

## **Response to the comments by the reviewers**

**Qingcai Chen, Haoyao Sun and Yanlin Zhang.**

We appreciate the comments from two reviewers. We have answered the comments of the two reviewers, and addressed the problems raised by the reviewers, such as inaccurate descriptions of the details of this article and logical problems. These improvements have a very positive effect on this article.

Our responses to all the comments from the reviewers and changes made in the paper are listed below.

### ***Reviewer #1:***

*EPFRs are widely present in atmospheric particulates, but there is a limited understanding of the size-resolved health risks of these radicals. Here, they reported the risks and sources of EPFRs for different particles in summer and winter. They found different types of sources of EPFRs in particles with different sizes. The experimental design was good, and evidence was solid. The results were useful for scientific community, so this paper can be published after the authors address the following comments.*

We appreciate the positive evaluation of this work.

### ***Specific comments:***

***(1) It is somehow surprising for me that biomass burning was a major emission source in summer. What was the major types of BB? Open burning?***

We appreciate this comment from the reviewer. The results from factor analysis shown that EPFRs mainly from the combustion sources both in winter and summer. However, the results also showed that the dominant factors were different in winter and summer. Obviously the winter is coal burning and thus the summer should be other combustion sources instead of burning coal. According to the production structure of Linfen area, wheat is the main local agricultural crop, and there is often a problem of burning wheat straw in summer, so we speculate that biomass burning may be an important source of EPFRs. We note that this result is speculative. We have modified this part as follows:

P2L22-24: “In both seasons, combustion sources are the main sources of EPFRs with coal combustion as the major contributor in winter, while other fuel combustions are the major source in summer.”

P11L258-260: “Factor 2 is different from factor 1; factor 2 is more likely the combustion of fossil fuels, while factor 1 should be other combustion sources instead of burning coal, such as biomass combustion.”

***(2) Sec. 2.2 and 2.4 more details should be given.***

We appreciate this comment from the reviewer. We have added more details about the analysis of EPFRs and PAHs in the sections of 2.2 and 2.4 as follows:

#### L103-111 (Sec. 2.2): *EPFR analysis*

The EPR spectrometer (MS5000, Freiberg, Germany) is used to detect EPFRs in atmospheric samples. Cut the sample filter into thin strips (5 mm × 28 mm), and put it into the sample tank of the quartz tissue cell (the size of the sample tank is 10 mm × 30 mm)., Then the quartz tissue cell with attached filter sample was placed in a resonant cavity and analyzed by an EPR spectrometer. The detection parameters were magnetic field strength, 335 - 342 mT; detection time, 60 s; modulation amplitude, 0.20 mT; number of detections, 1; and microwave intensity, 8.0 mW. Specific testing protocols have been described previously (Chen et al., 2018c).

#### L125-140 (Sec. 2.3): *PAH analysis*

PAHs were detected using gas chromatography/mass spectrometry (GC/MS) on a GC7890B/MS5977A (Agilent Technologies, Clara, CA). Quartz-fiber filter samples (8 mm in diameter) were cut from each 25-mm quartz-fiber filter substrates used on the ELPI impactor stages using a stainless-steel round punch over a clean glass dish and loaded into the TD glass tube. Next, the TD glass tube was heated to 310 °C at a rate of 12 °C/min and thermally desorbed at 310 °C for 3 min. The desorbed organic compounds were trapped on the head of a GC-column (DB-5MS: 5% diphenyl-95% dimethyl siloxane copolymer stationary phase, 0.25-mm i.d., 30-m length, and 0.25-mm thickness). Sixteen target PAHs were identified based on retention time and qualified ions of the standards, including 16 EPA parent PAHs (p-PAHs). The method detection limits (MDLs) ranged from 0.2 pg/mm<sup>2</sup> (Ace) to 0.6 pg/mm<sup>2</sup> (Incdp). Naphthalene-D8, acenaphthene-D10, phenanthrene-D10, chrysene-D12, and perylene D12 were used for the analytical recovery check. All compounds were recovered with a desorption recovery percentage of > 90%. Specific testing protocols have been described previously (Han et al., 2018).

#### (3) *Line 147: not to use active tense (we or I).*

We appreciate this comment from the reviewer. We have modified this part in the whole text as follows:

L161-162: “To assess the health risks of EPFRs, this study divided the ...”

L170-171: “In addition, the daily inhaled concentration... were converted.”

L208-210: “In another study, the results have shown ...”

L317-319: “To evaluate the health risks..., this study evaluated the ...”

L397-399: “Through this study, the results have shown....”

L401-403: “It is found that the upper respiratory tract ...”

#### (4) *Lines 181-182: evidence should be given to prove it was coal combustion.*

We appreciate this comment from the reviewer. We have modified this part as follows:

L192-196: In addition, the concentration of EPFRs in particulates  $<0.43 \mu\text{m}$  in winter is very high, but it is very low in summer. According to the results of factor analysis in part 3.2 of this study, this particulate matter is related to combustion, which indicates that coal combustion in winter may provide an important contribution to EPFRs.

**(5) Line 194: *it is not necessary dust, and biogenic aerosols can contribute to large particles.***

We appreciate this comment from the reviewer. As the reviewer said, biogenic aerosols may be an important contribution to coarse particulate matter, but there is no research on EPFRs in bioaerosols. Previous studies have demonstrated that dust is an important source of EPFRs in atmospheric particulates in these western Regions of China (Chen et al., 2018b and 2019b). So, this study highlights the sources of dust. Thus, as suggested by the reviewer we also add a possible source of bioaerosols as follows:

L207-208: For example, coarse particles are often associated with dust sources and biogenic aerosols.

**(6) Line 207: *more details should be given to explain g-factor.***

We appreciate this comment from the reviewer. We have modified this part as follows:

L220-226: The *g*-factor obtained by using EPR to detect the sample is an important parameter to distinguish the type of EPFR. It is the ratio of the electronic magnetic moment to its angular momentum (Shaltout et al., 2015; Arangio et al., 2016). The *g*-factor of carbon-centered persistent free radicals is generally less than 2.003, the *g*-factor of oxygen-centered persistent radicals is generally greater than 2.004, and the *g* factor of carbon-centered radicals with adjacent oxygen atoms is between 2.003 and 2.004 (Cruz et al., 2012).

**(7) *I do not suggest using “first” throughout the text.***

We appreciate this comment from the reviewer. As the reviewer said, it is not rigorous enough to use “first” in the full text. We have modified the relevant description in this article.

L17 and L139: “first” has been removed.

L330: “EPFRs were first found in...” has been replaced by “EPFRs were found early in...”

***(8) Line 380-389: what can be the implication for such seasonality? what is the driven factor ?***

We appreciate this comment from the reviewer. This seasonal characteristic is mainly affected by the source characteristics of EPFRs. For example, in winter, EPFRs are mainly found in fine particles. These EPFRs are not only easier to enter the human body, but also due to the smaller  $g$  factor and lower oxidation degree, which means that their reactivity is higher and the harm to the human body is greater. We have added more details about the driven factor and the implication of seasonal characteristic of EPFRs as follows:

L381-383: This seasonal characteristic of EPFRs is mainly affected by the PM sources, this result also indicates that the potential toxicity caused by EPFRs may also vary with particle size and season.

**Reviewer #2:**

*This paper reports measurements of environmentally persistent free radicals (EPFRs) in particulate matter sampled in Linfen, China. The measurements took place in 2 seasons and involved size-resolved samples. The work makes a contribution to our understanding of this unique group of health actors, so should be published pending the authors addressing some general and specifics comments.*

We appreciate the positive evaluation of this reviewer.

**General Comments:**

*There are sections of the Result and Discussion that are repetitious and could be better organized and made more concise. I will point those out in the specific comments, and I strongly recommend the authors go through the paper with an eye towards making it more clear. The authors use the term “formation mechanism” throughout the paper, but they present nothing that resembles the chemistry that would constitute a formation mechanism. I think the authors need to find a better term that describes what they mean, or show actual chemical mechanisms.*

We appreciate this comment from the reviewer. We have deleted the results and simplified the conclusion part of the article.

**Section 4:**

“This study systematically reported the particle size distribution of EPFRs in atmospheric PM in Linfen, which is one of the most polluted cities in China and is located in a typical coal-burning area. In addition, this study evaluated the comprehensive health risks of EPFRs, and reported possible sources and formation process of atmospheric EPFRs with respect to different particle sizes. The following main conclusions were obtained.

(1) This study found that EPFRs are widely present in atmospheric particles of different particle sizes and exhibit significant particle size distribution characteristics. The results of this study demonstrate that the concentrations and types of EPFRs are dependent on particle size and season. This seasonal characteristic of EPFRs is mainly affected by the PM sources, this result also indicates that the potential toxicity caused by EPFRs may also vary with particle size and season.

(2) This study reported the possible source and formation process of atmospheric EPFRs in different particle sizes. The results show that combustion is the most important source of EPFRs (>70%) in both winter and summer PM samples in Linfen. The graphite oxide-like process has the highest contribution (~70%) and is mainly distributed in particles with a size of > 0.43  $\mu\text{m}$ . These findings deepen our understanding of the pollution characteristics of atmospheric EPFRs and are useful for controlling EPFR generation in heavily polluted areas.

(3) This study assessed the exposure risk of EPFRs in different areas of the respiratory system. The results show that the upper respiratory tract is the area with

the highest EPFR exposure. The trachea and alveoli are also exposed to EPFRs, and the risk of exposure is equivalent to that of 8 cigarettes per person per day. Coarse particles are the main source of EPFRs in the upper respiratory tract, while fine particles are mainly involved in the alveoli.

Through this study, the results have shown that there are significant differences in the concentrations and types of EPFRs in particles of different sizes and these differences are due to the influence of the source and generation process. In the future, assessments of the particle size distribution and the seasonality of EPFRs in atmospheric PM should be considered. Health risks are another focus of this study. It is found that the upper respiratory tract is the key exposure area of EPFRs, and the traffic source is the main source of EPFRs in this area. This finding is significant for a systematic assessment of the health risks of EPFRs. In view of the complexity and diversity of the formation process of EPFRs in actual atmospheric particulates, the relative contributions of EPFRs generated by different process and their associated health risks should be more comprehensively studied in the future.”

