#### Anonymous Referee #1

This work present the results of the field measurement of air pollution during 2019 Spring Festival. Spring Festival is a special period to investigate the impact of emission reduction on air quality. The topic itself is very interesting. The authors provide very interesting data. However, there's major defect for the current manuscript. The authors compared the variation of various air pollutants, and gave the conclusion that the reduction of "Sharply reduced sulfur dioxide (SO2) and nitrogen dioxide (NO2) concentrations during the festival holiday resulted in an unexpected increase in the surface ozone (O3) concentration", and further promote the secondary formation. These conclusions are astonishing and new, but the authors did not provide enough convincing evidence. Besides, considering the quality of ACP, I will not recommand the publication of this manuscript.

**Response:** While we appreciate the critical comment of the review, it'd be much more helpful if the reviewer could have provided a more informative and insightful comment so that we know more about his/her concern. For any scientific research, a finding of "astonishing and new" should not be the reason for rejection.

In this study, we investigated the impact of emission reductions on the concentrations of several trace gases and their further impact on aerosol formation during the special period of the 2019 Spring Festival. It is clear that emission reductions could efficiently reduce primary pollutants (SO<sub>2</sub>, NO<sub>x</sub>, BC, etc.). The time series of O<sub>x</sub> (O<sub>3</sub>+NO<sub>2</sub>) added in Fig. 2 depicts a weak decrease of O<sub>x</sub> from the POL period to the BG period, suggesting that the possible appeared O<sub>3</sub>-titration made [O<sub>3</sub>] increase during the BG period. Simultaneously, the mass concentrations of secondary inorganics decreased but their reduction percentages were much lower than those of primary pollutants. With the further analysis of SOR, NOR, and their relationships with ambient RH and ALWC, we concluded that the enhancement of aqueous chemical reactions oxidized by the dissolved O<sub>3</sub> maybe the main reason causing the enhanced secondary inorganic aerosol (SIA) formation, especially for sulfate.

In recent years, the annual average  $PM_{25}$  concentration has decreased rapidly in China, benefitting from the implementation of many emission reduction measures taken by the Chinese government. However, the mass fraction of inorganics increased by more than 10 % during these years (H. Li et al., 2019; Y. Wang et al., 2019), implying the formation enhancement of secondary inorganic aerosols (SIA), which partly counteracted the decrease in  $PM_{25}$ . Xie et al. (2020) found that the aerosol pH level increased as  $PM_{25}$  decreased in urban Beijing because of the increased mass ratio of nitrate to sulfate. They also stated that the major chemical processes during haze events and the control target should be re-evaluated to obtain the most effective control strategy. As one possible consequence of the increased aerosol pH, the dissolved O<sub>3</sub> in particles may play a more important role in SIA formation, especially for sulfate (Seinfeld and Pandis, 2016). Our study provides sound evidence for this.

Other recent studies have also suggested that the role of  $O_3$  on SIA formation cannot be neglected. For example, Fang et al. (2019) found that relative humidity (RH) and  $O_3$  concentration were two important prerequisites for sulfate formation, based on a year-long set of field measurements made in Beijing. They found a rapid rise in the SOR at the RH

threshold of ~ 45% or an O<sub>3</sub> concentration threshold of ~35 ppb, similar to what we found in our study. As another example, Huang et al. (2020) investigated air quality during the COVID-19 lockdown using comprehensive measurements and modeling with a focus on China. They also found that a large reduction in emissions could enhance the concentration of O<sub>3</sub> in winter in the Beijing-Tianjin-Hebei (BTH) region, promoting SIA formation through the enhancement of nocturnal aqueous chemical reactions during the COVID-19 lockdown.

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### Anonymous Referee #2

Comments on manuscript entitled "Enhancement of secondary aerosol formation by reduced anthropogenic emissions during Spring Festival 2019 and enlightenment for regional PM2.5 control in Beijing"

General comments: This manuscript reported primary pollutant reduction but enhanced SIA formation in an emission reduction period during the 2019 Spring Festival in Beijing. The opposite trend of atmospheric oxidative capacity responding emission reduction was proposed the cause for enhanced SIA formation. Though the supporting discussion still appeared to be weak. Nevertheless, this study should call for the attention on SIA pollution control policy mitigation. I thus recommend publication of this manuscript on ACP with minor revision.

### Specific comments:

Lines 21-22: O3 control regime on a regional scale is still a controversial topic. This manuscript did not intend to discuss on such topic given no VOCs measurements were present. Therefore, it is a bit risky to go such far with current data available. I suggest to delete the statement on NOx and VOCs control strategy if no more discussion shall add. **Response:** Agreed. The sentence "The emission control of volatile organic compounds (VOCs) may be more suitable than the emission control of NO<sub>x</sub> to reduce O<sub>3</sub> because VOCs under current emission conditions likely control the formation of O<sub>3</sub> in winter in the BTH region" has been deleted.

# Line 56: be specific! Change to "secondary inorganic aerosol formation" **Response:** Revised.

### Line 92: consider to revise this sentence

**Response:** This sentence has been revised as: "Our measurements around this period of the field campaign are thus ideal for investigating the impact of reduced anthropogenic emissions on surface  $O_3$  and aerosol formation.".

# Line 160: as shown in Fig.2, O3 titration appeared to occur in both POL and BG period. Ox=O3+NO2 is thus suggested to add in Fig.2.

**Response:** The time series of  $O_x$  ( $O_3$ +N $O_2$ ) has been added to Fig. 2 in the manuscript (shown below as Fig. R1). It shows a weak variation of  $O_x$  from the POL period to the BG period, indicating that the presence of strong  $O_3$ -titration during Spring Festival 2019. The corresponding discussion about  $O_x$  and  $O_3$  titration has been added to section 3.1 of the revised manuscript.

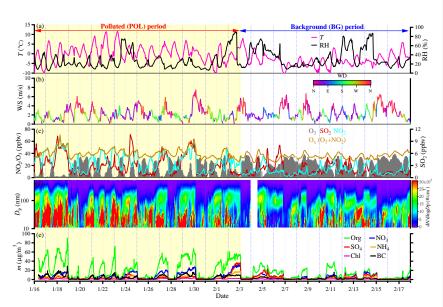


Figure R1. Time series of (a) ambient temperature (T) and relative humidity (RH), (b) wind direction (WD) and speed (WS), (c) volume mixing ratios of trace gases  $[O_3, SO_2, NO_2 \text{ and } O_x (O_3+NO_2)]$ , (d) the aerosol particle number size distribution measured by the SMPS, and (e) mass concentrations of aerosol chemical species in PM<sub>25</sub> measured by the ACSM and the AE-33. The trace gas information was from the Yizhuang station, and the others were observed at the experiment site in Beijing (16 January to 17 February 2019).

