

1 **The promotion effect of nitrous acid on aerosol formation in**  
2 **wintertime Beijing: possible contribution of traffic-related**  
3 **emission**

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5 Yongchun Liu<sup>1\*</sup>, Yusheng Zhang<sup>1</sup>, Chaofan Lian<sup>2,6</sup>, Chao Yan<sup>3</sup>, Zeming Feng<sup>1</sup>, Feixue  
6 Zheng<sup>1</sup>, Xiaolong Fan<sup>1</sup>, Yan Chen<sup>2,6</sup>, Weigang Wang<sup>2,6\*</sup>, Biwu Chu<sup>3,4</sup>, Yonghong Wang<sup>3</sup>,  
7 Jin Cai<sup>3</sup>, Wei Du<sup>3</sup>, Kaspar R. Daellenbach<sup>3</sup>, Juha Kangasluoma<sup>1,3</sup>, Federico Bianchi<sup>1,3</sup>,  
8 Joni Kujansuu<sup>1,3</sup>, Tuukka Petäjä<sup>3</sup>, Xuefei Wang<sup>6</sup>, Bo Hu<sup>5</sup>, Yuesi Wang<sup>5</sup>, Maofa Ge<sup>2</sup>,  
9 Hong He<sup>4</sup> and Markku Kulmala<sup>1,3\*</sup>

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11 1. Aerosol and Haze Laboratory, Advanced Innovation Center for Soft Matter Science and  
12 Engineering, Beijing University of Chemical Technology, Beijing, 100029, China

13 2. State Key Laboratory for Structural Chemistry of Unstable and Stable Species, Beijing  
14 National Laboratory for Molecular Sciences, Institute of Chemistry, Chinese Academy of  
15 Sciences, Beijing 100190, China

16 3. Institute for Atmospheric and Earth System Research/Physics, Faculty of Science, University  
17 of Helsinki, P.O. Box 64, FI-00014, Finland

18 4. State Key Joint Laboratory of Environment Simulation and Pollution Control, Research  
19 Center for Eco-Environmental Sciences, Chinese Academy of Sciences, Beijing, 100085, China

20 5. State Key Laboratory of Atmospheric Boundary Layer Physics and Atmospheric Chemistry,  
21 Institute of Atmospheric Physics, Chinese Academy of Sciences, Beijing, 100029, China

22 6. University of Chinese Academy of Sciences, Beijing 100049, PR China

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25 **Supplement information**

26 **Non-refractory PM<sub>2.5</sub> (NR-PM<sub>2.5</sub>) measurement.** Concentration of NR-PM<sub>2.5</sub> was  
27 measured with a ToF-ACSM (Aerodyne Co. Ltd., USA). The operation protocol and  
28 the configuration of ToF-ACSM has been described well in previous work <sup>1</sup>. Namely,  
29 PM<sub>2.5</sub> particles from the inlet were focused by a PM<sub>2.5</sub> aerodynamic lens <sup>2</sup>, and then  
30 vaporized by a standard vaporizer heated at 600 °C followed by electronic ionization  
31 (EI, 70 eV). The non-refractory components including chloride, nitrate, sulfate,  
32 ammonia and organics were measured using a time-of-flight mass spectrometer with  
33 unit mass resolution (UMR). The concentrations of the above species were calculated  
34 based on the measured fragments signals, the signal ions (SI), the fragment table, the  
35 measured ionization efficiency (IE) of nitrate and the corresponding relative ionization  
36 efficiency (RIE) for sulfate, chloride, ammonia and organics. IE calibration of nitrate  
37 was performed using 300 nm dry NH<sub>4</sub>NO<sub>3</sub> every month during this observation study.

38 **VOCs measurement.** VOCs were measured using a Single Photo Ionization Time-of-  
39 flight Mass spectrometer (SPI-ToF-MS 3000R, Hexin Mass Spectrometry). 0.8 L min<sup>-1</sup>  
40 of filtered air was sucked from the whole sampling tube and heated to 80 °C in the  
41 inlet. VOCs were selectively enriched continuously through a polydimethylsiloxane  
42 (PDMS) membrane, and then ionized by VUV light (10.5 eV) with a deuterium lamp.  
43 The concentration of VOCs was determined with the time-of-flight mass spectrometer  
44 (ToF-MS) based upon external standard curves of PAMS and TO-15 standard gases  
45 (Linde Electronics & Specialty Gases, USA). VOCs with m/z from 40 to 300 were  
46 recorded with 3 min of time resolution, while hourly averaged concentration were

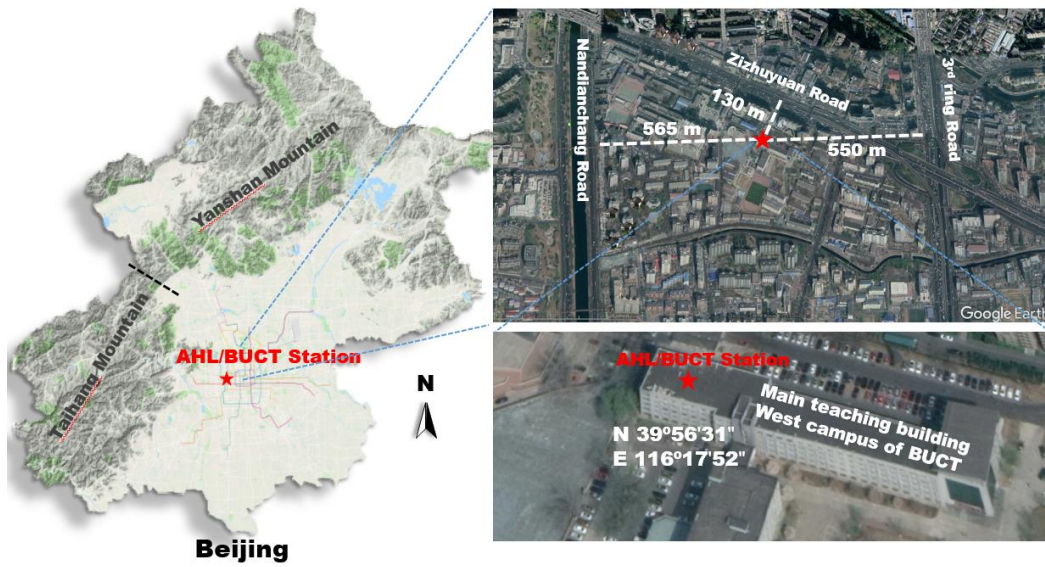
47 reported in this work. Calibration was performed every week.

