- 1 A comparison of PM_{2.5}-bound polycyclic aromatic hydrocarbons in summer
- 2

Beijing (China) and Delhi (India)

Atallah. Elzein¹, Gareth J. Stewart¹, Stefan J. Swift¹, Beth S. Nelson¹, Leigh R. Crilley^{2,6},
 Mohammed S. Alam², Ernesto. Reyes-Villegas³, Ranu. Gadi⁴, Roy M. Harrison^{2,7}, Jacqueline

5 F. Hamilton¹, Alastair C. Lewis⁵

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- ⁴ Indira Gandhi Delhi Technical University for Women, New Delhi, 110006, India.
- ⁵National Centre for Atmospheric Science, University of York, York, YO10 5DD, United Kingdom.
- ⁶ Currently at: Department of Chemistry, York University, Toronto, ON, Canada.
- ⁷ Also at: Department of Environmental Sciences / Center of Excellence in Environmental Studies, King
 Abdulaziz University, PO Box 80203, Jeddah, 21589, Saudi Arabia.
- 18 Abdulaziz University, PO Box 80203, Jeddah, 21589, Saudi Arabia.19
- 20

Correspondence to: Atallah. Elzein (atallah.elzein@york.ac.uk)

23 Abstract.

24 Polycyclic aromatic hydrocarbons (PAHs) are ubiquitous pollutants in air, soil and water and known to 25 have harmful effects on human health and the environment. The diurnal and nocturnal variation of 17 26 PAHs in ambient particle-bound PAHs were measured in urban Beijing (China) and Delhi (India) during the summer season using GC-Q-TOF-MS. The mean concentration of particles less than 2.5 27 28 microns ($PM_{2,5}$) observed in Delhi was 3.6 times higher than in Beijing during the measurement period in both the daytime and night-time. In Beijing, the mean concentration of the sum of the 17 PAHs ($\sum 17$ 29 30 PAHs) was 8.2 ± 5.1 ng m⁻³ in daytime, with the highest contribution from Indeno[1,2,3-cd]pyrene (12) %), while at night-time the total PAHs was 7.2 \pm 2.0 ng m⁻³, with the largest contribution from 31 Benzo[b]fluoranthene (14 %). In Delhi, the mean $\sum 17$ PAHs was 13.6 ± 5.9 ng m⁻³ in daytime, and 22.7 32 \pm 9.4 ng m⁻³ at night-time, with the largest contribution from Indeno[1,2,3-cd]pyrene in both the day 33 (17%) and night (20%). Elevated mean concentrations of total PAHs in Delhi observed at night were 34 attributed to emissions from vehicles and biomass burning and to meteorological conditions leading to 35 their accumulation from a stable and low atmospheric boundary layer. Local emission sources were 36 37 typically identified as the major contributors to total measured PAHs in both cities. Major emission 38 sources were characterised based on the contribution from each class of PAHs, with the 4, 5, and 6 ring 39 PAHs accounting ~ 95 % of the total PM_{2.5}-bound PAHs mass in both locations. The high contribution 40 of 5 ring PAHs to total PAH concentration in summer Beijing and Delhi suggests a high contribution 41 from petroleum combustion. In Delhi, a high contribution from 6 ring PAHs was observed at night, suggesting a potential emission source from the combustion of fuel and oil in power generators, widely 42 used in Delhi. The lifetime excess lung cancer risk (LECR) was calculated for Beijing and Delhi, with 43 44 the highest estimated risk attributed to Delhi (LECR = 155 per million people), 2.2 times higher than Beijing risk assessment value (LECR = 70 per million people). Finally, we have assessed the emission 45 control policies in each city and identified those major sectors that could be subject to mitigation 46 47 measures.

 ¹Wolfson Atmospheric Chemistry Laboratories, Department of Chemistry, University of York, York, YO10 5DD,
 United Kingdom.

 ⁹ ²Division of Environmental Health & Risk Management, School of Geography, Earth & Environmental Sciences,
 ¹⁰ University of Birmingham, Birmingham, B15 2TT, United Kingdom.

 ³Department of Earth and Environmental Science, The University of Manchester, M13 9PL, United
 Kingdom.

49 1 Introduction

50 The significant increase of particulate matter (PM) and gaseous pollutants over the past decades in some 51 rapidly expanding economies, has led to greater emphasis being placed on mitigation of emissions and management air quality health effects. To support such measures requires insight in both the sources of 52 pollution, and the composition of pollution so that most harmful sources may be tackled as a priority. 53 54 Although there have been recent improvements that have reduced primary particle concentrations in 55 some regions, concentrations of many damaging gases and fine particles continue to exceed WHO 56 guidelines (WHO, 2016), in megacities such as Beijing (Elzein et al., 2019; Lin et al., 2018; Gao and 57 Ji., 2018) and Delhi (Kanawade et al., 2019; Sharma et al., 2007) and in many other cities around the 58 world such as Cairo, Egypt (Cheng et al., 2016) and Islamabad, Pakistan (Mehmood et al., 2020). Growing populations, human activities, energy consumption and natural contributions (volcanic 59 eruptions and forest fires) are an important contributor to particles emissions. PM monitoring and 60 analysis become ever more important because of its adverse effect on human health. The chemical 61 62 composition of airborne particles influences the health impacts, particularly the abundance of primary and secondary organic matter, metals, and ions (WHO, 2016; Bond et al., 2004; Saikawa et al., 2009). 63 Fine particles become more harmful as particle size decreases (ultrafine), they enter the human body 64 65 through the lungs and may translocate to other organs causing respiratory diseases and cancer 66 (Schraufnagel., 2020). The greatest adverse effects on human health in epidemiological studies are currently associated with the mass of particles less than 2.5 microns in diameter (PM2.5) (Raaschou-67 Nielsen et al., 2013, Pun et al., 2017, Hamra et al., 2014). The organic component of PM_{2.5} consists of 68 thousands of compounds, among them polycyclic aromatic hydrocarbons (PAHs), a particular class of 69 70 species with high toxic potency. They are released into the atmosphere from both natural and 71 anthropogenic sources. PAHs are considered ubiquitous in the environment and can be found in soil 72 and water via dry or wet atmospheric deposition (Menzie et al., 1992, Meador et al., 1995). Their major 73 emissions come from anthropogenic sources and include incomplete combustion of fossil fuels, vehicle 74 exhaust emissions, cigarette emissions, agricultural burning and industrial activities (Saikawa et al., 75 2009). It has been shown that PAHs can react with atmospheric oxidants leading to the formation of 76 secondary species with direct-acting mutagenicity and carcinogenicity and thus they can be significant 77 contributors to the high toxicity of particles even at low PM levels (Nisbet and LaGoy 1992).

Beijing and Delhi often suffer from severe air pollution episodes, reaching high PM_{2.5} concentrations
and air quality index levels. The local government in Beijing has declared many different air quality
actions since September 2013, resulting in a decrease in the concentration of total PAHs as reported in
recent studies for the winter season in Beijing (Chen et al., 2017, Elzein et al., 2019, Feng et al., 2019).
This has been attributed to the efforts made by the municipal government of Beijing to improve air
quality and control emissions by reducing combustion sources and promoting the use of clean energy
sources and electric vehicles.

Several anti-pollution measures have been introduced in Delhi in the last two decades such as, Bharat
stage (equivalent to Euro standards), switching public transport from running on diesel to compressed
natural gas (CNG), and applying "odd-even" vehicle number plate restriction during working days
(Guttikunda et al., 2014; Goel and Guttikunda, 2015; Chowdhury et al., 2017). Despite the government
effort to tackle air pollution in India and especially in Delhi, recent studies have showed that the air
quality continue to be among the poorest in the world causing thousands of premature deaths (Tiwari
et al., 2015, Ghude et al., 2016, Chowdhury and Dey., 2016, Pant et al., 2017, Conibear et al., 2018).