In addition, the term generation mechanism is used in many places in this article, and we have changed it to the generation process (includes Key words, L351, L345, etc.). In addition, based on the reviewers' specific comments, we have rewritten some logically problematic parts as follows.

***Specific comments:***

***(1) Line 17: I am always skeptical when people claim 'firsts'. In addition this is phrased in the past tense. Why not just say "This study reports. . .?"***

We appreciate this comment from the reviewer. As the reviewer said, it is not rigorous enough to use “first” in the full text. We have modified this part as follows:

L17-19: This study reports the exposure risks and source of EPFRs in atmospheric particulate matter (PM) of different particle sizes ( $<10 \mu\text{m}$ ) in Linfen, a typical coal-burning city in China.

***(2) Lines 101-102: How long were the samples refrigerated before analysis?***

We appreciate this comment from the reviewer. The samples used in this study have a low temperature storage time of 1 year before testing. Our previous research has shown that the proportion of long-life EPFRs (with a lifetime of 3-5 years) in atmospheric samples exceeds 80% (Chen et al., 2019). In addition, we compared the EPR spectra of the same sample before and after storage for 1 year. The results showed that EPFRs did not change significantly. Therefore, long-term storage will not affect the conclusions of this study.

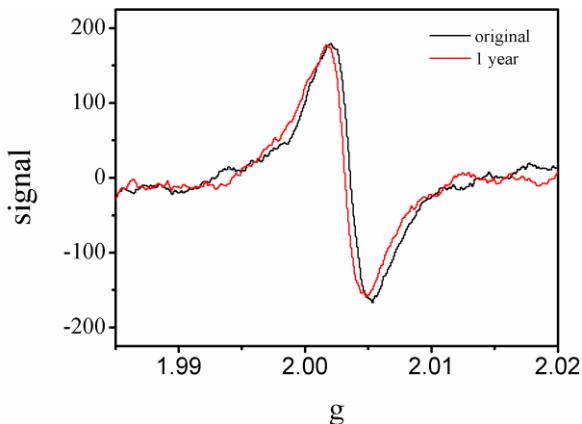


Figure S1 The average EPR spectrum of samples stored before and after 1 year ago. N=9. Sample date: January 25-27, April 20-22, July 11-13 2017. Original refers to the sample used in this article, 1 year means that these samples are stored for 1 year (Chen et al., 2019).

- Chen, Q., Sun, H., Mu, Z., Wang, Y., Li, Y., Zhang, L., Wang, M., Zhang, Z., 2019. Characteristics of environmentally persistent free radicals in PM2.5: Concentrations, species and sources in Xi'an, Northwestern China. Environ. Pollut. 247, 18–26.

**(3) Lines 141-142: The phrase “find a solution to the final solution” sounds awkward and should be rephrased.**

We appreciate this comment from the reviewer. We have added the discussion of g-factor changes and EPFRs decay as follows:

L153-156: Use the gradient-based multiplication algorithm to find a solution from multiple random starting values, and then use the first algorithm to find the final solution based on the least squares effective set algorithm.

**(4) Lines 190-192: The phrasing here is unclear. I think the authors mean the size-segregated contribution of EPFR concentration to the overall. Is this contribution by mass, it's not clear?**

We appreciate this comment from the reviewer. This sentence refers to the contribution of EPFRs in coarse and fine particles to the total concentration of EPFRs. We have changed this part as follows:

L204-206: Figure 1b shows the size-segregated contribution of EPFR concentration to the overall. The contribution of fine PM in summer is only 14.9%, while in winter is 58.5%.

**(5) Line 195: What kind of EPFRs are found in dust particles? Metals?**

We appreciate this comment from the reviewer. Such EPFRs are supposed

mainly as a type of metals. According to our previous research results, EPFRs in sand and dust have no correlation with EC, which may be due to the fact that dust and gravel contain many magnetic materials such as  $\text{Cu}^{2+}$ ,  $\text{Mn}^{2+}$  and  $\text{Zn}^{2+}$ . They are not only paramagnetic, but may also react with some organic matter to form EPFRs and attach to atmospheric particles. We have added the kind of EPFRs in the test as follows:

L209: "...particles contain large amounts of metallic EPFRs and that..."

- Chen, Q., Wang, M., Sun, H., Wang, X., Wang, Y., Li, Y., Zhang, L., Mu, Z., 2018b. Enhanced health risks from exposure to environmentally persistent free radicals and the oxidative stress of PM2.5 from asian dust storms in erenhot, Zhangbei and Jinan, China. Environ. Int. 123, 260–268.

*(6) Lines 207-233: This paragraph was hard to follow, I think because the authors skipped around from sentence to sentence in their discussion of g-factor, concentration, size fraction and season. Sometimes a sentence would be referring to the previous sentence, but in a way that was hard to follow. I would like to see this section rearranged so that it has a more logical and clear flow. Pick one feature at a time and make sure it is clear in each sentence what is being referred to.*

We appreciate this comment from the reviewer. There are some problems with the logic of this paragraph, we have rewritten it as follows:

L220-242: The  $g$ -factor obtained by using EPR to detect the sample is an important parameter to distinguish the type of EPFR. It is the ratio of the electronic magnetic moment to its angular momentum (Shaltout et al., 2015; Arangio et al., 2016). The  $g$ -factor of carbon-centered persistent free radicals is generally less than 2.003, the  $g$ -factor of oxygen-centered persistent radicals is generally greater than 2.004, and the  $g$  factor of carbon-centered radicals with adjacent oxygen atoms is between 2.003 and 2.004 (Cruz et al., 2012). Figure 2a shows the  $g$ -factor distribution characteristics of EPFRs in different particle sizes in summer and winter. The  $g$ -factor of fine particles and coarse particles shows different characteristics. The  $g$ -factor of EPFRs in fine particles (particle size  $< 2.1 \mu\text{m}$ ) ranges from 2.0034 to 2.0037, which may be from carbon-centered radicals with adjacent oxygen atoms. However, the  $g$ -factor of EPFRs in coarse particles (particle size  $> 2.1 \mu\text{m}$ ) is significantly less than that of fine particles. The  $g$ -factor ranges from 2.0031 to 2.0033, indicating that EPFRs in coarse particles are more carbon-centered than those in fine particles and are free of heteroatoms. As shown in Figure 2b, the variation in the  $g$ -factor with concentration in different season is different. The  $g$ -factor of summer PM showed a significant decreasing trend with increasing concentration, while the  $g$ -factor of winter PM showed a significant increasing trend with increasing EPFR concentration. Oyana et al. (2017) studied EPFRs in the surface dust of leaves in the Memphis region of the United States and found that the concentration of EPFRs was positively correlated with the  $g$ -factor, and they believed that this was related to the source of EPFRs. This

phenomenon indicates that the sources and toxicity of EPFRs in winter and summer are different.

**(7) Lines 244: Could it be that the POC in these samples is actually from secondary organic aerosol formation?**

We appreciate this comment from the reviewer. The dominant factor of factor 1 is WISOC, which is typical of a primary combustion source. On the one hand, according to the generation characteristics of EPFRs, the dominant component of the aromatic substances EPFRs produced by low-temperature combustion. On the other hand, according to the local pollution characteristics, summer burning mainly comes from the burning of straw and the catering process. Therefore, we believe that factor one may be mainly biomass combustion. According to previous studies, EPFRs generated by the secondary process are usually active, with a life span of only tens of minutes, so it is unlikely that they are secondary aerosols.

- Chen, Q., Sun, H., Wang, M., Wang, Y., Zhang, L., Han, Y., 2019. Environmentally persistent free radical (EPFR) formation by visible-light illumination of the organic matter in atmospheric particles. *Environ. Sci. Technol.*, 53 (17), 10053–10061.

**(8) Lines 254-256: Here the authors are talking about a graphite oxide formation mechanism – this would be greatly improved if they could should the actual chemical reactions – that is what constitutes a mechanism.**

We appreciate this comment from the reviewer. The research on the chemical reaction of the generation mechanism of graphene oxide to EPFRs has been carried out in our previous research. The research on the chemical reaction of the generation mechanism of graphene oxide to EPFRs has been carried out in our previous research. In that study, we conducted high-temperature treatment experiments on actual atmospheric samples and glucose, and performed EPR, OC/EC and FT-IR tests on the processed samples. The experimental results show that the processed sample can generate EPFRs and is rich in benzene ring structure (benzene ring C=C) and heteroatom functional groups. Its EPR spectrum and *g*-factor are similar to graphene oxide.

- Chen, Q., Sun, H., Wang, M., Mu, Z., Wang, Y., Li, Y., Wang, Y., Zhang, L., Zhang, Z., 2018. Dominant fraction of EPFRs from Nonsolvent-Extractable organic matter in fine particulates over Xi'an, China. *Environ. Sci. Technol.* 52 (17), 9646–9655.

**(9) Lines 243-314: These paragraphs have the problems as the discussion of *g*-factors. Everything is mixed together, with sentences that are hard to follow. I suggest really trying to reorganize this so that it is easier to follow.**

We appreciate this comment from the reviewer. We have rewritten this part.

L252-300 “As shown in Figure 3a1, the typical spectral characteristic of summer factor 1 is that it contains a small fraction of EC components and a large amount of OC components, which indicates that combustion may be the source associated with this factor. This factor has the highest loading of OC, especially WISOC; this fraction mainly contains macromolecular organic substances, which are considered to contribute to the main atmospheric particulate EPFRs and to be graphite oxide-like substances (Chen et al., 2017; Chen et al., 2018a). Factor 2 is different from factor 1; factor 2 is more likely the combustion of fossil fuels, while factor 1 should be other combustion sources instead of burning coal, such as biomass combustion. The generation process is similar to a hybrid process, which includes the graphite oxide-like substances produced by incomplete combustion and the EPFRs formed by some metal oxides. The typical characteristic of factor 3 is that the contribution of metal elements is relatively high, while the contributions of EC and OC are very low. Metal elements such as Al, Ti, Mn, and Co are typical crust elements, so this factor may represent dust sources (Pan et al., 2013; Srivastava et al., 2007; Trapp et al., 2010). The generation process of EPFRs. The others are likely derived from the electroplating metallurgy industry (detailed in S1). As shown in Figure 3a2, the contribution ratios of different factors show that the contribution ratios of factor 1 and factor 2 are the highest, and factor 3 has only a small contribution, which indicates that combustion sources, especially incomplete combustion, are the main sources of EPFRs. The particle size distribution characteristics show that factor 1 is mainly distributed in particles larger than 2.1  $\mu\text{m}$ , while factor 2 is mainly distributed in particles smaller than 0.43  $\mu\text{m}$ .