48 **HONO measurement.** HONO in ambient air directly sampled from the window of the  
49 laboratory was absorbed by a solution containing  $0.06 \text{ mol L}^{-1}$  sulfnilamide in  $1 \text{ mol L}^{-1}$   
50  $\text{HCl}$ , and then transformed into an azo dye by *N*-(1-naphthyl) ethylene-  
51 diaminedihydrochloride ( $0.8 \text{ mmol L}^{-1}$ ). The azo dye was pumped into Teflon absorption  
52 cells (Liquid Core Waveguide, LCW) and detected by a mini-spectrometer with a diode  
53 array detector (Ocean Optics, SD2000). The HONO concentrations was obtained by  
54 subtracting the calibrated signal of the second coil from the first coil using external  
55 nitrile standard solutions. Zero point calibration was performed every day using  
56 scrubbed zero air<sup>3</sup>.

57 **Photolysis rate constants of HONO and O<sub>3</sub>.** Photolysis rate constants of  $\text{NO}_2$  ( $J_{\text{NO}_2}$ ),  
58  $\text{HONO}$  ( $J_{\text{HONO}}$ ) and  $\text{O}_3$  ( $J_{\text{O}_3}$ ) for clear sky conditions were calculated according to the  
59 solar zenith angle and the location using a box model (FACSIMILE 4).  $\text{NO}_2$  photolysis  
60 sensor ( $J_{\text{NO}_2}$ , Metcon) was unavailable during our observation study. However, it was  
61 available in from Aug 17 to Sep 16, 2018. Calibration function between the measured  
62 UVB light intensity and  $J_{\text{NO}_2}$  from Aug 17 to Sep 16, 2008 was established to correct  
63 the influence the climatological  $\text{O}_3$  column, aerosol optical depth and cloud cover on  
64 surface UV light intensity. As shown in Figure S8, the model well predicted the  $J_{\text{NO}_2}$ .  
65 Then the  $J_{\text{NO}_2}$  during this campaign study was predicted using the model. We further  
66 confirmed the calculated  $J_{\text{NO}_2}$  by comprising the OH concentration estimated by the  
67  $J_{\text{O}1\text{D}}$  according to the equation ( $c_{\text{OH}}=J_{\text{O}1\text{D}}\times 2\times 10^{11} \text{ molecules cm}^{-3}$ )<sup>4</sup> and the measured  
68 OH concentration at Huairou, which is 60 km northeast from BUCT, form Jan 11 to

69 Mar 10, 2016. As shown in Figure S8C, the estimated diurnal curve of OH is  
70 comparable with that measured at Huairou.  
71

72 **Supplementary figures**



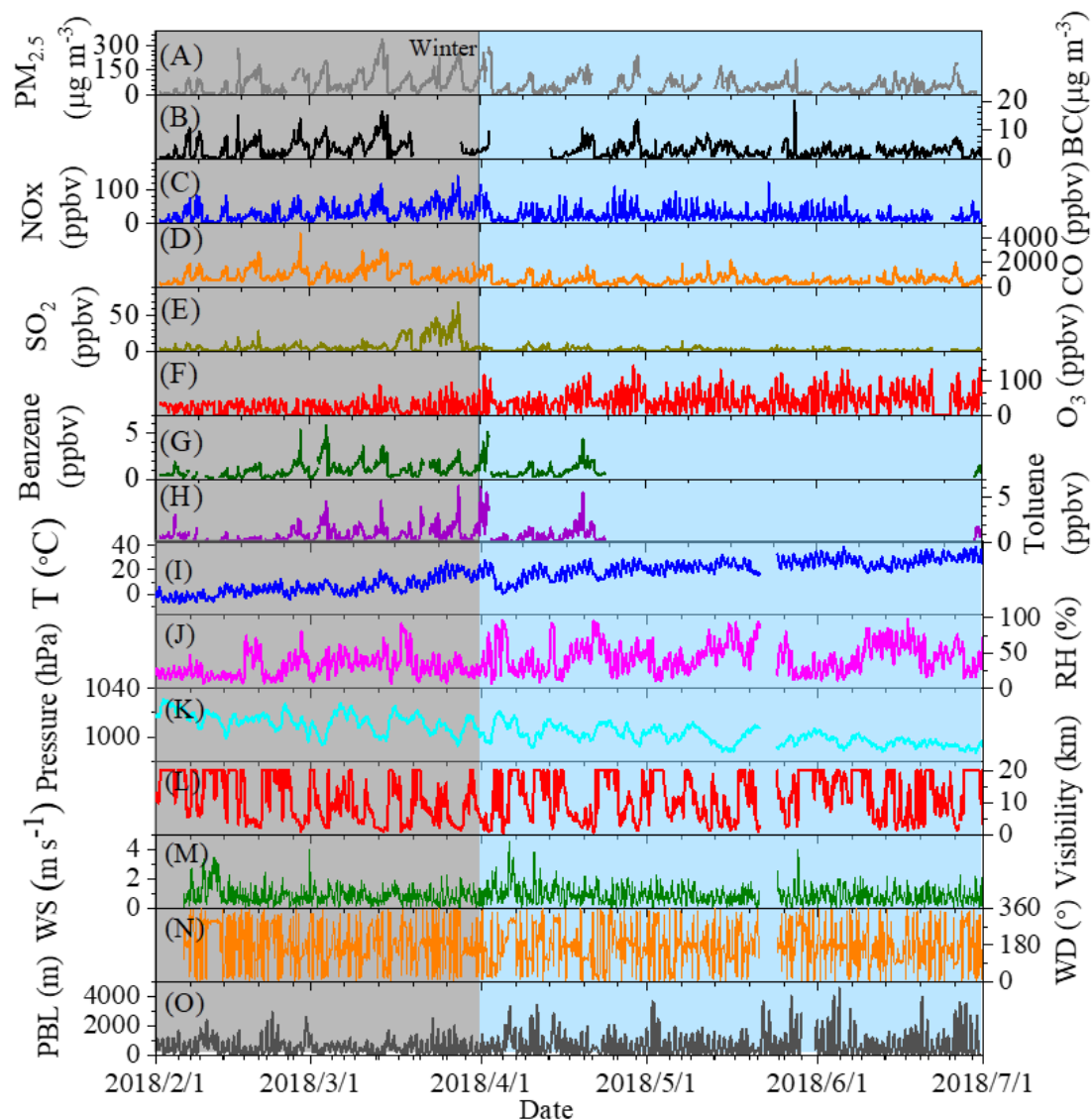
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74 Figure S1. Location of AHL/BUCT observation station. The map was

