

1 **Scattering and Absorption Cross Sections of Atmospheric Gases in the**
2 **Ultraviolet-Visible Wavelength Range (307 - 725 nm)**

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12

13 **Abstract**

14 Accurate Rayleigh scattering and absorption cross sections of atmospheric gases are essential for
15 understanding the propagation of electromagnetic radiation in planetary atmospheres. Accurate
16 extinction cross sections are also essential for calibrating high finesse optical cavities and
17 differential optical absorption spectroscopy and for accurate remote sensing. In this study, we
18 measured the scattering and absorption cross sections of carbon dioxide, nitrous oxide, sulfur
19 hexafluoride, oxygen, and methane in the continuous wavelength range of 307–725 nm using
20 Broadband Cavity Enhanced Spectroscopy (BBCES). The experimentally derived Rayleigh
21 scattering cross sections for CO₂, N₂O, SF₆, O₂, and CH₄ agree with refractive index-based
22 calculations, with a difference of (0.4±1.2)%, (−0.6±1.1)%, (0.9±1.4)%, (2.8±1.2)%, and
23 (0.9±2.2)%, respectively. The O₂-O₂ collision-induced absorption and absorption by methane are
24 obtained with high precision at the 0.8 nm resolution of our BBCES instrument in the 307–725 nm
25 wavelength range. New dispersion relations for N₂O, SF₆, and CH₄ were derived using data in the
26 UV-vis wavelength range. This study provides dispersion relations for refractive indices, *n*-based
27 Rayleigh scattering cross sections, and absorption cross sections based on more continuous and
28 more extended wavelength ranges that available in the current literature.

29 1. Introduction

30 The dominant interactions of gas-phase molecules with light in Earth's atmosphere can be divided
31 into absorption, where the light energy is converted to internal energy and generally (at
32 atmospheric pressures) transferred to the surrounding environment either as heat or as
33 photoemission, and light scattering where the gases redistribute the light energy in the atmosphere.
34 The knowledge of light extinction (scattering + absorption) by gases is essential for predicting the
35 radiative transfer in the atmospheres of the Earth and other planets. In addition, the light extinction
36 by gases is widely used for determining the effective optical pathlength of high-finesse optical
37 cavities that measure trace gases and aerosols (Washenfelder et al., 2013; Washenfelder et al.,
38 2008; Wilmoth and Sayres, 2019; Jordan et al., 2019) and for Differential Optical Absorption
39 Spectroscopy (DOAS) to infer information about the light extinction properties of aerosols and
40 clouds in the open atmosphere (Baidar et al., 2013; Platt and Stutz, 2008).

41 The interaction of light with a wavelength much larger than the size of a molecule/particle gives
42 rise to the scattering of light, which is known as Rayleigh scattering (Strutt, 1899). Rayleigh
43 scattering accounts for scattering, local field effects (Lorentz–Lorenz) (Strutt, 1920) as well as
44 depolarization from the non-sphericity of molecule/particles (King correction factor) (King and
45 Eve, 1923; Strutt, 1918). For a gas with known refractive index (n_ν) and King correction factor
46 ($F_k(\nu)$), the wavelength-dependent Rayleigh scattering cross section (σ_ν , $\text{cm}^2 \text{ molecule}^{-1}$) can be
47 calculated as follows (Sneep and Ubachs, 2005):

$$48 \sigma_\nu = \frac{24\pi^3 \nu^4}{N^2} \left(\frac{n_\nu^2 - 1}{n_\nu^2 + 2} \right)^2 F_k(\nu) \quad (1)$$

49 where N is the number density of the gas (molecules cm^{-3}) and ν is the wavenumber of the light
50 (cm^{-1}). Note that the cross section contains the gas number density but is not in fact dependent on
51 the number density since the refractive index also appears in the expression. This n -based method
52 is an advantageous approach for calculating Rayleigh scattering cross sections, but it is vital to
53 note that the accuracy of the calculated cross sections depends on the experimentally-determined
54 refractive indices and the King correction factors. In particular, cautions should be used when
55 applying a dispersion formula derived from measurements in one wavelength region to calculate
56 Rayleigh scattering cross sections in a different wavelength range.

57 Direct experimental measurement of Rayleigh scattering cross sections is essential given the
58 potential uncertainties in n -based calculations. While measurements of the King correction factors
59 and refractive index for gases are well known from the literature (Cuthbertson and Cuthbertson,
60 1932; Leonard, 1974; Strutt, 1920; Vukovic et al., 1996; Hohm, 1993), there are only a few direct
61 measurements of Rayleigh scattering cross sections (Fuchs et al., 2009; He et al., 2018; Ityaksov
62 et al., 2008a, b; Jordan et al., 2019; Naus and Ubachs, 2000; Sneep and Ubachs, 2005; Thalman
63 and Volkamer, 2013; Thalman et al., 2014; 2017; Wilmouth and Sayres, 2019; 2020), especially
64 measurements with a continuous spectrum from ultraviolet to visible.

65 Rayleigh scattering cross section measurements were previously performed at a single wavelength
66 (e.g., 458 nm, 532 nm, 632.8 nm) using Nephelometry (Shardanand and Rao, 1977) and cavity-
67 ring down spectroscopy (CRDS) (Ityaksov et al., 2008a, b; Naus and Ubachs, 2000; Sneep and
68 Ubachs, 2005; He et al., 2018). More recently, advanced Broadband Cavity Enhanced
69 Spectroscopy (BBCES) was used to determine the Rayleigh scattering cross sections of gases such
70 as Ar, CO₂, O₂, SF₆, and CH₄. The BBCES technique enables the measurement of Rayleigh
71 scattering cross sections over a broad wavelength range. Thalman et al. (2014) performed
72 measurements over selected wavelength regions between 350 and 660 nm using six BBCES
73 cavities for N₂, Ar, and O₂. The BBCES systems were calibrated with He and N₂ using Rayleigh
74 scattering cross sections calculated using their refractive index and from cavity-ring down
75 measurements, respectively. They found a good agreement with n -based values to within $0.2\pm 0.4\%$.
76 Recent studies using BBCES with 30 nm spectral range were also used for Rayleigh scattering
77 cross section measurement in the UV wavelength region and demonstrated excellent agreement
78 with n -based values for Ar and CO₂ (Wilmouth and Sayres, 2020, 2019). Recently, Rayleigh
79 scattering cross sections for CO₂ were measured using BBCES at visible wavelengths between
80 400 and 650 nm, and agreement with n -based values was within 2.4% on average. To the best of
81 our knowledge, there are no direct continuous wavelength measurements of extinction cross
82 sections of gases that covers the ultraviolet across the entire visible range (300–725 nm) as shown
83 in Table 1. Recently, Wilmouth and Sayres (2020) combined refractive index data in the UV region
84 (264-297 nm and 333-363 nm) and at several single wavelengths in the visible, and they derived
85 the dispersion relation of refractive index for SF₆ and CH₄ applicable in the wavelength range of
86 250-650 nm. However, more data in the visible range are needed in order to further validate these
87 dispersion relations.

88 In this study, we used a recently-developed BBCES instrument to measure the extinction cross
89 sections of CO₂, N₂O, SF₆, O₂, and CH₄ continuously across the wavelength region 307–725 nm.
90 All of the measurements were done at a single pressure to eliminate effects due to alignment.
91 This requires the use of two gases with different Rayleigh cross sections for the calibration of the
92 BBCES instrument since the reference state is not vacuum. In this study, He and N₂ were used to
93 calibrate the system. By using the *n*-based calculated Rayleigh scattering cross sections of He and
94 N₂ to calibrate the path length of the optical cavity, the other cross sections can be determined
95 relative to the difference between these two gases. We report high accuracy Rayleigh scattering
96 cross sections for all five gases and compared our results with previous *n*-based values. New
97 dispersion relations for N₂O, SF₆, and CH₄ are derived by incorporating data obtained by this study,
98 extinction cross section data in the deep UV, and previously available scattering cross section data
99 in the visible wavelength range.

100

101 **2. Methods**

102 **2.1 Extinction measurement using BBCES**

103 The BBCES systems used in this study are analogous to our previous studies (He et al., 2018;
104 Washenfelder et al., 2016; Bluvshstein et al., 2016). Briefly, our BBCES instrument consists of two
105 channels, one in the UV (BBCES_{UV}, 307-350 nm) and one in the UV-vis range (BBCES_{vis}, 338-
106 725 nm). The two channels of the BBCES instrument share a laser-driven Xenon arc lamp source
107 (LDLS EQ-99CAL, Energetiq Technology, Inc., MA, USA) coupled with a high transmission
108 UV-Vis optical fiber from which the light is collimated and focused (BBFIBERX-600-1M,
109 Energetiq Technology, Inc., MA, USA). The light source was purged with high purity N₂ and
110 cooled by an aluminum block (with 15°C circulating water inside) to maintain stable optical power
111 output. The UV light from the fiber was reflected by a low-pass dichroic mirror and filtered (Schott
112 Glass WG310 and UG11) into the BBCES_{UV} channel, which has a cavity with two 2.5 cm diameter,
113 1 m radius of curvature mirrors, with manufacturer’s reported reflectivity of 0.9995 (per pass loss
114 = 500 parts per million, ppm) at the nominal center wavelength of 330 nm (Advanced Thin Films,
115 Boulder, USA). The transmitted UV-vis light from the beam splitter was reflected and filtered
116 (Schott Glass WG345 and Edmund Optics 15-261) into the BBCES_{vis} channel consisting of two
117 2.5 cm, 1 m radius of curvature mirrors (FiveNine Optics, USA) with manufacturer’s reported

118 reflectivity above 0.9993 (loss < 700 ppm), see Figure S1. The light emerging through the rear
 119 mirror of the cavity was focused using a 0.1 cm F/2 fiber collimator (74–UV, Ocean Optics,
 120 Dunedin, FL, USA) into a high transmission UV-vis optical fiber which directs the light into a
 121 high-performance spectrometer (QEPro, Ocean Insight, USA). Before gas measurement, the
 122 wavelength of the spectrometer was calibrated using an HG-1 mercury argon calibration light
 123 source (Ocean Insight, USA) within the wavelength range of 302.15–727.29 nm. During these
 124 experiments, a 300 line mm⁻¹ grating and a 200 μm entrance slit width were used. The CCD array
 125 is a back-illuminated detector with 1024×56 pixels (Hamamatsu S7031-1006, Japan) thermo-
 126 electrically cooled to –10 °C to reduce thermal noise. Individual spectra at a wavelength resolution
 127 of 0.8 nm were acquired with 3.0 s integration time, and a total of 150 spectra were recorded during
 128 each measurement.

