Authors' Response

Reviewer #1:

We appreciate the helpful comments and feedback from Reviewer #1, which have helped improve our manuscript. We have carefully considered the recommendations of Reviewer #1 to shorten the manuscript, which was also suggested by Reviewer #2. To this end, we have shortened our introduction, moved the description of our denuder and filter preparation, handling, and extraction protocol to the supplement, and removed our discussion of the elevated vehicle [NH₃], which distracted from our main point of characterizing the isotopic composition of vehicle derived NH₃. Overall, these changes have shortened the manuscript by ~150 lines.

Reviewer #1 also pointed out that perhaps we should consider reporting the results of passive vs active collection for $\delta^{15}N(NH_3)$ characterization in a separate manuscript. We respectfully disagree with this suggestion and feel that this is an important result for this study, as pointed out by Reviewer #2. Specifically, this comparison helps put our measurements into context with previous studies that have reported very different $\delta^{15}N(NH_3)$ values derived from vehicle emissions. Our observation that a large $\delta^{15}N(NH_3)$ offset exists between active and passive sampling techniques reconciles differences in our measurements with previous literature reports and highlights the need for the reactive nitrogen isotope community to consider using robust, laboratory and field verified techniques shown to be accurate in characterizing $\delta^{15}N$. This is an incredibly important point that cannot be stressed enough. Therefore, we did not remove our comparison between active and passive NH₃ collection in the revised manuscript. Below we provide a point-by-point response to specific comments raised by Reviewer #1.

Specific Comments:

Comment: I think the title is somewhat misleading. You don't really constrain the vehicle ammonia emissions using N isotopes. The title as is suggests a source apportionment study, which is not the case. It should read: "Characterizing the isotopic composition of ammonia from vehicle plumes" or something like that.

Response: Thank you for this comment. We have changed the title in the revised manuscript to "Characterizing the Spatiotemporal Nitrogen Stable Isotopic Composition of Ammonia in Vehicle Plumes" to reflect the content of this work better.

Comment: Your abstract makes no mention of the comparison between active and passive collection techniques, which supports my previous point that you could remove that from your manuscript and have it in a separate paper. It reads as a sideways discussion in the present format and distracts the reader from the main findings. I am not saying it is not interesting and useful, just that it could be its own paper.

Response: We appreciate the comment, but we believe this comparison is an important finding and should be included in this manuscript, which is also aligned with the opinion of Reviewer #2. Thank you for pointing out that we did not mention this comparison in our abstract, which was an oversight, and also brought up by Reviewer #2. In the revised manuscript, we have changed the abstract to draw attention to the comparison between active and passive sampling, adding the following sentences, "Our recommended vehicle $\delta^{15}N(NH_3)$ signature is significantly different from previous reports. This is due to a large and consistent $\delta^{15}N(NH_3)$ bias of approximately -15.5 ‰ between commonly employed passive NH₃ collection techniques and the laboratory-tested active NH₃ collection technique." Additionally, we have restructured our section headings to help reduce the comparison between active and passive sampling to read as a sideways discussion. We have removed the subsection division in the results and discussion section between (1) Active NH_x Collection using a Denuder-Filter Pack and (2) Comparison Between Active and Passive Collection. We feel that this reduces the complicated subsection grouping and creates more streamlined results and discussion sections in the revised manuscript.

Comment: L37-39: I thought soil acidification is mostly due to HNO3. How can an alkaline compound like NH3 cause acidification?

Response: There are numerous processes in which NH_3 can cause soil acidification that has been well-documented, including plant uptake and assimilation, nitrification, and NH_3 volatilization. The uptake and assimilation of NH_4^+ results in a net release of H^+ as NH_4^+ is deprotonated during this process. Nitrification associated with the oxidation of NH_4^+ to NO_2^- and subsequent oxidation to NO_3^- will also lead to the net release of H^+ :

 $NH_4{}^+ + 2O_2 \rightarrow NO_3{}^- + H_2O + 2H^+$

Finally, during NH_3 volatilization, the pH of the soil surface will decrease as H^+ is released when NH_4^+ is converted to NH_3 :

 $N{H_4}^{\scriptscriptstyle +} \leftrightarrow N{H_3} + H^{\scriptscriptstyle +}$

To clarify how $NH_{3/4}$ can lead to acidification we have revised the sentence in question to the following, "Deposition of NH_3 and its secondary product, particulate ammonium (pNH_4^+), have critical environmental consequences, including soil acidification (via plant assimilation, nitrification, and NH_3 volatilization), eutrophication, and decreased biodiversity in sensitive ecosystems," and included an additional reference that does an excellent job reviewing soil acidification (Bolan, N.S., Hedley, M.J., White, R.E. Processes of soil acidification during nitrogen cycling with emphasis on legume based pastures. Plant and Soil, 134(1), 53-63, 1991.

Comment: L47: Helpful if you could indicate here NH3 atmospheric lifetime.

Response: Thank you for pointing this out. We have revised this sentence to include the NH₃ atmospheric lifetime to the following, "While agricultural activities are known to dominate the emission of NH₃, accounting for over 60 % of the global inventory (Bouwman et al., 1997), there

are significant spatiotemporal variabilities due to its short atmospheric lifetime that is on the order of a several hours to a day (Paulot et al., 2016) and its multitude of emission sources (e.g., Hu et al., 2014)."

Comment: L48-49: The Templer group in Boston has more recent studies highlighting large vehicle contributions to urban NH3 budget. Check out the Decina et al., papers, particularly relevant since you drove to Boston for this study

Response: We have added the following reference to the end of this sentence in the revised manuscript, "Decina, S. M., Templer, P. H., Hutyra, L. R., Gately, C. K. and Rao, P.: Variability, drivers, and effects of atmospheric nitrogen inputs across an urban area: emerging patterns among human activities, the atmosphere, and soils, Sci. Total Environ., 609, 1524–1534, 2017."

Comment: L61: Can you quantify here the contribution as a % at the global scale?

Response: To help shorten the manuscript, we have removed this sentence in the revised manuscript. We point out that vehicle emissions are an important urban source of NH_3 , "In urban regions, vehicle derived emissions have been identified as a major NH_3 source (Decina et al., 2017; Gong et al., 2011; Li et al., 2006; Livingston et al., 2009; Meng et al., 2011; Nowak et al., 2012; Sun et al., 2014, 2017). Recently, vehicle NH_3 emissions have been suggested to be a key driver of N deposition in urban and urban-affected regions (Fenn et al., 2018)."

Comment: L65: Once again, check the work lead by the Templer group in Boston about N deposition in urban areas

Response: Thank you for this comment. We have included additional references to Decina et al., 2017 and Decina, S. M., Hutyra, L. R. and Templer, P. H.: Hotspots of nitrogen deposition in the world's urban areas: a global data synthesis, Front. Ecol. Environ., 18(2), 92–100, 2020.

Comment: L66-67: how are "fuel-combustion" and "vehicle" source different? Isn't the latest included with the first?

Response: The original use of "fuel-combustion" was to refer to stationary fuel-combustion, such as electricity and heating generation. To improve clarity, we have changed "fuel-combustion" to "stationary fuel-combustion" in the revised manuscript.

Comment: L90-91 Didn't you just say that these techniques were shown to not accurately capture the δ 15N-NH3, based on work by Skinner et al.,? This seems contradictory.

Response: Thanks for the comment, and we have removed this sentence in the revised manuscript.

Comment: L. 158: How long is the inlet line?

Response: No inlet sampling line was used for the employed denuder-filter pack (ChemComb Speciation Cartridge), as the cartridges were directly exposed to ambient air as stated in section 2.2, "The samplers were directly exposed to ambient air without the use of an additional inlet tubing to prevent the loss of NH_3 ." The inlet described in Line 158 of the original manuscript refers to the air inlet of the ChemComb Speciation Cartridge, where ambient air is first introduced into the sampler. This piece is approximately 4 cm long, which has been indicated in the revised manuscript in Seciton 2.2, "The PTFE coated air inlet (~4 cm)…"

Comment: L.209: Did you characterize the potential inlet loss, and induced fractionation on NH3, to see if it was indeed negligible?

Response: Thank you for pointing this out, as we also feel that it is very important to consider potential inlet losses of reactive species to accurately characterize isotopic compositions, which had been previously largely ignored. In a previous study, we have conducted extensive laboratory experiments to document such potential sampling artifacts (Walters, W.W., Hastings, M.G. Collection of ammonia for high time-resolved nitrogen isotopic characterization utilizing an acidcoated honeycomb denuder. Anal. Chem., 90, 8051-8057, 2018). We evaluated the potential of the ChemComb inlet to induce fractionation by comparing NH₃ collections with (1) honeycomb denuders housed in the ChemComb Cartridge and (2) a gas scrubbing impinger that does not have an inlet (control), in which the NH₃ line was directly scrubbed in an acid solution. We found no statistical difference in $\delta^{15}N(NH_3)$ between the ChemComb sampler and the control, suggesting that inlet loss as a potential source of $\delta^{15}N(NH_3)$ fractionation was negligible. This result is also in agreement with Koutrakis et al., 1993 that reported no evidence for significant loss of NH₃ induced via the PTFE-coated sampling inlet. We have added the following sentence to the revised manuscript in Section 2.2 to demonstrate that we have considered the potential influence of the sampling inlet on inducing $\delta^{15}N(NH_3)$ fractionation, "The PTFE coated inlet has been shown to lead to a negligible loss of NH₃ and induce insignificant $\delta^{15}N(NH_3)$ fractionation (Koutrakis et al., 1993; Walters and Hastings, 2018)."

Comment: L.214: Any chance the denuders could trap a portion of the particulate phase as well on top of the gas phase?

Response: Thank you for raising this point. Particulates do not contribute to the final measurement from the denuder extracts due to the system design of the denuder-filter pack and operation conditions. We have added the following to the revised manuscript in section 2.2 to elaborate on this point, "Briefly, ambient air is drawn into the sampler and reactive gases are removed under laminar flow conditions such that radial mixing can only be achieved via diffusion-based processes. Particulates, with their much lower diffusion velocity compared to gases, cannot migrate to the walls of the denuder during the residence time within the unit and are collected on

a downstream filter pack. The samplers are also held vertically to limit the potential for gravitational settling of particles onto the denuder surfaces, such that particulates do not contribute to the denuder extract (Ali et al., 1989)."

Comment: L214: Can you give quantify your detection limits?

Response: Thank you for this comment. Limits of detection are based on off-line ion quantification as described in Section 2.3 in the revised manuscript. We have added the limit of detection (LOD) quantification for NH_4^+ , NO_2^- , NO_3^- , and SO_4^{2-} in section 2.3, "The limit of detection (LOD) of the quantified ions were no higher than 0.5, 0.2, 2.0, and 1.5 μ mol·L⁻¹ for $[NH_4^+]$, $[NO_2^-]$, $[NO_3^-]$, and $[SO_4^{2-}]$, respectively". To improve the clarity of the manuscript, we have moved the sentence on Line 214 in the original manuscript to after the instrumentation LOD was discussed in the revised manuscript.

Comment: L.219: pNO3-, but what about pNH4+?

Response: Thank you for raising this point. Nylon filters will quantitatively capture pNO_3^- but a significant fraction of pNH_4^+ will volatilize off this type of filter (see Walters, W.W, Blum, D.E., Hastings, M.G. Selective collection of particulate ammonium for nitrogen isotopic characterization using a denuder-filter pack sampling system, Anal. Chem, 91, 7586-7594, 2019 & Yu, X., Lee, T., Ayres, B., Kreidenweis, S.M., Malm, W., Collett, J.L. Loss of fine particulate ammonium from denuded nylon filters. Atmos Environ, 40, 4797-4807, 2006). However, a backup acid-coated filter will quantitatively capture any volatilized pNH_4^+ (Walters et al., 2019). We note that we had originally planned to quantify the inorganic anions collected on the filters in all measurement campaigns, which is why we planned to utilize both Nylon and citric acid coated filters. However, we found the extracted anion concentrations to be below our detection limits, such that this data was not reported for the stationary and mobile US measurements. If quantification of pNH_4^+ is the main goal (for concentration or isotopic analysis), a single acid-coated filter downstream from an acid-coated denuder should suffice, as we have pointed out in Walters et al., 2019.

