Concerted measurements of lipids in seawater and on submicron aerosol particles at the Cape Verde Islands: biogenic sources, selective transfer and high enrichments

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Abstract

~~Measurements~~In the marine environment, measurements of lipids as representative species ~~for~~within different lipid classes ~~in the marine environment~~ have been performed to characterize their oceanic sources and their transfer from the ocean into the atmosphere to marine aerosol particles. The set of lipid classes includes hydrocarbons (HC), fatty acid methyl esters (ME), free fatty acids (FFA), alcohols (ALC), 1,3-diacylglycerols (1,3 DG), 1,2-diacylglycerols (1,2 DG), monoacylglycerols (MG), wax esters (WE), triacylglycerols (TG), phospholipids (PP) including phosphatidylglycerols (PG), phosphatidylethanolamine (PE), phosphatidylcholines (PC), as well as glycolipids (GL) ~~including~~which cover sulfoquinovosyldiacylglycerols (SQDG), monogalactosyl-diacylglycerols (MGDG), digalactosyldiacylglycerols (DGDG) and sterols (ST). ~~Introduced~~These introduced lipid classes have been analyzed in the dissolved and particulate fraction of seawater, ~~while~~ differentiating between underlying water (ULW) and the sea surface microlayer (SML~~), and~~) on the one hand. On the other hand, they have been examined on ambient submicron aerosol particle samples (PM$_1$) which were collected ~~from the ambient~~ at the Cape Verde Atmospheric Observatory (CVAO) applying concerted measurements. ~~The~~These different lipids are found in all marine compartments but in different compositions. ~~At this point, a~~Along the campaign, certain ~~variability is~~variabilities are observed for the concentration of dissolved ($\sum$DL$_{ULW}$: 39.8-128.5 µg L^{-1}, $\sum$DL$_{SML}$: 55.7-121.5 µg L^{-1}) and particulate ($\sum$PL$_{ULW}$: 36.4-93.5 µg L^{-1}, $\sum$PL$_{SML}$: 61.0-118.1 µg L^{-1}) lipids in the seawater of the tropical North Atlantic Ocean ~~along the campaign~~. Only slight SML enrichments are observed for the lipids with an enrichment factor EF$_{SML}$ of 1.1-1.4 (DL) and 1.0-1.7 (PL). On PM$_1$ aerosol particles, a total lipid concentration between 75.2-219.5 ng m^{-3} (averaged: 119.9 ng m^{-3}) is measured. ~~Besides~~As also bacteria - besides phytoplankton sources~~, bacteria -~~ influence the lipid concentrations in seawater and on the aerosol particles,

~~so that the~~ the lipid abundance cannot be exclusively explained by the phytoplankton tracer (chlorophyll-*a*~~) cannot sufficiently explain the lipid abundance.~~). The concentration and enrichment of lipids in the SML ~~is~~are not related to physicochemical properties ~~describing~~which describe the surface activity. ~~For aerosols, however,~~On the ~~high enrichment of lipids (as a sum) corresponds well with the consideration of their high surface activity, thus the~~aerosol particles, an EF_{aer} (the enrichment factor on ~~the~~ submicron aerosol particles compared to ~~the~~ SML) ~~ranges~~ between $9 \cdot 10^4$-$7 \cdot 10^5$ is observed. Regarding the ~~single~~individual lipid groups on the aerosol particles, a ~~weak relation~~statistically significant correlation (R^2= 0.45, p= 0.028) was found between EF_{aer} and lipophilicity (expressed by the K_{OW} value~~) was identified,~~), which was ~~absent~~not present for the SML. ~~However, overall~~But simple physico-chemical descriptors are overall not sufficient to fully explain the transfer of lipids. As our findings show that additional processes such as formation and degradation influence the ocean-atmosphere transfer of both OM in general and of lipids in particular, they have to be considered in OM transfer models. Moreover, our data suggest that the ~~extend~~extent of ~~the~~ enrichment of ~~the~~ lipid classes constituents on the aerosol particles might be related to the distribution of the lipid within the bubble-air-water-interface. ~~Lipids,~~The lipids TG and ALC which are preferably arranged within the bubble interface~~, namely TG and ALC,~~ are transferred to the aerosol particles to the highest ~~extend~~extent. Finally, the connection between ice nucleation particles (INP) in seawater, which are active already at higher temperatures (-10 °C to -15 °C), and the lipid classes PE and FFA suggests that lipids formed in the ocean have the potential to contribute to (biogenic) INP activity when transferred ~~to~~into the atmosphere.

Keywords

Lipids, organic matter, submicron marine aerosol particles (PM_1), sea surface microlayer (SML), ice nucleating particles (INP), enrichment factor, concerted measurements, Cape Verde Atmospheric Observatory (CVAO)

1. Introduction

Lipids are a major biochemical class of organic matter (OM) in seawater along with carbohydrates and proteins. Their ocean concentrations are much lower, yet their surface affinity and enrichment are higher than for the other groups (Burrows et al., 2014). As lipids are rich in carbon and serve as energy storage compounds, they are important components of the cellular metabolisms of species, at least in the ocean (Wakeham et al., 1997). They are ~~universally~~ distributed throughout ~~in~~ the marine environment and are involved in numerous essential biological processes of both the dissolved and particulate OM pool (Arts et al., 2001;Frka et al., 2011). As regards the particulate lipid pool, important sources are phytoplankton cells with a contribution of up to 79 % in biologically productive surface water layers, while the contribution of lipids in phytoplankton cells ranges between ≤ 1 % and 46 % of dry weight (Frka et al. (2011), and references therein). Marine lipids can be produced by abiotic and biotic processes and play an important role as energy sources in the aquatic ecosystem (Parrish, 2013). As a whole, the analysis of lipids and their classes is a useful method to study the dynamics of the global carbon cycle in the ocean (Yoshimura et al., 2009), since lipid classes can be used as specific markers for the identification of OM sources and

biogeochemical cycles in the marine environment (Parrish et al., 1988;Frka et al., 2011). Phosphorus-containing lipids or phospholipids (PP), i.e. phosphatidylglycerols (PG), phosphatidylethanolamine (PE) and phosphatidylcholines (PC), belong to the organic substances associated with living organisms (Derieux et al., 1998) since they are a major component of cell membranes providing the structure and protection of cells (Khozin-Goldberg, 2016). The most common glycolipids (GL) in

5   plankton are mono- and di-galactosyldiacylglycerols (MGDG and DGDG) as well as sulfoquinovosyldiacylglycerol (SQDG) (Guschina and Harwood, 2009;Gašparović et al., 2013). Triacylglycerols (TG) indicate metabolic reserves (Frka et al., 2011) and are reported as storage compounds in phytoplankton (Becker et al., 2018). Wax esters (WE) are a major group of neutral lipids of some zooplankton species (Kattner, 1989). Fatty alcohols (ALC) mainly originate from zooplankton wax esters (Frka et al., 2011). Diacylglycerides (DG), monoacylglycerides (MG) and free fatty acids (FFA) are glyceride degradation products

10   which characterize the degradation level of lipids by means of the lipolysis index (LI) (Parrish et al., 1988;Goutx et al., 2003). ~~Using~~By using lipids as biomarkers, estimates of marine OM sinks and sources can be made and previous OM parameterizations, based only on chl-*a*, can be extended. Indeed, chl-*a* is often used as an indicator of marine biological productivity, representing the abundance of the major group of organisms, i.e. the photoautotrophs (Gantt et al., 2011;Rinaldi et al., 2013). To fully describe the biological control of the OM cycle, both autotrophic and heterotrophic organisms must be

15   considered. The information on the abundance of the main OM classes, namely lipids, proteins and carbohydrates, contributing to the marine OM pool and reflecting the OM sources, can therefore be used for advanced modelling approaches to predict OM on marine aerosol particles depending on the chemical composition of marine OM, as e.g. described by Burrows et al. (2014). Further spatio-temporal investigations of the organic classes in different oceanic areas are important for this purpose. The study of lipids is particularly important because their reactivity and physical surface properties contribute to the formation

20   and stabilization of the sea surface microlayer (SML) (Frka et al. (2012), and references therein). The SML represents a chemically distinct film enriched with surface active OM that accumulates at the air-water interface relative to the underlying water (ULW) (Wurl and Holmes, 2008;Wurl et al., 2011;Cunliffe et al., 2013). ~~Furthermore~~So, the SML~~,~~ as a natural interface between ocean and atmosphere~~,~~ could play an important role as OM source for aerosol particles in the marine environment (Engel et al., 2017). The presence of lipids at the air-water interface is the result of their high surface affinity, competitive

25   adsorption and segregation from other OM constituents (Frka et al., 2012).

OM can be transferred from the ocean ~~to~~into the atmosphere by wind-driven processes and bubble bursting, resulting in the formation of primary marine aerosol particles. Within this process, film drops leading to the formation of submicron particles are enriched mainly with OM compared to larger jet drops (Wilson et al., 2015). ~~Most~~According to most parameterizations, the transfer of OM and the prediction of the OM content on marine aerosol particles, is based on chl-*a* seawater concentrations

30   that are used as a broad indicator of biological productivity (Gantt et al., 2011;Rinaldi et al., 2013). However, especially in oligotrophic regions additional parameters besides wind speed and chl-*a* must be taken into consideration for ~~accurately~~an accurate prediction of OM on marine aerosol particles (van Pinxteren et al., 2017). In a new approach by Burrows et al. (2014) the parameterization/ OM prediction for marine aerosol particles is based on important compound classes of OM, e.g. lipids, carbohydrates~~,~~ and proteins, with different physico-chemical properties instead of the chl-*a* concentrations of the seawater.

This new approach requires information of distinct OM groups in both marine matrices (seawater and aerosol particles) such as the lipid concentrations performed in the present study.

Specific lipid classes such as long chain fatty acids and cholesterol as constituents of aerosol particles are already regarded as important factors for the activation of aerosol particles to cloud condensation nuclei (CCN) (Barati et al., 2019) or ice nucleation particles (INP) (Nguyen et al., 2017;DeMott et al., 2018). Cochran et al. (2016b) investigated the fatty acid composition ~~in~~on sub- and supermicron sea spray aerosol particles and reported that about 75 % of the submicron aerosol particles showed ~~strong~~clear signals for the presence of long-chain ~~fattiy~~fatty acids. In contrast, supermicrometer sea spray aerosol particles were dominated (up to 88 %) by oxygen-rich species (Cochran et al., 2016b).

Until now, lipids or individual lipid classes have often been ~~often~~ analyzed in only ~~in~~ one compartment of the ambient marine environment, i.e. seawater or marine aerosol particles (Gagosian et al., 1982;Simoneit and Mazurek, 1982;Frka et al., 2011;Gašparović et al., 2014). The possible transfer of total lipids (determined as a sum parameter) has so far only been described under controlled conditions in laboratory studies (Rastelli et al., 2017) or for certain lipid classes such as fatty acids, *n*-alkanes, total hydrocarbons (Marty et al., 1979) or as fatty acids and their derivatives (Cochran et al., 2016b) in freshly emitted sea spray aerosol. Unfortunately, comprehensive studies on marine lipid biogeochemistry and on a possible transfer of certain lipid classes into the atmosphere are missing.

The present work ~~aimed~~aims at investigating lipids at the Cape Verde Atmospheric Observatory (CVAO) as species representative for different lipid classes in the marine environment of the tropical Atlantic Ocean, to study their abundance, (biogenic) sources and selective transfer into the marine atmosphere. The lipid data set obtained for samples from different marine compartments at the CVAO is discussed with regard to its biological origin and its ice nucleation potential. Finally, the potential transfer of the lipids from seawater to aerosol particles will be investigated by relating the physico-chemical properties of the individual lipid classes to their respective observed atmospheric enrichment.

~~2.   Experimental~~

2.   Experiment

2.1 Study area and sampling sites

As part of the MarParCloud project (Marine biological production, organic aerosol particles and marine clouds: a Process chain) with a contribution from MARSU (MARine atmospheric Science Unravelled: Analytical and mass spectrometric techniques development and application), a field campaign was performed at the Cape Verde Atmospheric Observatory (CVAO, 16°51''49'N, 24°52'02'E) from 13 September to 13 October 2017 (Triesch et al., ~~2020~~2021;van Pinxteren et al., 2020). The CVAO, a remote marine station in the tropical Atlantic Ocean located on the northeast coast of Sao Vicente island (16º 51' 49" N. 24º 52' 02" S), is described in more detail in Carpenter et al. (2010) and Fomba et al. (2014). The ocean around the Cape Verde Islands belongs to the region 'North Atlantic Tropical Gyral Province (NATR)' according to the classification of Longhurst (2007), a region with the lowest surface chlorophyll in the North Atlantic and with a greater annual variability than seasonality. Wilson et al. (2015) reported that high concentrations of marine INPs can occur in the North Atlantic and

that comparatively high surface-level marine (INP_{15}) and OC concentrations have been predicted by models in this region around the Cape Verde Islands. ~~Concerted~~A concerted sampling was carried out during the campaign, including PM_1 aerosol particles at the CVAO and seawater at the ocean site (~16°53´17´N, ~24°54´25´E). The seawater sampling site was located upwind of the CVAO and had a minimal island influence. A map illustrating the seawater sampling station and the CVAO is shown in Figure S1.

2.1.1   SML and seawater sampling

The SML samples (n=6) were collected using the manual glass-plate technique, a standard SML sampling method whose correct application and specification are described in detail in the 'Guide to best practices to study the ocean's surface' by Cunliffe (2014). A glass plate (500x250 mm) with a sampling area of 2500 cm^2 was briefly immersed vertically in the seawater and then slowly pulled ~~upwards~~out. The surface film material adsorbed on the surface of the glass plate was then ~~moved~~ directly put into pre-cleaned bottles using Teflon wipers. The ULW samples (n=13) were collected from a depth of 1 m and also put in pre-cleaned plastic bottles, which were attached to a telescopic rod. The bottles were opened under water at the intended sampling depth in order to avoid influences by the SML. In addition, 5-6 litres of bulk surface water (at a depth between 10 and 50 cm) were collected with a plastic bottle for the pigment analysis (n=11).

2.1.2   Aerosol particles sampling

At the top of a 30 m high sampling tower at the CVAO, PM_1 aerosol particles (n=13) were ~~investigated with~~collected using a high-volume Digitel sampler DHA-80 (Walter Riemer Messtechnik, Germany) on 150 mm quartz ~~fibre~~fiber filters (Munktell, MK 360) at a flow rate of 700 L min^{-1} in 24 h. The filters used for sampling were preheated at 105 °C for 24 h to avoid contamination. The collected filters were stored in aluminium boxes at -20 °C up until ~~analysis~~analyzed.

