

# Profiling of formaldehyde, glyoxal, methylglyoxal, and CO over the Amazon: Normalized excess mixing ratios and related emission factors in biomass burning plumes

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**Abstract.** We report on airborne measurements of tropospheric mixing ratios and vertical profiles of formaldehyde ( $\text{CH}_2\text{O}$ ), glyoxal ( $\text{C}_2\text{H}_2\text{O}_2$ ), methylglyoxal and higher carbonyls ( $\text{C}_3\text{H}_4\text{O}_2^*$ ) (see below), and carbon monoxide (CO) over the Amazon Basin during the ACRIDICON-CHUVA campaign from the German High Altitude and Long-range research aircraft (HALO) in fall 2014. The joint observation of in situ CO and remotely measured  $\text{CH}_2\text{O}$ ,  $\text{C}_2\text{H}_2\text{O}_2$ , and  $\text{C}_3\text{H}_4\text{O}_2^*$ , together with visible imagery and air mass back trajectory modelling using NOAA HYSPLIT (National Oceanic Atmospheric Administration, HYbrid Single-Particle Lagrangian Integrated Trajectory) allow us to discriminate between the probing of background tropical air, in which the concentration of the measured species results from the oxidation of biogenically emitted VOCs (mostly isoprene), and measurements of moderately to strongly polluted air masses affected by biomass burning emissions or the city plume of Manaus. For twelve near surface measurements of fresh biomass burning plumes, normalized excess mixing ratios of  $\text{C}_2\text{H}_2\text{O}_2$  and  $\text{C}_3\text{H}_4\text{O}_2^*$  with respect to  $\text{CH}_2\text{O}$  are inferred and compared to recent studies. The mean glyoxal to formaldehyde ratio  $R_{\text{GF}}=0.07$  (range 0.02–0.11) is in good agreement with recent reports which suggest  $R_{\text{GF}}$  to be significantly lower than previously assumed in global chemical transport models (CTMs). The mean methylglyoxal to formaldehyde ratio  $R_{\text{MF}}=0.98$  (range 0.09–1.50) varies significantly during the different observational settings, but overall appears to be much larger (up to a factor of 5) than previous reports suggest even when applying a correction factor of  $2.0 \pm 0.5$  to account for the additional dicarbonyls included in the  $\text{C}_3\text{H}_4\text{O}_2^*$  measurements. Using recently reported emission factors of  $\text{CH}_2\text{O}$  for tropical forests, our observations suggest emission factors of  $\text{EF}_G=0.25$  (range 0.11 to 0.52)  $\text{g kg}^{-1}$  for  $\text{C}_2\text{H}_2\text{O}_2$ , and  $\text{EF}_M=4.7$  (range 0.5 to 8.64)  $\text{g kg}^{-1}$  for  $\text{C}_3\text{H}_4\text{O}_2^*$ . While  $\text{EF}_G$  agrees well with recent reports,  $\text{EF}_M$  is (like  $R_{\text{MF}}$ ) slightly larger than reported in other studies, presumably due to the different plume ages or fuels studied. Our observations of these critical carbonyls and intermediate oxidation products may support future photochemical modelling of air pollution over tropical vegetation, as well as validate past and present space-borne observations of the respective species.

## 1 Introduction

Emissions from biomass burning are a large source of reactive gases and particulate matter to the atmosphere on a local, regional and up to the global scale (Crutzen and Andreae, 1990; Andreae and Merlet, 2001a; Andreae et al., 2001; Andreae and Merlet, 2001b). Pyrogenic emissions form a complex mixture of species that can change over the duration of the fire and undergo further chemical reactions as emissions are transported downwind. In consequence, biomass burning may have adverse effects on the atmospheric photochemistry, secondary aerosol formation and composition, and hence cloud formation and radiation, as well as human health. Therefore, accurate measurements of pyrogenic emissions, the evolution of fire plume composition and the formation and aging of aerosols are needed, but respective data are lacking for many species (e.g., Andreae and Merlet (2001a); Akagi et al. (2011); Stockwell et al. (2015); Andreae and Merlet (2001b)).

30 Among the manifold of species emitted in large amounts by fires or formed in their plumes are carbonyl compounds such as formaldehyde ( $\text{CH}_2\text{O}$ ), glyoxal ( $\text{C}_2\text{H}_2\text{O}_2$ ), methylglyoxal ( $\text{C}_3\text{H}_4\text{O}_2$ ), 2,3-butanedione ( $\text{C}_3\text{H}_6\text{O}_2$ ), and many others (e.g., Andreae and Merlet (2001b)). Here and further on,  $\text{C}_3\text{H}_4\text{O}_2^*$  denotes  $\text{C}_3\text{H}_4\text{O}_2$  and other substituted dicarbonyls (cf. 2,3-butanedione) with visible absorption spectra similar to  $\text{C}_3\text{H}_4\text{O}_2$ , since they cannot be distinguished with the spectral resolution available in our limb measurements (sect. 2.1).

35 It is well known that in the atmosphere most of  $\text{CH}_2\text{O}$  is formed as an intermediate oxidation species from the degradation of  $\text{CH}_4$  and from a suite of non-methane hydrocarbons (NMHCs) (Seinfeld and Pandis, 2013).  $\text{CH}_2\text{O}$  can further be directly emitted into the troposphere by biomass burning (e.g., Finlayson-Pitts and Pitts (2000); Andreae and Merlet (2001a); Wagner et al. (2002); Akagi et al. (2011); Seinfeld and Pandis (2013); Stockwell et al. (2015)), and from incomplete combustion as well as the direct emission by vegetation (Carlier et al. (1986) and references therein). Due to its short lifetime in the 40 atmosphere of only several hours,  $\text{CH}_2\text{O}$  mixing ratios range from several 10 ppt to a few 100 ppt in the pristine (e.g., marine) environment. In the lower or polluted troposphere, mixing ratios might reach several ppb (e.g., Finlayson-Pitts and Pitts (2000); Jaeglé et al. (1997); Wagner et al. (2002); Borbon et al. (2012); Seinfeld and Pandis (2013) and others). Secondary formed  $\text{CH}_2\text{O}$  (for example from acetone photolysis) is an important source of  $\text{HO}_x$  radicals, in particular in the middle and upper troposphere (Jaeglé et al., 1997). Further, since  $\text{CH}_2\text{O}$  is considerably water soluble, it may participate in either acid 45 or base catalyzed aldol-condensation reactions in the aerosol phase, hence contributing to secondary organic aerosol (SOA) formation (Wang et al., 2010). Numerous studies of formaldehyde emissions from biomass burning both in the laboratory and the field were undertaken in the past, and its emission factor appears to be fairly well established for the various studied fuel and fire types (e.g., Finlayson-Pitts and Pitts (2000); Andreae and Merlet (2001a); Wagner et al. (2002); Akagi et al. (2011); Seinfeld and Pandis (2013); Stockwell et al. (2015); Andreae and Merlet (2001b)). Based on this knowledge, models have 50 generally been able to reproduce reasonably well the formaldehyde column densities observed by satellites (e.g., Dufour et al. (2009); Stavrakou et al. (2009b); Boeke et al. (2011); Bauwens et al. (2016), and others), or concentrations directly measured in the atmosphere (e.g., Arlander et al. (1995); Fried et al. (2008); Borbon et al. (2012); Kaiser et al. (2015); Chan Miller et al. (2017) and others).

Similar to formaldehyde, large parts of the atmospheric glyoxal (47 %) and methylglyoxal (79 %) are thought to be formed 55 during the oxidation of isoprene emitted by vegetation (Fu et al., 2008; Wennberg et al., 2018). Recent air-borne and satellite measurements identified the Amazon to be the world most prominant isoprene source region, with isoprene mixing ratios exceeding 6 ppb in the boundary layer by end of the dry season in September 2014 (Gu et al., 2017; Fu et al., 2019). These elevated mixing ratios are a consequence of the large isoprene emission from the rain forest combined with a significantly 60 prolonged atmospheric lifetime of up to 36 hours (Fu et al., 2008) over the Amazon Basin compared to the global mean of less than 4 hours (Fu et al. (2019), fig. 6a). Fu et al. (2019) suspected strong isoprene emissions combined with low NO<sub>x</sub> concentrations to lead to significantly supressed OH-levels as a possible reason for the observed slower isoprene oxidation in this regime. The longer isoprene lifetime and efficient vertical transport over the Amazon lead to elevated isoprene mixing 65 ratios not only within the boundary layer but also in higher altitudes (Fu et al. (2019), fig. 4d), where it is a possible source for in situ glyoxal and methylglyoxal production in the free troposphere. Satellite observations have shown strongly enhanced vertical column densities of atmospheric glyoxal above this region (e.g., Myriokefalitakis et al. (2008a)).

The second most important precursors of the trace gases are acetylene (mostly anthropogenically emitted) for glyoxal and acetone (mostly biogenic) for methylglyoxal, but both gases are also formed during the oxidation of other volatile organic compounds (VOCs), particularly alkenes, aromatics, and monoterpenes (Fu et al., 2008; Taraborrelli et al., 2020). For both gases, Fu et al. (2008) estimated their global source strengths to 45 Tg a<sup>-1</sup> and 140 Tg a<sup>-1</sup>, respectively. Atmospheric lifetimes of 70 glyoxal and methylglyoxal are only about 2 h, mostly due to photolysis and to a lesser degree due to reactions with OH radicals (Koch and Moortgat, 1998; Volkamer et al., 2005a; Tadić et al., 2006; Fu et al., 2008; Wennberg et al., 2018). Accordingly, in ambient air C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> and C<sub>3</sub>H<sub>4</sub>O<sub>2</sub> mixing ratios typically do not exceed several 100 ppt (Fu et al., 2008).

Glyoxal and methylglyoxal are additionally emitted in considerable quantities by biomass burning, their individual emission 75 strengths however are not yet well established (e.g., Andreae and Merlet (2001a); Akagi et al. (2011); Stockwell et al. (2015); Zarzana et al. (2017, 2018); Andreae and Merlet (2001b)). The recent compilation of emission factors by Andreae and Merlet (2001b) indicates for most reported fuels the complete lack of emission data for both dicarbonyls. Until the recent laboratory and field studies by Zarzana et al. (2017, 2018), previous information on the glyoxal and methylglyoxal emissions from biomass burning were based on only two laboratory studies (McDonald et al., 2000a; Hays et al., 2002). While Hays et al. (2002) reported larger emission factors for glyoxal compared to methylglyoxal, the recent findings of Zarzana et al. (2017, 2018) 80 indicated significantly smaller glyoxal emissions but larger emissions of methylglyoxal and other substituted dicarbonyls in fires of agricultural biomass than those previously assumed in models (Fu et al., 2008).

Satellite and aircraft measurements of formaldehyde and glyoxal complemented by modelling have been used to study their different atmospheric sources (e.g., Stavrakou et al. (2009b); Boeke et al. (2011); Lerot et al. (2010); Chan Miller et al. (2014); Bauwens et al. (2016); Stavrakou et al. (2016), and others), the secondary aerosol formation from carbonyls (typically in the 85 background atmosphere), as well as in biomass burning plumes (Knote et al., 2014; Li et al., 2016; Lim et al., 2019), and even more recently to estimate the organic aerosol abundance (Liao et al., 2019). Unlike for formaldehyde, the models had varying success in reproducing the glyoxal column densities (Myriokefalitakis et al., 2008b; Stavrakou et al., 2009a; Lerot et al., 2010). In comparison to satellite measurements from SCIAMACHY and GOME-2, several studies have found evidence

that the models underestimate global glyoxal emissions, when not considering additional biogenic sources (Myriokefalitakis  
90 et al., 2008a; Stavrakou et al., 2009a; Lerot et al., 2010).

The ratios of  $C_2H_2O_2$  to  $CH_2O$  ( $R_{GF}$ ), and  $C_3H_4O_2$  to  $CH_2O$  ( $R_{MF}$ ) can be used as indications of the precursor VOC species and to identify biomass burning in contrast to biogenic sources (Kaiser et al., 2015). In the past,  $R_{GF}$  has been studied based on multiple ground-based observations as well as satellite measurements (e.g., McDonald et al. (2000a); Kaiser et al. (2015); Chan Miller et al. (2017)). Kaiser et al. (2015) studied inferred  $R_{GF}$  over the south-eastern United States during  
95 the 2013 SENEX (Southeast Nexus) campaign, and compared these measurements to respective OMI (Ozone Monitoring Instrument) satellite retrievals. They concluded that in regimes rich of monoterpene emissions  $R_{GF}$  is generally larger than 3 %, while regions dominated by isoprene oxidation appear characterized by low  $R_{GF} < 3 \%$ . For a compilation of previously inferred  $R_{GF}$  see Kaiser et al. (2015), Table 1.

More recently, Stavrakou et al. (2016) examined emissions of crop residue fires in the North China Plain using data from  
100 the OMI satellite. Their inferred column emission ratio of 0.04–0.05 was comparable to the  $R_{GF}$  observed by Zarzana et al. (2017), but smaller than those reported by McDonald et al. (2000a) and Hays et al. (2002), which were used in the study of Fu et al. (2008). Nevertheless, the chemical transport model used by Stavrakou et al. (2016) was able to reproduce the measured formaldehyde column densities and the glyoxal enhancements observed over the North China Plain during the peak burning season.

105 Information on methylglyoxal in the atmosphere is even more rare, since to our knowledge satellite instruments measuring methylglyoxal are not (yet) available, and atmospheric measurements of methylglyoxal are still sparse. This is a result of the moderate spectral resolution of space- and air-borne optical measurements, which suffer from the spectral interference of the major but only weakly structured absorption bands of methylglyoxal with those of other dicarbonyls (mainly 2,3-butanedione) in the blue spectral region (Meller et al., 1991; Horowitz et al., 2001; Thalman et al., 2015; Zarzana et al., 2017). Potential  
110 interferences with the  $7\nu$  absorption band of water vapour at 442 nm further complicate the spectral retrieval (Thalman et al., 2015; Zarzana et al., 2017). In situ measurements of methylglyoxal using different techniques have only recently become available (Thalman and Volkamer, 2010; Pang et al., 2011; Lawson et al., 2015; Michoud et al., 2018). The very limited number of measurements in conjunction with insufficient detection limits are still preventing comprehensive studies of methylglyoxal.

115 Here we report on simultaneous measurements of  $CO$ ,  $CH_2O$ ,  $C_2H_2O_2$ , and  $C_3H_4O_2^*$  in the lower to upper troposphere over the Amazon Basin at the end of the dry season in 2014. All measurements were performed from the German research aircraft HALO operated by the Deutsches Zentrum für Luft- und Raumfahrt (DLR). The simultaneous near surface measurements of  $C_2H_2O_2$  and  $C_3H_4O_2^*$  allow us to estimate their emission ratios with respect to  $CH_2O$  as well as the normalized excess mixing ratios within the probed biomass burning plumes, and combined with previously reported emission factors of  $CH_2O$  (e.g., Andreae and Merlet (2001b)), to estimate their emission factors for tropical forest fires. The joint measurements of the  
120 carbonyls and  $CO$  in the middle and in the upper troposphere, more distant from their direct sources at the ground, provides insight into their tropospheric profiles and precursors in a pristine atmosphere and under low  $NO_x$  conditions. Additionally, air mass backward trajectory modelling classifies the probed air masses and their possible origins.

**Table 1.** Trace gas absorption cross sections used for the DOAS based spectral retrieval.

No.	Absorber	Temperature [K]	Reference	Uncertainty [%]
1	O <sub>4</sub>	273	Thalman and Volkamer (2013)	4
2	O <sub>3</sub>	203	Serdyuchenko et al. (2014)	3
3	O <sub>3</sub>	273	Serdyuchenko et al. (2014)	3
4	NO <sub>2</sub>	273	Burrows et al. (1998)	4
5	H <sub>2</sub> O	293	Rothman et al. (2009)	
6	CH <sub>2</sub> O	293	Chance and Orphal (2011)	10
7	C <sub>2</sub> H <sub>2</sub> O <sub>2</sub>	296	Volkamer et al. (2005b)	3
8	C <sub>3</sub> H <sub>4</sub> O <sub>2</sub>	296	Meller et al. (1991)	10
9	C <sub>4</sub> H <sub>6</sub> O <sub>2</sub>	223	Horowitz et al. (2001)	4

The paper is organized as follows. Chapter 2 briefly describes the measurement techniques and methods used in the study. Chapter 3 and 4 report on the ACRIDICON-CHUVA campaign and the actual measurements, and chapter 5 discusses the major 125 findings and results. Chapter 6 concludes and summarises the study.

## 2 Methods

### 2.1 DOAS Measurements

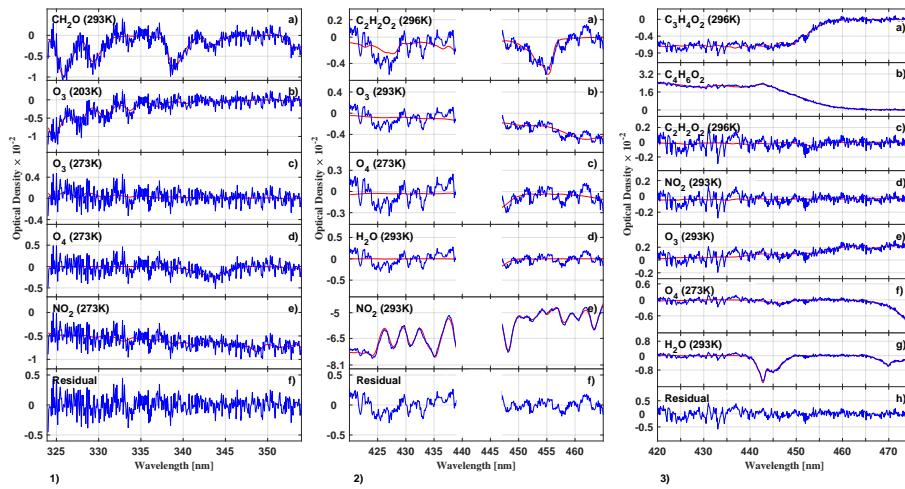
The mini-DOAS instrument is a UV/vis/near-IR six channel optical spectrometer, which detects scattered skylight received from scanning limb and nadir directions. These spectra can be analyzed for absorption structures of O<sub>4</sub>, CH<sub>2</sub>O, C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> and 130 C<sub>3</sub>H<sub>4</sub>O<sub>2</sub><sup>\*</sup> (among other species). A video camera (type uEye UI-1005XS from IDS Imaging Development Systems GmbH) with a field of view of 46 ° is installed into the telescope unit and co-aligned with the spectrometer's limb telescopes. It provides visual imagery of the probed atmospheric scene at a rate of 1 Hz. An extensive description of the instrument and its deployment on the HALO aircraft can be found in Hüneke (2016); Hüneke et al. (2017). The mini-DOAS instrument remotely probes the atmosphere with the limb telescopes (FOV 0.5° vertical, 3.15° horizontal) oriented perpendicular and right hand to the aircraft 135 main axis. Although the limb telescopes can be directed at any elevation angle for traditional atmospheric profiling, during the ACRIDICON-CHUVA campaign they were constantly directed at 0.3° below the horizon, for optimal measurements when using the O<sub>4</sub> scaling method (Hüneke (2016); Hüneke et al. (2017); Stutz et al. (2017); Werner et al. (2017), and below).

The post-flight analysis of the collected limb spectra includes a DOAS (Differential Optical Absorption Spectroscopy (Platt and Stutz, 2008)) analysis of the measured skylight spectra for the absorption structures of O<sub>4</sub>, CH<sub>2</sub>O, C<sub>2</sub>H<sub>2</sub>O<sub>2</sub>, and C<sub>3</sub>H<sub>4</sub>O<sub>2</sub><sup>\*</sup> 140 as well as the attribution of the measured absorption to the correct locations in the atmosphere. The latter is performed using the O<sub>4</sub> scaling method described in detail in Hüneke (2016); Hüneke et al. (2017); Stutz et al. (2017), as well as below. The DOAS analysis of the data is performed according to the settings in Tables 1 and 2.

