

Laboratory study of the collection efficiency of submicron aerosol particles by cloud droplets.

Part I - Influence of relative humidity

Alexis Dépée^{1,2}, Pascal Lemaître^{1*}, Thomas Gelain¹, Marie Monier^{2,3}, Andrea Flossmann^{2,3}

[1] {Institut de Radioprotection et de Sûreté Nucléaire (IRSN), PSN-RES, SCA, Gif-sur-Yvette, 91192, France}

[2] {Université Clermont Auvergne, Laboratoire de Météorologie Physique, Clermont-Ferrand, France}

[3] {CNRS, INSU, UMR 6016, LaMP, Aubière, France}

*Correspondence to: Pascal Lemaître (pascal.lemaître@irsn.fr)

ABSTRACT

A new In-Cloud Aerosol Scavenging Experiment (In-CASE) has been conceived to measure the collection efficiency (CE) of submicron aerosol particles by cloud droplets. In this setup, droplets fall at their terminal velocity through a one-meter-high chamber in a laminar flow containing aerosol particles. At the bottom of the In-CASE's chamber, the droplet train is separated from the aerosol particle flow - droplets are collected in an impaction cup whereas aerosol particles are deposited on a High Efficiency Particulate Air (HEPA) filter. The collected droplets and the filter are then analysed by fluorescence spectrometry since the aerosol particles are atomised from a sodium fluorescein salt solution ($C_{20}H_{10}Na_2O_5$). In-CASE fully controls all the parameters which affect the CE - the droplets and aerosol particles size distributions are monodispersed, the electric charges of droplets and aerosol particles are controlled, while the relative humidity is indirectly set via the chamber's temperature. This novel In-CASE setup is presented here as well as the first measurements obtained to study the impact of relative humidity on CE. For this purpose, droplets and particles are electrically neutralised. A droplet radius of $49.6 \pm 1.3 \mu\text{m}$ has been considered for six particle dry radii between 50 and 250 nm and three relative humidity levels of 71.1 ± 1.3 , 82.4 ± 1.4 and 93.5 ± 0.9 %. These new CE measurements have been compared to theoretical models from literature which adequately describe the relative humidity influence on the measured CE.

INTRODUCTION

Every year, several billion tons of particulate matter are emitted in the atmosphere, originating mainly from oceans, soils, gas-to-particle conversion, evaporating clouds and from human activities (Jaenicke, 1993). During the last decades, the lifecycle of these aerosol particles (APs) has been a key topic in atmospheric science for many reasons. First, APs play a key role in weather and climate. They act on cloud formation and their chemical composition, size distribution and number concentration affect the droplet size distributions and precipitation (Tao et al., 2012). They also have an impact on the cloud cover which in turn modulates albedo (Twomey et al., 1974) - influencing the Earth's energy budget. Moreover, anthropogenic APs have also been reported impacting human health (Dockery *et al.*, 1992). In fact, the Great Smog of London in 1952, one of the best-known related events, caused up to 12,000 deaths (Bell et al., 2004). Radioactive material released from a nuclear accident is another AP pollution hazard. Indeed, many studies revealed that radioactive material like caesium-137 isotopes can attach to the atmospheric APs and were transported over long distances on a continental scale both after the Chernobyl (Devell et al., 1986 ; Jost et al., 1986 ; Pöllänen et al., 1997) and the Fukushima (Kaneyasu et al., 2012 ; Adachi et al., 2013) nuclear accidents in 1986 and 2011, respectively. With a half-life up to thirty years, caesium-137 can remain for decades in the atmosphere - following resuspension cycles of the atmospheric APs - and jeopardise both ecosystems and human health.

Thus, it is essential to understand the two mechanisms which move atmospheric APs back to the ground. APs can settle through many effects like gravity, wind, surface forces, turbulence, etc. This is referred as dry AP deposition. There is also the wet AP deposition due to the interactions between

50 APs and clouds or their precipitations. The present paper deals with the wet removal since, far away
51 from the source, it is the main mechanism involved in the AP scavenging (Jaenicke, 1993). Note that
52 Flossmann (1998) numerically showed that the wet deposition is mainly induced by the in-cloud AP
53 collection since 70 % of the AP mass contained in raindrops reaching the soil comes from the cloud.
54 This result is consistent with the environmental measurements of Laguionie et al. (2014) who
55 evaluated the cloud contribution up to 60 % in the wet AP deposition. The in-cloud AP scavenging is
56 subdivided into two mechanisms - AP activation to form cloud hydrometeors and AP collection by
57 clouds hydrometeors. The in-cloud AP collection is therefore a fundamental climate, weather and
58 health issue. In most of current AP wet removal models - like DESCAM (Detailed SCAvenging Model,
59 Flossmann, 1985) - the AP collection is described through a microphysical parameter called
60 "collection efficiency" (CE) which quantifies the ability of a droplet to capture the APs present in its
61 surroundings during its fall. It is the ratio between the AP number (or mass) collected by the droplet
62 over the AP number (or mass) within the volume swept by the droplet for a given AP radius. Another
63 equivalent definition is the ratio of the cross-sectional area inside which the AP trajectories are
64 collected by the droplet over the cross-sectional area of the droplet.

65 Many microphysical effects influence this CE and their contribution is mainly depending on the AP
66 size. To be collected an AP has to deviate from the streamline around the falling droplet to make
67 contact with it. The nanometric AP's trajectory is affected by the collisions with air molecules -
68 referred as the Brownian diffusion. It results in random movement patterns (see Figure 1, A) which
69 tend to increase the CE when the AP radius decreases. For massive APs, there is an increase of CE as
70 they retain an inertia strong enough to deviate significantly from the streamline when it curves and
71 to move straight toward the droplet surface - known as inertial impaction (see Figure 1, B). When
72 considering intermediate AP size, the CE goes through a minimum value called the "Greenfield gap"
73 (Greenfield, 1957) where the AP diffusion and inertia are weaker. In this gap, other microphysical
74 effects can be involved to make the droplet encounter the AP like the interception for instance. It is
75 the collection of APs following a streamline that approaches the droplet within a distance equivalent
76 to the particle radii (a) - see Figure 1, C. Note that the electrostatic forces can have a significant
77 influence on the CE (Tinsley and Zhou, 2015; Dépée et al., 2019). This effect will be discussed in a
78 companion paper (Dépée et al., 2020) of this work.

79 There are also thermophoretic and diffusiophoretic effects which can influence the CE. In clouds,
80 they shall favour the CE increase when evaporation occurs and decrease CE during condensation (due
81 to a thermal equilibrium between the droplet and the air). Thermophoresis exists when a thermal
82 gradient prevails between the air and the droplet. When the relative humidity (RH) is below 100 %,
83 the evaporating droplet's surface temperature ($T_{d,s}$) is colder than the bulk air temperature (T_{air}).
84 The average kinetic energy of air molecules is then decreasing when approaching the droplet's
85 surface. An AP is thus attracted by a thermophoretic force near the evaporating droplet (see Figure
86 1, F) caused by the asymmetry in kinetic energy transferred during each collision. Diffusiophoresis
87 occurs in an environment where a gradient of vapor density in the air exists such as in the surrounding
88 of an evaporating droplet. In this case, water molecules diffuse toward the surrounding air meanwhile
89 the air molecules diffuse toward the droplet surface. In clouds, since the water molar mass is lower
90 than the air molar mass, there is an asymmetry in the momentum transferred to APs close to the
91 evaporating droplet produced by collisions with the molecules from the continuous phase. This
92 diffusion tends to attract the AP to the droplet. Nonetheless, in order to maintain a constant air
93 pressure at the droplet surface, a hydrodynamical flow directed toward the air is induced - this is
94 the Stefan flow. The hydrodynamical drag induced by the Stefan flow tends to repulse APs from an
95 evaporating droplet. Diffusiophoresis is the sum of the drag force produced by Stefan flow and the
96 momentum transferred to APs (located in a diffusion boundary layer), due to the dissymmetry of
97 molecular weight. Note that the Stefan flow (repulsive around an evaporating cloud droplet) is on
98 average five times larger than the addition to the diffusion flows (attractive around an evaporating
99 cloud droplet) as mentioned by Santachiara et al. (2012). So, diffusiophoresis repulses APs from the
100 evaporating droplet (see Figure 1, D) which in turn decreases the CE. Finally, since the amplitude of
101 the thermophoresis is on average twice larger than the diffusiophoresis (Tinsley et al., 2006), APs are
102 ultimately attracted toward droplets in subsaturated air due to these phoretic effects (see Figure 1,
103 E). Thus, the coupling of the thermophoresis and diffusiophoresis increases the CE when the relative
104 humidity decreases.

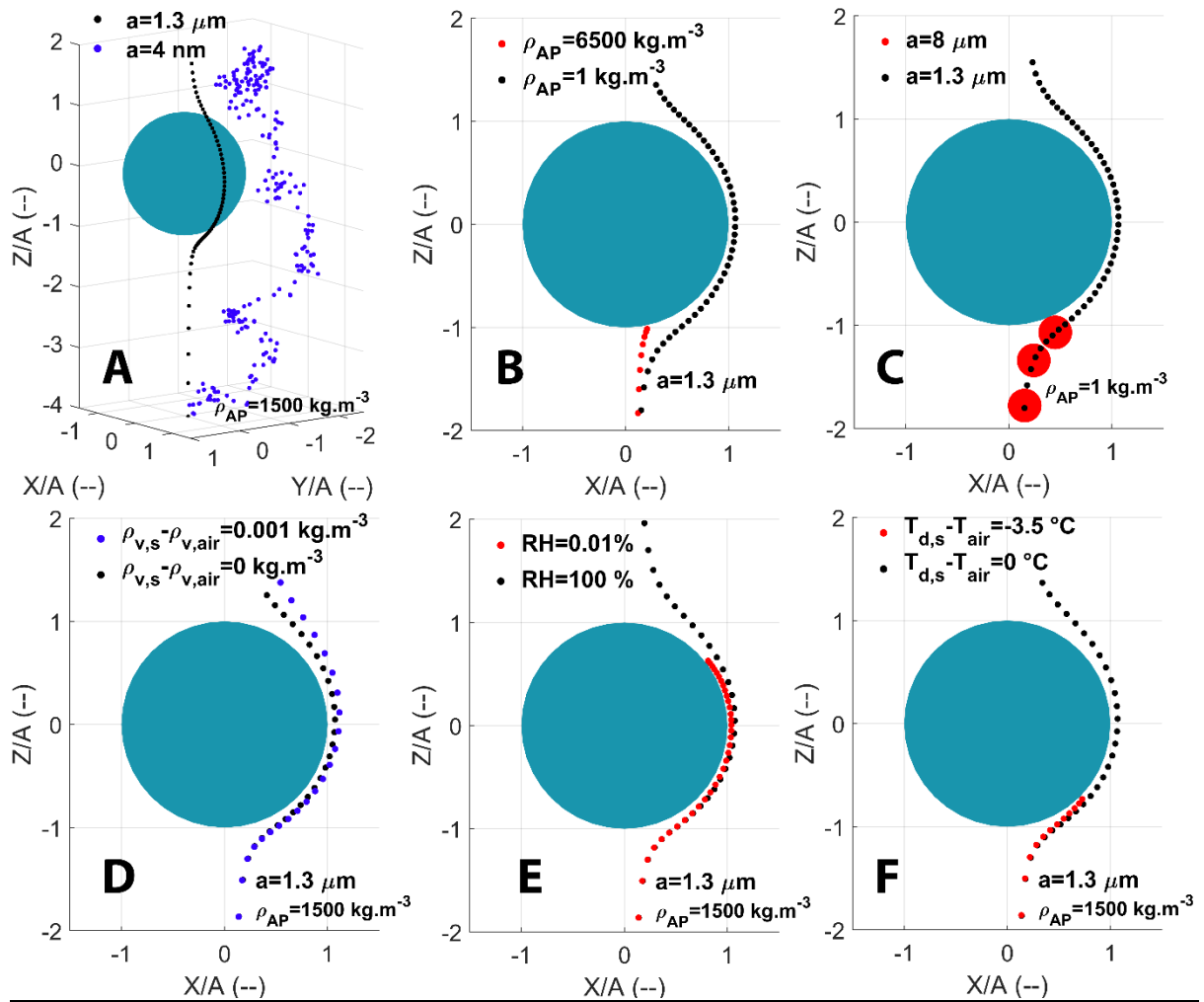
105 The influence of the relative humidity on the CE is described by the well-known Wang et al. (1978)
106 model which is used in many cloud models like DESCAM (Flossmann, 1985). Since their model predicts
107 an important contribution of thermophoresis and diffusiophoresis on the CE for cloud droplet radii
108 ($A < 100 \mu\text{m}$) and submicron AP radii, it is desirable to validate those theoretical CEs through
109 experiments. A review of available CE measurements can be found in Ardon-Dryer et al. (2015). The

110 only experimental study that tackles the influence of the relative humidity on the CE for cloud
111 droplets is the one of Ardon-Dryer et al. (2015), which tested two levels of relative humidity of 15
112 and 88 %. However, in their work they report that the measured electric charge on the droplets were
113 400 ± 400 elementary charges and on the APs were 1 elementary charge. Therefore, the electrostatic
114 forces should have had a significant influence on the measured CE for the droplet radius considered
115 ($A \approx 21.6 \mu\text{m}$) as numerically shown by Tinsley and Zhou (2015). Furthermore, there are no similar
116 measurements for other cloud droplet sizes neither for high levels of relative humidity as found in-
117 cloud.