2.2 Analytics

2.2.1   Seawater samples analytics: lipids, pigments, nutrients, microorganisms, INP activity

For the lipid analysis, the seawater samples (2 L) were passed through a 200 µm stainless steel screen to remove zooplankton and larger particles. Afterwards, they were filtered through pre-combusted (350 °C for 5 h) 0.7 µm GF/F filters (Whatman®, Sigma Aldrich). After the filtration step, the internal standard 2-hexadecanone (purity ≥ 98 %, Sigma Aldrich) (10 µg) was added to each filtrate, while the corresponding filters were stored in tubes at -78 °C until ~~analysis~~analyzed. Dissolved lipids (DL) were immediately extracted from the filtrate with dichloromethane (DCM; dichloromethane for liquid chromatography LiChrosolv®, Merck) in four extraction steps (twice at pH 8 and twice at pH 2). The final extract was concentrated using a rotary evaporator. Particulate lipids (PL) were extracted by a modified one-phase solvent mixture of DCM-methanol-water (Bligh and Dyer, 1959) after the addition of 2-hexadecanone as internal standard . DL and PL classes were further analyzed by Iatroscan thin layer chromatography-flame ionization detection (TLC- FID) (Iatroscan MK-VI, Iatron, Japan) as described

in Gašparović et al. (2015) and (2017). After the separation on Chromarod-SIII thin layer rods, lipid classes were identified and quantified by external calibration with a standard lipid mixture. The investigated lipid classes ~~investigated~~ included HC, ME, FFA, ALC, 1,3 DG, 1,2 DG, MG, WE, TG, PP (including PG, PE, PC), GL (including SQDG, MGDG, DGDG), ST and PIG. Each sample was measured ~~in duplicate~~twice with a relative standard deviation < 10 %. Also, blank seawater samples

5 (high purity water filled in pre-cleaned plastic bottles and handled the same as the seawater samples) were ~~analysed~~analyzed and the blank values were always below 15 % of ambient seawater samples. All presented lipid values for the SML and ULW samples were blank corrected by subtracting the field blank values from the samples.

For the pigment analysis, 5-6 L of bulk water were filtrated through GF/F filters. These were extracted in 5 mL ethanol~~,~~ and an aliquot (20 µL) was injected into a high performance liquid chromatograph system (HPLC) with fluorescence detection

10 (Dionex, Sunnyvale, CA. USA). The pigments, including chl-a, chl-b, fucoxanthin, pheophorbide, pheophytin a/b, zeaxanthin, diadinoxanthin, lutein, chlorophyllide, violaxanthin, β-carotene, were separated under gradient elution using methanol/acetonitrile/water systems as mobile phase, as described in van Pinxteren et al. (2019). Nutrients covering nitrogen oxides (N_2O_3, NO_2, NO_3), ~~phosphate~~phosphates (PO_4^{3-}) and silicates (SiO_4^{4+}) were measured colorimetrically according to Grasshoff et al. (1999) with a Seal Analytical QuAAtro constant flow analyzer. Microbial cell numbers were counted via flow

15 cytometry after the seawater samples were fixed, flash-frozen in liquid-nitrogen, and stored at -20 °C. For the prokaryotic cells counts, all samples were stained with SYBR Green solution. ~~Counting~~The counting was performed after the addition of latex beads serving as an internal standard. Further details can be found in Robinson et al. (2019). Small autotrophic cells were counted in a separate measurement after ~~addition of~~adding red fluorescent latex beads (Polysciences, Eppelheim, Germany). Cells were detected by their signature in a plot of red (FL3) vs. orange (FL2) fluorescence, and red fluorescence vs. side scatter

20 (SSC). This approach allows discrimination between different groups of prokaryotic and eukaryotic autotrophs (Marie et al., 2010), which in our case were size classes defined as *Synechococcus-like cells* and *Nanoeucaryotes.* Two droplet freezing devices called LINA (Leipzig Ice Nucleation Array) and INDA (Ice Nucleation Droplet Array) ~~were~~are used at TROPOS to obtain information on the ice nucleation activity as described in more detail by Gong et al. (2020), and references therein.

25 **2.2.2   Aerosol particle samples analytics**

The analysis of sodium (Na^+) was performed by ion chromatography (ICS3000, Dionex, Sunnyvale, CA, USA) as described in Mueller et al. (2010). For the analysis of the lipid classes, 28.27 cm^2 of the total PM$_1$ filter area were extracted and measured following the procedure for particulate lipids in seawater (see section 2.2.1). A chromatogram of the TLC-FID measurements of an aerosol particle sample and the standards is shown in Fig. S2. The field blanks (n=5) were prepared using pre-baked

30 quartz ~~fibre~~fiber filters without an active sampling and treated according to the same procedure as the field samples. The concentrations of the lipid classes were calculated by external calibration. Each sample was measured ~~in duplicate~~twice with a relative standard deviation < 10 % and field blanks, which were always below 20 % of the real aerosol particle sample, were subtracted. All presented values are ~~blank~~corrected for the field blank.

2.2.3 Lipid ratios

The Lipolysis Index (LI) is an index describing the lipid degradative state and the biological degradation processes of lipids (Goutx et al., 2003). ~~In this~~This concept~~, it is proposed that~~ describes the degradation process of marine acyl-lipids by the concentration ratio of free lipids/metabolites (ALC, FFA, MG, DG) to their precursors (TG, WE, glycolipids as MGDG, DGDG, SQDG and phospholipids as PG, PE, PC) ~~could be described~~ by equation (1).

$$LI = \frac{(ALC+FFA+MG+DG)}{(TG+WE+MGDG+DGDG+SQDG+PG+PE+PC)} \quad (1)$$

It is obvious that the concentration of precursor lipids and metabolites alone can already influence the LI and that there may be a natural variance between them. For example, metabolites can be quickly absorbed by existing microorganisms, which would lead to a reduction of the concentration and thus influence the LI. However, the LI is reported as a useful measure to characterize the degradation level of biogenic organic material (Goutx et al., 2003). Higher LI values are characteristic for an enhanced OM degradation and metabolite release, while lower LI values indicate that the appearing lipid classes are more fresh or resistant to degradation.

The PE/PG ratio can be used to determine the origin of the phospholipids that contribute to the OM pool in seawater (Goutx et al., 1993). Here, PG as the most important compound of the phospholipids of microalgae is an indicator for algae as potential sources, whereas PE is predominantly found in bacterial membranes and thus represents an indicator for bacterial sources (Goutx et al., 1993).

2.2.4 Enrichment factors

The enrichment factor in the SML (EF_{SML}) was calculated by dividing the concentration of the respective analyte in the SML by the concentration of the analyte in the ULW using equation (2):

$$EF_{SML} = \frac{c\ (analyte)_{SML}}{c(analyte)_{ULW}} \quad (2)$$

An enrichment in the SML is defined as $EF_{SML} > 1$, a depletion with $EF_{SML} < 1$.

The EF_{aer} is a quantitative metric for the comparison of compounds in the ocean and in the atmosphere. The EF_{aer} concept is mainly applied to closed systems (Quinn et al. (2015) and references therein, Rastelli et al. (2017)) since formation or degradation pathways on aerosol particles including biological or photochemical atmospheric reactions and possible transports from other than marine sources are excluded ~~for~~of this parameter. ~~However~~Nevertheless, for comparison purposes it is useful to ~~calculate~~apply the EF_{aer} also for open systems, as in the studies of e.g. Russell et al. (2010) or van Pinxteren et al. (2017). To calculate the enrichment factor of the different analytes ~~in~~on aerosol particles (EF_{aer}) relative to ~~the~~seawater (SW), here distinguished between SML and ULW, the atmospheric concentration of the analyte relative to the sodium concentration on

the PM$_1$ sample was divided by the analyte concentration relative to the sodium concentration in the corresponding ~~SML~~SW sample using equation (3):

$$EF_{aer} = \frac{\cancel{c\,(analyte)_{PM_1}/c\,(Na^+)_{PM_1}}\;c\,(analyte)_{PM_1}/c\,(Na^+)_{PM_1}}{\cancel{c\,(analyte)_{SML}/c\,(Na^+)_{SML}}\;c\,(analyte)_{SW}/c\,(Na^+)_{SW}} \tag{3}$$

The EF$_{aer}$ calculation was limited by the availability of the analyte concentration in both matrices, i.e. PM$_1$ and SML samples were collected simultaneously.

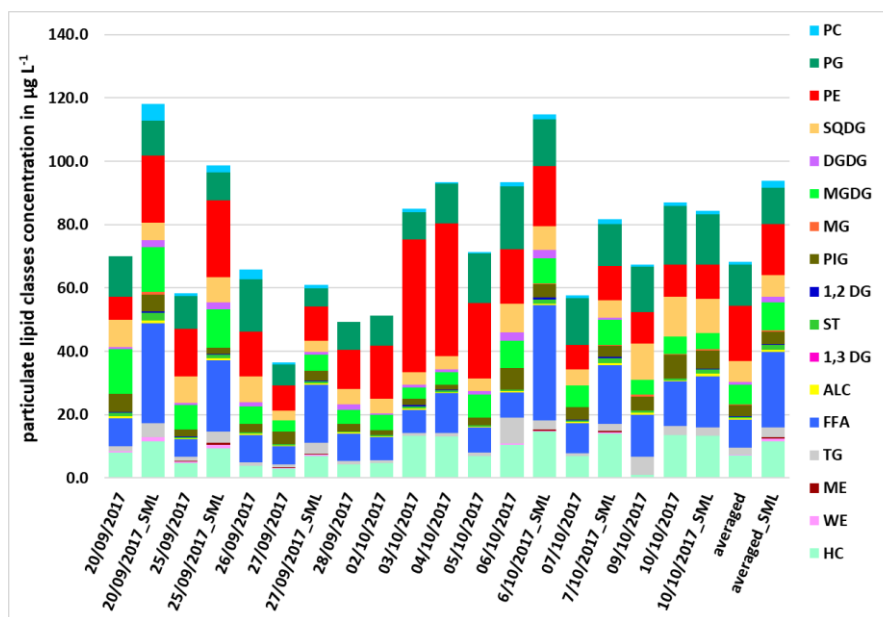
2.2.5 Statistical analysis

To investigate possible relationships between chemical, (micro)biological and physical parameters of the seawater samples (ULW, SML), listed in Fig. S3, a Pearson correlation analysis was performed. Figs. S4-S7 show the matrix of parameters when either ULW or SML samples of the dissolved or particulate fraction are considered. The correlation coefficient (R), the number of samples examined (n) and the p-value were used to validate the significance of the correlation. In particular, the p-value as a test for statistical hypothesis in research areas must be considered when defining statistical relevance (Bhattacharya and Habtzghi, 2002;Perezgonzalez, 2015). The statistical parameters for the performed analysis of the ULW and SML samples are defined as follows. For the ULW samples, the statistical relevance of a relationship ~~was~~is defined if n ≥ 4 (n$_{max}$=13), R ≥ 0.6 and p-value ≤ 0.05 (Perezgonzalez, 2015). We ~~considered~~consider a 'trend' to be valid if n ≥ 4 (n$_{max}$=13), R ≥ 0.4 and p-value ≥ 0.05. Due to the limited number of SML samples, p-values were always ≥ 0.05, so we could not define a statistical relevance, but considered a 'trend' to be valid if n ≥ 4 (n$_{max}$=6), R ≥ 0.4.

3. Results and Discussion

3.1 Seawater and SML samples

3.1.1 ~~3.1.1~~ Lipids and lipid classes in the particulate fraction

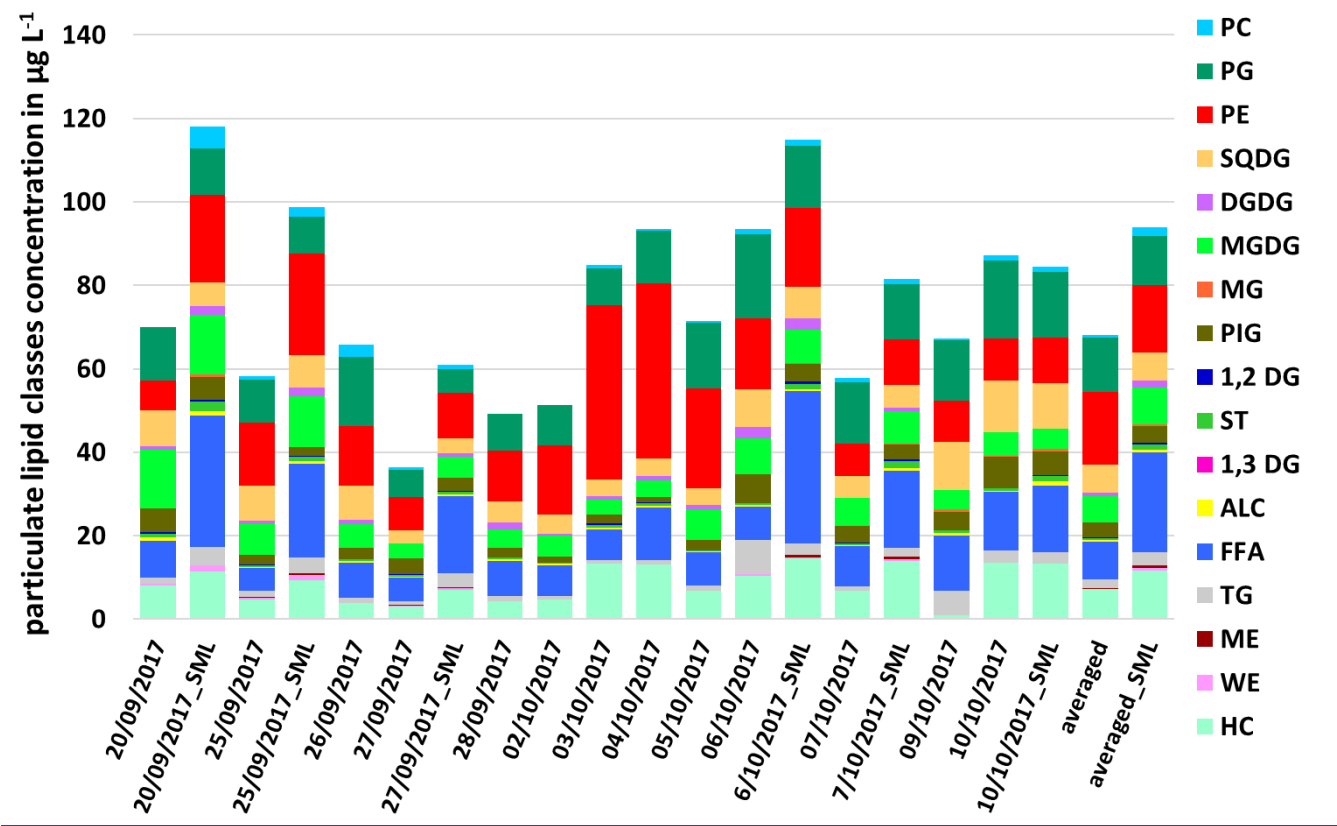
The particulate lipids showed a certain variance during the campaign, as shown in Fig.-1. For the ULW samples, the

concentration of ∑PL was between 36.4 and 93.5 µg L⁻¹, for the SML samples between 61.0 and 118.1 µg L⁻¹. The measured

