

1 **Size-resolved exposure risk of persistent free radicals (PFRs)**  
2 **in atmospheric aerosols and their potential sources**

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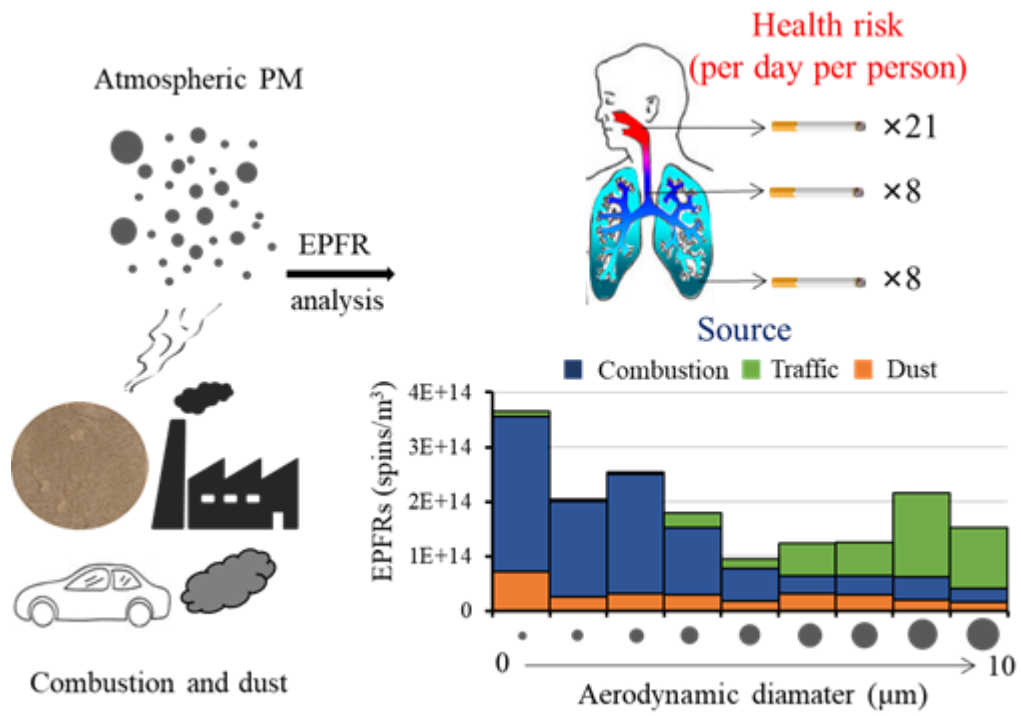
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14 **Abstract:** Environmentally persistent free radicals (EPFRs) are a new type of  
15 substance with potential health risks. EPFRs are widely present in atmospheric  
16 particulates, but there is a limited understanding of the size-resolved health risks of  
17 these radicals. This study reports the exposure risks and source of EPFRs in  
18 atmospheric particulate matter (PM) of different particle sizes ( $<10\ \mu\text{m}$ ) in Linfen, a  
19 typical coal-burning city in China. The type of EPFRs in fine particles ( $< 2.1\ \mu\text{m}$ ) is  
20 different from that in coarse particles ( $2.1\text{-}10\ \mu\text{m}$ ) in both winter and summer.  
21 However, the EPFR concentration is higher in coarse particles than in fine particles in  
22 summer, and the opposite trend is found in winter. In both seasons, combustion  
23 sources are the main sources of EPFRs with coal combustion as the major contributor  
24 in winter, while other fuels are the major source in summer. Dust contributes part of  
25 the EPFRs and it is mainly present in coarse particles in winter and the opposite in  
26 summer. The upper respiratory tract was found to be the area with the highest risk of  
27 exposure to EPFRs of the studied aerosols, with an exposure equivalent to that of  
28 approximately 21 cigarettes per person per day. Alveolar exposure to EPFRs is  
29 equivalent to 8 cigarettes per person per day, with combustion sources contributing  
30 the most to EPFRs in the alveoli. This study helps us to better understand the potential  
31 health risks of atmospheric PM with different particle sizes.

32 **Key words:** EPFRs; particle size distribution; source; generation process

33



## 36 **1. Introduction**

37 Free radicals are atoms or groups containing unpaired electrons, such as hydroxyl  
38 radicals and superoxide radicals, and they usually have strong chemical reactivity and  
39 short lifetimes (Pryor et al., 1986; Finkelstein., 1982). Free radicals with long  
40 lifetimes (months or even years) in the environment are currently called  
41 environmentally persistent free radicals (EPFRs), which have received much attention  
42 in recent years as new environmentally hazardous substances (Vejerano et al., 2018;  
43 Gehling, 2013; Chen et al., 2019c). EPFRs can be used as an active intermediate to  
44 catalyze the production of reactive oxygen species (ROS) by oxygen molecules, thus  
45 endangering human health (D'Arienzo et al., 2017; Thevenot et al., 2013; Harmon et  
46 al., 2018; Blakley et al., 2001; Khachatryan et al., 2011). Studies have found that  
47 EPFRs are present in different environmental media, such as water and soil, and even  
48 in the atmosphere (Dellinger et al., 2001; Truong et al., 2010; Vejerano et al., 2012).

49 A number of studies have investigated the occurrences, sources and formation  
50 process of EPFRs in atmospheric particulates in different regions. For example, in the  
51 studies of Rostock in Germany, Taif in Saudi Arabia and Xuanwei in China, the  
52 average concentration of EPFRs in atmospheric particulate matter (PM) was reported  
53 to be in the range of  $\sim 10^{16}$  -  $10^{18}$  spins/g (Wang et al., 2018; Arangio et al., 2016;  
54 Shaltout et al., 2015). Atmospheric EPFRs are mainly carbon-centered radicals with  
55 adjacent oxygen atoms (Gehling et al., 2013). EPFRs of different lifetimes are present  
56 in atmospheric PM, with only a few hours for short-lifetime EPFRs and several years  
57 for long-lifetime EPFRs that show no signs of decay (Gehling et al., 2013; Chen et al.,  
58 2019c). Most studies indicate that sources of transportation and combustion may be  
59 the primary EPFR sources in atmospheric PM (Wang et al., 2018; Yang et al., 2017;  
60 Chen et al., 2019b). Chen et al. (2018b and 2019b) found that strong atmospheric  
61 photochemical effects in summer and dust particles may also be important sources of  
62 EPFRs. The process of electron transfer and stabilization between the surface of metal  
63 oxides (such as iron, copper, zinc and nickel) and substituted aromatic molecules

64 under high temperatures is considered to be the main process for the formation of  
65 EPFRs in atmospheric particles (Truong., 2010; Vejerano et al., 2012a; Patterson et al.,  
66 2013; Vejerano., 2010; Vejerano et al., 2012b). However, the study by Chen et al.  
67 (2018a) suggests that EPFRs in atmospheric particulates are mainly derived from  
68 graphite oxide-like substances produced during combustion. In addition to primary  
69 sources such as combustion, secondary chemical processes in the atmosphere may  
70 also be an important source of EPFRs in atmospheric PM (Chen et al. 2019b and  
71 2019d; Tong et al., 2018).

72 Different particle sizes of atmospheric PM pose different health risks to humans,  
73 depending on the deposition efficiency of the particles and the chemical composition  
74 and concentrations of hazardous substances they contain (Strak et al., 2012;  
75 Valavanidis et al., 2008). Among various hazardous substances, EPFRs may also be  
76 involved in the toxicity of atmospheric particulates. Yang et al. (2017) studied the  
77 EPFRs that are extractable by dichloromethane in different particle sizes in Beijing in  
78 winter and found that the concentration of EPFRs was the highest in particles with  
79 sizes  $< 1 \mu\text{m}$ . Arangio et al. (2016) found that the concentration of EPFRs in 180 nm  
80 particles was the highest in the 56 nm -  $1.8 \mu\text{m}$  particle size range. Although several  
81 studies have examined the particle size distribution of EPFRs, systematic studies have  
82 not been conducted on the formation process, source and exposure assessment of  
83 EPFRs in atmospheric particles with different particle sizes.