**Table 2.** Details of the DOAS spectral analysis for the various trace gases.

Target gas	Spectral interval [nm]	Fitted absorbers (see Table 1)	Add. Param.	Polyn.	$\sigma$ [dSCD]
O <sub>4</sub>	348–369	1, 2, 3, 4, 6	$I_{\text{Ofs}}^{(i)}$ , $R^{(ii)}$ , $R \cdot \lambda^4$ <sup>(iii)</sup>	2	$8 \times 10^{41}$
	460–490	1, 2, 3, 4, 5	$I_{\text{Ofs}}$ , $R$ , $R \cdot \lambda^4$	2	$4 \times 10^{41}$
CH <sub>2</sub> O	324–354	1, 2, 3, 4, 6	$I_{\text{Ofs}}$ , $R$ , $R \cdot \lambda^4$	2	$5 \times 10^{15}$
C <sub>2</sub> H <sub>2</sub> O <sub>2</sub>	420–439 and 447–465	1, 3, 4, 5, 7	$I_{\text{Ofs}}$ , $R$ , $R \cdot \lambda^4$	2	$1 \times 10^{15}$
C <sub>3</sub> H <sub>4</sub> O <sub>2</sub> <sup>*</sup>	420–475	1, 3, 4, 5, 7, 8, 9	$I_{\text{Ofs}}$ , $R$ , $R \cdot \lambda^4$	2	$1 \times 10^{16}$

(i)  $I_{\text{Ofs}}$ : Offset spectrum; (ii)  $R$ : Ring spectrum; (iii)  $R \cdot \lambda^4$ : Ring spectrum multiplied by  $\lambda^4$ .



**Figure 1.** Inferred absorption spectra of CH<sub>2</sub>O (1), C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> (2), and C<sub>3</sub>H<sub>4</sub>O<sub>2</sub><sup>\*</sup> (3) for the measurements at 17:16, 15:06 and 14:53 UTC, respectively, during the flight on Sept. 11, 2014 (AC11). The traces shown in blue are the inferred atmospheric spectra together with the residual spectral structures, the red line shows the reference spectra of the respective gases. In order to avoid cross correlations with the 7 $\nu$  absorption band of H<sub>2</sub>O at 442 nm, C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> is simultaneously fitted over two separated spectral windows, ranging from 420 nm to 439 nm and 447 nm to 465 nm.

Exemplary sample retrievals of CH<sub>2</sub>O, C<sub>2</sub>H<sub>2</sub>O<sub>2</sub>, and C<sub>3</sub>H<sub>4</sub>O<sub>2</sub><sup>\*</sup> are shown in fig. 1. The detection of C<sub>3</sub>H<sub>4</sub>O<sub>2</sub><sup>\*</sup> (fig. 1, panel 3) deserves a brief discussion. It is known that substituted dicarbonyls such as biacetyl (CH<sub>3</sub>COCOCH<sub>3</sub>), with absorption cross sections in the visible spectral range similar to those of C<sub>3</sub>H<sub>4</sub>O<sub>2</sub>, are also co-emitted from biomass fires in significant quantities (e.g., Meller et al. (1991); Horowitz et al. (2001); Thalman et al. (2015); Zarzana et al. (2017)). Owing to the moderate resolution of our spectrometer in the visible spectral range (full width half maximum resolution of 1.1 nm or 8.4 pixels, see Hüneke (2016)), we cannot separately (and thus unambiguously) detect these substituted dicarbonyls in the atmosphere, unlike in studies employing higher resolving spectrometers (e.g., Thalman et al. (2015)). Accordingly, we report here the weighted

150 sum (absorption cross section times concentration) of  $\text{C}_3\text{H}_4\text{O}_2$  and that of other substituted dicarbonyls and express that quantity as  $\text{C}_3\text{H}_4\text{O}_2^*$ .

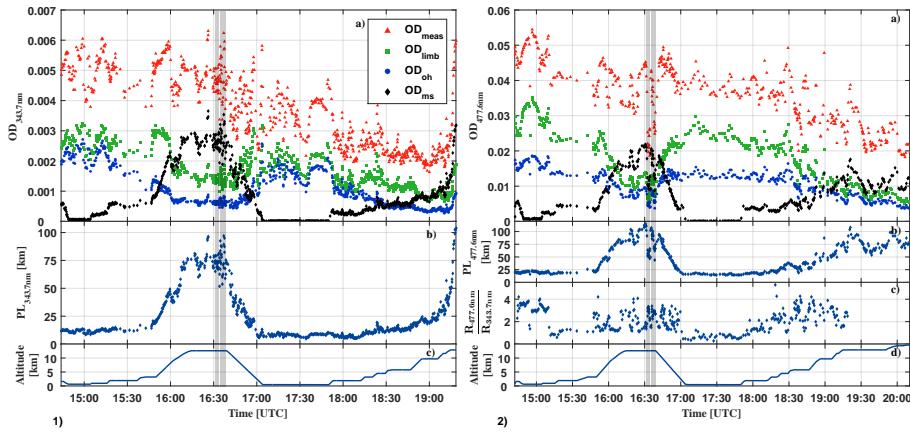
From the retrieved slant column densities, absolute concentrations of the targeted gases are inferred using the recently developed  $\text{O}_4$  scaling method. As shown in previous studies, the scaling method is based on simultaneous measurements of the target gas X and a scaling gas P of known concentration (here the collision complex  $\text{O}_2-\text{O}_2$ , briefly called  $\text{O}_4$ ) in the same 155 wavelength interval (Hüneke, 2016; Hüneke et al., 2017; Stutz et al., 2017; Werner et al., 2017). From the mini-DOAS limb measurements of the total slant column density, the concentration  $[\text{X}]_j$  of the trace gas X in the atmospheric layer j (i.e., the altitude of the aircraft) is then determined from

$$[\text{X}]_j = \frac{\alpha_{X_j}}{\alpha_{P_j}} \cdot \frac{\text{SCD}_X}{\text{SCD}_P} \cdot [\text{P}]_j \quad (1)$$

(see Stutz et al. (2017), eq. (14)). Here,  $[\text{P}]_j$  is the calculated extinction of  $\text{O}_4$  ( $\epsilon_{\text{O}_4} = \sigma \cdot k_{\text{eq}} \cdot [\text{O}_2]_j^2$ ) in the atmospheric layer j, 160 from which we take the sample.  $\text{SCD}_P$  is the measured optical depth of  $\text{O}_4$  and  $\text{SCD}_X$  is the measured slant column density of the target gas X. The  $\alpha$ -factors  $\alpha_{X_j}$  and  $\alpha_{P_j}$  each express the ratio of the absorption of the gas in layer j to the total atmospheric absorption. Usually, the  $\alpha$ -factors are simulated in supporting radiative transfer calculations using the Monte Carlo model 165 McArtim (Deutschmann et al., 2011; Knecht, 2015). When needed, the total atmospheric column density of the respective gases is approximated by integrating the measured lower and higher quartile profiles in incremental steps of 100 m each (fig. 9).

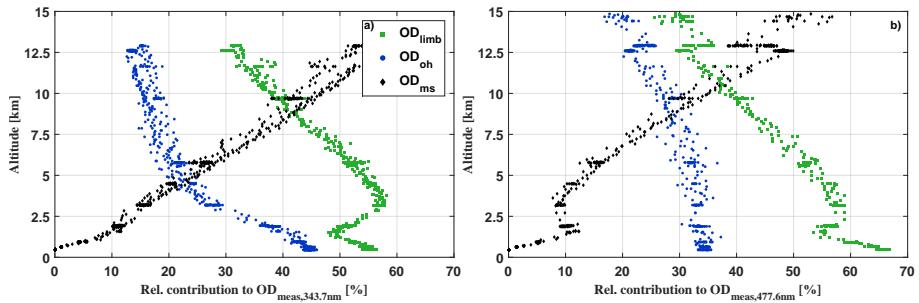
When applying the scaling method to air-borne UV/vis spectroscopy in limb geometry, a powerful radiative transfer model is necessary to treat the radiative transfer in 2D or 3D, calculate the refraction and also account for the sphericity of the Earth. We use the Monte Carlo radiative transfer model McArtim (Deutschmann et al., 2011), where each measurement is modelled forward while considering the actual atmospheric, instrumental, and observational (e.g., Sun position relative to the pointing) 170 parameters, the geolocation and the pointing of the telescope as described in Hüneke (2016); Hüneke et al. (2017). Further, a combination of climatological aerosol profiles obtained by LIVAS (Amiridis et al., 2015) and the Stratospheric Aerosol and Gas Experiment II (SAGE II) ([https://eosweb.larc.nasa.gov/project/sage3/sage3\\_table](https://eosweb.larc.nasa.gov/project/sage3/sage3_table)) is considered. For the calculation of the  $\alpha$ -factors, box air mass factors are simulated, and summed weighted by concentration (Stutz et al. (2017), eq. (11)). We calculate 175 the extinction  $[\text{P}]_j$  to obtain the vertical  $\text{O}_4$  profile. For the target gases  $\text{CH}_2\text{O}$ ,  $\text{C}_2\text{H}_2\text{O}_2$ ,  $\text{C}_3\text{H}_4\text{O}_2^*$ , we assume exponentially decaying profiles. The latter appears justified because of photochemical arguments, the source distribution (mostly surface), the photochemical lifetimes, the time scales of possible vertical transport, as well as when inspecting the profile shapes obtained in the post-analysis (fig. 9).

Evidently, UV/vis limb measurements naturally divide the atmosphere into three different layers, which are separated by the different contributions to the total absorption or optical depth (OD). These contributions are (a) the overhead absorption 180 ( $\text{OD}_{\text{oh}}$ ), (b) the absorption located within the line-of-sight of the telescopes ( $\text{OD}_{\text{limb}}$ ) (of which most, but not all is due to single scattering), and (c) the absorption below the line of sight ( $\text{OD}_{\text{ms}}$ ) (of which the photon paths are necessarily all of higher scattering order). In the following, we discuss the significance of the different contributions exemplarily for  $\text{O}_4$ . Evidently, the same tri-partitioning applies for all other gases of interest, however differently weighted as expressed by the



**Figure 2.** Attribution of the total measured optical depth of O<sub>4</sub> at 343.7 nm (panel 1 on the left) and at 477.6 nm (panel 2 on the right) to the various fractions, i.e., absorption from overhead the aircraft (OD<sub>oh</sub>), from multiple scattering below the aircraft (OD<sub>ms</sub>), within the line-of-sight of the telescopes (OD<sub>limb</sub>), and the total measured optical depth of O<sub>4</sub> (OD<sub>meas</sub>) for the HALO flight on Sept. 16, 2014 (AC11) (panels 1a, 2a). The average line-of-sight photon path length inferred from the measured OD<sub>meas</sub> at 343.7 nm and 477.6 nm, respectively, as a function of the flight time is plotted in panels 1b, 2b. The color ratio  $\frac{R_{343.7}}{R_{477.6}}$  of the measured radiances at 343.7 nm versus 477.6 nm is shown in panel 2c. The flight trajectory of the HALO aircraft is shown in panels 1c and 2d. The sharp drop of the optical depth between 16:32–16:34 UTC and 16:35–16:39 UTC (grey bars) coincides with the passage of a cumulonimbus cloud in front of the telescopes. For details see the text.

respective alpha-factors. For a high-flying aircraft, OD<sub>oh</sub> has the least uncertainty since O<sub>4</sub> absorption approximately scales 185 with [O<sub>2</sub>]<sup>2</sup> and the optical state of the overhead atmosphere is well known. Therefore, the remainder of the measured total O<sub>4</sub> absorption has to be due to the varying contributions of OD<sub>limb</sub> and OD<sub>ms</sub>, which are mostly modulated by the current low level cloud cover and ground albedo (Stutz et al. (2017), fig. 7). The knowledge of the  $\alpha$ -factor however gives a handle 190 on the relative fractions OD<sub>limb</sub> and OD<sub>ms</sub>, since contribution OD<sub>limb</sub> can be calculated from OD<sub>limb</sub>= $\alpha \cdot$ OD<sub>meas</sub>, and accordingly OD<sub>ms</sub>=OD<sub>meas</sub>–OD<sub>oh</sub>–OD<sub>limb</sub>. Still, the calculated attributions of OD<sub>limb</sub> and OD<sub>ms</sub> to OD<sub>meas</sub> can only 195 be approximated due to the assumptions made regarding the current aerosols distribution and the cloud coverage. Figure 2 illustrates the contributions to OD<sub>meas</sub> at 343.7 nm (panel 1a) and 477.3 nm (panel 2a) for O<sub>4</sub> as a function of the flight time for Sept. 16, 2014. Figure 3 shows the same data, but sorted by flight altitude to demonstrate the relative change of the contributions. Up to 10 km altitude and for both investigated wavelengths, the absorption within the line of sight of the telescopes contributes with more than 50 % to the total O<sub>4</sub> absorption. A relative minimum can be seen at the top of 200 the planetary boundary layer at approximately 2 km, where often stratocumulus clouds prevail over the Amazon. Further, a maximum of OD<sub>limb</sub> is visible at about 4 km, most probably a result of the increasing horizontal visibility with altitude and the moderate contribution of reflected multiple scattered photons from the lower atmosphere to the total O<sub>4</sub> absorption. In the upper troposphere, the relative contribution of OD<sub>limb</sub> to OD<sub>meas</sub> decreases while the effective horizontal light paths become 205 longer and reach maximum values around 100 km. But given all uncertainties in the details of the actual atmospheric aerosol



**Figure 3.** Relative contribution to the limb measured optical depth of O<sub>4</sub> at 343.7 nm (panel a) and at 477.6 nm (panel b) due to the absorption overhead the aircraft (OD<sub>oh</sub>), from multiple scattering (OD<sub>ms</sub>), and within the line of sight of the telescopes (OD<sub>limb</sub>) for the HALO flight on Sept. 11, 2014 (AC11). For details see the text.

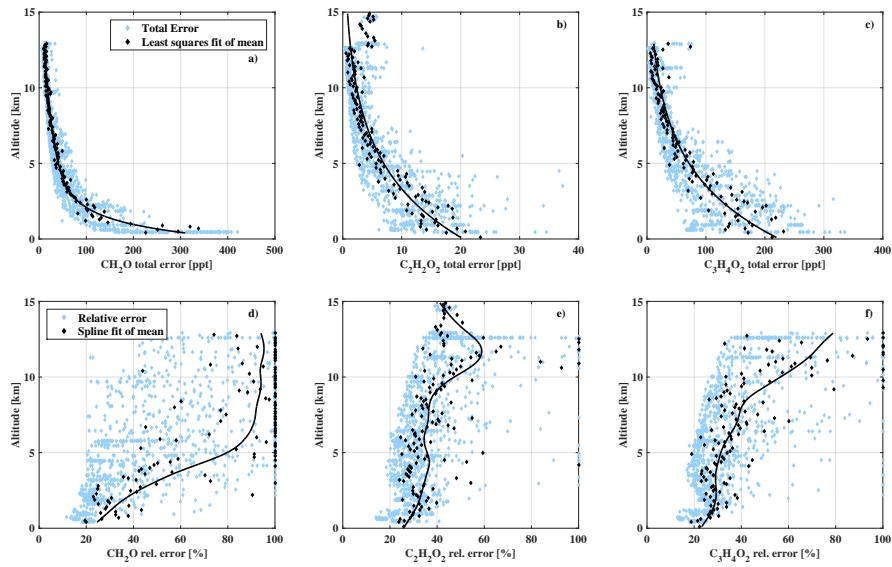
200 content, the cloud structure and coverage and their optical properties, it is more reasonable to speak here of an indication rather than a determination of the photon path lengths.

Fortunately, these uncertainties do not propagate into the uncertainty of the  $\alpha$ -factor ratio and hence into the determination of [X]<sub>j</sub>, provided the gases X and P have similar profile shapes (Knecht, 2015; Hüneke et al., 2017; Stutz et al., 2017). From the above discussion, it also becomes clear that our air-borne UV/vis limb measurements average over some atmospheric volume, 205 which is determined by the viewing angle of the telescope lenses (0.3 °), the light path length, and the aircraft displacement during the time of measurement, both of which are in the order of several kilometers (for details see sect. 3, and fig. 6). This large sampling volume precludes direct comparisons with in situ measured quantities on spatial scales smaller than the current averaging volume. This is of relevance to our study, for example when adding information from in situ measurements like CO to the analysis of our data.

210 Summing-up all described uncertainties, the precision error of the combined methods can be approximately calculated according to eq. (1) as

$$[\Delta X]_j = \sqrt{\left(\frac{\Delta \alpha_{R,j}}{\alpha_{R,j}}\right)^2 + \left(\frac{\Delta S_{X,j}}{S_{X,j}}\right)^2 + \left(\frac{\Delta S_{P,j}}{S_{P,j}}\right)^2 + \left(\frac{\Delta P_j}{P_j}\right)^2 \cdot [X]_j}, \quad (2)$$

where  $\Delta \alpha_{R,j}$  accounts for (a) random Mie extinction, (b) small scale variability of the target and scaling gas mixing ratios at flight level and (c) uncertainty in the sampling contribution as discussed in fig. 2 and 3. The slant column errors  $\Delta S_{X,j}$  and  $\Delta S_{P,j}$  account for (a) uncertainty of the inferred fraunhofer reference SCD and (b) the DOAS fit error.  $\Delta P_j$  accounts for the uncertainty of the in situ mixing ratio of the scaling gas, which in the case of O<sub>4</sub> scales with the air density and is accordingly limited by the measurement uncertainty of pressure and temperature. Additionally, the systematic errors of the individual absorption cross sections need to be added to  $\Delta S_{X,j}$  according to Table 1. An extensive discussion of the total error budget of the O<sub>4</sub> scaling can be found in Hüneke (2016). While  $\Delta P_j$  and  $\Delta \alpha_{R,j}$  contribute approximately constantly to the total error budget,  $\Delta S_{X,j}$  strongly increases with decreasing slant column density. For mixing ratios below 1 ppb (CH<sub>2</sub>O), 215 0.15 ppb (C<sub>2</sub>H<sub>2</sub>O<sub>2</sub>), and 1.3 ppb (C<sub>2</sub>H<sub>4</sub>O<sub>2</sub><sup>\*</sup>), the total precision error is strongly dominated by  $\Delta S_{X,j}$ . For higher mixing



**Figure 4.** Relative and absolute total precision error of all measurements as a function of the measurement altitude.

ratios,  $\Delta\alpha_{R,j}$  and  $\Delta S_{X,j}$  contribute in equal parts in the cases of  $C_2H_2O_2$  and  $C_2H_4O_2^*$ . For  $CH_2O$ ,  $\Delta\alpha_{R,j}$  is the major error factor for mixing ratios above 1 ppb, and  $\Delta S_{X,j}$  only makes up 30 % of the total error. For all gases,  $\Delta P_j$  never exceeds 7 % of the total precision error. Due to the strong dependence of  $\Delta S_{X,j}$  on the slant column density and the exponentially decreasing vertical profiles of the gases, the resulting total precision errors  $[\Delta X]_j$  are strongly altitude dependent (fig. 4). They range from 16 % to 100 % for  $CH_2O$ , 17 % to 100 % for  $C_2H_2O_2$ , and 16 % to 100 % for  $C_2H_4O_2^*$ . This study focuses on measurements below 2 km altitude, with mean total errors of the analyzed biomass burning events in the order of 21 % for  $CH_2O$  (upper and lower limit of the total precision errors are 350 ppt and 740 ppt), 25 % for  $C_2H_2O_2$  (total errors from 30 ppt to 45 ppt), and 22 % for  $C_2H_4O_2^*$  (total errors from 240 ppt to 560 ppt).

## 230 2.2 AMTEX CO Measurements

In situ CO measurements were performed using an Aero-Laser 5002 vacuum UV resonance fluorescence instrument (Gerbig et al., 1996, 1999) installed in the cabin of the HALO aircraft. Main components of the CO instrument include a fluorescence chamber, resonance lamp, optical filter, two dielectric mirrors, and a pump. Atmospheric air is sampled through a backward-facing inlet mounted on top of the HALO fuselage. CO molecules in the ambient air are excited in the chamber by radiation from a radio frequency discharge resonance lamp. The wavelength range (center line 151 nm with a bandwidth of 9 nm) is selected by the optical filter and the mirrors. The fluorescence signal is detected by a photomultiplier. The flow rate of the sample air through the fluorescence chamber is set to  $40 \text{ mL min}^{-1}$  (STP). The sample air is dried with a SICAPENT trap (granulated phosphorus pentoxide drying agent) to eliminate a cross-sensitivity to water vapour.

240 The CO measurements are made at 1 Hz with a precision of 1.5 ppb. The accuracy is given by the precision and systematic errors caused by the drift of the background signal and sensitivity of the instrument, and the uncertainty of the CO calibration standard. The total accuracy amounts to  $1.5 \text{ ppb} \pm 2.4\%$  (Gerbig et al., 1999). The CO instrument has been deployed during many aircraft campaigns using the DLR Falcon (e.g. Huntrieser et al. (2016a, b)) and was used for the first time used on board HALO during the ACRIDICON-CHUVA campaign (Wendisch et al., 2016).