We clarified this sentence in the revised manuscript to the following, "However, due to potential loss of semi-volatile NH₄NO₃, all subsequent campaigns utilized a Nylon filter (Cole-Parmer, 0.8 μ m pore, 47 mm diameter) which has been shown to collect and retain pNO₃⁻ quantitatively (Yu et al., 2005). A significant fraction of pNH₄⁺ collected on denuded Nylon filters may volatilize (Yu et al., 2006), such that a backup acid-coated (5 % citric acid (w/v) in water) cellulose filter (Whatman, 8 μ m pore, 47 mm diameter) is used to capture any volatilized NH₃ from the collected particles and/or NH₃ breakthrough during conditions of denuder saturation (Walters et al., 2019)."

Comment: L.241: What do you use ethanol for?

Response: This was used to wet the hydrophobic Teflon filter surface. We have clarified this sentence in the revised manuscript, "The PTFE filters were pre-wetted with 500 μ L of ethanol to wet its hydrophobic surface before extraction." We note that this text was moved to the Supplement (Text S1) in the revised manuscript to shorten the manuscript length.

Comment: L.380: Does it mean that the urban background NH3 has the isotopic composition of vehicle emissions?

Response: Thank you for this comment. I think concluding that urban background $\delta^{15}N(NH_3)$ is the same as vehicle emissions based on wind direction analysis at the near-highway stationary site would be incorrect since the measurement location is near a major NH₃ emission source. We have pointed this out in the revised manuscript, "Overall, the $\delta^{15}N(NH_3)$ values were not found to be significantly different when the monitoring site was upwind or downwind of I-95, with averages of 7.6±1.4 ‰ (n=3) and 7.1±1.8 ‰ (n=51), respectively (p>0.05), which is likely due to the proximity of the sampling location to airmasses significantly influenced by vehicle emissions."

Comment: L.395-401: I understand that you can't estimate f(NH3) accurately, but why can't you calculate the concentration of pNH4+ here? Were the Nylon filters also saturated? There is not mention of that aspecit it, and it should be expanded on.

Response: The Nylon filters were likely not "saturated", but pNH_4^+ collected on Nylon filters are subject to significant volatilization, as we have mentioned in section 2.2. Thus, we cannot quantitatively determine the NH_x speciation as pNH_4^+ extracted from the Nylon filter likely contains a negative artifact. The extracted pNH_4^+ extracted from the acid-coated filter represents both NH_3 breakthrough due to denuder saturation as well as some component of pNH_4^+ volatilization. We have further clarified why we can't quantitatively determine pNH_4^+ in section 3.2 of the revised manuscript, "Thus, our NH_x measurements are expected to be accurate, but there could be uncertainty in the NH_x speciation, because the NH_4^+ extracted from the acid-coated denuder and Nylon filter will have a low bias due to denuder saturation and pNH_4^+ volatilization, respectively, and NH_4^+ extracted from the acid-coated filter will derive from both NH_3 breakthrough and NH_3 volatilized from the Nylon filter."

Comment: L.409: An introduction sentence about what ISORROPIA is would be nice.

Response: In the revised manuscript, we have further elaborated on ISORROPIA as followed, " NH_x speciation was also estimated using ISORROPIA, which is a gas-aerosol equilibrium partitioning model (Fountoukis and Nenes, 2007; Nenes et al., 1998)."

Comment: L.421: I think it would be useful and interesting to provide, maybe in the SI, the isotopic composition for each component, especially the nylon-collected pNH4+. And maybe

expand on the different isotopic compositions of NHx and pNH4+, if such is the case (and I expect it to be).

Response: Thank you for this comment as this is an interesting point and one of the original goals of attempting to collect and speciate between NH₃ and NH₄⁺ simultaneously. However, due to our NH_x speciation problems in the tunnel measurements from NH₃ denuder saturation (as we have well-documented in section 3.2), it is impossible to relate the NH₄⁺ extracted from the acid-coated denuder, Nylon filter, and acid-coated filter to their atmospheric component due to numerous sampling artifacts. Therefore, we do not think it would be a good idea to discuss the $\delta^{15}N(NH_4^+)$ from the varying sampling media and attempt to relate them to NH₃ and pNH₄⁺, and this was the reason we presented the results in section 3.2 as $\delta^{15}N(NH_x)$. As requested, we have included a figure in the Supplement (Fig. S6) that shows the varying δ^{15} N values of NH₄⁺ extracted from the acid-coated denuder, Nylon filter, and acid-coated filter in the revised manuscript. In section 3.2 of the revised manuscript, we added the average δ^{15} N values of the varying sampling media, "The measured δ^{15} N from NH₄⁺ extracted from the acid-coated denuders, Nylon filters, and acid-coated filters averaged 6.0±5.6 ‰ (n=21), 1.0±10.7 ‰ (n=21), and -20.0±10.1 ‰ (n=21) (Figure S6)." Additionally, we note, "These $\delta^{15}N$ differences to some degree reflect differences in the $\delta^{15}N$ of ambient NH₃ and NH₄⁺ but are difficult to interpret due to the ambiguity in NH_x speciation." Since we have strong evidence that we captured 100% of NH_x, but did not accurately speciate NH_x , we focus our discussion in the text on $\delta^{15}N(NH_x)$.

Comment: L.431: Section title should be revised; it is the same as the previous section title

Response: Thank you for pointing this out. We have provided the correct subtitle name, "Mobile On-Road NH₃ Survey in Northeastern US" in the revised manuscript.

Comment: L.481: Please recall here what are elevated NH3 concentrations.

Response: Thank you for this comment. In the revised manuscript, we removed our discussion of the elevated vehicle [NH₃] to reduce the manuscript length and to draw attention to our $\delta^{15}N$ results, which is the main focus of this work.

Comment: L.521-523: Maybe recall that your f(NO3) is approximate in this case.

Response: Thank you for pointing this out. We believe Reviewer #1 is referring to $f(NH_3)$ and not $f(NO_3)$, and have included that the $f(NH_3)$ in our tunnel measurements were an approximation in the revised manuscript, "The temporal tunnel variability is not likely to be driven by $f(NH_3)$ partitioning influences as the estimated $f(NH_3)$ was not found to be significantly different between periods the tunnel was open or closed (p>0.05), indicating a significant change in NH₃/pNH₄⁺ partitioning did not occur during these periods."

Reviewer #2:

Thank you for your constructive feedback, comments, and suggestions, which have helped improve our manuscript. We agree that our finding of a large $\delta^{15}N$ offset between active and passive collection is significant, and we have revised our abstract to draw attention to this finding. In particular, we have added the following sentences, "Our recommended vehicle $\delta^{15}N(NH_3)$ signature is significantly different from previous reports. This is due to a large and consistent $\delta^{15}N(NH_3)$ bias of approximately -15.5 ‰ between commonly employed passive NH₃ collection techniques and the laboratory-tested active NH₃ collection technique," and added, "This work... and highlights the importance of utilizing verified collection methods for accurately characterizing $\delta^{15}N(NH_3)$ values," to the abstract.

We understand the concerns about the length of the manuscript, which was also raised by Reviewer #1. As suggested, we have simplified our discussion of background NH₃ influences from our onroad measurements in the revised manuscript to the following, "Furthermore, we do not expect background NH₃ contributions to have played a significant role in the spatial $\delta^{15}N(NH_3)$ variability observed from the on-road measurements in the Northeastern US. While lower $\delta^{15}N(NH_3)$ values in non-urban regions might be consistent with an increased contribution from background agricultural emissions which tend to have a low $\delta^{15}N(NH_3)$ signature (e.g., -31 to -14 ‰; Hristov et al., 2011), we expect these temperature-dependent emissions to be minimal during the winter when the on-road measurements were conducted." Additionally, we have shortened our introduction, moved the description of our denuder and filter preparation, handling, and extraction protocol to the supplement, and removed our discussion of the elevated vehicle [NH₃], which distracted from our main point of characterizing the isotopic composition of vehicle derived NH₃. Overall, these changes have shortened the manuscript by ~150 Lines. Below we provide a pointby-point response to specific comments raised by Reviewer #2:

Comment: Line 71 – clarify whether improvements refers to the sources or our understanding of them

Response: Here we aim to make the point that while $\delta^{15}N(NH_3)$ might be a potentially valuable tool for tracking NH₃ emissions, the number of $\delta^{15}N(NH_3)$ source characterization studies are limited. Thus, we need to enhance our $\delta^{15}N(NH_3)$ emission inventory before we can begin to utilize $\delta^{15}N(NH_3)$ as a quantitative tool for source apportionment. To clarify this point, we have changed this sentence in the revised manuscript to, "However, $\delta^{15}N(NH_3)$ source characterization studies are limited, particularly for non-agriculture NH₃ emissions (Chang et al., 2016; Felix et al., 2013; Freyer, 1978; Heaton, 1987; Smirnoff et al., 2012); thus, to quantitatively utilize this tracer for NH₃ source apportionment requires further improvements in $\delta^{15}N(NH_3)$ source emission signatures and an increased understanding of spatiotemporal variabilities."

Comment: Line 76-78 – This sentence is worded unclearly.

Response: Thank you for pointing this out. We have simplified this sentence in the revised manuscript as follows, "To quantitatively utilize $\delta^{15}N(NH_3)$ for NH₃ source apportionment requires distinguishable isotopic signatures, such that we need to understand the drivers behind the reported large variability in $\delta^{15}N(NH_3)$ from vehicle emissions."

Comment: Line 268-269 Is this sentence saying that the limit of detection for this method was higher than usual due to contamination? It's hard to follow the logic.

Response: When using the BrO-/azide chemical method for converting NH_4^+ to N_2O , we find a significant reagent N_2O -blank. This reagent blank makes it difficult to accurately and precisely characterize $\delta^{15}N$ for low concentration samples, such that we only conducted $\delta^{15}N$ analysis for samples with an $[NH_4^+] > 2 \mu mol \cdot L^{-1}$. We have clarified this point in the revised manuscript to the following, "Briefly, $\delta^{15}N(NH_4^+)$ was measured based on an established off-line wet-chemistry technique involving hypobromite (BrO⁻) oxidation and acetic acid/sodium azide reduction (Zhang et al., 2007), which was conducted for samples with $[NH_4^+] > 2 \mu mol \cdot L^{-1}$."

Comment: Line 319 Define what is meant by fblank

Response: The f_{Blank} refers to the fraction of collected NH₄⁺ that corresponds to the field blank. We have defined f_{Blank} in the text and rewrote the sentence in the revised manuscript to the following, "Blank $\delta^{15}N(NH_4^+)$ corrections were made for all samples when the fraction of the field blank ($f_{Blank} = [NH_4^+]_{blank}/[NH_4^+]_{total}$) were less than 30% of the total collected NH₄⁺, as the propagated $\delta^{15}N$ uncertainty generally did not exceed ±2.5 ‰ for this f_{Blank} value."

Comment: Line 430 Section 3.1.3 appears to have the wrong title

Response: Thank you for pointing this out. The correct subtitle "Mobile On-Road NH₃ Survey in Northeastern US" and has been changed appropriately in the revised manuscript.

Comment: Figure 7 – showing a median and interquartile range for two samples seems a bit excessive. Perhaps just report the two values as individual symbols.

Response: We agree and have adjusted Figure 7 in the revised manuscript, such that "On-Road (Trucking Routes)" only shows the two data points and not a statistical summary.

Comment: Line 671-Smirnoff is misspelled

Response: Thank you for pointing this out. We have fixed this mistake in the revised manuscript.