# Lines 202-203: cannot read from Figure 2 that morg and mBC increase by % at night from daytime is less in BG relative to POL Lines

**Response:** We're sorry that this sentence has confused the reviewer. Figure 2e depicts that the peaks of  $m_{\rm org}$  and  $m_{\rm BC}$  at night during the BG period were much lower than those during the POL period, caused by emission reductions during the BG period. Therefore, here we want to express that the enhancement of  $m_{\rm org}$  and  $m_{\rm BC}$  at night during the BG period was not as strong as that during the POL period.

This sentence has been revised as: "However, the increases in  $m_{org}$  and  $m_{BC}$  at night during the BG period were not as strong as those during the POL period.".

### 203-204: both mnitrate and msulfate varied!

**Response:** Figure 2e depicts that both  $m_{NO3}$  and  $m_{SO4}$  decreased from the POL period to the POL period. However, their reduction magnitudes differed considerably.

### Lines 246-24: decreased from what?

**Response:** This sentence has been revised as:" Table 2 also indicates that the mass concentrations (*m*) of aerosol chemical species in PM<sub>2.5</sub> were much less during the BG period than during the POL period".

Lines 255-257: From the context, I can only get that Org and BC reduction was sharper than sulfate and nitrate. If I can accept that "secondary (inorganic) aerosol" could replace "sulfate and nitrate", I am still reluctant to accept that Org and BC are all primary aerosol. **Response:** The reviewer asks a good question. BC is mainly from primary emissions, but organics were not. Part of the organics is from primary emissions (i.e., primary organic aerosols, or POA), but another part is from gas-to-particle transformations (i.e., secondary organic aerosols, or SOA). Unfortunately, we are not able to separate POA and SOA in organics using our measurement data from the campaign. In this paper, we were not trying to define BC and organics as the primary matter. They were simply regarded as representing primary matter because many of them were from primary emissions. To a certain extent, the mass variations of BC and organics can represent the mass variations of primary aerosols. Similarly, SIA matter (mainly sulfate, nitrate and ammonium) are important chemical components of secondary aerosols, so their mass variations can represent the mass variations of secondary aerosols.

# Figure 5: The high SIA and large PM2.5 number in POL were mostly seen at low RH, which is against the impression that heavy PM2.5 pollution was usually accompanied by high RH condition in literature. The author should at least address such unusual data.

**Response:** Some studies have found that heavy haze events are generally associated with high RH conditions and southerly winds. This is because the southerly winds are not only beneficial to the transport of pollutants from southern highly industrialized areas, but also to the transport of water vapor. In our study, the prevailing winds during both the POL and BG periods were northerly, which were beneficial to dispersing pollutants in Beijing, so no heavy haze episodes occurred during the two periods. However, the PM<sub>25</sub> during the POL period with ordinary emission conditions could reach moderate pollution level (over 100  $\mu$ g/m<sup>3</sup>) although the ambient RH was low.

The basic meteorological and environmental characteristics have been described in section 3.1.

# Figure 6: Given the discussion on RH or ALWC in the context, I would suggest to add one of the two parameters in one column.

**Response:** That is a good suggestion. A figure showing the diurnal variation in ambient relative humidity (RH) (Fig. R2 below) was added to the supplement (Fig. S4). It shows that the ambient RH levels at night are elevated during both the POL and BG periods, favorable for aqueous chemical reactions.

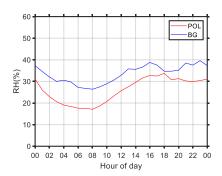


Figure R2. Diurnal variation of ambient relative humidity (RH) during the POL and BG periods.

Line 403: High O3 concentration itself will not surely lead to strong atmospheric oxidative capacity or even O3 production. The first reason is that O3 was titrated in Figure 2. The secondary reason is that O3 can be regionally transported as a relatively long-lived species. And the third, OH instead of O3 is the major oxidant in the atmosphere, which better represents the atmospheric oxidative capacity and does not differ significantly from pollution days to clean days in winter Beijing (see Eloise et al., Elevated levels of OH observed in haze events during wintertime in central Beijign). More data or discussion are needed here.

**Response:** Agreed. The analysis of  $O_x$  above shows that  $O_3$ -titration appeared during the special period studied. In this campaign, OH was not measured, so the atmospheric oxidation capacity wasn't analyzed accurately. For this reason, the discussion about atmospheric oxidation capacity in the manuscript has been deleted.

6

# Enhancement of secondary aerosol formation by reduced anthropogenic emissions during Spring Festival 2019 and enlightenment for regional PM<sub>2.5</sub> control in Beijing

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### 1 Abstract

A comprehensive field experiment measuring aerosol chemical and physical 2 properties at a suburban site in Beijing around the 2019 Spring Festival was carried out 3 to investigate the impact of reduced anthropogenic emissions on aerosol formation. 4 Sharply reduced sulfur dioxide (SO2) and nitrogen dioxide (NO2) concentrations during 5 the festival holiday resulted in an unexpected increase in the surface ozone (O<sub>3</sub>) 6 7 concentration caused by the strong O3-titration phenomenon, leading to enhancement 8 of the atmospheric oxidation capacity. Simultaneously, the reduced anthropogenic emissions resulted in massive decreases in particle number concentration at all sizes 9 10 and the mass concentrations of organics and black carbon. However, the mass concentrations of inorganics (especially sulfate) decreased weakly. Detailed analyses 11 of the sulfur oxidation ratio and the nitrogen oxidation ratio suggest that sulfate 12 formation during the holiday could be promoted by enhanced nocturnal aqueous-phase 13 14 chemical reactions between SO<sub>2</sub> and O<sub>3</sub> under moderate relative humidity (RH) 15 conditions (40 % < RH < 80 %). Daytime photochemical reactions in winter in Beijing mainly controlled nitrate formation, which was enhanced a little during the holiday. A 16 17 regional analysis of air pollution patterns shows that the enhanced formation of secondary aerosols occurred throughout the entire Beijing-Tian-Hebei (BTH) region 18 during the holiday, partly offsetting the decrease in particle matter with an aerodynamic 19 diameter less than 2.5 µm. Our results highlight the necessary control of O3 formation 20 21 to reduce secondary pollution in winter under current emission conditions. The 22 emission control of volatile organic compounds (VOCs) may be more suitable than the

