

Response to the Comments of the Reviewers

Dear Editor and Reviewers,

We would like to thank you and the reviewers for the great efforts and elaborate work on this manuscript.

We revised the manuscript by responding to each of the suggestions in the reviews. In our response, the questions of the reviewers are shown in *Italic* form and the responses in standard form.

We appreciate your help and time.

Sincerely yours,

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Manuscript Title: Spatiotemporal Variation, Sources, and Secondary Transformation Potential of VOCs in Xi'an, China.

Response to Reviewer #2

General comments

Understanding the sources of VOCs is vital to the mitigation of O₃ pollution. Song et al. performed comprehensive field observations and VOC grid sampling in Xi'an to elucidate the concentration levels, source characteristics, and secondary conversion ability of VOCs. They found that vehicle exhaust was the dominant source of VOC emissions in Xi'an. This paper has important implications for the control of O₃ pollution in megacities, so it is well within the scope of ACP. I recommend this paper to be published after addressing the comments below.

Response:

We would like to thank reviewer #2 for carefully reading our manuscript and for the valuable and constructive comments. The manuscript was significantly revised according to the reviewer's suggestions. Listed below are our point-by-point responses to reviewer's comments. Lastly, we would like to thank you for your comments and guidance.

Comments

1. Lines 33-38: The authors only present VOC sources for some specific cities. Can the authors summarize the results of previous studies for various regions of China, e.g. Beijing-Tianjin-Hebei, Yangtze River Delta, Pearl River Delta?

Response:

We appreciate the reviewer's comments. After investigating a large amount of literature, we found that different studies have very different PMF source resolution methods, including source numbers and source definition. We categorize the sources of VOCs in these studies into 7 sources, as shown in Figure R4. From the Figure R4 we found that the concentrations of VOCs in urban areas were seriously affected by vehicle exhaust. In addition, southern China was significantly affected by paint solvent usage sources. However, it is difficult to summarize the results for different regions of China.

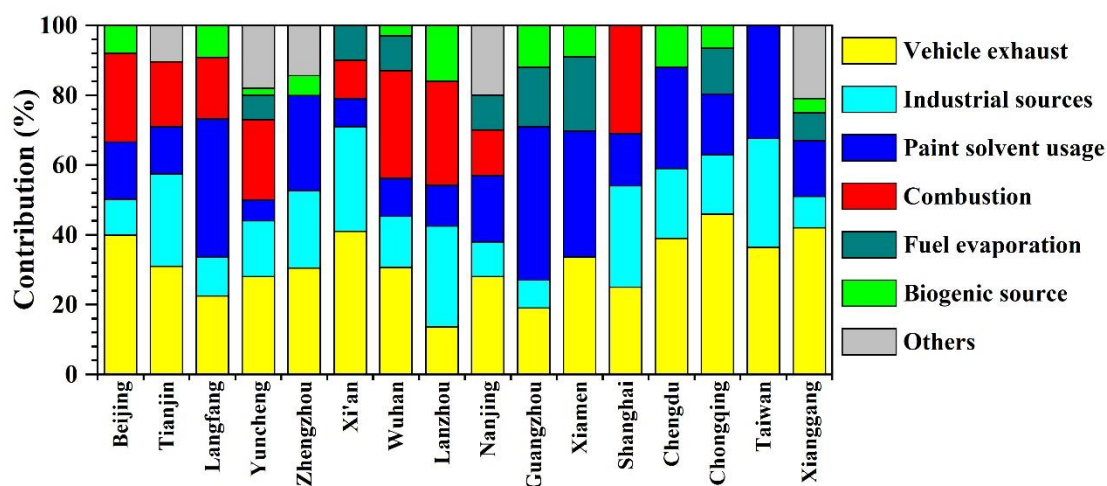


Figure R4. Newly researches about Source apportionment of VOCs at 16 sites in China

Note: The 16 sites include Beijing (Li et al., 2020), Tianjin (Yang et al., 2019), Langfang (Zhang et al., 2019a), Yuncheng (Gao et al., 2020), Zhengzhou (Zhang et al., 2019b), Xi'an (This study), Wuhan (Shen et al., 2020), Lanzhou (Zhou et al., 2019), Nanjing (An et al., 2017), Guangzhou (Ling et al., 2011), Xiamen (Zhuang et al., 2019), Shanghai (Wang et al., 2013), Chengdu (Song et al., 2018), Chongqing (Li et al., 2018), Taiwan (Chen et al., 2019), Hong Kong (Ling and Guo, 2014).

2. Lines 78-79: How often were external standards run? Please provide more details.

Response:

We appreciate the reviewer's comments. We agree with reviewer and carefully revised the manuscript according to the reviewer's suggestion. Now it reads as follows:

External and internal standard gas produced by The Linde Group in the United States were used to calibrate the GC-MS/FID. External standard gas were used to calibrate the GC-MS/FID weekly during the campaign to ensure quantitative accuracy. In addition, the instruments were also daily calibrated by internal standard gases (Bromochloromethane, 1,4-Dichlorobenzene, Chlorobenzene, and Fluorobromobenzene) to ensure the stability of the instrument.

3. Lines 88-89: What's the duration for sampling? Did the authors also test blank samples?

Response:

This study uses instantaneous sampling to collect air samples, and the sampling time is about 2 min. In addition, blank sample tests were performed on the VOC gridded sampling in each sampling period (July 1 7:00, July 1 15:00, July 14 7:00, and July 14 15:00). The concentration of all VOCs species in the blank sample is below the method detection limits (MDLs), indicating that the canisters were not contaminated during transportation.