129 During the extinction measurements, the entire 94.0 ± 0.1 cm pathlength between the mirrors was
 130 filled with He, N₂, CO₂, N₂O, SF₆, or CH₄. The gases were obtained from several vendors (Airgas,
 131 Linde) with the following purities: He, 99.995%; N₂, 99.999%; N₂O, 99.999%, CO₂, 99.999%;
 132 SF₆, 99.999%; CH₄, 99.9995%.

133 The reflectivity of the mirrors ($R(\lambda)$) can be determined as a function of wavelength (λ) by taking
 134 into account the difference in the extinction due to known literary data of Rayleigh scattering
 135 coefficient (α_{Ray}^{gas}) by two different gases such as N₂ ($\alpha_{Ray}^{N_2}(\lambda)$) and He ($\alpha_{Ray}^{He}(\lambda)$) (Washenfelder
 136 et al., 2008).

$$137 \quad R(\lambda) = 1 - d \frac{I_{N_2}(\lambda)(\alpha_{Ray}^{N_2}(\lambda)) - I_{He}(\lambda)(\alpha_{Ray}^{He}(\lambda))}{I_{He}(\lambda) - I_{N_2}(\lambda)} \quad (2)$$

138 where d is the length of the cavity filled by the gas. I_{gas} is the light intensity measured by filling
 139 the cavity with high purity N₂ ($I_{N_2}(\lambda)$) and He ($I_{He}(\lambda)$). Rayleigh scattering (α_{Ray}^{gas}) is the combined
 140 product of Rayleigh scattering cross section (σ) and the gas number density (N) during the
 141 measurements. Rayleigh scattering cross sections of N₂ and He were calculated using the data in
 142 Table 1. Figure S1 shows typical examples of light intensity when the BBCES cavities are filled
 143 with pure N₂. Reflectivity measurements were repeated every three sample measurements to track
 144 the stability of the system.

145 Once the reflectivity is determined, it is possible to calculate the wavelength-dependent extinction
146 cross sections of other gases ($\sigma(\lambda)$) as follows:

$$147 \quad \sigma(\lambda) = \left[\left(\frac{1-R(\lambda)}{d} \right) \left(\frac{I_{He}(\lambda) - I_{gas}(\lambda)}{I_{gas}(\lambda)} \right) + \left(\frac{I_{He}(\lambda)}{I_{gas}(\lambda)} \right) \left(\alpha_{Ray}^{He}(\lambda) \right) \right] / N \quad (3)$$

148 Where N is the number density of the gas during the measurements, and $I_{gas}(\lambda)$ is the light intensity
149 when a target gas fills the cavity. During our experiments, the purge flow of the high reflection
150 mirrors was shut down to ensure that the cavity was filled with target gas completely. To measure
151 the extinction cross sections of CO₂, N₂O, and SF₆, the cavity is filled with pure target gas. Mass
152 flow controller controlled O₂/CH₄ flow was mixed with He in a 2 m Teflon tube ($\Phi = 1/4$ inch) to
153 generate a gas mixture with total flow rate of 500 mL min⁻¹. For O₂ experiments, measurements
154 were performed for O₂ + He mixtures by varying the O₂ percentage between 10% and 100% with
155 a 10% step. The CH₄, measurements were performed for CH₄ + He mixtures with CH₄ percentage
156 ranges between 10% to 100% with a 10% step. Additional measurements were also performed for
157 15%, 25%, 35%, and 45% CH₄.

158 **2.2 Extinction measurements using cavity-ring down spectroscopy (CRDS) at 404 nm and** 159 **662 nm.**

160 To obtain independent measurements for the extinction cross sections and to cross-validate of the
161 BBCES technique, we conducted CRD measurements at two fixed wavelengths of 404 nm and
162 662 nm. CRDS is a highly sensitive technique and uses a different measurement principle than
163 BBCES. The CRDS setup measured the decay rate of light due to extinction rather than an absolute
164 absorbance (as in the BBCES instrument) and thus was immune to shot-to-shot source light
165 fluctuations. A detailed description of the CRD method for light extinction measurement can be
166 found in Bluvstein et al. (2016) and He et al. (2018). Briefly, diode lasers (110 mW 404 nm diode
167 laser, iPulse, Toptica Photonics, Munich, Germany; 120 mW 662 nm diode laser, HL6545MG,
168 Thorlabs Inc., NJ, USA) were used as the light source of these CRDS instrument. The 404 nm and
169 662 nm lasers were modulated at 1383 Hz and 500 Hz with a 50% duty cycle. The diode lasers
170 were optically isolated by quarter waveplates ($1/4 \lambda$) and polarizing beam splitters to prevent
171 damage to the laser head by back reflections from the highly reflective CRDS mirror. The back-
172 reflected light beam was directed into a photodiode, which serves as an external trigger source.
173 Light transmitted through the back mirror of the cavity was collected by an optical fiber and

174 detected by a photomultiplier tube (PMT), which sampled at a rate of 10 to 100 MHz. The time-
175 dependent intensity data was acquired with a 100MHz card (PCI-5122, National Instruments, USA)
176 and processed by data acquisition software in Labview. An exponential curve was fitted to each
177 intensity decay data set (Figure S2). Over 1000 decay time measurements were monitored and
178 averaged on a second basis. The residual of the fit for the averaged intensity decay was obtained
179 and further normalized to the averaged intensity. The derived relative residuals (Figure S2) showed
180 no apparent structure with other time constants, validating the application of CRDS as a good
181 measure of extinction. The resultant 1 Hz decay time was averaged over one measurement duration
182 of five minutes with standard error as the measurement uncertainty.

183 All of the CRDS measurements were performed under room temperature and pressure downstream
184 from the BBCES instrument. The gas temperature (K-type thermocouple) and cavity pressure
185 (Precision Pressure Transducer, Honeywell International Inc., MN, USA) were recorded between
186 the two cavities for gas number density (N) calculation. During the CRDS measurements, the full
187 cavity was filled with the investigated gases (He, CO₂, N₂O, SF₆, O₂, CH₄, or gas mixtures (O₂ +
188 He and CH₄ + He)). The extinction cross section ($\sigma(\lambda)$) of the studied gas was measured relative
189 to that of He and was calculated by equation (4):

$$190 \quad \sigma(\lambda) = \frac{L}{cIN} \left(\frac{1}{\tau_{gas}} - \frac{1}{\tau_{He}} \right) + \sigma_{He} \quad (4)$$

191 Where L is the total length of the cavity (l), c is the speed of light, and τ_{gas} and τ_{He} are the ring-down
192 time of the cavity when it is filled by target gas or by the reference gas, He.

193 **2.3 Data processing**

194 For comparison, the scattering cross sections of the gases investigated in this study were also
195 calculated with Equation (1) based on the refractive index and the King correction factors available
196 in the literature that are listed in Table 1. The King correction factors were taken as unity for mono-
197 atomic molecules and spherical molecules (with regards to the depolarization) but deviates for
198 non-spherical molecules. For the 307–725 nm wavelength range of this study, the n -based
199 calculated Rayleigh scattering cross sections from largest to smallest are SF₆ (Sneep and Ubachs,
200 2005; Wilmouth and Sayres 2020), N₂O (Sneep and Ubachs, 2005), CO₂ (Alms et al. 1975;
201 Bideau-Mehu et al. 1973), CH₄ (Sneep and Ubachs, 2005; Wilmouth and Sayres 2020), N₂ (Bates
202 1984), O₂ (Bates 1984; Sneep and Ubachs, 2005), and He (Abjean et al., 1970; Cuthbertson and

203 Cuthbertson, 1932). Additionally, the refractive indices of SF₆, N₂O, and CH₄ were calculated
204 based on Equation (1) using cross section results from this study and the King correction factors
205 listed in Table 1. Our measurements were performed under ~295K and ~1020 hPa. However, the
206 calculated refractive indices were scaled to 288.15K and 1013.25 hPa as in previous studies (Sneep
207 and Ubachs, 2005; Wilmouth and Sayres, 2020).

208 The extinction of O₂ + He mixtures (α_{O_2+He}) consists of the extinction by O₂ (α_{O_2}) and He (α_{He}),
209 and the O₂-O₂ collision-induced absorption ($\alpha_{O_2-O_2}$). The extinction of O₂ and He is a combined
210 product of extinction cross section (σ_{gas}) and gas number density (N_{gas}). Thus α_{O_2+He} can be
211 described with the following equation:

$$212 \alpha_{O_2+He} = \sigma_{O_2-O_2} \times N_{O_2}^2 + \sigma_{O_2} \times N_{O_2} + \sigma_{He} \times N_{He} \quad (5)$$

213 Where N_{O_2} and N_{He} are the number density of the O₂ and He in the cavities. Performing a 2nd order
214 polynomial fit to the extinction obtained by the BBCES system with respect to the gas number
215 density thus yields the extinction cross section of O₂ and the O₂-O₂ collision-induced absorption
216 (CIA) cross section.

217 In addition to the results from 2nd order polynomial fitting, we also used data from pure O₂
218 measurement to calculate the extinction by O₂ and by CIA of O₂-O₂. The real refractive index of
219 O₂ (n_{O_2}) derived from extinction data measured in the wavelength regions where there is no
220 absorption was fitted using the generalized expression of $(n_{O_2} - 1) \times 10^8 = A + \frac{B}{C-v^2}$. Based
221 on the refractive index, the scattering cross sections of O₂ in the wavelength range of 307-725 nm
222 were further calculated. By subtracting the scattering cross section of O₂ from the measured total
223 extinction, we derived the CIA of O₂-O₂. However, the O₂ absorption bands at 580, 630, and 690
224 nm overlap with those of O₂-O₂ collisions. Additional corrections are thus needed to split the
225 absorption by O₂ and O₂-O₂ collision, which is out of the scope of this study.

226 Methane has weak vibrational overtone absorption in the UV-vis wavelength range that is
227 comparable to or greater than its Rayleigh scattering. Previous high-resolution spectroscopy
228 studies have identified smooth and unstructured absorption bands across the UV-visible range
229 (Giver, 1978; Smith et al., 1990). The spectral features are substantially broader than 0.8 nm, thus
230 the absorption by CH₄ can be measured by our BBCES system. The measured extinction

231 coefficients of CH₄+He mixtures (α_{CH_4+He}) are linearly correlated with the number concentration
232 of CH₄ (N_{CH_4}) as described by the following equation:

$$233 \quad \alpha_{CH_4+He} = \sigma_{CH_4} \times N_{CH_4} + \sigma_{He} \times N_{He} \quad (6)$$

234 A linear fit was used for deriving the extinction cross section of CH₄. The absorption between 300
235 and 400 nm is negligible as compared to the Rayleigh scattering. Thus extinction data in this UV
236 wavelength range were used to calculate the real part of the refractive index of CH₄ which was
237 further fitted utilizing the expression of $(n_{CH_4} - 1) \times 10^8 = A + \frac{B}{C - \nu^2}$. By applying this
238 dispersion relation, the Rayleigh scattering cross sections in the entire wavelength range of 307–
239 725 nm were derived. Finally, the CH₄ absorption cross sections were calculated by subtraction of
240 the scattering cross section from the extinction cross section.