5   total lipid concentrations are within the concentration range of the previous measurement studies in different oceanic regions (Frka et al., 2011;Gašparović et al., 2014;Stolle et al., 2019). ~~The~~Within the PL, the lipid classes FFA (ULW: 5.4-14.0 µg L⁻¹; SML: 16.1-36.5 µg L⁻¹) and ~~phospholipids (PE, PG, PC) dominated within the PL~~PP (ULW: 15.2-54.9 µg L⁻¹; SML: 17.6-37.4 µg L⁻¹) had high concentrations in seawater, while other lipid classes ~~like~~such as TG, ~~HC,~~ (< 5.8 µg L⁻¹) and ST ~~showed not only~~(< 2.4 µg L⁻¹) had concentrations lowered by a ~~much lower concentration but also much less pronounced~~

10   ~~variance~~factor of 4-23. Another interesting feature is the opposite abundance of the two phospholipids PE and PG. On sampling days when the PE ~~had a high~~ concentration~~, PG was significantly less concentrated as found in the middle of the campaign~~ was high (e.g. 3/10/2017 ~~and 4~~: 41.8 µg L⁻¹; 04/10/2017: 41.9 µg L⁻¹), the PG concentration, however, was lower (3/10/2017: 8.8 µg L⁻¹; 04/10/2017: 12.6 µg L⁻¹, see Fig. 1), whereas towards the end of the campaign ~~(e.g. 07/10/2017 and 09/10/2017), when the concentration of PE becomes lower~~, the concentration of PE decreased by a factor of 4-5 (e.g. 7/10/2017: 7.6 µg L⁻¹;

15   9/10/2017: 9.8 µg L⁻¹), while the concentration of PG (7/10/2017: 14.8 µg L⁻¹; 9/10/2017: 14.4 µg L⁻¹) increased~~.~~ by a factor up to 2. The PE/PG ratio calculated from these data varied along the campaign with increasing values towards the middle of the campaign (maxima on 03/10/2017 and 04/10/2017) and decreasing values towards the end (Table S4), following the same trend as the total bacteria number (TCN, Table S3). This indicates a change in the lipid dominant biological contributions, with bacterial sources dominating in the first part and especially in the middle of the campaign. However, in the last part of

20   the campaign (~~afterwards on~~after the 05/10/2017) contributions to the lipid pool were rather dominated by phytoplankton. These differences between bacterial and phytoplankton sources are not reflected in the total lipid concentrations, because

degradation products ~~like~~such as FFA also contribute ~~strongly~~ to ~~the~~ total lipids with a high proportion (Fig. 1). For this reason, neither bacterial nor phytoplankton sources alone are the controlling drivers for the total lipid concentration, at least in the short term.

5 Figure 1: Concentration of individual particulate lipid classes in the ULW (sampling date) and the SML (sampling date_SML) samples along the campaign and as an averaged value in µg L^{-1}

3.1.2 Lipids and lipid classes in the dissolved fraction

Compared to the particulate fraction, ~~slightly~~ higher concentrations of total dissolved lipids were detected, by a factor between
10 1.1 and 1.4 with ∑DL: 39.8-128.5 µg L^{-1} in the ULW and with ∑DL: 55.7-121.5 µg L^{-1} in the SML samples (Fig. 2). The maximum concentrations ~~reported~~ here were ~~slightly~~also a factor of 1.3-1.4 higher than the total dissolved lipid concentrations reported by Frka et al. (2011) in the Mediterranean semi-enclosed temperate Adriatic sea (∑DL: 7.5-92.2 µg L^{-1}).

In contrast to the particulate lipids, HC showed the highest concentration and variation within the lipids ~~and~~as it varied between 6.6 up to 64.0 µg L^{-1} in the ULW and in the SML between 9.2-49.6 µg L^{-1} (Fig. 2). HC can have both anthropogenic and natural
15 sources (Scholz-Böttcher et al., 2009). Shorter *n*-alkanes (as 2-nonadecane) may have additional sources as mature organic

matter or petroleum contamination, but the main ~~sources~~source of the investigated HC surrogate (2-nonadecane) is marine phytoplankton (Scholz-Böttcher et al. (2009) and references therein).

Phospholipids, especially PE and PG, ~~as well as~~and FFA, which dominated ~~in~~ the particulate lipids, showed ~~significantly~~ lower concentrations by a factor of 1.1-2.1 within the total dissolved lipids. In contrast to the particulate fraction, both phospholipids

5   (PE and PG) showed ~~most~~ similar concentrations and percentages in the dissolved fraction~~.~~ (Fig. 2).

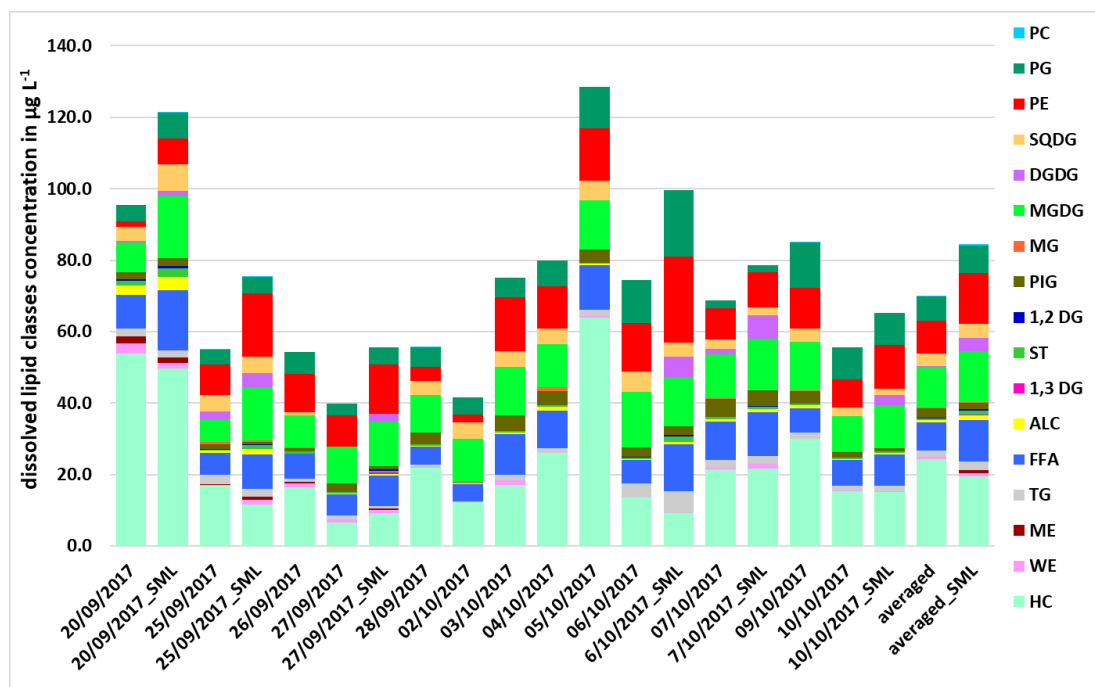
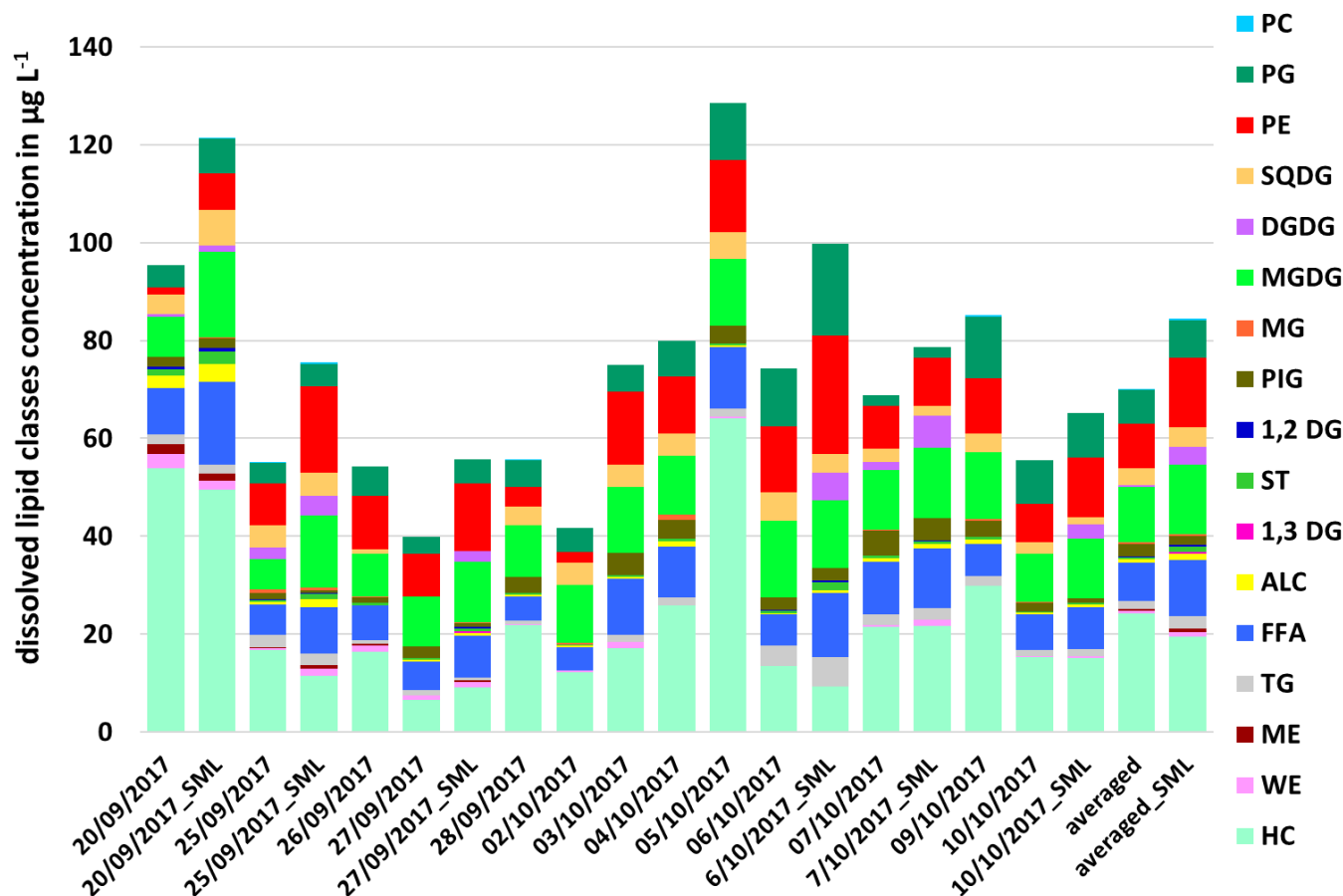



Figure 2: Concentration of individual dissolved lipid classes in the ULW (sampling date) and the SML (sampling date_SML) samples along the campaign and as an averaged value in µg L^{-1}

The percentage composition of the total dissolved lipids is in good agreement with the literature (Goutx et al., 2009;Marić et al., 2013). Altogether, our detailed analysis of the lipid classes shows that, although the concentration of PL and DL are overall very similar, the composition of the lipids compared to the PL and DL groups is partly different. This indicates that different production and degradation mechanisms for DL and PL contribute to the respective lipid composition.

3.1.3 Lipolysis Index

To evaluate the lipid degradative state and (bio)degradation processing, the LI (Eq. (1); Table S5) was calculated for the lipids in the seawater samples. The LI in the particulate fraction varied between 0.13-0.31 in the ULW and between 0.37-0.66 in the SML samples. van Wambeke et al. (2001) reported LI values of particulate lipids (0.21-0.39) in the north-western Mediterranean Sea during the end of a phytoplankton bloom up to pre-oligotrophic conditions. In general, the LI indicated that the intact lipid classes were dominant compared to the degradation indices/metabolites for the particulate lipids and that the

lipids were thus degraded only to a small extent. This coincides well with the low concentrations of chlorophyll degradation products (PIG), suggesting that only moderate grazing took place and the (pigment-containing) organisms were fresh and in healthy condition (van Pinxteren et al., 2020). However, on specific days, the LI_{SML} of PL was ≥ 0.5 (Table S5), indicating a slightlyan increased OM/lipid degradation and metabolite release in the SML compared to the ULW. A higher LI in the SML

5   was also observed by Gašparović et al. (2014) in the East Atlantic Ocean and can be attributed to both bacterial and photochemical abiotic degradation (Christodoulou et al., 2009). The LI of DL (Table S5) varied between 0.13-0.53 in the ULW and between 0.20-0.48 in the SML samples, suggesting that the dissolved lipid classes were somewhat more resistant to degradation than the particulate lipids.

10   **3.1.4 Pigments, nutrients and microbiological investigations in seawater**

To further elucidate the biological production and degradation state of lipids, the lipid concentrations were related to a set of biological parameters, including indicators for autotrophic organisms (namely marine pigments, chl-*a*, *Nanoeucaryotes* and *Synechococcus-like cells)* and TCN as a proxy for bacterial abundance. Altogether, the concentrations of pigments andas well as autotrophic and heterotrophic cells indicated an oligotrophic system (detailed values in Tables S2/S3, further information

15   isare given in van Pinxteren et al. (2020)). In addition, the chl-*a* concentration in seawater increased from 0.11 µg L^{-1} to 0.60 µg L^{-1} (Table S2) during the campaign, but was generally low compared to other subtropical/tropical regions or worldwide (Duhamel et al., 2019).

