^{mea}$  to 365 nm with the measured AAE<sub>470/530</sub> assumed it to be 20% of the measured absorption coefficient at 470 nm and approximated the AAE<sub>365/470,BrC</sub> with the 20% of  $\sigma_{abs,365,extrapolate}$  and the  $\sigma_{abs,470,BrC}^{cal}$  from the set of the set of the set of the the set of the
- 10 <u>homogeneous models. This method</u> and ealculated the AAE of BrC between 365 and 470 nm with the calculated BrC absorption coefficient at 470 nm from homogeneous models. This AAE is much lower than the actual AAE for BrCyields an underestimation of AAE<sub>365/470,BrC</sub> as the BrC absorption coefficient at 365 nm is greatly underestimated; while its value, can still reach 18, but it is still well abovemuch higher than the upper limit of the

AAE range, -2-11, reported in the literature (Laskin et al., 2015), suggesting that the use of homogeneous models

- 15 for RF10 may not be appropriate. As discussed in Section 3.1, the values of *R*<sub>BrC,470</sub> for RF10 from the CS model and RF05\_1, RF05\_2, RF05\_3, RF06\_1, RF06\_2, and RF11 from homogeneous models are more plausible-than those from the CS model. As discussed in Section 3.2, <u>P</u>particles of RF05\_1 and RF05\_2 may contain hematite, in which cases their R<sub>BrC,470</sub> may have beenbe overestimated (Section 3.2). Generally, in our study, the *R*<sub>BrC,470</sub> ranges from ~<u>85-2215</u> % at 470 nm (grey zone in Fig. 7)<sub>a</sub>. This value is lower slightly lower than the contribution of 13-26 % at 500 nm for relatively fresh OA in the wood smoke from Kirchstetter and Thatcher (2012) and
  - generally comparable or slightly higher than. Comparing to the result of ~11 % at 405 nm for highly aged aerosols from CLARIFY 2017, which were sampled downwind of ORACLES with 4-8 d aging time, our result is generally comparable while under a much longer wavelength, implying the bleaching of BrC during transport (Hems et al., 2021; Che et al., 2021).

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We calculated the contribution of BrC to the total absorption using the AAE attribution method  $(R_{BrC,470,AAE})$  and compared it with that from our optical closure method  $(R_{BrC,470})$ . In the AAE attribution method, the absorption coefficient of BC at the investigated wavelength (470 nm in our case) is determined by extrapolating the absorption coefficient at a longer wavelength (530 nm in our case) with the AAE of BC. For example, in this study, the average AAE at 530/660 wavelength pair (AAE<sub>530/660</sub>) for all flights is 0.93±0.16, which is just below unity and within the range expected for BC\_(Liu et al., 2018; Taylor et al., 2020). Thus, we assume that the absorption coefficients at 530 nm ( $\sigma_{abs,530}^{mea}$ ) and AAE<sub>530/660</sub> have no significant contribution from BrC and that the AAE for BC is independent of wavelength. The contribution of BrC to the total absorption,  $R_{BrC,470,AAE}$ , can then be calculated in below.

$$R_{Brc,470,AAE} = 1 - \frac{\sigma_{abs,470}^{cal,AAE}}{\sigma_{abs,470}^{mea}}$$
(12)

$$\sigma_{abs,470}^{cal,AAE} = \sigma_{abs,530}^{mea} (\frac{530}{470})^{AAE_{530/660}}$$
(13)

$$AAE_{530/660} = -\frac{\ln(\sigma_{abs,530}^{mea}/\sigma_{abs,660}^{mea})}{\ln(530/660)}$$
(14)

 $\sigma_{abs,470}^{mea}$ ,  $\sigma_{abs,530}^{mea}$ , and  $\sigma_{abs,660}^{mea}$  represent the measured absorption coefficients at 470, 530, and 660 nm, respectively.  $\sigma_{abs,470}^{cal,AAE}$  is the absorption coefficient calculated from the AAE attribution method at 470 nm. Actually, this assumption is not always true as the AAE varies with both the wavelength and core size and coating thickness of BC particles (Lack and Cappa, 2010; Wang et al., 2016). The results of  $R_{BrC,470,AAE}$  are shown in Fig.

5 7, with averages of all flights below 5\_%, roughly a factor of three smaller than the  $R_{BrC,470-}$  from the optical closure approach.

The AAE attribution method is widely used for its simplicity, while there is one caveat that need to be noted when applying this method. The absorption coefficients and AAE, in this case,  $\sigma_{abs,530}^{mea}$ ,  $\sigma_{abs,660}^{mea}$ , and AAE<sub>530/660</sub>, should be for BC only, without contributions from other absorbers, such as BrC (Lack and Cappa,