We appreciate the reviewer's comments. We have added the above sampling information in the revised manuscript. Now it reads as follows:

Before VOC gridded sampling, the SiloniteTM canisters were cleaned with high purity nitrogen using the Entech 3100 canister cleaning system, and then they were evacuated to a vacuum. Instantaneous sampling method was adopted for ambient air sample collection with a sampling duration of approximately 2min. VOCs in the sampled air were analyzed using a GC-MS/FID system, which was the same as that

used for online measurements, but it was running in off-line mode. In this study, blank sample tests were performed on the VOC gridded sampling in each sampling period (July 1 7:00, July 1 15:00, July 14 7:00, and July 14 15:00). The concentration of all VOCs species in the blank sample is below the MDLs, indicating that the canisters were not contaminated during transportation.

4. Lines 158-159: It is recommended to provide the VOC list in the supporting information.

Response:

We appreciate the reviewer's comments. We agree with reviewer and added VOC list in the supporting information. Now it reads as follows:

During the field observation campaign, 99 VOCs were measured, including 29 alkanes, 11 alkenes, 1 alkyne, 16 aromatics, 28 halohydrocarbons, 13 oxygenated VOCs (OVOCs), and 1 acetonitrile (Table S1).

Table S1: Measured VOC species in CB, DHS, and QL sites.

Classification	VOC Species		
Alkanes	Ethane	3-Methylpentane	Methylcyclohexane
	Propane	n-Hexane	2,3,4-Trimethylpentane
	Iso-butane	2,4-Dimethylpentane	2-Methylheptane
	n-Butane	Methylcyclopentane	3-Methylheptane
	Cyclopentane	2-Methylhexane	n-Octane
	Iso-pentane	Cyclohexane	n-Nonane
	n-Pentane	2,3-Dimethylpentane	n-Decane
	2,2-Dimethylbutane	3-Methylhexane	n-Undecane
	2,3-Dimethylbutane	2,2,4-Trimethylpentane	n-Dodecane
	2-Methylpentane	n-Heptane	
Alkenes	Ethene	Cis-butene	Isoprene
	propene	1,3-Butadiene	cis-2-Pentene
	Trans-2-butene	1-Pentene	1-Hexene
	1-Butene	trans-2-Pentene	
Alkynes	Ethyne		
Aromatics	Benzene	iso-Propylbenzene	1,2,4-Trimethylbenzene
	Toluene	n-Propylbenzene	1,2,3-Trimethylbenzene
	Ethylbenzene	m-ethyltoluene	m-diethylbenzene
	m/p-Xylene	p-ethyltoluene	p-diethylbenzene
	o-Xylene	1,3,5-Trimethylbenzene	
	Styrene	o-Ethyltoluene	
Halohydro-carbons	Freon114	cis-1,2-Dichloroethylene	1,1,2-trichloroethane
	Chloromethane	Chloroform	Tetrachloroethene
	Vinylchloride	1,1,1-Trichloroethane	1,2-Dibromoethane
	Bromomethane	Tetrachloromethane	Chlorobenzene
	Chloroethane	1,2-Dichloroethane	1,3-Dichlorobenzene
	Freon11	Trichloroethylene	1,4-Dichlorobenzene
	1,1-Dichloroethene	1,2-Dichloropropane	Benzylchloride

	Freon113	Bromodichloromethane	1,2-Dichlorobenzene
	Dichloromethane	trans-1,3-Dichloropropene	
	1,1-Dichloroethane	cis-1,3-Dichloropropene	
OVOCs	Acetaldehyde	Methyl Vinyl Ketone	3-Pentanone
	Acrolein	Methyl Ethyl Ketone	n-Hexanal
	Propanal	n-Butanal	MTBE
	Acetone	2-Pentanone	
	Methacrolein	n-Pentanal	
Others	Acetonitrile		

5. Lines 172-173: It is not clear to me how good the correlation between O₃ and temperature is. What is R²? It seems that the correlation is moderate.

Response:

We appreciate the reviewer's comments. From figure R5 we can see, the linear correlations between O₃ and temperature on polluted days ($R_{\text{Pearson}}=0.7$, $R^2=0.5$) is stronger than that on clean days ($R_{\text{Pearson}}=0.5$, $R^2=0.2$) at CB site. The correlation between O₃ and temperature at the DHS site shows similar characteristics, and the linear correlations (R^2) between O₃ and temperature on polluted days and clean days were 0.4 ($R_{\text{Pearson}}=0.7$) and 0.3 ($R_{\text{Pearson}}=0.5$), respectively.

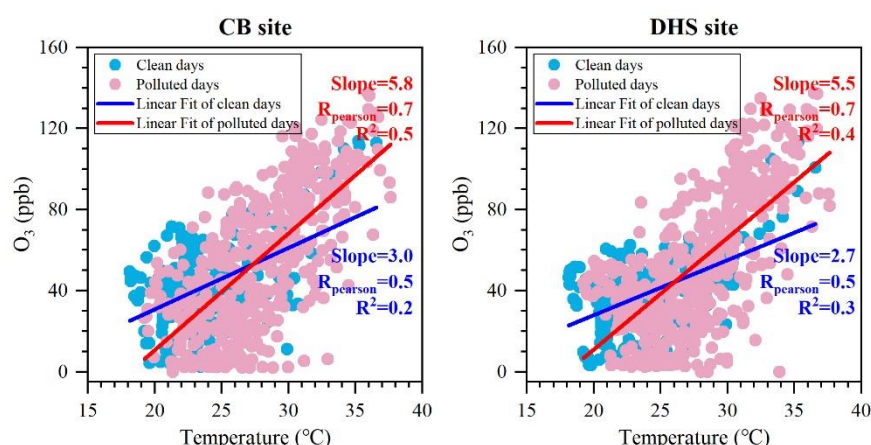


Figure R5: Linear correlations between O₃ and temperature on clean and polluted days.