84 This study takes Linfen as an example. Linfen is one of the cities in China with  
85 the most serious air pollution and is a typical coal-burning city. The particle size  
86 distribution of EPFRs in atmospheric PM in this region was studied by EPR  
87 spectrometry. The effects of particle size and season on the source, formation process,  
88 and health risk of EPFRs were revealed. In particular, the comprehensive health risks  
89 of EPFRs were evaluated, and it was found that the upper respiratory tract is the area  
90 with the highest risk of EPFRs exposure, which is equivalent to twenty-one cigarettes  
91 per person per day. This study is of great significance for understanding the source  
92 and formation process of EPFRs in atmospheric particulates as well as for health risk

93 assessments.

## 94 **2. Experimental section**

### 95 *2.1 Sample collection*

96 The sampling site for this study is located in Hongdong (36°23', 111°40'E) in  
97 Shanxi, China. To collect atmospheric particles of different sizes (0-10  $\mu\text{m}$ ), this study  
98 used a Thermo-Anderson Mark II sampler to collect aerosol samples of 9 sizes. The  
99 samples were collected on a prebaked quartz filter (450 °C, 4.5 hours), and the  
100 sampling dates were as follows: in winter, January 26 to February 4, 2017,  $n = 10$ ; and  
101 in summer, July 31 to August 24, 2017,  $n = 12$ . The samples were placed in a -20 °C  
102 refrigerator prior to analysis.

### 103 *2.2 EPFR analysis*

104 The EPR spectrometer (MS5000, Freiberg, Germany) is used to detect EPFRs in  
105 atmospheric samples. The filters were cut into thin strips (5 mm  $\times$  28 mm), and put it  
106 into the sample tank of the quartz tissue cell (the size of the sample tank is 10 mm  $\times$   
107 30 mm). Then the quartz tissue cell with attached filter sample was placed in a  
108 resonant cavity and analyzed by an EPR spectrometer. The detection parameters were  
109 magnetic field strength, 335 - 342 mT; detection time, 60 s; modulation amplitude,  
110 0.20 mT; number of detections, 1; and microwave intensity, 8.0 mW. Specific testing  
111 protocols have been described previously (Chen et al., 2018c).

### 112 *2.3 Carbon composition analysis*

113 The contents of organic carbon (OC) and elemental carbon (EC) in the filter  
114 samples were analyzed using a semicontinuous OC/EC analyzer (Model 4, Sunset Lab.  
115 Inc., Oregon, USA) with a NIOSH 5040 detection protocol (Lin et al., 2009).

116 The water-soluble organic carbon (WSOC) concentration was analyzed using an  
117 automatic TOC-LCPH analyzer (Shimadzu, Japan). The WSOC extraction was  
118 performed with ultrapure water under ultrasonication for 15 minutes, and all WSOC

119 concentrations were blank corrected. The concentration of OC in the MSM  
120 (Methanol-soluble materials) was calculated as the difference between the OC and  
121 WSOC (Water-soluble organic carbon) concentrations. This calculation assumes that  
122 all water-insoluble organic carbon (WISOC) in the aerosol can be extracted with  
123 MeOH, and the rationality of this assumption has been verified elsewhere (Mihara et  
124 al., 2011; Liu et al., 2013; Cheng et al., 2016; Chen et al., 2019a).

#### 125 *2.4 PAH analysis*

126 PAHs were detected using gas chromatography/mass spectrometry (GC/MS) on a  
127 GC7890B/MS5977A (Agilent Technologies, Clara, CA). Quartz-fiber filter samples  
128 (8 mm in diameter) were cut from each 25-mm quartz-fiber filter substrates used on  
129 the ELPI impactor stages using a stainless-steel round punch over a clean glass dish  
130 and loaded into the TD glass tube. Next, the TD glass tube was heated to 310 °C at a  
131 rate of 12 °C/min and thermally desorbed at 310 °C for 3 min. The desorbed organic  
132 compounds were trapped on the head of a GC-column (DB-5MS: 5% diphenyl-95%  
133 dimethyl siloxane copolymer stationary phase, 0.25-mm i.d., 30-m length, and  
134 0.25-mm thickness). Sixteen target PAHs were identified based on retention time and  
135 typical ion fragments of each PAH standards, including 16 EPA parent PAHs  
136 (p-PAHs). The method detection limits (MDLs) ranged from 0.2 pg/mm<sup>2</sup> (Ace) to 0.6  
137 pg/mm<sup>2</sup> (Incdp). Naphthalene-D8, acenaphthene-D10, phenanthrene-D10,  
138 chrysene-D12, and perylene D12 were used for the analytical recovery check. All  
139 compounds were recovered with a desorption recovery percentage of > 90%. Specific  
140 testing protocols have been described previously (Han et al., 2018; Song et al., 2020).

#### 141 *2.5 Metal element analysis*

142 The concentration of metal elements in the samples was determined by a Thermo  
143 X2 series inductively coupled plasma mass spectrometer (ICP-MS, Thermo, USA).  
144 The metal elements analyzed in summer were Na, Mg, K, Ca, Ti, V, Cr, Mn, Fe, Co,  
145 Ni, Cu, Zn, As, Cd, Pb, and Al, and those in winter were Al, Zn, V, Cr, Mn, Co, Ni, Cu,  
146 As, Se, Sr, Cd, Ba, and Pb. The specific measurement method is based on the study of

147 Qi et al (2016).

## 148 2.6. Data statistics method

149 The source and formation process of EPFRs in PM with different particle sizes  
150 were analyzed by nonnegative matrix factorization (NMF). The method is based on  
151 the study of Chen et al (2016 and 2019e). Briefly, NMF analysis of EPFR data, metal  
152 element contents, OC/EC contents and PAH contents was performed in MATLAB.  
153 The version of the NMF toolbox is 1.4 (<https://sites.google.com/site/nmftool/>). A  
154 gradient-based multiplication algorithm was used to find a solution from multiple  
155 random starting values, and then the first algorithm was used to find the final solution  
156 based on the least-squares effective-set algorithm. To find a global solution, the model  
157 was run 100 times, each time with a different initial value. By comparing the 1-12  
158 factor model (Figure S4) with the residual of the spectral load, the 6 factor (summer)  
159 and 10 factor (winter) NMF models were finally selected.

## 160 2.7. EPFR exposure evaluation

161 To assess the health risks of EPFRs, this study divided the respiratory system into  
162 three parts based on the human breathing model: extrathoracic (ET) areas, including  
163 the anterior nasal cavity, posterior nasal cavity, oral cavity, and throat;  
164 tracheobronchial (TB) areas, including the trachea, bronchi, bronchioles, and terminal  
165 bronchi; and pulmonary (P) areas, including the alveolar ducts and alveoli. Then, the  
166 sedimentation rates of different particle sizes in different areas of the respiratory  
167 system were determined to calculate the exposure risk of EPFRs. Here, the human  
168 respiratory system particulate deposition model of Salma et al. (2002) was used, and  
169 the specific data can be found in Table S3 and S4.

170 In addition, the daily inhaled concentration of EPFRs into the concentration of free  
171 radicals in cigarettes were converted. The specific conversion method is as follows:

$$172 \quad N_{\text{cig}} = (C_{\text{EPFRs}} \cdot V) / (RC_{\text{cig}} \cdot C_{\text{tar}}) \quad (1)$$

173 where  $N_{\text{cig}}$  represents the number of cigarettes (/person/day),  $C_{\text{EPFRs}}$  (spins/m<sup>3</sup>)



174 represents the atmospheric concentration of EPFRs in PM, and V represents the  
175 amount of air inhaled by an adult per day ( $20 \text{ m}^3/\text{day}$ ) (Environmental Protection  
176 Agency, 1988).  $RC_{\text{cig}}$  ( $4.75 \times 10^{16}$  spins/g) (Baum et al., 2003; Blakley et al., 2001;  
177 Pryor et al., 1983; Valavanidis and Haralambous, 2001) indicates the concentration of  
178 free radicals in cigarette tar, and  $C_{\text{tar}}$  (0.013 g/cig) indicates the amount of tar per  
179 cigarette (Gehling et al., 2013).