The pigment measurements of the bulk water indicated temporal changes in the composition of the community and an increasing trend in pigment concentration towards the end of the campaign. A correlation of total lipids with the abundances

20   of chl-*a, Nanoeucaryotes and Synechococcus-like* cells was not observed. However, with regard to specific pigments beyond chl-*a*, a statistically relevant correlation between particulate PE and the pigment zeaxanthin with R=0.69 (p-value=0.03, n=10) was found in the ULW (Fig. S8a). Zeaxanthin has been reported as a proxy for chlorophytes and cyanobacteria (Grant and Louda, 2010) and for some microalgae (Galasso et al. (2017), and references therein). Furthermore, the pigment fucoxanthin, a marker for diatoms (Descy et al., 2009), showed a weak trend with particulate FFA (R=0.53, p-value=0.12, n=10) in ULW

25   samples (Fig. S8b)**.** The observed correlations/trends of zeaxanthin and fucoxanthin with individual particulate lipid classes pointed to a contribution of chlorophytes, cyanobacteria and diatoms towithin the lipid pool in our study.

We observed a positive and statistically relevant correlation between PE and TCN in the dissolved fraction of ULW samples (R=0.79, p-value=0.006, n=10, Fig. S9a). We also found a positive correlation between the particulate PE and TCN in the ULW (R=0.72, p-value=0.02, n=10), a similar trend was noticed for PE and TCN in the SML (R=0.64, p-value=0.36, n=4)

30   (Fig. S9b). The contribution of bacteria to the PE pool most likely results from the fact, that PE is part of the bacterial membrane (Stillwell, 2016;Gašparović et al., 2018). Additional indications for bacteria influencing the lipid pool resultsresult from a negative correlation between the lipolysis index of total particulate lipids (LI_{PL}) and TCN in the ULW and the SML samples. In the ULW, a correlation between LI_{PL} and TCN with R= -0.73 (p-value=0.02, n=10) (Figure S9c) was observed. In the SML (Figure S9c), a similar trend was noticeable for LI_{PL} and TCN (R= -0.87, p-value=0.13, n=4). These relationships may result

from the passive contribution (higher cell abundance and thus higher concentration of PE) of TCN to the phospholipids pool via PE. Phospholipids contribute to LI as part of the intact lipids (section 2.2.3, Eq. 1), which leads to a negative correlation with LI_{PL}. On the other hand, the ~~'metabolites'~~metabolites could be actively taken up by bacteria, which most likely happens when more bacteria are present. Due to the low abundance of *Synechococcus-like* cells, we assume that most bacteria counted

5 as TCN are heterotrophic and have taken up the ~~'metabolites'.~~metabolites. Although it remains unclear whether the bacteria have a passive (i.e. via membrane) or active (i.e. metabolism of the lipid 'metabolites') effect on the observed correlation between LI_{PL} and TCN~~, it~~. It is most likely that bacteria have influenced the lipid pool which is consistent with the results obtained from the lipid composition.

10 ## 3.2 Transfer of lipids from the Oceans

3.2.1   The comparability of the different marine matrices (seawater and aerosol particles)

The concerted measurements performed here also included spot samplings in the ocean (ULW, SML) during the sampling period of PM_1 aerosol particles at the CVAO (24h). The air masses arriving at the CVAO often followed the water current (Peña-Izquierdo et al., 2012;van Pinxteren et al., 2017) and suggest an enhanced link between the upper ocean and the aerosol

15 particles, as mainly winds drive the ocean currents in the upper 100 m of the ocean. The backward trajectories as well as the concentrations of inorganic ions and mineral dust tracers on the aerosol particles measured during the campaign, suggested a ~~predominantly~~predominant marine origin with low to medium dust influences (Triesch et al., ~~2020~~2021;van Pinxteren et al., 2020).

20 ### 3.2.2   Enrichment in the SML

The EF_{SML} was calculated using Eq. (2) to compare the concentration of the lipid classes in the SML samples with the ones in the ULW samples. The EF_{SML} of the total lipids and of lipid class representatives in the particulate and dissolved fraction is listed in Table S6. For the total lipids in the particulate fraction, the EF_{SML} varied between 1.0-1.7 (averaged $EF_{SML(\sum PL)}$: 1.4), whereas in the dissolved fraction it varied between 1.1-1.4 (averaged $EF_{SML(\sum DL)}$: 1.3). The EF_{SML} of the total lipids are

25 therefore quite similar for PL and DL.

A major aspect contributing to lipid enrichment in the SML might be explained by the physico-chemical properties of the respective lipid classes~~, namely~~ such as the surface activity (Table S7). The parameters describing this characteristic, i.e. the density, the partitioning coefficient between octanol and water (K_{OW}) and the topological polar surface area (TPSA), were compared with lipid enrichment. As shown in Table S7, the nonpolar lipids such as FFA and ALC have a higher surface

30 enrichment potential compared to the more polar glycolipids and phospholipids, but a correlation between the enrichment of the lipids and the surface activity was absent. In the dissolved fraction neither such a gradation of enrichments at the surface, nor a correlation between the enrichment and parameters describing the surface activity were visible. As Table S6 shows, similar EF_{SML} were found in the dissolved fraction for the individual lipid classes (EF_{SML}: 1.5 (FFA), 1.7 (ALC) and 1.6 (PP)). A comparison of lipid enrichment with other OM compounds showed that the SML enrichment of lipids seemed to be less

pronounced in contrast to other organic species such as amino acids (Reinthaler et al., 2008;Triesch et al., ~~2020~~2021)~~. This is somehow surprising, since SML enrichment of lipids is likely to be strongest among organic groups due to their surface activity~~, despite the high surface activity of the lipids (Burrows et al. (2014), and references therein). It has to be considered that ~~a~~the SML described here represents a layer with a thickness of about 100 µm (van Pinxteren et al., 2017) and therefore

5   gradients within this layer (e.g. an enhanced enrichment of surfactants only in the top layer of a few µm) cannot be considered here. The fact, that other (less surface active) compounds are ~~stronger~~more enriched in the SML (upper 100 µm) underlines the need to consider additional parameters to describe the SML enrichment of lipids in the ambient marine environment.

To this end, besides the physical processes leading to SML enrichment, the potential importance of OM-producers and -degraders (bacteria, phytoplankton and their released metabolites) were further investigated. Regarding the enrichment in the

10 SML within the lipid classes or ~~within~~ both fractions (the dissolved and particulate~~), stronger~~ one), clearer differences ~~are~~were found when looking at the individual lipid classes. The bacterial marker PE was enriched in the SML in the DL and PL fractions with an EF$_{SML(PE)}$ of 1.6 (PF) and ~~with~~of 2.1 (DF). In contrast, the phytoplankton marker PG was always depleted in the SML (EF$_{SML}$ < 1) in the PL and mostly enriched in the DL (averaged EF$_{SML(PG)}$ of 1.3). This is consistent with the observed permanent abundance and the slight enrichment of TCN and may indicate an enhanced bacterial activity in the SML. Furthermore, the

15 degradation lipid class FFA showed high enrichments in the SML (averaged EF$_{SML(FFA):}$ of 3.1 (PL) and 1.5 (DL)). These high concentrations and enrichments point to an enhanced biodegradation in the SML, which is consistent with previous observations (Frka et al., 2011;Gašparović et al., 2018), that lipids are more degraded ~~more strongly~~ in the SML (high LI) than in the ULW. Besides the increased abundance of microbial cells, this may also be due to a different microbial community composition between the SML and the ULW and thus~~,~~ to different functional diversity (Cunliffe et al., 2011). The metabolic

20 reserves lipids, represented by TG, showed the highest variability of enrichment in the SML along the campaign in the particulate fraction. In this fraction, EF$_{SML(TG)}$ varied between 0.3 and 4.4, resulting in an averaged enrichment of 2.3. The enrichment in the lipid classes of the dissolved fraction was less pronounced and always showed an opposite trend to PL, i.e. if the TG was highly enriched in the PL, it ~~was~~would have been less enriched in the DL and vice versa. This indicates that lipid reserves such as TG are stored within the particulate pool and upon dissolution become part of the dissolved pool.

25 Altogether, our results indicate that physico-chemical descriptors alone, which are related to the surface activity of the lipids, are not sufficient to describe the SML enrichment of the lipids, at least not in the top 100 µm. In-situ formation and degradation by phytoplankton and mainly bacterial processes, as shown here from the lipid classes patterns, also contribute to the abundance and SML enrichment of lipids in the ambient marine seawater.

30 **3.2.2 Measured PM$_1$ aerosol particle composition**

Up to now, the discussion about lipids on (marine) aerosol particles has only covered distinct classes of lipids (mostly fatty acids). Given this fact, this work firstly presents a comprehensive analysis of several lipid classes on marine aerosol particles. The atmospheric concentration of the total lipids in the PM$_1$ samples at the CVAO varied between 75.2 and 219.5 ng m^{-3} (average 119.9 ng m^{-3}), as shown in Fig. 3. ~~The atmospheric concentration at the CVAO~~It was 18.5 ng m^{-3} (8.7-33.9 ng m^{-3})

for FFA and 6.3 ng m^{-3} (3.4-9.8 ng m^{-3}) for ALC. ~~This was~~ at the CVAO. These values are in good agreement with Kawamura et al. (2003), who reported atmospheric concentrations between 0.19-23 ng m^{-3} (average 2.0 ng m^{-3}) for ALC and between 2.5-38 ng m^{-3} (average 14 ng m^{-3}) for FAA on marine aerosol particles from the western North Pacific. Other than that, Mochida et al. (2002) observed atmospheric concentration between 0.8-24 ng m^{-3} for saturated fatty acids (C$_{14}$-C$_{19}$) on marine aerosol

5   particles collected ~~over~~over the northern Pacific.

A high percentage contribution of the lipid classes TG, FFA and ALC (26.3-64.0 %, average 39.8 %) to the total lipids was observed. Especially, high percentages for TG (11.9-29.1 %, average 18.8 %) and FFA (8.4-31.2 %, average 15.4 %) were noticeable.

Compared to the seawater lipids, the atmospheric composition showed the same classes of lipids with ~~a stronger agreement~~

10   ~~to~~an increased consistency of the DL composition (high contribution of HC and lower contributions of ~~phospholipids~~PP).

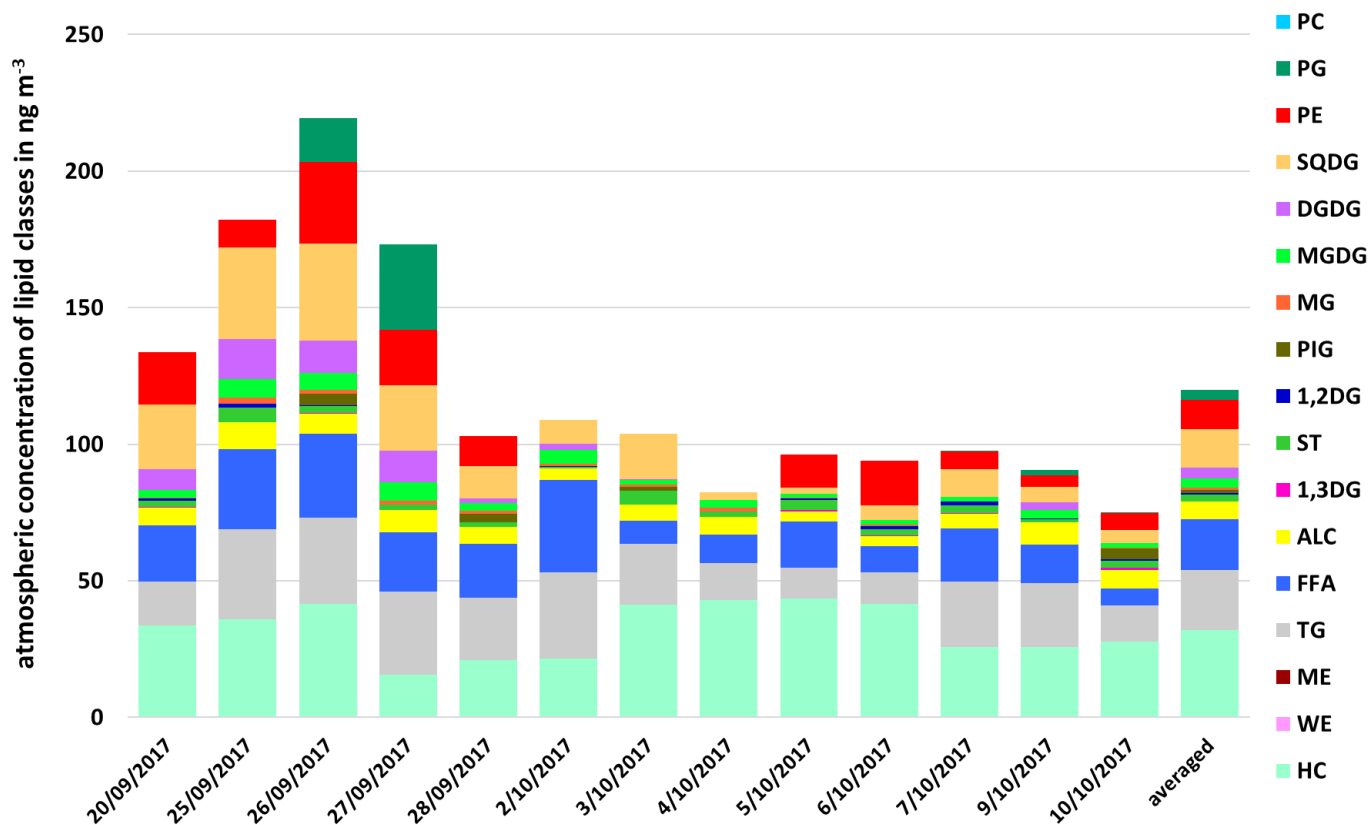
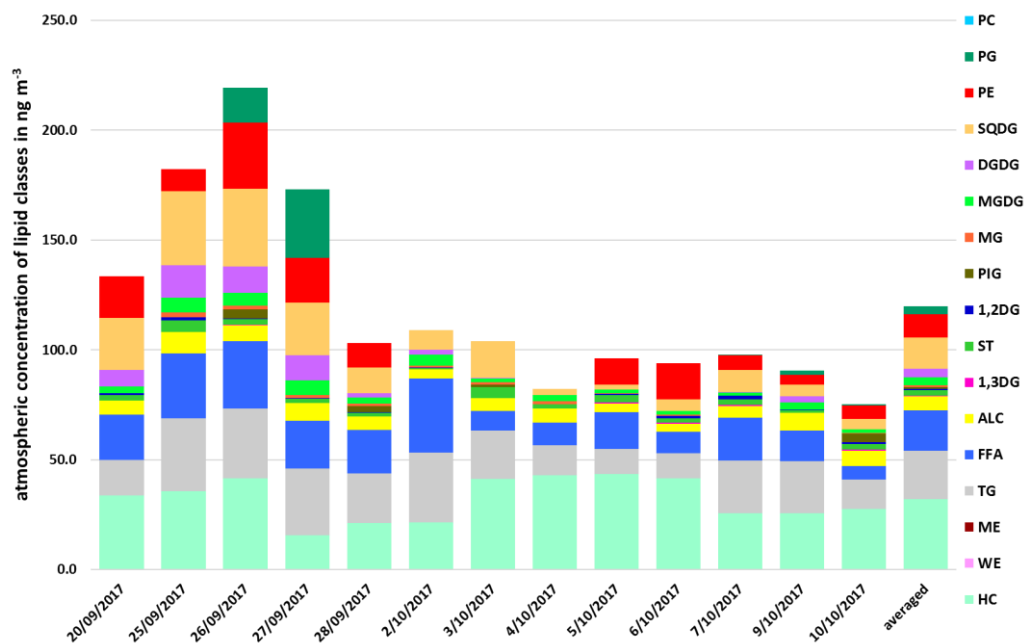



Figure 3: Atmospheric concentration of individual lipid classes in PM$_1$ aerosol particle samples and as an average at the CVAO in ng m^{-3}

However, although the atmospheric concentration of phospholipids was lower, ~~it was found that~~ PE was ~~always higher~~found to be more concentrated ~~as~~than PG, with only one exception on 27/09/2017 (Fig. 3). Since heterotrophic bacteria are reported

5 as a dominant source of PE (Michaud et al., 2018), this suggests that i) bacteria, possibly transported from the ocean into the atmosphere, contribute PE on the aerosol particles and/ or ii) PE is directly transferred from the ocean into the atmosphere, likely via bubble bursting.