We also agree with the reviewer that the statement was not very it may not be very appropriate to express a significant positive correlation between O₃ and temperature. We carefully revised the statement according to the reviewer's suggestion. Now it reads as follows:

The variation trends of O₃ and temperature display a positive correlation, and the linear correlations between O₃ and temperature on polluted days ($R_{\text{Pearson}}=0.7$) is stronger than that on clean days ($R_{\text{Pearson}}=0.5$). The value of temperature, O₃, and TVOCs all increased significantly on polluted days, indicating that the secondary transformation of VOCs to O₃ is more conducive at high temperatures.

6. Lines 175-180: The average temperature on polluted days is much higher than that on clean days, which will increase the emission of some VOCs, e.g. isoprene as well as solvent evaporation. As the precursor of MVK and MACR, did the concentrations of isoprene increase on the polluted days?

Response:

That's a very good comments. Because of the low concentration of isoprene, it was discussed as part of the alkene when comparison was considered in the original manuscript. According to the suggestions of reviewers, we analyzed isoprene separately from other alkenes and found that the emission of isoprene increased significantly during the polluted days, along with the increase of temperature (Figure 3c, 3f). However, concentrations of most aromatics which were regarded as tracers of solvent evaporation remained unchanged (Figure 3c, 3f). Therefore, we think there is no clear evidence that the emissions of solvent evaporation increased on the polluted days.

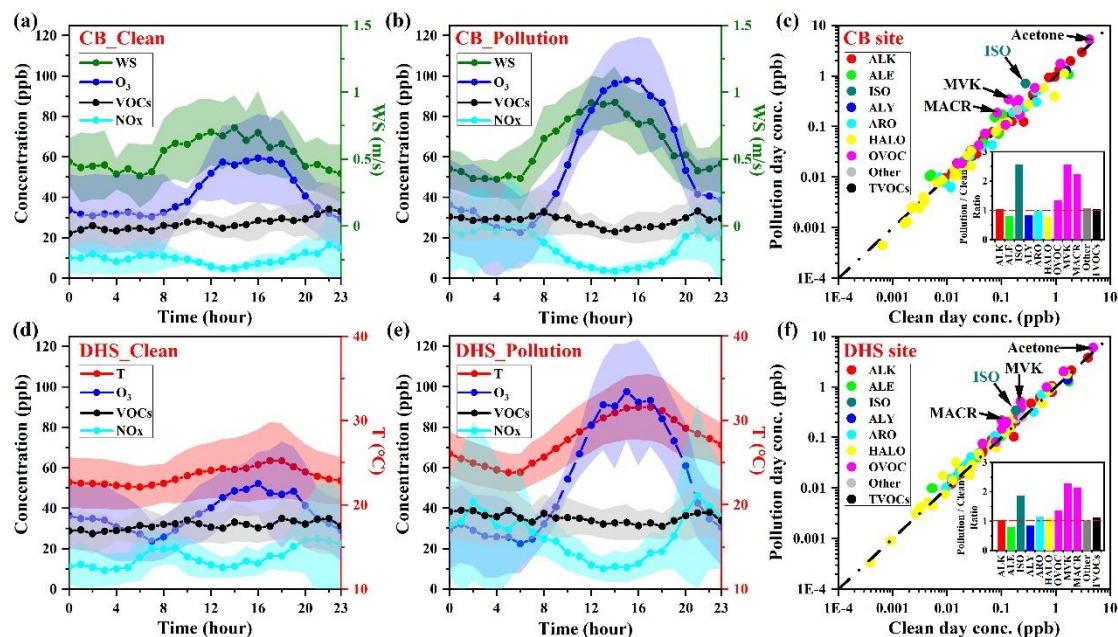


Figure 3: Diurnal variations in wind speed (WS), temperature (T), O₃, NO_x, and TVOCs on clean and polluted days at the (a) and (b) CB and (d) and (e) DHS sites. Differences in VOC concentrations between clean and polluted days at the (c) CB and (f) DHS sites.

Note: ALK = alkanes, ALE = alkenes (except isoprene), ISO = isoprene, ALY = alkynes, ARO = aromatics, HALO = Halohydrocarbons, MVK = Methyl Vinyl Ketone, and MACR = Methacrolein.

We have also modified this part in the revised manuscript. Now it reads as follows:

As shown in Figure 3, isoprene concentrations at urban sites increased significantly during the O₃ pollution day, which could due to the stronger plant emission at elevated temperature (Guenther et al., 1993; Guenther et al., 2012; Stavrou et al., 2014). Concentrations of isoprene oxidation products (i.e., MVK and MACR) as well as most OVOCs also increased in the same period. However, similar concentrations of anthropogenic VOCs are found in clean and polluted days. This indicates a stronger photochemical conversion of VOCs existed in O₃ pollution days, which could due to the more favorable meteorological conditions (i.e., higher temperature and solar radiation).

7. Based on Figure 3, the increase in the concentration of OVOC on O₃ pollution days is largely driven by the increased concentration of acetone. The authors also show in section 3.2.2 that acetone is mainly from vehicle exhaust and industrial sources. Both primary emission and/or secondary transformation may contribute to the increase of OVOC. Can the authors estimate the contribution from primary emissions?

Response:

We appreciate the reviewer's comments. The sources of OVOCs can be divided into anthropogenic primary sources, anthropogenic secondary sources, biogenic sources and background sources (Li et al., 2014; Wang et al., 2015). The multi-linear regression model was used to analyse the sources of OVOCs in different sites in Xi'an. Ethyne, PAN and isoprene were selected as the tracers of the anthropogenic primary source, the anthropogenic secondary source and the biogenic sources respectively. The equation of the multi-linear regression model is as follows:

$$[\text{OVOCs}] = k_0 + k_1 \times [\text{Ethyne}] + k_2 \times [\text{PAN}] + k_3 \times [\text{Isoprene}] \quad (\text{R1})$$

where [Ethyne] represents the concentration of Ethyne, [PAN] represents the concentration of PAN, [Isoprene] represents the concentration of isoprene, k_0 represents the background concentration, k_1 , k_2 and k_3 are the corresponding coefficients.

