



1 Scattering and Absorption Cross-sections of Atmospheric Gases in the

2 Ultraviolet-Visible Wavelength Range (307 - 725 nm)

- 3 Quanfu He¹, Zheng Fang¹, Ofir Shoshamin², Steven S. Brown^{3,4}, Yinon Rudich^{1,*}
- 4 ¹ Department of Earth and Planetary Sciences, Weizmann Institute of Science, Rehovot 76100,
- 5 Israel
- 6 ² Department of Environmental Physics, Institute for Biological Research, Ness-Ziona 74100,
- 7 Israel
- 8 ³ Chemical Sciences Division, Earth System Research Laboratory, National Oceanic and
- 9 Atmospheric Administration, 325, Broadway, Boulder, CO 80305, USA
- ⁴ Department of Chemistry, University of Colorado, 216 UCB, Boulder, CO 80309, USA
- 11 Correspondence to: Yinon Rudich (<u>vinon.rudich@weizmann.ac.il</u>)
- 12

13 Abstract

Accurate Rayleigh scattering and absorption cross-sections of atmospheric gases are essential for 14 understanding the propagation of electromagnetic radiation in planetary atmospheres. Accurate 15 extinction cross-sections are also essential for calibrating high finesse optical cavities and 16 differential optical absorption spectroscopy and for accurate remote sensing. In this study, we 17 measured the scattering and absorption cross-sections of carbon dioxide, nitrous oxide, sulfur 18 hexafluoride, oxygen, and methane in the continuous wavelength range of 307-725 nm using 19 20 Broadband Cavity Enhanced Spectroscopy (BBCES). The experimentally derived Rayleigh 21 scattering cross-sections for CO₂, N₂O, SF₆, O₂, and CH₄ agree with refractive index-based 22 calculations, with a difference of 1.5% and 1.1%, 1.5%, 2.9%, and 1.4% on average, respectively. 23 The O_2 - O_2 collision-induced absorption and absorption by methane are obtained with high precision at the 0.8 nm resolution of our BBCES instrument in the 307–725 nm wavelength range. 24 25 New dispersion relations for N₂O, SF₆, and CH₄ were derived using data in the UV-vis wavelength range. This study provides improved refractive index dispersion relations, n-based Rayleigh 26 27 scattering cross-sections, and absorption cross-sections for these gases.





28 **1. Introduction**

The dominant interactions of gas-phase molecules with light in Earth's atmosphere can be divided 29 into absorption, where the light energy is converted to internal energy and generally (at 30 atmospheric pressures) transferred to the surrounding environment as either as heat or as 31 photoemission, and light scattering where the gases redistribute the light energy in the atmosphere. 32 The knowledge of light extinction (scattering + absorption) by gases is essential for predicting the 33 radiative transfer in the atmospheres of the Earth and other planets. In addition, the light extinction 34 by gases is widely used for determining the effective optical pathlength of high-finesse optical 35 cavities that measure trace gases and aerosols (Washenfelder et al., 2013; Washenfelder et al., 36 2008; Wilmouth and Sayres, 2019; Jordan et al., 2019) and for Differential Optical Absorption 37 38 Spectroscopy (DOAS) to infer information about the light extinction properties of aerosols and clouds in the open atmosphere (Baidar et al., 2013; Platt and Stutz, 2008). 39

The interaction of light with a wavelength much larger than the size of a molecule/particle gives rise to the scattering of light, which is known as Rayleigh scattering(Strutt, 1899). Rayleigh scattering accounts for scattering, local field effects (Lorentz–Lorenz) (Strutt, 1920) as well as depolarization from the non-sphericity of molecule/particles (King correction factor) (King and Eve, 1923; Strutt, 1918). For gas with known refractive index (n_v) and King correction factor ($F_k(v)$), the wavelength-dependent Rayleigh scattering cross-section (σ_v , cm² molecule⁻¹) can be calculated as follows (Sneep and Ubachs, 2005):

47
$$\sigma_{\nu} = \frac{24\pi^{3}\nu^{4}}{N^{2}} \left(\frac{n_{\nu}^{2}-1}{n_{\nu}^{2}+2}\right)^{2} F_{k}(\nu)$$
(1)

where N is the number density of the gas (molecules cm^{-3}) and v is the wavenumber of the light 48 (cm⁻¹). Note that the cross-section contains the gas number density but is not in fact dependent on 49 50 the number density since the refractive index also appears in the expression. This *n*-based method is an advantageous approach for calculating Rayleigh scattering cross-sections, but it is vital to 51 note that the accuracy of the calculated cross-sections depends on the experimentally-determined 52 53 refractive index and the King correction factors. In particular, cautions should be used when 54 applying a dispersion formula derived from measurements in one wavelength region to calculate 55 Rayleigh scattering cross-sections in a different wavelength range.





56 Direct experimental measurement of Rayleigh scattering cross-sections is essential given the 57 potential uncertainties in *n*-based calculations. While measurements of the King correction factors 58 and refractive index for gases are well known from the literature (Cuthbertson and Cuthbertson, 1932; Leonard, 1974; Strutt, 1920; Vukovic et al., 1996; Hohm, 1993), there are only a few direct 59 measurements of Rayleigh scattering cross-sections (Ityaksov et al., 2008a, b; Jordan et al., 2019; 60 Naus and Ubachs, 2000; Sneep and Ubachs, 2005; Thalman and Volkamer, 2013; Thalman et al., 61 2014; Wilmouth and Sayres, 2019; He et al., 2018; Fuchs et al., 2009), especially measurements 62 with a continuous spectrum from ultraviolet to visible. 63

Rayleigh scattering cross-section measurements were previously performed at a single wavelength 64 (e.g., 458 nm, 532 nm, 632.8 nm) using Nephelometry (Shardanand and Rao, 1977) and cavity-65 66 ring down spectroscopy (CRDS) (Ityaksov et al., 2008a, b; Naus and Ubachs, 2000; Sneep and Ubachs, 2005; He et al., 2018). More recently, advanced Broadband Cavity Enhanced 67 Spectroscopy (BBCES) was used to determine the Rayleigh scattering cross-sections of gases such 68 69 as Ar, CO₂, O₂, SF₆, and CH₄. The BBCES technique enables the measurement of Rayleigh 70 scattering cross-sections over a broad wavelength range. Thalman et al. (2014) performed measurements over selected wavelength regions between 350 and 660 nm using six BBCES 71 cavities for N_2 , Ar, and O_2 . The BBCES were calibrated with He and N_2 using Rayleigh scattering 72 cross-sections calculated using their refractive index and from cavity-ring down measurements, 73 respectively. They found a good agreement with n-based values to within 0.2±0.4%. Recent studies 74 75 using BBCES with 30 nm spectral range were also used for Rayleigh scattering cross-section measurement in the UV wavelength region and demonstrated excellent agreement with n-based 76 values for Ar and CO₂ (Wilmouth and Sayres, 2020, 2019). Recently, Rayleigh scattering cross-77 78 sections for CO₂ were measured using BBCES at visible wavelengths between 400 and 650 nm, and agreement with *n*-based values was within 2.4% on average. To the best of our knowledge, 79 there is no direct continuous wavelength measurements of extinction cross-sections of gases that 80 covers the ultraviolet across the entire visible range (300–725 nm) as shown in Table 1. 81

In this study, we used a recently-developed BBCES instrument to measure the extinction crosssections of CO₂, N₂O, SF₆, O₂, and CH₄ continuously across the wavelength region 307–725 nm. All of the measurements were done at a single pressure to eliminate effects due to alignment. This requires the use of two gases with different Rayleigh cross-sections for the calibration of the BBCES instrument since the reference state is not vacuum. In this study, He and N₂ were used to





calibrate the system. By using the *n*-based calculated Rayleigh scattering cross-sections of He and N₂ to calibrate the path length of the optical cavity, the other cross-sections can be determined relative to the difference between these two gases. We report high accuracy Rayleigh scattering cross-sections for all five gases and compared our results with previous *n*-based values. New dispersion relations for N₂O, SF₆, and CH₄ are derived by incorporating data obtained by this study, extinction cross-section data in the deep UV, and previously available scattering cross-section data in the visible wavelength range.