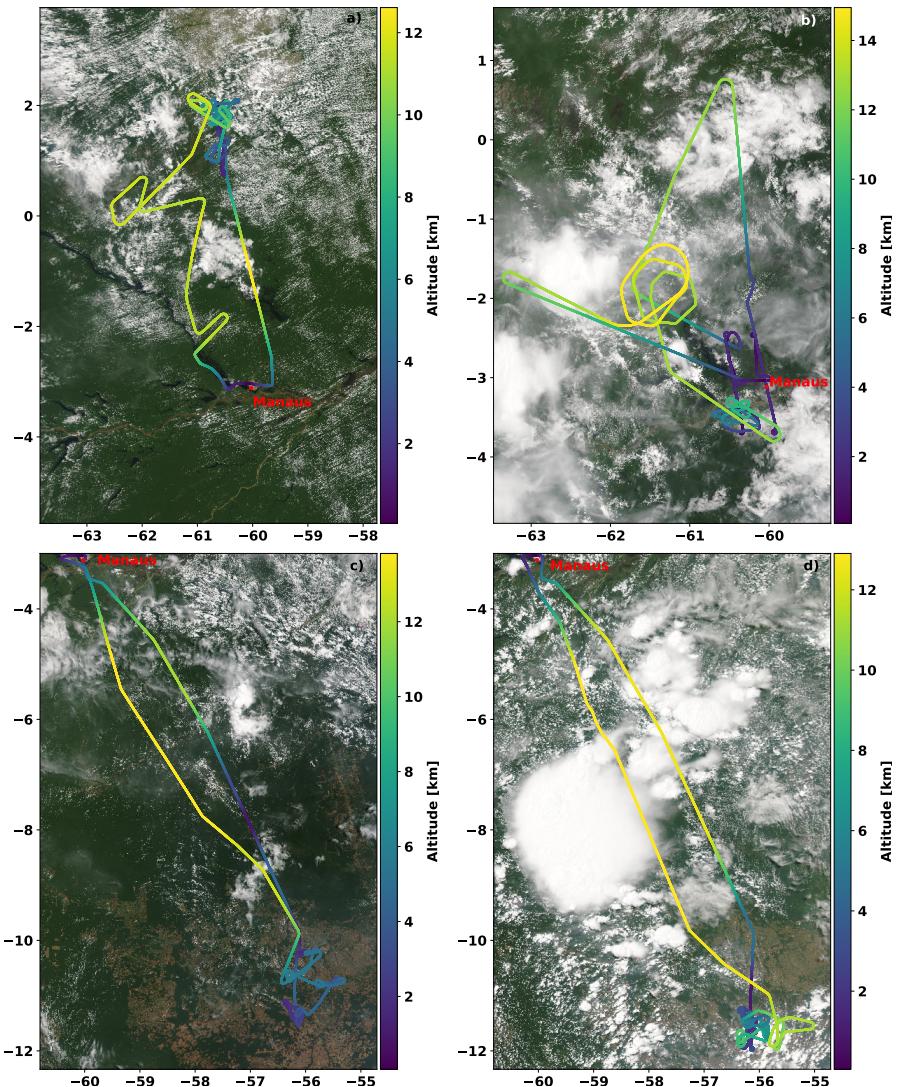
### 3 Measurements

245 The ACRIDICON-CHUVA aircraft campaign took place from the operational basis Manaus (Brazil) within the period September and early October 2014 with a total of 14 scientific flights over the Amazon region. An overview of all ACRIDICON-CHUVA flight tracks can be found in Wendisch et al. (2016), fig. 6. The campaign aimed to improve the understanding of tropical deep convective clouds and their interaction with biogenic and anthropogenic aerosols, and also included precipitation formation and air pollution studies. A detailed description of the campaign objectives, the participating instruments and 250 research groups is given in Wendisch et al. (2016).

Our study requires simultaneous measurements of  $\text{CH}_2\text{O}$ ,  $\text{C}_2\text{H}_2\text{O}_2$ ,  $\text{C}_3\text{H}_4\text{O}_2^*$  and  $\text{O}_4$  (performed by the mini-DOAS spectrometers in the UV and visible spectral ranges as described in sect. 2.1) as well as CO measurements (performed by the AMTEX instrument, see sect. 2.2), but the instruments occasionally failed. Therefore, only for a subset of flights joint data is available. Further, since a part of the flights focused on objectives other than near surface air pollution, here we only present 255 data from four out of the 14 ACRIDICON-CHUVA flights, i.e. those of Sept. 11th (AC09), 16th (AC11), 18th (AC12), and 19th (AC13) in 2014. Nevertheless, the data collected during these flights provides valuable information on the different measured pollutants, their sources, atmospheric transformation and transport in pristine background air in the troposphere as well as fresh and aged air affected by biomass burning in the Amazon.

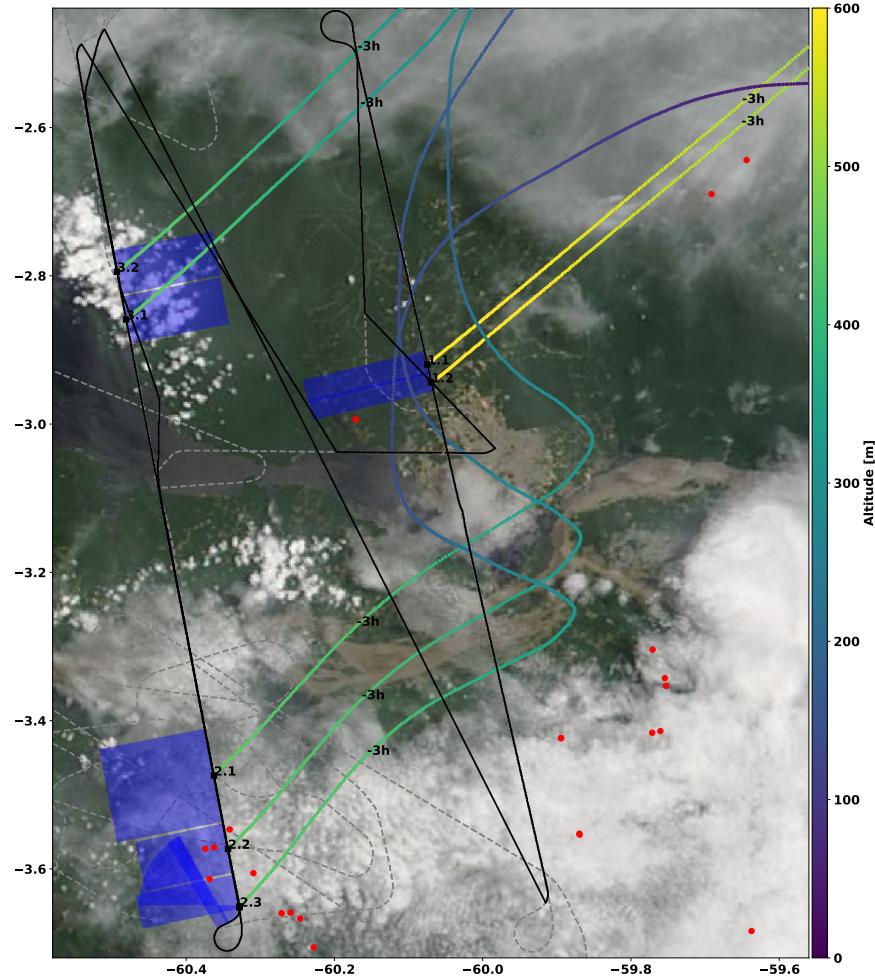
260 Each of these measurement flights took seven to eight hours, typically occurring between 14 UTC and 22 UTC, i.e. during daytime. The long flight time supported the sampling of a large geographical area of the rain forest during the four flights, reaching from approximately  $6^\circ \text{ S}$  to  $8^\circ \text{ N}$ , and  $49^\circ \text{ W}$  to  $63^\circ \text{ W}$ . The detailed flight tracks and respective altitudes are shown in fig. 5. During each of these flights, the probed altitudes ranged from several hundred meters up to 15 km, with flight sections inside the planetary boundary layer overpassing large agricultural areas (AC12 and AC13) as well as the rain forest and the Amazon river delta (AC09 and AC11, fig. 5).

265 Due to the relatively large, but patchy cloud cover over the Amazon, the UV and vis limb radiances varied significantly during the flights. Therefore, typical integration times of the measured co-added sets of skylight spectra range from seconds (under clear sky conditions) to several minutes (during cloudy sky and hazy conditions). In consequence, the along track resolution of the measurements was typically several kilometres in addition to the horizontal resolution perpendicular to the aircraft determined by radiative transfer (sect. 2.1). For the biomass burning events AC11-1 to AC11-3 (Table 3 and sect. 5.4), 270 a two-dimensional visualization of the resulting averaging (or probed) areas per single measurement is indicated by the blue areas in fig. 6. The identification of the atmospheric measurement conditions and in particular of each biomass burning event is



**Figure 5.** Tracks of the ACRIDICON-CHUVA flights (a) on Sept. 11, 2014, (b) on Sept. 16, 2014 (AC11), (c) on Sept. 18, 2014 (AC12), and (d) on Sept. 19, 2014 (AC13). All flights started and ended at the operational base Manaus airport in Brazil (red dot). The color coding indicates the flight altitude. The MODIS satellite images are from 14:10 UTC (Sept. 11, 2014), 14:30 UTC (Sept. 16, 2014), 14:20 UTC (Sept. 18, 2014), and 17:55 UTC (Sept. 19, 2014). The images were taken from NASA WORLDVIEW <https://worldview.earthdata.nasa.gov/>.

based on visual imagery using the recorded video data stream (fig. 7). In the following, only those measurements are claimed to be affected by biomass burning, for which the actual plumes are visible in the recorded images and are in the telescope's line of sight. These events are listed in Table 3. Depending on the flight pattern and the integration time of the spectra, some of these 275 plumes were intercepted more than once. In total, eight distinct plumes were probed during twelve individual measurements, all of which were recorded below 2 km flight altitude.



**Figure 6.** Zoom of the flight track for AC11 on Sept. 16, 2014. The flight tracks below/above 1000 m altitude are labelled by drawn/dashed lines, respectively. The take-off and landing of the aircraft was from Manaus airport ( $3.1^{\circ}$  S,  $60.0^{\circ}$  W) located in the center of the image. The biomass burning events detected by the MCD14 MODIS instrument on the Terra and Aqua satellites are indicated with red dots (not to scale). The biomass events probed by the mini-DOAS instrument are marked by black numbers next to the flight track, according to the labelling of Table 3. The blue rectangles approximately indicate the probed air masses in the horizontal plane, where the distance perpendicular to the flight track is given by the O<sub>4</sub> estimated photon path lengths (14 to 19 km), and the along track distance (3 to 15 km) by the spectrum integration time (33 to 166 s) multiplied by the aircraft ground speed (approximately 95 m s<sup>-1</sup>) (sect. 2.1). For each biomass burning event, 3 h backward air mass trajectories were calculated and plotted color-coded by the altitude of the airmass. The MODIS satellite image was taken from NASA WORLDVIEW, see <https://worldview.earthdata.nasa.gov/>.

The MODIS instrument on board the Terra and Aqua satellites (MCD14, collection 6, available online from <https://worldview.earthdata.nasa.gov/>,

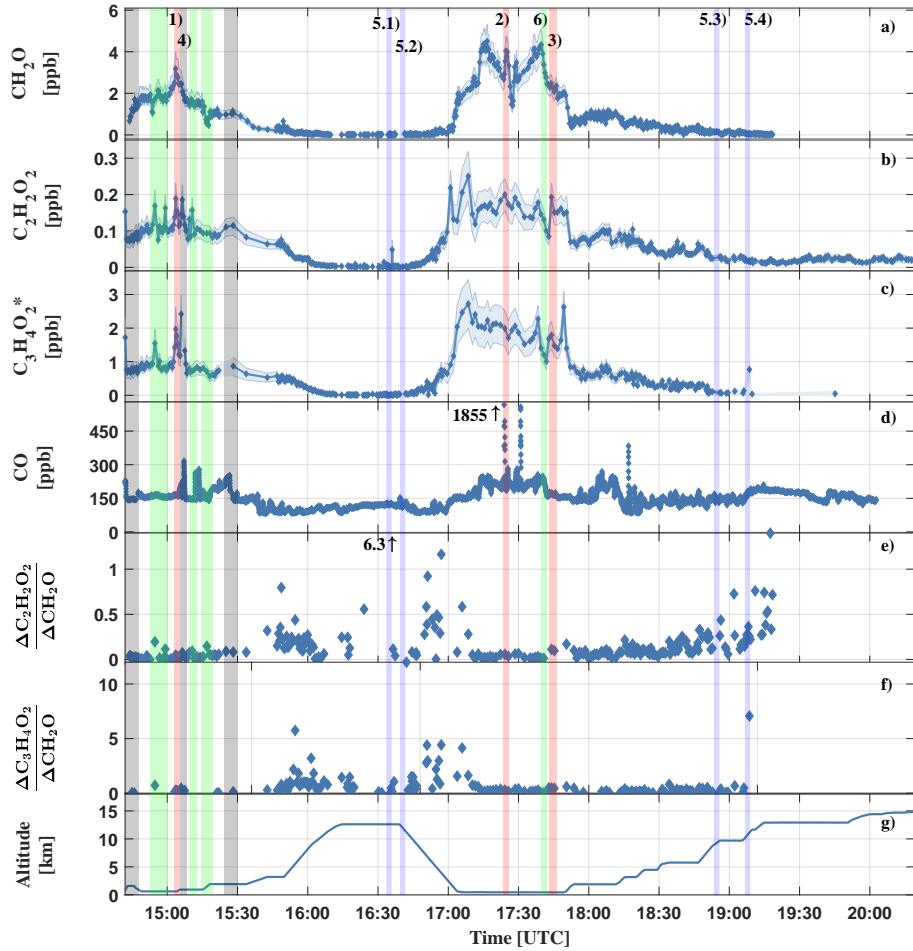
doi:10.5067/FIRMS/MODIS/MCD14DL.NRT.006) reports additional fires in the vicinity of the aircraft trajectory, which were



**Figure 7.** Images taken by the mini-DOAS camera during events 1–6 (as indicated in fig. 6 and 8) for the flight AC11 with the UTC time given in each image. The second image was taken from the respective flight report. The video camera points into the same direction as the three limb telescopes of the mini-DOAS spectrometer, however with a much larger field of view of  $\sim 46^\circ$  width.

280 not directly located within the field of view of the mini-DOAS telescopes, nor within the larger field of view of the camera. Therefore, these events are not further investigated in our analysis. However, the emissions of these fires are likely to have created an atmospheric background generally affected by differently aged biomass burning emissions within the boundary layer (fig. 6). Further insights into the recent photochemical past of the air masses are provided by 3 h backward air mass 285 trajectories calculated using the READY (Real-time Environmental Applications and Display sYstem) website of the NOAA HYSPLIT model (<https://www.ready.noaa.gov/HYSPLIT.php>) (Stein et al., 2015; Rolph et al., 2017). Exemplary 3 h backward trajectories are shown for events AC11-1.2 to AC11-3.2 in fig. 6. They indicate, that prior to detection all air masses resided well within the boundary layer, but did not directly pass over any additional reported fire.

However, all identified biomass burning events show large and correlated enhancements of  $\text{CH}_2\text{O}$ ,  $\text{C}_2\text{H}_2\text{O}_2$ , and  $\text{C}_3\text{H}_4\text{O}_2^*$ , with mixing ratios up to 4.0 ppb, 0.26 ppb, and 2.8 ppb, respectively (fig. 8). Due to the differently sized spatial resolutions of 290 both instruments (sect. 2), the detected peaks in CO and those of  $\text{CH}_2\text{O}$ ,  $\text{C}_2\text{H}_2\text{O}_2$ , and  $\text{C}_3\text{H}_4\text{O}_2^*$  do not (and are not expected to) strictly correlate. Therefore, we refrain from directly estimating emission ratios with respect to CO.



**Figure 8.** Measured mixing ratios of  $\text{CH}_2\text{O}$  (panel a),  $\text{C}_2\text{H}_2\text{O}_2$  (panel b),  $\text{C}_3\text{H}_4\text{O}_2^*$  (panel c), CO (panel d), inferred  $R_{\text{GF}} = \Delta\text{C}_2\text{H}_2\text{O}_2/\Delta\text{CH}_2\text{O}$  (panel e),  $R_{\text{MF}} = \Delta\text{C}_3\text{H}_4\text{O}_2^*/\Delta\text{CH}_2\text{O}$  (panel f), and the height versus time trajectory for the HALO flight AC11 (panel h). The shaded blue area in panels a to c shows the respective measurement uncertainty. All measurements within the Manaus city plume are marked by grey bars (e.g. event AC11-4). Examples for measurements in aged air masses of the upper troposphere are marked in blue (e.g. events AC11-5.1–5.4), and green for the general background atmosphere i.e. tropical forest affected air (e.g. event AC11-6). All measurements of biomass burning plumes are labelled in red (events AC11-1, AC11-2 and AC11-3). The event numbers correspond to the labelling in Table 3 and fig. 7.

#### 4 Results

Figure 8 shows the inferred volume mixing ratios of  $\text{CH}_2\text{O}$  (panel a),  $\text{C}_2\text{H}_2\text{O}_2$  (panel b),  $\text{C}_3\text{H}_4\text{O}_2^*$  (panel c), and CO (panel d) for flight AC11 on Sept. 16, 2014. The measured trace gas mixing ratios are the largest for flight sections within the planetary boundary layer (14:45 to 15:15 UTC and 17:05 to 17:45 UTC) and the lowest in the upper troposphere (16:15 to 16:45 UTC

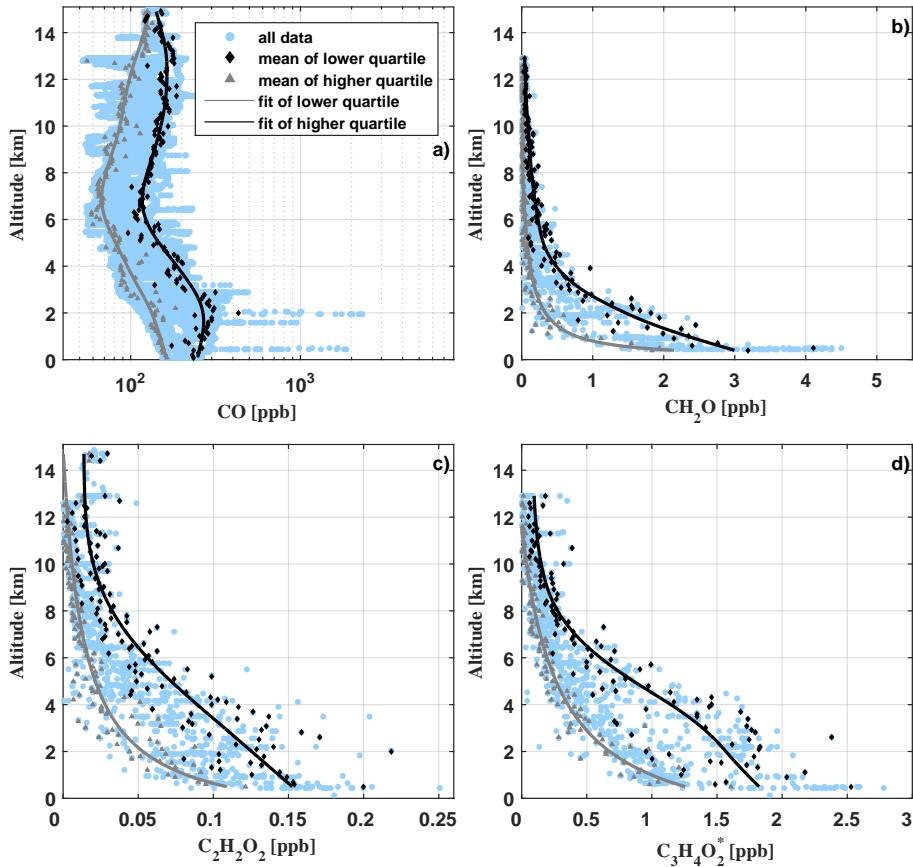
and 19:15 to 20:20 UTC). Notable are the peaks in CO while flying through the Manaus city plume (grey bars) or biomass burning plumes (red bars). Elevated mixing ratios of  $\text{CH}_2\text{O}$ ,  $\text{C}_2\text{H}_2\text{O}_2$ , and  $\text{C}_3\text{H}_4\text{O}_2^*$  can be related to biomass burning as well as to air masses of presumably aged background air (labelled with the green bars) as evidenced by the only moderately elevated CO. Here, the numbers (1) to (6) in panel a of fig. 8 denote biomass burning affected air (1–3), air of the Manaus city plume 300 (4), pristine air of the upper troposphere (5), and boundary layer air above the tropical forest (6). Sky images of all six different situations are shown in fig. 7.