92 PAHs emission sources in Delhi have previously been attributed to vehicle emissions, coal combustion,

- wood and burning leaves (Gadi et al., 2019; Shivani et al., 2019; Gupta et al., 2011; Sharma et al 2007).
- 94 To the best of our knowledge, data on $PM_{2.5}$ -bound PAHs in Delhi during the summer season (pre-
- 95 monsoon: March June) is scarce and limited to other periods of the year with low-time resolution

- 97 particle-bound PAH when collected over long sampling times are subject to higher uncertainties related
- 98 to sampling artefacts deriving from meteorological effects and oxidant concentrations such as ozone
- 99 (Goriaux et al., 2006; Tsapakis and Stephanou, 2003, 2007; Ringuet et al., 2012a; Keyte et al., 2013).
- Using shorter time periods for ambient particle sampling (e.g. 3 and 4 h) has been suggested as offering
 more accurate diagnosis of emission sources (Tian et al., 2017; Srivastava et al., 2018), shorter time
- sampling is still scarce and limited to studies outside China and India (Reisen and Arey, 2004;
- 103 Srivastava et al., 2018). Considering the above and that particles collected during 24-hour sampling
- time integrate both daytime and night-time chemistry together, we collected high frequency ambient air
- particle samples ($PM_{2.5}$) in urban Beijing (China) and Delhi (India) to determine the temporal diurnal
- and nocturnal variation of PAHs. A great advantage in this study is that all particle samples from both
- campaigns were collected, extracted and analysed using the same analytical method, which provide a
 better comparison of the variation in PAHs between cities and on the feasibility and efficiency of
- better comparison of the variation in PAHs between cities and on the feasibimplementing emission control policies to improve air quality in both cities.
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111 2 Methods

112 **2.1 Sampling campaigns**

Both measurement campaigns were part of the UK NERC / MRC Air Pollution and Human Health 113 (APHH) research programme. The sampling site in Beijing was located at the Institute of Atmospheric 114 Physics, Chinese Academy of Sciences in Beijing (39°58'28" N, 116°22'15" E) and the sampling site 115 in Delhi was located at Indira Gandhi Delhi Technical University for Women (28°39'52.6" N, 116 77°13'54.1" E). In both campaigns, the sampling equipment was installed on the roof of a 2-storey 117 building about 8 m above ground level. Prior to sampling the quartz microfiber filters (Whatman QM-118 119 A, 20.3×25.4 cm, supplied by VWR U.K.) were baked at 550 °C for 5 h in order to eliminate any organic matter, PM_{2.5} filter samples were collected every 3 hours during daytime and over 15 h at night-120 time, using a High-Volume Air Sampler (Ecotech HiVol 3000, Victoria, Australia) operating at 1.33 121 $m^3 min^{-1}$. The daytime sampling started at 8:30 in the morning and the filter was changed every 3 h. 122 Night-time sampling began at ~17:30 and ended at 08:30 the following day. Filters were collected for 123 20 days (22 May 2017 to 10 June 2017) totalling 80 filters during Beijing campaign, and for 9 days (28 124 May 2018 to 5 June 2018) totalling 35 filters during Delhi campaign. After sampling, filters were 125 126 wrapped in aluminium foil, sealed in polyethylene bags and stored at -20 °C until extraction and 127 analysis.

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129 2.2 Sample extraction

130 Collected filters were cut using a cube cutter (1/16 of the filter) measuring a surface area equivalent to 24 cm². Each section was then cut into small pieces to fit inside 5 mL stainless steel extraction cells 131 used by a pressurized solvent extractor (Dionex, ASE 350). All samples were extracted in acetonitrile 132 (HPLC-grade) using the following method: Oven at 120°C, pressure at 1500 psi, rinse volume 60 % 133 134 and 60 s purge time for three consecutive 5 min cycles. The extraction time of each cell was about 25 135 min for a final volume of 20 mL. Prior to purification, extracts (V = 20 mL) were evaporated to 136 approximately 6 mL under a gentle stream of nitrogen. All samples and blanks were purified on solid 137 phase extraction (SPE) silica normal phase cartridge (1g/6mL; Sigma Aldrich) to reduce the impacts of 138 interfering compounds in the matrix and to help maintain a clean GC injection inlet liner. After the purification step, the solution of each sample was evaporated to 1 mL under a gentle stream of nitrogen 139 at room temperature (20 °C) and transferred to a 1.5 mL autosampler amber vial. Each concentrated 140 141 sample was stored at 4 °C until analysis.

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145 2.3 Analytical procedures

- In this study, 17 PAHs were selected based on their presence within the particle phase and commercially 146 147 available standards. These are listed in Table 1 and standards purchased from Sigma Aldrich, Alfa Aesar and Santa Cruz Biotechnology in the UK with a minimum purity of 97 %. In parallel to individual 148 standards, a mixed solution of the 16 EPA PAHs (CRM47940, Supelco, Sigma Aldrich) of 10 µg ml⁻¹ 149 in acetonitrile was also used. Standard solutions for calibrations were prepared in acetonitrile (HPLC 150 151 grade, 99.9 % purity, Sigma Aldrich). Phenanthrene-d10 and pyrene-d10 were used as surrogate 152 standards and were spiked over two blank filters and two sample filters from both campaigns, with concentration on filters corresponding to 300 ng (V = 60 μ L from 5 ng μ L⁻¹ in acetonitrile). Spiked 153 filters solutions were analysed 10 times, and the average recovery efficiencies calculated from surrogate 154 standards was 96 %, ranging from 88 % to 107 % for both compounds, phenanthrene-d10 and pyrene-155 d10. PAH concentrations were corrected to the average recovery efficiencies. These two deuterated 156 compounds were supplied by C/D/N isotopes and distributed by QMX Laboratories Ltd (Essex, UK). 157
- 158 All PAH were quantified using a gas chromatography - time of flight - mass spectrometry system (GC Agilent 7890B coupled to an Agilent 7200 Q-TOF-MS). 1 µL of each sample was injected in pulsed 159 splitless mode at 320 °C using an automated liquid injection with the GERSTEL MultiPurpose Sampler 160 161 (MPS). Helium was used as carrier gas at 1.4 mL min⁻¹ and target compounds were eluted using the RXi-5ms (Restek GC column, Crossbond diphenyl dimethyl polysiloxane; length: 30 m, diameter: 0.25 162 mm, film thickness: 0.25 µm). The analysis time of each sample was set to 35 min using the following 163 GC oven temperature programme: 65 °C for 4 min as a starting point and then increased to 185 °C at a 164 heating rate of 40 °C min⁻¹ and held for 0.5 min, followed by a heating rate of 10 °C min⁻¹ to 240 °C 165 and then ramped at 5 °C min⁻¹ until 320 °C and held isothermally for further 6 min to ensure all analytes 166 167 eluted from the column. The MS was operated in Electron Ionisation (EI) mode at 70 eV with an emission current of 35 µA. 10-point calibration solutions were injected 4 times in the same sequence as 168 for samples and covered the range from 1 pg μ L⁻¹ to 1000 pg μ L⁻¹, with a correlation coefficient from 169 170 the linear regression between 0.970 and 0.999.
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172 **2.4 Error evaluation**

- As part of our method validation and in addition to recovery efficiency corrections, we have evaluated other possible factors that can affect our final true result. In this study, the solvent (acetonitrile) and field blanks (n = 3) were analysed following the same procedure as for the samples (Extraction, SPE, Evaporation) to determine any source of contamination during sample preparation and the analytical procedure. Whilst most target compounds were found to be below our limit of detection (S/N=3), three PAHs (Fluorene, Phenanthrene, Fluoranthene, and Pyrene) were quantified in field blanks and their contributions to the final data have been corrected.
- To evaluate the agreement between repeated measurements (precision error), we calculated the relative 180 181 standard deviations (%RSD) from replicate analysis (n = 10) of two samples. The %RSD average of total PAHs was 8.7 % (range: 3.36 – 13.71 %) for Beijing samples and 4.2 % (range: 2.64 – 10.12 %) 182 for Delhi samples. The %RSD average for deuterium labelled compounds spiked over two sample filters 183 was ~ 3.6 % for both campaigns. The %RSD of each compound is shown in Table S1. Moreover, the 184 calibration offset and the influence of the sample matrix on the quantification step are an important 185 186 source of systematic error and was estimated to be a maximum of 20 %. Therefore, the upper limit estimated error, combining the precision and the systematic effects, is 30 % for Beijing samples and 25 187 188 % for Delhi samples.
- 189 In addition, another type of error has been attributed to sampling artefacts. Previous studies (Schauer et
- al., 2003, Goriaux et al., 2006, Tsapakis and Stephanou, 2003, Brown and Brown, 2012) have reported
- 191 a chemical decomposition of PAHs depending on the ambient concentration of ozone and sampling
- time. Therefore, data from long sampling times and under very high ozone ambient concentrations (>70