118 The purpose of this work is to fill up the deficiency of data in this area. Thus, a novel experiment
119 has been developed in order to study the influence of the relative humidity on the CE to assess the
120 magnitude of the thermophoretic and diffusiophoretic processes. With this experiment, the influence
121 of electric charges can also be investigated and this is the object of a companion paper (D ep ee et
122 al., 2020).

123 In the first section of this paper, the experimental setup is detailed while the experimental method
124 to evaluate the CE and the uncertainties are described in the second one. The third section is
125 dedicated to the new CE measurements which are presented and compared to theoretical data from
126 the Wang et al. (1978) Eulerian model. Another comparison is made in the last section to the newer
127 Lagrangian model of D ep ee et al. (2019) since it can model every microphysical effect involved in
128 the AP collection by cloud droplets (like Brownian motion, inertial impaction, interception, etc.) and
129 especially their coupling. D ep ee et al. (2019) focused on the electrostatic forces but did not consider
130 the thermophoresis and the diffusiophoresis. Here, we extend the D ep ee et al. (2019) model by
131 adding these phoretic effects. Finally, this study experimentally validates the D ep ee et al. (2019)
132 model which provides consistent theoretical CEs for a convenient incorporation in cloud models,
133 pollution models, climate models, and so forth.

134
135
136
137
138
139
140
141
142
143
144
145
146
147
148
149
150
151



152
153
154
155
156
157
158
159
160
161
162
163
164
165
166
167
168
169
170
171
172
173
174
175
176
177
178

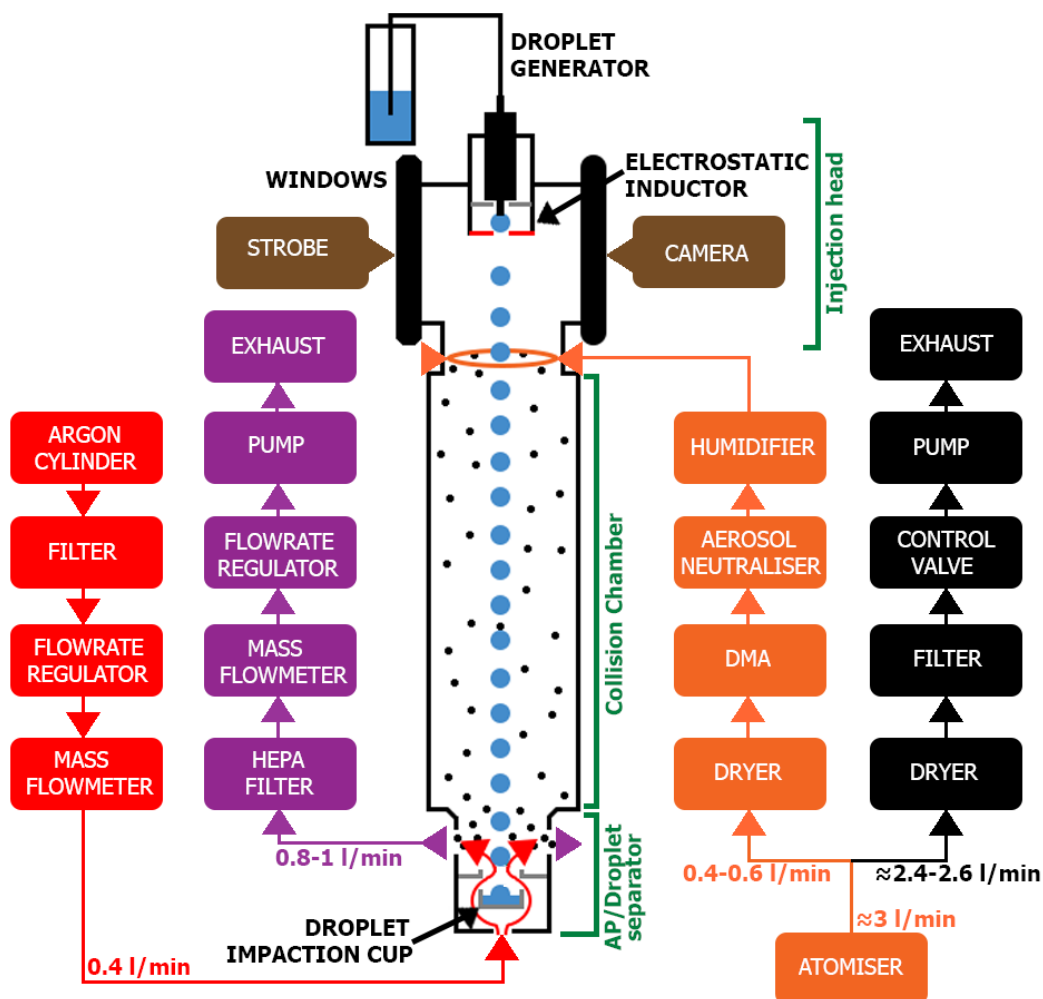
Figure 1 APs trajectories computed with the extended Dépée et al. (2019) model for a 50 μm droplet radius (A) and AP with various radii (a) and densities (ρ_{AP}). The air temperature (T_{air}) and the air pressure (P_{air}) are respectively -17°C and 540 hPa. The panels indicate the effects of Brownian motion (A), inertial impact (B), interception (C), diffusiophoresis (D), combined thermophoresis and diffusiophoresis (E), and thermophoresis (F). Red trajectories result in an AP collection. In Figure D and F, the gradients are equivalent to a relative humidity of 0.01 % (when there is no gradient the equivalent relative humidity is 100 %). $\rho_{v,s}$ and $\rho_{v,air}$ are the vapor densities at the droplet surface and in the bulk air, respectively.

1 EXPERIMENTAL SETUP

1.1 Overview

All measurements were conducted inside the In-Cloud Aerosol Scavenging Experiment (In-CASE). Figure 2 shows the airflow diagram with the different parts of the experiment in order to study the relative humidity influence on the CE. The major In-CASE's component is the collision chamber (Figure 2) where a laminar flow containing APs interacts with a train of droplets falling at terminal velocity. In this chamber, the droplet and AP size distributions are monodispersed and for this particular work the droplet and AP electric charges are neutralised. Droplets are generated through a piezoelectric inductor, neutralised with an electrostatic inductor (detailed in 1.4) and the radius is measured by optical shadowgraphy with a strobe and a camera (brown, Figure 2). In Figure 2, the orange shows the AP generation (which is described in subsection 1.3), the black - an exhaust used to evacuate the AP flowrate surplus at the atomiser's outlet, the red - an Argon flow injected into the In-CASE chamber's bottom part to separate droplets from the AP flow (this is detailed in

179 subsection 1.2.2.1) meanwhile the purple indicated the AP flow which leaves the chamber toward a
 180 High Efficiency Particulate Air (HEPA) filter.
 181 The relative humidity in the collision chamber is set through the temperature, this latter being
 182 controlled via a cooling system. In the next sections, the In-CASE's chamber as well as the droplets
 183 and AP characterisation are described.



184 Figure 2 In-CASE setup to study the influence of relative humidity. Colors represent
 185 different functions. Red - upward Argon flow against AP pollution in the droplet impaction
 186 cup. Purple - AP (and Argon) evacuation toward the HEPA filter. Orange - AP, generation,
 187 selection and neutralisation. Black - surplus evacuation and DMA flowrate control. Brown -
 188 droplet radius measurement. All the key features of the setup are detailed in Table 2.
 189

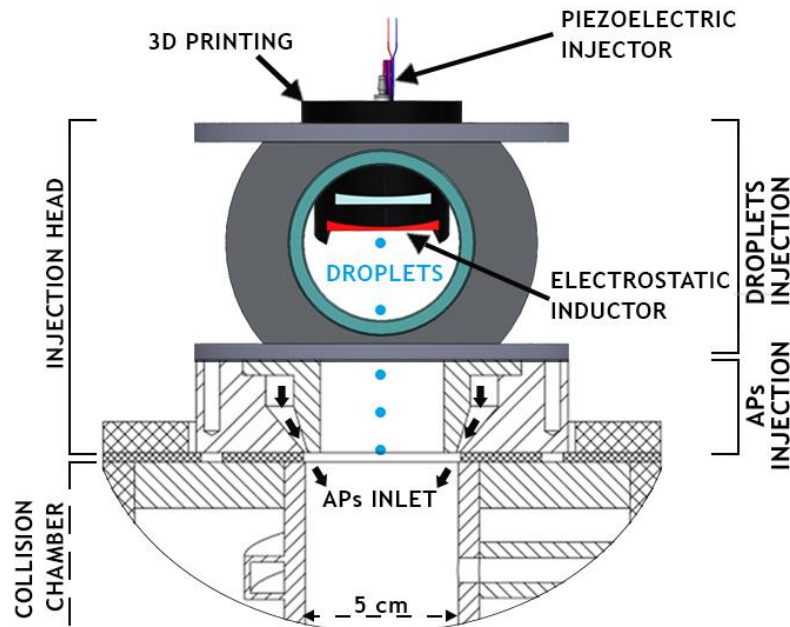
194 1.2 In-CASE chamber

196 The In-CASE chamber (see Figure 2) is subdivided into three stages - the injection head, the collision
 197 chamber and the In-CASE chamber's bottom part. These three parts will be detailed in the next
 198 subsections.

200 1.2.1 Injection head

202 The injection head is composed of two parts - the droplets and the APs injection. The upper part is
 203 used to inject the droplets while the APs are injected in the second part about 10 cm below. This
 204 distance is required to measure the droplet size through the two facing windows (see section 1.4.1)
 205 but also to let droplets decelerate and reach their terminal velocity.

206 The droplet train is injected through a housing made with a 3D printer set at the top of the droplet
 207 injector (see Figure 3). This housing has been constructed to precisely place the droplet generator
 208 and the electrostatic inductor together. Indeed, the electrostatic inductor has to keep the same
 209 position relative to the droplet generator to prevent changes in the electric field E_{ind} which in turn,
 210 can disturb the droplet charge and stop the neutralisation (detailed in 1.4.2).
 211 The APs are injected from the sides of the entire circumference through a flat torus inlet. This
 212 injection principle is based on the CLINCH experiment (CoLision Ice Nucleation Chamber, Ladino et
 213 al., 2011) which ensures a laminar flow and a great spatial APs mixture in the collision chamber's
 214 inlet.
 215



216
 217 Figure 3 View of the In-CASE chamber's top with the injection head where APs and droplets are
 218 injected into the collision chamber.
 219

220
 221
 222

223 1.2.2 In-CASE's bottom stage

224
 225
 226
 227
 228

The CE is calculated from the AP mass collected by the droplets during an experiment and the average AP mass concentration in the collision chamber. To obtain these quantities, the droplet train must be separated from the interstitial APs (which were not collected).