23 emission control of NO<sub>\*</sub> to reduce O<sub>3</sub> because VOCs under current emission conditions

- 24 likely control the formation of O<sub>3</sub> in winter in the BTH region.
- 25

## 26 1. Introduction

27 Aerosols consist of liquid and solid particles, and their mixture suspended in the 28 atmosphere. The massive increase in aerosol particles caused by human activities (e.g., 29 traffic, industrial production, and construction work) in urban areas (e.g., traffic, 30 industrial production, and construction work) can deteriorate air quality to the point of having a detrimental impact on human health (e.g., Chow et al., 2006; Matus et al., 31 32 2012; Gao et al., 2017; Zhong et al., 2018; An et al., 2019). Moreover, aerosols can change atmospheric optical and hygroscopic properties, altering the transfer of solar 33 radiation and the development of clouds, thereby changing weather and climate in both 34 aerosol source regions and their downstream areas (e.g., Altaratz et al., 2014; R. Zhang 35 36 et al., 2015; Z. Li et al., 2016, 2019; Y. Wang et al., 2018, 2019b; Jin et al., 2020).

37 With the rapid economic development and urbanization in recent decades in China, the scales of many cities have expanded quickly along with sharply increased 38 39 populations in urban areas, especially in the three most economically developed regions (the Beijing-Tianjin-Hebei (BTH) metropolitan region, the Yangtze River Delta, and 40 the Pearl River Delta). As a result, air pollution has become a severe problem in these 41 megacity regions (e.g., Chan and Yao, 2008; Han et al., 2014; Zhong et al., 2018). On 42 some heavy haze days, the mass concentration of particulate matter with an 43 44 aerodynamic diameter of less than 2.5 µm (PM2.5) dramatically increased from tens to 45 hundreds of micrograms per cubic meter in several hours (Guo et al., 2014; Sun et al.,

46 2016a).

47	Over the past a few years, many emission control measures have been taken in
48	China to mitigate air pollution. As a response, the mass concentration of $PM_{2.5}$ has
49	decreased in most cities in China since 2013, especially in the BTH region (Q. Zhang
50	et al., 2019; Vu et al., 2019; Zhai et al., 2019). Organics and black carbon (BC)
51	concentrations largely decreased during these years thanks to the reduction in coal
52	combustion and biomass burning (H. Li et al., 2019a; Xu et al., 2019). Simultaneously,
53	the mass concentrations of inorganics (mainly sulfate, nitrate, and ammonium) also
54	decreased due to the reduction in their gaseous precursors (especially sulfur dioxide, or
55	$SO_2$ ). However, the mass fraction of inorganics increased by more than 10 % during
56	these years (H. Li et al., 2019a; Y. Wang et al., 2019a), implying the enhancement of
57	secondary inorganic aerosol (SIA) formation, which partly counteracted the decrease
58	in PM2.5. All these variations would change aerosol physicochemical properties. For
59	example, Xie et al. (2020) found that the aerosol pH increased as PM <sub>2.5</sub> decreased in the
60	past few years in urban Beijing due to the enhanced mass ratio of nitrate to sulfate. As
61	a possible consequence of the elevated aerosol pH, the dissolved ozone (O3) in particles
62	would play a more important role to SIA formation, especially for sulfate formation
63	(Seinfeld and Pandis, 2016). Therefore, the major chemical processes during haze
64	events and the control target should be re-evaluated. Therefore, e Elaborating the
65	secondary aerosol formation mechanism under current emission conditions is important
66	for taking more proper measures to control PM <sub>2.5</sub> in the future.

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67	Some studies have argued that controlling emissions of nitrogen oxides $(NO_x)$ is
68	important because nitrate in $\ensuremath{\text{PM}_{2.5}}$ has had the weakest decrease relative to other
69	chemical species over the past several years (Q. Zhang et al., 2019; F. Zhang et al.,
70	2020). The transformation of $NO_x$ to nitrate is closely related to atmospheric oxidation
71	processes (Seinfeld and Pandis, 2016). Surface ozone (O <sub>3</sub> ) is an important secondary
72	gaseous pollutant and oxidizing agent in the atmosphere. Recent studies have found
73	that a reduction in $PM_{2.5}$ resulted in an increase in the $O_3$ volume mixing ratio ([O <sub>3</sub> ]) at
74	a rate of 3.3 ppbv per annum during the summer of the past few years in the BTH region
75	(K. Li et al., 2019, 2020). The increased [O <sub>3</sub> ] can enhance the atmospheric oxidation
76	capacity, thereby promoting the formation of secondary aerosols in summer (T. Wang
77	et al., 2017). However, less emphasis has been placed on the variation in $[O_3]$ in winter.
78	The formation of $O_3$ and its effect on secondary aerosol formation in a cold environment
79	is thus unclear.

Some special events held in China have provided unique opportunities to 80 investigate the impact of human activities on air quality by taking advantage of unusual 81 82 changes associated with short-term, drastic measures implemented by the Chinese government to reduce anthropogenic emissions, such as the 2008 Summer Olympic 83 84 Games (T. Wang et al., 2010; Guo et al., 2013), the 2014 Asia-Pacific Economic Cooperation (Sun et al., 2016b), the 2015 China Victory Day parade (Y. Wang et al., 85 2017; Zhao et al., 2017), and the 2016 G20 Summit (H. Li et al., 2019b). The annual 86 Spring Festival holiday is also a special occasion when the vast majority of the 87 population stops working for 2 to 4 weeks (Tan et al., 2009; Y. Zhang et al., 2016; C. 88

Wang et al., 2017). Investigating the impact of changes in anthropogenic emissions on
gaseous pollutants and aerosol formation during these special occasions may provide
useful guidance on more scientifically sound measures to take to control PM<sub>2.5</sub>.