193 ppb) may be biased by sampling artefacts of more than 100 % (Schauer et al., 2003, Goriaux et al., 2006). However, at low ozone levels (< 30 ppb), negative artefacts were considered not significant 194 195 (Tsapakis and Stephanou, 2003), whilst, at medium ozone levels (30-50 ppb) PAHs values were underestimated by 30 % (Schauer et al., 2003). Tsapakis and Stephanou (2003), have reported a loss of 196 PAHs by 28 % due to ozone atmospheric concentration about 60 ppb and long sampling times of 24-h. 197 They have also suggested that long sampling time under low ozone concentration (< 30 ppb) do not 198 199 affect the concentration of collected PAHs in the gas or particle phases, while short sampling time (2 200 h) under ozone concentration about 60 ppb will reduce the concentration of PAHs by 17 %. In this 201 study, ozone concentrations were measured in both campaigns and averaged to the filter sampling time 202 to provide more accurate estimation on the negative sampling artefacts. The ozone concentration in summer Beijing ranged between 3.7 and 140 ppb over the campaign (mean value: 56 ± 31 ppb), 203 approximately 5 times higher than that in winter (mean value: 10.4 ± 8.8 ppb, Elzein et al., 2019). 204 205 Daytime ozone concentration ranged from 12 to 140 ppb (mean value: 63 ± 30 ppb), while over the 206 night-time it ranged from 4 to 74 ppb (mean value: 34 ± 18 ppb). Therefore, based on Tsapakis and Stephanou (2003) study, the negative sampling artefacts due to ozone concentration was estimated to 207 be 20 % for daytime samples (3 h) and 10 % for the night-time samples (15 h) due to lower ozone 208 209 concentration at night. Using the same approach, daytime ozone concentration in Delhi ranged between 39 and 119 ppb, with a mean value: 75 ± 20 ppb, while the night-time concentration ranged from 14 to 210 50 ppb (mean value: 37 ± 12 ppb). Therefore, the estimation of the negative sampling artefacts on the 211 data from Delhi ranged between 15 and 30 %, with the highest error estimation attributable to daytime-212 213 time samples (3 h) because of the higher ozone concentration during the day.

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215 **3 Results and discussion**

216 **3.1** Concentration levels of PM_{2.5} and Benzo[a]pyrene in summer Beijing and Delhi

217 In 2016, the World Health Organisation (WHO, 2016) published an air quality guideline on outdoor air 218 pollution limits to help protect human health and reduce the risk of mortality due to fine particles. The air quality standards for PM2.5 were set by the ministry of environment in China and India, and published 219 in the WHO air quality guideline as an annual and 24-h mean concentration. The daily PM_{2.5} (mean 24-220 h) guideline concentration is currently set at 75 and 60 μ g m⁻³ for China and India, respectively, while 221 the annual mean guideline concentration is currently set at 35 and 40 $\mu g\ m^{-3}$ for China and India, 222 respectively. PM_{2.5} concentrations measured at the two sites were averaged to the filter sampling time 223 and are shown in Fig.1 and Fig. 2. In Beijing, the average 24-h PM_{2.5} concentration was $39 \pm 21 \ \mu g \ m^{-1}$ 224 ³ (range: $16 - 97 \mu \text{g m}^{-3}$), exceeding the Chinese 24-h limit value (75 $\mu \text{g m}^{-3}$) on 1 day of the 20 sampling 225 226 days. The average daytime and night-time PM_{2.5} concentrations in Beijing were $40 \pm 24 \ \mu g \ m^{-3}$ and 38 \pm 24 µg m⁻³, respectively. Delhi showed an average 24-h PM_{2.5} concentration of 143 \pm 27 µg m⁻³ (range: 227 110 - 195 µg m⁻³), exceeding the Indian 24-h limit value on all 9 sampling days. The average daytime 228 and night-time PM_{2.5} concentrations in Delhi were $141 \pm 51 \ \mu g \ m^{-3}$ and $140 \pm 26 \ \mu g \ m^{-3}$, respectively. 229 PM_{2.5} concentrations in Delhi have changed little in recent years; between 2008 and 2011 the daily 230 average of PM_{2.5} concentrations was $123 \pm 87 \ \mu g \ m^{-3}$ (Guttikunda and Calori., 2013), while the annual 231 average was reported to be $125.5 \pm 77.2 \ \mu g \ m^{-3}$ between January 2013 and May 2014 (Winter: 196 μg 232 m⁻³, Summer: 83.6 µg m⁻³, Monsoon: 58.8 µg m⁻³) (Sharma and Mandal., 2017). The limited change 233 234 seen in recent years may be associated with continued emissions from residential energy use, estimated 235 to contribute about 50 % of PM_{2.5} airborne concentrations (Conibear et al., 2018, Butt et al., 2016). The 236 construction sector is fast growing in India, and the dust emitted from various activities (demolition, 237 excavation, drilling etc.) is also considered an important source of particles (Guttikunda et al., 2014).

The WHO has not published a guideline limit value for PAHs, but benzo[a]pyrene (BaP) is often used as a marker of toxicity for all PAHs. The European Union has set an annual mean air quality limit of 1 ng m⁻³ for BaP (WHO, 2016). The daily (24-h) concentration of BaP in summer Beijing (this study)

- ranged from 0.49 to 1.18 ng m⁻³ (average 0.80 ± 0.17 ng m⁻³), about 19 times lower than previously
- observed at this site in winter (Elzein et al., 2019). This is below the 24-h average limit value of 2.5 ng
- m⁻³, set in China by the Ministry of Ecology and Environment in 2012, on all of the 20 days of sampling
 period. During the short summer measurement period in Delhi, BaP varied between 1.16 and 3.16 ng
- 245 m^{-3} (average 1.78 ± 0.67 ng m⁻³), and exceeded the threshold of 2.5 ng m⁻³ on 1 day.
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247 3.2 Variability of PAHs in summer Beijing and Delhi

A time series of the temporal variation of total PAHs in Beijing and Delhi are shown in Fig. 1 and Fig.
2, respectively. The box plots in Fig. 3 show a comparison of the measured concentrations of total 17
PAHs between daytime and night-time in both cities.

252 Beijing:

The mean 3 h daytime concentration of $\sum 17$ PAHs in Beijing was 8.2 ± 5.1 ng m⁻³ ranging from 2.6 to 253 254 31.2 ng m⁻³, while the mean 15 h night-time concentration was 7.2 ± 2.0 ng m⁻³ ranging from 2.8 to 11.4 255 ng m⁻³. The mean 24-h total concentration (combined results from daytime and night-time samples) of the total 17 PAHs was 7.6 \pm 1.9 ng m⁻³ (range: 3.9 – 11.3 ng m⁻³). This 24-h average is approximately 256 257 13 times lower than the average 24-h in winter time Beijing (97 ng m⁻³) reported in our previous study 258 at the same location (Elzein et al., 2019). Similarly, previous studies in urban Beijing (Feng et al., 2005; Wu et al., 2014; Gao and Ji., 2018; Song et al., 2019; Feng et al., 2019) have reported much lower 259 values of PAHs in summer than in winter, showing the important role of seasonal variation in 260 influencing ambient concentrations of PAHs. The dominant reason for this seasonal variation in Beijing 261 is the increase in energy consumption in winter and in particular the emissions from traditional 262 263 rural/urban heating methods using coal. The effect of dispersion in vertical and horizontal directions and long-range transport due to air mass trajectory can also influence local PAHs concentration. 264

265 The photochemical effects on particulate PAHs between summer and winter are not clear in the 266 literature and were considered to play a minor role in seasonal variation of PAHs (Wu et al., 2014). 267 However, an important factor that might affect PAH levels is their degradation or transformation due to high ozone ambient concentration level in summer, which were 5 times higher than in winter (Elzein 268 et al., 2019). This factor can negatively bias PAHs levels during filter sampling by more than 100 % 269 270 (discussed above in section 2.4). The gas phase concentrations of PAHs were not measured in this study, but the distribution of PAHs between the gaseous and particulate phases is an important factor affecting 271 272 their fate in the environment (Lohmann and Lammel, 2004). The distribution of highly volatile PAHs (e.g. 2-3 ring) are known to be influenced by temperature (Tsapakis and Stephanou., 2005; Gaga and 273 274 Ari., 2011; Verma et al., 2017), while the low volatile PAHs and PAHs-derivatives (e.g. oxygenated 275 and nitrated-PAHs) are mostly associated to the particle phase (Albinet et al., 2008; Liu et al., 2017, 276 Zhang et al., 2018). The spatial variation within urban Beijing might have little difference on pollutants 277 concentrations (He et al., 2001), the variation will increase when moving to suburban areas because it 278 consists of a large number of factories, airports, and power plant stations. Feng et al. (2005) compared the total PAHs concentration in $PM_{2.5}$ at urban and suburban sites in Beijing at the same period of the 279 280 year (July and November 2002), and they reported higher values at the suburban site than at the urban 281 site by a factor ~ 1.5 in summer and ~ 2 in winter. The sampling location in this study is an urban area 282 surrounded by busy roads, residential buildings, an underground railway, and restaurants. Therefore, it may be representative of the urban area of Beijing covering approximately half of the population in 283