241 **2.4 Error Propagation for Extinction Measurements**

242 The uncertainty for BBCES measurements can be assessed by the propagation of the errors
243 associated with the measurements. Rayleigh scattering cross section of N₂ is validated up to 468
244 nm with an uncertainty of 1%. Rayleigh scattering cross sections measurement by CRDs agree
245 well with those calculated from the refractive index with relative difference within 1% in the
246 wavelength range of 468–650 nm (Naus and Ubachs, 2000; Thalman et al., 2014). Thus, an uncertainty
247 of 1% was assigned for N₂ Rayleigh scattering cross section in the wavelength range of 307–725
248 nm. Each parameter (temperature, pressure, light intensity) was measured 150 times for each gas.
249 The standard deviation of the measurements (<0.3%) is combined with the uncertainties in the
250 pressure ($\pm 0.3\%$), temperature ($\pm 0.1\%$), the Rayleigh cross section uncertainties for N₂ ($\pm 1\%$) as
251 well as uncertainty in the measurements of the spectral signal by the spectrometer ($\ll 0.2\%$) to get
252 an overall relative uncertainty for the effective pathlength curve of $\pm 1.1\%$. This uncertainty is
253 further propagated to the target gas by consideration of the uncertainties of pressure ($\pm 0.3\%$),
254 temperature ($\pm 0.1\%$), and spectral intensity ($\ll 0.2\%$) of the target gas measurements, and the
255 standard deviation of the measurements (<1.2%). The overall 1- σ uncertainty of the gas extinction
256 cross section is within 1.7%. The precision of the mass flow controllers is 0.5 mL min⁻¹. When the
257 total flow rate is 500 mL min⁻¹, the resulting uncertainty in the gas concentration (10-100%) varies
258 from 0% to 1.0%. Thus, the overall 1- σ uncertainty of extinction coefficients measured for
259 CH₄+He and O₂+He varies from 1.1% to 1.5%. The detailed wavelength-dependent uncertainties

260 were calculated due to the wavelength-dependence of the spectral intensity. Moreover, due to the
261 highly-structured reflectivity curve of the high-reflection mirrors, additional uncertainty could be
262 introduced and this uncertainty can not be quantified in this study.

263

264 **3 Results and Discussion**

265 **3.1 Performance of the optical system**

266 The reflectivity of the cavity mirrors, measured across the entire range using the difference in
267 Rayleigh scattering of N₂ and He, was very stable throughout the experiments. The measured
268 mirrors reflectivity curves are shown in Figure S1. The mean peak reflectivity of the BBCES_{UV}
269 mirrors was 0.999328 ± 0.000006 (672 ± 6 ppm) at 330 nm, with a corresponding effective optical
270 pathlength of 1.40 ± 0.01 km. The reflectivity curve of the BBCES_{vis} is much more structured, with
271 reflectivity ranging between 0.999224 ± 0.000010 and 0.9999550 ± 0.0000006 (776 ± 10 ppm > loss >
272 45 ± 0.6 ppm) over a wide wavelength range of 338–725 nm. The reflectivity of the BBCES_{vis} is
273 much higher than that of our previous system (He et al., 2018) and also covers a much broader
274 wavelength range. Thus the effective pathlength of the BBCES_{vis} varies between 1.3 and 20.4
275 km, guaranteeing a high sensitivity of the extinction measurement.

276 **3.2 Rayleigh scattering cross sections of CO₂, N₂O, SF₆.**

277 Figure 1 shows the extinction cross sections of CO₂, N₂O, and SF₆ measured by the BBCES system.
278 The extinction cross sections of these gases monotonically decrease with increasing wavelength,
279 and no absorption (i.e., no structured extinction larger than the smoothly varying Rayleigh curve)
280 is observed in the wavelength range of 307–725 nm, indicating that the measured extinction is due
281 solely to the Rayleigh scattering of these gases. The wavelength-dependent relative standard
282 deviations of the measurements for each gas are shown in Figure 1d. The mean 1- σ uncertainty of
283 the reported cross sections for all three gases across the 307–725 nm wavelength range is 1.2–1.3%
284 for CO₂, 1.2–1.7% for N₂O, and 1.2–1.4% for SF₆. As mentioned above, the derived uncertainty
285 originates predominantly from the uncertainty in the N₂ Rayleigh scattering cross section.
286 Uncertainty in the Rayleigh cross sections of each gas varies with wavelength and generally tracks
287 the light intensity spectra, which is a combined product of light source spectrum and the mirror

288 reflectivity profile. The uncertainty is much higher when the transmitted light intensity is low
289 (Figure S1).

290 The BBCES system measured Rayleigh scattering cross sections for these three gases agree well
291 with those obtained by our CRDS system operating at 404 nm and 662 nm, with deviations smaller
292 than 0.9%. Table 2 lists the Rayleigh scattering cross sections at several wavelengths obtained by
293 the BBCES measurements (Exp) and by the calculations using the refractive index and $F_k(\nu)$ values
294 from Table 1 (n -based). The relative differences between these two sets of results are within 1.4%.

295 Figure 1a–c shows a comparison of the measured Rayleigh scattering cross sections for CO₂, N₂O,
296 and SF₆ with n -based calculations and with previous experimental results from the literature. There
297 are a few measurements for the Rayleigh scattering cross sections for CO₂ which cover a wide
298 spectral range (Jordan et al., 2019; Shardanand and Rao, 1977; Sneep and Ubachs, 2005; Wilmouth
299 and Sayres, 2019; 2020; He et al., 2018). There are fewer Rayleigh scattering measurements for
300 N₂O and SF₆ in the studied wavelength range. The measured Rayleigh scattering cross sections for
301 CO₂, N₂O, and SF₆ are in excellent agreement with n -based calculations. The wavelength-
302 dependent difference of our experimentally derived Rayleigh scattering cross sections with n -
303 based calculations are shown in Figure 1e. The mean ratios of our measurements to the n -based
304 values for the entire wavelength range of 307–725 nm are 1.00 ± 0.01 , 0.99 ± 0.01 , and 1.01 ± 0.01
305 for CO₂, N₂O, and SF₆, respectively. The relative difference between our measurements and the n -
306 based values are $(0.4 \pm 1.2)\%$, $(-0.6 \pm 1.1)\%$, $(0.9 \pm 1.4)\%$ (Mean \pm SD) for CO₂, N₂O, and SF₆,
307 respectively. Variability of the relative difference is due to structure in the mirror reflectivity that
308 does not fully cancel. The wavelength-dependent Rayleigh scattering cross section is generally
309 described in the form of $\sigma = A \times \lambda^B$, where σ and λ are the cross section and the wavelength. In
310 this study, the measured values and the n -based data were both fit to this function. The relative
311 difference between these two fitted functions is shown in Figure 1(f). That would be a measure of
312 the uncertainty comparing smooth functions to smooth functions. The relative differences were
313 $(0.5 \pm 0.5)\%$, $(-0.4 \pm 0.3)\%$, and $(0.9 \pm 0.2)\%$ (Mean \pm stdev), for CO₂, N₂O, and SF₆, respectively. The
314 mean values of the relative difference obtained from the fitting function are close to that obtained
315 from the measurements. However, the variabilities are much smaller, which may be related to the
316 cancellation of the influence by the structured mirror reflectivity. Notably, while our results for
317 N₂O agree well with the n -based calculations, previous results obtained by CRDS at 532 nm

318 (Sneep and Ubachs, 2005) and by absorption spectroscopy in the wavelength of 300–315 nm
319 (Bates and Hays, 1967) do not agree well with the n -based calculations. The measurements
320 between 300 and 315 nm were first published by Bates and Hays (1967), who obtained the results
321 from a doctoral thesis. However, the results from our BBCES system are in good agreement with
322 the n -based calculations and with experimental results from independent CRDS measurements,
323 thus increasing the confidence in our measured values.

324 **3. 3 Scattering and absorption cross sections of O₂.**

325 The UV-vis spectra of gas-phase molecular oxygen are characterized by discrete structured
326 absorption bands due to the electronic transition ($b^1 \Sigma_g^+(v' = 1/2/3) \leftarrow \Sigma_g^-(v'' = 0)$) of O₂
327 monomer, broader unstructured CIA of O₂–O₂, and structured dimer bands from the bound van
328 der Waals O₂ dimer (Newnham and Ballard, 1998). Under atmospheric conditions, the O₂–O₂ CIA
329 bands are frequently described as "O₄" bands, although absorption by O₂ dimer is thought to be
330 significant only under very low-temperature conditions (Thalman and Volkamer, 2013; Long and
331 Ewing, 1973). Within the wavelength range investigated in this work, the molecular oxygen B
332 band at 688 nm ($b^1 \Sigma_g^+(v' = 1) \leftarrow X^3 \Sigma_g^-(v'' = 0)$), γ overtone band at 629 nm ($b^1 \Sigma_g^+(v' = 2) \leftarrow$
333 $X^3 \Sigma_g^-(v'' = 0)$), and δ overtone band at 580 nm ($b^1 \Sigma_g^+(v' = 3) \leftarrow X^3 \Sigma_g^-(v'' = 0)$) overlap with
334 O₂–O₂ CIA bands of $^1 \Sigma_g^+(v = 1)$, $^1 \Delta_g + ^1 \Delta_g (v = 0)$, and $^1 \Delta_g + ^1 \Delta_g (v = 1)$, respectively.
335 These absorption bands can only be resolved by a high-resolution spectroscopic technique.
336 Absorption cross sections of the B, γ , and δ bands were convoluted from the HITRAN database
337 (Gordon et al., 2017) by considering the temperature, pressure, and wavelength resolution of the
338 instrument. The wings of the oxygen lines also show a quadratic dependence on the pressure due
339 to pressure broadening. However, due to the minimal O₂ absorption contribution below 680 nm
340 and the low instrument wavelength resolution, the extinction cross section of the O₂ monomer can
341 be treated as linearly correlated with the O₂ concentration. Moreover, the O₂–O₂ CIA cross section
342 is correlated with the square of the O₂ concentration. Therefore, these cross sections can be
343 retrieved from measurements at different O₂ concentrations. Due to the discrete structured
344 absorption bands and the wavelength resolution of the instrument, the range of absorption cross
345 sections spans several orders of magnitude within the spectral response of the instrument, limiting
346 the relevance of the absorption cross sections for other researchers. These results are not further

347 discussed here. However, the data for broader unstructured CIA of O₂-O₂ are still useful for
348 various applications.