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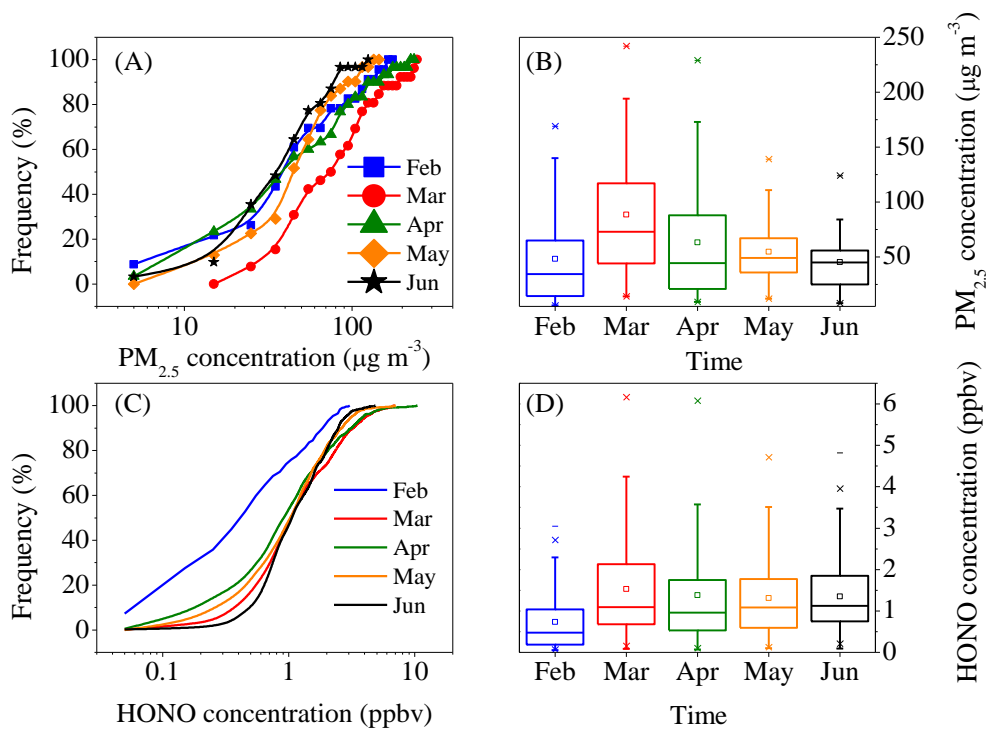
made from Wemap and © Google Earth.

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78 Figure S2. Hourly averaged (A)-(H) concentration of pollutants and (I)-(O)  
 79 meteorological parameters from Feb 1 to Jun 30, 2018. PM<sub>2.5</sub> concentration data were  
 80 obtained from surrounding National Environmental Monitoring Centre of China. PBL  
 81 data were obtained from the NOAA Hysplit model.