94

95 2. Methods

96 2.1 Extinction measurement using BBCES

The BBCES systems used in this study are analogous to our previous studies (He et al., 2018; 97 98 Washenfelder et al., 2016; Bluvshtein et al., 2016). Briefly, our BBCES consists of two channels, one in the UV (BBCES_{UV}, 307-350 nm) and one in the UV-vis range (BBCES_{Vis}, 338-725 nm). 99 The two channels of the BBCES share a laser-driven Xenon arc lamp source (LDLS EQ-99CAL, 100 101 Energetiq Technology, Inc., MA, USA) coupled with a high transmission UV-Vis optical fiber 102 from which the light is collimated and focused (BBFIBERX-600-1M, Energetiq Technology, Inc., 103 MA, USA). The light source was purged with high purity N_2 and cooled by an aluminum block (with 15°C circulating water inside) to maintain stable optical power output. The UV light from 104 105 the fiber is reflected by a low-pass dichroic mirror and filtered into the BBCES_{UV} channel, which 106 has a cavity with two 2.5 cm diameter, 1 m radius of curvature mirrors, with manufacturer's reported reflectivity of 0.9995 (per pass loss = 500 parts per million, ppm) at the nominal center 107 wavelength of 330 nm (Advanced Thin Films, Boulder, USA). The transmitted UV-vis light from 108 109 the beam splitter is reflected and filtered into the BBCESvis channel consisting of two 2.5 cm, 1 m radius of curvature mirrors (FiveNine Optics, USA) with manufacturer's reported reflectivity 110 above 0.9993 (loss < 700 ppm), see Figure S1. The light emerging through the rear mirror of the 111 cavity was focused using a 0.1 cm F/2 fiber collimator (74–UV, Ocean Optics, Dunedin, FL, USA) 112 into a high transmission UV-vis optical fiber which directs the light into a high-performance 113 spectrometer (QEPro, Ocean Insight, USA). Before gas measurement, the wavelength of the 114 115 spectrometer was calibrated using an HG-1 mercury argon calibration light source (Ocean Insight, USA) within the wavelength range of 302.15–727.29 nm. During these experiments, a 300 line 116





- 117 mm⁻¹ grating and a 200 μm entrance slit width were used. The CCD array is a back-illuminated
- detector with 1024×56 pixels (Hamamatsu S7031-1006, Japan) thermo-electrically cooled to -10 °C
- to reduce thermal noise. Individual spectra at a wavelength resolution of 0.8 nm were acquired
- 120 with 3.0 s integration time, and a total of 150 spectra were recorded during each measurement.
- 121 During the extinction measurements, the entire 94.0 ± 0.1 cm pathlength between the mirrors was
- filled with He, N₂, CO₂, N₂O, SF₆, or CH₄. The gases were obtained from several vendors (Airgas,
- Linde) with the following purities: He, 99.995%; N₂, 99.999%; N₂O, 99.999%, CO₂, 99.999%;
- 124 SF₆, 99.999%; CH₄, 99.9995%.

The reflectivity of the mirrors (R(λ)) can be determined as a function of wavelength (λ) by taking into account the difference in the extinction due to known literary data of Rayleigh scattering coefficient (α_{Ray}^{gas}) by two different gases such as N₂ ($\alpha_{Ray}^{N_2}(\lambda)$) and He ($\alpha_{Ray}^{He}(\lambda)$) (Washenfelder et al., 2008).

129
$$\frac{1-R(\lambda)}{d} = \frac{I_{N_2}(\lambda) \left(\alpha_{Ray}^{N_2}(\lambda)\right) - I_{He}(\lambda) \left(\alpha_{Ray}^{He}(\lambda)\right)}{I_{He}(\lambda) - I_{N_2}(\lambda)}$$
(2)

where d is the length of the cavity filled by the gas. In this study, the studied gas filled the entire 130 length of the cavity (94.0 \pm 0.1 cm) since no purge flows were used. I_{gas} is the light intensity 131 measured by filling the cavity with high purity N₂ ($I_{N_2}(\lambda)$) and He ($I_{He}(\lambda)$). Rayleigh scattering 132 (α_{Ray}^{gas}) is the combined product of Rayleigh scattering cross-section (σ) and the gas number density 133 134 (N) during the measurements. Rayleigh scattering cross-sections of N₂ and He are calculated using the data in Table 1. Figure S1 shows typical examples of light intensity when the BBCES cavities 135 are filled with pure N2. Reflectivity measurements were repeated every three sample measurements 136 137 to track the stability of the system.

Once the reflectivity is determined, it is possible to calculate the wavelength-dependent extinction cross-sections of other gases ($\sigma(\lambda)$) as follows:

140
$$\sigma(\lambda) = \left[\left(\frac{1 - R(\lambda)}{d} \right) \left(\frac{I_{He}(\lambda) - I_{gas}(\lambda)}{I_{gas}(\lambda)} \right) + \left(\frac{I_{He}(\lambda)}{I_{gas}(\lambda)} \right) \left(\alpha_{Ray}^{He}(\lambda) \right) \right] / N$$
(3)

141 Where *N* is the number density of the gas during the measurements, and $I_{gas}(\lambda)$ is the light intensity 142 when a target gas fills the cavity. During our experiments, the purge flow of the high reflection





143 mirrors was shut down to ensure that the cavity was filled with target gas completely. To measure 144 the extinction cross-sections of CO₂, N₂O, and SF₆, the cavity is filled with pure target gas. Mass flow controller controlled O₂/CH₄ flow was mixed with He in a 2 m Teflon tube ($\Phi = \frac{1}{4}$ inch) to 145 generate a gas mixture with total flow rate of 500 mL min⁻¹. For O_2 experiments, measurements 146 were performed for O_2 + He mixtures by varying the O_2 percentage between 10% and 100% with 147 a 10% step. The CH₄, measurements were performed for CH₄ + He mixtures with CH₄ percentage 148 149 ranges between 10% to 100% with a 10% step. Additional measurements were also performed for 150 15%, 25%, 35%, and 45% CH₄.

151 2.2 Extinction measurements using cavity-ring down systems (CRDS) at 404 nm and 662 nm.

To obtain independent measurements for the extinction cross-sections and to cross-validation of 152 the BBCES technique, we conducted CRD measurements at two fixed wavelengths of 404 nm and 153 662 nm. CRDS is a highly sensitive technique and uses a different measurement principle than 154 BBCES. The CRDS measured the decay rate of light due to extinction rather than an absolute 155 156 absorbance (as in the BBCES) and thus immune to shot-to-shot source light fluctuations. A detailed description of the CRD method for light extinction measurement can be found in 157 Bluvshtein et al. (2016) and He et al. (2018). Briefly, diode lasers (110 mW 404 nm diode laser, 158 iPulse, Toptica Photonics, Munich, Germany; 120 mW 662 nm diode laser, HL6545MG, Thorlabs 159 160 Inc., NJ, USA) are used as the light source of these CRDS. The 404 nm and 662 nm lasers are modulated at 1383 Hz and 500 Hz with a 50% duty cycle. The diode lasers are optically isolated 161 162 by quarter waveplates $(1/4 \lambda)$ and polarizing beam splitters to prevent damage to the laser head by back reflections from the highly reflective CRDS mirror. The back-reflected light beam is directed 163 164 into a photodiode, which serves as an external trigger source. Light transmitted through the back mirror of the cavity is collected by an optical fiber and detected by a photomultiplier tube (PMT), 165 166 which samples at a rate of 10 to 100 MHz. The time-dependent intensity data is acquired with a 100MHz card (PCI-5122, National Instruments, USA) and processed by data acquisition software 167 168 in Labview. An exponential curve is fitted to each intensity decay data set (Figure S2). Over 1000 decay time measurements are monitored and averaged on a second basis. The residual of the fit 169 170 for the averaged intensity decay is obtained and further normalized to the averaged intensity. The derived relative residuals (Figure S2) show no apparent structure with other time constants, 171 172 validating the application of CRDS as a good measure of extinction. The resultant 1 Hz decay time





- is averaged over one measurement duration of five minutes with standard error as the measurementuncertainty.
- 175 All of the CRDS measurements were performed under room temperature and pressure downstream 176 from the BBCES instrument. The gas temperature (K-type thermocouple) and cavity pressure 177 (Precision Pressure Transducer, Honeywell International Inc., MN, USA) were recorded for gas 178 number density (*N*) calculation. During the CRDS measurements, the full cavity was filled with 179 the investigated gases (He, CO₂, N₂O, SF₆, O₂, CH₄, or gas mixtures (O₂ + He and CH₄ + He)). 180 The extinction cross-section ($\sigma(\lambda)$) of the studied gas was measured relative to that of He and was 181 calculated by equation (4):

182
$$\sigma(\lambda) = \frac{L}{clN} \left(\frac{1}{\tau_{gas}} - \frac{1}{\tau_{He}} \right) + \sigma_{He}$$
(4)

183 Where *L* is the total length of the cavity (*l*), *c* is the speed of light, and τ_{gas} and τ_{He} are the ring-down 184 time of the cavity when it is filled by target gas or by the reference gas, He.

