

SUPPLEMENTARY MATERIALS

Measurement report: Spatial variations in snowpack ionic chemistry and water stable isotopes across Svalbard

Elena Barbaro^{1,2}, Krystyna Kozioł³, Mats P. Björkman⁴, Carmen P. Vega⁵, Christian Zdanowicz⁶, Tonu Martma⁷, Jean-Charles Gallet⁸, Daniel Kępski⁹, Catherine Larose¹⁰, Bartłomiej Luks⁹, Florian Tolle¹¹, Thomas V. Schuler^{12,13}, Aleksander Uszczyk¹⁴ and Andrea Spolaor^{1,2*}

¹Institute of Polar Sciences, ISP-CNR, Via Torino 155, 30170 Venice Mestre, Italy

²Department of Environmental Sciences, Informatics and Statistics, Ca' Foscari University of Venice, Via Torino 155, 30172, Venice, Italy.

³Department of Analytical Chemistry, Chemical Faculty, Gdańsk University of Technology, G. Narutowicza 11/12, 80-233 Gdańsk, Poland.

⁴Department of Earth Sciences, University of Gothenburg, Box 460, SE-40530 Göteborg, Sweden.

⁵Dirección Meteorológica de Chile, Dirección General de Aeronáutica Civil, Portales 3450, Santiago, Chile.

Previously at: Department of Earth Sciences, Uppsala University, Villavägen 16, Uppsala, Sweden.

⁶Department of Earth Sciences, Uppsala University, Villavägen 16, SE-76236, Uppsala, Sweden.

⁷Department of Geology, Tallinn University of Technology, Ehitajate tee 5, 19086 Tallinn, Estonia

⁸Norwegian Polar Institute, Tromsø, No-9296, Norway

⁹Institute of Geophysics, Polish Academy of Sciences, Księcia Janusza 64, 01-452 Warsaw, Poland

¹⁰Environmental MicrobialGenomics, Laboratoire Ampère, CNRS, University of Lyon, France

¹¹Université de Franche-Comté, Besançon, FEMTO-ST, UMR 6174 CNRS

¹²Departement of Geosciences, University of Oslo, Oslo, Norway

¹³Arctic Geophysics, University Center on Svalbard, UNIS, Longyearbyen, Svalbard, Norway

¹⁴University of Silesia in Katowice, Faculty of Natural Sciences, Będzińska 60, 41-200 Sosnowiec, Poland

*Corresponding author

Andrea Spolaor

Keywords

Snow, Svalbard, Arctic, inorganic ions, water isotopes

Figure S1. Pie diagrams showing relative ionic composition in the snow pits dug in the accumulation zones of the studied glaciers.