The results of the factor analysis in winter are different from those in summer. As shown in Figure 3b1, the typical spectral characteristic of factor 1 is that it contains a large amount of OC components and As and Se. As and Se are trace elements of coal combustion, as shown in many studies (Pan et al., 2013; Tian et al., 2010), so coal combustion may be the source represented by this factor. From the generation process viewpoint, the factor does not contain EC, but the content of OC is very high. In the particles with a particle size of less than 3.3, which is mainly present in factor 1, the concentration of OC is 16 times that of EC. So it may be mainly a graphite oxide-like substance formed by the agglomeration of gaseous volatile organic compounds (VOCs) generated during combustion. The typical spectral characteristics of factor 2 are due to a large amount of V and some Al, EC and OC. OC and EC are also typical combustion products. V is rich in fossil fuels, especially fuel oil (Karnae et al., 2011). Therefore, traffic is the source represented by this factor. The factor contains crust elements such as Al and Mn, so it is speculated that this factor may also include traffic-related dust. The typical spectral characteristics of factor 3 are similar to those of factor 1, and both contain relatively large amounts of As and Se, with the exception that factor 3 contains a large amount of EC, indicating that it is also mainly derived from incomplete combustion sources. The generation process of factor 3 should be different from factor 1, which may include both the graphite oxide-like

material generated by fuel coking and the EPFRs generated by the metal oxide. The other factors are mainly atmospheric dust and electroplating or metallurgy (see text S1). As shown in Figure 3b2, factor 1 and factor 2 have the highest proportions, and factor 3 also has a small contribution, which indicates that winter is the same as summer, and combustion sources are the main source of EPFRs. The particle size distribution characteristics show that factor 1 is mainly distributed in particles with a size of 0.43 - 3.3  $\mu\text{m}$ , while factor 2 is mainly distributed in particles larger than 3.3  $\mu\text{m}$ ."

**(10) Line 339: I believe this should be Gehling and Dellinger, (2013).**

We appreciate this comment from the reviewer. We have corrected this mistake.

**(11) Lines 402-403: This sentence is backwards, the trachea and alveoli are exposed to EPFRs not the other way around.**

We appreciate this comment from the reviewer. We have corrected this mistake as follows:

L393-394: The trachea and alveoli are also exposed to EPFRs, and the risk of exposure is equivalent to that of 8 cigarettes per person per day.

**(12) It is hard to distinguish the blue and green colors in the (a) panels of Figures 1 and 2. Please choose better colors.**

We appreciate this comment from the reviewer. We have changed the colors in panel (a) of Figures 1 and 2 to green and red as follows:

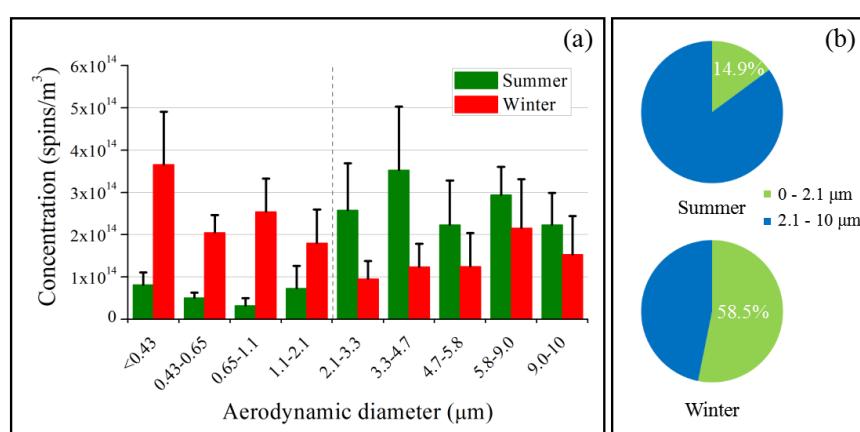


Figure 1. The concentration of EPFRs in PM with different particle sizes. (a) Atmospheric concentrations of EPFRs in different particle sizes in summer and winter. (b) The relative contribution of fine particles and coarse particles to the total EPFR concentration.

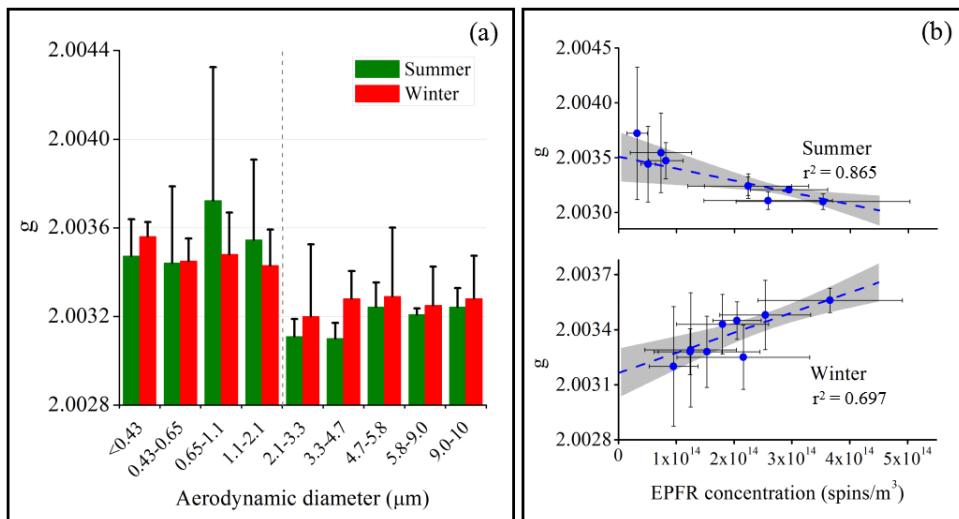
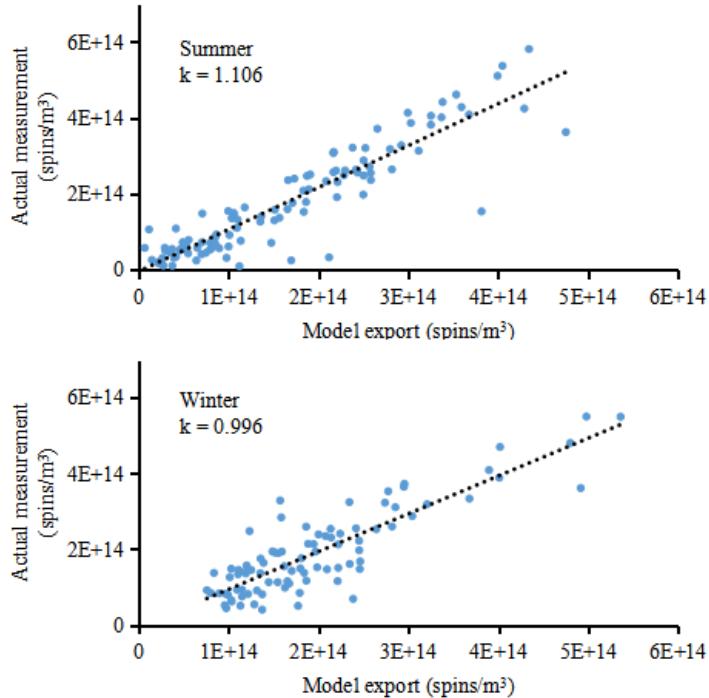


Figure 2. A *g*-factor comparison. (a) Comparison of *g*-factors of EPFRs in different particle sizes in different seasons. (b) Correlation analysis of *g*-factors and concentrations of EPFRs in summer and winter PM. The gray areas in the figure represent 95% confidence intervals.

(13) *Supplement In the first paragraph there is superscript 3 – is this supposed to be a reference? Figure S7 - the caption and axis – ‘model’ should be ‘model’.*

We appreciate this comment from the reviewer. The superscript 3 in the supplementary information is an error and has been deleted. We have modified Figure S7 as follows:



**Figure S7.** Comparison of the concentration of model export and actual measurement.

# 1 Size-resolved exposure risk of persistent free radicals (PFRs)

## 2 in atmospheric aerosols and their potential sources

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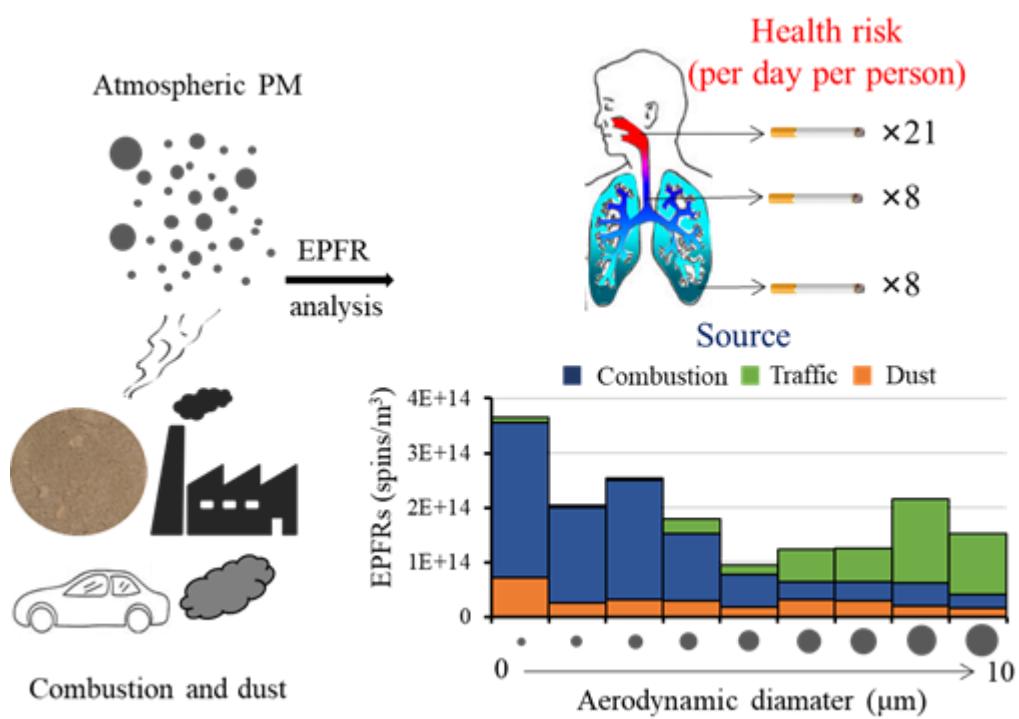
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14   **Abstract:** Environmentally persistent free radicals (EPFRs) are a new type of  
15   substance with potential health risks. EPFRs are widely present in atmospheric  
16   particulates, but there is a limited understanding of the size-resolved health risks of  
17   these radicals. This study ~~first reported~~reports the exposure risks and source of EPFRs  
18   in atmospheric particulate matter (PM) of different particle sizes ( $<10 \mu\text{m}$ ) in Linfen,  
19   a typical coal-burning city in China. The type of EPFRs in fine particles ( $< 2.1 \mu\text{m}$ ) is  
20   different from that in coarse particles ( $2.1\text{-}10 \mu\text{m}$ ) in both winter and summer.  
21   However, the EPFR concentration is higher in coarse particles than in fine particles in  
22   summer, and the opposite trend is found in winter. In both seasons, combustion  
23   sources are the main sources of EPFRs with coal combustion as the major contributor  
24   in winter, while ~~biomass combustion is~~other fuel combustions are the major source in  
25   summer. Dust contributes part of the EPFRs and it is mainly present in coarse  
26   particles in winter and the opposite in summer. The upper respiratory tract was found  
27   to be the area with the highest risk of exposure to EPFRs of the studied aerosols, with  
28   an exposure equivalent to that of approximately 21 cigarettes per person per day.  
29   Alveolar exposure to EPFRs is equivalent to 8 cigarettes per person per day, with  
30   combustion sources contributing the most to EPFRs in the alveoli. This study helps us  
31   to better understand the potential health risks of atmospheric PM with different  
32   particle sizes.

33   **Key words:** EPFRs; particle size distribution; source; generation ~~mechanism~~process

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## 37 1. Introduction

38 Free radicals are atoms or groups containing unpaired electrons, such as hydroxyl  
39 radicals and superoxide radicals, and they usually have strong chemical reactivity and  
40 short lifetimes (Pryor et al., 1986; Finkelstein., 1982). Free radicals with long  
41 lifetimes (months or even years) in the environment are currently called  
42 environmentally persistent free radicals (EPFRs), which have received much attention  
43 in recent years as new environmentally hazardous substances (Vejerano et al., 2018;  
44 Gehling, 2013; Chen et al., 2019c). EPFRs can be used as an active intermediate to  
45 catalyze the production of reactive oxygen species (ROS) by oxygen molecules, thus  
46 endangering human health (D'Arienzo et al., 2017; Thevenot et al., 2013; Harmon et  
47 al., 2018; Blakley et al., 2001; Khachatryan et al., 2011). Studies have found that  
48 EPFRs are present in different environmental media, such as water and soil, and even  
49 in the atmosphere (Dellinger et al., 2001; Truong et al., 2010; Vejerano et al., 2012).

50 A number of studies have investigated the occurrences, sources and formation  
51 mechanismsprocess of EPFRs in atmospheric particulates in different regions. For  
52 example, in the studies of Rostock in Germany, Taif in Saudi Arabia and Xuanwei in  
53 China, the average concentration of EPFRs in atmospheric particulate matter (PM)  
54 was reported to be in the range of  $\sim 10^{16} - 10^{18}$  spins/g (Wang et al., 2018; Arangio et  
55 al., 2016; Shaltout et al., 2015). Atmospheric EPFRs are mainly carbon-centered  
56 radicals with adjacent oxygen atoms (Gehling et al., 2013). EPFRs of different  
57 lifetimes are present in atmospheric PM, with only a few hours for short-lifetime  
58 EPFRs and several years for long-lifetime EPFRs that show no signs of decay  
59 (Gehling et al., 2013; Chen et al., 2019c). Most studies indicate that sources of  
60 transportation and combustion may be the primary EPFR sources in atmospheric PM  
61 (Wang et al., 2018; Yang et al., 2017; Chen et al., 2019b). Chen et al. (2018b and  
62 2019b) found that strong atmospheric photochemical effects in summer and dust  
63 particles may also be important sources of EPFRs. The process of electron transfer  
64 and stabilization between the surface of metal oxides (such as iron, copper, zinc and

nickel) and substituted aromatic molecules under high temperatures is considered to be the main [mechanismprocess](#) for the formation of EPFRs in atmospheric particles (Truong., 2010; Vejerano et al., 2012a; Patterson et al., 2013; Vejerano., 2010; Vejerano et al., 2012b). However, the study by Chen et al. (2018a) suggests that EPFRs in atmospheric particulates are mainly derived from graphite oxide-like substances produced during combustion. In addition to primary sources such as combustion, secondary chemical processes in the atmosphere may also be an important source of EPFRs in atmospheric PM (Chen et al. 2019b and 2019d; Tong et al., 2018).

Different particle sizes of atmospheric PM pose different health risks to humans, depending on the deposition efficiency of the particles and the chemical composition and concentrations of hazardous substances they contain (Strak et al., 2012; Valavanidis et al., 2008). Among various hazardous substances, EPFRs may also be involved in the toxicity of atmospheric particulates. Yang et al. (2017) studied the the EPFRs that are extractable by dichloromethane in different particle sizes in Beijing in winter and found that the concentration of EPFRs was the highest in particles with sizes  $< 1 \mu\text{m}$ . Arangio et al. (2016) found that the concentration of EPFRs in 180 nm particles was the highest in the 56 nm - 1.8  $\mu\text{m}$  particle size range. Although several studies have examined the particle size distribution of EPFRs, systematic studies have not been conducted on the formation [mechanismprocess](#), source and exposure assessment of EPFRs in atmospheric particles with different particle sizes.

This study takes Linfen as an example. Linfen is one of the cities in China with the most serious air pollution and is a typical coal-burning city. The particle size distribution of EPFRs in atmospheric PM in this region was studied by EPR spectrometry. The effects of particle size and season on the source, formation [mechanismprocess](#), and health risk of EPFRs were revealed. In particular, the comprehensive health risks of EPFRs were evaluated, and it was found that the upper respiratory tract is the area with the highest risk of EPFRs exposure, which is equivalent to twenty-one cigarettes per person per day. This study is of great

94 significance for understanding the source and formation mechanismprocess of EPFRs  
95 in atmospheric particulates as well as for health risk assessments.

## 96 **2. Experimental section**

### 97 *2.1 Sample collection*

98 The sampling site for this study is located in Hongdong (36°23', 111°40'E) in  
99 Shanxi, China. To collect atmospheric particles of different sizes (0-10  $\mu\text{m}$ ), this study  
100 used a Thermo-Anderson Mark II sampler to collect aerosol samples of 9 sizes. The  
101 samples were collected on a prebaked quartz filter (450 °C, 4.5 hours), and the  
102 sampling dates were as follows: in winter, January 26 to February 4, 2017,  $n = 10$ ; and  
103 in summer, July 31 to August 24, 2017,  $n = 12$ . The samples were placed in a -20 °C  
104 refrigerator prior to analysis.

### 105 *2.2 EPFR analysis*

106 The EPR spectrometer (MS5000, Freiberg, Germany) is used to detect EPFRs in  
107 atmospheric samples. Cut the Specific testing protocols have been described  
108 previously (Chen et al., 2018e. The sample filter was cut into thin strips (5 mm × 28  
109 mm), and elamped with a put it into the sample tank of the quartz piece, and then tissue  
110 cell (the size of the sample tank is 10 mm × 30 mm).. Then the quartz pieee tissue cell  
111 with attached filter sample was placed in a resonant cavity and analyzed by an EPR  
112 spectrometer (MS5000, Freiberg, Germany). The detection parameters were magnetic  
113 field strength, 335 - 342 mT; detection time, 60 s; modulation amplitude, 0.20 mT;  
114 number of detections, 1; and microwave intensity, 8.0 mW. Specific testing protocols  
115 have been described previously (Chen et al., 2018c).

### 116 *2.3 Carbon composition analysis*

117 The contents of organic carbon (OC) and elemental carbon (EC) in the filter  
118 samples were analyzed using a semicontinuous OC/EC analyzer (Model 4, Sunset Lab.  
119 Inc., Oregon, USA) with a NIOSH 5040 detection protocol (Lin et al., 2009).

120 The water-soluble organic carbon (WSOC) concentration was analyzed using an  
121 automatic TOC-LCPH analyzer (Shimadzu, Japan). The WSOC extraction was  
122 performed with ultrapure water under ultrasonication for 15 minutes, and all WSOC  
123 concentrations were blank corrected. The concentration of OC in the MSM  
124 (Methanol-soluble materials) was calculated as the difference between the OC and  
125 WSOC (Water-soluble organic carbon) concentrations. This calculation assumes that  
126 all water-insoluble organic carbon (WISOC) in the aerosol can be extracted with  
127 MeOH, and the rationality of this assumption has been verified elsewhere (Mihara et  
128 al., 2011; Liu et al., 2013; Cheng et al., 2016; Chen et al., 2019a).

129 *2.4 PAH analysis*

130 PAHs were detected using gas chromatography/mass spectrometry (GC/MS) on a  
131 GC7890B/MS5977A (Agilent Technologies, Clara, CA), ~~as described in detail in a~~  
132 ~~previously published study (Han et al.,).~~ Quartz-fiber filter samples (8 mm in diameter)  
133 were cut from each 25-mm quartz-fiber filter substrates used on the ELPI impactor  
134 stages using a stainless-steel round punch over a clean glass dish and loaded into the  
135 TD glass tube. Next, the TD glass tube was heated to 310 °C at a rate of 12 °C/min  
136 and thermally desorbed at 310 °C for 3 min. The desorbed organic compounds were  
137 trapped on the head of a GC-column (DB-5MS: 5% diphenyl-95% dimethyl siloxane  
138 copolymer stationary phase, 0.25-mm i.d., 30-m length, and 0.25-mm thickness).  
139 Sixteen target PAHs were identified based on retention time and qualified ions of the  
140 standards, including 16 EPA parent PAHs (p-PAHs). The method detection limits  
141 (MDLs) ranged from 0.2 pg/mm<sup>2</sup> (Ace) to 0.6 pg/mm<sup>2</sup> (Incip). Naphthalene-D8,  
142 acenaphthene-D10, phenanthrene-D10, chrysene-D12, and perylene D12 were used  
143 for the analytical recovery check. All compounds were recovered with a desorption  
144 recovery percentage of > 90%. Specific testing protocols have been described  
145 previously (Han et al., 2018).