284 Beijing metropolitan.

In Figure 1, the concentration of total PAHs in the first 3 h filter (08:30–11:30) of the day appear to be higher than the rest of the day. These elevated concentrations are potentially associated with the early morning rush hour time and vehicle emissions. The average night-time total PAHs concentration (7.2 ng m⁻³) was in the range of the second and third "3 h" daytime average values, 7.8 and 6.4 ng m⁻³ respectively. This indicates that the elevated total PAHs concentration in the first 3 h filter (08:30–

- 11:30) of the day is not related to accumulation of air pollutants at night-time but most likely related todirect emissions from traffic in the early morning and particularly petrol combustion emissions (see
- 292 section 3.3).
- 293 To the best of our knowledge, previous studies in Beijing have not reported higher time resolution data
- on total PAHs concentration, but mainly focused on the 24-h average concentration. A previous study
- from Feng et al., (2005) reported a 24-h mean value of 25 ± 6.8 ng m⁻³ in summer Beijing (July 2002), approximately 3.3 times higher than our 24-h mean value (7.6 ng m⁻³). A more recent study from Feng
- et al., (2019) reported a 24-h mean value of 11 ± 5.9 ng m⁻³ in Beijing in warm months (April to June
- 2015), 1.4 times higher than our 24-h mean value. In both studies of Feng et al., (2005 and 2019), the urban sampling site was located at the campus of Peking University health science centre, a short
- distance from our sampling site (~1 mile). In the same year of 2015 (July to September), Chen et al.,
- 301 (2017) and Zhang et al., (2020) reported the values of 9.7 ng m⁻³ and 8.9 ng m⁻³, respectively, for total
- 302 PAHs concentration in $PM_{2.5}$. In the study of Chen et al., (2017), the $PM_{2.5}$ samples were collected at 303 the campus of Beihang university (~2.5 miles from our sampling site). Zhang et al., (2020) sampling
- 304 site was located at a primary school in the Haidian district (~3.5 miles from our sampling site).
- So Furthermore, Gao and Ji (2018) reported 14.5 \pm 1.3 ng m⁻³ for total PAHs concentration in PM_{2.5},
- collected in summer Beijing (May-July, 2016) in the Haidian district (~4 miles from our sampling site).
 Accordingly, the majority of previous studies have focused on studying PAHs in the Haidian district
- 308 (Wu et al., 2014; Chen et al., 2017; Gao and Ji., 2018; Feng et al., 2019; Zhang et al., 2020) because of 309 its high population density. The variation of PAHs concentration from different site locations in the 310 Haidian district may suggest that the spatial variation of PAHs in this area is not significant. Future 311 studies in different districts and rural areas of the metropolitan of Beijing would be helpful for 312 comparison of population exposures and spatial variation.
- The results from studies of PAHs in summer Beijing in recent years (2015-2020) (Chen et al., 2017, Gao and Ji., 2018, Feng et al., 2019, Zhang et al., 2020) show a continuous decrease in PAH concentrations (range: 7 - 15 ng m⁻³) in comparison with the previous decade (2000-2010; range: 11 - 31 ng m⁻³) (Feng et al., 2005, Li et al., 2013, Wu et., 2014). This improvement in air quality (mitigating the emissions of PAHs) could be related to meteorological conditions (e.g. temperature, boundary layer height), but also to the anti-pollution actions adopted by the municipal government of Beijing in 2013 to continue tackling air pollution, by reducing combustion sources in the intervening years and
- 320 promoting the use of clean energy sources such as solar hot water heating systems, banning heavy duty 321 vehicles from circulating in daytime, public transport modernisation, promoting electric vehicles and
- 322 electric motorbikes.
- 323

324 Delhi:

- 325 The population in Delhi is projected to continue growing and to become the most populous city in the world with 39 million people in 2030 (United Nations, 2019), living within a geographic area of 1483 326 km², of which 783 km² is designated as rural and 700 km² as urban (Nagar et al., 2014). The summer or 327 pre-monsoon season (March-June) has very high temperatures and low precipitation. The city is 328 329 surrounded by different climatic zones; the Thar desert in the west, the central hot plains to the south, 330 the Himalayas to the north, and the Indo Gangetic plain in the east (Nagar et al., 2014). The northern 331 and Eastern part of India are considered to be the most polluted part of the country (Guttikunda et al., 332 2014).
- 333 Similar to Beijing, previous studies in Delhi mainly focused on the 24-h average concentration of total
- PAHs. They have addressed the problem of air pollution across India, focusing on $PM_{2.5}$ trends, $PM_{2.5}$
- health impact, and source apportionment (Chowdhury et al., 2007; Gupta et al., 2011; Chowdhury and
- Big Dey., 2016; Pant et al., 2017; Chen et al., 2020), however, PAHs emissions and variation between

337 daytime and night-time have not been measured or discussed previously, and only few studies (Sharma 338 et al., 2007; Singh et al., 2011, Gadi et al., 2019) reported the 24-h mean concentration of \sum n-PAHs (n > 10) in Delhi itself. In this study, the mean 24-h (combined results from daytime and night-time 339 samples) of the 17 PAHs was 19.3 ± 7.1 ng m⁻³ ranging from 13.3 to 35 ng m⁻³. Previous studies showed 340 a large spatial variation in PAH concentrations within Delhi; Sharma et al., (2007) reported a mean 24-341 h value of 624.04 ± 376.48 ng m⁻³ (~ 32 times higher than our mean value) for Σ 12-PAHs at the South 342 of Delhi during the summer season of 2003. Singh et al., 2011 reported a mean 24-h value of 45.8 \pm 343 344 22.1 ng m⁻³ (~ 2.4 times higher than our mean value) for Σ 16-PAHs at the East of Delhi during the summer season of 2008. Gadi et al., (2019) reported an annual average of 277 ± 126 ng m⁻³ for the 345 summation of 16-PAHs between December 2016 and December 2017, and 260 ± 111 ng m⁻³ for the 346 summer season of 2017 (~ 13.5 times higher than our mean value, and ~ 5.7 times higher than Singh et 347 al., 2011 mean value). In the study of Gadi et al., (2019), the sampling site was in the same campus as 348 349 this study (Indira Gandhi Delhi Technical University for Women), the low Σ PAHs value in this study 350 is most probably due to the short summer measurement period in Delhi (9 days, 35 samples), coupled with less pollution episodes for PAHs. It appears that the variation of PAHs concentration in Delhi 351 could be affected by multiple factors including the spatial variation, the input from multiple local 352 353 sources, the surroundings of the sampling site, and the meteorological conditions in the north of Delhi. Therefore, the concentrations of ambient particle-bound PAH over longer averaging period such as the 354 24-h are subject to high uncertainty related to the multiple factors cited above. Higher frequency filter 355 sampling during 24-h can provide a better assessment of PAH concentrations and is more likely to 356 357 reflect direct source emission signals as modified by meteorology. This perspective also applies to other Indian cities in future studies. 358

359 The mean 3 h Σ PAHs concentration in daytime samples in Delhi was 13.6 ± 5.9 ng m⁻³ (~1.7 higher than in Beijing) ranging from 8.4 to 36.6 ng m⁻³, and the mean 15 h night-time samples was 22.7 ± 9.4 360 ng m⁻³ (~3.2 higher than in Beijing) ranging from 13.8 to 42.9 ng m⁻³ as shown in Figure 3. During the 361 362 daytime, the total PAH concentrations were generally highest during the first filter sample (8:30 to 11:30 am) (Fig. 2), following the same trend as for Beijing suggesting vehicle emissions as a dominant 363 source. The mean total PAH concentration at night-time was ~ 1.7 times higher than the mean in 364 365 daytime. Higher total PAHs concentration at night could be related to emissions from biomass burning, waste burning, solid fuel cooking and heavy duty diesels entering the city at night-time. The higher 366 PAHs concentration at night may also be attributed to the relatively lower temperature at night (~10 °C) 367 368 and lower atmospheric mixing heights (Fig. 4), weaker turbulence leading to lower pollutant dispersion rates and absence of photodecomposition. Moreover, power cuts in India are frequent (Guttikunda et 369 al.,2014), especially when high demand occurs on air conditioners in summer (Harish et al., 2020), 370 371 which require in situ electricity generation using oil, diesel and petrol (Guttikunda et al., 2014). This 372 may be considered as an additional source of air pollution in a megacity like Delhi. PM_{2.5} concentrations increased on most nights in comparison with the preceding daytime sample. Residential energy use 373 across India has been reported to be an important source of PM2.5 emissions, contributing 62 % in 374 summer and 70 % in winter of anthropogenic emissions of $PM_{2.5}$ (Conibear et al., 2018). This was also 375 376 confirmed in the study of Butt et al., (2016), showing that the impact of residential combustion 377 emissions on atmospheric aerosol across India is very important, accounting for 63 % of anthropogenic 378 black carbon and 78 % of anthropogenic particle organic matter emissions.