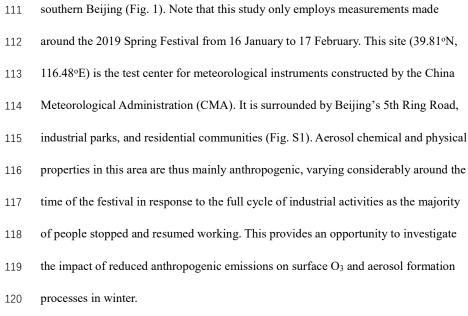
92 A comprehensive aerosol field experiment at a suburban site near the 5th Ring Road in the Daxing District of Beijing was carried out for more than two years, 93 including the 2019 Spring Festival. Beijing was one of the three top cities in China with 94 95 the largest migrating population during the 2019 Spring Festival holiday (https://cloud.tencent.com/developer/news/393324). In addition, fireworks were 96 prohibited throughout the Beijing metropolitan region within the 6th Ring of the Beijing 97 Beltway. The intensity of anthropogenic emissions was thus much weaker than usual 98 99 during this holiday. Our measurements-made around this period of the field campaign are thus ideal for investigating the impact of reduced anthropogenic emissions on 100 101 surface O<sub>3</sub> and aerosol formation.

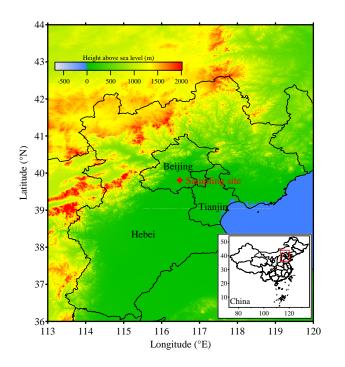
102 This paper is structured as follows. Section 2 describes the experiment site and the 103 measurement data used in this study. Section 3 presents the results and discussion, 104 mainly concerning the impact of reduced anthropogenic emissions during the holiday 105 on the variations in trace gases and aerosol chemical species in Beijing and the BTH 106 region. Section 4 presents the conclusions and their implications.

107

## 108 2. Experiment site and measurement data

- 109 A comprehensive field experiment measuring aerosol physical and chemical
- 110 properties was conducted from August 2017 to October 2019 at a suburban site in





**Figure 2**. Map showing the Beijing-Tianjin-Hebei region in China and the location of

the experiment site. The colored background shows the terrain height (unit: m abovesea level).

126

127 Table 1 lists the instruments used in this campaign. A scanning mobility particle sizer (SMPS) and an aerodynamic particle sizer (APS) measured the aerosol particle 128 number size distribution (PNSD) from 10 nm to 20 µm. The SMPS consists of a 129 differential mobility analyzer (model 3081, TSI Inc.) and a condensation particle 130 counter (model 3772, TSI Inc.). The aerodynamic diameter measured by the APS can 131 132 be converted to the Stokes diameter through division by the square root of the aerosol density. The aerosol density in this study was calculated following the method of Zhao 133 134 et al. (2017), using measured aerosol chemical composition information. An aerosol chemical speciation monitor (ACSM) equipped with a PM2.5 lens system, a capture 135 vaporizer, and a quadrupole mass spectrometer was used to measure mass 136 concentrations of non-refractory aerosol chemical species in PM2.5, including organics 137 (Org), nitrate ( $NO_3^-$ ), sulfate ( $SO_4^{2-}$ ), ammonium ( $NH_4^+$ ), and chlorine (Chl) (Peck et al., 138 139 2016; Xu et al., 2017; Y. Zhang et al., 2017). A seven-wavelength aethalometer (model AE-33, Magee Scientific Corp.) with a PM2.5 cyclone in the sample inlet was used to 140 141 retrieve the mass concentration of BC.

In addition to the above aerosol measurements, meteorological parameters were observed by the CMA at the experiment site. The Chinese Ministry of Ecology and Environment network and Beijing Municipal Environmental Monitoring Center (http://106.37.208.233:20035/ and http://www.bjmemc.com.cn/) provided PM<sub>2.5</sub> and trace gas (sulfur dioxide (SO<sub>2</sub>), nitrogen dioxide (NO<sub>2</sub>), carbon monoxide (CO), and

147	O <sub>3</sub> ) measurements made in different locations of the BTH region. Yizhuang in Beijing
148	is the nearest station to the experiment site (about 3.0 km to the southeast, Fig. S1). The
149	total mass concentrations of measured non-refractory aerosol chemical species and BC
150	mass concentrations in $PM_{2.5}$ show good consistency with the $PM_{2.5}$ mass
151	concentrations obtained from the Yizhuang station (Fig. S2).
152	

- 153 Table 1. Aerosol instruments used in this campaign and their observed parameters and
- 154 manufacturer information.

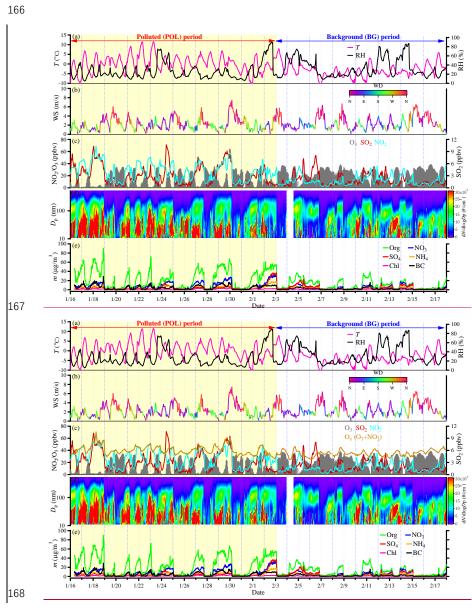
Instrument	Measured Parameters	Manufacturer	Model	Time Resolution
SMPS	Particle number size distribution (10–550 nm)	TSI	3938	5 min
APS	Particle number size distribution (0.5–20 µm)	TSI	3321	5 min
ACSM	Mass concentrations of non-refractory aerosol chemical species in PM <sub>2.5</sub>	Aerodyne	Q-ACSM	15 min
Aethalometer	Mass concentration of black carbon	Magee	AE-33	5 min

155

### 156 **3. Results and Discussion**

## 157 3.1. Basic meteorological and environmental characteristics

While the official Spring Festival holiday was from 4 February to 10 February 2019, many people left before 3 February and came back after the Lantern Festival (19 February). In this study, we regarded the days from 16 January to 2 February as the polluted (POL) period with high anthropogenic emissions, more representative of ordinary conditions, and the days from 3 February to 17 February as the background (BG) period with low anthropogenic emissions. Figure 2 shows the time series of meteorological parameters, trace gas volume mixing ratios, and aerosol properties 165 during the two periods.