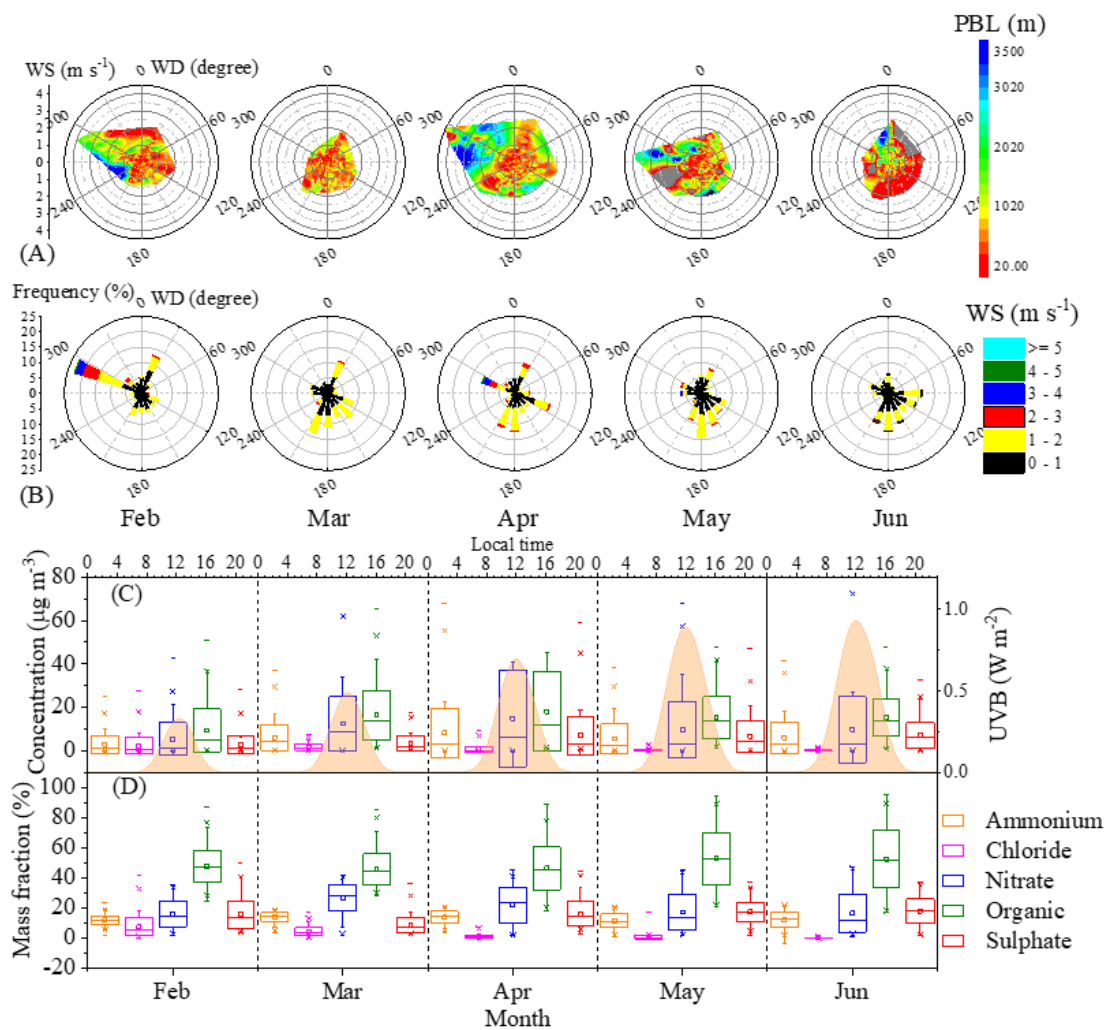
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84 Figure S3. The monthly cumulative frequency of PM<sub>2.5</sub> and HONO and the monthly

85 mean concentration of PM<sub>2.5</sub> and HONO.

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88 Figure S4. (A)-(B) monthly Windrose-PBL plots, and monthly averaged (C) UVB

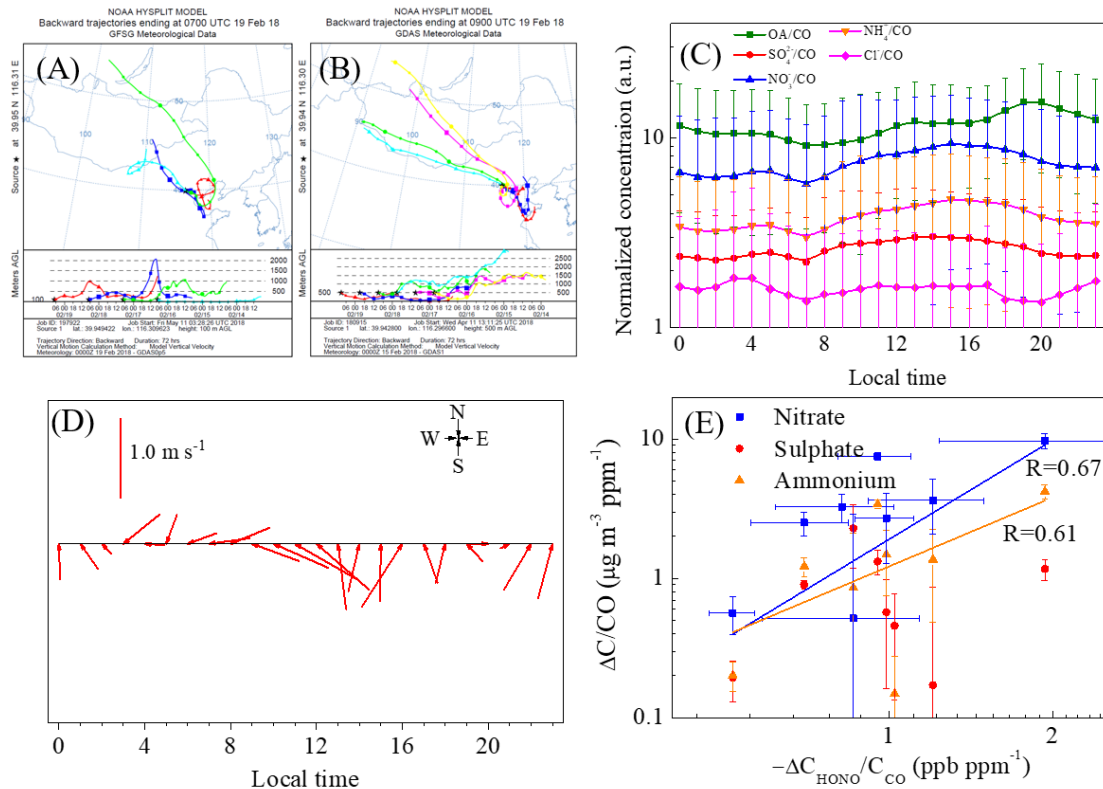
89 intensity, mass concentration and (D) fraction of individual component in NR-PM<sub>2.5</sub>

90 composition from Feb to Jun, 2018.