The mean 24-h values of \sum PAHs from this study (19.3 ± 7.1 ng m⁻³) and recent previous studies (45.8 ± 22.1 ng m⁻³ (Singh et al., 2011) and 260 ± 111 ng m⁻³ (Gadi et al., 2019)) are high. This suggests the need for the implementation of a residential emission control strategy through potentially more effective alternative technologies such as the use of non-fossil fuel (biofuel) and clean energy sources (solar, wind, hydro-electric power, natural gas) for domestic use, end the use of in situ power generators by increasing electricity supply and load, and cutting emissions from open waste burning by implementingefficient collection and disposal of waste.

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387 **3.3 Major PAH, particle aging and traffic emissions**

In both campaigns, a high proportion of 5 and 6-ring PAHs were found in the particle phase (Table 1). 388 In Beijing, Benzo[b]fluoranthene, Benzo[k]fluoranthene, Benzo[a]pyrene and Indeno[1,2,3-cd]pyrene 389 390 were the four most abundant particle-bound PAHs in daytime samples, while Benzo[b]fluoranthene, 391 Indeno[1,2,3-cd]pyrene, and Benzo[ghi]perylene were the three dominant particle-bound PAHs in 392 night-time samples. Six major compounds were found in Delhi day and night samples, 393 Benzo[b]fluoranthene, Benzo[k]fluoranthene, Benzo[a]pyrene (BaP), Benzo[e]pyrene (BeP), Indeno[1,2,3-cd]pyrene, and Benzo[ghi]perylene; the mean concentration of these compounds were 394 higher at night. Generally, the 2 and 3-ring PAHs are predominantly found in the gas phase and the 4-395 396 ring PAHs partition between the gas and particle phase (Liu et al., 2013, Zhang et al., 2018). The study 397 from Liu et al., (2013) suggested that ambient temperature largely controls the gas-particle partitioning 398 of 2, 3, and 4-ring PAHs concentration in the gas and particle phases.

A number of molecular diagnostic ratios for source apportionment have been proposed in literature and 399 400 are still under debate (Larsen and Baker., 2003; Katsoyiannis et al., 2011; Keyte et al., 2013; Zheng et 401 al., 2017). They were considered uncertain in some studies because the results were not consistent and 402 reproducible and did not reflect known differences in sources in space and time unless the source is very strong and the sampling measurements are made close to the known source. In addition, they may 403 404 be biased by atmospheric reactions and selective loss processes (Katsoyiannis et al., 2011; Zheng et al., 2017). Among the PAHs, BaP is known to be a particularly carcinogenic compound inducing a 405 406 mutagenic effect in experimental animals and is used as key PAH marker of total exposure (WHO, 2016, IARC, 2012). BaP is mostly emitted from coal and biomass burning, and vehicle exhaust 407 408 emissions (Harrison et al., 1996; WHO, 2016). The sampling site in both Beijing and Delhi was at a 409 short distance from busy roads with significant vehicle exhaust emissions. BaP lifetime is affected by 410 light and oxidants in the atmosphere, BaP is far more reactive than its isomer BeP (Butler and Crossley, 1981, Ringuet et al., 2012b), thus the decline of the ratio BaP/(BaP + BeP) can be regarded as an 411 412 indicator of particle aging (Liu et al., 2013; Watson et al., 2016). Considering the above, BaP/(BaP + 413 BeP) has been used in this study to assess the contribution of local sources on particle composition. Generally, there is no defined threshold value for BaP/(BaP + BeP) which can distinguish aged particles 414 415 from freshly emitted. However, Watson et al., (2016) suggested that similar amounts of BaP and BeP (ratio = 0.5) indicates that particles are freshly emitted and mostly affected by local emissions rather 416 417 than long range atmospheric transport. Moreover, Liu et al., (2013) compared BaP and BeP ratios at 418 remote sites and emission source regions and reported BaP/BeP lower than 0.4 means aged aerosol type, 419 while a value higher than 0.4 indicates local emission source. In this study, a ratio of BaP/(BaP + BeP)420 higher than 0.5 has been considered to characterize local emission sources. The ratio of BaP/(BaP + BeP) has been calculated for both campaigns and results are shown in Figure 5. This ratio mainly varied 421 between 0.5 and 0.6 in Beijing, indicating a dominant contribution from local sources. In Delhi, 25 % 422 of the data were below 0.5 indicating that aged particles might contribute to the air masses collected at 423 424 the sampling site but local emissions of PAH are still a significant source. However, this assumption 425 does not take into account the "on-filter" oxidation errors during sampling. Tsapakis and Stephanou 426 (2003) reported a relative reactivity of BaP of 1.6 times higher than BeP, with respect to degradation 427 on glass fibre filters. Using the reactivity factor of 1.6, the ratio of BaP/(BaP + BeP) will be affected 428 negatively by an average of 4 % (day) and 1.6 % (night) for Beijing, 7 % (day) and 3 % (night) for Delhi. This assumption will therefore be affecting Delhi results suggesting more contribution from long 429 430 range transport. Therefore, the indicator of particle aging should be used with careful in the summer

- 431 season unless ozone ambient concentrations are below 30 ppb, and consequently the negative artefacts432 are considered not significant (Tsapakis and Stephanou 2003).
- Since the influence from traffic emissions is very strong in both Beijing and Delhi, related diagnostic 433 ratios were used to distinguish between petrol and diesel, such as, Fluorene/Fluorene+Pyrene (< 0.5 for 434 Petrol engines and > 0.5 for diesel engines) and Pyrene/BaP (Ravindra et al., 2008; Tobiszewski and 435 Namiesnik., 2012; Watson et al., 2016; Zheng et al., 2017). We calculated the ratio value for Pyrene/BaP 436 437 using the data reported in a previous study (Rogge et al., 1993), where the authors have quantified more 438 than 100 organic compounds in exhaust emissions fine particulate matter. The ratio value for 439 Pyrene/BaP was ~ 0.7 for noncatalyst-equipped petrol cars, ~ 1.3 for catalyst-equipped petrol cars and >16 for heavy duty diesel engines. A recent study from Perrone et al., (2014) reported PM-phased PAH 440 emission factors for different types of vehicles (Euro 3 standards), the ratio value for Pyrene/BaP was 441 \sim 6 and 30 for petrol and diesel cars, respectively. The results from both ratios 442 443 (Fluorene/Fluorene+Pyrene and Pyrene/BaP) suggest high contribution from petrol engines to particle 444 composition in Beijing and Delhi (Fig S1). Tsapakis and Stephanou (2003) reported a relative reactivity for BaP, Pyrene and Fluorene of 0.86, 0.82, and 0.68 respectively. The relative reactivity of BaP and 445 Pyrene are similar and therefore does not affect the indicator Pyrene/BaP values. Pyrene is by 20 % 446 447 more reactive than Fluorene, the "on-filter" oxidation has little effect on the indicator 448 Fluorene/Fluorene+Pyrene values, because of the large difference in the defined threshold values which were 6 and 30 for petrol and diesel cars, respectively. 449