Characterizing the Spatiotemporal Nitrogen Stable Isotopic Composition of Ammonia in Vehicle Plumes

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Abstract. Vehicle emissions have been identified as an important urban source of ammonia (NH₃). However, there are large uncertainties regarding the contribution of vehicle emissions to urban NH_3 budgets, as well as its role in spatiotemporal fine particulate matter (PM_{2.5}) formation and nitrogen (N) deposition patterns. The N stable isotopic composition ($\delta^{15}N$) may be a useful observational constraint to track NH₃ emission sources and chemical processing, but previously reported vehicle $\delta^{15}N(NH_3)$ emission signatures have reported a wide range of values, indicating the need for further refinement. Here we have characterized $\delta^{15}N(NH_3)$ spatiotemporal variabilities from vehicle plumes in stationary and on-road measurements in the USA and China using an active NH₃ collection technique demonstrated to accurately characterize $\delta^{15}N(NH_3)$ on the order of hourly time resolution. Significant spatial and temporal δ^{15} N(NH₃) variabilities were observed and suggested to be driven by vehicle fleet composition and influences from NH₃ dry deposition on tunnel surfaces. Overall, a consistent δ^{15} N(NH₃) signature of 6.6±2.1 ‰ ($\bar{x}\pm 1\sigma$; n=80) was found in fresh vehicle plumes with fleet compositions typical of urban regions. Our recommended vehicle $\delta^{15}N(NH_3)$ signature is significantly different from previous reports. This difference is due to a large and consistent $\delta^{15}N(NH_3)$ bias of approximately -15.5 ‰ between commonly employed passive NH₃ collection techniques and the laboratorytested active NH₃ collection technique. This work constrains the δ^{15} N(NH₃) urban traffic plume signature, which has important implications for tracking vehicle NH_3 in urban-affected areas and highlights the importance of utilizing verified collection methods for accurately characterizing $\delta^{15}N(NH_3)$ values.

1. Introduction

Atmospheric ammonia (NH₃) is a critical component of the atmosphere and the global nitrogen (N) cycle (Behera et al., 2013; Galloway et al., 2004). As the primary atmospheric alkaline molecule, NH₃ plays an essential role in the neutralization of sulfuric acid (H₂SO₄) and nitric acid (HNO₃), leading to the formation of ammonium nitrate (NH₄NO₃), ammonium bisulfate (NH₄HSO₄), and ammonium sulfate ((NH₄)₂SO₄) (Behera and Sharma, 2012).

These compounds are the most abundant secondary components of inorganic fine particulate matter ($PM_{2.5}$), which has important implications for air quality, human health, visibility, and global climate change (Behera and Sharma, 2010; Updyke et al., 2012; Wang et al., 2015). Deposition of NH₃ and its secondary product, particulate ammonium (pNH₄⁺), have critical environmental consequences, including soil acidification (via plant assimilation, nitrification, and NH₃ volatilization), eutrophication, and decreased biodiversity in sensitive ecosystems (Bolan et al., 1991; Erisman et al., 2008; Galloway et al., 2004; Sutton et al., 2008). In recent years, N deposition in the form of NH_x (= NH₃ + pNH₄⁺) has come to dominate total inorganic reactive N deposition across most of the United States (Li et al., 2016). To evaluate the influence of NH₃ on climate and the environment, an accurate understanding of NH₃ atmospheric concentrations, emission sources, and spatiotemporal distributions are critical. However, the quantification of NH₃ emission budgets remains uncertain (Clarisse et al., 2009), and recent highresolution satellite NH₃ observations imply that anthropogenic emission inventories are substantially underestimated (Van Damme et al., 2018).

While agricultural activities are known to dominate the emission of NH₃, accounting for over 60 % of the global inventory (Bouwman et al., 1997), there are significant spatiotemporal variabilities due to its short atmospheric lifetime that is on the order of several hours to a day and its multitude of emission sources (e.g., Hu et al., 2014). In urban regions, vehicle derived emissions have been identified as a major NH₃ source (Decina et al., 2017; Gong et al., 2011; Li et al., 2006; Livingston et al., 2009; Meng et al., 2011; Nowak et al., 2012; Sun et al., 2014, 2017). Recently, vehicle NH₃ emissions have been suggested to be a key driver of N deposition in urban and urban-affected regions (Fenn et al., 2018). However, relating urban NH₃ emission sources to spatiotemporal N deposition patterns can be challenging due to the variety of potential emission sources that exist in the urban atmosphere including stationary fossil fuel combustion, waste containers, sewerage systems, transport from agricultural areas, and vehicles (Decina et al., 2017, 2020; Gong et al., 2011; Hu et al., 2014; Meng et al., 2011; Saylor et al., 2010; Sun et al., 2014, 2017; Sutton et al., 2000; Whitehead et al., 2007). The N stable isotopic composition (δ^{15} N) of NH₃ could be a valuable observational constraint to track source contributions and validate model apportionments (Felix et al., 2013; Felix et al., 2017). However, δ^{15} N(NH₃) source characterization studies are limited, particularly for nonagriculture NH₃ emissions (Chang et al., 2016; Felix et al., 2013; Freyer, 1978; Heaton, 1987; Smirnoff et al., 2012);

thus, to quantitatively utilize this tracer for NH₃ source apportionment requires further improvements in δ^{15} N(NH₃) source emission signatures and an increased understanding of spatiotemporal variabilities.

Tracking the contribution of vehicle NH₃ emissions might be possible using $\delta^{15}N(NH_3)$ (e.g., Felix et al., 2017). However, previous measurements of vehicle $\delta^{15}N(NH_3)$ signatures are limited and have reported a wide range of values from -17.8 to 0.4 ‰ (Chang et al., 2016; Felix et al., 2013; Smirnoff et al., 2012), which overlaps with agricultural derived NH₃ that has been measured to range from -15.2 to -8.9 ‰ in animal-sheds (Heaton, 1987; Frever, 1978). To quantitatively utilize δ^{15} N(NH₃) for NH₃ source apportionment requires distinguishable isotopic signatures, such that we need to understand the drivers behind the reported large variability in $\delta^{15}N(NH_3)$ from vehicle emissions. The previous vehicle δ^{15} N(NH₃) characterization studies have included tunnel monitoring in the United States (Felix et al., 2013), tunnel monitoring in China (Chang et al., 2016), and near-highway monitoring in Canada (Smirnoff et al., 2012), with reported $\delta^{15}N(NH_3)$ averages ($\bar{x}\pm 1\sigma$) of -3.4±1.2 ‰ (n=2), -14.2±2.6 ‰ (n=8), and -2.1 ± 1.9 % (n=11), respectively. We note that the observed variability may be related to spatiotemporal differences in the vehicle emitted $\delta^{15}N(NH_3)$, as the studies conducted in the US and Canada have reported relatively consistent values that are higher than that in China, but the factors influencing this potential spatiotemporal δ^{15} N(NH₃) pattern are unknown (Chang et al., 2016; Felix et al., 2013; Smirnoff et al., 2012). Notably, the reported $\delta^{15}N(NH_3)$ source measurements were conducted using a variety of NH₃ capture techniques for off-line $\delta^{15}N(NH_3)$ quantification that have included both passive samplers (Chang et al., 2016; Felix et al., 2013) and active collection using a filter pack (Smirnoff et al., 2012). Indeed, it has been shown that different active and passive NH_3 collection devices -including a gas-scrubbing bubbler, moss bag, shuttle sampler, and diffusion tube - resulted in significant δ^{15} N(NH₃) differences and variance when sampling the same emission source (Skinner et al., 2006). Thus, there could be inaccuracies in the previously reported $\delta^{15}N(NH_3)$ emission values related to the collection technique used to concentrate ambient NH₃ for off-line $\delta^{15}N(NH_3)$ characterization.

To improve the $\delta^{15}N(NH_3)$ source inventory for accurate NH₃ source apportionment, we need to quantify $\delta^{15}N(NH_3)$ using accurate methods and address spatiotemporal variabilities. In this study, $\delta^{15}N(NH_3)$ was characterized in a variety of integrated vehicle plumes with a combination of stationary and mobile on-road measurements, utilizing a laboratory-verified active collection technique shown to be accurate for $\delta^{15}N(NH_3)$ quantification (Walters and Hastings, 2018). Stationary measurements were conducted during the summer and winter at a near-highway monitoring site in Providence, RI, USA, and within a tunnel in Shenyang, Liaoning, China. A broad spatial survey of on-road mobile measurements was also conducted in the northeastern USA to evaluate the influences of a variety of real-world vehicle fleet compositions and driving modes on the traffic $\delta^{15}N(NH_3)$ signature. Passive NH₃ samplers, which have been used in previous $\delta^{15}N(NH_3)$ source characterization studies (Chang et al., 2016; Felix et al., 2013, 2017), were also deployed in the near-highway and tunnel monitoring campaigns and compared with the active collection technique verified for $\delta^{15}N(NH_3)$ accuracy (Walters and Hastings, 2018). Overall, this data will better define the $\delta^{15}N(NH_3)$ source signature for urban vehicle plumes, with implications for tracking emission contributions to urban atmospheric NH₃ concentrations and N deposition.

2. Site Description and Methods

2.1 Sampling Sites

2.1.1 Near-Highway Measurements (Providence, RI, USA)

Stationary measurements were conducted at an air monitoring station in Providence, RI, USA (41°49'46.0"N 71°25'03.0"W) maintained by the Rhode Island Department of Environmental Management (RI-DEM) and Rhode Island Department of Health (RI-DOH) during the summer and winter (Figure S1). The air monitoring station is located 4.62 m east of northbound I-95, a major interstate highway with a traffic volume of ~200,000 vehicles/day (HERE Traffic Analysis; https://company.here.com/automotive/traffic/traffic-analytics/), dominated by light-duty gasoline-powered vehicles. Continuous on-line measurements of CO (Thermo Scientific 48i) were monitored at the sampling location, and meteorological parameters, including temperature, relative humidity, wind speed, and wind direction, were recorded at the Urban League RI-DEM monitoring site, 2.4 km south of the near-highway site (Figure S1). Collections of speciated NH_x were conducted using an active sampling technique (denuder-filter pack; described in 2.2) with 6 h sampling intervals that included 00:30-6:30; 6:30-12:30; 12:30-18:30, and 18:30-00:30 during summer (August 9 to August 18, 2017) and 00:00-6:00, 6:00-12:00, 12:00-18:00, and 18:00-0:00 during winter (January 21 to February 1, 2018). During the sampling periods, NH_x collections were not conducted during

precipitation periods (or forecasted precipitation periods) due to the potential role of wet scavenging to alter $\delta^{15}N(NH_3)$ (Xiao et al., 2015). NH₃ was also collected using a passive sampler (ALPHA), in which NH₃ diffuses through a PTFE membrane and accumulates on an acid-coated (5 % citric acid (w/v) in water) cellulose filter (Albet, Grade 604, 24 mm diameter) housed in a protective case. Replicate passive samplers were deployed during the winter for NH₃ collection for two separate approximate 1-week collection periods during winter (February 10 – February 17 & February 17 – February 25 in 2018) for a total of four collected samples at the near-highway monitoring site. All samplers were secured on the roof (~3.85 m above ground) of the air monitoring station on the underside of a weatherproof shelter.

2.1.2 Tunnel Measurements (Shenyang, China)

From October 30 to November 5 in 2018, stationary tunnel measurements were conducted in the middle of an underground tunnel of North-South Expressway in Shenyang, Liaoning Province, China (41°48'16.0"N 123°26'54.0"E). This tunnel is approximately 2,360 m long, experiences approximately 28,804 vehicles/day during the weekday and 26,237 vehicles/day during the weekend (data from real-time traffic control system, Shenyang WuAi Tunnel Management co. LTD). The tunnel was open to vehicle passage from 5:00 to 23:00, and collections of speciated NH_x were conducted using a denuder-filter pack at 8 h intervals (approximately 6:00 to 14:00, 14:00 to 22:00, and 22:00 to 6:00). Sampling from 22:00 to 6:00 included the period that the tunnel was closed to vehicle passage (i.e., 23:00 to 5:00). The denuder-filter pack samplers were mounted on an elevated platform approximately 1.5 m above ground (Figure S2). Three ALPHA samplers were also mounted on the elevated platform and simultaneously collected NH₃ during the sampling campaign (~7 days). The relative humidity and temperature within the tunnel were monitored (iButton[®], DS1923, Wdsen Electronic Technology Co., Ltd) from October 31, 2018, at 14:00 to the end of the sampling campaign that included measurements for 16 out of the 21 collection periods.