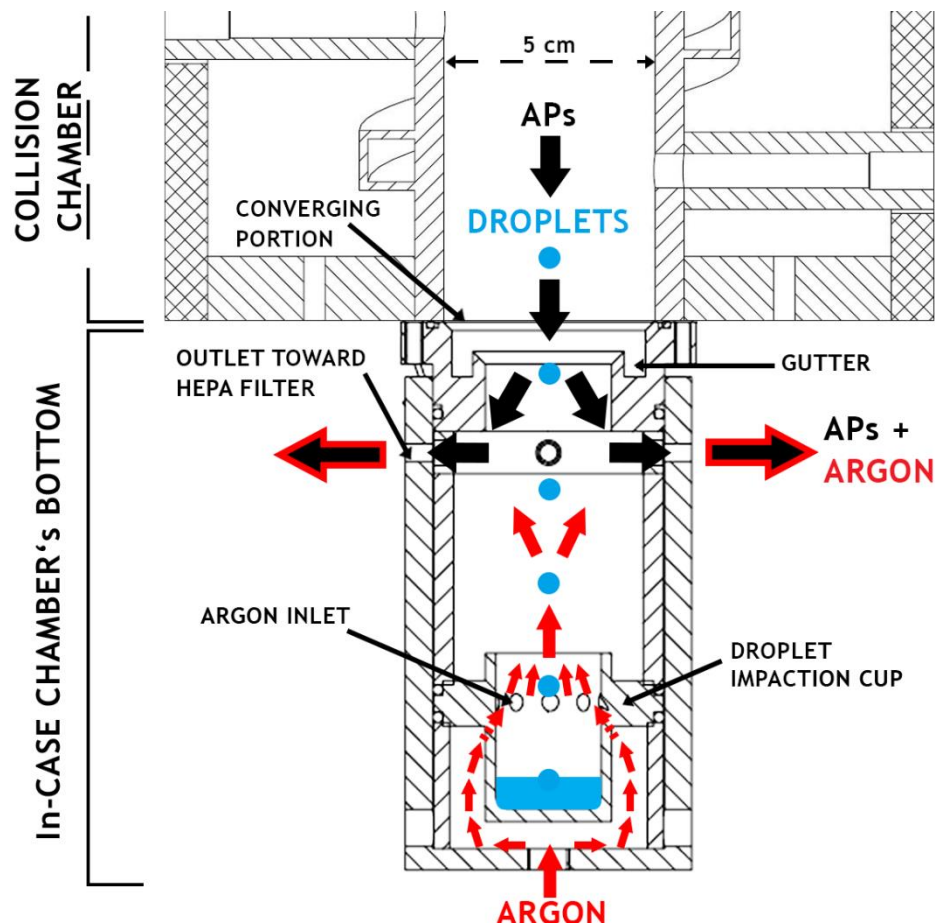
229 1.2.2.1 APs and droplets separation

230
 231
 232
 233
 234
 235
 236
 237
 238
 239
 240

The system developed to separate the droplet train from the AP flow is presented in Figure 4. It is composed of a converging portion (from 5 to 3 cm in diameter) where a gutter is inserted to prevent the water condensed on the wall from entering to the In-CASE's chamber bottom. The APs are directly vacuumed toward a HEPA filter (see Figure 2) at the upper part of the separator through four outlets while the droplets - containing collected APs - are impacted into a cup at the separator's lower part. To prevent AP pollution in the droplet impaction cup, a counter-flow is injected below the In-CASE's chamber and passes through the droplet impaction cup from nine holes set on its entire circumference. Since the counter-flow is injected at the laboratory temperature and the APs downward flow is colder, Argon - denser than the air - was selected to avoid any Rayleigh-Taylor instability (Sharp, 1983).

241 Argon is injected at 0.4 l/min. The diameter of the nine holes is 4 mm and the top of the droplet
 242 impaction cup is 2.5 cm. Thus, the upward Argon flow is injected at 5.9 and 1.4 cm/s, through the
 243 nine holes and the top of the impaction cup, respectively. Because the droplet velocity is about 25
 244 cm/s (for the 50 μm droplet radius studied) and the AP terminal velocity is less than 10⁻³ cm/s, APs
 245 can not settle into the impaction cup whereas droplets are impacted without undue disruption.

246
247
248



249
250
251
252
253
254
255
256
257
258
259
260
261
262
263
264
265
266
267
268
269
270
271
272
273
274
275
276

Figure 4 View of the In-CASE chamber's bottom - APs and droplets separation.

1.2.2.2 Validation

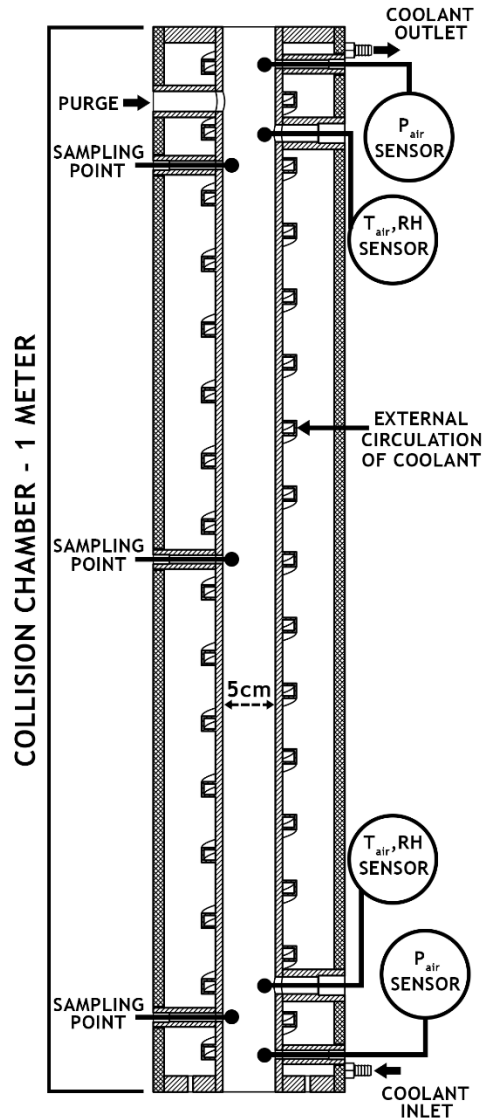
The droplets and APs separation were verified with two tests. First, In-CASE was run under standard experimental conditions except no droplets were generated. After five hours of experiment, a spectrometry analysis was performed in the droplet impaction cup and no fluorescein was detected. Thus, no AP had settled on the droplet impaction cup during the experiment.

The second test was to ensure that droplets were collected by the impaction cup. Then, In-CASE was again run like a typical experiment except the flow passing through the In-CASE chamber was clean air without any AP. Droplets were tracked by adding sodium fluorescein salt in the water supplying the piezoelectric injector. Since the concentration of sodium fluorescein salt in the water, the droplet generation frequency, the droplet size and the experiment time were known, the goal was to verify if the expected fluorescein mass in the droplets and the actual measured fluorescein mass were equal. After five hours (= 450,000 injected droplets), a discrepancy of 2 % between expected and measured fluorescein mass was obtained. Therefore, all droplets are considered impacted in the impaction cup.

Finally, this indicates that the AP mass detected in the droplet impaction cup after the experiment effectively results from collection by drops in the In-CASE collision chamber and not from contamination from other sources.

277 **1.2.3 Collision chamber**

278
 279 The collision chamber is a one-meter stainless steel cylinder with an inner diameter of 5 cm (see
 280 Figure 5). The collision chamber's temperature is controlled through a coolant which spirally
 281 circulates outside the chamber, from the bottom to the top of the collision chamber. The pressure
 282 (P_{air}), temperature (T_{air}) and relative humidity (RH) are measured at the top and the bottom by
 283 sensors. To clean the chamber, water or compressed dried air are injected via a purge. Three
 284 sampling points are available but were not used for these experiments.
 285 The temperature and the relative humidity discrepancies between top and bottom were respectively
 286 less than 1°C and 4 % in all the CE measurements - the mean values are then considered for the both
 287 parameters.
 288



289
 290 Figure 5 In-CASE collision chamber - 2D section plane.

291
 292 **1.2.3.1 Thermodynamic conditions**

293
 294 All the experiments were conducted at atmospheric pressure. To get comparable CE measurements,
 295 the temperature has been set to $0.58 \pm 0.50^{\circ}\text{C}$ - as constant as possible between experiments. Three
 296 levels of relative humidity (RH) were considered - 71.1, 82.4 and 93.5 %. To increase the relative
 297 humidity at a given collision chamber temperature, the temperature of the pure water in the
 298 humidifier (Figure 2) was increased. The relative humidity level of 71.1 % was obtained by completely
 299 removing the humidifier to get the driest AP flow possible at the collision chamber's inlet. At lab

300 temperature (about 22 °C), the relative humidity of the dry AP flow ranged from 10 to 20 % at the In-
301 CASE's chamber inlet.
302 Note that the AP flow before the injection head is also thermally set to inject APs with the same
303 temperature as in the collision chamber.

304 305 1.2.3.2 Droplet evaporation 306

307 The change in droplet radius due to evaporation in the collision chamber is calculated according to
308 the section 13.2 of Pruppacher and Klett (1997). The corresponding terminal velocity ($U_{A,\infty} \approx 25$ cm/s)
309 is computed from Beard (1976). The residence time of the droplet in the chamber (≈ 4 s) is computed
310 considering these two changes. Since the droplet radius only decreases around 3 % by evaporation
311 with the lower relative humidity considered in the experiments (71.1 %), the droplet evaporation in
312 the collision chamber is neglected.

313 314 1.2.3.3 AP hygroscopicity 315

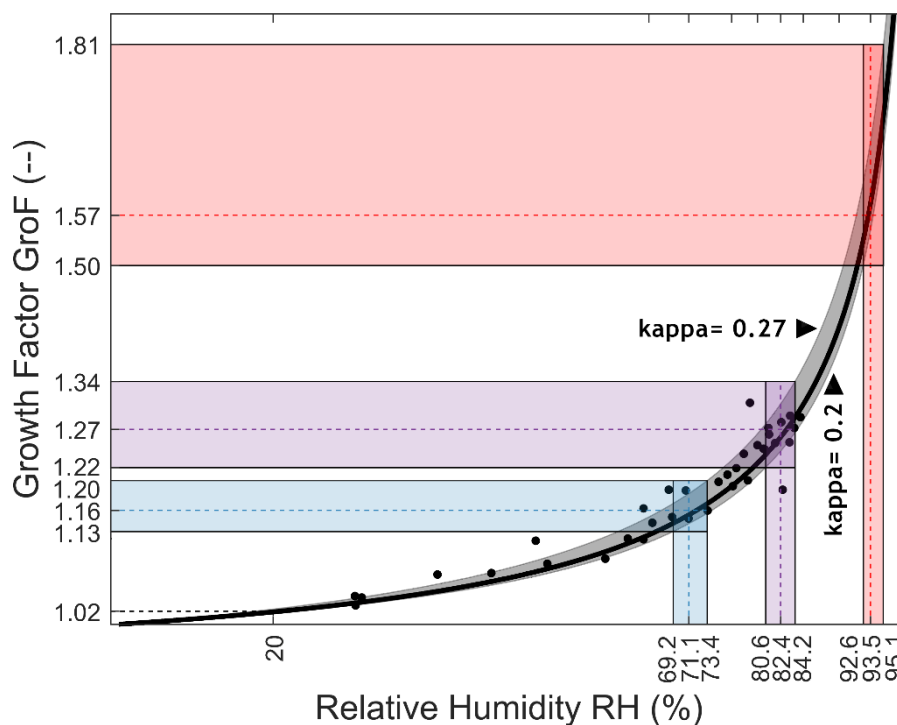
316 The APs are composed of pure sodium fluorescein salt which is a high hygroscopic chemical
317 compound. The APs inside the collision chamber then grow to reach their equilibrium size with the
318 relative humidity (RH). In order to evaluate the increase of size by humidification, the AP growth
319 factor ($GroF$) measured in Quérel et al. (2014) was considered. The growth factor is defined as the
320 ratio of the size of the wet AP over the size of the dry AP. Since their data are limited to relative
321 humidity levels below 90 %, the kappa-theory described in Petters and Kreidenweis (2007) is used to
322 extrapolate to the required values. To fit the measurements of Quérel et al. (2014) with the kappa-
323 theory, only their data with a relative humidity level less than 85 % were considered. Figure 6 shows
324 the AP growth factor related to the relative humidity for a kappa value of 0.23 and two extreme
325 values of 0.2 and 0.27 - fitting to the sodium fluorescein salt hygroscopicity.

326 Thus, for relative humidity levels of 71.1 %, 82.4 % and 93.5 % studied here, a dry AP radius of 50 nm
327 selected by the DMA grows with a growth factor ($GroF$) of 1.16, 1.27 and 1.57, respectively.
328 Consequently, the CE measured are applied for size of respectively 58.0, 63.5 and 78.5 nm AP radii.
329 Note that the AP density is not the one of sodium fluorescein salt ($\rho_{fluorescein} = 1580$ kg.m⁻³) since
330 APs contain water. Indeed, the water density (ρ_{water}) should be considered in the AP density (ρ_{AP})
331 calculation. At a given relative humidity (RH), the AP density inside the chamber is then deduced by
332 the equation (1):

$$\rho_{AP}(RH) = \frac{\rho_{fluorescein} + \rho_{water} \times [GroF(RH)^3 - 1]}{GroF(RH)^3} \quad (1)$$

333 Since the relative humidity after the dryer (see Figure 2) ranges from 10 to 20 %, the AP growth factor
334 is less than 1.02 (see Figure 6) in the DMA. APs are then considered dry when exiting the DMA.