349 Figure 2 shows the wavelength-dependent extinction coefficients of O₂+He mixtures. He was used
350 in these experiments to minimize extinction contributions from Rayleigh scattering. Nine
351 absorption peaks centered at 344 nm (CIA), 360 nm (CIA), 380 nm (CIA), 446 nm (CIA), 477 nm
352 (CIA), 532 nm (CIA), 577 nm (δ overtone and CIA), 629 nm (γ overtone and CIA), and 688 nm
353 (B band and CIA) were observed in the wavelength range of 307–725 nm. The absorption
354 coefficients of the central wavelengths for the first eight peaks increase non-linearly with O₂
355 concentration while that of the 688 nm peak increases in a more linear manner, indicating that the
356 O₂ B band absorption dominates the last absorption peak while the other peaks are mostly
357 associated with CIA of O₂-O₂.

358 The extinction coefficients obtained by the BBCES system correlated well with those measured
359 by the CRD setup, with slopes of 0.990 ($R^2=0.9994$) and 0.993 ($R^2 = 0.9996$) at the wavelengths
360 of 404 nm and 662 nm, respectively (Figure 3). This excellent agreement between the instruments
361 further substantiates the BBCES measurements and suggests that the accuracy of the BBCES
362 instrument at these two wavelengths is better than estimated in the error propagation above, where
363 the N₂ refractive index was the largest uncertainty. As explained in the data processing section,
364 the measured extinction coefficients were fitted with a 2nd order polynomial (selected wavelengths
365 at the peaks of the CIA absorption bands are shown in Figure 4). At 476.7, 577.2, and 629.2 nm,
366 the absorption is from the CIA of O₂-O₂. The fit generates positive values matching the absorption
367 cross section of O₂-O₂ CIA. At 687.7 nm where strong B-band absorption appears, the fit yields a
368 small negative coefficient for O₂-O₂ CIA.

369 Figure 5a shows the extinction cross section measured for 100% O₂. These results agree well with
370 previously reported results by Jordan et al. (2019). For wavelengths where no absorption is
371 detected, the measured extinction cross sections agree well with n -based calculations. Figure 5b-c
372 shows the determined extinction cross sections for molecular O₂ and the absorption cross sections
373 of O₂-O₂ CIA. For wavelength ranges without O₂ bands, our extinction cross sections agree well
374 with the n -based values with an average deviation of (2.8±1.2)%. The absorption cross sections
375 for O₂-O₂ CIA derived in this study mostly agree well with literature data from Thalman and
376 Volkamer (2013). The differences are within 1.1% at 477, 532, 577, and 629 nm but larger

377 deviations were found at 344 (4.2%), 360 (-29%), 380 (-21%), and 446 (4.2%) nm. These
378 absorption bands are the lowest intensity bands and therefore have the largest relative uncertainties
379 in either measurement. Moreover, the absorptions at 344, 360, 380, and 446 nm contribute a much
380 smaller fraction of the extinction as compared to that of 477, 532, 577, and 630 nm. Thus larger
381 discrepancies were observed during the apportionment of absorption from extinction.

382 The Rayleigh scattering cross sections of molecular O₂ derived from the 100% O₂ measurement
383 agree well with *n*-based calculations with an average difference of 1.2%. CIA of O₂-O₂ calculated
384 from this single measurement matches the results from the fitting method. Due to strong absorption
385 from O₂ B band and γ overtone band, this method cannot derive the cross sections of CIA of O₂-
386 O₂ at 630 and 688nm.

387 **3. 4 The scattering and absorption cross sections of CH₄.**

388 CH₄ has weak absorption in the UV-vis wavelength range, and these bands dominate the
389 photographic spectra of planets such as Uranus and Neptune (Adel and Slipher, 1934). Figure 6
390 presents the wavelength-dependent extinction coefficients of CH₄+He mixtures. A total of eleven
391 absorption bands were detected in the wavelength range of 307–725 nm. The extinction
392 coefficients increase as a function of increasing CH₄ concentration. Extinction coefficients
393 obtained by the BBCES system correlated well with those measured in parallel by the CRDS
394 instrument, with slopes of 1.002 ($R^2=0.9999$) and 0.99 ($R^2 = 0.999$) at the wavelengths of 404 nm
395 and 662 nm (Figure S3). The excellent agreement between these three systems further supports
396 the accuracy of BBCES extinction measurements over a wide working range. The measured
397 extinction coefficients were linearly fit against the CH₄ number concentration. Figure 7 shows the
398 fitted curves at five selected wavelengths. The extinction coefficients have a linear correlation with
399 CH₄ concentration ($R^2 > 0.9988$) without exception. The calculated slopes represent the extinction
400 cross sections of CH₄ and also indicate a wide dynamic range of our BBCES system.

401 The extinction cross sections for CH₄ retrieved from concentration-dependent measurements are
402 plotted in Figure 8a. BBCES results from this study agree well with results from previous studies
403 using BBCES (Jordan et al., 2019; Wilmouth and Sayres, 2019) and CRDS (Sneep and Ubachs,
404 2005). Previous studies using a Nephelometer (Shardanand and Rao, 1977) and interferometer
405 (Cuthbertson and Cuthbertson, 1920; Watson et al., 1936) obtained the scattering cross sections
406 and the refractive index of CH₄. The BBCES instrument measures the extinction cross section. For

407 wavelengths where extinction is dominated by Rayleigh scattering (< 475 nm), our BBCES results
 408 agree well with the results from Nephelometer and interferometer measurements. In this study, the
 409 refractive index of CH_4 was calculated using the extinction data in the wavelength range of 307-
 410 400 nm. The calculated refractive index was fitted to the general expression:

$$411 \quad (n_{\text{CH}_4} - 1) \times 10^8 = 5476 + \frac{4.1579 \times 10^{14}}{1.1568 \times 10^{10} - \nu^2} \quad (7)$$

412 As shown in Figure 8b, our calculated scattering cross sections are in good agreement with those
 413 derived from the newest refractive index developed by Wilmouth and Sayres (2020) (Table 2),
 414 with an average difference of $(0.9 \pm 2.2)\%$. The absorption cross section, which is the difference
 415 between the total extinction and the Rayleigh scattering cross section, is shown in Figure 8c. At
 416 most spectral ranges, our results are in better agreement with the results from previous studies by
 417 Giver, (1978) and Smith et al.(1990). For example, the difference as compared to the results from
 418 Giver (1978) at 542, 576.4, 598, 619, 665.7, and 703.6 nm is 4.0% on average. At several
 419 wavelength regions (e.g., 520–536nm, 580–605 nm), the results from Fink et al. (1977) differ from
 420 all of the other studies. In the wavelength range of 400–725 nm, absorption contributes up to 99.7%
 421 of the CH_4 extinction.

422 **3. 5 Dispersion relations for N_2O , SF_6 and CH_4 .**

423 **SF_6 :** Wilmouth and Sayres (2020) found that their measured Rayleigh scattering cross sections for
 424 SF_6 in the ultraviolet range were lower than those from the n -based expression of Sneep and
 425 Ubachs (2005). They generated the dispersion formula for SF_6 from the combined fit using
 426 refractive index data in the wavelength range of 264-297 nm and 333-363 nm by Wilmouth and
 427 Sayres (2020, 2019), and direct refractive index measurement at 632.99 nm (Vukovic et al., 1996)
 428 (Figure 9a). In our study, the refractive index of SF_6 in the wavelength range of 307-725 nm was
 429 calculated from the measured Rayleigh scattering cross section. The calculated refractive index
 430 was scaled to 288.15 K and 1013.25 hPa to be consistent with past studies. To constrain the
 431 dispersion formula when extrapolated over a broad wavelength range, we employed an fit of the
 432 form $A+B/(C-\nu^2)$ to our data and the data used by Wilmouth and Sayres (2020) for fitting. All data
 433 points were weighted equally. The resulting dispersion relation for SF_6 in the wavelength range of
 434 264–725 nm is

$$435 \quad (n_{\text{SF}_6} - 1) \times 10^8 = 18997.7 + \frac{8.27663 \times 10^{14}}{1.56833 \times 10^{10} - \nu^2} \quad (8)$$

436 **N₂O**: Sneep and Ubachs (2005) derived the refractive index based on polarizability measurements
 437 using interferometer at five single wavelengths (457.9, 488, 514.5, 568.2, 647.1 nm) by Alms et
 438 al. (1975). In this study, we calculated the refractive index of N₂O from the Rayleigh scattering
 439 cross sections in the wavelength range of 307–725 for 288.15K and 1013.25 hPa. Based on this
 440 refractive index data set, the dispersion relation (Eq (9)) for N₂O was retrieved for a much broader
 441 wavelength range (Figure 9b) compared to that generated by Sneep and Ubachs (2005).

$$442 \quad (n_{N_2O} - 1) \times 10^8 = 22095 + \frac{1.66291 \times 10^{14}}{6.75226 \times 10^9 - \nu^2} \quad (9)$$

443 **CH₄**: The previous study by Wilmouth and Sayres (2019) has shown that their measured Rayleigh
 444 scattering cross sections for CH₄ are in substantial disagreement (22%) with those calculated from
 445 the refractive index recommended by Sneep and Ubachs (2005). Sneep and Ubachs (2005)
 446 formulated the refractive index of CH₄ based on interferometric measurements at wavelengths of
 447 325, 543.5, 594.1, 612, and 633 nm by Hohm (Hohm, 1993). However, the Rayleigh scattering
 448 cross sections calculated from Sneep and Ubachs (2005) are much higher than all the measured
 449 values listed in Figure 9c. Using equally weighted Rayleigh scattering cross sections data sets in
 450 the wavelength range of 264–297 nm, 333–363 nm (Wilmouth and Sayres, 2019, 2020), 307–400
 451 nm from this study, and single wavelength measurements that are not impacted by absorption
 452 (Cuthbertson and Cuthbertson, 1920; Watson et al., 1936), we derived the dispersion formula for
 453 the refractive index of CH₄ in the combined UV/visible range (Figure 9c) as follows:

$$454 \quad (n_{CH_4} - 1) \times 10^8 = 3603.09 + \frac{4.40362 \times 10^{14}}{1.1741 \times 10^{10} - \nu^2} \quad (10)$$

455 The calculated Rayleigh scattering cross sections using the dispersion relations derived in this
 456 study were compared with those derived from previously recommended formulations listed in
 457 Table 1 (Figure 9). The difference increases significantly towards the longer wavelength in the
 458 region of 320–725 nm (Figure S4). The average deviations are 0.1%, 0.9%, and 0.1% for SF₆, N₂O,
 459 and CH₄, respectively. The fit performed by this study combines the data set used by Wilmouth
 460 and Sayres and the data acquired by this study. These small deviations support that our data
 461 acquired by the BBCES at a wide range agrees well with literature data. Notably, the difference
 462 for N₂O is more significant than for the other two gases. This study uses refractive index data in
 463 the continuous wavelength ranges of 307–725 nm to derive the dispersion relation, while the
 464 formulation for N₂O in Table 1 is derived by Sneep and Ubachs (2005) based on polarizability

465 measurements at five single wavelengths. For the formulation of the refractive index of CH₄,
466 Wilmouth and Sayres (2020) weighted the data sets from Watson and Ramaswamy (1936) and
467 Cuthbertson and Cuthbertson (1920) equally but gave more weight to their UV measurements
468 when deriving the formulation of the refractive index. In this study, all the CH₄ data set were
469 weighted equally. The derived dispersion relation agrees very well with that from Wilmouth and
470 Sayres (2020), as shown in Figure 9 (c-d).