146 *2.5 Metal element analysis*

147 The concentration of metal elements in the samples was determined by a Thermo

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148 X2 series inductively coupled plasma mass spectrometer (ICP-MS, Thermo, USA).  
149 The metal elements analyzed in summer were Na, Mg, K, Ca, Ti, V, Cr, Mn, Fe, Co,  
150 Ni, Cu, Zn, As, Cd, Pb, and Al, and those in winter were Al, Zn, V, Cr, Mn, Co, Ni, Cu,  
151 As, Se, Sr, Cd, Ba, and Pb. The specific measurement method is based on the study of  
152 Qi et al (2016).

153 *2.6. Data statistics method*

154 The source and formation ~~mechanismprocess~~ of EPFRs in PM with different  
155 particle sizes were analyzed by nonnegative matrix factorization (NMF). The method  
156 is based on the study of Chen et al (2016 and 2019e). Briefly, NMF analysis of EPFR  
157 data, metal element contents, OC/EC contents and PAH contents was performed in  
158 MATLAB. The version of the NMF toolbox is 1.4  
159 (<https://sites.google.com/site/nmftool/>). ~~First, a~~Use the gradient-based multiplication  
160 algorithm ~~is used~~ to find a solution from multiple random starting values, and then use  
161 the first algorithm ~~is used~~ to find ~~a solution to~~ the final solution ~~using a~~based on the  
162 least squares effective set algorithm. To find a global solution, the model was run 100  
163 times, each time with a different initial value. By comparing the 1-12 factor model  
164 (Figure S4) with the residual of the spectral load, the 6 factor (summer) and 10 factor  
165 (winter) NMF models were finally selected.

166 *2.7. EPFR exposure evaluation*

167 To assess the health risks of EPFRs, ~~we this study~~ divided the respiratory system  
168 into three parts based on the human breathing model: extrathoracic (ET) areas,  
169 including the anterior nasal cavity, posterior nasal cavity, oral cavity, and throat;  
170 tracheobronchial (TB) areas, including the trachea, bronchi, bronchioles, and terminal  
171 bronchi; and pulmonary (P) areas, including the alveolar ducts and alveoli. Then, the  
172 sedimentation rates of different particle sizes in different areas of the respiratory  
173 system were determined to calculate the exposure risk of EPFRs. Here, the human  
174 respiratory system particulate deposition model of Salma et al. (2002) was used, and  
175 the specific data can be found in Table S3 and S4.

176 In addition, ~~we converted~~ the daily inhaled concentration of EPFRs into the  
177 concentration of free radicals in cigarettes were converted. The specific conversion  
178 method is as follows:

179 
$$N_{cig} = (C_{EPFRs} \cdot V) / (RC_{cig} \cdot C_{tar}) \quad (1)$$

180 where  $N_{cig}$  represents the number of cigarettes (/person/day),  $C_{EPFRs}$  (spins/m<sup>3</sup>)  
181 represents the atmospheric concentration of EPFRs in PM, and  $V$  represents the  
182 amount of air inhaled by an adult per day (20 m<sup>3</sup>/day) (Environmental Protection  
183 Agency, 1988).  $RC_{cig}$  ( $4.75 \times 10^{16}$  spins/g) (Baum et al., 2003; Blakley et al., 2001;  
184 Pryor et al., 1983; Valavanidis and Haralambous, 2001) indicates the concentration of  
185 free radicals in cigarette tar, and  $C_{tar}$  (0.013 g/cig) indicates the amount of tar per  
186 cigarette (Gehling et al., 2013).

187 **3. Results and discussion**

188 *3.1 Concentrations and types of EPFRs*

189 Figure 1a shows the concentration distribution of EPFRs with different particle  
190 sizes in different seasons. EPFRs were detected in the particles of each tested size (the  
191 EPR spectrum is shown in Figure S1), but their EPFR concentration levels were  
192 different. In summer, the concentration of EPFRs in fine particles (particle size < 2.1  
193  $\mu\text{m}$ ) is  $(3.2 - 8.1) \times 10^{13}$  spins/m<sup>3</sup>, while the concentration of EPFRs in coarse  
194 particles (particle size > 2.1  $\mu\text{m}$ ) is 1-2 orders of magnitude higher than that of fine  
195 particles, reaching values of  $(2.2 - 3.5) \times 10^{14}$  spins/m<sup>3</sup>. Winter samples show  
196 completely different characteristics from summer samples. The concentration of  
197 EPFRs in fine particles (particle size < 2.1  $\mu\text{m}$ ) is  $(1.8 - 3.6) \times 10^{14}$  spins/m<sup>3</sup>, while the  
198 concentration of EPFRs in coarse particles (particle size > 2.1  $\mu\text{m}$ ) is smaller than that  
199 of fine particles, with values of  $(1.0 - 2.1) \times 10^{14}$  spins/m<sup>3</sup>. In addition, the  
200 concentration of EPFRs in particulates <0.43  $\mu\text{m}$  in winter is very high, but it is very  
201 low in summer. ThisAccording to the results of factor analysis in part 3.2 of this study,  
202 this particulate matter is related to combustion, which indicates that coal combustion

203 in winter may provide an important contribution to EPFRs. The EPFR concentration  
204 in the fine PM of Linfen reported above is equivalent to that in the fine PM of Xi'an,  
205 but it is ten times smaller than that in the fine PM of Beijing (Yang et al., 2017; Chen  
206 et al., 2019b). Although the particle size distribution characteristics of EPFRs in  
207 winter and summer are different, their concentration levels are similar, which  
208 indicates that the EPFR concentration is not related to the PM concentration, but is  
209 determined by the source characteristics. The source characteristics will be discussed  
210 in detail in the factor analysis section.

211 Figure 1b shows the size-segregated contribution of ~~the~~ EPFR concentration to the  
212 overall ~~EPFR concentration in coarse and fine particles~~. The contribution of fine PM  
213 in summer is only 14.9%, while ~~that of fine PM~~ in winter is 58.5%. The differences in  
214 EPFR concentrations with particle size may be related to the source of EPFRs. For  
215 example, coarse particles are often associated with dust sources ~~and biogenic aerosols~~.  
216 In another study, ~~we~~ the results have shown that dust particles contain large amounts  
217 of metallic EPFRs and that they can be transported over long distances (Chen et al.,  
218 2018b). EPFRs in fine particles may be mainly derived from the combustion process,  
219 such as traffic sources, which are considered to be an important source of EPFRs in  
220 atmospheric PM (Secrest et al., 2016; Chen et al., 2019b). Due to winter heating in the  
221 Linfen area, the amount of coal burning increases sharply in this season. In 2017, the  
222 nonclean heating (Coal-fired heating) rate of urban heating energy structures in Linfen  
223 was 40% (data source: <http://www.linfen.gov.cn/>). With the burning of coal, large  
224 amounts of EPFRs are produced, and in the summer, EPFRs emitted by burning coal  
225 should be much less than those emitted in winter. This can explain to a certain extent  
226 that the contribution of fine particles to summer EPFRs is small, and the contribution  
227 of winter EPFRs is very large.

228 The *g*-factor obtained by using EPR to detect the sample is ~~a~~an important parameter  
229 ~~used~~to distinguish the type of EPFR. It is the ratio of the electronic magnetic moment  
230 to its angular momentum (Shaltout et al., 2015; Arangio et al., 2016). The *g*-factor of  
231 carbon-centered persistent free radicals is generally less than 2.003, the *g*-factor of

oxygen-centered persistent radicals is generally greater than 2.004, and the g factor of carbon-centered radicals with adjacent oxygen atoms is between 2.003 and 2.004 (Cruz et al., 2012). Figure 2a shows the g-factor distribution characteristics of EPFRs in different particle sizes in summer and winter. The g-factor of fine particles and coarse particles ~~also~~ shows different characteristics. The g-factor of EPFRs in fine particles (particle size < 2.1  $\mu\text{m}$ ) ranges from 2.0034 to 2.0037, which may be from carbon-centered radicals with adjacent oxygen atoms. However, the g-factor of EPFRs in coarse particles (particle size > 2.1  $\mu\text{m}$ ) is significantly less than that of fine particles. The g-factor ranges from 2.0031 to 2.0033, indicating that EPFRs in coarse particles are more carbon-centered than those in fine particles and are free of heteroatoms. ~~Although the particle size characteristics of the g-factor of the EPFRs in summer and winter are the same, the~~ As shown in Figure 2b, the variation in the g-factor with concentration in different season is different. ~~As shown in Figure 2b, the~~ The g-factor of summer PM showed a significant decreasing trend with increasing concentration, while the g-factor of winter PM showed a significant increasing trend with increasing EPFR concentration. Oyana et al. (2017) studied EPFRs in the surface dust of leaves in the Memphis region of the United States and found that the concentration of EPFRs was positively correlated with the g-factor, and they believed that this was related to the source of EPFRs. This phenomenon indicates that the sources and toxicity of EPFRs in winter and summer are different. ~~Figure 1 shows that the summer EPFRs are mainly derived from coarse particles, while the g-factor of EPFRs in coarse particles is smaller than that in fine particles, so the g-factor of EPFRs in summer decreases with an increase in EPFR concentration. In winter, fine particles contribute more to EPFRs, so the g-factor of EPFRs in winter increases with the concentration of EPFRs.~~

### 3.2 Factor Analysis of EPFRs

To explore the possible sources and formation mechanismprocess of EPFRs in atmospheric particles with different particle sizes, the NMF model was used to statistically analyze EPFRs, carbon components, PAHs and metal elements in samples.