Based on the analysis of the multi-linear regression model, we have a deeper understanding of the source of OVOCs during the ozone pollution period. From Figure S1 and S2 we found that the contribution of anthropogenic primary sources to OVOCs on O_3 pollution days is more significant.

We have carefully revised this statement in the revised manuscript. Now it reads as follows:

The specialty of OVOCs is that in addition to the primary emissions, OVOCs can also be formed through photochemical oxidation with alkenes and aromatics (Birdsall and Elrod 2011). The sources of OVOCs can be divided into anthropogenic primary sources, anthropogenic secondary sources, biogenic sources and background sources (Li et al., 2014; Wang et al., 2015). Base on the multi-linear regression model results (Figure S1 and S2) we found that the contribution of anthropogenic primary sources to OVOCs on O_3 pollution days is more significant.

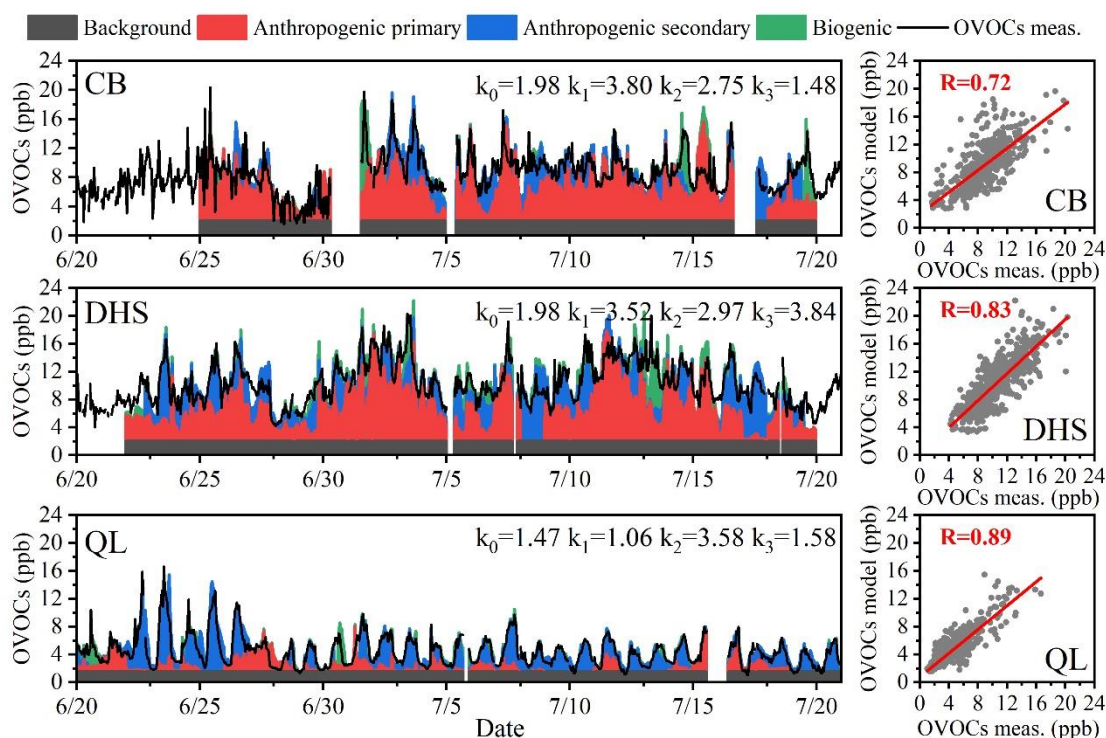


Figure S1. Time series of measured OVOCs concentrations and OVOCs calculated from the multi-linear regression model.

Note. The equation of the multi-linear regression model is:

$$[\text{OVOCs}] = k_0 + k_1 \times [\text{Ethyne}] + k_2 \times [\text{PAN}] + k_3 \times [\text{Isoprene}]$$

where [Ethyne] represents the concentration of Ethyne, [PAN] represents the concentration of PAN,

[Isoprene] represents the concentration of isoprene, k_0 represents the background concentration, k_1 , k_2 and k_3 are the corresponding coefficients, meas. represents measure.

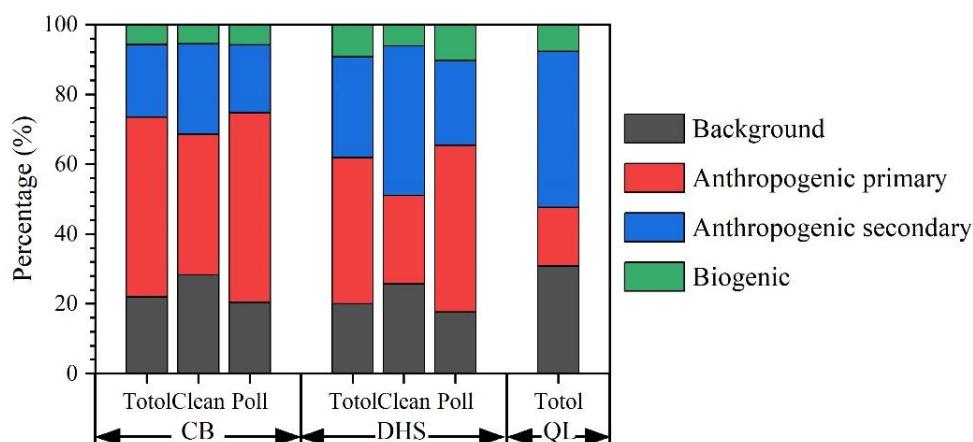


Figure S2. Contributions of different sources of OVOCs in different sites in Xi'an base on the multi-linear regression model.

8. Did the NO_x concentration change during the polluted and clean days? Did NO_x play a role in increasing the O₃ concentration on polluted days?