335
336
337



338
339

340 Figure 6 Growth factor (*GroF*) as function of the given relative humidity (*RH*). Data points (dots)
341 from Qu erel et al. (2014) and fits (lines) with the kappa-theory (Petters and Kreidenweis, 2007).
342

343

344

345

1.3 AP generation

346

347

348

349

350

351

352

353

354

355

356

357

358

359

360

361

APs are generated by the atomisation (atomiser, TSI 3076) of a sodium fluorescein salt solution ($C_{20}H_{10}Na_2O_5$). This molecule has been selected for its significant fluorescent properties, detectable at very low concentrations (down to 10^{-10} g/l). Once atomised, the fine droplets go through a dry diffuser to produce dry APs. In Figure 7, two AP size distributions are presented for two different concentrations of the sodium fluorescein salt solution considered - 36 and 100 g/l - during the experiments. Those two size distributions have been evaluated using a Scanning Mobility Particle Sizer (SMPS). It was observed that the size distribution mode passes from 41 to 67.9 nm in radius when the concentration becomes three times larger. Since the geometric standard deviation (σ_g) is above 1.75, a Differential Mobility Analyser (DMA; TSI 3080) is set between the atomiser and the In-CASE's chamber to reduce the dispersion of the AP size distribution. After exiting the DMA, the AP flow goes through a low-energy X-ray neutraliser (< 9.5 keV, TSI 3088) so that the AP charge distribution entering the In-CASE's chamber is similar to a Boltzmann distribution. After the neutralisation, the dry AP flow is humidified by a pure water container in order to get high relative humidity in the collision chamber.

362

363

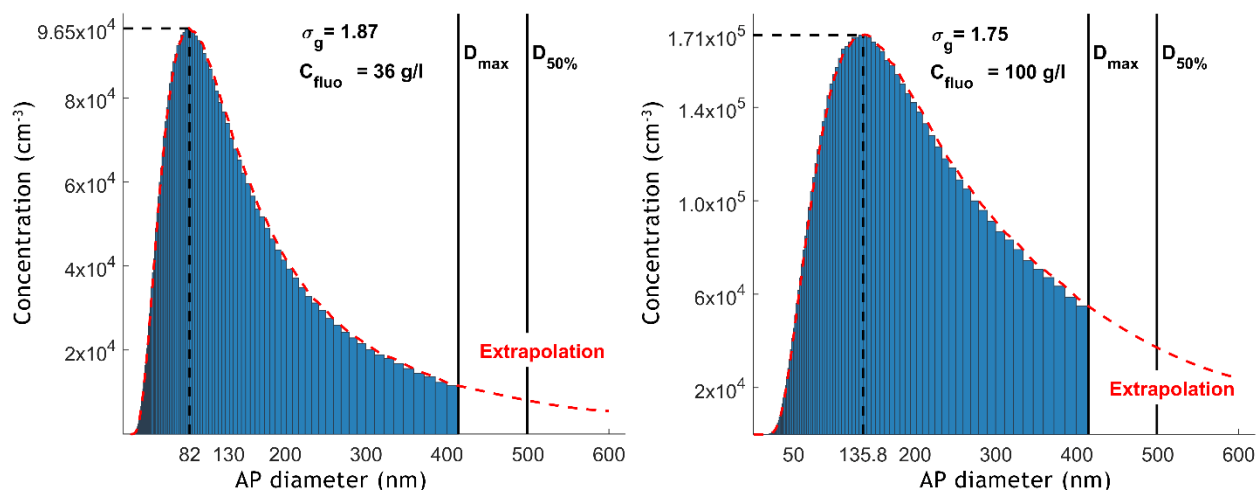
364

365

366

367

Note that, the DMA selects APs according to their electrical mobility - Z in equation (6) - assuming that only single charged APs can leave the DMA. Actually, depending on the AP size distribution and the AP flowrate in the DMA, larger AP radii carrying multiple charges than the one considered can also be selected. Sometimes those multiple charged APs cannot be neglected as discussed in section 2.2.



368
369

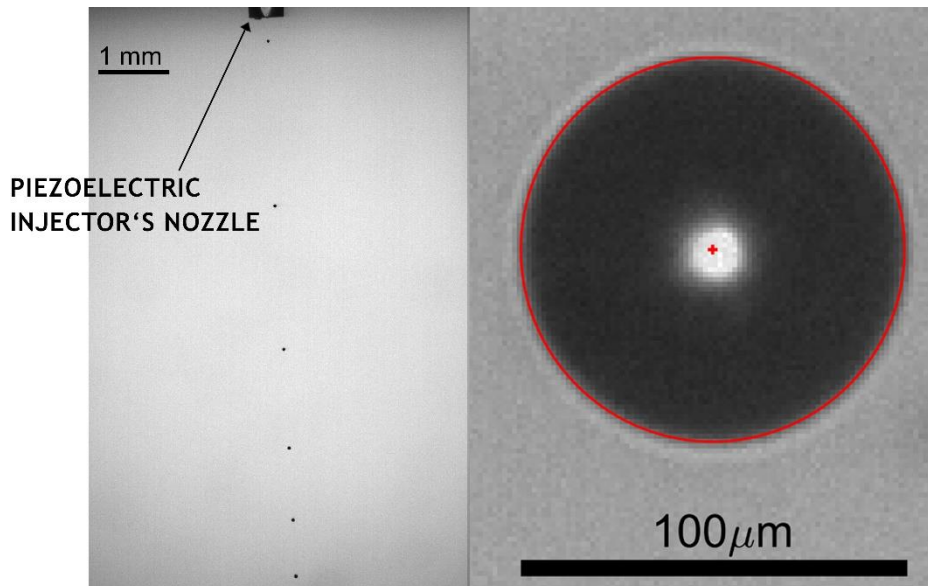
370 Figure 7 Two typical AP size distributions obtained with a SMPS at the atomiser's outlet. The
371 concentration of the sodium fluorescein salt solution is 36 g/l (Left) and 100 g/l (Right). D_{max} and
372 $D_{50\%}$ are respectively the maximum diameter selected by the DMA and the cut-off diameter of the
373 impactor at the DMA's inlet, at a given AP flowrate (0.6 l/min).
374

375 1.4 Droplet characterisation

376 377 1.4.1 Droplet generation frequency and size measurement

378
379 The droplet generator used for these experiments is a piezoelectric injector provided by Microfab -
380 the MJ-ABL-01 model with an internal diameter of 150 μm . This model has been used for its stability
381 over time, since the experiments can last up to 5 hours. This piezoelectric injector generates droplets
382 - at a given frequency - above their terminal velocity. The distance between two following droplets
383 reduces when droplets fall away from the injector's nozzle since the droplet velocity decreases (see
384 Figure 8, left). It was emphasised during *ex situ* experiments that droplet generation frequencies
385 greater than 25 Hz induce droplet coalescence since the inter-droplet space becomes too short to
386 prevent droplets from aerodynamically disturbing each other. This agrees with Ardon-Dryer et al.
387 (2015) who observed droplet coalescence for droplet generation frequency larger than 30 Hz
388 operating a similar piezoelectric injector. Thus, droplets were generated at 25 Hz in all experiments
389 presented in this current paper.

390 The droplet generator is placed at the top of the In-CASE's collision chamber, within an injection
391 head (see Figure 3). Few times during an experiment, droplet pictures are recorded by optical
392 shadowgraphy through two facing windows in the injection head (see Figure 3). A circle Hough
393 transform is then applied to evaluate the droplet radii in the recorded pictures. An example is given
394 in Figure 8 (right) for a 49.7 μm droplet radius. Note that the size distributions of the droplets
395 generated by the piezoelectric injector are considered monodispersed since the droplet size
396 dispersion is very low ($\sigma \sim 1\%$).
397
398
399



400 Figure 8 (Left) Droplet train at the piezoelectric injector's outlet obtained by optical shadowgraphy
 401 - the droplet generating frequency is 200 Hz. (Right) A droplet picture obtained by optical
 402 shadowgraphy - the droplet radius and centre are detected through a circle Hough transform (red
 403 cross and line).
 404

405 1.4.2 Droplet charge neutralisation

406 It is well-known that the piezoelectric droplet generator produces highly electrically charged droplets.
 407 With a similar device, Ardon-Dryer et al. (2015) measured up to 10^4 elementary charges on the
 408 generated droplets. Since this paper focused only on the relative humidity influence, the droplets,
 409 as well as APs, must be neutralised.

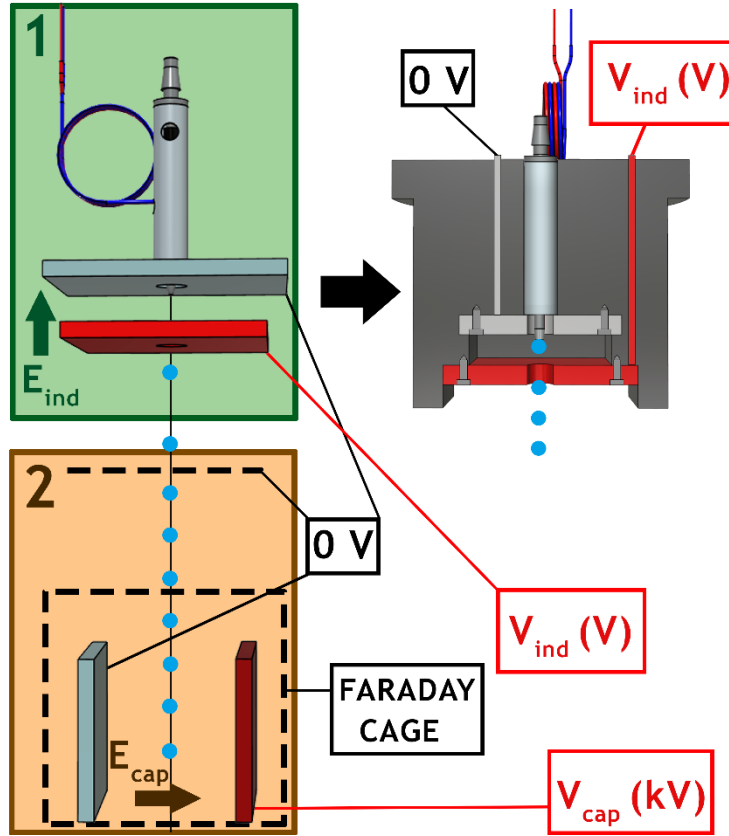
410 To do so, an electrostatic inductor was built following Reischl et al. (1977). Two parallel metal plate
 411 are placed at the droplet generator's nozzle - this is the electrostatic inductor shown in Figure 9
 412 (labelled 1, left). One plate is connected to a potential (V_{ind}) while the other is connected to the
 413 neutral potential - as presented in Figure 9 - in order to induce an electric field ($E_{ind} \sim 10^2 - 10^3$ V/m).
 414 Sodium chloride is added to the pure water that feeds the piezoelectric injector. According to the
 415 generated electric field polarity, the system can selectively attract negative or positive ions toward
 416 the nozzle where the droplet is formed. If V_{ind} is positive, the negative chloride ions (Cl^-) migrate
 417 toward the nozzle and the positive sodium ions (Na^+) are repulsed from the nozzle and inversely if
 418 the potential is negative. Following the electric field amplitude - through V_{ind} - the ion quantity can
 419 be set. This system can conclusively control the droplet charge. Note that the sodium chloride
 420 concentration has no effect on the principle of induction used here since the ion number is large
 421 enough for the entire experiment period (Reischl et al., 1977) - 3.3 g/l has been considered.

422 To evaluate the droplet charge and then, neutralise the droplets, an *ex situ* experiment has been
 423 conducted where the droplet train passed through a capacitor (labelled 2, Figure 9, left). One
 424 capacitor's plate is connected to the neutral whereas the other is connected to a high potential (V_{cap})
 425 - inducing an electric field ($E_{cap} \sim 10^5 - 10^6$ V/m). A Faraday cage surrounding the capacitor and a plate
 426 maintained at a neutral potential are set in order to prevent the electric field at the capacitor (E_{cap})
 427 from disturbing the electric field at the inductor (E_{ind}) which could change the droplet charge.
 428 Finally, the potential V_{ind} which electrically neutralises the droplet is found by selecting for the V_{ind}
 429 value which minimises the droplet train deflection.

430 Actually, this system can also be used to precisely evaluate the electric charges on the droplets (for
 431 both polarities), this method is applied and presented in **Dépée et al. 2020**.