3.2.3 Transfer of lipid classes from the ocean to the aerosol particles

10 An often applied parameter to quantify the transfer of OM from the ocean ~~to~~into the atmosphere, is the EF$_{aer}$ (e.g. Russell et al. (2010), van Pinxteren et al. (2017), ~~Triesch et al. (2020)). According to Eq. (3), the EF$_{aer(TL)}$ was calculated based on the dissolved total lipids in~~Triesch et al. (2021). According to Eq. (3), the EF$_{aer(TL)}$ was calculated based on the dissolved total lipids in the SML and varied between 9·10^4 and 7·10^5 with an average value of 3·10^5 (Fig. 4, Table S8). The EF$_{aer}$ based on the particulate total lipids in the SML was with an average of 2·10^5 very similar to the EF$_{aer}$ of the dissolved total lipids (3·10^5) as

15 discussed in the SI, Table S8. The data reported in the literature for enrichment factors of organic carbon or groups of OM ~~in~~on aerosol particles often originate from laboratory experiments, e.g. using controlled artificial bubbling ~~unit~~units (Quinn et al. (2015), and references therein). Rastelli et al. (2017) determined the enrichment of lipids~~,~~ as a sum parameter~~,~~ on submicron aerosol particles compared to seawater in a bubble-bursting experimental set-up and found an EF$_{aer}$ of 1·10^5. The good agreement of the EF$_{aer(TL)}$ derived from the ambient measurements reported here with the EF$_{aer}$ derived from laboratory

20 experiments under controlled conditions by Rastelli et al. (2017) indicates that the transfer of lipids from the SML to the aerosol phase under ambient conditions is consistent with processes described in laboratory studies.

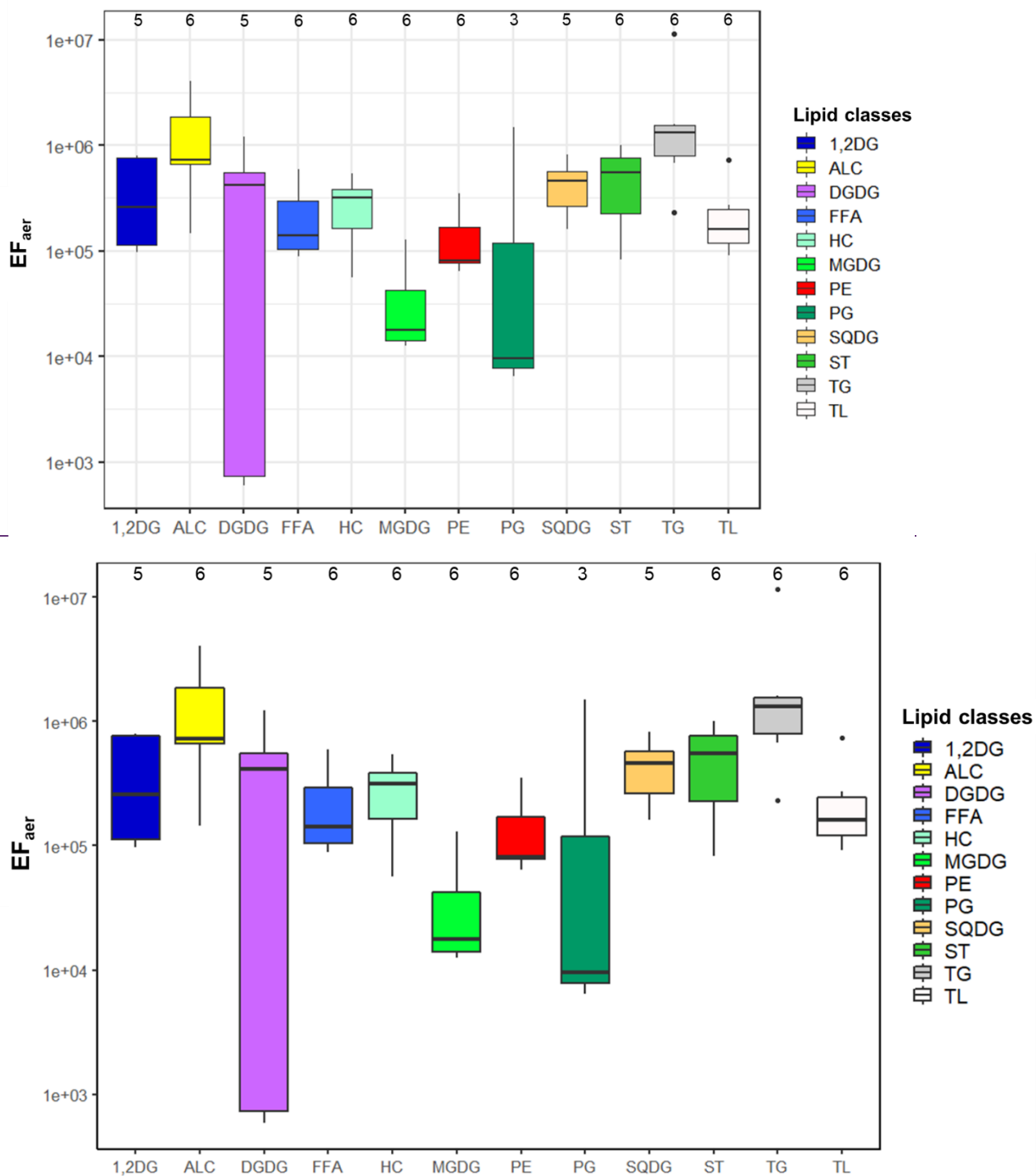



Figure 4: Boxplot of the enrichment factor aerosol (EF$_{aer}$) of the individual lipid classes and total lipids (TL) at the CVAO including the median, the 25th and 75th percentile; more explanation in Fig. S10

~~The EF_aer of~~The mean $EF_{aer(TL)}$ calculated and based on the ULW concentration is $3.4 \cdot 10^5$, and thus very similar to the mean $EF_{aer(TL)}$ based on the SML concentrations ($2.6 \cdot 10^5$). This can be attributed to the fact that the lipid concentrations in the ULW and SML were in the same concentration range, resulting in a comparatively low enrichment in the SML (EF_{SML}: 1.0-1.7, section 3.2.2).

5   All in all, the EF_{aer} of the total lipids was about one to two orders of magnitude higher than the EF_{aer} of free amino acids ($4 \cdot 10^2$-$3 \cdot 10^4$) on submicron aerosol particles measured during the same campaign ~~(Triesch et al., 2020), underlining the preferred transfer of lipids from the ocean to the atmosphere. In contrast to SML enrichment, the higher enrichment of lipids on aerosol particles observed here corresponds well with~~(Triesch et al., 2021) and underlining the preferred transfer of lipids from the ocean into the atmosphere. In contrast to the SML enrichment, the higher enrichment of the lipids on the aerosol particles

10 observed here corresponds to the high surface activity of the lipids and the preferred adsorption to (bubble) surfaces resulting in a strong sea-to-air-transfer (Tervahattu et al., 2002;Facchini et al., 2008;Cochran et al., 2016a;Schmitt-Kopplin et al., 2012;Rastelli et al., 2017), and their possible association with other compounds promoting co-aerosolization processes (Quinn et al., 2015;Hoffman and Duce, 1976;Rastelli et al., 2017). Further possible transport mechanisms are discussed in section 3.2.4.

15 A gradient regarding the EF_{aer} of the individual lipid classes was found, showing that some of them were enriched to a larger extent than others (Fig.4, Table S9). Such differences between the lipid classes have not been reported so far because the lipids were mainly measured as a sum (Rastelli et al., 2017) or only a specific lipid class (as FFA, HC) ~~has been~~was investigated (Cochran et al., 2016b;Marty et al., 1979). In our data set we observed that TG ($EF_{aer(TG)}$: $3 \cdot 10^6$) followed by ALC were most enriched ($EF_{aer(ALC)}$: $1 \cdot 10^6$), while MGDG showed a lower enrichment with the $EF_{aer(MGDG)}$: $4 \cdot 10^4$. For the SML enrichment,

20 the parameters describing the surface activity (density, K_{OW} and TPSA, Table S7), were compared with lipid enrichment. Lipid classes with a comparatively low surfactant activity (in relation to the TPSA value of the lipid class, Table S7) including PP and GL showed relatively lower enrichments ($EF_{aer(PP)}$: $2 \cdot 10^5$ and $EF_{aer(GL)}$: $3 \cdot 10^5$) compared to highly enriched ALC ($EF_{aer(ALC)}$: $1 \cdot 10^6$). Furthermore, a ~~mild connection~~statistically relevant correlation (R^2=0.45, p=0.028) was found between the log K_{OW} and the EF_{aer} of the individual lipid classes (Fig. S11), indicating that the compounds with higher log K_{OW} and ~~therefore~~

25 ~~stronger~~thus a higher lipophilicity are preferably enriched on the aerosol particles. For example, TG with the highest log K_{OW} value of 25.5 (Table S7) was observed to have the highest enrichment on aerosol particles ($EF_{aer(TG)}$: $3 \cdot 10^6$). In contrast, lipid classes with lower log K_{OW} values such as MGDG (log $K_{OW(MGDG)}$: -3.5) were characterized by lower enrichments ($EF_{aer(MGDG)}$: $4 \cdot 10^4$). The compounds that are highly enriched in the aerosol phase only partially correspond to their respective enrichment in the SML. TG and ALC showed a high enrichment in both ~~in~~the SML and ~~in~~on the aerosol particles. However, FFA~~,~~ that

30 showed a pronounced SML enrichment in PL ($EF_{SML(FFA)}$:3.1), exhibited only a medium enrichment ~~in~~on the aerosol particles compared to other lipid groups.

It needs to be emphasized~~,~~ that the calculated EF_{aer} provides the quantitative description of the transfer from the ocean ~~to~~into the atmosphere, but does not consider additional formation or degradation pathways of the lipids on the aerosol particles, including biological or photochemical atmospheric reactions and a transport from other than marine sources. In agreement

with the results of the SML enrichment, these results suggest that additional processes such as biotic formation and degradation influence the lipid abundance on the aerosol particles. It has been reported that microorganisms (Rastelli et al., 2017) and especially bacteria (Michaud et al., 2018) can be transported from the ocean into the atmosphere. Bacteria can be transferred to marine aerosol particles. Here, besides passive contribution (i.e. providing lipids to aerosol particles upon cell

5   disintegration), bacteria may also actively influence the OM composition of aerosols (i.e. lipid production or degradation). However, the extent of this passive and especially of ~~potential~~potentially active ~~bacterial~~bacteria contribution to the lipid pool of aerosols warrants further studies. In addition, photochemical oxidation processes can take place in the atmosphere, i.e. the conversion of FFA to ALC (Bikkina et al., 2019).

Overall, the detailed measurements of the lipid classes (within the concerted measurements) together with additional
10   parameters showed that although lipids are the ~~highly~~largest OM species ~~in~~on aerosols, which is in-line with their high surface activity, additional biological processes influence the lipid composition. ~~These need~~This needs to be further studied and considered in OM transfer models. More recent models recognized that the OM transfer must be modeled by including the individual OM groups (like lipids) rather than phytoplankton tracers like chl-*a* (Burrows et al., 2014). Indeed, these models describe the OM transfer on the basis of their physico-chemical properties, but our data suggest that in the ambient marine
15   environment, additional in-situ formation and degradation must also be considered in order to fully address the OM abundance in general and the lipid abundance in particular.

3.2.4 Discussion of possible transfer mechanisms

The transfer of the dissolved and particulate OM from the ocean ~~to~~into the atmosphere probably occurs via bubble bursting,
20   whereby bubbles rising through the water column absorb OM and, when bursting at the surface, release the OM to the aerosol particles via jet and film drops (Quinn et al. (2015), and references therein). The finding that both classes of lipids (DL and PL) are found on the aerosol particles (Fig. 5, Fig. S12) indicates that both types of lipids can be transferred from the seawater to the aerosol particles, e.g. via bubble bursting ~~the bubbles~~. A differentiation of the contribution of the dissolved and particulate lipid fractions in seawater to the formation of the aerosol particles was therefore not possible here.