2.1.3 Mobile On-road Measurements (Northeastern USA)

Mobile on-road measurements were conducted in the northeastern USA from February 20 to February 24, 2018, for approximately 21 hours and spanned ~2,125 km. The mobile laboratory consisted of a pick-up truck (Ford F-150) equipped with a denuder-filter sampling device, a CO analyzer (American Ecotech Serinus 30), a temperature and relative humidity probe (Elitech GSP-6), and a GPS tracking application (Map Plus). The denuder-filter pack samplers were placed in a weatherproof enclosure that was secured in the truck bed (~1 m above the truck bed), and

collections were conducted for approximately 1 h (Figure S3). Sampling was temporarily ceased during periods in which our vehicle speed was lower than 15 km hr⁻¹ to limit the possibility of sampling self-emissions. The CO analyzer was placed inside the truck and kept at a similar temperature to calibration conditions in the laboratory, and an air sampling inlet (PTFE tubing, 6.35 mm OD) was secured to the roof of the truck. Due to the significant power demands of the on-board instruments and collection equipment (i.e., vacuum pump), a gasoline-powered generator (Champion 1200-Watt Portable Generator) was used to power all equipment. The exhaust from the generator was diverted and emitted alongside the truck exhaust.

2.2 Active Collection of NH_x

Active speciated NH_x collection was conducted using a glass honeycomb denuder-filter pack sampling system (ChemComb Speciation Cartridge) during all campaigns. This collection system has been extensively described for its ability to speciate between reactive inorganic gases and particulate matter for off-line concentration determination (Koutrakis et al., 1988, 1993). Briefly, ambient air is drawn into the sampler, and reactive gases are removed under laminar flow conditions such that radial mixing can only be achieved via diffusion-based processes. Particulates, with their much lower diffusion velocity compared to gases, cannot migrate to the walls of the denuder during the residence time within the unit and are collected on a downstream filter pack. The samplers are also held vertically to limit the potential for gravitational settling of particles onto the denuder surfaces, such that particulates do not contribute to the denuder extract (Ali et al., 1989). The sampler consisted of a PTFE coated inlet to minimize reactive gas loss, a $PM_{2.5}$ impactor plate, a basic-coated honeycomb denuder (2% carbonate (w/v) + 1% glycerol (w/v) in 80:20 water: methanol (v/v) solution), acid-coated honeycomb denuder (2 % citric acid (w/v) + 1 % (w/v)glycerol (w/v) in 20:80 water:methanol (v/v) solution) to collect NH_3 , and a filter pack to collect pNH_4^+ . The basic-coated denuder was used to remove atmospheric acids (e.g., HNO₃, SO₂, and hydrochloric acid (HCl)) as a precaution to reduce collection-related gas-particle interactions. Recently, this sampling system has been shown to quantify $\delta^{15}N(NH_3)$ with a precision (±1 σ) of ±0.8 ‰ from laboratory experiments and field testing (Walters and Hastings, 2018). The PTFE coated air inlet (~4 cm) has been shown to lead to a negligible loss of NH₃ and induce insignificant δ^{15} N(NH₃) fractionation (Koutrakis et al., 1993; Walters and Hastings, 2018). The samplers were directly exposed to ambient air without the use of an additional inlet tubing to prevent the loss of NH₃.

In the first measurement campaign (near-highway monitoring in summer of 2017), pNH_4^+ was collected using a single Fluoropore PTFE membrane filter (Millipore, 1.0 µm pore, 47 mm diameter). However, due to potential loss of semi-volatile NH₄NO₃, all subsequent campaigns utilized a Nylon filter (Cole-Parmer, 0.8 µm pore, 47 mm diameter) that has been shown to collect and retain pNO_3^- quantitatively (Yu et al., 2005). A significant fraction of pNH_4^+ collected on denuded Nylon filters may volatilize (Yu et al., 2006), such that a backup acid-coated (5 % citric acid (w/v) in water) cellulose filter (Whatman, 8 µm pore, 47 mm diameter) is used to capture any volatilized NH₃ from the collected particles and/or NH₃ breakthrough during conditions of denuder saturation (Walters et al., 2019). All collections were conducted at a flow rate of 10 liters per minute (LPM) using a mass-flow controller (Dakota mass flow controller 6AGC1AL55-10AB2; precision ±1%) attached to an oil-less vacuum pump (Welch 2546B-01). All denuder and filter preparation, handling, and extraction techniques have been previously described (Walters et al., 2019; Walters and Hastings, 2018) and are summarized in the Supplement (Text S1).

2.3 Concentration and $\delta^{15}N(NH_x)$ Isotopic Analysis

The concentrations of the denuder and filter extraction solutions were analyzed using a combination of standardized colorimetry and/or ion chromatography analytical techniques. Colorimetric analysis included measurement of $[NH_4^+]$ based on the indophenol blue method with absorbance detection at 625 nm (e.g., US EPA Method 350.1), as well as $[NO_2^-]$ via diazotization with sulfanilamide dihydrochloride followed by detection of absorbance at 520 nm (e.g., US EPA Method 353.2) that was automated using a discrete UV-Vis spectrophotometer (Westco SmartChem Discrete Analyzer) at Brown University. These analyses were conducted for all samples collected in the US (i.e., near-highway and mobile measurements). Pooled standard deviations $(\pm 1\sigma)$ of replicate measurements of quality control standards were $\pm 0.35 \ \mu mol \cdot L^{-1}$ (n=48), and $\pm 0.23 \ \mu mol \cdot L^{-1}$ (n=60), and the average relative standard deviations (RSD) were 1.3 % and 0.81 % for $[NH_4^+]$ and $[NO_2^-]$, respectively. All samples collected in the Shenyang tunnel were analyzed for $[NH_4^+]$, $[NO_2^-]$, $[NO_3^-]$, and $[SO_4^{2-}]$ using ion chromatography (DionexTM CS12A column and CQ12A guard column with 10 mmol·L⁻¹ methanesulfonic acid as the eluent. Anions were quantified using a DionexTM AS22 column and AQ22 guard column with 4.5 mmol·L⁻¹ sodium carbonate and 1.4 mmol·L⁻¹ sodium bicarbonate as the eluent. For all quantified ions, the RSD was less than 1.5 %. The limit of detection (LOD) of the quantified ions were no higher

than 0.5, 0.2, 2.0, and 1.5 μ mol·L⁻¹ for [NH₄⁺], [NO₂⁻], [NO₃⁻], and [SO₄²⁻], respectively. The measured [NH₄⁺] was used to calculate the concentrations of NH₃ and pNH₄⁺ in the traffic plumes, while [NO₂⁻] was quantified because it will interfere with nitrogen isotopic analysis of NH₄⁺ (Zhang et al., 2007), but [NO₂⁻] was never measured above the LOD. The gases collected on the basic-coated denuder were generally below detection limits and were not reported in this work.

The quantification of $\delta^{15}N(NH_4^+)$ was performed separately for the acid-coated honeycomb denuder, the particulate filter, and the acid-coated cellulose filter extraction solutions, corresponding to NH₃, pNH₄⁺, and volatilized pNH₄⁺ (and/or NH₃ breakthrough during denuder saturation conditions), respectively. Briefly, $\delta^{15}N(NH_4^+)$ was measured based on an established off-line wet-chemistry technique involving hypobromite (BrO⁻) oxidation and acetic acid/sodium azide reduction (Zhang et al., 2007), which was conducted for samples with [NH₄⁺] > 2 µmol·L⁻¹. Samples were diluted to at least 10 µmol·L⁻¹ of NH₄⁺ and then oxidized to NO₂⁻ using BrO⁻ in an alkaline solution as previously described (Zhang et al., 2007). After a reaction time of at least 30 minutes, the reaction was stopped by 0.4 mL addition of 0.4 mol·L⁻¹ sodium arsenite to remove excess BrO⁻. The concentration of the product NO₂⁻ was then measured to confirm the quantitative conversion of NH₄⁺ to NO₂⁻. The product NO₂⁻ was reduced to nitrous oxide (N₂O) using sodium azide buffered in an acetic acid solution based on previously described chemical protocols (McIlvin and Altabet, 2005).

Samples were then analyzed for their $\delta^{15}N(N_2O)$ composition using an automated N₂O extraction system coupled to a continuous flow isotope ratio mass spectrometer for *m/z* 44, 45, and 46 measurements. These measurements were conducted at Brown University for samples collected within the USA and at the Institute of Applied Ecology, Chinese Academy of Sciences for samples collected within the Shenyang Tunnel. In each batch analysis, samples were calibrated relative to internationally recognized N isotopic NH₄⁺ reference materials. These reference materials underwent the same chemical processing as the samples and were used to correct for isotopic fractionation and blank effects resulting from the chemical conversion of NH₄⁺ to N₂O. At Brown University, two international reference materials were used that included IAEA-N2 and USGS25 with $\delta^{15}N(NH_4^+)$ values of 20.3 ‰ and -30.3 ‰, respectively (Böhlke et al., 1993; Gonfiantini, 1984). Repeated measurements of these reference materials yielded a standard deviation ($\pm 1\sigma$) of $\pm 0.65 \%$ (IAEA-N2; n=25) and $\pm 0.73 \%$ (USGS25; n=25) and an overall pooled standard deviation of $\pm 0.69 \%$ (n=50). At the Institute of Applied Ecology, Chinese Academy of Sciences, three reference materials were used that included IAEA-N1, USGS25, and USGS26 with δ^{15} N(NH₄⁺) values of 0.4 ‰, -30.3 ‰, and 53.7 ‰ (Böhlke et al., 1993; Gonfiantini, 1984), respectively. These materials had measured standard deviations of $\pm 0.53 \%$ (IAEA-N1; n=8), $\pm 0.24 \%$ (USGS25; n=8), and $\pm 0.45 \%$ (USGS26; n=8) and an overall pooled standard deviation of $\pm 0.42 \%$ (n=24). All N isotopic compositions are reported relative to reference standards using delta (δ) notation in units of per mil (‰).

$$\delta(\%_0) = 1000 \left(\frac{R_{sample}}{R_{ref}} - 1 \right)$$
(1)

where R is the ratio of the heavy to light isotope (i.e., ${}^{15}N/{}^{14}N$), for the sample and reference, respectively. Atmospheric nitrogen (N₂) is the established international delta-scale reference for N isotopic composition.

2.4 Data Analysis

The targeted analytes were corrected for field blanks, and ambient air concentrations were then calculated based on the volume of sampled air and reported in units of ppb_v and $\mu g \cdot m^{-3}$ for NH₃ and pNH_4^+ , respectively. The effective volume of air sampled by the ALPHA passive sampler was calculated as the following:

$$V = DAt/L$$

where V is the volume of sampled air (m³), D is the NH₃ diffusion constant (= $2.09 \times 10^{-5} \text{ m}^2 \text{ s}^{-1}$), A is the stationary air layer within the sampler (= $3.4636 \times 10^{-4} \text{ m}^2$), t is the time of exposure (h), and L is the cross-sectional area (= 0.006 m) (from ALPHA Sampler User Instructions).

The method detection limit (MDL) for $[NH_3]$ and $[pNH_4^+]$ determination for the active sampling technique (i.e., denuder-filter pack) was calculated as three times the standard deviation of the field blanks. The MDL was reported based on the typical collection times and reported separately for each sampling environment (Table 1). The reported $[NH_3]$ and $[pNH_4^+]$ precision using the denuder-filter pack sampling device was based on five separate replicate

sample collections conducted at the near-highway stationary site and expressed as the relative standard deviation (RSD %) (Table 1). The error bars of $[NH_3]$ and $[pNH_4^+]$ quantified using the denuder-filter pack in subsequent figures represent the ±RSD % when above the MDL. Some collections had $[pNH_4^+]$ below the MDL, and these samples were reported as 0.5MDL ± 0.5MDL. Multiple passive samplers (i.e., ALPHA) were always simultaneously collected, such that RSD % was not explicitly determined, and results were reported as $\bar{x}\pm 1\sigma$ of the multiple collections.