169Figure 3. Time series of (a) ambient temperature (T) and relative humidity (RH), (b)170wind direction (WD) and speed (WS), (c) volume mixing ratios of trace gases [( $O_3$ , SO<sub>2</sub>,

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171 and  $NO_2$  and  $O_x$  (O<sub>3</sub>+NO<sub>2</sub>)], (d) the aerosol particle number size distribution measured 172 by the SMPS, and (e) mass concentrations of aerosol chemical species in PM<sub>2.5</sub> 173 measured by the ACSM and the AE-33. The trace gas information was from the 174 Yizhuang station, and the others were observed at the experiment site in Beijing (16 175 January to 17 February 2019).

176

177 Ambient temperature (T) and relative humidity (RH) have clear diurnal cycles (Fig. 2a). The average T and RH during the BG period were slightly lower (- $3.3\pm3.4$  versus 178 0.2±4.2°C) and higher (33.2±20.1 versus 25.8±17.6 %) than those during the POL 179 period, respectively. This was caused by several short-term light snowfall events that 180 occurred on 6, 12, and 14 February during the BG period. Figures 2b and S3 display 181 similar wind patterns during the POL and BG periods, i.e., wind patterns that changed 182 183 periodically. The prevailing, strong northerly winds during the two periods were beneficial to dispersing pollutants in Beijing (Sun et al., 2016b; Y. Wang et al., 2017), 184 and thus no heavy haze episodes occurred during these periods. Overall, the 185 186 meteorological parameters were similar during the POL and BG periods.

Figure 2c <u>illustratesdepiets</u> that the volume mixing ratios of SO<sub>2</sub> and NO<sub>2</sub> ([SO<sub>2</sub>] and [NSO<sub>2</sub>]) during the BG period were lower than those during the POL period, suggesting less gaseous pollutants from anthropogenic emissions during the BG period. In addition, [O<sub>3</sub>] remained at a high level for several days during the BG period but not during the POL period. The average [O<sub>3</sub>] increased by 77.4 % during the BG period compared with the POL period ( $46.2\pm18.9$  versus  $26.1\pm22.2$  ppbv). The percent change

193	in [O <sub>3</sub> ] due to the "holiday effect" during this field campaign is much higher than that	
194	reported in other regions of China (K. Huang et al., 2012; C. Wang et al., 2017; S. Wang	
195	et al., 2019). Figure 2c also shows a weak variation in $O_x$ ( $O_3 + NO_2$ ) from the POL	
196	period to the BG period, indicating that strong O3-titration appeared during the Spring	

197 <u>Festival 2019.</u>

214

198 Many bursts of fine particles (Fig. 2d) occurring mainly during rush hours or at 199 night were observed during the POL period. This is likely related to the substantial increases in gasoline or diesel vehicles on two nearby roads at these times. Zhu et al. 200 (2017) found that efficient nucleation and partitioning of gaseous species from on-road 201 vehicles can promote new particle formation in the wintertime. However, this 202 phenomenon occurred much less frequently during the BG period, likely because of the 203 massive reduction in on-road vehicles. The few short-term bursts of fine particles 204 during the BG period occurred during the daytime, presumably because of enhanced 205 206 nucleation by photochemical processes.

The aerosol chemical species in PM<sub>2.5</sub> also differed during the POL and BG periods (Fig. 2e). During the POL period, the mass concentrations of aerosol chemical species readily accumulated, especially the organics ( $m_{org}$ ) with rapid increases at night. The mass concentration of BC ( $m_{BC}$ ) also clearly increased, likely associated with an increase in heavy-duty diesel vehicles and a decrease in the nocturnal planetary boundary layer at night (Y. Wang et al., 2017; Zhao et al., 2017; Z. Li et al., 2017). However, the increases in  $m_{org}$  and  $m_{BC}$  at night during the BG period were not as strong

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as those during the POL period. The mass concentration of nitrate  $(m_{NO3})$  largely

215 decreased during the BG period, while there was a weak variation in the mass 216 concentration of sulfate ( $m_{SO4}$ ).

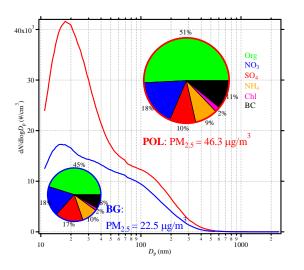
In summary, distinct differences existed in all observed trace gases and aerosol chemical and physical parameters during the POL and BG periods. However, the meteorological parameters (wind direction and speed, ambient temperature, and RH) and weather regimes were similar during these two periods. This helps to single out the impact of reduced anthropogenic emissions on trace gases and aerosol formation processes.

223

## 224 3.2. Impact of reduced anthropogenic emissions on aerosol formation processes

The average PM<sub>2.5</sub> mass concentrations were 46.3 and 22.5  $\mu$ g/m<sup>3</sup> during the POL 225 and BG periods, respectively. Figure 3 illustrates the average PNSD and aerosol 226 chemical species in PM2.5 during the two periods. The particle number concentrations 227 at all sizes were much higher during the POL period than during the BG period, 228 229 especially for ultrafine particles (with diameters, or  $D_{\rm p}$ , < 100 nm). The diurnal variation in PNSD during the POL period shown in Fig. 4a suggests that aerosol 230 231 particles with D<sub>p</sub> < 50 nm burst during rush hours and in the nighttime. The total particle number concentration (N) remained greater than 30,000 cm<sup>-3</sup> at these times. However, 232 during the BG period, the number concentration of ultrafine particles only increased 233 weakly during rush hours or nucleation times. N was always less than 20,000 cm<sup>-3</sup> on 234 all days during the BG period (Fig. 4b), probably linked with the reduction in on-road 235 236 vehicles during the holiday. As shown in Table 2, the ratio of BG to POL 10-50 nm particle number concentrations ( $N_{10-50 \text{ nm}}$ ) (0.47) is much smaller than the ratios for larger particles (0.78 for  $N_{50-100 \text{ nm}}$  and 0.67 for  $N_{>100 \text{ nm}}$ ). These all demonstrate the strong impact of reduced anthropogenic emissions on aerosol number concentrations, especially for nucleation-mode and small Aitken-mode particles.