471 The structure in derived refractive indices that arises from the residual structure in the reflectivity
472 curves is a limitation. However, the agreement between the CRDS data and the BBCES data at
473 405 nm and 662 nm is 0.24% and 0.8%, both values well within the stated uncertainty of the
474 BBCES data. CRDS is an absolute method that does not suffer from the apparent artifact due to
475 the structured mirror reflectivity. Moreover, the large number of data points in the BBCES data
476 easily compensates for the uncertainty in the structured mirror reflectivity when fitting a smooth
477 function to the data. Figure S5–7 show the refractive index fit using our raw BBCES data and a
478 series of fits to more highly averaged data. As the averaging increases, the apparent structure in
479 the data decreases, but the fit does not change.

480 **Conclusions and Implications**

481 Rayleigh scattering cross sections between 307 and 725 nm were determined for CO₂, N₂O,
482 SF₆, O₂, and CH₄ by simultaneous BBCES and CRDS measurements. Extinction coefficients
483 obtained by the BBCES system show high consistency with those measured by parallel CRDS
484 instruments at 404 and 662 nm (Figure 3 and figure S3), demonstrating that the BBCES
485 measurements provide results with both a wide wavelength range and high accuracy. Comparison
486 of our measurements with the *n*-based calculations for these gases in the entire wavelength range
487 of this study yields excellent agreement with relative differences of (0.4±1.2)%, (−0.6±1.1)%,
488 (0.9±1.4)%, (2.8±1.2)%, and (0.9±2.2)% for CO₂, N₂O, SF₆, O₂, and CH₄, respectively. The O₂-
489 O₂ CIA cross sections obtained from the BBCES measurements are compared with those published
490 by Thalman and Volkamer (2013). The relative differences are within 1.1% at 477, 532, 577, 630
491 nm. Larger relative differences occur at the weak bands at 344 (4.2%), 360 (−29%), 380 (−21%),
492 and 446 (4.2%) nm. The absorption cross sections of CH₄ in the wavelength range of 400-725 nm
493 agree well with those documented by Giver (1978).

494 Rayleigh scattering cross sections of CO₂ determined using BBCES system and CRDS
495 instruments in this study, and in other studies have shown that the refractive index recommended
496 by Snee and Ubachs (2005) is suitable for use in the wavelength range of 307–725 nm. By
497 incorporating the refractive index data from previous studies, we developed new dispersion
498 relations for the refractive index of N₂O (307–725 nm), SF₆ (264–725 nm), and CH₄ (264–671 nm).
499 The derived dispersion relations for SF₆ and CH₄ agree well with those provided by Wilmouth and
500 Sayres (2020).

501 Previous studies measured the Rayleigh scattering and absorption cross sections of CO₂, N₂O,
502 O₂, SF₆, and CH₄ at narrow spectral ranges or single wavelengths. In this study, we used BBCES
503 system that covers the broad wavelength range of 307–725 nm to measure total extinction (the
504 sum of absorption and scattering). The measurements validate that refractive index-based methods
505 for calculating Rayleigh extinction cross sections are accurate and provide new fits over more
506 continuous and extended wavelengths range than available in the literature to constrain such
507 methods. The Rayleigh scattering cross sections reported here are useful in several applications.
508 These include calibration standards based on extinction for optically-based instruments, such as
509 those designed for aerosol optical properties measurements or trace gas concentrations in the field
510 (Jordan et al., 2019; Min et al., 2016; Bluvshtein et al., 2017), especially when high-refractive
511 index gases are used for improved calibration. They will also improve the accuracy of Rayleigh
512 scattering parameterizations for major greenhouse gases in Earth’s atmosphere, CO₂, CH₄, and
513 N₂O. Accurate quantitative measurements of Rayleigh scattering coefficients and absorption cross
514 sections of atmospheric gases such as molecular N₂, O₂, CO₂ and the CIA of O₂–O₂ cross sections
515 in the UV-NIR range are of particular importance for the application of Rayleigh LIDAR systems,
516 especially at the Nd:YAG laser harmonics 1064, 532 & 366 nm. These systems analyze the
517 molecular backscattering contributions to the LIDAR’s attenuated backscatter signals to retrieve
518 the atmospheric profile of aerosols and clouds in the planetary boundary layer (Tomasi et al., 2005;
519 Herron, 2007). Recent NASA satellite missions have also aimed to measure global carbon dioxide
520 concentrations with high precision (0.25%) (Drouin et al., 2017). These CO₂ global missions use
521 the O₂–O₂ CIA underneath the structured O₂ A-band (760 nm) to evaluate the solar radiation
522 double pathlength in the Earth atmosphere and to determine the atmospheric pressure. The
523 measurements in this study validate the existing literature on the extinction of O₂ collision
524 complexes and molecular oxygen bands, and can be used for calibration purposes in both remote

525 sensing and *in-situ* spectroscopic applications in the atmosphere. In the future, gas extinction
526 measurements at extended wavelengths (near-infrared) and for additional gases (e.g., N₂) will
527 improve the spectroscopic applications in atmospheric studies. We also note that while acquiring
528 data at a broad wavelength range quickly using a single BBCES instrument, uncertainties were
529 also observed in the wavelength ranges where the mirror reflectivity changed significantly.
530 However, with appropriate averaging, this can be minimized without compromising the accuracy.
531 There is a tradeoff between obtaining data at a wide wavelength range and ensuring high accuracy
532 data. New mirrors with a smoother reflectivity curve will improve the performance of BBCES
533 instrument.

534 **Data availability.**

535 Data are available upon request from the corresponding author (yionon.rudich@weizmann.ac.il).

536 **Author contributions.**

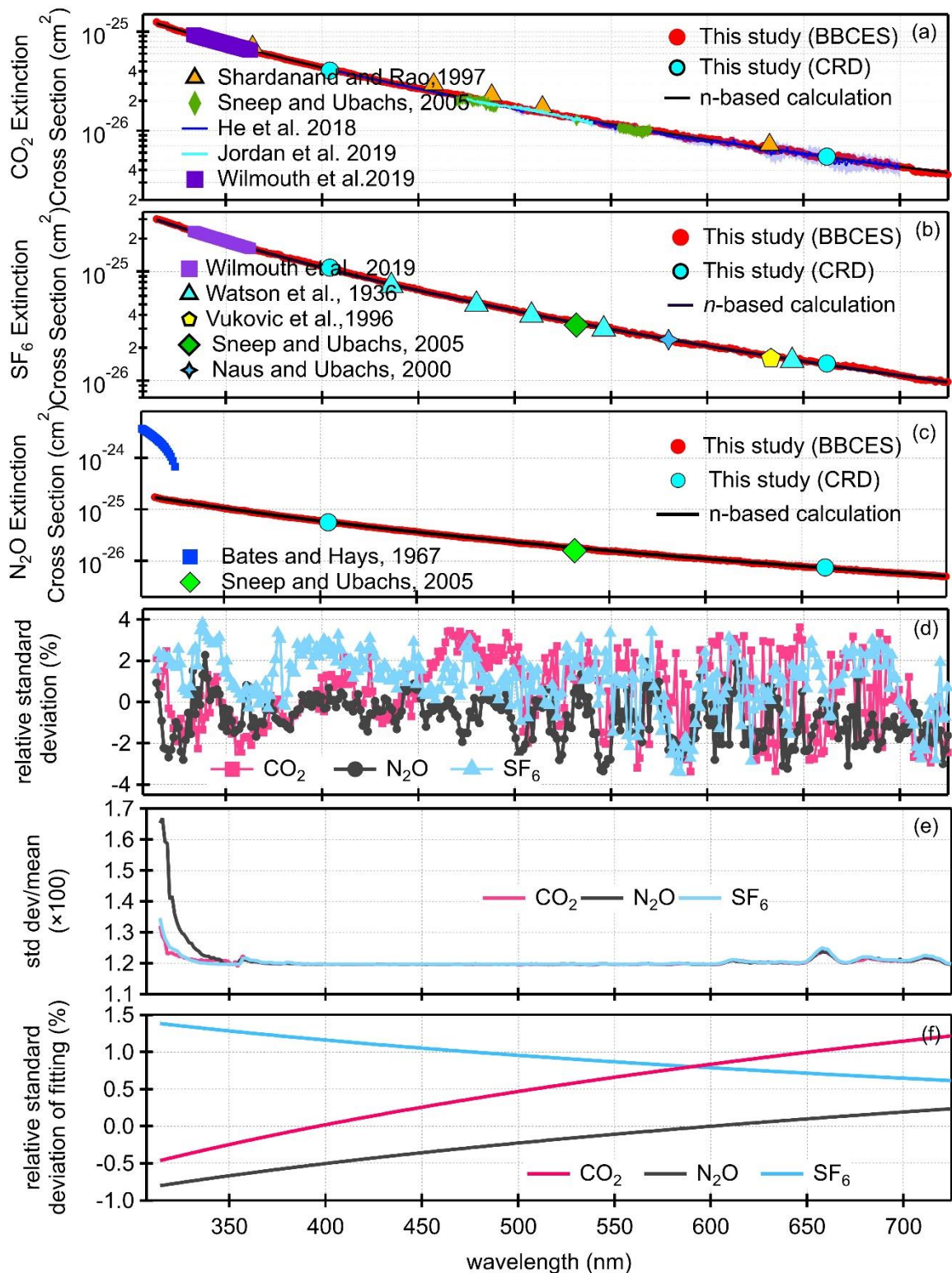
537 Q.H., S.S., and Y.R. designed this study. Q.H., Z.F., and O.S. conducted the experiments. Q.H.
538 prepared the draft and all of the co-authors reviewed it and provided comments.