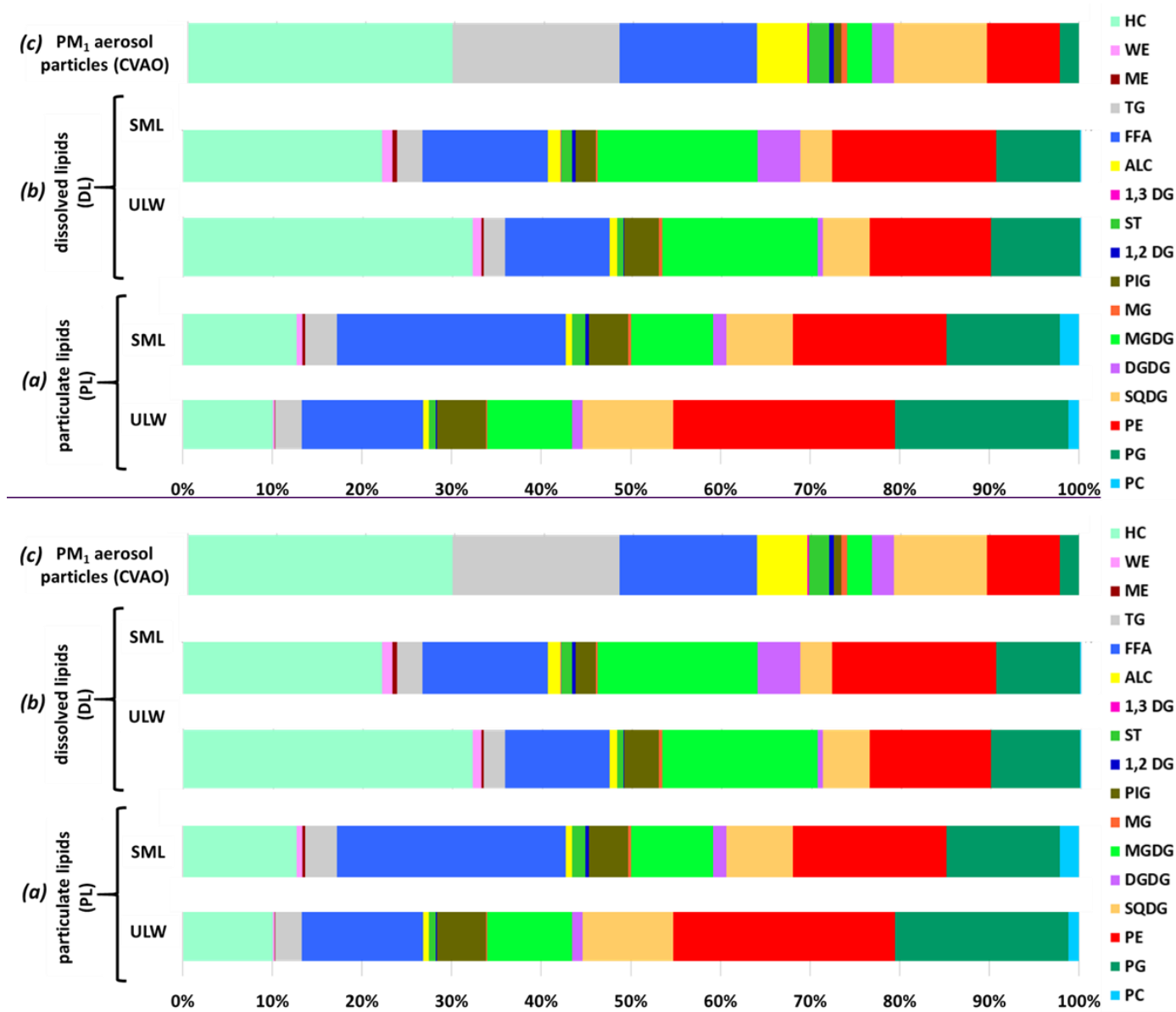
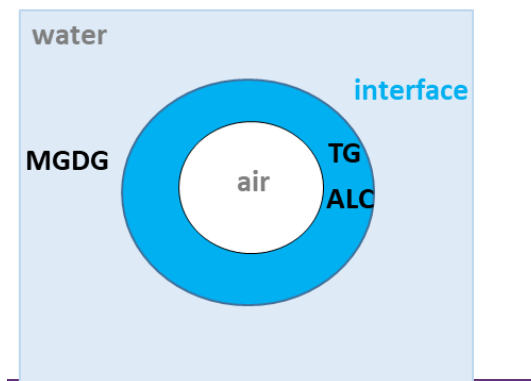



Figure 5: The percentage contribution of the individual lipid classes to the total lipids in the (a) particulate and (b) dissolved fraction in seawater (differentiation between ULW and SML) and on (c) PM₁ aerosol particles at the CVAO

However, regarding the process of the OM absorption on the bubbles in more detail, the OM absorbed on the bubble can be distributed either towards the gas or aqueous phase or can preferably reside within the bubble interface. To conceptually address the distribution of the OM towards the bubble-water-interface, we calculated the adsorption coefficient related to air (K_a) after Kelly et al. (2004) and additionally an adsorption coefficient of the analytes related to water (K_{aq}). The calculation included the measured SML concentrations, Henry's Law constants and the gas-phase saturation vapor pressure as explained and discussed in detail in Table S10. The comparison of the calculated K_a and K_{aq} values can give a hint on the distribution of

lipids at the bubble-air-water-interface. When $K_{aq}>>K_a$ (Fig. S17a), the analyte should be ~~preferred~~preferably distributed (from water) ~~to~~into air (inside the bubble). When $K_a>>K_{aq}$ (Fig. S17b) in turn, the analyte should be preferably distributed (from air) into water while the analyte should be ~~preferred~~preferably distributed within the bubble interface when $K_{aq} \sim K_a$ (Fig. S17c). Our data set showed that the analytes that had similar K_a and K_{aq} values (Table S10), namely TG and ALC, had the highest

5   EF_{aer} ($EF_{aer(TG)}$: $3 \cdot 10^6$ and $EF_{aer(ALC)}$: $1 \cdot 10^6$). This indicates that analytes, which are preferably distributed within the interface of the bubble, are transferred to the aerosol particles to a larger extent via the bubble bursting process, probably due to the higher stickiness to the interface (Fig. ~~6).~~

10   ~~Figure 6: Scheme of a bubble during the process of a bubble rising through the water column, distinguished between 'air' (inside the bubble), 'water' (surrounding the bubble), the 'interface' (bubble surface) and the distribution of the lipid classes MGDG, TG and ALC related to their K_a and K_{aq} values~~

S18). For MGDG, the lipid class with the lowest EF_{aer} ($4 \cdot 10^4$), the observed ratio was $K_a>>K_{aq}$, meaning that MGDG was preferably distributed ~~in~~into water. For the other lipid classes, however, we did not find such a connection between the adsorption coefficients (K_{aq}, K_a) and the aerosol enrichment (EF_{aer}). Nevertheless, the hypothesis that the transfer and

15   enrichment of the lipids ~~is~~are related to the distribution of ~~a~~ compounds within the bubble-air-water-interface, as observed for the lipid classes with extreme EF_{aer}, should be further investigated, preferably in controlled laboratory experiments.

3.3   Connection between lipids and INP activity in seawater

One main feature of biological components in general is their potential ability to contribute to ice nucleation and act as INP in

20   the atmosphere (Šantl-Temkiv et al. (2019) and references therein). To identify potential connections between lipid classes and INP activity in seawater, a statistical analysis was performed as described in section 2.2.5. As shown in detail in Gong et al. (2020), all samples collected for the present study contained both SML and ULW INP with concentrations of ~ 200 L^{-1} at temperatures of about -10 °C (Fig. S17), increasing to 10^7 L^{-1} at $\sim$ -25 °C. The existence of INP that are already ice active at temperatures above -15 °C indicated the presence of biogenic INP (Kanji et al., 2017;Šantl-Temkiv et al., 2019).

25   Both lipid fractions of the SML samples showed a positive trend towards the INP concentrations measured at -10 °C (R=0.72, p-value=0.28, n=4 for the total PL and R=0.69, p-value=0.31, n=4 for the total DL, Fig. S13a/b). This connection suggests the involvement of lipids in biogenic INP. Furthermore, the lipid classes which had shown a relationship with autotrophic or

heterotrophic organisms, namely PE and FFA (section 3.1.4), were investigated for their INP relationship. The relationship between PE and INP with regard to ULW cannot be discussed here, as the criteria defined in section 2.2.5 were not met (e.g. not enough matching data points were available). But a trend was found between the for the SML particulate PE and INP measurements at -10 °C (R=0.95, p-value=0.06, n=4, Fig. S13c). Similar relations between lipids and the INP activity have been reported previously (Govindarajan and Lindow, 1988;Palaiomylitou et al., 1998;DeMott et al., 2018). Palaiomylitou et al. (1998) reported that ice nucleation proteins were associated with phospholipids and showed that phospholipids, especially PE, not only contribute to the increased overall activity but also to the production of ice nuclei active at higher temperatures. To further test the biogenic INP activity, we analyzed the INP activity of the seawater samples before and after heating (95°C for 1 hour), since biogenic, especially proteinogenic, compounds are deactivated when heated to 100 °C (Šantl-Temkiv et al., 2019). It could be shown that a large proportion of INPs, that were active between -10 °C and -15 °C, lost their ice activity after the heating procedure (Fig. S15). As mentioned above, the deactivation of the INP function by heating is often associated with proteins. However, it has been shown that ice nucleating proteins have a connection to lipids and that interactions with membrane lipids, especially PE, are needed to maintain the conformational structure and functional activity of many membrane-bound proteins (Govindarajan and Lindow, 1988;Palaiomylitou et al., 1998). For this reason, the lipids might have a driving function in the INP activity of biogenic INP. In the SML samples, trends between INP measurements at -10 °C and the particulate FFA (R=0.84, p-value=0.16, n=4, Fig. S14a) and dissolved FFA (R=0.63, p-value=0.37, n=4, Fig. S14b) were observed. Moreover, a trend was found between the particulate FFA in the ULW and the INP measurements at -15 °C (R=0.64, p-value=0.025, n=12, Fig. S14c). DeMott et al. (2018) reported that the ice nucleation by particles containing long-chain fatty acids in a crystalline phase was relevant for the freezing of sea spray aerosols. Burrows et al. (2013) suggested that marine biogenic INP most probably played a dominant role in the INP concentrations studied in near-surface-air over the Southern Ocean. Wilson et al. (2015) proposed that there are ice active macromolecules in the OM of the SML. Moreover, they pointed out that global model simulations of marine organic aerosol in connection with their measurements indicated that marine OM might be an important source of INP in remote marine environments, e.g. the Southern Ocean, North Pacific Ocean and North Atlantic Ocean (Wilson et al. (2015), and references therein). The relationships presented here between the lipids in general and in particular the lipid classes with assigned biological context (PE, FFA) and INP activity at higher temperatures (-10 °C, -15 °C) in the ambient SML indicatingindicate that lipids in the tropical North Atlantic Ocean have the potential to contribute to (biogenic) INP activity when transferred tointo the atmosphere. However, it remains unclear to what extent INPs transferred from the ocean into the atmosphere contribute to the INP pool in the atmosphere, since further studies have identified other sources of INPs besides sea spray aerosol (Gong et al. (2020), and references therein).

4. Conclusion

Concerted measurements of lipids were performed in proximity to the Cape Verde Islands to compare the concentration of specific lipid classes inon submicron aerosol particles and in the dissolved and particulate fraction of seawater (ULW and SML). The analysis of lipid classes in seawater showed that, although concentrations in the particulate and dissolved fraction

are generally very similar, the contribution of lipids within both fractions differed. This suggests that different production and degradation mechanisms for DL and PL contribute to the respective lipid composition. On the aerosol particles, the lipid composition resembles the lipid composition of the dissolved fraction in seawater. ~~Although the lipids are reported as fast reactive compounds, our results suggest that the DL are somewhat more resistant to degradation.~~ The phytoplankton groups

5   chlorophytes, cyanobacteria and diatoms probably influence the lipid abundance as shown by pigment measurements. Our results indicate that, besides phytoplankton, bacteria also play an important role in lipid abundance in the oligotrophic North Atlantic, as shown by the PE/PG ratio and the abundance (and slight SML enrichment) of TCN. The concentration and enrichment of lipids in the ambient SML ~~is~~are not related to their physico-chemical properties describing the surface activity, at least not in the short term, probably due to parallel in-situ formation and degradation processes. This is underlined by the

10   fact that lipids in the top 100 µm of the SML are not as highly enriched as other (less surface active) ~~compound~~compounds, such as amino acids.

~~For aerosol, however, the high enrichment of total lipids corresponds well with the consideration of their high surface activity.~~ The EF_{aer} agrees with the considerations from modelling and laboratory studies that, among the marine OM groups, lipids are the most highly enriched compounds and indicates that the transfer of lipids from the SML to the aerosol phase in the complex

15   marine field is consistent with processes described in laboratory studies. In addition, the EF_{aer} of lipids was one to two orders of magnitude higher than the EF_{aer} of the less surface-active amino acids previously reported. In terms of the individual lipid groups on the aerosol particles, a ~~mild relation~~statistically significant correlation (R^2=0.45, p=0.028) between EF_{aer} and lipophilicity (expressed by the K_{OW} value) was observed, which was ~~missing in~~not present for the SML. In general, however, the parameters representing the surface activity of the lipid classes (density, K_{OW} value and TPSA) were not sufficient to

20   describe their transfer to the aerosol particles. The fact that bacteria are ~~strongly~~clearly involved in lipid abundance underlines that models using chl-*a* are not enough to describe OM in general. In addition, physico-chemical OM properties such as surface activity, are not sufficient to describe lipid abundance in the complex marine environment. Further processes such as biotic formation and degradation, as shown by the investigation of the individual lipid classes, contribute to lipid abundance in seawater, the SML and on aerosol particles in the marine environment and must be included in the consideration of lipid

25   transfer and finally in OM transfer models. Beyond that, our data suggest that the enrichment of the lipid classes on aerosol particles may be related to the distribution of the ~~lipid~~lipids on their respective adsorption coefficients in water (K_{aq}) and in air (K_a). Compounds which are preferably arranged within the bubble interface ($K_{aq}\sim K_a$), namely TG and ALC, are transferred to the aerosol particles to the highest extend. Finally, our results showed that lipids had the potential to contribute to (biogenic) INP activity when transferred ~~to~~into the atmosphere.

30   Altogether, we showed that the diverse group of lipids represent an important and complex OM group in seawater and on marine aerosol particles. To the best of the author's knowledge, the present study is the first to analyze several lipid classes simultaneously in seawater including the ULW and the SML and on submicron aerosol particles (PM_1) in such detail to obtain indications on their sources and sea-air linkage in the marine environment.