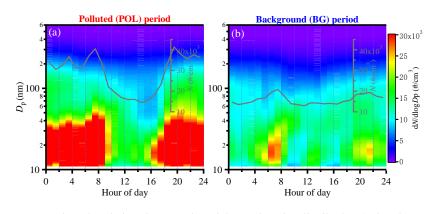
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Figure 4. Average aerosol particle number size distributions (red and blue curves) and mass fractions of aerosol chemical species in PM<sub>2.5</sub> (pie charts with red and blue outlines) during the POL (in red) and BG (in blue) periods.

246



248 Figure 5. Diurnal variations in aerosol particle number size distribution (colored

249 background) and total aerosol number concentration (N, shown as grey curves) during

250 the (a) POL and (b) BG periods.

251

247

Table 2. Summary of the average aerosol number concentration (N) in different size ranges, volume mixing ratios of trace gases, mass concentrations of PM<sub>2.5</sub> and different aerosol chemical species, sulfur oxidation ratios (SOR), and nitrogen oxidation ratios

255 (NOR) during the POL and BG periods and their ratios.

	$N_{10-50 \text{ nm}}$	N50-100 nm	N>100 nm	SO <sub>2</sub>	NO <sub>2</sub>	O <sub>3</sub>	PM <sub>2.5</sub>	
	(cm <sup>-3</sup> )	(cm <sup>-3</sup> )	(cm <sup>-3</sup> )	(ppbv)	(ppbv)	(ppbv)	$(\mu g/m^3)$	
POL	20,861±19,935	3,946±2,544	3,888±2,757	8.31±6.35	51.96±27.35	26.06±22.24	46.32±39.05	
BG	9,837±8,493	3,071±1,478	2,600±2,223	4.85±3.83	21.87±13.99	$46.23{\pm}18.86$	$22.52 \pm 20.28$	
BG/POL	0.47	0.78	0.67	0.58	0.42	1.77	0.49	
	morg	<i>m</i> <sub>NO3</sub>	m <sub>SO4</sub>	<i>m</i> <sub>NH4</sub>	mbc	SOR	NOR	
	(µg/m <sup>3</sup> )	$(\mu g/m^3)$	$(\mu g/m^3)$	$(\mu g/m^3)$	$(\mu g/m^3)$	SOR	NOK	
POL	23.55±19.58	8.25±7.91	4.59±6.20	3.96±3.83	5.05±4.51	0.27±0.17	0.09±0.08	
BG	10.17±9.13	4.09±4.25	3.82±4.08	2.18±2.14	1.91±1.74	0.32±0.18	$0.10 \pm 0.08$	
BG/POL	0.43	0.50	0.83	0.55	0.38	1.19	1.11	

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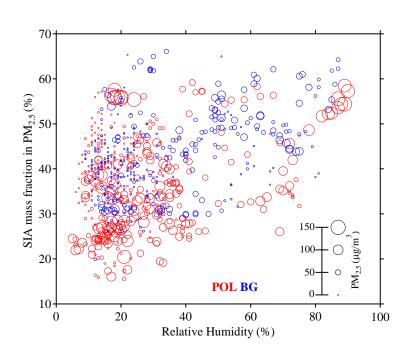
Table 2 also indicates that the mass concentrations (*m*) of aerosol chemical species

258 in PM<sub>2.5</sub> were much less during the BG period than during the POL periodelearly

259 decreased more during the BG period than during the POL period. The *m* related to

260	primary emissions ( $m_{\rm org}$ and $m_{\rm BC}$ ) decreased by more than 50 %, but the <i>m</i> of secondary
261	inorganic aerosols (SIA, including nitrate, sulfate, and ammonium) slightly decreased,
262	especially-for sulfate ( $m_{SO4}$ decreased by only 17 %). The pie charts in Fig. 3 show
263	significant differences in the aerosol chemical species of PM <sub>2.5</sub> during the two periods.
264	The mass fractions of Org and BC were lower during the BG period (45 $\%$ and 8 %,
265	respectively) than during the POL period (51 % and 11 %, respectively). By contrast,
266	the mass fraction of SIA was higher during the BG period (45 %) than during the POL
267	period (37 %). This indicates that the strongly reduced anthropogenic emissions during
268	the holiday caused sharp decreases in primary aerosols but not secondary aerosols. The
269	sulfur oxidation ratio (SOR) and nitrogen oxidation ratio (NOR) are usually calculated
270	to study the transformation of secondary aerosols (Sun et al., 2006; Y. Li et al., 2017).
271	SOR (NOR) is defined as the ratio of the molar concentration of sulfate (nitrate) to the
272	total molar concentration of sulfate (nitrate) and $SO_2$ (NO <sub>2</sub> ). Table 2 shows that SOR
273	and NOR were higher during the BG period than during the POL period, suggesting
274	that the formation of secondary inorganics was enhanced during the BG period. Figure
275	5 shows that most large $PM_{2.5}$ mass concentrations (>100 $\mu g/m^3)$ during the POL period
276	occurred along with low RH (< 40 %) and low SIA mass fractions, indicating the
277	important contribution of primary emissions to the accumulation of $PM_{2.5}$ in a polluted
278	environment. However, large $PM_{2.5}$ mass concentrations (> 50 $\mu\text{g/m^3})$ during the BG
279	period mainly appeared under moderate RH (40 $<$ RH $<$ 80 %) and high SIA mass
280	fraction conditions, likely caused by enhanced aqueous-phase chemical reactions
281	during this period.





283

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Figure 6. The variation in secondary inorganic aerosols (SIA) mass fraction in PM<sub>2.5</sub>
as a function of ambient relative humidity during the POL (in red) and BG (in blue)
periods. The different circle sizes denote different PM<sub>2.5</sub> mass concentrations.