#### 4.1 Vertical profiles

Below 450 m altitude, CO mixing ratios range from 88.6 ppb to 291.8 ppb (fig. 9, panel a), with a mean of  $[\text{CO}] = 188.1$  ppb. Within the planetary boundary layer, CO mixing ratios increase from 88.4 ppb to 2308 ppb (mean  $[\text{CO}] = 212.8$  ppb), with 305 peak mixing ratios above 500 ppb appearing when biomass burning plumes are directly passed. Lowest  $[\text{CO}]$  mixing ratios in the range of 55.1 ppb to 231.9 ppb (mean  $[\text{CO}] = 86.3$  ppb) are found in the middle troposphere between 6 and 8 km altitude. In the upper troposphere (12–14 km), CO increases from 52.5 ppb to 212.6 ppb (mean  $[\text{CO}] = 134.5$  ppb). In the convective tropics, C-shaped CO profiles are a well-known phenomenon, caused by the rapid transport of near surface CO rich air by meso-scale convective systems into the upper troposphere (e.g., Brocchi et al. (2018); Krysztofiak et al. (2018)). When 310 binning the data in 100 m altitude intervals and averaging over the lower and upper data quartiles within each interval, another maximum forms around the top of the boundary layer. Most likely, this suggests that over the Amazon biomass burning plumes often detrain their pollutants near the top of the boundary layer (cf. see for example the image at 17:24 UTC, fig. 7).

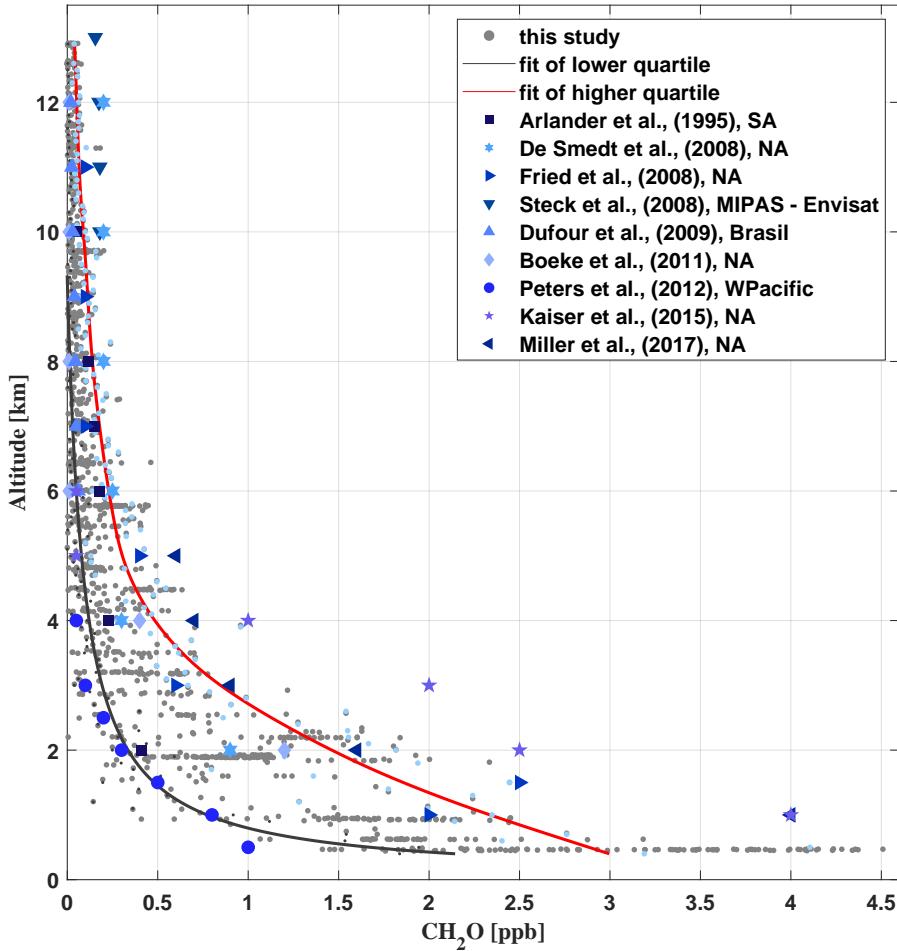
Inferred  $[\text{CH}_2\text{O}]$  ranges from 4.5 ppb near the surface to mean mixing ratios of 30 ppt in the upper troposphere (above 10 km). Mixing ratios within the boundary layer and middle troposphere vary significantly (fig. 9, panel b). In part, this 315 might be a manifestation of the direct emission of  $\text{CH}_2\text{O}$  into the lower part of the atmosphere, or the in situ formation of  $\text{CH}_2\text{O}$  during the oxidation of biogenic VOCs (BVOCs) as well as partly oxidized VOCs (OVOCs). Figure 10 puts our  $\text{CH}_2\text{O}$  measurements over the Amazon into the context of previously published  $\text{CH}_2\text{O}$  observations all over the world. Within this comparison, it is remarkable that the lowest  $\text{CH}_2\text{O}$  mixing ratios measured over the Amazon approximately agree with  $\text{CH}_2\text{O}$  measured within the boundary layer and middle troposphere over the West Pacific by Peters et al. (2012). Our measurements 320 in the upper troposphere compare well to findings from Dufour et al. (2009) and Boeke et al. (2011) over Brazil and North America, respectively. The lowest mixing ratios measured in the middle and upper troposphere may reflect  $\text{CH}_2\text{O}$  formed during the oxidation of mostly  $\text{CH}_4$  and, if available, some residual VOCs, OVOCs, and BVOCs (cf., see fig. 9, panel c for  $\text{C}_2\text{H}_2\text{O}_2$  and panel d for  $\text{C}_3\text{H}_4\text{O}_2^*$ ), since their photochemical and heterogeneous removal is certainly slower than the  $\text{CH}_2\text{O}$  photochemical lifetime of only a few hours (Frost et al., 2002). In contrast, the largest  $\text{CH}_2\text{O}$  mixing ratios measured over 325 the Amazon at any altitude appear slightly smaller than the average, and significantly smaller than the maximum mixing ratios reported from the joint SENEX aircraft observations of  $\text{CH}_2\text{O}$  and  $\text{C}_2\text{H}_2\text{O}_2$  over North America (Kaiser et al. (2015); Chan Miller et al. (2017)).

$\text{C}_2\text{H}_2\text{O}_2$  mixing ratios range from 85 ppt to 250 ppt below 500 m altitude and decrease in the upper troposphere (12–14 km) to mixing ratios from 5 ppt (our detection limit) to about 49 ppt. Close to the ground, our results are comparable



**Figure 9.** Vertical profiles of CO (panel a), CH<sub>2</sub>O (panel b), C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> (panel c), and C<sub>3</sub>H<sub>4</sub>O<sub>2</sub><sup>\*</sup> (panel d) as measured during flights AC09, AC11, AC12, and AC13 (blue circles). Note the logarithmic scale of the x-axis in panel a. For AC09, there are no CO data available. The grey lines indicate a spline fit (panel a) and least squares fits (panel b to d) through the lower quartile of the data (grey triangles) in 100 m altitude bins. The black lines show corresponding fits through the mean of the upper quartile (black diamonds).

330 to C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> mixing ratios found over North America by Kaiser et al. (2015) and Chan Miller et al. (2017), but much larger than those measured in pristine air masses of the South Pacific (7–23 ppt) by Lawson et al. (2015). Evidently, besides the C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> formation during the oxidation of BVOCs (Wennberg et al., 2018), a fraction of the near surface and boundary layer C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> is certainly due to direct emission by biomass burning (Zarzana et al., 2018; Andreae and Merlet, 2001b). In the middle troposphere, our inferred mixing ratios of C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> are significantly larger than observations during SENEX (which 335 reach to approximately 6 km altitude) and are also larger in the upper troposphere above 6 km when comparing to data extrapolated from SENEX. Between 6 and 14 km altitude, C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> enhancements of up to  $\approx 63$  ppt become apparent when compared to either the SENEX data or the binned lower quartile of our data. Such elevated mixing ratios of C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> in the middle and upper troposphere point to an efficient vertical transport of C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> and its precursors VOCs, OVOCs, and BVOCs from their emission sources at the ground or to direct formation from longer-lived VOCs (e.g., phenols) (Andreae et al., 2001;



**Figure 10.** Comparison of different CH<sub>2</sub>O measurements available from the literature (Arlander et al., 1995; De Smedt et al., 2008a; Fried et al., 2008; Dufour et al., 2009; Steck et al., 2008; Boeke et al., 2011; Peters et al., 2012; Kaiser et al., 2015; Chan Miller et al., 2017) with those of the present study (grey). The geographical regions of the measurements are Brazil, South America (SA), North America (NA), West Pacific (WPacific) and MIPAS-Envisat measurements covering orbit 8164, 14° S/46° W.

340 Taraborrelli et al., 2020). Further, C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> (and C<sub>3</sub>H<sub>4</sub>O<sub>2</sub><sup>\*</sup>) present at higher atmospheric altitudes may serve as a marker for the formation of ISOPOO (isoprene peroxy radicals), ISOPOOH (oligomer hydroxyhydroperoxides) and finally IEPOX (isoprene epoxydiols) within the isoprene oxidation chain (Wennberg et al., 2018). In fact, IEPOX mediated SOA formation in the upper troposphere over the Amazon has been reported from observations made within the framework of the ACRIDICON-CHUVA project (Andreae et al., 2018; Schulz et al., 2018).

345 The detection of C<sub>3</sub>H<sub>4</sub>O<sub>2</sub><sup>\*</sup>, here taken as the sum of methylglyoxal and larger carbonyls as described previously, appears elusive due to the spectral interference among the different species. Weighting of the measured total absorption for C<sub>3</sub>H<sub>4</sub>O<sub>2</sub><sup>\*</sup> with the relative absorption cross sections of the inferred C<sub>3</sub>H<sub>4</sub>O<sub>2</sub>, C<sub>4</sub>H<sub>6</sub>O<sub>2</sub> (biacetyl), and C<sub>4</sub>C<sub>8</sub>O<sub>4</sub> (acetylpropionyl) (Zarzana

et al. (2017), fig. S4) may be indicative of the relative abundance of each species. As in Zarzana et al. (2017), we recommend a weighting factor of  $2.0 \pm 0.5$  by which the inferred  $\text{C}_3\text{H}_4\text{O}_2^*$  needs to be divided for indicative  $\text{C}_3\text{H}_4\text{O}_2$  mixing ratios. However, we note that the factor of  $2.0 \pm 0.5$  may largely depend on the precursor concentrations, and thus on the geolocation, altitude, and geophysical regime, and therefore barely provides more than a hint on its true size.

Due to the lack of previous atmospheric  $\text{C}_3\text{H}_4\text{O}_2^*$  (or  $\text{C}_3\text{H}_4\text{O}_2$ ) measurements, it is more difficult to validate our inferred  $\text{C}_3\text{H}_4\text{O}_2^*$  profile (fig. 9, panel d). As for the other measured hydrocarbons, the largest  $\text{C}_3\text{H}_4\text{O}_2^*$  mixing ratios between 1.2 ppb and 2.8 ppb (mean  $[\text{C}_3\text{H}_4\text{O}_2^*] = 1.6$  ppb) are found near the ground and within the planetary boundary layer. Below 1 km altitude, our results of 0.8 ppb to 2.8 ppb (mean  $[\text{C}_3\text{H}_4\text{O}_2^*] = 1.5$  ppb) are considerably larger than recently reported  $\text{C}_3\text{H}_4\text{O}_2$  mixing ratios (28 ppt to 365 ppt) from a Mediterranean site with intense biogenic emissions and low levels of anthropogenic trace gases (Michoud et al., 2018), as well as measurements at Cape Grim ( $28 \pm 11$  ppt) and the Chatham Rise ( $10 \pm 10$  ppt) in pristine marine air (Lawson et al., 2015). Therefore, we expect a major fraction of the enhanced  $\text{C}_3\text{H}_4\text{O}_2^*$  to be related to the oxidation of BVOCs and biomass burning (e.g., Andreae and Merlet (2001a); Akagi et al. (2011); Stockwell et al. (2015); Zarzana et al. (2017, 2018); Andreae and Merlet (2001b)). Evidence that the former is overall the more relevant process (as compared to direct emissions by biomass burning) in the Amazonian troposphere is also provided by the relatively compact clustering of the inferred  $\text{C}_3\text{H}_4\text{O}_2^*$  along the fit of the lower quartile of the data.

## 5 Discussion

### 5.1 Vertical profiles and precursor VOCs

Above the boundary layer, significant enhancements in  $\text{C}_2\text{H}_2\text{O}_2$  and  $\text{C}_3\text{H}_4\text{O}_2^*$  mixing ratios above the inferred backgrounds are observed (fig. 9, panels c and d). After applying a correction factor of  $2.0 \pm 0.5$  to the measured  $\text{C}_3\text{H}_4\text{O}_2^*$  mixing ratios as discussed above, methylglyoxal exceeds the inferred glyoxal mixing ratios by up to a factor of five for all measurements. Because of many long flight sections above the remote rain forest, which were to our knowledge at least partly free of fires and plumes, as well as the comparatively different shape of the formaldehyde profile, not all of these enhancements can be attributed to direct emissions from biomass burning.

As their dominant source, isoprene globally accounts for 67 %, 47 %, and 79 % of the annual sources of formaldehyde, glyoxal, and methylglyoxal (Fu et al., 2008), respectively, leading to their rapidly decreasing vertical profiles. As argued by Fu et al. (2008) and Fu et al. (2019), the oxidation of isoprene under low  $\text{NO}_x$  conditions may be delayed by hours or even several days. This prolonged lifetime combined with the efficient vertical atmospheric transport in the tropics leads to isoprene oxidation above the boundary layer. As a consequence, we suspect the in situ formation of glyoxal and methylglyoxal and finally formaldehyde from isoprene oxidation to contribute significantly to the observed enhanced mixing ratios in the free troposphere. The methylglyoxal enhancements in the middle troposphere are approximately five times larger than the observed glyoxal mixing ratios. Molar yields of glyoxal and methylglyoxal from second and third generation isoprene oxidation combined, are 6.2 % and 34 %, respectively (Fu et al., 2008). Therefore, we expect the in situ formation of methylglyoxal

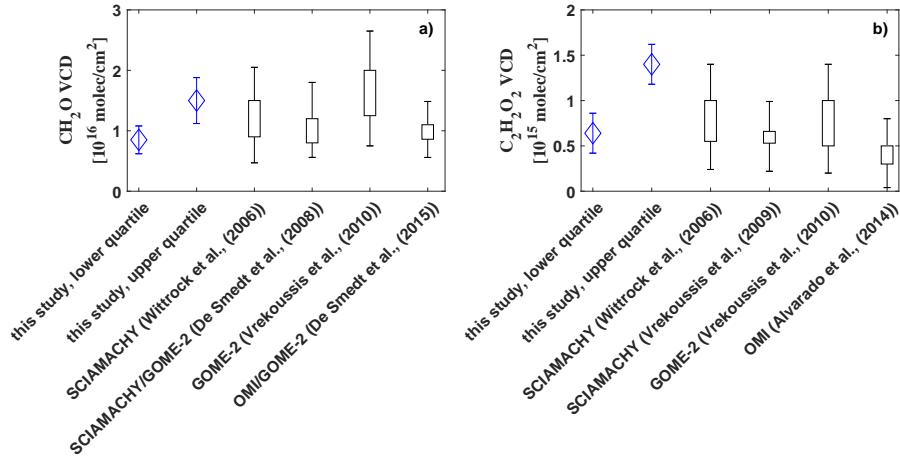
380 from isoprene oxidation to be approximately five times larger than the respective glyoxal production. In accordance with our measurements, this should lead to five times larger methylglyoxal mixing ratios compared to glyoxal in the free troposphere.

385 Besides their common dominant precursor, additional sources of the gases differently influence their local distribution (mostly combustion processes and oxidation of other biogenic/anthropogenic hydrocarbons in the case of formaldehyde (Lee et al., 1998; Liu et al., 2007; Fortems-Cheiney et al., 2012), oxidation of acetylene in the case of glyoxal, and oxidation of acetone in the case of methylglyoxal Fu et al. (2008)), and might therefore differently contribute to our measurements. A recent 390 study additionally discussed the oxidation of aromatics as possible relevant source of atmospheric glyoxal and methylglyoxal (Taraborrelli et al., 2020). While acetylene is mostly an anthropogenically emitted trace gas, acetone additionally has direct biogenic sources. Both gases have a long lifetime of up to 18 and 22 days, respectively (Fu et al., 2008). Therefore, vertically transported acetylene from ground-based combustion processes might be a further source of the observed glyoxal production in 395 the free troposphere. While acetone is also emitted in biomass burning processes (one fifth of the total acetone budget (Pöschl et al., 2001)), an equally large source of acetone emissions is decomposing plant material ( $10^{-4}$  g g<sup>-1</sup> (Warneke et al., 1999)). The remaining acetone is to a large part an oxidation product of hydrocarbons and also an oxidation product of isoprene under 400 low NO<sub>x</sub> conditions, which are typical for air above rain forests (Warneke et al., 2001). Mean acetone mixing ratios in the boundary layer above tropical rain forests are on the order of 2 ppb (Pöschl et al., 2001). Besides isoprene oxidation, a fraction 405 of the observed methylglyoxal enhancements in the free troposphere might therefore be a direct or secondary (through the oxidation of hydroacetone) product of acetone oxidation above the boundary layer or a result of longer-lived oxidized aromatics (Taraborrelli et al., 2020).

410 It is remarkable, that even though formaldehyde, glyoxal, and methylglyoxal have a common dominant source (isoprene) and sink (photolysis), the relative profile shapes are different. Especially above the boundary layer, the inferred formaldehyde mixing ratios decrease faster with increasing altitude than the vertical profiles of glyoxal and methylglyoxal do (fig. 9, panel b). It is thus to question whether the relative contribution of the different precursor gases, their sinks due to photolysis and reaction with OH radicals in a low-NO<sub>x</sub>, high-VOC, and low-HO<sub>x</sub> environment or the different efficiencies regarding the uptake into 415 aerosols or cloud particles may alter the relative profile shapes of the three species. The water solubility of the gases is largely different ( $H^{CP} = 32$  mol m<sup>-3</sup> Pa<sup>-1</sup> for formaldehyde,  $H^{CP} = 4100$  mol m<sup>-3</sup> Pa<sup>-1</sup> for glyoxal, and  $H^{CP} = 370$  mol m<sup>-3</sup> Pa<sup>-1</sup> for methylglyoxal, see Sander (2015)) and significantly smaller for formaldehyde than glyoxal. Heterogeneous uptake may 420 therefore not provide an explanation for the relative depletion of formaldehyde over glyoxal and methylglyoxal in the middle troposphere.

## 5.2 Comparison with satellite measurements

425 As neither CH<sub>2</sub>O nor C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> have relevant stratospheric sources, the integration of the profiles yields their total vertical column density (VCD). The uncertainty of the integrated profiles follows from the altitude weighted total error of our measurements as displayed in fig. 4. We compare our findings to measurements from GOME (Global Ozone Monitoring Experiment), 430 SCIAMACHY (SCanning Imaging Absorption spectrometer for Atmospheric CHartographY), OMI (Ozone Monitoring In-



**Figure 11.** Comparison of the integrated total column densities of  $\text{CH}_2\text{O}$  and  $\text{C}_2\text{H}_2\text{O}_2$ . The vertical bars indicate the range of the observed vertical column amounts as reported in the respective studies. The uncertainty range for Wittrock et al. (2006) and Alvarado et al. (2014) was estimated based on Vrekoussis et al. (2009).

strument), and GOME-2, which combined provide global  $\text{CH}_2\text{O}$  and  $\text{C}_2\text{H}_2\text{O}_2$  observations covering more than a decade (fig. 11).