- 10 2010). However, it is usually very difficult to completely exclude the impact of these absorbers in the calculation. extreme underestimation The homogeneous models are usually reported to overestimate BC absorption (Zhang et al., 2015a; Fierce et al., 2017b; Taylor et al., 2020) and hence underestimate BrC absorption; while the values of the contribution of BrC determined from by the AAE attribution method areis even lower than those from the homogeneous methods, which we suspect might be partly attributed to the impact of absorbers besides BC.
- 15 <u>Therefore, we evaluated the absorption attribution at detected wavelengths</u> <u>mainly due to the fact that the AAE</u> and absorption coefficients used in this method are not derived from BC alone, but include contributions from other absorbing substances. by tTakinge RF06\_1 as an example.<sub>5</sub> tThe contribution of BrC derived from the AAE attribution method is 2.% for RF06\_1, approximately a factor of five smaller than that from our the optical closure method with the BG model, 11\_% (Fig. 7 and 8). As illustrated in Fig. 8, σ<sup>mea</sup><sub>abs,530</sub> the absorption coefficient at 530
- 20 nm used in Eq. 14 and 15 includes the contribution of BrC, which accounts for 4\_% of the total absorption, and that of absorbers beyond BC and BrC, which accounts for 21\_%. Note this attribution is based on the BC absorption coefficient calculated from the BG model with  $m_{BC}$ =1.95+0.79i, which will vary with the value of  $m_{BC}$ yields an upper bound of BC absorption (Liu et al., 2021). All calculated parameters are subject to uncertainties as discussed in Section S3 in the supplement. If we remove the 4\_% BrC from  $\sigma_{abs,530}^{mea}$  in Eq. 1413, the BrC contribution to
- 25 <u>absorption at 470 nm</u> would increase to 6\_%, two times larger than 2\_%, indicating that a substantial impact of <u>BrC on the result of the AAE attribution method</u> even though <u>BrC-the BrC absorption may only</u> exist as a small portion at the long wavelength, its impact on the AAE attribution method can be substantial. Thus, we recommend that application of any optical properties-based attribution method to use absorption coefficients at the longest

possible wavelength to minimize the influence of BrC, and in the meanwhile, to account for potential contributions from other absorbing materials.

## **4** Conclusions

- We investigated the contribution of BrC to the total absorption of BB aerosols with different models
  utilizing measurements from ORACLES 2018 field campaign. An effective BC refractive index, *m*<sub>eBC</sub>=n<sub>eBC</sub>1.95+i*k*<sub>eBC</sub>, that was constrained by scattering and absorption measurements to account for all absorbing components at 660 nm, was introduced in this study. Most of the values derived for *k*<sub>eBC</sub> were greater than the commonly used *k*<sub>BC</sub> values, suggesting contributions from absorbing materials besides BC at 660 nm. TEM-EDX single particle analysis further suggests that these absorbers might include FeOx that absorbs at long visible
  wavelengths, i.e. magnetite, as Fe is only present for flights with large values of *k*<sub>eBC</sub>. RF05\_3 yielded the largest values of *k*<sub>eBC</sub>, approximately <u>640%-1040</u>% greater than that of other flights, implying a greater contribution of absorbers besides BC to the total absorption at 660 nm than other flights, which <u>could might</u> be due to a larger amount of magnetite in these samples or the result of possible absorbing titanium compounds, which are present on this filter.
- 15 As refractive indices of BC and magnetite are generally constant over the range of wavelengths considered, i.e. 470 and 660 nm, we assumed  $m_{eBC}$  to be independent of wavelength and calculated MAC<sub>OA</sub> and the contribution of BrC to total absorption at 470 nm, R<sub>BrC,470</sub>, using with different models. The values of  $MAC_{OA.470}$  and  $R_{BrC,470}$  from the three homogeneous mixing models were fairly close to each other, while those from the CS model were much higher, underscoring the importance of proper-model treatment. The values of 20 MAC<sub>0A.470</sub> and contribution of BrC to the total absorption, R<sub>BrC,470</sub>, from the CS model for RF10 and from homogeneous models for other flights are found to be more plausible, based on the morphology of BC-containing particles from TEM imagesanalysis, MR values from SP2, and the validation of the contribution of WSOC results to the total absorption at 365 nm. The values of  $R_{BrC,470}$  ranged from  $\sim 58-2215$  %, with MACOA.470 varying between  $0.30\pm0.270.25\pm0.34$  m<sup>2</sup> g<sup>-1</sup> and  $0.68\pm0.080.43\pm0.12$  m<sup>2</sup> g<sup>-1</sup> for the various flights. Values from our results 25 were generally comparable or slightly higher than the with  $MAC_{OA,470}$  of  $0.31\pm0.09$  m<sup>2</sup> g<sup>-1</sup> and  $R_{BrC}$  of ~11 % at 405 nm for highly aged aerosols from CLARIFY 2017, which were sampled downwind of ORACLES with 4-8 d aging time, while under a much longer wavelength, i.e. 470 nm, implying the bleaching of BrC during transport. High MAC<sub>0A,470</sub> values of  $1.84\pm0.64$   $1.18\pm0.54$  and  $2.38\pm1.891.07\pm1.60$  m<sup>2</sup> g<sup>-1</sup> were observed for RF05 1 and RF05 2, respectively, which is suspected to be due tocaused by the presence of hematite, an abundant FeOx in 30 African dust, which may have mixed with the BC particles in RF05 1 and RF05 2 during the 11-6 d transport. Measurements of a modified SP2 that can distinguish between hematite and magnetite and obtain FeOx concentrations is are highly recommended.

Contribution of BrC to the total absorption obtained from the AAE attribution method, R<sub>BrC.470,AAE</sub>, were calculated and compared to that from our optical closure method. R<sub>BrC,470,AAE</sub> is generally <5\_%, approximately a factor of three smaller than those from our optical closure method. From the absorption attribution of RF06 1 as an example, Absorption attribution using the Bruggeman mixing Mie model suggest a minor BrC contribution of

5 4 % at 530 nm for RF06 1, and its removal would triple the BrC contribution to the total absorption at 470 nm from 2 % to 6 % obtained using the AAE attribution method, suggesting a substantial impact of BrC to the result of AAE attribution method even though the BrC may only exist as a small portion. we found this underestimation is mainly due to the influence of absorbers besides BC in parameters used in this approach. Thus, when applying optical properties-based attribution methods, it is recommended to use the absorption coefficient at the longest possible wavelength to minimise the influence of BrC and to account for other absorbers in the meanwhile.

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