The diurnal variation in  $[O_3]$  (Fig. 6a) shows more accumulated  $O_3$  during the BG period than during the POL period at any time of the day, revealing a stronger atmospheric oxidation capacity during the BG period. In particular,  $[O_3]$  at night was two times higher during the BG period than during the POL period (Fig. 6b). Distinct diurnal variations in SOR were found during the two periods (Fig. 6c). The higher SOR at night during the BG period (Fig. 6c) indicates the enhanced transformation of SO<sub>2</sub> to sulfate, likely related to nocturnal aqueous-phase chemical reactions <u>due to the elevated</u>

i	
295	ambient RH at night (Fig. S4). Figure S54 indicates that SOR increased following an
296	increase in ALWC when <u>the</u> ambient RH was higher than $\sim 40$ % during the BG period.
297	Moreover, Fig. 6d shows that the diurnal variation in the SOR ratio during the two
298	periods was similar to that of the $[O_3]$ ratio. This suggests that sulfate formation during
299	the holiday was likely enhanced by nocturnal aqueous-phase chemical reactions
300	between $SO_2$ and $O_3$ . This is consistent with the study of Fang et al. (2019), which found
301	that ambient RH and the O3 concentration are two prerequisites for rapid sulfate
302	formation via aqueous-phase oxidation reactions. This result highlights that controlling
303	$O_3$ formation under current emission conditions in winter in Beijing is key to further
304	reducing the formation of sulfate and implies that the high underestimation of sulfate
305	at night in models (Miao et al., 2020) could be caused by <u>anthe</u> inaccurate simulation
306	of $[O_3]$ . The higher daytime NOR (Fig. 6e) than nighttime NOR during the two periods
307	illustrates that the formation of nitrate was mainly controlled by photochemical
308	reactions in winter. Figure 6f shows that the larger NOR difference (the higher NOR
309	ratio) during rush hours during the two periods likely occurred because a mass of
310	emitted NO <sub>x</sub> during rush hours could not be transformed to nitrate during the POL
311	period. Figure 6e and 6f also suggests nitrate formation was <u>somewhat</u> enhanced <del>a little</del>
312	during the holiday likely due to the enhanced daytime photochemical reactions.

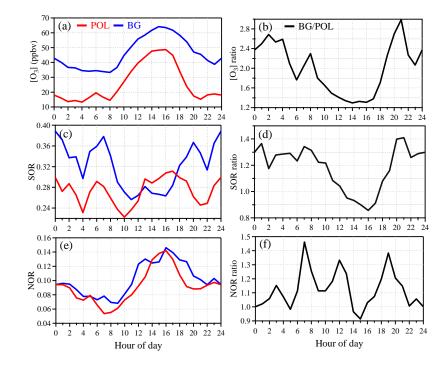




Figure 7. Diurnal variations in (a and b) O<sub>3</sub> volume mixing ratio and its ratio, (c and
d) sulfur oxidation ratio (SOR) and its ratio, and (e and f) nitrogen oxidation ratio

316 (NOR) and its ratio during the BG and POL periods. The ratio of a quantity is that

317 quantity during the BG period divided by that quantity during the POL period.

318

319 Overall, the reduced anthropogenic emissions led to a drastic decrease in aerosol

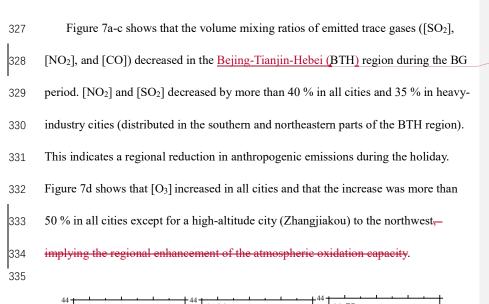
320 particle number concentration during the holiday. However, the general enhancement

321 of [O<sub>3</sub>]atmospheric oxidation capacity was enhanced during the holiday, thereby could

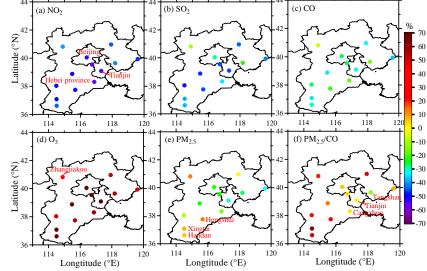
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322 promot<u>eing</u> the formation of secondary inorganics (especially sulfate).

- 323 324
- 325



3.3. Impact of reduced anthropogenic emissions on regional air pollution



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**Figure 8**. Percent changes in trace gas volume mixing ratios (NO<sub>2</sub>, SO<sub>2</sub>, CO, and O<sub>3</sub>),

the PM<sub>2.5</sub> mass concentration, and the PM<sub>2.5</sub>/CO ratio during the BG period relative to the POL period:  $100 \times \left(\frac{[BG]-[POL]}{[POL]}\right)$  in the Beijing-Tianjin-Hebei (BTH) region. 带格式的: 图案: 清除

341	The PM <sub>2.5</sub> mass concentration decreased at a much lower rate relative to the	
342	decrease in [NO <sub>2</sub> ] and [SO <sub>2</sub> ] in most cities during the BG period (Fig. 7e), while it	
343	increased slightly at Zhangjiakou and three southern heavy-industry cities (Hengshui,	
344	Xingtai, and Handan). The ratio $PM_{2.5}/CO$ is an indicator of aerosol secondary	
345	formation to primary emissions. An increase in $PM_{2.5}/CO$ was found during the BG	
346	period at all cities except for three coastal cities (Tangshan, Tianjin, and Cangzhou),	
347	revealing the regional enhancement of secondary aerosol formation during the	
348	holiday. The weak decrease in $PM_{2.5}/CO$ at the three coastal cities was likely due to	
349	the influence of mixed sea flows.	
350	The regional analysis of air pollution assumes that the findings from Beijing	
351	presented in section 3.2 are applicable to the entire BTH region. Regionally reduced	
352	anthropogenic emissions resulted in sharply decreased gaseous pollutants and	
353	increased O <sub>3</sub> . Higher atmospheric oxidation led to t <u>T</u> he enhanced formation of	
354	secondary aerosols likely due to enhanced aqueous chemical reactions involving O <sub>3</sub>	<b>带格式的:</b> 下标
355	<u>could</u> , thus counteracting the decrease in $PM_{2.5}$ mass concentration. There are two	
356	possible reasons explaining the high $[O_3]$ during the holiday: (1) the reduced gaseous	
357	precursors (NO <sub>x</sub> and SO <sub>2</sub> ) weakened the consumption of O <sub>3</sub> , and (2) O <sub>3</sub> formation in	
358	the BTH region is <u>likely</u> volatile organic compound (VOC)-controlled under current	
359	emission conditions, therefore the reduction in $NO_x$ would lead to higher $[O_3]$ . This-	
360	<u>All these results</u> demonstrates that it is more important to reduce $Q_3$ formation VOC	<b>带格式的:</b> 下标
361 362 363	emissions to further control $PM_{2.5}$ in winter in the BTH region.	