### 180 **3. Results and discussion**

#### 181 *3.1 Concentrations and types of EPFRs*

182 Figure 1a shows the concentration distribution of EPFRs with different particle  
183 sizes in different seasons. EPFRs were detected in the particles of each tested size (the  
184 EPR spectrum is shown in Figure S1), but their EPFR concentration levels were  
185 different. In summer, the concentration of EPFRs in fine particles (particle size  $< 2.1$   
186  $\mu\text{m}$ ) is  $(3.2 - 8.1) \times 10^{13}$  spins/ $\text{m}^3$ , while the concentration of EPFRs in coarse  
187 particles (particle size  $> 2.1 \mu\text{m}$ ) is 1-2 orders of magnitude higher than that of fine  
188 particles, reaching values of  $(2.2 - 3.5) \times 10^{14}$  spins/ $\text{m}^3$ . Winter samples show  
189 completely different characteristics from summer samples. The concentration of  
190 EPFRs in fine particles (particle size  $< 2.1 \mu\text{m}$ ) is  $(1.8 - 3.6) \times 10^{14}$  spins/ $\text{m}^3$ , while the  
191 concentration of EPFRs in coarse particles (particle size  $> 2.1 \mu\text{m}$ ) is smaller than that  
192 of fine particles, with values of  $(1.0 - 2.1) \times 10^{14}$  spins/ $\text{m}^3$ . In addition, the  
193 concentration of EPFRs in particulates  $< 0.43 \mu\text{m}$  in winter is very high, but it is very  
194 low in summer. According to the results of factor analysis in part 3.2 of this study, this  
195 particulate matter is related to combustion, which indicates that coal combustion in  
196 winter may provide an important contribution to EPFRs. The EPFR concentration in  
197 the fine PM of Linfen reported above is equivalent to that in the fine PM of Xi'an, but  
198 it is ten times smaller than that in the fine PM of Beijing (Yang et al., 2017; Chen et  
199 al., 2019b). Although the particle size distribution characteristics of EPFRs in winter  
200 and summer are different, their concentration levels are similar, which indicates that  
201 the EPFR concentration is not related to the PM concentration, but is determined by

202 the source characteristics. The source characteristics will be discussed in detail in the  
203 factor analysis section.

204 Figure 1b shows the concentration ratio of EPFRs in coarse and fine particles. The  
205 contribution of EPFRs in fine PM in summer is only 14.9%, while in winter is 58.5%.  
206 The differences in EPFR concentrations with particle size may be related to the source  
207 of EPFRs. For example, coarse particles are often associated with dust sources and  
208 biogenic aerosols. In another study, the results have shown that dust particles contain  
209 large amounts of metallic EPFRs and that they can be transported over long distances  
210 (Chen et al., 2018b). EPFRs in fine particles may be mainly derived from the  
211 combustion process, such as traffic sources, which are considered to be an important  
212 source of EPFRs in atmospheric PM (Secrest et al., 2016; Chen et al., 2019b). Due to  
213 winter heating in the Linfen area, the amount of coal burning increases sharply in this  
214 season. In 2017, the nonclean heating (Coal-fired heating) rate of urban heating  
215 energy structures in Linfen was 40% (data source: <http://www.linfen.gov.cn/>). With  
216 the burning of coal, large amounts of EPFRs are produced, and in the summer, EPFRs  
217 emitted by burning coal should be much less than those emitted in winter. This can  
218 explain to a certain extent that the contribution of fine particles to summer EPFRs is  
219 small, and the contribution of winter EPFRs is very large.

220 The  $g$ -factor obtained by using EPR to analyze the sample is an important  
221 parameter to distinguish the type of EPFR. It is the ratio of the electronic magnetic  
222 moment to its angular momentum (Shaltout et al., 2015; Arangio et al., 2016). The  
223  $g$ -factor of carbon-centered persistent free radicals is generally less than 2.003, the  
224  $g$ -factor of oxygen-centered persistent radicals is generally greater than 2.004, and the  
225  $g$  factor of carbon-centered radicals with adjacent oxygen atoms is between 2.003 and  
226 2.004 (Cruz et al., 2012). Figure 2a shows the  $g$ -factor distribution characteristics of  
227 EPFRs in different particle sizes in summer and winter. The  $g$ -factor of fine particles  
228 and coarse particles shows different characteristics. The  $g$ -factor of EPFRs in fine  
229 particles (particle size  $< 2.1 \mu\text{m}$ ) ranges from 2.0034 to 2.0037, which may be from  
230 carbon-centered radicals with adjacent oxygen atoms. However, the  $g$ -factor of

231 EPFRs in coarse particles (particle size  $> 2.1 \mu\text{m}$ ) is significantly less than that of fine  
232 particles. The  $g$ -factor ranges from 2.0031 to 2.0033, indicating that EPFRs in coarse  
233 particles are more carbon-centered than those in fine particles and are free of  
234 heteroatoms. As shown in Figure 2b, the  $g$ -factor varied differently depending on  
235 season. The  $g$ -factor of summer PM showed a significant decreasing trend with  
236 increasing concentration, while the  $g$ -factor of winter PM showed a significant  
237 increasing trend with increasing EPFR concentration. Oyana et al. (2017) studied  
238 EPFRs in the surface dust of leaves in the Memphis region of the United States and  
239 found that the concentration of EPFRs was positively correlated with the  $g$ -factor, and  
240 they believed that this was related to the source of EPFRs. This phenomenon indicates  
241 that the sources and toxicity of EPFRs in winter and summer are different.

### 242 *3.2 Factor Analysis of EPFRs*

243 To explore the possible sources and formation process of EPFRs in atmospheric  
244 particles with different particle sizes, the NMF model was used to statistically analyze  
245 EPFRs, carbon components, PAHs and metal elements in samples. The factors  
246 obtained by the NMF model should reflect the different sources and generation  
247 process of EPFRs. As shown in Figure 3a1 and b1, the three main contributing factors  
248 to EPFRs in summer and winter are shown (see Figure S5, S6 for spectra of other  
249 factors), which explain 94.5% and 83.8% of the EPFR concentrations in summer and  
250 winter, respectively.

251 As shown in Figure 3a1, the typical spectral characteristic of summer factor 1 is  
252 that it contains a small fraction of EC components and a large amount of OC  
253 components, which indicates that combustion may be the source associated with this  
254 factor. This factor has the highest loading of OC, especially WISOC; this fraction  
255 mainly contains macromolecular organic substances, which are considered to  
256 contribute to the main atmospheric particulate EPFRs and to be graphite oxide-like  
257 substances (Chen et al., 2017; Chen et al., 2018a). Factor 2 is different from factor 1;  
258 factor 2 is more likely the combustion of fossil fuels, while factor 1 should be other  
259 combustion sources instead of burning coal, such as biomass combustion. The

260 generation process is similar to a hybrid process, which includes the graphite  
261 oxide-like substances produced by incomplete combustion and the EPFRs formed by  
262 some metal oxides. The typical characteristic of factor 3 is that the contribution of  
263 metal elements is relatively high, while the contributions of EC and OC are very low.  
264 Metal elements such as Al, Ti, Mn, and Co are typical crust elements, so this factor  
265 may represent dust sources (Pan et al., 2013; Srivastava et al., 2007; Trapp et al.,  
266 2010). The generation process of EPFRs. The others are likely derived from the  
267 electroplating metallurgy industry (detailed in S1). As shown in Figure 3a2, the  
268 contribution ratios of different factors show that the contribution ratios of factor 1 and  
269 factor 2 are the highest, and factor 3 has only a small contribution, which indicates  
270 that combustion sources, especially incomplete combustion, are the main sources of  
271 EPFRs. The particle size distribution characteristics show that factor 1 is mainly  
272 distributed in particles larger than 2.1  $\mu\text{m}$ , while factor 2 is mainly distributed in  
273 particles smaller than 0.43  $\mu\text{m}$ .