Significant NH₄⁺ field blanks were found on the acid-coated honeycomb denuder and the acid-coated cellulose filter. A subset of these blanks was analyzed for δ^{15} N(NH₄⁺) and found to have relatively consistent values of -9.6±1.3 ‰ (n=3) and -10.9±1.4 ‰ (n=3) for the acid-coated honeycomb denuder and acid-coated cellulose filter, respectively. Corrections for δ^{15} N were made based on mass-balance to account for the observed blanks as previously described (Walters et al., 2019):

$$\delta^{15}N(NH_4^+)_{sample} = \frac{\delta^{15}N(NH_4^+)_{total}[NH_4^+]_{total} - \delta^{15}N(NH_4^+)_{blank}[NH_4^+]_{blank}}{[NH_4^+]_{total} - [NH_4^+]_{blank}}$$

(3)

Blank $\delta^{15}N(NH_4^*)$ corrections were made for all samples when the fraction of the field blank ($f_{Blank} = [NH_4^*]_{blank}/[NH_4^*]_{total}$) was less than 30 % of the total collected NH₄⁺, as the propagated $\delta^{15}N$ uncertainty generally did not exceed ±2.5 ‰ for this f_{Blank} value. Samples with an f_{Blank} that exceeded 30 % were not reported for $\delta^{15}N$. This requirement as well as the azide method detection limit (i.e., 2 µmol·L⁻¹) limited our ability to quantify $\delta^{15}N(pNH_4^+)$ for samples collected at the near-highway monitoring site and mobile on-road measurements, such that only $\delta^{15}N(NH_3)$ was reported for the collections conducted in the USA. The collection media blank also impacted the mobile $\delta^{15}N(NH_3)$ measurements, as 6 out of 20 samples had a $f_{Blank} > 30$ %. Error bars reported for subsequent $\delta^{15}N$ values represent the propagated uncertainty that includes the collection uncertainty and the blank contribution. Replicate collected samples at the near-highway site indicated that $\delta^{15}N(NH_3)$ from NH₃ collected using an acid-coated denuder had an average reproducibility within 0.8 ‰ (n=5) (Table 1), consistent with previous field measurements (Walters and Hastings, 2018).

3. Results

3.1 Near Highway-Measurements (Providence, RI, USA)

Seasonal NH_x collections at the near-highway monitoring site were performed under a variety of environmental conditions (Table 2). Overall, the near-highway $[NH_3]$ ranged between 5.8 and 20.2 ppb_y during summer and 2.4 and 20.9 ppb_v during winter at the near-highway monitoring location (Figure 1a). The average [NH₃] ($\bar{x}\pm 1\sigma$) was 14.0 ± 4.0 ppb_v (n=32) and 12.0 ± 4.8 ppb_v (n=22) for summer and winter, respectively (Table 2), which was not found to be significantly different (p>0.05). Diel [NH₃] patterns were observed during both summer and winter, with significantly lower values occurring during the night/early morning collection period (Table 2). The dependence of [NH₃] on the vector averaged wind direction is shown in Figure 2. Overall, the near-highway monitoring site was downwind of I-95 for 51 out of 54 NH_x collection periods (Figure 2). The [NH₃] was significantly lower when the wind direction indicated the monitoring site was upwind of I-95 compared to when downwind of I-95, with averages of 5.8 ± 2.7 ppb_v (n=3), and 13.6 ± 4.2 ppb_v, respectively (p < 0.01). Strong positive linear correlations were found between $[NH_3]$ and the mean [CO] during each collection period during summer (r = 0.736, p < 0.01) and winter (r = 0.821, p<0.01), with a slope (mol:mol) of 0.025 ± 0.005 and 0.027 ± 0.005 , respectively (Figure 3). These observed [NH₃]:[CO] relations were similar to previously reported values of 0.031±0.005 from on-road measurements in New Jersey and California in the United States using high-resolution open-pathway sensors (Sun et al., 2014, 2017), and 0.031 to 0.038 based off fitted NH_x and CO slopes from aircraft measurements in the California South Coast Air Basin (Nowak et al., 2012). The similarity of these measurements indicated that the traffic plumes measured in this study were representative of previous literature reports in the USA, and the active collection of NH₃ using a denuder-filter pack sampling technique was suitable for reproducing accurate [NH₃] under traffic plume environmental conditions.

Roadside $[pNH_4^+]$ ranged from 0.045 to 0.938 µg·m⁻³ and from 0.117 to 2.327 µg·m⁻³ during summer and winter, respectively (Figure 1b). The average $[pNH_4^+]$ was 0.302±0.208 µg·m⁻³ (n=32) and 0.530±0.468 µg·m⁻³ (n=22) during summer and winter, respectively (Table 2), which was significantly different (p<0.05). During winter, an $[pNH_4^+]$ outlier of 2.327 µg·m⁻³ was identified based on a Grub's t-test (p<0.05). However, even with the removal

of this outlier, the seasonal $[pNH_4^+]$ average was found to be significantly different (p<0.05). NH_x speciation was quantified as f(NH₃):

$$f_{NH_3} = \frac{[NH_3] (mol)}{[NH_3 + pNH_4^+] (mol)}$$
(4)

Overall, $f(NH_3)$ ranged from 0.889 to 0.996 during summer and 0.878 to 0.986 during winter (Figure 1c), indicating that NH₃ was the dominant NH_x species during both summer and winter. The average $f(NH_3)$ was 0.972±0.022 (n=32) and 0.944±0.029 (n=22) during summer and winter, respectively (Table 2). The average seasonal $f(NH_3)$ was found to be statistically different (p<0.05), indicating a greater extent of NH₃ partitioning to pNH₄⁺ during winter. Significant correlations were observed between $f(NH_3)$ and relative humidity for both summer (r = -0.533, p<0.01) (Figure S4) and winter (r = -0.613, p<0.01) (Figure S5).

The measured $\delta^{15}N(NH_3)$ ranged from 2.6 to 9.3 ‰ and from 4.9 to 10.1 ‰ during the summer and winter, respectively (Figure 1d). The $\delta^{15}N(NH_3)$ average was 6.4±1.7 ‰ (n=32) and 8.1±1.4 ‰ (n=22) during summer and winter, respectively (Table 2), which were significantly different (p<0.05). The dependence of $\delta^{15}N(NH_3)$ on the vector averaged wind direction is shown in Figure 2. Overall, the $\delta^{15}N(NH_3)$ values were not found to be significantly different when the monitoring site was upwind or downwind of I-95, with averages of 7.6±1.4 ‰ (n=3) and 7.1±1.8 ‰ (n=51), respectively (p>0.05), which is likely due to the proximity of the sampling location to airmasses significantly influenced by vehicle emissions. No statistical difference was found between the collection period and $\delta^{15}N(NH_3)$ during the winter (p>0.05), but significantly lower $\delta^{15}N(NH_3)$ values were observed during the summer for the night/early morning sample (0:30 to 6:30) (p<0.05) (Table 2). Significant correlations between $\delta^{15}N(NH_3)$ and f(NH_3) were observed for both summer (r = 0.349, p<0.05) (Figure S4) and winter (r = 0.355, p<0.05) (Figure S5). However, these correlations were found to be impacted by an influential f(NH_3) value during the summer and winter of 0.889 and 0.878, respectively (Figures S4 and S5). Removing these influential f(NH_3) observations, resulted in an insignificant correlation between $\delta^{15}N(NH_3)$ and f(NH_3) for both summer (r = 0.300, p >0.05) (Figure S4) and winter (r = 0.378, p >0.05) (Figure S5).

3.2 Tunnel Measurements (Shenyang, Liaoning, China)

Tunnel temperature and relative humidity conditions remained relatively consistent throughout our sampling campaign and averaged 19.3 ± 1.6 °C and 35.4 ± 6.7 %, respectively (Table 3). Due to the elevated concentrations in the tunnel, the amount of collected NH₃ on the acid-coated honeycomb denuders averaged 406 ± 125 µg, indicating the laboratory determined operative capacity of ~400 µg was often exceeded (Walters and Hastings, 2018). The citric acid-coated filter collected no more than 275 µg of NH₃, which was within the laboratory determined operative capacity of at least 350 µg (Walters et al., 2019). Thus, our NH₄ measurements are expected to be accurate, but there could be uncertainty in the NH₈ speciation, because the NH₄⁺ extracted from the acid-coated denuder and Nylon filter will have a low bias due to denuder saturation and pNH₄⁺ volatilization, respectively, and NH₄⁺ extracted from the acid-coated filter will derive from both NH₃ breakthrough and NH₃ volatilized from the Nylon filter. Therefore, our concentration results and analysis of samples collected in the Shenyang tunnel will focus on [NH_x]. Overall, [NH_x] ranged from 64.4 to 210.6 ppb_v and averaged 132.5±45.8 (n=21) (Figure 4a; Table 3). An obvious [NH_k] diel cycle was observed in which higher concentrations occurred during periods that the tunnel was open compared to sampling periods in which the tunnel was closed to vehicle passage, with averages of 136.8±18.8 ppb_v (n=7), 181.2±23.0 ppbv (n=7), and 79.4±14.4 ppb_v (n=7), for the 6:00 to 14:00, 14:00 to 22:00, and 22:00 to 6:00 collection periods, respectively (Table 3).

We have estimated $f(NH_3)$, assuming that the pNH_{4^+} in $PM_{2.5}$ was linked to the $SO_4^{2^-}$ — NO_3^- — NH_{4^+} thermodynamic system, and that the influence of other ions (e.g., Na^+ , Ca^{2+} , or Cl^-) had a negligible impact on the chemistry of this system (Shah et al., 2018). Ion-mass balance was utilized to calculate the expected [pNH_{4^+}] for each collection period based on the measured [pNO_3^-] and [$pSO_4^{2^-}$] (Figure 4b) from the aqueous filter extracts:

$$[pNH_4^+](mol) = (2[pSO_4^{2-}] + [pNO_3^-])(mol)$$
(5)

Utilizing the ion-mass balance approach, $f(NH_3)$ was estimated to range between 0.856 to 0.997 and averaged 0.956±0.038 (Figure 4c; Table 3). NH_x speciation was also estimated using ISORROPIA, which is a gas-aerosol equilibrium partitioning model (Fountoukis and Nenes, 2007; Nenes et al., 1998). Model inputs included the measured $[NH_x]$, $[pNO_3^-]$, and $[pSO_4^{2-}]$, and average relative humidity and temperature for each collection period,

and the model was run in the forward direction in the meta-stable state. The $f(NH_3)$ was then calculated based on the model output of $[NH_3]$ and $[pNH_4^+]$ (Table S1). Overall, there was a near-exact agreement in $f(NH_3)$ between the ion-mass balance and the ISORROPIA approaches, noting that ISORROPIA was not used for the first five collection periods due to the absence of relative humidity and temperature data (Figure 4c). Overall, this analysis indicated that NH_x in the tunnel was primarily in the form of NH_3 , consistent with the near-highway stationary observations.