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## 365 4. Conclusions and Implications

366 In recent years, the mass concentration of particulate matter with an aerodynamic 367 diameter of less than 2.5 µm (PM2.5) has shown a general decreasing trend, presumably due to the series of emission reduction measures taken in China in an attempting to 368 369 improve air quality. However, haze pollution episodes still occur from time to time, 370 including during some special events when primary emissions reduced drastically, such 371 as the Chinese New Year holiday and even during the COVID-19 lockdown when 372 anthropogenic activities diminished dramaticallydrastically (X. Huang et al., 2020). We conjecture that reductions through primary emissions may be offset by increases in the 373 374 formation of secondary aerosols.

375 To test this, we examined the secondary aerosol formation mechanism in a 376 comprehensive field experiment conducted in Beijing. Comprehensive aAerosol and 377 meteorological measurements were made for more than two years, but data around the 2019 Chinese Spring Festival from 16 January to 17 February were employed in this 378 study to single out the impact of emission reductions due to the holiday. The study 379 period was divided into polluted (POL) and background (BG) periods, with high and 380 low anthropogenic emissions before and during the festival holiday, respectively. 381 Investigated were the impacts of reduced anthropogenic emissions on trace gases and 382 383 PM<sub>2.5</sub> under similar meteorological conditions.

384 The average  $PM_{2.5}$  mass concentrations were 46.3 and 22.5  $\mu$ g/m<sup>3</sup> during the POL 385 and BG periods, respectively, with no heavy haze events occurring. The average aerosol

386	particle number size distribution shows that the reduced anthropogenic emissions	
387	during the holiday led to decreased aerosol number concentrations at all sizes,	
388	especially in the nucleation and Aitken modes (mobility diameters less than 50 nm).	
389	Simultaneously, the reduced anthropogenic emissions resulted in decreases in the	
390	volume mixing ratios of $SO_2$ and $NO_2$ and an unexpected increase in the volume mixing	
391	ratio of O <sub>3</sub> [O <sub>3</sub> ] during the BG period <u>caused by the appeared O<sub>3</sub>-titration phenomenon</u> .	Ħ
392	The analysis of the aerosol chemical species in $\ensuremath{\text{PM}_{2.5}}$ demonstrates that the large	
393	decreases in organics and black carbon mass concentrations during the BG period were	
394	likely caused by the large decrease in on-road vehicles. Moreover, the mass	
395	concentration of nitrate also decreased while that of sulfate decreased much less during	
396	the BG period. Comparisons of the sulfur oxidation ratio (SOR) and the nitrogen	
397	oxidation ratio (NOR) during the two periods imply that the transformation of gaseous	
398	precursors to secondary inorganics (especially the transformation of SO <sub>2</sub> to sulfate) was	
399	promoted during the BG period, likely due to the enhanced atmospheric aqueous	
400	chemical reactions involving the dissolved O20xidation capacity. The diurnal variation	Ħ
401	in the SOR ratio between the BG and POL periods was similar to that of the $[O_3]$ ratio,	
402	illustrating that sulfate formation was promoted by the enhanced nocturnal aqueous-	
403	phase chemical reactions between $SO_2$ and $O_3$ under moderate relative humidity (RH)	
404	conditions (40 % < RH < 80 %). The higher NOR in the daytime during the two periods	
405	points out that the formation of nitrate was mainly controlled by photochemical	
406	reactions and weakly affected by the increase in [O <sub>3</sub> ].	

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408	regional air pollution patterns during the holiday. The variation trends of trace gases in	
409	most cities in the Beijing-Tian-Hebei (BTH) region were similar to those in Beijing,	
410	indicating the regional influence of reduced anthropogenic emissions on the volume	
411	mixing ratios of trace gases during the holiday. The weak $PM_{2.5}$ variation and the	
412	increased $\text{PM}_{2.5}/\text{CO}$ ratio (an indicator of aerosol secondary formation to primary	
413	emissions) during the BG period both suggest that the enhanced formation of secondary	
414	aerosols offset the regional decrease in PM <sub>2.5</sub> during the holiday.	
415	Our findings provide evidence that decreases in anthropogenic emissions can	
416	promote the formation of secondary inorganics due to the enhancement of aqueous	
417	reactions likely caused by the increased [O <sub>3</sub> ]. This result implies that haze mitigation	
418	in north China needs a coordinated and balanced strategy for controlling multiple	
419	pollutantsthe atmospheric oxidation capacity (manifested by more accumulated O3).	
420	In the future, the simultaneous control of PM <sub>2.5</sub> and O <sub>3</sub> will be needed to further reduce	
421	air pollution. The O <sub>3</sub> formation in winter in the BTH region is possibly volatile organic	
422	compound (VOC) controlled under current emission conditions. Controlling VOC	
423	emissions may thus be more important than controlling emissions of nitrogen oxides.	
424		
425	Advantational Kay D&D Department of the	
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429	42005067), and the Open Fund of State Key Laboratory of Remote Sensing Science	
430	(grant no. 202015). Startup Foundation for Introducing Talent of NUIST (No.	[
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433	Data availability. Data from the Chinese Ministry of Ecology and Environment
434	network and Beijing Municipal Environmental Monitoring Center can be downloaded
435	from the websites given in the main text. The measurement data from the field
436	experiment used in this study are available from the first author upon request
437	(yuyingwang@nuist.edu.cn).
438	
439	Author contributions. ZL and PY designed the field experiment. YW and ZL
440	conceived the study and led the overall scientific questions. YW, QW, and XJ
441	processed the measurement data and prepared this paper. MC copyedited the article.
442	Other co-authors participated in the implementation of this experiment and the
443	discussion of this paper.
444	
445	Competing interests. The authors declare that they have no conflict of interest.
446	
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