539 **Competing interests.**

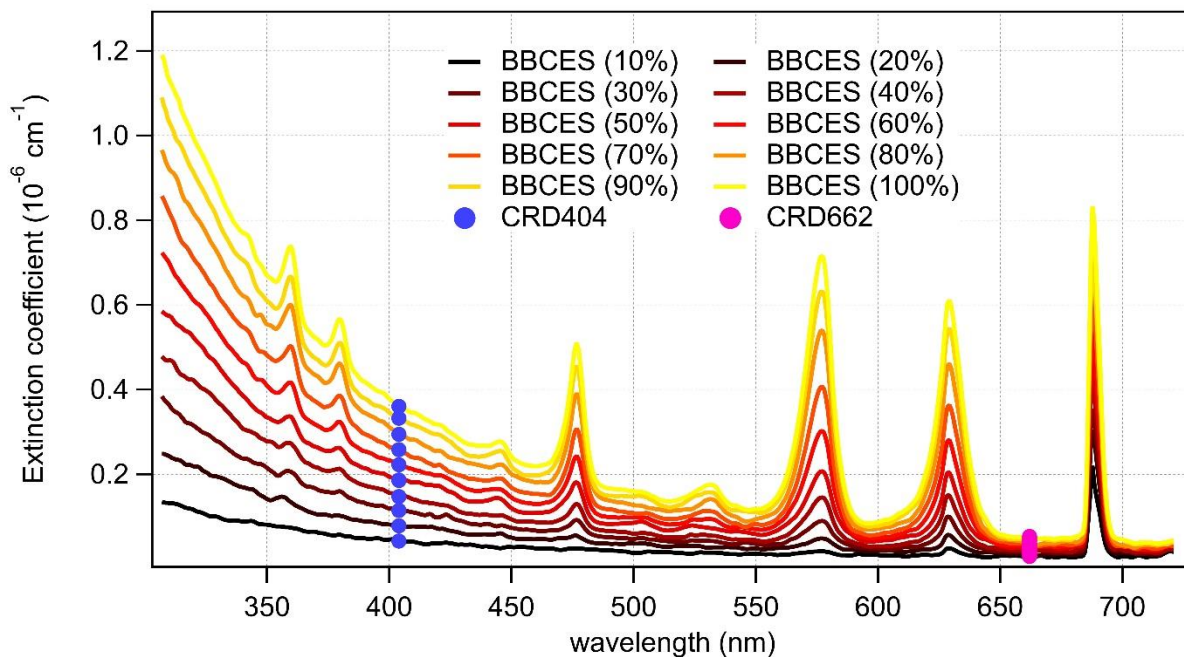
540 The authors declare that they have no conflict of interest.

541 **Acknowledgments**

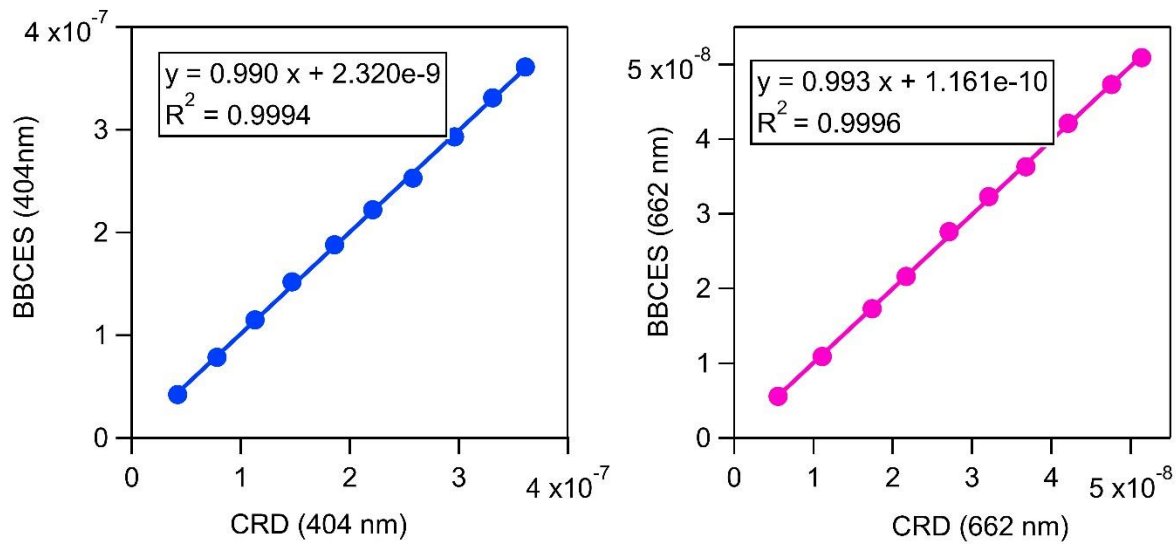
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545 Weizmann Institute.



546
 547 Figure 1. Rayleigh scattering cross sections of CO₂ (a), SF₆ (b), and N₂O (c). Panel (d) shows the
 548 relative standard deviations as a function of wavelength for each gas. The relative difference in the
 549 cross sections obtained by our measurements and calculations from the refractive index are
 550 displayed (e). Panel (f) shows the relative difference after fitting ($\sigma=A\lambda^B$).

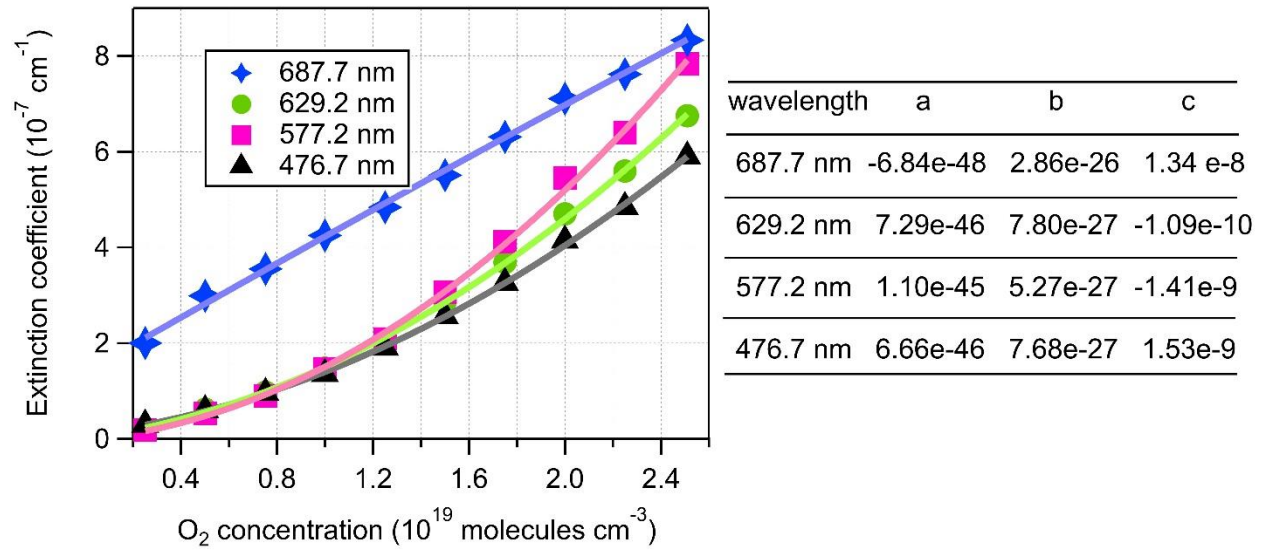


551
 552 Figure 2. Wavelength-dependent extinction coefficients of O₂ + He mixtures as a function of O₂
 553 concentration. The colored lines represent the extinction coefficients measured by BBCES system,
 554 and markers represent results from CRDS setup.



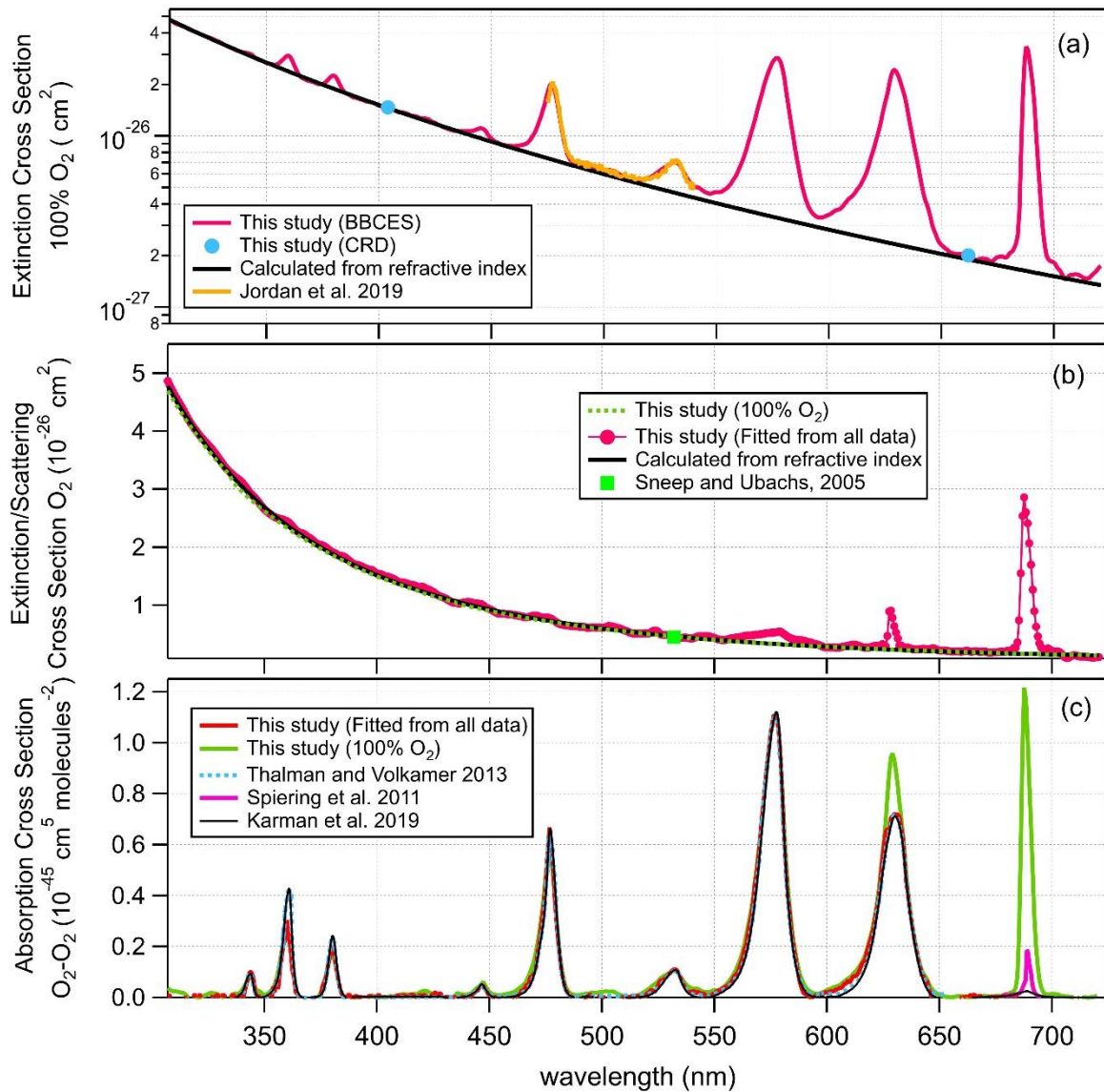
555

556 Figure 3. Correlations between the extinction coefficients (unit, cm^{-1}) measured by the BBCES
 557 system and CRDS setup.