Data availability. The data are available through the World Data Centre PANGAEA under the following link: https://doi.pangaea.de/10.1594/PANGAEA.921832.

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Competing interest. The authors declare that they have no conflict of interest.

References

Arts, M. T., Ackman, R. G., and Holub, B. J.: "Essential fatty acids" in aquatic ecosystems: a crucial link between diet and human health and evolution, Canadian Journal of Fisheries and Aquatic Sciences, 58, 122-137, 10.1139/f00-224, 2001.

Barati, F., Yao, Q., and Asa-Awuku, A. A.: Insight into the Role of Water-Soluble Organic Solvents for the Cloud Condensation Nuclei Activation of Cholesterol, ACS Earth and Space Chemistry, 3, 1697-1705, 10.1021/acsearthspacechem.9b00161, 2019.

Becker, K. W., Collins, J. R., Durham, B. P., Groussman, R. D., White, A. E., Fredricks, H. F., Ossolinski, J. E., Repeta, D. J., Carini, P., Armbrust, E. V., and Van Mooy, B. A. S.: Daily changes in phytoplankton lipidomes reveal mechanisms of energy storage in the open ocean, Nature Communications, 9, 5179, 10.1038/s41467-018-07346-z, 2018.

Bhattacharya, B., and Habtzghi, D.: Median of the p Value under the Alternative Hypothesis, The American Statistician, 56, 202-206, 2002.

5   Bikkina, P., Kawamura, K., Bikkina, S., Kunwar, B., Tanaka, K., and Suzuki, K.: Hydroxy Fatty Acids in Remote Marine Aerosols over the Pacific Ocean: Impact of Biological Activity and Wind Speed, ACS Earth and Space Chemistry, 3, 366-379, 10.1021/acsearthspacechem.8b00161, 2019.

Bligh, E. G., and Dyer, W. J.: A rapid method of total lipid extraction and purification, Canadian Journal of Biochemistry and Physiology, 37, 911-917, 1959.

10 Burrows, S. M., Hoose, C., Pöschl, U., and Lawrence, M. G.: Ice nuclei in marine air: biogenic particles or dust?, Atmos. Chem. Phys., 13, 245-267, 10.5194/acp-13-245-2013, 2013.

Burrows, S. M., Ogunro, O., Frossard, A. A., Russell, L. M., Rasch, P. J., and Elliott, S. M.: A physically based framework for modeling the organic fractionation of sea spray aerosol from bubble film Langmuir equilibria, Atmos. Chem. Phys., 14, 13601-13629, 10.5194/acp-14-13601-2014, 2014.

15 Carpenter, L. J., Fleming, Z. L., Read, K. A., Lee, J. D., Moller, S. J., Hopkins, J. R., Purvis, R. M., Lewis, A. C., Muller, K., Heinold, B., Herrmann, H., Fomba, K. W., van Pinxteren, D., Muller, C., Tegen, I., Wiedensohler, A., Muller, T., Niedermeier, N., Achterberg, E. P., Patey, M. D., Kozlova, E. A., Heimann, M., Heard, D. E., Plane, J. M. C., Mahajan, A., Oetjen, H., Ingham, T., Stone, D., Whalley, L. K., Evans, M. J., Pilling, M. J., Leigh, R. J., Monks, P. S., Karunaharan, A., Vaughan, S., Arnold, S. R., Tschritter, J., Pohler, D., Friess, U., Holla, R., Mendes, L. M., Lopez, H., Faria, B., Manning, A. J., and Wallace, D. W. R.: Seasonal characteristics of tropical marine boundary

20 layer air measured at the Cape Verde Atmospheric Observatory, J. Atmos. Chem., 67, 87-140, 10.1007/s10874-011-9206-1, 2010.

Christodoulou, S., Marty, J.-C., Miquel, J.-C., Volkman, J. K., and Rontani, J.-F.: Use of lipids and their degradation products as biomarkers for carbon cycling in the northwestern Mediterranean Sea, Mar. Chem., 113, 25-40, 10.1016/j.marchem.2008.11.003, 2009.

Cochran, R. E., Jayarathne, T., Stone, E. A., and Grassian, V. H.: Selectivity Across the Interface: A Test of Surface Activity in the Composition of Organic-Enriched Aerosols from Bubble Bursting, J. Phys. Chem. Lett., 7, 1692-1696, 10.1021/acs.jpclett.6b00489, 2016a.

25 Cochran, R. E., Laskina, O., Jayarathne, T., Laskin, A., Laskin, J., Lin, P., Sultana, C., Lee, C., Moore, K. A., Cappa, C. D., Bertram, T. H., Prather, K. A., Grassian, V. H., and Stone, E. A.: Analysis of Organic Anionic Surfactants in Fine and Coarse Fractions of Freshly Emitted Sea Spray Aerosol, Environ. Sci. Technol., 50, 2477-2486, 10.1021/acs.est.5b04053, 2016b.

Cunliffe, M., Upstill-Goddard, R. C., and Murrell, J. C.: Microbiology of aquatic surface microlayers, FEMS microbiology reviews, 35, 233-246, 10.1111/j.1574-6976.2010.00246.x, 2011.

30 Cunliffe, M., Engel, A., Frka, S., Gašparović, B., Guitart, C., Murrell, J. C., Salter, M., Stolle, C., Upstill-Goddard, R., and Wurl, O.: Sea surface microlayers: A unified physicochemical and biological perspective of the air–ocean interface, Progress in Oceanography, 109, 104-116, 10.1016/j.pocean.2012.08.004, 2013.

Cunliffe, M. a. W., O.: Guide to best practices to study the ocean's surface. , Plymouth, UK, Marine Biological Association of the United Kingdom for SCOR, 118pp., 2014.

35 DeMott, P. J., Mason, R. H., McCluskey, C. S., Hill, T. C. J., Perkins, R. J., Desyaterik, Y., Bertram, A. K., Trueblood, Jonathan V., Grassian, V. H., Qiu, Y., Molinero, V., Tobo, Y., Sultana, C. M., Lee, C., and Prather, K. A.: Ice nucleation by particles containing long-chain fatty acids of relevance to freezing by sea spray aerosols, Environmental Science: Processes & Impacts, 20, 1559-1569, 10.1039/C8EM00386F, 2018.

Derieux, S., Fillaux, J., and Saliot, A.: Lipid class and fatty acid distributions in particulate and dissolved fractions in the north Adriatic sea,
40 Organic Geochemistry, 29, 1609-1621, 10.1016/S0146-6380(98)00089-8, 1998.

Descy, J.-P., Sarmento, H., and Higgins, H. W.: Variability of phytoplankton pigment ratios across aquatic environments, European Journal of Phycology, 44, 319-330, 10.1080/09670260802618942, 2009.

Duhamel, S., Kim, E., Sprung, B., and Anderson, O. R.: Small pigmented eukaryotes play a major role in carbon cycling in the P-depleted western subtropical North Atlantic, which may be supported by mixotrophy, Limnol. Oceanogr., 64, 2424-2440, 10.1002/lno.11193, 2019.

Engel, A., Bange, H. W., Cunliffe, M., Burrows, S. M., Friedrichs, G., Galgani, L., Herrmann, H., Hertkorn, N., Johnson, M., Liss, P. S., Quinn, P. K., Schartau, M., Soloviev, A., Stolle, C., Upstill-Goddard, R. C., van Pinxteren, M., and Zäncker, B.: The Ocean's Vital Skin: Toward an Integrated Understanding of the Sea Surface Microlayer, Frontiers in Marine Science, 4, 10.3389/fmars.2017.00165, 2017.

Facchini, M. C., Rinaldi, M., Decesari, S., Carbone, C., Finessi, E., Mircea, M., Fuzzi, S., Ceburnis, D., Flanagan, R., Nilsson, E. D., de Leeuw, G., Martino, M., Woeltjen, J., and O'Dowd, C. D.: Primary submicron marine aerosol dominated by insoluble organic colloids and aggregates, Geophys. Res. Lett., 35, doi:10.1029/2008GL034210, 2008.

Fomba, K. W., Muller, K., van Pinxteren, D., Poulain, L., van Pinxteren, M., and Herrmann, H.: Long-term chemical characterization of tropical and marine aerosols at the Cape Verde Atmospheric Observatory (CVAO) from 2007 to 2011, Atmos. Chem. Phys., 14, 8883-8904, 10.5194/acp-14-8883-2014, 2014.

Frka, S., Gašparović, B., Marić, D., Godrijan, J., Djakovac, T., Vojvodić, V., Dautović, J., and Kozarac, Z.: Phytoplankton driven distribution of dissolved and particulate lipids in a semi-enclosed temperate sea (Mediterranean): Spring to summer situation, Estuar. Coast. Shelf Sci., 93, 290-304, 10.1016/j.ecss.2011.04.017, 2011.

Frka, S., Pogorzelski, S., Kozarac, Z., and Ćosović, B.: Physicochemical Signatures of Natural Sea Films from Middle Adriatic Stations, The Journal of Physical Chemistry A, 116, 6552-6559, 10.1021/jp212430a, 2012.

Gagosian, R. B., Zafiriou, O. C., Peltzer, E. T., and Alford, J. B.: Lipids in aerosols from the tropical North Pacific: Temporal variability, Journal of Geophysical Research: Oceans, 87, 11133-11144, 10.1029/JC087iC13p11133, 1982.

Galasso, C., Corinaldesi, C., and Sansone, C.: Carotenoids from Marine Organisms: Biological Functions and Industrial Applications, Antioxidants (Basel), 6, 96, 10.3390/antiox6040096, 2017.

Gantt, B., Meskhidze, N., Facchini, M. C., Rinaldi, M., Ceburnis, D., and O'Dowd, C. D.: Wind speed dependent size-resolved parameterization for the organic mass fraction of sea spray aerosol, Atmos. Chem. Phys., 11, 8777-8790, 10.5194/acp-11-8777-2011, 2011.

Gašparović, B., Godrijan, J., Frka, S., Tomažić, I., Penezić, A., Marić, D., Djakovac, T., Ivančić, I., Paliaga, P., Lyons, D., Precali, R., and Tepić, N.: Adaptation of marine plankton to environmental stress by glycolipid accumulation, Marine environmental research, 92, 120-132, 10.1016/j.marenvres.2013.09.009, 2013.

Gašparović, B., Frka, S., Koch, B. P., Zhu, Z. Y., Bracher, A., Lechtenfeld, O. J., Neogi, S. B., Lara, R. J., and Kattner, G.: Factors influencing particulate lipid production in the East Atlantic Ocean, Deep Sea Research Part I: Oceanographic Research Papers, 89, 56-67, 10.1016/j.dsr.2014.04.005, 2014.

Gašparović, B., Kazazić, S. P., Cvitešić, A., Penezić, A., and Frka, S.: Improved separation and analysis of glycolipids by Iatroscan thin-layer chromatography-flame ionization detection, J. Chromatogr. A, 1409, 259-267, 10.1016/j.chroma.2015.07.047, 2015.

Gašparović, B., Kazazić, S. P., Cvitešić, A., Penezić, A., and Frka, S.: Corrigendum to "Improved separation and analysis of glycolipids by Iatroscan thin-layer chromatography–flame ionization detection" [J. Chromatogr. A 1409 (2015) 259–267], J. Chromatogr. A, 1521, 168-169, 10.1016/j.chroma.2017.09.038, 2017.

Gašparović, B., Penezić, A., Lampitt, R. S., Sudasinghe, N., and Schaub, T.: Phospholipids as a component of the oceanic phosphorus cycle, Mar. Chem., 205, 70-80, 10.1016/j.marchem.2018.08.002, 2018.

Gong, X., Wex, H., van Pinxteren, M., Triesch, N., Fomba, K. W., Lubitz, J., Stolle, C., Robinson, T. B., Müller, T., Herrmann, H., and Stratmann, F.: Characterization of aerosol particles at Cape Verde close to sea and cloud level heights – Part 2: ice nucleating particles in air, cloud and seawater, Atmos. Chem. Phys., 20, 1451–1468, 10.5194/acp-20-1451-2020, 2020.

Goutx, M., Acquaviva, M., and Gérin, C.: Iatroscan-measured phospholipids from marine microalgae, bacteria and suspended particles. Inform-International news on fats, oils and related materials, American Oil Chemists Society Publishers, 4, 516-517, 1993.

Goutx, M., Guigue, C., and Striby, L.: Triacylglycerol biodegradation experiment in marine environmental conditions: definition of a new lipolysis index, Organic Geochemistry, 34, 1465-1473, 10.1016/S0146-6380(03)00119-0, 2003.

Goutx, M., Guigue, C., D, D. A., Ghiglione, J. F., Pujo-Pay, M., Raybaud, V., Duflos, M., and Prieur, L.: Short term summer to autumn variability of dissolved lipid classes in the Ligurian sea (NW Mediterranean), Biogeosciences, 6, 1229-1246, 10.5194/bg-6-1229-2009, 2009.

Govindarajan, A. G., and Lindow, S. E.: Phospholipid requirement for expression of ice nuclei in Pseudomonas syringae and in vitro, The Journal of biological chemistry, 263, 9333-9338, 1988.

Grant, C. S., and Louda, J. W.: Microalgal pigment ratios in relation to light intensity: Implications for chemotaxonomy, Aquatic Biology, 11, 10.3354/ab00298, 2010.

Grasshoff, K., Kremling, K., and Ehrhardt, M.: Methods of Seawater Analysis - 3rd edition, edited by: Grasshoff, K., Kremling, Klaus, Ehrhardt, Manfred, Wiley-VCH Weinheim , Germany, 1999.

Guschina, I. A., and Harwood, J. L.: Algal lipids and effect of the environment on their biochemistry, in: Lipids in Aquatic Ecosystems, edited by: Kainz, M., Brett, M. T., and Arts, M. T., Springer New York, New York, NY, 1-24, 2009.

5   Hoffman, E. J., and Duce, R. A.: Factors influencing the organic carbon content of marine aerosols: A laboratory study, Journal of Geophysical Research (1896-1977), 81, 3667-3670, 10.1029/JC081i021p03667, 1976.

Kanji, Z. A., Ladino, L. A., Wex, H., Boose, Y., Burkert-Kohn, M., Cziczo, D. J., and Krämer, M.: Overview of Ice Nucleating Particles, Meteorological Monographs, 58, 1.1-1.33, 10.1175/amsmonographs-d-16-0006.1, 2017.