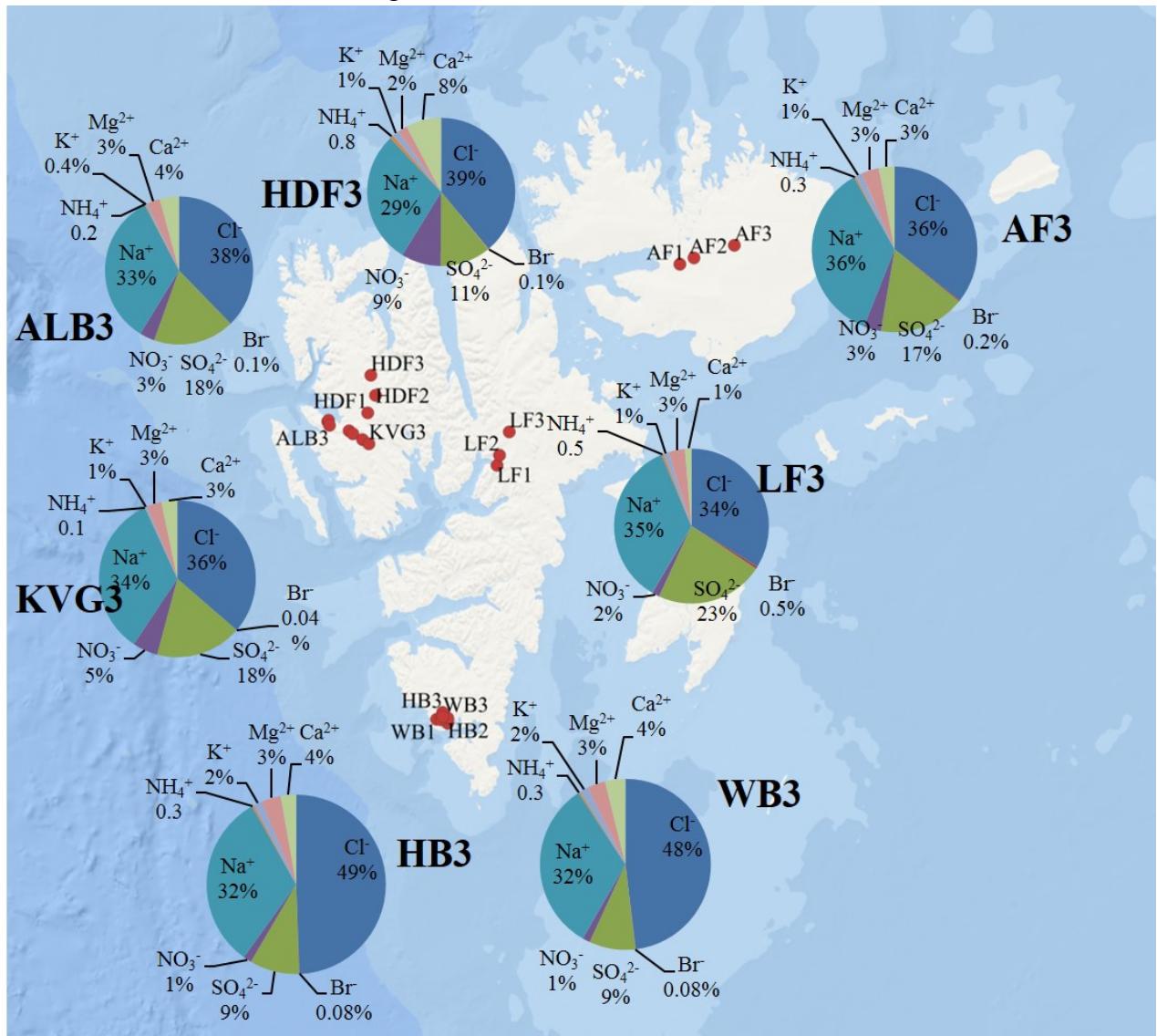


Figure S2. Box plots of stable water isotopes ($\delta^{18}\text{O}$ and $\delta^2\text{H}$) and deuterium excess (d) for each snow pit.