Integration of the  $\text{CH}_2\text{O}$  profiles yields total column densities of  $(0.9 \pm 0.2) \cdot 10^{16}$  molec cm $^{-2}$  (lower quartile as plotted in fig. 9) to  $(1.5 \pm 0.4) \cdot 10^{16}$  molec cm $^{-2}$  (upper quartile), with a mean of  $(1.1 \pm 0.3) \cdot 10^{16}$  molec cm $^{-2}$  (all data, fig. 11, panel a). Satellite observations generally report enhanced  $\text{CH}_2\text{O}$  VCDs over regions with large biogenic emissions, especially over the Amazon Basin, with maximal enhancements during the dry season. Monthly means of  $\text{CH}_2\text{O}$  VCDs measured by GOME and SCIAMACHY in the years 1996–2007 over the Amazon report maximal enhancements of  $1\text{--}2 \cdot 10^{16}$  molec cm $^{-2}$  during the dry season and much smaller vertical column densities of  $0.8 \cdot 10^{16}$  molec cm $^{-2}$  during the wet season (De Smedt et al., 2008b). Our inferred  $\text{CH}_2\text{O}$  vertical column densities based on the lower quartile agree well with their measurements during the wet season, while our integrated profile of the upper quartile lies within the observations during the dry season. Our results agree equally well with SCIAMACHY observations in 2005 over South America with an annual mean  $\text{CH}_2\text{O}$  VCD of approximately  $1.2 \cdot 10^{16}$  molec cm $^{-2}$  as reported by Wittrock et al. (2006) and also agree with GOME-2 observations, for which Vrekoussis et al. (2010) report  $\text{CH}_2\text{O}$  VCDs of  $1.2$  to  $1.6 \cdot 10^{16}$  molec cm $^{-2}$  over the Amazon Basin for the years 2007 to 2008. Finally, yearly averaged  $\text{CH}_2\text{O}$  from OMI and GOME-2 observations over Brazil between 2007 and 2013 by De Smedt et al. (2015) agree with our mean of  $1.1 \cdot 10^{16}$  molec cm $^{-2}$ .

The integrated  $\text{C}_2\text{H}_2\text{O}_2$  profiles range from  $(0.6 \pm 0.2) \cdot 10^{15}$  molec cm $^{-2}$  (lower quartile) to  $(1.4 \pm 0.4) \cdot 10^{15}$  molec cm $^{-2}$  (upper quartile), with a mean of  $(1.0 \pm 0.3) \cdot 10^{15}$  molec cm $^{-2}$ . Based on SCIAMACHY measurements in the years 2002 to 2007, Vrekoussis et al. (2009) report seasonal mean  $\text{C}_2\text{H}_2\text{O}_2$  VCDs of  $0.5 \cdot 10^{15}$  molec cm $^{-2}$  over northern South America in autumn (fig. 11, panel b). In good agreement with our lower limit, Wittrock et al. (2006) report slightly higher SCIAMACHY  $\text{C}_2\text{H}_2\text{O}_2$  observations of approximately  $0.6\text{--}0.7 \cdot 10^{15}$  molec cm $^{-2}$  over North Brazil for the year 2005. Corresponding to our

mean VCD, Vrekoussis et al. (2010) further report average VCDs of  $0.5 \cdot 10^{15}$  molec cm $^{-2}$  with maximal enhancements of  $1 \cdot 10^{15}$  molec cm $^{-2}$  based on GOME-2 measurements over South America in the years 2007 to 2008. Observations from OMI 435 yield slightly lower monthly mean C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> VCDs of  $0.3 \cdot 10^{15}$  molec cm $^{-2}$  (Alvarado et al., 2014). The VCD based on the upper quartile, which is indicative of the large amount of direct glyoxal emissions from various biomass burnings into the boundary layer, is larger than, but still within the uncertainty range of the SCIAMACHY and GOME-2 observations.

### 5.3 R<sub>XF</sub><sup>\*</sup> above the Amazon rain forest

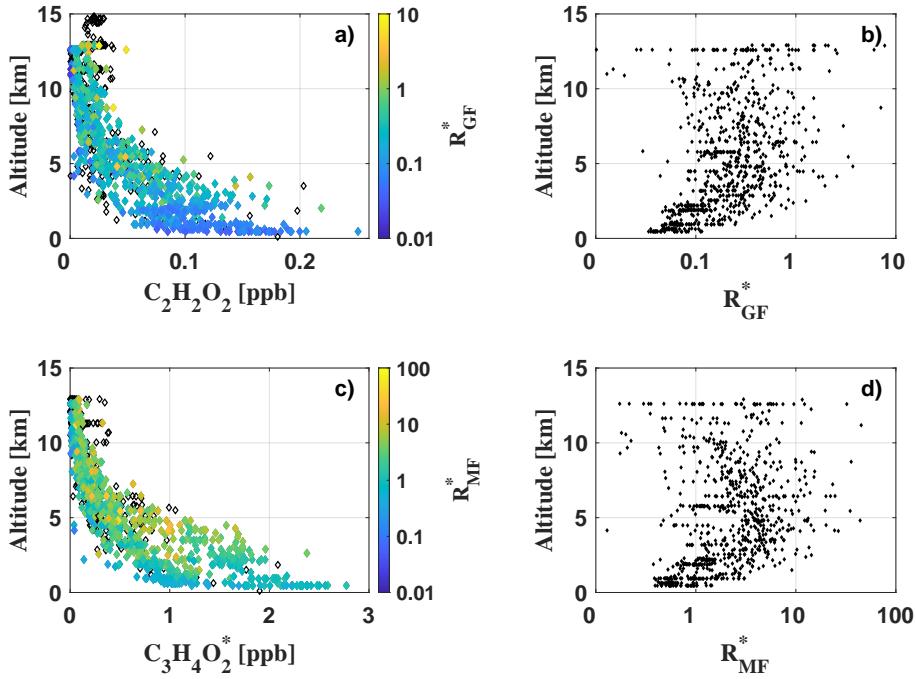
As outlined above, the different air masses probed with the mini-DOAS and AMTEX instrument prevent a definition of emission ratios relative to CO (sect. 2.1 and 2.2). Instead, we use the inferred mixing ratios to calculate the emission ratios R<sub>GF</sub><sup>\*</sup> and R<sub>MF</sub><sup>\*</sup> relative to CH<sub>2</sub>O, as often done in remote sensing studies (Fu et al. (2008); Kaiser et al. (2015); Stavrakou et al. (2016); Zarzana et al. (2017, 2018), and others). According to the literature, the background uncorrected emission ratio is defined as

$$R_{XF}^* = \frac{[X]}{[CH_2O]}. \quad (3)$$

In this study, [X] is either [C<sub>2</sub>H<sub>2</sub>O<sub>2</sub>] or [C<sub>3</sub>H<sub>4</sub>O<sub>2</sub><sup>\*</sup>], as often used in satellite and modelling studies related to hydrocarbon 445 precursors of the studied species (Fu et al. (2008); Kaiser et al. (2015), and others). R<sub>XF</sub><sup>\*</sup> is calculated for each measurement and analyzed with respect to the measurement altitude (fig. 12, panels a and b).

#### 5.3.1 R<sub>GF</sub><sup>\*</sup>

R<sub>GF</sub><sup>\*</sup> mostly remains smaller than 1.0, with a total mean of  $\bar{R}_{GF}^* = 0.35 \pm 0.16$ , but reaches maximum values  $> 5.0$  during several measurements. Small ratios  $< 0.05$  and large ratios  $> 1.0$  can be observed throughout all analyzed flights and nearly all 450 altitudes. The comparison with Kaiser et al. (2015) shows, that the inferred R<sub>GF</sub><sup>\*</sup> is notably larger than during most of their in situ measurements at altitudes ranging from the ground up to 6 km over the southeastern US in June–July 2013. Our results are in much better agreement with the R<sub>GF</sub><sup>\*</sup> inferred by MacDonald et al. (2012) from ground based DOAS measurements for altitudes between 0 and 1000 m over a southeast Asian tropical rainforest in April–July 2008. Their R<sub>GF</sub><sup>\*</sup> in the range 0.2 to 455 0.7 agrees with most of our measurements as well as with  $\bar{R}_{GF}^*$ . From measurements above the Kisatchie National Forest and the Mark Twain National Forest in the south-eastern US, Kaiser et al. (2015) conclude characteristically low R<sub>GF</sub><sup>\*</sup> in pristine regions with strong isoprene emissions, while regimes dominated by monoterpene emissions appear to have higher R<sub>GF</sub><sup>\*</sup>. Most of our measurements took place in pristine air masses above the rain forest and far away from any major population center or other anthropogenic emission sources. Monoterpene and isoprene emissions over the central Amazon were studied in multiple 460 investigations (e.g. Helmig et al. (1998) and Kesselmeier et al. (2000)). All these studies reported generally low mixing ratios of monoterpenes compared to isoprene, which composes 90 % of the total VOC budget in the Amazon (Kuhn et al., 2007). Apparently, emissions of monoterpenes, like alpha-pinene, are at least one order of magnitude smaller than isoprene emissions. Approximately the same relation was found for the total monoterpene emissions (Kuhn et al., 2007; Rizzo et al., 2010). Kaiser et al. (2015) report on approximately 1.2–2.0 times larger C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> yield from  $\alpha$ - and  $\beta$ -pinene oxidation with respect to CH<sub>2</sub>O. Still, the significant predominance of isoprene emissions over the Amazon clearly compensate for this effect. We



**Figure 12.** Vertical profiles of C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> (panel a) and C<sub>3</sub>H<sub>4</sub>O<sub>2</sub><sup>\*</sup> (panel c), color-coded by the ratios R<sub>GF</sub><sup>\*</sup> and R<sub>MF</sub><sup>\*</sup>, respectively, for each measurement during flights AC09, AC11, AC12, and AC13. Black diamonds indicate measurements with missing CH<sub>2</sub>O data (recorded with a different spectrometer) and consequently missing ratios. Panels b and d illustrate the vertical profiles of R<sub>GF</sub><sup>\*</sup> and R<sub>MF</sub><sup>\*</sup> throughout the troposphere. For panels b and d a logarithmic x-axis has been chosen for a better illustration of the altitude dependence of small R<sub>GF</sub><sup>\*</sup> and R<sub>MF</sub><sup>\*</sup>, respectively.

465 conclude that isoprene and not monoterpenes is the dominant precursor of CH<sub>2</sub>O and C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> for our observations. On  
 contrary to the measurements of Kaiser et al. (2015) in isoprene rich regimes, the results indicate significantly elevated R<sub>GF</sub><sup>\*</sup>  
 470 in the troposphere over the Amazon. This is a direct consequence of the much lower CH<sub>2</sub>O mixing ratios as compared to  
 their measurements over the south-eastern US. CH<sub>2</sub>O generally does not exceed 1–2 ppb (except for direct measurements in  
 biomass burning or in the Manaus city plume), which is four times less than reported by Kaiser et al. (2015) for pristine air  
 masses over the Mark Twain National Forest (ca. 8 ppb). Our C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> mixing ratios, on the other hand, are comparable to  
 their findings and typically 100 ppt in the lower troposphere. As a result, we obtain significantly higher R<sub>GF</sub><sup>\*</sup> than Kaiser et al.  
 (2015) for the pristine troposphere.

Based on the SENEX observations, Chan Miller et al. (2017) reported mean R<sub>GF</sub><sup>\*</sup> of 0.024 below 1 km altitude. Interestingly,  
 475 Chan Miller et al. (2017) obtained larger R<sub>GF</sub> of up to 0.06 when NO<sub>x</sub> was very low (0.1–0.5 ppb). Even though we are not  
 able to infer NO<sub>x</sub> concentrations for all measurements due to instrument failures, we were able to measure NO<sub>2</sub> during all  
 analyzed flights. Inferred NO<sub>2</sub> mixing ratios from mini-DOAS measurements were very low during all flights. Mixing ratios  
 above 1 ppb were inferred only in rare cases and exclusively over direct emission sources like Manaus city and in biomass fire

plumes. We conclude from these overall low  $\text{NO}_2$  mixing ratios, that the measurements during the campaign were generally under very low  $\text{NO}_x$  conditions. As described in detail by Chan Miller et al. (2017), this could lead to prompt glyoxal formation 480 through isoprene peroxy radicals and thus slightly enhanced  $R_{\text{GF}}$  compared to observations under high  $\text{NO}_x$ . Further, from our measurements over Manaus city, we cannot detect any sizeable influence of anthropogenic VOCs on the inferred  $R_{\text{GF}}^*$ . During several low overpasses of the city (< 2 km flight altitude),  $R_{\text{GF}}^*$  did not show any significant difference to measurements distant from anthropogenic emission sources. During the measurements in the city plume, the relative increase of  $\text{CH}_2\text{O}$  and  $\text{C}_2\text{H}_2\text{O}_2$  mixing ratios was similar, leading to consistent  $R_{\text{GF}}^*$ .

485 Finally, the vertical profile of  $R_{\text{GF}}^*$  indicates slightly elevated ratios above the boundary layer and in the free troposphere (fig. 12, panel b). Within the boundary layer,  $R_{\text{GF}}^*$  remains approximately constant. Both features were previously observed by Kaiser et al. (2015) for altitudes up to 6 km, however less pronounced. The increase in  $R_{\text{GF}}^*$  between 2–10 km appears most pronounced just above the boundary layer (at about 2 km), where  $\text{CH}_2\text{O}$  and  $\text{C}_2\text{H}_2\text{O}_2$  mixing ratios are still significantly above the detection limits. Notably, the correlation of  $\text{CH}_2\text{O}$  and  $\text{C}_2\text{H}_2\text{O}_2$  is larger within the boundary layer than in the free 490 troposphere. When discussing the profile shape of  $R_{\text{GF}}^*$ , one has to keep in mind the very low  $\text{CH}_2\text{O}$  and  $\text{C}_2\text{H}_2\text{O}_2$  mixing ratios in the upper troposphere and the increasing influence of measurement noise on the inferred mixing ratios. Above 6 km, we observe mean  $\text{CH}_2\text{O}$  and  $\text{C}_2\text{H}_2\text{O}_2$  of only  $54 \pm 40$  ppt and  $15 \pm 5$  ppt, respectively, and accordingly,  $R_{\text{GF}}^*$  varies on average by 60 % among the different measurements within the same altitude range.

### 5.3.2 $R_{\text{MF}}^*$

495 Generally,  $R_{\text{MF}}^*$  appears approximately 10 times larger than  $R_{\text{GF}}^*$  (fig. 12, panels c and d).  $R_{\text{MF}}^* < 1$  are observed only for a minority of the measurements and mostly for background  $\text{C}_3\text{H}_4\text{O}_2^*$  concentrations. Larger  $R_{\text{MF}}^* > 1$  are inferred at all altitudes, with a total mean  $\bar{R}_{\text{MF}}^* = 3.4 \pm 1.6$  and a maximum ratio  $R_{\text{MF},\text{max}}^* = 45.2 \pm 0.8$ . As discussed above for  $R_{\text{GF}}^*$ , the vertical profile of  $R_{\text{MF}}^*$  indicates slightly larger  $R_{\text{MF}}^*$  above the boundary layer and throughout the free troposphere. In the upper troposphere,  $R_{\text{MF}}^*$  decreases again, leading to a concave curvature of the vertical profile for small  $R_{\text{MF}}^*$ , as described above for  $R_{\text{GF}}^*$ . Several 500 studies were published on the emission ratio of  $\text{C}_3\text{H}_4\text{O}_2$  to  $\text{CH}_2\text{O}$  (or CO) in biomass burning plumes (e.g. Hays et al. (2002); Müller et al. (2016); Zarzana et al. (2017, 2018)). To our knowledge, comparable data for air masses of the free troposphere are not yet available.

## 5.4 Normalized excess mixing ratios in biomass burning plumes and inferred emission factors

During the four measurement flights, a total of twelve biomass burning plume intercepts was identified based on video imagery 505 (e.g. fig. 7 for events AC11-1.1/1.2 and AC11-3.1/3.2). Most of the plume measurements occurred south of Manaus, and during two extensive flight periods of AC11 and AC13 at altitudes below 2 km. The precise location of all plume intercepts during flight AC11, and the related horizontal averaging kernels of the telescopes are given in fig. 6. As indicated by the backward trajectories, three hours prior to detection all probed air masses moved well within the boundary layer, below 600 m altitude. This time frame more than covers the atmospheric lifetimes of  $\text{CH}_2\text{O}$ ,  $\text{C}_2\text{H}_2\text{O}_2$  and  $\text{C}_3\text{H}_4\text{O}_2$ . Consequently, the measured 510 mixing ratios of these gases are a result of fresh emissions at the surface. Each biomass burning plume encounter correlates

**Table 3.** Inferred normalized excess mixing ratios and emission factors for individual biomass burning events (range and mean).  $R_{GF}$  and  $R_{MF}$  are given in mole mole $^{-1}$ . The emission factors  $EF_G$  and  $EF_M$  are in units of g kg $^{-1}$ .

event	time interval [UTC]	location	altitude [km]	$R_{GF}$	$R_{MF}$	$EF_G$	$EF_M$
AC11-1.1	15:03:19–15:03:51	2.9° S, 60.1° W	0.624–0.632	0.05 ± 0.03	0.09 ± 0.13	0.23 ± 0.13	0.50 ± 0.76
AC11-1.2	15:03:48–15:04:16	2.9° S, 60.1° W	0.615–0.647	0.03 ± 0.02	0.11 ± 0.16	0.16 ± 0.12	0.65 ± 0.94
AC11-2.1	17:21:06–17:23:31	3.6° S, 60.3° W	0.455–0.471	0.06 ± 0.04	0.86 ± 0.39	0.26 ± 0.19	4.96 ± 2.6
AC11-2.2	17:23:31–17:24:48	3.6° S, 60.3° W	0.453–0.470	0.06 ± 0.04	0.66 ± 0.31	0.26 ± 0.18	3.8 ± 2.04
AC11-2.3	17:24:50–17:26:30	3.6° S, 60.3° W	0.452–0.470	0.03 ± 0.03	0.51 ± 0.26	0.16 ± 0.15	2.93 ± 1.68
AC11-3.1	17:43:27–17:44:41	2.9° S, 60.5° W	0.450–0.456	0.11 ± 0.10	1.50 ± 0.60	0.52 ± 0.47	8.64 ± 4.13
AC11-3.2	17:44:43–17:45:52	2.8° S, 60.5° W	0.449–0.454	0.10 ± 0.11	1.30 ± 0.70	0.45 ± 0.52	7.49 ± 4.49
AC13-1	16:33:41–16:34:20	11.0° S, 56.2° W	0.923–0.934	0.04 ± 0.05	1.20 ± 0.60	0.18 ± 0.23	6.91 ± 3.90
AC13-2	16:34:20–16:34:56	11.1° S, 56.2° W	0.921–0.932	0.08 ± 0.07	1.40 ± 0.70	0.37 ± 0.34	8.06 ± 4.55
AC13-3	16:35:39–16:36:18	11.2° S, 56.2° W	0.918–0.929	0.03 ± 0.04	0.76 ± 0.43	0.13 ± 0.17	4.35 ± 2.73
AC13-4	16:37:01–16:37:43	11.2° S, 56.2° W	0.915–0.932	0.05 ± 0.04	0.81 ± 0.43	0.23 ± 0.19	4.67 ± 2.76
AC13-5	16:52:15–16:53:03	11.2° S, 56.2° W	1.849–1.191	0.02 ± 0.02	0.56 ± 0.27	0.11 ± 0.10	3.25 ± 1.77
Range				0.02–0.11	0.09–1.50	0.11–0.52	0.50–8.64
Mean				0.07 ± 0.05	0.98 ± 0.42	0.25 ± 0.23	4.70 ± 2.70

with significant enhancements in  $CH_2O$ ,  $C_2H_2O_2$  and  $C_3H_4O_2^*$  over the background by 13 to 400 %. Based on the backward trajectories combined with video imagery and the short lifetimes of the gases, contributions from other major emission sources (e.g. Manaus city) can largely be excluded. The exact timing, location and altitude of each plume intercept is given in Table 3.

As indicated by the numbering in column 1, some of the plumes were probed more than once, giving a total of eight different 515 biomass burning plumes with 12 encounters (further on called events AC11-1.1 to AC13-5). Due to the lack of respective  $C_2H_2O_2$  and  $C_3H_4O_2$  measurements, we infer mean background mixing ratios  $[X]_{bkg}$  for all three gases by binning the data in 100 m altitude stacks and calculating the mean of the lower data quartile for each bin as displayed in fig. 9 (grey line). As fig. 10 shows for the case of formaldehyde, the such defined background mixing ratios approximately correspond to formaldehyde measurements of pristine air masses above the Western Pacific Ocean by Peters et al. (2012). In order to detect enhancements 520 due to the plumes,  $[X]_{bkg}$  is then subtracted from the measured mixing ratios. From the resulting enhancements, the normalized excess mixing ratio  $R_{XF}$  (or NEMR) is calculated according to

$$R_{XF} = \frac{\Delta[X]}{\Delta[CH_2O]} = \frac{[X_{fire}] - [X_{bkg}]}{[CH_2O_{fire}] - [CH_2O_{bkg}]} \quad (4)$$

with  $X$  being either  $C_2H_2O_2$  or  $C_3H_4O_2^*$ . The results for each individual biomass burning event as well as the mean and range of  $R_{GF}$  and  $R_{MF}$  are given in Table 3.