Kattner, G.: Lipid composition of Calanus finmarchicus from the north sea and the arctic. A comparative study, Comparative Biochemistry
10 and Physiology Part B: Comparative Biochemistry, 94, 185-188, 10.1016/0305-0491(89)90031-X, 1989.

Kawamura, K., Ishimura, Y., and Yamazaki, K.: Four years' observations of terrestrial lipid class compounds in marine aerosols from the western North Pacific, Global Biogeochemical Cycles, 17, 1003, 10.1029/2001GB001810, 2003.

Kelly, C. P., Cramer, C. J., and Truhlar, D. G.: Predicting Adsorption Coefficients at Air−Water Interfaces Using Universal Solvation and Surface Area Models, The Journal of Physical Chemistry B, 108, 12882-12897, 10.1021/jp037210t, 2004.

15 Khozin-Goldberg, I.: Lipid Metabolism in Microalgae, in: The Physiology of Microalgae, edited by: Borowitzka, M. A., Beardall, J., and Raven, J. A., 6, Springer International Publishing, 413-484, 2016.

Longhurst, A. R.: Chapter 9 - THE ATLANTIC OCEAN, in: Ecological Geography of the Sea (Second Edition), edited by: Longhurst, A. R., Academic Press, Burlington, 131-273, 2007.

Marić, D., Frka, S., Godrijan, J., Tomažić, I., Penezić, A., Djakovac, T., Vojvodić, V., Precali, R., and Gašparović, B.: Organic matter
20 production during late summer–winter period in a temperate sea, Continental Shelf Research, 55, 52-65, 10.1016/j.csr.2013.01.008, 2013.

Marie, D., Shi, X. L., Rigaut-Jalabert, F., and Vaulot, D.: Use of flow cytometric sorting to better assess the diversity of small photosynthetic eukaryotes in the English Channel, FEMS Microbiology Ecology, 72, 165-178, 10.1111/j.1574-6941.2010.00842.x, 2010.

Marty, J. C., Saliot, A., Buat-Ménard, P., Chesselet, R., and Hunter, K. A.: Relationship between the lipid compositions of marine aerosols, the sea surface microlayer, and subsurface water, Journal of Geophysical Research: Oceans, 84, 5707-5716, 10.1029/JC084iC09p05707,
25 1979.

Michaud, J. M., Thompson, L. R., Kaul, D., Espinoza, J. L., Richter, R. A., Xu, Z. Z., Lee, C., Pham, K. M., Beall, C. M., Malfatti, F., Azam, F., Knight, R., Burkart, M. D., Dupont, C. L., and Prather, K. A.: Taxon-specific aerosolization of bacteria and viruses in an experimental ocean-atmosphere mesocosm, Nature Communications, 9, 2017, 10.1038/s41467-018-04409-z, 2018.

Mochida, M., Kitamori, Y., Kawamura, K., Nojiri, Y., and Suzuki, K.: Fatty acids in the marine atmosphere: Factors governing their
30 concentrations and evaluation of organic films on sea-salt particles, Journal of Geophysical Research: Atmospheres, 107, AAC 1-1-AAC 1-10, doi:10.1029/2001JD001278, 2002.

Mueller, K., Lehmann, S., van Pinxteren, D., Gnauk, T., Niedermeier, N., Wiedensohler, A., and Herrmann, H.: Particle characterization at the Cape Verde atmospheric observatory during the 2007 RHaMBLe intensive, Atmos. Chem. Phys., 10, 2709-2721, 10.5194/acp-10-2709-2010, 2010.

35 Nguyen, Q. T., Kjær, K. H., Kling, K. I., Boesen, T., and Bilde, M.: Impact of fatty acid coating on the CCN activity of sea salt particles, Tellus B: Chemical and Physical Meteorology, 69, 1304064, 10.1080/16000889.2017.1304064, 2017.

Palaiomylitou, M. A., Kalimanis, A., Koukkou, A. I., Drainas, C., Anastassopoulos, E., Panopoulos, N. J., Ekateriniadou, L. V., and Kyriakidis, D. A.: Phospholipid Analysis and Fractional Reconstitution of the Ice Nucleation Protein Activity Purified fromEscherichia coli Overexpressing the in a Z Gene of Pseudomonas syringae, Cryobiology, 37, 67-76, 10.1006/cryo.1998.2102, 1998.

40 Parrish, C. C., Wangersky, P. J., Delmas, R. P., and Ackman, R. G.: Iatroscan-measured profiles of dissolved and particulate marine lipid classes over the Scotian Slope and in Bedford Basin, Mar. Chem., 23, 1-15, 10.1016/0304-4203(88)90019-9, 1988.

Parrish, C. C.: Lipids in Marine Ecosystems, ISRN Oceanography, 2013, 16, 10.5402/2013/604045, 2013.

Peña-Izquierdo, J., Pelegrí, J. L., Pastor, M. V., Castellanos, P., Emelianov, M., Gasser, M., Salvador, J., and Vázquez-Domínguez, E.: The continental slope current system between Cape Verde and the Canary Islands, 2012, 76, 14, 10.3989/scimar.03607.18C, 2012.

Perezgonzalez, J. D.: Fisher, Neyman-Pearson or NHST? A tutorial for teaching data testing

Front Psychol, 6, 223-223, 10.3389/fpsyg.2015.00223, 2015.

Quinn, P. K., Collins, D. B., Grassian, V. H., Prather, K. A., and Bates, T. S.: Chemistry and Related Properties of Freshly Emitted Sea Spray Aerosol, Chem. Rev., 115, 4383-4399, 10.1021/~~er5007139,~~cr500713g 2015.

5 Rastelli, E., Corinaldesi, C., Dell'Anno, A., Lo Martire, M., Greco, S., Cristina Facchini, M., Rinaldi, M., O'Dowd, C., Ceburnis, D., and Danovaro, R.: Transfer of labile organic matter and microbes from the ocean surface to the marine aerosol: an experimental approach, Scientific Reports, 7, 11475, 10.1038/s41598-017-10563-z, 2017.

Reinthaler, T., Sintes, E., and Herndl, G. J.: Dissolved organic matter and bacterial production and respiration in the sea-surface microlayer of the open Atlantic and the western Mediterranean Sea, Limnol. Oceanogr., 53, 122-136, 10.4319/lo.2008.53.1.0122, 2008.

10 Rinaldi, M., Fuzzi, S., Decesari, S., Marullo, S., Santoleri, R., Provenzale, A., von Hardenberg, J., Ceburnis, D., Vaishya, A., O'Dowd, C. D., and Facchini, M. C.: Is chlorophyll-a the best surrogate for organic matter enrichment in submicron primary marine aerosol?, Journal of Geophysical Research: Atmospheres, 118, 4964-4973, 10.1002/jgrd.50417, 2013.

Robinson, T.-B., Wurl, O., Bahlmann, E., Jürgens, K., and Stolle, C.: Rising bubbles enhance the gelatinous nature of the air–sea interface, Limnol. Oceanogr., 0, 10.1002/lno.11188, 2019.

15 Russell, L. M., Hawkins, L. N., Frossard, A. A., Quinn, P. K., and Bates, T. S.: Carbohydrate-like composition of submicron atmospheric particles and their production from ocean bubble bursting, Proc. Natl. Acad. Sci. U. S. A., 107, 6652-6657, 10.1073/pnas.0908905107, 2010.

Šantl-Temkiv, T., Lange, R., Beddows, D., Rauter, U., Pilgaard, S., Dall'Osto, M., Gunde-Cimerman, N., Massling, A., and Wex, H.: Biogenic Sources of Ice Nucleating Particles at the High Arctic Site Villum Research Station, Environ. Sci. Technol., 53, 10580-10590, 10.1021/acs.est.9b00991, 2019.

20 Schmitt-Kopplin, P., Liger-Belair, G., Koch, B. P., Flerus, R., Kattner, G., Harir, M., Kanawati, B., Lucio, M., Tziotis, D., Hertkorn, N., and Gebefügi, I.: Dissolved organic matter in sea spray: a transfer study from marine surface water to aerosols, Biogeosciences, 9, 1571-1582, 10.5194/bg-9-1571-2012, 2012.

Scholz-Böttcher, B., Ahlf, S., Vázquez-Gutiérrez, F., and Rullkötter, J.: Natural vs. anthropogenic sources of hydrocarbons as revealed through biomarker analysis: A case study in the southern Gulf of Mexico, Boletin de la Sociedad Geologica Mexicana, 61, 25 10.18268/BSGM2009v61n1a5, 2009.

Simoneit, B. R. T., and Mazurek, M. A.: Organic matter of the troposphere—II.∗∗For Part I, see Simoneit et al. (1977). Natural background of biogenic lipid matter in aerosols over the rural western united states, Atmospheric Environment (1967), 16, 2139-2159, 10.1016/0004-6981(82)90284-0, 1982.

Stillwell, W.: Chapter 5 - Membrane Polar Lipids, in: An Introduction to Biological Membranes (Second Edition), edited by: Stillwell, W., 30 Elsevier, 63-87, 2016.

Stolle, C., Ribas-Ribas, M., Badewien, T. H., Barnes, J., Carpenter, L. J., Chance, R., Damgaard, L. R., Quesada, A. M. D., Engel, A., Frka, S., Galgani, L., Gašparović, B., Gerriets, M., Mustaffa, N. I. H., Herrmann, H., Kallajoki, L., Pereira, R., Radach, F., Revsbech, N. P., Rickard, P., Saint, A., Salter, M., Striebel, M., Triesch, N., Uher, G., Upstill-Goddard, R. C., Pinxteren, M. v., Zäncker, B., Zieger, P., and 35 Wurl, O.: The MILAN campaign: Studying diel light effects on the air-sea interface, Bull. Amer. Meteorol. Soc., null, 10.1175/bams-d-17-0329.1, 2019.

Tervahattu, H., Juhanoja, J., and Kupiainen, K.: Identification of an organic coating on marine aerosol particles by TOF-SIMS, J. Geophys. Res.-Atmos., 107, 7, 10.1029/2001jd001403, 2002.

Triesch, N., van Pinxteren, M., Engel, A., and Herrmann, H.: Concerted measurements of free amino acids at the ~~Cape~~Cabo Verde ~~Islands: High~~islands: high enrichments in submicron sea spray aerosol particles and cloud droplets, Atmos. Chem. Phys.~~ Discuss., 2020, 1-24~~., 21, 40 163-181, 10.5194/acp-~~2019-976, 202~~21-163-2021, 2021.

van Pinxteren, M., Barthel, S., Fomba, K. W., Muller, K., von Tumpling, W., and Herrmann, H.: The influence of environmental drivers on the enrichment of organic carbon in the sea surface microlayer and in submicron aerosol particles - measurements from the Atlantic Ocean, Elementa-Sci. Anthrop., 5, 21, 10.1525/elementa.225, 2017.

van Pinxteren, M., Fomba, K. W., van Pinxteren, D., Triesch, N., Hoffmann, E. H., Cree, C. H. L., Fitzsimons, M. F., von Tümpling, W., 45 and Herrmann, H.: Aliphatic amines at the Cape Verde Atmospheric Observatory: Abundance, origins and sea-air fluxes, Atmos. Environ., 203, 183-195, 10.1016/j.atmosenv.2019.02.011, 2019.

van Pinxteren, M., Fomba, K. W., Triesch, N., Stolle, C., Wurl, O., Bahlmann, E., Gong, X., Voigtländer, J., Wex, H., Robinson, T. B., Barthel, S., Zeppenfeld, S., Hoffmann, E. H., Roveretto, M., Li, C., Grosselin, B., Daële, V., Senf, F., van Pinxteren, D., Manzi, M., Zabalegui, N., Frka, S., Gašparović, B., Pereira, R., Li, T., Wen, L., Li, J., Zhu, C., Chen, H., Chen, J., Fiedler, B., von Tümpling, W., Read, K. A., Punjabi, S., Lewis, A. C., Hopkins, J. R., Carpenter, L. J., Peeken, I., Rixen, T., Schulz-Bull, D., Monge, M. E., Mellouki, A., George, C., Stratmann, F., and Herrmann, H.: Marine organic matter in the remote environment of the Cape Verde islands – an introduction and overview to the MarParCloud campaign, Atmos. Chem. Phys., 20, 6921-6951, 10.5194/acp-20-6921-2020, 2020.

van Wambeke, F., Goutx, M., Striby, L., Sempéré, R., and Vidussi, F.: Bacterial dynamics during the transition from spring bloom to oligotrophy in the northwestern Mediterranean Sea: Relationships with particulate detritus and dissolved organic matter, Marine Ecology-progress Series - MAR ECOL-PROGR SER, 212, 89-105, 10.3354/meps212089, 2001.

Wakeham, S. G., Hedges, J. I., Lee, C., Peterson, M. L., and Hernes, P. J.: Compositions and transport of lipid biomarkers through the water column and surficial sediments of the equatorial Pacific Ocean, Deep Sea Research Part II: Topical Studies in Oceanography, 44, 2131-2162, 10.1016/S0967-0645(97)00035-0, 1997.

Wilson, T. W., Ladino, L. A., Alpert, P. A., Breckels, M. N., Brooks, I. M., Browse, J., Burrows, S. M., Carslaw, K. S., Huffman, J. A., Judd, C., Kilthau, W. P., Mason, R. H., McFiggans, G., Miller, L. A., Najera, J. J., Polishchuk, E., Rae, S., Schiller, C. L., Si, M., Temprado, J. V., Whale, T. F., Wong, J. P. S., Wurl, O., Yakobi-Hancock, J. D., Abbatt, J. P. D., Aller, J. Y., Bertram, A. K., Knopf, D. A., and Murray, B. J.: A marine biogenic source of atmospheric ice-nucleating particles, Nature, 525, 234-238, 10.1038/nature14986, 2015.

Wurl, O., and Holmes, M.: The gelatinous nature of the sea-surface microlayer, Mar. Chem., 110, 89-97, 10.1016/j.marchem.2008.02.009, 2008.

Wurl, O., Wurl, E., Miller, L., Johnson, K., and Vagle, S.: Formation and global distribution of sea-surface microlayers, Biogeosciences, 8, 121-135, 10.5194/bg-8-121-2011, 2011.

Yoshimura, K., Ogawa, T., and Hama, T.: Degradation and dissolution properties of photosynthetically-produced phytoplankton lipid materials in early diagenesis, Mar. Chem., 114, 11-18, 10.1016/j.marchem.2009.03.002, 2009.