185 2.3 Data processing

For comparison, the scattering cross-sections of the gases investigated in this study were also calculated with Equation (1) based on the refractive index and the King correction factors available in the literature that are listed in Table 1. The King correction factors are taken as unity for mono-atomic molecules and spherical molecules (with regards to the depolarization) but deviates for non-spherical molecules. For the 307–725 nm wavelength range of this study, the *n*-based calculated Rayleigh scattering cross-sections from largest to smallest are SF₆, N₂O, CO₂, CH₄, N₂, O₂, and He.

The extinction of O_2 + He mixtures (α_{O_2+He}) consists of the extinction by O_2 (α_{O_2}) and He (α_{He}), and the O_2-O_2 collision-induced absorption ($\alpha_{O_2-O_2}$). The extinction of O_2 and He is a combined product of extinction cross-section (σ_{gas}) and gas number density (N_{gas}). Thus α_{O_2+He} can be described with the following equation:

197
$$\alpha_{O_2+He} = \sigma_{O_2-O_2} \times N_{O_2}^2 + \sigma_{O_2} \times N_{O_2} + \sigma_{He} \times N_{He}$$
(5)

Where N_{O_2} and N_{He} are the number density of the O₂ and He in the cavities. Performing a 2rd order polynomial fit to the extinction obtained by the BBCES with respect to the gas number density





thus yields the extinction cross-section of O_2 and the O_2 - O_2 collision-induced absorption (CIA) cross-section.

In addition to the results from 2rd order polynomial fitting, we also used data from pure O₂ 202 measurement to calculate the extinction by O_2 and by CIA of O_2 -O₂. The real refractive index of 203 204 $O_2(n_{O_2})$ derived from extinction data measured in the wavelength regions where there is no absorption was fitted using the generalized expression of $(n_{O_2} - 1) \times 10^8 = A + \frac{B}{C - v^2}$. Based on 205 the refractive index, the scattering cross-sections of O2 in the wavelength range of 307-725 nm 206 207 were further calculated. By subtracting the scattering cross-section of O_2 from the measured total extinction, we derived the CIA of O₂-O₂. However, the O₂ absorption bands at 580, 630, and 690 208 nm overlaps with those of O_2 - O_2 collisions. Additional corrections are thus needed to split the 209 absorption by O_2 and O_2 - O_2 collision, which is out of the scope of this study. 210

211 Methane has weak vibrational overtone absorption in the UV-vis wavelength range that is 212 comparable to or greater than its Rayleigh scattering. Previous high-resolution spectroscopy 213 studies have identified smooth and unstructured absorption bands across the UV-visible range 214 (Giver, 1978; Smith et al., 1990). The spectral features are substantially broader than 0.8 nm, thus 215 the absorption by CH₄ can be measured by our BBCES. The measured extinction coefficients of 216 CH₄+He mixtures (α_{CH_4+He}) are linearly correlated with the number concentration of the CH₄ (N_{CH_4}) 217 as described by the following equation:

218
$$\alpha_{CH_4+He} = \sigma_{CH_4} \times N_{CH_4} + \sigma_{He} \times N_{He}$$
(6)

A linear fit was used for deriving the extinction cross-section of CH₄. The absorption between 300 and 400 nm is negligible as compared to the Rayleigh scattering. Thus extinction data in this UV wavelength range were used to calculate the real part of the refractive index of CH₄ which was further fitted utilizing the expression of $(n_{CH_4} - 1) \times 10^8 = A + \frac{B}{C - v^2}$. By applying this dispersion relation, the Rayleigh scattering cross-sections in the entire wavelength range of 307– 725 nm were derived. Finally, the CH₄ absorption cross-sections were calculated by subtraction of the scattering cross-section from the extinction cross-section.

226 2.4 Error Propagation for BBCES





227 The uncertainty for BBCES measurements can be assessed by the propagation of the errors 228 associated with the measurements. The pressure ($\pm 0.01\%$), temperature ($\pm 0.1\%$) and cavity length $(94.0 \pm 0.1 \text{ cm})$ are combined with the Rayleigh cross-section uncertainties for N₂ (±1%) as well 229 as uncertainty in the measurements of the spectral signal by the spectrometer ($\ll 0.2\%$) to get an 230 overall relative uncertainty for the mirror reflectivity curve of $\pm 1.03\%$. This uncertainty is further 231 232 propagated to the target gas by consideration of the uncertainties of pressure, temperature, and spectral intensity of the target gas measurements. The overall $1-\sigma$ uncertainty of the gas extinction 233 cross-section is 1.1%. The precision of the mass flow controllers is 0.5 mL min⁻¹. When the total 234 flow rate is 500 mL min⁻¹, the resulted uncertainty of the gas concentration (10-100%) varies from 235 0% to 1.0%. Thus, the overall 1- σ uncertainty of extinction coefficients measured for CH₄+He and 236 237 O₂+He varies from 1.1% to 1.5%. The detailed wavelength-dependent uncertainties were calculated due to the wavelength-dependence of the spectral intensity. The results are shown and 238 discussed in later sections. 239

240

241 3 Results and Discussion

242 **3. 1 Performance of the optical system**

The reflectivity of the cavity mirrors, measured across the entire range using the difference in 243 Rayleigh scattering of N_2 and He, was very stable throughout the experiments. The measured 244 mirrors reflectivity curves are shown in Figure S1. The mean peak reflectivity of the BBCES_{UV} 245 246 mirrors was 0.999328±0.000006 (672±6 ppm) at 330 nm, with a corresponding effective optical pathlength of 1.40±0.01 km. The reflectivity curve of the BBCES_{Vis} is much more structured, with 247 reflectivity ranging between 0.999224 ± 0.000010 and 0.9999550 ± 0.0000006 (776±10 ppm > loss > 248 249 45±0.6 ppm) over a wide wavelength range of 338–725 nm. The reflectivity of the BBCES_{Vis} is much higher than that of our previous system (He et al., 2018) and also covers a much broader 250 251 wavelength range. Thus the effective pathlength of the BBCESVis varies between 1.3 and 20.4 km, guaranteeing a high sensitivity of the extinction measurement. The mean uncertainty in the 252 253 effective pathlength across the measured wavelengths as determined from the mirror reflectivity was $\pm 1.03\%$, which is predominantly due to the uncertainty in the Rayleigh scattering cross section 254 255 for N₂ derived from *n*-based calculation.

256 3. 2 Rayleigh scattering cross-sections of CO₂, N₂O, SF₆.





257 Figure 1 shows the extinction cross-sections of CO_2 , N_2O , and SF_6 measured by the BBCES. The 258 extinction cross-sections of these gases monotonically decrease with increasing wavelength, and no absorption (i.e., no structured extinction larger than the smoothly varying Rayleigh curve) is 259 observed in the wavelength range of 307–725 nm, indicating that the measured extinction is due 260 solely to the Rayleigh scattering of these gases. The wavelength-dependent relative standard 261 deviations of the measurements for each gas are shown in Figure 1d. The mean $1-\sigma$ uncertainty of 262 the reported cross sections for all three gases across the 307–725 nm wavelength range is 1.5% for 263 264 CO_2 , 1.1% for N₂O, and 1.5% for SF₆. As mentioned above, the derived uncertainty originates 265 predominantly from the uncertainty in the N₂ Rayleigh scattering cross-section. Uncertainty in the Rayleigh cross-sections of each gas varies with wavelength and generally tracks the light intensity 266 spectra, which is a combined product of light source spectrum and the mirror reflectivity profile. 267 The uncertainty is much higher when the transmitted light intensity is low (Figure S1). 268

The BBCES measured Rayleigh scattering cross-sections for these three gases agree well with those obtained by our CRDS operating at 404 nm and 662 nm, with deviations smaller than 1.6%. Table 2 listed the Rayleigh scattering cross-sections at several wavelengths obtained by the BBCES measurements (Exp) and by the theoretical calculations using the refractive index and $F_k(v)$ values from Table 1 (*n*-based). The relative differences between these two sets of results are within 1.4%.