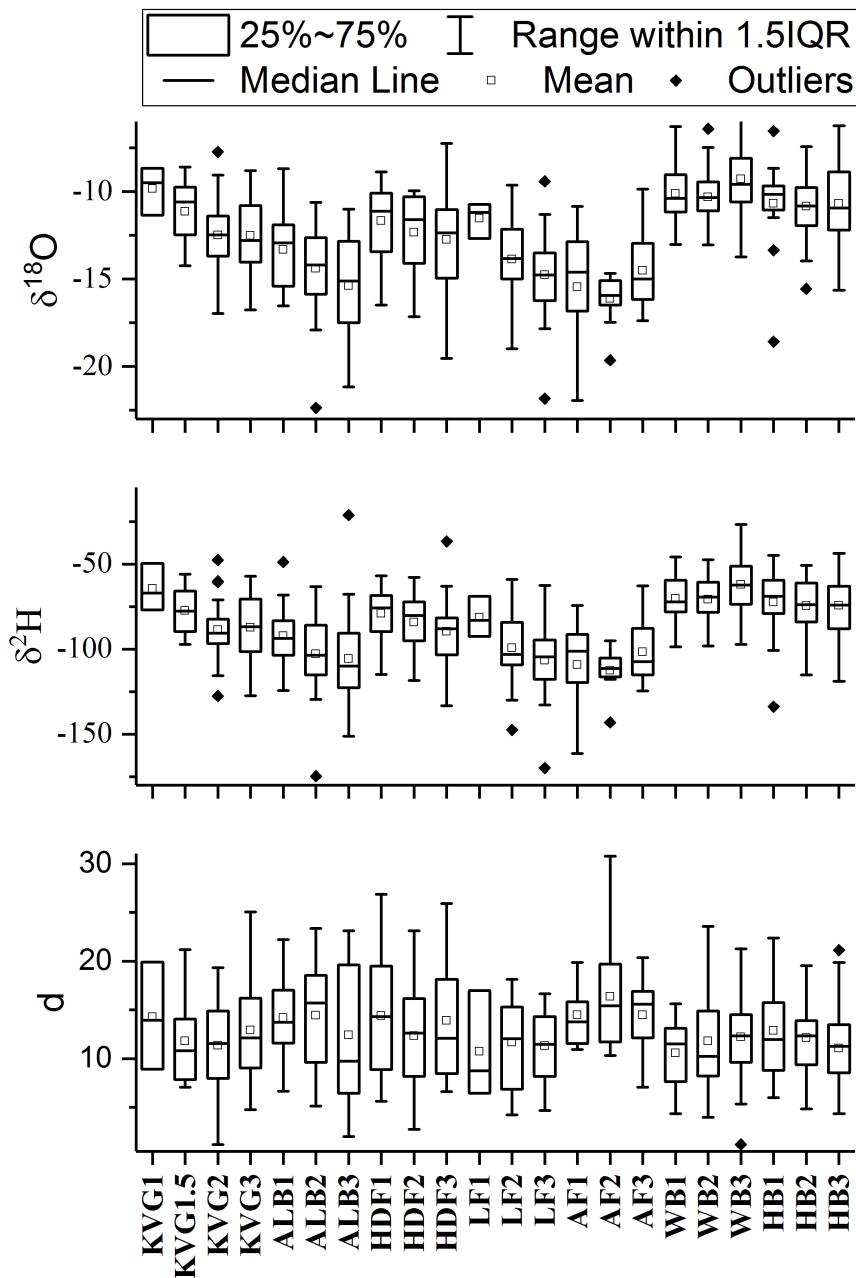


Figure S3. Ion balance of the total load for each glacier sampled during the C2S3 project. X^- denotes missing anions in the balance. Frequently in natural waters X^- is found to be composed mostly of HCO_3^- , which cannot be determined with ion chromatography.