274 The results of the factor analysis in winter are different from those in summer. As  
275 shown in Figure 3b1, the typical spectral characteristic of factor 1 is that it contains a  
276 large amount of OC components and As and Se. As and Se are trace elements of coal  
277 combustion, as shown in many studies (Pan et al., 2013; Tian et al., 2010), so coal  
278 combustion may be the source represented by this factor. From the generation process  
279 viewpoint, the factor does not contain EC, but the content of OC is very high. In the  
280 particles with a particle size of less than 3.3, which is mainly present in factor 1, the  
281 concentration of OC is 16 times that of EC. So it may be mainly a graphite oxide-like  
282 substance formed by the agglomeration of gaseous volatile organic compounds  
283 (VOCs) generated during combustion. The typical spectral characteristics of factor 2  
284 are due to a large amount of V and some Al, EC and OC. OC and EC are also typical  
285 combustion products. V is rich in fossil fuels, especially fuel oil (Karnae et al., 2011).  
286 Therefore, traffic is the source represented by this factor. The factor contains crust  
287 elements such as Al and Mn, so it is speculated that this factor may also include  
288 traffic-related dust. The typical spectral characteristics of factor 3 are similar to those

289 of factor 1, and both contain relatively large amounts of As and Se, with the exception  
290 that factor 3 contains a large amount of EC, indicating that it is also mainly derived  
291 from incomplete combustion sources. The generation process of factor 3 should be  
292 different from factor 1, which may include both the graphite oxide-like material  
293 generated by fuel coking and the EPFRs generated by the metal oxide. The other  
294 factors are mainly atmospheric dust and electroplating or metallurgy (see text S1). As  
295 shown in Figure 3b2, factor 1 and factor 2 have the highest proportions, and factor 3  
296 also has a small contribution, which indicates that winter is the same as summer, and  
297 combustion sources are the main source of EPFRs. The particle size distribution  
298 characteristics show that factor 1 is mainly distributed in particles with a size of 0.43 -  
299 3.3  $\mu\text{m}$ , while factor 2 is mainly distributed in particles larger than 3.3  $\mu\text{m}$ .

300 Based on the above analysis, it can be found that combustion sources are the main  
301 sources of EPFRs, and EPFRs from these sources are mainly graphite oxide-like  
302 substances generated by the polymerization of organic matter or fuel coking. Studies  
303 have shown that graphene oxide can cause cell damage by generating ROS (Seabra et  
304 al., 2014). The surface of these compounds contains not only carbon atoms but also  
305 some heteroatoms, which leads to disorder and the presence of defects in the  
306 carbon-based structure (Lyu et al., 2018; Chen et al., 2017a; Mukome et al., 2013;  
307 Keiluweit et al., 2010). The dust source is also a source of important EPFRs identified  
308 in this study (with a contribution of approximately 10%). It was shown in the above  
309 analysis that the concentration of EPFRs in coarse particles has a significant  
310 correlation with the concentration of metallic elements, particularly crustal elements.  
311 Some crustal elements, such as Al, and Fe, not only have their own paramagnetism  
312 (Li et al., 2017; Yu et al., 2013; Nikitenko et al., 1992), but also interact with aromatic  
313 compounds attached to the surface of the particles to produce a stable single-electron  
314 structure.

### 315 3.3 *Health risk of EPFRs*

316 To evaluate the health risks of EPFRs in PM with different particle sizes, this study  
317 evaluated the comprehensive exposure of EPFRs based on the deposition efficiency of

318 PM with different particle sizes in different parts of the human body. The results are  
319 shown in Figure 4a. The ET region is the region with the highest EPFR exposure,  
320 while the TB and P regions have relatively close EPFRs. This result shows that  
321 atmospheric EPFRs are the most harmful to the health of the human upper respiratory  
322 tract. Comparing the EPFR exposure in different seasons indicates that the exposure  
323 risk in the ET area in summer is significantly higher than that in winter. This  
324 difference occurs because the concentration of EPFRs in coarse particles is much  
325 higher than that of fine particles in summer and the deposition efficiency of large  
326 particles in the ET area is generally higher. Fine particles are more efficiently  
327 deposited in the P region, leading to a higher risk of EPFR exposure in the P region in  
328 winter.

329 EPFRs were found early in cigarette tar and are considered one of the health risk  
330 factors in cigarette smoke (Lyons et al., 1960); thus, in this study, the exposure risks  
331 of EPFRs in particles deposited in the human body were converted to the equivalent  
332 number of cigarettes inhaled per adult per day. As shown in Figure 4b, the ET area is  
333 the most contaminated area, with an average equivalence of twenty-one cigarettes  
334 (twenty-five in summer and sixteen in winter). The average values for the TB area  
335 (nine in summer and seven in winter) and P area (seven in summer and ten in winter)  
336 are eight. The results indicate that EPFRs pose significant health risks to human lungs  
337 in both winter and summer. Other similar studies, such as a study of the average  
338 amount of EPFRs in PM<sub>2.5</sub> inhaled per person per day in Xi'an in 2017, found values  
339 equivalent to approximately 5 cigarettes (Chen et al., 2018a). Gehring and Dellinger  
340 (2013) found that EPFR exposure in PM<sub>2.5</sub> is equivalent to approximately 0.3  
341 cigarettes per person per day in St. Joaquin County, the location with the worst air  
342 pollution in the United States. The average exposure risk of EPFRs in fine particles in  
343 the Linfen area (approximately 13 cigarettes) was higher than those in these two  
344 studies. However, these previous studies only studied the exposure risk of EPFRs in  
345 fine particles. The results of this study indicate that the health risks of EPFRs are  
346 significantly increased when the particle size distribution of EPFRs is taken into

347 account. Therefore, it is important to study the source characteristics and generation  
348 process of EPFRs with different particle sizes, which will be discussed in detail in the  
349 following sections.

350 This study calculated the proportion of EPFRs with different particle sizes in  
351 different parts of the respiratory system based on the deposition efficiency of particles  
352 with different particle sizes. As shown in Figure 4c, in the ET region and the TB  
353 region, coarse particles are the dominant component in summer and winter. In  
354 particular, in summer, the proportion of EPFRs in coarse particles in these two regions  
355 exceeds 95%. In the P region, there are significant differences between summer and  
356 winter. The P region in summer is still dominated by coarse particles, but its  
357 proportion is significantly lower than those in the ET and TB regions. In the P region  
358 in winter, fine particles are the dominant component (approximately 70%). These  
359 distribution characteristics indicate different sources of EPFRs in different regions. As  
360 shown in Figure 4d, in summer, combustion sources are the main source of EPFRs in  
361 the respiratory system. In winter, combustion and transportation sources contribute  
362 equally in the EP and ET regions, while in the alveoli, combustion sources are the  
363 main contributor. The ET region is the area with the highest risk of exposure to  
364 EPFRs (21 cigarettes). The generation process of these EPFRs is mainly attributable  
365 to graphene oxide-like substances. Studies have shown that graphene oxide is  
366 cytotoxic (Harmon et al., 2018). In the alveoli, the contribution of combustion sources  
367 is significantly increased (especially in winter). These EPFRs are mainly generated by  
368 the action of metal oxides and organic substances. Studies have shown that such  
369 EPFRs can generate ROS in the lung fluid environment (Khachatryan et al., 2011).

#### 370 **4. Conclusions and environmental implications**

371 This study systematically reported the particle size distribution of EPFRs in  
372 atmospheric PM in Linfen, which is one of the most polluted cities in China and is  
373 located in a typical coal-burning area. In addition, this study evaluated the  
374 comprehensive health risks of EPFRs, and reported possible sources and formation

375 process of atmospheric EPFRs with respect to different particle sizes. The following  
376 main conclusions were obtained.

377 (1) This study found that EPFRs are widely present in atmospheric particles of  
378 different particle sizes and exhibit significant particle size distribution characteristics.  
379 The results of this study demonstrate that the concentrations and types of EPFRs are  
380 dependent on particle size and season. This seasonal characteristic of EPFRs is mainly  
381 affected by the PM sources, this result also indicates that the potential toxicity caused  
382 by EPFRs may also vary with particle size and season.

383 (2) This study reported the possible source and formation process of atmospheric  
384 EPFRs in different particle sizes. The results show that combustion is the most  
385 important source of EPFRs (>70%) in both winter and summer PM samples in Linfen.  
386 The graphite oxide-like process has the highest contribution (~70%) and is mainly  
387 distributed in particles with a size of > 0.43  $\mu\text{m}$ . These findings deepen our  
388 understanding of the pollution characteristics of atmospheric EPFRs and are useful for  
389 controlling EPFR generation in heavily polluted areas.

390 (3) This study assessed the exposure risk of EPFRs in different areas of the  
391 respiratory system. The results show that the upper respiratory tract is the area with  
392 the highest EPFR exposure. The trachea and alveoli are also exposed to EPFRs, and  
393 the risk of exposure is equivalent to that of 8 cigarettes per person per day. Coarse  
394 particles are the main source of EPFRs in the upper respiratory tract, while fine  
395 particles are mainly involved in the alveoli.