261 The factors obtained by the NMF model should reflect the different sources  
262 mechanisms and generation process of EPFRs. As shown in Figure 3a1 and b1, the  
263 three main contributing factors to EPFRs in summer and winter are shown (see Figure  
264 S5, S6 for spectra of other factors), which explain 94.5% and 83.8% of the EPFR  
265 concentrations in summer and winter, respectively.

266 As shown in Figure 3a3a1, the typical spectral characteristic of summer factor 1 is  
267 that it contains a small fraction of EC components and a large amount of OC  
268 components, which indicates that combustion may be the source associated with this  
269 factor. This factor has the highest loading of OC, especially WISOC; this fraction  
270 mainly contains macromolecular organic substances, which are considered to  
271 contribute to the main atmospheric particulate EPFRs and to be graphite oxide-like  
272 substances (Chen et al., 2017; Chen et al., 2018a). ~~The result shows that factor 1 has~~  
273 ~~the highest contribution of all the factors to EPFRs in PM (69.6%), and they are~~  
274 ~~mainly distributed in particles with sizes > 2.1  $\mu\text{m}$ . Factor 2 is typically characterized~~  
275 ~~by a high contribution from EC and a small fraction of OC and metal elements, which~~  
276 ~~is a typical source of incomplete combustion.~~ Factor 2 is different from factor 1;  
277 factor 2 is more likely the combustion of fossil fuels, while factor 1 ~~may~~ should be  
278 other combustion sources instead of burning coal, such as biomass combustionsource.  
279 The generation mechanismprocess is similar to a hybrid mechanismprocess, which  
280 includes the graphite oxide-like substances produced by incomplete combustion and  
281 the EPFRs formed by some metal oxides. ~~The relative contribution of these EPFRs is~~  
282 ~~13.5% and is mainly distributed in particles with a size < 0.43  $\mu\text{m}$ .~~ The typical  
283 characteristic of factor 3 is that the contribution of metal elements is relatively high,  
284 while the contributions of EC and OC are very low. Metal elements such as Al, Ti, Mn,  
285 and Co are typical crust elements, so this factor may represent dust sources (Pan et al.,  
286 2013; Srivastava et al., 2007; Trapp et al., 2010). The generation mechanism may be  
287 ~~mainly due to the participation of metal oxides in the generation~~process of EPFRs.  
288 ~~Compared with the other factors, this factor also has a partial load on PAHs,~~  
289 ~~indicating that PAHs may be involved in the formation of metal oxide related EPFRs.~~

290 These EPFRs have a relatively low contribution to total EPFRs (approximately 12.4%)  
291 and are mainly distributed in particles with a size of 0.43–2.1  $\mu\text{m}$ . The EPFR  
292 contribution of other factors is 4.4%; they others are likely derived from the  
293 electroplating metallurgy industry (detailed in S1). As shown in Figure 3a2, the  
294 contribution ratios of different factors show that the contribution ratios of factor 1 and  
295 factor 2 are the highest, and factor 3 has only a small contribution, which indicates  
296 that combustion sources, especially incomplete combustion, are the main sources of  
297 EPFRs. The particle size distribution characteristics show that factor 1 is mainly  
298 distributed in particles larger than 2.1  $\mu\text{m}$ , while factor 2 is mainly distributed in  
299 particles smaller than 0.43  $\mu\text{m}$ .

300 The results of the factor analysis in winter are different from those in summer. As  
301 shown in Figure 3b3b1, the typical spectral characteristic of factor 1 is that it contains  
302 a large amount of OC components and As and Se. As and Se are trace elements of  
303 coal combustion, as shown in many studies (Pan et al., 2013; Tian et al., 2010), so  
304 coal combustion may be the source represented by this factor. From the generation  
305 mechanism process viewpoint, the factor does not contain EC, but the content of OC is  
306 very high. In the particles with a particle size of less than 3.3, which is mainly present  
307 in factor 1, the concentration of OC is 16 times that of EC. So it may be mainly a  
308 graphite oxide-like substance formed by the agglomeration of gaseous volatile organic  
309 compounds (VOCs) generated during combustion. These EPFRs are mainly  
310 distributed in particles with a size of 0.43–3.3  $\mu\text{m}$ , and their contribution to EPFRs in  
311 PM is up to 44.6%. Factor 2 contributes 25.7% to EPFRs. The typical spectral  
312 characteristics of factor 2 are due to a large amount of V and some Al, EC and OC.  
313 OC and EC are also typical combustion products. V is rich in fossil fuels, especially  
314 fuel oil (Karna et al., 2011). Therefore, traffic is the source represented by this factor.  
315 The factor contains crust elements such as Al and Mn, so it is speculated that this  
316 factor may also include traffic-related dust. The particle size distribution shows that  
317 such EPFRs are mainly present in large particles with a size of 3.3–10  $\mu\text{m}$ . The  
318 typical spectral characteristics of factor 3 are similar to those of factor 1, and both

319 contain relatively large amounts of As and Se, with the exception that factor 3  
320 contains a large amount of EC, indicating that it is also mainly derived from  
321 incomplete combustion sources. The generation mechanismprocess of factor 3 should  
322 be different from factor 1, which may include both the graphite oxide-like material  
323 generated by fuel coking and the EPFRs generated by the metal oxide. These EPFRs  
324 are mainly distributed in particles with a size of <0.43  $\mu\text{m}$ , and their total contribution  
325 to EPFRs in PM is 13.4%. In addition, the The other factors contribute 16.3% to  
326 EPFRs, and these factors are mainly atmospheric dust (11.4%) and electroplating or  
327 metallurgy (4.9%) (see text S1). The results of this studyAs shown in Figure 3b2,  
328 factor analysis were similar to the results of the study by Wang et al. (2019) on EPFRs  
329 in Xi'an. They found1 and factor 2 have the highest proportions, and factor 3 also has  
330 a small contribution, which indicates that coal, trafficwinter is the same as summer,  
331 and dust were combustion sources are the main sources of EPFRs and accounted for  
332 76.2% of the total source of EPFRs. The particle size distribution characteristics show  
333 that factor 1 is mainly distributed in particles with a size of 0.43 - 3.3  $\mu\text{m}$ , while factor  
334 2 is mainly distributed in particles lager than 3.3  $\mu\text{m}$ .

335 Based on the above analysis, it can be found that combustion sources are the main  
336 sources of EPFRs, and EPFRs from these sources are mainly graphite oxide-like  
337 substances generated by the polymerization of organic matter or fuel coking. Studies  
338 have shown that graphene oxide can cause cell damage by generating ROS (Seabra et  
339 al., 2014). The surface of these compounds contains not only carbon atoms but also  
340 some heteroatoms, which leads to disorder and the presence of defects in the  
341 carbon-based structure (Lyu et al., 2018; Chen et al., 2017a; Mukome et al., 2013;  
342 Keiluweit et al., 2010). The dust source is also a source of important EPFRs identified  
343 in this study (with a contribution of approximately 10%). It was shown in the above  
344 analysis that the concentration of EPFRs in coarse particles has a significant  
345 correlation with the concentration of metallic elements, particularly crustal elements.  
346 Some crustal elements, such as Al, and Fe, not only have their own paramagnetism  
347 (Li et al., 2017; Yu et al., 2013; Nikitenko et al., 1992), but also interact with aromatic

348 compounds attached to the surface of the particles to produce a stable single-electron  
349 structure.

350 *3.3 Health risk of EPFRs*

351 To evaluate the health risks of EPFRs in PM with different particle sizes, [we this](#)  
352 [study](#) evaluated the comprehensive exposure of EPFRs based on the deposition  
353 efficiency of PM with different particle sizes in different parts of the human body. The  
354 results are shown in Figure 4a. The ET region is the region with the highest EPFR  
355 exposure, while the TB and P regions have relatively close EPFRs. This result shows  
356 that atmospheric EPFRs are the most harmful to the health of the human upper  
357 respiratory tract. Comparing the EPFR exposure in different seasons indicates that the  
358 exposure risk in the ET area in summer is significantly higher than that in winter. This  
359 difference occurs because the concentration of EPFRs in coarse particles is much  
360 higher than that of fine particles in summer and the deposition efficiency of large  
361 particles in the ET area is generally higher. Fine particles are more efficiently  
362 deposited in the P region, leading to a higher risk of EPFR exposure in the P region in  
363 winter.

364 EPFRs were [first](#) found [early](#) in cigarette tar and are considered one of the health  
365 risk factors in cigarette smoke (Lyons et al., 1960); thus, in this study, the exposure  
366 risks of EPFRs in particles deposited in the human body were converted to the  
367 equivalent number of cigarettes inhaled per adult per day. As shown in Figure 4b, the  
368 ET area is the most contaminated area, with an average equivalence of twenty-one  
369 cigarettes (twenty-five in summer and sixteen in winter). The average values for the  
370 TB area (nine in summer and seven in winter) and P area (seven in summer and ten in  
371 winter) are eight. The results indicate that EPFRs pose significant health risks to  
372 human lungs in both winter and summer. Other similar studies, such as a study of the  
373 average amount of EPFRs in  $PM_{2.5}$  inhaled per person per day in Xi'an in 2017, found  
374 values equivalent to approximately 5 cigarettes (Chen et al., 2018a). Gehring [et al.](#) and  
375 [Dellinger](#) (2013) found that EPFR exposure in  $PM_{2.5}$  is equivalent to approximately  
376 0.3 cigarettes per person per day in St. Joaquin County, the location with the worst air

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377 pollution in the United States. The average exposure risk of EPFRs in fine particles in  
378 the Linfen area (approximately 13 cigarettes) was higher than those in these two  
379 studies. However, these previous studies only studied the exposure risk of EPFRs in  
380 fine particles. The results of this study indicate that the health risks of EPFRs are  
381 significantly increased when the particle size distribution of EPFRs is taken into  
382 account. Therefore, it is important to study the source characteristics and generation  
383 [mechanismprocess](#) of EPFRs with different particle sizes, which will be discussed in  
384 detail in the following sections.