The measured $\delta^{15}N$ from NH₄⁺ extracted from the acid-coated denuders, Nylon filters, and acid-coated filters averaged 6.0±5.6 ‰ (n=21), 1.0±10.7 ‰ (n=21), and -20.0±10.1 ‰ (n=21) (Figure S6), respectively. These $\delta^{15}N$ differences to some degree, reflect differences in the $\delta^{15}N$ of ambient NH₃ and NH₄⁺ but are difficult to interpret due to the ambiguity in NH_x speciation. Since NH_x speciation was not achieved in the tunnel collections due to denuder saturation, our reported isotopic results and analysis will focus on $\delta^{15}N(NH_x)$, with the expectation that it primarily represents NH₃. The $\delta^{15}N(NH_x)$ was calculated for each sampling period using mass-balance:

$$\delta^{15}N(NH_x) = f_{NH_4^+-denuder}\delta^{15}N(NH_4^+)_{denuder} + f_{NH_4^+-Nylon}\delta^{15}N(NH_4^+)_{Nylon} + f_{NH_4^+-acid\ filter}\delta^{15}N(NH_4^+)_{acid\ filter}$$

where $f_{NH4+-denuder}$, $f_{NH4+-Nylon}$, and $f_{NH4+-acid filter}$, represents the fraction of NH₄⁺ extracted from the denuder, Nylon filter, and acid-coated filter, respectively for each sampling event. Overall, the $\delta^{15}N(NH_x)$ ranged from -1.6 to 9.2 ‰ (Figure 4d) and had a numerical average of 2.9±2.5 ‰ (n=21) (Table 3). There was a strong diel cycle in $\delta^{15}N(NH_x)$ in which the 22:00 to 6:00 collection period that included the period the tunnel was closed to vehicle passage (i.e., 23:00 to 5:00) resulted in a statistically lower $\delta^{15}N(NH_x)$ of 0.1 ± 1.3 ‰ (n=7), relative to the 6:00 to 14:00 and 14:00 to 22:00 collection periods that averaged 3.6 ± 1.0 ‰ (n=7) and 4.8 ± 2.0 ‰ (n=7), respectively (p<0.05) (Table 3).

3.3 Mobile On-Road NH₃ Survey (Northeastern USA)

Overall, the on-road [NH₃] ranged from 2.3 to 23.2 ppb_v and averaged 7.3±4.7 ppb_v (n=20) (Figure 5b). The highest [NH₃] were found to occur during collection periods near urban cores that included Boston, MA, Providence, RI, New York City, NY, and Washington, DC (Figure 6a). The on-road [NH₃] was significantly correlated with [CO] (r = 0.821, p<0.01), and the linear relationship between NH₃ and CO had a slope (NH₃(mol):CO(mol)) of 0.026±0.005 (Figure 3), which was similar to the near-highway relation of 0.025 ± 0.005 and 0.0270 ± 0.005 , observed during summer and winter, respectively. On-road [NH₃] was found to be significantly correlated with vehicle speed (r = - 0.673, p < 0.01) (Figure S7). On-road [pNH₄⁺] ranged from 0.047 to 0.710 µg·m⁻³ (Figure 5c) and averaged 0.204±0.176 µg·m⁻³ (n=20). NH_x speciation indicated NH₃ was the dominant species, consistent with our stationary observations, as f(NH₃) ranged from 0.800 to 0.987 (Figure 5d) and averaged 0.934±0.050 (n=20).

On-road $\delta^{15}N(NH_3)$ ranged from -3.0 to 10.1 ‰ (Figure 5e) and averaged 5.7±3.5 ‰ (n=14). On-road $\delta^{15}N(NH_3)$ was not found to be significantly correlated with f(NH₃) (r = 0.249, p>0.05) or average vehicle speed (r = -0.179, p>0.05) (Figure S7). Spatial mapping of $\delta^{15}N(NH_3)$ indicated the highest values near urban cores (Figure 6b). Each collection period was categorized as either a trucking or highway route using the percentage of annual average daily truck traffic contributions to annual average daily traffic (U.S. Dept of Transportation, 2013) similar to that previously described (Miller et al., 2017). Routes on our on-road measurements with diesel trucks that comprised at least 25 % of the annual average daily traffic and at least a yearly average of 8,500 diesel trucks per day were identified (U.S. Dept of Transportation, 2013), which were located on rural highways typically outside of urban areas. This categorization technique was used to qualitatively identify differences in vehicle fleet compositions during our measurements since real-time vehicle count data was not collected. Two sampling collection periods were identified as a trucking route, including (1) from outside Harrisburg, PA to New Smithville, PA along I-81 and I-78, and (2) from Kirkwood, PA to Colliersville, NY along I-81 and I-88. Though the number of measurements conducted on trucking routes was limited in this case study, the average on-road $\delta^{15}N(NH_3)$ on highway and trucking routes were 6.9 ± 1.9 ‰ (n=12) and -1.5 ± 1.6 ‰ (n=2), respectively, which were found to be significantly different (p<0.01).

3.4 Comparison Between Active and Passive NH₃ Collection

A comparison between the active and passive collection of NH₃ for concentration and $\delta^{15}N(NH_3)$ characterization is summarized in Table 4. The active collection sampling technique resulted in an $[NH_3]$ of 12.0 ± 1.2 ppb_v and 127.1 ± 12.5 ppb_v over the entire winter near-highway and Shenyang tunnel sampling campaigns, respectively. These concentrations were calculated from the total collected NH_{4^+} over the sampling campaign divided by the total volume of collected air for each respective campaign, and the reported uncertainty represents the RSD of the active collection technique of 9.8 %. We note that the $[NH_3]$ in the Shenyang tunnel determined using the denuder-filter pack was not measured directly but was calculated from the measured $[NH_x]$ and estimated $f(NH_3)$. The passive collection resulted in an $[NH_3]$ of 11.6±1.4 ppb_v (n=4) and 124±3.6 ppb_v (n=3), during winter at the near-highway monitoring location and in the Shenyang tunnel respectively, which was in close agreement with the active collection technique. The mass-weighted $\delta^{15}N(NH_x)$ using the active collection technique was 8.0±1.1 ‰ and 3.5±0.8 % during winter at the near-highway monitoring location and in the Shenyang tunnel, respectively, where the uncertainty represents the propagated error (Table 4). We note that the tunnel $\delta^{15}N(NH_3)$ technically represents δ^{15} N(NH_x); however, due to the elevated estimated f(NH₃), δ^{15} N(NH_x) ~ δ^{15} N(NH₃). The passive NH₃ collection technique resulted in an average $\delta^{15}N(NH_3)$ of -7.7±0.1 ‰ (n=4) and -11.7±0.3 ‰ (n=3) during winter at the nearhighway monitoring location and in the Shenyang tunnel, respectively, which was found to be significantly different from the $\delta^{15}N(NH_3)$ measured using the active collection for each sampling campaign (p<0.01). The $\delta^{15}N(NH_3)$ difference between passive and active collection was calculated to be -15.7±1.1 ‰ and -15.2±0.9 ‰ during winter at the near-highway monitoring location and in the tunnel in Shenyang China, respectively (Table 4), indicating a consistent $\delta^{15}N(NH_3)$ off-set between the active and passive sampling collection techniques.

4. Discussion

4.1 Traffic-Plume δ¹⁵N(NH₃) Variability

Here we assess the drivers behind the $\delta^{15}N(NH_3)$ variabilities measured within each sampling campaign, including the seasonal difference measured at the near-highway monitoring site, the temporal variation observed during summer at the near-highway site and the Shenyang tunnel, and the spatial patterns observed from the on-road measurements. We

hypothesize that the observed $\delta^{15}N(NH_3)$ variabilities could be related to (1) f(NH₃) partitioning, (2) NH₃ dry deposition, (3) background NH₃ contributions and/or (4) vehicle fleet composition differences.

Previously, it has been theoretically estimated and shown from field observations and laboratory studies that isotopic N equilibrium and reactions between NH₃ and NH₄⁺ can scramble the 14 N and 15 N distributions between these molecules, leading to the preferential partitioning of ¹⁵N into NH₄⁺ (Kawashima and Ono, 2019; Savard et al., 2017; Urey, 1947; Walters et al., 2018). A significant positive correlation between $f(NH_3)$ and $\delta^{15}N(NH_3)$ during both summer and winter were observed at the near-highway monitoring location, which is consistent with influences from N isotopic equilibrium reactions. However, the $\delta^{15}N(NH_3)$ and $f(NH_3)$ relations were affected by a single influential f(NH₃) value during both summer and winter, and removal of these points resulted in an insignificant relation between δ^{15} N(NH₃) and f(NH₃) (Figures S4 and S5). The temporal tunnel variability is not likely to be driven by f(NH₃) partitioning influences as the estimated f(NH₃) was not found to be significantly different between periods the tunnel was open or closed (p>0.05), indicating a significant change in NH_3/pNH_4^+ partitioning did not occur during these periods. Thus, we do not expect $f(NH_3)$ partitioning and NH_3 reactive sink to have played a significant role in the δ^{15} N(NH₃) variability observed at the various sampling sites. We note that the influence of N isotopic exchange reactions on $\delta^{15}N(NH_3)$ depends on the degree of NH_3/pNH_4^+ partitioning. Typically, f(NH₃) was observed to be >0.934, which would limit the influence of equilibrium exchange reactions to alter the measured $\delta^{15}N(NH_3)$ values. We also note that there is an equilibration time needed before N isotopic equilibrium between NH_3 and pNH_4^+ is achieved, but this rate is currently unknown. Thermodynamic gas-fine aerosol equilibrium has been calculated to have an equilibration time on the order of 10s of minutes to several hours, dependent upon ambient conditions and particle characteristics (Meng and Seinfeld, 1996). Assuming a similar equilibration rate for N isotopic exchange between NH₃ and pNH₄⁺ would indicate that complete N isotopic equilibrium would likely not be achieved near NH₃ emission sources, which is consistent with our observations.

NH₃ dry deposition was not expected to contribute to the observed variability in the well-ventilated sampling conditions at the near-highway monitoring location and the on-road measurements. These measurements were conducted close to the emitted NH₃ (e.g., typically within 5 m at the near-highway monitoring site), which should have minimized NH₃ loss via dry deposition (Asman et al., 1998). However, NH₃ dry deposition may have played an

important role under the closed sampling environment of the tunnel and may explain the observed $\delta^{15}N(NH_s)$ temporal variability with higher values observed when the tunnel was open (4.2±1.7 ‰, n=14) compared to samples collected during periods that the tunnel was closed to traffic (0.1±1.3 ‰, n=7) (Table 3). Previously, lower NH₃ emission ratios were reported from traffic plumes in tunnels relative to on-road highway measurements, which was concluded to result from contributions of NH₃ dry deposition on the tunnel surfaces (Sun et al., 2017). If NH₃ dry deposition is influenced by N isotopic equilibrium reactions between NH₃ and the surface deposited NH₄⁺, this would have resulted in NH₃ depleted in ¹⁵N as it is removed from the atmosphere resulting in lower $\delta^{15}N(NH_3)$ values (Walters et al., 2018). Indeed, a previous NH₃ absorption-desorption study on minerals has shown the preferential removal of ¹⁵NH₃ from the gaseous phase, with the degree of ¹⁵N depletion of the gaseous NH₃ dependent upon the adsorbed NH₃ amount (Sugahara et al., 2017). Thus, as the traffic plume ages in the absence of fresh emissions, we would expect NH₃ dry deposition influences and the potential for N isotopic exchange reactions between the air and tunnel surface to be most significant, which might explain the lower $\delta^{15}N(NH_x)$ values observed during periods the tunnel was closed. Dry deposition of NH₃ during the day in the tunnel could have also impacted the measured $\delta^{15}N(NH_3)$ values, but the constant emission of NH₃ likely resulted in non-equilibrium conditions, such that N isotopic equilibrium between the ambient NH₃ and surface deposited NH₄⁺ would not have been fully achieved.