Figure 1a-c shows a comparison of the measured Rayleigh scattering cross-sections for CO₂, N₂O, 275 276 and SF₆ with *n*-based calculations and with previous experimental results from the literature. There are a few measurements for the Rayleigh scattering cross-sections for CO₂ which cover a wide 277 spectral range (Jordan et al., 2019; Shardanand and Rao, 1977; Sneep and Ubachs, 2005; Wilmouth 278 and Sayres, 2019; He et al., 2018). There are fewer Rayleigh scattering measurements for N₂O and 279 280 SF_6 in the studied wavelength range. The measured Rayleigh scattering cross-sections for CO_2 , 281 N_2O_1 , and SF_6 are in excellent agreement with *n*-based calculation. The wavelength-dependent difference of our experimentally derived Rayleigh scattering cross-sections with n-based 282 283 calculations are shown in Figure 1e. The mean ratios of our measurements to the n-based values 284 for the entire wavelength range of 307-725 nm are 1.00 ± 0.01 , 0.99 ± 0.01 , and 1.01 ± 0.01 for CO₂, N_2O , and SF_6 , respectively. Notably, while our results for N_2O agree well with the *n*-based 285 calculations, previous results obtained by CRDS at 532 nm (Sneep and Ubachs, 2005) and by 286 absorption spectroscopy in the wavelength of 300-315 nm (Bates and Hays, 1967) do not agree 287





well with the n-based calculations. The measurements between 300 and 315 nm were first published by Bates and Hays (1967), who obtained the results from a doctoral thesis. However, the results from our BBCES system are in good agreement with the *n*-based calculations and with experimental results from independent CRDS measurements, thus increasing the confidence in our measured values.

3.3 Scattering and absorption cross-sections of O2.

The UV-vis spectra of gas-phase molecular oxygen are characterized by discrete structured 294 absorption bands due to the electronic transition $(b^1 \sum_{q=1}^{+} (v' = 1/2/3) \leftarrow \sum_{q=1}^{-} (v'' = 0))$ of O_2 295 monomer, broader unstructured CIA of O2-O2, and structured dimer bands from the bound van 296 297 der Waals O₂ dimer (Newnham and Ballard, 1998). Under atmospheric conditions, the O₂–O₂ CIA bands are frequently described as "O₄" bands, although absorption by O₂ dimer is thought to be 298 significant only under very low-temperature conditions (Thalman and Volkamer, 2013; Long and 299 Ewing, 1973). Within the wavelength range investigated in this work, the molecular oxygen B 300 band at 688 nm $(b^1 \sum_{g}^+ (v' = 1) \leftarrow X^3 \sum_{g}^- (v'' = 0)), \gamma$ overtone band at 629 nm $(b^1 \sum_{g}^+ (v' = 2) \leftarrow 1)$ 301 $X^3 \sum_g^- (\nu'' = 0))$, and δ overtone band at 580 nm $(b^1 \sum_g^+ (\nu' = 3) \leftarrow X^3 \sum_g^- (\nu'' = 0)$ overlap with 302 O₂-O₂ CIA bands of ${}^{1}\Sigma_{g}^{+}(\nu = 1)$, ${}^{1}\Delta_{g} + {}^{1}\Delta_{g}(\nu = 0)$, and ${}^{1}\Delta_{g} + {}^{1}\Delta_{g}(\nu = 1)$, respectively. 303 These absorption bands can only be resolved by a high-resolution spectroscopic technique. 304 305 Absorption cross-sections of the B, γ , and δ bands were convoluted from the HITRAN database 306 (Gordon et al., 2017) by considering the temperature, pressure, and instrument's wavelength 307 resolution. The wings of the oxygen lines also show a quadratic dependence on the pressure due to pressure broadening. However, due to the minimal O_2 absorption contribution below 680 nm 308 and the low instrument wavelength resolution, the extinction cross-section of the O₂ monomer can 309 310 be treated as linearly correlated with the O₂ concentration. Moreover, the O₂–O₂ CIA cross-section is correlated with the square of the O_2 concentration. Therefore, these cross-sections can be 311 312 retrieved from measurements at different O₂ concentrations. Due to the discrete structured absorption bands and the instrument's wavelength resolution, the range of absorption cross-313 314 sections spans several orders of magnitude within the spectral response of the instrument, limiting the relevance of the absorption cross-sections for other researchers. These results are not further 315 discussed here. However, the data for broader unstructured CIA of O2-O2 are still useful for 316 317 various applications.





318 Figure 2 shows the wavelength-dependent extinction coefficients of O₂+He mixtures. He was used 319 in these experiments to minimize extinction contributions from Rayleigh scattering. Nine absorption peaks centered at 344 nm (CIA), 360 nm (CIA), 380 nm (CIA), 446 nm (CIA), 477 nm 320 321 (CIA), 532 nm (CIA), 577 nm (δ overtone and CIA), 629 nm (γ overtone and CIA), and 688 nm (B band and CIA) were observed in the wavelength range of 307-725 nm. The absorption 322 coefficients of the central wavelengths for the first eight peaks increase non-linearly with O_2 323 concentration while that of the 688 nm peak increases in a more linear manner, indicating that the 324 325 O₂ B band absorption dominates the last absorption peak while the other peaks are mostly 326 associated with CIA of O₂-O₂.

The extinction coefficients obtained by the BBCES correlated well with those measured by the 327 CRD, with slops of 0.990 (R^2 =0.9994) and 0.993 (R^2 = 0.9996) at the wavelengths of 404 nm and 328 329 662 nm, respectively (Figure 3). This excellent agreement between the instruments further substantiates the BBCES measurements and suggests that the accuracy of the BBCES is better 330 than estimated in the error propagation above, where the N_2 refractive index was the largest 331 332 uncertainty. As explained in the data processing section, the measured extinction coefficients were fitted with a 2rd order polynomial (selected wavelengths at the peaks of the CIA absorption bands 333 are shown in Figure 4). At 476.7, 577.2, and 629.2 nm, the absorption is from the CIA of O_2 – O_2 . 334 The fit generates positive values matching the absorption cross-section of O₂–O₂ CIA. At 687.7 335 nm where strong B-band absorption appears, the fit yields a small negative coefficient for O_2-O_2 336 337 CIA.

Figure 5a shows the extinction cross-section measured for 100% O₂. These results agree well with 338 previously reported results by Jordan et al. (2019). For wavelengths where no absorption is 339 detected, the measured extinction cross-sections agree well with *n*-based calculations. Figure 5b-340 341 c shows the determined extinction cross-sections for molecular O_2 and the absorption cross-342 sections of O_2 – O_2 CIA. For wavelength ranges without O_2 bands, our extinction cross-sections agree well with *n*-based values with an average deviation of 2.8%. The absorption cross-sections 343 344 for O₂–O₂ CIA derived in this study mostly agree well with literature data from Thalman and 345 Volkamer (2013). The differences are within 1.1% at 477, 532, 577, and 630 nm but larger deviations were found at 344 (4.2%), 360 (-29%), 380 (-21%), and 446 (4.2%) nm. These 346 absorption bands are the lowest intensity bands and therefore have the largest relative uncertainties 347 in either measurement. 348





- The Rayleigh scattering cross-sections of molecular O_2 derived from the 100% O_2 measurement agree well with *n*-based calculations with an average difference of 1.2%. CIA of O_2 – O_2 calculated from this single measurement matches the results from the fitting method. Due to strong absorption from O_2 B band and γ overtone band, this method cannot derive the cross-sections of CIA of O_2 -
- 353 O₂ at 630 and 688nm.

354 **3.4** The scattering and absorption cross-sections of CH4.

