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Interactive comment

## *Interactive comment on* "Exploring the drivers of the elevated ozone production in Beijing in summertime during 2005–2016" *by* Wenjie Wang et al.

## Anonymous Referee #2

Received and published: 30 August 2020

This manuscript investigated the trends of surface ozone metrics and their causes at a urban site in Beijing. Detailed measurements, TUV and RACM box model were employed. Long-term ozone trends in China is a key issue to take effective measures in ths situation of rapid increasing ozone backgroud. This manuscript investimated the impacts of meteorology, VOCs, NOx and PM2.5 on ozone trend, and would be helpful to understand the formation mechnisms in North China. This manuscript is within the scope of ACP and is well organised and executed and there is no doubt about the quality of the work. It can be accepted for publication after the following comments are addressed. 1. Regional transport is also a key source of surface ozone. This work tried to assess the impact of regional O3 by analyzing measurements at a

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regional background site. This is not suffcient bacause this site was largely affected by Beijing emissions. I suggested that other background sites can be employed or back-transjectories at typical year can be used to analyze the impact of regional transport.

2. Recent, a few heteorogeneous chemical reactions are thought to be potential factors of ozone. For example, photolysis of HNO3 (NO3–) adsorbed on the solid surface of aerosol particles effectively produces HONO and NOx in the gas phase (Salgado and Rossi et al., PCCP,2002; Ramazan, 2006). A short disscuss should be performed.

M. S. Salgado Muñoz and M. J. Rossi, Heterogeneous reactions of HNO3 with flame soot generated under different combustion conditions. Reaction mechanism and kinetics, Physical Chemistry Chemical Physics, Phys. Chem. Chem. Phys., 2002,4, 5110-5118.

Ramazan, K., Wingen, A.M., Miller, Y., Chaban, G.M., Gerber, R.B., Xantheas, S.S. and Finlayson-Pitts, B.J. (2006). New experimental and theoretical approach to the heterogeneous hydrolysis of NO2:âĂL' Key role of molecular nitric acid and its complexes. J. Phys. Chem. A 110: 6886–6897

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