Background NH₃ contributions are important to identify as a possible driver of $\delta^{15}N(NH_3)$ variability. At the nearhighway monitoring site, wind sector analysis found no statistical difference in $\delta^{15}N(NH_3)$ when sorted by wind direction for either summer or winter (Figure 2). This indicates that transport from local NH₃ point sources other than vehicle emissions played a minor role in the seasonal $\delta^{15}N(NH_3)$ difference. Additionally, the similar [NH₃]:[CO] seasonal relations at the near-highway monitoring site (Figure 3) indicates that seasonal variations in background NH₃ influences at the near-highway monitoring site were minor. While dilution by background air into the Shenyang tunnel during the periods that the tunnel was closed to traffic should be considered as a driver of the temporal $\delta^{15}N(NH_x)$ variability, both the average [NH_x] and f(NH₃) were not consistent with significant mixing in of background air. When the tunnel was closed [NH_x] averaged 79.4±14.4 ppb_v (Table 3), which is elevated compared to urban background [NH₃] measurements previously reported from a megacity in China (Beijing) during winter of $5.22\pm3.75 \ \mu g \cdot m^{-3}$ (or $6.9\pm4.9 \ ppb_v$) (Ianniello et al., 2010). Additionally, f(NH₃) was elevated during the collection period that the tunnel was closed, averaging 0.937±0.045 (Table 3), consistent with local emissions rather than contributions from background air that tends to have a lower $f(NH_3)$ value such as reported to be typically below 0.6 during November based on data collected from Beijing, China (Zhang et al., 2018). Thus, we do not expect background NH₃ contribution to have played a significant role in the tunnel temporal $\delta^{15}N(NH_x)$ variability. Furthermore, we do not expect background NH₃ contributions to have played a significant role in the spatial $\delta^{15}N(NH_3)$ variability observed from the on-road measurements in the Northeastern USA. While lower $\delta^{15}N(NH_3)$ values in non-urban regions might be consistent with an increased contribution from background agricultural emissions, which tend to have a low $\delta^{15}N(NH_3)$ signature (e.g., -31 to -14 ‰; Hristov et al., 2011), we expect these temperature-dependent emissions to be minimal during the winter when the on-road measurements were conducted.

Vehicle fleet compositions could have a strong influence on the measured $\delta^{15}N(NH_3)$ variabilities if gasoline and diesel-powered engines, which utilize different types of NO_x reduction technologies that lead to NH_3 emission (Suarez-Bertoa and Astorga, 2018), have different $\delta^{15}N(NH_3)$ emission signatures. Categorization of our on-road collection routes as either highway routes or trucking routes resulted in statistically significantly different $\delta^{15}N(NH_3)$ values of 6.9±1.9 ‰ (n=12) and -1.5±1.6 ‰ (n=2), respectively, supporting the idea that the $\delta^{15}N(NH_3)$ spatial variation was influenced by fleet composition. This would also be consistent with previous findings that vehicle fleet composition was the main driver of spatial on-road variability observed for $\delta^{15}N(NO_x)$ (Miller et al., 2017). Vehicle fleet NH₃ emissions driven by reduction technologies may have also influenced the seasonal $\delta^{15}N(NH_3)$ difference observed at the near-highway monitoring location. Under cold ambient conditions of -7 °C, diesel-powered vehicles equipped with selective catalytic reduction (SCR) technology were reported to have minimal emission of NH₃. In comparison, gasoline-powered vehicles equipped with three-way catalytic converter (TWCC) were reported to have increased NH₃ emission relative to warmer conditions at 23 °C (Suarez-Bertoa and Astorga, 2018). Vehicle fleet composition may also explain the significantly lower $\delta^{15}N(NH_3)$ values during the summer night/early morning collection period at the near-highway monitoring site (Table 2). Vehicle fleet composition was not monitored in this study, but a previous study has reported relatively higher truck traffic compared to gasoline vehicles from nearhighway measurements during the night/early morning before morning rush hour (Wang et al., 2018). A lower δ^{15} N(NH₃) signature from diesel emissions compared to gasoline, as supported by our on-road measurements, would explain both the seasonal differences in $\delta^{15}N(NH_3)$ and the temporal $\delta^{15}N(NH_3)$ variability observed primarily during summer. To date, there are neither direct tailpipe measurements of $\delta^{15}N(NH_3)$ from gasoline and diesel-powered

vehicle nor an explanation for the expected $\delta^{15}N(NH_3)$ signatures of vehicle derived emissions. Future work is needed to evaluate direct tailpipe $\delta^{15}N(NH_3)$ signatures from gasoline and diesel-powered vehicles to test our hypothesis. We note that while there was a statistically significant seasonal difference in the measured $\delta^{15}N(NH_3)$ at the near-highway monitoring site, the absolute difference of ~1.7‰ was small.

4.2 Comparison Between Active and Passive NH₃ Collection

A comparison between active and passive sampling was conducted to evaluate the performance of the varying NH₃ collection techniques. Overall, remarkably similar [NH₃] were determined using the active (i.e., denuder-filter pack) and passive (i.e., ALPHA) sampling techniques (Table 4). The finding of similar [NH₃] between passive (ALPHA) and active sampling of NH₃ is consistent with previous comparisons (Day et al., 2012; Pan et al., 2020), and provides support that passive collection of NH₃ may be a convenient approach for spatial documentation of near-surface [NH₃].

While the two sampling techniques produced consistent [NH₃], significant differences in $\delta^{15}N(NH_3)$ were observed. The mass-weighted $\delta^{15}N(NH_3)$ using the active sampling technique was 8.0 ± 1.1 ‰ and 3.5 ± 0.8 ‰, while the values using the passive sampling technique was -7.7 ± 0.1 ‰ (n=4) and -11.7 ± 0.3 ‰ (n=3) at the near-highway site (winter) and in the Shenyang tunnel, respectively (Table 4). The measured traffic derived $\delta^{15}N(NH_3)$ values via passive sampler were similar to previous measurements utilizing a similar sampling approach that included measurements in a tunnel in the USA and a tunnel in China with reported values of -3.4 ± 1.2 ‰ (n=2) (Felix et al., 2013) and -14.2 ± 2.6 ‰ (n=8) (Chang et al., 2016), respectively. While our passive $\delta^{15}N(NH_3)$ values were generally consistent with previous reports, there is a large off-set between the passive and active sampling techniques that were calculated to be -15.7 ± 1.1 ‰ and -15.2 ± 0.9 ‰ at the near-highway site and in the Shenyang tunnel, respectively (Table 4). This $\delta^{15}N(NH_3)$ off-set between passive and active NH₃ collection techniques is in agreement with a value of -15.4 ‰ observed from urban background measurements conducted in Beijing, which has been concluded to be due to a diffusive isotope fractionation (Pan et al., 2020). Overall, the large $\delta^{15}N(NH_3)$ off-set observed between passive and active NH₃ collection and potential $\delta^{15}N(NH_3)$ bias in the passive collection of NH₃ has several important implications. The majority of reported $\delta^{15}N(NH_3)$ source signatures have been characterized using passive sampling techniques and might be biased by approximately -15.5 ‰ under the environmental conditions during our sampling periods. These previous measurements could potentially be corrected, but the further characterization of the passive sampler $\delta^{15}N(NH_3)$ off-set is needed.

4.3 Urban Traffic Plume δ¹⁵N(NH₃) Signature

The measured $\delta^{15}N(NH_3)$ traffic plume signatures utilizing the active sampling technique demonstrates an overall range from -3.0 to 10.1 ‰ (Figure 7). Our analysis indicated that $\delta^{15}N(NH_3)$ variability was influenced by fleet composition and NH₃ dry deposition in aged vehicle plumes measured in a tunnel. Thus, for deriving an urban traffic plume $\delta^{15}N(NH_3)$ signature, we have considered measurements conducted under fresh plume conditions and on/near highway measurements, as representative of urban vehicle NH₃ emissions. These observations included the near-highway measurements conducted during both summer and winter, the mobile on-road measurements conducted on highways, and the Shenyang tunnel during operation. While there are $\delta^{15}N(NH_3)$ differences between sampling environments for this subset of observations (Figure 7), the absolute difference in the mean $\delta^{15}N(NH_3)$ was quite small (generally within ~3 ‰) and may reflect actual differences in urban vehicle fleet compositions. Overall, the constrained observations assumed to be representative of urban vehicle emissions reduces the $\delta^{15}N(NH_3)$ variability with a range of 2.1 to 10.1 ‰ (Figure 7). The constrained $\delta^{15}N(NH_3)$ has a combined numerical average of 6.6±2.1 ‰ (n=80) (Figure 7), which was found to not significantly differ from a normal distribution (Kolmogorov-Smirnov test of normality, p = 0.528), and is suggested to be the urban vehicle-derived traffic plume $\delta^{15}N(NH_3)$ source signature.

The recommended $\delta^{15}N(NH_3)$ vehicle-derived traffic signature of 6.6±2.1 ‰ (n=80) has a narrower range and higher value than previously reported vehicle signatures of -17.8 to 0.4 ‰ (Chang et al., 2016; Felix et al., 2013; Smirnoff et al., 2012). The difference between the recommended $\delta^{15}N(NH_3)$ vehicle-derived source signature and previous reports by Chang et al., 2016 and Felix et al., 2013, was found to be caused by a $\delta^{15}N(NH_3)$ bias from passive NH₃ collection that was suggested to be driven by a diffusion isotope effect. The recommended $\delta^{15}N(NH_3)$ vehicle-derived source signature was also found to be statistically different from a previous report that actively sampled NH₃ using a filter pack collection system, which reported an average $\delta^{15}N(NH_3)$ of -2.1±1.9 ‰ (Smirnoff et al., 2012). Differences between our recommended $\delta^{15}N(NH_3)$ value and previous reports by Smirnoff et al., 2012 are difficult to identify and

may be related to differences in vehicle fleet compositions. Additionally, we note that this difference may be related to the potential for a positive sampling artifact associated with filter pack collection using a particulate filter and subsequent acid-coated filter for separate pNH₄⁺ and NH₃ collection, respectively, as volatilization of the collected pNH₄⁺ could have resulted in an NH₃ collection bias (Yu et al., 2006). Indeed, previous laboratory experiments have shown that NH₃ volatilized from NH₄NO₃ particles collected from filters have a δ^{15} N(NH₃) value lower than the δ^{15} N(pNH₄⁺) by 28.6±2.7 ‰ (Walters et al., 2019). Thus, pNH₄⁺ volatilization could have artificially lowered the reported δ^{15} N(NH₃) value and may explain the lower δ^{15} N(NH₃) values reported in Smirnoff et al. 2014 compared to our results.

5. Conclusions

We characterized the $\delta^{15}N(NH_3)$ signatures from a variety of temporal and spatial traffic derived plumes utilizing a laboratory-verified active collection technique demonstrated to reflect accurate $\delta^{15}N(NH_3)$ values. Overall, our measurements indicate a $\delta^{15}N(NH_3)$ range of -3.0 to 10.1 ‰ from vehicle-derived plumes representing a variety of driving conditions and fleet compositions that included stationary measurements conducted in Providence, RI, USA, and Shenyang, Liaoning, China, and mobile on-road measurements performed in the Northeastern USA. These $\delta^{15}N(NH_3)$ values were found to be higher than previous reports of traffic derived measurements that ranged between -17.8 to 0.4 ‰. Our results indicate that the majority of these previously reported lower values were due to a $\delta^{15}N(NH_3)$ collection bias of approximately -15.5 ‰ associated with passive NH₃ collection, highlighting the critical need to utilize accurate $\delta^{15}N(NH_3)$ collection techniques.

Significant spatial and temporal $\delta^{15}N(NH_3)$ variabilities were observed in the seasonal and summer diel measurements conducted at the near-highway monitoring site, in aged traffic plumes in the Shenyang tunnel, and along rural trucking routes in the Northeastern USA. Vehicle fleet composition was suggested to drive significant $\delta^{15}N(NH_3)$ variability, as suspected higher diesel NH₃ emissions during summer relative to winter and mobile measurements conducted on trucking routes were found to result in lower $\delta^{15}(NH_3)$ values, which likely reflects differences in NH₃ production via three-way catalytic converter and selective catalytic reduction technologies. Additionally, physical processing associated with NH₃ dry deposition was suspected of having lowered the observed $\delta^{15}N(NH_3)$ values in the tunnel when vehicle passage was ceased. The reactive NH₃ sink associated with pNH₄⁺ formation was found to play a minor role in the $\delta^{15}N(NH_3)$ variability due to elevated f(NH₃). Accounting for these influences, our results constrain the $\delta^{15}N(NH_3)$ signature from urban traffic derived fresh plume emissions to 6.6±2.1 ‰ ($\bar{x}\pm 1\sigma$; n = 80). In addition to $\delta^{15}N(NH_3)$ characterization, our measurements demonstrate elevated NH₃ emissions from vehicle plumes and a strong relationship between [NH₃]:[CO] (mol:mol) with fitted slopes of 0.025±0.005, 0.027±0.005, and 0.026±0.005 for summer near-highway, winter near-highway, and on-road measurements, respectively, which are in agreement with recent measurements in other regions. Overall, our results highlight the significance of traffic derived NH₃ emissions and demonstrates the potential to use $\delta^{15}N(NH_3)$ to track its contributions to chemistry and N deposition budgets.