Response:

We appreciate the reviewer's comments. Compared with clean days, the NO_x concentration at CB and DHS sites on polluted days increased by 5.8 and 10.3 ppb, respectively. However, before analyzing the influence of NO_x changes on ozone, a VOCs-NO_x-O₃ sensitivity analysis is needed. We have added VOCs-NO_x-O₃ sensitivity analysis in section 3.4.2. Now it reads as follows:

The relationship between the ozone production rates ($P(O_3)$), anthropogenic VOCs (AVOCs) reactivity and NO_x reactivity of the CB, DHS, and QL sites during the observation period was shown in Figure 12. The black curve in the Figure 12 represents the $P(O_3)$ contour, and the black straight line represents the connection line of the $P(O_3)$ turning point (ridgeline), whose slope represents the photochemical parameter k_{NOx}/k_{AVOCs} (Jiang et al., 2018). When the site's k_{NOx}/k_{AVOCs} value is located above the ridgeline, it means that ozone formation is under VOCs-limited regime, otherwise it means that ozone formation is under NO_x-limited regime. It can be seen from Figure 12 that the ozone generation of QL site is located in the NO_x-limited regime, and reducing NO_x can effectively control ozone generation. The ozone generation of DHS site is located in the VOCs-limited regime, and reducing VOCs can effectively control ozone generation. However, CB site is located in the transition regime between VOC- and NO_x-limited regimes. Therefore, simultaneous reduction of VOCs and NO_x concentration should be considered at CB site to achieve the purpose of controlling O₃.

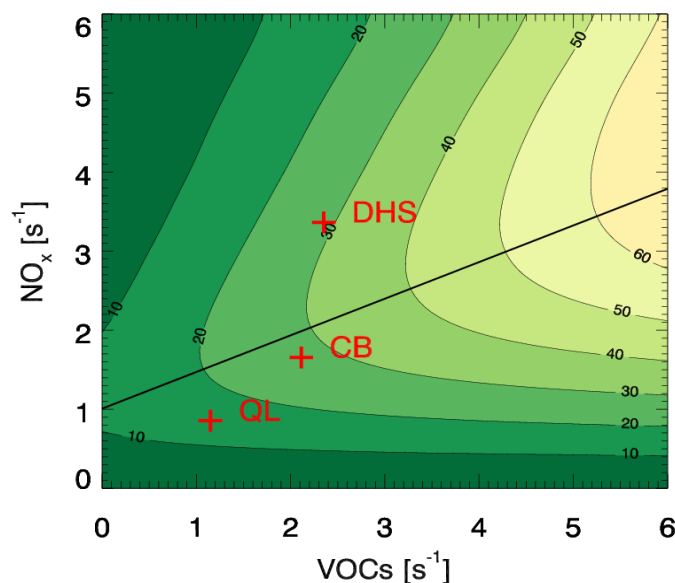


Figure 12. The ozone production rate ($P(O_3)$) contours diagram versus anthropogenic VOCs (AVOCs) and NO_x using Empirical Kinetic Modelling Approach at CB, DHS, and QL sites.

We also have added this part of the results in the conclusion section. Now it reads as follows:

The VOCs-NO_x-O₃ sensitivity analysis results showed that the ozone generation of DHS site is located in the VOCs-limited regime, CB in the transition regime between VOC- and NO_x-limited regimes, and QL sites is located in the NO_x-limited regime. Therefore, reducing VOCs concentration at DHS site, reducing VOCs and NO_x concentration at CB site, and reducing NO_x concentration at QL site can effectively control ozone generation.

9. Line 184: It is recommended to provide the VOC list and the grid sampling data in the supporting information.

Response:

We appreciate the reviewer's comments. We agree with reviewer and added VOC list and the grid sampling data of the grid sampling sites in the supporting information. Now it reads as follows:

In the VOC grid sampling, 106 VOCs were measured, including 29 alkanes, 11 alkenes, 1 alkyne, 17 aromatics, 35 halohydrocarbons, 12 OVOCs, and carbon disulfide (Table S2-S3).

Table S2: Measured VOC species in VOC grid sampling sites.

Classification	VOC Species		
Alkanes	Ethane	3-Methylpentane	Methylcyclohexane
	Propane	n-Hexane	2,3,4-Trimethylpentane
	Iso-butane	2,4-Dimethylpentane	2-Methylheptane
	n-Butane	Methylcyclopentane	3-Methylheptane
	Cyclopentane	2-Methylhexane	n-Octane
	Iso-pentane	Cyclohexane	n-Nonane
	n-Pentane	2,3-Dimethylpentane	n-Decane
	2,2-Dimethylbutane	3-Methylhexane	n-Undecane
	2,3-Dimethylbutane	2,2,4-Trimethylpentane	n-Dodecane

	2-Methylpentane	n-Heptane	
Alkenes	Ethene	Cis-butene	Isoprene
	propene	1,3-Butadiene	cis-2-Pentene
	Trans-2-butene	1-Pentene	1-Hexene
	1-Butene	trans-2-Pentene	
Alkynes	Ethyne		
Aromatics	Benzene	iso-Propylbenzene	1,2,4-Trimethylbenzene
	Toluene	n-Propylbenzene	1,2,3-Trimethylbenzene
	Ethylbenzene	m-ethyltoluene	m-diethylbenzene
	m/p-Xylene	p-ethyltoluene	p-diethylbenzene
	o-Xylene	1,3,5-Trimethylbenzene	Naphthalene
	Styrene	o-Ethyltoluene	
Halohydrocarbons	Freon114	Tetrachloromethane	Bromodichloromethane
	Freon11	1,2-Dichloroethane	Benzyl chloride
	Freon113	Trichloroethylene	1,2-Dichlorobenzene
	Chloromethane	1,2-Dichloropropane	trans-1,2-Dichloroethylene
	Vinylchloride	cis-1,3-Dichloropropene	Freon12
		trans-1,3-	
	Bromomethane	Dichloropropene	Trichloromethane
			Dibromo-monochloro-
	Chloroethane	1,2,4-trichlorobenzene	methane
	1,1-Dichloroethene	1,1,2-trichloroethane	Tribromomethane
	Dichloromethane	Tetrachloroethene	1,1,2,2-tetrachloroethane
	1,1-Dichloroethane	1,2-Dibromoethane	1,3-Dichlorobenzene
	cis-1,2-		
	Dichloroethylene	Chlorobenzene	Hexachloro-1,3-butadiene
	1,1,1-Trichloroethane	1,4-Dichlorobenzene	
OVOCs	Acrolein	2-Propanol	Tetrahydrofuran
	Methyl isobutyl		
	ketone	Vinyl Acetate	Methyl methacrylate
	Acetone	Methyl Ethyl Ketone	1,4-Dioxane
	MTBE	Ethyl Acetate	2-Hexanone
Others	Carbon Disulfide		