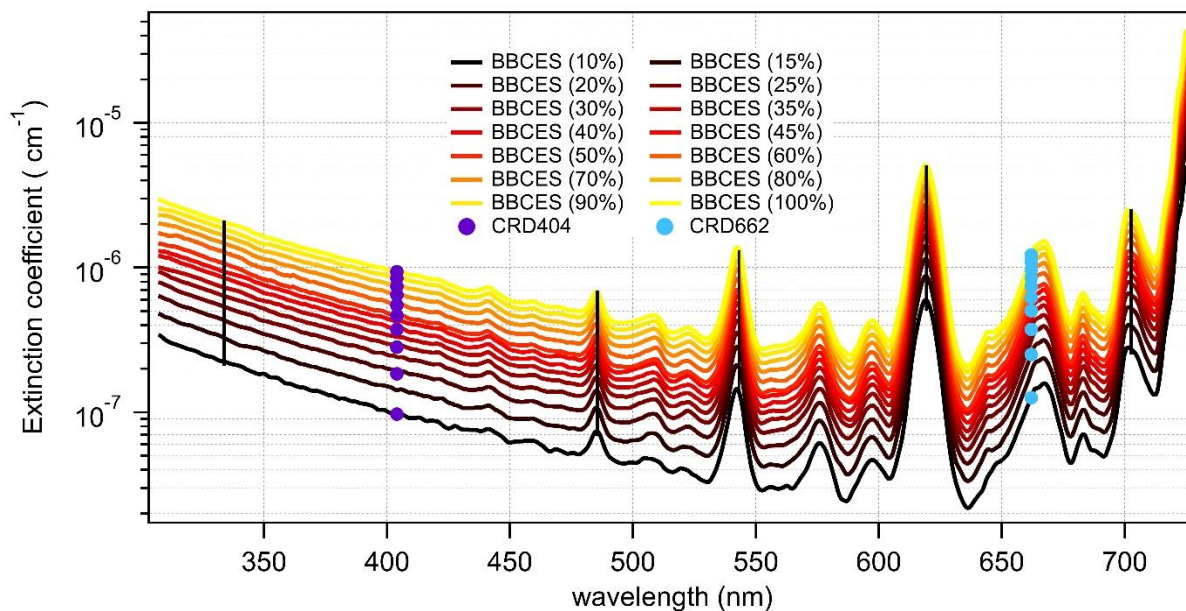
558

559 Figure 4. 2nd order polynomial fit of extinction coefficients measured by the BBCES system. The
 560 O₂ concentration-dependent extinction coefficients are contributed by the extinction coefficients
 561 of O₂ (σ_{O_2}), He (σ_{He}), and the O₂-O₂ CIA cross sections ($\sigma_{O_2-O_2}$).

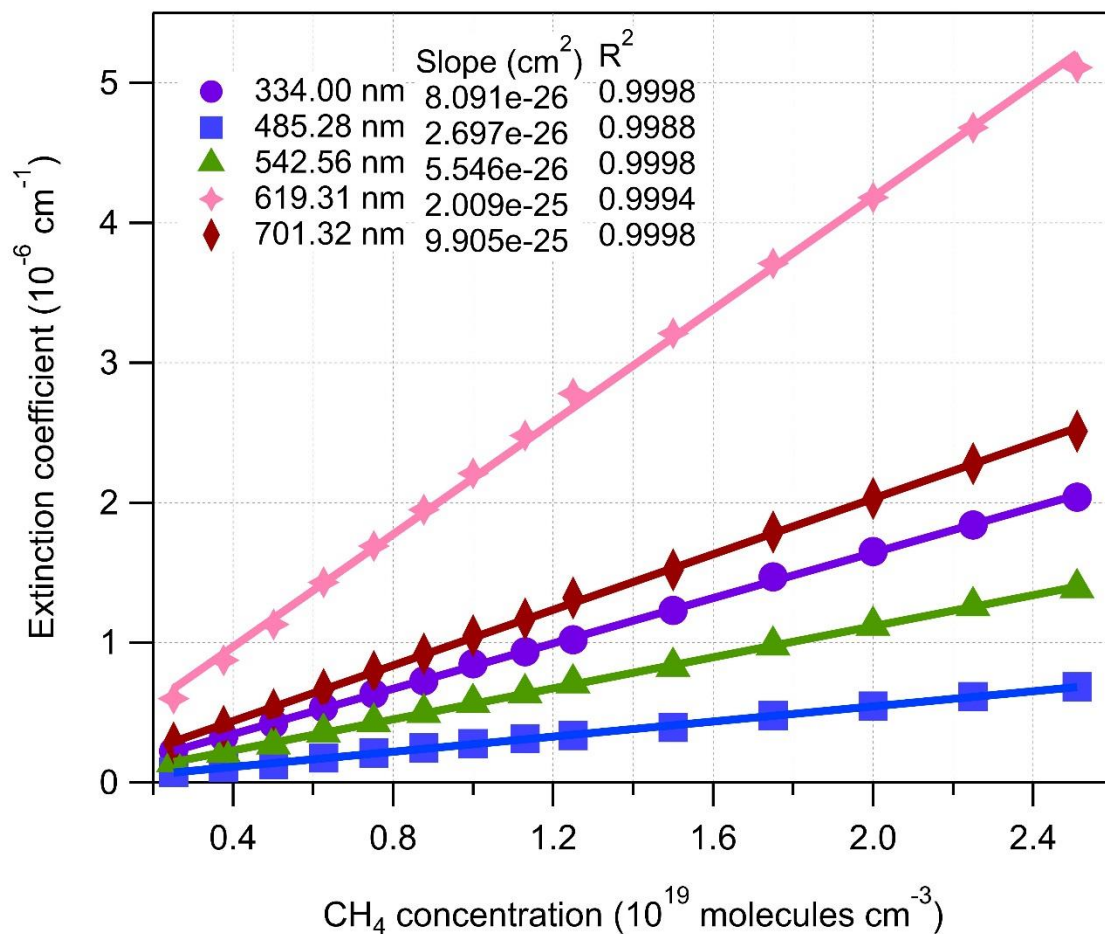


562

563 Figure 5. Wavelength-dependent extinction cross sections of 100% O₂ (a), extinction cross
 564 sections of O₂ (b), and O₂-O₂ CIA cross section (c).

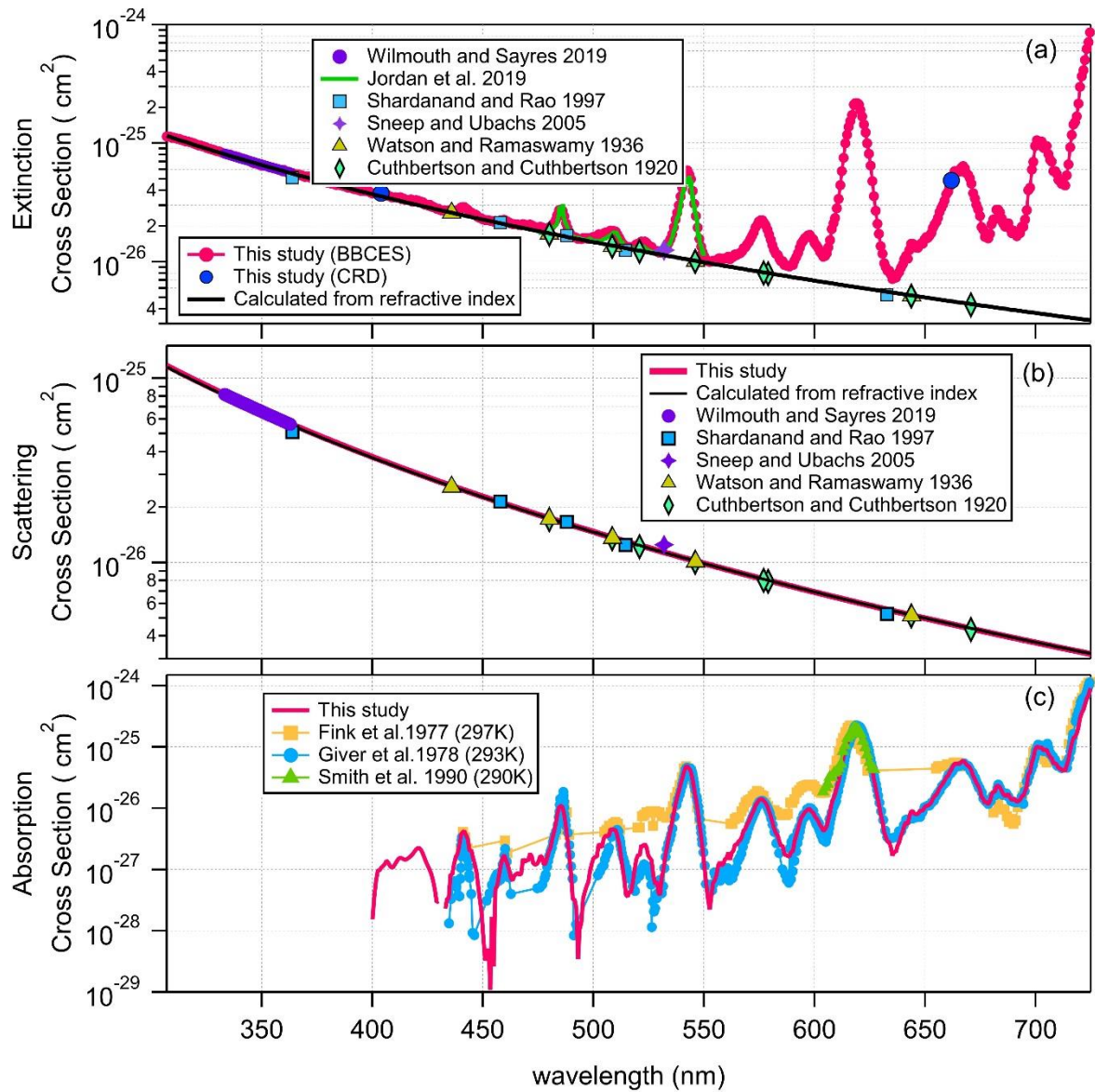


565
 566 Figure 6. Wavelength-dependent extinction coefficients of CH₄ + He mixtures as a function of
 567 CH₄ mixing ratio. The colored lines represent extinction coefficients obtained from BBCES system
 568 and markers represent results from CRDS setup. Measurements were performed with CH₄
 569 percentage within 10% and 100% with a 10% step. Moreover, BBCES measurements were also
 570 performed for 15%, 25%, 35%, and 45% CH₄. The number concentration of 100% methane was
 571 2.50143×10^{19} molecules cm⁻³. Data at selected wavelengths (vertical lines) are shown in Figure 7.