The results of this study have important implications for evaluating NH₃ budgets, particularly in urban regions. The measured $\delta^{15}N(NH_3)$ traffic signature (6.6±2.1 ‰, n=80) is unique as it is the only source that has a reported positive $\delta^{15}N(NH_3)$ value. Thus, $\delta^{15}N(NH_3)$ may be a useful tracer to evaluate the contribution of traffic derived emissions in urban regions and to evaluate the connection between urban NH₃ emissions and its role in PM_{2.5} formation. Our demonstrated approach for utilizing a laboratory-verified technique with potential for hourly time resolution is applicable for constraining other important NH₃ emissions sources and to produce a consistent database of $\delta^{15}N(NH_3)$ source signature values. Future work is needed to accurately characterize and improve upon the $\delta^{15}N(NH_3)$ source inventory and evaluate potential fractionation influences associated with NH_x plume aging and deposition.

Data availability. Data presented in this manuscript are available on the Brown University Digital Repository at https://doi.org/10.26300/q3h4-7s93, and the RI-DEM monitoring data are publicly available via the U.S. EPA Air Quality System Data Mart at https://aqs.epa.gov/api.

Author Contribution. WWW, LS, JC, YF, MGH designed varying aspects of the field sampling plan. WWW, LS, JC, NC were involved in carrying out the field measurements. WWW and LS conducted all laboratory analyses of data. WWW prepared the manuscript with contributions from all co-authors.

Competing interests. The authors declare that they have no conflict of interest.

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Tables

Table 1. Summary of method detection limit (MDL), pooled relative standard deviations (RSD), and δ^{15} N reproducibility of NH₃ determined from active sampling using a denuder-filter pack (ChemComb Speciation Cartridge). The MDL is reported in units of ppb_v for NH₃ and $\mu g \cdot m^{-3}$ for pNH₄⁺. The MDL is reported for each sampling environment, including the near-highway monitoring location in Providence, RI, USA during the summer (Summer-NH) and winter (Winter-NH), on-road mobile measurements in the northeastern USA (Mobile), and the Tunnel in Shenyang, Liaoning, China (Tunnel).

	М	$DL (ppb_v \text{ or } \mu)$						
			-	δ^{15} N-				
Species	Summer- NH	Winter-NH	Mobile	Tunnel	RSD(%)	Reproducibility		
Active Sampling (denuder-filter pack)								
NH_3	0.088	0.147	0.415	0.170	9.8	0.8 ‰		
pNH_{4}^{+}	0.090	0.234	0.093	0.118	8.5	N/A ^a		

^aSeparate measurement of $\delta^{15}N(pNH_4^+)$ was not conducted due to sample mass limitations.

Table 2. Summary of the near-highway (Providence, RI, USA) environmental conditions including temperature (Temp), relative humidity (RH), wind direction and NH_x data including [NH₃], [pNH₄⁺], f(NH₃), and δ^{15} N(NH₃) sorted by NH_x collection period for both summer and winter. Data are reported as $\bar{x}(\pm 1\sigma)$ for each collection period during summer and winter, respectively. The number of collections made during each collection period (n) is indicated.

Prevailing								
Collection	Temp		Wind	[NH ₃]	$[pNH_4^+]$		$\delta^{15}N(NH_3)$	
Period (n)	(°C)	RH (%)	Direction	(ppb _v)	$(\mu g \cdot m^{-3})$	f(NH ₃)	(‰)	
Summer (August 9 to August 18)								
0:30-6:30 (7)	20.1(1.0)	80.5(11.1)	WSW	9.8(3.7)	0.350(0.269)	0.956(0.032)	4.2(1.0)	
6:30-12:30 (8)	24.0(1.9)	63.6(10.2)	S	13.4(3.7)	0.301(0.221)	0.973(0.016)	7.3(1.5)	
12:30-18:30 (8)	27.4(2.0)	45.1(14.1)	SSE	16.0(3.3)	0.252(0.135)	0.980(0.012)	7.1(1.5)	
18:30-0:30 (9)	23.1(1.3)	65.8(13.5)	SSW	15.9(1.8)	0.310(0.183)	0.976(0.015)	6.9(0.7)	
Overall (32)	23.7(3.0)	63.3(17.4)	SSW	14.0(4.0)	0.302(0.208)	0.972(0.022)	6.4(1.7)	
Winter (January 21 to February 1)								
0:00-6:00 (5)	-3.7(2.8)	59.1(10.8)	WNW	6.3(1.7)	0.388(0.173)	0.925(0.017)	8.5(0.3)	
6:00-12:00 (5)	-0.8(4.3)	53.6(13.8)	WNW	13.4(1.5)	0.601(0.289)	0.947(0.024)	8.8(1.0)	
12:00-18:00 (5)	2.7(4.3)	43.7(12.5)	WNW	16.0(2.7)	0.447(0.191)	0.963(0.015)	7.8(1.5)	
18:00-0:00 (7)	0.8(4.3)	61.2(16.0)	NW	12.3(5.2)	0.640(0.739)	0.942(0.036)	7.7(1.7)	
Overall (22)	-0.2(4.6)	55.0(15.2)	WNW	12.0(4.8)	0.530(0.468)	0.944(0.029)	8.1(1.4)	

Table 3. Summary of the Shenyang, China tunnel data including temperature (Temp), relative humidity (RH), $[NH_x]$, $f(NH_3)$, $[NH_3]$, and $\delta^{15}N(NH_x)$. Data are reported as $\bar{x}(\pm 1\sigma)$ for each collection period and the overall monitoring period during summer and winter, respectively. The number of collections made during each collection period (n) is indicated.

Collection Period	Temp		[NH _x]		$\delta^{15}N(NH_x)^b$
(n)	(°C)	RH (%)	(ppb _v)	f(NH ₃) ^a	(‰)
6:00 - 14:00 (7)	19.2(1.1)	35.2(4.8)	136.8(18.8)	0.959(0.027)	3.6(1.0)
14:00 - 22:00 (7)	20.5(1.9)	36.2(6.4)	181.2(23.0)	0.973(0.028)	4.8(2.0)
22:00 - 6:00 (7)	18.3(0.9)	34.7(8.1)	79.4(14.4)	0.937(0.045)	0.1(1.3)
Overall (21)	19.3(1.6)	35.4(6.7)	132.5(45.8)	0.956(0.038)	2.9(2.5)

^af(NH₃) was calculated from $[pNH_4^+]$ estimated using ion-mass balance based on the $[NH_x]$, $[pNO_3^-]$, and $[pSO_4^{2-}]$ measurements (see Eq. 6)

^bDue to the elevated f(NH₃), the measured $\delta^{15}N(NH_x) \sim \delta^{15}N(NH_3)$.

Table 4. Summary of $[NH_3]$ and $\delta^{15}N(NH_3)$ from the passive and active collection of NH_3 at the winter near-highway and Shenyang stationary monitoring locations.

Location	[NH ₃] (ppb _v)		δ ¹⁵ N(NH _{3/x}) (‰)			
	Passive	Active	Passive	Active	Shift ^a	
Winter Near-Highway	11.6±1.4	12.0±1.2	-7.7±0.1	8.0±1.1	-15.7±1.1	
Shenyang Tunnel	124±3.6	127.1±12.5	-11.7±0.3	3.5 ± 0.8^{b}	-15.2±0.9	

^a Calculated as the $\delta^{15}N$ difference between passive and active NH₃ collection. The uncertainty represents the propagated error between these two measurements.

^b The Shenyang Tunnel active measurements represent $\delta^{15}N(NH_x)$; however, due to elevated $f(NH_3)$ that averaged, $\delta^{15}N(NH_x) \sim \delta^{15}N(NH_3)$.

Figures



Figure 1: Near-highway (Providence, RI, USA) data summary of (a) [NH₃], (b) [pNH₄⁺], (c) f(NH₃) (=[NH₃](mol)/[NH_x](mol)), and (d) $\delta^{15}N(NH_3)$. The NH_x data was generated from an active collection technique using a denuder-filter pack with a collection time of 6 h, and the error bars for concentrations and $\delta^{15}N(NH_3)$ measurements shown as black vertical lines represent the RSD (%) and propagated error, respectively. The break in the x-axis separates the summer and winter measurements.



Figure 2: Wind sector analysis of samples collected at the near-highway monitoring site (Providence, RI, USA) for [NH₃] (circles) in (a) summer and (b) winter. The data is size-coded for $f(NH_3)$ and color-coded for $\delta^{15}N(NH_3)$ (‰). The monitoring location is downwind of I-95 except for wind directions 15 to 150° (grey-shaded region).



Figure 3: Linear relations between [NH₃] and [CO] from the near-highway (Providence, RI, USA) and mobile on-road (northeastern USA) measurements. The [NH₃] data were based on acid-coated denuder collection and the [CO] represents the average of the on-line determined concentrations over the collection period. The linear regressions (solid lines) and Pearson's correlation coefficients (r) are provided for each respective measurement location.



Figure 4: Tunnel (Shenyang, Liaoning, China) data summary of (a) $[NH_x]$, (b) concentrations of $[pNO_3^-]$ (blue square) and $[pSO_4^{2-}]$ (red triangle), (c) $f(NH_3)$ calculated using ion-mass balance (open circle) and modelled using ISORROPIA (purple diamond), and (d) $\delta^{15}N(NH_x)$. The data was generated from using a denuder-filter pack with a collection time of approximately 8 h. Error bars for concentrations and $\delta^{15}N(NH_x)$ measurements shown as black vertical lines represent the RSD (%) and propagated error, respectively. ISORROPIA was not run for five collection periods, due to the absence of relative humidity and temperature data.



Figure 5: Mobile on-road (Northeastern USA) measurements including (a) spatial mapping of measurement path sorted by date and data summary of (b) [NH₃], (c) [pNH₄⁺], (d) f(NH₃) (=[NH₃](mol)/[NH_x](mol)), (e) δ^{15} N(NH₃) for highway (black circle) and trucking routes (red square). The NH_x data was generated using a denuder-filter pack with a collection time of approximately 1 h, and the error bars for concentrations and δ^{15} N(NH₃) measurements shown as black vertical lines represent the RSD (%) and propagated error, respectively. The break in the x-axis separates breaks in the mobile measurements. Image (a) was created using ArcGIS Copyright © 1995-2019 Esri.



Figure 6: Spatial maps of (a) mean [NH₃] (ppb_v) and (b) $\delta^{15}N(NH_3)$ (‰) from on-road collections in the northeastern US. Each color represents one concentration or isotope measurement for NH₃ collected over a highway segment at an approximate 1 h resolution using an acid-coated denuder. Note that there are fewer reported $\delta^{15}N(NH_3)$ values than [NH₃] because some samples had an elevated blank (i.e., f_{Blank} > 30 %) and were not measured for $\delta^{15}N(NH_3)$. Images were created using ArcGIS Copyright ©1995-2019 Esri.



Figure 7: Box and whisker plot summarizing the distribution (lower extreme, lower quartile, median (blue circle), upper quartile, upper extreme, and outliers (black diamond)) of $\delta^{15}N(NH_3)$ measurements from near-highway, on-road, and tunnel sampling. The "Urban Traffic Fresh Plume" category represents the combination of $\delta^{15}N(NH_3)$ measurements from the near-highway, on-road (highway), and Shenyang tunnel (fresh) sampling. No statistical summary is provided for "On-Road (Trucking Routes)", due to the limited number of samples in this category.