451 **3.4 Emission source fingerprints**

- The profiles of specific class of PAHs and their abundance vary largely, depending on the fuel types 452 453 and combustion conditions (IARC, 2012). Previous studies reported that 2 and 3 aromatic ring PAHs are mostly emitted from wood combustion (Khalili et al., 1995; Larsen and Baker., 2003; Liu et al., 454 455 2017); 2, 3 and 4 ring from diesel exhaust emission (Bourotte et al., 2005; Ravindra et al., 2007; De 456 Souza et al., 2016; Zheng et al., 2017); 3 and 4 ring from coal combustion (Harrison, et al., 1996; Liu et al., 2017); 4, 5, and 6 ring from vehicle emissions (Ravindra et al., 2007, Zhao et al., 2020); 5 and 6 457 ring from petrol and oil combustion (Harrison, et al., 1996; Ravindra et al., 2007). In this study, we 458 459 have classified the 17 PAHs based on their number of aromatic ring; we referred to previous studies 460 and to our knowledge of local sources, sampling site and surroundings, and analytical uncertainties to describe the emission source of each class. The 2 and 3 aromatic ring PAHs were below LOD and LOQ 461 462 in many samples, however, they are predominantly found in the gas phase and their partitioning to the particle phase is very small because of their high volatility; their percentage in the particle phase was 463 464 previously reported to be less than 10 % (Ravindra et al., 2007; Liu et al., 2017; Zhao et al., 2020).
- 465 In this study, the 4, 5, and 6 ring PAHs accounted ~ 95 % of the total $PM_{2.5}$ -bound PAH concentrations measured in both campaigns (Fig. 6). The mean contribution of the number of ring to the total of PAHs 466 in Beijing $PM_{2.5}$ was distributed in the order 5 > 4 > 6 ring, and in Delhi as 5 > 6 > 4 ring (Fig. 6). In 467 both, Beijing and Delhi, road traffic is known to be one of the largest emission source of gas and particle 468 phase pollutants (Fang et al., 2016; Zhang et al., 2020; Shivani et al., 2019), which might explain the 469 470 high contribution from 4, 5 and 6 ring to total PAHs, while diesel and coal combustion may also contribute to the emission of 4 ring PAHs. The 4 ring PAHs concentration in Delhi was higher than in 471 472 Beijing, the mean concentration of the 4 ring PAHs in daytime and night-time samples was 2.7 and 2.1 ng m⁻³ in Beijing; 3.1 and 4.1 ng m⁻³ in Delhi, respectively (Table 1). In contrast to Beijing (5 > 4 > 6)473 ring), the 4 ring PAHs total concentration in Delhi was lower than the 6 ring PAHs (Fig. 6) and a 474 475 potential emission source could be the common use of fuel and oil in power generators.
- The distribution of ring PAHs in Figure 6 shows a comparison between the results from this study and our previous study in winter Beijing (Elzein et al., 2019). The contributions from 2, 3 and 4 rings were higher in winter than in summer, and the ring PAHs are distributed as 4 > 5 > 6 > 2-3 ring (Fig. 6).

- 479 Ambient temperature highly affects the gas/particle partitioning of 2-3 ring PAH (Tsapakis and Stephanou., 2005; Gaga and Ari., 2011; Verma et al., 2017). Therefore, higher contributions from 2, 3 480 481 and 4 rings in winter Beijing are likely due to lower temperature and to the use of coal and wood combustion for residential heating. The 2-3 and 6 ring PAHs contributions in winter Beijing are 482 relatively similar, while in summer Beijing the contribution from 6 ring PAHs is ~ 20 % higher than 2-483 3 ring, most probably due to the effect of ambient temperature on 2-3 ring PAH. The 5 ring PAHs 484 485 (representative of vehicle emissions) contribute the most to the total PAH concentration in summer 486 Beijing and Delhi, suggesting a high contribution from petroleum combustion.
- Finally, identifying PAH markers emitted from specific emission sources (types of fuel, types of coal, types of waste, etc.) in ambient air is still complex due to the similarity of PAH profiles from different source types, which may quickly blend in the air with interferences from both nearby and remote emission sources. A more complete assessment of emission source types at specific locations would require the use of individual PAHs as source markers combined with other chemical constituents of PM_{2.5} (elements and ions) and with gas phase air pollutants known to be released from the same source such as VOCs markers.
- 494

495 **3.5 Health risk assessment**

- 496 Although PAHs have long been recognized as carcinogenic environmental pollutants, BaP is still the 497 only PAH allowing a quantitative risk assessment (WHO, 2000, Boström et al., 2002). BaP is used as the most common reference chemical as being representative for PAH mixtures from emissions of coke 498 ovens and similar combustion processes in urban air (WHO, 2000). Relative potencies of individual 499 PAHs (relative to BaP) have also been published as toxicity equivalency factors (TEF) (Nisbet and 500 501 LaGoy, 1992; Larsen et al, 1998; Durant et al, 1996, OEHHA., 1994). Thus, the carcinogenic risk of the mixture of PAHs can be expressed as BaP equivalents (BaPeq). The equivalent exposure to the index 502 compound (i.e. BaP) can be calculated from the TEF of each target compound (Table S2) multiplied by 503 504 its corresponding concentration in ng m⁻³.
- 505 $[BaP]_{eq} = \sum_{i=1}^{N} PAH_i \times TEF_i$ (1)
- To estimate the statistical potential of contracting cancer from inhalation and lifetime exposure to PM_{2.5}-506 bound PAHs, commonly known as the lifetime excess cancer risk (LECR) shown in Eq. (2). we have 507 used the WHO unit risk (UR) estimate of 8.7 x 10⁻⁵ (ng m⁻³)⁻¹ (WHO, 2000), meaning that 8.7 people 508 509 per 100 000 people may contract lung cancer when exposed continuously to 1 ng m⁻³ of BaP concentration over a lifetime of 70 years. This risk refers to the total PAH mixture and not only to the 510 BaP content (U.S.EPA, 2002, Boström et al., 2002) and is referred to as the surrogate approach. The 511 512 use of BaPeq instead of BaP in Eq.2, overestimates the LECR, and therefore the use of the actual 513 measured BaP concentration better assesses the lifetime cancer risk following Eq. (2)
- $LECR = BaP \times UR$ (2)

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Table 2. Mean concentration of BaP_{eq} , BaP and LECR assessment for Beijing and Delhi.

Sampling location	$Mean [BaP]_{eq} \pm SD^*$ (ng m ⁻³)	Mean BaP \pm SD* (ng m ⁻³)	LECR	LECR per million people
Beijing / 24-h	1.47 ± 0.35	0.8 ± 0.17	7 x 10 ⁻⁵	70
Delhi / 24-h	3.42 ± 1.35	1.78 ± 0.67	15.5 x 10 ⁻⁵	155

517 * SD : Standard deviation

As shown in Table 2, the LECR attributable to the 15 PAHs in urban air of Beijing and Delhi was 7 x

519 10^{-5} and 15.5 x 10^{-5} (> 10^{-6}), respectively, suggesting an elevated lifetime cancer risk for adults (Chen

and Liao., 2006; Bai et al., 2009), especially when considering the population size associated with each city. The LECR value for Delhi gives an estimate of 85 additional cancer cases per million people exposed, in comparison to Beijing. The LECR for Beijing in winter (Elzein et al., 2019) was much higher than in summer (this study) and showed 1235 additional cases, and this was mostly attributed to the increase in use of fossil fuels for central and residential heating, in addition to meteorological conditions such as lower volatilisation at low temperatures and lower photochemical transformation. It is however the annual mean which is directly related to cancer risk.

Although BaP is widely used as indicator of all PAHs carcinogenicity, this approach is still under debate
and may not give a very good representation of the whole mixture potency (U.S.EPA, 2002, Boström
et al., 2002). Delgado-Saborit et al., (2011) have used the TEFs to calculate the percentage contribution
of each PAH to total carcinogenicity following Eq. (3):

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$$(\text{\%Carc. Potential})_{i} = \frac{(\text{RC} \times \text{TEF})_{i}}{\sum_{i=1}^{N} (\text{RC} \times \text{TEF})_{i}} \times 100$$
(3)

where RC is the relative abundance marker of an individual PAH to the carcinogenic marker BaP (RC
= (PAH)_i/(BaP)). Using Eq.3, the compounds that contribute most to the total carcinogenic potential of
the PAH mixture in Beijing (B) and Delhi (D) are: Benzo[a]pyrene (B: 46 %; D: 48 %),
Dibenzo[a,h]anthracene (B: 23 %; D: 19 %), Benzo[b]fluoranthene (B: 15 %; D: 13 %),
Benzo[k]fluoranthene (B: 5 %; D: 5 %), and Indeno[1,2,3-cd]pyrene (B: 6 %; D: 10 %). The sum of all
other PAHs used in this study was about 5 %.