432 Note that, the droplet charge induced by the piezoelectric injector has been calculated to
 433 -8,400 elementary charges - in line with Ardon-Dryer et al. (2015) using a similar generator. Moreover,
 434 after the droplet neutralisation, an uncertainty of 600 elementary charges was estimated.
 435

436



438
439

440 Figure 9 (Left) 1 - Electrostatic inductor set at the piezoelectric injector's nozzle to electrically
441 neutralise the droplets. 2 - Capacitor used to analyse the droplet deviation caused by the electric
442 field in the capacitor (E_{cap}). (Right) Housing made with a 3D printer containing the piezoelectric
443 injector and the electrostatic inductor, set in the injection head (see Figure 3).
444

445
446

2 DATA ANALYSIS

447

2.1 Definition of the collection efficiency

448

449 At the end of an experiment, the collection efficiency (CE) is calculated from the equation (2):
450

$$CE(a, A, HR) = \frac{m_{AP,d}}{m_{AP,available}} \quad (2)$$

451 Where the AP mass collected by all droplets ($m_{AP,d}$) is directly measured by spectrometry analysis in
452 the droplet impaction cup (see Figure 4) while the mass of available APs in the volume swept by the
453 droplets ($m_{AP,available}$) is given by the equation (3) :

$$m_{AP,available} = \pi(A + GroF(RH) \times a)^2 \times F_d \times \Delta t \times H_{eff} \times C_{m,AP} \quad (3)$$

454 F_d and Δt are respectively the droplet generation frequency and the experiment duration - the
455 product of those two quantities is the number of droplets injected during an experiment. $C_{m,AP}$ is the
456 mean AP mass concentration in the In-CASE collision chamber. Note that a is the AP dry radius
457 corrected by the growth factor ($GroF$) which depends on the relative humidity (see section 1.2.3.3).
458 H_{eff} is the effective height of interaction between droplets and APs. Since the APs are also falling in
459 the In-CASE collision chamber, this height is not the In-CASE collision chamber's height ($H_{In-CASE}$) but
460 is equal to the equation (4):

$$H_{eff} = \frac{U_{A,\infty}}{U_{A,\infty} + V_Q} H_{In-CASE} \quad (4)$$

461 However, as the droplet terminal velocity ($U_{A,\infty}$) is about 25 cm/s and the maximum AP flow velocity
 462 (V_Q) considered in the In-CASE collision chamber during the experiment is 5 mm/s (for an AP flowrate
 463 of 0.6 l/min), these two heights are thus considered equal ($H_{eff} \sim H_{In-CASE}$).

464
 465 In equation (3), the mean AP mass concentration in the In-CASE collision chamber is estimated from
 466 the fluorescence spectrometry analysis of the HEPA filter through the equation (5):

$$C_{m,AP} = \frac{m_{AP,tot}}{\Delta t \times Q_{In-CASE,c}} \quad (5)$$

467 $Q_{In-CASE,c}$ is the AP flowrate within the In-CASE collision chamber.
 468
 469
 470
 471

472 2.2 DMA selection - multiple charged AP's principle

473
 474 As previously stated, the AP flow travels through a DMA to select the particles according to their
 475 electrical mobility (Z) which is defined by the equation (6):

$$Z = \frac{n e C u}{6 \pi \eta_{air} a} \quad (6)$$

476 Where n , Cu , η_{air} are respectively the number of elementary charge (e), the Cunningham correction
 477 coefficient and the air dynamic viscosity (expressed here in poise).
 478

479 Thus, for an AP radius selected by the DMA, all particles with the same $\frac{n Cu}{a}$ ratio are actually
 480 selected. For example, when an AP with a radius of 50 nm is selected (single charged), the AP radii
 481 of 75.8 nm (double charged) and 98.2 nm (triple charged) will also be selected and progress into the
 482 In-CASE collision chamber since they have the same electrical mobility. In this paper, "multiple
 483 charged APs" are referred as the APs with the same electrical mobility as those with single charge
 484 selected by the DMA.
 485

486 At the DMA's inlet, an aerodynamic impactor is placed to prevent the heaviest APs from entering the
 487 DMA. Thus, for a given AP flowrate in the DMA, the multiple charged APs can be impacted at the
 488 DMA's inlet and can then be neglected at the DMA's outlet. To evaluate this case, the cut-off radius
 489 of the impactor at the DMA's inlet must be considered (referred as $D_{50\%}/2$). This radius is defined as
 490 the one where 50 % of the APs are impacted. The Table 1 shows this parameter for every AP flowrate
 491 used during the experiment and for a given selected AP radius. The double charged AP radius with
 492 the same electrical mobility as the selected AP radius (single charged) is also indicated - when this
 493 latter size is large enough compared to the cut-off radius, it is assumed that there is no contribution
 494 of the multiple charged APs in the CE measurement. This is the case for a selected AP radius of 200
 495 or 250 nm where the AP size distribution at the DMA's outlet can be considered purely monodispersed.
 496

497 However, for a selected AP radius of 50 or 150 nm, according to Table 1, the multiple charged AP
 498 radii cannot be neglected. Different experiments were run to perform a deconvolution of their
 499 respective contributions in the final CE calculation. This method is presented in Appendix A.
 500
 501
 502
 503
 504
 505
 506
 507
 508

Table 1 AP selection parameters

Selected dry AP radius by the DMA (single charged)	Double charged dry AP with the same electrical mobility	AP flowrate in the DMA	Cut-off radius of the impactor at the DMA's inlet ($D_{50\%}/2$)
50 nm	75.8 nm	0.6 l/min	213 nm
150 nm	253.7 nm	0.6 l/min	213 nm
200 nm	348.3 nm	0.6 l/min	213 nm
250 nm	444.3 nm	0.4 l/min	268.5 nm

510

511

512 **2.3 Uncertainty evaluations**

513

514 **2.3.1 AP radius uncertainty**

515

516 The first AP radius uncertainty is related to the AP selection by the DMA. Nevertheless, this
 517 uncertainty has been neglected since the spectral bandwidth of the DMA is quite small compared to
 518 the AP radius uncertainty addressed below.

519 Indeed, the only significant AP radius uncertainty results from the effective AP radius inside the In-
 520 CASE collision chamber due to the hygroscopicity of the APs. For the relative humidity levels studied
 521 (71.1, 82.4 or 93.5 %), the extreme relative humidity levels measured in all experiments are
 522 considered - for 71.1 %, the minimum and maximum values are 69.2 % and 73.4 %, respectively. As a
 523 reminder, the kappa-value is assumed from the Quérel et al. (2014) data and ranges from 0.2 to 0.27
 524 (see Figure 6). The low uncertainty for the AP radius is then evaluated by considering the minimum
 525 growth factor ($GroF$) in Figure 6 for the lower level of relative humidity measured and the lower
 526 kappa value determined - respectively 69.2 % and 0.2. Similarly, for the same example ($RH=71.1\%$),
 527 the high uncertainty for the AP radius is estimated by evaluating the maximum growth factor - for
 528 the maximum level of relative humidity observed and the maximum kappa value assumed -
 529 respectively 73.4 % and 0.27. In this example, for a dry AP radius of 50 nm selected by the DMA, its
 530 wet radius in the In-CASE collision chamber is likely to be 58 nm ($GroF=1.16$) ranging from 56.5 nm
 531 ($GroF=1.13$) to 60 nm ($GroF=1.20$) resulting from the respective low and high uncertainties.

532

533 **2.3.2 Uncertainty of the collection efficiency**

534

535 Since the method of CE evaluation differs in the presence of multiple charged APs, the uncertainty
 536 calculation is also different depending on the situations. The method is described in the Appendix B.

537

538 When there are no multiple charged APs in the AP flow, the CE is directly estimated through the
 539 equation (3) which can be rewritten as the equation (7):

$$CE(a, A, RH) = \frac{m_{AP,d}}{\pi(A + GroF(RH) \times a)^2 \times N_d \times H_{eff} \times C_{m,AP}} \approx \frac{m_{AP,d}}{\pi A^2 \times N_d \times H_{eff} \times C_{m,AP}} \quad (7)$$

540

541 Where N_d is the number of injected droplets during the experiment. The relative CE uncertainty (u_{CE})
 542 is then evaluated according to Lira (2003) and summarised by the equation (8):

543

$$u_{CE} = \sqrt{u_A^2 + u_{H_{eff}}^2 + u_{N_d}^2 + u_{m_{AP,d}}^2 + u_{C_{m,AP}}^2} \quad (8)$$

544

545 With:

- 546 • The relative uncertainty related to the droplet radius measurement (u_A) which is the ratio
 547 between the standard-deviation and the mean droplet radius on 200 pictures obtained by
 548 optical shadowgraphy. This relative uncertainty is about 1 %;

- 549 • The relative uncertainty of the effective height of interaction between droplets and APs
550 ($u_{H_{eff}}$) which is 4 %. Indeed, it has been evaluated that maximum 4 cm are required to assure
551 a good AP mixing at the injection in the collision chamber of 1 m height ($H_{In-CASE}$);
- 552 • The relative uncertainty of the number of droplets (u_{N_d}) which can be correlated to the
553 droplet number effectively impacted on the droplet impaction cup. This relative uncertainty
554 was evaluated during the validation of APs and droplet train separation (section 1.2.2.2) and
555 is about 2 %;
- 556 • The relative uncertainty of the detected AP mass in the droplet impaction cup ($u_{m_{AP,d}}$) which
557 takes into account the relative uncertainty related to the spectrometry analysis ($u_{fluorimeter}$)
558 and the one caused by the dilution ($u_{dilution}$) - equation (9). Indeed, at the end of an
559 experiment the water contained in the droplet impaction cup is dried and the residual AP
560 mass is then dissolved in 2 ml volume of ammonia water.

$$u_{m_{AP,d}} = \sqrt{u_{fluorimeter}^2 + u_{dilution}^2} \quad (9)$$

561 $u_{dilution}$ is estimated at 1 % meanwhile $u_{fluorimeter}$ is the main source of uncertainty. In fact,
562 when the mass of AP collected by the droplet is close to the detection limit of the fluorimeter
563 (about 10^{-15} kg in the droplet sample volume analysed), $u_{fluorimeter}$ is up to 30 %.

- 564 • The relative uncertainty of the mean AP mass concentration in the In-CASE collision chamber
565 ($u_{C_{m,AP}}$) which can be evaluated, according to the equation (5), by the equations (10):

$$\begin{cases} u_{C_{m,AP}} = \sqrt{u_{m_{AP,tot}}^2 + u_{Q_{In-CASE,c}}^2 + u_{\Delta t}^2} \approx \sqrt{u_{m_{AP,tot}}^2 + u_{Q_{In-CASE,c}}^2} \\ u_{m_{AP,tot}} = \sqrt{u_{fluorimeter}^2 + u_{dilution}^2} \end{cases} \quad (10)$$

566 Where the relative uncertainty of the detected AP mass on the HEPA filter ($u_{m_{AP,tot}}$) depends
567 on the one on the fluorimeter ($u_{fluorimeter}$) and the one on the dilution ($u_{dilution} \sim 1\%$). In fact,
568 the spectrometry analysis is performed by diluting the AP mass on the HEPA filter in a 100 ml
569 ammonia water solution at the end of an experiment. The relative uncertainty of the AP
570 flowrate in the In-CASE collision chamber ($u_{Q_{In-CASE,c}}$) is given by the datasheet of the
571 constructor - about 1 %. Note that the relative uncertainty on the experiment time ($u_{\Delta t}$) is
572 neglected since the error is approximately one second on an experiment that can last more
573 than 5 hours.

574
575

576 3 RESULTS AND DISCUSSIONS

577

578 3.1 Extension of the Dépée et al. (2019) model

579

580 In all experiments, the droplet charge is 0 ± 600 elementary charges with a radius of about 50 μm .
581 Since the AP charge distribution is similar to a Boltzmann distribution, an AP charge of more than 5
582 elementary charges is thus highly unlikely in the radius range considered in the experiments.
583 Moreover, Dépée et al. (2019) numerically evaluated the contribution of the electrostatic forces on
584 the CE for a droplet of 50 μm radius with -1000 elementary charges and 5 elementary charges on the
585 AP. For these extreme values, they calculate an increase of the CE due to the electrostatic forces by
586 42 % and 22 % for an AP radius of 50 nm and 300 nm, respectively. Close to these two AP radii, a rise
587 of the CE by a factor of 3 and 4, respectively, is observed when the relative humidity goes from 93.5
588 $\pm 0.9\%$ to 71.1 $\pm 1.3\%$ (Figure 10). Consequently, it is assumed that the contribution of the
589 thermophoresis and the diffusiphoresis is of first order in the measurements and the electrostatic
590 forces can be neglected in the observed increase of CE.