CH₄ has weak absorption in the UV-vis wavelength range, and these bands dominate the 355 356 photographic spectra of planets such as Uranus and Neptune. Figure 6 presents the wavelengthdependent extinction coefficients of CH₄+He mixtures. A total of eleven absorption bands were 357 detected in the wavelength range of 307-725 nm. The extinction coefficients increase as a function 358 of increasing CH₄ concentration. Extinction coefficients obtained by the BBCES correlated well 359 360 with those measured in parallel by the CRDS, with slopes of 1.002 (R^2 =0.9999) and 0.99 (R^2 = 0.999) at the wavelengths of 404 nm and 662 nm (Figure S3). The excellent agreement between 361 these three systems further supports the accuracy of BBCES extinction measurements over a wide 362 working range. The measured extinction coefficients were linearly fit against the CH₄ number 363 concentration. Figure 7 shows the fitted curves at five selected wavelengths. The extinction 364 coefficients have a linear correlation with CH₄ concentration ($R^2 > 0.9988$) without exception. The 365 calculated slopes represent the extinction cross-sections of CH₄ and also indicate a wide dynamic 366 range of our BBCES. 367

The extinction cross-sections for CH₄ retrieved from concentration-dependent measurements are 368 369 plotted in Figure 8a. BBCES results from this study agree well with results from previous studies 370 using BBCES (Jordan et al., 2019; Wilmouth and Sayres, 2019) and CRDS (Sneep and Ubachs, 371 2005). Previous studies using a Nephelometer (Shardanand and Rao, 1977) and interferometer (Cuthbertson and Cuthbertson, 1920; Watson et al., 1936) obtained the scattering cross-sections 372 373 of CH₄. The BBCES measures the extinction cross-section. For wavelengths where extinction is dominated by Rayleigh scattering (< 475 nm), our BBCES results agree well with the results from 374 Nephelometer and interferometer measurements. In this study, the refractive index of CH₄ was 375 calculated using the extinction data in the wavelength range of 307-400 nm. The calculated 376 refractive index was fitted to the general expression: 377

378
$$(n_{CH_4} - 1) \times 10^8 = 5476 + \frac{4.1579 \times 10^{14}}{1.1568 \times 10^{10} - \nu^2}$$
 (7)





379 As shown in Figure 8b, our calculated scattering cross-sections are in good agreement with those 380 derived from the newest refractive index developed by Wilmouth and Sayres (2020) (Table 2), with an average difference of 1.4%. The absorption cross-section, which is the difference between 381 382 the total extinction and the Rayleigh scattering cross-section, is shown in Figure 8c. At most spectral ranges, our results are in good agreement with the results from previous studies (Giver, 383 1978; Smith et al., 1990). For example, the difference as compared to the results from Giver (1978) 384 at 542, 576.4, 598, 619, 665.7, and 703.6 nm is 4.0% on average. At several wavelength regions 385 386 (e.g., 520–536nm, 580–605 nm), the results from Fink et al. (1977) differ from all of the other 387 studies. In the wavelength range of 400-725 nm, absorption contributes up to 99.7% of the CH₄ 388 extinction.

389 3. 5 Dispersion relations for N₂O, SF₆ and CH₄.

390 SF6: Wilmouth and Sayres (2020) found that their measured Rayleigh scattering cross-sections for SF_6 in the ultraviolet range were much lower than those from the *n*-based expression of Sneep and 391 Ubachs (2005). To better constrain the dispersion formula when extrapolated over a broad 392 wavelength range, we employed an alternative fit of the form $A+B/(C-v^2)$ to our data. The 393 Rayleigh scattering derived refractive index in the wavelength range of 264-297 nm and 333-363 394 nm by Wilmouth and Sayres (2020, 2019), and direct refractive index measurement at 632.99 nm 395 (Vukovic et al., 1996) were used (Figure 9a). The resulting dispersion relation for SF_6 in the 396 wavelength range of 264-725 nm is 397

398
$$(n_{SF_6} - 1) \times 10^8 = 22871 + \frac{8.0021 \times 10^{14}}{1.6196 \times 10^{10} - \nu^2}$$
 (8)

N₂O: Sneep and Ubachs (2005) derived the refractive index based on polarizability measurements using interferometer at five single wavelengths (457.9, 488, 514.5, 568.2, 647.1 nm) by Alms et al. (1975). In this study, we calculated the refractive index of N₂O from the Rayleigh scattering cross-sections in the wavelength range of 307–725. Based on this refractive index data set, the dispersion relation (Eq (9)) for N₂O was retrieved for a much broader wavelength range (Figure 9b) compared to that generated by Sneep and Ubachs (2005).

405
$$(n_{N_20} - 1) \times 10^8 = 23154 + \frac{1.534 \times 10^{14}}{6.5069 \times 10^9 - \nu^2}$$
 (9)

406 CH4: The previous study by Wilmouth and Sayres (2019) has shown that their measured Rayleigh
407 scattering cross-sections for CH₄ are in substantial disagreement (22%) with those calculated from





408 the refractive index recommended by Sneep and Ubachs (2005). Sneep and Ubachs (2005) 409 formulated the refractive index of CH₄ based on interferometric measurements at wavelengths of 325, 543.5, 594.1, 612, and 633 nm by Hohm (Hohm, 1993). However, the Rayleigh scattering 410 cross-sections calculated from their refractive index are much higher than all the measured values 411 listed in Figure 9b. Using Rayleigh scattering cross-sections in the wavelength range of 264–297 412 nm, 333-363 nm (Wilmouth and Sayres, 2019, 2020), 307-400 nm from this study, and single 413 wavelength measurements which are not impacted by absorption (Cuthbertson and Cuthbertson, 414 415 1920; Shardanand and Rao, 1977; Watson et al., 1936), we derived the dispersion formula for the refractive index of CH₄ in the combined UV/visible range (Figure 9c) as follows: 416

417
$$(n_{CH_4} - 1) \times 10^8 = 7327.7 + \frac{4.1884 \times 10^{14}}{1.2208 \times 10^{10} - v^2}$$
 (10)

The calculated Rayleigh scattering cross-sections using the dispersion relations derived in this 418 study were compared with those derived from previously recommended formulations listed in 419 Table 1 (Figure 9). The difference increases significantly at the longer wavelength in the region of 420 421 320–725 nm (Figure S4). The average deviations are 0.8%, 0.9%, and 1.6% for SF₆, N₂O, and CH₄, respectively. Notably, the difference for CH_4 is much more significant than for the other two gases. 422 423 This study uses additional measurements and literature data in the wavelength ranges of 307-333nm and 363-400 nm than those used by Wilmouth and Sayres (2020). Therefore this fit 424 captures well our BBCES measurements (Figure 9d), and also the Wilmouth and Sayres (2020, 425 426 2019) data.

427 Conclusions and Implications

Rayleigh scattering cross-sections between 307 and 725 nm were determined for CO₂, N₂O, 428 429 SF₆, O₂, and CH₄ by simultaneous BBCES and CRDS measurements. Extinction coefficients 430 obtained by the BBCES show high consistency with those measured by parallel CRDS at 404 and 431 662 nm (Figure 3 and figure S3), demonstrating that the BBCES measurements provide results with both a wide wavelength range and high accuracy. Comparison of our measurements with n-432 433 based calculations for these gases in the entire wavelength range of this study yields excellent agreement with relative differences of 1.5% and 1.1%, 1.5%, 2.9%, and 1.4% on average, 434 respectively. The O₂-O₂ CIA cross-sections obtained from the BBCES measurements are 435 compared with those published by Thalman and Volkamer (2013). The relative differences are 436 within 1.1% at 477, 532, 577, 630 nm. Larger relative differences occur at the weak bands at 344 437





438 (4.2%), 360 (-29%), 380 (-21%), and 446 (4.2%) nm. The absorption cross-sections of CH₄ in the 439 wavelength range of 400-725 nm agree well with those documented by Giver (1978).

Rayleigh scattering cross-sections of CO₂ determined using BBCES in this study, and in other studies have shown that the refractive index recommended by Sneep and Ubachs (2005) is suitable for use in the wavelength range of 307–725 nm. By incorporating the refractive index data from previous studies, we developed new dispersion relations for the refractive index of N₂O (307-725 nm), SF₆ (264–725 nm), and CH₄ (264–671 nm). The new dispersion relation for CH₄ captures the measurements from BBCES more adequately.