396 Through this study, the results have shown that there are significant differences in  
397 the concentrations and types of EPFRs in particles of different sizes and these  
398 differences are due to the influence of the source and generation process. In the future,  
399 assessments of the particle size distribution and the seasonality of EPFRs in  
400 atmospheric PM should be considered. Health risks are another focus of this study. It  
401 is found that the upper respiratory tract is the key exposure area of EPFRs, and the  
402 traffic source is the main source of EPFRs in this area. This finding is significant for a  
403 systematic assessment of the health risks of EPFRs. In view of the complexity and



404 diversity of the formation process of EPFRs in actual atmospheric particulates, the  
405 relative contributions of EPFRs generated by different process and their associated  
406 health risks should be more comprehensively studied in the future.

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412 Innovation & Entrepreneurship Team.

## 413 **Appendix A. Supplementary data**

414 Appendix A contains additional details, including the EPR spectra of samples of  
415 different particle sizes, correlations between EPFRs and carbon in particles of  
416 different particle sizes, the results and errors of factor analysis, correlation analysis of  
417 EPFRs with metallic elements, and EPFR exposure in different areas of the human  
418 respiratory tract.

419 **Code/Data availability:** All data that support the findings of this study are  
420 available in this article and its Supplement or from the corresponding author on  
421 request.

422 **Author contribution:** Qingcai Chen: Research design, Methodology, Writing -  
423 Original Draft, Writing - Review & Editing, Project administration, Funding  
424 acquisition; Haoyao Sun: Investigation, Sample analysis, Writing - Original Draft,  
425 Writing - Review & Editing, Methodology, Formal analysis; Wenhui Song:  
426 Investigation, Sample collection, Chemical analysis; Fang Cao: Investigation, Sample  
427 collection; Chongguo Tian: Investigation, Chemical analysis; Yan-Lin Zhang:  
428 Conceptualization, Writing - Review & Editing, Formal analysis, Validation, Funding  
429 acquisition.

430 **Competing interests:** The authors declare that they have no conflict of interest.

## 431 **References**

432 Arangio, A. M., Tong, H., Socorro, J., Pöschl, U., Shiraiwa, M., 2016. Quantification of  
433 environmentally persistent free radicals and reactive oxygen species in atmospheric aerosol  
434 particles. *Atmos. Chem. Phys.* 16 (20), 13105–13119.

435 Blakley, R. L., Henry, D. D., Smith, C. J., 2001. Lack of correlation between cigarette mainstream  
436 smoke particulate phase radicals and hydroquinone yield. *Food. Chem. Toxicol.* 39 (4),  
437 401–406.

438 Baum, S.L., Anderson, I.G.M., Baker, R.R., Murphy, D.M., Rowlands, C.C., 2003. Electron spin  
439 resonance and spin trap investigation of free radicals in cigarette smoke: development of a  
440 quantification procedure. *Anal. Chim. Acta* 481, 1–13.

441 Cruz, A.L.N.D., Cook, R.L., Lomnicki, S.M., Dellinger, B., 2012. Effect of low temperature  
442 thermal treatment on soils contaminated with pentachlorophenol and environmentally  
443 persistent free radicals. *Environ. Sci. Technol.* 46, 5971–5978.

444 Chen, N., Huang, Y., Hou, X., Ai, Z., Zhang, L., 2017. Photochemistry of hydrochar: Reactive  
445 oxygen species generation and sulfadimidine degradation. *Environ. Sci. Technol.* 51 (19),  
446 11278–11287.

447 Chen, Q., Mu, Z., Song, W., Wang, Y., Yang, Z., Zhang, L., Zhang, Y., 2019a. Size-resolved  
448 characterization of the chromophores in atmospheric particulate matter in Linfen, China. *J.*  
449 *Geophys. Res-Atmos.* 124, DIO: 10.1029/2019JD031149.

450 Chen, Q., Ikemori, F., Nakamura Y., Vodicka, P., Kawamura, K., Mochida, M., 2017. Structural  
451 and light-absorption characteristics of complex water-insoluble organic mixtures in urban  
452 submicron aerosols. *Environ. Sci. Technol.* 51(15), 8293–8303.

453 Chen, Q., Miyazaki, Y., Kawamura, K., Matsumoto, K., Coburn, S., Volkamer, R., Iwamoto, Y.,  
454 Kagami, S., Deng, Y., Ogawa, S., 2016. Characterization of chromophoric water-soluble  
455 organic matter in urban, forest, and marine aerosols by HR-ToF-AMS analysis and  
456 excitation–emission matrix spectroscopy. *Environ. Sci. Technol.* 50 (19), 10351–10360.

457 Chen, Q., Sun, H., Mu, Z., Wang, Y., Li, Y., Zhang, L., Wang, M., Zhang, Z., 2019b.  
458 Characteristics of environmentally persistent free radicals in PM<sub>2.5</sub>: Concentrations, species  
459 and sources in Xi'an, Northwestern China. *Environ. Pollut.* 247, 18–26.

460 Chen, Q., Sun, H., Wang, J., Shan, M., Xue, J., Yang, X., Deng, M., Wang, Y., Zhang, L., 2019c.  
461 Long-life type — The dominant fraction of EPFRs in combustion sources and ambient fine  
462 particles in Xi'an. *Atmos. Environ.* 219, 117059.

463 Chen, Q., Sun, H., Wang, M., Mu, Z., Wang, Y., Li, Y., Wang, Y., Zhang, L., Zhang, Z., 2018a.  
464 Dominant fraction of EPFRs from Nonsolvent-Extractable organic matter in fine particulates  
465 over Xi'an, China. *Environ. Sci. Technol.* 52 (17), 9646–9655.

466 Chen, Q., Sun, H., Wang, M., Wang, Y., Zhang, L., Han, Y., 2019d. Environmentally persistent  
467 free radical (EPFR) formation by visible-light illumination of the organic matter in  
468 atmospheric particles. *Environ. Sci. Technol.* 53 (17), 10053–10061.

469 Chen, Q., Wang, M., Sun, H., Wang, X., Wang, Y., Li, Y., Zhang, L., Mu, Z., 2018b. Enhanced  
470 health risks from exposure to environmentally persistent free radicals and the oxidative stress  
471 of PM<sub>2.5</sub> from asian dust storms in erenhot, Zhangbei and Jinan, China. *Environ. Int.* 123,  
472 260–268.

473 Chen, Q., Wang, M., Wang, Y., Zhang, L., Li, Y., Han, Y., 2019e. Oxidative potential of  
474 water-soluble matter associated with chromophoric substances in PM<sub>2.5</sub> over Xi'an, China.  
475 *Environ. Sci. Technol.* 53 (17), 10053–10061.

476 Chen, Q., Wang, M., Wang, Y., Zhang, L., Xue, J., Sun, H., Mu, Z., 2018c. Rapid determination of  
477 environmentally persistent free radicals (EPFRs) in atmospheric particles with a quartz  
478 sheet-based approach using electron paramagnetic resonance (EPR) spectroscopy. *Atmos.*  
479 *Environ.* 184, 140–145.

480 Cheng, Y., He, K. B., Du, Z. Y., Engling, G., Liu, J. M., Ma, Y. L., Zheng, M., Weber, R. J., 2016.  
481 The characteristics of brown carbon aerosol during winter in Beijing. *Atmos. Environ.* 127,  
482 355–364.

483 Cormier, S. A., Lomnicki, S., Backes, W., Dellinger, B., 2006. Origin and health impacts of  
484 emissions of toxic by-products and fine particles from combustion and thermal treatment of  
485 hazardous wastes and materials. *Environ. Health. Perspect.* 114 (6), 810–817.

486 D'Arienzo, M., Gamba, L., Morazzoni, F., Cosention, U., Creco, C., Lasagni, M., Pitea, D., Moro,  
487 G., Cepek, C., Butera, V., Sicilia, E., Russo, N., Muñoz-García, A., Pavone, M., 2017.  
488 Experimental and theoretical investigation on the catalytic generation of environmentally  
489 persistent free radicals from benzene. *J. Phys. Chem. A.* 121 (17), 9381–9393.

490 Dellinger, B., Lomnicki, S., Khachatryan, L., Maskos, Z., Hall, R. W., Adoukpe, J., McFerrin, C.,  
491 Truong, H., 2007. Formation and stabilization of persistent free radicals. *Proc. Combust. Inst.*  
492 31 (1), 521–528.