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94 Figure S5. Transport of air mass during Chinese New Year based on back trajectory

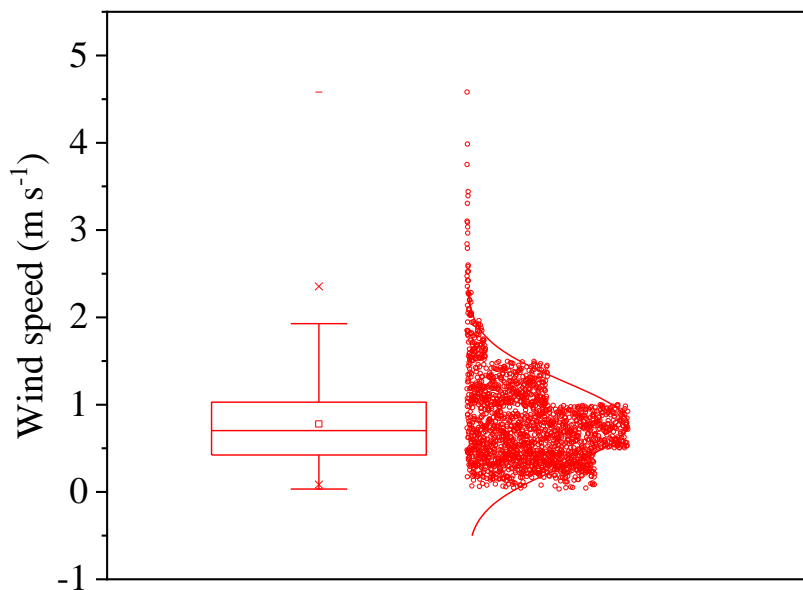
95 analysis (A) at 100 and (B) 500 m height; (C) Diurnal variation of NR-PM<sub>2.5</sub> normalized

96 to CO concentration from Feb 1 to March 31; (D) Hourly averaged wind speed variation

97 in the 12<sup>th</sup> episode; (E) Correlation of the concentration increment of individual

98 component and consumed HONO normalized to CO in the daytime.

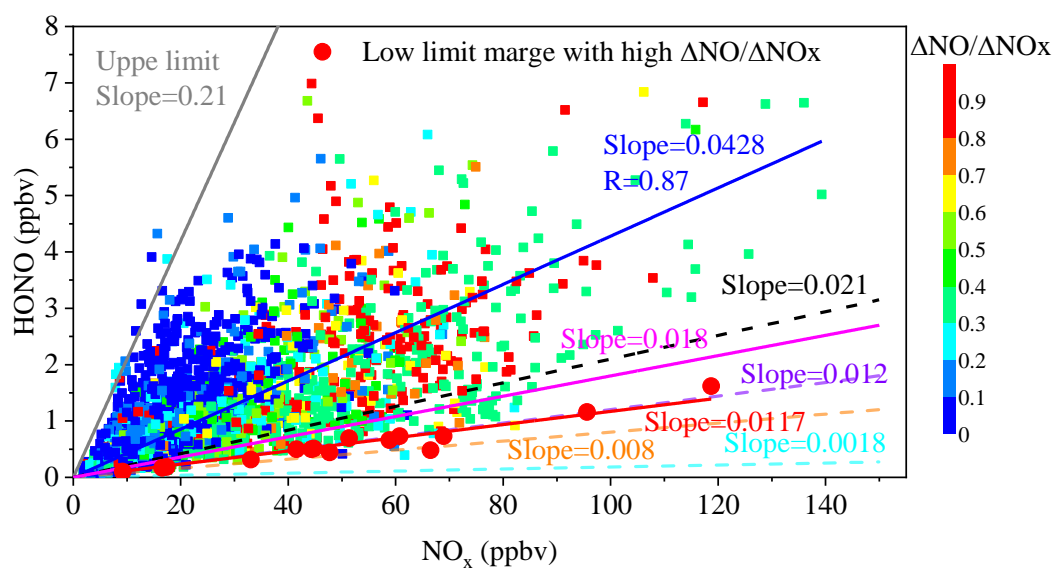
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101 Figure S6. Distribution of wind speed when the PM<sub>2.5</sub> concentration was larger than 50