539 Since the majority of people (~ 90%) spend most of their time indoor, the total PAH burden from 540 inhalation has been related to indoor air, and BaP is used as a marker for the carcinogenic potential of all PAHs irrespective of the environment (indoor or outdoor) (Delgado-Saborit et al., 2011). In this 541 542 study, the health risk evaluation was only based on inhalation exposure to PAHs in the particulate phase. 543 The risk values can increase due to the presence of PAHs derivatives in the particulate phase such as the nitrated-PAHs (Elzein et al., 2019), and in particular from 6-Nitrochrysene and 1,6-dinitropyrene 544 545 who have been attributed a high TEF value equal to 10 (OEHHA., 1994; WHO., 2003; Lundstedt et al., 2007). Dermal exposure to PAHs is also an important risk factor for skin cancer but toxicity values for 546 547 dermal exposures are still not available (U.S.EPA, 2002), in addition ingestion exposure to PAHs from 548 soil, sediments and water is high (Li et al., 2010), and both exposures (dermal+ingestion) can highly 549 exceed the risk from inhalation (U.S.EPA, 2002).

PAHs in the gas phase are mostly low molecular weight (2-3 ring PAHs) and their partitioning to the 550 particle phase is small (Ravindra et al., 2007; Liu et al., 2017; Zhao et al., 2020) with lower TEF values 551 (Table S2), therefore, their contribution to total carcinogenic potential is low (< 5 % in this study). 552 Previous studies may consider a variable number of PAHs and other aromatics with known TEF such 553 554 as the nitrated-PAHs, including different references for TEF values which make a direct comparison of the carcinogenic risk between studies not ideal. However, as particulate-PAH concentrations are lower 555 556 in warm months, related BaPeq values were also lower in warm months. Feng et al., 2019 reported an average BaPeq total concentration of 20 PAHs equal to 1.9 and 21.9 ng m⁻³ in summer and winter 557 Beijing, respectively. In this study, the summer BaPeq total concentration of 15 quantified PAHs in 558 Beijing was 1.47 ng m⁻³, while in our previous study for winter Beijing (Elzein et al., 2019), the BaPea 559 total concentration of 16 quantified PAHs and 7 derivatives was 23.6 ng m⁻³. The results from this study 560 suggest to focus attention on mitigating the emission of major contributors to the total carcinogenic 561 562 potential (Benzo[a]pyrene, Dibenzo[a,h]anthracene, Benzo[b]fluoranthene, Benzo[k]fluoranthene, and Indeno[1,2,3-cd]pyrene) in order to reduce adverse health effects from exposure to this class of air 563 pollution. These compounds are 5 and 6 ring PAHs, and were mostly related to emission from petrol 564

and oil combustion (Harrison, et al., 1996; Ravindra et al., 2007).

567 4 Conclusions

- 568 Diurnally-resolved samples of ambient $PM_{2.5}$ were collected in Beijing-China from 22 May 2017 to 10 569 June 2017 (20 days), and in Delhi from 28 May 2018 to 5 June 2018 (9 days). The 24-h average 570 concentration of $PM_{2.5}$ was $39 \pm 21 \ \mu g \ m^{-3}$ (range: 16 - 97 $\mu g \ m^{-3}$) in Beijing, exceeding the Chinese 571 24-h guideline value (75 $\mu g.m^{-3}$) on 1 day of the 20 sampling days, while in Delhi the 24-h average 572 concentration of $PM_{2.5}$ was $143 \pm 27 \ \mu g \ m^{-3}$ (range: 110 - 195 $\mu g \ m^{-3}$) exceeding the Indian 24-h 573 guideline value (60 $\mu g.m^{-3}$), on all 9 sampling days. High contribution to $PM_{2.5}$ emissions was attributed 574 to residential energy use emissions and to the construction sector.
- In Beijing and Delhi, 17 PAHs were quantified using a GC-Q-TOF-MS and the measured concentrations compared between daytime and night-time, showing a high relative proportion (~ 95 %) of 4, 5 and 6-ring PAHs in the particle phase. In Beijing, $\sum 17$ PAH concentrations varied between 2.6 and 31.2 ng m⁻³ (average 8.2 ± 5.1 ng m⁻³) in daytime, and from 2.8 to 11.4 ng m⁻³ (average 7.2 ± 2.0 ng m⁻³) at night-time. In Delhi, $\sum 17$ PAH concentrations varied between 8.4 and 36.6 ng m⁻³ (average
- 580 13.6 ± 5.9 ng m⁻³) in daytime, and from 13.8 to 42.9 ng m⁻³ (average 22.7 ± 9.4 ng m⁻³) at night-time.
- 581 In Beijing, Indeno[1,2,3-cd]pyrene was the highest contributor to the mean total PAHs during daytime
- (12 %) and Benzo[b]fluoranthene at night (14 %) at night-time. In Delhi, Indeno[1,2,3-cd]pyrene was
 the largest contributor to the total PAHs in both the day (17 %) and night-time (20 %).
- The elevated mean concentration of total PAHs in Delhi at night was attributed to emissions from biomass burning, waste burning, open fire cooking along with meteorological conditions facilitating the accumulation of air pollutants as a result of low atmospheric boundary layer heights.
- The ratio of BaP/(BaP + BeP) has been used to evaluate the contribution from local sources against
 long range atmospheric transport of particle-bound PAHs. This ratio suggests a larger contribution from
 local sources in both cities. Flu/Flu+Pyr and Pyr/BaP were used as diagnostic ratios to distinguish
 between petrol and diesel, and results suggest petrol combustion emissions as a major source in Both
 Beijing and Delhi.
- 592 PAHs were classified according to their number of aromatic rings to characterize major emission 593 sources. The 4, 5, and 6 ring PAHs accounted ~ 95 % of the total PM_{2.5}-bound PAH concentrations in 594 both campaigns. The 5 ring PAHs contribute the most to the total PAH concentration in summer Beijing 595 and Delhi, suggesting a high contribution from petroleum combustion. In Beijing, the 4 ring PAHs total concentration was higher than the 6 ring by 8 % during the day and 5 % at night, while in Delhi, the 6 596 ring PAHs total concentration was higher than the 4 ring PAHs by 7 % during the day and 18 % at 597 night; a potential emission source of 6 ring PAHs in Delhi could be the common use of fuel and oil in 598 599 power generators. Due to the similarity of PAH profiles from different source types, it would be 600 beneficial to use other source markers such as elements and ions in PM2.5 and VOCs markers from the gas phase to better identify the differences in emission sources. 601
- The lifetime excess lung cancer risk was calculated for Beijing and Delhi, with the highest estimated risk attributed to Delhi (LECR = 155 per million people), 2.2 times higher than Beijing risk assessment value (LECR = 70 per million people). The results from this study suggest focusing attention on mitigating the emission of major contributors to the total carcinogenic potential, being the 5 and 6 ring PAHs (mostly emitted from petrol and oil combustion), in order to reduce adverse health effects from inhalation exposure to PAHs in the particulate phase.
- Finally, in Beijing, the anti-pollution actions since 2013 appear to have had a positive effect on the air quality, while in Delhi, despite the government effort to mitigate air pollutants emission, a strict implementation of emission control policies is still needed with particular focus on mitigating residential emissions and burning, increasing the electricity supply to cover peak demand in summer and limiting the use of local power generators as well as, promoting cleaner vehicles. Future studies in

different districts of Beijing (other than Haidian) and Delhi (other than old Delhi) and rural areas would
be helpful for comparison of population exposures and spatial variation. Higher frequency filter
sampling (every 3 h) can provide a better assessment of PAH concentrations and photochemistry and
can lead to better conclusions on direct source emission signals as modified by meteorology during the
daytime and night-time.

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619 *Author contributions:* AE conducted the chemical analysis, analysed the data and prepared the 620 manuscript. ACL, JFH and RMH contributed to the interpretation, writing and corrections of the paper. 621 ERV calculated and provided the data on Delhi planetary boundary layer height. RG supported on site 622 filter collection and helped to set up the laboratory at the field site. AE and SJS conditioned and 623 collected the filter samples in Beijing. GJS and BSN conditioned and collected the filter samples in 624 Delhi. LRC and MSA measured and provided the data on $PM_{2.5}$ in Delhi. All authors reviewed and 625 contributed to corrections of the paper.

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627 *Competing interests.* The authors declare that they have no competing interests.