493 Dellinger, B., Pryor, W. A., Cueto, R., Squadrito, G. L., Hegde, V., Deutsch, W. A., 2001. Role of  
494 free radicals in the toxicity of airborne fine particulate matter. *Chem. Res. Toxicol.* 14 (10),  
495 1371–1377.

496 Environmental Protection Agency, 1988. Recommendations for and Documentation of Biological  
497 Values for Use in Risk Assessment. PB-179874. EPA 600/6-87/008. US Environmental  
498 Protection Agency, Cincinnati, OH.

499 Finkelstein, E., Rosen, G. M., Rauckman, E. J., 1982. Production of hydroxyl radical by  
500 decomposition of superoxide spin-trapped adducts. *Mol. Pharmacol.* 21 (2), 262–265.

501 Gehling, W., Dellinger, B., 2013. Environmentally persistent free radicals and their lifetimes in  
502 PM<sub>2.5</sub>. *Environ. Sci. Technol.* 47 (15), 8172–8178.

503 Han, Y., Chen, Y. J., Saud, A., Feng, Y. L., Zhang, F., Song, W. H., Cao, F., Zhang, Y., Yang, X., Li,  
504 J., Zhang, G., 2018. High time- and size-resolved measurements of PM and chemical  
505 composition from coal combustion: Implications for the EC formation process. *Environ. Sci.*  
506 *Technol.* 52 (11), 6676–6685.

507 Harmon, A. C., Hebert, V. Y., Cormier, S. A., Subramanian, B., Reed, J. R., Backes, W. L., Dugas,  
508 T. R., 2018. Particulate matter containing environmentally persistent free radicals induces  
509 AhR-dependent cytokine and reactive oxygen species production in human bronchial  
510 epithelial cells. *Plos. One.* 13 (10), e0205412.

511 Karnae, S., John, K., 2011. Source apportionment of fine particulate matter measured in an  
512 industrialized coastal urban area of South Texas. *Atmos. Environ.* 45 (23), 3769–3776.

513 Keiluweit, M., Nico, P. S., Johnson, M. G., Kleber, M., 2010. Dynamic molecular structure of  
514 plant biomass-derived black carbon (biochar). *Environ. Sci. Technol.* 44 (4), 1247–1253.

515 Khachatryan, L., Dellinger, B., 2011. Environmentally persistent free radicals (EPFRs)-2. Are free  
516 hydroxyl radicals generated in aqueous solutions?. *Environ. Sci. Technol.* 45 (21),  
517 9232–9239.

518 Li, G. L., Wu, S. Y., Kuang, M. Q., Hu, X. F., Xu, Y. Q., 2017. Studies on the g-factors of the  
519 copper(II)-oxygen compounds. *J. Struct. Chem.* 58 (4), 700–705.

520 Lin, P., Hu, M., Deng, Z., Slanina, J., Han, S., Kondo, Y., Takegawa, N., Miyazaki, Y., Zhao, Y.,  
521 Sugimoto, N., 2009. Seasonal and diurnal variations of organic carbon in PM<sub>2.5</sub> in Beijing  
522 and the estimation of secondary organic carbon. *J. Geophys. Res-Atmos.* 114, 1–41.

523 Liu, J., Bergin, M., Guo, H., King, L., Kotra, N., Edgerton, E., Weber, R. J., 2013. Size-resolved  
524 measurements of brown carbon in water and methanol extracts and estimates of their  
525 contribution to ambient fine-particle light absorption. *Atmos. Chem. Phys.* 13, 12389–12404.

526 Lomnicki, S., Truong, H., Vejerano, E., Delligner, B., 2008. Copper oxide-based model of  
527 persistent free radical formation on combustion-derived particulate matter. *Environ. Sci.*  
528 *Technol.* 42 (13), 4982–4988.

529 Lyu, L., Yu, G., Zhang, L., Hu, C., Sun, Y., 2018. 4-Phenoxyphenolfunctionalized reduced  
530 graphene oxide nanosheets: A metal-free fenton-like catalyst for pollutant destruction.  
531 *Environ. Sci. Technol.* 52 (2), 747–756.

532 Lyons, M.J., Spence, J.B., 1960. Environmental free radicals. *Br. J. Canc.* 14, 703–708

533 Mihara, T., Michihiro, M., 2011. Characterization of solvent-extractable organics in urban aerosols  
534 based on mass spectrum analysis and hygroscopic growth measurement. *Environ. Sci.*  
535 *Technol.* 45 (21), 9168–9174.

536 Mukome, F. N. D., Zhang, X., Silva, L. C. R., Six, J., Parikh, S. J., 2013. Use of Chemical and  
537 physical characteristics to investigate trends in biochar feedstocks. *J. Agric. Food Chem.* 61  
538 (9), 2196–2204.

539 Nikitenko, V. A., 1992. Luminescence and EPR of zinc oxide (review). *J. Appl. Spectrosc.* 57  
540 (5–6), 783–798.

541 Oyana, T. J., Lomnicki, S. M., Guo, C., Cormier, S. A., 2017. A scalable field study protocol and  
542 rationale for passive ambient air sampling: a spatial phytosampling for leaf data collection.  
543 *Environ. Sci. Technol.* 51 (18), 10663–10673.

544 Pan, Y., Wang, Y., Sun, Y., Tian, S., Cheng, M., 2013. Size-resolved aerosol trace elements at a  
545 rural mountainous site in Northern China: importance of regional transport. *Sci. total Environ.*  
546 461–462, 761–771.

547 Patterson, M. C., Keilbart, N. D., Kiruri, L. W., Thibodeaux, C. A., Lomnicki, S., Kurtz, R. L.,  
548 Poliakoff, E. D., Dellinger, B., Sprunger, P. T., 2013. EPFR formation from phenol adsorption  
549 on Al<sub>2</sub>O<sub>3</sub> and TiO<sub>2</sub>: EPR and EELS studies. *Chem. Phys.* 422, 277–282.

550 Pryor, W.A., Prier, D.G., Church, D.F., 1983. Electron-spin resonance study of mainstream and  
551 sidestream cigarette smoke: nature of the free radicals in gas-phase smoke and in cigarette tar.  
552 *Environ. Health Perspect.* 47, 345–355.

553 Pryor, W. A., 1986. Oxy-Radicals and Related Species: Their Formation, Lifetimes, and Reactions.  
554 *Annu. Rev. Physiol.* 48, 657–667.

555 Qi, L., Zhang, Y., Ma, Y., Chen, M., Ge, X., Ma, Y., Zheng, J., Wang, Z., Li, S., 2016. Source  
556 identification of trace elements in the atmosphere during the second Asian Youth Games in  
557 Nanjing, China: Influence of control measures on air quality. *Atmos. Pollut. Res.* 7, 547–556.

558 Shaltout, A. A., Boman, J., Shehadeh, Z. F., Al-Malawi, D. A. R., Hemeda, O. M., Morsy, M. M.,  
559 2015. Spectroscopic investigation of PM<sub>2.5</sub>, collected at industrial, residential and traffic  
560 sites in taif. Saudi Arabia. *J. Aerosol. Sci.* 79, 97–108.

561 Song, W., Cao F., Lin Y., Haque, M. M., Wu, X., Zhang, Y., Zhang, C., Xie, F., Zhang Y., 2020.  
562 Extremely high abundance of polycyclic aromatic hydrocarbons in aerosols from a typical

563 coal-combustion rural site in China: Size distribution, source identification and cancer risk  
564 assessment. *Atmos. Res.* 248, 105192.

565 Srivastava, A., Jain, V. K., 2007. Size distribution and source identification of total suspended  
566 particulate matter and associated heavy metals in the urban atmosphere of Delhi.  
567 *Chemosphere.* 68(3), 579–589.

568 Strak, M., Janssen, N. A. H., Godri, K. J., Gosens, I., Mudway, I. S., Cassee, F. R., Lebret, E.,  
569 Kelly F. J., Harrison, R. M., Brunekreef, B., Steenhof, M., Hoek, G., 2012. Respiratory health  
570 effects of airborne particulate matter: The role of particle size, composition, and oxidative  
571 potential—the RAPTES project. *Eviron. Health. Persp.* 120 (8), 1183–1189.

572 Salma, I., Balásházy, I., Winkler-Heil, R., Hofmann, W., Zárny, G. 2002. Effect of particle mass  
573 size distribution on the deposition of aerosols in the human respiratory tract. *J. Aerosol. Sci.*  
574 33(1), 119-132.