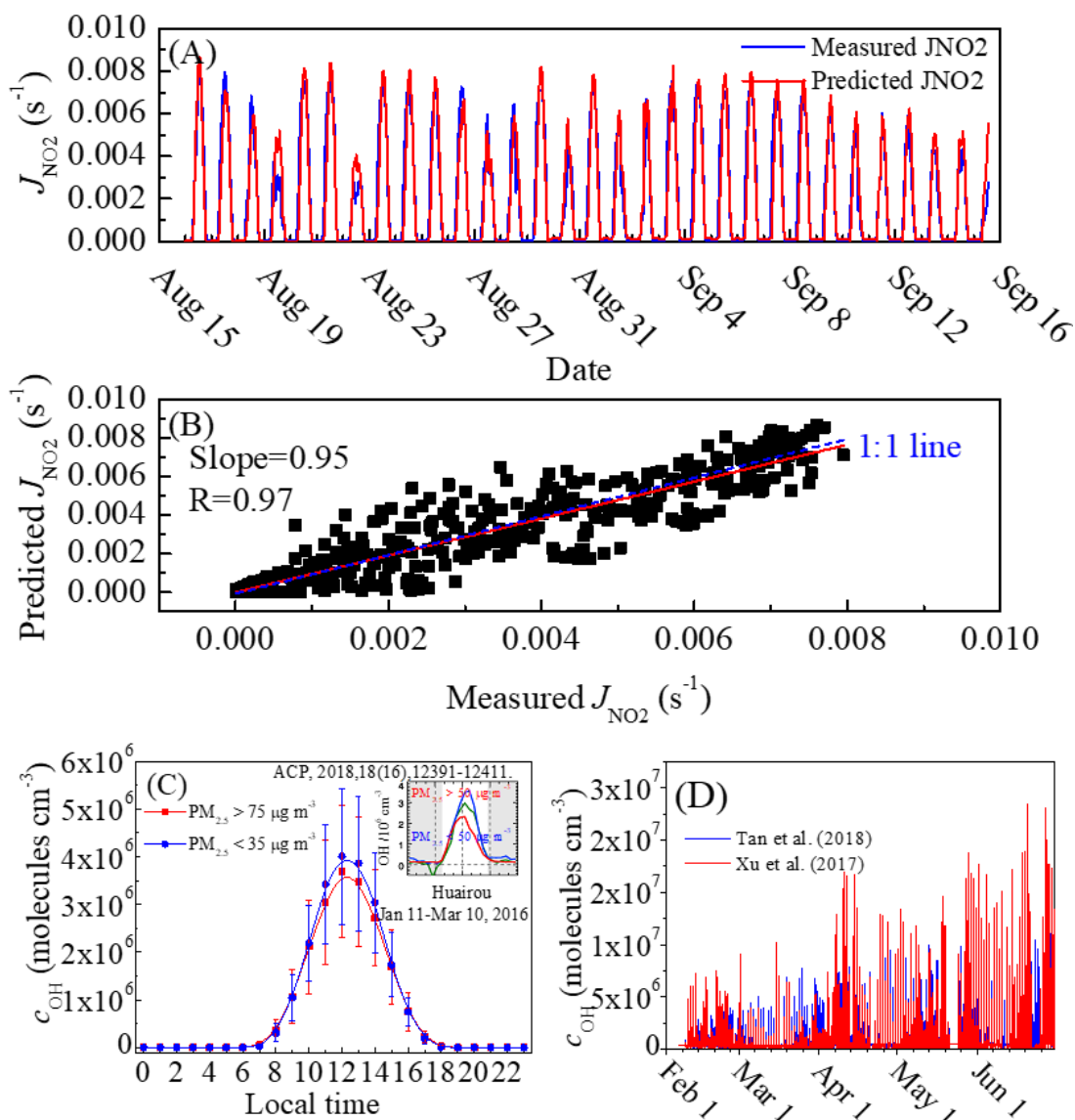
102 μg m<sup>-3</sup> and the RH was less than 90 %.



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104 Figure S7. Correlation of measured HONO concentration with NO<sub>x</sub> concentration.

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107 Figure S8. (A) Measured and predicted  $J_{\text{NO}_2}$  and (B) the correlation between measured

108 and predicted  $J_{\text{NO}_2}$  from Aug. 15 to Sep. 16; (C) calculated diurnal curve of OH

109 concentration based on  $J_{\text{OH}}$  compared with that measured at Huairou (60 km northeast

110 from BUCT) from Jan 11 to Mar 10, 2016; (D) Comparison of OH concentrations

111 estimated using different methods.

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## Supplementary tables

Table S1. ANOVA statistics analysis for the monthly mean fraction of the individual component in NR-PM<sub>2.5</sub> and HONO concentration.

Component	Fraction of NR-PM <sub>2.5</sub> (%) or Concentration of gaseous pollutants (ppbv)				
	Feb	Mar	Apr	May	Jun
Ammonium	Feb (12.2±2.9)				
	Mar (14.2±2.8)	Significant			
	Apr (14.0±4.0)	Significant	Not significant		
	May (11.6±4.6)	Not significant	Significant	Significant	
	Jun (12.2±5.2)	Not significant	Significant	Significant	Not significant
Chloride	Feb (7.7±6.1)				
	Mar (4.4±2.6)	Significant			
	Apr (1.1±1.2)	Significant	Significant		
	May (0.7±1.1)	Significant	Significant	Not significant	
	Jun (0.3±0.2)	Significant	Significant	Significant	Not significant
Nitrate	Feb (16.2±8.5)				
	Mar (26.7±8.8)	Significant			
	Apr (22.0±11.7)	Significant	Significant		
	May (17.3±11.8)	Not significant	Significant	Significant	
	Jun (16.7±12.8)	Not significant	Significant	Significant	Not significant
Organic	Feb (47.9±10.7)				
	Mar (45.9±10.2)	Not significant			
	Apr (46.5±14.2)	Not significant	Significant		
	May (52.9±17.0)	Not significant	Significant	Significant	
	Jun (52.6±18.7)	Significant	Significant	Significant	Not significant