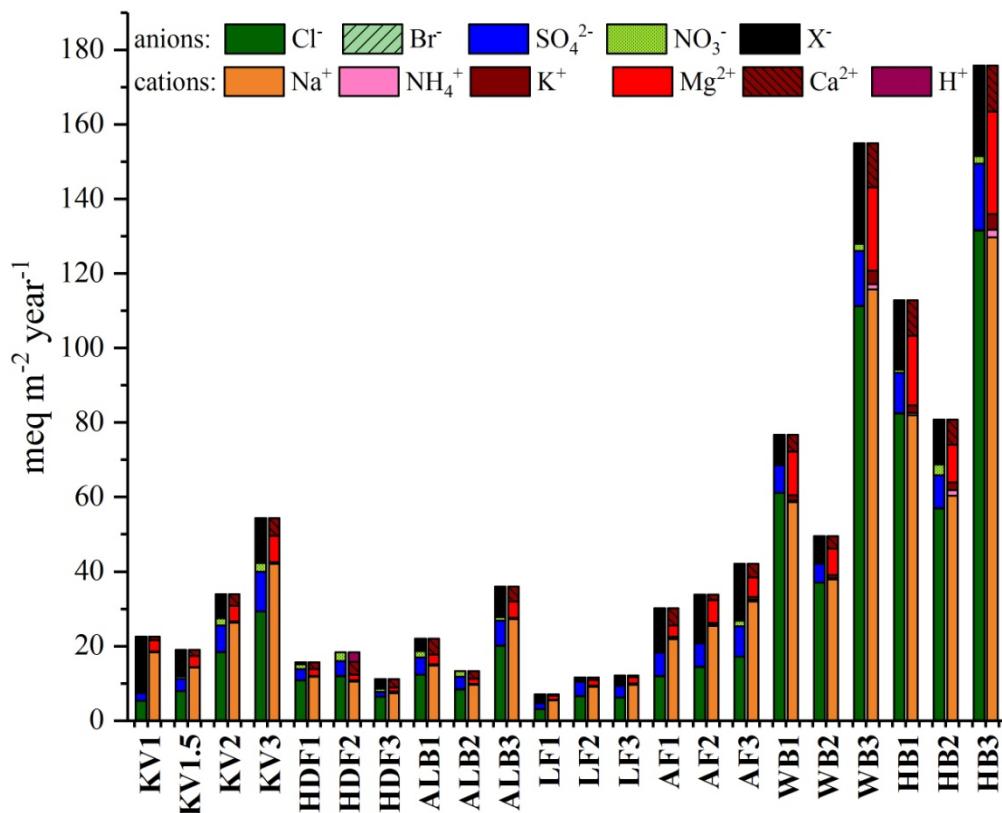


Figure S4. North Atlantic Chlorophyll-a concentration in April and May 2016. The data were obtained via an Aqua/MODIS NASA satellite continually orbiting the globe (https://neo.sci.gsfc.nasa.gov/view.php?datasetId=MY1DMM_CHLORA) with 1-month resolution and plot on © Google Earth platform. The data provides an estimate of the near-surface concentration of chlorophyll, calculated using an empirical relationship derived from in situ measurements of chlorophyll-a and remote sensing reflectance (Rrs) in the blue-to-green region of the visible spectrum.

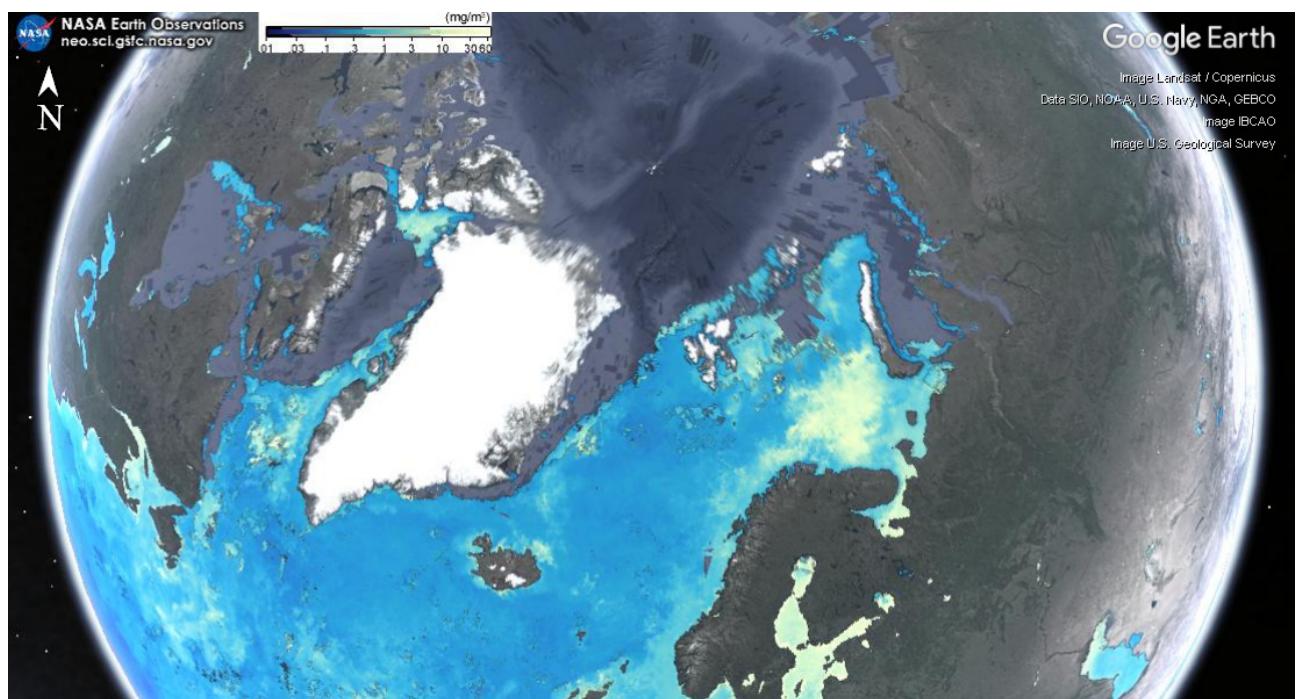


Figure S5. Bromide enrichment (Br_{enr}), calculated for each glacier. The snow pits have been ordered from lower to higher Br_{enr} value (bottom panel; the dashed line indicates the referenced Br/Na sea water ratio). The uppermost panel compares the Br_{enr} and the altitude, the upper middle with longitude, and the lower middle with latitude.