575 Seabra A.B., Paula A.J., Lima R. D., Alves. O.L., Durán. N. 2014. Nanotoxicity of graphene and  
576 graphene oxide. *Chem. Res. Toxicol.* 27 (2), 159–168.

577 Thevenot, P. T., Saravia, J., Jin, N., Giaimo, J. D., Chustz, R. E., Mahne, S., Kelley, M. A., Hebert,  
578 V. Y., Dellinger, B., Dugas, T. R., Demayo, F. G., Cormier, S. A., 2013. Radical-containing  
579 ultrafine particulate matter initiates epithelial-to-mesenchymal transitions in airway epithelial  
580 cells. *Am. J. Respir. Cell. Mol. Biol.* 48 (2), 188–197.

581 Tian, H., Wang, Y., Xue, Z., Cheng, K., Qu, Y., Chai, F., Hao, J., 2010. Trend and characteristics of  
582 atmospheric emissions of Hg, As, and Se from coal combustion in China, 1980–2007. *Atmos.*  
583 *Chem. Phys.* 10 (23), 11905–11919.

584 Tong, H., Lakey, P. S. J., Arangio, A. M., Socorro, J., Shen, F., Lucas, K., Brune, W. H., Pöschl, U.,  
585 Shiraiwa, M., 2018. Reactive oxygen species formed by secondary organic aerosols in water  
586 and surrogate lung fluid. *Environ. Sci. Technol.* 52 (20), 11642–11651.

587 Trapp, J. M., Millero, F. J., Prospero, J. M., 2010. Temporal variability of the elemental  
588 composition of African dust measured in trade wind aerosols at Barbados and Miami. *Mar.*  
589 *Chem.* 120 (1-4), 71–82.

590 Truong, H., Lomnicki, S., Dellinger, B., 2010. Potential for misidentification of environmentally  
591 persistent free radicals as molecular pollutants in particulate matter. *Environ. Sci. Technol.* 44  
592 (6), 1933–1939.

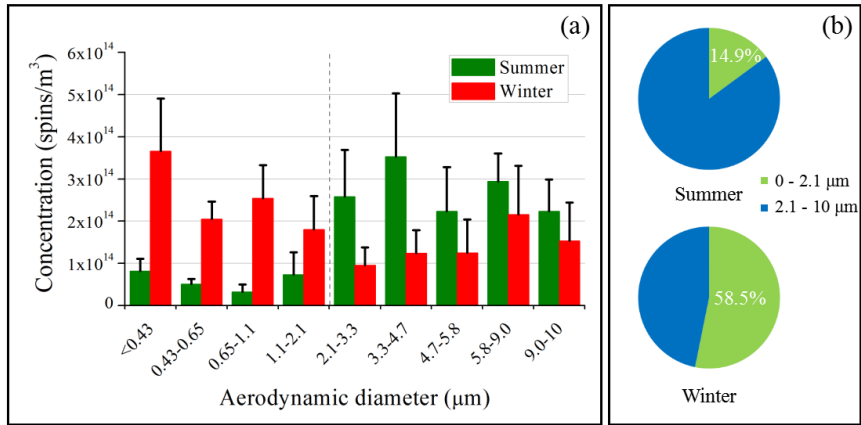
593 Valavanidis, A., Fiotakis, K., Vlachogianni, T., 2008. Airborne particulate matter and human health:  
594 toxicological assessment and importance of size and composition of particles for oxidative  
595 damage and carcinogenic mechanisms. *J. Environ. Sci. Heal. C.* 26 (4), 339–362.

596 Vejerano, E. P., Rao, G., Khachatryan, L., Cormier, S. A., Lomnicki, S., 2018. Environmentally  
597 persistent free radicals: Insights on a new class of pollutants. *Environ. Sci. Technol.* 52 (5),  
598 2468–2481.

599 Vejerano, E., Lomnicki, S. M., Dellinger, B., 2012a. Formation and stabilization of  
600 combustion-generated, environmentally persistent radicals on Ni(II)O supported on a silica  
601 surface. *Environ. Sci. Technol.* 46 (17), 9406–9411.

602 Vejerano, E., Lomnicki, S., Dellinger, B., 2011. Formation and stabilization of  
603 combustion-generated environmentally persistent free radicals on an Fe(III)2O3/silica surface.  
604 *Environ. Sci. Technol.* 45 (2), 589–594.

- 605 Valavanidis, A., Haralambous, E., 2001. A comparative study by electron paramagnetic resonance  
606 of free radical species in the mainstream and sidestream smoke of cigarettes with  
607 conventional acetate filters and 'bio-filters. *Redox. Rep.* 6, 161–171.
- 608 Vejerano, E., Lomnicki, S., Dellinger, B., 2010. Formation and stabilization of  
609 combustion-generated environmentally persistent free radicals on an Fe(III) 2O<sub>3</sub>/silica  
610 surface. *Environ. Sci. Technol.* 45 (2), 589–594.
- 611 Vejerano, E., Lomnicki, S., Dellinger, B., 2012b. Lifetime of combustion-generated  
612 environmentally persistent free radicals on Zn(II)O and other transition metal oxides. *J.*  
613 *Environ. Monit.* 14 (10), 2803–2806.
- 614 Wang, P., Pan, B., Li, H., Huang, Y., Dong, X., Fang, A., Liu, L., Wu, Min., Xing, B., 2018. The  
615 overlooked occurrence of environmentally persistent free radicals in an area with low-rank  
616 coal burning, Xuanwei, China. *Environ. Sci. Technol.* 52 (3), 1054–1061.
- 617 Wang, Y., Li, S., Wang, M., Sun, H., Mu, Z., Zhang, L., Li, Y., Chen, Q., 2019. Source  
618 apportionment of environmentally persistent free radicals (EPFRs) in PM<sub>2.5</sub> over Xi'an,  
619 China. *Sci. Total. Environ.* 689, 193–202.
- 620 Yang, L., Liu, G., Zheng, M., Jin, R., Zhu, Q., Zhao, Y., Wu, X., Yang, X., 2017. Highly elevated  
621 levels and particle-size distributions of environmentally persistent free radicals in  
622 haze-associated atmosphere. *Environ. Sci. Technol.* 51 (14), 7936–7944.
- 623 Yu, T., Wang, J., Shen, M., Li, W., 2013. NH<sub>3</sub>-SCR over Cu/SAPO-34 catalysts with various acid  
624 contents and low Cu loading. *Catal. Sci. Technol.* 3 (12), 3234–3241.

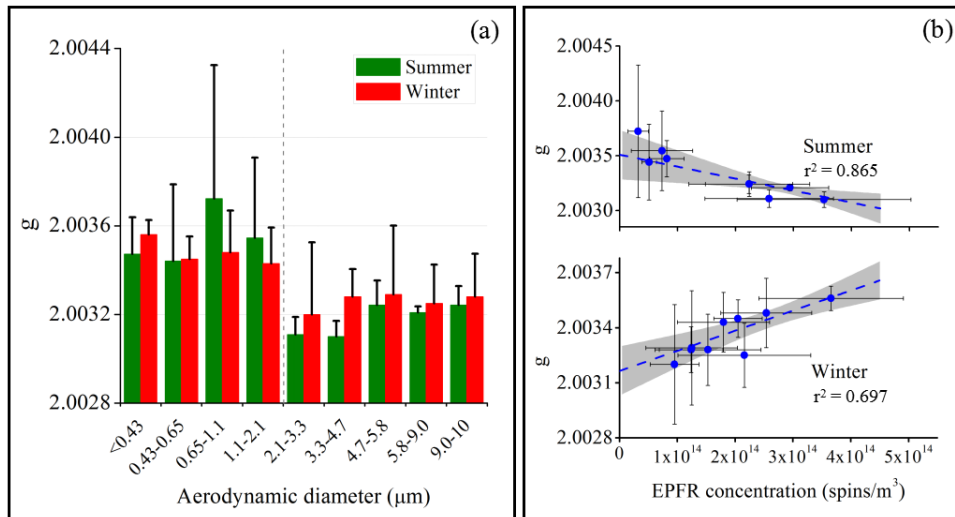


625

626 Figure 1. The concentration of EPFRs in PM with different particle sizes. (a) Atmospheric

627 concentrations of EPFRs in different particle sizes in summer and winter. (b) The relative