591

592 To extend the Dépée et al. (2019) model for the thermophoretic (F_{th}) and diffusiphoretic forces
593 (F_{df}), the resulting velocity at the AP location ($U_{f@p}^*$) given by the authors (in Equation 6) is replaced
594 by the equation (11):

$$U_{f@AP}^*(t) = U_{f@AP}(t) + \frac{\tau_{AP}}{m_{AP}}(F_{buoy} + F_{df} + F_{th}) \quad (11)$$

595 Where F_{buoy} is the buoyancy force, $U_{f@AP}$ is the fluid velocity at the AP location, τ_{AP} the AP relaxation
 596 time and m_{AP} the AP mass. The thermophoresis and the diffusiophoresis which are given by Brock
 597 (1962) and Waldmann and Schmitt (1966), respectively, summarised in the Equations (12):

$$\begin{cases} \mathbf{F}_{df} = -6\pi\eta_{air}a \frac{0,74D_v M_{air}}{C_u M_{water} \rho_{air}} \times \frac{\overset{\textcircled{1}}{(\rho_{v,air} - \rho_{v,s})f_v}}{Ar^{*2}} \mathbf{u}_r \\ \mathbf{F}_{th} = -\frac{12\pi\eta_{air}a}{5P_{air}} \frac{(k_{air} + 2,5k_{AP}K_n)k_{air}}{(1 + 3K_n)(2k_{air} + k_{AP} + 5k_{AP}K_n)} \times \frac{\underset{\textcircled{2}}{(T_{air} - T_{d,s})f_h}}{Ar^{*2}} \mathbf{u}_r \end{cases} \quad (12)$$

598 With \mathbf{u}_r - the unit vector in the radial direction from the droplet centre to the AP centre, r^* - the
 599 distance between the AP and droplet centres normalised by the droplet radius A , D_v - the diffusivity
 600 of vapor, K_n - the Knudsen number, M_{air} and M_{water} - the respective air and water molar masses, k_{air}
 601 and k_{AP} - the respective air and AP thermal conductivities. Note that the thermal conductivity of the
 602 sodium fluorescein salt is considered for k_{AP} - equal to $0.43 \text{ m.kg.s}^{-3}.\text{K}^{-1}$ (Al-Azzawi et Owen, 1984).
 603

604 In equations (12), the terms $\textcircled{1}$ and $\textcircled{2}$ represent the gradient of vapor density and the thermal
 605 gradient in the air, respectively. These two gradients are computed under the assumption that the
 606 temperature and vapor density profiles are spherically symmetric around the droplet (Wang et al.,
 607 1978). Because the droplet is falling in the air, f_v and f_h - which are the ventilation coefficient for
 608 the vapor and the heat respectively (Beard and Pruppacher, 1971) - correct the gradients since the
 609 profiles are actually disturbed by the airflow.
 610

611 3.2 Collection efficiency measurements and analysis

612 In Figure 10, the CEs are presented for the three levels of relative humidity studied - 71.1, 82.4 and
 613 93.5 % - and 6 dry AP radii ranging from 50 to 250 nm. The numerical values are presented in Table 3.
 614 As a reminder, all experiments were conducted with an air temperature of $0.58 \pm 0.50^\circ\text{C}$ at the
 615 atmospheric pressure, the AP charge distribution is similar to a Boltzmann distribution and the droplet
 616 charge is 0 ± 600 elementary charges. The droplet radius is $49.6 \pm 1.3 \mu\text{m}$. The key features of the
 617 experiments are summarised in Table 2. The measurements are compared to computed efficiencies
 618 using the models described in Wang et al. (1978) (dashed lines) as well as the extended version of
 619 Dépée et al. (2019) (solid lines). Note that the experimental conditions vary a little for the CE
 620 measurements at a given relative humidity level. For the modelling, air temperature and droplet
 621 radius are then the mean values of the three levels of relative humidity - $T_{air}=0.26, 0.27$ and 1.2°C -
 622 $A=49.3, 50.8$ and $48.8 \mu\text{m}$ - from the lowest to the highest, respectively. For $RH= 100 \%$, these
 623 parameters are those from table 2.
 624
 625
 626

627 Regarding the experimental results, it can be noted that the influence of the relative humidity via
 628 the thermophoresis and diffusiophoresis contribution on the CE is of first order. For the larger AP
 629 radii studied, the CE increases by a factor of 4 when the relative humidity passes from 93.5 to 71.1 %
 630 - filling up the Greenfield gap as the models predicts. A slight decline of the contribution of these
 631 two phoretic effects is observed when the AP radius decreases - the previous factor of 4 reducing to
 632 a factor of 3 for the smaller AP radii and for the same relative humidity range (from 93.5 to 71.1 %).
 633 Although this decrease is small, it is in line with the theory. Indeed, when the AP radius decreases
 634 the contribution of the Brownian motion on the CE increases and starts dominating over the
 635 thermophoretic and the diffusiophoretic forces. Consequently, the influence of the relative humidity
 636 on the CE is negligible for nanometric AP radii.

637 Moreover, the impact of the AP size is lower than the influence of the relative humidity for the
 638 experimental conditions considered. Indeed, between the larger and the smaller AP radii, the CE is
 639 only increased by a factor of 1.61, 1.59 and 2.03 for the respective relative humidity levels of 71.1,
 640 82.4 and 93.5 %. A decrease of the AP size effect on the CE is noticeable when the thermophoresis
 641 and the diffusiophoresis contributions intensify - in other words when the relative humidity declines.
 642 This observation is in line with the modelling of the CE when a threshold is more and more visible as
 643 the relative humidity decreases, for the submicron AP radii studied.

644 Finally, for the AP sizes and the droplet radius studied, both models describe relatively well the
 645 observed CE variations when changing relative humidity. For the two lowest levels of relative
 646 humidity (71.1 and 82.4 %), the CE modelling is really close between both models since the

647 thermophoresis and diffusiophoresis dominate the influence on the CE. Nevertheless, some significant
648 discrepancies appear for the highest relative humidity (93.5 %), where the Dépée et al. (2019)
649 extended model gives higher CE values. These differences result from the Wang et al. (1978) model
650 which does not consider dynamic effects such as AP inertia, AP weight and interception, in contrast
651 to the extended model of Dépée et al. (2019) which offers a complete description of the
652 microphysical effects involved in clouds.
653
654
655

Table 2 Key features of the In-CASE setup

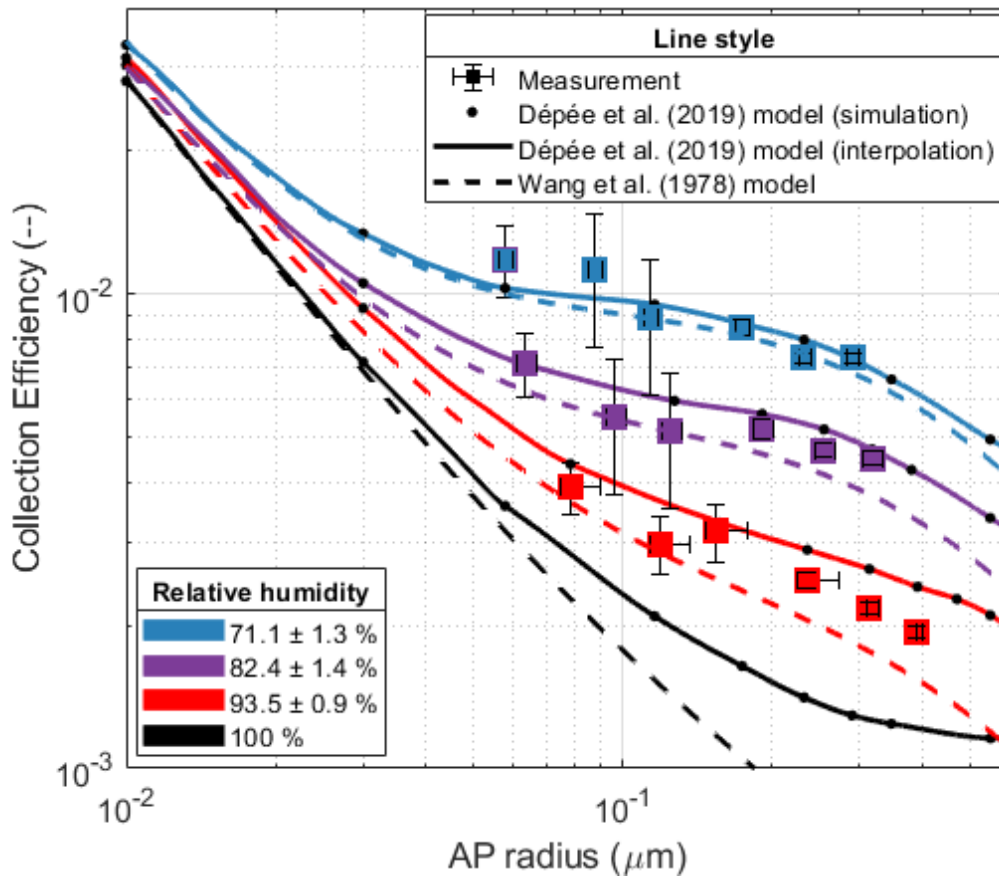
Feature	Numerical value
Collision chamber's parameters	
Height of the collision chamber ($H_{In-CASE}$)	1 m
Distance between droplet injection and AP injection	≈ 10 cm
Diameter of the collision chamber	5 cm
Impaction cup diameter	2.5 cm
AP flowrate in the collision chamber ($Q_{In-CASE,c}$)	0.4 or 0.6 l/min (following the selected AP radius)
Flow velocity in the collision chamber (V_Q)	3.4 or 5.1 mm/s (following the selected AP radius)
Flowrate of the upward Argon at the inlet of AP/droplet separator	0.4 l/min
Flowrate of the upward Argon in the impaction cup	1.4 cm/s
AP and Argon flowrate at the outlet of In-CASE chamber (toward the HEPA filter)	0.8 or 1 l/min (following the selected AP radius)
Air pressure in the collision chamber (P_{air})	1 atm
Temperature in the collision chamber (T_{air})	0.58 ± 0.50 °C
Relative humidity in the collision chamber (RH)	71.1 ± 1.3 , 82.4 ± 1.4 or 93.5 ± 0.9 %
Duration of experiments (Δt)	From 3 to 6 hours (related to the expected APs mass in droplets)
AP's parameters	
Selected dry AP radius during experiment (a)	50, 150, 200 or 250 nm
Dry AP radii considered for the CE evaluation (AP charge at the DMA's outlet)	50 (single charged), 75.8 (double charged), 98.2 (triple charged), 150 (single charged), 200 (single charged) and 250 nm (single or double charged)
Growth factor of the APs ($GroF$)	1.16, 1.27 or 1.57
Density of sodium florescein ($\rho_{fluorescein}$)	$1580 \text{ kg}\cdot\text{m}^{-3}$
Density of the wet APs (ρ_{AP})	1372, 1283 or $1150 \text{ kg}\cdot\text{m}^{-3}$
AP terminal velocity	$\leq 10^{-3}$ cm/s (equal to 8×10^{-4} cm/s for the larger selected dry AP radius 250 nm)
AP residence time in the collision chamber	≈ 200 or 300 s (following the selected AP radius)

Total AP concentration (single and multiple charged at the DMA's outlet)	From 5.10^4 cm^{-3} (for a selected dry AP radius of 50 nm) to 6.10^3 cm^{-3} (for a selected dry AP radius of 250 nm)
AP charge (q) distribution	Similar to Boltzmann distribution
Droplet's parameters	
Droplet radius (A)	$49.6 \pm 1.3 \text{ }\mu\text{m}$
Droplet generation frequency (F_d)	25 Hz
Droplet terminal velocity ($U_{A,\infty}$)	$\approx 25 \text{ cm/s}$
Number of injected droplets during experiments (N_d)	From 270,000 to 540,000 (related to the expected APs mass in droplets)
Observed distance between two successive droplets	$\approx 9 \text{ mm} \approx 180 \text{ droplet radii}$
Droplet residence time in the collision chamber	$\approx 4 \text{ s}$
Droplet charge before neutralisation (Q)	-8,400 elementary charges
Droplet charge after neutralisation (Q)	0 ± 600 elementary charges
Droplet evaporation between the injection and the end of the collision chamber	$\approx 3 \%$, $\approx 2 \%$ or $\approx 0.6 \%$ for the three levels of relative humidity considered
Sodium chloride concentration in the pure water	3.3 g/l

656
657
658
659
660
661

Table 3 CE measurements for the three levels of relative humidity (RH) and the wet AP radii (a_{wet}).
The droplet radius is $49.6 \pm 1.3 \text{ }\mu\text{m}$.