385 This study calculated the proportion of EPFRs with different particle sizes in  
386 different parts of the respiratory system based on the deposition efficiency of particles  
387 with different particle sizes. As shown in Figure 4c, in the ET region and the TB  
388 region, coarse particles are the dominant component in summer and winter. In  
389 particular, in summer, the proportion of EPFRs in coarse particles in these two regions  
390 exceeds 95%. In the P region, there are significant differences between summer and  
391 winter. The P region in summer is still dominated by coarse particles, but its  
392 proportion is significantly lower than those in the ET and TB regions. In the P region  
393 in winter, fine particles are the dominant component (approximately 70%). These  
394 distribution characteristics indicate different sources of EPFRs in different regions. As  
395 shown in Figure 4d, in summer, combustion sources are the main source of EPFRs in  
396 the respiratory system. In winter, combustion and transportation sources contribute  
397 equally in the EP and ET regions, while in the alveoli, combustion sources are the  
398 main contributor. The ET region is the area with the highest risk of exposure to  
399 EPFRs (21 cigarettes). The generation [mechanismprocess](#) of these EPFRs is mainly  
400 attributable to graphene oxide-like substances. Studies have shown that graphene  
401 oxide is cytotoxic (Harmon et al., 2018). In the alveoli, the contribution of  
402 combustion sources is significantly increased (especially in winter). These EPFRs are  
403 mainly generated by the action of metal oxides and organic substances. Studies have  
404 shown that such EPFRs can generate ROS in the lung fluid environment (Khachatrian  
405 et al., 2011). [Therefore, the health risks of EPFRs from different sources and](#)

406 mechanisms should be evaluated in the future in order to better assess the harm  
407 caused by EPFRs to the body.

408 **4. Conclusions and environmental implications**

409 This study systematically reported the particle size distribution of EPFRs in  
410 atmospheric PM in Linfen, which is one of the most polluted cities in China and is  
411 located in a typical coal-burning area. In addition, this study evaluated the  
412 comprehensive health risks of EPFRs, and reported possible sources and formation  
413 mechanismsprocess of atmospheric EPFRs with respect to different particle sizes. The  
414 following main conclusions were obtained.

415 (1) This study found that EPFRs are widely present in atmospheric particles of  
416 different particle sizes and exhibit significant particle size distribution characteristics.  
417 ~~EPFR concentrations are higher in coarse particles than in fine particles in summer~~  
418 ~~and vice versa in winter. Differences were also found in the g factors of EPFRs in~~  
419 ~~coarse particles and fine particles, indicating that the types of EPFRs of different~~  
420 ~~particle sizes were also different.~~ The results of this study demonstrate that the  
421 concentrations and types of EPFRs are dependent on particle size and season. ~~This~~  
422 This result~~This seasonal characteristic of EPFRs is mainly affected by the PM sources, this~~  
423 result also indicates that the potential toxicity caused by EPFRs may also vary with  
424 particle size and season.

425 (2) This study reported the possible source and formation mechanismsprocess of  
426 atmospheric EPFRs in different particle sizes. The results show that combustion is the  
427 most important source of EPFRs (>70%) in both winter and summer PM samples in  
428 Linfen. ~~Atmospheric dust also contributes to EPFRs (~10%), and they are mainly~~  
429 ~~found in fine particles in summer and coarse particles in winter.~~ The graphite  
430 oxide-like mechanismprocess has the highest contribution (~70%) and is mainly  
431 distributed in particles with a size of > 0.43  $\mu\text{m}$ , ~~while EPFRs in which metal oxides~~  
432 ~~participate are mainly distributed in particles with a size of < 0.43  $\mu\text{m}$ .~~ These findings  
433 deepen our understanding of the pollution characteristics of atmospheric EPFRs and

434 are useful for controlling EPFR generation in heavily polluted areas.

435 (3) This study assessed the exposure risk of EPFRs in different areas of the  
436 respiratory system. The results show that the upper respiratory tract is the area with  
437 the highest EPFR exposure (the value in summer is higher than that in winter), with a  
438 value equivalent to 21 cigarettes per person per day. EPFRs are equally exposed to the  
439 trachea and alveoli. The trachea and alveoli are also exposed to EPFRs, and the risk of  
440 exposure is equivalent to that of 8 cigarettes per person per day. Coarse particles are  
441 the main source of EPFRs in the upper respiratory tract, while fine particles are  
442 mainly involved in the alveoli. In summer, combustion sources are the main source of  
443 EPFRs in various parts of the respiratory system. In winter, traffic and other  
444 combustion sources are the main source of EPFRs in the upper respiratory tract, and  
445 combustion sources mainly contribute to the EPFRs in the alveoli.

446 Through this study, ~~we the results~~ have shown that there are significant differences  
447 in the concentrations and types of EPFRs in particles of different sizes and these  
448 differences are due to the influence of the source and generation ~~mechanism~~process.  
449 In the future, assessments of the particle size distribution and the seasonality of  
450 EPFRs in atmospheric PM should be considered. Health risks are another focus of this  
451 study. ~~We It is~~ found that the upper respiratory tract is the key exposure area of EPFRs,  
452 and the traffic source is the main source of EPFRs in this area. This finding is  
453 significant for a systematic assessment of the health risks of EPFRs. In view of the  
454 complexity and diversity of the formation ~~mechanisms~~process of EPFRs in actual  
455 atmospheric particulates, the relative contributions of EPFRs generated by different  
456 ~~mechanisms~~process and their associated health risks should be more comprehensively  
457 studied in the future.

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464 **Appendix A. Supplementary data**

465 Appendix A contains additional details, including the EPR spectra of samples of  
466 different particle sizes, correlations between EPFRs and carbon in particles of  
467 different particle sizes, the results and errors of factor analysis, correlation analysis of  
468 EPFRs with metallic elements, and EPFR exposure in different areas of the human  
469 respiratory tract.

470 **Code/Data availability:** All data that support the findings of this study are  
471 available in this article and its Supplement or from the corresponding author on  
472 request.

473 **Author contribution:** Qingcai Chen: Research design, Methodology, Writing -  
474 Original Draft, Writing - Review & Editing, Project administration, Funding  
475 acquisition; Haoyao Sun: Investigation, Sample analysis, Writing - Original Draft,  
476 Writing - Review & Editing, Methodology, Formal analysis; Wenhui Song:  
477 Investigation, Sample collection, Chemical analysis; Fang Cao: Investigation, Sample  
478 collection; Chongguo Tian: Investigation, Chemical analysis; Yan-Lin Zhang:  
479 Conceptualization, Writing - Review & Editing, Formal analysis, Validation, Funding  
480 acquisition.

481 **Competing interests:** The authors declare that they have no conflict of interest.