Previous studies measured the Rayleigh scattering and absorption cross-sections of CO₂, N₂O, 446 O₂, SF₆, and CH₄ at narrow spectral ranges or single wavelengths. In this study, we used BBCES 447 that covers the broad wavelength range of 307–725 nm to measure total extinction (the sum of 448 absorption and scattering). The measurements validate that refractive index-based methods for 449 calculating Rayleigh extinction cross-sections are accurate and provide new fits over more 450 451 continuous and extended wavelengths range than available in the literature to constrain such methods. The Rayleigh scattering cross-sections reported here are useful in several applications. 452 These include calibration standards based on extinction for optically-based instruments, such as 453 454 those designed for aerosol optical properties measurements or trace gas concentrations in the field (Jordan et al., 2019; Min et al., 2016; Bluvshtein et al., 2017), especially when high-refractive 455 index gases are used for improved calibration. They will also improve the accuracy of Rayleigh 456 scattering parameterizations for major greenhouse gases in Earth's atmosphere, CO₂, CH₄, and 457 458 N₂O. Accurate quantitative measurements of Rayleigh scattering coefficients and absorption cross-sections of atmospheric gases such as molecular N_2 , O_2 , CO_2 and the CIA of O_2 - O_2 cross-459 sections in the UV-NIR range are of particular importance for the application of Rayleigh LIDAR 460 systems, especially at the Nd:YAG laser harmonics 1064, 532 & 366 nm. These systems analyze 461 462 the molecular backscattering contributions to the LIDAR's attenuated backscatter signals to retrieve the atmospheric profile of aerosols and clouds in the planetary boundary layer (Tomasi et 463 464 al., 2005; Herron, 2007). Recent NASA satellite missions have also aimed to measure global carbon dioxide concentrations with high precision (0.25%) (Drouin et al., 2017). These CO₂ global 465 466 missions use the O_2 - O_2 CIA underneath the structured O_2 A-band (760 nm) to evaluate the solar 467 radiation double pathlength in the Earth atmosphere and to determine the atmospheric pressure. The measurements in this study validate the existing literature on the extinction of O_2 collision 468





- 469 complexes and molecular oxygen bands, and can be used for calibration purposes in both remote
- 470 sensing and *in-situ* spectroscopic applications in the atmosphere. In the future, gas extinction
- 471 measurements at extended wavelengths (near-infrared) and for additional gases (e.g., N₂) will
- 472 improve the spectroscopic applications in atmospheric studies.

473 Data availability.

474 Data are available upon request from the corresponding author (yinon.rudich@weizmann.ac.il).

475 Author contributions.

- 476 Q.H., S.S., and Y.R. designed this study. Q.H., Z.F., and O.S. conducted the experiments. Q.H.
- 477 prepared the draft and all of the co-authors reviewed it and provided comments.

478 Competing interests.

The authors declare that they have no conflict of interest.

480 Acknowledgments

- 481 This research was partially supported by the US-Israel Binational Science Foundation (BSF grant
- 482 #2016093). Dr. Q. H. is supported by the Koshland Foundation and the Center for Planetary
- 483 Sciences, Weizmann Institute of Science. Dr. Z.F. is supported by SAERI initiative of the
- 484 Weizmann Institute.









Figure 1. Rayleigh scattering cross-sections of CO_2 (a), SF_6 (b), and N_2O (c). Panel (d) shows the relative standard deviations as a function of wavelength for each gas. The relative difference in the cross-sections obtained by the BBCES and calculations from the refractive index are displayed (e).







489

490 Figure 2. Wavelength-dependent extinction coefficients of O_2 + He mixtures as a function of O_2

490 Figure 2. Wavelength-dependent extinction coefficients of O_2 + He mixtures as a function of O_2 491 concentration. The colored lines represent the extinction coefficients measured by BBCES, and 492 markers represent results from CRDS.





493



494 Figure 3. Correlations between the extinction coefficients (unit, cm⁻¹) measured by the BBCES
 495 and CRDS.









497 Figure 4. 2^{rd} order polynomial fit of extinction coefficients measured by the BBCES. The O₂ 498 concentration-dependent extinction coefficients are contributed by the extinction coefficients of 499 O₂ (σ_{O_2}), He (σ_{He}), and the O₂-O₂ CIA cross-sections ($\sigma_{O_2-O_2}$).





500



Figure 5. Wavelength-dependent extinction cross-sections of 100% O_2 (a), extinction crosssections of O_2 (b), and O_2 - O_2 CIA cross-section (c).







503

Figure 6. Wavelength-dependent extinction coefficients of CH₄ + He mixtures as a function of CH₄ mixing ratio. The colored lines represent extinction coefficients obtained from BBCES and markers represent results from CRDS. Measurements were performed with CH₄ percentage within 10% and 100% with a 10% step. Moreover, BBCES measurements were also performed for 15%, 25%, 35%, and 45% CH₄. The number concentration of 100% methane was 2.50143×10¹⁹ molecules cm⁻³. Data at selected wavelengths (vertical lines) are shown in Figure 7.





510



Figure 7. The relationship between BBCES measured extinction coefficients of CH₄+He mixtures
and CH₄ concentration. The selected wavelengths were located in Figure 6 by vertical lines.







513



514 Figure 8. Extinction (a), scattering (b), and absorption (c) cross-sections of CH₄.





515



Figure

9. Real refractive index (n) for SF_6 (a), N_2O (b), and CH_4 (c). Comparison of Refractive index 516 from this work with previous studies (Cuthbertson and Cuthbertson, 1920; Naus and Ubachs, 2000; 517 518 Shardanand and Rao, 1977; Sneep and Ubachs, 2005; Vukovic et al., 1996; Watson et al., 1936; Wilmouth and Sayres, 2019, 2020) over the wavelength range of 264-725 nm. The green line 519 represents the dispersion relation given in Table 1. The black line represents the dispersion relation 520 given in Eq. (8-10) derived from a fit to our data and references results. The *n* values for 521 Shardanand and Rao (1977), Sneep and Ubachs (2005), Naus and Ubachs (2000) were calculated 522 523 from their reported Rayleigh scattering cross-sections. Refractive index data from Sneep and 524 Ubachs (2005) are not used in the fitting since these results are away from others. Panel (d) is a close-up view of the panel (c) in the wavelength range of 264–363 nm. 525 526





- Table 1. Refractive index and King correction factors for calculating Rayleigh scattering cross sections and available measurements in the wavelength range of 300–725 nm. Measurements for
- 529 He and N_2 are not summarized in this table.

	Refractiv	Measurements				
Gas	$(n-1) \times 10^8$	$F_k(v)$	ν (cm ⁻¹)	References	λ (nm)	References
He	$2283 \\ + \frac{18102 \times 10^{13}}{1.5342 \times 10^{10} - v^2}$	1.0	14285- 33333	Thalman, 2014; Leonard, 1974; Cuthbertson, 1932		
N ₂	$5677.465 \\ + \frac{318.81874 \times 10^{12}}{1.44 \times 10^{10} - v^2}$	$1.034 + 3.17 \times 10^{-12} \nu^2$	21360- 39370	Sneep, 2005; Naus, 2000		
CO ₂	$\begin{aligned} & \frac{1.1427 \times 10^{11}}{5799.25} \\ & \times (\frac{5799.25}{(128908.9)^2 - v^2} \\ & + \frac{120.05}{(89223.8)^2 - v^2} \\ & + \frac{5.3334}{(75037.5)^2 - v^2} \\ & + \frac{4.3244}{(67837.7)^2 - v^2} \\ & + \frac{1.218145 \times 10^{-5}}{(2418.136)^2 - v^2}) \end{aligned}$	$1.1364 + 2.53 \times 10^{-11} v^2$	39417- 55340	Alms, 1975; Bideau-Mehu, 1973; Sneep, 2005	333-725	Jordan, 2019; Shardanand, 1977; Sneep, 2005; Wilmouth, 2019; He, 2018
CH4	$4869.8 + \frac{4.1023 \times 10^{14}}{1.133 \times 10^{10} - \nu^2}$	1.0	15385- 40000	Sneep, 2005; Wilmouth, 2020	333-363, 434-725	Cuthbertson 1920; Jordan, 2019; Shardanand, 1977; Sneep, 2005; Watson, 1936; Wilmouth, 2019; Smith, 1990; Giver, 1978; Fink, 1977
N ₂ O	$46890 + 4.12 \times 10^{-6} \nu^2$	$\frac{3.3462+70.8\times10^{-12}\ \nu^2}{2.7692-47.2\times10^{-12}\ \nu^2}$	15453- 21838	Alms, 1975; Sneep, 2005	300-320, 532	Johnston, 1975; Sneep, 2005
SF ₆	$18611.4 \\ + \frac{8.9566 \times 10^{14}}{1.680 \times 10^{10} - v^2}$	1.0	15385- 40000	Sneep, 2005; Vukovic, 1996; Wilmouth, 2020	333-363, 532, 633	Sneep, 2005; Vukovic, 1996; Wilmouth, 2019
O2 ^a	$20564.8 + \frac{2.480899 \times 10^{13}}{4.09 \times 10^9 - \nu^2}$	$\begin{array}{c} 1.09 + 1.385 \times 10^{-11} \ \nu^2 \\ + 1.448 \times 10^{-20} \ \nu^4 \end{array}$	18315- 34722	Hohm, 1993; Sneep, 2005	328-667	Thalman, 2013; Jordan, 2019; Hermans, 1999; Greenblatt, 1990; Spiering, 2011