Sulfate	Feb (16.0±9.1)					
	Mar (8.8±5.4)	Significant				
	Apr (16.4±8.2)	Not significant	Significant			
	May (17.5±6.6)	Significant	Significant	Not significant		
	Jun (18.2±8.0)	Significant	Significant	Significant	Not significant	
BC	Feb (3.0±2.8)					
	Mar (4.6±3.1)	Significant				
	Apr (3.2±2.6)	Not significant	Significant			
	May (2.8±2.1)	Not significant	Significant	Not significant		
	Jun (2.6±1.5)	Significant	Significant	Significant	Not significant	
HONO	Feb (0.73±0.70)					
	Mar (1.53±1.25)	Significant				
	Apr (1.38±1.35)	Significant	Not significant			
	May (1.31±1.00)	Significant	Significant	Not significant		
	Jun (1.35±0.80)	Significant	Significant	Not significant	Not significant	
NO <sub>x</sub>	Feb (20.4±17.3)					
	Mar (40.5±24.0)	Significant				
	Apr (22.8±18.6)	Not significant	Significant			
	May (25.0±15.9)	Significant	Significant	Not significant		
	Jun (19.0±12.1)	Not significant	Significant	Significant	Significant	Significant
SO <sub>2</sub>	Feb (3.8±3.3)					
	Mar (12.1±13.0)	Significant				
	Apr (2.8±2.4)	Significant	Significant			
	May (1.8±1.7)	Significant	Significant	Not significant		
	Jun (1.3±1.2)	Significant	Significant	Significant	Not significant	
CO	Feb (959.6±554.6)					

	Mar (1075.0±571.8)	Significant			
	Apr (546.6±378.1)	Significant	Significant		
	May (554.1±336.9)	Significant	Significant	Not significant	
	Jun (583.4±286.2)	Significant	Significant	Not significant	Not significant
O <sub>3</sub>	Feb (22.6±14.6)				
	Mar (23.8±19.2)	Not significant			
	Apr (43.5±29.0)	Significant	Significant		
	May (42.5±28.3)	Significant	Significant	Not significant	
	Jun (57.2±30.7)	Significant	Significant	Significant	Significant

Note: “Significant” or “Not significant” denotes that the difference of the monthly mean fractions or concentrations is significant or not significant at the 0.05 level.

Tab. S2. Mean concentrations of HONO and PM<sub>2.5</sub> in selected episodes

Episode No.	Duration	HONO (ppb)	Average PM <sub>2.5</sub> concentration	NR-PM <sub>2.5</sub> Concentration (%)									
				Chloride		Nitrate		Organic		Sulphate		Ammonium	
				(%)	( $\mu\text{g m}^{-3}$ )	(%)	( $\mu\text{g m}^{-3}$ )	(%)	( $\mu\text{g m}^{-3}$ )	(%)	( $\mu\text{g m}^{-3}$ )	(%)	( $\mu\text{g m}^{-3}$ )
1	Feb 2-5	0.38±0.28	9.3±4.5	4.0±2.3	0.26±0.39	12.3±5.6	0.80±1.17	51.1±10.0	2.68±3.00	20.6±9.2	0.69±0.24	12.0±3.2	0.54±0.49
2	Feb 8-9	0.90±0.72	44.5±3.5	6.3±2.9	1.59±1.46	15.8±7.9	4.20±3.87	49.9±4.8	9.63±7.64	17.3±8.8	2.31±1.42	10.8±1.0	2.14±1.69
3	Feb 10-12	0.31±0.40	9.0±0.8	5.2±3.5	0.18±0.22	6.8±3.9	0.30±0.44	48.6±10.6	1.75±1.72	28.1±11.5	0.74±0.38	11.2±2.5	0.35±0.23
4	Feb 16-19	1.38±0.86	101.5±26.8	15.5±4.2	9.04±4.94	25.0±4.1	13.15±7.73	32.2±3.8	18.21±8.25	14.4±3.7	7.82±4.39	12.9±1.5	6.85±3.78
5	Feb 21-24	0.64±0.58	24.3±7.0	5.5±4.1	0.60±0.51	14.9±6.3	1.80±1.38	56.3±10.0	5.83±2.94	11.8±5.0	1.17±0.67	11.6±2.8	1.24±0.77
6	Feb 25-28	0.87±0.64	108.8±42.9	5.2±1.4	2.94±1.97	27.1±3.9	15.3±8.77	42.5±6.8	22.83±9.68	10.4±3.8	6.44±5.78	14.7±1.8	8.34±5.30
7	Mar 2-3	1.41±0.84	120.0±47.0	8.3±2.2	4.23±1.72	26.5±4.8	15.29±9.44	44.4±6.2	23.40±10.49	7.2±1.9	4.36±3.37	13.5±1.9	7.74±4.76
8	Mar 8-10	1.36±0.89	88.7±34.2	4.8±1.8	1.87±1.09	28.3±5.2	11.00±6.20	43.0±7.0	15.65±7.15	9.0±2.8	3.10±1.42	14.9±2.0	5.58±2.92
9	Mar 11-14	2.27±1.68	170.3±75.4	3.5±0.9	2.48±1.32	34.8±4.3	28.32±19.09	36.8±5.0	27.90±15.78	8.1±1.8	6.60±4.72	16.8±1.5	13.57±8.99
10	Mar 16-19	1.88±1.38	66.0±25.7	3.8±1.7	1.99±1.18	30.2±6.3	17.40±12.45	35.9±2.8	20.87±10.52	13.5±5.1	7.00±4.92	16.5±1.0	9.17±5.86
11	Mar 21-23	1.41±0.72	83.7±22.1	5.3±2.8	2.54±2.30	31.5±3.8	12.23±5.22	45.1±6.7	18.02±5.46	4.4±1.0	1.67±0.92	13.7±1.6	5.38±2.08
12	Mar 25-27	2.22±1.34	129.5±51.9	2.0±0.7	0.94±0.64	35.3±3.6	16.32±9.90	41.5±5.4	20.46±10.18	5.7±1.2	2.56±1.68	15.6±1.6	7.11±4.37

1 Table S3. The summary of the HONO/NO<sub>x</sub> ratio from vehicles in this study and the  
 2 reported emission ratio of HONO/NO<sub>x</sub> from vehicles in China.