525 **5.4.1 R<sub>GF</sub>**

R<sub>GF</sub> ranges between 0.02 and 0.11, with an average of  $\bar{R}_{GF}=0.07 \pm 0.05$ . Repeated measurements of the same biomass burning plumes all agree well within the error, indicating the respective plumes were in fact the major emission sources throughout the measurements. The plume AC11-3 (fig. 7) yields the highest R<sub>GF</sub> of 0.10 and 0.11. Unfortunately, the plume was not reported in the MODIS fire database. Therefore, we do not know the precise location of the fire and cannot discuss the 530 results with respect to the distance of the aircraft from the plume. The smallest R<sub>GF</sub> of 0.02 is found during a measurement with an approximately 1000 m higher flight altitude than during the rest of the events.

Our results are consistent with recent laboratory studies on biomass burning emissions by Zarzana et al. (2018) as well as different field measurements and satellite observations by Chan Miller et al. (2014, 2017); DiGangi et al. (2012); Wittrock et al. (2006) and Zarzana et al. (2017). From laboratory measurements, Zarzana et al. (2018) reported an average R<sub>GF</sub> of 535 0.068 for fresh emissions and different kinds of fuels. Without further knowledge on the fuels burned during the observed events, we cannot make firm conclusions on the photochemistry within the fire plumes. While different fuel types may lead to changing emission factors of glyoxal, Zarzana et al. (2018) observed only little variance in R<sub>GF</sub> for different burning species. The correlation of C<sub>2</sub>H<sub>2</sub>O<sub>2</sub> and CH<sub>2</sub>O seems to be consistent for different kinds of fuel. The lack of knowledge of the exact 540 fuels burned in each fire is therefore not an issue when comparing R<sub>GF</sub> of the different plumes and intercepts. In two older studies, Hays et al. (2002) and McDonald et al. (2000b) report much higher R<sub>GF</sub> on the order of approximately 2.5 to 3 for Palmae and Poaceae (Hays et al., 2002) and even up to 4 (McDonald et al., 2000b). Our inferred R<sub>GF</sub> are at least one order of magnitude lower. In fact, results larger than 1 are only observed for the background-uncorrected R<sub>GF</sub><sup>\*</sup> in the free troposphere.

During recent field measurements over the US, Zarzana et al. (2017) report R<sub>GF</sub> in the range 0.008 to 0.110. Due to the 545 very nature of air-borne plume measurements, we do not know by how much the measurements differ with respect to plume age, air mass origin, atmospheric composition, measurement distance to the plume, etc. Despite the very different atmospheric backgrounds and conditions, our measurements over the rain forest agree well with their findings.

Measurements of an aged biomass burning plume yield slightly smaller R<sub>GF</sub> of 0.02 to 0.03 (DiGangi et al., 2012). Of all 550 events, event AC13-5 is most likely the oldest plume measured. This is a consequence of the high flight altitude in combination with the distance from the aircraft to the plume during the measurement. AC13-5 has the smallest R<sub>GF</sub> of only 0.02, which agrees with the findings of DiGangi et al. (2012). We cannot state by how much this plume measurement is influenced by the general atmospheric background. The air masses in the boundary layer are generally highly polluted with biomass burning emission of different ages. We do not know how much these differently aged emissions influenced our individual plume measurements. Therefore, we cannot draw a distinct relationship between R<sub>GF</sub> and the plume ages, which, due to the well-mixed condition of the boundary layer most probably varied between several minutes to hours during all measurements.

555 **5.4.2 R<sub>MF</sub>**

R<sub>MF</sub> ranges between 0.09 and 1.50, with an average of  $\bar{R}_{MF}=(0.98 \pm 0.42)$ . Three of the eight probed plumes yield R<sub>MF</sub> > 1 (both measurements of AC11-3, AC13-1, and AC13-2). All the other plumes yield R<sub>MF</sub> < 1, in the range of 0.09 to 0.86.

For event AC11-1,  $R_{GF}$  is comparable to the other results, while  $R_{MF}$  is significantly smaller than the average  $\bar{R}_{MF}$  by approximately a factor of 10. Based on the video imagery, the size of the observed plumes as well as the estimated distance 560 to the aircraft are comparable during events AC11-1 and AC11-3 (fig. 7). Still, they yield very different normalized excess mixing ratios. In fact, the largest  $R_{MF}$  and  $R_{GF}$  are obtained during event AC11-3. Accordingly, the emission of  $C_2H_2O_2$  and  $C_3H_4O_2^*$  with respect to  $CH_2O$  is the highest within this biomass burning plume. Interestingly, this is not connected to the size and quantity of the biomass burning events monitored by the telescopes. The highest emissions (i.e. the largest plumes) were 565 observed during event AC11-2, where three different large fires were simultaneously detected in a generally hazy atmosphere rich in fresh emissions (fig. 7). These plumes were also recorded by MCD14 on MODIS (fig. 6, red dots in the field of view of event 2.2). Laboratory measurements report  $R_{MF}$  approximately between 1 and 2 for different kinds of fuels (Hays et al., 2002). This is the upper range of our observations. Instead of  $R_{MF}$ , Zarzana et al. (2018) report the ratio  $C_3H_4O_2$  to  $C_2H_2O_2$  570 ( $R_{MG}$ ) to be on the order of 1.7–2.5 for burning rice straw and different kinds of canopy. Our measurements yield  $R_{MG}$  in the range 1–3, with the exception of event AC11-1, where  $R_{MG} < 1$  due to the small  $C_3H_4O_2^*$  mixing ratios. Event AC11-3 yields  $R_{MG}=1$  during both measurements. This reflects the correlated strong enhancement of both gases within the biomass burning plumes.

#### 5.4.3 Biomass burning emission factors

According to Andreae and Merlet (2001b), the biomass burning emission factors for  $C_2H_2O_2$  and  $C_3H_4O_2^*$  are defined by the normalized excess mixing ratio  $R_{XF}$  as

$$575 \quad EF_X = R_{XF} \frac{MW_X}{MW_{CH_2O}} EF_{CH_2O} \quad (5)$$

using the molecular weight MW of each species, and the mean emission factor  $EF_{CH_2O}$  of the reference trace gas  $CH_2O$ . As the latter was not measured during the campaign, we use the recent comprehensive compilation by Andreae and Merlet (2001b), which reports  $EF_{CH_2O}$  with respect to different combustion processes and fuel types. For tropical forest fires, Andreae and Merlet (2001b) inferred mean  $EF_{CH_2O}=2.4 \pm 0.63$ . The resulting  $EF_G$  and  $EF_M$  are listed in Table 3 for each biomass burning 580 plume intercept in units of g (target species)  $kg^{-1}$  (fuel).

$EF_G$  ranges from 0.11 to 0.52 g glyoxal per kg of fuel burned with a mean emission factor of  $(0.25 \pm 0.23) \text{ g kg}^{-1}$ . According to its high  $R_{GF}$ , event AC11-3 yields the highest emission factors of 0.52 and  $0.45 \text{ g kg}^{-1}$ . Five out of the twelve plume encounters yield emission factors of less than  $0.2 \text{ g kg}^{-1}$ . Corresponding to the largest  $EF_G$ , Andreae and Merlet (2001b) estimated  $EF_G=0.6 \text{ g kg}^{-1}$  for tropical forest fires. In the same study, glyoxal emissions from open burns of agricultural residues 585 were estimated to be approximately 60 % lower with  $EF_G=0.23 \text{ g kg}^{-1}$ . All biomass burning events encountered during flight AC13 were located between  $11.0\text{--}11.2^\circ \text{ S}$  and  $56.2^\circ \text{ W}$ . While other parts of the measurement flight were located over rain forest, this region is largely dominated by agricultural activities.  $EF_G$  of events AC13-1 to AC13-5 ranges from 0.11 to  $0.37 \text{ g kg}^{-1}$ . Assuming that the biomass burning plumes were dominated by agricultural residues during these measurements, we infer a corresponding mean emission factor  $EF_G=(0.25 \pm 0.23) \text{ g glyoxal per kg of open burns of agricultural residues}$ .

590 The range of  $EF_G$  further agrees with laboratory measurements by Zarzana et al. (2018), who reported  $EF_G$  in the range of 0.06 to  $0.55 \text{ g kg}^{-1}$  depending on the fuel type.

595  $EF_M$  ranges between 0.5 and  $8.64 \text{ g methylglyoxal per kg of fuel burned}$ , with a mean emission factor of  $(4.68 \pm 2.7) \text{ g kg}^{-1}$ . For each biomass burning event,  $EF_M$  is at least twice as high as  $EF_G$ . The largest  $EF_M$  were found during flight AC13, i.e. for fires dominated by agricultural residues. Even after applying the correction factor of 2 on the inferred  $EF_M$ , the results are significantly larger than reported by Zarzana et al. (2018) from laboratory measurements, who found maximal emission factors of approximately  $2 \text{ g kg}^{-1}$  for burns of duff. While their absolute  $EF_M$  are much lower, their results were equally variable with  $EF_M$  ranging from 0.11 to  $2.0 \text{ g kg}^{-1}$ .

## 6 Conclusions

600 We report on the first simultaneous measurements of  $\text{CH}_2\text{O}$ ,  $\text{C}_2\text{H}_2\text{O}_2$  and  $\text{C}_3\text{H}_4\text{O}_2^*$  over remote and polluted sections of the Amazon. The measurements were performed from on board the DLR HALO aircraft during the ACRIDICON-CHUVA campaign in fall 2014 (Wendisch et al., 2016). The observations took place in the troposphere between 500 m and 15 km altitude, where air masses of different compositions were encountered. These include continental background as well as aged and fresh biomass burning plumes.

605 The observed formaldehyde mixing ratios in the lower and free troposphere range from those previously reported over the remote Pacific (Peters et al., 2012) up to those measured over continental North and South America (Kaiser et al., 2015). The lower quartile of the measured mixing ratios appears to result from the oxidation of methane and to a lesser degree of VOCs (mostly isoprene). In the lower troposphere, enhanced formaldehyde mixing ratios have two major contributions: direct emission e.g. from biomass burning, and secondary formation during the degradation of short-lived VOCs, like isoprene. In the middle and upper troposphere, formaldehyde seems to have more constant sources and sinks than in the lower troposphere. 610 While the sink (mostly photolysis) may not vary significantly with altitude, the degradation of longer lived VOCs provides a comparably constant source. When integrating our profiles, our observations are in good agreement with previously inferred formaldehyde VCDs from satellites (De Smedt et al., 2008b; Wittrock et al., 2006; Vrekoussis et al., 2010; De Smedt et al., 2015).

615 A good agreement with previous satellite measurements is also found for the lower and upper range of profile integrated glyoxal mixing ratios (Wittrock et al., 2006; Vrekoussis et al., 2009, 2010; Alvarado et al., 2014). Analogously to the presence of formaldehyde, the lower bound of glyoxal is due to the oxidation of precursor VOCs. Besides direct emissions from biomass burning and secondarily formed glyoxal from short lived precursors, the oxidation of longer lived precursors like isoprene or acetone contributes to elevated glyoxal mixing ratios in the free troposphere. Taraborrelli et al. (2020) report in a recent study on long-lived VOCs like aromatics as a potential additional source of glyoxal in the free troposphere on a regional scale. Recent 620 findings by Alvarado et al. (2020) reveal long range transport of glyoxal and formaldehyde within biomass burning plumes reaching distances of 1500 km downwind and thus enlarging the assumed local influence of biomass burning events on the

tropospheric distribution of both gases. The time needed to reach such distances is in the order of several days, which implies the formation of glyoxal and formaldehyde from longer-lived precursors within the plumes.

In agreement with the study of Zarzana et al. (2017),  $C_3H_4O_2^*$  mixing ratios generally exceed those of glyoxal by at least a factor of 5 during the majority of the measurements. Yet, the shapes of the upper and lower bounds appear similar to the profile shape of glyoxal in contrast to formaldehyde, which appears to decline faster above the boundary layer in the low- $NO_x$ , high-VOC, and low- $HO_x$  environment probed during this study. The compact clustering as well as the smooth decrease of the lower quartile mixing ratios with altitude indicate the type and sink of the precursor molecules. Apparently, these precursors should have lifetimes of at least several days, and should either be constantly emitted by the biosphere (e.g. isoprene and acetone) or occasionally during biomass burning (e.g. acetylene, aromatics) before being vertically transported.  $R_{GF}^*$  and  $R_{MF}^*$  are found to increase with altitude due to the more rapid decrease of formaldehyde relative to glyoxal and methylglyoxal mixing ratios. Again, this behavior expresses the different and changing fractions of the short and long-lived precursors relevant for the formation of formaldehyde as compared to those of glyoxal and methylglyoxal. In particular, the increasing  $R_{GF}$  and  $R_{MF}$  in the middle and upper troposphere require significant sources for both gases which are supposed to result from the oxidation of longer-lived VOCs (Schulz et al., 2018). In fact, since both gases are known to contribute to SOA formation, the elevated mixing ratios of glyoxal and methylglyoxal in the upper troposphere further lend support to the proposed SOA formation from products of isoprene oxidation as observed by Schulz et al. (2018); Andreae et al. (2018) and Williamson et al. (2019) over the Amazon Basin and generally in the tropics.

Our inferred  $R_{GF}$  in the range of 2.4–11.2 ( $\bar{R}_{GF}=6.5$ ) for the observed biomass burning events agree well with previous studies, both in the field over North America (Zarzana et al., 2017) as well as in the laboratory fuel study (Zarzana et al., 2018). It is further in agreement with  $R_{GF}$  inferred from satellite measurements (Chan Miller et al., 2014, 2017; DiGangi et al., 2012; Wittrock et al., 2006). For  $R_{MF}$ , we infer a range of 0.09 to 1.5 with  $\bar{R}_{MF}=0.98$  (or respectively smaller results when correcting the  $C_3H_4O_2^*$  mixing ratios by a factor of  $2.0 \pm 0.5$ ), which overlaps with the lower end of the range of normalized excess mixing ratios previously reported by Hays et al. (2002) for different types of fuels in the laboratory. The presented study as well as the measurements by Hays et al. (2002) indicate a slightly larger  $R_{MF}$  than estimated from  $R_{MG}$  and  $R_{GF}$  (appr. 0.1–0.5) reported by Zarzana et al. (2018), even though this difference may in part be a result of additional dicarbonyls included in our retrieval. In fact, when correcting  $R_{MF}$  by a factor of 2, our results agree much better to the measurements by Zarzana et al. (2018). Based on  $R_{GF}$  and  $R_{MF}$  and average emission factors for formaldehyde estimated by Andreae and Merlet (2001b), we infer emission factors in the range of 0.11–0.52 ( $\bar{EF}_G=(0.25 \pm 0.23) \text{ g kg}^{-1}$ ) for glyoxal and 0.5–8.6 ( $\bar{EF}_M=(4.7 \pm 2.7) \text{ g kg}^{-1}$ ) for methylglyoxal for the probed biomass burning events over the Amazon. Our inferred  $EF_G$  agrees well with the range reported from laboratory measurements (Zarzana et al., 2018). Notably, our  $EF_M$  is significantly larger than found in their study, also when correcting the results for possibly included additional dicarbonyls.

The presented study may be complemented by future studies with data collected during several past HALO missions, i.e. ESMVal (fall 2012), OMO (over the Mediterranean in summer 2014), EMeRGe (Europe in summer 2017 and Eastern Asia in spring 2018), CAFE (over West Africa/tropical Atlantic in summer 2018), and Southtrac (Southern Argentina and northern Antarctica in fall 2019), which may provide further information on the role of carbonyls in global atmospheric chemistry.

*Data availability.* All the data are archived in the HALO data depository (<https://halo-db.pa.op.dlr.de/mission/5>), datasets 6495-6502 (mini-DOAS data), and 3440-3442 (AMTEX CO data). The data are accessible upon signing a data protocol.

*Author contributions.* TH, MK and KP operated the mini-DOAS instrument and ML and HS performed the AMTEX measurements during the ACRIDICON-CHUVA campaign. TH wrote the data retrieval software. FK performed the data analysis and wrote the paper with contributions from ML, KP and HS.

*Competing interests.* The authors declare that they have no conflict of interest.

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## References

Akagi, S. K., Yokelson, R. J., Wiedinmyer, C., Alvarado, M. J., Reid, J. S., Karl, T., Crounse, J. D., and Wennberg, P. O.: Emission factors for open and domestic biomass burning for use in atmospheric models, *Atmospheric Chemistry and Physics*, 11, 4039–4072, 680 <https://doi.org/10.5194/acp-11-4039-2011>, <https://www.atmos-chem-phys.net/11/4039/2011/>, 2011.

Alvarado, L., Richter, A., Vrekoussis, M., Wittrock, F., Hilboll, A., Schreier, S., and Burrows, J.: An improved glyoxal retrieval from OMI measurements, *Atmospheric Measurement Techniques*, 7, 4133, 2014.

Alvarado, L. M., Richter, A., Vrekoussis, M., Hilboll, A., Hedegaard, A. B. K., Schneising, O., and Burrows, J. P.: Unexpected long-range 685 transport of glyoxal and formaldehyde observed from the Copernicus Sentinel-5 Precursor satellite during the 2018 Canadian wildfires, *Atmospheric Chemistry and Physics*, 20, 2057–2057, 2020.

Amiridis, V., Marinou, E., Tsekeli, A., Wandinger, U., Schwarz, A., Giannakaki, E., Mamouri, R., Kokkalis, P., Binietoglou, I., Solomos, S., Herekakis, T., Kazadzis, S., Gerasopoulos, E., Proestakis, E., Kottas, M., Balis, D., Papayannis, A., Kountou, C., Kourtidis, K., Paganianopoulos, N., Mona, L., Pappalardo, G., Le Rille, O., and Ansmann, A.: LIVAS: a 3-D multi-wavelength aerosol/cloud database 690 based on CALIPSO and EARLINET, *Atmospheric Chemistry and Physics*, 15, 7127–7153, <https://doi.org/10.5194/acp-15-7127-2015>, <https://www.atmos-chem-phys.net/15/7127/2015/>, 2015.

Andreae, M. and Merlet, P.: Emission of trace gases and aerosols from biomass burning, *Global Biogeochemical Cycles*, 15, 955–96, 695 <https://doi.org/10.1029/2000GB001382>, <https://agupubs.onlinelibrary.wiley.com/doi/abs/10.1029/2000GB001382>, 2001a.

Andreae, M. O. and Merlet, P.: Emission of trace gases and aerosols from biomass burning, *Global biogeochemical cycles*, 15, 955–966, 700 2001b.

Andreae, M. O., Artaxo, P., Fischer, H., Freitas, S. R., Grégoire, J.-M., Hansel, A., Hoor, P., Kormann, R., Krejci, R., Lange, L., Lelieveld, J., Lindner, W., Longo, K., Peters, W., de Reus, M., Scheeren, B., Silva Dias, M. A. F., Ström, J., van Velthoven, P. F. J., and Williams, J.: Transport of biomass burning smoke to the upper troposphere by deep convection in the equatorial region, *Geophysical Research Letters*, 28, 951–954, <https://doi.org/10.1029/2000GL012391>, <https://agupubs.onlinelibrary.wiley.com/doi/abs/10.1029/2000GL012391>, 2001.

Andreae, M. O., Afchine, A., Albrecht, R., Holanda, B. A., Artaxo, P., Barbosa, H. M., Borrman, S., Cecchini, M. A., Costa, A., Dollner, 705 M., et al.: Aerosol characteristics and particle production in the upper troposphere over the Amazon Basin, *Atmospheric Chemistry and Physics*, 18, 921–961, 2018.

Arlander, D. W., Brüning, D., Schmidt, U., and Ehhalt, D. H.: The tropospheric distribution of formaldehyde during TROPOZ II, *Journal of Atmospheric Chemistry*, 22, 251–269, <https://doi.org/10.1007/BF00696637>, <https://doi.org/10.1007/BF00696637>, 1995.

Bauwens, M., Stavrakou, T., Müller, J.-F., De Smedt, I., Van Roozendael, M., van der Werf, G. R., Wiedinmyer, C., Kaiser, J. W., Sindelarova, K., and Guenther, A.: Nine years of global hydrocarbon emissions based on source inversion of OMI formaldehyde observations, 710 *Atmospheric Chemistry and Physics*, 16, 10 133–10 158, <https://doi.org/10.5194/acp-16-10133-2016>, <https://www.atmos-chem-phys.net/16/10133/2016/>, 2016.