Table S3: Concentrations of seven VOCs groups in VOC grid sampling sites.

site	Alkanes	Alkenes	Alkynes	Aromatics	Halohydrocarbons	OVOCs	Others	TVOCs
XF	19.2	8.4	2.3	4.4	4.9	14.6	0.2	54.0
CT	16.1	3.6	1.8	3.3	5.1	11.5	0.1	41.4
HC	15.2	2.4	1.2	2.8	4.5	12.1	0.1	38.2
ZYT	13.7	5.0	1.8	3.0	2.0	9.7	0.1	35.4
YT	8.7	3.3	1.1	1.1	3.4	16.1	0.2	33.9
XS	12.3	2.4	1.2	2.6	3.5	10.5	0.2	32.6
GYL	10.1	2.5	1.3	2.5	3.6	9.8	0.1	29.9

LTC	10.5	4.8	2.3	2.4	2.4	6.7	0.1	29.4
XY	11.6	2.7	1.5	1.7	3.4	8.3	0.1	29.1
JFT	9.8	2.8	1.1	1.4	2.5	10.6	0.1	28.2
RS	10.6	2.4	1.4	1.3	3.5	8.6	0.1	27.9
LT	8.7	3.2	0.9	0.9	3.4	10.5	0.2	27.7
GZ	10.1	2.6	1.4	1.7	3.6	7.2	0.1	26.6
ZZC	9.8	2.0	1.2	1.2	2.5	8.3	0.1	25.0
JDT	7.5	2.9	1.1	1.3	2.9	7.0	0.1	22.8
CAT	7.3	2.3	0.8	0.9	2.7	7.9	0.1	22.2
XX	8.0	1.8	1.3	0.9	2.9	5.5	0.1	20.6
YST	5.8	1.9	1.1	1.3	2.7	7.0	0.1	19.9
WQ	7.3	0.9	0.8	0.6	3.1	6.8	0.1	19.6
XHT	5.9	1.7	1.0	0.9	2.1	6.3	0.1	18.0

10. Line 190: Did “the overall level” mean “the average concentration”?

Response:

We appreciate the reviewer’s comments. We apologize for the vague expression and carefully revised the statement in the revised manuscript. Now it reads as follows:

Compared with the results of the field observation campaign, the VOC concentration at the CB site was closer to the average concentration in Xi'an, and the VOC concentration at the DHS site was significantly higher than the average concentration.

11. Lines 192-194: Are there any specific industrial sources near YT? Again, the contribution of primary emissions to OVOCs should be excluded to draw this conclusion.

Response:

We appreciate the reviewer’s comments. Based on the analysis of the multi-linear regression model, the anthropogenic secondary sources are more significant for OVOCs in rural sites. The YT site is a rural site and does not have many primary sources of VOCs (Figure R2), so the source of OVOCs is more from aging sources.



Figure R2. Geographic environment map of TY site.

12. Lines 209-210: These numbers are the slopes of the fitting lines, not the correlation coefficients. Please also revise other places accordingly.

Response:

We appreciate the reviewer's comments. We agree with reviewer and carefully revised the statement according to the reviewer's suggestion. Now it reads as follows:

The ratios of toluene to benzene at the CB, DHS, QL, and gridded sampling sites were 0.7 ($R_{\text{Pearson}}=0.5$), 2.3 ($R_{\text{Pearson}}=0.6$), 0.5 ($R_{\text{Pearson}}=0.6$), and 1.2 ($R_{\text{Pearson}}=0.9$), respectively.

13. Lines 215-217: For the grid sampling, only samples at 7:00 and 15:00 were collected. It is reasonable that vehicle exhaust greatly contributed to the overall VOCs because of the sampling time. The authors should state the weakness of the sampling as a caveat.

Response:

We appreciate the reviewer's comments. We agree with reviewer that the sample time may affect the ratio of T/B. In order to reduce the influence of sample time and photochemical reaction on the ratio of benzene to toluene, this study selected the weaker photochemical reaction period (3:00-7:00) for the analysis of toluene and benzene (Figure 5).