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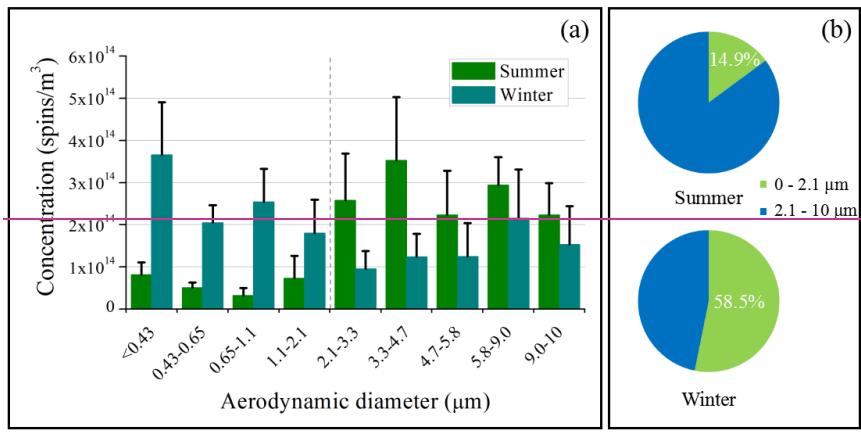
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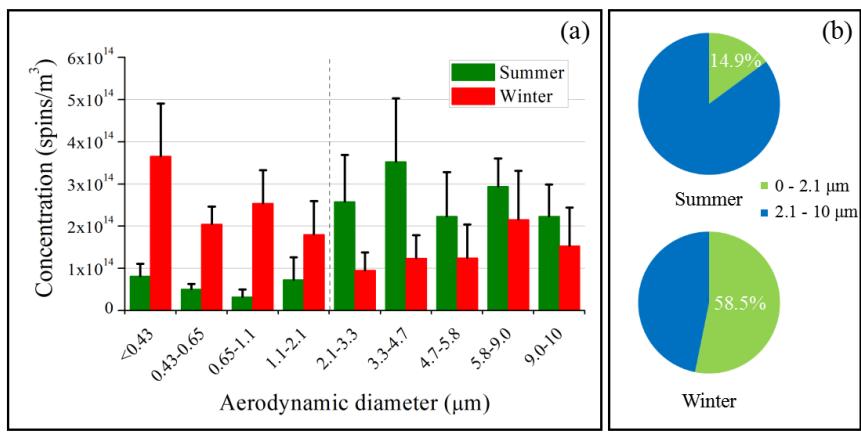
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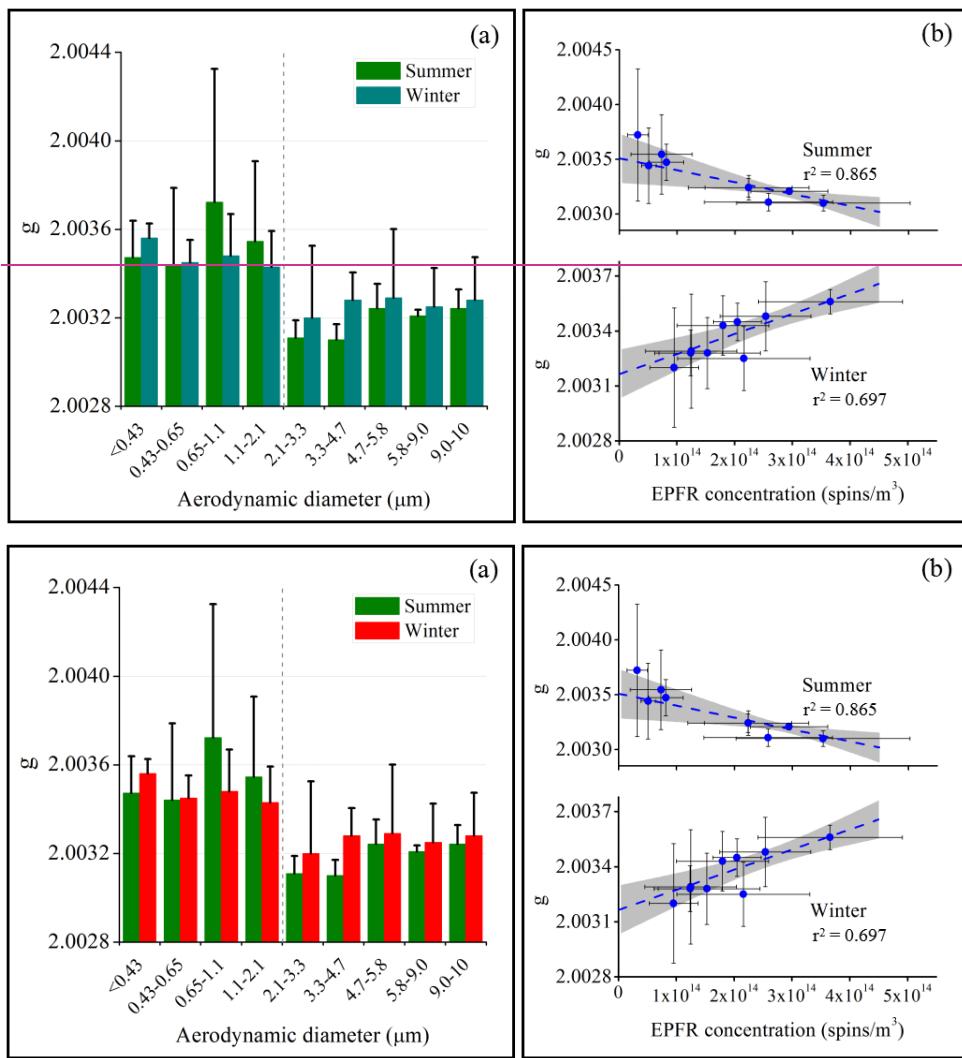
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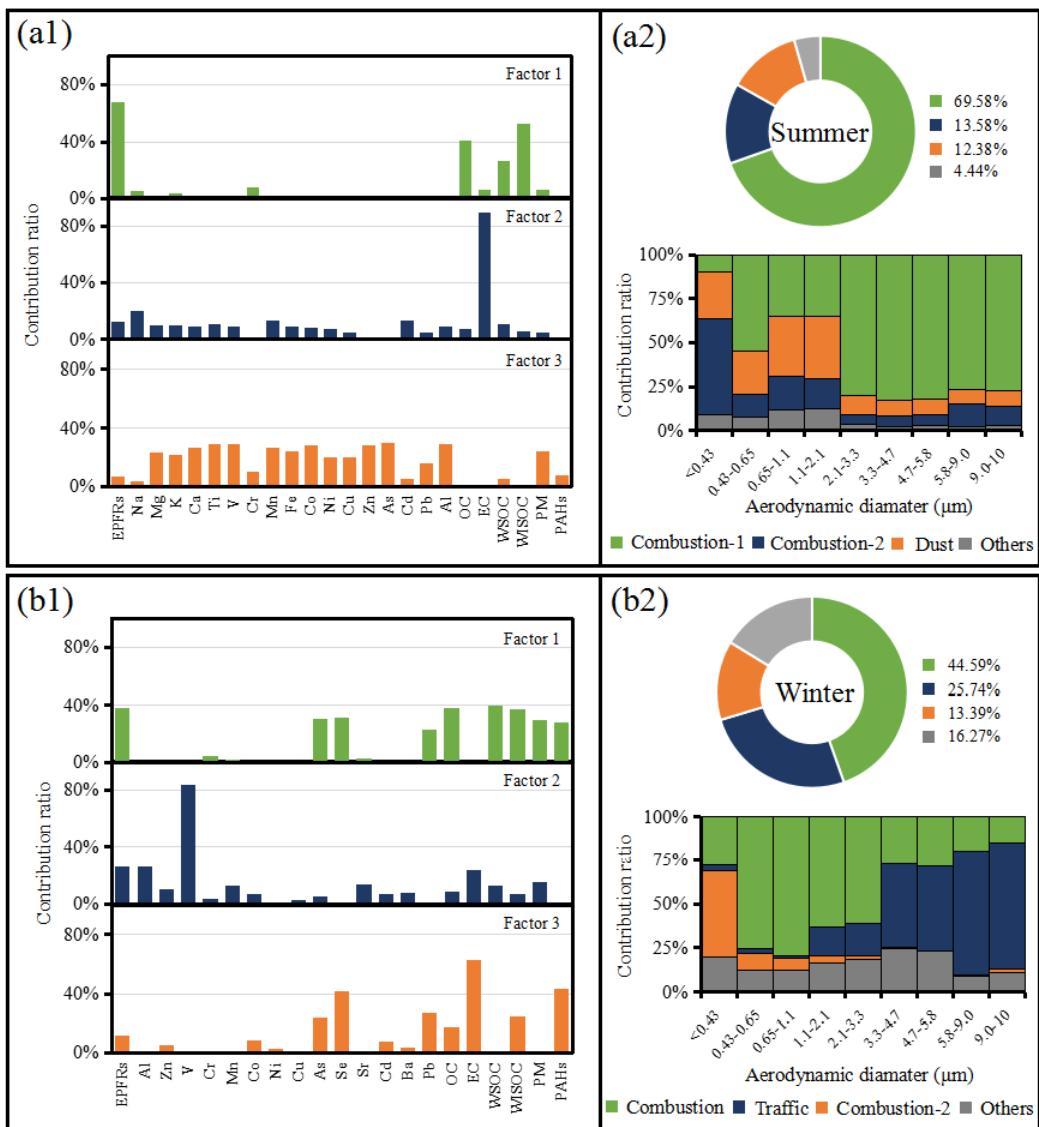
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674 Figure 1. The concentration of EPFRs in PM with different particle sizes. (a) Atmospheric  
 675 concentrations of EPFRs in different particle sizes in summer and winter. (b) The relative  
 676 contribution of fine particles and coarse particles to the total EPFR concentration.

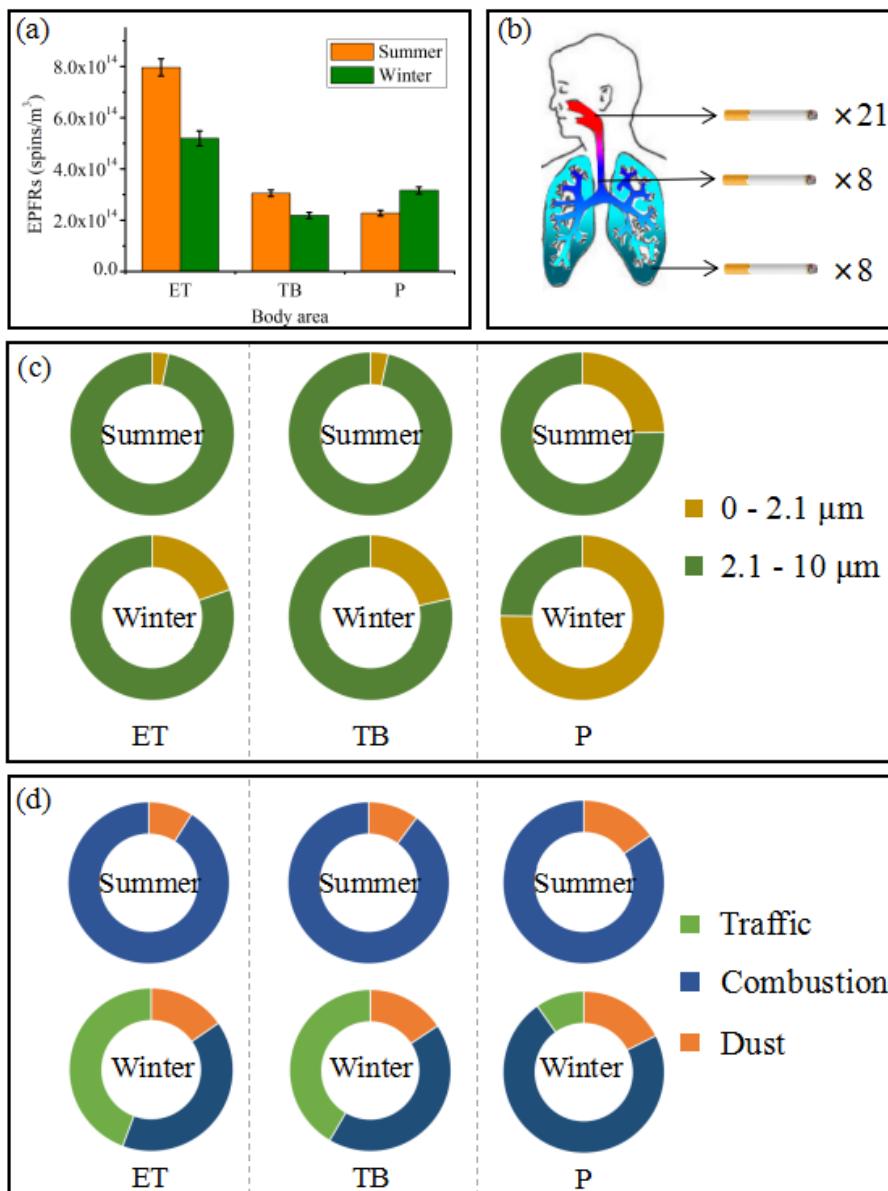


677 Figure 2. A *g*-factor comparison. (a) Comparison of *g*-factors of EPFRs in different particle sizes  
 678 in different seasons. (b) Correlation analysis of *g*-factors and concentrations of EPFRs in summer  
 681 and winter PM. The gray areas in the figure represent 95% confidence intervals.



682

683 Figure 3. Factor analysis of EPFRs in different particle sizes in different seasons. (a1) and (b1)  
 684 represent the results of factor analysis for summer and winter, respectively. (a2) and (b2) represent  
 685 the contribution of various factors in summer and winter, respectively, to EPFRs and the relative  
 686 contributions of each factor for different particle sizes.



687

688 Figure 4. Exposure risks to EPFRs. (a) EPFR exposure in the ET, TB, and P regions. (b) Cigarette  
 689 exposure to EPFRs in the human respiratory system. (c) Exposure ratio of EPFRs with different  
 690 particle sizes in different areas of the respiratory system. (d) Contribution of EPFRs from different  
 691 sources to different areas of the respiratory system.