530 Unless noted, the refractive index is scaled to 288.15 K and 1013.25 hPa. N = 2.546899×10^{19} 531 molecules cm⁻³.

532 Due to limited space, only the first name of each reference is shown in the table.

^a The refractive index was obtained at 273.15 K and 1013.25 hPa, N = 2.68678×10^{19} molecules

534 cm^{-3} is used in Eq. (1)





λ(nm)	CO ₂		SFe	SF_6		N ₂ O			O2		 CH ₄	
	n-based	Exp	n-based	Exp		n-based	Exp	_	n-based	Exp	n-based	Exp
330	98.22	96.8	241.5	239.4		137.9	136.7		34.71	35.1	 84.12	85.3
404	41.67	41.6	104.5	105.7		57.71	57.9		14.57	14.8	35.57	35.9
532	13.32	13.3	33.92	34.1		18.19	18.3		4.642	4.55	11.34	11.3
660	5.516	5.52	14.16	14.2		7.483	7.47		1.924	1.95	4.693	4.68
710	4.101	4.08	10.55	10.4		5.551	5.48		1.430	1.41	 3.487	3.47

Table 2. The Rayleigh scattering cross-sections (10^{-27} cm^2) calculated from the refractive index (*n*-based) and obtained from BBCES (Exp) of selected wavelengths.





References

- Alms, G. R., Burnham, A. K., and Flygare, W. H.: Measurement of the dispersion in polarizability anisotropies, J. Chem. Phys., 63, 3321-3326, 10.1063/1.431821, 1975.
- Baidar, S., Oetjen, H., Coburn, S., Dix, B., Ortega, I., Sinreich, R., and Volkamer, R.: The CU Airborne MAX-DOAS instrument: vertical profiling of aerosol extinction and trace gases, Atmos. Meas. Tech., 6, 719-739, 10.5194/amt-6-719-2013, 2013.
- Bates, D. R., and Hays, P. B.: Atmospheric nitrous oxide, Plan. Space Sci., 15, 189-197, 10.1016/0032-0633(67)90074-8, 1967.
- Bideau-Mehu, A., Guern, Y., Abjean, R., and Johannin-Gilles, A.: Interferometric determination of the refractive index of carbon dioxide in the ultraviolet region, Opt. Commun., 9, 432-434, 10.1016/0030-4018(73)90289-7, 1973.
- Bluvshtein, N., Flores, J. M., Segev, L., and Rudich, Y.: A new approach for retrieving the UV– vis optical properties of ambient aerosols, Atmos. Meas. Tech., 9, 3477-3490, 10.5194/amt-9-3477-2016, 2016.
- Bluvshtein, N., Lin, P., Flores, J. M., Segev, L., Mazar, Y., Tas, E., Snider, G., Weagle, C., Brown, S. S., Laskin, A., and Rudich, Y.: Broadband optical properties of biomass-burning aerosol and identification of brown carbon chromophores, J. Geophys. Res. Atmos., 122, 5441-5456, 10.1002/2016JD026230, 2017.
- Cuthbertson, C., and Cuthbertson, M.: On the refraction and dispersion of carbon dioxide, carbon monoxide, and methane, Proc. R. Soc. Lond. A, 97, 152-159, 10.1098/rspa.1920.0020, 1920.
- Cuthbertson, C., and Cuthbertson, M.: The refraction and dispersion of neon and helium, Proc. R. Soc. Lond. A, 135, 40-47, 10.1098/rspa.1932.0019, 1932.
- Drouin, B. J., Benner, D. C., Brown, L. R., Cich, M. J., Crawford, T. J., Devi, V. M., Guillaume, A., Hodges, J. T., Mlawer, E. J., Robichaud, D. J., Oyafuso, F., Payne, V. H., Sung, K., Wishnow, E. H., and Yu, S.: Multispectrum analysis of the oxygen A-band, J. Quant. Spectrosc. Radiat. Transf., 186, 118-138, 10.1016/j.jqsrt.2016.03.037, 2017.
- Fink, U., Benner, D. C., and Dick, K. A.: Band model analysis of laboratory methane absorption spectra from 4500 to 10500 Å, J. Quant. Spectrosc. Radiat. Transf., 18, 447-457, 10.1016/0022-4073(77)90077-2, 1977.
- Fuchs, H., Dube, W. P., Lerner, B. M., Wagner, N. L., Williams, E. J., and Brown, S. S.: A Sensitive and Versatile Detector for Atmospheric NO₂ and NOx Based on Blue Diode Laser Cavity Ring-Down Spectroscopy, Environ. Sci. Technol., 43, 7831-7836, 10.1021/es902067h, 2009.
- Giver, L. P.: Intensity measurements of the CH4 bands in the region 4350 Å to 10,600 Å, J. Quant. Spectrosc. Radiat. Transf., 19, 311-322, 10.1016/0022-4073(78)90064-X, 1978.
- Gordon, I. E., Rothman, L. S., Hill, C., Kochanov, R. V., Tan, Y., Bernath, P. F., Birk, M., Boudon, V., Campargue, A., Chance, K. V., Drouin, B. J., Flaud, J. M., Gamache, R. R., Hodges, J. T., Jacquemart, D., Perevalov, V. I., Perrin, A., Shine, K. P., Smith, M. A. H., Tennyson, J., Toon, G. C., Tran, H., Tyuterev, V. G., Barbe, A., Császár, A. G., Devi, V. M., Furtenbacher, T., Harrison, J. J., Hartmann, J. M., Jolly, A., Johnson, T. J., Karman, T., Kleiner, I., Kyuberis, A. A., Loos, J., Lyulin, O. M., Massie, S. T., Mikhailenko, S. N., Moazzen-Ahmadi, N., Müller, H. S. P., Naumenko, O. V., Nikitin, A. V., Polyansky, O. L., Rey, M., Rotger, M., Sharpe, S. W., Sung, K., Starikova, E., Tashkun, S. A., Auwera, J. V., Wagner, G., Wilzewski, J., Wcisło, P., Yu, S., and Zak, E. J.: The HITRAN2016 molecular spectroscopic database, J. Quant. Spectrosc. Radiat. Transf., 203, 3-69, https://doi.org/10.1016/j.jqsrt.2017.06.038, 2017.