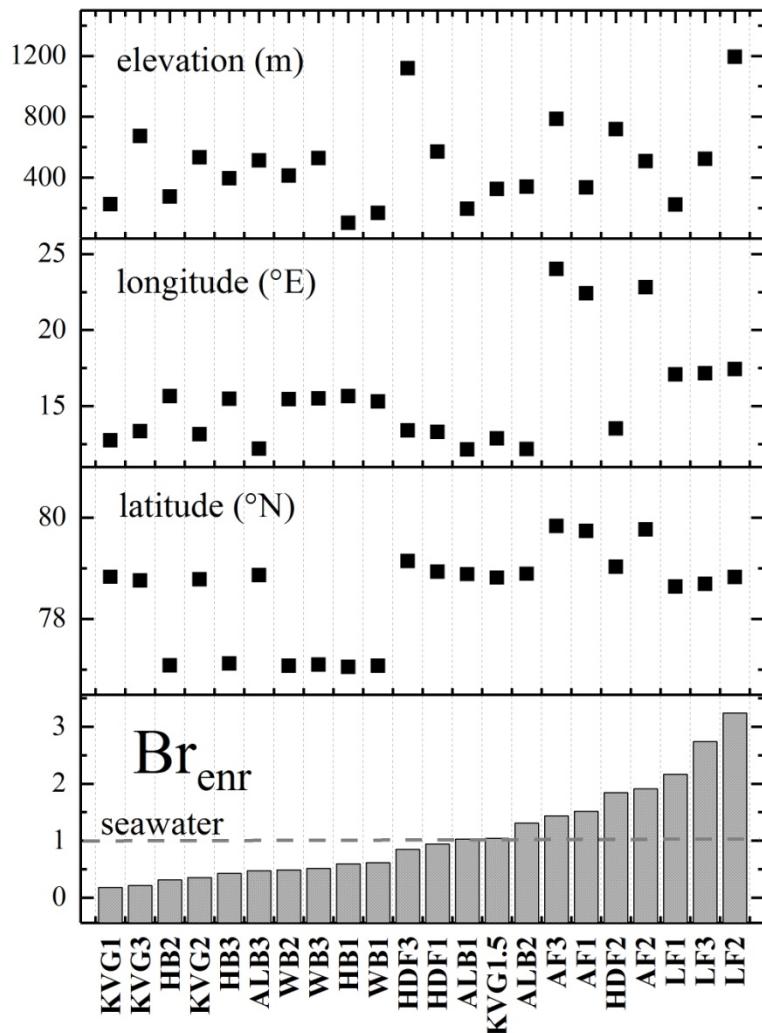


Table S1. Performance of analytical methods applied at Hornsund, Venice and Uppsala laboratories. Method detection limit (MDL) was calculated as three times the standard deviation of procedural blank (ultrapure water). Error (%) and relative standard deviation (RSD%) were determined to evaluate the trueness and precision of each method, respectively.

	Hornsund method				Venice method				Uppsala method			
	Linear range mg/L	MDL (mg/L)	Error %	RSD %	Linear range mg/L	MDL (mg/L)	Error %	RSD %	Linear range mg/L	MDL (mg/L)	Error% %	RSD %
Cl ⁻	0.003-20	0.003	-3	5	0.001-4	0.2	-9	7				
Br ⁻	0.004-1	0.004	0	1	0.002-0.2	0.002	-5	6				
SO ₄ ²⁻	0.003-20	0.003	-3	2	0.002-8	0.07	4	3				
NO ₃ ⁻	0.006-10	0.006	-2	1	0.002-2	0.04	-8	4				
Na ⁺	0.03-15	0.03	0	2	0.1-10	0.05	4	4	0.001-1	0.004	<±5%	1
NH ₄ ⁺	0.02-2	0.02	n.d.	n.d.	0.1-10	0.005	n.d.	n.d.	0.001-1	0.001	n.d.	0.5
K ⁺	0.05-5	0.05	-1	2	0.1-10	0.02	-8	1	0.001-1	0.002	<±5%	2.0
Mg ²⁺	0.05-5	0.05	-20	3	0.1-10	0.007	-9	1	0.001-1	0.001	<±5%	0.4
Ca ²⁺	0.08-25	0.08	-2	2	0.1-10	0.04	-1	1	0.001-1	0.006	<±5%	5