No.	Time	$\Delta\text{NO}/\Delta\text{NO}_x$	$R_{\Delta\text{NO}/\Delta\text{NO}_x}$	$\Delta\text{HONO}/\Delta\text{NO}_x$	$R_{\Delta\text{HONO}/\Delta\text{NO}_x}$
1	2018/2/6 5:00-8:00	1.00	0.99	1.3%	0.92
2	2018/2/8 5:00-8:00	0.94	0.99	1.8%	0.96
3	2018/3/3 5:00-8:00	0.98	0.99	2.4%	0.96
4	2018/3/13 5:00-8:00	1.00	0.99	1.4%	0.86
5	2018/4/15 5:00-7:00	0.82	0.97	2.3%	0.99
Mean		0.95±0.08	-	1.8±0.5%	-
Time	Place	Methods	$\Delta\text{HONO}/\Delta\text{NO}_x$		Reference
			Range	Mean	
2015/9/1- 2016/8/31	Ji'nan, Shandong	Empirical analysis of field data	0.19%- 0.87%	0.53±0.20%	5
2011/8/3- 2012/5/31	Hongkong	Empirical analysis of field data	0.5%- 1.6%	1.2±0.4%	6
2014	Beijing	Tunnel experiment	-	2.1%	7
2017	Beijing	Chassis dynamometer	0.03%- 0.42%	0.18%	8
2018/2/1- 2018/6/30	Beijing	Empirical analysis of field data	1.3-2.4%	1.8±0.5%	This study
2018/2/1- 2018/6/30	Beijing	Low limit correlation of field data	-	1.17±0.05%	This study

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#### 4 **References:**

- 5 1. Fröhlich, R.; Cubison, M. J.; Slowik, J. G.; Bukowiecki, N.; Prévôt, A. S. H.; Baltensperger, U.;  
 6 Schneider, J.; Kimmel, J. R.; Gonin, M.; Rohner, U.; Worsnop, D. R.; Jayne, J. T., The ToF-ACSM: a  
 7 portable aerosol chemical speciation monitor with TOFMS detection. *Atmos. Meas. Tech.* **2013**, *6*, (11),  
 8 3225-3241.
- 9 2. R. Williams, L.; A. Gonzalez, L.; Peck, J.; Trimborn, D.; McInnis, J.; R. Farrar, M.; Moore, K.; T.  
 10 Jayne, J.; A. Robinson, W.; Lewis, D.; B. Onasch, T.; R. Canagaratna, M.; Trimborn, A.; Timko, M.;  
 11 Magoon, G.; Deng, R.; Tang, D.; de la Rosa Blanco, E.; Prevot, A.; R. Worsnop, D., Characterization of  
 12 an aerodynamic lens for transmitting particles greater than 1 micrometer in diameter into the Aerodyne  
 13 aerosol mass spectrometer. *Atmos. Meas. Tech.* **2013**, *6*, 3271-3280.
- 14 3. Tong, S.; Hou, S.; Zhang, Y.; Chu, B.; Liu, Y.; He, H.; Zhao, P.; Ge, M., Exploring the nitrous acid  
 15 (HONO) formation mechanism in winter Beijing: direct emissions and heterogeneous production in  
 16 urban and suburban areas. *Faraday Discuss.* **2016**, *189*, 213-230.
- 17 4. Tan, Z. F.; Lu, K. D.; Jiang, M. Q.; Su, R.; Wang, H. L.; Lou, S. R.; Fu, Q. Y.; Zhai, C. Z.; Tan, Q.  
 18 W.; Yue, D. L.; Chen, D. H.; Wang, Z. S.; Xie, S. D.; Zeng, L. M.; Zhang, Y. H., Daytime atmospheric  
 19 oxidation capacity in four Chinese megacities during the photochemically polluted season: a case study  
 20 based on box model simulation. *Atmos. Chem. Phys.* **2019**, *19*, (6), 3493-3513.
- 21 5. Li, D.; Xue, L.; Wen, L.; Wang, X.; Chen, T.; Mellouki, A.; Chen, J.; Wang, W., Characteristics and  
 22 sources of nitrous acid in an urban atmosphere of northern China: Results from 1-yr continuous  
 23 observations. *Atmos. Environ.* **2018**, *182*, 296-306.



- 24 6. Xu, Z.; Wang, T.; Wu, J.; Xue, L.; Chan, J.; Zha, Q.; Zhou, S.; Louie, P. K. K.; Luk, C. W. Y.,  
25 Nitrous acid (HONO) in a polluted subtropical atmosphere: Seasonal variability, direct vehicle emissions  
26 and heterogeneous production at ground surface. *Atmos. Environ.* **2015**, *106*, 100-109.
- 27 7. Yang, Q.; Su, H.; Li, X.; Cheng, Y.; Lu, K.; Cheng, P.; Gu, J.; Guo, S.; Hu, M.; Zeng, L.; Zhu, T.;  
28 Zhang, Y., Daytime HONO formation in the suburban area of the megacity Beijing, China. *Sci. China-*  
29 *Chem.* **2014**, *57*, (7), 1032-1042.
- 30 8. Liu, Y.; Lu, K.; Ma, Y.; Yang, X.; Zhang, W.; Wu, Y.; Peng, J.; Shuai, S.; Hu, M.; Zhang, Y., Direct  
31 emission of nitrous acid (HONO) from gasoline cars in China determined by vehicle chassis  
32 dynamometer experiments. *Atmos. Environ.* **2017**, *169*, 89-96.
- 33