- Greenblatt, G. D., Orlando, J. J., Burkholder, J. B., and Ravishankara, A. R.: Absorption measurements of oxygen between 330 and 1140 nm, J. Geophys. Res. Atmos., 95, 18577-18582, 10.1029/JD095iD11p18577, 1990.
- He, Q., Bluvshtein, N., Segev, L., Meidan, D., Flores, J. M., Brown, S. S., Brune, W., and Rudich, Y.: Evolution of the Complex Refractive Index of Secondary Organic Aerosols during Atmospheric Aging, Environ. Sci. Technol., 52, 3456-3465, 10.1021/acs.est.7b05742, 2018.
- Hermans, C., Vandaele, A. C., Carleer, M., Fally, S., Colin, R., Jenouvrier, A., Coquart, B., and Mérienne, M.-F.: Absorption cross-sections of atmospheric constituents: NO2, O2, and H2O, Environ. Sci. Pollut. Res., 6, 151-158, 10.1007/BF02987620, 1999.
- Herron, J. P.: Rayleigh-Scatter Lidar Observations at USU's Atmospheric Lidar Observatory (Logan,UT) Temperature Climatology, Temperature Comparisons with MSIS, and Noctilucent Clouds, Doctor of Philosophy (PhD), Utah State University, 2007.
- Hohm, U.: Experimental determination of the dispersion in the mean linear dipole polarizability $\alpha(\omega)$ of small hydrocarbons and evaluation of Cauchy moments between 325 nm and 633 nm, Mol. Phys., 78, 929-941, 10.1080/00268979300100621, 1993.
- Ityaksov, D., Linnartz, H., and Ubachs, W.: Deep-UV Rayleigh scattering of N₂, CH₄ and SF₆, Mol. Phys., 106, 2471-2479, 10.1080/00268970802570334, 2008a.
- Ityaksov, D., Linnartz, H., and Ubachs, W.: Deep-UV absorption and Rayleigh scattering of carbon dioxide, Chem. Phys. Lett., 462, 31-34, 10.1016/j.cplett.2008.07.049, 2008b.
- Johnston, H. S., and Selwyn, G. S.: New cross sections for the absorption of near ultraviolet radiation by nitrous oxide (N₂O), Geophys. Res. Lett., 2, 549-551, 10.1029/GL002i012p00549, 1975.
- Jordan, N., Ye, C. Z., Ghosh, S., Washenfelder, R. A., Brown, S. S., and Osthoff, H. D.: A broadband cavity-enhanced spectrometer for atmospheric trace gas measurements and Rayleigh scattering cross sections in the cyan region (470–540 nm), Atmos. Meas. Tech., 12, 1277-1293, 10.5194/amt-12-1277-2019, 2019.
- King, L. V., and Eve, A. S.: On the complex anisotropic molecule in relation to the dispersion and scattering of light, Proc. R. Soc. Lond. A, 104, 333-357, 10.1098/rspa.1923.0113, 1923.
- Leonard, P. J.: Refractive indices, Verdet constants, and Polarizabilities of the inert gases, At. Data Nucl. Data Tables, 14, 21-37, 10.1016/s0092-640x(74)80028-8, 1974.
- Long, C. A., and Ewing, G. E.: Spectroscopic investigation of van der Waals molecules. I. The infrared and visible spectra of (O₂)₂, J. Chem. Phys., 58, 4824-4834, 10.1063/1.1679066, 1973.
- Min, K. E., Washenfelder, R. A., Dube, W. P., Langford, A. O., Edwards, P. M., Zarzana, K. J., Stutz, J., Lu, K., Rohrer, F., Zhang, Y., and Brown, S. S.: A broadband cavity enhanced absorption spectrometer for aircraft measurements of glyoxal, methylglyoxal, nitrous acid, nitrogen dioxide, and water vapor, Atmos. Meas. Tech., 9, 423-440, 10.5194/amt-9-423-2016, 2016.
- Naus, H., and Ubachs, W.: Experimental verification of Rayleigh scattering cross sections, Opt. Lett., 25, 347-349, 10.1364/OL.25.000347, 2000.
- Newnham, D. A., and Ballard, J.: Visible absorption cross sections and integrated absorption intensities of molecular oxygen (O2 and O4), J. Geophys. Res. Atmos., 103, 28801-28815, 10.1029/98JD02799, 1998.
- Platt, U., and Stutz, J.: Differential Optical Absorption Spectroscopy, in: Physics of Earth and Space Environments, Springer Berlin Heidelberg, 2008.
- Shardanand, S., and Rao, A. D. P.: Absolute Rayleigh scattering cross sections of gases and freons of stratospheric interest in the visible and ultraviolet regions, NASA Technical Note, 1977.





- Smith, W. H., Conner, C. P., and Baines, K. H.: Absorption-Coefficients for the 6190-a Ch4-Band between 290-Degrees-K and 100-Degrees-K with Application to Uranus Atmosphere, Icarus, 85, 58-64, 10.1016/0019-1035(90)90103-G, 1990.
- Sneep, M., and Ubachs, W.: Direct measurement of the Rayleigh scattering cross section in various gases, J. Quant. Spectrosc. Ra., 92, 293-310, 10.1016/j.jqsrt.2004.07.025, 2005.
- Spiering, F. R., Kiseleva, M. B., Filippov, N. N., van Kesteren, L., and van der Zande, W. J.: Collision-induced absorption in the O2 B-band region near 670 nm, Phys. Chem. Chem. Phys., 13, 9616-9621, 10.1039/C1CP20403C, 2011.
- Strutt, J. W.: XXXIV. On the transmission of light through an atmosphere containing small particles in suspension, and on the origin of the blue of the sky, London, Edinburgh Dublin Philos. Mag. J. Sci., 47, 375-384, 10.1080/14786449908621276, 1899.
- Strutt, R. J.: The light scattered by gases: its polarisation and intensity, Proc. R. Soc. Lond. A, 95, 155-176, 10.1098/rspa.1918.0057, 1918.
- Strutt, R. J.: A re-examination of the light scattered by gases in respect of polarisation. I.-Experiments on the common gases, Proc. R. Soc. Lond. A, 97, 435-450, 10.1098/rspa.1920.0044, 1920.
- Thalman, R., and Volkamer, R.: Temperature dependent absorption cross-sections of O₂–O₂ collision pairs between 340 and 630 nm and at atmospherically relevant pressure, Phys. Chem. Chem. Phys., 15, 15371-15381, 10.1039/C3CP50968K, 2013.
- Thalman, R., Zarzana, K. J., Tolbert, M. A., and Volkamer, R.: Rayleigh scattering cross-section measurements of nitrogen, argon, oxygen and air, J. Quant. Spectrosc. Ra., 147, 171-177, 10.1016/j.jqsrt.2014.05.030, 2014.
- Tomasi, C., Vitale, V., Petkov, B., Lupi, A., and Cacciari, A.: Improved algorithm for calculations of Rayleigh-scattering optical depth in standard atmospheres, Appl. Opt., 44, 3320-3341, 10.1364/AO.44.003320, 2005.
- Vukovic, D., Woolsey, G. A., and Scelsi, G. B.: Refractivities of and at wavelengths of 632.99 and 1300 nm, Journal of Physics D: Applied Physics, 29, 634-637, 10.1088/0022-3727/29/3/023, 1996.
- Washenfelder, R. A., Langford, A. O., Fuchs, H., and Brown, S. S.: Measurement of glyoxal using an incoherent broadband cavity enhanced absorption spectrometer, Atmos. Chem. Phys., 8, 7779-7793, 10.5194/acp-8-7779-2008, 2008.
- Washenfelder, R. A., Flores, J. M., Brock, C. A., Brown, S. S., and Rudich, Y.: Broadband measurements of aerosol extinction in the ultraviolet spectral region, Atmos. Meas. Tech., 6, 861-877, 10.5194/amt-6-861-2013, 2013.
- Washenfelder, R. A., Attwood, A. R., Flores, J. M., Zarzana, K. J., Rudich, Y., and Brown, S. S.: Broadband cavity-enhanced absorption spectroscopy in the ultraviolet spectral region for measurements of nitrogen dioxide and formaldehyde, Atmos. Meas. Tech., 9, 41-52, 10.5194/amt-9-41-2016, 2016.
- Watson, H. E., Ramaswamy, K. L., and Donnan, F. G.: The refractive index dispersion and polarization of gases, Proceedings of the Royal Society of London. Series A - Mathematical and Physical Sciences, 156, 144-157, 10.1098/rspa.1936.0140, 1936.
- Wilmouth, D. M., and Sayres, D. S.: Rayleigh scattering cross sections of argon, carbon dioxide, sulfur hexafluoride, and methane in the UV-A region using Broadband Cavity Enhanced Spectroscopy, J. Quant. Spectrosc. Radiat. Transf., 234, 32-39, 10.1016/j.jqsrt.2019.05.031, 2019.





Wilmouth, D. M., and Sayres, D. S.: Determination of Rayleigh scattering cross sections and indices of refraction for Ar, CO₂, SF₆, and CH₄ using BBCES in the ultraviolet, J. Quant. Spectrosc. Radiat. Transf., 107224, 10.1016/j.jqsrt.2020.107224, 2020.