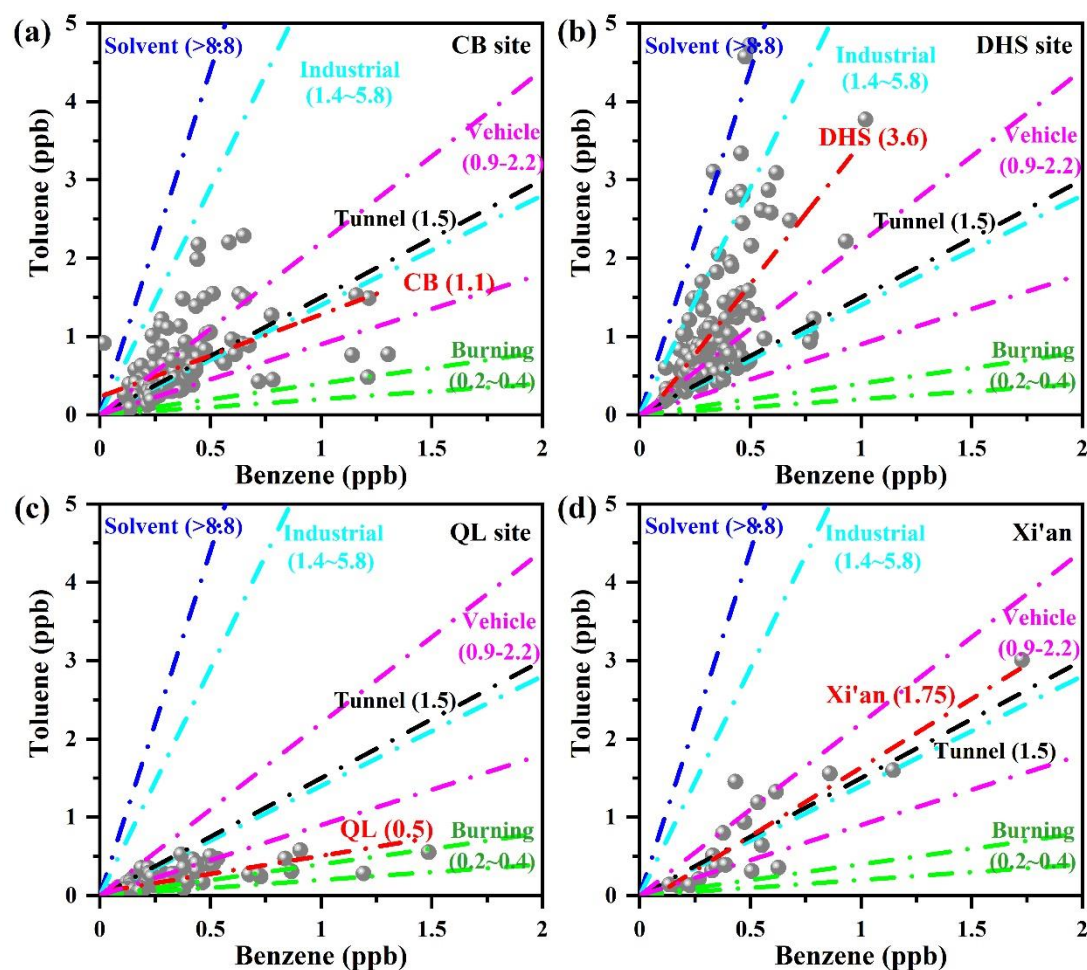


Figure 5: Linear correlations between toluene and benzene at the CB, DHS, QL, and gridded sampling sites between 3:00-7:00 during the observation period.

We agree with reviewer and carefully revised the statement in the revised manuscript. Now it reads as follows:

In order to reduce the influence of photochemical reaction on the ratio of benzene to toluene, this study selected the weaker photochemical reaction period (3:00-7:00) for the analysis of toluene and benzene (Figure 5). Figure 5 shows that the ratios of toluene to benzene at the CB, DHS, QL, and gridded sampling sites were 1.1 ($R_{\text{Pearson}}=0.5$), 3.6 ($R_{\text{Pearson}}=0.6$), 0.5 ($R_{\text{Pearson}}=0.8$), and 1.75 ($R_{\text{Pearson}}=0.9$), respectively. In the urban areas (CB and DHS sites), most of the T/B ratios were distributed within the reference range of vehicle emissions and industrial emissions (Figure 5a, 5b), implying that vehicle sources and industrial sources contribute significantly to the VOCs in Xi'an urban area. In addition, the T/B value of some samples is greater than 5.8 in urban area which may be affected by paint solvent usage source (Figure 5b). However, the detailed source contribution needs to be obtained through PMF source analysis results (Section 3.2.2). In the rural area (QL site), most of the T/B ratios were distributed within the reference range of vehicle emissions and burning emissions (Figure 5c), implying that vehicle sources and burning sources contribute significantly to the VOCs in Xi'an rural area. In the gridded sampling sites, the T/B ratio was predominately concentrated around 1.5, indicating that vehicle exhaust sources may greatly contribute to the overall VOCs in Xi'an (Figure 5d).

14. MVK is a photochemical product of isoprene. Why are most of the MVK attributed to vehicle exhaust at the CB site?

Response:

We appreciate the reviewer's comments. As the photooxidation product of isoprene, MVK could be emitted from secondary sources. However, the tunnel and vehicular exhaust emissions measurements have indicated that MVK could be emitted from vehicle exhaust sources including gasoline and diesel vehicles (Biesenthal and Shepson, 1997; Kean et al., 2001; Ling et al., 2019). In urban site with less affected by biological sources, MVK may be more affected by anthropogenic emissions

15. Figure 1: it is recommended to mark Feiwei plain.

Response:

We appreciate the reviewer's comments. We agree with reviewer and marked the Feiwei plain in Figure 1. Now it reads as follows:

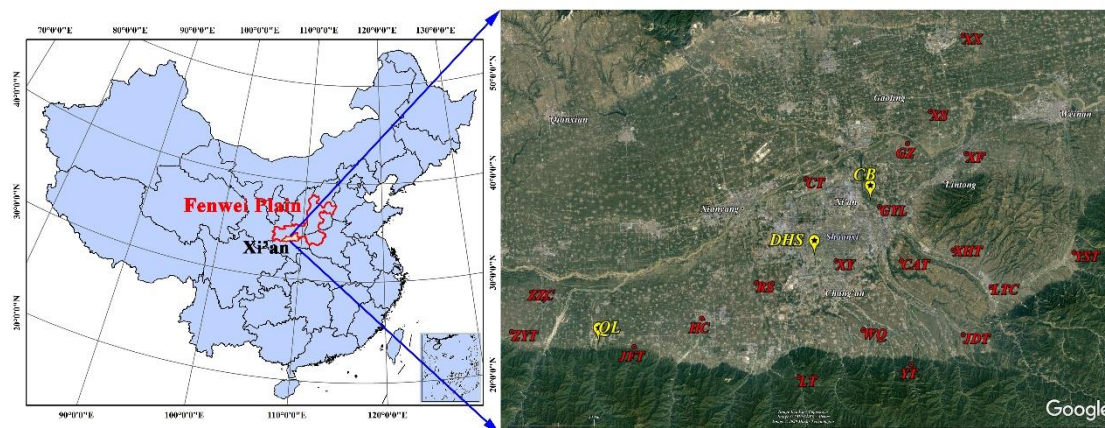


Figure 1: VOC field observation and grid sampling sites in Xi'an.