$RH = 93.5 \%$		$RH = 82.4 \%$		$RH = 71.1 \%$	
a_{wet} (nm)	CE (--)	a_{wet} (nm)	CE (--)	a_{wet} (nm)	CE (--)
79	3.92×10^{-3}	64	7.15×10^{-3}	58	1.18×10^{-2}
119	2.98×10^{-3}	96	5.52×10^{-3}	88	1.12×10^{-2}
154	3.17×10^{-3}	125	5.16×10^{-3}	114	8.94×10^{-3}
235	2.48×10^{-3}	191	5.20×10^{-3}	174	8.50×10^{-3}
314	2.18×10^{-3}	254	4.69×10^{-3}	232	7.31×10^{-3}
393	1.93×10^{-3}	318	4.51×10^{-3}	290	7.32×10^{-3}



662 Figure 10 CE measurements for three levels of relative humidity - 71.1, 82.4 and 93.5 % - compared
 663 to the extended model of Dépée et al. (2019) (solid lines) and the Wang et al. (1978) model (dashed
 664 line). Squares are the CE measurements summarised in Table 3. For the modelling, air temperature
 665 and droplet radius are then the mean values of the three levels of relative humidity - $T_{air}=0.26, 0.27$
 666 and 1.2°C - $A=49.3, 50.8$ and $48.8\ \mu\text{m}$ - from the lowest to the highest, respectively. For $RH=100\%$,
 667 the parameters are those from table 2.
 668

669 CONCLUSIONS

670 In-CASE (In-Cloud Aerosol Scavenging Experiment) was built to conduct a set of experiments
 671 quantifying the contribution of any microphysics effects involved in the AP collection by falling cloud
 672 droplets. For this purpose, all parameters influencing the collection efficiency (CE) are controlled -
 673 i.e. the AP and droplet sizes, the AP and droplet electric charges and the relative humidity.

674 This study focused on the influence of relative humidity since the literature lacks baseline data
 675 validating the theoretical models of CE implemented in cloud, climate and pollution models. Indeed,
 676 only the work of Ardon-Dryer et al. (2015) is dedicated to investigate the CE variation for two levels
 677 of relative humidity and cloud droplet sizes ($A \leq 100\ \mu\text{m}$). Nevertheless, for the droplet radius
 678 considered, the authors conclude that the electrostatic forces could have played a key role on their
 679 CE measurements, since the AP and droplet are charged, however slightly.

680 In the new measured CE dataset that is presented here, the APs and droplets are neutralised. There
 681 is no significant remaining electrostatic effect considering the maximum residual AP and droplet
 682 charges for the droplet radius examined ($A=49.6 \pm 1.3\ \mu\text{m}$), twice larger than the one studied by
 683 Ardon-Dryer et al. (2015). Here, three levels of relative humidity were investigated - 71.1, 82.4 and
 684 93.5 % which are typical in-cloud conditions.

685 From the measurements obtained, it is clear that the relative humidity - through the thermophoretic
 686 and diffusiophoretic forces - significantly impacts the CE. Indeed, an increase by a factor of 4 was
 687 observed for the CE when the relative humidity level declines from 93.5 to 71.1 %. Thus, it is quite
 688 important to consider these effects in cloud model since the levels of relative humidity are
 689 comparable from those used in this study. It was also shown that for the AP size considered in the
 690 present study, the impact of the AP size on the CE is a second order dependency. In fact, only a

691 doubling of the CE was highlighted - for a relative humidity of 93.5 % - from the larger to the smaller
692 AP radius considered. This impact of the AP size decreased when the influence of the relative
693 humidity increases.

694 The CE computed with the well-established model of Wang et al. (1978) as well as the new Lagrangian
695 model described in Dépée et al. (2019) and extended to phoretic effects were compared to the
696 measurements. The agreement was good. Nevertheless, significant discrepancies between both
697 models were revealed for high relative humidity (in subsaturated air) where the relative humidity
698 influence is weak. This can be attributed to the fact that the model of Wang et al. (1978) disregards
699 some microphysics effects such as AP weight, AP inertia and interception which have a significant
700 contribution near the Greenfield gap (Greenfield, 1957). Thus, the extended Lagrangian model of
701 Dépée et al. (2019) offers a more appropriate estimation of the CE.

702 In this study, the electrostatic effects were not considered. However, Dépée et al. (2019) have shown
703 an impact of several orders of magnitude on the CE, especially considering the electric charges of
704 cloud droplets and radioactive APs. Then, it is essential to investigate the AP collection by clouds
705 due to the electrostatic forces - referred as “electroscavenging”. Up to now, the analytical expression
706 of the electrostatic forces - based on the image charge theory developed by Jackson (1999) - has
707 never been experimentally validated or at least emphasised. Consequently, In-CASE was also used to
708 study the influence of the droplet and AP charge on CE which is addressed in a second paper (Dépée
709 et al., 2020).

710
711
712
713
714
715
716
717
718
719
720
721
722
723
724
725
726
727
728
729
730
731
732
733
734
735
736
737
738
739
740
741
742
743
744
745
746
747
748

749 **Appendix A - Evaluation method of the collection efficiency in the**
750 **presence of multiple charged APs**

751 This appendix presents the method used to evaluate the CE when the selected AP radius by the DMA
752 is 50 or 150 nm - when the multiple charged APs can not be neglected (see section 2.2).
753

754 **A.1 Ratio of multiple charged APs**

755 **A.1.1 Selected AP radius of 50 nm**

756 Before the AP selection, the DMA charges the APs following a known charging law (Wiedensohler,
757 1988) with an energy X-ray neutraliser (not presented in Figure 2).
758

759 The first step is to estimate the number and mass ratios of multiple charged APs in the mean AP mass
760 concentration measured in the In-CASE collision chamber ($C_{m,AP}$). For this purpose, the size
761 distribution of the APs produced by the atomiser is measured just before the DMA selection (Figure
762 7). The AP number concentration at the single (50 nm), double (75.8 nm), triple (98.2 nm), quadruple
763 (119.1 nm) and quintuple (139.1 nm) charged radii are deduced from the size distribution.
764

765 Those AP number concentrations are the total concentrations at a given multiple charged AP radius.
766 From those total concentrations, a fraction will be actually carrying the correct charge number to
767 have the exact electrical mobility selected by the DMA (1 charge for 50 nm, 2 charge for 75.8 nm, 3
768 charges for 98.2 nm, etc.). This fraction number ($F_{N,n}$) of an AP radius (a) carrying n elementary
769 charge(s) can be estimated through the APs charging law imposed by the energy X-ray neutraliser -
770 defined by Wiedensohler (1988). This similar Boltzmann distribution is defined in the equations (A.1):
771
772
773
774

$$\left\{ \begin{array}{l}
 F_{N,n}(a) = 10 \left[\sum_{i=1}^6 c_i(n) \left(\log \left(\frac{2a}{10^{-9}} \right) \right)^{i-1} \right] \quad \text{if } n < 3 \\
 F_{N,n}(a) = \frac{e}{\sqrt{8\pi^2 \epsilon_0 a k_b T_{air}}} \exp \left[\frac{n - \frac{4\pi\epsilon_0 a k_b T_{air} \ln \left(\frac{Z_{i+}}{Z_{i-}} \right)}{e^2}}{2 \frac{4\pi\epsilon_0 a k_b T_{air}}{e^2}} \right]^2 \quad \text{if } n \geq 3
 \end{array} \right. \quad \left\{ \begin{array}{l}
 c_{i \in [1,6]}(1) = \begin{bmatrix} -2,3484 \\ 0,6044 \\ 0,4800 \\ 0,0013 \\ -0,1553 \\ 0,0320 \end{bmatrix} \\
 c_{i \in [1,6]}(2) = \begin{bmatrix} -44,4756 \\ 79,3772 \\ -62,8900 \\ 26,4492 \\ -5,7480 \\ 0,5049 \end{bmatrix}
 \end{array} \right. \quad (A.1)$$

775 Where ϵ_0 , k_b and $T_{air} \approx 295.15$ K are the vacuum permittivity, the Boltzmann's constant and the lab
776 temperature. The ion mobility ratio $\left(\frac{Z_{i+}}{Z_{i-}} \right)$ is assumed to be equal to 0.875 (Wiedensohler, 1988).
777
778

779 Finally, the effective AP numbers for the respective multiple charged AP radii have been evaluated
780 in the AP flow at the DMA's outlet (corresponding to the AP flow going into the In-CASE collision
781 chamber). Thus, the mass fractions ($F_{m,n}$) for the single, double..., quintuple charged AP radii were
782 estimated. It was found that the quadruple and quintuple charged AP radii can be neglected since
783 their weight less than 6 % in the mean AP mass concentration in the In-CASE collision chamber ($C_{m,AP}$).
784 Moreover, since their number concentrations are really poor (less than 50 cm⁻³) compared to the
785
786
787

788 single, double and triple charged radius ($\sim 10^3\text{-}10^4 \text{ cm}^{-3}$), the likelihood of those APs to be collected
 789 by a droplet in the collision chamber is extremely unlikely.

790

791 A.1.2 Selected AP radius of 150 nm

792

793 For a selected AP radius of 150 nm, only the double charged APs are considered since the triple
 794 charged APs are assumed to be stopped by the impactor at the DMA's inlet (triple charged radius =
 795 353.4 nm and $D_{50\%}/2 = 213 \text{ nm}$, Table 1). The mass fractions ($F_{m,n}$) of the single and double charged
 796 are evaluated in the same way as a 50 nm selected AP radius.

797

798 A.2 Deduction of the collection efficiency

799

800 A.2.1 Selected AP radius of 50 nm

801

802 As explained in section 2.2, when the selected AP radius by the DMA is 50 nm, the AP mass collected
 803 at the In-CASE's chamber bottom ($m_{AP,d}$) is actually the sum of the masses of the single (50 nm),
 804 double (75.8 nm) and triple (98.2 nm) charged AP collected by the droplet train. This can also be
 805 defined as the linear combination of the collection efficiencies ($CE_i(a_i, A, RH)$) and the available AP
 806 mass in the volume swept by the droplets ($m_{AP,available}(a_i)$) at a given multiple charged dry AP radius
 807 (a_i) - equation (A.2):

$$m_{AP,d} = m_{50 \text{ nm},d} + m_{75.8 \text{ nm},d} + m_{98.2 \text{ nm},d} = \sum_{i=1}^3 CE_i(a_i, A, RH) \times m_{AP,available}(a_i) \quad (\text{A.2})$$

808

809 Where the respective available AP masses in the volume swept by the droplets are defined by the
 810 equation (A.3):

$$m_{AP,available}(a_i) = \pi(A + GroF(RH) \times a_i)^2 \times F_d \times \Delta t \times H_{eff} \times C_{m,AP} \times F_{m,n}(a_i) \quad (\text{A.3})$$

811

812 All the parameters given in equations (A.2) and (A.3) are either measured or initially known, except
 813 the collection efficiencies (CE_i) for the single, double and triple charged AP dry radius. To deduce
 814 those three unknown parameters, a set of j linearly independent experiments ($j \geq 3$) has been
 815 performed by varying the ratio of the multiple charged APs (by changing the AP size distribution mode
 816 in Figure 7). The matrix system is then described through the equation (A.4):

$$M_{collected \ mass} = M_{available} \otimes M_{CE} \quad (\text{A.4})$$

817

818 Where the one-dimension matrix of the collected mass ($M_{collected \ mass}$) for the set of j experiment is
 819 noted as the equation (A.5):

$$M_{collected \ mass} = \begin{bmatrix} m_{AP,d,1} \\ \vdots \\ m_{AP,d,j} \end{bmatrix} \quad (\text{A.5})$$

820

821 The two-dimension matrix of the available AP masses in the volume swept by the droplet ($M_{available}$)
 822 for the single (a_1), double (a_2) and triple (a_3) charged is defined as the equation (A.6):

$$M_{available} = \begin{bmatrix} m_{AP,available,1}(a_1) & m_{AP,available,1}(a_2) & m_{AP,available,1}(a_3) \\ \vdots & \vdots & \vdots \\ m_{AP,available,j}(a_1) & m_{AP,available,j}(a_2) & m_{AP,available,j}(a_3) \end{bmatrix} \quad (\text{A.6})$$

823

824 The one-dimension matrix containing all the unknown CEs (M_{CE}) is the equation (A.7):

$$M_{CE} = \begin{bmatrix} CE_1 \\ CE_2 \\ CE_3 \end{bmatrix} \quad (\text{A.7})$$

825

826 Finally, this matrix system (16) is numerically solved by the quasi-Newton method. The uniqueness
 827 of the solution was verified - the initial value was changed in the solving method, giving the same
 828 solution vector.

829 A.2.2 Selected AP radius of 150 nm

830

831 Like the same principle as before, the AP mass collected by the whole droplets ($m_{AP,d}$) is the linear
 832 combination of the single (150 nm) and double charged (253.7 nm), defined as the equation (A.8):

$$m_{AP,d} = m_{150\text{ nm},d} + m_{253.7\text{ nm},d} = \sum_{i=1}^2 CE_i(a_i, A, RH) \times m_{AP,available}(a_i) \quad (\text{A.8})$$

833

834 Nevertheless, to avoid additional experiments and numerically reverse a similar matrix system as
 835 (10), it was assumed that the CE of a dry AP radius of 253.7 nm is equivalent to the one for a dry AP
 836 radius of 250 nm. Then, the CE for a 150 nm dry AP radius is deduced by the equation (A.9):

$$CE_1(150\text{ nm}, A, RH) = \frac{m_{AP,d} - CE_2(253.7\text{ nm}, A, RH) \times m_{AP,available}(253.7\text{ nm})}{m_{AP,available}(150\text{ nm})} \quad (\text{A.9})$$

$$\approx \frac{m_{AP,d} - CE(250\text{ nm}, A, RH) \times m_{AP,available}(253.7\text{ nm})}{m_{AP,available}(150\text{ nm})}$$

837

838 The right term in equation (A.9) has no unknown since the CE of a 250 dry AP radius
 839 ($CE_2(250\text{ nm}, A, RH)$) has been previously calculated with the method developed in section 2.1.