Boeke, N. L., Marshall, J. D., Alvarez, S., Chance, K. V., Fried, A., Kuros, T. P., Rappenglück, B., Richter, D., Walega, J., Weibring, P., and Millet, D. B.: Formaldehyde columns from the Ozone Monitoring Instrument: Urban versus background levels and evaluation using aircraft data and a global model, *Journal of Geophysical Research: Atmospheres*, 116, <https://doi.org/10.1029/2010JD014870>, <https://agupubs.onlinelibrary.wiley.com/doi/abs/10.1029/2010JD014870>, 2011.

Borbon, A., Ruiz, M., Bechara, J., Aumont, B., Chong, M., Huntrieser, H., Mari, C., Reeves, C. E., Scialom, G., Hamburger, T., Stark, H., Afif, C., Jambert, C., Mills, G., Schlager, H., and Perros, P. E.: Transport and chemistry of formaldehyde by mesoscale convective systems in West Africa during AMMA 2006, *Journal of Geophysical Research: Atmospheres*, 117, , 2012.

715 Brocchi, V., Krysztofiak, G., Catoire, V., Guth, J., Marecal, V., Zbinden, R., Amraoui, L. E., Dulac, F., and Ricaud, P.: Intercontinental transport of biomass burning pollutants over the Mediterranean Basin during the summer 2014 ChArMEx-GLAM airborne campaign, *Atmospheric Chemistry and Physics*, 18, 6887–6906, 2018.

Burrows, J., Dehn, A., Deters, B., Himmelmann, S., Richter, A., Voigt, S., and Orphal, J.: Atmospheric remote-sensing reference data from GOME: Part 1. Temperature-dependent absorption cross-sections of NO<sub>2</sub> in the 231–794 nm range, *Journal of Quantitative Spectroscopy and Radiative Transfer*, 60, 1025–1031, 1998.

720 Carlier, P., Hannachi, H., and Mouvier, G.: The chemistry of carbonyl compounds in the atmosphere—A review, *Atmospheric Environment* (1967), 20, 2079 – 2099, [https://doi.org/https://doi.org/10.1016/0004-6981\(86\)90304-5](https://doi.org/https://doi.org/10.1016/0004-6981(86)90304-5), <http://www.sciencedirect.com/science/article/pii/0004698186903045>, 1986.

725 Chan Miller, C., Gonzalez Abad, G., Wang, H., Liu, X., Kurosu, T., Jacob, D. J., and Chance, K.: Glyoxal retrieval from the Ozone Monitoring Instrument, *Atmospheric Measurement Techniques*, 7, 3891–3907, <https://doi.org/10.5194/amt-7-3891-2014>, <https://www.atmos-meas-tech.net/7/3891/2014/>, 2014.

730 Chan Miller, C., Jacob, D. J., Marais, E. A., Yu, K., Travis, K. R., Kim, P. S., Fisher, J. A., Zhu, L., Wolfe, G. M., Hanisco, T. F., Keutsch, F. N., Kaiser, J., Min, K.-E., Brown, S. S., Washenfelder, R. A., González Abad, G., and Chance, K.: Glyoxal yield from isoprene oxidation and relation to formaldehyde: chemical mechanism, constraints from SENEX aircraft observations, and interpretation of OMI satellite data, *Atmospheric Chemistry and Physics*, 17, 8725–8738, <https://doi.org/10.5194/acp-17-8725-2017>, <https://www.atmos-chem-phys.net/17/8725/2017/>, 2017.

735 Chance, K. and Orphal, J.: Revised ultraviolet absorption cross sections of H<sub>2</sub>CO for the HITRAN database, *Journal of Quantitative Spectroscopy and Radiative Transfer*, 112, 1509–1510, 2011.

Crutzen, P. J. and Andreae, M. O.: Biomass Burning in the Tropics: Impact on Atmospheric Chemistry and Biogeochemical Cycles, *Science*, 250, 1669–1678, <https://doi.org/10.1126/science.250.4988.1669>, <http://science.sciencemag.org/content/250/4988/1669>, 1990.

740 De Smedt, I., Müller, J.-F., Stavrakou, T., van der A, R., Eskes, H., and Van Roozendael, M.: Twelve years of global observations of formaldehyde in the troposphere using GOME and SCIAMACHY sensors, *Atmospheric Chemistry and Physics*, 8, 4947–4963, <https://doi.org/10.5194/acp-8-4947-2008>, <https://www.atmos-chem-phys.net/8/4947/2008/>, 2008a.

De Smedt, I., Müller, J.-F., Stavrakou, T., Van Der A, R., Eskes, H., and Van Roozendael, M.: Twelve years of global observations of formaldehyde in the troposphere using GOME and SCIAMACHY sensors, *Atmos. Chem. Phys.*, 2008b.

745 De Smedt, I., Stavrakou, T., Hendrick, F., Danckaert, T., Vlemmix, T., Pinardi, G., Theys, N., Lerot, C., Gielen, C., Vigouroux, C., Hermans, C., Fayt, C., Veefkind, P., Müller, J.-F., and Van Roozendael, M.: Diurnal, seasonal and long-term variations of global formaldehyde columns inferred from combined OMI and GOME-2 observations, *Atmospheric Chemistry and Physics*, 15, 12 519–12 545, <https://doi.org/10.5194/acp-15-12519-2015>, <https://www.atmos-chem-phys.net/15/12519/2015/>, 2015.

750 Deutschmann, T., Beirle, S., Frieß, U., Grzegorski, M., Kern, C., Kritten, L., Platt, U., Pukite, J., Wagner, T., Werner, B., and Pfeilsticker, K.: The Monte Carlo Atmospheric Radiative Transfer Model McArtim: Introduction and Validation of Jacobians and 3D Features, *Journal of Quantitative Spectroscopy and Radiative Transfer*, 112, 1119–1137, 2011.

DiGangi, J., Henry, S., Kammerath, A., Boyle, E., Kaser, L., Schnitzhofer, R., Graus, M., Turnipseed, A., Park, J., Weber, R., et al.: Observations of glyoxal and formaldehyde as metrics for the anthropogenic impact on rural photochemistry, *Atmospheric Chemistry and Physics*, 12, 9529–9543, 2012.

750 Dufour, G., Szopa, S., Barkley, M. P., Boone, C. D., Perrin, A., Palmer, P. I., and Bernath, P. F.: Global upper-tropospheric formaldehyde: seasonal cycles observed by the ACE-FTS satellite instrument, *Atmospheric Chemistry and Physics*, 9, 3893–3910, <https://doi.org/10.5194/acp-9-3893-2009>, <https://www.atmos-chem-phys.net/9/3893/2009/>, 2009.

Finlayson-Pitts, B. J. and Pitts, J. N.: *Photochemistry of Important Atmospheric Species*, Academic Press, San 755 Diego, <https://doi.org/https://doi.org/10.1016/B978-012257060-5/50006-X>, <http://www.sciencedirect.com/science/article/pii/B978012257060550006X>, 2000.

Fortems-Cheiney, A., Chevallier, F., Pison, I., Bousquet, P., Saunois, M., Szopa, S., Cressot, C., Kurosu, T. P., Chance, K., and Fried, A.: The formaldehyde budget as seen by a global-scale multi-constraint and multi-species inversion system, *Atmospheric Chemistry and Physics*, 12, 6699–6721, <https://doi.org/10.5194/acp-12-6699-2012>, <https://www.atmos-chem-phys.net/12/6699/2012/>, 2012.

760 Fried, A., Olson, J. R., Walega, J. G., Crawford, J. H., Chen, G., Weibring, P., Richter, D., Roller, C., Tittel, F., Porter, M., Fuelberg, H., Halland, J., Bertram, T. H., Cohen, R. C., Pickering, K., Heikes, B. G., Snow, J. A., Shen, H., O'Sullivan, D. W., Brune, W. H., Ren, X., Blake, D. R., Blake, N., Sachse, G., Diskin, G. S., Podolske, J., Vay, S. A., Shetter, R. E., Hall, S. R., Anderson, B. E., Thornhill, L., Clarke, A. D., McNaughton, C. S., Singh, H. B., Avery, M. A., Huey, G., Kim, S., and Millet, D. B.: Role of convection in redistributing formaldehyde to the upper troposphere over North America and the North Atlantic during the summer 2004 INTEX campaign, *Journal of Geophysical Research: Atmospheres*, 113, <https://doi.org/10.1029/2007JD009760>, <https://agupubs.onlinelibrary.wiley.com/doi/abs/10.1029/2007JD009760>, 2008.

765 Frost, G., Fried, A., Lee, Y.-N., Wert, B., Henry, B., Drummond, J., Evans, M., Fehsenfeld, F. C., Goldan, P., Holloway, J., et al.: Comparisons of box model calculations and measurements of formaldehyde from the 1997 North Atlantic Regional Experiment, *Journal of Geophysical Research: Atmospheres*, 107, ACH–3, 2002.

770 Fu, D., Millet, D. B., Wells, K. C., Payne, V. H., Yu, S., Guenther, A., and Eldering, A.: Direct retrieval of isoprene from satellite-based infrared measurements, *Nature Communications*, 10, 1–12, 2019.

Fu, T.-M., Jacob, D. J., Wittrock, F., Burrows, J. P., Vrekoussis, M., and Henze, D. K.: Global budgets of atmospheric glyoxal and methylglyoxal, and implications for formation of secondary organic aerosols, *Journal of Geophysical Research: Atmospheres*, 113, , <https://doi.org/10.1029/2007JD009505>, <https://agupubs.onlinelibrary.wiley.com/doi/abs/10.1029/2007JD009505>, 2008.

775 Gerbig, C., Kley, D., Volz-Thomas, A., Kent, J., Dewey, K., and McKenna, D. S.: Fast response resonance fluorescence CO measurements aboard the C-130: Instrument characterization and measurements made during North Atlantic Regional Experiment 1993, *Journal of Geophysical Research: Atmospheres*, 101, 29 229–29 238, 1996.

Gerbig, C., Schmitgen, S., Kley, D., Volz-Thomas, A., Dewey, K., and Haaks, D.: An improved fast-response vacuum-UV resonance fluorescence CO instrument, *Journal of Geophysical Research: Atmospheres*, 104, 1699–1704, 1999.

780 Gu, D., Guenther, A. B., Shilling, J. E., Yu, H., Huang, M., Zhao, C., Yang, Q., Martin, S. T., Artaxo, P., Kim, S., et al.: Airborne observations reveal elevational gradient in tropical forest isoprene emissions, *Nature Communications*, 8, 1–7, 2017.

Hays, M. D., Geron, C. D., Linna, K. J., Smith, N. D., and Schauer, J. J.: Speciation of Gas-Phase and Fine Particle Emissions from Burning of Foliar Fuels, *Environmental Science & Technology*, 36, 2281–2295, <https://doi.org/10.1021/es0111683>, <https://doi.org/10.1021/es0111683>, pMID: 12075778, 2002.

785 Helmig, D., Balsley, B., Davis, K., Kuck, L. R., Jensen, M., Bognar, J., Smith Jr, T., Arrieta, R. V., Rodríguez, R., and Birks, J. W.: Vertical profiling and determination of landscape fluxes of biogenic nonmethane hydrocarbons within the planetary boundary layer in the Peruvian Amazon, *Journal of Geophysical Research: Atmospheres*, 103, 25 519–25 532, 1998.

Horowitz, A., Meller, R., and Moortgat, G. K.: The UV–VIS absorption cross sections of the  $\alpha$ -dicarbonyl compounds: pyruvic acid, biacetyl and glyoxal, *Journal of Photochemistry and Photobiology A: Chemistry*, 146, 19–27, 2001.

790 Hüneke, T.: The scaling method applied to HALO measurements: Inferring absolute trace gas concentrations from airborne limb spectroscopy under all sky conditions, Phd, University of Heidelberg, Heidelberg, Germany, [http://www.iup.uni-heidelberg.de/institut/forschung/groups/atmosphere/stratosphere/Theses/](http://www.iup.uni-heidelberg.de/institut/forschung/groups/atmosphere/stratosphere/Theeses/), 2016.

Hüneke, T., Aderhold, O.-A., Bounin, J., Dorf, M., Gentry, E., Grossmann, K., Groß, J.-U., Hoor, P., Jöckel, P., Kenntner, M., Knapp, M., Knecht, M., Lörks, D., Ludmann, S., Matthes, S., Raecke, R., Reichert, M., Weimar, J., Werner, B., Zahn, A., Ziereis, H., and 795 Pfeilsticker, K.: The novel HALO mini-DOAS instrument: inferring trace gas concentrations from airborne UV/visible limb spectroscopy under all skies using the scaling method, *Atmospheric Measurement Techniques*, 10, 4209–4234, <https://doi.org/10.5194/amt-10-4209-2017>, <https://www.atmos-meas-tech.net/10/4209/2017/>, 2017.

Huntrieser, H., Lichtenstern, M., Scheibe, M., Aufmhoff, H., Schlager, H., Pucik, T., Minikin, A., Weinzierl, B., Heimerl, K., Fütterer, D., et al.: On the origin of pronounced O<sub>3</sub> gradients in the thunderstorm outflow region during DC3, *Journal of Geophysical Research: Atmospheres*, 121, 6600–6637, 2016a.

800 Huntrieser, H., Lichtenstern, M., Scheibe, M., Aufmhoff, H., Schlager, H., Pucik, T., Minikin, A., Weinzierl, B., Heimerl, K., Pollack, I., et al.: Injection of lightning-produced NO<sub>x</sub>, water vapor, wildfire emissions, and stratospheric air to the UT/LS as observed from DC3 measurements, *Journal of Geophysical Research: Atmospheres*, 121, 6638–6668, 2016b.

Jaeglé, L., Jacob, D. J., Wennberg, P. O., Spivakovsky, C. M., Hanisco, T. F., Lanzendorf, E. J., Hintsa, E. J., Fahey, D. W., Keim, E. R., 805 Proffitt, M. H., Atlas, E. L., Flocke, F., Schaufler, S., McElroy, C. T., Midwinter, C., Pfister, L., and Wilson, J. C.: Observed OH and HO<sub>2</sub> in the upper troposphere suggest a major source from convective injection of peroxides, *Geophysical Research Letters*, 24, 3181–3184, <https://doi.org/10.1029/97GL03004>, <https://agupubs.onlinelibrary.wiley.com/doi/abs/10.1029/97GL03004>, 1997.

Kaiser, J., Wolfe, G. M., Min, K. E., Brown, S. S., Miller, C. C., Jacob, D. J., deGouw, J. A., Graus, M., Hanisco, T. F., Holloway, J., Peischl, J., Pollack, I. B., Ryerson, T. B., Warneke, C., Washenfelder, R. A., and Keutsch, F. N.: Reassessing the ratio of glyoxal to formaldehyde 810 as an indicator of hydrocarbon precursor speciation, *Atmospheric Chemistry and Physics*, 15, 7571–7583, <https://doi.org/10.5194/acp-15-7571-2015>, <https://www.atmos-chem-phys.net/15/7571/2015/>, 2015.

Kesselmeier, J., Kuhn, U., Wolf, A., Andreae, M., Cicioli, P., Brancaleoni, E., Frattoni, M., Guenther, A., Greenberg, J., Vasconcellos, P. D. C., et al.: Atmospheric volatile organic compounds (VOC) at a remote tropical forest site in central Amazonia, *Atmospheric Environment*, 34, 4063–4072, 2000.

815 Knecht, M.: Simulation of radiative field modification due to tropical clouds, Master thesis, Institut für Umweltphysik, Universität Heidelberg, [http://www.iup.uni-heidelberg.de/institut/forschung/groups/atmosphere/stratosphere/publications/pdf/MA\\_thesis\\_Matthias\\_final.pdf](http://www.iup.uni-heidelberg.de/institut/forschung/groups/atmosphere/stratosphere/publications/pdf/MA_thesis_Matthias_final.pdf), 2015.

Knote, C., Hodzic, A., Jimenez, J. L., Volkamer, R., Orlando, J. J., Baidar, S., Brioude, J., Fast, J., Gentner, D. R., Goldstein, A. H., Hayes, P. L., Knighton, W. B., Oetjen, H., Setyan, A., Stark, H., Thalman, R., Tyndall, G., Washenfelder, R., Waxman, E., and Zhang, Q.: 820 Simulation of semi-explicit mechanisms of SOA formation from glyoxal in aerosol in a 3-D model, *Atmospheric Chemistry and Physics*, 14, 6213–6239, <https://doi.org/10.5194/acp-14-6213-2014>, <https://www.atmos-chem-phys.net/14/6213/2014/>, 2014.

Koch, S. and Moortgat, G. K.: Photochemistry of methylglyoxal in the vapor phase, *The Journal of Physical Chemistry A*, 102, 9142–9153, 1998.

Krysztofiak, G., Catoire, V., Hamer, P. D., Marecal, V., Robert, C., Engel, A., Bönisch, H., Grossmann, K., Quack, B., Atlas, E., et al.: Evidence of convective transport in tropical West Pacific region during SHIVA experiment, *Atmospheric Science Letters*, 19, e798, 2018.

Kuhn, U., Andreae, M., Ammann, C., Araújo, A., Brancaleoni, E., Ciccioli, P., Dindorf, T., Frattoni, M., Gatti, L., Ganzeveld, L., et al.: Isoprene and monoterpene fluxes from Central Amazonian rainforest inferred from tower-based and airborne measurements, and implications on the atmospheric chemistry and the local carbon budget, *Atmospheric Chemistry and Physics*, 7, 2855–2879, 2007.

Lawson, S. J., Selleck, P. W., Galbally, I. E., Keywood, M. D., Harvey, M. J., Lerot, C., Helmig, D., and Ristovski, Z.: Seasonal in situ observations of glyoxal and methylglyoxal over the temperate oceans of the Southern Hemisphere, *Atmospheric Chemistry and Physics*, 15, 223–240, <https://doi.org/10.5194/acp-15-223-2015>, <https://www.atmos-chem-phys.net/15/223/2015/>, 2015.

Lee, Y.-N., Zhou, X., Kleinman, L., Nunnermacker, L., Springston, S., Daum, P., Newman, L., Keigley, W., Holdren, M., Spicer, C., et al.: Atmospheric chemistry and distribution of formaldehyde and several multioxxygenated carbonyl compounds during the 1995 Nashville/Middle Tennessee Ozone Study, *Journal of Geophysical Research: Atmospheres*, 103, 22 449–22 462, 1998.

Lerot, C., Stavrakou, T., De Smedt, I., Müller, J.-F., and Van Roozendael, M.: Glyoxal vertical columns from GOME-2 backscattered light measurements and comparisons with a global model, *Atmospheric Chemistry and Physics*, 10, 12 059–12 072, <https://doi.org/10.5194/acp-10-12059-2010>, <https://www.atmos-chem-phys.net/10/12059/2010/>, 2010.

Li, J., Mao, J., Min, K.-E., Washenfelder, R. A., Brown, S. S., Kaiser, J., Keutsch, F. N., Volkamer, R., Wolfe, G. M., Hanisco, T. F., Pollack, I. B., Ryerson, T. B., Graus, M., Gilman, J. B., Lerner, B. M., Warneke, C., Gouw, J. A., Middlebrook, A. M., Liao, J., Welti, A., Henderson, B. H., McNeill, V. F., Hall, S. R., Ullmann, K., Donner, L. J., Paulot, F., and Horowitz, L. W.: Observational constraints on glyoxal production from isoprene oxidation and its contribution to organic aerosol over the Southeast United States, *Journal of Geophysical Research: Atmospheres*, 121, 9849–9861, <https://doi.org/10.1002/2016JD025331>, <https://agupubs.onlinelibrary.wiley.com/doi/abs/10.1002/2016JD025331>, 2016.

Liao, J., Hanisco, T. F., Wolfe, G. M., St. Clair, J., Jimenez, J. L., Campuzano-Jost, P., Nault, B. A., Fried, A., Marais, E. A., Gonzalez Abad, G., Chance, K., Jethva, H. T., Ryerson, T. B., Warneke, C., and Wisthaler, A.: Towards a satellite formaldehyde – in situ hybrid estimate for organic aerosol abundance, *Atmospheric Chemistry and Physics*, 19, 2765–2785, <https://doi.org/10.5194/acp-19-2765-2019>, <https://www.atmos-chem-phys.net/19/2765/2019/>, 2019.

Lim, C. Y., Hagan, D. H., Coggon, M. M., Koss, A. R., Sekimoto, K., de Gouw, J., Warneke, C., Cappa, C. D., and Kroll, J. H.: Secondary organic aerosol formation from biomass burning emissions, *Atmos. Chem. Phys.*, 19, 12 797–12 809, 2019.