Note: The topographic image was obtained from Google Earth.

16. Figures 5 and 7: Linear correlations are shown, not correlation coefficients.

Response:

We appreciate the reviewer's comments. We agree with reviewer and carefully revised the statement according to the reviewer's suggestion. Now it reads as follows:

Figure 5: Linear correlations between toluene and benzene at the CB, DHS, QL, and gridded sampling sites.

Figure 7: Linear correlations between (a) iso-pentane and n-pentane and (b) propane and ethane at the CB (red), DHS (orange), QL (blue), and gridded sampling sites (light blue).

17. Figure 6: What does the green line represent?

Response:

We appreciate the reviewer's comments. The green line represents the initial emission ratio of m/p-xylene and ethylbenzene, which can be replaced by the highest concentration ratio in periods where the photochemical reaction is weak.

We have added the green line description in the caption in the revised manuscript. Now it reads as follows:

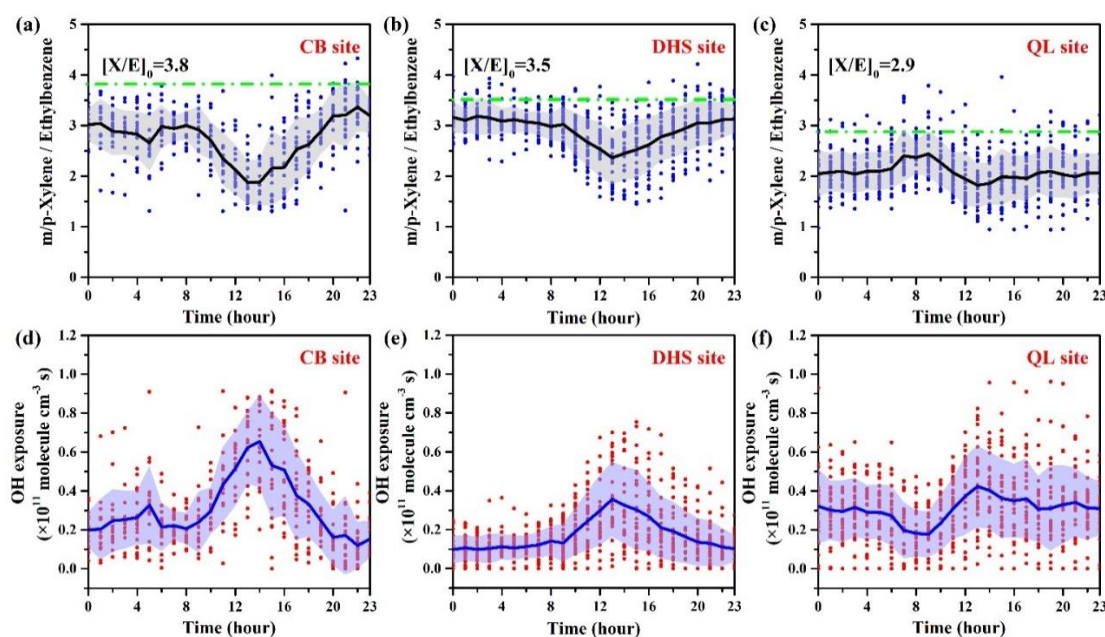


Figure 6: Diurnal variations in m/p-xylene to ethylbenzene and OH exposure at the CB, DHS, QL, and gridded sampling sites.

Note: Time is expressed in CST. The green line represents the initial emission ratio of m/p-xylene and ethylbenzene.

18. Figure 8: What do the bars and dots represent? Please explain in the caption.

Response:

We appreciate the reviewer's comments. We agree with reviewer and add the explanation in the caption.

Now it reads as follows:

Figure 8: Source profiles and contributions of VOCs in the CB, DHS, and QL sites during the observation period.

Note. Bars represent the concentration of each species apportioned to the factor, and black dots represent the percent of each species apportioned to the factor.

Technical comments

1. Line 10: “a critical precursors”... delete “a”.

Response:

We appreciate the reviewer’s comments. We are sorry for our mistakes, and we have removed “a” in this sentence.

2. Line 23: References are missing.

Response:

We appreciate the reviewer’s comments, and we have added references in this sentence in the revised manuscript. Now it reads as follows:

Atmospheric pollution in China is characterized by frequent secondary pollution, which is primarily reflected by the yearly increasing ozone (O₃) concentrations and proportion of secondary organic components (SOA) in PM_{2.5} (Lu et al., 2018; Huang et al., 2014).

3. Line 35: “source” should be “sources”

Response:

We appreciate the reviewer’s comments. We are sorry for our mistakes, and we have change “source” to “sources” in this sentence.

4. Line 36: “indicates” should be “indicate”

Response:

We appreciate the reviewer’s comments. We are sorry for our mistakes, and we have change “indicates” to “indicate” in this sentence.

5. Equation 1, lines 98 and 100: the rate coefficient is typically represented by the lowercase k.

Response:

We appreciate the reviewer’s comments. We are sorry for our mistakes, and we have change the rate coefficient K by the lowercase k in the revised manuscript.

Lastly, we would again express our appreciation to the reviewers and editor for their warm-hearted help. Thank you very much!!!!

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