840

841

842

843

844

845

846

847

848

849

850

851

852

853

854

855

856

857

858

859

860

861

862

863

864

865

866

867

868

869

870

871

872

873

874

875

876

877

878 **Appendix B - Uncertainty of the collection efficiency in the presence**
 879 **of multiple charged APs**

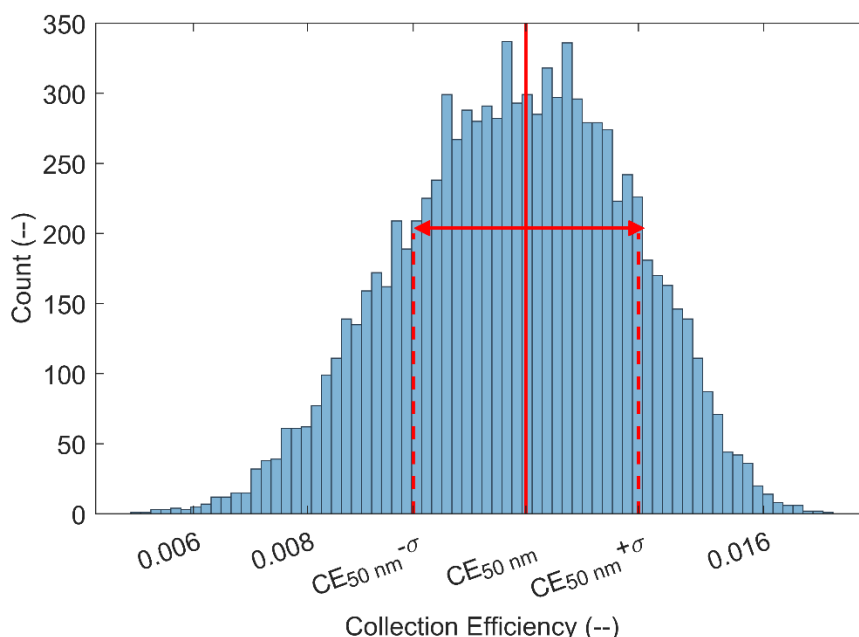
880 This appendix presents the method used to evaluate the CE uncertainty when the selected AP radius
 881 by the DMA is 50 or 150 nm - when the multiple charged APs can not be neglected (see section 2.2).
 882

883
 884 **B.1 With a selected dry AP radius of 150 nm**
 885

886 Since the CE of a selected dry AP radius of 150 nm ($CE(150\text{ nm}, A, RH)$) is calculated through the CE
 887 of a selected dry AP radius of 250 nm ($CE(250\text{ nm}, A, RH)$) - equation (A.9) - the uncertainty on the
 888 CE for the 150 nm ($u_{CE(150\text{ nm}, A, RH)}$) is evaluated by propagating the uncertainty on the CE for 250 nm
 889 ($u_{CE(250\text{ nm}, A, RH)}$). It means the term $u_{CE(250\text{ nm}, A, RH)}$ is added in equation (8) to deduce $u_{CE(150\text{ nm}, A, RH)}$.
 890

891 **B.2 With a selected dry AP radius of 50 nm**
 892

893 When the selected dry AP radius is 50 nm, the matrix system (16), solved by a quasi-Newton method,
 894 is composed of parameters each with their relative uncertainties. The relative CE uncertainties of
 895 the single (50 nm), double (75.8 nm) and triple (98.2 nm) charged dry AP radius are then deduced by
 896 randomly perturbing the terms of the matrix $M_{collected\ mass}$ and $M_{available}$ in equation (A.4) within the
 897 limits of their respective experimental relative uncertainties. 10,000 perturbed matrix systems were
 898 generated by the Monte-Carlo method and solved with the quasi-Newton method. From the 10,000
 899 solution vectors - shaped like the equation (A.5) - the ones with negative CEs were removed since
 900 they have no physical meaning. The Figure B.1 shows the set of the solutions for a relative humidity
 901 level of 71.1 % and a single charged dry AP radius (50 nm).
 902 Finally, the relative uncertainty of the CE is given by the standard deviation (σ) of the solution
 903 distribution.



904
 905 **Figure B.1 Distribution of 10,000 solutions (negative values were removed) for a relative humidity**
 906 **level of 71.1 % and a single charged dry AP radius (50 nm)**
 907

908
 909
 910
 911
 912
 913
 914

915 References

- 916 Adachi, K., Kajino, M., Zaizen, Y., & Igarashi, Y. (2013). Emission of spherical cesium-bearing particles
917 from an early stage of the Fukushima nuclear accident. *Scientific reports*, 3, 2554.
918
- 919 Al-Azzawi, H. K., & Owen, I. (1984). Measuring the thermal conductivity of uranin. *International journal*
920 *of heat and fluid flow*, 5(1), 57-59.
921
- 922 Ardon-Dryer, K., Huang, Y. W., & Cziczo, D. J. (2015). Laboratory studies of collection efficiency of sub-
923 micrometer aerosol particles by cloud droplets on a single-droplet basis. *Atmospheric Chemistry and*
924 *Physics*, 15(16), 9159-9171.
925
- 926 Beard, K. V., & Pruppacher, H. R. (1971). A wind tunnel investigation of the rate of evaporation of small
927 water drops falling at terminal velocity in air. *Journal of the atmospheric Sciences*, 28(8), 1455-1464.
928
- 929 Beard, K. V. (1976). Terminal velocity and shape of cloud and precipitation drops aloft. *Journal of the*
930 *Atmospheric Sciences*, 33(5), 851-864.
931
- 932 Bell, M. L., Davis, D. L., & Fletcher, T. (2004). A retrospective assessment of mortality from the London
933 smog episode of 1952: the role of influenza and pollution. *Environmental health perspectives*, 112(1),
934 6-8.
935
- 936 Brock, J. R. (1962). On the theory of thermal forces acting on aerosol particles. *Journal of Colloid*
937 *Science*, 17(8), 768-780.
938
- 939 Dépée, A., Lemaître, P., Gelain, T., Mathieu, A., Monier, M., & Flossmann, A. (2019). Theoretical study
940 of aerosol particle electroscavenging by clouds. *Journal of Aerosol Science*, 135, 1-20.
941
- 942 **Dépée, A., Lemaître, P., Gelain, T., Monier, M., & Flossmann, A. (2020). Laboratory study of the**
943 **collection efficiency of submicron aerosol particles by cloud droplets. Part II - Influence of**
944 **electric charges. Submitted to Atmospheric Chemistry and Physics,**
945 **<https://acp.copernicus.org/preprints/acp-2020-832/>**
946
- 947 Devell, L., Tovedal, H., Bergström, U., Appelgren, A., Chyssler, J., & Andersson, L. (1986). Initial
948 observations of fallout from the reactor accident at Chernobyl. *Nature*, 321(6067), 192-193.
949
- 950 Dockery, D. W., Schwartz, J., & Spengler, J. D. (1992). Air pollution and daily mortality: associations
951 with particulates and acid aerosols. *Environmental research*, 59(2), 362-373.
952
- 953 Flossmann, A. I., Hall, W. D., & Pruppacher, H. R. (1985). A theoretical study of the wet removal of
954 atmospheric pollutants. Part I: The redistribution of aerosol particles captured through nucleation and
955 impaction scavenging by growing cloud drops. *Journal of the Atmospheric Sciences*, 42(6), 583-606.
956
- 957 Flossmann, A. I. (1998). Interaction of aerosol particles and clouds. *Journal of the atmospheric*
958 *sciences*, 55(5), 879-887.
959
- 960 Greenfield, S. M. (1957). Rain scavenging of radioactive particulate matter from the
961 atmosphere. *Journal of Meteorology*, 14(2), 115-125.
962
- 963 Jackson, J. D. (1999). *Classical electrodynamics*.
964
- 965 Jaenicke, R. (1993). Tropospheric aerosols. In *International Geophysics* (Vol. 54, pp. 1-31). Academic
966 Press.
967
- 968 Jost, D. T., Gäggeler, H. W., Baltensperger, U., Zinder, B., & Haller, P. (1986). Chernobyl fallout in size-
969 fractionated aerosol. *Nature*, 324(6092), 22-23.
970
- 971 Kaneyasu, N., Ohashi, H., Suzuki, F., Okuda, T., & Ikemori, F. (2012). Sulfate aerosol as a potential
972 transport medium of radiocesium from the Fukushima nuclear accident. *Environmental science &*
973 *technology*, 46(11), 5720-5726.

974
975 Ladino, L., Stetzer, O., Hattendorf, B., Günther, D., Croft, B., & Lohmann, U. (2011). Experimental study
976 of collection efficiencies between submicron aerosols and cloud droplets. *Journal of the Atmospheric*
977 *Sciences*, 68(9), 1853-1864.
978
979 Laguionie, P., Roupsard, P., Maro, D., Solier, L., Rozet, M., Hébert, D., & Connan, O. (2014).
980 Simultaneous quantification of the contributions of dry, washout and rainout deposition to the total
981 deposition of particle-bound ⁷Be and ²¹⁰Pb on an urban catchment area on a monthly scale. *Journal*
982 *of Aerosol Science*, 77, 67-84.
983
984 Lira, I. (2003). Evaluating the measurement uncertainty: fundamentals and practical guidance.
985
986 Petters, M. D., & Kreidenweis, S. M. (2007). A single parameter representation of hygroscopic growth
987 and cloud condensation nucleus activity. *Atmospheric Chemistry and Physics*, 7(8), 1961-1971.
988
989 Pöllänen, R., Valkama, I., & Toivonen, H. (1997). Transport of radioactive particles from the Chernobyl
990 accident. *Atmospheric Environment*, 31(21), 3575-3590.
991
992 Pruppacher, H. R., & Klett, J. D. (1997). *Microphysics of Clouds and Precipitation*.
993 (Dordrecht/Boston/London).
994
995 Quérel, A., Lemaitre, P., Monier, M., Porcheron, E., Flossmann, A. I., & Hervo, M. (2014). An experiment
996 to measure raindrop collection efficiencies: influence of rear capture. *Atmospheric Measurement*
997 *Techniques*, 7(5), 1321-1330.
998
999 Reischl, G. P. W. W., John, W., & Devor, W. (1977). Uniform electrical charging of monodisperse
1000 aerosols. *Journal of Aerosol Science*, 8(1), 55-65.
1001
1002 Santachiara, G., Prodi, F., & Belosi, F. (2012). A review of thermo- and diffusio-phoresis in the
1003 atmospheric aerosol scavenging process. Part 1: Drop scavenging. *Atmospheric and Climate*
1004 *Sciences*, 2(02), 148.
1005
1006 Sharp, D. H. (1983). *Overview of Rayleigh-taylor instability* (No. LA-UR-83-2130; CONF-8305110-2).
1007 Los Alamos National Lab., NM (USA).
1008
1009 Tao, W. K., Chen, J. P., Li, Z., Wang, C., & Zhang, C. (2012). Impact of aerosols on convective clouds
1010 and precipitation. *Reviews of Geophysics*, 50(2).
1011
1012 Tinsley, B. A., Zhou, L., & Plemmons, A. (2006). Changes in scavenging of particles by droplets due to
1013 weak electrification in clouds. *Atmospheric Research*, 79(3-4), 266-295.
1014
1015 Tinsley, B. A., & Zhou, L. (2015). Parameterization of aerosol scavenging due to atmospheric
1016 ionization. *Journal of Geophysical Research: Atmospheres*, 120(16), 8389-8410.
1017
1018 Twomey, S. (1974). Pollution and the planetary albedo. *Atmospheric Environment* (1967), 8(12), 1251-
1019 1256.
1020
1021 Waldmann, L., & Schmitt, K. H. (1966). Thermo-phoresis and diffusio-phoresis of aerosols. *Aerosol*
1022 *Science*, 137-162.
1023
1024 Wang, P. K., Grover, S. N., & Pruppacher, H. R. (1978). On the effect of electric charges on the
1025 scavenging of aerosol particles by clouds and small raindrops. *Journal of the Atmospheric*
1026 *Sciences*, 35(9), 1735-1743.
1027
1028 Wiedensohler, A. (1988). An approximation of the bipolar charge distribution for particles in the
1029 submicron size range. *Journal of Aerosol Science*, 19(3), 387-389.
1030
1031
1032