Liu, L., Flatøy, F., Ordóñez, C., Braathen, G., Hak, C., Junkermann, W., Andreani-Aksoyoglu, S., Mellqvist, J., Galle, B., Prévôt, A., et al.: Photochemical modelling in the Po basin with focus on formaldehyde and ozone, 2007.

MacDonald, S., Oetjen, H., Mahajan, A., Whalley, L., Edwards, P., Heard, D., Jones, C., and Plane, J.: DOAS measurements of formaldehyde and glyoxal above a south-east Asian tropical rainforest, *Atmospheric Chemistry and Physics*, 12, 5949–5962, 2012.

McDonald, J. D., Zielinska, B., Fujita, E. M., Sagebiel, J. C., Chow, J. C., and Watson, J. G.: Fine Particle and Gaseous Emission Rates from Residential Wood Combustion, *Environmental Science & Technology*, 34, 2080–2091, <https://doi.org/10.1021/es9909632>, <https://doi.org/10.1021/es9909632>, 2000a.

McDonald, J. D., Zielinska, B., Fujita, E. M., Sagebiel, J. C., Chow, J. C., and Watson, J. G.: Fine particle and gaseous emission rates from residential wood combustion, *Environmental Science & Technology*, 34, 2080–2091, 2000b.

860 Meller, R., Raber, W., Crowley, J., Jenkin, M., and Moortgat, G.: The UV-visible absorption spectrum of methylglyoxal, *Journal of Photochemistry and Photobiology A: Chemistry*, 62, 163–171, 1991.

Michoud, V., Sauvage, S., Léonardis, T., Fronval, I., Kukui, A., Locoge, N., and Dusanter, S.: Field measurements of methylglyoxal using proton transfer reaction time-of-flight mass spectrometry and comparison to the DNPH–HPLC–UV method, *Atmospheric Measurement Techniques*, 11, 5729–5740, <https://doi.org/10.5194/amt-11-5729-2018>, <https://www.atmos-meas-tech.net/11/5729/2018/>, 2018.

865 Müller, M., Anderson, B. E., Beyersdorf, A. J., Crawford, J. H., Diskin, G. S., Eichler, P., Fried, A., Keutsch, F. N., Mikoviny, T., Thornhill, K. L., et al.: In situ measurements and modeling of reactive trace gases in a small biomass burning plume, *Atmospheric Chemistry and Physics*, 16, 3813–3824, 2016.

Myrokefalitakis, S., Vrekoussis, M., Tsigaridis, K., Wittrock, F., Richter, A., Brühl, C., Volkamer, R., Burrows, J., and Kanakidou, M.: The influence of natural and anthropogenic secondary sources on the glyoxal global distribution, 2008a.

870 Myrokefalitakis, S., Vrekoussis, M., Tsigaridis, K., Wittrock, F., Richter, A., Brühl, C., Volkamer, R., Burrows, J. P., and Kanakidou, M.: The influence of natural and anthropogenic secondary sources on the glyoxal global distribution, *Atmospheric Chemistry and Physics*, 8, 4965–4981, <https://doi.org/10.5194/acp-8-4965-2008>, <https://www.atmos-chem-phys.net/8/4965/2008/>, 2008b.

Pang, X., Lewis, A., and Hamilton, J.: Determination of airborne carbonyls via pentafluorophenylhydrazine derivatisation by GC-MS and its comparison with HPLC method, *Talanta*, 85, 406–414, <https://doi.org/10.1016/j.talanta.2011.03.072>, 2011.

875 Peters, E., Wittrock, F., Großmann, K., Frieß, U., Richter, A., and Burrows, J. P.: Formaldehyde and nitrogen dioxide over the remote western Pacific Ocean: SCIAMACHY and GOME-2 validation using ship-based MAX-DOAS observations, *Atmospheric Chemistry and Physics*, 12, 11 179–11 197, <https://doi.org/10.5194/acp-12-11179-2012>, <https://www.atmos-chem-phys.net/12/11179/2012/>, 2012.

Platt, U. and Stutz, J.: Differential Optical Absorption Spectroscopy (DOAS), Principle and Applications, no. ISBN 978-3-642-05946-9 in ISSN 1610-1677, Springer Verlag, Heidelberg, <https://doi.org/10.1007/978-3-540-75776-40>, 2008.

880 Pöschl, U., Williams, J., Hoor, P., Fischer, H., Crutzen, P., Warneke, C., Holzinger, R., Hansel, A., Jordan, A., Lindinger, W., et al.: High acetone concentrations throughout the 0–12 km altitude range over the tropical rainforest in Surinam, *Journal of atmospheric chemistry*, 38, 115–132, 2001.

Rizzo, L., Artaxo, P., Karl, T., Guenther, A., and Greenberg, J.: Aerosol properties, in-canopy gradients, turbulent fluxes and VOC concentrations at a pristine forest site in Amazonia, *Atmospheric Environment*, 44, 503–511, 2010.

Rolph, G., Stein, A., and Stunder, B.: Real-time environmental applications and display system: READY, *Environmental Modelling & Software*, 95, 210–228, 2017.

885 Rothman, L. S., Gordon, I. E., Barbe, A., Benner, D. C., Bernath, P. F., Birk, M., Boudon, V., Brown, L. R., Campargue, A., and Champion et al., J.-P.: The HITRAN 2008 molecular spectroscopic database, *Journal of Quantitative Spectroscopy and Radiative Transfer*, 110, 533–572, 2009.

Sander, R.: Compilation of Henry's law constants (version 4.0) for water as solvent, *Atmos. Chem. Phys*, 15, 4399–4981, 2015.

890 Schulz, C., Schneider, J., Amorim Holanda, B., Appel, O., Costa, A., Sá, S. S. d., Dreiling, V., Fütterer, D., Jurkat-Witschas, T., Klimach, T., et al.: Aircraft-based observations of isoprene-epoxydiol-derived secondary organic aerosol (IEPOX-SOA) in the tropical upper troposphere over the Amazon region, *Atmospheric Chemistry and Physics*, 18, 14 979–15 001, 2018.

Seinfeld, J. H. and Pandis, S. N.: Atmospheric chemistry and physics : from air pollution to climate change, ISBN 9780471720171, J. Wiley & Sons, Inc., third edition edn., 2013.

895 Serdyuchenko, A., Gorshelev, V., Weber, M., Chehade, W., and Burrows, J. P.: High spectral resolution ozone absorption cross-sections  
- Part 2: Temperature dependence, *Atmospheric Measurement Techniques*, 7, 625–636, <https://doi.org/10.5194/amt-7-625-2014>, <http://www.atmos-meas-tech.net/7/625/2014/>, 2014.

900 Stavrakou, T., Müller, J.-F., De Smedt, I., Van Roozendael, M., Kanakidou, M., Vrekoussis, M., Wittrock, F., Richter, A., and Burrows,  
Chemistry and Physics, 9, 8431–8446, <https://doi.org/10.5194/acp-9-8431-2009>, <https://www.atmos-chem-phys.net/9/8431/2009/>, 2009a.

905 Stavrakou, T., Müller, J.-F., De Smedt, I., Van Roozendael, M., van der Werf, G. R., Giglio, L., and Guenther, A.: Evaluating the performance  
of pyrogenic and biogenic emission inventories against one decade of space-based formaldehyde columns, *Atmospheric Chemistry and*  
*Physics*, 9, 1037–1060, <https://doi.org/10.5194/acp-9-1037-2009>, <https://www.atmos-chem-phys.net/9/1037/2009/>, 2009b.

910 Stavrakou, T., Müller, J.-F., Bauwens, M., De Smedt, I., Lerot, C., Van Roozendael, M., Coheur, P.-F., Clerbaux, C., Boersma, K. F., van der  
A, R., and Song, Y.: Substantial Underestimation of Post-Harvest Burning Emissions in the North China Plain Revealed by Multi-Species  
Space Observations, *Scientific Reports*, 6, <https://doi.org/10.1038/srep32307>, <https://doi.org/10.1038/srep32307>, 2016.

915 Steck, T., Glatthor, N., von Clarmann, T., Fischer, H., Flaud, J. M., Funke, B., Grabowski, U., Höpfner, M., Kellmann, S., Linden, A.,  
Perrin, A., and Stiller, G. P.: Retrieval of global upper tropospheric and stratospheric formaldehyde ( $\text{H}_2\text{CO}$ ) distributions from high-  
resolution MIPAS-Envisat spectra, *Atmospheric Chemistry and Physics*, 8, 463–470, <https://doi.org/10.5194/acp-8-463-2008>, <https://www.atmos-chem-phys.net/8/463/2008/>, 2008.

920 Stein, A., Draxler, R. R., Rolph, G. D., Stunder, B. J., Cohen, M., and Ngan, F.: NOAA's HYSPLIT atmospheric transport and dispersion  
modeling system, *Bulletin of the American Meteorological Society*, 96, 2059–2077, 2015.

925 Stockwell, C. E., Veres, P. R., Williams, J., and Yokelson, R. J.: Characterization of biomass burning emissions from cooking fires, peat,  
crop residue, and other fuels with high-resolution proton-transfer-reaction time-of-flight mass spectrometry, *Atmospheric Chemistry and*  
*Physics*, 15, 845–865, <https://doi.org/10.5194/acp-15-845-2015>, <https://www.atmos-chem-phys.net/15/845/2015/>, 2015.

930 Stutz, J., Werner, B., Spolaor, M., Scalone, L., Festa, J., Tsai, C., Cheung, R., Colosimo, S. F., Tricoli, U., Raecke, R., Hossaini, R., Chipper-  
field, M. P., Feng, W., Gao, R.-S., Hints, E. J., Elkins, J. W., Moore, F. L., Daube, B., Pittman, J., Wofsy, S., and Pfeilsticker, K.: A new Dif-  
ferential Optical Absorption Spectroscopy instrument to study atmospheric chemistry from a high-altitude unmanned aircraft, *Atmospheric*  
*Measurement Techniques*, 10, 1017 – 1042, <https://doi.org/10.5194/amt-10-1017-2017>, <http://www.atmos-meas-tech.net/10/1017/2017/>,  
2017.

Tadić, J., Moortgat, G. K., and Wirtz, K.: Photolysis of glyoxal in air, *Journal of Photochemistry and Photobiology A: Chemistry*, 177,  
116–124, 2006.

935 Taraborrelli, D., Cabrera-Perez, D., Bacer, S., Gromov, S., Lelieveld, J., Sander, R., and Pozzer, A.: Influence of aromatics on tropospheric  
gas-phase composition, *Atmospheric Chemistry and Physics Discussions*, 2020, 1–26, <https://doi.org/10.5194/acp-2020-461>, <http://www.atmos-chem-phys-discuss.net/acp-2020-461/>, 2020.

940 Thalman, R. and Volkamer, R.: Inherent calibration of a blue LED-CE-DOAS instrument to measure iodine oxide, glyoxal, methyl gly-  
oxal, nitrogen dioxide, water vapour and aerosol extinction in open cavity mode, *Atmospheric Measurement Techniques*, 3, 1797–1814,  
<https://doi.org/10.5194/amt-3-1797-2010>, <https://www.atmos-meas-tech.net/3/1797/2010/>, 2010.

945 Thalman, R. and Volkamer, R.: Temperature dependent absorption cross-sections of O<sub>2</sub>-O<sub>2</sub> collision pairs between 340 and 630 nm and at  
atmospherically relevant pressure, *Phys. Chem. Chem. Phys.*, 15, 15 371–15 381, <https://doi.org/10.1039/C3CP50968K>, <http://dx.doi.org/10.1039/C3CP50968K>, 2013.

Thalman, R., Baeza-Romero, M. T., Ball, S. M., Borrás, E., Daniels, M. J. S., Goodall, I. C. A., Henry, S. B., Karl, T., Keutsch, F. N., Kim, S., Mak, J., Monks, P. S., Muñoz, A., Orlando, J., Peppe, S., Rickard, A. R., Ródenas, M., Sánchez, P., Seco, R., Su, L., Tyndall, G., Vázquez, M., Vera, T., Waxman, E., and Volkamer, R.: Instrument intercomparison of glyoxal, methyl glyoxal and NO<sub>2</sub> under simulated atmospheric conditions, *Atmospheric Measurement Techniques*, 8, 1835–1862, <https://doi.org/10.5194/amt-8-1835-2015>, <https://www.atmos-meas-tech.net/8/1835/2015/>, 2015.

Volkamer, R., Molina, L. T., Molina, M. J., Shirley, T., and Brune, W. H.: DOAS measurement of glyoxal as an indicator for fast VOC chemistry in urban air, *Geophysical Research Letters*, 32, <https://doi.org/10.1029/2005GL022616>, <https://agupubs.onlinelibrary.wiley.com/doi/abs/10.1029/2005GL022616>, 2005a.

Volkamer, R., Spietz, P., Burrows, J., and Platt, U.: High-resolution absorption cross-section of glyoxal in the UV–vis and IR spectral ranges, *Journal of Photochemistry and Photobiology A: Chemistry*, 172, 35–46, 2005b.

Vrekoussis, M., Wittrock, F., Richter, A., and Burrows, J.: Temporal and spatial variability of glyoxal as observed from space, *Atmospheric Chemistry and Physics*, 9, 4485–4504, 2009.

Vrekoussis, M., Wittrock, F., Richter, A., and Burrows, J. P.: GOME-2 observations of oxygenated VOCs: what can we learn from the ratio glyoxal to formaldehyde on a global scale?, *Atmospheric Chemistry and Physics*, 10, 10 145–10 160, <https://doi.org/10.5194/acp-10-10145-2010>, <https://www.atmos-chem-phys.net/10/10145/2010/>, 2010.

Wagner, V., von Glasow, R., Fischer, H., and Crutzen, P. J.: Are CH<sub>2</sub>O measurements in the marine boundary layer suitable for testing the current understanding of CH<sub>4</sub> photooxidation?: A model study, *Journal of Geophysical Research: Atmospheres*, 107, ACH 3–1–ACH 3–14, <https://doi.org/10.1029/2001JD000722>, <https://agupubs.onlinelibrary.wiley.com/doi/abs/10.1029/2001JD000722>, 2002.

Wang, M., Shao, M., Chen, W., Yuan, B., Lu, S., Zhang, Q., Zeng, L., and Wang, Q.: Atmospheric nanoparticles formed from heterogeneous reactions of organics, *Nature Geoscience*, 3, 238–24, <https://doi.org/10.1038/ngeo778>, <https://www.nature.com/articles/ngeo778>, 2010.

Warneke, C., Karl, T., Judmaier, H., Hansel, A., Jordan, A., Lindinger, W., and Crutzen, P. J.: Acetone, methanol, and other partially oxidized volatile organic emissions from dead plant matter by abiological processes: Significance for atmospheric HO<sub>x</sub> chemistry, *Global Biogeochemical Cycles*, 13, 9–17, 1999.

Warneke, C., Holzinger, R., Hansel, A., Jordan, A., Lindinger, W., Pöschl, U., Williams, J., Hoor, P., Fischer, H., Crutzen, P., et al.: Isoprene and its oxidation products methyl vinyl ketone, methacrolein, and isoprene related peroxides measured online over the tropical rain forest of Surinam in March 1998, *Journal of Atmospheric Chemistry*, 38, 167–185, 2001.

Wendisch, M., Pöschl, U., Andreae, M. O., Machado, L. A. T., Albrecht, R., Schlager, H., Rosenfeld, D., Martin, S. T., Abdelmonem, A., Afchine, A., Araújo, A. C., Artaxo, P., Aufmhoff, H., Barbosa, H. M. J., Borrman, S., Braga, R., Buchholz, B., Cecchini, M. A., Costa, A., Curtius, J., Dollner, M., Dorf, M., Dreiling, V., Ebert, V., Ehrlich, A., Ewald, F., Fisch, G., Fix, A., Frank, F., Fütterer, D., Heckl, C., Heidelberg, F., Hüneke, T., Jäkel, E., Järvinen, E., Jurkat, T., Kanter, S., Kästner, U., Kenntner, M., Kesselmeier, J., Klimach, T., Knecht, M., Kohl, R., Kölling, T., Krämer, M., Krüger, M., Krisna, T. C., Lavric, J. V., Longo, K., Mahnke, C., Manzi, A. O., Mayer, B., Mertes, S., Minikin, A., Molleker, S., Münch, S., Nillius, B., Pfeilsticker, K., Pöhlker, C., Roiger, A., Rose, D., Rosenow, D., Sauer, D., Schnaiter, M., Schneider, J., Schulz, C., de Souza, R. A. F., Spanu, A., Stock, P., Vila, D., Voigt, C., Walser, A., Walter, D., Weigel, R., Weinzierl, B., Werner, F., Yamasoe, M. A., Ziereis, H., Zinner, T., and Zöger, M.: ACRIDICON–CHUVA Campaign: Studying Tropical Deep Convective Clouds and Precipitation over Amazonia Using the New German Research Aircraft HALO, *Bulletin of the American Meteorological Society*, 97, 1885–1908, <https://doi.org/10.1175/BAMS-D-14-00255.1>, <https://doi.org/10.1175/BAMS-D-14-00255.1>, 2016.

Wennberg, P. O., Bates, K. H., Crounse, J. D., Dodson, L. G., McVay, R. C., Mertens, L. A., Nguyen, T. B., Praske, E., Schwantes, R. H., Smarte, M. D., St Clair, J. M., Teng, A. P., Zhang, X., and Seinfeld, J. H.: Gas-Phase Reactions of Isoprene and Its Major Oxidation Prod-

970     ucts, *Chemical Reviews*, 118, 3337–3390, <https://doi.org/10.1021/acs.chemrev.7b00439>, <https://doi.org/10.1021/acs.chemrev.7b00439>, pMID: 29522327, 2018.

975     Werner, B., Stutz, J., Spolaor, M., Scalone, L., Raecke, R., Festa, J., Colosimo, S. F., Cheung, R., Tsai, C., Hossaini, R., Chipperfield, M. P., Taverna, G. S., Feng, W., Elkins, J. W., Fahey, D. W., Gao, R.-S., Hintsas, E. J., Thornberry, T. D., Moore, F. L., Navarro, M. A., Atlas, E., Daube, B. C., Pittman, J., Wofsy, S., and Pfeilsticker, K.: Probing the subtropical lowermost stratosphere and the tropical upper troposphere and tropopause layer for inorganic bromine, *Atmospheric Chemistry and Physics*, 17, 1161–1186, <https://doi.org/10.5194/acp-17-1161-2017>, <http://www.atmos-chem-phys.net/17/1161/2017/>, 2017.

980     Williamson, C. J., Kupc, A., Axisa, D., Bilsback, K. R., Bui, T., Campuzano-Jost, P., Dollner, M., Froyd, K. D., Hodshire, A. L., Jimenez, J. L., et al.: A large source of cloud condensation nuclei from new particle formation in the tropics, *Nature*, 574, 399–403, 2019.

985     Wittrock, F., Richter, A., Oetjen, H., Burrows, J. P., Kanakidou, M., Myriokefalitakis, S., Volkamer, R., Beirle, S., Platt, U., and Wagner, T.: Simultaneous global observations of glyoxal and formaldehyde from space, *Geophysical Research Letters*, 33, , <https://doi.org/10.1029/2006GL026310>, <https://agupubs.onlinelibrary.wiley.com/doi/abs/10.1029/2006GL026310>, 2006.

990     Zarzana, K. J., Min, K.-E., Washenfelder, R. A., Kaiser, J., Krawiec-Thayer, M., Peischl, J., Neuman, J. A., Nowak, J. B., Wagner, N. L., Dubè, W. P., St. Clair, J. M., Wolfe, G. M., Hanisco, T. F., Keutsch, F. N., Ryerson, T. B., and Brown, S. S.: Emissions of Glyoxal and Other Carbonyl Compounds from Agricultural Biomass Burning Plumes Sampled by Aircraft, *Environmental Science & Technology*, 51, 11 761–11 770, <https://doi.org/10.1021/acs.est.7b03517>, <https://doi.org/10.1021/acs.est.7b03517>, pMID: 28976736, 2017.

995     Zarzana, K. J., Selimovic, V., Koss, A. R., Sekimoto, K., Coggon, M. M., Yuan, B., Dubé, W. P., Yokelson, R. J., Warneke, C., de Gouw, J. A., Roberts, J. M., and Brown, S. S.: Primary emissions of glyoxal and methylglyoxal from laboratory measurements of open biomass burning, *Atmospheric Chemistry and Physics*, 18, 15 451–15 470, <https://doi.org/10.5194/acp-18-15451-2018>, <https://www.atmos-chem-phys.net/18/15451/2018/>, 2018.