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Figure 1 Temporal variation of total PAHs and $PM_{2.5}$ concentrations in summer Beijing. $PM_{2.5}$ concentrations were averaged to the filter sampling time, approximately 3 h in daytime and 15 h at night. The 3 h midpoint time tick labels at noon (~13:00) and in the afternoon (~16:00) have been omitted for clarity.



¹⁰¹⁰ 1011

1012 **Figure 2** Temporal variation of total PAHs and $PM_{2.5}$ concentrations in summer Delhi. $PM_{2.5}$ 1013 concentrations were averaged to the filter sampling time, approximately 3 h in daytime and 15 h at 1014 night.





Figure 3. Concentrations of 17 PAHs in PM_{2.5} samples during the daytime (3 h) and night-time (15 h).
Box plots represents the 25th and 75th percentiles range of the observed concentrations and the whisker
numbers reflect the data within 1.5 times the interquartile range (IQR). Red square symbols represent
the mean concentration, and the short dash line within the boxes represent the median. Empty Triangles
correspond to the data measured over 3 h and 15 h in Beijing (Blue) and in Delhi (Purple). The lines
between data points and boxes reflect a normal distribution curve.





Figure 4 Planetary boundary layer height in summer Delhi 2018 (Lat 28.625, Lon 77.25; source:
 ECMWF ERA5 in 0.25°, 1-hour time resolution). Square symbols represent the temporal variation of
 total PAHs in daytime and night-time.



Figure 5. Ratio of BaP/(BaP + BeP) in $PM_{2.5}$ samples during the daytime (every 3 h) and night-time (15 h). Violin plots represent the data distribution between day and night and the wider sections represent the higher density of data smoothed by a kernel density estimator. Red rectangle symbols represent the mean concentration, and the red line connect the mean values between day and night.

- 1033
- 1034

	Summer Beijing: PAH concentrations (ng m ⁻³)		Summer Delhi: PAH concentrations (ng m ⁻³)				
Compound/ring	Variation	Mean \pm SD*	Variation	Mean \pm SD*			
	Daytime / Night-time	Daytime / Night-time	Daytime / Night-time	Daytime / Night-time			
17 PAHs	3 h / 15 h	3 h / 15 h	3 h / 15 h	3 h / 15 h			
Naphthalene/2	0.07-0.29 / 0.02-0.15	$0.14 \pm 0.06 \: / \: 0.06 \pm 0.04$	0.07-0.62 / 0.35-0.58	$0.29 \pm 0.14 \: / \: 0.45 \pm 0.09$			
Acenaphthylene/3	0.08-0.12 / 0.02-0.08	$0.10 \pm 0.03 \: / \: 0.04 \pm 0.02$	0.05-0.15 / 0.15-0.29	$0.09 \pm 0.02 \: / \: 0.19 \pm 0.06$			
Acenaphthene/3	< LOD	< LOD	< LOD	< LOD			
Fluorene/3	0.09-0.15 / 0.02-0.07	$0.13 \pm 0.03 \: / \: 0.05 \pm 0.01$	0.04-0.09 / 0.03-0.09	$0.07 \pm 0.02 \: / \: 0.05 \pm 0.02$			
Phenanthrene/3	0.16-1.28 / 0.07-0.42	$0.37 \pm 0.25 \: / \: 0.28 \pm 0.09$	0.14-0.58 / 0.56-0.86	$0.26 \pm 0.12 \: / \: 0.70 \pm 0.10$			
Anthracene/3	< LOD	< LOD	< LOD	< LOD			
Total 2-3 ring PAHs	0.04-1.30 / 0.06-0.62	$0.38 \pm 0.29 \:/\:\: 0.41 \pm 0.13$	0.15-1.24 / 0.98-1.87	$0.62 \pm 0.28 \: / \: 1.36 \pm 0.29$			
Fluoranthene/4	0.15-5.78 / 0.25-1.58	$0.74 \pm 0.99 \: / \: 0.66 \pm 0.31$	0.36-1.76 / 0.8-1.28	$0.67 \pm 0.29 \: / \: 1.04 \pm 0.16$			
Pyrene/4	0.12-2.84 / 0.19-1.12	$0.56 \pm 0.58 \: / \: 0.56 \pm 0.21$	0.37-2.10 / 0.82-1.36	$0.69 \pm 0.36 \: / \: 1.07 \pm 0.18$			
Benzo[a]anthracene/4	0.43-1.22 / 0.12-0.53	$0.56 \pm 0.16 / 0.30 \pm 0.10$	0.59-1.59 / 0.49-1.32	$0.75 \pm 0.20 \: / \: 0.72 \pm 0.27$			
Chrysene/4	0.42-3.27/0.28-1.12	$0.81 \pm 0.51 / 0.62 \pm 0.20$	0.64-2.35 / 0.88-1.95	$0.94 \pm 0.34 \: / \: 1.20 \pm 0.35$			
Total 4 ring PAHs	1.14-13.4 / 0.84-4.24	$2.67 \pm 2.19 \; / \; 2.14 \pm 0.77$	2.04-7.83 / 3.05-6.17	$3.10 \pm 1.19 \ / \ 4.10 \pm 1.02$			
Benzo[b]fluoranthene/5	0.44-3.76/0.41-1.78	$0.97 \pm 0.65 \ / \ 1.02 \pm 0.33$	0.87-3.73 / 1.34-4.86	$1.40 \pm 0.59 / 2.41 \pm 1.12$			
Benzo[k]fluoranthene/5	0.56-2.67/0.31-1.06	$\textbf{0.93} \pm \textbf{0.43} \ / \ 0.68 \pm 0.19$	0.96-3.38 / 1.05-3.44	$1.39 \pm 0.48 \: / \: 1.80 \pm 0.75$			
Benzo[a]pyrene/5	0.63-2.56/0.34-1.15	$\textbf{0.95} \pm \textbf{0.38} \ / \ 0.71 \pm 0.19$	0.95-3.41 / 1.10-3.92	$1.40 \pm 0.51 \ / \ 2.01 \pm 0.90$			
Benzo[e]pyrene/5	0.55-2.08/0.12-0.92	$0.80 \pm 0.32 \: / \: 0.47 \pm 0.22$	0.88-3.67 / 1.04-5.70	$1.37 \pm 0.61 / 2.14 \pm 1.49$			
Dibenzo[a,h]anthracene/5	0.69-0.78/0.14-0.26	$0.74 \pm 0.05 \: / \: 0.19 \pm 0.03$	0.64-0.91 / 0.12-1.08	$0.76 \pm 0.11 \: / \: 0.39 \pm 0.27$			
Total 5 ring PAHs	0.65-12 / 1.21-4.24	$3.57 \pm 1.97 \; / \; 2.98 \pm 0.80$	3.68-15.11 / 4.81-19.0	$5.69 \pm 2.36 \ / \ 8.88 \pm 4.49$			
Indeno[1,2,3-cd]pyrene/6	0.65-3.27 / 0.39-1.57	$1.03 \pm 0.47 \: / \: 0.93 \pm 0.31$	1.32-6.36 / 2.12-8.62	$2.29 \pm 1.11 / 4.45 \pm 2.03$			
Benzo[ghi]perylene/6	0.44-2.45 / 0.14-1.37	$0.77\pm0.37/\textbf{0.80}\pm\textbf{0.3}$	1.08-6.08 / 2.01-7.53	$1.97 \pm 1.13 \ / \ 3.90 \pm 1.76$			
Total 6 ring PAHs	0.43-5.71 / 0.66-2.72	$1.76 \pm 0.86 \:/\:\: 1.72 \pm 0.54$	2.47-12.44 / 4.12-15.7	$4.26 \pm 2.24 \; / \; 8.34 \pm 3.64$			
Total 17 PAHs	2.6-31.2 / 2.8-11.4	$8.2\pm 5.1/7.2\pm 2.0$	8.4-36.6 / 13.8-42.9	$13.6 \pm 5.9 \: / \: 22.7 \pm 9.4$			
Total 16-PAHs Winter Beijing (Elzein et al., 2019)	18-297 / 23-165	$87.3 \pm 58 \ / \ 107 \pm 51$					
037 * SD: Standard Deviation							

Table 1. Minimum, maximum and mean concentrations of individual PAHs in PM_{2.5}. Compounds in
 bold represent the highest mean contribution to the sum of all compounds.



Figure 6. Distribution of PAHs compounds by number of cyclic rings in PM_{2.5} samples collected during
the daytime (every 3 h) and night-time (15 h) for summer Beijing 2017 and Delhi 2018 (this study) and
winter Beijing 2016 (Elzein et al., 2019).