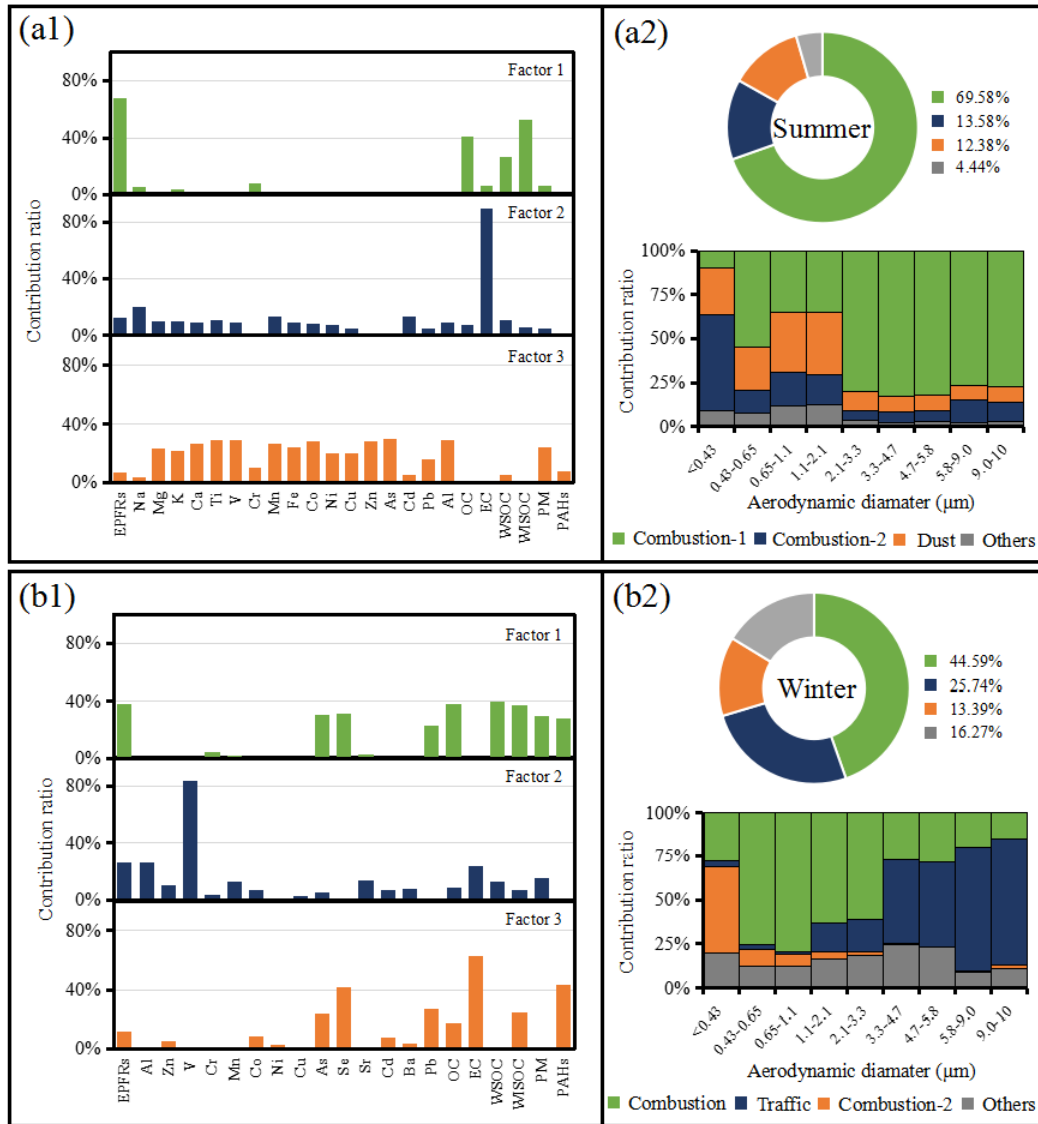
628 contribution of fine particles and coarse particles to the total EPFR concentration.



629

630 Figure 2. A  $g$ -factor comparison. (a) Comparison of  $g$ -factors of EPFRs in different particle sizes  
 631 in different seasons. (b) Correlation analysis of  $g$ -factors and concentrations of EPFRs in summer  
 632 and winter PM. The gray areas in the figure represent 95% confidence intervals.





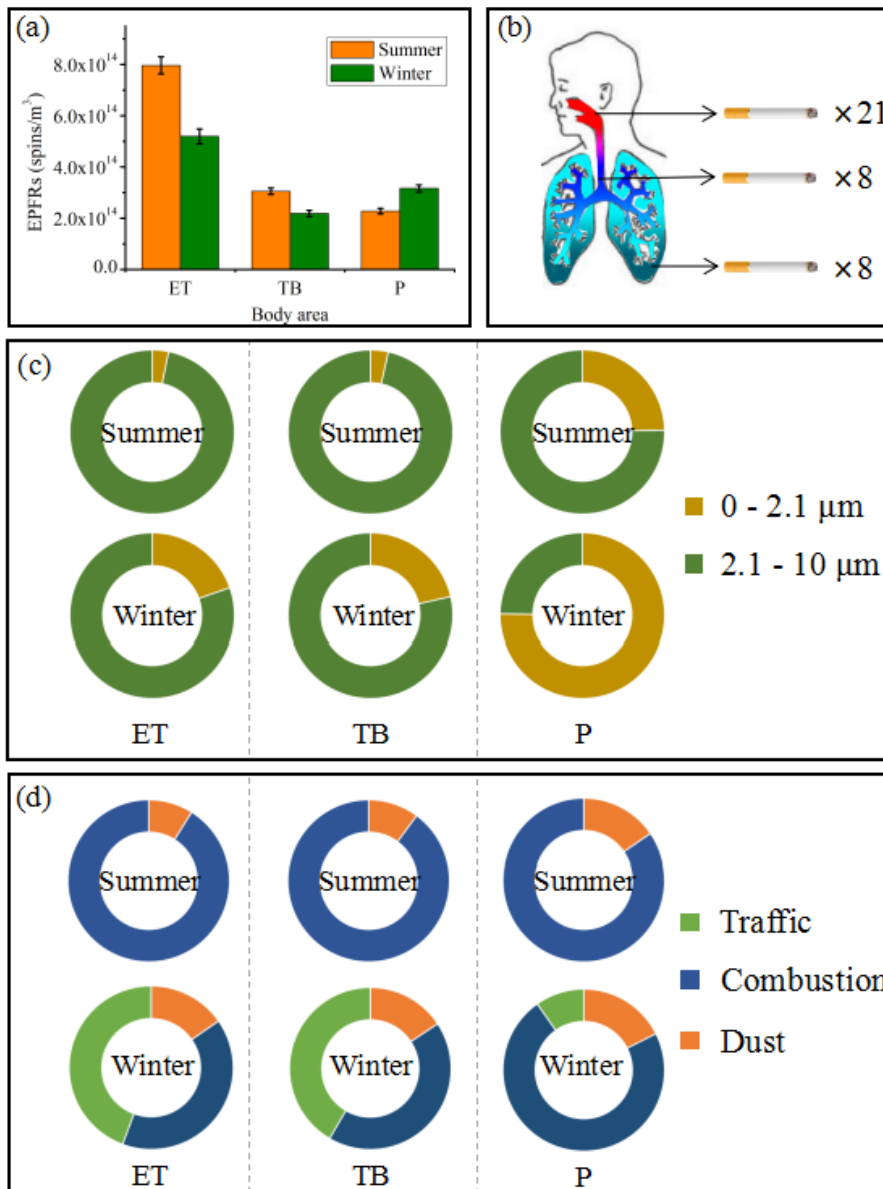
633

634 Figure 3. Factor analysis of EPFRs in different particle sizes in different seasons. (a1) and (b1)

635 represent the results of factor analysis for summer and winter, respectively. (a2) and (b2) represent

636 the contribution of various factors in summer and winter, respectively, to EPFRs and the relative

637 contributions of each factor for different particle sizes.



638

639 Figure 4. Exposure risks to EPFRs. (a) EPFR exposure in the ET, TB, and P regions. (b) Cigarette

640 exposure to EPFRs in the human respiratory system. (c) Exposure ratio of EPFRs with different

641 particle sizes in different areas of the respiratory system. (d) Contribution of EPFRs from different

642 sources to different areas of the respiratory system.