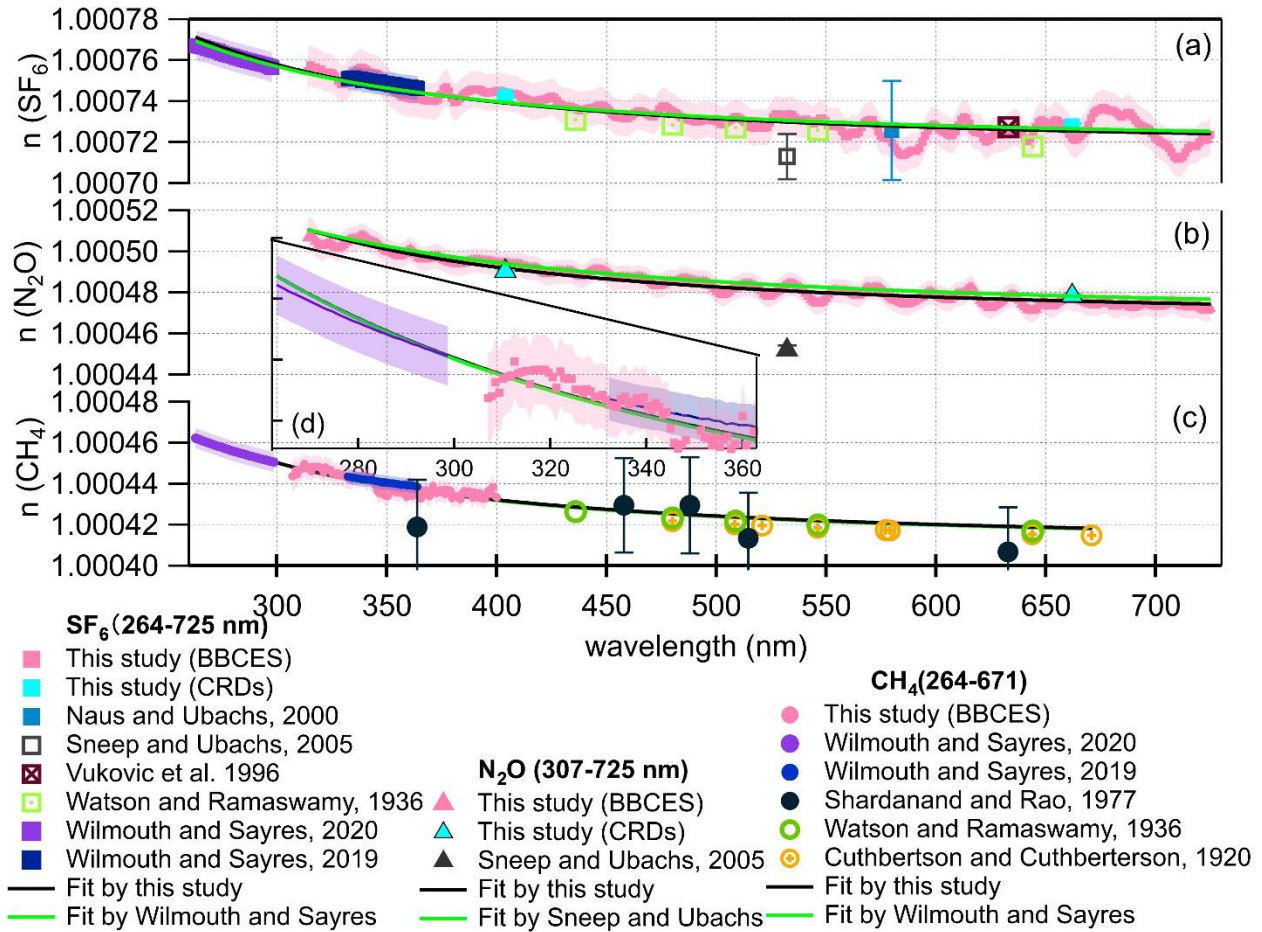
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573 Figure 7. The relationship between BBCES system measured extinction coefficients of CH₄+He
 574 mixtures and CH₄ concentration. The selected wavelengths were located in Figure 6 by vertical
 575 lines.



576

577 Figure 8. Extinction (a), scattering (b), and absorption (c) cross sections of CH₄.



578
 579 Figure 9. Real refractive index (n) for SF_6 (a), N_2O (b), and CH_4 (c). Comparison of Refractive
 580 index from this work with previous studies (Cuthbertson and Cuthbertson, 1920; Naus and Ubachs,
 581 2000; Shardanand and Rao, 1977; Sneep and Ubachs, 2005; Vukovic et al., 1996; Watson et al.,
 582 1936; Wilmouth and Sayres, 2019, 2020) over the wavelength range of 264–725 nm. The green
 583 line represents the dispersion relation given in Table 1. The black line represents the dispersion
 584 relation given in Eq. (8–10) derived from a fit to our data and references results. The shading
 585 represents 1- σ uncertainty of the n . The n values for Shardanand and Rao (1977), Sneep and
 586 Ubachs (2005), Naus and Ubachs (2000) were calculated from their reported Rayleigh scattering
 587 cross sections. Refractive index data from Sneep and Ubachs (2005) are not used in the fitting
 588 since these results are away from others. Data from Shardanand and Rao (1977) are not used due
 589 to large uncertainties. All of the data sets are equally weighted during fitting. Panel (d) is a close-
 590 up view of panel (c) in the wavelength range of 264–363 nm.

591 Table 1. Refractive index and King correction factors for calculating Rayleigh scattering cross-
592 sections and available measurements in the wavelength range of 300–725 nm. Measurements for
593 He and N₂ are not summarized in this table.

Gas	Refractive index and King correction factors				Measurements	
	$(n-1) \times 10^8$	$F_k(v)$	v (cm ⁻¹)	References	λ (nm)	References
He	2283 + $\frac{1.8102 \times 10^{13}}{1.5342 \times 10^{10} - v^2}$	1.0	14285- 33333	Abjean, 1970; Leonard, 1974 ; Cuthbertson, 1932		
N ₂	5677.465 + $\frac{318.81874 \times 10^{12}}{1.44 \times 10^{10} - v^2}$	$1.034 + 3.17 \times 10^{-12} v^2$	21360- 39370	Bates 1984 Sneep, 2005; Naus, 2000		
CO ₂	$\frac{1.1427 \times 10^{11}}{5799.25}$ $\times \left(\frac{(128908.9)^2 - v^2}{120.05} \right)$ + $\frac{(89223.8)^2 - v^2}{5.3334}$ + $\frac{(75037.5)^2 - v^2}{4.3244}$ + $\frac{(67837.7)^2 - v^2}{1.218145 \times 10^{-5}}$ + $\frac{(2418.136)^2 - v^2}{1.218145 \times 10^{-5}}$	$1.1364 + 2.53 \times 10^{-11} v^2$	39417- 55340	<i>Alms, 1975</i> ; Bideau-Mehu, 1973 ; Sneep, 2005	333-725	Jordan, 2019; Shardanand, 1977; Sneep, 2005; Wilmouth, 2019; He, 2018
CH ₄	4869.8 + $\frac{4.1023 \times 10^{14}}{1.133 \times 10^{10} - v^2}$	1.0	15385- 40000	<i>Sneep, 2005</i> ; Wilmouth, 2020	264-297 333-363, 434-725	Cuthbertson 1920; Jordan, 2019; Shardanand, 1977; Sneep, 2005; Watson, 1936; Wilmouth, 2019;2020; Smith, 1990; Giver, 1978; Fink, 1977
N ₂ O	46890 + $4.12 \times 10^{-6} v^2$	$\frac{3.3462 + 70.8 \times 10^{-12} v^2}{2.7692 - 47.2 \times 10^{-12} v^2}$	15453- 21838	<i>Alms, 1975</i> ; Sneep, 2005	300-320, 532	Johnston, 1975; Sneep, 2005
SF ₆	18611.4 + $\frac{8.9566 \times 10^{14}}{1.680 \times 10^{10} - v^2}$	1.0	15385- 40000	<i>Sneep, 2005</i> ; Vukovic, 1996; Wilmouth, 2020	264-297 333-363, 532, 633	Sneep, 2005; Vukovic, 1996; Wilmouth, 2019,2020
O ₂ ^a	20564.8 + $\frac{2.480899 \times 10^{13}}{4.09 \times 10^9 - v^2}$	$1.09 + 1.385 \times 10^{-11} v^2$ + $1.448 \times 10^{-20} v^4$	18315- 34722	Bates 1984 ; Hohm, 1993; <i>Sneep, 2005</i>	328-667	Thalman, 2013; Jordan, 2019; Hermans, 1999; Greenblatt, 1990; Spiering, 2011

594 Unless noted, the refractive index is scaled to 288.15 K and 1013.25 hPa. $N = 2.546899 \times 10^{19}$
595 molecules cm⁻³.

596 Due to limited space, only the first name of each reference is shown in the table.

597 The references in bold and italics describe the formulation of refractive index and King correction
598 factor for n -based calculation, respectively.

599 ^a The refractive index was obtained at 273.15 K and 1013.25 hPa, $N = 2.68678 \times 10^{19}$ molecules
600 cm⁻³ is used in Eq. (1)

601 Table 2. The Rayleigh scattering cross sections (10^{-27} cm²) calculated from the refractive index
 602 (*n*-based) and obtained from BBCES system (Exp) of selected wavelengths.

$\lambda(\text{nm})$	CO ₂		SF ₆		N ₂ O		O ₂		CH ₄	
	<i>n</i> -based	Exp	<i>n</i> -based	Exp	<i>n</i> -based	Exp	<i>n</i> -based	Exp	<i>n</i> -based	Exp
330	98.22	96.8	241.5	239.4	137.9	136.7	34.71	35.1	84.12	85.3
404	41.67	41.6	104.5	105.7	57.71	57.9	14.57	14.8	35.57	35.9
532	13.32	13.3	33.92	34.1	18.19	18.3	4.642	4.55	11.34	11.3
660	5.516	5.52	14.16	14.2	7.483	7.47	1.924	1.95	4.693	4.68
710	4.101	4.08	10.55	10.4	5.551	5.48	1.430	1.41	3.487	3.47

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